Air Quality and Atmospheric Deposition in the Eastern US: 20 Years of Change

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14	Abstract. Data collected in the eastern US between 1990 and 2009 at 34 paired dry and wet
15	monitoring sites are examined. A goal is to evaluate the air quality impacts occurring between
16	1990 and 2009 that are associated with concurrent legislatively-mandated changes in emissions.
17	Four 5-year periods, 1990–1994 (P1), 1995–1999 (P2), 2000–2004 (P3), and 2005–2009 (P4) are
18	considered, with a primary focus on P1-to-P4 changes. Results suggest that legislatively-
19	mandated air pollution mitigation strategies have been successful in improving air quality and
20	reducing atmospheric deposition in the eastern US.
21	Respective P1-to-P4 reductions of estimated sulfur dioxide (SO ₂) and nitrogen oxides
22	(NOx) emissions in the eastern US are 50% and 42%. Corresponding behavior of the following
23	metrics associated with these emissions reductions are examined: monitored atmospheric
24	concentrations of SO ₂ , aerosol sulfate (SO ₄), and oxidized sulfur (S); dry, wet, and total
25	deposition of S; monitored atmospheric concentrations of nitric acid (HNO ₃), aerosol nitrate
26	(NO ₃), and their sum, oxidized nitrogen (OxN); dry, wet, and total deposition of OxN; monitored
27	atmospheric concentration of aerosol ammonium (NH4); dry, wet, and total deposition of NH4;
28	summed monitored atmospheric concentration of oxidized and reduced nitrogen (N); dry, wet,
29	and total deposition of N; wet deposition of hydrogen ion (H ⁺); monitored atmospheric
30	concentration of ozone (O ₃); dry deposition of O ₃ ; and the summed monitored atmospheric
31	concentration of aerosol NO ₃ , SO ₄ , and NH ₄ (CASTNET PM). Other metrics (e.g., ratios of dry

32 to total deposition) are also considered.

33 Selected period-to-period changes of air quality and deposition metrics at site, regional, 34 and seasonal scales are discussed. As an example, despite P1-to-P3 reductions in estimated 35 emissions of both SO₂ and NOx, aerosol NO₃ concentration increased in the east, with 36 widespread wintertime numerical increases in both aerosol NO₃ concentration and CASTNET 37 PM. However, a reversal of this behavior is associated with continuing P3-to-P4 reductions of 38 SO₂ and NOx emissions. Thus, additional P3-to-P4 reductions of these emissions, especially 39 NOx, appear to have made progress in altering the chemical regime of the wintertime eastern US 40 atmosphere so that future emissions reductions and their resulting reductions in aerosol 41 concentrations may no longer be accompanied by sub-linear changes (or actual increases) in 42 CASTNET PM. 43

Keywords: Trends, dry deposition, wet deposition, total deposition, flux, deposition velocity,
concentration, ozone, sulfur dioxide, nitric acid, sulfate, nitrate, ammonium, hydrogen ion,
CASTNET, NADP

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- 49 **1.** Introduction
- 50

51 In the United States (US), pollution control legislation has been established to protect the 52 public health and welfare by reducing emissions of selected pollutants. Both sulfur dioxide (SO₂) 53 and nitrogen dioxide (NO₂) are designated by the US Environmental Protection Agency (EPA) 54 as criteria pollutants, and they contribute to acidic deposition as well as airborne particulate 55 matter (PM). In addition, nitrogen oxides (NOx), the sum of nitric oxide (NO) and NO₂, act as 56 precursors of another criteria pollutant, ozone (O₃).

57 Nationwide, in 1990, 69% of SO₂ emissions came from electric generating units (EGUs); 58 whereas, 82% of NOx emissions came from the sum of EGUs (25%) and mobile sources (57%) 59 (Xing et al., 2013). Sources in the eastern states contributed 76% and 58% of the US total 1990 60 SO₂ and NOx emissions. The Clean Air Act Amendments of 1990 (CAAA) established phased 61 year-round controls for SO₂ and NOx emissions from EGUs that became effective in 1995 for 62 SO_2 and 1996 for NOx (Phase I); while additional controls became effective under Phase II in 63 2000. Over half of the EGUs targeted by the CAAA are located in six states along the Ohio 64 River (source region (SR), Table 1). Beginning in 1999, O₃ season (i.e., summertime) NOx 65 controls focusing on EGUs became effective in selected eastern states under the Ozone Transport 66 Commission (OTC) and were superceded in 2003 by the NOx State Implementation Plan (SIP) Call. The affected states chose to meet their mandatory reductions by participating in the NOx 67 68 Budget Trading Program (BTP), a market-based cap and trade program for EGUs and large 69 industrial units. The BTP was replaced in 2009 by the Clean Air Interstate Rule (CAIR). Various 70 mobile source NOx emissions control programs also began in the mid- to late 1990s. Fully 71 implemented in 2006, the Tier 2 Gasoline Sulfur Program targeted both SO₂ and NOx emissions 72 from mobile sources. These mobile source programs have a cumulative effect of reducing 73 emissions over time as the mobile fleet is replaced. 74 Figure 1 shows estimates of emissions from 1990 to 2010 for both the total US and the

eastern states (Xing et al., 2013). Both SO₂ and NOx emissions show weak reductions between
1990 and 1994. There is a sharp reduction in SO₂ emissions in 1995, when Phase I controls
became effective. Although additional reductions of both SO₂ and NOx emissions begin to
appear in 1999, they are apparent in later years (e.g., after 2005 for SO₂). Although a substantial
portion of the long-term emissions reductions since 1990 has resulted from legislatively driven

emissions controls, economic factors (e.g., coal-to-natural gas fuel switching and recessioninduced reduction in demand for electricity and transportation fuel) played a sizeable role
between 2008 and 2009 (Lu et al., 2012).

83 A major objective of the current study is to examine the monitored and estimated air 84 quality and deposition behaviors in the eastern US (E) between 1990 and 2009 associated with 85 changes in emissions. The time period of interest is divided into four 5-year periods: 1990–1994 86 (period one, P1), 1995–1999 (period two, P2), 2000–2004 (period three, P3), and 2005–2009 87 (period four, P4). Focusing on these 5-year periods, average P1 US SO₂ and NOx emissions have 88 declined by 19% and 6% by P2, by 33% and 19% by P3, and by 49% and 37% by P4 (Xing et 89 al., 2013). In the current study, the east is taken to represent all states east of the Mississippi 90 River and is divided into three geographical regions: midwest (MW), northeast (NE), and south 91 (SO) (Fig. 2). Mean period-specific emissions in these regions are illustrated for each 5-year 92 period relative to P1 in Fig. 3.

93 This paper builds on the previous work (Sickles and Shadwick, 2007a and b), hereafter 94 known as SS07a and SS07b. In SS07a seasonal and regional distributions of concentration and 95 deposition of selected pollutants, were examined for the 1990s (i.e., P1 and P2) at eastern rural 96 US Clean Air Status and Trends Network (CASTNET) and paired National Atmospheric 97 Deposition Program (NADP) monitoring sites. In SS07b temporal changes in air quality and 98 deposition in the eastern states between the early 1990s and early 2000s (i.e., P1-to-P3) were 99 described and discussed. The record is expanded in the current paper to include a fourth 5-year 100 period. Here, emphasis is on temporal changes occurring between the early 1990s and late 2000s 101 (i.e., P1-to-P4) as new and continuing pollution controls have become mature and their impacts 102 apparent. Results are displayed in a consistent manner to previous work (SS07a and b) to 103 facilitate comparison. See Table 2 for definitions of terms and pollutant metrics used in the text, 104 tables, and figures. 105 106

107 **2.** Approach

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109 Monitoring data collected in the east between 1990 and 2009 at 34 CASTNET

110 [http://epa.gov/castnet/javaweb/index.html] dry monitoring sites and paired NADP

111 [http://nadp.sws.uiuc.edu/] wet monitoring sites provide a sufficiently complete 20-year record to 112 permit examination. Weekly concentrations of gaseous SO_2 and nitric acid (HNO₃), and aerosol 113 sulfate (SO₄), nitrate (NO₃), and ammonium (NH₄) (using 3-stage filter packs) and hourly 114 concentrations of O_3 (using continuous monitors) are determined at 10 m. Dry deposition of 115 airborne species is estimated using the inferential model by multiplying modeled deposition 116 velocities (Vd) by the monitored atmospheric concentration for the gas or aerosol species of 117 interest. Deposition velocities for gases and aerosols are estimated using the multilayer model 118 (MLM) (Meyers et al., 1998; Wesely et al., 1985). Monitoring data are composed from reported 119 weekly (hourly for O_3) measurements into estimates of means for year, site, and season. These 120 year-site-season means, computed for atmospheric concentration, Vd, precipitation rate (PR), 121 and dry, wet and total deposition, are averaged to period-site-season means and used to examine 122 differences between 5-year periods for seasons, sites, and predefined groupings of sites. The 123 data, methodologies, data quality (i.e., precision and accuracy), covariance of concentration and Vd, impacts of non-ideal landscape features, adjustments for sampling artifacts, and data 124 125 completeness criteria for accepting mean estimates of measurement data are described in SS07a. 126 Monitoring site locations, 10 in the MW and NE, and 14 in the SO, are identified in Fig.

2 and Table 1. These 34 sites provide sparse coverage, especially in the southern part of the SO
and the northeastern part of the NE. Three additional environmental groupings of monitoring
sites are also defined as subsets of the eastern sites: background (BKG); high elevation (HE); and
SR. These sites are chosen to reveal distinctions in pollutant behavior characteristic of their
environments (Table 1 and SS07a).

132 CASTNET monitoring locations were initially sited to avoid the influence of major cities,
133 highways, local activities, and point and area sources of pollutants and to be regionally
134 representative (i.e., similar to a characteristic 80 by 80 km area surrounding the site) (Clarke et
135 al., 1997; Baumgardner et al., 2002). In the current study, most of the CASTNET sites are
136 located within 50 km of a paired NADP/NTN wet deposition site.

137 Collocated precision estimates (as CVs) of weekly CASTNET concentration, Vd, and dry
138 deposition at eastern sites are within 5% to 10% for the major species (Clarke et al., 1997;
139 Sickles and Shadwick, 2002). Corresponding precision estimates are: 5% for weekly NADP PR,
140 wet S (i.e., wet deposition of sulfate, see Table 2), and wet OxN; 10% for wet H⁺; and 15% for
141 wet NH₄ (Nilles et al., 1994). These estimates for weekly network results are likely to be

142 conservative for corresponding longer-duration 5-year period results.

Total oxidized nitrogen (NO_Y) includes the primary NOx species $(NO \text{ and } NO_2)$ plus the secondary species $(HNO_3, \text{ nitrous acid } (HNO_2), \text{ aerosol } NO_3, \text{ peroxyacetylnitrate } (PAN), \text{ as well}$ $as other oxidized organic nitrogen species}). Total reduced nitrogen <math>(NHx)$ includes both gaseous ammonia (NH_3) and aerosol NH_4 . In the current study, only two of the NO_Y species, HNO_3 and aerosol NO_3 , (their sum is OxN), and only one of the NHx species, aerosol NH_4 , are monitored. As a result, reported estimates of dry deposition of OxN and NH_4 underestimate corresponding values for NO_Y and NHx.

150 The contribution of coarse (i.e., $>2.5 \mu m$) particles to total particulate deposition likely 151 depends on location and proximity to sources of crustal dusts, road salt, and marine aerosols. 152 Since the contribution of coarse particles is ignored in the MLM, where all particles are modeled 153 as fine aerosol SO₄ particles, current estimates of particulate dry deposition are biased low. 154 However, the non-size selective CASTNET filter pack sampler likely oversamples the fine tail of 155 coarse particle distribution. Although this may not be too important for aerosol SO₄ and NH₄, 156 which occur mainly as fine particles; it would underestimate the contribution of dry deposition of 157 aerosol NO₃, where appreciable coarse aerosol NO₃ can occur (Wolff, 1984; Lefer and Talbot, 158 2001).

159 As noted in SS07a (and references therein), use of the CASTNET filter pack sampler 160 introduces sampling artifacts, especially to nitrogen-containing species. Findings from Sickles II 161 and Shadwick (2008) as well as wind tunnel and modeling studies (unpublished) of particle 162 sampling characteristics of the non-size selective open-faced CASTNET filter pack inside its 163 weather shield over a range of wind speeds suggest appreciable collection of both fine ($<2.5 \mu m$) 164 and coarse particles on the first filter (Teflon) of the filter pack. Although both aerosol SO₄ and 165 NH₄ exist mainly in the fine fraction (<1 µm) year round; aerosol NO₃ is multi-modal, and in 166 non-marine environments exists mainly (>75%) in the fine fraction in cool seasons and mainly as 167 larger (>4 µm) particles during warm seasons (Wolff, 1984; Lefer and Talbot, 2001; Zhang et al., 168 2008; Lee et al., 2008). Sampling artifacts in opposing directions may arise through possible 169 loss of aerosol NH₄NO₃ during sampling and by aspiration of the fine tail of coarse aerosol NO₃. 170 Aerosol NH₄NO₃ (nominally $<2.5 \mu$ m) collected on the Teflon filter may be lost during sampling 171 by volatilization into gaseous HNO₃ and NH₃, driven by departures from equilibrium 172 (exaggerated at higher temperatures) and/or by reaction with strong acids. Results from

173 comparison studies (Sickles II and Shadwick, 2002; Sickles II and Shadwick 2008; Lee et al., 174 2008; Zhang et al., 2009) are consistent with loss of NH_4NO_3 from the Teflon filter followed by 175 collection of volatilized HNO₃ on the downstream nylon filter of the CASTNET filter pack 176 during warm season sampling and with minimal volatilization loss during cool season sampling. 177 Coarse alkaline particles (e.g., sea/road salt or soil/mineral dust) may contain aerosol NO₃ 178 (possibly formed by scavenging HNO₃ while airborne), and to the extent that they are collected, 179 these larger particles may provide a high bias to CASTNET aerosol NO₃ relative to its fine 180 $(PM_{2,5})$ contribution. Some of these larger particles (if unreacted and retained on the Teflon 181 filter) may scavenge airborne HNO₃ from sampled air during week-long sampling periods, also 182 providing a high bias to aerosol NO₃ and a low bias to HNO₃.

183 Some of these concepts are illustrated by the performance of CASTNET and IMPROVE 184 (with 2.5 µm size selective inlet) samplers collocated at 10 eastern monitoring sites (Sickles II 185 and Shadwick 2008). At the eight non-marine sites good agreement was found for aerosol SO₄ 186 (CASTNET was biased high by 4 to 7%), and more variability was found for aerosol NO₃ 187 (CASTNET ranged between 25% higher and 39% lower). Seasonally, aerosol NO₃ showed 188 median relative biases ranging between 5 and -10% for all seasons, except summer, where the 189 IMPROVE aerosol NO₃ exceeded the corresponding CASTNET values by >40% (this was at very low concentrations where the median bias was 0.07 μ g m⁻³). Finally, although CASTNET 190 191 and IMPROVE aerosol NO₃ displayed mixed comparisons at eight inland sites, CASTNET sites 192 showed consistently higher aerosol NO₃ (40 - 100%) at the two marine-influenced sites, which 193 was attributed to over-sampling the fine tail of coarse sea salt aerosols by the CASTNET 194 sampler.

195 Thus, in the current study, minimal CASTNET sampling bias is expected for aerosol SO₄ 196 and NH4, which occur mainly as fine particles. However, HNO3 and aerosol NO3 as well as other 197 metrics derived from them are subject to sampling biases that act in opposing directions. 198 Volatilization of collected NH₄NO₃ during sampling could lead to a high bias of HNO₃ and a low 199 bias of aerosol NO₃. In contrast, the non-size selective sampler inlet could lead to an 200 oversampling of the fine tail of coarse aerosol NO₃ with a high bias of aerosol NO₃ and a 201 potentially low bias of HNO_3 . The impacts of these biases are expected to be minimal during 202 cool season sampling at non-marine sites. As a result, in the current study HNO₃, aerosol NO₃, 203 and metrics derived from them (e.g., CASTNET PM, Dry HNO₃, Dry NO₃, RNO₃ and other

204 ratios of concentration and deposition) should be viewed with caution, especially during summer. 205 Uncertainties in Vd, rather than concentration, likely limit the accuracy of dry deposition 206 estimates (Hicks et al., 1991; Lovett and Lindberg, 1993). Accuracy of Vd estimates has not 207 been well established. Based on comparisons with field measurements, the MLM may provide 208 fairly accurate estimates of Vd for SO₂, O₃, and HNO₃ to many surfaces (e.g., crops (Meyers et 209 al., 1998)); however, Vd for SO₂ to forested canopies may be underestimated by $\approx 35\%$ 210 (Finkelstein et al., 2000). Recent comparison at the same site of weekly CASTNET MLM Vd 211 estimates with those from a similar model used by the Canadian Air and Precipitation 212 Monitoring Network (CAPMoN) show CASTNET Vd to be low (relative to CAPMoN) by 213 approximately 50% for SO₂ and 35% for HNO₃; differences of hourly estimates between the two 214 models could be as large as a factor of two to three (Schwede et al., 2011). Annual average 215 small-scale spatial variability (0.5 km) of model-estimated Vd was estimated at 10% for SO₂ and 216 O₃, 30% for aerosol SO₄, and 40% for HNO₃ (Brook et al., 1997). Uncertainty of the inferential 217 model has been estimated at 25% for O₃, 30% for SO₂ and \geq 40% for HNO₃ and particles (Clarke 218 et al., 1997). These estimates may be appropriate for areas with ideal terrain and uniform land 219 use; however, nonideal landscape features (e.g., irregular terrain and/or forested canopies) likely 220 contribute low biases to concentration, estimated Vd, and dry deposition estimates, and increase 221 their resulting uncertainty (Clarke et al., 1997; Hicks, 2006). The distribution of influential 222 landscape features across sites suggests that their impacts result in the greatest underestimates at 223 sites in the south and northeast geographical regions, and that their impacts should be smaller at 224 sites in the midwestern grouping. Much of the above evidence suggests that CASTNET Vd 225 estimates, while uncertain, likely provide conservative lower bound estimates of dry deposition. 226 Internetwork comparisons of annual wet deposition have been conducted for the periods 227 1986-1993 (Sirois et al., 2000) and 1995-2004 (Wetherbee et al., 2010) at collocated NADP and 228 CAPMoN sites. NADP results for precipitation depth and wet deposition of H⁺, NH₄, S, and

OxN appear to be biased low relative to CAPMoN by less than 30%. Although there are nostandards for determining accuracy, current NADP results are possibly biased low and likely

accurate within 25%.

The east has large heterogeneities in its spatial distribution of land cover (i.e., crops, forests, and suburban), of terrain (i.e., flat, rolling, complex, and montane) and of pollution sources (e.g., EGU sources in the Ohio Valley and mobile sources in urban centers). As a result, 235 it is unlikely that the sparse coverage of CASTNET (i.e., 34 sites) can provide truly accurate 236 regional representations. This is supported by a recent modeling study of wet deposition (Sickles 237 et al., 2009) where current study site locations were found to over represent geographical areas 238 subject to both high emissions and wet deposition, yielding overestimations of corresponding 239 regional values by 5 to 25%. Nevertheless, for the current study, values representing areas with 240 common geographical and environmental characteristics are approximated as the means of 241 values from sites associated with each classification. It is assumed that these estimates are useful 242 as rough indices of their respective geographical regions and environmental groupings.

As indicated above, the airborne concentration, estimated Vd, and dry and wet deposition estimates may be subject to sizeable uncertainties. However, assuming constant bias, their precision to a first approximation is expected to control the uncertainty associated with the current examination of period-to-period (e.g., P1-to-P4) temporal changes.

247 Data have been composed from reported measurements into estimates of means for year, 248 site, and season. Year-site-season means were computed for atmospheric concentration, Vd, PR, 249 and dry, wet and total deposition. Consistent with previous work (SS07a and b) these values, 250 averaged to period-site-season means, are used to examine differences between 5-year emission 251 periods for seasons, sites, and predefined site groupings. Seasons (e.g., data for winter (W) 1990 252 includes data from December 1989 through February 1990; spring (Sp); summer (Su); and fall 253 (F)) are used to define the years in periods. Five-year periods have been chosen to coincide with 254 emission patterns (noted earlier), to minimize the impacts of missing data on seasonal means, to 255 reduce the impact of meteorological variability, and to maintain consistency with earlier studies. 256 For example, Table 3 presents averages of year-site-season means averaged over all the years in 257 P4 and all the sites in each specified site grouping. Analogous tabulations for P1, P2, and P3 may 258 be found in SS07a and b.

The approach to data analysis, described in detail in SS07b, uses mean squared error derived from analysis of variance of mean estimates for each pollutant metric to examine periodto-period differences in the monitoring data and metrics derived from them. Use of the term, "significant," implies comparison where a statistical test is satisfied at the p=0.05 significance level, and "NS" implies a non-significant test result (p>0.05). Comparison with no mention of significance, using the term "numerically," implies nominal comparison where one comparator is numerically different from another, without statistical testing. The statistical tests that were performed apply only to absolute differences. Relative differences (changes) are used as a
 convenient construct to enable discussion of the absolute differences.

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3. Results

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272 Means of each species-specific value of P4 atmospheric concentration; Vd; PR; dry, wet, 273 and total deposition; and selected ratios are presented in Table 3 for each site grouping. 274 Corresponding across-site variabilities are given as coefficients of variation (CV). Results are 275 displayed for each grouping of monitoring sites over all seasons and for the east by specific 276 season. P1-to-P4 absolute changes in both native units and relative differences (%) are displayed 277 in Fig. 4 for each geographical region. To facilitate identification of patterns of P1-to-P4 278 changes, significant ($p \le 0.05$) decreases are colored blue, and increases are colored pink. Six 279 sets of period-to-period differences are considered (e.g., P1-to-P2, P1-to-P3, etc.); however, they 280 are not tabulated due to space limitations. Although not shown comprehensively, P3-to-P4 281 differences are discussed in a few instances to contrast with P1-to-P4 differences. Graphical 282 illustrations of period-to-period comparisons of concentration, deposition, and related metrics as 283 well as relative changes by site and region have been created for each of the entries shown in 284 Table 3. Example illustrations for selected airborne concentration (i.e., S, OxN, N and 285 CASTNET PM) and deposition (i.e., total S, total OxN, total N and wet H⁺) are shown for all 286 sites in Figs. 5 and 6.

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288 **3.1.** Deposition velocity and precipitation rate

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Deposition velocity and its analog, PR, may be viewed as velocities at which pollutants are delivered (i.e., deposited via dry and wet deposition) to the earth's surface. Both Vd and PR are strongly influenced by meteorology. One of the reasons for selecting 5-year averaging periods in the current study is to reduce the confounding influences of year-to-year meteorological variability in assessing the impacts of changes in pollutant emissions on resulting ambient atmospheric concentrations, dry deposition, and wet deposition.

296 The variabilities of Vd and PR are fairly low. In the case of Vd, this is indicated by the

297 2010 CASTNET decision [http://epa.gov/castnet/javaweb/docs/CASTNET_Factsheet_2013.pdf]

298 to discontinue meteorological monitoring and use site-specific historical average Vd at all but 299 five monitoring sites (Bowker et al., 2011). In the current study, period-site-season-specific CVs, 300 averaged across all periods, sites, and seasons for Vd for aerosol, O₃, SO₂, and HNO₃ are each 301 <5%, and the corresponding value for PR is <9%. For the east and for each of the three 302 geographic regions, entity-specific relative differences were determined for each of the six 303 possible period-to-period comparisons (i.e., $4 \ge 6 = 24$ relative differences per entity), yielding 304 median absolute relative differences $\leq 6\%$ for Vd of each of the four species and PR. Similar 305 comparisons at the site-specific level ($34 \times 6 = 204$ relative differences per entity) yield median 306 absolute relative differences $\leq 7\%$ for Vd of each of the four species and PR.

307 Since period-to-period changes in Vd are usually small, the behavior of P1-to-P4 changes 308 in species atmospheric concentration and dry deposition are generally similar, and the relative 309 behavior of the dry deposition of a species is often mirrored by that of its airborne concentration. However, there are instances where opposing changes in Vd are sufficient to cause the resulting 310 311 change in species deposition to differ appreciably and sometimes significantly from its 312 corresponding atmospheric concentration. For example, wintertime increases in aerosol Vd yield 313 changes in dry deposition that tend to moderate P1-to-P4 reductions apparent in atmospheric 314 concentration (e.g., dry SO₄, Fig. 4) and exaggerate increases (e.g., dry NO₃).

In the east and in each geographic region, seasonal P1-to-P4 changes in mean PR are not significant ($p\leq0.05$, Fig. 4), and they are usually small ($\leq10\%$), except in winter. These wintertime changes contribute to a small, but significant, 1% increase in PR in the east, 9% increase in the northeast, and 2% decrease in the south. Reduced significance of wintertime changes in wet deposition (e.g., wet S, wet OxN, wet H⁺, and wet N, Fig. 4) appear to be associated with these numerical wintertime P1-to-P4 changes in PR.

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- 322 **3.2.** Oxidized sulfur species
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As is illustrated in Fig. 4, significant P1-to-P4 reductions occur in atmospheric SO₂ concentration and dry deposition, atmospheric aerosol SO₄ concentration, atmospheric S concentration and dry deposition, and total S in each region in each season and over all seasons. For atmospheric SO₂ concentration and dry deposition, the numerically largest significant 328 seasonal absolute reductions usually occur in winter when SO₂ concentration is at its seasonal 329 maximum (largest in northeast); the smallest occur in summer when SO₂ concentration is at its 330 seasonal minimum (smallest in south). Both SO₂ concentration and dry deposition show similar 331 P1-to-P4 relative reductions of approximately 50% across all seasons and regions.

332 As noted earlier, P1-to-P4 changes in atmospheric aerosol SO₄ concentrations show 333 significant reductions in each region both in each season and over all seasons. Seasonally, the 334 numerically largest absolute and relative reductions tend to occur in summer when aerosol SO4 335 concentration is at its seasonal maximum, with regionally uniform summertime absolute reductions of $\approx 3.4 \ \mu g SO_4/m^3$ and relative reductions ranging regionally between 38 and 42%. 336 337 The regionality of aerosol SO₄ concentration, a secondary pollutant, is also illustrated in Table 3 338 by contrasting its across-site variability (CV) with that of its precursor, SO₂ (e.g., 22% versus 339 51%).

340 Dry deposition of aerosol SO₄ also shows significant P1-to-P4 reductions regionally. 341 Overall relative reductions in dry SO₄ are numerically less than the corresponding values for 342 concentration of aerosol SO₄ (i.e., for the east, 29% versus 33%). This pattern is likely due in 343 part to significant period-to-period increases in aerosol Vd, noted earlier. Seasonally, the 344 numerically largest absolute and relative reductions in dry SO₄ tend to occur in summer when 345 aerosol SO₄ concentration is at its seasonal maximum, with an average relative reduction of 37% 346 in summer in the east. Dry SO₄ shows significant reductions in most seasons, except winter, 347 when its concentration is at a seasonal minimum.

348 Since atmospheric S concentration is the sum of atmospheric SO_2 and aerosol SO_4 349 concentrations (expressed as mass of sulfur) and $\approx 67\%$ of atmospheric S is present as SO₂ 350 (RSO₂, Table 3), the behavior of atmospheric S concentration is dominated by that of SO₂. 351 Additionally, since $\approx 85\%$ of the dry S is usually contributed by SO₂ (RFSO₂, Table 3), the 352 behavior of dry S is likewise dominated by that of dry SO₂. As a result, P1-to-P4 changes of both 353 atmospheric S concentration and dry deposition are similar to those for atmospheric SO_2 354 concentration and dry deposition, and they are similar to each other (Fig. 4). Atmospheric S 355 concentration and dry S are reduced significantly in each region in each season and over all 356 seasons. Seasonally, the numerically largest significant absolute reductions usually occur in 357 winter when atmospheric SO₂ concentration is at its seasonal maximum, and the largest of these 358 occurs in the northeast. Regionally, P1-to-P4 relative reductions are similar for atmospheric S

359 concentration (42 to 47%) and dry deposition (48 to 51%).

360 In contrast to atmospheric SO_2 concentration (with its summertime minimum), wet S has 361 a seasonal summer maximum. This coincides with scavenging of SO₄ by precipitation when both 362 atmospheric aerosol SO₄ concentration and PR are generally at their seasonal maxima. The 363 numerically largest significant absolute and relative P1-to-P4 reductions in wet S also generally 364 occur in summer. Changes in wet S are numerically smaller and less significant in winter due to 365 seasonal minima of wet S and PR and to wintertime P1-to-P4 increases in PR. In general, 366 relative reductions in wet S are similar to those of aerosol SO₄ concentration (e.g., 33 and 32% in 367 the east).

368 Except during winter, wet S generally constitutes more than 50% of the total S ((D/T) S, 369 Table 3) and exerts a corresponding influence on total S. As noted earlier, P1-to-P4 total S is 370 reduced significantly in each region in each season and over all seasons (Fig. 4). Since seasonal 371 behaviors of dry S and wet S are different but complementary, the seasonal behavior of their 372 sum, total S, shows less seasonal variability than its constituents. Nevertheless, the numerically 373 largest significant P1-to-P4 absolute reductions for total S occur in summer when wet S is at its 374 seasonal maximum. Relative P1-to-P4 reductions are similar across regions, and average 40% in 375 the east.

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377 3.3. Oxidized nitrogen species

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379 Sampling artifacts described in Sect. 2 suggest that sampling artifacts may provide biases 380 in opposing directions for concentrations of HNO₃ and aerosol NO₃, but these impacts would be 381 minimized during non-summer sampling at non-marine sites. Regional and seasonal changes in 382 atmospheric HNO3 concentration have more consistent, substantial, and significant reductions 383 for P1-to-P4 (Fig. 4) than for P1-to-P2 or P1-to-P3 (not shown). With recent summertime 384 reductions in NOx emissions, the seasonal timing of peak HNO₃ concentration has shifted from 385 summer in P1 and P2 to spring in P3 and P4. In addition, P1, P2, and P3 summertime peaks of 386 dry deposition of both HNO₃ and OxN have shifted to spring in P4. The overall P1-to-P4 relative 387 reductions in atmospheric HNO₃ concentration in the east are 34% versus 0% (NS) for P1-to-P2 388 and 13% for P1-to-P3. Regionally, the numerically largest significant P1-to-P4 reductions occur 389 in the northeast (39%), with smaller reductions in the south (32%) and midwest (30%).

390 Numerically, maximum absolute and relative seasonal reductions consistently occur in summer,

391 with the largest of these in the northeast. Summertime P1-to-P4 relative reductions in

392 atmospheric HNO₃ concentration are 48% in northeast, 40% in south, 38% in midwest, and 42%

393 in east. Wintertime P1-to-P4 absolute reductions in atmospheric HNO₃ concentration are smaller,

394 40 to 46% of those in summer.

395 Dry HNO₃ exhibits P1-to-P4 reductions that are of similar magnitude and consistency 396 with those noted above for atmospheric HNO₃ concentration. The only exception is winter in the 397 midwest, where P4 Vd was 9% higher than in P1. Regionally, significant reductions occur in the 398 northeast (41%), south (35%), and midwest (28%), averaging 35% in the east. Seasonal behavior 399 of dry HNO₃ is similar to that of atmospheric HNO₃ concentration, with numerically maximum 400 absolute and relative seasonal reductions generally occurring in summer. Summertime P1-to-P4 401 relative reductions in dry HNO₃ are 50% in northeast, 41% in south, 34% in midwest, and 42% 402 in east. Wintertime P1-to-P4 absolute reductions in dry HNO₃ are smaller, 20 to 28% of those in 403 summer.

404 In contrast to P1-to-P3 comparisons, where atmospheric aerosol NO₃ concentration and dry deposition generally display significant increases (SS07b), P1-to-P4 comparisons in Fig. 4 405 406 show more reductions. Significant P1-to-P4 absolute and relative increases in atmospheric 407 aerosol NO₃ concentration and dry NO₃ occur only in winter in the east (e.g., 0.20 µgNO₃/m³ 408 and 12%, and 0.03 kgN/(ha y) and 31%) when atmospheric aerosol NO₃ concentration is at its 409 seasonal maximum. The increased magnitude of winter relative increases in dry NO₃ over those 410 of airborne concentration are likely due to previously noted significant period-to-period increases 411 in aerosols Vd. Wintertime regional absolute P1-to-P4 increases in concentration and dry 412 deposition are numerically strongest in the midwest (e.g., $0.27 \mu g NO_3/m^3$ and 0.05 k g N/(ha y)). 413 The wintertime P1-to-P4 increases in aerosol NO₃ concentration are considerably smaller than 414 the corresponding P1-to-P3 increases (Fig. 7). Significant P1-to-P4 reductions in absolute and 415 relative atmospheric aerosol NO₃ concentration occur in many of the remaining seasons, 416 especially summer. Significant P3-to-P4 reductions in relative atmospheric aerosol NO₃ 417 concentration occur in all seasons in the east, consistent with recent aggressive reductions of 418 NOx emissions.

419 Atmospheric OxN concentration is the sum of atmospheric HNO₃ and aerosol NO₃
420 concentrations (expressed as mass of nitrogen), and almost 60% of the OxN is generally present

421 as HNO₃ (RHNO₃ Table 3). As a result, the behavior of atmospheric OxN concentration reflects 422 the influence of both atmospheric HNO₃ and aerosol NO₃ concentrations. Also, since well over 423 90% of the dry OxN is generally contributed by HNO₃ (RFHNO₃, Table 3), the behavior of dry 424 OxN is strongly influenced by that of dry HNO₃. Seasonally in the east, the numerically largest 425 P1-to-P4 absolute and relative reductions in atmospheric OxN concentration and dry deposition (e.g., $0.26 \mu g N/m^3$ and 38%, and 1.23 k g N/(ha y) and 43%, Fig. 4) occur in summer when 426 427 atmospheric HNO₃ is the dominant OxN species. Summertime relative reductions are 428 numerically more pronounced in the northeast (43% and 50%, respectively) than the south (35% 429 and 43%) or midwest (36% and 34%). In contrast, in winter, when atmospheric aerosol NO₃ 430 concentration is at its seasonal maximum and significant P1-to-P4 increases of both atmospheric 431 aerosol NO₃ concentration and dry deposition occur, reductions in OxN concentration and dry 432 deposition also occur, but they are smaller in magnitude and in some cases not significant. Over 433 all seasons, regional P1-to-P4 relative reductions in atmospheric OxN concentration and dry 434 deposition are numerically larger in the northeast (29% and 40%) than the south (24% and 34%) 435 or the midwest (21% and 27%), and average 24% and 34%, respectively, in the east.

Wet OxN and PR in the east are at their seasonal maxima in summer (followed closely by
spring). Numerically, the largest significant absolute reductions in wet OxN occur in summer (or
spring). Reduced significance of some wintertime P1-to-P4 reductions in wet OxN is associated
with numerical increases in wintertime PR. Significant regional P1-to-P4 reductions in wet OxN
are numerically larger in the northeast (34%) than midwest (28%) or south (25%), and average
29% in the east.

442 Considering the dry deposition of a comprehensive suite of NO_Y species, field 443 measurements suggest that 40 to 85% of total NO_Y deposition at eight rural Canadian sites 444 occurred as wet deposition (Zhang et al., 2009); whereas, a modeling study suggests that for the 445 continental US this figure is closer to 31% for NO_Y and 44% for OxN (Zhang et al., 2012). In the 446 current study, over 60% of the total OxN generally is contributed by wet OxN ((D/T) OxN, 447 Table 3); however, this estimate may be biased because it neglects previously noted uncertainties 448 in Vd for HNO₃ and contributions to dry deposition by non-monitored NO_Y species. 449 Nevertheless, wet OxN appears to exert a moderate-to-strong influence on the behavior of total 450 OxN. Both dry OxN and wet OxN display strong seasonal P1-to-P4 reductions in summer that 451 become weaker in winter (Fig. 4). Seasonally, the numerically largest absolute and relative

reductions in total OxN occur in summer and range from 42% in the northeast, to 35% in the
south, to 33% in the midwest, and average 37% in the east. Regional P1-to-P4 reductions in total
OxN are 36% in the northeast, 29% in the south, and 28% in the midwest, and average 31% in
the east. Thus, over all seasons the numerically largest absolute and relative reductions in
atmospheric HNO₃ concentration and dry deposition, atmospheric OxN concentration and dry
deposition, and total OxN generally occur in the northeast, and seasonally they tend to occur
there in summer.

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3.4. Reduced nitrogen species

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462 Since NH₃ is not monitored in CASTNET, aerosol NH₄ is the reduced nitrogen species 463 considered in the current study. Numerically, the regional maximum for atmospheric aerosol 464 NH₄ concentration occurs near emissions sources in the source region and midwest, and the 465 seasonal maximum in the east occurs in summer. As is illustrated in Fig. 4, between P1 and P4, 466 atmospheric aerosol NH₄ concentration shows significant regional relative reductions that are 467 similar, ranging from 29% in the midwest, to 27% in the northeast, to 24% in the south, and 468 averaging 26% in the east. Seasonally, the numerically largest absolute and relative reductions of 469 atmospheric aerosol NH₄ concentration occur in summer (when atmospheric aerosol SO₄ 470 concentration is at its seasonal maximum and is also experiencing its largest absolute and relative 471 reductions). Summertime P1-to-P4 relative reductions in atmospheric aerosol NH₄ concentration 472 range from 40% in the midwest, to 32% in the northeast, to 26% in the south, and average 32% 473 in the east. In contrast, reductions are smaller in winter (e.g., 14% in the east) when atmospheric 474 aerosol NH₄ concentration is small and atmospheric aerosol SO₄ concentration is also at its 475 seasonal minimum.

Although generally showing significant reductions, dry deposition of atmospheric aerosol NH₄ shows numerically weaker P1-to-P4 changes than its atmospheric concentration. The overall P1-to-P4 relative reduction for dry NH₄ in the east (22%) is numerically smaller than the corresponding value for its atmospheric concentration. This pattern of weaker reductions is likely due to the significant P1-to-P4 increases in aerosol Vd, noted earlier. Regional P1-to-P4 relative reductions in dry NH₄ are 26% in the northeast and 21% in the midwest and south. Seasonally, in winter, when the atmospheric aerosol NH₄ concentration is low, changes in its dry deposition show changes which tend to be small and not significant regionally. Mirroring aerosol NH₄
concentration, the numerically largest P1-to-P4 absolute and relative reductions in dry NH₄ also
occur in summer. They range from 34% in the northeast, to 33% in the midwest, to 24% in the
south, and average 30% in the east.

Numerically, wet NH₄ is also high near emissions sources in the midwest and source
region, and in the east in the spring and summer seasons when PR is also relatively high. At the
regional and seasonal scales, wet NH₄ shows P1-to-P4 changes that are usually not significant.
The only two reductions that are significant occur overall in the midwest (12%) and east (9%).
Two wintertime increases occur in the midwest and the east, but they are not significant.

492 Since the magnitude of wet NH₄ is over seven times larger than dry deposition ((D/T)
493 NH₄, Table 3), the behavior of total NH₄ is strongly influenced by the behavior of wet NH₄.
494 Similar to wet NH₄, total NH₄ shows mostly P1-to-P4 reductions that are usually not significant,
495 but with non-significant wintertime increases in the midwest and the east. The only significant
496 reduction is overall in the east (11%).

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- 498 **3.5.** Oxidized plus reduced nitrogen species
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500 The sum of monitored oxidized and reduced nitrogen species is represented as N. The 501 behavior of the resulting N-metrics (i.e., atmospheric N concentration, dry N, wet N, and total N) 502 reflects the behavior of the weighted sum of the individual constituents, described previously. 503 Aerosol NH₄ constitutes $\approx 67\%$, HNO₃ $\approx 19\%$, and aerosol NO₃ $\approx 14\%$ of the monitored 504 airborne N concentration (RNNH₄ and RNHNO₃, Table 3). As a result, aerosol NH₄ exerts a 505 moderate-to-strong influence on its behavior. Seasonally, the numerically strongest absolute and 506 relative P1-to-P4 reductions of atmospheric N concentration occur in summer (midwest, 39%; 507 northeast, 34%; south, 28%; and east, 33%). This seasonality coincides with the summer peaks 508 of atmospheric N concentration and P1-to-P4 reductions in the concentration of aerosol NH₄, 509 HNO₃, and OxN. Regionally, atmospheric N concentration shows relative reductions of 26% in 510 the midwest, 27% in the northeast, 24% in the south, and 26% in the east. 511 Due to the large difference in deposition velocities, $\approx 75\%$ of dry N is contributed by dry 512 HNO₃, but only $\approx 21\%$ by dry NH₄ and $\approx 4\%$ by dry NO₃ (RNFHNO₃ and RNFNH₄, Table 3).

513 Seasonally, the numerically strongest absolute and relative P1-to-P4 reductions in dry N mirror

those of aerosol NH₄, HNO₃, and OxN concentration and dry deposition (and N concentration); they occur in summer (northeast, 47%; south, 39%; midwest, 34%; and east, 40%). Regional non-significant wintertime P1-to-P4 reductions may result in part from corresponding significant regional increases in dry NO₃. Regionally, the numerically largest P1-to-P4 reductions occur in the northeast (38%), with smaller reductions in the south (32%) and midwest (26%), and they average 32% in the east.

520 Wet OxN and wet NH₄ contribute about equally to wet N, with maxima in spring or 521 summer (Table 3). The numerically largest significant P1-to-P4 absolute reductions in wet N also 522 occur in spring or summer, and peak seasonal relative reductions range between 20 and 30%. 523 Non-significant numerical wintertime P1-to-P4 reductions in wet N occur in the northeast and 524 south, with a non-significant increase in the midwest. This behavior is likely associated with 525 previously noted numerical wintertime P1-to-P4 increases in PR. Regional P1-to-P4 relative 526 reductions in wet N are 24% in the northeast, 20% in the midwest, 17% in the south, and average 527 20% in the east.

528 Since wet N contributes $\approx 75\%$ of total N ((D/T) N, Table 3), the behavior of total N is 529 strongly influenced by the behavior of wet N. The numerically largest significant P1-to-P4 530 absolute reductions in total N generally occur in summer, with corresponding seasonal relative 531 reductions of 31% in northeast, 25% in midwest, and 26% in south. Non-significant P1-to-P4 532 reductions in total N occur only in wintertime and are associated with similar behavior of wet 533 OxN and wet N and numerical wintertime P1-to-P4 increases in PR. Regionally, the numerically 534 largest P1-to-P4 reductions occur in the northeast (28%), with smaller reductions in the midwest 535 and south (22%). The overall P1-to-P4 relative reduction of total N in the east is 24%.

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537 **3.6.** Wet H⁺

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Regionally, wet H⁺ deposition is generally heaviest in the northeast, and seasonally in the
east it is heaviest during summer (Table 3). Seasonally, the numerically largest absolute
reductions occur in summer, with relative reductions of similar magnitude in spring and summer;
while generally smaller reductions occur in winter (Fig. 4). The numerically largest seasonal
relative reductions range from 65% in the midwest (summer), to 55% in the northeast (spring), to
50% in the south (spring). Regionally, P1-to-P4 regional relative reductions of wet H⁺ are

numerically larger near sources in the midwest (56%) and smaller in the south (46%) and northeast (44%). The overall P1-to-P4 relative reduction in wet H^+ in the east is 47%.

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548 **3.7.** Atmospheric O₃ concentration and dry deposition

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550 In the current study, hourly atmospheric O_3 concentrations, averaged to weekly values, 551 are treated as other weekly CASTNET data (i.e., averaged to period-site-season means); however 552 this does not represent the relevant time period for determining compliance with the US ambient 553 air quality standard. As is illustrated in Fig. 4, mostly non-significant P1-to-P4 seasonal increases 554 in atmospheric O_3 concentration occur in winter in the northeast (13%, NS), midwest and south 555 (7%, NS), and east (9%, significant). Summertime atmospheric O₃ concentrations do show 556 substantial and mostly significant absolute and relative reductions in each region: 6.7 ppb (17%) 557 in the northeast; 5.3 ppb (14%) in the south; 2.4 ppb (6%, NS) in the midwest; and 4.9 ppb (12%) in the east. The numerically largest reduction in atmospheric O₃ concentration occurs in 558 559 summer and in the northeast, coinciding with aggressive O_3 -season NOx emissions controls 560 during P3 and P4 in this region and upwind in the east. Unpublished examination of seasonal 561 NOx emissions in the eastern US for the period between 1990-1992 and 2006-2008 (roughly 562 corresponding to P1-to-P4) suggests that reductions in summertime emissions are approximately 563 6% greater than those for the aggregate of the remaining three seasons over the same time 564 period. Regional reductions of atmospheric O₃ concentration over all seasons are small and only 565 significant in the south (5%) and east (4%).

Moderate P1-to-P4 increases in Vd for O₃ act in opposition to corresponding reductions in atmospheric O₃ concentration to yield P1-to-P4 changes in dry deposition of O₃ that are sometimes significant. Relatively large, significant seasonal reductions in dry O₃, ranging from 5 to 15%, occur regionally in summer; however, corresponding increases ranging between 15 and 25% occur in winter. Regionally, P1-to-P4 reductions in dry O₃ occur in the northeast (8%) and south (4%), and a non-significant increase occurs in the midwest (1%, NS). The P1-to-P4 relative reduction in dry O₃ in the east is 4%.

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575 **4. Discussion**

576

- 577 4.1. Changes in atmospheric aerosol SO₄ and SO₂ concentration
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579 Both P1-to-P4 absolute and relative changes in atmospheric aerosol SO₄ concentration, 580 shown in Fig. 4, are numerically substantially less than the corresponding values for atmospheric 581 SO₂ concentration, consistent with the findings of Reid et al. (2001), Holland et al. (2004), and 582 SS07b for earlier time periods. For regional groupings of sites and in the east, the year-round 583 ratios of P1-to-P4 relative reductions in atmospheric aerosol SO₄ concentration to atmospheric 584 SO₂ concentration are similar ($\approx 0.7\%$ decrease in aerosol SO₄ for each 1% decrease in SO₂). In 585 the east these ratios range numerically between a wintertime low ($\approx 0.5\%$ SO₄ per 1% SO₂) and a 586 high ($\approx 0.8\%$ SO₄ per 1% SO₂) in summer when aerosol SO₄ levels are at their seasonal maxima 587 and SO₂ levels are at their seasonal minima. Year-round ratios of P1-to-P4 absolute reductions of 588 atmospheric aerosol SO₄ concentration to those of SO₂ concentration (both expressed as mass of 589 sulfur) are similar in regions of higher emissions density, the midwest and northeast (≈ 0.2 590 μ gS/m³ decrease in aerosol SO₄ for each 1 μ gS/m³ decrease in SO₂). This ratio is numerically 591 higher in the south (≈ 0.3), and it may be due in part to concurrent reductions of SO₂ emissions in 592 areas of higher emissions density with subsequent transport of reduced amounts of oxidized 593 sulfur, especially aerosol SO₄, into the south. 594 RSO₂ (Table 3), is the ratio of atmospheric SO₂ concentration to total atmospheric S

595 concentration, and 1-RSO₂ is the relative amount of atmospheric S concentration present as 596 aerosol SO₄, representing a rough index of the extent of conversion of SO₂ to other oxidized 597 sulfur species (i.e., aerosol SO₄). Significant P1-to-P4 RSO₂ reductions (not shown) of 8% in the 598 south, 5% in the northeast, and 7% in the midwest and east correspond to increases in the 599 conversion index (e.g., 17% in the east). This suggests that as the P1-to-P4 atmospheric SO₂ 600 concentration in the east has decreased, the relative amount of airborne S present as SO₂ has 601 decreased, and both the relative amount present as aerosol SO₄ and the apparent extent of 602 conversion of SO₂ to aerosol SO₄ have increased.

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4.2.

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As noted in Section 3.2, wet S exceeds dry S in the east, except in winter. The P4 ratio of

Changes in dry and wet deposition

607 dry S to total S in the east is 0.37 ((D/T) S, Table 3), down from 0.44 in P1; qualitatively similar 608 behavior is noted for the BKG sites (i.e., P4 ratio 0.23 versus 0.26 in P1) and HE sites (i.e., P4 609 ratio 0.40 versus 0.50 in P1). In the east, the ratio of dry S to wet S (calculated from (D/T) S) has 610 diminished by 24% from P1 (0.78) to P4 (0.60). This reduction is strongest (>30%) in regions of 611 high emissions density (midwest and northeast) in winter, where SO_2 concentrations are at their peak. Year-round, in the east and regionally, both P1-to-P4 absolute and relative reductions in 612 613 dry S are numerically larger (by ≈ 20 to 70%, Fig. 4) than corresponding values for wet S. 614 However, in summer when PR is high and atmospheric aerosol SO₄ concentration is at its 615 seasonal maximum, the absolute P1-to-P4 reductions are larger (by ≈ 25 to 100%) for wet S than 616 dry S. In other seasons, the roles are different, and in winter, when atmospheric SO_2 617 concentration is at its seasonal maximum and PR is at its seasonal minimum, P1-to-P4 absolute 618 and relative reductions in dry S, are substantially larger (by over a factor of ≈ 2) than the 619 corresponding reductions for wet S. Thus, strong emissions controls of the primary pollutant SO_2 620 have reduced dry S to a greater extent than wet S, permitting the latter to become an increasing 621 relative contributor to total S. This, along with the increased apparent conversion of SO_2 to 622 aerosol SO₄, noted above, suggests that wet processes are becoming more prominent than dry 623 processes at removing oxidized sulfur from the current atmosphere in the east.

624 Whereas deposition estimates of both primary and secondary oxidized sulfur species are 625 available for comparison of their dry and wet P1-to-P4 changes, only changes in deposition of 626 monitored secondary oxidized nitrogen species can be compared in the current study. In the east, 627 dry OxN appears to be numerically smaller than wet OxN (Section 3.3), and the P4 ratio of dry 628 OxN to total OxN is 0.35 ((D/T) OxN, Table 3), down slightly from 0.37 in P1. Qualitatively 629 similar behavior is noted for the BKG sites (i.e., P4 ratio 0.27 versus 0.30 in P1) and HE sites 630 (i.e., P4 ratio 0.41 versus 0.51 in P1). Absolute P1-to-P4 reductions in dry OxN are generally 631 numerically smaller than corresponding values for wet OxN (Fig. 4). However, with the recent, 632 more aggressive NOx emissions controls, the relative P1-to-P4 reductions of dry OxN now 633 exceed those for wet OxN more frequently than in the earlier P1-to-P3 comparison (SS07b). 634 These findings suggest that wet OxN is becoming an increasing relative contributor to total OxN. 635 4.3. **CASTNET PM**

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637

Although PM_{2.5} is not monitored in CASTNET, many of the major inorganic constituents (i.e., aerosol SO₄, NO₃, and NH₄) are monitored. In contrast to aerosol SO₄ and NH₄, which occur mainly as fine particles, there is evidence suggesting that aerosol NO₃ can be multimodal in the east and is subject to sampling artifacts in opposing directions with the CASTNET sampler, but they are thought to be minimal during cooler seasons at the non-marine, eastern CASTNET sites (see Sect. 2). In the current study, CASTNET PM is taken to be the sum of atmospheric aerosol SO₄, NO₃, and NH₄ mass concentrations.

645 Significant P1-to-P4 (Fig. 5) and P3-to-P4 reductions occur in CASTNET PM 646 concentration in each region in each season and over all seasons. This is in contrast to P1-to-P3 647 behavior where non-significant wintertime changes occurred (SS07b), with both numerical and 648 significant increases in the midwest (Fig. 8). Year-round absolute and relative reductions in the east are 2.3 μ g/m³ and 28% (P1-to-P4) and 1.1 μ g/m³ and 16% (P3-to-P4). Seasonally, they peak 649 650 in summer when aerosol SO₄ and NH₄ concentrations are at their maxima and are also 651 experiencing their largest reductions. Regionally, the largest absolute and relative reductions in 652 CASTNET PM occur in the midwest in summer (4.9 μ g/m³ and 41% (P1-to-P4) and 1.9 μ g/m³ 653 and 21% (P3-to-P4)).

654 For the chemically complex NH₃-HNO₃-H₂SO₄-water system, conditions can arise where 655 the availability of NH₃ limits the formation of aerosol NH₄NO₃. Here, subsequent reductions of 656 airborne aerosol SO₄, by tying up less NH₃ as aerosol (NH₄)₂SO₄, may make more NH₃ available 657 to react with HNO₃, increasing the level of aerosol NH₄NO₃. Although atmospheric NH₃ 658 concentration is not monitored in CASTNET, when the index, RNO₃ (i.e., the ratio of 659 atmospheric aerosol NO₃ to OxN concentrations), is less than 0.7, then formation of aerosol 660 NH₄NO₃ may be NH₃-limited (Blanchard et al., 2000). Using this index and focusing on non-661 summer season results to minimize the impacts of sampling artifacts (see Sect. 2), results in 662 Table 3 suggest that NH₃-limited conditions are present in the east but are weakest in winter and 663 year-round in the midwest.

664 Significant P1-to-P2-to-P3-to-P4 increases in RNO₃ occur in the east. Regionally, the 665 numerically largest absolute and relative P1-to-P4 increases occur in the northeast (40%), with 666 smaller increases in the south (28%), midwest (10%), and east (22%). Seasonally, the 667 numerically largest absolute increases occur in winter. The numerically largest regional P1 668 winter RNO₃ value (0.70) occurs in the midwest, and this value increases significantly by 12% to

- 669 0.78 in P4. These results suggest that between P1 and P4, NH₃ availability has increased in the
- east and the formation of aerosol NO₃ has become less NH₃-limited. The least NH₃-limited
- 671 situation appears to occur in winter in the midwest (with its strong sources of NH₃). In addition,
- the relative amount of monitored OxN present as HNO₃, RHNO₃ (i.e., 1-RNO₃), diminished
- 673 significantly in the east between P1 (0.66) and P4 (0.59) (Section 3.3).

674 For a chemical system initially containing, NH_3 , HNO_3 , H_2SO_4 , and water, an index of 675 the extent to which ambient sulfuric acid (H₂SO₄) has been neutralized by atmospheric NH₃ may 676 be defined as the neutralization index (NI). Correcting for the presence aerosol NH₄NO₃, it may 677 be computed as ([NH₄] - [NO₃]) / 2[SO₄] (Blanchard and Hidy, 2005). The value of NI can range 678 between zero (unneutralized H_2SO_4) and unity (100% neutralization to (NH₄)₂SO₄). As shown in 679 Table 3, NI averages ≈ 0.8 in the east. Calculation of NI assumes that aerosol NO₃ is present as 680 fine aerosol NH_4NO_3 and neglects the possible presence other forms of aerosol NO_3 (e.g., from 681 large particle sea/road salt or soil/mineral dust). As suggested in Sect. 2, sampling artifacts may 682 provide biases in opposing directions for aerosol NO₃. Their impacts are expected to be minimal 683 during cooler season sampling at non-marine sites. Although small (but significant) P1-to-P4 684 increases in NI occur in the east, large increases occur in summer in the northeast (15%) and 685 south (18%), while small non-significant decreases occur in winter. These results suggest that 686 between P1 and P4 an acidic aerosol is approaching neutralization due to increased relative 687 availability of NH3 associated with reductions of aerosol SO4 concentration. These findings are 688 consistent with those noted in previous paragraphs and with those for P1-to-P3 (SS07b), 689 indicating that conditions in the east are becoming less NH₃-limited.

690 Modeling efforts have indicated that nonlinear changes in PM_{2.5} concentrations can occur 691 in response to reductions in atmospheric aerosol SO₄ concentrations in the east (Ansari and 692 Pandis, 1998; West et al., 1999; Blanchard and Hidy, 2005). Under NH₃-limited conditions 693 Blanchard et al. (2000) suggest: that reduction of aerosol SO₄ concentrations can increase aerosol 694 NO_3 concentrations by amounts that exceed the aerosol SO_4 reductions on a mass basis; and that 695 reductions of ambient oxidized nitrogen concentrations (that may accompany reductions in NOx 696 emissions) may not reduce aerosol NO₃ concentrations. These modeling studies suggest that for 697 the NH₃-HNO₃-H₂SO₄-water system, constituent concentration and temperature conditions can 698 exist in the east so that when the atmospheric aerosol SO₄ concentration is reduced, the 699 corresponding change in $PM_{2.5}$ concentration can be influenced nonlinearly. At low

temperatures, aerosol NH₄NO₃ will likely form when NH₃ is made available from a reduction in atmospheric aerosol SO₄ concentration, and negative to nonlinear responses ranging from -0.3 to $<1.0 (\Delta \mu g PM_{2.5} / \Delta \mu g SO_4)$ can result (Ansari and Pandis, 1998; West et al., 1999). Using model estimates, West et al. (1999) have predicted that these conditions would exist most commonly in winter in the midwest and would be uncommon in summer in the east. A 2009 field study (Stanier et al., 2012) has focused on wintertime episodes of elevated aerosol NO₃ concentrations in the midwest (Wisconsin).

707 Ratios of P1-to-P4 absolute changes in CASTNET PM concentrations to corresponding 708 changes in aerosol SO₄ (($\Delta(\mu g \text{ CASTNET PM/m}^3) / \Delta(\mu g \text{ SO}_4/m^3)$) have been determined. Year 709 round, these ratios are of similar magnitude, numerically falling between 1.2 and 1.6, both for the 710 east and for each regional grouping of sites. Seasonally in the east, these ratios range between 1.3 711 and 1.5 in spring, summer, and fall. However, in winter they are numerically close to 1.0 in the 712 east and each region. This represents a change from P1-to-P3 behavior, where the wintertime 713 ratio averaged close to 0.0 in the east, and in the midwest was negative and accompanied by a 714 numerical increase in CASTNET PM (SS07b). Wintertime numerical increases in CASTNET 715 PM occurred at 15 sites between P1 and P3, but at only one site between P1 and P4 and at only 716 one site between P3 and P4 (Fig. 8). This wintertime behavior of CASTNET PM appears to be 717 driven largely by that of aerosol NO₃ (Fig. 7). These findings suggest that additional P3-to-P4 718 reductions in emissions of SO₂, and especially NOx, have made progress in altering the chemical 719 regime of the wintertime eastern US atmosphere so that future emissions reductions and their 720 resulting reductions in aerosol concentrations may no longer be accompanied by sub-linear 721 changes (or actual increases) in CASTNET PM.

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4.4.

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Emissions estimates for SO₂ and NOx (Xing et al., 2013) have been aggregated to regional levels and have been used to determine current mean estimates of P1, P2, P3 and P4 emissions as well as period-to-period absolute and relative changes in regional SO₂ and NOx emissions in the east (see Section 2.1 for identification of the states composing each geographical region). Although current P1, P2, and P3 emissions estimates have been updated and may be slightly different compared to those reported in SS07b; using the same current

Comparison with emissions estimates

version permits period-to-period comparisons to be made using consistent emissions estimates.

732 In addition, subsequent analyses repeated using emissions estimates (US EPA, National

733 Emissions Inventory, Emissions Factor and Inventory Group, Office of Air Quality Planning and

734 Standards, Research Triangle Park, NC) [http://www.epa.gov/ttn/chief/net/2008inventory.html,

accessed 02/12/13], yield reasonably consistent findings with no appreciable changes to

736 conclusions.

Continuous emissions monitors used on most EGUs provide reasonably accurate
estimates of SO₂ and NOx emissions from these sources. Although EGUs are the main source of
SO₂ emissions; EGUs, on-road, and off-road sources contribute substantially to total NOx
emissions. Uncertainty in on-road and other non-EGU NOx emissions make estimates of total
NOx emissions more uncertain than those for SO₂ (Butler et al., 2005; Parrish, 2006; Blanchard
et al., 2013).

Estimates of both P1 and P4 emissions density are shown in Table 4 in the same units as deposition. For example, the P1 emissions density for SO₂ is over 60% higher in the midwest, with its EGU sources in the Ohio Valley, than in the south (i.e., 51 versus 31 kgS/(ha y)). P1-to-P4 relative change results for regional SO₂ and NOx emissions (Fig. 3) are also shown in Table 4 with corresponding values for selected measures of atmospheric concentration and total deposition from Fig. 4.

749

750 **4.4.1. Oxidized sulfur species**

The fairly universal reductions in the S-related concentration and deposition metrics
 noted previously in Section 3.2 largely reflect the year-round SO₂ emissions controls that have

been continuing since 1995 (Fig. 1). Reductions in SO₂ emissions in the east have led to

r54 significant reductions in monitored SO₂ concentrations (i.e., P1-to-P2 (21 versus 27%), P1-to-P3

755 (35 versus 33%), P1-to-P4 (50 versus 48%), and P3-to-P4 (23 versus 23%)). Reductions in SO₂

emissions in the east have led to significant, but smaller, reductions in aerosol SO₄

757 concentrations (i.e., P1-to-P2 (21 versus 12%), P1-to-P3 (35 versus 21%), P1-to-P4 (50 versus

758 33%), and P3-to-P4 (23 versus 15%)).

As indicated in Table 4, the P1-to-P4 relative reductions in SO₂ emissions in the east (50%) are in closer numerical agreement with reductions of atmospheric concentration of the monitored primary pollutant, SO₂ (48%), than the monitored reaction product, aerosol SO₄ 762 (33%). Ratios of the P1-to-P4 relative change in monitored concentration to emissions estimates 763 for SO_2 in the east are approximately 0.97 for SO_2 and 0.66 for aerosol SO_4 , with similar 764 findings for P3-to-P4 comparisons. Relative P1-to-P4 comparisons also show significant 765 reductions for SO₂ emissions (50%) and dry S (49%). These results suggest that at the current 766 scale, the responses of both atmospheric SO₂ concentration and dry deposition to changes in SO₂ 767 emissions are close to 1:1; however, the relationship appears to be less than 1:1 for aerosol SO₄. 768 In addition, as indicated here and in Section 4.1, the relationship between changes in the 769 concentration of the reaction product, aerosol SO₄, and that of the primary pollutant, SO₂, 770 appears to be less than 1:1.

771 These findings are generally consistent with earlier findings for P1-to-P3 comparisons 772 (SS07b) as well as those of those of others. Butler et al. (2001) reported slightly less than a 1:1 773 relationship between changes in downwind atmospheric S concentration and changes in SO₂ 774 emissions from specific upwind source regions. Blanchard et al. (2013) found southeastern US 775 trends of SO_2 ambient concentrations and emissions to be decreasing with a qualitative 1:1 776 proportionality, while the relationship for trends of ambient aerosol SO_4 concentrations and SO_2 777 emissions were decreasing but at less than 1:1 proportionality. Hand et al. (2012) reported linear 778 relationships between aerosol SO₄ concentrations and SO₂ emissions that depended on location 779 in the US, with aerosol SO₄ concentration showing the weakest response to SO₂ emissions in a 780 region in the eastern US roughly included in the currently defined source region, midwest, and 781 northeast.

782 Atmospheric pollutants are transported and transformed by physical and chemical 783 processes between their release and deposition. For example, gaseous SO_2 may be transformed 784 into aerosol SO₄ by several pathways, including gas-phase oxidation by the hydroxyl radical and, 785 after dissolution in water droplets, via oxidation by hydrogen peroxide and O₃ (McHenry and 786 Dennis, 1994). As SO₄-containing aqueous particles undergo atmospheric processing (i.e., 787 evaporation and dissolution), there is a close linkage between the secondary pollutant's 788 atmospheric concentration and wet removal (i.e., they have similar atmospheric time scales). 789 Although SO₄ is then subject to both dry and wet deposition, the latter dominates (e.g., east P4 790 wet S is 5.2 kgS/(ha y) versus 0.4 kgS/(ha y) for dry SO₄ (Table 3)). This may explain the 791 relatively good agreement between P1-to-P4 (33%) and P3-to-P4 (15%) reductions in the 792 atmospheric concentration of secondary pollutant, aerosol SO₄, and corresponding changes (32%)

793 and 13%) in wet S.

Since atmospheric S concentration is comprised of SO_2 and aerosol SO_4 , reductions of SO₂ emissions may be expected to yield responses of atmospheric S concentration and total S that fall between corresponding values for SO_2 and aerosol SO_4 . This is confirmed in Table 4 where reductions in SO_2 emissions in the east correspond to significant reductions of both atmospheric S concentration (i.e., P1-to-P4 (50 versus 45%)) and total S (i.e., P1-to-P4 (50 versus 40%)).

800 For a pollutant where all primary and secondary species are monitored (e.g., S), 801 neglecting other sources and sinks, emissions density may be compared with total deposition to 802 construct rough budgets for oxidized sulfur, revealing insights about the relative amounts of the 803 species deposited and exported. Model predictions for North America (Wojcik and Chang, 1997) 804 suggest approximately 50% of S emissions are deposited. In the current study, P1, P2, P3 and P4 805 S emissions densities in the east are 38, 30, 25, and 19 kgS/(ha y), while corresponding total S 806 values are 15, 12, 11 and 8.7 kgS/(ha y). Between P1 and P4 the relative amount of deposited S 807 emissions increased by $\approx 8\%$ from 39 to 47%, and the net exported S emissions decreased by a 808 corresponding amount from 61% to 53%. Thus, the strong P1-to-P4 SO₂ emissions controls are 809 accompanied by increased apparent conversion of SO₂ to aerosol SO₄ (Section 4.1), possible 810 increased relative contribution of wet versus dry deposition (Section 4.2), increased fraction of 811 deposited S emissions, and decreased fraction of exported S emissions.

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813 4.4.2. Oxidized nitrogen species

814 Changes in pollutant emissions generally lead to more direct changes in the atmospheric 815 concentration of primary rather than secondary pollutant species, where emissions as well as 816 variable meteorological and atmospheric chemical processes are influential. In contrast to 817 oxidized sulfur species, where the behaviors of both primary and secondary pollutants are 818 monitored, the atmospheric concentration and deposition of only a few of the secondary oxidized 819 nitrogen species (i.e., HNO₃, aerosol NO₃ and aqueous NO₃ ion) are monitored in the current 820 study. This likely propagates more uncertainty into comparisons with primary NOx emissions 821 than is present in earlier comparisons with SO₂ emissions.

Recent aggressive NOx emissions controls in the east have led to significant reductions in concentrations of the important secondary oxidized nitrogen pollutant, HNO₃ (i.e., P1-to-P2 (6% 824 versus NS), P1-to-P3 (22 versus 13%), P1-to-P4 (42 versus 34%), and P3-to-P4 (25 versus 825 (24%)). The non-linear behavior of aerosol NO₃ concentration noted in Section 4.3 is apparent in 826 many comparisons of reductions in NOx emissions with aerosol NO₃ behavior but appears to be 827 diminishing with time (i.e., P1-to-P2 (6 versus -5%), P1-to-P3 (22 versus -11%), P1-to-P4 (42 828 versus 8%), and P3-to-P4 (25 versus 17%), where the minus sign indicates increase). 829 The numerically largest P1-to-P4 relative reductions in estimated NOx emissions (46%), 830 year-round atmospheric concentrations of HNO₃ (39%) and OxN (29%), dry OxN (40%), wet 831 OxN(34%), total OxN(36%), and summertime total OxN(42%) occur in the northeast, the 832 region with the largest P1 emissions density (Table 4). As noted in Section 3.3, numerical 833 maximum absolute and relative seasonal reductions in atmospheric HNO₃ concentrations also 834 occur in summer, with largest reductions in the northeast. These changes are likely reflecting 835 contributions from year-round control of NOx emissions by Phases I and II of the CAAA and 836 continuing mobile source controls. However, the timing (e.g., changes most apparent in summer 837 with peak concentrations moving from summer to spring) and the location (i.e., strongest 838 reductions in northeast) suggest that major contributors are summertime NOx emissions control 839 activities instituted under OTC, SIP Call, BTP and CAIR that focused on EGU sources in the 840 eastern states.

841 As noted above and in SS07b, the 22% reduction in the east of P1-to-P3 NOx emissions 842 is associated with a 13% reduction in HNO₃ concentration, an 8% reduction in dry OxN, a 12% 843 reduction in wet OxN, an 11% reduction in total OxN, but an 11% increase in aerosol NO₃ 844 concentration. Here, the ratio of P1-to-P3 relative changes of monitored HNO₃ concentration to 845 those of estimated NOx emissions is 0.6, qualitatively consistent with the value of 0.7 found by 846 Butler et al. (2005) for a similar time period (1991-2001). Recent aggressive NOx emissions 847 controls, however, appear to have altered these relationships. As indicated in Table 4, the 42% 848 P1-to-P4 reduction of NOx emissions in the east corresponds to 34% reductions in HNO₃ 849 concentration and dry OxN, a 29% reduction in wet OxN, a 31% reduction in total OxN, and an 850 8% reduction in aerosol NO₃ concentration. Here, the ratio of P1-to-P4 relative changes of 851 monitored HNO₃ concentration and dry OxN to those of estimated NOx emissions is 0.8, but the 852 corresponding ratio for aerosol NO_3 concentration is 0.2. These findings are consistent with the 853 recent observations in the southeastern US of decreasing 1999-2010 trends of NO_Y 854 concentration that exceed those for aerosol NO₃ concentration and are qualitatively proportional

- to decreasing NOx emissions trends (Blanchard et al., 2013).
- 856 In the east, the 42% P1-to- P4 reduction of NOx emissions also corresponds to significant 857 reductions of deposition (i.e., dry HNO₃ (35%), dry OxN (34%), wet OxN (29%) and total OxN 858 (31%)). The relatively good agreement between the P1-to-P4 changes in the atmospheric 859 concentration of HNO₃, dry HNO₃, and dry OxN can be explained by the high RFHNO₃ (Section 860 3.3; Table 3). The favorable comparison with wet OxN may be related to the high relative 861 abundance and high water solubility of HNO₃ in comparison to other NO_Y species in the 862 atmosphere. It could also be related to a possible high correlation of atmospheric concentration 863 of HNO₃ with those of other airborne NOx reaction products that are removed from the 864 atmosphere and measured as aqueous NO₃ ion in precipitation. 865 Model predictions for the continental US (Zhang et al., 2012) suggest approximately 63% of 2006-2008 NOx emissions are deposited as NO_Y and 45% as OxN. In the current study, none 866
- 867 of the primary NOx species and only two of the secondary species (HNO₃ and aerosol NO₃) are 868 monitored. Nevertheless, neglecting other sources and sinks, it is possible to compare NOx
- 869 emissions density with total OxN deposition to determine the fraction of NOx emissions
- 870 deposited as OxN. Using the approach described in Section 4.4.1 for oxidized sulfur,
- approximately 30% of the NOx emissions appear to be deposited as OxN during the study
- period. Although there are hints of a numerically small P1-to-P4 increase in the relative amount
- 873 of NOx emissions deposited as OxN, the large relative uncertainty associated with NOx
- 874 emissions estimates (Section 4.4) precludes a conclusion in this instance. Nevertheless, the
- strong P1-to-P4 reduction of NOx emissions (42%) is accompanied by reduced atmospheric
- 876 concentration of HNO₃ and OxN as well as RHNO₃ (Sections 3.3 and 4.3), and by increased
- relative contribution of wet OxN versus dry OxN (Section 4.2).
- 878

879 **4.4.3. Wet H**⁺

In the east the P1-to-P2 reductions in estimated emissions are 21% for SO₂ and 6% for NOx. Concurrent changes in deposition include reductions of 17% for wet S, 3% (NS) for both wet OxN and wet NH₄, and 24% for wet H⁺ (not shown). In the absence of sizeable P1-to-P2 changes in NOx emissions, it appears that P1-to-P2 reductions in SO₂ emissions played a major role in the resulting reductions of both wet S and wet H⁺ in the east. P1-to-P3 reductions in estimated SO₂ and NOx emissions are 35% and 22%, with corresponding reductions of 22% in

- wet S, 12% in wet OxN, no change in wet NH₄, and 34% in wet H⁺ (SS07b). For P1-to-P4,
- reductions in estimated SO_2 and NOx emissions are 50% and 42%, and corresponding reductions
- are 32% for wet S, 29% for wet OxN, 9% for wet NH₄, and 47% for wet H⁺ (Fig. 4). Respective
- 889 P3-to-P4 reductions in estimated SO₂ and NOx emissions are 23% and 25%, with corresponding
- reductions of 13% in wet S, 19% in wet OxN, 9% in wet NH₄, and 20% in wet H⁺.
- 891 Changes in wet NH₄ between P1 and P3 are generally small, variable, and non-
- significant. Although significant P1-to-P4 reductions in wet NH₄ are also relatively small (\approx 9%),
- they occur mainly between P3 and P4. In the east between P1 and P4, with the continuing
- reduction in SO₂ and NOx emissions, the fraction of wet S plus wet OxN that could be
- neutralized on an equivalent basis by wet NH₄ increased by approximately 8% from 25% in P1,
- to 33% in P4.

897 Neglecting the relatively small temporal impacts of wet NH₄ (above) and those of other 898 wet cations, examination of the behavior of wet S plus wet OxN (on an equivalent basis) permits 899 rough estimation of the relative contributions of reductions in SO₂ and NOx emissions to 900 corresponding reductions in wet H⁺. Year-round in the east, P1-to-P3 reductions in wet S 901 contribute most (\approx 80%) of the relative reduction of wet S plus wet OxN associated with reduced 902 wet H⁺. Between P1 and P4, this relative contribution diminishes (to $\approx 70\%$), and between P3 and 903 P4 the relative contribution of wet S (\approx 55%) approaches that of wet OxN. This finding illustrates 904 the impact of recent aggressive NOx emissions controls along with continuing SO₂ controls on 905 wet H⁺ and the increasing relative contribution of reductions in NOx emissions to the significant 906 and substantial reduction of wet H⁺ in the eastern US.

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909 5. Summary, conclusions, and recommendations

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Data collected in the eastern US between 1990 and 2009 at 34 CASTNET dry monitoring sites and paired NADP wet monitoring sites are examined. A major objective is to evaluate the monitored air quality impacts occurring between 1990 and 2009 that are associated with concurrent legislatively-mandated changes in emissions. Four 5-year periods (P1, P2, P3, and P4) are considered. Period-to-period changes in selected pollutant metrics are examined, focusing on P1-to-P4 changes. Data are composed from reported weekly measurements into

- 917 period-site-season means. These means, computed for atmospheric concentration, Vd, PR, and
- 918 dry, wet and total deposition, are used to examine differences between 5-year periods for
- 919 seasons, sites, and predefined regional groupings of sites. The approach to data analysis uses the
- 920 mean squared error derived from analysis of variance of mean estimates for each pollutant metric
- 921 to examine differences in the monitoring data and metrics derived from them.
- 922 Estimated SO₂ emissions for P4 are 50% lower than for P1. This is accompanied in each 923 regional site grouping in each season and over all seasons by significant reductions in 924 atmospheric SO₂ concentration and dry deposition, atmospheric aerosol SO₄ concentration, 925 atmospheric S concentration and dry deposition, wet S (except in winter), and total S. For 926 atmospheric SO₂ concentration and dry deposition, the numerically largest significant seasonal 927 absolute reductions usually occur in winter when atmospheric SO₂ concentration is at its 928 seasonal maximum. Analogous behavior occurs in summer for atmospheric aerosol SO₄ 929 concentration, dry SO₄, and wet S, when atmospheric aerosol SO₄ concentration is at its seasonal 930 maximum. In the east, significant reductions of 48%, 33%, and 45% for atmospheric 931 concentration of SO_2 , aerosol SO_4 and atmospheric S, and of 49%, 32%, and 40% for dry, wet, 932 and total deposition of atmospheric S are associated with the 50% P1-to-P4 reduction in 933 estimated SO₂ emissions. Consistent with earlier findings (SS07b), results suggest that at the 934 current scale, the responses of both atmospheric SO₂ concentration and dry deposition to changes in SO₂ emissions are close to 1:1; and the relationship between changes in the concentration of 935 936 the secondary reaction product, aerosol SO₄, and that of the primary pollutant, SO₂, appears to be 937 less than 1:1. The strong P1-to-P4 reduction of SO_2 emissions (50%) is accompanied by 938 increased apparent conversion of SO₂ to aerosol SO₄, possible increased relative contribution of 939 wet versus dry deposition as a sink for S, increased fraction of deposited S emissions, and 940 decreased fraction of exported S emissions. 941 In the east, significant P1-to-P4 reductions of 34%, 8%, and 24% for atmospheric 942 concentration of HNO₃, aerosol NO₃ and OxN, and of 34%, 29%, and 31% for dry, wet, and total 943 deposition of OxN are associated with the 42% reduction in estimated NOx emissions. 944 Atmospheric aerosol NO₃ concentration in the east displays a significant P1-to-P3 increase 945 (11%), with the numerically strongest seasonal relative increase (31%) in winter (SS07b). Recent
- 946 aggressive reductions of NOx emissions have been accompanied by significant P1-to-P4 and P3-
- 947 to-P4 reductions of 8% and 17% in atmospheric aerosol NO₃ concentration over all seasons. In

948 addition, the smaller P1-to-P4 wintertime increase in atmospheric aerosol NO₃ concentration 949 (i.e., 12% versus 31% for P1-to-P3) and the P3-to-P4 wintertime reduction of 14% suggest that 950 recent continuing NOx emissions controls are having a desirable impact. Over all seasons the 951 numerically largest absolute and relative reductions in atmospheric HNO₃ concentration and dry 952 deposition, atmospheric OxN concentration and dry deposition, and total OxN generally occur in 953 the northeast (the geographical region with the highest P1 NOx emissions density), and 954 seasonally they tend to occur there in summer. In the east, coincident timing and location suggest 955 that aggressive summertime NOx emissions reductions by EGU sources contributed substantially 956 to this observed behavior. The strong P1-to-P4 reduction of NOx emissions (42%) is 957 accompanied by reduced atmospheric concentration of HNO₃ relative to OxN, and increased

958 contribution of wet OxN relative to dry OxN.

959 Estimated annual 1990 to 2009 NH₃ emissions are relatively constant, ranging between 960 3.5 and 3.9 Tg/y (Xing et al., 2013). However, this is not confirmed through monitoring in the 961 current study, because only the atmospheric concentration of the secondary reduced nitrogen 962 species, aerosol NH₄, is monitored. In the east, both atmospheric aerosol NH₄ concentration and 963 dry deposition show significant P1-to-P4 reductions (i.e., 26 and 22%). The numerically largest 964 absolute and relative reductions of atmospheric aerosol NH₄ concentration and dry deposition 965 tend to occur regionally in the midwest and northeast and seasonally in the summer (when 966 atmospheric aerosol SO₄ concentration is at its seasonal maximum and is also experiencing its 967 largest reductions). Although showing a significant (9%) P1-to-P4 reduction over all in the east, 968 wet NH₄ shows changes at the regional and seasonal scales that are usually not significant. Since 969 the magnitude of wet NH₄ is over seven times larger than dry deposition, the resulting reduction 970 in total NH₄ in the east (11%) is similar to that of wet NH₄.

971 The atmospheric concentration of N, the sum of monitored oxidized and reduced nitrogen 972 species, is dominated in the east by aerosol NH₄ (\approx 67%) with smaller contributions from HNO₃ 973 (\approx 19%) and aerosol NO₃ (\approx 14%). However, a high Vd permits HNO₃ to dominate dry deposition 974 of N (\approx 75%). Both the atmospheric concentration and dry deposition of N show significant 975 overall P1-to-P4 reductions (26% and 32%) which tend to be seasonally higher in summer. Wet 976 deposition of OxN and NH₄ are fairly comparable, and their sum (wet N) dominates total N 977 (\approx 75%). The overall P1-to-P4 relative reductions of wet N and total N in the east are 20% and 978 24%, with the strongest reductions in summer and in the northeast.

- In the east, a 47% reduction in wet deposition of H^+ ion is associated with P1-to-P4 reductions of 50% and 42% in SO₂ and NOx emissions. The relative contributions of reduced NOx versus SO₂ emissions have increased over time, until currently (P3-to-P4), they are almost equal. Thus, recent aggressive reductions of NOx emissions along with continuing reductions of SO₂ emissions appear to contribute similarly to the significant and substantial reduction of wet deposition of H⁺ ion in the eastern US.
- Although both atmospheric O₃ concentration and dry deposition in the east show
 significant summertime P1-to-P4 reductions of 12% each, significant wintertime respective
 increases of 9% and 19% lead to overall significant reductions of 4% each. The numerically
 largest seasonal and regional reductions of both atmospheric O₃ concentration and dry deposition
 (17% and 15%) occur in summer in the northeast, coinciding with aggressive O₃-season NOx
 emissions controls during P3 and P4 in the region and upwind in the east.
- 991 Although conditions in the east permitting the formation of aerosol NO₃ may be limited 992 by the availability of NH₃, this appears to be diminishing over time. Increasing NI between P1 993 and P4 suggests that an acidic aerosol exists in the east, but is approaching neutralization due to 994 increased relative availability of NH₃ associated with reductions of aerosol SO₄ concentration. 995 Widespread wintertime numerical increases in the atmospheric concentrations of both aerosol 996 NO₃ and CASTNET PM are present between P1 and P3, despite reductions in estimated 997 emissions of both SO₂ and NOx (SS07b). Currently, this behavior appears to be largely reversed, 998 and this reversal is associated with continuing reductions of SO2 and NOx emissions between P3 999 and P4. Our findings suggest that additional P3-to-P4 reductions in emissions of SO₂, and 1000 especially NOx, have made progress in altering the chemical regime of the wintertime eastern 1001 US atmosphere so that future emissions reductions and their resulting reductions in aerosol 1002 concentrations may no longer be accompanied by sub-linear changes (or actual increases) in 1003 CASTNET PM.
- Monitoring data for several atmospheric species not collected by CASTNET, including
 NH₃, NO, NO₂, PAN, other oxidized organic nitrogen species, and NO_Y, as well as aerosol size
 distributions would have facilitated data analysis and interpretation. It is recommended that
 addition of these determinations be considered in future monitoring network upgrades.
 Uncertainties in estimates of many species' deposition velocities and emissions (especially NOx)
 argue strongly for their continued improvement.

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Table Captions

 Table 1. Summary of characteristics of 34 CASTNET monitoring sites providing data in current

 study (see Fig. 2) listed by region.

 Table 2. Notation.

Table 3. Mean (CV) atmospheric concentration; deposition velocity; precipitation rate; dry, wet and total deposition; and concentration and deposition metrics at 34 eastern US CASTNET and selected subsets of sites for the 5-year period 2005-2009 (see Table 2 for definitions).

Table 4. P1 and P4 emissions density and P1-to-P4 relative changes (%) in oxidized sulfur and nitrogen emissions, atmospheric concentration, and dry, wet, and total deposition.

Site	Name	Land Use ^a	Terrain ^b	Elevation (m)	Region ^d	Special Category ^e	Latitude (°N)	Longitude (°W)
ALH	Alhambra, IL	А	FL	164	MW	SR	38.9	89.6
ANA	Ann Arbor, MI	F	FL	267	MW		42.6	83.9
BVL	Bondville, IL	А	FL	212	MW	SR	40.1	88.4
DCP	Deer Creek State Park, OH	А	R	265	MW	SR	39.6	83.3
LYK	Lykens, OH	А	FL	303	MW	SR	40.9	83.0
OXF	Oxford, OH	А	Rc	284	MW	SR	39.5	84.7
PRK	Perkinstown, WI	А	R	472	MW	BKG	45.2	90.6
SAL	Salamonie Reservoir, IN	А	FL	249	MW	SR	40.8	85.7
UVL	Unionville, MI	А	FL	210	MW		43.6	83.4
VIN	Vincennes, IN	А	R	134	MW	SR	38.7	87.5
ARE	Arendtsville, PA	А	R	269	NE	SR	39.9	77.3
ASH	Ashland, ME	А	FL	235	NE	BKG	46.6	68.4
BEL	Beltsville, MD	S/A	FL	46	NE		39.0	76.8
СТН	Connecticut Hill, NY	F	R ^c	515	NE	HE	42.4	76.7
KEF	Kane Experimental Forest, PA	F	R	622	NE	SR	41.6	78.8
LRL	Laurel Hill State Park, PA	F	Cc	615	NE	SR	40.0	79.2
MKG	M.K. Goddard State Park, PA	F	Rc	384	NE	SR	41.4	80.2
PSU	Penn. State University, PA	А	R	378	NE	SR	40.7	77.9
WSP	Washington's Crossing, NJ	A/S	R	58	NE		40.3	74.9
WST	Woodstock, NH	F	Cc	258	NE		43.9	71.7
CAD	Caddo Valley, AR	F	Rc	71	SO	BKG	34.2	93.1
CDR	Cedar Creek State Park, WV	F	Cc	234	SO	SR	38.9	80.9
CND	Candor, NC	F	R	198	SO		35.3	79.8
COW	Coweeta, NC	F	Cc	686	SO		35.1	83.4
CVL	Coffeeville, MS	F	R	134	SO		34.0	89.8
ESP	Edgar Evins State Park, TN	F	R ^c	302	SO		36.0	85.7
MCK	Mackville, KY	А	R	353	SO	SR	37.7	85.0
PAR	Parsons, WV	F	Cc	510	SO	SR	39.1	79.7
PED	Prince Edward, VA	F	R	146	SO		37.2	78.3
SHN	Big Meadows, VA	F	М	1073	SO	HE	38.5	78.4
SND	Sand Mountain, AL	А	R	352	SO		34.3	86.0
SPD	Speedwell, TN	А	R	361	SO		36.5	83.8
SUM	Sumatra, FL	F	FL	14	SO	BKG	30.1	85.0
VPI	Horton Station, VA	А	Mc	920	SO	HE	37.3	80.6

 Table 1. Summary of characteristics of 34 CASTNET monitoring sites providing data in current study (see Fig. 2) listed by region.

^aLand use category within 10 km of site: A=agricultural; F=forested; S=suburban;

^bTerrain classification (local ground slope): FL=flat (<5°); R=rolling (5–15°); C=complex (>15°); M=montane;

^cAssumptions for applying the MLM may not be satisfied;

^dGeographical regions: MW=midwest; NE=northeast; SO=south; individual sites are identified by region in the same order shown in Figs. 5-8; ^eSpecial category sites: BKG=background; HE=high elevation; SR=source region. Table 2. Notation.

BKG	background
С	concentration
CV	coefficient of variation (100 SD/ \bar{x})
CASTNET PM	sum of aerosol SO4, NO3, and NH4 concentrations in total mass units
D	deposition
Drv HNO ₃	dry deposition of monitored oxidized nitrogen from HNO3, computed as (14 x HNO3 deposition/63), in units of mass of nitrogen
Dry N	dry deposition of nitrogen from HNO ₃ , aerosol NO ₃ , and aerosol NH ₄ , computed as (Dry OxN + Dry NH ₄)
, Drv NH₄	dry deposition of monitored reduced nitrogen from aerosol NH ₄ , computed as (14 x dry NH ₄ ⁺ ion deposition/18), in units of mass of nitrogen
Dry NO₂	$d_{\rm r}$ deposition of monitored oxidized nitrogen from aerosol NO ₂ computed as (14 x dry NO ₂ ion deposition/62) in units of mass of nitrogen
$Dry \Omega_2$	dry deposition of nonzone in units of mass of O
Dry OxN	ary deposition of monitored ovidized aitrogen from HNO ₂ and aerosol NO ₂ computed as ($Dry HNO_2 + Dry NO_2$)
Dry S	any deposition of nonidized suffur from SO ₂ and zerosol SO ₂ computed as (D_1 × D_2 + D_1 × D_2)
Dry SO.	dry deposition of oxidized suffur from SO ₂ computed as (22 x SO ₂ deposition /64) in units of mass of suffur
Dry SO2	dry deposition of oxidized suffur from 502, computed as $(32.35)^2$ deposition (04^2) in direction from $(350, 350)$ deposition $(140, 150)$ in this of mass of suffur from $(140, 150)$ in the second suffur from $(140, 150)$ is the second suffur from $(140, 150)$ in the second suffur from
	ary deposition of oxidized summinor actions of a computed as (52 × all y 504 into deposition y 50), in units of mass of summi-
	ratio of dry NH deposition to total N deposition, computed as Dry N/ total N
(D/T) NH4 (D/T) Over	ratio of dry Nr_4 deposition to total (dry plus wel) Nr_4 deposition, computed as Dry Nr_4 / Total Nr_4
	ratio of dry CAN deposition to total (dry plus wet) CAN deposition, computed as Dry CAN / Total CAN
(D/T) S	ratio of dry 5 deposition to total (dry plus wet) 5 deposition, computed as Dry 5/ lotal 5
E	
F	Tail (Sep, Uct, NoV)
HE	high elevation
HNO ₃	nitric acid (HNO ₃) concentration
MW	midwest
n	number of site-specific period values comprising the mean
N	monitored oxidized plus reduced nitrogen present as HNO3, aerosol NO3, and aerosol NH4, in units of mass of nitrogen
NE	northeast
NH ₄	ammonium (NH4*) ion concentration
NI	neutralization index, computed as [(NH4/18) – (NO $_3$ /62)] / 2(SO $_4$ /96), using airborne concentrations (in molar units)
NO ₃	nitrate (NO ₃ ⁻) ion concentration
O3	ozone (O $_3$) concentration, in ppb
OA	over all = period-site means averaged over sites ≈ annual average for 5-γ period
OxN	monitored oxidized nitrogen present as HNO3 and aerosol NO3, computed as (14 x HNO3/63) + (14 x NO3/62), in units of mass of nitrogen
PR	precipitation rate, in units of cm yr ¹ or cm season ¹
RFHNO ₃	ratio of dry HNO $_3$ deposition to dry OxN deposition, computed as Dry HNO $_3$ / Dry OxN
RFSO ₂	ratio of dry SO $_2$ deposition to dry S deposition, computed as Dry SO $_2$ / Dry S
RHNO ₃	ratio of oxidized nitrogen concentration from HNO₃ to oxidized nitrogen present as OxN, computed as (14 x HNO₃/63) / OxN
RNFHNO ₃	ratio of dry HNO $_3$ deposition to dry N deposition, computed as Dry HNO $_3$ / Dry N
RNFNH ₄	ratio of dry NH₄ deposition to dry N deposition, computed as Dry NH₄ / Dry N
RNHNO ₃	ratio of oxidized nitrogen concentration from HNO $_3$ to nitrogen present as N, computed as (14 x HNO $_3$ /63) / N
RNNH ₄	ratio of reduced nitrogen concentration from aerosol NH $_4$ to nitrogen present as N, computed as (14 x NH $_4$ /18) / N
RNO ₃	ratio of oxidized nitrogen concentration from aerosol NO ₃ to oxidized nitrogen present as OxN, computed as (14 x NO ₃ /62) / OxN, also = 1-
	RHNO3
RSO ₂	ratio of airborne oxidized sulfur concentration from SO ₂ to airborne oxidized sulfur present as S, computed as (32 x SO ₂ /64) / S
S	monitored oxidized sulfur present as SO ₂ and aerosol SO ₄ , computed as (32 x SO ₂ /64) + (32 x SO ₄ /96), in units of mass of sulfur
SO	south
SO ₂	sulfur dioxide (SO ₂) concentration
SO4	sulfate (SQ_a^2) ion concentration
Sp	spring (Mar, Apr, May)
SR	source region
Su	summer (Iun Iul Aug)
Total N	sum of total dry plus wet deposition of monitored nitrogen, computed as (Dry N + Wet N)
Total NH4	sum of dry plus wet deposition of monitored reduced nitrogen computed as (Dry NH4+ Wet NH4) in units of mass of pitrogen
Total OxN	sum of dry plus wet deposition of monitored oxidized nitrogen, computed as ($Dry OxN + Wet OxN$) in units of mass of nitrogen
	stand dry plus wet oxidized sulfur denosition computed integers, computed as (Dry S+ Wet S) in units of mass of sulfur
W/	winter (Der lan Sch)
₩ ₩/et H+	wet denosition of aqueous H ⁺ ion in units of mass of hydrogen
Wet N	wet deposition of aqueous through it units of mass of figurogen wat deposition of nitrogen from squeous NO $_{\rm e}$ ion and squeous NH, tion, computed as (Wet OvN + Wet NH.)
	wet deposition of neurogen norm aqueous No3 ion and aqueous Ni14 ion, computed as (Wet IOAN + Wet INTA) we have a finance of pitragen
	wet deposition of reduced hitrogen from aqueous NO-tion, computed as (14 x aqueous NO-tion deposition/16), in units of mass of mass of introgen
Wet OXN	wet deposition of oxidized nitrogen from aqueous NO3 ion, computed as (14 x aqueous NO3 ion deposition/b2), in units of mass of nitrogen
wet S	wet deposition of oxidized sulfur from aqueous SU ₄ ² ion, computed as (32 x aqueous SU ₄ ² ion deposition/96), in units of mass of sulfur

 Table 3. Mean (CV) atmospheric concentration; deposition velocity; precipitation rate; dry, wet and total deposition; and concentration and deposition metrics at 34 eastern US CASTNET and selected subsets of sites for the 5-year period 2005-2009 (see Table 2 for definitions).

	Site Grouping													
Region			E			NE	MW	SO	SR	HE	BKG			
Season	OA	W	Sp	Su	F	OA	OA	OA	OA	OA	OA			
n	34	34	34	34	34	10	10	14	15	3	4			
	2005-2009 Atmospheric Concentration, μg m ⁻³													
SO ₂	5.63 (51)	8.83 (55)	5.00 (47)	3.75 (54)	4.98 (52)	7.22 (49)	5.87 (41)	4.33 (48)	7.61 (28)	6.45 (27)	1.54 (16)			
SO ₄	3.52 (22)	2.49 (18)	3.33 (21)	5.13 (28)	3.12 (21)	3.32 (32)	3.41 (23)	3.74 (12)	3.94 (9)	3.49 (15)	2.32 (39)			
S (as S)	4.00 (40)	5.25 (48)	3.61 (37)	3.59 (39)	3.53 (41)	4.72 (45)	4.08 (35)	3.42 (34)	5.12 (22)	4.39 (23)	1.55 (25)			
HNO ₃	1.32 (32)	1.26 (35)	1.56 (33)	1.35 (40)	1.12 (35)	1.33 (42)	1.46 (24)	1.22 (28)	1.51 (20)	1.55 (10)	0.68 (43)			
NO ₃	1.04 (64)	1.85 (72)	0.92 (68)	0.55 (36)	0.85 (69)	0.78 (50)	1.92 (17)	0.60 (48)	1.38 (52)	0.56 (25)	0.73 (83)			
OxN (as N)	0.53 (40)	0.70 (41)	0.55 (41)	0.42 (35)	0.44 (44)	0.47 (43)	0.76 (15)	0.41 (30)	0.65 (31)	0.47 (1)	0.32 (49)			
NH ₄	1.35 (26)	1.20 (39)	1.20 (26)	1.80 (28)	1.19 (25)	1.23 (36)	1.56 (17)	1.28 (22)	1.58 (13)	1.22 (6)	0.78 (37)			
N (as N)	1.58 (29)	1.63 (40)	1.49 (30)	1.82 (26)	1.36 (30)	1.43 (37)	1.97 (16)	1.40 (23)	1.88 (19)	1.42 (4)	0.92 (40)			
O₃ (ppb)	30.8 (14)	24.6 (16)	38.4 (11)	34.5 (21)	25.6 (18)	30.3 (10)	30.6 (6)	31.4 (20)	30.4 (11)	39.4 (10)	28.0 (14)			
Vd, cm s ⁻¹														
HNO ₃	1.16 (21)	0.96 (27)	1.31 (19)	1.35 (19)	1.04 (24)	1.13 (28)	1.23 (11)	1.13 (21)	1.19 (21)	1.29 (7)	1.14 (24)			
SO ₂	0.31 (29)	0.31 (40)	0.32 (30)	0.33 (22)	0.29 (32)	0.30 (30)	0.36 (21)	0.28 (32)	0.36 (22)	0.30 (9)	0.25 (27)			
O ₃	0.17 (17)	0.09 (18)	0.17 (24)	0.26 (20)	0.15 (18)	0.16 (18)	0.15 (15)	0.17 (17)	0.17 (17)	0.19 (9)	0.16 (19)			
Aerosol	0.10 (21)	0.08 (28)	0.13 (19)	0.12 (22)	0.09 (24)	0.10 (28)	0.11 (12)	0.10 (22)	0.10 (22)	0.11 (15)	0.10 (18)			
PR (cm time ⁻¹)	119 (19)	25.7 (27)	30.3 (24)	32.7 (20)	29.6 (24)	119 (12)	100 (20)	132 (15)	115 (14)	126 (17)	124 (29)			
					Depos	ition, kg (ha y	/r) ⁻¹							
Dry SO ₂ (as S)	3.13 (62)	4.84 (70)	2.83 (61)	2.20 (58)	2.58 (62)	3.91 (60)	3.68 (51)	2.18 (61)	4.61 (35)	2.96 (14)	0.67 (12)			
Dry SO4 (as S)	0.40 (33)	0.21 (35)	0.46 (30)	0.63 (37)	0.29 (34)	0.36 (45)	0.40 (31)	0.42 (27)	0.44 (25)	0.43 (23)	0.23 (28)			
Dry S (as S)	3.53 (58)	5.06 (68)	3.29 (55)	2.83 (51)	2.87 (58)	4.27 (58)	4.09 (48)	2.60 (55)	5.05 (33)	3.40 (14)	0.91 (13)			
Wet S (as S)	5.22 (28)	3.59 (35)	5.65 (29)	7.53 (38)	3.97 (26)	5.96 (31)	4.97 (30)	4.86 (20)	6.25 (18)	5.13 (11)	3.16 (31)			
Total S (as S)	8.75 (36)	8.65 (47)	8.93 (35)	10.4 (38)	6.84 (35)	10.2 (38)	9.06 (37)	7.47 (25)	11.3 (19)	8.53 (5)	4.06 (27)			
Dry HNO₃ (as N)	1.28 (41)	0.89 (45)	1.65 (41)	1.60 (45)	0.98 (45)	1.24 (54)	1.48 (35)	1.16 (36)	1.48 (34)	1.46 (18)	0.62 (41)			
Dry NO ₃ (as N)	0.07 (66)	0.11 (73)	0.09 (68)	0.05 (48)	0.05 (75)	0.06 (67)	0.13 (21)	0.04 (62)	0.10 (56)	0.04 (20)	0.05 (86)			
Dry OxN (as N)	1.35 (41)	1.00 (42)	1.74 (42)	1.65 (44)	1.03 (45)	1.30 (54)	1.61 (33)	1.21 (36)	1.58 (34)	1.50 (18)	0.66 (36)			
Wet OXN (as N)	2.35 (19)	2.06 (28)	2.76 (22)	2.86 (22)	1.73 (23)	2.56 (24)	2.45 (15)	2.14 (11)	2.67 (12)	2.17 (17)	1.78 (26)			
	3.71 (22)	3.00 (25)	4.50 (25)	4.51 (25)	2.76 (24)	3.80 (29)	4.06 (19)	3.35 (13)	4.25 (14)	3.07 (5) 0.25 (10)	2.45 (28)			
Dry NH4 (as N)	0.35 (38)	0.24 (49)	0.38 (35)	0.52 (38)	0.26 (39)	0.31 (51)	0.41 (26)	0.34 (37)	0.41 (30)	0.35 (19)	0.18 (32)			
Wet NH ₄ (as N)	2.30 (27)	1.29 (33)	3.14 (35)	3.10 (31)	1.65 (34)	2.23 (29)	2.87 (21)	1.94 (14)	2.64 (22)	2.33 (25)	1.79 (54)			
Total NH ₄ (as N)	2.65 (26)	1.53 (32)	3.53 (33)	3.62 (28)	1.91 (31)	2.54 (29)	3.28 (19)	2.28 (15)	3.05 (21)	2.68 (20)	1.97 (52)			
Dry N (as N)	1.70 (40)	1.24 (40) 2 26 (26)	2.12 (40) 5 00 (27)	2.17 (41) 5 96 (24)	1.29 (43)	1.01 (53)	2.02 (31) 5 22 (17)	1.54 (35)	1.98 (33) 5 22 (14)	1.85 (18)	0.85 (32)			
Total N (as N)	4.00 (21) 6 36 (22)	5.50 (20) 4 59 (24)	5.90 (27) 8 02 (26)	5.90 (24) 8 13 (25)	5.56 (20) 4 67 (24)	4.79 (23) 6 41 (28)	5.52 (17) 7 34 (16)	4.08 (11) 5 63 (13)	5.52 (14) 7 30 (15)	4.50 (21) 6 35 (11)	5.56 (57) A A2 (3A)			
Wet H ⁺	0.26 (27)	0.22 (24)	0.22 (20)	0.37 (49)	0 21 (24)	0.34 (21)	0 20 (41)	0 25 (25)	0 31 (33)	0.26 (8)	0 15 (50)			
	39 5 (27)	16 3 (28)	<u>48</u> 2 (20)	66 1 (22)	28 5 (21)	28 2 (28)	36 6 (18)	42 3 (20)	40 4 (25)	56 4 (10)	37 3 (18)			
	55.5 (27)	10.3 (20)	+0.J (2J)	(00.1 (33)	Concentration	and denosit	ion metrics	+2.3 (30)	40.4 (23)	JU. 4 (10)	52.5 (10)			
RSO ₂	0.67 (16)	0.81 (11)	0.66 (16)	0 49 (23)	0.66 (17)	0 73 (12)	0.71 (7)	0.60 (18)	0 73 (7)	0 73 (5)	0 51 (17)			
RESO ₂	0.85 (9)	0.01 (11)	0.00(10) 0.82(12)	0.43(23)	0.86 (9)	0.89 (6)	0.71(7)	0.00 (10)	0.73(7)	0.75 (3)	0.75 (7)			
(D/T) S	0.85 (3)	0.53 (3)	0.82 (12)	0.74 (14)	0.80 (3)	0.89 (0)	0.88 (7)	0.80 (11)	0.91 (3)	0.87 (3)	0.73(7)			
NI	0.79 (10)	0.70 (13)	0.75 (12)	0.85 (10)	0.80 (9)	0.80 (8)	0.78 (6)	0.78 (13)	0.80 (7)	0.81 (5)	0.67 (22)			
RHNO₂	0 59 (22)	0 47 (43)	0.65 (19)	0.69(12)	0.60 (23)	0.62 (12)	0 42 (18)	0.68 (12)	0.55 (23)	0.73 (9)	0.51 (33)			
	0.41 (32)	0.53 (37)	0.35 (36)	0.31 (28)	0.00 (25)	0.38 (21)	0.58 (13)	0.32 (24)	0.35 (23)	0.75 (5)	0.49 (34)			
	0.41 (32)	0.33(37)	0.33 (30)	0.51 (20)	0.40 (33)	0.30 (21)	0.16 (15)	0.32 (24)	0.43 (23)	0.27 (20)	0.45 (34)			
	0.13 (10)	0.58 (6)	0.23 (17)	0.10(27)	0.10 (22)	0.20 (10)	0.10(13)	0.13(17)	0.10(13)	0.24 (0)	0.17 (20)			
	0.07 (0)	0.30 (0)	0.04 (3)	0.70(0)	0.05 (10)	0.06(0)	0.01 (2)	0.0 = (1)	0.00 (0)	0.07 (2)	0.00 (0)			
	0.95 (3)	0.72 (10)	U.30 (3)	0.37 (2)	0.95 (3)	0.90 (2)	0.31 (4)	0.30(1)	0.94 (3)	0.97 (1)	0.92(7)			
	0.75 (6)	0.72 (18)	0.77(7)	0.73 (8)	0.75(7)	0.70(5)	0.72(7)	0.75 (5)	0.75 (4)	0.79(1)	0.72 (14)			
	0.21 (15)	0.19 (35)	0.19 (21)	0.25 (22)	0.21 (22)	0.20 (17)	0.21 (15)	0.22 (15)	0.21 (10)	0.19 (2)	0.22 (25)			
	0.35 (28)	0.32 (31)	0.37 (27)	0.35 (30)	0.30 (32)	0.32 (38)	0.33 (20)	0.35 (27)	0.30 (25)	0.41 (19)	0.27 (12)			
(U/T) NH4	0.13 (32)	0.16 (39)	0.11 (34)	0.15 (39)	0.14 (38)	0.12 (39)	0.13 (29)	0.15 (30)	0.13 (28)	0.14 (31)	0.10 (18)			
	0.20 (29)	0.27 (32)		U.20 (31)	0.27 (34)	U.24 (38)	0.27(25)	0.27(27)	0.27 (20)	0.50 (25)	0.13(17)			
CASTNETPIM	5.91 (24)	5.55 (36)	5.45 (24)	7.49 (26)	5.16 (25)	5.33 (34)	6.89 (17)	5.62 (16)	6.90 (13)	5.28 (9)	3.83 (32)			

	E	missions	P1-to-P4 (P1-to-P4 Change in atmospheric concentration P1-to-P4 Change in depositic							
Region	ρ ₁ ^a (ρ ₄)	P1-to-P4 Change	SO ₂	SO ₄	S	Dry S	Wet S	Total S	Wet H⁺		
E	38 (19)	-50	-48	-33	-45	-49	-32	-40	-47		
MW	51 (23)	-55	-51	-33	-47	-48	-36	-42	-56		
NE	36 (19)	-47	-48	-34	-45	-51	-30	-40	-44		
SO	31 (16)	-47	-47	-32	-42	-49	-31	-39	-46		
-	ρ ₁ ^b (ρ ₄)	P1-to-P4 Change	HNO ₃	NO ₃	OxN as N	Dry OxN	Wet OxN	Total OxN	Wet H⁺		
E	19 (11)	-42	-34	-8	-24	-34	-29	-31	-47		
MW	21 (12)	-43	-30	-13	-21	-27	-28	-28	-56		
NE	23 (12)	-46	-39	-1 ^c	-29	-40	-34	-36	-44		
SO	17 (10)	-39	-32	-2 ^c	-24	-34	-25	-29	-46		

Table 4. P1 and P4 emissions density and P1-to-P4 relative changes (%) in oxidized sulfur and nitrogen emissions, atmospheric concentration, and dry, wet, and total deposition.

 $^{a}SO_{2}$ emissions density in units of kgS (ha yr)⁻¹. ^{b}NOx emissions density in units of kgN (ha yr)⁻¹. $^{c}p > 0.05$, NS.

Figure Captions.

Figure 1. Annual US SO₂ (TgS) and NOx (TgN) emissions estimates between 1990 and 2010 (Xing et al., 2013).

Figure 2. CASTNET monitoring sites and geographical regions considered in the current study (see Table 1).

Figure 3. Estimates of regional mean annual SO₂ and NOx emissions for four 5 year periods (P1, P2, P3, and P4) between 1990 and 2009, expressed as percentages of the 1990-1994 (P1) regional means. In the SO, NE, and MW mean P1 SO₂ emissions are 3.48, 1.68, and 3.27 TgS yr⁻¹, and mean P1 NOx emissions are 1.89, 1.06, and 1.35 TgN yr⁻¹ (Xing et al., 2013).

Figure 4. P1-to-P4 changes expressed in Table 3 units and as relative differences (%); see Table 2 for definitions of column headings.

Figure 5. Site- and region-specific concentration (μ g m⁻³, left (upper bar, P1; and lower bar, P4)) and P1-to-P4 relative changes (%, right and bottom) for: (**a**) S: monitored oxidized sulfur (μ gS m⁻³); (**b**) OxN: monitored oxidized nitrogen (μ gN m⁻³); (**c**) N: monitored oxidized plus reduced nitrogen (μ gN m⁻³); and (**d**) CASTNET PM: sum of aerosol SO₄, NO₃, and NH₄ (μ g m⁻³). Significance level of period-to-period relative changes indicated by bar fill: black, p≤0.05; and open, p>0.05 (NS). Sites are identified by region in the same order as shown in Table 1.

Figure 6. Site- and region-specific total deposition (kg (ha yr)⁻¹), left (upper bar, P1; and lower bar, P4)) and P1-to-P4 relative changes (%, right and bottom) for: (**a**) Total S: total dry plus wet oxidized sulfur deposition; (**b**) Total OxN: sum of dry plus wet deposition of monitored oxidized nitrogen; (**c**) Total N: sum of total dry plus wet deposition of monitored oxidized plus reduced nitrogen; and (**d**) Wet H⁺: wet deposition of H⁺. Significance level of period-to-period relative changes indicated by bar fill: black, $p \le 0.05$; and open, p > 0.05 (NS). Sites are identified by region in the same order as shown in Table 1.

Figure 7. Site- and region-specific seasonal aerosol NO₃ concentration (μ g m⁻³, left (upper bar: early period; and lower bar: later period)) and period-to-period relative changes (%, right and bottom) for: (a) P1-to-P3, W; (b) P3-to-P4, W; (c) P1-to-P4, W; and (d) P1-to-P4, Su. Significance level of period-to-period relative changes indicated by bar fill: black, p≤0.05; and open, p>0.05 (NS). Sites are identified by region in the same order as shown in Table 1.

Figure 8. Site- and region-specific wintertime CASTNET PM concentration (μ g m⁻³, left (upper bar: early period; and lower bar: later period)) and period-to-period relative changes (%, right and bottom) for: (a) P1-to-P3, W; (b) P1-to-P4, W; and (c) P3-to-P4, W. Significance level of period-to-period relative changes indicated by bar fill: black, p≤0.05; and open, p>0.05 (NS). Sites are identified by region in the same order as shown in Table 1.







Figure 4. P1-to-P4 changes expressed in Table 2 units and as relative differences (%); see Notation Section for definitions.

Region	Season	С	D	С	D	С	D	D	D	С	D	С	D	D	D
		SO2	Dry SO ₂	SO4	Dry SO ₄	S	Dry S	Wet S	Total S	03	03	NH4	$Dry NH_4$	Wet NH ₄	Total NH ₄
E	OA	-5.28 (-48)	-3.29 (-51)	-1.73 (-33)	-0.17 (-29)	-3.22 (-45)	-3.45 (-49)	-2.45 (-32)	-5.90 (-40)	-1.13 (-4)	-1.59 (-4)	-0.48 (-26)	-0.10 (-22)	-0.22 (-9)	-0.32 (-11)
E	W	-8.03 (-48)	-4.38 (-48)	-0.78 (-24)	-0.03 (-13)	-4.28 (-45)	-4.41 (-47)	-0.76 (-17)	-5.17 (-37)	2.04 (9)	2.65 (19)	-0.20 (-14)	0.00 (0)	0.05 (4)	0.05 (3)
E	Sp	-4.45 (-47)	-3.24 (-53)	-1.50 (-31)	-0.16 (-26)	-2.73 (-43)	-3.40 (-51)	-2.79 (-33)	-6.19 (-41)	-0.16 (0)	0.74 (2)	-0.48 (-29)	-0.11 (-23)	-0.29 (-8)	-0.40 (-10)
E	Su	-3.47 (-48)	-2.31 (-51)	-3.42 (-40)	-0.38 (-37)	-2.88 (-45)	-2.69 (-49)	-4.19 (-36)	-6.88 (-40)	-4.85 (-12)	-9.10 (-12)	-0.85 (-32)	-0.22 (-30)	-0.27 (-8)	-0.49 (-12)
E	F	-5.15 (-51)	-2.99 (-54)	-1.24 (-28)	-0.12 (-29)	-2.99 (-46)	-3.11 (-52)	-2.17 (-35)	-5.28 (-44)	-1.66 (-6)	-3.56 (-11)	-0.41 (-26)	-0.09 (-26)	-0.25 (-13)	-0.34 (-15)
MW	OA	-6.00 (-51)	-3.67 (-50)	-1.68 (-33)	-0.14 (-25)	-3.56 (-47)	-3.81 (-48)	-2.76 (-36)	-6.57 (-42)	-0.16 (-1)	0.34 (1)	-0.63 (-29)	-0.11 (-21)	-0.39 (-12)	-0.50 (-13)
MW	W	-8.55 (-50)	-4.90 (-47)	-0.90 (-26)	-0.02 (-7)	-4.58 (-47)	-4.92 (-46)	0.10 (3)	-4.82 (-34)	1.47 (7)	2.89 (25)	-0.24 (-12)	0.03 (9)	0.19 (14)	0.21 (13)
MW	Sp	-4.85 (-47)	-3.67 (-52)	-1.53 (-32)	-0.11 (-20)	-2.94 (-44)	-3.79 (-50)	-3.01 (-34)	-6.80 (-41)	0.93 (3)	4.27 (12)	-0.68 (-32)	-0.11 (-20)	-0.45 (-10)	-0.56 (-11)
MW	Su	-4.68 (-54)	-2.99 (-55)	-3.38 (-42)	-0.33 (-35)	-3.47 (-49)	-3.32 (-52)	-4.68 (-43)	-7.99 (-46)	-2.44 (-6)	-3.11 (-5)	-1.14 (-40)	-0.26 (-33)	-0.38 (-9)	-0.64 (-13)
MW	F	-5.87 (-52)	-3.74 (-53)	-0.94 (-23)	-0.06 (-15)	-3.25 (-46)	-3.80 (-51)	-2.69 (-39)	-6.49 (-45)	-0.91 (-4)	-1.14 (-4)	-0.46 (-25)	-0.06 (-17)	-0.59 (-22)	-0.65 (-21)
NE	OA	-6.61 (-48)	-4.18 (-52)	-1.74 (-34)	-0.18 (-33)	-3.89 (-45)	-4.36 (-51)	-2.55 (-30)	-6.91 (-40)	-1.12 (-4)	-3.43 (-8)	-0.45 (-27)	-0.11 (-26)	-0.18 (-7)	-0.29 (-10)
NE	W	-10.3 (-45)	-6.25 (-49)	-0.80 (-24)	-0.05 (-19)	-5.43 (-43)	-6.30 (-48)	-0.76 (-19)	-7.05 (-41)	2.93 (13)	2.82 (22)	-0.18 (-14)	-0.02 (-7)	-0.06 (-5)	-0.08 (-6)
NE	Sp	-5.41 (-48)	-4.07 (-57)	-1.50 (-34)	-0.17 (-29)	-3.21 (-45)	-4.24 (-55)	-3.33 (-37)	-7.57 (-45)	0.27 (1)	0.49 (1)	-0.43 (-29)	-0.11 (-25)	-0.42 (-14)	-0.54 (-15)
NE	Su	-4.30 (-48)	-3.01 (-53)	-3.45 (-41)	-0.43 (-42)	-3.31 (-45)	-3.44 (-51)	-4.27 (-30)	-7.71 (-37)	-6.65 (-17)	-11.9 (-15)	-0.82 (-32)	-0.25 (-34)	-0.28 (-8)	-0.54 (-13)
NE	F	-6.39 (-51)	-4.10 (-58)	-1.29 (-32)	-0.13 (-37)	-3.63 (-48)	-4.23 (-56)	-2.56 (-37)	-6.79 (-47)	-0.87 (-3)	-3.11 (-10)	-0.39 (-28)	-0.10 (-33)	-0.08 (-5)	-0.18 (-9)
SO	OA	-3.81 (-47)	-2.37 (-52)	-1.75 (-32)	-0.17 (-29)	-2.49 (-42)	-2.55 (-49)	-2.16 (-31)	-4.71 (-39)	-1.82 (-5)	-1.65 (-4)	-0.40 (-24)	-0.09 (-21)	-0.15 (-7)	-0.23 (-9)
SO	W	-6.04 (-48)	-3.28 (-52)	-0.69 (-22)	-0.03 (-12)	-3.25 (-45)	-3.31 (-50)	-1.44 (-28)	-4.75 (-40)	1.81 (7)	2.36 (15)	-0.18 (-16)	-0.01 (-4)	-0.15 (-10)	-0.16 (-10)
SO	Sp	-3.49 (-46)	-2.61 (-54)	-1.48 (-29)	-0.19 (-28)	-2.24 (-41)	-2.80 (-51)	-2.12 (-28)	-4.92 (-37)	-1.24 (-3)	-1.60 (-3)	-0.38 (-25)	-0.11 (-23)	-0.06 (-2)	-0.17 (-5)
SO	Su	-2.02 (-41)	-1.29 (-43)	-3.42 (-38)	-0.38 (-36)	-2.16 (-39)	-1.67 (-41)	-3.36 (-33)	-5.03 (-36)	-5.28 (-14)	-11.5 (-15)	-0.66 (-26)	-0.16 (-24)	-0.07 (-3)	-0.23 (-7)
SO	F	-3.74 (-50)	-2.17 (-56)	-1.43 (-30)	-0.16 (-34)	-2.35 (-44)	-2.33 (-53)	-1.66 (-32)	-3.99 (-42)	-2.75 (-9)	-5.62 (-15)	-0.39 (-26)	-0.10 (-29)	-0.18 (-12)	-0.28 (-16)
		HNO3	Dry HNO ₃	NO3	Dry NO ₃	OxN	Dry OxN	Wet OxN	Total OxN	PR	Wet H ⁺	Ν	Dry N	Wet N	Total N
E	OA	-0.67 (-34)	-0.69 (-35)	-0.09 (-8)	0.00 (-1)	-0.17 (-24)	-0.69 (-34)	-0.97 (-29)	-1.66 (-31)	1.61 (1)	-0.24 (-47)	-0.55 (-26)	-0.79 (-32)	-1.19 (-20)	-1.98 (-24)
E	W	-0.41 (-24)	-0.26 (-23)	0.20 (12)	0.03 (31)	-0.04 (-6)	-0.24 (-19)	-0.32 (-13)	-0.56 (-15)	1.37 (6)	-0.11 (-33)	-0.20 (-11)	-0.24 (-16)	-0.27 (-7)	-0.51 (-10)
E	Sp	-0.66 (-30)	-0.72 (-30)	-0.21 (-19)	-0.01 (-8)	-0.19 (-26)	-0.73 (-29)	-1.22 (-31)	-1.94 (-30)	-0.07 (0)	-0.28 (-55)	-0.57 (-28)	-0.84 (-28)	-1.50 (-20)	-2.34 (-23)
E	Su	-0.97 (-42)	-1.15 (-42)	-0.17 (-24)	-0.01 (-22)	-0.26 (-38)	-1.23 (-43)	-1.43 (-33)	-2.66 (-37)	0.98 (3)	-0.36 (-50)	-0.91 (-33)	-1.46 (-40)	-1.70 (-22)	-3.16 (-28)
E	F	-0.64 (-37)	-0.60 (-38)	-0.19 (-18)	-0.01 (-14)	-0.19 (-30)	-0.60 (-37)	-0.84 (-33)	-1.45 (-34)	0.61 (2)	-0.20 (-49)	-0.51 (-27)	-0.69 (-35)	-1.09 (-24)	-1.79 (-28)
MW	OA	-0.63 (-30)	-0.58 (-28)	-0.29 (-13)	0.00 (-2)	-0.21 (-21)	-0.59 (-27)	-0.96 (-28)	-1.55 (-28)	1.75 (2)	-0.24 (-56)	-0.70 (-26)	-0.70 (-26)	-1.35 (-20)	-2.04 (-22)
MW	W	-0.44 (-30)	-0.22 (-22)	0.27 (8)	0.05 (32)	-0.04 (-3)	-0.17 (-15)	-0.08 (-4)	-0.25 (-8)	5.20 (33)	-0.06 (-21)	-0.23 (-9)	-0.14 (-10)	0.11 (3)	-0.04 (-1)
MW	Sp	-0.51 (-22)	-0.47 (-20)	-0.64 (-27)	-0.03 (-14)	-0.26 (-24)	-0.49 (-19)	-1.12 (-27)	-1.62 (-24)	1.61 (6)	-0.29 (-62)	-0.78 (-29)	-0.60 (-19)	-1.57 (-18)	-2.18 (-19)
MW	Su	-1.08 (-38)	-1.10 (-34)	-0.34 (-30)	-0.02 (-21)	-0.32 (-36)	-1.12 (-34)	-1.40 (-32)	-2.52 (-33)	-1.43 (-5)	-0.39 (-65)	-1.20 (-39)	-1.38 (-34)	-1.78 (-21)	-3.16 (-25)
MW	F	-0.46 (-27)	-0.41 (-26)	-0.42 (-21)	-0.01 (-13)	-0.20 (-23)	-0.43 (-26)	-1.05 (-36)	-1.48 (-32)	-2.30 (-9)	-0.22 (-54)	-0.55 (-24)	-0.49 (-24)	-1.63 (-29)	-2.13 (-28)
NE	OA	-0.84 (-39)	-0.88 (-41)	-0.01 (-1)	0.00 (2)	-0.19 (-29)	-0.88 (-40)	-1.31 (-34)	-2.19 (-36)	9.81 (9)	-0.26 (-44)	-0.54 (-27)	-0.98 (-38)	-1.49 (-24)	-2.48 (-28)
NE	W	-0.48 (-25)	-0.43 (-31)	0.22 (19)	0.02 (29)	-0.06 (-8)	-0.41 (-29)	-0.58 (-20)	-0.99 (-23)	2.95 (13)	-0.11 (-30)	-0.19 (-12)	-0.43 (-26)	-0.64 (-16)	-1.07 (-18)
NE	Sp	-0.90 (-38)	-1.00 (-39)	-0.04 (-6)	0.00 (3)	-0.21 (-30)	-0.99 (-38)	-1.86 (-41)	-2.85 (-40)	-0.03 (0)	-0.33 (-55)	-0.54 (-29)	-1.11 (-36)	-2.28 (-30)	-3.39 (-31)
NE	Su	-1.20 (-48)	-1.54 (-50)	-0.12 (-21)	-0.01 (-23)	-0.29 (-43)	-1.55 (-50)	-1.78 (-36)	-3.33 (-42)	3.21 (11)	-0.39 (-42)	-0.93 (-34)	-1.81 (-47)	-2.06 (-25)	-3.87 (-31)
NE	F	-0.80 (-42)	-0.81 (-47)	-0.09 (-13)	-0.01 (-14)	-0.20 (-34)	-0.81 (-46)	-1.06 (-35)	-1.87 (-39)	2.87 (10)	-0.25 (-50)	-0.50 (-30)	-0.91 (-44)	-1.14 (-23)	-2.05 (-30)
SO	OA	-0.58 (-32)	-0.63 (-35)	-0.01 (-2)	0.00 (-1)	-0.13 (-24)	-0.63 (-34)	-0.70 (-25)	-1.33 (-29)	-3.12 (-2)	-0.22 (-46)	-0.44 (-24)	-0.72 (-32)	-0.85 (-17)	-1.57 (-22)
SO	W	-0.34 (-20)	-0.24 (-21)	0.15 (18)	0.02 (32)	-0.04 (-7)	-0.22 (-19)	-0.43 (-19)	-0.65 (-19)	-3.56 (-11)	-0.17 (-43)	-0.18 (-13)	-0.23 (-16)	-0.57 (-16)	-0.81 (-16)
SO	Sp	-0.59 (-29)	-0.78 (-34)	-0.03 (-6)	0.00 (-3)	-0.14 (-24)	-0.78 (-33)	-0.71 (-21)	-1.49 (-26)	-0.24 (-1)	-0.23 (-50)	-0.43 (-25)	-0.89 (-31)	-0.77 (-12)	-1.66 (-18)
SO	Su	-0.74 (-40)	-0.88 (-41)	-0.09 (-16)	-0.01 (-20)	-0.18 (-35)	-1.00 (-43)	-1.12 (-30)	-2.11 (-35)	0.78 (2)	-0.31 (-46)	-0.69 (-28)	-1.20 (-39)	-1.19 (-19)	-2.41 (-26)
SO	F	-0.66 (-39)	-0.69 (-44)	-0.09 (-16)	-0.01 (-16)	-0.17 (-33)	-0.70 (-43)	-0.56 (-28)	-1.25 (-35)	0.76 (2)	-0.16 (-45)	-0.47 (-28)	-0.80 (-40)	-0.73 (-21)	-1.53 (-28)

Colored entries are significant at p=0.05 level: blue=significant decrease; pink=significant increase.









