

1

2 **Seasonal differences in oxygenated organic aer-** 3 **osol (OOA) composition: implications for emis-** 4 **sions sources and factor analysis**

5 F. Canonaco, J. G. Slowik, U. Baltensperger, A. S. H. Prévôt

6 Laboratory of Atmospheric Chemistry, Paul Scherrer Institute, CH-5232 Villigen
7 PSI, Switzerland

8 **Abstract**

9 Aerosol chemical speciation monitor (ACSM) measurements were performed in
10 Zurich, Switzerland for 13 months (February 2011 through February 2012). Many
11 previous studies using this or related instruments have utilized the fraction of or-
12 ganic mass measured at m/z 44 (f_{44}), which is typically dominated by the CO_2^+ ion
13 and related to oxygenation, as an indicator of atmospheric aging. The current
14 study demonstrates that during summer afternoons, when photochemical pro-
15 cesses are most vigorous as indicated by high oxidant OX ($\text{O}_3 + \text{NO}_2$), f_{44} for ambi-
16 ent SOA is not higher but is rather similar or lower than on days with low OX. On
17 the other hand, f_{43} (less oxidized fragment) tends to increase. These changes are
18 discussed in the f_{44}/f_{43} space frequently used to interpret ACSM and aerosol
19 mass spectrometer (AMS) data. This is likely due to the formation of semi-volatile
20 oxygenated aerosol produced from biogenic precursor gases, whose emissions in-
21 crease with ambient temperature.

22 In addition, source apportionment analyses conducted on winter and summer da-
23 ta using positive matrix factorization (PMF) yield semi-volatile oxygenated organic
24 aerosol (SV-OOA) factors that retain source-related chemical information. Winter
25 SV-OOA is highly influenced by biomass burning, whereas summer SV-OOA is to a
26 high degree produced from biogenic precursor gases. These sources contribute to
27 substantial differences between the winter and summer f_{44}/f_{43} data, suggesting

28 that PMF analysis of multi-season data employing only two OOA factors cannot
29 capture the seasonal variability of OOA.

30

31 **1. Introduction**

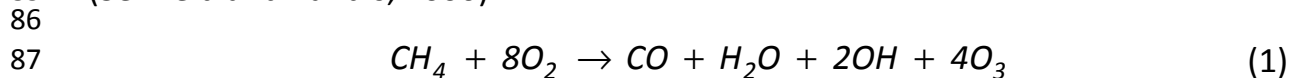
32 Atmospheric aerosols are at the center of scientific and political discussions due to
33 their highly uncertain direct and indirect climate effects (IPCC, 2007), their ad-
34 verse impacts on human health (Peng et al., 2005), and their influence on our in-
35 habited (Watson, 2002) and agricultural areas (Matson et al., 2002). Reliable iden-
36 tification and quantification of aerosol sources is essential for developing control
37 strategies. The concentrations of particulate matter have generally decreased in
38 the last 10-20 years in Europe and Switzerland but legal thresholds are still often
39 exceeded (Barnpadimos et al., 2011; Barnpadimos et al., 2012). Atmospheric
40 aerosols are classified based on their formation processes as primary and second-
41 ary aerosols, which are directly emitted into the atmosphere and formed from gas
42 to particle conversion, respectively. Recently, the scientific focus has shifted to-
43 wards submicron particulate matter (PM₁) (Hallquist et al., 2009), especially the
44 organic fraction, which typically comprises 20-90% of the total submicron aerosol
45 mass (Jimenez et al., 2009).

46 Aerodyne aerosol mass spectrometers (AMS), including the aerosol chemical spe-
47 ciation monitor (ACSM), have become important and widely employed instru-
48 ments for the chemical characterization of submicron organic aerosol
49 (Canagaratna et al., 2007; Ng et al., 2011b). These instruments provide on-line
50 quantitative mass spectra of the non-refractory (inorganic and organic) aerosol
51 composition with high time resolution. Frequently, the organic fraction is further
52 analyzed (Lanz et al., 2007; Ulbrich et al., 2009; Zhang et al., 2011) using the posi-
53 tive matrix factorization algorithm (PMF) proposed by Paatero and Tapper (1994),
54 which represents the organic mass spectral matrix as a set of source/process-
55 related factor mass spectra and time series. Compilation and comparison of
56 northern hemispheric datasets led to the characterization of secondary organic
57 aerosol (SOA)-related factors as semi-volatile and low volatility oxygenated organ-
58 ic aerosol (SV-OOA and LV-OOA) (Jimenez et al., 2009; Ng et al., 2010). The SV-
59 OOA mass spectra have a higher fraction of m/z 43 to organic mass (f_{43}) and a

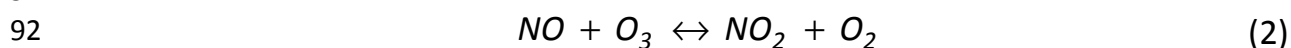
60 lower fraction of m/z 44 to organic mass (f_{44}) relative to LV-OOA. SV-OOA and LV-
61 OOA factors derived from ambient PMF analyses yield a triangle in the f_{44}/f_{43}
62 space (Ng et al., 2010). SV-OOA usually represents freshly formed OOA, whereas
63 LV-OOA may result from photochemical aging of SV-OOA, direct gas-phase to LV-
64 OOA conversion (Ehn et al., 2014), or aqueous-phase chemistry, all of which lead
65 to a net increase of OOA f_{44} with atmospheric age (Hallquist et al., 2009).

66 Evidence from several smog chamber and ambient studies suggests that the loca-
67 tion in the f_{44}/f_{43} space could carry information on the source of SV-OOA. The
68 generation of SOA from smog chamber experiments indicates that for a given f_{44} ,
69 SOA formed from wood burning experiments (Hennigan et al., 2011; Heringa et
70 al., 2011) yields lower f_{43} than from biogenic precursors (Alfarra et al., 2012;
71 Chhabra et al., 2011; Ng et al., 2010; Pfaffenberger et al., 2013). Recently, some
72 ambient studies also showed that the f_{43} and f_{44} points lie in specific regions in
73 the triangular space depending on the season (Crippa et al., 2014; Freney et al.,
74 2011; Freney et al., 2014; Ge et al., 2012). However, the ambient studies showing
75 the raw data points consider the total f_{43} and f_{44} fraction rather than the model-
76 derived OOA. Thus, the position of these points is affected by the contribution of
77 the primary sources, whereas in Ng et al. (2010) the triangular space referred to
78 the modelled OOA factors, i.e. SV-OOA and LV-OOA only.

79 Photochemical oxidation constitutes a major production pathway for OOA. The
80 tropospheric ozone concentration is mainly generated from the oxidation of vola-
81 tile organic compounds (VOCs) initiated by the OH radical and is thus a useful indi-
82 cator for photochemical activity. The net oxidation of the simplest VOC, methane
83 (CH_4) with oxygen (O_2) leading to carbon monoxide (CO), water (H_2O), hydroxyl
84 radical (OH) and ozone (O_3), is represented in the presence of NO_x as follows
85 (Seinfeld and Pandis, 2006):



87
88
89 The ozone molecules produced by these oxidation reactions participate in a rapid
90 equilibrium between NO and NO_2 , involving the photolysis of NO_2 , summarized by:



92
93

94 The sum of O₃ and NO₂ concentrations, defined as the oxidant OX (Alghamdi et al.,
95 2014; Clapp and Jenkin, 2001; Jenkin, 2014), is therefore a proxy for atmospheric
96 aging in the troposphere.

97 Lagrangian studies investigating the evolution of air parcels moving downwind
98 from a city have shown an increase of the LV-OOA to SV-OOA ratio (implying an
99 increase in total OOA *f*₄₄) as a function of the distance from the city center
100 (Jimenez et al., 2009). However, field campaigns and monitoring networks typical-
101 ly rely on stationary measurements, representing Eulerian studies instead, for
102 which the relationship between aging (i.e. OX) and *f*₄₄ is not strictly given.

103 In this study, ACSM aerosol mass spectra measured in Zurich between February
104 2011 and February 2012 were analyzed using the PMF algorithm in the multilinear
105 engine (ME-2) implementation (Paatero, 1999). The transformation of OOA during
106 the aging processes in summer is related to OX and temperature. Moreover, the
107 OOA composition is characterized in terms of *f*₄₄ and *f*₄₃, investigating the extent
108 to which precursor sources can be inferred from these values.

109 **2. Materials and Methods**

110 **2.1. Measurements**

111 The instruments and the methods employed for this study were described in de-
112 tail by Canonaco et al. (2013), and only a brief overview is presented here. An
113 ACSM (Aerodyne Research, Inc., Billerica, MA, USA) was deployed at the Kaserne
114 station, an urban background station in the city center of Zurich (Switzerland),
115 from February 2011 to February 2012. The ACSM is a compact aerosol mass spec-
116 trometer designed for long-term measurements of non-refractory particulate
117 matter with vacuum aerodynamic diameters between approximately 60 and 600
118 nm, typically denoted as NR-PM1. The instrument is described in detail by Ng et al.
119 (2011b). For a detailed description of AMS operation and analysis principles, the
120 reader is referred to Jayne et al. (2000), Jimenez et al. (2003), Allan et al. (2003),
121 Allan et al. (2004), and Canagaratna et al. (2007).

122 The meteorological data and trace gases were measured with conventional in-
123 struments by the Swiss National Air Pollution Monitoring Network, NABEL (Empa,
124 2011). The time resolution of all these instruments was ten minutes. NO_x was
125 measured by chemiluminescence spectroscopy (Horiba APNA 360), whereas UV

126 absorption was employed to measure the concentration of ozone (Thermo Envi-
127 ronmental Instruments (TEI) 49C, Thermo Electron Corp., Waltham, MA). An ae-
128 thalometer (AE 31, Magee Scientific Inc.) was utilized to retrieve the concentration
129 of equivalent black carbon (EBC).

130 **2.2. The multilinear engine (ME-2)**

131 ME-2 (Paatero, 1999) is an engine for solving the positive matrix factorization al-
132 gorithm (Paatero and Tapper, 1994) where a measured matrix \mathbf{X} is deconvolved
133 into two matrices \mathbf{G} and \mathbf{F} and the remaining residual matrix \mathbf{E} :

$$134 \quad \mathbf{X} = \mathbf{GF} + \mathbf{E} \quad (3)$$

136
137 In the measured matrix \mathbf{X} , the columns j are the m/z 's and each row i represents a
138 single mass spectrum per unit time. Note that p is defined as the number of fac-
139 tors of the chosen model solution, i.e. the number of columns of \mathbf{G} and the num-
140 ber of rows of \mathbf{F} . Each column of the matrix \mathbf{G} represents the time series of a fac-
141 tor, whereas each row of \mathbf{F} represents the factor profile (mass spectrum).

142 In PMF, the entries in \mathbf{G} and \mathbf{F} are fit using a least squares algorithm that minimiz-
143 es iteratively the quantity Q^m , defined as:

$$144 \quad Q^m = \sum_{i=1}^m \sum_{j=1}^n \left(\frac{e_{ij}}{\sigma_{ij}} \right)^2 \quad (4)$$

146
147 Here e_{ij} are the elements of the residual matrix \mathbf{E} and σ_{ij} are the measurement un-
148 certainties for the input points x_{ij} .

149 It is well known that PMF solutions suffer from rotational ambiguity (Paatero et
150 al., 2002), i.e. multiple combinations of \mathbf{G} and \mathbf{F} can be found that yield similar Q^m .
151 Thus, the solution space needs to be explored in order to find the most environ-
152 mentally reasonable and interpretable solution according to the recommenda-
153 tions discussed in Ulbrich et al. (2009), Canonaco et al. (2013) and Crippa et al.
154 (2014).

155 In this study, rotations are explored using the α -value approach, which was first
156 introduced by Lanz et al. (2008) for AMS data, employed for ACSM data in Canon-
157 aco et al. (2013) and systematically tested on 25 AMS data sets in Crippa et al.
158 (2014). Within this method, the user directs the algorithm towards useful rota-

159 tions by constraining factor profiles (as done here) and / or factor time series
160 based on a priori information (Paatero and Hopke (2009)).

161 Briefly, the α -value determines the extent to which a given factor profile ($f_{j,solution}$)
162 is allowed to vary with respect to its predefined profile value (f_j) during the model
163 iteration:

$$f_{j,solution} = f_j \pm \alpha \cdot f_j \quad (5)$$

164
165
166
167 where the index j denotes a measured variable (i.e. m/z) and the α -value is its sca-
168 lar product. As an example, an α -value of 0.1 allows for a variability of approxi-
169 mately $\pm 10\%$.

170 Generally, primary aerosol components are assumed to be unaffected by meteor-
171 ological and chemical aging processes, since they represent fresh emissions. This
172 assumption is empirically supported by the similarities in POA profiles retrieved in
173 PMF analyses at many sites, e.g. in Ng et al. (2011a). This allows the model to be
174 constrained using POA factor profiles from other PMF studies and allowing da-
175 taset-specific optimizations with the ME-2 solver using the α -value technique. In
176 contrast to POA, OOA profiles (i.e. SOA composition) depend on many parame-
177 ters, e.g. precursor gases, aging processes and atmospheric conditions. To appro-
178 priately constrain an OOA profile, the effect of these considerations on the spec-
179 tral profiles must be known, but this is not currently possible. Therefore, OOA fac-
180 tors are not constrained but rather allowed to be modeled by PMF.

181 For this study we conducted the source apportionment employing the ME-2 solver
182 and constraining the primary aerosol components, i.e. traffic (HOA), cooking (COA)
183 and biomass burning (BBOA) using the α -value approach and allowed for two ad-
184 ditional free factors representing the secondary components. HOA and COA an-
185 chor profiles were taken from Crippa et al. (2013), where these primary sources
186 were successfully resolved in an unconstrained PMF run. The BBOA anchor is the
187 averaged BBOA mass spectrum reported in Ng et al. (2011a). The employed α -
188 values are 0.1 for HOA and COA and 0.3 for BBOA. This was based on different
189 sensitivity tests performed on the winter and summer data set separately, similar
190 to those presented in Canonaco et al. (2013). The higher α -value for BBOA ac-
191 counts for the fact that the biomass burning sources are more variable, as they
192 depend strongly on the burning material and burning conditions (Hennigan et al.,
193 2011; Heringa et al., 2012; Weimer et al., 2008).

194 **2.3. Estimating OOA f_{43} and f_{44}**

195 The approach used in this study for estimating f_{43} and f_{44} for OOA involved the
196 subtraction of the contributions from the primary sources arising at m/z 43 and
197 m/z 44. This is slightly different from the method of Ng et al. (2010), where only
198 the modeled f_{43} and f_{44} for SV-OOA and LV-OOA were considered. The two
199 methods produce slightly different results because specific sources/processes are
200 not perfectly represented by individual factors, as evidenced by the residual ma-
201 trix **E**. In the current method, the variability not captured by the model (i.e. resid-
202 uals), propagates into the calculated f_{43} and f_{44} . This means that the total OOA
203 variability is more fully captured by the current method, but at the cost of unin-
204 tentionally including variability due to imperfectly modeled POA.

205 **3. Results and Discussion**

206 **3.1. Source apportionment in winter and summer 2011**

207 Source apportionment was conducted separately for winter (February and March)
208 and summer (June to August) 2011. These two seasons represent extreme cases,
209 whereas spring and fall may be conceptualized as intermediate cases. A complete
210 source apportionment analysis of the ACSM data of 2011 / 2012 is in preparation
211 (Canonaco et al., in prep.). The summer / winter 2011 results are summarized in
212 the supplementary material. Note that the solutions are environmentally reason-
213 able, e.g. the traffic factor correlates with NO_x and EBC, the cooking factor peaks
214 during mealtimes (noon and evening), BBOA is higher at night and lower during
215 the day accounting for nocturnal heating in winter and barbecuing and possible
216 local fire events in summer, the daily cycle of SV-OOA is anticorrelated with tem-
217 perature for the winter and summer data (Fig. S3 and S6). In addition, the daily cy-
218 cle of LV-OOA is rather flat (winter) or shows an increase during the afternoon
219 (summer), representing either the conversion of SV-OOA to LV-OOA or direct LV-
220 OOA formation from the gas-phase and thus compensating the effect of boundary
221 layer dilution and / or advecting air masses containing background LV-OOA.

222 Figures S3 and S6 reveal an increase in concentrations for all factors except LV-
223 OOA during the late evening and at night in winter and summer. This is mostly
224 governed by a smaller boundary layer in the evening compared to midday which
225 tends to concentrate all emissions.

226 The OOA f_{44}/f_{43} data together with the ratios of the SV-OOA and LV-OOA in the
227 f_{44}/f_{43} space are summarized for both seasons in Fig. 1. This figure highlights the
228 fact that the majority of the OOA points, especially those with high masses (data
229 with weak signal to noise is not expected to be explained by the model), are well-
230 captured by the seasonal PMF run, since they are equally scattered (unimodal re-
231 sidual) around the connection line between SV-OOA and LV-OOA and hardly ex-
232 tend beyond these endpoints. Hence no major systematic over- or underestima-
233 tions occur. Visual inspection of Fig. 1 indicates that the SOA points do not form a
234 cloud, but rather suggest a line. This indicates that a single OOA factor cannot ad-
235 equately represent SOA spectral variability in both seasons and at least two OOA
236 factors with consistent characteristics of SV-OOA and LV-OOA are required. Note
237 that the different location (and apparent slope) in the f_{44}/f_{43} space of the winter
238 and summer data indicates that a combined winter/summer PMF would fail to
239 capture the seasonal variability in OOA. The winter SV-OOA lies more on the left
240 side of the triangular space, whereas the summer SV-OOA is on the right side of
241 the triangular space. These locations are comparable with the location of SOA
242 from smog chamber studies conducted with biomass burning (Heringa et al.,
243 2011) and α -pinene (Pfaffenberger et al., 2013). The data from these two studies
244 are represented with orange rectangles in Figs. 1a and 1b, respectively.

245 **3.2. Variations in the f_{44}/f_{43} space for winter 2011**

246 The variation of the winter 2011 OOA points in the f_{44}/f_{43} space shown in Fig. 1a
247 is due to the linear combination between the winter LV-OOA and the biomass
248 burning-related SV-OOA. Adopting the nomenclature convention proposed in
249 Murphy et al. (2014) the winter SV-OOA, called SV-bbSOA (biomass burning SV-
250 OOA), would be due to aging of biomass burning-related VOC's emitted primarily
251 by domestic heating, which peaks at night. Fig. S1 supports the interpretation of
252 the winter SV-OOA as originating mainly from biomass burning emissions, due to
253 the presence of m/z 60, the biomass burning tracer (Alfarra et al., 2007) that has
254 been shown to be substantial in SV-OOA (Heringa et al., 2011). SV-OOA arising
255 from traffic emissions is likely to be a minor contribution, as the total estimated
256 contribution of the traffic source (combined POA and SOA) for Zurich winter
257 2011/2012 is on average less than 20% to the total OA (Zotter et al., 2014). In

258 comparison, the PMF result in this study ascribed on average 12% of OA to prima-
259 ry traffic contributions, 25% to SV-OOA and 40% to LV-OOA. Even if the remaining
260 traffic SOA contribution were completely assumed to be SV-OOA, this would still
261 be a minor part of the total SV-OOA.

262 SV-OOA has a lower f_{44} compared to the SV-OOA in summer (see Sect. 3.3).
263 Nonetheless, the f_{44} of LV-OOA in winter is higher than that of LV-OOA in summer
264 despite reaching similarly low f_{43} values. A possible explanation could be aque-
265 ous-phase production of LV-OOA (either directly or via processing of SV-OOA) in
266 clouds or humidified aerosols, which are believed to increase oxygenation above
267 that predicted by gas-phase reaction/condensation mechanisms (Ervens et al.,
268 2011; George et al., 2008; Slowik et al., 2012). However, due to the lack of exper-
269 imental data and ambient tracers for such mechanisms, this hypothesis remains
270 speculative.

271 **3.3. Variations in the f_{44}/f_{43} space for the summer 2011**

272 **3.3.1. General trends**

273 The variation of the summer 2011 OOA data in the f_{44}/f_{43} space shown in Fig. 1
274 can be described as a linear combination of the summer LV-OOA and SV-OOA (see
275 Sect. 3.2). The relation between temperature and OX for the measured data is
276 shown in Fig. 2a. Note that OX is plotted as a function of the maximal daily tem-
277 perature ± 2 hours to capture the period of highest photochemical activity. This
278 strong relation implies that the photochemical oxidation is highest for days with
279 high temperature. However, the relation between f_{44} and OX is rather flat if not
280 slightly inversely proportional as highlighted in Fig. 2b. On the contrary, f_{43} shows
281 a proportional dependence on the afternoon OX values (Fig. 2c). This is consistent
282 with increased production of SV-OOA relative to LV-OOA, i.e. with increased VOC
283 precursors in the atmosphere. Given the season and elevated temperatures, bio-
284 genic emissions are a likely source (Guenther, 1997). Fig. 3a illustrates the clus-
285 tered afternoon data on top of the summer OOA data in the f_{44}/f_{43} space. Only
286 the values in the interval of four hours around the maximal daily temperature
287 (T_{max}) were considered for the clustering. This figure elucidates the fact that the
288 main horizontal movement of the f_{44}/f_{43} OOA data is driven mainly by tempera-
289 ture and thus by biogenic emissions. Therefore, the modeled SV-OOA is predomi-

290 nantly of biogenic nature and has been referred to as SV-bSOA (biogenic SOA) by
291 Murphy et al. (2014).

292 **3.3.2. Day and night variations**

293 Fig. 3b shows the OOA data together with two grouped families. The red points
294 are the same as in Fig. 3a, i.e. the afternoon values only. The blue points are the
295 clustered points between 0 to 5 AM of the following morning. Only early morning
296 points are considered to avoid dilution effects from a rising boundary layer after
297 sunrise. The comparison of the two grouped families (afternoon and subsequent
298 early morning) suggests that semi-volatile organics generated during the day con-
299 dense to the aerosol phase at night, increasing f_{43} and decreasing f_{44} . This effect
300 was already described Lanz et al. (2007) showing that the condensation of fresh
301 oxygenated organic compounds (SV-OOA) was enhanced during the night and the
302 early morning following hot summer days during a three-week campaign with the
303 AMS in Zurich. The fact that the ACSM data from the entire summer season in Zur-
304 ich in 2011 shows the same temperature-driven partitioning for SV-OOA reinforc-
305 es the interpretation of the semi-volatile character of OOA2 from Lanz et al.
306 (2007). As discussed in the previous section, the summer SV-OOA is likely gov-
307 erned by biogenic SV-OOA (SV-bSOA). Moreover, oxidation processes enhancing
308 the LV-OOA fraction during the afternoon, as highlighted by the diurnal cycle in
309 Figure S6, will increase the diurnal f_{44} / f_{43} ratio leading to a stronger separation
310 of the their day versus night points. Only a single pair of points in Fig. 3b, at the
311 lowest temperature (and which has high statistical uncertainty due to the small
312 number of measurements it contains) violates this trend.

313 **3.3.3. SV-OOA vs LV-OOA**

314 Figures 4a and 4b show summer SV-OOA and LV-OOA against the maximum after-
315 noon temperature as calculated above (red bars) and the data between midnight
316 and 5 AM of the following morning (blue bars). The SV-OOA concentration tends
317 to increase as a function of temperature, both during the afternoon and early
318 morning. LV-OOA shows a similar but less pronounced behavior. In Fig. 4c, the ra-
319 tio (SV-OOA / total OOA) is plotted against the total OOA mass concentration. For
320 low OOA concentrations, the fraction of SV-OOA increases with increasing total
321 OOA. However, the fraction levels off for total OOA mass concentrations above 5

322 $\mu\text{g}\cdot\text{m}^{-3}$. One possible explanation could involve the departure of the exponential
323 dependency of the biogenic VOC's emission rate from the temperature, which oc-
324 curs at temperatures between 30 and 35°C (Smiatek and Steinbrecher, 2006).
325 Besides the positive trend of SV-OOA with respect to the summer afternoon tem-
326 perature, the SV-OOA fraction increases as a function of total organic aerosol
327 mass (OA), highlighted in Fig. 4c and Fig. S7. Recently, Pfaffenberger et al. (2013)
328 showed that the partitioning of biogenic semi-volatile organic compounds (SV-
329 OOA) to the aerosol phase is enhanced for increased aerosol mass concentrations,
330 resulting in a decrease of f_{44} and an increase in f_{43} . Fig. S7 shows f_{44} and f_{43}
331 plotted as a function of total OA mass for the summer data together with the ex-
332 perimental data (Pfaffenberger, personal communication). Our results in the high-
333 er mass range (above $5 \mu\text{g}\cdot\text{m}^{-3}$) suggest a similar behavior for f_{43} , though less pro-
334 nounced for f_{44} . One reasonable explanation could involve a slightly higher total
335 f_{44} value for the ambient data at low concentrations, due to higher photochemi-
336 cal aging. In addition, other reaction pathways, e.g. nitrate oxidation, or aqueous
337 phase reactions might also affect the ambient composition and finally, the pres-
338 ence of other ambient VOCs than those tested in the above-mentioned smog
339 chamber study, would also lead to a different slope of the ambient data.

340 **3.3.4. SOA formation in summer**

341 The main ambient emission and photochemical oxidation processes are summa-
342 rized in Fig. 5. The four sources relevant for this study are represented at the bot-
343 tom, i.e. biogenic, traffic, cooking and biomass burning. These sources emit VOCs
344 which are transformed to SV-OOA and further to LV-OOA or directly to LV-OOA
345 (orange arrows in the figure). Due to a substantial amount of VOCs and SV-OOA in
346 summer, the conversion rate generating LV-OOA from either SV-OOA or directly
347 from VOC's, typically occurring on a timescale of hours (Jimenez et al., 2009), is
348 rather small. The concentration of OX in the atmosphere is related to the net ag-
349 ing processes and is therefore linked to the conversion from VOC to SV-OOA
350 and/or LV-OOA and from SV-OOA to LV-OOA.

351 The strong relation between OX and temperature highlighted in Fig. 2a suggests
352 that photochemistry is more active during the summer afternoon. However, Fig.
353 2b indicates that the SOA f_{44} is rather flat with increasing OX in favor of the total

354 OOA f_{43} (SV-OOA), as also shown in Fig. 2c. Higher temperatures will enhance the
355 biogenic emissions relative to the other emission rates. As a consequence the bio-
356 genic path will dominate and the resulting SV-OOA will most likely be predomi-
357 nantly of biogenic nature, i.e. SV-OOA, represented as the orange path in Fig. 5.

358 **4. Conclusions**

359 This study shows that the SV-OOA modeled for ambient data by the means of the
360 multilinear engine (ME-2) in winter and in summer retains some chemical infor-
361 mation related to its precursor source(s). For a given f_{44} , biomass burning-related
362 SV-OOA exhibits lower f_{43} relative to biogenic SV-OOA (SV-OOA), locating these
363 two SOA factors on the left and right-hand sides of the triangular space identified
364 by Ng et al. (2010), respectively.

365 Periods of high photochemical activity in summer do not increase the SOA f_{44}/f_{43}
366 ratio because temperature-driven biogenic emissions and subsequent SV-OOA
367 formation dominate over the conversion rate of SV-OOA to LV-OOA or direct for-
368 mation of LV-OOA. The f_{44}/f_{43} ratio is consistently lower at night than during the
369 previous day due to the condensation of semi-volatile compounds produced dur-
370 ing the day, predominantly from reaction of biogenic VOC's.

371 In summer, the OOA composition depends strongly on temperature and mass
372 concentration for values below $5 \mu\text{g}\cdot\text{m}^{-3}$. This highlights the importance of biogen-
373 ic VOC emissions and of the biogenic SOA production.

374 In addition, there were substantial differences between the winter and summer
375 f_{44}/f_{43} data indicating that a PMF result over the whole data employing two OOA
376 factors only would fail to fully represent the seasonal variability of OOA.

377 **Acknowledgements**

378 The ACSM measurements were supported by the Swiss Federal Office for the Envi-
379 ronment (FOEN). The authors would like to thank L. Pfaffenberger and M. Heringa
380 for compiling and providing their smog chamber results in the f_{44}/f_{43} space, rele-
381 vant for this publication. Thanks are also due to the Environmental group of the
382 Swiss Federal Laboratories for Materials and Testing (EMPA) for their support and
383 to M. Canonaco and S. Canonaco-Franceschini for a critical reading of this manu-
384 script. J. Slowik acknowledges support from the Swiss National Science Foundation

385 (SNF) in the form of Ambizione and Starting grants (PZ00P2_131673 and
386 BSSGIO_155846).

387 **Literature**

388 Alfarra, M. R., Hamilton, J. F., Wyche, K. P., Good, N., Ward, M. W., Carr, T., Barley,
389 M. H., Monks, P. S., Jenkin, M. E., Lewis, A. C., and McFiggans, G. B.: The effect of
390 photochemical ageing and initial precursor concentration on the composition and
391 hygroscopic properties of beta-caryophyllene secondary organic aerosol, *Atmos.*
392 *Chem. Phys.*, 12, 6417-6436, 2012.

393 Alfarra, M. R., Prévôt, A. S. H., Szidat, S., Sandradewi, J., Weimer, S., Lanz, V. A.,
394 Schreiber, D., Mohr, M., and Baltensperger, U.: Identification of the mass spectral
395 signature of organic aerosols from wood burning emissions, *Environ. Sci. Technol.*,
396 41, 5770-5777, 2007.

397 Alghamdi, M. A., Khoder, M., Harrison, R. M., Hyvarinen, A. P., Hussein, T., Al-
398 Jeelani, H., Abdelmaksoud, A. S., Goknil, M. H., Shabbaj, I. I., Almeahadi, F. M.,
399 Lihavainen, H., Kulmala, M., and Hameri, K.: Temporal variations of O₃ and NO_x in
400 the urban background atmosphere of the coastal city Jeddah, Saudi Arabia,
401 *Atmos. Environ.*, 94, 205-214, 2014.

402 Allan, J. D., Delia, A. E., Coe, H., Bower, K. N., Alfarra, M. R., Jimenez, J. L.,
403 Middlebrook, A. M., Drewnick, F., Onasch, T. B., Canagaratna, M. R., Jayne, J. T.,
404 and Worsnop, D. R.: A generalised method for the extraction of chemically
405 resolved mass spectra from Aerodyne aerosol mass spectrometer data, *J. Aerosol*
406 *Sci.*, 35, 909-922, 2004.

407 Allan, J. D., Jimenez, J. L., Williams, P. I., Alfarra, M. R., Bower, K. N., Jayne, J. T.,
408 Coe, H., and Worsnop, D. R.: Quantitative sampling using an Aerodyne aerosol
409 mass spectrometer: 1. Techniques of data interpretation and error analysis, *J.*
410 *Geophys. Res.-Atmos.*, 108, 4090, 2003.

411 Barmpadimos, I., Hueglin, C., Keller, J., Henne, S., and Prevot, A. S. H.: Influence of
412 meteorology on PM₁₀ trends and variability in Switzerland from 1991 to 2008,
413 *Atmos. Chem. Phys.*, 11, 1813-1835, 2011.

414 Barmpadimos, I., Keller, J., Oderbolz, D., Hueglin, C., and Prevot, A. S. H.: One
415 decade of parallel fine (PM_{2.5}) and coarse (PM₁₀-PM_{2.5}) particulate matter

416 measurements in Europe: trends and variability, *Atmos. Chem. Phys.*, 12, 3189-
417 3203, 2012.

418 Canagaratna, M. R., Jayne, J. T., Jimenez, J. L., Allan, J. D., Alfarra, M. R., Zhang, Q.,
419 Onasch, T. B., Drewnick, F., Coe, H., Middlebrook, A., Delia, A., Williams, L. R.,
420 Trimborn, A. M., Northway, M. J., DeCarlo, P. F., Kolb, C. E., Davidovits, P., and
421 Worsnop, D. R.: Chemical and microphysical characterization of ambient aerosols
422 with the Aerodyne aerosol mass spectrometer, *Mass Spectrom. Rev.*, 26, 185-222,
423 2007.

424 Canonaco, F., Crippa, M., Slowik, J., Baltensperger, U., and Prévôt, A. S. H.: SoFi, an
425 IGOR-based interface for the efficient use of the generalized multilinear engine
426 (ME-2) for the source apportionment: ME-2 application to aerosol mass
427 spectrometer data, *Atmos. Meas. Tech.*, 2013.

428 Canonaco, F., Dällenbach, K., ElHaddad, I., Crippa, M., Bozzetti, C., Huang, R.-J.,
429 Slowik, J., Baltensperger, U., Hüglin, C., Herich, H., and Prévôt, A. S. H.: A novel
430 strategy for the source apportionment of long-term ACSM data based on ME-2
431 with SoFi: Automatic Rolling SoFi (AuRo-SoFi), in prep.

432 Chhabra, P. S., Ng, N. L., Canagaratna, M. R., Corrigan, A. L., Russell, L. M.,
433 Worsnop, D. R., Flagan, R. C., and Seinfeld, J. H.: Elemental composition and
434 oxidation of chamber organic aerosol, *Atmos. Chem. Phys.*, 11, 8827-8845, 2011.

435 Clapp, L. J. and Jenkin, M. E.: Analysis of the relationship between ambient levels
436 Of O₃, NO₂ and NO as a function of NO_x in the UK, *Atmos. Environ.*, 35, 6391-
437 6405, 2001.

438 Crippa, M., Canonaco, F., Lanz, V. A., Äijälä, M., Allan, J. D., Carbone, S., Capes, G.,
439 Ceburnis, D., M., D. O., Day, D. A., DeCarlo, P. F., Ehn, M., Eriksson, A., Freney, E.,
440 Hildebrandt Ruiz, L., Hillamo, R., Jimenez, J. L., Junninen, H., Kiendler-Scharr, A.,
441 Kortelainen, A.-M., Kulmala, M., Laaksonen, A., Mensah, A. A., Mohr, C., Nemitz,
442 E., O'Dowd, C., Ovadnevaite, J., Pandis, S. N., Petäjä, T., Poulain, L., Saarikoski, S.,
443 Sellegri, K., Swietlicki, E., Tiitta, P., Worsnop, D. R., Baltensperger, U., and Prevot,
444 A. S. H.: Organic aerosol components derived from 25 AMS data sets across
445 Europe using a consistent ME-2 based source apportionment approach, *Atmos.*
446 *Chem. Phys.*, 14, 6159–6176, 2014.

447 Crippa, M., DeCarlo, P. F., Slowik, J. G., Mohr, C., Heringa, M. F., Chirico, R.,
448 Poulain, L., Freutel, F., Sciare, J., Cozic, J., Marco, C. F. D., Elsasser, M., José, N.,

- 449 Marchand, N., Abidi, E., Wiedensohler, A., Drewnick, F., Schneider, J., Borrmann,
450 S., Nemitz, E., Zimmermann, R., Jaffrezo, J.-L., Prévôt, A. S. H., and Baltensperger,
451 U.: Wintertime aerosol chemical composition and source apportionment of the
452 organic fraction in the metropolitan area of Paris, *Atmos. Chem. Phys.*, 13, 961-
453 981, 2013.
- 454 Ehn, M., Thornton, J. A., Kleist, E., Sipila, M., Junninen, H., Pullinen, I., Springer, M.,
455 Rubach, F., Tillmann, R., Lee, B., Lopez-Hilfiker, F., Andres, S., Acir, I.-H., Rissanen,
456 M., Jokinen, T., Schobesberger, S., Kangasluoma, J., Kontkanen, J., Nieminen, T.,
457 Kurten, T., Nielsen, L. B., Jorgensen, S., Kjaergaard, H. G., Canagaratna, M., Dal
458 Maso, M., Berndt, T., Petaja, T., Wahner, A., Kerminen, V.-M., Kulmala, M.,
459 Worsnop, D. R., Wildt, J., and Mentel, T. F.: A large source of low-volatility
460 secondary organic aerosol, *Nature*, 506, 476-+, 2014.
- 461 Empa: Technischer Bericht zum Nationalen Beobachtungsnetz für Luftfremdstoffe,
462 (NABEL), available at <http://www.empa.ch>, 2011.
- 463 Ervens, B., Turpin, B. J., and Weber, R. J.: Secondary organic aerosol formation in
464 cloud droplets and aqueous particles (aqSOA): a review of laboratory, field and
465 model studies, *Atmos. Chem. Phys.*, 11, 11069-11102, 2011.
- 466 Freney, E. J., Sellegri, K., Canonaco, F., Boulon, J., Hervo, M., Weigel, R., Pichon, J.
467 M., Colomb, A., Prévôt, A. S. H., and Laj, P.: Seasonal variations in aerosol particle
468 composition at the puy-de-Dome research station in France, *Atmos. Chem. Phys.*,
469 11, 13047-13059, 2011.
- 470 Freney, E. J., Sellegri, K., Canonaco, F., Colomb, A., Borbon, A., Michoud, V.,
471 Doussin, J. F., Crumeyrolle, S., Amarouche, N., Pichon, J. M., Bourianne, T., Gomes,
472 L., Prévôt, A. S. H., Beekmann, M., and Schwarzenboeck, A.: Characterizing the
473 impact of urban emissions on regional aerosol particles: airborne measurements
474 during the MEGAPOLI experiment, *Atmos. Chem. Phys.*, 14, 1397-1412, 2014.
- 475 Ge, X. L., Setyan, A., Sun, Y. L., and Zhang, Q.: Primary and secondary organic
476 aerosols in Fresno, California during wintertime: Results from high resolution
477 aerosol mass spectrometry, *J. Geophys. Res.-Atmos.*, 117, 2012.
- 478 George, I. J., Slowik, J., and Abbatt, J. P. D.: Chemical aging of ambient organic
479 aerosol from heterogeneous reaction with hydroxyl radicals, *Geophysical*
480 *Research Letters*, 35, 2008.

- 481 Guenther, A.: Seasonal and spatial variations in natural volatile organic compound
482 emissions, *Ecological Applications*, 7, 34-45, 1997.
- 483 Hallquist, M., Wenger, J. C., Baltensperger, U., Rudich, Y., Simpson, D., Claeys, M.,
484 Dommen, J., Donahue, N. M., George, C., Goldstein, A. H., Hamilton, J. F.,
485 Herrmann, H., Hoffmann, T., Iinuma, Y., Jang, M., Jenkin, M. E., Jimenez, J. L.,
486 Kiendler-Scharr, A., Maenhaut, W., McFiggans, G., Mentel, T. F., Monod, A.,
487 Prevot, A. S. H., Seinfeld, J. H., Surratt, J. D., Szmigielski, R., and Wildt, J.: The
488 formation, properties and impact of secondary organic aerosol: current and
489 emerging issues, *Atmos. Chem. Phys.*, 9, 5155-5236, 2009.
- 490 Hennigan, C. J., Miracolo, M. A., Engelhart, G. J., May, A. A., Presto, A. A., Lee, T.,
491 Sullivan, A. P., McMeeking, G. R., Coe, H., Wold, C. E., Hao, W. M., Gilman, J. B.,
492 Kuster, W. C., de Gouw, J., Schichtel, B. A., Collett, J. L., Jr., Kreidenweis, S. M., and
493 Robinson, A. L.: Chemical and physical transformations of organic aerosol from the
494 photo-oxidation of open biomass burning emissions in an environmental chamber,
495 *Atmos. Chem. Phys.*, 11, 7669-7686, 2011.
- 496 Heringa, M. F., DeCarlo, P. F., Chirico, R., Lauber, A., Doberer, A., Good, J.,
497 Nussbaumer, T., Keller, A., Burtscher, H., Richard, A., Miljevic, B., Prévôt, A. S. H.,
498 and Baltensperger, U.: Time-resolved characterization of primary emissions from
499 residential wood combustion appliances, *Environ. Sci. Technol.*, 46, 11418-11425,
500 2012.
- 501 Heringa, M. F., DeCarlo, P. F., Chirico, R., Tritscher, T., Dommen, J., Weingartner,
502 E., Richter, R., Wehrle, G., Prevot, A. S. H., and Baltensperger, U.: Investigations of
503 primary and secondary particulate matter of different wood combustion
504 appliances with a high-resolution time-of-flight aerosol mass spectrometer,
505 *Atmos. Chem. Phys.*, 11, 5945-5957, 2011.
- 506 IPCC: IPCC Fourth Assessment Report: The Physical Science Basis, Working Group
507 I, Final Report, Geneva, Switzerland., 2007.
- 508 Jayne, J. T., Leard, D. C., Zhang, X. F., Davidovits, P., Smith, K. A., Kolb, C. E., and
509 Worsnop, D. R.: Development of an aerosol mass spectrometer for size and
510 composition analysis of submicron particles, *Aerosol Sci. Tech.*, 33, 49-70, 2000.
- 511 Jenkin, M. E.: Investigation of an oxidant-based methodology for AOT40 exposure
512 assessment in the UK, *Atmos. Environ.*, 94, 332-340, 2014.

Paul Scherrer Institute (PSI) Switzerland

- 513 Jimenez, J. L., Canagaratna, M. R., Donahue, N. M., Prévôt, A. S. H., Zhang, Q.,
514 Kroll, J. H., DeCarlo, P. F., Allan, J. D., Coe, H., Ng, N. L., Aiken, A. C., Docherty, K. S.,
515 Ulbrich, I. M., Grieshop, A. P., Robinson, A. L., Duplissy, J., Smith, J. D., Wilson, K.
516 R., Lanz, V. A., Hueglin, C., Sun, Y. L., Tian, J., Laaksonen, A., Raatikainen, T.,
517 Rautiainen, J., Vaattovaara, P., Ehn, M., Kulmala, M., Tomlinson, J. M., Collins, D.
518 R., Cubison, M. J., Dunlea, E. J., Huffman, J. A., Onasch, T. B., Alfarra, M. R.,
519 Williams, P. I., Bower, K., Kondo, Y., Schneider, J., Drewnick, F., Borrmann, S.,
520 Weimer, S., Demerjian, K., Salcedo, D., Cottrell, L., Griffin, R., Takami, A., Miyoshi,
521 T., Hatakeyama, S., Shimono, A., Sun, J. Y., Zhang, Y. M., Dzepina, K., Kimmel, J. R.,
522 Sueper, D., Jayne, J. T., Herndon, S. C., Trimborn, A. M., Williams, L. R., Wood, E.
523 C., Middlebrook, A. M., Kolb, C. E., Baltensperger, U., and Worsnop, D. R.:
524 Evolution of organic aerosols in the atmosphere, *Science*, 326, 1525-1529, 2009.
- 525 Jimenez, J. L., Jayne, J. T., Shi, Q., Kolb, C. E., Worsnop, D. R., Yourshaw, I.,
526 Seinfeld, J. H., Flagan, R. C., Zhang, X. F., Smith, K. A., Morris, J. W., and Davidovits,
527 P.: Ambient aerosol sampling using the Aerodyne aerosol mass spectrometer,
528 *Geophys. Res. Atmos.*, 108, 8425, 2003.
- 529 Lanz, V. A., Alfarra, M. R., Baltensperger, U., Buchmann, B., Hueglin, C., and
530 Prévôt, A. S. H.: Source apportionment of submicron organic aerosols at an urban
531 site by factor analytical modelling of aerosol mass spectra, *Atmos. Chem. Phys.*, 7,
532 1503-1522, 2007.
- 533 Lanz, V. A., Alfarra, M. R., Baltensperger, U., Buchmann, B., Hueglin, C., Szidat, S.,
534 Wehrli, M. N., Wacker, L., Weimer, S., Caseiro, A., Puxbaum, H., and Prévôt, A. S.
535 H.: Source attribution of submicron organic aerosols during wintertime inversions
536 by advanced factor analysis of aerosol mass spectra, *Environ. Sci. Technol.*, 42,
537 214-220, 2008.
- 538 Matson, P., Lohse, K. A., and Hall, S. J.: The globalization of nitrogen deposition:
539 Consequences for terrestrial ecosystems, *Ambio*, 31, 113-119, 2002.
- 540 Murphy, B. N., Donahue, N. M., Robinson, A. L., and Pandis, S. N.: A naming
541 convention for atmospheric organic aerosol, *Atmos. Chem. Phys.*, 14, 5825-5839,
542 2014.
- 543 Ng, N. L., Canagaratna, M. R., Jimenez, J. L., Zhang, Q., Ulbrich, I. M., and Worsnop,
544 D. R.: Real-time methods for estimating organic component mass concentrations
545 from aerosol mass spectrometer data, *Environ. Sci. Technol.*, 45, 910-916, 2011a.

- 546 Ng, N. L., Canagaratna, M. R., Zhang, Q., Jimenez, J. L., Tian, J., Ulbrich, I. M., Kroll,
547 J. H., Docherty, K. S., Chhabra, P. S., Bahreini, R., Murphy, S. M., Seinfeld, J. H.,
548 Hildebrandt, L., Donahue, N. M., DeCarlo, P. F., Lanz, V. A., Prevot, A. S. H., Dinar,
549 E., Rudich, Y., and Worsnop, D. R.: Organic aerosol components observed in
550 northern hemispheric datasets from aerosol mass spectrometry, *Atmos. Chem.*
551 *Phys.*, 10, 4625-4641, 2010.
- 552 Ng, N. L., Herndon, S. C., Trimborn, A., Canagaratna, M. R., Croteau, P. L., Onasch,
553 T. B., Sueper, D., Worsnop, D. R., Zhang, Q., Sun, Y. L., and Jayne, J. T.: An aerosol
554 chemical speciation monitor (ACSM) for routine monitoring of the composition
555 and mass concentrations of ambient aerosol, *Aerosol Sci. Tech.*, 45, 770-784,
556 2011b.
- 557 Paatero, P.: The multilinear engine - A table-driven, least squares program for
558 solving multilinear problems, including the n-way parallel factor analysis model, *J.*
559 *Comput. Graph. Stat.*, 8, 854-888, 1999.
- 560 Paatero, P. and Hopke, P. K.: Rotational tools for factor analytic models, *J.*
561 *Chemometrics*, 23, 91-100, 2009.
- 562 Paatero, P., Hopke, P. K., Song, X. H., and Ramadan, Z.: Understanding and
563 controlling rotations in factor analytic models, *Chemometr. Intell. Lab.*, 60, 253-
564 264, 2002.
- 565 Paatero, P. and Tapper, U.: Positive matrix factorization - a nonnegative factor
566 model with optimal utilization of error-estimates of data values, *Environmetrics*, 5,
567 111-126, 1994.
- 568 Peng, R. D., Dominici, F., Pastor-Barriuso, R., Zeger, S. L., and Samet, J. M.:
569 Seasonal analyses of air pollution and mortality in 100 US cities, *Am. J. Epidemiol.*,
570 161, 585-594, 2005.
- 571 Pfaffenberger, L., Barmet, P., Slowik, J. G., Praplan, A. P., Dommen, J., Prevot, A. S.
572 H., and Baltensperger, U.: The link between organic aerosol mass loading and
573 degree of oxygenation: an alpha-pinene photooxidation study, *Atmos. Chem.*
574 *Phys.*, 13, 6493-6506, 2013.
- 575 Seinfeld, J. H. and Pandis, S. N.: *Atmospheric Chemistry And Physics, From Air*
576 *Pollution To Climate Change*, Second Edition ed., John Wiley & Sons, Hoboken.,
577 2006.

578 Slowik, J. G., Wong, J. P. S., and Abbatt, J. P. D.: Real-time, controlled OH-initiated
579 oxidation of biogenic secondary organic aerosol, *Atmospheric Chemistry and*
580 *Physics*, 12, 9775-9790, 2012.

581 Smiatek, G. and Steinbrecher, R.: Temporal and spatial variation of forest VOC
582 emissions in Germany in the decade 1994-2003, *Atmos. Environ.*, 40, S166-S177,
583 2006.

584 Ulbrich, I. M., Canagaratna, M. R., Zhang, Q., Worsnop, D. R., and Jimenez, J. L.:
585 Interpretation of organic components from positive matrix factorization of aerosol
586 mass spectrometric data, *Atmos. Chem. Phys.*, 9, 2891-2918, 2009.

587 Watson, J. G.: Visibility: Science and regulation, *Journal of the Air & Waste*
588 *Management Association*, 52, 628-713, 2002.

589 Weimer, S., Alfarra, M. R., Schreiber, D., Mohr, M., Prévôt, A. S. H., and
590 Baltensperger, U.: Organic aerosol mass spectral signatures from wood-burning
591 emissions: Influence of burning conditions and wood type, *J. Geophys. Res.-*
592 *Atmos.*, 113, 2008.

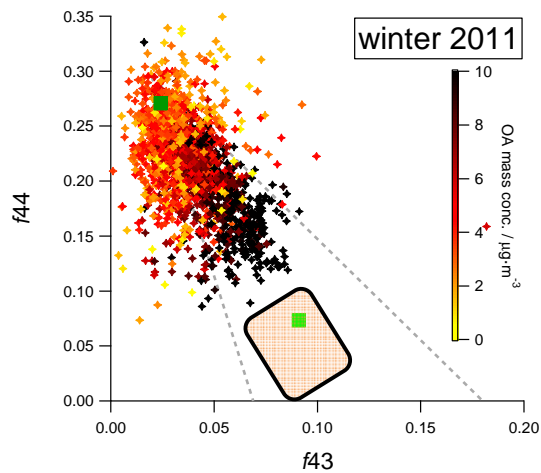
593 Zhang, Q., Jimenez, J. L., Canagaratna, M. R., Ulbrich, I. M., Ng, N. L., Worsnop, D.
594 R., and Sun, Y.: Understanding atmospheric organic aerosols via factor analysis of
595 aerosol mass spectrometry: a review, *Analytical and Bioanalytical Chemistry*, 401,
596 3045-3067, 2011.

597 Zotter, P., Ciobanu, V. G., Zhang, Y. L., El-Haddad, I., Macchia, M., Daellenbach, K.
598 R., Salazar, G. A., Huang, R. J., Wacker, L., Hueglin, C., Piazzalunga, A., Fermo, P.,
599 Schwikowski, M., Baltensperger, U., Szidat, S., and Prévôt, A. S. H.: Radiocarbon
600 analysis of elemental and organic carbon in Switzerland during winter-smog
601 episodes from 2008 to 2012 – Part 1: Source apportionment and spatial variability,
602 *Atmos. Chem. Phys.*, 14, 13551-13570, 2014.

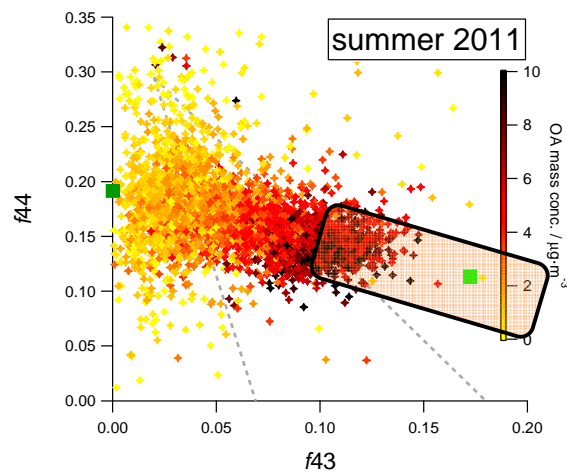
603

604

605 a)



b)

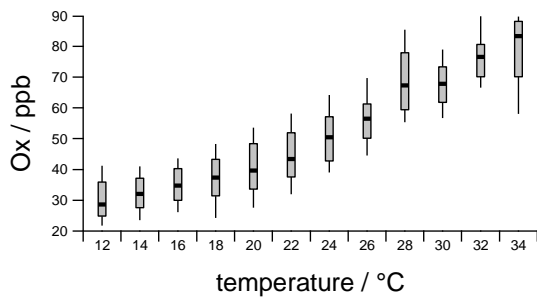


606

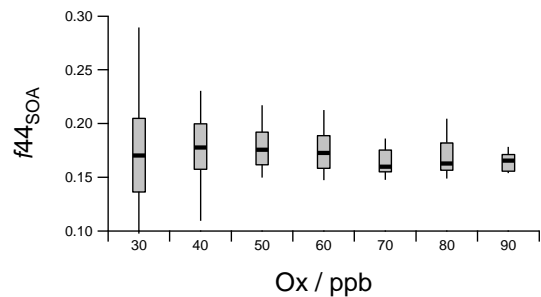
607 **Fig. 1** Estimated SOA f_{44} and f_{43} for winter 2011 (a) and summer 2011 (b).
608 The data is color-coded based on the total OA mass concentration.
609 The green points are the f_{44} and f_{43} ratios of SV-OOA (light green)
610 and LV-OOA (dark green). Orange rectangles represent the composition
611 of SOA from smog chamber experiments using biomass burning
612 (a) and α -pinene (b) precursors (Heringa et al. (2011), Pfaffenberger
613 et al. (2013)).

614

615 a)



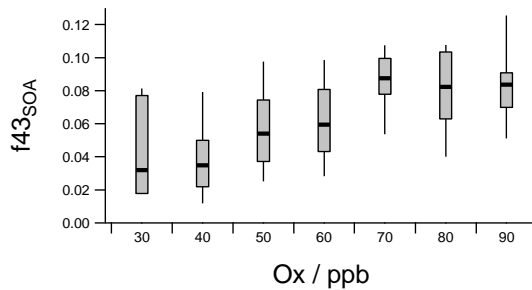
b)



616

617

c)



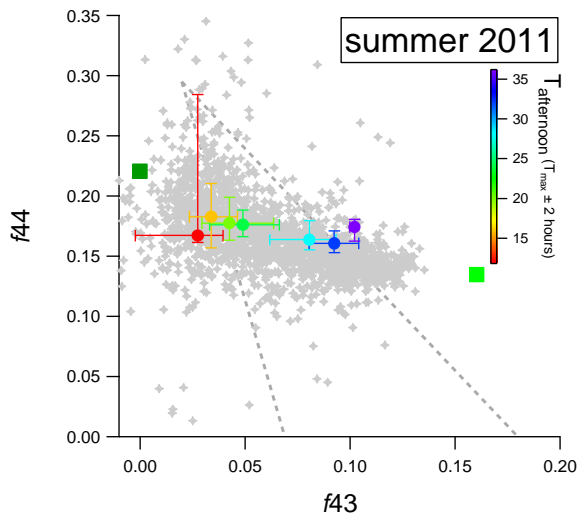
618

619

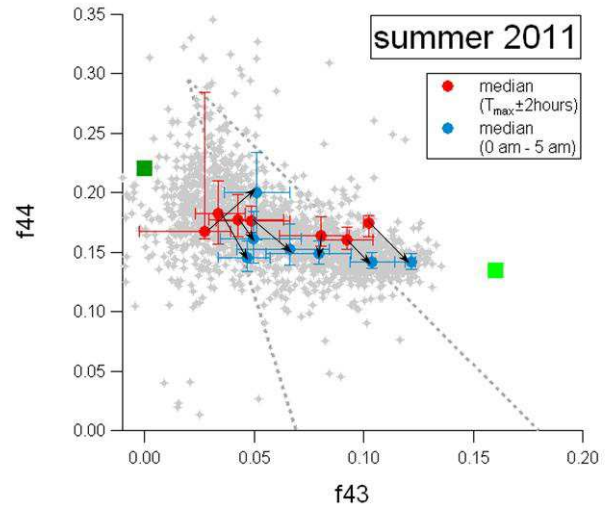
620 **Fig. 2** Box plots for the summer data: a) OX as a function of afternoon tem-
621 perature ($T_{\max} \pm 2$ hours), b) SOA f_{44} as a function of afternoon OX
622 and c) SOA f_{43} as a function of afternoon OX. The horizontal lines de-
623 note the median, the boxes span the quartiles and the whiskers rep-
624 resent the 10th and 90th percentiles.

625

626 a)



b)



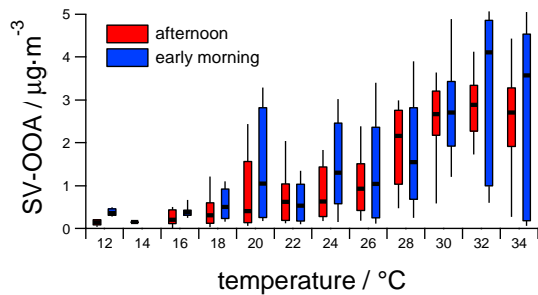
627

628 **Fig. 3** SOA f_{44} vs. f_{43} for all data points in summer 2011 (gray dots) and LV-
 629 OOA/SV-OOA factors (green squares). a) Color-coded circles denote
 630 averages at the daily maximum temperature (T_{\max}) \pm 2 hours. b) Red
 631 circles denote daily $T_{\max} \pm 2$ hours, while blue circles denote the average
 632 over the following midnight to 5 AM period. Black arrows connect
 633 corresponding day and night averages.

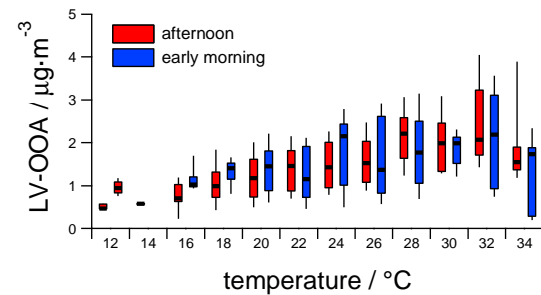
634

635

636 a)



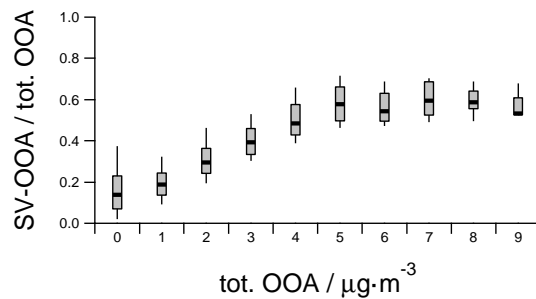
b)



637

638

c)



639

640 **Fig. 4** Box plots describing OOA composition, where horizontal lines indicate median values, boxes denote quartiles, and whiskers represent
 641 10^{th} and 90^{th} percentiles. Quantities plotted are for the summer data:
 642 SV-OOA vs. temperature (a), LV-OOA vs. temperature (b), and SV-
 643 OOA fraction vs. total OOA mass (c). The afternoon and early morning
 644 points are estimated as $T_{\text{max}} \pm 2$ hours and midnight to 5 AM, respec-
 645 tively.
 646

647

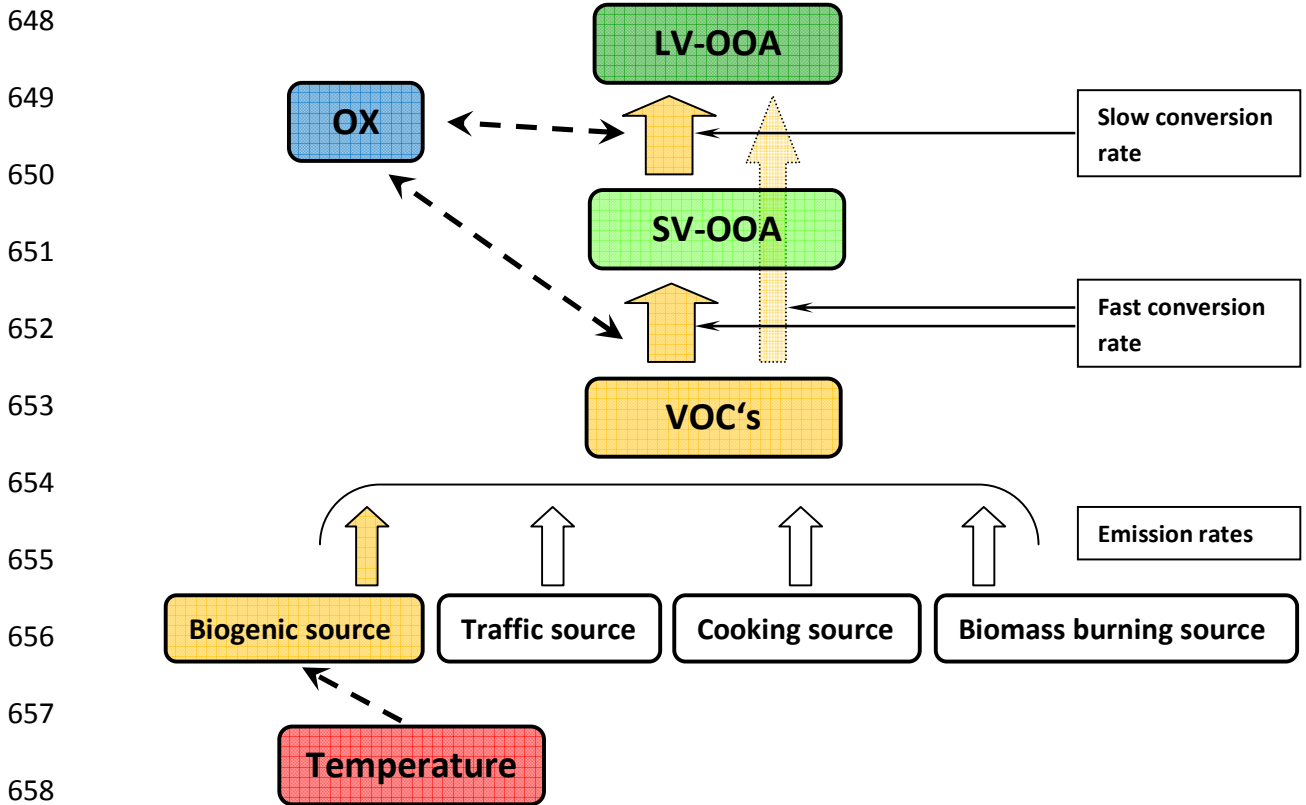


Fig. 5 The simplified scheme represents the emissions and aging processes occurring in ambient during the summer afternoons. The big arrows stand for the emission/conversion rates and the dashed arrows show qualitative dependencies, with the arrow pointing towards the dependent quantity. Higher temperatures in summer primarily enhance the biogenic path (highlighted in orange).