How much is particulate matter near the ground influenced by upper level processes within and above the PBL? A summertime case study in Milan (Italy)

Formattato: Italiano (Italia)

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Response to Anonymous Referee #1

Review of “How much is particulate matter near the ground influenced by upper level processes within and above the PBL? A summertime case study in Milan (Italy)” by Curci et al.

This study uses field data in Milan, Italy in 2007 along with WRF/CHEM modeling to investigate what the composition is of aerosol layers formed at the top of the boundary layer during the day, how much of the surface aerosol layer is from entrainment of aerosols aloft, and how much these layers aloft contribute to surface concentrations on the following day. The topic is of interest to this journal. This topic has not received that much attention and this paper adds to the growing body of knowledge about how aerosol layers above the PBL can influence the surface layer. The authors focus on one case event and treat it in depth and reach interesting conclusions that are worthy of publication. The paper requires minor English editing. Prior to publication, the authors should address the issues below.

The authors thank the reviewer for careful reading and valuable comments which improved the clarity of the manuscript. The paper was revised following reviewer’s suggestions as detailed below. Moreover, English was edited in the revised version of the manuscript.

Specific Comments:

Page 26405, Line 2: I suggest changing “yield” to “lead”

Done.

Page 26405, Line 11: change “evidences” to “evidence”

Done.

Pg 26406, Line 2: remove “the” before “atmospheric”

Done.

Page 26407, Line 26: Could another reason for why winter PM is high is low-level residential combustion for heating purposes?

There are certainly more emissions for heating in winter. Actually, biomass burning was reported as a dominant winter PM source, at least for the organic fraction (Gelencser et al., 2007; Gilardoni et al., 2011). This source includes wood burning for domestic heating. However, in summer there are likely enhanced biogenic and soil erosion sources, with respect to winter. Thus changed primary emissions were not listed among the driving factors of winter/summer PM differences in the Po Valley in the studies we refer to, in that part of the manuscript (Barnaba et al.). We thus preferred to leave the sentence unchanged in the revised manuscript.

Gelencser, A., B. May, D. Simpson, A. Sanchez-Ochoa, A. Kasper-Giebl, H. Puxbaum, A. Caseiro, C. Pio, and M. Legrand (2007), Source apportionment of PM2.5 organic aerosol over Europe: Primary/secondary, natural/anthropogenic, and fossil/biogenic origin, J. Geophys. Res., 112, D23S04, doi:10.1029/2006JD008094.

Gilardoni, S., Vignati, E., Cavalli, F., Putaud, J. P., Larsen, B. R., Karl, M., Stenström, K., Genberg, J., Henne, S., and Dentener, F.: Better constraints on sources of carbonaceous aerosols using a combined 14C – macro tracer analysis in a European rural background site, Atmos. Chem. Phys., 11, 5685-5700, doi:10.5194/acp-11-5685-2011, 2011.

Page 26408, Line 1-6: The authors should improve the balance of this paragraph's discussion to also mention other airborne data that have shown that secondarily produced species can be higher aloft than at the surface. Such studies shown below have pointed to the importance of clouds and relative humidity in leading to higher mass concentrations of aerosol species aloft. Duong et al. (2011). Water-soluble organic aerosol in the Los Angeles Basin and outflow regions: Airborne and ground measurements during the 2010 CalNex field campaign, J. Geophys. Res., 116, D00V04, doi:10.1029/2011JD016674.

Wonaschuetz et al. (2012). Aerosol and gas re-distribution by shallow cumulus clouds: an investigation using airborne measurements, J. Geophys. Res., 117, D17202, doi:10.1029/2012JD018089.

Heald et al. (2006), Concentrations and sources of organic carbon aerosols in the free troposphere over North America, J. Geophys. Res., 111, D23S47, doi:10.1029/2006JD007705.

Novakov et al. (1997), Airborne measurements of carbonaceous aerosols on the East Coast of the United States, J. Geophys. Res., 102(D25), 30,023–30,030, doi:10.1029/97JD02793.

We thank the reviewer for bringing to our attention those interesting studies, which add further observational evidence of the presence of enhanced aerosol layers above the surface. We added a related sentence at the end of the paragraph: "Moreover, the presence of aerosol layers enriched with sulfate and water-soluble carbonaceous matter was observed above the boundary layer or in convective clouds during several aircraft campaigns over North America (Novakov et al., 1997; Heald et al., 2006; Duong et al., 2011; Wonaschuetz et al., 2012), and attributed to biomass burning plumes or aqueous-chemistry processes."

Page 26410, last line and into next page: this sentence confuses me about why nitrate is in the coarse fraction due to ammonium being neutralized by sulfate. If sulfate is fully neutralized by ammonium it would seem that if there is any more ammonia present that it can then interact with nitric acid to produce ammonium nitrate, which typically is in the fine mode. Based on literature (see for instance: Lee et al. (2008), Observations of fine and coarse particle nitrate at several rural locations in the United States, Atmos. Environ., 42, 2720–2732, doi:10.1016/j.atmosenv.2007.05.016.), coarse mode nitrate originate from interactions of nitric acid with coarse particle types such as mineral dust and sea salt. This sentence should be revised as it currently confuses this reviewer.

We agree with the reviewer that the original sentence may be confusing, and acknowledge that the reviewer correctly guessed the real intended meaning. We reformulated the sentence as follows: "In summer, nitrate can exhibit a broader size distribution as a larger fraction may also form in the coarse mode. Higher temperatures, lower humidity, higher load of sulphate competing for the uptake of ammonia, are less favourable to ammonium nitrate accumulation in the fine mode. As a consequence, more nitric acid is available to react with soil dust or sea salt leading to the formation of mineral nitrate on coarse particles.". We also added the suggested reference Lee et al., 2008.

Page 26414, Line 9: A period is missing after “measurements”

Done.

Page 26424, Line 1: Can the authors be more clear when they say it is “visible e.g. on 16-17 July in both observations and models”. Are readers supposed to know this clearly from looking at specific figures? If so, what figures and be specific as to how readers will know that this is visible.

First of all we corrected a wrong reference to Hodzic et al. (2006) to Hodzic et al. (2004) at the beginning of the same paragraph. We agree with the reviewer that the sentence is obscure and ambiguous. We now revised the last part of the paragraph for a more clear and sharp guidance to the features we would like to bring to reader’s eyes. We now focus on days 14-16 July, in place of 16-17 July, because they better illustrate those features. Here is the revised part of the paragraph: “Then, in the afternoon, the mountain-valley breeze cleans the lower PBL (note the abrupt abatement of both the Lidar and the model aerosol signals in the second part of the day), often leaving an upper air aerosol residual layer above. Model simulations also reproduce such residual layers (note the afternoon increase of PM2.5 values in the upper levels, particularly visible on July 15-16). When such residual layers persists overnight, the Lidar shows these to entrain into the developing PBL the day after (note the merging of the upper level aerosol layers with the growing, aerosol-traced PBL in Figure 6a, particularly evident in the morning of July 14 and 15). There are hints of the same features also in model simulations.”

Page 26427, Lines 4-8 and subsequent text: It is argued that nitrate is produce at higher altitudes due to more favorable RH values that allow for aqueous conversion of nitric acid to nitrate. But one major aspect of this argument that requires discussion is that if such aerosols aloft are then brought down to the surface, wouldn’t the aerosols get dried at lower RHs and then the nitrate evaporates back out? For the case of cloud droplets for instance, it has been shown in past studies that evaporation of drops leads to losses of nitrate back to the gas phase. Also, in the discussion in this same section, I was hoping the authors could discuss whether the ambient multicomponent particles would be expected to deliquesce at the DRH point of ammonium nitrate or at a different RH. Past work has shown that the DRH varies from the pure salt form when there are other components included, and it can be argued that it is more practical to assume that the aerosols in the region will deliquesce at a point other than the DRH of ammonium nitrate.

The nitrate layer formed in the upper PBL during the day indeed forms because nitrate re-evaporates back to the gas phase when brought back near the ground by vertical turbulent motions. It's exactly the reason why the nitrate profile is not homogeneous, as it is the case for sulfate. Thus the contribution to surface nitrate from this upper layer is expected to be very small, if not negative, during daytime. This point is indeed highlighted for days 13 and 16 July in the subsequent paragraph of the manuscript, regarding the budget analysis of ground PM as contributed from different layers of the lower atmosphere (lower PBL, upper PBL, and above PBL). As also mentioned in that paragraph, if enough aerosol (including nitrate) survives the night in the layer above the PBL, it may entrain into the PBL the following morning and then effectively contribute to surface PM levels. However, we haven't pushed the latter budget to the detail of single species, since we feel that more observational constraints on the aerosol vertical profile should be available (which is not the case here) to support the modelling results.

Regarding the multicomponent DRH point we are aware that it is different from that of pure nitrate. However, it is known that multicomponent DRH point is always lower than that of single components (e.g. Seinfeld and Pandis, 2006), thus a more accurate calculation of the DRH would just make a broader layer for the potential aqueous formation of nitrate, but it will not change the substance of the discussion, i.e. that nitrate preferentially forms in the upper PBL and in aqueous form. The thickness of the layer favourable to nitrate formation deducible from Figure 12 may thus be regarded as a conservative lower estimate.

In order to improve clarity on these points, we added the following sentences at the end of the paragraph: "Although the real multicomponent DRH point will differ from that of pure nitrate, it is known that the DRH of mixtures is always lower than that of pure salts (Seinfeld and Pandis, 2006). The thickness of the layer favourable to aqueous nitrate formation deducible from Figure 12 may thus be regarded as a conservative lower estimate. During daytime, the nitrate formed in the upper boundary layer re-evaporates back to the gas phase when brought to the ground by vertical motions, and that's the origin of the

inhomogeneous vertical profile of nitrate. For further discussion on how much the upper aerosol layer contributes to ground PM we point the reader to the next paragraph."

Figure 2 caption: should say ". . .shown in inset are. . ."

Figure 2 caption: should say ". . .shown in inset to. . ."

We believe the reviewer here refers to Figures 3 and 4. We added the sentence "Shown inset are statistical indices defined in Appendix A." in the caption of both figures.

Figure S5 caption: "Rolf" is mis-spelled and should be "Rolph"

Done.

Figure 12: I recommend moving the text box off of the data curves as it is distracting.

Done.

Response to Anonymous Referee #2

This manuscript combines observations and modeling to interpret the vertical profile and budget of aerosol over Milan. The study is clear and thorough and provides some interesting insights about the vertical profile of nitrate. I have only minor comments below. However the article does need to be edited for language prior to publication (numerous grammatical and phrasing errors).

The authors thank the reviewer for careful reading and valuable comments which improved the clarity of the manuscript. The paper was revised following reviewer's suggestions as detailed below. Moreover, English was edited in the revised version of the manuscript.

1. Abstract, lines 19-23: These results are somewhat overstated given that they are based on one event, and not necessarily generalizable. I suggest clarifying that numbers in particularly are based on the one case study examined here.

All the analysis presented in the manuscript is limited to one short case study in one location is clear from the title. However, we added this statement in front of the bottom line of the abstract: "Although the results presented here are relative to one relatively short period at one location, …"

2. Page 26411, lines 9-11: what size ranges do the 2 submicron modes cover?

We overlooked this information. The dry diameter of the two modes is in the range 50-200 nm, but they have different hygroscopic properties, one hydrophobic and the other hydrophilic, as already mentioned in the manuscript.

3. Page 26413: lines 5 and 17: please comment on why differing OM:OC ratios (1.6 and 1.8) are applied.

1.6 and 1.8 are applied as factors to convert OC to OM and WSOC to WSOM, respectively. These conversion factors are reported in the cited literature as reasonable values for urban environment.

4. Page 26413, lines 7-9: how far away are the weather and monitoring stations from the aerosol measurement site?

The gas and weather station run by the local environmental agency is in the same suburban area (Via Juvara) of the Torre Sarca aerosol site. The distance of the two sites is about 5 km, thus well within the 10 km of the model horizontal resolution.

5. Page 26415, lines 20-21: what types of SOA are included in the simulation?

SOA in the VBS mechanism implemented in WRF/Chem originates from the oxidation of anthropogenic and biogenic VOC currently believed relevant for SOA production. These include alkanes, alkenes, xylenes, aromatics, isoprene, mono- and sesqui-terpenes. Full details are provided in the cited reference Ahmadov et al. (2012). We added a sentence in the text " …, which include the oxidation of anthropogenic and biogenic VOC currently believed to be important for SOA production (alkanes, alkenes, xylenes, aromatics, isoprene, monoterpenes, and sesquiterpenes)."

6. Page 26415, line 19: the name of the thermodynamic model is MAR-A not RPMARES

We thank the reviewer for the clarification, the actual name was not clear from the references and from comments embedded into the code.

7. Page 26417, lines 3-5: Has this model been validated using TNO emissions? If so please include references.

The model was validated with TNO emissions in the frame of AQMEII intercomparison. We added the two relevant references (Im et al., 2014a,b).

8. Page 26420, lines 25-29 & page 26422, lines 17-22: Please comment on how the lack of dust in the model may impact the comparison to observations.

Regarding the Saharan dust event after 17 July, we excluded the days from the analysis because of the lack of representation of that contribution. We added this sentence at the end of paragraph on page 26420: "Since Saharan dust intrusions are not modelled here,

these days are excluded from the analysis.". Regarding the bias on PM10 commented on page 26422, we added the following sentence: "The negative bias of PM10 could be partly explained by the missing source from soil dust erosion and resuspension in the model."

9. Page 26423, lines 8-9: Please comment/explain the poor model performance for nitrate shown in Figure 5.

We added the following comment: "Recently reported hourly measurements of PM composition in the Po Valley indeed confirm the same "pulsed" behaviour of nitrate near the ground, with values near zero during daytime, and irregular peaks nighttime (Decesari et al., 2014). This highlights the inherent difficulties in simulating the nitrate concentrations at sub-daily frequency."

Decesari, S., Allan, J., Plass-Duelmer, C., Williams, B. J., Paglione, M., Facchini, M. C., O'Dowd, C., Harrison, R. M., Gietl, J. K., Coe, H., Giulianelli, L., Gobbi, G. P., Lanconelli, C., Carbone, C., Worsnop, D., Lambe, A. T., Ahern, A. T., Moretti, F., Tagliavini, E., Elste, T., Gilge, S., Zhang, Y., and Dall'Osto, M.: Measurements of the aerosol chemical composition and mixing state in the Po Valley using multiple spectroscopic techniques, Atmos. Chem. Phys., 14, 12109-12132, doi:10.5194/acp-14-12109-2014, 2014.

10. Page 26423, line16: what kind of SOA? Biogenic? Anthropogenic? From where?

Both biogenic and anthropogenic with about the same share, from the larger region around Milan. We haven't added more comments in the manuscript, since this is distracting from the main discussion.

11. Page 26423, lines 20-29: comment on how the model compares to the lidar profiles

The paragraph was revised as follows. First of all we corrected a wrong reference to Hodzic et al. (2006) to Hodzic et al. (2004) at the beginning of the same paragraph. We now revised the last part of the paragraph for a more clear and sharp guidance to the features we would like to bring to reader's eyes. We now focus on days 14-16 July, in place of 16-17 July, because they better illustrate those features. Here is the revised part of the paragraph:

"Then, in the afternoon, the mountain-valley breeze cleans the lower PBL (note the abrupt abatement of both the Lidar and the model aerosol signals in the second part of the day), often leaving an upper air aerosol residual layer above. Model simulations also reproduce such residual layers (note the afternoon increase of PM2.5 values in the upper levels, particularly visible on July 15-16). When such residual layers persists overnight, the Lidar shows these to entrain into the developing PBL the day after (note the merging of the upper level aerosol layers with the growing, aerosol-traced PBL in Figure 6a, particularly evident in the morning of July 14 and 15). There are hints of the same features also in model simulations."

12. Figure 7: color scale makes it difficult to see features. I suggest different color bars be used for different panels, as appropriate

We agree that the scales somewhat hide the simulated features, however the same color scale makes the intercomparison of PM species contribution very direct. Moreover, the relevant features discussed in the text still emerge (e.g. homogeneity of sulfate and SOA profile in the PBL, correlation of ammonium with nitrate). We prefer to leave the Figure 7 as is.

13. Page 26424, line 11: what do you mean by primary? Only those emitted particles? Is this BC and POA in your simulation? Please clarify.

This is also primary inorganic and unspeciated anthropogenic fraction. We changed the (inorganic and organic) in parentheses to (unspeciated anthropogenic, black carbon, and primary organic carbon).

14. Figures 9, 10, 11: orange and pink are difficult to distinguish. I suggest changing one of these colors.

Done. Pink changed to black.

15. Page 26425, line14: why was this time chosen? Is it representative of the entire period? Could the authors instead show an average over a longer period of time?

The time chosen in representative of the typical budget during the central part of the day, and highlights the more interesting features emerging in the vertical profile budget discussed in the rest of the manuscript. We believe that averaging on several hours will not change the picture emerging from this snapshot view.

16. Page 26426, line 23: please add temperature profile to Figure 12

Done.

Abstract

Chemical and dynamical processes ~~yield~~ lead to the formation of aerosol layers in the upper planetary boundary layer (PBL) and above it. Through vertical mixing and entrainment into the PBL these layers may contribute to the ground-level particulate matter (PM), ~~but~~ however to date a quantitative assessment of such a contribution ~~–~~has not been carried out~~is still missing~~. This study investigates this aspect by combining chemical and physical aerosol measurements with WRF/Chem model simulations. The observations were collected in the Milan urban area (Northern Italy) during summer of 2007. The period coincided with the passage of a meteorological perturbation that cleansed the lower atmosphere, followed by a high pressure period ~~that~~ favouring~~ed~~ pollutant accumulation. Lidar observations revealed the formation of elevated aerosol layers and ~~show~~ evidence~~s~~ of their entrainment into the PBL. We analyzed the budget of ground-level PM2.5 (particulate matter with an aerodynamic diameter less than 2.5 µm) with the help of the online meteorology-chemistry WRF/Chem model, ~~with~~ focusing in particular ~~focus~~ on the contribution of upper level processes. ~~We~~ Our findings show that an important player in determining the upper PBL aerosol layer is particulate nitrate, which may reach higher values in the upper PBL (up to 30% of the aerosol mass) than in the lower PBL. The nitrate formation process is predicted to be largely driven by the relative humidity vertical profile, ~~that~~ which may trigger efficient aqueous nitrate formation when exceeding the ammonium nitrate deliquescence point. Secondary PM2.5 produced in the upper half of the PBL may contribute up to 7-8 µg/m^3 (or 25%) to ground level concentrations on an hourly basis. A large potential role is also found to be played by the residual aerosol layer above the PBL, which may occasionally contribute up to 10-12 µg/m^3 (or 40%) to hourly ground level PM2.5 concentrations during the morning hours. Although the results presented here refer to one relatively short period in one location, t~~T~~his study highlights the importance of considering the interplay between chemical and dynamical processes occurring within and above the PBL when interpreting ground level aerosol observations.

1 Introduction

The understanding of processes governing ~~the~~ atmospheric aerosols is primarily motivated by their adverse effects on health and their contribution to the radiative budget of the atmosphere. Diseases ~~to~~ affecting the respiratory system have been linked to inhalation of aerosols, especially their finer and more numerous fraction (Beelen et al., 2014~~3~~; Oberdorster, 2001), although the mechanisms underlying the health effect associated to size, number and composition of particulate matter have ~~just started~~only recently begun to be disclosed (Harrison and Yin, 2000; Daher et al., 2012; Perrone et al., 2013). Aerosols affect the atmospheric energy balance directly, by scattering and absorbing radiation (Yu et al., 2006), indirectly, by serving as cloud condensation nuclei (Lohmann and Feichter, 2005), and semi-directly, by heating the air through absorption of radiation and reducing low cloud cover (Johnson et al., 2004). The assessment of these effects caused by~~of~~ aerosols is still characterized by large uncertainties, since ~~it is still uncertain~~ our knowledge of the processes determining their abundance, size distribution, and chemical composition, which strongly vary in space and time, is still limited (Raes et al., 2000; Poschl, 2005). Here we focus on the interplay between dynamical and chemical processes in the vertical direction, in order to better understand the budget terms making up the ground level particulate matter, a common measure to evaluate ~~the~~ air quality. The study ~~is~~ focuses~~d~~ on the urban environment of Milan, ~~placed~~ situated in the ~~middle~~ center of ~~the~~ Italy's Po Valley ~~(Italy)~~, a~~n~~ European hot-spot for atmospheric pollution.

The correlation between pollutants at the surface and meteorological variables is well established and the fundamental role played by the variables associated to the vertical mixing in the planetary boundary layer (PBL) has been highlighted ~~for~~ both for ozone (Di Carlo et al., 2007, and references therein) and particulate matter (Tai et al., 2010, and references therein). Moreover, Zhang and Rao (1999) analyzed aircraft and tower measurements over the Eastern United States and showed that elevated nocturnal layers rich ~~of~~ in ozone and its precursors aloft, remnant of the previous day's mixed layer, may strongly affect ground-level ozone levels on the following morning as ~~the~~ vertical motions mix upper and surface air. The same authors suggested that a reduction of ozone and precursors aloft may be more effective in reducing pollution than local emission cuts~~ in reducing pollution~~, thus calling for a region-wide strategy for emissions control. Aerosols are also known to form layers above or near the top of the mixing layer, especially when stability and presence of clouds increase (e.g.

O'Dowd and Smith, 1996). Similarly to ozone, an aerosol residual layer aloft is often observed (e.g. Di Giuseppe et al., 2012), which may influence the aerosol at the surface, as witnessed by similar size-distributions (Maletto et al., 2003). A significant contribution to surface aerosol from entrainment and vertical dilution and chemical net production in the boundary layer ~~was ~~has also been pointed out in recent studies using single-column models (van Stratum et al., 2012; Ouwersloot et al., 2012).

The nontrivial relationship between ground- and upper-level aerosols burden is illustrated by ~~the ~~comparing~~son~~ of surface particulate matter (PM) mass concentrations ~~with ~~to aerosol optical depth (AOD), which is proportional to the aerosol column load (typically measured by ground-based sun-photometers or retrieved from satellites). In a well mixed PBL, the AOD may exhibit a high correlation with surface PM, especially with its fine fraction, and indeed this assumption is often exploited to infer surface PM2.5 (PM with diameter < 2.5 µm) from satellite AOD observations (e.g. van Donkelaar et al., 2010). However, that assumption does not always hold true, ~~because of~~due to the presence of significant aerosol stratification aloft, and noticeable differences which may occur between AOD and surface PM behaviour may occur, such as in the timing of daily peak values or in multi-day trends (Barnaba et al., 2007, 2010; Boselli et al., 2009; Estelles et al., 2012; He et al., 2012). Analyzing two-year measurements in the Po Valley (Italy), Barnaba et al. (2010) indeed pointed out that annual cycles of AOD and surface PM10 (PM with diameter < 10 µm) display a remarkable opposite phase. While PM10 peaks in winter, ~~because of~~due to the reduced dilution by a shallower PBL and ~~of ~~to the condensation of semi-volatile species favoured by the lower temperatures, AOD peaks in summer, because of a more persistent presence of an aerosol residual layer aloft, which contributes up to 30% of the total AOD.

Aircraft measurements also showed intriguing features of aerosol vertical gradients in the lower troposphere, in particular when looking at different chemical components. Several studies reported a generally constant or slightly decreasing profile in the convective boundary layer of sulfate and organic matter as opposed to an increasing profile of nitrate (Neuman et al., 2003; Cook et al., 2007; Crosier et al., 2007; Morgan et al., 2009; Ferrero et al., 2012). Neuman et al. (2003) attributed the enhanced nitrate layer near the top of the PBL to the lower temperatures that favour gas-phase nitric acid (HNO_3) and ammonia (NH_3) conversion to particulate ammonium nitrate. The same authors also pointed out that nitrate and HNO_3 display sharp vertical gradients in the PBL, as opposed to other directly emitted (carbon monoxide) or secondary (ozone) species that are relatively uniform, and this~~interpreted the~~ observation was interpreted as an indication that thermodynamic equilibrium between gas and particle phases occurs faster than vertical mixing. However, the issue is still under debate~~d~~ as subsequent model studies found that an instantaneous thermodynamic equilibrium between HNO_3 and nitrate yields ~~too~~ excessively steep and unrealistic vertical gradients (Morino et al., 2006; de Brugh et al., 2012). Moreover, the presence of aerosol layers enriched with sulfate and water-soluble carbonaceous matter was observed above the boundary layer or in convective clouds during several aircraft campaigns over North America (Novakov et al., 1997; Heald et al., 2006; Duong et al., 2011; Wonaschuetz et al., 2012), and attributed to biomass burning plumes or aqueous-chemistry processes.

A quantitative assessment of the contribution of elevated aerosol layers and related dynamical and chemical processes to ground-level particulate matter level is still lacking. Recent modelling studies that reported budget (or process) analyses of the simulated aerosol mainly focused on terms of the continuity equation at the surface or on integrated values over the whole boundary layer. Surface and PBL total PM2.5 mass is calculated to be mainly produced by direct emissions and secondary formation by aerosol processes (e.g. condensation and absorption) and removed by horizontal and vertical transport and wet deposition (Zhang et al., 2009; Liu et al., 2011). The controlling processes are different for surface PM number, which is accumulated mainly by homogeneous nucleation and vertical transport and it is lost mainly by dry deposition and coagulation (Zhang et al., 2010).

For primary components such as black carbon (BC) the fate is similar to that of total PM2.5, while for secondary species it is more intricate. Sulfate is generally produced in the PBL by aerosol and clouds processes (the latter being very important) and exported out of the PBL throughout the year (de Meij et al., 2007; Zhang et al., 2009; de Brugh et al., 2011; Liu et al., 2011). Averaged over the year, the nitrate budget is similar to that of sulfate, with the difference that cloud processes (wet deposition) are a sink (de Brugh et al., 2011; Liu et al., 2011). However, during the summer there might be ~~a~~ competition between PM production (e.g. condensation and absorption) and destruction (e.g. evaporation and desorption) processes, and PBL may become a sink and not a source for nitrate (Zhang et al., 2009). The same competition between PM production and destruction processes affect the secondary organic aerosols (SOA) throughout the year (Zhang et al., 2009). Moreover, SOA are strongly influenced by biogenic volatile organic compounds (BVOC) emissions, through semi-volatile products of the ~~oxidation of~~ isoprene and terpenes oxidation, which also have a marked seasonal cycle (Zhang et al., 2007; Hodzic et al., 2009).

In the present study, we examined the formation of aerosol near the surface in the particular perspective of the boundary layer vertical processes outlined above. We analyzed ~~observations of~~ aerosol mass observations, composition, number and optical properties in the month of July 2007 in Milan (45°N, 9°E, Northern Italy) during the intensive campaigns carried ~~out in the~~within the framework of the ~~Italian projects~~ QUITSAT (“Air Quality by the Integration of Ground- and Satellite-based Observations and Multiphase Chemistry-Transport Modelling”, funded by the Italian Space Agency, ASI) and AeroClouds (“Study of Direct and Indirect Effect of Aerosols and Clouds on Climate”, funded by the Italian Ministry for Higher Education) projects. The experimental results ~~are~~ were then complemented/interpreted ~~by~~ through WRF/Chem model simulations.

~~We~~ firstly, ~~briefly review~~ what is known about the aerosol phenomenology in the investigated domain is briefly reviewed in section 2. ~~Then w~~We describe the experimental setup in section 3 and the model setup in section 4. In section 5, ~~we conduct~~ a preliminary analysis of the observations is carried out~~,~~ in order to characterize the relevant features of the case study and pose questions arising from the picture given by the measurements. ~~We t~~Then, this questions are addressed ~~these questions~~ using WRF/Chem model simulations. After a model validation against available observations, we analyzed the budget of aerosol species as calculated by the model, focusing in particular on the vertical dimension. ~~We summarize~~The main results are summarixed in conclusive section 6.

2 The investigated domain

Milan is the largest urban area in Italy (c.a. 5 million people) and lies in one of the most polluted ~~places~~ areas of Europe, the Po Valley (Putaud et al., 2010). The topography of the valley (closed by the Alps to the North and West, and by the Apennine to the South), under high-pressure systems, favour stagnant atmospheric conditions and recirculation of air through the typical mountain-valley breeze (Dosio et al., 2002). The local circulation in combination with elevated anthropogenic emissions especially from traffic, residential combustion, and agriculture (Lonati et al., 2005; Carnevale et al., 2008; Perrone et al., 2012; Saarikoski et al., 2012) makes it a nitrogen dioxide and aerosol hot-spot ~~well~~ clearly visible from space (e. g. Chu et al., 2003; Barnaba and Gobbi, 2004; Ordonez et al., 2006; van Donkelaar et al., 2010).

At the surface, PM10 annual mean in Milan has ~~stabilized~~ been stable between 50 and 60 μg/m^3 in the last decade (Carnevale et al., 2008; Silibello et al., 2008), thus systematically above the European limit of 40 μg/m^3 for human protection (EC, 2008). The winter average values are roughly double than those in the summer, and peak values are up to 200 μg/m^3 (Marcazzan et al., 2001). The main aerosol components are sulfate, nitrate, and organic matter (OM), which account for roughly 20%, 15%, 40%, respectively, of PM10 mass in summer, and 10%, 30%, 50%, respectively, in winter (Marcazzan et al., 2001; Putaud et al., 2002; Lonati et al., 2005; Carbone et al., 2010; Perrone et al., 2010; Daher et al., 2012). These values are similar to those in other urban areas in the Po Valley (Matta et al., 2003; Carbone et al., 2010; Squizzato et al., 2013). Most of the mass of those species is distributed in the accumulation mode (particle diameter in the range 0.14-1.2 μm), while the coarse mode (1.2-10 μm diameter) has a larger fraction of crustal material and sea salts (Matta et al., 2003; Carbone et al., 2010). In summer, nitrate can exhibit a broader size distribution as a larger fraction may also form in the coarse mode. Higher temperatures, lower humidity, higher load of sulphate competing for the uptake of ammonia, are less favourable to ammonium nitrate accumulation in the fine mode. As a consequence, more nitric acid is available to react with soil dust or sea salt leading to the formation of mineral nitrate on coarse particles.~~In summer a significant fraction of nitrate may also form in the coarse mode, because most of the ammonium is neutralized by sulfate~~ (Matta et al., 2003; Hodzic et al., 2006; Lee et al., 2008 ; Carbone et al., 2010). The total number concentration of aerosol is of the order of 10^4 cm^{-3}, with the ultrafine (diameter d < 100 nm) and submicron (100 < d < 1000 nm) particles ~~making~~ constituting up to 80% and 20% of the total, respectively (Lonati et al., 2011). The aerosol number concentration is usually distributed in three modes (Balternsperger et al., 2002; Lonati et al., 2011). One mode with diameters in the range of 20-30 nm, consisting of hydrophobic and highly volatile organic material originating from combustion (Baltensperger et al., 2002), plus new particles from nucleation events that occur on about 35% of the days in Po Valley (Hamed et al., 2007). The other two modes are in the submicron range (dry diameters 50-200 nm), one almost hydrophobic, related to primary emissions (e.g. soot), and the other hydrophilic, related to secondary aerosols (Balternsperger et al., 2002).

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The aerosol vertical profile in Milan and in the wider Po Valley region was characterized by means of aircraft, Lidar, and tethered balloon measurements (Highwood et al. 2007; Barnaba et al., 2007; Barnaba et al., 2010; Crosier et al., 2007; Angelini et al., 2009; Ferrero et al., 2010; Ferrero et al., 2011). Similarly to other polluted ~~places located in a~~ valley areas, two layers with distinct characteristics are often found. One in the PBL which is~~,~~ humid, rich in~~of~~ fresh emissions, ~~and~~ with ~~a~~ the nitrate profile increasing with height, and another layer, above the PBL, with more aged aerosols enriched in the sulfate and organic matter fraction (Highwood et al., 2007; Crosier et al., 2007; Ferrero et al., 2010). This~~e~~ decoupling into two layers is attributed to the mountain-valley breeze dynamics (Angelini et al., 2009) and to the sporadic arrival of long-range transported Saharan dust (Barnaba et al., 2007) or biomass burning plumes (Barnaba et al., 2011). The number concentration of fine mode (d < 1.6 µm) particles are found to be relatively constant with height in the PBL, and it decreases by a factor of 2-3 above the PBL. In contrast, ~~C~~coarse particle (d > 1.6 µm) number concentrations~~, in contrast,~~ display a decrease with height also in the PBL, due to sedimentation processes (Ferrero et al., 2010).

3 Experimental setup

Ground-based and vertical profiles measurements used in this study were conducted at Torre Sarca site which is located on the northern side of Milan (45°31’19”N, 9°12’46”E; within the Milano-Bicocca University campus), in the midst of an extensive conurbation that is the most industrialized and heavily-populated area in the Po Valley. We report here a brief description of the experimental setup and provide relevant references for further details.

3.1 Particulate matter bulk composition and number size distribution, and gas-phase composition

At ground level, $PM_{2.5}$ and PM_1 (EN-14907) samples were gravimetrically collected using the FAI-Hydra dual channel Low-Volume-Sampler (LVS; 2.3 m^3 h^{-1}, 24 hours of sampling time, PTFE filters for PM_1, ore-fired Quartz fibre filters for $PM_{2.5}$, ∅=47 mm), while the aerosol number-size distribution was constantly monitored using an Optical Particle Counter (OPC; Grimm 1.107 “Environcheck”, 31 class-sizes ranging from 0.25 µm to 32 µm). Further details are given in Ferrero et al. (2014).

The aerosol chemistry was assessed on $PM_{2.5}$ samples for the ionic fraction, EC and OC, respectively. For the purpose of ions' analysis, $PM_{2.5}$ samples were extracted in 3 mL of ultrapure water (Milli-Q®; 18.2 MΩ·cm) for 20 minutes using an ultrasonic bath (SONICA, Soltec, Italy). The obtained solutions were then analysed using a coupled ion chromatography system~~s~~ consisting of: 1) a Dionex ICS-90 (CS12A-5 Analytical column) with an isocratic elution of methanesulfonic acid (20 Mm; 0.5 mL/min) whose signal was suppressed by means of tetrabutylammonium hydroxide (0.1 M; CMMS III 4 mm MicroMembrane Suppressor) for cations (Na^+, K^+, Ca^{++}, Mg^{++} NH_4^+) and, 2) a Dionex ICS-2000 (AS14A-5 analytical columns) with an isocratic solution of $Na_2CO_3/NaHCO_3$ (8.0 mM/1.0 mM; 1 mL/min) whose signal was suppressed by means of sulphuric acid (0.05 M; AMMS III 2 mm MicroMembrane Suppressor) for anions (F^-, Cl^-, NO_3^-, $SO_4^=$).

EC and OC were determined in $PM_{2.5}$ using the Thermal Optical Transmission method (TOT, Sunset Laboratory inc.; NIOSH 5040 procedure, http://www.cdc.gov/niosh/nmam/pdfs/5040f3.pdf). The organic matter (OM) fraction was then estimated from OC using a coefficient to account for the presence of hetero-atoms (H, O, N, etc.). Following the work of Turpin and Lim (2001), the chosen factor was 1.6 for the urban Torre Sarca site.

Finally, meteorological and gas-phase (NO_x, O_3) observations at ground-level were taken from the weather and monitoring stations operated in Milan by the local regional environmental protection agency (ARPA Lombardia).

3.2 Size-segregated aerosol composition

From July 14 (8:00 local time LT) to 18 (8:00 LT)~~, July~~ 2007, size segregated daytime (8:00 to 21:00 LT) and night-time (21:00 to 8:00 LT) aerosol samples were collected by means of a five-stage Berner impactor (LPI 80/0.05) with 50% size cut at 0.05, 0.14, 0.42, 1.2, 3.5 and 10 µm aerodynamic diameter. Substrates were off-line analyzed for the determination of the carbonaceous - water soluble organic (WSOC) and water insoluble (WINC) carbon - and soluble inorganic components (NH_4^+, Na^+, K^+, Ca^{2+}, Mg^{2+}, Cl^-, NO_3^-, SO_4^{2-}). Mass-to-carbon ratios of 1.8 and 1.2 were used to convert WSOC to the corresponding mass, WSOM (water-soluble organic matter) and WINC to WINCM (water-insoluble carbonaceous matter), respectively. A complete description of the sampling and analytical methods adopted is reported in Carbone et al. (2010) and references therein. In the analysis presented here, we ~~use~~ only used the total mass of aerosol components (sum over size bins).

3.3 Lidar-ceilometer profiles

Lidar-ceilometers (called Lidar for brevity in the manuscript) operate on the same physical basis of more complex research-type lidars, but are compact systems, generally with a lower laser energy power, capable of operating 24 hours per day, unattended and in all weather conditions. Initially developed for cloud-base determination, the technology of these system is now mature enough to represent a very convenient and widely used tool for the operational monitoring of the atmospheric aerosol and of relevant meteorological parameters (e.g. Haeffelin et al., 2012).

A lidar-ceilometer (Vaisala LD-40) operating at 855 nm collected aerosol profiles at the Milan Torre Sarca site in the ~~period~~ January 2007-February 2008 period. The system was switched on during selected dates (and mostly when meteorological conditions allowed the contemporary launch of balloon-borne aerosol instruments, Ferrero et al. 2010), collecting a database of more than 200 days of measurements. During the selected dates, the Lidar-ceilometer operated 24 hours per day, collecting aerosol profiles every 15 seconds that were afterwards averaged over 15 minutes to achieve a better signal-to-noise ratio. Due to the instrumental limitations, the lowest altitude the system can observe is about 60 m. After the background noise is subtracted from the collected backscattered signal, the range-corrected signal (RCS, i.e., the signal S times the square of the system-to-target distance R) is derived to extract information on the aerosol vertical distribution. More details on the system and measurements capabilities can be found in Angelini et al. (2009) and Di Giuseppe et al. (2012).

4 WRF/Chem model

4.1 Description and Setup

The ~~version~~ 3.4.1 version of the Weather Research and Forecasting model with Chemistry (WRF/Chem), with some updates, is used in order to interpret the observed concentrations of aerosol and its composition at surface and along vertical profile of PBL. WRF/Chem is a coupled on-line model where meteorological and chemical processes are fully consistent (Grell et al., 2005).

The model is configured with two 1-way nested domains centred on Northern Italy (Po Valley). The mother domain covers Western Europe with 131 × 95 cells at a horizontal resolution of 30 km, the nested domain covers Northern Italy with 109 × 91 cells at a resolution of 10 km. The vertical grid is made of 33 eta levels up to 50 hPa, with first five levels centred approximately at 12, 36, 64, 100 and 140 m above the ground and 12 levels below 1 km.

The physical and chemical parameterizations used are the same for the two domains, and are listed in Table 1~~Table 1~~. These include the Rapid Radiative Transfer Model for short and long wave radiation (Iacono et al., 2008), the Mellor-Yamada Nakanishi-Niino boundary layer parameterization (Nakanishi and Niino, 2006), the Noah Land Surface Model (Chen and Dudhia, 2001), the Morrison cloud microphysics scheme (Morrison et al., 2009), and the Grell 3D ensemble cumulus scheme, which is an update version of the Grell-Devenyi scheme (Grell and Devenyi, 2002). Cumulus clouds feedback with radiation is activated.

The gas-phase chemistry is simulated with an updated version of the Regional Atmospheric Chemistry Mechanism (RACM) that includes a wide range of chemical and photolytic reactions for organic and inorganic species (Stockwell et al., 1997). Aerosol parameterization adopted is the Modal Aerosol Dynamic for Europe (Ackermann et al., 1998) that uses three overlapping lognormal modes for Aitken, accumulation and coarse particles. Thermodynamic equilibrium for inorganic species is calculated ~~with~~ using the ~~RPMARES~~ MAR-A module (Saxena et al., 1986; Binkowski and Roselle, 2003). The Secondary Organic Aerosol (SOA) production is calculated using the Volatility Basis Set (VBS) scheme implemented in WRF/Chem by Ahmadov et al. (2012), which include the oxidation of anthropogenic and biogenic VOC currently believed to be important for SOA production (alkanes, alkenes, xylenes, aromatics, isoprene, monoterpenes, and sesquiterpenes).~~.~~ To our knowledge, this study is the first application over Europe of this new parameterization for SOA yield ~~with~~ by means of WRF/Chem. Photolysis rates are estimated with the Fast-J scheme (Wild at al., 2000). The dry deposition flux is simulated with the scheme ~~of~~ by Wesely et al. (1989), and the dry deposition velocity of organic vapours is assumed to be the 25% that of nitric acid (HNO_3). Cloud chemistry in convective updraft is parameterized following Walcek and Taylor (1986). Wet deposition ~~by~~ due to convective and large scale precipitation is also included in our simulations. The aerosol optical properties are calculated online with the package ~~of~~ by Barnard et al. (2010), using the volume average internal mixing assumption. We included the direct effect of aerosol on radiation, but ~~we switch off~~ the indirect aerosol effects on clouds were switched of since~~, a~~ this function ~~that~~ is still under testing with the SOA VBS scheme (Tuccella et al., manuscript ~~in preparation~~submitted to Geoscientific Model Development Discussion).

In order to ~~help~~ enhance the understanding of the influence of the upper-~~-~~level~~s~~ processes on the pollutant budget at surface, we use the diagnostic of the tendency terms in the continuity equation for chemical species following Wong et al. (2009). We extended the original module, which included only some gas-phase compounds, ~~also~~ to include aerosol species and processes as well. Diagnosed terms are: emission, horizontal and vertical advection, photochemistry (gases and aerosols), vertical mixing plus dry deposition (these cannot be separated in the WRF/Chem implementation), convective transport, aqueous chemistry, and wet deposition.

We have simulated the period from the ~~25~~ June 25 to the ~~18~~ July 18, 2007, ~~and~~ discarding the first 10 days as spin up. Simulation on the mother domain uses initial and boundary meteorological conditions provided by the National Center for Environmental Prediction (NCEP) 6-hourly analyses, having a~~n~~ horizontal resolution of 1° × 1°. For the mother domain, chemical boundary conditions are provided with WRF/Chem default idealized vertical profiles, representative of Northern hemispheric, mid-latitude and clean environmental conditions (McKeen et al., 2002; Grell et al., 2005; Tuccella et al., 2012), while boundary conditions to the nested domain are provided by the mother domain. The simulations are carried out at 24 hours time-slots, starting at 12:00 UTC of each day and then run for 30 hours, with the first 6 hours considered as model spin-up. Chemical fields are restarted from previous runs.

4.2 Emissions

Total annual 2007 anthropogenic emissions of nitrogen oxides (NO_x), carbon monoxide (CO), sulphur oxides (SO_x), ammonia (NH_3), Non-Methane Volatile Organic Compounds (NMVOC), unspeciated particulate matter ($PM_{2.5}$ and coarse PM), primary organic carbon (OC), and elemental carbon (EC) are taken from the Netherlands Organization for Applied Scientific Research (TNO) database (Kuenen et al., 2014). Annual TNO anthropogenic emissions consist of gridded data from ten source types (SNAP sectors) with horizontal resolution of 1/16° latitude by 1/8° longitude (about 7 × 7 km^2).

TNO emissions are adapted to WRF/Chem following the methodology used by Tuccella et al. (2012), with minor changes derived from the second phase of the Air Quality Modelling Evaluation International Initiative (AQMEII) (Alapaty et al., 2012, Im et al., 2014a,b).

Biogenic emissions are calculated online using the Model of Emissions of Gases and Aerosols from Nature (MEGAN) (Guenther et al., 2006). Sea salt flux is calculated online, while dust source is not included.

5 Results

5.1 Preliminary analysis of the observations

In Figure 1 ~~we show~~ time-series of ground-based meteorological and physical-chemical observations performed in Milan in ~~the period 5-20~~ July 5-20, 2007 period are shown. The large scale circulation is illustrated in Figure S1, while the evolution of cloud cover over Northern Italy is illustrated by MODIS-Aqua true colour images in Figure S2. The period starts with a low-pressure system over Germany, rapidly moving Eastward, and allowing a pressure increase over Northern Italy from ~~5 to 8 of~~ July 5 to 8, associated with fair weather and sparse clouds. From ~~9 to 11~~ July 9 to 11, a North Atlantic low-pressure system induces a significant increase of cloud cover over Milan with light rain on July 10. From July 12, a wide anticyclonic system forms over the Western Mediterranean, warranting clear sky and stable conditions until July 20 and ~~beyond~~later. Maximum daily temperature is around 30°C before the Atlantic perturbation, then it increases steadily (from 25° to 35° C) at a rate of ~ 2 °/day from 11 to 15 as the high-pressure system settles. Humidity is high at night (above 70%) on the days following the low-pressure passage, then the atmosphere gradually dries out under the anticyclone.

During the period preceding the Atlantic perturbation (~~5-8~~ July 5-8, 2007), wind is prevalently westerly daytime, forced by the large scale circulation, with wind speed around 2.5-3 m/s. Wind is slowed down to less than 1 m/s at night, because the downward transport of momentum toward the surface is inhibited by the nighttime vertical stratification (Stull, 1988; Whiteman, 1990). Wind speed increases up to 5 m/s at the passage of the low-pressure system (~~9-11~~ July 9-11, 2007), and stays above 2 m/s also nighttime. From July 11, when the high-pressure over the Mediterranean begins to settle, the wind field adjusts to a typical mountain-valley breeze regime (Whiteman, 1990). Starting from midnight, the slow (~ 1 m/s) northerly flow gradually accelerates and rotates clockwise, reaching peak speeds of ~ 3 m/s in the afternoon at south-westerly direction, then gradually slows down and return northerly. This wind pattern favours conditions of stagnation and recirculation of air within the valley, allowing the build-up of pollutants from a day to the next. Figure S3 shows the simple stagnation and recirculation indices proposed by Allwine and Whiteman (1994) and confirms that the only ventilated period is that of the Atlantic perturbation.

The passage of the Atlantic low-pressure system on ~~9-10~~ July 9-10 marks a sort of “restart” for the atmospheric composition at ground level. Indeed, relatively longer lived (few days) chemical species, such as ozone and PM, first accumulate during the days preceding the perturbation, then are suppressed in perturbed weather, and finally re-accumulate afterwards (Figure 1~~Figure 1~~ c,d). Outside the perturbed period, ozone and nitrogen oxides (NO_x) follow a daily cycle typical of that observed in many urban areas (Mavroidis and Ilia, 2012, and references therein). The primary pollutant nitric oxide (NO) displays a sharp peak during morning rush hours (between 6 and 9 Local Solar Time), then gradually decreases during the day. It displays a secondary small peak during evening rush hours (20-22 LST), then ~~stays~~ remains at low values until the following morning. Nitrogen dioxide (NO_2) is mainly originated from the oxidation of NO by ozone and peroxy radicals (Jenkin and Clemitshaw, 2000), and displays peaks delayed by ~ 1 hour with respect to those of NO. It shows a plateau between the morning and the evening peak, because concentrations are sustained daytime by photochemistry. The photolysis of NO_2 is the main tropospheric source of atomic oxygen (O) that readily reacts with molecular oxygen (O_2) to produce ozone. Indeed, during daylight hours, NO, NO_2 and O_3 equilibrate on the so called “photostationary equilibrium” on time scales of minutes (Clapp and Jenkin, 2001).

Ozone is depleted during the morning rush hours by reaction with NO, then it is photochemically formed during the day and peaks during late afternoon (14-16 LST), and thereafter gradually decreases to lower nighttime levels. In fair weather, the daily cycle of ozone and NO_x is regulated by the solar radiation, the dilution of fresh emissions from the surface in the growing daytime PBL, the vertical mixing with air entrained from the residual layer and the free troposphere above the PBL, and the dry deposition at the surface. Past studies pointed out that the entrainment from ozone-rich residual layer may be as important as the photochemical production in the PBL during pollution events even in urban atmospheres (e.g. Zhang and Rao, 1999). In the present case, the build-up of ozone in the days following the perturbation is evident, but it is difficult to discern the relative role played by the local photochemical production and by the vertical mixing on the ozone trend observed at the surface.

Accumulation and cleansing of the atmosphere near the surface is even more evident from aerosol time-series (Figure 1~~Figure 1~~ d-g). PM2.5 and PM1 follow a similar trend, while PM10 often show a different behaviour, pointing out the presence of additional sources to the coarse fraction, most probably the erosion and resuspension of soil material by vehicles and wind. The aerosol mass is shown to build up before the Atlantic perturbation (PM10 around 20-30 $\mu g/m^3$) and to abruptly decrease (PM10 below 10 $\mu g/m^3$) during the low-pressure system passage (probably because of a combination of enhanced ventilation, wet deposition processes, and soil erosion inhibited by increased soil moisture). Afterwards, PM concentration keep increasing after the low-pressure passage (maximum PM10 values of more than 60 $\mu g/m^3$ reached on July 18-19). Daily cycle of the fine aerosol mass (PM2.5 and PM1) displays similarities with that of NO, in particular a similar morning peak, indicating the important role played by primary emissions. This is confirmed by the analysis of aerosol speciation (Figure 1~~Figure 1~~ e), which shows high values of elemental carbon (EC, 2-4 $\mu g/m^3$) and insoluble carbonaceous matter (WINCM, 2-10 $\mu g/m^3$). The latter makes, on average, 40-50% of the PM1 mass (Carbone et al., 2010). Major secondary species are inorganic ions (sulfate, nitrate, and ammonium) and part of the organic matter, which may be associate with its water soluble fraction (WSOM, Carbone et al., 2010). Similarly to ozone, secondary aerosol accumulates during the days preceding and following the perturbation.

Cleansing of the atmosphere after the perturbation and subsequent recover of the aerosol load is also clearly visible in the number concentration timeseries. At the passage of the perturbation, aerosol number rapidly decreases by more than an order of magnitude at all observed size ranges, then returns to the pre-perturbation levels on a time scale of about two days. We note, however, differences in the aerosol regime before and after the perturbation. Before the cleansing, the aerosol size distribution is locked to a fixed shape, with no or little daily variability. Conversely, in the stable conditions of July 12-19, it displays a clear daily cycle with a growth towards larger sizes in daytime, and a return to narrower distributions nighttime.

As mentioned in sec. 3.3, Lidar observations are only available in the days following the perturbation and give useful indications on the aerosol vertically-resolved infra- and inter-diurnal variability (e.g. Angelini et al., 2009). During the morning hours, a layer of aerosol is formed under the growing boundary layer. There, fresh emissions from the surface are diluted and mixed vertically in the PBL. Throughout the period, but especially on some days such as in the mornings of ~~13 and 15 ~~July 13 and 15, an enhanced layer of aerosol is visible in the upper levels near the top of the PBL. Aerosol is subsequently partly removed in the second half of the day by the mountain breeze, while a residual layer with relatively high aerosol content may survive above the nocturnal PBL (e.g. on July 13, 15, and 16~~ July~~). This layer may potentially be entrained the following morning into the PBL and contribute to the surface aerosol budget. On the last days displayed in Figure 1, a further aerosol layer between 2 and 3 km appears in the Lidar signal. As indicated by increased coarse fraction AOD at Modena AERONET station (Figure S4) and model backtrajectories (Figure S5), it is a Saharan dust incursion which is probably entrained at ground level, as indicated by the enhancement of PM10 levels on days of July 18-19~~ of July~~. Since Saharan dust intrusions are not modelled here, these days are excluded from the analysis.

From the measurements ~~we ~~reported here~~, at least a few~~ some questions~~, ~~ emerged~~which we address in what follows, emerge~~:

1. What is the composition of the aerosol layer formed during the day in the upper PBL?
2. How much of the aerosol burden measured at the ground is due to localized processes and how much is conversely due to processes occurring in the upper PBL and to the subsequent mixing in the lowermost levels? In other words, how important is the interplay

between surface and upper layers in shaping the aerosol mass we measure near the ground?

3. How much may the residual layer above the PBL ~~may~~ contribute to the aerosol budget at ground level the next day?

We attempted to provide answers to these questions using simulations with the WRF/Chem model and relevant comparison with the observational dataset.

5.2 Model verification against available observations

Before drawing conclusions on the scientific questions outlined at the end of the previous section, we verified~~y~~ our model simulations against the dataset of observations depicted in Figure 1~~Figure 1~~ and~~.~~ ~~We~~ onnnly displayed results only for the nested domain over Northern Italy, ~~and~~ using~~e~~ statistical indices defined in Appendix A~~B~~ as a guidance to quantify model biases.

In Figure 2~~Figure 2~~ we compared observed and simulated meteorological variables at ground level in Milan for the period July 5-17, ~~July~~ 2007. The temperature is underestimated by about 2.5°C, which is probably due to not well resolved dynamics and heat fluxes in the urban boundary layer. The overestimation of relative humidity of about 10% is mostly attributable to the underestimation of temperature. Wind speed is overestimated by 0.8 m/s and has a relatively low correlation of 0.29 with observations, thus fitting to typical characteristics of current mesoscale models (e.g Misenis and Zhang, 2010). The simulated wind speed is also more variable than that observed as denoted by the RMSE of 1.7 m/s. The wind direction is generally captured well, in particular the mountain-valley cycles after the passage of the perturbation of July 9.

In Figure 3~~Figure 3~~ we show comparison of gas-phase observations and simulation near the ground. The daily cycle of NO is reproduced quite well (r = 0.52), the timings of the morning peak and the subsequent decrease are captured by the model. The magnitude of the morning peak does not show a tendency neither to underestimation nor to overestimation, while NO values for the rest of the day are underestimated, resulting in a bias of -4.1 ppb (-60%). The model is also able to capture the basic features of the NO_2 daily cycle, i.e. the morning and evening peaks and the minimum at night. However, values are generally underestimated (bias of -8.3 ppb or -34%) and the trend on weekly time scale display much less variability than that observed. Ozone display a very low systematic bias (-2.3 ppb), but less variability than observations (RMSE of 11.3 ppb), and a correlation of 0.65. The timing of the daily cycle is captured well, with a maximum in the afternoon, a secondary peak around midnight, and a minimum during the morning rush hour.

In Figure 4~~Figure 4~~ we compare PM10 and PM2.5 simulated mass to hourly observations at ground. The PM10 trend is qualitatively captured by the model, displaying the sharp decrease at the passage of the perturbation on July 10 and the subsequent gradual accumulation in the following days. This leans confidence in the simulated removal and production terms, and the resulting negative bias is ~~small~~ low (-4 µg/m^3 or -10%). The model also captures some of the characteristics of the daily cycle (r = 0.57), however the observed signal is quite irregular, and the model does not reproduce all the variability. The negative bias of PM10 could be partly explained by the missing source from soil dust erosion and resuspension in the model. For PM2.5 the general features of the comparison are similar to PM10, but the model has a positive bias (+4 µg/m^3 or +70%), mostly attributable to few spurious peaks in the simulation. The overestimation of PM2.5 partly compensates and masks the underestimation of coarse particles (PM2.5-10). The comparison of the simulated number size distribution against that observed with the OPC (not shown) suggests that the high bias of PM2.5 is attributable to aerosol in the size range 0.5-1 µm.

In Figure 5~~Figure 5~~ we show the comparison of simulated PM2.5 composition with daily and bi-daily samplings near the ground. In the period precedent to the perturbation (July 5-9), the model underestimates the magnitude of the observed peak of sulfate and ammonium, but it reproduces ~~well the~~ subsequent “restart” and recovery well. Observed nitrate displays a relatively smooth trend, with a slight decrease at the passage of the perturbation and almost constant levels during the rest of the period. Modelled nitrate has a much more variable behaviour, which ~~looks like~~seems to be characterized by sudden and irregular pulses. The bi-daily observations indeed suggest that the daily average observation masks much of the underlying variability associated to nitrate. Recently reported hourly measurements of PM composition in the Po Valley indeed confirm the same “pulsed” behaviour of nitrate near the ground, with values near zero during daytime, and irregular peaks at nighttime (Decesari et al., 2014). This highlights the inherent difficulties in simulating the nitrate concentrations at sub-daily frequency. Elemental carbon, being primary and almost hydrophobic, is largely unaffected by the perturbation. This feature is captured by the model, but EC values are underestimated by a factor of two, probably due to underestimated emissions. Interestingly, the bi-daily observations of WINCM (EC plus primary insoluble organic material) display a large diurnal cycle (maximum at night and minimum during the day) which is not captured by the model. Organic carbon trend and magnitude is reproduced quite well, with the exception of a large spurious peak on July 8-9 not seen in the observations. The peak is associated with secondary organic aerosol (not shown). The bi-daily observations of soluble organic material (WSOM) do not show the strong daily cycle of primary carbonaceous matter, and confirm a tendency of the model at overestimating the SOA fraction.

In Figure 6~~Figure 6~~ we qualitatively compare the Lidar profiles with the simulated PM2.5 profiles. A quantitative comparison would require the calculation of optical properties of simulated PM2.5 and subsequent solution of the Lidar equation (Hodzic et al., 2004~~6~~). However, in first approximation Lidar signal may be associated to PM2.5 mass. The model captures some of the basic features of the previously described aerosol profile cycle observed in this period (sec. 5.1). Every morning a plume of fresh aerosol detaches from the ground and follows/traces the growing boundary layer until its maximum extension in the central part of the day. ~~Then, in the evening, the mountain-valley breeze cleans the lower PBL, often leaving an upper air aerosol residual layer above the PBL. When the residual layer survives the night, it might be entrained down to the ground on the subsequent morning. This is visible e.g. on July 16-17, in both observations and simulations.~~ Then, in the afternoon, the mountain-valley breeze cleans the lower PBL (note the abrupt abatement of both the Lidar and the model aerosol signals in the second part of the day), often leaving an upper air aerosol residual layer above. Model simulations also reproduce such residual layers (note the afternoon increase of PM2.5 values in the upper levels, particularly visible on July 15-16). When such residual layers persist overnight, the Lidar shows these to entrain into the developing PBL the day after (note the merging of the upper level aerosol layers with the growing, aerosol-traced PBL in Figure 6a, particularly evident in the morning of July 14 and 15). There are also hints of the same features in model simulations.

5.3 Insights into the budget of aerosol vertical profile over Milan

The “chemical restart” caused by the passage of the perturbation on July 9-10, and the following settl~~e~~ing of an almost periodic circulation pattern, naturally creates favourable conditions for a study of the processes yielding aerosol production and accumulation in the area of Milan. ~~We thus~~Our analysis shall now focus ~~our analysis~~ on the days ~~that~~ following the perturbation (July 12-17).

Using model output, we firstly examined the composition of the aerosol layers noted in the Lidar profiles of Figure 6~~Figure 6~~. In Figure 7~~Figure 7~~, we show the composition of PM2.5 simulated over Milan. The model predicts a major role played by the primary fraction (unspeciated anthropogenic, black carbon, and primary organic carbon~~inorganic and organic~~), which is largely responsible for the two rush hours peaks (morning and evening) and the bulk of aerosol mass in the PBL. Fresh emissions are mostly concentrated near the ground and turbulent transport dilutes them in the PBL during the day. A relatively small fraction (~ 30%) of primary aerosol remains above the PBL overnight and contributes to the upper aerosol layers seen by the Lidar.

The sum of secondary species contributes 40-60% of the aerosol mass in the PBL, but with remarkable differences in the vertical distribution of single components. Sulfate and Secondary Organic Aerosol (SOA) start to form and dilute under the PBL a few hours after ~~the~~ sunrise, contributing in a relatively homogeneous way to the aerosol column in the PBL. Anthropogenic SOA (ASOA) contributes more than biogenic SOA (BSOA) to the SOA budget. The concentration of those secondary species are similar also above the PBL, thus significantly contributing to the upper aerosol layers. ASOA are slightly more persistent than BSOA and sulfate in the free troposphere.

Nitrate displays a peculiar profile, with enhanced concentrations in the upper part of the PBL formed during the central part of the day. These concentrations may largely exceed those found near the ground (i.e. on July 13, 16, 17). Moreover, nitrate is predicted to be the major secondary species contributing to the formation of the residual aerosol layers above the PBL. Enhanced upper level concentrations of nitrate into PM1 were also reported at Monte Cimone (a mountain peak of 2160 m at the southern border of the Po Valley) by Carbone et al. (2010, 2014).

In Figure 8~~Figure 8~~ we show the maps of simulated sulfate and nitrate over the Po Valley on ~~13~~ July 13, 2007 at 16 LST at the surface and at 750 m height. It can be seen that the main features of the composition of the aerosol profile outlined above are not peculiar of the Milan area, but are suggested to be representative of the larger area of the Po Valley.

In order to better understand the processes underlying the predicted characteristics of the aerosol over Milan, we ~~look at~~analysed the terms of the continuity equation for chemical species. Budget terms considered are horizontal and vertical advection, chemistry, turbulent mixing and dry deposition, emission. Terms related to cloud processes (convection, aqueous chemistry, wet deposition) make a very small contribution in the dry period under investigation and will not be shown to improve the figure’s clarity. In Figure 9~~Figure 9~~ we show the vertical profile of the budget terms for sulfate and nitrate at 16 local time of July 13 over Milan. For sulfate, the dominant terms are those related to advection, indicating the presence of spatially distributed sources and a relatively long lifetime, making it a regional scale pollutant. Locally, sulfate is both directly emitted and produced by secondary pathways throughout the PBL. Turbulent mixing distributes it vertically in the PBL and dry deposition removes it from the atmosphere near the ground, determining an almost homogeneous sulfate profile in the PBL. Conversely, nitrate has relatively low contribution from advection, while the largest terms are chemistry and vertical mixing. In the simulation, nitrate is produced only in the upper half of the PBL and destroyed in the lower half. The vertical transition between the nitrate destruction and production zone is quite sharp. Turbulent mixing is nearly in equilibrium with chemical production, indicating that model simulates a very rapid adjustment to the thermodynamic equilibrium for the sulfate-nitrate-ammonium system. This results in nitrate concentrations higher in the upper part of the PBL ~~with respect~~compared to the lower part.

Similarly to nitrate, ~~also~~ SOA also displays an enhanced net chemical production in the upper part of the PBL and destruction in the lower part (Figure 10~~Figure 10~~), but since the chemical and vertical mixing terms are of the same order of the advection terms the resulting vertical profile is almost constant with height, similar to that of sulfate.

Further insights into the simulated sharp transition to an environment favourable to nitrate formation in the upper part of the PBL, is investigated by means of several model sensitivity tests as outlined in Table 2~~Table 2~~. In Figure 10~~Figure 10~~ we first look at the gas phase precursor of nitrate, nitric acid (HNO_3). The left panel shows the vertical profile of the budget terms for HNO_3 at the same instant of Figure 9~~Figure 9~~. The chemical and vertical mixing terms mirror those of particulate nitrate, resulting in a decreasing concentration profile with height. The right panel of Figure 9 shows the budget profile from a sensitivity simulation where aerosol chemistry is switched off (AERO, see Table 2~~Table 2~~). The chemistry and vertical mixing terms are greatly reduced and are the same order of magnitude of advective terms, indicating that the sharp gradients in net chemical production of HNO_3 (and nitrate) are dominated by aerosol processes, and not by gas-phase processes.

In Figure 12~~Figure 12~~ we provide further elements to evaluate the simulated particulate nitrate thermodynamics. Ambient relative humidity increases with height in the PBL, from a minimum of ~50% near the ground to a maximum of ~80% at an altitude of 1000 m (~400 m below the PBL top). The nitrate chemical production term shown in Figure 9~~Figure 9~~ is reported for ease of comparison, and displays the already noted peak between 500 and 1000 m. The sulfate ratio (ratio of total ammonia and sulfate) is well above the threshold of 2 along the profiles (not shown), thus suitable for particulate nitrate formation (Seinfeld and Pandis, 2006). The profile of equilibrium constants for both the aqueous and solid nitrate increase with height, in response to a decreasing temperature profile (not shown), indicating that conversion of nitric acid to particulate is favoured with increasing height. However, no sharp transitions, correlated to the nitrate net chemical term, can be noticed in the profiles of those equilibrium constants.

The profile of ammonium nitrate's deliquescence relative humidity (DRH) helps disclosing the possible reason for such a transition. At ground level, ambient RH is well below the ammonium nitrate DRH, indicating an environment thermodynamically favourable only to the solid form of nitrate. However, since the RH gradient with height is steeper than that of DRH, the two curves intersect at an altitude of ~500 m, and then again at ~1300 m, because of the RH decrease near the PBL top. Ambient RH is thus higher than ammonium nitrate DRH in the same altitude range (~500-1000 m) where the nitrate net chemical production peaks. This indicates that, over Milan and in the period under consideration, the nitrate chemical production is dominated by aqueous conversion of nitric acid to nitrate ion, condition that is reached only in the upper part of the PBL, where RH levels are high enough to sustain the formation of an aqueous solution containing nitrate. Although the real multicomponent DRH point will differ from that of pure nitrate, it is known that the DRH of mixtures is always lower than that of pure salts (Seinfeld and Pandis, 2006). The thickness of the layer favourable to aqueous nitrate formation deducible from Figure 12 may thus be regarded as a conservative lower estimate. During daytime, the nitrate formed in the upper boundary layer re-evaporates back to the gas phase when brought to the ground by vertical motions, and that's the origin of the inhomogeneous vertical profile of nitrate. For further discussion on how much the upper aerosol layer contributes to ground PM we point the reader to the next paragraph.

The budget analysis we have presented so far reveals of a complex interplay between chemical processes and vertical mixing taking place at different altitude ranges. In order to better quantify the impact of chemical production at upper layers on particulate matter at ground level, we perform three tests alternatively switching on/off the chemical process at selected altitude ranges (namely within the lower half of the PBL, the upper half of the PBL and above the PBL, see Table 2~~Table 2~~). Results are shown in Figure 13~~Figure 13~~ for PM2.5, and its components sulfate, nitrate and SOA. In the figure, the contribution to the ground PM2.5 of the chemical processes in the different altitude ranges is positive/negative when the associated sensitivity line is below/above the CTRL. For PM2.5, we ~~find~~ have found that chemical process in all regions positively contribute to the ground level concentration. During the first days after the passage of the perturbation, the shutdown of secondary chemical formation makes very little difference, indicating a dominance of primary emissions. As ~~the~~ time goes by, secondary processes gain importance, but primary fraction remains the main driver of PM2.5 concentration even after a week. Interestingly, the magnitude of the relative contribution of the different layers (lower PBL, upper PBL, above PBL) to ground level PM2.5 is comparable, and of the order of up to 7-8 µg/m^3 each, on hourly basis. Exceptions are noted on afternoons of July 13 and 16, when a negative contribution from secondary processes in the lower PBL is simulated (note the blue dashed line above the red line). These peaks are associated with the nitrate sink in the lower PBL (see panel c). Sulfate has an identical contribution from lower and upper PBL chemical production, and may also have a very important contribution from the region above the PBL, even higher than processes in the PBL (e.g. on July 17). SOA budget is similar to that of sulfate, but with an enhanced contribution from PBL processes versus those above it. As expected, nitrate displays distinctive features. Chemical production in the lower PBL positively contributes to ground level concentration in the first part of the day, then in the afternoon results in a net destruction. On the other hand, processes in the upper PBL and above PBL always positively contribute to the ground level nitrate concentrations.

A further quantitative assessment of the impact of upper aerosol layers on ground concentrations can be estimated combining information in Figure 14~~Figure 14~~ and Figure 6~~Figure 6~~. In Figure 14~~Figure 14~~ we show the time-series of the difference in the simulated

$PM_{2.5}$ profile between APBL and CTRL runs. When a residual layer is visible, we may roughly estimate from the figure the related change near the surface on the subsequent morning. We focus ~~the~~ our attention on July 17, when the presence of a residual layer is clearly visible. The concentration change (APBL – CTRL) in the residual layer is about 8-10 µg/m^3. The following morning the concentration change near the surface is 4-5 µg/m^3, thus we may estimate a 50% sensitivity of ground $PM_{2.5}$ to a change in the residual layer. In Figure 6b, we see that on July 17 the PM2.5 concentration in residual layer is 20-24 µg/m^3, thus the expected impact on hourly concentrations near the ground is of the order of 10-12 µg/m^3, or about 40% of the PM2.5 concentration near the ground. This is the extreme case in the short period analyzed here, but gives a feeling of the potential importance that entrainment of aerosol layers aloft may occasionally have on PM2.5 observed near the surface.

6 Conclusions

The object of this study is the analysis of the role played by the combination of chemical and dynamical processes occurring throughout and above the PBL in determining the aerosol concentration and composition we observe near the ground. We analyzed the observations of the atmospheric composition during a period of two weeks carried out in Milan (Northern Italy) ~~during~~ in July 2007. The period was characterized by the passage of a perturbation that favoured cleansing of the Po Valley, providing a natural “chemical restart”. After the perturbation, stable high-pressure conditions determined the establishment of a nearly repetitive meteorological pattern, driven by a mountain-valley breeze system, that allowed for a gradual re-accumulation of pollutants.

Lidar observations after the “chemical restart” revealed intriguing features of the aerosol vertical profile over Milan. Every morning, a plume of fresh emissions from the ground is dispersed in the growing convective boundary layer. In the afternoon, an enhanced aerosol layer appears in the upper part of the PBL, while in the evening the bottom part of the PBL is cleansed by the mountain breeze. A residual aerosol layer may form and survive the night above the PBL, and may be entrained again down to the ground the day after. We investigated how this “vertical” sequence of processes affect the aerosol concentrations observed at ground level.

With the help of simulations from the state-of-art online meteorology-chemistry model WRF/Chem we attempted to answer ~~to~~ three main questions suggested by the observations. The questions and the relat~~ed~~ive answers are summarized here below:

- What is the composition of the aerosol layer formed during the day in the upper PBL?

  Model simulations suggest that 40-60% of the fine aerosol in ~~the~~ Milan's summer PBL is of primary origin, consistently with previous studies (e.g. Carbone et al., 2010). This primary fraction displays a decreasing concentration profile with height in the PBL, since the sources are concentrated near the ground and species are vertically mixed by turbulence. Sulfate and secondary organic aerosol are produced throughout the PBL and have a nearly homogeneous profile there. Nitrate and ammonium have a distinct profile, with enhanced values in the upper PBL, where concentrations may be much higher than those near the ground. The low temperature and the relative humidity above the ammonium nitrate deliquescence point in the upper PBL is ~~predicted~~ thought to determine this peculiar profile. Nitrate is the major component of the upper PBL aerosol layer, contributing up to 30% of the aerosol mass.

- How much of the aerosol burden measured at the ground is due to localized processes and how much is conversely due to processes occurring in the upper PBL and to the subsequent mixing in the lowermost levels? In other words, how important is the interplay between surface and upper layers in shaping the aerosol mass we measure near the ground?

  For PM2.5 mass, our calculations indicate that in the upper PBL secondary aerosol are formed and then mixed in the PBL by turbulence. The importance of the secondary fraction increases with the aging of air masses, as shown by the progression of days from the “chemical restart”. A week after the perturbation, secondary PM2.5 produced in the upper PBL may contribute up to 7-8 μg/m^3 (or 25%) to ground level hourly concentrations. Sulfate and SOA production is equally shared by bottom and upper PBL, while nitrate is mostly produced in the upper PBL, the bottom PBL acting as a sink during the afternoon.

- How much may the residual layer above the PBL ~~may~~ contribute to the aerosol budget at ground level the next day?

~~We~~ It is calculated that the chemical production above the PBL significantly impacts aerosol levels near the ground, sometimes overtaking the contribution from the production term in the PBL (especially for sulfate and SOA). We estimate that the residual layer above the PBL, which is formed by both primary and secondary species, may occasionally contribute up to 10-12 μg/m^3 (or 40%) to ground level PM2.5 hourly concentrations during the following morning.

The peculiar features of the vertical profile of aerosol nitrate have already emerged in past studies. Neuman et al. (2003) reported aircraft observations of increasing nitrate profiles with height, ~~and~~ attributing~~ed~~ th~~os~~em to the favourable lower temperature in the upper layers, ~~with respect~~compared to bottom PBL, ~~for~~ due to the conversion of nitric acid to aerosol nitrate. We confirm their conclusion, and add that a key role in shaping the aerosol nitrate production profile is played by the relative humidity. In particular, nitrate production may be enhanced when RH is above the ammonium nitrate deliquescence point.

This study has put ~~the~~ emphasis on some less obvious and recognized aspects of the aerosol vertical profile budget. Since ~~I~~it is based on the analysis of a short period of high pressure conditions in summer over the area of Milan, ~~thus~~ further analyses are recommended for winter periods and different meteorological and geographical conditions. Moreover, it clearly ~~reminds~~ underlines the fact that the interplay between chemical and dynamical processes must be considered when interpreting atmospheric chemistry observations near the ground, and that more observational constraints (e.g. profiles of the aerosol composition in and above the PBL) would certainly be helpful ~~for~~ to achieve a better simulation of those processes.

Appendix A: definition of statistical indices used in model to observations comparison

Let Obs_i and Mod_i be the observed and modeled values at time *i*, and *N* the number of observations.

- The Pearson's Correlation (*r*):

$$r = \frac{1}{N}\sum_{i=1}^{N} Z_i(Mod) \bullet Z_i(Obs)$$
$$Z(X) = \frac{X - \langle X \rangle}{\sigma_X}$$

where *X* is a generic vector, *Z(X)* is its standard score, and σ_X is the standard deviation.

- Bias:

$$Bias = \frac{1}{N}\sum_{i=1}^{N} Mod_i - Obs_i$$

- Normalized Mean Bias (NMB):

$$NMB = \frac{1}{N}\sum_{i=1}^{N}\frac{Mod_i - Obs_i}{Obs_i} \times 100$$

- Root Mean Square Error (RMSE):

$$RMSE = \sqrt{\frac{1}{N}\sum_{i=1}^{N}(Mod_i - Obs_i)^2}$$

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Table 1. Main physical and chemical parameterizations used in WRF/Chem simulations.

Process	Scheme
Short-wave radiation	RRTM
Long-wave radiation	RRTM
Surface Layer	Monin-Obukov
Boundary Layer	MYNN
Land surface model	Noah LSM
Cumulus convection	Grell scheme G3
Cloud microphysics	Morrison
Gas-phase mechanism	RACM-ESRL
Aerosol mechanism	MADE/SOA-VBS
Photolysis	Fast-J
Cloud chemistry and wet deposition	On
Biogenic emissions	MEGAN
Direct aerosol effect	On
Indirect aerosol effects	Off

Table 2. Description of sensitivity tests with WRF/Chem model.

Label	Description
CTRL	Reference run, see Table 1~~Table 1~~.
AERO	Aerosol chemical processes switched off
LPBL	Gas and aerosol chemical processes switched off in the Lower half of the PBL
UPBL	Gas and aerosol chemical processes switched off in the Upper half of the PBL
APBL	Gas and aerosol chemical processes switched off Above the PBL

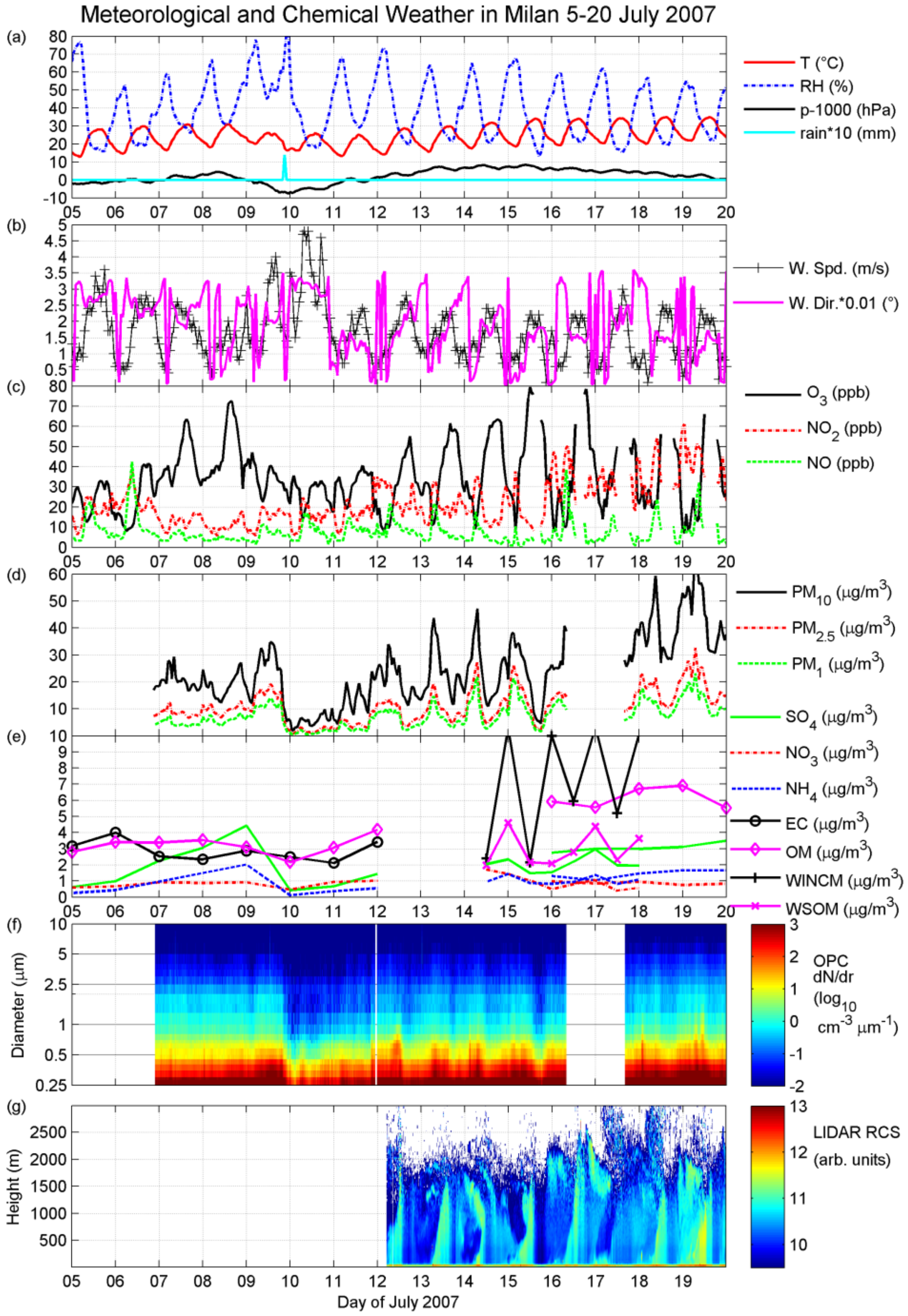
Meteorological and Chemical Weather in Milan 5-20 July 2007
(a)
T (°C)
RH (%)
p-1000 (hPa)
rain*10 (mm)
(b)
W. Spd. (m/s)
W. Dir.*0.01 (°)
(c)
O$_3$ (ppb)
NO$_2$ (ppb)
NO (ppb)
(d)
PM$_{10}$ (μg/m^3)
PM$_{2.5}$ (μg/m^3)
PM$_1$ (μg/m^3)
(e)
SO$_4$ (μg/m^3)
NO$_3$ (μg/m^3)
NH$_4$ (μg/m^3)
EC (μg/m^3)
OM (μg/m^3)
WINCM (μg/m^3)
WSOM (μg/m^3)
(f)
Diameter (μm)
OPC dN/dr (log$_{10}$ cm^{-3} μm^{-1})
(g)
Height (m)
LIDAR RCS (arb. units)
05 06 07 08 09 10 11 12 13 14 15 16 17 18 19 20
Day of July 2007

Figure 1. Ground-based observations in Milan during July 5-20, ~~July~~ 2007. Panel (a) shows hourly measurements of temperature, relative humidity, pressure and precipitation. Pressure is subtracted by 1000 and precipitation is multiplied by 10 in order to fit the same y-axis. (b) Hourly wind speed and wind direction (0° from the North, 90° from the East), the latter divided by 100 to fit the same y-axis. (c) Hourly ozone, nitrogen dioxide and nitrogen oxide. (d) Particulate matter mass. Hourly observations of PM10, PM2.5, and PM1. (d) Particulate matter composition. Daily data of sulfate, nitrate, ammonium, elemental carbon and organic matter collected during QUITSAT campaign. Night-time (21 to 08 local solar time) and day-time (08 to 21 LST) samples of sulfate, nitrate, ammonium, water-insoluble carbonaceous matter (WINCM) and water-soluble organic matter (WSOM) collected during AeroClouds campaign (July 14-17 ~~July~~). (e) Particulate matter number size distribution. Optical particle counter (OPC) hourly average measurements, y-axis denotes the size bin. (f) Particulate matter vertical profile. LIDAR Range Corrected Signal, y-axis denotes the height above ground level.

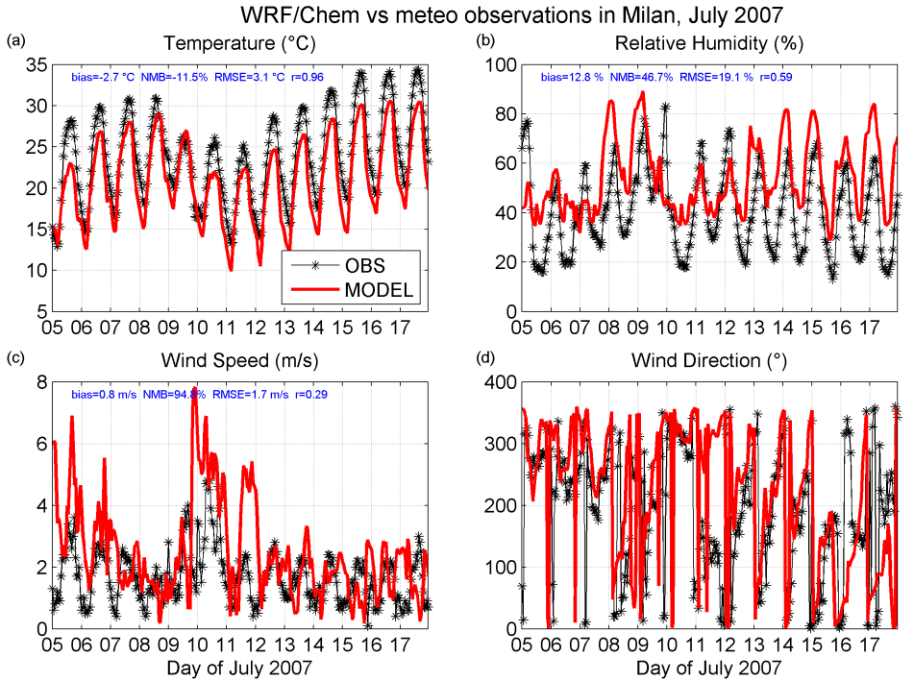

Figure 2. Comparison of observed and simulated hourly meteorological variables at ground level in Milan on July 5-17~~, July~~ 2007. Simulations are carried out with WRF/Chem model and results are shown for the nested domain over Northern Italy at 10 km horizontal resolution. Statistical indices shown inset are defined in Appendix A~~B~~.

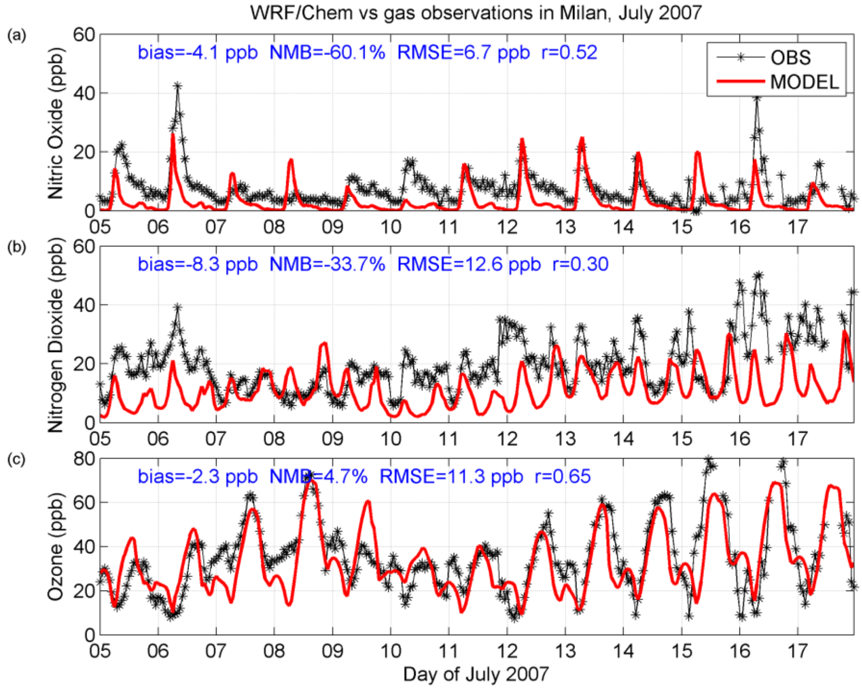

Figure 3. Same as Figure 2~~Figure 2~~, but for hourly gas-phase variables at ground level in Milan on July 5-17, ~~July~~ 2007. Shown inset are statistical indices defined in Appendix A.

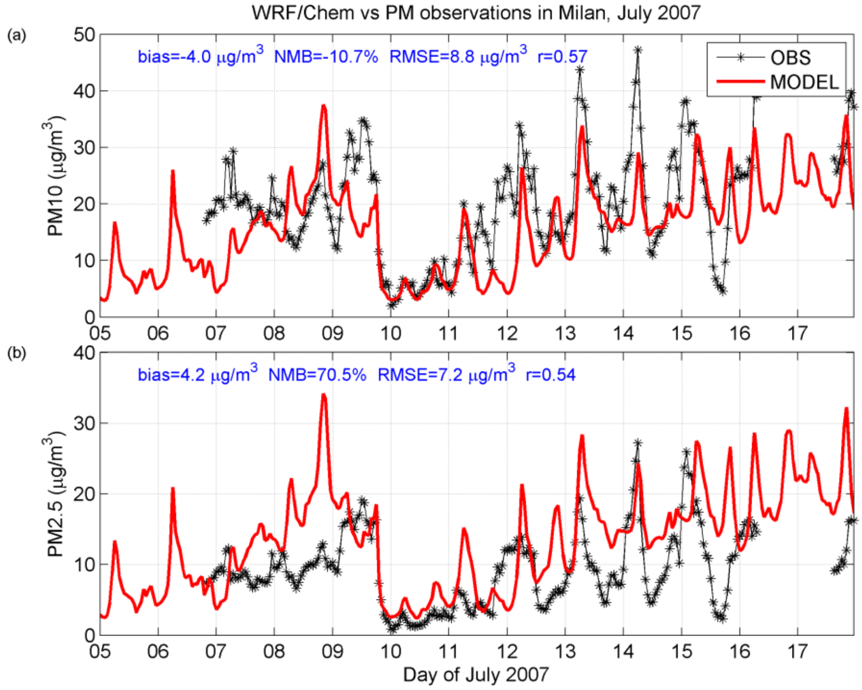

Figure 4. Same as Figure 2~~Figure 2~~, but for hourly particulate matter at ground level in Milan on July 5-17, ~~July~~ 2007. Shown inset are statistical indices defined in Appendix A.

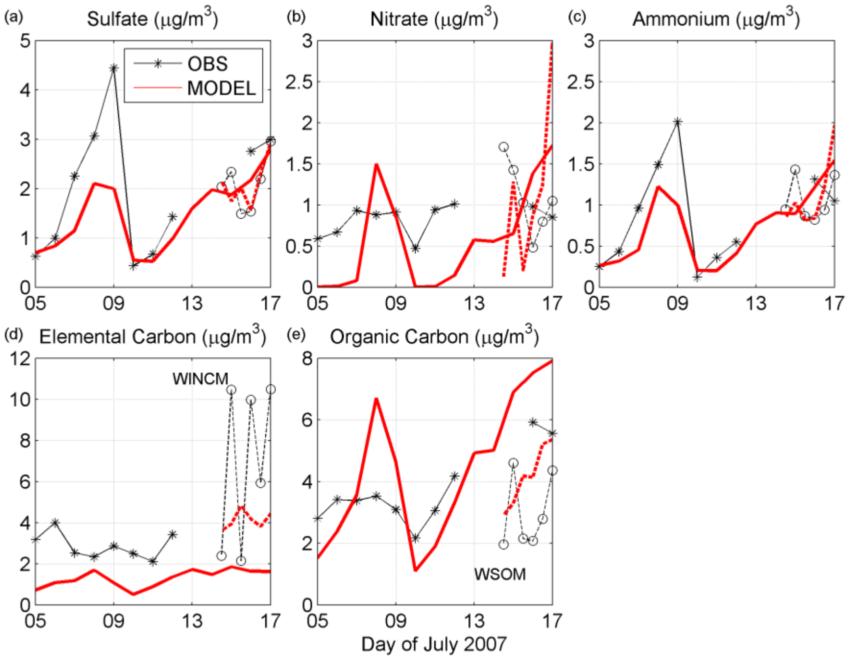

Figure 5. Same as Figure 2~~Figure 2~~, for daily and bi-daily particulate matter composition at ground level in Milan on July 5-17, ~~July~~ 2007. Bi-daily observations (dashed lines) are available only from July 14 to 17~~July~~. In panel (d), WINCM is the Water Insoluble Carbon Mass (EC + mostly primary OC), in panel (e) WSOM is Water Soluble Organic Mass (mostly secondary organic aerosol, Carbone et al., 2010).

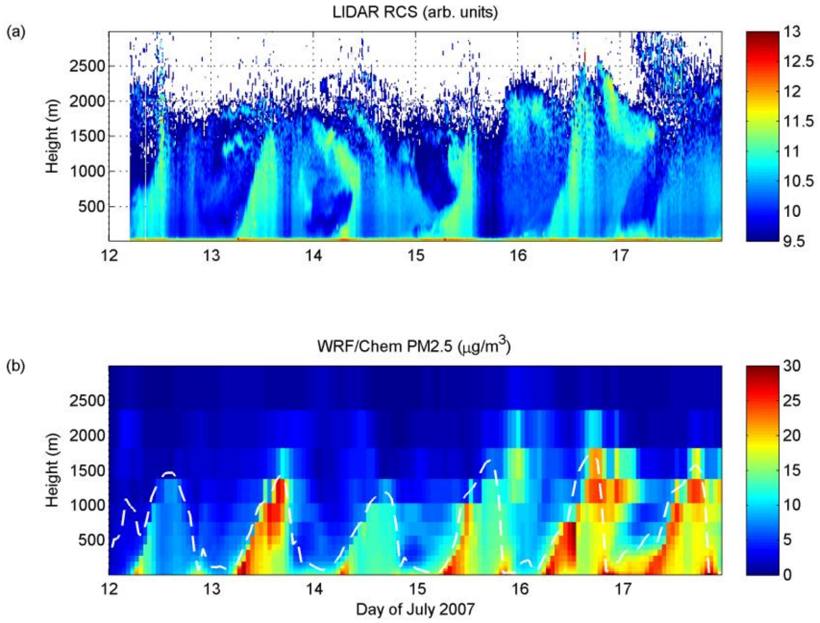

Figure 6. Qualitative comparison of (a) LIDAR Range Corrected Signal and (b) simulated PM2.5 vertical profile over Milan on July 12-17~~, July~~ 2007.

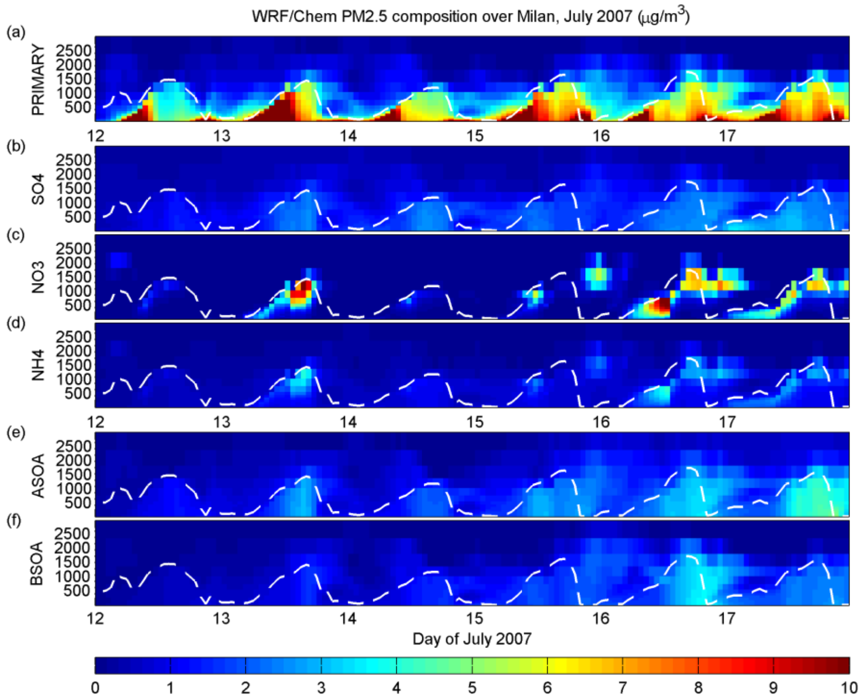

Figure 7. Simulated composition of PM2.5 profile shown in Figure 6~~Figure 6~~. ASOA and BSOA in panels (e) and (f) are anthropogenic and biogenic secondary organic aerosol, respectively.

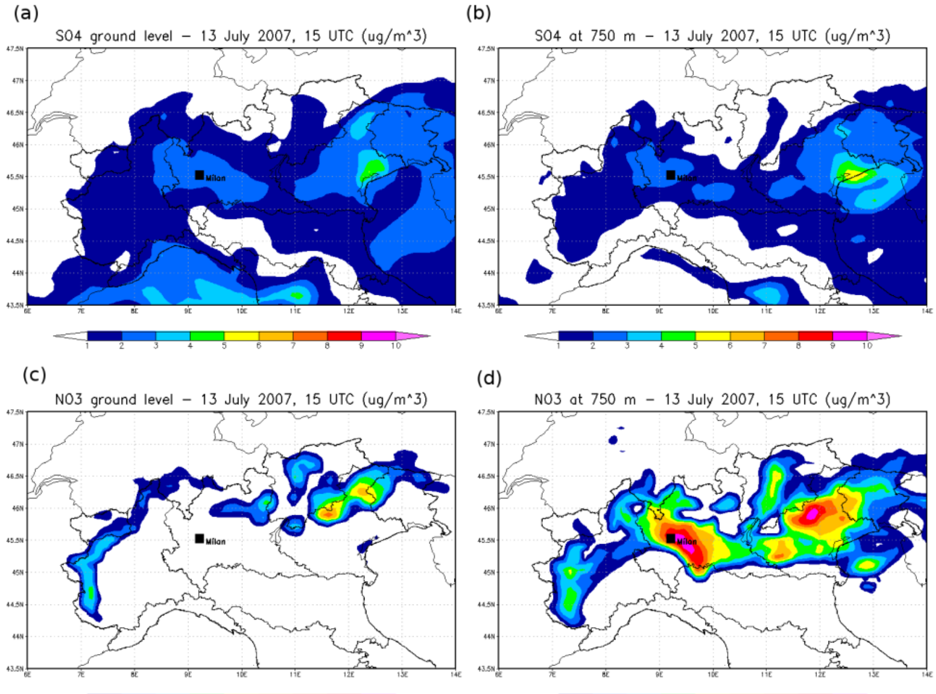

Figure 8. Maps of the concentration of PM2.5 sulfate (a-b) and nitrate (c-d) components simulated at 16 LST of July 13, ~~July~~ 2007 over Po Valley. Panels (a-c) are at ground level, panels (b-d) at 750 m height.

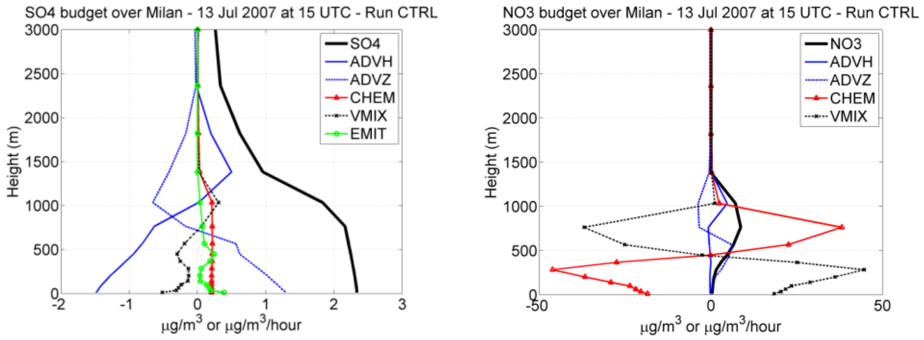

Figure 9. Simulated vertical profile of concentration (μg/m^3) and continuity equation terms (μg/m^3/h) for particulate sulfate (left) and nitrate (right) at 16 LST of July 13~~,~~ ~~July~~ 2007 over Milan. Budget terms are: horizontal advection (ADVH), vertical advection (ADVZ), chemistry (CHEM), turbulent mixing and dry deposition (VMIX), emission (EMIT).

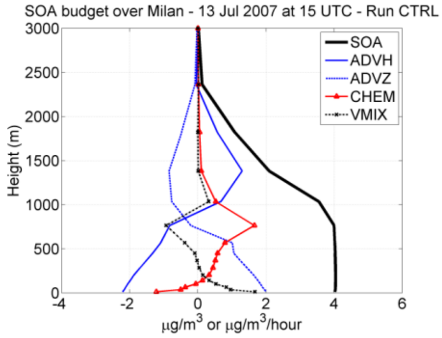

Figure 10. Same as Figure 9~~Figure 9~~, but for Secondary Organic Aerosol (SOA).

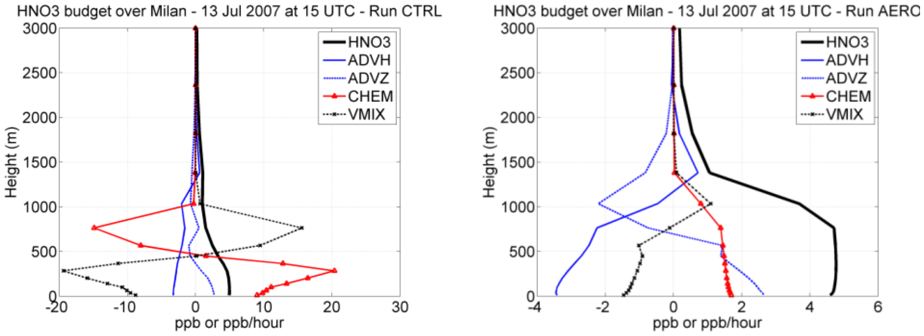

Figure 11. Same as Figure 9~~Figure 9~~, but for nitric acid (HNO$_3$) and units in ppb. On the left the reference simulation (CTRL), on the right a sensitivity simulation with aerosol chemistry switched off (AERO). Please notice the different abscissa range.

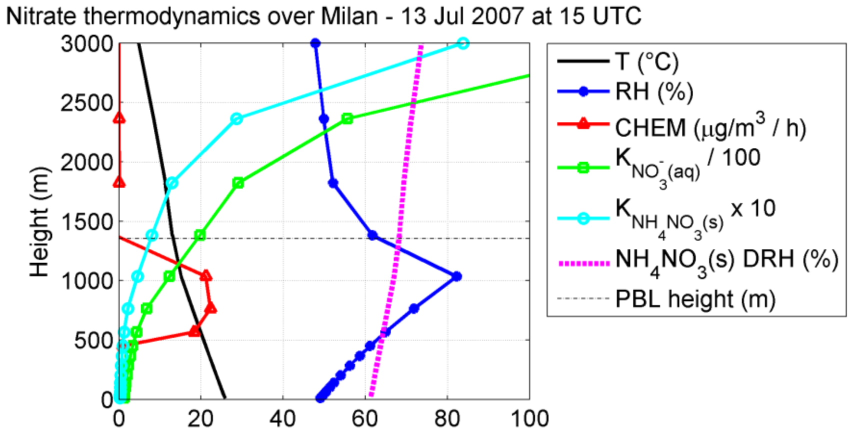

Figure 12. Simulated vertical profile of relative humidity (blue) and particulate nitrate net chemical production term (red, triangles) at 16 LST of July 13, ~~July~~ 2007 over Milan. Also shown, vertical profiles of equilibrium constants of aqueous phase nitrate (green, squares) and solid ammonium nitrate (cyan, circles), and ammonium nitrate deliquescence relative humidity (magenta, dashed). The height of PBL is denoted by the horizontal black dashed line. Please note that equilibrium constants are scaled by the constant factors shown inset to fit on the same abscissa range.

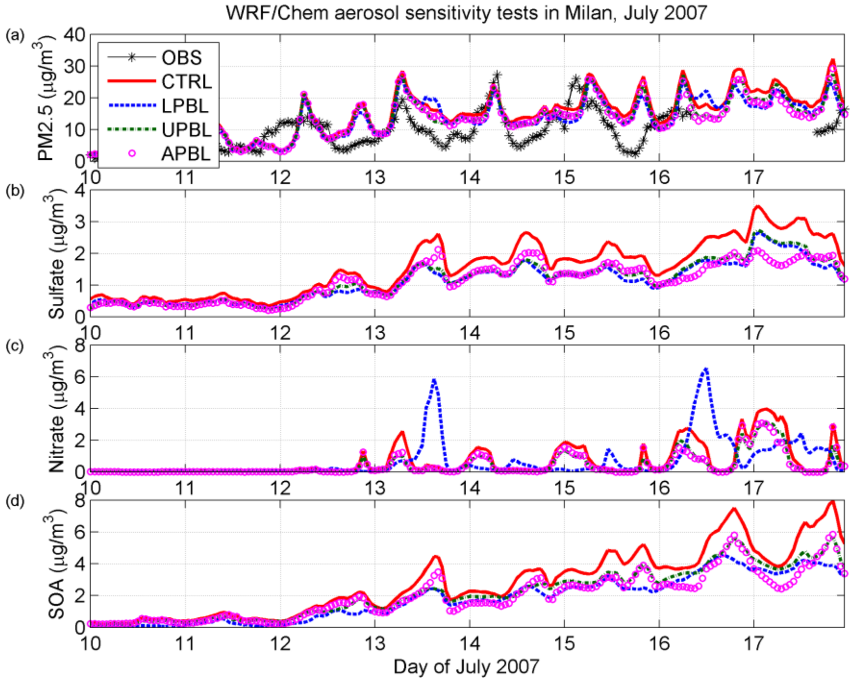

Figure 13. Sensitivity tests on chemical production in different vertical layers (see Table 2~~Table 2~~ for explanation of labels), at ground level over Milan on July 10-17, ~~July~~ 2007. Hourly observations (black line-star) are only available for PM2.5 (top panel).

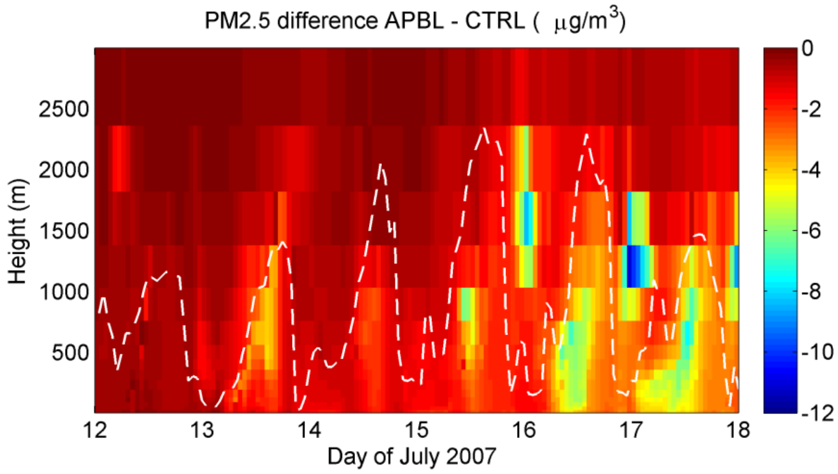

Figure 14. Difference of the simulated $PM_{2.5}$ profile over Milan between APBL and CTRL runs (see Table 2~~Table 2~~). Useful to estimate the impact of aerosol residual layer on ground concentrations, in combination with Figure 6~~Figure 6~~b.