1	Chemical and stable carbon isotopic composition of PM _{2.5} from
2	on-road vehicle emissions in the PRD region and implication for
3	vehicle emission control policy
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19 Abstract

Vehicle emission is a major source of urban air pollution. In recent decade, the 20 21 Chinese government has introduced a range of policies to reduce the vehicle emission. 22 In order to understand the chemical characteristics of PM2.5 from on-road vehicle 23 emission in the Pearl River Delta (PRD) region and to evaluate the effectiveness of 24 control policies on vehicles emission, the emission factors of PM_{2.5} mass, elemental carbon (EC), organic carbon (OC), water-soluble organic carbon (WSOC), 25 26 water-soluble inorganic ions (WSII), metal elements, organic compounds and stable 27 carbon isotopic composition were measured in the Zhujiang Tunnel of Guangzhou, in the PRD region of China in 2013. Emission factors of PM_{2.5} mass, OC, EC, and 28 WSOC were 92.4, 16.7, 16.4, and 1.31 mg vehicle⁻¹ km⁻¹ respectively. Emission 29 factors of WSII were 0.016 (F⁻) ~ 4.17 (Cl⁻) mg vehicle⁻¹ km⁻¹, totally contributing 30 31 about 9.8% to the PM_{2.5} emission. The sum of 27 measured metal elements accounted 32 for 15.2% of the PM_{2.5} emission. Fe was the most abundant metal element, with an emission factor of 3.91 mg vehicle⁻¹ km⁻¹. Emission factors of organic compounds 33 34 including n-alkanes, PAHs, hopanes, and steranes were 91.9, 5.02, 32.0 and 7.59 µg vehicle⁻¹ km⁻¹, respectively. Stable carbon isotopic composition δ^{13} C value was 35 36 measured and it was -25.0% on average. An isotopic fractionation of 3.2% was found 37 during fuel combustion. Compared with a previous study in Zhujiang Tunnel in year 2004, emission factors of PM_{2.5} mass, EC, OC, WSII except Cl⁻, and organic 38 compounds decreased by 16.0-93.4%, which could be attributed to emission control 39

40 policy from 2004 to 2013. However, emission factors of most of the metal elements 41 increased significantly, which could be partially attributed to the changes in motor oil 42 additives and vehicle conditions. There are no mandatory national standards to limit metal content from vehicle emission, which should be a concern of the government. A 43 44 snapshot of the 2013 characteristic emission of PM2.5 and its constituents from 45 on-road vehicular fleet in the PRD region retrieved from our study would be helpful for the assessment of past and future implementation of vehicle emission control 46 47 policy.

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Keywords: Tunnel; PM_{2.5}; vehicle emission; emission factor; water-soluble organic
carbon; stable carbon isotope; PRD

51 **1. Introduction**

52 Vehicle emission is a major source of urban air pollution and it accounts for 53 approximately 14~50% of total fine particle mass in urban areas (Sheesley et al., 2007; 54 Wang et al., 2008; Yu et al., 2013). The environmental and health effects of vehicle 55 emission have been our concern during the last decades. Numerous studies have been 56 conducted to characterize vehicular particulate matter (PM) emission in many 57 countries, with respect to emission factors, chemical composition, and size 58 distribution (Chiang and Huang, 2009; Laschober et al., 2004; Pio et al., 2013). The 59 characteristics of vehicle emission in China were studied by tunnel experiments, dynamometer tests and/or road monitoring (He et al., 2008; Jin et al., 2014; Song et 60 al., 2012). Because of the differences in fuel qualities, engine conditions, and 61 62 operation practices, the PM emission from vehicles varied from region to region and 63 time to time.

64 Pearl River Delta (PRD) region, located in the southern coast of China has experienced serious atmospheric pollution with its rapid urbanization and 65 industrialization in the last few decades. Vehicle emission accounts for approximately 66 67 25-30% of fine PM in PRD total the region 68 (http://epaper.southcn.com/nfdaily/html/2014-01/03/content 7261687.htm). Peer 69 reviewed papers had reported emission factors and chemical characteristic of PM_{2.5} from vehicle emission in the PRD region, by means of tunnel studies in Zhujiang 70 71 Tunnel (Guangzhou) and Wutong Tunnel (Shenzhen) (He et al., 2006; He et al., 2008; 72 Huang et al., 2006b). However, the sampling in these studies was conducted in 2004. During the past decade, the Environment Protect Agency of Guangdong Province 73 74 revised the "Motor vehicle exhaust pollution prevention and control regulations of 75 Guangdong Province" in 2008 and released the "PRD Regional Air Quality Management Plan" and "A Clean Air Plan" in 2010, to improve the relevant air 76 77 quality through policies and measures. The emission standards for newly registered vehicles were tightened to China IV and the better quality of gasoline and diesel were 78 79 supplied in 2013. Therefore, the characteristics of PM emission from vehicles in the 80 PRD region might have changed throughout these years.

Tunnel experiments and chassis dynamometer tests were widely used to measure various pollutants emitted from vehicles (He et al., 2006; Heeb et al., 2003). However, dynamometer test has the defect that it cannot account for vehicle fleet composition and emissions characteristics related to break and tire wear and re-suspension of road dust (Thorpe and Harrison, 2008). Tunnel studies have been demonstrated to be a suitable setup to measure PM emissions from on-road mixed fleets (Chiang and Huang, 2009; Laschober et al., 2004; Pio et al., 2013).

This study was carried out in a roadway tunnel located in the PRD region. We report here the emission factors of $PM_{2.5}$ mass, organic carbon (OC), elemental carbon (EC), water-soluble inorganic ions (WSII), metal elements, water-soluble organic carbon (WSOC), organic compounds and stable carbon isotope. WSOC has the potential to modify the hygroscopicity of particles, PM size and cloud condensation

93	nuclei activities (Shulman et al., 1996), however, it is often ignored in previous
94	studies owing to the hydrophobic nature of the organic aerosol from primary vehicle
95	emission. Stable carbon isotope (δ^{13} C) is very useful for tracing sources
96	(Lopez-Veneroni, 2009; Widory, 2006), and it was also less reported for vehicular
97	exhaust emissions (Ancelet et al., 2011; Widory, 2006). The objectives of this study
98	are: (1) to obtain comprehensive information on the chemical and stable carbon
99	isotopic composition of $PM_{2.5}$ emission from on-road vehicles in the PRD region; (2)
100	to compare our results with the previous study conducted in the same tunnel in 2004;
101	(3) to evaluate the effectiveness of the implementation of vehicle emission control
102	policies from 2004 to 2013 in the PRD region. Although the fleet composition in this
103	tunnel was probably different from the vehicle composition in the PRD region, it does
104	not affect the conclusions in this paper.

106 **2. Experimental**

107 2.1. Tunnel sampling

108 $PM_{2.5}$ samples were collected during August 10 to 14, 2013 from the roadway 109 tunnel (Zhujiang Tunnel) located in Guangzhou City, China. It has two bores, each of 110 which has three lanes with traffic in the same direction, as shown in Fig. 1. Two 111 high-volume $PM_{2.5}$ samplers (GUV-15HBL1, Thermo, USA) were placed at a 112 distance of 75 m from the entrance and 75 m from the exit, respectively. The vehicle 113 speed in the Zhujiang Tunnel was 18 to 45 km h⁻¹, with an average vehicle speed of

33.4 km h⁻¹ during the sampling. The PM samples were collect at about 1.13 m³ min⁻¹ 114 115 through the quartz fiber filters (QFFs, 20.3 cm×25.4 cm, Whatman). Other devices 116 such as diffusion denuders and foam plugs were not used due to the difficulties in 117 applying these devices. Consequently, volatilization losses or adsorption artifacts may occur on the filter for semi-volatile organic compounds especially for the low 118 119 molecular weight compounds due to their high volatility (Kavouras et al., 1999). However, the calculation of emission factors was based on the concentration 120 121 differences between the exit and entrance of the tunnel, the potential losses or 122 adsorption artifacts of semi-volatile organic compounds would be partly deducted. 123 Field blank samples were also collected by loading filters into the samplers but 124 without pulling air through. The ventilation system of the tunnel was turned off during 125 the sampling period, thus the dispersion of air pollutants in the tunnel was mainly 126 brought from the piston effect arising from the traffic flow. The sampled filters were wrapped with annealed aluminum foil and stored in a refrigerator at -40 °C till 127 128 analysis. The meteorological parameters were synchronously recorded. A video 129 camera was placed at the exit to record the passing vehicles during the sampling 130 periods. The videotapes were then used to determine the vehicle counts and to classify the vehicles into three categories, namely, diesel vehicles (DV) (heavy-duty trucks, 131 light-duty trucks and large passenger cars), gasoline vehicles (GV) (small cars and 132 motorcycles), and liquefied petroleum gas vehicles (LPGV) (bus and taxies). The 133 average traffic density during sampling was 1797 per hour with DV, GV and LPGV 134

proportion of $13.7 \pm 2.7\%$, $59.8 \pm 8.8\%$ and $26.5 \pm 7.9\%$ respectively. More details of the vehicle counts and meteorological conditions are summarized in Table S1 of the Supplement.

138

139 2.2 Chemical analysis

The PM_{2.5} mass concentrations were determined gravimetrically by weighing the quartz filters before and after sampling. The samples were conditioned in an electronic hygrothermostat for 24 h at 25 °C and 50% relative humidity (RH) before weighing. Then, samples were analyzed for OC/EC, WSOC, WSII, metal elements, organic compounds and stable carbon isotope. The experimental methods of the chemical analysis are available in the Supplement.

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147 2.3 Calculation of emission factor

Average emission factor (*EF*) was calculated for each sampling period on the basis of the concentration difference between the exit and entrance of the tunnel by the following equation (Handler et al., 2008):

$$EF = (C_{out} - C_{in})V/NL$$

where EF is the emission factor of a species in unit of mg vehicle⁻¹ km⁻¹, N is the number of vehicles passing through the tunnel, L is the distance between inlet and outlet sampling locations, C_{out} and C_{in} are the measured species concentration at the tunnel outlet and inlet, respectively, and V is the corresponding air volume calculated

156	from the cross-sectional area of the tunnel, the average wind speed, and the sampling
157	duration of each filter. The average concentrations of all measured species at the inlet
158	and outlet sampling locations and the corresponding emission factors in this study are
159	presented in Tables S2-4 of the Supplement.
160	
161	3. Results and discussion
162	3.1 Characteristics of $PM_{2.5}$ emissions from vehicles in the PRD region
163	3.1.1 PM _{2.5} mass, OC, EC, WSOC, WSII, metal elements
164	The $PM_{2.5}$ mass emission factors ranged from 79.8 to 107 mg vehicle ⁻¹ km ⁻¹ ,
165	with an average of 92.4 ± 8.9 mg vehicle ⁻¹ km ⁻¹ . Average OC and EC emission factors
166	were 16.7 \pm 1.9 and 16.4 \pm 2.1 mg vehicle ⁻¹ km ⁻¹ , respectively, and they accounted for
167	$19\pm2\%$ and $18\pm2\%$ of $PM_{2.5}$ mass emission. The ratio of OC to EC in the Zhujiang
168	Tunnel ranged from 0.77 to 1.35, with an average of 1.03. Previous studies have
169	shown that the OC/EC ratio is useful to separate gasoline engine emissions from
170	diesel emissions. Higher values (> 2) are associated with GV and LPGV exhaust, and
171	lower values (0.3 to \sim 0.9) associated with DV exhaust (Cadle et al., 1999; Cheng et
172	al., 2010; Gillies and Gertler, 2000). Therefore, the low OC/EC ratios in this study,
173	which are closer to that from DV exhaust, indicate that diesel vehicles played an

- 174 important role in the $PM_{2.5}$ emission although the proportion of DV was only 13.7%
- 175 during the sampling. Additionally, it should be noted that emissions of EC from heavy
- 176 duty trucks are expected to be relatively low under the low speed operating conditions

177	in the tunnel (Kweon et al., 2002). Therefore, the ratio could be lower at the actual
178	driving condition of vehicle fleet with a higher speed on the road. The concentration
179	of WSOC at the inlet was 6.21 μg m $^{-3}$ (Table S2 of the Supplement) with a percentage
180	of 31.1% of OC, which is close to that of ambient air (Ding et al., 2008; Ho et al.,
181	2006). At the outlet of the tunnel, the concentration of WSOC was 8.00 $\mu g\ m^{-3},$
182	representing 17.9% of OC. The WSOC had been reported to contribute on average 20%
183	to OC in the exit of Marseille roadway tunnel (El Haddad et al., 2009), in which
184	background influence was included. The calculated emission factor of WSOC in this
185	study ranged from 0.5 to 2.8 mg vehicle ⁻¹ km ⁻¹ , with an average of 1.31 mg vehicle ⁻¹
186	km ⁻¹ , which consisted of 7.84% of OC. Such a WSOC fraction is considerably lower
187	than that previously measured for biomass burning particles (71%) (Mayol-Bracero et
188	al., 2002). However, it could influence the hygroscopicity of particles and the
189	formation of secondary aerosols (Ho et al., 2006; Rogge et al., 1993b; Weber et al.,
190	2007) and is worthy of more attention and in-depth research.

The sum of WSII comprised about 9.8% of the $PM_{2.5}$ emission, with emission factors of 4.17, 0.104, 0.609, 2.88, 0.165, 0.177 and 0.953 mg vehicle⁻¹ km⁻¹ for Cl⁻, NO₃⁻, SO₄²⁻, Na⁺, NH₄⁺, Mg²⁺ and Ca²⁺, respectively. The other WSII had a minor contribution (< 0.1 mg vehicle⁻¹ km⁻¹). Totally 27 measured metal elements contributed 15.2% to the PM_{2.5} emission. Fe was the most abundant element, with an emission factor of 3.91 mg vehicle⁻¹ km⁻¹, followed by Na 3.53 mg vehicle⁻¹ km⁻¹, Al 3.15 mg vehicle⁻¹ km⁻¹, Ca 1.93 mg vehicle⁻¹ km⁻¹, Mg 0.496 mg vehicle⁻¹ km⁻¹, and K 198 0.338 mg vehicle⁻¹ km⁻¹, which accounted for 4.2%, 3.8%, 3.4%, 2.1%, 0.5%, and 0.4% 199 of PM_{2.5} mass emission respectively. These six elements contributed 95.0% to the 200 total metal emission. Emission factors of other metals ranged from 0.0001 (Ag) to 0.25 (Ba) mg vehicle⁻¹ km⁻¹, with a sum of 0.71 mg vehicle⁻¹ km⁻¹. It is worth noting 201 that emission factors of elements including Na, K, Mg and Ca were significantly 202 203 higher than that of their corresponding water-soluble parts (Table S3 of the Supplement). The differences can be attributed to the water-insoluble matter with 204 205 these metal elements, such as calcium and magnesium carbonates and Na-K-Mg 206 bearing aluminosilicate species (Pio et al., 2013).

207 PM_{2.5} mass was also obtained by summing OM, EC, geological component, sea salt, and major water soluble inorganic ions (NH4⁺, SO4²⁻, NO3⁻). OC was multiplied 208 209 by 1.4 to estimate mass of OM (He et al., 2008). The geological component of 35 mg vehicle⁻¹ km⁻¹ was estimated based on the Al emission data as present in Table S3 of 210 211 the Supplement. A typical road dust Al composition is 9% on average (Tiittanen et al., 1999). Sea salt of 9 mg vehicle⁻¹ km⁻¹ was estimated by Na assuming sea salt contains 212 213 32% of Na. Thus, the average PM_{2.5} reconstructed mass was 91.8% of the gravimetric 214 value. This discrepancy can be attributed to the uncertainties in the weighing process, 215 the estimation methods and uncalculated components.

216

217 *3.1.2 Organic compounds*

218 The average emission factors and abbreviated names of 67 individual organic

219 compounds identified in the Zhujiang Tunnel, including n-alkanes, polycyclic 220 aromatic hydrocarbons (PAHs), hopanes, and steranes are listed in Table S4 of the 221 Supplement. These organic compounds accounted for 0.59% of the OM and 0.11% of 222 the PM_{2.5} mass emissions, The distributions of organic molecular markers associated 223 with PM_{2.5} are known to be source indicative despite of their small mass fractions 224 (Schauer et al., 1996; Simoneit, 1986). n-Alkanes are an important class of organic compounds in atmospheric aerosols, and their homologue distribution may indicate 225 226 different pollution sources (Rogge et al., 1993a). In this study, the *n*-alkane traces 227 were dominated by C11-C36 with no odd-even carbon number predominance and the 228 maximum was at C24, consistent with the characteristics of vehicle emissions 229 reported by Simoneit (Simoneit, 1984, 1985). The emission factors of of individual n-alkanes were in the range of 0.22 (C13)~ 13.3 (C24) μ g vehicle⁻¹ km⁻¹ (Table S4 of 230 231 the Supplement).

232 There has been a worldwide concern to PAHs due to their known carcinogenic 233 and mutagenic properties. PAHs are thought to be the result of incomplete combustion. 234 Totally 15 priority PAHs (the results of naphthalene have not been discussed in this 235 study due to its low recovery) were identified and quantified. The emission factor of total PAHs varied from 4.56 to 5.54 μ g vehicle⁻¹ km⁻¹ in this study. The emission 236 factor of benzo[a]pyrene (BaP), which is often used as an indicator of PAHs and 237 regarded by World Health Organization as a good index for whole PAHs 238 carcinogenicity, was in the range of 0.37 to 0.46 µg vehicle⁻¹ km⁻¹. The emission 239

240	factors for other compounds ranged from 0.006 (acenaphthene) to 0.89 (pyrene) μg
241	vehicle ⁻¹ km ⁻¹ (Table S4 of the Supplement). Pyrene was the most abundant
242	compound, followed by chrysene (CHR), benzo[ghi]perylene (BghiP) and
243	benz[a]anthracene (BaA), which is different from biomass burning and coal
244	combustion (Huang et al., 2014; Shen et al., 2012). PAHs diagnostic ratios have been
245	used as a tool for identifying pollution emission sources including ANT/(ANT+PHE),
246	FLA/(FLA+PYR), BaA/(BaA+CHR), BbF/(BbF+BkF), IcdP/(IcdP+BghiP) and
247	BaP/(BaP+BghiP) (Tobiszewski and Namiesnik, 2012; Yunker et al., 2002; Zhang et
248	al., 2005). We summarized PAHs ratios mentioned above in Fig. 2 for three
249	combustion sources including vehicle emission, biomass burning and coal combustion.
250	On the whole, the six ratios in this study are similar to the other tunnel experiments,
251	though environmental conditions of tunnels are different to some extent. It is also
252	suggested that the ratio of FLA/(FLA+PYR) and IcdP/(IcdP+BghiP) might be more
253	suitable to distinguish vehicle emission from biomass burning and coal combustion.
254	Hopanes and steranes are known molecular markers of aerosol emissions from
255	fossil fuel utilization (Simoneit, 1985). Rogge et al. (1993a) and Schauer et al. (1996)
256	had shown that these petroleum biomarkers can be used to trace motor vehicle
257	exhaust contributions to airborne PM in southern California atmosphere. Fourteen
258	major hopanes homologues with emission factors ranging from 0.46 \sim 9.14 μg
259	vehicle ⁻¹ km ⁻¹ and twelve steranes homologues ranging from $0.31 \sim 0.97 \ \mu g \ vehicle^{-1}$
260	km ⁻¹ were identified in this study. $17\alpha(H)$, $21\beta(H)$ -hopane (HP30) was the most

abundant component with the emission factor of 9.14 μ g vehicle⁻¹ km⁻¹. The emission 261 factor of total hopanes was 32.0 µg vehicle⁻¹ km⁻¹. Emissions of the S hopanes for the 262 263 extended $17\alpha(H), 21\beta(H)$ -hopane homologues > C31 were always higher than those of 264 the corresponding R pairs. All these characteristics of hopanes in the Zhujiang Tunnel 265 are consistent with those in gasoline and diesel exhausts (Rogge et al., 1993a; 266 Simoneit, 1985) and in other tunnel studies (see Fig. S1 of the Supplement). Emission factors of individual sterane ranged from 0.31 to 0.97 ng vehicle⁻¹ km⁻¹, and the sum 267 of their emission factors was 7.58 µg vehicle⁻¹ km⁻¹. The most abundant homologue 268 269 was C29 $\alpha\beta\beta$ -stigmastane (20R) (29 $\alpha\beta\beta$ R), followed by 29 $\alpha\alpha\alpha$ S and 29 $\alpha\beta\beta$ S.

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271 *3.1.3 Stable carbon isotope*

272 Stable carbon isotope analysis of vehicle emissions in Zhujiang Tunnel yielded δ^{13} C values ranging from -25.5 to -24.7‰ with an average value of -25.0 ± 0.2‰, and 273 274 is comparable to previously reported ranges of -29 to -24.6‰ (Table 1) for vehicular 275 fuel emission. Generally, the variation in $\delta^{13}C_{Fuel}$ could affect the $\delta^{13}C$ of 276 hydrocarbons (Keppler et al., 2004; Yamada et al., 2009). In the PRD region, the δ^{13} C value of gasoline was on average -28.6 \pm 0.6‰, and the δ^{13} C value of diesel was – 277 278 $27.8 \pm 0.2\%$, and small variation of fuel δ^{13} C was observed (Hu et al., 2014). We calculated the isotopic differences between $\delta^{13}C_{PM2.5}$ and $\delta^{13}C_{Fuel}$, which represents 279 the apparent isotopic fractionation occurring during fuel burning. It expressed as 280 Δ^{13} C (‰), and is defined by the following equation (Yamada et al., 2009). 281

282
$$\Delta^{13}C_{PM2.5-Fuel} = \left(\frac{\delta^{13}C_{PM2.5} + 1000}{\delta^{13}C_{Fuel} + 1000} - 1\right) \times 1000$$

283 In this study, the value of $\Delta^{13}C_{PM2.5-Fule}$ was from 2.7 to 3.5 ‰, with an average of 3.2 ‰, indicating an isotopic fractionation occurred during fuel combustion. 284 285 Comparing the stable isotopic carbon value of vehicular fuel emission with other 286 particulate emission sources (see Table 1), it is found that different emission sources 287 showed different stable carbon isotopic composition. For total carbon in PM_{2.5} sample, δ^{13} C (‰) of coal and fuel oil combustion are -23.9‰ and -26.0‰ respectively, while 288 that of vehicle emission is -25.9~-25.0%. Obviously, the $\delta^{13}C$ (%) of vehicle 289 290 emission is not significantly different from that of coal and fuel oil combustion. 291 However, they are obviously different from other sources, like dust particle (-21~-18.4‰), C3 plant (-19.3~-13‰), and C4 plant (-34.7~-27‰). Therefore, δ^{13} C 292 293 might be used to distinguish the fossil fuel combustion from the other sources.

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295 **3.2** Comparison with previous studies conducted in the same tunnel

To investigate the variation of chemical emission characteristics from vehicles in the PRD region over the past decade, we compared the chemical emission characteristics of this study with that of previous study (He et al., 2008) for the same tunnel in 2004 (see Figs 3 and 4). Fig. 3 shows that $PM_{2.5}$ mass, OC, EC decreased significantly from 2004 to 2013. The reason can be partly attributed to the implementation of pollution control measures for Chinese vehicle emission. During this 9-year period, vehicle emission standard have raised two levels (from China II in 303 2004 to China IV in 2013) (Table 2). Additionally, comparing the fleet composition of 2013 with 2004 in Zhujiang Tunnel, we found that the proportion of DV and GV 304 305 decreased while that of LPGV increased. LPG is a type of clean energy, and LPGV is 306 known to emit much less PM mass than GV and DV, while DV emits the most PM mass (Allen et al., 2001; Myung et al., 2014; Yang et al., 2007). LPG could be 307 308 combusted more completely than gasoline and diesel. Changes mentioned above contributed greatly to the decrease of emission factors of OC and EC (31.3% and 309 310 66.9%) and PM_{2.5} mass (16.0%) from 2004 to 2013. However, the emissions of PM_{2.5} 311 mass, OC and EC are still quite higher than those measured in other countries (see 312 Table S5 of the Supplement). The implication of these high emission levels is that 313 both the fuel quality and engine technologies in the PRD region need to be further 314 improved.

It is also found from Fig. 3 that emission factors of NO_3^- , SO_4^{2-} and NH_4^+ 315 decreased from 2004 to 2013. Improvement of fuel quality resulted in decreasing of 316 sulfate emission factor from 3.18 to 0.61 mg vehicle⁻¹ km⁻¹, since the amount of sulfur 317 in fuel is slashed by 81.5~95 % in China IV (2013) when compared that in China II 318 319 (2004) (Table 3). The emission levels of nitrate and ammonium were about one-tenth 320 of those observed in 2004, possibly because NOx emission standard is tightened from 321 2004 to 2013 (Table 2) leading to the less production of ammonium nitrate. Emission factor of chloride is quite higher than that obtained from Zhujiang Tunnel in 2004 and 322 other tunnels. Chloride was found up to 74 mg vehicle⁻¹ km⁻¹ in PM₁₀ in the Howell 323

Tunnel, due to the application of salt to melt ice on roadways in the winter (Lough et al., 2005). However, it is not applicable in Guangzhou City. The good correlation between Cl⁻ and Na⁺ ($r^2=0.992$) indicates the re-suspension of sea salt particles combined with vehicle emission PM might be a major source (He et al., 2008).

328 Emission factors of most of the metal elements increased in Zhujiang Tunnel from 2004 to 2013 except Cd and Pb. Na emission increased 3.18 mg vehicle⁻¹ km⁻¹ in 329 2013 than in 2004. Na correlated weakly with Cl⁻ ($r^2=0.374$) and Na⁺ ($r^2=0.429$). This 330 331 indicates that Na emission had other sources and it was not only from the 332 re-suspension of sea salt particles. The other four most abundant elements including 333 Fe, Ca, Mg, K increased by one to three times, probably because of resuspended road 334 dust. However, the wind speed in 2013 was found to be higher but not significantly than that in 2004 (3.8 m s⁻¹ in 2013 versus 3.0 m s⁻¹ in 2004). This minor difference in 335 336 wind speed could not account for the large increase. Furthermore, examination of the 337 number of vehicles per hour in 2013 and 2004 suggests that there were fewer vehicles 338 per hour in 2013. Therefore, a more plausible explanation is that there was a lot more 339 dust on the road in 2013. Other sources would cause the increased emissions of these 340 elements, such as oil additive (Mg, Ca, Cu, Zn. (Cadle et al., 1997)) and the wear of engines (Fe) (Cadle et al., 1997; Garg et al., 2000), brakes and tires (Al, Fe,Cu, Mn, 341 342 Cd, Ni, Pb and Zn. (Garg et al., 2000; Pio et al., 2013)). Additionally, emissions of Zn, Cu, Mn, Cr, Ni, V, As, Co, U, and Tl increased by 0.5 to 4.5 times. Although the sum 343 of these elements did not exceed 0.5% of PM2.5 mass, they are important for health 344

effects. Lower emission factor of Pb $(0.01\pm0.0007 \text{ mg vehicle}^{-1} \text{ km}^{-1})$ in 2013 than in 2004, could be a result of the phase out of leaded gasoline across China in the late 1990s.

348 Figure 4 shows comparison of organic compounds emission in Zhujiang Tunnel 349 between 2004 and 2013. The n-alkane homologues exhibited a smooth hump-like 350 distribution with the most abundance at C24, as shown in Fig. 4a. Such a distribution 351 pattern was similar to patterns observed in Zhujiang Tunnel 2004. However, there are 352 some differences. Firstly, the highest abundant *n*-alkane shifted from C23 in 2004 to 353 C24 in 2013. This difference might be explained by the shift of gas-particle 354 partitioning as alkanes of < C26 are semi-volatile. However, T-test showed that the 355 temperatures were not significant different (p = 0.14) between this study ($33.0 \pm 2.3^{\circ}$ C) 356 and that in 2004 (31.8 \pm 1.0 °C). Thus the differences due to different C_{max} between 357 this study and the study in 2004 cannot be regarded as a result of temperature differences. Furthermore, C_{max} was found to be C24 in every test of this study, 358 although the temperature ranged from 28.6 to 36.1 °C. It was reported that the 359 *n*-alkane in the highest abundance was C20 for DV and C25 or C26 for GV in 360 361 dynamometer tests (Rogge et al., 1993a; Schauer et al., 1999, 2002). As the emissions 362 collected in tunnel studies present a composite result of emissions from a mixed 363 vehicle fleet, the lower fraction of DV in 2013 was more likely the cause for the shift of the C_{max}. Secondly, emission factors of C16-C26 in 2013 were quite lower than 364 365 those in 2004, while this trend reversed gradually after C27. Emission factors of the

PAHs decreased by 67.6% ~ 93.4%. BaP equivalents (BaPeq) emission factors 366 decreased by 88.1% from 2004 to 2013 (Table S6 of the Supplement). This could be 367 368 attributed to the variation of fleet composition between 2004 and 2013. PAHs emitted 369 from LPGV are about one-third of that from GV (Yang et al., 2007), while DV emit more PAHs than GV (Phuleria et al., 2006). Therefore, higher proportion of LPGV 370 371 and lower proportion of DV resulted in the lower emission factor of PAHs in 2013 than that in 2004. Emission factors of hopanes also decreased from 2004 to 2013, the 372 373 percentage of decrease ranged from 56.2% to 68.7%. However, the distributions of 374 hopane series derived from different tunnel studies were very similar (see Fig. S2 of 375 the Supplement). This suggests that the hopane emission characteristics might be independent of the fleet composition. This is a reasonable result given that hopanes 376 377 originate from the lubricating oil used in DV, GV and LPGV rather than from the fuel 378 (He et al., 2008; Phuleria et al., 2006). Owing to more units in heavy-duty vehicle 379 need lubrication, emission factors of hopanes attributable to heavy-duty vehicle were 380 higher than that to light-duty vehicle (Phuleria et al., 2006). Reduction of the 381 proportion of heavy-duty vehicle (bus, heavy-duty truck, large passenger cars) 382 proportion in fleet composition in 2013 (11.3%) compared that in 2004 (20%) might be the reason that emission factors of hopanes decreased. 383

384

385 3.3 Implication for vehicle emission control policy

386 Vehicle emission control strategies and policies adopted by Guangdong Province

387 can be classified as emission control on vehicles and fuel quality improvements or alternative fuel utilization. PM emission standards for newly registered vehicles were 388 389 tightened from China II in 2004 to China IV in 2013 (Table 2). The reduction of on 390 road high PM emitting vehicles, the phasing of lower PM emitting vehicles, and more 391 environmental friendly vehicles on road with better advanced engines following the 392 implementation of these emission standards were effective for decreasing PM emission. Emission factors of PM decreased by 16% from 2004 to 2013. Also for 393 394 NOx, the emission limit was reduced to about half from 2004 to 2013. This change in 395 emission standard which limit NOx emission was a major factor for emission factors 396 of nitrate and ammonium decreased by about 90%. On the other hand, National 397 Standard has been revised several times to improve fuel quality to adapt to stringent 398 vehicle emission standards (Table 3). Taken sulfur content as an example, it has a 399 sharp decrease by over 90% from 2004 to 2013, resulting in the decrease of emission 400 factor of sulfate by 70%. Additionally, LPG and liquefied natural gas (LNG) are 401 gradually taking the place of diesel and gasoline as the fuel of taxi and bus after 2004, 402 and now seldom taxi and bus use diesel and gasoline fuel as 403 (http://www.southcn.com/news/gdnews/nanyuedadi/200707040173.htm). The 404 application of clean fuel led to nearly complete combustion, and resulted in much less 405 emission from taxis and buses. In general, our results suggest that these strategies are 406 effective to reduce emission factors of PM_{2.5} mass, as well as OC, EC, WSII, and 407 organic compounds in PM_{2.5}. However, the total vehicle population increased year by

year. As shown in Fig. 5a, the total vehicle population increased by 49.1% from 2004 408 to 2013. Total emission of vehicle exhaust of PM2.5 mass (calculated as emission 409 410 factors multiply by annual average driving distance per car and vehicle population 411 (Wu et al., 2012)) increased by 25.2% from 2004 to 2013 (Fig. 5b). Consequently, it 412 is demonstrated that more stringent emission standards and higher quality of fuel or 413 more utilization of clean fuels will be necessary to offset the impacts on the growth in 414 vehicle population and to improve air quality in the PRD region. Additionally, owing 415 to no mandatory national standards to limit metal content from vehicle emission, the 416 emission of majority of metals increased from 2004 to 2013 (Fig. 3 and 5b). In China, 417 heavy metals, including As, Cr, Cu, Ni and Tl, had been listed as key substances that 418 should be preferentially monitored in the atmospheric environment (SEPA, 2003) and 419 thus the increases of metal elements should raise the awareness of the government due 420 to their health concern.

421

422 4. Conclusions

PM_{2.5} samples were collected between August 10 and 14, 2013 in Guangzhou Zhujiang Tunnel to acquire a comprehensive snapshot of the chemical characteristics of vehicle emission. The average emission factors of PM_{2.5} mass, EC, OC, WSOC, WSII, metal elements, organic compounds and stable carbon isotope were measured. Stable carbon isotope δ^{13} C value indicates an isotopic fractionation of 3.2‰ occurred during fuel combustion. Compared with a previous study in Zhujiang Tunnel in year 429 2004, emission factors of PM2.5 mass, EC, OC, and major WSII decreased due to control policy induced changes throughout the nine years from 2004 to 2013, that is, 430 change of fleet composition, implementation of more stringent gasoline and diesel 431 432 emission standards, improvement in fuel quality, and clean fuel used for taxi and 433 buses. The shift in n-alkanes distribution and decreased PAHs emission were due to 434 the lower proportion of DV in 2013 than in 2004, and the decrease in emission factor of hopanes was due to the reduction of proportion of heavy duty vehicles. Our study 435 436 shows that control polices for vehicles emission by the government were effective to 437 decrease the emission factors of PM2.5, EC, OC, and WSII from on-road vehicular 438 fleets. However, the increase in emission of most metal elements should raise the 439 awareness of the government, since metal elements, especially heavy metals could 440 affect human health. Also, in order to offset the impacts in the growth of vehicle 441 population and to improve air quality in the PRD region, more stringent emission and 442 aggressive control policies are necessary.

443

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Table 1. δ^{13} C values (‰) of PM from vehicle emission in this study and other emission sources

Emission sources and Sampling site	Particle types	δ^{13} C values	Sampling time	Reference
Vehicular fuel emission				
Vehicle emissions (Zhujiang Tunnel, China)	$PM_{2.5}/TC$	-25.0 ± 0.3	August 2013	This study
Vehicle emissions (Tunnel of Rio de Janeiro, Brazil)	PM/OC	-25.4	April 1985	(Tanner and Miguel, 1989)
Vehicle emissions (Tunnel of Rio de Janeiro, Brazil)	PM/EC	-24.8	April 1985	(Tanner and Miguel, 1989)
Complete combustion of diesel	PM/TC	-29	N/A	(Widory, 2006)
Complete combustion of gasoline	PM/TC	-27	N/A	(Widory, 2006)
Vehicle emissions (Cassier Tunnel, Canada)	PM _{2.5} /OC	-27.1	N/A	(Huang et al., 2006a)
Vehicle emissions (Cassier Tunnel, Canada)	PM _{2.5} /EC	-26.9	N/A	(Huang et al., 2006a)
Diesel vehicle emissions (Central Camionera del Norte, Mexico)	$PM_{2.5}/TC$	-24.6 ± 0.3	March 2002	(Lopez-Veneroni, 2009)
Gasoline vehicle emissions (Tunnel of Avenida Chapultepec,	PM ₂₅ /TC	-25 5+0 1	March 2002	(Lopez-Veneroni 2009)
Mexico)	1 1012.5/ 1 C	25.5-0.1		(Lopez Veneroni, 2009)
Vehicle emissions (Mount Victoria Tunnel, New Zealand)	$PM_{2.5}/TC$	-25.9 ± 0.8	December 2008 to March 2009	(Ancelet et al., 2011)
Non-vehicular fuel sources				
Coal combustion (Paris, Franch)	PM _{2.5} /TC	-23.9 ± 0.5	May to September 2002	(Widory et al., 2004)
Coal combustion (Yurihonjo City, Japan)	PM _{2.5} /EC	-23.3	N/A	(Kawashima and Haneishi, 2012)
Charcoal combustion (Yurihonjo City, Japan)	PM _{2.5} /EC	-27.4±1.7	N/A	(Kawashima and Haneishi, 2012)
Fireplace soot (Yurihonjo City, Japan)	PM/EC	-26.5 ± 0.1	N/A	(Kawashima and Haneishi, 2012)
Fuel oil combustion (Paris, Franch)	PM _{2.5} /TC	-26.0 ± 0.5	May to September 2002	(Widory et al., 2004)
Dust particles				
Street dust (Mexico City, Mexico)	$PM_{2.5}/TC$	-21±0.2	March 2002	(Lopez-Veneroni, 2009)
Street dust (Yurihonjo City, Japan)	PM _{2.5} /EC	-18.4~ -16.4	November 2009	(Kawashima and Haneishi, 2012)
Biomass burning				
C4 Plant	PM/TC	-13±4	N/A	(Boutton, 1991)
C4 Plant (Yurihonjo City, Japan)	PM _{2.5} /EC	-19.3~ -16.1	April to November 2009	(Kawashima and Haneishi, 2012)
C3 Plant	PM/TC	-27±6	N/A	(Boutton, 1991)
C3 Plant (Yurihonjo City, Japan)	PM _{2.5} /EC	-34.7~ -28.0	April to November 2009	(Kawashima and Haneishi, 2012)

Emission	Vear ^{<i>a</i>}	Limit f	for PM	Limit fo	Limit for NO _x	
standard	i cai	g km ^{-1 b}	g kWh ^{-1 c}	g km ^{-1 b}	g kWh ⁻¹ ^c	
China I	2001	0.14~0.40	0.40~ 0.68	-	8.0~9.0	
China II	2004	0.08~0.20	0.15	-	7.0	
China III	2007	0.05~ 0.10	0.10~ 0.21	0.15~ 0.78	5.0	
China IV	2010	0.025~0.060	0.02~ 0.03	0.08~ 0.39	3.5	

663 **Table 2.** Vehicle emission standard and limit for PM and NO_x implemented in

665 ^{*a*} Year of implementation; ^{*b*} for light duty vehicle; ^{*c*} for compression ignition and gas fuelled positive ignition

666 engines of vehicles.

Guangzhou after 2000

Standard	China I		China II		China III		China IV	
	Limit	Year ^a	Limit	Year	Limit	Year	Limit	Year
Gasoline	1000	2001	500	2005	150	2006	50	2010
Diesel	2000	2002	500	2003	350	2010	50	2013
LPG	-	-	270^{b}	2003	-	-	50	2013

Table 3. Vehicle fuel standard and limit for sulfur content (mg kg⁻¹) implemented in

669 ^{*a*} Year of implementation; ^{*b*} unit: mg m⁻³

Guangzhou after 2000

670 Figure captions:

Fig. 1. Sampling schematic diagram of the Zhujiang Tunnel.

672

673	Fig. 2	. ANT/(ANT+PHE),	FLA/(FLA+PYR),	BaA/(BaA+CHR),	BbF/(BbF+BkF),
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674 BaP/(BaP+BghiP) and IcdP/(IcdP+BghiP) ratios for three source emissions. The

675 vehicle emission (VE) composition is from data collected in roadway tunnel((this

676 study), (He et al., 2008), (Ancelet et al., 2011), (He et al., 2006), (Ho et al., 2009),

677 (Oda et al., 2001). The biomass burning profiles are obtained from 9 straws (Shen et

al., 2011), 26 firewood (Shen et al., 2012), 3 plant leaves and branches (Sheesley et al.,

679 2003) and 2 biomass briquettes burning (Sheesley et al., 2003). The coal combustion

680 profiles are selected from 5 coals imitate combustion (Shen et al., 2011)and main

681 coal-mining regions in China (Zhang et al., 2008Zhang et al., 2008).

682

Fig. 3. Comparison of PM_{2.5}, OC, EC, WSII and metal emissions in the Zhujiang
Tunnel sampling in 2004 and 2013.

685

Fig. 4. Comparison of organic compounds emissions in the Zhujiang Tunnel samplingin 2004 and 2013.

688

Fig. 5. (a) Growth in total vehicle population in Guangzhou during 2004-2013. (b)

Total exhaust emission of PM_{2.5} mass, OC, EC, WSII and metal in 2004 and 2013.







Fig. 1











Fig. 4



Fig. 5