

The authors thank the anonymous referees for their time to review our manuscript and particularly for their valuable comments and suggestions that have significantly improved the manuscript. We have made most of the changes suggested by the reviewers and have outlined these in detail below.

Anonymous Referee #2

The manuscript by Ho et al describes the organic acid and carbonyl concentrations measured in two locations in Beijing in 2007. The aim of the study was to determine the roles of regional transport, local emissions and secondary formation of particulate matter in the air in Beijing. Beijing is one of the largest cities in the world with severe atmospheric pollution and every effort towards cleaner air is certainly beneficial. They managed to show that traffic restrictions are useful in reducing primary pollutants although secondary products were not reduced due to enhanced photochemical aging. These issues are important when planning efficient strategies for air pollution control in China. The paper is well written and clear and I think it is suitable for publication in ACP after minor corrections. My main concern deals with the stability of the samples. The samples were collected seven years ago, which is really long storage time for organic molecules even when kept in -20 °C. When were they analyzed? If they were analyzed only recently, how was the stability of the samples confirmed? The other concern deals detection using FID since there is very often overlapping of other compounds when using FID for detection. The occasional MS analysis does not confirm the purity of the peaks. It would have been better to analyze all the samples using MS detection. The number of samples is small, only ten samples/site and two of the samples were taken during traffic restrictions. This limits the confidence to the results.

Response:

We agree with the reviewer that care must be taken in the stability of organic compounds during sample storage. Our manuscript was prepared in 2014, but samples were measured in 2010, around 2.5 years after samples collection and storage at -20 °C. We consider that it is within the range of uncertainties. The data reported in this study were determined and quantified by both Agilent 6890GC/FID (Palo Alto, CA, USA) and ThermoQuest Trace MS (Austin, TX, USA). We agree with the reviewer that the number of samples is small, which is limited by the CAREBeijing-2007 campaign itself. The campaign was a pilot study to study the effects of the traffic restriction on the air quality of Beijing and to get experience and scientific evidence for the preparation of the 2008 Olympic game. There were only three consecutive days with traffic restriction (17-19 August), and we took measurements on two days (17 and 19 August). The rest was made before and after the traffic control events.

Minor comments:

What is meant by the expression C18:1, C25:0 etc.? I assume the first number is C-number and the latter one refers to the amount of double bonds, but please mention it in the first place.

Response:

It is true that the first number is C-number and the latter one refers to the amount of double bonds. We have explained it in the first place of the revised manuscript.

The Fig.1 is too small in the printed version and the font size too small. You do not need to include the explanations in every panel, once is enough.

Response:

Fig. 1 is revised as suggested by the reviewer.

Fig. 2 should show also the trajectory for 19th August, but it does not.

Response:

Fig. 2c shows the trajectory for 17 and 19 August. As they have similar trajectory, so we put two trajectories into one.

Figure 3 would be clearer with colors.

Response:

Fig. 3 is revised (with colors) as suggested by the reviewer.

In Fig. 5 caption please do not use abbreviation R/N, but the whole word.

Response:

We use the whole word instead of abbreviation in the revised version.

When calculating the ratios, did you use all measured compounds with 3 or 4 C atoms?

Response:

We use malonic and succinic acids when calculating the ratios. We have added the full name of the chemical in the first place.

7. The number of samples collected at Yufa was 10 and not 1 as indicated in Table 1?

Response:

Sorry for the typo. It is 10. We have revised in Table 1.

Anonymous Reviewer#3

Comment on “Dicarboxylic acids, ketocarboxylic acids, α -dicarbonyls, fatty acids and benzoic acid in PM_{2.5} aerosol collected during CAREBeijing-2007: an effect of traffic restriction on air quality” by K. F. Ho et al. The manuscript describes the presence of dicarboxylic acids, carbonyls, fatty acids and benzoic acids and OC, EC, and WSOC concentrations in PM_{2.5} samples collected in 2007 in two sites in the metropolitan area of Beijing.

One of the outcomes is that traffic restrictions decrease the impact of ‘primary’ organics, while secondary products are not influenced. These secondaries seem to be ‘regional’. The amount of data (samples) is limited, which could bias the results and discussion. But the sampling sites are interesting for publication in ACP. However, it is not clear whether the analytical data was already obtained in 2007 (or 2008) and is presented now, or that the chemical analyses were performed recently in stored (7 years) samples. This later issue may have affected to state of the organic compound in the samples, and the results and discussion.

Response:

As replied to Referee #2 above, we agree that the number of samples is small, which is limited by the CAREBeijing-2007 campaign itself. The campaign was a pilot study to study the effects of the traffic restriction on the air quality of Beijing and to get experience and scientific evidence for the preparation of the 2008 Olympic game. There were only three consecutive days with traffic restriction (17-19 August), and we took measurements on two days (17 and 19 August). The rest was made before and after the traffic control events.

As for the stability of organic aerosol, the samples were measured in 2010, around 2.5 years after samples collection and storage at -20 °C. We consider that it is within the range of uncertainties.

The manuscript is well written and logically structured. However, there are some comments that need to be taken into account in order to improve the manuscript. The main issue is the fact that the PM concentrations and chemical concentrations in the “clean air” samples are still (very) high in both sites. For example, where does one find average PM_{2.5} concentrations under clean air conditions between 60 and 70 $\mu\text{g}/\text{m}^3$?

Response:

Such PM_{2.5} level (e.g., ~60-70 $\mu\text{g}/\text{m}^3$) was defined as “clean air” by comparing our data with previous PM_{2.5} data in Beijing with ‘blue sky’ occurring. This level is also comparable to the recently released Chinese pollution standard (75 of $\mu\text{g}/\text{m}^3$ PM_{2.5}). Certainly, the concentrations are still much higher than the values measured in urban cities in the developed countries (e.g., US and Europe). Nevertheless, to avoid the misleading to the readers, we changed ‘clean air’ to ‘less polluted air’ in the revised manuscript.

Figure 1 is very small, but with a ‘zoom’ one can see it well.

Response:

The resolution of Figure 1 has been increased in the revised manuscript.

Often the influence of local vs. regional contamination can be observed by the ‘correlations’ between the same chemical in two sites. If regional influences (photochemical aging) is dominant over local (emissions) than one observes similar and correlated concentration variations in time. Is this the case in the present study? Based on figure 1 there seems not to be much correlation and there is also not much variation between days....with vs without restriction...

Response: We realize that the different start time at Beijing and Yufa (i.e., the first three days), as shown in Fig. 1, may mislead the readers. We now point out this in Fig. 1 caption. In general, the aerosol composition at Beijing and Yufa follows a similar trend from 13-31 August. There are a few data points (e.g., for benzoic acid) that do not correlate between Beijing and Yufa. This could be explained by the wind direction and enhanced local emissions (e.g., coal and biomass burning at Yufa), that is, the air mass we measured at these two sites could be different (Yufa is at the south of Beijing).

From Fig. 1 we can see the large variation at different wind sectors. Also the effects of traffic control on PM should be considered when the air mass from same wind sectors.

Page 14866 and 14867 states that in Figure 3 it is visible that there are “substantially” higher concentrations of chemical species, OC and EC under “pollution event” conditions than under “clean air” conditions. The differences are not that much and “clean air” is maybe not the best name for this event, since all levels are high compared to other urban sites.

Response: Following the reviewer’s suggestion, we remove “substantially”. Also we change “clean air” to “less polluted air”.

Any comparison with other “megacities” is missing and would be welcome to understand the high concentrations found in the present study.

Response: We did not tabulate a comparison with previous studies. However, we compared our results with previous results in the main text (see Section 3.1 and 3.2).

The OC/EC ratios are “slightly” different (line 26), but the authors claim that “the low OC/EC ratio during pollution episodes..”. First, the ratios are very similar and, second, the OC/EC of 2.05-2.52 are not low.

Response: the “low” should be “lower” when comparing the OC/EC during pollution episodes with that during less polluted air. We change the description to the following: “The average OC/ EC ratios at less polluted air (PKU: 2.63; Yufa: 2.19) events were slightly higher than those found at the pollution episodes (PKU: 2.52; Yufa: 2.05) at both sites. The

slightly lower OC/EC ratio during pollution episodes is likely associated to high combustion emissions, especially from traffic exhaust. The slightly higher OC/EC ratios observed during less polluted air events suggest that secondary formation of OA was critical during less polluted air event.”

Page 14866 The ratios of C16:0 and C18:0 were used to determine that cooking is “a dominant source” (line 14), however, the observed ratios between 0.6-1.2 could be other: : :such as unpaved/paved roads (line 11). Moreover, normally cooking is accompanied with high levels of C18:1 (relative to C18:0). This was not the case here. There is too much contradiction in the results to point to cooking as a dominant (local) source. Generally, it is not clear whether the storage of the samples affected the results, if not, the authors should justify better that “traffic restrictions” do not affect the air quality much and that concentrations of chemicals and EC, OC, WSOC are high in Beijing at the time of sampling. This is a tough job when there are (only) ten samples.

Response: We agree with the reviewer that it is not straightforward to conclude that cooking is a dominant source by simply using the C18:0 /C16:0 ratios. In the revised manuscript, it reads *“The C18:0/C16:0 ratios observed in this study had a range between 0.64 – 1.17 (average value: 0.85 in both locations) in PKU and Yufa, indicating that contribution of cooking emissions and paved/unpaved road dust cannot be ruled out.”*

As discussed above, we believe the effects from sample storage are within the range of measurement uncertainties. The “traffic restrictions” can reduce the emissions of some primary pollutants (e.g., EC in PKU, see Fig. 5a) and certain secondary aerosol species formed from oxidation of traffic emitted volatile organic compounds (see Fig 5). However, the effect of “traffic restrictions” on air quality could be limited given emission from other sources (e.g., biomass burning and cooking). Further, a very recent study from Huang et al. (2014) shows a majority of secondary aerosol during high pollution events at Beijing. This also explains the high OC and WSOC measured in this study.

Reference:

Huang R.J., et al., High secondary aerosol contribution to particulate pollution during haze events in China. *Nature*, 514, 218-222, 2014.

1 **Dicarboxylic Acids, Ketocarboxylic acids, α -Dicarbonyls, Fatty Acids**
2 **and Benzoic Acid in PM_{2.5} aerosol collected during CAREBeijing-2007:**
3 **an effect of traffic restriction on air quality**

4
5 K. F. Ho^{1,*}, R.-J. Huang^{2,3,*}, Kimitaka Kawamura⁴³, Eri Tachibana⁴³, S. C. Lee⁵⁴, Steven Sai
6 Hang Ho^{1,65}, Tong Zhu⁷⁶, and Linwei Tian⁴⁸

7
8
9 ¹ The Jockey Club School of Public Health and Primary Care, The Chinese University of
10 Hong Kong, Shatin, Hong Kong

11
12 ² Key Laboratory of Aerosol Chemistry and Physics, Institute of Earth Environment, Chinese
13 Academy of Sciences, Xi'an 710061, China

14
15 ³² Laboratory of Atmospheric Chemistry, Paul Scherrer Institute, 5232 Villigen, Switzerland

16
17 ⁴³ Institute of Low Temperature Science, Hokkaido University, Sapporo, Japan

18
19 ⁵⁴ Research Center of Urban Environmental Technology and Management, Department of
20 Civil and Structural Engineering, The Hong Kong Polytechnic University, Hun Hom,
21 Kowloon, Hong Kong

22
23 ⁶⁵ Division of Atmospheric Sciences, Desert Research Institute, 2215 Raggio Pathway, Reno,
24 NV 89512, United States

25
26 ⁷⁶ College of Environmental Sciences and Engineering, Peking University

27
28 ⁸ School of Public Health, The University of Hong Kong, Hong Kong, China

29
30 *Corresponding author: K. F. Ho

31 E-mail address: kfho@cuhk.edu.hk

32 *Corresponding author: R.-J. Huang

33 E-mail address: rujin.huang@psi.ch

34 |
35 | [Jan, 2014](#)
36

37 **Abstract**

38 Thirty water-soluble organic species, including dicarboxylic acids, ketocarboxylic
39 acids, α -dicarbonyls, fatty acids, and benzoic acid were determined as well as organic carbon
40 (OC), elemental carbon (EC) and water-soluble organic carbon (WSOC) in PM_{2.5} samples
41 collected during the Campaign of Air Quality Research in Beijing 2007 (CAREBeijing-2007)
42 in the urban and suburban areas of Beijing. The objective of this study is to identify the
43 influence of traffic emissions and regional transport to the atmosphere in Beijing during
44 summer. PM_{2.5} samples collected with or without traffic restriction in Beijing are selected to
45 evaluate the effectiveness of local traffic restriction measure on air pollution reduction. The
46 average concentrations of the total quantified bifunctional organic compounds (TQBOC),
47 total fatty acids and benzoic acid during the entire sampling period were $1184 \pm 241 \text{ ng m}^{-3}$,
48 $597 \pm 159 \text{ ng m}^{-3}$ and $1496 \pm 511 \text{ ng m}^{-3}$ in PKU, and $1050 \pm 303 \text{ ng m}^{-3}$, $475 \pm 114 \text{ ng m}^{-3}$
49 and $1278 \pm 372 \text{ ng m}^{-3}$ in Yufa. Oxalic acid (C₂) was found as the most abundant dicarboxylic
50 acid at PKU and Yufa, followed by phthalic acid (Ph). A strong even carbon number
51 predominance with the highest level at palmitic acid (C_{16:0}), followed by stearic acid (C_{18:0})
52 was found for fatty acids. According to the back trajectories modeling results, the air masses
53 were found to originate mainly from northeast, passing over southeast or south of Beijing
54 (heavily populated, urbanized and industrialized areas), during heavier pollution events,
55 whereas they are mainly from north or northwest sector (mountain areas without serious
56 anthropogenic pollution sources) during cleaner-less pollution events. The data with wind
57 only from the same sector (minimizing the difference from regional contribution) but with
58 and without traffic restriction in Beijing were analyzed to evaluate the effectiveness of local
59 traffic restriction measure on the reduction of local air pollution in Beijing. The results
60 suggested that the “traffic restriction” measure can reduce the air pollutants, but the decrease
61 of pollutants is generally smaller in Yufa compared to that in PKU. Moreover, an
62 enhancement of elemental carbon (EC) value indicates elevated primary emissions in Yufa
63 during restriction period than non-restriction period. This study demonstrates that even when
64 primary exhaust was controlled by traffic restriction, the contribution of secondary organic
65 species formed from photochemical processes was critical with long-range atmospheric
66 transport of pollutants.

67

68 Keywords: dicarboxylic acids, ketocarboxylic acids, α -dicarbonyls, fatty acids, secondary
69 organic carbon, Chinese aerosols

70

71 **1. Introduction**

72 Organic aerosol (OA) typically constitutes 20–90% of submicron aerosol ([Huang et al.,](#)
73 [2014](#); Jimenez et al., 2009) and is influencing Earth's climate directly by absorbing and
74 scattering radiation and indirectly by acting as cloud condensation. OA also adversely affects
75 air quality and human health. However, uncertainties exist in the effect of OA on health and
76 climate, due in large part to the complexity of the OA composition (Hallquist et al., 2009;
77 [Hoffmann et al., 2011](#); Poschl, 2005) . OA can be primary emitted, or secondary produced by
78 photochemical reactions of gas-phase precursors. Due to polar functional groups formation
79 (e.g., carbonyl, carboxyl and hydroxyl), a major fraction of the SOA is thought to be water-
80 soluble which, together with some water-soluble POA, accounts for about 40–80% of the OA
81 (Jaffrezo et al., 2005; Saxena and Hildemann, 1996) .

82 Despite the dominant presence of WSOC in the atmosphere, there exist large
83 uncertainties associated with sources, the chemical composition, removal mechanisms and
84 atmospheric formation processing of aerosol WSOC. This is particularly evident in polluted
85 megacities where multiple sources of local and regional origins may significantly change the
86 chemical and physical properties of aerosol and therefore influence the air quality, climate
87 and human health. Dicarboxylic acids (diacids) are the most abundant organic compounds in
88 OA, which can be derived from primary emissions and/or secondary formation from different
89 precursor species via photochemical reactions (Glasius et al., 2000; Kawamura et al., 1996;
90 Kundu et al., 2010; Legrand et al., 2007). Fossil fuel combustion and biomass burning
91 (Falkovich et al., 2005; Ho et al., 2006; [Huang et al., 2014](#); Kundu et al., 2010) are the major
92 primary sources whereas photochemical oxidation of volatile organic compounds (VOCs)
93 from biogenic and anthropogenic emissions (Kawamura et al., 1996; Mkoma and Kawamura,
94 2013) are the major secondary sources.

95 Beijing is one of the largest metropolitan cities in Asia and has become a heavily
96 polluted area due to the fast urbanization and industrialization over the past two decades. In
97 2009, more than 17.5 million residents and 4.0 million vehicles were reported in Beijing
98 (BMBS, 2010). Besides local emissions, the air flowed into Beijing from polluted
99 neighboring regions can have significant impact to the air quality in Beijing (Hatakeyama et
100 al., 2005; Luo et al., 2000; Mauzerall et al., 2000). Especially, the gas-to-particle partitioning
101 of semi-volatile organic compounds (SVOCs) and their subsequent aging via photochemical
102 processing during transport has been recognized to be a major air pollution source (Ding et
103 al., 2008; Guttikunda et al., 2005). Atmospheric aerosols have been investigated extensively
104 in China (An et al., 2007; Cao et al., 2003; ; [Huang et al., 2014](#); Xu et al., 2008). However,

105 relevant studies on organic acids are still very scarce. With such limited information available
106 | on organic acids despite the rapid urbanization and development (especially the increase in
107 traffic density), it is essential to seek a better understanding of organic acids in Beijing. For
108 the promised “Green Olympic Game” in 2008, many pollution control measures, such as
109 controlling traffic, halting industrial/construction activities, and sweeping roads, was taken to
110 improve the air quality. The ‘traffic restriction’ measure, which only allowed vehicles to be
111 on road in alternative business days according to their even and odd plate numbers, was
112 proposed to reduce air pollution.

113 To investigate the effects of the traffic restriction on the air quality of Beijing and to
114 accumulate experience and scientific evidence for the preparation of the 2008 Olympic game,
115 we conducted aerosol ($PM_{2.5}$) monitoring at two sites in Beijing during August 3–31, 2007. In
116 this study, $PM_{2.5}$ samples collected were analyzed by gas chromatography flame ionization
117 detector (GC-FID) and gas chromatography mass spectrometry (GC-MS) to determine the
118 composition of low molecular weight (MW) diacids (C_2 – C_{12}), ketocarboxylic acids (ωC_2 – ωC_9 ,
119 pyruvic acid), α -dicarbonyls (C_2 – C_3), benzoic acid and fatty acids (C_{12} – C_{25}). Moreover, OC,
120 EC, and WSOC were also analyzed. Through the intensive sampling campaign, the roles of
121 regional transport, local emissions and secondary formations of particulate matter in the
122 atmosphere of Beijing were investigated.

123

124 **2. Experiment**

125 *2.1 $PM_{2.5}$ sampling*

126

127 Two sampling locations, Peking University (PKU) (39.98°N, 116.35°E) and Yufa
128 (39.51°N, 116.31°E) were selected in this study. The detailed descriptions of the sampling
129 locations were reported elsewhere (Ho et al., 2010). The air samplers were placed on the top
130 floor of the buildings (PKU: a 6-story building; Yufa: 4-story building). The meteorological
131 data such as wind speed, wind direction, relative humidity, and temperature were collected
132 during the sampling period. North and northwest of PKU are enclosed by mountains whereas
133 south and southeast of Yufa are surrounded by heavily industrialized and urbanized areas
134 such as Hebei province and Tianjin city.

135 | Pre-heated (800 °C, 3h) quartz-fiber filters (47 mm QM-A Whatman quartz filters)
136 were used to collect 24 h integrated $PM_{2.5}$ samples by Airmetrics mini-volume $PM_{2.5}$ samplers
137 at a flow rate of 5 L min^{-1} . A DryCal[®] flow meter (BIOS International, Butler, NJ, USA) was
138 used to calibrate the sampling flows before and after the sampling. Sampling was carried out

139 simultaneously from 09:00 a.m. to 09:00 a.m. local time at the two sampling locations from
140 3rd to 31st August, 2007. The samples were properly kept in a freezer (−20°C) to prevent
141 evaporation of semi volatile components and microbial degradation of organics.

142

143 2.2 Chemical analysis

144 OC and EC were analyzed (on a 0.526 cm² punch) by thermal analysis with optical
145 detection following the IMPROVE protocol on a Desert Research Institute (DRI) Model 2001
146 Thermal/Optical Carbon Analyzer (Atmoslytic Inc., Calabasas, CA, USA) (Cao et al., 2003;
147 Chow et al., 2005). The method detection limit (MDL) of OC and EC analysis is 0.8 and 0.4
148 μgC cm⁻², respectively. To determine the WSOC, a total area of 2.63 cm² of the sample filter
149 was cut from each filter and 5 ml of Milli-Q water (18 MΩ) was added into a 15 ml vial
150 where the sample was placed. Ultra-sonic water bath was used to extract the particles on the
151 filter for 1 hr. Syringe filters (0.2 μm PTFE membrane) were used to remove the insoluble
152 particles from the extracts. Filtered extract was then transferred into clean vials and analyzed
153 total organic carbon (TOC) by using a Shimadzu TOC-V CPH Total Carbon Analyzer
154 (Columbia, MD, USA). The MDL is 0.01 μgC m⁻³, with a precision of ±5%. The data
155 reported in this study were all corrected by the blanks.

156 The analytical procedures for water-soluble organic species were well reported
157 elsewhere (Kawamura and Yasui, 2005). Briefly, the sample was extracted with organic-free
158 water (10 ml × 3) to isolate bifunctional organic compounds as well as fatty acids and benzoic
159 acid. After the extracts were concentrated using a vacuum rotary evaporator, 14% BF₃/*n*-
160 butanol were added at 100 °C to convert the aldehyde groups to dibutoxy acetals and carboxyl
161 groups to butyl esters. Homologous series of fatty acids were analyzed as butyl esters
162 (Mochida et al., 2007). No serious contamination (< 5% of real samples) was observed in our
163 analysis. The data reported in this study were corrected by the field blanks. The derivatized
164 samples were determined by a Agilent 6890GC/FID (Palo Alto, CA, USA) equipped with a
165 split/splitless injector, HP-5 fused silica capillary column (25 m × 0.2 mm i.d. x 0.5 μm film
166 thickness) and an FID detector. Peak identifications were relied on the retention times of
167 authentic standards. ThermoQuest Trace MS (Austin, TX, USA) with a similar GC conditions
168 was used for mass spectral confirmation of the compounds. The reproducibility of the
169 methods was < ±15%; recoveries of the bifunctional organic compounds fatty acids and
170 benzoic acid were > 70% (Kawamura and Yasui, 2005; Mochida et al., 2007). Field blanks
171 concentrations were < 15% of real samples, except for phthalic acid (up to 30%). The results
172 shown in this study were all corrected by the field blanks.

173

174 3. Results and discussion

175 3.1 Overview of molecular compositions of bifunctional organic compounds in PKU and 176 Yufa

177 Average OC, EC, and WSOC concentrations in PKU and Yufa are illustrated in Table
178 1 and their levels during the entire sampling period were 14.9 ± 2.47 , 6.21 ± 1.90 and $5.59 \pm$
179 $1.49 \mu\text{gC m}^{-3}$ in PKU, and 11.1 ± 3.68 , 5.6 ± 1.83 and $4.55 \pm 1.79 \mu\text{gC m}^{-3}$ in Yufa,
180 respectively. The WSOC accounted for $37 \pm 7\%$ and $40 \pm 7\%$ of OC in PKU and Yufa,
181 respectively. It was consistent with the WSOC/OC ratios (20–40%) at other metropolitan
182 cities (Ho et al., 2007; Yang et al., 2005), suggesting that WSOC is one of the main
183 components in OA in China. Yufa is located at southern Beijing, which is close to the border
184 of Beijing Municipality and Hebei Province. Regional pollution from heavy industrialized and
185 urbanized areas, like Hebei province and Tianjin city, have a great impact to the air quality of
186 Yufa area.

187 The concentrations of bifunctional organic compounds measured in PKU and Yufa are
188 presented in Table 1. The concentrations of total quantified bifunctional organic compounds
189 (TQBOC) varied from 730 to 1455 ng m^{-3} (average concentration: $1184 \pm 241 \text{ ng m}^{-3}$) in PKU,
190 and from 554 to 1621 ng m^{-3} (average concentrations: $1050 \pm 303 \text{ ng m}^{-3}$) in Yufa. The results
191 are higher than measurements (average 813 ng m^{-3} in PKU; average 771 ng m^{-3} in Yufa)
192 reported in 2006 in same sampling locations (Ho et al., 2010), reflecting that there were
193 continuous increases of primary emissions and more aging of aerosols in Beijing. However,
194 the concentrations are close to those megacities studied recently (Ho et al., 2007).

195 Oxalic acid (C_2) was the most abundant diacid ($435 \pm 124 \text{ ng m}^{-3}$ and $418 \pm 130 \text{ ng m}^{-3}$
196 at PKU and Yufa, respectively) determined in this study, followed by phthalic acid (Ph)
197 ($209 \pm 28.8 \text{ ng m}^{-3}$ and $176.3 \pm 91.5 \text{ ng m}^{-3}$), and succinic acid (C_4) ($89.9 \pm 27.7 \text{ ng m}^{-3}$ and
198 $80.9 \pm 26.9 \text{ ng m}^{-3}$). These three species accounted for 65% of TQBOC in PKU and Yufa,
199 respectively. Oxalic acid was also recognized as predominant diacid in previous studies in
200 China (Ho et al. 2010, 2011). C_2 can be either released from combustion processes (e.g., fossil
201 fuel and biomass burning) (Kawamura and Kaplan, 1987; Narukawa et al., 1999) or
202 secondary produced by the oxidation of VOCs (Carlton et al., 2006; Warneck, 2005).

203 The average phthalic acid (Ph) concentrations measured in this study are substantially
204 higher than those reported by other studies (Ho et al., 2007; Wang and Kawamura, 2005).
205 Three phthalic acids (phthalic acid (*o*-isomer), terephthalic acid (*p*-isomer) and isophthalic
206 acid (*m*-isomer)) were determined and these isomer species distribution was dominated by *o*-

207 isomer, followed by *p*-isomer and *m*-isomer, which are consistent with studies measured in
208 Mt. Tai, China and Pearl River Delta region (Fu et al., 2008; Ho et al., 2011). The abundant
209 phthalic acid (Ph) can be released from incomplete combustion processes or secondary
210 formed by oxidation of aromatic compounds (e.g., naphthalenes, NAP) (Kawamura and
211 Kaplan, 1987; Kawamura and Yasui, 2005). In some previous studies, high levels of NAP
212 were found in Beijing urban areas (Liu et al., 2007; Tao et al., 2007) and it can be thus one of
213 the potential precursors to phthalic acid (Ph) formation (Ho et al., 2007).

214 Besides diacids (C₂-C₄), azelaic acid (C₉) was the most abundant species among the
215 saturated diacids in both sampling locations (71.4 ± 8.91 ng m⁻³ in PKU; 49.2 ± 8.99 ng m⁻³ in
216 Yufa). C₉ is recognized as a photochemical reaction product of biogenic unsaturated fatty
217 acids, such as oleic (C_{18:1}) and linoleic (C_{18:2}) acids [\[the first number is carbon number and the](#)
218 [latter one refers to the amount of double bond\]](#) (Kawamura and Gagosian, 1987) and is
219 generally abundant in the high molecular weight homologues. The unsaturated fatty acids
220 are commonly determined in marine micro-organism or higher plant leaves. However, these
221 acids could be released by meat charbroiling also (Rogge et al., 1991). During long range
222 transport, photochemical oxidation of C_{18:1} to C₉ via oxidants (e.g., ozone and/or OH radicals)
223 may occur in the air (Stephanou and Stratigakis, 1993). The C₉/C_{18:1} ratio determined was
224 lower in Yufa (average value: 2.1) than that in PKU (average value: 5.12) ~~which suggests~~
225 that significant secondary production of C₉ occurred in urban area of Beijing.

226 Diacids can be formed when ketocarboxylic acids, which are regarded as
227 intermediates product of mono-carboxylic acids oxidation, react with other pollutants in the
228 air (He et al., 2013; Kawamura et al., 1996). The concentrations of total measured
229 ketocarboxylic acids varied from 87.4 to 169 ng m⁻³, (average value: 122 ± 28.8 ng m⁻³) in
230 PKU and from 52.0 to 131 ng m⁻³ (average value: 97.0 ± 22.9 ng m⁻³) in Yufa (Table 1). The
231 concentrations in both sampling sites are higher than those measured in rural site in Gosan,
232 South Korea (53 ng m⁻³) and megacities in China (summer: 37 ng m⁻³) (Ho et al., 2007; 2010;
233 Kawamura et al., 2004). These results reveal that the organic aerosols in PKU and Yufa were
234 likely more photochemically aged than that in other urban sites caused by photochemical
235 reaction during transportation (He et al., 2013). Glyoxylic acid (ωC₂) was found as the most
236 abundant ketocarboxylic acid, followed by pyruvic acid (Pyr). Their concentration levels are
237 similar to previous measurement in Tokyo, Japan (Kawamura and Yasui, 2005).

238 Concentrations of total measured α-dicarbonyls varied from 35.5 to 99.5 ng m⁻³
239 (average value: 51.8 ± 17.9 ng m⁻³) in PKU and from 29.0 to 61.4 ng m⁻³ (average value: 44.2
240 ± 10.3 ng m⁻³) in Yufa. The two simplest α-dicarbonyl compounds (Glyoxal and

241 methylglyoxal) have recently attracted much attention as potential SOA precursors. These
242 compounds are formed by ~~both~~ photochemical oxidation of both biogenic (e.g., isoprene and
243 terpenes) and anthropogenic VOCs (e.g., toluene, xylene) (Fick et al., 2004; Volkamer et al.,
244 2001). They have been identified as the significant precursors in the heterogeneous processes
245 for SOA formation (Kroll et al., 2005). High concentrations of glyoxal and methylglyoxal
246 observed indicate the greater SOA formation potential in this region. α -Dicarbonyls levels
247 measured in PKU and Yufa were higher than previous results in other cities of China (average
248 value: 12 ng m^{-3}) (Ho et al., 2007). It indicates that the biogenic sources such as oxidation of
249 isoprene are more important than other urban cities in China.

250

251 3.2 Overview of molecular compositions of fatty acids and benzoic acid in PKU and Yufa

252 Table 1 presents the average concentrations of straight chain saturated fatty acids
253 ($\text{C}_{12:0}$ - $\text{C}_{25:0}$, the first number is carbon-number and the latter one refers to the amount of
254 double bonds), unsaturated fatty acid and benzoic acid. Total measured fatty acids
255 concentrations varied from 459 to 1003 ng m^{-3} (average value: $597 \pm 159 \text{ ng m}^{-3}$) in PKU and
256 from 375 to 684 ng m^{-3} (average value: $475 \pm 114 \text{ ng m}^{-3}$) in Yufa. The distributions of fatty
257 acids were dominated by even carbon number with maximum at palmitic acid ($\text{C}_{16:0}$),
258 followed by stearic acid ($\text{C}_{18:0}$). This finding is consistent with previous measurements
259 reported in megacities of China (Fu et al., 2008; Ho et al., 2010). Both natural biogenic and
260 anthropogenic emissions represent the major sources of fatty acids, whereas, homologs $< \text{C}_{20}$
261 are partially released from microbial sources (Simoneit and Mazurek, 1982). Additionally,
262 low MW fatty acids ($< \text{C}_{18}$) can be emitted by tire wear debris and traffic exhaust. Biomass
263 burning also produces high fractions of fatty acids which are the major components of plant
264 tissues and surface waxes. $\text{C}_{16:0}$ and $\text{C}_{18:0}$ were also the major organic compounds emitted
265 from the meat cooking (Schauer et al., 1999, 2002; Zhao et al., 2007a, b). Higher
266 concentrations of fatty acids observed in PKU can be explained by the mixed contributions of
267 regional and local emissions in urban area. Interestingly the contributions of total quantified
268 fatty acids to OC are similar in both sites (3.1% in PKU and 3.2% in Yufa, respectively).

269 The even-over-odd carbon number preference in fatty acid ($\text{C}_{12:0}$ to $\text{C}_{25:0}$) is measured
270 by Carbon Preference Index (CPI) :

271

$$\text{CPI}_{\text{fatty acid}} = \frac{\Sigma \text{ Even carbon number fatty acids}}{\Sigma \text{ Odd carbon number fatty acids}}$$

272

273 CPI is a measure to differentiate anthropogenic and biogenic sources and the values are 43.3
274 in PKU and 45.9 in Yufa, respectively. High CPI values observed in this study indicate that
275 biological sources such as vascular plant has significant influence in this region (Simoneit,
276 1984).

277 In this study, $C_{18:1}$ was detected in all samples which can be directly emitted from
278 higher plants and soils. In urban areas, biomass burning and cooking are likely to be the main
279 anthropogenic sources for this acid (Rogge et al., 1993). Its concentrations varied from 2.94 to
280 33.0 ng m^{-3} (average value: $24.3 \pm 8.93 \text{ ng m}^{-3}$) and from 13.0 to 47.9 ng m^{-3} (average value:
281 $24.6 \pm 9.23 \text{ ng m}^{-3}$) in PKU and Yufa, respectively. Oleic acid is a good tracer for unsaturated
282 organic aerosol and a representative compound for reactivity model (Rudich et al., 2007). The
283 diagnostic ratio of $C_{18:1}/C_{18:0}$ was used to determine the level of aerosol aging in this study.
284 Low values indicate that the air masses are more aged. The ratios in PKU and Yufa were 0.12
285 and 0.14, respectively, which suggests that unsaturated fatty acids are depleted by the
286 enhanced photochemical degradation in PKU (Wang et al., 2006). Moreover, the diagnostic
287 ratio of $C_{18:0}/C_{16:0}$ was applied as an indicator for source evaluation. Low ratios observed (<
288 0.25) in $\text{PM}_{2.5}$ were likely originated from wood smoke, waxy leaf surface abrasions, and
289 foliar vegetation combustion; ratios that ranged between 0.25-0.5 were indicated for vehicle
290 exhausts; while ratios that ranged between 0.5-1 were obtained from hamburger charbroiling
291 and paved/unpaved road dust (Oliveira et al., 2007; Rogge et al., 2006). The $C_{18:0}/C_{16:0}$ ratios
292 observed in this study had a range between 0.64 – 1.17 (average value: 0.85 in both locations)
293 in PKU and Yufa, indicating ~~a dominant source from~~ that the contribution of cooking
294 emissions and paved/unpaved road dust cannot be ruled out.

295 Almost all $\text{PM}_{2.5}$ samples collected contained benzoic acid which has been identified
296 as a direct pollutant from the traffic emissions (Kawamura et al., 1985) and a indirect
297 pollutant produced from photo-degradation of aromatic compounds (e.g., toluene) released
298 from traffic exhausts (Suh et al., 2003). The average benzoic acid concentrations were $1496 \pm$
299 511 ng m^{-3} in PKU and $1278 \pm 372 \text{ ng m}^{-3}$ in Yufa, respectively. Although, benzoic acid is
300 semi-volatile organic species and is mainly found in gas phase (Fraser et al., 2003), it can be
301 ~~formed~~ observed in particulate phase via gas-to-particle partitioning. During ozone episode in
302 August 2006, high concentration of toluene was determined in Beijing ($11.4 \mu\text{g m}^{-3}$) (Duan et
303 al., 2008), which suggests that oxidation of toluene is one of the significant sources of benzoic
304 acid in the air.

305

306 3.3 *Less polluted ~~Clean~~ air versus pollution events*

307 Figure 1a and 1b show the temporal variation of mass concentrations of EC, OC and
308 WSOC in PKU and Yufa from 3 to 31 August 2007, respectively. Heavier air pollution events
309 were observed during 3, 5, 9, 15 and 31 August, as reflected by the elevated PM_{2.5}
310 concentrations (i.e., range 96-191 $\mu\text{g m}^{-3}$, average 124 $\mu\text{g m}^{-3}$ in PKU and range 100-127 μg
311 m^{-3} , average 110 $\mu\text{g m}^{-3}$ in Yufa, respectively). The concentrations of OC, EC, and WSOC
312 significantly increased during these pollution events, but generally decreased for the ~~less~~
313 ~~polluted clean~~-air mass events on 7, 13, 21 and 27 August, consistent with lower PM_{2.5}
314 concentrations (i.e., a range of 65-77 $\mu\text{g m}^{-3}$, average 71 $\mu\text{g m}^{-3}$ in PKU and a range of 39-179
315 $\mu\text{g m}^{-3}$, average 62 $\mu\text{g m}^{-3}$ in Yufa, respectively). Similar temporal variations in total
316 quantified bifunctional organic compounds and fatty acids ~~werehave-been~~ observed in both
317 PKU and Yufa (Figure 1c and 1d). However, the temporal variation of benzoic acid is
318 different from the other compounds measured, indicating a different source or atmospheric
319 processing for benzoic acid.

320 Ensemble 3 day air mass back trajectory analysis shows that the pollutants were
321 mainly from northeast, passing over southeast or south of Beijing, during heavier pollution
322 events, whereas they were mainly from north or northwest sector during ~~less pollution cleaner~~
323 events (see Figure 2). South and southeast areas of Beijing are located close to the heavily
324 industrialized ~~areaszooms~~ (e.g., Tianjin city, Shandong and Hebei province) whereas north
325 and northwest areas of Beijing are enclosed by the massive mountain ranges ~~that~~ without ~~the~~
326 impact of anthropogenic pollution sources (Ho et al., 2010). As seen in Figures 3a ~~and~~ -b, the
327 concentration levels of EC, OC, WSOC, diacids and ketocarboxylic acids in PKU and Yufa
328 are ~~substantially~~ higher for heavier pollution episodes compared to the ~~less polluted clean~~-air
329 events, suggesting that high emission of carbonaceous aerosols and their precursor gases from
330 neighboring provinces and the subsequent transport to Beijing is one of the major sources
331 responsible for the elevated particulate pollutants in Beijing.

332 The OC to EC ratio (OC/EC) was used to estimate the transformation and emission
333 properties of carbonaceous aerosol. The average OC/EC ratios at ~~less polluted clean~~-air (PKU:
334 2.63; Yufa: 2.19) events were slightly higher those found at the pollution episodes (PKU: 2.52;
335 Yufa: 2.05) at both sites. The ~~slightly lower low~~-OC/EC ratio during pollution episodes is
336 likely associated to high combustion emissions, especially from traffic exhaust. The ~~slightly~~
337 ~~higher high~~-OC/EC ratios observed during ~~less polluted clean~~-air events suggest that
338 secondary formation of OA was critical during ~~less polluted clean~~-air event. Bendle et al.
339 (2007) reported that the unsaturated-over-saturated C₁₈ fatty acids (C_{18:n}/C_{18:0}) ratio could be
340 used as a good indicator to estimate the freshness of OM in marine samples. In this study,

341 high ratios were recorded in samples associated with pollution episode, whereas low ratios
342 were observed in less polluted ~~clean~~-air event with air masses originated from the north and
343 northwest of Beijing. Low ratios observed in less polluted ~~clean~~-air event represent an aged
344 air mass, ~~i.e.~~, indicating longer residence -time for particle transformation and transportation
345 (Alves et al., 2007).

346 Moreover, malonic acid (C_3) can be a byproduct of photochemical breakdown of
347 succinic acid (C_4) in the air. The C_3/C_4 ratio, which was used as a tracer of the enhanced
348 photochemical aging of OA (Kawamura and Ikushima, 1993), observed during less polluted
349 ~~clean~~-air event was higher than pollution episode in both sites (0.66 versus 0.58 in PKU and
350 0.57 versus 0.52 in Yufa). Higher C_3/C_4 ratios in less polluted ~~clean~~-air event suggest that
351 secondary formation of diacids are more significant in less polluted ~~clean~~-air event, which
352 further indicates secondary photochemical formation of particulate diacids is also critical
353 during less polluted ~~clean~~-air event.

354 It should be noted, however, that the concentrations of α -dicarbonyls and benzoic acid
355 in both PKU and Yufa are higher during less polluted air ~~clean~~-episodes compared to
356 pollution episodes. This indicates that local production or secondary formation could be
357 important source for these compounds. It is known that α -dicarbonyls are intermediate
358 reaction products (via photochemical oxidation) of a wide range of biogenic and
359 anthropogenic VOCs (Galloway et al., 2009). More distant sources lead to longer transport
360 time and therefore to increase chemical oxidation of glyoxal and methylglyoxal to their
361 corresponding acids and further reaction products. This potentially reduces the local
362 contribution of α -dicarbonyls in Beijing. Positive correlation was observed between α -
363 dicarbonyls and benzoic acid ($R^2 = 0.82$ in PKU and $R^2 = 0.65$ in Yufa) at both sites (Figure
364 4a and 4b), which further suggest~~ing~~ that a major fraction of α -dicarbonyls and benzoic acid
365 are most likely produced in the local atmosphere of Beijing through photochemical processing.

367 3.4 *Influence of local traffic on air quality between restriction and non-restriction periods*

368 One goal of this sampling campaign is to study the traffic controls influence on the air
369 quality in Beijing given the use of a large number of vehicles and the resulting high emission
370 of particulate matter and precursor gases. As described above, the level of particulate
371 pollutants in Beijing is significantly influenced by regional transport depending on the wind
372 sector. Therefore, in the following discussion, only events with wind from the same sector
373 (minimizing the difference from regional contribution) but with and without traffic restriction
374 in Beijing are selected to evaluate the effectiveness of local traffic restriction measure on air

375 | pollution reduction. Measurements taken on 17 and 19 August represent the restriction events
376 | and those taken on 3, 5, 9, 15 and 31 August represent the non-restriction events.

377 | The concentration ratios of the restriction to the non-restriction periods (R/N) are
378 | shown in Figure 5. A value of close to unity represents that the restriction does not have any
379 | impact in the pollution controls. In PKU, the R/N ratios of EC, OC, WSOC, total diacids, total
380 | ketocarboxylic acids and total α -dicarbonyls are much lower than 1, suggesting that these
381 | pollutants ~~or their~~ precursors are closely related to the traffic emissions and that the
382 | “traffic restriction” measure can reduce primary pollutants (e.g., EC) and the precursors of
383 | secondary pollutants (e.g., diacids and α -dicarbonyls). A previous study (Zhang et al., 2011)
384 | also indicated the reduction of anthropogenic elements in Beijing during the traffic restriction
385 | period of August, 2007. The average OC/EC ratios observed at traffic restriction period (PKU:
386 | 2.69) was slightly higher than that found at non-restriction period (PKU: 2.52). The slightly
387 | lower OC/EC ratio during non-restriction period was mainly due to the higher EC emissions
388 | from traffic exhaust, while EC emissions were reduced during traffic restriction period.
389 | ~~combustion emissions, especially those from traffic exhaust.~~ However, the R/N ratios of
390 | benzoic acid and total fatty acids are higher than 1. A possible explanation for this elevated
391 | R/N ratios is that these organics are mainly derived from regional emissions. An alternative is
392 | that they are mainly produced from sources other than vehicle emissions. For example,
393 | cooking emission that was not controlled under traffic restriction period is a significant source
394 | of fatty acids in the air. More household cooking activities can be found if the residents
395 | trended to stay home during the restriction period.

396 | The profile of R/N ratio in Yufa is different from that in PKU. The concentrations of
397 | OC, WSOC, total diacids and total fatty acids were lower during restriction period than those
398 | during non-restriction period, suggesting that the “traffic restriction” measure indeed reduced
399 | particulate pollutants. However, the decrease is generally smaller in Yufa compared to that in
400 | PKU, indicating that the contribution of local traffic emission to air pollution in Yufa is
401 | smaller. The R/N value >1 occurred to EC, total ketocarboxylic acids, total α -dicarbonyls and
402 | benzoic acid. An enhanced EC value indicates elevated primary emissions in Yufa during
403 | restriction period than non-restriction period. The potential contribution could be local rural
404 | emissions (e.g., biomass burning and coal burning) and/or regional transport from polluted
405 | neighboring provinces that are closer to Yufa. The average OC/EC ratios at traffic non-
406 | restriction period (Yufa: 2.05) events were slightly higher than those found at restriction
407 | period (Yufa: 1.89). The lower OC/EC ratios during restriction period further suggest the
408 | elevated ~~primary~~ emissions of EC from sources other than traffic at Yufa.

409

410 3.5 Ratios of selected species

411 The C_3/C_4 ratios measured in this study varied from 0.28 to 0.84 (average value: 0.59)
412 which are close to those measured in Northern China (0.61) (Ho et al., 2007), but higher than
413 that observed from traffic exhausts (0.3-0.5) (Kawamura and Kaplan, 1987). However, the
414 ratios determined in this study are much lower than the marine particles measured from
415 Pacific Ocean, where photochemical processing is commonly more intensive (Kawamura and
416 Sakaguchi, 1999). Higher C_3/C_4 ratios were observed in PKU (0.62) than in Yufa (0.56),
417 additionally, the ratios observed during traffic restriction period were higher than non-
418 restriction periods in both sites (0.65 versus 0.58 in PKU and 0.61 versus 0.52 in Yufa). This
419 result suggests that C_3 is vigorously produced in traffic restriction period by photochemical
420 reaction of C_4 (Kawamura and Ikushima, 1993). Even though variations of the ratio were
421 small, these are sufficiently representatives to any minor rotations and vibrations of emission
422 sources. The results also suggested that secondary formation of diacids by photochemical
423 oxidation was critical during traffic restriction period despite primary exhaust was controlled.

424 Adipic acid (C_6) is considered as a reaction product of the photochemical oxidation of
425 cyclohexene, whereas C_9 is mainly emitted from unsaturated fatty acids (Hatakeyama et al.,
426 1987; Kawamura and Gagosian, 1987). Therefore C_6/C_9 ratio has been applied to evaluate the
427 abundances of biogenic and anthropogenic sources to OA (Kawamura and Yasui, 2005).
428 C_6/C_9 ratios show higher values in non-restriction period (PKU: 0.40; Yufa: 0.61) than in
429 restriction period (PKU: 0.36; Yufa: 0.38) in this study. Higher C_6/C_9 ratios observed in non-
430 restriction period support that anthropogenic organic compounds, especially from vehicles,
431 are the major source of OA during that period of time.

432 EC is a major component of vehicle exhaust, whereas C_2 is a major secondary organic
433 species in the air. Therefore, C_2/EC ratio can be used to assess the aging of the air mass. The
434 average C_2/EC ratios were 0.075 and 0.078 at PKU and Yufa, respectively, (which has a
435 range of 0.044 to 0.113), which are much higher than previously reported traffic exhaust ratio
436 (0.0022), but similar to those measured in the air over Shenzhen (0.063 in summer) (Huang
437 and Yu, 2007). The C_2/EC ratios generally showed higher values in restriction period (PKU:
438 0.081; Yufa: 0.077) than in non-restriction period (PKU: 0.067; Yufa: 0.074). The results are
439 consistent with the notion that the “traffic restriction” measure can reduce primary pollutants
440 (e.g., EC), ~~but enhance the photochemical oxidation of the precursors of C_2 in the air.~~

441 Moreover, C_2 /total diacids ratio can be applied as an indicator to assess the aging of
442 OA (Kawamura and Sakaguchi, 1999). In this study, the abundances of C_2 in total diacids

443 varied from less than 30% to 54%. Interestingly, the ratios of C₂/total diacids generally
444 showed higher values in restriction period than in non-restriction period. The result indicates
445 that oxalic acid is preferentially formed in restriction period by the oxidation of its precursors
446 (other than anthropogenic VOCs, biogenic VOCs and their oxidation products may serve as
447 important precursors in restriction period) in the atmosphere. Further, ωC₉ is generated by
448 biogenic unsaturated fatty acids oxidation, revealing higher concentrations in restriction
449 period (PKU: 3.47 ng m⁻³; Yufa: 2.49 ng m⁻³) than in non-restriction period (PKU: 1.82 ng m⁻³;
450 ³; Yufa: 2.12 ng m⁻³) (Yokouchi and Ambe, 1986). This result indicates that biogenic
451 emissions are important source for the formation of ωC₉ in restriction period, which can
452 further breakdown to produce lower molecular weight diacids including C₄, C₃, and C₂. The
453 results further indicate that secondary formation of diacids by atmospheric oxidation was also
454 critical during traffic restriction period despite primary exhaust was controlled.

455

456 **4. Summary and Conclusions**

457 During the CAREBeijing-2007 in summer, molecular compositions of bifunctional
458 organic compounds, fatty acids and benzoic acid were studied in Beijing. Oxalic acid (C₂)
459 was detected as the most abundant diacid followed by phthalic (Ph) acid. Low MW
460 bifunctional organic compounds were found as the major water-soluble organic fraction,
461 accounting for more than 8.9% and 10.3% of WSOC in PKU and Yufa, respectively.
462 Additionally, total fatty acids and benzoic acid contributed 3.1% and 7.2% of OC in PKU and
463 3.2% and 9.3% of OC in Yufa, respectively. Bifunctional organic compounds can be released
464 from primary emissions (e.g., traffic exhaust and biomass burning) or formed by atmospheric
465 oxidation of VOCs in the Beijing atmosphere. Both natural biogenic (e.g., microbial) and
466 anthropogenic (e.g., traffic exhaust, ~~meat~~-cooking) sources provide the major inputs of fatty
467 acids, whereas benzoic acid was mainly formed by the photodegradation of aromatic
468 compounds such as toluene from traffic emission.

469 The concentrations of OC, EC and WSOC significantly increased during the heavy
470 pollution events, but generally decreased during the less pollution ~~clean~~-events. Results of
471 back trajectory analyses indicated that the air masses were originated mainly from northeast,
472 passing over heavily populated, urbanized and industrialized areas during the heavy pollution
473 events, whereas they were mainly from mountain clean areas during less pollution ~~cleaner~~
474 events.

475 In PKU, the restriction to non-restriction period (R/N) ratios of OC, EC, WSOC, total
476 diacids, total ketocarboxylic acids and total α-dicarbonyls were much lower than 1,

477 suggesting that the “traffic restriction” measure can reduce primary pollutants (e.g., EC) and
478 the precursors of secondary pollutants (e.g., diacids and α -dicarbonyls). The R/N ratios of OC,
479 WSOC, total diacids and total fatty acids in Yufa were lower than 1, however, the values are
480 generally larger than those in PKU. Moreover, the R/N value >1 occurred to EC, total
481 ketocarboxylic acids, total α -dicarbonyls and benzoic acid, indicating that there are higher
482 contribution of local emissions (e.g., coal and biomass burning) and/or regional transport
483 from polluted neighboring provinces than local traffic emission in Yufa.

484 The C_3/C_4 , C_2/EC and $C_2/\text{total diacids}$ ratios observed during traffic restriction period
485 were higher than those of non-restriction periods at both sites. This result suggests that C_2 and
486 C_3 are secondarily more produced in traffic restriction period by the photochemical oxidation
487 of their precursors, indicating that even when primary exhaust was controlled, secondary
488 photochemical formation of particulate diacids was not controlled during traffic restriction
489 period. This study demonstrates that atmospheric oxidizing capability (photochemical aging)
490 is enhanced by the reduction of atmospheric loading of aerosol particles during the traffic
491 restriction period possibly due to the increased solar radiation reaching to the ground surface.

492

493 **Acknowledgements**

494 This study is partially supported by Research Grants Council of the Hong Kong
495 Special Administrative Region (Project No. 412612), the Strategic Priority Research Program
496 of the Chinese Academy of Science (XDA05100401), and also by a Grant-in-Aid No.
497 19204055 from the Japan Society for the Promotion of Science.

498

499 **Reference**

- 500 Alves, C., Oliveira, T., Pio, C., Silvestre, A. J. D., Fialho, P., Barata, F., and Legrand, M.:
501 Characterisation of carbonaceous aerosols from the Azorean Island of Terceira, *Atmos.*
502 *Environ.*, 41, 1359-1373, doi:<http://dx.doi.org/10.1016/j.atmosenv.2006.10.022>, 2007.
- 503 An, X., Zhu, T., Wang, Z., Li, C., and Wang, Y.: A modeling analysis of a heavy air pollution
504 episode occurred in Beijing, *Atmos. Chem. Phys.*, 7, 3103-3114, 2007.
- 505 Bendle, J., Kawamura, K., Yamazaki, K., and Niwai, T.: Latitudinal distribution of terrestrial
506 lipid biomarkers and n-alkane compound-specific stable carbon isotope ratios in the
507 atmosphere over the western Pacific and Southern Ocean, *Geochim. Cosmochim. Acta*,
508 71, 5934-5955, doi:10.1016/j.gca.2007.09.029, 2007.
- 509 Beijing statistical yearbook: <http://www.bjstats.gov.cn/>, access: Feb 2014, 2010.
- 510 Cao, J. J., Lee, S. C., Ho, K. F., Zhang, X. Y., Zou, S. C., Fung, K., Chow, J. C., and Watson,
511 J. G.: Characteristics of carbonaceous aerosol in Pearl River Delta Region, China
512 during 2001 winter period, *Atmos. Environ.*, 37, 1451-1460, doi:10.1016/s1352-
513 2310(02)01002-6, 2003.
- 514 Carlton, A. G., Turpin, B. J., Lim, H. J., Altieri, K. E., and Seitzinger, S.: Link between
515 isoprene and secondary organic aerosol (SOA): Pyruvic acid oxidation yields low
516 volatility organic acids in clouds, *Geophys. Res. Lett.*, 33, L06822,
517 doi:10.1029/2005gl025374, 2006.
- 518 Chow, J. C., Watson, J. G., Chen, L. W. A., Paredes-Miranda, G., Chang, M. C. O., Trimble,
519 D., Fung, K. K., Zhang, H., and Yu, J. Z.: Refining temperature measures in
520 thermal/optical carbon analysis, *Atmos. Chem. Phys.*, 5, 2961-2972, 2005.
- 521 Ding, A. J., Wang, T., Thouret, V., Cammas, J. P., and Nedelec, P.: Tropospheric ozone
522 climatology over Beijing: analysis of aircraft data from the MOZAIC program, *Atmos.*
523 *Chem. Phys.*, 8, 1-13, 2008.
- 524 Duan, J. C., Tan, J. H., Yang, L., Wu, S., and Hao, J. M.: Concentration, sources and ozone
525 formation potential of volatile organic compounds (VOCs) during ozone episode in
526 Beijing, *Atmospheric Research*, 88, 25-35, doi:10.1016/j.atmosres.2007.09.004, 2008.
- 527 Falkovich, A. H., Graber, E. R., Schkolnik, G., Rudich, Y., Maenhaut, W., and Artaxo, P.:
528 Low molecular weight organic acids in aerosol particles from Rondonia, Brazil,
529 during the biomass-burning, transition and wet periods, *Atmos. Chem. Phys.*, 5, 781-
530 797, 2005.
- 531 Fick, J., Nilsson, C., and Andersson, B.: Formation of oxidation products in a ventilation
532 system, *Atmos. Environ.*, 38, 5895-5899, doi:10.1016/j.atmosenv.2004.08.020, 2004.
- 533 Fraser, M. P., Cass, G. R., and Simoneit, B. R. T.: Air quality model evaluation data for
534 organics. 6. C-3-C-24 organic acids, *Environ. Sci. Technol.*, 37, 446-453,
535 doi:10.1021/Es0209262, 2003.
- 536 Fu, P. Q., Kawamura, K., Okuzawa, K., Aggarwal, S. G., Wang, G. H., Kanaya, Y., and
537 Wang, Z. F.: Organic molecular compositions and temporal variations of summertime
538 mountain aerosols over Mt. Tai, North China Plain, *J. Geophys. Res-Atmos.*, 113,
539 D19107, doi:10.1029/2008jd009900, 2008.
- 540 Galloway, M. M., Chhabra, P. S., Chan, A. W. H., Surratt, J. D., Flagan, R. C., Seinfeld, J. H.,
541 and Keutsch, F. N.: Glyoxal uptake on ammonium sulphate seed aerosol: reaction
542 products and reversibility of uptake under dark and irradiated conditions, *Atmos.*
543 *Chem. Phys.*, 9, 3331-3345, 2009.
- 544 Glasius, M., Lahaniati, M., Calogirou, A., Di Bella, D., Jensen, N. R., Hjorth, J., Kotzias, D.,
545 and Larsen, B. R.: Carboxylic acids in secondary aerosols from oxidation of cyclic
546 monoterpenes by ozone, *Environ. Sci. Technol.*, 34, 1001-1010,
547 doi:10.1021/Es990445r, 2000.

548 Guttikunda, S. K., Tang, Y. H., Carmichael, G. R., Kurata, G., Pan, L., Streets, D. G., Woo, J.
549 H., Thongboonchoo, N., and Fried, A.: Impacts of Asian megacity emissions on
550 regional air quality during spring 2001, *J. Geophys. Res-Atmos.*, 110, D20301,
551 doi:10.1029/2004jd004921, 2005.

552 Hallquist, M., Wenger, J. C., Baltensperger, U., Rudich, Y., Simpson, D., Claeys, M.,
553 Dommen, J., Donahue, N. M., George, C., Goldstein, A. H., Hamilton, J. F.,
554 Herrmann, H., Hoffmann, T., Iinuma, Y., Jang, M., Jenkin, M. E., Jimenez, J. L.,
555 Kiendler-Scharr, A., Maenhaut, W., McFiggans, G., Mentel, T. F., Monod, A., Prevot,
556 A. S. H., Seinfeld, J. H., Surratt, J. D., Szmigielski, R., and Wildt, J.: The formation,
557 properties and impact of secondary organic aerosol: current and emerging issues,
558 *Atmos. Chem. Phys.*, 9, 5155-5236, 2009.

559 Hatakeyama, S., Takami, A., Wang, W., and Tang, D. G.: Aerial observation of air pollutants
560 and aerosols over Bo Hai, China, *Atmos. Environ.*, 39, 5893-5898,
561 doi:<http://dx.doi.org/10.1016/j.atmosenv.2005.06.025>, 2005.

562 Hatakeyama, S., Ohno, M., Weng, J., Takagi, H., and Akimoto, H.: Mechanism for the
563 formation of gaseous and particulate products from ozone-cycloalkene reactions in air,
564 *Environ. Sci. Technol.*, 21, 52-57, doi:10.1021/es00155a005, 1987.

565 He, N., Kawamura, K., Okuzawa, K., Kanaya, Y., and Wang, Z. F.: Diurnal variations of total
566 carbon, dicarboxylic acids, ketoacids and α -dicarbonyls in aerosols in the northern
567 vicinity of Beijing, *Atmos. Chem. Phys. Discuss.*, 13, 16699-16731,
568 doi:10.5194/acpd-13-16699-2013, 2013.

569 Ho, K. F., Lee, S. C., Cao, J. J., Kawamura, K., Watanabe, T., Cheng, Y., and Chow, J. C.:
570 Dicarboxylic acids, ketocarboxylic acids and dicarbonyls in the urban roadside area of
571 Hong Kong, *Atmos. Environ.*, 40, 3030-3040, 2006.

572 Ho, K. F., Cao, J. J., Lee, S. C., Kawamura, K., Zhang, R. J., Chow, J. C., and Watson, J. G.:
573 Dicarboxylic acids, ketocarboxylic acids, and dicarbonyls in the urban atmosphere of
574 China, *J. Geophys. Res-Atmos.*, 112, D22S27, doi:10.1029/2006jd008011, 2007.

575 Ho, K. F., Lee, S. C., Ho, S. S. H., Kawamura, K., Tachibana, E., Cheng, Y., and Zhu, T.:
576 Dicarboxylic acids, ketocarboxylic acids, α -dicarbonyls, fatty acids, and benzoic acid
577 in urban aerosols collected during the 2006 Campaign of Air Quality Research in
578 Beijing (CAREBeijing-2006), *J. Geophys. Res.*, 115, D19312,
579 doi:10.1029/2009jd013304, 2010.

580 Ho, K. F., Ho, S. S. H., Lee, S. C., Kawamura, K., Zou, S. C., Cao, J. J., and Xu, H. M.:
581 Summer and winter variations of dicarboxylic acids, fatty acids and benzoic acid in
582 PM_{2.5} in Pearl Delta River Region, China, *Atmos. Chem. Phys.*, 11, 2197-2208,
583 doi:10.5194/acp-11-2197-2011, 2011.

584 Hoffmann, T., Huang, R.-J., and Kalberer, M.: Atmospheric analytical chemistry, *Anal.*
585 *Chem.*, 83, 4649-4664, 2011.

586 Huang, R.-J., Zhang, Y. L., Bozzetti, C., Ho, K. F., Cao, J. J., Han, Y. M., Dällenbach, K. R.,
587 Slowik, J. G., Platt, S. M., Canonaco, F., Zotter, P., Wolf, R., Pieber, S. M., Bruns, E.
588 A., Crippa, M., Ciarelli, G., Piazzalunga, A., Schwikowski, M., Abbazade, G.,
589 Schnelle-Kreis, J., Zimmermann, R., An, Z. S., Szidat, S., Baltensperger, U., El
590 Haddad, I., Prévôt, A. S. H.: High secondary aerosol contribution to particulate
591 pollution during haze events in China, *Nature*, 514, 218-222, 2014.

592 Huang, X. F., and Yu, J. Z.: Is vehicle exhaust a significant primary source of oxalic acid in
593 ambient aerosols?, *Geophys. Res. Lett.*, 34, L02808, doi:10.1029/2006gl028457, 2007.

594 Jaffrezo, J.-L., Aymoz, G., Delaval, C., and Cozic, J.: Seasonal variations of the water soluble
595 organic carbon mass fraction of aerosol in two valleys of the French Alps, *Atmos.*
596 *Chem. Phys.*, 5, 2809-2821, 2005.

597 Jimenez, J. L., Canagaratna, M. R., Donahue, N. M., Prevot, A. S. H., Zhang, Q., Kroll, J. H.,
598 DeCarlo, P. F., Allan, J. D., Coe, H., Ng, N. L., Aiken, A. C., Docherty, K. S., Ulbrich,
599 I. M., Grieshop, A. P., Robinson, A. L., Duplissy, J., Smith, J. D., Wilson, K. R., Lanz,
600 V. A., Hueglin, C., Sun, Y. L., Tian, J., Laaksonen, A., Raatikainen, T., Rautiainen, J.,
601 Vaattovaara, P., Ehn, M., Kulmala, M., Tomlinson, J. M., Collins, D. R., Cubison, M.
602 J., Dunlea, E. J., Huffman, J. A., Onasch, T. B., Alfarra, M. R., Williams, P. I., Bower,
603 K., Kondo, Y., Schneider, J., Drewnick, F., Borrmann, S., Weimer, S., Demerjian, K.,
604 Salcedo, D., Cottrell, L., Griffin, R., Takami, A., Miyoshi, T., Hatakeyama, S.,
605 Shimono, A., Sun, J. Y., Zhang, Y. M., Dzepina, K., Kimmel, J. R., Sueper, D., Jayne,
606 J. T., Herndon, S. C., Trimborn, A. M., Williams, L. R., Wood, E. C., Middlebrook, A.
607 M., Kolb, C. E., Baltensperger, U., and Worsnop, D. R.: Evolution of Organic
608 Aerosols in the Atmosphere, *Science*, 326, 1525-1529, doi:10.1126/science.1180353,
609 2009.

610 Kawamura, K., and Kaplan, I. R.: Motor exhaust emissions as a primary source for
611 dicarboxylic-acids in Los-Angeles ambient air, *Environ. Sci. Technol.*, 21, 105-110,
612 doi:10.1021/Es00155a014, 1987.

613 Kawamura, K., and Gagosian, R. B.: Implications of [omega]-oxocarboxylic acids in the
614 remote marine atmosphere for photo-oxidation of unsaturated fatty acids, *Nature*, 325,
615 330-332, 1987.

616 Kawamura, K., and Ikushima, K.: Seasonal-changes in the distribution of dicarboxylic-acids
617 in the urban atmosphere, *Environ. Sci. Technol.*, 27, 2227-2235,
618 doi:10.1021/Es00047a033, 1993.

619 Kawamura, K., and Sakaguchi, F.: Molecular distributions of water soluble dicarboxylic acids
620 in marine aerosols over the Pacific Ocean including tropics, *J. Geophys. Res-Atmos.*,
621 104, 3501-3509, doi:10.1029/1998jd100041, 1999.

622 Kawamura, K., and Yasui, O.: Diurnal changes in the distribution of dicarboxylic acids,
623 ketocarboxylic acids and dicarbonyls in the urban Tokyo atmosphere, *Atmos. Environ.*,
624 39, 1945-1960, doi:10.1016/j.atmosenv.2004.12.014, 2005.

625 Kawamura, K., Ng, L. L., and Kaplan, I. R.: Determination of organic-acids (C1-C10) in the
626 atmosphere, motor exhausts, and engine oils, *Environ. Sci. Technol.*, 19, 1082-1086,
627 doi:10.1021/Es00141a010, 1985.

628 Kawamura, K., Kasukabe, H., and Barrie, L. A.: Source and reaction pathways of
629 dicarboxylic acids, ketoacids and dicarbonyls in arctic aerosols: One year of
630 observations, *Atmos. Environ.*, 30, 1709-1722, doi:10.1016/1352-2310(95)00395-9,
631 1996.

632 Kawamura, K., Kobayashi, M., Tsubonuma, N., Mochida, M., Watanabe, T., and Lee, M.:
633 Organic and inorganic compositions of marine aerosols from East Asia: Seasonal
634 variations of water-soluble dicarboxylic acids, major ions, total carbon and nitrogen,
635 and stable C and N isotopic composition, *Geo Soc S P*, 9, 243-265,
636 doi:[http://dx.doi.org/10.1016/S1873-9881\(04\)80019-1](http://dx.doi.org/10.1016/S1873-9881(04)80019-1), 2004.

637 Kroll, J. H., Ng, N. L., Murphy, S. M., Varutbangkul, V., Flagan, R. C., and Seinfeld, J. H.:
638 Chamber studies of secondary organic aerosol growth by reactive uptake of simple
639 carbonyl compounds, *J. Geophys. Res-Atmos.*, 110, D23207,
640 doi:10.1029/2005jd006004, 2005.

641 Kundu, S., Kawamura, K., Andreae, T. W., Hoffer, A., and Andreae, M. O.: Molecular
642 distributions of dicarboxylic acids, ketocarboxylic acids and alpha-dicarbonyls in
643 biomass burning aerosols: implications for photochemical production and degradation
644 in smoke layers, *Atmos. Chem. Phys.*, 10, 2209-2225, 2010.

645 Legrand, M., Preunkert, S., Oliveira, T., Pio, C. A., Hammer, S., Gelencser, A., Kasper-Giebl,
646 A., and Laj, P.: Origin of C-2-C-5 dicarboxylic acids in the European atmosphere

647 inferred from year-round aerosol study conducted at a west-
648 10.1029/2006jd008019east transect, *J. Geophys. Res-Atmos.*, 112, D23S07,
649 doi:10.1029/2006jd008019, 2007.

650 Liu, Y. N., Tao, S., Yang, Y. F., Dou, H., Yang, Y., and Coveney, R. M.: Inhalation exposure
651 of traffic police officers to polycyclic aromatic hydrocarbons (PAHs) during the
652 winter in Beijing, China, *Sci. Total Environ.*, 383, 98-105,
653 doi:10.1016/j.scitotenv.2007.05.008, 2007.

654 Luo, C., John, J. C. S., Zhou, X. J., Lam, K. S., Wang, T., and Chameides, W. L.: A nonurban
655 ozone air pollution episode over eastern China: Observations and model simulations, *J.*
656 *Geophys. Res-Atmos.*, 105, 1889-1908, doi:10.1029/1999jd900970, 2000.

657 Mauzerall, D. L., Narita, D., Akimoto, H., Horowitz, L., Walters, S., Hauglustaine, D. A., and
658 Brasseur, G.: Seasonal characteristics of tropospheric ozone production and mixing
659 ratios over East Asia: A global three-dimensional chemical transport model analysis, *J.*
660 *Geophys. Res-Atmos.*, 105, 17895-17910, doi:10.1029/2000jd900087, 2000.

661 Mkomu, S. L., and Kawamura, K.: Molecular composition of dicarboxylic acids,
662 ketocarboxylic acids, alpha-dicarbonyls and fatty acids in atmospheric aerosols from
663 Tanzania, East Africa during wet and dry seasons, *Atmos. Chem. Phys.*, 13, 2235-
664 2251, doi:10.5194/acp-13-2235-2013, 2013.

665 Mochida, M., Umemoto, N., Kawamura, K., Lim, H. J., and Turpin, B. J.: Bimodal size
666 distributions of various organic acids and fatty acids in the marine atmosphere:
667 Influence of anthropogenic aerosols, Asian dusts, and sea spray off the coast of East
668 Asia, *J. Geophys. Res-Atmos.*, 112, D15209, doi:10.1029/2006jd007773, 2007.

669 Narukawa, M., Kawamura, K., Takeuchi, N., and Nakajima, T.: Distribution of dicarboxylic
670 acids and carbon isotopic compositions in aerosols from 1997 Indonesian forest fires,
671 *Geophys. Res. Lett.*, 26, 3101-3104, doi:10.1029/1999gl010810, 1999.

672 Oliveira, C., Pio, C., Alves, C., Evtugina, M., Santos, P., Goncalves, V., Nunes, T., Silvestre,
673 A. J. D., Palmgren, F., Wahlin, P., and Harrad, S.: Seasonal distribution of polar
674 organic compounds in the urban atmosphere of two large cities from the North and
675 South of Europe, *Atmos. Environ.*, 41, 5555-5570,
676 doi:10.1016/j.atmosenv.2007.03.001, 2007.

677 Poschl, U.: Atmospheric aerosols: Composition, transformation, climate and health effects,
678 *Angew Chem Int Edit*, 44, 7520-7540, doi:10.1002/anie.200501122, 2005.

679 Rogge, W. F., Medeiros, P. M., and Simoneit, B. R. T.: Organic marker compounds for
680 surface soil and fugitive dust from open lot dairies and cattle feedlots, *Atmos.*
681 *Environ.*, 40, 27-49, doi:10.1016/j.atmosenv.2005.07.076, 2006.

682 Rogge, W. F., Hildemann, L. M., Mazurek, M. A., Cass, G. R., and Simoneit, B. R. T.:
683 Sources of fine organic aerosol .1. Charbroilers and meat cooking operations, *Environ.*
684 *Sci. Technol.*, 25, 1112-1125, doi:10.1021/Es00018a015, 1991.

685 Rogge, W. F., Hildemann, L. M., Mazurek, M. A., Cass, G. R., and Simoneit, B. R. T.:
686 Sources of Fine Organic Aerosol .2. Noncatalyst and Catalyst-Equipped Automobiles
687 and Heavy-Duty Diesel Trucks, *Environ. Sci. Technol.*, 27, 636-651, doi:Doi
688 10.1021/Es00041a007, 1993.

689 Rudich, Y., Donahue, N. M., and Mentel, T. F.: Aging of organic aerosol: Bridging the gap
690 between laboratory and field studies, *Annu. Rev. Phys. Chem.*, 58, 321-352,
691 doi:10.1146/annurev.physchem.58.032806.104432, 2007.

692 Saxena, P., and Hildemann, L. M.: Water-soluble organics in atmospheric particles: A critical
693 review of the literature and application of thermodynamics to identify candidate
694 compounds, *J. Atmos. Chem.*, 24, 57-109, doi:10.1007/Bf00053823, 1996.

695 Schauer, J. J., Kleeman, M. J., Cass, G. R., and Simoneit, B. R. T.: Measurement of emissions
696 from air pollution sources. 1. C1 through C29 organic compounds from meat
697 charbroiling, *Environ. Sci. Technol.*, 33, 1566-1577, doi:10.1021/Es980076j, 1999.

698 Schauer, J. J., Kleeman, M. J., Cass, G. R., and Simoneit, B. R. T.: Measurement of emissions
699 from air pollution sources. 4. C1-C27 organic compounds from cooking with seed oils,
700 *Environ. Sci. Technol.*, 36, 567-575, doi:10.1021/Es002053m, 2002.

701 Simoneit, B. R. T.: Organic-matter of the troposphere .3. Characterization and sources of
702 petroleum and pyrogenic residues in aerosols over the western United-States, *Atmos.*
703 *Environ.*, 18, 51-67, doi:10.1016/0004-6981(84)90228-2, 1984.

704 Simoneit, B. R. T., and Mazurek, M. A.: Organic-matter of the troposphere .2. Natural
705 background of biogenic lipid matter in aerosols over the rural western United-States,
706 *Atmos. Environ.*, 16, 2139-2159, doi:10.1016/0004-6981(82)90284-0, 1982.

707 Stephanou, E. G., and Stratigakis, N.: Oxocarboxylic and alpha,omega-dicarboxylic acids -
708 Photooxidation products of biogenic unsaturated fatty-acids present in urban aerosols,
709 *Environ. Sci. Technol.*, 27, 1403-1407, doi:10.1021/Es00044a016, 1993.

710 Suh, I., Zhang, R., Molina, L. T., and Molina, M. J.: Oxidation mechanism of aromatic peroxy
711 and bicyclic radicals from OH-toluene reactions, *J. Am. Chem. Soc.*, 125, 12655-
712 12665, doi:10.1021/ja0350280, 2003.

713 Tao, S., Wang, Y., Wu, S. M., Liu, S. Z., Dou, H., Liu, Y. N., Lang, C., Hu, F., and Xing, B.
714 S.: Vertical distribution of polycyclic aromatic hydrocarbons in atmospheric boundary
715 layer of Beijing in winter, *Atmos. Environ.*, 41, 9594-9602,
716 doi:10.1016/j.atmosenv.2007.08.026, 2007.

717 Volkamer, R., Platt, U., and Wirtz, K.: Primary and secondary glyoxal formation from
718 aromatics: Experimental evidence for the bicycloalkyl-radical pathway from benzene,
719 toluene, and p-xylene, *J. Phys. Chem. A*, 105, 7865-7874, doi:10.1021/Jp010152w,
720 2001.

721 Wang, G. H., and Kawamura, K.: Molecular characteristics of urban organic aerosols from
722 Nanjing: A case study of a mega-city in China, *Environ. Sci. Technol.*, 39, 7430-7438,
723 doi:10.1021/Es051055+, 2005.

724 Wang, G. H., Kawamura, K., Lee, S., Ho, K., and Cao, J.: Molecular, seasonal, and spatial
725 distributions of organic aerosols from fourteen Chinese cities, *Environ. Sci. Technol.*,
726 40, 4619-4625, 2006.

727 Warneck, P.: Multi-phase chemistry of C2 and C3 organic compounds in the marine
728 atmosphere, *J. Atmos. Chem.*, 51, 119-159, doi:10.1007/s10874-005-5984-7, 2005.

729 Xu, J., Zhang, Y. H., Fu, J. S., Zheng, S. Q., and Wang, W.: Process analysis of typical
730 summertime ozone episodes over the Beijing area, *Sci. Total Environ.*, 399, 147-157,
731 doi:10.1016/j.scitotenv.2008.02.013, 2008.

732 Yang, H., Yu, J. Z., Ho, S. S. H., Xu, J. H., Wu, W. S., Wan, C. H., Wang, X. D., Wang, X.
733 R., and Wang, L. S.: The chemical composition of inorganic and carbonaceous
734 materials in PM_{2.5} in Nanjing, China, *Atmos. Environ.*, 39, 3735-3749,
735 doi:<http://dx.doi.org/10.1016/j.atmosenv.2005.03.010>, 2005.

736 Yokouchi, Y., and Ambe, Y.: Characterization of polar organics in airborne particulate matter,
737 *Atmospheric Environment (1967)*, 20, 1727-1734, doi:[http://dx.doi.org/10.1016/0004-
738 6981\(86\)90121-6](http://dx.doi.org/10.1016/0004-6981(86)90121-6), 1986.

739 Zhang, R. J., Shen, Z. X., Zhang, L. M., Zhang, M. G., Wang, X., and Zhang, K.: Elemental
740 Composition of Atmospheric Particles during Periods with and without Traffic
741 Restriction in Beijing: The Effectiveness of Traffic Restriction Measure, *Sola*, 7, 061-
742 064, doi:10.2151/sola.2011-016, 2011.

- 743 Zhao, Y. L., Hu, M., Slanina, S., and Zhang, Y. H.: Chemical compositions of fine particulate
744 organic matter emitted from Chinese cooking, *Environ. Sci. Technol.*, 41, 99-105,
745 doi:10.1021/Es0614518, 2007a.
- 746 Zhao, Y. L., Hu, M., Slanina, S., and Zhang, Y. H.: The molecular distribution of fine
747 particulate organic matter emitted from Western-style fast food cooking, *Atmos.*
748 *Environ.*, 41, 8163-8171, doi:10.1016/j.atmosenv.2007.06.029, 2007b.
- 749

750 **Table 1.** Concentrations of dicarboxylic acids, ketocarboxylic acids and α -dicarbonyls, fatty acids and
 751 benzoic acid in PM_{2.5} samples during CAREBeijing 2007.

Compounds (ng m ⁻³)	PKU (n=10)			Yufa (n=10)		
	Range	Average	S.D.	Range	Average	S.D.
Dicarboxylic acids						
Oxalic, C2	212-586	435	124	226-632	418	130
Malonic, C3	30.0-73.5	54.9	14.0	17.1-68.6	43.5	15.0
Succinic, C4	52.8-147	89.9	27.7	44.8-129	80.9	26.9
Glutaric, C5	13.7-59.2	36.0	14.4	16.0-168	41.3	46.2
Adipic, C6	15.1-35.1	26.7	6.03	10.8-73.1	24.5	18.1
Pimelic, C7	MDLs-6.44	2.79	2.56	MDLs-3.38	0.34	1.07
Suberic, C8	MDLs	MDLs		MDLs	MDLs	
Azelaic, C9	58.8-85.8	71.4	8.91	37.5-64.6	49.2	8.99
Sebacic, C10	MDLs-3.91	0.69	1.47	MDLs-2.78	0.28	0.88
Undecanedioic, C11	MDLs	MDLs		MDLs	MDLs	
Dodecanedioic, C12	MDLs	MDLs		MDLs	MDLs	
Methylmalonic, iC4	MDLs	MDLs		MDLs	MDLs	
Methylsuccinic, iC5	MDLs-10.6	3.82	2.75	MDLs-5.21	3.05	2.22
2-Methylglutaric, iC6	MDLs-10.5	6.09	2.79	MDLs-8.92	4.80	2.40
Maleic, M	9.18-20.5	15.6	3.39	9.69-17.2	13.5	2.03
Fumaric, F	MDLs	MDLs		MDLs	MDLs	
Methylmaleic, mM	7.52-12.2	9.60	1.40	5.75-8.83	7.21	0.92
Phthalic, Ph	171-250	209	28.9	80.6-415	176	91.5
Isophthalic, iPh	MDLs-15.1	7.27	5.76	MDLs-12.8	4.65	5.22
Terephthalic, tPh	12.2-51.8	30.4	10.1	6.83-53.2	30.9	13.8
Malic, hC4	MDLs-3.21	0.32	1.02	MDLs	MDLs	
Ketomalonic, kC3	2.79-11.8	6.56	2.79	MDLs-10.4	6.67	3.07
4-Ketopimelic, kC7	MDLs-13.0	3.71	5.06	MDLs-11.0	3.21	4.33
Total diacids	599-1287	1010	220	473-1429	909	278
Ketocarboxylic acids						
Pyruvic	17.9-70.2	30.3	15.9	12.7-27.3	20.1	4.18
Glyoxylic, ω C2	49.2-105	72.9	16.1	32.8-81.7	59.7	14.5
3-Oxopropanoic, ω C3	MDLs-3.29	1.53	1.23	MDLs-1.31	0.59	0.62
4-Oxobutanoic, ω C4	5.91-25.0	15.8	5.55	6.59-23.3	15.0	4.83
9-Oxononanoic, ω C9	MDLs-7.28	1.70	2.50	MDLs-5.03	1.61	2.19
Total ketoacids	87.4-169	122	28.8	52.0-131	97.0	22.9
α -Dicarbonyls						
Glyoxal, Gly	1.40-21.4	13.1	5.72	2.93-17.3	11.5	4.04
Methylglyoxal, MeGly	23.3-81.3	38.7	16.2	21.3-51.3	32.7	8.76
Total dicarbonyls	35.5-99.5	51.8	17.9	29.0-61.4	44.2	10.3
Sum of bifunctional species	730-1455	1184	241	554-1621	1050	303
Fatty acids						
Tridecanoic acid, C _{13:0}	5.08-16.4	10.8	3.92	MDLs-11.8	7.56	3.31
Tetradecanoic acid, C _{14:0}	54.5-97.9	68.7	13.6	36.5-74.0	50.0	13.6
Hexadecanoic acid, C _{16:0}	199-393	249	57.0	149-310	201	48.9
Heptadecanoic acid, C _{17:0}	MDLs-13.3	4.32	4.53	MDLs-12.8	3.90	5.06
Octadecanoic acid, C _{18:0}	134-462	219	94.0	122-237	170	36.6
Octadecenoic acid, C _{18:1}	2.91-33.0	24.3	8.93	13.0-47.9	24.6	9.23
Eicosanoic acid, C _{20:0}	MDLs-7.84	4.01	3.51	MDLs-7.71	3.28	3.50
Docosanoic acid, C _{22:0}	5.69-13.6	9.24	2.43	MDLs-15.4	7.01	4.95
Tetracosanoic acid, C _{24:0}	MDLs-10.5	6.51	3.59	MDLs-12.1	7.11	3.09
Sum of fatty acids	459-1003	597	159	375-684	475	114
Benzoic acid	933-2754	1496	511	724-1930	1278	372
PM _{2.5} (μ g m ⁻³)	64.9-191	103	35.1	38.6-160	98.3	34.1
OC (μ g m ⁻³)	12.1-19.3	14.9	2.47	4.63-17.0	11.1	3.68
EC (μ g m ⁻³)	4.48-11.0	6.21	1.90	2.96-8.62	5.57	1.83
WSOC (μ g m ⁻³)	2.69-8.20	5.59	1.49	1.72-7.16	4.55	1.79

MDL: Method Detection Limit

752
 753
 754
 755

756 **Figure Captions**

757

758 **Figure 1.** Temporal variations of EC, OC, WSOC and various organic compound classes at
759 PKU and Yufa during CAREBeijing-2007 [[Note the difference in start time at PKU and Yufa](#)
760 [\(i.e., the first three days\)](#)].

761

762 **Figure 2.** 3-day air mass back trajectories on a) 05 August b) 13 August and c) 17 and 19
763 August.

764

765 **Figure 3.** Pollution events versus [less polluted clean](#)-air at PKU (a and b) and Yufa (c and d),
766 showing the variation of particulate pollutants.

767

768 **Figure 4.** Positive correlation between α -dicarbonyls and benzoic acid observed at PKU (a)
769 and Yufa (b).

770

771 **Figure 5.** The R/N ratio of particulate compounds observed at PKU (a) and Yufa (b)

772