

1 **Assessing the Regional Impact of Indonesian Biomass**
2 **Burning Emissions Based on Organic Molecular Tracers**
3 **and Chemical Mass Balance Modeling**

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14

15 **Abstract**

16 Biomass burning activities commonly occur in Southeast Asia (SEA), and are particularly
17 intense in Indonesia during dry seasons. The effect of biomass smoke emissions on air quality
18 in the city state of Singapore was investigated during a haze episode in October 2006.
19 Substantially increased levels of airborne particulate matter (PM) and associated chemical
20 species were observed during the haze period. Specifically, the enhancement in the
21 concentration of molecular tracers for biomass combustion such as levoglucosan by as much
22 as two orders of magnitude and diagnostic ratios of individual organic compounds indicated
23 that biomass burning emissions caused a regional smoke haze episode due to their long-range
24 transport by prevailing winds. With the aid of air mass back trajectories and chemical mass
25 balance modeling, large-scale forest and peat fires in Sumatra and Kalimantan were identified
26 as the sources of the smoke aerosol, exerting a significant impact on air quality in downwind
27 areas, such as Singapore.

1 **1 Introduction**

2 Air quality in SEA is affected by multiple emission sources, including combustion of fossil
3 fuels for power generation, transportation and industrial processes, biomass burning, and
4 primary as well as secondary emissions from biogenic sources (Balasubramanian et al., 1999;
5 Balasubramanian et al, 2003; See et al., 2006). Biomass burning is particularly an important
6 source of carbonaceous aerosol, resulting from the open burning of agricultural residues,
7 slash-and-burn practices, grassland and forest fires, and residential combustion of biofuels for
8 cooking and heating. Some of the most intense fires, associated with substantial smoke
9 emissions, have occurred many times in Indonesia since the 1990s. Among the different types
10 of biomass burned in Indonesia, peat fires are the key source of smoke haze (Page et al., 2002;
11 See et al., 2007), particularly in El Nino years. Indonesia has the largest area of tropical peat
12 lands in the world, with 27 million hectares scattered around the different provinces (Page et
13 al., 2002). The burning of biomass and organic soil produces large quantities of carbonaceous
14 particles comprising burnt or partially burnt carbon.

15 Lowland forests and plantations are burnt annually, especially during dry seasons (March
16 through April and July through October), by farmers and plantation owners as part of land-
17 clearing activities. Such fires can easily get out of control and spread to adjacent forest and
18 peat areas, resulting in large-scale fires. Moreover, the ground in these areas often consists of
19 peat, which is known to sustain smoldering fires for extended periods of time, partly
20 occurring underground, and thus releasing large amounts of smoke. The most intensive fire
21 event recorded in the literature occurred in Indonesia in 1997 (Levine, 1999; Heil and
22 Goldammer, 2001), followed by another severe regional haze episode in June 2013 which
23 received considerable international attention.

24 Investigations of emissions from the combustion of peat are limited to laboratory-scale
25 burning of peat in combustion chambers (e.g., Muraleedharan et al., 2000), and few field
26 studies (See et al., 2007; Betha et al., 2013). The field studies showed elevated levels of PM
27 number and mass concentrations as well as various chemical species at downwind locations
28 from the biomass burning source regions in Indonesia. Christian et al. (2003) and Iinuma et al.
29 (2007) quantified gaseous as well as particulate components of smoke derived from the
30 combustion of Indonesian peat. It is important to note that combustion conditions are an
31 important factor, influencing the physical and chemical properties of the resulting smoke. In
32 the case of peat, due to its chemical nature and high moisture content in typical environments,

1 the combustion process typically proceeds as a smoldering burn, characterized by low
2 combustion efficiency. As a result, peat smoke emissions are high in particulate organic
3 compounds and low in black carbon (BC) content. Despite the environmental significance of
4 wildfires and related chemistry, biomass burning emissions in SEA have received much less
5 attention than other tropical regions such as South America (e.g. Lelieveld et al., 2008; Martin
6 et al., 2010), Africa (e.g. Marticorena et al., 2011; Ferreira et al., 2010), and the Indian
7 subcontinent (e.g. Kulshrestha et al., 2001). More importantly, the quantitative contribution
8 of Indonesian biomass burning to the levels of PM in other countries within SEA in relation to
9 local air pollution sources remains poorly known.

10 We carried out ambient measurements of total suspended particulate matter (TSP) and its
11 chemical composition in Singapore during a haze episode in 2006. The objectives of this
12 research include: (1) to identify suitable organic molecular tracers for studying the influence
13 of peat and biomass fires on regional air quality, and (2) to quantitatively estimate the
14 fractional contribution of these fires to the atmospheric PM loading at a strategically located
15 downwind site based on chemical mass balance modeling.

16 **2 Experimental Procedures**

17 **2.1 Sampling Site and Sample Collection**

18 Aerosol samples were collected during two observation periods between September 2006 and
19 January 2007 at the Tropical Marine Science Institute on St. John's Island (SJI), Singapore
20 (1° 13' 10" N; 103° 50' 54" E). The sampling site was located in an open coastal area in the
21 southern part of Singapore, and was not influenced by local pollution sources in the main
22 island, but potentially by biomass burning emissions from Indonesia. Additional information
23 about the sampling location has been reported previously (Sundarambal et al., 2010). Total
24 suspended particle (TSP) samples were collected for 24 h three times per week, using a High-
25 Volume Air Sampler (HVAS, model 3800 AFC, HI-Q Environmental Products Company,
26 USA). The air flow of the HVAS was maintained at $68 \text{ m}^3 \text{ h}^{-1}$ by a mass flow controller. TSP
27 samples were collected on quartz fiber filters (QM-A, 20.3×25.4 cm, Whatman, UK). All
28 filters used in this study were inspected for defects under bright illumination and were
29 handled with a pair of stainless steel forceps. The TSP filters were folded in half lengthwise
30 after sampling, so that only surfaces with collected particulate matter were in contact, when
31 placed in the filter holder (glassine envelope). The filters were pre-equilibrated in a dry box at

1 constant temperature (22–25 °C) and relative humidity (30–35%) for at least 24 h before
2 weighing. The filters were weighed on a microbalance (model MC5, Sartorius AG,
3 Goettingen, Germany) with a sensitivity of 0.0001 mg. The balance was regularly checked
4 with NIST-traceable standard calibrated weights. The pre- and post-sampling weights were
5 used to obtain the particulate mass collected on the filters. All filters were stored in a
6 refrigerator at 4 °C until analysis. Two field blank filters, one during each observation period,
7 were collected by placing the filters in the sampler for a day without running the sampler and
8 processed together with filter samples.

9 **2.2 Chemical Analyses**

10 The TSP filter samples were subjected to a series of chemical analyses, including
11 quantification of the common inorganic ions, trace metals and various organic species. Blank
12 filters were processed according to the same procedures as the sample filters, and the blank
13 values were used to correct the ambient concentrations of the individual chemical species,
14 provided that the blank values were above the method detection limits.

15 The carbonaceous content of the individual samples was measured in the form of organic
16 carbon (OC) and elemental carbon (EC) by a thermo optical transmittance (TOT) method,
17 using a Sunset semi-continuous carbon analyzer (Sunset Labs, Model 4, Tigard, OR, USA) in
18 off-line mode, based on a modified NIOSH protocol (Birch and Cary, 1996). Polycyclic
19 aromatic hydrocarbons (PAHs) were determined by high-performance liquid chromatography
20 with mass spectrometric detection (HPLC-MS) upon solvent extraction of an additional
21 portion of the filter samples. A Waters Alliance 2695 HPLC system (Waters, Milford, MA,
22 USA), coupled with an API-3000 triple-quadrupole mass spectrometer (Applied
23 Biosystems/MDS SCIEX, Toronto, Canada), equipped with an APPI source, was employed
24 for the HPLC-MS analyses. Chromatographic separation of the individual PAHs was
25 achieved with a Waters PAH C18 (250 mm x 4.6 mm x 5 mm) column (Waters, Milford, MA,
26 USA), using a binary mobile phase system of water and acetonitrile (Zhang et al. (2010a).

27 Various carbohydrates, including anhydrosugars, polyols and monosaccharides were
28 separated, identified and quantified by high-performance anion exchange chromatography
29 coupled with pulsed amperometric detection (HPAEC-PAD). The detailed description of the
30 analytical method can be found elsewhere (Engling et al., 2006; Iinuma et al., 2009). Briefly,
31 a portion (2.2 cm²) of each quartz fiber filter was extracted with 2.0 mL of de-ionized water

1 (>18.2 MΩ resistivity) under ultrasonic agitation for 60 min. The aqueous extracts were
2 passed through syringe filters (0.45 μm, Pall Corporation, NY, USA) to remove insoluble
3 material. All extract solutions were stored at 4 °C until sample analysis. The carbohydrate
4 analyses were performed on a Dionex ICS-3000 system (Dionex, Sunnyvale, CA, USA),
5 consisting of a DP pump module, DC detector/chromatography compartment and
6 autosampler.

7 The separation of the individual carbohydrate species was carried out on a Dionex CarboPac
8 MA1 analytical column (250 x 4 mm) and guard column with an aqueous sodium hydroxide
9 (NaOH, 480 mM) eluent at a flow rate of 0.4 mL min⁻¹. Method detection limits for the
10 individual analytes ranged from 0.2 to 4.1 ng. Quantitative (nearly 100%) extraction of the
11 various carbohydrates was shown by analysis of spiked blank filters. None of the
12 carbohydrate species were detected in the blank samples.

13 The common inorganic ions were determined in the same aqueous extracts as those prepared
14 for HPAEC analysis, using a compact Dionex IC system (Model ICS-2000, Dionex Corp.
15 USA). It consists of an eluent generator (EG with CR-TC), isocratic pump, conductivity
16 detector, anion self-regenerating suppressor (ASRS Ultra II-4mm), rheodyne six-port
17 injection valve with 25 μL injection loop, and AS-40 autosampler. An AS-11-HC (4 mm)
18 analytical column in conjunction with an AG-11-HC guard column was used for anion
19 measurement. For cations, the cation ion-exchange column, CS-12 (4 mm), was used with
20 11.0 mM H₂SO₄ as the eluent. The peaks corresponding to different analytes in individual
21 standards were identified according to their retention times. The whole IC operation including
22 calibration and sample analysis was carried out using Chromeleon software.

23 Both water-soluble and insoluble metals were measured by a Perkin-Elmer Elan 6100 ICP-
24 MS (Perkin-Elmer Inc., USA). The ICP-MS was equipped with a cross flow nebulizer and a
25 quartz torch. The instrument with Pt sampler and skimmer cones was optimized daily and
26 operated as recommended by the manufacturer. The nebulizer gas flow rate was adjusted to
27 keep the CeO/Ce and BaO/Ba ratios less than 2%. A Gilson Miniplus 2 peristaltic pump was
28 used to remove the waste from the nebulizer. An eight point calibration (1, 2, 5, 10, 20, 40, 60
29 and 100 μg L⁻¹) was performed for 13 elements, and the regression coefficients for all 11
30 elements were better than 0.999. The coefficient of variations for the 13 elements was found
31 to be less than 5% (n = 10) at 5 μg L⁻¹ concentration level. Sample handling and preparation
32 were carried out in a laminar flow hood equipped with a HEPA (high-efficiency particulate

1 air) filter to prevent contamination. In all experiments, reagent blanks were taken separately.
2 The filter blanks and the final concentrations of metals in PM are reported after reagent blank
3 correction.

4 **2.3 Source Apportionment**

5 The chemical mass balance air quality receptor model CMB 8.2 (U.S. EPA, 2004) was used
6 to estimate the relative contribution of peat fires and other sources to ambient TSP measured
7 at the sampling site during the clear and hazy days. Source profiles of inorganic ions and trace
8 elements used in the model were obtained from SPECIATE 4.3 (SPECIATE, 2011). The
9 source profiles for petroleum refinery (Cooper et al., 1987) and diesel emissions (Vega et al.,
10 2004; Chow et al., 2002; Vega et al., 2000) obtained from the USEPA database are applicable
11 to Singapore, since the process of refining crude oil, the engineering practices adopted in oil
12 refineries, and the type of diesel used are similar to those in the studies included in the
13 database. In Singapore, most of the power plants continue to use oil as a fuel of choice, or as a
14 fuel in tandem with natural gas. Therefore, the source profiles of oil fired powerplants (Henry
15 and Knapp, 1980; Howes et al., 1983) included in the USEPA database were used in the
16 model. Source profiles for Indonesian peat fires were obtained from our previous study (See
17 et al., 2007) while that for ship emissions was obtained from Moldanova et al. (2009) and
18 Popovicheva et al. (2009). The chemical species used in the CMB model were potassium,
19 aluminum, cobalt, chromium, iron, manganese, lead, nickel, cadmium, titanium, vanadium,
20 arsenic, chloride, nitrate, sulfate, ammonium, nitrite, calcium, and sodium.

21 **2.4 Backward Trajectories**

22 The latest, updated Hybrid Single-Particle Lagrangian Integrated Trajectory (HYSPLIT)
23 model (Version 4.9) (Draxler and Rolph, 2003; Rolph, 2003), developed by the National
24 Oceanic and Atmospheric Administration (NOAA), was used to compute backward
25 trajectories for air samples taken during the haze period (October 2006) and clear days
26 (December 2006 – January 2007) in Singapore. Meteorological data were obtained from
27 National Centers for Environmental Prediction (NCEP) Global Data Assimilation System
28 (GDAS, global, 2006 – present). Kinematic 3D trajectories were used as they are reported to
29 provide an accurate description of the history of air masses in comparison with all of the other
30 approaches (isentropic, isobaric) (Stohl, 1998; Stohl and Seibert, 1998). Air mass backward
31 trajectories reaching 4 days (96 h) back were calculated using start times of 08:00 of each

1 sampling period. The calculation heights were fixed at 500, 1000 and 1500 m above the
2 ground. The atmospheric level, 500 m-agl, is frequently used (Lee et al., 2006; Erel et al.,
3 2007) and ensures that the trajectory ends in the planetary boundary layer (ABL) (Dvorská et
4 al., 2009).

5

6 **3 Results and Discussion**

7 **3.1 Trajectory Analysis and Temporal Variations in TSP Concentrations**

8 The representative air mass backward trajectories for both haze and clear days are presented
9 in Figure 1 with a haze map in October 2006. The haze period (October 2006) occurred at the
10 end of the Southwest Monsoon season, within which the climate in Singapore is governed
11 mainly by Southwest (SW) or Southeast (SE) winds. In this study, the predominant airflow
12 direction was consistent with the general wind pattern and direction from SE. In October
13 2006, Singapore had been affected by smoke haze transported from uncontrolled biomass and
14 peat burning due to "slash and burn" cultivation in Indonesia, where specifically dense
15 hotspots were observed in Kalimantan and south Sumatra (Figure 1(d) and 1(e)). During this
16 period, some of the air masses that arrived at Singapore originated from Sumatra (Figure 1(a))
17 while others were advected from Kalimantan with thick haze (Figure 1(b)); in addition, on
18 some sampling occasions, the air masses descended (from 1500 m altitude) or rose (from 500
19 m altitude) during the transport, indicating mixing of air masses from different layers on the
20 way to Singapore. During the clear period (December 2006 – January 2007), on the other
21 hand, the prevalent wind direction was northeasterly and the major air masses associated with
22 cleaner source regions originated from the South China Sea, characteristic of the Northeast
23 Monsoon season.

24 Figure 2 shows the daily average concentrations of total suspended particles (TSP) and
25 Pollution Standard Index (PSI) readings (for south Singapore) which are typically based on
26 PM₁₀ concentrations during haze episodes (Singapore National Environment Agency). It is
27 evident that the daily variation of PSI was well correlated with the variation in TSP
28 concentrations during both haze and clear days.

1 3.2 TSP mass balance

2 Average concentrations of EC, OC, PAHs, ions, and trace elements are presented in Table 1,
3 as observed during haze (October 2006) and clear (December 2006 – January 2007) days on
4 St. John's Island in Singapore. The mean concentration of TSP during the haze period was
5 higher by a factor of 3.5 as compared to clear days. Similarly, the levels of the speciated
6 chemical components of TSP were significantly higher during the haze period than those
7 during clear days. The relative contributions of various chemical species to the total mass
8 concentrations of TSP are shown in Figure 3. In this figure, "others" refers to the fraction of
9 TSP that was not identifiable (See et al., 2007), where

10

$$11 \text{ others} = \text{TSP} - (\text{EC} + \text{OC} + \text{inorganic ions} + \text{total metals}) \quad (1)$$

12

13 During haze days, a noticeably enhanced carbonaceous fraction (OC + EC) was observed
14 relative to clear days, accounting for 43% of TSP mass on average (versus 25% on clear days,
15 Figure 3). It is also worthwhile estimating the amount of organic matter (OM), accounting for
16 hetero atoms in the organic compounds such as oxygen and nitrogen, by applying suitable
17 OM/OC conversion factors. In the case of aerosols measured downwind of biomass burning
18 activities, OM/OC conversion factors of more than 1.4 have been suggested in literature for
19 organic aerosols in urban areas (e.g., White and Roberts, 1977; Turpin and Lim, 2001). When
20 applying a factor of 1.4 to the OC levels detected during the haze period, the content of OM in
21 TSP was found to be extremely high (50% on average), while it was only around 25% during
22 clear days. Aside from the substantial increase in the OC or OM fraction, the OC/EC ratio
23 was very high on haze days (15.5 versus 2.6) compared to clear days, which was higher than
24 the values from most studies reported in the literature (Lee et al., 2004; See et al., 2006; See et
25 al., 2007), suggesting a strong influence by biomass burning emissions. It should be noted that
26 high OC/EC ratios can also be due to contributions of organic species derived from biogenic
27 sources and secondary organic aerosol (SOA) in particular; however, the source strength of
28 such processes is expected to be fairly constant throughout the year at a tropical location such
29 as Singapore. Therefore, the substantially higher OC/EC ratios observed during the haze
30 period (compared to clear days) confirmed the significant contributions of carbonaceous
31 material derived from biomass burning.

1 3.3 Chemical Speciation of TSP

2 As presented in Table 1, the chemical speciation data include a large number of compounds
3 and elements, such as inorganic ions, trace metals, and various organic species
4 (carbohydrates, PAHs, and carboxylic acids). The substantial increase in the OC mass (by
5 about one order of magnitude) is obviously also reflected in elevated ambient concentrations
6 of individual organic species, specifically in terms of carbohydrates and carboxylic acids.
7 Anhydrosugars, such as levoglucosan, mannosan and galactosan, are well established
8 molecular tracers for biomass burning emissions, as they are solely derived from the thermal
9 decomposition of cellulose and hemicelluloses (Simoneit et al., 1999). Although recent
10 laboratory studies have revealed that these tracers may be degraded in the presence of
11 oxidants, such as OH radicals (Hennigan et al., 2010; Hoffmann et al., 2010), the
12 anhydrosugars can be used at least as qualitative indicators of biomass burning influence, but
13 also for quantitative estimates of biomass smoke contributions when transport distances are
14 shorter (Mochida et al., 2010), as was the case with the haze episode in this study. Hennigan
15 et al. (2010) reported that nearly 30-75% of levoglucosan would react within one day at
16 typical atmospheric OH levels while Hoffmann et al. (2010) estimated the half-life of
17 levoglucosan to be 12.7 – 83.2 h (0.5 – 3.5 days) at 90% relative humidity. From the
18 backward trajectory analysis, it was observed that the air masses from the source regions of
19 biomass burning reached Singapore within 2 – 3 days, which falls in the range of the half-life
20 of levoglucosan according to Hoffmann et al. (2010). We would like to emphasize here that
21 the ambient concentrations of anhydrosugars were up to two orders of magnitude higher
22 during the haze period relative to the clear days, indicating a massive impact of biomass
23 smoke emissions. In comparison to levoglucosan, the inorganic biomass tracer, potassium,
24 did not show much increase during the haze period. Levoglucosan is a source-specific tracer
25 for biomass burning, as it is generated during thermal breakdown of cellulose and
26 hemicellulose molecules. Potassium, on the other hand, can be derived from various emission
27 sources, including biomass burning, but also from cooking, vegetation, sea salt and soil,
28 especially in the coarse particle fraction. Since in this study we collected TSP, there is a
29 likely substantial contribution of potassium from other sources than biomass burning, i.e., sea
30 salt and soil, for which we did not apply a correction. Thus, the increase in potassium
31 concentrations during the haze episode is not expected to be as large as that for levoglucosan.

1 The inspection of the relative abundance of individual tracers revealed characteristic patterns,
2 which can be used to distinguish periods with smoke influence from the clear days.
3 Specifically, the ratio of the anhydrosugars levoglucosan and mannosan (LG/MN) can be
4 used for qualitative assessments of contributions from specific biomass species (Engling et
5 al., 2009, Fabbri et al., 2009). For instance, smoke particles derived from softwood
6 combustion are characterized by low LG/MN values of 3 – 5, while hard wood smoke
7 LG/MN ratios range from 15 to 25 and those for rice straw are even higher (~40). Unlike
8 during the clear days, the LG/MN ratios showed very little variation throughout the haze
9 episode, implying that only one specific biomass type or a steady mix of species was subject
10 to burning. The average LG/MN ratio (10) observed during the haze period is in excellent
11 agreement with chamber burn data from peat combustion (Iinuma et al., 2007) and ambient
12 observations from Malaysia (Simoneit et al., 2004), confirming the biomass burning source to
13 be the peat lands in Indonesia.

14 Two additional carbohydrates, the sugar alcohols arabitol and mannitol were quantified in the
15 TSP samples in Singapore. These polyols have recently been proposed as bioaerosol
16 indicators and molecular tracers for fungal spores in particular (Bauer et al., 2008a; Elbert et
17 al., 2007). Meanwhile, fungal spore contributions to the aerosol burden have been estimated
18 based on ambient concentrations of these tracers in a few recent studies (Bauer et al., 2008b;
19 Liang et al., 2013; Zhang et al., 2010b), revealing relatively large fungal spore content
20 specifically in coarse particles in tropical environments (Zhang et al., 2010b). The average
21 TSP content of these tracers during the haze period in Singapore was not significantly
22 different from that on clear days, indicating no apparent association of fungal spores with
23 biomass smoke particles derived from peat combustion.

24 Analogous to the carbohydrates, ambient PAH concentrations were substantially higher
25 during the haze period compared to clear days (by factors of 2 – 5). While PAHs are common
26 combustion products of all carbonaceous materials, including fossil fuels and biomass,
27 diagnostic ratios (DRs) of specific PAH species, although dependent on aerosol age, can be
28 used to constrain the predominant influence of emissions from certain types of combustion.
29 For instance, $\text{Ind}/(\text{Ind} + \text{B(g,h,i)P}) > 0.5$ indicates sources of solid fuel combustion (such as
30 coal, wood, grass, etc.) and ratios < 0.2 indicate petroleum origin, while the ratios between 0.2
31 – 0.5 indicate liquid fossil fuel combustion (vehicles and crude oil) (Yunker et al., 2002).
32 Similarly, $\text{Flt}/(\text{Flt} + \text{Pyr}) < 0.4$ is attributable to unburned petroleum, and a ratio at and above

1 0.5 indicates coal, grass and wood combustion origins, while ratios between 0.4 – 0.5 indicate
2 liquid fossil fuel combustion (Yunker et al., 2002). In this study the ratios of Ind/(Ind +
3 B(g,h,i)P) and Flt/(Flt + Pyr) were found to be in the range of 0.51 – 0.57 and 0.47 – 0.55
4 respectively during haze days, while they ranged from 0.41 – 0.47 and 0.40 – 0.45 during
5 clear days, suggesting that the dominant contributor is biomass burning during the hazy
6 period and local sources, such as ships and diesel engines, during clear days.

7 All of the common inorganic ions were present at higher levels in TSP during the haze period,
8 with enhancements of up to one order of magnitude in some cases (e.g., ammonium). Aside
9 from ammonium, potassium and nitrate also showed significantly higher ambient
10 concentrations in the haze versus clear periods, confirming the strong influence of biomass
11 burning activities on the local aerosol burden.

12 During the haze period, dicarboxylic acids (succinate, malonate, glutarate and oxalate) were
13 much more abundant than monocarboxylic acids (acetate and formate); oxalate was the
14 predominant dicarboxylic acid. Analogous to other organic species, the mass concentrations
15 of organic acids observed during the haze period were much higher than those obtained
16 during the clear days, suggesting that biomass burning is an important source of organic acids.
17 These acids could have been directly emitted from the biomass burning source (e.g., peat
18 fires) and/or formed in the atmosphere as a result of chemical conversions in the biomass
19 burning smoke plume (See et al., 2007). In case of the haze episode observed in this study, a
20 substantial portion of the dicarboxylic acids was likely derived from atmospheric oxidation,
21 i.e., aging processes, during transport of the smoke particles to the sampling site in light of the
22 favorable conditions. A substantial increase in water-soluble organics in smoke aerosols,
23 including organic acids, has also been reported from previous studies in SEA and elsewhere
24 (Agarwal et al., 2010; Graham et al., 2002; Gao et al., 2003; Kundu et al., 2010).

25 Both total and water-soluble metal concentrations were determined in the TSP samples
26 collected during the haze period and clear days in Singapore. The concentrations of most trace
27 metals were clearly enhanced during the haze period in relation to those on clear days. Cu, Al,
28 Fe, and Ti were the most abundant trace metals found in haze samples. The substantial
29 increase in the metallic species concentrations seems to be associated with re-mobilization of
30 local soils via the air medium upon high-temperature combustion of the peat/vegetation, as
31 these three metals are also the top three most abundant metals in the Earth's crust (Lide,
32 2005). As for copper, total Cu concentrations during the haze period were 30 times higher on

1 average compared to those on clear days. This phenomenon may be due to the strong fixation
2 ability of water-soluble copper compounds by rich humic and fulvic acids in the peat soils
3 (Mutert et al., 1999); upon combustion, this part of copper can be subsequently discharged
4 into the ambient air. The low solubility of this metal could be due to the chemical form in
5 which it exists such as metal oxides produced during high-temperature combustion. These
6 oxides are in general less soluble as compared to metal nitrates/sulphates. There is a copper
7 mine (Beutong mine, one of the largest copper mine in Indonesia) in the Sumatra region
8 where peat fires occurred during the haze episode which could be one of the additional
9 sources of copper apart from peat emissions. However, whether or not the soil in this local
10 area contains high concentration of copper needs to be further investigated. In addition, the
11 water-soluble fraction of most metals was larger by factors of 5 or more during haze versus
12 clear periods. This could be due to strong (more effective) attachment of organic matter to the
13 metals in the newly formed particles emitted during biomass burning processes. As mentioned
14 earlier, organic acids were much more enriched in the haze aerosols than those on clear days.
15 The presence of organic ligands can result in the formation of soluble heavy metal complexes
16 (See et al., 2007; Gaberell et al., 2003). Some of the metals such as Al have shown a
17 significantly higher proportion of the water soluble fraction (~30 times) during the haze
18 compared to clear days. In general, the water solubility of a metal depends on the chemical
19 form in which it is present. The higher water-soluble fraction of Al observed during the haze
20 might be due to the presence of more soluble Al chemical species in haze aerosols. However,
21 additional investigation is necessary to examine the chemical fractionation of particulate-
22 bound metals using a sequential extraction procedure. This investigation will account for (1)
23 soluble and exchangeable metals; (2) carbonates, oxides, and reducible metals; (3) metals
24 bound to organic matter, oxidizable and sulfidic metals; and (4) residual metals, while the
25 current study quantified only the soluble and exchangeable metals. The outcome of the
26 investigation will be published elsewhere.

27 **3.4 Source Apportionment**

28 Contributions of particulate matter from the different emission sources can be assessed by a
29 variety of methods. As a large number of chemical species was quantified in the TSP samples
30 as part of this study, the chemical mass balance (CMB) modeling approach was applied here,
31 providing reliable estimates of the individual sources (Watson et al., 2002). Figure 4 shows
32 the relative source contributions to ambient TSP at the sampling site during haze and clear

1 days. The CMB performance measure indices are all within the target ranges recommended
2 by the U.S. EPA (r^2 : 0.8 – 1; χ^2 : 0 – 4; % mass: 80 – 120%; *degrees of freedom* > 5, *t-stat* > 2
3 for all the contribution sources). The major pollution sources that affected the air quality at
4 the sampling site were identified to be peat fires, diesel exhaust, petroleum refineries, ships,
5 and power plants during the sampling period. From Figure 4, it can be seen that diesel
6 exhaust was the dominating source on a typical clear day. The sampling site is remotely
7 located away from the main land of Singapore, so that the power supply for the entire island is
8 provided by diesel generators which could be the reason for the diesel exhaust to be a major
9 contributor to PM (~62%) at the sampling site. Ships pass by the island to reach the Singapore
10 port also contributed to TSP (27%), followed by a nearby petroleum refinery (~6%) and
11 power plants (~5%). During the haze episode, peat fires in Indonesia were the predominant
12 source (~76%) of TSP followed by diesel exhaust (19%), and ship emissions (~5%).

13

14 **4 Conclusions**

15 In this study, the impact of a major smoke haze episode, caused by biomass burning and peat
16 fires in Indonesia, on regional air quality in Singapore was investigated during autumn of
17 2006. It was observed that the TSP concentration increased by a factor of 3.5 with the major
18 contribution from the carbonaceous fraction (~43%) during the haze episode when compared
19 to clear days (carbonaceous fraction ~25%). Very high OC/EC ratios (15.5), concentrations of
20 organic acids, PAHs, and trace metals, especially Cu, Al, Fe and Ti, in TSP observed during
21 the haze period reveal the impact of Indonesian biomass burning activities on regional air
22 quality in SEA. Enhancement in the concentrations of organic molecular tracers of biomass
23 burning, as well as a diagnostic ratio of LG/MN of 10 observed during the haze period is in
24 excellent agreement with the laboratory data from controlled peat combustion, confirming the
25 biomass burning source to be the peat lands in Indonesia. Similarly, the diagnostic ratios of
26 PAH species indicate that the TSP during the haze period was predominantly influenced by
27 biomass burning while it was mainly influenced by local fossil fuel combustion and diesel
28 engine emissions during clear days. Source apportionment using the CMB model indicated
29 that on clear days, TSP at the sampling location was mainly derived from exhaust (~62%)
30 from local diesel generators which is the only source of power on the Island, followed by
31 ships (27%), petroleum refinery (~6%) and power plants (~5%). On the other hand, during the
32 haze period nearly 76% of the TSP was derived from peat fires in Indonesia, reducing the

1 source contribution of diesel exhaust and ship emissions at the sampling site to 19% and 5%,
2 respectively.

3 **Acknowledgements**

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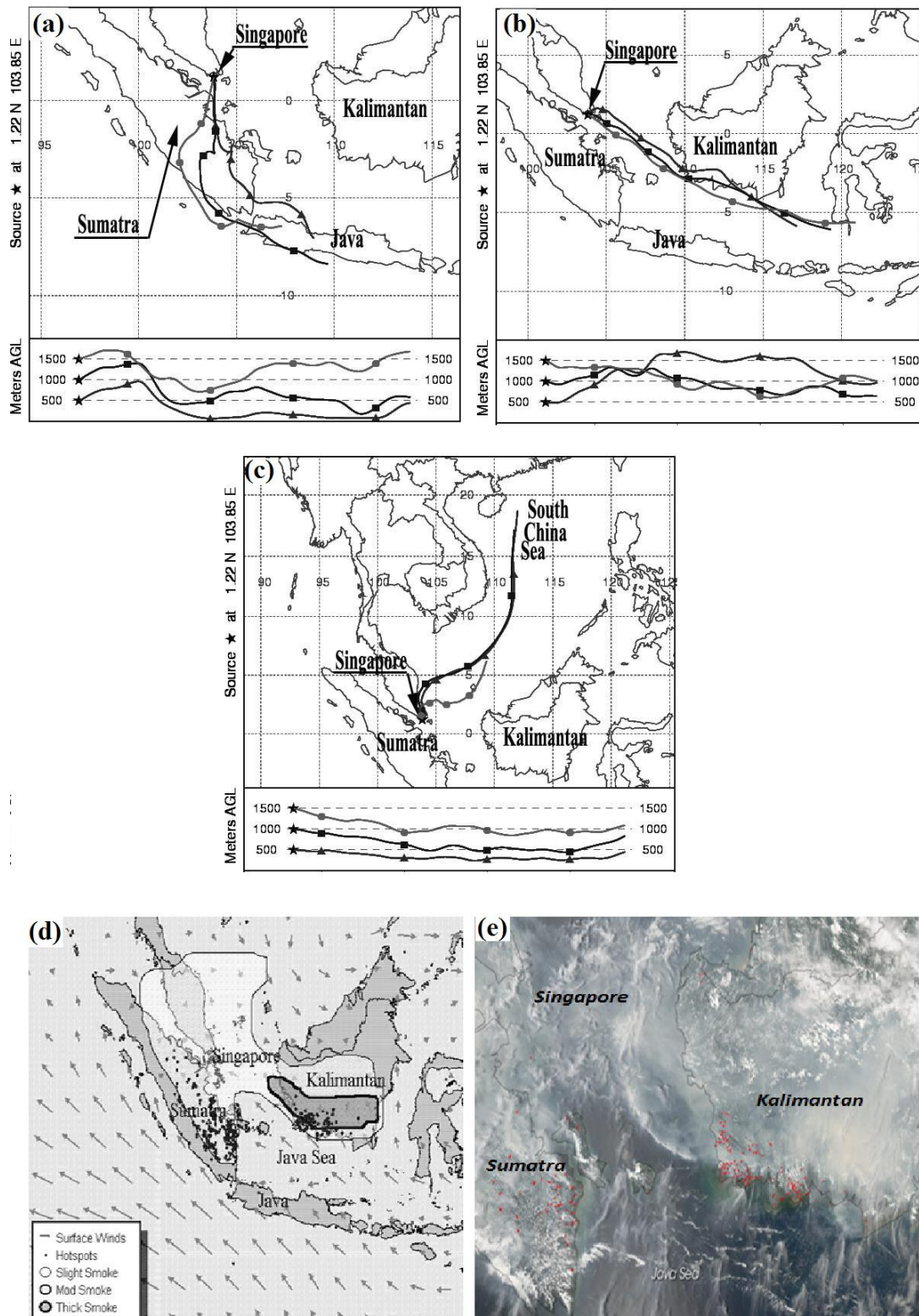
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1 **Table 1.** Mass concentrations of TSP and associated chemical components during haze and
 2 clear periods at St. John Island in Singapore

	Haze period Concentration (average \pm standard deviation)	Clear days
	TSP ($\mu\text{g m}^{-3}$)	
TSP	94.1 \pm 33.4	24.0 \pm 5.2
	Carbonaceous components ($\mu\text{g m}^{-3}$)	
EC	2.3 \pm 1.5	1.7 \pm 0.50
OC	33.6 \pm 18.9	4.2 \pm 0.53
OC/EC	15.5	2.6
	Polycyclic aromatic hydrocarbons (PAHs) (ng m^{-3})	
Benzo	0.11 \pm 0.05	0.08 \pm 0.06
Naph	N.A	N.A
Acy	0.11 \pm 0.04	0.04 \pm 0.02
Ace	0.05 \pm 0.04	0.02 \pm 0.01
Flu	0.06 \pm 0.08	0.02 \pm 0.03
Phe	0.62 \pm 0.33	0.08 \pm 0.05
Ant	0.02 \pm 0.03	N.A
Flt	0.54 \pm 0.18	0.12 \pm 0.08
Pyr	0.59 \pm 0.19	0.15 \pm 0.13
B(b)N	0.13 \pm 0.01	0.02 \pm 0.01
Retene	0.22 \pm 0.06	0.13 \pm 0.05
C(c,d)P	0.58 \pm 0.19	0.20 \pm 0.09
B(a)A	0.11 \pm 0.03	0.04 \pm 0.02
Chr	0.05 \pm 0.03	0.04 \pm 0.03
B(e)P	0.47 \pm 0.13	0.17 \pm 0.10
B(b)F	0.55 \pm 0.13	0.19 \pm 0.12
B(k)F	0.32 \pm 0.08	0.12 \pm 0.09
B(a)P	0.43 \pm 0.10	0.14 \pm 0.09
D(a,h)A	0.17 \pm 0.09	0.08 \pm 0.02
B(g,h,i)P	0.80 \pm 0.32	0.17 \pm 0.14
Ind	0.63 \pm 0.23	0.19 \pm 0.08
Cor	0.34 \pm 0.10	0.09 \pm 0.07
	Polar organic compounds (ng m^{-3})	
Levoglucosan	1150.9 \pm 917.2	15.3 \pm 5.7
Mannosan	115.5 \pm 101.5	4.6 \pm 2.9
Galactosan	54.3 \pm 47.7	N.A.
Arabitol	22.5 \pm 10.6	17.3 \pm 3.5
Mannitol	35.4 \pm 5.3	50.9 \pm 5.5
	Water-soluble ions ($\mu\text{g m}^{-3}$)	
F ⁻	0.18 \pm 0.08	0.01 \pm 0.003
Cl ⁻	1.0 \pm 0.48	0.38 \pm 0.25
NO ₂ ⁻	0.38 \pm 0.21	0.01 \pm 0.003
SO ₄ ²⁻	19.6 \pm 8.5	3.0 \pm 0.74
NO ₃ ⁻	6.4 \pm 2.2	0.76 \pm 0.36
PO ₄ ³⁻	0.46 \pm 0.23	0.09 \pm 0.02
Na ⁺	3.7 \pm 0.64	1.1 \pm 0.40
NH ₄ ⁺	3.0 \pm 2.2	0.07 \pm 0.04

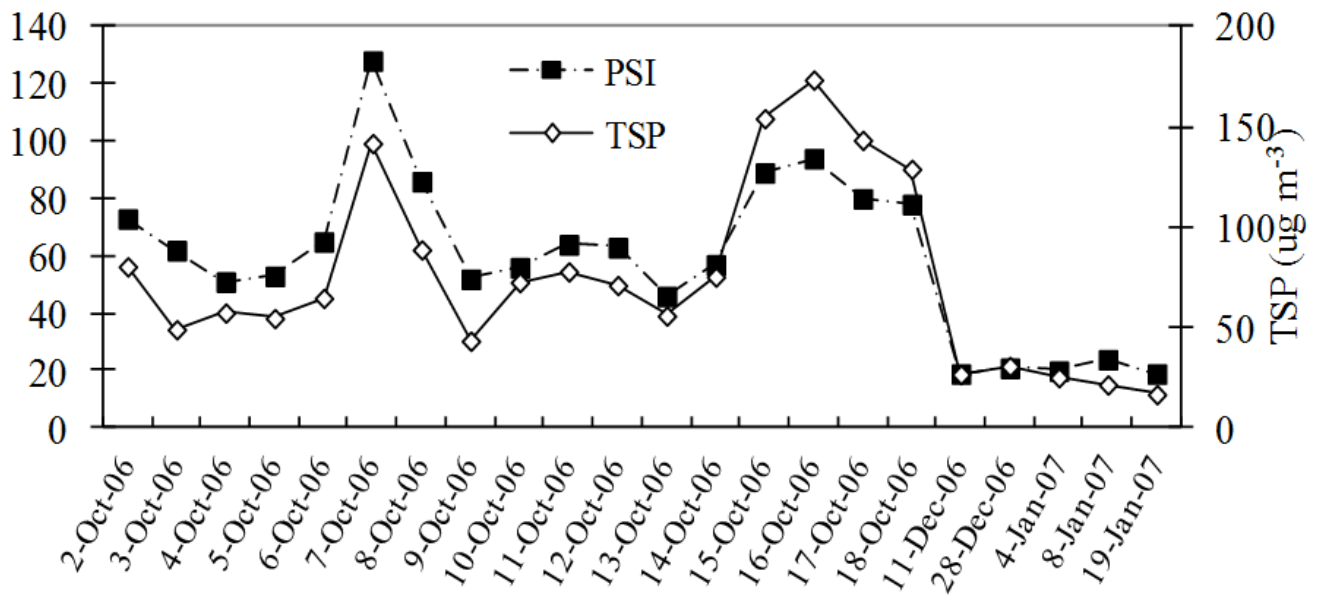
K ⁺	1.6 ± 0.40	0.42 ± 0.07
Ca ²⁺	1.7 ± 1.4	1.1 ± 0.21
Acetate	0.59 ± 0.32	0.025 ± 0.015
Formate	0.49 ± 0.27	0.005 ± 0.009
Methylsulfonate	0.15 ± 0.10	N.A.
Succinic	0.40 ± 0.25	0.003 ± 0.003
Glutamate	0.51 ± 0.33	0.005 ± 0.005
Malonic acid	2.3 ± 1.6	0.008 ± 0.007
Oxalate	5.0 ± 3.1	0.23 ± 0.13
Total metals (water-soluble metals) (ng m⁻³)		
Al	1658.1 ± 487.7 (113.9 ± 28.3)	1200.0 ± 697.9 (4.9 ± 1.8)
Co	1.4 ± 1.1 (0.48 ± 0.42)	0.77 ± 0.16 (0.17 ± 0.08)
Cr	19.2 ± 10.4 (4.9 ± 3.0)	8.9 ± 4.0 (1.4 ± 0.10)
Cu	1129.0 ± 734.2 (13.9 ± 11.3)	38.7 ± 19.1 (3.14 ± 1.26)
Fe	1646.9 ± 1258.5 (379.5 ± 159.4)	1018.8 ± 389.9 (54.0 ± 10.9)
Mn	27.2 ± 12.4 (17.3 ± 10.9)	18.4 ± 3.6 (5.7 ± 2.7)
Pb	27.1 ± 13.6 (14.5 ± 6.3)	18.2 ± 3.7 (1.3 ± 0.46)
Zn	68.1 ± 39.8	96.2 ± 30.7
Cd	0.99 ± 0.50	0.66 ± 0.29
Ni	24.2 ± 19.2	15.7 ± 5.2
Ti	626.1 ± 472.3	534.3 ± 248.7
V	71.1 ± 42.2	47.1 ± 10.9
As	5.6 ± 3.1	2.1 ± 0.86

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 2 **Figure 1.** Air mass backward trajectories during haze and clear days in Singapore on
 3 (a) 7th Oct 2006; (b) 15th Oct 2006; (c) 8th Jan 2007; as well as (d) haze and fire hotspot map
 4 (e) Satellite image showing smoke due biomass burning in Sumatra and Kalimantan envelope
 5 the region.

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3 **Figure 2.** Temporal changes in total suspended particulate matter (TSP) concentrations and
4 Pollutants Standard Index (PSI) during haze (Oct 2006) and clear (Dec 2006 – Jan 2007) days
5 in St. John's island of Singapore

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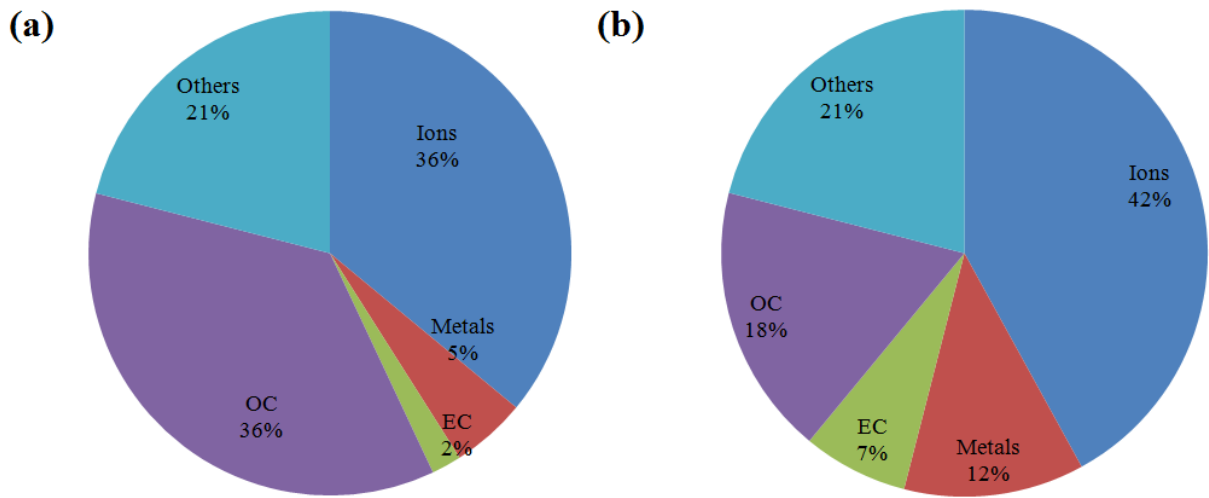
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4 **Figure 3.** Mass balance of TSP sampled during haze (3a) and clear days (3b) at St. John's
5 Island in Singapore

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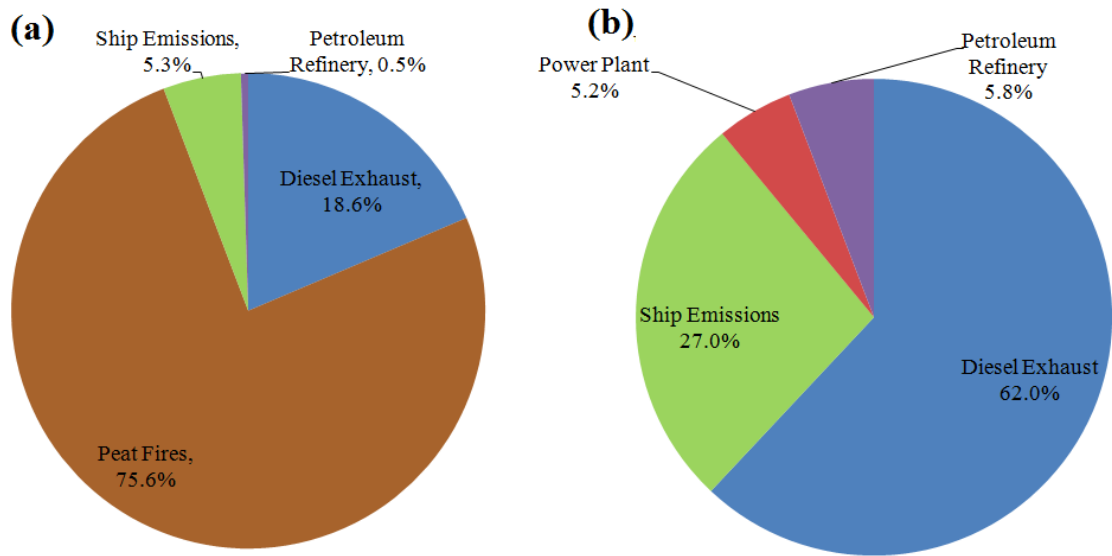
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($\mu\text{g}\cdot\text{m}^{-3}$)	Haze	Clear days
Ship Emissions	4.46 ± 1.77	6.48 ± 1.4
Petroleum Refinery	0.42 ± 0.17	1.39 ± 0.3
Power Plant	-	1.25 ± 0.27
Diesel Exhaust	15.64 ± 6.21	14.88 ± 3.22
Peat Fires	63.58 ± 25.25	-

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4 **Figure 4.** Source contributions to TSP during haze (4a) and clear days (4b) at St. Johns Island
 5 in Singapore

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