## **Response to Reviewer 3**

Reviewer 3# comments

Main points:

Improving and completing the description of the CCN data analysis:
The correction of the CN and CCN data for multiple charges and the transfer function is not trivial. The authors should either describe exactly how their data were corrected (e.g. in appendix, if not wished in the text) or at least reference the papers that describe the methods they used.

Response: A multiple charge correction and transfer function similar to that described in Deng et al. (2011) has been applied to each CN size distribution spectrum as well as to the CCN efficiency spectrum. As the reviewer suggested, the reference has been cited in the revised paper and added to the reference section:

Deng, Z., Zhao, C., Ma, N., Liu, F., Ran, L., Xu, W., Liang, Z., Liang, S., Huang, M., Ma, X., Zhang, Q., Quan, J., and Yan, P.: Size- resolved and bulk activation properties of aerosols in the North China Plain. Atmos. Chem. Phys., 11, 3835-3846, 2011.

• It is not always clearly distinguished between bulk and size-resolved CCN measurements. This could be easily done, e.g., by using different symbols and/or using distinct terms for the derived parameters (e.g. total CCN number concentration vs. CCN number concentration (measured behind the DMA, in size-resolved mode) and activation ratio vs. size-resolved activation ratio).

Response: As per the suggestion, we now use distinct terms to describe the parameters derived from bulk CCN measurements and size-resolved CCN measurements.

- Improving the Results section:

• The whole section on the CCN closure is totally inscrutable. It needs to be clarified which data sets exactly were used for which type of closure. I have read this section at least ten times and I still cannot figure out from the text which parameters are plotted in Figs. 5 and 6. I have done this kind of closure tests myself several times. Thus, I have an idea of what might be shown in the two figures but it is certainly not clearly described in the text. Also, because the method is not described properly, it is hard to understand what the results tell us. Therefore the description of the method as well as the discussion of the closure results needs to be revised.

The parameters I would calculate and compare with each other for your study would be the following:

a. The actually measured total CCN concentration directly measured by the CCNC.

b. The observed total CCN concentration obtained from size-resolved CCN measurements: the actually measured CCN efficiency spectrum is multiplied by the actually measured CN size distribution, which yields the CCN size distribution; then this is integrated over the whole size range to obtain the total CCN concentration.

c. The predicted total CCN concentration obtained from the average CCN spectrum: the average measured CCN efficiency spectrum (would correspond to the spectra plotted in your Fig. 1) is multiplied by the actually measured CN size distribution, which yields the CCN size distribution; then this is integrated over the whole size range to obtain the total CCN concentration.

d. The predicted total CCN concentration obtained from the average CN size distribution: the actually measured CCN efficiency spectrum is multiplied by the average measured CN size distribution, which yields the CCN size distribution; then this is integrated over the whole size range to obtain the total CCN concentration.

The following comparisons of parameters would be interesting:

o a and b: This would be the easiest way to check if in general the size-resolved CCN measurements are consistent with bulk measurements. This comparison cannot be done in this study since no parallel bulk and size-resolved measurements are available.

Response: The reviewer is right. We may conduct such measurements in the future.

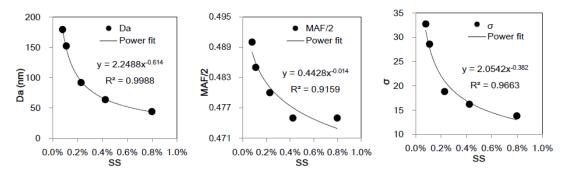
o b and c: For the time period of the size-resolved CCN measurements the total CCN concentration is 1) derived from the actually measured CCN spectrum and 2) calculated from the averaged CCN spectrum. With this comparison the influence of the variation of the chemical composition on the CCN concentration can be investigated, since the CN size distribution is the same for both the parameters. I guess this is what is shown in Fig. 5.

Response: The reviewer is right. Figure 5 in our study shows a comparison of "b" and "c", which is clearly described by the reviewer here. We call such a comparison a PO closure test, with the aim to investigate the influence of variations in chemical composition on CCN concentrations.

%a and c: For the time period when the bulk CCN measurements were

performed the measured total CCN concentration is compared with the calculated concentration from the averaged CCN spectrum. With this comparison again the variation of the chemical composition can be investigated but also how the bulk compares with the size-resolved CCN method. I guess this is what is shown in Fig. 6 but if so which average CCN spectra were then used for the calculation? The size-resolved CCN measurements, from which the CCN spectra can be derived, were made at different supersaturations than shown in Fig. 6.

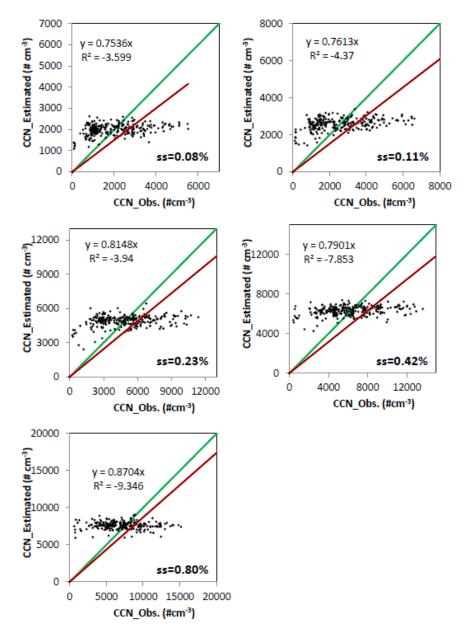
Response: The reviewer is right. Figure 6 in our study shows a comparison of "a" and "c", which is clearly described by the reviewer here. We call such a comparison an NPO closure test. The reviewer asked which average CCN spectra were used for the estimation of CCN number concentration. This was not described clearly in the paper. The SS levels were set at 0.2%, 0.5%, and 0.8% when making the bulk CCN measurements. These values should be corrected as effective SS, thus they are are 0.23%, 0.51% and 0.80. The averaged CCN efficiency spectra at SS = 0.23% and 0.80% (Fig. 1) is directly used to estimate the total CCN concentration from 1-25 June. The mean CCN efficiency spectra at SS = 0.51% is calculated using the exponential relationships developed from plotting the three CDF fit parameters as a function of SS ( $R^2 > 0.90$ ) (see the followingFigure).



**Fig. 1.**  $D_a$ , MAF, and  $\sigma$  as a function of SS.

% b and d: This comparison would be interesting for investigating the influence of the variation of the size distribution on the CCN concentration because here the CCN spectrum is the same for both the parameter. I guess that this reveals an even weaker correlation than for Fig. 6.

Response: As suggested by the reviewer, we plotted the comparison (see Fig. 2 below). Poor correlations are obtained, as expected by the reviewer. Particle size plays a key role in CCN activation, as shown in the literature. However, in this study, we mainly focus on investigating the influence of chemical composition on CCN activation. Also, in view of the current length of the paper, we prefer to not add



#### new material such as the suggested comparison.

Fig. 2. Estimated  $N_{CCN}$  as a function of observed  $N_{CCN}$  in the parallel observation (PO) closure test (under the assumption of an unchanged particle size distribution). The green solid line is the 1:1 line.

Overall, the whole section on the CCN closure has been revised following the reviewer's comments and suggestions. Which data sets were used for which type of closure test has been stated in the revised paper.

• If MAF<1 is only due to an error of the CDF fit (p. 16, l.9-10), why the whole discussion on Da vs. Dcut is needed? It should be enough to report Da values only since Dcut is anyhow practically the same. Btw,

whether MAF<1 is an error of the CDF fit could be shown by plotting the data points in addition to the fit lines in Fig. 1!

Response: The discussion has been removed. The measured data points have been added to Fig. 1.

- Improving the presentation quality:

• The manuscript is written in bad English. In some cases, grammar mistakes even change the meaning of the sentence. Apart from that, the methods and results are not presented in a clear and well-structured way. While revising this manuscript you should make sure that "the description of experiments and calculations is sufficiently complete and precise to allow their reproduction by fellow scientists" (cf. ACP review aspects).

Response: The manuscript has been proofread by a native English speaker.

Specific remarks:

p. 2, l. 5-7: "A gradual increase..." This sentence is weird. I think what you mean is "A gradual increase of the activation ratio (AR) with particle diameter suggests that aerosol particles consisted of a variety of different hygroscopicities."

Response: The sentence has been corrected.

p.2, I.7-10: Better formulation: "During pollution events the activation diameter (Da) measured at low supersaturation (SS) was significantly increased compared to background conditions; the increase was not observed when SS>0.4%" Response: Revised.

p. 4, l. 5: "aerosol particle formation" instead of "aerosol formation" Response: Revised.

p. 4, l. 17: add "particle number" before "size distribution" Response: Revised.

p. 5, l. 23-24: it must be "Rose et al., 2010; 2011; Gunthe et al., 2011" instead of "Rose et al., 2008; Gunthe et al., 2009" Response: Corrected.

p. 7, l. 12-13: "Aerosol chemical composition..." I would shift this sentence to the end of this paragraph. Response: Revised.

p. 8, l. 12: Which temperature sensor do you mean, inlet or sample temperature?

Response: Revised. It is the sample temperature.

p. 15, l. 15: Please give an error estimation for the supersaturation levels.

Response: Revised. The overall error  $(1\sigma)$  for the supersaturation levels is estimated to be < 3.5%.

p. 9, l. 22-23: What means "CCN data ... were filtered according to the instrument recorded parameters"? Do you mean that you sorted the data? Did you throw out any data points? If so, explain why?

Response: Some data was discarded due to instrument fluctuations or malfunction. For example, if the relative difference between the actual and set sample flow is larger than 4%, the data are flagged as invalid data. If the "Temp. Stability" equaled "0", the data are also flagged as invalid data. These flagged data are not used in the analysis.

p. 10, l. 24: please add "of a particle" after "describe the ability" Response: Revised.

p. 14, l. 16: it says that Fig.1 shows the "measured CCN efficiency", but this is not true. The figure shows the "CDF fit lines of the average CCN efficiency spectra". In fact, I would really appreciate if you could show the data points of the average spectra, too.

Response: Revised. The actual measured data points have been added.

p. 14, l. 21-23: "A gradual increase...". This is basically the same weird sentence as in the abstract. Please revise it according to my above suggestions.

Response: Revised.

p. 15, l. 1-2: Please reword this sentence: "The slope of AR around Da is less steep during polluted events than under background conditions, especially for low SS."

Response: It has been reworded as "The gentler slopes of AR around  $D_a$  during polluted events shows that the particle composition was more heterogeneous than the particle composition under background conditions."

p. 16, l. 5-7: "It should be noticed that... is equal to 1" I have no idea what these two sentences mean. Please revise.

Response: Revised.

p. 17, l. 18: change to "... may be due to the high amounts of organics freshly emitted..."

#### Response: Corrected.

p. 18, l. 18-19: I do not see that "Da\_BG and Da\_POL show larger variations at lower SS". Relatively seen, I would say that the largest variation is seen at SS = 0.11% and 0.23%. This would be also consistent with the variations for Da given in Tab. 1.

Response: Revised according to the reviewer's suggestion.

p. 21, l. 14-18: Here you discuss the correlation of AR at SS=0.2% with kappa\_chem, which is also shown in Fig. 8a. However, Fig. 9, which you also refer to at this point, shows AR at SS=0.5%. Can you explain?

Response: Corrected. The SS in Fig. 9 has been changed to 0.23%.

p. 23, l. 16-20: "Based on the CDF fit method,... was not observed when SS>0.4%." These are the same strange sentence as in the abstract. Please see my comments above.

Response: The sentence has been revised.

p.23, I. 25: Rephrase as "However, the case is more complex for particles originating from heavy pollution..."

Response: Rephrased.

Tab. 1:

- Please indicate the meaning of the values in this table. Are they the arithmetic mean values +/- standard deviation over the entire measurement period?

Response: The values are arithmetic mean values  $\pm$  one standard deviation.

- I think it would be sufficient to write only one digit after the comma for Da, Dcut, AR.

- Your MAF values decrease with increasing supersaturation. I think it makes physically no sense that all particles with ~ 300nm activate at SS=0.1% but not at 0.8%. If a particle is activated at SS=0.1% it is activated even more easily at 0.8%. Since the MAF values smaller than one seem to be an artifact, which you also mention on p. 16, 1.9-10, the whole distinction between Da and Dcut is meaningless. In my point of view, in this manuscript, you should present the Da and respective kappa\_a values only.

Response: The reviewer makes a good point. We have presented the  $D_a$  and respective kappa\_a values only in the revised version of the manuscript.

Fig. 1:

- Please plot the data points of the average measured CCN efficiency spectra, too.

Response: Revised.

Fig. 2:

- Please indicate in the caption the meaning of the error bars and that this graph is depicting the size resolved CCN data.

- Btw., it would be also interesting to calculate the kappa from the bulk measurements.

Response: Revised. In the manuscript, kappa\_chem, which represents the overall hygroscopicity of particles with particle sizes <1000 nm (PM1.0), was calculated from bulk chemical composition

#### measurements.

Fig. 4:

- What is plotted here? kappa\_cut as mentioned in the caption or kappa\_a as indicated in the x-axis label?

- I guess the particle size ranges as indicated in the individual panels should be consistent with the values of Tab. 1, but they are not. Response: The typos have been fixed.

#### Fig. 7:

- Why do you write "AR (\*100%)" in the y-axis labels? Is it not just the unitless AR that is plotted here?

- Please define the exact size ranges for the individual modes you integrated the particle number concentration (panel g, h).

- Please revise the figure caption. It is enough to write only once that the background conditions are plotted on the left and polluted conditions on the right.

Response: Revised.

#### Fig. 8:

- Please label the individual panels and indicate in the text accordingly. Response: Revised.

- In fact, it would be also interesting to see ALL the combinations of AR (for all SS) versus kappa\_chem, f44, and N\_CN,Acc, N\_CN,Ait, N\_CN, Nuc; also the cases for which no good correlation is found. Therefore I suggest either to add panels for the other combinations or, what is probably more convenient, to add a table showing the correlation coefficients of all the combinations.

Response: We are currently working on another paper which will include such comparisons. We also plan to include data measured at several other sites. The manuscript underway aims to optimize CCN parameterization schemes in climate models.

Fig. 9: y-axis label: please remove "(\*100%)". This is not valid, not for AR and especially not for kappa\_chem.

Response: Revised.

### Aerosol hygroscopicity and CCN activity during the AC<sup>3</sup>Exp13 campaign: <u>implicationsImplications</u> for CCN parameterization

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#### Abstract

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Aerosol hygroscopicity and cloud condensation nuclei (CCN) activity under background conditions and during polluted events are investigated during the Aerosol-CCN-Cloud Closure Experiment (AC<sup>3</sup>Exp) campaign conducted at Xianghe, China in summer 2013. A gradual increase in size-resolved activation ratio (AR) dependence onwith particle diameter  $(D_p)$  suggests that aerosol particles have lower hygroscopicity and heterogeneity of the composition during the campaign. Both had different hygroscopicities. During pollution events, the activation diameter (Da) and eut-off-diameter (D<sub>eut</sub>) were increased measured at low supersaturation (SS) was significantly at lower super saturations (SS) (<0.2%) dueincreased compared to

pollutions (e.g. biomass burnings); the background conditions. An increase was not observed when SS> $\geq$  0.4%, Hygroscopicity The hygroscopicity parameter kappa ( $\kappa$ ) are about was ~0.31-0.38 for particles in accumulation mode under background conditions and about. This range in magnitude of  $\kappa$  is ~20% higher than that  $\kappa$  derived under polluted conditions; however,  $\kappa$  are about 0.20-0.34 for <u>For</u> particles in 15 nucleation or Aitken mode at, k ranged from 0.20-0.34 for background cases, showing slightly lower than that duringand polluted eventscases. Larger particles were on average more hygroscopic than smaller particles. But the caseThe situation is more complex for heavy pollution particles with heavy pollutions due to the 20 diversities because of the diversity in particle compositions composition and mixing state. A non-parallel observation (NPO) CCN closure test shows about 30% 40% showed that uncertainties in N<sub>CCN</sub> prediction CCN number concentration estimates

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ranged from 30%-40%, which are associated with the changes of particle composition- as well as measurement uncertainties associated with bulk and size-resolved CCN methods. A case study showshowed that bulk CCN activation rationation increased with the increase of as total condensation nuclei (CN) number concentrations (N<sub>CN</sub>) inincreased on background days. In the The case, study also showed that bulk AR exhibited good correlation correlated well with  $\kappa_{ehem}$ , which is the hygroscopicity parameter calculated from chemical volume fractions. On the contrary, bulk AR declineddecreased with increase of increasing total N<sub>CN</sub> during polluted events, but it is was closely related to f44, which is usually associated with the particle

organic oxidation level. Our study highlights the importance of chemical composition on determining the particlesparticle activation properties, underlining and underlines the significance of long-term observationobservations of CCN under different atmospheric environments, especially those regions with heavy pollutions.

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pollution.

#### 1. Introduction

Indirect<u>The indirect</u> influence of aerosol particles on the radiative balance of the atmosphere through changes in cloud droplet number and <u>the persistence of clouds</u> (Twomey, 1974; Albrecht, 1989), carries the largest uncertainty amongst the presently known causes of radiative forcing (IPCC, 2007, 2013). Thus, better understanding of aerosol particle formation, growth, and activation is essential.

Field and laboratory experiments have been conducted with the aim of better characterizing the particle physical and chemical parameters impacting on cloud condensation nuclei (CCN) activation. Studies have addressed the relative importance

- of the size distribution, particle composition, and mixing state in determining CCN activation, but there are disagreements on the relative importance of these parameters
  (e.g<sub>7</sub>, Roberts et al., 2002; Feingold, 2003; Ervens et al., 2005; Mircea et al., 2005; Dusek et al., 2006a; Anttila and Kerminen, 2007; Hudson, 2007; Quinn et al., 2008; Zhang et al., 2008; Deng et al., 2013, Ma et al<sub>7</sub>, 2013). CCN closure studies are a
- useful approach to test our knowledge of the controlling physical and chemical factors
  and to help verify experimental results. The CCN number concentration, -\_\_(N<sub>CCN7</sub>) is
  usually predicted from measured aerosol properties such as <u>particle number</u> size
  distribution and composition or hygroscopicity based on K öhler theory. The closure
  between the measured and estimated N<sub>CCN</sub> is often <u>able to achieve in theachieved</u>
  under background <u>atmosphereatmospheric</u> conditions without heavy
  pollutionspollution (Chuang et al., 2000; Dusek et al., 2003; VanReken et al., 2003;
  Rissler et al., 2004; Gasparini et al., 2006; Stroud et al., 2007; Bougiatioti et al., 2009).

In urban/polluted areas, the particle size distribution is more complex with various composition of air mass (Lee et al., 2003; Alfarra et al., 2004; Zhang et al., <del>2004a</del>2004;

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Salcedo et al., 2006). The particleParticle activation is affected by the composition and the mixing state of the aerosol particles. It has been demonstrated that particles are more difficult to be activatedactivate during biomass burning plumesplume events (Mircea et al., 2005; Lee et al., 2006; Clarke et al., 2007; Rose et al., 2010, 2011;

- Paramonov et al., 2013; Lathem et al., 2013). Also, their activation ratios wereare reduced by secondary organics formed from oxidation of common biogenic emissions (VanReken et al., 2005; Varutbangkul et al., 2006; Mei et al., 2013) and black carbon (Dusek et al., 2006b; Kuwata et al., 2007). Other organic components (e.g-, organic acids) arehave been shown to activate more easily (Raymond and Pandis, 2002; Hartz
- et al., 2006; Bougiatioti et al., 2011), but still much less than inorganic species.
  Therefore, heavy polluted areas represent one of the most challenging cases to testtesting the understanding of parameters controlling CCN activation and growth. is challenging in heavily polluted areas. Furthermore, the main uncertainty in predicting the magnitude of the global aerosol indirect effects arises from those regions
  withunder the influence of urban emissions (Sotiropoulou et al., 2007). The study of aerosol-CCN closure and relationships insidewithin and in the outflow of heavily polluted areas is thus important.

East Asia, especially the Jing (Beijing)-Jin (Tianjin)-Ji (Hebei) region, is a fast developing and densely populated region including numerous megacities, where
anthropogenic aerosol emissions have increased significantly over recent years (Streets et al., 2008) and where aerosol loading is high and chemical composition is complex (Li et al., 2007a<sub>7</sub>; Xin et al., 2007). The high aerosol loading would have <u>a</u> significant influence on radiative properties, cloud microphysics, and precipitation (Xu, 2001; Li et al., 2007b; Xia et al., 2007; Rosenfeld et al., 2007; Lau et al., 2008;

Li et al., 2011).

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Field measurements of CCN have been made in East Asia where megacities are likely to be major sources of pollutants and CCN (Yum et al., 2007; Rose et al., 20082010, 2011; Gunthe et al., 2009; Yue et al., 2011; Liu et al., 2011; Zhang et al., 2012; Deng et al., 2011; Leng et al., 2013). Despite the significant accomplishments achieved by

these studies, limitations and uncertainties exist. As a recent example forover the 5 region of interest, Deng et al. (2011) over--predicted the concentrations of CCN at a site in the North China Plain by 19% when compared with direct measurements.

The aim of this paper is to study the aerosol hygroscopisity hygroscopicity and CCN activity under high aerosol loading in the polluted regions, and get some implications<u>conditions</u> and to parameterize CCN number concentrations by using CCN activation rationation (AR) to the as a proxy of the total number of aerosol particles in the atmosphere. A-A cumulative Gaussian distribution function (CDF) fit model is applied to bothaerosol data collected under clean and polluted aerosolsconditions to probeexamine the influencesinfluence of size distribution, 15 heterogeneity of chemical composition, and mixing state on CCN activity. The hygroscopicity parameter ( $\kappa$ ) is derived using K öhler theory to study <u>aerosolsaerosol</u> hygroscopicity duringon clean days and during polluted events. In the CCN closure study, besides in addition to the parallel observation (PO) closure test, we employapply the CCN efficiency spectrum to non-simultaneously-simultaneous condensation nuclei 20 (CN) and <u>bulk\_CCN</u> observation<u>observations</u>, namely <u>the</u> non-parallel observation (NPO) closure test, to estimate  $N_{CCN}$  in order to test the impacts from particle composition. Finally, the relationship between bulk\_AR and aerosol physical and chemical properties is examined to imply a possibility of CCN parameterization in climate models. 25

#### 2. Measurement<u>Measurements</u> and data

An intensive <u>field\_campaign, named\_called\_the</u> Aerosol-CCN-Cloud Closure Experiment (AC<sup>3</sup>Exp<del>),</del>) was conducted during June and July of 2013 at the Xianghe Atmospheric Observatory (39.<del>798N<u>798</u>°N</del>, 116.<del>958E<u>958</u>°E</del>; 35 m above sea level<del>),</del>)

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located about 60 km southeast of the Beijing metropolitan area. This site is surrounded by agricultural land, densely occupied residences, and light industry. Sitting between two megacities (with Beijing to the northwest and Tianjin to the southeast) and less than 5 km west of the local town center (with a population of 50,000), the site experiences frequent pollution plumes. Depending on the wind

10 direction, instruments at the Xianghe site detect pollutants of urban, rural, or mixed origins, experiencing both fresh biomass burning emissions and advected aged aerosols. <u>More informationDetails</u> about the measurement location and meteorological conditions has been described byat the site can be found in Li et al. (2007, 2011).

#### 15 2.1 Instrument Instruments and measurement measurements

In the campaign, bulk<u>Bulk</u> CCN activation was measured from 1 June to 25 June 2013. Size-resolved CCN <u>waswere</u> measured from 7 July to 21 July 2013. Aerosl <u>Aerosol</u> particle size <u>distributiondistributions</u> (10-700 nm) <u>waswere</u> measured from 1 June to 25 June 2013 and <u>from</u> 7 July to 21 July 2013. During 1-25 June 2013, a Scanning Mobility Particle Sizer (SMPS) (DMA; TSI 3081, CPC; TSI 3776)-was used independently for size distribution measurement; and frommeasurements. From 7 July to 21 July 2013, it was combined with a Droplet Measurement Technologies - Cloud Condensation Nuclei Counter (DMT-CCNc) (Lance et al., 2006) and used for size-resolved CCN measurement. Aerosol chemical composition was measured from

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31 May to 30 June 2013. Therefore, the<u>measurements. The</u> CCN efficiency spectrum (was<u>Fig.-1) is</u> derived from the-size-resolved CCN observationobservations made from 7 July to 21 July 2013. The aerosol particle size distribution data independently measured by the SMPS <u>and bulk CCN measurements</u> from 1 June to 25 June 2013 combingcombined with the derived CCN efficiency spectraspectrum (Fig. 1) is used for the NPO CCN closure test. <u>Aerosol chemical composition was measured from 31</u> May to 30 June 2013.

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The aerosol inlet for the size distribution measurements was equipped with a TSI Environmental Sampling System (Model 3031200), which consists of a standard PM10 inlet, a sharp-cut PM1 cyclone, and a bundled nafion dryer. After dried through the nafion bundle, the sample flow with relative humidity (RH) of <\_30% was sent into the SMPS for the aerosol size distribution measurements (10-700 nm). Meanwhile, the <u>CCN number concentrations (bulk N<sub>CCN</sub>)</u> at different super-saturations were(<u>SS</u>) was measured, using a continuous-flow CCN counter from the DMT-CCN<sub>C</sub>. Each <u>bulk</u> CCN measurement cycle included three super-saturations<u>SS levels</u>: 0.223%, 0.551% and 0.880%. The scanning times for those <u>super saturationsSS levels</u> were set asat 7, 5, and 5 minutes, respectively.

The size-resolved CCN efficiency spectra were measured by coupling the same DMT-CCN<sub>C</sub> used with the SMPS (Rose et al., 2008). In this setup the particles are rapidly dried with RH of  $\ll$  30% upon entering the Differential Mobility Analyzer (DMA<sub>7</sub>). Thus, size selection is effectively performed under dry conditions, and the relative deviations in particle diameter should be <\_1% except for potential kinetic limitations (Mikhailov et al., 2009). The sample flow exiting the DMA was split into two parts, with 0.3 lpm for the CPC and 0.5 lpm for the CCN<sub>C</sub>. The DMA, controlled

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<sup>25</sup> by the TSI-AIM software, scanned one size distribution every five minutes. The

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CCN<sub>C</sub> was operated at a total flow rate of 0.5 lpm with thea sheath-to-aerosol flow ratio of 10. The inlet RH for CCN is < was < 30%. During the field campaign, the averaging mean sample temperature and pressure as measured by the  $CCN_{C}$  sensors were  $(23.5\pm1.6)$ - $\mathbb{C}$  and  $(985.5\pm3.6)$  hPa. The deviations were determined by the

- 5 measurement uncertainties. ForSS levels of CCN<sub>C</sub> were calibrated with ammonium sulfate before and after the field campaign, following procedures outlined in Rose et al. (2008). During each CCN measurement cycle, SS was set to 5 different values: calibrated effective SS are 0.08,%, 0.11,%, 0.23,%, 0.42,%, and 0.80%. The overall error  $(1\sigma)$  for the SS levels was estimated to be < 3.5%. The completion of a
- full measurement cycle took 60 min (20 min for the SS = 0.08% and 10 min for each 10 of the rests). The supersaturations of CCN<sub>C</sub>-were calibrated with ammonium sulfate both before and after the campaign, following Rose's procedures (Rose et al., <del>2008).</del>the other SS levels).
- The measurement of non-refractory submicron (NR-PM1)-aerosol species including organics, sulfate, nitrate, ammonium, and chloride were made with an Aerodyne 15 Aerosol Chemical Speciation Monitor (ACSM) (Sun et al., 2012) is also conducted during the campaign.). The ACSM uses the same aerosol sampling, vaporization and ionization modules as the Aerosol Mass Spectrometer (AMS) (DeCarlo et al., 2006), but removes the size components, i.e., no size information. During the field campaign, ambient aerosol was aerosols were drawn inside through a 1/2 inch (outer diameter) 20 stainless steel tube at a flow rate of ~3 L min<sup>-1</sup>, of which ~84 cc min<sup>-1</sup> was sub-sampled into the ACSM. An URG cyclone (Model: URG-2000-30ED) was also supplied in front of the sampling inlet to remove coarse particles with a size cut-off of 2.5 mm. Before sampled into the ACSM, aerosol particles are dried by Silica gel desiccant. The residence time in the sampling tube is ~5 s. The ACSM was operated at

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a time resolution of ~15 min with a scan rate of mass spectrometer at 500 ms amu<sup>-1</sup> from m/z 10 to 150. Regarding the calibration of the ACSM, the mono-dispersed, size-selected 300 nm ammonium nitrate particles within a range of concentrations were sampled into both the ACSM and a condensation particle counter (CPC). The ionization efficiency (IE) was then determined by comparing the response factors of the ACSM to the mass calculated with the known particle size and the number concentrations from the CPC. Once the IE is determined, the changes of in the internal standard naphthalene or air ions, e.g., m/z 28 (N<sub>2</sub><sup>+</sup>) or m/z 32 (O<sub>2</sub><sup>+</sup>)<sup>+</sup>), can be used to account for the degradation of the detector. The other Other details including the sampling instrument, aerosol setup, operations, and calibrations were detailed calibration are given in Sun et al. (2012) and Ng et al. (2011).

In addition to the ACSM, the black carbon (BC) in PM2.5 was simultaneously measured at a time resolution of 5 min by a BC analyzer (Aethalometer, Model AE22, Magee Scientific Corporation). The campaign averaged mass concentrations of BC are ~4.2  $\mu$ g m<sup>-3</sup>, and the averaged mass fraction is about 6%, with maximum of 18% and minimum of 2%. During the experiment period, the campaign area was generally hot and wetmoist, with an average temperature of 23.6 °C and an average ambient relative humidity (RH) of 72.3%.

#### 2.2 Data

20 The raw CCN data for both bulk and size-resolved CCN measurements were firstly<u>first</u> filtered according to the instrument recorded parameters (e.g<sub> $\tau$ </sub>, temperature and flow). For example, if the relative difference between the actual and set sample flows was larger than 4%, the data are flagged as invalid data. If the "Temp. Stability" was recorded as "0", the data is also flagged as invalid data due to instrument

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Manuscript for Atmos. Chem. Phys., Zhang et al.: CCN measurement during AC<sup>3</sup>Exp <u>16 April,24 Oct.</u> 2014 fluctuations. These flagged data are not used for further analysis. A multiple charge correction and transfer function (Deng et al., 2011) is applied forto each CN size distribution spectrum as well as to the CCN efficiency spectrum. The CCN activation ratio (AR) is just the ratio of N<sub>CCN</sub>/N<sub>CN</sub>. In order to To examine the CCN activity under different cases, we classified conditions, the size-resolved CCN efficiency data was classified as polluted andor as background conditions based on the aerosol loading as well as the synchronism synchronal surface horizontal winds wind data. Basically, the polluted Polluted conditions are with-identified when the total CN number <u>concentration</u> ( $N_{CN}$ ) > 15000 cm<sup>-3</sup> and when the airflow came from southeast/east and the background. Background cases are withidentified when N<sub>CN</sub> <\_15000 cm<sup>-3</sup> always associated with theand when winds were from the west or northwest. NCN is the total aerosol number concentrations concentration with a particle size range of 10-700 nm. Here, the background refers to a regional background condition which represents a well-mixed atmosphere without influencedunaffected by local emissions, e.g., biomass burning. Bulk measurementmeasurements of total CCN number concentrations at SS levels of 0.2%, 0.5% and 0.8% could lead to a considerable underestimation of CCNN<sub>CCN</sub> under polluted conditions (Deng et al., 2011) due to water depletion inside the column (Lathern and Nenes, 2011). Therefore, in this study the, data points with  $N_{CN} > 25000$  cm<sup>-3</sup> were excluded. In the closure study, CCN size distributions were calculated by multiplying the fitted campaign-averaged CCN efficiency spectra (spectrum (a 3-parameter CDF fit) with the aerosol particle number size distribution. TotalThe total N<sub>CCN</sub> was then obtained by integrating the CCNsize-resolved N<sub>CCN</sub> over the whole size range. The Aerosol mass concentrations and mass spectra were processed using ACSM standard data analysis software (version 1.5.1.1). The detailed Detailed procedures for the data analysis have been

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described in Ng et al. (2011) and Sun et al. (2012). The campaign averaged mass concentrations of BC are ~4.2 µg m<sup>3</sup>, and the averaged mass fraction are about 6%, with maximum of 18% and minimum of 2%.

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#### 3. Theory

5 As proposed by Petters and Kreidenweis (2007),  $\kappa$  can be used to describe the ability of particles to absorb water vapor and act as CCN. Based on K öhler theory (K öhler, 1936),  $\kappa$  relates the dry diameter of aerosol particles to the critical water vapor SS. According to measurements and thermodynamic models,  $\kappa$  is zero for insoluble materials like soot or mineral dust, however. However, their hygroscopicity would be <del>changed</del><u>changes</u> due to the aging process of the soot and mineral dust and, so the  $\kappa$ value <u>then is thus  $\geq 0$ . The magnitude of</u>  $\kappa$  is 0.1 for secondary organic aerosols, 0.6 for ammonium sulfate and nitrate, 0.95-1 for sea salt (Niedermeier et al., 2008), and 1.28 for sodium chloride aerosols. The effective hygroscopicity of mixed aerosols can be approximated by a linear combination of the  $\kappa$ -values of the individual chemical components weighted by the volume or mass fractions, respectively (Kreidenweis et 15 al., 2008; Gunthe et al., 2009). In this study, we calculated  $\kappa$  based on both size-resolved CCN measurements and bulk chemical composition observations made during the <u>field</u> campaign. The method to <u>derived</u> derive  $\kappa$  is described below.

#### 3.1 Derivation of Ka and Keut-

Particle hygroscopicity The magnitude of  $\kappa$  were was derived from the measured size 20 -resolved CCN activated fraction using  $\kappa$ -Köhler theory (Petters and Kreidenweis, 2007). In  $\kappa$ -K öhler theory, the water vapor saturation ratio over an aqueous solution droplet S is given by:

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$$S = \frac{D^3 - D_p^3}{D^3 - D_p^3 (1 - \kappa)} \exp\left(\frac{4\sigma_w M_w}{RT\rho_w D}\right)$$
(1)

where *D* is the droplet diameter,  $D_p$  is the dry diameter of the particle,  $M_w$  is the molecular weight of water,  $\sigma_w$  is the surface tension of pure water,  $\rho_w$  is the density of water, *R* is the gas constant, and *T* is the absolute temperature. When  $\kappa$  is greater than 0.1, it can be conveniently derived expressed as:

 $\kappa = \frac{4A^3}{27D_p^3 S_c^2}$ 

 $A = \frac{4\sigma_{\rm w}M_{\rm w}}{RT\rho_{\rm w}}$ 

(2)

where S<sub>c</sub> is the particle critical supersaturation and is derived using the approach described by Rose et al<sub>72</sub> (2008). The characteristic S<sub>c</sub> of the size selected CCN is represented by the supersaturation<u>SS level</u> at which AR reaches 50%. For the parameters listed above,  $T = _298.15K_{15}K$ ,  $R = _ 8.315 J K^{-1} mol^{-1}$  (gas constant),  $\rho_w = _ 997.1 \text{ kg m}^{-3}$ ,  $M_w = _ 0.018015 \text{ kg mol}^{-1}$ , and  $\sigma_w = _ 0.072 JmJ m^{-2}$ . Note that values derived from CCN measurement data through K öhler model calculations assuming the surface tension of pure water have to be regarded as "effective hygroscopicity parameters" that account not only for the reduction of water activity by the solute ("effective Raoult parameters") but also for surface tension effects (Petters and Kreidenweis, 2007).

In this study, a parameter <u>called</u>  $\kappa_a$ , which characterizes the average hygroscopicity of CCN-active particles in the size range around activated diameters (D<sub>a</sub>), is calculated

from the data pairs of SS and  $D_a$  based on the  $\kappa$ -K öhler theory. Similarly, a parameter  $\kappa_{eut}$  is also derived from the data pairs of SS and a critical dry particle diameter ( $D_{eut}$ )

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#### 3.2 Derivation of K<sub>chem</sub>

For a given internal mixture,  $\kappa$  can be predicted by a simple mixing rule on the basis of chemical volume fractions,  $\varepsilon_i$  (Petters and Kreidenweis, 2007; Gunthe et al., 2009);):

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based on the *k* Köhler theory, which characterizes the average hygroscopicity of

aerosol particles in the size range around D<sub>cut</sub>. Note the D<sub>cut</sub> is the diameter when the

AR=50% regardless of maximum activation fraction (MAF) smaller than 1 or equal to

1. Whereas, D<sub>a</sub> is the diameter when the AR=MAF/2. The discrepancy of D<sub>cut</sub> and D<sub>a</sub>

can reflect the mixing state and chemical heterogeneity of aerosol particles,

$$\kappa_{\rm chem} = \sum_{i} \varepsilon_i \kappa_i$$

(4)

where,  $\kappa_i$  and  $\varepsilon_i$  are the hygroscopicity parameter and volume fraction for the individual (dry) componentcomponents in the mixture withand *i*\_is the number of components in the mixture. We derive  $\varepsilon_i$  from the particle chemical composition measured by the ACSM. Measurements from the ACSM show that the composition of submicron particles was dominated by organics, followed by nitrate, ammonium, and sulfate. The contribution of chloride was negligible (with a\_volume fraction of about < 2%). The analysis of the anion and cation balance suggests that anionic species (NO<sub>3</sub><sup>-7</sup>, SO<sub>4</sub><sup>2-7</sup>) were essentially neutralized by NH<sub>4</sub><sup>+</sup> over the relevant size range. For refractory species, BC represented a negligible fraction of the total submicron aerosol volume (less than about-3%). The seaSea salt and dust are usually with coarse mode particles with particle size of >sizes > 1 ++µm (Whitby, 1978), the). The contribution of such types of aerosols areis thus expected to be negligible for the size range of

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<<u>sizes < 1000 nm</u>. Therefore, <u>thosethe</u> submicron particles measured by <u>the ACSM</u>

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were-mainly consisting consisted of Organics organics, (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>, and NH<sub>4</sub>NO<sub>3</sub>. The particle hygroscopicity is thus the volume average of the three participating species:

 $\kappa_{\text{chem}} = \kappa_{\text{Org}} \epsilon_{\text{Org}} + \kappa_{(\text{NH4})2\text{SO4}} \epsilon_{(\text{NH4})2\text{SO4}} + \kappa_{\text{NH4NO3}} \epsilon_{\text{NH4NO3},}$ (5)

The values of  $\kappa$  are 0.67 and 0.61 for (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> and NH<sub>4</sub>NO<sub>3</sub>, respectively, and are

5 derived from previous laboratory experiments (Petters et al., and Kreidenweis, 2007). For Organics, we used a The following linear function between  $\kappa_{Org}$  and  $f_{44}$  derived by Mei et al., (2013) was used to estimate the  $\kappa_{Org}$  in our study. The formula is written as:  $\kappa_{\text{Org}} = 2.10 \times f_{44} - 0.11$  (Mei et al., 2013). By applying the function, calculated  $\kappa_{\text{Org}}$ . The mean value of  $\kappa_{Org}$  during the field campaign is 0.115±0.019 during the observed 10 period. Species volume fractions were derived from mass concentrations and densities of participating species. The densities of  $(NH_4)_2SO_4$  and  $NH_4NO_3$  are 1770 kg m<sup>-3</sup> and 1720 kg m<sup>-3</sup>, respectively. And the The density of organics is 1200 kg m<sup>-3</sup> (Turpin et

#### 4. Results and discussion

al., 2001).

#### 4.1 CDF fit and parameters derived from the CCN efficiency 15

The spectra of measured CCN efficiency atunder both polluted and background conditions were fitted with a cumulative Gaussian distribution function (CDF; \_\_(Rose et al., 2008):

$$f_{N_{\rm CCN}/N_{\rm CN}} = a \left( 1 + \operatorname{erf}\left(\frac{D - D_{\rm a}}{\sigma_{\rm a}\sqrt{2}}\right) \right),\tag{6}$$

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where, the maximum activated fraction, MAF=, is equal to 2a,  $D_a$  is the midpoint activation diameter  $\underline{D}_a$ , and  $\underline{\sigma}_a$  is the CDF standard deviation  $\underline{\sigma}_a$ . The <u>These</u> parameters were determined for each spectrum. If MAF=1 by changing the parameter "a" to 0.5, the spectrum is characteristic for internally mixed aerosols with

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homogeneous composition and hygroscopicity of the particles. The 3-parameter fit results represent the average activation properties of the aerosol particle fraction. During the field campaign period of AC<sup>3</sup>Exp 2013, we measured, about 1200 size-resolved CCN efficiency spectra for atmospheric aerosols at SS levels of 0.08% to 0.80%. Fig.% were measured. Figure 1 shows campaign-averaged spectra of both measured and fitted CCN efficiency at SS levels of 0.08%, 0.11%, 0.23%, 0.42%%, and 0.80% for background and polluted conditions. The slope of AR with respect to diameterdiameters near  $D_a$  in Fig. 1 provides information on about the heterogeneity of the composition for the size-resolved particles. For an ideal case when all CCN-active particles have the same composition and size, a steep change of AR from 0 to MAF would be observed as D reaches D<sub>a</sub>. A gradual increase in size-resolved AR with particle diameter suggests that some of theaerosol particles have lower hygroscopicity and/or heterogeneity of the composition and are not able to activate at the given SS than others, consisted of different hygroscopicities. The slope gentler slopes of AR around D<sub>a</sub> during polluted events shows gentler increase compared with thoseshow that the particle composition was more heterogeneous than the particle composition under background conditions, especially at low SS. Overall, significant. Significant differences of thein size-resolved CCN efficiency spectra forunder polluted and clean conditions at lower SS levels have been derived. The different shape just suggests aerosols hygroscopisity and shapes suggest that aerosol hygroscopicity and CCN activity would be affected by the local emission sources (a e.g., biomass burning).

The three parameters (MAF,  $D_a$ , and  $\sigma$ ) of CCN efficiency spectra derived from the 3 parameter CDF fits as well as  $D_{eut}$ ,  $\kappa_a$  and  $\kappa_{eut}$  under polluted and clean conditions were also summarized in Table 1. Da\_POL and Da\_BG in Table 1 are defined as

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*Manuscript for Atmos. Chem. Phys.,* Zhang et al.: CCN measurement during AC<sup>3</sup>Exp <u>16 April,24 Oct.,</u> 2014 activation diameter under polluted and clean conditions respectively. Dcut\_POL and Dcut\_BG are defined as cut-off diameter under polluted and clean conditions respectively.

4.1.1 Activation diameter (D<sub>a</sub>)

The three parameters (MAF, D<sub>a</sub>, and σ) describing the CCN efficiency spectra derived from the 3-parameter CDF fits, as well as κ<sub>a</sub> under polluted and clean conditions, are summarized in Table 1. Activation diameters under polluted and clean conditions are denoted as Da POL and Da BG in Table 1, respectively. As expected, Da decreaseddecreases with increasing SS under both background and polluted conditions. At a given SS, D<sub>a</sub>POL are largeris greater than D<sub>a</sub>BG. That means the suggesting that particles at under polluted easesconditions would be activated at a larger diameter. ButAs SS increases, the difference of D<sub>a</sub>-POL and D<sub>a</sub>-BG reduced and was close to each other with the increasing of SS. Accordingly, D<sub>eut</sub>between D<sub>a</sub>-POL and D<sub>a</sub>-BG dependence on SS showed similar changes to D<sub>a</sub>-But, as stated previously, D<sub>eut</sub> shows a little bit larger than D<sub>a</sub>-Because D<sub>eut</sub> is defined as the diameter when AR is up to 50%, but based on the CDF fit, the "MAF/2" is smaller

than 50% and thus results in a smaller D<sub>a</sub>. decreases.

4.1.2 Maximum Activated Fraction (MAF)

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GenerallyIn general, aerosols with <u>a</u> more uniform and homogenous chemical composition or with <u>a</u> core-shell structure would have a higher MAF. <u>MAF\_POL and MAF\_BG in Table 1 are defined as maximum activation fractionThe MAF</u> under polluted and background conditions <u>is denoted by MAF\_POL and MAF\_BG in Table 1</u>, respectively-the. Values of MAF\_BG and MAF\_POL are from 0.95-0.98 and

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<u>from 0.94-0.98</u>, respectively at different SS. No significant discrepancies of <u>in</u> MAF were are observed between polluted and background conditions. It should be noticed that MAF are just closer to 1 but equal to 1. Because MAF for pure ammonium sulphate particles (0.05mol  $L^{-1}$ ) is equal to 1; Furthermore, the observations have

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indicated <u>Observations show</u> that particles can be able to activate to CCN completely when particle <u>diameter</u> ><u>diameters are greater than</u> 300 nm even at SS= = 0.08%. Therefore, <u>This suggests that</u> a smaller portion (<u>of</u> 1–MAF, \_\_(2-6%) is <u>just</u>-caused by the error <u>of in the</u> CDF fit, which will lead to <u>a</u> lower MAF than <u>it should be expected</u>.

4.1.3 CDF standard deviations ( $\sigma$ )

The CDF standard deviations ( $\sigma$ ) are general indicators for the extent of external 10 mixing and the heterogeneity of particle composition for the investigated aerosolaerosols in the size range around  $D_a$ .  $\sigma_POL$  and  $\sigma_BG$  in Table 1 are defined as CDF standard deviations or under polluted and background conditions are denoted as  $\sigma$  POL and  $\sigma$  BG in Table 1, respectively. Under ideal conditions, the CDF standard deviations should beg equals zero for an internally mixed, fully 15 mono-dispersed aerosol with particles of homogeneous chemical compositions composition. According to Rose et al. (. (2008), even after correcting for the DMA transfer function, however, calibration aerosols composed of high-purity ammonium sulfate exhibit small non-zero  $\sigma$  values that correspond to ~3% of D<sub>a</sub>. HtThis can be attributed to heterogeneities of the water vapor SS profile in the CCNC 20 or other non-idealities, such as DMA transfer function and particle shape effects. Thus, "heterogeneity parameter" values of  $\sigma/D_a = 3\%$  indicate internally mixed CCN, whereas higher values indicate external mixtures of particles. or/Da\_POL and o/Da BG in Table 1 are defined as heterogeneity parameterHeterogeneity parameters

Manuscript for Atmos. Chem. Phys., Zhang et al.: CCN measurement during AC<sup>3</sup>Exp <del>16 April,24 Oct.,</del> 2014 under polluted and background conditions are denoted as  $\sigma/Da$  POL and  $\sigma/Da$  BG in <u>Table 1, respectively. According to Table 1,  $\sigma$  /D<sub>a</sub>\_POL and  $\sigma$  /D<sub>a</sub>\_BG are with values</u> of They range from 17%-30%, which is much higher than the 3% observed for pure ammonium sulfate, indicating that the particles were externally mixed with respect to their solute content.

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#### 4.2 Derived $\kappa_a$ dependence on $D_a$

Fig.Figure 2 shows the dependence of  $\kappa_a$  on  $D_a$  under both-background and polluted conditions.  $\kappa_a$ \_POL—\_and  $\kappa_a$ \_BG are defined as the average hygroscopicity of CCN-active particles in the size range around  $D_a$  under polluted and background

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conditions-, respectively. For background days, larger particles were on average more hygroscopic than smaller particles; i.e.,  $\kappa_a$  BG increased increases substantially from about 0.22 at 30-60 nm to about 0.38 at the size range of 120-180 nm. Our result This is consistent with the field results observed in Guangzhou, South China by Rose et al., (2010). However, compared to  $\kappa_a$ \_BG,  $\kappa_a$ \_POL exhibited shows a relatively flat trend

15 withas the increase of particles particle size diameter. The  $\kappa_{a}$  POL didn't exhibited significant increase but with larger increases and error bars around a given D<sub>p</sub>, suggesting the complex and diversities of particle compositions are larger. This suggests that under polluted conditions, particle composition and their mixing state at polluted cases, is complex and diverse. In this case, larger particles are even less hygroscopic than the smaller particles. One of the possible reason for the changes of in 20  $\kappa_a$  atunder polluted eases conditions may be due to the presence of high amount of organics freshly emitted from biomass burning (Andreae and Rosenfeld, 2008; Petters et al., 2009; Rose et al., 2010,)) which would coat on those the larger particles and then lead the particles with render them less hygroscopic. Overall,  $\kappa$  for polluted

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aerosols are about 20% lower than that <u>offor</u> clean aerosols for particles in <u>the</u> accumulation size range; for these. For particles in <u>the</u> nucleation or Aitken size range,  $\kappa_a$  for polluted particles is slightly higher than that <u>atfor particles in the</u> background cases. Based on laboratory <u>experimentexperiments</u>, Petters et al., (2009) examined

the hygroscopic properties of particles freshly emitted from biomass burning. They found that κ was a function of particle size, with 250 nm particles being generally weakly hygroscopic and sub-100 nm particles being more hygroscopic. During the field campaign at Xianghe, the, polluted cases represent cases where particles were mainly biomass burning aerosols are the lead particles for the selected polluted cases.
The laboratory results, to some extent, can thus explain our field measurements. But furtherFurther investigations, including laboratory experiments and field measurements of size-resolved chemical composition, are needed to confirm and clarify this. The changes of κ<sub>eut</sub>\_POL and κ<sub>eut</sub>\_BG dependence on D<sub>eut</sub>, which are not shown here, displayed similar characteristics. But the derived κ<sub>eut</sub>\_POL and κ<sub>eut</sub>\_BG

(given in Table 1) are generally slightly smaller than  $\kappa_a$ \_POL and  $\kappa_a$ \_BG.

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#### 4.4 PDF of D<sub>a</sub> and $\kappa_a$

Fig.Figure 3 exhibits the shows probability distribution function functions (PDF) of  $D_a$  under background conditions and during polluted events throughout the campaign. D<sub>a</sub>\_POL are mainly distributed in the ranges of about 185-205, 163-180, 95-120, 65-75 and 45-55 nm at SS <u>levels</u> of 0.08%, 0.11%, 0.23%, 0.42% and 0.80%, respectively. At each SS<sub>7</sub>\_PDF\_level, the PDFs of D<sub>a</sub>\_POL showshave a wider distribution than the PDFs of D<sub>a</sub>\_BG. The PDF for D<sub>a</sub>\_POL moved several to dozens of nanometer and extended to<u>At</u> each SS level, the sidePDFs of D<sub>a</sub>\_POL extend to large particle size at all SS sizes indicating the impact by pollutions. In addition, both the pollution. The largest variation in Da\_BG and Da\_POL show larger variations at lower SS. This can be explained by two reasons: one is weakenseen at SS = 0.08%and 0.11%, respectively. One reason for this is the weakened impact (solute effect) of chemical composition on CCN activity at high SS; levels, i.e., the solute effect. The other onereason is probably due to the relatively larger uncertainties for measuring

Fig.Figure 4 exhibits the PDFshows PDFs of  $\kappa_a$  under background conditions and

lowerthat arise from making measurements at low SS levels.

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during polluted events-throughout the campaign... The PDF of  $\kappa_a$ \_POL presentshas a large variationsvariation at each size range around  $D_a$ . At a given  $D_{p_a}$  range, they are distributed withand shows two modes. For example,  $\kappa_a$ \_POL for particles around the size diameter of 48 nm is withshows two hygroscopic-peaks of at ~0.15 and ~0.23; it is with peak. Peak values of ~0.26 and ~0.32 are seen for particles around  $D_p==198$  nm. Because a large amounts of  $\kappa_a$ -are <0.3 but a very small portion of particles are with  $\kappa_a$ <0.1 at these cases. Such distribution mode for  $\kappa_a$  at polluted cases thus indicated anMost  $\kappa_a$ \_POL values are less than 0.3. This suggests externally mixed, but with-less hygroscopic particles during the-polluted cases. Compared to the PDFevents. Less variation is seen in the PDFs of  $\kappa_a$ \_POL,  $\kappa_a$ \_BC displays much less significant variations. It distributed with oneBG. One mode and is seen with peak values of 0.23,

0.30, 0.35, 0.35, and 0.38 around the size diameter of for  $D_p = 46$ -nm, 64-nm, 92-nm,

152-nm, and 179 nm. Few particles are with  $\kappa_a <$ , respectively. Most  $\kappa_a$ \_BG values are

<u>greater than 0.3</u> at when  $D_p \ge 60$  nm under background conditions, reflecting,

indicating that the particles are more hygroscopic with a relatively homogeneous

composition of the particles under background conditions.

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#### 4.5 CCN closure tests

In this section, we compare <u>observed total  $N_{CCN}$  observations</u> with corresponding values that were estimated on the basis of aerosol particle number size distributions measured in parallel and <u>non-not in parallel and assuming a uniform particlesparticle</u> composition. <u>If theClosure is achieved if</u> estimated and observed  $N_{CCN}$  agree quantitatively within the range of their <u>uncertainty-uncertainties</u>.

<u>4.5.1 PO</u> closure is achieved. By using averaged CDF fit curve method with an assumption of uniform chemical composition, CCN size distributions were firstly calculated by multiplying the CCN efficiency spectra with the total aerosol particle (CN) number size distributions measured in parallel and non-parallel. Estimated N<sub>CCN</sub> were calculated by stepwise integration of the CCN size distributions from 10 to 700

<del>nm. <u>test</u></del>

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In PO closure tests, the measured CCN efficiency spectrum is first multiplied by the measured CN size distribution, which yields the CCN size distribution. This is then

15integrated over the whole size range to obtain the observed total CCN concentration<br/>(CCN Obs). Size-resolved N<sub>CCN</sub> are calculated by multiplying the campaign-averaged<br/>CCN efficiency spectrum with simultaneously measured CN number size distributions.<br/>Estimated total N<sub>CCN</sub> (CCN Estimated) are obtained by the stepwise integration of<br/>size-resolved N<sub>CCN</sub> from 10 to 700 nm. With this comparison, the influence of the

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Fig. 4.5.1 Parallel observation (PO) closure tests

because the CN size distribution is the same for both parameters.

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The comparison of estimated and parallel observed N<sub>CCN</sub>-5 shows CCN Estimated as a function of CCN\_Obs. at five SS of levels ranging from 0.08% to 0.80% is shown in

variation in chemical composition on the CCN concentration can be investigated

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	Fig. 5. In this case, a good agreement between estimated%. Estimated and measured		
	total $N_{CCN}$ was obtained agree well. The mean slope and correlation coefficient ( $R^2$ )		
	are 0.99 and 0.97, respectively, at the five super saturations. At SS of 0.08% and	 带格式的	
	0.11%, the results showed a little bit (~levels. A ~ $3-4$ %)% underestimation of CCN		
5	number concentrations.occurs at SS levels of 0.08% and 0.11%. One reason for		
	thethis slight worse closure between measured and estimated CCN number		
	concentrations at lower SS lies in underestimation is that size-resolved activation		
	ratios exhibit a larger variabilityARs are more variable at low SS levels than that at		
	higher SS <u>levels</u> . Also, compared to thetotal activated CCN number concentrations at		
10	high SS, the $N_{CCN}$ is with a levels, there are less amounts $N_{CCN}$ at low SS levels,		
	which would lead to larger uncertainties or to a lower correlation. Overall, the CCN		
	closure can be achieved by using campaign-averaged CCN efficiency spectra. In this		
	PO closure test-indicated that the estimation on a basis, the influence of mean CDF fit	 带格式的	
	AR curve methods can estimate the observed N <sub>CCN</sub> pretty well when SS is high,		
15	although uniform and internally mixed variations in chemical composition throughout	 带格式的	
	the size range being assumed on CCN concentrations is insignificant.	 带格式的	
	4.5.2 Non-parallel observation (NPO) closure teststest	( <b>带格式的:</b> 字体:加粗	
	In this study, we also Mean CCN efficiency spectra derived from size-resolved CCN		
	measurements taken on 7-21 July 2013 are used to estimate total CCN number		
20	concentrations based on non-parallel-CN size distribution measurements (taken on		
	1-25 June, 2013) 2013. This is referred to as an NPO closure test. The average		
	measured CCN efficiency spectrum (corresponding to spectra in Fig. 1) is multiplied	 带格式的:字体颜色:黑色	
	by the measured CN size distribution which yields the CCN size distribution. This is		
	integrated over the whole size range (10-700 nm) to obtain the estimated total CCN		
l			

concentration. The mean CCN efficiency spectra at SS levels of 0.23% and 0.80% (Fig. 1) is used to estimate the total CCN concentration during 1-25 June 2013. The mean CCN efficiency spectra at SS = 0.51% is derived using the exponential relationships developed from plotting the three CDF fit eurve derived from the

size resolved CCN measurements (7 21 July, 2013). The comparison of estimated

parameters as a function of SS (see Fig. 7).

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Estimated total N<sub>CCN</sub> and the non-parallel as a function of measured bulk N<sub>CCN</sub> at SS levels of 0.223%, 0.5%51%, and 0.880% are shown in Fig. 6. A reasonable agreement between estimated and measured N<sub>CCN</sub> was obtained by using CDF fit curve method. The <u>The lower slope at SS = = 0.223</u>% indicates that the estimation on <u>athe</u> basis of NPO closure underestimates about 7% of the observed N<sub>CCN</sub>. The closure is considerably improved at higher SS. However, only a levels. A reasonable correlation with R<sup>2</sup> of about 0.6 0.8 between estimated and measured total N<sub>CCN</sub> was achieved, is seen ( $R^2 = 0.6-0.8$ ), which suggests temporal variations both-in chemical composition orand mixing state of aerosol particles. In addition, there are uncertainties due to measuring bulk and size-resolved CCN. Overall, the-uncertainties in suchthis NPO CCN closure study are aboutrange from 30%-40%. Thus, it is important to conduct such field experiment to measure CCN under different environmental conditions. Likewise, caution needs to be exercised to useCaution is needed when using data from any experiment of short periods term experiment at a single site to do CCN parameterization parameterizations for any large-scale applications. It is necessary to conduct long-term CCN measurements at more regional sites, especially those with

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heavy pollution of high CNthat are heavily polluted.

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4.6 Case study: <u>implications of</u> CCN activation <u>combiningand</u> chemical composition

To understand the The behavior of CCN activation under different surrounding circumstance, twobackground and polluted conditions is examined. Two cases are selected: one case with total N<sub>CN</sub> << 15000 cm<sup>-3</sup> and >(background) and another case 5 with total N<sub>CN</sub> > 15000 cm<sup>-3</sup> during the campaign are selected for investigation, which are defined as background and polluted condition, respectively. Interestingly, (polluted). Trends in bulk CCN activation exhibits completelyas N<sub>CN</sub> increases are different changes with N<sub>CN</sub>-from for the two background and polluted cases: the. Bulk AR at allthe three SS of levels (0.223%, 0.5%51%, and 0.8% increase with the 10 increase of 80%) increases as total N<sub>CN</sub> at increases for background cases (Fig-7a), whereas they decline with increase of. 8a) and decreases as total N<sub>CN</sub> duringincreases for polluted eventscases (Fig-7b). At. 8b). For the background cases, changes of in <u>bulk</u> AR are apparently dependence dependent on changes of  $\kappa_{chem}$  (Fig. 7c), showing well<u>8c). A good correlation between AR\_0.223 and  $\kappa_{\text{chem}}$  with  $(R^2 \ge 0.7 - ()$ </u> 15 is seen in Fig. 8). Such 9. A high correlations correlation between bulk AR and  $\kappa_{chem}$  have when total N<sub>CN</sub> is low has been observed when N<sub>CN</sub> are low during the whole campaign (Fig. 910). In these cases, organics account for relatively low amounts (-----30%)% of the total particle mass concentrations butconcentration and concentrations of soluble inorganics are high (Fig. 7e8e). In particular, the mass 20 concentration of nitrate is higher than that for organics accountingand accounts for the largest mass fraction of all compositions when  $\kappa_{chem}$  reaches the maximum with a mean value of ~0.45. The f44, which is the fraction of total organic mass signal at m/z44, is not correlated with AR (Fig.  $\frac{7e8c}{2}$ ). The m/z 44 signal is due-mostly due to acids (Takegawa et al., 2007; Duplissy et al., 2011) or acid-derived species, such as esters, 25

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	and f44 is closely related to the organic oxidation level-(i.e., O:C ratio) (Aiken et
	al.,2008), i.e., O:C ratio (Aiken et al., 2008). Oxidized/aged acids are generally more
	hygroscopic and easily activated. Therefore, the lower correlation between f44 and
	AR implies that organics under low N <sub>CN</sub> conditions are less hygroscopic. CN number
5	concentrations in the nucleation, Aitken, and accumulation modes are shown in Fig.
	8g (polluted) and Fig. 8h (background). Under background conditions, bulk AR at SS
	= 0.23% is more correlated ( $R^2$ = 0.5) with changes in N <sub>CN</sub> in the accumulation mode
	(Fig. 9), suggesting that most aerosol particles with sizes > 100 nm can be activated.
	Smaller particles with Aitken diameters of < 40 nm at the given SS levels
10	(0.23%-0.80%) are not as easily activated, if at all, so no correlation is seen (Fig. 8g).
	Under polluted conditions, there is little dependence of changes in bulk AR with
	changes in $\kappa_{chem}$ (Fig. 8d). Bulk AR at SS = 0.23% is moderately correlated with f44
	$(\mathbf{R}^2 = 0.5, \text{Fig. 9})$ . As stated above, f44 is always related to the organic oxidation level.
	Usually, the oxidized/aged acids are more hygroscopic and easily activated. Therefore,
15	the deteriorated correlations between f44 and AR implied that organics at low N <sub>CN</sub>
	cases are less hygroscopic. In addition, CN number concentrations at size range of
	nucleation, Aitken and accumulated mode are also shown in Fig 7g and 7h for
	background and polluted conditions respectively. In background days, AR at SS=0.2%
	is more correlated with the changes of $N_{CN}$ at accumulated mode with $R^2$ of ~0.5 (Fig.
20	8), suggesting aerosol particles at the size range of >100 nm can be mostly activated.
	Smaller particles with Aitken diameter of <40 nm at the given SS (0.2% 0.8%) are
	difficult or cannot be activated, thus there are no correlations indicated in Fig. 7g
	During polluted events, changes of AR didn't displayed apparent dependence on
	changes of $\kappa_{\text{chem}}$ (Fig. 7d). But AR at SS=0.2% showed moderate correlation with f44
25	with $R^2$ of ~0.5 (Fig. 8). As stated above, the f44 is always related to the organic

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oxidation level. Usually, the oxidized/aged acids are more hygroscopic and easily activated. The correlation between f44 and <u>bulk AR suggested suggests</u> that the significant roles of organics contributions contribution from oxidized or aged aerosols toplay a significant role in CCN activity. Because the impact from organics on the

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- 5 particles activation is complicated (Jimenez et al., 2009). In addition, it will introduce definiteA bias is introduced by using a parameterized function derived from observations made at other sites with different aerosol types to describe the particle hygroscopicity and activation properties due to the complexity of the organic aerosol fraction and its tendency to evolve with atmospheric oxidative processing and aerosol
- 10 aging based on previous studies (e.g., Padró et al., 2010; Engelhart et al., 2011, 2012; Asa-Awuku et al., 2011). Similarly, under Under polluted conditions, the bulk AR\_0.2 is more correlated with the changes of accumulated in accumulation mode particles, with  $(\mathbf{R}^2 \text{ of } = \sim 0.3.)$ .
- Overall, based on the case investigationstudy, one cannot use a parameterized formula combining with using only total N<sub>CN</sub> to estimate total CCN number concentrations. 15 However, if If observations such as size-resolve CCN as well as and size-resolved chemical compositions composition are not available, athe possibility of using bulk  $\kappa_{\rm chem}$  and f44 by combining in combination with bulk N<sub>CN>100nm</sub> > 100 nm to parameterize CCN number concentrations when with low and high N<sub>CN</sub> respectively has been is implied. But further by the case study. Further field investigations or 20 examinations are needed to demonstrate and confirm the relationship between bulk AR and particle size and compositionscomposition.

#### 5. Summary and conclusions

Atmospheric aerosol particles acting as CCN are pivotal elements of the hydrological cycle and climate change. In this study, we measured and characterized CCNNccn in



relatively clean and polluted air during the AC<sup>3</sup>Exp campaign conducted at Xianghe,

China during summer 2013, with the. The aim of understanding thewas to examine CCN activation properties under high aerosol loading conditions in thea polluted regionsregion and implying to assess the impacts of particle size and chemical 5 composition on the CCN activation ratio to the AR which acts as a proxy of the total number of aerosol particles in the atmosphere. Based on the CDF fit method, a gradual increase in AR dependence on D<sub>p</sub>size-resolved AR with particle diameter suggests that aerosol particles have lower hygroscopicity and heterogeneity of the composition during the campaign. Bothdifferent hygroscopicities. Da and Deut were increased significantly at lower SS (< levels (< 0.223%) due to pollutions (pollution, 10 e.g., biomass burnings); the. This increase was not observed when  $SS \ge 0.4\%$ . The value of  $\kappa$  are about 0.31-0.38 for For particles in the accumulation mode-, values of  $\kappa$ range from 0.31-0.38 under background conditions-and, which is about 20% higher than that derived under polluted conditions; however,  $\kappa$  are about 0.20-0.34 for. For 15 particles in the nucleation or Aitken mode-at,  $\kappa$  range from 0.20-0.34 under both background cases, showing slightly lower than that during and polluted events conditions. Larger particles were on average more hygroscopic than smaller particles. ButHowever, the case is more complex for particles withoriginating from heavy pollutionspollution due to the diversities of diversity in particle compositions composition and mixing state. –The low  $R^2$  for the NPO CCN closure 20 test suggests abouta 30%-40% uncertainties uncertainty in total N<sub>CCN</sub> prediction that is mainly caused by changes in particle composition. By combining analyses of estimation. Using bulk chemical composition measurementmeasurements, the relationship between bulk AR and the physical and chemical properties of the atmospheric aerosolaerosols is investigated. Based on thea 25

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case investigationstudy, we conclude that one cannot use a parameterized formula combining with using only total N<sub>CN</sub> to estimate total N<sub>CCN</sub>. We have implied ashown the possibility of using bulk  $\kappa_{chem}$  and f44 by combining in combination with bulk N<sub>CN>100nm</sub> > 100 nm to parameterize CCN number concentrations when with low and high N<sub>CN</sub> respectively. Further field investigations or examinations are needed to demonstrate and confirm the relationship between bulk AR and particle size and compositionscomposition.

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#### **Acknowledgement**

### **Acknowledgements**

10 We would thank the two-reviewers very much for their suggestions and comments which have greatly improved the paper. This work was funded by the National Basic Research Program of China '973' (Grant No. 2013CB955801, 2013CB955804), the National Science Foundation (1118325), and the Fundamental Research Funds for the Central Universities (Grant No. 2013YB35). The We greatly appreciate the support of the entire  $AC^{3}Exp$  team has been much appreciated. 15

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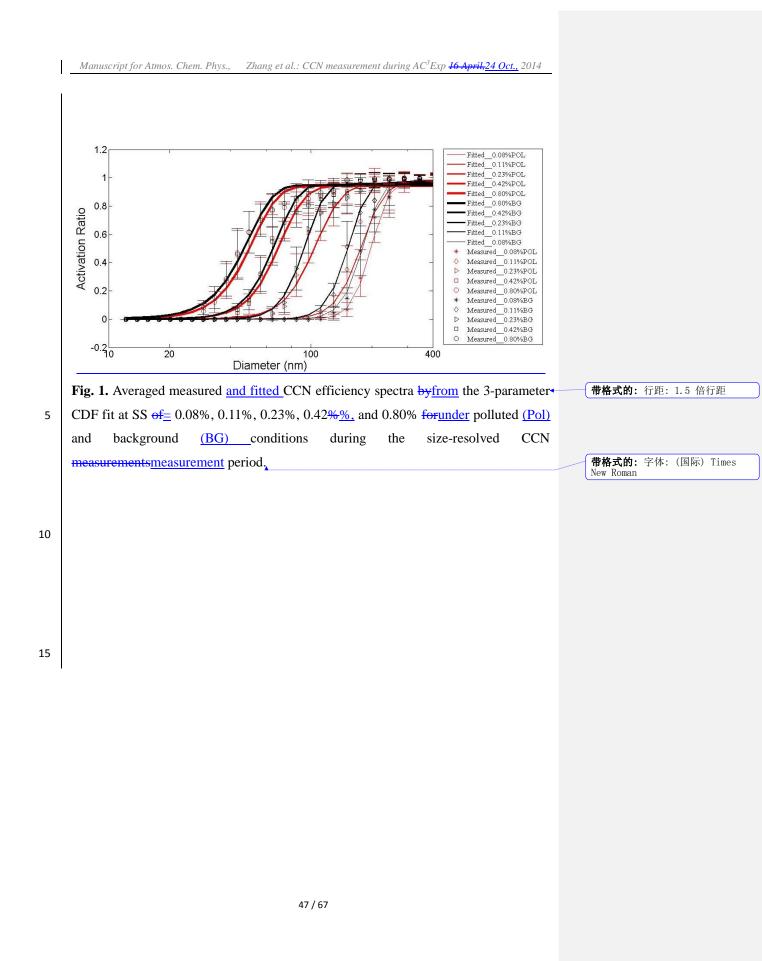
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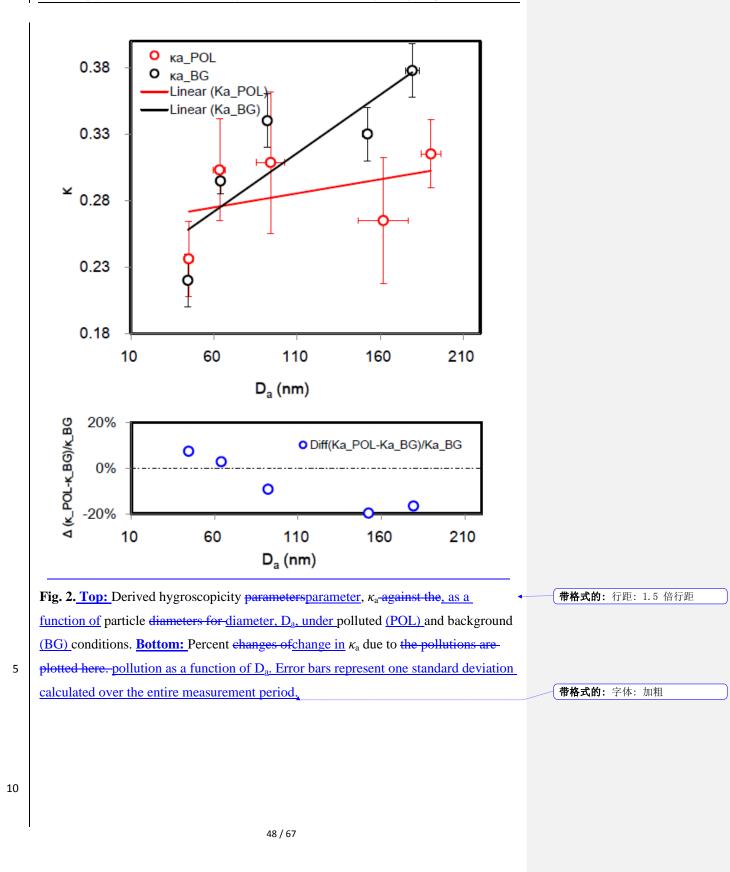
## **Tables and Figures**

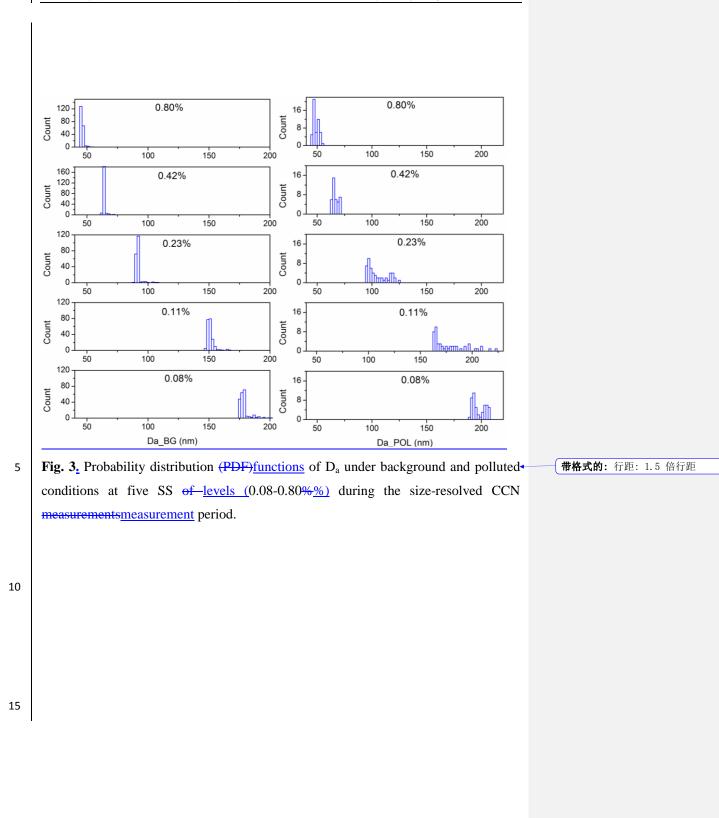
**Table 1.** Characteristic of Parameters describing the CCN efficiency spectra basic spectral parameters and hygroscopicity for polluted (\*\_POL) and background aerosols during(\*\_BG) cases: the campaign for different super saturation. Quantities are midpoint activation diameters diameters ( $D_{a, -} D_{eut}$ ), the maximum activated fractions fraction (MAF), the CDF standard deviations deviation ( $\sigma$ ), the heterogeneity parameters parameter ( $\sigma/D_a$ ), and the hygroscopicity parameters parameter ( $\kappa_a, \kappa_{eut}$ ). Values shown are arithmetic mean values with one standard deviation averaged over the entire measurement period. **Figure captions:** 

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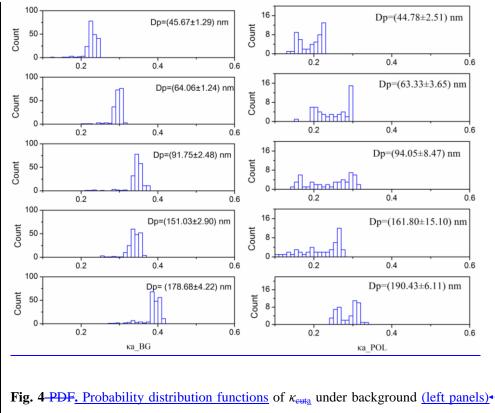
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1	<u>0.08%</u>	<u>190.43±6.11</u>	<u>0.98±0.01</u>	<u>33.34±4.49</u>	<u>0.17±0.02</u>	<u>0.32±0.03</u>		178.68±4.22	<u>0.98±0.01</u>	<u>32.73±2.07</u>	<u>0.18±0.01</u>	<u>0.38±0.02</u>
İ	<u>0.11%</u>	<u>161.80±15.10</u>	<u>0.98±0.01</u>	<u>38.61±7.62</u>	<u>0.22±0.03</u>	<u>0.26±0.05</u>		<u>151.03±2.90</u>	<u>0.97±0.01</u>	28.56±1.97	<u>0.19±0.01</u>	<u>0.33±0.02</u>
İ	<u>0.23%</u>	<u>94.05±8.47</u>	<u>0.96±0.01</u>	<u>27.87±6.30</u>	<u>0.26±0.04</u>	<u>0.31±0.05</u>		<u>91.75±2.48</u>	<u>0.96±0.00</u>	<u>18.81±1.53</u>	<u>0.20±0.01</u>	<u>0.34±0.02</u>
İ	<u>0.42%</u>	<u>63.33±3.65</u>	<u>0.94±0.01</u>	<u>18.02±2.84</u>	<u>0.26±0.03</u>	<u>0.30±0.04</u>		<u>64.06±1.24</u>	<u>0.95±0.00</u>	<u>16.21±0.81</u>	<u>0.25±0.01</u>	<u>0.29±0.01</u>
İ	<u>0.80%</u>	44.78±2.51	<u>0.94±0.01</u>	<u>14.08±0.98</u>	<u>0.29±0.01</u>	<u>0.24±0.03</u>	_	45.67±1.29	<u>0.95±0.01</u>	13.82±1.17	<u>0.30±0.02</u>	<u>0.22±0.02</u>
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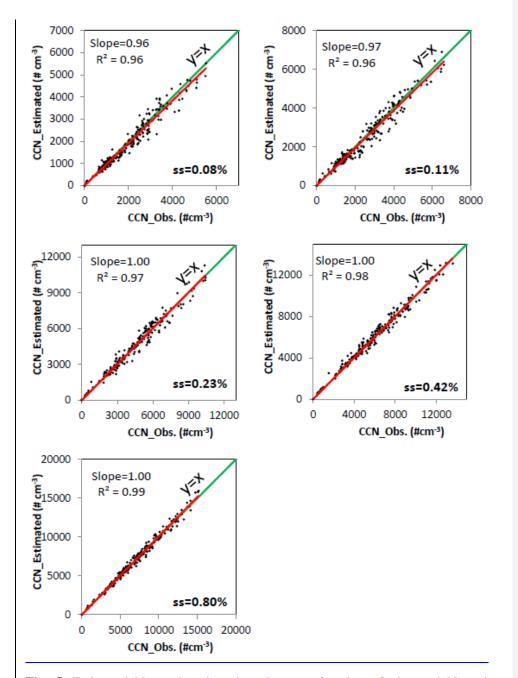
**Fig. 4**-PDF. Probability distribution functions of  $\kappa_{\text{euta}}$  under background <u>(left panels)</u> and polluted <u>(right panels)</u> conditions <u>atfor</u> different particle size ranges during the size-resolved CCN measurements measurement period.

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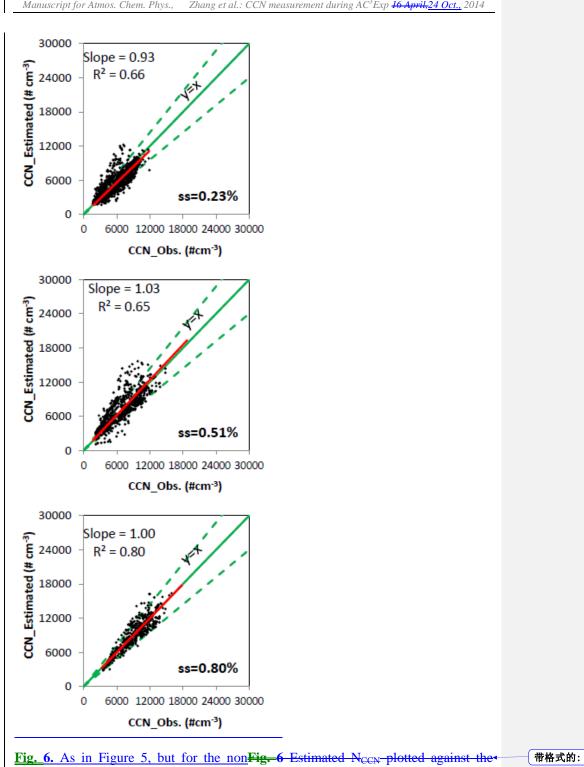
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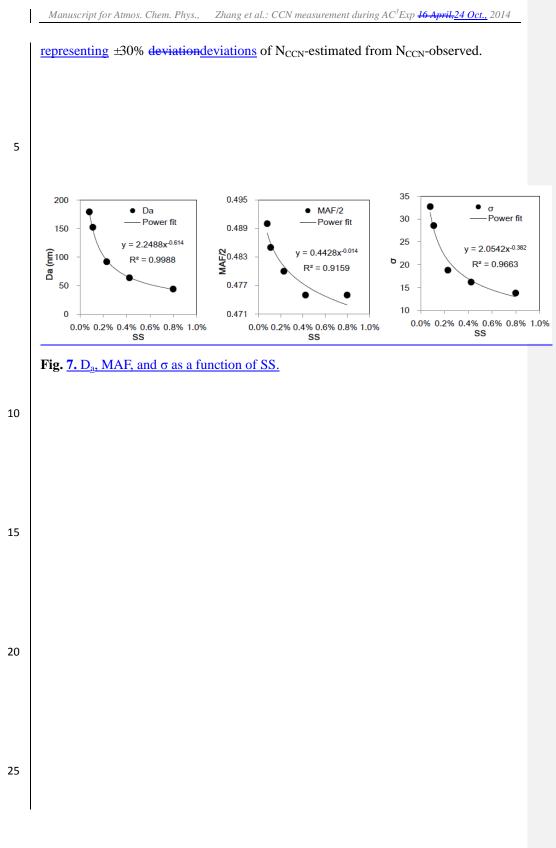
**Fig. 5**. Estimated  $N_{CCN}$  plotted against the <u>as a function of observed  $N_{CCN}$  inat</u> <u>different SS levels in the parallel observation (PO) closure test.</u> The green solid line is the 1:1 line.

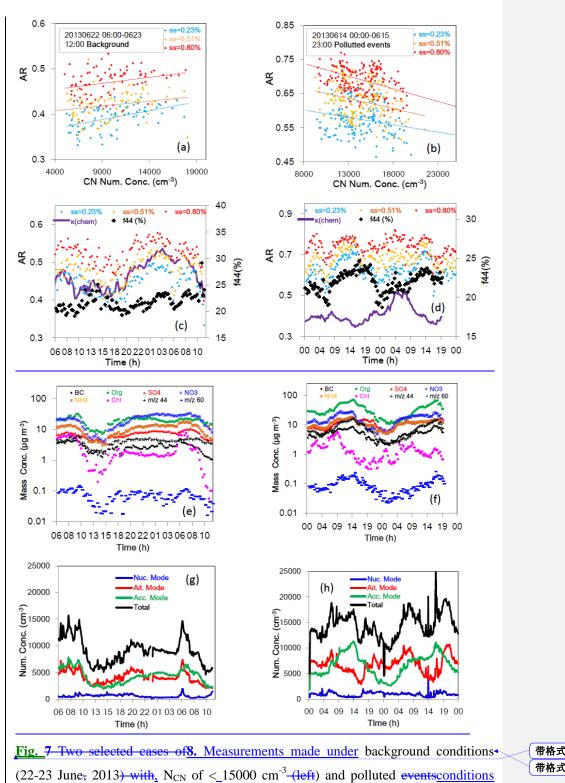
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observed N<sub>CCN</sub> in Non-parallel observation (NPO) closure test. The green solid line is 1:1 line, and the dashed green lines indicate the band of aboutboundaries the

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(14-15 June, 2013) with  $N_{CN}$  of > 15000 cm<sup>-3</sup> (right) during the campaign.). Bulk

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CCN activation ratios (AR) at all three supersaturations of SS = 0.2%, 0.5%, and 0.8% against as a function of N<sub>CN</sub> in clear days under background and polluted daysconditions are shown in Fig-7a. 8a and Fig. 7b8b, respectively. Diurnal variations

of $\underline{in}$ AR, derived $\underline{from} \kappa_{chem}$ and $\underline{the}$ fraction of total organic mass signal at $\frac{m/z}{2}$ 44	
(f44) are shown in Fig. 7c (, under background and polluted conditions) and are	带格式的
shown in Fig. 7d (polluted events).8c and Fig. 8d, respectively. Mass concentrations	带格式的
of black carbon (BC), organics, nitrate (NO <sub>3</sub> <sup><math>-</math></sup> ), ammonium (NH <sub>4</sub> <sup>+</sup> ), sulfate (SO <sub>4</sub> <sup>2<sup><math>-),</math></sup></sup>	
chloride (Cl <sup>-</sup> ) ironsions, etc. are shown in Fig. 7e (., under background and polluted	带格式的
conditions) are shown in Fig. 8e and Fig.7f (polluted events). 8f, respectively. N <sub>CN</sub> at	带格式的
size range of for nucleation, (10-30 nm), Aitken and accumulated mode are shown in	
Fig 7g and Fig. 7h for (30-130 nm), and accumulation modes (130-700 nm) under	

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conditions) are shown in Fig. 8e and Fig.7f (polluted events). 8f, respectively. N<sub>CN</sub> at size range of for nucleation, (10-30 nm), Aitken and accumulated mode are shown in Fig 7g and Fig. 7h for (30-130 nm), and accumulation modes (130-700 nm) under background conditions and polluted events conditions are shown in Fig. 8g and Fig. 8h respectively.

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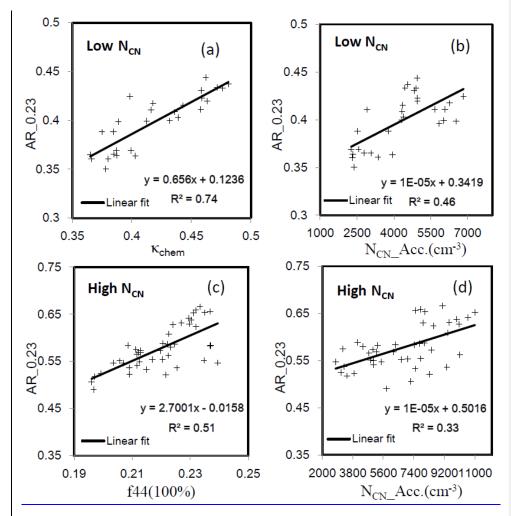
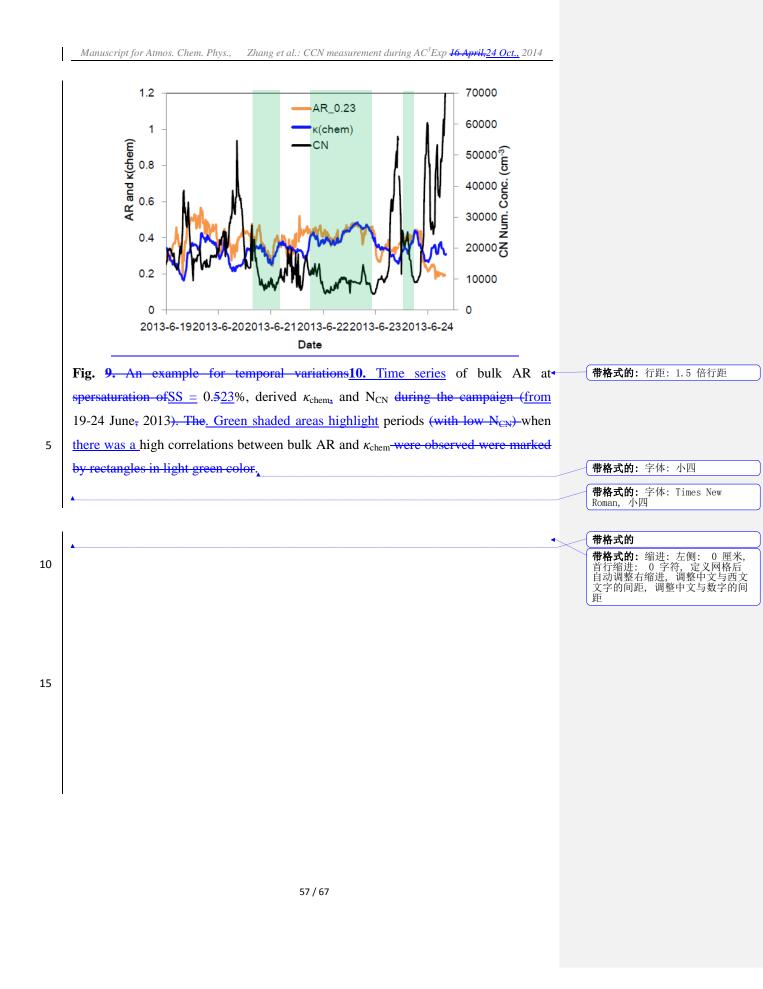


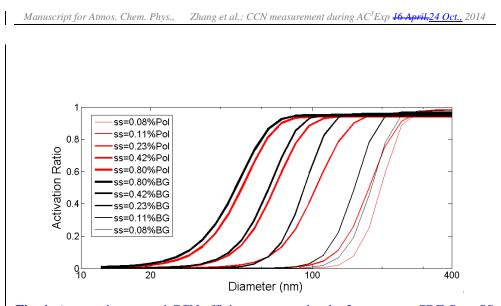
Fig. 8 Correlations 9. AR measured at SS = 0.23% as a function of AR with(a)  $\kappa_{\text{chem}}$  f44 and N<sub>CN</sub>\_Acc. at\_under background (with low N<sub>CN</sub>) and conditions, (b) accumulation mode N<sub>CN</sub> under background conditions, (c) f<sub>44</sub> under polluted (with high N<sub>CN</sub>) cases.conditions, and (d) accumulation mode N<sub>CN</sub> under polluted conditions. The ARaccumulation mode size range is measured at SS=0.2%, the N<sub>CN</sub>\_Acc. is CN number concentrations at accumulated modes.130-700 nm in this study.\_

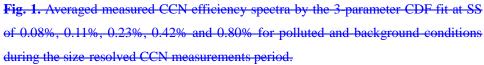
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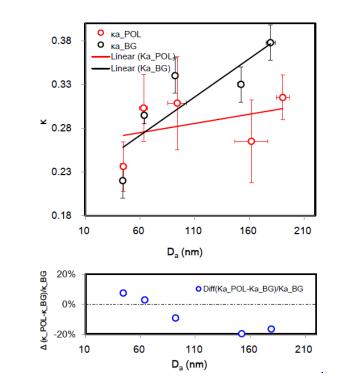


**Table 1.** Characteristic of spectra basic spectral parameters for polluted and background aerosols during the campaign for different super saturation. Quantities are midpoint activation diameters ( $D_{a}$ ,  $D_{eut}$ ), maximum activated fractions (MAF), CDF standard deviations ( $\sigma$ ), heterogeneity parameters ( $\sigma/D_{a}$ ), hygroscopicity parameters ( $\kappa_{a}$ ,  $\kappa_{eut}$ ).

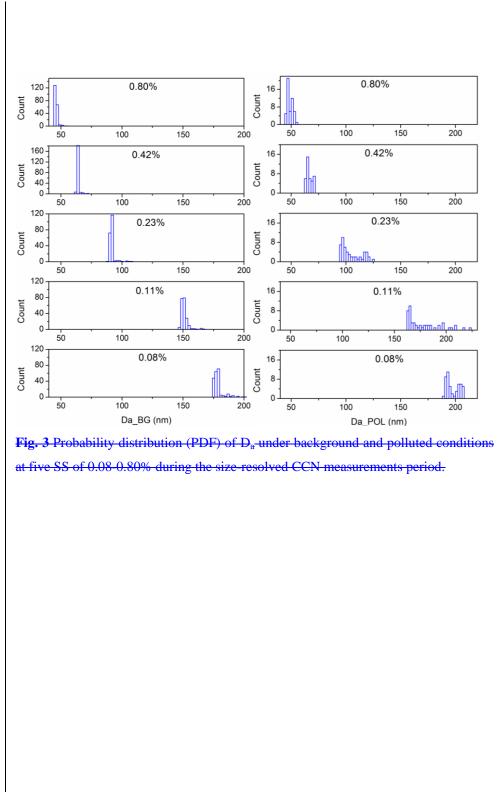
i T	<del>SS</del>	Ð <u>₊_POL</u>	D <sub>eut</sub> _POL	MAF_POL	<del>σ_POL</del>	<del>σ/D<sub>e</sub>_POL</del>	<del>κ<sub>e</sub>_POL</del>	<del>к<sub>ен</sub>_POL</del>	_	<mark>D<sub>e</sub>_BG</mark>	Ð <sub>ew⊂</sub> BG	MAF_BG	<del>σ_BG</del>	<del>σ/D<sub>a</sub>_BG</del>	<del>ĸ<sub>a</sub>_BG</del>	<del>ĸ<sub>eut_</sub>BG</del>
	<del>0.079%</del>	<del>190.43±6.11</del>	<del>191.27±5.23</del>	<del>0.98±0.01</del>	<del>33.34±4.49</del>	<del>0.17±0.02</del>	<del>0.32±0.03</del>	<del>0.31±0.02</del>		<del>178.68±4.2</del> 2	<del>180.22±3.9</del> 7	<del>0.98±0.01</del>	<del>32.73±2.07</del>	<del>0.18±0.01</del>	<del>0.38±0.02</del>	<del>0.37 ±0.03</del>
	<del>0.109%</del>	<del>161.80±15.1</del> ₽	<del>162.93±15.1</del> 4	<del>0.98±0.01</del>	<del>38.61±7.62</del>	<del>0.22±0.03</del>	<del>0.26±0.05</del>	<del>0.26±0.05</del>		<del>151.03±2.9</del> 0	<del>153.62±2.6</del> 9	<del>0.97±0.01</del>	<del>28.56±1.97</del>	<del>0.19±0.01</del>	<del>0.33±0.02</del>	<del>0.31±0.02</del>
	<del>0.228%</del>	<del>94.05±8.47</del>	<del>95.31±7.01</del>	<del>0.96±0.01</del>	<del>27.87±6.30</del>	<del>0.26±0.04</del>	<del>0.31±0.05</del>	<del>0.30±0.05</del>		<del>91.75±2.48</del>	<del>93.40±2.27</del>	<del>0.96±0.00</del>	<del>18.81±1.53</del>	<del>0.20±0.01</del>	<del>0.34±0.02</del>	<del>0.32±0.02</del>
i	<del>0.423%</del>	<del>63.33±3.65</del>	<del>64.76±3.1</del> 4	<del>0.94±0.01</del>	<del>18.02±2.84</del>	<del>0.26±0.03</del>	<del>0.30±0.04</del>	<del>0.28±0.04</del>		<del>64.06±1.2</del> 4	<del>65.86±1.13</del>	<del>0.95±0.00</del>	<del>16.21±0.81</del>	<del>0.25±0.01</del>	<del>0.29±0.01</del>	<del>0.27±0.02</del>
l	<del>0.797%</del>	44 <del>.78±2.51</del>	47.55 <u>±2.65</u>	<del>0.94±0.01</del>	<del>14.08±0.98</del>	<del>0.29±0.01</del>	<del>0.24±0.03</del>	<del>0.20±0.04</del>	_	4 <del>5.67±1.29</del>	4 <del>7.25±1.16</del>	<del>0.95±0.01</del>	<del>13.82±1.17</del>	<del>0.30±0.02</del>	<del>0.22±0.02</del>	<del>0.20±0.02</del>

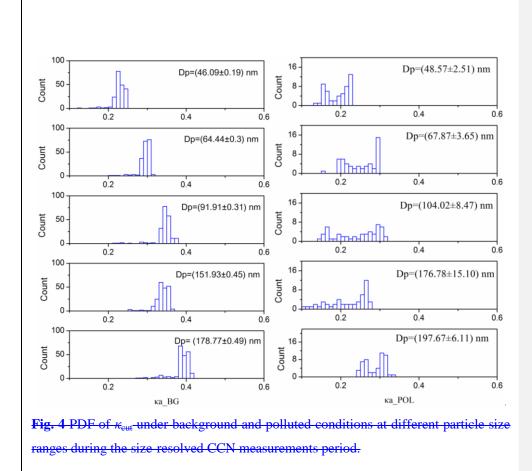


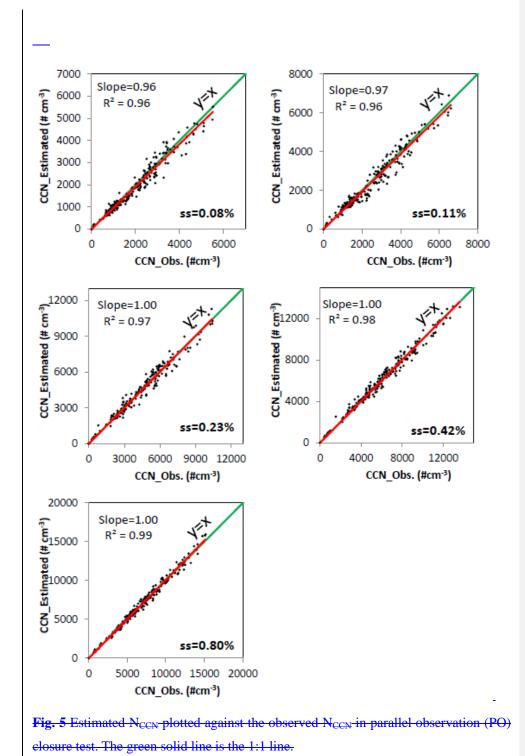




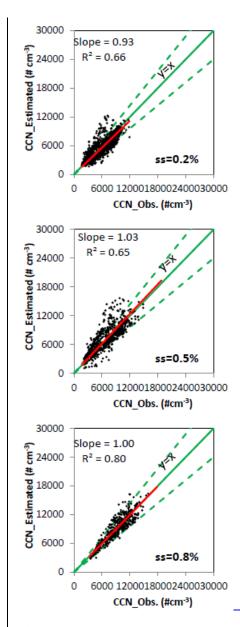
**Fig. 2.** Derived hygroscopicity parameters,  $\kappa_{a}$  against the particle diameters forpolluted and background conditions. Percent changes of  $\kappa_{a}$  due to the pollutions areplotted here.



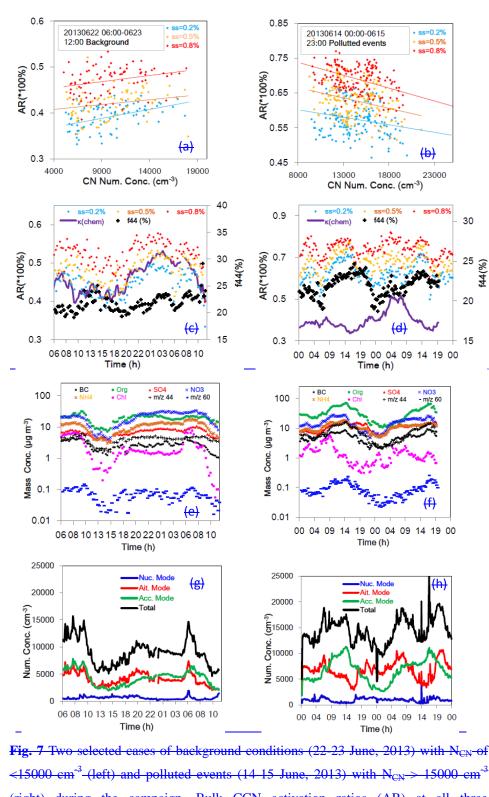




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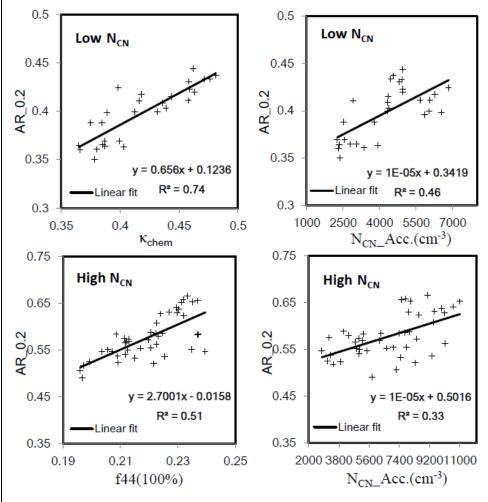


**Fig. 6** Estimated  $N_{CCN}$  plotted against the observed  $N_{CCN}$  in Non-parallel observation (NPO) closure test. The green solid line is the 1:1 line, and the dashed green lines indicate the band of about ±30% deviation of  $N_{CCN}$ -estimated from  $N_{CCN}$ -observed.



(right) during the campaign. Bulk CCN activation ratios (AR) at all three

supersaturations of 0.2%, 0.5% and 0.8% against  $N_{CN}$  in clear days and polluted days are shown in Fig 7a and Fig. 7b respectively. Diurnal variations of AR, derived  $\kappa_{chem}$ and fraction of total organic mass signal at m/z 44 (f44) are shown in Fig. 7c (background conditions) and Fig. 7d (polluted events). Mass concentrations of black carbon (BC), organics, nitrate (NO<sub>3</sub><sup>=</sup>), ammonium (NH<sub>4</sub><sup>+</sup>), sulfate (SO<sub>4</sub><sup>2=</sup>), chloride (CI<sup>-</sup>) irons etc. are shown in Fig. 7e (background conditions) and Fig.7f (polluted events). N<sub>CN</sub> at size range of nucleation, Aitken and accumulated mode are shown in Fig 7g and Fig. 7h for background conditions and polluted events respectively.



**Fig. 8** Correlations of AR with  $\kappa_{\text{chem}}$ , f44 and  $N_{\text{CN}}$ -Acc. at background (with low  $N_{\text{CN}}$ ) and polluted (with high  $N_{\text{CN}}$ ) cases. The AR is measured at SS=0.2%, the  $N_{\text{CN}}$ -Acc. is CN number concentrations at accumulated modes.

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**Fig. 9.** An example for temporal variations of bulk AR at spersaturation of 0.5%, derived  $\kappa_{\text{chem}}$  and N<sub>CN</sub> during the campaign (19-24 June, 2013). The periods (with low N<sub>CN</sub>) when high correlations between bulk AR and  $\kappa_{\text{chem}}$  were observed were marked by rectangles in light green color.

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