

1           **Characteristics of Trace Metals in Traffic-Derived Particles in**  
2           **Hsuehshan Tunnel, Taiwan: Size Distribution, Potential Source, and**  
3                           **Fingerprinting Metal Ratio**

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22

1 **Abstract**

2 Traffic emissions are a significant source of airborne particulate matter (PM) in  
3 ambient environments. These emissions contain high abundance of toxic metals and  
4 thus pose adverse effects on human health. Size-fractionated aerosol samples were  
5 collected from May to September 2013 by using micro-orifice uniform deposited  
6 impactors (MOUDI). Sample collection was conducted simultaneously at the inlet  
7 and outlet sites of Hsuehshan Tunnel in northern Taiwan, which is the second  
8 longest freeway tunnel (12.9 km) in Asia. Such endeavor aims to characterize the  
9 chemical constituents and size distributions, as well as fingerprinting ratios of  
10 particulate metals emitted by vehicle fleets. A total of 36 metals in size-resolved  
11 aerosols were determined through inductively coupled plasma mass spectrometry.  
12 Three major groups, namely, tailpipe emissions (Zn, Pb, and V in fine mode), wear  
13 debris (Cu, Cd, Fe, Ga, Mn, Mo, Sb, and Sn), and resuspended dust (Ca, Mg, K, and  
14 Rb), of airborne PM metals were categorized on the basis of the results of  
15 enrichment factor, correlation matrix, and principal component analysis. Size  
16 distributions of wear-originated metals resembled the pattern of crustal elements,  
17 which were predominated by super-micron particulates (PM<sub>1-10</sub>). By contrast,  
18 tailpipe exhaust elements such as Zn, Pb, and V were distributed mainly in  
19 submicron particles. By employing Cu as a tracer of wear abrasion, several  
20 inter-metal ratios, including Fe/Cu (14), Ba/Cu (1.05), Sb/Cu (0.16), Sn/Cu (0.10),  
21 and Ga/Cu (0.03), served as fingerprints for wear debris. However, the data set  
22 collected in this work is useful for further studies on traffic emission inventory and  
23 human health effects of traffic-related PM.

24

25 **Keywords:** Traffic-related Metal, Size-fractionated Aerosol, Fingerprinting Metal

**1. Introduction**

Traffic emissions are an important source of particulate matter (PM) (Sternbeck et al., 2002; Birmili et al., 2006; Lough et al., 2005; Johansson et al., 2009) in urban atmosphere. Exposure to traffic-derived PM poses adverse effects on human health and increases the risk of respiratory illness, cardiovascular diseases, and asthma (Brauer et al., 2002; Defino et al., 2005), resulting in increased mortality (Nel, 2005). Airborne traffic-related PM is emitted mainly by tailpipe exhaust from gasoline and diesel engines (exhaust emissions), wear from brake linings and tires, as well as re-suspension of road dust (non-exhaust emissions) by moving vehicles (Rogge et al., 1993; Cadle et al., 1999; Garg et al., 2000; Wåhlin et al., 2006; Lawrence et al., 2013). Exhaust emissions contribute a large amount of fine particulate matter (aerodynamic diameter less than 2.5  $\mu\text{m}$ ,  $\text{PM}_{2.5}$ ), whereas non-exhaust emissions mainly consist of larger particles (Abu-Allaban et al., 2002; Sanders et al., 2003). With regard to elemental compositions, Pb, Zn, Ni, and V in submicron particles were commonly attributed to pipe emissions and fuel oil combustion of both gasoline and diesel engines as shown in Table S1 (Lin et al., 2005; Wang et al., 2003; Shafer et al., 2012). Silicon (Si), Fe, Ca, Na, Mg, Al, and K are essentially found in larger particles and are associated with re-suspension of road dust. Large amounts of Ca and K observed in submicron particles occasionally originate from the tailpipe emission of lubricating oil as well as the vaporization of volatile K-compounds and potassium titanate ( $\text{K}_2\text{O}\cdot\text{nTiO}_2$ ), which is used for improving heat resistance and wear characteristics (Hee and Filip, 2005; Iijima et al., 2007; Kuo et al., 2009). Meanwhile, Cu, Ba, Sb, Fe, Cd, Cr, Ga, Sn, and Zn, which are commonly associated with wear dust from brake

1 linings and tires, are predominant in coarse PM (Lough et al., 2005; Grieshop et al.,  
2 2006; Thorpe and Harrison, 2008).

3 A number of studies investigated the chemical and physical properties of  
4 traffic-originated PM by performing conventional dynamometric tests and field  
5 measurements near roads and inside tunnels (Sternbeck et al., 2002; Sanders et al.,  
6 2003; Birmili et al., 2006; Wählín et al., 2006; Iijima et al., 2007; Ning et al., 2007;  
7 Harrison et al., 2012; Dall'Osto et al., 2013; Lawrence et al., 2013). Dynamometric  
8 tests may allow optimal control of experimental conditions; however, the limitations  
9 of such tests are the costs and inadequate representative of real-world traffic  
10 emissions on the roads (Jamriska et al., 2004). A field measurement nearby roadside  
11 is another method to well characterize the traffic-derived PM (Ning et al., 2004), but it  
12 may be influenced by local meteorological conditions and traffic activities (Jamriska  
13 et al., 2004; Ntziachristos et al., 2007). Accordingly, tunnel study may be an  
14 alternative way to address this issue.

15 Tunnel aerosol sampling is designed to explore size distributions, chemical  
16 compositions, and emission factors of traffic-related aerosols and their associated  
17 compositions (Weingartner et al., 1997; Funasaka et al., 1998; Gillies et al., 2001;  
18 Sternbeck et al., 2002; Grieshop et al., 2006; Chiang and Huang, 2009; Pio et al.,  
19 2013). Pio et al. (2013) discriminated three main types of aerosols in Marquês tunnel,  
20 Portugal, namely, carbonaceous, soil component, and vehicle mechanical wear. They  
21 also suggested that Cu is a good tracer for wear emissions of road traffic. Wear  
22 emission elements such as Zn, Sb, and Ba exhibited a peak mode in the size range of  
23 3.2  $\mu\text{m}$  to 5.6  $\mu\text{m}$ . In comparison, Pb, Ca, and Fe partitioned within 0.1  $\mu\text{m}$  are mostly  
24 emitted from combustion of fuel and lubricant oil or vaporization from hot brake  
25 surface (Lough et al., 2005). Sternbeck et al. (2002) collected aerosol samples in two

1 tunnels in Sweden and analyzed trace metals through inductively coupled plasma  
2 mass spectrometry (ICP-MS). They concluded that vehicle-related metals, such as Cu,  
3 Zn, Cd, Sb, Ba, and Pb, originated mainly from wear rather than from combustion,  
4 and that heavy-duty vehicles (HDV), rather than light-duty vehicles (LDV), are the  
5 leading emitter of Ba and Sb. They further suggested that a Sb/Cu ratio of  $\sim 0.22$   
6 indicates the presence of brake wear-related particles.

7 In this work, a series of aerosol sampling was conducted at two sites in  
8 Hsuehshan Tunnel by using micro-orifice uniform deposited impactors (MOUDI) to  
9 characterize the physical and chemical properties of metallic aerosols under real  
10 driving conditions. In the past several intensive measurements of aerosols have been  
11 carried out inside Hsuehshan Tunnel (Chang et al., 2009; Chen et al., 2010; Cheng et  
12 al., 2010a; Zhu et al., 2010); for instance, Zhu et al. (2010) characterized different  
13 temperature carbonaceous aerosols in fine PM and then identified their sources by  
14 positive matrix factorization (PMF) approach. Moreover, number concentrations of  
15 ultrafine particle (UFP) measured by Cheng et al. (2010) indicate that UFP, on  
16 average, were about  $1.0 \times 10^5 - 3.0 \times 10^5$  particles/cm<sup>3</sup> while higher UFP numbers  
17 were found at a traffic jam. They also suggested that gas-to-particle conversion is a  
18 crucial way to produce nucleation PM at entrance of the tunnel, and coagulation  
19 growth of nucleation particles is an important mechanism for forming Aitken mode  
20 PM at the middle and exit section. Besides, gaseous pollutants, including VOC, O<sub>3</sub>,  
21 CO and NO<sub>x</sub>, inside this tunnel have been also studied previously (Chang et al., 2009;  
22 Li et al., 2011, Lai and Peng, 2012). Thus, Hsuehshan Tunnel is a suitable study area  
23 for characterizing the behaviors of air pollutants associated with vehicle fleets. During  
24 the experimental campaigns, a total of 24 sets of size-resolved aerosol samples were  
25 collected; 36 target metals were analyzed by ICP-MS. Elemental compositions, size

1 distributions, and fingerprinting metal ratios in traffic aerosols are reported in this  
2 paper. The resulting comprehensive dataset would provide useful insight into health  
3 effect studies, source apportionment of atmospheric metals, and emissions inventory  
4 of traffic-related particulate metals.

5

6 **2. Methodology**

7 **2.1 Site description**

8 With a length of 12.9 km, Hsuehshan Tunnel is the second longest road tunnel  
9 in Asia and the fifth longest in the world. Opened to traffic on June 2006, Hsuehshan  
10 Tunnel connects Pingling in New Taipei City and Toucheng in Yilan County. The  
11 tunnel has two separate two-lane bores and ascends steadily from 44 m a.m.s.l.  
12 (meters above mean sea level) at the south end (Toucheng) to 208 m a.m.s.l. at the  
13 north end (Pingling), that is, a slope of 1.26 %. Only passenger cars and light-duty  
14 trucks (which are both classified under LDV) as well as shuttle buses (categorized  
15 under HDV) are allowed to travel inside the tunnel with vehicle speed limited to 90  
16 km/h. Four aerosol sampling campaigns were conducted in the northbound bore  
17 between May and September 2013; each campaign lasted for three days: Friday to  
18 Sunday. During the sampling period, the traffic volume passing through Hsuehshan  
19 Tunnel at the northbound, in general, approximated 1800 vehicles per hour on the  
20 weekend, which was 1.3 times higher than the workdays (see in Table 1). However,  
21 the traffic flow increased to 2300 vehicles/h from Sunday afternoon to evening when  
22 people traveled back to Taipei, as a result, a traffic jam always occurred inside the  
23 tunnel since Sunday afternoon, probably influencing traffic-related PM metal  
24 emissions.

25 A ventilation system composed of three air exchange stations and three air

1 interchange stations was built inside the tunnel to maintain air quality. Exchange and  
2 interchange stations are located alternatively at intervals of nearly 2 km. In exchange  
3 stations, polluted air is exchanged with outer fresh air by using separate fresh and  
4 exhaust shafts equipped with two sets of fans. Fans are typically triggered at  
5 temperatures higher than 40 °C or CO concentrations higher than 75 ppm. In  
6 interchange stations, the air in each bore is diverted into another bore by two sets of  
7 fans, which are also triggered when CO concentration exceeds 75 ppm. During the  
8 sampling periods, the ventilation system operated regularly, particular in July and  
9 August campaigns when the air temperature near the outlet site is frequently more  
10 than 40 °C.

11

## 12 **2.2 Sampling and analysis**

13 During the sampling campaigns, two aerosol samplers were installed at 1.7 and  
14 10.6 km from the entrance. The intakes of both aerosol instruments were placed 1.6 m  
15 above the pavement. MOUDIs (model 100, MSP Corporation, Minneapolis,  
16 Minnesota) equipped with pre-weighed Teflon filters (PTFE, 47 mm in diameter and  
17 1.0 mm in pore size, Pall Gelman, East Hills, New York) were used to collect  
18 size-resolved aerosol samples. MOUDI consists of 10 size-fractionating stages with  
19 50 % cut-off diameters of 10, 5.6, 3.2, 1.8, 1.0, 0.56, 0.32, 0.18, 0.10, and 0.056  $\mu\text{m}$ ,  
20 plus an inlet (nominal cut size of 18  $\mu\text{m}$ ) and an after-filter ( $< 0.018 \mu\text{m}$ ) at the base.  
21 Flow rate was calibrated prior to each sampling run and maintained at 30 L/min. Each  
22 sample was collected for 12 h (typically from 9 a.m. to 9 p.m.) daily.

23 After sampling, filter samples were conditioned for 48 h, followed by gravimetric  
24 measurement at 23 °C and RH  $30 \pm 5\%$  with a microbalance (METTLER TOLEDO,  
25 MX5, AX205, precision 1  $\mu\text{g}$ ) to determine the net mass of collected aerosol particles,

1 which is needed to calculate the PM mass concentration. The samples were then  
2 subjected to acid digestion with the use of an ultra-high throughput microwave  
3 digestion system (MARSPress, CEM Corporation, Matthews, NC). The vessels  
4 were acid-cleaned thoroughly prior to sample digestion. A half of each sample filter  
5 was digested in an acid mixture (1.5 ml 60% HNO<sub>3</sub> and 1.5 ml 48% HF). After  
6 digestion, the vessels were transferred to the XpressVap<sup>TM</sup> accessory sets (CEM) for  
7 evaporation of the remaining acids. When nearly dried, 2 mL concentrated HNO<sub>3</sub> was  
8 added into each vessel and reheated. The resulting solution was then diluted with  
9 Milli-Q water to a final volume of 15 mL for analysis. The digestion procedure has  
10 been detailed in previous studies (Hsu et al., 2008; 2009; Zhang et al., 2013).

11 A total of 36 target elements in aerosols were analyzed by ICP-MS (Elan 6100,  
12 Perkin Elmer<sup>TM</sup> SCIEX, USA). For each run, a blank reagent and three filter  
13 membrane blanks were subjected to the same procedure as that for the samples.  
14 Indium (In) was added to the digests as an internal standard with a final concentration  
15 of 10 ng/mL for ICP-MS analysis. The QA/QC of data is guaranteed by the analysis  
16 of a standard reference material, SRM 1648 (urban atmospheric particulate matter  
17 prepared by the National Institute Standards and Technology (NIST)). The recoveries  
18 of target elements mostly fell within 10 % (n = 7) of certified or reference values  
19 (Table S2). The method detection limits (MDLs) for the analyzed elements are also  
20 presented. Details of the ICP-MS analysis has been extensively discussed by Hsu et al.  
21 (2010) and Zhang et al (2013).

22

### 23 **2.3 Enrichment factor and principal component analysis**

24 In addition to size distribution, three approaches, namely, enrichment factor (EF),  
25 correlation matrix, and principal component analysis (PCA) were applied to explore



1 the possible sources and associations of elements. EF is used to assess the influence of  
2 crustal source on a given metal ( $X_i$ ), which can be calculated by using the following  
3 equation:

$$4 \quad EF(X_i) = \frac{(X_i / Al)_{PM}}{(X_i / Al)_{Crust}} \quad (1)$$

6 where  $(X_i/Al)_{PM}$  is the concentration ratio of a given element X to Al in tunnel  
7 particulate matters and  $(X_i/Al)_{Crust}$  is the concentration ratio of an interested element X  
8 to Al in the average crustal abundance (Taylor, 1964).

9 PCA can elucidate variance in a given dataset in terms of minimum number of  
10 significant component. This technique has been employed in the tunnel studies  
11 concerning source apportionment of airborne metals (Lin et al., 2005; Lawrence et al.,  
12 2013). The software used here is STATISTICA 12 (Statsoft Inc.). A factor loading of >  
13 0.7 was adopted in this study to assign source identification to a given principal  
14 component.

15

### 16 **3. Result and Discussions**

#### 17 **3.1 Chemical compositions**

18 Table 1 summarizes the data on PM mass concentrations in size-resolved  
19 aerosols at both the inlet and outlet sites in Hsuehshan Tunnel. The aerosols are  
20 treated into three size bins: submicron ( $PM_1$ ), fine ( $PM_{1-1.8}$ ), and coarse ( $PM_{1.8-10}$ )  
21 modes. During the sampling periods, the mass concentrations of  $PM_{10}$ , which were  
22 determined as the sum of aerosol masses at all corresponding stages with a cut-off  
23 diameter less than 10  $\mu m$ , ranged from 35 to 68  $\mu g/m^3$  (average:  $54 \pm 9 \mu g/m^3$ ) at the  
24 inlet site and from 106 to 241  $\mu g/m^3$  (average:  $162 \pm 42 \mu g/m^3$ ) at the outlet site.  
25 Submicron particles were the predominant fraction, accounting for  $60 \pm 6 \%$  and 82

1  $\pm 3$  % of  $PM_{10}$  mass at the entrance and the exit, respectively. The abundance of  
2 submicron PM may indicate that combustion processes are significant sources of  
3 tunnel aerosols, which are presumably dominated by carbonaceous particles (Zhu et  
4 al., 2010; Pio et al., 2013). Compared with the inlet site, higher concentrations of  
5  $PM_{1-1.8}$  and  $PM_1$  were observed at the outlet site by a factor of 2.5 and 4.4,  
6 respectively. For  $PM_{1.8-10}$ , the concentration at the outlet site was nearly equal to that  
7 at the inlet site (outlet-to-inlet ratio:  $\sim 1.1$ ). The outlet-to-inlet ratio of PM mass  
8 concentration increases with decreasing PM size, indicating that smaller particles are  
9 relatively efficiently transported from the entrance to the exit; previous studies have  
10 attributed such efficient transport to “piston effect” (Chang et al., 2009; Cheng et al.,  
11 2010; Moreno et al., 2014). These authors suggested that passing vehicles pick up  
12 air pollutants emitted from vehicle fleets and the flows lead them to the exit,  
13 resulting in the accumulation of large quantities of air pollutants in that area. On the  
14 other hand, the discrepancies of out-to-inlet ratios in  $PM_{10}$ ,  $PM_{1.8-10}$  and  $PM_1$  could  
15 also be ascribed to the distinct emission rates for PM in different size fractions  
16 inside the tunnel.

17 Figure 1a shows the average elemental concentrations of  $PM_{10}$  at the two sites  
18 in Hsuehshan Tunnel, and Figure 1b depicts the partitioning of trace elements  
19 among three size bins. As shown in Figure 1a, Fe was the most abundant element,  
20 with a mean concentration of  $2384 \pm 1416$  ng/m<sup>3</sup>. In addition to Na, Ca, and Al (300  
21 to 500 ng/m<sup>3</sup>), Zn, K, Ba, Cu, and Mg (up to 100 ng/m<sup>3</sup>) were also major metals in  
22  $PM_{10}$ , followed by Ti (73 ng/m<sup>3</sup>), Mn (29 ng/m<sup>3</sup>), Sb (23 ng/m<sup>3</sup>), and then followed  
23 by Mo, Pb, Ga, Sr, Ni, V, and Ce (1 ng/m<sup>3</sup> to 10 ng/m<sup>3</sup>). The rest of the elements  
24 have concentrations less than 1 ng/m<sup>3</sup> (i.e., 0.9 ng/m<sup>3</sup> for Bi to 0.02 ng/m<sup>3</sup> for U).  
25 Most elements exhibited significantly higher concentrations at the exit than at the

1 entrance ( $p < 0.05$ , Figure 1c), with the exception of a number of crustal elements  
2 such as Al, K, Mg, and Rb. This suggests that a lower road dust reservoir is present  
3 inside the freeway tunnel (Amato et al., 2012). Considerably high outlet-to-inlet  
4 ratios (ranging from 2.2 for Sr to 4.3 for Zn) were found for traffic-derived elements,  
5 including Zn, Cu, Ba, Mn, Sb, Sn, Pb, Ga, Sr, and Cd.

6

### 7 **3.2 Size distributions**

8 The average size distributions of some of the analyzed metals are shown in  
9 Figures 1b, 2 and S1. Barium (Ba), Cd, Cu, Fe, Ga, Mn, Mo, Sb, and Sn were  
10 predominant in coarse mode at the inlet site (Figure 1b). These elements displayed a  
11 typical mono-modal distribution with a major peak in the size range of 3.2 - 5.6  $\mu\text{m}$ ,  
12 while they had another small peak at 1.0 - 1.8  $\mu\text{m}$  at times (Figures 2 and S2). The  
13 size distribution patterns of these metals were consistent with the results observed by  
14 Harrison et al. (2012) at a curbside in central London. The authors assigned the  
15 elements Fe, Cu, Sb, Ba, and Zn to the non-exhaust traffic particles. At the outlet site,  
16 those elements (Ba, Cd, Cu, Fe, Ga, Mn, Mo, Sb, and Sn) similarly had a  
17 mono-modal size distribution, but the main peak shifted to 1.0 - 1.8  $\mu\text{m}$  (Figures 2  
18 and S2). Similar to that in PM mass, this shift was perhaps due to “piston effect,”  
19 which, as previously mentioned, facilitated the transport of finer PM to the exit.

20 Zinc (Zn) showed a bi-modal distribution for most samples at the entrance, with  
21 a major peak in the size range of 3.2 - 5.6  $\mu\text{m}$  and a second peak in the size range of  
22 0.56 - 1.0  $\mu\text{m}$ . Meanwhile, a mono-modal pattern with a major peak at 1.0 - 1.8  $\mu\text{m}$   
23 was found at the exit. Lead (Pb) displayed two peaks at the inlet site: one at 0.56 -  
24 1.0  $\mu\text{m}$  and another one at 3.2 - 5.6  $\mu\text{m}$ . However, Pb exhibited a typical  
25 mono-modal distribution at the outlet site, peaking at 0.32 - 0.56  $\mu\text{m}$ . Vanadium (V)

1 revealed a bi-modal size pattern with a major peak at 0.32 - 0.56  $\mu\text{m}$  and a second  
2 peak at 3.2 - 5.6  $\mu\text{m}$  at the inlet site, whereas it peaked at 0.18 - 0.32  $\mu\text{m}$  or 0.32 -  
3 0.56  $\mu\text{m}$  at the exit.

4 Aluminum (Al), Ca and Mg of predominant geological origins showed a  
5 typical mono-modal size distribution at the inlet site with a major peak at 3.2 - 5.6  
6  $\mu\text{m}$ ; however, a peak was occasionally found in the submicron particles (Figures S1  
7 and S3). For example, the abundance of Al, Ca and K was observed at submicron  
8 size in two sets of samples (July 21 and August 10). Such abundance was ascribed to  
9 non-crustal sources such as vaporization from lubricating oil and diesel emissions  
10 (Wang et al., 2003), which perhaps alters the size distributions of these crustal  
11 elements. Submicron mode, which is an indicator of combustion or high temperature  
12 processes, contributes non-negligible Ca and K, which are usually regarded as  
13 crustal elements. Potassium titanate and a number of volatile compounds are known  
14 to contain K and are therefore may be the possible sources of submicron K (Hee and  
15 Filip, 2005; Iijima et al., 2007). Submicron Ca probably originated from tailpipe  
16 emissions of lubricating oil (Kuo et al., 2009). Like traffic elements, these crustal  
17 elements had a major peak that shifted to 1.0 - 1.8  $\mu\text{m}$  at the outlet site, which was  
18 also arisen from the “piston effect”. At the inlet site, rare earth elements (REEs),  
19 such as La, Ce, Nd, Pr, and Sm, revealed a mono-modal size distribution with a  
20 major peak at 3.2 - 5.6  $\mu\text{m}$ . At the exit, such elements essentially showed a  
21 mono-modal distribution that peaked at 1.0 - 1.8  $\mu\text{m}$ .

22

23 **3.3 Sources of trace metals**

24 Figure 3 presents the results of enrichment factor analysis for all analyzed  
25 elements in three size bins of size-segregated particles at both the inlet and outlet sites.

1 EF values for all species were higher at the outlet than at the inlet site, suggesting that  
2 the influence of re-suspended road dust were insignificant for most metals at the exit.  
3 Enrichment factor values for Ca, K, Mg, Rb, Sr, and Ti in the three size-resolved  
4 particles were generally close to unity at both sites, demonstrating that these elements  
5 originated mainly from the resuspension of soil and road dust. EF values for these  
6 geological metals increased with decreasing size, indicating that these elements in  
7 smaller particles would be significantly influenced by anthropogenic sources such as  
8 diesel emissions, lubricating oil, and additive in oil fuels. For lanthanides, lower  
9 enrichment was found for La, Pr, Nd, and Sm in all three sized PM, although high EFs  
10 were occasionally found. This indicates that although such elements mainly originate  
11 from geological sources, they sometimes from mixed sources of dust and  
12 anthropogenic emissions such as automotive catalyst (Kulkarni et al., 2006). Cerium  
13 (Ce), which is one of the lanthanides, had higher EF values ( $>10$ ) in all size-resolved  
14 particles than La, Pr, Nd, and Sm, demonstrating that Ce is highly influenced by  
15 anthropogenic emissions. For the three size-resolved particles, Ce is highly correlated  
16 not only with La, Pr, Nd, and Sm but also with a number of anthropogenic elements,  
17 again implying that Ce originated from traffic emissions such as automotive catalyst  
18 and fuel additive of diesel vehicles as well as from a crustal source (Kulkarni et al.,  
19 2006; Cassee et al., 2011).

20 High EFs ( $> 10$ ) were obtained for As, Ba, Cd, Cu, Cr, Ga, Mo, Sb, Se, and Sn,  
21 indicating their anthropogenic origins. Among these elements, Cu is an additive in  
22 high temperature lubricant and is present in brake linings, approximately 1 - 10 % by  
23 weight (Sanders et al., 2003), and it has been used successfully as a good tracer for  
24 wear emission of road traffic (Pio et al., 2013). Correlation analyses (Table 2)  
25 illustrate that Ba, Cd, Ga, Mo, Sb, and Sn are well correlated with Cu ( $r > 0.93$ ) in

1 both coarse and fine modes, suggesting that, similar to Cu, these elements in  
2 Hsuehshan Tunnel originated mainly from wear abrasive sources. This could be  
3 supported by the presence of both BaSO<sub>4</sub> and Sb<sub>2</sub>S<sub>3</sub>-containing particles in both brake  
4 lining materials, in which the former is utilized as a filler and the latter is utilized as  
5 an alternative to asbestos (Ingo et al., 2004). Moreover, the use of organic Sb  
6 compounds in grease and motor oil is another road traffic emission source of Sb  
7 (Huang et al., 1994; Cal-Prieto, 2001).

8       Lead (Pb) and Zn show high enrichment in all size fractions, indicating that both  
9 elements are contributed primarily by traffic emissions, rather than a natural origin.  
10 According to the bimodal distribution (Figure 2) and the good correlations with Cu,  
11 Ba, and Sb ( $r > 0.63$ ) in PM<sub>1.8-10</sub> (Table 2), Zn appears to originate from traffic  
12 emissions, and two traffic sources could account for the observed Zn. For the coarse  
13 mode, Zn is associated with wear tire debris because Zn is added to tires during  
14 vulcanization and is responsible for 1 - 2 % of the tires by weight (Degaffe and Tuner  
15 2011; Taheri et al., 2011). This is in concert with previous results (Adachi and  
16 Tainosho, 2004; Councill et al., 2004; Tanner et al., 2008, Harrison et al., 2012). For  
17 the fine mode, Zn is probably contributed by lubrication oil via pipe emissions  
18 (Huang et al., 1994). Emissions from vehicle exhaust and wear abrasion are both  
19 important sources of Pb. In this work, Pb showed good correlation with Cu, Sb, Ba  
20 and Zn ( $r > 0.60$ ) in coarse PM, indicating wear debris origins. In fine mode, Pb  
21 correlated well with Cu, Sb and Ba ( $r > 0.69$ ), suggesting wear abrasion dust. Moreover,  
22 good correlation between Pb and Zn ( $r = 0.75$ , in Table 2) was also found, revealing  
23 that Pb in fine PM might also be derived from diesel engines since both species were  
24 detected together and constituted up to 2 % in fresh diesel PM (Sharma et al., 2005;  
25 Agarwal et al., 2014). On the contrary, Pb only correlated well with Zn ( $r = 0.77$ ) in

1 submicron size (in Table S3), reflecting that Pb was preferentially contributed by  
2 combustion process from vehicle fleets (Wang et al., 2003).

3 Iron (Fe), which is considered an important crustal element, exhibited enrichment  
4 factors of 5 to 11 at the entrance and 12 to 21 at the exit, indicating that Fe in the  
5 tunnel was mainly produced from anthropogenic emissions other than road dust.  
6 Previous studies have pointed out that in addition to road dust, wear debris from brake  
7 linings and tires as well as diesel engine emissions are main sources of Fe in areas  
8 near traffic emissions (Cadle et al., 1997; Garg et al., 2000; Wang et al., 2003). In the  
9 present study, Fe correlated well with Cu, Ba, and Sb in all sizes ( $r > 0.87$ , Tables 2  
10 and S3), demonstrating that wear dust is a major anthropogenic source of Fe in  
11 Hsuehshan Tunnel, as is the case for those elements.

12 PCA results are presented in Table 3, in which the data (samples) are divided into  
13 three size groups. Two possible sources are identified for coarse PM. As seen, PC1  
14 was associated with Fe, Ba, Mn, Cu, Mo, Cd, Sb, Ti, V and Ga; moderate loadings  
15 were found for Zn and Pb. This indicates that PC1 was likely attributed to wear debris  
16 (Wåhlin et al., 2006; Lawrence et al., 2013; Pio et al., 2013). In PC2, high loadings  
17 were found for Na, Mg, K, Ca and Rb, inferring road dust origins. For fine particles,  
18 Fe, Ba, Mn, Cu, Mo, Cd, Sb, Mg, K, Ca, Rb, La, and Ce all had high loadings,  
19 whereas Pb had moderate loadings in PC1; brake abrasion mixed with re-suspended  
20 dust and gasoline emissions might explain this factor. In PC2, high positive loading  
21 was found for Zn; moderate loading for Pb; thus, PC2 could be explained by diesel  
22 emissions (Sharma et al., 2005; Agarwal et al., 2014). The third component was  
23 identified as road dust because of the correlations among Na, Al, and Mg (loadings  $>$   
24 0.6). For submicron particles, high loadings were found for Fe, Ba, Cu, Mo, Sb, Ga  
25 and Ce in PC 1. As previously mentioned, Ce in smaller PM may be associated with

1 catalyst converter and fuel additives; therefore, PC 1 might be grouped into mixed  
2 sources of wear abrasion and auto catalyst. In PC 2, high positive loadings were found  
3 for Pb and Zn, illustrating that exhaust from diesel engine was a potential source in  
4 this component. However, PC 3, which had a high loading of Al and a moderate  
5 loading of Ca indicates that road dust could be the potential source. PC4 is a  
6 component with high loading for V and Ni. Previous studies have suggested that V  
7 and Ni in submicron particles were commonly attributed to fuel oil combustion of  
8 gasoline and diesel engines (Wang et al., 2003; Shafer et al., 2008), but higher  
9 emission rates for gasoline exhaust compared to diesel engines (Cheng et al., 2010b).  
10 Consequently, PC4 in submicron PM may be associated preferentially from gasoline  
11 engines. Overall, wear abrasion dust and road dust are major sources of many airborne  
12 metals through all sized PM ranges inside Hsuehshan Tunnel, and combustion  
13 processes from vehicle fleets are additional sources of fine and submicron particles  
14 bound metals.

15

### 16 **3.4 Fingerprinting ratios of traffic-derived metals**

17 Cu is used as an indicator for wear debris, and the ratios of wear-derived  
18 elements to Cu obtained by linear regression approach can be applied to determine  
19 the contribution of specific metals from wear debris in urban atmosphere. Figure 4  
20 presents the scatter plots of Fe, Ba, Sb, Sn, Ga and Mo against Cu in PM<sub>1</sub>, PM<sub>1-1.8</sub>,  
21 and PM<sub>1.8-10</sub> at the two sites. These elements had strong correlations ( $r > 0.9$ ), and  
22 these ratios were constant in different size-resolved PM, strongly suggesting that  
23 these ratios can be applied as good fingerprinting ratios of wear emissions. The  
24 mean mass ratios of Fe/Cu, Ba/Cu, Sb/Cu, Sn/Cu, and Ga/Cu were 14, 1.05, 0.16,  
25 0.10 and 0.03, respectively. Table 4 compares our ratios to those established by other



1 tunnel studies. The ratios of Fe/Cu held around 14 to 15 over all sizes in the present  
2 work, which agrees with that (14) acquired by dynamometer tests (Sanders et al.,  
3 2003) and is also comparable to those observed in different tunnels (Gillies et al.,  
4 2001; Fabretti et al., 2009; Cheng et al., 2010b; Pio et al., 2013). However, the  
5 Fe/Cu ratio is also significantly distinct from those (37 to 60) found in other tunnels;  
6 such difference may have arisen from discrepancies in ingredients of brake pads and  
7 in driving conditions (Garg et al., 2000). Ba/Cu ratios of 0.8 - 1.1 were similar to  
8 those found in Europe but slightly lower than that ( $> 2$ ) found in the United States.  
9 Our Sb/Cu ratio of 0.16 is consistent with the result obtained in Hong Kong but  
10 lower than that (0.76 to 0.88) occasionally measured in American countries (Gillies  
11 et al., 2001; Mancilla and Menodza, 2012). In Japan, Iijima et al. (2007), with the  
12 use of dynamometer tests, reported Sb/Cu ratios ranging from 0.05 to 0.11 for  
13 different brake pads. They also pointed out that Sb-free brake pads have been  
14 utilized recently in Japanese passenger cars. According to the Taiwan Transportation  
15 Vehicle Manufactures Association 44 % and 13 % of vehicle fleets in Taiwan are  
16 Japanese and American cars, respectively. The abundance of Japanese cars in Taiwan  
17 may have caused the lower Sb/Cu values in this work. For the Mo against Cu scatter  
18 plot, two slopes are obtained: 0.05 for coarse and fine particles and 0.12 for particles  
19 with aerodynamic diameter less than  $0.56 \mu\text{m}$ . The enhancement of Mo in such  
20 submicron particles is perhaps attributed to an additional source of Mo such as  
21 diesel exhausts (Kuo et al., 2009). Previous studies show that the ratio of V/Ni has  
22 been widely used as a fingerprinting ratio of specific anthropogenic origins. For  
23 example, heavy oil combustion shows a narrow range of V/Ni ratio (3 to 4)  
24 (Hedberg et al., 2005; Mazzei et al., 2008). Combustion origins from gasoline and  
25 diesel vehicles have smaller V/Ni ratios ( $< 2.0$ ) (Qin et al., 1997; Watson et al.,

1 2001). On the other hand, small quantities of V and Ni are also found in soil with a  
2 V/Ni ratio of  $< 1.5$  (Hsu et al., unpublished data). In this work, V/Ni ratios were  
3 typically lower than 2.0 in fine and submicron PM; the ratios which were  
4 alternatively acquired directly from their mass concentrations (instead of linear  
5 regression) because V is not strongly correlated with Ni ( $r < 0.5$ , Tables 2 and S3) in  
6 three different sizes. In fine and submicron PM, the lower V/Ni ratios with higher  
7 EF value ( $>10$ ) for both elements suggest that they were contributed mostly by oil  
8 combustion from traffic fleets. In coarse PM, a low V/Ni ratio ( $<2$ ) with a low EF  
9 value ( $\sim 2$ ) for V indicate that V was associated with soil origins; however, high EF  
10 for Ni suggests that Ni was contributed by combustion sources. The Pb/Cu ratios in  
11 the tunnel particles averaged at 0.07, which is much lower than those (much higher  
12 than unity) usually observed in ambient air (Fang et al., 2005). In addition, the  
13 tunnel particles had As/Sb and Se/Sb ratios of 0.1 and 0.05, respectively, which are  
14 also evidently lower than those (around unity) measured in ambient aerosols (Querol  
15 et al., 2007). These results imply that traffic emissions are not major sources of Pb,  
16 As, and Se in ambient atmospheres.

17 Figure 5 illustrates the relationships of La against Ce, Pr, Nd, and Sm. Their  
18 correlations weaken with decreasing particle size, suggesting that the REEs in  
19 smaller particles were disturbed by certain anthropogenic sources. A ratio of La/Ce  
20 has been successfully used to distinguish natural sources from anthropogenic origins  
21 (Kulkarni et al., 2006). In this work, the La/Ce ratios that range from 0.15 to 0.18  
22 and from 0.10 to 0.12 at the inlet and outlet sites, respectively, are expectedly  
23 significantly lower than that of average crust ( $\sim 0.50$ ) (Taylor, 1964) and soils ( $\sim 0.7$ )  
24 (Kulkarni et al., 2006). Such values agree with those of Kulkarnu et al. (2006) and  
25 Huang et al. (1994) who reported that La/Ce ratios for traffic emissions were 0.20

1 and 0.13, respectively. As discussed in Section 3.2, the EF values of Ce were mostly  
2 higher than unity at both the inlet and outlet sites, with even some of the values  
3 being one order of magnitude higher (Figure 3), revealing that soil dust is not the  
4 sole source of Ce. Thus, the low La/Ce values found in the present study could be  
5 attributed to an additional supply of Ce from vehicular emissions.

6

7 **4. Summary and concluding remarks**

8 Size-fractionated aerosol samples were collected in Hsuehshan Tunnel to  
9 characterize particulate metals emitted by vehicle fleets. A total of 36 elements were  
10 analyzed by ICP-MS. Compared to the entrance, enhanced concentrations for most  
11 metals at the exit are due to “piston effect”. With regard to enrichment factor,  
12 correlation matrix, and principal component analysis, the analyzed metals were  
13 categorized into three groups, namely, wear abrasion (Cu, Cd, Cu, Fe, Ga, Mn, Mo,  
14 Sb, and Sn), re-suspended dust (Ca, Mg, K and Rb), and pipe emissions (Zn, Pb and  
15 V in fine mode). Size distributions of these elements were significantly different  
16 because of their origins. For wear-related metals and geological elements, a  
17 mono-modal size distribution was found and the major peak shifted from the range  
18 of 3.2 - 5.6  $\mu\text{m}$  at the entrance to the range of 1 - 1.8  $\mu\text{m}$  at the exit. However,  
19 elements attributed to combustion sources were predominant mainly in submicron  
20 particles and peaked at 0.56 - 1.0  $\mu\text{m}$  at the inlet site and at 0.18 - 0.32  $\mu\text{m}$  or 0.32 -  
21 0.56  $\mu\text{m}$  at the outlet site. By adopting Cu as an indicator element of wear debris,  
22 fingerprinting ratios were constructed, including Fe/Cu, Ba/Cu, Sb/Cu, Sn/Cu and  
23 Ga/Cu. These ratios can effectively apportion the source of specific elements in  
24 urban environment from wear abrasion.

25 In this work, we characterized traffic-derived PM metals using a tunnel study.

1 The data would be useful for future studies on traffic emission inventory and health  
2 effects, especially for submicron PM. Wear abrasion appeared to be a major source  
3 of specific toxic elements. While the government focuses on exhaust emission  
4 control, the contribution of wear from brake linings and tires could not be ignored.  
5 Thus, stringent implementations of measures for reducing wear emissions are  
6 needed in the future.

7

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15

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## Table captions

Table 1. Summary of sampling dates, mass concentrations ( $\mu\text{g}/\text{m}^3$ ) of  $\text{PM}_{1.8-10}$ ,  $\text{PM}_{1-1.8}$  and  $\text{PM}_1$  as well as traffic flow and wind speed in Hsuehshan Tunnel during the sampling periods in 2013.

Table 2. Correlation matrix of selected elements in coarse (top side triangle) and fine particles (lower side triangle) observed in Hsuehshan Tunnel. Correlation coefficients higher than 0.8 are marked in bold.

Table 3. Summaries of principal component analysis for trace metals in coarse, fine and submicron particles observed in Hsuehshan Tunnel. Factor loadings lower than  $\pm 0.4$  are not given. Loading factor greater than 0.7 is marked by bold.

Table 4. Ratios of specific elements to Cu in tunnel PM.

## Figure captions

Figure 1. (a) Elemental compositions of PM<sub>10</sub> collected at both the inlet and outlet sites in Hsuehshan Tunnel; (b) partitioning of trace metals within three sized PM; (c) outlet-to-inlet ratio for each element in PM<sub>10</sub>. The sequence of metallic species is in order of decreasing concentrations (ng/m<sup>3</sup>) at the outlet site. N denotes the number of aerosol samples.

Figure 2. Average size distributions of traffic-derived elements observed at the inlet and outlet sites inside Hsuehshan Tunnel.

Figure 3. Enrichment factors of trace metals in (a) PM<sub>1</sub>, (b) PM<sub>1-1.8</sub> and (c) PM<sub>1.8-10</sub> observed at the inlet and outlet sites in Hsuehshan Tunnel.

Figure 4. Scatter plots of (a) Fe, (b) Ba, (c) Sb, (d) Sn, (e) Ga, (f) Mo against Cu concentrations (ng/m<sup>3</sup>) in different size-segregated particles observed in Hsuehshan Tunnel.

Figure 5. Scatter plots of La and (a) Ce, (b) Pr, (c) Nd and (d) Sm concentrations (ng/m<sup>3</sup>) in different size-segregated particles observed in Hsuehshan Tunnel.



Table 1

Summary of sampling dates, mass concentrations ( $\mu\text{g}/\text{m}^3$ ) of  $\text{PM}_{1.8-10}$ ,  $\text{PM}_{1-1.8}$  and  $\text{PM}_1$  as well as traffic flow and wind speed in Hsuehshan Tunnel during the sampling periods in 2013.

Sampling No.	Date	Inlet Site			Outlet Site			Vehicle fleet		Wind Speed (m/s)
		$\text{PM}_{1.8-10}$	$\text{PM}_{1-1.8}$ ( $\mu\text{g}/\text{m}^3$ )	$\text{PM}_1$	$\text{PM}_{1.8-10}$	$\text{PM}_{1-1.8}$ ( $\mu\text{g}/\text{m}^3$ )	$\text{PM}_1$	LDV (No./hr)	HDV (No./hr)	
1	2013/5/17	17	4	32	17	9	155	1272	72	4.7
2	2013/5/18	18	7	43	18	11	128	1777	88	4.6
3	2013/5/19	19	6	35	21	12	208	1843	109	4.7
4	2013/7/19	16	4	27	26	9	83	1277	104	4.3
5	2013/7/20	16	3	34	15	9	142	1400	118	4.8
6	2013/7/21	13	3	33	20	9	168	1680	126	4.7
7	2013/8/8	17	4	26	15	11	142	1354	109	4.7
8	2013/8/9	19	4	39	9	10	87	1460	133	5.2
9	2013/8/10	9	3	23	16	10	126	1712	81	4.9
10	2013/9/27	27	4	22	28	10	125	1334	81	4.7
11	2013/9/28	22	4	39	16	9	85	1764	101	5.0
12	2013/9/29	15	4	34	18	10	180	1909	121	4.7

Table 2

Correlation matrix of selected elements in coarse (top side triangle) and fine particles (lower side triangle) observed in Hsuehshan Tunnel.

Correlation coefficients higher than 0.8 are marked in bold.

	Al <sup>a</sup>	Fe	Mg	K	Ca	Sr	Ba	Ti	Mn	Ni	Cu	Zn	Mo	Cd	Sn	Sb	Pb	V	Cr	Rb	Cs	Ga	La	Ce	Pr	Nd
Al		0.29	0.42	0.44	0.44	0.45	0.29	0.35	0.34	0.05	0.25	0.43	0.16	0.19	0.17	0.17	0.49	0.20	0.10	0.47	0.41	0.30	0.40	0.20	0.45	0.25
Fe	0.69		0.27	0.31	0.43	<b>0.88</b>	<b>0.97</b>	<b>0.96</b>	<b>1.00</b>	-0.03	<b>0.99</b>	0.66	<b>0.98</b>	<b>0.97</b>	<b>0.97</b>	<b>0.98</b>	0.64	0.74	0.57	0.41	0.38	<b>0.96</b>	0.71	<b>0.91</b>	0.73	<b>0.90</b>
Mg	0.78	<b>0.84</b>		0.74	0.61	0.62	0.36	0.44	0.30	-0.09	0.24	0.31	0.23	0.29	0.29	0.24	0.62	0.42	0.03	0.75	0.62	0.38	0.66	0.44	0.55	0.50
K	0.71	<b>0.89</b>	<b>0.84</b>		0.61	0.63	0.45	0.44	0.36	0.25	0.22	0.56	0.25	0.32	0.31	0.26	0.68	0.45	0.37	0.91	<b>0.88</b>	0.45	0.68	0.48	0.69	0.55
Ca	0.70	<b>0.86</b>	<b>0.82</b>	<b>0.86</b>		0.75	0.57	0.56	0.48	-0.05	0.38	0.59	0.39	0.48	0.47	0.46	<b>0.90</b>	0.49	0.13	<b>0.81</b>	0.74	0.61	<b>0.82</b>	0.47	0.74	0.55
Sr	0.64	<b>0.99</b>	<b>0.86</b>	<b>0.89</b>	<b>0.88</b>		<b>0.94</b>	<b>0.93</b>	<b>0.90</b>	-0.04	<b>0.83</b>	0.76	<b>0.83</b>	<b>0.88</b>	<b>0.87</b>	<b>0.86</b>	<b>0.87</b>	0.75	0.45	0.75	0.67	<b>0.94</b>	<b>0.90</b>	<b>0.86</b>	<b>0.88</b>	<b>0.90</b>
Ba	0.60	<b>0.98</b>	<b>0.81</b>	<b>0.87</b>	<b>0.82</b>	<b>0.99</b>		<b>0.95</b>	<b>0.97</b>	-0.04	<b>0.93</b>	0.77	<b>0.94</b>	<b>0.96</b>	<b>0.96</b>	<b>0.96</b>	0.75	0.73	0.52	0.56	0.52	<b>1.00</b>	<b>0.81</b>	<b>0.91</b>	<b>0.82</b>	<b>0.92</b>
Ti	0.68	<b>0.99</b>	<b>0.85</b>	<b>0.87</b>	<b>0.84</b>	<b>0.98</b>	<b>0.97</b>		<b>0.96</b>	0.02	<b>0.96</b>	0.70	<b>0.95</b>	<b>0.96</b>	<b>0.96</b>	<b>0.96</b>	0.75	0.79	0.50	0.55	0.51	<b>0.95</b>	<b>0.80</b>	<b>0.88</b>	0.75	<b>0.89</b>
Mn	0.63	<b>0.95</b>	<b>0.80</b>	<b>0.90</b>	<b>0.91</b>	<b>0.95</b>	<b>0.95</b>	<b>0.93</b>		0.45	<b>0.98</b>	0.69	<b>0.97</b>	<b>0.96</b>	<b>0.97</b>	<b>0.97</b>	0.68	0.75	0.60	0.46	0.43	<b>0.97</b>	0.74	<b>0.90</b>	0.76	<b>0.90</b>
Ni	0.01	0.08	0.02	0.11	-0.01	0.05	0.06	0.06	0.12		-0.09	-0.02	-0.07	-0.11	-0.10	-0.09	-0.03	-0.02	0.73	0.15	0.16	-0.05	0.05	0.00	0.06	0.03
Cu	0.66	<b>0.99</b>	<b>0.83</b>	<b>0.85</b>	<b>0.82</b>	<b>0.98</b>	<b>0.97</b>	<b>1.00</b>	<b>0.93</b>	0.06		0.63	<b>0.99</b>	<b>0.97</b>	<b>0.98</b>	<b>0.98</b>	0.60	0.73	0.51	0.32	0.30	<b>0.93</b>	0.66	<b>0.87</b>	0.64	<b>0.85</b>
Zn	0.22	0.50	0.34	0.55	0.61	0.50	0.52	0.47	0.72	0.43	0.45		0.63	0.72	0.67	0.67	0.76	0.49	0.31	0.56	0.51	0.78	0.67	0.56	0.65	0.60
Mo	0.61	<b>0.98</b>	<b>0.81</b>	<b>0.84</b>	<b>0.82</b>	<b>0.98</b>	<b>0.98</b>	<b>0.99</b>	<b>0.93</b>	0.05	<b>0.99</b>	0.47		<b>0.98</b>	<b>0.99</b>	<b>0.99</b>	0.60	0.75	0.54	0.34	0.33	<b>0.93</b>	0.67	<b>0.88</b>	0.65	<b>0.86</b>
Cd	0.56	<b>0.96</b>	0.76	<b>0.86</b>	<b>0.87</b>	<b>0.95</b>	<b>0.96</b>	<b>0.95</b>	<b>0.98</b>	0.17	<b>0.95</b>	0.70	<b>0.96</b>		<b>1.00</b>	<b>0.99</b>	0.68	0.73	0.49	0.43	0.41	<b>0.96</b>	0.72	<b>0.87</b>	0.69	<b>0.86</b>
Sn	0.60	<b>0.98</b>	<b>0.80</b>	<b>0.84</b>	<b>0.83</b>	<b>0.98</b>	<b>0.97</b>	<b>0.99</b>	<b>0.93</b>	0.05	<b>0.99</b>	0.48	<b>1.00</b>	<b>0.96</b>		<b>0.99</b>	0.66	0.74	0.51	0.41	0.40	<b>0.95</b>	0.72	<b>0.89</b>	0.69	<b>0.88</b>
Sb	0.63	<b>0.99</b>	<b>0.81</b>	<b>0.85</b>	<b>0.85</b>	<b>0.98</b>	<b>0.98</b>	<b>0.99</b>	<b>0.94</b>	0.05	<b>0.99</b>	0.50	<b>0.99</b>	<b>0.97</b>	<b>1.00</b>		0.64	0.74	0.52	0.38	0.36	<b>0.95</b>	0.71	<b>0.87</b>	0.68	<b>0.86</b>
Pb	0.59	0.73	0.76	<b>0.84</b>	<b>0.89</b>	0.75	0.70	0.71	<b>0.85</b>	0.21	0.69	0.75	0.68	<b>0.80</b>	0.70	0.70		0.62	0.27	<b>0.81</b>	0.73	0.77	<b>0.91</b>	0.64	0.78	0.70
V	0.28	0.39	0.31	0.49	0.35	0.38	0.38	0.41	0.37	0.22	0.40	0.20	0.42	0.40	0.39	0.38	0.44		0.45	0.54	0.52	0.73	0.71	0.72	0.65	0.75
Cr	0.20	0.41	0.28	0.30	0.27	0.38	0.39	0.38	0.44	<b>0.84</b>	0.39	0.60	0.38	0.49	0.39	0.38	0.40	0.11		0.31	0.32	0.49	0.37	0.53	0.44	0.54
Rb	0.64	<b>0.81</b>	0.74	<b>0.92</b>	<b>0.92</b>	<b>0.83</b>	0.79	0.78	<b>0.89</b>	0.07	0.75	0.64	0.75	<b>0.82</b>	0.76	0.77	<b>0.90</b>	0.40	0.27		<b>0.96</b>	0.56	<b>0.82</b>	0.57	<b>0.83</b>	0.65
Cs	0.50	0.65	0.56	<b>0.80</b>	<b>0.82</b>	0.67	0.64	0.61	0.77	0.11	0.58	0.65	0.58	0.70	0.60	0.61	<b>0.84</b>	0.44	0.23	<b>0.95</b>		0.51	0.74	0.53	0.77	0.61
Ga	0.60	<b>0.99</b>	<b>0.81</b>	<b>0.85</b>	<b>0.85</b>	<b>0.99</b>	<b>0.99</b>	<b>0.98</b>	<b>0.95</b>	0.06	<b>0.98</b>	0.53	<b>0.98</b>	<b>0.97</b>	<b>0.99</b>	<b>0.98</b>	0.71	0.38	0.40	0.79	0.63		<b>0.82</b>	<b>0.90</b>	<b>0.82</b>	<b>0.91</b>
La	0.71	<b>0.87</b>	<b>0.81</b>	<b>0.88</b>	<b>0.94</b>	<b>0.88</b>	<b>0.82</b>	<b>0.85</b>	<b>0.88</b>	0.04	<b>0.83</b>	0.52	<b>0.83</b>	<b>0.84</b>	<b>0.84</b>	<b>0.84</b>	<b>0.87</b>	0.44	0.32	<b>0.89</b>	0.77	<b>0.84</b>		0.79	<b>0.89</b>	<b>0.84</b>
Ce	0.60	<b>0.89</b>	<b>0.80</b>	<b>0.81</b>	0.78	<b>0.90</b>	<b>0.86</b>	<b>0.88</b>	<b>0.81</b>	0.00	<b>0.89</b>	0.30	<b>0.90</b>	<b>0.81</b>	<b>0.89</b>	<b>0.87</b>	0.67	0.36	0.32	0.72	0.54	<b>0.87</b>	<b>0.87</b>		<b>0.83</b>	<b>0.99</b>
Pr	0.68	<b>0.90</b>	<b>0.81</b>	<b>0.89</b>	<b>0.82</b>	<b>0.92</b>	<b>0.89</b>	<b>0.87</b>	<b>0.87</b>	0.01	<b>0.86</b>	0.43	<b>0.86</b>	<b>0.82</b>	<b>0.85</b>	<b>0.85</b>	0.70	0.28	0.31	<b>0.85</b>	0.68	<b>0.87</b>	<b>0.85</b>	<b>0.87</b>		<b>0.88</b>
Nd	0.62	<b>0.91</b>	<b>0.82</b>	<b>0.83</b>	<b>0.82</b>	<b>0.92</b>	<b>0.88</b>	<b>0.91</b>	<b>0.84</b>	0.01	<b>0.91</b>	0.32	<b>0.92</b>	<b>0.84</b>	<b>0.91</b>	<b>0.89</b>	0.70	0.36	0.32	0.76	0.57	<b>0.89</b>	<b>0.90</b>	<b>1.00</b>	<b>0.90</b>	

Table 3

Summaries of principal component analysis for trace metals in coarse, fine and submicron particles observed in Hsuehshan Tunnel. Factor loadings lower than  $\pm 0.4$  are not given. Loading factor greater than 0.7 is marked by bold.

	Coarse		Fine			Submicron			
	PC1	PC2	PC1	PC2	PC3	PC1	PC2	PC3	PC4
Al <sup>a</sup>		0.55	0.52		0.68			<b>0.88</b>	
Fe	<b>0.98</b>		<b>0.94</b>			<b>0.82</b>	0.52		
Na		<b>0.81</b>			<b>0.93</b>				
Mg		<b>0.89</b>	<b>0.70</b>		0.66	0.69			
K		<b>0.88</b>	<b>0.77</b>		0.44		0.57		
Ca		<b>0.75</b>	<b>0.81</b>			0.65		0.53	
Ba	<b>0.94</b>		<b>0.95</b>			<b>0.96</b>			
Ti	<b>0.93</b>		<b>0.94</b>			<b>0.73</b>			
Mn	<b>0.97</b>		<b>0.90</b>				<b>0.96</b>		
Ni				<b>0.76</b>				0.56	<b>0.72</b>
Cu	<b>0.98</b>		<b>0.95</b>			<b>0.96</b>			
Zn	0.65	0.42	0.44	<b>0.79</b>			<b>0.97</b>		
Mo	<b>0.99</b>		<b>0.96</b>			<b>0.96</b>			
Cd	<b>0.97</b>		<b>0.92</b>				<b>0.90</b>		
Sb	<b>0.99</b>		<b>0.96</b>			<b>0.90</b>			
Pb	0.58	<b>0.71</b>	0.62	0.61			<b>0.84</b>		
V	<b>0.72</b>								<b>0.92</b>
Rb		<b>0.90</b>	<b>0.73</b>	0.44			0.63		
Ga	<b>0.93</b>		<b>0.96</b>			<b>0.94</b>			
La	0.65	0.68	<b>0.80</b>			<b>0.81</b>			
Ce	<b>0.86</b>		<b>0.86</b>			<b>0.92</b>			
Potential source	Wear debris	Dust	Wear debris + Dust + Gasoline	Diesel	Dust	Wear debris + Auto catalyst	Diesel	Dust	Fuel oil

Table 4

Ratios of specific elements to Cu in tunnel PM.

Tunnel studies	Size	Fe/Cu	Ba/Cu	Sb/Cu	Sn/Cu	Reference <sup>b</sup>
Hatfield Tunnel (United Kingdom)	PM <sub>10</sub>	19	1.23	0.13		1
Marquês de Pombal Tunnel (Portugal)	PM <sub>0.5-10</sub>	16	0.27	0.08	0.23	2
Tsngstad Tunnel (Sweden)	PM <sub>10</sub>	28	0.74	0.18		3
Lundby Tunnel (Sweden)	PM <sub>10</sub>	60	1.34	0.24		3
Malraux Tunnel (France)	PM <sub>2.5</sub>	15		0.14	0.14	4
Squirrel Hill Tunnel (USA) <sup>a</sup>	PM <sub>2.5</sub>	37	2.48	0.21	0.48	5
Sepulveda Tunnel (USA) <sup>a</sup>	PM <sub>2.5</sub>	16	2.12	0.88	0.82	6
Loma Largo Tunnel (Mexico)	PM <sub>2.5</sub>	7	0.13	0.76	0.49	7
Jãnio Tunnel (Brazil)	PM <sub>2.5</sub>	20		0.12		8
Belway Rodonael Mário Covas Tunnel (Brazil)	PM <sub>2.5</sub>	45		0.36		8
Shing Mun Tunnel (Hong Kong)	PM <sub>2.5</sub>	17	0.58	0.14	0.29	9
Zhuijiang Tunnel (China)	PM <sub>2.5</sub>	28	1.08			10
Hsuehshan Tunnel (Taiwan)	PM <sub>1.8-10</sub>	14	0.80	0.14	0.09	This study
	PM <sub>1-1.8</sub>	14	1.07	0.16	0.09	
	PM <sub>1</sub>	15	1.10	0.16	0.11	

<sup>a</sup>. The ratios of Squirrel Hill Tunnel and Sepulveda Tunnel are obtained from the ratios of elemental emission factors.

<sup>b</sup>. 1.Lawrence et al. (2013); 2.Pio et al. (2013); 3.Sternbeck et al. (2002); 4. Fabretti et al. (2009); 5.Grieshop et al. (2006); 6.Gillies et al. (2001); 7.Mancilla and Mendoza (2012); 8Brito et al. (2013); 9.Cheng et al. (2010); 10.He et al. (2008).

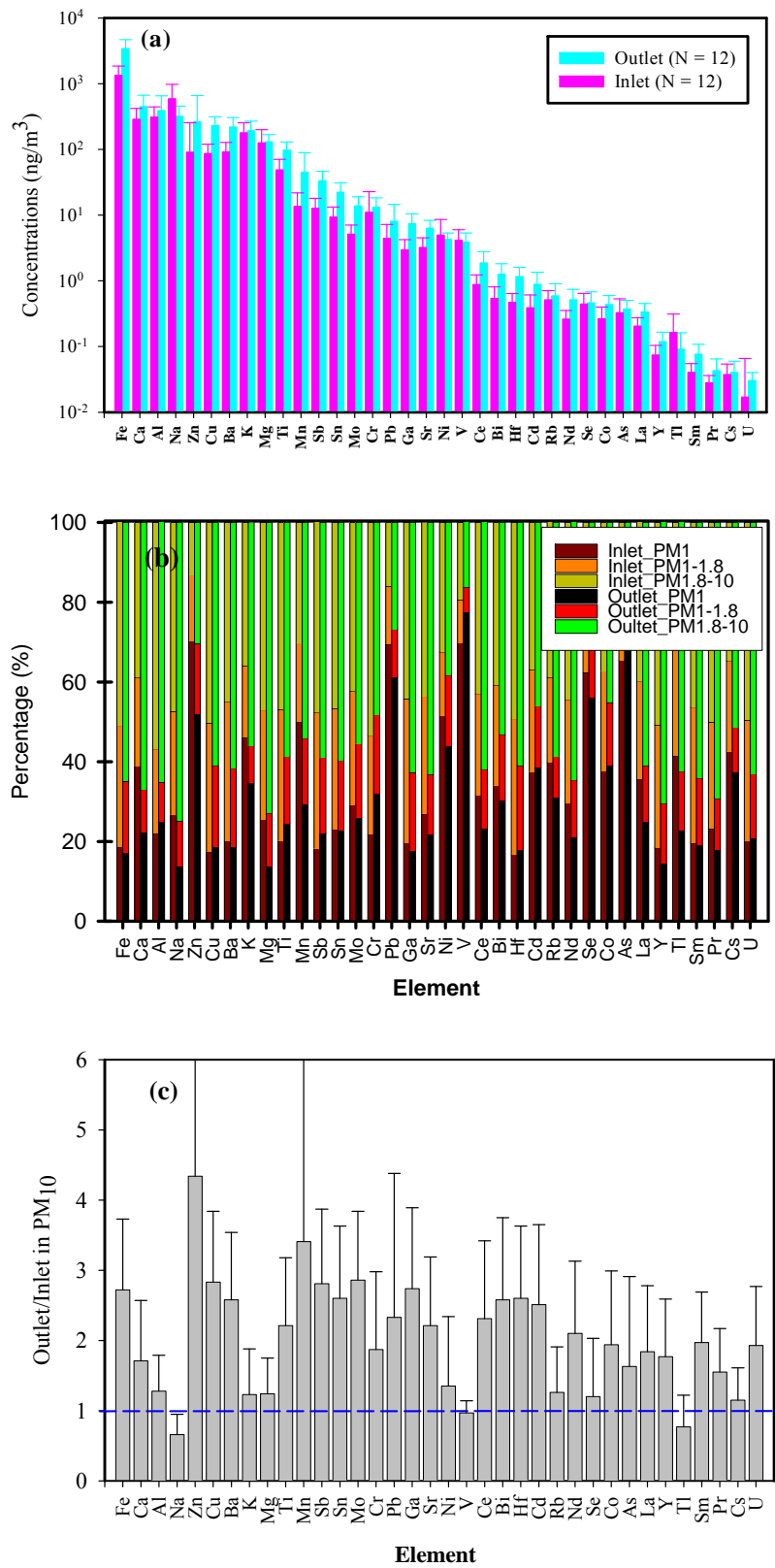


Figure 1.

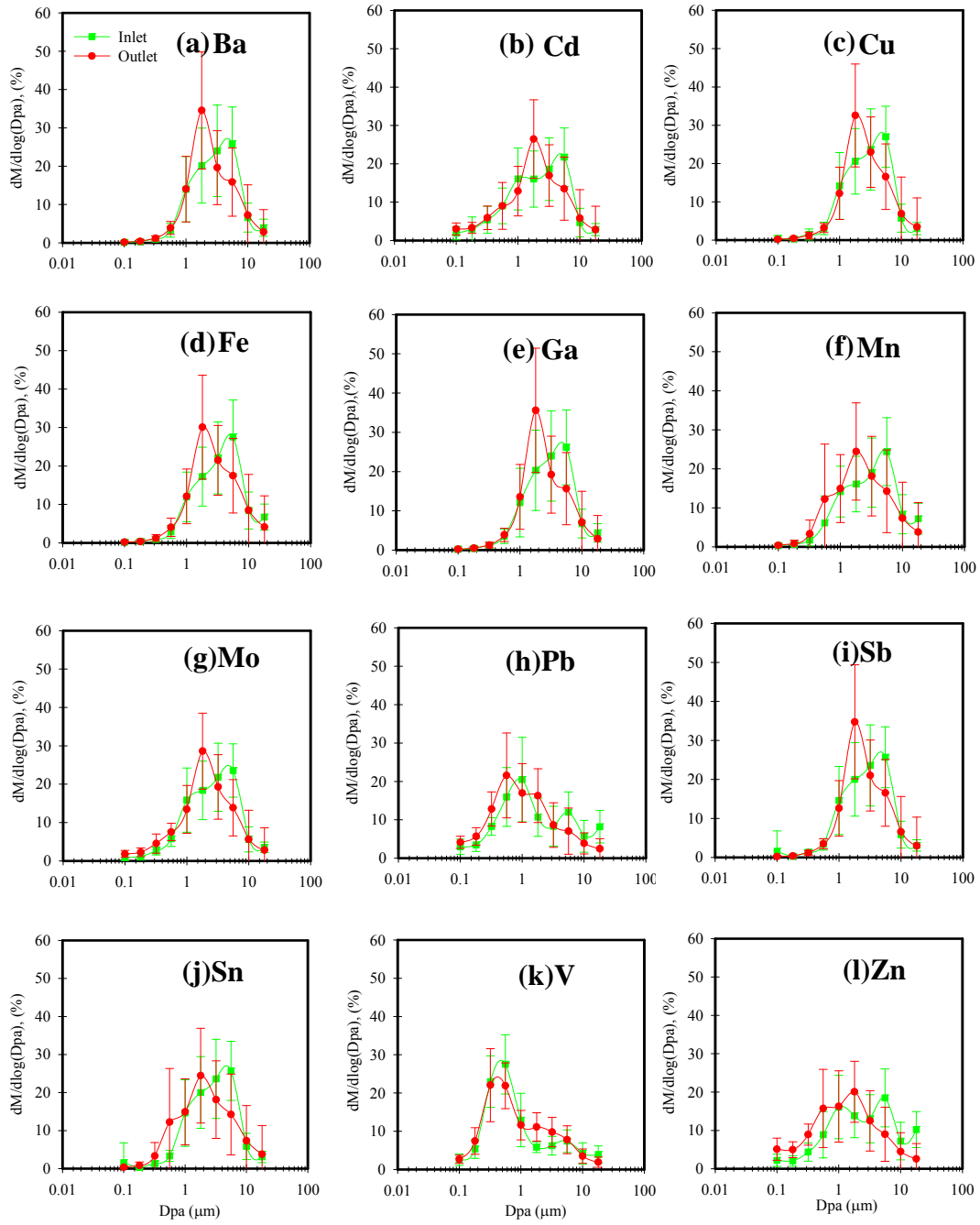


Figure 2.

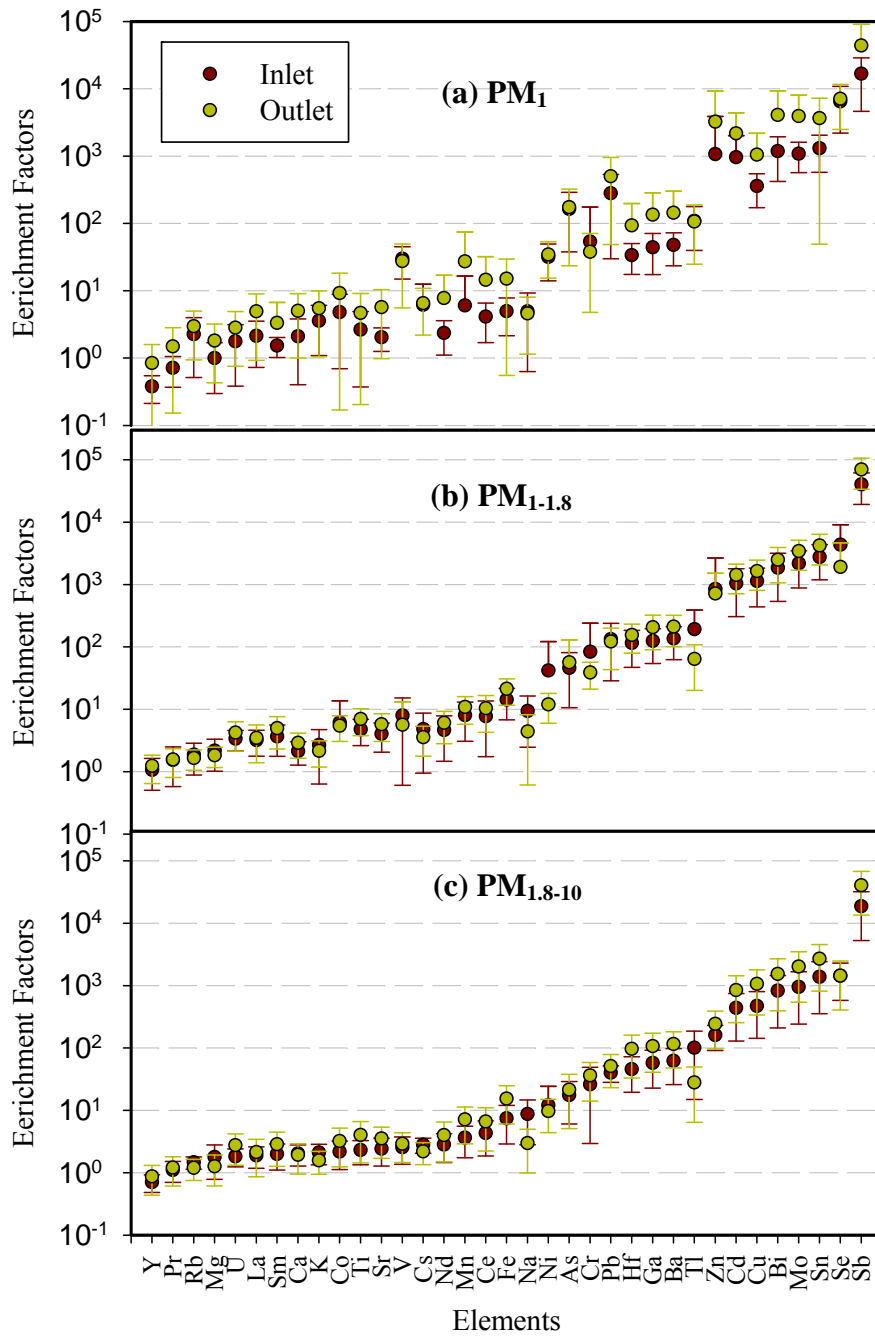


Figure 3.

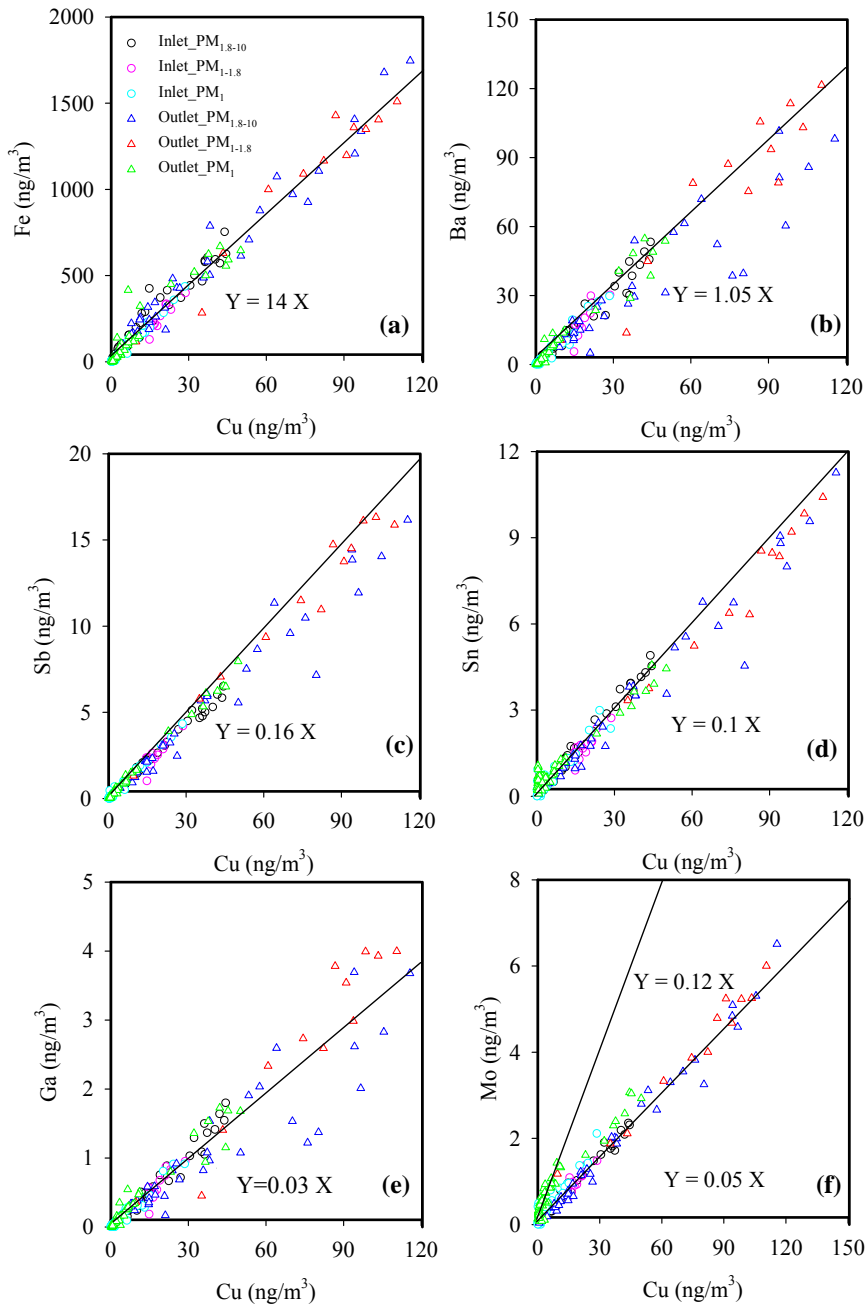


Figure 4.



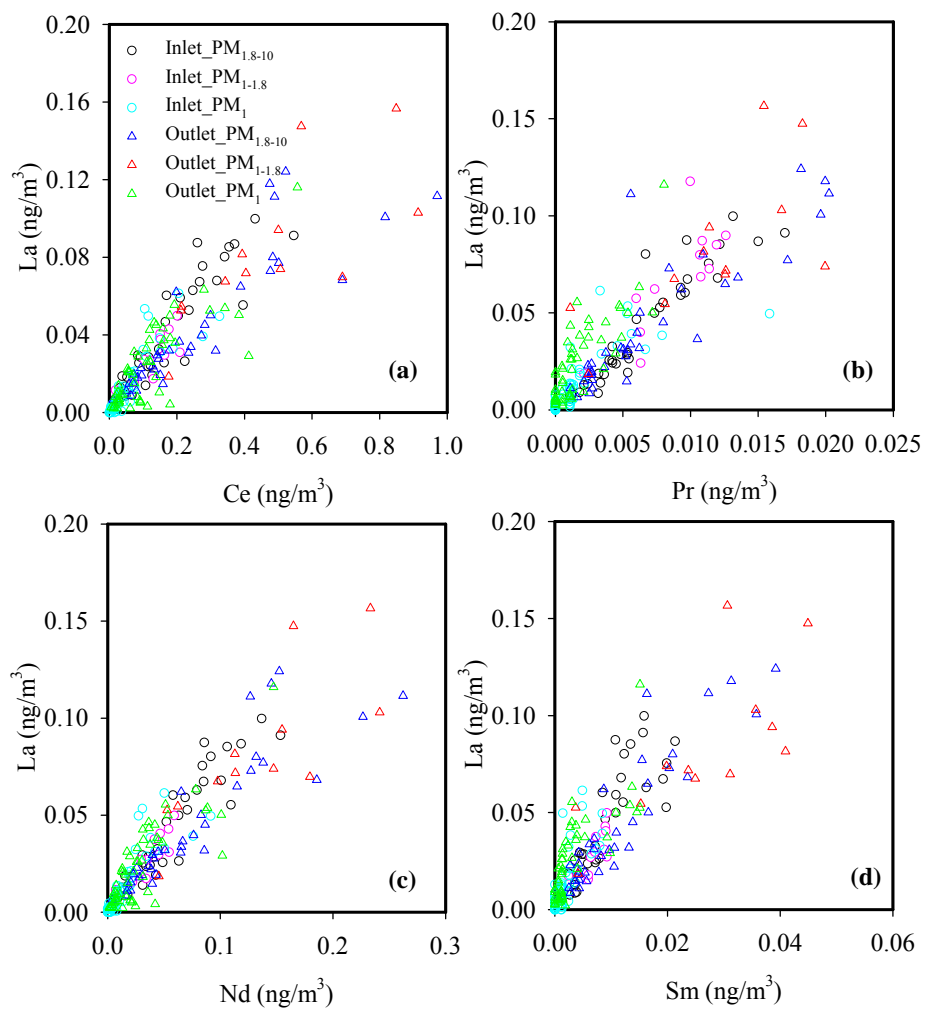


Figure 5.