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2 **RECEPTOR MODELLING OF FINE PARTICLES IN**
3 **SOUTHERN ENGLAND USING CMB INCLUDING**
4 **COMPARISON WITH AMS-PMF FACTORS**

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31 **ABSTRACT**

32 PM_{2.5} was collected during a winter campaign at two southern England sites, urban background
33 North Kensington (NK) and rural Harwell (HAR), in January-February 2012. Multiple organic and
34 inorganic source tracers were analysed and used in a Chemical Mass Balance (CMB) model, which
35 apportioned seven separate primary sources, that explained on average 53% (NK) and 56% (HAR)
36 of the organic carbon (OC), including traffic, woodsmoke, food cooking, coal combustion,
37 vegetative detritus, natural gas and dust/soil. With the addition of source tracers for secondary
38 biogenic aerosol at the NK site, 79% of organic carbon was accounted for. Secondary biogenic
39 sources were represented by oxidation products of α -pinene and isoprene, but only the former made
40 a substantial contribution to OC. Particle source contribution estimates for PM_{2.5} mass were
41 obtained by the conversion of the OC estimates and combining with inorganic components
42 ammonium nitrate, ammonium sulphate and sea salt. Good mass closure was achieved with 81%
43 (92% with the addition of the secondary biogenic source) and 83% of the PM_{2.5} mass explained at
44 NK and HAR respectively, with the remainder being secondary organic matter. While the most
45 important sources of OC are vehicle exhaust (21% and 16%) and woodsmoke (15% and 28%) at
46 NK and HAR respectively, food cooking emissions are also significant, particularly at the urban NK
47 site (11% of OC), in addition to the secondary biogenic source, only measured at NK, which
48 represented about 26%. In comparison, the major source components for PM_{2.5} at NK and HAR are
49 inorganic ammonium salts (51% and 56%), vehicle exhaust emissions (8% and 6%), secondary
50 biogenic (10% measured at NK only), woodsmoke (4% and 7%) and sea salt (7% and 8%), whereas
51 food cooking (4% and 1%) showed relatively smaller contributions to PM_{2.5}. Results from the
52 CMB model were compared with source contribution estimates derived from the AMS-PMF
53 method. The overall mass of organic matter accounted for is rather similar for the two methods.
54 However, appreciably different concentrations were calculated for the individual primary organic
55 matter contributions, although for most source categories the CMB and AMS-PMF results were
56 highly correlated ($r^2 = 0.69-0.91$). In comparison with the CMB model, the AMS appears to over-

57 estimate the biomass burning/coal and food cooking sources by a factor of around 1.5 to 2 while
58 estimates of the traffic source are rather similar for each model. The largest divergence is in the
59 primary/secondary organic matter split, with the AMS estimating an appreciably smaller secondary
60 component. Possible reasons for these discrepancies are discussed, but despite these substantial
61 divergences, the strong correlation of the two methods gives some confidence in their application.

62

63 **Keywords:** PM_{2.5}; aerosol; source apportionment; CMB model; AMS; PMF model; urban site; rural
64 site.

65 **1. INTRODUCTION**

66 Reduction of the airborne concentrations of particulate matter remains a high priority. The main
67 drivers are European Union (EU) Limit Values and the health benefits to be gained from lower
68 concentrations. In particular, the exposure reduction targets of the EU for fine particle PM_{2.5} (a
69 15% reduction to be achieved by the UK by 2020 from 2009-2011 average concentrations) provide
70 tough challenges for abatement measures. Cost-effective abatement depends upon a clear
71 knowledge of the contributions of individual sources and source sectors to airborne concentrations.
72 Currently in the UK, components of PM_{2.5} for which data are particularly weak are woodsmoke (or
73 solid fuel burning smoke more generally) (Harrison et al., 2012), cooking aerosol (especially in city
74 centres with a high concentration of restaurants) (Allan et al., 2010), abrasion particles from road
75 vehicles (Thorpe and Harrison, 2008; Pant and Harrison, 2013) and secondary organic fractions,
76 which need additional research in order to fully understand their source contributions.

77

78 It has been established that significant amounts of the fine particulate matter (PM_{2.5}) are comprised
79 of organic matter at sites within and outside Europe, representing around 25-31% in the UK West
80 Midlands (Harrison et al., 2004), 21-33% in Ireland (Yin et al., 2005), 27-47% in Australia (Chan et
81 al., 1997), 38-47% in France (Bressi et al., 2013) and 50% in Michigan, USA (Pancras et al., 2013).
82 Organic matter is derived from both primary sources from which it is directly emitted to the
83 atmosphere, and secondary production through oxidation of volatile organic compounds (VOCs) in
84 the atmosphere. While numerous studies have been carried out upon the primary organic
85 compounds in terms of their speciation and sources (e.g. Schauer et al., 1996; Stone et al., 2008;
86 Yin et al., 2010; El Haddad et al., 2011a; Hasheminassab et al., 2013), the contribution of secondary
87 organic aerosol to the total organic carbon and particle mass remains less clear due to its complex
88 origins, composition and formation mechanisms in the atmosphere (Turpin et al., 2000; Hallquist et
89 al., 2009). A number of studies have been carried out over mainland Europe on secondary organic
90 aerosols formed through oxidation of biogenic volatile organic compounds (BVOCs) (Böge et al.,

91 2006; Plewka et al., 2006; Wagener et al., 2012a,b), since their global emissions have been
92 estimated to be 10 times higher than those of anthropogenic VOCs (Guenther et al., 1995). The
93 major molecular markers for biogenic secondary organic aerosol constituents identified/used
94 include a) pinic and pinonic acid (the major oxidation products of α -pinene) (Presto et al., 2005),
95 and b) 2-methyltetrols (i.e., 2-methylthreitol and 2-methylerythritol: oxidation products of isoprene)
96 (Claeys et al., 2004; Kourtchev et al., 2005; Clements and Seinfeld, 2007; Stone et al., 2009).

97

98 Receptor modelling methods have been used for quantitative source apportionment of both primary
99 and secondary particulate matter using chemically discriminated composition to provide source
100 attribution. The widely used receptor models include Principal Component Analysis with Multiple
101 Linear Regression (PCA-MLR), Positive Matrix Factorization (PMF), UNMIX and Chemical Mass
102 Balance (CMB). The molecular marker-based CMB model requires aerosol chemical composition
103 data from both the pollution sources and the receptor site, and has proved able to distinguish
104 different primary sources of carbonaceous aerosols (Schauer et al., 1996; Zheng et al., 2002; Fraser
105 et al., 2003; Stone et al., 2008; Yin et al., 2010; El Haddad et al., 2011a).

106

107 The contribution of secondary organic aerosol (SOA) has been estimated based simply upon
108 laboratory-derived ratios of secondary organic carbon (SOC) mass to individual secondary organic
109 marker compounds from the precursors isoprene, α -pinene, β -caryophyllene and toluene
110 (Kleindienst et al., 2007; Lewandowski et al., 2008; Kourtchev et al., 2009; El Haddad et al.,
111 2011b), although this method is subject to considerable uncertainties due to the simplification of
112 replacing the complex atmospheric chemical reactions responsible for SOA formation with a
113 laboratory-derived single-value mass fraction. The CMB model has also been used to apportion
114 both primary and secondary sources (e.g. Stone et al., 2009) by the addition of specific secondary
115 organic molecular markers derived from isoprene, α -pinene, β -caryophyllene and toluene, with the

116 highest ambient concentrations observed for derivative species of isoprene and α -pinene
117 (Lewandowski et al., 2008; Stone et al., 2009).

118
119 Our previous study at two sites in the West Midlands area of the UK (Yin et al., 2010) identified
120 eight primary sources that contribute about 56-85% on average to fine particulate organic carbon,
121 including vehicular emissions (diesel engines, gasoline engines, smoking engines), wood smoke,
122 vegetative detritus, natural gas combustion, coal combustion and road dust/soil. Vehicle exhaust
123 emissions from all engines contributed up to 57% of the fine OC, with a relatively smaller amount
124 up to 14% from other known sources, whilst a large amount (up to 34%) of the OC remained
125 unexplained (termed as Other-OC), and was inferred to be mostly associated with secondary
126 organic compounds.

127
128 A further study, reported here, has been carried out in southeast England at urban background and
129 rural sites in order to obtain updated and extended information. Ambient aerosol samples have been
130 analysed for multiple organic and inorganic source tracers, specifically including a number of
131 additional markers for food cooking and secondary biogenic aerosols, in addition to the primary
132 molecular markers previously analysed in the earlier study (Yin et al., 2010). This new dataset,
133 particularly the food cooking and biogenic secondary markers, has allowed further evaluation of the
134 concentration and sources of those components and the possibility for the first time to estimate, with
135 the CMB method, the contributions of the main groups of biogenic VOC to secondary organic
136 aerosol in the UK atmosphere. The CMB results have been compared with source contribution
137 estimates derived from an Aerosol Mass Spectrometer (AMS), with an emphasis on sources of food
138 cooking and secondary particles.

139
140 The comparison of CMB and AMS-PMF results is very challenging as the two techniques are very
141 different. The CMB method takes measured chemical source profiles and constructs linear

142 combinations to give a best fit to the composition of the ambient particles. It thus requires prior
143 knowledge of the contributing sources. On the other hand, the application of Positive Matrix
144 Factorization (PMF) to AMS mass spectral data is a multivariate statistical fitting method which
145 identifies covariations within the mass spectral data itself, and outputs mass spectral profiles of
146 contributing sources without any *a priori* assumptions about the contributing sources. The CMB
147 model will work best when the contributing sources are well characterised in terms of a constant
148 chemical composition of each. If any source is omitted, or has a different chemical profile to that
149 used to describe it in the model, or varies in composition, uncertainties and errors are introduced.
150 The AMS method does not depend upon known source profiles (although use of ME-2 rather than
151 PMF for data analysis allows them to be introduced), but achieving a reliable separation of factors
152 with PMF requires considerable skill, and attribution to sources relies upon subjective judgements.
153 The AMS-PMF method is also subject to other uncertainties, such as those introduced by rotational
154 ambiguities, discussed in the body of the paper. Hence both have their own strengths and
155 weaknesses, with intercomparisons being a useful means of evaluating performance with a view to
156 their optimisation.

157

158 **2. Methodology**

159 **2.1 Site Location and Aerosol Sampling**

160 Sampling was carried out during the winter ClearfLo campaign in January-February 2012 in
161 southeast England, UK at two sites, an urban background site, North Kensington, (NK) London and
162 a rural site, Harwell, (HAR), Oxfordshire. The NK site (51° 31''N, 0° 12''W) is situated in the
163 grounds of a school in a residential area, 7 km to the west of central London and is widely accepted
164 as representative of air quality across a large part of London. The air pollution climate at the NK
165 site has been previously analysed in detail by Bigi and Harrison (2010). The HAR monitoring
166 station (51° 34''N, 1° 20''W) is situated to the west of London. The nearest road is a minor road

167 located approximately 140 metres from the station and the surrounding area is generally open with
168 agricultural fields, with the nearest trees at a distance of approximately 25 metres.

169
170 There were two collocated instruments at NK and HAR sites, a dichotomous Partisol 2025 sampler
171 and a Digitel DHA-80 sampler for the purpose of both chemical and physical analyses. The Partisol
172 sampler was used to collect 24-hour fine ($PM_{2.5}$) particles onto 47 mm PTFE filters used for
173 gravimetric and metal analyses. The Digitel was used for collecting also 24-hour fine particles on
174 150 mm diameter quartz fibre filters, which were analysed for organic molecular markers, total
175 organic carbon (OC), elemental carbon (EC), anions and cations. In addition to the samplers at NK,
176 an Aerodyne High-Resolution Time-of-Flight Aerosol Mass Spectrometer was deployed. The
177 sampling record, instrument/filter media used for ambient sampling, chemical and physical
178 properties analysed and the methodologies/instrumentation used in the laboratory are summarised in
179 Table 1. Most of the detailed procedures have been outlined in previous studies (Harrison and Yin,
180 2010; Yin et al., 2010) and are briefly described here along with the new procedures which are
181 described in greater detail.

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183
184

185 **2.2 Methodologies**

186 **2.2.1 $PM_{2.5}$ mass and metals**

187 The Partisol PTFE filters collected at NK and HAR were conditioned and weighed in a controlled
188 environment room ($20\pm 2^\circ\text{C}$ and 35-45% R.H.) before and after exposure to obtain the gravimetric
189 mass of $PM_{2.5}$. After gravimetric analysis, those samples were analysed for elements Fe, Si and Al
190 using a Bruker S8 Tiger WD-XRF (X-ray Fluorescence Spectrometer) instrument, and then for
191 metals Ti, Mn, Ni, Cu, Zn and Ba by Agilent 7500ec ICP-MS, after extraction using an aqua regia
192 acid solution (Harper et al., 1983; Allen et al., 2001; Birmili et al., 2006).

193 2.2.2 Ions, OC, EC and organic markers

194 The Digital PM_{2.5} samples on quartz filters at NK and HAR were analysed for ions SO₄²⁻, NO₃⁻, Cl⁻,
195 Na⁺, K⁺, Ca²⁺, Mg²⁺ and NH₄⁺ using a Dionex ion chromatograph, OC and EC by Sunset
196 Laboratory thermal-optical OC/EC analyser and organic markers (Table S1) by GC-MS, including
197 12 n-alkanes C₂₄ – C₃₅, 9 hopanes, 10 PAHs, 2 sterols (cholesterol and levoglucosan), 6 fatty acids
198 and 4 secondary biogenic molecular markers (at NK only), i.e. oxidation products of α-pinene
199 (pinonic acid, pinic acid) and isoprene (2-methyltetrols: 2-methylthreitol and 2-methylerythritol).
200 Similar methods to those reported by Yue and Fraser (2004) and Yin et al. (2010) have been applied
201 for the sample extraction and analysis procedures, but a modified derivation/quantification method
202 from Wagener et al.(2012a) was used for the secondary biogenic markers. In brief, one quarter of
203 the Digital sample was spiked with internal standards octacosane-d₅₈, aaa-20R-cholestane-d₄,
204 dibenz(ah)anthracene-d₁₄,cholesterol-2,2,3,4,4,6-d₆, methyl-beta-D-xylopyranoside, heptadecanoic
205 acid-d₃₃and meso-erythritol and extracted with DCM and methanol (2:1) by undergoing mild
206 ultrasonic treatment at room temperature. The combined extract was reduced in volume to
207 approximately 5mL using a turbo evaporator, then filtered/dried and further concentrated down to
208 300μL. One aliquot of the extract was analysed directly using an Agilent GC-MS system for non-
209 polar compounds, n-alkanes, hopanes and PAHs, whilst the polar organics needed to be derivatised
210 before the GC-MS analysis. For fatty acids, one aliquot of the extract was evaporated to near
211 dryness and derivatised by addition of methanol and 2.0M trimethylsilyldiazomethane (TMS-DM)
212 in diethyl ether. For sterols and biogenic markers, another aliquot of the extract was concentrated
213 down to near dryness and derivatised by addition of N,O-bis(trimethylsilyl)trifluoroacetamide plus
214 1% trimethylchlorosilane (BSTFA + 1% TMCS) and pyridine at 70°C for 1h, and finally cooled in a
215 desiccator before being run on the GCMS. The analytical precision and detection limit for
216 individual compounds calculated using repeated measurement of the lowest standard are listed in
217 Table S2. Blank values higher than the DL were subtracted from the sample results.

218
219

220 2.2.3 The CMB model

221 The US EPA CMB8.2 software was used for CMB modelling, with mostly similar source profiles
222 to our earlier work, including vegetative detritus (Rogge et al., 1993a), natural gas combustion
223 (Rogge et al., 1993b), wood smoke/biomass burning (Fine et al., 2004), dust/soil (Schauer, 1998)
224 and coal combustion (Zhang et al., 2008). For traffic, the split of source profiles for gasoline, diesel
225 and smoking engines may be incorrect as it is based on old engine source profiles from Los Angeles
226 in 2001 (Lough et al., 2007) and tends to overestimate the emissions from the UK traffic fleet (Yin
227 et al., 2010; Pant et al., 2014). Therefore they were not applied here, and instead a single traffic
228 source profile was generated from a twin site measurement from London (roadside site minus
229 background site) (Pant et al., 2014). Additional source profiles used were food cooking (Zhao et al.,
230 2007b) and secondary biogenic emissions, which was generated from ambient measurements in
231 Germany (Wagener et al., 2012a,b). Selected fitting species used in the model are elemental carbon,
232 silicon, aluminium, levoglucosan, C₂₉–C₃₅*n*-alkanes, 17a(H)-22,29,30-trisnorhopane, 17a(H)-
233 21b(H)-hopane, 17b(H)-21a(H)-30-norhopane, benzo(b)fluoranthene, benzo(k)fluoranthene,
234 benzo(e)pyrene, indeno(1,2,3-cd)pyrene, benzo(ghi)perylene, picene, *n*-hexadecanoic acid, *n*-
235 octadecanoic acid, 9-octadecenoic acid, 9,12-octadecadienoic acid, pinonic acid, pinic acid, 2-
236 methylthreitol and 2-methylerythritol. Detailed model performance measures can be found in Yin
237 et al. (2010).

238

239 2.2.4 AMS data analyses

240 The chemical composition of non-refractory PM₁ species was measured by an Aerodyne High-
241 Resolution Time-of-Flight Aerosol Mass Spectrometer (HR-ToF-AMS, hereafter AMS), which
242 operated in the standard configuration at NK, taking both MS and PToF data. A detailed description
243 of the instrument can be found elsewhere (DeCarlo et al., 2006; Canagaratna et al., 2007). As the
244 AMS sampled in an alternating sequence with other black carbon and aerosol volatility

245 measurements, 5-minute averaged ambient samples in ‘V mode’ were only obtained every 30
246 minutes. Full details of the measurements are given in Young et al. (2014).
247
248 Data were analysed within Igor Pro (Wave metrics) using the standard analysis software packages,
249 SQUIRREL v1.52J and PIKA v1.11J. A time and composition dependent collection efficiency (CE)
250 was applied to the data based on the algorithm by Middlebrook et al. (2012) and was validated by
251 comparing the volume concentration with that of the DMPS measurements. The AMS was
252 calibrated using 350nm monodisperse ammonium nitrate particles.
253
254 Positive matrix factorization (PMF) (Paatero, 1997) was performed on the organic data matrix from
255 the ‘V-mode’ data, permitting analysis of peaks according to elemental composition (Sun et al.,
256 2011). While the ‘W-mode’ data could in theory provide a more detailed analysis, too low a fraction
257 of peaks were consistently fit by PIKA (due to the lower signal-to-noise) to permit a meaningful
258 PMF analysis. A front-end for using the related ME-2 algorithm (Lanz et al., 2008; Paatero, 1999)
259 is currently available (Canonaco et al., 2013), which in some circumstances can produce more
260 accurate data. However, the benefits of this approach are most significant when applied to unit mass
261 resolution (UMR) data (from the Q-AMS, C-TOF-AMS and ACSM), where key peaks (such as
262 $C_3H_7^+$ and $C_2OH_3^+$) cannot be explicitly separated and therefore contribute to rotational ambiguity
263 under normal PMF analysis. As this is not an issue with the HR-TOF-AMS data presented here, it
264 was decided that it would be most appropriate to use PMF, so the results would not be influenced
265 by a priori assumptions regarding the aerosol’s behaviour.
266
267 The data were pre-processed in the recommended method of practice as described by Ulbrich et al.
268 (2009). Isotopes were not included in the organic matrix and nitrogen-containing peaks were not
269 deemed to have been successfully retrieved using PIKA. Five factors were identified: oxygenated
270 OA (OOA), cooking OA (COA), hydrocarbon-like OA (HOA) and two solid fuel OA (SFOA 1 and

271 SFOA 2), which had the appearance of ‘split’ factors. While the 4-factor solution (which contained
272 only one SFOA factor) seemed to be valid, the 5-factor solution gave improvements to diagnostics
273 (e.g. Q) and correlations with ancillary data (e.g. NO_x, BC and CO), so it was deemed that the 5-
274 factor solution with the split SFOA factors was the most appropriate. The 6-factor solution was
275 discarded due to its significant dependency on initialisation seed (unlike the solutions with 5 or
276 fewer factors) and as well as the production of a factor that did not appear physically meaningful.
277 Further details are presented in Young et al. (2014).

278

279 **2.2.5 Rotational ambiguity**

280 Ambiguity due to rotational freedom within the solutions is a problem inherent to PMF, in common
281 with many multivariate analyses; subtle changes in the mass spectral profiles can alter the mass
282 concentrations of the factors, while still producing mathematically viable solutions (Paatero et al.,
283 2002). This ambiguity was explored through varying the ‘fpeak’ parameter and it was found that
284 values between -0.6 and 1.0 produced solutions that could be considered valid (see Table S3).
285 Outside of this range, solutions produced nonphysical factors or failed to converge properly. It was
286 found that between these values, the concentrations of HOA and COA showed some variation,
287 however the ambiguity was not a direct rotation between the two factors as would be expected for
288 factors derived from UMR data. Because the high-resolution data is good at distinguishing HOA
289 (which is mainly hydrocarbons) from COA (which contains oxygenated peaks), the HOA profile
290 was consistent between all values of fpeak. Instead, the exchange of signals between profiles
291 seemed to involve the COA and two SFOA factors, with HOA variance accounting for changes in
292 the hydrocarbon peaks in the SFOA.

293

294 This range of variation can be seen as indicative of the amount of rotational ambiguity present in
295 the solutions (Allan et al., 2010). However, the solution for fpeak=0 is used for further analysis, as

296 this is most likely to be physically meaningful according to the recommendations of Paatero et al.
297 (2002), which does not advocate the use of nonzero values of f_{peak} for environmental data.

298

299

300 **3. RESULTS AND DISCUSSION**

301 **3.1 Measured Ambient Concentration Levels at NK and HAR**

302 Average concentration statistics are shown in Table 2 for the measured components that are used in
303 the CMB modelling. The mean concentrations of $PM_{2.5}$ and its chemical components were mostly
304 higher at the NK urban site than at the HAR rural site except for woodsmoke (levoglucosan) and
305 vegetative detritus (n-alkanes) marker compounds which showed the opposite, whilst similar
306 concentrations were observed for chloride, nitrate and sulphate.

307

308 **3.1.1 Secondary organic marker components at NK**

309 Quantifiable concentration levels of the secondary biogenic compounds were detected (0.19-1.3 ng
310 m^{-3}) (Table 2) at the London NK urban background site, but these levels are lower than those
311 measured from other European sites (Table 3). Wagener et al. (2012a) conducted measurements at
312 three sites (HV – high vegetation influenced site, LV – low vegetation influenced site and regV –
313 regional vegetation influenced site) in Berlin, Germany, and reported higher levels of those
314 molecular markers for PM_{10} and $PM_{1.0}$ at HV, PM_{10} and $PM_{1.0}$ at LV and PM_{10} at regV.
315 Regardless of the different particle size fraction measured, the concentration levels in Berlin are
316 roughly 2-10 times higher than those at the London site, presumably due to influences from the
317 surrounding forest area at the Berlin sites. Higher levels were also found for pinonic and pinic acids
318 in the PM_{10} fraction at a background station in southern Sweden, , with higher levels in summer
319 than in winter (Hyder et al., 2012), and for 2-methylthreitol and 2-methylerythritol in PM_{10} at a
320 rural background site at Hyytiälä, Finland (Kourtchev et al., 2005). It is interesting to note that
321 similar concentrations have been observed at the rural background site, Hyytiälä, Finland in autumn

322 2004 for the two isoprene-oxidation products to those at the UK NK site, although the former is
323 surrounded by forests. Clearly meteorological/seasonal effects as well as source proximity can
324 influence the levels of the biogenic secondary organic markers, which explain the low
325 concentrations detected at our site in the winter months. In particular, the isoprene-derived
326 compounds show much higher levels in summer than in winter, whilst similar concentrations were
327 observed for α -pinene derived products (Wagener et al., 2012a), which may explain the higher
328 concentrations of pinic and pinonic acids than 2-methyltetrols at our NK site. Another study in
329 summer 2002, in a coniferous forest in Germany (Plewka et al., 2006) indicated very different day
330 and night concentrations, with higher night-time levels for pinic acid, but higher daytime levels for
331 pinonic acid, 2-methylthreitol and 2-methylerythritol.

332

333 Apart from the European data, sampling has also been reported from four sites in a heavily wooded
334 region in the south-eastern US in June 2004, and also showed higher average concentrations of 2-
335 methylthreitol and 2-methylerythritol of 4.8 and 11.9 ng m⁻³ at the inland sites, and 1.6 and 4.9 ng
336 m⁻³ at the coastal site (Clements and Seinfeld, 2007). It is noticeable that the concentrations of 2-
337 methylerythritol are always higher than those of 2-methylthreitol at both our NK site and in the
338 published work, whilst higher levels of pinonic acid than pinic acid were observed at NK and in
339 Berlin, Germany (Wagener et al., 2012a) but not at the background station in Southern Sweden
340 (Hyder et al., 2012) where similar mean concentrations were observed.

341

342 **3.1.2 Primary organic components at NK and HAR**

343 Concentrations of both biogenic and anthropogenic primary molecular marker compounds were
344 mostly similar or higher in comparison with those of secondary marker compounds at the UK NK
345 site, where anthropogenic sources such as traffic emissions, wood smoke and food cooking markers
346 play an important role. Higher levels were found for levoglucosan (Levo) (73.9 ng m⁻³ and 94.5 ng
347 m⁻³), hopanes (0.25-0.50 ng m⁻³ and 0.079-0.36 ng m⁻³) and PAHs (0.10-0.67 ng m⁻³ and 0.044-0.51

348 ng m⁻³) at the current southeast England sites NK and HAR in winter 2012 than that measured at the
349 UK West Midlands urban background monitoring site, EROS, (Levo: 9.2 ng m⁻³; hopanes: 0.08-
350 0.18 ng m⁻³; PAHs: 0.06-0.27 ng m⁻³) and rural site CPSS (Levo: 7.7 ng m⁻³; hopanes: 0.07-0.15ng
351 m⁻³; PAHs: 0.05-0.21ng m⁻³) in 2007-2008 (Harrison and Yin, 2010), but lower levels were
352 observed for n-alkanes from the current study (0.58-2.1 ng m⁻³ and 1.2-3.7 ng m⁻³ for NK and
353 HAR), presumably due to a seasonal effect, since earlier results (1.0-5.2 ng m⁻³ and 1.8-4.7 ng m⁻³
354 for EROS and CPSS) cover both summer and winter periods. Similar or higher n-alkane
355 concentration levels can be found from the current study if compared with winter periods only for
356 EROS (0.73-1.9 ng m⁻³) and CPSS (0.47-1.7 ng m⁻³).

357

358 **3.2 CMB Model Results**

359 Source contributions to fine particulate OC and PM_{2.5} were calculated with the CMB model for the
360 averaged samples for the whole sampling periods and for the individual daily samples for NK and
361 HAR sites (Table 4 and Figures 1-4).

362

363 **3.2.1 Source contributions to fine OC from the CMB model**

364 Seven primary pollution sources were apportioned initially using the average concentration data
365 (Table 4 and Figure S1) that contribute on average about 53% at NK and 56% at HAR of the
366 particulate organic carbon including traffic, wood smoke/biomass burning, food cooking, vegetative
367 detritus, coal combustion, natural gas combustion and road dust/soil. The most significant sources
368 identified are vehicle exhaust and woodsmoke emissions contributing about 21% and 15% of
369 organic carbon (OC) at the London urban background NK site, and 16% and 28% at the rural HAR
370 site. Other sources together contribute a relatively smaller amount of about 18% for NK and 12%
371 for HAR respectively, including a significant amount of food cooking particularly at NK (NK: 11%
372 and HAR: 3%), coal combustion (NK: 2% and HAR: 2%), vegetative detritus (NK: 2% and HAR:
373 5%), natural gas combustion (NK: 1% and HAR: 2%) and dust/soil (NK: 1% and HAR: 1%). As

374 expected, most of the source contribution estimates, such as traffic, food cooking, coal combustion,
375 dust/soil were larger at the urban site NK than that at the rural site HAR, where dust/soil (in bold
376 figures) was not statistically significant over the winter period. The unidentified sources, referred to
377 as 'Other-OC', calculated as the difference between the measured total organic carbon and the sum
378 of all source contribution estimates has been considered as being mostly secondary organic aerosol
379 (SOA) (Yin et al., 2010) and any unidentified primary sources that are not accounted for in the
380 CMB modelling. These represent about 47% at NK and 44% at HAR of the measured particulate
381 OC over the whole sampling period. Daily source contributions fluctuate at both sites with a
382 tendency that higher percentage mass explained by the model mostly occurred when ambient OC
383 levels were low, and on the other hand, a large un-apportioned Other-OC component was often
384 associated with high OC levels, indicative that secondary sources played an important role in these
385 samples.

386

387 In order to apportion the Other-OC component, a source profile of the secondary biogenic
388 component was generated using the mean ambient measurement data from Berlin, Germany
389 (Wagener et al., 2012a). Both PM_{10} and $PM_{1.0}$ fractions were available, but the $PM_{1.0}$ was used
390 since previous data obtained by Wagener et al. (2012a) suggested that those biogenic markers were
391 present mostly in the fine rather than the coarse fraction. The newly measured data on secondary
392 biogenic molecular markers at NK, the 2-methyltetrols and the α -pinene-oxidation products,
393 pinonic and pinic acid was combined with those source markers used earlier, and the CMB
394 calculation was repeated to estimate an OC source contribution from secondary biogenic sources.
395 Table 4 and Figure S1 show the mean results with (NK2012b) and without (NK2012a) the addition
396 of these new molecular markers.

397

398 On average, organic carbon was much better accounted at NK (79%) with the addition of the
399 secondary biogenic components than without (54%). In addition to the major primary source

400 components, traffic ($0.73 \mu\text{g m}^{-3}$), wood smoke ($0.54 \mu\text{g m}^{-3}$) and food cooking ($0.39 \mu\text{g m}^{-3}$), the
401 secondary biogenic concentration was estimated at $0.90 \mu\text{g m}^{-3}$, representing about 26% of the total
402 organic carbon mass for North Kensington in winter 2012. Similar concentration estimates were
403 obtained using a PMF model in Berlin with ranges for PM_{10} of $0.34\text{-}0.84 \mu\text{g m}^{-3}$ and $\text{PM}_{1.0}$ of 0.43-
404 $1.03 \mu\text{g m}^{-3}$ in the colder months, and as $0.9\text{-}1.5 \mu\text{g m}^{-3}$ for PM_{10} and $1.1\text{-}1.2 \mu\text{g m}^{-3}$ for $\text{PM}_{1.0}$ in the
405 warmer months (Wagener et al., 2012b). Relative source contributions to OC in Berlin were mostly
406 similar to the UK site for the colder months (6.3-32.2%), but higher values were obtained for the
407 warmer period (20.0-54.5%) in Berlin. Daily CMB results (Figures 1 and 2) showed, as expected,
408 fluctuations for the source contribution, with a few days when more OC was apportioned by the
409 model than was measured (Figure 2). Over 30% of the OC was attributed to Other-OC on the days
410 starting January 13th, 17th, 30- 31st and February 3-5th when higher pollution levels of OC occurred,
411 which is likely due to other biogenic and anthropogenic primary or secondary sources that are not
412 accounted for on those days. Air mass back trajectories (Figure S2) over those periods indicated that
413 the high OC levels were strongly influenced by pollutants transported from mainland Europe on
414 January 17th, 30th and 31st, whilst during January 13th local or regional sources within the UK were
415 dominant as the air mass travelled across the Atlantic Ocean and passed through southern England
416 before reaching the site. On February 3-5th, both sources from mainland Europe and UK
417 regional/local may have contributed.

418

419 **3.2.2 Source contributions to $\text{PM}_{2.5}$ from the CMB model**

420 Source contribution estimates were calculated by multiplication of the fine OC source
421 apportionment concentrations by the ratios of $\text{PM}_{2.5}$ mass to fine OC obtained from the same source
422 profiles used for the primary OC apportionment (Pant et al., 2014; Rogge et al., 1993a,b; Fine et al.,
423 2004; Zhang et al., 2008; Schauer, 1998; Zhao et al., 2007b). Whilst the aerosol mass to OC ratio is
424 not available for the secondary biogenic sources, a ratio of 1.8 was used to obtain this source
425 contribution estimate for the NK site (Utembe et al., 2009). In addition to the seven/eight source

426 components calculated from OC source contribution estimates, other organic matter (Other-OM)
427 was estimated by multiplying the 'Other-OC' by a factor of 1.8 (Utembe et al., 2009), sea salt
428 calculated as $1.65 \cdot \text{Cl}^-$, ammonium sulphate as $1.38 \cdot \text{SO}_4^{2-}$ and ammonium nitrate as $1.29 \cdot \text{NO}_3^-$
429 were added into the $\text{PM}_{2.5}$ source apportionment (Harrison et al., 2003).

430
431 The output of the CMB model is critically dependent upon the source profiles used. The starting
432 point was those used by Yin et al. (2010) which were mostly derived from earlier work in North
433 America. A sensitivity study was conducted in which three ways were used of estimating the
434 profile for road traffic following the work of Pant et al. (2014). The first two methods used
435 dynamometer data, one using profiles of gasoline and diesel exhaust from dynamometer tests
436 reported by Schauer et al. (1999; 2002). Secondly, more recent dynamometer data, reported by
437 Lough et al. (2007), were utilised. Thirdly, a profile for road traffic created from measurements in a
438 heavily-trafficked street canyon in London after subtraction of the local urban background as
439 reported in Pant et al., (2014) was tested. This profile seems more likely to be reflective of the
440 current vehicle fleet in London as the data are relatively recently collected. For each of these three
441 traffic profiles, CMB was run with two different woodsmoke profiles, both taken from the USEPA
442 SPECIATE database, one for USEPA Region 4 and the other for USEPA Region 5. In order to
443 judge which source profiles gave the best results, the estimate for road traffic exhaust from CMB
444 was compared with an estimate based on elemental carbon using an OC/EC ratio of 0.63 as
445 measured in recent London data as well as an OC/EC ratio of 0.35 as reported for roadside sites in
446 Europe by Pio et al. (2011). The derivations from elemental carbon concentration gave estimated
447 traffic OC of $0.96 \mu\text{g m}^{-3}$ and $0.54 \mu\text{g m}^{-3}$ from the two OC/EC ratios. The CMB estimate of traffic
448 exhaust OC from the application of the traffic source profile derived from data collected in London
449 by Pant et al. (2014) of $0.73 \mu\text{g m}^{-3}$ lay comfortably between these values, and hence the measured
450 London profile was our final choice.

451

452 There was also some sensitivity of the model output to the choice of woodsmoke profile, with the
453 EPA Region 5 profile giving an estimate for woodsmoke OC of $0.53 \mu\text{g m}^{-3}$ and that for EPA
454 Region 4, giving $0.78 \mu\text{g m}^{-3}$ of organic carbon. The estimate for woodsmoke organic carbon based
455 upon mean levoglucosan for the campaign times a factor of 7.35 (Puxbaum et al., 2007) gave an OC
456 concentration of $0.54 \mu\text{g m}^{-3}$ which lay very much closer to the estimate using the USEPA Region 5
457 source profile and therefore this profile was adopted for the final runs of the model.

458

459 Once the optimal source profiles had been selected using the campaign averaged dataset, source
460 contribution estimates to OC and $\text{PM}_{2.5}$ were run both on the averaged dataset and on the data for
461 each separate day of the campaign. Since the model needs to be separately optimised for each
462 measurement day, the average of the daily model runs presented in Table S4 is slightly different
463 from that derived from the overall campaign averaged concentration data which is reported in Table
464 5. Unless stated otherwise, source contribution estimates reported are derived from the overall
465 average dataset.

466

467 Concentrations of woodsmoke $\text{PM}_{2.5}$ were found by the CMB model to be an average of $0.64 \mu\text{g}$
468 m^{-3} at North Kensington in the winter 2012 campaign and $0.77 \mu\text{g m}^{-3}$ at Harwell during the same
469 campaign (from daily data). These concentrations are slightly higher but comparable in magnitude
470 to those measured in London in the winter of 2011 (Harrison et al., 2012). As they were measured
471 at the coldest time of the year, it can be anticipated that annual average concentrations of
472 woodsmoke would be appreciably lower than from those measured during the winter campaign and
473 probably no more than 50% of these concentrations.

474

475

476 $\text{PM}_{2.5}$ source apportionment results for both mean and daily samples at NK and HAR sites are
477 shown in Table 4 and Figures 3 and 4. $\text{PM}_{2.5}$ mass was well explained by those source components
478 which represented about 81%/92% without/with the addition of secondary biogenic component at

479 the urban NK site over the winter period. This comprised on average of 37% ammonium nitrate, 14%
480 ammonium sulphate, 8% exhaust emissions, 10% secondary biogenic, 7% sea salt and 24% of all
481 other identified/unidentified source components (vegetative detritus, wood smoke, natural gas, coal,
482 dust/soil, food cooking, Other-OM and the unidentified component). The rural HAR site, with 83%
483 total explained $PM_{2.5}$ mass, showed similar relative source contributions from ammonium salts (37%
484 ammonium nitrate and 19% ammonium sulphate) but a smaller relative contribution from vehicle
485 exhaust emissions (6%). In comparison, ammonium salts were also predominant in the UK West
486 Midlands sites for both winter (urban background: 33.6%, rural: 37.7%) and summer (urban
487 background: 52.5%, rural: 44.2%) periods (Yin et al., 2010), with much higher estimated
488 contributions from vehicular emissions in the winter months (urban background: 29.0%, rural:
489 23.7%), due to apportionment with separate traffic source profiles for diesel, gasoline and smoking
490 engines which lead to an overestimation (Pant et al., 2014). In addition, the source contribution
491 estimates from food cooking (previously not apportioned) were not large but significant particularly
492 at the urban NK site (OC: $0.39 \mu\text{g m}^{-3}$, $PM_{2.5}$: $0.69 \mu\text{g m}^{-3}$) representing about 11% of the OC and 4%
493 of the $PM_{2.5}$ mass. As discussed for OC above, the secondary biogenic source contribution, 26% for
494 OC and 10% for $PM_{2.5}$ at the UK site NK, cannot be ignored, particularly as during summer months
495 this component may be significantly larger (Wagener et al., 2012b). Heal et al. (2011) studied the
496 carbon-14 content in $PM_{2.5}$ samples from a UK West Midlands urban background site (EROS) in
497 2007/2008, and found that the fraction of contemporary total carbon f_c (TC) was positively
498 correlated to the ratio SOC/TC, which were both related to air mass origin, suggesting that
499 secondary organic aerosol is substantially associated with the oxidation of biogenic VOC emissions
500 from terrestrial contemporary carbon sources from continental Europe. An average estimate of
501 about 40% of the total carbon and 9-29% of the $PM_{2.5}$ was attributed to biogenic SOC or biogenic
502 SOA (Heal et al., 2011). Those contribution estimates were higher in comparison with the current
503 CMB estimates from the NK site (26% and 10% for OC and $PM_{2.5}$), which is likely due to (a) use of
504 a different sampling season, i.e. the study at NK only involve winter months while both winter and

505 summer months were included in the study by Heal et al. (2011), and (b) estimates of contemporary
506 carbon using carbon-14 also contain sources of non fossil OC/OA other than biogenic SOC/SOA,
507 whilst the estimates from the current study include only those components which correlate with the
508 oxidation products of α -pinene or isoprene.

509
510 The results from NK indicate an average secondary/Other-OC organic component of PM_{2.5} mass of
511 2.95 $\mu\text{g m}^{-3}$, of which 1.63 $\mu\text{g m}^{-3}$ (55%) is accounted for by the inclusion of oxidation products of
512 α -pinene and isoprene. The CMB method can address the composition and origin of this material,
513 but does not explicitly consider its mechanism of formation. It seems very probable that production
514 of other biogenic VOC and their oxidation products would correlate strongly with α -pinene and
515 isoprene, which are recognised as two of the main precursors of biogenic secondary organic aerosol.
516 This leads to the tentative conclusion that at least 55% of the secondary organic aerosol (SOA) is
517 biogenic in origin, even during winter, consistent with the conclusions of both Heal et al. (2011)
518 and Charron et al. (2013) that SOA in the south and Midlands of the UK is dominated by the
519 biogenic component. By inference, up to 45% of SOA may arise from anthropogenic precursors
520 such as toluene.

521

522 **4. COMPARISON BETWEEN CMB AND AMS-PMF ESTIMATES**

523 The AMS data collected at NK during winter 2012 has been analysed and apportioned using the
524 PMF (Positive Matrix Factorisation) model based on the method used by Allan et al. (2010). Full
525 details of the methods are available from Young et al. (2014). The results identified five source
526 emission components that contribute to organic aerosols/matter (OA/OM), including one secondary
527 component (OOA) and four primary components, two biomass burning/solid fuel burning organic
528 aerosol (SFOA) factors, cooking organic aerosol (COA) and traffic related/hydrocarbon-like
529 organic aerosol (HOA). Estimates of OC from the CMB model are then converted into source
530 contribution estimates to PM_{2.5} (Table 4) using conversion factors established in earlier work (Yin

531 et al., 2010 and references therein). The conversion of organic carbon to PM_{2.5} mass is to allow for
532 other elements (H, N, O) associated with carbon in the organic compounds, as well as other
533 chemical constituents associated with that source. For example, in the case of road traffic exhaust,
534 there is a component of elemental carbon which is included, and for dust/soil, major contributions
535 from inorganic constituents of soil (Si, Al etc.) are included. For comparison with the AMS results,
536 OC values calculated from the CMB modelling have been converted to OM by applying OM/OC
537 ratios considered appropriate to the source, as illustrated in Table 6 and Figure 5. The two SFOA
538 factors have been summed as this gave the strongest correlation to both the woodsmoke and
539 woodsmoke plus coal contributions derived from the CMB. Table 6 used OC average estimates
540 calculated from the daily CMB results, shown in Table S4, in order to be consistent with Figure 5.
541 The conversion of OC to PM_{2.5} in Table 4 used factors taken from earlier CMB studies (see Section
542 3.2.2). Different factors were used to estimate values of OM in Table 6 based upon extensive
543 literature. These may be derived from the ratio of OM for CMB in Table 6 to OC for that source in
544 Table S4.

545

546 **4.1 Woodsmoke Particles**

547 The mean concentration of SFOA ($1.63 \mu\text{g m}^{-3}$) derived from AMS-PMF was almost 2 times the
548 CMB woodsmoke (CMB-WS) concentration estimate ($0.85 \mu\text{g m}^{-3}$) (Table 6), possibly due to the
549 SFOA factor also including particles from other solid fuel combustion apart from wood burning.
550 The AMS-SFOA value remained larger at 1.7 times the CMB value when the CMB coal
551 combustion component was also included (CMB-WS + Coal: $0.97 \mu\text{g m}^{-3}$). The conversion to OM
552 in Table 6 used an OM/OC ratio of 1.6 which is smaller than some literature values, although larger
553 than that in Table 4. This is broadly consistent with the estimated OM/OC ratios of 1.69 and 1.33
554 derived using the high-resolution profiles of SFOA1 and SFOA2 respectively, using the method
555 described by Aiken et al. (2007) and also consistent with the values of 1.91 and 1.40 respectively if
556 the modified algorithm proposed by Canagaratna et al. (2014) is used. Using a larger factor would

557 obviously reduce the divergence between the CMB and AMS estimates from woodsmoke. Turpin
558 and Lim (2001) suggest an OM/OC ratio as high as 2.2-2.6 for an aerosol heavily impacted by
559 woodsmoke, which would have contributed significantly to reducing the divergence of OM masses.
560 Nevertheless good correlation was observed between CMB-WS and AMS-SFOA ($r^2 = 0.75$) (Table
561 5 and Figure 5a), indicating that the SFOA component is closely related to woodsmoke. It is
562 notable that a slightly stronger correlation is observed between the AMS-SFOA and the CMB
563 component (WS+Coal) ($r^2 = 0.78$), which may indicate other sources in the AMS component SFOA
564 apart from woodsmoke. It is interesting to note that both CMB-WS and CMB-(WS+Coal) are
565 correlated more closely to SFOA at low levels of those components (CMB-WS/WS+Coal <0.9/1.2
566 and SFOA < 3.0) as compared to high levels when data points are more scattered from the
567 regression line (Figures 5a and b). This seems most likely to relate to a change in composition at
568 higher concentrations.

569

570 Previous work in the UK has obtained lower biomass smoke OC values for Birmingham EROS
571 2009-2010 ($0.23 \mu\text{g m}^{-3}$), London NK 2010/2011 ($0.33 \mu\text{g m}^{-3}$) and Budbrooke, Warwickshire
572 2009-2010 ($0.42 \mu\text{g m}^{-3}$) (Harrison et al., 2012), whilst six European sites showed biomass smoke
573 OC concentrations in the range of $0.039\text{-}3.1 \mu\text{g m}^{-3}$ annually and $0.048\text{-}7.7 \mu\text{g m}^{-3}$ for winter
574 months (Gelencsér et al., 2007).

575

576 Based on the CMB-WS estimates, woodsmoke can represent on average about 15% and 28% of the
577 OC, and 4% and 7% of the $\text{PM}_{2.5}$ for NK and HAR respectively. The relative contributions of
578 woodsmoke to OC are similar to those measured in Belgium by Maenhaut et al. (2012), and the
579 woodsmoke contributions to $\text{PM}_{2.5}$ are in line with the lower range calculated by Zhang et al. (2010)
580 who conducted measurements at 15 urban/rural sites in the south-eastern US and estimated that the
581 relative contribution of biomass burning to $\text{PM}_{2.5}$ were 13% annually, 27% in winter and 2% in the
582 summer months.

583 4.2 Food Cooking Particles

584 Many studies have indicated that food cooking is one of the most important aerosol emission
585 sources in the indoor environment (Kamens et al., 1991; Zhao et al., 2007a; Buonanno et al., 2009;
586 Wan et al., 2011; Massey et al., 2012). A detailed review has shown that cooking aerosol is a
587 significant PM source for both indoors and outdoors (Abdullahi et al., 2013), and may arise from
588 both residential and commercial food cooking. The AMS has been used to characterise PM and
589 identify organic aerosols from cooking by means of application of PMF to mass spectral data
590 (Kleeman et al., 2008; Allan et al., 2010; Huang et al., 2010; Williams et al., 2010; He et al., 2010
591 and 2011; Sun et al., 2011; Zhang et al., 2011; Mohr et al., 2009 and 2011; Clougherty et al., 2011),
592 whilst CMB is able to calculate the food cooking concentration estimate using appropriate
593 molecular markers (Zheng et al., 2002; Fraser et al., 2003; Schauer et al., 1996; Robinson et al.,
594 2006; Wang et al., 2009). Food cooking estimates were calculated for the first time in the UK using
595 a CMB model with this source profile at both sites NK and HAR. Earlier work (Yin et al., 2010)
596 used only cholesterol as a tracer of meat cooking, suggesting much lower concentrations. AMS-
597 PMF method data were available for the NK site only, for which the results were compared.

598

599 The CMB-derived mean food cooking concentration estimates for OC and OM/PM_{2.5} from
600 averaging the daily CMB results are 0.32 and 0.56 $\mu\text{g m}^{-3}$ at NK (Tables S4 and 5), representing
601 about 9% of the fine OC and 4% of the PM_{2.5} mass. In comparison, the PMF apportioned results
602 from the AMS data gave a value of 0.87 $\mu\text{g m}^{-3}$ for cooking organic aerosol, which is about 1.6
603 times the value of the PM_{2.5} CMB estimate. A strong correlation ($r^2 = 0.80$, Pearson) was found for
604 the daily food cooking estimates between the AMS-PMF and CMB datasets, with a gradient of 2.40
605 (the gradient is 1.76 with a small intercept of -0.13 after removing the high value point) (Table 5
606 and Figure 5c). A much lower food cooking contribution was obtained from the CMB method for
607 HAR (0.12 $\mu\text{g m}^{-3}$ or 1% of PM_{2.5}), where no immediate local sources were present.

608

609 Similar food cooking concentration estimate ranges have been observed in many previous studies
610 for outdoor measurements from both CMB and AMS-PMF source apportionment. Fraser et al.
611 (2003) conducted source apportionment using CMB for both urban and background sites in
612 Houston, Texas, and identified a $PM_{2.5}$ source component of meat cooking of $0.9\text{-}1.3 \mu\text{g m}^{-3}$ at an
613 urban and $0.7 \mu\text{g m}^{-3}$ at a background site. Robinson et al. (2006) used CMB to apportion ambient
614 fine OC, and indicated that 10% or $0.32 \mu\text{g m}^{-3}$ of the OC is attributable to food cooking in
615 Pittsburgh, Pennsylvania. Zheng et al. (2002) estimated that about 5-12% of the fine OC arose from
616 meat cooking emissions in the South-eastern US, whilst Lee et al. (2008) estimated that 12% of the
617 $PM_{2.5}$ mass was from meat cooking in Korea. Sun et al. (2011) apportioned $PM_{1.0}$ AMS data using
618 PMF and obtained $1.02 \mu\text{g m}^{-3}$ of cooking organic aerosol in New York City, which contributes
619 30% to the primary OA. Williams et al. (2010) analysed the AMS data collected in Southern
620 California with PMF source apportionment, which identified 10.4% or $0.98 \mu\text{g m}^{-3}$ of cooking
621 emissions from the organic aerosol in summer. Huang et al. (2010) concluded that about 24.4% of
622 the OA was related to cooking emissions during the Beijing 2008 Olympic Games. There appears
623 to be a trend, with AMS studies estimating larger contributions to OM and $PM_{2.5}$ than CMB studies.
624 Summertime concentrations for cooked meat-related air particles are normally higher than in
625 wintertime, presumably due to increased outdoor cooking activities and open kitchen windows.

626

627 **4.3 Traffic Related Particles**

628 The CMB mean concentration estimates (Table 4) of total traffic are $0.73 \mu\text{g m}^{-3}$ for OC and $1.26 \mu\text{g}$
629 m^{-3} for $PM_{2.5}$ at North Kensington, while about a half of these levels were observed at rural Harwell
630 (OC: $0.36 \mu\text{g m}^{-3}$ and $PM_{2.5}$: $0.61 \mu\text{g m}^{-3}$). The relative contribution of total traffic to OC and $PM_{2.5}$
631 at the urban site is about 21% and 8%, and at the rural site about 16% and 6% respectively. Very
632 strong correlations ($r^2 = 0.90\text{-}0.99$) have been observed for the CMB traffic component and other
633 related variables such as aethalometer BC, measured EC and calculated primary organic carbon
634 (obtained based on the method of Castro et al., 1999) at both urban and rural sites (Pant et al.,

635 2014). The AMS-PMF derived component HOA is also strongly correlated with the CMB OM
636 traffic component ($r^2 = 0.80$) for NK, and if the two outlying points are removed, the correlation
637 improves ($r^2 = 0.86$) and the gradient reduces to 0.98. The absolute OM concentration levels are
638 very similar from the two methods, showing a CMB traffic estimate of $0.98 \mu\text{g m}^{-3}$ and an AMS
639 value of $0.86 \mu\text{g m}^{-3}$ (Table 6).

640

641 **4.4 Secondary Particles**

642 The CMB component Other-OC/Other-OM is regarded as mostly secondary OC/OM (Yin et al.,
643 2010). The Other-OM, with concentration estimates of $2.92 \mu\text{g m}^{-3}$ at North Kensington and 1.85
644 $\mu\text{g m}^{-3}$ at Harwell (Table S4), represented about 46% and 45% of the total organic aerosol (OA)
645 mass, and 19% and 17% of the $\text{PM}_{2.5}$ mass at those two sites. The secondary component derived
646 from the AMS-PMF results (OOA) is $0.99 \mu\text{g m}^{-3}$ for the same period at NK, which is well below
647 the CMB Other-OM level. Docherty et al. (2008) studied secondary organic aerosol (SOA) at
648 Riverside, Southern California in the summer period using different methods, and showed very
649 similar proportions of SOA/OA estimated by the CMB (77%) and AMS-PMF ($74 \pm 19\%$) methods,
650 but with much higher relative contributions of SOA comparing with our UK NK site. However, the
651 study also summarised results from previous studies in the Eastern LA Basin area with different
652 methods, found that the SOA/OA ratios were mostly less than 50% (range: about 15-50%), with
653 only one exception (75%) by Schauer et al. (2002), and attributed those large differences to
654 variations in sampling season, location, duration and methodology.

655

656 The CMB Other-OC/OM is strongly correlated with the calculated Sec-OC (obtained based on the
657 elemental carbon tracer method of Castro et al., 1999) at both sites ($r^2 = 0.84$ and 0.62 for North
658 Kensington and Harwell respectively), confirming that this component is mainly secondary
659 organics, and lesser but still significant correlations were observed for Other-OM with secondary
660 inorganic components, sulphate and nitrate. A strong relationship was found for AMS OOA with

661 the CMB Other-OM ($r^2 = 0.68$), Sec-OC ($r^2 = 0.64$) and inorganic components SO_4^{2-} and NO_3^- ($r^2 =$
662 0.79 and 0.80) as expected (Table 5).

663

664 Regression analyses showed low to moderate correlation between the measured biogenic secondary
665 marker 2-methyltetrols and the calculated Sec-OC/CMB Other-OC ($r^2 = 0.25$ - 0.41), whilst higher
666 correlations were found between the measured α -pinene oxidation products and Sec-OC/CMB
667 Other-OC ($r^2 = 0.31$ - 0.82). Pinic acid particularly showed strong correlation with Sec-OC/CMB
668 Other-OC (Figure 6). In comparison, no correlation was found between 2-methyltetrols and AMS
669 OOA showing close to zero coefficients, whilst pinic acid again exhibited a good relationship with
670 the AMS OOA component ($r^2 = 0.55$) (Figure 6). The results of Kleindienst et al. (2007) from
671 measurements of VOC oxidation products at a US site attribute far greater importance to α -pinene
672 than isoprene as an SOA precursor during the cooler months of the year. The species of trees and
673 shrubs present locally will also be influential.

674

675 **4.5 Overview of CMB Comparison with AMS-PMF Results**

676 There are few previous published studies that have compared the results between the AMS-PMF
677 and the CMB methods, and this first comparison study in the UK at the NK site has shown some
678 inconsistent results for individual primary component estimates, and a different split between the
679 total primary and secondary source components. In a study in Mexico, Aiken et al. (2009) found
680 similar average OA/OM apportionment values from the two methods AMS-PMF and CMB for
681 HOA/Vehicle ($4.5 \mu\text{g m}^{-3}$, 28%/ $4.5 \mu\text{g m}^{-3}$, 29%), BBOA/Woodsmoke ($2.7 \mu\text{g m}^{-3}$, 17%/ $1.7 \mu\text{g m}^{-3}$,
682 12%) and OOA/Other-OM ($7.4 \mu\text{g m}^{-3}$, 46%/ $9.2 \mu\text{g m}^{-3}$, 58%), but the source components
683 calculated were different to our study.

684

685 Generally speaking, overall correlations between the CMB and PMF based estimates here are very
686 good, but the quantitative agreement is lacking, with PMF estimates generally greater than CMB for

687 the primary species (in particular, cooking and solid fuel burning) and CMB assigning a much
688 larger proportion of organic matter to secondary aerosol. Agreement between the methods is
689 relatively good for the traffic source, and fair for food cooking (Table 6). Measurements of
690 levoglucosan and fine potassium made during the campaign (Crilley et al., 2014) give an estimate
691 for biomass burning particles consistent with the CMB results, and other work based upon
692 elemental carbon data (arising mostly from diesel emissions) suggests that the traffic estimate in the
693 CMB model is reasonable (Pant et al., 2014). The AMS estimates of SFOA and HOA also correlate
694 very well with a Single Particle Soot Photometer (SP2) -based apportionment of the black carbon
695 particles present, based on the mass of individual particles and coating thicknesses (Liu et al.,
696 2014).

697

698 It is important to remember that the CMB model is applied to organic carbon, which is apportioned
699 into the eight categories listed at the top of Table 4. Table 6 shows a comparison of the CMB with
700 the AMS data, attempting where possible to express the CMB results as solely the organic matter
701 content so as to be comparable with the AMS data. The AMS factors do not include vegetation,
702 natural gas and dust/soil in their apportionments, however this is not unexpected; vegetation and
703 dust/soil particles are generally too large to be detected by the AMS and natural gas does not
704 contribute a sufficient mass of particulate organics to the overall loading.

705

706 There are also a number of technical reasons why both techniques may deliver inaccurate estimates.
707 In the case of the CMB model, uncertainties arise from the concentration measurements, the source
708 profiles and the amount of colinearity amongst the source profiles. This is expressed quantitatively
709 as a standard deviation in Table 4, and varies substantially in proportion to the Source Contribution
710 Estimate between source categories. The uncertainties in AMS source apportionment results are
711 more difficult to quantify, and possible sources of error in both techniques are discussed below. The
712 sum of the two sets of measurements when expressed as organic matter is greater for the CMB than

713 the AMS (Table 6). The CMB model is applied to samples of PM_{2.5} whereas the AMS samples
714 particles up to around 0.8 µm, and consequently fails to sample larger particles, which are possibly
715 of different composition. It is conceivable that the larger estimate of SOA by CMB may be caused
716 in part by condensation of secondary material onto supermicron particles.

717

718 One issue that may affect the AMS is an ambiguous collection efficiency (CE). A time-dependent
719 and composition-dependent parameterisation of CE was used, with a value of 0.5 used for most of
720 the data, in line with the parameterisation of Middlebrook et al. (2012). However, it is possible that
721 if a certain particulate population is externally mixed with the inorganic fraction in the
722 accumulation mode, it may exhibit a different collection efficiency. While the overall CE was
723 validated against a DMPS, it is possible that if a fraction makes a low enough contribution to the
724 total volume concentration, it may not be picked up through this test. In the event that cooking
725 particles are liquid, which is deemed likely given that many of the constituents such as oleic acid
726 are of this phase at room temperature, their concentration could be overestimated by the AMS by up
727 to a factor of 2. Note that this will only apply to particles that have not diffused onto the
728 accumulation mode, so the level of overestimation could be less than 2, even if the true CE of pure-
729 component particles is unity. However, this could account for the majority of the discrepancies
730 with the primary particles.

731

732 PMF analysis is subject to inherent uncertainties associated with rotational ambiguity (Allan et al.,
733 2010) referred to in Section 2.2.5. This can result in an amount of the mass being misattributed
734 between factors and it is conceptually possible that some secondary material could be wrongly
735 interpreted as solid fuel burning; indeed, there is evidence for this occurring to an extent in the
736 CTOF instrument, which suffers from this phenomenon more than the HR-AMS used here (Young
737 et al., 2014). Dall'Osto et al. (2014) showed that the COA factor from AMS data in Cork, Ireland
738 exhibited an association with other urban aerosol sources, possibly indicating that it contained a

739 contribution from these rather than cooking. These issues should be manifested as rotational
740 freedoms within the solution sets and by varying the f_{peak} parameter, a tangible amount of
741 uncertainty in the PMF outputs can be attributed to this (see Table S3) but this in itself is not large
742 enough to explain the discrepancies. It is worth noting in particular that the ambiguity identified
743 using this method consists mainly of a redistribution of mass between the primary factors, so would
744 not explain in isolation an overestimation of both SFOA and COA. Nevertheless, it could be that
745 this might explain at least part of the discrepancies reported.

746

747 There are a number of problems that may cause the CMB model to be inaccurate. Firstly, there is a
748 general uncertainty surrounding how representative the source profiles assumed are of the aerosols
749 encountered in this environment, but the sensitivity study of CMB profiles discussed above was
750 intended to probe and minimise such effects. That said, it is recognised that the application of CMB
751 to the secondary fraction is inherently highly uncertain, owing to the broad range of precursors and
752 the complexity of the chemistry. It is also possible that some of the marker compounds are being
753 lost from the aerosols between emission and measurement, either through repartitioning to the gas
754 phase as the aerosol undergoes dilution in the atmosphere, chemical reactions through atmospheric
755 processing, or evaporating from the filter during or after sampling. Levoglucosan could be an
756 example, contributing to a CMB underestimate of woodsmoke mass. The Partisol sampler used to
757 collect samples for the CMB estimation is subject to both positive (adsorption of vapour on the
758 filter) and negative (evaporation of semi-volatile material from the filter) artefacts.

759

760 **5. CONCLUSIONS**

761 The CMB and AMS-PMF methods use entirely different processes to apportion organic carbon and
762 organic matter respectively to source categories. The CMB method is able to attribute carbon to a
763 larger number of sources, but depends upon prior knowledge of source profiles, which must add a
764 significant element of uncertainty. The AMS-PMF method makes no *a priori* assumptions, but

765 depends upon the PMF to separate components with many similarities in their mass spectra, and is
766 able to apportion into a smaller number of classes.

767

768 The use of other marker elements/compounds to estimate source contributions from biomass
769 burning and road traffic gives some confidence in the estimates from the CMB approach. It also
770 attributes carbon to sources not recognised by AMS-PMF, which must be in some way included in
771 the factors output by this method.

772

773 Although the average mass estimates for primary components differ appreciably between the CMB
774 and AMS-PMF results, the estimated daily average concentrations for each generic source category
775 show generally good correlations. In summary, the PMF estimations were higher than CMB by over
776 a factor of two for solid fuel burning and cooking, slightly higher for traffic and lower for the
777 secondaries. While no single issue with either technique can explain the discrepancies, they are
778 within the scope of a combination of known problems and ambiguities (such as AMS collection
779 efficiency, PMF rotation, inhomogeneous distribution of components across size fractions, Partisol
780 collection artefacts and uncertainties in CMB profiles). Work needs to be performed to better
781 constrain all of these issues. This would include obtaining further local source profiles for use in
782 the CMB model, and designing air sampling so that the bulk chemical data used in the CMB model
783 derive from the same particle size range as is sampled by the AMS.

784

785

786 **ACKNOWLEDGEMENTS**

787 Funding was received from the Natural Environment Research Council as part of the ClearfLo
788 project (NERC NE/H003142/1). Raw data are available from the authors on request.

789

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1161 **TABLE LEGENDS**

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1178 **FIGURE LEGENDS**

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1180 **Figure 1** Daily OC source contributions at NK and HAR (without secondary biogenic
1181 component identified separately).
1182
1183 **Figure 2** Daily OC source contribution estimates with secondary biogenic components at NK.
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1185 **Figure 3** Daily $\text{PM}_{2.5}$ source contribution estimates with secondary biogenic components at
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1190 **Figure 5** Primary component comparison between the AMS-PMF and CMB methods at NK.
1191
1192 **Figure 6** Secondary component comparison at NK showing relationships between pinic acid
1193 concentrations and estimated secondary OC from the elemental carbon tracer method
1194 (Sec-OC) and the CMB model (Other-OC), and with the AMS OOA factor.
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1198 **Table 1: Air sampling and analytical instrumentation**

Site	Instrument/ Filter media	Particle size	Chemical analyses /Measurements	Analyser/ Methods	Sample intervals
NK	Partisol/ PTFE filter	PM _{2.5}	PM mass Al, Si, Fe Ti, Mn, Ni, Cu, Zn, Ba	Balance XRF ICP-MS	Daily
	Digitel/ Quartz filter	PM _{2.5}	Organic markers (including secondary) OC & EC SO ₄ ²⁻ , NO ₃ ⁻ , Cl ⁻ , Na ⁺ , K ⁺ , Mg ²⁺ , Ca ²⁺ , NH ₄ ⁺	GCMS OCEC analyser Dionex	Daily
	TOF-AMS	PM _{0.8}	Mass size segregated Organic aerosol OOA, BBOA, HOA, COA	PMF	1-10 minutes
HAR	Partisol/ PTFE filter	PM _{2.5}	PM mass Al, Si, Fe Ti, Mn, Ni, Cu, Zn, Ba	Balance XRF ICP-MS	Daily
	Digitel/ Quartz filter	PM _{2.5}	Organic markers (not including secondary) OC & EC SO ₄ ²⁻ , NO ₃ ⁻ , Cl ⁻ , Na ⁺ , K ⁺ , Mg ²⁺ , Ca ²⁺ , NH ₄ ⁺	GCMS OCEC analyser Dionex	Daily

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1200

Table 2: Concentration summary of measured components at NK and HAR

Components	NK			HAR		
	Mean	Median	10%ile/90%ile	Mean	Median	10%ile/90%ile
<i>PM2.5</i> ($\mu\text{g m}^{-3}$)	15.7	12.6	4.9/29.8	11.0	10.2	3.1/19.6
<i>OC</i>	3.5	2.6	1.3/6.7	2.3	2.1	0.46/5.0
<i>EC</i>	1.5	1.2	0.67/2.7	0.68	0.56	0.12/1.3
<i>Cl⁻</i>	0.67	0.60	0.33/1.3	0.50	0.45	0.23/0.76
<i>SO₄²⁻</i>	1.8	1.3	0.58/3.8	1.7	1.3	0.46/4.4
<i>NO₃⁻</i>	3.5	2.6	0.69/7.6	3.2	2.7	0.42/6.3
<i>Al</i>	0.044	0.035	0.017/0.086	0.027	0.019	0.008/0.060
<i>Si</i>	0.14	0.12	0.056/0.25	0.077	0.056	0.023/0.15
<i>Levo</i> (ng m^{-3})	73.9	69.5	42.7/118	94.5	99.0	27.2/152
<i>C29</i>	2.1	1.7	1.2/4.0	3.7	3.6	1.6/5.4
<i>C31</i>	1.7	1.3	0.84/3.1	2.9	3.0	0.93/4.3
<i>C33</i>	0.95	0.75	0.43/2.0	1.2	1.2	0.54/2.0
<i>C35</i>	0.58	0.47	0.28/1.1	3.1	3.5	0.4/4.3
<i>17αTNohop</i>	0.25	0.19	0.14/0.36	0.079	0.075	0.042/0.12
<i>17$\beta$$\alpha$Nohop</i>	0.50	0.41	0.24/0.84	0.36	0.36	0.25/0.47
<i>17$\alpha$$\beta$Hop</i>	0.33	0.26	0.20/0.39	0.17	0.16	0.14/0.22
<i>B(k)F</i>	0.67	0.50	0.15/1.4	0.49	0.38	0.053/1.0
<i>B(b)F</i>	0.54	0.39	0.15/1.1	0.51	0.48	0.12/0.94
<i>B(e)P</i>	0.48	0.35	0.14/0.94	0.33	0.29	0.048/0.66
<i>IP</i>	0.40	0.28	0.11/0.84	0.29	0.23	0.041/0.62
<i>PIC</i>	0.10	0.081	0.045/0.17	0.044	0.034	0.005/0.10
<i>B(ghi)PER</i>	0.47	0.35	0.15/0.94	0.25	0.20	0.039/0.52
<i>Palma</i>	60.2	50.1	29.0/110	19.0	13.7	7.2/34.5
<i>LinoA</i>	7.9	2.0	0.17/14.2	6.7	0.66	0.31/6.3
<i>OleiA</i>	11.8	2.3	1.1/22.4	3.0	1.6	0.66/3.2
<i>SteaA</i>	26.6	22.7	12.6/43.1	10.7	7.3	5.4/18.7
<i>MethT</i>	0.19	0.13	0.05/0.27	-	-	-
<i>MethE</i>	0.31	0.26	0.12/0.39	-	-	-
<i>PinoA</i>	1.3	0.96	0.55/2.7	-	-	-
<i>PinicA</i>	0.94	0.56	0.11/2.5	-	-	-

Note: $\text{PM}_{2.5}$ mass and inorganic constituents in $\mu\text{g m}^{-3}$; Organic markers in ng m^{-3}

Table 3: Comparison of data for mean concentrations of secondary biogenic molecular markers

Site	Season/ Fraction	2-Methyl -threitol	2-Methyl -erythritol	2-Methyl -tetrols	Pinonic acid	Pinic acid	Pinonic + Pinic acids	References
NK, London, UK, urban background	Jan-Feb2012/ PM _{2.5}	0.19	0.31	0.50	1.3	0.94	2.3	-
HV (high vegetation site), Berlin, Germany	Feb-Oct2010/ PM ₁₀	0.8	1.4	2.2	13.2	3.9	17.1	Wagener et al., 2012a,b
LV (roadside, low vegetation), Berlin, Germany	Feb-Oct2010/ PM ₁₀	0.6	1.2	1.8	8.4	2.2	10.6	Wagener et al., 2012a,b
RegV (background, regional influence), Berlin, Germany	Feb-Oct2010/ PM ₁₀	0.8	1.2	2.0	15.3	5.6	20.9	Wagener et al., 2012a,b
HV (high vegetation site), Berlin, Germany	Feb-Oct2010/ PM ₁	0.6	1.1	1.7	11.9	3.1	15.0	Wagener et al., 2012a,b
LV (roadside, low vegetation), Berlin, Germany	Feb-Oct2010/ PM ₁	0.3	0.8	1.1	6.9	2.3	9.2	Wagener et al., 2012a,b
Vavihill, background, southern Sweden	2008-2009/ PM ₁₀	-	-	-	3.02	3.03	6.1	Hyder et al., 2012
Hyytiälä, Finland, rural with forests surrounded	summer2004/ PM ₁	5.1	21.2	26.3	-	-	-	Kourtchev et al., 2005
Hyytiälä, Finland, rural with forests surrounded	autumn2004/ PM ₁	0.18	0.29	0.47	-	-	-	Kourtchev et al., 2005
Southeastern US, urban & urban background	June2004	4.8	11.9	16.7	-	-	-	Clements and Seinfeld, 2007
Southeastern US, rural	June2004	1.6	4.9	6.5	-	-	-	Clements and Seinfeld, 2007

Table 4: Source contribution estimates (SCE) ($\mu\text{g m}^{-3}$) and standard deviation (S.D.) for fine particulate OC and PM_{2.5} at NK and HAR from the CMB model

<i>Source Name</i>		OC			PM _{2.5}			OC/PM _{2.5}
		NK ^a	NK ^b	HAR ^a	NK ^a	NK ^b	HAR ^a	
<i>Vegetation</i>	<i>SCE</i>	0.069	0.069	0.11	0.21	0.21	0.35	0.324
	<i>S.D.</i>	0.010	0.010	0.015	0.030	0.030	0.048	-
<i>Woodsmoke</i>	<i>SCE</i>	0.53	0.53	0.64	0.64	0.64	0.76	0.836
	<i>S.D.</i>	0.11	0.11	0.14	0.14	0.14	0.16	-
<i>Natural Gas</i>	<i>SCE</i>	0.046	0.046	0.042	0.054	0.054	0.049	0.849
	<i>S.D.</i>	0.009	0.009	0.007	0.011	0.011	0.008	-
<i>Dust/Soil</i>	<i>SCE</i>	0.044	0.044	0.016	0.34	0.34	0.12	0.131
	<i>S.D.</i>	0.036	0.036	0.015	0.27	0.27	0.11	-
<i>Coal</i>	<i>SCE</i>	0.074	0.074	0.041	0.17	0.17	0.094	0.432
	<i>S.D.</i>	0.020	0.020	0.009	0.046	0.046	0.021	-
<i>Food Cooking</i>	<i>SCE</i>	0.39	0.39	0.072	0.69	0.69	0.13	0.566
	<i>S.D.</i>	0.066	0.066	0.013	0.12	0.12	0.023	-
<i>Total Traffic</i>	<i>SCE</i>	0.73	0.73	0.36	1.26	1.26	0.61	0.579
	<i>S.D.</i>	0.35	0.35	0.16	0.86	0.86	0.29	-
<i>Biogenic Secondary</i>	<i>SCE</i>	-	0.90	-	-	1.63	-	0.556
	<i>S.D.</i>	-	0.17	-	-	0.31	-	-
<i>Other-OC/OM</i>	<i>SCE</i>	1.64	0.73	1.02	2.95	1.32	1.84	0.556
	<i>S.D.</i>	-	-	-	-	-	-	-
<i>Sea Salt</i>	<i>SCE</i>	-	-	-	1.1	1.1	0.82	-
	<i>S.D.</i>	-	-	-	0.020	0.020	0.020	-
<i>Ammonium Sulphate</i>	<i>SCE</i>	-	-	-	2.2	2.2	2.1	-
	<i>S.D.</i>	-	-	-	0.028	0.028	0.028	-
<i>Ammonium Nitrate</i>	<i>SCE</i>	-	-	-	5.8	5.8	4.1	-
	<i>S.D.</i>	-	-	-	0.072	0.072	0.072	-
<i>Measured OC/PM_{2.5}</i>	<i>Mass</i>	3.5	3.5	2.3	15.7	15.7	11.0	-

Note: Figures in bold were not statistically different from zero; *a* - Modelled without biogenic secondary source profile; *b* - Modelled with biogenic secondary source profile; *c* - Conversion factor

Table 5: RMA regression results for CMB and AMS-PMF organic matter estimates and related variables

Pair of Variables	Slope	NK2012	
		Intercept	r^2
<i>AMS-SFOA/CMB-WS</i>	2.81	-0.69	0.75
<i>AMS-SFOA/CMB-(WS+Coal)</i>	2.40	-0.64	0.78
<i>AMS-COA/CMB Food Cooking</i>	2.24	-0.33	0.80
<i>AMS-HOA/CMB Traffic</i>	1.24	-0.32	0.80
<i>AMS (SFOA+COA+HOA)/CMB (WS+Coal+Food+Traffic)</i>	1.85	-1.14	0.91
<i>AMS-OOA/CMB Other-OM</i>	0.39	-0.06	0.69
<i>AMS-OOA/Sec-OC</i>	0.81	0.18	0.64
<i>AMS OOA/Measured SO₄²⁻</i>	0.71	-0.27	0.79
<i>AMS OOA/Measured NO₃⁻</i>	0.33	-0.16	0.80
<i>CMB Other-OC/Sec-OC</i>	0.99	0.66	0.84
<i>CMB Other-OM/Measured SO₄²⁻</i>	1.13	0.78	0.39
<i>CMB Other-OM/Measured NO₃⁻</i>	0.58	0.84	0.42

Table 6: Comparison of apportionment of organic matter by the AMS-PMF and CMB method at NK ($\mu\text{g m}^{-3}$)

Source	CMB Estimate	AMS-PMF Estimate
Woodsmoke	0.85	1.63
Coal	0.12	-
Food cooking	0.56	0.87
Traffic	0.98	0.86
Vegetation	0.11	-
Natural gas	0.055	-
Dust/soil	0.10	-
Primary-total	2.77	3.36
Secondary	2.92	0.99
TOTAL	5.69	4.35

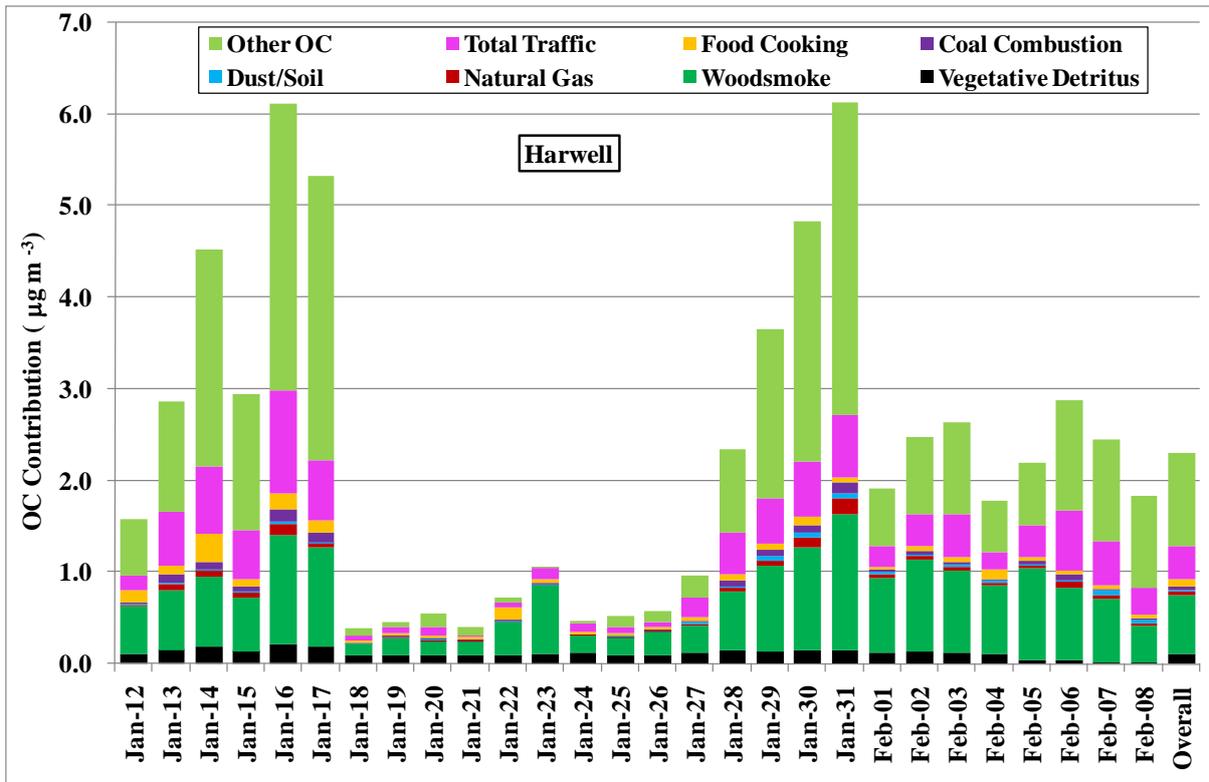
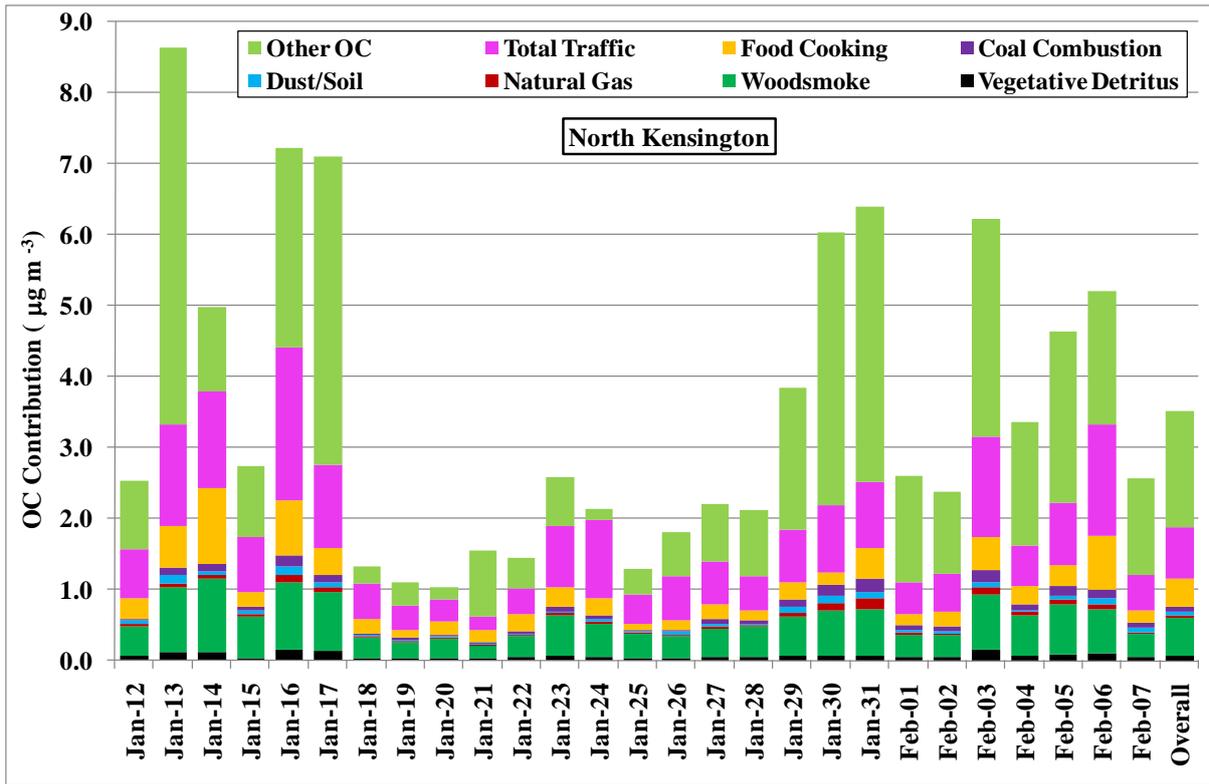


Figure 1: Daily OC source contributions at NK and HAR (without secondary biogenic component identified separately)

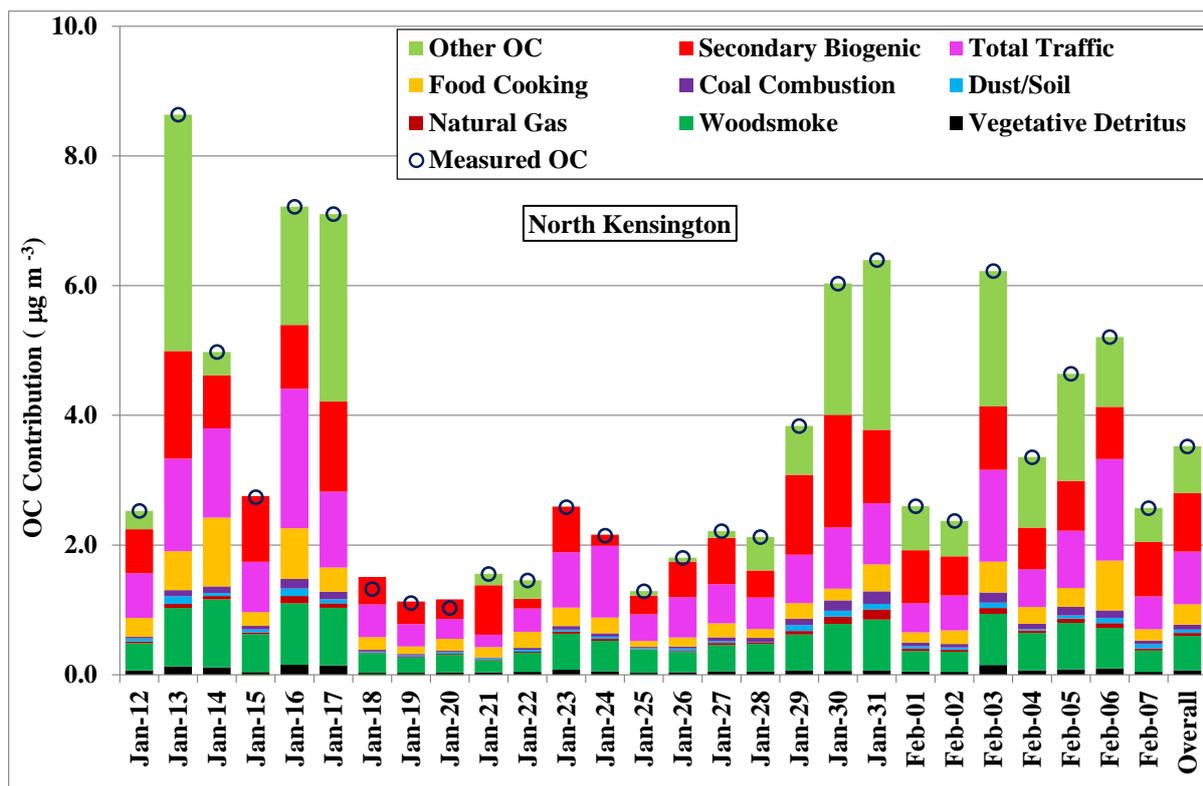


Figure 2: Daily OC source contribution estimates with secondary biogenic components at NK

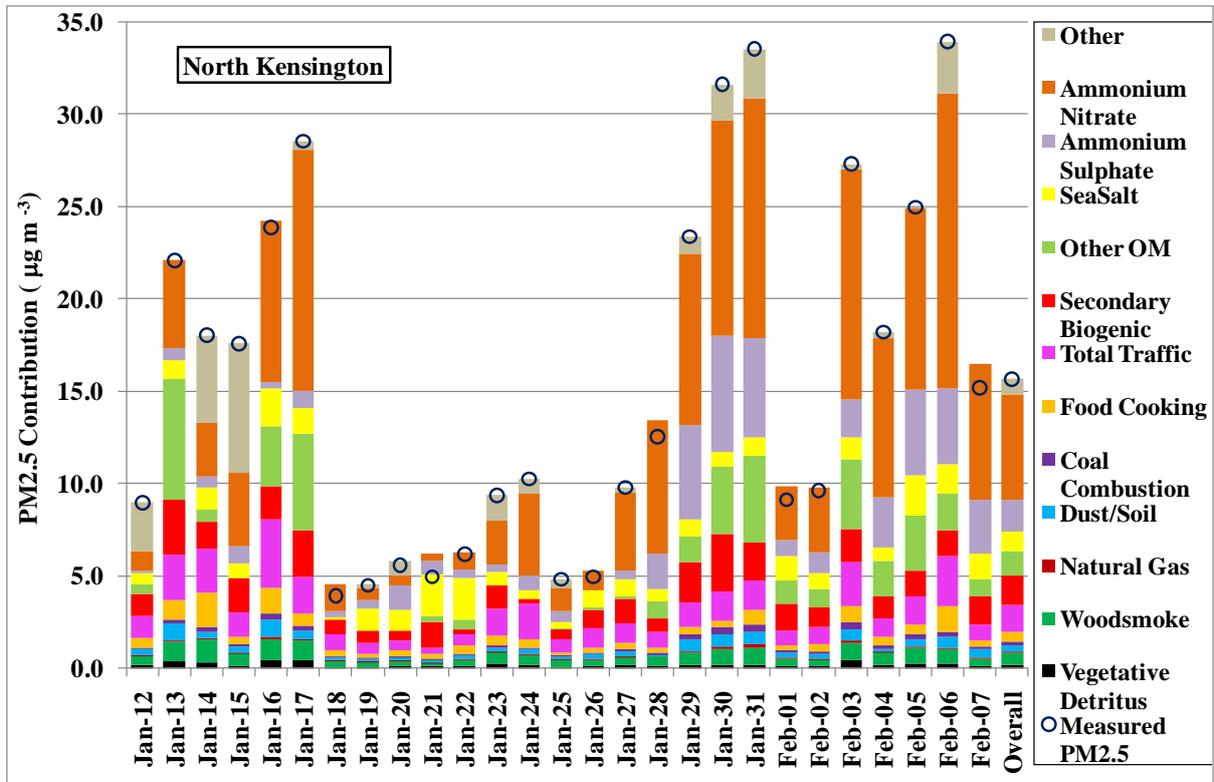


Figure 3: Daily PM_{2.5} source contribution estimates with secondary biogenic components at NK

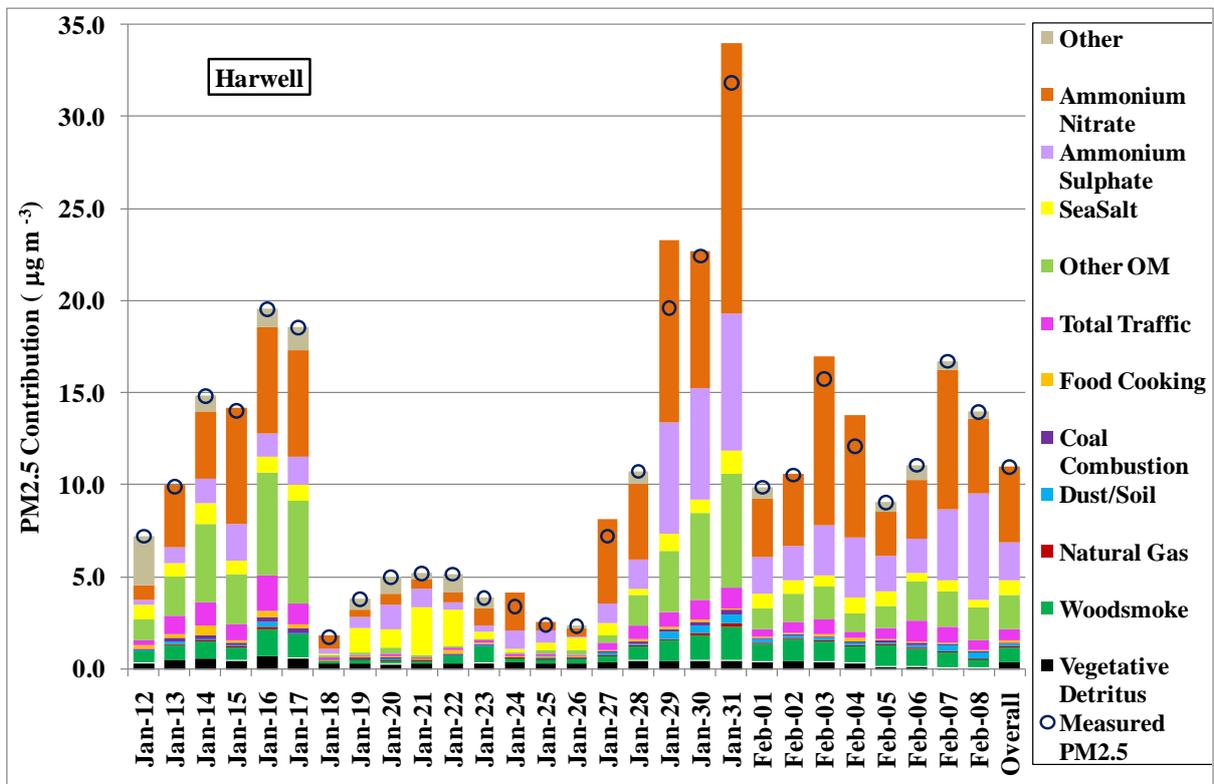
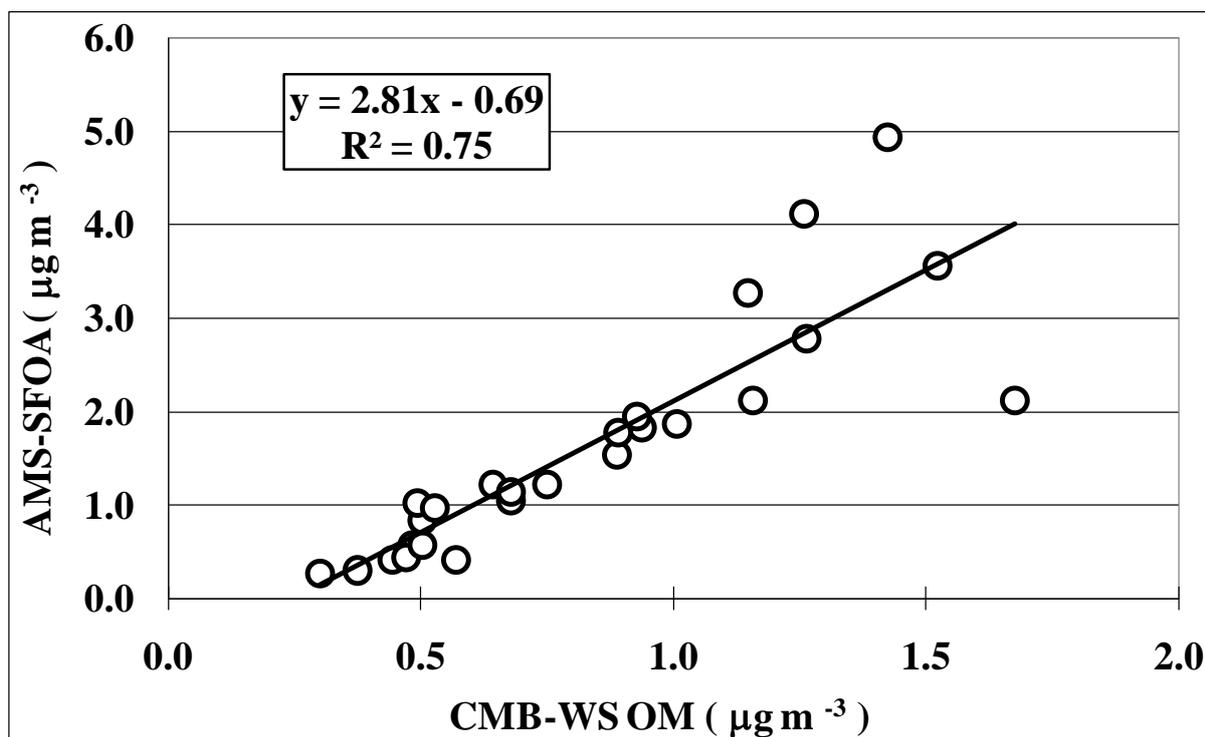
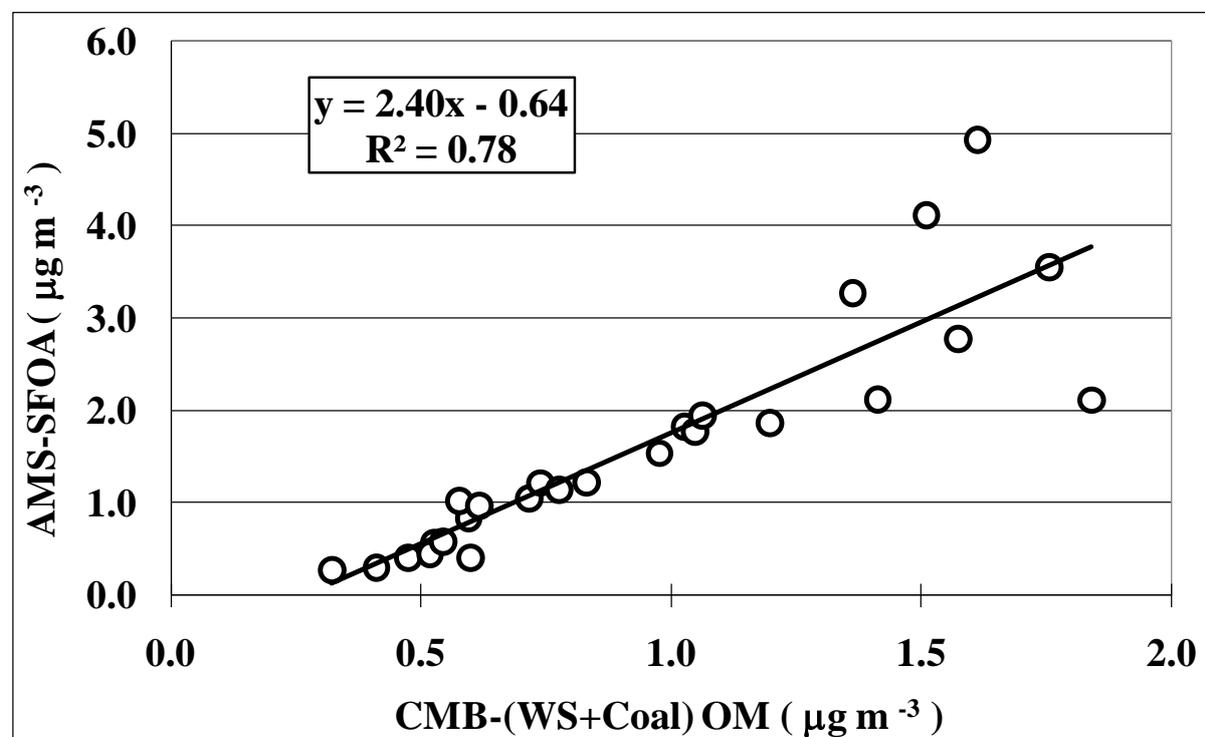


Figure 4: Daily PM_{2.5} source contribution estimates at HAR

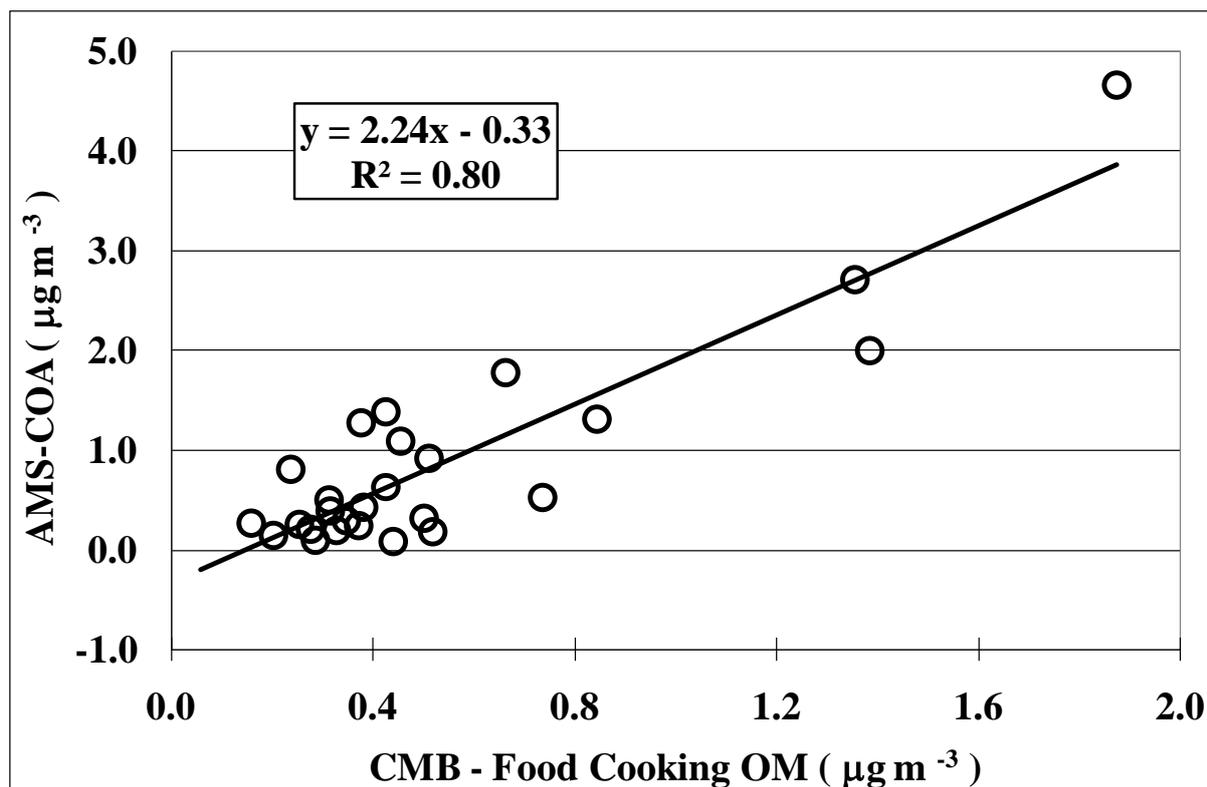
(a)



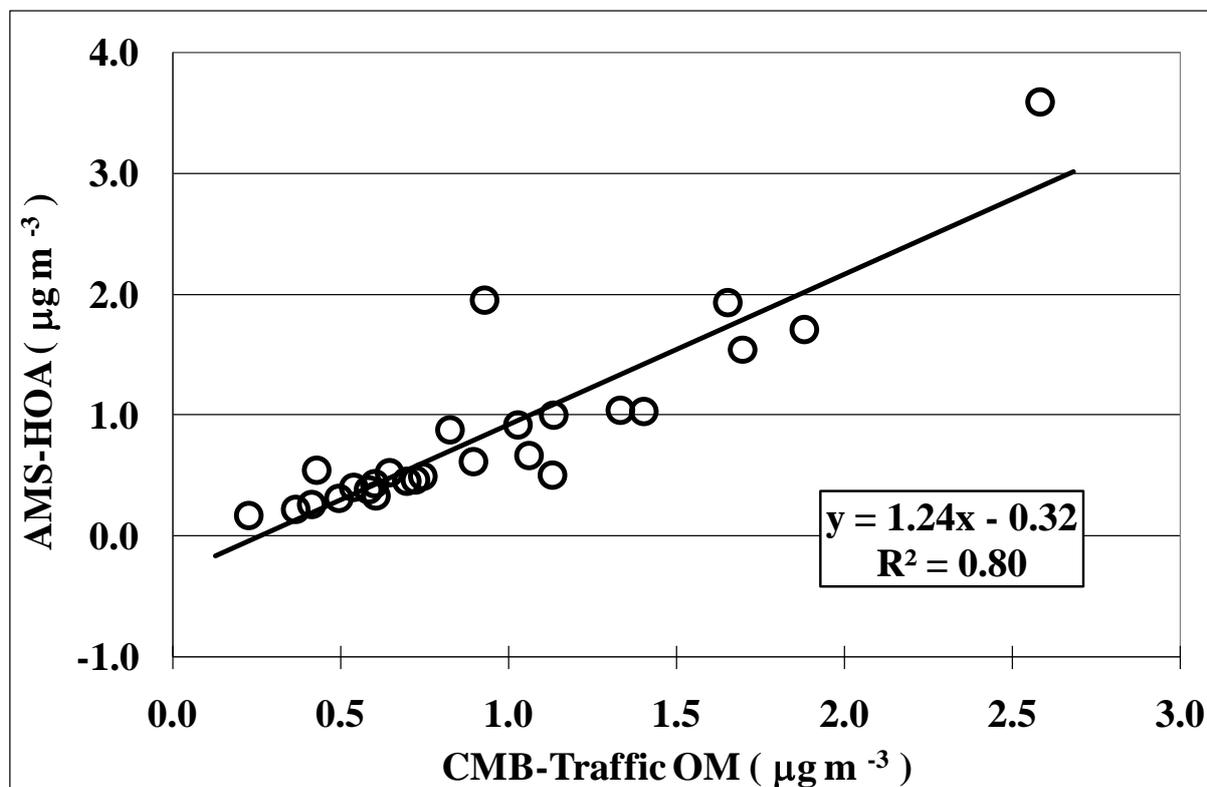
(b)



(c)



(d)



(e)

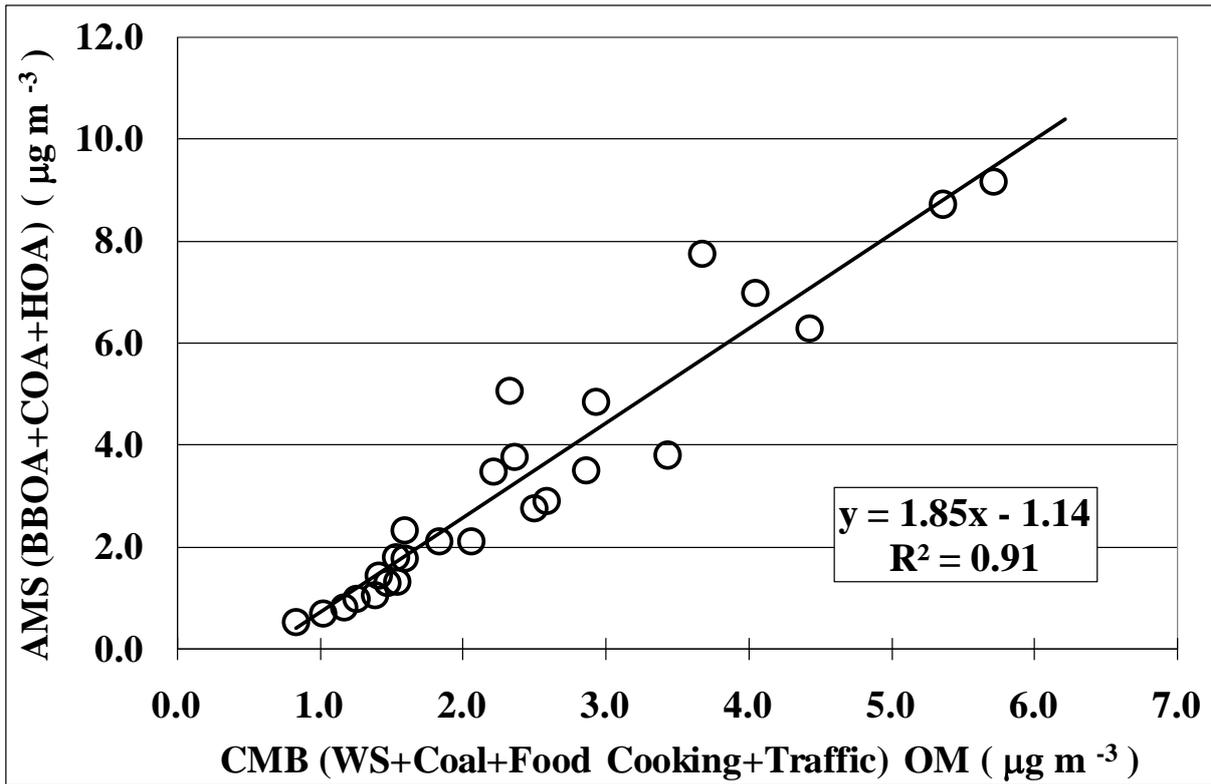


Figure 5: Primary component comparison between the AMS-PMF and CMB methods at NK

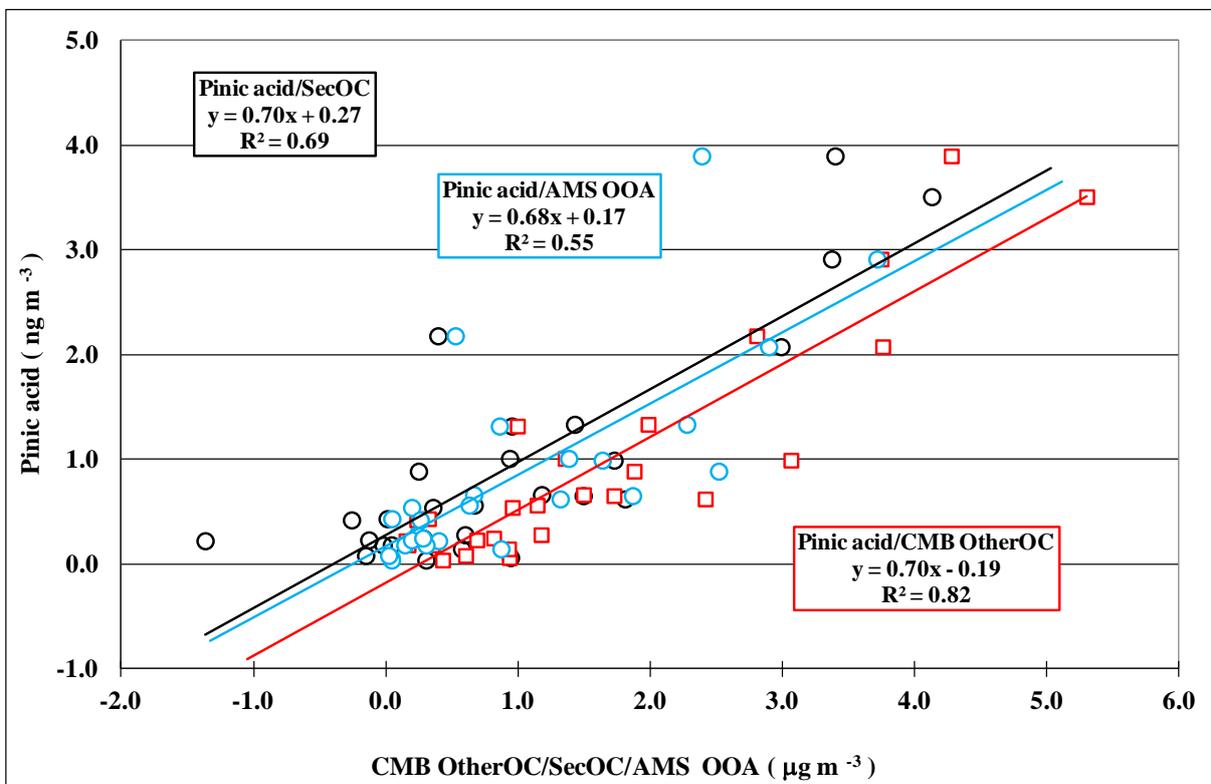


Figure 6: Secondary component comparison at NK showing relationships between pinic acid concentrations and estimated secondary OC from the elemental carbon tracer method (Sec-OC) and the CMB model (Other-OC), and with the AMS OOA factor