

Absorption and scattering properties of organic carbon

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Absorption and scattering properties of organic carbon vs. sulfate dominant aerosols at Gosan climate observatory in Northeast Asia

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Abstract

Carbonaceous and soluble ionic species of $PM_{1.0}$ and PM_{10} were measured along with the absorption and scattering properties and aerosol number size distributions at Gosan climate observatory (GCO) from January to September 2008. The daily averaged equivalent black carbon (EBC) measured as aerosol absorption exhibited two types of spectral dependence with a distinct maximum (peak) at either 370 nm or 880 nm, by which two subsets were extracted and classified into the respective groups (370 nm and 880 nm). The 370 nm group was distinguished by high organic carbon (OC) concentrations relative to elemental carbon (EC) and sulfate, but sulfate was predominant for the 880 nm group. The $PM_{1.0}$ OC of the 370 nm group was mainly composed of refractory and pyrolyzed components that correlated well with $PM_{1.0}$ EC₁, referred to as char EC, which suggests biofuel and biomass combustion as the source of these OC fractions, particularly during winter. The scanning electron microscope (SEM) images and the number size distributions implied that aerosols of the 370 nm group were externally mixed upon transport in fast-moving air masses that passed through the Beijing area in about one day. In contrast, the aerosols of the 880 nm group were characterized by high sulfate concentrations, and seemed to be internally mixed during slow transport over the Yellow Sea region over approximately two to four days. The absorption and scattering coefficients of the 880 nm group were noticeably higher compared to those of the 370 nm group. The average absorption ångström exponent (AAE) was estimated to be 1.29 and 1.0 for the 370 nm and 880 nm groups, respectively, in the range 370–950 nm. These results demonstrated that the optical properties of aerosols were intimately linked to chemical composition and mixing state, characteristics determined both by source and atmospheric aging processes. In OC dominant aerosols, absorption was enhanced in the UV region, which was possibly due to refractory and pyrolyzed OC compounds. Under sulfate dominant conditions, the sulfate coating on BC particles contributed to the absorption of the longer visible light. Consequently, single scattering albedo (SSA) was higher for the 880 nm group than for the

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370 nm group, emphasizing that the relative abundances of absorbing and scattering constituents are also important in estimating the climate effect of aerosols.

1 Introduction

Atmospheric aerosols play an important role in climate change by altering the global radiation balance both directly and indirectly, and the relative importance of these processes depends on the chemical composition and size distribution of aerosols (Forster et al., 2009). Carbonaceous and sulfate particles are the two most important anthropogenic aerosol constituents influencing climate (Chuang et al., 1997; Haywood and Ramaswamy, 1998; Schult et al., 1997). A positive component of this radiative forcing is largely due to black carbon (BC) produced from pyrolysis during incomplete combustion of biomass and fossil fuels (Ramanathan and Carmichael, 2008; Roessler and Faxvog, 1979). The magnitude of the direct radiative forcing from BC exceeds that due to CH₄, suggesting that, BC may be the second most important component of global warming after CO₂, in terms of direct forcing (Bond et al., 2013).

However, the definition and measurement techniques for atmospheric BC (commonly referred to as soot) and elemental carbon (EC) have long been subjects of scientific controversy. In general, BC and EC are identified by their different analytical methods: BC is measured using its optical properties, with the main property being strong absorption of visible light with a mass absorption cross section (MAC) at a wavelength of $\lambda = 550 \text{ nm}$ above $7.5 \pm 1.2 \text{ m}^2 \text{ g}^{-1}$ for freshly produced particles (Bond et al., 2013). In contrast, EC is usually determined by thermal methods based on its chemical properties, and is defined as the carbonaceous fraction of particulate matter that can only be gasified by oxidation starting at temperatures above 340 °C (Petzold et al., 2013). Among the types of BC, the BC derived from the light-absorption coefficient (σ_{ap}) using a MAC is called equivalent BC (EBC). Given that this measurement technique detects particles that absorb light, EBC measurements could include other absorbing particles and possibly lead to slightly different mass concentrations than EC measure-

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ments, even though they should theoretically be the same. The light absorption of BC is generally considered to be weakly dependent on wavelength (Bergstrom et al., 2002). Conversely, other light-absorbing aerosol components such as organics and dust show stronger absorption spectral dependence that contributes to absorption at shorter wavelengths (Bergstrom et al., 2007; Kirchstetter, 2004). Recent studies have reported that light-absorbing organic carbon that is not black (“brown carbon”), tends to absorb light strongly in the blue and ultraviolet (UV) spectral regions (Andreae and Gelencsér, 2006; Jacobson, 1998, 1999). In the atmosphere, fresh BC ages as it is transported, leading to mixing with other aerosol components such as organics, minerals, and sulfates (Hasegawa and Ohta, 2002). It has been found that organic or sulfate coating around an inner BC core enhances the light-absorption of BC by a “lensing” effect (Cross et al., 2010; Lack et al., 2009). Thus, the presence of light-absorbing organics and aerosols, either as intrinsic components of or as coatings on the BC particles makes it imperative to reassess and redefine BC and other light-absorbing carbonaceous matter in the atmosphere (Andreae and Gelencsér, 2006).

In regions influenced by various emissions from deserts, oceans, biomass burning, and human and industrial activities, aerosol chemical composition and optical properties are extremely complicated. Therefore, it is critical to understand the link between chemical composition and optical properties, particularly for policy making related to climate change. East Asia is a typical example of regions described above and is one of the strongest BC emission source regions (Bond et al., 2004). Here, we examined light-absorbing and scattering properties of aerosols in relation to their major components including carbonaceous and soluble ions using data obtained at GCO located in northeast Asia.

2 Measurements

The chemical compositions of $PM_{1.0}$ and PM_{10} were measured along with aerosol optical properties and particle number distributions from August 2007 to September 2008

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at GCO (33.17° N, 126.10° E, 70 m a.s.l.) on Juju island, which is known as an ideal location for monitoring Asian outflows. For this study, a subset of 29 samples was chosen from January to September 2008, for which the measurements of chemical composition, aerosol absorption and scattering, and number size distribution were all available.

The sample set included five samples from January, three from February, four from March, eight from April, four from May, one from June, two from August, and two from September 2008. Details of the sampling procedures and methods can be found in Lim et al. (2012).

For PM_{1.0} and PM₁₀, sampling was conducted once every six days, starting at 09:00 LST and lasting for 24 h. Eight inorganic species were determined by using ion chromatography (Dionex 4500, Dionex, USA), and EC and organic carbon (OC) were quantified by the Interagency Monitoring of Protected Visual Environments (IMPROVE) thermal/optical reflectance (TOR) protocol, which produced four OC fractions: OC1, OC2, OC3, and OC4 at temperatures of 120, 250, 450, and 550 °C, respectively, in a non-oxidizing He atmosphere; and three EC fractions: EC1, EC2, and EC3 at 550, 700, and 800 °C, respectively, in an oxidizing atmosphere of 2 % O₂ in He. In the He mode, a fraction of OC was pyrolyzed (OP) and this evolved in the oxidizing atmosphere until the reflected light achieved its initial value. OP is assumed to be originally organic carbon, so OC is considered to be the sum of the four OC fractions and the OP fraction, and OP is subtracted from the sum of EC fractions.

Light-absorption coefficients (σ_{ap}) were obtained every 10 min from the seven wavelength aethalometer measurements (AE-31, Magee scientific Corp., USA) and EBC mass concentrations were derived with the “Magee BC” calibration factor (Hansen, 2005). The aethalometer collects the aerosol sample on a quartz fiber filter tape and measures the light attenuation of samples at wavelengths covering the UV to the near-IR ranges (i.e., 370, 470, 520, 590, 660, 880, and 950 nm). Similar to other filter-based absorption instruments, the aethalometer needs to be corrected for scattering by the fiber filter substrates, scattering of aerosols embedded in the filter, and filter loading by accumulation of light absorbing particles (Collaud Coen et al., 2010). In this study, σ_{ap}

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was determined by following the correction process presented by Arnott et al. (2005). The absorption ångström exponent (AAE) was determined by statistical regression to fit the σ_{ap} data with a power law equation. The MAC of EBC along wavelengths was calculated by dividing the light absorption coefficient by EC mass concentration. The scattering coefficient was obtained from a nephelometer (model 3563, TSI Inc., USA) every 10 min at three wavelengths: 450 nm, 550 nm, and 700 nm, which were corrected to STP and for truncation according to Anderson and Ogren (1998). The absorption and scattering data were averaged daily for comparison with chemical compositions that were measured daily. Aerosol number size distributions from 10.4 nm to 469.8 nm in diameter were measured every 10 min using a scanning mobility particle sizer (SMPS, model 3034, TSI Inc., USA). The scanning electron microscope (SEM) image was obtained from quartz filters laden with $PM_{1.0}$ after being coated with Pt-Pd.

3 Results and discussion

3.1 Spectral dependence of absorption and chemical characteristics

We compared daily averaged EBC concentrations with $PM_{1.0}$ EC concentrations and examined the spectral dependence of EBC. The EC concentrations of $PM_{1.0}$ agreed well with EBC at 880 nm, which corresponds to 92 % of $PM_{1.0}$ EC. At other wavelengths, the EBC concentrations were less than 90 % of $PM_{1.0}$ EC. The daily averaged EBC concentrations varied along wavelengths, some of which showed clear maxima either at 370 nm or at 880 nm. The EBC concentrations of the others did not significantly change along wavelengths, but tended to be slightly higher at 660 nm. While the maxima at 370 nm were apparent regardless of EBC concentration, they were less distinguished at 880 nm or 660 nm as EBC concentration decreased. Thus, we only considered EC concentrations higher than mean for the classification of the 880 nm maximum. Based on the wavelength peak of EBC, five samples were selected for both the 370 nm and 880 nm groups. Then, chemical characteristics of each group were

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compared for major constituents, including EC, OC, sulfate, and inorganic salts. In particular, their relative abundance was distinguished between the two groups such as OC/EC, OC/sulfate, and EC/sulfate ratios. It is noteworthy that all three ratios were higher for the 370 nm group than for the 880 nm group. While the 370 nm group was characterized by relatively higher OC concentration, sulfate was predominant for the 880 nm group. However, there was one exception in the 880 nm group (9 January 2008), in which the OC/EC ratio was high with relatively low ratios of OC/sulfate and EC/sulfate. Its maximum absorption was also not as distinct as the other four in the group. Thus, it was excluded and the other four samples were finally classified as the 880 nm group. The five samples of the 370 nm and the four samples of the 880 nm group were collected in winter and spring, respectively (Table 1).

Figure 1a shows the EBC concentrations relative to $PM_{1.0}$ EC concentrations along wavelengths. Although the concentrations of EBC and $PM_{1.0}$ EC were expected to be equal, they were found to be slightly different. The EBC/ $PM_{1.0}$ EC ratio varied from 0.73 to 1.04 and from 0.76 to 1.18 for the 370 nm and 880 nm groups, respectively. For the former, the EBC/ $PM_{1.0}$ EC ratio was the highest at 370 nm and decreased toward 950 nm by 0.07 ~ 0.18. For the latter, the EBC/ $PM_{1.0}$ EC ratio was at its minimum at 370 nm and gradually increased by 0.15 ~ 0.22 to its maximum at 880 nm. The chemical characteristics of the two groups are summarized in Table 1. While EBC concentrations were lower for the 370 nm group compared to those of the 880 nm group, OC concentrations were similar in the two groups. The difference in meteorological variables and precursor gases reflect seasonal characteristics of each group.

3.2 Optical properties in relation to chemical composition

3.2.1 Absorption in organic carbon dominant regime

The OC/EC ratio of $PM_{1.0}$ was higher for the 370 nm group (2.68) than for the 880 nm group (1.81) (Fig. 1b). The average OC/EC ratio of the total sample set was 2.26. For the 370 nm group, the contribution of OC3, OC4, and OP to $PM_{1.0}$ OC was noticeable.

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The higher PM_{1.0} OC/EC ratio of the 370 nm group was mostly due to the greater contribution of refractory OC, including OC3 and OC4 that are resistant to volatilization and evolve at higher temperatures (450 and 550 °C), and OP. The average ratio of PM_{1.0} (OC3 + OC4 + OP)/EC for the 370 nm group (2.00) was higher than that of the 880 nm group (1.20) and the total sample set (1.51). The average PM_{1.0} OP/EC was also greater for the 370 nm group (0.87) than the 880 nm group (0.53).

In this study, the larger fraction of PM_{1.0} refractory OC and OP against EC (e.g., (OC3 + OC4 + OP)/EC and OP/EC in Fig. 1b) was associated with the enhanced ratio of EBC at 370 nm to EC, which implied the strong light-absorption of aerosols in the UV region. It has been reported that brown carbon shows steeper wavelength dependence and has various origins in the atmosphere, e.g., soil humics, humic-like substances (HULIS), tarry materials from combustion, and bioaerosols (Andreae and Gelencsér, 2006 and references therein). These components are considered to be the class of HULIS (Hoffer et al., 2006; Lukács et al., 2007) which is associated with the water soluble organic carbon (WSOC) fraction with high molecular weight (Miyazaki et al., 2007; Yu et al., 2004). In addition, it has been reported that the WSOC fraction of ambient aerosols is prone to charring, and it accounts for a significant fraction of overall OP (Andreae and Gelencsér, 2006; Yu et al., 2002). In the recent study by Kondo et al. (2011) that evaluated the charring of four kinds of laboratory HULIS, more than half evolved as the OC fraction resistant to volatilization in the He mode and released in He/O₂ mixtures. Clarke et al. (2007) also argued that the measured refractory OC, for which they used a density of bulk HULIS, remaining at high temperature (400 °C) in thermal analysis was responsible for enhanced shortwave absorption in biomass burning plumes. These studies support the observation that the refractory OC lead to more charring in the He mode and, subsequently, to enhanced OP fractions. In this study, a direct link between refractory OC and HULIS or other light-absorbing organics was not confirmed. However, the higher contribution of refractory OC and OP to total OC was one of the main attributes of the 370 nm group that showed distinctively enhanced absorption at the UV region.

strong winds. Therefore, the coarse particles of the two groups were possibly derived from different sources such as salt deposits and soil minerals, contributing to scattering for the 370 nm group and to absorption for the 660 nm group. In addition, the local emission of these inorganic salts should not be disregarded for all samples because

Gosan is located right by the sea.

For the 370 nm group, the inorganic salts of PM_{10} ($Na^+ + Ca^{2+} + Mg^{2+} + Cl^-$) showed a good correlation with the light-scattering coefficient at 550 nm ($R^2 = 0.85$) (Fig. 2a). In addition, PM_{10} OC was well correlated with the scattering coefficient, while $PM_{1.0}$ OC showed a weak correlation. Thus, the coarse OC possibly originated from salt deposits in dry lakes, and it could be agglomerated with inorganic salts. It is thought that the chemical effect on light scattering is outweighed by the size effect of the aerosols, although sulfate is the most efficient light-scattering single species (Tang, 1996). Moreover, when one of the aerosol components is much larger than the others, it dominates the total optical characteristics of the mixture, especially for semi-externally or externally mixed particles (Mishchenko et al., 2004).

In the 370 nm group, three types of particles recognized by their distinct shapes and sizes seemed to be mixed externally (Fig. 3a). One type consisted of agglomerates that were more irregular in size and shape than soot agglomerates (Fig. 3b) and were thought to be organic particles coating soot (Adachi and Buseck, 2008; Adachi et al., 2010). It is likely that soot particles, being relatively less aged, increased in size through organic coatings under high OC/EC and OC/sulfate conditions. Two other distinct particle shapes were a thin cube and a sphere, thought to be a sea-salt particle and a tar-ball, respectively. In particular, a tar-ball is considered to possibly be brown carbon (Alexander et al., 2008) as well as an indicator of biomass combustion (Pósfai et al., 2004). It is distinguished in shape and structure from soot aggregates, typically made up of spherules 20 to 60 nm in diameter, often with an internal structure of curved graphene-like layers (Fig. 3b) (Faeth and Köylü, 1995; van Poppel et al., 2005; Vander Wal and Tomasek, 2004). The air masses of the 370 nm group underwent fast transport, passing through the Beijing area in about one day, and as a result, inorganic

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particles being externally mixed with EC and organic aerosol might have been able to play a part in light scattering under low sulfate conditions (Fig. 1c and d).

The majority of samples classified into the 370 nm, 880 nm, or 660 nm groups were collected under the influence of Chinese outflows in winter and spring and thus carbonaceous compounds were mainly associated with coal and biomass combustion for residential heating (Lu et al., 2011). Their study pointed out biofuel from residential sector as a predominant source of OC, especially in winter. In our study, for the 370 nm and 660 nm groups, the concentrations of PM_{1,0} EC₁, referred to as char EC (Han et al., 2010; Lim et al., 2012) were well correlated with the sum of OC₃, OC₄, and OP in PM_{1,0} ($R^2 = 0.6$), whereas this was not the case ($R^2 < 0.1$) in the 880 nm group. The char EC is known to be produced from incomplete combustion at lower temperatures; e.g., biofuel combustion (Han et al., 2010). This implies that the major source of PM_{1,0} OC for the 370 nm and 660 nm groups were biofuel combustion. However, the enhanced absorption of the short wavelengths was observed only for the 370 nm group. This optical trait of the 370 nm group would be intimately coupled with chemical characteristics such as high ratios of OC/EC, (OC₃ + OC₄ + OP)/EC, OP/EC, OC/sulfate, EC/sulfate in PM_{1,0}, in conjunction with meteorological conditions such as low temperature and high wind speed causing aerosols to be less aged and externally mixed. This also represents favorable conditions for semi-volatile organics to be partitioned into the particle phase. In particular, the number distributions of the 370 nm group showed a distinct peak between 100 and 200 nm in SMPS measurements (Fig. 4a and b), which is bigger than freshly produced urban soot particles (< 100 nm) and a size that has been suggested for brown carbon (100–200 nm) (Schwarz et al., 2008). The agglomerates shown in Fig. 3a could comprise refractory OC or OP as a main component of OC and account for the peak distribution at 100–200 nm for the 370 nm group. In particular, the ratio of OP to OC was high in PM_{1,0} at GCO (Lim et al., 2012). Therefore, the PM_{1,0} refractory OC and OP of the 370 nm group are most likely to be responsible for enhanced light absorption in the UV region and to be a constituent part of brown carbon.

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3.2.2 Scattering in sulfate dominant regime

In contrast to the 370 nm group, the 880 nm group was distinguished by higher sulfate concentrations, which resulted in lower ratios of EC/sulfate and OC/sulfate in PM_{1.0} (Fig. 1b). The average concentrations of PM_{1.0} sulfate were 3.25 μg m⁻³, 9.83 μg m⁻³, and 5.16 μg m⁻³ for the 370 and 880 nm groups and the total sample set, respectively. The average ratios of EC/sulfate and OC/sulfate in PM_{1.0} were 0.78 and 2.20 for the 370 nm group; 0.23 and 0.42 for the 880 nm group; and 0.37 and 0.90 for the total sample set. The high sulfate concentrations were coupled with stagnant conditions over the Yellow Sea, which supplies sufficient humidity and time for sulfate conversion in Chinese outflows (Lim et al., 2012) (Table 1).

In general, scattering coefficients at 520 nm were well correlated with sulfate concentrations of PM_{1.0}, which was more pronounced in the 880 nm group ($R^2 = 0.90$; Fig. 2b). Although the scattering coefficients and sulfate concentrations of the 880 nm group were greater than those of the 370 nm group, the scattering coefficient per sulfate mass (slope in Fig. 2b) was less for the 880 nm group. This result is in accordance with the findings of previous studies in which BC incorporated or randomly positioned within the sulfate aerosol can lead to absorption enhancement and reduce the expected direct cooling effect due to sulfate (Bond et al., 2013; Chýlek et al., 1995; Fuller et al., 1999; Martins et al., 1998). The mixing of BC with non-absorbing materials such as sulfate has been known to alter the scattering properties of aerosols (Bond et al., 2006). For the 880 nm group, BC was internally mixed with sulfate during the aging processes, which modified the absorption properties of the aerosols and resulted in spectral dependence of EBC (Fig. 1a).

In the present study, it was not determined if the BC core was surrounded by a well-mixed shell including sulfate or simply incorporated into other components. However, the daily variation of the particle size distribution (top panels in Fig. 4c and d) suggested some characteristics of the mixing state of the aerosols. For the 880 nm group, the mode and number concentration remained unchanged through the day, which im-

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plies that aerosols were internally mixed (top panels in Fig. 4c and d). In general, the atmospheric BC of remote areas is often internally mixed with other materials such as sulfate (Clarke et al., 1997; Pósfai et al., 1999) through intensive processing due to longer residence times (Hasegawa and Ohta, 2002). The air masses classified into the 880 nm group were slowly transported over the Yellow Sea over two to four days (bottom panels in Fig. 4c and d), during which BC particles must have been mixed with other types of aerosols such as sulfate-forming BC coatings or intrinsic components.

3.3 Chemical composition in relation to climate effect of aerosol

3.3.1 AAE in relation to organic carbon and sulfate

The change in EBC concentrations with wavelength (Fig. 1a) is related to the intrinsic properties of BC types such as composition, size, mixing state, and source. AAE, denoting the spectral dependence of light-absorption, was derived for each group by fitting the measured σ_{ap} at seven wavelengths from 370 nm to 950 nm with a power law equation (Fig. 5a). While the AAE of the 370 nm group was 1.29 (1.24–1.4) for the averaged light-absorption coefficients ranging from 19.97 Mm^{-1} to 5.72 Mm^{-1} , the AAE of the 880 nm group was close to 1.0 (0.95–1.05), the theoretical value of BC, with higher light-absorption coefficients from 22.69 Mm^{-1} to 8.75 Mm^{-1} .

In several laboratory and field studies (Kirchstetter, 2004; Schnaiter et al., 2003, 2005), it was observed that BC produced from high-temperature combustion processes (e.g., diesel combustion) has a low spectral dependence with $\text{AAE} \approx 1.0$, whereas BC from low-temperature combustion (e.g., biomass burning) exhibits a much stronger spectral dependence with $\text{AAE} > 2.0$. Moreover, organic aerosols were reported to contribute to strong light-absorption at shorter wavelengths (i.e., UV), leading to higher AAE (Dubovik et al., 1998; Kirchstetter, 2004; Schnaiter et al., 2005). Kirchstetter (2004) demonstrated that in addition to BC, OC contributed considerably to measured light absorption in the UV and visible spectral regions. Theoretical and measurement studies also revealed that coating with scattering aerosol increased the absorption of

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visible light (Chung et al., 2012; Lack and Cappa, 2010; Schnaiter et al., 2005). Coating BC with scattering materials such as sulfate was reported to increase (Lack and Cappa, 2010) or decrease AAE to less than one (Schnaiter et al., 2005). Lack and Cappa (2010) pointed out that AAE was 1 for sufficiently small BC (~ 10 nm) and was subject to change by techniques and analysis methods, thereby suggesting that AAE estimated from measurements should be discussed in view of the analysis method.

The average AAE of the 370 nm group was 1.29, which was higher than the diesel BC AAE (~ 1.0) but lower than biomass burning BC AAE (~ 2.0). As discussed above, these samples were influenced by the Beijing plumes, and open biomass burning events were not detected. Consequently, we consider this group to be OC-rich aerosols transported from densely populated areas. In fact, our results lie in the upper limit of AAE (1.0–1.5) derived from OC rich pollution cases at GCO (Lee et al., 2012).

The AAE of the 880 nm group had a narrow range between 0.95 and 1.05 with a mean of 1.0, which was measured in regions close to the source of BC emitted from high temperature combustion and where externally mixed BC dominates absorption (Bond et al., 2013; Kirchstetter, 2004; Schnaiter et al., 2003). AAE from measurements was found to be ~ 1.0 near an urban roadway (Kirchstetter, 2004) and 0.9–1.3 in samples from urban plumes (Gyawali et al., 2009). In a previous study performed at GCO, AAE of 0.8–1.5 were derived for aerosols during sulfate rich pollution (Lee et al., 2012), which is in accordance with our result for the 880 nm group. At GCO, sulfate rich aerosols are often found in aged air masses. Therefore, our AAE of 1.0 for the 880 nm group likely reveals an influence of sulfate on absorption of BC particles. It is also noteworthy to mention that our AAEs were derived from regression method using seven wavelengths, whereas they were calculated using wavelength pairs in Lee et al. (2012).

To examine the different absorption tendencies of the two groups, we calculated BC MAC along wavelengths for the two groups assuming the measured absorption was attributed solely to BC particles measured as EC (Fig. 5b). If no other absorbing aerosol than BC is found, then the BC MAC would be the same for the two groups. However,

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the difference became substantial at wavelengths shorter than 520 nm, where the MAC of the 370 nm group was considerably higher than that of the 880 nm group. At wavelengths longer than 520 nm, MAC was larger for the 880 nm group than for the 370 nm group and their difference was nearly constant along wavelengths. With consideration to the Lack and Cappa et al. (2010) argument, we compared AAEs calculated for several different wavelength ranges: 370–520 nm, 450–660 nm, and 660–950 nm (Fig. 5c). For the 370 nm group, the AAE was increased only at UV region (370–520 nm) by 15 % relative to that of the visible range (470–660 nm) with no considerable difference in the other ranges. In contrast, the AAE of the 880 nm group was decreased by 6 % at UV region (370–520 nm) but increased by 20 % at IR region (660–950 nm), compared to that of the visible range (470–660 nm). While the former indicates no enhanced absorption at UV region, the latter possibly results from relatively less absorption at IR region compared to the visible range. At the same time, the both can be attributed to the enhanced absorption at longer visible range compared to the UV and IR regions. These results imply that the enhanced absorption of UV and longer visible light were responsible for the spectral dependence observed in the EBC of the 370 nm and 880 nm groups, respectively.

It should also be mentioned that this analysis only considered the most dominant species. Therefore, dominant species such as OC and sulfate are likely responsible for enhanced absorption of UV and longer visible light for the 370 nm and 880 nm groups, respectively. The enhanced absorption of OC dominant aerosols in the UV region is consistent with what was observed from other measurements, which highlights the contribution of OC to aerosol spectral dependence. When sulfate was predominant, particularly in aged air masses, sulfate likely contributed to light absorption as a coating on BC particles.

3.3.2 SSA in relation to major aerosol composition

In this study, enhanced light absorption at 370 nm was found in a fast-moving winter air masses transported from the Beijing region within approximately one day, whereas

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the high mass absorption at 880 nm was observed in air masses that slowly passed through the Yellow Sea region (bottom panels in Fig. 4). In the northeast Pacific rim, outflows transported from areas around Beijing tended to show a higher ratio of black carbon to sulfate than those from other parts of China, exerting a strong positive influence on the net warming (Lim et al., 2012; Ramana et al., 2010). In order to examine how chemical composition alters the optical properties of aerosols, thereby affecting climate, SSA was calculated and compared for the two groups (Fig. 5d). The SSA was higher for the 880 nm group than for the 370 nm group, despite the higher absorption of the 880 nm group. This result highlights the importance of considering relative abundance as well as absolute concentrations when estimating radiative forcing of aerosols.

4 Conclusions

PM_{1.0} and PM₁₀ samples were collected daily for the analysis of soluble ions, OC, and EC from January to September 2008, in conjunction with continuous measurements of absorption and scattering properties and number size distributions of aerosols at GCO. In the comparison of daily averaged EBC with PM_{1.0} EC concentrations, two types of spectral dependence were identified with clear maxima (peaks) at either 370 nm or 880 nm, by which two subsets of each five samples were extracted and classified into 370 nm and 880 nm groups. The groups were distinguished by the relative abundance of major constituents such as OC/EC, OC/sulfate and EC/sulfate ratios. While the 370 nm group was characterized by high ratios of OC/EC, OC/sulfate, and EC/sulfate, sulfate was predominant for the 880 nm group.

In the PM_{1.0} of the 370 nm group, the ratios of OC/EC and (OC₃ + OC₄ + OP)/OC were higher than those of the 880 nm group. The main trait of the 370 nm group was the enhanced light absorption at 370 nm and greater contribution of refractory OC and OP to total OC, as previous studies reported that increased absorption, particularly in the UV regions, was from light-absorbing organic matter, frequently referred to as brown carbon. In addition, the refractory and pyrolyzed OC were well correlated with

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PM_{1.0} EC1 (char EC), which indicates biofuel and biomass combustion from residential heating as the source, particularly during winter. Conversely, the refractory OC in PM_{1.0} of the 370 nm group was likely derived from alkaline soil in dry lakes, contributing to light scattering. The SEM images and number size distributions together with trajectory analysis suggested that these aerosols were externally mixed in fast-moving air masses from the Beijing area passing over Mongolia in approximately one day. In comparison, some samples with a chemical composition similar to the PM_{1.0} of the 370 nm group but having different PM_{1.0} due to the influence of soil minerals, showed a slight increase in absorption at 660 nm without a clear trend in spectral dependence.

In the sulfate dominant regime, the chemical characteristics of the 880 nm group resulted in the lowest ratios of EC/sulfate and OC/sulfate in PM_{1.0} among all of the measurements. The OC/EC ratio was lower in the 880 nm group than the 370 nm group; while OC was similar in the two groups, EC was higher in the 880 nm group. The scattering coefficients at 520 nm were higher for the 880 nm group owing to its high sulfate concentrations in both PM_{1.0} and PM₁₀. However, the scattering coefficient per sulfate mass was less in the 880 nm group than in the 370 nm group. In addition, the aerosols were highly likely to be internally mixed because the mode and number distributions of aerosols remained unchanged through the day and air masses were transported slowly over the Yellow Sea.

The average AAE estimated from σ_{ap} at seven wavelengths from 370 nm to 950 nm was 1.29 (1.24–1.4) for the 370 nm and 1.0 (0.95–1.05) for the 880 nm groups, respectively. AAEs calculated for several different wavelength confirmed that the absorption was enhanced in the UV or longer visible light regions, leading to the maxima at 370 nm or 880 nm in EBC concentrations, respectively. In turn, the dominant species such as OC and sulfate were likely responsible for enhanced absorption of UV and longer visible light for the 370 nm and the 880 nm groups, respectively. In the OC dominant regime, the enhanced absorption of UV highlights the contribution of OC to aerosol spectral dependence. When sulfate was predominant, particularly in aged air masses, sulfate likely contributed to light absorption by coating BC particles. Finally, SSA was higher

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for the 880 nm group compared to that of the 370 nm group. The results of this study demonstrated that the optical properties of aerosols are intimately linked with their composition and mixing state and revealed the importance of the relative abundance as well as absolute concentrations of absorbing and scattering constituents in determining the climate effect of aerosols.

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Table 1. Comparison of the 370 nm group and the 880 nm group.

	370 nm group		880 nm group	
Date	3, 15, and 27 Jan, 14 Feb, 3 Mar		11 Mar, 14 and 17 Apr, 21 May	
Meteorological parameters				
WS (ms ⁻¹)	7.1–12.6 (11.6) ^a		2.6–5.7 (4.0)	
WD	NW, N		SE, NE, NW	
Temp. (°)	4.4–8.5 (5.9)		10.6–18.6 (13.9)	
RH (%)	45–59 (53)		58–79 (71)	
Precursor gases (ppbv)				
SO ₂	1.1–12.4 (5.8)		1.3–9.5 (3.9)	
CO	350–746 (604)		458–875 (573)	
O ₃	33–51 (40)		49–82 (65)	
PM _{1.0} (µgm ⁻³)				
Mass ^b	7.42–20.32 (13.47)		21.73–26.22 (23.36)	
Ions	2.80–11.11 (6.50)		11.79–17.80 (13.93)	
EC	0.65–2.62 (1.61)		2.08–2.31 (2.18)	
OC	1.95–4.92 (4.07)		3.22–5.30 (3.95)	
OC3+OC4	0.67–2.69 (1.75)		1.03–1.86 (1.47)	
OP	0.74–1.93 (1.31)		0.82–1.87 (1.15)	
Sulfate	0.67–5.88 (3.25)		7.99–12.33 (9.83)	
Nitrate	0.12–2.43 (1.37)		0.63–1.69 (1.07)	
PM ₁₀ (µgm ⁻³)				
Mass ^b	11.85–44.18 (29.83)		39.66–46.91 (42.18)	
Ions	5.68–20.76 (13.72)		24.97–32.55 (27.59)	
sodium	0.05–0.21 (1.38)		0.40–0.74 (0.52)	
chloride	0.63–3.54 (2.05)		0.10–0.25 (0.18)	
magnesium	0.14–0.53 (0.37)		0.16–0.22 (0.19)	
calcium	0.13–0.69 (0.42)		0.28–0.58 (0.37)	
Na ⁺ +Cl ⁻ +Mg ²⁺ +Ca ²⁺	1.40–6.67 (4.22)		1.03–1.46 (1.27)	
EBC				
	EBC (µgm ⁻³)	Absorption coeff. (Mm ⁻¹)	EBC (µgm ⁻³)	Absorption coeff. (Mm ⁻¹)
370 nm	0.67–2.09 (1.43)	8.48–27.51 (19.97)	1.63–1.99 (1.78)	20.28–26.40 (22.69)
470 nm	0.64–2.04 (1.37)	5.92–19.58 (14.15)	1.81–2.21 (1.96)	16.72–21.65 (18.60)
520 nm	0.63–2.01 (1.34)	5.11–16.88 (12.15)	1.86–2.26 (2.01)	15.13–19.43 (16.68)
590 nm	0.64–2.03 (1.35)	4.43–14.67 (10.49)	1.95–2.34 (2.09)	13.59–17.33 (14.93)
660 nm	0.64–2.04 (1.36)	3.89–12.79 (9.19)	2.02–2.40 (2.15)	12.26–15.41 (13.41)
880 nm	0.64–2.01 (1.33)	2.81–9.13 (6.52)	2.09–2.45 (2.20)	9.23–11.24 (9.91)
950 nm	0.61–1.92 (1.27)	2.49–7.98 (5.72)	2.04–2.35 (2.14)	8.17–9.78 (8.75)
Scattering coefficient (Mm ⁻¹)				
450 nm	58–257 (147)		257–360 (308)	
550 nm	47–199 (113)		200–278 (235)	
700 nm	34–135 (76)		132–179 (150)	

^a The numbers in parentheses are averaged values.

^b Of the total mass, eight ionic soluble ions, OC, and EC were measured. Detailed information on chemical composition can be found in Lim et al. (2012).

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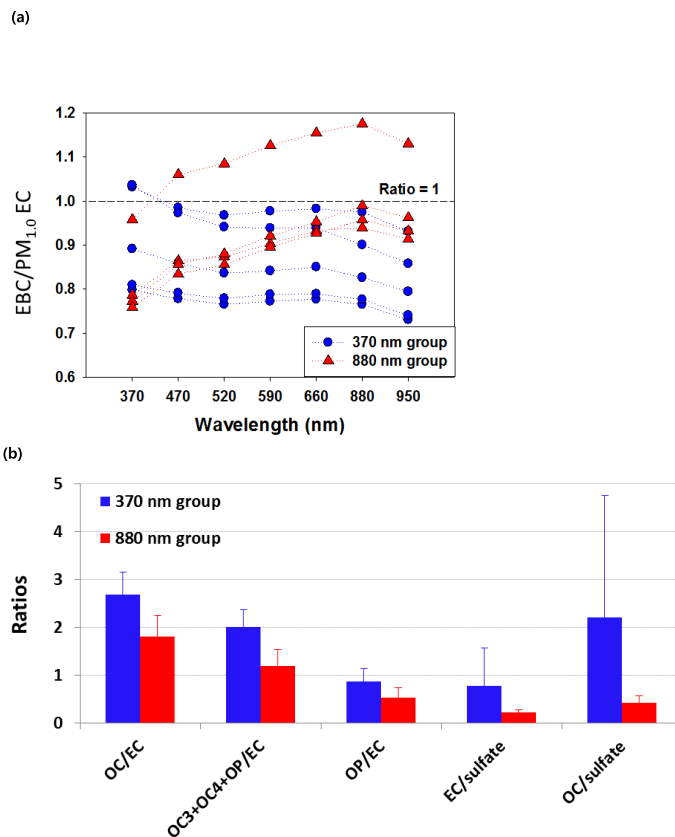


Fig. 1. Comparison of the 370 nm group with the 880 nm group. **(a)** EBC/PM_{1.0} EC ratios as a function of wavelength. **(b)** Ratios of OC/EC, OC₃ + OC₄ + OP/EC, OP/EC, EC/sulfate, and OC/sulfate in PM_{1.0}. Ratios are shown as mean values and error bars indicate standard deviation (1 σ).

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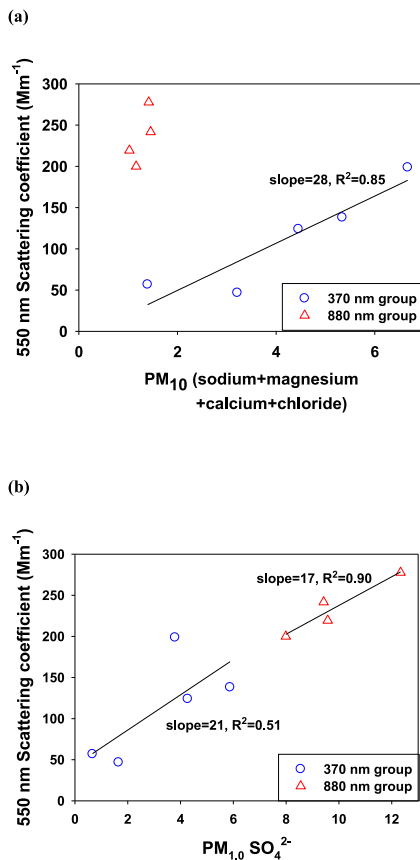


Fig. 2. (a) Correlation of PM₁₀ ($\text{Na}^+ + \text{Ca}^{2+} + \text{Mg}^{2+} + \text{Cl}^-$) with 550 nm scattering coefficient. (b) Correlation of PM_{1,0} sulfate with 550 nm scattering coefficient. Scattering coefficient data obtained by nephelometer were collected at 10 min intervals, and were averaged for each day to be compared with our observed PM data. Lines show linear regressions, for which slopes and R^2 values are given.

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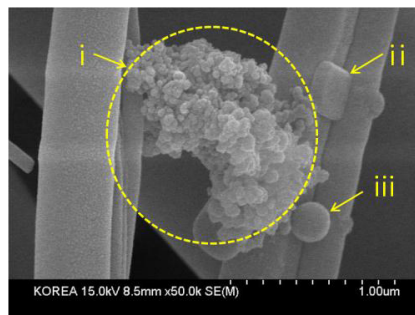
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(a)



(b)

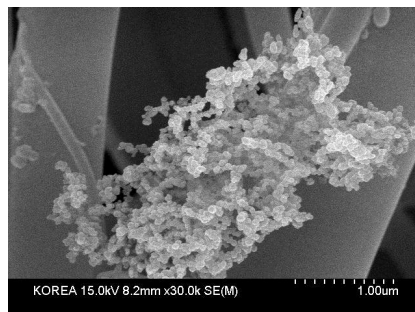


Fig. 3. SEM images of ambient particles. **(a)** Particles in the 370 nm group, collected on 3 January 2008. Regions and/or particles marked by i, ii, and iii appear to be organic matter, sea-salt, and a tar-ball, respectively. **(b)** Particles in the 880 nm group, collected on 14 April 2008. In this sample, individual spherules are organized into large, irregular, or chain-like aggregates.

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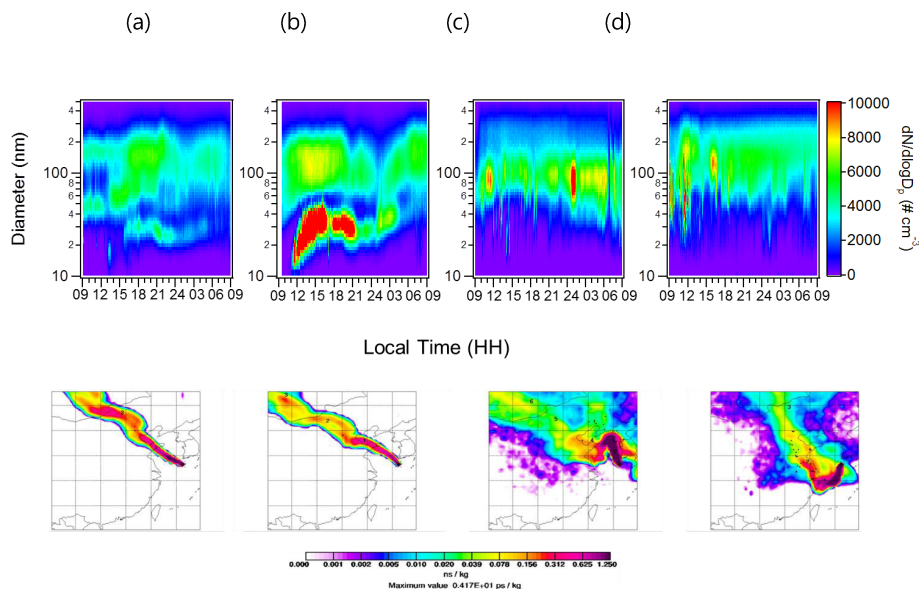


Fig. 4. Comparison of particle number size distribution (top panels) and air mass trajectories (bottom panels) for the 370 nm and 880 nm groups. **(a)** 3 January and **(b)** 14 February are included in the 370 nm group, and **(c)** 17 April and **(d)** 21 May are included in the 880 nm group. Particle number concentrations were measured by SMPS. Air trajectories were obtained by Flexpart model footprint images in (Stohl et al., 2005; <http://zardoz.nilu.no/~andreas/STATIONS/GOSAN/index.html>). The model output (skg^{-1}) is a potential emission sensitivity distribution of 40 000 particles released in a particular grid cell at the measurement location during the measurement interval and followed backward in time. This is considered proportional to the particle residence time in that cell.

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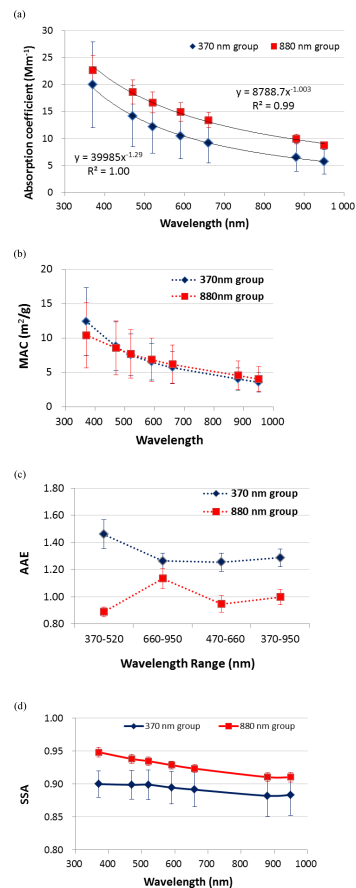


Fig. 5. (a) Absorption coefficients of the 370 nm and 880 nm groups and fitted absorption ångstrom exponent (AAE). (b) AAE for the two groups in different wavelength ranges. (c) Mass absorption cross section (MAC) of the two groups (please see text for details of the calculation). (d) Single scattering albedo (SSA) along the wavelengths for the two groups.