



Firework impacts on  
air quality and human  
health: tracers,  
sources and aging  
processes

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# The impacts of fireworks burning at Chinese Spring Festival on air quality and human health: insights of tracers, source evolution and aging processes

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## Abstract

To understand the impact of fireworks burning (FW) particles on air quality and human health during winter haze period, thirty-nine elements, ten water-soluble ions and eight fractions of carbonaceous species in atmospheric PM<sub>2.5</sub> at Nanjing were investigated during 2014 Chinese Spring Festival (SF). Serious regional haze pollution persisted throughout the entire sampling period, PM<sub>2.5</sub> averaging at  $113 \pm 69 \mu\text{g m}^{-3}$  and visibility at  $4.8 \pm 3.2 \text{ km}$ . The holiday effect led to almost all the chemical species decreasing during the SF, except for Al, K, Ba and Sr which were related to FW. The source contributions of coal combustion, vehicle emission and road dust decreased dramatically, whereas FW contributed to about half of the PM<sub>2.5</sub> during SF period. The intensive emission of FW particles at New Year's Eve accounted for 60.1 % of the PM<sub>2.5</sub>. They also significantly modified the chemical compositions of PM<sub>2.5</sub>, with 39.3 % contributed by increased organic matter, followed by steadily increased loadings of secondary inorganic ions. The aging processes of the FW particles lasted for at least six days reflected by the variation of SO<sub>4</sub><sup>2-</sup>, characterized by heterogeneous reactions of SO<sub>2</sub> and NO<sub>x</sub> on crustal materials directly from FW, the replacement of Cl<sup>-</sup> by NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup>, coating of NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup> on soot, formation of secondary organic aerosols and metal-catalyzed formation of NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup> at higher relative humidity. During aging, the main contributors to the extinction coefficient shifted from elemental carbon and organic matter to sulfate ammonium. The particles raised higher cancer risks by heavy metals (especially for Cd and As) as  $1.62 \times 10^{-6}$ . This study provided detailed composition data and first comprehensive analysis of the aging processes of FW particles at serious haze pollution period and their potential impact on human health.

## 1 Introduction

Atmospheric pollutants emitted from fireworks burning (FW) at festivals or special celebration events around the world have recently received wide attention, such as Guy

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istry of Environmental Protection, at 31 January 2014, about 80 % of 161 cities held average  $\text{PM}_{2.5}$  concentrations higher than  $150 \mu\text{g m}^{-3}$  (Ye et al., 2014). The addition of pollutants from FW deteriorates ambient air quality during the SF and its impact can vary significantly with FW duration and meteorological conditions (Vecchi et al., 2008; Yu et al., 2013).

Previous studies have extensively characterized the physicochemical signatures of FW emissions, such as gaseous pollutants (Wehner et al., 2000; Attri et al., 2001; Wang et al., 2007; Vecchi et al., 2008; Tan et al., 2009; Godri et al., 2010), particle size distribution and number concentration (Wehner et al., 2000; Vecchi et al., 2008; Zhang et al., 2010; Dutschke et al., 2011; Yang et al., 2014), chemical components including trace elements (Wang et al., 2007; Moreno et al., 2007; Vecchi et al., 2008; Estrellan and lino, 2010; Godri et al., 2010; Joly et al., 2010; Do et al., 2012; Tsai et al., 2012; Tian et al., 2014; Yang et al., 2014), organic carbon (OC) and elemental carbon (EC) (Estrellan and lino, 2010; Tsai et al., 2012; Feng et al., 2012; Tian et al., 2014; Yang et al., 2014) and water-soluble ions (Wang et al., 2007, 2013; Vecchi et al., 2008; Shen et al., 2009; Estrellan and lino, 2010; Tsai et al., 2012; Jiang et al., 2014; Tian et al., 2014; Yang et al., 2014). Recently, single particle's chemical compositions (Allan et al., 2010; Jiang et al., 2014), morphology and mixing properties (Li et al., 2013) and optical properties (Yu et al., 2013) of FW particles had been reported. However, there are to date still two shortages.

Firstly, no studies reported the chemical compositions (crustal elements, trace elements, water-soluble ions, OC and EC) of FW particles completely, considering the complex manufacture materials of FW (Supplement S2). Although Estrellan and lino (2010), Feng et al. (2012), Yang et al. (2014) and Tian et al. (2014) reported the ions, elements and carbonaceous species synchronously, some important species were missed, such as Na, Mg, K, Ti,  $\text{NH}_4^+$  and  $\text{Cl}^-$  in Tian et al. (2014), OC in Estrellan and lino (2010), Si and some heavy elements in Feng et al. (2012) and crustal elements in Yang et al. (2014). These species are all highly elevated during FW events. Crustal elements including Na, Mg, Al, Si, K, Ca, Ti, Fe and Mn were always missing

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or not sufficiently studied in FW particles. However, except for inorganic and organic chemicals such as charcoal, potassium nitrate, potassium chlorate, potassium perchlorate, sulfur, manganese, sodium oxalate, aluminum, iron powder, strontium nitrate, and barium nitrate (Shen et al., 2009; Jing et al., 2014), clay is also used to seal the top and bottom of fireworks as shown in Supplement S3. 90 % of the total mineral aerosol was from the emission of FW on the lantern night in Beijing (Wang et al., 2007). The incomplete compositions of FW particles may bias the identification of particle sources during FW period and limit our understanding of its aging processes. For example, crustal elements-Al, Si and Ca were thought to result from the resuspension of materials already deposited on the ground (caused by pyrotechnic device explosions) and hereby the resuspended dust was regarded as a contributor to atmospheric particles during SF (Tian et al., 2014). Without detecting the crustal and trace elements, secondary particulate matter accounted for 63–82 % of  $PM_{1.0}$  during FW periods in Beijing (Jiang et al., 2014).

Secondly, all the former studies unanimously agreed with that FW contributed to elevated concentrations of particles and associated chemical species, but no studies concerned the aging processes of particles after emitted from intensive FW. The cocktail of primary pollutants released may exhibit varied interactions among themselves, and if aided by favourable atmospheric conditions, may lead to the formation of secondary pollutants (Sarkar et al., 2010). Do et al. (2012) pointed that the sub-micron aerosol or accumulation-mode particles from FW can suspend in the air for very long periods, from days to weeks, potentially causing pollution to large areas. Data in literature verified the existence of aging processes of FW particles, though it has not been discussed. For example, Li et al. (2013) indicated that emissions from FW significantly changed the transformation pathway from  $SO_2$  to  $SO_4^{2-}$  and the FW particles can influence the air at downwind site (50 km far away). And after about two days, the elevated  $PM_{2.5}$  mass concentrations at CNY's Eve decreased to the level of the day before CNY (pre-CNY). In Beijing,  $PM_{2.5}$  mass concentrations at CNY's Eve decreased to the level of pre-CNY in one day, while secondary ions ( $SO_4^{2-}$ ,  $NH_4^+$  and  $NO_3^-$ ) and organic mat-

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ter (OM) increased first and then decreased to the level of pre-CNY after three days (Wang et al., 2007). In Jinan, a “tailing” phenomena was found, indicating that  $\text{NH}_4^+$  and  $\text{NO}_3^-$  did not increase (peaking at 08:00 and 12:00 local time (LT), respectively) immediately with the concentrations of particles (peaking at 00:00 LT) after emitted at CNY’s Eve (Yang et al., 2014). In Shanghai,  $\text{NH}_4^+$  and  $\text{NO}_3^-$  also increased in the first day after emitted at CNY’s Eve and then decreased to the level of pre-CNY in the third day and the decreasing rate of  $\text{PM}_{2.5}$  was faster than that for OC (Feng et al., 2012). Therefore, we still need to answer the following questions: (1) are the decreasing trends of particles and associated chemical components the same? (2) How long and how far can FW influence the air quality at certain meteorological conditions? (3) During aging and transport, which type of chemical reactions will the FW particles undergo and how will specific chemical ratios vary?

Due to rapid economic expansion and urbanization, the occurrence frequency of haze has increased rapidly in recent 30 years at the Yangtze River Delta region (Wang et al., 2014). As one of the central megacities of this region, Nanjing is suffering from serious air pollution and the occurrence of hazy days increased from 1961 to 2005 (Kang et al., 2013). Thus, it provides a unique site and opportunity to study the chemical composition evolution and source variation during haze-clear days with the injection of intensive FW at SF. The main purpose of this study is to: (1) characterize in detail the chemical compositions of atmospheric  $\text{PM}_{2.5}$  before, during and after intensive FW events; (2) identify how long and how much can the FW particles influence  $\text{PM}_{2.5}$  by tracer analysis and receptor models; (3) emphasize how can the FW particles affect visibility and human cancer and non-cancer risks; (4) analyze the FW particle aging processes by specific species and ratios. The data and analysis will improve the knowledge of chemical compositions of FW particles, their evolution during serious haze pollution periods and their influence on visibility and human health.

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## 2 Methodology

### 2.1 PM<sub>2.5</sub> collection

From 24 January to 21 February 2014, a sampling campaign of PM<sub>2.5</sub> was conducted on the rooftop (40 m high) of a building in Nanjing University of Information Science & Technology (Kang et al., 2013; Wang et al., 2014). It is a suburban site, surrounded by residential communities at the west, south, north and southeast directions. There is a steel factory 2 km to the east of the campus and a chemical industry park about 10 km to the northeast. It faced one road with heavy traffic, about 200 m in the east. Location of the sampling site was shown in Fig. 1 and Supplement S4. During the sampling period, CNY at 30 January and Lantern Festival (LF) at 14 February were included. In CNY's and LF's Eve, numerous fireworks were consumed. The SF holiday was from 30 January to 7 February. PM<sub>2.5</sub> samples were collected for about 24 h using two medium-volume air samplers (TH-150C, Wuhan Tianhong Ltd., China) on quartz fiber filters (baked at 800 °C for 2 h) and polypropylene fiber filters (baked at 80 °C for 0.5 h) at a flow rate of 100 L min<sup>-1</sup>. The hourly online PM<sub>2.5</sub> mass concentrations for the nine monitoring sites set by Jiangsu Environmental Monitoring Center were collected from the public platform (<http://218.94.78.75/jsair/>) (their locations can be found in Supplement S4). Twenty four pairs of filter samples were collected (Supplement S5). By using a microbalance (Ohaus Discovery DV214CD) with balance sensitivity as ±0.010 mg, filters were weighed before and after sampling under controlled environment with temperature and relative humidity (RH) of 22 °C and 35 %. Then they were stored at -20 °C until chemical analysis. The quartz fiber filters were for analyzing water-soluble inorganic ions, OC and EC. Polypropylene fiber filters were for elemental analysis.

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## 2.2 Chemical analysis

Inductively coupled plasma-mass spectroscopy (ICP-MS) (Agilent 7500a, Agilent Co. USA) was used for analyzing Li, Be, Na, Mg, Al, P, K, Ca, Sb, Sc, Ti, V, Cr, Mn, Co, Ni, Cu, Zn, As, Rb, Y, Mo, Cd, Sn, Cs, La, Ce, Sm, W, Tl, Pb, Bi, Th and U. Nine elements including Si, Al, Ca, Mg, Fe, Ti, Ba, Sr and Zr were analyzed by inductively coupled plasma-optical emission spectrometer (ICP-OES). Al, Ca, Mg and Ti were analyzed by both ICP-MS and ICP-OES, and results from the latter were used as the analysis accuracy for the four elements by ICP-OES was better than that by ICP-MS (Kong et al., 2014a). Ten ions including  $\text{NH}_4^+$ ,  $\text{Na}^+$ ,  $\text{Mg}^{2+}$ ,  $\text{K}^+$ ,  $\text{Ca}^{2+}$ ,  $\text{F}^-$ ,  $\text{Cl}^-$ ,  $\text{NO}_2^-$ ,  $\text{NO}_3^-$  and  $\text{SO}_4^{2-}$  were analyzed by a professional Ion Chromatograph (Wan Tong 850, Switzerland). DRI Model 2001 (Thermal/Optical Carbon Analyzer) with the IMPROVE thermal/optical reflectance protocol was used for OC and EC analysis (Han et al., 2008, 2010; Huang et al., 2012; Li et al., 2012; Wang et al., 2013; Kong et al., 2014a). A  $0.188\text{ cm}^2$  punch area from the quartz filter was heated to produce four OC fractions: OC1, OC2, OC3 and OC4 at temperatures of 120, 250, 450 and  $550^\circ\text{C}$  in a non-oxidizing He atmosphere, three EC fractions: EC1, EC2 and EC3 at 550, 700 and  $800^\circ\text{C}$  in an oxidizing atmosphere of 2%  $\text{O}_2/98\%$  He and optically detected pyrolyzed carbon (OPC). OC is defined as  $\text{OC1} + \text{OC2} + \text{OC3} + \text{OC4} + \text{OPC}$  and EC is calculated by  $\text{EC1} + \text{EC2} + \text{EC3} - \text{OPC}$ . Char-EC is defined as EC1 minus OPC, and soot-EC is the sum of EC2 and EC3 (Han et al., 2008, 2010). The pre-treatment and chemical analysis procedures and quality assurance and control are described detailedly in our previous works (Li et al., 2012; Kong et al., 2014a, b; Li et al., 2014).

## 2.3 Meteorological parameters

The meteorological parameters including relative humidity (RH), visibility, wind speed, wind direction and temperature were recorded by the meteorological observatory of our university (<http://qxt.nuist.edu.cn/>). The rainfall information was obtained from <http://www.wunderground.com/>. According to China Meteorological Administration, fog

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where  $\bar{X}$  is the arithmetic mean of the log-transformed data,  $s$  is the standard deviation of the log-transformed data,  $H$  is the  $H$ -statistic and  $n$  is the number of samples. InhR: inhalation rate, 7.6 and 20 m<sup>3</sup> day<sup>-1</sup> for children and adult; EF: exposure frequency, 4 day year<sup>-1</sup> in this study (according to the discussion in Sect. 3.4.2); ED: exposure duration, 6 and 24 years for children and adult; BW: average body weight; 15 and 70 kg for children and adult; AT: the averaging time, for non-carcinogens, AT (days) = ED × 365; for carcinogens, AT (days) = 70 × 365 = 25 550.

After the  $D_{\text{inh}}$  was calculated, a Hazard Quotient (HQ) for non-cancer toxic risk can be obtained by (Kong et al., 2012):

$$\text{HQ} = D/\text{RfD} \quad (4)$$

Considering the sensitive group, the reference dose (RfD) (mg kg<sup>-1</sup> day<sup>-1</sup>) is estimated as the maximum permissible risk on human by daily exposure. The threshold values of RfD indicate whether there is adverse health effect during a life time. Then Hazard Index (HI) can be obtained by summing up the individual HQ to estimate the total risks of all elements considered:

$$\text{HI} = \sum \text{HQ}_i, \quad (5)$$

where  $i$  denotes different heavy metals. RfD values are  $7 \times 10^{-3}$ ,  $2.86 \times 10^{-5}$ ,  $1.43 \times 10^{-5}$ ,  $4.0 \times 10^{-2}$ ,  $3.0 \times 10^{-1}$  and  $3.5 \times 10^{-3}$  for the cancer risk estimation of V, Cr, Mn, Cu, Zn and Pb. For non-cancer risk estimation of Co, Ni, As and Cd, the RfD values are  $5.71 \times 10^{-6}$ ,  $2.00 \times 10^{-2}$ ,  $3.00 \times 10^{-4}$  and  $1.00 \times 10^{-3}$ , respectively (Kong et al., 2012). If  $\text{HI} \leq 1$ , there is no adverse health effects; if  $\text{HI} > 1$ , likely adverse health effects exist. For carcinogens, the LADD is multiplied by the corresponding RfD value. If the cancer risk is higher than  $10^{-6}$ , risk management decisions should be made.

### 2.4.4 Mass closure of PM<sub>2.5</sub>

To better understand the chemical compositions before, during and after SF, the chemical mass closure analysis was conducted. The chemical components are divided into

six classes as follows: mineral matter (MIN), trace elements (TE), OM, EC, sea salt (SS), secondary inorganic aerosol (SIA) and unidentified matter (UM). MIN is the sum of the common oxides of Al, Mg, Mn, Si, Na, K, Ca, Ti and Fe as:

$$\begin{aligned} \text{MIN} = & 2.14 \times \text{Si} + 1.67 \times \text{Ti} + 1.89 \times \text{Al} + 1.59 \times \text{Mn} + 1.67 \times \text{Mg} + 1.95 \times \text{Ca} \\ & + 1.35 \times \text{Na} + 1.21 \times \text{K} + 1.43 \times \text{Fe} \end{aligned} \quad (6)$$

As the existence of CaO and CaCO<sub>3</sub>, a factor of 1.95 for Ca is used (Terzi et al., 2010). Except for the above elements in MIN, all other elements are summed up to act as TE. OM is calculated by multiplying OC of a conversion factor, in accordance with the organic molecular carbon weight per carbon weight. Here, 2.0 is applied, same to the value used for spring festival period in Shanghai, another megacity in Yangtze River Delta region (Huang et al., 2012). In Allan et al. (2010), during bonfires and fireworks burning period at the Guy Fawkes' night, the OM/OC ratio ranged around 2.0 or higher than 2.0. The marine contribution is calculated based on a standard sea water composition, assuming that soluble Na<sup>+</sup> in aerosols only come from sea salt. Then,

$$\text{sea salt} = [\text{Na}^+] + [\text{ss-Cl}^-] + [\text{ss-Mg}^{2+}] + [\text{ss-K}^-] + [\text{ss-Ca}^{2+}] + [\text{ss-SO}_4^{2-}], \quad (7)$$

where  $\text{ss-Cl}^- = 1.8 \times [\text{Na}^+]$ ,  $\text{ss-Mg}^{2+} = 0.12 \times [\text{Na}^+]$ ,  $\text{ss-K}^- = 0.036 \times [\text{Na}^+]$ ,  $\text{ss-Ca}^{2+} = 0.038 \times [\text{Na}^+]$ , and  $\text{ss-SO}_4^{2-} = 0.252 \times [\text{Na}^+]$  (Terzi et al., 2010). SIA is the sum of nss-SO<sub>4</sub><sup>2-</sup>, NO<sub>3</sub><sup>-</sup> and NH<sub>4</sub><sup>+</sup>. The concentrations of all these species in μg m<sup>-3</sup> are adopted.

#### 2.4.5 Visibility re-construction by chemical components

The mass scattering efficiencies of spherical particles is a function of water and chemical components including (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>, sea salt, mineral materials and carbonaceous species (Kim et al., 2001). To find which types of chemical components are the key for the visibility degradation during sampling period and how can the injection of FW particles change the visibility, the IMPROVE equation was used to calculate the light extinction

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coefficient ( $b_{\text{ext}}$ ) (Kim et al., 2001; Yang et al., 2007):

$$b_{\text{ext}} = 3f(\text{RH})[\text{ammonium sulfate}] + 3f(\text{RH})[\text{ammonium nitrate}] + 4[\text{OM}] + 1[\text{soil}] + 10[\text{BC}] \quad (8)$$

The visibility can be calculated by  $V_s = 3.91/b_{\text{ext}}$  (Yang et al., 2007). We used [ammonium sulfate] =  $0.944 \times [\text{NH}_4^+] + 1.02 \times [\text{SO}_4^{2-}]$ , [ammonium nitrate] =  $1.29 \times [\text{NO}_3^-]$ , [OM] =  $2.0 \times [\text{OC}]$ , [soil] = MIN and [BC] = [EC] (Yang et al., 2007; Tao et al., 2009).  $f(\text{RH})$  is the RH growth function indicating how scattering efficiencies increase for  $\text{SO}_4^{2-}$  and  $\text{NO}_3^-$  as they absorb liquid water and detailed data can be found in Tao et al. (2009).

#### 2.4.6 WRF-FLEXPART modeling

To see the transportation distance and decreasing trend of the highest mass concentrations of  $\text{PM}_{2.5}$  at CNY's Eve, Weather Research and Forecast model (WRF) version 3.4 was used to provide meteorological inputs of FLEXPART, with the NCEP global reanalysis meteorological dataset ( $1^\circ \times 1^\circ$ ) as initial and boundary conditions. The time step is set as 180 s. The grid system is divided into 28 layers in the vertical direction. Two domains are adopted, with the grid resolutions of 30 and 10 km. The simulating time period is 02:00–11:00 LT of 31 January, 2014. The output of 10 km evolution wind farms is used as the input of FLEXPART. For forward simulating of FLEXPART model, Nanjing ( $31^\circ 14' - 32^\circ 37' \text{ N}$ ,  $118^\circ 22' - 119^\circ 14' \text{ E}$ , height as 50 m) is considered as a whole source region. In this study, we only considered the dry and wet deposition of particles at CNY's Eve. The average mass concentration of  $572 \mu\text{g m}^{-3}$  is used as the initial particle concentration.



orated the air quality, with visibilities decreased to 1.4–6.2 km in the following three days. Therefore, the PM<sub>2.5</sub> pollution raised by FW can last for at least three days under unfavorable weather conditions (visibility was strongly related to PM<sub>2.5</sub> mass in 30 January–3 February,  $R^2 = 0.82$ ). At 3 and 5 February, PM<sub>2.5</sub> was cleaned effectively by the rainfall (precipitation as 0.3 and 9 mm), decreasing by 5.6–10.6 times for the ten sites.

### 3.1.2 Characteristic of chemical species

Tables 1 and 2 summarized the statistics of PM<sub>2.5</sub>, water-soluble ions, OC and EC before, during and after SF. The data at CNY and LF were listed separately for intensive FW activities. The ratios of PM<sub>2.5</sub> and chemical species at CNY and LF with those for the day before them were also listed. For elements, K, Si, Al, Na, Ca and Fe were the most abundant species, totally accounting for 85–90% for all of the elements at these five periods. Without considering the data for CNY, elements Al, K, Ba and Sr at SF still elevated when compared with those for Pre-SF, whereas all other elements decreased during SF. This was related to the weakened sources, like construction activities, vehicle emission and industrial activities in the national holiday. For ions, NO<sub>3</sub><sup>-</sup> was most abundant, averaged as 5.9 (±3.7) μg m<sup>-3</sup> for the days without intensive FW activities, then followed by SO<sub>4</sub><sup>2-</sup> and NH<sub>4</sub><sup>+</sup>, with average concentrations of 5.0 (±3.7) and 4.8 (±2.4) μg m<sup>-3</sup>, respectively. K<sup>+</sup> and Cl<sup>-</sup> also had relatively high concentrations of 0.9 (±1.0) and 0.8 (±0.5) μg m<sup>-3</sup>. It was similar to the results of Wang et al. (2014) that the secondary aerosols in Nanjing are dominated by nitrate in winter haze periods. Na<sup>+</sup>, Ca<sup>2+</sup>, Mg<sup>2+</sup>, F<sup>-</sup> and NO<sub>2</sub><sup>-</sup> accounted for a minor fraction (totally of 4.2%). For carbonaceous species, OC3, OC4 and EC1 were most abundant, in combination accounting for 80% of the total carbon, indicating coal combustion and gasoline exhaust were important sources for PM<sub>2.5</sub> at Nanjing (Cao et al., 2005). OPC and OC1 also showed a higher values which may be related to biomass burning (Cao et al., 2005). NO<sub>3</sub><sup>-</sup>, OC and EC were highest for Pre-SF period, as the “spring travel rush” effect, characterized by extremely high traffic flows (Huang et al., 2012). The PM<sub>2.5</sub> concentrations

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ing Chinese Spring Festival at Beijing, organics, nitrate, sulfate, ammonium, BC and chloride, accounted for 43, 22, 14, 13, 5 and 3 % of  $PM_{1.0}$  with no elements considered (Jiang et al., 2014). The combination of filter sampling-offline chemical analysis and online real-time monitoring of chemical species are desired for atmospheric aerosol studies.

Figure 7a shows the calculated average values of extinction coefficient ( $b_{\text{ext}}$ ). They varied between 44 and  $525 \text{ Mm}^{-1}$ . The calculated visibility ( $V_{\text{cal}}$ ) exhibited good correlations with the measured one ( $V_{\text{meas}}$ ) ( $V_{\text{cal}} = 0.96V_{\text{meas}} + 1.4$ ,  $R^2 = 0.45$ ,  $P < 0.001$ ). The discrepancy between the measured and calculated values can be attributed to the influence of ambient water vapor (Huang et al., 2012) which was a key component of aerosol particles in this study as discussed before. At higher RH, the role of ambient water vapor on light extinction was more important (Huang et al., 2012). The  $b_{\text{ext}}$  exhibited higher values at CNY and the following three days, in the range of  $301\text{--}525 \text{ Mm}^{-1}$ .  $(\text{NH}_4)_2\text{SO}_4$  had the largest contribution to  $b_{\text{ext}}$ , accounting for  $36.5 \pm 11.7\%$ , followed by  $\text{NH}_4\text{NO}_3$  ( $25.8 \pm 8.2\%$ ), EC ( $21.8 \pm 9.0\%$ ), OM ( $10.8 \pm 4.2\%$ ) and soil ( $5.1 \pm 1.8\%$ ) (Fig. 7b). These results implied that sulfate was the largest contributor to visibility degradation especially at SF period. Before SF, the contribution of EC can be as high as 34–44 % during serious haze periods (24–26 January,  $\text{RH} < 51\%$ ). It highlights the importance of controlling vehicle emissions in haze days. Sulfate and soot were also found as the main contributors to visibility degradation in other studies (Tao et al., 2009). The results obtained here were different from those in Shanghai that organic aerosol had the largest contribution to the aerosol extinction as 47 %, then followed by sulfate ammonium (22 %), nitrate ammonium (14 %) and EC (12 %) (Huang et al., 2012). The FW at CNY obviously changed the contributions of chemical species to scattering coefficient. At CNY, the influence of FW particles to visibility was mainly controlled by  $(\text{NH}_4)_2\text{SO}_4$  (36 %),  $\text{NH}_4\text{NO}_3$  (26 %), EC (15 %) and OM (15 %). During FW particle aging processes, the contribution of  $(\text{NH}_4)_2\text{SO}_4$  increased from 36 % (30 January) to 67 % (3 February), while for  $\text{NH}_4\text{NO}_3$ , its contribution increased first to 28 % at 31 January and then decreased to 10 % (3 February). Similar trend was found for EC,

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it increased to 22 % (1 February) and then decreased to 7.6 % (3 February). For OM, it exhibited decreasing trend, to the lowest value at 2 February (5.5 %).

### 3.3 Source evolution

#### 3.3.1 Contributions of fireworks burning to PM<sub>2.5</sub> at CNY and LF

The PM<sub>2.5</sub> concentrations peaked at CNY and LF in this study with intensive FW activities. Assuming that the PM<sub>2.5</sub> collected on CNY and LF was a simple combination of particles from FW and other emission sources, the contribution and compositions of the particles from FW can be estimated by subtracting the contribution of the non-firework sources (Feng et al., 2012). An accumulation factor of 0.88 (mixing layer heights at 30 and 29 January were 227 and 257 m, respectively) and 0.51 (mixing layer heights at 14 and 13 February were 353 and 696 m, respectively) were used to eliminate the disperse impact caused by mixing height layer increasing (Feng et al., 2012; Deka and Hoque, 2014). About 60.1 and 10.9 % of the PM<sub>2.5</sub> masses at CNY and LF were estimated to be from FW. The contributions to almost all species of FW at CNY were more obvious than those at LF (Fig. 8). At CNY, OPC, EC3, Ba, Sr, Soot-EC, EC2 and OC1 maintained the higher contributions from FW, larger than 80 %, followed by K, K<sup>+</sup>, Al, Bi and Char-EC, with the contributions higher than 70 %. At LF, 45 % of Ba and 41 % of Sr were from FW and the contributions to V, Bi, Fe, Be, K, K<sup>+</sup>, Al, NO<sub>2</sub><sup>-</sup> and Sb were higher than 20 %.

#### 3.3.2 Source variations by tracers and PCA analysis

In order to further identify the sources of particles, PCA results are shown in Fig. 9 and Supplement S8.

Five, five and eight factors were extracted for the dataset of Pre-SF, SF and After-SF period, which explained 100, 99.4 and 97.5 % of the variance, suggesting almost all the sources have been explained. Fireworks burning was the most important source for

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PM<sub>2.5</sub>, contributing 24.0% for the whole period. Coal combustion (including both local and regional transport) was also important to PM<sub>2.5</sub> at this site in winter (contributing 19.3%). It was verified by the time series of tracers (As and Sb, highly correlated with  $R^2 = 0.88$ ) for coal combustion in Fig. 10a. Extremely high values can be found at 24 January, even higher than those for the intensive FW day. This can be explained by two reasons: (1) on 24 January, it was a haze day with visibility and mixing layer height as 8.2 km and 321 m, facilitating the acculumation of pollutants; (2) the air masses on 24 January were classified as cluster 2 from central China where the Chang-Zhu-Tan region with intensive non-ferrous metal metallurgy industries with huge coal consumption is located. On this day, higher concentrations were also found for metals like Mg (228 ng m<sup>-3</sup>), Cr (17 ng m<sup>-3</sup>), Ca (782 ng m<sup>-3</sup>), Mn (85 ng m<sup>-3</sup>), Fe (1014 ng m<sup>-3</sup>), Co (0.61 ng m<sup>-3</sup>), Ni (7.6 ng m<sup>-3</sup>), Cu (124 ng m<sup>-3</sup>), Mo (4.4 ng m<sup>-3</sup>), Cd (8.8 ng m<sup>-3</sup>), Sn (16 ng m<sup>-3</sup>) and Pb (359 ng m<sup>-3</sup>). FW contributed about half of the PM<sub>2.5</sub> during SF. For Pre-SF and After-SF periods, its contributions were 9 and 14%. As shown in Fig. 10b, the tracers of Ba and Sr exhibited much higher values at SF, and there were also some small peaks before SF (27 January) and after SF (14 February). During the one-week holiday, most industrial activities shut down in China with low energy consumption during this period (Feng et al., 2012; Huang et al., 2012; Li et al., 2013). As holiday effect, when compared the source contributions at SF to those at Pre-SF, the contributions of coal combustion decreased from 46 to only 16%, the contributions of vehicle emissions decreased from 17 to 14% and the contributions of dust also decreased significantly. For After-SF period, along with the gradually re-starting of industrial plants and contrustion activities and increasing traffic density, the contributions from coal combustion, heavy oil burning, industrial processes and road dust increased, to 37, 4, 3.4 and 5%, respectively. OC and EC wers selected as the tracers of vehicle emission and their correlation was higher to 0.80 without the data at CNY and LF. Ca was selected as the tracer for construction activities or road dust (the main road was covered by abundant dust from subway construstion). Na<sup>+</sup> was used as the tracer of sea salt. Li, Be and Si were selected as the indicators of soil ( $R^2$  higher than 0.65 between them). The vari-





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for phenol formaldehyde resin is 80–85 °C, all lower than the measuring temperature for OC2 (230 °C). Therefore, OC1, OC2 and OPC are more easily and directly emitted from FW. The initially decreasing OC/EC ratio indicated the gradual formation of secondary (aged) organic aerosols (Feng et al., 2012). To summarize, the heterogeneous reactions of SO<sub>2</sub> and NO<sub>x</sub> on crustal materials directly from FW, the replacement of Cl<sup>-</sup> by NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup>, coating of NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup> on soot and formation of secondary organic aerosols are the main of FW particles' aging mechanisms.

Previous studies also indicated that metals like Fe, Cu and Mn can catalyze the formation of NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup> (Wang et al., 2007; Do et al., 2012; Feng et al., 2012). Wang et al. (2007) indicated that the heterogeneous formation of sulfate through reacting with α-Fe<sub>2</sub>O<sub>3</sub> exists under moist atmosphere and is a function of RH and HNO<sub>3</sub>. As shown in Fig. 12 and Supplement S9, significant correlations are found between NO<sub>3</sub><sup>-</sup> and Fe, Cu and Mn and between SO<sub>4</sub><sup>2-</sup> and Fe, Cu and Mn at higher RH. It indicated the mechanism of metal-catalyzed formation of NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup> was more likely to occur at higher RH. At the day of CNY and 2 February, the RH was 88 and 87 %, implying the occurrence of metal-catalyzed reactions, which may also explain the higher NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup> concentrations at these days. The *R*<sup>2</sup> increased with the elevated RH. There are significant increases of *R*<sup>2</sup> for Fe-NO<sub>3</sub><sup>-</sup>, Mn-NO<sub>3</sub><sup>-</sup> and Cu-NO<sub>3</sub><sup>-</sup> when RH were > 85 %, > 85 % and > 65 %, respectively. For the correlations of Fe-SO<sub>4</sub><sup>2-</sup>, Mn-SO<sub>4</sub><sup>2-</sup> and Cu-SO<sub>4</sub><sup>2-</sup>, clear increasings in *R*<sup>2</sup> were found when RH were > 85 %, > 85 % and > 55 %. Then we can conclude that for Fe-catalyzed and Mn-catalyzed reactions, the threshold of RH was around 85 %; for Cu-catalyzed formation of NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup>, the thresholds of RH were around 65 and 55 %, respectively. For RH higher than 90 %, the *R*<sup>2</sup> decreased for Fe-NO<sub>3</sub><sup>-</sup>, Cu-NO<sub>3</sub><sup>-</sup>, Fe-SO<sub>4</sub><sup>2-</sup> and Mn-SO<sub>4</sub><sup>2-</sup> which needs more data in the future to be verified and explained.

### 3.4.2 Potential health risks during the aging of fireworks burning particles

After emitted from the intensive FW at CNY's Eve, heavy metals including Vi, Cr, Mn, Co, Ni, Cu, Zn, As, Cd and Pb decreased directly in the following four days (Supplement S10). It indicated that they were emitted from FW primarily and were removed mainly by dispersion and deposition during aging. Then we can assume that: (1) other emission sources (coal combustion, vehicle exhaust and industrial processes were their main sources) for these metals were stable during SF period; (2) the highest concentrations of them were just the combinations of particles from the FW on 30 January and other sources on 29 January. We subtract the concentrations of elements at 29 January from the corresponding values on 30 and 31 January, 1 and 2 February for each element, respectively, to obtain the elements concentrations just related with the FW particles. Considering the variations of mixing layer height (as 306, 257, 227, 284 and 248 m for the five days of 29 January–2 February) and accumulation effect, the concentrations of these elements on 30 and 31 January, 1 and 2 February were transferred by factors of 0.84, 0.74, 0.93 and 0.81, respectively. Then the  $C_{95\% \text{ UCL}}$  values raised by FW particles were 7.4, 21.2, 12.0, 23.7, 0.42, 5.2, 54.5, 11.7, 126 and 837  $\text{ngm}^{-3}$  of the ten elements, respectively, for the four days. The non-cancer risks raised just by FW at CNY's Eve were below 1 (0.03 and 0.02 for children and adult) (Fig. 13), indicating no adverse health effects. However the cancer risk was  $1.62 \times 10^{-6}$ , higher than  $10^{-6}$ , suggesting the FW particles were important carcinogens and should be controlled effectively especially for the contained Cd and As. Meanwhile, the risk level for children was higher than that for adult, which meant that children were more sensitive to non-carcinogenic effects and should be kept from possible exposure to them (Yang et al., 2014). Therefore, the FW should be restricted at SF from the view of their cancer risks to human health.

## 4 Conclusions

Thirty-nine elements, ten water-soluble ions and eight fractions of carbonaceous species were measured to fully characterize the chemical compositions in atmospheric PM<sub>2.5</sub> in Nanjing during the 2014 Chinese Spring Festival (SF). Serious regional haze pollution occurred and lasted during the whole sampling period. At the Chinese New Year (CNY)'s Eve, after it peaked at 02:00 LT, PM<sub>2.5</sub> exponentially decreased in the following 11 h to the level before extensive fireworks burning (FW) activities. Due to holiday effect, almost all elements decreased during SF except for Al, K, Ba and Sr which were related with FW. As the "spring travel rush" effect, NO<sub>3</sub><sup>-</sup>, OC and EC showed highest values for periods before SF, indicating the extremely high traffic flows. At the New Year's Eve, about 60.1 % of the PM<sub>2.5</sub> mass was estimated to be from FW. Highly elevated Ba and Sr were also found and highly correlated, indicating they can be used as the tracers of FW. The intensive FW at CNY's Eve obviously changed the chemical compositions of PM<sub>2.5</sub>, with elevated organic matter (OM) immediately, contributing 39.3 % of PM<sub>2.5</sub>. The contributions of secondary ions formed by gas-particle transformations gradually increased during FW particles' aging processes. After FW particles emitted at CNY, the contribution of (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> to extinction coefficient increased from 36 to 67 %; while for NH<sub>4</sub>NO<sub>3</sub> and EC, their contributions increased first and then decreased. According to tracers and principle component analysis, fireworks burning was the most important source of PM<sub>2.5</sub> at this site. FW contributed about half of PM<sub>2.5</sub> during SF. Tracers for various sources (As and Sb for coal combustion, Ba and Sr for FW, OC and EC for vehicle emission, Ca for construction or road dust, Na<sup>+</sup> for sea salt and Li, Be and Si for soil) performed well and they varied accordantly with the contributions of corresponding sources. The FW particles emitted from CNY's Eve may undergo at least six days aging processes from the decreasing trend of SO<sub>4</sub><sup>2-</sup>. The aging processes were characterized by heterogeneous reactions of SO<sub>2</sub> and NO<sub>x</sub> on crustal materials directly from FW, the replacement of Cl<sup>-</sup> by NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup>, coating of NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup> on soot, formation of secondary organic aerosols and metal-catalyzed formation

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**Table 1.** Mass concentrations of elements in PM<sub>2.5</sub> before, during and after 2014 Chinese Spring Festival at Nanjing (ngm<sup>-3</sup>).

Elements	Pre-SF	SF (without CNY)	After-SF (without LF)	CNY	LF	CNY/pre-CNY	LF/pre-LF
Li	1.9 ± 0.7	1.1 ± 0.8	1.4 ± 1.0	2.3	1.4	1.8	1.3
Be	0.1 ± 0.0	0.0 ± 0.0	0.0 ± 0.0	0.0	0.0	2.1	1.9
Na	826 ± 209	417 ± 196	435 ± 172	563	506	1.3	1.1
Mg	124 ± 79	97 ± 89	106 ± 30	196	124	2.6	1.3
Al	612 ± 234	802 ± 550	530 ± 230	3127	811	7.9	1.7
Si	1105 ± 420	752 ± 147	847 ± 422	1431	943	2.3	1.4
P	66 ± 35	35 ± 20	29 ± 13	42	35	1.5	1.3
K	1940 ± 725	3330 ± 3848	1276 ± 817	14 336	2940	9.3	1.8
Ca	454 ± 198	355 ± 95	383 ± 193	454	469	1.9	1.6
Sc	0.1 ± 0.0	0.1 ± 0.0	0.1 ± 0.1	0.1	0.1	1.5	1.4
Ti	31 ± 27	14 ± 8	24 ± 28	20	15	1.2	1.7
V	7.7 ± 2.3	7.0 ± 7.8	6.6 ± 4.6	9.4	10.5	1.2	2.4
Cr	10.4 ± 4.0	7.9 ± 5.7	8.7 ± 3.4	17.4	10.2	2.9	1.3
Mn	48 ± 22	22 ± 19	67 ± 67	47	50	1.9	1.3
Fe	385 ± 342	160 ± 140	261 ± 172	304	300	1.8	1.9
Co	0.4 ± 0.1	0.4 ± 0.4	0.4 ± 0.2	0.6	0.5	2.6	1.6
Ni	8.5 ± 6.0	4.5 ± 3.1	8.9 ± 5.6	5.9	7.6	1.3	1.5
Cu	102 ± 30	81 ± 63	77 ± 25	170	116	1.3	1.3
Zn	257 ± 145	93 ± 81	194 ± 154	242	174	1.6	1.3
As	17 ± 13	9.0 ± 9.8	8.0 ± 4.6	22	8.8	2.2	1.3
Rb	8.5 ± 2.2	4.7 ± 4.2	5.7 ± 4.5	12	6.5	1.7	1.5
Sr	5.1 ± 3.9	34 ± 49	4.9 ± 5.9	181	20.5	79.4	5.7
Y	0.2 ± 0.1	0.1 ± 0.1	0.2 ± 0.1	0.1	0.2	1.5	1.5
Zr	1.9 ± 0.7	1.6 ± 0.6	1.9 ± 1.3	1.6	2.3	1.2	1.4
Mo	2.7 ± 1.0	1.3 ± 1.2	2.9 ± 1.8	2.8	2.7	1.6	0.6
Cd	3.4 ± 2.8	1.3 ± 1.4	1.1 ± 0.7	4.1	1.3	1.7	1.3
Sn	9.0 ± 5.6	5.0 ± 6.1	4.6 ± 3.5	18.0	4.5	3.7	1.1
Sb	11 ± 6.5	4.8 ± 5.4	3.0 ± 1.6	13.9	4.8	2.9	1.7
Cs	1.8 ± 1.0	0.7 ± 0.8	1.4 ± 1.5	1.9	1.2	1.4	1.4
Ba	14 ± 12	152 ± 216	14.4 ± 20.4	850	70.1	99.1	9.7
La	0.7 ± 0.6	0.6 ± 0.5	0.6 ± 0.5	0.6	0.5	2.5	0.8
Ce	0.8 ± 0.5	0.4 ± 0.3	0.6 ± 0.6	0.6	0.5	2.3	1.0
Sm	0.1 ± 0.0	0.0 ± 0.0	0.0 ± 0.0	0.0	0.0	1.4	1.2
W	4.3 ± 2.4	3.1 ± 2.9	3.2 ± 1.3	4.7	5.4	0.6	1.1
Tl	1.4 ± 0.7	0.5 ± 0.5	0.8 ± 0.5	1.4	0.8	1.4	1.2
Pb	187 ± 96	116 ± 132	91 ± 54	425	127	2.6	1.0
Bi	7.3 ± 5.4	9.1 ± 13.0	3.0 ± 2.2	36.9	6.5	7.2	2.3
Th	0.6 ± 0.0	0.5 ± 0.0	0.6 ± 0.0	0.6	0.6	1.0	1.0
U	0.1 ± 0.0	0.0 ± 0.0	0.1 ± 0.1	0.1	0.1	1.4	1.2

Pre-SF indicated the period before 2014 Chinese Spring Festival (SF), covering the days from 24 to 29 January; SF indicated the Spring Festival period, covering the days from 30 January to 6 February; After-SF indicated the period after Spring Festival, covering the days from 12 to 21 February. CNY indicates the Chinese New Year (CNY) day and in 2014, it is 30 January; pre-CNY is the day before CNY and is 29 January. LF indicates the Lantern Festival (LF) day and in 2014, it is 14 February; pre-LF is the day before LF and is 14 February. At CNY and LF, intensive fireworks were burned. Therefore, the mass concentrations of PM<sub>2.5</sub> and associated chemical species of the two days were listed separately. CNY/pre-CNY indicates the ratios between mass concentrations of PM<sub>2.5</sub> and associated chemical species at CNY and the day before CNY (pre-CNY). LF/pre-LF indicates the ratios between mass concentrations of PM<sub>2.5</sub> and associated chemical species at LF and the day before LF (pre-LF).

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**Table 2.** Mass concentrations of ions and carbonaceous species in PM<sub>2.5</sub> before, during and after 2014 Chinese Spring Festival at Nanjing ( $\mu\text{g m}^{-3}$ ).

Elements	Pre-SF	SF (without CNY)	After-SF (without LF)	CNY	LF	CNY/pre-CNY	LF/pre-LF
Na <sup>+</sup>	0.65 ± 0.17	0.37 ± 0.07	0.31 ± 0.09	0.40	0.29	0.7	1.0
NH <sub>4</sub> <sup>+</sup>	4.26 ± 1.12	4.81 ± 3.68	5.33 ± 1.98	10.5	6.85	2.4	1.3
K <sup>+</sup>	0.49 ± 0.31	1.59 ± 1.55	0.67 ± 0.22	4.81	1.06	8.9	1.8
Ca <sup>2+</sup>	0.09 ± 0.05	0.23 ± 0.14	0.19 ± 0.09	0.25	0.18	3.3	0.8
Mg <sup>2+</sup>	0.03 ± 0.03	0.20 ± 0.31	0.08 ± 0.04	0.10	0.09	1.3	1.5
F <sup>-</sup>	0.03 ± 0.01	0.04 ± 0.02	0.03 ± 0.01	0.05	0.04	2.2	1.4
Cl <sup>-</sup>	1.16 ± 0.26	0.72 ± 0.54	0.75 ± 0.40	3.04	1.09	3.8	1.2
NO <sub>2</sub> <sup>-</sup>	0.02 ± 0.01	0.04 ± 0.01	0.04 ± 0.01	0.02	0.05	0.8	1.7
NO <sub>3</sub> <sup>-</sup>	5.81 ± 2.54	5.27 ± 5.38	6.59 ± 2.98	14.6	8.36	2.5	1.3
SO <sub>4</sub> <sup>2-</sup>	3.46 ± 1.28	7.74 ± 5.17	4.05 ± 2.34	16.1	4.54	4.4	1.1
OC	18.2 ± 7.45	9.38 ± 6.70	8.95 ± 3.15	56.9	12.2	4.1	1.1
EC	6.93 ± 4.64	3.12 ± 3.06	2.96 ± 1.20	8.29	3.38	1.8	0.9
OC1	1.87 ± 1.40	0.40 ± 0.23	0.50 ± 0.16	16.1	0.64	33.2	1.1
OC2	4.32 ± 1.65	2.56 ± 1.75	2.36 ± 0.74	14.8	2.86	4.2	1.1
OC3	5.46 ± 2.69	2.77 ± 2.09	2.90 ± 1.15	9.10	4.28	2.4	1.0
OC4	5.44 ± 1.52	3.27 ± 3.03	2.94 ± 1.36	5.81	4.46	0.9	1.1
EC1	7.92 ± 5.07	3.44 ± 2.64	3.15 ± 1.00	18.8	3.30	4.1	0.9
EC2	0.12 ± 0.07	0.07 ± 0.06	0.06 ± 0.05	0.50	0.08	37.7	1.6
EC3	0.00 ± 0.00	0.00 ± 0.00	0.00 ± 0.00	0.11	0.00	–	–
OPC	1.12 ± 2.43	0.39 ± 0.39	0.24 ± 0.43	11.09	0.00	–	–
PM <sub>2.5</sub>	119.6 ± 43.4	106.9 ± 84.1	88.9 ± 37.4	318.0	118.4	3.1	1.3

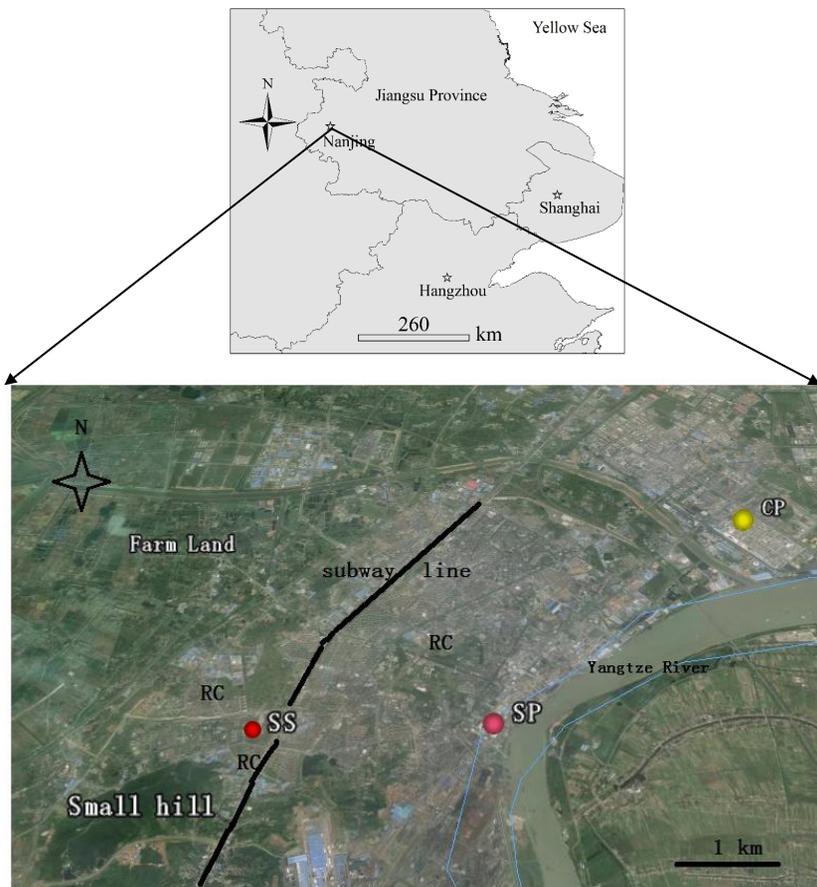
– Not detected in the day before CNY and LF.

Pre-SF indicated the period before 2014 Chinese Spring Festival (SF), covering the days from 24 to 29 January; SF indicated the Spring Festival period, covering the days from 30 January to 6 February; After-SF indicated the period after Spring Festival, covering the days from 12 to 21 February. CNY indicates the Chinese New Year (CNY) day and in 2014, it is 30 January; pre-CNY is the day before CNY and is 29 January. LF indicates the Lantern Festival (LF) day and in 2014, it is 14 February; pre-LF is the day before LF and is 14 February. At CNY and LF, intensive fireworks were burned. Therefore, the mass concentrations of PM<sub>2.5</sub> and associated chemical species of the two days were listed separately. CNY/pre-CNY indicates the ratios between mass concentrations of PM<sub>2.5</sub> and associated chemical species at CNY and the day before CNY (pre-CNY). LF/pre-LF indicates the ratios between mass concentrations of PM<sub>2.5</sub> and associated chemical species at LF and the day before LF (pre-LF).



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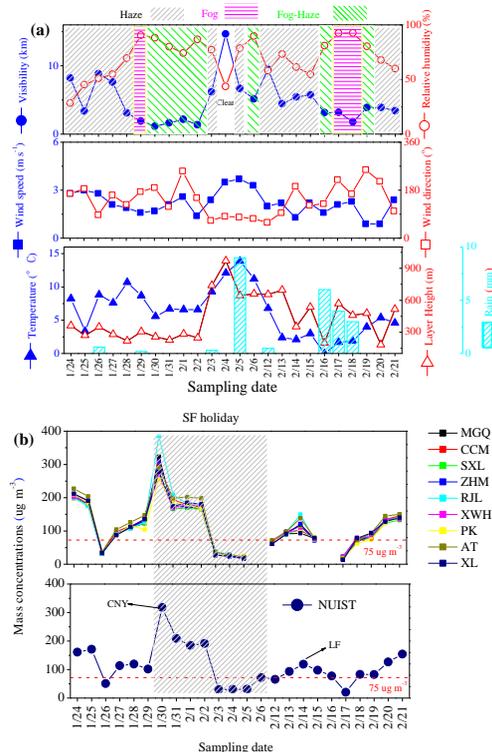
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**Figure 1.** Location of the sampling site (the red dot, SS). SP: iron smelt plant; CP: chemical industrial park; RC: residential community. The black line indicates the subway line near SS.

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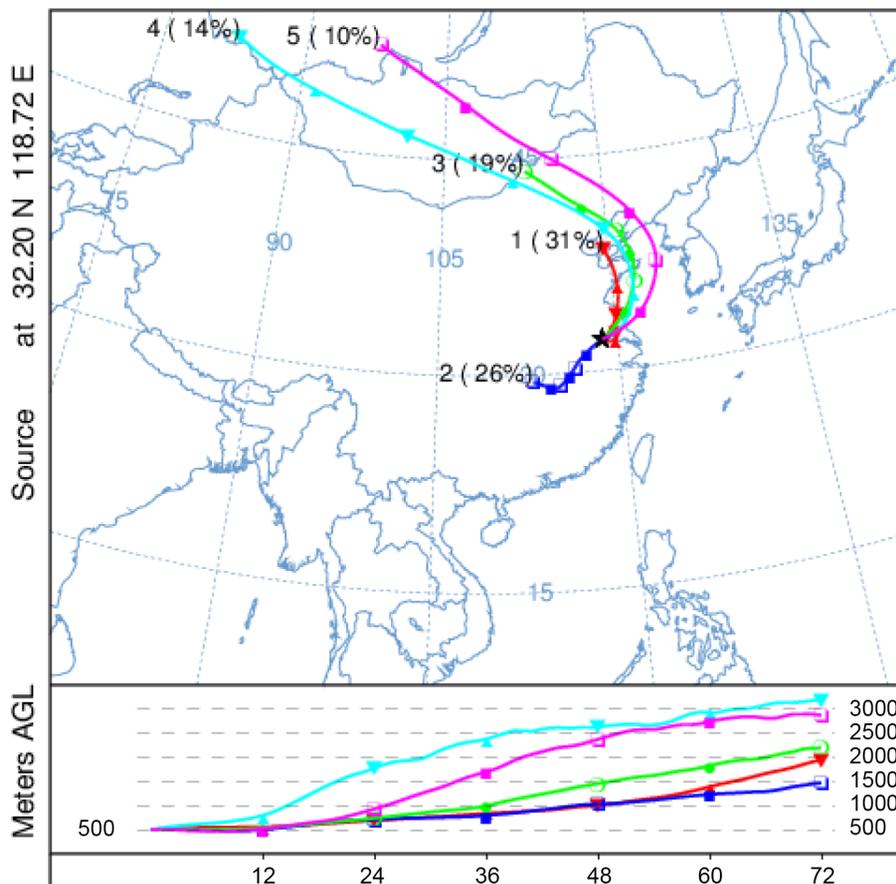
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**Figure 2.** Daily variation of meteorological factors (a) and mass concentrations of  $PM_{2.5}$  (b) during sampling periods. MGQ, CCM, SXL, ZHM, RJL, XWH, PK, AT and XL indicate the nine urban air quality monitoring sites set by Jiangsu Environmental Monitoring Center as Supplement S4 shown. NUIST indicates the sampling site of this study at Nanjing University of Information Science and Technology. SF means Spring Festival (between 30 January–7 February); CNY means Chinese New Year day (at 30 January); LF means Lantern Festival day (at 14 February).  $75 \mu m^{-3}$  is the 24 h-averaged secondary standard for  $PM_{2.5}$  of China National Ambient Air Quality.

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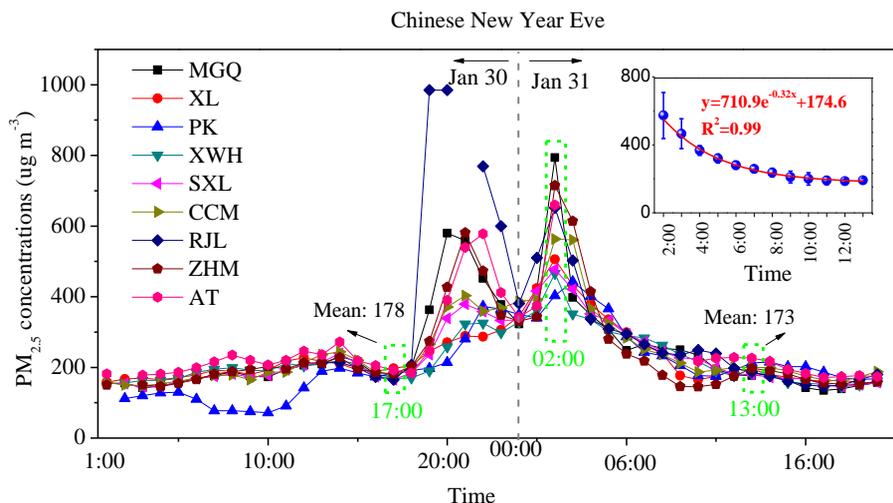
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**Figure 3.** Cluster analysis for backward trajectories of air masses during the whole sampling periods. The different colors indicated different clusters of the backward trajectories.

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**Figure 4.** Hourly  $PM_{2.5}$  concentrations before, during and after Chinese New Year (CNY)'s Eve of 2014 at Nanjing. The figure at the top-right corner indicates the decreasing trend of averaged  $PM_{2.5}$  mass concentrations after 02:00 LT of CNY's Eve (at 30 January 2014). Hourly data for the nine sites in urban Nanjing (locations were shown in Supplement S4) were collected from the public platform at <http://218.94.78.75/jsair/>.

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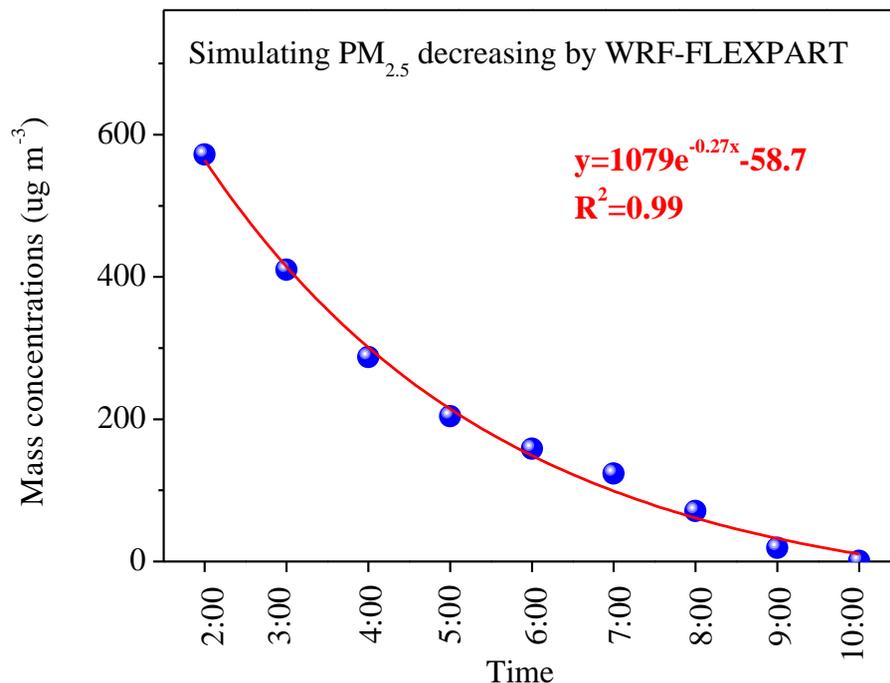
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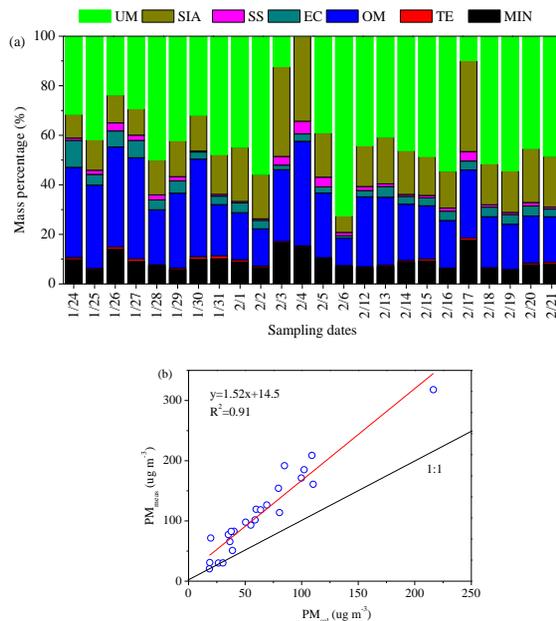


**Figure 5.** Particles decreasing trend at Chinese New Year's Eve by WRF-FLEXPART simulating. The particle concentrations decreased to  $0.76 \mu\text{g m}^{-3}$  after nine hours, faster than the real decrease as Fig. 4 shows. It can be explained as that we did not consider the injections of particles during simulating course, which biases the result.

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**Figure 6.** Mass balance of chemical species in  $PM_{2.5}$  at sampling periods **(a)** and relationship between the mass calculated by adding individual components and the gravimetrically measured particulate mass **(b)**.  $MIN = 2.14 \times Si + 1.67 \times Ti + 1.89 \times Al + 1.59 \times Mn + 1.67 \times Mg + 1.95 \times Ca + 1.35 \times Na + 1.21 \times K + 1.43 \times Fe$ . Trace elements (TE) indicated the sum of all other elements except for those used in calculating MIN. Sea salt (SS) =  $[Na^+] + [ss-Cl^-] + [ss-Mg^{2+}] + [ss-K^-] + [ss-Ca^{2+}] + [ss-SO_4^{2-}]$ ;  $ss-Cl^- = 1.8 \times [Na^+]$ ;  $ss-Mg^{2+} = 0.12 \times [Na^+]$ ;  $ss-K^- = 0.036 \times [Na^+]$ ;  $ss-Ca^{2+} = 0.038 \times [Na^+]$ ;  $ss-SO_4^{2-} = 0.252 \times [Na^+]$ . Secondary inorganic aerosol (SIA) was calculated as the sum of  $nss-SO_4^{2-}$ ,  $NO_3^-$  and  $NH_4^+$ . Organic matter (OM) =  $2.0 \times OC$ . UM indicated unidentified matter. 30 January–7 February belongs to the Chinese Spring Festival in 2014; 30 January is the Chinese New Year day; 14 February is the Lantern Festival day.

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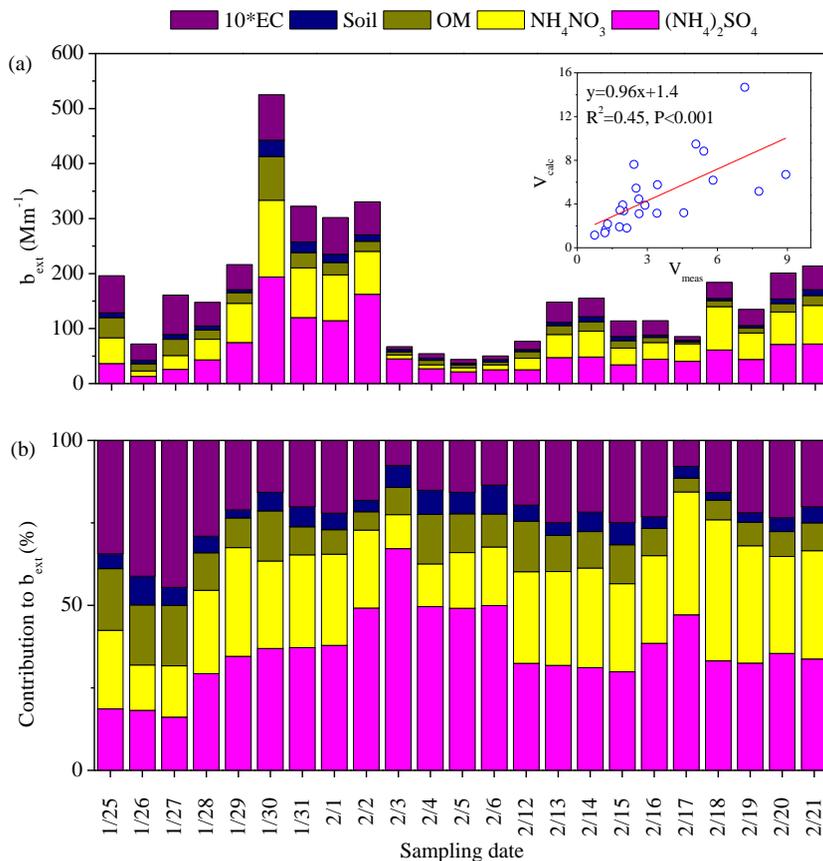
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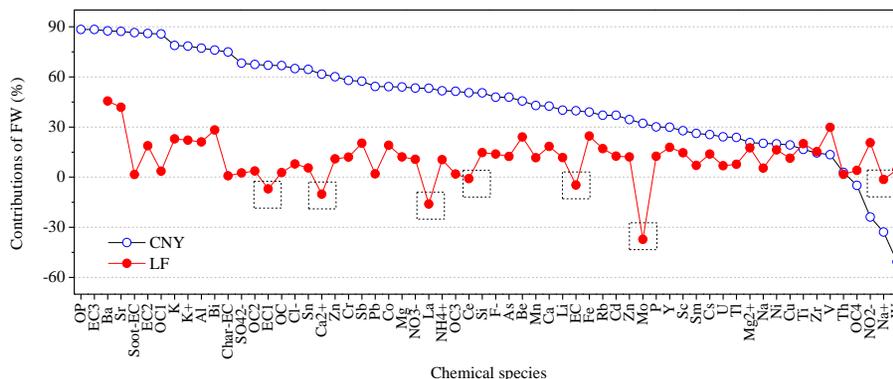
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**Figure 7.** Extinction coefficients of chemical components in  $\text{PM}_{2.5}$  at Nanjing **(a)** and relative contribution to light extinction of chemical components in  $\text{PM}_{2.5}$  **(b)**. 30 January–7 February belongs to the Chinese Spring Festival in 2014; 30 January is the Chinese New Year day; 14 February is the Lantern Festival day.  $V_{\text{calc}}$  indicates the visibility calculated by adding individual components;  $V_{\text{meas}}$  is the measured visibility.

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**Figure 8.** Contributions of fireworks burning (FW) at CNY and LF to chemical species in PM<sub>2.5</sub>. CNY means Chinese New Year day (30 January); LF means Lantern Festival day (14 February).

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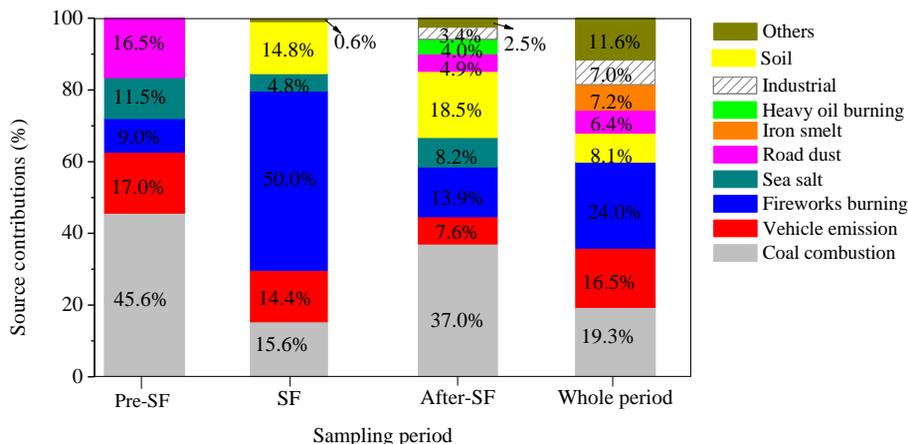
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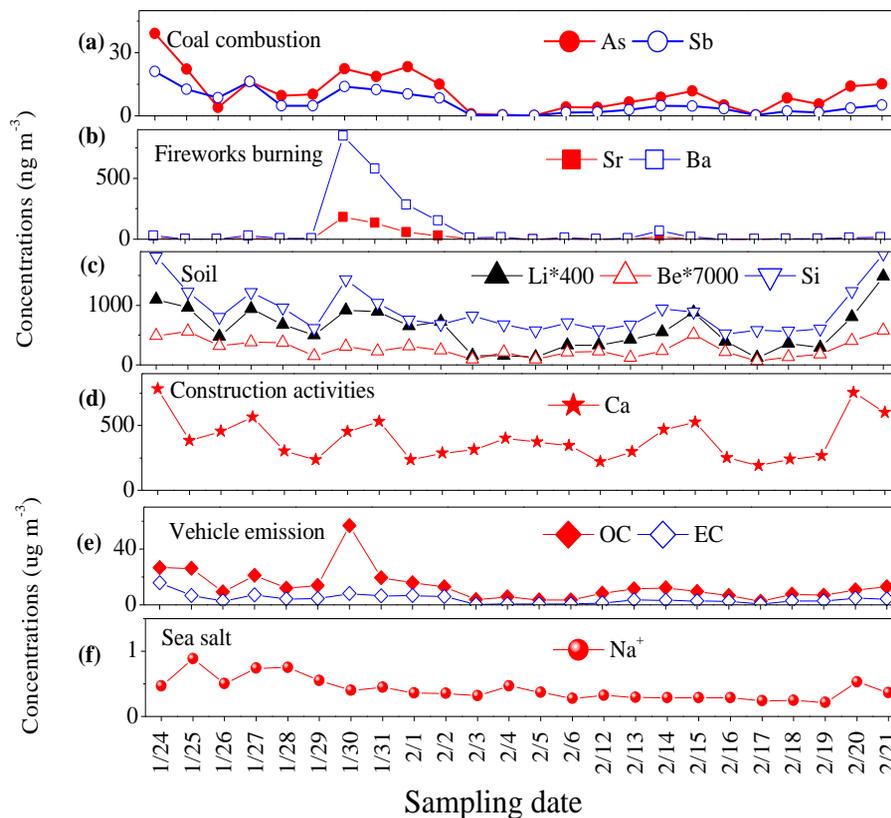
**Figure 9.** Source contributions to atmospheric  $PM_{2.5}$  before, during, after 2014 Spring Festival (SF) and the whole period by principal component analysis. Pre-SF indicated the period before SF, covering the days from 24 to 29 January; SF indicated the period during SF, covering the days from 30 January to 6 February; After-SF indicated the period after SF, covering the days from 12 to 21 February.

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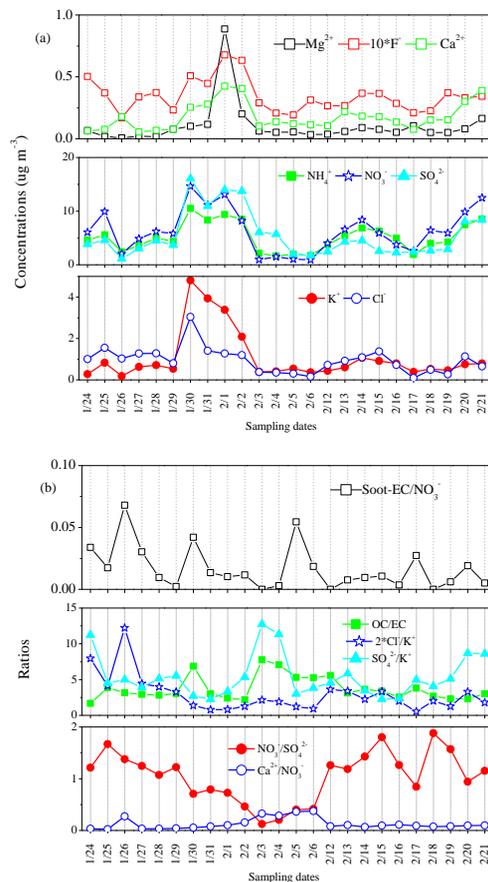
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**Figure 10.** Time series of tracers for **(a)** coal combustion, **(b)** fireworks burning, **(c)** soil, **(d)** construction activities or road dust, **(e)** vehicle emission and **(f)** sea salt. The y axis units for **(a–d)** are  $\text{ng m}^{-3}$  and are  $\mu\text{g m}^{-3}$  for **(e and f)**. 30 January–7 February belongs to the Chinese Spring Festival in 2014; 30 January is the Chinese New Year day; 14 February is the Lantern Festival day.

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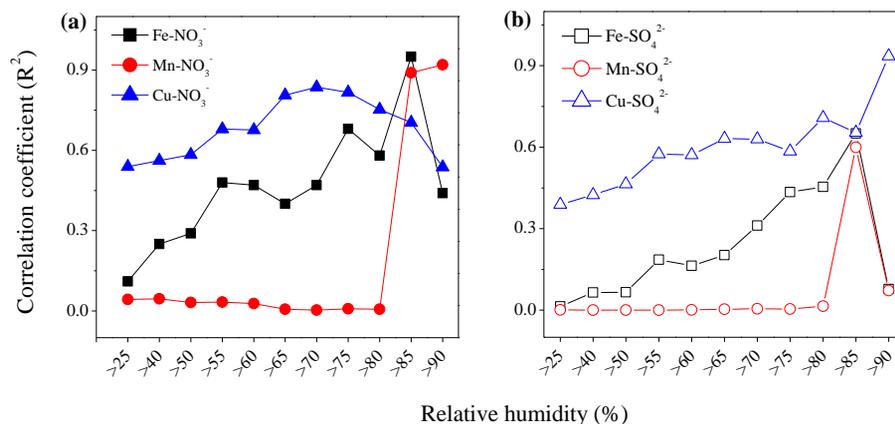
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**Figure 11.** Time series of ions **(a)** and specific ratios **(b)** during sampling period. 30 January–7 February belongs to the Chinese Spring Festival in 2014; 30 January is the Chinese New Year day; 14 February is the Lantern Festival day.

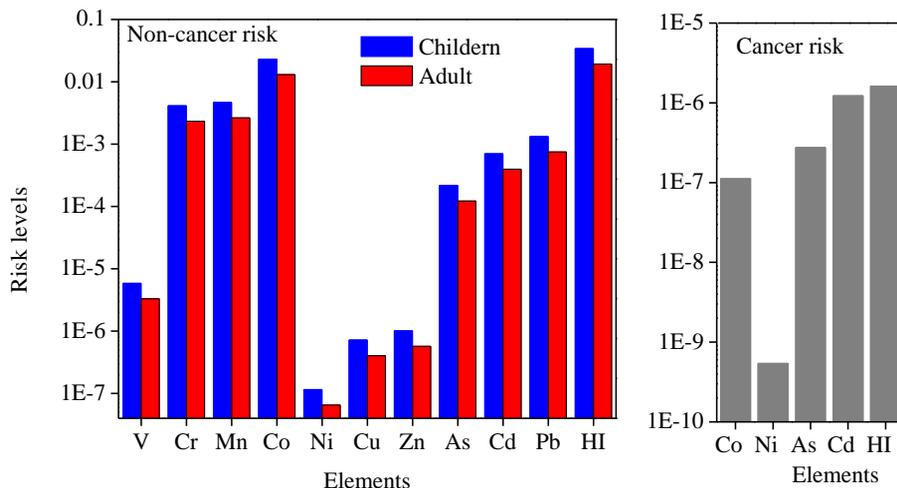
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**Figure 13.** Non-cancer and cancer risks for heavy metals raised by intensive fireworks burning at Nanjing.

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