

1 **Supplementary Information (SI)**

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3 **Dependence of particle nucleation and growth on high molecular weight gas phase**
4 **products during ozonolysis of α -pinene**

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13 The supplementary material contains three sections.

- 1 **Section I:** contains 3 tables and 6 figures.
- 2 Table S1. Summary of experimental conditions in this study.

Exp.	α -pinene	Ozone		Cluster CIMS reagent ions
	(ppb)	first 3 hours (ppb min ⁻¹)	steady state (ppb)	
E1	20	0.2	75	Nitrate dimer
E2	5	0.1	50	Nitrate dimer
E3	20	N/A	75	Acetate dimer

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1 Table S2. Summary of all measured Category II products for *EI*: ion (m/z), peak concentration
 2 (in cm⁻³), corresponding neutral identities, and correlation coefficients with > 20 nm particles.

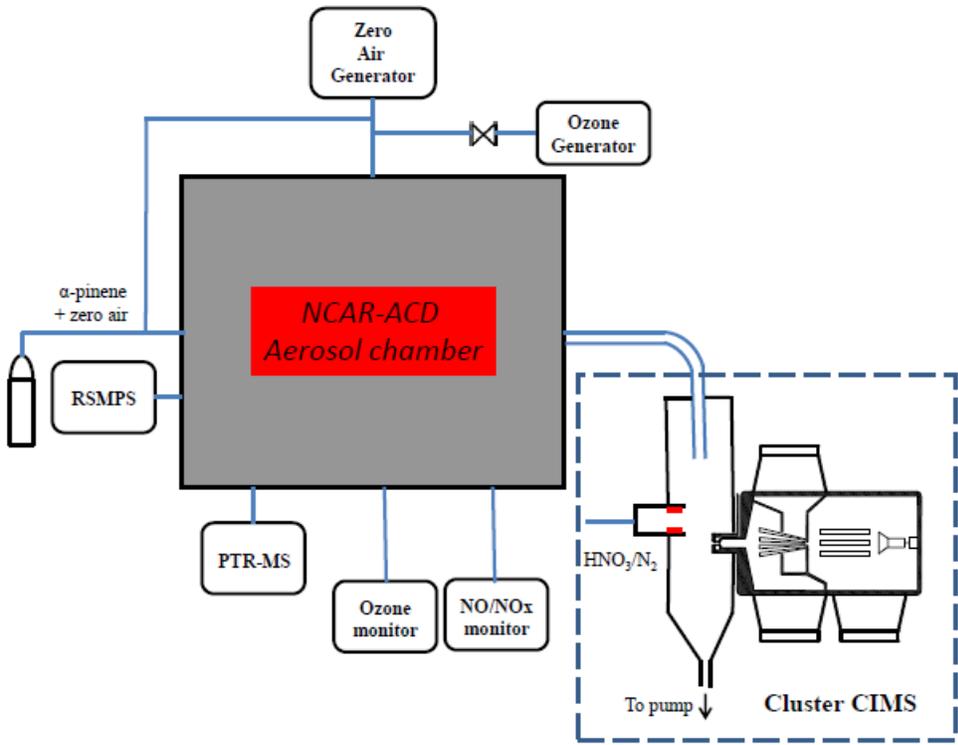
Ion (m/z)	Peak conc. ^a (cm ⁻³)	NO ₃ ⁻ Clusters? ^b	Corresponding neutral ^b		Correlation coefficient ^c
			Mass (amu)	Formula	
142	2.5x10 ⁵				0.63
158	2.1x10 ⁵				0.67
228	3.0x10 ⁵				0.66
240	4.2x10 ⁶	Y	178	C ₅ H ₆ O ₇	0.80
241	3.9x10 ⁵				0.60
257	2.6x10 ⁵				0.75
282	8.9x10 ⁵	Y	220	C ₇ H ₈ O ₈ or C ₈ H ₁₂ O ₇	0.51
287	5.2x10 ⁵				0.82
297	5.3x10 ⁵				0.81
298	4.3x10 ⁵	Y	236	C ₈ H ₁₂ O ₈	0.69
308	2.6x10 ⁶	Y	246	C ₁₀ H ₁₄ O ₇	0.78
309	5.2x10 ⁵				0.66
310	1.1x10 ⁶	Y	248	C ₉ H ₁₂ O ₈ or C ₁₀ H ₁₆ O ₇	0.77
326	7.5x10 ⁵			C ₉ H ₁₂ O ₉ or C ₁₀ H ₁₆ O ₈	0.61
327	5.5x10 ⁵				0.80
329	2.5x10 ⁵				0.82
339	8.0x10 ⁵				0.72
341	4.6x10 ⁵				0.61
343	3.0x10 ⁵				0.50
355	5.5x10 ⁵				0.74
356	4.3x10 ⁵	Y	294	C ₁₀ H ₁₄ O ₁₀	0.61
371	3.1x10 ⁵				0.53
375	2.0x10 ⁵				0.57

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 4 ^a Concentration: background subtracted peak concentration (in cm⁻³) at time c in Fig. 1b;
 5 ^b from Ehn et al. 2011;
 6 ^c Correlation was performed with the concentration of >20 nm particles for time period t=0-10 hr
 7 in Fig. 1b (t=0 was defined as the time when ozone was added to the chamber).

1 Table S3. Summary of all measured m/z of category III products for *EI*.

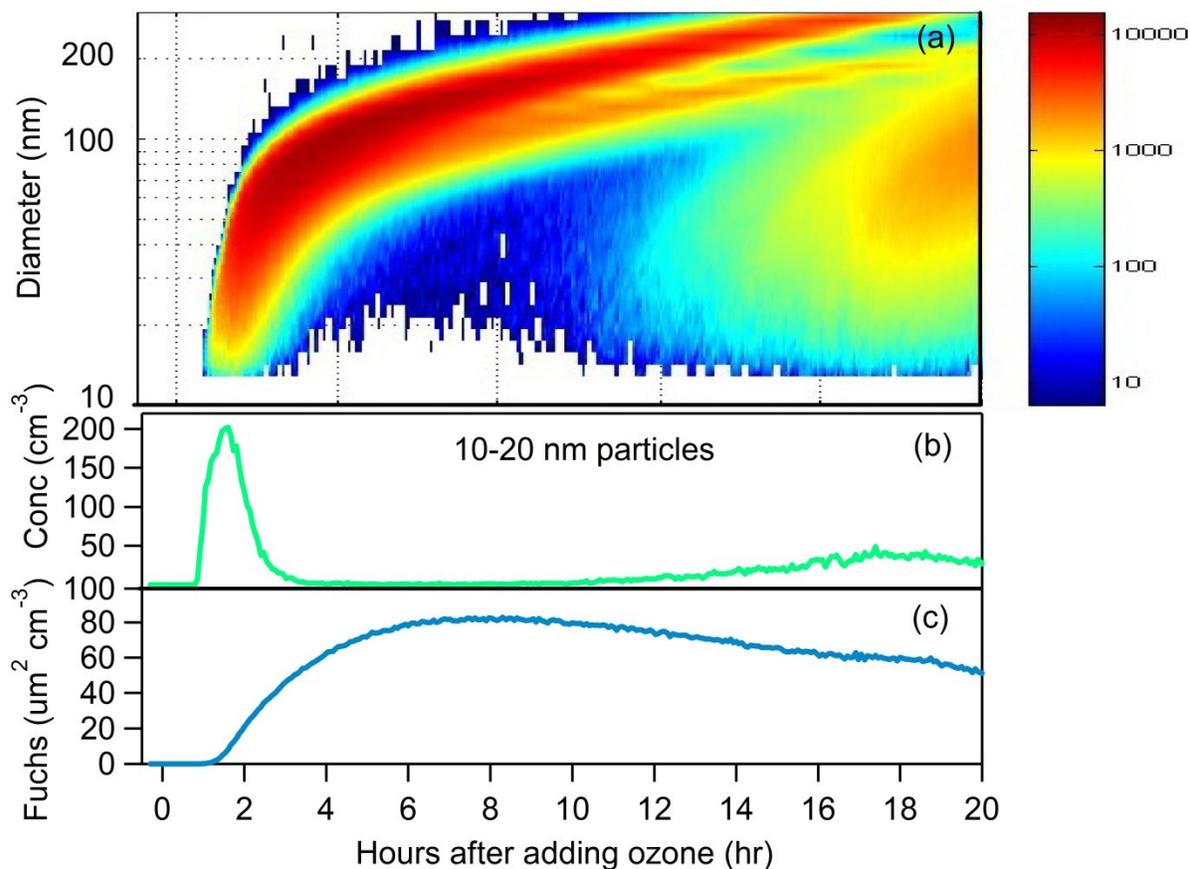
Ion (m/z)
152, 154, 160, 164, 165, 166, 167, 168, 178, 179, 180, 182, 183, 192, 194, 196, 198, 206, 208, 209, 210, 212, 213, 214, 217, 218, 220, 222, 223, 224, 225, 226, 227, 234, 235, 236, 238, 239, 244, 246, 248, 249, 250, 251, 252, 253, 254, 262, 264, 265, 266, 267, 268, 270, 277, 278, 279, 280, 281, 285, 292, 293, 294, 295, 296, 307, 311, 313, 314, 324, 325, 357

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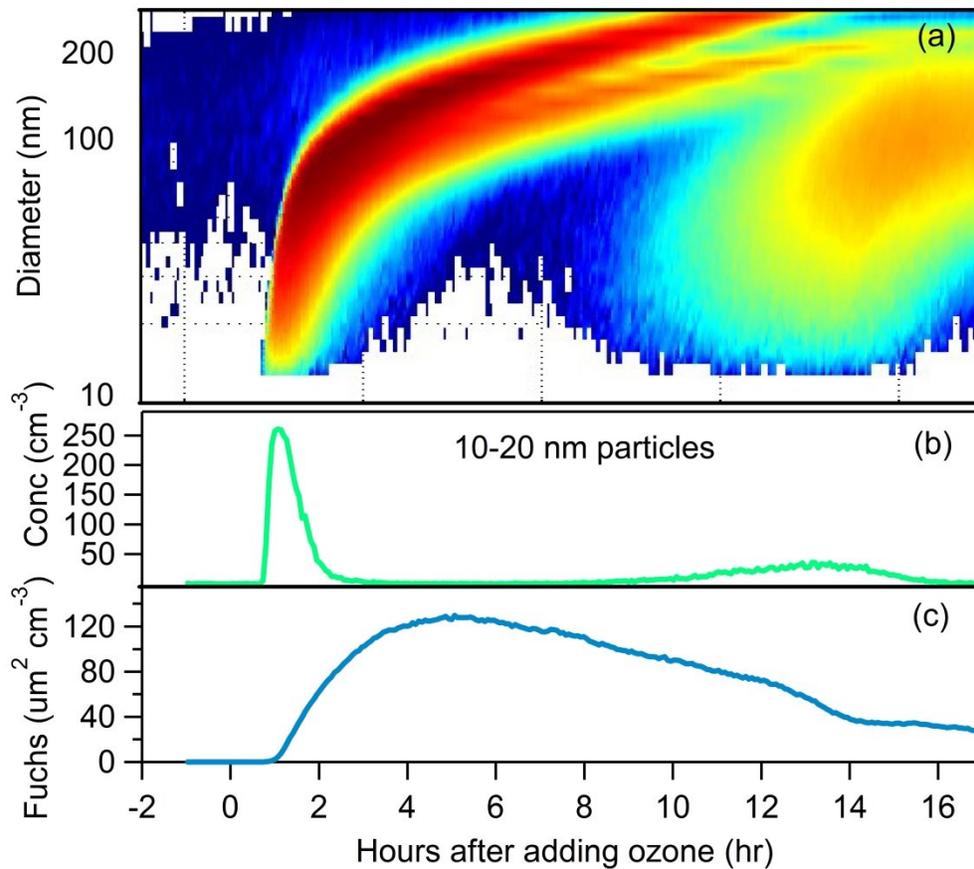


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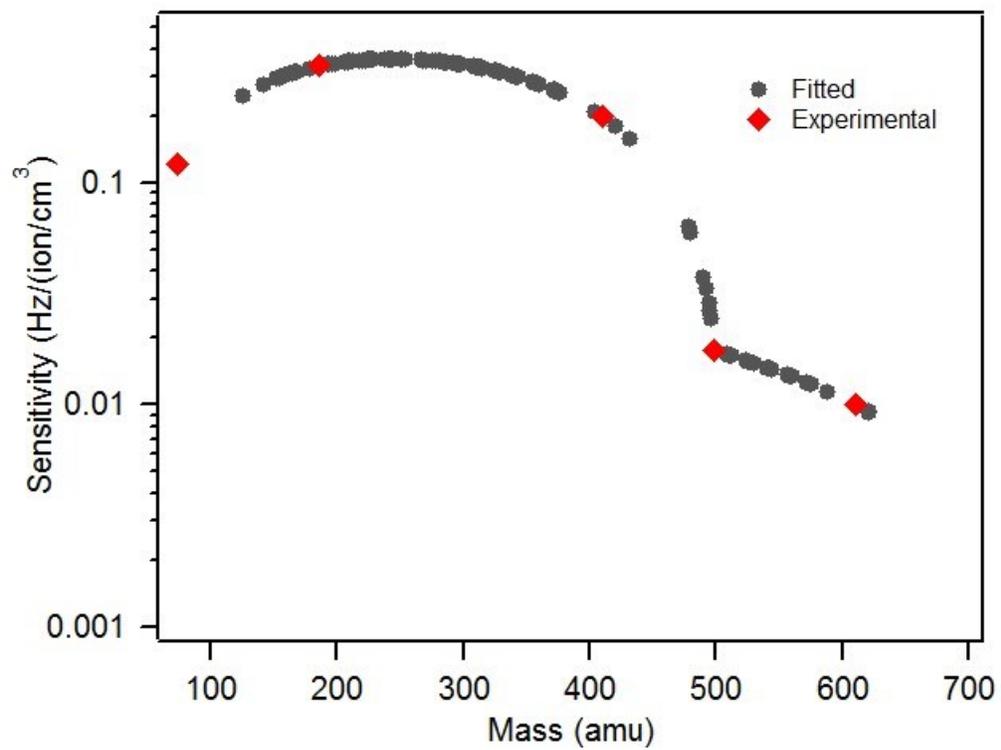
2 Fig. S1. Schematic diagram of the NCAR reaction chamber and the instruments used in this
 3 study.



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 2 Fig. S2. Characteristics of particle formation from chamber α -pinene ozonolysis for *E2*: (a)
 3 Contour plot of the particle size distribution measured with the SMPS in the 10-350 nm diameter
 4 range; (b) Temporal total concentration of 10-20 nm particles; (c) Estimated time-dependent
 5 Fuchs surface area (in $\mu\text{m}^2 \text{cm}^{-3}$).

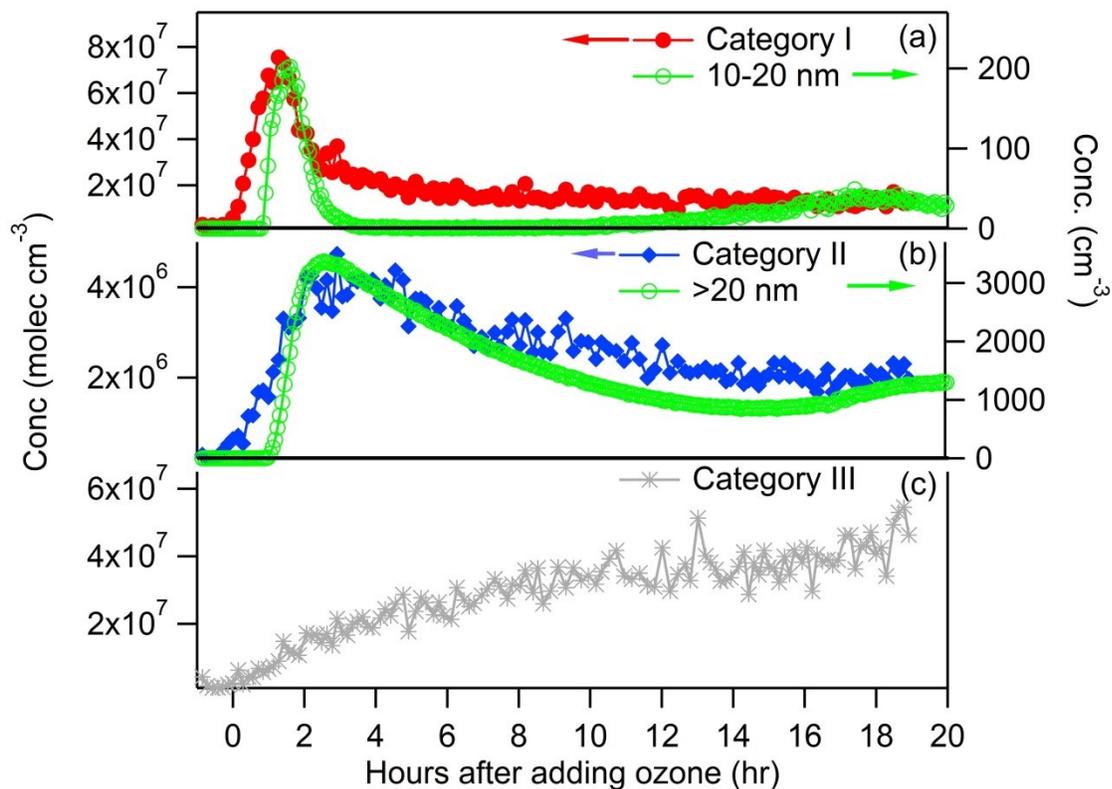


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 2 Fig. S3. Characteristics of particle formation from chamber α -pinene ozonolysis for *E3*: (a)
 3 Contour plot of the particle size distribution measured with the SMPS in the 10-350 nm diameter
 4 range; (b) Temporal total concentration of particles between 10 and 20 nm; (c) Estimated time-
 5 dependent Fuchs surface area (in $\mu\text{m}^2 \text{cm}^{-3}$).

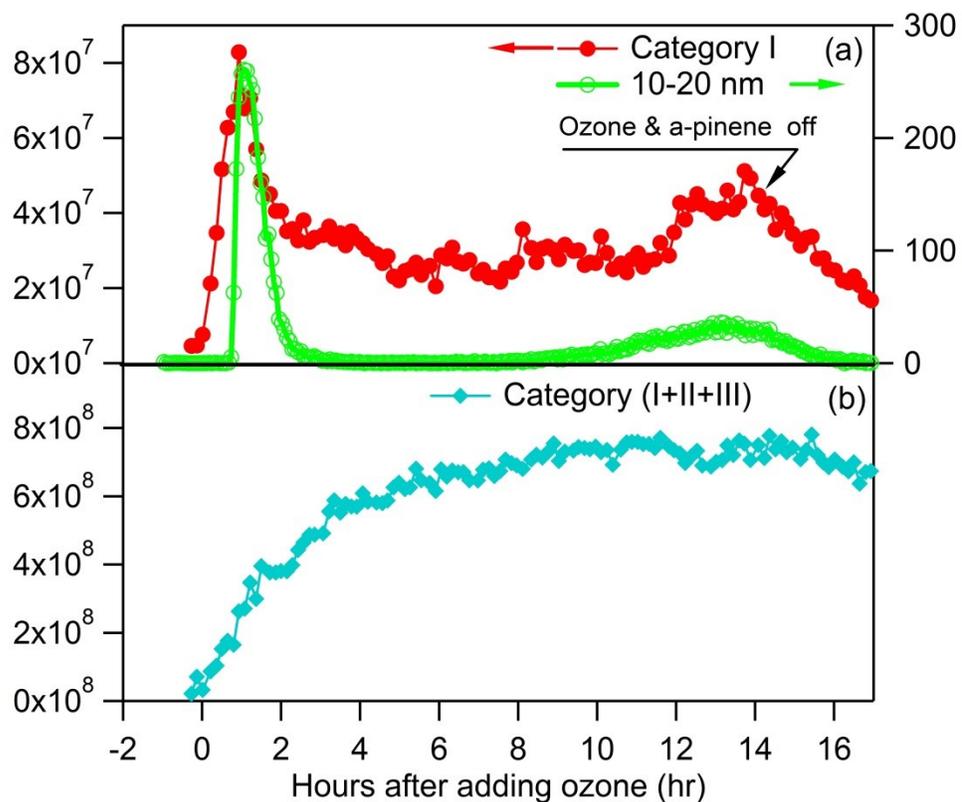


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2 Fig. S4. Mass-dependent sensitivities used for estimating concentrations of oxidation products
3 (Zhao et al., 2010).



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 2 Fig. S5. Time-dependent total concentrations of the category I-III products measured with the
 3 Cluster CIMS, along with the total concentrations of 10-20 nm particles and particles larger than
 4 20 nm measured with the SMPS for *E2*. (a) Category I products and 10-20 nm particles; (b)
 5 Category II products and particles larger than 20 nm; (c) Category III.



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2 Fig. S6. Time-dependent total concentrations of the category I-III products measured with the
 3 Cluster CIMS, along with the total concentrations of 10-20 nm particles measured with the
 4 SMPS for *E3*. (a) Category I products and 10-20 nm particles; (b) Category (I+II+III) products.

1 **Section II:** This section describes the method used to estimate the formation and growth rates of
 2 the first and second events for *E1-E3*.

3 1. Method:

4 The evolution of particle concentrations between 10 nm and 20 nm can be estimated from the
 5 following general dynamic equation (eq. 1) (Gelbard and Seinfeld, 1974).

$$6 \quad \frac{dN_{10-20}}{dt} = J_{10nm} - J_{20nm} + CoagSrc - CoagSnk \quad (1)$$

7 $\frac{dN_{10-20}}{dt}$ is the rate of change of particle concentrations between 10 and 20 nm. For *E1-E3*, this
 8 rate was assumed to increase linearly with time during the first half an hour when nucleation
 9 occurs. Thus it can be approximated as $\frac{\Delta N_{10-20}}{\Delta t}$, the concentration change divided by the time

10 (half an hour). J_{20nm} is the formation rate of particles at 20 nm, which can be expressed as

$$11 \quad \left. \frac{dN}{dDp} \right|_{20nm} \cdot GR.$$

12 GR is the modal growth rate at 20 nm, and is estimated with the method described by

13 Stolzenburg (2005). *CoagSrc* and *CoagSnk* are the coagulation source and coagulation sink

14 terms respectively. These two terms are calculated according to Kuang et al. (2012).

15 Rearranging eq. 1, the particle formation rate at 10 nm (J_{10nm}) can then be calculated by eq. 2.

$$16 \quad J_{10nm} \approx \frac{\Delta N_{10-20}}{\Delta t} + \left. \frac{dN}{dDp} \right|_{20nm} \cdot GR - CoagSrc + CoagSnk \quad (2)$$

17 2. Summary of particle formation rates and growth rates for *E1-E3*.

18 Table S4. Summary of formation rates and growth rates for *E1-E3*

Exp.	1st event		2nd event	
	Form. Rate (cm ⁻³ sec ⁻¹)	Growth Rate (nm hr ⁻¹)	Form. Rate (cm ⁻³ sec ⁻¹)	Growth Rate (nm hr ⁻¹)
E1	0.42	36	0.038	28
E2	0.28	16	0.026	12
E3	0.53	23	0.022	13

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1 **Section III:** This section describes how to estimate the minimum concentrations for growth of
2 particles at the measured growth rates for *E1-E3*.

3 The minimum concentration of a monomer N_1 required to grow particles at a certain rate can be
4 estimated by eq. 3. (Kuang et al., 2010).

$$5 \quad GR = \frac{1}{2} v_1 N_1 \bar{c} \quad (3)$$

6 Where GR is the diameter growth rate, v_1 is the molecular volume for monomer and \bar{c} is the
7 thermal velocity of the monomer. A molecular weight of 500 amu is assumed for Category I
8 products.

9 Table S5. Summary of minimum concentrations of a low molecular weight vapor (i.e. $N_1=N_{98\text{amu}}$)
10 or a high molecular weight vapor (i.e. $N_1=N_{500\text{amu}}$) required to grow particle at the observed
11 growth rates for *E1-E3*

Exp.	1st event		2nd event	
	$N_{98\text{amu}} (\text{cm}^{-3})$	$N_{500\text{amu}} (\text{cm}^{-3})$	$N_{98\text{amu}} (\text{cm}^{-3})$	$N_{500\text{amu}} (\text{cm}^{-3})$
E1	8.3×10^8	3.3×10^8	6.5×10^8	2.5×10^8
E2	3.7×10^8	1.4×10^8	2.8×10^8	1.1×10^8
E3	5.3×10^8	2.1×10^8	3.0×10^8	1.2×10^8

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14 References:

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