

1 Seasonal and Diurnal Trends in Black Carbon Properties and Co-Pollutants in Mexico City

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1 ABSTRACT

2 The Mexico City Metropolitan Area (MCMA) is a region that continues to grow in population and
3 vehicular traffic as well as being the largest urban source of criteria and short lived climate pollutants
4 (SLCP) in Latin America. The local city government has made significant progress in controlling some
5 of these pollutants, i.e. ozone (O₃) and carbon monoxide (CO), but particulate matter (PM_{2.5} and PM₁₀)
6 and black carbon (BC) have shown less positive response to mitigation strategies that have been in
7 place for almost three decades. For the first time, extended measurements have been made of
8 equivalent black carbon (eBC), derived from light absorption measurements made with a Photoacoustic
9 Extinctionmeter (PAX), over a 13 month period from March, 2013 through March, 2014. The daily trends
10 in workday (Monday through Saturday) and Sunday eBC, PM_{2.5} and the co-pollutants CO, O₃ and NO_x
11 are evaluated with respect to the three primary seasons in the MCMA: rainy, cold and dry and warm
12 and dry.

13 The maximum values in all of the particle and gas concentrations were significantly larger (Student's t-
14 test, P<0.05) during the dry periods than in the rainy season. The changes from rainy to dry seasons
15 for eBC, PM_{2.5}, CO, O₃, and NO_x were 8.8 to 13.1 µg m⁻³ (40%), 49 to 73 µg m⁻³ (40%), 2.5 to 3.8 ppm
16 (40%), 73 to 100 ppb (30%) and 144 to 252 ppb (53%), respectively.

17 The primary factors that lead to the large differences between the wet and dry seasons are the
18 accelerated vertical mixing of boundary layer and free tropospheric air by the formation of clouds that
19 dilutes the concentration of the SLCPs, the decreased actinic flux that reduces the production of ozone
20 by photochemical reactions and the heavy, almost daily rain that removes particulate matter.

21 A significant "weekend effect" was identified, particularly the decrease in BC due to fewer large
22 transport vehicles, that are fueled by diesel, that produce a large fraction of the BC. The other co-
23 pollutant concentrations are also significantly less on weekends except for O₃ that shows no change in
24 maximum values from workday to Sunday. This lack of change is a result of the balancing effects of
25 lower precursor gases, i.e. VOCs, offset by lower concentrations of NO_x that is an O₃ inhibitor.

26 A comparison of the average, maximum value of eBC measured during the one year period of the
27 current study, with maximum values measured in shorter field campaigns in 2000 and 2006, show no
28 significant change in the eBC emissions over a 14 year period. This suggests that new mitigation
29 strategies may need to be developed to decrease potentially toxic levels of this particulate pollutant.

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31 **1. Background**

1 Short-lived climate pollutants (SLCP) are those gases and particles whose atmospheric
2 lifetimes range from less than a day to a few years (Shindell et al, 2012; Bowerman et al., 2013) and
3 exert a positive radiative forcing to the global climate. Black carbon (BC), methane (CH₄), tropospheric
4 ozone (O₃) and selected hydrofluorocarbons (HFCs) also impact the environment and have negative
5 effects on human health. Although the SLCPs play a role in the radiative balance of the earth and have
6 global impacts, their more immediate effects are observed locally and regionally.

7 Black carbon has been identified as the atmospheric component whose regional warming effect
8 on climate is second only to carbon dioxide (Bond et al., 2013). Unlike CO₂ whose increasing global
9 concentration and long lifetime require an international scale effort to reduce concentrations, BC is a
10 contaminant whose negative effects locally can be significantly decreased through local and regional
11 mitigation (Shindell et al, 2012; Bowerman et al., 2013). This is an optimistic possibility since BC is a
12 product of inefficient combustion from motor vehicles (cars, buses, and trucks), commercial processes
13 (e.g. brick making, cooking, oil and gas production), residential heating and cooking and open biomass
14 burning (e.g. agricultural clearing and inadvertent forest fires). All of these emission sources lend
15 themselves to a greater level of control through regulatory actions and with technologies that are
16 already available. Likewise, the other co-pollutant SLCPs may be similarly reduced through the
17 measures taken to reduce BC.

18 The Mexico City Metropolitan Area (MCMA), referred to as Greater Mexico City, has a
19 population that now exceeds 22 million people according to the 2014 census, making it the largest
20 urban area in North America. The MCMA is also one of the most polluted megacities in the western
21 hemisphere and a major source of SLCPs that not only affect the local and regional environment but
22 can also contribute to global climate change (Barth and Church, 1999; Singh et al., 2009). According to
23 the most recent assessment by the city government (SMA-GDF, 2012) annual emissions of BC, PM_{2.5},
24 CO and NO_x are 2 kilotons (Kt), 9.4 Kt, 1606 Kt and 239 Kt, respectively.

25 The Mexico City government established a network of automatic air quality stations in 1986
26 (Red Automática de Monitoreo Atmosférico - RAMA, <http://www.aire.df.gob.mx/>) to monitor O₃, CO,
27 NO_x and SO₂, adding mass concentration of particles with aerodynamic diameter less than 10 μm
28 (PM₁₀) in 1991 and less than 2.5 μm (PM_{2.5}) in 2003. RAMA currently has 29 stations located in the
29 MCMA (Fig. 1, courtesy of the city government's environmental agency,
30 <http://www.aire.df.gob.mx/default.php>). Until recently, however, there were no measurements of BC at
31 any of the RAMA stations. The few BC measurements previously available had been taken during short
32 field campaigns. The first published data on BC were by Baumgardner et al. (2000) who analyzed
33 measurements with a Particle Soot Aerosol Photometer (PSAP) to derive equivalent black carbon

1 (eBC), the name that has been given to BC derived from measurements of light absorption (Petzold et
2 al., 2013). Those measurements were made during a two week period in November, 1997 at an
3 elevated site (400 m above the city) in the southwest sector of Mexico City. Additional measurements
4 were made, also with a PSAP, at three RAMA sites in 2000 over a three week period in January and
5 February (Baumgardner et al., 2002). In the spring of 2003 and again in 2005, measurements were
6 made with a co-located PSAP and a single particle soot photometer (Baumgardner et al., 2007). These
7 were very short measurement campaigns of one week each. In 2003, as part of the MCMA-2003
8 campaign (Molina et al., 2007), a mobile laboratory was deployed at various locations around the city.
9 Filter samples were evaluated with electron microscopy and aerosol mass spectrometry (Johnson et
10 al., 2005) and eBC was derived from an aethalometer, a filter based instrument like the PSAP (Jiang
11 et al., 2005). Marley et al. (2007) also made measurements of eBC during MCMA-2003 with an
12 aethalometer and compared them with measurements that they had made in 1997.

13 During the MILAGRO campaign in March 2006 (Molina et al., 2010), measurements of eBC and
14 elemental carbon (EC) were made at various locations within the city (Marley, 2009a,b) and in the
15 regions around the urban area (Baumgardner et al., 2009; Subramanian et al., 2010).

16 In 2011, a new instrument, the Photoacoustic Extinctionmeter (PAX) was installed at the RAMA
17 supersite located in the north central section of the city (Fig. 1). From 2011 to 2013 the instrument was
18 evaluated as a possible addition to the RAMA air quality network. In March 2013, the PAX was certified
19 by RAMA as an operational instrument for measuring the optical properties of BC relevant for climate,
20 i.e. the light scattering and absorption coefficients, B_{scat} and B_{abs} , and the derived eBC. We report in
21 this paper the results from 13 months of continuous observations made by this instrument since March
22 2013.

23 Numerous studies have focused on a better understanding of the meteorological processes
24 that transport pollutants from Mexico City and the chemical processes that lead to the production of
25 ozone and aerosol particle mass (e.g. Stephens et al, 2008; Molina et al, 2010 and references therein).
26 There have been no studies, however, that look at how the chemical processes are linked to the
27 seasonal meteorology. The objective of this study is to evaluate the temporal trends in some of the
28 SLCPs (BC and O_3) and related co-pollutants (CO, NO_x and $PM_{2.5}$) and show how they are modulated
29 by seasonal variations in temperature, humidity, precipitation and radiation.

30 This study is an extension of the evaluation of Mexico City pollutants by Stephens et al. (2008)
31 who analyzed daily trends but did not have information on BC. In the following analysis, the diurnal
32 trends in the concentrations of SLCPs over a 13 month period are compared with respect to seasonal
33 climate changes and linked to the underlying physical processes that lead to seasonal differences.

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2. Measurement Methodology

The RAMA “super-site” (Fig. 1) is located in an area of the city that is a mixture of private residences and industry (19° 29.0’ N, 99° 9.85’ W, 2243 m a.m.s.l). The site is approximately 300 m from a major highway that is a conduit for north-south traffic. Automobiles, local and regional buses and heavy trucks are responsible for most of the primary pollutants (CO, NO_x, BC and primary PM_{2.5}). The cars and buses run primarily on unleaded gasoline whereas the trucks mostly consume diesel that is the major producer of BC. Hence, although the majority of the BC comes from the trucks, by the time the air reaches the measurement site, the BC is well mixed with the CO, NO_x and other particles that are produced by the cars and buses as well as by the trucks.

2.1 Meteorology, radiation, gas and PM_{2.5} measurements

Temperature and relative humidity are measured using a Met One (Met One Instrument, Grants Pass, OR) model 083E sensor located inside an aspirated radiation shield. Wind speed and direction are measured with a Met One model 010C lightweight three-cup anemometer, and a Met One model 020C lightweight airfoil vane. The temperature and relative humidity sensors are positioned in a meteorological tower 4 meters above the ground. The anemometer and vane are located at the top of the 10-meter meteorological tower. All sensors are calibrated annually. The ultraviolet radiation, over the wavelength range from 400 – 320 nm (UV-A), is measured with a Solar Light UVA model 501 radiometer that outputs its data in mW cm⁻². The UV-A measurements used in this paper are not made at the “Supersite”, but are an average from six stations across the city (see Fig. 1): Merced (MER), Montecillo (MON), Pedregal (PED), San Agustin, (SAG), Santa Fe (SFE), and Tlalnepantla (TLA). Precipitation was measured with a tipping bucket pluviometer at five stations (see Fig. 1): Legaria (IBM), Metropolitan Museum (MCM), Cerro de Tepayac (TEC), Tlalnepantla (TLA) and Xalostoc (XAL).

Ozone is measured by ultraviolet photometry (Teledyne API model 400E, San Diego CA). The NO_x, defined operationally, is measured by chemiluminescence following conversion in a heated molybdenum NO₂-to-NO converter (Teledyne-API model 200E, San Diego, CA). The CO is measured by a gas filter correlation, non-dispersive infrared analyzer (Teledyne API model 300E, San Diego, CA). The gas analyzers are tested every two weeks with a known standard and calibrated quarterly.

PM_{2.5} is measured with a TEOM 1400A ambient particulate monitor 8500C FDMS (Thermo Scientific, Franklin, MA), with a BGI VSCC PM_{2.5} cyclone inlet for size selection. PM_{2.5} hourly concentrations are reported at local temperature and pressure conditions. The TEOM is operated at a temperature of 35°C in order to measure dry particle mass.

2.2 The Photoacoustic Extinctionmeter

The eBC, B_{scat} and B_{abs} are derived from the PAX, a sensor that evolved from the Photoacoustic Soot Spectrometer (PASS) developed at the University of Nevada (e.g., Arnott et al., 2005, 2006; Moosmüller et al., 2009) and commercialized by Droplet Measurement Technologies (Chan et al., 2011; Holder et al., 2014; Liu et al., 2014; Nakayama et al., 2015). The PAX performs simultaneous measurements of aerosol scattering and absorption coefficients using a single diode laser modulated at approximately 1500 Hz. The 375, 405, 532 and 870 nm single wavelength versions of the instrument have identical measurement cells and differ only in a few optical components. Measurements reported in the current study were made by a PAX with the 870 nm laser.

Figure 2 shows a diagram of the PAX measurement cells. The nominal 1 LPM sample flow first passes a solenoid pinch valve (not shown) that allows for periodic filter sampling to determine the instrument background signals. The lengths of sample tubing between the filter line and sample line are the same to provide the same acoustic properties along both channels. The sample flow passes into an acoustically insulated enclosure containing the sample cells, where it is then split, making opposing 90-degree turns into the scattering and absorption measurement regions. The sample is exhausted through two ports at the ends of the absorption cells before being filtered and pumped out of the instrument. The flows are controlled using critical orifices and do not affect the calculated absorption and scattering coefficients, which depend only on the properties of the sampled particles and geometry of the measurement region. Sample flow does affect the residence time of particles in the measurement cells, which is approximately 7 seconds.

The absorption cell includes a 0.635 cm diameter tube 10.8 cm long called the resonator. A small microphone located at the top of the resonator detects the pressure perturbations induced by the heating of the light absorbing particles. The microphone signal is passed through a fast Fourier transform circuit that gives the peak power at the resonator frequency, which is calculated based on the resonator geometry, air pressure, temperature and dew point temperature measured in the cell. The microphone raw pressure signal (p_{mic}) is converted to the raw light absorption coefficient using the equation:

$$b_{\text{abs,raw}} = \frac{p_{\text{mic}} A \pi^2 f}{P_L (\gamma - 1) Q} \cos \phi_{\text{raw}} \quad (1)$$

1 where A is the resonator cross-section, f is the calculated resonator frequency, P_L is the laser power,
2 γ is the ratio of isobaric and isochoric specific heat for air, Q is the calculated resonator quality factor
3 (also calculated from measured cell pressure, temperature and dew point temperature), and ϕ_{raw} is
4 the phase raw absorption signal relative to the phase of the laser power signal plus the phase
5 correction. The phase correction accounts for the difference between the speed of light and sound that
6 translates to a phase shift in the acoustic pressure measured by the microphone relative to the
7 modulated laser power incident on the absorbing material. See Arnott et al. (2005) for more details on
8 applying phase corrections to photoacoustic measurements of light absorption coefficients.

9 The raw absorption and scattering signals must be corrected for background absorption and
10 scattering signal in the cell to give the desired light absorption and scattering coefficients due to the
11 sampled aerosol particles. Background absorption arises from absorbing gases (not important for the
12 870 nm version of the PAX but relevant for shorter wavelength versions), absorption by the cell and
13 particles deposited to the cell surfaces (especially the cell windows) and acoustic and electrical noise
14 that contributes to signal at the resonator frequency. The background scattering is not sensitive to
15 acoustical noise, but is affected by electrical noise and light scattering from the cell surfaces and
16 particles deposited to the cell surfaces and windows. The background signals are measured
17 simultaneously at a user-specified interval, typically once every ten minutes, by passing the sample air
18 through a filter before it enters the measurement cell. The background measurement period occurs
19 after a 20 second flush with particle-free air to mix sampled particles out of the measurement cells and
20 lasts 30 seconds, followed by another 20 second flush before regular measurements resume.
21 Background absorption and scattering values are saved in the data record and automatically subtracted
22 from the measured raw signals to give the absorption and scattering coefficients in real-time. Typically
23 the background drift is small ($<3 \text{ Mm}^{-1}$) compared to the observed scattering and absorption
24 coefficients, especially in more polluted environments like Mexico City. Background drift occurs due to
25 small changes in sampling environment, such as temperature and relative humidity. Over longer
26 periods of time backgrounds are affected by contamination on the cell windows and minor changes in
27 the optical alignment. Cleaning and re-alignment of the system can be necessary when the magnitude
28 of the background drift becomes comparable to changes in the scattering and absorption values on the
29 timescales of the filtered background measurements. Post-processing of background values to apply
30 a linearly-interpolated correction (using background measurements before and after a sample point)
31 provides a better correction for background drift compared to the standard background correction
32 performed in real-time by the instrument.

1 The PAX absorption and scattering sensitivity depends on the sampling conditions and laser
2 power in the cell. The laser power in turn depends on the quality of the alignment and level of
3 contamination on the cell windows. The scattering sensitivity is mainly limited by the small
4 measurement volume and variability in the particle sample stream. The absorption sensitivity is mainly
5 limited by electrical and acoustical noise that cannot be distinguished from the true particle signal. While
6 noise in the sampling environment can reach the acoustic cell through vibrations in the case, a bigger
7 factor is acoustical noise transmitted through the sampling line directly into the resonator. Helmholtz
8 notch filters can be inserted into the sampling line to filter noise at the measurement frequency in
9 environments with high acoustic noise and relatively low aerosol loadings. In Mexico City, however,
10 these were not necessary because the PAX was installed in a temperature and noise controlled
11 environment with minimal acoustic noise.

12 Figure 3 shows the corrected absorption coefficients and Allan variance plot for the absorption
13 signal measured during overnight sampling of filtered air with an 870 nm PAX. The instrument
14 performed a background measurement approximately every 10 minutes, which corrected for the
15 instrument drift on timescales longer than 10 minutes, so the variance is only shown up to a five-minute
16 averaging interval. At the highest temporal resolution of one second, the variance was 0.14 Mm^{-1} ,
17 equivalent to about $.03 \mu\text{g m}^{-3}$ black carbon using the factory applied mass-specific absorption cross-
18 section (MAC) of $4.74 \text{ m}^2 \text{ g}^{-1}$. This is the value at 870 nm, derived using the λ^{-1} correction to the 7.5 m^2
19 g^{-1} recommended by Bond and Bergstrom (2006). The three-sigma sensitivity for one second
20 averaging was 1.1 Mm^{-1} ($0.23 \mu\text{g m}^{-3}$ BC). The values are a “best case” scenario where the
21 environmental conditions were very stable (running overnight in an empty laboratory) and because the
22 particle filter also acted to block acoustical noise from the sampling environment from reaching the cell.
23 We have observed one second variance in PAX absorption signals ranging from $\sim 1\text{-}5 \text{ Mm}^{-1}$ depending
24 on sample environment.

25 The PAX uses a wide-angle ($6^\circ\text{-}174^\circ$) integrating reciprocal nephelometer to measure the light
26 scattering coefficient. The scattering detector consists of a Teflon diffuser placed in front of a
27 photodiode (Fig. 2). The scattering measurement responds to all particle types regardless of chemical
28 makeup, mixing state, or morphology.

29 The PAX is calibrated through a two-step process by first introducing high concentrations of
30 purely light scattering particles followed by high concentrations of partially absorbing particles to the
31 measurement cells (Arnott et al., 2000; Nakayama et al., 2015). The concentrations must be high
32 enough to allow direct measurement of the extinction coefficient (b_{ext}) from the reduction in laser power
33 measured at the laser power monitor (Fig. 2) using Beer’s law:

$$b_{\text{ext}} = -\frac{1}{l} \ln \frac{I}{I_0} 10^6 [\text{Mm}^{-1}] \quad (2)$$

where I is the average laser power measured when the high concentration of particles is being sampled, I_0 is the average laser power immediately before and after the high concentrations are introduced to the cell, and l is the path length of the laser beam through the entire optical cavity (scattering region and absorption region) between the two windows, which for the PAX is 0.354 m.

Ignoring truncation in the nephelometer cell, an error that is of order 6% (Nakayama et al., 2015), the measured extinction coefficient equals the measured scattering coefficient for purely scattering particles (e.g. nebulized, dry ammonium sulfate). The relationship between the measured extinction and scattering coefficients are fit using a linear regression giving a calibration factor for the scattering measurement. Introducing calibration particles that have non-zero absorption means the measured extinction equals the now-calibrated scattering coefficient plus the absorption coefficient. Subtracting the measured scattering coefficient from the extinction coefficient determined from Beer's law gives the absorption coefficient, which can be regressed against the measured absorption coefficient to obtain a second calibration factor for the absorption cell. The major uncertainty introduced by this method of calibration is that it assumes that the measured particles have similar scattering phase functions so the truncation error in the scattering cell cancels.

The PAX measures the light scattering and absorption coefficients, B_{scat} and B_{abs} , directly using the in-line nephelometer and photoacoustic technique, respectively. The single scattering albedo (SSA), defined as the ratio of B_{scat} to the sum of B_{scat} and B_{abs} (extinction coefficient) is also derived and recorded, along with the eBC that is derived from the B_{abs} as described above.

The PAX was operated with a $\text{PM}_{2.5}$ cyclone particle separator on the inlet in order to remove larger particles and the particle stream was also dried using a diffusion drier in order to minimize measurement effects at high relative humidity (e.g., Lewis et al., 2009; Murphy et al., 2009) and also to provide the scattering coefficients of dry particles only. The PAX was connected to the cyclone by 3 m of 3/8" conductive tubing. Transmission losses are insignificant at the flowrate of 1 Lpm, based on calculations made using the Aerocalc program developed by Baron and Willeke (2001).

The uncertainty in the measured B_{scat} is ca. $\pm 10\%$ due to the accuracy with which the instrument can be calibrated and the truncation error. The uncertainty in the measured B_{abs} is ca. $\pm 20\%$. This uncertainty stems primarily from the accuracy of the calibration but the losses in the inlet system also contribute to the overall accuracy. Given the range of possible MAC values that depend on the

1 composition and size distribution of the black carbon (Bond and Bergstrom, 2006), the uncertainty in
2 the MAC is of order 50% and hence the derived eBC from B_{abs} is approximately 55-60%. In addition,
3 Bond et al. (2007) show that BC coated by non-light absorbing material can cause a lensing effect that
4 leads to enhanced absorption, perhaps as much as a factor of 1.5; hence, the derived eBC mass
5 concentrations may be as much as 50% higher than the actual concentration of BC due to this coating
6 effect.

7 **3. Measurements and Analysis**

8 The meteorology, gas and particle measurements were made at the RAMA supersite from
9 March 6, 2013 to March 31, 2014. Mexico City is located in a sub-tropical zone where the seasons can
10 be generally separated into three periods: 1) the rainy season extends from approximately June until
11 October with an average annual rainfall exceeding 1000 mm; 2) the cool, dry season, from November
12 to March and 3) the warm, dry period from April through May. The starting and ending dates for these
13 seasons will vary from year to year but these three seasonal periods will be used in the analysis as an
14 operational definition, abbreviated from hereon as “Rainy Season” (RS), cold and dry season (CDS)
15 and warm and dry season (WDS). During the dry months, clear sky conditions lead to a strong thermal
16 inversion at night (Collins and Scott, 1993). This persists until several hours after sunrise when it is
17 eroded by turbulent mixing, generated by strong solar heating of the surface, due to Mexico City’s
18 tropical location and high elevation.

19 During the RS, the interaction of the predominant easterly winds with the mountains that
20 surround the Mexico City basin force upward motions, leading to convection. Early mornings are
21 typically clear during this season, but thermal inversions are infrequent due to the high moisture content
22 in the atmosphere. Solar heating during the morning leads to the development of turbulent eddies that
23 vertically mix the pollutants emitted at the surface. Convection develops over the mountains, resulting
24 in further dilution of pollutants due to the vertical updrafts within clouds. The precipitation that develops
25 from the convective clouds has a large gradient within the basin (Jáuregui, 1971), from around 400 mm
26 yr^{-1} in the northeast to almost 1200 mm yr^{-1} in the south-western part of the city. Figure 4 shows the
27 daily precipitation during the period of the research study, showing that in 2013 the rain began in the
28 middle of May and ended the first week of November. It rained almost every day, some days more
29 heavily than others. In this study, the RS extends from May 11 to November 7, 2013. Only days when
30 the rain exceeded 2 mm are included in the present study.

31 The concentration of gases and aerosols within the boundary layer is controlled by the balance
32 between the emission rates of primary pollutants (i.e. CO, NO_x, volatile organic compounds, VOC, and
33 eBC), the rate of secondary production due to photochemical reactions of O₃ and PM_{2.5}, and the rate

of dilution by mixing with tropospheric air. Not only does this balance vary with the season, but because the emissions patterns change from the workdays (Monday through Saturday) to Sunday, we would expect to observe differences in the trends of SLCPs when comparing these two seasons.

3.1 Seasonal Trends

The average, median, daily average maxima and the quantile values at 5%, 25%, 75% and 95% were calculated for CO, NO_x and O₃ and are shown in Figs 5a-c by season, workdays and Sundays. The box and whisker plots show median (horizontal line within the box), 25% quantile (bottom of box), 75% quantile (top of box), 5% quantile (bottom whisker) and 95% quantile (top whisker). In addition the average of the daily maxima are shown with the red, filled circles. The same information for eBC, PM_{2.5}, B_{scat} and SSA is shown in Figs. 6a-d, respectively. The values used to generate these graphs are tabulated in Table I.

Table II summarizes the average and standard deviations about the average, maximum daily values tabulated by season, workdays and Sundays. A Student's t-test was applied to the differences and the significance evaluated at a confidence level of $P < 0.10$, i.e. that there is less than a 10% probability that the observed differences are due to chance. For each season and each measured parameter the significance of the differences between the workday and Sunday values was tested using the hypothesis that the average values were from the same population. This hypothesis was rejected when the value of $T > |1.7|$, the critical value for $P < 0.10$ with the pooled degrees of freedom of the two samples in each comparison. In the Table, underlined numbers correspond to Sunday values significantly smaller than on workdays, i.e. the null hypothesis that they were from the same population was rejected. The numbers in bold indicate the cases when the Sunday values exceeded the workday values.

Using the same approach the difference in the average maximum values between each of the three seasons was tested (last three columns in Table II). The underlined numbers indicate a significant decrease in a parameter value from one season to the next and the numbers in bold highlight significant increases. The value in parentheses in each of these cells is the computed value of the t statistic.

Median concentrations of CO and NO_x (Figs. 5a and 5b, respectively) do not significantly change with season, whereas there is a clear increase in O₃ (Fig. 5c) during the WDS compared to either the RS or CDS. The maximum concentrations for all three of the gas species increase significantly (Table II) in the dry seasons (cold and warm) compared to the RS. This increase is seen for workdays as well as Sundays. A significant increase from the CDS to the WDS is only observed in the O₃ concentrations, due to the larger actinic flux late in the spring. Of note is that the maximum CO

1 concentrations on Sundays actually decrease between the CDS and WDS (numbers in underlined
2 italics in Table II).

3 A “weekend effect” is observed for CO, NO_x and O₃. The maximum CO and NO_x concentrations
4 decrease between workdays and Sundays by more than 30% during all three seasons (Table II). As
5 discussed by Stephens et al. (2008) the lack of a decrease in the O₃ concentrations on Sundays is a
6 “weekend effect” because with the decrease in VOCs (CO is a surrogate for VOCs), the O₃ was
7 expected to decrease as well. As detailed by Lawson (2003) for cities in general, and by Stephens et
8 al. (2008) for Mexico City in particular, there are several possible mechanisms responsible for this
9 behavior; however, the one that was discussed as most plausible is related to the inhibition of O₃
10 formation by NO_x under VOC-limited conditions. The production of O₃ is positively correlated with VOC
11 emissions and negatively correlated with NO_x. Since the CO (VOC) and NO_x are both decreasing, there
12 is a balancing effect on O₃ production, i.e. less production from VOC reactions but less inhibition from
13 NO_x

14 Figures 6a and b display the same grouping of comparisons as Fig. 5 but for the mass
15 concentrations of PM_{2.5} and eBC. The median PM_{2.5} concentrations increase between RS and CDS
16 and between CDS and WDS. In contrast, the median concentration of eBC does not show a seasonal
17 trend. When evaluating the seasonal trends of the maximum concentrations, both the PM_{2.5} and eBC
18 have much larger values in the dry seasons than during the rainy season; however, unlike the PM_{2.5}
19 that also shows a significant increase between the CDS and WDS, the eBC actually decreases slightly
20 between these two seasons, but not significantly (Table II). There is a clear “weekend effect” that is
21 seen in the maximum concentrations drawn in Fig 6, and highlighted in Table II. The concentrations
22 decrease from the workdays to Sundays over all seasons; however, unlike the eBC whose decreases
23 are statistically significant over all seasons, the PM_{2.5} decreases are only significant during the RS and
24 CDS.

25 The seasonal impacts on the optical properties of the aerosol particles are illustrated in Figs.
26 6c and d that show B_{scat}, and SSA, respectively. The figures suggest that the differences between
27 seasons of B_{scat} and SSA are small; however, as seen in Table II, there are statistically quantifiable
28 differences, although for B_{scat}, only the increase from RS to WDS is significant. The SSA values are
29 much more sensitive to season, even though the median values do not appear that different. The SSA
30 decreases from the RS to CDS but increases from the CDS to WDS. A decrease in the SSA can be a
31 result of decreases in B_{scat} or from increases in B_{abs}, since SSA is directly proportional to B_{scat} and
32 inversely proportional to B_{abs}; however, since B_{scat} is seen to be relatively insensitive to seasonal
33 changes, the changes with SSA are primarily a result of the seasonal sensitivity of B_{abs}, i.e. changes in

1 the eBC. It should be noted that other aerosols like certain organics, as well as dust, will also absorb
2 light, but at shorter wavelengths than the 870 nm used by the PAX. Hence, the majority of the absorption
3 measured in this study is by BC.

4 **3.2 Daily Trends**

5 Figures 7a-c show the differences in the average daily UV-A radiation (Fig. 7a), temperature
6 (Fig. 7b) and relative humidity (RH, Fig. 7c). The UV radiation is averaged over the daylight hours from
7 10:00 to 16:00 local standard time (LST) whereas the temperature and RH are hourly averages. In
8 Figs. 7b and c the red, green and blue curves differentiate the trends by season: WDS, RS and CDS,
9 respectively. There is on average 30% more UV radiation in the CDS and WDS than during the RS.
10 The WDS temperature (Fig. 7b) is generally about 5°C warmer than either the RS or CDS except in the
11 early mornings just before sunrise when the RS and WDS are the same. The WDS is always much
12 drier throughout the day (Fig. 7c) than during the other seasons while the humidity during the rainy
13 season is 30-50% higher than the dry seasons. As will be further discussed below, the seasonal
14 radiation, temperature and humidity difference are principal factors that underpin the associated
15 seasonal differences in gas and particle properties.

16 An examination of the average, hourly properties of the pollutants, differentiated by season and
17 by workday/Sunday, provides a different perspective than the comparison of just the daily quantiles
18 and maxima that are shown in Figs. 5 and 6. All of the pollutants have daily cycles that are linked to
19 the traffic patterns, photochemical processes and boundary layer depth (vertical mixing with clean air).
20 Figures 8a-c illustrate these cycles for CO, NO_x and O₃, respectively. The solid and dashed lines
21 separate the measurements into the workdays (Monday-Saturday) and Sunday. The red, green and
22 blue curves represent the seasons as described for Figs. 7b and c.

23 The average hourly CO and NO_x (Figs. 8a,b) have very similar patterns over all three seasons.
24 The maximum concentrations are reached between 7:00 and 8:00 LST during the RS and WDS and
25 between 8:00 and 9:00 in the CDS, respectively. A secondary, broader maximum is seen near midnight,
26 formed after the concentration begins increasing after the minimum at 16:00. As was illustrated in Figs.
27 5a,b, the maximum CO and NO_x values during the RS are c.a. 40% lower than the dry seasons;
28 however, when comparing the RS with the WDS, this difference appears to be dominant only from
29 about 4:00 to 11:00 LST. Outside of this time period the concentrations are nearly the same. Likewise,
30 after 13:00 LST there is no longer a significant difference in the CO and NO_x concentrations in the RS
31 and CDS. This is a result of the manner in which the vertical mixing progresses and the boundary layer
32 depth increases during the three seasons. The shift in the time of the peaks between the CDS and
33 other seasons is due to the shift in Mexico from daylight savings time (DST) the first Sunday in

1 November to standard time then back to DST the first Sunday in April; however, the measurement time
2 base does not shift with changes in DST.

3 In the RS, as discussed previously, the boundary layer growth is accelerated by the vertical
4 motions as convective clouds develop, resulting in much larger updrafts that reach higher in the
5 troposphere than during the dry season. This leads to more rapid dilution of the primary emissions. At
6 the same time, the clouds reduce the solar radiation hence limiting photolysis and the photochemical
7 reactions that produce O₃ (Fig. 8c). In the dry seasons there is also vertical mixing but not as intense
8 as during the rainy season and it is only due to dry turbulent eddies within the boundary layer that slowly
9 erode the inversion at the top by mixing with the free tropospheric air. There is less O₃ during the CDS
10 because the solar zenith angle is larger with subsequent decreased photochemical activity.

11 The minimum in concentrations at 16:00 LST, and steady increase thereafter, is due to the
12 maximum boundary layer depth in late afternoon (Stull, 1988; Perez-Vidal and Raga, 1997), followed
13 by its gradual collapse as the solar insolation diminishes and the dilution decreases while the daytime
14 primary emissions remain more or less constant. There is actually an increase in emissions from traffic
15 during the evening rush hour in Mexico City that contributes to the increase in CO and NO_x at night
16 (Edgerton et al., 1997; Shifter et al., 2003)

17 A “weekend effect” (only Sunday) is also evident during all seasons whereby the maximum
18 concentrations on Sundays are 50-70% lower than on workdays (See Table II) regardless of the time
19 of day. Given that the meteorology on Sundays is not expected to differ from workdays, i.e. the
20 boundary layer depth and solar radiation should not be dependent on the day of the week, this decrease
21 over all seasons must be due to decreased, work-related traffic on Sundays.

22 The hourly average O₃ (Fig. 8c) reaches the maximum concentrations between 13:00 and
23 14:00 LST during the rainy season and between 14:00 and 15:00 LST during the dry periods.
24 Differences in the O₃ concentrations between the rainy and cold-dry seasons are only obvious between
25 the hours of approximately 14:00 and 20:00 LST and between 10:00 and midnight LST when comparing
26 the RS and WDS. The differences in O₃ are primarily being driven by the photochemical production of
27 O₃. As summarized in Table I, the CDS maximum O₃, is 13% larger than the RS whereas the maximum
28 during the warm-dry season is 47% larger. The impact of available solar radiation on the production of
29 O₃ from the precursor gases (NO_x and VOCs) can be better understood by comparing measurements
30 of the UV-A that are shown in Fig. 7a. The summer months of June-August are a period with minimum
31 solar zenith angle, and yet, the daily averaged RS UV-A is about 20% less than the CDS and 30% less
32 than the WDS. Hence, a major fraction of the difference between RS and WDS O₃ can be explained by

1 the decreased actinic flux in the RS with the remaining difference due to the increased vertical mixing
2 produced by the cloud formation.

3 There are large differences in the hourly trends in PM_{2.5} and eBC (Fig. 9a-b), not only in the
4 hourly concentrations that vary with season, but also in the hours during which these PM_{2.5} mass
5 concentrations peak. These peak hours change with season, although to a lesser extent for the eBC.
6 The differences between the rainy and dry season PM_{2.5} maximum average mass concentrations are
7 36% and 56%, respectively, when compared to the CDS and WDS (Table II). For the eBC, these
8 differences are 38% and 48%. Whereas the PM_{2.5} concentrations during the CDS and WDS are always
9 larger than those in the RS, regardless of the time of day, the eBC concentrations in the dry seasons
10 only exceed those during the rainy season from approximately midnight until 12:00 LST. After this time
11 the concentrations are on average the same. This is a reflection of the same atmospheric process that
12 was discussed for CO and NO_x, i.e. the difference in vertical mixing between the rainy and dry seasons.
13 There is much greater dilution of the eBC during the rainy season since clouds are responsible for the
14 vertical mixing.

15 The daily cycles in PM_{2.5} have seasonal patterns that are much more distinctive than either the
16 gases (Figs. 8a-c) or the eBC. There is a slight shift in the hour of maximum concentrations for the
17 eBC, between 6:00 and 7:00 for the RS and WDS, and between 7:00 and 8:00 for the CDS. This shift
18 is only a result of the change from daylight savings time (DST) in November and April, so that in the
19 CDS the major commuter traffic begins an hour later relative to the clock used on the RAMA data that
20 doesn't change to DST. The PM_{2.5} reaches maximum concentrations between 7:00 and 8:00, 9:00 and
21 10:00 and 10:00 and 11:00, respectively for the WDS, CS, and RS. The large shift in the peaks of the
22 daily cycles are the result of both the atmospheric dynamics, i.e. boundary layer growth, and the
23 chemical processes behind the formation and growth of the particles that make up the PM_{2.5}.

24 There is a very distinctive "weekend effect" displayed in the eBC and PM_{2.5} concentrations. As
25 summarized in Table II, the maximum eBC concentrations decrease by more than 50% from workdays
26 to Sundays during all seasons. As can be observed in Fig. 9b, the difference is distinct over all hours
27 of the day but is the most predominant between from 5:00 to 18:00. One of the major differences in
28 primary emissions of BC between the workdays and Sundays is that the use of large, diesel burning
29 trucks and machinery is much less on Sundays. Although vehicular traffic in general is also less on
30 Sundays, since combustion of diesel fuel is the major contributor of BC emissions in the city, the large
31 decrease in BC is linked more to the decrease in diesel combustion than gasoline. To support this
32 assertion, looking only at the dry season to remove any effects of precipitation, the average eBC to CO
33 ratio on workdays was 3.6 µg m⁻³ of eBC to 1.0 ppm of CO. This compared to the Sunday ratio that is

1 2.1 $\mu\text{g m}^{-3}$ of eBC to 1.0 ppm of CO; hence, the eBC decreases by a much larger percentage than the
2 CO.

3 The hourly trends in the optical properties of the aerosol particles, B_{scat} and SSA, are shown in
4 Figs. 10a and b, respectively. The maximum B_{scat} falls between 9:00 and 10:00 LST during the WDS
5 and between 10:00 and 11:00 LST in the RS and CDS. Unlike the particle mass concentrations of $\text{PM}_{2.5}$
6 and eBC, there are no distinct differences from season to season in the maximum values. The
7 differences are instead seen in the morning hours before reaching the maximum, and then again in the
8 afternoon. During both of these periods the dry season values are about 30% larger than during the
9 rainy period. In the same way that the trends in $\text{PM}_{2.5}$ are a balance between the processes that drive
10 particle growth and the dynamic processes that dilute the number concentration, likewise the trends in
11 the scattering are driven by complex interactions. The intensity of light scattering and the $\text{PM}_{2.5}$ mass
12 concentrations are strongly correlated because both are proportional to particle concentration and size
13 (albeit to different power, squared and cubed, respectively). The B_{scat} and $\text{PM}_{2.5}$ values peak at the
14 same time periods during the three seasons (Figs. 9a. and 10a).

15 The SSA reaches its minimum value in the morning between 6:00 and 7:00 LST in the WDS
16 and RS and between 7:00 and 8:00 LST in the CDS. As mentioned previously, the SSA appears to be
17 much more sensitive to changes in eBC, i.e. light absorption, than in B_{scat} . Since the minimum in the
18 SSA corresponds to the maximum in the eBC this further reinforces the sensitivity of SSA to changes
19 in the eBC concentration in this environment. The changes with season of the SSA do not mirror those
20 of eBC. Whereas the maximum eBC in the dry seasons was significantly larger than the rainy season
21 (Table II), the trends in the SSA are more complicated with the RS and WDS values being nearly equal
22 until they separate after midday at 13:00 LST with the RS SSA decreasing while the WDS maintains
23 an almost constant value. These trends can be best understood by comparing the trends in $\text{PM}_{2.5}$, the
24 parameter that is correlated with B_{scat} , and eBC and observing that while the eBC decreases rapidly
25 after its morning maximum during all seasons, the $\text{PM}_{2.5}$ is decreasing much more slowly, due to
26 photochemical reactions leading to the growth of pre-existing particles (Baumgardner et al, 2004) and
27 to the secondary production of particles (e.g. de Gouw et al., 2008; Herndon et al., 2008). This
28 illustrates that whereas the minimum in SSA observed during the morning is being driven by the
29 presence of particles with eBC, the afternoon SSA is being driven mainly by changes in non-absorbing
30 particles represented by B_{scat} .

31 **3.4 Correlations and Causal Links between Co-Pollutants**

1 In addition to deducing the potential relationships between co-pollutants by comparing the
2 hourly trends in Figs. 8, 9 and 10, these links can be further investigated by calculating the cross-
3 correlations to highlight the temporal shifts that may change with season and might offer additional
4 evidence of how photochemical and thermodynamic processes drive the daily trends in gas
5 concentrations and particle properties.

6 Figures 11-13 are correlations between pairs of pollutants as a function of the lag time that is
7 varied from -10 hours to +10 hours, in 10 minute increments. The vertical black dashed line is drawn at
8 the zero lag time and the horizontal black dashed line is drawn at the crossover point from positive to
9 negative correlation. The colors of the lines denote the season using the same coding as previous
10 figures. Figure 11a illustrates the relationship between CO and NO_x and indicates that these two
11 pollutants are highly correlated ($R=0.9$) with no lag time, i.e. they are being produced at the same time
12 and likely by the same sources. The correlation between O₃ and NO_x (Fig. 11b), shows that their
13 highest correlations are at lag times of -8, -7.5 and -7 hours, for the WDS, CDS and RS, respectively.
14 The maximum correlation at negative lag indicates that O₃ is linked to the NO_x that was formed
15 approximately eight hours earlier, an expected correlation. These lag times are just another way of
16 quantifying the time differences between the peaks in NO_x and O₃ that were seen in Figs. 8b and c.
17 The one hour difference between the CDS and WDS is caused by the change in DST as was explained
18 earlier for the daily trends.

19 There is additional information contained in these cross-correlations by evaluating the time from
20 maximum to zero correlation. This time, when the correlation coefficient crosses the zero point, is a
21 measure of a characteristic time scale, referred to as the integral time scale, that is related to the rate
22 at which atmospheric processes produce, dilute or remove the pollutants. This integral time scale
23 indicates how quickly two pollutants are being de-correlated by dilution and mixing with other
24 atmospheric components. For example, there is a single, season-independent integral time scale for
25 the CO and NO_x correlation, 4-5 hours, whereas for the O₃ and NO_x there are three time scales: 5, 4.5
26 and 4 hours, respectively, for the CDS, WDS and RS.

27 The eBC and CO concentrations are highly correlated ($R>0.8$) with no lag as seen in Fig. 12a,
28 indicating that their sources are located in the same region. The high correlation between CO and BC
29 mass (derived by an alternative method) in Mexico City was already reported by Baumgardner et al
30 (2000, 2002). The integral time scales are 6, 7 and 8.5 hours during the WDS, RS and CDS,
31 respectively.

32 The cross-correlation between the eBC and PM_{2.5} (Fig. 12b) presents a more complicated
33 picture that is related to the complex relationship between the black carbon that is produced by primary

1 emissions and particle mass that is a mixture of primary and secondary processes (de Gouw et al.,
2 2009). Looking at the average maximum values in Table II, we see that eBC is approximately 20% of
3 the mass of PM_{2.5}. Hence, the good correlation at zero lag for all seasons represents the fraction of the
4 PM_{2.5} that is made up of BC. The trends in the correlation coefficients rapidly diverge as the lag time
5 increases. This reflects the secondary processes that either produce new particles or lead to the
6 increase in mass of existing particles. These secondary processes can be aqueous, photochemical or
7 a combination of both so the rate at which particles grow will depend on the temperature, relative
8 humidity, pH and UV radiation flux. Since BC is a primary particle onto which organics, sulfates or
9 nitrates can condense, there will remain a correlation between BC and PM_{2.5}. As the BC particle evolves
10 and takes on a coating or mixture of other substances, when it is measured by the PAX it will continue
11 to absorb energy and be identified as eBC, i.e. there is still a “memory” of the original particles that are
12 being emitted even over very long lag times. As Fig. 12b illustrates the integral time scales exceed 10
13 hours during all seasons. The much larger integral time scale of the rainy season is most likely due to
14 the higher humidity during this period that promotes aqueous phase reactions and particle growth.
15 During the dry seasons, the secondary processes are only related to photochemical reactions. These
16 interpretations should be considered a testable hypothesis using a chemical growth model, which is
17 beyond the scope of this current study.

18 The correlations between the particle optical properties (B_{scat} and SSA) and the eBC and PM_{2.5}
19 are displayed in Figs. 13a,b. The relationship between B_{scat} and PM_{2.5} that was discussed previously is
20 highlighted in Fig. 13a, showing a correlation coefficient >0.8 at zero lag time. The very long integral
21 time scale is a result of the same complex processes that led to the greater than 10 hour time scales
22 for the eBC and PM_{2.5} (Fig. 12b). Figure 13b underscores the earlier discussion about the sensitivity of
23 the SSA to eBC. The negative correlation is due to the inverse relationship between SSA and eBC. The
24 correlations are higher in the dry months than in the rainy season because of the additional contribution
25 to light scattering by the particles that are formed under conditions of high humidity in the RS. The
26 separation in the integral time scales is also linked to the relative rate by which the B_{scat} increases as
27 secondary processes promote particle growth in the different seasons.

28 29 **4.0 Summary and Conclusions**

30 Measurements of SLCPs and precursor gases, made over a 13 month period from March 6,
31 2013 to March 31, 2014 have been evaluated to document and explain the seasonal trends related to
32 changes in meteorology and radiative fluxes. The SLCPs that were analyzed are the eBC, O₃ and PM_{2.5}
33 along with the co-pollutant gases of CO and NO_x. The eBC data are the longest, continuous

1 measurements that have ever been conducted in Mexico City. These data extend over the three,
2 primary seasons in Mexico City, i.e. rainy, cold and dry and warm and dry, and provide the basis for
3 linking daily trends to the underlying physico-chemical processes that drive them.

4 The maximum concentrations of the gases and particles were significantly less in the rainy
5 season compared to the dry seasons. The maximum concentration changes from the rainy to dry
6 seasons were: eBC 8.8 to 13.1 $\mu\text{g m}^{-3}$ (44%), $\text{PM}_{2.5}$ from 49 to 73 $\mu\text{g m}^{-3}$ (61%), NO_x from 144 to 252
7 ppm (78%), O_3 from 73 to 100 ppb (51%) and CO from 2.5 to 3.8 ppm (75%). The primary factor that
8 leads to lower concentrations in the rainy season is morning cloud formation that produces vigorous
9 vertical motions and result in the dilution of all the pollutants. The clouds also reduce solar energy (UV
10 radiation) and decrease the photochemical reactions that produce ozone and secondary aerosol
11 particles. The other factor that will primarily impact the particle population is the precipitation during the
12 rainy season that can remove the aerosol particles by inertial scavenging. The magnitude of this effect
13 requires an analysis beyond the scope of the current study but given the almost daily rain of more than
14 2 mm, this source of particle removal could be of a similar magnitude as vertical mixing as the cause
15 for decreased particle concentrations in the rainy season.

16 A “weekend effect” is observed that is linked to changes in traffic patterns and the type of fuel
17 burned. Significant decreases (t-test) are observed between workdays (Monday through Saturday) and
18 Sundays in the concentrations of CO, NO_x , eBC and $\text{PM}_{2.5}$. This decrease is most likely a result of
19 reduced emissions by diesel burning vehicles. A significant increase is seen in the SSA because of its
20 sensitivity and inverse relationship to changes in the light absorbing eBC. A significant effect was also
21 seen in O_3 concentrations that did not change from workday to Sunday because the decreased
22 production from VOCs was balanced by the decreased inhibition by NO_x .

23 Cross correlations and the derived integral time scales were calculated between the co-
24 pollutants in order to establish the temporal links and de-correlation times. Shorter integral time scales
25 are related to rapid vertical mixing that erodes the “memory” of the pollutants that are originally coupled
26 to the same source or location where they are produced. Cross correlations between pollutants where
27 one or both can be affected by secondary chemical reactions typically have much longer integral time
28 scales since these secondary reactions extend the memory of the coupled co-pollutants.

29 The maximum concentrations of eBC measured in 2000 (Baumgardner et al., 2002) and 2006 (Marley
30 et al., 2009a,b), were 9.1 and 9.4 $\mu\text{g m}^{-3}$, respectively, compared to 8.8 and 13.1 $\mu\text{g m}^{-3}$ in the wet and
31 dry seasons between March, 2013 and March, 2014. Not only are these exceptionally large
32 concentrations of a type of particle that has been specifically identified by the World Health Organization
33 as hazardous to health, there has been no significant change in the eBC emissions, within the estimated

1 uncertainties of more than 60%, over a 15 year period. The efforts of the city government to reduce
2 some of the pollutants have been highly successful in reducing CO and O₃; however, the mitigation
3 strategies target gases more than particles. Given that the fleet of vehicles that emit BC has grown
4 from 210,000 trucks and buses in 2000 to almost 350,000 in 2014, it is encouraging that the BC has
5 remained more or less constant over this same time period, since the newer vehicles have more
6 efficient engines. This would suggest also that some of the strategies to reduce emissions are partially
7 mitigating the effects of BC despite the increase in vehicles. As of this writing, Mexico is planning to
8 adopt the European standard for diesel fuel (EURO VI) that will decrease the emissions of BC and
9 possibly counter the impact of a very likely higher number of vehicles in the future.

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Figure Captions

Figure 1 This map of the Metropolitan Area of Mexico City shows the locations of the RAMA air quality monitoring stations and the location of the supersite where the measurements were made for this paper. Map courtesy of the Mexico City government (<http://www.aire.df.gob.mx/default.php>).

Figure 2 Diagram showing the scattering (blue dashed region) and absorption (green dashed region) cells in the photoacoustic extinctions (PAX).

Figure 3 Background-corrected absorption coefficients (a) and Allan variance (b) measured by an 870 nm PAX sampling filtered laboratory air overnight in a temperature-controlled environment. The dashed line in (b) gives the variance for white noise.

Figure 4 Time evolution of the daily average rain (blue curve), along with the accumulated precipitation beginning at the first of May (red curve).

Figure 5 The box and whisker plots show median (horizontal line within the box), 25% quantile (bottom of box), 75% quantile (top of box), 5% quantile (bottom whisker) and 95% quantile (top whisker). In addition the average of the daily maxima are shown with the red, filled circles. The trends in these statistics as a function season, workdays and Sundays are plotted for (a) CO, (b) NO_x and (c) O₃.

Figure 6 The same as Fig. 5 but for (a) PM_{2.5}, (b) eBC, (c) B_{scat} and (d) SSA.

Figure 7 (a) The average, daily accumulated UV-A is shown in the bar chart illustrating the impact of clouds during the rainy season when the sun reaches its maximum elevation angle at the Mexico City latitude. The (b) average hourly and seasonal temperature and (c) relative humidity separated by season: warm-dry (red), rainy (green) and cold-dry (blue).

Figure 8 The average hourly values are shown separated by workday (solid) and weekend (dashed) and by season: warm-dry (red), rainy (green) and cold-dry (blue) for (a) CO, (b) NO_x and (c) O₃ concentrations.

Figure 9 As in Fig. 8, the average hourly (a) PM_{2.5} mass and (b) equivalent black carbon concentrations are shown differentiated by seasons and day of the week.

Figure 10 As in Figs. 8 and 9 but for (a) light scattering coefficient, B_{scat} and (b) single scattering albedo, SSA.

1 **Figure 11** These plots illustrate the correlations between (a) CO and NO_x, (b), O₃ and NO_x as a
2 function of the lag time and differentiated by season. The vertical dashed line shows
3 the correlation of the parameters at no lag time. The horizontal dashed line demarks
4 the cross over from positive to negative correlation. The lag time was varied in 10
5 minute intervals.

6 **Figure 12** As in Fig. 11, these plots illustrate the correlations between (a) eBC and CO, (b), eBC
7 and PM_{2.5} as a function of the lag time and differentiated by season.

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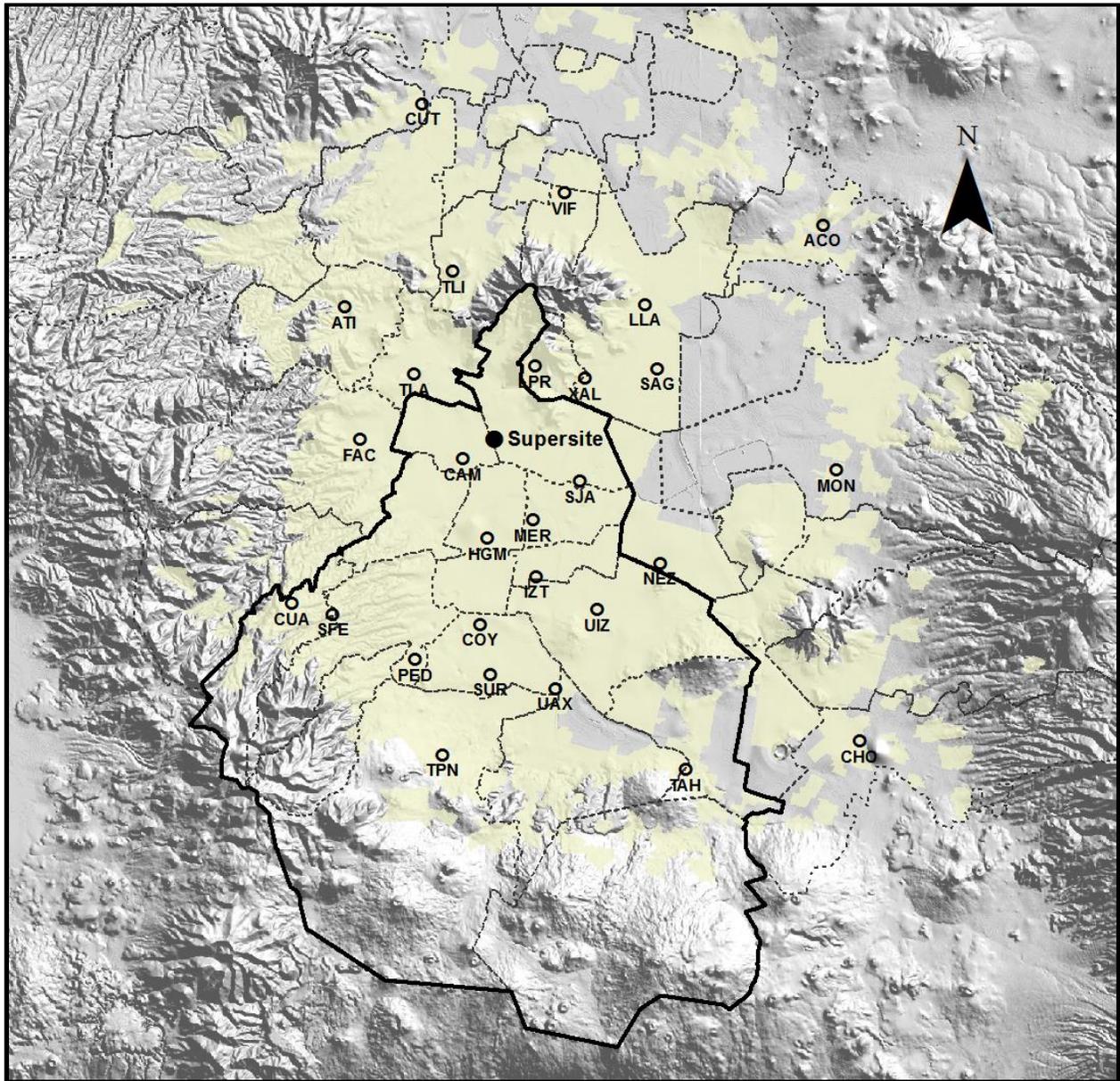
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Table II

Daily and Seasonal Maxima

Underlining indicates significant decrease at P<0.10, Bold indicates significant increase

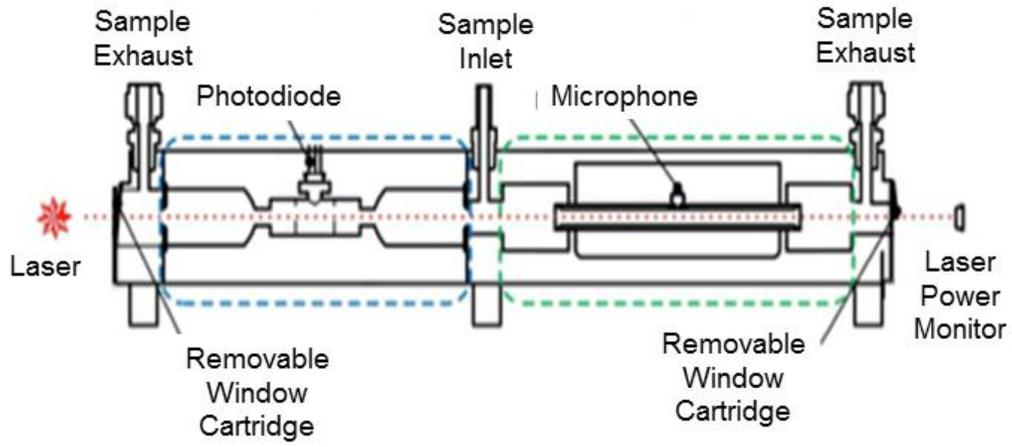
Parameter	Day of the week	Rainy Maximum (Std. - %)	Cold-Dry Maximum (Std. - %)	Warm-Dry Maximum (Std. - %)	Rainy to Cold	Rainy to Warm	Cold to Warm
					% difference (T-Test Values)		
CO (ppm)	Mon-Sat.	2.6 (61)	4.0 (64)	4.0 (50)	41 (3.9)	43 (3.3)	0 (0.3)
	Sunday	<u>1.3 (30)</u>	<u>3.5 (68)</u>	<u>2.8 (32)</u>	92 (4.0)	74 (3.1)	-21 (-1.0)
O ₃ (ppb)	Mon-Sat.	73 (38)	82 (37)	124 (25)	12 (2.2)	52 (8.1)	41 (6.9)
	Sunday	67 (43)	88 (31)	123 (13)	28 (2.1)	59 (5.0)	34 (3.8)
NO _x (ppb)	Mon-Sat.	169 (42)	304(41)	330(37)	58 (10.3)	65 (7.2)	9 (1.2)
	Sunday	<u>98 (32)</u>	<u>246 (45)</u>	<u>226 (26)</u>	87 (6.1)	80 (4.5)	-8 (-0.5)
PM _{2.5} μg m ⁻³	Mon-Sat.	47 (36)	67 (31)	83 (30)	36 (7.7)	56 (7.8)	22 (3.5)
	Sunday	<u>41 (39)</u>	<u>55 (34)</u>	<u>71 (21)</u>	30 (2.3)	54 (3.7)	26 (2.1)
eBC μg m ⁻³	Mon-Sat.	9.3 (46)	13.6 (47)	15.1 (59)	38 (6.1)	48 (2.5)	-11 (-0.7)
	Sunday	<u>6.3 (114)</u>	<u>8.7 (70)</u>	<u>11.1 (59)</u>	32 (1.1)	66(2.5)	-25 (-2.0)
SSA	Mon-Sat.	0.75 (14)	0.71 (15)	0.75 (14)	<u>-5 (-2.5)</u>	0 (0)	0 (0)
	Sunday	0.84 (11)	0.77 (15)	0.81 (11)	<u>-8 (-1.8)</u>	0.3 (0.3)	-3 (-0.3)
B _{scat} Mm ⁻¹	Mon-Sat.	103 (43)	98 (45)	95 (51)	-6 (-1.0)	8 (0.5)	2 (0.1)
	Sunday	<u>74 (48)</u>	102 (68)	71 (22)	32 (1.7)	4 (0.3)	<u>-35 (-2.2)</u>



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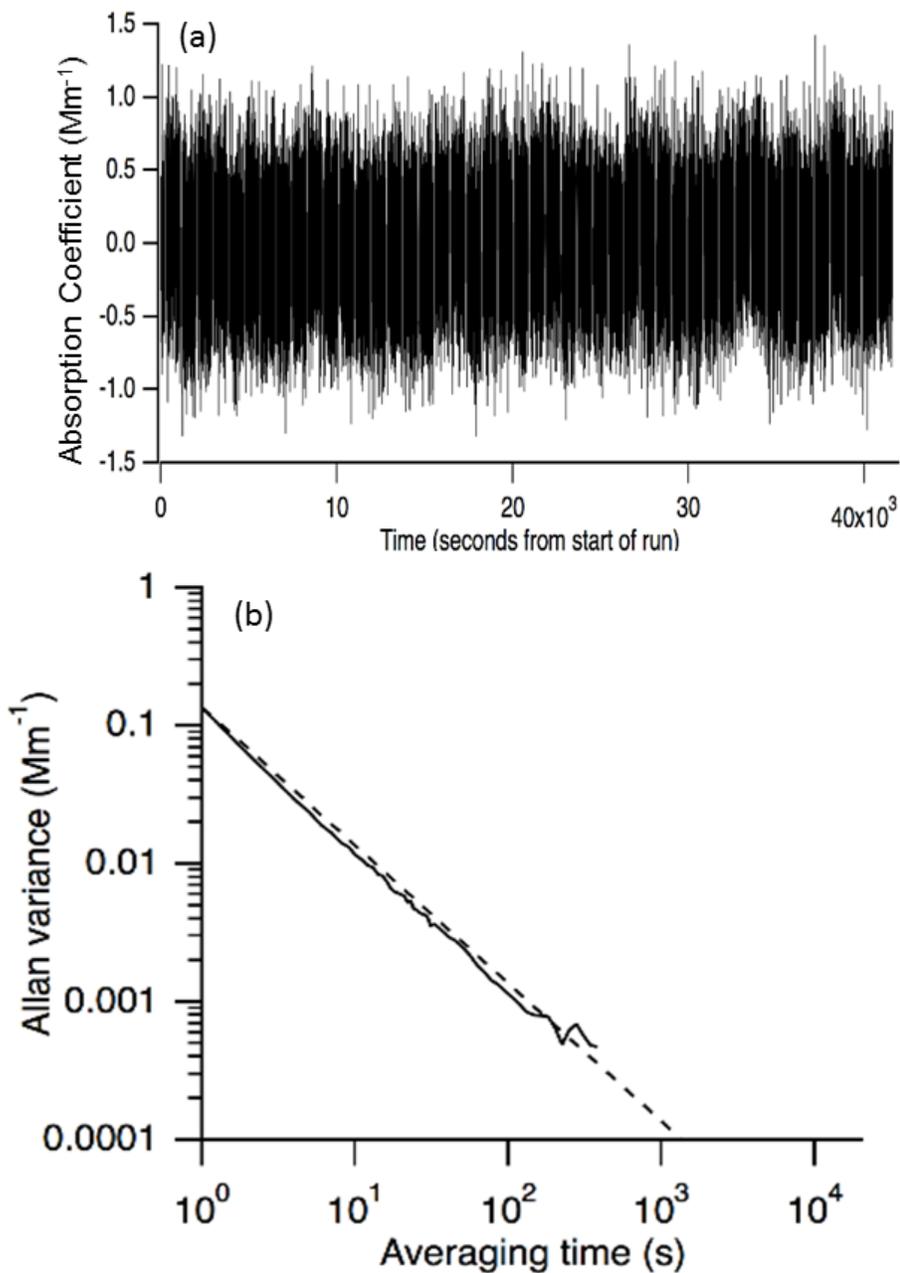
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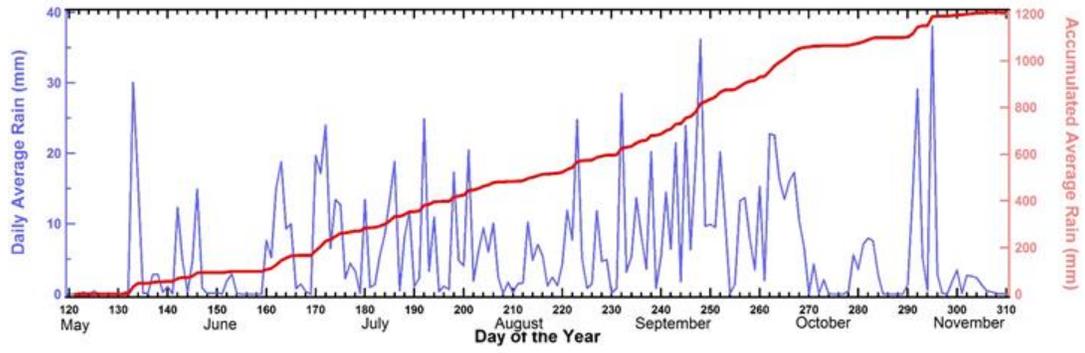
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3 **Figure 2** Diagram showing the scattering (blue dashed region) and absorption (green dashed
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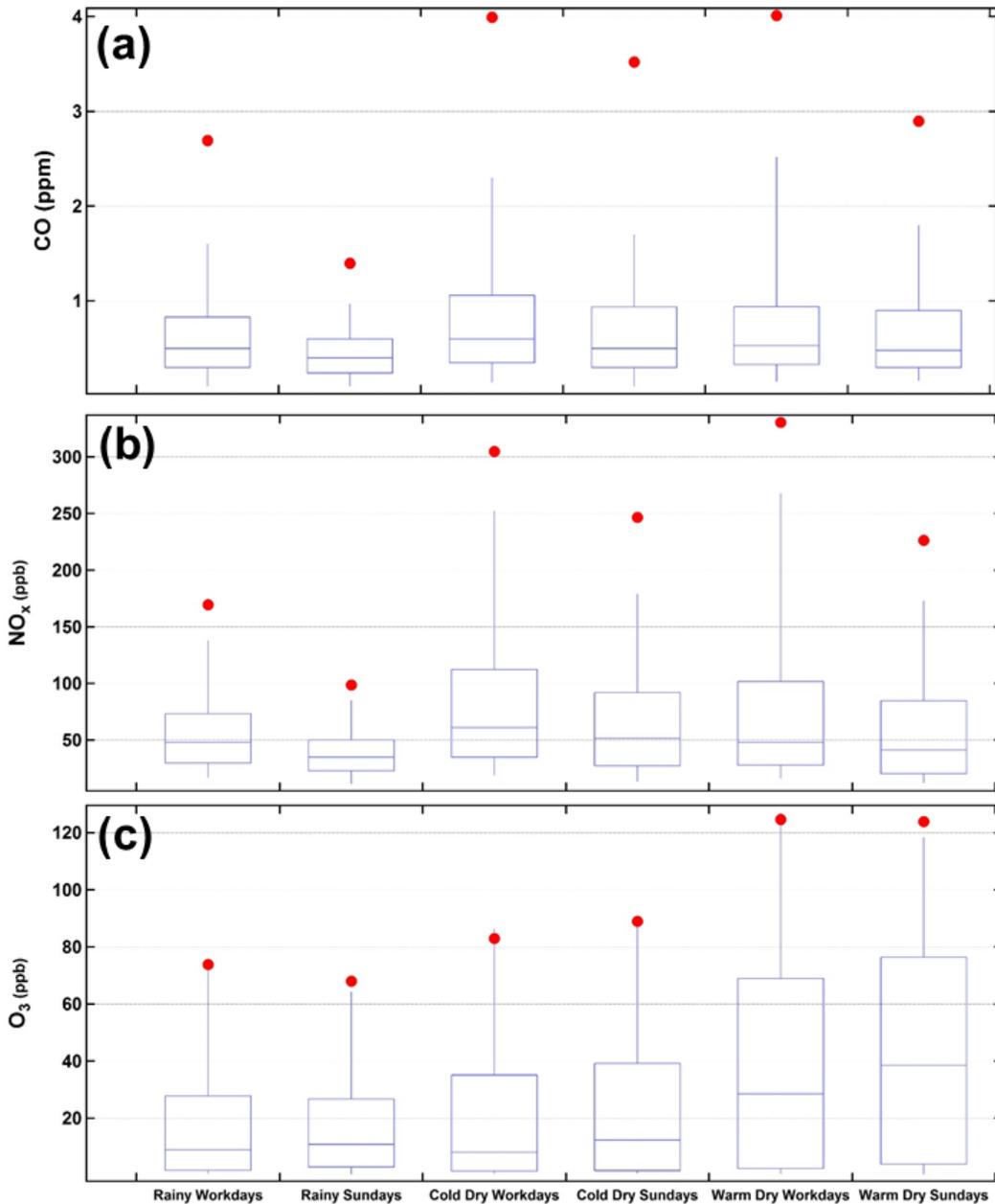


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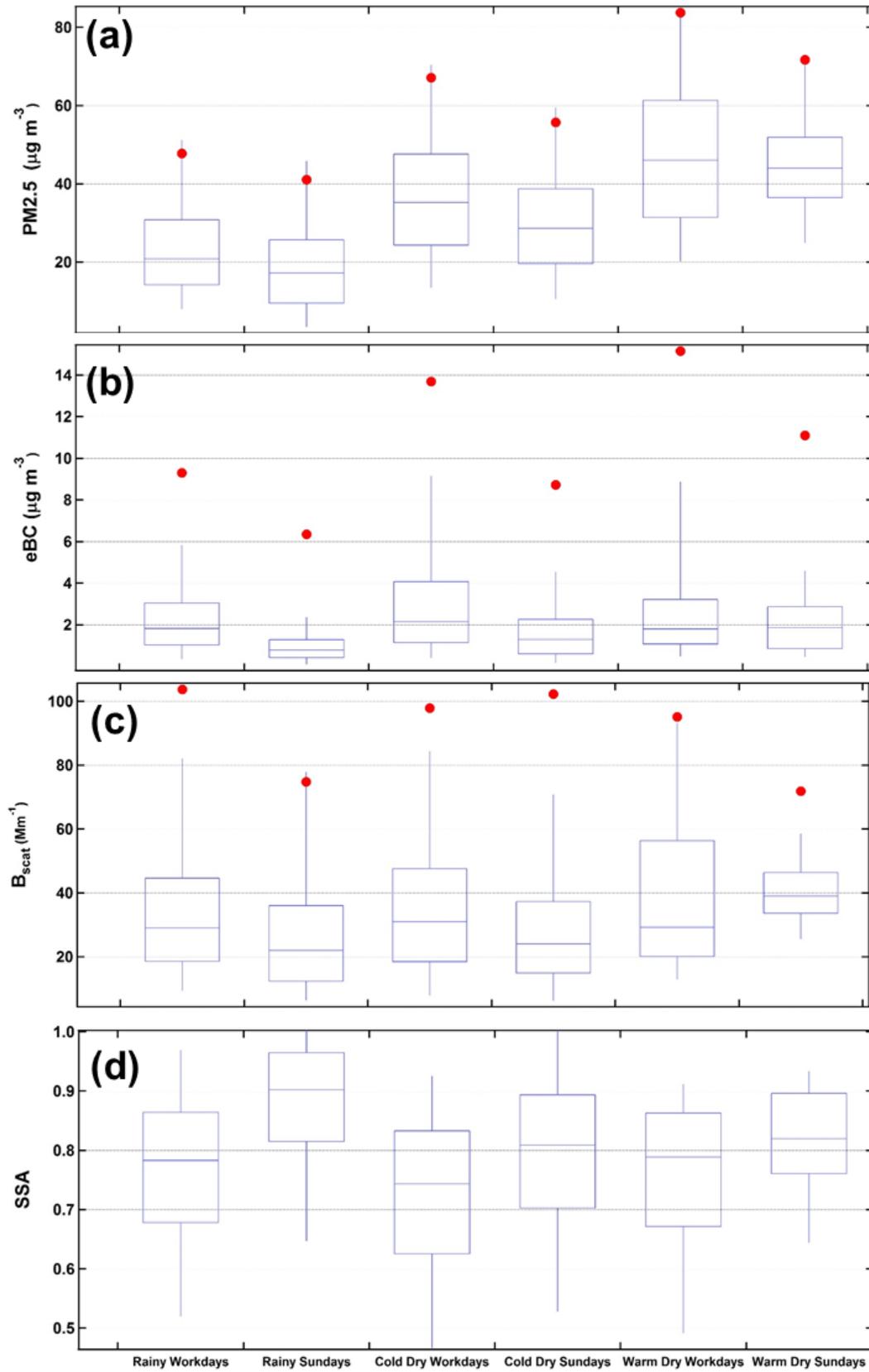
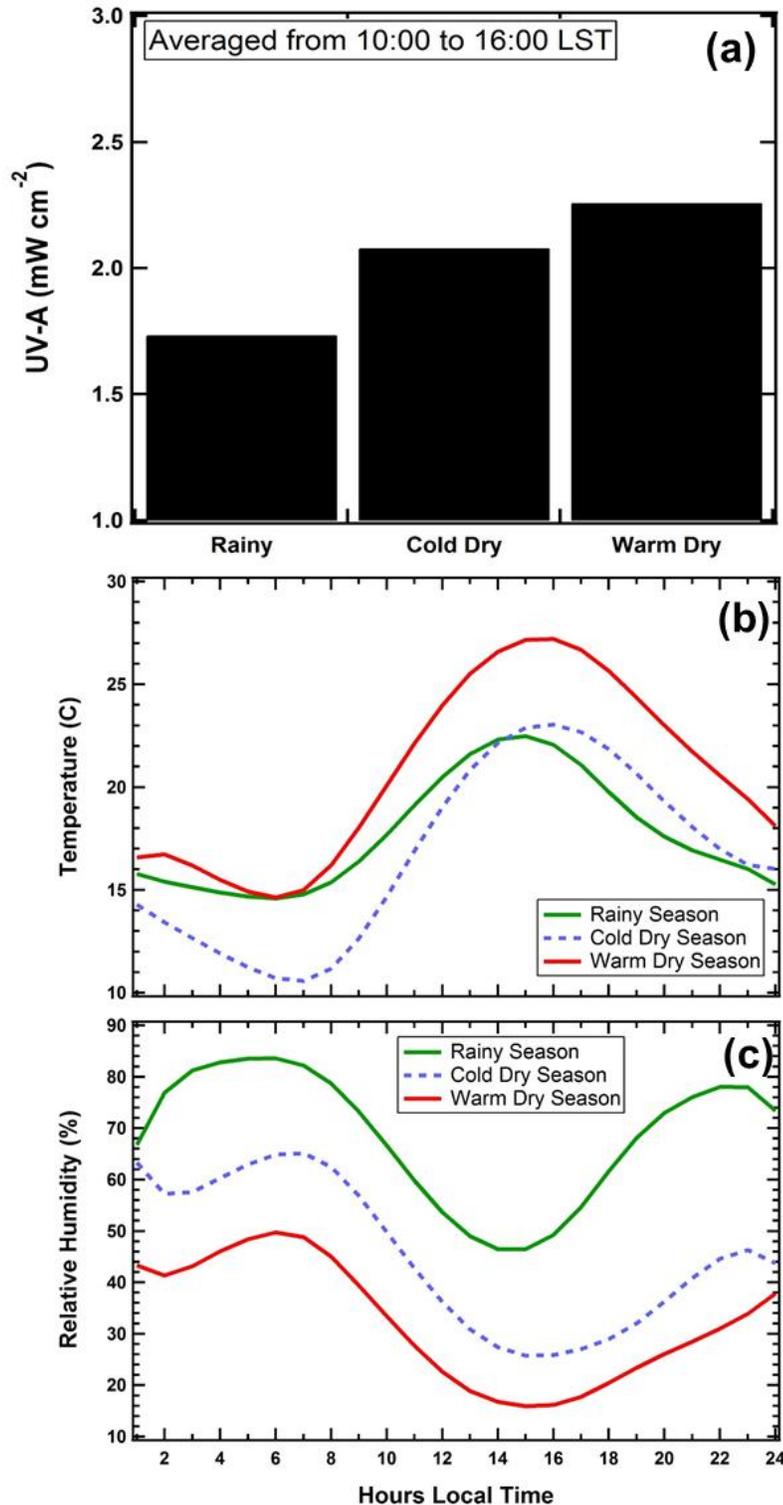


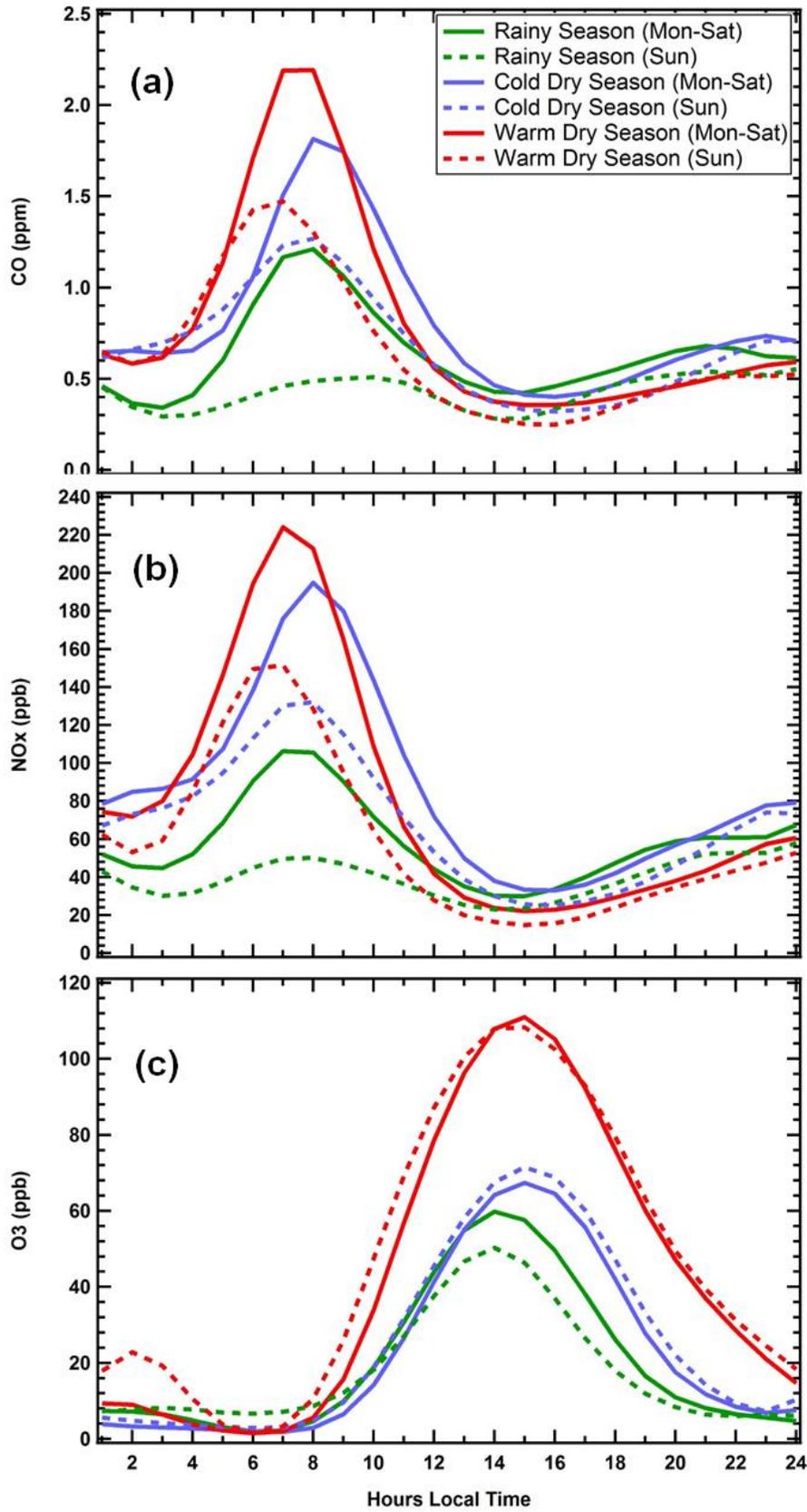
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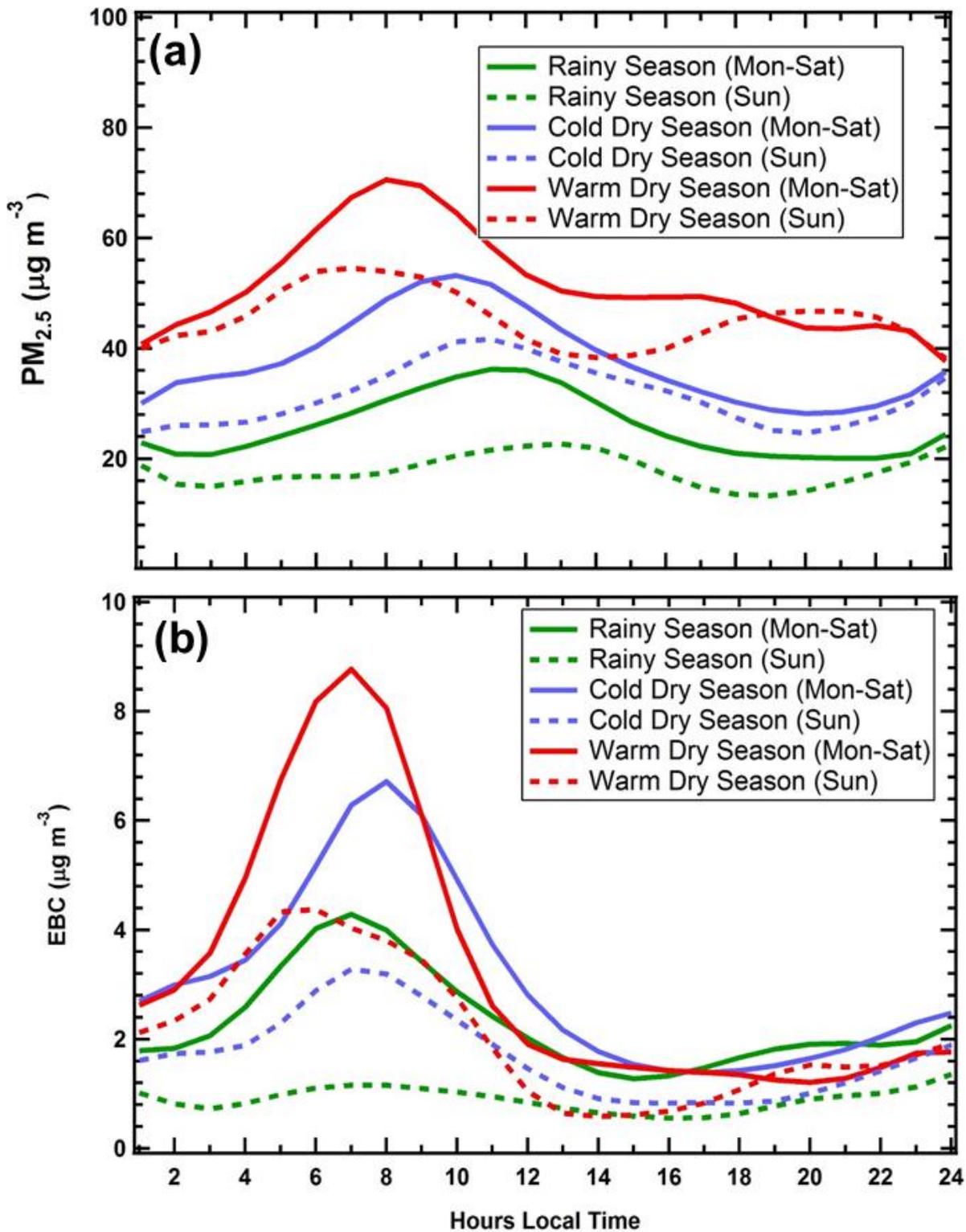


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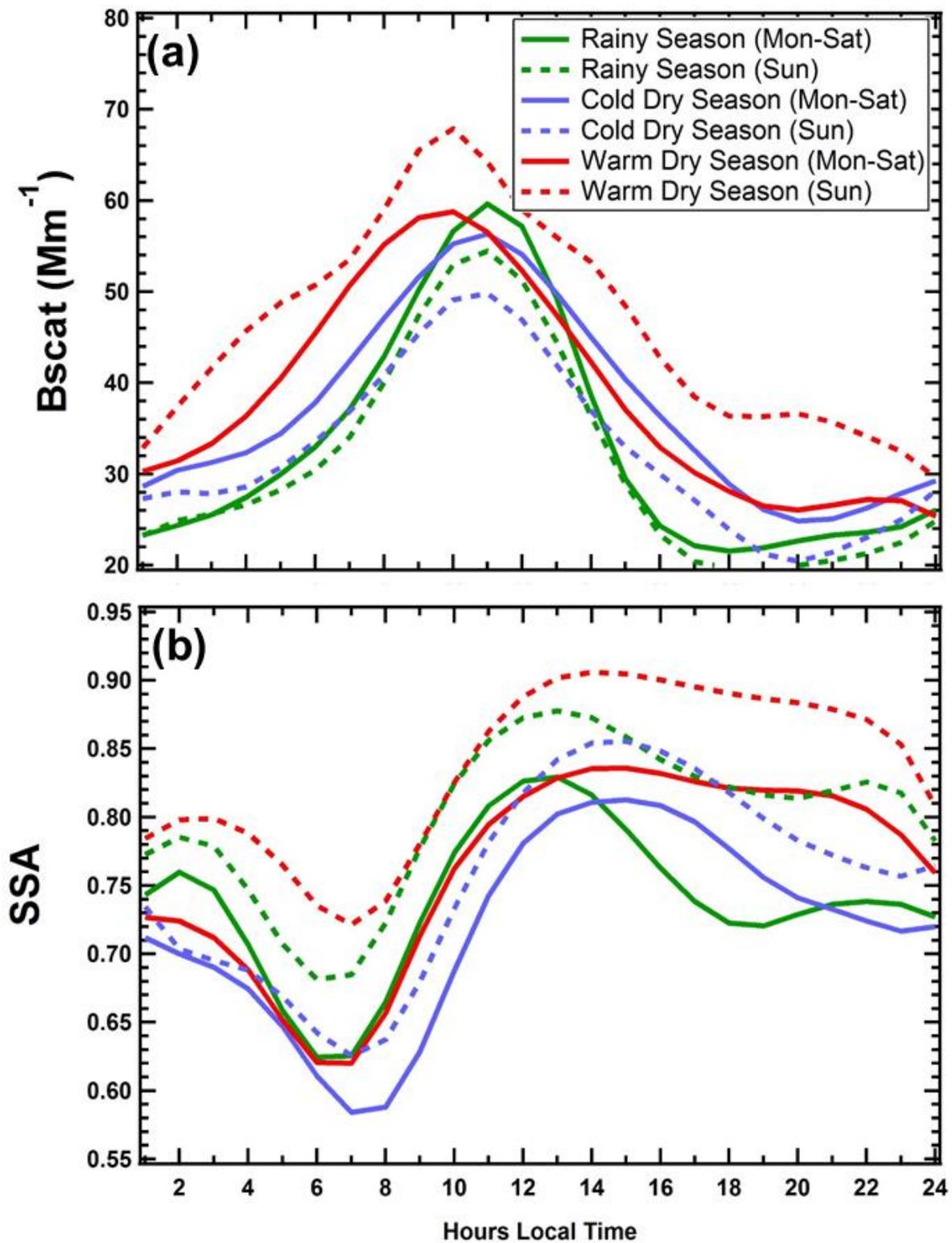
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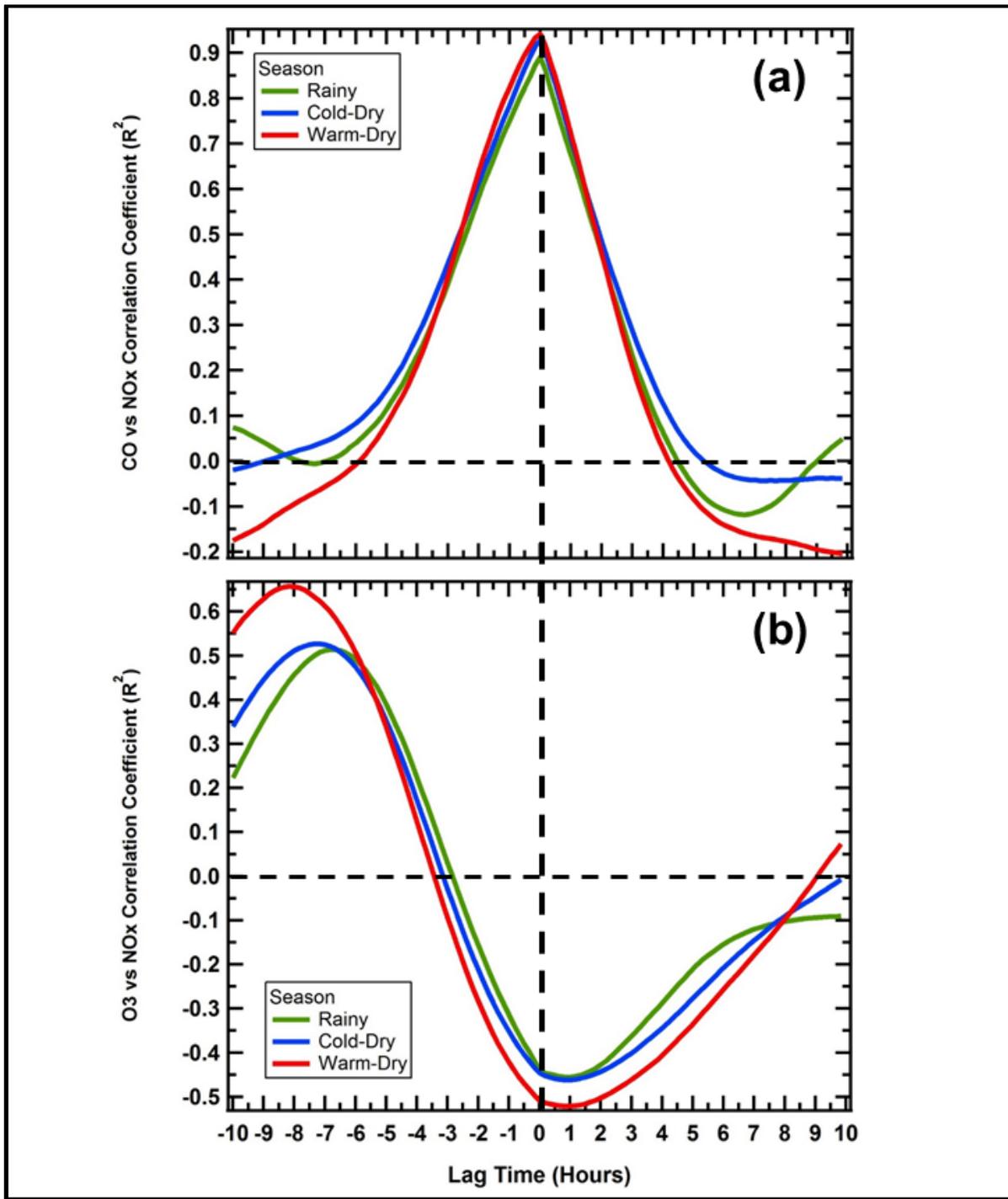


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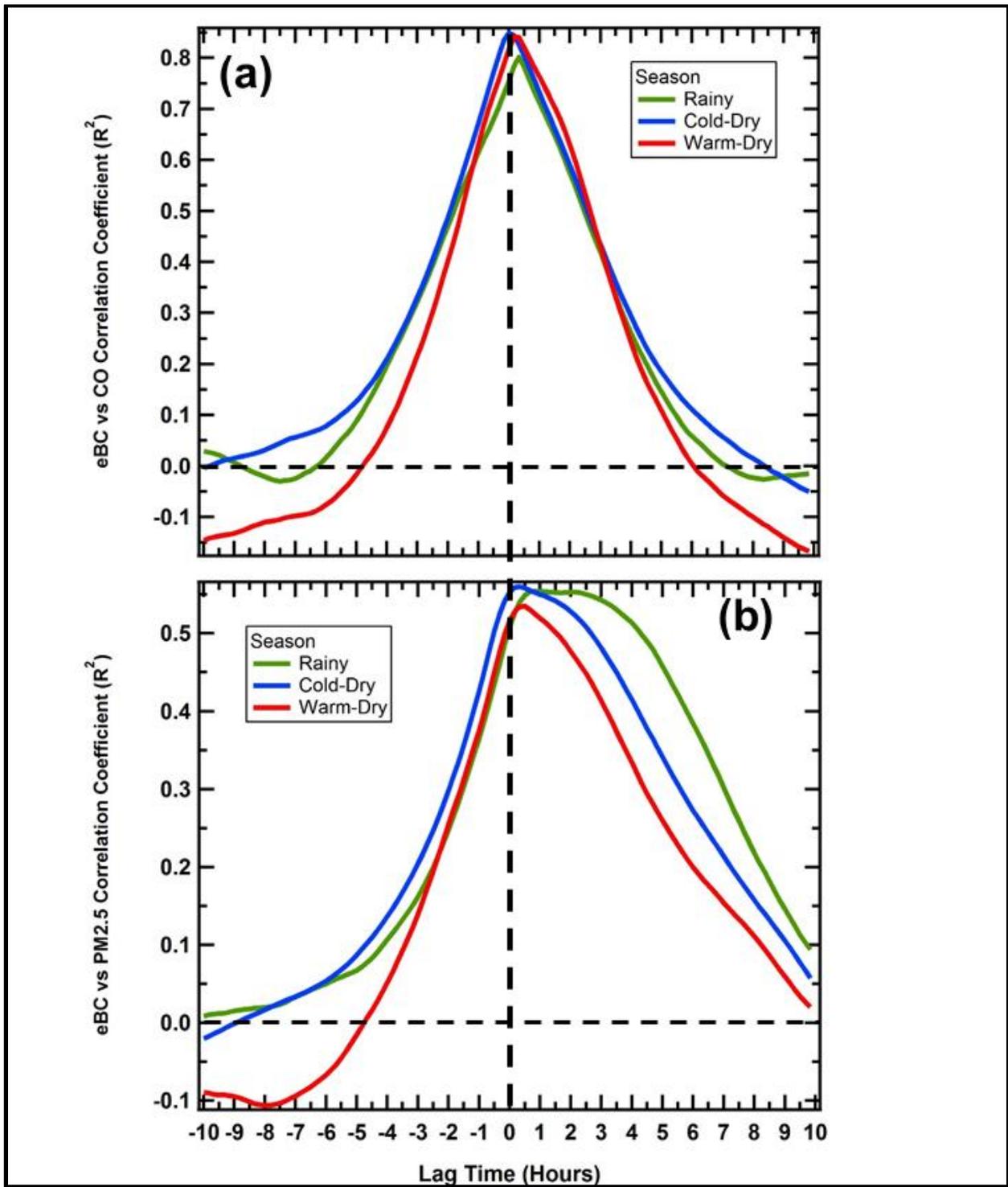


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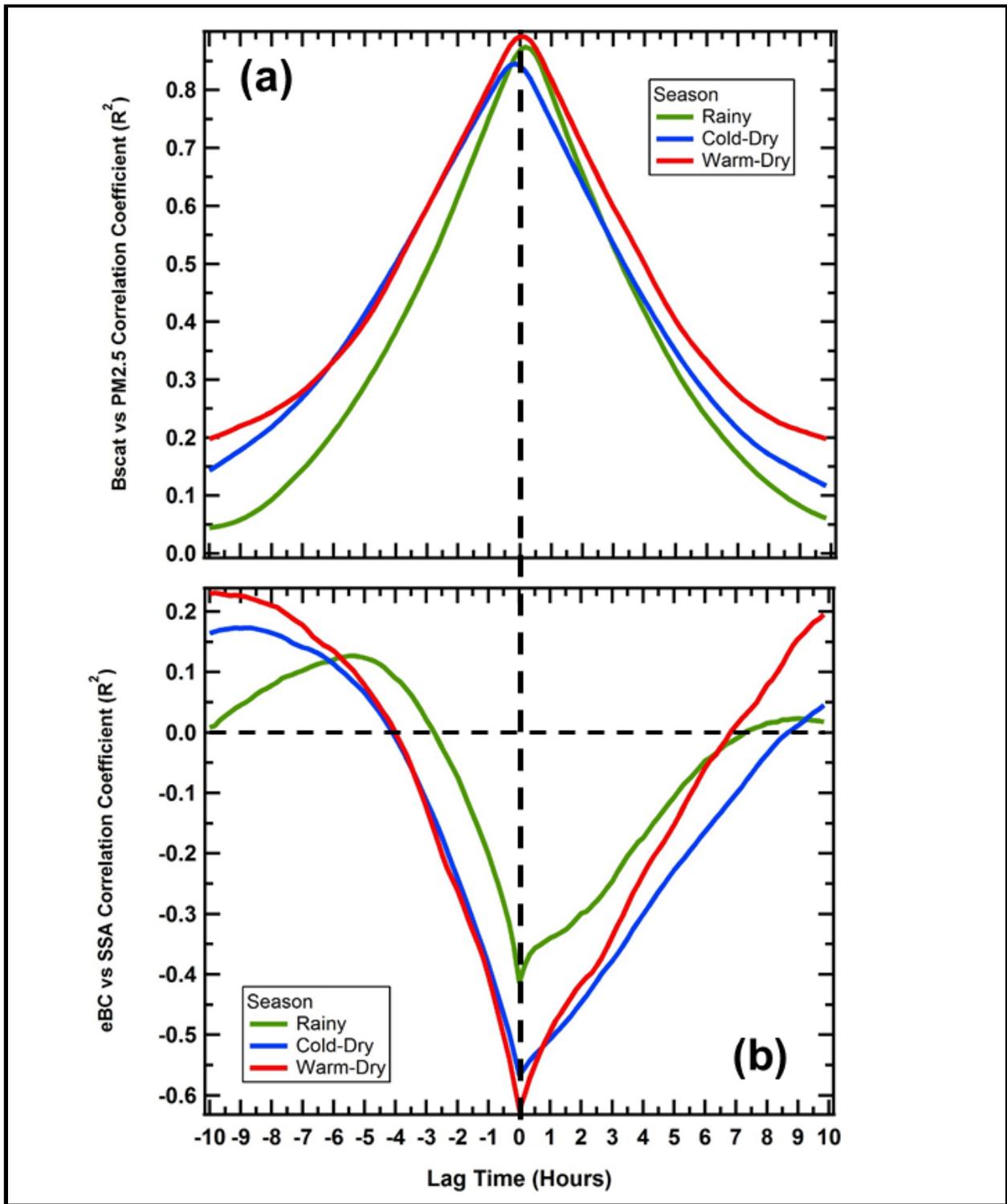


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