1 Improvement and Further Development in CESM/CAM5: Gas-Phase Chemistry and Inorganic 2 Aerosol Treatments J ian He¹ and Yang Zhang^{1,*} ¹ Air Quality Forecasting Laboratory, Department of Marine, Earth, and Atmospheric Sciences, 5 North Carolina State University, Raleigh, NC, 27695, USA 6 7

8 **Abstract:** Gas-phase chemistry and subsequent gas-to-particle conversion processes such as new 9 particle formation (J), condensation, and thermodynamic partitioning have large impacts on air 10 quality, climate, and public health through influencing the amounts and distributions of gaseous 11 precursors and secondary aerosols. Their roles in global air quality and climate are examined in 12 this work using the Community Earth System Model version 1.0.5 (CESM1.0.5) with the 13 Community Atmosphere Model version 5.1 (CAM5.1) (referred to as CESM1.0.5/CAM5.1). 14 CAM5.1 includes a simple chemistry that is coupled with a 7-mode prognostic Modal Aerosol 15 Model (MAM7). MAM7 includes classical homogenous nucleation (binary and ternary) and 16 activation nucleation (empirical first-order power law) parameterizations, and a highly-simplified 17 inorganic aerosol thermodynamics treatment that only simulates particulate-phase sulfate and 18 ammonium. In this work, a new gas-phase chemistry mechanism based on the 2005 Carbon 19 Bond Mechanism for Global Extension (CB05_GE) and several advanced inorganic aerosol 20 treatments for condensation of volatile species, ion-mediated nucleation (IMN), and explicit 21 inorganic aerosol thermodynamics for sulfate, ammonium, nitrate, sodium, and chloride have 22 been incorporated into CESM/CAM5.1-MAM7. Comparing to the simple gas-phase chemistry, 23 CB05 GE can predict many more gaseous species, and thus could improve model performance 24 for $PM_{2.5}$, PM_{10} , PM components, and some PM gaseous precursors such as SO_2 and NH₃ in 25 several regions, as well as aerosol optical depth (AOD) and cloud properties (e.g., cloud fraction

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26 (CF), cloud droplet number concentration (CDNC), and shortwave cloud forcing (SWCF)) on 27 globe. The modified condensation and aqueous-phase chemistry could further improve the 28 predictions of additional variables such as $HNO₃$, $NO₂$, and $O₃$ in some regions, and new particle 29 formation rate (J) and AOD over globe. IMN can improve the predictions of secondary $PM_{2.5}$ 30 components, $PM_{2.5}$, and PM_{10} over Europe, as well as AOD and CDNC over globe. The explicit 31 inorganic aerosol thermodynamics using ISORROPIA II improves the predictions of all major 32 PM2.5 components and their gaseous precursors in some regions, as well as downwelling 33 shortwave radiation, SWCF, and cloud condensation nuclei at a supersaturation of 0.5% over 34 globe. For simulations of 2001-2005 with all the modified and new treatments, the improved 35 model predicts that on a global average, SWCF increases by 2.7 W $m²$, reducing NMBs of 36 SWCF from -5.4% to 1.2%. Uncertainties in emissions can explain largely the inaccurate 37 predictions of precursor gases (e.g., SO₂, NH₃, and NO) and primary aerosols (e.g., black carbon 38 and primary organic matter). Additional factors leading to discrepancies between model 39 predictions and observations include assumptions associated with equilibrium partitioning for 40 fine particles assumed in ISORROPIA II, irreversible gas/particle mass transfer treatment for 41 coarse particles, uncertainties in model treatments such as dust emissions, secondary organic 42 aerosol formation, multiple-phase chemistry, cloud microphysics, aerosol-cloud interaction, dry 43 and wet deposition, and model parameters (e.g., accommodation coefficients and prefactors of 44 the nucleation power law), as well as uncertainties in model configuration such as the use of a 45 coarse grid resolution.

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47 **Keywords**: CESM/CAM5.1, CB05_GE, New particle formation, Aerosol thermodynamics, Ion-48 mediated nucleation, ISORROPIA II, Earth system modeling

49
50 50 **1. Introduction**

94 2012b; Yu et al., 2012). Limited observations make it difficult to validate predicted J values and 95 appropriateness of various parameterizations.

96 A number of thermodynamic aerosol modules have been developed to understand physical and 97 chemical properties of inorganic aerosols. For example, EQUISOLV II (Jacobson, 1999) has 98 been used in a one-way nested (from global to local scales) gas, aerosol, transport, radiation, 99 general circulation, mesoscale, and ocean model (GATOR-GCMOM) (Jacobson, 2010). 100 EQUISOLV II uses analytical equilibrium iteration and mass flux iteration to solve equilibrium 101 problems (Jacobson, 1999), which requires relatively large computational cost. SCAPE2 is used 102 in the California Institute of Technology (CIT) model (Meng et al., 1998). ISORROPIA (Nenes 103 et al., 1998) has been used in several global models such as GEOS-Chem (Bey et al., 2001), the 104 GISS Caltech (Liao et al., 2003), and the GU-WRF/Chem (Zhang et al., 2012b) and regional 105 models such as the Community Multiscale Air Quality model (CMAQ) (Byun and Schere, 2006) 106 and the Comprehensive Air Quality Model with Extensions (CAMx) (ENVIRON, 2010). An 107 updated version, ISORROPIA II (Fountoukis and Nenes, 2007), has also been implemented in 108 recent versions of CMAQ (e.g., CMAQ v4.7-Dust (Wang et al., 2012) and CMAQ v5.0 (Appel et 109 al., 2013)), GEOS-Chem (Fountoukis and Nenes, 2007), and ECHAM5 with MESSy 110 Atmospheric Chemistry and Global Modal-aerosol eXtension (EMAC/GMXe) (Metzger et al., 111 2011). The Multicomponent Equilibrium Solver for Aerosols (MESA) (Zaveri et al., 2005) has 112 been used in the mesoscale WRF/Chem (Fast et al., 2006). These modules assume that particles 113 simulated in a given particle size range have the same composition (i.e., internal mixture). 114 Different aerosol thermodynamic models can lead to different aerosol predictions (Nenes et al., 115 1998; Zhang et al., 2000; Zaveri et al., 2005). Zhang et al (2000) reported average absolute 116 differences of 7.7% - 12.3% in total PM predictions between different thermodynamic modules

117 under 400 test conditions but the differences could be as large as 68% under some cases (e.g., 118 high nitrate/chloride and low/medium relative humidity (RH)). Fountoukis and Nenes (2007) 119 found the largest discrepancies between ISORROPIA II and SCAPE2 in water concentration 120 predictions exist under low RH conditions (RH < 60%), primarily from differences in the 121 treatment of water uptake and solid state composition. The 3-D atmospheric models with these 122 modules include explicit thermodynamic treatments for sulfate, ammonium, nitrate, sodium, and 123 chloride. The equilibrium assumption, however, is not valid under some conditions (e.g., coarse 124 particles and cooler conditions) (Meng and Seinfeld, 1996). Kinetic approaches are therefore 125 needed to treat gas/particle mass transfer under such conditions. Kinetic approaches, on the 126 other hand, are computationally expensive (Zhang et al., 2004; Hu et al., 2008) and have only 127 been implemented in a few 3-D models (e.g., Meng and Seinfeld, 1996; Jacobson, 2005; Zhang 128 and Wexler, 2006; Zaveri et al., 2008). A hybrid approach that assumes equilibrium for fine 129 particles and solves gas/particle mass transfer for coarse particles has been thus developed, 130 which provides the best compromise between numerical accuracy and computational efficiency 131 (Capaldo et al., 2000; Kelly et al., 2010). A simple approach for gas/particle mass transfer used 132 in some GCMs, such as CAM5 is to treat sulfate and ammonium only with a full neutralization 133 (the NH_4^+ / SO_4^2 molar ratio of 2 for a mode) through an irreversible condensation. 134 In this work, a comprehensive gas-phase chemical mechanism and detailed inorganic 135 aerosol treatments for nucleation and aerosol thermodynamics are incorporated into CAM

136 version 5.1 (CAM5.1) in the CESM version 1.0.5 (CESM1.0.5). Several modifications are also

137 made to the existing treatments such as condensation and aqueous-phase chemistry. The

138 objectives are to improve the representations of gas-phase chemistry and inorganic aerosol

139 treatments in CESM/CAM5.1, and reduce associated uncertainties. The improved model with

140 enhanced capabilities can be applied for decadal simulations to study interactions among

141 atmospheric chemistry, aerosols, and climate change.

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143 **2. Model Development and Improvement**

144 CESM is a fully-coupled global Earth system model, which includes land, ocean, 145 atmosphere, and sea ice components. The atmosphere component used in this study is CAM5.1.

146 Existing and new model treatments related to this study are described in this section. Further

147 details on CAM5.1 can be found at http://www.cesm.ucar.edu/models/cesm1.0/cam/.

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149 2.1 Existing Gas-Phase Chemistry and Aerosol Treatments in CESM/CAM5.1

150 CAM5.1 uses a simple gas-phase chemistry for sulfur species, which includes 1 151 photolytic reaction and 7 kinetic reactions among 6 gas-phase species (i.e., hydrogen peroxide 152 (H₂O₂), sulfuric acid (H₂SO₄), sulfur dioxide (SO₂), dimethylsulfide (DMS), ammonia (NH₃), 153 and semi-volatile organic gas (SOAG)). A more comprehensive gas-phase mechanism with 40 154 photolytic reactions and 172 kinetic reactions among 103 species, i.e., the Model of OZone and 155 Related chemical Tracers version 4 (MOZART-4) of Emmons et al. (2010), has been 156 incorporated into the official released CAM5.1. It was only coupled with the bulk aerosol 157 module (BAM) in CAM5.1 implemented in CESM 1.0.5 that is used in this work (It was coupled 158 with MAM in CESM v1.1). In addition to BAM, CAM5.1 contains the modal aerosol model 159 (MAM) that is based on modal representations of aerosols. In this study, MAM is used because it 160 can represent more accurate size distributions as compared to BAM. There are two versions of 161 MAM, one with seven lognormal modes (MAM7), and the other with three lognormal modes 162 (MAM3) (Liu et al., 2012), and both are coupled with the simple gas-phase chemistry in the

164 and size distributions for dust, sea-salt, and primary carbon compared to MAM3. MAM7 165 explicitly treats sulfate, ammonium, sea-salt, dust, BC, POM, and SOA. It simulates 166 condensational growth of aerosol, nucleation, coagulation, dry deposition, wet removal, and 167 water uptake. Condensation is simulated based on a kinetic approach, in which MAM7 treats 168 H2SO4, NH3, and methanesulfonic acids (MSA) as completely non-volatile species and treats 169 SOAG as a volatile species, using a constant accommodation coefficient of 0.65 for all these 170 condensing species based on Adams and Seinfeld (2002). NH3 condensation stops when the $171 \text{ NH}_4^+/\text{SO}_4^2$ molar ratio of a particle mode reaches 2 (i.e., fully neutralized by SO_4^2 ions). The 172 net uptake rate, *Inet*, due to gas to particle mass transfer for each species to each mode is 173 simulated as,

$$
I_{net} = \int dx \frac{dN}{dx} I_{cond} \tag{1}
$$

174

$$
I_{\text{cond}} = 2 \times \pi \times D_{g} \times D_{p} \times F(Kn, \alpha)
$$
\n⁽²⁾

$$
F(Kn, \alpha) = \frac{0.75 \times (1 + Kn)}{Kn \times \left(\frac{1 + Kn}{\alpha} + 0.283\right) + 0.75}
$$
(3)

175 where D_p is the particle diameter; *x* is the logarithmic diameter of particle, = $ln(D_p)$; dN/dx is the 176 log-normal particle number density distribution; *Inet* is the gas condensation rate; *Kn* is the 177 Knudsen number; α is the accommodation coefficient of condensable vapor; D_g is the gas 178 diffusivity, and F(*Kn*, α) is the Fuchs-Sutugin correction factor that describes the resistance to 179 uptake caused by gas-phase diffusion. Equation (1) is solved using the Gauss-Hermite quadrature 180 of order 2. Based on equation (3), as *α* approaches zero, F(*Kn*, *α*) approaches zero. Consequently, 181 *Icond* (i.e., the uptake rate) approaches zero in equation (1).

204 reactions among 93 gas-phase species in this study. The gas-phase chemical system is solved 205 using an implicit backward Euler method.

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- 207 2.2.2 Ion-mediated nucleation parameterization

208 Ions generated by cosmic radiation and natural radioactive decay have been studied for a 209 long time as an important source to enhance nucleation (Raes et al., 1986). An IMN model is 210 developed by Yu (2010) (Yu10) for $H_2SO_4-H_2O$ system, and explicitly solves the dynamic 211 equations in terms of temperature, RH, H₂SO₄ vapor concentration, ionization rate, and surface 212 area of preexisting particles. Different from classic binary nucleation theory, which is based on 213 the minimization of changes in Gibbs free energy (Seinfeld and Pandis, 2006), IMN is based on a 214 kinetic model that considers the interactions among ions, neutral and charged clusters, vapor 215 molecules, and preexisting particles (Yu, 2010). The global ionization rates due to cosmic rays 216 are calculated based on the schemes given in Usoskin and Kovaltsov (2006) and the contribution 217 of radioactive materials from soil to ionization rates is parameterized based on the profiles given 218 in Reiter (1992). To reduce the computing cost using IMN in 3-D models, Yu et al. (2008) 219 developed lookup tables with simple interpolation subroutines to calculate nucleation rates under 220 typical atmospheric conditions. In this work, IMN based on YU10 is implemented into MAM7 221 and combined with default nucleation parameterizations (VE02, ME07, and WP09) in order to 222 improve the J predictions and aerosol number concentrations in upper troposphere. The J value 223 above PBL is taken as the maximum value among predictions from IMN (YU10) and 224 homogeneous nucleation (VE02 or ME07), and the J value within PBL is taken as the maximum 225 value among predictions from IMN (YU10), homogeneous nucleation (VE02 or ME07), and the 226 first-order parameterization (WP09).

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228 2.2.3 Inorganic Aerosol Thermodynamics

229 Gas-particle partitioning is an important process in the formation and evolution of 230 secondary aerosols. Several factors affect gas-particle partitioning, such as temperature, RH, 231 saturation vapor pressures of species, the physical state of the condensed-phase, and the 232 interactions among aerosol components (Zuend et al., 2010). Most models focus on inorganic 233 aerosols. Fountoukis and Nenes (2007) developed a computationally-efficient thermodynamics 234 equilibrium model, ISORROPIA II, for the magnesium (Mg^{2+}) - potassium (K^+) - calcium (Ca^{2+}) 235 - NH₄⁺- Na⁺ - SO₄²- NO₃ - Cl⁻- H₂O aerosol system. An important difference between 236 ISORROPIA II and most other thermodynamics equilibrium models is that ISORROPIA II 237 simulates crustal species, such as Mg^{2+} , K^+ , and Ca^{2+} , which are important constituents of 238 atmospheric aerosols, in particular, mineral dust. Therefore, to explicitly simulate aerosol 239 thermodynamics, ISORROPIA II has been implemented into MAM7 and applied for 240 accumulation, Aitken, fine sea-salt, and fine dust modes to explicitly simulate thermodynamics 241 of SO_4^2 , NH₄⁺, NO₃⁻, Cl⁻, and Na⁺ as well as the impact of crustal species associated with fine 242 dust modes on aerosol thermodynamics. The concentrations of K^+ , Ca^{2+} , and Mg^{2+} as the input 243 for ISORROPIA II are calculated from dust concentrations, using the mass ratios of 1.022×10^{-3} , 244 1.701×10⁻³, and 7.084×10⁻⁴, respectively (Van Pelt and Zobeck, 2007). The resulted 245 concentrations of aerosol components from ISORROPIA are mapped back to fine aerosol modes 246 based on their mass ratios to the total mass over all fine modes at the previous time step. 247 Aerosol thermodynamics involving coarse particles (in coarse sea-salt and dust modes) is 248 currently not treated explicitly in this work, given the high computational cost (by at least a 249 factor of 3) for solving the non-equilibrium system involving coarse particles. Instead, the simple

250 kinetic approach used in the default CAM5.1 is used to simulate the condensation of inorganic 251 gases onto coarse modes (see section 2.2.4). For fine mode particles, before thermodynamic 252 calculation using ISORROPIA II, the condensation and nucleation processes are simulated to 253 allow a more realistic allocation of gaseous H_2SO_4 between these two competing processes. 254 Such a treatment for fine mode particles is similar to the kinetic approach used in regional air 255 quality models, except that the condensation is assumed to be irreversible with lower limit values 256 of mass accommodation coefficients in this work. 257 2.2.4 Modifications of Existing Aerosol Treatments

258 MAM 7 does not treat NO_3 ⁻ and it treats NaCl as one species. In this work, MAM7 is 259 modified to explicitly simulate NO_3 , Cl, and Na⁺ using a similar method to the condensation of 260 H₂SO₄ and NH₃. NO₃ and Cl are simulated in all modes except for primary carbon mode. Na⁺ is 261 simulated in sea-salt modes. The source of $Na⁺$ is calculated based on the mass ratio of Na and Cl 262 from sea-salt emissions. The source of Cl includes sea-salt emissions, and the condensation of 263 HCl resulted from HCl emissions and gas-particle partitioning of total chloride.

264 Species-dependent accommodation coefficients are used for H_2SO_4 , NH_3 , HNO_3 , and 265 HCl, with the values of 0.02, 0.097, 0.0024, and 0.005 (Zhang et al., 1998; Sander et al., 2002), 266 respectively. Since by default the model treats the condensation of inorganic volatile gas species 267 as irreversible process (no evaporation) (see equation (1)), the lower limit values of mass 268 accommodation coefficients are used for these species to represent their net fluxes from the gas-269 phase to the liquid/solid phases. Such lower limit values correspond to uptake coefficients, which 270 represent the net fluxes and are smaller than mass accommodation coefficients. To ensure 271 electroneutrality in each mode after kinetically condensing H_2SO_4 , NH₃, HNO₃, and HCl at 272 different condensation (or uptake) rates, the condensation of NH₃ will stop when the mole

273 concentration of cations (i.e., NH₄⁺) is equal to sum of those of anions (i.e., [NH₄⁺] = $2 \times [SO_4^{2-}$ 274]+[NO₃]+[Cl⁻]). While such an approach allows the gas/particle portioning of those volatile 275 species over both fine and coarse modes, the irreversible condensation with lower limit mass 276 accommodation coefficients assumed in this work, however, may contribute to model biases in 277 simulating condensation of volatile species on coarse mode particles. A more accurate method 278 (i.e., reversible condensation) should be used for volatile species for future work. The original 279 MAM7 treats $NH_3(g)/NH_4^+$ in cloud water. In this work, the dissolution and dissociation of 280 HNO₃ and HCl to produce NO₃⁻ and Cl⁻ in cloud water are added in the model based on Marsh 281 and McElroy (1985) and Seinfeld and Pandis (2006), i.e.,

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284 The concentration of H^+ (thus the pH value of the solution) is obtained by solving the electro-285 neutrality equation using the bisection method. The aqueous-phase chemical system is solved

 $HNO_{3(g)} \rightleftharpoons HNO₃(aq) \rightleftharpoons H⁺ + NO₃$

 $HCl_{(g)} \rightleftharpoons HCl_{(aq)} \rightleftharpoons H^+ + Cl^-$

287

286 analytically.

288 **3. Model Configurations and Evaluation Protocols**

289 3.1 Model Setup and Simulation Design

290 Table 1 summarizes the CESM/CAM5.1 simulations that are designed to examine the 291 impacts of individual new and modified treatments on model predictions. The first set of 292 simulations includes two simulations with the same default MAM7 coupled with different gas-293 phase mechanisms: one uses the simple gas-phase chemistry (MAM_SIM) with a total of 37 294 prognostic species and one uses the CB05_GE (MAM_CB05_GE) with a total of 127 prognostic

295 species. A comparison of the two simulations provides an estimate of the impacts of gas-phase 296 chemical mechanisms. The second set of simulations consists of five simulations that use the 297 same CB05_GE gas-phase mechanism but with modified and new aerosol treatments 298 individually and jointly. The first one is MAM_CON that uses an explicit treatment for $NO₃$, Cl, 299 and $Na⁺$ and species-dependent mass accommodation coefficients for condensation and that 300 includes the aqueous-phase chemistry of $HNO₃/NO₃$ and HCl/CI . This simulation includes a 301 total of 139 prognostic species. The second one is MAM_CON/IMN that uses the same 302 treatments as MAM_CON but with IMN as one of the nucleation mechanisms and a prefactor of 1.0×10^{-8} in WP09. The third one is MAM CON/ISO that uses the same treatments as 304 MAM_CON but with ISORROPIA II for aerosol thermodynamics assuming metastate 305 equilibrium (i.e., liquid only). The fourth one is MAM_NEWA that uses the same treatments as 306 MAM CON but with all new and modified aerosol treatments and a prefactor of 1.0×10^{-9} for 307 WP09. The fifth one is MAM_NEWB that uses the same treatments as MAM_NEWA, but with 308 ISORROPIA II assuming a stable condition (i.e., solid and liquid coexist). A comparison of 309 MAM_CB05_GE with MAM_CON indicates the impact of modified condensation and aqueous-310 phase chemistry. A comparison of MAM_CON/IMN, MAM_CON/ISO, and MAM_NEWA with 311 MAM_CON indicates the impacts of IMN, ISORROPIA II, and combined new and modified 312 aerosol treatments, respectively. Comparison of MAM_NEWB with MAM_NEWA indicates the 313 impacts of thermodynamic conditions on gas-aerosol partitioning. The $3rd$ set of simulation 314 includes one simulation using the same configuration as MAM_NEWA but with adjusted 315 emissions (MAM_NEW/EMIS). Its comparison with MAM_NEWA indicates the impacts of 316 uncertainties in emissions on model predictions. The $4th$ set of simulation includes one simulation 317 using the same configuration as MAM_SIM but with prescribed SST for a 5-yr period during 318 2001-2005 (MAM_SIM_5Y), and two simulations both using the same configuration as 319 MAM_NEW/EMIS for 2001-2005, but one with prescribed SST (MAM_NEW_5YA), and the 320 other in a fully-coupled mode (MAM_NEW_5YB).

321 All these simulations use the same approach for photolytic rates calculations based on 322 Lamarque et al. (2012), the same aqueous-phase chemistry of Barth et al. (2000), and the same 323 physical options as those in MAM_SIM. Major physical options include the cloud microphysics 324 parameterization of Morrison and Gettelman (2008), the moisture PBL scheme of Bretherton and 325 Park (2009), the shallow convection scheme and deep convection scheme of Park and Bretherton 326 (2009) and Zhang and McFarlane (1995), respectively, the aerosol activation parameterization of 327 Abdul-Razzak and Ghan (2000), and the Rapid Radiative Transfer Model for GCMs (RRTMG) 328 of Iacono et al. (2003, 2008) for long and short-wave radiation. The land surface processes are 329 simulated by the Community Land Model (CLM) of Lawrence et al. (2011) in CESM that is 330 coupled with CAM5.1.

331 All simulations except for MAM_SIM_5Y and MAM_NEW_5YA are performed with 332 fully-coupled CESM1.0.5 with standard B_1850-2000_CAM5_CN configuration, which 333 represents 1850 to 2000 transient conditions and includes all active components in CESM with 334 biogeochemistry in the land model. MAM_SIM_5Y and MAM_NEW_5YA are performed with 335 standard F_AMIP_CAM5 configuration, which uses a climatological dataset for SST provided 336 by NCAR for ocean model. The simulations are conducted for the full-year of 2001 and 2001- 337 2005 at a horizontal resolution of $0.9^{\circ} \times 1.25^{\circ}$ and a vertical resolution of 30 layers for CAM5.1. 338 The initial conditions for ice and ocean models are from CESM default settings. The initial 339 conditions for the land model are based on the output from the NCAR's CESM/CAM4 B_1850- 340 2000 CN simulation. The initial conditions for CAM5 are derived from a 10-yr (1990-2000)

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364 3.2 Available Measurements for Model Validation

365 A number of observational datasets from surface networks and satellites are used for 366 model evaluation. They are summarized along with the variables to be evaluated in Table A1 in 367 the supplementary material. Global surface networks include the Baseline Surface Radiation 368 Network (BSRN) and the National Oceanic and Atmospheric Administration Climate 369 Diagnostics Center (NOAA/CDC). The satellite datasets include the Moderate Resolution 370 Imaging Spectroradiometer (MODIS), the Clouds and Earth's Radiant Energy System (CERES), 371 the Total Ozone Mapping Spectrometer/the Solar Backscatter UltraViolet (TOMS/SBUV), the 372 Measurements Of Pollution In The Troposphere (MOPITT), and the Global Ozone Monitoring 373 Experiment (GOME). Other satellite-based data include the MODIS-derived CDNC from 374 Bennartz (2007) (BE07). 375 Regional observational networks include the Clean Air Status and Trends Network 376 (CASTNET), the Interagency Monitoring of Protected Visual Environments (IMPROVE), and 377 the Speciation Trends Network (STN) over CONUS; the European Monitoring and Evaluation 378 Program (EMEP), the Base de Données sur la Qualité de l'Air (BDQA), and the European air 379 quality database (AirBase) over Europe; the Ministry of Environmental Protection of China 380 (MEP of China), the National Institute for Environmental Studies of Japan (NIES of Japan), and 381 Taiwan Air Quality Monitoring Network (TAQMN) over East Asia. The observational data for 382 particle formation rate J is compiled from Kulmala et al. (2004) and Yu et al. (2008), which 383 include land-, ship-, and aircraft-based measurements. 384

385 3.3 Evaluation Protocol

- 406 Tables 2 and 3 show MBs and NMBs of radiative/cloud and chemical predictions,
- 407 respectively. The model performance of the baseline simulation, MAM_SIM, is discussed below,
- 408 and that for all other simulations will be discussed in section 5.

409 As shown in Table 2, radiative variables such as LWD and SWD are underpredicted by 410 3.4 W m⁻² (\sim -1.1%) and 2.0 W m⁻² (\sim -1.1%), respectively, whereas OLR and SWCF are 411 overpredicted by 8.8 W m⁻² (\sim 4.1%) and 3.2 W m⁻² (\sim 7.9%) respectively. Cloud variables such 412 as CF and PWV are slightly underpredicted, whereas COT, CWP, column CCN at a 413 supersaturation of 0.5% (CCN5), and CDNC are largely underpredicted, with NMBs of -77.8% 414 to -55.6%, which is likely due to the limitations in the current model treatments of cloud 415 microphysics and aerosol-cloud interactions in CAM5.1. 416 AOD is also underpredicted by 36.1%, which is likely due to inaccurate predictions of 417 aerosol concentrations and uncertainties in the assumed hygroscopicity of aerosol components in 418 the calculation of optical properties and water uptake. For example, as shown in Table 3, $PM_{2.5}$ 419 concentrations over CONUS and Europe, and PM₁₀ concentrations over CONUS, Europe, and 420 East Asia are underpredicted, with NMBs of -67.5% to -31.8%, which is due to the inaccurate 421 predictions of SO_4^2 , NH₄⁺, and organic aerosols, and missing major inorganic aerosol species 422 such as nitrate and chloride. The concentrations of BC, OC, and TC are underpredicted (by \sim 423 50%), which is likely due to the uncertainties in the BC and primary OC emissions as well as 424 treatments for SOA formation. In particular, the SOA treatment used in CAM5.1 is based on a 425 highly-simplified aerosol yield approach with a single lumped semi-volatile organic gas (i.e., 426 SOAG). For gaseous species, SO_2 concentrations over CONUS and Europe are significantly 427 overpredicted by 10.3 μg m⁻³ (\sim 264.8%) and 6.6 μg m⁻³ (\sim 97.5%), respectively, whereas SO₂ 428 concentrations over East Asia are largely underpredicted by 7.9 μ g m⁻³ (by ~63.0%). NH₃ 429 concentrations over Europe are also largely underpredicted by 82.0%. These large biases in $SO₂$ 430 and NH₃ are likely due in part to the uncertainties in the emissions of SO_2 and NH₃, which in 431 turn affect the predictions of SO_4^2 and NH_4^+ . The J values in PBL are highly underpredicted by

432 99.6%, which is mainly due to the inaccurate calculation of H_2SO_4 vapor concentration that 433 participates in the nucleation and uncertainties in the nucleation parameterizations used in the 434 default CESM/CAM5.1.

435

436 **5. Sensitivity Simulations**

437 5.1 Impacts of New Gas-Phase Chemistry

438 Compared to simple gas-phase chemistry, many more gaseous species and chemical 439 reactions simulated in CB05_GE can affect secondary aerosol formation through gas-to-particle 440 mass transfer and aqueous-phase chemistry and affect climatic variables through chemistry 441 feedbacks to the climate system. Figure 1a shows the absolute differences of H_2O_2 , SO_2 , SO_4^2 ², 442 and SOA between MAM_CB05_GE and MAM_SIM. MAM_CB05_GE treats more gaseous 443 species and chemical reactions than MAM_SIM, leading to large changes in the concentrations 444 of gaseous and PM species. Compared with MAM_SIM, MAM_CB05_GE predicts higher H_2O_2 445 by 0.4 ppb, SO₂ by 7.3 ppt, SO₄²⁻ by 0.01 μ g m⁻³, and SOA by 0.06 μ g m⁻³ in terms of global 446 mean. Those changes are mainly caused by different gas-phase chemical mechanisms used in 447 MAM SIM and MAM CB05 GE. While MAM CB05 GE explicitly simulates OH, HO₂, NO₃, 448 and O_3 , MAM SIM uses climatology data for these species. OH simulated by MAM CB05 GE 449 is lower than that prescribed by MAM_SIM by up to 2.8×10^6 molecules cm⁻³, or higher by up to 3.0×10^6 molecules cm⁻³ in different regions (Figure not shown), with a higher global mean by 451 MAM CB05 GE. MAM SIM includes the production of H_2O_2 from the self-destruction of HO_2 452 and the loss of H_2O_2 through its photolytic reaction and its reaction with OH. Higher H_2O_2 in 453 MAM CB05 GE is mainly due to greater production of H_2O_2 from additional chemical 454 reactions (e.g., OH+OH) than loss of H_2O_2 through the reactions of OH + H_2O_2 , O + H_2O_2 , Cl +

455 H₂O₂, and Hg + H₂O₂. Different predictions in H₂O₂ can in turn affect OH mixing ratios in 456 MAM CB05 GE but not in MAM SIM. In addition, the photolytic reactions of VOCs (e.g., 457 HCHO, peroxyacyl nitrates (PAN), and peroxyacetic and higher peroxycarboxylic acids (PACD)) 458 and other gases (e.g., HNO₃, HONO, HNO₄, HOCl, and HOBR) treated in MAM CB05 GE can 459 produce OH. Figure 1b shows the absolute differences between the mixing ratios of major 460 oxidants predicted from MAM_CB05_GE and climatology values used in MAM_SIM. The 461 global mean mixing ratios of oxidants are higher in MAM_CB05_GE than climatology data in 462 MAM SIM, leading to more oxidation of VOCs and therefore more SOA in MAM CB05 GE. 463 Higher O_3 predicted from MAM CB05 GE over most of the domain is mainly due to more O_3 464 precursors (e.g., $NO₂$ and VOCs) treated in the model. Despite higher OH mixing ratios in 465 MAM CB05 GE, many gaseous species such as NO_x , $SO₂$, HNO₃, HONO, and other VOCs are 466 oxidized by OH to form secondary inorganic and organic aerosols. Those oxidation reactions 467 compete for limited OH, leading to less oxidation of SO_2 , thus higher SO_2 mixing ratios over 468 most land areas by MAM CB05 GE. Lower SO_2 mixing ratios over the oceanic areas in 469 MAM CB05 GE is due to the combined effects of less production of SO_2 from lower DMS 470 mixing ratios (due to increased OH levels) and greater SO_2 oxidation from higher OH mixing 471 ratios.

472 The changes in the concentrations of PM and its components are due to the change in the 473 mixing ratios of gaseous precursors. CB05_GE contains more photolytic reactions, which affect 474 the mixing ratios of OH, SO₂, and H₂SO₄, and subsequently the concentration of SO₄² through 475 condensation and homogeneous nucleation. Higher SO_2 mixing ratios in MAM CB05 GE result 476 in more H₂SO₄ thus more SO₄². For example, both SO₂ mixing ratios and SO₄² concentrations 477 are higher over eastern China in MAM_CB05_GE. More SO_4^2 over the oceanic areas is mainly

501 oceanic emissions of HCl, leading to higher HCl over ocean. Higher isoprene mixing ratios over 502 South Africa, central Africa, and Oceania are mainly due to higher isoprene emissions in those 503 regions, which also contribute to the formation of SOA in those regions.

504 The aforementioned changes in the concentrations of gaseous species and PM due to new 505 gas-phase chemistry implemented in the model and its feedbacks to radiation through the climate 506 system result in a change in predicted cloud properties and radiation balance that in turn affect 507 the predictions of all chemical species during subsequent time steps. As a consequence of 508 interwoven changes due to complex feedback mechanisms, the two simulations perform 509 differently, with noticeable improvement by MAM_CB05_GE. As shown in Table 2, compared 510 with MAM SIM, MAM CB05 GE reduces MB of LWD by 17.6%, OLR by 8.0%, CF by 511 28.6%, COT by 1.0%, PWV by 28.0%, AOD by 5.5%, and CDNC by 1.8%, leading to 0.3-2.2% 512 absolute reduction in their NMBs. Although MAM CB05 GE increases MB of SWD by 26.2%, 513 the increases in their NMBs are only 1.2%. As shown in Table A1 in the supplementary material, 514 the changes in most cloud and radiative variables between MAM_SIM and MAM_CB05_GE are 515 statistically significant. As shown in Table 3, MAM CB05 GE also reduces MBs of SO_2 by 516 2.5% and PM₁₀ by 8.1% over East Asia, NH₃ by 1.3% and SO₄²⁻ by 12.5% over Europe, OC by 517 11.1%, TC by 8.3%, and $PM_{2.5}$ by 3.3% over CONUS, leading to 0.8-6.5% absolute reductions 518 in NMBs. Despite the model improvement by CB05_GE, large biases still remain for some 519 chemical species. For example, CO over East Asia is largely underpredicted with an NMB of - 520 82.1% (see Table 3), which results from the uncertainties in the CO emissions over East Asia. 521 However, the column CO over globe is predicted very well, with an NMB of -5.7%. Large biases 522 in SO2 predictions over CONUS, Europe, and East Asia are mainly due to the uncertainties in the 523 SO₂ emissions over those regions. Large biases in O₃ over Europe are likely due to the

525 over Europe. In particular, the large underpredictions in $NO₂$ concentrations (likely due to the 526 uncertainties in the NO_x emissions and overpredictions in radiation, see section 5.5 for more 527 detailed discussions) indicate insufficient NO_x for titration of $O₃$, leading to a large 528 overprediction in O_3 concentrations in Europe. The large biases in HNO_3 are due to no treatment 529 for gas-particle partitioning in both simulations. 530 531 5.2 Impacts of Condensation and Aqueous-Phase Chemistry 532 The mass accommodation coefficient (α) for H_2SO_4 vapor is subject to considerable 533 uncertainty. The default condensation module with a default α value of 0.65 gives a very low 534 concentration of H_2SO_4 , resulting in very low nucleation rates and aerosol number 535 concentrations. Considering that the original model treats H_2SO_4 and NH_3 condensation as an 536 irreversible process, the default α value of 0.65 for H₂SO₄ and NH₃ is reduced to 0.02 and 0.097, 537 respectively, based on Zhang et al. (1998). This change in α value provides sufficient H₂SO₄ and 538 NH₃ for nucleation with a typical H₂SO₄ concentration range of $10^6 \sim 10^8$ molecules cm⁻³. 539 Because HNO₃ and HCl are semi-volatile species, the lower limits of α (0.0024 and 0.005, 540 respectively) based on Sander et al. (2002) are selected for their irreversible condensation 541 process. NH₄⁺ from NH₃ condensation will be constrained by the available SO_4^2 ², NO₃², and 542 condensed Cl⁻ to neutralize the system. 543 Figure 3 shows the absolute differences of NH_3 , SO_2 , HNO_3 , HCl , H_2SO_4 , total 544 particulate ammonium (TNH₄), total particulate sulfate (TSO₄), total particulate nitrate (TNO₃), 545 and total particulate chloride (TCL) in all the modes except primary carbon mode, and $PM_{2.5}$ 546 between MAM_CON and MAM_CB05_GE in June, July, and August (JJA), 2001. Due to the

524 uncertainties in the O_3 precursor emissions (e.g., NO_x) and inaccurate predictions of radiation

547 inclusion of $HNO₃$ and HCl condensation in MAM CON, the concentrations of $HNO₃$ and HCl 548 decrease by 0.1 ppb $(\sim 72\%)$ and 0.097 ppb $(\sim 84\%)$, respectively. NO₃ is not simulated in the 549 original model and the concentration of $NO₃$ is assumed as zero in MAM_CB05_GE. Therefore, 550 the concentration of NO_3 increases due to the condensation of HNO_3 in MAM_CON . The 551 concentration of TCL in MAM_CB05_GE is calculated from the mass ratio of chloride in sea-552 salt. Over land, TCL increases significantly due to the condensation of HCl to form Cl. The 553 change of TCL over ocean is mainly due to the change of sea-salt emissions. The changes of SO_2 554 mixing ratios are mainly due to the differences in mixing ratios of species in sulfur chemistry in 555 the two simulations. For example, compared to MAM CB05 GE , the increase of SO₂ over 556 eastern U.S. in MAM CON is likely due to less $SO₂$ oxidation in clouds (Figure not shown), 557 which results from lower CF. The decrease of SO_2 mixing ratios over most oceanic areas is likely 558 due to the combined effects of DMS oxidation and SO_2 oxidations in MAM CON. More SO_2 559 can result in more H_2SO_4 and therefore more SO_4^2 through condensation and homogeneous 560 nucleation of H₂SO₄. The changes in H₂SO₄ concentrations are the results of changes of SO₂ 561 mixing ratios. The mass accommodation coefficient of H_2SO_4 is reduced significantly (by a 562 factor of 32.5), allowing more H_2SO_4 to participate in binary/ternary homogeneous nucleation 563 and produce more secondary SO_4^2 , improving predictions of SO_4^2 over CONUS but degrading 564 the performance of SO_4^2 over Europe (see Table 3). Although the mass accommodation 565 coefficient of NH₃ is reduced significantly (by a factor of 67), more available NH₃ can participate 566 in the ternary homogeneous nucleation and produce secondary NH_4^+ . Meanwhile, the secondary 567 NH₄⁺ formed from NH₃ condensation is also constrained by available SO_4^2 ², NO₃⁻, and 568 condensed Cl. As a result, the concentrations of NH₃ decrease and those of NH₄⁺ increase. Due 569 to more available H_2SO_4 participating in the nucleation, J has been improved significantly,

- 592 uncertainties in treating $HNO₃$ and HCl as non-volatile species using their lower limits of 593 accommodation coefficients, and lack of treatments for $NO₃$ and Cl thermodynamics.
- 594
- 595 5.3 Impacts of New Particle Formation

596 Figure 4 shows the annual-mean vertical distributions of particle formation rate (J) values 597 and aerosol number concentrations, and simulated J values averaged between the ground level 598 and 1000-m overlaid with observations within the same layers. In MAM_CON/IMN, IMN is 599 combined with three default nucleation parameterizations to predict J throughout the atmosphere. 600 In MAM_CON, J over ocean is overpredicted by factors of 5-50, despite a seeming good NMB 601 of -12.8% in the globe mean (see Table 3). J values at several sites over land are underpredicted 602 by factors of 1-10, which compensates the large overpredictions at most sites over ocean. The 603 large underpredictions at those sites are likely due to the uncertainties in SO_2 emissions and 604 nucleation parameterizations, and the missing species that may have participated in nucleation. 605 For example, several other species may contribute to the new particle formation, including 606 methanesulfonic acid (van Dingenen and Raes, 1993), hydrochloric acid (Arstila et al., 1999), 607 organic compounds (Berndt, et al., 2013), iodine-containing compounds (Hoffmann et al., 2001), 608 and amines (Berndt et al., 2013). Limited observations also introduce some uncertainties in the 609 model validation. The overprediction of J over ocean is mainly due to the use of the prefactor of 1×10^{-6} in WP09. This prefactor is derived from limited in-situ measurements (Sihto et al., 2006). 611 It can vary by up to 3-4 orders of magnitude based on measurements in different areas and 612 seasons (Zhang et al., 2010), introducing a large uncertainty for its application to the global 613 scale. In MAM CON/IMN, a prefactor of 1×10^{-8} is used in WP09 in PBL over the globe, which 614 then decreases J and aerosol number concentrations in PBL (see Figure 4). J in PBL is very

615 sensitive to the prefactor in WP09, and the uncertainty of the prefactor can result in a large bias 616 in predictions of J and aerosol number in PBL. With the implementation of IMN, J values in the 617 troposphere increase by factors of 2-10, which in turn increase the aerosol number concentrations 618 in the troposphere. Due to a stronger radiation in the upper layer, more available ions can 619 contribute to the new particle formation, therefore increasing the aerosol number concentrations 620 in the middle/upper troposphere and lower stratosphere by factors of 2-4.

621 Figure 5 shows the absolute differences of $PM_{2.5}$, AOD, column CCN5, CF, SWCF, and 622 SWD between MAM_CON and MAM_CON/IMN for 2001. Aerosol number can directly affect 623 CCN, which can affect cloud formation and properties as well as radiation. Changes of PM 624 concentrations also have impacts on AOD, CCN, CF, COT, and SWCF through both aerosol 625 direct and indirect effects. As a net result of all those interwoven changes initially triggered by 626 the increase of aerosol number concentrations in troposphere/stratosphere, AOD and column 627 CCN5 increase by 0.004 (or by 3.3%) and 2.1×10^7 cm⁻² (or by 11.9%), respectively, and SWCF 628 and SWD decrease by 0.1 W m⁻² (or by 0.2%) and 0.8 W m⁻² (or by 0.5%), respectively, in terms 629 of global mean. As shown in Table A1, the changes in SWD, AOD, and cloud variables such as 630 column CCN5, CDNC, and COT between MAM_CON and MAM_CON/IMN are statistically 631 significant, indicating the significant impacts of IMN on aerosol number concentration and cloud 632 prediction.

633 Compared with MAM_CON, IMN (MAM_CON/IMN) improves the predictions of $SO₂$, 634 NO₃⁻, and PM_{2.5} over CONUS, SO₂, SO₄²⁻, NH₄⁺, NO₃⁻, Cl⁻, PM_{2.5}, and PM₁₀ over Europe, PM₁₀ 635 over East Asia (see Table 3). The improved performance in aerosol concentrations and increased 636 aerosol numbers in the troposphere and lower stratosphere contribute to the improved 637 performance of aerosol and cloud parameters, with increased AOD, CCN, and CDNC, and

638 consequently increased CF, COT, CWP, and SWCF, as shown in Table 2. However, there are still 639 large biases for some chemical species predictions. For example, CO mixing ratio is 640 underpredicted over East Asia, which is mainly due to the uncertainty in CO emissions in this 641 region. Large biases in SO2 predictions over CONUS, Europe, and East Asia are mainly due to 642 the uncertainties in SO_2 emissions in those regions. Large biases in NO_2 and HNO_3 predictions 643 over Europe are mainly due to the uncertainties in NO_x emissions and inaccurate predictions of 644 radiation over this region. The performance of J degrades with NMBs from -21.8% to -49.6% in 645 the globe, which is due to the use of a smaller prefactor of WP09 in MAM_CON/IMN than in 646 MAM CON. J in PBL is very sensitive to the prefactor in WP09. Although the prediction of J 647 over ocean in PBL has been improved in MAM_CON/IMN, J over land areas in PBL is largely 648 underpredicted by factors of 1-100, resulting in degraded J performance in terms of globe mean. 649 The underprediction of J over land in PBL is likely due to the uncertainties in the nucleation 650 parameterizations (e.g., the missing species as mentioned previously). Large NMBs still remain 651 for COT, CWP, and CCN, indicating the uncertainties in the treatments of related atmospheric 652 processes such as cloud microphysics and aerosol-cloud interactions.

653

654 5.4 Impacts of Gas-Aerosol Partitioning

655 The inclusion of ISORROPIA II changes the mass concentrations of major $PM_{2.5}$ species 656 and their gaseous precursors. Changes in PM concentrations then affect predictions of cloud 657 variables and therefore radiation. Changes of radiation can also affect $SO₂$ oxidation by OH, 658 which affects H_2SO_4 concentrations. Figure 6 shows the absolute differences of H_2SO_4 , fine 659 particulate sulfate (SO4f), NH₃, fine particulate ammonium (NH4f), HNO₃, fine particulate 660 nitrate (NO3f), HCl, and fine particulate chloride (CLf) for summer 2001 between MAM_CON

726 Large biases in some variables remain in MAM_NEWA due to uncertainties in model 727 inputs (e.g., meteorology and emissions) and model treatments (e.g., multi-phase chemistry, dust 728 emission scheme, cloud microphysics, aerosol activation, SOA formation, and dry and wet

729 deposition). The large NMBs of CO and SO_2 over East Asia, SO_2 , NH₃, and NO₂ over Europe, 730 SO2, and BC over CONUS are likely due to the uncertainties of emissions and the interpolation 731 of emissions from a fine-grid scale in the original emission inventories (e.g., county-based 732 emissions over CONUS) to a large-grid scale used in this work, which can result in large NMBs 733 in secondary aerosols (e.g., SO_4^2 , NH_4^+ , NO_3 , thus $PM_{2.5}$ and PM_{10}). Heterogeneous reactions 734 are not included in this work, which may help explain to some extent less oxidation and 735 underpredictions for PM species predictions (e.g., sulfate and nitrate) and overpredictions for 736 gaseous species. The large NMB of O_3 predictions over Europe in MAM NEWA (with an NMB 737 of 62.7%) is mainly due to a lack of NO_x titration (as indicated by large underpredictions in NO_2) 738 and more production of O_3 from the photolytic reaction of NO₂ resulted from overpredictions of 739 SWD particularly in autumn and winter. Table 4 shows the seasonal statistics for O_3 , NO_2 , and 740 HNO₃ over Europe in MAM NEWA. During autumn and winter, O_3 is overpredicted by about 741 100% ~ 140%, whereas NO₂ is underpredicted by about -85% ~ -20%, indicating insufficient 742 NO_x for titration of O₃ titration. SWD is overpredicted by 45.0 W m⁻² (or by 58.4%), favoring the 743 photolytic reactions of NO₂ to produce O₃. Due to the uncertainties in the NO_x emissions, NO₂ is 744 underpredicted, causing less $NO₂$ to be oxidized to produce HNO₃, which results in an 745 underprediction of HNO₃ in winter. In autumn, SWD is overpredicted by 42.8 W m⁻² (or by 746 37.9%). However, in autumn, although $NO₂$ is underpredicted due to the uncertainties in the NO_x 747 emissions, HNO₃ mixing ratios are overpredicted. SWD is stronger in autumn than in winter, and 748 mixing ratios of OH are higher due to photolytic reactions of overpredicted O_3 and additional 749 photolytic reactions of VOCs. Therefore, OH can oxidize $NO₂$ to produce HNO₃, resulting in the 750 overprediction of HNO3. Simple aqueous-phase chemistry is included in this work, which could 751 result in high uncertainty in predicting aerosols in clouds. Decreased aerosol number

752 concentrations can result in a decrease of CCN and AOD directly. The underpredictions of 753 CDNC are likely due to uncertainties in the model treatments for aerosol activation and cloud 754 microphysics, which then result in large NMBs in COT and CWP. The large biases in OC and TC 755 indicate the uncertainties in the emissions of BC and primary OC, and the treatments for SOA 756 formation. The large NMB in particle formation rate J is likely due to uncertainties in model 757 inputs (e.g., SO_2 emissions) and model treatments (e.g., the accommodation coefficient of H_2SO_4 758 and missing participants in the current nucleation schemes).

759

760 5.6 Impacts of Adjusted Emissions

761 The evaluation and analyses of MAM_NEWA indicate that some large biases are caused 762 by inaccuracies in the emissions of CO , SO_2 , BC , OC , and NH_3 . The sensitivity simulation with 763 adjusted emissions of CO, SO_2 , BC, OC, and NH₃ (MAM NEW/EMIS) is performed to further 764 look into such impacts. For example, with 30% increase in CO emissions and 20% increase in 765 NH3 emissions over Europe, the NMBs of surface concentrations of CO and NH3 change from-766 3.4% to 12.1%, -84.3% to -77.5%, respectively. On a global scale, the increased CO emissions 767 result in 3.0% absolute reduction in the NMB of column CO. The 30% reduction in SO_2 768 emissions and 20% increase in OC and BC emissions over CONUS result in 139.6%, 8.6%, and 769 24.9% absolute reduction in their NMBs. The 30% increase in CO emissions and 20% increase 770 in SO₂ over East Asia result in 3.3% and 7.8% absolute reduction in their NMBs. 771 As shown in Table 3, compared with MAM_NEWA, MAM_NEW/EMIS shows an 772 improved performance in the concentrations of SO_2 , HNO_3 , SO_4^2 ⁻, NH_3 , and NH_4^+ over Europe, SO_2 , HNO₃, BC, OC, TC, NO₃⁻, and Cl⁻ over CONUS, CO and SO₂ over Asia, and column CO

774 over globe. However, it degrades to some extent the performance of SO_4^2 and NH_4 ⁺ over

- 795 0.4-4.2% absolute reduction in their NMBs). MAM_NEW_5YA also shows slight improvement
- 796 for the predictions of SO_4^2 and BC over CONUS and SO_2 over East Asia (with 0.3-2.3%)
- 797 absolute reduction in their NMBs), but moderate-to-large improvements for the predictions of

820 multi-phase chemistry, dust emission scheme, cloud microphysics, aerosol activation, SOA

821 formation, and dry and wet deposition), which have been illustrated in Section 5.5. Large biases 822 in Cl predictions over Europe are likely due to the combined effects of a low concentration of 823 observed Cl, uncertainties in HCl emissions, and inaccurate predictions of coarse Cl in the 824 model since ISORROPIA II is only implemented for fine particles. Uncertainties in the mass 825 accommodation coefficients of volatile gas species can also result in uncertainties in predictions 826 of condensable gases.

827

828 6.2 Impact of New and Modified Treatments on 5-year (2001-2005) Simulations

829 Figure 7 shows the absolute differences of surface SO_2 , NH₃, SO_4^2 ⁻, NH₄⁺, TC, PM_{2.5}, 830 PM₁₀, J, and aerosol number (PM_{num}) and Figure 8 shows the absolute differences of radiative 831 variables between MAM_SIM_5Y and MAM_NEW_5YA. The new and modified model 832 treatments in MAM_NEW_5YA cause changes in the concentrations of PM and precursor gases, 833 which affect radiative variables through aerosol direct and indirect effects. The changes in 834 radiative variables in turn affect gas-phase chemistry and aerosol processes. As shown in Figure 835 7, the difference of SO_2 between the two simulations varies from -1.7 to 3.8 ppb, with a global 836 mean difference of 4.2 ppt. The decrease of $SO₂$ over most oceanic area is mainly due to the 837 decrease of DMS resulted from less oxidation by OH radicals. The increase of SO_4^2 over East 838 • Asia and eastern U.S. drives more NH₃ from gas-phase to particulate phase to form NH₄⁺ 839 through thermodynamic equilibrium, increasing the concentrations of NH_4^+ over these regions. 840 However, the concentrations of SO_4^2 decrease over Europe due in part to less oxidation of SO_2 . 841 Despite such a decrease, the concentrations of NH_4^+ are higher over Europe due to the 842 neutralization of NH₃ by Cl⁻ and NO₃ that are treated in MAM_NEW_5YA but not treated in in 843 MAM_SIM_5Y. Compared with MAM_SIM_5Y, J from MAM_NEW_5YA increases over

845 accommodation coefficient of H_2SO_4 in MAM_NEW_5YA, resulting in more available H_2SO_4 846 vapor participating in nucleation. The increases in J result in an increase in aerosol mass and 847 number concentrations and thus higher concentrations of $PM_{2.5}$ and PM_{10} (which improve 848 appreciably their performance, see Table 5). 849 As shown in Figure 8, compared with MAM_SIM_5Y, AOD increases by 0.007, column 850 CCN5 increases by 3.8×10^7 cm⁻², and CDNC increases by 16.1 cm⁻³ in MAM_NEW_5YA. 851 Higher PM_{num} in MAM NEW 5YA allows more aerosol to grow into the CCN size, leading to 852 higher CCN in MAM_NEW_5YA. Higher aerosol concentrations in MAM_NEW_5YA result in 853 higher AOD. The increased aerosol number and mass concentration result in an increase in the 854 predictions of cloud variables through the aerosol-cloud interactions. For example, with all the 855 modified and new treatments, COT increases by 0.8, CWP increases by 4.1 g m⁻², and PWV 856 increases by 0.026 cm on global average. Due to the aerosol direct and indirect effects, the 857 difference in simulated SWD varies from -19.3.0 to 10.4 W m⁻² and decreases by 3.4 W m⁻² (\sim 858 2.0%) on a global average. The difference in LWD varies from -4.2 to 8.5 W m⁻² and increases by 859 1.0 W m⁻² (\sim 0.4%) on a global average (Figure not shown). The difference in SWCF varies from 860 -8.4 to 17.9 W m⁻² with a net increase of 2.7 W m⁻² (\sim 6.4%) on a global average. The absolute 861 differences of surface chemical species and major cloud/radiative variables for JJA average of 862 2001-2005 are shown in Figures A4 and A5 in the supplementary material. Compared with 5- 863 year average, the absolute changes of most radiative variables are smaller in JJA. The absolute 864 changes of PM₁₀ are smaller in JJA, which is mainly due to the dust events during other months 865 (e.g., March-May over East Asia).

844 globe with a global mean difference of 0.066 cm⁻³ s⁻¹, due to the use of a lower mass

867 6.3 Global Burden Analysis

868 Table 7 shows the simulated global burdens of major gas and aerosol species for 2001- 869 2005. The global burdens of most gaseous precursors of aerosol are higher in MAM_NEW_5YA 870 than MAM SIM 5Y (except for NH₃), due mainly to the incorporation of ISORROPIA II in 871 MAM NEW 5YA. The global burden of tropospheric O_3 is higher in MAM NEW 5YA than 872 MAM SIM 5Y, which is due to higher mixing ratios of O_3 precursors (e.g., NO₂ and VOCs) that 873 are simulated in MAM_NEW_5YA. The global burdens of most gas species are comparable with 874 previous studies (Horowitz et al.2006; Lamarque et al., 2006; Williams et al., 2009; Liu et al., 875 2012) with absolute differences of less than 20%. One exception is H_2SO_4 , which is higher by a 876 factor 5 in MAM_NEW_5YA than in MAM_SIM_5Y. The higher burden of H_2SO_4 in 877 MAM NEW 5YA is likely due to the less condensation of H_2SO_4 resulted from the use of a 878 lower mass accommodation coefficient. SO_4^2 burden is higher by 8.3% in MAM_NEW_5YA 879 than MAM SIM 5Y, which is likely due to greater SO_2 oxidation in MAM NEW 5YA. Higher 880 SO_4^2 burden results from higher SO_2 burden. Higher SO_2 burden leads to more SO_2 to be 881 oxidized to produce SO_4^2 , which overweighs the impacts from less H_2SO_4 condensation due to 882 lower mass accommodation coefficient. More SO_4^2 result in more NH₄⁺. The burdens of BC and 883 POM are lower by 16.5% and 23.8%, respectively, in MAM_NEW_5YA than in MAM_SIM_5Y, 884 which is likely due in part to greater dry deposition fluxes and in part to a slower primary carbon 885 aging rate resulted from reduced condensation of gas species in MAM_NEW_5YA. 886 Condensation onto the primary carbon mode produces aging of the particles in this mode. A 887 lower accommodation coefficient is used in MAM_NEW_5YA, which results in less 888 condensation. Therefore, the fraction of aged particles has decreased. The global burdens of most 889 aerosol species are in the range of previous studies. For example, global burdens of SO_4^2 and

901

902 **7. Conclusions and Future work**

903 In this work, a new gas-phase chemistry mechanism and several advanced inorganic 904 aerosol treatments have been incorporated into CESM/CAM5.1-MAM7. These include (1) the 905 CB05_GE gas-phase chemical mechanism coupled with MAM7; (2) the condensation and 906 aqueous-phase chemistry involving $HNO₃/ NO₃$ and $HCl/ Cl⁺$; (3) an ion-mediated nucleation 907 (IMN) parameterization for the new particle formation from ions, (4) an inorganic 908 thermodynamic module, ISORROPIA II, that explicitly simulates thermodynamics of SO_4^2 , 909 NH₄⁺, NO₃⁻, Cl⁻, and Na⁺ as well as the impact of crustal species, such as Ca²⁺, K⁺, and Mg²⁺, on 910 aerosol thermodynamics. CB05 GE with new and modified inorganic aerosol treatments in 911 MAM7 simulates 139 species with 273 chemical reactions, which is more accurate than simple 912 gas chemistry coupled with default MAM7. Seven 1-yr simulations for 2001 and three 5-yr

913 simulations for 2001-2005 with different model configurations are performed to evaluate the 914 capabilities of the original and improved CESM/CAM5 and the mechanisms underlying 915 differences among model predictions.

916 Comparing to the simple gas-phase chemistry, the 2001 simulation with CB05 GE can 917 predict many more gaseous species, and give improved performance for predictions of organic 918 carbon and $PM_{2.5}$ over CONUS, NH₃ and SO₄² over Europe, SO₂ and PM₁₀ over East Asia, and 919 cloud properties such as CF, CDNC, and SWCF. MAM_CON simulates $NO₃$ and Cl, which are 920 important inorganic aerosols. With species-dependent accommodation coefficients for gas 921 condensation, more H_2SO_4 can participate in homogeneous nucleation, resulting in the 922 improvement of predictions of $PM_{2.5}$, $PM₁₀$, J, CDNC, and SWCF. IMN can increase the 923 predictions of J and PM_{num} in the upper atmosphere and thus improve the predictions of AOD, 924 CCN, and cloud properties, and SWCF over globe, $PM_{2.5}$ over CONUS and Europe, PM_{10} over 925 Europe and East Asia, and PM composition over Europe. The 2001 simulation with ISORROPIA 926 II can improve the predictions of major gas and aerosol species significantly, including $HNO₃$, 927 NH₄⁺, NO₃⁻, Cl⁻, BC, OC, TC, and PM_{2.5} over CONUS, SO₂, NH₃, NO₂, SO₄², NH₄⁺, NO₃⁻, and 928 Cl⁻ over Europe, and CO and $SO₂$ over East Asia. Such improvements lead to improved 929 predictions of SWD, SWCF, and CCN5 over globe. The 2001 simulation with the new and 930 modified inorganic aerosol treatments appreciably improve the predictions of OLR, CF, COT, 931 CWP, PWV, CCN, CDNC, SWCF, J over globe, and $HNO₃$, $NH₄⁺$ (CONUS), $PM_{2.5}$, and $PM₁₀$. 932 The 2001 sensitivity simulation with adjusted emissions further improves model predictions of 933 CO and SO_2 over East Asia, SO_2 , HNO₃, NO₃⁻, Cl⁻, BC, OC, and TC over CONUS, SO_2 , NH₃, 934 NH_4^+ , HNO₃, NO₃⁻, and Cl⁻ over Europe, and column CO and SWD over globe. The change of 935 emissions can affect primary gaseous precursors directly, and secondary gaseous species

936 indirectly through gas-phase chemistry. Meanwhile, secondary aerosols can be affected by 937 gaseous precursors, and therefore have impacts on cloud properties as well as direct and indirect 938 effects on radiation. Reducing the uncertainty of emissions can thus help reduce the model biases

939 significantly.

940 The comparison of the 5-yr simulations with prescribed SST shows that

941 MAM NEW 5YA with CB05 GE can appreciably improve the predictions of AOD, COT, CWP,

942 CCN5, CDNC, SWD, LWD, OLR, and SWCF on global scale and OC, TC, and PM_{2.5} over

943 CONUS, PM_{10} over East Asia, and SO₂, $PM_{2.5}$, and PM_{10} over Europe. The performance is

944 overall similar for all radiative variables and most chemical species between MAM_NEW_5YA

945 with prescribed SST and MAM_NEW_5YB in a fully-coupled mode.

946 In addition to uncertainties in emissions, additional uncertainties exist in the model 947 treatments. For example, the large biases in the predictions of O_3 over Europe and East Asia are 948 mainly due to insufficient NO_x titration resulting from the uncertainties in the NO_x emissions. 949 The large biases in PM_{10} over East Asia and Europe may be mainly due to the inaccurate predictions of dust. The large bias in Cl- 950 over Europe may be due to the inaccurate predictions of 951 HCl and coarse Cl, resulted from the irreversible condensation of HCl over coarse mode 952 particles, and the uncertainty in the mass accommodation coefficient of HCl used. A reversible 953 condensation treatment should be used for volatile species in the future, which can more 954 accurately simulate the gas/particle partitioning of those volatile species over coarse mode 955 particles. Assumptions associated with equilibrium partitioning for fine particles such as 956 metastable conditions may be responsible for biases over desert/arid regions under low RH 957 conditions. In the default and modified nucleation treatments, it only considers H_2SO_4 , NH₃, 958 H2O, and ions involving in the new particle formation. Missing species (e.g., organics, iodine

959 compounds, and DMS) may also contribute to the new particle formation. Uncertainties in 960 treating organic gas-aerosol partitioning may contribute to the inaccurate predictions of SOA, 961 OC, TC, and PM. The large biases in CDNC, COT, and LWP indicate the uncertainties in cloud 962 microphysics schemes and aerosol-cloud interaction parameterizations, which also limit the 963 ability of climate and Earth system models to quantify aerosol indirect effects (Stephens, 2005; 964 Gettelman et al., 2008). In addition to uncertainties in the model treatments, uncertainties in the 965 model simulation settings such as the use of a coarse grid resolution and a large model time step 966 of 1800 seconds for solving the chemical system in this work may contribute to the model biases. 967 The representations of some of the aforementioned uncertain processes in CESM/CAM5.1 are 968 being further improved by the authors' group. Decadal simulations using improved 969 CESM/CAM5.1 will be conducted in the future to study the interactions among atmospheric 970 chemistry, aerosol, and climate change and reduce associated uncertainties.

971

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Table 2. Mean Bias (MB) and Normalized Mean Bias (NMB, in %) of Radiative/Cloud Predictions for the 2001 Simulations Table 2. Mean Bias (MB) and Normalized Mean Bias (NMB in %) of Radiative/Cloud Predictions for the 2001 Simulations

cThe values of modeled results (Sim), MBs, and NMBs are expressed as Sim/MB/NMB. $\frac{\left(\frac{http://www.pangaea.de\right)}{http://www.pangaea.de\right)}$.

"The values of modeled results (Sim), MBs, and NMBs are expressed as Sim/MB/NMB. (http://www.pangaea.de).

^bThe pair of observation and simulation is removed in the statistical calculation if the observed SWD value is lower than -10 or higher than 3000 W m⁻²

(http://www.pangaea.de).

Table 3 Mean Rias (MR) and Normalized Mean Rias (NMR in $\%$) of Chemical Predictions for the 2001 Simulations Table 3. Mean Bias (MB) and Normalized Mean Bias (NMB, in %) of Chemical Predictions for the 2001 Simulations

	Network		Obs $(\mu g \, m^{-3})$	Sim	MB/NMB
					$37.5/99.6^a$
	Airbase		26.0		$-18.4/-70.9$
			31.0		43.2/139.2
Winter	BDQA		30.6		$-25.0/-81.9$
			50.7	75.7	25.0/49.3
	EMEP	NO ₂	9.0	8.3	$-0.7/-7.8$
		NO ₂ O_3 NO ₂ O ₃ HNO ₃ O_3 NO ₂ O_3 NO ₂ O_3 NO ₂ HNO ₃ O_3 NO ₂ O_3 NO ₂ O_3 NO ₂ HNO ₃ O_3 NO ₂ O_3 NO ₂ O_3 NO ₂	0.5	0.5	$-4.9\times10^{-3}/1.0$
Spring			63.1	100.8	37.7/59.7
	Airbase		20.0	4.6	$-15.4/-77.1$
	BDQA		59.6	98.9	39.3/65.9
			23.6	3.1	$-20.5/-87.0$
	EMEP		75.0	101.9	26.9/35.9
			5.9	4.9	$-1.0/-17.2$
			0.4	0.9	0.5/144.5
	Airbase		64.9	93.5	28.6/44.0
			16.2	4.4	$-11.8/-72.8$
	BDQA		64.5	94.5	30.0/46.5
Summer			18.7	$(\mu g \, m^{-3})$ 75.2 7.6 74.2 5.6 3.6 91.2 4.4 1.3 79.5 5.3 80.9 3.7 78.2 5.2 0.9	$-15.1/-80.9$
			72.2		19.0/26.3
	EMEP		4.7		$-0.3/-6.2$
			0.5		0.8/169.6
	Airbase		40.5		39.0/96.4
			21.7		$-16.4/-75.6$
	BDQA		35.7		45.2/126.5
Autumn			24.8		$-21.1/-85.2$
	EMEP		51.7		26.5/51.2
			6.6		$-1.4/-21.1$
	^a The values of MBs and NMBs are expressed as MB/NMB.	HNO ₃	0.6		0.3/45.0

Table 4. The observed values and the mean bias (MB) and normalized mean bias (NMB, in %) of predictions of O₃ NO₂, and HNO₃ mixing ratios over Europe in MAM_NEWA

(http://www.pangaea.de).
°The values are expressed as Sim/MB/NMB/NME/RMSE. Sim: simulated values; MB: mean bias; NMB: normalized mean bias (%); NME: normalized mean
error (%); RMSE: root mean squared error. cThe values are expressed as Sim/MB/NMB/NME/RMSE. Sim: simulated values; MB: mean bias; NMB: normalized mean bias (%); NME: normalized mean error (%); RMSE: root mean squared error.(http://www.pangaea.de).

Species/	Domain	Obs.	Simulations				
variables ^a			MAM_SIM_5Y	MAM_NEW_5YA	MAM_NEW_5YB		
\bf{CO}	East Asia	562.0		139.7/-422.3/-75.1/75.1/451.8 ^b	137.0/-425.0/-75.6/75.6/454.0		
	CONUS	3.4	9.6/6.2/183.9/184.6/9.9	10.1/6.7/198.8/199.1/10.6	10.3/6.9/203.1/203.5/10.9		
SO ₂	Europe	6.6	$6.0/-0.6/-9.3/73.3/7.9$	$6.6/-0.06/-0.9/77.2/8.3$	$6.2/-0.4/-5.5/74.6/8.0$		
	East Asia	3.4	3.4/0.04/1.1/76.0/5.0	3.4/0.01/0.4/76.2/5.0	$3.4/-0.05/-1.6/73.1/4.8$		
NH ₃	Europe	6.3	3.0/-3.3/-52.0/81.0/25.3	2.4/-3.9/-61.3/79.7/25.3	2.4/-3.9/-62.0/79.3/25.3		
NO ₂	Europe	23.5		5.8/-17.7/-75.4/76.5/21.5	5.5/-18.0/-76.7/77.7/21.7		
	East Asia	13.5	$\frac{1}{2}$	2.3/-11.2/-83.3/83.3/12.2	2.3/-11.2/-83.6/83.6/12.2		
	CONUS	35.1	$\overline{}$	43.9/8.8/25.1/27.3/11.3	44.1/9.0/25.7/27.7/11.6		
O ₃	Europe	52.7	$\qquad \qquad \blacksquare$	86.6/33.9/64.5/64.6/36.4	89.2/36.5/69.3/69.4/38.8		
	East Asia	27.4	$\frac{1}{2}$	45.6/18.2/66.4/66.4/19.2	45.5/18.1/66.0/66.0/19.1		
	CONUS	1.4	$\overline{}$	1.6/0.2/16.3/39.5/0.7	1.6/0.2/12.1/38.2/0.7		
HNO ₃	Europe	0.7	$\overline{}$	1.0/0.3/45.8/83.5/0.8	1.0/0.3/37.9/79.8/0.8		
SO ₄ ²	CONUS	2.6	$2.3/-0.3/-13.4/26.9/1.0$	$2.3/-0.3/-13.1/23.0/0.8$	2.3/-0.3/-12.8/24.2/0.9		
	Europe	2.3	2.3/-0.04/-1.9/37.3/1.4	$2.0/-0.3/-11.1/34.1/1.3$	$2.0/-0.3/-13.0/35.5/1.4$		
$\mathbf{NH_4}^+$	CONUS	1.2	$0.9/-0.3/-20.8/33.4/55.0$	1.5/0.3/22.2/43.2/0.8	1.5/0.3/26.4/44.3/0.8		
	Europe	1.0	$0.8/-0.2/-16.8/36.9/0.5$	1.6/0.6/62.8/68.7/0.9	1.5/0.5/53.8/60.3/0.8		
NO ₃	CONUS	1.1	\overline{a}	1.6/0.5/41.3/85.4/1.4	1.6/0.5/49.8/90.2/1.5		
	Europe	1.8	$\qquad \qquad \blacksquare$	2.3/0.5/30.3/51.1/1.2	2.2/0.4/24.7/47.0/1.1		
CI	CONUS	0.1	\overline{a}	$0.1/3.1 \times 10^{-3}/2.7/105.8/0.4$	$0.1/8.7 \times 10^{-3}/7.8/110.1/0.4$		
	Europe	0.3	\overline{a}	2.4/2.1/681.2/681.2/2.9	2.3/2.0/663.3/663.6/2.8		
BC	CONUS	0.4	$0.3/-0.1/-17.9/44.4/0.3$	$0.3/-0.1/-15.6/44.0/28.2$	$0.3/-0.1/-17.7/44.3/0.2$		
OC	CONUS	1.2	$0.9/-0.3/-23.2/59.3/1.0$	$1.1/-0.1/-7.7/56.7/1.0$	$1.1/-0.1/-11.0/54.3/0.9$		
TC	CONUS	3.1	1.4/-1.7/-54.4/62.8/2.8	1.7/-1.4/-45.7/57.1/2.6	$1.6/-1.5/-47.1/57.1/2.7$		
PM _{2.5}	CONUS	$8.8\,$	7.2/-1.6/-17.9/37.0/4.3	9.2/0.4/4.1/33.5/3.9	$8.7/-0.1/-1.1/29.4/3.6$		
	Europe	14.6	6.7/-7.9/-53.9/54.6/10.6	9.7/-4.9/-33.8/37.6/8.6	10.0/-4.6/-31.7/36.1/8.4		
PM_{10}	Europe	26.3	15.1/-11.2/-42.6/46.8/15.9	18.7/-7.6/-28.8/36.1/13.9	19.9/-6.4/-24.4/33.5/13.1		
	East Asia	107.9	45.4/-62.5/-58.0/59.3/70.7	52.5/-57.4/-53.2/54.2/66.0	57.8/-50.1/-46.5/50.0/61.6		
Col. CO	Globe	1.4×10^{18}		$1.3\times10^{18}/-1.4\times10^{17}$ /	$1.2\times10^{18}/-1.5\times10^{17}/$		
				$-10.2/16.5/3.1 \times 10^{17}$	$-11.0/17.2/3.2 \times 10^{17}$		
Col. NO ₂	Globe	5.3×10^{14}		$8.4\times10^{14}/3.1\times10^{14}$	$8.3\times10^{14}/3.0\times10^{14}$		
				$59.2/70.0/5.4 \times 10^{14}$	$57.6/69.2/5.4 \times 10^{14}$		
TOR	Globe	30.4	29.9/-0.5/1.6/16.3/6.1	30.5/0.1/0.4/15.0/5.8	29.9/-0.5/-1.7/16.4/6.1		

Table 6. Statistical Performance of Chemical Predictions (Average of from the 5-yr (2001-2005) Simulations)

^aThe units are CO, ppm (over East Asia); SO₂, ppb (over East Asia) and μ g m⁻³ (over CONUS); O₃, ppb (over CONUS) and μ g m⁻³ (over Europe); column CO and NO₂, molecules cm⁻²; TOR, DU. All other concentrations are in μ g m⁻³.
^bThe values are expressed as Sim/MB/NMB/NME/RMSE. MB: mean bias; NMB: normalized mean bias (%); NME: normalized

mean error (%); RMSE: root mean square error.

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	MAM SIM 5Y	MAM NEW 5YA	Previous studies				
Tropospheric CO $(Tg)^a$	N.A. ^c	322.06	337-354 d				
Tropospheric O_3 (DU) ^a	29.7°	30.5	34.04 ^e				
Tropospheric $O_3(Tg)^a$	324.14 ^c	332.87	372 $^{\circ}$				
DMS (Tg S)	0.051	0.058	0.067 ^f				
$SO2$ (Tg S)	0.276	0.281	0.34 ^f				
H ₂ SO ₄ (Tg S)	3.8×10^{-4}	1.9×10^{-3}	4.2×10^{-4} f				
Tropospheric NOx ^{a, b}		0.116 Tg N	7.6×10^{14}				
	$N.A.$ ^c	$(8.24\times10^{14}$ molecules cm ⁻²)	molecules cm ^{-2 g}				
NOy (Tg N) ^b	$N.A.$ ^c	3.26	$N.A.$ ^c				
NH3(Tg N)	0.074	0.059	0.064 ^f				
VOCs (Tg C) ^b	$N.A.$ ^c	7.63	$N.A.$ ^c				
Tropospheric HCHO (Tg C) ^a	$N.A.$ ^c	0.391	$0.\overline{335-0.349}$ ^d				
$\overline{SO_4^{2-}}$ (Tg S)	0.36	0.39	$0.84 \frac{e}{0.47 \frac{f}{1}} 0.66 \frac{h}{h}$				
$NO3- (Tg N)$	$N.A.^c$	0.11	$0.01 - 0.14i$				
$\overline{\text{NH}_4}^+$ (Tg N)	0.20	0.21	0.24 ^f , $(0.27-0.44)$ ⁱ				
$\overline{Na}^+(Tg)$	2.93	3.04	$2.98e$, $(0.38-5.19)i$				
$Cl^{r}(Tg)$	4.52	4.47	$4.60^{\circ}, (0.59-8.02)^{\circ}$				
BC(Tg)	0.091	0.076	$0.28^{\circ}, 0.093^{\circ}$				
\overline{OC} (Tg)	0.45	0.61	1.28 ^e				
POM (Tg)	0.63	0.48	$0.68^{\text{ f}}$, 1.70 ^h				
SOA(Tg)	N.A. ^c	0.38	1.15^{f} , 0.59 ^j				
Dust (Tg)	25.78	26.43	$24.7 f$, $(7.9-35.9) i$				

Table 7. Global Burdens of Major Gaseous and Aerosol Species from the 5-yr (2001-2005) Simulations

^a CESM/CAM5 simulations use 30 model layers, with atmospheric pressures from \sim 1000 mb (layer 30) to \sim 3 mb (layer 1). Troposphere refers to model layers below tropopause height.

 $b_{\rm NOx} = NO + NO_2$; NO_y = NO_x + nitrogen trioxide (NO₃) + dinitrogen pentoxide (N₂O₅) + nitrous acid (HONO) + nitric acid (HNO₃) + pernitric acid (HNO₄) + peroxyacyl nitrate (PAN) + \geq C₃ peroxyacyl nitrate (PANX) + other organic nitrate (NTR); VOCs-volatile organic compounds including acetaldehyde (ALD2), carboxylic acid(AACD), long-chain alkanes (ALKH), Cresol and higher phenols (CRES), ethene (ETH), ethane (ETHA), ethanol (ETOH), formaldehyde (FORM), internal olefinic carbon bond (IOLE), methanol (MEOH), olefinic carbon bond (OLE), paraffin carbon bond (PAR), polycyclic aromatic hydrocarbons (PAH), toluene (TOL), xylene (XYL), isoprene (ISOP), and terpene (TERP).

 $\rm c$ N.A – not available, it refers to the species that are not treated in MAM_SIM_5Y or species having no burden data from previous studies. Tropospheric O_3 burden in MAM_SIM_5Y is from climatology. N.A. in SOA is due to no SOAG emission for MAM_SIM_5Y.

^d Williams et al. (2009)

e Horowitz et al. (2006)

 f Liu et al. (2012)

^g Lamarque et al. (2006)

 h Textor et al. (2006)

ⁱ Tsigaridis et al. (2006)

^j Heald et al. (2008)

Figure captions

Figure 1a. Absolute differences of H_2O_2 , SO_2 , SO_4^2 , and SOA between MAM_CB05_GE and MAM_SIM for 2001.

Figure 1b. Absolute differences between the mixing ratios of surface OH, HO_2 , NO_3 , and O_3 predicted from MAM_CB05_GE and climatology values used in MAM_SIM for 2001.

Figure 2. Spatial distributions of CO , O_3 , NO_2 , HNO_3 , HCl , and isoprene (ISOP) at surface simulated by MAM_CB05_GE for 2001.

Figure 3. Spatial distributions of total ammonium, total sulfate, total nitrate, total chloride, $PM_{2.5}$, NH₃, SO₂, H₂SO₄, HNO₃, and HCl at surface between MAM_CON and MAM_CB05_GE for summer (June, July, and August (JJA)), 2001.

Figure 4. Vertical distributions of new particle formation rate (J) (row 1) and aerosol number (PM_{num}) (row 3) simulated by MAM_CON/IMN for 2001. The overlay plots in row 2 show the distributions of simulated and observed J in bottom 1000-m in the atmosphere. Circles on overlay plots represent observations for J. Different colors of circles represent different values of J, using the same color scale as simulated J.

Figure 5. Absolute differences of PM_{2.5}, AOD, column CCN5, CF, COT, and SWCF between MAM_CON/IMN and MAM_CON for 2001.

Figure 6. Absolute differences of major PM species and their gas precursors between MAM_CON/ISO and MAM_CON for summer, 2001.

Figure 7. Absolute differences of major aerosol species and their gas precursors, new particle formation rate (J), and aerosol number between MAM_NEW_5YA and MAM_SIM_5Y for 2001-2005.

Figure 8. Absolute differences of major cloud and radiative variables between MAM_NEW_5YA and MAM_SIM_5Y for 2001-2005.

Figure 1a. Absolute differences of H_2O_2 , S O_2 , S O_4^2 -, and SOA A between MA M $\frac{\text{CB05}}{\text{GE}}$ and MAM SIM for 2001.

Figure 2. Surface distribution of CO , O_3 , NO_2 , Figure 2. Surface distribution of CO, O_3 , NO₂, HNO₃, HCl, and isoprene (ISOP) in MAM_{_}CB05_{_}GE for 2001.

Figure 3. Surface distribution of total ammonium, total sulfate, total nitrate, total chloride, PM_{2.5}, NH₃, SO₂, H₂SO₄, HNO₃, and HCl between MAM_CON and MAM_CB05_GE for summer $(June, July, and August (JJA)), 2001.$

Figure 4. Vertical distribution of new particle formation rate (J) and aerosol number (PM_{num}) Figure 4. Vertical distribution of new particle formation rate (J) and aerosol number (PM_{num})
simulated by MAM_CON/IMN for 2001. The overlay plots show the distribution of J in bottom 1000-m. Circles on overlay plots represent observations for J. Different colors of circles represent different values of J, using the same color scale as simulated J.

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