Investigation into the Chemistry of New Particle Formation and Growth in a Subtropical Urban Environment

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12 Abstract

13 The role of different chemical compounds, particularly organics, involved in the new particle 14 formation (NPF) and its consequent growth are not fully understood. Therefore, this study 15 was conducted to investigate the chemistry of aerosol particles during NPF events in an urban 16 subtropical environment. Aerosol chemical composition was measured along with particle 17 number size distribution (PNSD) and several other air quality parameters at five sites across an urban subtropical environment. An Aerodyne compact Time-of-Flight Aerosol Mass 18 19 Spectrometer (c-TOF-AMS) and a TSI Scanning Mobility Particle Sizer (SMPS) measured aerosol chemical composition (particles above 50nm in vacuum aerodynamic diameter) and 20 21 PNSD (particles within 9-414 nm in mobility diameter), respectively. Five NPF events, with 22 growth rates in the range 3.3-4.6 nm, were detected at two of the sites. The NPF events 23 happened on relatively warmer days with lower condensation sink (CS). Temporal percent fractions of organics increased after the particles grew enough to have a significant 24 25 contribution to particles volume, while the mass fraction of ammonium and sulphate decreased. This uncovered the important role of organics in the growth of newly formed 26 particles. Three organic markers, factors f_{43} , f_{44} and f_{57} , were calculated and the f_{44} vs f_{43} 27 28 trends were compared between nucleation and non-nucleation days. K-means cluster analysis 29 was performed on f_{44} vs f_{43} data and it was found that they follow different patterns on 1 nucleation days compared to non-nucleation days, whereby f_{43} decreased for vehicle emission 2 generated particles, while both f_{44} and f_{43} decreased for NPF generated particles. It was found 3 for the first time that vehicle generated and newly formed particles cluster in different 4 locations on f_{44} vs f_{43} plot and this finding can be potentially used as a tool for source 5 apportionment of measured particles.

6 **1. Introduction**

7 Aerosol particles affect the Earth's climate, air quality and public health, both directly and 8 indirectly (e.g. Stevens and Feingold, 2009;Lohmann and Feichter, 2005;Pope II and 9 Dockery, 2006). Knowledge about their formation, transformation and physical/chemical 10 properties helps in understanding their effects on climate and human health. New particle 11 formation (NPF; also known as nucleation) events have been observed in different locations, 12 including coastal, forested, rural and urban areas (O'Dowd et al., 2002;Kulmala et al., 2007;Petäjä et al., 2007;Stanier et al., 2004;Woo et al., 2001;Salma et al., 2010;Cheung et al., 13 14 2010;Mejía and Morawska, 2009;Modini et al., 2009), and these events are one of the main sources of ultrafine particles (UFPs; particles smaller than 100nm), in addition to combustion 15 16 emitted particles (Morawska et al., 2009). The significant increase in the number of UFPs 17 after NPF events can potentially cause adverse effects on human health, given the increasing 18 evidence of the toxicity of UFPs and their role in human mortality and morbidity (Donaldson 19 et al., 2002;WHO, 2005;Ning et al., 2003).

20 Over the past years, the traditional view of the NPF has evolved in light of the recent 21 breakthroughs in this field. The main findings leading to the current understanding of NPF 22 events were observations and theoretical prediction of molecular clusters and their size 23 dependent growth via nano-Kohler-type processes (Kulmala et al., 2007;Kulmala et al., 2013;Kulmala et al., 2004). It is now understood that this process includes the following 24 steps (I) production of low and extremely low volatile vapours as a results of chemical 25 reaction in the gas phase; (II) clustering; (III) nucleation; (IV) activation of clusters by a 26 27 different group of vapours; and (V) multicomponent condensation growth of the activated 28 clusters (Ehn et al., 2014;Riccobono et al., 2014;Kulmala et al., 2014b;Kulmala et al., 2014a).

The chemistry and dynamics of UFPs during NPF events in urban environments have been investigated in only a handful of studies as it was thought that these events tend to happen more in pristine air. However, NPF events have been observed frequently in Brisbane, an urban area in Eastern Australia, and were associated with precursors emitted from traffic and

solar radiation (Cheung et al., 2010). Cheung et al. (Cheung et al., 2010) characterised the 1 2 evolution of particle number size distribution (PNSD) in Brisbane and found 65 nucleation events over a one year period, with an average particle growth rate of 4.6 (nmh⁻¹). Leigh et al 3 4 (2014) found ammonium and sulphate as the dominant species on particle formation days in a 5 short term study based on measurements at a fixed site. More studies need to be carried out, particularly in less investigated areas, using direct measurements techniques, and including 6 7 comprehensive and advanced analysis, in order to determine the species involved in NPF 8 events and the nature of their contribution. To address this gap in knowledge, the main aim of 9 this study was to determine the role of chemical species in NPF events growth process in a 10 subtropical urban environment.

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12 **2.** Materials and Methods

13 2.1. Background

This study was performed as part of the 'Ultrafine Particles from Traffic Emissions and Children's Health' project (UPTECH;Salimi et al., 2013). Air quality measurements were conducted for two consecutive weeks at each of 25 sites across the Brisbane Metropolitan Area, in Australia, during the period October 2010 to August 2012. Chemical composition of the aerosol particles were monitored in real-time at five sites, referred to as s1, s4, s11, s12 and s25.

20 2.2. Instrumentation

Meteorological parameters including wind speed/direction, relative humidity and solar 21 radiation were measured using a 'Monitor Sensor' weather station. A TSI SMPS measured 22 23 PNSD within the size range 9-414 nm. The SMPS system consisted of a TSI 3071 Differential Mobility Analyser (DMA) and a TSI 3782 water-based Condensation Particle Counter (CPC). 24 25 Particle size detection accuracy of the SMPS system was calibrated at the start of 26 measurements at each site using polystyrene latex (PSL) particles. In addition, the system 27 sheath and aerosol flow rate were checked every second day. The measured PNSD data were 28 corrected for diffusion loss inside the system and sampling tube. Further information 29 regarding the SMPS's operating conditions and calibration can be found in Salimi et al. 30 (2014). An Aerodyne compact Time-of-Flight Aerosol Mass Spectrometer (c-ToF-AMS) was

deployed to monitor the chemical composition of aerosol particles in the PM₁ fraction in real 1 2 time. A detailed description of the sampling method and c-TOF-AMS operation can be found in Crilley et al. (2013). Briefly, the c-TOF-AMS was housed in a vacant room at each site and 3 4 sampled for 2-3 weeks. The sampling interval was 5 minutes, alternating equally between 5 particle time of flight and mass spectrometer modes. The instrument was calibrated for both 6 ion efficiency and particle size at the beginning of measurements at each site, and ion 7 efficiency calibration was also performed in the middle and at the end of measurement 8 campaigns at each site. All calibrations were performed according to the standard procedures 9 proposed in the literature (Drewnick et al., 2005;Jayne et al., 2000;Jimenez et al., 2003c). 10 Solar radiation and other meteorological parameters were measured by a Monitor Sensors 11 µSmart Series weather station, a TSI 3781 water-based CPC was used for particle number concentration (PNC) measurements, and a TSI DustTrak measured particle mass 12 13 concentration ($PM_{2.5}$ and PM_{10}).

14 **2.3.** Data analysis

15 <u>Characterisation of New Particle Formation Events:</u> NPF events were identified and 16 classified using the procedure developed by Dal Maso et al (Dal Maso et al., 2005). PNSD 17 data were analysed to identify a distinct new mode of particles in the nucleation mode size 18 range (particles with diameter less than 30 nm). If the mode grew in size and was persistent 19 for more than one hour, it was assigned as a NPF event. Once identified, the growth rate of 20 the nucleation events were calculated following the procedure described in (Creamean et al., 2011).

22 <u>Condensation Sink:</u> The surface area of aerosol particles available for condensation can be 23 estimated by the condensation sink (CS) parameter which is calculated from the measured 24 PNSD. The CS determines the rate of condensation of gaseous molecules on pre-existing 25 particles and is a function of PNSD (Pirjola et al., 1999;Lehtinen et al., 2003). CS, with unit s⁻¹, was calculated using the following formula (Lehtinen et al., 2003):

27 1)
$$CS = 2\pi D \int_{0}^{\pi} d_{p} \beta_{M}(d_{p}) n(d_{p}) dd_{p} = 2\pi D \sum_{i} \beta_{M_{i}} d_{p,i} N_{i}$$

28 where *D* is the diffusion coefficient, d_p is the particle diameter, N_i is the concentration of 29 particles and β_M can be expressed as (Fuks et al., 1971):

1 1)
$$\beta_M = \frac{Kn+1}{0.377Kn+1+\frac{4}{3}\alpha^{-1}Kn^2+\frac{4}{3}\alpha^{-1}Kn}$$

- 2 where α is the sticking coefficient and is assumed to be 1 (Clement et al., 1996) and *Kn* is the 3 Knudsen number, which is equal to $\frac{2\lambda_{\nu}}{d_{p}}$. The mean free path (λ_{ν}) is a function of pressure
- 4 and temperature and can be calculated using the following formula (Willeke, 1976):

5 2)
$$\lambda_{\nu} = \lambda_{o}(\frac{101}{p})(\frac{T}{296.2})(\frac{(1+\frac{110}{296.2})}{(1+\frac{110}{T})})$$

6 where *P* is in *kPa* and *T* in *K*. $\lambda_0 = 0.039 \,\mu$ m, which is the mean free path of H_2SO_4 at standard

7 conditions (Bae et al., 2010). CS's were calculated using the above mentioned procedure on

- 8 the PNSD data.
- 9

10 <u>Particle Size</u>: Due to the difference in detection limit between the two instruments (D_{va} = 50nm for AMS and $D_m = 10$ nm for SMPS), measurement of the mass distribution for the 11 12 smallest particles detected during each nucleation was delayed for the AMS in order for 13 particles to grow to the detactable size . Ratio of mobility diameter (D_m) , measured by the SMPS, to vacuum aerodynamic diameter (D_{va}) is a function of size, composition, shape and 14 15 relative humidity for ambient particles. This ratio can be simplified to be equal to particle density, assuming that particles are spherical (Jimenez et al., 2003a). The newly formed 16 particles were assumed to have the density of around 1.8 g/cm³ and therefore, $D_m/D_{va} = 1.8$ 17 for newly formed particles (Zhang et al., 2004). Therefore, the AMS's lowest detection limit 18 19 will be equivalent to a $D_m = 30$ nm.

<u>Particle Chemical Composition:</u> Concentrations of sulphate, nitrate, organics and ammonium
were measured by a c-TOF-AMS, as these have been identified as the main contributing
species to the NPF events and subsequent particulate growth in literature (Zhang et al., 2004).
These data were processed and analysed using Squirrel v1.51 in Igor Pro v6.22. The organic
aerosol (OA) can be divided into oxygenated OA (OOA) and hydrocarbon-like OA (HOA),
based upon key m/z ions that have been shown to be surrogates for the different components.
OOA can be separated further into low-volatility OOA (LV-OOA) and semi-volatile (SV-

OOA). Two main ions, m/z 44 (CO₂⁺) and m/z 43 (mostly C₂H₃O⁺), can characterise the 1 2 evolution of OA in the atmosphere (Ng et al., 2010b). The ratios of m/z 44 and m/z 43 to total 3 organic signal (f_{44} and f_{43} , respectively) indicate the degree of oxidation of OA, since the SV-OOA component spectra have a lower f_{44} and higher f_{43} compared to LV-OOA. OA has been 4 5 shown to occupy a triangular region of the f_{44} vs f_{43} plot with younger OA occupying the 6 lower part of the triangle, while more oxidised and subsequently more aged OA are concentrated in the upper part (Ng et al., 2010b;Ng et al., 2011). Therefore, f_{43} and f_{44} were 7 8 calculated and plotted against each other to determine the type and evolution of OA during 9 NPF events. m/z 57 ions have been shown to be a tracer of HOA (primary organics from 10 combustion sources) (Ng et al., 2010a) and therefore, f_{57} was calculated to investigate the OA 11 originated from vehicle emissions.

12 LOESS Smoothing: In order to better understand the trends and patterns of the data, LOESS

13 technique as a locally weighted polynomial regression smoothing method was selected and

14 applied to the data. The resulting smooth functions, and their 95% confidence intervals,

15 indicate the trends of the relevant data throughout the paper.(Cleveland and Devlin, 1988)

16 K-means clustering: This clustering algorithm minimises the within-cluster sum of squares in 17 order to divide M observations with N dimensions into K clusters (Hartigan and Wong, 1979). 18 A common method of choosing the number of clusters is to compare visually a measure of 19 error, such as sum of squared error (SSE), with sequential number of clusters by plotting the 20 measure of error on y axis and the number of clusters on x axis (Everitt et al., 2010). Optimum number of clusters is the point at which the measure of error flattens and form an 21 22 elbow. The abovementioned clustering technique were applied to the f_{44} and f_{43} to assess the possible clustering for days with and without nucleation. 23

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25 **3. Results and discussion**

26 **3.1.** Identified New Particle Formations

PNSD data was available for 6, 18, 11, 13, and 8 days in s1, s4, s11, s12, and s25 respectively. After the analysis of PNSD measured at five sites, five NPF events were detected, where three events were observed in s12 and two in s25, while no events were observed at s1, s4 and s11. Therefore, we focused only on s12 and s25, for which the NPF growth rates varied from 3.3 to 4.6 (nmh⁻¹) (Table 1).

Figure 1 summarises the average solar radiation, relative humidity, temperature and CS 1 2 calculated using LOESS method for days with NPF events (nucleation days), as well as for the days where no NPF events were observed (non-nucleation days) at s12 and s25. 3 Nucleation days had higher solar radiation intensity compared to non-nucleation days at s25 4 5 while opposite trend was observed at s12. Nucleation days had slightly higher temperature compared to non-nucleation days at both sites. Higher relative humidity was observed on 6 7 nucleation days compared to non-nucleation days at s12, whereas opposite trend was 8 observed at S25. The wind speed/direction have not been plotted as they did not show a typical trend during the nucleation and non-nucleation days. CS was found to be lower on 9 10 nucleation days compared to non-nucleation days, about two hours before the start of 11 nucleation at both sites. Similar trends have been observed in other studies (Salma et al., 12 2010:Hussein et al., 2008). CS is a measure of the available surface area for condensation of 13 vapours, as well as for the scavenging of particle clusters. A small CS is favourable for 14 nucleation as it produces less surface area for vapours and newly formed clusters, in 15 agreement with the observations in Figure 1.

Figure 2 illustrates the average mass concentration of organics, nitrate, sulphate and ammonium during nucleation and non-nucleation days at s12 and s25. Mass concentration of organics and nitrate followed relatively the same trend through the day until 4-5 pm (6-7 hours after the start of the NPF event) at s12 and s25.

20 However, mass concentration of organics and nitrate increased dramatically between 3-10 pm

21 on nucleation days, at higher rate compared to non-nucleation days at s12, and the same trend

22 was observed at s25, with an hours delay in the increase of the mass concentration of organics

and nitrate. At s25, Sulphate and ammonium mass concentration started to increase at 12pm,

- 24 reached a peak and decreased subsequently, however, a significant increase was observed
- 25 around 4pm.

3.2. Evolution of Chemical Composition of Newly Formed Particles

The analysis and interpretation of the chemical composition of the newly formed particles focusses mostly on the particles in the growth phase due to the lowest detection limit of the C-TOF AMS which does not permit to see them in the formation phase. Particle volume distribution (PVD) was calculated from the PNSD data as it is a better measure for

31 comparison with the mass.

It has been previously determined in the literature that the main contributing species to NPF 1 2 events and subsequent particulate growth are sulphate, nitrate, organics and ammonium (Zhang et al., 2004). Therefore, the percent fractions of each of these chemical species were 3 4 calculated by dividing the mass concentration of each chemical species by the total (sulphate 5 + nitrate + ammonium + organics). 6 Figure 3 illustrates the evolution of PNSD, PVD, and mass concentration of chemical species 7 (organics, nitrate, sulphate and ammonium) during the three consecutive NPF events at s12. 8 In general, mass concentration of the aerosol species followed the evolution of particle 9 volume distribution as expected. Ammonium, sulphate and nitrate mass fractions peaked just before the particles volume increase due to the growth of newly formed particles. However, 10 11 organics mass fraction followed the opposite trend, with a significant rise after the increase in particles volume due to the growth of new particles. In other words, the fraction of organics 12 13 increased and the fraction of ammonium, sulphate and nitrate decreased when the newly formed particles grew enough to dominate the particles volume. This shows the import 14 15 contribution of organics to the growth of newly formed particles. 16 Time series of PNSD and mass concentration of particle species during two NPF events 17 happening on two consecutive days at s25 are illustrated in Figure 4. The mass concentrations 18 of the chemical species and their fractions followed similar trends as the ones at s12. At s12, 19 the magnitude of mass fractions changed from almost 50, 30, 10, and 10% before the

20 nucleation to 70, 20, 6, and 4% after the event for organics, sulphate, ammonium and nitrate

and respectively. At s25, the changes in mass fractions were more dramatic as they changed

22 from 40, 25, 25, and 10 % before the event to 85, 5, 5, and 5% after the nucleation occurred.

23 **3.3.** Role of organics

24 As indicated in the previous section, five NPF events were observed at s12 and the 25 contribution of organics to the growth of newly formed particles was high, therefore, further 26 data analysis regarding the role of organics were performed. Firstly, f_{43} and f_{44} were calculated 27 for the total chemical mass data on nucleation days and non-nucleation days. f_{44} was plotted against f_{43} at each hour of the day, as well as the triangle space proposed by Ng et al. (Ng et 28 29 al., 2010b) (Figure 5). During the non-nucleation days, the OA was concentrated in the 30 middle and right border of the triangular region, which is the beginning of the LV-OOA region. As expected, the f_{44} decreased during morning and afternoon rush hours, as the 31

1 particles were younger and less oxidised, while, f_{43} remained almost unchanged during the 2 day.

3 During the nucleation days, several hours after the start of nucleation, when the particles grew 4 enough to have significant effect on the total signal, f_{44} and f_{43} both decreased and the aerosol 5 components reached the bottom left side of the triangular region. It is expected that the 6 particles originating from nucleation events cluster somewhere at top of the triangle, however, 7 as the particles measured in our study have been originating from nucleation events and 8 primary vehicle emissions, they clustered at the middle left hand side of the triangle. The 9 abovementioned visual inspection of the figure revealed potential clustering on the graph during nucleation and non-nucleation days. 10

11 In order to assess the possible different clustering of f_{44} vs. f_{43} for nucleation days compared 12 to non-nucleation days, K-means clustering technique was applied on the f_{43} and f_{44} data. A period between 3-5 pm was selected for this purpose as this was the initial stage where the 13 14 newly formed particles grew enough to have significant contribution to the total signal. In order to find the optimum number of clusters, SSE was plotted against the sequential number 15 of clusters and five number of clusters was found to be appropriate as it was located at the 16 17 elbow in the plot (Figure 6). 18 The five identified clusters as well as their 95% confidence ellipse are illustrated in Figure

19 7.a, 93% of the data measured in nucleation days were in clusters 3-5 (54%, 15%, and 24% in

20 clusters3,4,and 5 respectively) while clusters1 and 2 contained 77% of the data measured in

21 non-nucleation days (Figure 7). In addition, cluster 1 and 5 contained less than 1% of

22 nucleation and non-nucleation days respectively. These show a distinct clustering on f_{44} vs. f_{43}

23 for nucleation days compared to non-nucleation days and indicating a potential application of

24 f_{44} vs. f_{43} plot for identification of the newly formed particles.

It has been reported in the literature that f_{57} is associated with combustion generated primary organics (Canagaratna et al., 2004;Jimenez et al., 2003b). f_{57} was calculated and its diurnal variation was modelled using a LOESS smoothing model. The measured data and their smoothed average showed that f_{57} followed exactly the same trend on nucleation and nonnucleation days, with an increase during the morning and afternoon rush hours (Figure 8). All non-nucleation days did not have exactly the same meteorological conditions but the effects

31 of their variation on f_{57} were minimal.

4. Summary and Conclusions

2 In summary, PNSD, chemical composition and meteorological parameters were measured at five sites across the Brisbane Metropolitan Area. Five NPF events, with growth rates ranging 3 from 3.3-4.6 nm.hr⁻¹, were observed at two of the five sites, and the NPF events happened on 4 days with lower CS and higher temperature than non-event days. Higher sulphate, nitrate, 5 6 ammonium and organics were observed on nucleation days compared with days when no 7 nucleation was observed. Percent fractions of nitrate, sulphate, ammonium and organic 8 chemical species were calculated and their diurnal trends were modelled using the LOESS. 9 Ammonium, sulphate and nitrate mass fractions increased before the newly formed particles 10 grew enough to have a significant contribution to the particles volume, peaked around that 11 time and decreased after that. Conversely, the organics percent fraction increased significantly 12 after the contribution of new particles to total volume, indicating the important role of 13 organics in the growth phase of NPF events. f_{44} and f_{43} were analysed to investigate the role of organics more in depth, as the f_{44} vs f_{43} would reveal information regarding the level of 14 15 oxidation and volatility of OOA. f_{43} and f_{44} both decreased after the start of nucleation, while f_{44} decreased only in case of particles generated from vehicle emission. K-means clustering 16 17 analysis revealed that the aerosol particles generated by vehicle emissions and NPF events 18 clustered in different locations on the f_{44} vs f_{43} plot. This determined the potential application 19 of the f_{44} vs f_{43} plot for identification of the source/s and transformation of the OOA 20 components as they clearly followed different patterns. However, it should be noted that the 21 analysis were based on only five NPF events and the results may not be a general rule.

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1 References

- 2 Bae, M. S., Schwab, J. J., Hogrefe, O., Frank, B. P., Lala, G. G., and Demerjian, K. L.: 3 Characteristics of size distributions at urban and rural locations in New York, Atmos. Chem. 4 Phys., 10, 4521-4535, 2010.
- 5 Canagaratna, M. R., Jayne, J. T., Ghertner, D. A., Herndon, S., Shi, Q., Jimenez, J. L., Silva, P. J., Williams, P., Lanni, T., and Drewnick, F.: Chase studies of particulate emissions from 6
- 7 in-use New York City vehicles, Aerosol Science and Technology, 38, 555-573, 2004.
- 8 Cheung, H. C., Ristovski, Z. D., and Morawska, L.: Observation of new particle formation in 9 subtropical urban environment, Atmospheric Chemistry and Physics Discussions, 10, 22623-10 22652, 2010.
- 11 Clement, C. F., Kulmala, M., and Vesala, T.: Theoretical consideration on sticking probabilities, Journal of Aerosol Science, 27, 869-882, http://dx.doi.org/10.1016/0021-12 13 8502(96)00032-8, 1996.
- 14 Cleveland, W. S., and Devlin, S. J.: Locally Weighted Regression: An Approach to 15 Regression Analysis by Local Fitting, Journal of the American Statistical Association, 83, 16 596-610, 10.1080/01621459.1988.10478639, 1988.
- 17 Creamean, J. M., Ault, A. P., Ten Hoeve, J. E., Jacobson, M. Z., Roberts, G. C., and Prather,
- 18 K. A.: Measurements of Aerosol Chemistry during New Particle Formation Events at a 19 Remote Rural Mountain Site, Environmental Science & Technology, 45, 8208-8216, 20 10.1021/es103692f, 2011.
- 21 Crilley, L. R., Ayoko, G. A., Jayaratne, E. R., Salimi, F., and Morawska, L.: Aerosol mass 22 spectrometric analysis of the chemical composition of non-refractory PM1 samples from 23 school environments in Brisbane, Australia, Science of The Total Environment, 458-460, 81-24 89, http://dx.doi.org/10.1016/j.scitotenv.2013.04.007, 2013.
- 25 Crilley, L. R., Jayaratne, E. R., Ayoko, G. A., Miljevic, B., Ristovski, Z., and Morawska, L.:
- 26 Observations on the Formation, Growth and Chemical Composition of Aerosols in an Urban
- 27 Environment, Environmental Science & Technology, 48, 6588-6596, 10.1021/es5019509,
- 28 2014.
- 29 Dal Maso, M., Kulmala, M., Riipinen, I., Wagner, R., Hussein, T., Aalto, P. P., and Lehtinen,
- 30 K. E. J.: Formation and growth of fresh atmospheric aerosols: eight years of aerosol size
- distribution data from SMEAR II, Hyytiälä, Finland, Boreal Environment Research, 10, 323-31
- 32 336, 2005.
- 33 Donaldson, K., Brown, D., Clouter, A., Duffin, R., MacNee, W., Renwick, L., Tran, L., and
- 34 Stone, V.: The pulmonary toxicology of ultrafine particles, Journal of aerosol medicine, 15, 35 213-220, 2002.
- 36 Drewnick, F., Hings, S. S., DeCarlo, P., Javne, J. T., Gonin, M., Fuhrer, K., Weimer, S.,
- 37 Jimenez, J. L., Demerjian, K. L., and Borrmann, S.: A new time-of-flight aerosol mass 38 spectrometer (TOF-AMS)-Instrument description and first field deployment, Aerosol
- 39 Science and Technology, 39, 637-658, 2005.
- 40 Ehn, M., Thornton, J. A., Kleist, E., Sipilä, M., Junninen, H., Pullinen, I., Springer, M.,
- 41 Rubach, F., Tillmann, R., Lee, B., Lopez-Hilfiker, F., Andres, S., Acir, I. H., Rissanen, M.,
- 42 Jokinen, T., Schobesberger, S., Kangasluoma, J., Kontkanen, J., Nieminen, T., Kurtén, T.,
- 43 Nielsen, L. B., Jørgensen, S., Kjaergaard, H. G., Canagaratna, M., Maso, M. D., Berndt, T.,

- 1 Petäjä, T., Wahner, A., Kerminen, V. M., Kulmala, M., Worsnop, D. R., Wildt, J., and
- 2 Mentel, T. F.: A large source of low-volatility secondary organic aerosol, Nature, 506, 476-
- 3 479, 2014.
- Everitt, B. S., Landau, S., Leese, M., and Stahl, D.: Cluster Analysis, in, 5 ed., Wiley,
 Hoboken, 2010.
- Fuks, N., Sutugin, A. G., and Soo, S.: Topics in current aerosol research, Pergamon press,1971.
- 8 Hartigan, J. A., and Wong, M. A.: Algorithm AS 136: A K-Means Clustering Algorithm,
- Journal of the Royal Statistical Society. Series C (Applied Statistics), 28, 100-108, 10.2307/2346830, 1979.
- 11 Hussein, T., Martikainen, J., Junninen, H., Sogacheva, L., Wagner, R., Dal Maso, M.,
- Riipinen, I., Aalto, P. P., and Kulmala, M.: Observation of regional new particle formation in the urban atmosphere, Tellus B, 60, 509-521, 10.1111/j.1600-0889.2008.00365.x, 2008.
- 14 Jayne, J. T., Leard, D. C., Zhang, X., Davidovits, P., Smith, K. A., Kolb, C. E., and Worsnop,
- 15 D. R.: Development of an aerosol mass spectrometer for size and composition analysis of
- 16 submicron particles, Aerosol Science & Technology, 33, 49-70, 2000.
- Jimenez, J. L., Bahreini, R., Cocker, D. R., Zhuang, H., Varutbangkul, V., Flagan, R. C.,
 Seinfeld, J. H., O'Dowd, C. D., and Hoffmann, T.: New particle formation from
- photooxidation of diiodomethane (CH2I2), Journal of Geophysical Research: Atmospheres
 (1984–2012), 108, 2003a.
- 21 Jimenez, J. L., Jayne, J. T., Shi, Q., Kolb, C. E., Worsnop, D. R., Yourshaw, I., Seinfeld, J.
- 22 H., Flagan, R. C., Zhang, X., and Smith, K. A.: Ambient aerosol sampling using the aerodyne
- aerosol mass spectrometer, Journal of Geophysical Research: Atmospheres (1984–2012), 108,
- 24 2003b.
- Jimenez, J. L., Jayne, J. T., Shi, Q., Kolb, C. E., Worsnop, D. R., Yourshaw, I., Seinfeld, J.
 H., Flagan, R. C., Zhang, X., Smith, K. A., Morris, J. W., and Davidovits, P.: Ambient aerosol
 sampling using the Aerodyne Aerosol Mass Spectrometer, Journal of Geophysical Research:
 Atmospheres, 108, 8425, 10.1029/2001JD001213, 2003c.
- Kulmala, M., Kerminen, V.-M., Anttila, T., Laaksonen, A., and O'Dowd, C. D.: Organic
 aerosol formation via sulphate cluster activation, Journal of Geophysical Research:
 Atmospheres, 109, D04205, 10.1029/2003JD003961, 2004.
- 32 Kulmala, M., Riipinen, I., Sipilä, M., Manninen, H. E., Petäjä, T., Junninen, H., Maso, M. D.,
- 33 Mordas, G., Mirme, A., Vana, M., Hirsikko, A., Laakso, L., Harrison, R. M., Hanson, I., 34 Leung, C., Lehtinen, K. E. J., and Kerminen, V.-M.: Toward direct measurement of
- atmospheric nucleation, Science (New York, N.Y.), 318, 89-92, 2007.
- 36 Kulmala, M., Kontkanen, J., Junninen, H., Lehtipalo, K., Manninen, H. E., Nieminen, T.,
- Petäjä, T., Sipilä, M., Schobesberger, S., and Rantala, P.: Direct observations of atmospheric
 aerosol nucleation, Science, 339, 943-946, 2013.
- Kulmala, M., Petäjä, T., Ehn, M., Thornton, J., Sipilä, M., Worsnop, D., and Kerminen, V.M.: Chemistry of Atmospheric Nucleation: On the Recent Advances on Precursor
 Characterization and Atmospheric Cluster Composition in Connection with Atmospheric New
 Particle Formation, Annual review of physical chemistry, 65, 21-37, 2014a.
- Kulmala, M., Petäjä, T., Ehn, M., Thornton, J., Sipilä, M., Worsnop, D. R., and Kerminen, V.
 M.: Chemistry of atmospheric nucleation: On the recent advances on precursor

- 1 characterization and atmospheric cluster composition in connection with atmospheric new
- 2 particle formation, in: Annual Review of Physical Chemistry, 21-37, 2014b.
- Lehtinen, K. E. J., Korhonen, H., Dal Maso, M., and Kulmala, M.: On the concept of condensation sink diameter, Boreal Environment Research, 8, 405-411, 2003.
- Lohmann, U., and Feichter, J.: Global indirect aerosol effects: a review, Atmos. Chem. Phys.,
 5, 715-737, 2005.
- Mejía, J. F., and Morawska, L.: An investigation of nucleation events in a coastal urban
 environment in the Southern Hemisphere, Atmos. Chem. Phys., 9, 7877-7888, 2009.
- 9 Modini, R., Ristovski, Z., Johnson, G. R., He, C., Surawski, N., Morawska, L., Suni, T., and
- Kulmala, M.: New particle formation and growth at a remote, sub-tropical coastal location,Atmos. Chem. Phys, 9, 7607-7621, 2009.
- 12 Morawska, L., Wang, H., Ristovski, Z., Jayaratne, E. R., Johnson, G., Cheung, H. C., Ling,
- 13 X., and He, C.: JEM Spotlight: Environmental monitoring of airborne nanoparticles, Journal
- 14 of Environmental Monitoring, 11, 1758-1773, 10.1039/B912589M, 2009.
- Ng, N., Canagaratna, M., Jimenez, J., Zhang, Q., Ulbrich, I., and Worsnop, D.: Real-time
 methods for estimating organic component mass concentrations from aerosol mass
 spectrometer data, Environmental science & technology, 45, 910-916, 2010a.
- 18 Ng, N., Canagaratna, M., Zhang, Q., Jimenez, J., Tian, J., Ulbrich, I., Kroll, J., Docherty, K.,
- Chhabra, P., and Bahreini, R.: Organic aerosol components observed in Northern
 Hemispheric datasets from Aerosol Mass Spectrometry, Atmos. Chem. Phys., 10, 4625-4641,
 2010b.
- 22 Ng, N., Canagaratna, M., Jimenez, J., Chhabra, P., Seinfeld, J., and Worsnop, D.: Changes in
- organic aerosol composition with aging inferred from aerosol mass spectra, Atmos. Chem.
 Phys., 11, 6465-6474, 2011.
- Ning, L., Sioutas, C., Cho, A., Schmitz, D., Misra, C., Sempf, J., Meiying, W., Oberley, T.,
 Froines, J., and Nel, A.: Ultrafine Particulate Pollutants Induce Oxidative Stress and
 Mitochonrial Damage, Environmental Health Perspectives, 111, 455, 2003.
- O'Dowd, C. D., Jimenez, J. L., Bahreini, R., Flagan, R. C., Seinfeld, J. H., Hämeri, K., Pirjola,
 L., Kulmala, M., Jennings, S. G., and Hoffmann, T.: Marine aerosol formation from biogenic
 iodine emissions, Nature, 417, 632-636, 2002.
- 31 Petäjä, T., Kerminen, V., Dal Maso, M., Junninen, H., Koponen, I., Hussein, T., Aalto, P.,
- 32 Andronopoulos, S., Robin, D., and Hämeri, K.: Sub-micron atmospheric aerosols in the
- 33 surroundings of Marseille and Athens: physical characterization and new particle formation,
- 34 Atmos. Chem. Phys, 7, 2705-2720, 2007.
- 35 Pirjola, L., Kulmala, M., Wilck, M., Bischoff, A., Stratmann, F., and Otto, E.: FORMATION
- 36 OF SULPHURIC ACID AEROSOLS AND CLOUD CONDENSATION NUCLEI: AN 37 EXPRESSION FOR SIGNIFICANT NUCLEATION AND MODEL COMPRARISON,
- EXPRESSION FOR SIGNIFICANT NUCLEATION AND MODEL COMPRARISON,
 Journal of Aerosol Science, 30, 1079-1094, <u>http://dx.doi.org/10.1016/S0021-8502(98)00776-</u>
- 39 <u>9</u>, 1999.
- 40 Pope II, C. A., and Dockery, D. W.: Health effects of fine particulate air pollution: lines that 41 connect, Air & Waste Management Association, 709-742, 2006.
- 42 Riccobono, F., Schobesberger, S., Scott, C. E., Dommen, J., Ortega, I. K., Rondo, L.,
- 43 Almeida, J., Amorim, A., Bianchi, F., Breitenlechner, M., David, A., Downard, A., Dunne, E.

- M., Duplissy, J., Ehrhart, S., Flagan, R. C., Franchin, A., Hansel, A., Junninen, H., Kajos, M., 1 Keskinen, H., Kupc, A., Kürten, A., Kvashin, A. N., Laaksonen, A., Lehtipalo, K., 2 3 Makhmutov, V., Mathot, S., Nieminen, T., Onnela, A., Petäjä, T., Praplan, A. P., Santos, F. 4 D., Schallhart, S., Seinfeld, J. H., Sipilä, M., Spracklen, D. V., Stozhkov, Y., Stratmann, F., 5 Tomé, A., Tsagkogeorgas, G., Vaattovaara, P., Viisanen, Y., Vrtala, A., Wagner, P. E., 6 Weingartner, E., Wex, H., Wimmer, D., Carslaw, K. S., Curtius, J., Donahue, N. M., Kirkby, J., Kulmala, M., Worsnop, D. R., and Baltensperger, U.: Oxidation Products of Biogenic 7 8 Emissions Contribute to Nucleation of Atmospheric Particles, Science, 344, 717-721, 9 10.1126/science.1243527, 2014.
- 10 Salimi, F., Mazaheri, M., Clifford, S., Crilley, L., Laiman, R., and Morawska, L.: Spatial 11 variation of particle number concentration in school microscale environments and its impact
- 12 on exposure assessment, Environmental Science & Technology, 10.1021/es400041r, 2013.
- 13 Salimi, F., Ristovski, Zoran, Mazaheri, Mandana, Laiman, Rusdin, Crilley, Leigh R., He,
- 14 Congrong, Clifford, Sam and Morawska, Lidia: Assessment and Application of Clustering
- 15 Techniques to Atmospheric Particle Number Size Distribution for the Purpose of Source
- 16 Apportionment, Under review in Atmospheric Chemistry and Physics, 2014.
- 17 Salma, I., Borsós, T., Weidinger, T., Aalto, P., Hussein, T., Maso, M. D., and Kulmala, M.:
- 18 Production, growth and properties of ultrafine atmospheric aerosol particles in an urban
- 19 environment, Atmospheric Chemistry and Physics Discussions, 10, 13689-13721, 2010.
- 20 Stanier, C. O., Khlystov, A. Y., and Pandis, S. N.: Nucleation Events During the Pittsburgh
- 21 Air Quality Study: Description and Relation to Key Meteorological, Gas Phase, and Aerosol
- 22 Parameters Special Issue of Aerosol Science and Technology on Findings from the Fine
- 23 Particulate Matter Supersites Program, Aerosol Science and Technology, 38, 253-264, 2004.
- Stevens, B., and Feingold, G.: Untangling aerosol effects on clouds and precipitation in a
 buffered system, Nature, 461, 607-613, 2009.
- 26 <u>http://www.ilaqh.qut.edu.au/Misc/UPTECH%20Home.htm.</u>
- WHO, W. H. O.: Air Quality Guidelines: Global Update 2005: Particulate Matter, Ozone,
 Nitrogen Dioxide, and Sulfur Dioxide, World Health Organization Europe, 2005.
- Willeke, K.: Temperature dependence of particle slip in a gaseous medium, Journal of Aerosol Science, 7, 381-387, <u>http://dx.doi.org/10.1016/0021-8502(76)90024-0</u>, 1976.
- Woo, K., Chen, D., Pui, D., and McMurry, P.: Measurement of Atlanta aerosol size
 distributions: observations of ultrafine particle events, Aerosol Science and Technology, 34,
 75-87, 2001.
- 34 Zhang, Q., Stanier, C. O., Canagaratna, M. R., Jayne, J. T., Worsnop, D. R., Pandis, S. N., and
- 35 Jimenez, J. L.: Insights into the Chemistry of New Particle Formation and Growth Events in
- 36 Pittsburgh Based on Aerosol Mass Spectrometry, Environmental Science & Technology, 38,
- 37 4797-4809, 10.1021/es035417u, 2004.



Figure 1: Average diurnal pattern of solar radiation (SR), humidity, temperature and condensation sink (CS) as well as the measured data on nucleation and non-nucleation day at s12 and s25. Shaded areas represent the 95% confidence interval in the mean. Data includes 5 and 51 nucleation and non-nucleation days respectively.



Figure 2: Average diurnal pattern of organics, nitrate, sulphate and ammonium concentrations as well as the measured data on nucleation and non-nucleation days at s12 and S25. Shaded areas represent the 95% confidence interval in the mean. Data includes 5 and 51 nucleation and non-nucleation days respectively.



Figure 3: Time series of the particle number/volume size distributions (top two graphs) and mass concentration of particle species and their fraction to the total (bottom two graphs) at s12. Solid lines are calculated using LOESS smoothing technique and the shaded areas represent the 95% CI.



Figure 4: Time series of the particle number/volume size distributions (top two graphs) and mass concentration of particle species and their fraction to the total (bottom two graphs) at s25. Solid lines are calculated using LOESS smoothing technique and the shaded areas represent the 95% CI.



Figure 5: f_{44} vs f_{43} at each hour of the day for all data measured during nucleation and non-nucleation days. The triangle from Ng et al. (2010b) is drawn as a visual aid. Within the period of 15-17, the newly formed particles grew enough to have significant contribution to total signal. Data includes 5 and 51 nucleation and non-nucleation days respectively.



2 Figure 6: SSE versus the number of clusters. The sharp decrease in SSE stops at 5 clusters forming an elbow.



1 2 3 4

Figure 7: The f_{44} vs f_{43} plot for nucleation days and non-nucleation days with the five clusters and their 95% confidence epsilon (a). The contributing proportion of each cluster to the total for nucleation and non-nucleation days (b). The triangle from Ng et al. (2010b) is drawn as a visual aid.

33%

50 Fraction (%)

13%

75

19%

100

34%

25

no-

0

5

4

5



Figure 8: Diurnal variation of the *f*₅₇ during the nucleation and non-nucleation days. Solid lines have been calculated using the LOESS smoothing method and the shaded areas represent the 95% confidence interval. Data includes 5 and 51 nucleation and non-nucleation days. The data includes weekdays and weekend as no exclusions were applied to increase the statistical power.

Table 1: Duration of each nucleation event as well as their average growth rates. Growth rates were calculated by a linearfit from 9nm to 20nm.

Site	start of event	end of event	GR(nm/hr)
	12/08/2011	13/08/2011	
s12	11:00	5:00	3.9
	13/08/2011	14/08/2011	
s12	12:30	6:20	4.6
	14/08/2011	15/08/2011	
s12	15:20	6:00	3.3
	25/08/2012	26/08/2012	
s25	12:24	2:14	4.3
	26/08/2012	27/08/2012	
s25	12:00	1:00	3.8