



*Supplement of*

## **Importance of aerosol composition and mixing state for cloud droplet activation in the high Arctic**

**C. Leck and E. Svensson**

*Correspondence to:* C. Leck (lina@misu.su.se)

# 1 Importance of aerosol composition and mixing state for 2 cloud droplet activation in the high Arctic

3 C. Leck<sup>1</sup> and E. Svensson<sup>1</sup>

4 [1]{Department of Meteorology and Bert Bolin Centre for Climate Research, Stockholm  
5 University, SE-10691 Stockholm, Sweden}

6 Correspondence to: C. Leck ([lina@misu.su.se](mailto:lina@misu.su.se))

7

## 8 Supplementary material

9 We developed a new implementation (written in Matlab) of the adiabatic parcel model  
10 described by Pruppacher and Klett (1997). It consists of essentially the same equations as the  
11 model presented in Leitch et al. (1986), with the difference that we use the full implicit  
12 dynamic diffusional growth equation for particles in each size bin:

13

$$14 \quad r \frac{dr}{dt} = \frac{D' M_w e_{sat,w}}{\rho_s RT} \left( S - \frac{1}{1+\delta} \exp \left[ \frac{L_e M_w}{RT} \left( \frac{\delta}{1+\delta} \right) + \frac{2M_w \sigma_{s/a}}{RT(1+\delta)\rho_w r_i} - \frac{v\Phi_s \epsilon_m M_w \rho_N r_N^3}{M_s^2 w (r^3 - r_N^3)} \right] \right) \quad (1),$$

15 where:

16  $r$  - radius of particle

17  $t$  - time

18  $M_w$  - molecular weight of water

19  $e_{sat,w}$  - saturation pressure of water vapor above a water surface

20  $\rho_s$  - density of the droplet solution

21  $R$  - universal gas constant

22  $T$  - temperature of the gas surroundings

23  $S$  - supersaturation

24  $L_e$  - latent heat of evaporation of water

1  $\sigma_{s/a}$  - surface tension at solution/air interface

2  $\rho_w$  - density of pure water

3  $v$  - ion number of the salt

4  $\Phi_s$  - osmotic coefficient of droplet solution

5  $\varepsilon_m$  - mass fraction of soluble material

6  $\rho_N$  - density of dry nucleus

7  $M_s$  - molecular mass of nucleus material

8  $r_{N,1}$  - radius of dry nucleus

9

10  $\delta = T_d/T - 1$ , where  $T_d$  is the droplet temperature, gives a measure of the heating or cooling  
11 caused by latent heat released when the droplet grows or shrinks. It is given by:

12

13 
$$\delta = \frac{L_e \rho_s}{TK'} r \frac{dr}{dt} \quad (2)$$

14

15 The  $D'$  and  $K'$  are modified diffusion and thermal conduction coefficients:

16 
$$D' = \frac{D}{\left[ \frac{r}{r + \lambda} + \frac{D}{r\alpha_c} \sqrt{\frac{2\pi M_w}{RT}} \right]} \quad (3)$$

17

18 
$$K' = \frac{k}{\left[ \frac{r}{r + \lambda} + \frac{k}{r\alpha_t \rho_{pa}} \sqrt{\frac{2\pi M_A}{RT}} \right]} \quad (4)$$

19

20 where:

21  $D$  - pressure and temperature dependent diffusivity of water vapor in air

- 1  $\lambda$  - constant ( $1.2 \cdot 10^{-7}$  m)  
 2  $\alpha_c$  - condensation accommodation coefficient  
 3  $k$  - heat conductivity  
 4  $\alpha_t$  - condensation accommodation coefficient  
 5  $M_A$  - mean molecular mass of air

6

7 We simplify the Raoult term in the particle growth equation by identifying it as the water  
 8 activity.

9

$$\hat{M}\Phi = -\ln(a_w)$$

$$10 \frac{\nu\Phi_s \varepsilon_m M_w \rho_N r_N^3}{M_s \rho_w (r^3 - r_N^3)} = \nu M \quad (5)$$

11 Recent research has shown that both of the accommodation coefficients are likely to have  
 12 values close to unity for pure water surfaces (e.g. Winkler et al., 2004, Winkler et. al., 2006,  
 13 Morita et al., 2004).

14 Due to the dependence of  $\delta$  on  $dr/dt$ , the system of ordinary differential equations is implicit,  
 15 and hence we need to provide both starting values of  $T$ ,  $p$ ,  $S$ ,  $w$  and  $r_i$  and their time  
 16 derivatives. There are a few numerical caveats in the model, such as particle sizes shrinking  
 17 below the dry size or even becoming negative. Unphysical developments like these are  
 18 avoided by artificially setting the particle size derivatives positive when the droplet size  
 19 approaches the size of the dry particle.

20 In contrast to the model of Leitch et al. (1986) we make no assumptions of log-normal  
 21 aerosol distributions. Instead, we directly as input use the measured size distributions by the  
 22 TDMPS. The lower cut-off for the chemical information is 22 nm in diameter, which is well  
 23 below the size required to act as CCN in the conditions of his study. Chemical mass  
 24 concentration data from LPI impactor samples are interpolated on to the TDMPS size bins  
 25 and thereafter converted to number concentration assuming spherical particles.

1

$$r_i \frac{dr_i}{dt} = \frac{D' M_w e_{sat,w}}{\rho_s RT} \left( S - \frac{1}{1+\delta} \exp \left[ \frac{L_e M_w}{RT} \left( \frac{\delta}{1+\delta} \right) + \frac{2M_w \sigma_{s/a}}{RT(1+\delta)\rho_w r_i} - \frac{\nu \Phi_s \varepsilon_m M_w \rho_N r_{N,i}^3}{M_s \rho_w (r_i^3 - r_{N,j}^3)} \right] \right) \quad (6)$$

3

4 For an isolated air parcel ascending adiabatically with the vertical speed V, the temperature T,  
5 pressure p, supersaturation S and liquid water content are described by the equations:

6

7

8

$$-\frac{dT}{dt} = \frac{gV}{c_{pa}} + \frac{L_e}{c_{pa}} \frac{dw}{dt} \quad (7)$$

10

$$\frac{dp}{dt} = -\frac{gpV}{R_a T} \quad (8)$$

12

$$\frac{dS}{dt} = \frac{p}{\varepsilon e_{sat,w}} - (1+S) \left[ \frac{L_e}{R_a T^2} \frac{dT}{dt} + \frac{gV}{R_a T} \right] \quad (9)$$

14

$$\frac{dw}{dt} = \sum_i 4\pi n_i r_i^2 \frac{dr_i}{dt} \quad (8)$$

16

17 In this study, however, we use V=0 and artificially change S in the way the CCN counter  
18 does.