1	Long-term real-time chemical characterization of submicron aerosols		
2	at Montsec (Southern Pyrenees, 1570 m a.s.l.)		
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Abstract. Real-time measurements of inorganic (sulfate, nitrate, ammonium,
 chloride and black carbon (BC)) and organic submicron aerosols (aerosols with an
 aerodynamic diameter less than 1 μm) from a continental background site (Montsec,
 MSC, 1570 m a.s.l.) in the Western Mediterranean Basin (WMB) were conducted for 10
 months (July 2011 - April 2012). An Aerosol Chemical Speciation Monitor (ACSM) was
 co-located with other on-line and off-line PM<sub>1</sub> measurements. Analyses of the hourly,
 diurnal, and seasonal variations are presented here, for the first time for this region.

25 Seasonal trends in PM<sub>1</sub> components are attributed to variations in: evolution of 26 the planetary boundary layer (PBL) height, air mass origin, and meteorological 27 conditions. In summer, the higher temperature and solar radiation increases 28 convection, enhancing the growth of the PBL and the transport of anthropogenic 29 pollutants towards high altitude sites. Furthermore, the regional recirculation of air 30 masses over the WMB creates a continuous increase in the background concentrations 31 of PM<sub>1</sub> components and causes the formation of reservoir layers at relatively high 32 altitudes. The combination of all these atmospheric processes results in a high 33 variability of PM<sub>1</sub> components, with poorly defined daily patterns, except for the organic 34 aerosols (OA). OA was mostly composed (up to 90%) of oxygenated organic aerosol 35 (OOA), split in two types: semi-volatile (SV-OOA) and low-volatile (LV-OOA), the rest 36 being hydrocarbon-like OA (HOA). The marked diurnal cycles of OA components 37 regardless of the air mass origin indicates that they are not only associated with 38 anthropogenic and long-range-transported secondary OA (SOA), but also with recently-39 produced biogenic SOA.

40 Very different conditions drive the aerosol phenomenology in winter at MSC. 41 The thermal inversions and the lower vertical development of the PBL leave MSC in 42 the free troposphere most of the day, being affected by PBL air masses only after 43 midday, when the mountain breezes transport emissions from the adjacent valleys and 44 plains to the top of the mountain. This results in clear diurnal patterns of both organic and inorganic concentrations. OA was also mainly composed (71%) of OOA, with 45 46 contributions from HOA (5%) and biomass burning OA (BBOA; 24%). Moreover, in 47 winter sporadic long-range transport from mainland Europe is observed.

The results obtained in the present study highlight the importance of SOA formation processes at a remote site such as MSC, especially in summer. Additional research is needed to characterize the sources and processes of SOA formation at remote sites.

52 53

Keywords: high altitude, mountain, remote, continental background, ACSM.

### 55 **1** Introduction

56 Earth's climate system is modulated by atmospheric aerosols. Submicron 57 particles (< 1 µm in aerodynamic diameter) play a dominant role in both cloud 58 formation and scattering or absorbing solar radiation (IPCC, 2013). The complexity of 59 aerosol sources and processes results in an uncertainty in the radiative forcing of 60 climate. Aerosol optical properties are connected to direct and indirect climate forcing 61 effects, and they are dependent on particle composition. Moreover, aerosol 62 composition may provide valuable information on aerosol sources and processes. 63 Consequently, long-term measurements of PM<sub>1</sub> chemical composition are needed to 64 better understand aerosol sources, to quantify their lifetime in the atmosphere and to 65 constrain the uncertainties of their climatic influence.

66 Long-term PM<sub>1</sub> chemical composition measurements are relatively scarce both 67 off-line and on-line. In the last decade, on-line PM1 chemical composition 68 measurements have been performed using aerosol mass spectrometers (AMS) at a 69 number of locations. Measurements of on-line chemical composition are useful to study 70 hourly variations and daily patterns. Most of these studies, however, correspond to 71 short-term measurement campaigns (typically a month) (e.g. Crippa et al., 2014; 72 Jimenez et al., 2009; Lanz et al., 2010; Ng et al., 2010; Zhang et al., 2007) given the 73 intensive instrument maintenance required and the need of highly-qualified personnel 74 for a good quality dataset.

75 In contrast to the use of the AMS in relatively short campaigns, the more 76 recently developed Aerodyne Aerosol Chemical Speciation Monitor (ACSM) is 77 becoming a widely used on-line instrument for long-term measurements of PM1 78 chemical composition (Budisulistiorini et al., 2014; Canonaco et al., 2013; Petit et al., 79 2014; Tiitta et al., 2014). The ACSM is built upon the same technology as the AMS, in 80 which an aerodynamic particle focusing lens is combined with high vacuum thermal 81 particle vaporization, electron impact ionization, and mass spectrometry. Modifications 82 in the ACSM design (e.g. lack of particle sizing chamber and components, use of 83 simple and compact RGA mass spectrometer detector), however, allow it to be smaller, 84 lower cost, and simpler to operate than the AMS (Ng et al., 2011c). The ACTRIS 85 (Aerosols, Clouds, and Trace gases Research InfraStructure) European network is 86 evaluating the use of the ACSM as a reliable instrument, which will provide the 87 opportunity to study long-term datasets of PM<sub>1</sub> chemical composition across the 88 continent.

Recent publications have investigated most of the existing worldwide AMS
databases (e.g. Crippa et al., 2014; Jimenez et al., 2009; Lanz et al., 2010; Ng et al.,
2010; Zhang et al., 2007) and reflected a prevalence of organic aerosols (20 to 90%) in

92 the submicron fraction, largely independent of the region and type of environment. However, our knowledge on organic aerosol (OA) formation, sources, and atmospheric 93 94 processing is still very incomplete, especially for secondary organic aerosols (SOA) 95 formed from chemical reactions of gas-phase species (e. g. Donahue et al., 2014; 96 Hallquist et al., 2009; Kroll and Seinfeld, 2008; Robinson et al., 2007; Volkamer et al., 97 2006). Recent progress has been made in identifying primary organic aerosols (POA) 98 sources (e.g. Elbert et al., 2007; Zhang et al., 2005), but significant gaps still remain in 99 our understanding on the atmospheric evolution of POA after emission (de Gouw and 100 Jimenez, 2009). For these reasons, OA measurements and analysis are required to 101 better understand its chemical evolution in the atmosphere.

102 The lack of long-term on-line PM<sub>1</sub> chemical composition measurements is 103 especially evident in the Western Mediterranean Basin (WMB), which is characterized 104 by particular atmospheric dynamics strongly influenced by its topography (Jorba et al., 105 2013; Millan et al., 1997). Over this region, arrival of natural and anthropogenic 106 aerosols as a result of long-range transport from Africa and Europe is frequent (e.g. 107 Pey et al., 2013; Querol et al., 2009; Ripoll et al., 2014; Rodríguez et al., 2001) and 108 accumulation and recirculation processes are frequently observed (Rodriguez et al., 109 2002). The sources and meteorological controls of PM in the regional background of 110 the WMB have been recently investigated during the DAURE study (Pandolfi et al., 111 2014a) using an AMS and <sup>14</sup>C analyses (Crippa et al., 2014; Minguillón et al., 2011). 112 Furthermore, Ripoll et al. (2015) studied the  $PM_1$  and  $PM_{10}$  chemical composition with 113 daily time resolution in the continental and regional background environments in the 114 WMB. In that study, a higher mineral contribution was identified in the continental 115 background due to the preferential transport of African dust at high altitude layers and 116 to the increased regional dust resuspension enhanced by the drier surface and higher 117 convection. Nevertheless, aerosol chemical characterization with higher time resolution 118 is needed to study the origin of specific PM components and the local and/or regional 119 processes, in particular to exploit the information contained in diurnal cycles that is 120 typically not accessible with off-line measurements.

121 In this study we deployed an ACSM at a high altitude site (Montsec, 1570 m 122 a.s.l.) in the NE of the Iberian Peninsula (42° 03' N, 0° 44' E), representative of the 123 continental background conditions of the Western Mediterranean Basin (WMB) (Ripoll 124 et al., 2014). This environment is under free tropospheric (FT) influence most of the 125 time, although it is exposed to regional pollutants during the summer time and/or under 126 the influence of mountain breezes, and it is affected by trans-boundary incursions of 127 natural and anthropogenic aerosols from Europe and North Africa (Ripoll et al., 2014, 128 2015). Co-located on-line and off-line PM<sub>1</sub> measurements were also carried out to

129 complement the ACSM dataset. Hence, the work presented here interprets the real-130 time variation of inorganic and organic submicron components during 10 months (July 131 2011 - April 2012), and the types of OA are also studied. Special emphasis is placed 132 on the analysis of diurnal pattern and seasonal variations of chemical components and 133 the main factors influencing these variations.

# 134 2 Methodology

### 135 2.1 Sampling site

Montsec site (MSC) is located on the highest part of the Montsec d'Ares mountain, at an altitude of 1570 m a.s.l., in a plain near the edge of a 1000 m cliff to the south, with no wind obstructions present around. It is located in the NE of the Iberian Peninsula (42°03'N, 0°43'E), 50 km S of the Pyrenees and 140 km NW of Barcelona (Fig.S1). A detailed description of this site can be found in Ripoll et al. (2014). <u>This</u> station is a member of the GAW (global atmospheric watch) network as a regional station named MSA.

143 The daily classification of atmospheric episodes affecting MSC was made 144 following the procedure described by Ripoll et al. (2014) using HYSPLIT model from 145 the NOAA Air Resources Laboratory (ARL). Air masses reaching MSC are mainly from 146 the Atlantic (62% of the days) all along the year. From March to October, North African 147 (NAF) episodes are more frequent (17% of the days) and very often are alternated with 148 the summer regional (SREG) scenarios (12% of the days). The winter regional 149 (WREG) scenarios are detected from October to March (11% of the days), as well as 150 the European (EU) episodes (11% of the days). Conversely, the Mediterranean (MED) 151 episodes are detected sporadically (4% of the days).

152 The boundary layer height was calculated using the Global Data Assimilation 153 System (GDAS1) model from the NOAA Air Resources Laboratory 154 (http://www.ready.noaa.gov/READYamet.php) (Fig. 1).

## 155 2.2 ACSM sampling and data analysis

156 The aerosol chemical speciation monitor (ACSM) (Aerodyne Research Inc.) 157 was measuring continuously from July 2011 to April 2012. The ACSM provides real-158 time mass concentration of submicron particulate organics, nitrate, sulfate, ammonium 159 and chloride via thermal vaporization and electron impact ionization, with detection by a 160 quadrupole mass spectrometer (Ng et al., 2011c). The mass concentration of a given 161 species is determined from the sum of the ion signals at each of its mass spectral 162 fragments and its Ionization Efficiency (IE) (Canagaratna et al., 2007). Since calibration 163 of IEs for all species is not feasible, the Relative Ionization Efficiency (RIE) (compared

164 to that of nitrate) is used (Jimenez et al., 2003). The ammonium nitrate calibration 165 described by Ng et al. (2011b) was performed using an atomizer (TSI, Constant Output 166 Atomizer model 3076) for primary aerosol generation, followed by a silica gel diffusion 167 dryer, a differential mobility analyzer (DMA) model TSI 3081, and a condensation 168 particle counter (CPC, TSI 3772). Monodisperse 300 nm ammonium nitrate aerosol 169 particles were used, covering a range of nitrate concentrations from 2 to 15  $\mu$ g m<sup>-3</sup>. 170 Several calibrations were conducted throughout the sampling period, and average values of 2.2 x 10<sup>-11</sup> for nitrate IE and 5.4 for RIE for ammonium were used for the 171 172 whole dataset. The RIE values used in this study for the rest of the species were those 173 usually applied in AMS ambient concentrations: 1.4 for OA and 1.1, 1.2, and 1.3 for 174 nitrate, sulfate, and chloride, respectively (Canagaratna et al., 2007). RIE for sulfate 175 was experimentally determined one year later and was found to be 1.26, although the 176 default value was used for the current dataset. A time resolution of 62 minutes was 177 used as a result of 12 scans (1 open and 1 filtered) per data point with a scan speed of 178  $1 \, {\rm s} \, {\rm amu}^{-1}$ .

179 The ACSM data were analyzed with the standard ACSM data analysis software 180 version 1.5.3.2 (Aerodyne Research Inc.) written in Igor Pro 6 (WaveMetrics, Inc., Lake 181 Oswego, OR, USA). As the ACSM was measuring continuously for a long time, the 182 standard correction for instrument sensitivity drifts was applied to the dataset based on 183 the inlet pressure and  $N_2$  signal. Finally, mass concentrations were corrected using a 184 Collection Efficiency (CE) to account for the particle bounce of aerosols on the 185 vaporizer. The composition-dependent CE was calculated as described by Middlebrook 186 et al. (2012) and was close to 0.45 for most of the time. Since for most ambient studies 187 a 0.5 CE value is found to be representative with data uncertainties generally within 188 ±20% (Canagaratna et al., 2007), and since our ACSM concentrations using CE=0.5 189 were in good agreement with concentrations from other co-located instruments, a CE 190 of 0.5 was used.

191 The organic components were further investigated by applying Multilinear 192 Engine (ME-2) (Paatero, 1999) to the organic mass spectra. With the ME-2, the user 193 can introduce a priori information about sources with the so-called a-value approach. 194 Hence, the user inputs one or more factor profiles and a constraint defined by the a-195 value, which determines the extent to which the output profile can differ from the profile 196 fed to the model. The source apportionment of OA was performed applying ME-2 using 197 the custom software tool of Source Finder (SoFi) version 4.8 developed by Canonaco 198 et al. (2013). The ME-2 analysis was carried out separately for the summer period (14 199 Jul 11 – 24 Sep 11) and the winter period (10 Jan 12 – 7 Mar 12). Only  $m/z \le 100$  were 200 used for source apportionment of OA because: a) the signals of m/z>100 account for a

minor fraction of the total organic mass (on average, 2 %), b) the m/z>100 have larger uncertainties, and c) the large interference of naphthalene signals (used for m/zcalibration of the ACSM) at these m/z (e.g., m/z 127, 128, and 129) (Sun et al., 2012).

### 204 2.3 Co-located measurements used in this study

205 MSC has been permanently equipped with aerosol monitoring instrumentation 206 since January 2010 and some of these measurements were used in this study. The 24 207 h PM<sub>1</sub> samples were collected every 4 days on 150 mm guartz micro-fiber filters (Pallflex QAT) using high volume samplers (30 m<sup>3</sup> h<sup>-1</sup>, MCV CAV-A/MSb) equipped 208 209 with MCV PM<sub>1</sub> cut-off inlets. Daily PM<sub>1</sub> mass concentrations were determined by off-210 line gravimetric procedures according to the EN 12341 standard (CEN, 1999), i.e. at 211 20°C temperature and 50% relative humidity. Furthermore, PM<sub>1</sub> chemical composition 212 was obtained as described by Ripoll et al. (2015) using inductively coupled plasma 213 atomic emission spectroscopy (ICP-AES) and mass spectrometry (ICP-MS) for major 214 and trace elements, respectively, ion high performance liquid chromatography (HPLC) 215 and selective electrode for ions concentrations, and thermal-optical method (using the 216 EUSSAR 2 protocol) for elemental carbon (EC) and organic carbon (OC) 217 concentrations. Real-time PM<sub>1</sub> mass concentrations were continuously measured by an 218 optical particle counter (OPC, model GRIMM 1.107 calibrated with different latex PSL 219 refraction index 1.59). PM<sub>1</sub> 30-minute data were daily averaged and subsequently 220 corrected by comparison with 24 h standard gravimetric mass measurements. The 221 absorption coefficient was measured continuously at 637 nm using a multi-angle 222 absorption photometer (MAAP, model 5012, Thermo). Equivalent black carbon (BC) 223 mass concentrations (Petzold et al., 2013) were calculated by the MAAP instrument software by dividing the measured absorption coefficient  $\sigma_{ap}$  ( $\lambda$ ) by 6.6 m<sup>2</sup> g<sup>-1</sup>, which is 224 the instrument default mass absorption cross section (MAC) at 637 nm (Müller et al., 225 226 2011; Petzold and Schönlinner, 2004). Particle scattering ( $\sigma_{sp}$ ; 0° - 360°) and 227 hemispheric backscattering ( $\sigma_{bsp}$ ; 90° - 270°) coefficients at three wavelengths (450nm, 228 525nm, 635nm) were measured with a LED-based integrating nephelometer (model 229 Aurora 3000, ECOTECH Pty, Ltd, Knoxfield, Australia).

Finally, all meteorological data were measured by the Catalonian Meteorological Service from the Montsec d'Ares station. Gaseous pollutants ( $O_3$ , NO, NO<sub>2</sub>, CO and SO<sub>2</sub>) were measured using real-time monitors belonging to the Network of Control and Surveillance of Air Quality of the Autonomous Government of Catalonia. NO and NO<sub>2</sub> concentrations were measured using a Thermo Scientific instrument, model 42i-TL; CO using a Teledyne 300 EU Gas filter correlation analyzer; O<sub>3</sub> using a MCV 48AV UV photometry analyzer; and SO<sub>2</sub> using a Teledyne 100 EU UV
fluorescence analyzer.

In addition to these routine measurements, 2 intensive campaigns were 238 performed in July - August 2011 and January - February 2012. During these intensive 239 240 campaigns PM<sub>1</sub> filters were collected daily and a scanning mobility particle sizer 241 (SMPS) was installed to measure particle number size distribution of mobility diameters 242 11-350 nm in the summer campaign, and 8-450 nm in the winter campaign. The SMPS 243 system comprises a classifier unit (model TSI 3080) and a differential mobility analyzer 244 (DMA, model TSI 3081) connected to a condensation particle counter (CPC, model TSI 245 3772). The SMPS data for the winter campaign were also used to estimate the mass 246 concentration to compare with the ACSM data. To this end, the volume size 247 distributions were calculated from the measured particle number distributions assuming 248 sphericity. The total volume concentrations were computed by integrating over the 249 measured particle range and converted to mass concentration using the estimated 250 composition-dependent density, calculated using the chemical composition given by 251 the ACSM and the equation of Salcedo et al. (2006). Average concentrations shown in 252 the whole paper are arithmetic averages unless otherwise specified.

253 3 Results and discussion

### 254 3.1 Submicron aerosol mass concentrations

255 In order to establish the consistency of the different measurements during this 256 study, the sum of the ACSM species (= sulfate + nitrate + ammonium + OA + chloride) 257 and the BC mass concentrations was compared with the co-located PM1 and light 258 scattering measurements (Fig. 2). The scatter plots of ACSM plus BC concentrations 259 versus PM<sub>1</sub> concentrations from the OPC and SMPS showed strong correlations (R<sup>2</sup>=0.72 and R<sup>2</sup>=0.87, respectively) and slopes close to unity (slope=0.94 and 260 intercept=0.09 for the ACSM+BC vs. OPC, and slope=1.21 and intercept=0.86 for the 261 262 ACSM+BC vs. SMPS) (¡Error! No se encuentra el origen de la referencia.). The 263 differences in the particle size range measured by the different instruments needs to be 264 considered when assessing these comparisons. Moreover, ACSM plus BC 265 concentrations were also highly correlated with light scattering at 525 nm determined by the nephelometer (R<sup>2</sup>=0.85, **¡Error! No se encuentra el origen de la referencia.**). 266 267 The high degree of agreement is also apparent in the time series plots of PM<sub>1</sub> shown in 268 Fig. 2.

The average concentration  $(25^{th}, 50^{th}, 75^{th} \text{ percentiles})$  of the ACSM + BC mass during during this study (July 2011 - April 2012) was 4.9 µg m<sup>-3</sup> (0.9, 2.8, 7.9 µg m<sup>-3</sup>) (

- Table 1), which is similar to the 2010-2012 average reported by Ripoll et al. (2014)
- from OPC measurements (5.0  $\mu$ g m<sup>-3</sup>). For the sake of brevity only summer (14 Jul 11
- 273 24 Sep 11) and winter (10 Jan 12 7 Mar 12) hourly variation will be discussed in the
- following sections, given that hourly variation in spring was similar to summer and that
- in fall was similar to winter. The seasonal average PM<sub>1</sub> concentrations were higher in
- 276 summer (7.5  $\mu$ g m<sup>-3</sup> (3.4, 7.1, 10.5  $\mu$ g m<sup>-3</sup>),

277 Table 1) than in winter (4.1  $\mu$ g m<sup>-3</sup> (0.8, 1.7, 5.6  $\mu$ g m<sup>-3</sup>),

278 Table 1). A similar seasonal pattern has been described at other high altitude 279 sites in Europe (e.g. Carbone et al., 2014; Cozic et al., 2008; Freney et al., 2011; Tositti 280 et al., 2013), being associated with differences in the air mass origin from summer to 281 winter, and also to variations in the planetary boundary layer (PBL) height. The 282 seasonal variation at MSC has been described in detail in recent works (Pandolfi et al., 283 2014b; Ripoll et al., 2014, 2015), and it has been principally attributed to the seasonal 284 variation of the PBL (Fig. 1). In summer, the stable anticyclonic conditions over this 285 continental area enhance convection increasing the development of the PBL and 286 favoring the transport of anthropogenic pollutants towards high altitude sites such as 287 MSC. The situation in winter is notably different, as the lower vertical development of 288 the PBL over this area leaves high altitude sites in the FT, isolating MSC from polluted 289 air masses. The seasonal variation of PM concentrations at MSC has been also 290 connected to mesoscale and synoptic processes. At MSC, southern flows and regional 291 recirculation episodes are more frequent in summer, whereas clean Atlantic advections 292 and northeastern winds from mainland Europe are more common in winter (Fig.S2 and 293 Table S1) (Pandolfi et al., 2014b; Ripoll et al., 2014, 2015). Moreover, the summer 294 maximum has been ascribed to the more vigorous photochemistry in the atmosphere 295 that enhances the formation of secondary inorganic and organic aerosols (Querol et al., 296 1999).

#### 297 3.2 Submicron aerosol chemical composition

298 Concentrations of ACSM species were daily averaged and compared with off-299 line measurements from 24-h PM<sub>1</sub> samples, and all species showed strong correlations 300 (R<sup>2</sup> between 0.77 and 0.96, Fig.S3). Different slopes (ACSM vs off-line measurements) 301 were found for each of the species: 1.12 for sulfate, 1.31 for ammonium and 1.35 for 302 nitrate. The relatively higher slope for nitrate, with respect to sulfate, could be attributed 303 to a sampling negative artifact due to the volatilization of nitrate on the off-line samples 304 (Schaap et al., 2004). Ammonium is present as a counterion for sulfate and nitrate, and 305 thus its slope is in between those of the two species.

306

The OM-to-OC ratio can not be determined for ACSMs based on f44 following 307 Aiken et al. (2008) as concluded from an intercomparison of 13 ACSMs (Fröhlich et al., 308 2015). Therefore, the OM-to-OC in the present study was estimated from the slope of 309 OA, measured by ACSM, vs. OC, obtained from off-line measurements. This slope was 310 found to be 3.39, however values of 2.2 are more common for aged aerosol (e.g. Aiken 311 et al., 2008; Minguillón et al., 2011; Takahama et al., 2011). This disagreement of a 312 factor of 1.54 may be attributed to different reasons. A negative volatilization artifact 313 may occur in the filters, hence resulting in an underestimation of OC. Alternatively it is

314 possible that the RIE for OA in the ACSM is larger than the value of 1.4 determined for 315 the AMS, a topic currently being investigated by the ACSM manufacturer, which would 316 result in an overestimation of OA. Similar series of intercomparisons with a similar 317 discrepancy for OA has been found for a one-year dataset (June 2012 – July 2013) 318 with the same instrument at Montseny site (Minguillón et al., 2015) and for a recently 319 reported study in Atlanta, US by Budisulistiorini et al. (2014). Assuming that the 320 disagreement was due to the overestimation by the ACSM, the OA concentrations 321 were corrected dividing by the disagreement factor (1.54) to compare the results with 322 co-located measurements (¡Error! No se encuentra el origen de la referencia.). The 323 resulting slopes were very similar and hence OA concentrations reported in the present 324 paper were not corrected since further research is needed to better estimate the RIE 325 for OA in the ACSM. 326 The average of PM<sub>1</sub> chemical composition at MSC during this study (July 2011 - April

- 327 April 2012) is given in Fig. 4. On average, OA was the largest  $PM_1$  constituent (50%),
- 328 followed by sulfate (20%), nitrate (14%), ammonium (12%), BC (4%) and chloride (1%).

329 As was the case of PM<sub>1</sub> concentrations, all chemical components increased in summer

and decreased in winter, with the exception of nitrate (Fig. 4 and 330

331 Table 1). The higher nitrate concentrations in winter than in summer were also 332 observed in other studies in the Mediterranean region (e.g. Pey et al., 2009; Querol et 333 al., 2009; Ripoll et al., 2015) and this variation was attributed to the high volatility of 334 ammonium nitrate at low humidity and high temperature (Zhuang et al., 1999). At MSC, 335 the summer maximum of the rest of PM components has been mainly ascribed to the 336 higher temperature and solar radiation in summer (Table S1), which enhances 337 atmospheric photochemistry, promoting the formation of secondary inorganic and 338 organic aerosols. All these seasonal characteristics are described in detail in Ripoll et 339 al. (2015).

## 340 3.2.1 <u>Summer trends</u>

341 Time series of  $PM_1$  components during summer time (14 Jul – 24 Sep 2011) are 342 shown in Fig. 5. Wind direction and wind speed, temperature, precipitation and 343 concentrations of nitrogen oxides, sulfur dioxide, and ozone are also depicted. The 344 daily classification of atmospheric episodes affecting MSC is also illustrated in different 345 background colors. On average, during summer the lowest concentrations of all PM1 346 components and gases were recorded under the Atlantic advection conditions since 347 these air masses are associated with precipitation, decreased temperature and solar 348 radiation, and strong winds, leading to cleaner atmospheric conditions. Conversely, 349 summer regional episodes lasted for 6 to 11 consecutive days and led to sustained 350 increases of the background concentrations of sulfate, OA and BC at MSC. Despite the 351 limited ACSM data availability during North African episodes, relatively high 352 concentrations of PM<sub>1</sub> components were observed under this type of episodes, 353 especially BC (Fig. 5).

354 The diurnal cycles of PM<sub>1</sub> components, gaseous pollutants and meteorological 355 variables are shown in Fig. 6. The summer concentrations of PM<sub>1</sub> components and 356 gases showed no clear diurnal patterns, except for ozone and OA. This lack of defined 357 daily patterns is similar to the findings obtained at the high altitude Puy-de-Dôme 358 station in central France (Freney et al., 2011, the only similar study found at a remote 359 site). For a high altitude site as MSC, the lack of diurnal cycles can be explained by a 360 combination of factors. In summer, the recirculation of air masses over the WMB 361 induced by an abrupt orography (Fig.S1) causes the formation of reservoir layers at 362 any time at relatively high altitudes (Millan et al., 1997; Rodríguez et al., 2003). 363 Moreover, long-range transport from North Africa, which can be more intense at high 364 altitude layers (Ripoll et al., 2015), could also blur the daily patterns since the 365 occurrence of this transport does not depend on the time of the day. These factors 366 result in a lack of well-defined daily patterns but in a high variability of diurnal cycles

even within the same type of episode, which is reflected in the similar average daily
evolutions and the high standard deviations calculated for the average daily patterns
when separated by air mass origin (Fig.S4).

370 The ozone and OA concentrations had a marked diurnal cycle regardless of the 371 air mass origin (Fig. 6 and Fig.S4) in summer. These different daily patterns with 372 respect to the rest of the gases and chemical components points to the fact that ozone 373 and OA variations are strongly influenced by local/regional processes and not just 374 dominated by long-range transport. Minimum ozone concentrations were recorded 375 between 8:00 and 9:00 UTC (Coordinated Universal Time, which is local time - 1:00h 376 and local summer time - 2:00h), whereas maximum concentrations were measured 377 between 16:00 and 17:00 UTC. In contrast, the highest OA concentrations were 378 observed around 12:00 UTC, and the lowest during the night and in the early morning. 379 The ozone variations may influence those of OA, although the complete understanding 380 of the ozone diurnal evolution is outside the scope of this study. The average increase 381 in OA during the day is likely due to the photooxidation of volatile organic compounds 382 (VOCs). Given that MSC is a remote site, in summer VOCs are most likely dominated 383 by local biogenic emissions (BVOCs), as it was found in the Mediterranean forested 384 area of Montseny (Seco et al., 2011). Hence, the midday increase is likely due to 385 recently-produced biogenic SOA, and to a lesser degree, photooxidation of 386 anthropogenic VOCs.

387 Despite the marked diurnal cycle of OA regardless of air mass origin, the 388 average increase during the day with respect to average concentrations during the night was higher under summer regional (2.6  $\mu$ g m<sup>-3</sup>) and North African (3.0  $\mu$ g m<sup>-3</sup>) 389 episodes than during Atlantic advections (1.3 µg m<sup>-3</sup>) (Fig.S4). This difference could be 390 391 caused by the higher SOA formation. This is due to the increase in BVOCs emissions 392 and atmospheric photooxidation caused by the higher temperature and solar radiation 393 (Paasonen et al., 2013; Seco et al., 2011) under summer regional and North African 394 episodes. Furthermore, under these episodes higher concentrations of ozone were 395 measured, which also favors the formation of SOA (via direct oxidation and also by 396 leading to higher OH concentrations). The SOA formation registered at MSC is 397 relatively high when compared to other high altitude sites such as Puy-de-Dôme 398 (Freney et al., 2011). This is in agreement with the modeled SOA emissions over 399 Europe, which identified higher SOA concentrations in Mediterranean environments 400 (Bessagnet et al., 2008). This higher SOA formation is probably due to the higher 401 emissions of BVOCs in the Mediterranean forested areas (up to 3 times higher than 402 Boreal forested areas) (Bessagnet et al., 2008; Lang-Yona et al., 2010; Steinbrecher et 403 al., 2009) and the comparable concentrations of tropospheric ozone with other high 404 altitude European sites (Chevalier et al., 2007). On the other hand, the extra formation
405 of SOA under summer regional and North African episodes might also have a
406 contribution from the photooxidation of anthropogenic VOCs, since Atlantic advections
407 are associated with cleaner atmospheric conditions.

## 408 3.2.2 Winter trends

Similar to summer, the lowest concentrations of all PM<sub>1</sub> components and gases in winter (10 Jan 12 – 7 Mar 12) were recorded under the Atlantic advections, whereas the highest were measured when MSC was affected by air masses from mainland Europe and sporadically under regional conditions (Fig. 7). Mediterranean air masses were detected very infrequently and therefore conclusions on their characteristics will not be drawn in the present paper. The relative contribution of different components was similar, with OA representing a somewhat smaller fraction than in summer.

416 In contrast to what was found in summer, in winter concentrations of most PM<sub>1</sub> 417 components and gaseous pollutants showed much clearer diurnal patterns, with a 418 minimum around 7:00 UTC and a maximum between 14:00 and 15:00 UTC (Fig. 6). 419 Similar patterns have been observed at the Puy-de-Dôme station during a winter 420 campaign (Freney et al., 2011). These daily cycles are probably caused by the fact that 421 MSC is located most of the day within the FT in winter, whereas PBL air masses are 422 only injected upwards after midday (Fig. 1). Moreover, thermal inversions are very 423 frequent from 20:00 to 07:00 UTC. These situations prevent the transport of pollutants 424 from the lower populated areas towards higher altitudes, especially at night. During the 425 morning, the thermal inversions dissipate due to the radiative warming of the ground 426 and mountain upslope winds develop (e.g. Henne et al., 2004). These mountain winds 427 transport anthropogenic emissions from the adjacent valleys and plains to the top of 428 the mountain, with a maximum upslope transport in the afternoon. Moreover, biogenic 429 emissions influence cannot be ruled out as average winter temperatures are high 430 enough for them to occur (Seco et al., 2011; Steinbrecher et al., 2009). Thus, mountain 431 breezes play an important role in determining the diurnal variation of PM<sub>1</sub> components 432 in winter (Fig. 6), especially under regional conditions. A clear example of the  $PM_1$ 433 components diurnal pattern under winter regional episodes was observed from 22 to 25 434 February 2012 with PM<sub>1</sub> concentrations (and NO<sub>x</sub>) increasing several fold during the 435 afternoon (Fig. 7).

The study of the daily cycles as a function of air mass origin (Fig.S5) showed clear diurnal patterns under winter regional episodes, as mentioned above, and less marked daily patterns when MSC is affected by Atlantic advections and long-range transport from mainland Europe. Under Atlantic episodes the concentrations of PM<sub>1</sub>

440 components were very low and the standard deviations with respect to the average 441 pattern were quite high, resulting in unclear diurnal patterns compared to those under 442 winter regional conditions. During European episodes, which can be more intense at 443 high altitude layers (Ripoll et al., 2015; Sicard et al., 2011), background concentrations 444 of PM<sub>1</sub> components were higher and the midday increment was lower compared to 445 those under winter regional conditions, resulting in less marked daily patterns. These 446 less-marked diurnal cycles are probably due to the fact that the increase of PM1 447 components occurs during these episodes regardless of the time of the day. A good 448 example of this less-marked diurnal variation during European episodes was observed 449 from 17 to 19 February 2012 (Fig. 7).

## 450 3.3 Characterization of OA components

451 In order to better characterize the profiles of OA components, the ME-2 analysis 452 was performed separately for the summer period (14 Jul – 24 Sep 2011) and the winter 453 period (10 Jan 12 – 7 Mar 12), since OA components are expected to vary throughout 454 the year. The solution of ME-2 analysis selected for each season was based on several 455 tests, with different number of factors and different a-values, taking into account the 456 correlations with external tracers (including nitrate, sulfate, BC and ozone), the daily 457 patterns of each factor, and the residuals. As a result, a solution of 3 factors was 458 selected for each season. In summer, a hydrocarbon-like OA (HOA), a semi-volatile 459 oxygenated OA (SV-OOA) and a low-volatile oxygenated OA (LV-OOA) (Fig.S6) were 460 resolved. The HOA factor was constrained using an average HOA factor from different 461 datasets (Ng et al., 2011b), with an a-value of 0.1. The SV-OOA was characterized by 462 a high 43-to-44 ratio, and the LV-OOA was defined by having a dominant peak at m/z463 44. In winter the 3 factors identified were: hydrocarbon-like OA (HOA), biomass burning 464 OA (BBOA) and oxygenated OA (OOA) (Fig.S6). The HOA and the BBOA factors were 465 constrained based on the profiles from different datasets (Ng et al., 2011b), with an a-466 value of 0.1 in both factors, and including BC as an additional variable to the ME-2 467 analysis. The OOA was characterized by a 43-to-44 ratio between those found for the 468 LV-OOA and the SV-OOA in summer, and by having a dominant peak at m/z 44. This 469 high signal of m/z 44 in the winter OOA indicates a high degree of oxidation, and 470 therefore a dominant aged character during winter. A solution with 4 factors in winter 471 was investigated in order to split the OOA into SV-OOA and LV-OOA, but the resulting 472 profiles did not represent two different OOA types, and the time series showed that one 473 of the factors was mainly representing noise. The HOA profiles obtained for the 474 summer and winter periods were similar and showed similar deviations from the Ng et

475 al. (2011b) spectrum (Fig.S7), since they were constrained with the same anchor HOA476 profile.

477 On average, LV-OOA dominated the OA fraction in summer contributing 64%, 478 followed by SV-OOA (26%) and HOA (10%) (Fig. 8), whereas in winter OOA accounted 479 for 71%, BBOA contributed 24%, and HOA contribution decreased to 5% (Fig. 9). The 480 high contribution of OOA components confirms the initial hypothesis of MSC organic 481 components being mostly secondary in their origin, and it is in agreement with what 482 was found in other remote sites (Freney et al., 2011; Raatikainen et al., 2010). 483 Furthermore, the origin of OA has been recently investigated in the Mediterranean 484 forested area of Montseny, and it has been found that SOA accounted for 91% 485 (Minguillón et al., 2011). The low contribution of primary organic components at MSC is 486 in agreement with the location, since the primary organic emissions are mixed and 487 oxidized during their transport from industrial and urban areas to the remote site of 488 MSC.

489 The diurnal cycles of OA components were studied as a function of air mass 490 origin (Fig. 8 and Fig. 9). A clear daily pattern of OA components was found regardless 491 of the air mass origin, except for Atlantic advections in winter. In summer, the 492 maximum concentrations of LV-OOA and HOA were measured between 12:00 and 493 13:00 UTC, whereas those of SV-OOA were observed between 11:00 and 12:00 UTC. 494 The LV-OOA has been generally associated with highly oxidized, aged, and long-495 range-transported aerosol particles (Lanz et al., 2010). Conversely, the SV-OOA has 496 been described as the less oxygenated and semi-volatile fraction of OOA (Ng et al., 497 2011a) and therefore it has been mostly attributed to SOA formation from more local 498 emissions (Jimenez et al., 2009). For this reason, the LV-OOA and HOA hourly 499 variations are more influenced by long-range transport than those of SV-OOA, which 500 are strongly influenced by local/regional processes. In winter, the maximum 501 concentrations of the OA components were observed simultaneously around 14:00 502 UTC. The different daily patterns between seasons can be attributed to the higher 503 production of SOA in summer as opposed to winter, when the maximum daily 504 concentrations are reached later driven by the mountain breezes.

### 505 4 Conclusions

506 This work interprets the real-time variation of inorganic and organic submicron 507 components during 10 months (July 2011 - April 2012) at a high altitude site in 508 southern Europe (Montsec, 1570 m a.s.l.). The aerosol chemical composition was 509 obtained with an ACSM, and co-located on-line and off-line  $PM_1$  measurements were 510 also carried out. The average concentration of the ACSM plus BC mass during this study was 4.9 µg m<sup>-3</sup>, and on average OA was the foremost PM<sub>1</sub> constituent (50%),
followed by sulfate (20%), nitrate (14%), ammonium (12%), BC (4%) and chloride (1%).
Discrepancies of OA determined by ACSM with co-located measurements pointed to
an overestimation by the ACSM probably caused by the use of the default RIE for OA,
which could be lower than the actual one. Further research is needed to better address
this issue.

517 The seasonal variation of PM<sub>1</sub> mass and chemical components concentrations 518 showed similar patterns, with an increase in summer and a decrease in winter, except 519 for nitrate which has high volatility in summer. The seasonal variation was attributed to 520 the evolution of the PBL height throughout the year and to synoptic circulation and 521 meteorological factors. At MSC the higher temperature and solar radiation in summer 522 enhances the convection processes, incrementing the development of the PBL, and 523 augments atmospheric photochemistry, promoting the formation of secondary inorganic 524 and organic aerosols.

525 The diurnal variation of PM<sub>1</sub> components had no clear diurnal patterns in 526 summer, except for organics. This lack of defined daily patterns was ascribed to the 527 recirculation of air masses that causes the formation of reservoir layers at relatively 528 high altitudes, and to the long-range transport from North Africa. These factors result in 529 a high variability of diurnal cycles even within the same type of episode. Nevertheless, 530 organic concentrations had a marked diurnal cycle regardless of the air mass origin, 531 with maximum concentrations around 12:00 UTC. The OA was dominated by LV-OOA 532 (64%), followed by SV-OOA (26%), and HOA (10%). Hence, the midday increase with 533 respect to average concentrations during the night was attributed to the formation of 534 SOA.

In winter under regional conditions, concentrations of all PM<sub>1</sub> components showed much clearer diurnal patterns than in summer, with a maximum between 14:00 and 15:00 UTC. These daily cycles were connected to the fact that MSC is located most of the day within the FT, whereas PBL air masses are only injected upwards after midday. However, when MSC was affected by long-range transport from mainland Europe, less marked daily patterns of PM<sub>1</sub> components were observed.

541 The OA in winter was also mainly secondary (71%), with contributions from 542 BBOA (24%), and HOA (5%). The hourly variation of these factors showed a clear 543 diurnal pattern regardless of the air mass origin, except for Atlantic advections.

544 To the authors' knowledge, this is one of the first times when real-time 545 submicron aerosol chemical composition is characterized and its variation is interpreted 546 during almost a year in a continental background environment. The results obtained in 547 the present study highlight the importance of the SOA formation processes at such 548 remote site as MSC, which could be the objective of further investigations.

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885 Fig. 1 Diurnal variation of the boundary layer height (computed with HYSPLIT model)

averaged for each month during the study period at Montsec. Variation bars indicate ±
 standard deviation.

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890 Fig. 2 Time series of PM<sub>1</sub> total mass from co-located measurements and light scattering

891 at 525 nm.

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Fig. 3 Top: Scatter plots of ACSM plus BC mass versus PM<sub>1</sub> from (a) OPC and (b) SMPS
(winter campaign), and versus light scattering at 525 nm from (c) nephelometer. Bottom:
Scatter plots of ACSM plus BC mass after dividing OA by 1.54 versus PM<sub>1</sub> from (d) OPC
and (e) SMPS (winter campaign), and versus light scattering at 525 nm from (f)
nephelometer. Data points correspond to hourly values. Equations and red lines
correspond to linear regression fits.



904  $\,$  Fig. 4 Average concentrations of  $\text{PM}_1$  chemical species measured at Montsec during the

905 whole study, in summer and in winter.

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Fig. 5 Time series of wind direction, precipitation, temperature, wind speed, concentrations of nitrogen oxides ( $NO_x$ ), sulfur dioxide ( $SO_2$ ), ozone ( $O_3$ ), and  $PM_1$ chemical species (OA, sulfate, nitrate, ammonium, chloride and black carbon (BC)) in summer (14 Jul 11 – 24 Sep 11). Background colors correspond to daily classification of atmospheric episodes (summer regional (SREG), North African (NAF), and Atlantic (AT)) and the pie chart correspond to the average chemical composition for the summer period.



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917 Fig. 6 Diurnal cycles of  $PM_1$  chemical species (black carbon (BC), sulfate, nitrate, 918 ammonium, chloride and OA), gaseous pollutants (ozone (O<sub>3</sub>), nitrogen oxides (NO<sub>x</sub>), and 919 sulfur dioxide (SO<sub>2</sub>)), and meteorological parameters (relative humidity, temperature and 920 solar radiation) averaged for the whole period, summer and winter. Variation bars 921 indicate ± standard deviation.



Fig. 7 Time series of wind direction (WD) and speed (WS), temperature (T), precipitation (PP), concentrations of nitrogen oxides (NO<sub>x</sub>), sulfur dioxide (SO<sub>2</sub>), ozone (O<sub>3</sub>), and PM<sub>1</sub> chemical species (organics, sulfate, nitrate, ammonium, chloride and black carbon (BC)) in winter (10 Jan 12 – 7 Mar 12). Background colors correspond to daily classification of atmospheric episodes (European (EU), winter regional (WREG), Mediterranean (MED) and Atlantic (AT)) and the pie chart correspond to the average chemical composition for the winter period.



Fig. 8 Top: Time series of organic species (hydrocarbon-like organic aerosol (HOA),
semi-volatile oxygenated organic aerosol (SV-OOA) and low-volatility oxygenated
organic aerosol (LV-OOA)) concentrations in summer (14 Jul 11 – 24 Sep 11).
Background colors correspond to daily classification of atmospheric episodes and the
pie chart correspond to the average organic species composition for the summer period.
Bottom: Diurnal cycles of organic species concentrations averaged as a function of
atmospheric episode for the summer period. Variation bars indicate ± standard deviation.





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Fig. 9 Top: Time series of organic species (hydrocarbon-like organic aerosol (HOA), biomass burning organic aerosol (BBOA) and oxygenated organic aerosol (OOA)) concentrations in winter (10 Jan 12 – 7 Mar 12). Background colors correspond to daily classification of atmospheric episodes and the pie chart correspond to the average organic species composition for the winter period. Bottom: Diurnal cycles of organic species concentrations averaged as a function of atmospheric episode for the winter period. Variation bars indicate ± standard deviation.

Table 1 Average (25th, 50th, 75th percentiles) concentrations of all chemical components
measured at Montsec during the study. Note that the whole period averages include also
spring and fall.

	Whole period (µg m <sup>-3</sup> )	Summer (µg m <sup>-3</sup> )	Winter (µg m <sup>-3</sup> )
Period	14 Jul 11 - 23 Apr 12	14 Jul 11 - 24 Sep 11	10 Jan 12 -7 Mar 12
Organics	<b>2.4</b> (0.5, 1.4, 3.8)	<b>4.0</b> (2.1, 4.0, 5.6)	<b>1.9</b> (0.4, 1.0, 2.6)
Sulfate	<b>1.0</b> (0.2, 0.5, 1.4)	<b>1.9</b> (0.7, 1.6, 2.7)	<b>0.5</b> (0.1, 0.3, 0.7)
Nitrate	<b>0.7</b> (0.05, 0.2, 0.7)	<b>0.5</b> (0.1, 0.3, 0.6)	<b>1.0</b> (0.05, 0.2, 1.0)
Ammonium	<b>0.6</b> (0.09, 0.3, 0.9)	<b>0.9</b> (0.3, 0.8, 1.2)	<b>0.5</b> (0.06, 0.2, 0.7)
Chloride	<b>0.05</b> (0.01, 0.02, 0.05)	<b>0.04</b> (0.01, 0.02, 0.05)	<b>0.1</b> (0.004, 0.02, 0.1)
BC	<b>0.2</b> (0.04, 0.1, 0.3)	<b>0.2</b> (0.1, 0.2, 0.3)	<b>0.2</b> (0.03, 0.1, 0.2)
Total	<b>4.9</b> (0.9, 2.8, 7.9)	<b>7.5</b> (3.4, 7.1, 10.5)	<b>4.1</b> (0.8, 1.7, 5.6)