

1 **Analysis of feedbacks between nucleation rate, survival** 2 **probability and cloud condensation nuclei formation**

3 **D.M. Westervelt^{1,*}, J.R. Pierce², and P.J. Adams¹**

4 [1]{Center for Atmospheric Particle Studies (CAPS), Carnegie Mellon University, Pittsburgh,
5 PA, USA}

6 [2]{Department Atmospheric Science, Colorado State University, Fort Collins, CO}

7 *Now at the Program in Science, Technology, and Environmental Policy, Woodrow Wilson
8 School of Public and International Affairs, Princeton University, Princeton, NJ, USA

9 Correspondence to: P.J. Adams (petera@andrew.cmu.edu)

10 **Abstract**

11 Aerosol nucleation is an important source of particle number in the atmosphere. However, in
12 order to become cloud condensation nuclei (CCN), freshly nucleated particles must undergo
13 significant condensational growth while avoiding coagulation scavenging. In an effort to
14 quantify the contribution of nucleation to CCN, this work uses the GEOS-Chem-TOMAS
15 global aerosol model to calculate changes in CCN concentrations against a broad range of
16 nucleation rates and mechanisms. We then quantify the factors that control CCN formation
17 from nucleation, including daily nucleation rates, growth rates, coagulation sinks,
18 condensation sinks, survival probabilities, and CCN formation rates, in order to examine
19 feedbacks that may limit growth of nucleated particles to CCN. Nucleation rate
20 parameterizations tested in GEOS-Chem-TOMAS include ternary nucleation (with multiple
21 tuning factors), activation nucleation (with two pre-factors), binary nucleation, and ion-
22 mediated nucleation. We find that nucleation makes a significant contribution to boundary
23 layer CCN(0.2%), but this contribution is only modestly sensitive to choice of nucleation
24 scheme, ranging from 49-78% increase in concentrations over a control simulation with no
25 nucleation. Moreover, a two order-of-magnitude increase in the globally averaged nucleation
26 rate (via changes to tuning factors) results in small changes (less than 10%) to global
27 CCN(0.2%) concentrations. To explain this, we present a simple theory showing that survival
28 probability has an exponentially-decreasing dependence on the square of the condensation
29 sink. This functional form stems from a negative correlation between condensation sink and
30 growth rate and a positive correlation between condensation sink and coagulation

1 scavenging. Conceptually, with a fixed condensable vapor budget (sulfuric acid and organics),
2 any increase in CCN concentrations due to higher nucleation rates necessarily entails an
3 increased aerosol surface area in the accumulation mode resulting in a higher condensation
4 sink, which lowers vapor concentrations and growth rates. As a result, slowly growing nuclei
5 are exposed to a higher frequency of coagulative scavenging for a longer period of time,
6 thus reducing their survival probabilities, and closing a negative feedback loop that dampens
7 the impact of nucleation on CCN. We confirm quantitatively that the decreases in survival
8 probability predicted by GEOS-Chem-TOMAS due to higher nucleation rates are in
9 accordance with this simple theory of survival probability.

10 **1 Introduction**

11 Aerosols affect climate directly by scattering incoming solar radiation and indirectly by
12 modifying cloud properties. The largest uncertainty in climate forcing is the aerosol indirect
13 effect, which consists of the cloud brightness (albedo) effect and the cloud lifetime effect and
14 is thought to have an overall cooling influence on global temperature (Twomey, 1977;
15 Albrecht, 1989). If water vapor amount and cloud dynamics are held constant, brighter clouds
16 with longer lifetimes are formed with enhanced aerosol number concentrations. The subset of
17 particles that serve as sites for cloud droplet formation are known as cloud condensation
18 nuclei (CCN). The ability of a particle to function as a CCN mainly depends on three factors:
19 the maximum supersaturation of relative humidity reached in the cloud, the particle diameter,
20 and the particle composition. In order for particles to affect clouds, they are either introduced
21 into the atmosphere by direct emission or by formation of new particles (aerosol nucleation).
22 Nucleated particles and ultrafine particles from primary emissions must undergo significant
23 growth to achieve the sizes required to function as CCN (Kerminen, 2005; Pierce and Adams,
24 2007; Kuang et al., 2009). The representation of nucleation in models is uncertain with
25 orders-of-magnitude differences in nucleation rates between commonly used schemes, leading
26 to uncertainty in estimates of the nucleation contribution to CCN (Makkonen et al., 2009;
27 Merikanto et al., 2009; Pierce and Adams, 2009b; Wang and Penner, 2009; Yu and Luo,
28 2009; Spracklen et al., 2010; Reddington et al., 2011). The role of sulfuric acid vapor as an
29 essential nucleating species has been reported by many studies (Doyle, 1961; Kulmala and
30 Laaksonen, 1990; Weber et al., 1995, 1997; Noppel et al., 2002; Berndt et al., 2005; Sihto et
31 al., 2006; Kuang et al., 2008; Sipilä et al., 2010; Vuollekoski et al., 2010; Pierce et al., 2012;
32 Kulmala et al., 2013). Other vapors that may assist in the initial stages of nucleation include

1 low volatility organic vapors (Zhang et al., 2004; Metzger et al., 2010; Paasonen et al., 2010),
2 amines (Kurtén et al., 2008; Bzdek et al., 2010; Kirkby et al., 2011; Chen et al., 2012;
3 Almeida et al., 2013), and ammonia (Ball et al., 1999; Erupe et al., 2010; Kirkby et al., 2011).
4 Chen et al. (2012) recently proposed an acid-base nucleation mechanism involving sulfuric
5 acid and amines and achieved nucleation rate closure to within a factor of 10, which is
6 significantly better than classical theories which differ by as much as a factor of 10^{10} .
7 Additionally, a laboratory study by Almeida et al. (2013) recently showed that amines may
8 help explain observed nucleation rates in the lower atmosphere.

9 Many parameterizations have been developed for calculating nucleation rates in global
10 models; we highlight a few that are commonly used below. The binary nucleation
11 parameterization of Vehkamäki et al. (2002) is often used to represent particle nucleation in
12 the free troposphere. In this scheme, supersaturated solutions of water vapor and sulfuric
13 acid form thermodynamically stable clusters. The Napari et al. (2002) parameterization adds
14 ammonia (NH_3) as a third nucleating species. Because the original formulation of Napari et al.
15 (2002) showed high biases in predictions of nucleation rates and aerosol number
16 concentrations (Jung et al., 2006; Merikanto et al., 2007), a scaled version with a globally
17 constant nucleation rate tuning prefactor of 10^{-5} can be used as in Westervelt et al. (2013).
18 This modified ternary parameterization has been incorporated into a both a regional and
19 global aerosol model and shows reasonable agreement with observations (Jung et al., 2010;
20 Westervelt et al., 2013). The activation nucleation mechanism (Kulmala et al., 2006) is an
21 empirical formulation which is often applied in the planetary boundary layer (PBL) in
22 conjunction with the binary scheme of Vehkamäki et al. (2002) in the free troposphere.
23 Although the nucleation rate is simply proportional to sulfuric acid concentration, activation
24 nucleation has been shown to agree well with ambient nucleation observations at five
25 locations (Westervelt et al., 2013). Ion-mediated nucleation, (Yu, 2010) in which atmospheric
26 ionization enhances the nucleating ability of precursor vapors is also considered in the present
27 work. Ion-mediated nucleation rates and number concentrations have compared favorably to
28 observations in a global model (Yu and Turco, 2011), although other studies have found that
29 ions do not likely play a large role in boundary layer nucleation (Laakso et al., 2007;
30 Manninen et al., 2009; Gagné et al., 2010; Kirkby et al., 2011; Almeida et al., 2013).

31 Understanding the growth and loss processes of fresh nuclei is a critical step in determining
32 the contribution of nucleation events to aerosol number and CCN concentrations. Whether or

1 not a nucleated particle can act as a CCN depends on its survival probability (SP): the
2 likelihood that the particle will grow to large enough sizes (typically at least 50 – 100 nm)
3 without being coagulationally scavenged (Kerminen, 2005; Pierce and Adams, 2007; Kuang
4 et al., 2009). The concept of survival probability is defined rigorously in Westervelt et al.
5 (2013) and in Sect. 2.3 of this paper. Although both primary and nucleated particles undergo
6 the same microphysical processes (condensation, coagulation), their survival probabilities
7 may be vastly different due to their very different initial particles sizes. Initial sizes of
8 nucleating clusters are typically ~1 nm in size, which is much smaller than any primary
9 emission size ranges (Mäkelä et al., 1997; Vehkamäki et al. 2002; Kulmala et al., 2000,
10 2004a, 2013). Because of this size disadvantage, particles formed via nucleation have further
11 to grow through a larger range of sizes and are exposed to coagulation scavenging for longer
12 periods of time than primary emissions. Additionally, these smaller particles are highly
13 diffusive and more likely to collide with pre-existing particles. Coagulation frequency is
14 therefore higher between fresh nuclei and larger pre-existing particles, adding to the
15 disadvantage that nucleated particles have to grow to CCN sizes. Kuang et al. (2009) inferred
16 survival probabilities from size distribution measurements and found that at least 80% of the
17 nucleated particles measured at Atlanta, GA and Boulder, CO were lost by coagulation before
18 the nucleation mode reached CCN sizes in the cases that they studied (20% survival
19 probability), even during days with high growth rates.

20 Most attempts to quantify the contribution of nucleation events to global CCN concentrations
21 have come in the form of sensitivity studies in which nucleation is zeroed-out as a control
22 simulation and then is compared to simulations with nucleation active. Modeling studies have
23 reported CCN sensitivities to nucleation ranging from roughly 5-60% (Makkonen et al., 2009,
24 2012; Merikanto et al., 2009; Pierce and Adams, 2009b; Reddington et al., 2011; Spracklen et
25 al., 2008, 2010; Wang and Penner, 2009; Yu and Luo, 2009). Each of these studies used
26 different models and often significantly different inputs, assumptions, and metrics for
27 assessing CCN sensitivity, making model intercomparison difficult. The details of these
28 studies are compared in depth in Westervelt et al. (2013) and also in Table 1. Recently, Lee et
29 al. (2013) compiled 28 model parameters covering several important aerosol processes and
30 ran Monte Carlo simulations to determine the magnitude of uncertainty in CCN
31 concentrations caused by each parameter. They find that while 45% of CCN are attributed to
32 nucleation, the CCN are generally insensitive to the details of the nucleation rates across a
33 wide, but sensible, range of boundary layer and free-tropospheric nucleation assumptions. On

1 the other hand, the most important factors contributing to uncertainties in CCN include the
2 emissions size distribution of primary particles, the amount of carbonaceous emissions, sub-
3 grid sulfate formation, and aerosol deposition.

4 The relative insensitivity of CCN concentrations to very large increases in the nucleation rate
5 can be explained in four steps: 1) For fixed condensable vapor production rates, an increase in
6 aerosol number due to nucleation and subsequent growth causes an increase in surface area in
7 the accumulation mode and a higher condensation sink. 2) The higher condensation sink
8 depletes vapors needed for particle growth; thus, the concentrations of condensable vapors are
9 reduced and growth rates are slowed. 3) The higher condensation sink also correlates with
10 higher coagulation scavenging frequencies; thus, a larger fraction of the growing particles are
11 scavenged (even if growth rates were held constant, which they are not) 4) This combination
12 of the slower growth rates and the faster coagulation scavenging lower particle survival
13 probabilities and CCN formation rates. Although CCN concentrations will typically not
14 decrease as a result of increased nucleation, the increase in CCN will be dampened such that
15 their fractional increase in CCN will be much smaller than the fractional increase in the
16 nucleation rate.

17 The increase in surface area due to nucleation (step 1 above) requires elaboration since
18 conventional wisdom holds that the surface area of the nucleation mode is too small to
19 contribute significantly to surface area. There are two main reasons the surface area (and thus
20 condensation sink) will increase from enhanced nucleation rates. First, the atmosphere has a
21 limited budget of condensable vapors, sulfuric acid and secondary organics, that dominate the
22 mass concentration of CCN mode particles. Given the fixed amount of aerosol mass
23 produced, any increase in CCN number concentrations (e.g. due to nucleation) implies a shift
24 in the CCN mode to smaller sizes and, therefore, an increase in aerosol surface area and
25 condensation sink. This effect is analogous to the aerosol indirect effect in which a fixed
26 water vapor budget implies an increase in cloud surface area if number concentrations
27 increase. Second, the direct contribution of nucleated particles to surface area and
28 condensation sink is non-negligible. For events with high nucleation and growth rates, enough
29 nucleation mode particles will form to compensate for their small surface area contribution
30 (we will show an example of this). This can result in a small but non-negligible enhancement
31 in the surface area over the course of a nucleation burst.

1 In this work, we quantify the global sensitivity of CCN to uncertainties in nucleation rates
2 across a wide range of nucleation rates in a global aerosol microphysics model (GEOS-Chem-
3 TOMAS) in an attempt to help unify the previous studies of this sensitivity. We also compare
4 CCN sensitivities for simulations with biogenic SOA only (19 Tg yr⁻¹, default treatment in
5 GEOS-Chem-TOMAS) and simulations with an extra 100 Tg yr⁻¹ of anthropogenic SOA
6 (Spracklen et al., 2011; D'Andrea et al., 2013). Here, CCN sensitivity refers to the percent
7 increase in CCN concentrations between two simulations with differing nucleation rates and
8 mechanisms (generally the comparison is between a simulation with no nucleation to a
9 simulation with a certain nucleation scheme turned on). We also investigate the microphysics
10 of CCN formation in detail with one year of modeled size distribution output to determine
11 quantitatively the feedback factors that are controlling our sensitivity results. We specifically
12 test the hypothesis that higher nucleation rates will lead to a lower survival probability and
13 will dampen CCN concentrations using global modeling results and the nucleated particle
14 analysis code presented in Westervelt et al. (2013). We propose a simplified theory for the
15 hypothesized CCN dampening in which particle survival probability is inversely related to
16 condensation sink and show that model results are generally consistent with this simple
17 theory.

18 **2 Models and analysis**

19 **2.1 GEOS-Chem**

20 The Goddard Earth Observing System global chemical transport model (GEOS-Chem)
21 version 8.2.2 is used for this study (Bey et al., 2001; <http://geos-chem.org>). The version of
22 GEOS meteorological fields used was GEOS-3 for all simulations. In all simulations, 4°
23 latitude by 5° longitude resolution is used with 30 vertical sigma-coordinate layers extending
24 from the surface to 0.01 hPa. We describe a few key model setup features here but refer the
25 reader to Trivitayanurak et al. (2008) and Westervelt et al. (2013) for full details.
26 Anthropogenic emissions are treated with the Emissions Database for Global Atmospheric
27 Research (EDGAR) inventory and regional inventories (Olivier et al., 1996). These regional
28 inventories include Big Bend Regional Aerosol and Visibility Observational Study (BRAVO)
29 emissions inventory for Mexico and southwestern US, Criteria Air Contaminants (CAC) for
30 anthropogenic emissions over Canada (<http://www.ec.gc.ca/inrp-npri/>), the Cooperative
31 Programme for Monitoring and Evaluation of the Long-range Transmission of Air Pollutants
32 in Europe (EMEP), EPA National Emissions Inventory (NEI) for the United States

1 (<http://www.epa.gov/oar/data/neidb.html>), and the Streets inventory for Asian emissions
2 (Streets et al., 2003; Auvray and Bey, 2005). Biogenic emissions in the model follow the
3 MEGAN database, and biomass burning emissions use the Global Fire Emissions Database
4 version 2 (GFEDv2) (Guenther et al., 2006). NO_x emissions from aircraft, lightning, and soil
5 are considered in the global model. Shipping SO_x emissions are considered within EDGAR
6 and EMEP.

7 **2.2 TwO-Moment Aerosol Sectional (TOMAS) algorithm**

8 As in Westervelt et al. (2013), aerosol microphysical processes such as condensation,
9 coagulation, and nucleation are calculated using the TwO Moment Aerosol Sectional
10 algorithm (TOMAS) (Adams and Seinfeld, 2002). TOMAS was introduced as a regular
11 component of the host model GEOS-Chem in version 8.2.2 and 8.3.1 and is available for
12 download (www.geos-chem.org). Advantages of the TOMAS algorithm and GEOS-Chem
13 implementation include the fact that all aerosol species have explicit, interactive microphysics
14 and TOMAS conserves number and mass concentrations allowing calculation of aerosol
15 number budgets. Generally, we employ the work of Trivitayanurak et al. (2008) and
16 Westervelt et al. (2013) with the organic aerosol additions of Pierce et al. (2007), the dust
17 additions of Lee et al. (2009), and the nucleation implementations of Pierce and Adams
18 (2009a). There are a number of nucleation theories added to the model, which are described in
19 Sect. 2.2.1. TOMAS computes the effects of nucleation, coagulation,
20 condensation/evaporation, cloud processing, size-resolved dry and wet deposition, and
21 emissions on the number and mass size distribution of aerosols (Tzivion et al., 1987; Adams
22 and Seinfeld, 2002). Aerosol chemical composition is represented by 9 species including
23 sulfate, sea salt, hydrophilic and hydrophobic organic carbon, externally and internally mixed
24 elemental carbon, mineral dust, ammonium, and aerosol water. Each of the 9 species is
25 tracked across 40 logarithmically spaced size sections covering a diameter range of 1.1 nm to
26 10 μm, resulting in 360 TOMAS-specific model tracers. Primary sulfate aerosol emissions are
27 1% of anthropogenic SO₂ emissions and use the size distributions described in
28 Adams and Seinfeld (2003). Sea salt emissions are treated in the same manner as in
29 Trivitayanurak et al. (2008) and are described in detail in Pierce and Adams (2006). Organic
30 aerosols are unchanged from Westervelt et al. (2013), except for some changes to SOA
31 condensation which are discussed in Sect. 2.2.3. Advection, chemistry, and deposition have
32 remained largely unchanged from the work of Trivitayanurak et al. (2008), although periodic

1 minor updates in both advection and chemistry (e.g. newer reaction rate constants and
2 photolysis constants) have been implemented into successive versions of GEOS-Chem.

3 2.2.1 Nucleation simulations

4 We have implemented a number of nucleation mechanisms into the model. As in Westervelt
5 et al. (2013), binary homogenous nucleation (Vehkamäki et al., 2002), ternary homogenous
6 nucleation (Napari et al., 2002), and activation nucleation (Kulmala et al., 2006; Sihto et al.,
7 2006) are included. The various simulations that we perform with the different nucleation
8 theories are described in Table 2. Binary nucleation (BHN) alone is one simulation. For
9 ternary nucleation, we employ 3 different simulations: one with a globally applied 10^{-5} tuning
10 factor on the nucleation rate (TER5), another with a 10^{-3} tuning factor (TER3), and a third
11 with no tuning factor (the original formulation, TER). Ternary nucleation defaults to binary
12 nucleation when NH_3 mixing ratios are below 0.1 pptv since the ternary nucleation scheme
13 predicts no nucleation below this NH_3 mixing ratio. Activation is also broken up into multiple
14 simulations: one with the A prefactor equal to $2 \times 10^{-6} \text{ s}^{-1}$, and another with $A = 1 \times 10^{-6} \text{ s}^{-1}$. The
15 factor-of-two change in the A value is smaller than order-of-magnitude changes evaluated in
16 previous studies (Spracklen et al., 2008, 2010). We use this smaller perturbation in nucleation
17 rates to determine how much CCN will increase due to a relatively smaller enhancement in
18 nucleation rates compared to our ternary simulations (TER, TER3, TER5). The activation
19 nucleation schemes only act in the boundary layer, and binary nucleation is used in the free
20 troposphere in these simulations. We employ the ion-mediated nucleation (IMN) of (Yu,
21 2010) as another simulation. Finally, we perform a control simulation (NONUC) where all
22 nucleation is turned off in the global model, which gives 8 simulations total. Simulations were
23 run for a total of 13 months, with 1 month of spinup, which is not used in the analysis.

24 As in Westervelt et al. (2013), gas-phase sulfuric acid concentrations are calculated using a
25 pseudo-steady state approach for each time step (Pierce and Adams, 2009a). Additionally, the
26 lower boundary on the size distribution of 1.1 nm allows for explicit simulation of the
27 dynamics of fresh nuclei (Lee et al., 2013).

28

1 2.2.2 CCN calculations

2 Cloud condensation nuclei formation in the GEOS-Chem-TOMAS model is similar to the
3 methods described in past model versions (Pierce et al., 2007; Trivitayanurak et al., 2008; Lee
4 et al., 2009). We calculate CCN concentrations at two fixed supersaturations of 1.0% and
5 0.2%, representative of convective and stratiform clouds. A comprehensive form of Köhler
6 theory (Raymond and Pandis, 2003) is employed via look-up tables that take particle
7 composition as input and yield critical activation diameters at various supersaturations as
8 output for every combination of particle compositions. We do not consider the effects of
9 surfactants on the Kelvin effect and CCN formation (Facchini et al., 1999).

10 CCN(0.2%) and CCN(1.0%) are both calculated for every nucleation simulation in this work.
11 Concentrations are calculated both spatially (latitude and longitude) and zonally-averaged
12 (latitude and atmospheric pressure/altitude). We then take percent differences between the
13 CCN concentrations for each simulation and the concentrations for the NONUC, control
14 simulation.

15 2.2.3 Secondary organic aerosol

16 We use the simple secondary organic aerosol (SOA) fixed yield approach in which SOA is
17 considered to be essentially non-volatile and does not thermodynamically partition between
18 the vapor and condensed phase, which has been shown to be a better approximation for
19 representing the growth of ultrafine particles (Pierce et al., 2011; Riipinen et al., 2011;
20 D'Andrea et al., 2013). SOA is essentially a “pseudo-primary” source in TOMAS, calculated
21 as 10% of modeled monoterpene emissions. The globally averaged flux using this approach is
22 approximately 19 Tg yr^{-1} , which is on the low end of estimated SOA fluxes (Goldstein and
23 Galbally, 2007; Heald et al., 2011; Spracklen et al., 2011) Similar to sulfuric acid
24 condensation, SOA condenses to all particles based on their Fuchs surface area (Pandis et al.,
25 1991). Despite strong evidence for the partitioning of semi-volatile organic aerosols between
26 the gas and particle phase (Donahue et al., 2006), the SOA treatment used here is simple and
27 performed well in earlier nucleation studies that compared to observed aerosol number
28 concentrations and growth rates (Riipinen et al., 2011, Pierce et al., 2011, Westervelt et al.,
29 2013) and is not inconsistent with equilibrium partitioning if low volatility organics are
30 formed quickly (Donahue et al., 2011).

1 Recent studies have pointed to a large missing source of SOA in global models, perhaps
2 anthropogenic or anthropogenically-controlled in nature (Heald et al., 2011; Spracklen et al.,
3 2011). As a result, we include a series of 8 more simulations in which 100 Tg yr⁻¹ of an
4 additional SOA precursor is emitted at locations coincident with anthropogenic SO₂ emissions
5 (Spracklen et al., 2011; D'Andrea et al., 2013). The generic precursor has a timescale of 12
6 hours until it forms SOA with unit yield in the condensed phase. Although this is a simple
7 approach, it is sufficient enough for our purposes in testing the sensitivity CCN concentrations
8 from nucleation.

9 **2.3 Nuclei fate analysis**

10 We output one year of size distribution data from the model every 30 minutes from the model
11 grid cell corresponding to Hyytiälä, Finland and calculate nucleation-relevant parameters in
12 the same manner done in Westervelt et al (2013), which evaluated the global model using
13 many of the same parameters. In Westervelt et al. (2013), the model showed good agreement
14 with observations, as average growth and nucleation rates, survival probabilities, and CCN
15 formation rates were biased by less than 50%.

16 We refer the reader to Westervelt et al. (2013) for details, but briefly, the nucleation and
17 growth rates are based on a single-day analysis first outlined in Dal Maso et al. (2005). The
18 nucleation rate, J_3 , is a number balance between the rate of formation of nucleation mode
19 particles and losses of those particles due to coagulation and growth out of the size range. For
20 purposes of the nucleation rate calculation, we define the nucleation mode as the 3-25 nm size
21 range. Diameter growth rates are also calculated for the 3-25 nm range as well as a 25-100 nm
22 range that is used for the survival probability calculation. Growth rate (GR) is the rate of
23 change in size (diameter) over time during a nucleation and growth event (see Fig. 2 of
24 (Westervelt et al., 2013)).

25 Two important quantities for particle and vapor loss, which partially determine CCN
26 sensitivity to nucleation, are the coagulation sink and condensation sink. The coagulation
27 sink, $CoagS$ (units of s⁻¹), of particles of size i to a larger size j is dependent on a coagulation
28 coefficient (K_{ij}) and the number concentration in the larger size range, N_j (Eq 1). In our
29 calculations, we calculate coagulation coefficients for all particles larger than size i . The
30 coagulation coefficient is based on Fuchs equation (Seinfeld and Pandis, 2006).

$$1 \quad CoagS_i = \frac{1}{2} K_{ii} N_i + \sum_{j=i+1}^{\max} K_{ij} N_j \quad (1)$$

2 The condensation sink (CS, also units of s^{-1}) is proportional to the aerosol surface area in the
 3 kinetic regime and particle diameter in the continuum regime. It describes the first-order rate
 4 of uptake of sulfuric acid or other condensable vapors to aerosols (Eq. 2). In Eq. 2, D refers to
 5 the gas-phase diffusion constant, D_{pi} is the particle diameter in size bin i , N_i is the number
 6 concentration in size i , and β_i is the non-continuum correction factor, which is a function of
 7 the Knudsen number (Seinfeld and Pandis, 2006).

$$8 \quad CS = 2\pi D \sum_{i=1}^{\max} \beta_i D_{pi} N_i \quad (2)$$

9 Calculation of survival probability is adapted from the Probability of Ultrafine Growth (PUG)
 10 theory, introduced by Pierce and Adams (2007). We define survival probability as the ratio of
 11 particle formation rates at the initial point of growth (typically J_3) and the CCN-relevant size
 12 or endpoint of growth (J_n , with $n = 50$ or 100 nm typically). It is calculated according to Eq. 3
 13 below (see also Eqs. 4-6 in Westervelt et al. (2013)). It is a highly time and size resolved
 14 method that involves calculating timescales of condensation growth and coagulation loss at
 15 each step in the growing nucleation mode. The formation rate of 100 nm particles (J_{100}) is
 16 calculated as the 3 nm formation rate multiplied by the survival probability from 3 to 100 nm
 17 (Eq. 4). Likewise, J_{50} is calculated as J_3 multiplied by the survival probability to 50 nm. These
 18 two particle sizes are within the range of typical activation diameters for CCN concentrations.

$$19 \quad SP_{m,n} = \prod_{k=m}^{n-1} \exp\left(-\frac{\tau_{k,k+1}^{cond}}{\tau_k^{coag}}\right) \quad (3)$$

$$20 \quad J_n = SP_{3-n} J_3 \quad (4)$$

21 We extend our survival probability calculation beyond the end of the first day and estimate a
 22 multi-day survival probability. This estimate extrapolates the 25 - 100 nm growth rate and
 23 coagulation loss rates at the end of the growth period and applies it to subsequent days,
 24 allowing particles to either grow to CCN sizes or be lost via coagulation at a later time. We
 25 judge that the multi-day survival probability estimate probably overstates the ultimate CCN
 26 formation for several reasons. First, while real nuclei do not grow as quickly overnight but are
 27 lost to coagulation, we extrapolate the daytime growth rate for all subsequent hours, day or

1 night. Second, it is expected that growth rates on nucleation days are somewhat higher than
 2 average. Third, it is expected that the coagulation sink on nucleation days is somewhat lower
 3 than average (Gong et al., 2010; Wu et al., 2011). As air masses have often shifted over a
 4 given location on days after nucleation days, it is difficult to track explicitly the actual
 5 evolution of the growing nucleation mode after the first day. Thus, this extrapolation method
 6 allows us to better estimate the growth to sizes beyond what is reached in the first day.

7 **2.4 Simplified model for dependence of survival probability on condensation** 8 **sink**

9 In addition to the methods described in Sect. 2.3, the survival probability can also be
 10 represented in another way based on simple scaling arguments, which we will make use of in
 11 Sect. 3.2.3. Here we propose a simple model that will be used to explain how changes in
 12 coagulation and condensation result in decreases in survival probability. We start with the
 13 theory presented by Lehtinen et al. (2007) for survival probability:

$$14 \quad SP_{i-x} = \frac{J_x}{J_i} = \exp\left(-\gamma \cdot d_i \cdot \frac{CoagS(d_i)}{GR}\right) \quad (5)$$

$$15 \quad \gamma = \frac{1}{m+1} \left[\left(\frac{d_x}{d_i}\right)^{m+1} - 1 \right] \quad (6)$$

16 Eqs. 5 and 6 (from Lehtinen et al. (2007)) are an updated form of the original equations in
 17 Kerminen and Kulmala (2002). In the equations, d_i and d_x are the initial and final particle
 18 diameters (e.g. 3-100 nm), $CoagS$ is the coagulation sink, GR is the growth rate, J_i is the
 19 particle nucleation rate at reference size i , J_x is the eventual formation rate of larger, CCN-
 20 sized particles from those nuclei, γ is a parameter that accounts for size-dependent coagulation
 21 as the particle grows from initial to final sizes, m is a constant that ranges between -1 and -2
 22 (see Lehtinen et al. (2007) for details), and SP is survival probability. Leaving aside the γ
 23 parameter, the exponential decay of particles predicted by Eq. 5 may be understood as
 24 follows. The time it takes for a fresh nuclei to grow to its final size is inversely proportional to
 25 the growth rate, GR . Over this time, a first-order loss of particles due to coagulation occurs
 26 with frequency, $CoagS$.

1 Recognizing that the coagulation and condensation sinks are linearly correlated (Lehtinen et
2 al., 2007) (see also Sect. 3.2.3 below) while the growth rate and the condensation sink are
3 linearly anti-correlated (since, assuming steady-state, the concentrations of condensable
4 vapors are inversely proportional to the condensation sink), we can simplify Eqs. 5 and 6. For
5 given bounds in the survival probability calculation, Eqs. 5 and 6 can now be written as solely
6 a function of the CS (Eq. 7).

$$7 \quad SP = \exp(-a CS^2) \quad (7)$$

8 Eq. 7 shows the basic form of the inferred relationship between survival probability and
9 condensation sink. This equation is similar to Eq. A7 in McMurry et al. (2005), which related
10 the particle survival probability (P) to the ratio between CS and GR. Here, we have simply
11 taken one step further, noting that the GR (growth rate) is inversely proportional to CS,
12 yielding an expression for SP as a function of CS and the constant a that depends on the initial
13 and final particle diameters and the relationships between CoagS, GR, and CS. We note,
14 however, that the GR is also proportional to the production rate of condensable vapors. If this
15 production rate varies widely between nucleation events, this simplified model for the
16 survival probability will not generally yield good predictions. It is also possible that the same
17 sources that add to the vapor production rate also may contribute to condensation sink, adding
18 another source of uncertainty in the simple model. However, to the extent that this simple
19 formulation captures the actual changes in survival probability predicted by the full model
20 (where vapor production does contribute to condensation sink), we can conclude that this
21 effect is small. We will show later that this model does generally fit our full survival-
22 probability calculations well and that variability in the production rate of condensable vapors
23 is relatively minor in GEOS-Chem-TOMAS.

24 **3 Results**

25 Global maps of N_{10} , CCN(1.0%), and CCN(0.2%) for the TER5 simulation are shown in Fig.
26 1. All maps and figures, except for Fig. 4, only include the base case 19 Tg yr^{-1} SOA source
27 and do not include the 100 Tg yr^{-1} anthropogenically enhanced SOA source. Westervelt et al.
28 (2013) showed that the TER5 nucleation mechanism performed well against observed
29 nucleation rates, growth rates, particle survival probabilities, and CCN formation efficiency
30 from several field-campaign sites with biases within 50% for all metrics. Thus, the TER5
31 simulation is shown here as a possible “best guess” to CCN and number concentrations. In
32 Fig. 1, annually averaged N_{10} concentrations are highest over the continents, with specific

1 hotspots in Eastern North America, Western Europe, and China. CCN(1.0%) and CCN(0.2%)
2 generally follow the same pattern, although with lower concentrations. CCN(0.2%) are less
3 abundant than CCN(1.0%), which is expected due to the larger diameter required for
4 activation at $S = 0.2\%$.

5 **3.1 Global sensitivity of N_{10} and CCN to nucleation**

6 Figure 2 shows the percent change in annually averaged N_{10} for a selection of the sensitivity
7 simulations (as listed in Table 2) compared to the base case simulation in which nucleation is
8 turned off (NONUC). Both boundary layer spatial distributions (left column) and zonal
9 averages (right column) are shown. Table 3 shows the global, annual average percent changes
10 for each of the sensitivity simulations. The N_{10} percent differences between the binary and the
11 NONUC simulations, as seen in panels A and B of Fig. 2, are quite small in the boundary
12 layer. In fact, there are some regions of percent decrease in N_{10} , such as off the western coast
13 of South America. The reason for this decrease is likely due to the demand for condensable
14 vapors by nucleated particles that are lost by coagulation to larger particles before they grow
15 to 10 nm. This demand for condensable vapor limits the ability of the primary particles to
16 grow and survive.

17 Panel B of Fig. 2 shows that while the binary nucleation parameterization of Vehkamäki et al.,
18 (2002) does not produce a large fractional increase in N_{10} in the boundary layer, it does have a
19 more pronounced fractional effect in and around the tropical upper troposphere.

20 Panels C and D of Fig. 2 show the percent changes for the ACT2 simulation. The ACT1 maps
21 are not shown but are similar to ACT2. One major characteristic of the ACT2 simulation is
22 that the activation mechanism predicts a large enhancement of N_{10} in the boundary layer,
23 especially over the oceans where increases from nucleation may not be expected. This is
24 partially due to the activation simulation lacking a third nucleating species that helps regulate
25 whether or not nucleation occurs in certain environments. For example, in TER simulations,
26 low amounts of ammonia over the oceans prevent nucleation from occurring. The zonal plot
27 shows the strong N_{10} enhancements near the surface. Above about 400 hPa, the binary
28 nucleation scheme (same as in the BHN simulation), which is also used in the ACT2
29 simulation, dominates the N_{10} increases, as evidenced by the similarities in panels B and D.

30 Panels E and F show the percent enhancements for TER5. TER3 is omitted here due to the
31 spatial patterns being similar (albeit with different intensities in particle number

1 concentration). The TER5 simulation gets similar N_{10} increases across the continents as
2 ACT2, but without the large increases over the oceans. The nucleation contribution to N_{10} is
3 larger near the surface and decreases in importance with altitude.

4 Panels G and H show the enhancements for TER. Past results have found overprediction in
5 nucleation rates and N_{10} concentrations when using the ternary parameterization without a
6 scaling factor (Jung et al., 2010). Not surprisingly, the largest continental enhancements occur
7 in the TER sensitivity simulation. It is interesting to note, however, that TER is not the
8 highest simulation in N_{10} enhancements over the ocean (ACT2 has the larger contribution)
9 due to the generally low concentrations of ammonia over the oceans in the model.

10 Figure 3 is the same as Fig. 2 except the enhancements are calculated for CCN(0.2%) instead
11 of N_{10} . Figure 3 shows that in most polluted regions of the world, all nucleation schemes we
12 simulated at least double the number of CCN compared to the NONUC simulation.
13 Interestingly, the CCN(0.2%) spatial maps across the four sensitivity studies are all relatively
14 similar and uniform, with exceptions mostly in the Middle East and the southern United
15 States. The uniformity between the different nucleation mechanisms suggests that the
16 influence of each specific nucleation scheme is only moderately important. There are no
17 instances of decreases in either the spatial distributions or the zonal plots.

18 The BHN scenario (panels A and B of Fig. 3) has a weak relative increase in boundary layer
19 N_{10} (23% global average, see Table 3), yet has a larger relative increase for boundary layer
20 CCN(1.0%) (27%) and CCN(0.2%) (49%). Although seemingly counterintuitive, this is due
21 to nucleation aloft in the free troposphere and subsequent growth to larger sizes during
22 downward subsidence before entraining into the boundary layer. Thus, the absolute increase
23 in N_{10} and CCN(0.2%) in the boundary layer are similar because a large fraction the nucleated
24 particles are CCN(0.2%) sizes by the time they reach the boundary layer (and the boundary-
25 layer CCN(0.2%) concentrations are lower than the boundary layer N_{10} concentrations).
26 However, because the CCN(0.2%) concentrations are $\sim 10x$ lower, the relative enhancement of
27 CCN(0.2%) in the boundary layer are $\sim 10x$ larger. When averaging through the entire
28 troposphere, the N_{10} increase is larger than both CCN(1.0%) and CCN(0.2%), as is expected.
29 This implies that at least in the BHN scheme, nucleation in the free troposphere can be a
30 major source of CCN in the boundary layer, which has also been observed in the GLOMAP
31 model (Merikanto et al., 2009).

1 The ACT2 simulation, shown in panels C and D of Fig. 3, has stronger CCN(0.2%)
2 enhancements in some regions across the boundary layer compared to the BHN simulation.
3 Since the ACT2 simulation is coupled with BHN, we see large increases in CCN(1.0%) and
4 CCN(0.2%) (see Table 3) even in the free troposphere. In these simulations, CCN
5 enhancement is due to both effects of BHN as described above as well as the boundary layer
6 enhancement provided by the activation simulation.

7 The TER5 boundary layer and zonally averaged plots, seen in panels E and F of Fig. 3, look
8 similar to ACT2. However, the TER5 simulation has less of a pronounced increase across the
9 Pacific Ocean in between the tropics. As a result, TER5 has a slightly smaller CCN percent
10 increase when compared to ACT2 (see Table 3). Unlike binary nucleation, ternary nucleation
11 is dominant in the boundary layer and weak in the free troposphere. The boundary layer
12 average percent increase in CCN(0.2%) is 56%, confirming some importance of boundary
13 layer nucleation for TER5. Finally, the TER simulations have the strongest CCN
14 enhancements for the boundary layer, as seen in Fig. 2 panel G as well as Table 3 due to the
15 unrealistically high nucleation rates in these simulations (see Fig. 4).

16 In all of the results plotted in Figs. 2 and 3, it appears that the fractional CCN(0.2%) increase
17 in the free and upper troposphere (up to 200 hPa) is larger than the N_{10} enhancement.
18 Although physical explanations such as aerosol transport through deep convection are
19 plausible, this is likely an artifact of the differencing against the control simulation, NONUC.
20 Without nucleation active (in the NONUC simulation), there is no particle source in the
21 tropical free troposphere and thus CCN(0.2%) concentrations are small (sometimes less than
22 $10 \text{ particles cm}^{-3}$). Thus, a small addition of particles due to nucleation in the various
23 simulations can lead to large relative increases. The addition of N_{10} from the nucleation-active
24 simulations is compared against a relatively higher N_{10} control value than is expected. In
25 summary, globally averaged CCN(0.2%) increases range from about 49-78% for the boundary
26 layer above the NONUC case. The TER5 simulation, which has performed well against
27 observations, had a 56% increase in boundary layer-averaged CCN compared to a simulation
28 without nucleation active, but this enhancement only increases to 78% for a much faster
29 nucleation scheme (TER). The range of these increases is small considering the nucleation
30 rates varied by as many as 4 orders of magnitude, as seen in Fig 4 and discussed below.

31 Figure 4 shows globally averaged CCN(0.2%) and CCN(1.0%) concentrations against
32 tropospheric average nucleation rates (J_1). The NONUC control simulation gives us a rough

1 estimate of the primary CCN, which is represented by the dashed black lines for comparison.
2 For the base SOA simulation (panel A), primary CCN(0.2%) are roughly 100 cm^{-3} globally
3 averaged, whereas CCN(1.0%) are about twice that. Consistent with the global map results
4 shown in Figs. 2 and 3, CCN(0.2%) do not appear to be very sensitive to large changes in the
5 nucleation rate. Specifically, the black and blue triangles in Fig. 4, representative of the BHN
6 and ION simulations, respectively, are about 4 orders of magnitude lower in the nucleation
7 rate than the TER simulation (furthest right red triangle). However, the CCN(0.2%) increases
8 from the BHN to TER cases are comparatively small at $40 \text{ particles cm}^{-3}$, or a 29% increase.
9 On the other hand, the slowest nucleation mechanism, BHN, gives a global boundary layer
10 enhancement of 49% over the NONUC simulation. From this, we conclude that the details of
11 a particular nucleation theory seem to be less important than the use of any theory within the
12 global model for predicting CCN(0.2%). Although this 29% sensitivity to nucleation
13 mechanism is not insignificant enough to ignore, it is important to note that it is smaller than
14 the sensitivity between any nucleation mechanism compared to a no-nucleation case. The
15 insensitivity is also evidenced by the green solid lines, which represent the two activation
16 cases (ACT1 and ACT2), and the red solid lines, which signify TER5, TER3, and TER. Both
17 of these lines are relatively flat for CCN(0.2%), indicating a small increase in CCN for a large
18 increase in nucleation. Additionally, the 29% cited above represents the largest possible range
19 in CCN changes. Although no nucleation mechanism is without faults, the CCN(0.2%)
20 sensitivity between two more reasonable nucleation simulations (for example, TER3 and
21 TER5) is much smaller (less than 10%).

22 As expected, CCN(1.0%) is more sensitive to changes in nucleation rate, as evidenced by the
23 steeper slopes in the CCN(1.0%) data (square markers). For example, between the TER5 and
24 TER simulation, roughly a 2 order-of-magnitude increase in the nucleation rate results in a
25 $200 \text{ particles cm}^{-3}$ increase in the CCN(1.0%) concentrations. The ACT1 and ACT2
26 simulations also show an increased CCN sensitivity to the same changes in nucleation. The
27 difference in globally averaged nucleation rate between BHN and ION is only about 0.01 cm^{-3}
28 s^{-1} , but the change in CCN(1.0%) is $70 \text{ particles cm}^{-3}$. This difference is likely due to
29 differences in the spatial dependence of nucleation. For these two simulations, the specific
30 nucleation mechanism does matter, since BHN and ION result in very similar nucleation rates
31 but different CCN. This implies that in regions where ION nucleation dominates the survival
32 probabilities are higher than in regions where BHN dominates, on average. In particular, the
33 mid-to-upper troposphere has the largest differences between BHN and ION in CCN

1 concentrations (not shown). This is consistent with recent work suggesting ion-based
2 nucleation may be relevant in the upper troposphere regions where colder temperatures
3 prevail (Kirkby et al., 2011). Although the sensitivities are low for most of the other
4 simulations, the mechanisms are still important for other reasons, such as spatial and temporal
5 variability. Spatial effects are clearly important as seen in the ACT2 and TER5 simulations in
6 which ACT2 predicts large N_{10} and CCN(0.2%) enhancements over the oceans (Figs. 2 and
7 3). Additionally, although aerosol indirect forcing cannot be estimated with the current model
8 setup, nucleation impacts on aerosol forcing are likely larger in certain regions. Future work
9 should seek to explore the sensitivity of forcing to nucleation. It also appears that mechanisms
10 matter for seasonal and daily variability in nucleation event frequency, growth rates,
11 nucleation rates, survival probabilities, and CCN formation (see Figs. 8 and 9 and Sect. 3.2.2,
12 and Westervelt et al. (2013)).

13 Finally, we include CCN-nucleation rate results in Fig. 4 for the same nucleation mechanisms
14 but with an extra 100 Tg yr^{-1} of SOA available for condensation (panel B). We find that these
15 cases (using the same color and symbol scheme as the base SOA simulations) result in higher
16 CCN(1.0%) and CCN(0.2%), due to the increased condensational growth. For CCN(0.2%),
17 the increase between low and high SOA simulations increases as the nucleation rate increases.
18 In other words, specifically for the TER5-TER3-TER line, the slope of the high SOA line (red
19 in panel B) is greater than the slope of the low SOA line (panel A). This suggests that the
20 additional SOA is able to grow nucleated particles to CCN activation sizes that would not
21 otherwise be activated and make survival probabilities less sensitive to changes in nucleation
22 rates. However, the change in the slope of CCN(1.0%) is small because nucleated particles
23 require less growth to reach CCN(1.0%) sizes.

24 **3.2 Microphysical feedbacks responsible for lower survival probabilities at** 25 **higher nucleation rates**

26 Despite introducing large differences in nucleation rates using the various parameterizations
27 listed in Table 2, CCN concentrations are only modestly increased. Figure 5 shows
28 qualitatively how an increase in the average nucleation rate in the model can ultimately cause
29 a negative feedback loop that minimizes CCN changes. Larger nucleation rates increase the
30 Fuchs aerosol surface area, which increases the coagulation and condensation sinks. The
31 increased condensation sink reduces condensable vapor concentrations, which slows down

1 particle growth. This slowed growth, in combination with increased coagulation, reduces the
2 survival probability and limits CCN formation.

3 The following three sections (Sects. 3.2.1, 3.2.2, and 3.2.3) provide the quantitative evidence
4 for the feedback phenomenon we have qualitatively described above. Small particles are not
5 traditionally thought to have a large impact on aerosol surface area (Fuchs) due to the square-
6 dependence on diameter. However, when considering that modeled nucleation events are not
7 isolated occurrences and can have compounding impacts that feedback on each other, the
8 effect is significant enough to increase the condensation sink, especially when the growth of
9 these nucleated particles to Aitken and accumulation modes is considered. There are two main
10 reasons why nucleated particles can contribute to the aerosol surface area enough to buffer the
11 CCN concentrations. First, with large increases in the nucleation rate, the CCN mode will
12 shift to smaller sizes over the long-term of many nucleation events due to a larger presence of
13 nucleated particles. This allows nucleated particles to contribute a larger fraction of surface
14 area than they might otherwise without the nucleation rate enhancements. For constant aerosol
15 mass production rates, an increase in CCN number concentration from nucleation leads to
16 aerosol Fuchs surface area enhancements, much like how smaller and more numerous cloud
17 droplets result in larger cloud surface area with a fixed water vapor budget (cloud albedo
18 aerosol indirect effect, or Twomey effect). Second, even in the short-term, for very high
19 nucleation rates, there is a non-negligible surface area increase associated with fresh nuclei
20 themselves. Figures 6-9 and the following sections will provide quantitative evidence for the
21 increase in surface area and condensation sink due to increases in the average nucleation rate.

22 3.2.1 Sample nucleation day

23 We first present a case study of a single day: April 3 at Hyytiälä. The first row of Fig. 6
24 (panels A, B, and C) shows three similar plots of the number size distribution evolution in
25 time (“banana plots”). Color contours represent the logarithm (base 10) of the number size
26 distribution function ($\log_{10}(dN/d\log D_p)$). The initial background size distributions are
27 different between the three simulations due to differences in nucleation, growth and survival
28 probability and previous days (shown quantitatively in Fig. 7). The black line in each panel of
29 Fig. 6 demarcates the 50 nm size threshold. Notice that within the first day, only the TER5
30 simulation (panel A), reaches the 50 nm size considered to be relevant for CCN. In panels B
31 and C, the growing nucleation mode does reach the 50 nm size, but not until the second or
32 third day after nucleation. Panel D shows the nucleation rate (J_3) for each of the sensitivity

1 cases for April 3, panel E shows the mean growth rates, and panel F shows the survival
2 probability to 50 nm. As the nucleation rates increase from TER5 to TER3 and TER, both
3 growth rates and survival probabilities decrease. This is not only due to feedbacks occurring
4 during this particular event, but also includes feedbacks from previous nucleation events, as
5 discussed previously. On this particular day, the survival probability is over 90% in TER5 and
6 less than 10% in TER3 and TER.

7 Figure 7 shows the condensation sink over the course of the nucleation event and Fuchs
8 surface area size distributions for the sample day at Hyytiälä. For all three ternary simulations,
9 the overall condensation sink increases throughout the day (panel A) consistent with
10 condensational growth of the particle size distribution. The TER and TER3 simulations have a
11 short-lived enhancement in condensation sink that corresponds with the onset of the
12 nucleation event; the same enhancement is not visible in the TER5 simulation. This indicates
13 that, at least under very high nucleation rates, there is an immediate impact of nucleation on
14 Fuchs surface area and condensation sink even if it is short-lived. It can be seen in Fig. 7
15 panel B that, during the middle of the nucleation event, the nucleation mode (centered at 10
16 nm at noon) makes a significant contribution to Fuchs surface area, but this goes away after a
17 few hours as these particles coagulate. It is also important to note that the condensation sink in
18 the TER simulations starts and remains higher than both the TER3 and TER5, which shows
19 that prior nucleation events in the long-term affect the initial condensation sink at the start of a
20 new nucleation event. The reason for the higher starting condensation sink in the TER
21 simulation is the same as described in the introduction: for a fixed budget of condensable
22 aerosol mass, a higher particle concentrations imply a shift of the CCN model to smaller sizes
23 and an increase in its surface area. This can be seen in panel B of Fig. 7, where the
24 accumulation mode Fuchs surface area is higher in TER5 compared to TER. Although Fig. 7
25 panel B shows this for noon during the event, the Fuchs surface area was already higher at the
26 beginning of the day. Comparing the two more realistic nucleation rates, TER5 vs. TER3, this
27 persistent increase in condensation sink is more important than the short-term contribution
28 from the nucleation mode.

29 Finally, using the nucleation rate (J_3) and the survival probability (SP) to 50 nm we can
30 calculate the CCN formation rate at 50 nm (J_{50}). For this particular event, the J_{50} for TER5,
31 TER3, and TER are 0.09, 0.1, and 0.11 $\text{cm}^{-3} \text{s}^{-1}$, respectively. Despite order-of-magnitude
32 variation in nucleation rates, the decreased survival probabilities in TER and TER3

1 compensate for the differences, resulting in similar values of CCN formation rate. Although
2 the fastest nucleation rates in the TER simulation lead to the largest CCN formation rates,
3 these CCN formation rates are only marginally (10-20%) larger than the slower nucleation
4 cases, TER3 and TER5.

5 3.2.2 Full year of nucleation events

6 To further quantify the dampening of CCN changes to changes in nucleation, we expand our
7 discussion with a yearlong time series of nucleation events in the TER5, TER3, and TER
8 simulations at the sample location of Hyytiälä, Finland. The TER5 simulation was already
9 considered in Westervelt et al. (2013), but we add in the TER3 and TER simulations to allow
10 us to look at how incremental changes in the nucleation rates effect growth rates, coagulation
11 sink (for 3 nm particles), condensation sink, survival probabilities, and ultimately CCN
12 concentrations. Figure 8 shows cumulative distribution functions for the three ternary
13 simulations at Hyytiälä. Panel A shows roughly order-of-magnitude increases in J_3 from the
14 TER5 to TER3 to TER simulations (note that the nucleation rate scale factors are two and
15 three orders of magnitude apart). This increase in nucleation causes an increase in both the
16 coagulation and condensation sinks, which follow the same order of $TER > TER3 > TER5$
17 (panels C and D). Because of the enhancement of the condensation sink due to the increased
18 Fuchs surface area from faster nucleation rates, condensable vapor concentrations are reduced.
19 Because of the reduction in condensable vapor concentrations, the TER simulation has the
20 slowest growth rates by as much as a factor of 2 (Panel B). The TER5 simulation, on the other
21 hand, has fast growth rates that allow particles to more effectively survive to CCN sizes (50 or
22 100 nm) within one day, as was the case for the April 3rd event shown above.

23 Figure 9 shows CDFs for survival probabilities and CCN formation rates. Survival
24 probabilities are calculated for multi-day growth, as is explained in Section 2.2.3 Because of
25 the increased coagulation and condensation sink, particle survival to 50 or 100 nm (panels A
26 and C) is diminished in the TER3 and TER simulations compared to the TER5. The survival
27 probabilities tend to be quite sensitive to the growth rate changes seen in Fig. 8. As a result,
28 large increases in the nucleation rate (Fig. 8a) are offset by low survival probabilities (e.g.
29 more than half the cases for TER have <1% chance of surviving to grow to 100 nm). As a
30 result, the annual-mean J_{50} CCN formation rates from Hyytiälä nucleation are within 5% of
31 each other for the three sensitivity cases (Table 4) despite a large change in the nucleation pre-
32 factor and more than a six-fold increase in the J_3 nucleation rate. For the 100 nm threshold,

1 annual-mean CCN formation rates due to nucleation are still within 10% of each other across
2 simulations. While it is true that higher nucleation rates tend to result in higher CCN
3 concentrations, the decreases in survival probability offset most of the increases in nucleation
4 rates.

5 3.2.3 Linking changes in condensation sink to changes in survival probability

6 Although we have shown in the previous two sections that the condensation sink increases
7 with faster nucleation rates to diminish survival probabilities, we have yet to directly link
8 condensation sink to growth rates, coagulation sinks, and survival probability. Figure 10a is a
9 scatterplot of simulated growth rates for one year at Hyytiälä. The red circles plotted represent
10 the values of the TER growth rates scaled by the ratio of the condensation sinks in TER and
11 TER5. We find that the growth rates for TER5 and “scaled TER” (which is equal to the
12 product of the TER growth rate and the ratio of the TER and TER5 condensation sinks, i.e.
13 $CS_{ter}/CS_{ter5} \times GR_{ter}$), are now in good agreement (within 10% of each other on average).

14 This suggests that the growth rate decrease from the TER5 to the TER simulations can largely
15 be explained by the increases in the condensation sink. In Fig. 10b, we find that the
16 coagulation sink and the condensation sink are strongly correlated in our model simulations.
17 This result has been seen in ambient measurements (Gong et al., 2010; Lehtinen et al., 2007;
18 Dal Maso et al., 2002) and is consistent with the idea that larger particles wield the biggest
19 influence in both the uptake of condensable vapors and the collision and combination of
20 particles. Therefore, changes in condensation sink can be used as proxies for changes in both
21 factors that determine survival probabilities: growth rates and coagulation scavenging
22 frequencies.

23 Figs. 10c and d show the relationship between survival probability (to 50 nm, SP50) and
24 condensation sink. In panel C, we plot survival probability as a function of condensation sink
25 for the TER5 simulations and notice a strong nonlinear decreasing trend, consistent with our
26 conceptual arguments and the theory presented in Sect. 2.4 above. Since the survival
27 probability is expected to scale with the exponential of the square of the condensation sink in
28 the absence of large variability in the production rate of condensable vapors (Eq. 7), we
29 attempt to fit such a function to our model output. With a tuned value of α , we find fairly
30 consistent agreement ($R^2=0.9$) for the negative exponential fit. Panel D is the same as C
31 except we plot model output from the TER simulation and compare it to the same fit derived

1 for the TER5 case (same equation as in panel C). We note that although there are some
2 discrepancies, the TER5 model fit still represents the TER output well ($R^2=0.73$). This
3 supports our claim that, to first order, the decrease in survival probability can be quantified
4 simply from basic theory and the change in the condensation sink. This is also consistent with
5 our claim that CCN concentrations do not change much with changes in nucleation theory or
6 nucleation rate, since the same fit applies, in this specific example at least to first order, to two
7 nucleation schemes with vastly different nucleation rates (see Fig. 8).

8 There are three likely reasons for the slight discrepancy in the survival probabilities and
9 condensation sinks between our model output and simple theory. First and foremost, if there is
10 variability in the production rate of condensable vapors (which are not accounted for in the
11 simplified fit), this will lead errors in the fit. There is generally little scatter in the data in
12 panels C and D of Fig. 10, which leads us to believe that the production rate generally does
13 not vary randomly between event days. However, we found that in the GEOS-Chem-TOMAS
14 model at this particular location, the production rate of condensable vapors is lower on days
15 with lower condensation sinks (i.e. a clean aerosol background correlates with a clean gas-
16 phase background). This correlation may be responsible for the regions of poor fit in Fig. 10d
17 for the TER simulation. The survival probability is lower in GEOS-Chem-TOMAS than the
18 simplified model at low condensation sinks, which may be due to lower condensable vapor
19 production when the condensation sink is low, which would lead to slower growth rates and
20 lower survival probabilities. However, the simple model overprediction at low condensation
21 sinks is not evident in Fig. 10c for the TER5 simulation. This lack of overprediction maybe
22 due to nucleation being less likely in the TER5 simulation than the TER simulation during
23 low vapor concentrations due to the orders-of-magnitude slower nucleation rates in TER5
24 (Fig. 4), or alternatively it may be due to the production rate of condensable vapors being less
25 significant as the survival probability approaches 1 (as it does for small values of
26 condensation sink in the TER5 in Fig. 10c; this is not the case for the TER simulation).
27 Second, condensation sink varies throughout the day (as seen in Fig. 7), but here we use the
28 24-hour average in Eq. 7 to predict changes in survival probability. Since the condensation
29 sink is larger shortly after a nucleation burst, our model points (GEOS-Chem-TOMAS) may
30 be shifted slightly to the right in Fig. 10d, leading to better agreement. Third, the correlation
31 between CoagS and CS assumed by our theory is mostly valid only for the kinetic regime and
32 may not be applicable to the transition regime (~ 100 nm). If the CoagS is actually smaller
33 than predicted by our simple theory, the model output survival probability may agree better.

1 Despite these shortcomings, the simple relationship between CS and SP is broadly confirmed
2 within the GEOS-Chem-TOMAS framework. However, we have yet to confirm if this
3 relationship holds for field data where it is possible that the production rate of condensable
4 vapors varies more strongly and independently from the condensation sink.

5 **4 Conclusions**

6 Using the GEOS-Chem-TOMAS global aerosol microphysics model, we performed a series
7 of 16 simulations with 8 differing nucleation mechanisms (or pre-factors) and two SOA
8 budgets in order to test the sensitivity of CCN to nucleation and growth. In addition to
9 determining the sensitivity, we analyzed one year of modeled nucleation events at Hyytiälä in
10 order to explain and quantify the reasons for the apparently modest sensitivity of CCN to
11 nucleation. We calculated modeled growth rates, condensation sinks, coagulation sinks,
12 survival probabilities, and CCN formation rates, and used these properties to explain the
13 microphysical feedbacks that lead to a buffered response in CCN due to increased nucleation
14 rates. Global modeling results showed that N_{10} is generally more sensitive to nucleation than
15 $CCN(0.2\%)$ consistent with the fact that nuclei are lost by coagulation during growth but
16 some primary particles are emitted already at or near CCN sizes. The global, boundary layer
17 averaged increase in N_{10} due to adding nucleation schemes to a simulation with no nucleation
18 varied from 23% in the BHN scheme to 190% in the TER5 scheme. Spatially, the activation
19 simulation (ACT2) predicted large increases in N_{10} in most boundary-layer locations,
20 including over the oceans, while other cases (binary (BHN), ternary with the 10^{-5} prefactor
21 (TER5)) predicted less dramatic increases limited mostly to the continents. In particular, the
22 lack of sufficient boundary layer nucleation in the BHN simulation caused BHN to be the
23 least sensitive in N_{10} to nucleation.

24 Enhancements of $CCN(0.2\%)$ above the NONUC simulation were less sensitive to the choice
25 of nucleation parameterization, ranging from 49-78% average increases in the boundary layer.
26 This 29% difference in $CCN(0.2\%)$ is non-negligible, but not especially large considering the
27 large uncertainties (factors of 2) in other aerosol indirect effect predictions. Because our
28 results showed that binary nucleation alone increases $CCN(0.2\%)$ almost as much as ternary
29 or activation nucleation, this implies that free tropospheric nucleation contributes a significant
30 amount of the nucleation-derived CCN, similar to what was found in (Merikanto et al., 2009).

31 We explored the details of the aerosol physics in the global model to explain further the weak
32 sensitivity of CCN to the specific nucleation mechanism and nucleation rate. Given a fixed

1 budget of condensable aerosol mass (sulfate and SOA), any increase in CCN concentrations
2 due to nucleation and growth imply that the CCN or accumulation mode shifts to smaller
3 sizes. Therefore, overall surface area is higher, leading to two negative feedbacks on
4 subsequent nucleation and growth events. First, higher surface area means higher
5 condensation sink and slower growth rates. Second, higher condensation sink correlates with
6 faster coagulation scavenging. The net result of these two feedbacks is that it takes nucleated
7 particles longer to grow to CCN sizes, during which time they are subject to more intense
8 coagulation scavenging. Therefore, with faster nucleation mechanisms, nuclei grow more
9 slowly and are exposed to higher coagulation loss frequencies for longer periods of time,
10 limiting their survival to CCN sizes, and reducing the impact that nucleation rates would
11 otherwise have on CCN concentrations. Both feedbacks lower survival probabilities, and the
12 net effect is often dramatic. We showed that the full three-dimensional model's predicted
13 decrease in survival probability with higher nucleation rates quantitatively matched what
14 would be expected using a simple theory that accounts for these two feedbacks.

15 We use one year of model output for Hyytiälä as an example of these feedback processes at
16 work. We analyzed one year of size distribution output for three ternary simulations, TER5,
17 TER3, and TER with different pre-factors tuning their nucleation rates. These three scenarios
18 represented incremental changes in the nucleation rates that allowed us to observe effects of
19 increased nucleation on other quantities that effect growth to CCN sizes. Specifically, the
20 growth rates in the faster nucleation cases (TER, TER3) were diminished by 50% on average
21 because of an increased sink of condensable vapors. Particle survival probability was
22 decreased due to the decrease in growth rate and increase in coagulation, evening out the CCN
23 formation rates across the three simulations. The faster nucleation mechanisms (TER, TER3)
24 did have higher CCN formation rates, but the annual-average CCN formation rates
25 attributable to nucleation were within 10% of each other at both 50 nm and 100 nm CCN size
26 thresholds.

27 We explored the relationship between CCN concentrations and nucleation rates, J_1 , on a
28 global scale for all of the nucleation scenarios plus an additional series of simulations that
29 included an extra 100 Tg yr^{-1} of SOA available for condensation. The boundary layer
30 $\text{CCN}(0.2\%)$ without the extra SOA varied only by $40 \text{ particles cm}^{-3}$ (29%) between all of the
31 nucleation simulations—a nucleation rate range of about 4 orders of magnitude. The
32 sensitivity range drops to $\sim 12\%$ when we eliminate nucleation parameterizations (BHN and

1 TER) known to exhibit serious biases compared to observed nucleation rates. Despite this,
2 even the slowest nucleation simulation (BHN) increased CCN by about 50% over the
3 NONUC control simulation. This suggests that the presence of any nucleation theory in the
4 model is has a larger effect than the differences resulting from choice of nucleation
5 parameterization. There are interesting exceptions, however, such as the BHN and ION
6 simulations, which have nearly identical global-average nucleation rates but noticeably
7 different CCN formation, implying that survival probability is different depending on when
8 and where nucleation is occurring. With extra SOA, CCN(0.2%) are much more sensitive to
9 nucleation, indicating that particles that normally would not reach activation sizes are indeed
10 reaching those sizes when additional SOA is available to enhance growth. We also find that
11 CCN(1.0%) are more sensitive to nucleation due to the capability of smaller particles (30 – 60
12 nm, depending on composition) to be activated and the greater ease with which nuclei can
13 reach these moderate sizes.

14 Our model results indicate that nucleation makes a significant contribution to aerosol number
15 concentrations, including CCN, but that boundary layer CCN are not very sensitive to the
16 choice of nucleation parameterization once obvious outliers (BHN and TER) are excluded.
17 Comparing the results of BHN and ACT simulations, it is clear that much of the contribution
18 to boundary layer CCN results from nucleation in the free troposphere, consistent with
19 previous work (Merikanto et al., 2009). The overall modest impact is due to a dampening
20 effect on CCN from an increased aerosol Fuchs surface area, subsequent loss of condensable
21 vapors, and the accompanying increase in coagulation scavenging. While sensitivity studies
22 such as the one we have presented here are useful, they are limited by the nonlinear nature of
23 aerosol model processes. Future work must be aimed at directly determining the nucleation
24 contribution to CCN through other methods.

25 **Acknowledgements**

26 This research was pursued with funds from the US Environmental Protection Agency under
27 EPA Agreement RD-83337401 and STAR Grant RD-83503501. We acknowledge the
28 Atlantic Computational Excellence Network (ACEnet) for providing the computational
29 resources required for this work.

30

31

1 References

- 2 Adams, P. J. and Seinfeld, J. H.: Predicting global aerosol size distributions in general
3 circulation models, *J Geophys Res-Atmos*, 107(D19), -, doi:Artn 4370 Doi
4 10.1029/2001jd001010, 2002.
- 5 Albrecht: Aerosols, Cloud Microphysics, and Fractional Cloudiness, *Science* (80-),
6 245(4923), 1227 –1230, doi:10.1126/science.245.4923.1227, 1989.
- 7 Almeida, J., Schobesberger, S., Kürten, A., Ortega, I. K., Kupiainen-Määttä, O., Praplan, A.
8 P., Adamov, A., Amorim, A., Bianchi, F., Breitenlechner, M., David, A., Dommen, J.,
9 Donahue, N. M., Downard, A., Dunne, E., Duplissy, J., Ehrhart, S., Flagan, R. C., Franchin,
10 A., Guida, R., Hakala, J., Hansel, A., Heinritzi, M., Henschel, H., Jokinen, T., Junninen, H.,
11 Kajos, M., Kangasluoma, J., Keskinen, H., Kupc, A., Kurtén, T., Kvashin, A. N., Laaksonen,
12 A., Lehtipalo, K., Leiminger, M., Leppä, J., Loukonen, V., Makhmutov, V., Mathot, S.,
13 McGrath, M. J., Nieminen, T., Olenius, T., Onnela, A., Petäjä, T., Riccobono, F., Riipinen, I.,
14 Rissanen, M., Rondo, L., Ruuskanen, T., Santos, F. D., Sarnela, N., Schallhart, S.,
15 Schnitzhofer, R., Seinfeld, J. H., Simon, M., Sipilä, M., Stozhkov, Y., Stratmann, F., Tomé,
16 A., Tröstl, J., Tsagkogeorgas, G., Vaattovaara, P., Viisanen, Y., Virtanen, A., Vrtala, A.,
17 Wagner, P. E., Weingartner, E., Wex, H., Williamson, C., Wimmer, D., Ye, P., Yli-Juuti, T.,
18 Carslaw, K. S., Kulmala, M., Curtius, J., Baltensperger, U., Worsnop, D. R., Vehkamäki, H.
19 and Kirkby, J.: Molecular understanding of sulphuric acid-amine particle nucleation in the
20 atmosphere., *Nature*, 502(7471), 359–63, doi:10.1038/nature12663, 2013.
- 21 Auvray, M. and Bey, I.: Long-range transport to Europe: Seasonal variations and implications
22 for the European ozone budget, *J. Geophys. Res.*, 110, 22 PP.,
23 doi:200510.1029/2004JD005503, 2005.
- 24 Ball, S. M., Hanson, D. R., Eisele, F. L. and McMurry, P. H.: Laboratory studies of particle
25 nucleation: Initial results for H₂SO₄, H₂O, and NH₃ vapors, *J. Geophys. Res.*, 104(D19), PP.
26 23,709–23,718, doi:199910.1029/1999JD900411, n.d.
- 27 Berndt, T., Böge, O., Stratmann, F., Heintzenberg, J. and Kulmala, M.: Rapid Formation of
28 Sulfuric Acid Particles at Near-Atmospheric Conditions, *Science* (80-), 307(5710), 698 –
29 700, doi:10.1126/science.1104054, 2005.
- 30 Bey, I., Jacob, D. J., Yantosca, R. M., Logan, J. A., Field, B. D., Fiore, A. M., Li, Q. B., Liu,
31 H. G. Y., Mickley, L. J. and Schultz, M. G.: Global modeling of tropospheric chemistry with
32 assimilated meteorology: Model description and evaluation, *J. Geophys. Res.*, 106(D19),
33 23073–23095, 2001.
- 34 Bzdek, B. R., Ridge, D. P. and Johnston, M. V.: Size-Dependent Reactions of Ammonium
35 Bisulfate Clusters with Dimethylamine, *J. Phys. Chem. A*, 114(43), 11638–11644,
36 doi:10.1021/jp106363m, 2010.
- 37 Chen, M., Titcombe, M., Jiang, J., Jen, C., Kuang, C., Fischer, M. L., Eisele, F. L., Siepmann,
38 J. I., Hanson, D. R., Zhao, J. and McMurry, P. H.: Acid-base chemical reaction model for
39 nucleation rates in the polluted atmospheric boundary layer., *Proc. Natl. Acad. Sci. U. S. A.*,
40 109(46), 18713–8, doi:10.1073/pnas.1210285109, 2012.

- 1 D'Andrea, S. D., Häkkinen, S. A. K., Westervelt, D. M., Kuang, C., Levin, E. J. T.,
2 Kanawade, V. P., Leaitch, W. R., Spracklen, D. V., Riipinen, I., and Pierce, J. R.:
3 Understanding global secondary organic aerosol amount and size-resolved condensational
4 behavior, *Atmos. Chem. Phys.*, 13, 11519-11534, doi:10.5194/acp-13-11519-2013, 2013.
- 5 Dal Maso, M., Kulmala, M., Riipinen, I., Wagner, R., Hussein, T., Aalto, P. P. and Lehtinen,
6 K. E. J.: Formation and growth of fresh atmospheric aerosols: eight years of aerosol size
7 distribution data from SMEAR II, Hyytiälä, Finland, *Boreal Environ. Res.*, 10(5), 323–336
8 [online] Available from: //000233128500002, 2005.
- 9 Dal Maso, M., Kulmala, M., Lehtinen, K. E. J., Mäkelä, J. M., Aalto, P. and O'Dowd, C. D.:
10 Condensation and coagulation sinks and formation of nucleation mode particles in coastal and
11 boreal forest boundary layers, *J. Geophys. Res.*, 107, 10 PP.,
12 doi:200210.1029/2001JD001053, 2002.
- 13 Donahue, N. M., Robinson, A. L., Stanier, C. O. and Pandis, S. N.: Coupled Partitioning,
14 Dilution, and Chemical Aging of Semivolatile Organics, *Environ. Sci. Technol.*, 40(8), 2635–
15 2643, doi:10.1021/es052297c, 2006.
- 16 Donahue, N. M., Trump, E. R., Pierce, J. R. and Riipinen, I.: Theoretical constraints on pure
17 vapor-pressure driven condensation of organics to ultrafine particles, *Geophys. Res. Lett.*,
18 38(16), L16801, doi:10.1029/2011GL048115, 2011.
- 19 Doyle, G. J.: Self-Nucleation in the Sulfuric Acid-Water System, *J. Chem. Phys.*, 35(3), 795,
20 doi:10.1063/1.1701218, 1961.
- 21 Erupe, M. E., Benson, D. R., Li, J., Young, L.-H., Verheggen, B., Al-Refai, M., Tahboub, O.,
22 Cunningham, V., Frimpong, F., Viggiano, A. A. and Lee, S.-H.: Correlation of aerosol
23 nucleation rate with sulfuric acid and ammonia in Kent, Ohio: An atmospheric observation, *J.*
24 *Geophys. Res.*, 115(D23), doi:10.1029/2010JD013942, 2010.
- 25 Facchini, M. C., Mircea, M. and Fuzzi, S.: Cloud albedo enhancement by surface-active
26 organic solutes in growing droplets, , 401(September), 257–259, 1999.
- 27 Gagné, S., Nieminen, T., Kurtén, T., Manninen, H. E., Petäjä, T., Laakso, L., Kerminen, V.-
28 M., Boy, M. and Kulmala, M.: Factors influencing the contribution of ion-induced nucleation
29 in a boreal forest, Finland, *Atmos. Chem. Phys.*, 10(8), 3743–3757, 2010.
- 30 Goldstein, A. H. and Gallaby, I. E.: Known and unexplored organic constituents in the earth's
31 atmosphere, *Environ. Sci. Technol.*, 41, 1514–1521, doi:10.1021/es072476p, 2007.
- 32 Gong, Y., Hu, M., Cheng, Y., Su, H., Yue, D., Liu, F., Wiedensohler, a., Wang, Z., Kalesse,
33 H. and Liu, S.: Competition of coagulation sink and source rate: New particle formation in the
34 Pearl River Delta of China, *Atmos. Environ.*, 44(27), 3278–3285,
35 doi:10.1016/j.atmosenv.2010.05.049, 2010.
- 36 Guenther, A., Karl, T., Harley, P., Wiedinmyer, C., Palmer, P. I. and Geron, C.: Estimates of
37 global terrestrial isoprene emissions using MEGAN (Model of Emissions of Gases and
38 Aerosols from Nature), *Atmos. Chem. Phys.*, 6(11), 3181–3210, 2006.

- 1 Hand, J. L., Kreidenweis, S. M., Sherman, D. E., Collett Jr., J. R., Hering, S. V., Day, D. E.,
2 and Malm, W. C.: Aerosol size distributions and visibility estimates during the Big Bend
3 regional aerosol and visibility observational (BRAVO) study, *Atmos. Environ.*, 36, 5043–
4 5055, doi:10.1016/S1352-2310(02)00568-x, 2002.
- 5 Heald, C. L., Coe, H., Jimenez, J. L., Weber, R. J., Bahreini, R., Middlebrook, a. M., Russell,
6 L. M., Jolleys, M., Fu, T.-M., Allan, J. D., Bower, K. N., Capes, G., Crosier, J., Morgan, W.
7 T., Robinson, N. H., Williams, P. I., Cubison, M. J., DeCarlo, P. F. and Dunlea, E. J.:
8 Exploring the vertical profile of atmospheric organic aerosol: comparing 17 aircraft field
9 campaigns with a global model, *Atmos. Chem. Phys.*, 11(24), 12673–12696, doi:10.5194/acp-
10 11-12673-2011, 2011.
- 11 Jung, J., Adams, P. J. and Pandis, S. N.: Simulating the size distribution and chemical
12 composition of ultrafine particles during nucleation events, *Atmos. Environ.*, 40(13), 2248–
13 2259, doi:10.1016/j.atmosenv.2005.09.082, 2006.
- 14 Jung, J., Fountoukis, C., Adams, P. J. and Pandis, S. N.: Simulation of in situ ultrafine particle
15 formation in the eastern United States using PMCAMx-UF, *J. Geophys. Res.*, 115, 13 PP.,
16 doi:201010.1029/2009JD012313, 2010.
- 17 Kerminen, V.-M.: Direct observational evidence linking atmospheric aerosol formation and
18 cloud droplet activation, *Geophys. Res. Lett.*, 32(14), L14803, doi:10.1029/2005GL023130,
19 2005.
- 20 Kirkby, J., Curtius, J., Almeida, J., Dunne, E., Duplissy, J., Ehrhart, S., Franchin, A., Gagne,
21 S., Ickes, L., Kurten, A., Kupc, A., Metzger, A., Riccobono, F., Rondo, L., Schobesberger, S.,
22 Tsagkogeorgas, G., Wimmer, D., Amorim, A., Bianchi, F., Breitenlechner, M., David, A.,
23 Dommen, J., Downard, A., Ehn, M., Flagan, R. C., Haider, S., Hansel, A., Hauser, D., Jud,
24 W., Junninen, H., Kreissl, F., Kvashin, A., Laaksonen, A., Lehtipalo, K., Lima, J., Lovejoy, E.
25 R., Makhmutov, V., Mathot, S., Mikkilä, J., Minginette, P., Mogo, S., Nieminen, T., Onnela,
26 A., Pereira, P., Petäjä, T., Schnitzhofer, R., Seinfeld, J. H., Sipilä, M., Stozhkov, Y.,
27 Stratmann, F., Tome, A., Vanhanen, J., Viisanen, Y., Virtala, A., Wagner, P. E., Walther, H.,
28 Weingartner, E., Wex, H., Winkler, P. M., Carslaw, K. S., Worsnop, D. R., Baltensperger, U.
29 and Kulmala, M.: Role of sulphuric acid, ammonia and galactic cosmic rays in atmospheric
30 aerosol nucleation, *Nature*, 476(7361), 429–433, doi:10.1038/nature10343, 2011.
- 31 Kuang, C., McMurry, P. H. and McCormick, A. V.: Determination of cloud condensation
32 nuclei production from measured new particle formation events, *Geophys. Res. Lett.*, 36,
33 doi:10.1029/2009GL037584, 2009.
- 34 Kuang, C., McMurry, P. H., McCormick, A. V. and Eisele, F. L.: Dependence of nucleation
35 rates on sulfuric acid vapor concentration in diverse atmospheric locations, *J. Geophys. Res.*,
36 113, 9 PP., doi:200810.1029/2007JD009253, 2008.
- 37 Kulmala, M., Kontkanen, J., Junninen, H., Lehtipalo, K., Manninen, H. E., Nieminen, T.,
38 Petäjä, T., Sipilä, M., Schobesberger, S., Rantala, P., Franchin, A., Jokinen, T., Järvinen, E.,
39 Äijälä, M., Kangasluoma, J., Hakala, J., Aalto, P. P., Paasonen, P., Mikkilä, J., Vanhanen, J.,
40 Aalto, J., Hakola, H., Makkonen, U., Ruuskanen, T., Mauldin, R. L., Duplissy, J., Vehkamäki,
41 H., Bäck, J., Kortelainen, A., Riipinen, I., Kurtén, T., Johnston, M. V, Smith, J. N., Ehn, M.,

- 1 Mentel, T. F., Lehtinen, K. E. J., Laaksonen, A., Kerminen, V.-M. and Worsnop, D. R.: Direct
2 observations of atmospheric aerosol nucleation., *Science*, 339(6122), 943–6,
3 doi:10.1126/science.1227385, 2013.
- 4 Kulmala, M. and Laaksonen, A.: Binary Nucleation of Water Sulfuric-Acid System -
5 Comparison of Classical-Theories with Different H₂SO₄ Saturation Vapor-Pressures, *J. Chem.*
6 *Phys.*, 93(1), 696–701, 1990.
- 7 Kulmala, M., Lehtinen, K. E. J. and Laaksonen, A.: Cluster activation theory as an
8 explanation of the linear dependence between formation rate of 3nm particles and sulphuric
9 acid concentration, *Atmos. Chem. Phys.*, 6(3), 787–793, doi:10.5194/acp-6-787-2006, 2006.
- 10 Kulmala, M., Pirjola, L. and Makela, J. M.: Stable sulphate clusters as a source of new
11 atmospheric particles, *Nature*, 404(6773), 66–69, doi:10.1038/35003550, 2000.
- 12 Kurtén, T., Loukonen, V., Vehkamäki, H. and Kulmala, M.: Amines are likely to enhance
13 neutral and ion-induced sulfuric acid-water nucleation in the atmosphere more effectively than
14 ammonia, *Atmos. Chem. Phys.*, 8(14), 4095–4103, doi:10.5194/acp-8-4095-2008, 2008.
- 15 Laakso, L., Gagné, S., Petäjä, T., Hirsikko, A., Aalto, P. P., Kulmala, M. and Kerminen, V.-
16 M.: Detecting charging state of ultra-fine particles: instrumental development and ambient
17 measurements, 2007.
- 18 Lee, L. A., Pringle, K. J., Reddington, C. L., Mann, G. W., Stier, P., Spracklen, D. V., Pierce,
19 J. R. and Carslaw, K. S.: The magnitude and causes of uncertainty in global model
20 simulations of cloud condensation nuclei, *Atmos. Chem. Phys.*, 13(17), 8879–8914,
21 doi:10.5194/acp-13-8879-2013, 2013a.
- 22 Lee, Y. H., Chen, K. and Adams, P. J.: Development of a global model of mineral dust
23 aerosol microphysics, *Atmos. Chem. Phys.*, 9(7), 2441–2458, 2009.
- 24 Lee, Y. H., Pierce, J. R. and Adams, P. J.: Representation of nucleation mode microphysics in
25 a global aerosol model with sectional microphysics, *Geosci. Model Dev.*, 6(4), 1221–1232,
26 doi:10.5194/gmd-6-1221-2013, 2013.
- 27 Lehtinen, K. E. J., Dal Maso, M., Kulmala, M. and Kerminen, V.-M.: Estimating nucleation
28 rates from apparent particle formation rates and vice versa: Revised formulation of the
29 Kerminen–Kulmala equation, *J. Aerosol Sci.*, 38(9), 988–994,
30 doi:10.1016/j.jaerosci.2007.06.009, 2007.
- 31 Mäkelä, J. M., Aalto, P., Jokinen, V., Pohja, T., Nissinen, A., Palmroth, S., Markkanen, T.,
32 Seitsonen, K., Lihavainen, H. and Kulmala, M.: Observations of ultrafine aerosol particle
33 formation and growth in boreal forest, *Geophys. Res. Lett.*, 24(10), PP. 1219–1222,
34 doi:199710.1029/97GL00920, 1997.
- 35 Makkonen, R., Asmi, A., Kerminen, V.-M., Boy, M., Arneth, A., Guenther, A. and Kulmala,
36 M.: BVOC-aerosol-climate interactions in the global aerosol-climate model ECHAM5.5-
37 HAM2, *Atmos. Chem. Phys.*, 12(21), 10077–10096, doi:10.5194/acp-12-10077-2012, 2012.

- 1 Makkonen, R., Asmi, A., Korhonen, H., Kokkola, H., Järvenoja, S., Räisänen, P., Lehtinen,
2 K. E. J., Laaksonen, A., Kerminen, V.-M., Järvinen, H., Lohmann, U., Bennartz, R., Feichter,
3 J. and Kulmala, M.: Sensitivity of aerosol concentrations and cloud properties to nucleation
4 and secondary organic distribution in ECHAM5-HAM global circulation model, *Atmos.*
5 *Chem. Phys.*, 9(5), 1747–1766, doi:10.5194/acp-9-1747-2009, 2009.
- 6 Manninen, H. E., Nieminen, T., Riipinen, I., Yli-Juuti, T., Gagné, S., Asmi, E., Aalto, P. P.,
7 Petäjä, T., Kerminen, V.-M. and Kulmala, M.: Charged and total particle formation and
8 growth rates during EUCAARI 2007 campaign in Hyytiälä, *Atmos. Chem. Phys. Discuss.*,
9 9(1), 5119–5151, doi:10.5194/acpd-9-5119-2009, 2009.
- 10 McMurry, P. H., Fink, M., Sakurai, H., Stolzenburg, M. R., Mauldin, R. L., Smith, J., Eisele,
11 F., Moore, K., Sjostedt, S., Tanner, D., Huey, L. G., Nowak, J. B., Edgerton, E. and Voisin,
12 D.: A criterion for new particle formation in the sulfur-rich Atlanta atmosphere, *J. Geophys.*
13 *Res.*, 110(D22), D22S02, doi:10.1029/2005JD005901, 2005.
- 14 Merikanto, J., Napari, I., Vehkamäki, H., Anttila, T. and Kulmala, M.: New parameterization
15 of sulfuric acid-ammonia-water ternary nucleation rates at tropospheric conditions, *J.*
16 *Geophys. Res.*, 112, 9 PP., doi:200710.1029/2006JD007977, 2007.
- 17 Merikanto, J., Spracklen, D. V., Mann, G. W., Pickering, S. J. and Carslaw, K. S.: Impact of
18 nucleation on global CCN, *Atmos. Chem. Phys.*, 9(21), 8601–8616, 2009.
- 19 Metzger, A., Verheggen, B., Dommen, J., Duplissy, J., Prevot, A. S. H., Weingartner, E.,
20 Riipinen, I., Kulmala, M., Spracklen, D. V., Carslaw, K. S. and Baltensperger, U.: Evidence
21 for the role of organics in aerosol particle formation under atmospheric conditions., *Proc.*
22 *Natl. Acad. Sci. U. S. A.*, 107(15), 6646–51, doi:10.1073/pnas.0911330107, 2010.
- 23 Napari, I., Kulmala, M. and Vehkamäki, H.: Ternary nucleation of inorganic acids, ammonia,
24 and water, *J. Chem. Phys.*, 117(18), 8418–8425, doi:10.1063/1.1511722, 2002.
- 25 Noppel, M., Vehkamäki, H. and Kulmala, M.: An improved model for hydrate formation in
26 sulfuric acid–water nucleation, *J. Chem. Phys.*, 116(1), 218, doi:10.1063/1.1423333, 2002.
- 27 Olivier, J. G. J., Bouwman, A. F., Van der Maas, C. W. M., Berdowski, J. J. M., Veldt, C.,
28 Bloos, J. P. J., Visschedijk, A. J. H., Zandveld, P. Y. J., and Haverlag, J. L.: Description of
29 EDGAR Version 2.0: A set of global emission inventories of greenhouse gases and ozone
30 depleting substances for all anthropogenic and most natural sources on a per country basis and
31 on 1×1 grid, National Institute of Public Health and the Environment (RIVM) report no.
32 771060 002, Netherlands Organization for Applied Scientific Research (TNO), 1996.
- 33 Paasonen, P., Nieminen, T., Asmi, E., Manninen, H. E., Petäjä, T., Plass-Dülmer, C., Flentje,
34 H., Birmili, W., Wiedensohler, A., Hõrrak, U., Metzger, A., Hamed, A., Laaksonen, A.,
35 Facchini, M. C., Kerminen, V.-M. and Kulmala, M.: On the roles of sulphuric acid and low-
36 volatility organic vapours in the initial steps of atmospheric new particle formation, *Atmos.*
37 *Chem. Phys.*, 10(22), 11223–11242 2010.

- 1 Pandis, S. N., Baltensperger, U., Wolfenbarger, J. K., Seinfeld, J.H., Inversion of aerosol data
2 from the epiphaniometer, *J. Aerosol Sci.*, 22, 417–428, doi:10.1016/0021-8502(91)90002-Y,
3 1991.
- 4 Pierce, J. R. and Adams, P. J.: Efficiency of cloud condensation nuclei formation from
5 ultrafine particles, *Atmos. Chem. Phys.*, 7(5), 1367–1379, doi:10.5194/acp-7-1367-2007,
6 2007.
- 7 Pierce, J. R. and Adams, P. J.: A Computationally Efficient Aerosol Nucleation/ Condensation
8 Method: Pseudo-Steady-State Sulfuric Acid, *Aerosol Sci. Technol.*, 43(3), 216,
9 doi:10.1080/02786820802587896, 2009a.
- 10 Pierce, J. R. and Adams, P. J.: Uncertainty in global CCN concentrations from uncertain
11 aerosol nucleation and primary emission rates, *Atmos. Chem. Phys.*, 9(4), 1339–1356 [online]
12 Available from: //000263642000018, 2009b.
- 13 Pierce, J. R., Chen, K. and Adams, P. J.: Contribution of primary carbonaceous aerosol to
14 cloud condensation nuclei: processes and uncertainties evaluated with a global aerosol
15 microphysics model, *Atmos. Chem. Phys.*, 7(20), 5447–5466, doi:10.5194/acp-7-5447-2007,
16 2007.
- 17 Pierce, J. R., Leaitch, W. R., Liggio, J., Westervelt, D. M., Wainwright, C. D., Abbatt, J. P.
18 D., Ahlm, L., Al-Basheer, W., Cziczo, D. J., Hayden, K. L., Lee, A. K. Y., Li, S.-M., Russell,
19 L. M., Sjostedt, S. J., Strawbridge, K. B., Travis, M., Vlasenko, A., Wentzell, J. J. B., Wiebe,
20 H. A., Wong, J. P. S. and Macdonald, A. M.: Nucleation and condensational growth to CCN
21 sizes during a sustained pristine biogenic SOA event in a forested mountain valley, *Atmos.*
22 *Chem. Phys.*, 12(7), 3147–3163, doi:10.5194/acp-12-3147-2012, 2012.
- 23 Pierce, J. R., Riipinen, I., Kulmala, M., Ehn, M., Petäjä, T., Junninen, H., Worsnop, D. R. and
24 Donahue, N. M.: Quantification of the volatility of secondary organic compounds in ultrafine
25 particles during nucleation events, *Atmos. Chem. Phys.*, 11(17), 9019–9036, doi:10.5194/acp-
26 11-9019-2011, 2011.
- 27 Raymond, T. M. and Pandis, S. N.: Formation of cloud droplets by multicomponent organic
28 particles, *J. Geophys. Res.*, 108, 8 PP., doi:200310.1029/2003JD003503, 2003.
- 29 Reddington, C. L., Carslaw, K. S., Spracklen, D. V., Frontoso, M. G., Collins, L., Merikanto,
30 J., Minikin, a., Hamburger, T., Coe, H., Kulmala, M., Aalto, P., Flentje, H., Plass-Dülmer, C.,
31 Birmili, W., Wiedensohler, a., Wehner, B., Tuch, T., Sonntag, a., O’Dowd, C. D., Jennings, S.
32 G., Dupuy, R., Baltensperger, U., Weingartner, E., Hansson, H.-C., Tunved, P., Laj, P.,
33 Sellegri, K., Boulon, J., Putaud, J.-P., Gruening, C., Swietlicki, E., Roldin, P., Henzing, J. S.,
34 Moerman, M., Mihalopoulos, N., Kouvarakis, G., Ždímal, V., Ziková, N., Marinoni, a.,
35 Bonasoni, P. and Duchi, R.: Primary versus secondary contributions to particle number
36 concentrations in the European boundary layer, *Atmos. Chem. Phys.*, 11(23), 12007–12036,
37 doi:10.5194/acp-11-12007-2011, 2011.
- 38 Riipinen, I., Pierce, J. R., Yli-Juuti, T., Nieminen, T., Häkkinen, S., Ehn, M., Junninen, H.,
39 Lehtipalo, K., Petäjä, T., Slowik, J., Chang, R., Shantz, N. C., Abbatt, J., Leaitch, W. R.,
40 Kerminen, V.-M., Worsnop, D. R., Pandis, S. N., Donahue, N. M. and Kulmala, M.: Organic

- 1 condensation: a vital link connecting aerosol formation to cloud condensation nuclei (CCN)
2 concentrations, *Atmos. Chem. Phys.*, 11(8), 3865–3878, doi:10.5194/acp-11-3865-2011,
3 2011.
- 4 Sihto, S.-L., Kulmala, M., Kerminen, V.-M., Dal Maso, M., Petäjä, T., Riipinen, I.,
5 Korhonen, H., Arnold, F., Janson, R., Boy, M., Laaksonen, A., and Lehtinen, K. E. J.:
6 Atmospheric sulphuric acid and aerosol formation: implications from atmospheric
7 measurements for nucleation and early growth mechanisms, *Atmos. Chem. Phys.*, 6, 4079-
8 4091, doi:10.5194/acp-6-4079-2006, 2006.
- 9 Sipilä, M., Berndt, T., Petäjä, T., Brus, D., Vanhanen, J., Stratmann, F., Patokoski, J.,
10 Mauldin, R. L., Hyvärinen, A.-P., Lihavainen, H. and Kulmala, M.: The role of sulfuric acid
11 in atmospheric nucleation., *Science*, 327(5970), 1243–6, doi:10.1126/science.1180315, 2010.
- 12 Spracklen, D. V., Carslaw, K. S., Kulmala, M., Kerminen, V.-M., Sihto, S.-L., Riipinen, I.,
13 Merikanto, J., Mann, G. W., Chipperfield, M. P., Wiedensohler, A., Birmili, W. and
14 Lihavainen, H.: Contribution of particle formation to global cloud condensation nuclei
15 concentrations, *Geophys. Res. Lett.*, 35, 5 PP., doi:200810.1029/2007GL033038, 2008.
- 16 Spracklen, D. V., Carslaw, K. S., Merikanto, J., Mann, G. W., Reddington, C. L.,
17 Pickering, S., Ogren, J. A., Andrews, E., Baltensperger, U., Weingartner, E., Boy, M.,
18 Kulmala, M., Laakso, L., Lihavainen, H., Kivekäs, N., Komppula, M., Mihalopoulos, N.,
19 Kouvarakis, G., Jennings, S. G., O'Dowd, C., Birmili, W., Wiedensohler, A., Weller, R.,
20 Gras, J., Laj, P., Sellegri, K., Bonn, B., Krejci, R., Laaksonen, A., Hamed, A., Minikin, A.,
21 Harrison, R. M., Talbot, R., and Sun, J.: Explaining global surface aerosol number
22 concentrations in terms of primary emissions and particle formation, *Atmos. Chem. Phys.*, 10,
23 4775-4793, doi:10.5194/acp-10-4775-2010, 2010.
- 24 Spracklen, D. V., Jimenez, J. L., Carslaw, K. S., Worsnop, D. R., Evans, M. J., Mann, G. W.,
25 Zhang, Q., Canagaratna, M. R., Allan, J., Coe, H., McFiggans, G., Rap, A., and Forster, P.:
26 Aerosol mass spectrometer constraint on the global secondary organic aerosol budget, *Atmos.*
27 *Chem. Phys.*, 11, 12109-12136, doi:10.5194/acp-11-12109-2011, 2011
- 28 Streets, D.G., T.C. Bond, G.R. Carmichael, S.D. Fernandes, Q. Fu, Z. Klimont, S.M. Nelson,
29 N.Y. Tsai, M.Q. Wang, J-H. Woo, and K.F. Yarber, An inventory of gaseous and primary
30 aerosol emissions in Asia in the year 2000, *J. Geophys. Res.*, 108, D21,
31 doi:10.1029/2002JD003093, 2003.
- 32
- 33 Trivitayanurak, W., Adams, P. J., Spracklen, D. V. and Carslaw, K. S.: Tropospheric aerosol
34 microphysics simulation with assimilated meteorology: model description and intermodel
35 comparison, *Atmos. Chem. Phys.*, 8(12), 3149–3168 [online] Available from:
36 //000257153400011, 2008.
- 37 Tzivion (Tzitzvashvili), S., Feingold, G., and Levin, Z.: An efficient numerical solution to the
38 stochastic collection equation, *J. Atmos. Sci.*, 44, 3139–3149, doi:10.1175/1520
39 0469(1987)044<3139:AENSTT>2.0.CO;2, 1987

- 1 Twomey, S. A.: Pollution and Cloud Albedo, *Eos T Am Geophys Un*, 58(8), 797–797 [online]
2 Available from: //a1977dp36300943, 1977.
- 3 Vehkamäki, H., Napari, I., Kulmala, M. and Noppel, M.: Stable ammonium bisulfate clusters
4 in the atmosphere, *Phys. Rev. Lett.*, 93(14), doi:10.1103/PhysRevLett.93.148501, 2004.
- 5 Vuollekoski, H., Nieminen, T., Paasonen, P., Sihto, S.-L., Boy, M., Manninen, H., Lehtinen,
6 K., Kerminen, V.-M. and Kulmala, M.: Atmospheric nucleation and initial steps of particle
7 growth: Numerical comparison of different theories and hypotheses, *Atmos. Res.*, 98(2-4),
8 229–236, doi:10.1016/j.atmosres.2010.04.007, 2010
- 9 Wang, M. and Penner, J. E.: Aerosol indirect forcing in a global model with particle
10 nucleation, *Atmos. Chem. Phys.*, 9(1), 239–260, doi:10.5194/acp-9-239-2009, 2009.
- 11 Weber, R. J., Marti, J. J., McMurry, P. H., Eisele, F. L., Tanner, D. J. and Jefferson, a.:
12 Measurements of new particle formation and ultrafine particle growth rates at a clean
13 continental site, *J. Geophys. Res.*, 102(D4), 4375, doi:10.1029/96JD03656, 1997.
- 14 Weber, R. J., McMurry, P. H., Eisele, F. L. and Tanner, D. J.: Measurement of Expected
15 Nucleation Precursor Species and 3–500-nm Diameter Particles at Mauna Loa Observatory,
16 Hawaii, *J. Atmos. Sci.*, 52(12), 2242–2257, doi:10.1175/1520-
17 0469(1995)052<2242:MOENPS>2.0.CO;2, 1995.
- 18 Westervelt, D. M., Pierce, J. R., Riipinen, I., Trivittayanurak, W., Hamed, A., Kulmala, M.,
19 Laaksonen, A., Decesari, S., and Adams, P. J.: Formation and growth of nucleated particles
20 into cloud condensation nuclei: model–measurement comparison, *Atmos. Chem. Phys.*, 13,
21 7645–7663, doi:10.5194/acp-13-7645-2013, 2013.
- 22 Wu, Z., Hu, M., Yue, D., Wehner, B. and Wiedensohler, A.: Evolution of particle number size
23 distribution in an urban atmosphere during episodes of heavy pollution and new particle
24 formation, *Sci. China Earth Sci.*, 54(11), 1772–1778, doi:10.1007/s11430-011-4227-9, 2011.
- 25 Yu, F.: Ion-mediated nucleation in the atmosphere: Key controlling parameters, implications,
26 and look-up table, *J. Geophys. Res.*, 115, 12 PP., doi:10.1029/2009JD012630, 2010.
- 27 Yu, F. and Luo, G.: Simulation of particle size distribution with a global aerosol model:
28 contribution of nucleation to aerosol and CCN number concentrations, *Atmos. Chem. Phys.*,
29 9(20), 7691–7710, 2009.
- 30 Yu, F. and Turco, R. P.: The size-dependent charge fraction of sub-3-nm particles as a key
31 diagnostic of competitive nucleation mechanisms under atmospheric conditions, *Atmos.*
32 *Chem. Phys.*, 11, 9451–9463, doi:10.5194/acp-11-9451-2011, 2011
- 33 Zhang, R., Suh, I., Zhao, J., Zhang, D., Fortner, E. C., Tie, X., Molina, L. T. and Molina, M.
34 J.: Atmospheric New Particle Formation Enhanced by Organic Acids, *Science (80-.)*,
35 304(5676), 1487–1490, doi:10.1126/science.1095139, 2004.

36

1 Table 1: Summary of previous nucleation-CCN formation studies. Values are CCN(0.2%)
 2 except for Yu and Luo (2009) which used CCN(0.4)%

Study	Model	Bounds	Pri. Sulf.	% change
Spraklen et al. (2008)	GLOMAP	ACT vs NONUC	2.5% of SO ₂ emis.	20
Merikanto et al. (2009)	GLOMAP	BHN + ACT vs BHN	2.5% of SO ₂	45
Makkonen et al. (2009)	ECHAM5-HAM	BHN + ACT vs BHN	2.5% SO ₂	50
Yu and Luo (2009)	GEOS-Chem-APM	ION vs NONUC	None	60
Wang and Penner (2009)	IMPACT/CCSM3	BHN + ACT vs BHN	2% of SO ₂	5
Pierce and Adams (2009)	GISS-TOMAS	TER vs BHN	1% of SO ₂	12
Makkonen et al. (2012)	ECHAM5.5-HAM2	ACT vs BHN	2.5% of SO ₂	19
Lee et al. (2013)	GLOMAP	ACT + BHN vs NONUC	0-1% of SO ₂	45

3
 4
 5
 6
 7
 8
 9

1 Table 2: List of simulations. All 8 simulations were also run at higher SOA production rates,
 2 as described in the text.

Name	Nucleation	Reference
NONUC	None (control)	N/A
BHN	Binary homogenous	Vehkamäki et al. (2002)
ION	Ion-mediated (IMN)	Yu (2010)
ACT1	Activation ($A=10^{-6}$)	Kulmala et al. (2006), Sihto et al. (2006)
ACT2	Activation ($A=2 \times 10^{-6}$)	Kulmala et al. (2006), Sihto et al. (2006)
TER5	Ternary (10^{-5} tuning factor)	Napari et al. (2002), Jung et al. (2005), Westervelt et al. (2013)
TER3	Ternary (10^{-3} tuning factor)	Napari et al. (2002), Jung et al. (2005), Westervelt et al. (2013)
TER	Ternary (no tuning factor)	Napari et al. (2002)

3
 4
 5
 6
 7
 8
 9
 10
 11
 12

1 Table 3. Global-average percent increases in a given nucleation simulation compared to a
 2 simulation with no nucleation. Values are averaged for boundary layer only. See also Figs. 2-
 3 3

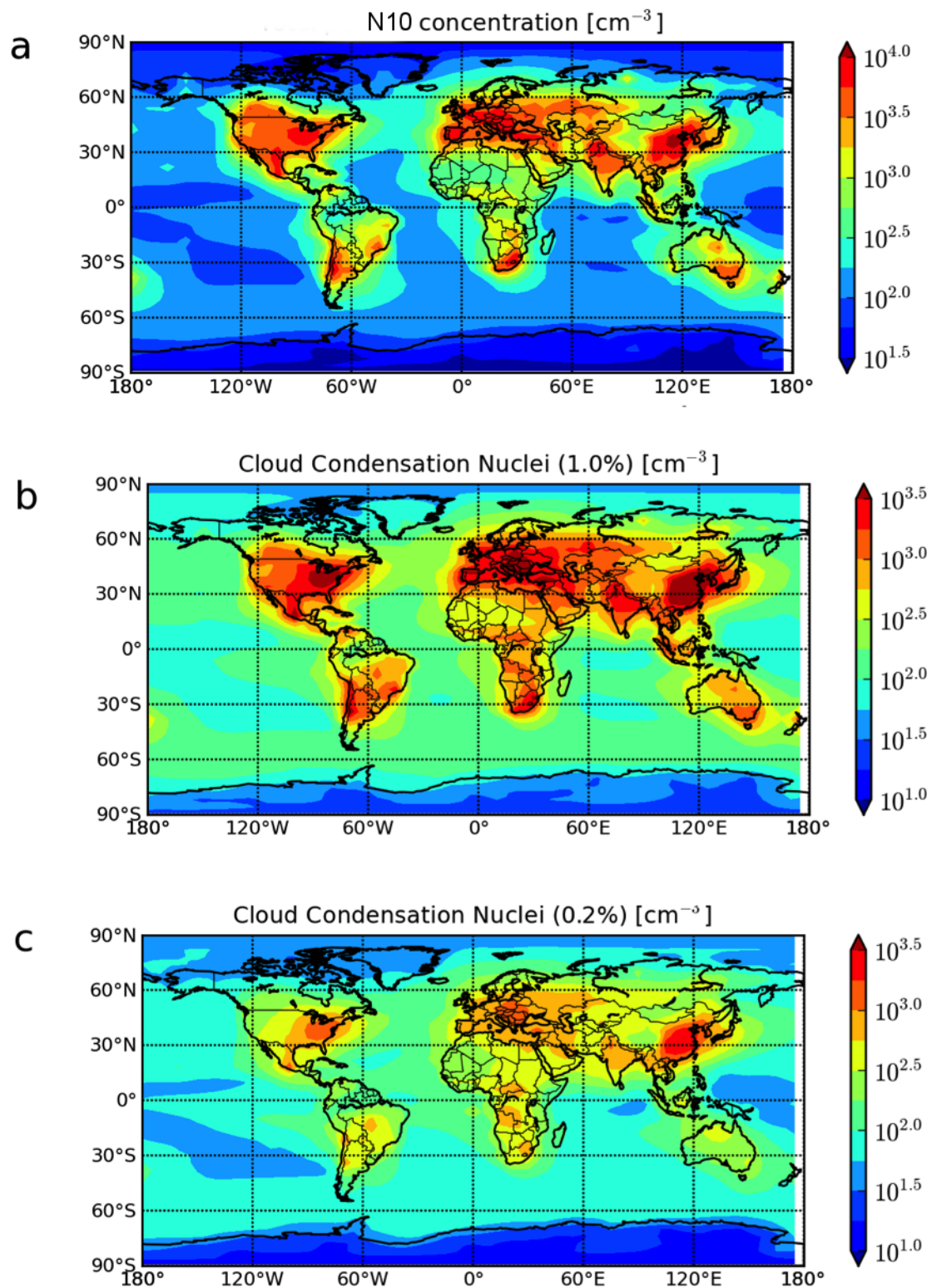
	N_{10} (%)	CCN(1.0%) (%)	CCN(0.2%) (%)
BHN	23	27	49
ION	55	50	60
ACT1	140	79	66
ACT2	170	88	69
TER5	48	46	56
TER3	103	70	64
TER	190	99	78

4
 5
 6
 7
 8
 9
 10
 11
 12

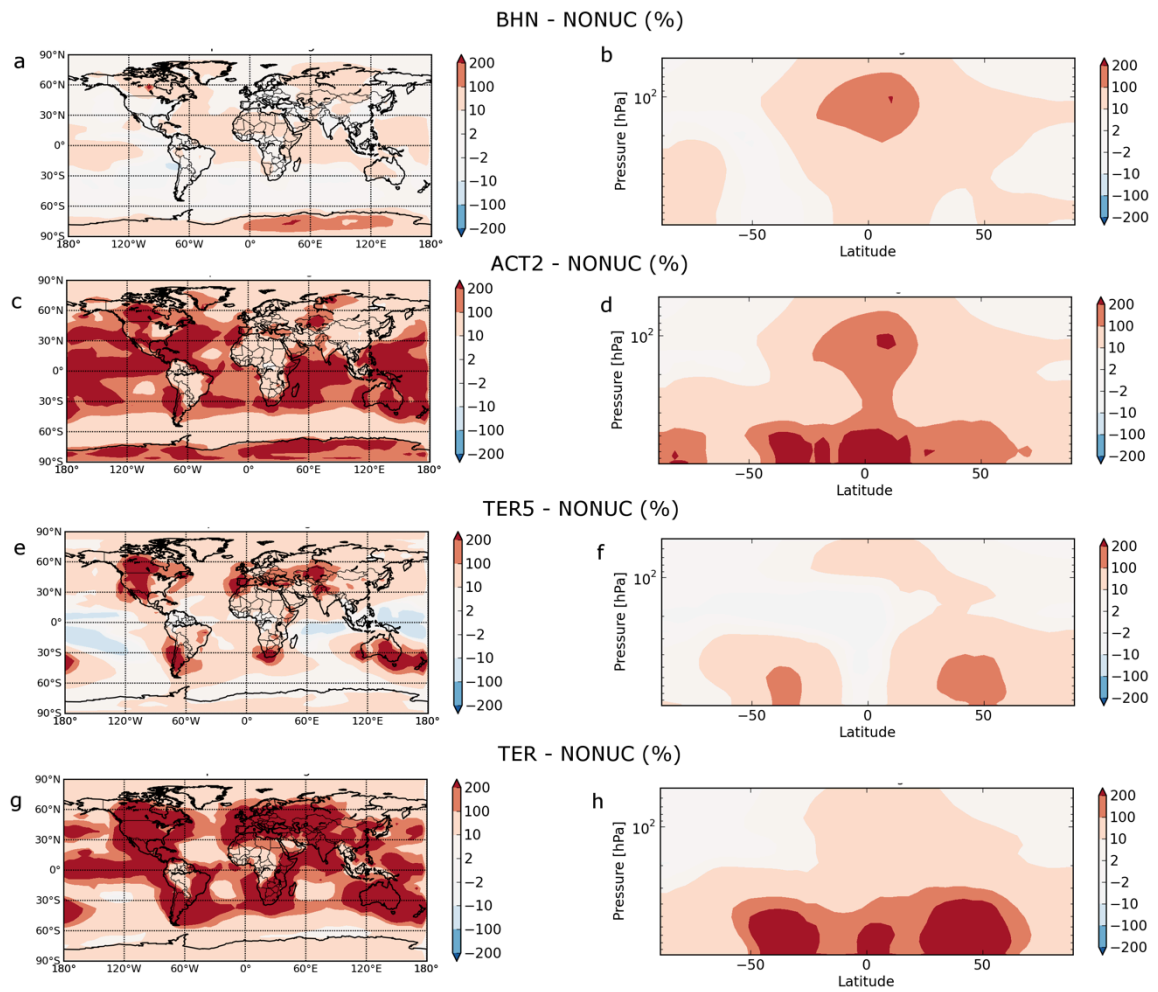
1 Table 4. Mean and median values for the three ternary simulations at Hyytiälä for the multi-
 2 day survival probability calculation. Median values in parentheses.

	Ternary x 10 ⁻⁵	Ternary x 10 ⁻³	Ternary
J ₃ (cm ⁻³ s ⁻¹)	2.3 (0.3)	7.4 (3.7)	15.1 (13.6)
GR (nm hr ⁻¹)	2.3 (1.9)	1.7 (1.5)	1.5 (1.3)
SP ₅₀ (%)	37 (36)	12 (14)	4.9 (4.5)
J ₅₀ (10 ⁻² cm ⁻³ s ⁻¹)	7.1 (6.0)	7.2 (7.0)	7.4 (8.0)
SP ₁₀₀ (%)	2.9 (2.0)	0.4 (0.4)	0.2 (0.2)
J ₁₀₀ (10 ⁻² cm ⁻³ s ⁻¹)	1.2 (0.23)	1.3 (0.41)	1.3 (0.81)

3

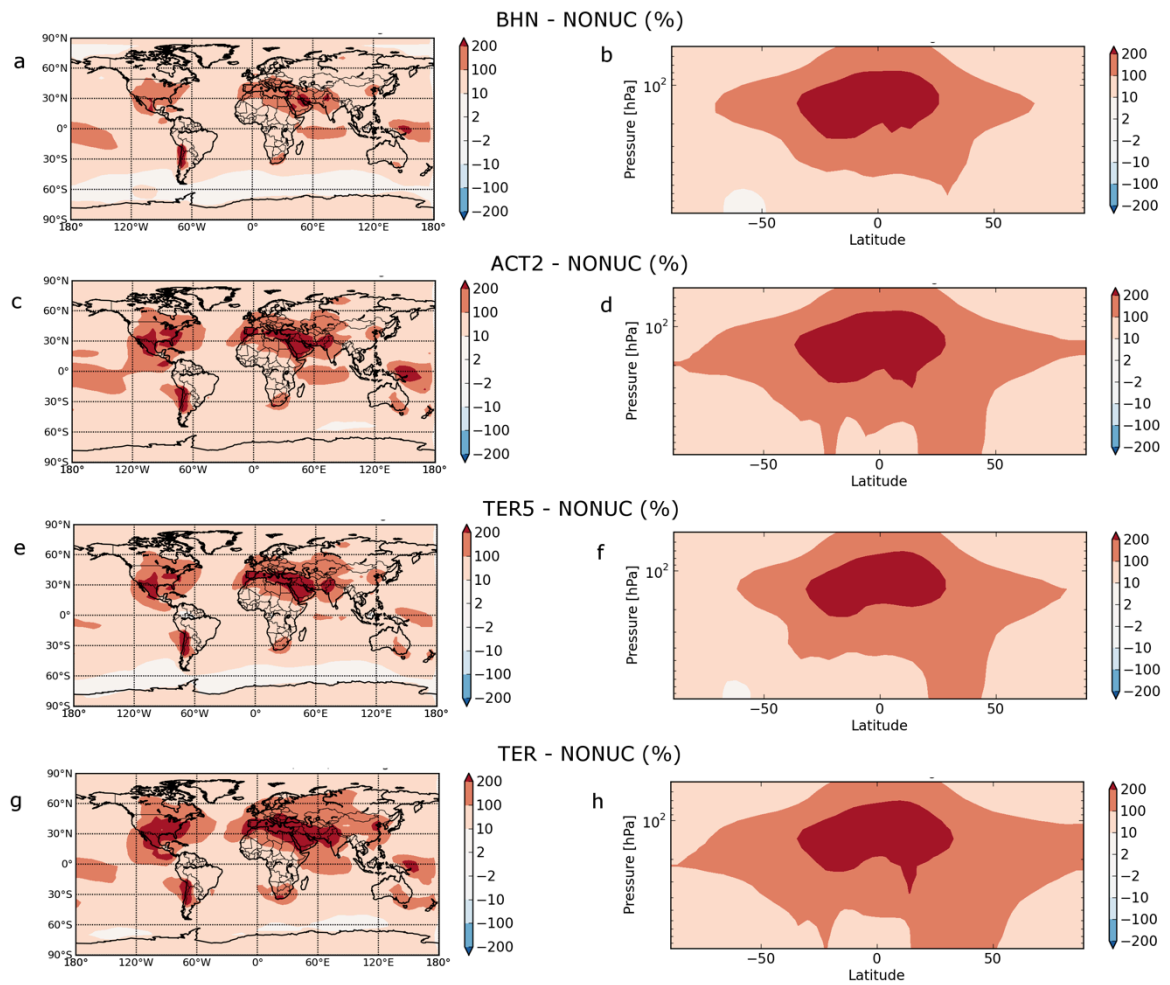


1
 2 Figure 1: Annual-average boundary layer global distribution of N_{10} (a), CCN(1.0%) (b), and
 3 CCN(0.2%) (c) for the ternary simulation with a 10^{-5} tuning factor (TER5 simulation).



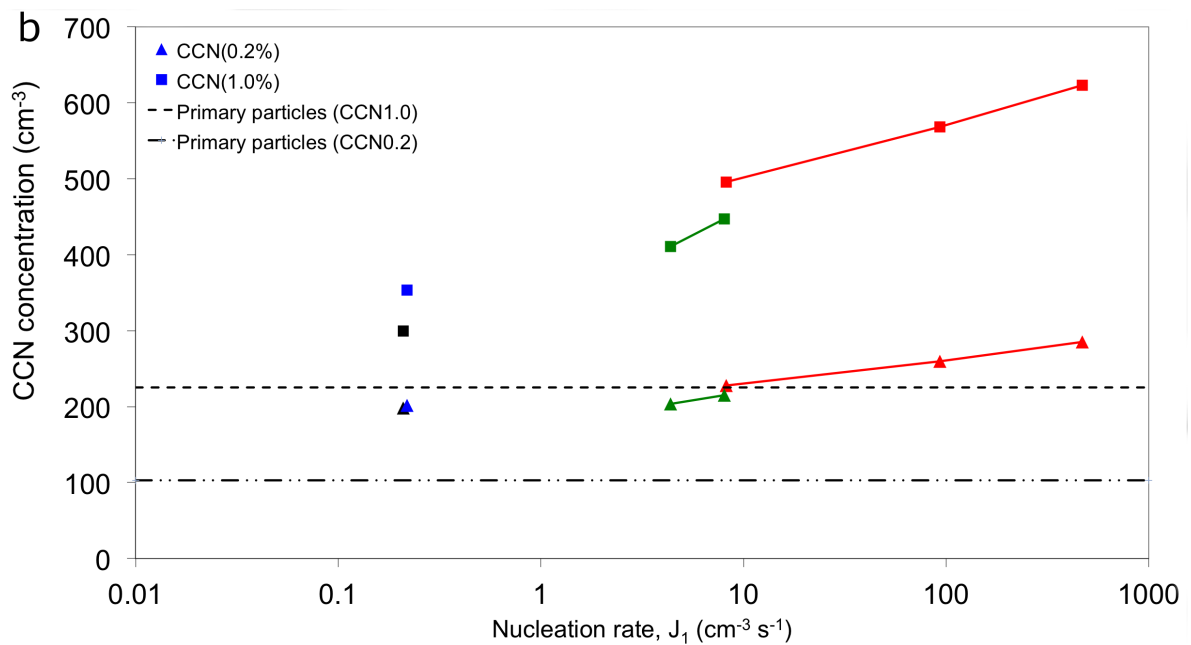
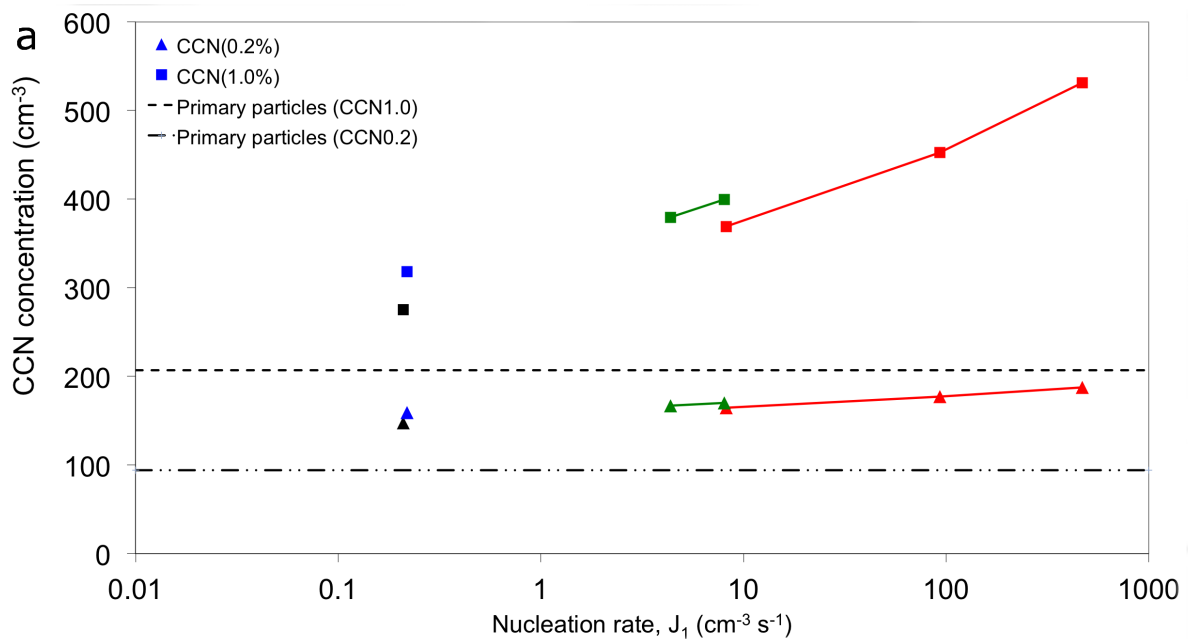
1

2 Figure 2: Global percent change in total particle number concentration (N_{10}) surface and zonal
 3 plots. Color contours in all panels represent the percent change in number concentration
 4 between the listed nucleation simulations and the simulation with nucleation off: **a,b**) binary;
 5 **c,d**) Activation ($A=2E-6$); **e,f**) Ternary (10^{-5} factor); **g,h**) Ternary.



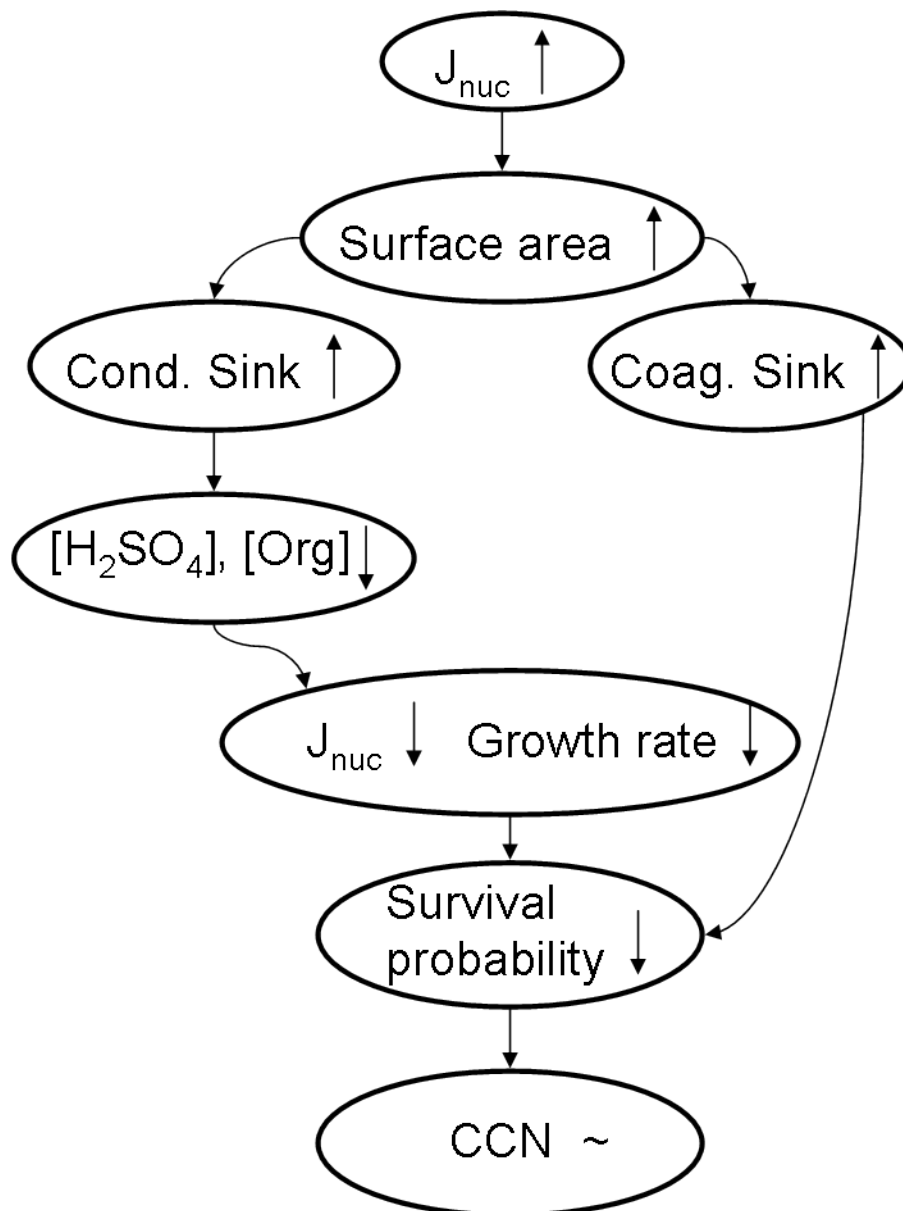
1

2 Figure 3: Global cloud condensation nuclei concentration at $S=0.2\%$ (CCN(0.2%)) percent
 3 change surface and zonal plots. Color contours in all panels represent the percent change in
 4 number concentration between the various nucleation simulations and the simulation with
 5 nucleation off: **a,b**) binary; **c,d**) Activation ($A=2E-6$); **e,f**) Ternary (10^{-5} factor); **g,h**) Ternary.



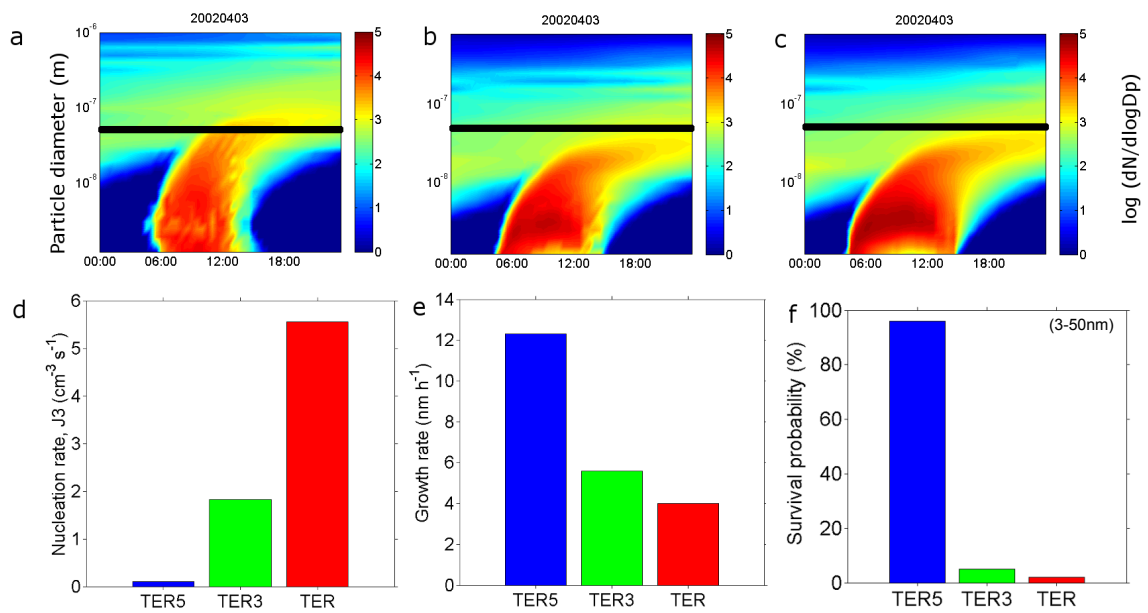
1
 2 Figure 4: Global average cloud condensation nuclei concentrations in the boundary layer for
 3 all nucleation simulations. Squares represent CCN(1.0%), and triangles show CCN(0.2%).
 4 Panel (a) shows the base SOA simulation results while (b) shows the high SOA simulation
 5 results (extra 100 Tg yr⁻¹). In both panels, the dashed black lines represent the CCN(1.0%)
 6 concentrations for the simulation with no nucleation (nominally, primary particle
 7 contribution). The dash-dot black lines show the same for CCN(0.2%). Blue lines refer to ion
 8 nucleation, black is binary, green is activation, and red is ternary.

9



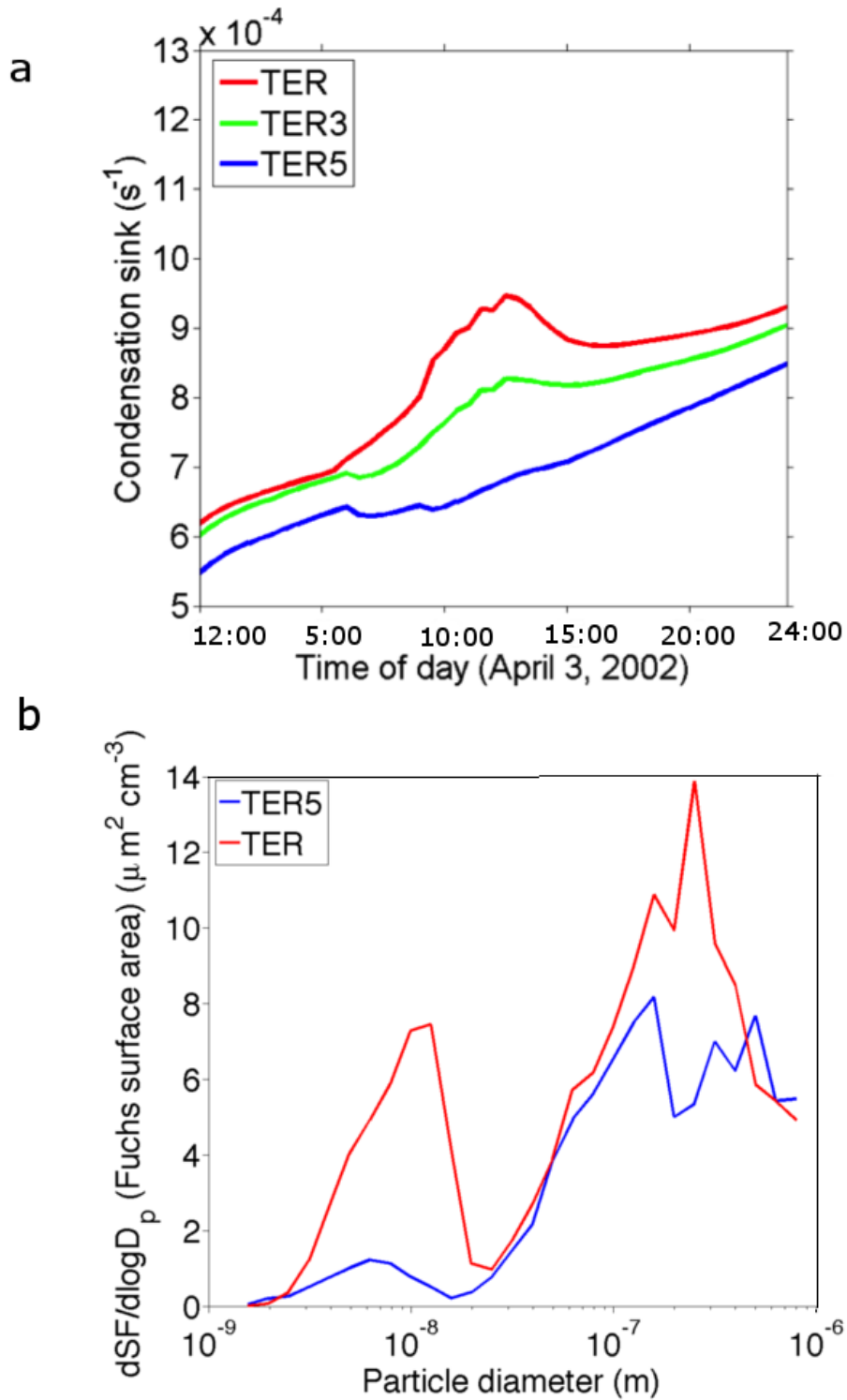
1

2 Figure 5. Flowchart of microphysical feedbacks that decrease survival probability with
 3 increasing nucleation rates, limiting the formation of CCN from nucleation events.



1
 2 Figure 6: Nucleation metrics and number size evolution plots for April 3 (local time) at
 3 Hyytiälä. Panel (a) is the size distribution evolution for the TER5 simulation, panel (b) is the
 4 same for the TER3 simulation, and panel (c) is the TER simulation. The bottom row (panels d
 5 through f) shows the nucleation rate, growth rate (3-25 nm), and survival probability to 50 nm
 6 (multi-day) for each of the three ternary nucleation scenarios. The heavy black line indicates
 7 the 50 nm size threshold used for this analysis.

8

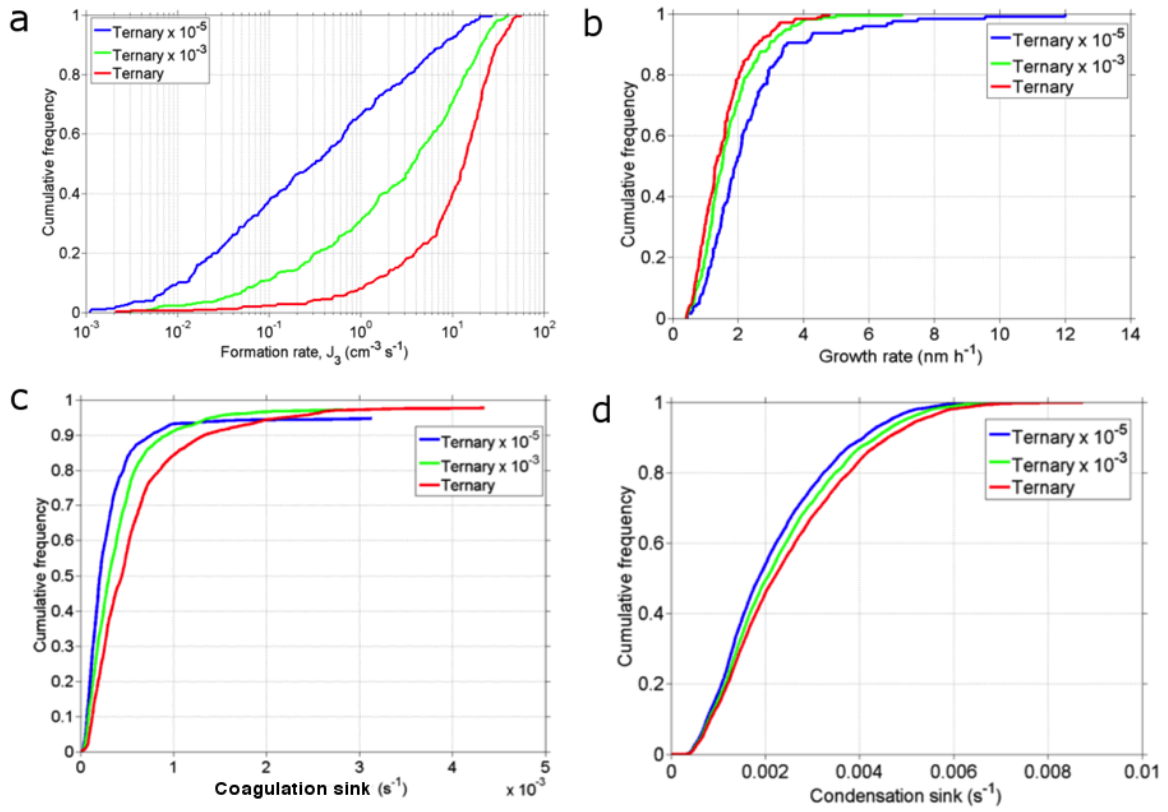


1

2 Figure 7: Visualization of condensation sink (**a**) and particle surface area size distribution (**b**),
 3 for April 3 at Hyytiälä. Fuchs surface area size distributions are shown for a 12:00 noon
 4 timeslice in the nucleation event on April 3, 2002

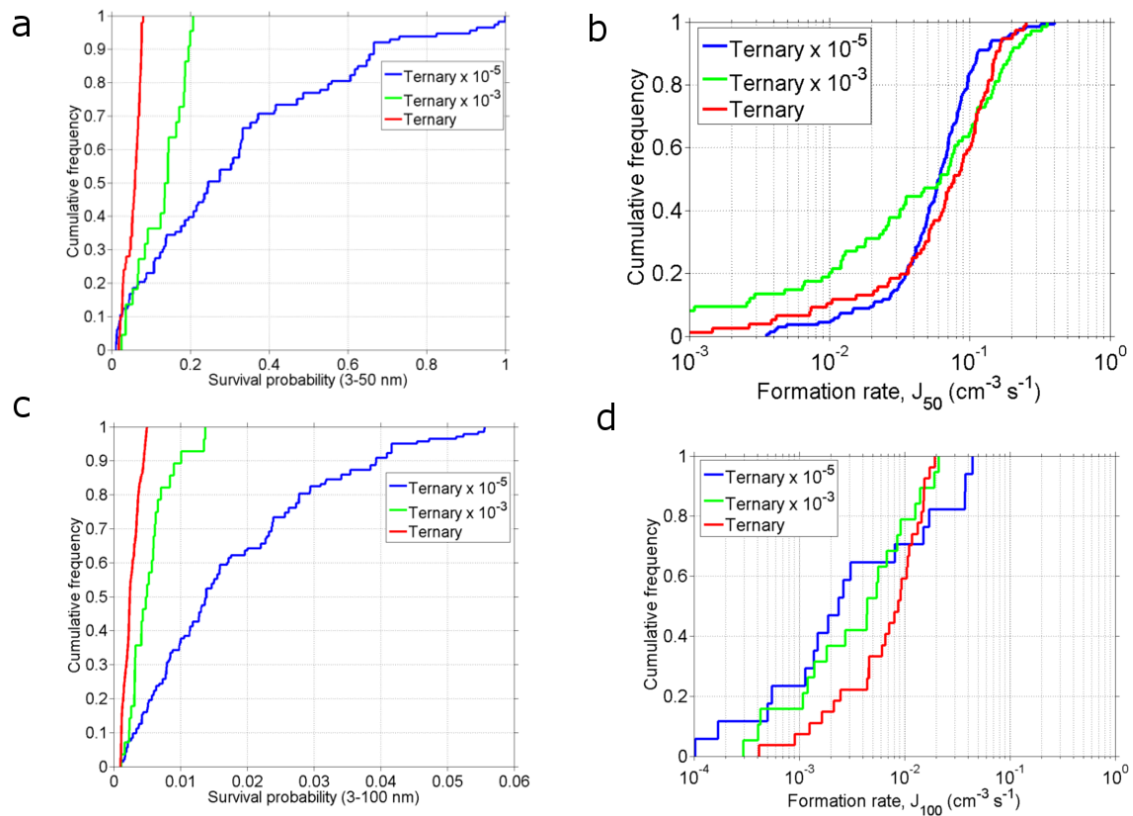
5

1
2



3

4 Figure 8: Cumulative distribution functions of nucleation event metrics at Hyytiälä. Red
5 curves represent the original ternary simulation (Napari et al. 2002). Green curves represent
6 ternary nucleation with a 10^{-3} tuning factor. Blue curves represent ternary nucleation with a
7 10^{-5} tuning factor. (a) 3 nm particle formation rate, J_3 ($\text{cm}^{-3} \text{s}^{-1}$); (b) 3-25 nm growth rate, GR
8 (nm h^{-1}); (c) Coagulation frequency (s^{-1}); (d) Condensation sink (s^{-1}).

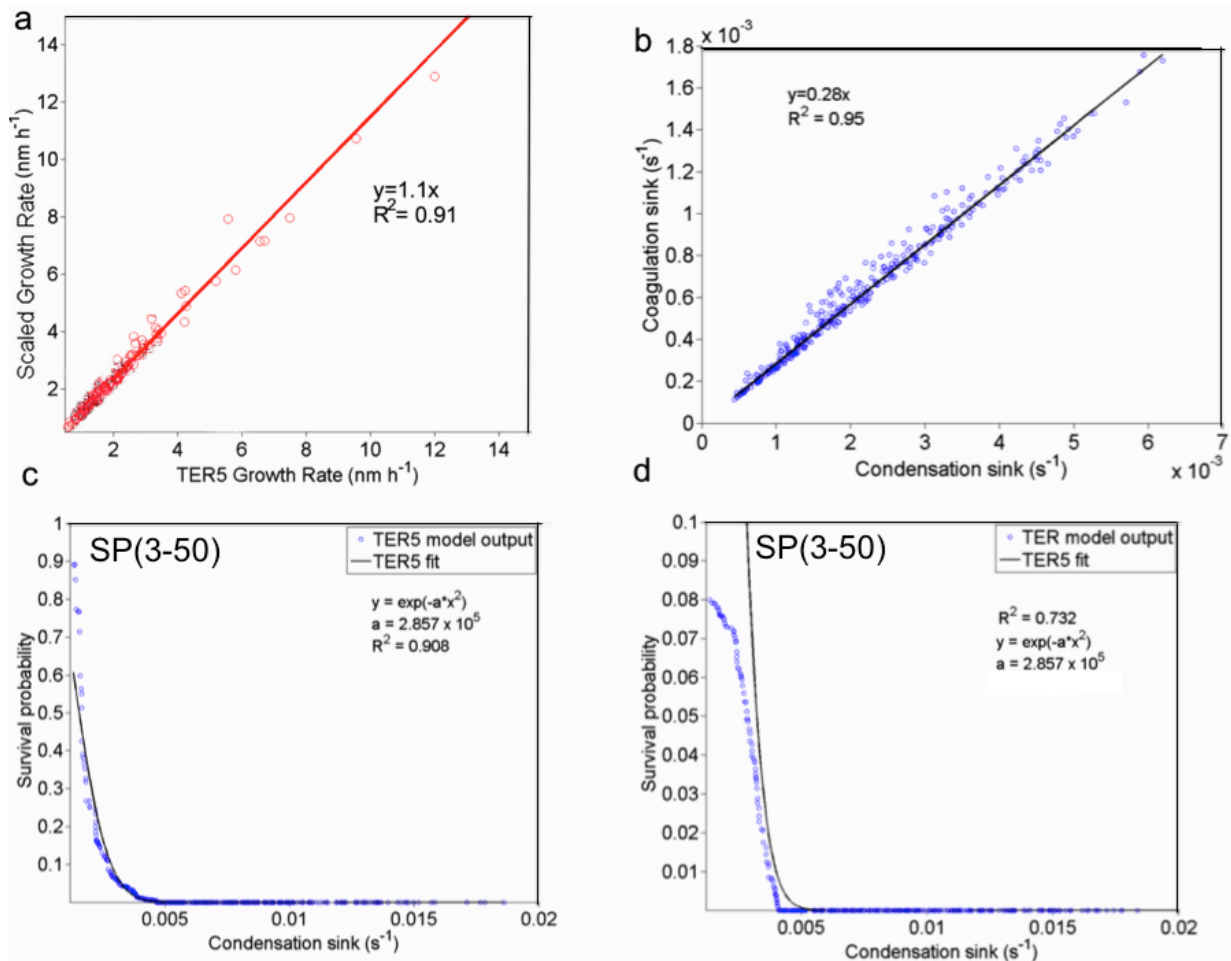


1

2 Figure 9. Cumulative distribution functions of nucleation event parameters at Hyttiälä. Red
 3 curves represent the original ternary simulation (Napari et al. 2002). Green curves represent
 4 ternary nucleation with a 10⁻³ tuning factor. Blue curves represent ternary nucleation with a
 5 10⁻⁵ tuning factor. **(a)** Survival probability to 50 nm, SP₃₋₅₀ **(b)** 50 nm particle formation rate,
 6 J₅₀ (cm⁻³ s⁻¹); **(c)** Survival probability to 100 nm, SP₃₋₁₀₀ ; **(d)** 100 nm particle formation rate,
 7 J₁₀₀ (cm⁻³ s⁻¹)

8

9



1

2 Figure 10: Explanation of the dependence of growth rates and survival probabilities on the
 3 condensation sink for one year of nucleation events at Hyytiälä. (a) Scaled TER vs TER5
 4 growth rates (red circles). The ternary growth rates have been scaled by the ratio of the two
 5 nucleation mechanisms' condensation sinks. Solid red line represents the line of best fit ($y =$
 6 $1.1x$, $r^2 = 0.91$). (b) Correlation between coagulation sink and condensation sink. Solid blue
 7 line is best fit ($y = 0.28x$, $r^2 = 0.95$). (c) Plot of survival probability (to 50 nm) versus
 8 condensation sink for the TER5 simulations (blue circles). Solid black line represents a model
 9 fit to the TER5 data of the form $y = \exp(-a \cdot x^2)$ as described in the text. Value of a is given in
 10 the figure. (d) Same as C, except that the TER model output (blue circles) is compared to the
 11 same TER5 fit from panel (C) (same equation and coefficients).

12

13

14