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*Supplement of*

## **Evidence for an unidentified ground-level source of formaldehyde in the Po Valley with potential implications for ozone production**

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## 1 **Additional consideration of deposition**

2 In the base case scenario, the model is constrained to all measurements with the exception of  
3 HCHO. Given the extensive constraints, deposition, emission, and advection are not treated  
4 explicitly. While such simplification is commonly performed in box models, the coupling  
5 between the surface layer and the overlying atmosphere can drive the behavior of many trace  
6 gases while chemistry plays a smaller role. The assumption is that with small enough time steps,  
7 constraining calculations to observed mixing ratios is functionally equivalent to having an inflow  
8 or outflow rate that accounts for net flux from the surface or emission sources. However, the  
9 impact on species that are not constrained could be a function of what the model sources and  
10 sinks of a given other species. For example, if an unconstrained HCHO precursor has a high  
11 deposition rate which is neglected, the model may overestimate loss to oxidation and create high  
12 concentrations of unconstrained oxidation products. This could lead to HCHO production, while  
13 deposition would not lead to HCHO formation. Furthermore, reaction with OH would be over-  
14 expressed, leading to erroneously high calculated OH reactivity.

15 To investigate the effect of neglecting model deposition, a model scenario which includes dry  
16 deposition of O<sub>3</sub>, HNO<sub>3</sub>, NO<sub>x</sub>, H<sub>2</sub>O<sub>2</sub>, CH<sub>3</sub>OOH, peroxyacetyl nitrate, CH<sub>3</sub>CHO, HCHO,  
17 CH<sub>3</sub>CO<sub>3</sub>H, HCOOH, isoprene, methyl vinyl ketone, methacrolein, methylacetate, acetic acid,  
18 and methanol was constructed. Deposition velocities were calculated using the SCM without  
19 imposing any constraint on the soil moisture level. The deposition rates are generally based on  
20 species solubility and reactivity according to Wesely (1989), except for species which have been  
21 modified according to Ganzeveld and Lelieveld (1995) (O<sub>3</sub>, HNO<sub>3</sub>, NO<sub>x</sub>), and peroxides  
22 (Ganzeveld et al., 2006). The base case model domain was extended down to the ground (rather  
23 than stopping at 100 m), and a first order deposition loss rate was applied to the surface layer. As  
24 observations were not available at these altitudes, the observed averages between 50-150 m are  
25 applied to the surface layer. All species except HCHO are constrained to measurements.

26 The results of this model run are shown in Fig. S1. Including deposition has negligible effects on  
27 calculated OH reactivity; however, calculated HCHO decreases significantly in the boundary  
28 layer, creating an inverted profile. To compensate for this potential depositional sink, an  
29 emission rate eight times greater than that used in the main text would be required. This suggests  
30 either net deposition and emission occur on separate surfaces, or the existence of a HCHO

1 compensation point at which the surface as a whole acts as a net source rather than a sink.  
2 Because calculated OH reactivity is largely unaffected by including deposition and because our  
3 base model scenario represents an upward bound on the calculated HCHO mixing ratio, the  
4 conclusions stated in the main text are unaffected by the deposition term. Our study represents a  
5 lower limit on the non-photochemical HCHO source, and therefore a lower limit on the  
6 sensitivity of ozone production to agricultural emissions of HCHO.

7

### 8 **Model sensitivity to turbulent mixing**

9 The eddy-diffusivities for tracer transport are taken from the SCM which simulates online the  
10 turbulent fluxes of momentum, heat and moisture transport according to the local-closure  
11 turbulent transport scheme of the climate model ECHAM4 including modifications for non-local  
12 contributions to the turbulent kinetic energy budget (Brinkop and Roeckner, 1995) and updated  
13 mixing length formulation (Lenderink and Holtslag, 2004). The simulated meteorology  
14 resembles the actual observed meteorological conditions quite well. The SCM simulations are  
15 constrained with the ECMWF (European Centre for Medium range Weather forecast) re-analysis  
16 data for the grid point resembling the location of the SPC site covering (e.g. Ganzeveld et al.,  
17 2006). The eddy diffusion constants as are shown in context of the HCHO lifetime with respect  
18 to photolysis/reaction with OH in Fig. S2.

19 As seen in the case where direct HCHO emission is included in the model, the eddy diffusion  
20 constants may not accurately capture the speed of growth of the mixed layer. As a sensitivity  
21 analysis, the coefficients are increased uniformly by a factor of 10 in both the base case scenario  
22 and the scenario where emissions are included (Fig. S3). While faster mixing better captures the  
23 early growth of the mixed layer in the emissions included case, the effect on the base case is  
24 minimal. Therefore, we conclude that the uncertainties in eddy diffusion have minimal impact on  
25 the conclusion that HCHO is underestimated at the lowest altitudes while OH reactivity is well  
26 captured.

27 Finally, as a boundary condition, no mixing occurs between the uppermost model layer and the  
28 overlying atmosphere (i.e. all boundaries are reflective). Any mixing would only affect non-  
29 constrained species. It is probable that the concentrations of these species (e.g. RO<sub>2</sub>, HCHO) is  
30 lower in the overlying layer and additional mixing would further dilute HCHO. Because our base

1 case model represents an upper estimate of model HCHO, i.e., a conservative estimate of missing  
2 HCHO, including mixing with the upper atmosphere would not affect our conclusions.  
3 Additionally, the reflective boundary conditions restrict emitted species to the total model  
4 volume. This could artificially inflate the concentrations of emitted species if the actual  
5 atmospheric mixing volume is greater than the model volume. According to the eddy diffusion  
6 coefficients, mixing significantly decreases at highest model altitude. Therefore, it is unlikely that  
7 the reflective boundary significantly inflates the increase in HCHO mixing ratios caused by  
8 emissions in the early morning.

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10 **References**

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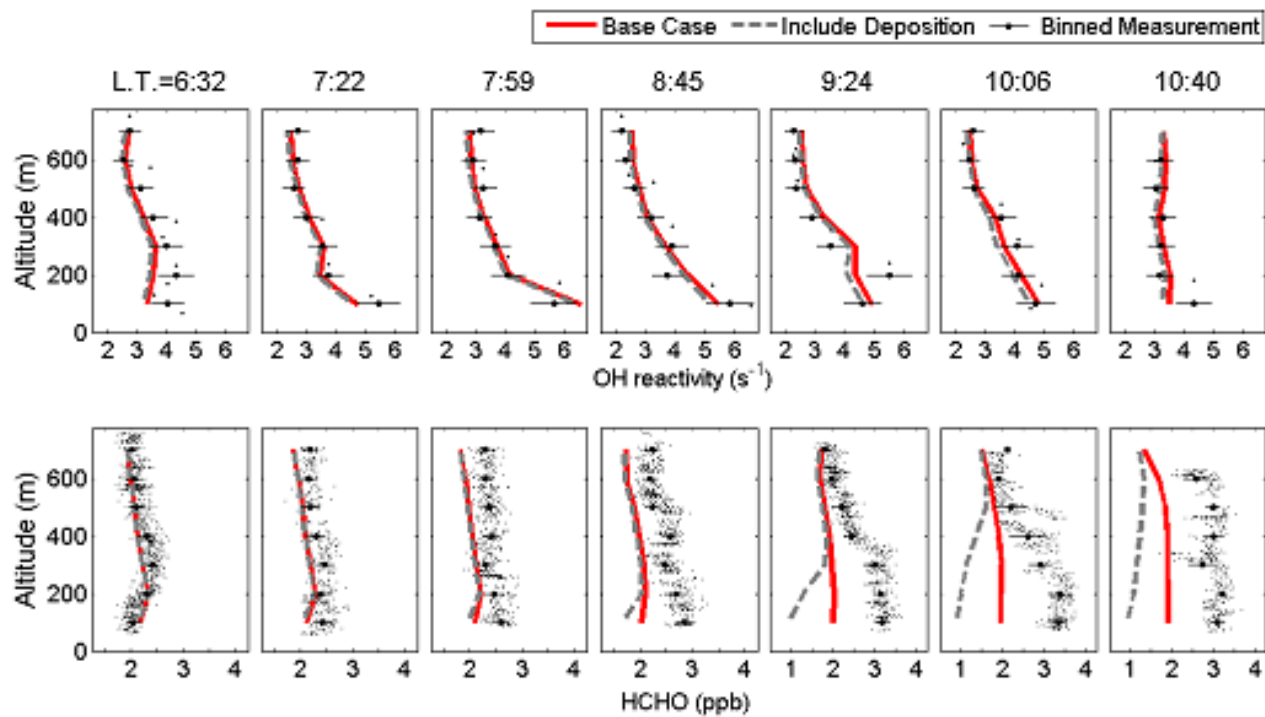
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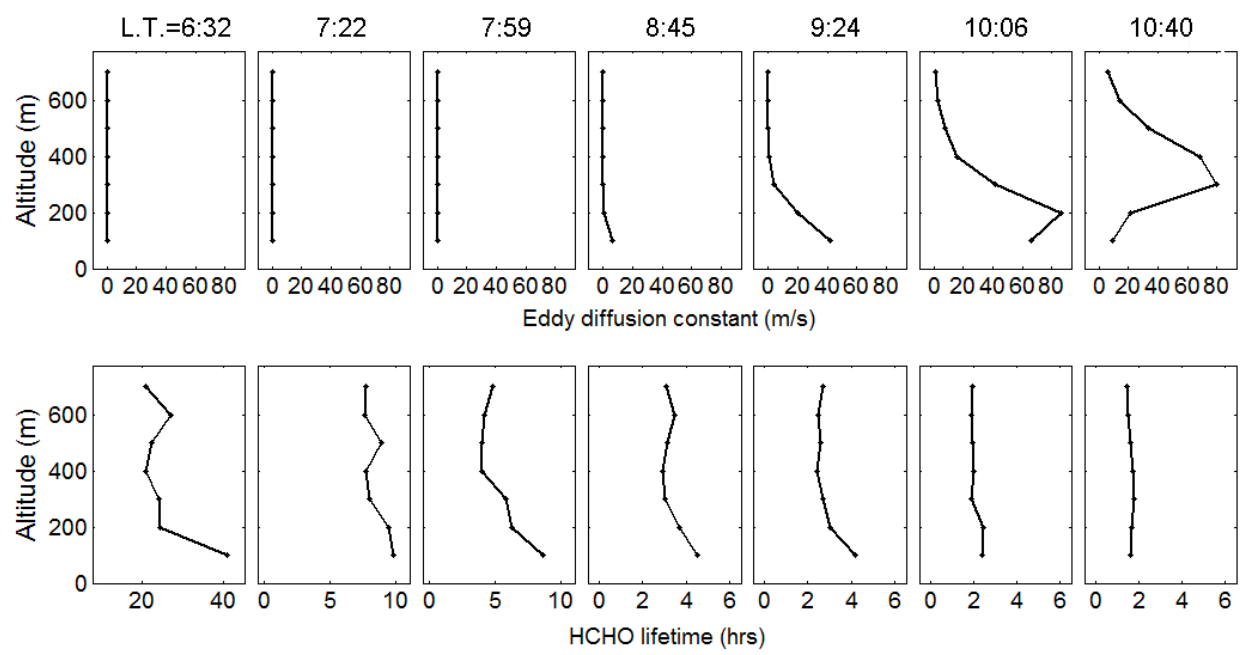
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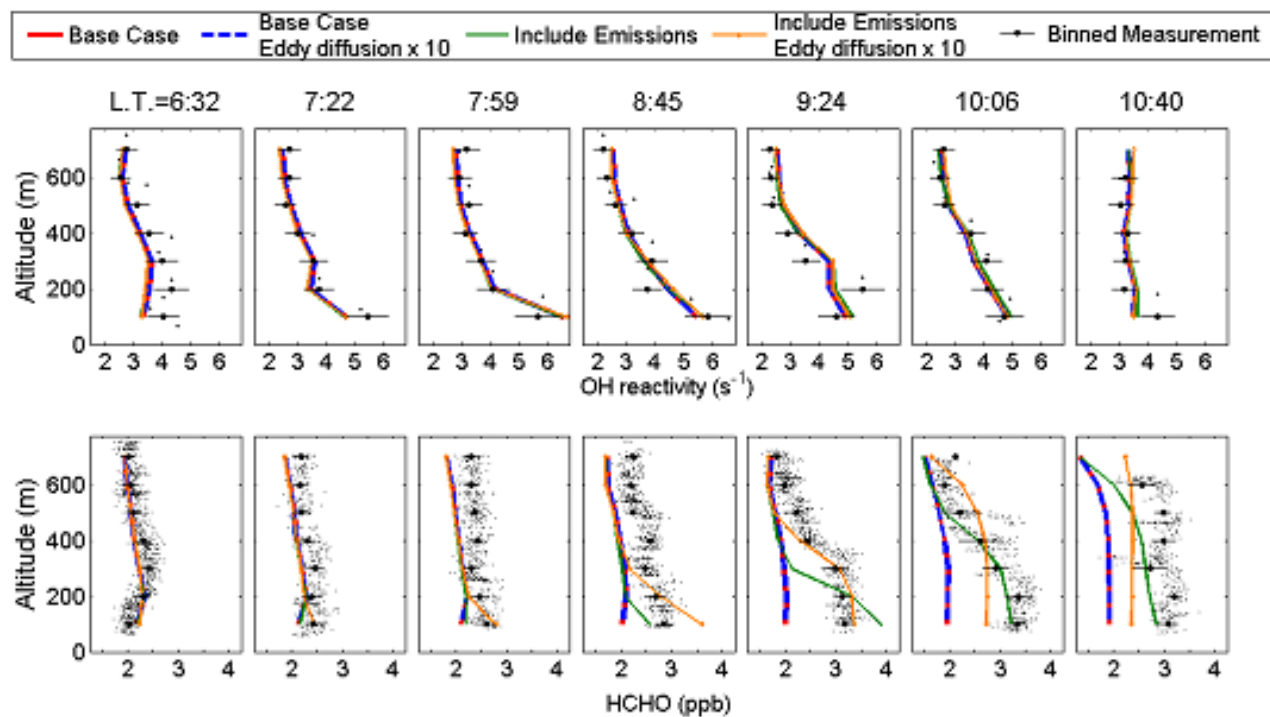
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 2 Figure S1. Measured and calculated OH reactivity and HCHO vertical profiles for every other  
 3 Zeppelin ascent. Error bars on OH reactivity represent the measurement precision. Error bars on  
 4 HCHO represent the standard deviation of the measurements in the given altitude bin.

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Figure S2. Calculated eddy diffusion constants and HCHO lifetime with respect to photolysis/reaction with OH.



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 2 Figure S3. Sensitivity to eddy diffusion constants for calculated OH reactivity and HCHO  
 3 vertical profiles in the base case scenario and including direct HCHO emissions. Increased eddy  
 4 diffusivity in most cases does not change the average model-measurement discrepancy.