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## Convective transport of very-short-lived bromocarbons to the stratosphere

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17 **Abstract.** We use the NASA GEOS Chemistry Climate Model (GEOSCCM) to quantify the  
18 contribution of two most important brominated very short-lived substances (VSLs), bromoform  
19 ( $\text{CHBr}_3$ ) and dibromomethane ( $\text{CH}_2\text{Br}_2$ ), to stratospheric bromine and its sensitivity to  
20 convection strength. Model simulations suggest that the most active transport of VSLs from the  
21 marine boundary layer through the tropopause occurs over the tropical Indian Ocean, the tropical  
22 western Pacific, and off the Pacific coast of Mexico. Together, convective lofting of  $\text{CHBr}_3$  and  
23  $\text{CH}_2\text{Br}_2$  and their degradation products supplies  $\sim 8$  ppt total bromine to the base of the Tropical  
24 Tropopause Layer (TTL,  $\sim 150$  hPa), similar to the amount of VSLs organic bromine available in  
25 the marine boundary layer ( $\sim 7.8$ - $8.4$  ppt) in the above active convective lofting regions. Of the  
26 total  $\sim 8$  ppt VSLs bromine that enters the base of TTL at  $\sim 150$  hPa, half is in the form of organic  
27 source gases and half as inorganic product gases. Only a small portion ( $<10\%$ ) of the VSLs-  
28 originated bromine is removed via wet scavenging in the TTL before reaching the lower  
29 stratosphere. On global and annual average,  $\text{CHBr}_3$  and  $\text{CH}_2\text{Br}_2$ , together, contribute  $\sim 7.7$  pptv  
30 to the present-day inorganic bromine in the stratosphere. However, varying model deep  
31 convection strength between maximum (strongest) and minimum (weakest) convection  
32 conditions can introduce a  $\sim 2.6$  pptv uncertainty in the contribution of VSLs to inorganic  
33 bromine in the stratosphere ( $\text{Br}_y^{\text{VSLs}}$ ). Contrary to the conventional wisdom, minimum  
34 convection condition leads to a larger  $\text{Br}_y^{\text{VSLs}}$  as the reduced scavenging in soluble product gases,  
35 thus a significant increase in product gas injection (2-3 ppt), greatly exceeds the relative minor  
36 decrease in source gas injection (a few 10ths ppt).

37

## 38 **1. Introduction**

39 Very short-lived (VSL) bromocarbons originate mostly from ocean biogenic sources, and  
40 when transported into the stratosphere, they exert a significant impact on the bromine budget and  
41 stratospheric ozone depletion (Kurylo and Rodriguez, 1999; Sturges et al., 2000). Recent years  
42 have seen significant progress in modeling efforts to quantify the contribution of brominated  
43 VSL substances (VSLs) to stratospheric inorganic bromine ( $\text{Br}_y^{\text{VSLs}}$ ) (e.g. Warwick et al., 2006;  
44 Liang et al., 2010; Aschmann et al., 2011; Aschmann and Sinnhuber, 2013; Hossaini et al.,  
45 2012a, 2012b, 2013). These modeling studies suggest that brominated VSLs contribute 4.8-7 ppt  
46 to reactive stratospheric bromine, within the 1-8 ppt estimate range from satellite and balloon-  
47 borne observations (WMO 2011; Sinnhuber et al., 2005; Sioris et al., 2006; Dorf et al., 2006a,  
48 2008; Salawitch et al., 2010).

49 The most important VSLs pathway to the stratosphere is via convective lofting through the  
50 tropical tropopause layer (TTL). Atmospheric chemistry and transport of brominated VSLs  
51 involves the coupling of various complex processes, e.g. highly un-uniform ocean emissions,  
52 convective transport of source gases and product gases, as well as wet scavenging of soluble  
53 product gases. An accurate modeling representation of these processes remains a challenging  
54 task. Modeling of these processes in general requires some extent of simplified assumptions  
55 and/or parameterizations, which may differ greatly from one model study to another. These  
56 differences have led to significant differences in the modeled  $\text{Br}_y^{\text{VSLs}}$  estimates. For example, a  
57 recent model study by Hossaini et al. (2013) found that modeled  $\text{Br}_y^{\text{VSLs}}$  could vary by a factor of  
58 2 when using four recently published emission inventories (Liang et al., 2010; Pyle et al., 2011;  
59 Ordóñez et al., 2012; Ziska et al., 2013). In general, model studies agree relatively well on  
60 source gas injection (SGI), suggesting about  $\sim 50\%$  of bromoform ( $\text{CHBr}_3$ ) and  $\sim 90\%$  of  
61 dibromomethane ( $\text{CH}_2\text{Br}_2$ ) can reach the stratosphere through SGI (e.g. Dvortsov et al., 1999;  
62 Nielsen and Douglass, 2001; Aschmann et al., 2011; Hossaini et al., 2010, 2012a). However,

63 modeled contribution on product gas injection (PGI) is highly uncertain depending on how wet  
64 scavenging is implemented. Many early model studies assume a uniform washout lifetime  
65 against wet scavenging (e.g. Dvortsov et al., 1999; Nielsen and Douglass, 2001; Hossaini et al.,  
66 2010; Aschmann et al., 2009), which was inadequate and lead to an underestimate in modeled  
67  $\text{Br}_y^{\text{VLS}}$  (Hossaini et al., 2012a; Aschmann et al., 2011). A recent study by Aschmann and  
68 Sinnhuber (2013) shows that treating  $\text{Br}_y^{\text{VLS}}$  as a single soluble tracer (e.g. Liang et al., 2010)  
69 can also lead to an underestimate. They found that  $\text{Br}_y^{\text{VLS}}$  increased from 3.4 ppt to 5 ppt when  
70 switching from an idealized setup with a single soluble inorganic bromine tracer to a full  
71 chemistry scheme. Thus, an accurate modeling of transport and wet scavenging of PGI seems to  
72 be the key in narrowing the uncertainty of model estimate of  $\text{Br}_y^{\text{VLS}}$ .

73 Understanding how the contribution of VLS to stratospheric bromine varies with  
74 convection strength has significant climate implications. A recent modeling analysis by Hossaini  
75 et al (2012b) suggests that VLS SGI increases from  $\sim 1.7$  ppt in 2000 to  $\sim 2.0$ - $2.7$  ppt in 2100  
76 using the Intergovernmental Panel on Climate Change (IPCC) representative concentration  
77 pathways (RCPs) scenarios, as the future simulations feature stronger tropical deep convection  
78 transport to the lower stratosphere. On the other hand, the overall response of  $\text{Br}_y^{\text{VLS}}$  to  
79 convection strength is somewhat murky. Earlier studies deploying a uniform washout lifetime  
80 found different washout rates result a significant range in the contribution of  $\text{CHBr}_3$  to  $\text{Br}_y$ , 0.5-3  
81 ppt in Sinnhuber and Folkins (2006) and 1.6-3 ppt in Aschmann et al., (2011). Liang et al.  
82 (2010) which deploys explicit wet scavenging in convective updrafts found that convective  
83 scavenging only accounts for  $\sim 0.2$  ppt difference in modeled  $\text{Br}_y^{\text{VLS}}$ . Similar results were  
84 reported in Aschmann et al. (2011), who showed while SGI is highly correlated with convective  
85 activity, the impact on total stratospheric bromine in a full chemistry scheme is nearly insensitive  
86 to dehydration, likely due to convection dilution and increased scavenging.

87 In this study, we use a 3-dimensional Chemistry Climate Model with fully interactive  $\text{CHBr}_3$   
88 and  $\text{CH}_2\text{Br}_2$ , the two most important brominated VLS, to better understand how VLS and their  
89 degradation products enter the stratosphere. We will also test the uncertainty in modeled  $\text{Br}_y^{\text{VLS}}$   
90 due to varying strength in convection and scavenging.

91

## 92 **2. Model and Simulations**

### 93 **2.1 Model Description**

94 We conduct model simulations using the NASA GEOS Chemistry Climate Model  
95 (GEOSCCM) Version 2, which couples the GEOS-5 GCM (Reinecker et al., 2008) with a  
96 stratospheric chemistry module. Model simulations have a horizontal resolution of  $2.5^\circ \times 2^\circ$   
97 (longitude by latitude) with 72 vertical layers from surface to 0.01 hPa. The GEOS-5 model uses  
98 a flux-form semi-Lagrangian dynamical core (Lin, 2004) and the Relaxed Arakawa Schubert  
99 (RAS) parameterization for convection (Moorthi and Suarez, 1992). The GEOS-5 moist  
100 processes are represented using a convective parameterization and prognostic cloud scheme.

101 The GEOSCCM V2 stratospheric chemistry module includes all important gas phase  
102 stratospheric reactions as described in Douglass and Kawa (1999), and chemical calculations are  
103 carried out above the 27<sup>th</sup> eta layer (approximately 350 hPa on global average). All chemical  
104 kinetics and photolysis rates are calculated following JPL 2010 (Sander et al., 2011). We modify  
105 the standard V2 chemistry scheme to include  $\text{CHBr}_3$  and  $\text{CH}_2\text{Br}_2$ , both are interactive with the  
106 full stratospheric chemistry scheme. The two organic source gases are released at the ocean  
107 surface following the geographically resolved emission distribution described in Liang et al.  
108 (2010) and are destroyed in the atmosphere via photolysis and reaction with the hydroxyl radical

109 (OH). OH above ~350 hPa is calculated online in the stratospheric chemistry module. OH in the  
110 lowest 26 layers (troposphere) is relaxed to a monthly mean climatological fields documented in  
111 Spivakovsky et al. (2000). Global annual mean OH from Spivakovsky et al. (2000) is  $1.16 \times 10^6$   
112 molecules  $\text{cm}^{-3}$ , yielding an atmospheric methyl chloroform ( $\text{CH}_3\text{CCl}_3$ ) lifetime of ~5.5 years. In  
113 most of the troposphere, only simple bromine chemistry is considered which partitions 80% of  
114 inorganic bromine products into HBr and the remaining 20% as HOBr, ratios adopted from Yang  
115 et al. (2005). Above the 27<sup>th</sup> eta layer (~350 hPa), bromine from  $\text{CHBr}_3$  and  $\text{CH}_2\text{Br}_2$  degradation  
116 is released as Br and interacts fully with stratospheric chemistry. Compare to the idealized case  
117 in Liang et al. (2010) that tracks the atmospheric transport of  $\text{Br}_y^{\text{VLS}}$  in a single highly soluble  
118 tracer, in this experiment, we deploy a detailed speciation of  $\text{Br}_y^{\text{VLS}}$  in both soluble forms (HBr,  
119 HOBr, and  $\text{BrONO}_2$ ) and insoluble forms (Br, BrO, BrCl) in the fully interactive stratospheric  
120 chemistry scheme in the TTL and stratosphere. Although  $\text{BrONO}_2$  is not produced in the  
121 troposphere in the current simple chemistry scheme,  $\text{BrONO}_2$  produced in the stratosphere is  
122 allowed to transport to lower altitudes via large-scale descent. A previous detailed tropospheric  
123 chemistry model study using the Harvard GEOS-Chem model and VLS emissions from Liang  
124 et al (2010) shows a small fraction of  $\text{Br}_y$  exists as BrO (0.1-0.2 ppt) below 10 km in the tropics  
125 (Parrella et al., 2012). However, the impact of the absence of this tropospheric BrO on  
126 stratospheric bromine is likely small. No heterogeneous chemistry for VLS is included in our  
127 simulations, but results from Aschmann and Sinnhuber (2013) suggest that the impact of  
128 heterogeneous chemistry is minor and the inclusion of heterogeneous activation prevents loss in  
129 scavenging and can increase  $\text{Br}_y^{\text{VLS}}$  by 10%. While heterogeneous chemistry can shift  $\text{Br}_y$   
130 partition and increase HBr up to 4 times between 12-18 km in the tropics (Aschmann et al,  
131 2011), the absence of heterogeneous chemistry in GEOSCCM is likely to have only a small  
132 impact on  $\text{Br}_y^{\text{VLS}}$  since the model HBr is present at very low abundance during daytime when the  
133 majority of convective scavenging take place (see section 3.1 and Figure 4) and the overall  
134 convective scavenging in the TTL is rather inefficient (see section 3.2).

135 The wet scavenging of the soluble inorganic bromine are the same as that detailed in Liang et  
136 al. (2010). It includes scavenging in rainout (in-cloud precipitation) and washout (below-cloud  
137 precipitation) in both large-scale precipitation (Giorgi and Chameides, 1986) and deep  
138 convective updrafts (Balkanski et al., 1993). We assume high solubility for all three inorganic  
139 bromine reservoir species, HBr, HOBr, and  $\text{BrONO}_2$ . At each time step, we follow the Giorgi  
140 and Chameides (1986) parameterization to compute the wet scavenging using the GEOS-5 model  
141 calculated large-scale and convective precipitation rates and the parameterized fraction of grid  
142 square area,  $F$ , that actually experience precipitation (Balkanski et al., 1993). Previous studies  
143 using the Giorgi and Chameides (1986) parameterization showed that, for large-scale  
144 precipitation, the global mean  $F$  is about 2.5% for all grid boxes and the median value is 10%,  
145 with values exceeding 40% in ~10% of the grid boxes (Balkanski et al., 1993; Liu et al., 2001).  
146 The global mean  $F$  for convective precipitation is much smaller, only 0.4% (Liu et al., 2001).  
147 When evaporation occurs during large-scale and convective transport, the corresponding fraction  
148 of the dissolved inorganic bromine is released back to the atmosphere. This wet scavenging  
149 scheme has been applied in many atmospheric modeling studies of soluble trace gases and  
150 aerosols, e.g.  $^{210}\text{Pb}$ ,  $^7\text{Be}$ , sulfate, sea salt, dust, and the simulated concentrations compare well  
151 with surface observations at many observation sites around the globe (e.g. Balkanski et al., 1993;  
152 Chin et al., 2000; Ginoux et al., 2001; Liu et al., 2001).

153 In this study, we present results from two 51-year simulations between 1960-2010, one with  
154 brominated VLS chemistry ( $\text{R}_{\text{VLS}}$ ) and one without ( $\text{R}_{\text{BASE}}$ ), to examine troposphere to

155 stratosphere transport (TST) of VSLS and their contribution to stratospheric bromine. It takes a  
156 considerable long time (~15 years) for the full impact of VSLS to reach the upper stratosphere,  
157 thus the first 16 years are considered as spin-up. A full model evaluation and discussion of the  
158 impact of VSLS on stratospheric ozone are presented in a separate paper, currently in  
159 preparation. For this study, we focus on model results from the last 10 years of the two  
160 simulations.

## 161 **2.2 Convection Sensitivity Simulations**

162 To quantify the uncertainty in  $Br_y^{VSLS}$  due to changes in convection strength, we conduct two  
163 sensitivity simulations from 1980 to 2010 to represent minimum convection ( $R_{MINCNV}$ ) and  
164 maximum convection ( $R_{MAXCNV}$ ) conditions by varying five convective parameters in RAS that  
165 impact the strength of deep convection, clouds, convective condensate and re-evaporation (Ott et  
166 al., 2009; 2011). These five convective parameters are RASAL1 and RASAL2 (regulates the  
167 strength and vertical profile of the relaxation time scale for deep convection), ACRTIFAC (used  
168 to compute the critical value of the cloud work function which determines the initiation of  
169 convection), BASE\_EVAP\_FAC (regulates the amount of rain evaporated into the environment  
170 below the cloud base), and AUTO\_CN (used to calculate the autoconversion of convective  
171 condensate). These five are identified as the most strongly influencing out of the total 16  
172 parameters examined using a large number of “Monte Carlo” type simulations and ensemble  
173 simulations in both a Single-Column Model as well as the GEOS-5 GCM (Ott et al., 2011). In  
174 the two sensitivity simulations, the five parameters are varied to produce the strongest  
175 (MAXCNV) and the weakest (MINCNV) representations of convection considered reasonable.  
176 In general, compared to the minimum convection condition, the maximum convection condition  
177 yields significantly large increases in shallow convection below 5 km and ~20-30% stronger  
178 vertical mass flux and more horizontal divergence between 12.5-16.5 km (Ott et al., 2011). The  
179 values used for the five convective parameters for the minimum and maximum conditions have  
180 been tested extensively in the previous ensemble analysis that they produce reasonable  
181 precipitation patterns when compared with data from the Global Precipitation Climatology  
182 Project (GPCP) that were compiled from satellite and rain gauge observations (Ott et al., 2011).  
183 The correlation coefficients between the standard run, MAXCNV and MINCNV simulations and  
184 the GPCP data sets are 0.65, 0.64, 0.62, respectively (Ott et al., 2011).

185 We aim to use these minimum and maximum conditions to bound the range of uncertainty  
186 that can be introduced due to variations in the strength of convection and wet scavenging.

187

## 188 **3. Results**

### 189 **3.1 The Contribution of VSLS to Stratospheric Bromine**

190 The simulated  $CHBr_3$  and  $CH_2Br_2$  have been evaluated extensively in Liang et al. (2010) and  
191 compares well with aircraft and surface observations in their atmospheric distribution,  
192 geographically and vertically, as well as the associated seasonality. Here we present a summary  
193 comparison of observed and modeled  $CHBr_3$  in the troposphere for 30-60°S, 30°S-30°N, 30-  
194 60°N, and 60-90°N latitude bands (Figure 1). The observed profiles are compiled using Whole  
195 Air Sampler canister measurements (Schauffler et al., 1999; Blake et al., 2003) from eight NASA  
196 aircraft missions, as detailed in Liang et al. (2010), and averaged within each latitude band at 1-  
197 km vertical interval. The comparison clearly shows that the model captures well the source gas  
198 concentrations and vertical gradients in the tropics, mid and high latitudes in both hemispheres.

199 Figure 2 shows a comparison of our simulated BrO with balloon measurements from the  
200 LPMA (Limb Profile Monitor of the Atmosphere) / DOAS (Differential Optical Absorption

201 Spectroscopy) at Teresina (5.1°S), Aire sur l'Adour (43.7°N), and Kiruna (67.9°N) collected  
202 between 2003-2005 (Dorf et al., 2006a, 2006b, 2008; Rozanov et al., 2011). The lower levels of  
203 BrO from  $R_{BASE}$  in comparison to the measurements indicate the essential role of VLSL in  
204 completing the stratospheric bromine budget. With VLSL, the GEOSCCM model simulates well  
205 the observed BrO at all sampled locations in the tropics, mid and high latitudes. Currently the  
206 GEOSCCM does not include the remaining 3 of the 5 major brominated VLSL,  $CH_2BrCl$ ,  
207  $CHBr_2Cl$  and  $CHBrCl_2$  (lifetimes between 70-150 days). Assuming a total tropospheric  
208 abundance of 1 ppt of these three VLSL and ~90% (value for  $CH_2Br_2$ ) of the source gases  
209 survive the TST to the stratosphere, this adds ~0.9 ppt to the current model estimate of  
210 stratospheric  $Br_y$  of ~24 ppt. With the current model  $BrO/Br_y$  ratio, such an increase will lead to  
211 an increase in BrO in the stratosphere up to 0.6 ppt (not shown). This results a better agreement  
212 with the DOAS BrO measurements in Teresina and Aire sur l'Adour and reasonable agreement  
213 at Kiruna, considering the large spatial variability in BrO in the high latitudes and rather  
214 localized balloon measurements. Since the model is running freely by specifying only surface  
215 source gas emissions, the fact that the model simulates well the observed concentrations for both  
216 the source gases and BrO suggests that the model presents a credible representation of  
217 stratospheric bromine chemistry.

218 Comparing results from the  $R_{VLSL}$  and  $R_{BASE}$  runs (Figure 3), we find that the inclusion of  
219  $CHBr_3$  and  $CH_2Br_2$  adds a uniform ~7.7 ppt  $Br_y$  throughout most of the stratosphere. The current  
220  $Br_y^{VLSL}$  estimate is 55% higher than our previous estimate of ~5 ppt (Liang et al., 2010) and ~6  
221 ppt estimate in Hossaini et al. (2013), though the latter two are driven by the same VLSL  
222 emissions. Compared to the idealized case in Liang et al. (2010) that tracks  $Br_y^{VLSL}$  in a single  
223 highly soluble tracer,  $Br_y^{VLSL}$  in this study is present in the fully interactive stratospheric  
224 chemistry scheme in both soluble forms (HBr, HOBr, and  $BrONO_2$ ) and insoluble forms (Br,  
225 BrO, BrCl) in the TTL. The ratio of insoluble and soluble  $Br_y$  varies with time of the day as well  
226 as altitude and location. Figure 4 shows the contrast of model inorganic bromine speciation in  
227 the tropics between daytime and nighttime. In the TTL region, at daytime, when most convective  
228 lofting occurs, the majority of  $Br_y$  exists as insoluble Br and BrO (Figure 4a), therefore greatly  
229 increases the amount of  $Br_y^{VLSL}$  that survives wet scavenging during the TST and ultimately  
230 reaches the stratosphere. The increase from ~5 ppt to ~7.7 ppt when switching from an idealized  
231 case to fully interactive stratospheric chemistry is consistent with results from Aschmann and  
232 Sinnhuber (2013), who found that  $Br_y^{VLSL}$  increased from 3.4 ppt to 5 ppt when switching from  
233 an idealized setup with a single soluble inorganic bromine tracer to a full chemistry scheme. The  
234 different estimates between Hossaini et al. (2013) and this work are likely due to how the ratio of  
235 soluble and insoluble inorganic bromine is determined inside the two models. Compared to this  
236 work which partitions inorganic bromine in its various forms based on chemical reaction rates,  
237 Hossaini et al. (2013) used a mean altitude-dependent HBr: $Br_y$  ratio. This likely leads to an  
238 excessive washout of  $Br_y^{VLSL}$  as during daytime when most of the convective lofting occurs, the  
239 majority of the inorganic bromine exists as insoluble Br and BrO. This suggests that a close  
240 approximate of soluble and insoluble inorganic product gases and the associated diurnal variation  
241 are critical to accurately quantify the contribution of VLSL to stratospheric halogen in VLSL  
242 modeling.

243 A recent study by Kreytcy et al. (2013), using balloon borne DOAS BrO measurements over  
244 Kiruna (67.9°N, 22.1°E), suggests that the ratio of  $J(BrONO_2)/k_{BrO+NO_2}$  should be a factor of 1.7  
245 larger than the JPL 2010 recommendations and is likely to shift more  $BrONO_2$  into BrO.  
246 However, this result will have only a small impact on our  $Br_y^{VLSL}$  estimates. Currently,  $BrONO_2$

247 only accounts for  $\sim 0.5$  ppt of the total  $\text{Br}_y^{\text{VLS}}$  in the TTL during daytime. In addition, convective  
248 scavenging of the inorganic bromine in the TTL is not an efficient removal process (section 3.2).  
249 Increasing the ratio of  $J(\text{BrONO}_2)/k_{\text{BrO}+\text{NO}_2}$  in the model will shift  $\text{Br}_y^{\text{VLS}}$  partition from  $\text{BrONO}_2$   
250 to  $\text{BrO}$ , which in turn will result an increase in  $\text{Br}_y^{\text{VLS}}$  due to less  $\text{BrONO}_2$  scavenging. However,  
251 this increase will be small (at most a few tenths ppt).

### 252 **3.2 Troposphere-to-Stratosphere Transport of VLS**

253 We use the simulated monthly mean  $\text{CHBr}_3$  distribution on the 355 K potential temperature  
254 surface (just below TTL) to show important tropical regions where active TST initiates (Figure  
255 5). Theoretically, trace gas distribution on the 365 K surface should be more indicative of TST  
256 as 365 K marks the zero radiative heating and air mass elevated above this level can enter the  
257 lower stratosphere through slow radiative ascent (e.g. Gettelman and Forster, 2002; Fueglistaler  
258 et al., 2009). However, it is difficult to identify active TST regions on the 365 K map (not  
259 shown) as a significant portion of  $\text{CHBr}_3$  is converted to product gases. This is not surprising for  
260 a short-lived compound with lifetime  $\sim 26$  days (WMO 2011) while on average it takes about 10  
261 days for air to transport by  $\pm 10$  K (Fueglistaler et al., 2004; Levine et al., 2007). Compared to  
262  $\text{CHBr}_3$ , it is much more difficult to identify active TST regions on a  $\text{CH}_2\text{Br}_2$  map, as  $\text{CH}_2\text{Br}_2$   
263 distribution appears more zonal with a smaller meridional gradient due to a longer lifetime,  $\sim$   
264 120 days (WMO 2011) and up to  $\sim 450$  days locally in the TTL (Hossaini et al., 2010), and thus  
265 more mixing with the surrounding background air (not shown). The 355K  $\text{CHBr}_3$  map shows  
266 three active regions that can efficiently deliver VLS to the base of TTL: 1) the tropical Indian  
267 Ocean, 2) the tropical western Pacific, and 3) off the Pacific coast of Mexico (Figure 5). There is  
268 significant seasonality associated with each entry region. The Indian Ocean appears as the most  
269 active region for the TST of  $\text{CHBr}_3$  and occurs all year long with a maximum in boreal winter  
270 (DJF). Lofting in the tropical western Pacific reaches its maximum in boreal summer (JJA),  
271 while the TST off the Mexico coast occurs mostly in boreal summer (JJA) and fall (SON). The  
272 importance of convective lofting in the Western Pacific warm pool in TST has been noted in  
273 many previous studies (e.g. Hatsushika and Yamazaki, 2003; Fueglistaler et al., 2004; Aschmann  
274 et al., 2009; Hossaini et al., 2012a; Ashfold et al., 2012) with several others suggesting that the  
275 Indian Ocean is also an important region in the TST of VLS (Levine et al., 2007, 2008; Brioude  
276 et al., 2010; Hoyle et al., 2011).

277 We show in Figure 6 the vertical profiles of organic source gases ( $\text{CH}_2\text{Br}_2 \times 2$ ,  $\text{CHBr}_3 \times 3$ ) and  
278 inorganic product gases in the three critical convective lofting regions to illustrate the transport  
279 and wet scavenging of the brominated VLS during the TST and the relative importance of SGI  
280 vs. PGI. Although tropospheric  $\text{Br}_y$  are assumed highly soluble, only a fraction of the grid boxes  
281 and a fractional area of the precipitating grid boxes actually experiencing precipitation and  
282 scavenging. On average, of all inorganic bromine ( $\sim 4$  ppt) produced from  $\text{CHBr}_3$  and  $\text{CH}_2\text{Br}_2$  in  
283 the tropical troposphere, about 50% ( $\sim 2$  ppt) is removed below 500 hPa, mainly by large-scale  
284 precipitation (Liang et al., 2010), with an additional few tenths ppt scavenged in the upper  
285 troposphere. Together,  $\sim 1.5$  ppt of  $\text{HBr}$  and  $\text{HOBr}$  survive large-scale and convective  
286 scavenging and remain in the tropical upper troposphere.  $\text{Br}$ ,  $\text{BrO}$ , and  $\text{BrONO}_2$  produced from  
287  $\text{CHBr}_3$  and  $\text{CH}_2\text{Br}_2$  degradation in the tropical upper troposphere adds an additional  $\sim 2$  ppt to  
288  $\text{Br}_y^{\text{VLS}}$ . Over the tropical Indian Ocean, on annual average, the mixing ratio of total bromine  
289 from VLS (organic + inorganic) at 150 hPa ( $\sim 355\text{K}$ ) is  $\sim 8.5$  ppt, the same as its surface  
290 abundance ( $\sim 8.5$  ppt). This implies convective lofting in this region is so efficient that the  
291 amount of VSL bromine that enters the base of TTL is set by the concentration at the marine  
292 boundary layer. As air ascend to higher altitudes, a small portion of  $\text{Br}_y^{\text{VLS}}$  ( $\sim 0.8$  ppt, 10%) is

293 gradually removed from the atmosphere via wet scavenging until reaching a constant 7.7 ppt  
294 above 10 hPa. The somewhat wavy structure in the total bromine profile in the lower  
295 stratosphere implies that not all wet scavenging happens locally and the dip in total bromine  
296 concentration between 100-50 hPa is likely associated with recirculation of air back from the  
297 mid-latitude lower stratosphere where additional scavenging occurs via in-cloud rainout. Results  
298 from the tropical western Pacific and off the coast of Mexico are similar, with a smaller surface  
299 abundance of organic bromine (7.8~8.1 ppt) and slightly less washout around 100 hPa. The  
300 global averaged total bromine from VLS (Br<sup>VLS</sup>) profile shows a maximum value of ~8 ppt at  
301 ~100 hPa. To supply this amount of Br<sup>VLS</sup> to the lower stratosphere, transport from the marine  
302 boundary layer has to initiate from tropical regions where active convective lofting is co-located  
303 with high surface concentrations, where the collective CHBr<sub>3</sub> and CH<sub>2</sub>Br<sub>2</sub> abundance exceeds 8  
304 ppt Br.

305 Mapping streamlines on top of the VLS organic bromine distribution in the tropics clearly  
306 illustrates the importance of co-location of deep convection with high surface concentration  
307 regions (Figure 7). Among the three ascending branches of the Walker circulation (Webster  
308 1983), while ascent in the tropical western Pacific penetrates deepest into the TTL, ascent in the  
309 Indian Ocean is capable in delivering more VLS bromine to the base of TTL due to higher  
310 surface concentrations. The surface abundance of VLS bromine is largely dependent on the  
311 emission distribution used. Liang et al. (2010) assumed uniform zonal emission strength across  
312 all longitudes, but the rate of bromocarbon emissions for the coastal regions per unit area are  
313 much higher than that in the open oceans. In the tropics, the prevailing surface easterly trade  
314 winds tend to bring recent emissions of CHBr<sub>3</sub> from the adjacent Indonesian coastal regions  
315 while the tropical western Pacific Ocean sees recent emissions of CHBr<sub>3</sub> from the open ocean.  
316 Therefore, the surface concentrations are higher in the Indian Ocean than the tropical western  
317 Pacific. However, it is important to point out that only a few surface observation constraints from  
318 the tropical western Pacific and none from the tropical Indian Ocean were available to derive the  
319 Liang et al., (2010) emission scenario. Hossaini et al. (2013) conducted a recent model study  
320 comparing four independent brominated VLS emission estimates. Of all four emission  
321 inventories examined (Liang et al., 2010; Pyle et al., 2011; Ordóñez et al., 2012; Ziska et al.,  
322 2013), modeled CHBr<sub>3</sub> and CH<sub>2</sub>Br<sub>2</sub> concentrations using the emission estimates from Liang et al.  
323 (2010) compare very well against tropical observations from multiple years of National Oceanic  
324 and Atmospheric Administration/Earth System Research Laboratory (NOAA/ESRL) surface  
325 flask measurements. The global mean model biases (model – observations) are –0.25 ppt for  
326 CHBr<sub>3</sub> (ranging from +0.12 ppt in the tropics to –0.65 ppt in the northern high latitudes) and –  
327 0.02 ppt for CH<sub>2</sub>Br<sub>2</sub> (ranging from +0.14 ppt in the tropics to -0.11 ppt in the northern and  
328 southern high latitudes). Comparison with the HIAPER Pole-to-Pole Observations (HIPPO)  
329 measurements between 2009–2011 supported by the National Science Foundation (NSF) yields  
330 even better agreement, with little biases in global mean CHBr<sub>3</sub> ( $\Delta_{\text{model-obs}} = +0.04$  ppt) and  
331 CH<sub>2</sub>Br<sub>2</sub> ( $\Delta_{\text{model-obs}} = -0.01$  ppt). The modeled biases for each latitude band are also small, –0.16 -  
332 +0.30 ppt for CHBr<sub>3</sub> and –0.10 - +0.12 ppt for CH<sub>2</sub>Br<sub>2</sub>. When compared against the aircraft  
333 measurements collected in the tropical western Pacific during the Stratospheric Ozone: Halogen  
334 Impacts in a Varying Atmosphere (SHIVA) campaign, Liang et al. (2010) yields the best CH<sub>2</sub>Br<sub>2</sub>  
335 (mean bias of +0.2 ppt compared with observations), compared to the other three emissions, but  
336 the simulated CHBr<sub>3</sub> is ~ +0.76 ppt too high on average. Note the high bias in modeled CHBr<sub>3</sub>  
337 in the tropical western Pacific will likely lead to a high bias in the model Br<sup>VLS</sup>, although the  
338 lofting in Indian Ocean seems to be more critical in determining the total amount of VLS



339 bromine in the lower stratosphere in this study. The model simulated high concentration of  
340 brominated VSLs over the Indian Ocean is yet to be evaluated when more surface observations  
341 become available and the importance of the Indian Ocean in delivering higher amounts of VSLs  
342 bromine into the TTL needs to be assessed. The simulated high surface concentrations of  $\text{CHBr}_3$   
343 and  $\text{CH}_2\text{Br}_2$  near tropical Central America were validated with the NASA TC<sup>4</sup> and INTEX-B  
344 measurements (Liang et al., 2010). This region, compared to the above two, is much less  
345 efficient in delivering VSLs bromine into the TTL due to the relatively weaker ascent.  
346 However, this ascending branch can be important in particular seasons and in individual years as  
347 the Walker circulation moves in the east-west direction between different phases of the El Niño  
348 Southern Oscillation (ENSO). For example, using trajectory calculations, Levine et al. (2008)  
349 found a clear shift in the TTL origin of air parcels from the tropical western Pacific and  
350 Indonesia to those from the Eastern Pacific and South America in El Niño years.

351 Of the total VSLs bromine that enters the base of TTL, about half (~4 ppt) is in the form of  
352 inorganic product gases (Figure 6) despite our simple assumption that all inorganic bromine  
353 reservoir species are highly soluble and all inorganic bromine produced below 350 hPa are  
354 partitioned into the soluble forms - HBr and HOBr. A commonly adopted approach in many  
355 previous modeling studies is to prescribe the VSLs concentration at the base of TTL with  
356 observed organic source gas mixing ratios in the upper troposphere and tracks the subsequent  
357 chemistry and transport (e.g. Sinnhuber and Folkins, 2006; Aschmann et al. 2011, Aschmann and  
358 Sinnhuber, 2013). Our result implies that  $\text{Br}_y^{\text{VSLs}}$  estimate from such approach is not complete as  
359 it misses an important component associated with PGI, which is as large as SGI at the base of  
360 TTL in our model simulation. A second implication of this result is that if we were to use  
361 measurements from the TTL region to quantify the contribution of VSLs to stratospheric  
362 bromine, it is necessary to make measurements of both organic and inorganic forms to fully  
363 account for the impact of VSLs on the atmospheric bromine budget.

### 364 **3.3 The Impact of Convection Strength**

365 We examine the difference between the two convection sensitivity simulations,  $R_{\text{MINCNV}}$  and  
366  $R_{\text{MAXRUN}}$ , to illustrate the impact of deep convection on  $\text{Br}_y^{\text{VSLs}}$  (Figure 8). Difference between  
367 these two simulations ( $R_{\text{MINCNV}} - R_{\text{MAXRUN}}$ ) suggests that in regions that mattered most (the three  
368 tropical convection centers), surprisingly, the minimum convection condition is more favorable  
369 for the TST of VSLs. While weaker convection slightly decreases SGI (a few tenths ppt), PGI  
370 increases significantly (2-3 ppt) due to less scavenging under minimum convection condition.  
371 TST of the longer-lived  $\text{CH}_2\text{Br}_2$  is less sensitive to convection strength compared to that of  
372  $\text{CHBr}_3$  because of its longer lifetime. The significant increase in  $\text{CHBr}_3$  and  $\text{CH}_2\text{Br}_2$  in the mid-  
373 latitude bands under minimum convection mainly reflects the compensating responses in large-  
374 scale descent in a general circulation model as a result of changing convection strength (Ott et  
375 al., 2011). The decrease of the descent does not impact TST of VSLs. Aschmann et al. (2011)  
376 found a qualitatively similar behavior for changes of VSLs SGI and PGI under El Niño vs. La  
377 Niña conditions - while SGI increased slightly under enhanced deep convection, the sum of SGI  
378 and PGI decreased. However, the net decrease in the  $\text{Br}_y^{\text{VSLs}}$  (<0.5 ppt) due to changes in  
379 convection strength between El Niño vs. La Niña conditions reported in Aschmann et al. (2011)  
380 is much smaller than our results above, most likely due to a much higher  $\text{Br}_y^{\text{VSLs}}$  abundance in  
381 the TTL in this work.

382 Globally, the differences in convection strength and wet scavenging introduce a ~2.6 ppt  
383 uncertainty in  $\text{Br}_y^{\text{VSLs}}$  (~6.5 ppt in  $R_{\text{MAXRUN}}$  and ~9.1 ppt in  $R_{\text{MINRUN}}$ ) (Figure 3). Although these  
384 two simulations represent the two extreme parameterization conditions within reasonable range,

385 BrO from both simulations still falls within the uncertainty range of the DOAS balloon  
386 observations of  $\pm 2.5$  ppt (Figure 2, Dorf et al., 2006b).

387

#### 388 4. Conclusions

389 We use the NASA Goddard 3-D Chemistry Climate Model, GEOSCCM, to quantify the  
390 contribution of brominated very short-lived substances (VSLs) to reactive stratospheric bromine  
391 and to test its sensitivity to the strength of deep convection and the associated scavenging.

392 The inclusion of  $\text{CHBr}_3$  and  $\text{CH}_2\text{Br}_2$  in a fully interactive stratospheric chemistry module  
393 deployed in the TTL and the stratosphere adds  $\sim 7.7$  pptv to the present-day stratospheric  
394 inorganic bromine. The most active transport of VSLs from the marine boundary layer through  
395 the tropical tropopause layer (TTL) occurs where high surface concentrations of VSLs co-locate  
396 with deep convection centers: 1) the tropical Indian Ocean, 2) the tropical western Pacific warm  
397 pool, and 3) off the Pacific coast of Mexico. On annual average, almost all VSLs-originated  
398 bromine available in the marine boundary layer ( $\sim 7.8$ - $8.4$  ppt) at these active convective lofting  
399 regions enters the TTL, half in the form of source gas injection (SGI) and half in the form of  
400 product gas injection (PGI). After lofting above 150 hPa ( $\sim 355$ K), the majority of the VSLs-  
401 originated bromine survives TST and reaches the lower stratosphere, with only a small portion  
402 ( $\sim 0.8$  ppt, 10%) removed by wet scavenging. Our model results point to a clear need for more  
403 surface measurements in the active VSLs lofting regions, in particular the tropical Indian Ocean  
404 and the tropical western Pacific, for improved emissions estimate as well as a better  
405 quantification of how much VSLs bromine is available before ultimately enters the stratosphere  
406 through the TTL.

407 Our current estimate of the contribution of  $\text{CHBr}_3$  and  $\text{CH}_2\text{Br}_2$  to stratospheric bromine  
408 ( $\text{Br}_y^{\text{VSLs}}$ ) is higher than previous modeling estimates, mainly reflecting the differences in how  
409 transport and wet scavenging of product gases are treated in different models (Hossaini et al.,  
410 2012a; Aschmann and Sinnhuber, 2013). Interestingly, even though the models are driven with  
411 same emission estimates (Liang et al., 2010; Hossaini et al., 2013) and/or with similar source gas  
412 abundance at the base of TTL (Aschmann et al. 2011; Aschmann and Sinnhuber, 2013), this  
413 study still yields higher  $\text{Br}_y^{\text{VSLs}}$  for two main reasons. Modeling VSLs and their degradation  
414 products in a full stratospheric chemistry scheme in the TTL and stratosphere leads to a better  
415 representation of the partition between soluble and insoluble product gases and the associated  
416 diurnal variation, both are critical in an accurate  $\text{Br}_y^{\text{VSLs}}$  estimate. Secondly, tracking PGI from  
417 the free troposphere to the base of TTL, which is as large as SGI at the base of TTL in our model  
418 simulation, is also important. This was not included in many earlier model studies (e.g.  
419 Aschmann et al. 2011; Aschmann and Sinnhuber, 2013), thus implying a missing component in  
420 these earlier model VSLs contribution estimates.

421 Differences in model deep convection strength can introduce  $\sim 30\%$  (6.6 ppt - 9.2 ppt)  
422 uncertainty in simulated  $\text{Br}_y^{\text{VSLs}}$  within the same convection scheme. It is important to point out  
423 that this uncertainty is likely larger among different models or a same model but different  
424 versions when different convective parameterizations are used.

425 Contrary to conventional wisdom that the VSLs impact on stratospheric bromine is larger  
426 under more intense deep convection, our simulations suggest that minimum convection condition  
427 is favorable for TST of VSLs due to reduced scavenging of soluble product gases under weaker  
428 convection. The impact of convection strength on PGI greatly outweighs the impact on SGI with  
429 the change in PGI  $\sim$  ten times larger than that of SGI.

430

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438

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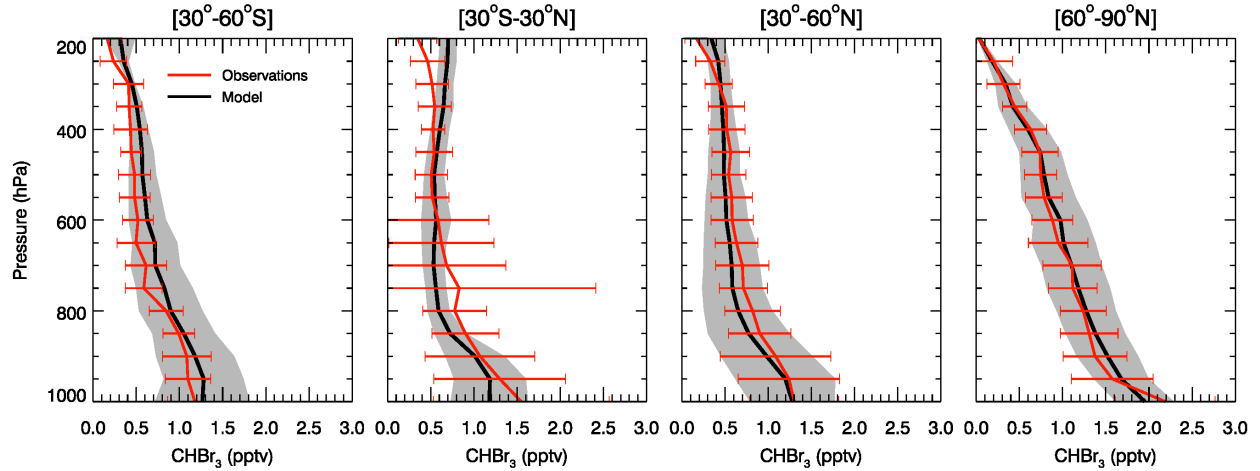
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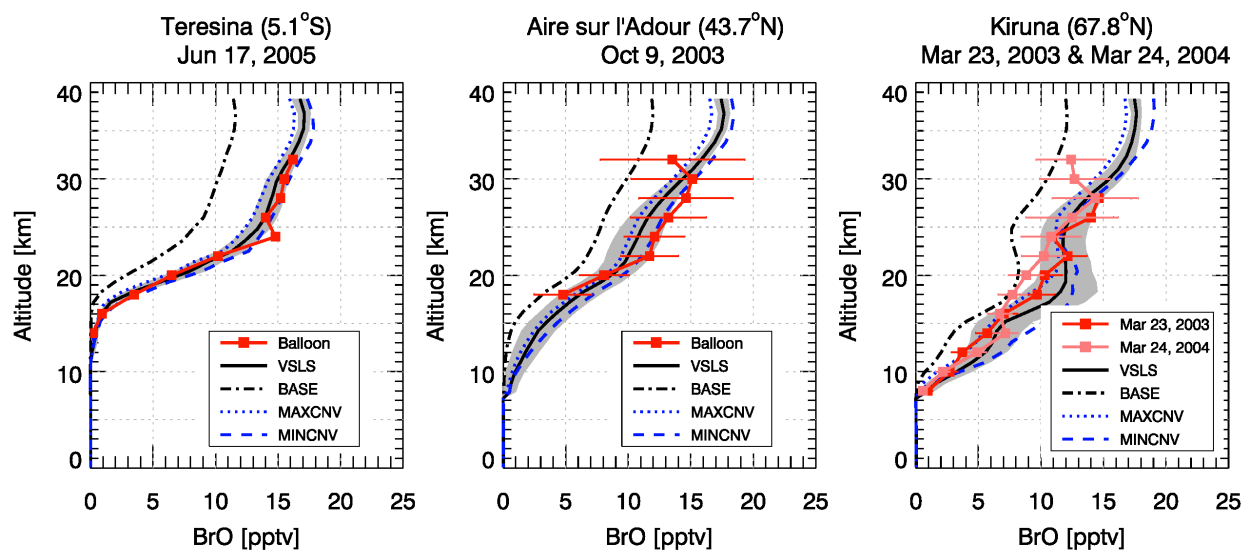
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**Figure 1.** Comparison between the observed (red lines with horizontal bars indicating one standard deviation) and simulated (black lines with gray shading showing one sigma variability) vertical profiles of CHBr<sub>3</sub> in the troposphere. Observations are compiled using Whole Air Sampler canister measurements from eight NASA aircraft missions are averaged for 30-60°S, 30°S-30°N, 30-60°N, and 60-90°N latitude bands at 1-km vertical interval. Model is sampled at the same location as the observations in the corresponding month in 2010.



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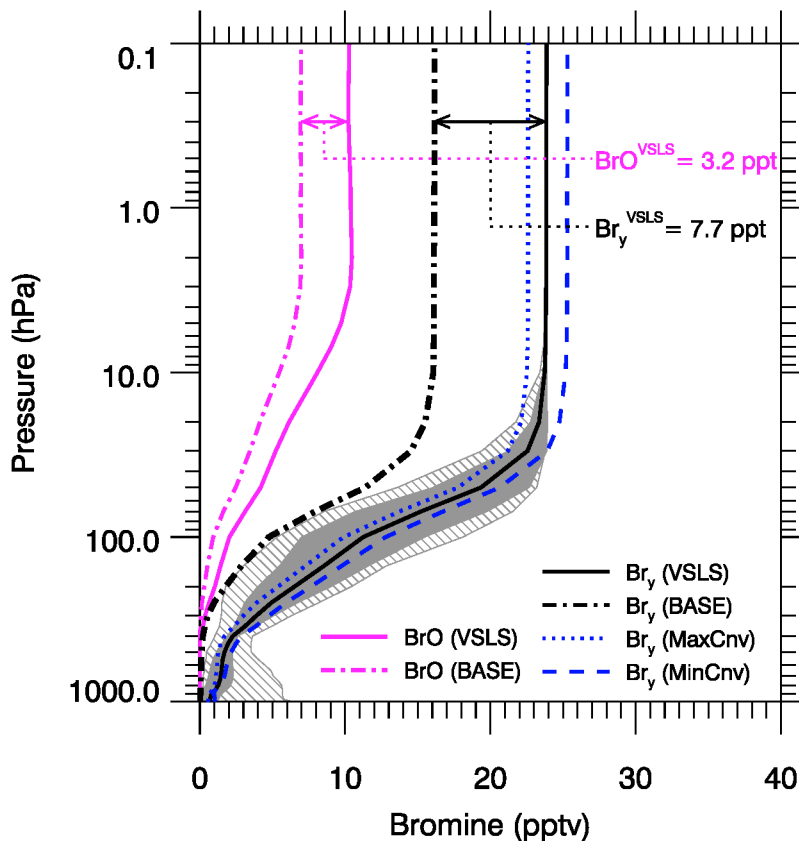
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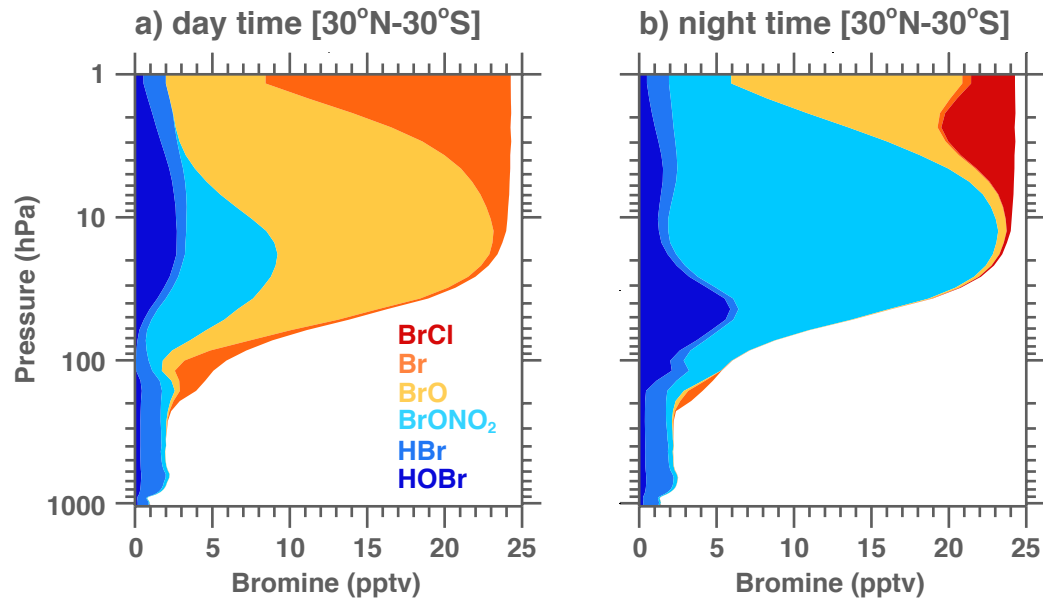
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**Figure 2.** Comparison of simulated BrO from the GEOSCCM run  $R_{VSLs}$  (black solid line with 1-sigma variance in gray shading) with balloon measurements (red line with 1-sigma uncertainty in horizontal bars) from the LPMA/DOAS Spectrometers in the stratosphere. The black dash-dotted line shows simulated BrO from the run without VSLs ( $R_{BASE}$ ). Simulated BrO from the two convective sensitivity simulations are also shown (dotted line for maximum convection condition and dashed line for minimum convection condition). The model is sampled at the same month and latitude as the balloon measurements but only daytime profiles from all available longitude grid points are collected to calculate the 1-sigma variance of simulated BrO.



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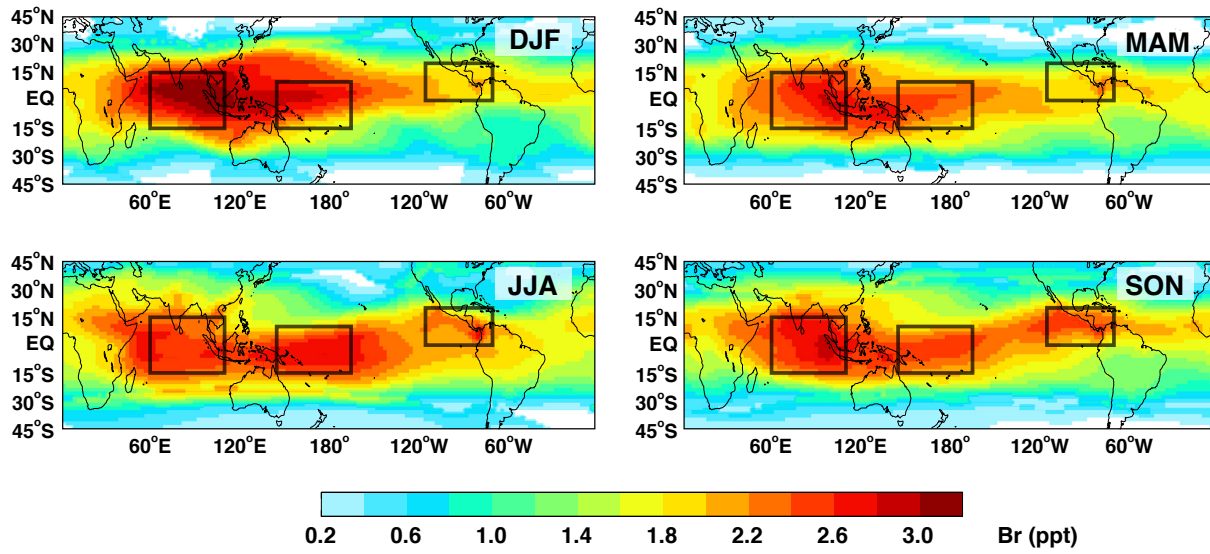
**Figure 3.** Model simulated global and annual mean BrO and Br<sub>y</sub> profiles from the simulation with VLSL ( $R_{VLSL}$ , solid lines) and the base simulation without VLSL ( $R_{BASE}$ , dash-dotted lines) for Year 2010. The gray shadings indicate the spread of annual mean Br<sub>y</sub> in the  $R_{VLSL}$  run with the hatched areas indicate the minimum to maximum range and the filled shadings indicate 1-sigma variance. Br<sub>y</sub> from the two convective sensitivity simulations are also shown (blue dashed and dotted line for minimum and maximum convection conditions, respectively).



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**Figure 4.** Annual mean model inorganic bromine (BrCl, Br, BrO, BrONO<sub>2</sub>, HBr, HOBr) tropical vertical profiles for daytime (left panel) and nighttime (right panel), averaged between 30°N-30°S.

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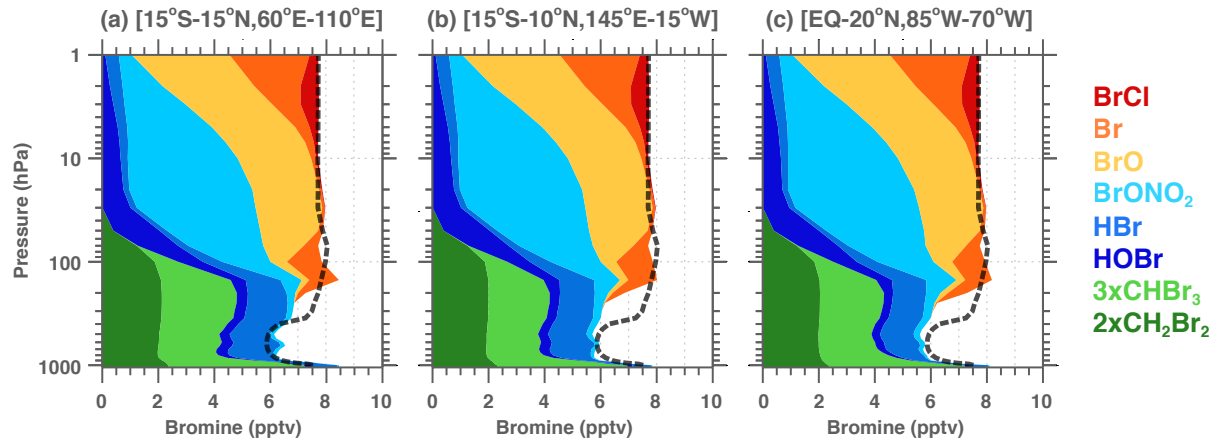
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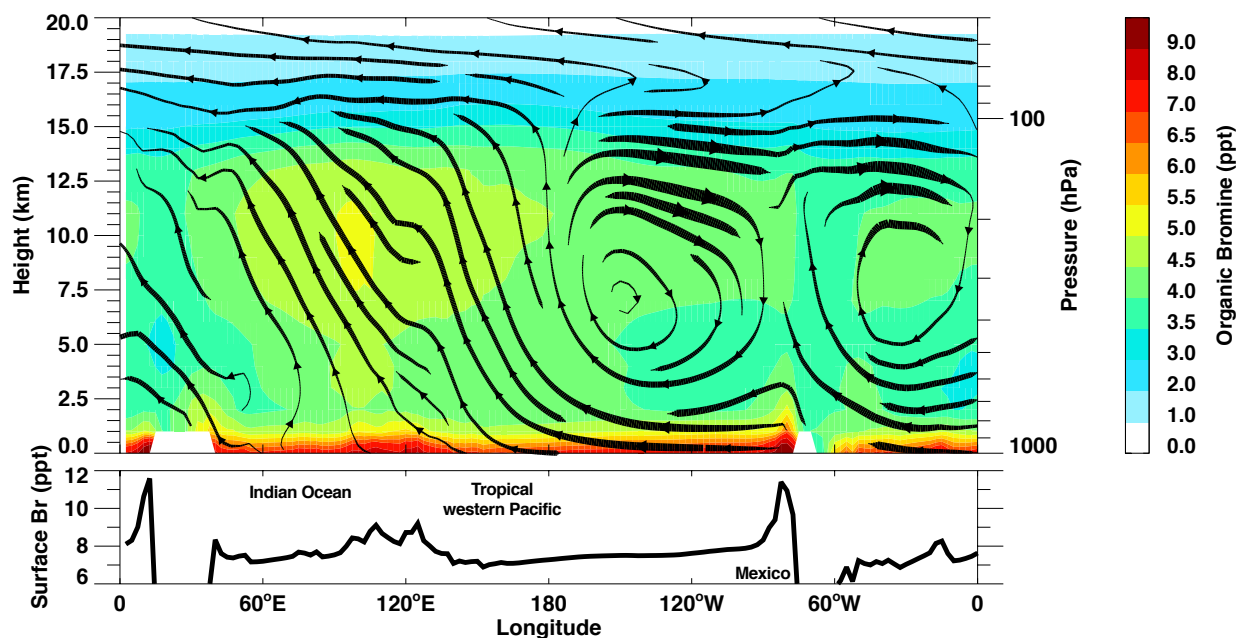
**Figure 5.** 10-year averaged (2001-2010) seasonal mean distribution of simulated CHBr<sub>3</sub> (unit ppt Br) at the 355K potential temperature layer (just below tropopause) in DJF, MAM, JJA and SON. The black boxes outline the three regions where most active troposphere-to-stratosphere transport occurs.



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**Figure 6.** The contribution of organic bromine ( $\text{CH}_2\text{Br}_2 \times 2$ ,  $\text{CHBr}_3 \times 3$ ) and inorganic bromine ( $\text{Br}_y^{\text{VSLs}}$ ) from  $\text{CHBr}_3$  and  $\text{CH}_2\text{Br}_2$  degradation to atmospheric bromine in three active TST regions (a) the tropical Indian Ocean (left), (b) the tropical western Pacific warm pool (middle), and (c) the Pacific Coast of Mexico (right). The soluble product gases (HBr, HOBr,  $\text{BrONO}_2$ ) and insoluble product gases (Br, BrO, BrCl) are shown in blue colors and warm colors, respectively. The global mean sum of organic and inorganic bromine from VSLs is also shown (thick black dashed line). The model results are 10-year annual averages from 2001-2010.

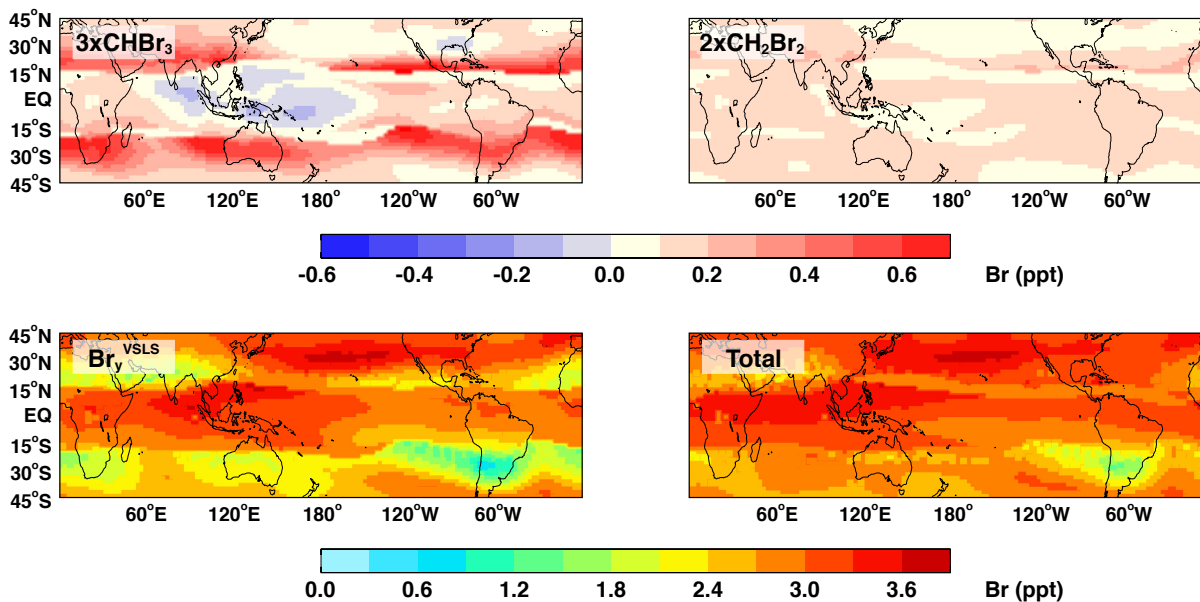
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**Figure 7.** Longitude-height cross-section of modeled total VSL organic bromine ( $\text{CHBr}_3 \times 3 + \text{CH}_2\text{Br}_2 \times 2$ ) (color contours) and streamlines (black lines) in the deep tropics. The organic bromine and wind streams are 10-year annual averages (2001-2010) between  $10^\circ\text{S}$ - $10^\circ\text{N}$ . Surface organic bromine abundance is shown as the black solid line at the bottom panel.

MinCnv - MaxCnv: 355K - 380K



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**Figure 8.** The simulated annual mean difference between the  $R_{\text{MINCNV}}$  and  $R_{\text{MAXRUN}}$  runs in organic bromine ( $3x\text{CHBr}_3$ ,  $2x\text{CH}_2\text{Br}_2$ ), inorganic bromine and total bromine from VSLs between 355-380K. The model results are 10-year annual averages from 2001-2010.