

1 **Air-sea exchange and gas-particle partitioning of polycyclic aromatic hydrocarbons in**
2 **the Mediterranean**

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12
13 **Abstract**

14 Polycyclic aromatic hydrocarbons concentration in air of the central and eastern
15 Mediterranean in summer 2010 was 1.45 (0.30-3.25) ng m⁻³ (sum of 25 PAHs), with 8 (1-17)
16 % in the particulate phase, almost exclusively associated with particles <0.25 µm. The total
17 deposition flux of particulate PAHs was 0.35-0.80 µg m⁻² year⁻¹. The diffusive air-sea
18 exchange fluxes of fluoranthene and pyrene were mostly found net-depositional or close to
19 phase equilibrium, while retene was net-volatilisation in a large sea region. Regional fire
20 activity records in combination with box model simulations suggest that seasonal
21 depositional input of retene from biomass burning into the surface waters during summer is
22 followed by an annual reversal of air-sea exchange, while inter-annual variability is
23 dominated by the variability of the fire season. One third of primary retene sources to the sea

24 region in the period 2005-2010 returned to the atmosphere as secondary emissions from
25 surface seawaters. It is concluded that future negative emission trends or interannual
26 variability of regional sources may trigger the sea to become a secondary PAH source through
27 reversal of diffusive air-sea exchange.

28

29 **Capsule:** Polycyclic aromatic hydrocarbons phase distributions in marine aerosols, direction
30 of air-sea exchange and open fires as a possible source characterised in the Mediterranean

31

32 **Keywords:** polycyclic aromatic hydrocarbons, long-range transport, air-sea exchange, open
33 fires

34

35 **1. Introduction**

36 The marine atmospheric environment is a receptor for polycyclic aromatic hydrocarbons
37 (PAHs) which are advected from combustion sources on land (power plants, biomass
38 burning, road transport, domestic heating). Marine sources may be significant near transport
39 routes (ship exhaust). Long-range transport from urban and industrial sources on land are the
40 predominant sources of PAHs in the Mediterranean (Masclet et al., 1988; Tsapakis et. al,
41 2003 and 2006; Tsapakis and Stephanou, 2005a). A number of PAHs are semivolatile (vapour
42 pressures at 298 K in the range 10^{-6} - 10^{-2} Pa) and, hence partition between the gas and
43 particulate phases of the atmospheric aerosol, influenced by temperature, particulate phase
44 chemical composition and particle size (Keyte et al., 2013). Upon deposition to surface water
45 PAHs partition between the aqueous and particulate (colloidal and sinking) phases and may
46 bioaccumulate in marine food chains (Lipiatou and Saliot, 1991; Dachs et al., 1997; Tsapakis

47 et. al, 2003; Berrojalbiz et al., 2011). They were also found enriched in the sea-surface
48 microlayer relative to subsurface water (Lim et al., 2007; Guitart et al., 2010). Semivolatile
49 PAHs may be subject to re-volatilisation from the sea surface (reversal of air-sea exchange),
50 similar to chlorinated semivolatile organics (Bidleman and McConnell, 1995), in case high
51 concentrations in surface water would build up. This had been predicted by
52 multicompartamental modelling for 2-4 ring PAHs for polluted coastal waters and also the
53 open ocean (Greenfield and Davis, 2005; Lammel et al., 2009a) and was indeed observed in
54 coastal waters off the northeastern United States (Lohmann et al., 2011). Field studies in the
55 open sea found net-deposition to prevail whenever determined (e.g. Tsapakis et al., 2006;
56 Balasubramanian and He, 2010; Guitart et al., 2010; Castro-Jiménez et al., 2012; Mai, 2012).
57 However, some 3-4 ring parent PAHs, among them fluorene (FLU), fluoranthene (FLT) and
58 pyrene (PYR), were reported to be close to phase equilibrium in the Mediterranean, Black
59 and North seas (Castro-Jiménez et al., 2012; Mai, 2012), and net volatilisation of FLT and
60 PYR was observed in the open southeastern Mediterranean Sea in spring 2007 (Castro-
61 Jiménez et al., 2012).

62 The aim of the measurements on board RV Urania was to investigate levels of organic
63 pollutants in summer in Mediterranean air and gain insights about sources and phase
64 partitioning in the aerosol of these substances. This study is on the cycling of PAHs in the
65 marine atmosphere.

66

67 **2. Methods**

68 **2.1 Sampling**

69 Samples were taken during the RV Urania cruise, 27 August – 12 September 2010 (see
70 Supplementary Material (SM), Fig. S1). PAHs were collected in the gaseous and particulate
71 phases using high volume samplers (Digitel) equipped with one glass fibre filter (Whatman)
72 and one polyurethane foam (PUF) plug (Gumotex Břeclav, density 0.030 g cm⁻³, 50 mm
73 diameter, cleaned by extraction in acetone and dichloromethane, 8 h each, placed in a glass
74 cartridge) in series. Particle size was classified in the particulate phase using high-volume
75 filter sampling ($F = 68 \text{ m}^3 \text{ h}^{-1}$, model HVS110, Baghirra, Prague) and low-volume impactor
76 sampling ($F = 0.54 \text{ m}^3 \text{ h}^{-1}$, Sioutas 5-stage cascade, PM₁₀ inlet, cutoffs 2.5, 1.0, 0.5, 0.25 μm
77 of aerodynamic particle size and back-up filter, impaction on quartz fibre filters (QFF), SKC
78 Inc., Eighty Four, USA, sampler Baghirra PM₁₀₋₃₅). In total 15 high-volume filter samples,
79 exposed 8-36 h (230-1060 m³ of air), and 3 low-volume impactor samples, exposed 5 d, were
80 collected. Water sampling was performed using the stainless steel ROSETTE active sampling
81 device equipped with 24 Niskin bottles (volume of 10 l) deployed in water at 1.5 m depth for
82 surface water sampling.

83

84 PAH sampling on GFF and in PUF can be subject to losses related to oxidation of sorbed
85 PAH by ozone (Tsapakis and Stephanou, 2003). This artifact is species-specific and the more
86 pronounced the higher the ozone concentration and the longer the sampling time. Among the
87 PAHs addressed benzo(a)pyrene and pyrene have been identified as particularly vulnerable to
88 oxidation. Based on such sampling artefact quantification studies (Tsapakis and Stephanou,
89 2003; Galarneau et al., 2006) and ozone levels (Table 1a) and sampling times (Table S1) we
90 expect that total PAHs are underestimated by up to 50% in the gas-phase and by up to 25% in
91 the particulate phase.

92 With the aim to characterize the potential influences of ship-bourne emissions on the
93 samples, passive air samplers with PUF disks were exposed at 5 different locations on board
94 during 16 days. The PAH levels of these samples indicated that ship-based contamination was
95 negligible.

96

97 **2.2 PAHs analyses and quality assurance**

98 For PAH analysis all samples were extracted with dichloromethane in an automatic extractor
99 (Büchi B-811). Surrogate recovery standards (D8-naphthalene, D10-phenanthrene, D12-
100 perylene) were spiked on each PUF and GFF prior to extraction. The volume was reduced
101 after extraction under a gentle nitrogen stream at ambient temperature, and fractionation
102 achieved on a silica gel column.

103 The first portion of the extract was fractionated on a silica column (5 g of silica 0.063 – 0.200
104 mm, activated 12 h at 150°C). The first fraction (10 mL n-hexane) containing aliphatic
105 hydrocarbons was discarded. The second fraction (20 mL DCM) containing PAHs was
106 collected and then reduced by stream of nitrogen in a TurboVap II (Caliper LifeSciences,
107 USA) concentrator unit and transferred into an insert in a vial. Terphenyl was used as syringe
108 standard, final volume was 200 µL. GC-MS analysis was performed on a 6890N GC
109 equipped with a 60m x 0.25mm x 0.25µm DB5-MS column (Agilent J&W, USA) coupled to
110 5973N MS (Agilent, USA). The MS was operated in EI+ mode with selected ion recording
111 (SIR).The targeted compounds are the 16 EPA priority PAHs (i.e., naphthalene (NAP),
112 acenaphthylene (ACY), acenaphthene (ACE), fluorene (FLN), phenanthrene (PHE),
113 anthracene (ANT), fluoranthene (FLT), pyrene (PYR), benzo(a)anthracene (BAA), chrysene
114 (CHR), benzo(b)fluoranthene (BBF), benzo(k)fluoranthene (BKF), benzo(a)pyrene (BAP),

115 indeno(123cd)pyrene (IPY), dibenzo(ah)anthracene (DBA), benzo(ghi)perylene (BPE)), 10
116 more parent PAHs (i.e., benzo(ghi)fluoranthene (BGF), cyclopenta(cd)pyrene (CPP),
117 triphenylene (TPH), benzo(j)fluoranthene (BJF), benzo(k)fluoranthene (BKF),
118 benzo(e)pyrene (BEP), perylene (PER), dibenz(ac)anthracene (DCA), anthranthrene (ATT),
119 and coronene (COR)), and one alkylated PAH, retene (RET). The injection volume was 1 μ L.
120 Terphenyl was used as internal standard.

121 Field blank values, b , were gained from GFFs and PUFs manipulated in the field, as far as
122 possible identical to the samples, except without switching the high-volume sampler on. No
123 QFF field blank was taken for impactor sampling. As no PAHs were detected in the stages
124 corresponding to 2.5-10 μ m (all PAHs < limit of detection in all such samples), instead the
125 mean of values of the QFF substrates of the 2 uppermost impactor stages (in total 6) was
126 taken. The respective b value was subtracted from sample values. The limit of quantification
127 needs to take the accuracy of the blank level into account. In lack of a measure for the
128 variation of the field blank, the relative standard deviation (SD) of field blanks from earlier
129 field campaigns, (σ_c/b_c) , on a high-mountain site (high-volume sampling summer 2007, $n = 5$;
130 Lammel et al., 2009b) and in the Mediterranean (impactor sampling summer 2008, $n = 6$;
131 Lammel et al., 2010a) was used ($\sigma = (\sigma_c/b_c) \times b$). The identical samplers, sampling and
132 analysis protocols for all analytes had been applied. Values below the sum of the field blank
133 value (from this campaign) and 3 relative SDs of the field blank values (from the previous
134 campaigns) were considered <LOQ ($LOQ = b + 3\sigma$). NAP and ACY were excluded from the
135 data set, because of the lack of blank values. The field blank values of most other analytes
136 were below instrument LOQ in high-volume PUF and GFF samples. However, higher field
137 LOQs, up to (6-25) pg m^{-3} (according to sampled volume of air) resulted for ANT, PYR and

138 RET, and up to (45-180) pg m^{-3} for ACE, FLN, PHE and FLT in PUF. Field LOQs of PAHs in
139 impactor QFF samples were below instrumental LOQ for most substances, but in the range
140 (8-15) pg m^{-3} for ACE, ANT, and FLT, $\approx 55 \text{ pg m}^{-3}$ for FLN, and 120-140 pg m^{-3} for NAP and
141 PHE.

142 The instrument limit of quantification (LOQ), which is based on the lowest concentration of
143 calibration standards used, was 0.5 ng, corresponding to 0.5-2.5 pg m^{-3} for high-volume
144 samples, $\approx 8 \text{ pg m}^{-3}$ for impactor samples, 6-10 pg m^{-3} for semivolatile PAHs determined in
145 passive air samples and up to 200 pg m^{-3} for non volatile PAHs in passive air samples.

146 Water samples (2-2.5 L) were extracted immediately after their collection using solid phase
147 extraction on C_{18} Empore discs using a vacuum manifold device. Disks were stored closed in
148 glass vials in a freezer and transported to the processing laboratory, PAHs were eluted from
149 disks using 40 mL of dichloromethane. The above listed PAHs were analysed on GC/MS
150 (Agilent GC 6890N coupled to an Agilent single quadrupole MS 5973N operating in electron
151 impact ionisation mode). LOQ was 0.1 ng L^{-1} .

152

153 **Other trace constituents and meteorological parameters**

154 Ozone was measured with an absorption method (Teledyne-API model 400A UV) on the top
155 deck (10 m above sea surface). Meteorological parameters (air temperature, humidity, wind
156 direction and velocity) and oceanographic parameters were determined aboard.

157

158 **2.3 Models of gas-particle partitioning**

159 The data set (15 high-volume samples of separate gas and particulate phase concentrations) is
160 used to test gas-particle partitioning models for semivolatile organics in terms of the organics'

161 mass size distribution and size dependent particulate matter (PM) composition. The models
162 assume different processes to determine gas-particle partitioning, i.e. an adsorption model
163 (Junge-Pankow; Pankow, 1987), and two absorption models (i.e. K_{OA} models; Finizio et al.,
164 1997; Harner and Bidleman, 1998). Absorption is into particulate OM. Adsorption to soot is a
165 significant gas---particle partitioning processes for PAHs, but no soot data or PM chemical
166 composition data are available. We, therefore, refrain from testing dual adsorption and
167 absorption models (e.g. Lohmann and Lammel, 2004). Particulate mass fraction, θ , and
168 partitioning coefficient, K_p , are defined by the concentrations in the 2 phases:

169

$$170 \quad \theta = c_p / (c_p + c_g)$$

171

$$172 \quad K_p = c_p / (c_g \times c_{TSP}) = \theta / [(1 - \theta) \times c_{TSP}]$$

173

174 with c_p and c_g in units of ng m^{-3} , c_p representing the whole particle size spectrum.

175 Different models describe different processes to quantify differences in ad- and absorption
176 between compounds. The Junge-Pankow model uses the vapour pressure of the subcooled
177 liquid p_L^0 , $\theta = c_j S / (p_L^0 + c_j S)$, (data taken from Lei et al., 2002), c_j should be approximately
178 171 Pa cm for PAHs (Pankow, 1987). The aerosol particle surface, S , was not measured and
179 a typical value for maritime aerosols is adopted instead ($4.32 \times 10^{-7} \text{ cm}^{-1}$; Jaenicke, 1988).
180 Harner and Bidleman, 1998, use the $\log K_{OA}$ and f_{OM} : $\log K_p = \log K_{OA} + \log f_{OM} - 11.91$; and
181 Finizio et al., 1997, uses only the K_{OA} as predictor (data taken from Ma et al., 2010): $\log K_p =$
182 $0.79 \times \log K_{OA} - 10.01$. The range of the fraction of OM used here is based on Putaud et al.,
183 2004 (16% lower limit) and Spindler et al., 2012 (45% upper limit).

184

185 **2.4 Air-sea diffusive mass exchange calculations**

186 State of phase equilibrium is addressed by fugacity calculation, based on the Whitman two-
187 film model (Liss and Slater, 1974; Bidleman and McConnell, 1995). The fugacity ratio (FR)
188 is calculated as:

189

$$190 \text{ FR} = f_a/f_w = C_aRT_a / (C_wH_{\text{Tw,salt}})$$

191

192 with gas-phase concentration C_a (ng m^3), dissolved aqueous concentration C_w (ng m^3),
193 universal gas constant R ($\text{Pa m}^3 \text{ mol}^{-1} \text{ K}^{-1}$), water temperature and salinity corrected Henry's
194 law constant $H_{\text{Tw,salt}}$ ($\text{Pa m}^3 \text{ mol}^{-1}$), and air temperature T_a (K). C_w is derived from the bulk
195 seawater concentration:

196

$$197 C_w = C_{\text{bulk}} / (1 + K_{\text{POC}} C_{\text{POC}} + K_{\text{DOC}} C_{\text{DOC}})$$

198

199 with C_{POC} and C_{DOC} from Pujó-Pay et al., 2011, K_{POC} and K_{DOC} from Karickhoff et al, 1981,
200 Lüers and ten Hulscher 1996, Rowe et al, 2009, and Ma et al, 2010. Values $0.3 < \text{FR} < 3.0$ are
201 conservatively considered to not safely differ from phase equilibrium, as propagating from
202 the uncertainty of the Henry's law constant, $H_{\text{Tw,salt}}$, and measured concentrations (e.g., Bruhn
203 et al., 2003; Castro-Jiménez et al., 2012; Zhong et al., 2012). This conservative uncertainty
204 margin is also adopted here, while $\text{FR} > 3.0$ indicates net deposition and $\text{FR} < 0.3$ net
205 volatilisation. The diffusive air–seawater gas exchange flux (F_{aw} , $\text{ng m}^{-2} \text{ day}^{-1}$) is calculated

206 according to the Whitman two-film model (Bidleman and McConnell, 1995; Schwarzenbach
207 et al., 2003):

208

$$209 \quad F_{aw} = k_{ol} (C_w - C_a RT_a / H_{Tw,salt})$$

210

211 with air-water gas exchange mass transfer coefficient k_{ol} (m h^{-1}), accounting for resistances to
212 mass transfer in both water (k_w , m h^{-1}) and air (k_a , m h^{-1}), defined as

213

$$214 \quad 1/k_{ol} = 1/k_w + RT_a / (k_a H_{Tw,salt})$$

215

216 with $k_a = (0.2U_{10} + 0.3) * (D_{i,air} / D_{H_2O,air})^{0.61} \times 36$, $k_w = (0.45U_{10}^{1.64}) \times (Sc_i / Sc_{CO_2})^{-0.5} \times 0.01$. U_{10}

217 is the wind speed at 10 meter height above sea level (m s^{-1}), $D_{i,air}$ and $D_{H_2O,air}$ are the

218 temperature dependent diffusivities of substance i and H_2O in air, and Sc_i and Sc_{CO_2} are the

219 Schmidt numbers for substance i and CO_2 (see Bidleman and McConnell, 1995; Zhong et al.

220 2012; and references therein). U_{10} , T_a , T_w and air pressure are taken from the ship based

221 measurements.

222 **2.5 Non-steady state 2-box model**

223 The air–sea mass exchange flux of RET is simulated by a non-steady state zero-dimensional

224 model of intercompartmental mass exchange (Lammel, 2004). RET is selected, because of

225 the prevalence of one dominating source. This 2-box model predicts concentrations by

226 integration of two coupled ordinary differential equations that solve the mass balances for the

227 two compartments, namely the atmospheric marine boundary layer (MBL) and seawater

228 surface mixed layer. Processes considered in air are dry (particle) deposition, removal from

229 air by reaction with the hydroxyl radical, and air-sea mass exchange flux (dry gaseous
230 deposition), while in seawater export (settling) velocity, deposition flux from air, air-sea mass
231 exchange flux (volatilisation), and degradation (as 1st order process) are considered. All input
232 parameters are listed in the SM, Table S2.

233 Atmospheric depositions related to emissions from open fires are assumed to provide the
234 only source for seawater RET. These are available as daily time series for the East
235 Mediterranean domain (28-45°N, 8-30°E) through the fire-related PM_{2.5} emissions as
236 provided by the Global Fire Assimilation System (GFASv1.0; Kaiser et al., 2012) in
237 combination with an emission factor (207 mg RET in PM_{2.5} (kg fuel burnt)⁻¹; Schmidl et al.,
238 2008). The fire emissions are averaged over the domain and assumed to disperse within the
239 MBL only. This is justified due to the assumed underestimation of the fire related emissions
240 and ignorance of other (emission) sources. The 2-box model is run for the years 2005-2010,
241 for the east Mediterranean domain (28-45°N, 8-30°E) with a 1 h time resolution. Fluxes in
242 the range $F_{em} = (0.30 \pm 1.46) \text{ ng m}^{-2} \text{ h}^{-1}$ (positive defined upward) are simulated (using the
243 initially estimated parameter set, Table S2). GFAS uses global satellite observations of fire
244 radiative power to estimate daily dry matter combustion rates and fire emission fluxes. The
245 GFAS system partly corrects for observational gaps (e.g. due to cloud cover) and detects fires
246 in all biomes, except for very small fires (lower detection limit of around 100-1000 m²
247 effective fire area).

248

249 **2.6 Analysis of long-range advection of air**

250 Distributions of potential sources can be identified by inverse modelling using meteorological
251 input data (Stohl et al., 2003; Eckhardt et al., 2007). So-called retroplumes are generated

252 using operational weather prediction model data and a Lagrangian particle dispersion model,
253 FLEXPART (Stohl et al., 1998, 2005). Hereby, 50000 virtual particles per hour were
254 'released' and followed backwards in time for 5 days. The model output is a 3-D distribution
255 of residence time.

256

257 **3. Results and discussion**

258 **3.1 PAH concentrations in air and seawater**

259 The mean total (i.e., sum of gaseous and particulate) $\Sigma 25$ PAHs concentration is 1.45 ng m^{-3}
260 (time-weighted; 1.54 with values $< \text{LOQ}$ replaced by $\text{LOQ}/2$, see Table 1a), and ranged from
261 $0.30\text{-}3.25 \text{ ng m}^{-3}$. The spatial variability of PAH levels in the Mediterranean is large,
262 determined by long-range advection (Tsapakis and Stephanou, 2005a; Tsapakis et al., 2006).
263 The levels found in this study in the southeastern Mediterranean are for most substances
264 lower than found earlier (Table 2). In the ISS some PAHs are found somewhat higher than
265 previously measured i.e., FLT and PYR (in the gas-phase) and BAP and PER (in the
266 particulate phase). Due to a sampling artefact BAP and other particulate phase PAHs could be
267 underestimated by up to 25% (aforementioned, section 2.1). The seasonality of emissions and
268 the variability of advection or advection in combination with different cruise routes being
269 influenced differently by coastal or ship emission plumes can have a large influence and may
270 explain these differences. On the other hand, the duration of temporal averaging atmospheric
271 concentrations was similar across the various studies. Diagnostic ratios ($\text{BAA}/(\text{BAA}+\text{CHR})$,
272 $\text{FLT}/(\text{FLT}+\text{PYR})$; Dvorská et al., 2011) in some of the samples (No. 2, 4, 7, 8, and 15) reflect
273 the influence of traffic and industrial sources. We investigated the potential source
274 distribution of individual samples collected along the cruise (section 2.6) and found that

275 indeed maxima of PAH concentrations corresponded with air masses having resided over
 276 large urban areas, and, vice versa, low concentrations corresponded with air masses without
 277 apparent passage of such areas (illustrated in Fig. S5). This finding is supported by the ozone
 278 data i.e., 53 (47-65) ppbv during influence from urban areas but 37 (33-62) ppbv otherwise.
 279 It had been pointed out that the source distribution around the Mediterranean may cause a
 280 west-east gradient, leading to higher concentrations found in the Ionian Sea and Sicily region
 281 (ISS) than in the southeastern Mediterranean (SEM; Berrojalbiz et al., 2011). This gradient is
 282 somewhat reflected in our results, as levels in the ISS exceeded levels in the SEM (Table 2a).
 283 Most PAH concentrations in surface seawater were <LOQ, while FLT, PYR and RET were
 284 quantified in at least part of the samples (Table 1b). These observed seawater contamination
 285 levels are comparable to levels found in the region 2 and 1 decades ago (Lipiatou et al., 1997;
 286 Tsapakis et al., 2003). The concentrations near Crete (samples No. 7 and 8a) are very similar
 287 to those found in fall 2001 and winter-spring 2002 (Tsapakis et al., 2006; FLT = 0.15 (0.11-
 288 0.21) ng L⁻¹, PYR = 0.12 (0.07-0.17) ng L⁻¹.

289

290 Table 1. Concentrations of PAHs found in (a.) air (total, i.e. sum of gas and particulate
 291 phases, ng m⁻³) and (b.) seawater (total, i.e. sum of dissolved and particulate, ng L⁻¹) as time-
 292 weighted mean (min-max). n_{LOQ} = number of samples > LOQ (out of 15 air and 23 seawater
 293 samples). PAHs with concentrations <LOQ in all samples not listed. For calculation of mean
 294 values <LOQ were replaced by LOQ/2. Ozone levels are given, too (ppbv).

295 a)

	n _{LOQ}	mean (min-max)
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ACE	4	0.025 (<0.020–0.089)
FLN	10	0.137 (<0.030–0.396)
PHE	15	0.581 (0.144–1.41)
ANT	13	0.043 (0.008–0.22)
RET	14	0.016 (0.006–0.030)
FLT	15	0.262 (0.053–0.795)
PYR	15	0.203 (0.044–0.564)
BAA	15	0.01 (0.0014–0.031)
CHR	15	0.04 (0.012–0.092)
TPH	15	0.018 (0.007–0.032)
BBN	11	0.018 (0.001–<0.085)
BBF	15	0.021 (0.004–0.102)
BKF	14	0.012 (0.002–<0.085)
BAP	12	0.015 (0.001–<0.085)
BGF	15	0.021 (0.005–0.067)
CPP	7	0.012 (0.001–<0.085)
BJF	15	0.016 (0.002–0.079)
BEP	14	0.019 (0.004–0.088)
PER	7	0.012 (0.001–0.1)
IPY	7	0.022 (0.008–0.094)
BPE	6	0.02 (0.009–0.085)
COR	5	0.016 (0.002–0.1)

Σ25 PAHs		1.539 (0.44–4.694)
Ozone		42 (33 – 65)

296

297 b)

	n _{LOQ}	mean (min-max)
PHE	1	1.1
RET	12	0.1 (<0.1–0.5)
FLT	10	0.1 (<0.1–0.3)
PYR	7	0.2 (<0.2–0.9)

298

299 Table 2 Gaseous (a) and particulate (b) concentrations in air (time-weighted mean (min-max),
300 ng m⁻³) of selected PAH compared to other studies in the Ionian Sea and Sicily region (ISS)
301 and in the southeastern Mediterranean (SEM). For calculation of means values <LOQ were
302 replaced by LOQ/2. RV = research vessel cruise.

303 a)

	ISS		SEM			
	RV August- September 2010	RV June 2006, May 2007	RV August- September 2010	RV June 2006, May 2007	Finokalia September- October 2001, February, March and July 2002 ⁽¹⁾	Finokalia November 2000- February 2002 ⁽²⁾
	this study	Castro- Jiménez 2012	this study	Castro-Jiménez 2012	Tsapakis et al. 2006	Tsapakis & Stephanou 2005

FLN	0.18 (0.050– 0.40)	2.25 (1.27– 5.65)	0.11 (<0.027– 0.34)	0.69 (0.36– 1.23)	1.05 (0.15– 1.67)	1.8 (0.2–5.7)
PHE	0.59 (0.14– 1.10)	7.00 (3.52– 15.45)	0.51 (0.14– 1.41)	3.94 (2.50– 6.35)	4.78 (1.75– 7.78)	7.3 (1.5–27.7)
ANT	0.064 (<0.021– 0.10)	0.37 (0.18– 0.55)	0.067 (<0.013– 0.22)	0.20 (0.16– 0.30)	0.61 (0.12– 1.31)	0.9 (0.1–4.5)
FLT	0.15 (0.053– 0.31)	0.05 (0.02– 0.07)	0.15 (0.061– 0.37)	0.007 (0.003– 0.011)	0.82 (0.12– 1.69)	1.8 (0.07–6.0)
PYR	0.15 (0.044– 0.29)	0.04 (0.02– 0.06)	0.19 (0.058– 0.56)	0.006 (0.003– 0.009)	0.65 (0.14– 0.97)	0.9 (0.1–2.8)
CHR	0.013 (0.0071– 0.021)	0.09 (0.03– 0.23)	0.018 (0.012– 0.037)	0.03 (0.02– 0.05)	0.18 (0.06– 0.33)	0.2 (<0.001–0.6)
Sum of 6 PAHs	1.1	9.8	1.0	4.9	8.1	12.9

304

305 b)

	ISS		SEM		
	RV August- September 2010	RV June 2006, May 2007	RV August- September 2010	RV June 2006, May 2007	Finokalia November 2000- February 2002
	this study	Castro-Jiménez 2012	this study	Castro- Jiménez 2012	Tsapakis & Stephanou 2005

FLN	<0.92 (<0.60– <1.1)	0.001 (0.0009– 0.002)	<0.66 (<0.33– <1.4)	0.0013 (0.0011– 0.0016)	0.02 (<0.001– 0.01)
PHE	<1.9 (<1.2– <2.3)	0.06 (0.01– 0.12)	<1.6 (<0.66– <2.7)	0.04 (0.01– 0.13)	0.05 (0.004–0.2)
ANT	<0.21 (<0.14– <0.26)	0.007 (0.0009– 0.012)	<0.16 (<0.07– <0.32)	0.009 (0.0007– 0.023)	0.004 (<0.001– 0.02)
FLT	<0.85 (<0.56– <1.0)	0.099 (0.01– 0.19)	<0.62 (<0.30– <1.3)	0.049 (0.01– 0.12)	0.1 (0.04–0.2)
PYR	<0.11 (<0.070– <0.13)	0.109 (0.016– 0.216)	<0.08 (<0.044– <0.16)	0.057 (0.012– 0.142)	0.04 (0.01–0.01)
BAA	0.0065 (<0.0018– 0.025)	0.013 (0.006– 0.023)	0.0030 (<0.0006– 0.0080)	0.018 (0.004– 0.046)	0.03 (0.003–0.1)
CHR	0.021 (0.0030– 0.076)	0.04 (0.01– 0.08)	0.010 (0.0033– 0.020)	0.043 (0.012– 0.101)	0.1 (0.02–0.3)
BBF	0.028 (<0.0018– 0.010)	0.029 (0.012– 0.045)	0.014 (0.0042– 0.033)	0.033 (0.010– 0.060)	0.04 (<0.001–0.2)
BKF	0.015 (<0.0018– 0.057)	0.015 (0.005– 0.027)	0.0065 (0.0018– 0.015)	0.089 (0.005– 0.333)	0.04 (<0.001–0.2)

BAP	0.016 (<0.0009– 0.072)	0.009 (0.04– 0.016)	0.0052 (<0.0011– 0.0098)	0.034 (0.005– 0.081)	0.02 (0.01–0.05)
BJF	0.021 (<0.0018– 0.079)	0.015 (0.014– 0.016)	0.011 (0.0023– 0.031)	0.010 (0.008– 0.011)	-
BEP	0.023 (<0.0018– 0.088)	0.03 (0.02– 0.05)	0.011 (0.0035– 0.025)	0.046 (0.017– 0.093)	0.04 (0.01–0.1)
PER	0.0028 (<0.00096 –0.011)	0.002 (0.0005– 0.004)	0.00085 (<0.0006– 0.0021)	0.026 (0.0001– 0.068)	0.004 (<0.001– 0.01)
IPY	0.020 (<0.00096 –0.094)	0.018 (0.006– 0.032)	0.0043 (<0.00052– 0.019)	0.009 (0.002– 0.013)	0.03 (0.009–0.2)
BPE	<0.0014 (<0.00096 –0.0018)	0.026 (0.017– 0.042)	0.0067 (<0.00052– 0.020)	0.081 (0.012– 0.210)	0.03 (0.010–0.09)
Sum of 15 PAHs	0.15	1.06	0.06	0.54	0.54

306 ⁽¹⁾ months Sep and Oct 2001, Feb, Apr and May 2002. No particulate data reported.

307 ⁽²⁾ 24h per month between Feb 2000 and Feb 2002

308

309 **3.2 Gas-particle partitioning**

310 Only a small mass fraction of the total, $\theta = 0.08$, is found in the particulate phase, confirming
311 earlier findings from remote sites in the region (Tsapakis and Stephanou, 2005a; Tsapakis et
312 al., 2006; Table 3c). The particulate mass fraction, θ , of four semivolatile PAHs varied
313 considerably along track (see SM Fig. S2). θ is thought to be strongly influenced by
314 temperature and doubling per 13 K cooling was found in a Mediterranean environment
315 (Lammel et al., 2010b) apart from PM composition. We refrain from an exploration of the
316 vapour pressure (p_{L0}) dependence of θ (or K_p): a low time resolution implies lack of
317 representativeness of the temperature measurement for the phase change (Pankow and
318 Bidleman, 1992), non-equilibrium conditions cannot be excluded (but are likely as a
319 consequence of time resolution; Hoff et al., 1998) and supporting physical and chemical
320 aerosol parameters, necessary to relate are lacking.

321 of only little temperature variation during the cruise ($T_{\text{mean_sample}} = 21\text{-}27^\circ\text{C}$, Fig. S2). For
322 similar temperatures higher θ values had been observed at sites on the region influenced by
323 urban and industrial sources (Mandalakis et al., 2002; Tsapakis and Stephanou, 2005b; Akyüz
324 and Cabuk, 2010), which is probably related to the influence of higher organic and soot PM
325 mass fractions. Gas-particle partitioning models (Table 3) underpredict θ , except the Finizio
326 et al., 1997, model for one substance, TPH. θ predicted by the Junge-Pankow (JP) model
327 comes closest. A number of semivolatile PAHs could not be included in this test of gas-
328 particle partitioning models as concentrations in either the gas-phase (CPP, BBF, BJF), or the
329 particulate phase (FLT, PYR, BBN) did not exceed LOQ or no insufficient input data were
330 available (BBF). The neglect of adsorption to soot, not covered by the gas-particle
331 partitioning models tested, may explain at least part of the underprediction (Lohmann and
332 Lammel, 2004). Due to the lack of OC/EC data an extended examination is not possible.

333 In size-segregated samples particulate PAH mass was almost exclusively found in the size
 334 fraction <0.25 μm a.e.d. (<LOQ in the other stages, except 0.002 ng m^{-3} CPP in the size
 335 fraction corresponding to 0.5-1.0 μm ; S2.1, Table S4). Most particulate phase PAHs, 40%,
 336 have been found associated with particles <0.5 μm out of 5 size ranges in the marine
 337 background aerosol of the sea region (coast of Crete, November 1996 – June 1997; Kavouras
 338 and Stephanou, 2002). At continental sites in central and southern Europe mass median
 339 diameters of PAHs were found to be in the accumulation range, mostly 0.5-1.4 μm (Schnelle
 340 et al., 1995; Kiss et al., 1998; Lammel et al., 2010b and 2010c), but also a second, coarse
 341 mode was found (up to 2.4 μm ; Chrysikou et al., 2009).

342

343 Table 3. Gas-particle partitioning of selected PAHs (mean \pm sd (median)), observed and
 344 predicted by the models Junge-Pankow, 1987 (JP), Harner and Bidleman, 1998 (HB), and
 345 Finizio et al., 1997 (F), expressed as (a) particulate mass fraction, θ , and (b) log K_p of this
 346 study.

347 a)

	Observed	JP	HB	F
BAA	0.51 \pm 0.28 (0.47)	0.18 \pm 0.07 (0.18)	0.08 – 0.20	0.18
TPH	0.27 \pm 0.13 (0.26)	0.24 \pm 0.10 (0.24)	0.23 – 0.46	0.37
CHR	0.35 \pm 0.15 (0.35)	0.31 \pm 0.13 (0.32)	0.09 – 0.21	0.19
BBF	0.88 \pm 0.40 (0.94)	0.91 \pm 0.40 (0.97)	0.49 – 0.73	0.59

348

349 b)

	Observed	JP	HB	F
BAA	-1.28 ± 1.00 (-0.96)	-1.97 ± 1.14 (-1.84)	-2.43 – -1.98	-1.89
TPH	-1.77 ± 1.27 (1.45)	-1.80 ± 1.07 (-1.63)	-1.91 – -1.46	-1.48
CHR	-1.59 ± 1.18 (1.34)	-1.65 ± 1.01 (-1.46)	-2.41 – -1.96	-1.87
BBF	-0.94 ± 0.19 (-0.24)	-0.52 ± 0.66 (-0.74)	-1.41 – -0.96	-1.08

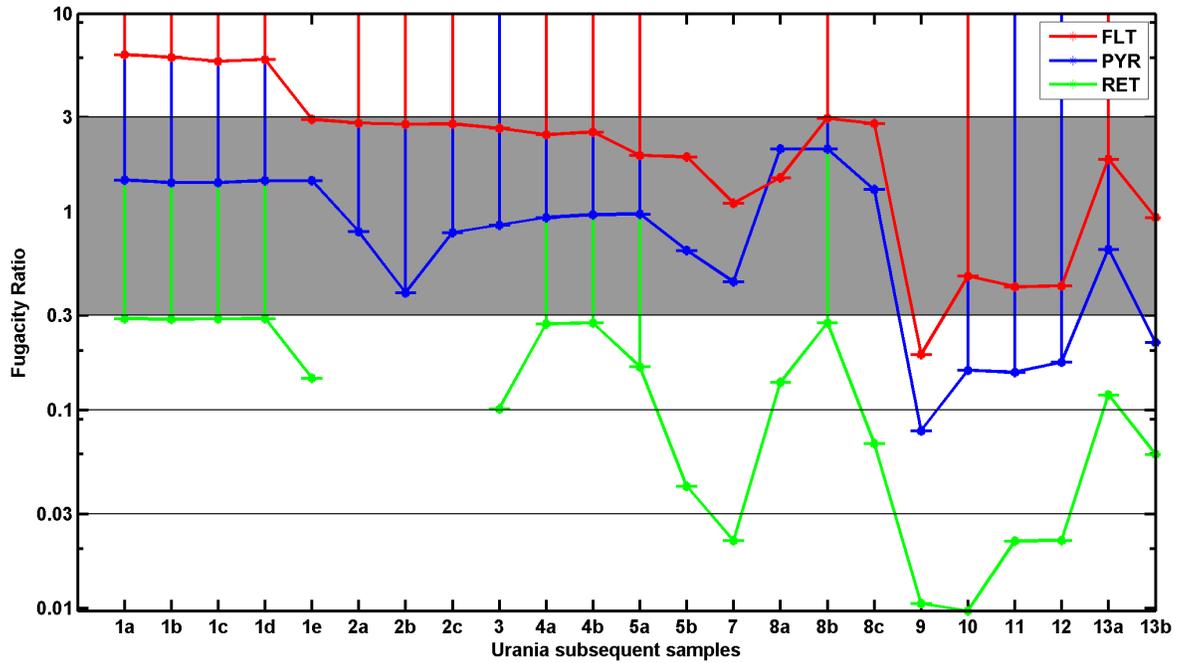
350

351 3.3 Fugacity ratio and air-sea exchange flux

352 Fugacity ratios (Fig. 1a) and vertical fluxes (Fig. 1b) could be quantified for FLT, PYR and
 353 RET. The uncertainty window of $FR = f_a/f_w = 0.3 - 3.0$ is based on the uncertainty of $H_{Tw,salt}$.
 354 Values $FR > 3.0$ indicate net deposition, $FR < 0.3$ indicate net volatilisation. For RET both
 355 water and air concentrations of sample No. 2 were <LOQ. Transfer coefficients were $k_w \ll$
 356 k_a .

357 Fig. 1. Air-sea exchange, (a) fugacity ratios $FR = f_a/f_w$ (volatilisation > 3 , deposition < 0.3 ,
 358 grey area insignificant deviation from phase equilibrium) and (b) flux F_{aw} ($ng\ m^{-2}\ d^{-2}$;
 359 volatilisation > 0 , deposition < 0) of FLT, PYR and RET along the cruise of RV Urania. Error
 360 bars indicate sea water concentration $C_w < LOQ$. The x-axis depicts the correspondence of
 361 sequential pairs of air samples (1-13) and water samples (a-e).

362 a.



363

364 b.



365

366

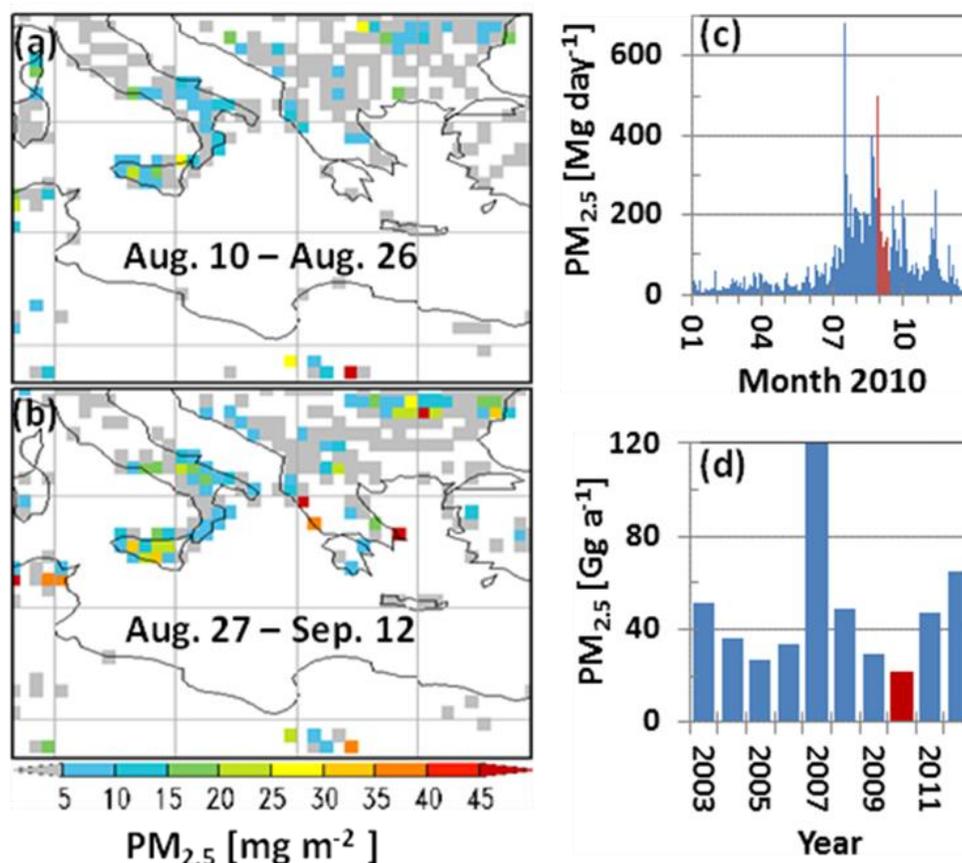
367 FLT and PYR were found to be close to phase equilibrium, with most of the FR values within
 368 the uncertainty range, one sample (No. 1) indicating deposition of FLT and one or two (No. 9
 369 and 13) indicating volatilisation of FLT and PYR, respectively. In comparison with earlier

370 observations of FLT and PYR air-sea exchange in the SEM in 2001-02 and 2007 (Tsapakis et
371 al., 2006; Castro-Jiménez et al., 2012) and considering spatial and temporal variabilities no
372 trend, in particular no reversal of air-sea exchange is indicated. This comparison is detailed in
373 the SM, S2.2.1. RET, however, is found net-volatilisation throughout most of the cruise
374 (Fig. 1). Among the highest fluxes ($> 50 \text{ ng m}^2 \text{ d}^{-1}$) are some samples with very low FR,
375 < 0.03 . Fugacity of RET from water is supported by its Henry's law coefficient ($11 \text{ Pa m}^3 \text{ mol}^{-1}$
376 1 at 298 K) which is higher than for CHR ($0.53 \text{ Pa m}^3 \text{ mol}^{-1}$) and FLT ($2.0 \text{ Pa m}^3 \text{ mol}^{-1}$). RET
377 is commonly considered as biomarker for coniferous wood combustion (Ramdahl, 1983). A
378 decrease in wildfires could explain the suspected RET volatilisation. Integrated over the
379 domain and the year 2010, fires released 7.2 PJ fire radiative energy, which translates into
380 around 22.2 Gg of $\text{PM}_{2.5}$ emitted (Fig. 2). Compared to the $\text{PM}_{2.5}$ emissions of the years 2003
381 to 2012, the year 2010 had the lowest emissions, equivalent to 46% of the 2003-2012 mean,
382 and only 18% of the peak emissions of the year 2007 (Fig. 2d). As typical for the East
383 Mediterranean region, the fire season in 2010 started by the end of June and ended by early
384 October. The Urania cruise measurements took place between 27.8. and 12.9., i.e. towards the
385 end of the main burning season (Fig. 2c). During the first half of the Urania cruise,
386 widespread fire activity was observed in the entire domain, with most intense fires occurring
387 in Southern Italy, Sicilia and along the East coast of the Adriatic and the Ionian Sea (notably
388 in Albania and Greece) (Fig. 2a).

389

390 Fig. 2. Spatial pattern of fire-related $\text{PM}_{2.5}$ emissions (Global Fire Assimilation System
391 GFASv1.0; Kaiser et al., 2012) for the East Mediterranean ($28-45^\circ\text{N}/8-30^\circ\text{E}$), (a) time
392 integral of August 10-26, (b) time integral of August 27 - September 12, 2010, given as sum

393 over each period in mg m^{-2} . Areas with no observed fire activity are displayed in white.
394 Temporal pattern of domain-integrated (c) daily total $\text{PM}_{2.5}$ emissions over 2010 (c) and
395 yearly total $\text{PM}_{2.5}$ emissions over 2003 to 2012. Labelled in red is (c) the the period of the
396 Urania cruise (27 August – 11 September 2010) (d) and the year 2010.



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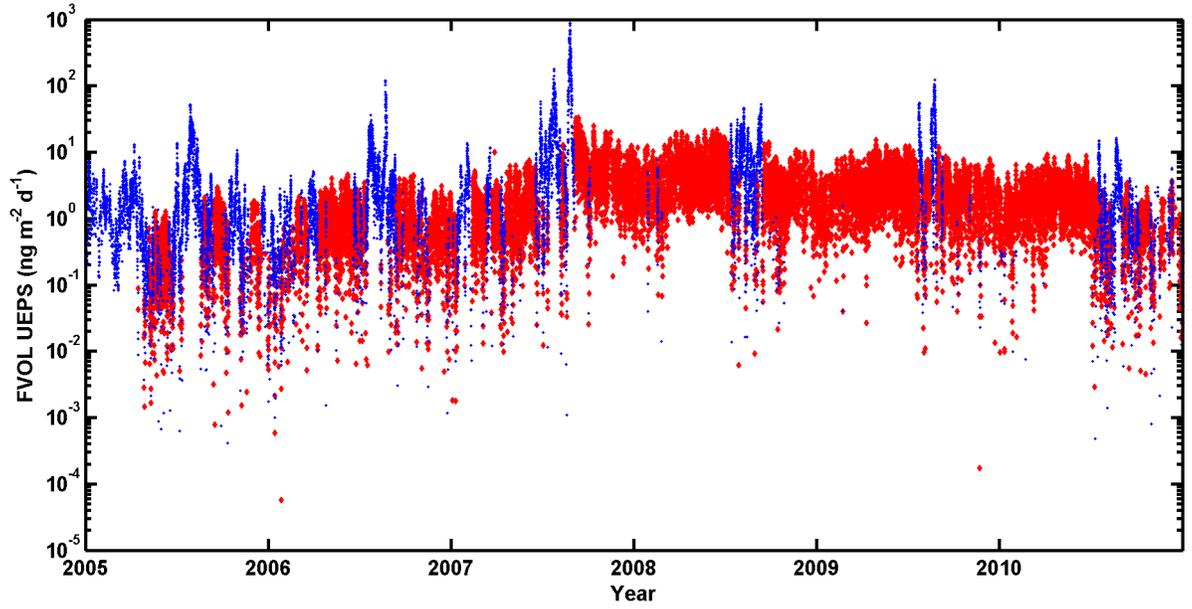
399 The hypothesis that seasonal depositional input of RET into the surface waters during the fire
400 season (summer) triggers reversal of diffusive air-sea exchange, at least in the year 2010, are
401 tested by box model (sections 2.5 and S1.3) runs. Two scenarios are considered, an 'Initially
402 Estimated Parameter Set' (IEPS) representing mean values for environmental parameters, and
403 an 'Upper Estimate Parameter Set' (UEPS) which represents realistic environmental
404 conditions favouring seawater pollution (SM, Table S3). Simulated diffusive air-sea exchange

405 flux, F_{aw} , during 2005-2010 initialised by the UEPS is shown in Fig. 3a and by the IEPS in
406 the SM, Fig. S3, and during the observations (cruise of RV Urania, 27.8.-9.9.2010) initialised
407 by the UEPS in Fig. 3b.

408 The model confirms the hypothesis that seasonal depositional input of RET into the surface
409 waters during the fire season (July-September, typically in the range $F_{aw} = 10^{-2}$ - 10^1 $\text{ng m}^{-2} \text{d}^{-1}$
410 under IEPS) is followed by a period of prevailing flux reversal, typically $F_{aw} = 10^{-2}$ - 10^0 ng m^{-2}
411 d^{-1} , which in the years 2008-10 started in October and lasted until the onset of the fire
412 season, but eventually started later in the years 2005-07 (at least under IEPS). The
413 volatilisation flux is predicted smaller in magnitude than the net-deposition flux during the
414 fire season, but correspondingly, i.e. higher after intense fire seasons. The high RET
415 volatilisation flux, indicated by measured C_a and C_w , seems to be dominated by biomass
416 burning in the region in the previous fire season. F_{aw} is predicted highly fluctuating, also
417 during the observational period (Fig. 3b). Even under UEPS the model is underpredicting F_{aw}
418 (Fig. 3b). The sensitivity to input uncertainties (SM S1.2) may explain part of the
419 underestimate, but not up to one order of magnitude. Neglected RET sources to seawater,
420 such as riverine input may explain part of the discrepancy.

421

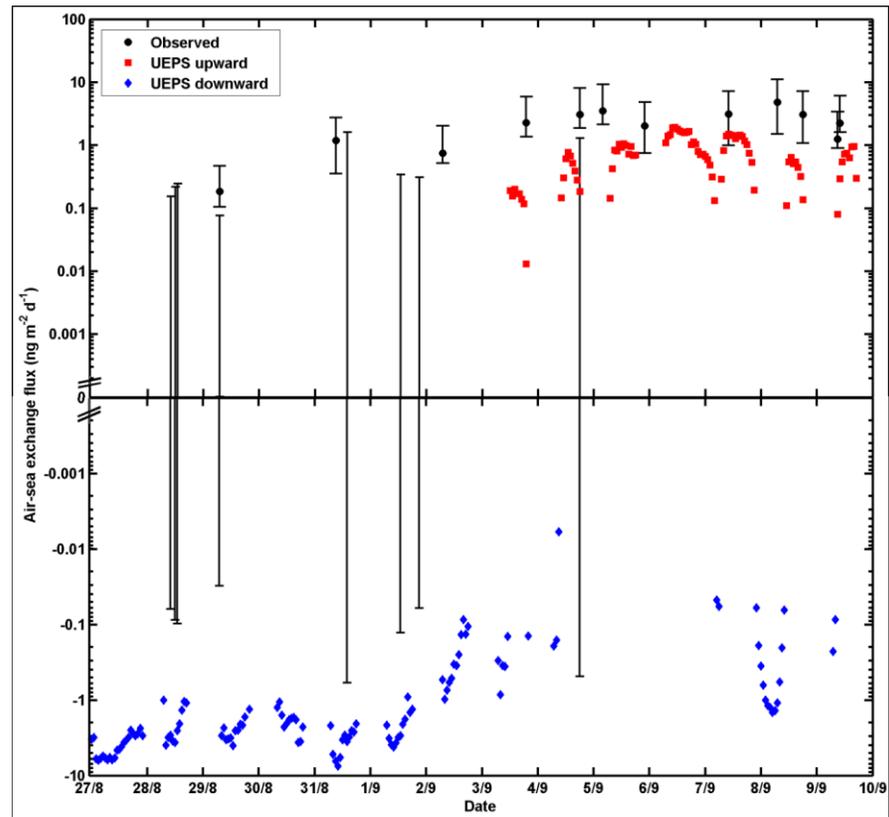
422 Fig. 3. Diffusive air-sea exchange flux, F_{aw} , of RET ($\text{ng m}^{-2} \text{d}^{-1}$; downward in blue and
423 upward in red) using the upper estimate parameter set (UEPS) for the Eastern Mediterranean
424 (28 - $45^\circ\text{N}/8$ - 30°E) (a.) model predicted for 1.1.2005-31.12.2010 and (b.) model predicted and
425 observed (black) for 27.8.-9.9.2010. Hourly mean data filtered against off-shore winds (see
426 text). Error bars including both signs of F_{aw} reflect $C_w < \text{LOQ}$.a.



427

428

429 b.



430

431

432 **4. Conclusions**

433

434 PAH pollution of the atmospheric Mediterranean environment was below previous
435 observations at the beginning of the decade (2001-02; Tsapakis and Stephanou, 2005a;
436 Tsapakis et al., 2006), also considering possible losses during sampling. This might reflect
437 emission reductions. The particulate phase PAHs were concentrated in the size fraction <
438 0.25 μm a.e.d. The residence time in the troposphere is longest for particles around 0.2 μm of
439 size, with $\approx 0.01 \text{ cm s}^{-1}$ being a characteristic corresponding dry deposition velocity (Franklin
440 et al., 2000), which translates into a residence time of ≈ 120 days in the MBL (depth of 1000
441 m; see Table S3) and deposition flux $F_{\text{dep}} = c \times v = 0.03\text{-}0.13 \mu\text{g m}^{-2} \text{ year}^{-1}$ for the individual
442 PAHs associated with the particulate phase ($c = 0.01\text{-}0.04 \text{ ng m}^{-3}$; Table 2b), such as BAP,
443 and 0.80 and 0.35 $\mu\text{g m}^{-2} \text{ year}^{-1}$, respectively, for the total flux of particulate phase PAHs in
444 the ISS and SEM in summer, respectively. The flux will be higher in winter, because of the
445 seasonality of the emissions.

446 Three gas-particle partitioning models were tested and found to underpredict the particulate
447 mass fraction in most of the samples (four PAHs i.e., BAA, TPH, CHR and BBF). Although
448 input parameters were incomplete these results confirm the earlier insight that additional
449 processes on the molecular level need to be included, beyond adsorption (Junge-Pankow
450 model) and absorption in OM (K_{oa} models), namely both adsorption and absorption
451 (Lohmann and Lammel, 2004) or even a complete description of molecular interactions
452 between sorbate and PM matrix (Goss and Schwarzenbach, 2001).

453 Simulations with a non-steady state 2-box model confirm the hypothesis that seasonal
454 depositional input of RET from biomass burning into the surface waters during summer is
455 followed by a period of flux reversal. The volatilisation flux is smaller in magnitude than the

456 net-deposition flux during the previous months, but correspondingly, i.e. higher after intense
457 fire seasons. Future negative emission trends or interannual variability of regional sources
458 may trigger the sea to become a secondary PAH source through reversal of diffusive air-sea
459 exchange. For the wood burning marker RET it is found that the secondary source became
460 significant in recent years: While the flux of secondary RET emissions (from surface
461 seawaters) in the study area was $1.0 \mu\text{g m}^{-2} \text{ year}^{-1}$ (mean of years 2005-2010, _UEPS), the
462 primary sources amounted to $3.1 \mu\text{g m}^{-2} \text{ year}^{-1}$. Because of non-diffusive emission from the
463 sea surface, such as aerosol suspension from sea spray and bubble bursting (Woolf, 1997;
464 Qureshi et al., 2009; Albert et al., 2012), the true volatilisation may have exceeded the
465 diffusive flux significantly.

466

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475

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