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Variations in global methane sources and sinks during 1910–2010

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Abstract

Atmospheric methane (CH₄) increased from ~900 ppb (parts per billion, or nanomoles per mole of dry air) in 1900 to ~1800 ppb during the 2000s at a rate unprecedented in any observational records. However, the causes of the CH₄ increase are poorly understood. Here we use initial emissions from bottom-up inventories for anthropogenic sources, emissions from wetlands and rice paddies simulated by a terrestrial biogeochemical model, and an atmospheric general circulation model (AGCM)-based chemistry-transport model (i.e. ACTM) to simulate atmospheric CH₄ concentrations for 1910 to 2010. The ACTM simulations are compared with the CH₄ concentration records

- ¹⁰ reconstructed from Antarctic and Arctic ice cores and firn air samples, and from direct measurements since the 1980s at multiple sites around the globe. The differences between ACTM simulations and observed CH₄ concentrations are minimized to optimize the global total emissions using a mass balance calculation. During 1910–2010, the global total CH₄ emission increased from ~290 Tg yr⁻¹ to ~580 Tg yr⁻¹. Compared to
- ¹⁵ optimized emission the bottom-up emission dataset underestimates the rate of change of global total CH₄ emissions by ~30 % during the high growth period of 1940–1990, while it overestimates by ~380 % during a low growth period of 1990–2010. Further, using the CH₄ stable carbon isotopic data (δ^{13} C), we attribute the emission increase during 1940–1990 primarily to enhancement of biomass burning. The total lifetime of
- ²⁰ CH₄ shortened from 9.4 yr during 1910–1919 to 9 yr during 2000–2009 by the combined effect of increasing abundance of atomic chlorine radicals (Cl) and increases in average air temperature. We show that changes of CH₄ loss rate due to increased tropospheric air temperature and CH₄ loss due to Cl in the stratosphere are important sources of uncertainty to more accurately estimate global CH₄ budget from δ^{13} C ²⁵ observations.



1 Introduction

Methane (CH₄), the second most important anthropogenic greenhouse gas, plays an important role in the chemical and radiative balances in the Earth's atmosphere. Due to its main removal by reaction with hydroxyl (OH) radical, which is a major atmospheric oxidant, CH₄ actively participates in tropospheric air-pollution chemistry. In the stratosphere, CH_4 is the predominant source of water vapour (Jones and Pyle, 1984) and is the primary sink for chlorine radicals (Cicerone and Oremland, 1988). The global warming potential (GWP) of CH_4 is 28 over a time horizon of 100 years (Myhre et al., 2013). Methane is released into the atmosphere from both anthropogenic and natural sources (Patra et al., 2011; Kirschke et al., 2013; and references therein). The 10 present-day concentration of CH_4 (~1800 ppb) is unprecedented over at least the past 800 000 years (Loulergue et al., 2008). Measurements of CH₄ in air bubbles trapped in Antarctic and Greenland ice sheets suggest that after a very slow increase (average rate of ~0.04 ppb yr⁻¹) during preindustrial times (1–1750 AD), CH₄ concentrations have increased rapidly (average rate of $\sim 4 \text{ ppb yr}^{-1}$ during 1750–1990 AD) until the 15

- ¹⁵ have increased rapidly (average rate of ~4 ppb yr duning 1750–1990 AD) until the early 1990s (Rasmussen and Khalil, 1984; Etheridge et al., 1998; Nakazawa et al., 1993; MacFarling Meure et al., 2006). The rapid rise of CH_4 levels paused in the 1990s (Dlugokencky et al., 2003) and resumed strong growth again starting in 2007 (Rigby et al., 2008; Dlugokencky et al., 2009).
- ²⁰ The factors responsible for the observed changes of CH₄ concentration on historic (century to millennium) time scales as well as recent decades remain unclear in relation with CH₄ emissions from natural and anthropogenic activities, and climate change. The stable carbon isotopic ratio (δ^{13} C) of CH₄ is useful for separating different CH₄ sources, since individual sources have characteristic δ^{13} C signatures (e.g., Quay et al., 1999). ²⁵ Ferretti et al. (2005) reported the significant δ^{13} C depletion during the last millennium (circa 1000 to 1700 AD) and ascribed it to changes in biomass burning emissions
- probably influenced by both human activities and natural climate change. Houweling et al. (2008) suggested that the δ^{13} C variation could be alternatively explained by in-



creasing anthropogenic CH₄ emissions over the past millennium, in combination with a shift from non-fossil to fossil sources during the industrial period. Analyzing a Greenland ice core for δ^{13} C and applying a two-box model, Sapart et al. (2012) attributed centennial-scale variations in δ^{13} C between 100 BC and 1600 AD to changes in py-

- ⁵ rogenic and biogenic sources driven by both natural climate variability and changes in human population and land use. It has been also suggested that changes in CH₄ emission from natural wetlands were the dominant process, in the absence of anthropogenic sources, for CH₄ variations during the Last Glacial Maximum (18 000 year before present, BP) to preindustrial Holocene (9000–200 year BP) (e.g., Chappellaz et al.,
- ¹⁰ 1993). All these studies, with competing theories, are focused on the preindustrial era or before. They are not tested for the latest century, which is a more complex situation due to the combination of natural and anthropogenic sources of CH₄, and is the focus of this study.
- The period between 1900 and the 1970s has no precise direct observations of CH_4 ¹⁵ concentration and $\delta^{13}C$. Ice core studies have succeeded in reconstructing the CH_4 concentration and $\delta^{13}C$ history until the mid-20th century, and using firn air, the measurement record can be extended back to the mid-20th century (Etheridge et al., 1998; Ferretti et al., 2005; Francey et al., 1999; MacFarling Meure et al., 2006; Sapart et al., 2013). Both the ice core and firn air measurements can have relatively coarse time
- ²⁰ resolution (the air age spread $(1\sigma) \sim 5$ years or longer time, Etheridge et al., 1998; Trudinger et al., 2002) and are mainly limited to polar regions, except for a recent ice core measurement from the Himalayan glacier which however has potential contaminations for CH₄ in the air bubbles (Hou et al., 2013). Direct measurements of concentration at high precision were initiated in the late 1970s (e.g., Rasmussen and Khalil,
- 1984), and the measurement networks have grown to more than 100 sites worldwide (e.g., Cunnold et al., 2002; Dlugokencky et al., 2011). Here we use combined emissions from bottom-up inventories, emissions from wetlands and rice paddies simulated by a terrestrial biogeochemical model, and a 3-dimensional chemistry-transport model to simulate atmospheric CH₄ for the past 100 years (1910–2010). With limited obser-



vations, attempts are made to estimate the optimized global total CH_4 emissions by using mass balance calculations and to infer the possible variations in biomass burning emissions by introducing an additional constraint based on $\delta^{13}C$, those being useful for developing Representative Concentration Pathways (RCPs) for climate modeling with policy implications, e.g., the Intergovernmental Panel on Climate Change (IPCC) (Myhre et al., 2013; Lamarque et al., 2010). Section 2 describes the materials and methods, followed by results and discussion in Sect. 3, and finally, conclusions are given in Sect. 4.

2 Materials and methods

10 2.1 Model description

The Center for Climate System Research/National Institute for Environmental Studies/Frontier Research Center for Global Change (CCSR/NIES/FRCGC) Atmospheric General Circulation Model (AGCM)-based Chemistry Transport Model (i.e., JAM-STEC's ACTM) is used for simulating CH₄ in the atmosphere (Patra et al., 2009).
The model resolutions are T42 spectral truncations (~2.8° × 2.8°) in horizontal and 67 sigma-pressure vertical layers (surface to ~90 km). In the control case, the ACTM meteorology is driven by interannually varying (IAV) sea-surface temperature (SST) and sea ice at monthly mean time intervals, based on the gridded analysis by the Hadley Centre (Rayner et al., 2003). We have checked that the AGCM simulated zonal-mean horizontal winds and temperatures in the troposphere are within ±5 m s⁻¹ and ±1 K, respectively, when compared with ACTM simulations nudged to the Japan Meteorological Agency (JMA) 25 year reanalysis (JRA-25) (Onogi et al., 2007). These differences

in meteorology do not appreciably affect our long-term simulation results, because only about $5 \text{ Tg CH}_4 \text{ yr}^{-1}$ higher loss is simulated in the ACTM driven only by SST compared to when the ACTM is nudged to JRA-25. The ACTM also realistically represents inter-



hemispheric transport, stratosphere-troposphere exchange, and SST driven climate

variations such as the El Niño Southern Oscillation. Annual mean concentrations are used in this analysis, although the model integration time step is about 20 min.

2.2 CH₄ emissions

We constructed global total CH₄ emissions by combining: (1) the interannually varying annual mean anthropogenic emissions from the Emission Database for Global Atmospheric Research (EDGAR) – Hundred Year Database for Integrated Environmental Assessments (HYDE; version 1.4) (van Aardenne et al., 2001) and EDGAR 3.2 (Olivier and Berdowski, 2001), (2) interannually and seasonally varying emissions from rice paddies and wetland simulated by the Vegetation Integrative Simulator for Trace Gases (VISIT) terrestrial ecosystem model (Ito and Inatomi, 2012), and (3) natural emissions, such as those from biomass burning, termites based on the GISS inventory (Fung et al., 1991); emissions due to oceanic exchange near the coastal region

- (Lambert and Schmidt, 1993); and mud volcano emissions (Etiope and Milkov, 2004) as the major emission components (Fig. 1a). Scaling factors for emissions due to ter-
- ¹⁵ mites, oceanic exchange, mud volcano, biomass burning, rice paddies and wetlands are 0.77, 0.40, 1.00, 0.4, 0.95, and 0.85. Scaling factors are set to simulate the CH₄ growth rate approximately for the first decade 1901–1910, and are in close agreement with Patra et al. (2011) for the period 1990–2008. For 1970–2000, interannually varying anthropogenic CH₄ emissions from EDGAR 3.2 and EDGAR 3.2FT data are used
- ²⁰ and the data have been extended for 1901–1970 following the sector-wise trends recommended in EDGAR HYDE. For 2001–2010, the EDGAR 3.2FT emissions map for 2000 is used. EDGAR 3.2 and EDGAR 3.2FT emissions for biomass burning and rice sectors (SAV, DEF, AGR, AGL sectors) are excluded from the initial CH₄ emissions (E_{ini}) , since they are given from different datasets as described above. The combina-²⁵ tion of different categories (Table 1) and interpolation/extrapolation of EDGAR dataset

are similar to that used by Patra et al. (2011).

All of the 4 main categories of anthropogenic emissions (oil and gas, coal, animals and landfills) have increased steadily in the last 110 years; according to the EDGAR



inventories (HYDE, v3.2 and v3.2FT), oil and gas emissions increased from 12 to 78 Tg yr^{-1} , coal from 9 to 33 Tg yr^{-1} , animals from 30 to 89 Tg yr^{-1} , and landfills from 6 to 59 Tg yr^{-1} for 1901–2010. The highest increases in these emissions took place during 1940–1990. Animal emissions were the dominant contributor to this rapid to-

- tal increase for 1940–1960, while oil and gas controlled the increasing trend for the next 3 decades (1960–1990). There was a decrease in oil and gas emissions during the early 1990s from the former Soviet Union (Fig. 1a). We kept annual total biomass burning emissions constant over time (49.7 Tg yr⁻¹; monthly varying GISS dataset) because no consensus on the amplitude and trends has been achieved in literature (see
- Sect. 3.4). The VISIT model simulated CH₄ emissions from rice paddies and wetlands using a scheme by Cao et al. (1998). The simulated total emissions from rice paddies increased from 18 to 37 Tg yr⁻¹ and that from wetlands varied from 141 to 159 Tg yr⁻¹ for the period 1901–2010. The trends in emissions from rice paddies are mainly due to the increase in rice cropping, and that for the wetlands are due to warming of the Earth's surface and inundation levels due to rainfall variations (Ito and Inatomi, 2012;
- Patra et al., 2013). The fraction of paddy field was derived from the cropland fraction in land use data (Hurtt et al., 2006).

2.3 CH₄ loss processes

The primary loss process for atmospheric CH_4 (~90%) is oxidation by hydroxyl radicals (OH), mostly in the troposphere. The remaining ~10% of the sinks include consumption by methanotrophic bacteria in soils, and reactions with chlorine radicals (CI) and electronically-excited atomic oxygen (O(¹D)) in the stratosphere. These reactions are modeled in ACTM with reaction rates taken from Sander et al. (2006). The climatological monthly mean tropospheric OH concentrations are taken from Spivakovsky

et al. (2000), and stratospheric OH and Cl concentrations are obtained from a stratospheric chemistry simulation by the CCSR/NIES AGCM (Takigawa et al., 1999). Concentration of O(¹D) is calculated online in ACTM using climatological ozone distribution. Trends in Cl concentration over the period of our simulation are introduced using the



estimated changes in Effective Equivalent tropospheric CI for the period 1992–2012 (Montzka et al., 1999; updates on the NOAA/ESRL website) and by simple extrapolation to 1901 following annual fluorocarbon production report of Alternative Fluorocarbons Environmental Acceptability Study (AFEAS) (www.afeas.org). This method ig-

- ⁵ nores the changes in CI vertical distribution due to the differences in CI production rate from different species, which is altitude dependent. A delay of about 5 years between emissions of the halocarbons at Earth's surface and CI release in the stratosphere is used based on average "age" of stratospheric air in ACTM. No trends in OH are considered in this study because of lack of consensus between models, e.g., 6 out 14
 ¹⁰ models show increases in OH concentrations in the period of 1850–1980, even though
- the models used consistent set of anthropogenic emissions since the preindustrial era (Naik et al., 2013).

The time series of CH₄ chemical loss as calculated with ACTM simulation with E_{ini} for 1901–2010 is shown in Fig. 1b. Loss due to OH is the dominant contributor (244–466 Tgyr⁻¹), followed by soil (18–36 Tgyr⁻¹ as simulated by the VISIT model), O(¹D) (4.6–8 Tgyr⁻¹), and Cl (1.4–15.6 Tgyr⁻¹) over 1901–2010. Consideration of trends in Cl concentration in the ACTM results in a dramatic increase in CH₄ loss by Cl since the 1950s (Fig. 1b). We show later that the trends in stratospheric feedback of ¹³C-enriched CH₄ cause a large imbalance in the tropospheric budget of the emission categories.

2.4 CH₄ observations: ice core/firn air data and direct measurements

The observed CH_4 concentrations in the Arctic and Antarctic regions were used for evaluating the ACTM simulations for two different emission scenarios. Two different types of data were used in the present study:

Ice core/Firn air measurements of CH₄ concentration – we used the Law Dome ice core records (DSS, DE08 and DE08-2) (Etheridge et al., 1998; Ferretti et al., 2005; MacFarling Meure et al., 2006) (1894–1980), and the firn records from



DE08–2 (Etheridge et al., 1998) and DSSW20K (MacFarling Meure et al., 2006 and references therein) (1944–1995). The NEEM firn data¹ (1949–1996) are based on Buizert et al. (2012) and the NGRIP firn air data (1951–2001) are obtained from Tohoku University (Umezawa et al., unpublished data).

Direct atmospheric measurements from sampling networks or air archive – direct measurements of atmospheric CH₄ from two representative sites are used: (1) the Southern Hemisphere high latitude (Cape Grim; CGO; 40.7° S, 144.7° E) and (2) the Northern Hemisphere high latitude (Summit; SUM; 72.6° N, 38.5° W). We used station data whenever available from the NOAA Earth System Research Laboratory (NOAA/ESRL) (Dlugokencky et al., 1994); and from the Global Atmospheric Gases Experiment (GAGE) and the Advanced Global Atmospheric Gases Experiment (AGAGE) program (Prinn et al., 2000; Cunnold et al., 2002). Annual mean values are calculated from flask-based or continuous measurement data for these sites available on the World Data Centre for Greenhouse Gases (WDCGG) website (http://ds.data.jma.go.jp/gmd/wdcgg). The archived air samples (1978–1995) from CGO are taken from Etheridge et al. (1998).

We applied scaling factors following Dlugokencky et al. (2005), for harmonizing all data onto the Tohoku University (TU) scale (Aoki et al., 1992; Umezawa et al., 2014). It is noted that the ice core and firn air measurements have time resolution of ~5 years or more (the air age spread (1 σ), Etheridge et al., 1998; Trudinger et al., 2002), while the direct measurements are available at either weekly or sub-hourly time intervals. The measurement accuracy for air samples is 5 ppb or better (Etheridge et al., 1998; Umezawa et al., 2014; Dlugokencky et al., 1994; Cunnold et al., 2002). To prepare

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¹We apply a spline interpolation technique to calculate the age of CH_4 using (1) depth vs. CH_4 concentration and (2) CH_4 concentration time series data available in Buizert et al. (2012). This could be an approximate method applicable only for locked-in zone where CH_4 concentration monotonically decreases with depth and is less affected by diffusion of atmospheric CH_4 . See Buizert et al. (2012) for more accurate calculation.



the time series of observed CH_4 concentration at uniform annual intervals, first the annual average of all observations were estimated and then the annual average data at uneven time intervals (due to data gaps) were interpolated using spline (order 4) fitting. The spline fitting is adopted using the SPLINEFIT program by Jonas Lundgren (www.mathworks.com/matlabcentral/fileexchange/13812-splinefit). The smoothing effect is controlled by the judicious selection of breaks or knots at different years. The knots are needed for finding optimal fit through data at irregular time intervals and changing concentration growth rate.

2.5 Calculation of optimized CH₄ emission: mass balance

¹⁰ The global mass balance equation for total emissions (*E*), loss (*L*) and burden (*B*) of CH_4 in the atmosphere is given by:

$$\frac{\mathrm{d}B}{\mathrm{d}t} = E - L \tag{1}$$

A conversion factor H can be calculated from the ratio of modeled B and an average CH₄ concentration at the lower-most model level ([CH₄]) as follows:

¹⁵ $B = H \times [CH_4]$

This gives the value of 2.8696 ± 0.0031 (average \pm interannual variation) Tg CH₄ ppb⁻¹ for the conversion factor *H* for the period 1910–2010. This value of *H* is about 3.7 % higher than common value of 2.767 calculated by Fung et al. (1991). This is mainly because in the present calculation we have used smaller than global mean CH₄ concentration from Antarctica. *L* is calculated by summing up loss at all ACTM grids. Because our knowledge for developing accurate initial emissions (*E*_{ini}) is incomplete, the simulated [CH₄] time series is likely to deviate from the observation. The "correction" term ΔE to initial emission time series is calculated by applying



(2)

Eq. (1) on the difference (Δ [CH₄]) between observed and simulated [CH₄]:

$$H\frac{d([CH_4])}{dt} + L = \Delta E$$

where ΔL is the difference in loss terms which is calculated using Δ [CH₄] and the ratio $L/[CH_4]$ as a conversion factor (0.2907 ± 0.0055) from model simulation for individual ⁵ years. The optimized emissions (E_{opt}) are given by

 $E_{\rm opt} = E_{\rm ini} + \Delta E$

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2.6 Isotope analysis

The contribution of different emission categories (E_{ini_i}) to the bottom-up emissions as used in this study is known (Fig. 1a and Table 1). However, the relative contribution of different emission categories to the emission correction ΔE is unknown. We thus introduce an additional constraint based on δ^{13} C for distributing ΔE between two hypothetical source categories with lighter (13 C-depleted) and heavier (13 C-enriched) isotopic signatures.

The isotopic ratio of ¹³C to ¹²C in CH₄ (δ^{13} C) is defined as:

¹⁵
$$\delta^{13}C = \left(\frac{R_{\text{sample}}}{R_{\text{std}}} - 1\right) \times 1000$$
(5)
$$R = {}^{13}C/{}^{12}C$$
(6)

where R_{sample} is the isotopic molar ratio in the methane sample, and R_{std} is the corresponding ratio in the international isotope standard (Vienna Peedee Belemnite (VPDB)) with an accepted value of 0.0112372 (Craig, 1957). δ^{13} C is expressed in "per mil" (‰) notation.

We use a one-box model (e.g., Lassey et al., 2000) to estimate the isotopic signature for global emission $E(\delta^{13}C_E)$ using (1) global atmospheric burden (*B*) and loss 27630



(3)

(4)

(*L*) taken from ACTM simulation and (2) observed atmospheric isotope ratio $\delta^{13}C_{atmos}$. We used $\delta^{13}C_{atmos}$ observations from the Law Dome ice core (1885–1976)/firn (1944–1998) records (Ferretti et al., 2005); from air archive samples (1978–1994) (Francey et al., 1999) and NOAA-ESRL network direct observations (1998–2010) at CGO (Miller et al., 2002; White and Vaughn, 2011). Schmitt et al. (2013) reported a possible interfering effect by Kr on $\delta^{13}C$ measurements using continuous-flow isotope ratio mass spectrometry systems. We assume that possible bias caused by the Kr-interference is not significant in deducing the observed $\delta^{13}C$ trends over the past century.

 $\delta^{13} \tilde{C}_E$ can be calculated using the observed atmospheric isotope ratio $\delta^{13} C_{atmos}$ and mass balance Eq. (1) as follows:

$${}^{13}E = \frac{d}{dt}({}^{13}B) + \sum_{i}^{13}L_i$$
(7)

$${}^{13}B = \left(\delta^{13}C_{\text{atmos}} \times 0.001 + 1\right) \times R_{\text{std}} \times B$$
$${}^{13}L_{j} = L_{j} \times \alpha_{j} \times R_{\text{atmos}}$$

$${}_{15} \quad \delta^{13} \mathcal{C}_E = \left[\frac{{}^{13}E}{E \times R_{\text{std}}} - 1\right] \times 1000 \tag{10}$$

where L_i are the loss processes due to reactions with OH, O(¹D), CI and soil oxidation, $\alpha_i = {}^{13}k_i/{}^{12}k_i$ is the isotopic fractionation factors for different loss processes, k is the rate coefficient of chemical reactions. Superscript 13 are used for carbon isotope 13 C in CH₄. 12 CH₄ is approximated by CH₄ = 12 CH₄ + 13 CH₄ excluding the more minor isotopes which has a negligible effect on our results. Our results are consistent with the calculation of δ^{13} C_E under non-steady-state conditions (Lassey et al., 2000).



(8)

(9)

Alternatively $\delta^{13}C_E$ can also be calculated by considering the relative fraction of individual emission categories (E_i) as follows:

$$\delta^{13}C_E = \sum \left(\delta^{13}C_{E_i} \times E_i \right) / E \tag{11}$$

where $E = \sum E_i$ and $\delta^{13}C_{E_i}$ is the isotopic signature for emission category E_i . For the optimized case, $E = E_{opt} = \sum E_{ini_i} + \Delta E$, and $L_i = L_{opt_i}$ and $B = B_{opt}$ from ACTM simulation will represent the observed condition. Using Eqs. (5)–(10), we calculate the isotope signature for $E_{opt}(\delta^{13}C_{E_{opt}})$. Again $\delta^{13}C_{E_{opt}}$ can be estimated using Eq. (11) with $E_{opt} = \sum E_{ini_i} + \Delta E$. Here we assume that ΔE is distributed between two hypothetical emissions ΔE_I and ΔE_h with lighter ($\delta^{13}C_{\Delta E_I}$) and heavier ($\delta^{13}C_{\Delta E_h}$) isotopic signatures, respectively. The $\delta^{13}C_{E_i}$ are taken from Monteil et al. (2011) and references therein (Table 1). Using Eq. (11), we obtain,

$$\delta^{13} C_{E_{opt}} = \frac{\sum \delta_{E_{ini_j}}^{13} \times E_{ini_j} + \delta^{13} C_{\Delta E_l} \times \Delta E_l + \delta^{13} C_{\Delta E_h} \times \Delta E_h}{E_{opt}}$$
(12)

And we have an additional constraint on ΔE_{l} and ΔE_{h} as follows:

$$\Delta E = \Delta E_{\rm I} + \Delta E_{\rm h}$$

¹⁵ The value of $\delta^{13}C_{E_{opt}}$ as calculated using Eq. (10) is substituted into Eq. (12). There is, however, no unique solution for Eqs. (12) and (13) as they contain 4 unknown variables $(\Delta E_{\rm h}, \Delta E_{\rm l}, \delta^{13}C_{\Delta E_{\rm h}}$ and $\delta^{13}C_{\Delta E_{\rm l}}$). For simplicity, we assume that $\delta^{13}C_{\Delta E_{\rm h}}$ and $\delta^{13}C_{\Delta E_{\rm l}}$ represent emissions from biomass burning (δ^{13} C is -21.8% from Monteil et al., 2011) and residual biogenic sources (e.g., wetland, rice, animals, etc., mean δ^{13} C near to -60% from Sapart et al., 2012), respectively. Equation (12) can now be modified to:

$$\Delta E_{l} = \frac{\delta^{13} C_{E_{opt}} \times E_{opt} - \sum \delta_{E_{ini_{j}}}^{13} \times E_{int_{j}} - \delta^{13} C_{\Delta E_{h}} \times \Delta E}{\left(\delta^{13} C_{\Delta E_{l}} - \delta^{13} C_{\Delta E_{h}}\right)}$$
(14)
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(13)

Using Eq. (14) we can estimate $\Delta E_{\rm I}$ and then $\Delta E_{\rm h}$ is calculated using Eq. (13). Estimation of emissions due to biomass burning is our primary interest here because no direct statistics are available over the past century and it was assumed constant at 49.7 Tg yr⁻¹ in $E_{\rm ini}$. It may be reiterated here that the anthropogenic emissions varied as per the EDGAR inventories, and wetland and rice emissions are taken from a terrestrial ecosystem model simulation.

Although CH₄ losses due to reactions with Cl and O(¹D), which mainly take place in the stratosphere, are small compared to the total loss, the strong isotopic fractionations ($\varepsilon_i = (\alpha_i - 1) \times 1000$; ε_i is also known as "kinetic isotope effect" (KIE)) in these reactions have a large impact on the isotopic budget (Lassey et al., 2007a; Wang et al., 2002). The isotopic fractionation factors (α_i) for CH₄ + CI reaction ($\alpha_{CI} = 0.935$) is

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much smaller than that of CH₄ + OH and CH₄ + O(¹D) reactions, e.g., $\alpha_{OH} = 0.9961$ and $\alpha_{O(^{1}D)} = 0.9872$, respectively (Saueressig et al., 1995). Previous studies have shown that there exists a large vertical gradient in δ^{13} C from the troposphere to the strato-

- ¹⁵ sphere due to the stronger fractionations effect in $CH_4 + O(^1D)$ and $CH_4 + CI$ reactions during passage through the stratosphere (Rice et al., 2003; Röckmann et al., 2011; Sugawara et al., 1997). Stratosphere air returning to the troposphere is enriched in ¹³ CH_4 , but the re-entry flux and consequent ¹³ CH_4 enrichment in the troposphere are not well quantified (Lassey et al., 2007a). In a modeling study, Wang et al. (2002) esti-
- ²⁰ mated that the tropospheric ¹³CH₄ enrichment for 1992 due to stratospheric CI without assuming steady state was 0.23‰ (0.18–0.54‰). If the α_i for CI and O(¹D) loss processes as mentioned above (Saueressig et al., 1995) are used in the one-box model, because the strong isotopic fractionations, the isotopic effect of stratospheric CH₄ loss on the tropospheric δ^{13} C budget will be overestimated. To avoid such overestimation,
- ²⁵ we assume that α_i in the troposphere are mainly due to OH loss, and a smaller fractionation effect of the stratospheric loss on δ^{13} C at the surface. In fact, the magnitude of fractionation during CH₄ loss, both in the troposphere and stratosphere, is uncertain since the published values of α_i in the literature are significantly different. The values of α_{CI} ranges from about 0.935 (Saueressig et al., 1995) to 0.966–0.974 (Tanaka



et al., 1996; Gupta et al., 1997), $\alpha_{O^1(D)}$ from 0.9872 (Saueressig et al., 2001) to 0.999 (Davidson et al., 1987) and α_{OH} from 0.9946 (Cantrell et al., 1990) to 0.9961 (Saueressig et al., 2001). In the present study, α_{OH} is the mean of two published α_{OH} values (Cantrell et al., 1990; Saueressig et al., 2001). We scaled α_{CI} so that the reaction with ⁵ Cl has an impact of +0.23% on surface δ^{13} C in 1992, to be consistent with Wang et al. (2002). $\alpha_{O(1D)}$ is also adjusted by keeping the ratio $(\alpha_{CI} - \alpha_{OH})/(\alpha_{O(1D)} - \alpha_{OH})$ and total loss by reactions with Cl and O(¹D) unchanged. The weighted mean isotopic fractionation factor ($\bar{\alpha}$) for all loss processes is 0.9943 and the global mean $\delta^{13}C_E$ value in 1990 is -52.1%, which are in agreement with other estimates, e.g., 0.9941 and -52.3%, respectively (e.g., Lassey et al., 2000). The values of α_{CI} and $\alpha_{O^1(D)}$ used in this study are now much closer to that of α_{OH} producing the smaller fractionation effect of stratospheric loss on the estimation of δ^{13} C at the surface (Table 1). Due to the increase in atomic Cl in stratosphere, the effect of α_{CI} on surface δ^{13} C is estimated to increase from +0.01 to +0.38% during the period 1910–2010.

3 Results and discussion

3.1 Model-observation comparisons of CH₄ concentrations

The model simulations are compared with the observed CH_4 concentration time series constructed from ice core, firn air, air archives and ambient air measurements (Fig. 2). The first 9 years (1901–1909) of the simulation are used to spin-up the ACTM, to guarantee that the results are independent of the initial conditions. The initial ACTM simulation using E_{ini} (Fig. 2) underestimates the growth rate by 0.6 ppb yr⁻¹ (4.8 ppb yr⁻¹) for the period of slow (rapid) growth during 1910–1950 (1950–1990), and it also fails to capture the slowdown of the observed CH_4 growth rate during the 1990s. Apparently, the E_{ini} is a better first-guess in the first half of the 20th century compared to the latter half, when the CH_4 growth rate changed dramatically. The model-observation mismatches are attributed to incomplete knowledge of E_{ini} as used in the ACTM simu-



lations, assuming no significant uncertainties in chemistry and transport. Thus we have estimated optimized global total CH_4 emissions using the mass balance calculation as described in Sect. 2.5.

- A new ACTM simulation using optimized emissions (*E*_{opt}) was performed and the modeled CH₄ concentrations for the Arctic and Antarctic regions (blue and green, respectively) are shown in Fig. 2. Simulated and observed CH₄ are in good agreement for the ACTM using *E*_{opt} (see Table 2 for detailed statistics). Observations in the Antarctic region reveal that the growth rates are: moderate (5.1 ppb yr⁻¹) during 1910–1950; fastest (13.6 ppb yr⁻¹) during 1950–1990; moderate (6.7 ppb yr⁻¹) during the 1990s; near-steady in the early 2000s; and moderate again (5.7 ppb yr⁻¹) since 2007. These are all simulated well by ACTM within the measurement uncertainties of 2–5 ppb. CH₄ observations in the Arctic region follow a similar trend as the Antarctic region: 5.6 ppb yr⁻¹ during 1910–1950; 15.2 ppb yr⁻¹ during 1950–1990; 5.2 ppb yr⁻¹ during
- ¹⁵ ing the 1990s; near-steady state during the early 2000s; and 5.8 ppb yr⁻¹ since 2007. The inter-polar difference (IPD) of CH₄ is, however, smaller in the model simulation (102.2±17.9 ppb) compared to observations (117±16.7 ppb) for the period 1949–2010. A detailed discussion on IPD is given in Sect. 3.3.

3.2 Trends in methane lifetime

- ²⁰ As no consensus has been reached for the trends in global mean OH concentration simulated by state-of-the-art CTMs (e.g., John et al., 2012), we used monthly varying climatological OH concentrations for our ACTM simulations. This OH distribution, from Spivakovsky et al. (2000), also fits well with ACTM transport for simulating interhemispheric gradients in CH₃CCl₃ and CH₄ for the period 1988–2010 (Patra et al.,
- ²⁵ 2011). For E_{opt} the total CH₄ lifetime is given by $\tau_{Total} = \frac{B_{opt}}{\left(E_{opt} \frac{dB_{opt}}{dt}\right)}$. The trends in CH₄

lifetime and tropospheric mean temperature anomaly are shown in Fig. 3. The average CH₄ total lifetime during 1910–1919 and 2000–2009 are 9.4 ± 0.09 and 9.0 ± 0.09 years,



respectively. The ACTM simulated air temperature anomaly is very similar to that of the observed temperature anomaly produced at GISS (Hansen et al., 2010). This illustrates that the long-term simulation of CH₄ by ACTM, driven by analyzed SST only, is close to that simulated by ACTM nudged to the reanalysis meteorology. To examine the factors causing change in CH₄ lifetime, we have calculated the temporal change of "apparent reaction rate" $k_{a,i} = L_i/B$, where *i* is CH₄ reaction with OH, O(¹D), CI or soil oxidation. Between the 1910s and the 2000s, contributions of CI, OH, O(¹D) reactions and soil oxidation to CH₄ lifetime change are +61.7, +48.7, -1.6 and -8.8%, respectively. Thus, the ~4% shorter average CH₄ total lifetime from the first to last decades of the last 100 years (1910–2009) is mainly caused by the large increase in CI concentration and the increase in the tropospheric air temperature.

3.3 Estimation of global total CH₄ emissions for improved concentration simulations

Global total CH₄ emissions are optimized by minimizing the model-observation mismatches of CH₄ concentrations over the Antarctic region (combined measurements from Law Dome ice core/firn air and direct air sampling from Cape Grim). Figure 4 shows a comparison of both the initial and optimized emissions with TransCom-CH₄ emissions (CH₄_EXTRA) for 1988–2010 (Patra et al., 2011). The increase rate of global total emissions is underestimated (overestimated) by ~30% (~380%) during 1940–1989 (1990–2009) in the bottom-up emission data. Optimized emissions (*E*_{opt}) are in overall agreement with TransCom CH₄ emissions for 1988–2010. This indicates

- robustness of the mass-balance based optimization used in this study. For quantitative assessment of ACTM simulations with all the emissions scenarios (E_{ini} and E_{opt}), the model-observation CH₄ concentration biases and SDs (1 σ) (in ppb) of the biases over
- ²⁵ the Antarctic region, and averages of E_{ini} , E_{opt} and B_{opt} for each decade are summarized in Table 2. Both the bias and 1σ are reduced drastically when optimized global total emissions are used in ACTM (Table 2). It may be reiterated here that the global



total emissions for mass balance is dependent on the $\rm CH_4$ loss rates as parameterized in ACTM (Eq. 1).

- Figure 5 shows inter-polar differences (IPD) of CH_4 concentration using the ACTM simulations with E_{ini} and E_{opt} over Arctic and Antarctic regions for 1910–2010. For the observations, we used a combination of datasets from Arctic (NGRIP firn air: 1953–2001, NEEM firn air: 1949–1996 and direct measurements at SUM since 1998) and Antarctic (Law Dome ice cores: 1901–1980, DE08-2 and DSSW20K firn air: 1978–1993, and direct measurements and archive tanks at CGO since 1978) regions. It is noted that no observation for 1910–1948 is available from the Arctic region, so the IPD of observed CH_4 concentration is limited to 1949–2010. For ACTM simulations with both E_{ini} and E_{opt} IPDs are underestimated compared to observations by 23 ± 10 and 15.3 ± 8.6, respectively. The difference between observed and simulated IPDs is more prominent for the period before the 1990s than the recent time (Fig. 5). This underesting the statement is the statement time (Fig. 5).
- timation also indicates that although the global total emission is optimized (E_{opt}) using observations from Antarctic region, the northern–southern hemispheric totals of CH₄ emissions in E_{opt} are imbalanced, particularly for the period of 1950–1990, assuming that the other factors affecting IPD of CH₄ concentration such as the latitudinal distribution of CH₄ sinks and the mixing rates between hemispheres remained unchanged. Using a two-box model (e.g., Baumgartner et al., 2012), we estimate an increase in NH
- ²⁰ emission by 7.2±2.3% and decrease in SH emission by 17.2±3.9% for fitting the large observation-model mismatch of IPDs during 1950–1990. The atmospheric lifetime of CH₄ and inter-hemispheric exchange time used for the two-box model calculation are 9.27 yr (mean total lifetime for 1949–2009, this study) and 1.39 yr (Patra et al., 2011), respectively. However, a systematic bias of -1 year in assigning effective age to the ice
- ²⁵ core/firn air would overestimate the CH₄ IPD during 1950–1990. This possibility cannot be ruled out because the model-observation mismatch in CH₄ IPD is much smaller for period of direct measurements (1998 onward). We also found a high correlation (R = 0.99) between differences of IPDs (IPD_{opt} – IPD_{int}) and difference of NH-SH emissions (Δ ENS_{opt} – Δ ENS_{int}) in optimized and initial guess cases (Fig. 5 inset), i.e., the



larger the change in difference of NH-SH emissions the larger the change in IPD. The change in latitudinal distribution of emissions is the dominant driver of IPD, which is in agreement with a previous study (Mitchell et al., 2013).

3.4 Application of δ^{13} C for separation of source category in ΔE

- The difference between initial and optimized emissions ($\Delta E = E_{opt} E_{int}$) ranges from 5 -15 to 0 and from 0 to 60 Tg yr⁻¹ for the first and second halves of the last 100 years, respectively (Fig. 4). Though E_{opt} reproduces the CH₄ concentration for the last 100 years fairly well, it does not verify how individual source categories have evolved over the period. Here we constrain different emission categories based on evolution of δ^{13} C by separating ΔE into isotopically lighter and heavier sources (Sect. 2.6). We set the goal here to infer trends in emissions from biomass burning, because this emission category was kept constant over the whole simulation period due to the lack of consensus among different estimations (Mieville et al., 2010; Ito and Penner, 2005). The detailed interannual variability cannot be calculated from $\delta^{13}C_{atmos}$ over the Antarctic region, because a smoothed fitted curve is used to interpolate between observations. 15 For the sake of consistency, we also fitted smoothed curves for $\delta^{13}C_{E_{out}}$ and ΔE , and then redistributed the smoothed ΔE between ΔE_1 and ΔE_h . Observations and smooth time series of $\delta^{13}C_{atmos}$ and $\delta^{13}C_{E_{out}}$ are shown in Fig. 6a. We assumed $\delta^{13}C_{\Delta E_{1}}$ and $\delta^{13}C_{\Delta E_h}$ values of -60% (representing biogenic sources) and -21.8% (representing biomass burning emissions), respectively (Table 1). Interestingly, $\Delta E_{\rm l}$ and $\Delta E_{\rm h}$ show 20
- different trends, which indirectly illustrates the novelty of this method to provide independent pieces of information (Fig. 6b). As $\Delta E_{\rm I}$ is representing, in general, biogenic sources, a group of sources rather than an individual source, we treat it as a separate supplement source with $\delta^{13}C_{E_i} = -60\%$, while $\Delta E_{\rm h}$, which represents biomass
- ²⁵ burning, an individual source, is added to the initial biomass emission (49.7 Tgyr⁻¹) for estimating corrected biomass burning emission (E_{bb} = 49.7 (initial biomass burning emission) + ΔE_h (correction term) Tgyr⁻¹). We have taken this approach because 27638



the level of confidence for estimations of emission variations from biomass burning is relatively low compared to all other emission categories, which are either estimated based on statistical data of human activities or model simulations. E_{bb} is also shown in Fig. 6b. $E_{\rm bb}$ has an increasing trend (varies from 23 to 51 Tgyr⁻¹) during the period 1910–1990, followed by a decreasing trend (from 51 to 38 Tg yr^{-1}) from 1990s onward. 5 The CO based reference for open biomass burning by Ito et al. (2005), which is scaled by using the ratio of CH₄ and CO emissions from biomass burning based on Global Inventory for Chemistry-Climate studies (GICC) dataset (Mieville et al., 2010), shows a similar increasing trend. However, both the trend and mean values in the present study are larger than the GICC dataset. The variation in biomass burning emissions during the 20th century is influenced by both the warmer climate and human activities (e.g., agricultural expansion, land-use change, domestic fuel use, fire management). The human population increased at the fastest rate since 1950 as per the United Nations statistics. However, saturation in cropland expansion in Asia, shift in domestic fuel use and improved fire management practices since the late 1980s 15 would have suppressed the growth of CH_{4} emissions from human activities (Patra et al., 2013; Li et al., 2002; Hurtt et al., 2006; Sathaye and Tyler, 1991; Montiel and Kraus, 2010). Uncertainties remain in the estimation of biomass burning emissions for the 20th century due to assumptions of $\delta^{13}C_{E_i}$ and α_i values. However, a sensitivity analysis using the estimated biomass burning emissions by varying $\delta^{13}C_{\Delta E}$, from –55

²⁰ analysis using the estimated biomass burning emissions by varying $\delta^{15}C_{\Delta E_1}$ from -55 to -65% (Sapart et al., 2012) suggests the trends in CH₄ emissions from biomass burning is robust and only the magnitude of this emission could change by ± (5–15)% (Appendix A). The decreasing trend in biomass burning emission has also been reported for the recent years but typically at lower absolute levels (Fig. 6) (e.g., Kirschke et al., 2013). The inclusion of δ^{13} C in inversions and global mass balances has also in the past yielded higher emissions for global biomass burning than inventories (Miller et al., 2002; Bousquet et al., 2006).

The Supplement biogenic source (ΔE_I) follows a similar trend (a slow increasing trend during 1910–1950 and a rapid increase for the period 1950–1980, Fig. 6b) to



that of the biogenic sources in the initial emissions (E_{ini}) for the period 1910–1980 (Fig. 1a). Between 1981–2006 ΔE_{I} shows a decreasing trend, followed by an increase from 2007 onward (Fig. 6b), which is different from the biogenic sources in E_{ini} showing an increasing trend during this whole period (Fig. 1a). Recent studies suggested

- ⁵ a likely reduction in emissions from wetlands (e.g., due to more frequent El Niño events in the last three decades compared to the decades before 1980, Hodson et al., 2011, or due to the cooling effect of increased anthropogenic sulfur pollution, Gauci et al., 2004, or volcanic eruptions, Hogan et al., 1994) and changes in rice agricultural practices (Li et al., 2002). The increase in atmospheric CH₄ since 2007 may be ascribed
- ¹⁰ to enhanced emissions from wetlands combined with an increasing trend of fossil fuel use (Dlugokencky et al., 2009; Bousquet et al., 2011; Kirschke et al., 2013; Bergamaschi et al., 2013). Apparently, ΔE_{l} is able to capture these detailed features of biogenic emissions in recent decades, which are otherwise different in E_{ini} . On the contrary, the rapid increase in emissions during 1950–1980, which is reflected in both ΔE_{l} and bio-15 genic sources in E_{ini} , is likely to be driven by increasing anthropogenic activities (e.g.,
- ¹⁵ genic sources in *E*_{ini}, is likely to be driven by increasing anthropogenic activities (e.g., agriculture, ruminants and termites, organic waste deposits etc.) related to increasing human population during this period.

The present analysis limits the unaccounted emissions ΔE to be from only biomass burning (heavy) and biogenic (light) sources, but there could be other combinations of

²⁰ different categories of emissions. As we have two Eqs. (12) and (13), unique solutions are only possible for two unknown categories of emissions assuming rest of the emissions are all known. One reasonable scenario is distributing ΔE into biomass burning and biogenic sources as examined in this study. To calculate for other possible combinations, such as fossil fuel (heavy) and biogenic (light) sources, we need to know ²⁵ the correct biomass burning emissions in E_{ini} and leaving the correction terms ΔE_{I} and ΔE_{h} erroneous. We also attempted a combination of biomass burning and fossil fuel sources, but it produced unrealistic emission values (negative). We need additional





of CH₄, which are presently very limited, to find solutions for more than two variables, e.g., distributing ΔE among biomass burning, fossil fuel and biogenic sources.

4 Conclusions

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We have simulated CH₄ concentration and its δ^{13} C for 1910–2010. The major findings of our study are as follows:

1. We used initial emissions (*E*_{ini}) from bottom-up inventories for anthropogenic sources, emissions from wetlands and rice paddies simulated by a terrestrial biogeochemical model (VISIT), and the ACTM to simulate atmospheric CH₄ concentrations for the period 1910 to 2010. By minimizing the difference between ACTM simulations and observed CH₄ concentrations (over the Antarctic region) using a mass balance calculation, we estimated the optimized global emission (*E*_{opt}) for atmospheric CH₄ for the past 100 years. The ACTM simulation with the optimized emissions reproduced the moderate (5 ppb yr⁻¹) and rapid (14.3 ppb yr⁻¹) growth rates for 1910–1950 and 1950–1990, respectively, and the drop in growth rate (5.3 ppb yr⁻¹) during 1990s, near-steady state in the early 2000s, and the regrowth in 2007 (5.2 ppb yr⁻¹) fairly well. However, the ACTM underestimated IPDs, particularly for the period before the 1990s compared to the last 2 decades. Such underestimation in IPD can be corrected by increasing the NH-SH gradient in the optimized emission (*E*_{opt}), suggesting that the change in latitudinal distribution of CH₄ emissions is the dominant driver of IPD.

2. To further investigate the strength and variability of different CH_4 emission categories, we introduced an additional constraint from $\delta^{13}C$. This helps us in inferring the possible trends in biomass burning emissions, which was initially kept constant over the period. The optimized result suggests an increasing biomass burning emissions until 1990 and a decrease afterwards. These variations are comparable with biomass burning emissions previously reported in the literature.



We also estimated a supplementary biogenic source, which is likely to fill the incomplete information of biogenic sources in the initial emissions. Further details about CH₄ sources could not be inferred due to limited observations covering the past 100 years and without measurements of additional constraints on CH₄ source categories, such as the ¹⁴C and δD .

Appendix A:

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As the splitting of ΔE into $\Delta E_{|}$ and ΔE_{h} depends on the values of $\delta^{13}C_{\Delta E_{|}}$ used in the calculation, a sensitivity of estimated proposed biomass burning emission with respect to varying $\delta^{13}C_{\Delta E_{|}}$, from -55 to -65 ‰ (Sapart et al., 2012), is shown in Fig. A1. Here the estimated biomass burning emission is expressed as a decadal mean so that it is consistent with other datasets (GICC: Mieville et al., 2010; CO based reference: Ito et al., 2005). The uncertainty in estimation of possible biomass burning emissions increases from 1950 onward reaching a peak in 1990s.

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Table 1. Average source and sink strengths of CH₄ for 1980–1989, isotopic ratios ($\delta^{13}C_{E_i}$) and fractionation factors (α_i).

Source/Sink	Annual Flux (Tgyr ⁻¹)	$\delta^{13}C_{E_i}$ (‰)	a ^a
Sources			
Wetland	146.8	-59	
Rice	34.2	-63	
Animals	83.5	-62	
Termites	20.7	-57	
Biomass Burning	49.7	-21.8	
Coal	30.6	-35	
Oil and Gas	59.2	-40	
Landfills	46.0	-55	
Ocean	7.4	-59	
Mud Volcanoes	7.5	-40	
ΔE^{b}	60.0	-21.8	
		-60.0	
Total Source	545.6		
Sinks			
OH	-451.5		0.995350
O(1D)	-7.5		0.994940
CI	-14.5		0.992532
Soil	-28.6		0.978000
Total Sinks	-502.1		

^a $\alpha_{\rm OH}$ is the mean of two published values (Cantrell et al., 1990; Saueressig et al., 2001), while the values of $\alpha_{\rm Cl}$ and $\alpha_{\rm O(^1D)}$ are modified assuming the smaller effect of isotopic fractionation in the stratosphere at the surface.

^b ΔE is assumed to consist of isotopically heavier ($\delta^{13}C = -21.8\%$) and lighter ($\delta^{13}C = -60\%$) sources.

Decade	Model-observation CH ₄ concentration (ppb)		E _{ini}	E _{opt}	B _{opt}
	with <i>E</i> _{ini}	with <i>E</i> _{opt}	(Tgyr ⁻¹)	(Tgyr ⁻¹)	(Tg)
1910–1919	40.94 ± 3.29	-3.31 ± 0.51	315.0	300.1	2686.4
1920–1929	37.33 ± 5.30	-2.07 ± 0.37	326.4	319.9	2846.7
1930–1939	21.21 ± 3.25	-0.44 ± 0.67	337.0	334.1	3002.7
1940–1949	13.92 ± 1.85	0.05 ± 0.39	351.2	349.2	3132.0
1950–1959	0.13 ± 6.84	-0.61 ± 0.44	375.7	381.9	3331.7
1960–1969	-38.96 ± 16.31	0.11 ± 1.26	407.2	434.4	3675.5
1970–1979	-101.88 ± 20.66	-1.04 ± 0.75	442.4	491.6	4109.6
1980–1989	-160.65 ± 14.15	0.15 ± 1.59	485.6	545.6	4572.7
1990–1999	-161.76 ± 8.37	2.54 ± 2.12	515.0	557.8	4878.8
2000–2009	-122.40 ± 15.34	2.67 ± 3.25	531.0	555.0	4973.2

Table 2. Average bias and 1 SD (in ppb) of model-observed CH_4 concentration for each decade over the Antarctic region. The averages of E_{ini} , E_{opt} and B_{opt} for each decade are also shown.







Figure 1. (a) Time series of CH_4 emission inventory estimates from different categories during our simulation. **(b)** Chemical loss of CH_4 calculated using ACTM simulation with initial emissions (E_{ini}).



Figure 2. ACTM simulations of CH_4 concentration compared with ice core, firn and direct measurements. ACTM simulated CH_4 concentrations with E_{ini} (E_{opt}) for Arctic and Antarctic regions are shown in magenta (blue) and purple (green), respectively. Red (black) symbol and text are for Arctic (Antarctic) region. Annual average ice core data (Law Dome: DSS, DE08 and DE08-2), firn records (Law Dome (DE08-2 and DSSW20K); NGRIP firn data; and NEEM firn data) and direct observations (CGO: air archive, flask sampling and GAGE/AGAGE; SUM: flask sampling) are prepared and presented separately here. All the observation data are referenced to the Tohoku University (TU) CH_4 scale (Aoki et al., 1992; Umezawa et a., 2014).





Figure 3. Five-year running means of CH_4 total lifetime and global tropospheric temperature anomaly (with the base period 1951–1980) during 1910–2010.





Figure 4. Time series of initial (E_{ini}) and optimized (E_{opt}) emissions. Emissions for a shorter period from the TransCom-CH₄ experiment (Patra et al., 2011) are also shown for a comparison.

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Interactive Discussion



Figure 5. Comparison of model simulated and observed inter-polar differences (IPD) of CH₄ concentration. Inset: correlation between difference of IPD (IPD_{opt} – IPD_{int}) and difference of NH-SH emissions (Δ ENS_{opt} – Δ ENS_{int}) in optimized and initial guess cases.





Figure 6. (a) Observations (circles) and smooth time series of $\delta^{13}C_{atmos}$ and $\delta^{13}C_{Eopt}$ (b) ΔE (smoothed) is split into ΔE_{I} and ΔE_{h} . Here $\delta^{13}C_{\Delta E_{I}}$ and $\delta^{13}C_{\Delta E_{h}}$ are assumed to be -60% (biogenic sources, a supplementary source) and -21.8% (biomass burning emission), respectively. Annual average $\delta^{13}C_{atmos}$ observations from Law Dome (ice core (red circle)/firn air (blue circle) records) and CGO (air archive samples and NOAA-ESRL network direct observations; (magenta circle)) are shown separately here. The estimated biomass burning emission is given by $E_{bb} = 49.7$ (initial biomass burning emission) + ΔE_{h} (correction term) Tgyr⁻¹. A recent estimation of biomass burning emissions (Kirschke et al., 2013) for last three decades is also shown for comparison (filled diamond).





Figure A1. Sensitivity of estimated proposed biomass burning emissions with respect to varying $\delta^{13}C_{\Delta E_1}$ (biogenic sources) from -55 to -65 ‰.

