1 Response to Reviewer #1

We thank the reviewer for taking the time to review our paper and for the constructive comments.

(1.1) [...] Nonetheless, there are serious problems with the physics the authors have achieved a mathematical definition of mixing state not a physical one. If discussions of physical entropy and the second law are omitted, the paper is fine. [...]

We agree with the reviewer that it is important to distinguish the thermodynamic and the informationtheoretic definition of entropy. To improve this aspect of our paper, we followed the reviewer's suggestion and made the following changes:

- In the introduction, page 3, line 24, changed the sentence to: "In this paper we present the first quantitative measure of aerosol mixing state, the mixing state index χ , based on diversity measures derived from the information-theoretic entropy of the chemical species distribution among particles."
- In the introduction, page 3, line 26, added "using information-theoretic entropy measures". The sentence now reads: "The measurement of species diversity and distribution using information-theoretic entropy measures has a long history in many scientific fields."
- We removed the sentence at the end of section 2 (page 7): "The case of a coagulating population represents a closed system as the total mass is conserved. Hence, according to the second law of thermodynamics, the total entropy of the system $H = \sum \mu_i H_i$ has to increase, which translates to an increase in D_{α} and the mixing state index χ ."
- We added sentence on intensive/extensive quantities at the end of section 2 (page 7, lines 7–11): "The population mixing results (Table 5 and Theorem 3) show that the diversities and entropies are intensive quantities. For example, doubling the size of particle i leaves H_i unchanged, and doubling the population leaves H_{α} unchanged. Extensive versions of these quantities can be defined by mass-weighting, so that the total mass-extensive entropy is $H = \sum_i \mu_i H_i$, for example."