Supplementary material for

## One-year Observations of Size Distribution Characteristics of Major Aerosol Constituents at a Coastal Receptor Site in Hong Kong: I. Inorganic Ions and Oxalate

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This supplementary document contains two tables and eight figures.

	1996-1997 <sup>1</sup>	200	08 <sup>2</sup>	2011-2012 (this study)	2011-2012 (this study)			
	Winter	Summer	Winter	Summer Winter				
SO4 <sup>2-</sup>	0.20	0.27±1.30	0.27±1.30	0.25±1.34 0.26±1.25				
	0.58	$0.80{\pm}1.46$	$0.95 \pm 1.53$	0.83±1.54 0.84±1.50				
	4.20	$3.34 \pm 2.27$	$4.17 \pm 2.07$	5.13±2.06 5.03±2.16				
$\mathrm{NH_4}^+$	0.21	$0.26 \pm 1.36$	$0.24{\pm}1.37$	0.26±1.35 0.26±1.26				
	0.56	$0.78 \pm 1.54$	$0.94{\pm}1.69$	0.82±1.53 0.84±1.50				
	5.70	$2.14{\pm}1.50$	$4.69 \pm 2.07$	5.02±2.35 5.06±2.17				
NO <sub>3</sub> <sup>-</sup>	0.14	$0.15 \pm 1.39$	$0.24{\pm}1.40$					
	0.46	$0.78 \pm 1.75$	$0.85 \pm 1.46$	0.77±2.18 0.88±1.44				
	3.95	$3.37 \pm 1.50$	$2.50{\pm}1.52$	3.40±1.49 2.96±1.53				
		$7.16 \pm 1.41$	$6.94{\pm}1.51$	7.19±1.42 6.83±1.44				
$C \cap 2^{-}$		0.25 + 1.65	0.24+1.27	0.12+1.51 0.20+1.28				
$C_2O_4$		$0.23 \pm 1.03$	$0.24 \pm 1.57$	$0.12\pm1.51$ $0.29\pm1.28$				
		$0.88 \pm 1.70$	0.90±1.64	$0.92\pm2.02$ $0.77\pm1.49$				
		4.35±2.04	4.22±2.05	5.25±1.64 4.58±2.12				

Table S1 Mass median aerodynamic diameter of distribution modes at UST (unit:  $\mu$ m)

<sup>1</sup>Zhuang et al., 1999b; <sup>2</sup> Yu et al, 2010.

Table S2. Comparisons of PMF-modeled and measured concentrations of species used in the PMF analysis of the coarse-mode species data.

	a a 2-	110 -	<b>NAXX</b> +	<b>x</b> z+	NT +	2+	017	a 2+	
	$SO_4^-$	$NO_3$	$NH_4$	K	Na	Mg	Cl	Ca	$PM_{2.5} S1^{-1}$
$\mathbb{R}^2$	0.79	0.99	0.97	0.79	0.99	0.97	0.99	0.83	0.90
Slope	0.73	0.94	1.03	1.28	1.01	0.98	0.96	0.98	0.67
Intercept	0.19	0.15	-0.015	-0.039	0.014	0.005	0.068	-0.013	0.12

<sup>1</sup> PM<sub>2.5</sub> Si data are used as tracers for bulk dust particles, as coarse mode Si data is not available.



Figure S1. Sampling location in this study



Figure S2 Hourly temperature (°C), RH (%) and wind vector during the sampling days.



**Figure S3** The Twomey algorithm results in skewed log-normal distribution curves of  $Ca^{2+}$  (red curves) when assuming one fine mode and one coarse mode. The measured data are shown as histograms.



**Figure S4** Time series of  $[Na^+]+2[Ca^{2+}]$  abundance in coarse particles (3.2–18 µm), modified NH<sub>4</sub>NO<sub>3</sub> dissociation equilibrium constant (K<sub>e</sub>'), and percentage of nitrate in the fine mode (P<sub>n</sub>) (<1.8 µm). K<sub>e</sub>' is dependent on temperature, relative humidity, and ion strength in aerosol liquid phase.



**Figure S5** Correlations between gaseous HCl and HNO<sub>3</sub> in the hours during collection samples of low  $P_{n_{fine}}$  and high  $P_{n_{fine}}$ .  $P_{n_{fine}}$  is the percentage of nitrate in the fine mode (<30%).



**Figure S6** Time series of gas concentration level for gas species including NO,  $NO_2$ , CO,  $O_3$ , and SO<sub>2</sub> from March 2011 to February 2012. Date is in the format of YYYYMMDD.



**Figure S7** Robust Q as a function of Fpeak value from 0 to 0.5 in the PMF analysis of the size-segregated sulfate data. The dot line represents the theoretical value as a reference.



**Figure S8** Comparison between seawater composition and PMF-modeled factors of fresh sea salt and aged sea salt. All the species are normalized against Na<sup>+</sup>. The source profile of seawater with salinity of 35‰ is taken from Millero and Sohn (1992).