

Supplementary Material

Chemical Insights, Explicit Chemistry and Yields of Secondary Organic Aerosol from Methylglyoxal and Glyoxal

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The supporting information contains 10 pages with following information: the chemical model (Table S1); the simulated concentration of dissolved oxygen during an experiment (Fig. S1); atmospheric CSTR simulated SOA yields (Fig. S2).

Table S1. Reactions and rate/equilibrium constants used in the full kinetic model of unified glyoxal/methylglyoxal + OH

	Reactions	Rate constants (M ¹⁻ⁿ s ⁻¹)	Ref
1	H ₂ O ₂ → 2OH	1.1e-4×Trans ^a	T, e
2	OH + H ₂ O ₂ → HO ₂ + H ₂ O	2.7e7	T
3	HO ₂ + H ₂ O ₂ → OH + H ₂ O + O ₂	3.7	T
4	2 HO ₂ → H ₂ O ₂ + O ₂	8.3e5	T
5	OH + HO ₂ → H ₂ O + O ₂	7.1e9	T
6	HO ₂ + O ₂ ⁻ + H ⁺ → H ₂ O ₂ + O ₂	1e8	T
7	2OH → H ₂ O ₂	5.5e9	T
8	OH + O ₂ ⁻ → OH [·] + O ₂	1e10	T
9	O _{2g} ↔ O ₂	K _{eq} = 1.3e-3 k _r = 5.3e2	T, W
10	CO _{2g} ↔ CO ₂	K _{eq} = 3.4e-2 k _r = 5.3e2	T, W
11	CO ₂ ↔ H ⁺ + HCO ₃ ⁻	K _{eq} = 4.3e-7 k _r = 5.6e4	T
12	HCO ₃ ⁻ → H ⁺ + CO ₃ ⁻²	K _{eq} = 4.69e-11 k _r = 5.0e10	T
13	CO ₂ ⁻ + O ₂ → O ₂ ⁻ + CO ₂	2.4e9	T
14	HCO ₃ ⁻ + OH → CO ₃ ⁻ + H ₂ O	1e7	T
15	CO ₃ ⁻ + O ₂ ⁻ → CO ₃ ⁻² + O ₂	6.5e8	T
16	CO ₃ ⁻ + HCO ₂ ⁻ → HCO ₃ ⁻ + CO ₂ ⁻	1.5e5	T
17	CO ₃ ⁻ + H ₂ O ₂ → HCO ₃ ⁻ + HO ₂	8e5	T
18	GCOLAC + OH → GCOLAC* + H ₂ O	6.0e8	T
19	GCOLAC* + O ₂ → GCOLACOO*	1e6	G, L'
20	GCOLACOO* → GLYAC + HO ₂	5e1	C
21	2GCOLACOO* → 2GCOLACO* + O ₂	3e8*0.95	L', e
22	2GCOLACOO* → GLYAC + OXLAC + O ₂	3e8*0.05	L', e
23	GCOLACO* → HCO ₂ H + CO ₂	I	Gi, e
24	GCOLACO* → GLYAC*	1e7	Gi, e
25	GCOLAC ↔ H ⁺ + GCOLAC ⁻	K _{eq} = 1.48e-4 k _r = 2.0e10	T
26	GCOLAC ⁻ + OH → GCOLAC* + H ₂ O	6.0e8	T
27	GCOLAC* + O ₂ → GCOLACOO*	1e6	G, L'
28	GCOLACOO* → GLYAC ⁻ + HO ₂	5e1	C
29	2GCOLACROO* → 2GCOLACO* + O ₂	3e8×0.95	L', e
30	2 GCOLACROO* → GLYAC ⁻ + OXLAC ⁻ + O ₂	3e8×0.05	L', e
31	GCOLACO* → HCO ₂ H + CO ₂ ⁻	I	Gi, e
32	GCOLACO* → GLYAC ⁻	1e7	Gi, e
33	GLY + OH → GLY* + H ₂ O	1.1e9	T
34	GLY* + O ₂ → GLYOO*	1e6	G, L'
35	GLYOO* → GLYAC + HO ₂	5e1	C
36	2GLYOO* → 2*CHOHOH + 2CO ₂ + O ₂ + 2H ₂ O	3e8	L'
37	*CHOHOH + O ₂ → HCO ₂ H + HO ₂	5e6	G, L'
38	GLYAC + OH → GLYAC* + H ₂ O	3.62e8	T
39	GLYAC* + O ₂ → GLYACOO*	1e6	G, L'
40	GLYACOO* → OXLAC + HO ₂	5e1	C

41	2GLYACOO* → 2CO ₂ + 2COOH	3e8	L'
42	*COOH + O ₂ → CO ₂ + HO ₂	5e6	G, L'
43	GLYAC ↔ H ⁺ + GLYAC ⁻	K _{eq} = 3.47e-4 K _r = 2.0e10	T
44	GLYAC ⁻ + OH → GLYAC* + H ₂ O	1.28e7	T
45	GLYAC ⁻ + OH → GLYAC* + OH ⁻	2.9e9	T
46	GLYAC* + O ₂ → GLYACOO*	1e6	G, L'
47	GLYACOO* → OXLAC + HO ₂	1e2	C, L'
48	2GLYACOO* → 2CO ₂ + 2*COOH	3e8	L'
49	MGLY + OH → MGLY* + H ₂ O	7.0e8×0.92	T
50	MGLY + OH → *MGLY + H ₂ O	7.0e8×0.08	T
51	MGLY* + O ₂ → MGLYOO*	1e6	G, L'
52	MGLYOO* → PYRAC + HO ₂	5e1	C
53	2MGLYOO* → 2CO ₂ + 2CH ₃ CO ₂ H + O ₂	3e8	L'
54	*MGLY + O ₂ → *OOMGLY	1e6	G, L'
55	2*OOMGLY → 2*OMGLY + O ₂	3e8×0.95	L', e
56	2*OOMGLY → HOMGLY + OMGLY + O ₂	3e8×0.05	L', e
57	*OMGLY → HCHO + GLY*	I	Gi, e
58	*OMGLY → *HOMGLY	1e7	Gi, e
59	HOMGLY + OH → *HOMGLY + H ₂ O	4.10e7	M
60	*HOMGLY + O ₂ → *OOHOMGLY	1e6	G, L'
61	*OOHOMGLY → OMGLY + HO ₂	5e1	C
62	OMGLY + OH → *OMGLY + H ₂ O	6.17e9	M
63	*OMGLY + O ₂ → *OOOMGLY	5e1	C
64	GLY* + *CHOHOH → C3D	1.3e9	G, L'
65	2GLY* → C4D	1.3e9	G, L'
66	GLY* + *COOH → C3D	1.3e9	G, L'
67	GLYAC* + *COOH → C3D	1.3e9	G, L'
68	GLYAC* + *CHOHOH → C3D	1.3e9	G, L'
69	2GLYAC* → C4D	1.3e9	G, L'
70	GLYAC* + GLY* → C4D	1.3e9	G, L'
71	GLYAC* + GLY* → C4D	1.3e9	G, L'
72	GLYAC* + GLYAC* → C4D	1.3e9	G, L'
73	2GLYAC* → C4D	1.3e9	G, L'
74	GLYAC* + *COOH → C3D	1.3e9	G, L'
75	GLYAC* + *CHOHOH → C3D	1.3e9	G, L'
76	GLYCOL* ¹ + *CHOHOH → C3D	1.3e9	G, L'
77	GLYCOL* ¹ + GLY* → C4D	1.3e9	G, L'
78	GLYCOL* ¹ + *COOH → C3D	1.3e9	G, L'
79	GLYCOL* ¹ + GLYAC* → C4D	1.3e9	G, L'
80	GLYCOL* ¹ + GLYAC* → C4D	1.3e9	G, L'
81	GLYCOL* ² + *CHOHOH → C3D	1.3e9	G, L'
82	GLYCOL* ² + GLY* → C4D	1.3e9	G, L'
83	GLYCOL* ² + *COOH → C3D	1.3e9	G, L'
84	GLYCOL* ² + GLYAC* → C4D	1.3e9	G, L'
85	GCOLAC* + *CHOHOH → C3D	1.3e9	G, L'
86	GCOLAC* + GLY* → C4D	1.3e9	G, L'
87	GCOLAC* + *COOH → C3D	1.3e9	G, L'
88	GCOLAC* + GLYAC* → C4D	1.3e9	G, L'
89	GCOLAC* + GLYAC* → C4D	1.3e9	G, L'

90	GCOLAC* + GLYCOL* ¹ → C4D	1.3e9	G, L'
91	GCOLAC* + GLYCOL* ² → C4D	1.3e9	G, L'
92	GCOLAC* + GCOLAC* → C4D	1.3e9	G, L'
93	GCOLAC* + *CHOHOH → C3D	1.3e9	G, L'
94	GCOLAC* + GLY* → C4D	1.3e9	G, L'
95	GCOLAC* + *COOH → C3D	1.3e9	G, L'
96	GCOLAC* + GLYAC* → C4D	1.3e9	G, L'
97	GCOLAC* + GLYAC* → C4D	1.3e9	G, L'
98	GCOLAC* + GLYAC* → C4D	1.3e9	G, L'
99	GCOLAC* + GLYCOL* ¹ → C4D	1.3e9	G, L'
100	GCOLAC* + GLYCOL* ² → C4D	1.3e9	G, L'
101	GCOLAC* + GCOLAC* → C4D	1.3e9	G, L'
102	2 GCOLAC* → C4D	1.3e9	G, L'
103	2MGLY* → C6D	1.3e9	G, L'
104	MGLY* + *CHOHOH → C4D	1.3e9	G, L'
105	MGLY* + GLY* → C5D	1.3e9	G, L'
106	MGLY* + *COOH → C4D	1.3e9	G, L'
107	MGLY* + GLYAC* → C5D	1.3e9	G, L'
108	MGLY* + GLYAC* → C5D	1.3e9	G, L'
109	MGLY* + GLYCOL* ¹ → C5D	1.3e9	G, L'
110	MGLY* + GLYCOL* ² → C5D	1.3e9	G, L'
111	MGLY* + GCOLAC* → C5D	1.3e9	G, L'
112	MGLY* + GCOLAC* → C5D	1.3e9	G, L'
113	MGLY* + CH ₃ CO* → C5D	1.3e9	G, L'
114	MGLY* + *HOPYRAC → C6D	1.3e9	G, L'
115	2*HOPYRAC → C6D	1.3e9	G, L'
116	MGLY* + *HOPYRAC → C6D	1.3e9	G, L'
117	*HOPYRAC + *HOPYRAC → C6D	1.3e9	G, L'
118	*HOPYRAC + *HOPYRAC → C6D	1.3e9	G, L'
119	CH ₃ CO* + *HOPYRAC → C6D	1.3e9	G, L'
120	CH ₃ CO* + *HOPYRAC → C6D	1.3e9	G, L'
121	2LA* → C6D	1.3e9	G, L'
122	LA* + MGLY* → C6D	1.3e9	G, L'
123	LA* + *CHOHOH → C4D	1.3e9	G, L'
124	LA* + GLY* → C5D	1.3e9	G, L'
125	LA* + *COOH → C4D	1.3e9	G, L'
126	LA* + GLYAC* → C5D	1.3e9	G, L'
127	LA* + GLYAC* → C5D	1.3e9	G, L'
128	LA* + GLYCOL* ¹ → C5D	1.3e9	G, L'
129	LA* + GLYCOL* ² → C5D	1.3e9	G, L'
130	LA* + GCOLAC* → C5D	1.3e9	G, L'
131	LA* + GCOLAC* → C5D	1.3e9	G, L'
132	LA* + CH ₃ CO* → C5D	1.3e9	G, L'
133	2CH ₃ CO* → C4D	1.3e9	G, L'
134	LA* + *HOPYRAC → C6D	1.3e9	G, L'
135	LA* + *HOPYRAC → C6D	1.3e9	G, L'
136	OXLAC + OH → COOH + CO ₂ + H ₂ O	1.4e6	T
137	OXLAC ↔ H ⁺ + OXLAC ⁻	K _{eq} = 5.67e-2 k _r = 5.0e10	T
138	OXLAC ⁻ + OH → COOH + CO ₂ ⁻ + H ₂ O	2.0e7	T, L'

139	$\text{OXLAC}^- \leftrightarrow \text{H}^+ + \text{OXLAC}^{-2}$	$K_{\text{eq}} = 5.42\text{e-}5$ $k_r = 5\text{e}10$	T
140	$\text{OXLAC}^{-2} + \text{OH} \rightarrow * \text{COOH} + \text{CO}_2^- + \text{OH}^-$	4.0e7	T, L'
141	$\text{LA} + \text{OH} \rightarrow \text{LA}^* + \text{H}_2\text{O}$	4.3e8	H
142	$\text{LA}^* + \text{O}_2 \rightarrow \text{LAOO}^*$	1e6	G, L'
143	$\text{LAROO}^* \rightarrow \text{PYRAC} + \text{HO}_2$	5e1	C
144	$\text{LA} \leftrightarrow \text{LA}^- + \text{H}^+$	$K_{\text{eq}} = 1.38\text{e-}4$ $k_r = 5.0\text{e}10$	E&C
145	$\text{LA}^- + \text{OH} \rightarrow \text{LA}^* + \text{H}_2\text{O}$	3e8	B
146	$\text{LA}^* + \text{O}_2 \rightarrow \text{LAOO}^{*-}$	1e6	G, L'
147	$\text{LAOO}^{*-} \rightarrow \text{PYRAC}^- + \text{HO}_2$	5e1	C
148	$\text{PYRAC} + \text{OH} \rightarrow \text{PYRAC}^* + \text{H}_2\text{O}$	$6.0\text{e}7 \times 0.85$	T
149	$\text{PYRAC} + \text{OH} \rightarrow \text{CH}_3\text{CO}^* + \text{CO}_2 + \text{H}_2\text{O}$	$6.0\text{e}7 \times 0.15$	T
150	$\text{CH}_3\text{CO}^* + \text{O}_2 \rightarrow \text{CH}_3\text{C(O)OO}^*$	1e6	G, L'
151	$\text{CH}_3\text{C(O)OO}^* \rightarrow \text{CH}_3\text{CO}_2\text{H} + \text{HO}_2$	5e1	C
152	$2\text{CH}_3\text{C(O)OO}^* \rightarrow 2\text{CH}_3\text{C(O)O}^* + \text{O}_2$	3e8	L'
153	$\text{CH}_3\text{C(O)O}^* \rightarrow \text{CO}_2 + \text{HCHO}$	1e7	Gi
154	$\text{PYRAC}^* + \text{O}_2 \rightarrow \text{PYRACOO}^*$	1e6	G, L'
144	$2\text{PYRACOO}^* \rightarrow 2\text{PYRACO}^* + \text{O}_2$	$3\text{e}8 \times 0.95$	L', e
145	$2\text{PYRACOO}^* \rightarrow \text{HOPYRAC} + \text{OPYRAC} + \text{O}_2$	$3\text{e}8 \times 0.15$	L', e
146	$\text{PYRACO}^* \rightarrow \text{HCHO} + \text{GLYAC}^*$	I	Gi, e
147	$\text{PYRACO}^* \rightarrow * \text{HOPYRAC}$	1e7	Gi, e
148	$\text{HOPYRAC} + \text{OH} \rightarrow * \text{HOPYRAC} + \text{H}_2\text{O}$	3.6e8	H
149	$* \text{HOPYRAC} + \text{O}_2 \rightarrow * \text{OOHOPYRAC}$	1e6	G, L'
150	$* \text{OOHOPYRAC} \rightarrow \text{OPYRAC} + \text{HO}_2$	5e1	C
151	$\text{OPYRAC} + \text{OH} \rightarrow * \text{OPYRAC} + \text{H}_2\text{O}$	5e7	e
152	$* \text{OPYRAC} + \text{O}_2 \rightarrow * \text{OO(O)PYRAC}$	1e6	G, L'
153	$* \text{OO(O)PYRAC} \rightarrow \text{MOXLAC} + \text{HO}_2$	5e1	C
154	$\text{PYRAC} \leftrightarrow \text{PYRAC}^- + \text{H}^+$	$K_{\text{eq}} = 3.2\text{e-}3$ $k_r = 2\text{e}10$	T
155	$\text{PYRAC}^- + \text{OH} \rightarrow \text{PYRAC}^{*-} + \text{H}_2\text{O}$	$6.0\text{e}7 \times 0.95$	T
156	$\text{PYRAC}^- + \text{OH} \rightarrow \text{CH}_3\text{CO}^* + \text{CO}_2 + \text{OH}^-$	$6.0\text{e}7 \times 0.05$	T
157	$\text{PYRAC}^{*-} + \text{O}_2 \rightarrow \text{PYRACOO}^*$	5e1	C
158	$2\text{PYRACOO}^* \rightarrow 2\text{PYRACO}^{*-} + \text{O}_2$	$3\text{e}8 \times 0.95$	L', e
159	$2\text{PYRACOO}^{*-} \rightarrow \text{HOPYRAC}^- + \text{OPYRAC}^- + \text{O}_2$	$3\text{e}8 \times 0.05$	L', e
160	$\text{PYRACO}^{*-} \rightarrow \text{HCHO} + \text{GLYAC}^{*-} + \text{O}_2$	I	Gi, e
161	$\text{PYRACO}^{*-} \rightarrow * \text{HOPYRAC}^-$	1e7	Gi, e
162	$\text{HOPYRAC} \leftrightarrow \text{HOPYRAC}^- + \text{H}^+$	$K_{\text{eq}} = 3.2\text{e-}3$ $k_r = 2\text{e}10$	e
163	$\text{OPYRAC} \leftrightarrow \text{OPYRAC}^- + \text{H}^+$	$K_{\text{eq}} = 3.2\text{e-}3$ $k_r = 2\text{e}10$	e
164	$\text{HOPYRAC}^- + \text{OH} \rightarrow * \text{HOPYRAC}^- + \text{H}_2\text{O}$	2.6e9	H
165	$* \text{HOPYRAC}^- + \text{O}_2 \rightarrow * \text{OOHOPYRAC}^-$	1e6	G, L'
166	$* \text{OOHOPYRAC}^- \rightarrow \text{OPYRAC}^- + \text{HO}_2$	5e1	C
167	$\text{OPYRAC}^- + \text{OH} \rightarrow * \text{OPYRAC}^- + \text{H}_2\text{O}$	5e7	M
168	$* \text{OPYRAC}^- + \text{O}_2 \rightarrow * \text{OO(O)PYRAC}^-$	1e6	G, L'
169	$* \text{OO(O)PYRAC}^- \rightarrow \text{MOXLAC}^- + \text{HO}_2$	5e1	C
170	$\text{MOXLAC} + \text{OH} \rightarrow \text{GLYAC}^* + \text{CO}_2 + \text{H}_2\text{O}$	5.7e7	Gl
171	$\text{MOXLAC}^- + \text{OH} \rightarrow \text{GLYAC}^{*-} + \text{CO}_2 + \text{H}_2\text{O}$	7.85e7	e
172	$\text{MOXLAC}^{-2} + \text{OH} \rightarrow \text{GLYAC}^{*-} + \text{CO}_2 + \text{OH}^-$	1.0e8	H

173	$\text{MOXLAC} \leftrightarrow \text{MOXLAC}^{\cdot} + \text{H}^{+}$	$K_{\text{eq}} = 3.16e-3$ $k_r = 5e10$	H
174	$\text{MOXLAC}^{\cdot} \leftrightarrow \text{MOXLAC}^{\cdot 2} + \text{H}^{+}$	$K_{\text{eq}} = 1.5e-2$ $k_r = 5e10$	V
175	$\text{CH}_3\text{CO}_2\text{H} + \text{OH} \rightarrow * \text{CH}_2\text{CO}_2\text{H} + \text{H}_2\text{O}$	1.36e7	T
176	$\text{CH}_3\text{CO}_2\text{H} + \text{OH} \rightarrow \text{CO}_2 + \text{HCHO} + \text{HO}_2 + \text{H}_2\text{O}$	2.40e6	T
177	$* \text{CH}_2\text{CO}_2\text{H} + \text{O}_2 \rightarrow * \text{OOCH}_2\text{CO}_2\text{H}$	1e6	G, L'
178	$2 * \text{OOCH}_2\text{CO}_2\text{H} \rightarrow 2 * \text{OCH}_2\text{CO}_2\text{H} + \text{O}_2$	3e8*0.95	L', e
179	$2 * \text{OOCH}_2\text{CO}_2\text{H} \rightarrow \text{GLYAC} + \text{GCOLAC} + \text{O}_2$	3e8*0.05	L', e
180	$* \text{OCH}_2\text{CO}_2\text{H} \rightarrow 2\text{CO}_2 + 2\text{HCHO}$	I	Gi, e
181	$* \text{OCH}_2\text{CO}_2\text{H} \rightarrow \text{GCOLAC}^{\cdot}$	1e7	Gi, e
182	$\text{CH}_3\text{CO}_2\text{H} \leftrightarrow \text{CH}_3\text{CO}_2^{\cdot} + \text{H}^{+}$	$K_{\text{eq}} = 1.75e-5$ $k_r = 5.0e10$	T
183	$\text{CH}_3\text{CO}_2^{\cdot} + \text{OH} \rightarrow * \text{CH}_2\text{CO}_2^{\cdot} + \text{H}_2\text{O}$	7.23e7	T
184	$\text{CH}_2\text{CO}_2^{\cdot} + \text{OH} \rightarrow \text{CO}_2 + \text{HCHO} + \text{HO}_2 + \text{OH}^{\cdot}$	1.28e7	T
185	$* \text{CH}_2\text{CO}_2^{\cdot} + \text{O}_2 \rightarrow * \text{OOCH}_2\text{CO}_2^{\cdot}$	1e6	G, L'
186	$2 * \text{OOCH}_2\text{CO}_2^{\cdot} \rightarrow 2 * \text{OCH}_2\text{CO}_2^{\cdot} + \text{O}_2$	3e8*0.95	L', e
187	$2 * \text{OOCH}_2\text{CO}_2^{\cdot} \rightarrow \text{GLYAC}^{\cdot} + \text{GCOLAC}^{\cdot} + \text{O}_2$	3e8*0.05	L', e
188	$* \text{OCH}_2\text{CO}_2^{\cdot} \rightarrow 2\text{CO}_2^{\cdot} + 2\text{HCHO}$	I	Gi, e
189	$* \text{OCH}_2\text{CO}_2^{\cdot} \rightarrow \text{GCOLAC}^{\cdot}$	1e7	Gi, e
190	$\text{H}_2\text{O} \leftrightarrow \text{H}^{+} + \text{OH}^{-}$	$K_{\text{eq}} = 1.0e-14$ $k_r = 1.4e11$	T
191	$\text{HO}_2 \leftrightarrow \text{H}^{+} + \text{O}_2^{-}$	$K_{\text{eq}} = 1.6e-5$ $k_r = 5.0e10$	T
192	$\text{HCO}_2\text{H} + \text{OH} \rightarrow * \text{COOH} + \text{H}_2\text{O}$	1e8	T
193	$\text{HCO}_2^{\cdot} + \text{OH} \rightarrow \text{CO}_2^{\cdot} + \text{H}_2\text{O}$	2.4e9	T
194	$\text{HCO}_2\text{H} \leftrightarrow \text{H}^{+} + \text{HCO}_2^{\cdot}$	$K_{\text{eq}} = 1.77e-4$ $k_r = 5.0e10$	T
195	$\text{GLYAC} + \text{H}_2\text{O}_2 \rightarrow \text{HCO}_2\text{H} + \text{CO}_2 + \text{H}_2\text{O}$	0.3	T
196	$\text{PYRAC} + \text{H}_2\text{O}_2 \rightarrow \text{CH}_2\text{CO}_2\text{H} + \text{H}_2\text{O} + \text{CO}_2$	0.11	T
197	$\text{PYRAC}^{\cdot} + \text{H}_2\text{O}_2 \rightarrow \text{CH}_2\text{CO}_2^{\cdot} + \text{H}_2\text{O} + \text{CO}_2$	0.11	T
198	$\text{MOXLAC} + \text{H}_2\text{O}_2 \rightarrow \text{OXLAC} + \text{CO}_2 + \text{H}_2\text{O}$	0.5	T
199	$\text{MOXLAC}^{\cdot} + \text{H}_2\text{O}_2 \rightarrow \text{OXLAC}^{\cdot} + \text{CO}_2 + \text{H}_2\text{O}$	0.5	T
200	$\text{HCO}_2\text{H} + \text{OH} \rightarrow \text{COOH} + \text{H}_2\text{O}$	1e8	T
201	$\text{HCO}_2^{\cdot} + \text{OH} \rightarrow \text{CO}_2^{\cdot} + \text{H}_2\text{O}$	2.4e9	T
202	$\text{HCO}_2\text{H} \leftrightarrow \text{H}^{+} + \text{HCO}_2^{\cdot}$	$K_{\text{eq}} = 1.77e-4$ $k_r = 5.0e10$	T
203	$2 * \text{CHOHOH} \rightarrow \text{GLY}$	1.3e9	G, L'
204	$* \text{CHOHOH} + * \text{COOH} \rightarrow \text{GLYAC}$	1.3e9	G, L'
205	$2 * \text{COOH} \rightarrow \text{OXLAC}$	1.3e9	G, L'
206	$\text{C3D} \leftrightarrow \text{MA} + \text{H}_2\text{O}$	$K_{\text{eq}} = 1e5$ $k_r = 1e-8$	L'
207	$\text{MA} + \text{OH} \rightarrow \text{C3D}^{\cdot} + \text{H}_2\text{O}$	1.6e7	E
208	$\text{TA} + \text{OH} \rightarrow \text{C4D}^{\cdot} + \text{H}_2\text{O}$	3.1e8	M
209	$2 * \text{COOH} \rightarrow \text{OXLAC}$	1.3e9	G, L'
210	$\text{CO}_2^{\cdot} + * \text{COOH} \rightarrow \text{OXLAC}^{\cdot}$	1.3e9	G, L'
211	$2\text{CO}_2^{\cdot} \rightarrow \text{OXLAC}^{\cdot 2}$	1.3e9	G, L'
212	$\text{PYRAC}^{\cdot} \rightarrow 0.45\text{CH}_3\text{CO}_2^{\cdot}$ ^b	1e-4 ^b	C, e

^aTrans = Transmittance = $10^{-18.4 \times 0.80 \times [\text{H}_2\text{O}_2]}$; * = radical (e.g., glyoxal* = glyoxal radical); *ⁿ = radical type n (e.g., GLYCOLAC*¹ = glycolic acid radical type 1); O* (or *O) = alkoxy radical ; OO* (or *OO) = peroxy radical; C_nD = C_n dimer (e.g., C2D = C₂ dimer); X_g = X in the gas phase (e.g., O_{2g} = O₂ in the gas

phase); MGLY = methylglyoxal, PYRAC = pyruvic acid, GLYAC = glyoxylic acid, GLYCOL = glycolaldehyde, GLYCOLAC = glycolic acid, LA = lactic acid, MOXLAC = mesoxalic acid, OXLAC = oxalic acid; n = nth order; K_{eq} = the equilibrium constant (M), k_r = the reverse rate constant for corresponding K_{eq}. Thus, the forward rate constant can be calculated by K_{eq} × k_r; (g) = in the gas phase; I (= the decomposition rate constant from alkoxy radicals) = 5e6 s⁻¹ for ~10μM acetic acid/methylglyoxal, 8e6 s⁻¹ for ~10² μM acetic acid/methylglyoxal, and 2e7 s⁻¹ for ~10³ μM acetic acid/ 3.2e7 s⁻¹ for ~10³ μM methylglyoxal;^b PYRAC is assumed to photolyze to produce only 45% acetic acid with 5 times slower than the literature value (Carlton et al., 2006).

Reference

T = Tan et al., 2009, 2010 and 2012

G = Guzman et al., JPCA, 2006

C = Carter et al., JPC, 1979

H = Herrmann et al., AE, 2005

E = Ervens et al., PCCP, 2003

M = Monod et al., AE, 2005, 2008

L = Lim et al., EST, 2005

L' = Lim et al., ACP, 2010

W = Warneck, PCCP, 1999

E&C = Eyal and Canari, Ind. Eng. Chem. Res., 1995

B = Buxton et al., JPCRD, 1988

Gi = Gilbert et al., 1976 and 1981

V = Volgger et al., J. Chrom. A, 1997

e = Estimation by fitting

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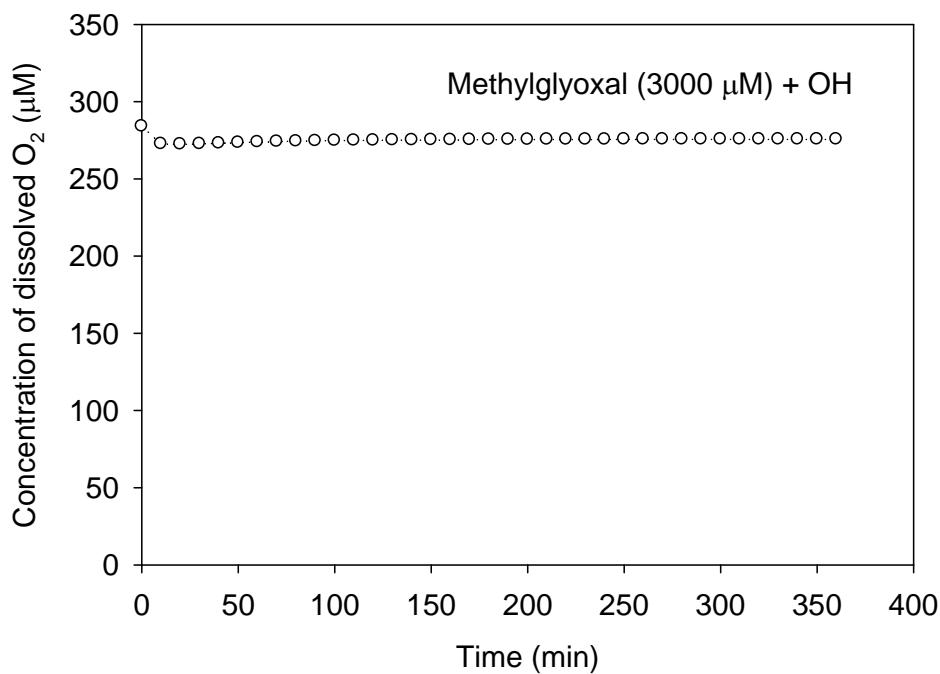


Figure S1. The simulated concentration of dissolved O_2 during the reaction of methylglyoxal (3000 μ M) + OH

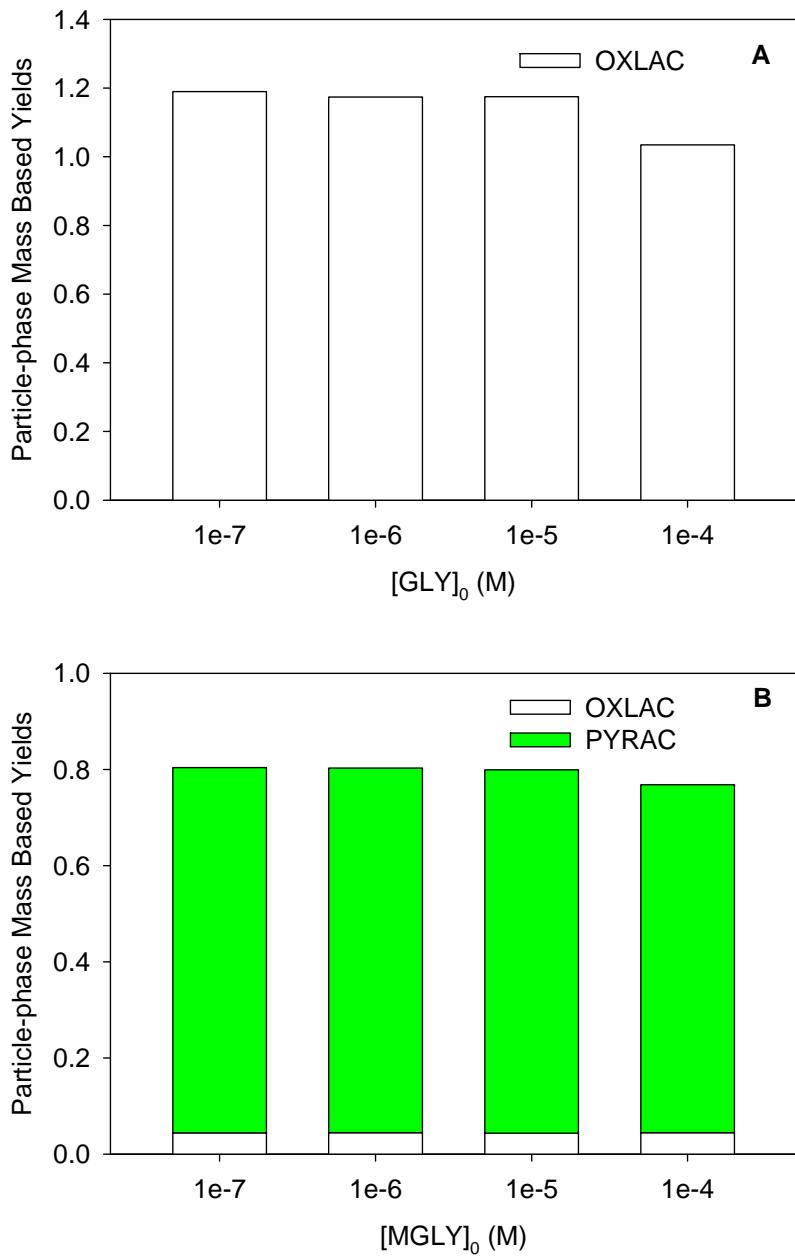


Figure S2. Atmospheric CSTR simulations (**A**) for particle-phase mass yields of oxalate (Y_{OXLAC}) with increasing initial concentrations of glyoxal ($[GLY]_0$) for aqueous-phase OH radical reactions ($Y_{OXLAC} = 1.19/(1+1450[GLY]_0)$; $Y_{SOA(GLY)} = Y_{OXLAC}$), and (**B**) for particle-phase mass yields of oxalate (Y_{OXLAC}) and pyruvate (Y_{PYRAC}) with increasing initial concentrations of methylglyoxal ($[MGLY]_0$) for aqueous-phase OH radical reactions($Y_{PYRAC} = 0.759/(1+495[MGLY]_0)$; $Y_{OXLAC} = 0.0439/(1-127[MGLY]_0)$); $Y_{SOA(MGLY)} = Y_{PYRAC} + Y_{OXLAC}$.