

1 **Atmospheric chemistry of nitrogenous aerosols in Northeast Asia:**
2 **biological sources and secondary formation**

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11 **Abstract.** To better understand the sources of nitrogenous aerosols, particularly
12 water-soluble organic nitrogen (WSO_N) and water-insoluble organic nitrogen (WIO_N), in
13 Northeast Asia, we measured total nitrogen (TN) and water-soluble total nitrogen (WSTN) as
14 well as nitrogen isotope ratios ($\delta^{15}\text{N}$) of TN ($\delta^{15}\text{N}_{\text{TN}}$) and WSTN ($\delta^{15}\text{N}_{\text{WSTN}}$) in the total
15 suspended particles (TSP) collected from Sapporo, northern Japan for one-year period. In
16 general, WIO_N was more abundant ($126 \pm 117 \text{ ng m}^{-3}$) whereas WSO_N ($89.7 \pm 80.6 \text{ ng m}^{-3}$),
17 accounting for $14 \pm 11\%$ and $9.2 \pm 7.3\%$ of TN, respectively. WSO_N peaked in late autumn
18 to winter (maximum 288 ng m^{-3}) and WIO_N peaked in mid spring to early summer (454 ng
19 m^{-3}). $\delta^{15}\text{N}_{\text{TN}}$ ($21.9 \pm 4.1\text{‰}$) and $\delta^{15}\text{N}_{\text{WSTN}}$ ($25.8 \pm 8.2\text{‰}$) showed peaks in summer with
20 relatively high ratios in late autumn. Based on the seasonal variations of WSO_N and WIO_N
21 together with organic tracers, fossil fuel combustion and biomass burning are found to be two
22 major sources of WSO_N whereas emissions of biological particles and secondary formation
23 by reactions of biogenic secondary organic species (carbonyls) with NH_3 are suggested as
24 important source of WIO_N. The seasonality of $\delta^{15}\text{N}_{\text{TN}}$ and $\delta^{15}\text{N}_{\text{WSTN}}$, together with the
25 comparisons to literature values, implies that chemical aging (including gas/particle
26 partitioning) and biomass burning are the causes of the enhanced values in summer and
27 autumn, respectively. This study demonstrates that contributions of aerosol N from fossil fuel
28 combustion and biomass burning dominate in autumn/winter whereas emission of terrestrial
29 biological particles and secondary formation from biogenic hydrocarbons and subsequent
30 chemical aging in the atmosphere are important in spring/summer in Northeast Asia.

31 **1 Introduction**

32 In East Asia, high loading of aerosol nitrogen (N) and its significant deposition onto the Earth
33 surface are associated with enhanced agricultural usage of nitrogenous fertilizer: 40% of
34 vegetation receives nitrogen deposition in excess of $1000 \text{ mg-N m}^{-2} \text{ yr}^{-1}$ (threshold level) and
35 even more (~50%) in Japan (Dentener et al., 2006; Liu et al., 2013; Morino et al., 2011).
36 Emissions of NH_3 and NO_x to the air have an impact on atmospheric chemistry, visibility
37 impairment, human health and the Earth's climate system (Rollins et al., 2012; Vitousek et al.,
38 1997). Further, N deposition over the Earth surface has large influence on terrestrial and
39 aquatic ecosystems via the changes in the global biogeochemical cycles (Duce et al., 2008;
40 Galloway et al., 2004; Kim et al., 2014). Most studies of aerosol N in East Asia and other
41 regions in the world have been focused on inorganic species (NO_3^- and NH_4^+) (Dentener et al.,
42 2006; Duce et al., 2008; Liu et al., 2013). Although organic nitrogen (ON) represents a
43 significant fraction (up to 80%) of total aerosol N and may play a critical role in
44 biogeochemical cycles, ON has rarely been studied (Cape et al., 2011; Matsumoto et al., 2014;
45 Zhang et al., 2008).

46 Aerosol ON is comprised of a wide range of nitrogenous compounds from semi-volatile
47 amines to proteins and macromolecules (Cape et al., 2011; Laskin et al., 2009; Wang et al.,
48 2010). In addition to biomass burning, the primary sources including biological particles
49 emitted from soil, vegetation, pollen, bacteria and the ocean surface are important (Cape et al.,
50 2011; Jickells et al., 2013; Miyazaki et al., 2014; Neff et al., 2002). ON can also be produced
51 in the atmosphere via the reactions of NO_3 radical or NH_3 with biogenic/anthropogenic
52 volatile organic compounds (VOCs) followed by gas-to-particle conversion (Cornell et al.,
53 2001; Miyazaki et al., 2014; Pratt et al., 2012). Recently, oxidation of isoprene and
54 monoterpenes by NO_3 is considered as a substantial source of organic nitrates and secondary
55 organic aerosols (SOA) (Fry et al., 2009; Pratt et al., 2012; Rollins et al., 2009; Rollins et al.,

2012). Further, chemical aging of SOA derived from both biogenic and anthropogenic precursors in the presence of NH_3 in gas phase and NH_4^+ salts in aqueous phase can produce light-absorbing nitrogenous compounds (defined as brown carbon) (Bones et al., 2010; Updyke et al., 2012).

However, the relative importance of anthropogenic and biogenic emissions including biomass burning is largely unknown and secondary formation of ON has been poorly characterized (Cape et al., 2011; Jickells et al., 2013; Kanakidou et al., 2012) and thus land-atmosphere interactions of aerosol N are not fully understood yet because of limited measurements of aerosol ON. Previous studies of ON have focused on water-soluble ON (WSO_N) but very little is known about the abundances and sources of water-insoluble ON (WIO_N) particularly in continental aerosols. Interestingly, WIO_N is much more abundant (on average $34.0 \text{ nmol N m}^{-3}$) than WSO_N ($4.5 \text{ nmol N m}^{-3}$) in aerosols from Lewes, Delaware on the mid-Atlantic US coast (Russell et al., 2003) and marine aerosols ($55 \pm 16\%$ of total N) over the western North Pacific (Miyazaki et al., 2011). Hence, it is of interest to investigate the seasonal variations of WSO_N and WIO_N.

Here, we present total N (TN), water-soluble TN (WSTN), WSO_N and WIO_N and nitrogen isotope ratios ($\delta^{15}\text{N}$) of TN ($\delta^{15}\text{N}_{\text{TN}}$) and WSTN ($\delta^{15}\text{N}_{\text{WSTN}}$) measured in ambient aerosols (TSP) collected from Sapporo, northern Japan over one-year period. The isotope studies are helpful in understanding the origins and atmospheric processing of N (Kawamura et al., 2004; Pavuluri et al., 2010). Based on WSO_N and WIO_N data together with various organic tracers, $\delta^{15}\text{N}$ values and air mass trajectories as well as the comparison with previous studies, we discuss the atmospheric chemistry of nitrogenous aerosols and their possible sources with a major focus on WSO_N and WIO_N in Northeast Asia.

79

80 **2 Materials and methods**

81 2.1 Aerosol sampling

82 Total suspended particles (TSP) were collected from Sapporo in the western part of Hokkaido
83 Island, northern Japan (43.07°N, 141.36°E), an ideal location for collecting the air masses
84 delivered from Northeast Asian regions: Siberia, the Russian Far East, China, and
85 surrounding oceans (Aggarwal and Kawamura, 2008; Yamamoto et al., 2011). The
86 geographic details of the sampling site are described elsewhere (Pavuluri et al., 2013). TSP
87 sampling ($n = 21$) was performed on the rooftop of the three-story building of the Institute of
88 Low Temperature Science (ILTS), Hokkaido University, Sapporo during 2 September 2009
89 and 5 October 2010 using a pre-combusted (450°C, 4 h) quartz fiber filter and high-volume
90 ($\sim 65 \text{ m}^3 \text{ h}^{-1}$) air sampler. Each sample was collected for ca. 2 consecutive weeks in order to
91 obtain sufficient amount of carbon for radiocarbon (^{14}C) analyses at organic molecular levels,
92 another objective of this research. Filter samples were placed in a pre-combusted glass jar
93 with a Teflon-lined screw cap and stored in a dark room at -20°C prior to analysis.

94 It should be noted that aerosol samples collected on quartz fiber filters might have
95 positive (adsorption of gaseous HNO_3 and NH_3 and WSON) and negative (evaporation of
96 particulate NH_4^+ salts such as NH_4NO_3 and WSON) sampling artifacts (Matsumoto et al.,
97 2014; Squizzato et al., 2013). Since the sampling time is longer in this study, the evaporative
98 loss from the particles should be more significant than the adsorbed gases by quartz fiber
99 filter and thus the reported concentrations may be underestimated. However, the ambient
100 temperatures encountered in Sapporo are rather low (range of averages of each sample period:
101 -3.30°C to 24.5°C) (Pavuluri et al., 2015) that may not cause a significant evaporative loss of
102 N species during the sampling period, because quartz filters show a good retention for
103 semi-volatile NH_4NO_3 sampled at an air temperature up to 21°C and $\sim 30\%$ at even an
104 elevated air temperature of 35°C (Schaap et al., 2004). Therefore, we believe that our
105 sampling technique does not have serious sampling artifacts even in summer.

106

107 **2.2 Chemical analyses**108 **2.2.1 Determination of TN and WSTN and their N isotope ratios**

109 TN (and WSTN) contents and their isotope ratios ($\delta^{15}\text{N}_{\text{TN}}$ and $\delta^{15}\text{N}_{\text{WSTN}}$, respectively) in TSP
110 samples were determined using an elemental analyzer (EA) (Carlo Erba NA 1500) and
111 EA/isotope ratio mass spectrometer (IRMS) (Finnigan MAT Delta Plus), respectively, as
112 described elsewhere (Kawamura et al., 2004; Pavuluri et al., 2010). An aliquot of filter
113 sample (1.8 cm in diameter disc \times 3) was extracted for WSTN with \sim 15 ml Milli Q water
114 ($>18.3 \text{ M}\Omega$) under ultrasonication for 20 min and filtrated with a syringe filter (GL Sciences
115 Chromatodisc Type A, 0.45 μm). The extracts were adjusted to pH 8~9 with \sim 0.5 ml of 0.05
116 M KOH and concentrated to dryness using a rotary evaporator under vacuum and then
117 re-dissolved in 200 μl Milli Q water. An aliquot of filter sample (1.0 cm diameter disc for TN
118 and 50 μl of water extract, a sub-sample of the 200 μl re-dissolved extract, adsorbed onto \sim 20
119 mg of pre-combusted Chromosorb for WSTN) was placed in a pre-cleaned tin cup and
120 introduced into EA. The resulting N_2 gas was measured with thermal conductivity detector in
121 EA and transferred to IRMS via an interface (ConFlo II) for $^{15}\text{N}/^{14}\text{N}$ measurement. The blank
122 levels in TN measurements are found to be 0.0 μgN . The analytical uncertainties (i.e., RSD)
123 in a single set of duplicate analyses of filter samples were within 6% for TN, 3% for WSTN,
124 0.28‰ for $\delta^{15}\text{N}_{\text{TN}}$ and 1.34‰ for $\delta^{15}\text{N}_{\text{WSTN}}$. The detection limit of EA determined by the
125 sensitivity of the detector is 0.4 $\mu\text{g N}$ (Nieuwenhuize et al., 1994).

126 We also measured WSTN using a total organic carbon (TOC)/total nitrogen (TN)
127 analyzer (Shimadzu TOC-Vcsh), as reported by Miyazaki et al. (2011). Briefly, an aliquot of
128 filter (1.4 cm in diameter disc) was extracted with 10 ml Milli Q water under ultrasonication
129 for 20 min. The extracts were filtered with syringe filter (GL Sciences Chromatodisc Type A,

130 0.45 μm) and then injected into TOC/TN analyzer. The analytical uncertainty in duplicate
131 analyses of filter samples was within 5%.

132 Concentrations of WSTN measured by EA are lower by $23 \pm 5\%$ than those measured by
133 TOC/TN analyzer. The difference was significant when the concentration of WSTN was
134 greater than $0.5 \mu\text{g m}^{-3}$. The difference may be arisen from different combustion temperature:
135 1400°C for EA v.s. 680°C for TOC/TN analyzer and different procedures. We consider that
136 the WSTN measured by EA may be underestimated due to possible evaporative loss of N
137 species such as NH_3 and HNO_3 during the concentration step and/or sticking loss of N
138 species on glass walls. In fact, if the evaporative loss of N species, e.g., NH_3 , is significant,
139 $\delta^{15}\text{N}$ of the remaining WSTN should be significantly varied in proportionate to loss of
140 WSTN, because of isotopic fractionation caused by NH_4^+ - NH_3 exchange (phase partitioning)
141 reaction (Hoefs, 2009). Under equilibrium conditions, $\delta^{15}\text{N}$ values of NH_4^+ become higher
142 than those of NH_3 , where the isotopic enrichment factor ($\epsilon_{\text{NH}_4^+-\text{NH}_3}$) estimated to be +33‰
143 (Heaton et al., 1997). However, $\delta^{15}\text{N}_{\text{WSTN}}$ did not show any abnormal deviation in its
144 temporal trend, which is comparable to that of $\delta^{15}\text{N}_{\text{TN}}$ (see Fig. 2b), suggesting that the
145 evaporative loss of N species and thus the impact of WSTN loss on its $\delta^{15}\text{N}$ are not important.
146 Here, we use WSTN measured by TOC/TN analyzer to minimize the uncertainties in the
147 estimation of WSON and WION.

148

149 **2.2.2 Measurements of inorganic ions**

150 Details in the measurements of inorganic ionic species such as NO_3^- , NH_4^+ and
151 methanesulfonate: MSA^- (CH_3SO_3^-) are described elsewhere (Pavuluri et al., 2015). Briefly,
152 an aliquot of filter sample was extracted with Milli Q water under ultrasonication and
153 filtrated with a syringe filter (GL Sciences Chromatodisc Type A, $0.45 \mu\text{m}$). The filtrates
154 were then injected into an ion chromatograph (761 Compact IC, Metrohm). A calibration

155 curve was prepared with a set of authentic standards to calculate the concentrations for the
156 samples. The analytical precision was within 4%.

157

158 **2.2.3 Estimation of WSON and WION**

159 Amounts of inorganic N (IN), i.e., sum of NO_3^- -N and NH_4^+ -N, were calculated from the
160 concentrations of NO_3^- and NH_4^+ ions (Pavuluri et al., 2015). The amounts of ON, WSON
161 and WION were estimated by the following equations (Miyazaki et al., 2011);

$$162 \quad \text{ON} = \text{TN} - \text{IN}$$

$$163 \quad \text{WSON} = \text{WSTN} - \text{IN}$$

$$164 \quad \text{WION} = \text{TN} - \text{WSTN}$$

165 Using propagating errors of each parameter, the precision of ON, WSON and WION were
166 estimated to be within 8%.

167

168 **2.2.4 Measurements of organic tracers**

169 Organic tracers such as hopanes, isoprene- and monoterpene-oxidation products were
170 determined using a capillary gas chromatograph (Hewlett-Packard 6890) coupled to mass
171 spectrometer (Hewlett-Packard 5973) (GC/MS) as described elsewhere (Fu et al., 2010).

172 Briefly, organic tracer compounds were extracted from the filter samples with
173 dichloromethane/methanol (2:1; v/v) under ultrasonication. The hydroxyl and carboxyl
174 functional groups were derivatized to trimethylsilyl (TMS) ethers and esters, respectively,
175 with 50 μl of N,O-bis-(trimethylsilyl)trifluoroacetamide (BSTFA) containing 1%
176 trimethylsilyl chloride and 10 μl of pyridine. The TMS derivatives were then measured using
177 a GC/MS. The analytical errors in duplicate analyses were less than 10%.

178 All the data reported here are corrected for the field blank collected in each season.

179

180 **3 Results and Discussion**

181 **3.1 Source regions of Sapporo aerosols**

182 Ten-day backward air mass trajectories arriving in Sapporo at 500 m above the ground level
183 (AGL) were computed for every 48 h during each sample period using HYSPLIT model
184 (Draxler and Rolph, 2012) and are plotted in Fig. 1. The trajectories show that the air masses
185 were mostly originated from Siberia passing over Northeast Asian regions and the Japan Sea
186 during autumn, winter and spring (Fig. 1). In summer, they originated from the East China
187 Sea and/or western North Pacific passing over the coastal region and/or the Main Island of
188 Japan (Fig. 1). The air parcels were mostly travelled at lower than 2000 m AGL in all seasons
189 and hence their advection from distant source regions to the receptor site should be
190 significant.

191 In fact, the radiocarbon analyses showed that percent modern carbon (pMC) of total
192 carbon and water-soluble organic carbon in our aerosols started to increase from mid to late
193 winter toward spring (Pavuluri et al., 2013), although the growing season starts in May in
194 Hokkaido when daily average temperatures are $\geq 10^{\circ}\text{C}$ (Toma et al., 2011). Because the
195 contribution of modern carbon from local vegetation is unlikely in winter, such increase in
196 pMC during winter has been mainly attributed to the contributions of biomass burning
197 aerosols via long-range transport from East Asia (Pavuluri et al., 2013). In addition, temporal
198 variations of crustal metals, e.g., Al, Ca and Fe, in Sapporo aerosols are consistent with those
199 of TSP from winter to mid summer with a peak in early spring, indicating a significant
200 atmospheric transport of Asian dust from arid regions in Mongolia and China (Pavuluri et al.,
201 2015). Therefore, we consider that the Sapporo aerosols should have largely influenced by
202 long-range transported air masses from distant sources over Northeast Asia. However, the
203 characteristics and seasonal variations of Sapporo aerosols could reflect the regional scenario
204 and we do not preclude an impact from the local sources during the campaign.

205

206 **3.2 Total nitrogen (TN) and water-soluble total nitrogen (WSTN)**

207 Concentrations of TN and WSTN ranged from 348 to 1,750 ng m⁻³ (average 1020 ± 466 ng
208 m⁻³) and 293 to 1,520 ng m⁻³ (893 ± 443 ng m⁻³), respectively, in Sapporo aerosols during the
209 study period (n = 21). The mass fractions of TN in TSP ranged from 1.20 to 8.66% (average
210 3.81 ± 2.28%). As shown in Fig. 2a, although the temporal trends of TN and WSTN did
211 follow a seasonal pattern, except for few samples, their increasing and decreasing trends are
212 not quite distinct from season to season. However, on average they were higher in winter
213 followed by spring, summer and autumn (Table 1). Similarly, TN/TSP mass ratios did not
214 show any clear temporal trend during the study period but showed a drastic increase from
215 mid autumn (1.31%) to winter (8.66%). TN/TSP mass ratios (average 3.8%, range 1.2–8.7%)
216 observed in Sapporo aerosols during the campaign are lower than those (average 5%, range
217 2–15%) reported for urban aerosols from Tokyo (Kawamura et al., 1995), where
218 anthropogenic source contributions are expected to be the major source. However, wintertime
219 TN/TSP mass ratios in Sapporo aerosols (≥7.8%) are close to that reported from urban Tokyo,
220 suggesting that the contributions from anthropogenic sources such as fossil fuel combustion
221 may be significant in winter.

222 As seen from Fig. 3, WSTN contains mostly IN (i.e., $\sum\text{NO}_3^- \text{-N} + \text{NH}_4^+ \text{-N}$) (range 70.8–
223 100%, average 89.7 ± 7.74%) whereas TN consists of a significant amount of ON (average
224 22.7 ± 17.7%, maximum 44.7%,) that ranged from below detection limit (BDL)–569 ng m⁻³
225 (215 ± 150 ng m⁻³) during the study period. High abundances of ON in TN rather than WSTN
226 indicate a significance of WION in Sapporo aerosols. Further, WSTN showed a stronger
227 linear relation with IN ($r^2 = 0.97$) than that ($r^2 = 0.90$) between TN and IN in Sapporo
228 aerosols (Fig. 3). Such deviation in the linearity suggests that the origins and/or secondary
229 formation processes of WION (i.e., TN - WSTN) may be different from that of WSTN

230 (including WSON). Therefore, it is apparent that the atmospheric loading and impacts of
231 aerosol N in Northeast Asia is significantly influenced by the sources and seasonality of ON.

232

233 **3.3 Water-soluble organic nitrogen (WSON) and water-insoluble organic nitrogen** 234 **(WION)**

235 **3.3.1 Concentrations and mass fractions in TN**

236 Concentrations of WSON and WION in Sapporo aerosols ranged from BDL to 288 ng m⁻³
237 and BDL to 454 ng m⁻³, respectively. On average, WION is more abundant (126 ± 117 ng m⁻³)
238 than WSON (89.7 ± 80.6 ng m⁻³). Although the temporal variations did not show any clear
239 seasonal pattern, WSON maximized in late autumn to mid winter whereas WION peaked in
240 mid spring to early summer (i.e., in growing season) (Fig. 4a). Further, seasonally averaged
241 concentrations of WSON are higher in winter followed by autumn, spring and summer
242 whereas those of WION are higher in summer followed by spring and much lower in autumn
243 and winter, although the seasonal differences are not statistically significant (Table 1). On
244 average, mass fraction of WION in TN is $13.9 \pm 11.1\%$ (range BDL–33.1%) and WSON/TN
245 ratio is $9.13 \pm 7.29\%$ (BDL–30.2%) during the study period. Temporal variations of the mass
246 fractions of WSON and WION in TN (Fig. 4b) are similar to those of the concentrations (Fig.
247 4a).

248

249 **3.3.2 Comparisons with previous studies from different locale**

250 As seen from Table 2, average concentration of WSON in Sapporo aerosols is lower than that
251 reported in urban aerosols from Davis, California and Kofu, Japan as well as from the coastal
252 sites: Erdemli in Turkey, Crete in Greece, and Qingdao in China. It is also lower than that
253 reported in the forest aerosols collected from Rondônia, Brazil during intensive biomass
254 burning period (dry season) (Mace et al., 2003a) and from Sapporo, Japan, and in the marine

255 aerosols over the Asian outflow regions: the Yellow Sea, South China Sea and western North
256 Pacific (Table 2). In contrast, average concentration of WSON in Sapporo aerosols is
257 comparable to that of forest aerosols from Fujiyoshida, Japan, but higher than that from
258 Rondônia, Brazil during wet season. It is also higher than that of the marine aerosols from
259 pristine oceanic regions: Cape Grim, Australia (Mace et al., 2003b) and Oahu, Hawaii
260 (Cornell et al., 2001) and over the western North Pacific (Miyazaki et al., 2011) (Table 2). In
261 addition, the higher end (maximum 288 ng m^{-3}) of WSON is comparable to average WSON
262 concentration reported in urban aerosols from Davis, California and Kofu, Japan and in
263 coastal urban aerosols from Crete, Greece (Table 2). However, average mass fraction of
264 WSON in WSTN in Sapporo aerosols is found to be comparable or close to those reported
265 for urban aerosols from Kofu, Japan, coastal urban aerosols from Crete, Greece, forest
266 aerosols from Fujiyoshida, Japan and the marine aerosols over the Yellow Sea, China Sea
267 and western North Pacific (Table 2).

268 Higher concentrations of WSON reported for urban aerosols from Davis, California,
269 during late fall and winter are attributed to increased amounts of atmospheric liquid water,
270 which promote a partition of gaseous WSON to particles (Zhang et al., 2002). In Kofu and
271 Fujiyoshida (forest site), Japan, emissions from combustion sources including biomass
272 burning and plant-derived particles as well as secondary formation through the reaction of
273 NO_2 with volatile organics are considered as potential sources of WSON (Matsumoto et al.,
274 2014). In the Mediterranean atmosphere, atmospheric dust from north Africa was expected as
275 dominant source of WSON at Erdemli, Turkey (Mace et al., 2003c) whereas in Crete, Greece,
276 fossil fuel combustion and biomass burning are also considered (Violaki and Mihalopoulos,
277 2010). Over the marginal seas of China (Asian outflow region): Qingdao, the Yellow Sea and
278 Sea of China, anthropogenic sources are considered as dominant origin of WSON with minor
279 contributions from marine and crustal sources (Nakamura et al., 2006; Shi et al., 2010). In the

280 forest aerosols from Sapporo, Japan, emissions of biological particles are considered as major
281 source of WSON in autumn whereas secondary formation of WSON from biogenic
282 hydrocarbons is considered as dominant source in summer (Miyazaki et al., 2014).

283 On the other hand, average concentration of WION ($126 \pm 117 \text{ ng m}^{-3}$) found in Sapporo
284 aerosols is comparable to that ($112 \pm 61 \text{ ng m}^{-3}$) reported in the marine aerosols over the
285 western North Pacific, where the sea-to-air emissions are considered as the major source as
286 well as the secondary production via the reactions of gaseous hydrocarbons with NH_3
287 (Miyazaki et al., 2011). In addition, the higher end (maximum 454 ng m^{-3}) of WION in
288 Sapporo aerosols is comparable to the average WION (476 ng m^{-3}) reported in the coastal
289 aerosols from Lewes, Delaware, where biological and mineral particles are expected as major
290 source (Russell et al., 2003).

291 Therefore, we consider that WSON in Sapporo aerosols may be mainly derived from
292 anthropogenic emissions including biomass burning, although emissions from biological
293 sources and secondary formation from gaseous hydrocarbon precursors cannot be excluded.
294 Meanwhile, WION may be originated from emissions of biological particles and secondary
295 formation from biogenic hydrocarbons. We further examine the possible contributions from
296 such sources in the following section based on their comparisons with the source tracers.

297

298 **3.4 Comparisons of WSON and WION with organic tracers**

299 **3.4.1 Possible sources of WSON**

300 The temporal trend of WSON is found to be somewhat similar to that of hopanes ($\text{C}_{27}\text{-C}_{32}$
301 hopanoid hydrocarbons), specific biomarkers of petroleum and coal (Schauer et al., 2002),
302 particularly in autumn and winter time Sapporo aerosols (Fig. 4c). Further, WSON showed a
303 linear relationship with hopanes with moderate but significant correlation during the
304 campaign (Fig. 5a). On the other hand, temporal variation of WSON from mid autumn to mid

305 winter is somewhat similar to that of levoglucosan (Fig. 4d), a tracer for biomass burning
306 (Simoneit, 2002), and showed a weak linear relation throughout the campaign (Fig.5b).
307 Contributions of WSON to TN are higher in the above seasons (Fig. 4b). In contrast, WSON
308 was abundant in few spring (from mid April to mid May) and summer (late June and late July)
309 samples (Fig. 4a), despite the lower levels of both hopanes and levoglucosan throughout the
310 spring and summer (Fig. 4c,d). The lower levels of levoglucosan in spring and summer, and
311 thus a weak correlation of WSON with levoglucosan during the campaign, should have been
312 caused, in part, by the decomposition of levoglucosan (Hoffmann et al., 2010), because
313 contributions of biomass burning emissions to Sapporo aerosols were considered to be
314 significant during these periods (Pavuluri et al., 2015).

315 Temporal trend of WSON from late spring to early summer (Fig. 4a) is similar to that
316 of sucrose (Fig. 4e), which is a tracer of pollens emitted from terrestrial higher plants (Fu et
317 al., 2012). The seasonal trend of WSON (Fig. 4a) is also similar to those of biogenic
318 secondary organic aerosols (SOA), i.e., α -pinene-SOA tracers (Kleindienst et al., 2007), and
319 MSA^- in late spring to early autumn (Fig. 4f,g). MSA^- is produced in the atmosphere by
320 photochemical oxidation of dimethyl sulfide (DMS) (Bates et al., 1992) that emits from
321 terrestrial higher plants (Geng and Mu, 2006), soil in forests (Yi et al., 2010) and
322 phytoplankton bloom in oceans (Bates et al., 1992). Therefore, MSA^- can be considered as a
323 tracer to understand the importance of secondary production of WSON (and WION) from
324 biogenic organics, although both DMS and MSA^- do not contain nitrogen. Interestingly,
325 WSON showed a linear relation with sucrose ($r = 0.71$, $p \leq 0.05$), α -pinene SOA tracers and
326 MSA^- during spring and summer (Fig. 5c-e). In fact, the Spearman's rank correlations of
327 WSON with α -pinene SOA tracers ($r = 0.71$) and MSA^- ($r = 0.62$) in spring and summer ($n =$
328 10) are statistically significant ($p \leq 0.05$). Further, it showed a positive Spearman's rank
329 correlation with isoprene SOA tracers ($r = 0.33$) and β -caryophyllinic acid ($r = 0.37$).

330 However, WSON shows no relation with sucrose, isoprene- and α -pinene-SOA tracers,
331 β -caryophyllinic acid and MSA^- when we consider the total data set ($n = 21$) that include all
332 seasons during the campaign (Table 3).

333 The above comparisons of WSON with source tracers imply that anthropogenic
334 emissions including biomass burning are major sources of WSON in Sapporo aerosols. Their
335 contributions to Sapporo aerosols may have been enhanced in autumn and winter when the
336 air masses enriched with forest fire and fossil fuel combustion products are often delivered
337 from Siberia passing over Northeast China (Fig. 1a, b). In fact, fossil fuel consumption is
338 significantly higher in winter than any other season in East Asia (Zhang et al., 2009). Further,
339 emissions of biological particles and secondary production by the reaction of biogenic
340 hydrocarbons with NO_3 radicals (Fry et al., 2009; Pratt et al., 2012) could contribute to
341 WSON in late spring and late spring-early autumn, respectively, although we do not preclude
342 a minor contribution from biomass burning emissions. In fact, levels of NO_3^- in Sapporo
343 aerosols declined in late spring to summer followed by a gradual increase toward early
344 autumn (Pavuluri et al., 2015). The lower levels of NO_3^- in summer may be in part caused by
345 its active involvement in the reactions with hydrocarbons (Fry et al., 2009; Pratt et al., 2012;
346 Rollins et al., 2009) to result in secondary WSON, although clean air masses from oceanic
347 regions often arrive in Sapporo during summer (Fig. 1d).

348

349 **3.4.2 Possible source and formation processes of WION**

350 The results of Spearman's rank correlation analysis between WION and source tracers are
351 presented in Table 3. WION shows a significant inversed correlation with hopanes and
352 levoglucosan during the campaign (Table 3). On the contrary, temporal trend of WION is
353 similar to those of sucrose (Fig. 4e) during late spring to early summer and of biogenic SOA
354 tracers: isoprene- and α -pinene-SOA tracers and β -caryophyllinic acid, (Fig. 4f) and MSA^-

355 (Fig. 4g) throughout the year. Further, WION showed a moderate significant positive
356 Spearman's correlation with sucrose, isoprene SOA and MSA⁻ during the study period (n =
357 21) (Table 3) and also with β -caryophyllinic acid in spring and summer ($r = 0.62$, $p \leq 0.05$; n
358 = 10).

359 Based on the above comparisons of WION with source tracers, we suggest that the
360 WION in Sapporo aerosols is mainly derived from emissions of biological particles such as
361 pollens and secondary production by the reactions of biogenic hydrocarbons (containing
362 carbonyls) with NH₃ in the atmosphere (Bones et al., 2010; Updyke et al., 2012). WSON
363 could be abundantly produced if oxidation of biogenic hydrocarbons by NO₃ radicals is
364 prominent. However, the SOA preferably produced by O₃-initiated oxidation of biogenic
365 hydrocarbons (e.g., isoprene and monoterpenes) contains carbonyl groups that can react with
366 reduced nitrogen species such as NH₃ (g), NH₄⁺ and amino acids and produce high molecular
367 weight N containing organics (Bones et al., 2010; Updyke et al., 2012), which may not be
368 fully water-soluble. For example, glyoxal has been reported to form an imidazole (Galloway
369 et al., 2009) and also undergo oligomerization reactions (Noziere et al., 2009) to form high
370 molecular weight N containing organics in the presence of NH₃. In fact, Wang et al. (2010)
371 observed high molecular weight N containing organic salts (*m/z* range of 250-500) at
372 substantial intensities in the positive-ion mass spectra in urban aerosols from Shanghai, China.
373 They interpreted the formation of such high molecular weight organic salts by Mannich
374 reaction rather than imidazole salts alone.

375 Based on observations under controlled environmental conditions, Husted and
376 Schjoerring (1996) reported that NH₃ emission from plants is enhanced with increasing leaf
377 temperature and light intensity, which is similar to the emission of organic compounds from
378 higher plants (Guenther, 1997). On the other hand, net emission of NH₃ from soil is enhanced
379 in summer due to increasing soil temperatures (Wentworth et al., 2014). Therefore, enhanced

380 emissions of biogenic hydrocarbons and NH_3 during growing season (mid spring to mid
381 summer) are very likely in spring and summer followed by the subsequent reactions and
382 production of WION in the atmosphere. In fact, the air masses arriving to Sapporo originate
383 from Siberia in spring and oceanic region passing over the Japanese Main Island in summer
384 (Fig. 1c, d). They should be enriched with the components derived from terrestrial and/or
385 oceanic emissions. In addition to biogenic SOA and MSA^- (Fig. 4f,g), abundances of NH_4^+
386 are relatively high in spring and summer (Pavuluri et al., 2015). Hence, secondary production
387 of WION by the reactions of SOA (carbonyls) with NH_3 ($\text{NH}_4^+ \leftrightarrow \text{NH}_3$) (Noziere et al., 2009)
388 should be accelerated during the warmer seasons in Northeast Asia. However, there is a need
389 to conduct the measurements of WION and source tracers at different locale in order to fully
390 understand the main source(s) of WION in atmospheric aerosols, which is a subject of future
391 research.

392

393 **3.5 Nitrogen isotope ratios of TN and WSTN: implications for source and chemical** 394 **aging**

395 **3.5.1 Seasonal variations**

396 $\delta^{15}\text{N}_{\text{TN}}$ ranged from 15.5 to 29.4‰ (average $21.9 \pm 4.1\%$) whereas $\delta^{15}\text{N}_{\text{WSTN}}$ ranged from
397 12.2 to 39.1‰ ($25.8 \pm 8.2\%$) in Sapporo aerosols during the campaign period ($n = 21$). As
398 shown in Fig. 2b, temporal trends of $\delta^{15}\text{N}_{\text{TN}}$ and $\delta^{15}\text{N}_{\text{WSTN}}$ were highly comparable. They
399 both show peaks in summer with relatively high values in late autumn (Fig. 2b). Their
400 averages, except for $\delta^{15}\text{N}_{\text{TN}}$ in autumn and winter, were significantly varied from
401 season-to-season with higher $\delta^{15}\text{N}$ values in summer followed by spring, winter and autumn
402 (Table 1). The seasonal trends of $\delta^{15}\text{N}_{\text{TN}}$ and $\delta^{15}\text{N}_{\text{WSTN}}$ suggest that aerosol N is significantly
403 influenced by season-specific source(s) and/or chemical aging of N species.

404 We also found that $\delta^{15}\text{N}_{\text{WSTN}}$ are higher than $\delta^{15}\text{N}_{\text{TN}}$ throughout the campaign, except
405 for few samples in autumn (Fig. 2b). An enrichment of ^{15}N in WSTN over TN is more
406 significant in summer than other seasons (Table 1). It has been reported that the aerosol
407 particles collected over the controlled laboratory burns of the vegetation, which significantly
408 contribute to WSTN, are enriched with ^{15}N ranging from -1.3 to 13.1‰ (average 6.6‰)
409 compared to the source vegetation (Turekian et al., 1998). Further, chemical aging of N
410 species, including gas-to-particle exchange (e.g., $\text{NH}_4^+ \leftrightarrow \text{NH}_3$) reactions during long-range
411 transport causes the enrichment of ^{15}N in aerosol N (Pavuluri et al., 2010). In addition, $\delta^{15}\text{N}$
412 of terrestrial plants, which significantly contribute to WION, is relatively low (e.g., -3.4 to
413 12.2‰ in plants from the San Francisco Bay area) (Cloern et al., 2002). Therefore, higher
414 $\delta^{15}\text{N}_{\text{WSTN}}$ values than $\delta^{15}\text{N}_{\text{TN}}$ in Sapporo aerosols suggest that contributions of biomass
415 burning to WSTN and biological sources to WION (i.e. TN-WSTN) may be significant and
416 aerosol N has been subjected to significant chemical aging during long-range transport. The
417 enhanced aging (particularly $\text{NH}_4^+ \leftrightarrow \text{NH}_3$) of nitrogenous aerosols under higher ambient
418 temperature in summer may have caused further enrichment of ^{15}N in WSTN (Pavuluri et al.,
419 2010).

420 As seen from Fig. 6, $\delta^{15}\text{N}_{\text{TN}}$ (and $\delta^{15}\text{N}_{\text{WSTN}}$) showed an inverse relation with the mass
421 fractions of NO_3^- -N and WSON in TN, but showed a linear relation with NH_4^+ -N/TN and
422 WION/TN during the study period. Their relations with NO_3^- -N/TN and NH_4^+ -N/TN are
423 statistically significant at 99% ($p \leq 0.01$) level but not in other two cases (Fig. 6). Such
424 relations imply that the $\delta^{15}\text{N}_{\text{TN}}$ and $\delta^{15}\text{N}_{\text{WSTN}}$ are mainly controlled by their sources, chemical
425 aging of nitrogenous species and abundances of NO_3^- -N and NH_4^+ -N. It is well recognized
426 that NO_3^- originates mainly from fossil fuel combustion and biomass burning whereas NH_4^+
427 from biomass burning, agricultural activities and biological emissions (Delon et al., 2012;
428 Hertel et al., 2012). In fact, NO_3^- is more abundant in mid autumn to early spring whereas

429 NH_4^+ maximize in mid spring to summer (Pavuluri et al., 2015). These results are consistent
430 with those of source tracers (Fig. 4), again suggesting that the contributions from biomass
431 burning and fossil fuel combustion are important in autumn and winter whereas the biological
432 particles in spring and biogenic emissions of gaseous species (hydrocarbons and NH_3) and
433 subsequent secondary production in spring/summer are important.

434

435 **3.5.2 Comparison with literature values**

436 Figure 7 compares the range (or mean) of $\delta^{15}\text{N}$ values reported for the particles emitted from
437 point sources in mid-latitudes as well as those observed in atmospheric aerosols from
438 different locale over the world including Sapporo aerosols. $\delta^{15}\text{N}_{\text{TN}}$ values observed in
439 Sapporo aerosols are significantly higher than those (-19.4‰ to 15.4‰) reported for the
440 particles emitted from point sources of fossil fuel combustion and waste incineration burning
441 (Fig. 7). They are also higher than those reported in urban aerosols from Paris, France, where
442 fossil fuel combustion was expected as a major source (Widory, 2007), and the marine
443 aerosols over the western North Pacific, which are considered to be mainly derived from
444 sea-to-air emissions (Miyazaki et al., 2011) (Fig. 7).

445 However, they are comparable to the higher ends of the $\delta^{15}\text{N}_{\text{TN}}$ reported in atmospheric
446 aerosols from Jeju Island, Korea (Fig. 7), which were attributed to significant atmospheric
447 processing of N species, emitted mainly from China, during long-range transport (Kundu et
448 al., 2010b). Further the lower ends of $\delta^{15}\text{N}_{\text{TN}}$ in Sapporo aerosols are comparable to the
449 higher ends of $\delta^{15}\text{N}_{\text{TN}}$ reported for the particles emitted from controlled burning of C_3 (range,
450 2.0‰ to 19.5‰) and C_4 (9.8‰ to 22.7‰) plant species in a laboratory study and those of
451 atmospheric aerosols from Piracicaba and Amazon basin, Brazil, where biomass burning is a
452 dominant source (Martinelli et al., 2002) (Fig. 7). In addition, they are well comparable to
453 those reported in the forest aerosols from Rondônia, Brazil, where biomass burning was

454 expected as a dominant source (Kundu et al., 2010a), and urban aerosols from Chennai and
455 Mumbai, India, which were considered to be mainly originated from biomass burning and
456 subjected for aging (Aggarwal et al., 2013; Pavuluri et al., 2010) (Fig. 7).

457 On the other hand, lower values of $\delta^{15}\text{N}_{\text{WSTN}}$ observed in Sapporo aerosols are
458 comparable to those reported in the forest aerosols from Sapporo experimental forest, Japan
459 (Fig. 7). Although their seasonal patterns (winter minimum and summer maximum) are
460 similar, the enrichment of ^{15}N in Sapporo aerosols (Fig. 2b) is significantly higher than that
461 (range of monthly averages, 9.0‰ to 26.0‰) reported in the Sapporo forest aerosols
462 (Miyazaki et al., 2014). Further, the range of $\delta^{15}\text{N}_{\text{WSTN}}$ values observed in Sapporo aerosols
463 are highly comparable with $\delta^{15}\text{N}$ of aerosol NH_4^+ , which is much higher than that of gaseous
464 NH_3 , in the atmosphere from the Maki monitoring station, Niigata, Japan (Fig. 7). The large
465 difference in $\delta^{15}\text{N}$ between NH_3 and NH_4^+ has been attributed to gas-to-particle (equilibrium)
466 exchange reactions in the atmosphere (Hayasaka et al., 2004) because under equilibrium
467 conditions, particulate NH_4^+ is more enriched with ^{15}N than gaseous NH_3 , where the isotopic
468 enrichment factor ($\epsilon_{\text{NH}_4^+ \cdot \text{NH}_3}$) was estimated to be +33‰ (Heaton et al., 1997). Such
469 comparisons suggest that aerosol N in Sapporo aerosols might have significantly originated
470 from biogenic sources including biomass burning and subjected to chemical aging during
471 long-range transport, although we do not preclude a contribution from fossil fuel combustion.

472

473 **4 Summary and Conclusions**

474 Water-soluble organic nitrogen (WSON) and water-insoluble organic nitrogen (WION) and
475 N isotope ratios of total nitrogen ($\delta^{15}\text{N}_{\text{TN}}$) and water-soluble TN ($\delta^{15}\text{N}_{\text{WSTN}}$) were obtained
476 for TSP aerosol samples collected from Sapporo, northern Japan during September 2009 and
477 October 2010. WSON and WION ranged from 0.00 to 288 ng m^{-3} (average 89.7 ± 80.6 ng
478 m^{-3}) and from 0.00 to 454 ng m^{-3} (126 ± 117 ng m^{-3}). Although their temporal variations did

479 not show clear seasonal trends, average WSON was found to be higher in winter followed by
480 autumn, spring and summer whereas WION maximized in summer followed by spring with
481 much lower concentration in autumn and winter. $\delta^{15}\text{N}_{\text{TN}}$ ranged from 15.5 to 29.4‰ (average
482 $21.9 \pm 4.1\%$) whereas $\delta^{15}\text{N}_{\text{WSTN}}$ ranged from 12.2 to 39.1‰ ($25.8 \pm 8.2\%$). Their seasonal
483 variations showed high ratios in summer and late autumn. Comparisons of seasonal variations
484 of WSON and WION with anthropogenic and biological source tracers together with air mass
485 trajectories suggest that fossil fuel combustion and biomass burning are major sources of
486 WSON whereas emissions of biological particles and secondary reactions of biogenic
487 hydrocarbons (carbonyls) with NH_3 significantly contribute to WION in Northeast Asia.
488 Seasonal variations of $\delta^{15}\text{N}_{\text{TN}}$ and $\delta^{15}\text{N}_{\text{WSTN}}$ and their relations to mass fractions of
489 nitrogenous components in TN suggest that aerosol N in Sapporo aerosols is mainly
490 originated from biogenic sources including biomass burning and subjected to chemical aging
491 during long-range transport, although a contribution from fossil fuel combustion is important
492 particularly in winter.

493

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743 **Table 1.** Seasonal and annual averages with standard deviation of concentrations of nitrogenous
 744 components and ^{15}N isotope ratios of TN and WSTN in atmospheric aerosol (TSP) samples
 745 collected during 2 September 2009 and 5 October 2010 from Sapporo, northern Japan.

Component	Autumn (n = 8)	Winter (n = 3)	Spring (n = 5)	Summer (n = 5)	Annual (n = 19)
Concentrations (ng m^{-3})					
TN	790 \pm 480	1456 \pm 298	1116 \pm 434	1023 \pm 433	1074 \pm 454
WSTN	714 \pm 496	1388 \pm 247	986 \pm 356	789 \pm 351	950 \pm 427
NO_3^- -N	284 \pm 120	362 \pm 55.2	395 \pm 182	73.8 \pm 24.1	279 \pm 172
NH_4^+ -N	338 \pm 347	896 \pm 97.2	511 \pm 191	647 \pm 310	576 \pm 314
ON	168 \pm 77.1	198 \pm 186	212 \pm 221	302 \pm 149	220 \pm 157
WSON	91.6 \pm 73.6	130 \pm 139	84.7 \pm 99.5	67.8 \pm 39.8	96.0 \pm 81.9
WION	77.5 \pm 50.9	68.4 \pm 51.3	130 \pm 141	234 \pm 144	125 \pm 123
Isotope ratios (‰)					
$\delta^{15}\text{N}_{\text{TN}}$	18.91 \pm 3.17	19.72 \pm 3.56	23.42 \pm 1.63	26.82 \pm 1.95	22.47 \pm 3.98
$\delta^{15}\text{N}_{\text{WSTN}}$	19.17 \pm 6.45	23.84 \pm 4.13	26.82 \pm 2.46	36.88 \pm 2.46	26.71 \pm 8.16

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748 **Table 2.** Concentrations of water-soluble organic nitrogen (WSON) and the mass fractions of WSON in
 749 water-soluble total nitrogen (WSTN) in Sapporo aerosols together with those in atmospheric aerosols from
 750 different sites in the world. ^aTotal of fine and coarse mode particles, ^bDry (biomass burning) season, ^cAsian
 751 outflow region.

Locale	Study Period	Aerosol Size	WSON (ng m ⁻³)	WSON/WSTN (%)	Reference
<i>Urban sites</i>					
Sapporo, Japan	Sept 2009 to Oct 2010	TSP	89.7 ± 80.6	10.5 ± 7.5	This study
Davis, California	Aug 1997 to Jul 1998	PM _{2.5}	265 ± 190	23	Zhang et al. (2002)
Kofu, Japan	Aug 2009 to Jan 2013	PM ₁₀ ^a	221	~14	Matsumoto et al. (2014)
<i>Forest sites</i>					
Rondônia, Brazil	Mar–May 1999	PM ₁₀	49 ± 64.4	45 ± 35	Mace et al. (2003a)
Rondônia, Brazil	Sept–Oct 1999 ^b	PM ₁₀	854 ± 938	43 ± 31	Mace et al. (2003a)
Sapporo, Japan	Jun 2009 to Dec 2011	TSP	157 ± 127	20 ± 11	Miyazaki et al. (2014)
Fujiyoshida, Japan	Aug 2009 to Jan 2013	PM ₁₀ ^a	101	~11	Matsumoto et al. (2014)
<i>Coastal sites</i>					
Erdeмли, Turkey	Mar–May 2000	TSP	406 ± 588	~26 ± 28	Mace et al., (2003c)
Crete, Greece	2005–2006	PM ₁₀ ^a	~239	13	Violaki & Mihalopoulos (2010)
Qingdao, China	Mar–Apr 2006	TSP	2520 ± 1760	~20	Shi et al. (2010)
<i>Oceanic sites</i>					
Yellow Sea ^c	Mar 2005	TSP	2860 ± 1780	~17	Shi et al. (2010)
Yellow Sea ^c	Apr 2006	TSP	1220 ± 812	~17	Shi et al. (2010)
South China Sea ^c	Apr 2005	TSP	910 ± 280	~34	Shi et al. (2010)
East China Sea ^c	Sept–Oct 2002	TSP ^a	756 ± 504	~24	Nakamura et al. (2006)
East China Sea and western North Pacific ^c	Mar 2004	TSP ^a	224 ± 266	~10	Nakamura et al. (2006)
Cape Grim, Australia	Nov–Dec 2000	TSP	50.4 ± 79.8	~21	Mace et al. (2003b)
Cape Grim, Australia	Nov–Dec 2000	TSP	13.0 ± 13.3	~25	Mace et al. (2003b)
Oahu, Hawaii	Jul–Aug 1998	PM ₁₀	46.2 ± 28	~33	Cornell et al. (2001)
Western North Pacific	Aug–Sept 2008	TSP	~17.5		Miyazaki et al. (2011)

752

753 **Table 3.** Spearman's rank correlation of WSON and WION with source tracers in Sapporo
 754 aerosols during the study period (n = 21). Significant correlation coefficients at 0.05 level are
 755 shown in bold.

Source tracer	Correlation coefficient (ρ_r)	
	WSON	WION
Hopanes	0.48	-0.50
Levogluconan	0.18	-0.57
Sucrose	-0.02	0.38
Isoprene-SOA tracers	-0.12	0.50
α -Pinene-SOA tracers	0.21	0.14
β -Caryophyllinic acid	0.06	0.30
MSA ⁻	0.09	0.59

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758 **Figure captions**

759

760 **Fig. 1.** Plots of 10-day backward air mass trajectories arriving over Sapporo at 500 AGL.

761

762 **Fig. 2.** Seasonal variations in (a) concentrations of TN and WSTN and (b) N isotope ratios of
763 TN ($\delta^{15}\text{N}_{\text{TN}}$) and WSTN ($\delta^{15}\text{N}_{\text{WSTN}}$).

764

765 **Fig. 3.** Linear relations of IN (sum of NO_3^- -N and NH_4^+ -N) to (a) WSTN and (b) TN.

766

767 **Fig. 4.** Seasonal variations of (a) concentrations of WSON and WION, (b) mass fractions of
768 WSON and WION in TN, and (c-e) concentrations of organic molecular tracers and MSA^- in
769 Sapporo aerosols. Concentrations of hopanes are sum of the concentrations of C_{27} - C_{32}
770 hopanoid hydrocarbons. Concentrations of isoprene-SOA tracers are sum of the
771 concentrations of 2-methylglyceric acid, cis-2-methyl-1,3,4-trihydroxy-1-butene,
772 3-methyl-2,3,4-trihydroxy-1-butene, trans-2-methyl-1,3,4-trihydroxy-1-butene,
773 2-methylthreitol and 2-methylerythritol whereas concentrations of α -pinene-SOA tracers are
774 sum of the concentrations of 3-hydroxyglutaric, pinonic, pinic and
775 3-methyl-1,2,3-butanetricarboxylic acids. The data of organic molecular tracers are from
776 Pavuluri et al. (2013) and MSA^- is from Pavuluri et al. (2015).

777

778 **Fig. 5.** Linear relations of WSON with (a) hopanes (C_{27} - C_{32} hopanoid hydrocarbons) and (b)
779 levoglucosan during campaign period ($n = 21$) and with (c) sucrose, (d) α -pinene-SOA
780 tracers (sum of 3-hydroxyglutaric, pinonic, pinic and 3-methyl-1,2,3-butanetricarboxylic
781 acids), and (e) MSA^- in spring and summer ($n = 10$) in Sapporo aerosols.

782

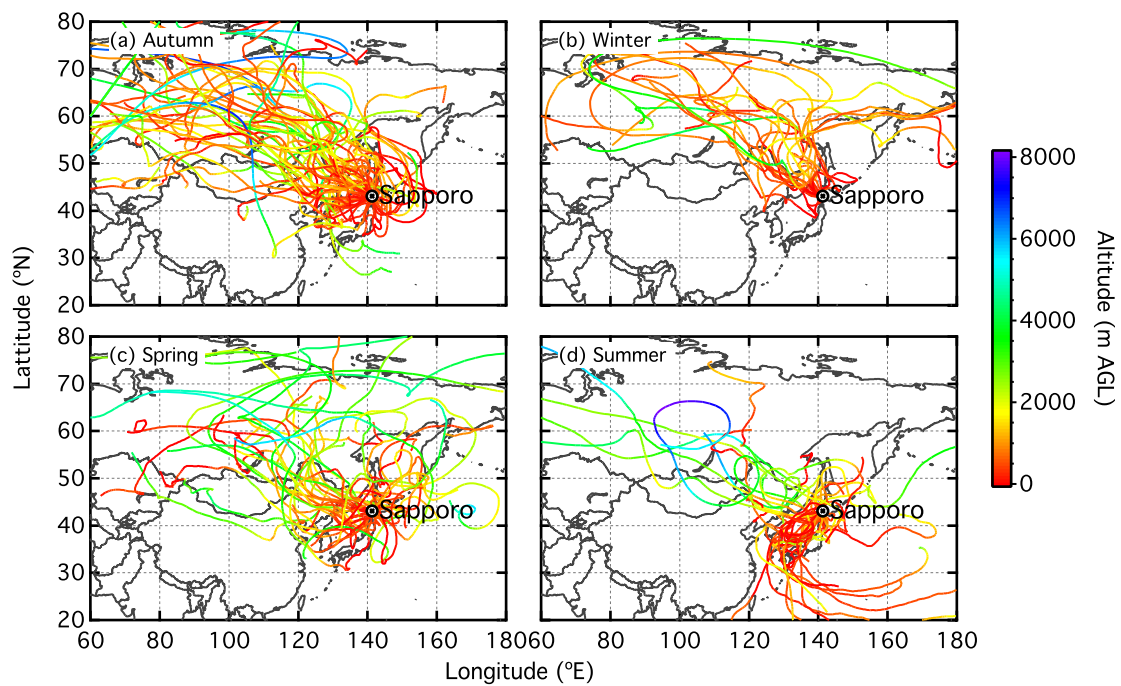
783 **Fig. 6.** Scatter plots between $\delta^{15}\text{N}_{\text{TN}}$ (and $\delta^{15}\text{N}_{\text{WSTN}}$) and mass fractions of nitrogenous
784 components (NO_3^- -N, NH_4^+ -N, WSON and WION) in TN in Sapporo aerosols.

785

786 **Fig. 7.** Range or mean N isotope ratios in the particles emitted from point sources, source
787 substance, and atmospheric aerosols from different sites in the world. *, **, and *** show
788 gaseous NH_3 , particulate NH_4^+ and WSTN, respectively. ^aWidory (2007); ^bTurekian et al.
789 (1998); ^{c,g}Kundu et al. (2010a, 2010b); ^dMartinelli et al. (2002); ^ePavuluri et al. (2010);
790 ^fAggarwal et al. (2013); ^{h,j}Miyazaki et al. (2011, 2014); ⁱHayasaka et al. (2004); ^kThis study.

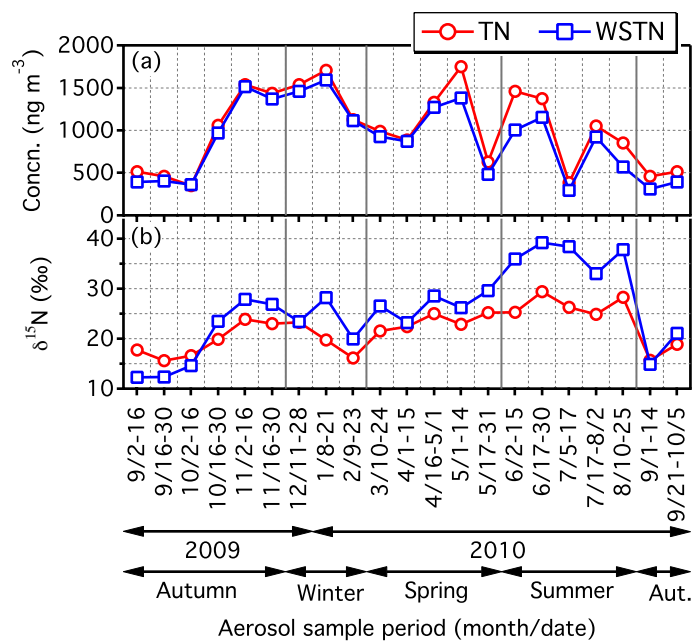
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792 **Figure 1.**



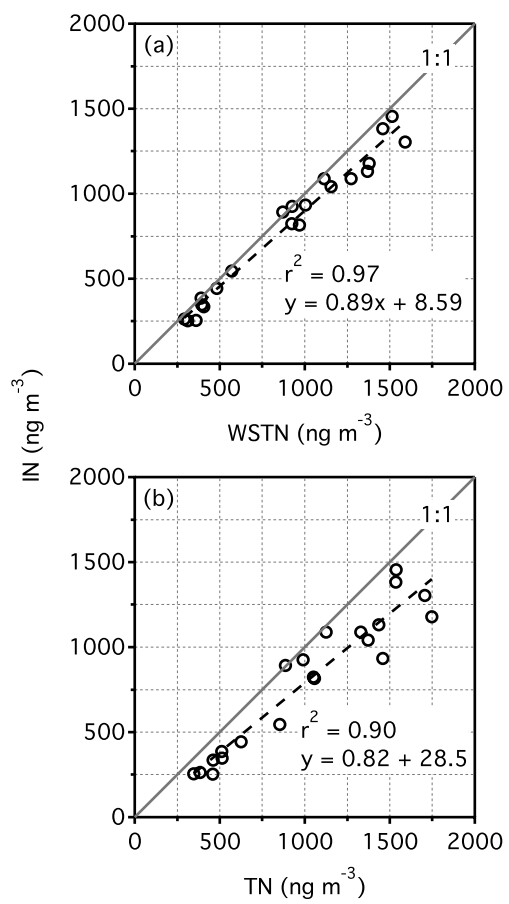
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795 **Figure 2.**

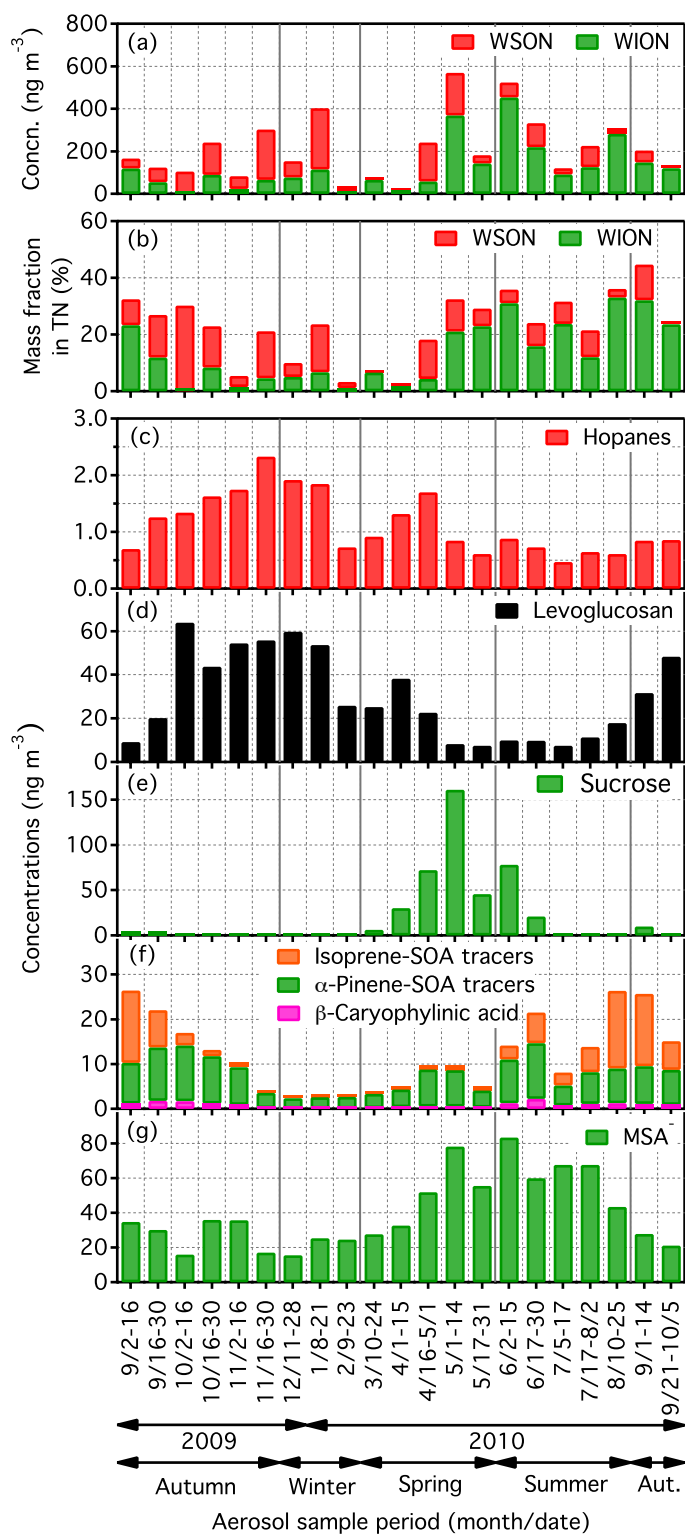
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798 **Figure 3.**

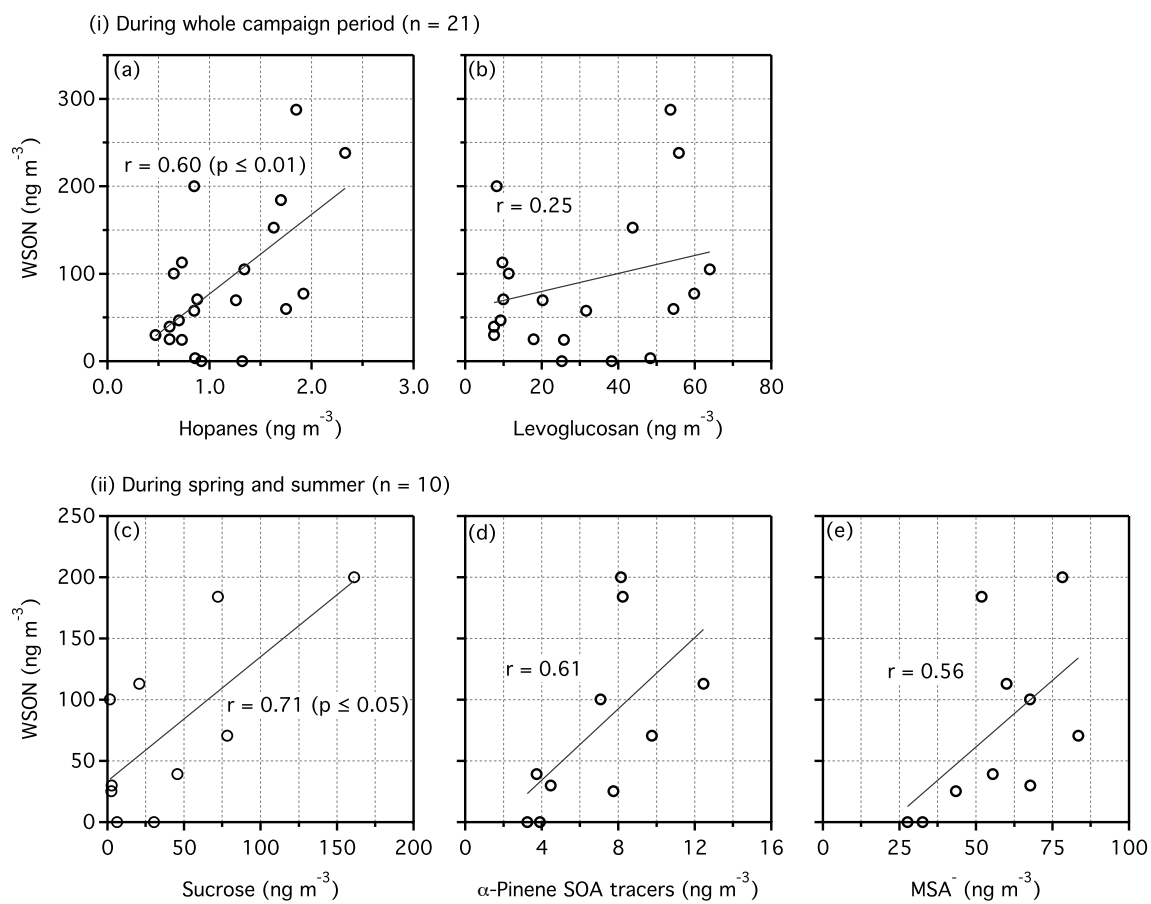
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801 **Figure 4.**

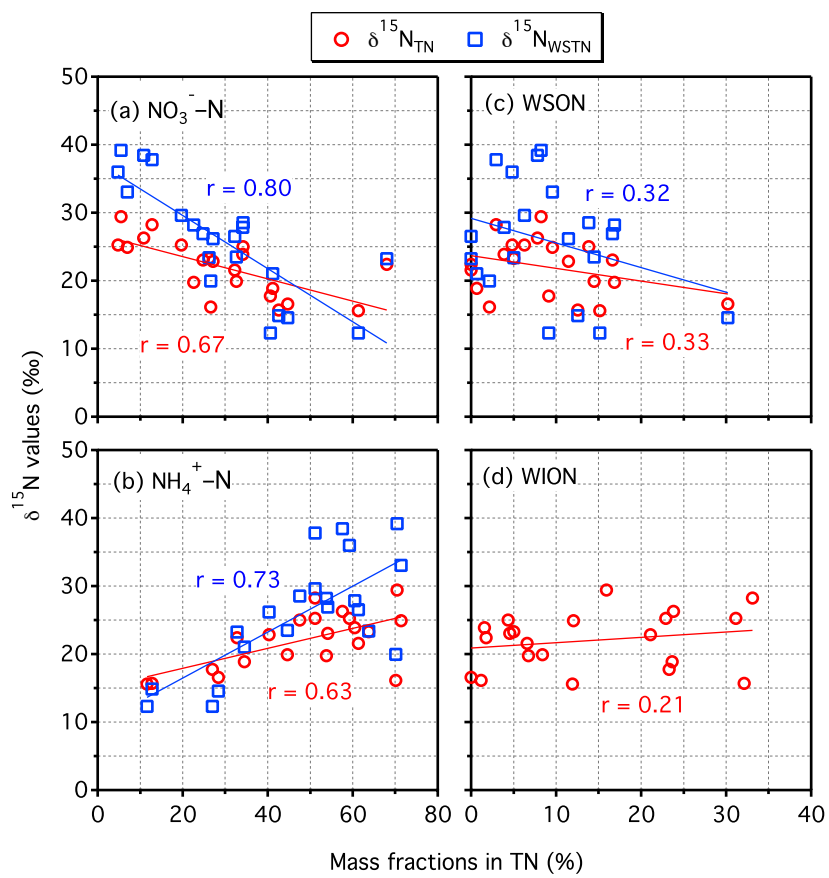
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804 **Figure 5.**

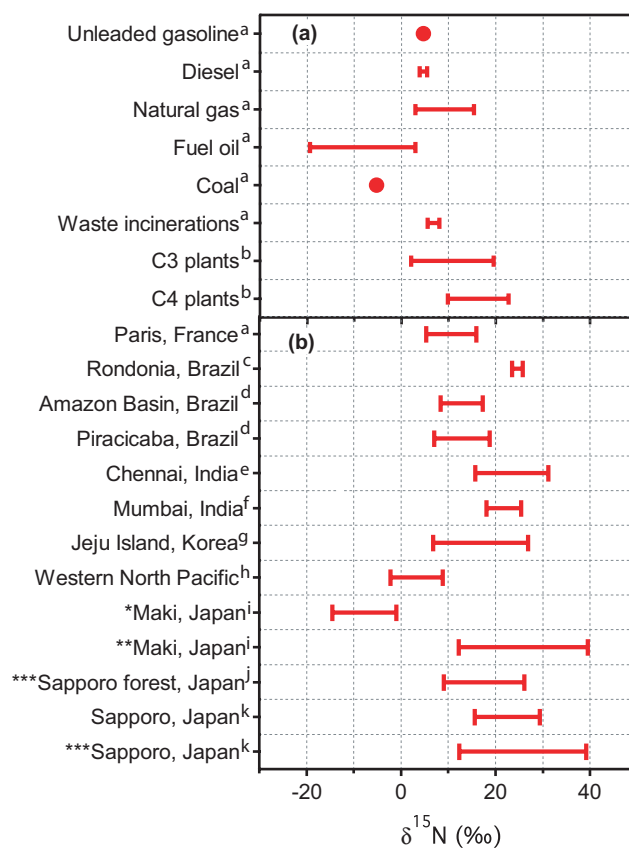
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807 **Figure 6.**

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810 **Figure 7.**

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