



The AeroCom OA  
model evaluation

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# The AeroCom evaluation and intercomparison of organic aerosol in global models

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## Abstract

This paper evaluates the current status of global modeling of the organic aerosol (OA) in the troposphere and analyzes the differences between models as well as between models and observations. Thirty-one global chemistry/transport and general circulation models have participated in this intercomparison, in the framework of AeroCom phase II. The simulation of OA varies greatly between models in terms of the magnitude of primary emissions, secondary OA (SOA) formation, the number of OA species used (2 to 62), the complexity of OA parameterizations (gas-particle partitioning, chemical aging, multiphase chemistry, aerosol microphysics), and the OA physical, chemical and optical properties. The diversity of the global OA simulation results has increased since earlier AeroCom experiments, mainly due to the increasing complexity of the SOA parameterization in models, and the implementation of new, highly uncertain, OA sources. Diversity of over an order of magnitude exists in the modeled vertical distribution of OA that deserves a dedicated future study. Furthermore, although the OA/OC ratio depends on OA sources and atmospheric processing and is important for model evaluation against OA and OC observations, it is resolved only by few global models.

The median global primary OA (POA) source strength is  $56 \text{ Tga}^{-1}$  (range  $34\text{--}144 \text{ Tga}^{-1}$ ) and the median secondary OA source strength (natural and anthropogenic) is  $19 \text{ Tga}^{-1}$  (range  $13\text{--}121 \text{ Tga}^{-1}$ ). Among the models that take into account the semi-volatile SOA nature, the median source is calculated to be  $51 \text{ Tga}^{-1}$  (range  $16\text{--}121 \text{ Tga}^{-1}$ ), much larger than the median value of the models that calculate SOA in a more simplistic way ( $19 \text{ Tga}^{-1}$ ; range  $13\text{--}20 \text{ Tga}^{-1}$ , with one model at  $37 \text{ Tga}^{-1}$ ). The median atmospheric burden of OA is  $1.4 \text{ Tg}$  (24 models in the range of  $0.6\text{--}2.0 \text{ Tg}$  and 4 between  $2.4\text{--}3.8 \text{ Tg}$ ) with a median OA lifetime of 5.4 days (range  $3.8\text{--}9.6$  days). In models that reported both OA and sulfate burdens, the median value of the OA/sulfate burden ratio of is calculated to be 0.77; 13 models calculate a ratio lower than 1, and 9 models higher than 1. For 26 models that reported OA deposition fluxes, the median

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wet removal is  $70 \text{ Tga}^{-1}$  (range  $28\text{--}209 \text{ Tga}^{-1}$ ), which is on average 85 % of the total OA deposition.

Fine aerosol organic carbon (OC) and OA observations from continuous monitoring networks and individual field campaigns have been used for model evaluation. At urban locations the model-observation comparison indicates missing knowledge on anthropogenic OA sources, both strength and seasonality. The combined model/measurements analysis suggests the existence of increased OA levels during summer due to biogenic SOA formation over large areas of the USA that can be of the same order of magnitude as the POA, even at urban locations, and contribute to the measured urban seasonal pattern.

Global models are able to simulate the high secondary character of OA observed in the atmosphere as a result of SOA formation and of POA aging, although, the amount of OA present in the atmosphere remains largely underestimated, with a mean normalized bias (MNB) equal to  $-0.62$  ( $-0.51$ ) based on the comparison against OC (OA) urban data of all models at surface,  $-0.15$  ( $+0.51$ ) when compared with remote measurements, and  $-0.30$  for marine locations with OC data. The correlations overall are low when comparing with OC (OA) measurements:  $0.47$  ( $0.52$ ) for urban stations,  $0.39$  ( $0.37$ ) for remote, and  $0.25$  for marine stations with OC data. The combination of high (negative) MNB and higher correlation at urban stations when compared with the low MNB and lower correlation at remote sites suggests that the knowledge about the processes, on top of the sources, are important at the remote stations. There is no clear change in model skill with increasing model complexity with regard to OC or OA mass concentration. However, the complexity is needed in models in order to separate between anthropogenic and natural OA and accurately calculate the impact of OA on climate.

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## 1 Introduction

Atmospheric aerosols are important drivers of air quality and climate. The organic component of aerosols can contribute 30–70 % of the total submicron dry aerosol mass, depending on location and atmospheric conditions (Kanakidou et al., 2005; Murphy et al., 2006). The majority of fine aerosol mass ( $PM_{10}$ : particulate matter of dry diameter smaller than  $1 \mu\text{m}$ ) consists of non-refractory material and has been found to contain large amounts of organic matter (Zhang et al., 2007; Jimenez et al., 2009), as measured by the Aerosol Mass Spectrometer (AMS).

Global model estimates of the dry organic aerosol (OA) direct radiative forcing at the top of the atmosphere vary between  $-0.06$  and  $-0.24 \text{ W m}^{-2}$  based on AeroCom phase I experiments (Schulz et al., 2006), which was decomposed during AeroCom phase II to  $-0.03 \pm 0.01 \text{ W m}^{-2}$  for primary organic aerosol (POA) from fossil fuel and biofuel,  $-0.02 \pm 0.09 \text{ W m}^{-2}$  for secondary organic aerosol (SOA) and  $0.00 \pm 0.05 \text{ W m}^{-2}$  for the combined OA and black carbon from biomass burning (Myhre et al., 2013). IPCC (2013) assessed the contribution of anthropogenic primary and secondary organic aerosols to the radiative forcing from aerosol-radiation interactions (RFari) to be  $-0.12$  ( $-0.4$  to  $+0.1$ )  $\text{W m}^{-2}$ . Spracklen et al. (2011) estimated the climate forcing of the anthropogenically driven natural SOA alone (including the presence of water on hydrophilic OA) at  $-0.26 \pm 0.15 \text{ W m}^{-2}$  (direct effect) and  $-0.6^{+0.24}_{-0.14} \text{ W m}^{-2}$  (indirect effect). These amounts largely depend on the atmospheric loadings of OA simulated by the models under past, present and future climate conditions, and on the properties they attribute to them. Indeed, Myhre et al. (2013) calculated a SOA load of  $0.33 \pm 0.32 \text{ Tg}$ , while Spracklen et al. (2011) estimated a SOA load of  $1.84 \text{ Tg}$ , which resulted in an order of magnitude higher radiative forcing. There is therefore urgent need for a consensus between models and agreement with observations, in order to constrain the large variability between models and, consequently, the OA impact on climate.

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## 1.1 Definitions

OA can be emitted directly as POA or formed via gas-phase reactions and subsequent condensation of semi-volatile vapors, resulting in SOA. Emissions of volatile organic compounds (VOCs) from terrestrial vegetation are 10 times larger than from anthropogenic sources (Guenther et al., 1995; Kanakidou et al., 2005 and references therein). In addition, the mass of organic carbon emitted in the gas phase exceeds by more than a factor of 10 that emitted directly as primary particulate matter (Goldstein and Galbally, 2007; Kanakidou et al., 2012). VOCs therefore have a large potential to contribute to SOA formation. However, the exact formation processes and composition of OA are poorly understood. Fuzzi et al. (2006) and Hallquist et al. (2009) provided a number of marker compounds and observations that could be used to distinguish the various OA sources. Most OA observational techniques measure the particulate organic carbon content of OA mass, either total (OC) or the water soluble component (WSOC), while some of the variability of OA is accounted for by oxygen, nitrogen and other elements in the organic compounds. Significant discrepancies in OC concentrations determined by different techniques have been identified (Kanakidou et al., 2005), and have been addressed by protocols of the definition of OC/EC (Elemental Carbon) measurements (Cavalli et al., 2010). The use of OC historically corresponded to its easier measurement. Recently, Aerosol Mass Spectrometer (AMS) observations started providing very high temporal resolution information on the OA mass of the non-refractory PM<sub>1</sub> (particles of diameter smaller than 1 μm). It has to be emphasized that it is the OA mass, not the OC, which determines aerosol properties such as chemical composition, size, hygroscopicity and hygroscopic growth, each of which is an important factor affecting aerosol scattering, absorption and ability to act as cloud condensation nuclei (CCN). Therefore, the ratio of OA to OC mass (Turpin and Lim, 2001; Aiken et al., 2008) requires careful investigation. Furthermore, OA components differ in their volatility, solubility, hygroscopicity, chemical reactivity and their physical and optical properties. Due to the chemical complexity of the organic component of aerosols (Goldstein

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and Galbally, 2007), only simplified representations are introduced in global chemistry climate models (Kanakidou et al., 2005; Hallquist et al., 2009). As a compromise between simplicity and accuracy, the net effect of the complex mixture of OA is described by only a limited number of representative compounds or surrogates.

## 1.2 Sources

Kanakidou et al. (2005) reviewed how organic aerosols were incorporated in global chemistry transport (CTM) and general circulation models (GCMs), and identified gaps of knowledge that deserved further investigation. The POA sources include fossil fuel, biofuel and biomass burning, as well as the less understood sources of marine OA, biological particles and soil organic matter on dust (Kanakidou et al., 2012 and references therein). Biogenic VOCs (BVOCs) greatly contribute to OA formation (e.g. Griffin et al., 1999b; Kanakidou et al., 2012), implying that significant feedbacks exist between the biosphere, the atmosphere and climate that affect the OA levels in the atmosphere, which was also demonstrated by more recent studies (Tsigaridis et al., 2005; Arneth et al., 2010; Carslaw et al., 2010; Paasonen et al., 2013). In addition, oxidant and pollutant enhancement by human induced emissions is expected to increase OA levels, even those chemically formed by BVOC (Hoyle et al., 2011 and references therein); it is therefore conceivable that some portion of the ambient biogenic SOA, which would had been absent under preindustrial conditions, can be removed by controlling emissions of anthropogenic pollutants (Carlton et al., 2010). Goldstein and Galbally (2007) estimated that SOA formation could be as high as  $910 \text{ TgCa}^{-1}$ , which is at least an order of magnitude higher than any SOA formation modeling study, as shown here. Spracklen et al. (2011), were able to reconcile AMS observations (mostly from the Northern Hemisphere mid-latitudes during summer) with global CTM simulations by estimating a large SOA source ( $140 \text{ Tg a}^{-1}$ ).  $100 \text{ Tg a}^{-1}$  was characterized as anthropogenically controlled, 90 % of which was possibly linked to anthropogenically-enhanced SOA formation from BVOC oxidation. Similar conclusions have been reached by Heald et al. (2011), by comparing aircraft AMS observations of submicron OA with the results

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of another global model, and by Heald et al. (2010) by accounting for the satellite-measured aerosol optical depth that could be possibly due to OA. Recently, Carlton and Turpin (2013) have shown that anthropogenically-enhanced aerosol water in the Eastern USA could lead to an increase in WSOC from BVOC. Although large uncertainties still exist in SOA modeling, there is a need for models to document and improve treatments of solubility, hygroscopicity, volatility and optical properties of the OA from different sources. The SOA formation from anthropogenic VOCs, despite a recent estimate of  $13.5 \text{ Tga}^{-1}$  which makes it a non-negligible SOA source in polluted regions (De Gouw and Jimenez, 2009), is mostly neglected by global models.

### 1.3 Atmospheric processing

Improvement in our understanding and quantification of the emissions of POA and SOA precursors demonstrated from earlier review studies (Kanakidou et al., 2005; Fuzzi et al., 2006) motivated a number of experimental, chamber and field studies that have also significantly enhanced our knowledge on the OA atmospheric cycle. Aging, both physical (e.g. condensation and coagulation) and chemical (in any phase), has been suggested as a significant contributor to the observed OA levels (Kanakidou et al., 2005; Fuzzi et al., 2006; Hallquist et al., 2009), which influences the amount and properties of organic material in the aerosol phase, and occurs at different rates and via different mechanisms in the various atmospheric compartments (e.g. urban/rural/marine boundary layers, low/middle/upper troposphere) (e.g. Molina et al., 2004; Ervens et al., 2011). Despite these advances in understanding, such OA processing remains to date either missing or very poorly parameterized in global models, since advances in OA parameterizations are limited by weak observational constraints. Zhang et al. (2007) and Jimenez et al. (2009) compiled experimental evidence showing that most of the OA in the atmosphere has undergone chemical aging, most likely via SOA formation, and is significantly oxygenated, with lower volatility and higher hygroscopicity than its precursors. To explain these large amounts of oxygenated OA several chemical pathways have been suggested (Hallquist et al., 2009; Jimenez et al., 2009), which differ

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5 in the O/C atomic ratio and in the volatility changes they induce to the parent compounds. Donahue et al. (2006) suggested lumping organic compounds according to their volatility and developed the volatility-basis set (VBS) algorithm to parameterize the many organic compounds present in the atmosphere into several lumped OA species  
10 of different volatility. Chemical aging via gas-phase reactions in the parameterization resulted in changes in the volatility of the species; this has been implemented for SOA from VOCs (e.g. Tsimpidi et al., 2010) and also for SOA from semivolatile and intermediate volatility species (Robinson et al., 2007). However, the implementation of VBS into global models is hindered both by the large number of tracers required, and the underlying uncertainties and free parameters involved. The VBS method was recently  
15 expanded to account for the degree of oxidation of OA, by tracking the O/C content of the organics per volatility class; the method is called 2-D VBS (Donahue et al., 2011) and has been successfully used to simulate the evolution of OA in field campaigns (Murphy et al., 2011, 2012). Unfortunately, this new approach needs an even larger number of tracers, which makes it extremely difficult to implement in a global climate model without a large performance penalty. Still, it certainly adds value to our OA understanding, since the ratio of organic matter (OM) to organic carbon (OM/OC), an alternative way to describe the degree of oxidation of OA, does greatly vary in time and  
20 space (Turpin and Lim, 2001). This variability is either neglected or taken into account in a very simplistic way in models.

25 Yu (2011) extended the two-product SOA formation scheme in the GEOS-Chem model by taking into account the volatility changes of secondary organic gases arising from the oxidative aging process (Jimenez et al., 2009) as well as the kinetic condensation of low-volatility secondary organic gases. It was shown that, over many parts of the continents, low volatility secondary organic gases concentrations are generally a factor of ~ 2–20 higher than those of sulfuric acid gas and the kinetic condensation of low volatility secondary organic gases significantly enhances particle growth rates. Based on this computationally efficient new SOA formation scheme, annual mean SOA mass concentrations in many parts of the boundary layer increase by a factor of 2–10,

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in better agreement with aerosol mass spectrometer (AMS) SOA measurements (Yu, 2011).

Hallquist et al. (2009) also summarized new laboratory data that provided insight into the chemical reaction pathways for the formation of oligomers and other higher molecular weight products observed in SOA. They determined higher production rates of SOA from their precursors' oxidation than earlier measurement studies and linked the dependence of SOA yield from VOC oxidation on the oxidant levels. In chamber experiments, Volkamer et al. (2009) have shown that even small ( $C_2$ ) molecules undergoing aqueous phase reactions can produce low volatility material and contribute to SOA formation in the atmosphere, a process that was reviewed by Ervens et al. (2011) and Lim et al. (2013). The global modeling study of Myriokefalitakis et al. (2011) has shown that multiphase reactions of organics significantly increase the OA mass and its oxygen content, while Murphy et al. (2012) suggested that these reactions are not enough to explain the observed O/C content of OA.

## 1.4 Losses

Hallquist et al. (2009) used the VBS concept and estimated the atmospheric deposition of OA to be  $150 \text{ Tg a}^{-1}$ , higher than earlier estimates and similar to the total particulate OC deposition of  $147 \text{ Tg a}^{-1}$  ( $109 \text{ Tg a}^{-1}$  of WSOC) calculated by Kanakidou et al. (2012). Dry and wet removal of organic vapors that are in thermodynamic equilibrium with SOA becomes increasingly important with atmospheric processing (Hodzic et al., 2013) and was found to lead to 10–30 % (up to 50 %) removal of anthropogenic (biogenic) SOA (Hodzic et al., in 2014). Volatilization of OA upon heterogeneous oxidation has been observed for laboratory and ambient particles (George and Abbatt, 2010) and might be a significant OA sink (Heald et al., 2011).

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## 1.5 Motivation and aim

During the AeroCom phase I modeling experiments (Textor et al., 2006), although most of the models considered both primary and secondary OA sources, OA was simulated in a very simplified way in which both primary and secondary OA were treated as non-volatile. OA was only allowed to age via hydrophobic-to-hydrophilic conversion, and be removed from the atmosphere by particle deposition. Comparisons of individual models with OA observations have shown a large underestimate of the organic aerosol component by models, especially in polluted areas (Volkamer et al., 2006 and references therein). That review showed that the underestimate of SOA by models increases during long-range transport, with largest discrepancies in the free troposphere, suggesting missing sources or underestimated atmospheric processing of organics in models.

Several global models now treat SOA as semi-volatile, as detailed below, which enables potentially more accurate model calculations. Some models also account for intermediate volatility organics, multiphase chemistry and semi-volatile POA (e.g. Pye and Seinfeld, 2010; Jathar et al., 2011; Myriokefalitakis et al., 2011; Lin et al., 2012), with encouraging results in reducing the difference between models and observations. Indeed, the modeled SOA concentrations in Mexico City were much closer to observations when intermediate volatility organics were taken into account in a regional model, although it was unclear if the model-observation gap was reduced for the right reasons (Hodzic et al., 2010). However, OA simulations have many degrees of freedom due to incomplete knowledge on the behavior and fate of OA in the troposphere. Thus, several assumptions made are translated to model tuning parameters that vary greatly between models.

This organic aerosol AeroCom intercomparison aims to update the evaluation of OA modeling by documenting the current status of global modeling of OA in the troposphere, identifying weaknesses that still exist in models, as well as explaining the similarities and differences that exist between models and observations. It quantifies the uncertainties in surface OA concentrations and attributes them to major contributors. It

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also attempts to identify and analyze potential model systematic biases. The ensemble of the simulations is used to build an integrated and robust view of our understanding of organic aerosol sources and sinks in the troposphere.

## 1.6 Terminology

In atmospheric OA research, several naming conventions and abbreviations are used, often ambiguously and inconsistently between authors. To avoid confusion, we clarify here the conventions adopted in this paper, which we use throughout. Note that some aspects of our terminology are different from the very recent VBS-centered attempt by Murphy et al. (2013) to clarify this ambiguity systematically; new model development is required from modelers to adopt the new naming convention in future model simulations.

- Organic Aerosol (OA) and the main OA components, like i.e. Primary and Secondary OA (POA and SOA, respectively): we use these terms to refer to the total mass that organic compounds have in the aerosol phase, including H, O and potentially other elements like N, S and P. Other authors have used the term organic matter (OM), which is synonymous to our OA definition. The units used are  $\mu\text{g m}^{-3}$  for surface mass concentrations at ambient conditions and Tg for burden and budget calculations. OA amounts exclude the water associated with it (assuming that OA is hygroscopic), an important additional component that affects particle size, refractive index and light scattering efficiency.
- Organic Carbon (OC), together with other OC components, like i.e. Primary and Secondary OC (POC and SOC, respectively): these terms refer to the mass of carbon present in OA, instead of the total OA mass. The units used here are  $\mu\text{gC m}^{-3}$  for surface mass concentrations. This is typically the terminology that is used when comparing model results with filter measurements analyzed by thermal-optical methods.

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OA mass can increase for constant OC, due to oxidative aging; this is something that very few models calculate, and should be improved in the future. The OA/OC ratio is discussed in more detail in Sect. 2.4. Care should also be taken for the case of methane sulfonic acid (MSA), since the letter A stands for Acid, not Aerosol, as in OA.

- 5 When reporting MSA results, we refer to the total methane sulfonic acid mass present in OA and not its carbon mass only, unless clearly stated otherwise.

## 2 Description of models

The models participating in the present study differ in (a) the spatial resolution, both horizontal and vertical, (b) the underlying model with which the aerosol calculations are coupled, which can be either a CTM or a GCM, and will be named “host model” from now on, (c) the emissions used, both for POA and SOA precursors, as well as for other gaseous and aerosol tracers, (d) the inclusion or not of aerosol microphysics, which are implemented in multiple ways (Mann et al., 2013), and (e) the OA processes simulated, i.e. the chemical and physical processes that change existing OA (such as oxidative aging), and the representation of SOA formation.

15 The complexity of the OA calculations varies greatly between models (Table 1). There are differences in OA emission source strength, both for primary particles (Table 2) and precursors of secondary OA (Table 3), as well as in the total number of OA tracers used (2 to 62; Table 4) and their properties. Although a classification is difficult, one can categorize the models in various groups when considering OA modeling from different perspectives. The classification used here will be presented later (Sect. 2.3).

20 Some models using the same host model have very specific (and not necessarily few) differences. ECHAM5-HAM2, ECHAM5-HAMMOZ, ECHAM5-SALSA and EMAC use the same host model (ECHAM5) but different aerosol parameterizations: the first two use M7 (modal), ECHAM5-SALSA uses SALSA (sectional) and EMAC uses a modified version of M7. ECHAM5-HAMMOZ uses the previous version of the HAM aerosol module, which does not take into account the detailed SOA formation introduced in

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ECHAM5-HAM2 (O'Donnell et al., 2011). GEOS-Chem and GEOS-Chem-APM are using the same host model (GEOS-Chem) but different aerosol representations: the first uses the default bulk aerosol scheme, while the latter uses a size-resolved (bin) advanced particle microphysics (APM) module (Yu and Luo, 2009). GISS-CMU-VBS and GISS-CMU-TOMAS use the same host GCM (GISS-II'), with the only difference being in the calculation of OA: the first one uses a bulk aerosol scheme with the VBS approach (Donahue et al., 2006; Jathar et al., 2011), and the second one the aerosol microphysics scheme TOMAS (Adams and Seinfeld, 2002; Lee and Adams, 2010, 2012). Similarly, GISS-MATRIX, the two GISS-modelE models and GISS-TOMAS use the same host GCM (GISS-E2), but they have different aerosol representations: the former uses the aerosol microphysics module MATRIX (Bauer et al., 2008), the two modelE versions have a bulk aerosol scheme (Koch et al., 2006; Miller et al., 2006; Koch et al., 2007; Tsigaridis and Kanakidou, 2007; Tsigaridis et al., 2013) and the latter uses the same aerosol microphysics scheme as GISS-CMU-TOMAS (Lee and Adams, 2012; Lee et al., 2014). GISS-modelE-G and GISS-modelE-I only differ in the emissions used; they both have CMIP5 anthropogenic emissions for all tracers (Lamarque et al., 2010), but GISS-modelE-G uses GFED3 (van der Werf et al., 2010) for biomass burning. GLOMAPbin and GLOMAPmode use the same host CTM (TOMCAT; Chipperfield, 2006), with the only difference being the sectional and modal aerosol microphysics calculations (Mann et al., 2012). TM4-ECPL-FNP is almost identical with TM4-ECPL-F, but also takes into account the contribution to OA from primary biological particles and soil dust in the fine and coarse modes (Kanakidou et al., 2012). These two models also use different biogenic and anthropogenic VOC emission inventories (Table 2 and Table 3).

All model results presented here come from monthly mean data, while measurements are averaged in monthly mean values, prior to any comparison with model data.



## 2.1 Meteorology

One major difference between the configurations of the models is the meteorology and meteorological year used. This affects aerosol transport, removal, chemistry (e.g. temperature dependence of reaction rates) and gas-particle partitioning of semi-volatile species. In some models meteorology also directly affects natural aerosol emissions, like wind-driven sea salt, marine organic aerosol, dust and VOC emissions from the vegetation and oceans. Indirectly, meteorology affects MSA sources, since MSA is produced via dimethyl sulfide (DMS) oxidation, whose source is affected by wind speed and its oxidation depends on chemical rates.

Several climate models that participated in this intercomparison calculate the meteorology online. These are BCC, CAM4-Oslo, CAM5-MAM3, CanAM-PAM, GISS-CMU-VBS and GISS-CMU-TOMAS. In addition, the climate models GISS-MATRIX, GISS-modelE-G, GISS-modelE-I and SPRINTARS are nudged to the NCEP reanalysis (Kalnay et al., 1996), GISS-TOMAS is nudged to MERRA meteorology (Rienecker et al., 2011), HadGEM2-ES and ECHAM5-HAMMOZ are nudged to the operational ECMWF meteorology (<http://www.ecmwf.int/products/data/archive/descriptions/od>), and LMDz-INCA is nudged to ECMWF reanalysis from the Integrated Forecast System. The remaining models use a variety of prescribed meteorology datasets for the year 2006 (Table 1), except that GISS-CMU-VBS uses 2008, IMPACT uses 1997, and TM4-ECPL-FNP uses 2005.

## 2.2 Emissions

All participating models include POA in their simulations. The sources are both anthropogenic and biogenic, and can be classified as follows:

1. Fuel emissions. These exclusively anthropogenic sources include fossil fuel and biofuel burning. All models include these sources, but the emission inventories used are not always the same (Table 2). A number of models used emissions for the year 2000; others used emissions for the year 2006, and one for the year 2005

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(TM4-ECPL-FNP). Cooking emissions, which contribute by up to 50 % of the POA in many urban areas (Mohr et al., 2012) are not included in any model.

2. Biomass burning. As in the case of fuel emissions, not all models use the same sources or representative years. Only about half of the models use biomass burning emissions from the year 2006 (Table 2), which is the reference year in the present study. Biomass burning is the largest POA source; it has significant inter-annual and strong seasonal variability and is the most uncertain POA source on a global scale (Andreae and Merlet, 2001), making it extremely important for comparison with measurements, especially at remote sites, to properly represent this source. Comparisons of several model simulations with the smoke aerosol optical depth (AOD) observed by MODIS have indicated a systematic underestimation when emissions from bottom-up inventories like GFED, used by several models here, are used. The underestimation may be as high as a factor of three on the global scale (Kaiser et al., 2012 and references therein) and strongly varies by region (Petrenko et al., 2012).
3. “Pseudo” primary non-volatile SOA fluxes. A number of models parameterize SOA chemical production in the atmosphere as a source of non-volatile aerosol emitted directly from vegetation. SOA is then modified similarly to POA by processes like transport, chemical aging, growth, coagulation and condensation, among others, depending on the model. BCC, CanAM-PAM, ECHAM5-HAMMOZ, ECHAM5-SALSA, ECMWF-GEMS, EMAC, GISS-CMU-TOMAS, LMDz-INCA and TM5 use a global source of  $19.1 \text{ Tg a}^{-1}$  (Dentener et al., 2006). This source is equivalent to a 15 % yield from the year 1990 monoterpenes emissions (Guenther et al., 1995) and is identical with the source used during the AeroCom phase I experiments. GISS-CMU-TOMAS, GISS-TOMAS, GLOMAPbin and GLOMAPmode also use the same approach (based on the Guenther et al. (1995) emissions, except GISS-TOMAS which is based on Lathière et al. (2005)) but with SOA produced according to an assumed molar yield following oxidation (see Sect. 2.3 and Table 4),

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which results in a calculated SOA source of 19.1, 17.1, 23.1, and 23.0 Tga<sup>-1</sup>, respectively. GISS-MATRIX and GISS-TOMAS use a 10 % yield (17.1 Tga<sup>-1</sup>) from monoterpenes emissions for the year 1990 from Lathièrè et al. (2005), while GMI and GOCART assume a 10 % yield (12.7 Tga<sup>-1</sup>) from the Guenther et al. (1995) monoterpene emissions. In the case of CAM4-Oslo the strength of the secondary source suggested by Dentener et al. (2006) has been scaled up to 37.5 Tga<sup>-1</sup>, based on Hoyle et al. (2007).

4. Marine sources. Few models take into account marine sources of organic aerosols (see Sect. 2.3); these depend on sea spray emissions. The GISS-modelE-G and GISS-modelE-I source depends on SeaWiFS chlorophyll *a* measurements from the year 2000 (Tsigaridis et al., 2013), while IMPACT and TM4-ECPL-F/FNP calculations use the MODIS chlorophyll *a* data from the corresponding simulated year. However, recent observations indicate the presence of marine organic aerosol over oligotrophic areas; this can be either due to long-range transport, or a missing source not accounted for with the current source parameterizations, or both. CAM4-Oslo also has marine organic emissions, with a global flux based on Spracklen et al. (2008), and spatial distribution given by the prescribed AeroCom phase I fine mode sea salt emissions (Dentener et al., 2006).
5. Other primary sources. TM4-ECPL-FNP (Kanakidou et al., 2012) includes some fine-mode POA sources that do not exist in any other global model in this inter-comparison. These consist of primary biological particle emissions from plants (25 Tga<sup>-1</sup>) and soil organic matter on dust (0.2 Tga<sup>-1</sup>).

In addition to the primary aerosol emissions, the inventories used for the precursors of secondary organic aerosols are also both very diverse and of great importance. These are presented in Table 3.

## 2.3 Organic aerosol speciation

In the present work, we have separated organic aerosols into five categories, as described below and summarized in Table 4. The models are then grouped based on their OA parameterizations in the figures.

1. tPOA, for terrestrial primary organic aerosol, which includes primary emissions from fossil fuel, biofuel and biomass burning. All models participating in this intercomparison include these three tPOA sources. Several models also consider a biogenic secondary organic aerosol source that is included in tPOA (BCC, CAM4-Oslo, CanAM-PAM, ECHAM5-HAMMOZ, ECHAM5-SALSA, ECMWF-GEMS, EMAC, GISS-CMU-TOMAS, GISS-MATRIX, GISS-TOMAS, GMI, GOCART, LMDz-INCA, SPRINTARS and TM5), as discussed earlier. This is considered to be linked with monoterpene emissions (Guenther et al., 1995), producing non-volatile aerosol mass with a fixed yield as discussed in Sect. 2.2. Some models have a simplified chemistry which produces non-volatile SOA, also included in tPOA: in GISS-CMU-TOMAS and GISS-TOMAS a generic SOA precursor is emitted in the gas phase representing all SOA precursor gases (Dentener et al., 2006; 15 % of the monoterpenes emissions, emitted in the gas-phase) which forms a non-volatile SOA tracer (which is included in tPOA) with a first-order loss rate of 12 h. In GISS-TOMAS the SOA precursor emissions are based on terpenes, with a 10 % yield, while *a*-pinene oxidation by all major oxidants (OH, O<sub>3</sub>, NO<sub>3</sub>) produces non-volatile SOA (included in tPOA) with a 13 % yield in GLOMAPbin and GLOMAPmode. SPRINTARS has a 9.2 % yield of non-volatile SOA (Griffin et al., 1999a; Griffin et al., 1999b) from monoterpenes emissions, and considers this tracer as inert and tracks it separately, in contrast to the other models that produce non-volatile SOA and track it together with tPOA. SOA from anthropogenic VOCs is included in only a few models.

2. mPOA, for primary organic aerosol from marine sources. CAM4-Oslo has a primary marine organic source of 8 Tg a<sup>-1</sup> (Spracklen et al., 2008) with the same

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emissions distribution as sea salt (provided by Dentener et al., 2006) included in tPOA. IMPACT includes a mPOA source of  $35 \text{ Tg a}^{-1}$  (Gantt et al., 2009b) which scales with chlorophyll *a* and sea salt as a proxy of marine biological activity (O'Dowd et al., 2004), while GISS-modelE-G/I and TM4-ECPL-F/FNP include a similar source of submicron mPOA based on Vignati et al. (2010). The GISS-modelE-G/I source is described in Tsigaridis et al. (2013) and the TM4-ECPL-F/FNP mPOA source in Myriokefalitakis et al. (2010). It has to be noted that these two studies have a factor of 10 difference in submicron mPOA source strength, despite having very similar source function parameterizations. This results from differences in sea-spray size distribution assumptions, as discussed in Tsigaridis et al. (2013). In addition to the fine mode mPOA source, TM4-ECPL-FNP accounts for about  $30 \text{ Tg Ca}^{-1}$  of coarse mode mPOA (Kanakidou et al., 2012), but that was not taken into account in the present study.

3. trSOA, for “traditional” secondary organic aerosol, which is produced by gas to particle mass transfer of secondary organic material, either assuming the material has a finite vapor pressure (a gas-particle partitioning process) or that it has zero vapor pressure (a condensation process). The most common precursors of SOA used across models are isoprene and terpenes, although few models have other precursors as well (Table 3). All models have some form of trSOA, either included in tPOA (as explained above), or via an explicit treatment of the semi-volatile oxidation products of the precursor VOCs. For the models other than the ones presented in (a) above which treat SOA as part of tPOA, the approach used and species taken into account differ. CAM5-MAM3 prescribes mass yields from 5 trSOA precursor categories (isoprene, terpenes, aromatics, higher molecular weight alkanes and alkenes, with yields of 6.0, 37.5, 22.5, 7.5, and 7.5%, respectively), which then reversibly and kinetically partition to the aerosol phase. GISS-CMU-VBS uses the volatility-basis set, but without aging for the biogenic trSOA. The rest of the models use the 2-product model approach to calculate trSOA; see the references column in Table 1 for more details. GEOS-Chem-APM

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considers the volatility changes of the gaseous semi-volatile compounds arising from the oxidation aging process, as well as the kinetic condensation of low volatility gases (Yu, 2011). HadGEM2-ES does not calculate trSOA online; instead, it uses an offline 3-dimensional monthly mean trSOA climatology obtained from the STOCHEM CTM (Derwent et al., 2003). The 2-product model implemented in IM-AGES was modified to account for the effect of water uptake on the partitioning of semi-volatile organics, through activity coefficients parameterized using a detailed model for alpha-pinene SOA (Ceulemans et al., 2011). IMPACT predicts semi-volatile SOA from organic nitrates and peroxides using the gas-particle partitioning parameterization with an explicit gas-phase organic chemistry. These condensed semi-volatile compounds are assumed to undergo further aerosol-phase reactions to form non-evaporative SOA with a fixed 1 day e-folding time (Lin et al., 2012). The two TM4-ECPL models account for SOA aging by gas-phase oxidation by OH with a rate of  $10^{-12} \text{ cm}^3 \text{ molec}^{-1} \text{ s}^{-1}$ , while the conversion of insoluble POA to soluble is parameterized as described by Tsigaridis and Kanakidou (2003) with a decay rate that depends on  $\text{O}_3$  concentration and water vapor availability, which corresponds to about 1 day global mean turnover time, with strong spatial variability.

4. ntrSOA, for non-traditional secondary organic aerosol, which comes from a variety of sources, as explained below. GISS-CMU-VBS includes the VBS (Robinson et al., 2007), which allows tPOA to evaporate and age (via oxidation) in the gas-phase, producing less volatile gas-phase products, which can again partition between the gas and aerosol phases. This model, which also takes into account the intermediate volatility species as additional sources of OA, enables the application of the partitioning theory to POA and its associated vapors as well, not only SOA. The aerosol phase of these oxidized products is termed ntrSOA. The impact of this process strongly affects the chemical composition of SOA and will be discussed later (Sect. 4.3.3). Other models, namely IMAGES, IMPACT, and TM4-ECPL-F/FNP, include an aqueous-phase oxidation pathway of small organic

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molecules like glyoxal and methylglyoxal that produces low volatility compounds and oligomers in cloud and aerosol water (Fu et al., 2008, 2009; Stavrou et al., 2009; Myriokefalitakis et al., 2011), with the two TM4-ECPL models having a primary glyoxal source from the oceans of  $4.1 \text{ TgCa}^{-1}$ , which is not present in the other two models. Glyoxal and methylglyoxal are highly reactive species in the aqueous phase. The aqueous-phase reactions can occur both in aerosol water and cloud droplets; after droplet evaporation, the residual organic compounds remain in the aerosol phase in the form of OA. By applying a reactive uptake ( $\gamma$ ) of glyoxal and methylglyoxal on aqueous particles and cloud drops (Liggio et al., 2005), IMAGES and IMPACT parameterized the irreversible surface-controlled uptake of these soluble gas-phase species. On the other hand, Myriokefalitakis et al. (2011) applied a much more detailed aqueous-phase chemical scheme in cloud droplets in order to produce oxalate. IMPACT also includes ntrSOA formation from the uptake of gas-phase epoxides onto aqueous sulfate aerosol (Paulot et al., 2009), a potentially important contributor to SOA (Lin et al., 2012).

5. MSA, an oxidation product of DMS, is also a SOA component. Although a minor organic aerosol component on the global scale, MSA can be very important at remote oceanic regions, especially when mPOA is relatively low: observations indicate that MSA can be at least 10% of the total WSOC mass (Sciare et al., 2001; Facchini et al., 2008) at marine locations. Only CAM4-Oslo, GEOS-Chem-APM, GISS-modelE-G/I, IMPACT, LMDz-INCA, TM4-ECPL-F/FNP and TM5 have this tracer, which has been typically neglected from the organic aerosol budget in modeling studies. In CAM4-Oslo, MSA is included in tPOA, in IMPACT is included in mPOA (which is in turn included in tPOA), whereas in the other models it is individually tracked.

A summary of the OA processes included in the models is presented in the figures. The total organic aerosol mass is calculated as follows:

$$\text{OA} = \text{tPOA} + \text{mPOA} + \text{trSOA} + \text{ntrSOA} + \text{MSA} \quad (1)$$

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In addition to this categorization, in order to compare with AMS data (see Sect. 3) we separate the modeled OA into HOA (hydrocarbon-like OA) and OOA (oxygenated OA) as defined by Zhang et al. (2005), when sufficient information on hydrophobic/hydrophilic speciation from the models is available. We use the terminology HOA/OOA instead of water soluble/insoluble OC (WSOC/WIOC), and compare only with AMS organic aerosol data, in order to contrast with the OC measurements that refer to organic carbon. The separation into HOA and OOA has been provided by only a few models: ECHAM5-HAM2, ECMWF-GEMS, EMAC, GISS-modelE-G, GISS-modelE-I, GISS-TOMAS, GLOMAPbin, GLOMAPmode, IMAGES, LMDz-INCA, TM4-ECPL-F, TM4-ECPL-FNP and TM5. From the AMS perspective, the total OA is calculated as follows:

$$\text{OA} = \text{HOA} + \text{OOA} \quad (2)$$

Further subdivisions in other categories of OOA (Jimenez et al., 2009) are neglected in this study. In addition, the term POA used in Zhang et al. (2011) as a surrogate of different HOA categories is also not taken into account here.

## 2.4 OA/OC and O/C ratios

To calculate the total amount of organic aerosol mass concentration for each model, we apply the following equation:

$$\text{OA}_i = \text{OC}_i \cdot (\text{OA}/\text{OC})_i \quad (3)$$

where  $(\text{OA}/\text{OC})_i$  is the organic aerosol to organic carbon ratio for aerosol tracer  $i$  (Table 4). OA/OC, frequently termed as OM/OC in the literature (OM: Organic Matter), was found to correlate extremely well with the O/C ratio in Mexico City and chamber data (Aiken et al., 2008), because of low N/C ratios. A low OA/OC ratio is also indicative of “fresh” OA as deduced from observations (Turpin and Lim, 2001; Philip et al., 2014). The OA/OC ratio varies greatly between models, with many of them setting



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OA/OC = 1.4 as a constant for all OA sources. Some models use different OA/OC ratios for every OA tracer: IMAGES, IMPACT, and the two TM4-ECPL models calculate the specific OA/OC ratio for each of their aerosol tracers, depending on their sources and chemical identity. CAM4-Oslo uses 1.4 for fossil fuel and biofuel, OsloCTM2 and SPRINTARS use 1.6, while all three models use 2.6 for biomass burning. In the case of CAM4-Oslo and SPRINTARS, it is not possible to accurately calculate the OC concentration from the model fields, since they only track one tracer. For this, we used a single value, that of the fossil fuel each model is using, which will lead to an underestimation of their OC concentration (but not of OA) close to biomass burning sources. The remaining models use a constant OA/OC ratio: GEOS-Chem and GEOS-Chem-APM use a specified value of 2.1, GISS-CMU-VBS and GISS-CMU-TOMAS use 1.8, and all other models use 1.4. Observations (Turpin and Lim, 2001; Aiken et al., 2008) suggest that OA/OC values of  $1.6 \pm 0.2$  and  $2.1 \pm 0.2$  are good approximations for urban and non-urban aerosols, respectively, indicating that most models might use OA/OC values that are low. The study of both the OA/OC and O/C ratios is extremely important and warrants a dedicated investigation; although this will be mentioned in the present work, it will be studied in detail in the future.

### 3 Measurements

The compilations of  $PM_{2.5}$  OC measurements by Bahadur et al. (2009) and  $PM_1$  OA measurements by Zhang et al. (2007) form the basis for the present study. Additional OC and OA observations from continuous monitoring networks and individual case studies reported in the literature have been used to increase the spatial and temporal coverage of the observational database for model evaluation.

The OC measurements reported by Bahadur et al. (2009) include data from:

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- The Interagency Monitoring of Protected Visual Environments (IMPROVE; <http://vista.cira.colostate.edu/IMPROVE>), which is the American monitoring network for national parks and wilderness areas.
- The Speciated Trends Network (STN) administered by the Environmental Protection Agency (Air Quality System Environmental Protection Agency (AQSEPA); <http://www.epa.gov/ttn/airs/airsaqs>), which mainly consists of urban monitoring stations within the USA.
- The California Regional PM<sub>10</sub>/PM<sub>2.5</sub> Air Quality Study (CRPAQS) (Chow et al., 2006).
- The North American Research Strategy for Tropospheric Ozone (NARSTO; <http://www.narsto.org>), which consists of measurements in Mexico, USA and Canada.
- The New England Air Quality Studies (NEAQS; <http://www.esrl.noaa.gov/csd/projects/neaqs>), which contains measurements from the New England region, as a part of NOAA field studies.
- The Southeastern Aerosol Research and Characterization Study (SEARCH; Hansen et al., 2003), which is a monitoring network for the southeastern United States.
- The European Monitoring and Evaluation Programme (EMEP; <http://www.emep.int>). EMEP is a European monitoring network with a few hundred monitoring stations all over Europe; only a few measure OC, which are used here.
- The Construction, Use and Delivery of a European Aerosol Database (CREATE; <http://www.nilu.no/projects/ccc/create>). CREATE is a database that compiles aerosol data from 8 European countries.
- The Hong Kong Environmental Protection Agency (<http://www.epd.gov.hk/epd/eindex.html>), with measurements from the extended area of Hong Kong.

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These datasets have been extended by numerous new measurements from published studies (Chow et al., 1993; Smith et al., 1996; Zappoli et al., 1999; Kim et al., 2000; Eatough et al., 2001; Krivacsy et al., 2001; Artaxo et al., 2002; Balasubramanian et al., 2003; Eatough et al., 2003; Gatari and Boman, 2003; Graham et al., 2003; Long et al., 2003; Fan et al., 2004; He et al., 2004; Ho et al., 2004; Jeong et al., 2004; Salma et al., 2004; Sawant et al., 2004; Ward et al., 2004; Xu et al., 2004; Hueglin et al., 2005; Kim et al., 2005; Fuzzi et al., 2006; Koulouri et al., 2008; Pindado et al., 2009; Sciare et al., 2009; Li et al., 2010; Shakya et al., 2010; Zhang et al., 2012c) enhancing primarily the spatial, but also the temporal availability of comparison points.

The IMPROVE and AQSEPA networks cover more than adequately most of the United States. The EMEP monitoring network together with the European Integrated project on Aerosol, Cloud, Climate, and Air Quality Interactions (EUCAARI) and CRE-ATE datasets and other studies found in the literature provide good coverage of a large part of Europe, with stations in 17 countries. Although the spatial and temporal coverage is not as extensive as in the USA, it provides a comprehensive representation of different sources and chemical environments over Europe. There are limited measurements from Asia, with many of them being at urban or urban-influenced locations in India and China. South America, Africa and Oceania have very poor spatial and temporal coverage, despite the importance of the tropical forests of the former two on the global OA budget. Marine areas are almost exclusively covered by short-term measurement campaigns, with the exception of Amsterdam Island in the southern Indian Ocean (Sciare et al., 2009). All OC measurements are  $PM_{2.5}$  or smaller sizes, e.g.  $PM_{1.8}$  (Koulouri et al., 2008).

A rapidly increasing number of AMS OA has been reported in the literature since the work of Zhang et al. (2007). Most of these AMS measurements are available online, in a web page created and maintained by Qi Zhang and Jose-Luis Jimenez (<http://tinyurl.com/ams-database>). We include in this analysis most of the ground-based data available as of January 2013. These data include the only AMS measurements so far available for a whole year (using the ACSM instrument, which is a monitoring version of

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the AMS; Ng et al., 2011), from Welgegund, South Africa (Tiitta et al., 2014); all other stations were measuring for about a month or less. The geographical coverage of the AMS stations is far less dense than the OC measurement locations, but the number of stations is rapidly increasing. Longer records are also starting to appear in the literature (Tiitta et al., 2014), and are expected to increase in the near future.

All station data have been classified in three main categories: urban, remote and marine (Fig. 1). Urban sites are defined as those that are either in cities or highly influenced by them. AMS stations characterized as “urban downwind” fall in this category. Remote sites are defined as those not influenced by local anthropogenic activities and include forested regions, mountains, rural areas, etc. Marine sites are all measurements from ships or from coastal stations that are highly influenced by the marine atmosphere. Only two AMS stations fall in this category (Okinawa, Japan, and Mace Head, Ireland), and for simplicity they were classified in the “remote” category.

The two databases (OC and OA measurements) have been kept separate because of the added complexity related to the OA/OC ratio (Sect. 2.4). Almost all models calculate OA mass concentration, integrated across the fine-mode size distribution where appropriate, which can be compared with AMS measurements without any unit conversion. To compare with filter measurements of OC, we used the models’ assumptions on the OA/OC ratio to convert the modeled OA to OC. As mentioned earlier, the importance of the OA/OC ratio will be explored in the future. The cutoff diameter of aerosols can also be an issue in the present study (Koulouri et al., 2008), but it is not expected to be significant. No model adds fine OA mass from coarse mode sources, and no model allows partitioning of semi-volatile gases to the coarse mode; thus, the difference between the  $PM_{2.5}$  filter measurements and  $PM_1$  AMS data is not expected to be properly resolved by models, even if they include aerosol microphysics calculations.

## 4 Results

### 4.1 Global budgets

The large number of models used in this study adds a significant level of complexity to the interpretation of results, due to the large diversity of inputs and configurations used by the different modeling groups. Despite the large differences between model formulations, on the global scale several interesting similarities and patterns appear, which are frequently associated with the parameterizations and emission inventories used.

#### 4.1.1 Emissions

Global mean model primary organic aerosol emissions used in the models are in the range 34–144 Tg a<sup>-1</sup>. The emissions from most models lie below 80 Tg a<sup>-1</sup> (Fig. 2), with a median value of 56 Tg a<sup>-1</sup>. Notable exceptions are the two GISS-modelE models (G and I), in which about two thirds of the POA emissions come from marine sources (Tsigaridis et al., 2013); without this source, these two models have the same emissions as GISS-MATRIX (39.5 Tg a<sup>-1</sup>) which falls below the 25 % quantile. CAM4-Oslo also has the highest terrestrial sources of all models (144 Tg a<sup>-1</sup>), followed by IMPACT (98 Tg a<sup>-1</sup>) and EMAC (92 Tg a<sup>-1</sup>). All models appear to have similar seasonality in POA emissions that are driven by tPOA, with increased emissions during the Northern Hemisphere summer due to the enhanced contribution of Northern Hemisphere biomass burning emissions from temperate and boreal forests to the total POA fluxes. In addition, several models include SOA sources in tPOA as explained earlier, scaled by BVOC emissions, which also peak during the Northern Hemisphere summer (Guenther et al., 1995, 2006); this contributes to a seasonal cycle of tPOA which is caused by the trSOA treatment as part of tPOA, and should not be interpreted as a tPOA seasonality. Also note that anthropogenic tPOA sources have no seasonality in emission inventories, only biomass burning sources have. The IMPACT model appears to

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have the opposite seasonality, with maximum POA emissions during winter and minimum during late spring-early summer, due to the fossil fuel emissions scaling to fit observations (Wang et al., 2009). The minimum of the emissions for all models except IMPACT is during Northern Hemisphere spring, when neither biomass burning nor the photochemical trSOA sources (included in tPOA by many models) are high.

#### 4.1.2 Chemical production

The chemical production of SOA is much more complex compared to the POA emissions. Firstly, many models include SOA sources as primary emissions, which are included in tPOA (see Sect. 2.3 and Table 4). This type of source was used during AeroCom phase I experiments (Dentener et al., 2006). For AeroCom phase II, 13 out of 31 models still use this source parameterization (Table 5), while 5 models use a simple SOA production rate based on gas-phase oxidation which then forms non-volatile SOA. These 18 models have a median SOA source strength of  $19.1 \text{ Tga}^{-1}$  (mean  $20 \text{ Tga}^{-1}$ ) and a standard deviation of  $4.9 \text{ Tga}^{-1}$  (Fig. 3). Very few models that include this source have provided budget information on the seasonal variability of its SOA source, since it is implicitly included in the tPOA sources and is not tracked separately. However, it has virtually identical seasonality with that of the monoterpene emissions adopted in each model. Two models appear as outliers: CAM5-MAM3 with a chemical production of  $103 \text{ Tga}^{-1}$ , and CAM4-Oslo with  $37 \text{ Tga}^{-1}$ . The reason is that both of these models have scaled up their SOA chemical production.

From the other models that include a more complex calculation of SOA chemical production, there is a large inter-model variability in the source flux, with median  $51 \text{ Tga}^{-1}$  (mean  $58 \text{ Tga}^{-1}$ ) and  $39 \text{ Tga}^{-1}$  standard deviation, based on the 12 out of 14 models that include such parameterizations and have submitted budget information. This is more than twice as high as the models that use the AeroCom phase I parameterization, but with much larger model diversity. The seasonality of OA emissions in all these models peaks during Northern Hemisphere summer (Fig. 3) when VOC fluxes from temperate and boreal forests are at a maximum, while emissions from tropical forests

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are high year-round. Six models (IMAGES, IMPACT, GISS-CMU-VBS, HadGEM2-ES, OsloCTM2 and TM4-ECPL-F) include very strong SOA sources of 120, 119, 79, 64, 53 and 49  $\text{Tg a}^{-1}$ , respectively, followed by CCSM4-Chem (33  $\text{Tg a}^{-1}$ ) and GEOS-Chem (31  $\text{Tg a}^{-1}$ ). About 42 % (50  $\text{Tg a}^{-1}$ ) in IMAGES are due to non-traditional sources (glyoxal and methylglyoxal). The traditional SOA source in IMAGES accounts for water up-  
take, which is found to increase the partitioning of semi-volatile intermediates (Müller, 2009). Monoterpenes alone account for about 40  $\text{Tg a}^{-1}$ . This large contribution is due to the very high SOA yields ( $\sim 0.4$ ) in the oxidation of monoterpenes by OH in low- $\text{NO}_x$  conditions, which are justified by the formation of low-volatility compounds like hydroxy di-hydroperoxides. IMPACT has several non-traditional SOA sources from aqueous chemistry, which can locally contribute as much as 80 % of the total OA mass. CAM5-MAM3 and IMPACT include also anthropogenic precursors. CAM5-MAM3 also uses a factor 1.5 SOA yield increase in order to reduce anthropogenic aerosol indirect forcing, by elevating the importance of SOA during the preindustrial period (Liu et al., 2012). As mentioned before, HadGEM2-ES does not calculate SOA production explicitly; instead, it uses the Derwent et al. (2003) climatology from STOCHEM, which calculates a SOA formation of 64  $\text{Tg a}^{-1}$ . For comparison, satellite-constrained studies estimate that the total OA formation (primary and secondary) can be as high as 150  $\text{Tg a}^{-1}$  with 80 % uncertainty (Heald et al., 2010); AMS-constrained estimates put the total SOA formation rate between 50–380  $\text{Tg a}^{-1}$ , with 140  $\text{Tg a}^{-1}$  being the best estimate (Spracklen et al., 2011), while (Hallquist et al., 2009) estimated using a top-down approach that the best estimate for the total biogenic SOA formation is 88  $\text{Tg Ca}^{-1}$ , out of a total 150  $\text{Tg Ca}^{-1}$  of OC.

The case of GISS-CMU-VBS deserves focus. This model calculates SOA production based on the VBS approach. Its secondary source of 79  $\text{Tg a}^{-1}$  includes not only newly formed SOA both from POA and intermediate volatility organics, but also gas-phase chemical conversion of organic mass that has evaporated from emitted POA, to produce less volatile organics, i.e. mass that has undergone aging in the atmosphere. The traditional SOA sources from biogenic VOC are included in this model like

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in other models that use the 2-product model, but also the chemical conversion of intermediate volatility organics to less volatile OA is taken into account, again with the use of the VBS. Overall, GISS-CMU-VBS presents a similar seasonal pattern of SOA chemical production as other models, but shifted by one month, i.e. peaking in August, when biomass burning is at its maximum in the Northern Hemisphere, instead of maximizing in July, when photochemical activity and biogenic VOC emissions are higher globally. This might be due to the inclusion of the intermediate volatility compounds as SOA precursors, which also have large biomass burning sources. CCSM4-Chem and GEOS-Chem also have a shift in the seasonal maximum. For CCSM4-Chem this is due to strong production from biomass burning sources, while in the case of GEOS-Chem the seasonal cycle seems to be driven by production from Amazonia, which is related with both biogenic and biomass burning emissions.

### 4.1.3 Burden

From the models that have submitted POA burden data (also termed load; the mean total mass in the atmosphere), both its seasonality and amplitude largely follows that of the corresponding POA emissions (Fig. 4), with two notable differences. The two GISS-modelE models have much lower POA burdens (but similar seasonality) than their emissions would imply. The reason is that the mPOA fraction of POA has a very short lifetime of  $\sim 1.5$  days, since mPOA is assumed to be internally mixed with fine mode sea salt, which is removed efficiently due to wet scavenging (Tsigaridis et al., 2013). This keeps the overall load of POA fairly low, and comparable with the models that do not have mPOA. The other difference is GISS-CMU-VBS, which also has a much lower POA load than their emissions would suggest. This is due to the POA aging parameterization, which converts POA into SOA, drastically reducing the POA burden. The other models appear to have the expected POA load, given their emissions, including IMPACT, whose different seasonal variability of the emissions is also reflected on its OA load.



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For the computed SOA load (Fig. 5), all models assume that SOA is very soluble, with 80–100% of its total mass considered soluble, which results in similar globally averaged removal rates across the models. This means that the differences in the SOA loads are expected to be driven primarily by the SOA chemical production, similar to how the POA load is driven by emissions. This is indeed the case for almost all models, with GISS-CMU-VBS, IMAGES, IMPACT, CCSM4-Chem and CAM5-MAM3 having the highest loads, exceeding 1 Tg, with the first two models being as high as 2.3 and 2.2 Tg, respectively, and GEOS-Chem being just below 1 Tg. Spracklen et al. (2011) estimated a global SOA burden of 1.84 Tg, similar to the high-end models that participate in the current intercomparison, but for a SOA formation rate of  $140 \text{ Tg a}^{-1}$ , which is about 20% higher than IMPACT and IMAGES (the models with the strongest SOA formation here), and about 3 times higher than the median SOA formation rate of the models that have a complex SOA parameterization. ECHAM5-HAM2 calculates an increasing load over the course of one year, which is related to the short spin-up time of 3 months, which is not sufficient for the upper tropospheric SOA to reach equilibrium. GEOS-Chem simulates an inverse seasonality when compared with other models, with the maximum load calculated during Northern Hemisphere winter and the minimum during Northern Hemisphere summer. The cycle seems to be dominated by the SOA load over the Southern Ocean; probably the removal processes are slower than other models there, thus SOA may form a uniform band between 30–50° S during the whole austral summer.

With regard to the total OA load, a median of 1.4 Tg (mean 1.6 Tg) and standard deviation of 0.8 Tg is calculated; half the models lie within the range of 1–1.6 Tg (Fig. 6). CAM4-Oslo calculates a global burden of 3.8 Tg, reflecting the very high POA emissions, while IMAGES, IMPACT, GISS-CMU-VBS and CCSM4-Chem calculate a burden of 3.7, 2.6, 2.4 and 2 Tg, respectively, as a result of their high SOA production. Overall, the models calculate very similar total OA load seasonality, which peaks during the Northern Hemisphere summer season, when both primary (biomass burning) and secondary (chemical production) OA sources are high, and minimizes during Northern

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Hemisphere spring, when neither biomass burning nor SOA chemical production is significant in the Northern Hemisphere. The tropical biomass burning and SOA production around December and January both contribute to the secondary maximum that all models calculate during that time. The relative importance of SOA over POA will be discussed in Sect. 4.3.3.

The total OA load is calculated to be mostly lower than the sulfate load in the models that reported budget values for both aerosol components, with a median value of the  $\text{OA}/\text{SO}_4^{2-}$  mass load ratio of 0.77 (mean 0.95). The ratio lies in the range 0.26–2.0; CAM4-Oslo, CAM5-MAM3, GEOS-Chem, GISS-modelE-G/I, IMAGES, IMPACT, and TM4-ECPL-F/FNP calculate values above 1, which means that annually on the global scale OA dominates over sulfate aerosols. That was the case for 5 out of 16 models during AeroCom phase I (Textor et al., 2006). Note however that AeroCom phase I models were simulating the year 2000, while here we simulate the year 2006; interactive chemistry, new sources (mPOA and ntrSOA) and different emission inventories also contribute to significant differences between the two studies. It has to be reminded that even in AeroCom phase II many models are using some emission inventories from a year other than 2006 (Table 2 and Table 3).

### 4.1.4 Deposition

Dry deposition is a minor removal pathway for OA, accounting for a median of  $13 \text{ Tga}^{-1}$  (range  $2\text{--}36 \text{ Tga}^{-1}$ ) and a mean  $15 \text{ Tga}^{-1}$  (standard deviation of  $10 \text{ Tga}^{-1}$ ; Fig. 7). On average, dry deposition is responsible for 15 % of the total OA removal across models. The two TOMAS models and TM5 are calculating by far the lowest dry deposition flux of all, followed by three of the ECHAM5 models, excluding EMAC. The two TOMAS models use essentially the same aerosol microphysics parameterization in two different host models, GISS-II' for GISS-CMU-TOMAS and GISS-E2 for GISS-TOMAS. GISS-modelE-G/I and GISS-MATRIX use the same host model and identical emissions as GISS-TOMAS, a fact that suggests the TOMAS aerosol module (Adams and Seinfeld, 2002) either is less efficient in scavenging OA via dry deposition, or is more efficient

in removing OA from the system via wet deposition, or both. The latter, though, would mean that the OA load (Fig. 6) would be much smaller in GISS-TOMAS in order to have low enough dry deposition fluxes, which does not appear to be the case.

Other than the two TOMAS models, of the remaining models that have submitted dry deposition flux data, three models calculate very low fluxes: ECHAM5-HAM2, ECHAM5-HAMMOZ, and TM5, with the latter already mentioned earlier. The first two models use ECHAM5 as the host model, and all three use the M7 aerosol microphysics module (Vignati et al., 2004). As for the TOMAS case, this is strong evidence that the M7 module does not allow OA to deposit as fast as in most other models; ECHAM5-SALSA, which uses the same host model as ECHAM5-HAM2 and ECHAM5-HAMMOZ, calculates higher dry deposition fluxes than the two ECHAM5 models with M7. The largest difference in dry deposition between the two aerosol microphysics schemes comes from the treatment of external mixing of OA in the accumulation sized particles. ECHAM5-SALSA includes soluble and insoluble OA in the accumulation mode while ECHAM5-HAMMOZ and ECHAM5-HAM2 include only soluble OA. In addition, EMAC, which uses a sectional version of M7 called GMXe, does not calculate as low dry deposition as the models that use the modal version of M7. The fact that there are other models with aerosol microphysics parameterizations in this inter-comparison, both modal and sectional, that do not calculate such low dry deposition fluxes suggests that it is not a general aerosol microphysics calculation issue. Whether the above explanation suffices to explain the low dry deposition, or other processes are involved as well, like very strong wet removal that does not allow time to dry deposition to become effective, the calculated aerosol size distribution, the aerosol properties that impact dry deposition rates, or something else, remains to be explored by dedicated deposition flux model/data comparisons. Also note that we have not assessed this feature of the models against observations, so we do not know which models are closer to observations.

CAM4-Oslo has the highest dry deposition flux of  $36 \text{ Tga}^{-1}$ , which is due to the high OA load. BCC follows with  $33 \text{ Tga}^{-1}$ , which is then followed by the two GISS-modelE

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models and IMAGES with  $\sim 28 \text{ Tga}^{-1}$ . In the case of the two GISS-modelE models, this is due to the strong removal of mPOA, which is internally mixed with sea salt (as explained earlier), while for IMAGES it is due to the high OA load, as a result of strong trSOA formation. BCC is using a smaller mass mean diameter as the size distribution of POA emissions which can explain the high dry deposition flux (Zhang et al., 2012a). Despite these large differences between models, the calculated dry deposition fluxes follow the same seasonal pattern as the aerosol load, presented earlier (Sect. 4.1.3 and Fig. 6).

The effective dry deposition rate coefficient, defined as the ratio of the dry deposition flux over the aerosol burden that is being deposited (Textor et al., 2006), ranges between  $0.005\text{--}0.13 \text{ days}^{-1}$ , with a median value of  $0.025 \text{ days}^{-1}$ , a mean value of  $0.029 \text{ days}^{-1}$  and a standard deviation of  $0.046 \text{ days}^{-1}$ . The diversity (defined as the standard deviation over the mean) has increased since AeroCom phase I, from 0.62 to 0.87. BCC has the largest effective dry deposition rate coefficient,  $0.13 \text{ days}^{-1}$ , more than double than any other model. The models with very low dry deposition fluxes are the ones that have the lowest effective dry deposition rate coefficients, all below  $0.014 \text{ days}^{-1}$ , supporting the hypothesis that their dry deposition flux is probably too low.

By far the most important removal mechanism across all models is wet deposition (Fig. 8). Due to similar OA solubility assumptions across all models, the wet deposition flux largely follows the OA load, both in the annual budget and the seasonality. IMPACT has the highest wet deposition flux of all models ( $209 \text{ Tga}^{-1}$ ) followed by IMAGES ( $163 \text{ Tga}^{-1}$ ), CAM4-Oslo ( $146 \text{ Tga}^{-1}$ ), CAM5-MAM3 ( $134 \text{ Tga}^{-1}$ ). OsloCTM2 ( $128 \text{ Tga}^{-1}$ ) and GISS-modelE-G/I ( $120/125 \text{ Tga}^{-1}$ , respectively). These are the models with the highest OA sources (Fig. 6), thus also with the highest sinks. Wet removal of OA is simulated to range between  $28\text{--}209 \text{ Tga}^{-1}$  for the 26 of the models that reported fluxes, with mean (median) standard deviation values of 86 (70)  $43 \text{ Tga}^{-1}$ , which is on average 85 % of the total OA deposition.

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The effective wet deposition rate coefficient ranges between 0.09–0.24 days<sup>-1</sup>, with a median value of 0.15 days<sup>-1</sup>, a mean value of 0.16 days<sup>-1</sup> and a standard deviation of 0.04 days<sup>-1</sup>. The diversity since AeroCom phase I has virtually not changed, with a slight increase from 0.27 to 0.28. OsloCTM2 has the highest effective wet deposition rate coefficient and LMDz-INCA the lowest.

### 4.1.5 Lifetime

The combination of all sources and sinks affects the load and lifetime of OA, either directly or indirectly. The lifetime of a species is calculated as the ratio of the species burden over its total removal; in the case of aerosols, the removal is dry and wet deposition. Unfortunately, while most model groups have submitted total OA diagnostics to calculate the OA lifetime, few have submitted the diagnostics required to calculate the global mean POA and SOA lifetimes.

The calculated median POA lifetime from the 13 models that reported relevant data is 4.8 days (mean 4.8 ± 1.4 days). The modeled lifetime ranges from 2.7 days for the two GISS-modelE models to 7.6 days for IMAGES (Fig. 9). The GISS-modelE models have the lowest lifetime, which is consistent with roughly two-thirds of POA being removed rapidly with sea salt (as mPOA). There is no clear seasonal signal on the calculated POA lifetime.

The SOA lifetime calculated by 12 out of 31 models also lacks a clear seasonal signal (Fig. 10). The GISS-modelE-G/I models, CCSM4-Chem, ECHAM5-HAM2 and GISS-CMU-VBS have the highest SOA lifetime of 15/14, 14, 13 and 10 days, respectively, which is related to large amounts of SOA in the upper troposphere, where there is virtually no removal mechanism and therefore SOA lifetime is enhanced, until atmospheric circulation or sedimentation brings it to lower layers where it becomes susceptible to removal. For the remaining models that provide information, the calculated SOA lifetime ranges from 2.4 to 6.8 days. The median SOA lifetime from all models that provide budget information is calculated to be 6.1 days (range 2.4–14.8 days), higher than the

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median POA lifetime. Anthropogenic POA, which in general is more hydrophobic than SOA, is almost exclusively emitted close to surface and below clouds, making it more susceptible to dry and wet removal; biomass burning POA can be emitted at higher altitudes (Dentener et al., 2006), while a significant amount of SOA is formed above clouds in the models, where temperatures are low. For instance, in TM4-ECPL-FNP about 42 % of the total SOA mass is formed in the free troposphere, while 98 % of POA mass is emitted in the boundary layer. Furthermore, although one might expect that SOA is more soluble, thus more susceptible to removal, this does not appear to be reflected in the model results; the reason is that SOA can be formed above clouds and avoid removal for long periods of time.

Twenty-four models provide sufficient information to calculate the total OA lifetime, which lies in the range of 3.8–9.6 days, with a median of 5.4 days and a mean of  $5.7 \pm 1.6$  days (Fig. 11). GISS-CMU-TOMAS has a very strong seasonality in the calculated OA lifetime, with maximum during late Northern Hemisphere spring and minimum during late Northern Hemisphere fall, and GISS-CMU-VBS has the highest OA lifetime of all models. As in the case of POA and SOA, there is no clear seasonality in the OA lifetime across models.

### 4.1.6 Optical depth

The aerosol-cloud interactions that comprise the indirect effect have been studied with many of the models used here (e.g. Quaas et al., 2009), and the direct effect has been studied previously, both during AeroCom phase I (Kinne et al., 2006; Schulz et al., 2006) and phase II (Myhre et al., 2013; Samset et al., 2013). The impact of the direct and indirect effects of organic aerosols on climate is beyond the scope of the present study. Still, for completeness, we performed a comparison of the OA optical depth at 550 nm (Fig. 12). It has to be noted that this is not always straightforward, or even possible: models that include aerosol microphysics or internally mixed aerosols cannot always separate the aerosol optical depth (AOD) of the organic component of the aerosol alone, and subtracting simulations with and without OA does not give the right

answer, due to non-linearities in the aerosol microphysics calculations. Such distinction is prohibited by the multi-component aerosol mixtures and water uptake that are taken into account, as well as the non-linear response of the aerosol-radiation interactions caused by such mixtures (e.g. Bond and Bergstrom, 2006). The models that use M7 microphysics (ECHAM5-HAM2, ECHAM5-HAMMOZ and TM5) and thus consider internally mixed aerosols, for diagnostic purposes calculate an OA AOD assuming external mixing in each aerosol mode, although this is not very accurate for estimating the OA contribution to the total AOD; their results are presented in Fig. 12, but should be interpreted with caution. For models that can calculate the organic AOD and have submitted results for both quantities, the organic AOD presents very similar behavior to the OA load, since it is a strong function of the OA column burden, given that most models use very similar optical properties for OA and water uptake parameterizations. Excluding CAM4-Oslo, which calculates a global mean organic AOD of 0.06 due to the computed very high OA load, the other models have organic AOD spanning almost an order of magnitude, from 0.004 to 0.023, with a median value of 0.014. This is 8 % of the total AOD calculated by the same models.

#### 4.1.7 Comparison with AeroCom phase I

Many global models have evolved significantly since the AeroCom phase I intercomparison studies. During phase I, there were two major experiments. The first one, AeroCom A (ExpA), was designed in a very similar way to the AeroCom phase II model simulations described here (Schulz et al., 2009): a simulation year was selected (2000, compared to 2006 for phase II) and the modelers were asked to perform model simulations for that specific year. For the second, AeroCom B (ExpB), all models used the same emission inventories, in order to eliminate the impact the different emission input fields have on the results. The outcomes of these studies have been summarized by Textor et al. (2006) for ExpA and Textor et al. (2007) for ExpB. These two studies focused on the total aerosol budget, but the individual aerosol components were also studied. Sixteen models participated in ExpA and twelve in ExpB, most of which are

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earlier versions of the models that participated in the present intercomparison. During AeroCom phase I, no models simulated semi-volatile SOA; all SOA was considered as to be emitted directly as non-volatile primary particles. As mentioned earlier, the intensity of this source was calculated from terpenes emissions, using a constant yield (15 %) for all models. These non-volatile OA pseudo-emissions were treated the same way as POA. Many models still use this approach (Sect. 2.2). The direct consequence of this assumption is that any uncertainties resulting from the OA sources in ExpA are only related to the POA emissions, since the SOA sources were identical across models.

The total OA sources during expA were very similar to the total sources from the phase II experiments (median  $97 \text{ Tga}^{-1}$  both in expA and here), while expB had much lower total OA sources,  $67 \text{ Tga}^{-1}$  (Fig. 13). All of these sources include SOA, either as pseudo-emissions (phase I) or from a variety of parameterizations (phase II). Note that in expB, although all models were supposed to have the same OA emissions, two deviated from this rule. The models from phase II present a much higher variability in their total OA sources, which is primarily attributed to the SOA chemical production variability that was not present in expA. The POA emissions variability from phase II is roughly the same as that of the OA variability from expA, which indicates that the significant uncertainties in the POA emissions in global models since AeroCom phase I have not been reduced. However, some models have very high POA emissions, due to the recently developed parameterizations of mPOA sources in global models. These highly uncertain sources were absent in AeroCom phase I.

One notable difference is that the dry deposition flux range of OA has been greatly increased in AeroCom phase II since both expA and expB AeroCom phase I experiments, by a factor of 2 or more (Fig. 13). As explained before (Sect. 4.1.4), the M7 and TOMAS aerosol microphysics parameterizations recently adopted by some models are a probable explanation for the range broadening. Comparisons of phase I models results for expA and expB strengthens this conclusion, since the model with the lowest OA dry deposition flux of expA (MPI\_HAM;  $5 \text{ Tga}^{-1}$ ) and that of expB (TM5;



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1.7 Tg a<sup>-1</sup>) both use the aerosol microphysics module (M7). This scheme appears to be responsible for the lowest dry deposition fluxes calculated by the models that participate in the present intercomparison: the updated versions of these two phase I models, ECHAM5-HAM2, ECHAM5-HAMMOZ and TM5, participate in the phase II experiment and simulated the lowest dry deposition fluxes among all phase II models, together with the two GISS-CMU-TOMAS models that did not participate in phase I. Although no comparison with measurements has been made in the present study to evaluate the modeled dry deposition fluxes, and a strong wet deposition flux in these models cannot be excluded as an explanation of the low dry deposition, it is evident that the models with either M7 or TOMAS aerosol microphysics are defining the lowest bounds of the OA dry deposition across all AeroCom experiments (Fig. 7 and Fig. 13). Whether these models are biased low or are closer to reality than the other models should be explored in a dedicated study that will evaluate the dry deposition schemes of different models and compare the calculated fluxes with field measurements.

Wet removal, which together with aerosol sources is a major driver of the calculated aerosol lifetime and load, presents a much higher variability in the phase II models (Fig. 13). This is largely due to the consideration of SOA formation, which is responsible for the large variability in OA sources and burden in the models, as well as to differences in the assumptions on SOA solubility and aging. The high wet removal variability is also reflected in the total OA load and lifetime, where SOA presents a very high variability between models, especially in the case of SOA lifetime. This slightly increases the calculated variability of the total OA by the phase II models compared to phase I. This change is not so pronounced in the OA burden, due to the relatively low contribution of SOA to the OA load calculated by the models. This might change in the future, though, since SOA is believed to be significantly underestimated in global models (Spracklen et al., 2011), as also supported by observations that indicate large amounts of processed OA in the atmosphere (Jimenez et al., 2009).

## 4.2 Median model annual mean

### 4.2.1 Surface distribution

The composite annual mean OC and OA surface air concentrations, defined as the median of the regridded model fields to a  $5^\circ \times 5^\circ$  horizontal resolution, exceed  $0.5 \mu\text{g m}^{-3}$  across most continental regions, with maximum concentrations primarily over biomass burning regions and secondarily over industrialized areas (Fig. 14). The model diversity, defined as the ratio of the standard deviation of all models over their corresponding mean value calculated on the same grid, is smallest over and downwind continental regions, with ratios below 1 over most continental areas, and above 1 over the remote oceans (Fig. 14).

Diversity that exceeds 2 is evident over most of the oceanic regions south of  $30^\circ \text{S}$  and Antarctica, which is a result of the marine OA sources being present in only a few models. Ratios approaching 2 are also found over the northern Pacific and Atlantic oceans, and are also related with the marine OA sources. However, these local maxima are not as pronounced as in the Southern Hemisphere, due to (a) the much stronger seasonality in the Northern Hemisphere, and (b) the stronger influence of continental aerosol sources.

Over and close to the continents, the model diversity is low, except in three areas which present striking differences. Two are located over biomass burning regions, Indonesia and the Pacific borders of the USA and Canada, where the different emissions used by the models produce a large local diversity in concentrations. The third case is off the Pacific coast of Mexico; although this might also be related with biomass burning, the exact reason for the high model diversity is not clear, since this is not over an aerosol source area. Marine sources or different precipitation patterns in the models can also be part of the explanation; however, there are very few measurements (Shank et al., 2012) over that region, which hinders a definite conclusion.

Overall, it appears that the model diversity is low over and downwind of continental source regions, since the primary sources of aerosols are constrained by the availability

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with very slow removal. The modeled vertical distribution of OA presents a diversity that spans over an order of magnitude.

The model diversity, defined as the standard deviation of all models over their mean at a common grid, is relatively low in the lower troposphere (below 600 hPa) between 30° S and 60° N, but very high over the poles and near the tropopause (Fig. 16). A similar pattern was found for BC, sulfate aerosol and particles larger than 100 μm in dry diameter in another AeroCom phase II intercomparison study that focused on aerosol microphysics (Mann et al., 2013). This points out three important features: (a) the areas directly affected by strong primary and secondary sources around the tropics and northern mid-latitudes do not present a large diversity, due to the fairly similar emission inventories in the different models; (b) the primary marine sources of OA however are both highly uncertain and not present in many models, resulting in the high model diversity close to the surface over the Southern Ocean; and (c) the processes that involve low temperatures (which favor condensation of semi-volatile compounds) are not well constrained either, and they are also absent in many models, leading to very high model diversity over the poles and above 200 hPa. The vertical distribution of OA is thus very poorly understood, much less than its surface concentration, and deserves a dedicated study with thorough analysis.

### 4.3 Comparison with measurements

Many model-measurement comparisons can be performed with the extensive dataset used here. The focus of the comparisons in the present study is to identify model strengths and weaknesses, and try to explain where and why the models are failing to simulate the measured concentrations. This will provide insight to directions for future model improvements. In parallel, we are also interested in understanding where and why the models successfully reproduce the observations, and focus on these areas in order to understand the role of the different model complexities on simulations with comparable skill. It is not within the scope of this work to identify which model is the “winner” in simulating OA concentrations, especially since one model is unlikely to

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outperform the others on all metrics, but to provide information on the robustness of the model results. The present study focuses on the surface OC and OA concentrations. The sources and amount of OA in the upper layers of the atmosphere are not explicitly studied here, although accounted for in the OA budget terms discussed above. The detailed analysis of the vertical distribution of OA will be the topic of a future study.

Due to the very inhomogeneous spatial variability of measurements (Fig. 1 and Supplement) only a general global model performance benchmark is performed here. Most data have been collected in the USA, followed by Europe and China. The rest of the world, including some very important regions with regard to OA, are severely under-represented, or not represented at all. Such regions include all tropical forest areas (the Amazon basin, Africa and Southeast Asia) and the boreal forests of Canada and Russia. Long-term measurements in these areas are extremely scarce, with the only notable exception being Alta Floresta in the Amazon, where OC measurements for more than ten years are available.

### 4.3.1 Model skill

One of the major challenges when comparing global models with observations is whether the measurement locations are representative of the regional levels of the measured quantity in question. For most urban measurements this is not the case, since the aerosol concentrations at urban centers are usually much higher than the regional background concentrations. Even a model with a very high horizontal resolution for a global model (like SPRINTARS) is not expected to capture the measurements at urban locations, since its grid cells are of the order of  $100\text{ km} \times 100\text{ km}$ , which is still too coarse to accurately resolve urban pollution. Many of the “urban downwind” AMS data are also expected to fall into this category; thus we included them in the “urban” category.

For all stations, there are several instances where more than one measurement location is present in a given grid box for a certain model. When this is the case, we use the

arithmetic mean of the measurements for that specific grid box, in order to compare the single aerosol concentration the model is providing with a single measurement value.

When discussing the model ensemble results we use the median of all models, while we also analyze the slope and mean normalized bias (MNB) of the models against measurements. The perfect comparison should have a slope = 1, MNB = 0 and correlation  $r = 1$ . The normalized bias (NB) at a given grid box is calculated as follows:

$$NB_i = \frac{C_{\text{model},i} - C_{\text{meas},i}}{C_{\text{meas},i}} \quad (4)$$

where  $C_{\text{model},i}$  is the modeled concentration in grid box  $i$ , and  $C_{\text{meas},i}$  is the measured concentration in the same grid box. If more than one station exists in the same grid box,  $C_{\text{meas},i}$  is the arithmetic mean of the individual stations. The model's MNB is derived as the arithmetic mean of all  $NB_i$  values.

## Urban locations

The models perform poorly at urban locations, as expected. Most models strongly underestimate the measurements, having a median MNB of  $-0.64$  (mean  $-0.62$ , range  $-0.04$  to  $-0.86$ ) for OC (Fig. 17) and  $-0.51$  (mean  $-0.48$ , range  $-0.1$  to  $-0.85$ ) for OA (Fig. 18). CAM5-MAM3 appears in both OC and OA as an outlier, with a slightly negative MNB for OC and  $+0.24$  for OA. As mentioned earlier, CAM5-MAM3 has an enhancement factor of 1.5 for the SOA formation, which might be part of the reason for the generally higher OA concentrations, which result in increased MNB, compared to the other models. The slope of the regression line between models and measurements, calculated by the least absolute deviation (LAD) technique (Branham, 1982), which gives less weight to outliers compared to the regular least-squares regression, varies substantially between models. It ranges from  $-0.07$  to  $2.49$  for OC (median  $0.61$ , mean  $0.74$ ) and  $0.14$  to  $4.92$  for OA (median  $1.21$ , mean  $1.35$ ). CAM5-MAM3 appears again as an outlier with very high slopes in both OC and OA; TM4-ECPL-F has a slightly negative slope for OC, and GISS-CMU-TOMAS a slope of  $0.14$  for OA. The same slope

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calculated with the least-squares regression technique, which gives a more stable numerical solution but is more vulnerable to major outliers, ranges from  $-0.16$  to  $3.57$  for OC (median  $0.85$ , mean  $1.07$ ) and  $0.4$  to  $4.71$  for OA (median  $1.57$ , mean  $1.67$ ), without excluding any model; all the slopes calculated by the Pearson least-square correlation, including the remote and marine ones discussed below, are presented in the Supplement. Interestingly, the correlation of model results against measurements is slightly higher for the OA data; a median value of  $0.54$  is calculated for OA (mean  $0.52$ , range  $0.11$  to  $0.77$ ), compared to  $0.47$  for OC (mean  $0.43$ , range  $-0.09$  to  $0.70$ ). Note though that the locations and temporal resolution of OC and OA measurements greatly differ, making a conclusive comparison between them impossible. In addition, these results are not representative for the overall performance of the models on the global scale; they only represent the models' ability to capture the available measurements, which are very inhomogeneously distributed in space and time (Fig. 1).

### Remote locations

The models show a completely different behavior when compared with measurements of OC (Fig. 19) and OA (Fig. 20) at remote locations. Compared with the models' performance at urban stations, more models have more negative than positive MNB in the case of OC at remote locations, with the range spanning from  $-0.61$  to  $1.29$  (median  $-0.15$ , mean  $-0.02$ ), while most models have a positive MNB in the case of OA, with a range from  $-0.38$  to  $2.17$  (median  $0.51$ , mean  $0.70$ ). It has to be noted, though, that the locations and times of OC and OA measurements are not the same, which means the model performance for OC and OA data are not directly comparable, due to the different spatial and temporal coverage of the stations. Only four models present relatively high positive MNB values when compared with the OC data: CAM5-MAM3 ( $1.3$ ), EMAC ( $0.9$ ), ECHAM5-SALSA ( $0.7$ ) and ECMWF-GEMS ( $0.6$ ). CAM5-MAM3 has the third highest SOA source of all models (Fig. 3), but none of the other three models with strong positive MNB has exceptionally high POA (Fig. 2) or SOA sources and sinks. All of EMAC, ECHAM5-SALSA and ECMWF-GEMS present a very strong maximum in the

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OC concentrations at the western border of the USA with Canada; monthly mean concentrations exceeding  $200 \mu\text{gC m}^{-3}$  in EMAC (Fig. S3 in the Supplement) might be the reason for the positive MNB. Also note that EMAC emits all biomass burning aerosols at surface, while most other models distribute them to a number of layers above the surface, typically within the boundary layer. The other models that present a positive correlation are all linked with either strong POA sources (CAM4-Oslo; Fig. 2) or strong SOA sources (HadGEM2-ES and IMPACT; Fig. 3), but that is not the case for IMAGES, which has the highest SOA source but presents a MNB of only +0.1, and TM4-ECPL-FNP, which has the 7th strongest SOA source from the models that submitted their SOA chemical production, but presents the second strongest negative MNB of all the models.

Many models have a lower correlation with remote OC and OA measurements than with urban OC and OA. Although this might appear unexpected, a possible explanation might be that urban pollution probably adds a large offset in the comparison, which does not affect correlation; in remote sites on the other hand, long-range transport adds one additional level of uncertainty in the model calculations, which can result in lower correlation of the model results with measurements. The correlation coefficient against OC remote measurements rarely exceeds 0.5, with the correlation for about half of the models lying below 0.4 (median 0.39, mean 0.40, range 0.11–0.67), while when compared against the remote OA measurements the correlations are slightly lower, with a median and mean value of 0.37 (range 0.07–0.55). It is possible that either a remote source is missing or treated in a too simplistic way, or that the transport and lifetime (which largely depend on solubility, representation of precipitation from clouds, and poorly represented ageing processes) of organic aerosols in the regional and remote atmosphere are not properly calculated in models, or that the seasonality of sources is not accurately represented, or a combination of any of these reasons. High (negative) MNB and high correlation (–0.61 and 0.47, respectively for OC) for the urban stations support the missing sources hypothesis. Low (negative) MNB and low correlation (–0.15 and 0.4, respectively for OC) for remote stations support the



conclusion that the knowledge about the processes, on top of the sources, contributes to the OA modeling uncertainty at remote stations.

## Marine locations

Since there are only two AMS OA marine stations categorized as remote in the global AMS database, only the OC model results have been compared against the marine OC measurements (Fig. 21). Very few models include a marine organic aerosol source: CAM4-Oslo, the two GISS-modelE models, IMPACT and the two TM4-ECPL models. Even with or without the primary marine source, rather poor statistics are calculated for most of the models. Most models have a negative MNB (median  $-0.30$ , mean  $-0.15$ , range  $-0.64$  to  $+0.90$ ), with a few exceptions: The two GISS-modelE models, with MNB  $\sim 0.85$ – $0.90$ , have a strong mPOA source, the strongest of all models that participate in this intercomparison; HadGEM2-ES, whose strong SOA source which is based on a climatology might be the reason for the high MNB; IMPACT and IMAGES, which have a simplified multiphase chemistry source that might be responsible for the increased remote marine OA; and EMAC, which is among the models with the highest POA sources (Fig. 2). All models present a very low slope, typically less than 0.4, but very frequently less than 0.2 (median 0.15, mean 0.21, range  $-0.01$  to  $+0.78$ ).

Although the GISS-modelE models appear to have the best slope, the correlation with measurements appears to be worse than other models. The reason might be the variability of the source of marine organics that may not be captured by the models: both GISS-modelE models that present the lowest correlation against marine OC measurements calculate the marine OC sources as a function of chlorophyll; this might not be the optimal parameterization of the marine POA source. The IMPACT and the TM4-ECPL models, which include similar mPOA sources, do not produce such low correlations. These models include aqueous production of OA, which acts as an additional source in the remote atmosphere. IMAGES, which also has an aqueous OA source, produces a rather high correlation with the marine OC measurements, a positive MNB and a slope close to 1. Although more marine observations are needed to verify this

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hypothesis, it appears that a multiphase source does improve the model comparison with remote marine measurements, as also discussed by Myriokefalitakis et al. (2011). One cannot dismiss though that an increase in SOA sources via gas-phase production is not the missing source in these locations, which might be able to improve the correlation there. It has to be reminded that IMAGES and IMPACT have a different source parameterization compared with that in TM4-ECPL-F/FNP, which results in a stronger aqueous OA source that degrades correlation, but not MNB, compared to the same model-measurements comparison when excluding the multiphase aerosol contribution (not shown). In TM4-ECPL-F/FNP the multiphase OA source is weaker ( $13\text{--}29\text{ Tg a}^{-1}$ ) than in the other two models and no statistically significant improvement is seen in the model's performance at surface when accounting for this source. Additional models able to simulate aqueous phase OA formation and comparison with targeted observations are needed to consolidate the importance of this process on the OA budget. The primary marine source also improves the comparison over the oceans (Fig. 36), but further work is needed to constrain this source. Overall, the median and mean correlations are very close (0.25 and 0.24, respectively) and the correlation range is between  $-0.03$  and  $+0.41$ .

### Importance of model complexity

In the comparisons of model results with urban station data, the correlations against OA observations were higher than those against OC. Urban aerosols are mostly fresh, compared to the more aged ones at remote locations. All models simulate OA, and then the OA/OC ratio is used to convert from OA to OC, in order to compare with OC data. Emission inventories however are frequently in units of carbon, not organic matter, adding an additional conversion, thus uncertainty, in the models. Using the same OA/OC ratio to convert emissions and then the simulated concentrations implies that the OA/OC ratio has not changed with atmospheric processing. This is clearly a weak assumption, since OA/OC is different at emission time and after atmospheric processing. Since all models have some aging parameterization in their calculations, this





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formation potential (Carlton et al., 2010), due to both vicinity of sources and abundance in solar radiation, especially during summer. One explanation might be that wintertime emissions are much stronger there than in other areas in USA, enhancing the wintertime OA levels and masking the summertime SOA contribution. Additionally, enhanced anthropogenic aerosols like sulfate might increase aerosol water content substantially in the southeast USA (Dick et al., 2000), counterbalancing the photochemical production of SOA, an effect currently absent from all models participating in this study that do not take into account aqueous SOA formation. All these hypotheses need to be investigated in the future by both field and modeling studies in more detail.

The absence of seasonality measured at several urban locations might be due to a combination of stronger anthropogenic primary sources and reduced dispersion during winter and enhanced SOA formation during summer, as well as missing processes from the models, flattening the seasonal cycle. Whether SOA dominates over anthropogenic POA, appears to be the decisive factor for the seasonal pattern. However, this is only a hypothesis that is driven by the model results, that needs to be explored in the field. The fact that the models appear to be (a) missing an urban source, and (b) underestimating the pollution levels in cities, is also supported by the comparison of the model results with remote stations close to the urban ones presented in Fig. 23, where the models are able to capture both the magnitude and seasonality of measurements much better (Fig. 24). An important thing to note is that the measurements are roughly a factor of 5 lower in these remote stations compared to their urban counterparts, except the case of Ohio, where the remote station appears to be influenced by urban pollution: its levels are only half that of the Ohio urban station, while its seasonality resembles the seasonality present in several urban stations discussed earlier.

### 4.3.3 Chemical composition

Unfortunately, it is impractical to present and analyze every individual station used in the present study. Instead, a number of stations have been selected, based on a number of criteria: they must be far enough away from each other geographically, have



emissions. SPRINTARS, which has a very fine grid, simulates virtually all OC as trSOC at the remote station, while tPOC dominates during winter at the urban station, presumably due to high anthropogenic emissions that are resolved on the fine grid of that model. The same can be seen with the coarser grid CCSM4-Chem model, but to a lesser extent.

The OA concentrations at the remote station LinAn, China (Fig. 27), are underestimated by the models. All models calculate a high contribution of tPOC to the total OC levels, a clear indication of very strong regional anthropogenic pollution. Not all models agree on the trSOC contribution to the total OC, with about half of the models that resolve the OC chemical composition calculating a contribution that approaches 50% during summer, and the other half calculating a much smaller contribution. IMPACT and IMAGES suggest that a significant amount of OC comes from multiphase chemistry, while the two TM4-ECPL models do not. Recall that TM4-ECPL has a more detailed aqueous chemistry scheme than the other two models, which probably explains the difference in the results.

The remote station Finokalia, Greece, has both OC and OA (AMS) measurements. The OC data (Fig. 28) do not exhibit any seasonality, contrary to all models that underestimate the wintertime measurements by simulating a wintertime minimum and a summertime maximum. The measured OA concentrations (Fig. 29), although from only four out of twelve months, appear to be higher during summer, a feature that is captured both in shape and magnitude by a small number of models. The air masses that arrive at Finokalia are aged, since there are no significant sources upwind for at least 300 km (Mihalopoulos et al., 1997). This is also evident from the GISS-CMU-VBS results, where virtually all POA is calculated to be ntrSOA (aged primary), which means that photochemistry, which is expected to be higher during summer, has already contributed to the aging of the air masses arriving at the station. If this is indeed the case, it means that the OA/OC ratio during summer is higher than the winter value, a fact that is implied by the measurements. Note however that it is not trivial to compare the PM<sub>1.8</sub> OC data with the PM<sub>1</sub> AMS data and calculate an OA/OC ratio (Koulouri et al.,

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is not exactly the same with that of the data, or the simulations should use emission inventories and meteorology for the specific year that the measurements have been performed. There is agreement among the models that the September maximum is due to POA, while SOA is fairly constant year-round; aqueous chemistry also contributes a small amount to the total OA, which is enhanced during the wet season. GISS-CMU-VBS calculates that most of the POA is already aged, although during the biomass burning season there is a non-negligible amount that is still fresh.

In Alaska, USA (Fig. 31), many models simulate a summer maximum, in agreement with the measurements, which is due to biomass burning sources. TM4-ECPL-FNP calculates a very strong contribution from primary biological particles to the total OC, resulting in a slight overestimation of measurements throughout the year. The four models that have provided mPOA concentrations (two GISS-modelE and two TM4-ECPL models) suggest that marine organics are present in significant quantities. Multiphase chemistry is also calculated to contribute during the summer months. ECMWF-GEMS shows a very wide peak during summer, in contrast with the other models, resulting in higher concentrations than the measured ones for half of the year. This might be caused by the averaging of biomass burning emissions over six fire seasons that this model uses which exhibit a large interannual variability and broadens the biomass burning contribution over many months. The remaining models generally underestimate the measurements, although they capture the observed seasonality rather well; more than half of the models have a correlation coefficient against measurements greater than 0.8. An interesting pattern is that of the two GISS-modelE models, which simulate significantly contribution of trSOA to the total OC, especially during winter. These two models are the only models that include semi-volatile SOA and use the Lathière et al. (2005) VOC emissions, in which strong summer emissions in south Alaska are present (Tsigaridis et al., 2005). It is very likely that the distribution of VOC sources (which differs from that of the other models), when combined with the low temperatures in Alaska during winter (which favors partitioning to the aerosol phase), leads to the enhanced trSOA formation.

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A region where biomass burning is also important is the Amazon basin. Alta Floresta, Brazil (Fig. 32), is a site with very long-term measurements and a clear biomass burning peak during the dry season, which the models reproduce quite successfully, with the exception of GMI that peaks during May. Two other models, BCC and IMPACT, do not capture the dry season peak, which shows potential problems with their biomass burning emissions in the region. Care has to be taken regarding the year simulated and compared with measurements. Through the years Alta Floresta has experienced a significant reduction in biomass burning intensity in the region, while the data shown in Fig. 32 are the climatology of all years of measurements. During the biomass burning period, GISS-CMU-VBS simulates equal amounts of fresh and aged POA, reflecting the close vicinity of POA sources. Multiphase chemistry appears not to play a significant role here, in contrast with the Manaus station, Brazil (Fig. 33), where IMPACT and IMAGES calculate significant amounts of ntrSOA; the location of the inter-tropical convergence zone (ITCZ) and the potentially very different availability of cloud and aerosol water at these stations appears to be the reason behind this important difference. At the same station, many models also calculate large amounts of trSOA during the dry season.

In Melpitz, Germany (Fig. 34), most models agree that there is no clear seasonality, other than a winter and a weak summer peak. The measurements do not present a clear seasonality either, although the spread of the concentrations is larger than that calculated by most models, and there are gaps in the data that make the comparison more difficult. In any case, emissions during winter appear to be significant, although fossil fuel emissions from different models are not the same (e.g. IMPACT vs. TM4-ECPL-F, where the models calculate very similar distributions with different absolute amounts due to different emission inventories). GISS-CMU-VBS has almost no POA, either fresh or aged; the calculated OA is mostly trSOA, which is present during summer in the models. Multiphase chemistry also forms a significant fraction of the total OC. TM4-ECPL-FNP calculates a high contribution from primary biological particles, but its seasonality does not agree with that of the measurements.

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The Mace Head station on the west coast of Ireland (Fig. 35) receives mostly oceanic air masses. Among the models that include an mPOA source, GISS-modelE-G/I and IMPACT exceed the measured OA, except a high value in May, which might be either an outlier or something related to a source that the models are unable to capture.

Among the other models that include mPOA, CAM4-Oslo has a mPOA source (added to tPOA) that follows the offline sea salt fluxes from AeroCom phase I (not the ones calculated online in the model) and TM4-ECPL-F/FNP have a low global flux; these models are closer to the measurements than the rest of the models that include mPOA. The models that do not include any mPOA parameterization calculate much lower OA concentrations. Products from multiphase chemistry are also present, while primary anthropogenic and biomass burning aerosols are virtually absent. Caution has to be taken because Mace Head is at the coast; many models that have their gridbox edge at 10° W use a continental box for representing Mace Head (longitude 9.9° W), which might not be representative for its predominantly marine environment.

As expected, only the models that include a marine source of mPOA are able to capture the OA concentrations at remote marine stations. This is particularly true for the two versions of GISS-modelE (Tsigaridis et al., 2013), which have the strongest source of mPOA of all models that participate in the intercomparison. Although most of the remote marine data we have are single measurements and their seasonality cannot be studied, it is important to note that their chemical composition is dominated by mPOA. Fortunately, there is one station with five years of data in a remote marine environment: Amsterdam Island, in the southern Indian Ocean (Fig. 36). As at Mace Head, the models that include mPOA sources are closer to the measurements, while the rest of the models simulate extremely low OC concentrations. There are three notable exceptions: one is the two GISS-modelE models, which strongly overestimate the measurements, as discussed by Tsigaridis et al. (2013). Second, the ECMWF-GEMS model, which although it does not have a marine OA source, simulates higher-than-expected OC concentrations there. Third, the IMAGES model which is able to capture some of the measured data due to high ntrSOA amounts calculated there.

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Multiphase chemistry appears to significantly contribute to the OC mass calculated at Amsterdam Island in other models as well, which reproduce the long-range transport of biomass burning aerosol from southern Africa during August–October (Fig. 36), which is also seen in the observations (Sciare et al., 2009). The meteorology used appears to significantly affect ntrSOA production in the two TM4-ECPL models, due to differences in the availability of water in aerosols and the distribution of clouds between the years simulated: 2005 for TM4-ECPL-F and 2006 for TM4-ECPL-FNP.

Okinawa, Japan (Fig. 37), is a remote marine site that is influenced by long-range transport of pollution aerosol from East Asia. AMS measurements at four different months suggest that  $2\text{--}3\ \mu\text{g m}^{-3}$  are present over Okinawa, an amount that only models that include either multiphase chemistry of organics, or mPOA sources, or both, can reproduce. IMAGES appears to agree best with measurements, and the two TM4-ECPL models, although having both ntrSOA and mPOA, are rather low. All models agree that there is a minimum in OA concentrations during summer, but unfortunately no data are available to validate this trend. Small but significant OA amounts come from multiphase chemistry. The models also calculate an important contribution of very aged POA.

## 5 Conclusions and perspectives

This study shows that the diversity of the global OA modeling results has increased since AeroCom phase I, mainly due to both the increased complexity, as well as the increased diversity of the OA parameterizations and sources in the models, which is evident in the different chemical compositions simulated by the models at the various stations analyzed here. Increased number of tracers, however, does not necessarily mean increased complexity of OA parameterizations; models with aerosol microphysics must have a large number of organic aerosol tracers, even when they may simulate OA production in a very simplistic way. At present, about half of the thirty-one participating models include explicit treatment of semi-volatile SOA formation in the atmosphere.

Four models also account for multiphase chemistry and 6 models for natural sources of POA, in particular the marine source, with one model including the emissions of primary biological particles.

The POA sources in the thirty-one AeroCom models range between 34 and 144  $\text{Tga}^{-1}$  with a median value of 56  $\text{Tga}^{-1}$ . Secondary OA sources show larger model diversity spanning from 12.7 to 121  $\text{Tga}^{-1}$ , with a median value for 12 out of 14 of the models that parameterize SOA chemical production of 51  $\text{Tga}^{-1}$  (mean 59  $\text{Tga}^{-1}$  with standard deviation of 38  $\text{Tga}^{-1}$ ). In the four models that account for multiphase chemistry of organics, its contribution to SOA levels is calculated to be significant (up to 50 % of total SOA formation), at least regionally.

The wet removal of OA is simulated to range between 28–209  $\text{Tga}^{-1}$  for 26 of the models, with median 70  $\text{Tga}^{-1}$ , which is on average 85 % of total OA deposition. The high wet removal variability, together with the large variability of OA sources, are attributed primarily to the diversity of SOA formation, which affects the total OA load and lifetime. The SOA very high variability between models is especially evident in SOA lifetime (2.4 days to 15 days). This slightly increases the calculated variability of the total OA by the phase II models compared to phase I, where the SOA model diversity was essentially zero.

The treatment of aerosol microphysics in the models appears to have a significant impact on the calculated OA load and dry deposition. The range in dry deposition flux for OA (2–36  $\text{Tga}^{-1}$  in the present study) has been greatly increased since both AeroCom expA and expB, by a factor of 2 or more, while the M7 and TOMAS aerosol microphysics parameterizations, used by three and two models, respectively, simulate very low dry deposition rates when compared to the other models and thus contribute a lot to this change in diversity.

The annual median atmospheric burden of OA is calculated to be 1.4 Tg by the AeroCom phase II models, with values that vary mostly between 0.6 Tg and 1.8 Tg. Four models simulate loadings higher than 2.0 Tg, up to 3.8 Tg. The models calculate very similar OA load seasonality, which maximizes during the Northern Hemisphere

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summer, when both primary (biomass burning) and secondary (chemical production) OA are high and minimize during Northern Hemisphere spring. A median OA lifetime of about 5.4 days (ranging from 3.8 to 9.6 days) is derived from the present study. The median POA lifetime of 4.8 days (ranging from 2.7 to 7.6 days) from this study is slightly shorter than the median SOA lifetime of 6.1 days (range from 2.4 to 14.8 days).

For all models that reported both OA and  $\text{SO}_4^{2-}$  loads, the OA load is calculated to be lower than that of  $\text{SO}_4^{2-}$ , with a median value of the OA/ $\text{SO}_4^{2-}$  mass load ratio of 0.77. Simulated values of this ratio span from 0.25 to 2.0, indicating that there is a low level of understanding of the relative importance of OA and  $\text{SO}_4^{2-}$  aerosol components between models, although modeling studies indicate that this ratio will increase in the future due to sulfur emission controls. This ratio is also affected by multiphase chemistry of organics and deserves further attention in the future.

A significant (up to 45 %) but highly variable contribution of multiphase chemistry to global SOA formation is calculated by models that account for this process. The comparison with observations indicates that the lower estimate of this source might be closer to reality, but this has to be revisited when more models will include multiphase SOA formation. In addition, a gas-phase source of SOA, either new or an enhanced pre-existing one, has the potential to improve the comparison with measurements in the same way multiphase chemistry does; OA chemical composition measurements can help identify which one of the two, or both, is the case. Further investigation of the importance of multiphase chemistry on the global scale and evaluation against targeted observations and field campaigns is needed.

The models show a large diversity (about two orders of magnitude) in the free troposphere, pointing to uncertainties in the temperature-dependent partitioning of SOA, uncertainties in free tropospheric sources, and impact of meteorology and transport. A systematic comparison of model results with the limited available free tropospheric observations can give important insights in the large model differences in the middle and upper troposphere. This will be the topic of future study.

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Despite the increasing diversity between models since AeroCom phase I experiments, the models are now able to simulate the secondary nature of OA observed in the atmosphere as a result of SOA formation and POA aging, although the absolute amount of OA present in the atmosphere remains underestimated. The median MNB of all models against urban measurements at surface is calculated to be  $-0.62$  for OC and  $-0.51$  for OA and with correlations  $0.47$  and  $0.54$ , respectively, while for remote surface measurements the MNB is  $-0.15$  for OC and  $+0.51$  for OA with correlations  $0.39$  and  $0.37$ , respectively. Due to the different locations, number of stations and measurement times where OA and OC data are available, a direct comparison between the OC and OA statistics results is not straightforward and should be avoided.

Comparison of model results with OA and OC, where available, shows that the models capture the submicron OA mass better than the  $PM_{2.5}$  OC mass near the surface. Although this indicates a possible overestimate of the OA/OC ratio by the models, this is not necessarily the case, since virtually all OC and OA measurements were taken at different locations and different times. Most models use a constant value of  $1.4$  and only four models in this study calculate it prognostically. The limited number of observations that can be used to derive the OA/OC ratio indicate dependence on sources, atmospheric conditions and season; this will be revisited in a future study.

The flat seasonality measured at several urban locations is not reproduced by the models. The comparisons indicate a missing or underestimated source of OA in the models, either anthropogenic primary (for instance domestic wood burning), or secondary, primarily during winter. Improvements in the seasonality and strength of the anthropogenic POA sources in models can reduce the differences between model results and observations.

Available OC and OA observations and thus model evaluations are concentrated in USA and Europe. Additional long-term observations from tropical, boreal, Southern Hemisphere and remote marine regions but also from the free troposphere are needed to complement the global OA observational database.

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Natural POA sources are important components of the OA global budget, however, among the thirty-one models participating in this intercomparison, only six account for the marine source of OA and one for the primary biogenic particles. Comparison of model results to observations over remote marine locations can provide constraints on our understanding of the marine POA source. The statistics on model performance calculated here are not able to quantify the importance or the understanding of this source because seasonal data from remote marine locations are limited. The magnitude of the marine source and the properties of marine OA remain highly uncertain and are an active area of research.

Primary biogenic particles can also be significant contributors to OA, particularly over land, but are taken into account only in one model. While the parameterization of the primary biogenic source of OA is extremely uncertain, model comparison with measurements is improved when accounting for this source in that model, by reducing the MNB and/or increasing the slope of the correlation towards 1. The correlation of the model results against observations is not changing significantly when including or excluding this source. However, station-by-station comparison indicates a low level of understanding of the spatial and seasonal variability of this natural source, which deserves further investigation and improvement.

Both the model diversity that increased with increasing model complexity over the past decade, as well as the comparison of model results with station data, reveal important gaps in our understanding of OA concentrations, sources and sinks in the atmosphere, and point towards the need for better understanding of sources and chemical aging of OA. Although the increasing complexity did not significantly improve the model performance, model complexity is imposed by the need to provide information for future developments that will help quantify the anthropogenic impact to climate via the aerosol direct and indirect effects. The existence of significant secondary sources of OA that are enhanced by interactions of natural with anthropogenic emissions remains an open question that cannot be answered by a simple OA parameterization. Furthermore the OA impact on climate depends on the OA physical, chemical and optical properties, as



well as the OA distribution in the atmosphere, which is affected by continuous evaporation/condensation processes of semi-volatile organic material and consequent change of hygroscopicity.

In this respect, new information from dedicated field campaigns that either occurred over the past few years or are planned to take place soon, is expected to shed light on the OA formation processes and how these are altered in the presence of anthropogenic pollution. The model development related to OA is expected to accelerate in the near future and must be performed in parallel with extensive model evaluation.

**Supplementary material related to this article is available online at**

**<http://www.atmos-chem-phys-discuss.net/14/6027/2014/acpd-14-6027-2014-supplement.pdf>.**

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**Table 1.** Host model description and year of simulation. If multiple years were simulated, the data used for the present study are those from the year 2006, if available; otherwise, the year closest to 2006 was selected.

Model (AeroCom experiment name)	Simulated year(s)	Horizontal resolution (lat/lon)	Vertical resolution	Meteorology	Model references
BCC (BCC_AGCM2.0.1_CAM.A2.HCA-FIX)	2000 <sup>1</sup>	2.8125 × 2.8125	26 (hybrid sigma) to 2.9 hPa	online	Zhang et al. (2012a)
CAM4-Oslo (CAM4-Oslo-Vcmip5.A2.CTRL)	2006 <sup>2</sup>	1.875 × 2.5	26 (hybrid sigma) to 2.19 hPa	online <sup>2</sup>	Kirkevåg et al. (2013)
CAM5-MAM3 (CAM5.1-MAM3-PNNL.A2.CTRL)	2000 <sup>1</sup>	1.875 × 2.5	30 (hybrid sigma) to 2.3 hPa	online	Liu et al. (2012)
CanAM-PAM (CCCma.A2.CTRL)	2006	3.75 × 3.71	35 (hybrid sigma) to 1 hPa	online	von Salzen et al. (2005); von Salzen (2006)
CCSM4-Chem (CCSM4)	2006	1.9 × 2.5	26 (sigma) to 3.5 hPa	GEOS 5	Lamarque et al. (2012)
ECHAM5-HAM2 (MPIHAM_V2_KZ.A2.CTRL)	2006– 2008	1.875 × 1.875	31 (hybrid sigma) to 10 hPa	ERA	Stier et al. (2005); Kazil et al. (2010); O'Donnell et al. (2011); Zhang et al. (2012b)
ECHAM5-HAMMOZ (ECHAM5-HAMMOZ.A2.HCA-0)	2000– 2005	2.8125 × 2.8125	31 (hybrid sigma) to 10 hPa	ECMWF ERA40 and operational	Stier et al. (2005); Pozzoli et al. (2008); Pozzoli et al. (2011)
ECHAM5-SALSA (SALSA_V1_TB.A2.CTRL)	2006	1.875 × 1.875	31 (hybrid sigma) to 10 hPa	ECMWF opera- tional	Stier et al. (2005); Kokkola et al. (2008); Bergman et al. (2012)
ECMWF-GEMS (n/a)	2003– 2008	1.5 × 1.5 <sup>3</sup>	60 (hybrid sigma) to 0.1 hPa	ECMWF-GEMS operational	Benedetti et al. (2009); Morcrette et al. (2009)
EMAC (ECHAM-MESSy-GMx.A2.CTRL)	2006	2.8125 × 2.8125	19 (hybrid) to 10 hPa	ECMWF reanaly- sis	Jockel et al. (2005); Pringle et al. (2010)
GEOS-Chem (GEOSCHEM-v822.AEROCOM_ A2.CTRL)	2006	2 × 2.5	47 (hybrid sigma) to 0.01 hPa	GMAO version 5 (GEOS-5)	Bey et al. (2001); Park et al. (2003); Henze and Seinfeld (2006); Liao et al. (2007)
GEOS-Chem-APM (GEOS-Chem-APM.A2.CTRL)	2006	2 × 2.5	47 (hybrid sigma) to 0.01 hPa	GEOS-5.2.0	Bey et al. (2001); Park et al. (2003); Yu and Luo (2009); Yu (2011)
GISS-CMU-TOMAS (GISS-TOMAS.A2.CTRL)	2006 <sup>1</sup>	4 × 5	9 (hybrid sigma) to 10 hPa	online	Hansen et al. (1983); Adams and Seinfeld (2002); Lee and Adams (2010); Lee and Adams (2012)
GISS-CMU-VBS (GISS-CMU.A2.CTRL)	2008 <sup>1</sup>	4 × 5	7 (hybrid sigma) to 10 hPa	online	Hansen et al. (1983); Farina et al. (2010); Jathar et al. (2011)
GISS-MATRIX (GISS-MATRIX.A2.CTRL)	2006– 2008 <sup>4</sup>	2 × 2.5	40 (sigma) to 0.1 hPa	NCEP reanalysis and online	Bauer et al. (2008)
GISS-modelE-G (GISS-modelE.A2.CTRL)	2000– 2008 <sup>4</sup>	2 × 2.5	40 (sigma) to 0.1 hPa	NCEP reanalysis and online	Koch et al. (2007); Tsigaridis and Kanakidou (2007); Tsigaridis et al. (2013)
GISS-modelE-I (GISS-modelE.A2.HCA-IPCC)	2000– 2008 <sup>4</sup>	2 × 2.5	40 (sigma) to 0.1 hPa	NCEP reanalysis and online	Koch et al. (2007); Tsigaridis and Kanakidou (2007); Tsigaridis et al. (2013)

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Table 1. Continued.

Model (AeroCom experiment name)	Simulated year(s)	Horizontal resolution (lat/lon)	Vertical resolution	Meteorology	Model references
GISS-TOMAS (n/a)	2006 <sup>5</sup>	2 × 2.5	40 (sigma) to 0.1 hPa	MERRA reanalysis and online	Lee and Adams (2010); Lee et al. (2014)
GLOMAPbin (GLOMAPbin1pt1.A2.CTRL)	2006	2.8125 × 2.8125	31 (hybrid sigma) to 10 hPa	ECMWF opera- tional	Mann et al. (2012)
GLOMAPmode (GLOMAPmodev6R.A2.CTRL)	2006	2.8125 × 2.8125	31 (hybrid sigma) to 10 hPa	ECMWF opera- tional	Mann et al. (2012)
GMI (GMI-v3.A2.CTRL)	2006	2 × 2.5	42 (hybrid sigma) to 0.01 hPa	NASA GEOS4	GMAO Liu et al. (2007); Bian et al. (2009)
GOCART (GOCART-v4.A2.HCA-0)	2000– 2007	2 × 2.5	30 (hybrid sigma) to 0.01 hPa	NASA GEOS4	GMAO Chin et al. (2000); Ginoux et al. (2001); Chin et al. (2002)
HadGEM2-ES (HadGEM2-ES.A2.CTRL)	2006– 2008	1.25 × 1.875	38 (hybrid height) to 39 km	ECMWF opera- tional and online	Bellouin et al. (2011) and references therein
IMAGES (n/a)	2006	2 × 2.5	40 (hybrid) to 44 hPa	ECMWF Interim	ERA- Müller (2009); Stavrou et al. (2009); Ceulemans et al. (2012)
IMPACT (IMPACT-C.A2)	1997	4 × 5	46 (hybrid sigma) to 0.147 hPa	NASA GEOS-STRAT	DAO Lin et al. (2012)
LMDz-INCA (LSCE-vRV.A2.CTRL)	2006	1.875 × 3.75	19 (sigma) to 3 hPa	ECMWF IMF and online	Schulz (2007); Balkanski (2011); Szopa et al. (2013)
OsloCTM2 (OsloCTM2-v2.A2.CTRL)	2006	2.8125 × 2.8125	60 (hybrid sigma) to 2 hPa	ECMWF IFS	Hoyle et al. (2007); Hoyle et al. (2009); Myhre et al. (2009)
SPRINTARS (SPRINTARS-v384.A2.CTRL)	2006	1.125 × 1.125	56 (sigma) to ~ 1 hPa	NCEP reanalysis and online	Takemura et al. (2000); Takemura et al. (2002); Takemura et al. (2005); Takemura et al. (2009)
TM4-ECPL-F (TM4-ECPL-F.A2.CTRL)	2006	2 × 3	34 (hybrid sigma) to 0.1 hPa	ECMWF Interim	ERA- Myriokefalitakis et al. (2008); Myriokefalitakis et al. (2010); Myriokefalitakis et al. (2011)
TM4-ECPL-FNP (TM4-ECPL-FNP.A2.CTRL)	2005	2 × 3	34 (hybrid sigma) to 0.1 hPa	ECMWF Interim	ERA- Myriokefalitakis et al. (2008); Myriokefalitakis et al. (2010); Myriokefalitakis et al. (2011); Kanakidou et al. (2012)
TM5 (TM5-V3.A2.HCA-IPCC)	2000– 2009	2 × 3	34 (hybrid sigma) to 0.5 hPa	ECMWF Interim	ERA- Huijnen et al. (2010); Aan de Brugh et al. (2011); van Noije et al. (2014)

<sup>1</sup> Meteorology calculated by the model's climate.<sup>2</sup> 5 yr mean of model's calculated meteorology, driven by offline CAM4 aerosols and cloud droplet number concentration.<sup>3</sup> The model is run at a TL159 L60 resolution, meaning a reduced physical grid of 1.125 × 1.125. The data extraction was carried out on a 1.5 × 1.5 regular grid.<sup>4</sup> Horizontal winds are nudged to NCEP reanalysis (Kalnay et al., 1996), with the rest of the climate parameters being calculated online.<sup>5</sup> Horizontal winds are nudged to MERRA reanalysis (Rienecker et al., 2011), with the rest of the climate parameters being calculated online.

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Model	Fossil fuel and biofuel	Year	Biomass burning	Year	Other/comments
BCC	Bond et al. (2004)	2000	GFED	2000	
CAM4-Oslo	AeroCom; mPOA based on Spracklen et al. (2008)	2006	AeroCom	2006	Emitted particle sizes (with some adjustments) and fire emission injection heights from Dentener et al. (2006).
CAM5-MAM3	CMIP5	2000	CMIP5	2000	Emitted particle sizes (with some adjustments) and fire emission injection heights from Dentener et al. (2006).
CanAM-PAM	AeroCom	2006	AeroCom	2006	
CCSM4-Chem	POET, REAS over Asia	2006	GFED2	2006	
ECHAM5-HAM2	AeroCom	2006	AeroCom	2006	
ECHAM5-HAMMOZ	AeroCom	2000–2005	GFED2	2000–2005	For 1980–2000 (not studied here), the model is using RETRO emissions.
ECHAM5-SALSA	Dentener et al. (2006)	2000	GFED	2000	
ECMWF-GEMS	Dentener et al. (2006)	2000	GFED2	2003–2008	Biofuel emissions have a prescribed diurnal cycle.
EMAC	Dentener et al. (2006)	2000	GFED	2000	
GEOS-Chem	Bond et al. (2007)	2000	GFED2	2006	
GEOS-Chem-APM	Bond et al. (2007)	2000	GFED2	2006	
GISS-CMU-TOMAS	AeroCom	2006	AeroCom	2006	
GISS-CMU-VBS	Bond et al. (2004) <sup>1</sup>	2000	GFED2	2005	
GISS-MATRIX	CMIP5 RCP4.5	2000	GFED3	2006–2008	
GISS-modelE-G	CMIP5 RCP4.5	2000–2008	GFED3	2000–2008	mPOA calculated online (Tsigaridis et al., 2013)
GISS-modelE-I	CMIP5 RCP4.5	2000–2008	CMIP5 RCP4.5	2000–2008	mPOA calculated online (Tsigaridis et al., 2013)
GISS-TOMAS	CMIP5 RCP4.5	2006	CMIP5 RCP4.5	2006	
GLOMAPbin	AeroCom	2000	GFED2	climatology (Dentener et al., 2006)	
GLOMAPmode	AeroCom	2000	GFED2	climatology (Dentener et al., 2006)	
GMI	CMIP5	2000	CMIP5	2000	
GOCART	AeroCom	2006	AeroCom	2006	Details about emissions from different sources are in Chin et al. (2009).
HadGEM2-ES	CMIP5	2000	CMIP5	2000	
IMAGES	Bond et al. (2004)	2000	GFED2	2000	
IMPACT	Ito and Penner (2005)	2000	Ito and Penner (2005)	2000	mPOA calculated online, based on Gantt et al. (2009a). Fossil fuel adjusted as in Wang et al. (2009)
LMDz-INCA	CMIP5	2000	CMIP5	2000	
OsloCTM2	CMIP5	2000	CMIP5	2000	
SPRINTARS	AeroCom	2006	AeroCom	2006	
TM4-ECPL-F	CIRCE	2006	GFED2	2006	mPOA calculated online (Myriokefalitakis et al., 2010)
TM4-ECPL-FNP	CMIP5	2005	CMIP5	2005	mPOA calculated online (Myriokefalitakis et al., 2010); POA from primary biogenic sources are scaled on leaf area index and dust distribution (Kanakidou et al., 2012).
TM5	CMIP5 RCP4.5	2000–2009	CMIP5 RCP4.5	2000–2009	

<sup>1</sup> North America emissions come from Park et al. (2003).



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Table 4. Organic aerosol representation in the models.

Model	OA types <sup>1</sup>	# of OA tracers	trSOA precursors	trSOA calculations	OA/OC	Comments
BCC	IPOA	12	Monoterpenes	15% yield from terpenes emissions (Dentener et al., 2006), included in IPOA	1.4	
CAM4-Oslo	IPOA <sup>2,3</sup>	3	Monoterpenes	37.5 Tg a <sup>-1</sup> from terpenes emissions based on Dentener et al. (2006) distribution, included in IPOA	1.4	1.4 for fossil and biofuel and 2.6 for biomass burning
CAM5-MAM3	IPOA, trSOA	3 <sup>4</sup>	Isoprene, terpenes, aromatics, higher molecular weight alkanes and alkenes	Prescribed mass yields for the 5 trSOA precursor categories (6.0, 37.5, 22.5, 7.5, and 7.5% respectively) that form a single semi-volatile species which then kinetically but reversibly partitions to the OA phase	1.4	Precursor VOCs are lumped species from MOZART. Yields listed include a 1.5X increase to reduce anthropogenic aerosol indirect forcing. The single semi-volatile gas has saturation mixing ratio of 0.1 ppbv at 298 K. Includes aerosol microphysics (MAM3; modal)
CanAM-PAM	IPOA <sup>3</sup>	3	Monoterpenes	15% yield from terpenes emissions (Dentener et al., 2006), included in IPOA	1.4	1 tracer in 3 size classes, internally mixed with BC and ammonium sulfate (2-moments)
CCSM4-Chem	IPOA, trSOA	7	Isoprene, monoterpenes, toluene, benzene, xylene	2-product model	1.4	Isoprene+OH is using the high-NO <sub>x</sub> pathway
ECHAM5-HAM2	IPOA, trSOA	24	Isoprene, monoterpenes, benzene, toluene, xylene	2-product model	1.4	Includes aerosol microphysics (M7; modal)
ECHAM5-HAMMOZ	IPOA <sup>3</sup>	4	Monoterpenes	15% yield from terpenes emissions (Dentener et al., 2006), included in IPOA	1.4	Includes aerosol microphysics (M7; modal)
ECHAM5-SALSA	IPOA <sup>3</sup>	11	Monoterpenes	15% yield from terpenes emissions (Dentener et al., 2006), included in IPOA	1.4	Includes aerosol microphysics (SALSA; sectional)
ECMWF-GEMS	IPOA <sup>3</sup>	2	Monoterpenes	15% yield from terpenes emissions (Dentener et al., 2006), included in IPOA	1.4	
EMAC	IPOA <sup>3</sup>	2	Monoterpenes	15% yield from terpenes emissions (Dentener et al., 2006), included in IPOA	1.4	Includes aerosol microphysics (GMXe, based on M7; sectional)
GEOS-Chem	IPOA, trSOA	5	Isoprene, monoterpenes, sesquiterpenes	2-product model	2.1	
GEOS-Chem-APM	IPOA, trSOA, MSA	24	Isoprene, monoterpenes, limonene, sesquiterpenes, alcohols, benzene, toluene, xylene	2-product model + aging/condensation	2.1	Considers the volatility changes of the gaseous semi-volatile compounds arising from the oxidation aging process, as well as the kinetic condensation of low volatility gases: Includes aerosol microphysics (bins)
GISS-CMU-TOMAS	IPOA <sup>3</sup>	24	Terpenes	A generic SOA precursor (Dentener et al., 2006) representing all SOA precursor gases is emitted and forms non-volatile SOA (included in IPOA) with a first-order loss rate of 12 h	1.8	Includes aerosol microphysics (sectional)
GISS-CMU-VBS	IPOA, trSOA, ntrSOM	26	Isoprene, monoterpenes, sesquiterpenes, alkanes, alkenes and aromatics (VOCs with C <sup>*</sup> < 10 <sup>6</sup> μg m <sup>-3</sup> )	Volatility-basis set	1.8	IPOA is treated as semi-volatile and reactive. ntrSOA is formed from the gas-phase oxidation of IPOA.

Table 4. Continued.

Model	OA types <sup>1</sup>	# of OA tracers	trSOA precursors	trSOA calculations	OA/OC	Comments
GISS-MATRIX	IPOA <sup>3</sup>	3	Monoterpenes	10 % yield from monoterpenes emissions (Lathière et al., 2005), included in tPOA	1.4	Includes aerosol microphysics (moments)
GISS-modelE-G	IPOA, mPOA, trSOA, MSA	9	Isoprene, terpenes	2-product model	1.4	
GISS-modelE-I	IPOA, mPOA, trSOA, MSA	9	Isoprene, terpenes	2-product model	1.4	
GISS-TOMAS	IPOA <sup>3</sup>	24	Terpenes	A generic SOA precursor (Lathière et al., 2005) representing all SOA precursor gases is emitted and forms non-volatile SOA (included in IPOA) with a first-order loss rate of 12h	1.4	Includes aerosol microphysics (sectional)
GLOMAPbin	IPOA <sup>3</sup>	40	Monoterpenes	<i>a</i> -pinene + all oxidants → 13 % non-volatile SOA (included in IPOA)	1.4	Includes aerosol microphysics (bin)
GLOMAPmode	IPOA <sup>3</sup>	5	Monoterpenes	<i>a</i> -pinene + all oxidants → 13 % non-volatile SOA (included in IPOA)	1.4	Includes aerosol microphysics (modal)
GMI	IPOA <sup>3</sup>	3	Monoterpenes	10 % yield from monoterpenes emissions (GEIA), included in IPOA	1.4	
GOCART	IPOA <sup>3</sup>	2	Monoterpenes	10 % yield from monoterpenes emissions (GEIA), included in IPOA	1.4	50 % of anthropogenic and biomass burning OC is emitted as hydrophobic and 50 % as hydrophilic (Cooke et al., 1999); hydrophobic OC becomes hydrophilic in an e-folding time of 2.5 days.
HadGEM2-ES	IPOA, trSOA	3	Terpenes	Fixed 3-D monthly climatology obtained from STOCHEM (Derwent et al., 2003)	1.4	3 tracers for fossil fuel organic carbon aerosols (fresh, aged, dissolved in cloud water).
IMAGES	IPOA, trSOA, ntrSOA	26	Isoprene, <i>a</i> -pinene, sesquiterpenes, benzene, toluene, xylene	2-product model	varying	trSOA includes the effect of water uptake on partitioning, ntrSOA is glyoxal and methylglyoxal from cloud chemistry and aqueous aerosol processing.
IMPACT	IPOA <sup>5</sup> , trSOA, ntrSOA	33	Isoprene, terpenes, aromatics	SOA comes from organic nitrates and peroxides using the traditional gas-particle partitioning with an explicit full chemistry. The condensed SOA is further assumed to form oligomers with a 1 day e-folding time.	varying	ntrSOA from the uptake of gas-phase glyoxal and methylglyoxal onto clouds and aqueous sulfate aerosol (Fu et al., 2008; Fu et al., 2009) and uptake of gas-phase epoxides onto aqueous sulfate aerosol (Paulot et al., 2009).
LMDz-INCA	IPOA <sup>3</sup> , MSA	3	Monoterpenes	15 % yield from terpenes emissions (Dentener et al., 2006), included in IPOA	1.4	
OscioCTM2	IPOA, trSOA	62	Isoprene, 5 classes of terpenoid compounds (Griffin et al., 1999b), 2 classes of aromatics	2-product model	1.6 for fossil and biofuel and 2.6 for biomass burning	

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Table 4. Continued.

Model	OA types <sup>1</sup>	# of OA tracers	trSOA precursors	trSOA calculations	OA/OC	Comments
SPRINTARS	tPOA, trSOA	2	Monoterpenes <sup>6</sup>	9.2% yield of non-volatile trSOM (Griffin et al., 1999a; Griffin et al., 1999b) from monoterpenes emissions (GEIA) <sup>6</sup>	1.6 for fossil fuel and biofuel, 2.6 for other	
TM4-ECPL-F	tPOA, mPOA, trSOA, ntr-SOA, MSA	22	Isoprene, terpenes, aromatics	2-product model	varying	ntrSOA is oxalic acid, glyoxilic acid and glyoxal oligomers from cloud chemistry
TM4-ECPL-FNP	tPOA, mPOA, trSOA, ntr-SOA, MSA	24	Isoprene, terpenes, aromatics	2-product model	varying	tPOA includes primary biogenic particles and organics associated with soil dust; ntrSOA is oxalic acid, glyoxilic acid and glyoxal oligomers from cloud chemistry.
TM5	tPOA <sup>3</sup> , MSA	4	Monoterpenes	15% yield from terpenes emissions (Dentener et al., 2006), included in tPOA	1.4	Includes aerosol microphysics (M7; modal)

<sup>1</sup> tPOA: terrestrial primary organic aerosol mass; mPOA: marine primary organic aerosol mass; trSOA: traditional secondary organic aerosol mass; ntrSOA: non-traditional secondary organic aerosol mass; MSA: methane sulfonic acid.

<sup>2</sup> tPOA also includes mPOA and MSA.

<sup>3</sup> tPOA also includes trSOA.

<sup>4</sup> tPOA in accumulation mode; trSOA in accumulation and Aitken modes. Aitken mode mass is minor.

<sup>5</sup> tPOA also includes mPOA.

<sup>6</sup> SPRINTARS also has a 2-product model configuration (not presented here), with trSOA coming from isoprene and  $\alpha$ -pinene.



**Table 5.** Summary of organic aerosol processes taken into account by the models.

Model	SOA like tPOA	mPOA in tPOA	Simple SOA, irreversible partitioning	Reversible partitioning (equilibrium)	Includes ntrSOA	MSA in tPOA	Microphys. aging
BCC	X						
CAM4-Oslo	X	in tPOA				in tPOA	X
CAM5-MAM3				kinetically			X
CanAM-PAM	X						
CCSM4-Chem				X			
ECHAM5-HAM2				X			X
ECHAM5-HAMMOZ	X						X
ECHAM5-SALSA	X						X
ECMWF-GEMS	X						
EMAC	X						X
GEOS-Chem				X			
GEOS-Chem-APM				X		X	X
GISS-CMU-TOMAS			X				X
GISS-CMU-VBS				X	VBS		
GISS-MATRIX	X						X
GISS-modelE-G		X		X		X	
GISS-modelE-I		X		X		X	
GISS-TOMAS			X				X
GLOMAPbin			X				X
GLOMAPmode			X				X
GMI	X						
GO CART	X						
HadGEM2-ES				offline			
IMAGES				X	Aqueous		
IMPACT		in tPOA		X	Aqueous	in tPOA	
LMDz-INCA	X					X	
OsloCTM2				X			
SPRINTARS	X						
TM4-ECPL-F		X		X	Aqueous	X	chemical
TM4-ECPL-FNP		X		X	Aqueous	X	chemical
TM5	X					X	X

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**Table A1.** List of acronyms.

AeroCom	Aerosol Comparisons between Observations and Models. For hindcast emissions, see (Diehl et al., 2012).
ACSM	Aerosol Chemical Specification Monitor, a mini-AMS (Ng et al., 2011).
AMS	Aerosol Mass Spectrometry (Jayne et al., 2000).
BVOC	Biogenic VOC.
CIRCE	Climate Change and Impact
Research	the Mediterranean Environment ( <a href="http://www.circeproject.eu">http://www.circeproject.eu</a> ; Doering et al., 2009).
CMIP5	Coupled Model Intercomparison Project phase 5 ( <a href="http://cmip-pcmdi.llnl.gov/cmip5">http://cmip-pcmdi.llnl.gov/cmip5</a> ). For historical emissions, see (Lamarque et al., 2010).
DMS	DiMethyl Sulfide, CH <sub>3</sub> SCH <sub>3</sub> .
ECMWF	European Center for Medium-Range Weather Forecasts.
GEIA	Global Emissions Inventory Activity ( <a href="http://geiacenter.org">http://geiacenter.org</a> ). For BVOC emissions, see (Guenther et al., 1995).
GFED	Global Fire Emissions Database (van der Werf et al., 2003).
GFED2	Global Fire Emissions Database, version 2 (van der Werf et al., 2006).
GFED3	Global Fire Emissions Database, version 3 (van der Werf et al., 2010).
HOA	Hydrocarbon-like OA.
LAD	Least Absolute Deviation technique.
MEGAN	Model of Emissions of Gases and Aerosols from Nature (Guenther et al., 2006).
mPOA/mPOC	marine POA/POC.
MERRA	Modern-Era Retrospective analysis for Research and Applications.
MNB	Mean Normalized Bias.
MSA	Methane Sulfonic Acid, CH <sub>3</sub> SO <sub>3</sub> H.
NCEP	National Centers for Environmental Prediction.
ntrSOA/ntrSOC	non-traditional SOA/SOC. For IMAGES, IMPACT and TM4-ECPL-F/FNP this is OA produced from multiphase chemistry, while for GISS-CMU-VBS it is OA formed from the VBS gas-phase chemistry.
OA	Organic Aerosol and Organic Aerosol mass (as appropriate)
OC	Organic Carbon.
OOA	Oxygenated OA.
ntrSOA/ntrSOC	non-traditional Secondary Organic Aerosol mass/Carbon.
POA/POC	Primary OA/OC.
POET	Present and future surface emissions of atmospheric compounds ( <a href="http://accent.aero.jussieu.fr/POET.php">http://accent.aero.jussieu.fr/POET.php</a> ; Granier et al., 2003)
RETRO	REanalysis of the TROpospheric chemical composition over the past 40 yr ( <a href="http://retro.enes.org">http://retro.enes.org</a> ; Schultz et al., 2007).
SOA	Secondary Organic Aerosol.
tPOA/tPOC	terrestrial POA/POC.
trSOA/trSOC	traditional SOA/SOC.
VBS	Volatility-Basis Set.
VOC	Volatile Organic Compounds.
WSOC	Water Soluble Organic Compounds.
WIOC	Water Insoluble Organic Compounds.

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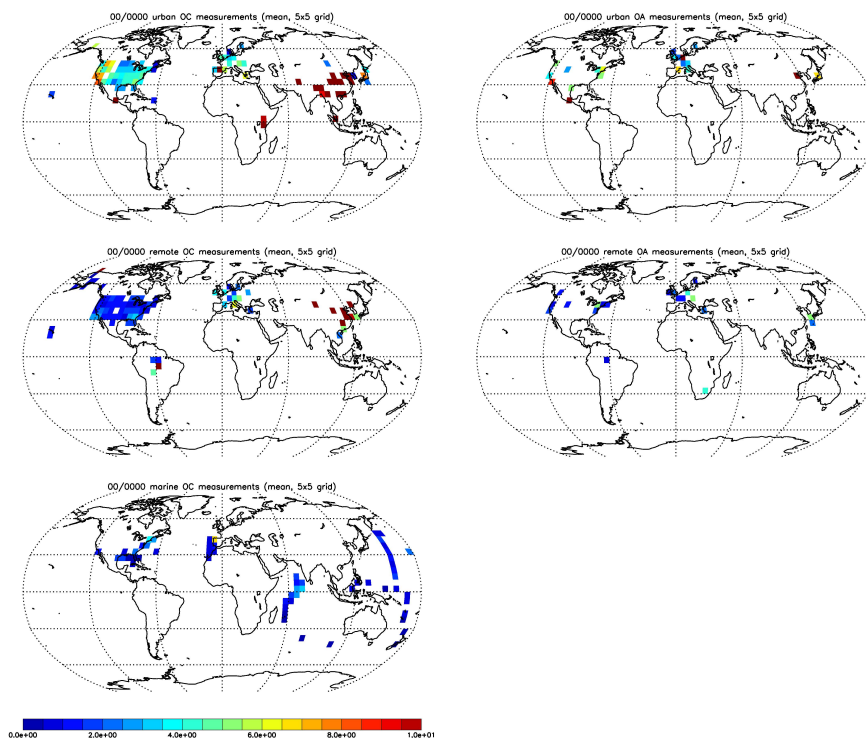
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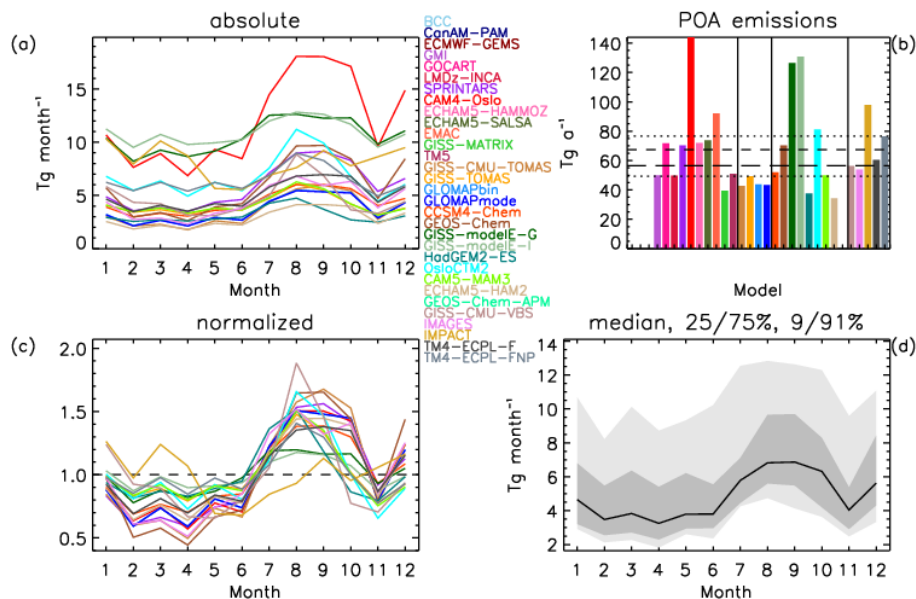


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**Fig. 1.** Mean measured OC ( $\mu\text{g C m}^{-3}$ ; left) and OA ( $\mu\text{g m}^{-3}$ ; right) concentrations on a  $5^\circ \times 5^\circ$  grid for urban (top), remote (middle) and marine (bottom) locations. The number of data points per grid cell, as well as the individual stations, are presented in the Supplement. The date 00/0000 means all months and all years of data.



**Fig. 2.** POA emissions included in models. Absolute seasonal variability **(a)**, annual mean **(b)**; short dashes: mean; long dashes: median, dotted lines: 25/75 % of the data), normalized seasonal variability over each model's mean **(c)**, and full model ensemble range **(d)**; black line: median; dark grey: 25/75 % of the data; light grey: 9/91 % of the data). Note that not all models have submitted budget data, and less have submitted seasonal information. The models are grouped based on their complexity in **(c)**, as separated by vertical solid lines. Groups from left to right are: SOA is directly emitted as a non-volatile tracer; SOA is chemically formed in the atmosphere but is considered non-volatile; SOA is semi-volatile; SOA is semi-volatile and also has VBS (GISS-CMU-VBS) or multiphase chemistry sources.

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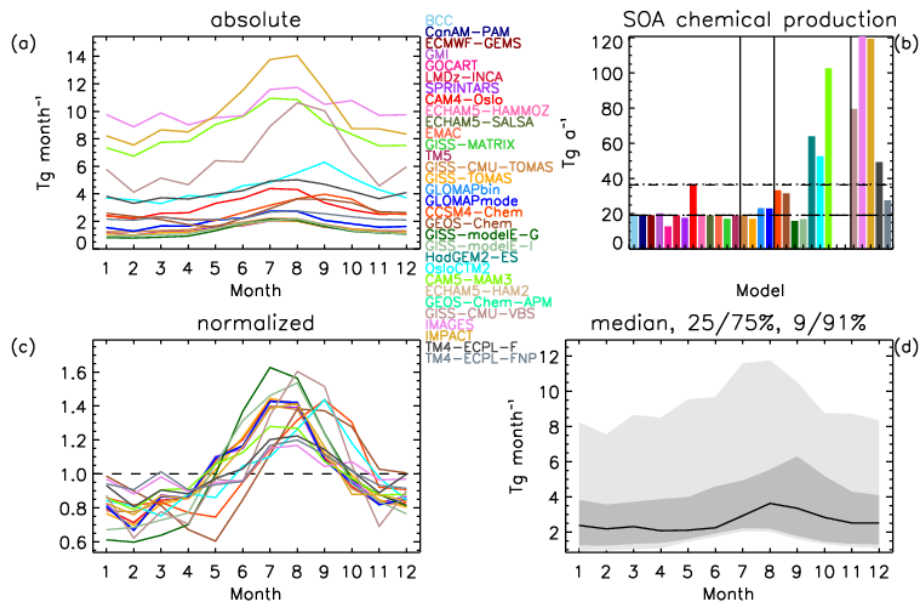


Fig. 3. Same as in Fig. 2 for SOA chemical production.

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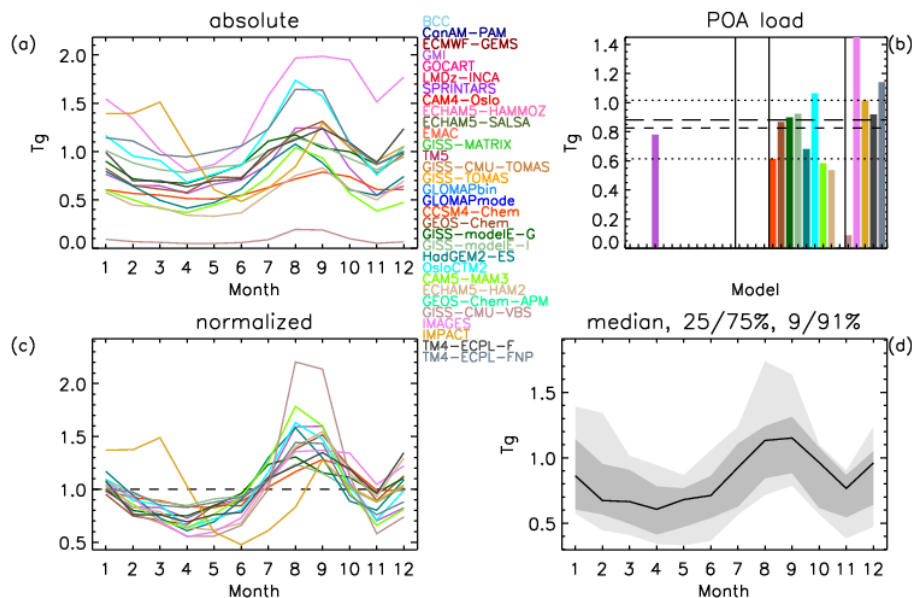
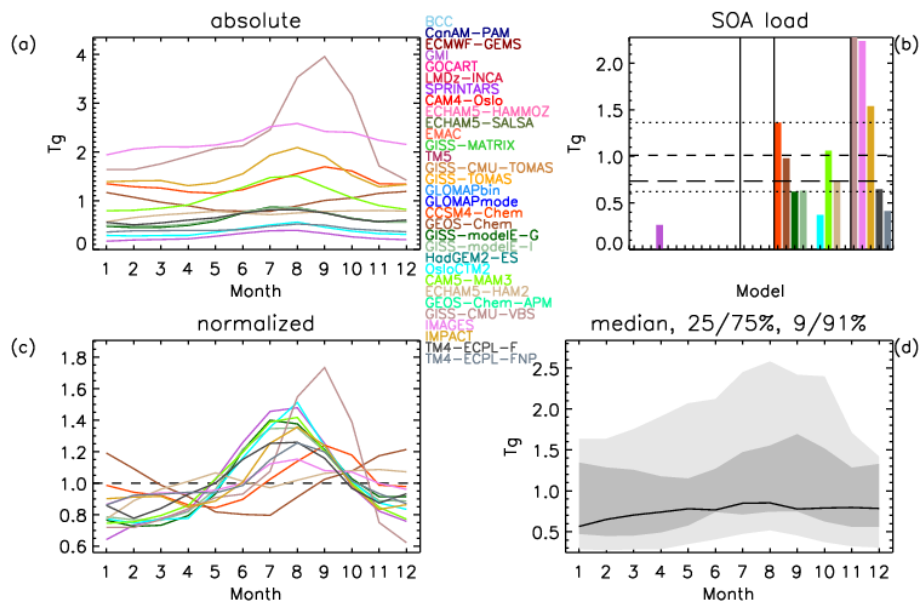


Fig. 4. Same as in Fig. 2 for POA load.

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**Fig. 5.** Same as in Fig. 2 for SOA load.

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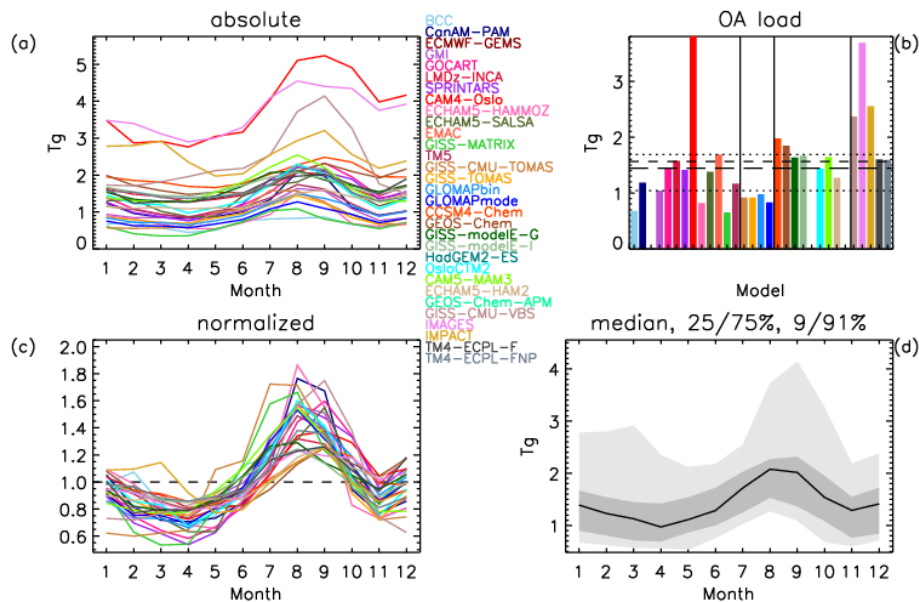


Fig. 6. Same as in Fig. 2 for OA load.

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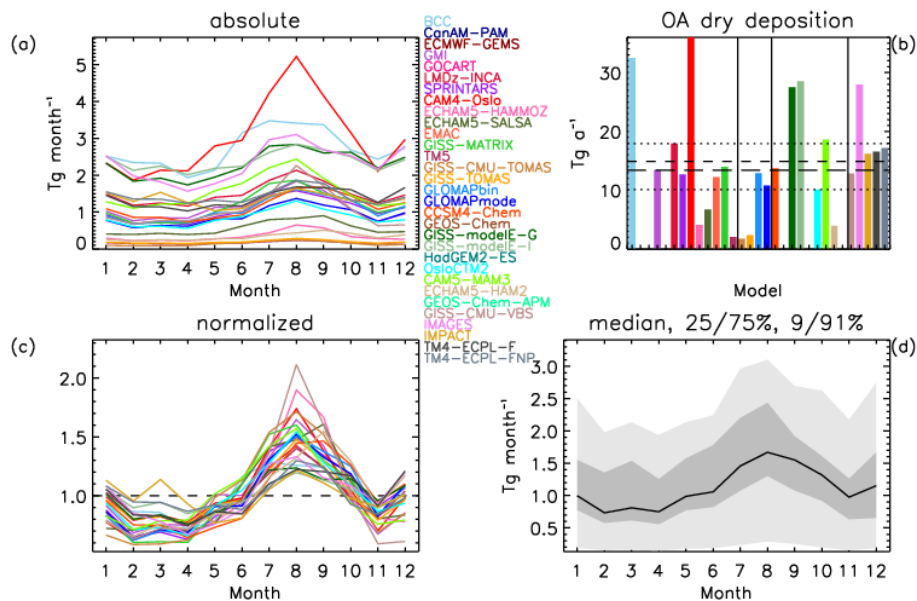


Fig. 7. Same as in Fig. 2 for OA dry deposition.

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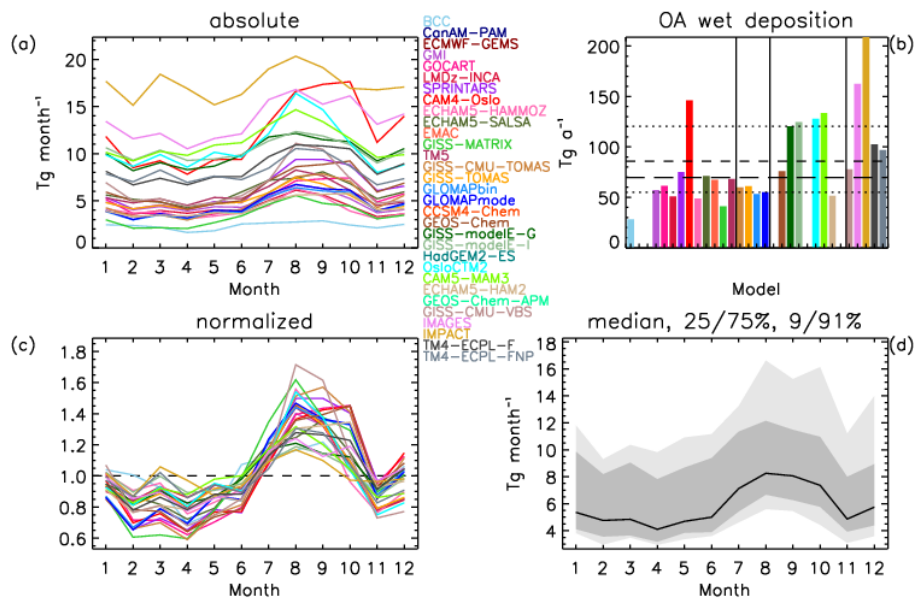
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**Fig. 8.** Same as in Fig. 2 for OA wet deposition.

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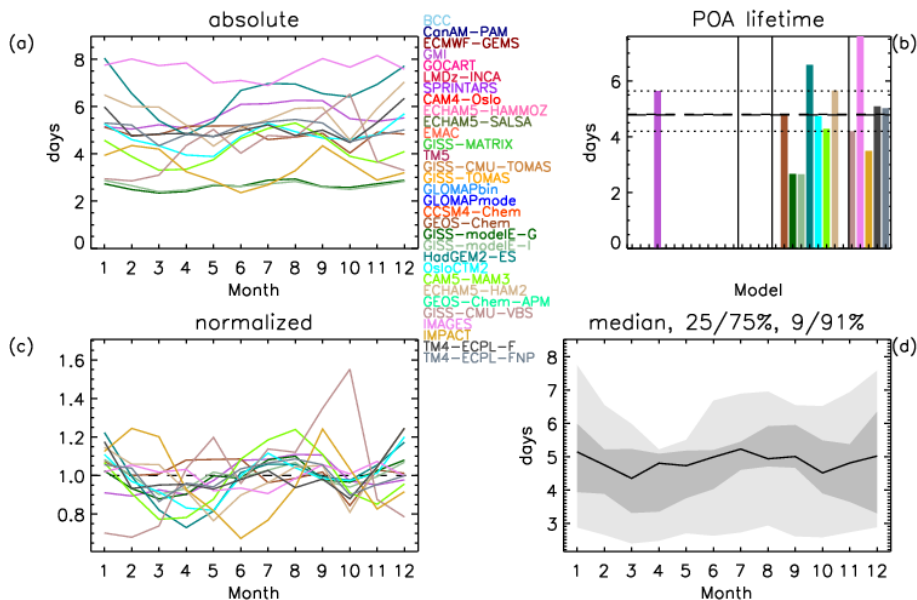


Fig. 9. Same as in Fig. 2 for POA lifetime.

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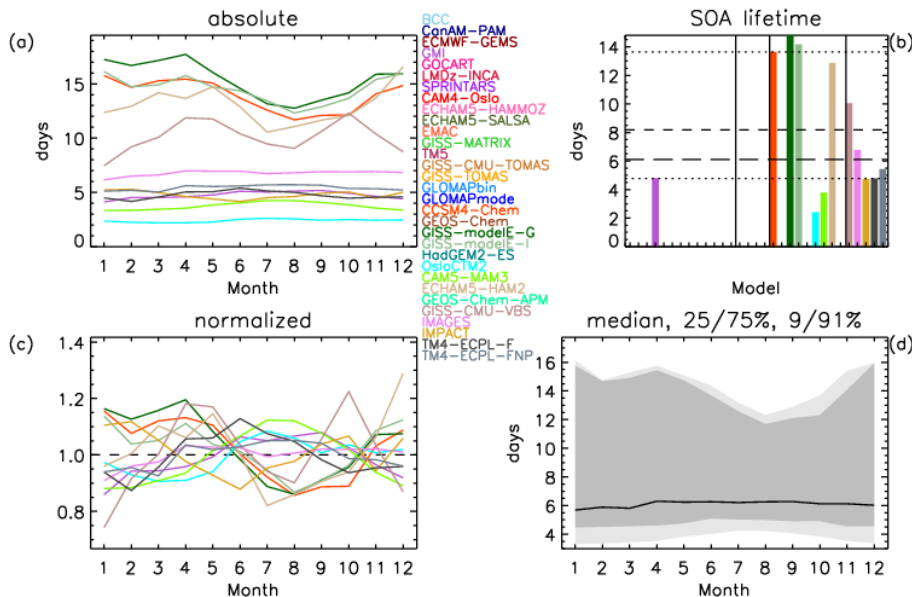


Fig. 10. Same as in Fig. 2 for SOA lifetime.

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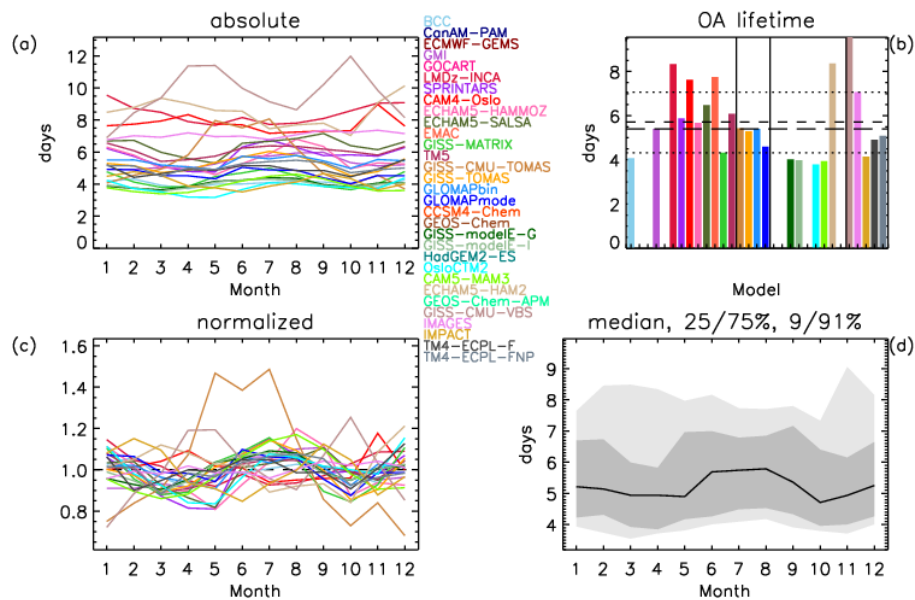


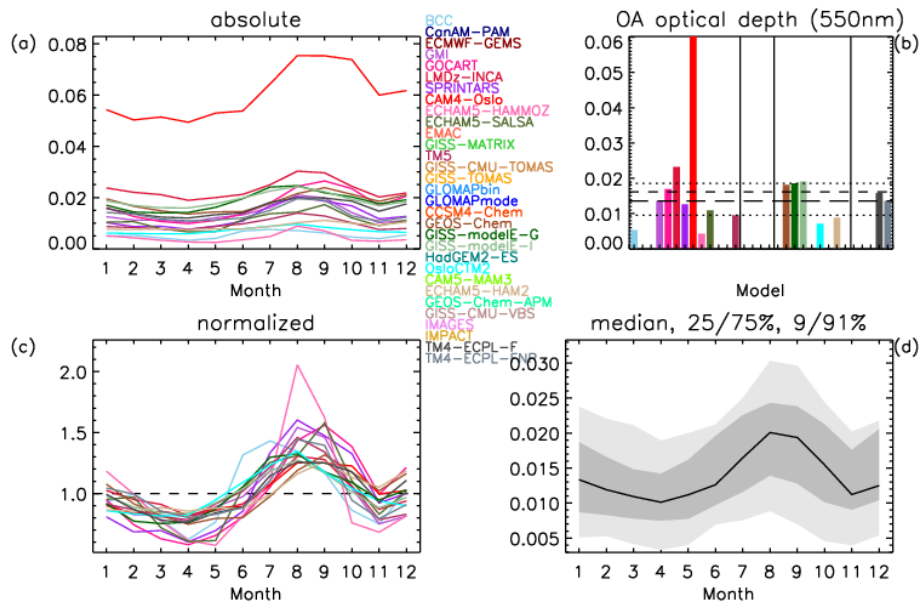
Fig. 11. Same as in Fig. 2 for OA lifetime.

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**Fig. 12.** Same as in Fig. 2 for OA all-sky aerosol optical depth at 550 nm.

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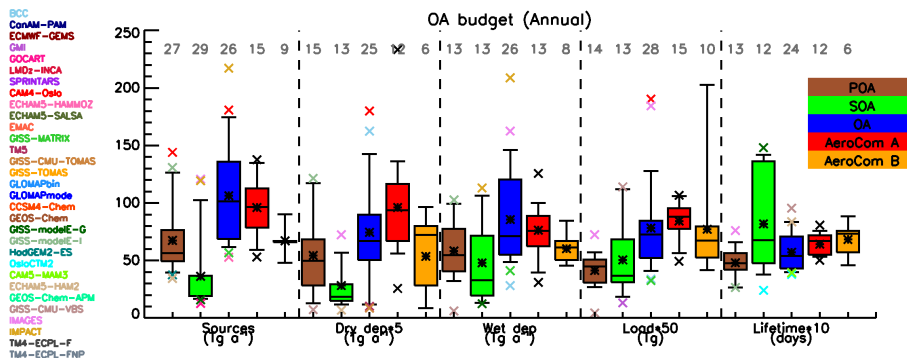
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**Fig. 13.** Box and whisker plot for all POA, SOA and OA global budgets and comparison with AeroCom phase I (Textor et al., 2006, 2007) results. The boxes represent the first and third quartile range (50% of the data), the line is the median value, the star is the mean, and the error bars represent the 9/91% of the data. Outliers are presented with x-symbols, with the corresponding color of the model, and the numbers of models participating in each bars statistics are presented with a grey number at the top. The AeroCom phase I outliers are presented with black color, since there is no direct correspondence with the models that participate in the present study. Bar colors are POA (brown), SOA (green), OA (blue), AeroCom A (red; Textor et al., 2006), and AeroCom B (orange; Textor et al., 2007).

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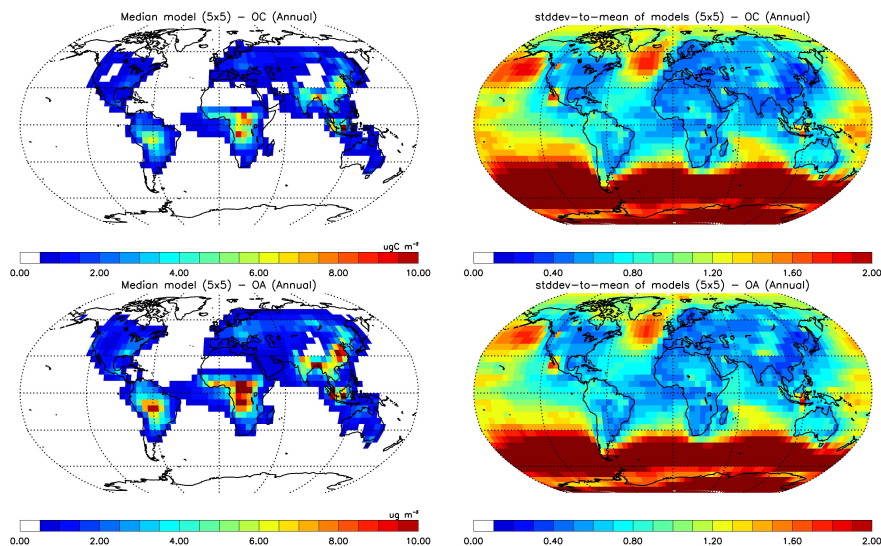
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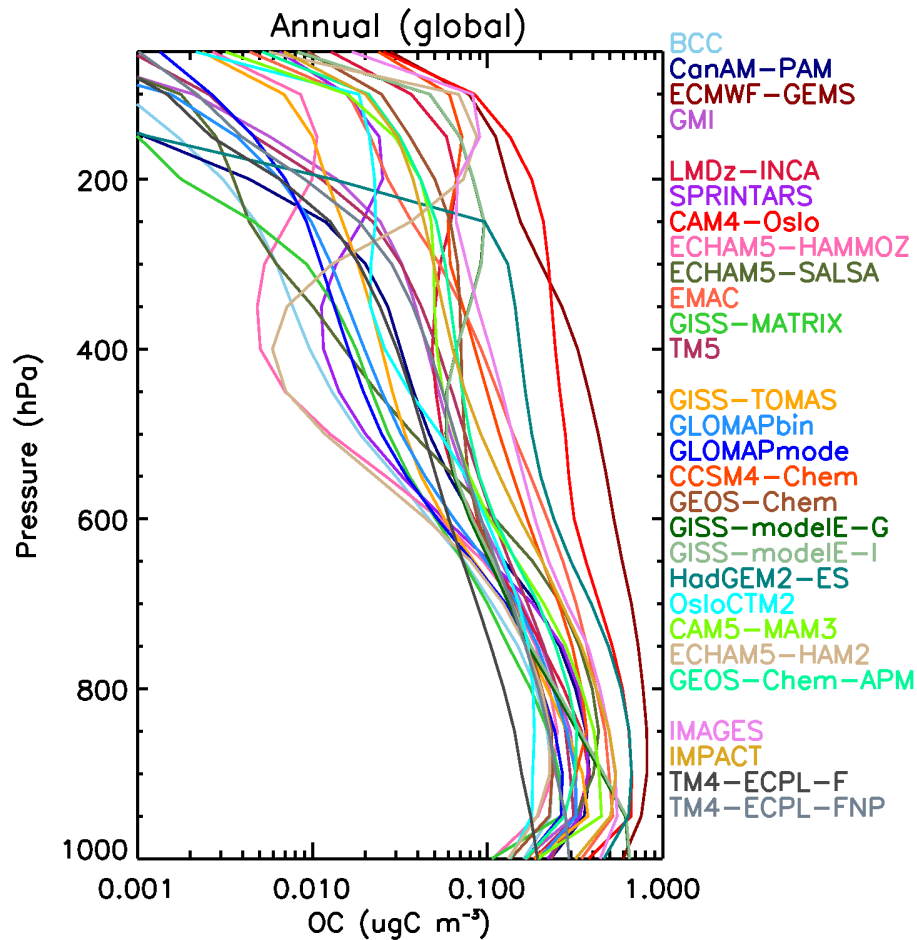
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**Fig. 14.** Annual mean (left) and model diversity (right), defined as the standard deviation of the models over their mean, of the median model surface air concentration of for OC (top) and OA (bottom) on a  $5^\circ \times 5^\circ$  grid.

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**Fig. 15.** Annual global mean vertical profile of OC (at ambient conditions) interpolated at 50 hPa steps from surface to 50 hPa for OC.

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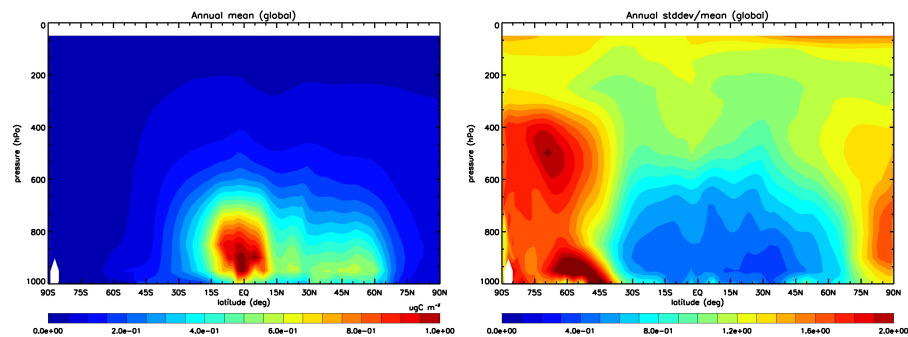
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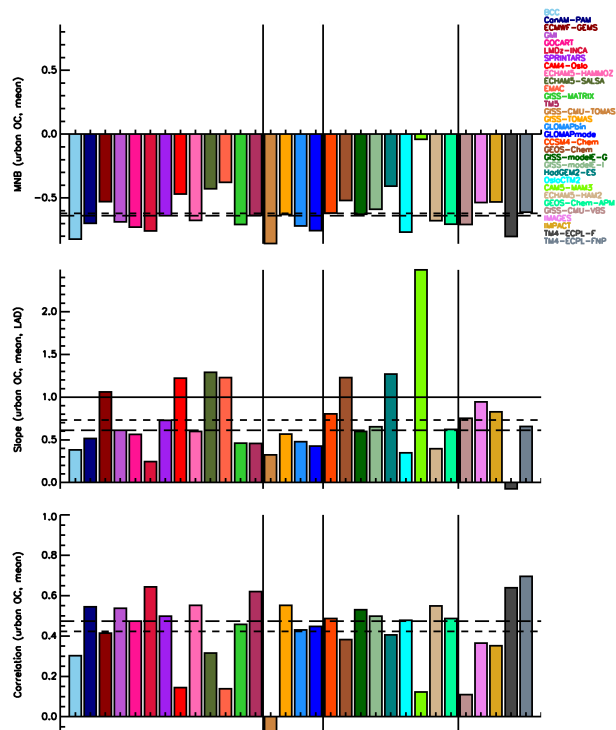


**Fig. 16.** Annual zonal mean (left) and standard deviation over mean (right) of the median model results interpolated at 50 hPa steps for OC.

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**Fig. 17.** Mean normalized bias (top), slope, calculated by the least absolute deviation (LAD) technique (middle), and correlation (bottom) of all models against urban OC measurements. When more than one measurement data point exists in a model grid, the corresponding mean of the measurements was used (see text). The mean across all models is shown with the short-dashed line, the median with the long-dashed line, and the slope of 1 with the solid line. The perfect model has MNB = 0, slope = 1 and correlation = 1. The models are grouped like in Fig. 2.

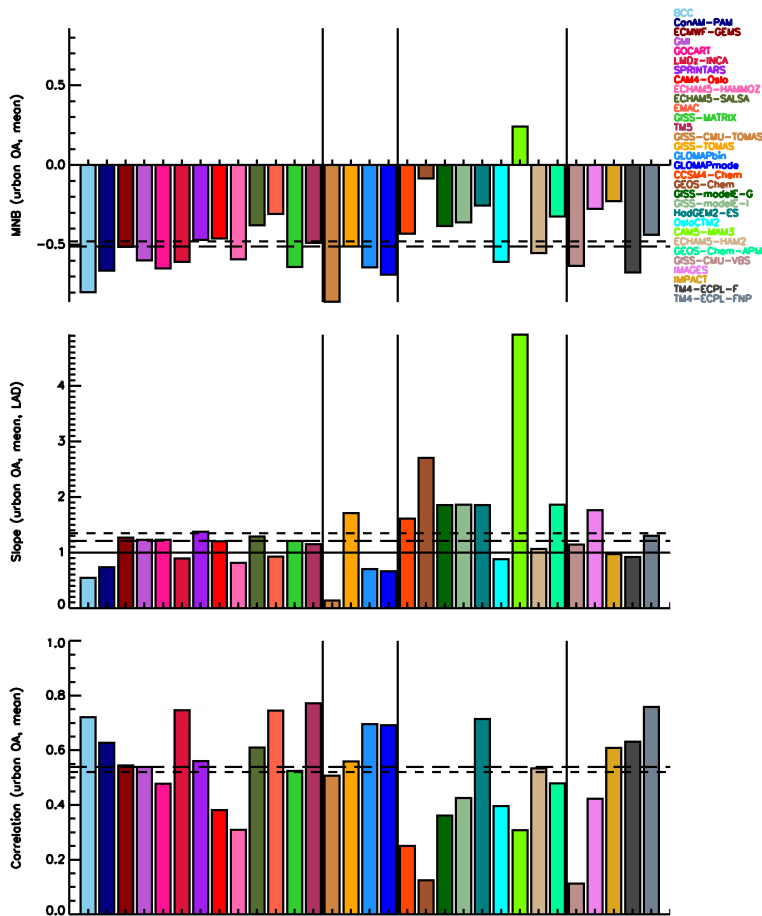


Fig. 18. Same as Fig. 17 for urban OA measurements.

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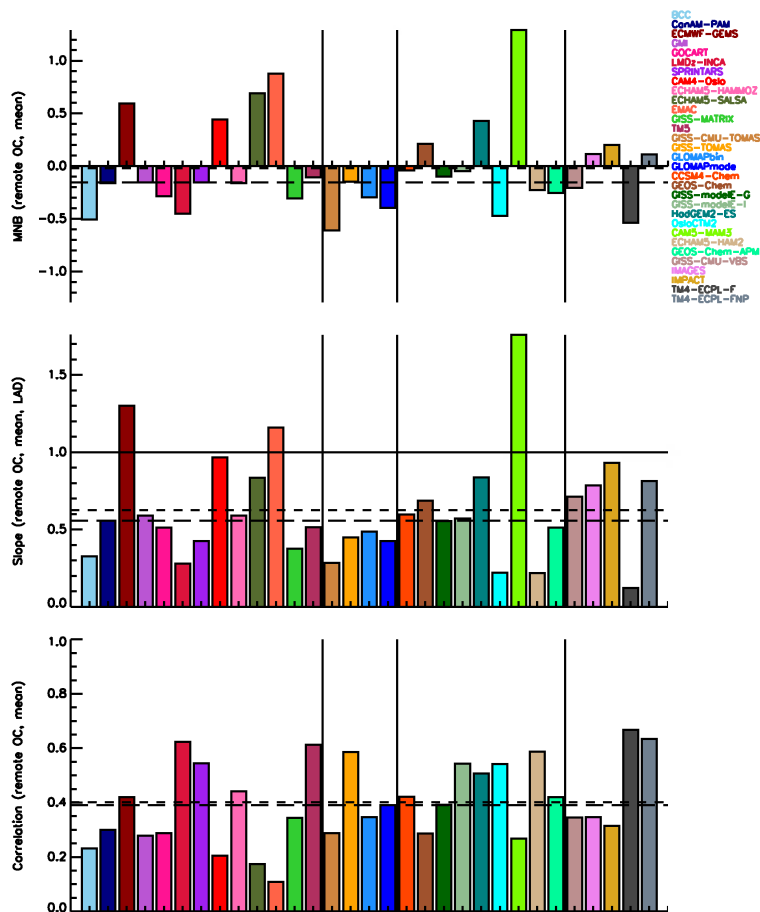


Fig. 19. Same as Fig. 17 for remote OC measurements.

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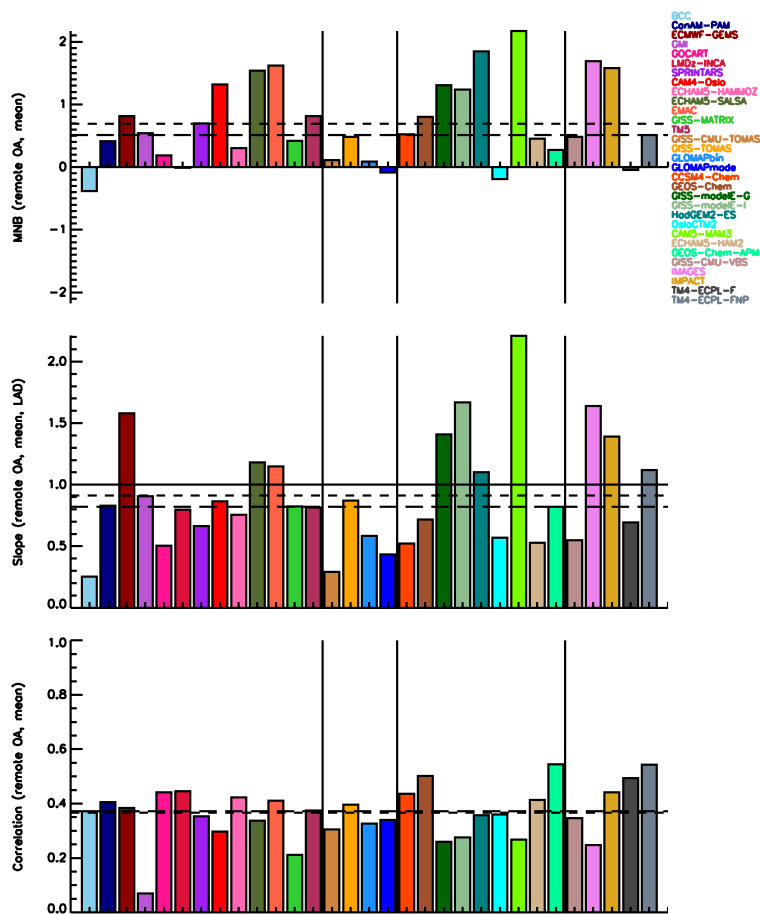


Fig. 20. Same as Fig. 17 for remote OA measurements.

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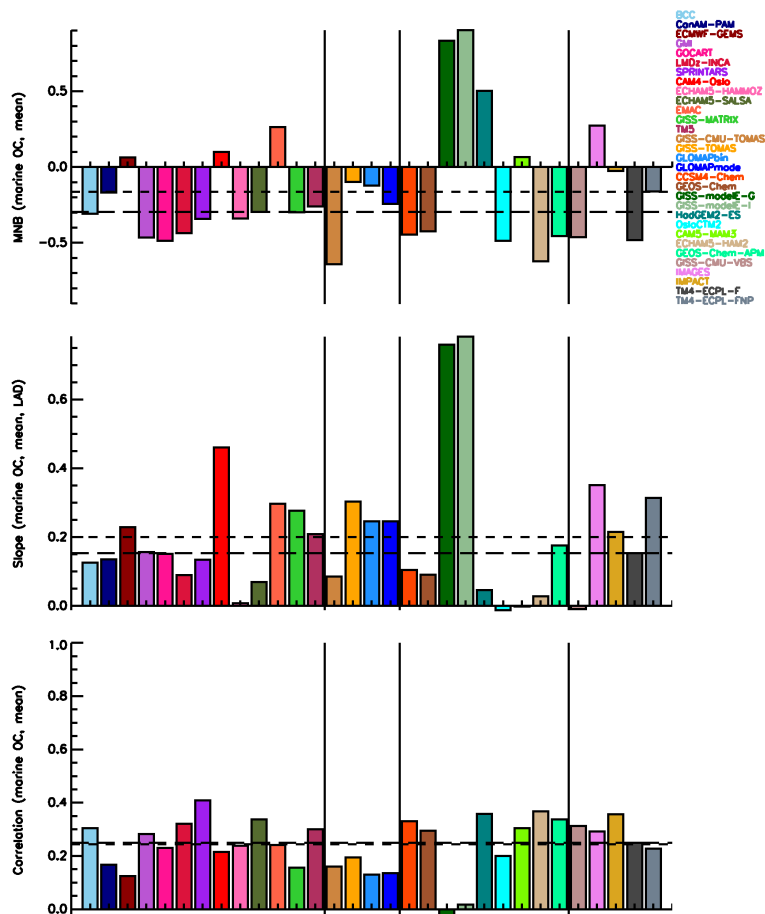
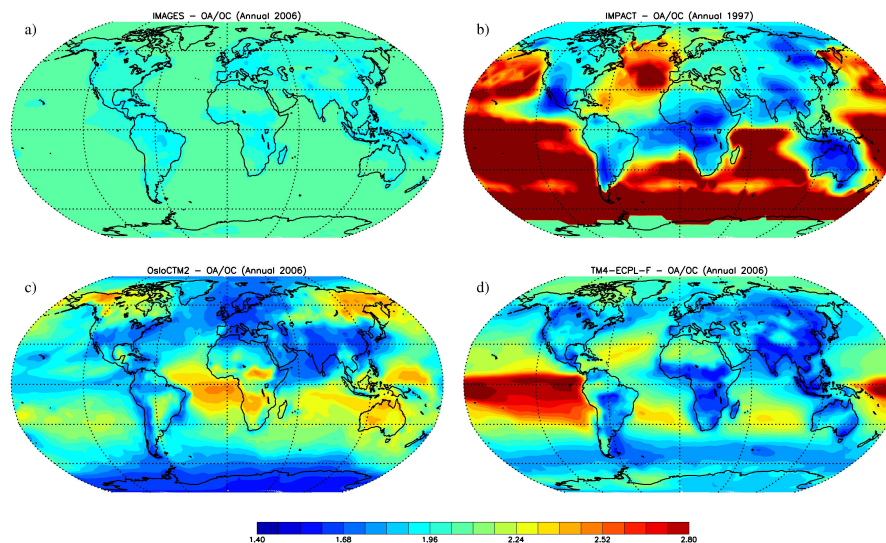


Fig. 21. Same as Fig. 17 for marine OC measurements.

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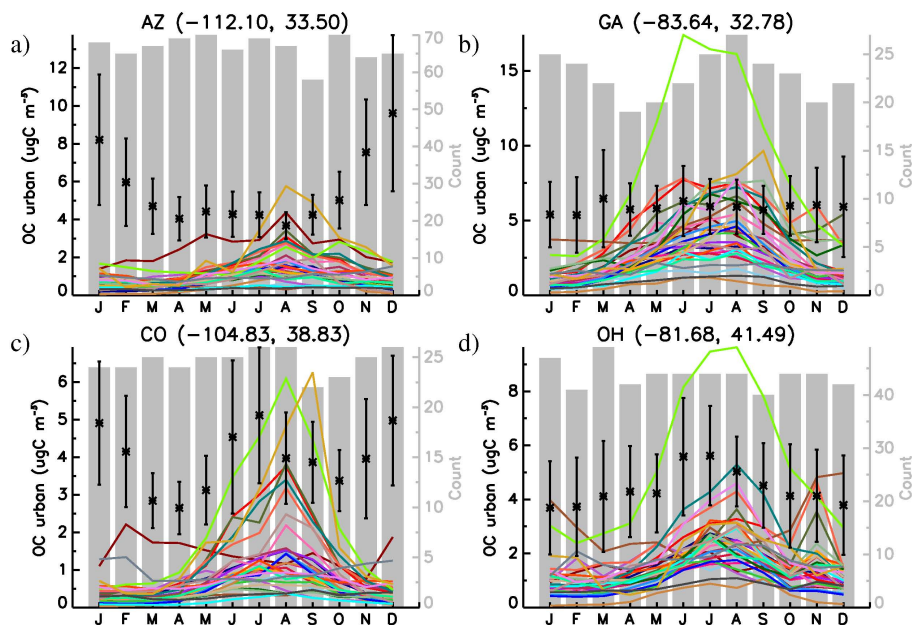
**Fig. 22.** Annual mean OA/OC as calculated by IMAGES (a), IMPACT (b), OsloCTM2 (c) and TM4-ECPL-F (d).

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**Fig. 23.** Typical seasonal distribution of OC measurements and comparison with model results for urban stations. Stars show the monthly mean of all measurements from all years that data are available, error bars present the standard deviation of the averaged measurements per month, and lines show model results, colored as in the previous figures. The grey bars show the number of measurements per month. The stations used are Arizona (**a**; 112.1° W, 33.5° N, years 2000–2007); Georgia (**b**; 83.64° W, 32.78° N, years 2001–2008); Colorado (**c**; 104.83° W, 38.83° N, years 2002–2006); Ohio (**d**; 81.68° W, 41.49° N, years 2001–2003 and 2005–2007).

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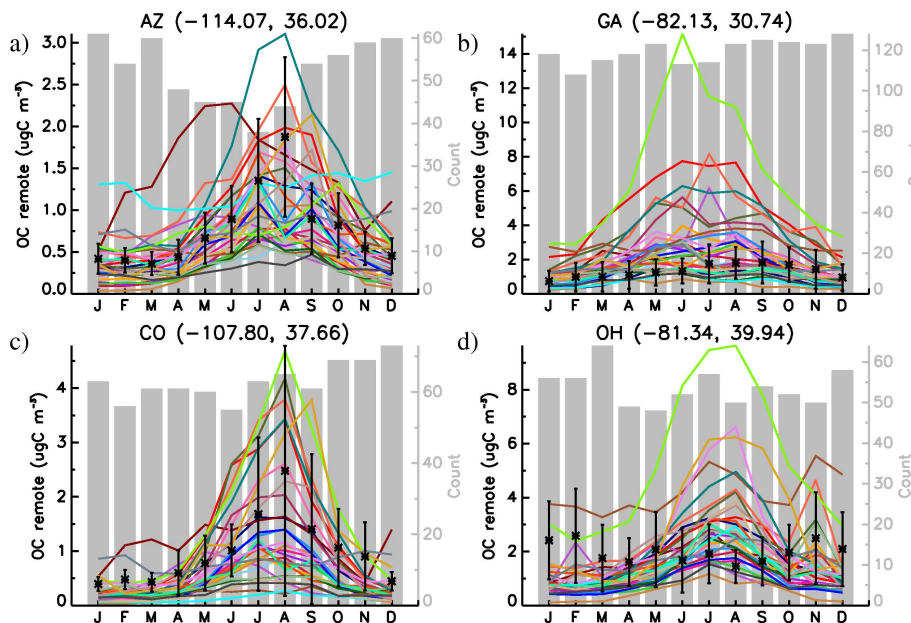
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**Fig. 24.** Same as in Fig. 23 for remote stations. Arizona (**a**;  $114.07^\circ$  W,  $36.02^\circ$  N, years 2000–2006); Georgia (**b**;  $82.13^\circ$  W,  $30.74^\circ$  N, years 1993–2006); Colorado (**c**;  $107.80^\circ$  W,  $37.66^\circ$  N, years 2000–2006); Ohio (**d**;  $81.34^\circ$  W,  $39.94^\circ$  N, years 1998–2004).

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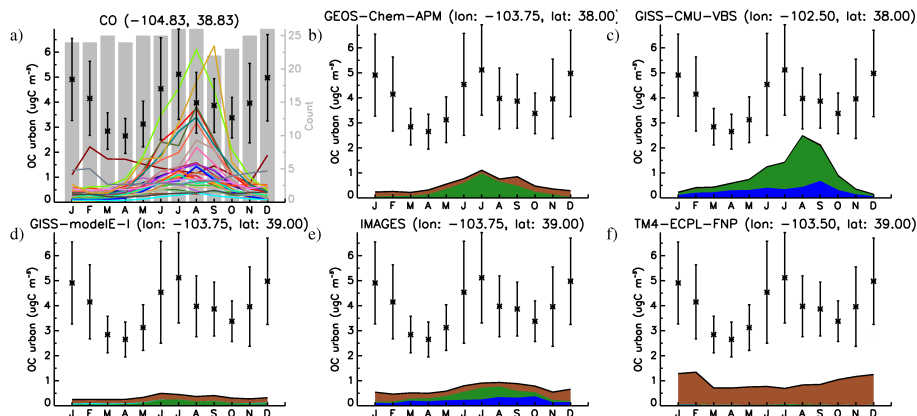
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**Fig. 25.** OC seasonality as calculated by all models **(a)** and chemical composition in GEOS-Chem-APM **(b)**, GISS-CMU-VBS **(c)**, GISS-modelE-I **(d)**, IMAGES **(e)** and TM4-ECPL-FNP **(f)** for Colorado, USA (urban, years 2002–2006). The coordinates in panel **(a)** show the location of the station, while those in panels **(b–f)** show the center of the grid box of the corresponding model. Panel **(a)** is similar with those presented in Figs. 23 and 24; for the chemical composition in panels **(b–f)**, brown is tPOC, green is trSOC, cyan is mPOC, blue is ntrSOC (aged tPOC for GISS-CMU-VBS, OA formed via multiphase chemistry for all other cases), and orange is MSA. The chemical composition of the remaining models that have submitted at least both tPOC and trSOC data are presented in the Supplement. Note the different scales on the y-axes.

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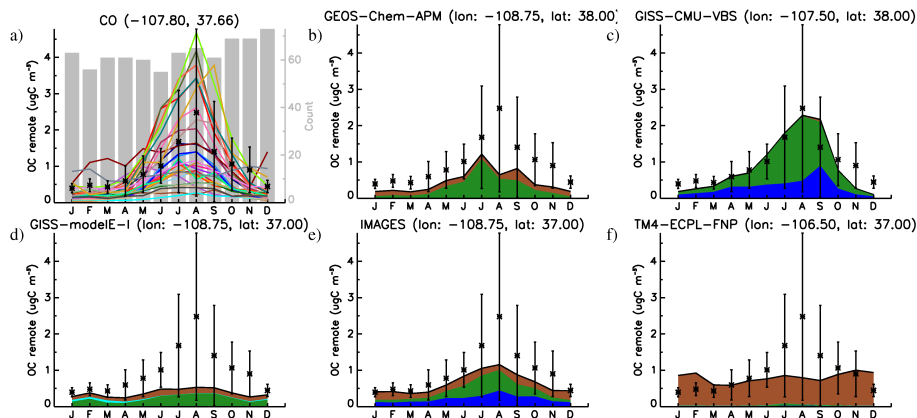
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**Fig. 26.** Same as in Fig. 25, for Colorado, USA (remote, years 2000–2006). For the chemical composition in panels (b–f), brown is tPOC, green is trSOC, cyan is mPOC, blue is ntrSOC, and orange is MSA.

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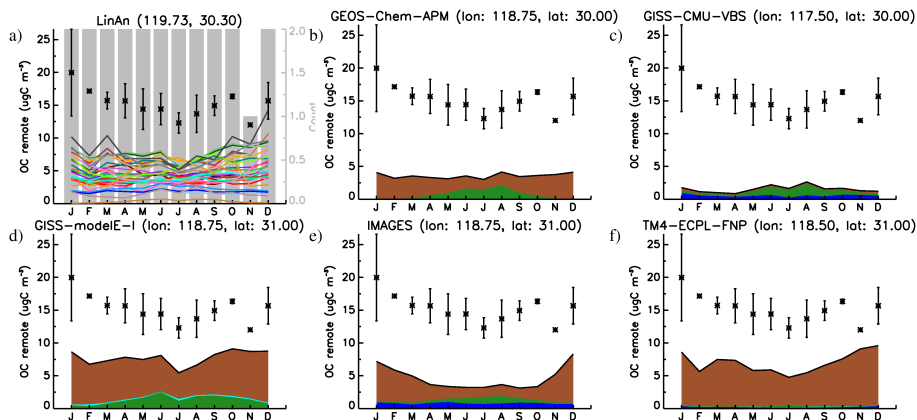
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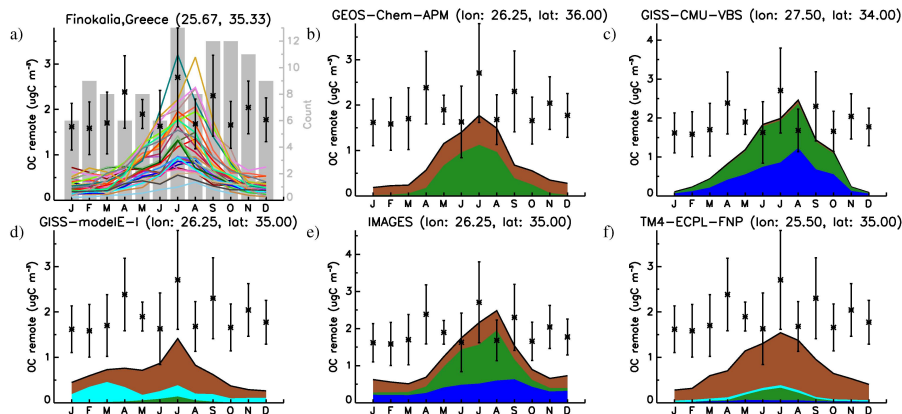
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**Fig. 27.** Same as in Fig. 25, for LinAn, China (remote, years 2006–2007). For the chemical composition in panels (b–f), brown is tPOC, green is trSOC, cyan is mPOC, blue is ntrSOC, and orange is MSA.

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**Fig. 28.** Same as in Fig. 25, for Finokalia, Greece (remote, years 2004–2007). For the chemical composition in panels (b–f), brown is tPOC, green is trSOC, cyan is mPOC, blue is ntrSOC, and orange is MSA.

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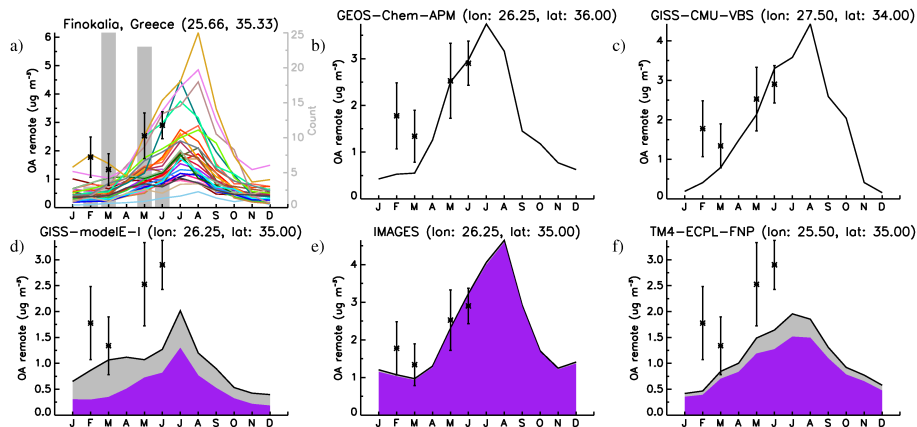
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**Fig. 29.** Same as in Fig. 28, for OA (years 2008 and 2009). The chemical composition in panels (b–f) (where available) is presented as defined by the AMS: HOA (grey) and OOA (purple).

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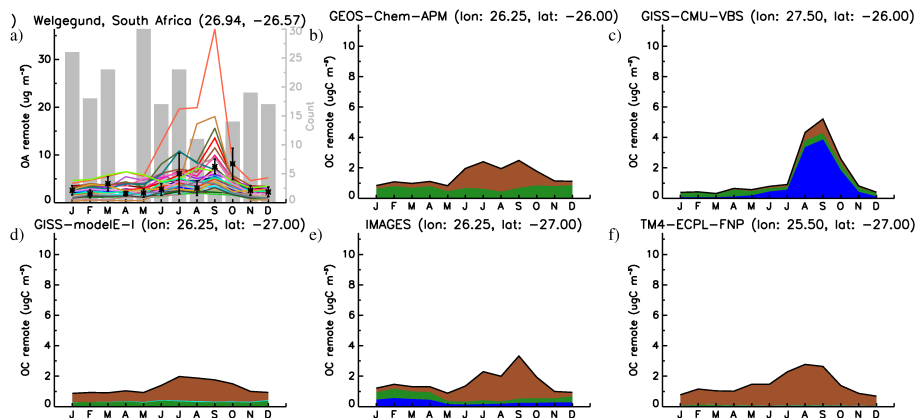
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**Fig. 30.** Same as in Fig. 25, at Welgegund, South Africa (remote, years 2010–2011). No measurements are plotted at the chemical composition panels, since measurements are OA and the chemical composition data from the models are OC. For the chemical composition in panels (b–f), brown is tPOC, green is trSOC, cyan is mPOC, blue is ntrSOC, and orange is MSA.

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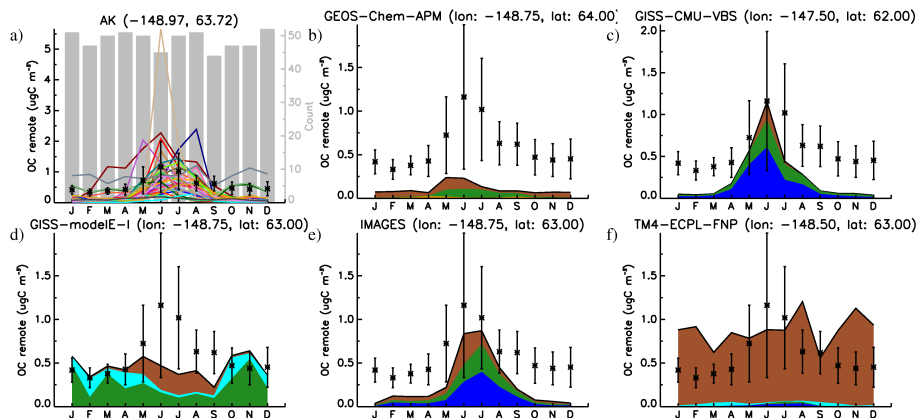
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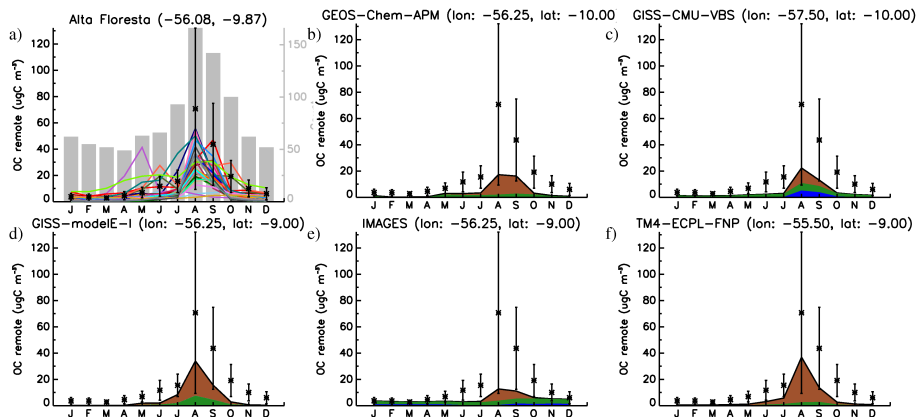
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**Fig. 31.** Same as in Fig. 25, for Alaska, USA (remote, years 2002–2006). For the chemical composition in panels (b–f), brown is tPOC, green is trSOC, cyan is mPOC, blue is ntrSOC, and orange is MSA.

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**Fig. 32.** Same as in Fig. 25, for Alta Floresta, Brazil (remote, years 1992–2005). For the chemical composition in panels (b–f), brown is tPOC, green is trSOC, cyan is mPOC, blue is ntrSOC, and orange is MSA.

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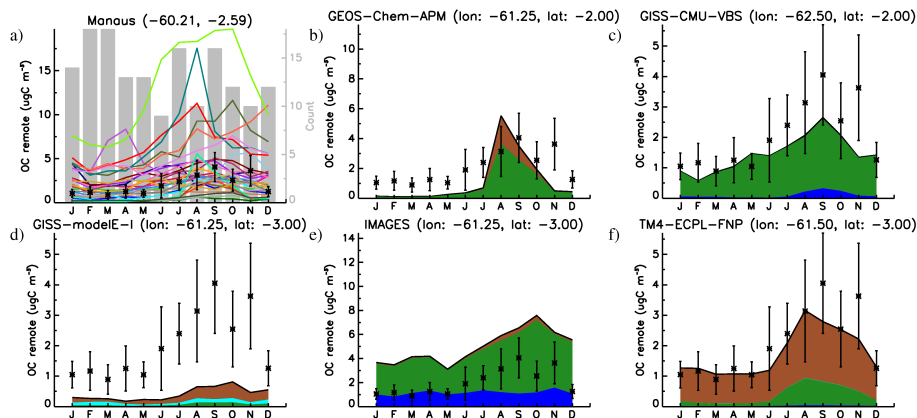
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**Fig. 33.** Same as in Fig. 25, for Manaus, Brazil (remote, years 2008–2010). For the chemical composition in panels (b–f), brown is tPOC, green is trSOC, cyan is mPOC, blue is ntrSOC, and orange is MSA.

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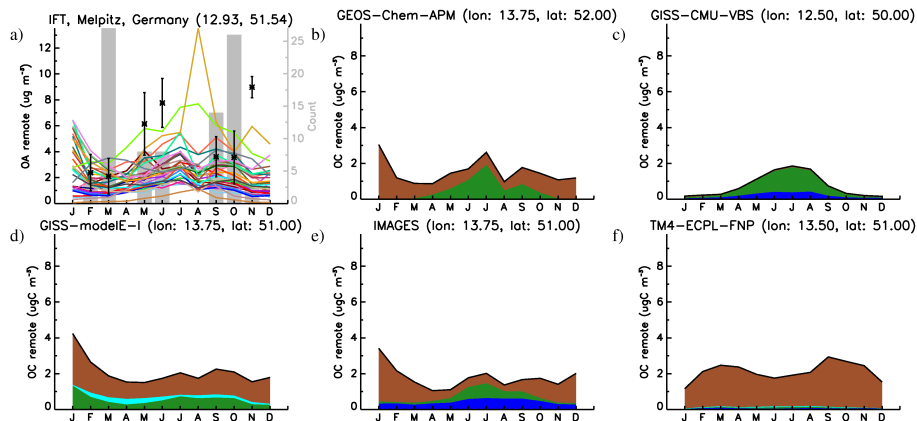
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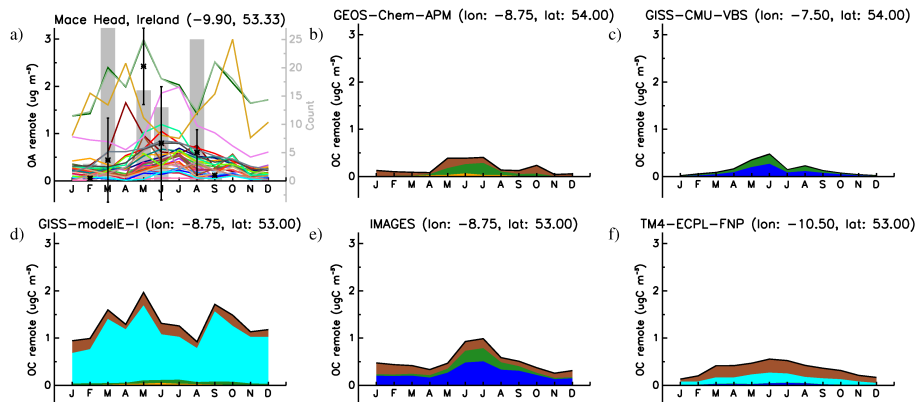
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**Fig. 34.** Same as in Fig. 25, for Melpitz, Germany (remote, years 2008 and 2009). For the chemical composition in panels (b–f), brown is tPOC, green is trSOC, cyan is mPOC, blue is ntrSOC, and orange is MSA.

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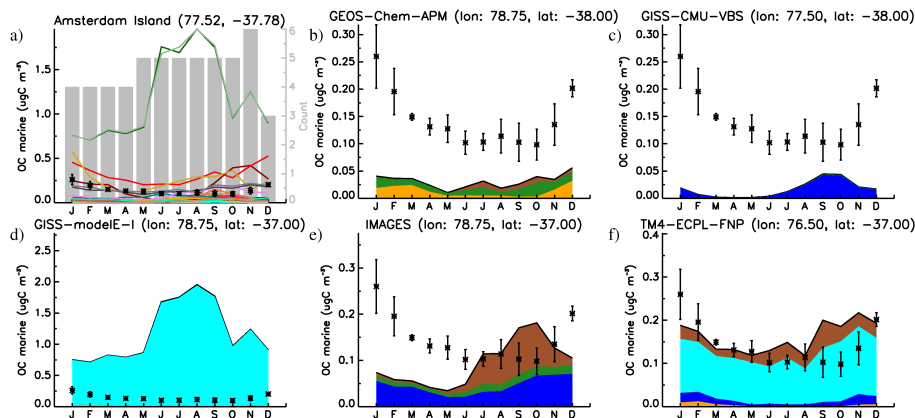
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**Fig. 35.** Same as in Fig. 30, for Mace Head, Ireland (remote, years 2002, 2008 and 2009). For the chemical composition in panels (b–f), brown is tPOC, green is trSOC, cyan is mPOC, blue is ntrSOC, and orange is MSA.

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**Fig. 36.** Same as in Fig. 25, for Amsterdam Island, Indian Ocean (marine, years 2003–2007). For the chemical composition in panels (b–f), brown is tPOC, green is trSOC, cyan is mPOC, blue is ntrSOC, and orange is MSA.

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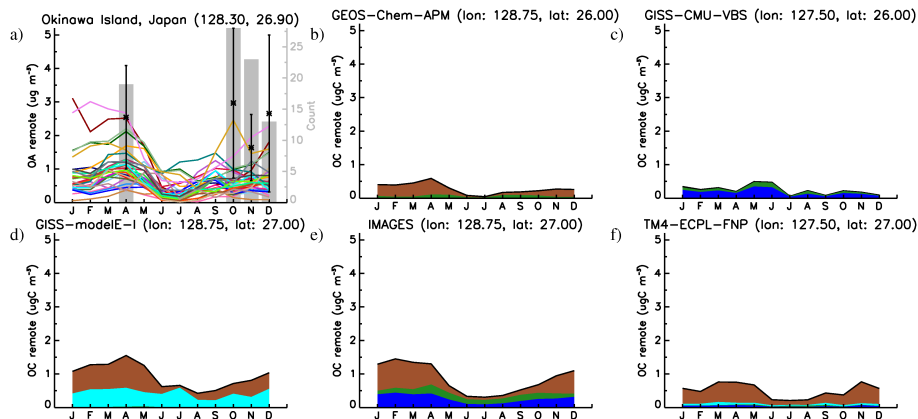
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**Fig. 37.** Same as in Fig. 30, for Okinawa, Japan (marine, years 2003 and 2004). For the chemical composition in panels (b–f), brown is tPOC, green is trSOC, cyan is mPOC, blue is ntrSOC, and orange is MSA.