

Seasonal variation of aerosol water uptake and its impact on the direct radiative effect at Ny-Ålesund, Svalbard

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Abstract

In this study we investigated the impact of water uptake by aerosol particles in ambient atmosphere on their optical properties and their direct radiative effect (ADRE, Wm^{-2}) in the Arctic at Ny-Ålesund, Svalbard, during 2008. To achieve this, we combined three models, a hygroscopic growth model, Mie model and a radiative transfer model, with an extensive set of observational data. We found that the seasonal variation of dry aerosol scattering coefficients showed minimum values during the summer season and the beginning of fall (July, August and September), when small particles ($< 100\text{nm}$ in diameter) dominate the aerosol size distribution. The maximum scattering by dry particles was observed during Arctic haze period (March, April and May) when average size of the particles was larger. Considering the hygroscopic growth of aerosol particles in the ambient atmosphere had a significant impact on the aerosol scattering coefficients: the aerosol scattering coefficients were enhanced by on average a factor of 4.30 ± 2.26 (mean \pm standard deviation), with lower values during the haze period (March, April,

May) as compared to summer and fall. Hygroscopic growth of aerosol particles was found to cause 1.6 to 3.7 times more negative ADRE on the surface, with the smallest effect during the haze period (March, April and May) and the highest during late summer and beginning of fall (July, August and September).

1 Introduction

Atmospheric aerosol particles influence the Earth's energy budget directly by scattering and absorbing radiation (McCormick and Ludwig, 1967; Charlson and Pilat, 1969; Atwater, 1970; Mitchell Jr., 1971; Coakley et al., 1983) and indirectly by acting as cloud condensation nuclei and thereby modifying cloud properties (Twomey, 1977; Albrecht, 1989; Charlson et al., 1992; Hegg, 1994; Boucher and Lohmann, 1995). A better understanding of the radiative impacts of atmospheric aerosols is needed for quantifying the factors determining the Earth's energy balance and driving changes in global climate (IPCC, 2013). In this study we focus on the Aerosol Direct Radiative Effect (ADRE), whose magnitude is determined by the chemical composition, size distribution, shape, and particle concentration profiles of the atmospheric aerosols, the Earth's surface albedo and the solar zenith angle (Yu et al., 2006).

Water is an important chemical component in atmospheric aerosol particles, and thus can affect ADRE (e.g. Myhre et al., 2004). For example, it has been estimated that increasing the relative humidity (RH) from 40% to 80-90% could double the direct negative radiative forcing caused by aerosols (Piliinis et al., 1995; Fierz-Schmidhauser et al., 2010a). The water content of a given atmospheric aerosol population is determined by the ambient RH together with the composition, particularly water-solubility, and dry size distribution of the aerosol particles. In situ measurements of aerosol size distributions and optical properties, however, often take place at dry or nearly dry conditions. Therefore, to evaluate the impact of aerosol water content on ADRE, the measurements at dry conditions need to be corrected for the hygroscopic growth of the aerosol particles under humid ambient atmospheric conditions. The water uptake (hygroscopicity) of aerosol particles in equilibrium with the atmospheric water vapour can be modeled using the κ -Köhler theory (e.g. Petters and Kreidenweis, 2007), where the aerosol water uptake is represented with a single hygroscopicity parameter κ .

Numerous experimental and modeling studies have investigated the influence of RH on optical properties of aerosol particles, which is often described with the enhancement factor $f(\text{RH})$, defined as the ratio of aerosol scattering coefficient at a given RH and the scattering coefficient at dry conditions (see e.g. Zieger et al., 2010). $f(\text{RH})$ has been investigated in a number of studies at various locations (see Table 1), typically by comparing the signal of a nephelometer operated at a given RH to a corresponding instrument at dry conditions. The reported values vary from almost no enhancement ($f=1$) to a considerable effect on the optical properties ($f>3$), depending on the location.

Temperature variability and climate trends in the Arctic region tend to be more pronounced than the corresponding trends and variability for the northern hemisphere or the globe as a whole, resulting from the different feedbacks active in the Arctic environment. This characteristic feature of the climate system is referred to as the Arctic amplification and it is expected to become stronger in the upcoming decades (Serreze and Barry, 2011). The impacts of Arctic amplification can also extend outside the Arctic region (Lawrence et al., 2008). Arctic temperatures have increased at almost twice the global average rate over the past 100 years (IPCC, 2013), contributing to a continuous reduction of Arctic summer sea ice cover and surface albedo since 1979 (Serreze et al., 2007). The Arctic region thus appears to be more sensitive to greenhouse gas-induced warming than the rest of the globe. Shindell and Faluvegi (2009) also showed that the Arctic climate is particularly sensitive to changes in northern hemisphere aerosol forcing, induced both by altered particle and precursor emissions as well as atmospheric water content.

In this manuscript we investigate the seasonality of the enhancement of the direct aerosol forcing in the Arctic caused by aerosol hygroscopic growth, focusing on the year 2008. We calculate seasonal enhancement factors $f(\text{RH})$ by driving a coupled hygroscopic growth and aerosol light scattering model with measured atmospheric aerosol size distribution, composition, temperature, and RH data collected at the Mt. Zeppelin station in Ny-Ålesund, Svalbard. We evaluate the model calculations using campaign data on the hygroscopic growth and aerosol optical properties (Silvergren et al., 2014; Zieger et al., 2010). Furthermore, we investigate the influence of the hygroscopic growth on the direct radiative forcing.

2 Mt Zeppelin station, Ny-Ålesund, Svalbard

All the measurements except for the soundings and surface albedo used in this study (see Sect. 4) were conducted at the Mt Zeppelin station. The observatory is located in the Arctic on Zeppelin Mountain, close to Ny-Ålesund, in the archipelago of Svalbard at 78°54' N, 11°53' E (Fig. 1). The station is located in an almost pristine Arctic environment, away from major pollution sources. Influence from local pollution sources, such as from the nearby community of Ny-Ålesund, is also limited by the location of the observatory at 474 meters above sea level (m.a.s.l.). The unique location of the observatory makes it an ideal platform for monitoring global atmospheric change and long-range pollution transport. The observatory belongs to the Norwegian Polar Research Institute (NP) and the Norwegian Institute for Air Research (NILU) is responsible for the scientific program performed at the station (Ström et al., 2003; Tunved et al., 2013). The soundings and surface albedo measurements were conducted at the village of Ny-Ålesund.

3 Model setup

To examine the effect of hygroscopic growth on aerosol optical properties and the aerosol direct effect in the Arctic, three different models were utilized. First, we modeled the hygroscopic growth of aerosol particles in ambient atmosphere using the κ -Köhler theory (Petters and Kreidenweis, 2007). In the next step, we investigated the effect of this hygroscopic growth on aerosol particle optical properties by coupling the hygroscopic model to a Mie scattering model (Wiscombe, 1979). Finally, a radiative transfer model (Santa Barbara Disort Atmospheric Radiation Transfer, SBDART, Ricchiazzi et al., 1998) was used to look into the local effect of hygroscopicity on direct radiative effect of aerosol particles. A scheme of the models and their required inputs is shown in Fig. 2. All the input data were taken from the year 2008 from which an extensive set of chemico-physical observations was available. The three models are described in more detail in the following subsections.

3.1 Hygroscopic growth model

If the atmospheric RH is high enough, aerosol particles containing soluble material are capable of absorbing water, thus becoming saturated aqueous solution droplets (Seinfeld and Pandis, 1998). The hygroscopicity of an aerosol particle is defined by its growth factor (GF), which is the ratio between the aerosol particle diameter after absorbing water (i.e. the wet droplet

diameter), and its dry diameter. Water uptake of an aerosol particle can be modeled by the κ - Köhler theory assuming thermodynamic equilibrium between atmospheric water vapour and the aerosol particle, where the aerosol water uptake is represented with a single hygroscopicity parameter, κ . Typical values of κ vary from 0 for nonhygroscopic components to about 1.4 for highly hygroscopic salts such as sodium chloride (Petters and Kreidenweis, 2007). According to the κ -Köhler theory, the saturation ratio (S) over a solution droplet is related to the ambient relative humidity (RH) and can be described by:

$$S(D_p) = \frac{RH}{100} = \frac{D_p^3 - D_d^3}{D_p^3 - D_d^3(1 - \kappa)} \exp\left(\frac{4\sigma_{s/a}M_w}{RT\rho_w D_p}\right), \quad (1)$$

where D_d (m) is the dry diameter of the aerosol particle, D_p (m) is the wet diameter, ρ_w (kg m⁻³) is the density of water, M_w (kg mol⁻¹) is the molar mass of water, T is the temperature, R is the universal molar gas constant and $\sigma_{s/a}$ is equal to the surface tension of the solution/air interface. In the following the surface tension of pure water 0.072Jm² was applied. The total hygroscopicity parameters κ for the multi-component aerosol particles considered in this study were calculated using the simple mixing rule,

$$\kappa = \sum_i \varepsilon_i \kappa_i. \quad (2)$$

ε_i and κ_i are the volume fraction and hygroscopicity parameter of each component, respectively. RH values above 95% were fixed as 95% in the calculations, due to the uncertainties in the measurements at high values. This might lead to a small negative bias in the GFs at high RHs.

3.2 Mie model

Aerosol optical properties such as extinction coefficient (scattering + absorption) are functions of particle size, chemical composition (which defines the complex refractive index of the particle) and the wavelength of the incident light (Ouimette and Flagan., 1982). The interaction of a single spherical particle with radiation can be computed from Mie theory (van de Hulst, 1957; Kerker, 1969; McCartney, 1976). In the present study, the Mie model, MIEV0 by Wiscombe (1979) was used. The entire package of numerical code is available from the internet server <http://www.scattport.org/index.php/light-scattering-software?start=100>. The Mie model was run for the whole year of 2008 with input as defined in Fig. 2, assuming aerosol particles

as homogeneously mixed spheres. Two base cases were investigated: the “Dry” base case where RH was assumed to be 0 (and $GF = 1$), and the “Wet” base case using ambient RH and the corresponding hygroscopic growth factors (see Table 2).

3.3 Radiative transfer model

The Santa Barbara DISORT (discrete ordinate) Atmospheric Radiative Transfer model was used to calculate the solar irradiance for clear sky conditions (SBDART, Ricchiazzi et al., 1998). The investigated wavelength range covers 0.25 to 4 μm using a wavelength increment of 0.005 μm . The radiative transfer model requires the atmospheric profiles of pressure (hPa), temperature (K), water vapor density (gm^{-3}) and ozone density (gm^{-3}) (see Sect. 4.1.3 for more information). In the current setup the model also requires specification of the aerosol optical depth (AOD), single scattering albedo ω , and the asymmetry parameter g of the phase function at each atmospheric layer. These parameters were calculated using the Mie model (see Sect. 3.2) over the indicated wavelength range. The solar zenith angle was predefined in the code according to the time of the day, time of year and geographical coordinates.

Instantaneous aerosol direct radiative effect (ADRE, Wm^{-2}) can be calculated from the outputs provided by the SBDART-model. Herein, we designate a perturbation of net (downward minus upward) radiant energy by total aerosol (natural plus anthropogenic) on the surface as aerosol direct radiative effect (ADRE) while the direct radiative forcing (RF) only considers the anthropogenic components (see IPCC, 2013). A positive radiative effect indicates addition of energy to the Earth system (i.e., a warming effect) whereas a negative effect indicates a net loss of energy (i.e., a cooling effect). Daily values of the ADRE were calculated based on the Dry and Wet base case calculations (see Table 2) from 8:00 am to 12:00 pm local time (to correspond to the timing of the RH soundings) for six days (1st, 5th, 10th, 15th, 20th, 25th) of each month. We focused on the radiative forcing in the morning due to the availability of the RH soundings which were reported around 10:00 am to 12:00 pm for each day. Monthly averages based on these six values were constructed from March to September, which were the months with sufficient sunlight available. These are the months with daytime solar radiation higher than a threshold of 10 Wm^{-2} (see sect. 4.1.4 for more details).

4 Experimental data

In the following subsections we describe the measurements used as inputs for the models (see Fig. 2, Sect. 4.1) or model evaluation (Sect. 4.2).

4.1 Model input data

4.1.1 Aerosol size distribution and relative humidity measurements

The aerosol number size distribution measurements (between 10 and 790 nm) have been conducted since March 2002 at Mt Zeppelin (Tunved et al., 2013), using a closed loop Differential Mobility Particle Sizer (DMPS) with a medium size Hauke differential mobility analyzer (DMA) (Knutson and Whitby, 1976; Jokinen and Makela, 1997). The particles are counted using a condensation particle counter (TSI3010). In the present study, one year (2008) of hourly averaged aerosol number size distributions was used. The surface ambient RH measurements were obtained on an hourly basis using Relative Humidity Sensor 3445-Aanderaa (sensor operated by NILU).

4.1.2 Aerosol chemical composition

To calculate the hygroscopic growth of aerosol particles, aerosol chemical composition determined from filter measurements was used. Chemical speciation was made using two different observational data sets: one for the division between organic and elemental carbon (OC/EC) and inorganic aerosol components, and one for attaining the composition of the inorganic aerosol fraction.

First, aerosol particles ($D_a < 10 \mu\text{m}$) were collected at the Zeppelin station on a monthly basis from 1 September 2007 to 9 September 2008, using a Sierra Andersen (Sierra Instruments Inc.) high-volume sampler equipped with a PM₁₀ inlet and operating with an air flow rate of approximately $1.7 \text{ m}^3 \text{ min}^{-1}$. Whatman sheets quartz filter Grade QM-A 20×25 cm (8×10 inches) were used. All filters were preheated to 800 °C over 12 hours before sampling. Filters were extracted in 200 ml of Milli-Q water and 6 % of the extract was removed for H-TDMA analysis. The filter samples were analyzed for the organic and elemental carbon (OC/EC) concentration using a Sunset Laboratories Thermo-Optical Transmittance Carbon Aerosol Analysis Instrument (Wallén et al., 2010).

1 Subsamples of each filter (1.5 cm^2) were analyzed for OC/EC before and after extraction in
2 Milli-Q water (2 ml cm^{-2}). The OC remaining on the filter after extraction was considered as
3 Less Water Soluble OC (LWS-OC). The difference of the amount of OC between non-extracted
4 and the extracted filter subsamples is an indirect way to measure the water soluble organic
5 carbon, and was denoted as More Water Soluble organic carbon (MWS-OC). MWS-OC was
6 also determined directly on subsamples of the 200 ml water extracts and an average of the
7 methods was used in the following work (Silvergren et al, 2014). These analyses provided us
8 with monthly mass fraction of inorganics, MWS-OC, LWS-OC and EC. The OC/EC
9 composition for the period from 10 September 2008 to 31 December 2008 was assumed to be
10 the same as for the corresponding period during the previous year.

11 In the next step, the inorganic fraction was assumed to consist of a sulfate (NO_3^- , NH_4^+ , SO_4^{2-} ,
12 Ca^{2+} , K^+) and a sea salt (Na^+ , Cl^- , Mg^{2+}) fraction. The fractions were determined using daily
13 samples, collected with an open face filter pack system (no particle size cut-off, but shielded
14 by a cylinder, which reduces the sampling efficiency of particles larger than $10 \text{ }\mu\text{m}$) and
15 analyzed by Ion Chromatography (Hjellbrekke and Fjæraa, 2010; Aas et al., 2009; Ström et al.,
16 2003).

17 The final chemical aerosol components are thus: OC (divided into MWS-OC and LWS-OC),
18 sulfate, sea salt and EC. The physical and chemical properties of these components needed as
19 input in the model calculations are presented in Table 3. For sulfate and sea salt we assumed
20 the properties of ammonium sulfate and sodium chloride, respectively. The averaged chemical
21 composition (Fig. 3) is dominated by inorganics, the contribution of EC to aerosol composition
22 is very small ($< 2\%$) throughout the year. This implies that the aerosol light extinction is
23 dominated by the scattering over the absorbing component (see the refractive indices of the
24 chemical components in Table 3).

25 Besides assuming the OC/EC division to be similar in the falls of 2008 and 2007, internally
26 mixed aerosol particles with homogenous chemical composition over the whole size range were
27 assumed. While these are certainly simplifications, it has been shown in previous studies that
28 Arctic aerosol particles at Ny-Ålesund, Svalbard are largely internally mixed, at least in March
29 and April (Covert and Heintzenberg, 1993; Engvall et al., 2009). Also, as shown later in this
30 work, the size dependence of the chemical composition does not appear to be a major factor
31 dominating the optical properties and the direct radiative effect of the aerosol.

4.1.3 Vertical profiles

Atmospheric profiles of pressure, temperature, RH and ozone were estimated using a combination of available daily routine radio soundings performed at Ny-Ålesund by the Alfred Wegener Institute (AWI) and standard atmospheric profiles for polar summer and polar winter (<http://www.atm.ox.ac.uk/RFM/atm>). The SBDART model is divided into 60 vertical levels. For the first 40 levels the increment is 0.5 km, and above this the increment is 20 km for each layer. By using linear interpolation, the various profiles were harmonized to match the vertical levels used in SDBART.

Since no direct measurements on the vertical profiles of aerosol particle number distributions were conducted in Ny-Ålesund, we assumed a vertical scale factor that relates the aerosol number concentrations at a given altitude to the surface measurements (at 474 m.a.s.l.). The vertical profile of the aerosol particle number size distribution was estimated based on mean extinction coefficient profiles obtained from observations with the spaceborne Cloud-Aerosol Lidar and Infrared Pathfinder Satellite Observations (CALIPSO) lidar over the Arctic (Di Pierro et al., 2013). Winker et al. (2013) present Arctic extinction profiles that show an exponential decrease with height. The Zeppelin observations were considered as being representative for the lowermost kilometer of the atmospheric column. Above this height, we scaled the in-situ findings by assuming an exponential decay in aerosol concentration with height. This leads to a scale factor that is unity at the height of Zeppelin station and decreases exponentially to zero at 10 km height. The chemical composition was kept the same for all vertical layers.

4.1.4 Surface albedo

Surface albedo data were taken from ground-based measurements at Ny-Ålesund using CMP11 pyranometers at 11 m.a.s.l (<http://doi.pangaea.de/10.1594/PANGAEA.808703>). During the polar night which starts and ends around mid-October and mid-February, respectively, no albedo measurements were available. Data was further reduced by only allowing measurements exceeding 10 W m^{-2} . This value is chosen to be approximately five times more than the typical variation from the instrument zero point. The daily mean values for the year 2008 from March to September were used as input to SBDART.

4.2 Model evaluation data

4.2.1 H-TDMA measurements of aerosol particle hygroscopic growth

The hygroscopic growth calculations were evaluated using data collected with a Hygroscopic Tandem Differential Mobility Analyzer (H-TDMA) between September 2007 and August 2008. TDMA was first introduced by Liu et al. (1978) as a technique to study the change in particle diameter as a response to changes in surrounding conditions (i.e. temperature or humidity). H-TDMA instruments have successfully been used in a multitude of studies to investigate particle size changes associated with changes in humidity (e.g. Sekigawa, 1983; McMurry and Stolzenburg, 1989; Swietlicki et al., 2008; Nilsson et al., 2009; Achtert et al., 2009). In the current study water extracts of the monthly filter samples of aerosol particles were analyzed by an H-TDMA by atomizing the extracts and measuring the hygroscopic growth factor of the dried 80, 90, 100, 110 and 120 nm particles. Note that the growth factors inferred from the H-TDMA measurements do not represent the size-dependent chemical composition at the Zeppelin site, but rather an average bulk composition. During the measurements, the humidity was set to approximately 90% RH in the second DMA, and the temperature was set to 293.15 K. Each scan took 300 seconds and at least four scans were averaged for each size bin (Silvergren et al., 2014).

4.2.2 Dry scattering coefficient

The Mie calculations (see Fig. 2) for dry aerosol particles were evaluated using data from a 3-wavelength integrating nephelometer (TSI Inc., Model 3563) operated at wavelengths 450, 550, and 700 nm (Anderson et al., 1996) throughout the year 2008 at almost dry conditions, with the RH inside the instrument below 20%. The scattering coefficients were averaged over 10 minutes.

4.2.3 Wet scattering coefficient

A field campaign was carried out at the Zeppelin station from July 15th to October 12th 2008, where a humidified nephelometer, hereafter referred to as the wet nephelometer, was used to measure light scattering coefficients at 450, 550, and 700 nm. The RH was changed in the instrument between 20% and 95% (Zieger et al., 2010; Fierz-Schmidhauser et al., 2010b). The wet nephelometer measurements were used to evaluate the Mie model in the humidified conditions. Furthermore, an estimate for the GF could be back-calculated from comparing the

predicted scattering enhancement factors $f(\text{RH})$ for different hygroscopic growth factors to the measured values of the humidified nephelometer (see Zieger et al., 2010 for the procedure).

5 Results and discussion

5.1 Model evaluation

5.1.1 Monthly HTDMA growth factor measurements vs. the hygroscopic model

To evaluate the hygroscopic growth model, monthly growth factor measurements were compared to model calculations for the period September 2007 to August 2008 (Fig.4). The RH in the model was set to 90% and the temperature to 293.15 K, i.e. the same as in the HTDMA set-up, and the averaged growth factors for a particle size range of 80-120 nm were calculated for each month. The model results show a very good agreement with the measurements for autumn and early winter (September-January) with the predicted values within about 2% of the measurements, but a positive bias of 4% to 15% for spring and summer (February-August). The good agreement for fall and winter gives confidence on our assumed chemical composition during this time, and is probably due to the dominance of sea salt in the total κ -value and thus the GF. During the other months, the sulfate and organic fractions are larger, leading to larger uncertainty in the assumed κ -values. The large deviation for the June sample is probably due to the fact that the high-volume sampler was out of order during part of June. This period coincided with high sea salt concentrations, causing apparent difference between the average predicted and measured GFs. The most likely explanation for the other discrepancies is the simplifications we have made regarding the chemical composition. For instance, due to the lack of information on the size dependence of chemical composition, we assumed a homogenous chemical composition over the whole size range. On the other hand, the H-TDMA data have been collected using dissolved, atomized and dried filter samples, thus yielding particles size and composition distributions that might be different from the ambient aerosol. Furthermore, while the HTDMA instrument had a size range of 80-120 nm the filter samples included contributions from considerably larger particles.

Previous studies on the seasonal trends of chemical composition at several monitoring sites in the Arctic with marine influence have showed a winter/early spring increase in sulfate (Radke

et al., 1984; Quinn et al., 2007), maximum concentration of submicrometer sea salt from November to February and maximum concentration of supermicron sea salt during summer months (Quinn et al., 2002). The hygroscopic growth model is very sensitive to the amount of inorganics due to their relatively high hygroscopicity parameter κ . Assuming the same relative amount of sea salt and sulfate in all particles throughout the year can explain the overestimation of growth factor calculations by the model for the size range of 80-120 nm compared to the HTDMA measurements. Considering these uncertainties, the agreement between the modeled and measured growth factors is reasonable.

5.1.2 Dry scattering coefficient measurements vs. the Mie model for the year 2008

Due to the low contribution of EC (see Fig. 3), typically less than 2%, the aerosol extinction coefficient is in practice equal to the scattering coefficient. The comparison between the dry scattering coefficients calculated with the Mie model for the Dry case (see Table 2) and those measured with the dry nephelometer is presented in Fig. 5a. The modeled and measured scattering coefficients show a good agreement ($R^2=0.95$). For most of the days the modeled scattering coefficients are within 20% of the measured values (see Fig. 5b), which gives confidence in modeling the optical properties of the aerosols using Mie theory.

5.1.3 Wet scattering coefficient measurements vs. the Mie model during the campaign

The comparison between the calculated and measured wet scattering coefficients during the campaign is presented in Fig. 6a. The calculated and modeled coefficients show a reasonable agreement with $R^2=0.64$. The histogram in Fig. 6b shows that for most of the days the deviation between the modeled and measured scattering coefficient is less than 40%, with an average bias of -10%. This negative bias is probably explained by particles > 790 nm not covered by the DMPS-based size distribution that we used as an input for the model – thus not accounting for potential contribution from the coarse mode.

The average modeled enhancement factor $f(RH=85\%)$ during the campaign period was 4.03 ± 0.50 (mean \pm standard deviation), which is higher than 3.24 ± 0.63 reported in Zieger et al. (2010). One possible reason for this could be an overestimation of the apparent hygroscopicity

(i.e. sea salt only attributed to the fine mode below 790 nm) leading to an overestimation of the resulting $f(\text{RH})$ (see also Zieger et al., 2013). Another reason for this bias could be the different dry scattering coefficient data used in the studies. The different dry values can be partly due to the different operating conditions, and partly due to different inlet structures and resulting losses – particularly for the coarse mode. The measured size distribution and dry nephelometer data were taken from instruments connected to the SU (Stockholm University) inlet (without a size cut), while Zieger et al. (2010) performed their measurements on their own total inlet. However, their scattering coefficient and size distribution measurements were approximately 25% higher compared to the SU inlet (see Zieger et al, 2010 for more details).

Zieger et al. (2010) parameterized their measured $f(\text{RH})$ -factors by an empirical γ -fit. In addition, they used their measured $f(\text{RH})$ and size distributions together with an assumption about dry refractive index to retrieve the apparent hygroscopic growth factors. These growth factors retrieved from the humidified nephelometer measurements were compared to the averaged growth factors (diameter >100 nm), calculated using the hygroscopic growth model (see Fig. 7). The deviation of our model calculations from these retrieved growth factors during the campaign is between -5% and 10%, which is in line with the comparisons to the H-TDMA data.

5.2 Seasonal variations in 2008

5.2.1 Relative humidity and hygroscopic growth factors

The seasonal variation of the RH measurements and the modeled GFs for the year 2008 are presented in Fig. 8. The RH measurements show no clear seasonal trend, except during March and the beginning of April when average RH values are in general lower (<80%) compared to the rest of the year. RH varies significantly not only from day to day but also during the day. The error bars in Fig. 8a indicate the standard deviation for each day. The low RH values coincide with the Arctic haze period (see e.g. Tunved et al., 2013 and references therein) and the smallest sea salt fraction in the particles (see Fig. 3), when polluted air masses from lower latitudes are transported to the Arctic. The annual variability in RH values during 2008 is similar to observations for other years as well.

The daily averaged GF calculated with the hygroscopic growth model follow the behavior of the RH, as expected (see Fig.8b). To separate the effect of RH and chemical composition on

growth factor calculations, we also looked into the modeled GF at a fixed relative humidity (85%) and dry diameter (200 nm) (see Fig. 8c). These results suggest that the particles were less hygroscopic during spring (March-May) as compared with other seasons (June-February). Comparison between Figs. 8b and 8c shows, however, that while RH is the main parameter controlling the magnitude of the ambient growth factor values, the chemical composition plays an important role in affecting the seasonal variation of the hygroscopic growth.

The annual mean GF(RH=ambient) and GF(RH=85%) averaged over the whole size distribution were calculated to be 1.64 ± 0.28 (mean \pm standard deviation) and 1.60 ± 0.05 , respectively.

5.2.2 Number size distributions, scattering coefficients and enhancement factors

Seasonal variations of measured aerosol number size distributions, and modeled scattering coefficients σ_{sp} and enhancement factors $f(\text{RH})$ are presented in Fig. 9.

Figure 9a shows the domination of particles larger than 100 nm during the haze period (March, April and May) and high concentrations of particles smaller than 100 nm during summer (June, July and August). The winter period from October to February displays extremely low particle concentrations. The same type of seasonal variation can be distinguished in size distribution measurements from the Zeppelin station for other years as well (Tunved et al., 2013).

Figure 9b shows a clear seasonal variability in dry σ_{sp} calculated by the Mie model, with minimum values during late summer and the beginning of fall (July, August and September). These low values are most likely related to the low concentration of particles larger than 100 nm in diameter. Summer is followed by a moderate increase of dry σ_{sp} towards fall and winter. The gradual increase continues until March and is then followed by a more abrupt increase. The maximum dry scattering is observed during March, April and May, associated with the increase in number concentration of larger particles (diameter > 100 nm). The overall seasonal changes in the scattering coefficients are similar for the wet (ambient RH) and the dry (RH=0%) case, except for late August and early September when the wet σ_{sp} is almost as high as March, April and May.

The enhancement factor $f(\text{RH=ambient})$ displays less distinct seasonal variation than the scattering coefficient itself (see Fig. 9c) although there is a tendency of systematically lower

values during March to early April. These low values coincide with both less hygroscopic aerosol particles and lower values of atmospheric RH as compared with the rest of the year, along with the dominance of larger particles over smaller particles. To separate the effects of RH and chemical composition, enhancement factors were also calculated for a fixed RH (85%). Like $f(\text{RH}=\text{ambient})$, $f(\text{RH}=85\%)$ is lower during the haze period as compared with the summer and early fall (see Fig. 9d). Comparison between $f(\text{RH}=85\%)$ and $f(\text{RH}=\text{ambient})$ values shows the large impact of RH variation. The seasonal trends in σ_{sp} and $f(\text{RH}=85\%)$ show an anti-covariation during the haze period, with the largest values of σ_{sp} and lowest values of $f(\text{RH}=85\%)$ (Figs. 9b and 9d). The calculated annual average $f(\text{RH}=85\%)$ and $f(\text{RH}=\text{ambient})$ values for the whole year 2008 were 3.84 ± 0.37 and 4.30 ± 2.26 (mean \pm standard deviation), respectively.

In Zieger et al. 2010, the same relation of a slight decrease in $f(\text{RH}=85\%)$ with increasing particle size was observed. In that study they found no clear shift in $f(\text{RH})$ during the campaign, while the size and chemical composition clearly changed in time. This was attributed to compensating effects between size and chemical composition: smaller and less hygroscopic particles had the same magnitude in scattering enhancement as larger but more hygroscopic particles like sea salt.

5.2.3 Sensitivity of aerosol light scattering to RH, particle dry size and composition

The sensitivity of the calculated wet scattering coefficients to RH, particle dry size and composition as compared with the Wet base case (see Table 2) is demonstrated in Fig. 10. The ambient RH was varied by $\pm 5\%$ of the base values and the particle dry size by $\pm 10\%$. The sensitivity to the aerosol chemical composition was tested in two ways: the daily averaged chemical compositions were replaced by monthly averaged chemical compositions or by pure ammonium sulfate. Figure 10 shows that the RH and dry size of the particle play the most important roles in determining the scattering coefficient. Increasing the RH by 5% of the base values, increases the hourly mean values of σ_{sp} by 10 to 100%, although in most cases the deviation is below 50%. Decreasing the RH by 5% decreases the hourly mean values of σ_{sp} by 0 to 40%. Increasing the initial dry diameter (D_d) by 10%, increases the hourly mean values of σ_{sp} by 20 to 50%, and decreasing the size by 10%, decreases the hourly mean values of σ_{sp} by

1 10 to 40%. As the whole particle size distribution is shifted with the factor, changes in the dry
2 diameter are equivalent to changing number concentrations of optically active particles.
3 Replacing the daily varying chemical composition of the particles by monthly varying chemical
4 compositions changes the hourly mean values of σ_{sp} by -10 to 30% and replacing the daily
5 varying chemical composition by pure ammonium sulfate changes the hourly mean values of
6 σ_{sp} by -20 to 10%, with most of the values being between -5 and 5%. The latter result implies
7 that assuming a composition of pure ammonium sulfate in calculations of the optical properties
8 of Arctic aerosol particles results in most cases in a deviation from the true value by only 5%,
9 which is in line with the findings of Zieger et al. (2010) for their summer and fall campaign.

10 **5.3 Effect of aerosol water uptake on the direct radiative effect of aerosols**

11 **5.3.1 Vertical profiles for April 11th, 2008**

12 Example vertical profiles of the number size distribution scale factor (see section 4.1.3), RH,
13 scattering coefficient (σ_{sp}) and absorption coefficient (σ_{ap}) for April 11th 2008 are presented in
14 Fig. 11a, b, and c, respectively. RH values of about 50% up to 2 km and lower values above
15 were measured on this example day. A comparison between the absorption coefficients
16 calculated for the Dry and Wet cases shows the negligible impact of RH (< 1%) on absorption
17 properties of aerosol particles at the Zeppelin station. In contrast, a significant difference
18 between the scattering coefficients calculated for the Dry and Wet cases is predicted, especially
19 below 2 km (about 50%), where both RH and the aerosol particle concentrations are high. The
20 magnitude of the enhancement is comparable to typical values reported by e.g. Zieger et al.
21 (2010). It is worth noting that the surface-level RH on this example day is somewhat towards
22 the lower end of typical values observed in April (see Fig. 8a), so the expected difference
23 between Dry and Wet cases is larger on days with higher RHs.

24 **5.3.2 Aerosol direct radiative effect (ADRE)**

25 Comparison between the Dry and Wet monthly and annual averaged ADRE at the surface is
26 presented in Fig. 12a. ADRE is calculated from March to September, using the daily mean
27 surface albedo, aerosol size distribution data, and the vertical profiles described in Sect. 4.1.3.
28 Larger particles backscatter more light (Bohren and Huffman, 1983), which results in less
29 downward solar flux and a cooling effect at the surface. Therefore, the monthly mean ADRE
30 (Wm^{-2}) calculated for the Wet case is always more negative than the Dry case and differs from

1 month to month due to the changes in solar zenith angle, surface albedo, amount of solar
2 radiation, RH, aerosol composition and number concentration profiles. The values of the
3 monthly mean ADRE vary from -0.44 to -1.09 Wm^{-2} for the Dry case and from -0.83 to -2.60
4 Wm^{-2} for the Wet case. The dry ADRE peaks in April when the scattering coefficients are
5 highest (see Fig. 9b). The wet ADRE is the highest in July, August and September. Humidity
6 observations in the Arctic troposphere over Ny-Ålesund shows highest RH values below 1 km
7 during July, August and September as compared with the other months, while there are no
8 significant monthly differences at the higher altitudes ($> 1\text{km}$) (Treffeisen et al., 2007). The
9 hygroscopic growth of aerosol particles, reflected in the ratio between the wet and dry ADRE,
10 results in about 1.6 to 3.7 times more negative ADRE at the surface (Fig. 12b), with less impact
11 of RH during the haze period (March, April and May) and higher impact during late summer
12 and early fall (July, August and September). This is reasonable, since the haze period is
13 characterized by less hygroscopic larger (diameter $>100\text{nm}$) particles, while after the haze
14 period the size distribution shifts to primarily smaller particles (diameter $<100\text{nm}$), and the
15 overall composition is dominated by sea salt (see Fig. 9a, 8b, 8c). The annual mean ADRE for
16 the Wet case is -0.92Wm^{-2} , which is more than two times more negative than the Dry case, for
17 which the ADRE is -0.41Wm^{-2} .

18 It is interesting to note that the seasonal variation of the direct aerosol effect displays somewhat
19 different behavior from the aerosol scattering coefficients displayed in Fig. 9a, which peak
20 during the haze period. This can be explained by the fact that the ADRE is the combined result
21 of the magnitude of solar insolation, surface albedo and the scattering coefficient and vertical
22 distribution of aerosol particles. The scattering coefficient of the aerosol population is in turn
23 controlled by the particle concentration and the scattering efficiency of the individual particles.
24 The latter is controlled by particle size (governed by their dry size, ambient RH, and
25 hygroscopicity) and refractive index (governed by the chemical composition). Thus, ADRE is
26 a complex function of season, aerosol properties, RH, and surface albedo. While we believe
27 that the seasonal trends in the calculated ADRE values are representative, their exact magnitude
28 is subject to larger uncertainties due to lack of information about the exact vertical distribution
29 of the aerosol particle concentration, their size distribution and chemical composition, as well
30 as the missing coarse mode.

5.3.3 Sensitivity of ADRE to RH, particle dry size, composition and surface albedo

The sensitivity of the calculated ADRE to RH, particle size, composition and surface albedo as compared with the Wet base case (see Table 2) is presented in Fig. 13. The variations in RH, particle dry size and composition are the same as those presented in Sect. 5.2.3 and Fig. 10. The surface albedo was varied by $\pm 10\%$. A ratio higher than one means a higher negative ADRE, therefore more cooling at the surface, as compared with the Wet base case. The relative importance of RH and dry particle size are reversed for ADRE and surface layer scattering coefficients (see Fig. 10). For example, the effect of changing RH on the ADRE is at most 20% whereas the enhancement of σ_{sp} was calculated to be up to 100% (most of the cases below 50%). On the other hand, σ_{sp} changed by less than 50% when aerosol size was changed, whereas the changes in ADRE are in some cases above 80%. This can be explained by the fact that the ADRE is integrated over the whole vertical column and the largest effect of RH is near the surface (see Fig. 11c), while at higher altitudes the aerosol direct forcing is governed by the concentration and dry diameter of the particles. Figure 13 also demonstrates the importance of knowing the surface albedo for accurate predictions of ADRE, particularly during the early spring months when surface albedo is higher due to the snow covered surface. Surface albedo at Ny-Ålesund changes because of snow melting and exposing to bare ground. During the transitional months, from snow cover to rock and vice versa we have high uncertainty in ADRE.

6 Summary and conclusions

We have investigated the seasonality and impact of hygroscopic growth on aerosol optical properties and the aerosol direct radiative effect (ADRE) in the Arctic at Ny-Ålesund, Svalbard, using a comprehensive set of observational data combined with model calculations for the year 2008. An aerosol hygroscopic growth model based on the κ -Köhler theory was utilized to calculate the aerosol particle hygroscopic growth. The optical properties and ADRE were investigated by coupling the hygroscopic model to a Mie scattering model and a radiative transfer model. Measured aerosol size distributions, ambient RH together with aerosol chemical composition from filter samples were used as input to the model calculations. Comparisons between modeled and measured aerosol hygroscopicity and optical properties showed an

1 agreement that gave confidence regarding the capability of the model setup to predict seasonal
2 variations in aerosol hygroscopic growth, optical properties and ADRE.

3 The ambient aerosol scattering coefficients at the surface showed a clear seasonal variation with
4 the highest values during the haze period (March, April and May) and the lowest values during
5 summer (June and July). The hygroscopic growth of the aerosol particles was found to have a
6 significant impact on the surface level scattering coefficients, with annual averaged
7 enhancement factor $f(\text{RH})$ of 4.30 ± 2.26 at ambient RH compared to dry conditions. The
8 impact was largest during summer and fall and smallest during the haze period in spring. The
9 ambient RH was found to be the most important factor determining the ambient GF and $f(\text{RH})$
10 as compared with the aerosol particle dry size and composition. In most cases, the deviation
11 from the true value of the aerosol scattering coefficient was less than 5% when assuming a
12 composition of pure ammonium sulphate instead of using real composition measurements. The
13 seasonal behaviour of the ADRE showed a different pattern compared to the scattering
14 coefficients at the surface: the most negative values (i.e. the largest cooling effect) were found
15 during July, August and September. The sensitivity of ADRE to ambient RH vs. aerosol
16 properties was also different from the surface-level scattering coefficients with larger influence
17 of aerosol size on the predicted ADRE. This is related to the fact that the ADRE is an integrated
18 measure of the scattering over the whole vertical column as compared with the surface level
19 observations of scattering coefficients. Humidity effects on the particle scattering are in general
20 largest in the boundary layer. All in all, including the hygroscopicity of the aerosol particles
21 increased the predicted ambient ADRE with a factor of about 1.6 to 3.7 compared to the dry
22 ADRE, depending on the season.

23 Besides the strong seasonality of aerosol optical properties and ADRE at the Ny-Ålesund, our
24 results demonstrate the importance of a correct predictions of aerosol hygroscopic growth for
25 determining the direct aerosol effect on the Arctic radiative forcing and climate. Although the
26 model results in this study were obtained specifically for the Zeppelin station during 2008, the
27 developed method may be applied for other regions and time periods in future studies.

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References

- Aas, W., Solberg, S., Manø, S. og Yttri, K.E.: Overvåking av langtransportert forurensset luft og nedbør. Atmosfærisk tilførsel, 2008. Norsk institutt for luftforskning, Kjeller, OR 22/2009 (SFT (Klif) rapport nr 1051/2009), 2009.
- Achtert, P., Birmili, W., Nowak, A., Wehner, B., Wiedensohler, A., Takegawa, N., Kondo, Y., Miyazaki, Y., Hu, M., Zhu, T.: Hygroscopic growth of tropospheric particle number size distributions over the North China Plain, *J. Geophys. Res.*, 114, D00G07, doi:10.1029/2008JD010921, 2009.
- Albrecht, B. A.: Aerosols, Cloud Microphysics, and Fractional Cloudiness, *Science* 245, 1227–1230, 1989.
- Anderson, T. L., Covert, D. S., Marshall, S. F., Laucks, M. L., Charlson, R. J., Waggoner, A. P., Ogren, J. A., Caldow, R., Holm, R. L., Quant, F. R., Sem, G. J., Wiedensohler, A., Ahlquist, N. A. and Bates, T. S.: Performance Characteristics of a High-Sensitivity, Three-Wavelength, Total Scatter/Backscatter Nephelometer, *J. Atmos. Ocean. Technol.*, 13(5), 967–986, doi:10.1175/1520-0426(1996)013<0967:PCOAHS>2.0.CO;2, 1996.
- Atwater, M. a: Planetary albedo changes due to aerosols. *Science*, 170(3953), 64–6, doi:10.1126/science.170.3953.64, 1970.

- 1 Bohren, C. F., Huffman, D.R.: Absorption and scattering of light by small particles, John Wiley
2 & Sons, INC, New York, 1983.
- 3 Boucher, O. and Lohmann, U.: The sulfate-CCN-cloud albedo effect, *Tellus B*, 47(3), 281–300,
4 doi:10.1034/j.1600-0889.47.issue3.1.x, 1995.
- 5 Carrico, C. M., Rood, M. J., Ogren, J. A., Neususs, C., Wiedensohler, A., and Heintzenberg, J.:
6 Aerosol optical properties at Sagres, Portugal during ACE-2, *Tellus B*, 52, 498–525, 2000.
- 7 Chang, H., Charalampopoulos, T.T.: Determination of the wavelength dependence of refractive
8 indices of flame soot, *Proc. R. Soc. Lond A*, 430, 577–591, 1990.
- 9 Charlson, R. J. and Pilat, M. J.: Climate: The Influence of Aerosols, *J. Appl. Meteorol.*, 8(6),
10 1001–1002, doi:10.1175/1520-0450(1969)008<1001:CTIOA>2.0.CO;2, 1969.
- 11 Charlson, R. J., Schwartz, S. E., Hales, J. M., Cess, R. D., Coakley, J. A., Hansen, J. E.,
12 Hofmann, D. J.: Climate Forcing by Antropogenic Aerosols, *Science* 255, 423-430, 1992.
- 13 Coakley, J. A., Cess, R. D. and Yurevich, F. B.: The Effect of Tropospheric Aerosols on the
14 Earth's Radiation Budget: A Parameterization for Climate Models, *J. Atmos. Sci.*, 40(1), 116–
15 138, doi:10.1175/1520-0469(1983)040<0116:TEOTAO>2.0.CO;2, 1983.
- 16 Covert, D. S and Heintzenberg, J.: Size distributions and chemical properties of aerosol at Ny-
17 Ålesund, Svalbard, *Atmos. Environ. Vol. 27A*, No 17/18, 2989-2997, 1993.
- 18 Di Pierro, M., Jaeglé, L., Eloranta, E. W., and Sharma, S.: Spatial and seasonal distribution of
19 Arctic aerosols observed by the CALIOP satellite instrument (2006–2012), *Atmos. Chem.*
20 *Phys.*, 13, 7075-7095, doi:10.5194/acp-13-7075-2013, 2013.
- 21 Eldering, a.: Aerosol optical properties during INDOEX based on measured aerosol particle
22 size and composition, *J. Geophys. Res.*, 107(D22), 8001, doi:10.1029/2001JD001572, 2002.
- 23 Engelhart, G. J., Asa-Awuku, A., Nenes, A., and Pandis, S. N.: CCN activity and droplet growth
24 kinetics of fresh and aged monoterpene secondary organic aerosol, *Atmos. Chem. Phys.*, 8,
25 3937–3949, doi:10.5194/acp-8-3937-2008, 2008.
- 26 Engvall, A.-C., Ström, J., Tunved, P., Krejci, R., Schlager, H. and Minikin, A.: The radiative
27 effect of an aged, internally mixed Arctic aerosol originating from lower-latitude biomass
28 burning, *Tellus B*, 61(4), 677–684, doi:10.1111/j.1600-0889.2009.00431.x, 2009.

1 Fierz-Schmidhauser, R., Zieger, P., Gysel, M., Kammermann, L., DeCarlo, P. F.,
2 Baltensperger, U. and Weingartner, E.: Measured and predicted aerosol light scattering
3 enhancement factors at the high alpine site Jungfraujoch, *Atmos. Chem. Phys.*, 10(5), 2319–
4 2333, doi:10.5194/acp-10-2319-2010, 2010a.

5 Fierz-Schmidhauser, R., Zieger, P., Vaishya, a., Monahan, C., Bialek, J., O'Dowd, C. D.,
6 Jennings, S. G., Baltensperger, U. and Weingartner, E.: Light scattering enhancement factors
7 in the marine boundary layer (Mace Head, Ireland), *J. Geophys. Res.*, 115(D20), D20204,
8 doi:10.1029/2009JD013755, 2010b.

9 Fitzgerald, J. W., Hoppel, W. A. and Vietti, M. A.: The Size and Scattering Coefficient of Urban
10 Aerosol Particles at Washington, DC as a Function of Relative Humidity, *J. Atmos. Sci.*, 39(8),
11 1838–1852, doi:10.1175/1520-0469(1982)039<1838:TSASCO>2.0.CO;2, 1982.

12 Hegg, D. A.: Cloud condensation nucleus-sulfate mass relationship and cloud albedo. *J.*
13 *Geophys. Res.*, 99,25903-25907, 1994.

14 Hess, M., Koepke, P. and Schult, I.: Optical Properties of Aerosols and Clouds: The Software
15 Package OPAC, *Bull. Am. Meteorol. Soc.*, 79(5), 831–844, doi:10.1175/1520-
16 0477(1998)079<0831:OPOAAC>2.0.CO;2, 1998.

17 Hjellbrekke, A. G. and Fjæraa, A.M.: Data Report 2008 Acidifying and eutrophying
18 compounds and particulate matter, Norwegian Institute for Air Research, Kjeller, EMEP/CCC-
19 Report 1/2010, 2010.

20 Intergovernmental Panel on Climate Change (IPCC), *Climate Change 2013: The Scientific*
21 *Basis: Contribution of Working Group I to the Fifth Assessment Report of the*
22 *Intergovernmental Panel on Climate Change*, Cambridge Univ. Press, New York, 2013.

23 Jokinen, V., and Makela, J. M.: Closed-loop arrangement with critical orifice for DMA sheath
24 excess flow system, *J Aerosol Sci*, 28, 643-648, 1997.

25 Kerker, M.: *The scattering of light and other electromagnetic radiation*. New York, London:
26 Academic Press, 1969.

27 Kim, J., Yoon, S.-C., Jefferson, A. and Kim, S.-W.: Aerosol hygroscopic properties during
28 Asian dust, pollution, and biomass burning episodes at Gosan, Korea in April 2001, *Atmos.*
29 *Environ.*, 40(8), 1550–1560, doi:10.1016/j.atmosenv.2005.10.044, 2006.

- 1 Knutson, E. O., Whitby, K. T.: Anomalous unipolar diffusion charging of polystyrene latex
2 aerosols. *Journal of Colloid and Interface Science*, 53, 493–495, 1975.
- 3 Koehler, K. A., Kreidenweis, S. M., DeMott, P. J., Prenni, A. J., Carrico, C. M., Ervens, B., and
4 Feingold, G.: Water activity and activation diameters from hygroscopicity data – Part II:
5 Application to organic species, *Atmos. Chem. Phys.*, 6, 795–809, 2006.
- 6 Kotchenruther, R. A. and Hobbs, P. V.: Humidification factors of aerosols from biomass
7 burning in Brazil, *J. Geophys. Res. Atmos.*, 103(D24), 32081–32089, doi:10.1029/98JD00340,
8 1998.
- 9 Lawrence, D. M., Slater, A. G., Tomas, R. a., Holland, M. M. and Deser, C.: Accelerated Arctic
10 land warming and permafrost degradation during rapid sea ice loss, *Geophys. Res. Lett.*, 35(11),
11 L11506, doi:10.1029/2008GL033985, 2008.
- 12 Liu, B. Y. H., Pui, D. Y. H., Whitby, K. T., Kittelson, D. B., Kousaka, Y.: Aerosol mobility
13 chromatograph - new detector for sulphuric-acid aerosols. *Atmos. Environ.* 12, 99–104, 1978.
- 14 Liu, X., Cheng, Y., Zhang, Y., Jung, J., Sugimoto, N., Chang, S.-Y., Kim, Y. J., Fan, S. and
15 Zeng, L.: Influences of relative humidity and particle chemical composition on aerosol
16 scattering properties during the 2006 PRD campaign, *Atmos. Environ.*, 42(7), 1525–1536,
17 doi:10.1016/j.atmosenv.2007.10.077, 2008.
- 18 McCartney, E. J.: *Optics of the Atmosphere*, John Wiley, New York, 1976.
- 19 McCormick, R. A. and Ludwig, J. H.: Climate Modification by Atmospheric Aerosols, *Science*,
20 156(3780), 1358–1359, doi:10.1126/science.156.3780.1358, 1967.
- 21 McMurry, P. H. and Stolzenburg, M. R.: On the sensitivity of particle-size to relative humidity
22 for Los-Angeles aerosols. *Atmos. Environ.* 23, 497–507, 1989.
- 23 Mitchell Jr., J. M.: The effect of atmospheric aerosols on climate with special reference to
24 temperature near the Earth's surface, *J. Appl. Meteorol.*, 10, 703–714, 1971.
- 25 Myhre, G., Stordal, F., Berglen, T. F., Sundet, J. K. and Isaksen, I. S. a.: Uncertainties in the
26 Radiative Forcing Due to Sulfate Aerosols, *J. Atmos. Sci.*, 61(5), 485–498, doi:10.1175/1520-
27 0469(2004)061<0485:UITRFD>2.0.CO;2, 2004.

1 Nessler, R., Weingartner, E. and Baltensperger, U.: Adaptation of dry nephelometer
2 measurements to ambient conditions at the Jungfraujoch., *Environ. Sci. Technol.*, 39, 2219–
3 2228, 2005.

4 Nilsson, E., Swietlicki, E., Sjogren, S., Löndahl, J., Nyman, M., and Svenningsson, B.:
5 Development of an H-TDMA for long-term unattended measurement of the hygroscopic
6 properties of atmospheric aerosol particles, *Atmos. Meas. Tech.*, 2, 313–318, 2009.

7 Petters, M. D. and Kreidenweis, S. M.: A single parameter representation of hygroscopic
8 growth and cloud condensation nucleus activity, *Atmos. Chem. Phys.*, 7(8), 1961–1971,
9 doi:10.5194/acp-7-1961-2007, 2007.

10 Pilinis, C., Pandis, S. N. and Seinfeld, J. H.: Sensitivity of direct climate forcing by atmospheric
11 aerosols to aerosol size and composition, *J. Geophys. Res.*, 100(D9), 18739,
12 doi:10.1029/95JD02119, 1995.

13 Quimette, J. R., Flagan, R. C.: The extinction coefficient of multicomponent aerosols, *Atmos.*
14 *Environ.*, 16, 2405-2416, 1982.

15 Quinn, P. K., Miller, T. L., Bates, T. S., Ogren, J. A., Andrews, E., and Shaw, G. E.: A 3- year
16 record of simultaneously measured aerosol chemical and optical properties at Barrow, Alaska,
17 *J. Geophys. Res. Atmos.*, 107, 4130, doi:10.1029/2001JD001248, 2002.

18 Quinn, P. K., Shaw, G., Andrews, E., Dutton, E. G., Ruoho-Airola, T. and Gong, S. L.: Arctic
19 haze: current trends and knowledge gaps, *Tellus B*, 59(1), 99–114, doi:10.1111/j.1600-
20 0889.2006.00238.x, 2007.

21 Radke, F. S., Lyons, J. H., Hegg, D. A., Hobbs, P. V. and Bailey, I. H.: Airborne observations
22 of Arctic aerosols, I, Characteristics of Arctic haze. *Geophys. Res. Lett.* 11, 393-396, 1984.

23 Ricchiazzi, P., Yang, S., Gautier, C., and Soble, D.: SBDART: a research and teaching software
24 tool for plane-parallel radiative transfer in the Earth's atmosphere, *B. Am. Meteorol. Soc.*, 79,
25 2101–2114, 1998.

26 2101–2114, 1998Schkolnik, G., Chand, D., Hoffer, A., Andreae, M.O., Erlick, C., Swietlicki,
27 E., Rudich, Y.: Constraining the density and complex refractive index of elemental and organic
28 carbon in biomass burning aerosol using optical and chemical measurements. *Atmos. Environ.*,
29 41, 1107–1118, 2007.

1 Seinfeld, J. H. and Pandis, S. N.: Atmospheric Chemistry and Physics. Wiley, New York, 1998.

2 Sekigawa, K.: Estimation of the volume fraction of water soluble material in submicron
3 aerosols in the atmosphere, *J. Meteorol. Soc. Japan*, 61(3), 359–367, 1983.

4 Serreze, M. C. and Barry, R. G.: Processes and impacts of Arctic amplification: A research
5 synthesis, *Glob. Planet. Change*, 77(1-2), 85–96, doi:10.1016/j.gloplacha.2011.03.004, 2011.

6 Serreze, M. C., Holland, M. M. and Stroeve, J.: Perspectives on the Arctic's shrinking sea-ice
7 cover, *Science*, 315(5818), 1533–6, doi:10.1126/science.1139426, 2007.

8 Sheridan, P. J., Delene, D. J. and Ogren, J. A.: Four years of continuous surface aerosol
9 measurements from the Department of Energy's Atmospheric Radiation Measurement Program
10 Southern Great Plains Cloud and Radiation Testbed site, *J. Geophys. Res.*, 106, 20735–20747,
11 2001.

12 Shindell, D. and Faluvegi, G.: Climate response to regional radiative forcing during the
13 twentieth century. *Nat. Geo.*, 2, 294-300, 2009.

14 Silvergren, S; Wideqvist, U; Ström, J; Sjogren, S; Svenningsson, B.: Hygroscopic growth and
15 cloud forming potential of Arctic aerosol based on observed chemical and physical
16 characteristics (a 1 year study 2007-2008), *J. Geophys. Res. Atmos.*, in review process.

17 Ström, J., Umegård, J., Tørseth, K., Tunved, P., Hansson, H.-C., Holmén, K., Wismann, V.,
18 Herber, a. and König-Langlo, G.: One year of particle size distribution and aerosol chemical
19 composition measurements at the Zeppelin Station, Svalbard, March 2000–March 2001, *Phys.*
20 *Chem. Earth, Parts A/B/C*, 28(28-32), 1181–1190, doi:10.1016/j.pce.2003.08.058, 2003.

21 Svenningsson, B., Rissler, J., Swietlicki, E., Mircea, M., Bilde, M., Facchini, M. C., Decesari,
22 S., Fuzzi, S., Zhou, J., Monster, J., and Rosenorn, T.: Hygroscopic growth and critical
23 supersaturations for mixed aerosol particles of inorganic and organic compounds of
24 atmospheric relevance, *Atmos. Chem. Phys.*, 6, 1937–1952, 2006.

25 Swietlicki, E., Hansson, H.-C., Hämeri, K., Svenningsson, B., Massling, A., Mcfiggans, G.,
26 McMurry, P. H., Petäjä, T., Tunved, P., Gysel, M., Topping, D., Weingartner, E., Baltensperger,
27 U., Rissler, J., Wiedensohler, A. and Kulmala, M.: Hygroscopic properties of submicrometer
28 atmospheric aerosol particles measured with H-TDMA instruments in various environments -
29 a review, *Tellus B*, 60(2):432–469, 2008.

1 Treffeisen, R., Krejci, R., Strom, J., Engvall, A., Herber, A., Thomason, L.: Humidity
2 observations in the Arctic troposphere over Ny-Ålesund, Svalbard based on 15 years of
3 radiosonde data, *Atmos. Chem. Phys.*, 7, 2721–2732, 2007.

4 Tunved, P., Strom, J., and Krejci, R.: Arctic aerosol life cycle: linking aerosol size distributions
5 observed between 2000 and 2010 with air mass transport and precipitation at Zeppelin station,
6 Ny-Alesund, Svalbard, *Atmos. Chem. Phys.*, 13, 3643–3660, 10.5194/acp-13-3643-2013, 2013.

7 Twomey, S.: The Influence of Pollution on the Shortwave Albedo of Clouds, *J. Atmos. Sci.*,
8 34(7), 1149–1152, doi:10.1175/1520-0469(1977)034<1149:TIOPOT>2.0.CO;2, 1977.

9 Wallén, A., Lidén, G., and Hansson, H.-C.: Measured elemental carbon by thermo-optical
10 transmittance analysis in water-soluble extracts from diesel exhaust, woodsmoke, and ambient
11 particulate samples. *J. Occup. Environ. Hyg.* 7(1), 35–45, 2010.

12 Van de Hulst, H. C.: *Light Scattering By Small Particles*, John Wiley & Sons, Inc., New York,
13 1957.

14 Winker, D. M., Tackett, J. L., Getzewich, B. J., Liu, Z., Vaughan, M. A., and Rogers, R. R.:
15 The global 3-D distribution of tropospheric aerosols as characterized by CALIOP, *Atmos.*
16 *Chem. Phys.*, 13, 3345–3361, doi:10.5194/acp-13-3345-2013, 2013.

17 Wiscombe, W.: *Mie scattering calculations: Advances in technique and fast, vector-speed*
18 *computer codes*, NCAR Tech. Note TN-140+STR, Natl. Cent. for Atmos. Res., Boulder,
19 Colorado, 1979.

20 Yu, H., Kaufman, Y. J., Chin, M., Feingold, G., Remer, L. a., Anderson, T. L., Balkanski, Y.,
21 Bellouin, N., Boucher, O., Christopher, S., DeCola, P., Kahn, R., Koch, D., Loeb, N., Reddy,
22 M. S., Schulz, M., Takemura, T. and Zhou, M.: A review of measurement-based assessments
23 of the aerosol direct radiative effect and forcing, *Atmos. Chem. Phys.*, 6(3), 613–666,
24 doi:10.5194/acp-6-613-2006, 2006.

25 Zieger, P., Fierz-Schmidhauser, R., Gysel, M., Ström, J., Henne, S., Yttri, K. E., Baltensperger,
26 U., Weingartner, E.: Effects of relative humidity on aerosol light scattering in the Arctic,
27 *Atmos. Chem. Phys.*, 10, 3875–3890, 2010.

28 Zieger, P; Weingartner, E; Henzing, J; Moerman, M; de Leeuw, G; Mikkila, J; Ehn, M; Petaja,
29 T; Clemer, K; van Roozendaal, M; Yilmaz, S; Friess, U; Irie, H; Wagner, T; Shaiganfar, R;

Beirle, S; Apituley, A; Wilson, K; Baltensperger, U.: Comparison of ambient aerosol extinction coefficients obtained from in-situ, MAX-DOAS and LIDAR measurements at Cabauw, Atmos. Chem. Phys., 11, 2603–2624, 2011.

Zieger, P., Fierz-Schmidhauser, R., Weingartner, E., Baltensperger, U.: Effects of relative humidity on aerosol light scattering: results from different European sites, Atmos. Chem. Phys., 13, 10609–10631, 2013.

Zieger, P; Fierz-Schmidhauser, R; Poulain, L; Müller, T; Birmili, W; Spindler, G; Wiedensohler, A; Baltensperger, U; Weingartner, E.: Influence of water uptake on the aerosol particle light scattering coefficients of the Central European aerosol, Tellus B, 66, 1-14, 2014.

Table 1. Enhancement factors $f(\text{RH})$ reported in previous studies.

Reference	$f(\text{RH})$	RH	Site	Time period
Carrico et al. (2000)	1.46 ± 0.1	82%	Sagres, Portugal	Jun–Jul 1997
Eldering et al. (2002)	1.5 to 2	ambient	Kaashidhoo Island, Republic of Maldives	Feb 1999
Fierz-Schmidhauser et al. (2010a)	1.2 to 3.3	85%	Jungfrauoch, Switzerland	May 2008
Fierz-Schmidhauser et al. (2010b)	2.22 ± 0.17 (clean marine) 1.77 ± 0.31 (polluted air)	85%	Mace Head, Ireland	Jan–Feb 2009
Fitzgerald et al. (1982)	factor of 3.5 (size range:30-80 nm)	30-95%	Washington, DC	Jul 1979
Kim et al. (2006)	2.75 ± 0.38	85%	Gosan, Korea	Apr 2001
Kotchenruther et al. (1998)	1.01 to 1.51	80%	Brazil	

Liu et al. (2008)	2.04±0.28 (urban) 2.29±0.28 (mixed) 2.68±0.59 (marine)	80%	Guangzhou city, China	Jul 2006
Nessler et al. (2005)	1.2 to 2.7 (summer) 1.4 to 3.8 (winter)	85%	Jungfrauoch, Switzerland	
Sheridan et al. (2001)	1.0 to 3.3	85%	North Oklahoma	1999
Zieger et al. (2010)	3.24± 0.63	85%	Zeppelin station, Ny-Ålesund, Svalbard	Jul-Oct 2008
Zieger et al. (2011)	3.38±0.31 (Maritime) 1.86±0.17 (Continental) 1.95±0.14 (Maritime polluted)	85%	Cabauw	Jun-Oct 2009
Zieger et al. (2014)	2.77±0.37 (Continental)	85%	Melpitz	Feb-Mar 2009
Current study	3.84 ±0.37 4.32.26	85% Ambient RH	Zeppelin station, Ny-Ålesund, Svalbard	2008

Table 2. The Dry and Wet base cases used in the model calculations. Please note that some of the individual chemical components are available on a monthly basis only (see text for details).

Chemical composition		RH	Particle size distribution
Wet case	Daily mean	Hourly mean	Hourly mean
Dry case	Daily mean	RH= 0%	Hourly mean

Table 3. Density (ρ), hygroscopicity parameter (κ) and refractive index (at 550nm) for considered chemical components.

Component	ρ (g/cm ³)	κ	Refractive index (550 nm)
Sulfate (Ammonium Sulfate)	1.77 ^[1]	0.53 ^[4]	$1.43+1\times 10^{-8}i$ ^[6]
Sea Salt (Sodium chloride)	2.17 ^[1]	1.12 ^[4]	$1.50+1\times 10^{-8}i$ ^[6]
More Water Soluble Organics (MWS-OC)*	1.56 ^[2]	0.27 ^[4]	$1.53+6\times 10^{-3}i$ ^[6]
Less Water Soluble Organics (LWS-OC)	1.50 ^[3]	0.10 ^[4]	$1.53+8\times 10^{-3}i$ ^[6]
Elemental Carbon (EC)	1.80 ^[5]	0.00	$1.74+6\times 10^{-1}i$ ^[7]

*mean value for glutaric acid, malonic acid and levoglucosan

[1] Svenningsson et al., 2006

[2] Koehler et al., 2006; Svenningsson et al., 2006

[3] Engelhart et al., 2008

[4] Petters et al., 2007

[5] Schkolnik et al., 2007

[6] Hess et al., 1998, refractive indices as a function of wavelength from the OPAC database were used in calculations.

[7] Chang and Charalampopoulos, 1990, refractive index as a function of wave length was used in calculations

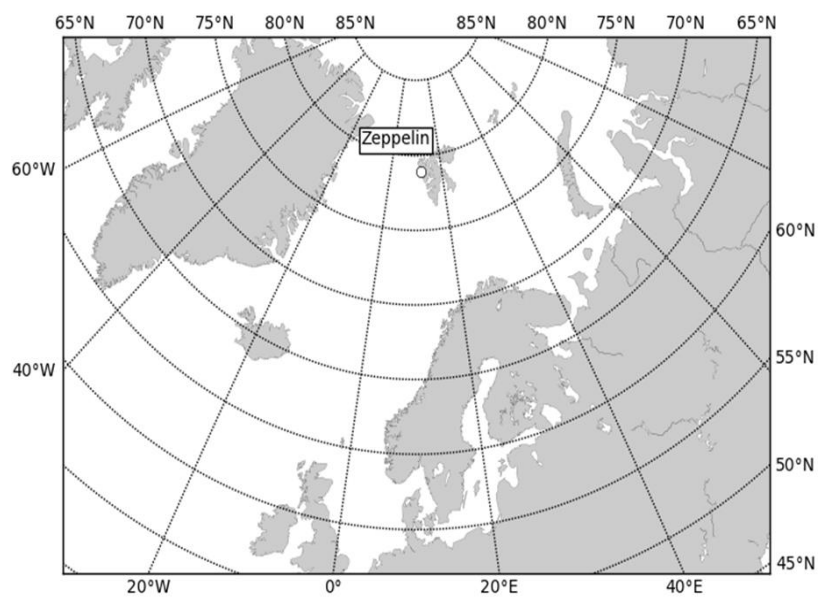
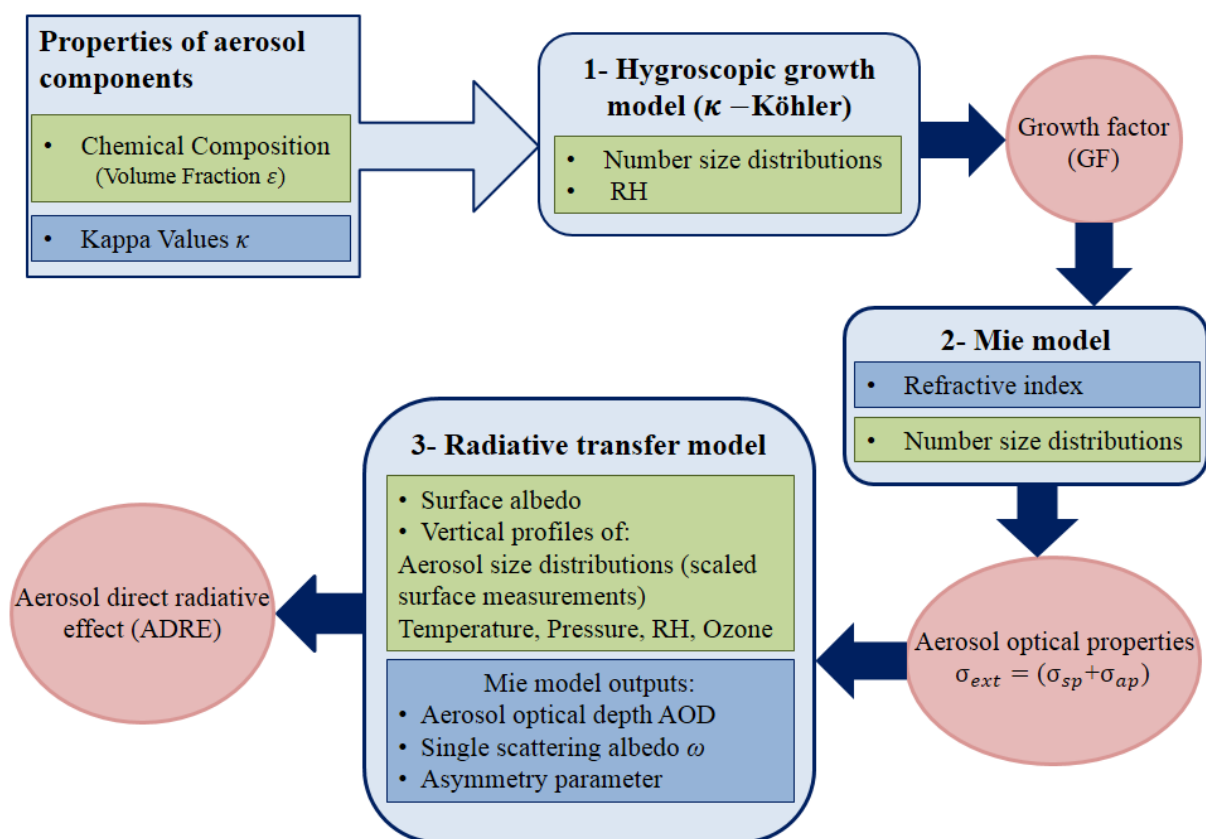


Figure 1. Mt Zeppelin station, Ny Ålesund, Svalbard at 78°54' N, 11°53' E (474 meters above sea level).

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3 Figure 2. Scheme of the models and their required input, starting with the hygroscopic growth
 4 model and ending with the radiative transfer model to calculate the aerosol direct radiative
 5 effect (ADRE). The light blue boxes refer to the different model calculations, the green boxes
 6 to experimental input data, and the dark blue boxes to additional input data (e.g. from literature),
 7 and the red circles denote model output.

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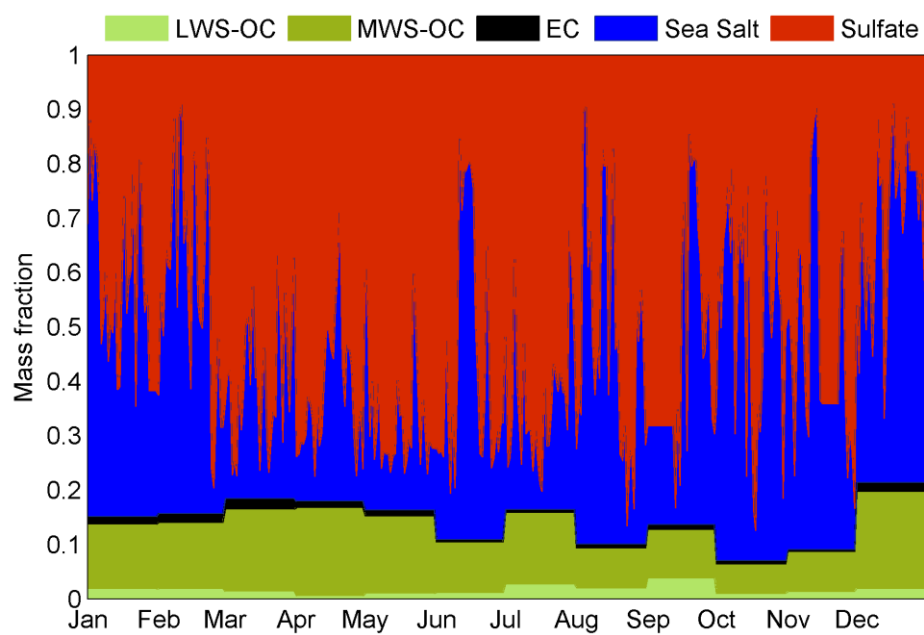
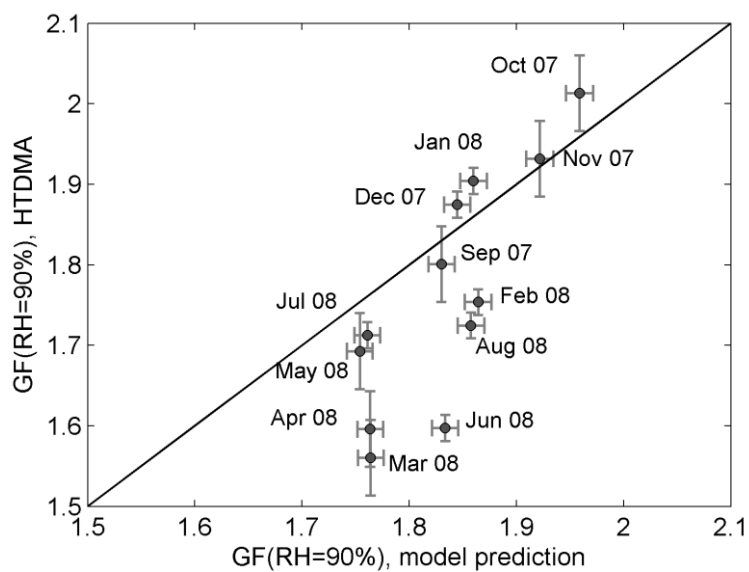


Figure 3. The averaged chemical composition for year 2008 at Zeppelin station based on filter measurements, daily basis for inorganics (Sea Salt and Sulfate) and monthly basis for Organics (Less Water Soluble Organics (LWS-OC), More Water Soluble Organics (MWS-OC), Elemental Carbon (EC)).

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3 Figure 4. Comparison of calculated monthly growth factors using the hygroscopic model and
 4 the sampled HTDMA measurements at RH = 90% for aerosol particles in the size range of 80-
 5 120 nm, from September 2007-August 2008.

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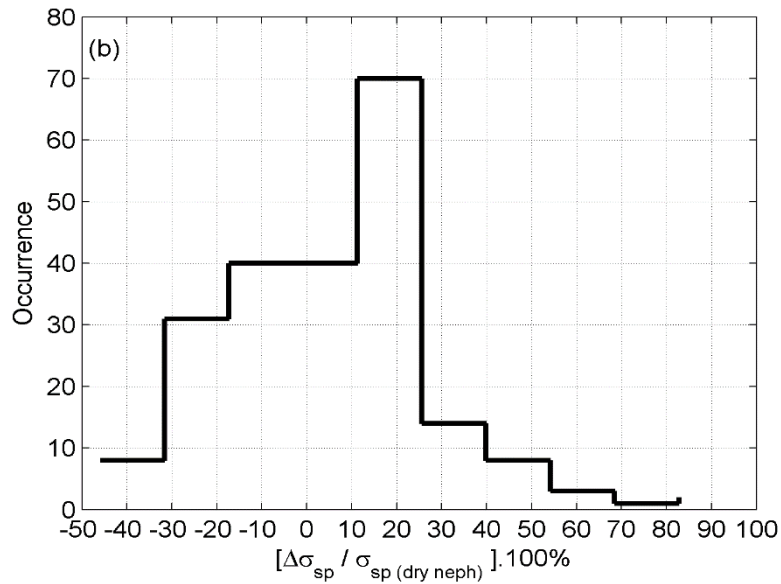
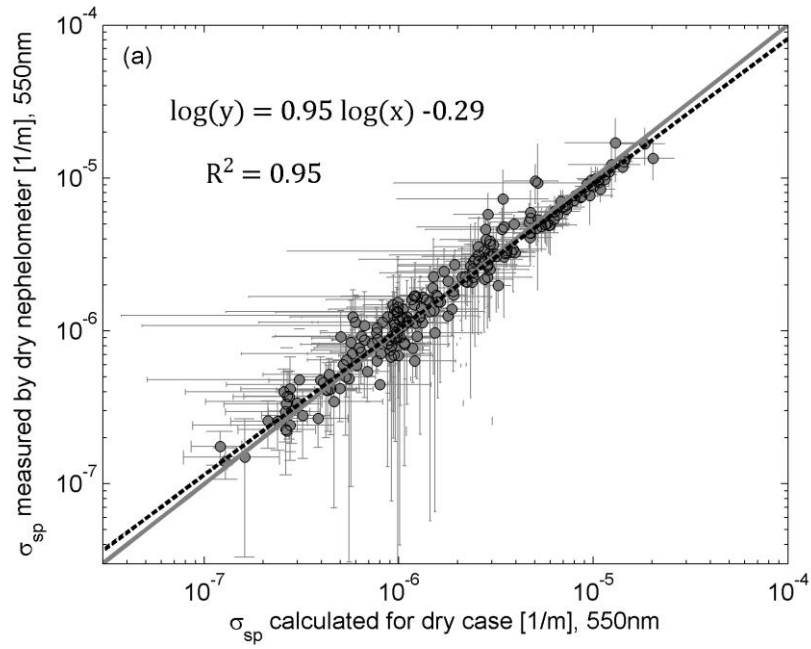
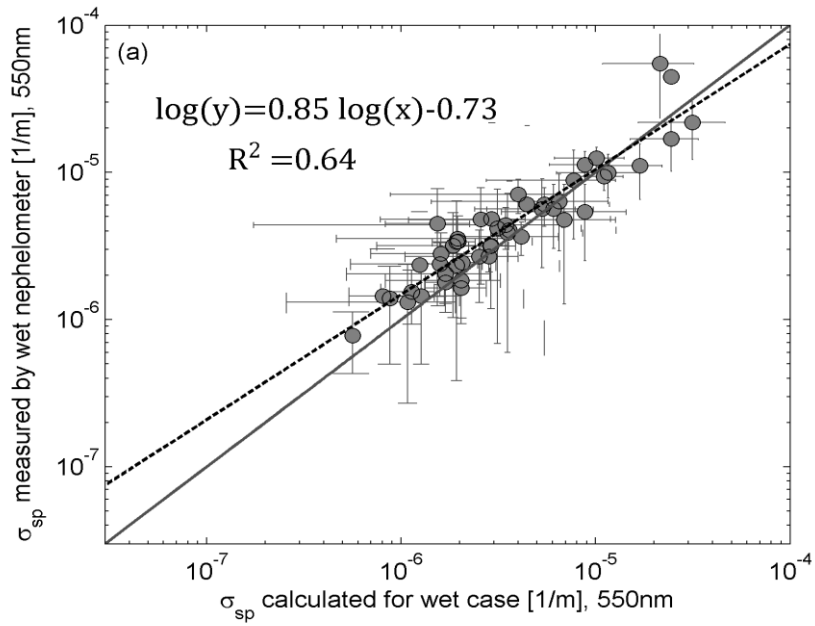
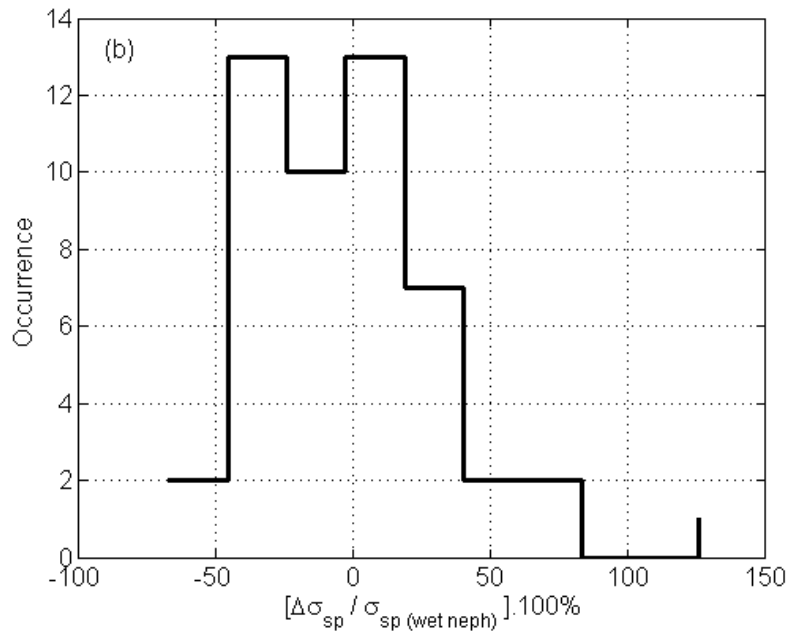


Figure 5. (a): Correlation between modeled and measured daily mean dry scattering coefficients $\sigma_{sp}(550\text{nm})$ for the year 2008. The error bars indicate the standard deviations of the daily averages. (b): Histogram of the deviation of modeled scattering coefficient from measurements in percentage (%).

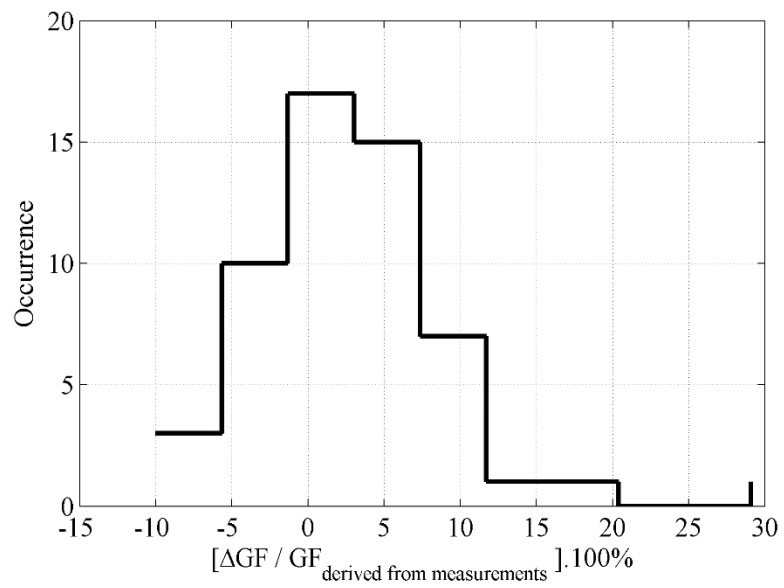


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3 Figure 6. (a): Correlation between modeled and measured daily mean wet scattering coefficients
 4 $\sigma_{sp}(550\text{nm})$ at the ambient RH for the campaign (July 15th – October 12th, 2008). The error bars
 5 indicate the standard deviations of the daily averages. (b): Histogram of the deviation of
 6 modeled scattering coefficient from measurements in percentage (%).



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2 Figure 7. Histogram of modeled growth factors deviation from values derived from the
3 humidified nephelometer measurements in percentage [%], during the campaign.

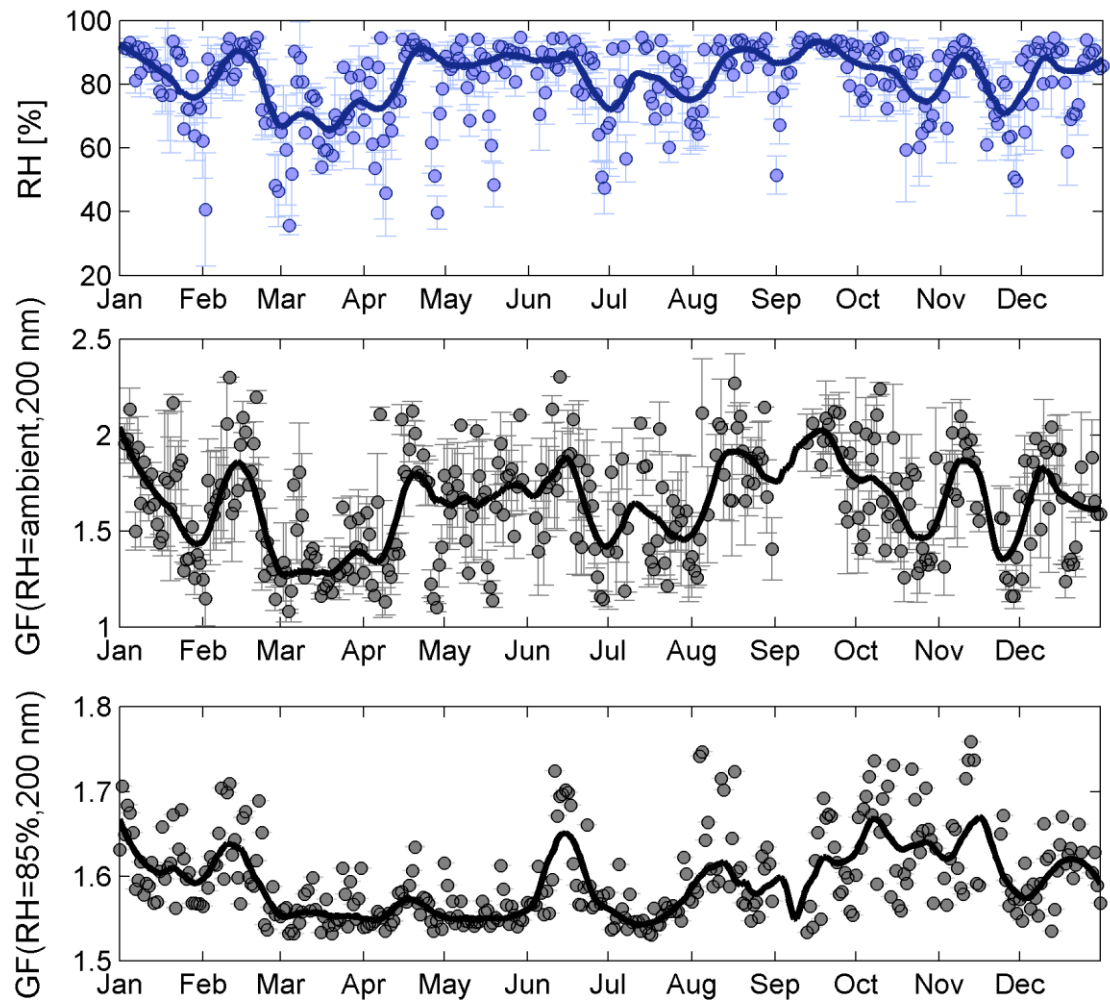


Figure 8. (a): Daily mean relative humidity (%) measured at Zeppelin station for year 2008. (b): The calculated daily mean GFs assuming ambient RH for initial size of 200 nm. (c): The calculated daily mean GFs at RH=85% for initial size of 200 nm. In figures 9a and 9b the error bars indicate standard deviations.

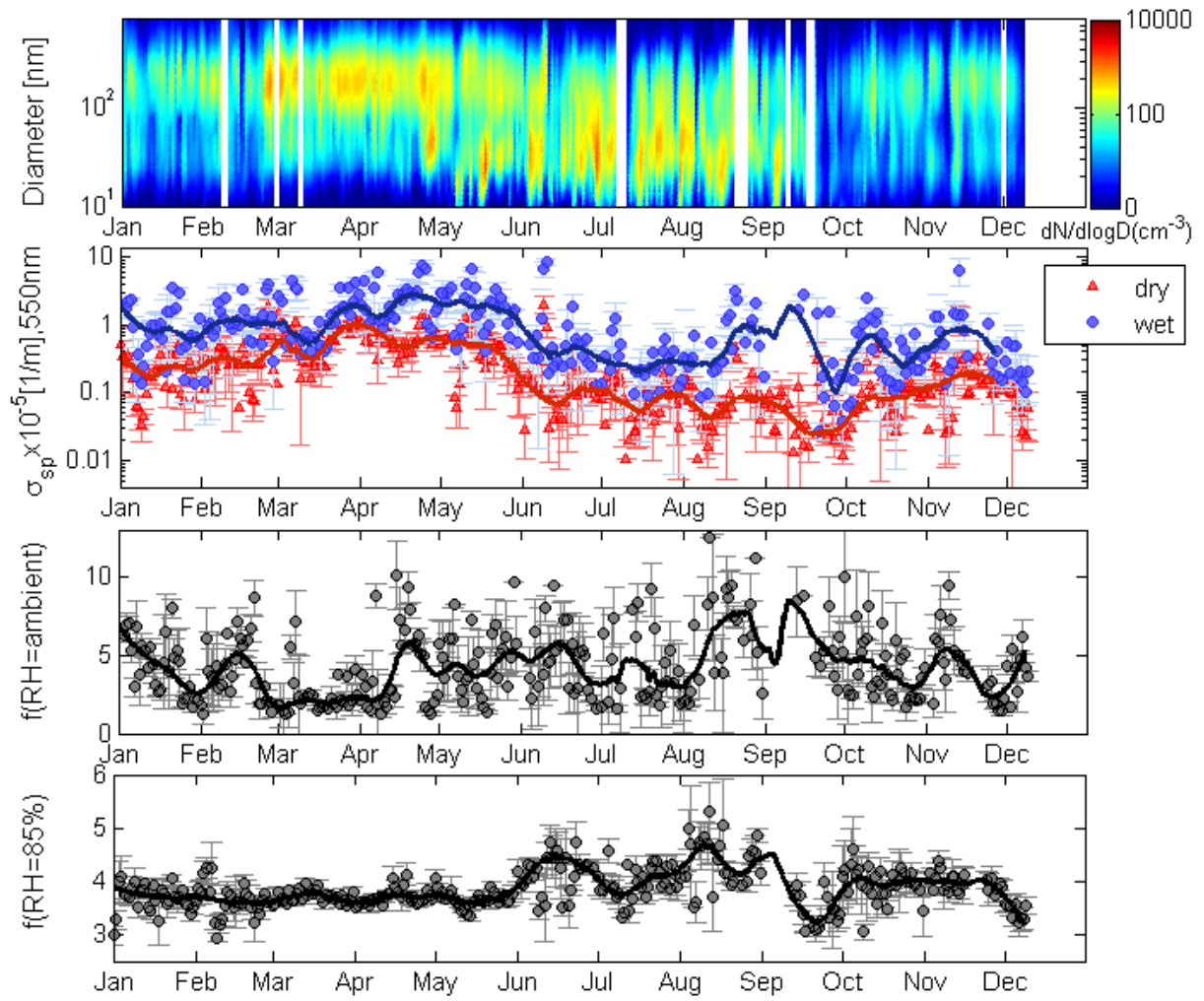


Figure 9. (a): Number size distributions measured at Zeppelin station for year 2008. (b): The calculated daily mean scattering coefficients for the Dry and Wet cases (see Table 2). (c): The calculated daily mean enhancement factors f (RH=ambient) for the Wet case. (d): The calculated daily mean enhancement factors f (RH=85%). In all figures the error bars indicate standard deviations. Note that the scale is different in 9c and 10d.

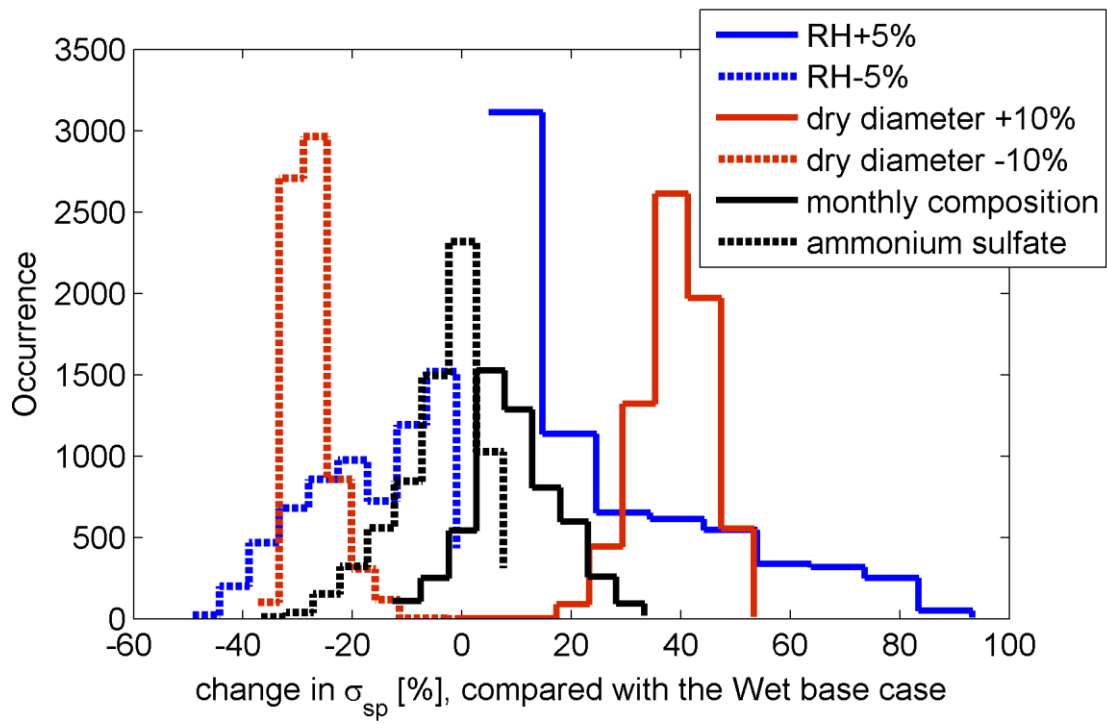
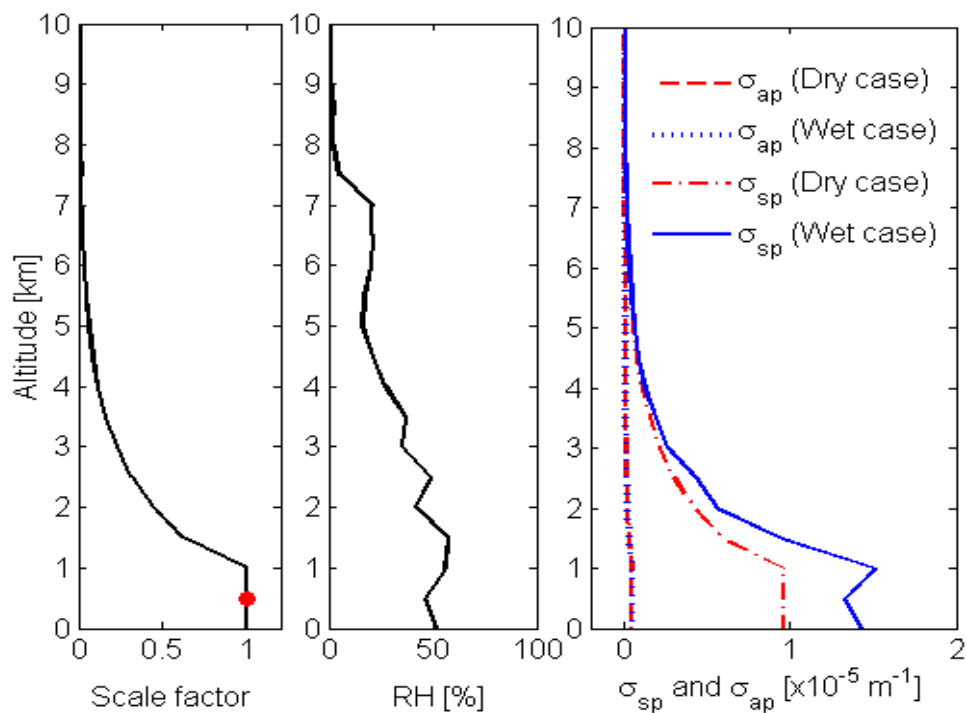


Figure 10. Sensitivity of the hourly calculated scattering coefficients σ_{sp} to RH, aerosol dry size and chemical composition as compared with the Wet base case (see Table 2).

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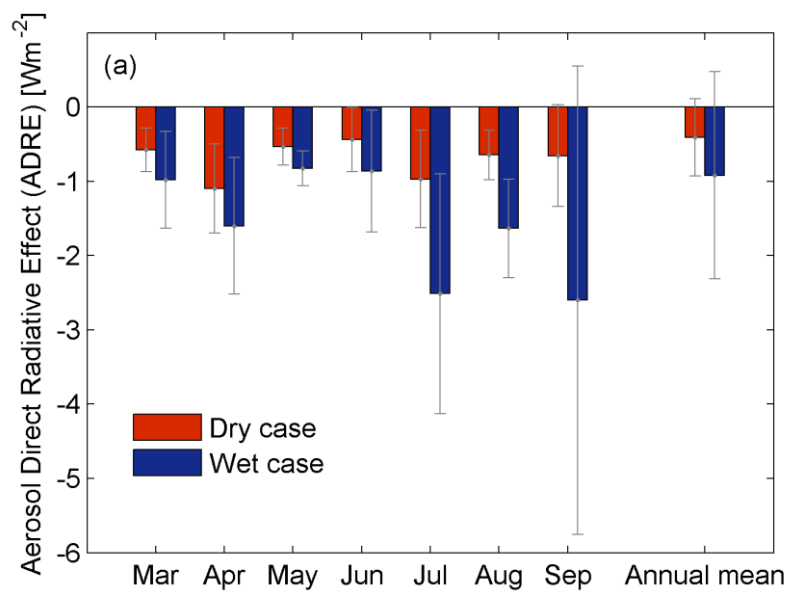


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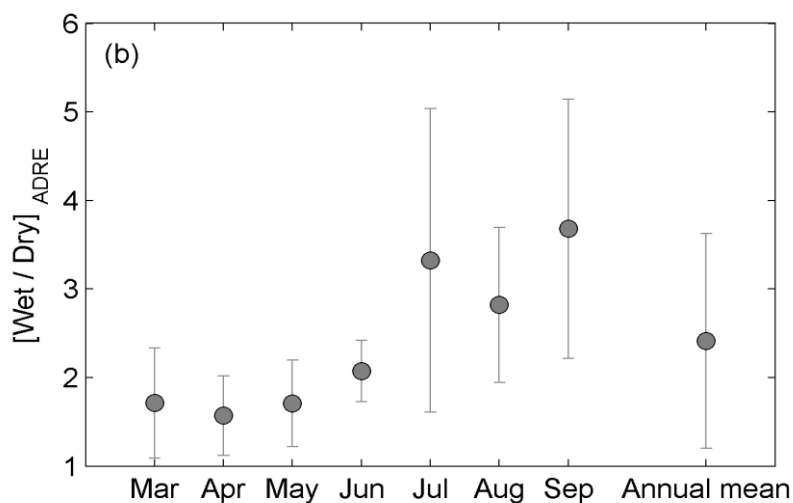
3 Figure 11. Vertical profile of (a): scale factor for particle number size distributions, the red
 4 point shows the level of the station. (b): Ambient RH measured by soundings. (c): Scattering
 5 coefficient (σ_{sp}) and absorption coefficient (σ_{ap}) m^{-1} , modeled for Dry and Wet cases for April
 6 11th, 2008 at the Zeppelin station.

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4 Figure 12. (a) Monthly and annual averaged aerosol direct radiative effects (ADRE) for the Dry
5 and Wet cases (see Table 2) at the Zeppelin station for 2008. For the months not shown the
6 ADRE is assumed to be zero due to lack of sunlight. (b) The ratio between Wet and Dry ADRE.
7 The error bars indicate the standard deviations.

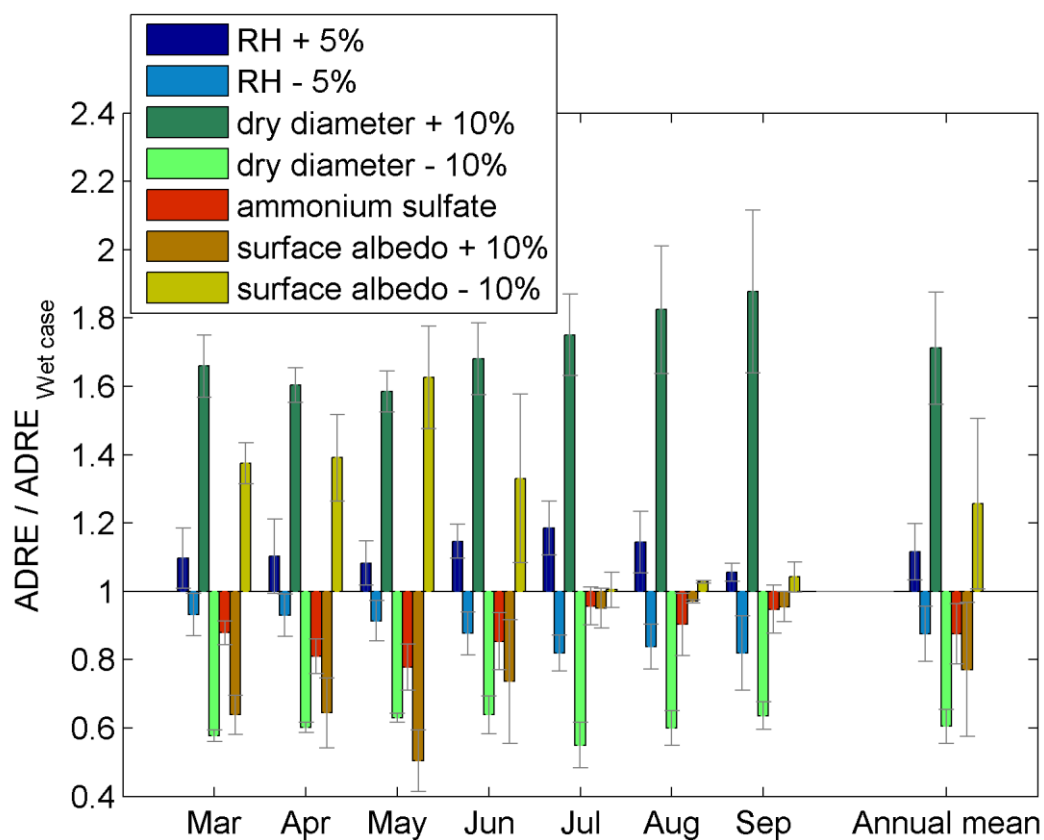


Figure 13. The sensitivity of the ratio between the calculated ADRE (new cases) and the ADRE (Wet case) to the parameters: RH, particle dry size, surface albedo and aerosol chemical composition.