



Particulate matter
and vertical
processes

G. Curci et al.

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How much is particulate matter near the ground influenced by upper level processes within and above the PBL? A summertime case study in Milan (Italy)

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Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



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**Particulate matter
and vertical
processes**

G. Curci et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



Abstract

Chemical and dynamical processes yield to the formation of aerosol layers in the upper planetary boundary layer (PBL) and above it. Through vertical mixing and entrainment into the PBL these layers may contribute to the ground-level particulate matter (PM), but a quantitative assessment of such contribution is still missing. This study investigates this aspect combining chemical and physical aerosol measurements with WRF/Chem model simulations. The observations were collected in the Milan urban area (Northern Italy) during summer of 2007. The period coincided with the passage of a meteorological perturbation that cleansed the lower atmosphere, followed by a high pressure period that favoured pollutant accumulation. Lidar observations reveal the formation of elevated aerosol layers and show evidences of their entrainment into the PBL. We analyze the budget of ground-level $PM_{2.5}$ (particulate matter with aerodynamic diameter less than $2.5\ \mu\text{m}$) with the help of the online meteorology-chemistry WRF/Chem model, with particular focus on the contribution of upper level processes. We find that an important player in determining the upper PBL aerosol layer is particulate nitrate, which may reach higher values in the upper PBL (up to 30 % of the aerosol mass) than the lower. The nitrate formation process is predicted to be largely driven by the relative humidity vertical profile, that may trigger efficient aqueous nitrate formation when exceeding the ammonium nitrate deliquescence point. Secondary $PM_{2.5}$ produced in the upper half of the PBL may contribute up to $7\text{--}8\ \mu\text{g m}^{-3}$ (or 25 %) to ground level concentrations on hourly basis. A large potential role is also found to be played by the residual aerosol layer above the PBL, which may occasionally contribute up to $10\text{--}12\ \mu\text{g m}^{-3}$ (or 40 %) to hourly ground level $PM_{2.5}$ concentrations during the morning. This study highlights the importance of considering the interplay between chemical and dynamical processes occurring within and above the PBL when interpreting ground level aerosol observations.

Particulate matter and vertical processes

G. Curci et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



1 Introduction

The understanding of processes governing the atmospheric aerosols is primarily motivated by their adverse effects on health and their contribution to the radiative budget of the atmosphere. Diseases to the respiratory system have been linked to inhalation of aerosols, especially their finer and more numerous fraction (Beelen et al., 2013; Oberdorster, 2001), although the mechanisms underlying the health effect associated to size, number and composition of particulate matter have just started to be disclosed (Harrison and Yin, 2000; Daher et al., 2012; Perrone et al., 2013). Aerosols affect the atmospheric energy balance directly, by scattering and absorbing radiation (Yu et al., 2006), indirectly, by serving as cloud condensation nuclei (Lohmann and Feichter, 2005), and semi-directly, by heating the air through absorption of radiation and reducing low cloud cover (Johnson et al., 2004). The assessment of these effects of aerosols is still characterized by large uncertainties, since it is still uncertain our knowledge of the processes determining their abundance, size distribution, and chemical composition, which strongly vary in space and time (Raes et al., 2000; Poschl, 2005). Here we focus on the interplay between dynamical and chemical processes in the vertical direction, in order to better understand the budget terms making up the ground level particulate matter, a common measure to evaluate the air quality. The study is focused on the urban environment of Milan, placed in the middle of the Po Valley (Italy), an European hot-spot for atmospheric pollution.

The correlation between pollutants at the surface and meteorological variables is well established and the fundamental role played by the variables associated to the vertical mixing in the planetary boundary layer (PBL) has been highlighted for both ozone (Di Carlo et al., 2007, and references therein) and particulate matter (Tai et al., 2010, and references therein). Moreover, Zhang and Rao (1999) analyzed aircraft and tower measurements over the Eastern United States and showed that elevated nocturnal layers rich of ozone and its precursors aloft, remnant of the previous day mixed layer, may strongly affect ground-level ozone levels the following morning as the vertical motions

Particulate matter and vertical processes

G. Curci et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



**Particulate matter
and vertical
processes**

G. Curci et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



mix upper and surface air. The same authors suggested that a reduction of ozone and precursors aloft may be more effective than local emission cuts in reducing pollution, thus calling for a region-wide strategy for emissions control. Aerosols are also known to form layers above or near the top of the mixing layer, especially when stability and presence of clouds increase (e.g. O'Dowd and Smith, 1996). Similarly to ozone, an aerosol residual layer aloft is often observed (e.g. Di Giuseppe et al., 2012), which may influence the aerosol at the surface, as witnessed by similar size-distributions (Maletto et al., 2003). A significant contribution to surface aerosol from entrainment and vertical dilution and chemical net production in the boundary layer was also pointed out in recent studies using single-column models (van Stratum et al., 2012; Ouwersloot et al., 2012).

The nontrivial relationship between ground- and upper-level aerosols burden is illustrated by the comparison of surface particulate matter (PM) mass concentrations with aerosol optical depth (AOD), which is proportional to the aerosol column load (typically measured by ground-based sun-photometers or retrieved from satellites). In a well mixed PBL, the AOD may exhibit a high correlation with surface PM, especially with its fine fraction, and indeed this assumption is often exploited to infer surface $PM_{2.5}$ (PM with diameter $< 2.5 \mu m$) from satellite AOD observations (e.g. van Donkelaar et al., 2010). However, that assumption does not always hold, because of the presence of significant aerosol stratification aloft, and noticeable differences between AOD and surface PM behaviour may occur, such as in the timing of daily peak values or in multi-day trends (Barnaba et al., 2007, 2010; Boselli et al., 2009; Estelles et al., 2012; He et al., 2012). Analyzing two-year measurements in the Po Valley (Italy), Barnaba et al. (2010) indeed pointed out that annual cycles of AOD and surface PM_{10} (PM with diameter $< 10 \mu m$) display a remarkable opposite phase. While PM_{10} peaks in winter, because of the reduced dilution by a shallower PBL and of the condensation of semi-volatile species favoured by the lower temperatures, AOD peaks in summer, because of a more persistent presence of an aerosol residual layer aloft, which contributes up to 30 % of the total AOD.

**Particulate matter
and vertical
processes**

G. Curci et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



Aircraft measurements also showed intriguing features of aerosol vertical gradients in the lower troposphere, in particular when looking at different chemical components. Several studies reported a generally constant or slightly decreasing profile in the convective boundary layer of sulfate and organic matter as opposed to an increasing profile of nitrate (Neuman et al., 2003; Cook et al., 2007; Crosier et al., 2007; Morgan et al., 2009; Ferrero et al., 2012). Neuman et al. (2003) attributed the enhanced nitrate layer near the top of the PBL to the lower temperatures that favour gas-phase nitric acid (HNO_3) and ammonia (NH_3) conversion to particulate ammonium nitrate. The same authors also pointed out that nitrate and HNO_3 display sharp vertical gradients in the PBL, as opposed to other directly emitted (carbon monoxide) or secondary (ozone) species that are relatively uniform, and interpreted the observation as an indication that thermodynamic equilibrium between gas and particle phases occurs faster than vertical mixing. However, the issue is still debated as subsequent model studies found that an instantaneous thermodynamic equilibrium between HNO_3 and nitrate yields too steep and unrealistic vertical gradients (Morino et al., 2006; de Brugh et al., 2012).

A quantitative assessment of the contribution of elevated aerosol layers and related dynamical and chemical processes to ground-level particulate matter level is still lacking. Recent modelling studies that reported budget (or process) analyses of the simulated aerosol mainly focused on terms of the continuity equation at the surface or on integrated values over the whole boundary layer. Surface and PBL total $\text{PM}_{2.5}$ mass is calculated to be mainly produced by direct emissions and secondary formation by aerosol processes (e.g. condensation and absorption) and removed by horizontal and vertical transport and wet deposition (Zhang et al., 2009; Liu et al., 2011). The controlling processes are different for surface PM number, which is accumulated mainly by homogeneous nucleation and vertical transport and it is lost mainly by dry deposition and coagulation (Zhang et al., 2010).

For primary components such as black carbon (BC) the fate is similar to that of total $\text{PM}_{2.5}$, while for secondary species it is more intricate. Sulfate is generally produced in the PBL by aerosol and clouds processes (the latter being very important) and ex-

**Particulate matter
and vertical
processes**

G. Curci et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



ported out of the PBL throughout the year (de Meij et al., 2007; Zhang et al., 2009; de Brugh et al., 2011; Liu et al., 2011). Averaged over the year, the nitrate budget is similar to that of sulfate, with the difference that cloud processes (wet deposition) are a sink (de Brugh et al., 2011; Liu et al., 2011). However, during the summer there might be a competition between PM production (e.g. condensation and absorption) and destruction (e.g. evaporation and desorption) processes, and PBL may become a sink and not a source for nitrate (Zhang et al., 2009). The same competition between PM production and destruction processes affect the secondary organic aerosols (SOA) throughout the year (Zhang et al., 2009). Moreover, SOA are strongly influenced by biogenic volatile organic compounds (BVOC) emissions, through semi-volatile products of the oxidation of isoprene and terpenes, which also have a marked seasonal cycle (Zhang et al., 2007; Hodzic et al., 2009).

In the present study, we examine the formation of aerosol near the surface in the particular perspective of the boundary layer vertical processes outlined above. We analyze observations of aerosol mass, composition, number and optical properties in July 2007 in Milan (45° N, 9° E, Northern Italy) during the intensive campaigns carried out in the frame of the Italian projects QUITSAT (“Air Quality by the Integration of Ground- and Satellite-based Observations and Multiphase Chemistry-Transport Modelling”, funded by Italian Space Agency, ASI) and AeroClouds (“Study of Direct and Indirect Effect of Aerosols and Clouds on Climate”, funded by the Italian Ministry for Higher Education) projects. The experimental results are then complemented/interpreted by WRF/Chem model simulations.

We first briefly review what is known about the aerosol phenomenology in the investigated domain in Sect. 2. Then we describe the experimental in Sect. 3 and the model setup in Sect. 4. In Sect. 5, we conduct a preliminary analysis of the observations, in order to characterize the relevant features of the case study and pose questions arising from the picture given by the measurements. We then address these questions using WRF/Chem model simulations. After a model validation against available observations,

we analyze the budget of aerosol species as calculated by the model, focusing in particular on the vertical dimension. We summarize main results in conclusive Sect. 6.

2 The investigated domain

Milan is the largest urban area in Italy (c.a. 5 million people) and lies in one of the most polluted places of Europe, the Po Valley (Putaud et al., 2010). The topography of the valley (closed by the Alps to the North and West, and by the Apennine to the South), under high-pressure systems, favour stagnant atmospheric conditions and recirculation of air through the typical mountain-valley breeze (Dosio et al., 2002). The local circulation in combination with elevated anthropogenic emissions especially from traffic, residential combustion, and agriculture (Lonati et al., 2005; Carnevale et al., 2008; Perrone et al., 2012; Saarikoski et al., 2012) makes it a nitrogen dioxide and aerosol hot-spot well visible from space (e. g. Chu et al., 2003; Barnaba and Gobbi, 2004; Ordonez et al., 2006; van Donkelaar et al., 2010).

At the surface, PM_{10} annual mean in Milan has stabilized between 50 and 60 $\mu\text{g m}^{-3}$ in the last decade (Carnevale et al., 2008; Silibello et al., 2008), thus systematically above the European limit of 40 $\mu\text{g m}^{-3}$ for human protection (EC, 2008). The winter average values are roughly double than those in the summer, and peak values are up to 200 $\mu\text{g m}^{-3}$ (Marcazzan et al., 2001). The main aerosol components are sulfate, nitrate, and organic matter (OM), which account for roughly 20, 15, 40 %, respectively, of PM_{10} mass in summer, and 10, 30, 50 %, respectively, in winter (Marcazzan et al., 2001; Putaud et al., 2002; Lonati et al., 2005; Carbone et al., 2010; Perrone et al., 2010; Daher et al., 2012). These values are similar to other urban areas in the Po Valley (Matta et al., 2003; Carbone et al., 2010; Squizzato et al., 2013). Most of the mass of those species is distributed in the accumulation mode (particle diameter in the range 0.14–1.2 μm), while the coarse mode (1.2–10 μm diameter) has a larger fraction of crustal material and sea salts (Matta et al., 2003; Carbone et al., 2010). In summer a significant fraction of nitrate may also form in the coarse mode, because most of the ammonium

is neutralized by sulfate (Matta et al., 2003; Hodzic et al., 2006; Carbone et al., 2010). The total number concentration of aerosol is of the order of 10^4 cm^{-3} , with the ultrafine (diameter $d < 100 \text{ nm}$) and submicron ($100 < d < 1000 \text{ nm}$) particles making up to 80 and 20 % of the total, respectively (Lonati et al., 2011). The aerosol number concentration is usually distributed in three modes (Balternsperger et al., 2002; Lonati et al., 2011). One mode with diameters in the range of 20–30 nm, consisting of hydrophobic and highly volatile organic material originating from combustion (Baltensperger et al., 2002), plus new particles from nucleation events that occur on about 35 % of the days in Po Valley (Hamed et al., 2007). The other two modes are in the submicron range, one almost hydrophobic, related to primary emissions (e.g. soot), and the other hydrophilic, related to secondary aerosols (Balternsperger et al., 2002).

The aerosol vertical profile in Milan and in the wider Po Valley region was characterized by means of aircraft, Lidar, and tethered balloon measurements (Highwood et al., 2007; Barnaba et al., 2007, 2010; Crosier et al., 2007; Angelini et al., 2009; Ferrero et al., 2010, 2011). Similarly to other polluted places located in a valley, two layers with distinct characteristics are often found. One in the PBL, humid, rich of fresh emissions and with a nitrate profile increasing with height, and another above the PBL with more aged aerosols enriched in the sulfate and organic matter fraction (Highwood et al., 2007; Crosier et al., 2007; Ferrero et al., 2010). The decoupling in two layers is attributed to the mountain-valley breeze dynamics (Angelini et al., 2009) and to the sporadic arrival of long-range transported Saharan dust (Barnaba et al., 2007) or biomass burning plumes (Barnaba et al., 2011). The number concentration of fine mode ($d < 1.6 \mu\text{m}$) particles are found to be relatively constant with height in the PBL, and it decreases by a factor of 2–3 above the PBL. Coarse particle ($d > 1.6 \mu\text{m}$) number concentrations, in contrast, display a decrease with height also in the PBL, due to sedimentation processes (Ferrero et al., 2010).

Particulate matter and vertical processes

G. Curci et al.

[Title Page](#)[Abstract](#)[Introduction](#)[Conclusions](#)[References](#)[Tables](#)[Figures](#)[Back](#)[Close](#)[Full Screen / Esc](#)[Printer-friendly Version](#)[Interactive Discussion](#)

3 Experimental setup

Ground-based and vertical profiles measurements used in this study were conducted at Torre Sarca site which is located on the northern side of Milan (45°31'19" N, 9°12'46" E; within the Milano-Bicocca University campus), in the midst of an extensive conurbation that is the most industrialized and heavily-populated area in the Po Valley. We report here a brief description of the experimental setup and provide relevant references for further details.

3.1 Particulate matter bulk composition and number size distribution, and gas-phase composition

At ground level, PM_{2.5} and PM₁ (EN-14907) samples were gravimetrically collected using the FAI-Hydra dual channel Low-Volume-Sampler (LVS; 2.3 m³ h⁻¹, 24 h of sampling time, PTFE filters for PM₁, ore-fired Quartz fibre filters for PM_{2.5}, $\emptyset = 47$ mm), while the aerosol number-size distribution was constantly monitored using an Optical Particle Counter (OPC; Grimm 1.107 "Environcheck", 31 class-sizes ranging from 0.25 to 32 μ m). Further details are given in Ferrero et al. (2014).

The aerosol chemistry was assessed on PM_{2.5} samples for the ionic fraction, EC and OC, respectively. For the purpose of ions' analysis, PM_{2.5} samples were extracted in 3 mL of ultrapure water (Milli-Q[®]; 18.2 M Ω cm) for 20 min using an ultrasonic bath (SONICA, Soltec, Italy). The obtained solutions were then analysed using a coupled ion chromatography systems consisting of: (1) a Dionex ICS-90 (CS12A-5 Analytical column) with an isocratic elution of methanesulfonic acid (20 mM; 0.5 mL min⁻¹) whose signal was suppressed by means of tetrabutylammonium hydroxide (0.1 M; CMMS III 4 mm MicroMembrane Suppressor) for cations (Na⁺, K⁺, Ca⁺⁺, Mg⁺⁺ NH₄⁺) and, (2) a Dionex ICS-2000 (AS14A-5 analytical columns) with an isocratic solution of Na₂CO₃/NaHCO₃ (8.0 mM/1.0 mM; 1 mL min⁻¹) whose signal was suppressed by means of sulphuric acid (0.05 M; AMMS III 2 mm MicroMembrane Suppressor) for anions (F⁻, Cl⁻, NO₃⁻, SO₄⁻).

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



EC and OC were determined in $PM_{2.5}$ using the Thermal Optical Transmission method (TOT, Sunset Laboratory inc.; NIOSH 5040 procedure, <http://www.cdc.gov/niosh/nmam/pdfs/5040f3.pdf>). The organic matter (OM) fraction was then estimated from OC using a coefficient to account for the presence of hetero-atoms (H, O, N, etc.).

5 Following the work of Turpin and Lim (2001), the chosen factor was 1.6 for the urban Torre Sarca site.

Finally, meteorological and gas-phase (NO_x , O_3) observations at ground-level were taken from the weather and monitoring stations operated in Milan by the local regional environmental protection agency (ARPA Lombardia).

10 3.2 Size-segregated aerosol composition

From 14 (8:00 local time LT) to 18 (8:00 LT) July 2007, size segregated daytime (8:00 to 21:00 LT) and night-time (21:00 to 8:00 LT) aerosol samples were collected by means of a five-stage Berner impactor (LPI 80/0.05) with 50 % size cut at 0.05, 0.14, 0.42, 1.2, 3.5 and $10\mu m$ aerodynamic diameter. Substrates were off-line analyzed for the determination of the carbonaceous – water soluble organic (WSOC) and water insoluble (WINC) carbon – and soluble inorganic components (NH_4^+ , Na^+ , K^+ , Ca^{2+} , Mg^{2+} , Cl^- , NO_3^- , SO_4^{2-}). Mass-to-carbon ratios of 1.8 and 1.2 were used to convert WSOC to the corresponding mass, WSOM (water-soluble organic matter) and WINC to WINCM (water-insoluble carbonaceous matter), respectively. A complete description of the sampling and analytical methods adopted is reported in Carbone et al. (2010) and references therein. In the analysis presented here, we use only the total mass of aerosol components (sum over size bins).

3.3 Lidar-ceilometer profiles

25 Lidar-ceilometers (called Lidar for brevity in the manuscript) operate on the same physical basis of more complex research-type lidars, but are compact systems, generally with a lower laser energy power, capable of operating 24 h per day, unattended and in

Particulate matter and vertical processes

G. Curci et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

⏪

⏩

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



all weather conditions. Initially developed for cloud-base determination, the technology of these system is now mature enough to represent a very convenient and widely used tool for the operational monitoring of the atmospheric aerosol and of relevant meteorological parameters (e.g. Haeffelin et al., 2012).

5 A lidar-ceilometer (Vaisala LD-40) operating at 855 nm collected aerosol profiles at the Milan Torre Sarca site in the period January 2007–February 2008. The system was switched on during selected dates (and mostly when meteorological conditions allowed the contemporary launch of balloon-borne aerosol instruments, Ferrero et al., 2010), collecting a database of more than 200 days of measurements During the selected
10 dates, the Lidar-ceilometer operated 24 h per day, collecting aerosol profiles every 15 s that were afterwards averaged over 15 min to achieve a better signal-to-noise ratio. Due to the instrumental limitations, the lowest altitude the system can observe is about 60 m. After the background noise is subtracted from the collected backscattered signal, the range-corrected signal (RCS, i.e., the signal S times the square of the system-to-
15 target distance R) is derived to extract information on the aerosol vertical distribution. More details on the system and measurements capabilities can be found in Angelini et al. (2009) and Di Giuseppe et al. (2012).

4 WRF/Chem model

4.1 Description and setup

20 The version 3.4.1 of Weather Research and Forecasting model with Chemistry (WRF/Chem), with some updates, is used in order to interpret the observed concentrations of aerosol and its composition at surface and along vertical profile of PBL. WRF/Chem is a coupled on-line model where meteorological and chemical processes are fully consistent (Grell et al., 2005).

25 The model is configured with two 1-way nested domains centred on Northern Italy (Po Valley). The mother domain covers Western Europe with 131×95 cells at a hori-

Particulate matter and vertical processes

G. Curci et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



zontal resolution of 30 km, the nested domain covers Northern Italy with 109×91 cells at a resolution of 10 km. The vertical grid is made of 33 eta levels up to 50 hPa, with first five levels centred approximately at 12, 36, 64, 100 and 140 m above the ground and 12 levels below 1 km.

5 The physical and chemical parameterizations used are the same for the two domains, and are listed in Table 1. These include the Rapid Radiative Transfer Model for short and long wave radiation (Iacono et al., 2008), the Mellor-Yamada Nakanishi-Niino boundary layer parameterization (Nakanishi and Niino, 2006), the Noah Land Surface Model (Chen and Dudhia, 2001), the Morrison cloud microphysics scheme (Morrison et al., 2009), and the Grell 3-D ensemble cumulus scheme, which is an update version of the Grell-Devenyi scheme (Grell and Devenyi, 2002). Cumulus clouds feedback with radiation is activated.

10 The gas-phase chemistry is simulated with an updated version of the Regional Atmospheric Chemistry Mechanism (RACM) that includes a wide range of chemical and photolytic reaction for organic and inorganic species (Stockwell et al., 1997). Aerosol parameterization adopted is the Modal Aerosol Dynamic for Europe (Ackermann et al., 1998) that uses three overlapping lognormal modes for Aitken, accumulation and coarse particles. Thermodynamic equilibrium for inorganic species is calculated with the RPMARES module (Saxena et al., 1986; Binkowski and Roselle, 2003). The Secondary Organic Aerosol (SOA) production is calculated using the Volatility Basis Set (VBS) scheme implemented in WRF/Chem by Ahmadov et al. (2012). To our knowledge, this study is the first application over Europe of this new parameterization for SOA yield with WRF/Chem. Photolysis rates are estimated with the Fast-J scheme (Wild et al., 2000). The dry deposition flux is simulated with the scheme of Wesely et al. (1989), and the dry deposition velocity of organic vapours is assumed to be the 25% that of nitric acid (HNO_3). Cloud chemistry in convective updraft is parameterized following Walcek and Taylor (1986). Wet deposition by convective and large scale precipitation is also included in our simulations. The aerosol optical properties are calculated online with the package of Barnard et al. (2010), using the volume average

Particulate matter and vertical processes

G. Curci et al.

[Title Page](#)[Abstract](#)[Introduction](#)[Conclusions](#)[References](#)[Tables](#)[Figures](#)[Back](#)[Close](#)[Full Screen / Esc](#)[Printer-friendly Version](#)[Interactive Discussion](#)

internal mixing assumption. We include the direct effect of aerosol on radiation, but we switch off the indirect aerosol effects on clouds, a function that is still under testing with the SOA VBS scheme (Tuccella et al., 2014).

In order to help the understanding of the influence of the upper levels processes on the pollutant budget at surface, we use the diagnostic of the tendency terms in the continuity equation for chemical species following Wong et al. (2009). We extend the original module, which included only some gas-phase compounds, also to aerosol species and processes. Diagnosed terms are: emission, horizontal and vertical advection, photochemistry (gases and aerosols), vertical mixing plus dry deposition (these cannot be separated in the WRF/Chem implementation), convective transport, aqueous chemistry, and wet deposition.

We simulate the period from the 25 June to the 18 July 2007, and discard the first 10 days as spin up. Simulation on the mother domain uses initial and boundary meteorological conditions provided by the National Center for Environmental Prediction (NCEP) 6 hourly analyses, having an horizontal resolution of $1^\circ \times 1^\circ$. For the mother domain, chemical boundary conditions are provided with WRF/Chem default idealized vertical profiles, representative of Northern hemispheric, mid-latitude and clean environmental conditions (McKeen et al., 2002; Grell et al., 2005; Tuccella et al., 2012), while boundary conditions to the nested domain are provided by the mother domain. The simulations are carried out at 24 h time-slots, starting at 12:00 UTC of each day and then run for 30 h, with first 6 h considered as model spin-up. Chemical fields are restarted from previous runs.

4.2 Emissions

Total annual 2007 anthropogenic emissions of nitrogen oxides (NO_x), carbon monoxide (CO), sulphur oxides (SO_x), ammonia (NH_3), Non-Methane Volatile Organic Compounds (NMVOC), unspiciated particulate matter ($\text{PM}_{2.5}$ and coarse PM), primary organic carbon (OC), and elemental carbon (EC) are taken from the Netherlands Organization for Applied Scientific Research (TNO) database (Kuenen et al., 2014). Annual

Particulate matter and vertical processes

G. Curci et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



TNO anthropogenic emissions consist of gridded data from ten source types (SNAP sectors) with horizontal resolution of $1/16^\circ$ latitude by $1/8^\circ$ longitude (about $7\text{ km} \times 7\text{ km}$).

TNO emissions are adapted to WRF/Chem following the methodology used by Tuccella et al. (2012), with minor changes derived from the second phase of the Air Quality Modelling Evaluation International Initiative (AQMEII) (Alapaty et al., 2012).

Biogenic emissions are calculated online using the Model of Emissions of Gases and Aerosols from Nature (MEGAN) (Guenther et al., 2006). Sea salt flux is calculated online, while dust source is not included.

5 Results

5.1 Preliminary analysis of the observations

In Fig. 1 we show timeseries of ground-based meteorological and physical-chemical observations performed in Milan in the period 5–20 July 2007. The large scale circulation is illustrated in Fig. S1, while the evolution of cloud cover over Northern Italy is illustrated by MODIS-Aqua true colour images in Fig. S2. The period starts with a low-pressure system over Germany, rapidly moving Eastward, and allowing a pressure increase over Northern Italy from 5 to 8 July, associated with fair weather and sparse clouds. From 9 to 11 July, a North Atlantic low-pressure system induces a significant increase of cloud cover over Milan with light rain on 10 July. From 12 July, a wide anticyclonic system forms over the Western Mediterranean, warranting clear sky and stable conditions until 20 July and beyond. Maximum daily temperature is around 30°C before the Atlantic perturbation, then it increases steadily (from 25° to 35°C) at a rate of $\sim 2^\circ\text{ day}^{-1}$ from 11 to 15 as the high-pressure system settles. Humidity is high at night (above 70%) on the days following the low-pressure passage, then the atmosphere gradually dries out under the anticyclone.

During the period preceding the Atlantic perturbation (5–8 July 2007), wind is prevalently westerly daytime, forced by the large scale circulation, with wind speed around

Particulate matter and vertical processes

G. Curci et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



$2.5\text{--}3\text{ m s}^{-1}$. Wind is slowed down to less than 1 m s^{-1} at night, because the downward transport of momentum toward the surface is inhibited by the nighttime vertical stratification (Stull, 1988; Whiteman, 1990). Wind speed increases up to 5 m s^{-1} at the passage of the low-pressure system (9–11 July 2007), and stays above 2 m s^{-1} also nighttime. From 11 July, when the high-pressure over the Mediterranean begins to settle, the wind field adjusts to a typical mountain-valley breeze regime (Whiteman, 1990). Starting from midnight, the slow ($\sim 1\text{ m s}^{-1}$) northerly flow gradually accelerates and rotates clockwise, reaching peak speeds of $\sim 3\text{ m s}^{-1}$ in the afternoon at south-westerly direction, then gradually slows down and return northerly. This wind pattern favours conditions of stagnation and recirculation of air within the valley, allowing the build-up of pollutants from a day to the next. Figure S3 shows the simple stagnation and recirculation indices proposed by Allwine and Whiteman (1994) and confirms that the only ventilated period is that of the Atlantic perturbation.

The passage of the Atlantic low-pressure system on 9–10 July marks a sort of “restart” for the atmospheric composition at ground level. Indeed, relatively longer lived (few days) chemical species, such as ozone and PM, first accumulate during the days preceding the perturbation, then are suppressed in perturbed weather, and finally re-accumulate afterwards (Fig. 1c and d). Outside the perturbed period, ozone and nitrogen oxides (NO_x) follow a daily cycle typical of that observed in many urban areas (Mavroidis and Iliia, 2012, and references therein). The primary pollutant nitric oxide (NO) display a sharp peak during morning rush hours (between 6:00 and 9:00 LST), then gradually decreases during the day. It displays a secondary small peak during evening rush hours (20:00–22:00 LST), then stays at low values until the following morning. Nitrogen dioxide (NO_2) is mainly originated from the oxidation of NO by ozone and peroxy radicals (Jenkin and Clemitshaw, 2000), and displays peaks delayed by $\sim 1\text{ h}$ with respect to those of NO. It shows a plateau between the morning and the evening peak, because concentrations are sustained daytime by photochemistry. The photolysis of NO_2 is the main tropospheric source of atomic oxygen (O) that readily reacts with molecular oxygen (O_2) to produce ozone. Indeed, during daylight hours, NO,

NO₂ and O₃ equilibrate on the so called “photostationary equilibrium” on time scales of minutes (Clapp and Jenkin, 2001).

Ozone is depleted during the morning rush hours by reaction with NO, then it is photochemically formed during the day and peaks during late afternoon (14:00–16:00 LST), and thereafter gradually decreases to lower nighttime levels. In fair weather, the daily cycle of ozone and NO_x is regulated by the solar radiation, the dilution of fresh emissions from the surface in the growing daytime PBL, the vertical mixing with air entrained from the residual layer and the free troposphere above the PBL, and the dry deposition at the surface. Past studies pointed out that the entrainment from ozone-rich residual layer may be as important as the photochemical production in the PBL during pollution events even in urban atmospheres (e.g. Zhang and Rao, 1999). In the present case, the build-up of ozone in the days following the perturbation is evident, but it is difficult to discern the relative role played by the local photochemical production and by the vertical mixing on the ozone trend observed at the surface.

Accumulation and cleansing of the atmosphere near the surface is even more evident from aerosol timeseries (Fig. 1d–g). PM_{2.5} and PM₁ follow a similar trend, while PM₁₀ often show a different behaviour, pointing out the presence of additional sources to the coarse fraction, most probably the erosion and resuspension of soil material by vehicles and wind. The aerosol mass is shown to build up before the Atlantic perturbation (PM₁₀ around 20–30 μg m⁻³) and to abruptly decrease (PM₁₀ below 10 μg m⁻³) during the low-pressure system passage (probably because of a combination of enhanced ventilation, wet deposition processes, and soil erosion inhibited by increased soil moisture). Afterwards, PM concentration keep increasing after the low-pressure passage (maximum PM₁₀ values of more than 60 μg m⁻³ reached on 18–19 July). Daily cycle of the fine aerosol mass (PM_{2.5} and PM₁) displays similarities with that of NO, in particular a similar morning peak, indicating the important role played by primary emissions. This is confirmed by the analysis of aerosol speciation (Fig. 1e), which shows high values of elemental carbon (EC, 2–4 μg m⁻³) and insoluble carbonaceous matter (WINCM, 2–10 μg m⁻³). The latter makes, on average, 40–50 % of the PM₁ mass (Carbone et al.,

Particulate matter and vertical processes

G. Curci et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



2010). Major secondary species are inorganic ions (sulfate, nitrate, and ammonium) and part of the organic matter, which may be associated with its water soluble fraction (WSOM, Carbone et al., 2010). Similarly to ozone, secondary aerosol accumulates during the days preceding and following the perturbation.

5 Cleansing of the atmosphere after the perturbation and subsequent recover of the aerosol load is also clearly visible in the number concentration timeseries. At the passage of the perturbation, aerosol number rapidly decreases by more than an order of magnitude at all observed size ranges, then returns to the pre-perturbation levels on a time scale of about two days. We note, however, differences in the aerosol regime
10 before and after the perturbation. Before the cleansing, the aerosol size distribution is locked to a fixed shape, with no or little daily variability. Conversely, in the stable conditions of 12–19 July, it displays a clear daily cycle with a growth towards larger sizes in daytime, and a return to narrower distributions nighttime.

As mentioned in Sect. 3.3, Lidar observations are only available in the days following
15 the perturbation and give useful indications on the aerosol vertically-resolved infra- and inter-diurnal variability (e.g. Angelini et al., 2009). During the morning hours, a layer of aerosol is formed under the growing boundary layer. There, fresh emissions from the surface are diluted and mixed vertically in the PBL. Throughout the period, but especially on some days such as in the mornings of 13 and 15 July, an enhanced layer
20 of aerosol is visible in the upper levels near the top of the PBL. Aerosol is subsequently partly removed in the second half of the day by the mountain breeze, while a residual layer with relatively high aerosol content may survive above the nocturnal PBL (e.g. on 13, 15, and 16 July). This layer may potentially be entrained the following morning into the PBL and contribute to the surface aerosol budget. On the last days displayed
25 in Fig. 1, a further aerosol layer between 2 and 3 km appears in the Lidar signal. As indicated by increased coarse fraction AOD at Modena AERONET station (Fig. S4) and model backtrajectories (Fig. S5), it is a Saharan dust incursion which is probably entrained at ground level, as indicated by the enhancement of PM_{10} levels on days 18–19 July.

Particulate matter and vertical processes

G. Curci et al.

[Title Page](#)[Abstract](#)[Introduction](#)[Conclusions](#)[References](#)[Tables](#)[Figures](#)[Back](#)[Close](#)[Full Screen / Esc](#)[Printer-friendly Version](#)[Interactive Discussion](#)

**Particulate matter
and vertical
processes**

G. Curci et al.

[Title Page](#)[Abstract](#)[Introduction](#)[Conclusions](#)[References](#)[Tables](#)[Figures](#)[Back](#)[Close](#)[Full Screen / Esc](#)[Printer-friendly Version](#)[Interactive Discussion](#)

From the measurements we reported here, at least a few questions, which we address in what follows, emerge:

1. What is the composition of the aerosol layer formed during the day in the upper PBL?
2. How much of the aerosol burden measured at the ground is due to localized processes and how much is conversely due to processes occurring in the upper PBL and to the subsequent mixing in the lowermost levels? In other words, how important is the interplay between surface and upper layers in shaping the aerosol mass we measure near the ground?
3. How much the residual layer above the PBL may contribute to the aerosol budget at ground level the next day?

We attempt to answer these questions using simulations with the WRF/Chem model and relevant comparison with the observational dataset.

5.2 Model verification against available observations

Before drawing conclusions on the scientific questions outlined at the end of the previous section, we verify our model simulations against the dataset of observations depicted in Fig. 1. We display results only for the nested domain over Northern Italy, and use statistical indices defined in Appendix A as a guidance to quantify model biases.

In Fig. 2 we compare observed and simulated meteorological variables at ground level in Milan for the period 5–17 July 2007. The temperature is underestimated by about 2.5°C , which is probably due to not well resolved dynamics and heat fluxes in the urban boundary layer. The overestimation of relative humidity of about 10% is mostly attributable to the underestimation of temperature. Wind speed is overestimated by 0.8 m s^{-1} and has a relatively low correlation of 0.29 with observations, thus fitting to typical characteristics of current mesoscale models (e.g Misenis and Zhang, 2010). The simulated wind speed is also more variable than that observed as denoted by

the RMSE of 1.7 m s^{-1} . The wind direction is generally captured well, in particular the mountain-valley cycles after the passage of the perturbation of 9 July.

In Fig. 3 we show comparison of gas-phase observations and simulation near the ground. The daily cycle of NO is reproduced quite well ($r = 0.52$), the timings of the morning peak and the subsequent decrease are captured by the model. The magnitude of the morning peak does not show a tendency neither to underestimation nor to overestimation, while NO values for the rest of the day are underestimated, resulting in a bias of -4.1 ppb (-60%). The model is also able to capture the basic features of the NO₂ daily cycle, i.e. the morning and evening peaks and the minimum at night. However, values are generally underestimated (bias of -8.3 ppb or -34%) and the trend on weekly time scale display much less variability than that observed. Ozone display a very low systematic bias (-2.3 ppb), but less variability than observations (RMSE of 11.3 ppb), and a correlation of 0.65 . The timing of the daily cycle is captured well, with a maximum in the afternoon, a secondary peak around midnight, and a minimum during the morning rush hour.

In Fig. 4 we compare PM₁₀ and PM_{2.5} simulated mass to hourly observations at ground. The PM₁₀ trend is qualitatively captured by the model, displaying the sharp decrease at the passage of the perturbation on 10 July and the subsequent gradual accumulation in the following days. This leans confidence in the simulated removal and production terms, and the resulting negative bias is small ($-4 \mu\text{g m}^{-3}$ or -10%). The model also captures some of the characteristics of the daily cycle ($r = 0.57$), however the observed signal is quite irregular, and the model does not reproduce all the variability. For PM_{2.5} the general features of the comparison are similar to PM₁₀, but the model has a positive bias ($+4 \mu\text{g m}^{-3}$ or $+70\%$), mostly attributable to few spurious peaks in the simulation. The overestimation of PM_{2.5} partly compensates and masks the underestimation of coarse particles (PM_{2.5-10}). The comparison of the simulated number size distribution against that observed with the OPC (not shown) suggests that the high bias of PM_{2.5} is attributable to aerosol in the size range $0.5\text{--}1 \mu\text{m}$.

Particulate matter and vertical processes

G. Curci et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

⏪

⏩

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



**Particulate matter
and vertical
processes**

G. Curci et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



In Fig. 5 we show the comparison of simulated $\text{PM}_{2.5}$ composition with daily and bi-daily samplings near the ground. In the period precedent to the perturbation (5–9 July), the model underestimates the magnitude of the observed peak of sulfate and ammonium, but it reproduces well the subsequent “restart” and recovery. Observed nitrate displays a relatively smooth trend, with a slight decrease at the passage of perturbation and almost constant levels during the rest of the period. Modelled nitrate has a much more variable behaviour, which looks like characterized by sudden and irregular pulses. The bi-daily observations indeed suggest that the daily average observation masks much of the underlying variability associated to nitrate. Elemental carbon, being primary and almost hydrophobic, is largely unaffected by the perturbation. This feature is captured by the model, but EC values are underestimated by a factor of two, probably due to underestimated emissions. Interestingly, the bi-daily observations of WINCM (EC plus primary insoluble organic material) display a large diurnal cycle (maximum at night and minimum during the day) which is not captured by the model. Organic carbon trend and magnitude is reproduced quite well, with the exception of a large spurious peak on 8–9 July not seen in the observations. The peak is associated with secondary organic aerosol (not shown). The bi-daily observations of soluble organic material (WSOM) do not show the strong daily cycle of primary carbonaceous matter, and confirm a tendency of the model at overestimating the SOA fraction.

In Fig. 6 we qualitatively compare the Lidar profiles with the simulated $\text{PM}_{2.5}$ profiles. A quantitative comparison would require the calculation of optical properties of simulated $\text{PM}_{2.5}$ and subsequent solution of the Lidar equation (Hodzic et al., 2006). However, in first approximation Lidar signal may be associated to $\text{PM}_{2.5}$ mass. The model captures some of the basic features of the previously described aerosol profile cycle observed in this period (Sect. 5.1). Every morning a plume of fresh aerosol detaches from the ground and follows/traces the growing boundary layer until its maximum extension in the central part of the day. Then, in the evening, the mountain-valley breeze cleans the lower PBL, often leaving an upper air aerosol residual layer above the PBL. When the residual layer survives the night, it might be entrained down to the ground on

the subsequent morning. This is visible e.g. on 16–17 July, in both observations and simulations.

5.3 Insights into the budget of aerosol vertical profile over Milan

The “chemical restart” caused by the passage of the perturbation on 9–10 July, and the following settle of an almost periodic circulation pattern, naturally creates favourable conditions for a study of the processes yielding aerosol production and accumulation in the area of Milan. We thus now focus our analysis on the days that follow the perturbation (12–17 July).

Using model output, we first examine the composition of the aerosol layers noted in the Lidar profiles of Fig. 6. In Fig. 7, we show the composition of $PM_{2.5}$ simulated over Milan. The model predicts a major role played by the primary fraction (inorganic and organic), which is largely responsible for the two rush hours peaks (morning and evening) and the bulk of aerosol mass in the PBL. Fresh emissions are mostly concentrated near the ground and turbulent transport dilutes them in the PBL during the day. A relatively small fraction ($\sim 30\%$) of primary aerosol remains above the PBL overnight and contributes to the upper aerosol layers seen by the Lidar.

The sum of secondary species contributes 40–60% of the aerosol mass in the PBL, but with remarkable differences in the vertical distribution of single components. Sulfate and Secondary Organic Aerosol (SOA) start to form and dilute under the PBL a few hours after the sunrise, contributing in a relatively homogeneous way to the aerosol column in the PBL. Anthropogenic SOA (ASOA) contributes more than biogenic SOA (BSOA) to the SOA budget. The concentration of those secondary species are similar also above the PBL, thus significantly contributing to the upper aerosol layers. ASOA are slightly more persistent than BSOA and sulfate in the free troposphere.

Nitrate displays a peculiar profile, with enhanced concentrations in the upper part of the PBL formed during the central part of the day. These concentrations may largely exceed those found near the ground (i.e. on 13, 16, 17 July). Moreover, nitrate is predicted to be the major secondary species contributing to the formation of the residual aerosol

Particulate matter and vertical processes

G. Curci et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



layers above the PBL. Enhanced upper level concentrations of nitrate into PM₁ were also reported at Monte Cimone (a mountain peak of 2160 m at the southern border of Po Valley) by Carbone et al. (2010, 2014).

In Fig. 8 we show the maps of simulated sulfate and nitrate over Po Valley on 13 July 2007 at 16:00 LST at the surface and at 750 m height. It can be seen that the main features of the composition of the aerosol profile outlined above are not peculiar of Milan area, but are suggested to be representative of the larger area of Po Valley.

In order to better understand the processes underlying the predicted characteristics of the aerosol over Milan, we look at the terms of the continuity equation for chemical species. Budget terms considered are horizontal and vertical advection, chemistry, turbulent mixing and dry deposition, emission. Terms related to cloud processes (convection, aqueous chemistry, wet deposition) make a very small contribution in the dry period under investigation and will not be shown to improve figure's clarity. In Fig. 9 we show the vertical profile of the budget terms for sulfate and nitrate at 16:00 LT of 13 July over Milan. For sulfate, the dominant terms are those related to advection, indicating the presence of spatially distributed sources and a relatively long lifetime, making it a regional scale pollutant. Locally, sulfate is both directly emitted and produced by secondary pathways throughout the PBL. Turbulent mixing distributes it vertically in the PBL and dry deposition removes it from the atmosphere near the ground, determining an almost homogeneous sulfate profile in the PBL. Conversely, nitrate has relatively low contribution from advection, while the largest terms are chemistry and vertical mixing. In the simulation, nitrate is produced only in the upper half of the PBL and destroyed in the lower half. The vertical transition between the nitrate destruction and production zone is quite sharp. Turbulent mixing is nearly in equilibrium with chemical production, indicating that model simulates a very rapid adjustment to the thermodynamic equilibrium for the sulfate-nitrate-ammonium system. This results in nitrate concentrations higher in the upper part of the PBL with respect to the lower part.

Similarly to nitrate, also SOA displays an enhanced net chemical production in the upper part of the PBL and destruction in the lower part (Fig. 10), but since the chemical

Particulate matter and vertical processes

G. Curci et al.

[Title Page](#)[Abstract](#)[Introduction](#)[Conclusions](#)[References](#)[Tables](#)[Figures](#)[Back](#)[Close](#)[Full Screen / Esc](#)[Printer-friendly Version](#)[Interactive Discussion](#)

and vertical mixing terms are of the same order of the advection terms the resulting vertical profile is almost constant with height, similar to that of sulfate.

Further insights into the simulated sharp transition to an environment favourable to nitrate formation in the upper part of the PBL, is investigated by means of several model sensitivity tests as outlined in Table 2. In Fig. 10 we first look at the gas phase precursor of nitrate, nitric acid (HNO_3). The left panel shows the vertical profile of the budget terms for HNO_3 at the same instant of Fig. 9. The chemical and vertical mixing terms mirror those of particulate nitrate, resulting in a decreasing concentration profile with height. The right panel of Fig. 9 shows the budget profile from a sensitivity simulation where aerosol chemistry is switched off (AERO, see Table 2). The chemistry and vertical mixing terms are greatly reduced and are the same order of magnitude of advective terms, indicating that the sharp gradients in net chemical production of HNO_3 (and nitrate) are dominated by aerosol processes, and not by gas-phase processes.

In Fig. 12 we provide further elements to evaluate the simulated particulate nitrate thermodynamics. Ambient relative humidity increases with height in the PBL, from a minimum of $\sim 50\%$ near the ground to a maximum of $\sim 80\%$ at an altitude of 1000 m (~ 400 m below the PBL top). The nitrate chemical production term shown in Fig. 9 is reported for ease of comparison, and displays the already noted peak between 500 and 1000 m. The sulfate ratio (ratio of total ammonia and sulfate) is well above the threshold of 2 along the profiles (not shown), thus suitable for particulate nitrate formation (Seinfeld and Pandis, 2006). The profile of equilibrium constants for both the aqueous and solid nitrate increase with height, in response to a decreasing temperature profile (not shown), indicating that conversion of nitric acid to particulate is favoured with increasing height. However, no sharp transitions, correlated to the nitrate net chemical term, can be noticed in the profiles of those equilibrium constants.

The profile of ammonium nitrate's deliquescence relative humidity (DRH) helps disclosing the possible reason for such a transition. At ground level, ambient RH is well below the ammonium nitrate DRH, indicating an environment thermodynamically favourable only to the solid form of nitrate. However, since the RH gradient with height

Particulate matter and vertical processes

G. Curci et al.

[Title Page](#)[Abstract](#)[Introduction](#)[Conclusions](#)[References](#)[Tables](#)[Figures](#)[Back](#)[Close](#)[Full Screen / Esc](#)[Printer-friendly Version](#)[Interactive Discussion](#)

**Particulate matter
and vertical
processes**

G. Curci et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



is steeper than that of DRH, the two curves intersect at an altitude of ~ 500 m, and then again at ~ 1300 m, because of the RH decrease near the PBL top. Ambient RH is thus higher than ammonium nitrate DRH in the same altitude range (~ 500 – 1000 m) where the nitrate net chemical production peaks. This indicates that, over Milan and in the period under consideration, the nitrate chemical production is dominated by aqueous conversion of nitric acid to nitrate ion, condition that is reached only in the upper part of the PBL, where RH levels are high enough to sustain the formation of an aqueous solution containing nitrate.

The budget analysis we presented so far reveals of a complex interplay between chemical processes and vertical mixing taking place at different altitude ranges. In order to better quantify the impact of chemical production at upper layers on particulate matter at ground level, we perform three tests alternatively switching on/off the chemical process at selected altitude ranges (namely within the lower half of the PBL, the upper half of the PBL and above the PBL, see Table 2). Results are shown in Fig. 13 for $PM_{2.5}$, and its components sulfate, nitrate and SOA. In the figure, the contribution to the ground $PM_{2.5}$ of the chemical processes in the different altitude ranges is positive/negative when the associated sensitivity line is below/above the CTRL. For $PM_{2.5}$, we find that chemical process in all regions positively contribute to the ground level concentration. During the first days after the passage of the perturbation, the shutdown of secondary chemical formation makes very little difference, indicating a dominance of primary emissions. As the time goes by, secondary processes gain importance, but primary fraction remains the main driver of $PM_{2.5}$ concentration even after a week. Interestingly, the magnitude of the relative contribution of the different layers (lower PBL, upper PBL, above PBL) to ground level $PM_{2.5}$ is comparable, and of the order of up to 7 – $8 \mu g m^{-3}$ each, on hourly basis. Exceptions are noted on afternoons of 13 and 16 July, when a negative contribution from secondary processes in the lower PBL is simulated (note the blue dashed line above the red line). These peaks are associated with the nitrate sink in the lower PBL (see panel c). Sulfate has an identical contribution from lower and upper PBL chemical production, and may also have a very important

contribution from the region above the PBL, even higher than processes in the PBL (e.g. on 17 July). SOA budget is similar to that of sulfate, but with an enhanced contribution from PBL processes vs. those above it. As expected, nitrate displays distinctive features. Chemical production in the lower PBL positively contributes to ground level concentration in the first part of the day, then in the afternoon results in a net destruction. On the other hand, processes in the upper PBL and above PBL always positively contribute to the ground level nitrate concentrations.

A further quantitative assessment of the impact of upper aerosol layers on ground concentrations can be estimated combining information in Figs. 14 and 6. In Fig. 14 we show the timeseries of the difference in the simulated $PM_{2.5}$ profile between APBL and CTRL runs. When a residual layer is visible, we may roughly estimate from the figure the related change near the surface on the subsequent morning. We focus the attention on 17 July, when the presence of a residual layer is clearly visible. The concentration change (APBL – CTRL) in the residual layer is about $8\text{--}10\ \mu\text{g m}^{-3}$. The following morning the concentration change near the surface is $4\text{--}5\ \mu\text{g m}^{-3}$, thus we may estimate a 50 % sensitivity of ground $PM_{2.5}$ to a change in the residual layer. In Fig. 6b, we see that on 17 July the $PM_{2.5}$ concentration in residual layer is $20\text{--}24\ \mu\text{g m}^{-3}$, thus the expected impact on hourly concentrations near the ground is of the order of $10\text{--}12\ \mu\text{g m}^{-3}$, or about 40 % of the $PM_{2.5}$ concentration near the ground. This is the extreme case in the short period analyzed here, but gives a feeling of the potential importance that entrainment of aerosol layers aloft may occasionally have on $PM_{2.5}$ observed near the surface.

6 Conclusions

The object of this study is the analysis of the role played by the combination of chemical and dynamical processes occurring throughout and above the PBL in determining the aerosol concentration and composition we observe near the ground. We analyzed the observations of the atmospheric composition during a period of two weeks carried

Particulate matter and vertical processes

G. Curci et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



**Particulate matter
and vertical
processes**

G. Curci et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



out in Milan (Northern Italy) during July 2007. The period was characterized by the passage of a perturbation that favoured cleansing of the Po Valley, providing a natural “chemical restart”. After the perturbation, stable high-pressure conditions determined the establishment of a nearly repetitive meteorological pattern, driven by mountain-valley breeze system, that allowed for a gradual re-accumulation of pollutants.

Lidar observations after the “chemical restart” reveal intriguing features of the aerosol vertical profile over Milan. Every morning, a plume of fresh emissions from the ground is dispersed in the growing convective boundary layer. In the afternoon, an enhanced aerosol layer appears in the upper part of the PBL, while in the evening the bottom part of the PBL is cleansed by the mountain breeze. A residual aerosol layer may form and survive the night above the PBL, and may be entrained again down to the ground the day after. We investigated how this “vertical” sequence of processes affect the aerosol concentrations observed at ground level.

With the help of simulations from the state-of-art online meteorology-chemistry model WRF/Chem we attempted to answer to three main questions suggested by the observations. The questions and the related answers are summarized here below:

- What is the composition of the aerosol layer formed during the day in the upper PBL?

Model simulations suggest that 40–60 % of the fine aerosol in the Milan summer PBL is of primary origin, consistently with previous studies (e.g. Carbone et al., 2010). This primary fraction displays a decreasing concentration profile with height in the PBL, since the sources are concentrated near the ground and species are vertically mixed by turbulence. Sulfate and secondary organic aerosol are produced throughout the PBL and have a nearly homogeneous profile there. Nitrate and ammonium have a distinct profile, with enhanced values in the upper PBL, where concentrations may be much higher than those near the ground. The low temperature and the relative humidity above the ammonium nitrate deliquescence point in the upper PBL is predicted to determine this peculiar profile. Nitrate

is the major component of the upper PBL aerosol layer, contributing up to 30 % of the aerosol mass.

- How much of the aerosol burden measured at the ground is due to localized processes and how much is conversely due to processes occurring in the upper PBL and to the subsequent mixing in the lowermost levels? In other words, how important is the interplay between surface and upper layers in shaping the aerosol mass we measure near the ground?

For $PM_{2.5}$ mass, our calculations indicate that in the upper PBL secondary aerosol are formed and then mixed in the PBL by turbulence. The importance of the secondary fraction increases with the aging of air masses, as shown by the progression of days from the “chemical restart”. A week after the perturbation, secondary $PM_{2.5}$ produced in the upper PBL may contribute up to $7\text{--}8\ \mu\text{g m}^{-3}$ (or 25 %) to ground level hourly concentrations. Sulfate and SOA production is equally shared by bottom and upper PBL, while nitrate is mostly produced in the upper PBL, the bottom PBL acting as a sink during the afternoon.

- How much the residual layer above the PBL may contribute to the aerosol budget at ground level the next day?

We calculate that the chemical production above the PBL significantly impacts aerosol levels near the ground, sometimes overtaking the contribution from the production term in the PBL (especially for sulfate and SOA). We estimate that the residual layer above the PBL, which is formed by both primary and secondary species, may occasionally contribute up to $10\text{--}12\ \mu\text{g m}^{-3}$ (or 40 %) to ground level $PM_{2.5}$ hourly concentrations during the following morning.

The peculiar features of the vertical profile of aerosol nitrate already emerged in past studies. Neuman et al. (2003) reported aircraft observations of increasing nitrate profiles with height, and attributed those to the favourable lower temperature in the upper layers, with respect to bottom PBL, for the conversion of nitric acid to aerosol nitrate.

Particulate matter and vertical processes

G. Curci et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



We confirm their conclusion, and add that a key role in shaping the aerosol nitrate production profile is played by the relative humidity. In particular, nitrate production may be enhanced when RH is above the ammonium nitrate deliquescence point.

This study put the emphasis on some less obvious and recognized aspects of the aerosol vertical profile budget. It is based on the analysis of a short period of high pressure conditions in summer over the area of Milan, thus further analyses are recommended for winter periods and different meteorological and geographical conditions. Moreover, it clearly reminds that the interplay between chemical and dynamical processes must be considered when interpreting atmospheric chemistry observations near the ground, and that more observational constraints (e.g. profiles of the aerosol composition in and above the PBL) would certainly be helpful for a better simulation of those processes.

Appendix A: Definition of statistical indices used in model to observations comparison

Let Obs_i and Mod_i be the observed and modeled values at time i , and N the number of observations.

– The Pearson's Correlation (r):

$$r = \frac{1}{N} \sum_{i=1}^N Z_i(\text{Mod}) \cdot Z_i(\text{Obs})$$

$$Z(X) = \frac{X - \langle X \rangle}{\sigma_X}$$

where X is a generic vector, $Z(X)$ is its standard score, and σ_X is the standard deviation.

Particulate matter and vertical processes

G. Curci et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



– Bias:

$$\text{Bias} = \frac{1}{N} \left(\sum_{i=1}^N \text{Mod}_i - \text{Obs}_i \right)$$

– Normalized Mean Bias (NMB):

$$\text{NMB} = \frac{1}{N} \sum_{i=1}^N \frac{\text{Mod}_i - \text{Obs}_i}{\text{Obs}_i} \times 100$$

– Root Mean Square Error (RMSE):

$$\text{RMSE} = \sqrt{\frac{1}{N} \left(\sum_{i=1}^N (\text{Mod}_i - \text{Obs}_i)^2 \right)}$$

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Particulate matter and vertical processes

G. Curci et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



Particulate matter and vertical processes

G. Curci et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



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Particulate matter and vertical processes

G. Curci et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



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**Particulate matter
and vertical
processes**

G. Curci et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



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**Particulate matter
and vertical
processes**

G. Curci et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



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**Particulate matter
and vertical
processes**

G. Curci et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



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- 30

**Particulate matter
and vertical
processes**

G. Curci et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



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**Particulate matter
and vertical
processes**

G. Curci et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



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- 30

**Particulate matter
and vertical
processes**

G. Curci et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



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**Particulate matter
and vertical
processes**

G. Curci et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



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**Particulate matter
and vertical
processes**

G. Curci et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



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ACPD

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**Particulate matter
and vertical
processes**

G. Curci et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



**Particulate matter
and vertical
processes**

G. Curci et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion

**Table 1.** Main physical and chemical parameterizations used in WRF/Chem simulations.

7	Process	8	Scheme
9	Short-wave radiation	10	RRTM
11	Long-wave radiation	12	RRTM
13	Surface Layer	14	Monin–Obukov
15	Boundary Layer	16	MYNN
17	Land surface model	18	Noah LSM
19	Cumulus convection	20	Grell scheme G3
21	Cloud microphysics	22	Morrison
23	Gas-phase mechanism	24	RACM-ESRL
25	Aerosol mechanism	26	MADE/SOA-VBS
27	Photolysis	28	Fast-J
29	Cloud chemistry and wet deposition	30	On
31	Biogenic emissions	32	MEGAN
33	Direct aerosol effect	34	On
35	Indirect aerosol effects	36	Off

Particulate matter and vertical processes

G. Curci et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion

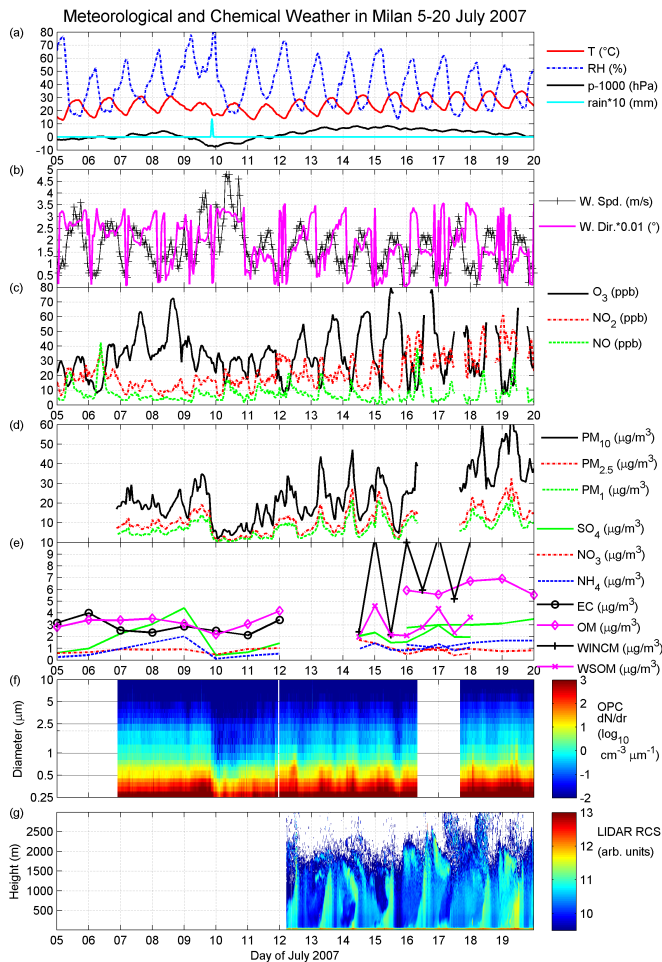


Table 2. Description of sensitivity tests with WRF/Chem model.

37	Label	38	Description
39	CTRL	40	Reference run, see Table 1.
	AERO		Aerosol chemical processes switched off
	LPBL		Gas and aerosol chemical processes switched off in the lower half of the PBL
	UPBL		Gas and aerosol chemical processes switched off in the upper half of the PBL
	APBL		Gas and aerosol chemical processes switched off above the PBL

Particulate matter and vertical processes

G. Curci et al.



Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



Figure 1. Ground-based observations in Milan during 5–20 July 2007. Panel **(a)** shows hourly measurements of temperature, relative humidity, pressure and precipitation. Pressure is subtracted by 1000 and precipitation is multiplied by 10 in order to fit the same y axis. **(b)** Hourly wind speed and wind direction (0° from the North, 90° from the East), the latter divided by 100 to fit the same y axis. **(c)** Hourly ozone, nitrogen dioxide and nitrogen oxide. **(d)** Particulate matter mass. Hourly observations of PM_{10} , $PM_{2.5}$, and PM_1 . **(d)** Particulate matter composition. Daily data of sulfate, nitrate, ammonium, elemental carbon and organic matter collected during QUITSAT campaign. Night-time (21:00 to 8:00 LST) and day-time (8:00 to 21:00 LST) samples of sulfate, nitrate, ammonium, water-insoluble carbonaceous matter (WINCM) and water-soluble organic matter (WSOM) collected during AeroClouds campaign (14–17 July). **(e)** Particulate matter number size distribution. Optical particle counter (OPC) hourly average measurements, y axis denotes the size bin. **(f)** Particulate matter vertical profile. LIDAR Range Corrected Signal, y axis denotes the height above ground level.

**Particulate matter
and vertical
processes**

G. Curci et al.

[Title Page](#)[Abstract](#)[Introduction](#)[Conclusions](#)[References](#)[Tables](#)[Figures](#)[Back](#)[Close](#)[Full Screen / Esc](#)[Printer-friendly Version](#)[Interactive Discussion](#)

Particulate matter and vertical processes

G. Curci et al.

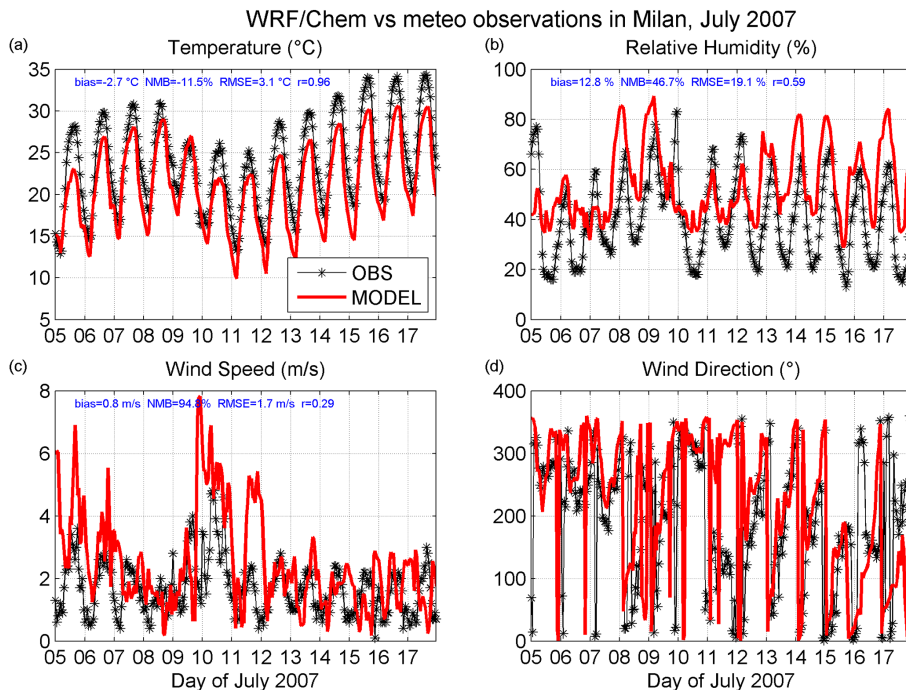


Figure 2. Comparison of observed and simulated hourly meteorological variables at ground level in Milan 5–17 July 2007. Simulations are carried out with WRF/Chem model and results are shown for the nested domain over Northern Italy at 10 km horizontal resolution. Statistical indices shown inset are defined in Appendix A.

[Title Page](#)[Abstract](#)[Introduction](#)[Conclusions](#)[References](#)[Tables](#)[Figures](#)[Back](#)[Close](#)[Full Screen / Esc](#)[Printer-friendly Version](#)[Interactive Discussion](#)

Particulate matter
and vertical
processes

G. Curci et al.

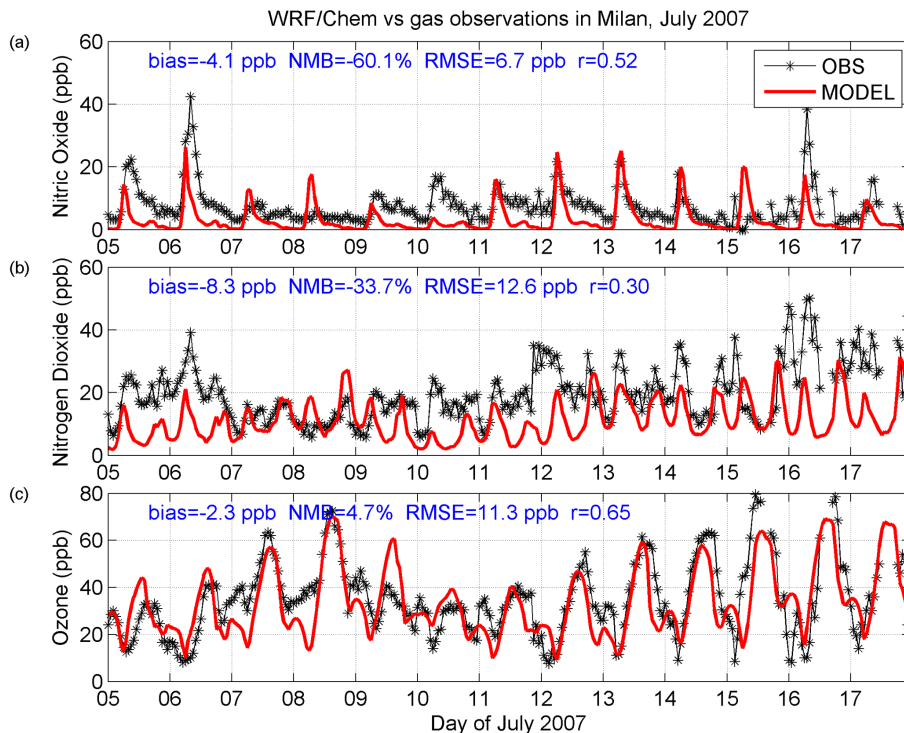


Figure 3. Same as Fig. 2, but for hourly gas-phase variables at ground level in Milan 5–17 July 2007.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



Particulate matter
and vertical
processes

G. Curci et al.

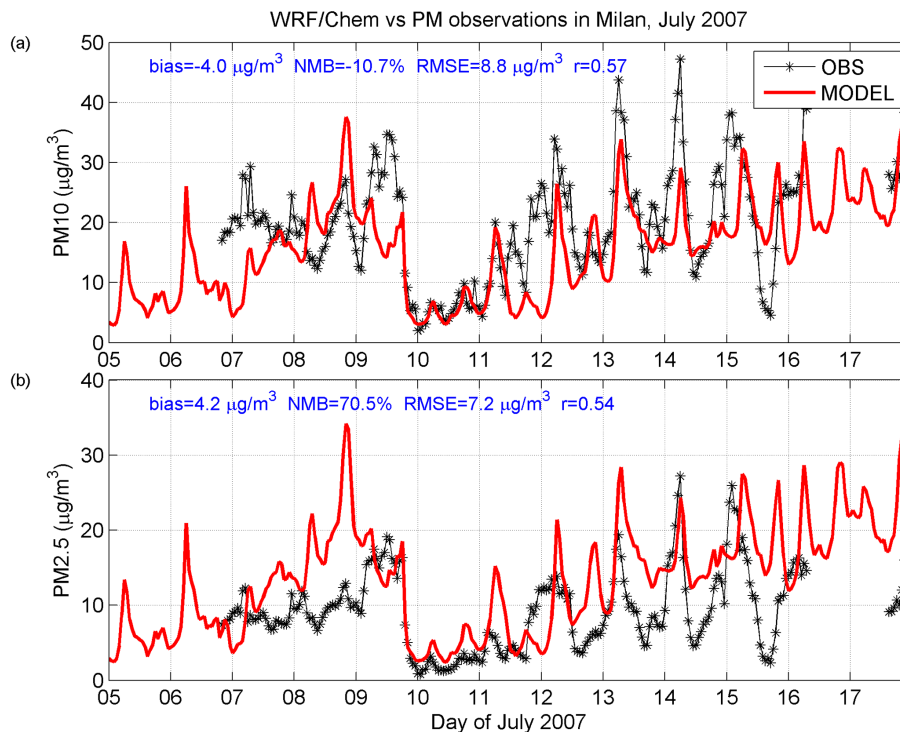


Figure 4. Same as Fig. 2, but for hourly particulate matter at ground level in Milan 5–17 July 2007.

[Title Page](#)[Abstract](#)[Introduction](#)[Conclusions](#)[References](#)[Tables](#)[Figures](#)[◀](#)[▶](#)[◀](#)[▶](#)[Back](#)[Close](#)[Full Screen / Esc](#)[Printer-friendly Version](#)[Interactive Discussion](#)

Particulate matter and vertical processes

G. Curci et al.

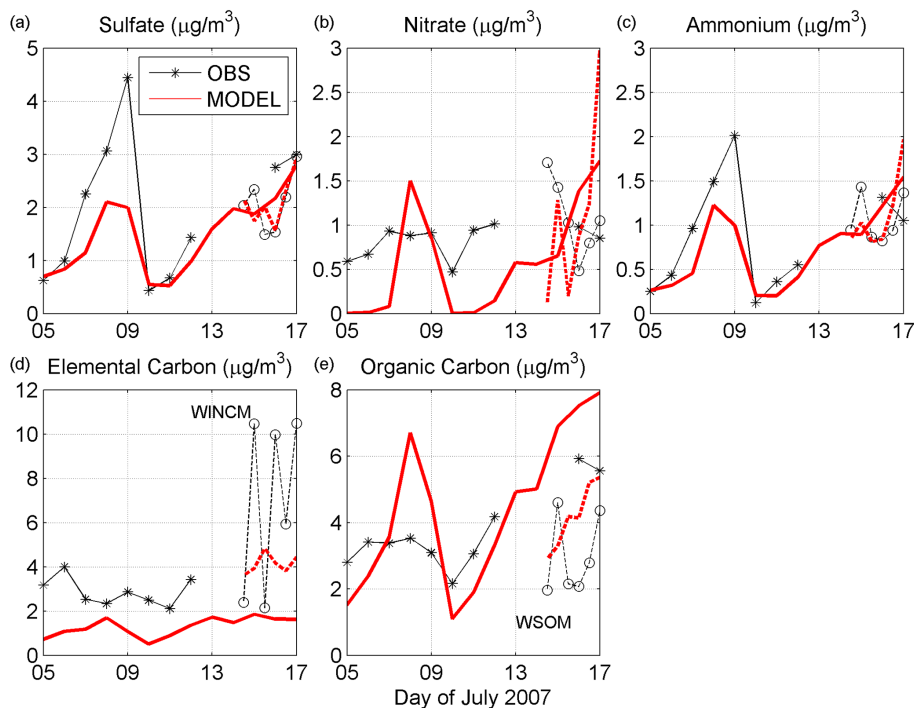


Figure 5. Same as Fig. 2, for daily and bi-daily particulate matter composition at ground level in Milan 5–17 July 2007. Bi-daily observations (dashed lines) are available only from 14 to 17 July. In panel (d), WINCM is the Water Insoluble Carbon Mass (EC + mostly primary OC), in panel (e) WSOM is Water Soluble Organic Mass (mostly secondary organic aerosol, Carbone et al., 2010).

[Title Page](#)
[Abstract](#)
[Introduction](#)
[Conclusions](#)
[References](#)
[Tables](#)
[Figures](#)
[Back](#)
[Close](#)
[Full Screen / Esc](#)
[Printer-friendly Version](#)
[Interactive Discussion](#)

Particulate matter
and vertical
processes

G. Curci et al.

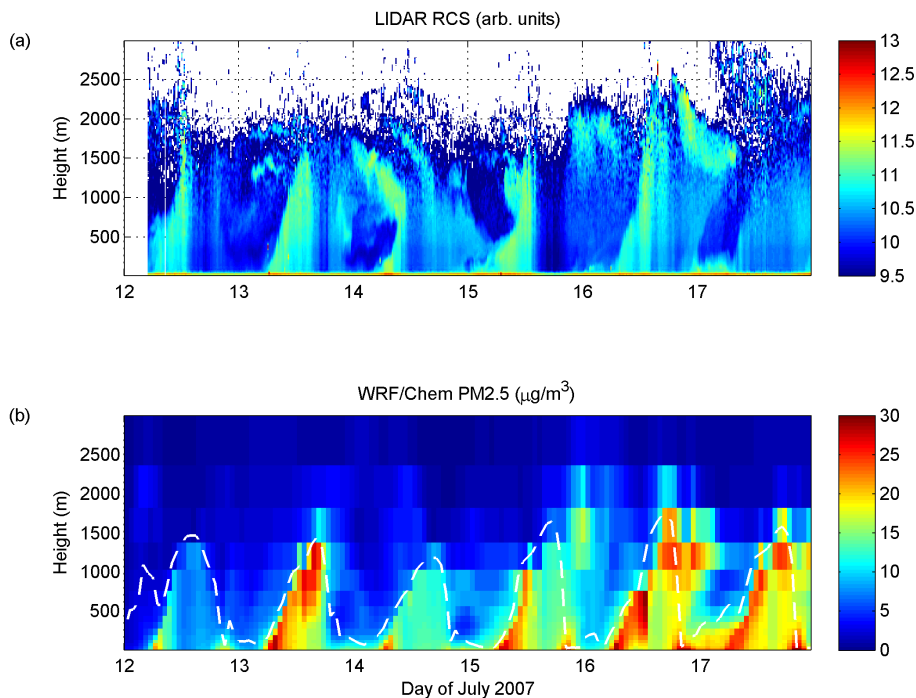


Figure 6. Qualitative comparison of **(a)** LIDAR Range Corrected Signal and **(b)** simulated PM_{2.5} vertical profile over Milan 12–17 July 2007.

Particulate matter
and vertical
processes

G. Curci et al.

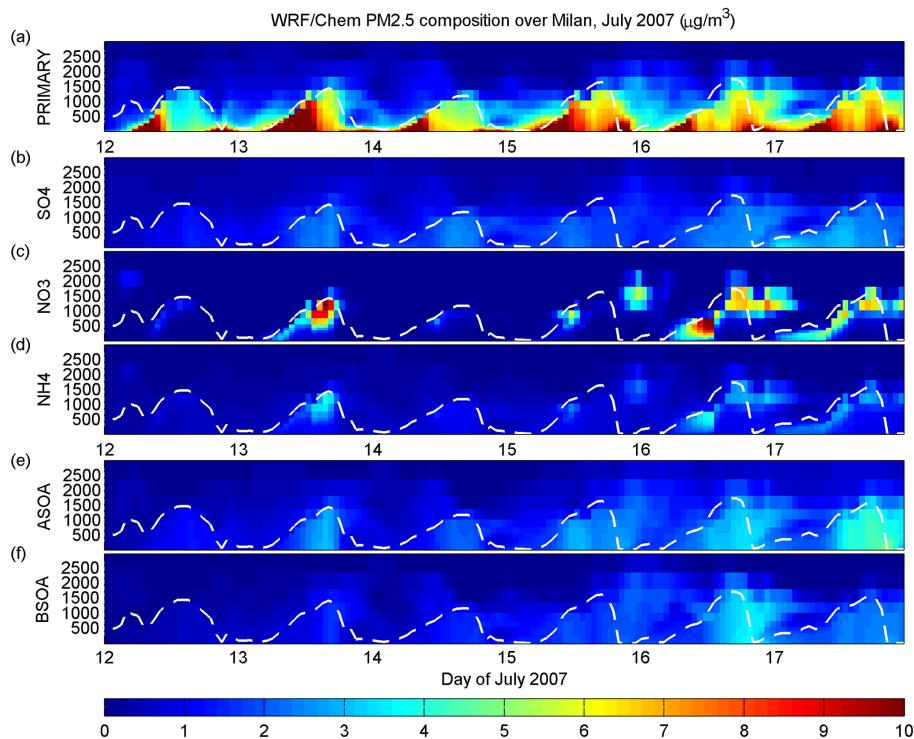


Figure 7. Simulated composition of PM_{2.5} profile shown in Fig. 6. ASOA and BSOA in panels (e) and (f) are anthropogenic and biogenic secondary organic aerosol, respectively.

Particulate matter
and vertical
processes

G. Curci et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion

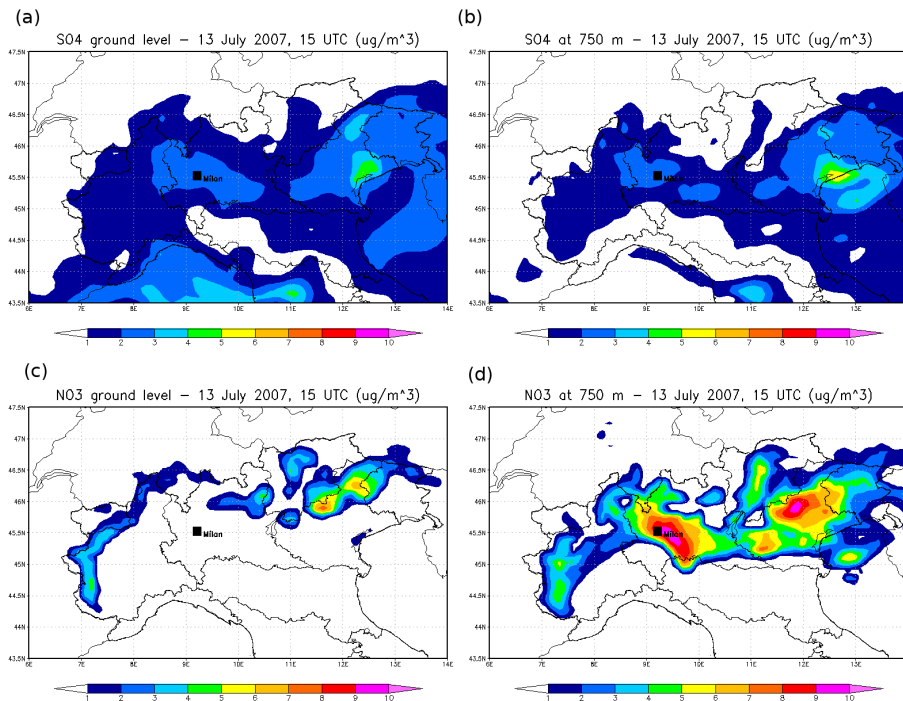


Figure 8. Maps of the concentration of PM_{2.5} sulfate (a–b) and nitrate (c–d) components simulated at 16:00 LST of 13 July 2007 over Po Valley. Panels (a–c) are at ground level, panels (b–d) at 750 m height.

Particulate matter and vertical processes

G. Curci et al.

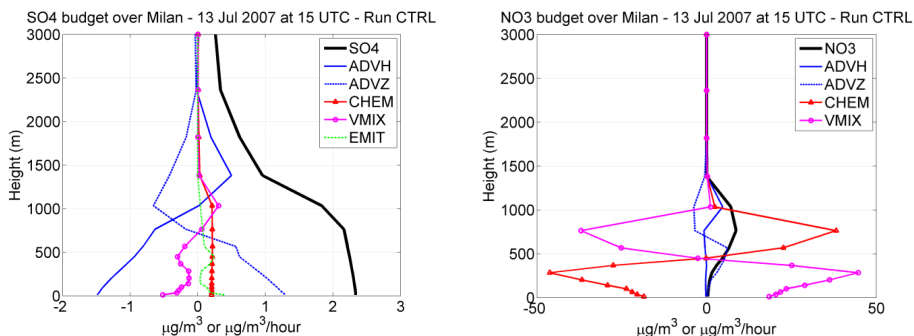


Figure 9. Simulated vertical profile of concentration ($\mu\text{g m}^{-3}$) and continuity equation terms ($\mu\text{g m}^{-3} \text{h}^{-1}$) for particulate sulfate (left) and nitrate (right) at 16:00 LST of 13 July 2007 over Milan. Budget terms are: horizontal advection (ADVH), vertical advection (ADVZ), chemistry (CHEM), turbulent mixing and dry deposition (VMIX), emission (EMIT).

Title Page

Abstract Introduction

Conclusions References

Tables Figures

◀ ▶

◀ ▶

Back Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



Particulate matter
and vertical
processes

G. Curci et al.

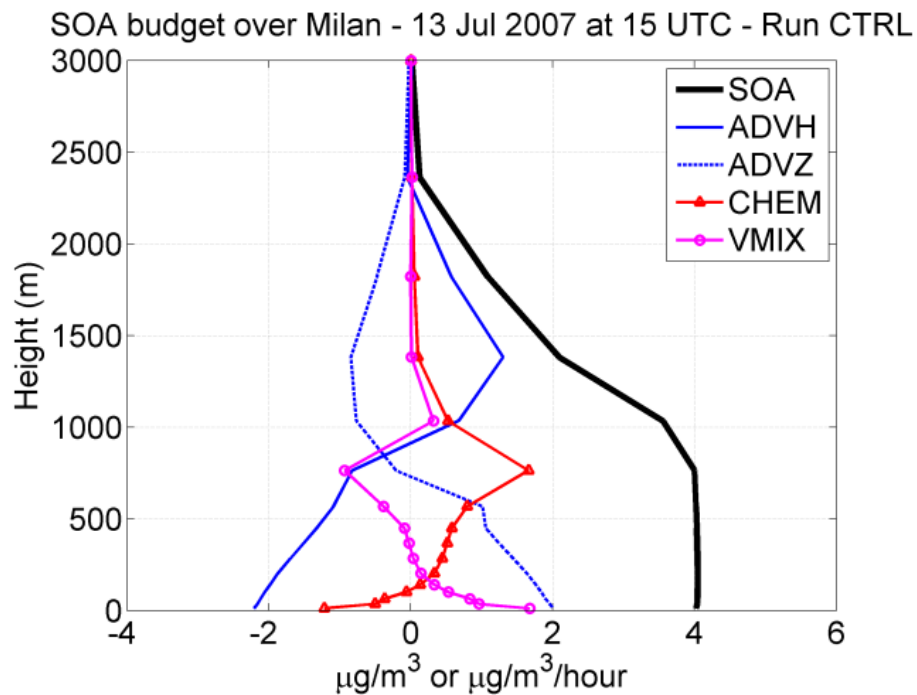


Figure 10. Same as Fig. 9, but for Secondary Organic Aerosol (SOA).

[Title Page](#)[Abstract](#)[Introduction](#)[Conclusions](#)[References](#)[Tables](#)[Figures](#)[◀](#)[▶](#)[◀](#)[▶](#)[Back](#)[Close](#)[Full Screen / Esc](#)[Printer-friendly Version](#)[Interactive Discussion](#)

Particulate matter
and vertical
processes

G. Curci et al.

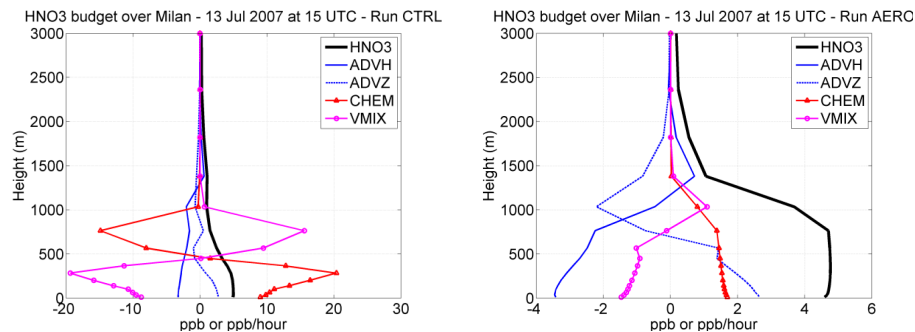


Figure 11. Same as Fig. 9, but for nitric acid (HNO_3) and units in ppb. On the left the reference simulation (CTRL), on the right a sensitivity simulation with aerosol chemistry switched off (AERO). Please notice the different abscissa range.

[Title Page](#)[Abstract](#)[Introduction](#)[Conclusions](#)[References](#)[Tables](#)[Figures](#)[Back](#)[Close](#)[Full Screen / Esc](#)[Printer-friendly Version](#)[Interactive Discussion](#)

Particulate matter and vertical processes

G. Curci et al.

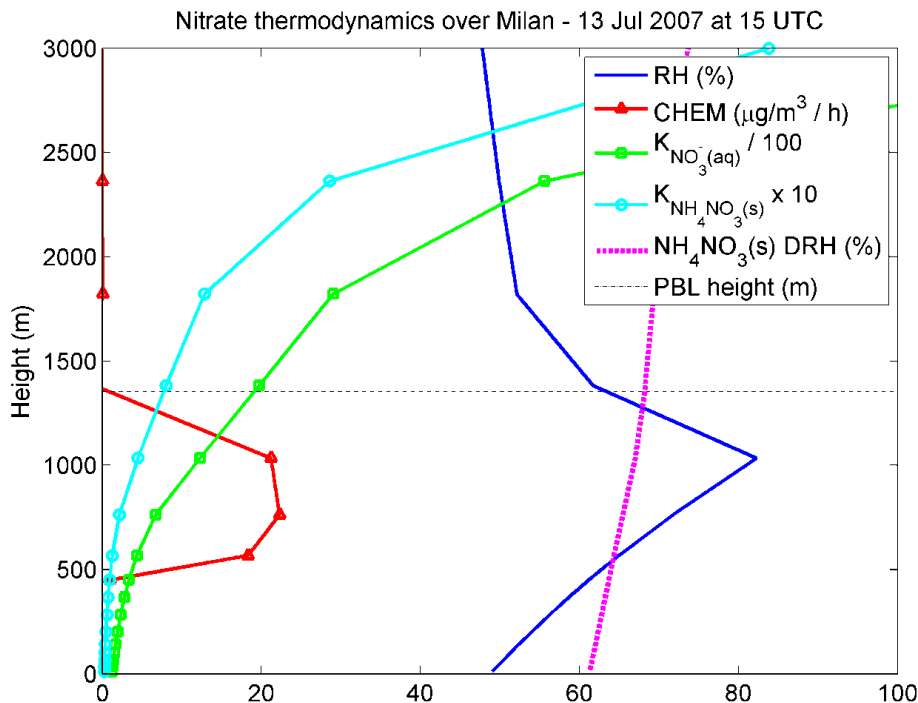


Figure 12. Simulated vertical profile of relative humidity (blue) and particulate nitrate net chemical production term (red, triangles) at 16:00 LST of 13 July 2007 over Milan. Also shown, vertical profiles of equilibrium constants of aqueous phase nitrate (green, squares) and solid ammonium nitrate (cyan, circles), and ammonium nitrate deliquescence relative humidity (magenta, dashed). The height of PBL is denoted by the horizontal black dashed line. Please note that equilibrium constants are scaled by the constant factors shown inset to fit on the same abscissa range.

Particulate matter
and vertical
processes

G. Curci et al.

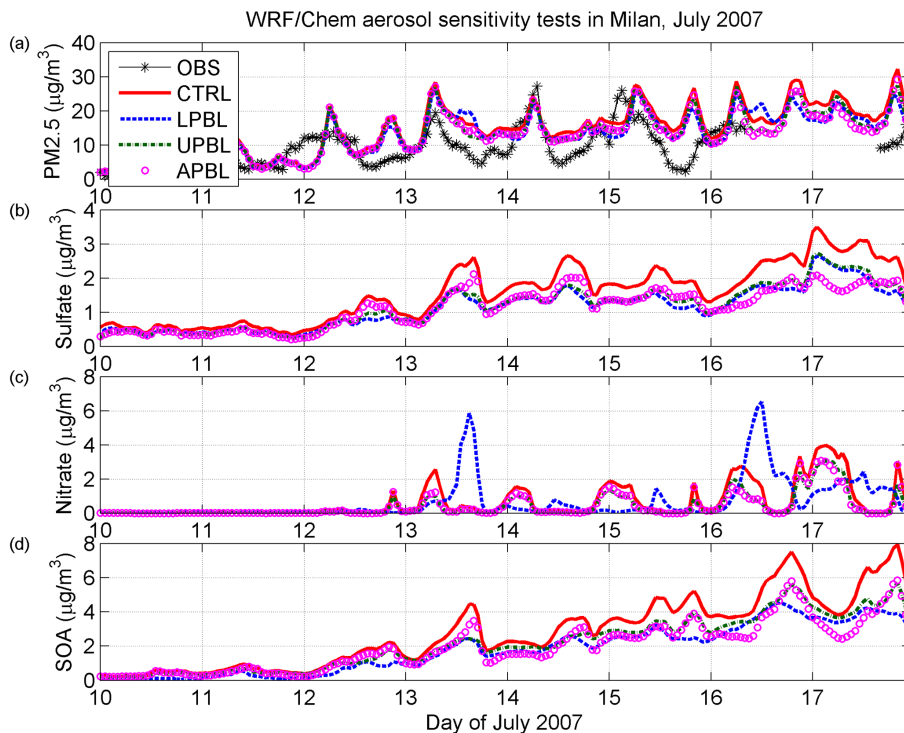


Figure 13. Sensitivity tests on chemical production in different vertical layers (see Table 2 for explanation of labels), at ground level over Milan 10–17 July 2007. Hourly observations (black line-star) are only available for $\text{PM}_{2.5}$ (top panel).

[Title Page](#)[Abstract](#)[Introduction](#)[Conclusions](#)[References](#)[Tables](#)[Figures](#)[Back](#)[Close](#)[Full Screen / Esc](#)[Printer-friendly Version](#)[Interactive Discussion](#)

