

1 **The climate impact of ship NO_x emissions: an improved estimate accounting for plume**
2 **chemistry**

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8

9 **Abstract**

10 Nitrogen oxide (NO_x) emissions from maritime shipping produce ozone (O₃) and hydroxyl
11 radicals (OH), which in turn destroy methane (CH₄). The balance between this warming (due to
12 O₃) and cooling (due to CH₄) determines the net effect of ship NO_x on climate. Previous estimates
13 of the chemical impact and radiative forcing (RF) of ship NO_x have generally assumed that
14 plumes of ship exhaust are instantly diluted into model grid cells spanning hundreds of
15 kilometers, even though this is known to produce biased results. Here we improve the parametric
16 representation of exhaust-gas chemistry developed in the GEOS-Chem chemical transport model
17 (CTM) to provide the first estimate of RF from shipping that accounts for sub-grid-scale ship
18 plume chemistry. The CTM now calculates O₃ production and CH₄ loss both within and outside
19 the exhaust plumes and also accounts for the effect of wind speed. With the improved modeling
20 of plumes, ship NO_x perturbations are smaller than suggested by the ensemble of past global
21 modeling studies, but if we assume instant dilution of ship NO_x on the grid scale, the CTM
22 reproduces previous model results. Our best estimates of the RF components from increasing ship
23 NO_x emissions by 1 Tg(N) yr⁻¹ are smaller than given in the past literature: $+3.4 \pm 0.85 \text{ mW m}^{-2}$
24 (1- σ confidence interval) from the short-lived ozone increase, $-5.7 \pm 1.3 \text{ mW m}^{-2}$ from the CH₄
25 decrease, and $-1.7 \pm 0.7 \text{ mW m}^{-2}$ from the long-lived O₃ decrease that accompanies the CH₄
26 change. The resulting net RF is $-4.0 \pm 2.0 \text{ mW m}^{-2}$ for emissions of 1 Tg(N) yr⁻¹. Due to non-
27 linearity in O₃ production as a function of background NO_x, RF from large changes in ship NO_x
28 emissions, such as the increase since preindustrial times, is about 20% larger than this RF value
29 for small marginal emission changes. Using sensitivity tests in one CTM, we quantify sources of
30 uncertainty in the RF components and causes of the $\pm 30\%$ spread in past model results. The main
31 source of uncertainty is the composition of the background atmosphere in the CTM, which is
32 driven by model formulation (± 10 to 20%) and the plausible range of anthropogenic emissions
33 ($\pm 10\%$).

34 **1. Introduction**

35 Maritime shipping affects climate through emissions of CO₂, nitrogen oxides (NO_x ≡
36 NO+NO₂), and SO₂, the latter two of which indirectly influence methane, ozone, aerosols, and
37 clouds. Other climate impacts due to ship emissions of CO, volatile organic compounds (VOCs),
38 and primary aerosols have significant uncertainties, but are much smaller (Eyring et al., 2010).
39 While ships produce only 3% of anthropogenic CO₂, they emit 17% of anthropogenic NO_x and
40 10% of anthropogenic SO₂ due to high engine temperatures and efficiencies, use of high-sulfur
41 fuel, and general lack of emission controls (Lamarque et al., 2010). CO₂ unambiguously warms
42 the climate and sulfate aerosol derived from SO₂ unambiguously cools it; the net forcing from
43 NO_x, however, involves both warming and cooling components. NO_x emissions, whether from
44 ships or other sources, favor ozone production (warming) as well as hydroxyl (OH) production
45 that destroys methane (cooling). The net balance of these competing effects is cooling for most
46 ground-based NO_x emission sources, including ships (e.g. Fiore et al., 2012), but can be warming
47 for aviation NO_x (Holmes et al., 2011). Here, we summarize all previous reports of methane and
48 ozone radiative forcing (RF) from ship NO_x and then calculate an improved RF that accounts for
49 non-linear chemistry in the exhaust plumes.

50 Rapidly growing international trade has spurred rising ship traffic in recent decades,
51 maintaining around 4% annual growth in the 2000s decade, with important impacts on air quality
52 as well as climate (Dalsøren et al., 2010; Eyring et al., 2010). Ozone generated by increased
53 shipping may explain much of the observed rise in background ozone concentrations reported at
54 coastal sites (Chan, 2009; Dalsøren et al., 2010; Parrish et al., 2009). Holmes et al. (2013) found
55 that rising ship NO_x emissions since 1980 have been one of the most important drivers of
56 decreasing atmospheric methane lifetime and that the wide range of modeled sensitivities to ship
57 emissions is one of the larger uncertainties in calculating trends in methane lifetime. These ship
58 emissions and impacts on climate and air quality are projected to continue growing rapidly
59 through the coming decades unless major changes in emission control technology are adopted
60 (Corbett et al., 2010; Dalsøren et al., 2013; Eyring et al., 2005a; Hodnebrog et al., 2011; Koffi et
61 al., 2010; Paxian et al., 2010).

62 Early efforts to include ship NO_x emissions in global 3-D chemical transport models (CTMs)
63 found that NO_x concentrations were severely overestimated (Davis et al., 2001; Kasibhatla et al.,
64 2000). Subsequent work indicated that the problem was not caused by emission inventory errors,
65 but instead arose from the expedient but inaccurate modeling assumption that ship exhaust
66 instantly mixes into a model grid cell, which is typically hundreds of kilometers across. Under

67 this instant dilution assumption coarse-resolution models bypass the early stages of plume
68 dilution when high NO_x concentrations intensify NO_x chemical losses and suppress O_3 formation
69 several fold (Chen et al., 2005; Kim et al., 2009; Song et al., 2003). While the non-linear nature
70 of NO_x - HO_x - O_3 chemistry in plumes is well known (e.g. Lin et al., 1988) and numerical
71 techniques have been developed for modeling sub-grid-scale plumes from other sources (e.g.
72 Paoli et al., 2011; Sillman et al., 1990), many global CTMs have continued to use the instant
73 dilution assumption for ship NO_x while acknowledging its deficiency. As a result, these models
74 overestimate O_3 and OH production by ships and generate biased impacts on climate and air
75 quality. To date, all estimates of RF due to ship NO_x come from models that assume instant
76 dilution (Eyring et al., 2010).

77 Large-eddy simulations at various spatial resolutions suggest the errors in surface O_3 and OH
78 enhancements caused by instant dilution of ship emissions in global CTMs are as large as 60%
79 (Charlton-Perez et al., 2009), although some Gaussian plume models find larger errors in
80 Northern Hemisphere shipping corridors (Franke et al., 2008; von Glasow et al., 2003). In a
81 European regional CTM, parameterizing ship plume chemistry reduces ship-caused surface O_3 by
82 20% over the North Atlantic Ocean and more near coasts, as compared to instant dilution (Huszar
83 et al., 2010). Vinken et al. (2011; 2013), using a different plume-in-grid approach in the GEOS-
84 Chem global CTM, found similar magnitude reductions in O_3 and also showed that the
85 parameterization improved the model's agreement with NO_x observations across several ocean
86 basins.

87 In this paper we further develop the plume parameterization in GEOS-Chem to better
88 represent CH_4 oxidation within ship exhaust plumes. We then calculate the global impact of ship
89 NO_x on abundances of O_3 and CH_4 and on RF. These impact estimates change under different
90 plume modeling assumptions and accounting for sub-grid-scale chemistry reduces the RF of ship
91 NO_x compared to the ensemble of past studies. We also identify major sources of uncertainty in
92 ship NO_x RF using similar methods to our earlier work on aviation NO_x (Holmes et al., 2011): by
93 decomposing the RF into factors that can be assessed individually and by reproducing the spread
94 of past results in a single model.

95

96 **2. Model description**

97 GEOS-Chem is a global tropospheric CTM driven by assimilated meteorological data from
98 the NASA Goddard Earth Observing System (GEOS-5) (Rienecker et al., 2008). The version

99 used here (9-01-03, www.geos-chem.org) has $2^\circ \times 2.5^\circ$ horizontal resolution and 47 layers. The
100 tropospheric chemical mechanism simulates $\text{HO}_x\text{-NO}_x\text{-VOC-O}_3$ reactions, including bromine
101 (Parrella et al., 2012). Anthropogenic emissions are based on the EDGAR and RETRO global
102 inventories (Olivier and Berdowski, 2001; van Aardenne et al., 2005; van Donkelaar et al., 2008;
103 van het Bolscher, 2008), which are replaced with regional inventories over the United States
104 (NEI2005), Canada (CAC), Mexico (BRAVO), Europe (EMEP) and East Asia (Streets). Ship
105 emissions are described further below.

106 Figure 1 shows ship NO_x emissions in GEOS-Chem which are $5.0 \text{ Tg(N) yr}^{-1}$ and distributed
107 according to ship locations in the AMVER-ICoads database for each month (Lee et al., 2011;
108 Wang et al., 2008). This is close to the best estimate of $5.4 \text{ Tg(N) yr}^{-1}$ for year 2000 (Eyring et al.,
109 2010), and well within the plausible range of $3.0\text{-}10.4 \text{ Tg(N) yr}^{-1}$ (Corbett and Koehler, 2003;
110 Endresen et al., 2007; 2003; Eyring et al., 2005b). GEOS-Chem also includes ship emissions of
111 SO_2 ($8.5 \text{ Tg(S) yr}^{-1}$; Eyring et al., 2005b), CO (1.1 Tg yr^{-1} ; Wang et al., 2008), and VOCs,
112 although ship CO and VOCs are small compared to other sources of those gases. We quantify the
113 effects of ship emissions by comparing a simulation with the base inventory to one with a
114 uniform 5% increase in ship NO_x emissions and another with zero ship NO_x emissions. Results
115 are derived from a simulation of year 2006 after spin up from July 2005.

116

117 **2.1 Plume chemistry and dispersion**

118 Previous versions of GEOS-Chem assumed that sub-grid chemistry in ship plumes convert
119 each mole of NO_x emissions into 10 mole of O_3 and 1 mole of HNO_3 —an ozone production
120 efficiency (OPE) of 10—based on observations of aged ship plumes (Chen et al., 2005). Imposing
121 a globally constant effective emission factor obviously neglects diurnal, seasonal, and regional
122 influences on plume chemistry. In addition, using this method underestimates NO_x concentrations
123 in ship tracks, since some NO_x survives oxidation until the plume has expanded to the global grid
124 resolution. To redress these shortcomings, Vinken et al. (2011) used a Gaussian plume chemistry
125 model to calculate the dilution and chemical evolution of the exhaust over 5 hours, at which point
126 the plume approximately fills a grid cell in the global CTM. The final OPE and fraction of NO_x
127 oxidized to HNO_3 are tabulated for various environmental conditions in a look-up table that
128 GEOS-Chem uses to determine locally appropriate emission factors for ship NO_x , O_3 , and HNO_3 .
129 These aged plume emissions are then injected into the global CTM, which then accounts for the
130 subsequent grid-scale photochemistry and large-scale advection. **Although Gaussian plume**
131 **models poorly simulate the first several minutes of plume aging, when turbulent transport limits**

132 the rates of fast $\text{NO}_x\text{-O}_3$ chemical reactions (Galmarini et al., 1995; Sykes et al., 1992), they can
133 provide a good representation of plume composition after about ten minutes (several kilometers)
134 of aging, once turbulent dispersion homogenizes the plume (Galmarini et al., 1995). Indeed
135 Vinken et al. (2011) demonstrated that their Gaussian plume model predicts NO_x , O_3 , and OH
136 concentrations consistent with field observations over several hours of ship plume aging (Chen et
137 al., 2005).

138 In this work, we update the Gaussian plume model to calculate CH_4 oxidation within the ship
139 plume and verify that the updated model still reproduces field observations of NO_x , O_3 , and OH
140 concentrations (Fig. S1). We also add wind speed as a factor in the look-up table, since CH_4
141 oxidation and O_3 production can vary by a factor of 2 between wind speeds of 2 and 18 m s^{-1} . Our
142 updated plume-in-grid parameterization depends on 8 meteorological and chemical factors:
143 ambient concentrations of NO_x and O_3 , solar zenith angle at emission time and 5 hours later,
144 photolysis rates of NO_2 and O_3 , temperature, and wind speed. Figure S2 shows how the
145 parameterization responds to each of these factors. Clouds affect the parameterized plume
146 chemistry through photolysis rates, but not through dispersion rates (Verzijlbergh et al., 2009).
147 The global CTM with updated plume chemistry has up to 3% less NO_x and 1% less O_3 in the
148 marine boundary layer compared to the earlier parameterization. Therefore, comparisons of the
149 CTM to observations over the North Atlantic and North Pacific Oceans shown by Vinken et al.
150 (2011; their Figs. 4,5) are unchanged. Specifically, in regions that are impacted by ship emissions
151 but outside distinct plumes, the parametric plume chemistry predicts median NO_x abundances
152 within 30% of observed values while instant dilution over predicts NO_x by a factor of 2. Ozone
153 observations in the same regions are consistent with the plume parameterization but unable to
154 falsify other model variants.

155 We compare the chemical and climate impact of shipping under three different modeling
156 assumptions about plume dilution and chemistry:

- 157 1. Instant dilution. We neglect sub-grid chemistry and emit NO_x into the CTM grid, at the rate
158 specified by the emission inventory, as done in previous studies with other models.
- 159 2. Fixed OPE. We assume that sub-grid chemistry converts each mole of ship NO_x to 10 moles
160 of O_3 and 1 mole of HNO_3 .
- 161 3. Parametric plume chemistry. This is our best representation of sub-grid plume chemistry
162 using the lookup tables described above.

163

164 2.2 Radiative forcing calculations

165 The global mean RF (F) from ship NO_x emissions consists of a short-lived O_3 component
166 (F_{O_3}) that decays within months after emissions and long-lived CH_4 and O_3 components that
167 persist for over a decade (F_{CH_4} and $F_{\text{long-O}_3}$, respectively):

$$168 \quad F = F_{\text{O}_3} + F_{\text{CH}_4} + F_{\text{long-O}_3}. \quad (1)$$

169 We calculate these components in steady state from the CTM output using a similar
170 decomposition as Holmes et al. (2011):

$$171 \quad F_{\text{O}_3} = (d[\text{O}_3]/dE) (dF/d[\text{O}_3]) \Delta E, \quad (2)$$

$$172 \quad F_{\text{CH}_4} = (d \ln \tau_{\text{total}}/dE) f[\text{CH}_4] (dF/d[\text{CH}_4]) S \Delta E, \text{ and} \quad (3)$$

$$173 \quad F_{\text{long-O}_3} = a F_{\text{CH}_4} / S, \quad (4)$$

174 where $d[\text{O}_3]/dE$ is the steady-state response to ship NO_x emissions (E) while holding $[\text{CH}_4]$
175 constant and $d \ln \tau_{\text{total}}/dE$ is the accompanying change in total atmospheric CH_4 lifetime, $dF/d[\text{O}_3]$
176 and $dF/d[\text{CH}_4]$ are the radiative efficiencies of tropospheric O_3 and CH_4 , $[\text{CH}_4] = 1.76$ ppm is the
177 global-mean CH_4 mole fraction in 2000, f is the CH_4 feedback factor that prolongs the lifetime of
178 CH_4 perturbations, $S = 1.15$ is the enhancement of F_{CH_4} due to CH_4 -derived stratospheric water
179 vapor (Myhre et al., 2007), and $a = 0.34 \pm 0.13$ describes the perturbations to O_3 abundance and
180 RF that accompany global CH_4 changes. The a term derives from a literature survey of multiple
181 CTMs and radiative transfer models (see SI and Holmes et al. 2011). This CTM and most prior
182 publications report changes in the CH_4 lifetime due to tropospheric OH (τ) rather than the total
183 atmospheric lifetime; these are related via $(d \ln \tau_{\text{total}}) = b (d \ln \tau)$, where the best estimates of all
184 atmospheric methane losses imply $b = 0.82 \pm 0.03$ (Prather et al., 2012).

185 CTM diagnostics provide $d[\text{O}_3]/dE$ and $d \ln \tau/dE$ for each plume chemistry treatment based on
186 5% perturbations to global ship NO_x emissions. Values and one-sigma (68%) confidence intervals
187 for other factors in equations 2-4 are given in Holmes et al. (2011), with the following updates.
188 Recent data suggest a smaller feedback factor, $f = 1.34 \pm 0.06$ (Holmes et al., 2013). We use a
189 ship-specific radiative efficiency for O_3 , which is smaller than that of aviation O_3 and that of
190 long-lived O_3 changes (cf. Fuglestad et al., 2008; Holmes et al., 2011) because ship O_3 is mostly
191 confined to low altitudes and high latitudes (e.g. Hoor et al., 2009). We adopt a value of 33 ± 4
192 $\text{mW m}^{-2} \text{DU}^{-1}$, based on the mean of recent studies (Table 1; Hoor et al., 2009; Myhre et al.,
193 2011), recognizing that radiative efficiency depends on the distribution of the O_3 burden and that
194 radiative transfer models differ by about 10% (Myhre et al., 2011). While models assuming

195 instant dilution of ship NO_x were used to calculate the ozone radiative efficiency, we show below
196 that the pattern of ship ozone perturbations is similar with the parametric plume assumption.

197

198 **3. Chemical response to ship NO_x emissions**

199 **3.1 Ozone production**

200 Figure 2 shows simulated, time-averaged OPE of ship NO_x with parametric plume chemistry.
201 OPE is defined here as $\Delta P(O_x) / \Delta L(NO_x)$, where P(X) and L(X) are the time-integrated
202 production and loss of species X, O_x is the odd oxygen family (O + O₃ + NO₂ + 2NO₃ + many
203 reservoirs of NO₂ and NO₃, see e.g. Parrella et al., 2012), and Δ refers to a steady-state change
204 caused by a 5% increase in ship NO_x emission. Chemical oxidation to HNO₃ and nitrate is the
205 main NO_x loss process, but surface deposition of NO₂, N₂O₅ and organic nitrates are about 6% of
206 global L(NO_x) and 2% of $\Delta L(NO_x)$. The parametric plume chemistry calculates a global-mean
207 OPE of 1.7 during young plumes. Some ship NO_x survives beyond the 5-hour scope of the plume
208 parameterization and its subsequent chemical effects are calculated with the grid-resolved
209 chemistry. The global-mean total OPE (plume plus grid chemistry) is 8.5. Although OPE is
210 negative at night and episodically in polluted continental outflow, the annual-mean OPE is
211 positive everywhere, both in the first 5 hours and total. The busiest shipping corridors in the
212 North Atlantic and North Pacific Oceans have an OPE of 4-8 while values around 40 are found in
213 the least-trafficked areas of the Equatorial Pacific Ocean. Figure 3 shows zonal-mean OPE for the
214 other plume assumptions. With the instant dilution assumption OPE is 12.5. Thus, the parametric
215 plume chemistry has the intended effect of suppressing O₃ production by 30%, relative to instant
216 dilution. The fixed OPE scenario assumes OPE to be 10 in the plume, but subsequently global O₃
217 production is suppressed in grid-resolved chemistry since no NO_x is released so the total OPE is
218 8.0.

219 O₃ enhancements generated by ship NO_x concentrate over the major emission regions in the
220 Atlantic and Pacific Oceans and a narrow strip in the Indian Ocean, as seen in Figure 4. The
221 largest O₃ enhancements are displaced eastward relative to the emissions in each ocean basin, as
222 found in previous studies, reflecting cumulative downwind production (Endresen et al., 2003;
223 Eyring et al., 2007). Figure 3 compares the zonal-mean O₃ column enhancements across the three
224 plume chemistry simulations. The pattern is similar in the instant dilution and parametric plume
225 simulations and this justifies our use of O₃ radiative efficiencies that were derived in models with
226 instant dilution. O₃ column enhancements in the fixed OPE simulation are qualitatively different

227 and concentrated mainly in the high northern latitudes, because OPE is not suppressed in winter
228 or by high NO_x emissions in this scenario. As Table 2 reports, the global-mean O₃ column change
229 for a 1 Tg(N) yr⁻¹ increase in ship NO_x is 0.10 DU in the parametric plume model, compared to
230 0.16 DU under the instant dilution assumption and 0.12 DU under fixed OPE assumption. These
231 column perturbations are not strictly proportional to the OPE across scenarios because the
232 lifetime of O₃ increases towards the poles. Previous CTM studies using instant dilution found
233 0.14-0.2 DU enhancements for emissions of 1 Tg(N) yr⁻¹ (Hodnebrog et al., 2011; Hoor et al.,
234 2009), which encompasses our estimate under instant dilution.

235

236 **3.2 Methane oxidation**

237 NO_x emissions affect OH concentrations and CH₄ oxidation in two general ways: directly by
238 recycling HO₂ and RO₂ back to OH, which increases the OH/HO₂ ratio and reduces the HO_x sink
239 via HO₂ self reaction; and indirectly by increasing O₃, which is a primary source of OH through
240 photolysis in the presence of water vapor. We define a time-averaged CH₄ oxidation efficiency
241 (MOE) similar to OPE above, as $\Delta L(\text{CH}_4) / \Delta L(\text{NO}_x)$. The MOE is 0.42 in the first 5 hours of
242 plume aging, as calculated by the parametric plume chemistry (Figure 2). MOE is low in the
243 young plume because of rapid NO_x loss, despite high OH concentrations in the plume that can be
244 double the ambient values (Chen et al., 2005; Song et al., 2003). The NO_x lifetime and MOE
245 increase in the grid-scale chemistry, so that the total MOE is 3.1 with parametric plume
246 chemistry. The instant dilution assumption raises the overall MOE to 4.4, while in the fixed OPE
247 simulation the MOE is only 1.2 because the direct chemical effects of ship NO_x on OH are
248 neglected.

249 Table 3 reports the sensitivity of CH₄ lifetime to increasing ship NO_x emissions by 1 Tg(N) yr⁻¹
250 ¹. The largest response occurs under instant dilution (-1.0%) and the smallest under fixed OPE (-
251 0.26%), with the parametric plume falling in the middle (-0.7%). The instant dilution value is
252 similar to those that we previously found in the University of California, Irvine (UCI) CTM and
253 Oslo CTM3, -0.8% and -0.9%, respectively (Holmes et al., 2013) and within the range of values
254 in literature (-0.9 ± 0.3%, Table 3). In past studies, the largest CH₄ lifetime changes per Tg(N) yr⁻¹
255 (-1.3 to -1.4%) derived from models that used small inventories of ship emissions and calculated
256 sensitivities by removing all ship NO_x emissions (Endresen et al., 2003; Lawrence and Crutzen,
257 1999). This suggests that the non-linear aspects of NO_x-ozone chemistry influence the spread of
258 model results and we investigate this further in Section 4.

259 Treatment of ship plume chemistry has an important effect on the current CH₄ lifetime, as well
260 as its perturbations (Table 3). In many CTMs, the CH₄ lifetime due to tropospheric OH is shorter
261 than observed and the cause of this discrepancy remains unknown (e.g. Holmes et al., 2013; Naik
262 et al., 2013). We find that instant dilution produces the shortest CH₄ lifetime of the three plume
263 chemistry scenarios (9.2 yr). The more realistic treatment of ship plume chemistry afforded by the
264 parametric plume model raises the lifetime to 9.4 yr, but the discrepancy with observations
265 remains (Prather et al., 2012). While the fixed OPE model is longer (9.7 yr), it cannot be
266 considered more realistic. Neglecting all ship NO_x emissions, the CH₄ lifetime is 9.8 yr. Thus,
267 with parametric plume chemistry, ship NO_x drives about 4% of all CH₄ oxidation by tropospheric
268 OH and 13% of that occurs in the first 5 hours of plume aging. Our results here include bromine
269 chemistry, which acts as an O₃ and HO_x sink. Removing bromine chemistry from a simulation
270 with fixed OPE shortens the CH₄ lifetime by about 0.5 yr, but the chemical impact of ship
271 emissions is nearly unchanged from the values in Tables 2 and 3.

272

273 **4. Radiative forcing from ship NO_x emissions**

274 Figure 5 shows all past reports of the CH₄ and short-lived O₃ RF components from ship NO_x
275 emissions. These include CTMs (Dalsøren et al., 2010; 2013; 2009; 2007; Eide et al., 2013;
276 Endresen et al., 2003; Eyring et al., 2007; Fuglestvedt et al., 2008; Hodnebrog et al., 2011; Hoor
277 et al., 2009; Lawrence and Crutzen, 1999; Myhre et al., 2011) and global climate models with
278 chemistry (Eyring et al., 2007; Hoor et al., 2009; Myhre et al., 2011; Olivie et al., 2012; Unger et
279 al., 2010), as well as some derived from literature synthesis (Borken-Kleefeld et al., 2010; Lee et
280 al., 2007). Where possible, we calculate F_{CH_4} from the reported changes in CH₄ lifetime using Eq.
281 (3), in order to use consistent assumptions about CH₄ lifetime, feedback and radiative efficiency.
282 RF values are scaled to emissions of 1 Tg(N) yr⁻¹ and are for steady-state conditions. We report
283 results from individual models in multi-model studies where possible. All of these RF estimates
284 have assumed instant dilution of ship emissions, which biases the RF values as we show below.

285 From the literature ensemble, we estimate the O₃ RF to be $+6.0 \pm 1.9 \text{ mW m}^{-2}$ and the CH₄ RF
286 to be $-8.0 \pm 2.4 \text{ mW m}^{-2}$ for 1 Tg(N) yr⁻¹. This average neglects two studies with small absolute
287 magnitudes that were clearly related to unjustified modeling assumptions. Early work by Lee et
288 al. (2007) used CH₄ and O₃ sensitivities to land NO_x emissions, rather than ship-specific
289 sensitivities that tend to be higher. The ship emission inventory used by one multi-model study
290 (Eyring et al., 2007) was subsequently found to unrealistically concentrate ship emissions along
291 narrow corridors and underestimate emissions in the tropics, as acknowledged in their work, both

292 of which tend to underestimate O₃ production. Figure 5 also shows three RFs estimates derived
293 from our previous analysis of CH₄ lifetime (Holmes et al., 2013). Two of these estimates lie
294 within the cluster of literature values and are based on CTMs that assume instant dilution, while
295 the outlying third estimate based on an earlier version of GEOS-Chem with fixed OPE
296 demonstrates that plume chemistry significantly influences the climate impact of ships.

297 In this work, the short-lived O₃ and CH₄ RFs with parametric plume chemistry are +3.4 and -
298 5.7 mW m⁻², respectively, for emissions of 1 Tg(N) yr⁻¹. With instant dilution, the RF components
299 are close to the central estimate from past literature and about 40% larger than our best estimate:
300 +5.3 and -8.5 mW m⁻². The fixed OPE model, unlike the others, predicts that warming from
301 short-lived O₃ (+3.8 mW m⁻²) exceeds the CH₄ cooling (-2.2 mW m⁻²) because instantly
302 converting NO_x emissions to HNO₃ neglects its direct effect on OH. The radiative efficiency in
303 the fixed OPE model is likely smaller than assumed here (see Sect. 2.3 and 3.1), but we have not
304 recalculated it because the fixed OPE model is not used to derive our best estimate.

305 Global aerosol impacts of ship NO_x have been identified as a knowledge gap that we briefly
306 estimate (Eyring et al., 2010). Ship NO_x increases oxidative production of nitrate and sulfate in
307 our simulations by 9% and 0.4%, respectively, compared to a simulation with no ship NO_x. Some
308 of these products absorb onto sea-salt aerosols, but this makes a negligible contribution to sea-salt
309 aerosol mass, so no RF is expected. The largest changes in aerosol column concentration occur
310 over land, however, due to long-range transport of O₃ and H₂O₂ perturbations from ship NO_x.
311 These oxidants mainly convert SO₂ to sulfate in cloud water, which is quickly followed by wet
312 deposition, so that ship NO_x drives sulfate aerosol burdens down over anthropogenic SO₂ source
313 regions, despite the increased oxidation. Chemical teleconnections initiated by land-based NO_x
314 emissions have been reported previously in which NO_x emissions increase sulfate burdens over
315 distant continents (Leibensperger et al., 2011). Land-based and ship NO_x emissions may have
316 opposite sign teleconnections with sulfate due to larger H₂O₂ perturbations produced in moist
317 marine air, but these effects should be evaluated in other CTMs. Averaged globally, the sulfate
318 burden falls by -6.3 μg m⁻² for ship emissions of 1 Tg(N) yr⁻¹, and the nitrate burden increases by
319 +7.9 μg m⁻² to consume available ammonium. Applying radiative efficiencies (direct effect only)
320 of these species (Myhre et al., 2013), the direct RF from these individual changes is +/- 1.2 mW
321 m⁻², with nearly perfect cancellation between sulfate and nitrate components. The aerosol direct
322 RF from ship NO_x is therefore only 2% of the O₃ and CH₄ RF. **Aerosol indirect effects, black
323 carbon, and organic carbon also contribute to radiative forcing from ships (Eyring et al., 2010)
324 but are beyond the scope of this study of ship NO_x.**

325 Our global RF calculation using parametric plume chemistry is the first to account for sub-
326 grid-scale ship NO_x chemistry. Being based on a single model, we are unable to estimate
327 uncertainties using the common approach of model ensembles. Instead we develop confidence
328 intervals by propagating uncertainties through Eqs. (2)-(3). Among the ensemble of models with
329 instant dilution, the one-sigma ranges of $(d[\text{O}_3]/dE)$ and $(d \ln \tau_{\text{total}}/dE)$ are 20% of their respective
330 means. Assuming the same proportional uncertainty for these factors with parametric plume
331 chemistry, and the one-sigma ranges for other factors given in Section 2.2, the one-sigma
332 confidence intervals for CH₄ and short-lived O₃ RFs are 22% and 25%, respectively.

333 An alternative approach to uncertainty analysis is to probe the causes for spread among the
334 past model studies. Much of this spread can be reproduced through several variants of the GEOS-
335 Chem model, which are shown in Figure 5. Given the non-linear nature of NO_x-O₃ chemistry, we
336 recalculate the ship NO_x RF against a reference simulation without any ship NO_x. To this point,
337 all results derived from 5% emission perturbations, which describes the climate response to small
338 marginal increases or decreases in emissions. Removing all ship NO_x from the simulation reveals
339 the average RF of all ship NO_x and is a common way to calculate the ship NO_x RF since
340 preindustrial times. With complete removal of ship NO_x, we find O₃ and CH₄ RF components per
341 Tg(N) yr⁻¹ are about 20% larger than with the 5% perturbations for both parametric plume
342 chemistry and instant dilution. The RF components shift along the model ensemble's major axis
343 of variability. Indeed past studies using complete removal have on average predicted 10% larger
344 ship RF than studies with 5-30% emission perturbations, so combined effects of non-linearities
345 and different perturbations might explain up to half of the model ensemble spread. We note that
346 the RF components are insensitive to size of the ship NO_x perturbation when assuming fixed
347 OPE. This demonstrates that the non-linear aspects of O₃ chemistry are generated almost entirely
348 during NO_x loss and O₃ production, and that adding O₃ alone does not significantly change the O₃
349 lifetime.

350 Model grid resolution is known to influence climatically important chemical fluxes, such as
351 O₃ production (Wild and Prather, 2006), so we test whether resolution affects RF from ship NO_x.
352 Doubling the GEOS-Chem grid size to 4° × 5° reduces the CH₄ lifetime and O₃ burden compared
353 to 2° × 2.5° resolution. Nevertheless, the O₃ and CH₄ responses to 5% increases in ship NO_x
354 emissions are unchanged from the finer resolution. Given this resolution-independence, we
355 conduct additional sensitivity tests at coarse resolution for computational expediency. A previous
356 version of the model (9-01-02) differs by less than 10% in terms of ship-NO_x perturbations from
357 the current version, despite improvements to wet scavenging, sea-salt aerosol, and stratospheric

358 chemistry (Jaeglé et al., 2011; Murray et al., 2012; Wang et al., 2011), indicating minimal
359 sensitivity of ship NO_x impacts to these processes. The ship RF is quite sensitive to anthropogenic
360 emissions, however. Using the Climate Model Intercomparison Project (CMIP5) inventory for
361 year 2000 (Lamarque et al., 2010) rather than the standard inventory described in Section 2, the
362 ship NO_x RF components are about 15% larger. The shift in RF components lies along the major
363 axis of variability in the model ensemble, indicating global (not ship) inventory differences could
364 contribute substantially to the ensemble spread presumably by generating background
365 atmospheres with different levels of NO_x and HO_x precursors. The CMIP5 inventory prescribes
366 more CO emissions (610 vs. 580 Tg(CO) yr⁻¹) and slightly less NO_x emissions (27.2 vs. 27.6
367 Tg(N) yr⁻¹) with large changes in their spatial distributions. These emission differences tend to
368 reduce background OH and NO_x and make ozone production more NO_x sensitive, which is
369 consistent with the direction of RF changes in the simulations. Our earlier work on climate
370 forcing from aviation NO_x similarly identified background NO_x levels as a driver of model
371 uncertainty. Although emission inventories are routinely updated and improved, reasonable
372 inventories continue to differ by 10% for NO_x and 20% for CO at the global level; differences are
373 often larger for biomass burning and natural emissions (Granier et al., 2011). If the two
374 inventories in GEOS-Chem exhibit typical differences, then inventory uncertainty may account
375 for ±10 % range in ship NO_x RF. Uncertainties in chemistry, transport and other processes that
376 control background atmospheric composition contribute as much or more than emissions to the
377 range in RF responses across models, since multi-model studies using common emissions still
378 exhibit ±20 % ranges in RF components (e.g. Eyring et al., 2007; Hoor et al., 2009; Myhre et al.,
379 2011). Three sources of uncertainty, when combined in quadrature, are therefore sufficient to
380 explain the ±30 % range of ship RF components in the literature ensemble: non-linearity from the
381 ship emission perturbation magnitude (±10 to 20%), emissions from other sources (±10%), and
382 other processes that control background atmospheric composition (±10 to 20%).

383

384 **5. Conclusions**

385 The non-linear chemistry governing O₃ and OH production in emission plumes has been
386 recognized for decades. In spite of this knowledge, global modeling studies of ship NO_x
387 emissions and their impacts on climate and air quality are usually made under the assumption that
388 emissions are instantly diluted into large grid volumes, which overestimates production of
389 tropospheric O₃ and OH. We present a suite of model simulations that quantify this error, one of
390 which uses an improved, more physically realistic treatment of plume chemistry on time and

391 spatial scales smaller than the global model grid. The limited observations of ship plume
392 composition during aging hamper efforts to widely evaluate the parameterization, but we have
393 shown that it is consistent with available data. With parametric plume chemistry, OPE from ship
394 NO_x is 30% smaller than under instant dilution. Methane perturbations from ship NO_x are
395 likewise reduced 30%. Parametric plume chemistry also increases the global atmospheric CH₄
396 lifetime compared to instant dilution, which brings the model closer to observations, but it is still
397 too short.

398 Our best estimate of the ship NO_x RF from the short-lived O₃ increase is $+3.4 \pm 0.85 \text{ mW m}^{-2}$
399 for steady-state emissions of 1 Tg(N) yr^{-1} . The RF from the CH₄ decrease is $-5.7 \pm 1.3 \text{ mW m}^{-2}$,
400 and RF from the long-lived O₃ reduction accompanying the CH₄ decrease is $-1.7 \pm 0.7 \text{ mW m}^{-2}$.
401 For each component the central estimate is similar to the smallest magnitude of previously
402 published RF estimates, due to our treatment of sub-grid-scale chemistry in ship emission plumes.
403 Combining all these components and accounting for correlations caused by common factors, our
404 best estimate of the total RF from ship NO_x is $-4.0 \pm 2.0 \text{ mW m}^{-2}$. Our RF estimate derives from
405 marginal (5%) changes in ship NO_x emissions. Scaling the marginal RF up to year 2010 total
406 emissions of $6.8 \text{ Tg(N) yr}^{-1}$ (Eide et al., 2013) suggests an RF of $-27.2 \pm 13.6 \text{ mW m}^{-2}$, but the
407 average RF of all ship NO_x emissions is likely about 20% larger (-33 mW m^{-2}) because of non-
408 linearity in O₃ production. Our best estimates of individual RF components have one-sigma
409 (68%) confidence intervals of $\pm 20\%$ to $\pm 30\%$. The largest contribution to this uncertainty arises
410 from differing abundances of photochemical oxidants in the background atmosphere, which when
411 entrained into ship plumes can alter their chemistry. Global emissions and model formulation
412 both contribute to these differences in the background atmosphere. Further reductions in RF
413 uncertainty are therefore unlikely without stronger observational constraints on radical sources
414 and sinks in the remote marine atmosphere and additional observational case studies of ship
415 plume aging.

416

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649

650

651 **Table 1. Radiative efficiency of O₃ generated from ship NO_x**

Source	Value or range, mW m ⁻² DU ⁻¹
Eyring et al. (2007)	33.2-33.7 ^{a,b}
Fuglestedt et al. (2008)	27.8 ^a
Hoor et al. (2009)	36.7 ± 3.0 ^{b,c}
Myhre et al. (2011)	30.1 ± 1.6 ^{b,c}
This work	33.0 ± 4 ^d

652 ^a Calculated by removing all ship NO_x emissions.

653 ^b Derived from reported O₃ burden and RF.

654 ^c Calculated for 5% perturbation to year 2000 ship NO_x emissions.

655 ^d Average of models used by Hoor et al. (2009) and Myhre et al. (2011).

656

657 **Table 2: Effect of ship NO_x emissions on O₃ column and RF.^a**

	Instant dilution		Fixed OPE		Parametric plume chemistry	
	$\frac{d[O_3]}{dE}$	$\frac{dF_{O_3}}{dE}$	$\frac{d[O_3]}{dE}$	$\frac{dF_{O_3}}{dE}$	$\frac{d[O_3]}{dE}$	$\frac{dF_{O_3}}{dE}$
<i>This work</i>						
GEOS-Chem ^b	0.16	5.3	0.12	3.8	0.10	3.4
<i>Holmes et al. (2013)</i>						
GEOS-Chem ^{c,d}			0.12	3.8		
UCI CTM ^d	0.17	5.6				
Oslo CTM3 ^d	0.23	7.7				
<i>Literature</i>						
models ^e	0.18 ± 0.04	6.0 ± 1.9				

658 ^a $d[O_3]/dE$ is the derivative of ozone column density with respect to ship NO_x emissions,
 659 reported in DU [Tg(N) yr⁻¹]⁻¹. dF_{O_3}/dE is the derivative of O₃ RF with respect to ship NO_x
 660 emissions, reported in mW m⁻² [Tg(N) yr⁻¹]⁻¹. Both $d[O_3]/dE$ and dF_{O_3}/dE are calculated from
 661 steady-state 5% perturbations to ship NO_x emissions on top of the base emission inventory.

662 ^b Emissions as described in Section 2. Ship NO_x emissions are 5.0 Tg(N) yr⁻¹.

663 ^c Holmes et al. (2013) used GEOS-Chem version 9-01-02, while this work uses version 9-01-03.

664 ^d Configured as described by Holmes et al. (2013), using Representative Concentration Pathway
 665 (RCP) emissions for year 2000. These include 5.4 Tg(N) yr⁻¹ from ships.

666 ^e From modeling studies by Endresen et al. (2003), Hoor et al. (2009), Dalsoren et al. (2010),
 667 Borken-Kleefeld et al. (2010), Unger et al. (2010), Myhre et al. (2011), Olivie et al. (2012),
 668 Dalsoren et al. (2013), and Eide et al. (2013) excluding Lee et al. (2007) and Eyring et al. (2007)
 669 for reasons given in Section 4. Emission inventories and perturbation magnitudes differ, but all
 670 assume instant dilution.

671

672 **Table 3: Effect of ship NO_x emissions on CH₄ lifetime and RF.^a**

	Instant dilution			Fixed OPE			Parametric plume chemistry		
	τ	$\frac{d \ln \tau}{dE}$	$\frac{dF_{\text{CH}_4}}{dE}$	τ	$\frac{d \ln \tau}{dE}$	$\frac{dF_{\text{CH}_4}}{dE}$	τ	$\frac{d \ln \tau}{dE}$	$\frac{dF_{\text{CH}_4}}{dE}$
<i>This work</i>									
GEOS-Chem ^b	9.2	-1.0	-8.5	9.7	-0.26	-2.2	9.4	-0.7	-5.7
<i>Holmes et al. (2013)</i>									
GEOS-Chem ^{c,d}				10.0	-0.31	-2.6			
UCI CTM ^c	8.5	-0.8	-6.8						
Oslo CTM3 ^c	8.7	-0.9	-7.6						
<i>Literature</i>									
models ^e		-0.9±0.3	-8.0±2.4						
observations ^f	11.2±1.3			11.2±1.3			11.2±1.3		

673 ^a τ is the CH₄ lifetime due to tropospheric OH, reported in years. $(d \ln \tau)/dE$ is the derivative of τ
674 with respect to ship NO_x emissions, reported in % [Tg(N) yr⁻¹]⁻¹. dF_{CH_4}/dE is the derivative of
675 CH₄ RF with respect to ship NO_x emissions, reported in mW m⁻² [Tg(N) yr⁻¹]⁻¹. Both $d\tau/dE$ and
676 dF_{CH_4}/dE are calculated from steady-state 5% perturbations to ship NO_x emissions on top of the
677 base emission inventory.

678 ^b Emissions as described in Section 2. Ship NO_x emissions are 5.0 Tg(N) yr⁻¹.

679 ^c Holmes et al. (2013) used GEOS-Chem version 9-01-02, while this work uses version 9-01-03.

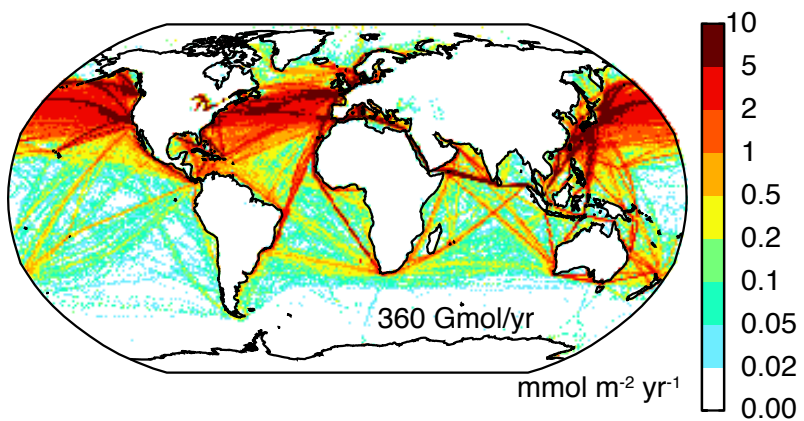
680 ^d Configured as described by Holmes et al. (2013), using Representative Concentration Pathway
681 (RCP) emissions for year 2000. These include 5.4 Tg(N) yr⁻¹ from ships.

682 ^e From modeling studies used in Table 2, plus Lawrence and Crutzen (1999), Dalsoren et al.

683 (2007), Dalsoren et al. (2009), and Hodnebrog et al. (2011). Emission inventories and

684 perturbation magnitudes differ, but all assume instant dilution.

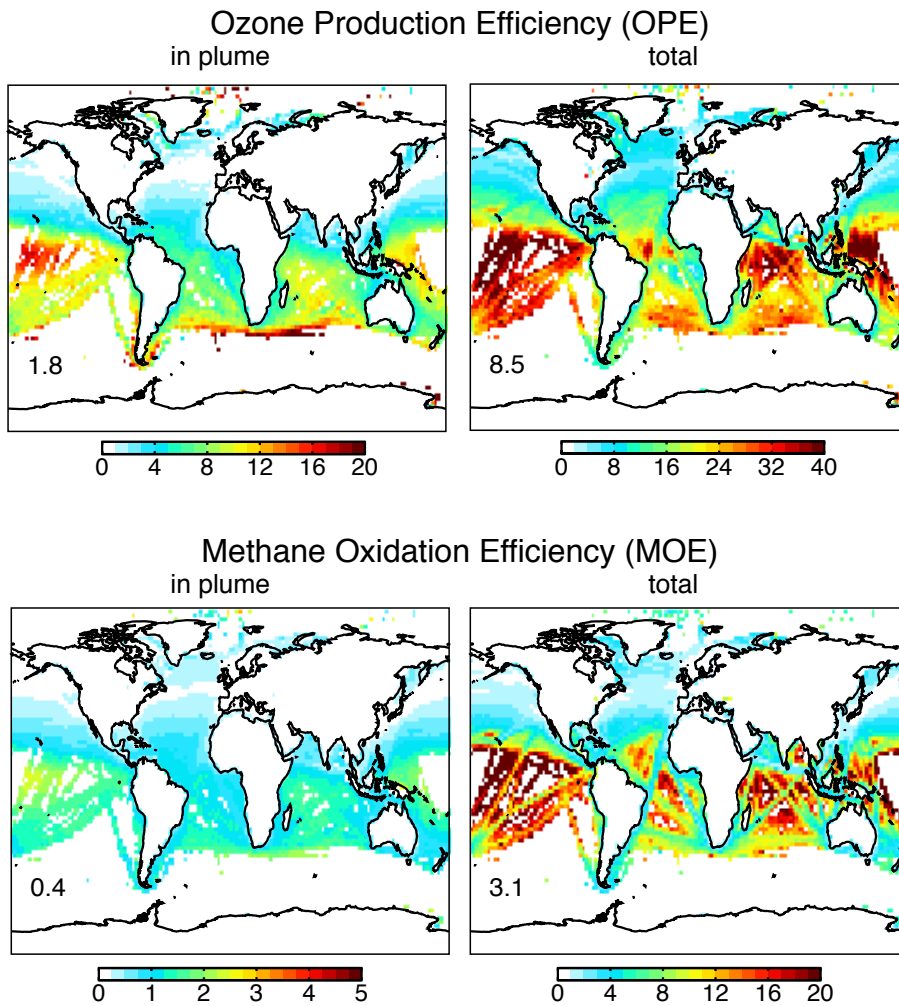
685 ^f From observations of methyl chloroform (Prather et al., 2012).



686

687 **Figure 1:**

688 Annual ship NO_x emissions (mmol m⁻² yr⁻¹) used in GEOS-Chem. The CTM includes monthly
689 variations in locations and magnitude (not shown).

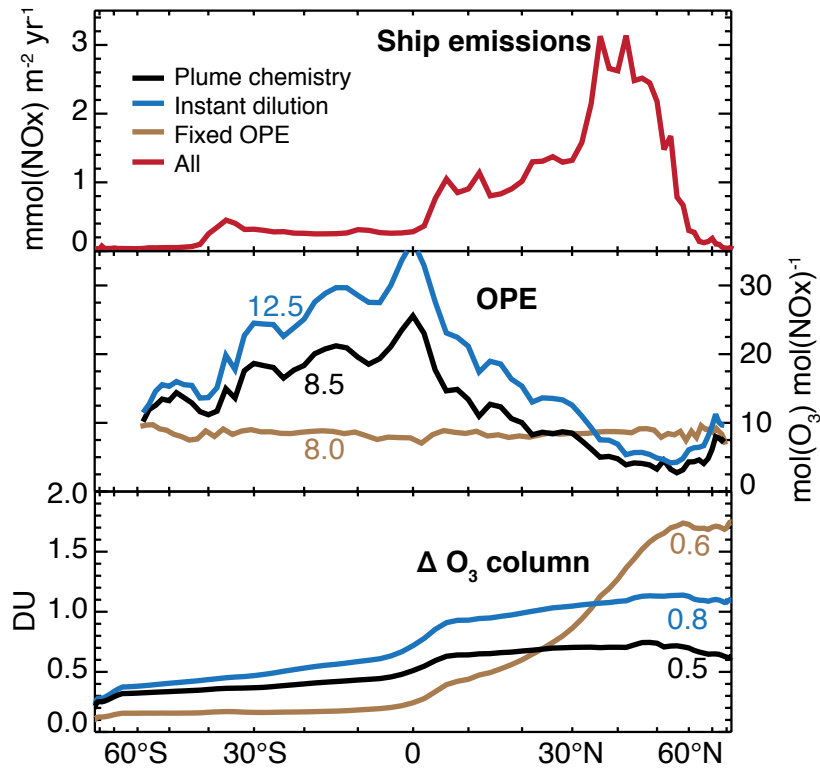


690

691 **Figure 2:**

692 Time-averaged ozone production efficiency (top row) and methane oxidation efficiency
 693 (bottom row) for marginal increases in ship NO_x emissions within the sub-grid plume (left panels)
 694 and total (plume plus grid chemistry, right panels). Values are shown from the CTM with
 695 parametric plume chemistry only where ship NO_x emissions exceed 10⁴ molec m⁻² s⁻¹. Inset
 696 numbers give global averages.

697



698

699

Figure 3:

700

Annual and zonal-mean NO_x emission and ozone changes caused by maritime shipping plotted

701

against sine latitude for each plume dilution assumption. OPE is averaged over regions where

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ship NO_x emissions exceed $10^4 \text{ molec m}^{-2} \text{ s}^{-1}$. Ozone column changes (ΔO_3) are calculated from

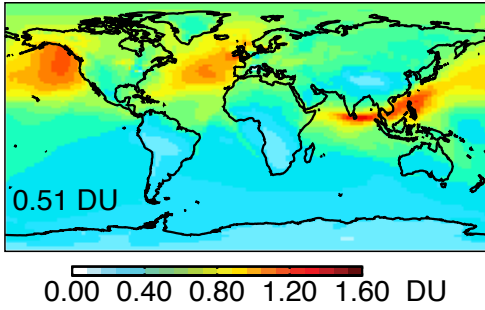
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5% perturbations to full ship emissions, then multiplied by 20. Inset numbers give global

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averages.

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Figure 4:

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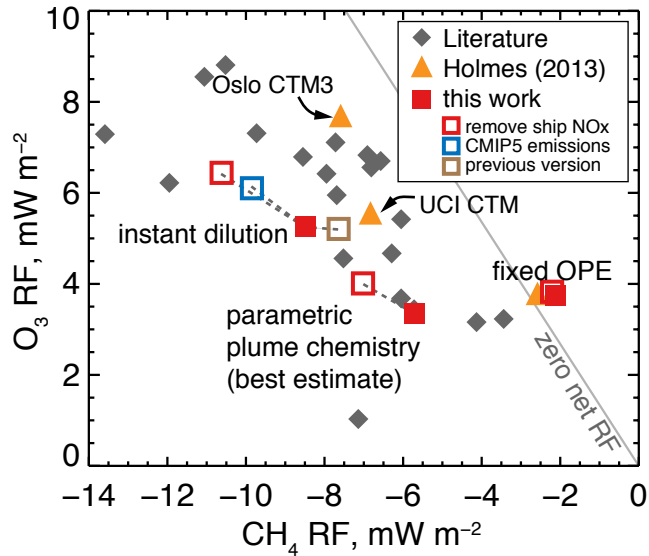
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Annual-mean O₃ column enhancements due to ship NO_x emissions with parametric plume chemistry. Values are calculated from a 5% emission perturbation and multiplied by 20. Inset number gives the global-mean change. Patterns are similar in the instant dilution simulation, but with fixed OPE, the O₃ enhancements shift toward high northern latitudes (Fig. 3).



714

715 **Figure 5.**

716 Steady-state RF (mW m^{-2}) from O_3 and CH_4 caused by ship NO_x emissions. Values are scaled
 717 to emissions of 1 Tg(N) yr^{-1} . Dashed lines link estimates that are made with 5% increases in ship
 718 NO_x (filled squares) to others made with the same plume dilution and chemistry assumption
 719 (open squares). These other estimates use CMIP5 emissions, a previous GEOS-Chem model
 720 version, or complete removal of ship NO_x emissions. The zero net RF line accounts for long-
 721 lived O_3 changes that enhance the CH_4 RF by approximately 34% (Eq. 4). Literature values are
 722 from studies listed in Tables 2 and 3.