Improvement and Further Development in CESM/CAM5: Gas-Phase Chemistry and Inorganic Aerosol Treatments
 Jian He¹ and Yang Zhang^{1,*}
 ¹Air Quality Forecasting Laboratory, Department of Marine, Earth, and Atmospheric Sciences, North Carolina State University, Raleigh, NC, 27695, USA

8 Abstract: Gas-phase chemistry and subsequent gas-to-particle conversion processes such as new 9 particle formation (J), condensation, and thermodynamic partitioning have large impacts on air 10 quality, climate, and public health through influencing the amounts and distributions of gaseous 11 precursors and secondary aerosols. Their roles in global air quality and climate are examined in 12 this work using the Community Earth System Model version 1.0.5 (CESM1.0.5) with the Community Atmosphere Model version 5.1 (CAM5.1) (referred to as CESM1.0.5/CAM5.1). 13 14 CAM5.1 includes a simple chemistry that is coupled with a 7-mode prognostic Modal Aerosol 15 Model (MAM7). MAM7 includes classical homogenous nucleation (binary and ternary) and 16 activation nucleation (empirical first-order power law) parameterizations, and a highly-simplified inorganic aerosol thermodynamics treatment that only simulates particulate-phase sulfate (SO_4^{2-}) 17 18 and ammonium (NH_4^+) . In this work, a new gas-phase chemistry mechanism based on the 2005 19 Carbon Bond Mechanism for Global Extension (CB05 GE) and several advanced inorganic 20 aerosol treatments for condensation of volatile species, ion-mediated nucleation (IMN), and 21 explicit inorganic aerosol thermodynamics have been incorporated into CESM/CAM5.1-MAM7. 22 Comparing to the simple gas-phase chemistry, CB05 GE can predict many more gaseous 23 species, and thus could improve model performance for PM_{2.5}, PM₁₀, PM components, and some 24 PM gaseous precursors such as SO₂ and NH₃ in several regions, as well as aerosol optical depth (AOD) and cloud properties (e.g., cloud fraction (CF), cloud droplet number concentration 25

^{*} Email: <u>vang_zhang@ncsu.edu</u>, Phone: (919)-515-9688

26 (CDNC), and shortwave cloud forcing (SWCF)) on globe. The modified condensation and 27 aqueous-phase chemistry could further improve the predictions of additional variables such as 28 HNO₃, NO₂, and O₃ in some regions, and new particle formation rate (J) and AOD over globe. 29 IMN can improve the predictions of secondary PM_{2.5} components, PM_{2.5}, and PM₁₀ over Europe, 30 as well as AOD and CDNC over globe. The explicit inorganic aerosol thermodynamics using 31 ISORROPIA II improves the predictions of all major PM2.5 components and their gaseous 32 precursors in some regions, as well as downwelling shortwave radiation, SWCF, and cloud 33 condensation nuclei at a supersaturation of 0.5% over globe. For simulations of 2001-2005 with 34 all the modified and new treatments, the improved model predicts that on a global average, SWCF increases by 2.7 W m⁻², reducing NMBs of SWCF from -5.4% to 1.2%. Uncertainties in 35 36 emissions can explain largely the inaccurate predictions of precursor gases (e.g., SO₂, NH₃, and 37 NO) and primary aerosols (e.g., black carbon (BC) and primary organic matter (POM)). 38 Additional factors leading to discrepancies between model predictions and observations include 39 uncertainties in model treatments such as dust emissions, secondary organic aerosol (SOA) 40 formation, multiple-phase chemistry, cloud microphysics, aerosol-cloud interaction, dry and wet 41 deposition, and model parameters (e.g., accommodation coefficients and prefactors of the 42 nucleation power law), as well as uncertainties in model configuration such as the use of a coarse 43 grid resolution.

44

Keywords: CESM/CAM5.1, CB05_GE, New particle formation, Aerosol thermodynamics, Ionmediated nucleation, ISORROPIA II, Earth system modeling

```
48 1. Introduction
```

49 Atmospheric gases and aerosols play important roles in climate change due to their 50 ability to directly or indirectly alter the Earth's radiation balance. Atmospheric chemistry 51 determines the distribution of important oxidants and gaseous precursors for secondary air 52 pollutants such as ozone (O_3) and fine particular matter $(PM_{2,5})$. Meanwhile, climate change can 53 strongly influence atmospheric chemistry and air quality. Therefore, gas-phase chemistry is an 54 important component for atmospheric and Earth system models. Different chemical reactions and 55 kinetic parameters can lead to differences in the predictions of gases, secondary aerosols, new 56 particle formation rate (J), as well as climatic variables such as cloud condensation nuclei 57 (CCN), cloud droplet number concentration (CDNC), and radiative forcing (Faraji et al., 2008; 58 Luecken et al., 2008; Sarwar et al., 2008; Kim et al., 2011a; Zhang et al., 2012a; Lamarque et al., 59 2013; Young et al., 2013; Shindell et al., 2013).

60 Aerosol can influence the Earth's radiative balance by directly scattering and absorbing 61 radiation and indirectly affecting cloud properties through acting as CCN and ice nuclei (IN). 62 Therefore, it is important to accurately simulate aerosol size distribution, chemical composition, 63 and properties, which can determine the magnitude of aerosol radiative forcing (Koloutsou-64 Vakakis et al., 1998). Aerosol and its influence on climate have been included in many global 65 climate models (GCMs) such as the Community Climate System Model (CCSM) (Collins et al., 2006; Gent et al., 2010), the 5th generation of global climate model modified from European 66 67 Centre for Medium-Range Weather Forecasts in Hamburg (ECHAM5) (Roeckner et al., 2003, 68 2006a; Stier et al., 2005), and Earth system models such as the Community Earth System Model 69 (CESM) (Ghan et al., 2012; Liu et al., 2012), the Integrated Global System Model (IGSM) 70 (Dutkiewicz et al., 2005; Sokolov et al., 2005; Monier et al., 2013), and the Earth System Model 71 (ESM) (Dunne et al., 2012, 2013). However, due to the complexity of aerosol microphysical

processes and their interactions with cloud processes, it remains a challenge to accurately
represent those properties and processes in GCMs.

Inorganic aerosols comprise 25-50% of fine aerosol mass (Heintzenberg, 1989), which 74 mainly includes sulfate (SO_4^{2-}) , ammonium (NH_4^+) , nitrate (NO_3^-) , chloride (CI^-) , and sodium 75 76 (Na^{+}) . The physical and chemical properties of these aerosol constituents have been understood 77 reasonably well, making it possible to simulate aerosol physical and chemical processes in 78 GCMs. Major gas-to-particle conversion processes of inorganic aerosols include condensation, 79 nucleation, and thermodynamics. An important factor that determines the condensation of gases 80 is the mass accommodation coefficient (α), which can be measured through laboratory 81 experiments. The measured α values, however, are subject to large uncertainty and may vary in 82 several orders of magnitudes under different laboratory conditions. To simulate aerosol condensational growth, a constant value of α is therefore often assumed in GCMs, which is a 83 84 source of uncertainty in model predictions.

85 Homogeneous nucleation of H₂SO₄ vapor produces new particles that can grow to form 86 CCN. Different nucleation parameterizations are used in GCMs or global aerosol models. For 87 example, Kulmala et al. (2006), Sihto et al. (2006), and Kuang et al. (2008) derived empirical power laws with the first- or second-order dependencies of new particle formation rates (J) on 88 89 H₂SO₄ vapor concentration from observations based on cluster-activation or barrierless kinetic 90 mechanisms, which have been used in the Community Atmosphere Model (CAM) (Wang and 91 Penner, 2009), the Global-through-Urban Weather Research and Forecasting model with 92 Chemistry (GU-WRF/Chem) (Zhang et al., 2012b), and Global Model of Aerosol Processes 93 (GLOMAP) (Spracklen et al., 2006). An ion-mediated nucleation (IMN) model was developed 94 to calculate J based on ambient atmospheric conditions, H₂SO₄ vapor concentrations, ionization

95 rate, and surface area of preexisting particles. It has been used in GEOS-Chem (Yu et al., 2008; 96 Yu et al., 2010), CAM (Yu et al., 2012), and GU WRF/Chem (Zhang et al., 2012). Different 97 nucleation parameterizations lead to significant differences in J predictions by regional and 98 global models (Zhang et al., 2010) and CCN/CDNC (Yu and Luo, 2009; Pierce and Adams, 99 2009; Kuang et al., 2009; Zhang et al., 2012; Yu et al., 2012). Limited observations make it 100 difficult to validate predicted J values and appropriateness of various parameterizations. 101 A number of thermodynamic aerosol modules have been developed to understand 102 physical and chemical properties of inorganic aerosols. For example, EOUISOLV II (Jacobson, 103 1999) has been used in a one-way nested (from global to local scales) gas, aerosol, transport, 104 radiation, general circulation, mesoscale, and ocean model (GATOR-GCMOM) (Jacobson, 105 2010). EQUISOLV II uses analytical equilibrium iteration and mass flux iteration to solve 106 equilibrium problems (Jacobson, 1999), which requires relatively large computational cost. 107 SCAPE2 is used in the California Institute of Technology (CIT) model (Meng et al., 1998). 108 ISORROPIA (Nenes et al., 1998) has been used in several global models such as GEOS-Chem 109 (Bey et al., 2001), the GISS Caltech (Liao et al., 2003), and the GU-WRF/Chem (Zhang et al., 110 2012) and regional models such as the Community Multiscale Air Quality model (CMAQ) 111 (Byun and Schere, 2006) and the Comprehensive Air Quality Model with Extensions (CAMx) 112 (ENVIRON, 2010). An updated version, ISORROPIA II (Fountoukis and Nenes, 2007), has also 113 been implemented in recent versions of CMAQ (e.g., CMAQ v4.7-Dust (Wang et al., 2012) and 114 CMAQ v5.0 (Appel et al., 2013)), GEOS-Chem (Fountoukis and Nenes, 2007), and ECHAM5 115 with MESSy Atmospheric Chemistry and Global Modal-aerosol eXtension (EMAC/GMXe) 116 (Metzger et al., 2011). The Multicomponent Equilibrium Solver for Aerosols (MESA) (Zaveri et 117 al., 2005) has been used in the mesoscale WRF/Chem (Fast et al., 2006). The Equilibrium

118 Simplified Aerosol Model (EOSAM) has been updated in EMAC/GMXe in the past decade 119 (Metzger et al., 2002, 2007, and 2011). These modules assume that particles simulated in a given 120 particle size range have the same composition (i.e., internal mixture) and most of the modules 121 assume that thermodynamic equilibrium exists between the gas and particulate phases for 122 volatile compounds. Different aerosol thermodynamic models can lead to different aerosol 123 predictions (Nenes et al., 1998; Zhang et al., 2000; Zaveri et al., 2005; Metzger et al., 2011). 124 Zhang et al (2000) reported average absolute differences of 7.7% - 12.3% in total PM predictions 125 between different thermodynamic modules under 400 test conditions but the differences could be 126 as large as 68% under some cases (e.g., high nitrate/chloride concentrations and low/medium 127 relative humidity (RH)). Fountoukis and Nenes (2007) found the largest discrepancies between 128 ISORROPIA II and SCAPE2 in water concentration predictions exist under low RH conditions 129 (RH < 60%), primarily from differences in the treatment of water uptake and solid state 130 composition. The 3-D atmospheric models with these modules include explicit thermodynamic 131 treatments for sulfate, ammonium, nitrate, sodium, and chloride. For comparison, some GCMs, 132 such as CAM, use highly-simplified thermodynamics that treats sulfate and ammonium only. 133 Most thermodynamic modules assume thermodynamic equilibrium between the gas and 134 particulate phases for volatile compounds. However, if the time needed for the system achieving 135 chemical equilibrium is much longer than the time step used in the model, the equilibrium 136 assumption is not valid, which often occurs for coarse particles and cooler conditions (Wexler 137 and Seinfeld, 1991; Meng and Seinfeld, 1996). Therefore, it remains a challenge to simulate 138 thermodynamics for coarse particles.

In this work, a comprehensive gas-phase chemical mechanism and detailed inorganic
aerosol treatments for nucleation and aerosol thermodynamics are incorporated into CAM

141	version 5.1 (CAM5.1) in the CESM version 1.0.5 (CESM1.0.5). Several modifications are also
142	made to the existing treatments such as condensation and aqueous-phase chemistry. The
143	objectives are to improve the representations of gas-phase chemistry and inorganic aerosol
144	treatments in CESM/CAM5.1, and reduce uncertainties in the chemical and radiative predictions
145	associated with those processes. The improved model with enhanced capabilities can be applied
146	for decadal simulations to study interactions among atmospheric chemistry, aerosols, and climate
147	change.
148	
149	2. Model Development and Improvement
150	CESM is a fully-coupled global Earth system model, which includes land, ocean,
151	atmosphere, and sea ice components. The atmosphere component used in this study is CAM5.1.
152	Existing and new model treatments related to this study are described in this section. Further
153	details on CAM5.1 can be found at http://www.cesm.ucar.edu/models/cesm1.0/cam/.
154	
155	2.1 Existing Gas-Phase Chemistry and Aerosol Treatments in CESM/CAM5.1
156	CAM5.1 uses a simple gas-phase chemistry for sulfur species, which includes 1
157	photolytic reaction and 7 kinetic reactions among 6 gas-phase species (i.e., hydrogen peroxide
158	(H ₂ O ₂), sulfuric acid (H ₂ SO ₄), sulfur dioxide (SO ₂), dimethylsulfide (DMS), ammonia (NH ₃),
159	and semi-volatile organic gas (SOAG)). A more comprehensive gas-phase mechanism with 40
160	photolytic reactions and 172 kinetic reactions among 103 species, i.e., the Model of OZone and
161	Related chemical Tracers version 4 (MOZART-4) of Emmons et al. (2010), has been
162	incorporated into the official released CAM5.1. It is, however, only coupled with the bulk
163	aerosol module (BAM) in CAM5.1. In addition to BAM, CAM5.1 contains the modal aerosol

164 model (MAM) that is based on modal representations of aerosols. In this study, MAM is used 165 because it can represent more accurate size distributions as compared to BAM. There are two 166 versions of MAM, one with seven lognormal modes (MAM7), and the other with three 167 lognormal modes (MAM3) (Liu et al., 2012), and both are coupled with the simple gas-phase 168 chemistry in the default model. MAM7 is used in this study because it contains explicit 169 treatments for ammonium and size distributions for dust, sea-salt, and primary carbon compared 170 to MAM3. MAM7 explicitly treats sulfate, ammonium, sea-salt, dust, BC, POM, and SOA. It 171 simulates condensational growth of aerosol, nucleation, coagulation, dry deposition, wet 172 removal, and water uptake. Condensation is simulated based on a kinetic approach, in which 173 MAM7 treats H₂SO₄, NH₃, and methanesulfonic acids (MSA) as completely non-volatile species 174 and treats SOAG as a volatile species, using a constant accommodation coefficient of 0.65 for all 175 these condensing species based on Adams and Seinfeld (2002). NH₃ condensation stops when the NH_4^+/SO_4^{2-} molar ratio of a particle mode reaches 2 (i.e., fully neutralized by SO_4^{2-} ions). The 176 177 net uptake rate, Inet, due to gas to particle mass transfer for each species to each mode is 178 simulated as,

$$I_{net} = \int dx \frac{dN}{dx} I_{cond} \tag{1}$$

179

$$I_{cond} = 2 \times \pi \times D_g \times D_p \times F(Kn, \alpha)$$
(2)

$$F(Kn,\alpha) = \frac{0.75 \times (1+Kn)}{Kn \times \left(\frac{1+Kn}{\alpha} + 0.283\right) + 0.75}$$
(3)

180 where D_p is the particle diameter; *x* is the logarithmic diameter of particle, = $ln(D_p)$; dN/dx is the 181 log-normal particle number density distribution; I_{net} is the gas condensation rate; *Kn* is the 182 Knudsen number; α is the accommodation coefficient of condensable vapor; D_g is the gas 183 diffusivity, and F(*Kn*, α) is the Fuchs-Sutugin correction factor that describes the resistance to

184	uptake caused by gas-phase diffusion. Equation (1) is solved using the Gauss-Hermite quadrature
185	of order 2. Based on equation (3), as α approaches zero, F(<i>Kn</i> , α) approaches zero.
186	Consequently, I_{cond} (i.e., the uptake rate) approaches zero in equation (1).
187	There are three nucleation parameterizations in MAM7. The empirical power law of
188	Wang and Penner (2009) (WP09) is used in the planetary boundary layer (PBL), which includes
189	a first-order dependence on H_2SO_4 vapor with a prefactor of 1×10^{-6} . The binary H_2SO_4 - H_2O
190	homogeneous nucleation of Vehkamaki et al. (2002) (VE02) and ternary H_2SO_4 -NH ₃ -H ₂ O
191	homogeneous nucleation of Merikanto et al. (2007) (ME07) are used above PBL. MAM7 also
192	includes simplified inorganic aerosol thermodynamics that only involves sulfate and ammonium.
193	A more detailed description of MAM can be found in Liu et al. (2012).
194	
195	2.2 New and Modified Model Treatments Implemented in This Work
196	2.2.1 Gas-Phase Chemical Mechanism
197	Highly simplified gas-phase mechanism can result in large uncertainties in the predictions
198	of oxidants and gaseous precursors for secondary aerosols. Therefore, a new gas-phase
199	mechanism, the 2005 Carbon Bond Mechanism for Global Extension (CB05_GE)
200	(Karamchandani et al., 2012) has been implemented into CAM5.1 using the same chemical
201	preprocessor as MOZART-4 (Lamarque et al., 2012) and coupled with both MAM3 and MAM7.
202	CB05_GE was developed to simulate major chemical reactions for global-through-urban
203	applications as illustrated in Zhang et al. (2012). A more detailed description of CB05_GE can be
204	found in Karamchandani et al. (2012). In this study, gas precursors for SOA in CB05_GE are
205	mapped to SOAG to make it compatible in MAM7. As the first study of CESM/CAM5.1 with
206	CB05_GE, this work focuses on the impact of gas-phase chemistry. The heterogeneous

chemistry on the surface of aerosol is turned off. CB05_GE implemented in CESM/CAM5
contains a total of 273 reactions including 50 photolytic reactions and 223 kinetic reactions
among 93 gas-phase species in this study. The gas-phase chemical system is solved using an
implicit backward Euler method.

211

212 2.2.2 Ion-mediated nucleation parameterization

213 Ions generated by cosmic radiation and natural radioactive decay have been studied for a 214 long time as an important source to enhance nucleation (Raes et al., 1986). An IMN model is 215 developed by Yu (2010) (Yu10) for H_2SO_4 -H₂O system, and explicitly solves the dynamic 216 equations in terms of temperature, relative humidity, H₂SO₄ vapor concentration, ionization rate, 217 and surface area of preexisting particles. Different from classic binary nucleation theory, which is 218 based on the minimization of changes in Gibbs free energy (Seinfeld and Pandis, 2006), IMN is 219 based on a kinetic model that considers the interactions among ions, neutral and charged clusters, 220 vapor molecules, and preexisting particles (Yu and Turco, 2000, 2001; Yu, 2006, 2010). The 221 global ionization rates due to cosmic rays are calculated based on the schemes given in Usoskin 222 and Kovaltsov (2006) and the contribution of radioactive materials from soil to ionization rates is 223 parameterized based on the profiles given in Reiter (1992). To reduce the computing cost using 224 IMN in 3-D models, Yu et al. (2008) developed lookup tables with simple interpolation 225 subroutines to calculate nucleation rates under typical atmospheric conditions. In this work, IMN 226 based on YU10 is implemented into MAM7 and combined with default nucleation 227 parameterizations (VE02, ME07, and WP09) in order to improve the J predictions and aerosol 228 number concentrations in upper troposphere. The J value above PBL is taken as the maximum 229 value among predictions from IMN (YU10) and homogeneous nucleation (VE02 or ME07), and

the J value within PBL is taken as the maximum value among predictions from IMN (YU10),

homogeneous nucleation (VE02 or ME07), and the first-order parameterization (WP09).

232

233 2.2.3 Inorganic Aerosol Thermodynamics

234 Gas-particle partitioning is an important process in the formation and evolution of 235 secondary aerosols. Several factors affect gas-particle partitioning, such as temperature, relative 236 humidity, saturation vapor pressures of species, the physical state of the condensed-phase, and 237 the way in which aerosol components interact each other (Cappa et al., 2008; Zuend et al., 2010). 238 Most models focus on inorganic aerosols. Fountoukis and Nenes (2007) developed a 239 computationally-efficient thermodynamics equilibrium model, ISORROPIA II, for the magnesium (Mg²⁺) - potassium (K⁺) - calcium (Ca²⁺) - NH₄⁺ - Na⁺ - SO₄²⁻ - NO₃⁻ - Cl⁻ - H₂O 240 241 aerosol system. An important difference between ISORROPIA II and most other 242 thermodynamics equilibrium models is that ISORROPIA II simulates crustal species, such as Mg^{2+} , K^+ , and Ca^{2+} , which are important constituents of atmospheric aerosols, in particular, 243 mineral dust. Therefore, to explicitly simulate aerosol thermodynamics, ISORROPIA II has been 244 245 implemented into MAM7 and applied for accumulation, Aitken, fine sea-salt, and fine dust modes to explicitly simulate thermodynamics of $SO_4^{2^-}$, NH_4^+ , NO_3^- , CI^- , and Na^+ as well as the 246 247 impact of crustal species associated with fine dust modes on aerosol thermodynamics. The concentrations of K⁺, Ca²⁺, and Mg²⁺ as the input for ISORROPIA II are calculated from dust 248 concentrations, using the mass ratios of 1.022×10^{-3} , 1.701×10^{-3} , and 7.084×10^{-4} , respectively 249 250 (Van Pelt and Zobeck, 2007). Aerosol thermodynamics involving coarse particles (in coarse sea-251 salt and coarse dust modes) is currently not treated in this work, given the high computational 252 cost (by at least a factor of 3) for solving the non-equilibrium system involving coarse particles.

253

254 2.2.4 Modifications of Existing Aerosol Treatments

255	MAM 7 does not treat NO_3^- and it treats NaCl as one species. In this work, MAM7 is
256	modified to explicitly simulate NO_3^- , Cl^- , and Na^+ using a similar method to the condensation of
257	H_2SO_4 and NH_3 . NO_3^- and Cl^- are simulated in all modes except for primary carbon mode. Na^+ is
258	simulated in sea-salt modes. The source of Na^+ is calculated based on the mass ratio of Na and Cl
259	from sea-salt emissions. The source of Cl ⁻ includes sea-salt emissions, and the condensation of
260	HCl resulted from HCl emissions and gas-particle partitioning of total chloride.
261	Species-dependent accommodation coefficients are used for H ₂ SO ₄ , NH ₃ , HNO ₃ , and
262	HCl, with the values of 0.02, 0.097, 0.0024, and 0.005 (Zhang et al., 1998; Sander et al., 2002),
263	respectively. Since by default the model treats the condensation of inorganic volatile gas species
264	as irreversible process (no evaporation) (see equation (1)), the lower limit values of mass
265	accommodation coefficients are used for these species to represent their net fluxes from the gas-
266	phase to the liquid/solid phases. Such lower limit values correspond to uptake coefficients, which
267	represent the net fluxes and are smaller than mass accommodation coefficients. To ensure
268	electroneutrality in each mode after kinetically condensing H ₂ SO ₄ , NH ₃ , HNO ₃ , and HCl at
269	different condensation (or uptake) rates, the condensation of NH3 will stop when the mole
270	concentration of cations (i.e., NH_4^+) is equal to sum of those of anions (i.e., $[NH_4^+] = 2 \times [SO_4^{2-1}]^2$
271]+[NO ₃ ⁻]+[Cl ⁻]).
272	The original MAM7 treats $NH_3(g)/NH_4^+$ in cloud water. In this work, the dissolution and
273	dissociation of HNO ₃ and HCl to produce NO_3^- and Cl^- in cloud water are added in the model

based on Schwartz (1984), Marsh and McElroy (1985), and Seinfeld and Pandis (2006), i.e.,

$$HNO_{3(g)} \rightleftharpoons HNO_{3}(aq) \rightleftharpoons H^{+} + NO_{3}^{-}$$

 $HCl_{(g)} \rightleftharpoons HCl_{(aq)} \rightleftharpoons H^{+} + Cl^{-}$

276

The concentration of H^+ (thus the pH value of the solution) is obtained by solving the electroneutrality equation using the bisection method. The aqueous-phase chemical system is solved analytically.

280

3. Model Configurations and Evaluation Protocols

282 3.1 Model Setup and Simulation Design

283 Table 1 summarizes the CESM/CAM5.1 simulations that are designed to examine the 284 impacts of individual new and modified treatments on model predictions. The first set of 285 simulations includes two simulations with the same default MAM7 coupled with different gas-286 phase mechanisms: one uses the simple gas-phase chemistry (MAM SIM) with a total of 37 287 prognostic species and one uses the CB05 GE (MAM CB05 GE) with a total of 127 prognostic 288 species. A comparison of the two simulations provides an estimate of the impacts of gas-phase 289 chemical mechanisms. The second set of simulations consists of five simulations that use the 290 same CB05 GE gas-phase mechanism but with modified and new aerosol treatments 291 individually and jointly. The first one is MAM CON that uses an explicit treatment for NO₃, Cl., and Na⁺ and species-dependent mass accommodation coefficients for condensation and that 292 293 includes the aqueous-phase chemistry of HNO₃/NO₃⁻ and HCl/Cl⁻. This simulation includes a 294 total of 139 prognostic species. The second one is MAM CON/IMN that uses the same 295 treatments as MAM CON but with IMN as one of the nucleation mechanisms and a prefactor of 1.0×10^{-8} in WP09. The third one is MAM CON/ISO that uses the same treatments as 296 297 MAM CON but with ISORROPIA II for aerosol thermodynamics assuming metastate

298 equilibrium (i.e., liquid only). The fourth one is MAM NEWA that uses the same treatments as 299 MAM CON but with all new and modified aerosol treatments and a prefactor of 1.0×10^{-9} for 300 WP09. The fifth one is MAM NEWB that uses the same treatments as MAM NEWA, but with 301 ISORROPIA II assuming a stable condition (i.e., solid and liquid coexist). A comparison of 302 MAM CB05 GE with MAM CON indicates the impact of modified condensation and aqueous-303 phase chemistry. A comparison of MAM CON/IMN, MAM CON/ISO, and MAM NEWA with 304 MAM CON indicates the impacts of IMN, ISORROPIA II, and combined new and modified 305 aerosol treatments, respectively. Comparison of MAM NEWB with MAM NEWA indicates the impacts of thermodynamic conditions on gas-aerosol partitioning. The 3rd set of simulation 306 307 includes one simulation using the same configuration as MAM NEWA but with adjusted 308 emissions (MAM NEW/EMIS). Its comparison with MAM NEWA indicates the impacts of uncertainties in emissions on model predictions. The 4th set of simulation includes one simulation 309 310 using the same configuration as MAM SIM but with prescribed SST for a 5-yr period during 311 2001-2005 (MAM SIM 5Y), and two simulations both using the same configuration as 312 MAM NEW/EMIS for 2001-2005, but one with prescribed SST (MAM NEW 5YA), and the 313 other in a fully-coupled mode (MAM NEW 5YB).

All these simulations use the same approach for photolytic rates calculations based on Lamarque et al. (2012), the same aqueous-phase chemistry of Barth et al. (2000), and the same physical options as those in MAM_SIM. Major physical options include the cloud microphysics parameterization of Morrison and Gettelman (2008), the moisture PBL scheme of Bretherton and Park (2009), the shallow convection scheme and deep convection scheme of Park and Bretherton (2009) and Zhang and McFarlane (1995), respectively, the aerosol activation parameterization of Abdul-Razzak and Ghan (2000), and the Rapid Radiative Transfer Model for GCMs (RRTMG) of Mlawer et al. (1997) and Iacono et al. (2003, 2008) for long and short-wave radiation. The
land surface processes are simulated by the Community Land Model (CLM) of Lawrence et al.
(2011) in CESM that is coupled with CAM5.1.

324 All simulations except for MAM SIM 5Y and MAM NEW 5YA are performed with 325 fully-coupled CESM1.0.5 with standard B 1850-2000 CAM5 CN configuration, which 326 represents 1850 to 2000 transient conditions and includes all active components in CESM with 327 biogeochemistry in the land model. MAM SIM 5Y and MAM NEW 5YA are performed with 328 standard F AMIP CAM5 configuration, which uses a climatological dataset for SST provided 329 by NCAR for ocean model. The simulations are conducted for the full-year of 2001 and 2001-2005 at a horizontal resolution of $0.9^{\circ} \times 1.25^{\circ}$ and a vertical resolution of 30 layers for CAM5.1. 330 331 The initial conditions for ice and ocean models are from CESM default settings. The initial 332 conditions for the land model are based on the output from the NCAR's CESM/CAM4 B 1850-333 2000 CN simulation. The initial conditions for CAM5 are derived from a 10-yr (1990-2000) 334 CAM5 standalone simulation with the MOZART chemistry provided by NCAR. A 1-year 335 (January 1-December 31, 2000) CESM/CAM5 simulation using NCAR's CESM B 1850-336 2000 CAM5 CN component set is performed as spinup to provide the initial conditions for 337 meteorological variables and chemical species that are treated in both MOZART and CB05 GE. 338 An additional 3-month (October 1-December 31, 2000) CESM/CAM5 simulation based on a 10-339 month (January-October, 2000) CESM/CAM5 output using initial conditions from NCAR's 340 CESM B 1850-2000 CAM5 CN is performed as spinup to provide initial conditions for 341 chemical species that are treated in CB05 GE but not in MOZART. All production simulations 342 of 2001 are from January 1 - December 31, 2001 and those of 2001-2005 are from January 1, 343 2001- December 31, 2005. The offline anthropogenic emissions used in all simulations except

344	for MAM_NEW/EMIS are taken from Zhang et al. (2012) (see Table 2 of Zhang et al. (2012) for
345	the sources of those anthropogenic emissions). Anthropogenic emissions used in
346	MAM_NEW/EMIS are adjusted emissions based on those of Zhang et al. (2012), with
347	adjustment factors of 0.7, 0.5, and 1.2 for SO ₂ over CONUS, Europe, and Asia, respectively, and
348	1.2 for NH ₃ , BC, and organic carbon (OC), and 1.3 for carbon monoxide (CO) over all three
349	regions. Those emissions are adjusted based on the comparison with the emission inventories
350	from the Representative Concentration Pathways (RCPs), the MOZART version 4 (MOZART-
351	4), the Reanalysis of the TROpospheric chemical composition (RETRO), the Global Fire
352	Emissions Database (GFED) version 2, and preliminary evaluation of CESM/CAM5.1 with
353	modified and new gas and aerosol treatments using available observations. The online emissions
354	include biogenic volatile organic carbon (Guenther et al., 2006), mineral dust (Zender et al.,
355	2003), and sea-salt (Martensson et al., 2003).
356	
357	3.2 Available Measurements for Model Validation
358	A number of observational datasets from surface networks and satellites are used for
359	model evaluation. They are summarized along with the variables to be evaluated in Table 2.
360	Global surface networks include the Baseline Surface Radiation Network (BSRN) and the
361	National Oceanic and Atmospheric Administration Climate Diagnostics Center (NOAA/CDC).
362	The satellite datasets include the Moderate Resolution Imaging Spectroradiometer (MODIS), the
363	Clouds and Earth's Radiant Energy System (CERES), the Total Ozone Mapping
364	Spectrometer/the Solar Backscatter UltraViolet (TOMS/SBUV), the Measurements Of Pollution
365	In The Troposphere (MOPITT), and the Global Ozone Monitoring Experiment (GOME). Other
366	satellite-based data include the MODIS-derived CDNC from Bennartz (2007) (BE07).

367 Regional observational networks include the Clean Air Status and Trends Network 368 (CASTNET), the Interagency Monitoring of Protected Visual Environments (IMPROVE), and 369 the Speciation Trends Network (STN) over CONUS: the European Monitoring and Evaluation 370 Program (EMEP), the Base de Données sur la Qualité de l'Air (BDQA), and the European air 371 quality database (AirBase) over Europe; the Ministry of Environmental Protection of China 372 (MEP of China), the National Institute for Environmental Studies of Japan (NIES of Japan), and 373 Taiwan Air Quality Monitoring Network (TAQMN) over East Asia. The observational data for 374 particle formation rate J is compiled from Kulmala et al. (2004) and Yu et al. (2008), which 375 include land-, ship-, and aircraft-based measurements. 376 377 3.3 Evaluation Protocol The protocols for performance evaluation include spatial distributions and statistics. 378 379 following the approach of Zhang et al. (2012). The analysis of the performance statistics will 380 focus on mean bias (MB), normalized mean bias (NMB), normalized mean error (NME), and 381 root mean square error (RMSE). The radiative variables are evaluated annually, including

downwelling shortwave radiation (SWD) and downwelling longwave radiation (LWD) from

383 BSRN; outgoing longwave radiation (OLR) from NOAA/CDC; shortwave cloud forcing

382

384 (SWCF) from CERES; cloud fraction (CF), aerosol optical depth (AOD), cloud optical thickness

385 (COT), cloud water path (CWP), precipitating water vapor (PWV), and CCN from MODIS; as

386 well as CDNC from BE07. Chemical concentrations evaluated include seasonal and annual

387 averaged concentrations of CO, O₃, SO₂, NH₃, NO₂, HNO₃, PM, and its major components (i.e.,

SO₄²⁻, NO₃⁻, and NH₄⁺, BC, OC, total carbon (TC) for CONUS and Europe). The chemical 388

389 observations over East Asia are very limited, and they only include surface concentrations of CO, SO₂, NO₂, O₃, and PM₁₀. Column concentrations of tropospheric CO and NO₂, and tropospheric
O₃ residual (TOR) are evaluated for globe.

All observational data used for evaluating 2001 simulations are based on 2001 only except for particle formation rates (J) that are based on different years compiled from Kulmala et al. (2004) and Yu et al. (2008). All observational data used for evaluating 2001-2005 simulations are based on 2001-2005.

396

397 4. Model Evaluation for MAM_SIM Based on Original Model Treatments

398 Tables 3 and 4 show MBs and NMBs of radiative/cloud and chemical predictions,

399 respectively. The model performance of the baseline simulation, MAM_SIM, is discussed below,

400 and that for all other simulations will be discussed in section 5.

401 As shown in Table 3, radiative variables such as LWD and SWD are underpredicted by

402 3.4 W m⁻² (~ -1.1%) and 2.0 W m⁻² (~ -1.1%), respectively, whereas OLR and SWCF are

403 overpredicted by 8.8 W m⁻² (~ 4.1%) and 3.2 W m⁻² (~ 7.9%) respectively. Cloud variables such

404 as CF and PWV are slightly underpredicted, whereas COT, CWP, column CCN at a

405 supersaturation of 0.5% (CCN5), and CDNC are largely underpredicted, with NMBs of -77.8%

406 to -55.6%, which is likely due to the limitations in the current model treatments of cloud

407 microphysics and aerosol-cloud interactions in CAM5.1.

408 AOD is also underpredicted by 36.1%, which is likely due to inaccurate predictions of

409 aerosol concentrations and uncertainties in the assumed hygroscopicity of aerosol components in

410 the calculation of optical properties and water uptake. For example, as shown in Table 4, PM_{2.5}

411 concentrations over CONUS and Europe, and PM₁₀ concentrations over CONUS, Europe, and

412 East Asia are underpredicted, with NMBs of -67.5% to -31.8%, which is due to the inaccurate

predictions of SO₄²⁻, NH₄⁺, and organic aerosols, and missing major inorganic aerosol species 413 414 such as nitrate and chloride. The concentrations of BC, OC, and TC are underpredicted (by \sim 415 50%), which is likely due to the uncertainties in the BC and primary OC emissions as well as 416 treatments for SOA formation. In particular, the SOA treatment used in CAM5.1 is based on a 417 highly-simplified aerosol yield approach with a single lumped semi-volatile organic gas (i.e., 418 SOAG). For gaseous species, SO₂ concentrations over CONUS and Europe are significantly overpredicted by 10.3 μ g m⁻³ (~ 264.8%) and 6.6 μ g m⁻³ (~ 97.5%), respectively, whereas SO₂ 419 concentrations over East Asia are largely underpredicted by 7.9 μ g m⁻³ (by ~63.0%). NH₃ 420 421 concentrations over Europe are also largely underpredicted by 82.0%. These large biases in SO₂ 422 and NH₃ are likely due in part to the uncertainties in the emissions of SO₂ and NH₃, which in turn affect the predictions of SO_4^{2-} and NH_4^{+} . The J values in PBL are highly underpredicted by 423 424 99.6%, which is mainly due to the inaccurate calculation of H₂SO₄ vapor concentration that 425 participates in the nucleation and uncertainties in the nucleation parameterizations used in the 426 default CESM/CAM5.1.

427

428 **5. Sensitivity Simulations**

429 5.1 Impacts of New Gas-Phase Chemistry

Compared to simple gas-phase chemistry, many more gaseous species and chemical
reactions simulated in CB05_GE can affect secondary aerosol formation through gas-to-particle
mass transfer and aqueous-phase chemistry and affect climatic variables through chemistry
feedbacks to the climate system. Figure 1a shows the absolute differences of H₂O₂, SO₂, SO₄²⁻,
and SOA between MAM_CB05_GE and MAM_SIM. MAM_CB05_GE treats more gaseous
species and chemical reactions than MAM_SIM, leading to large changes in the concentrations

436 of gaseous and PM species. Compared with MAM SIM, MAM CB05 GE predicts higher H₂O₂ by 0.4 ppb, SO₂ by 7.3 ppt, SO₄²⁻ by 0.01 μ g m⁻³, and SOA by 0.06 μ g m⁻³in terms of global 437 438 mean. Those changes are mainly caused by different gas-phase chemical mechanisms used in MAM_SIM and MAM_CB05_GE. While MAM CB05 GE explicitly simulates OH, HO₂, NO₃, 439 440 and O₃, MAM SIM uses climatology data for these species. OH simulated by MAM CB05 GE is lower than that prescribed by MAM SIM by up to 2.8×10^6 molecules cm⁻³, or higher by up to 441 3.0×10^6 molecules cm⁻³ in different regions (Figure not shown), with a higher global mean by 442 443 MAM CB05 GE. MAM SIM includes the production of H₂O₂ from the self-destruction of HO₂ 444 and the loss of H_2O_2 through its photolytic reaction and its reaction with OH. Higher H_2O_2 in 445 MAM CB05 GE is mainly due to greater production of H_2O_2 from additional chemical 446 reactions (e.g., OH+OH) than loss of H_2O_2 through the reactions of OH + H_2O_2 , O + H_2O_2 , Cl + 447 H_2O_2 , and $H_g + H_2O_2$. Different predictions in H_2O_2 can in turn affect OH mixing ratios in 448 MAM CB05 GE but not in MAM SIM. In addition, the photolytic reactions of VOCs (e.g., 449 HCHO, peroxyacyl nitrates (PAN), and peroxyacetic and higher peroxycarboxylic acids (PACD)) 450 and other gases (e.g., HNO₃, HONO, HNO₄, HOCl, and HOBR) treated in MAM CB05 GE can 451 produce OH. Figure 1b shows the absolute differences between the mixing ratios of major 452 oxidants predicted from MAM CB05 GE and climatology values used in MAM SIM. The 453 global mean mixing ratios of oxidants are higher in MAM CB05 GE than climatology data in 454 MAM SIM, leading to more oxidation of VOCs and therefore more SOA in MAM CB05 GE. 455 Higher O_3 predicted from MAM CB05 GE over most of the domain is mainly due to more O_3 456 precursors (e.g., NO₂ and VOCs) treated in the model. Despite higher OH mixing ratios in 457 MAM CB05 GE, many gaseous species such as NO_x, SO₂, HNO₃, HONO, and other VOCs are oxidized by OH to form secondary inorganic and organic aerosols. Those oxidation reactions 458

compete for limited OH, leading to less oxidation of SO₂, thus higher SO₂ mixing ratios over
most land areas by MAM_CB05_GE. Lower SO₂ mixing ratios over the oceanic areas in
MAM_CB05_GE is due to the combined effects of less production of SO₂ from lower DMS
mixing ratios (due to increased OH levels) and greater SO₂ oxidation from higher OH mixing
ratios.

464 The changes in the concentrations of PM and its components are due to the change in the 465 mixing ratios of gaseous precursors. CB05 GE contains more photolytic reactions, which affect the mixing ratios of OH, SO₂, and H₂SO₄, and subsequently the concentration of SO_4^{2-} through 466 467 condensation and homogeneous nucleation. Higher SO₂ mixing ratios in MAM CB05 GE result in more H_2SO_4 thus more SO_4^{2-} . For example, both SO_2 mixing ratios and SO_4^{2-} concentrations 468 are higher over eastern China in MAM CB05 GE. More SO₄²⁻ over the oceanic areas is mainly 469 470 due to more oxidation of SO₂ by OH. Due to the simplification of aerosol thermodynamics in default MAM7, the concentrations of SO_4^{2-} can affect the concentrations of NH_4^+ directly and 471 therefore NH₃ mixing ratios and PM number concentrations (PM_{num}). For example, the increase 472 of SO_4^{2-} results in an increase in NH_4^+ and PM_{num} , and a decrease in NH_3 . The increase of SO_4^{2-} 473 474 and PM_{num} can increase AOD, CF, COT, CWP, PWV, and CDNC and therefore affect radiation 475 by increasing LWD and SWD (Figures not shown, see changes in performance statistics of these 476 affected variables in Table 3). The increase of SOA is due to the inclusion of more gaseous 477 precursor emissions (e.g., isoprene, terpene, xylene, and toluene) in MAM CB05 GE, which 478 contribute to SOAG and thus SOA through gas-to-particle conversion. 479 Figure 2 shows the spatial distributions of CO, O₃, NO₂, HNO₃, hydrochloric acid (HCl),

480 and isoprene (ISOP) that can be predicted by MAM CB05 GE but not by MAM SIM. CO

481 mixing ratio is higher in most Asia, central Africa, South Africa, and eastern U.S., which is

482	mainly due to higher CO emissions in those regions and the production of CO from the
483	photolytic reactions of VOCs (e.g., formaldehyde, acetaldehyde, and isoprene). Higher O ₃
484	mixing ratios in the northern hemisphere than southern hemisphere are mainly due to much
485	higher mixing ratios of O ₃ precursors. Higher O ₃ mixing ratios over Mediterranean Sea are
486	mainly due to the transport of O_3 and its precursors from source regions and less deposition onto
487	ocean surface. Higher O ₃ mixing ratios over Tibet are mainly due to the stratospheric influences
488	from high altitude and no titration of O_3 due to low NO mixing ratios (< 0.2 ppb) in this region.
489	Higher mixing ratios of NO ₂ over most Asia, eastern U.S, Europe, and Central Africa are mainly
490	due to higher NO_x emissions over those regions, which also result in higher HNO_3 in those
491	regions. Higher mixing ratios of HCl over Europe, India, and East Asia are mainly due to the
492	higher anthropogenic HCl emissions in those regions. In addition, MAM_CB05_GE includes
493	oceanic emissions of HCl, leading to higher HCl over ocean. Higher isoprene mixing ratios over
494	South Africa, central Africa, and Oceania are mainly due to higher isoprene emissions in those
495	regions, which also contribute to the formation of SOA in those regions.
496	The aforementioned changes in the concentrations of gaseous species and PM due to new
497	gas-phase chemistry implemented in the model and its feedbacks to radiation through the climate
498	system result in a change in predicted cloud properties and radiation balance that in turn affect
499	the predictions of all chemical species during subsequent time steps. As a consequence of
500	interwoven changes due to complex feedback mechanisms, the two simulations perform
501	differently, with noticeable improvement by MAM_CB05_GE. As shown in Table 3, compared
502	with MAM_SIM, MAM_CB05_GE reduces MB of LWD by 17.6%, OLR by 8.0%, CF by
503	28.6%, COT by 1.0%, PWV by 28.0%, AOD by 5.5%, and CDNC by 1.8%, leading to 0.3-2.2%
504	absolute reduction in their NMBs. Although MAM_CB05_GE increases MB of SWD by 26.2%,

505	the increases in their NMBs are only 1.2%. As shown in Table A1 in the supplementary material,
506	the changes in most cloud and radiative variables between MAM_SIM and MAM_CB05_GE are
507	statistically significant. As shown in Table 4, MAM_CB05_GE also reduces MBs of SO ₂ by
508	2.5% and PM_{10} by 8.1% over East Asia, NH_3 by 1.3% and SO_4^{2-} by 12.5% over Europe, OC by
509	11.1%, TC by 8.3%, and $PM_{2.5}$ by 3.3% over CONUS, leading to 0.8-6.5% absolute reductions
510	in NMBs. Despite the model improvement by CB05_GE, large biases still remain for some
511	chemical species. For example, CO over East Asia is largely underpredicted with an NMB of -
512	82.1% (see Table 4), which results from the uncertainties in the CO emissions over East Asia.
513	However, the column CO over globe is predicted very well, with an NMB of -5.7%. Large biases
514	in SO ₂ predictions over CONUS, Europe, and East Asia are mainly due to the uncertainties in the
515	SO ₂ emissions over those regions. Large biases in O ₃ over Europe are likely due to the
516	uncertainties in the O ₃ precursor emissions (e.g., NO _x) and inaccurate predictions of radiation
517	over Europe. In particular, the large underpredictions in NO ₂ concentrations (likely due to the
518	uncertainties in the NO_x emissions and overpredictions in radiation, see section 5.5 for more
519	detailed discussions) indicate insufficient NO _x for titration of O ₃ , leading to a large
520	overprediction in O ₃ concentrations in Europe. The large biases in HNO ₃ are due to no treatment
521	for gas-particle partitioning in both simulations.
522	

523 5.2 Impacts of Condensation and Aqueous-Phase Chemistry

524 The mass accommodation coefficient (α) for H₂SO₄ vapor is subject to considerable 525 uncertainty. The calculation in the default condensation module with a default α value of 0.65 526 gives a very low concentration of H₂SO₄, resulting in very low nucleation rates and aerosol 527 number concentrations. Considering that the original model treats H₂SO₄ and NH₃ condensation

528	as an irreversible process, the default α value of 0.65 for H_2SO_4 and NH_3 is reduced to 0.02 and
529	0.097, respectively, based on Zhang et al. (1998). This change in α value provides sufficient
530	H_2SO_4 and NH_3 for nucleation with a typical H_2SO_4 concentration range of $10^6 \sim 10^8$ molecules
531	cm ⁻³ . Because HNO ₃ and HCl are semi-volatile species, the lower limits of α (0.0024 and 0.005,
532	respectively) based on Sander et al. (2002) are selected for their irreversible condensation
533	process. NH_4^+ from NH_3 condensation will be constrained by the available $SO_4^{2^-}$, NO_3^- , and
534	condensed Cl ⁻ , following the approach that is used in the default simplified thermodynamics to
535	neutralize the cations in the system.
536	Figure 3 shows the absolute differences of NH ₃ , SO ₂ , HNO ₃ , HCl, H ₂ SO ₄ , total
537	particulate ammonium (TNH ₄), total particulate sulfate (TSO ₄), total particulate nitrate (TNO ₃),
538	and total particulate chloride (TCL) in all the modes except primary carbon mode, and $PM_{2.5}$
539	between MAM_CON and MAM_CB05_GE in June, July, and August (JJA), 2001. Due to the
540	inclusion of HNO ₃ and HCl condensation in MAM_CON, the concentrations of HNO ₃ and HCl
541	decrease by 0.1 ppb (~72%) and 0.097 ppb (~84%), respectively. NO ₃ ⁻ is not simulated in the
542	original model and the concentration of NO ₃ ⁻ is assumed as zero in MAM_CB05_GE. Therefore,
543	the concentration of NO ₃ ⁻ increases due to the condensation of HNO ₃ in MAM_CON. The
544	concentration of TCL in MAM_CB05_GE is calculated from the mass ratio of chloride in sea-
545	salt. Over land, TCL increases significantly due to the condensation of HCl to form Cl ⁻ . The
546	change of TCL over ocean is mainly due to the change of sea-salt emissions. The changes of SO_2
547	mixing ratios are mainly due to the differences in mixing ratios of species in sulfur chemistry in
548	the two simulations. For example, compared to MAM_CB05_GE, the increase of SO ₂ over
549	eastern U.S. in MAM_CON is likely due to the less SO ₂ oxidation in clouds (Figure not shown),
550	which results from lower CF. The decrease of SO ₂ mixing ratios over most oceanic areas is likely

551 due to the combined effects of DMS oxidation and SO₂ oxidations in MAM CON. More SO₂ can result in more H_2SO_4 and therefore more SO_4^{2-} through condensation and homogeneous 552 553 nucleation of H₂SO₄. The changes in H₂SO₄ concentrations are the results of changes of SO₂ 554 mixing ratios. The mass accommodation coefficient of H₂SO₄ is reduced significantly (by a 555 factor of 32.5), allowing more H₂SO₄ to participate in binary/ternary homogeneous nucleation and produce more secondary SO_4^{2-} , improving predictions of SO_4^{2-} over CONUS but degrading 556 the performance of SO_4^{2-} over Europe (see Table 4). Although the mass accommodation 557 558 coefficient of NH₃ is reduced significantly (by a factor of 67), more available NH₃ can participate in the ternary homogeneous nucleation and produce secondary NH_4^+ . Meanwhile, the secondary 559 NH_4^+ formed from NH₃ condensation is also constrained by available SO₄²⁻, NO₃⁻, and 560 condensed Cl⁻. As a result, the concentrations of NH₃ decrease and those of NH₄⁺ increase. Due 561 562 to more available H₂SO₄ participating in the nucleation, J has been improved significantly, 563 reducing the NMB from -99.5% to -12.8%. With an inclusion of the dissolution and dissociation 564 of HNO₃ and HCl in cloud water, more NH₃ is required to dissolve to maintain cation-anion 565 equilibrium in the cloud water, which further reduces the mixing ratios of NH₃, HNO₃, and HCl. 566 As shown in Table 4, compared with MAM CB05 GE, MAM CON gives better performance against observations in terms of CO, NO₂, O₃, HNO₃, PM_{2.5}, and PM₁₀ over Europe, 567 CO and PM₁₀ over East Asia, O₃, HNO₃, SO₄²⁻, NH₄⁺, BC, OC, TC, and PM₂₅ over CONUS, and 568 column CO, column NO₂, TOR, and J over globe. As also shown in Table 3, the improved 569 570 chemical predictions improve the predictions of OLR, SWCF, CF, COT, CWP, AOD, and CDNC. 571 As shown in Table A1, the changes in most cloud /radiative variables between MAM CB05 GE 572 and MAM CON are statistically significant, indicating the significant impacts of the modified 573 condensation and aqueous-phase chemistry treatments on radiation. Treating condensation and

574	aqueous-phase chemistry of HNO_3 and HCl enables an explicit simulation of NO_3^- and Cl^- in
575	MAM7. However, the mass concentrations of SO ₂ remain significant overpredictions, with
576	NMBs of 301.2% for CONUS, and 123.0% for Europe, mainly because of the uncertainties in
577	SO ₂ emissions over those regions. Due to the simplified irreversible treatment for gas
578	condensation, the mass concentrations of SO_4^{2-} , NH_4^+ , NO_3^- , and CI^- are overpredicted, although
579	the lower limit of mass accommodation coefficient for each precursor is used in MAM_CON. As
580	shown in Table 4, the concentrations of $SO_4^{2^-}$, NH_4^+ , NO_3^- , and Cl^- from MAM_CON are
581	overpredicted by 1.7%, 20.0%, 198.2%, and 359.9%, respectively, for CONUS, and 40.3%,
582	85.0%, 67.8%, and 102.8%, respectively, for Europe. The large NMBs of NO_3^- and Cl^- in
583	MAM_CON are due to the small observed values for NO ₃ ⁻ (i.e., 1.0 μ g m ⁻³ over CONUS and 2.0
584	$\mu g~m^{\text{-3}}$ over Europe) and Cl $(i.e.,0.1~\mu g~m^{\text{-3}}$ over CONUS and 0.7 $\mu g~m^{\text{-3}}$ over Europe), the
585	uncertainties in treating HNO ₃ and HCl as non-volatile species using their lower limits of
586	accommodation coefficients, and lack of treatments for NO3 ⁻ and Cl ⁻ thermodynamics.
587	

588 5.3 Impacts of New Particle Formation

589 Figure 4 shows the annual-mean vertical distributions of particle formation rate (J) values 590 and aerosol number concentrations, and simulated J values averaged between the ground level 591 and 1000-m overlaid with observations within the same layers. In MAM CON/IMN, IMN is 592 combined with three default nucleation parameterizations to predict J throughout the atmosphere. 593 In MAM CON, J over ocean is overpredicted by factors of 5-50, despite a seeming good NMB 594 of -12.8% in the globe mean (see Table 4). J values at several sites over land are underpredicted 595 by factors of 1-10, which compensates the large overpredictions at most sites over ocean. The 596 large underpredictions at those sites are likely due to the uncertainties in SO₂ emissions and

597 nucleation parameterizations, and the missing species that may have participated in nucleation. 598 For example, several other species may contribute to the new particle formation, including 599 methanesulfonic acid (van Dingenen and Raes, 1993), hydrochloric acid (Arstila et al., 1999), 600 organic compounds (Berndt, et al., 2013), iodine-containing compounds (Hoffmann et al., 2001; 601 O'Dowd et al., 2002; Burkholder, et al., 2004; Pechtl et al., 2006), and amines (Kurtén et al., 602 2008; Berndt et al., 2013). Limited observations also introduce some uncertainties in the model validation. The overprediction of J over ocean is mainly due to the use of the prefactor of 1×10^{-6} 603 604 in WP09. This prefactor is derived from limited in-situ measurements (Sihto et al., 2006). It can 605 vary by up to 3-4 orders of magnitude based on measurements in different areas and seasons 606 (Zhang et al., 2010), introducing a large uncertainty for its application to the global scale. In MAM CON/IMN, a prefactor of 1×10^{-8} is used in WP09 in PBL over the globe, which then 607 608 decreases J and aerosol number concentrations in PBL (see Figure 4). J in PBL is very sensitive 609 to the prefactor in WP09, and the uncertainty of the prefactor can result in a large bias in 610 predictions of J and aerosol number in PBL. With the implementation of IMN, J values in the 611 troposphere increase by factors of 2-10, which in turn increase the aerosol number concentrations 612 in the troposphere. Due to a stronger radiation in the upper layer, more available ions can 613 contribute to the new particle formation, therefore increasing the aerosol number concentrations 614 in the middle/upper troposphere and lower stratosphere by factors of 2-4. 615 Figure 5 shows the absolute differences of PM_{2.5}, AOD, column CCN5, CF, SWCF, and

617 CCN, which can affect cloud formation and properties as well as radiation. Changes of PM

616

618 concentrations also have impacts on AOD, CCN, CF, COT, and SWCF through both aerosol

619 direct and indirect effects. As a net result of all those interwoven changes initially triggered by

SWD between MAM CON and MAM CON/IMN for 2001. Aerosol number can directly affect

the increase of aerosol number concentrations in troposphere/stratosphere, AOD and column CCN5 increase by 0.004 (or by 3.3%) and 2.1×10^7 cm⁻² (or by 11.9%), respectively, and SWCF and SWD decrease by 0.1 W m⁻² (or by 0.2%) and 0.8 W m⁻² (or by 0.5%), respectively, in terms of global mean. As shown in Table A1, the changes in SWD, AOD, and cloud variables such as column CCN5, CDNC, and COT between MAM_CON and MAM_CON/IMN are statistically significant, indicating the significant impacts of IMN on aerosol number concentration and cloud prediction.

627 Compared with MAM CON, IMN (MAM CON/IMN) improves the predictions of SO₂, NO_3^- , and PM_{25} over CONUS, SO_2 , SO_4^{2-} , NH_4^+ , NO_3^- , Cl^- , PM_{25} , and PM_{10} over Europe, PM_{10} 628 629 over East Asia (see Table 4). The improved performance in aerosol concentrations and increased 630 aerosol numbers in the troposphere and lower stratosphere contribute to the improved performance of aerosol and cloud parameters, with increased AOD, CCN, and CDNC, and 631 632 consequently increased CF, COT, CWP, and SWCF, as shown in Table 3. However, there are still 633 large biases for some chemical species predictions. For example, CO mixing ratio is 634 underpredicted over East Asia, which is mainly due to the uncertainty in CO emissions in this 635 region. Large biases in SO₂ predictions over CONUS, Europe, and East Asia are mainly due to 636 the uncertainties in SO₂ emissions in those regions. Large biases in NO₂ and HNO₃ predictions 637 over Europe are mainly due to the uncertainties in NO_x emissions and inaccurate predictions of 638 radiation over this region. The performance of J degrades with NMBs from -21.8% to -49.6% in 639 the globe, which is due to the use of a smaller prefactor of WP09 in MAM CON/IMN than in 640 MAM CON. J in PBL is very sensitive to the prefactor in WP09. Although the prediction of J 641 over ocean in PBL has been improved in MAM CON/IMN, J over land areas in PBL is largely 642 underpredicted by factors of 1-100, resulting in degraded J performance in terms of globe mean.

The underprediction of J over land in PBL is likely due to the uncertainties in the nucleation parameterizations (e.g., the missing species as mentioned previously). Large NMBs still remain for COT, CWP, and CCN, indicating the uncertainties in the treatments of related atmospheric processes such as cloud microphysics and aerosol-cloud interactions.

647

648 5.4 Impacts of Gas-Aerosol Partitioning

649 The inclusion of ISORROPIA II changes the mass concentrations of major PM_{2.5} species 650 and their gaseous precursors. Changes in PM concentrations then affect predictions of cloud 651 variables and therefore radiation. Changes of radiation can also affect SO₂ oxidation by OH, 652 which affects H_2SO_4 concentrations. Figure 6a shows the absolute differences of H_2SO_4 , fine 653 particulate sulfate (SO4f), NH₃, fine particulate ammonium (NH4f), HNO₃, fine particulate 654 nitrate (NO3f), HCl, and fine particulate chloride (CLf) for summer 2001 between MAM CON 655 and MAM CON/ISO. Similar plots for winter (December, January, and February (DJF)) 2001 656 are shown in Figure 6b. Compared to MAM CON, MAM CON/ISO gives higher H₂SO₄ mixing ratios but lower SO4f concentrations. SWD increases with the global mean of 8.9 W m^{-2} (~ 657 658 5.8%) in MAM CON/ISO, which allows more production of OH from photolytic reactions of 659 VOCs, HONO, HNO₃, HNO₄, H₂O₂, HOCl, and HOBr, and therefore enhanced oxidation of SO₂ 660 to form H_2SO_4 . As shown in Figure 6a, the mixing ratios of H_2SO_4 either increase up to 0.76 ppt 661 or decrease as large as 1.14 ppt, leading to a net increase of 0.002 ppt in terms of global mean. 662 The mass concentration of SO4f is mainly affected by H_2SO_4 condensation. Although the mixing 663 ratios of H₂SO₄ increase with the global mean change of 0.002 ppt, SO4f concentrations decrease with the global mean of $0.02 \ \mu g \ m^{-3}$, which are mainly due to less condensation of H₂SO₄ under 664 665 higher temperature conditions. In summer, the increase or decrease of H_2SO_4 can result in an

666 increase or a decrease of SO4f (e.g., over most oceanic areas). However, the decrease of SO4f 667 with the increase of H_2SO_4 over the India Ocean is mainly due to less H_2SO_4 condensation. For 668 the regions where SO4f increases over land, the increase of SO4f is due to more oxidation of SO_2 669 by OH.

670 Compared to MAM CON, the concentrations of NH_3 , HNO_3 , and HCl increase 671 significantly over most land areas, whereas NH4f, NO3f, and Clf decrease significantly over 672 most land areas in MAM CON/ISO. The chemical regimes is the controlling factor for gasaerosol equilibrium partitioning, which is determined based on the ratio of SO_4^{2-} molar 673 674 concentrations to total molar concentrations of cations and their respective gases (referred to as 675 TCAT/TSO4) (Zhang et al., 2000). Three regimes are defined based on the values of 676 TCAT/TSO4: (1) if TCAT/TSO4 < 2, the system contains excess sulfate and is in a sulfate-rich 677 regime; (2) if TCAT/TSO4 = 2, the system contains just sufficient sulfate to neutralize the cation 678 species and is in sulfate-neutral regime; (3) if TCAT/TSO4 > 2, the system contain insufficient 679 sulfate to neutralize the cation species and is in sulfate-poor regime. Over land, the major cation is NH_4^+ , and there are also crustal species (K⁺, Ca²⁺, and Mg²⁺) associated with dust emissions, 680 whereas over ocean, the major cation is Na⁺, which is a non-volatile species. Therefore, the gas-681 682 aerosol equilibrium partitioning behaves differently over land and over ocean. Figure 7 shows 683 the distributions of TCAT/TSO4 in MAM CON and MAM CON/ISO, and their absolute 684 differences for summer and winter, 2001. In summer, compared to MAM CON, TCAT/TSO4 in 685 MAM CON/ISO either increases up by 80.1 (mostly over ocean) or decreases up by 51.8 (over 686 both land and ocean), leading to a net increase of 0.7. In MAM CON, most regions are in 687 sulfate-poor regime, whereas Greenland, southeast U.S., North Africa, a small portion of Asia 688 and North Atlantic Ocean, and some areas in North Pole are in sulfate-rich regime in summer.

689 However, due to the simplified thermodynamics treatment in MAM CON, NH₃ is underpredicted and NH₄⁺ is overpredicted (see Table 4). With the inclusion of ISORROPIA II, 690 691 most sulfate-poor regions over land and over part of Pacific Ocean and most Atlantic Ocean 692 become less sulfate-poor. The sulfate-poor regime can drive HNO₃/HCl to produce NO₃/Cl⁻ by neutralizing excess NH_4^+ . If the amount of NO_3^-/Cl^- is insufficient to neutralize NH_4^+ , sulfate-693 poor regime can drive NH_4^+ to the gas phase to produce NH_3 . Therefore, the increase of NH_3 and 694 decrease of NH_4^+ in MAM_CON/ISO are mainly due to insufficient NO_3^-/Cl^- to neutralize NH_4^+ 695 696 under sulfate-poor regime. Insufficient NO_3^{-}/Cl^{-} results from the thermodynamic partitioning 697 under higher temperature conditions that favors the production of HNO₃ and HCl from NO₃/Cl⁻ 698 to produce HNO₃ and HCl under higher temperature conditions. The slight increase of NO₃⁻ over 699 Pacific Ocean and South Atlantic Ocean is due to much higher Na⁺ concentrations yet insufficient SO_4^{2-} in those regions compared with those over the land areas. Unlike a sulfate-poor 700 701 regime, a sulfate-rich regime (e.g., small portion of North Atlantic Ocean, South China Sea, and Greenland), requires more cations such as NH_4^+ and Na^+ to neutralize excess SO_4^{2-} in the system 702 and the thermodynamics favors the partitioning of volatile species such as NO₃⁻ and Cl⁻ in the 703 gas phase as HNO₃ and HCl. Therefore, despite the increased temperatures, the decrease of NH_4^+ 704 705 due to its evaporation back to the gas-phase is not as significant as that of NO_3^- and Cl^- , because NH_4^+ needs to stay in the system to neutralize SO_4^{2-} . In winter, as shown in Figure 6b, compared 706 707 with MAM CON, the mixing ratios of H₂SO₄ in MAM CON/ISO either increase by up to 4.3 708 ppt, or decrease by up to 1.0 ppt, leading to a net increase with the global mean of 0.001 ppt. 709 NH₃ increases over most regions except Europe, eastern China, and some regions in North Pole. 710 HNO₃ decreases over most oceanic areas, Northeastern China, and East Europe, whereas 711 increases over South Asia, North Pole, southern U.S., Africa, and most land areas in southern

hemisphere. HCl increases over most areas except the northeastern portion of Asia and easternEurope.

Compared with MAM CON, MAM CON/ISO predicts higher HNO₃ and HCl over some 714 715 land areas. As shown in Figure 7, in MAM CON, most regions are in sulfate-poor regime, 716 whereas Greenland, North Pole, North Africa, some portions of Asia and western Pacific Ocean 717 are in sulfate-rich regime. For example, northeastern China is in sulfate-poor regime, driving HNO_3 and HCl partitioning to the aerosol phase to neutralize excess NH_4^+ . This results in an 718 719 increase in NO3f and Clf, changing sulfate-poor regime to less sulfate-poor. North Pacific Ocean 720 and southern oceanic areas are also in sulfate-poor regime, and the increase of NO3f is due to the partitioning HNO₃ to the aerosol phase to neutralize Na⁺, whose concentration is relatively 721 722 higher compared to that over land areas. Therefore, more anions such as NO₃⁻ are needed to 723 neutralize the system. However, the decrease Cl⁻ over these regions is due to the equilibrium 724 state of HCl under different atmospheric conditions. The western Pacific Ocean is in sulfate-rich 725 regime, driving NO_3^- and Cl^- partition to the gas phase, which results in a decrease in NO3f and 726 Clf, and an increase in HNO₃ and HCl over this region. With the inclusion of ISORROPIA II, the 727 western Pacific Ocean changes from sulfate-rich regime to less sulfate-rich regime. 728 Compared to MAM CON, the prediction of SWD in MAM CON/ISO is improved with the NMB decreasing from -6.5% to -2.2%. The predictions of involved species such as NH_4^+ , 729 730 NO_3^- , and Cl⁻ are improved significantly by 13.6%~345.4%, although there is a slight degradation in the predictions of SO_4^{2-} and O_3 over CONUS, CO, O_3 , PM_{2.5}, and PM₁₀ over 731 732 Europe, PM₁₀ over East Asia, and column CO, NO₂, TOR, and J over globe. MAM CON/ISO

improves the predictions of HNO₃, NH₄⁺, NO₃⁻, Cl⁻, BC, OC, TC, and PM_{2.5} over CONUS, SO₂,

NH₃, NO₂, SO₄²⁻, NH₄⁺, NO₃⁻, and Cl⁻ over Europe, and CO and SO₂ over East Asia, which leads

735	to improved performance in SWD, column CCN5, and SWCF over globe, as shown in Table 3.
736	As shown in Table A1, the changes in most radiative and cloud variables between MAM_CON
737	and MAM_CON/ISO are statistically significant, indicating the significant impacts of
738	ISORROIA II on the predictions of radiation, aerosol, and cloud. ISORROPIA II calculates gas-
739	aerosol partitioning under different atmospheric conditions, significantly improving predictions
740	of major gas precursor (e.g., HNO ₃) over CONUS and secondary aerosols (e.g., NO ₃ ⁻ and Cl ⁻)
741	over CONUS and Europe. Large decreases in the concentrations of NO ₃ ⁻ and Cl ⁻ result in a
742	decrease in NH_4^+ , $PM_{2.5}$, and PM_{10} , thus decreasing CCN, CDNC, AOD, and the absolute value
743	of SWCF.
744	Figure 8 shows the absolute differences of major inorganic gas and aerosol species
745	between metastable (MAM_NEWA) and stable (MAM_NEWB) conditions. Compared with
746	MAM_NEWA, the global average changes predicted by MAM_NEWB are overall small (within
747	5%) for most gaseous and aerosol species. For example, the global average changes are 0.01 μ g
748	m^{-3} (by 4.2%) for SO ₄ ²⁻ , 0.005 µg m ⁻³ (by 12.8%) for NH ₄ ⁺ , 0.006 µg m ⁻³ (by -0.01%) for NO ₃ ⁻ ,
749	and $-4 \times 10^{-4} \ \mu g \ m^{-3}$ (by 2.0%) for Cl ⁻ . The increase of SO ₄ ²⁻ results in an increase in NH ₄ ⁺ (e.g.,
750	East Asia and Northeast U.S.). The differences between stable and metastable conditions may be
751	more significant under low RH conditions ($RH < 50\%$ for nitrate, Fountoukis et al., 2009).
752	However, based on the simulated global annual mean RH values, most regions have RH values >
753	60-70% (exceptions are over desert/arid regions such as Australia, the northern Africa, Arabian
754	Desert, northwestern China, and western U.S.). These results indicate that the assumption of
755	metastable conditions is not a significant sources of uncertainty for global model predictions of
756	gaseous and aerosol species.
757	

758 5.5 Overall Impacts of All New and Modified Model Treatments

759 Compared to MAM CB05 GE, the simulations with modified or new aerosol treatments 760 (MAM CON, MAM CON/IMN, MAM CON/ISO, MAM NEWA) slightly degrade the 761 prediction of LWD (increasing NMB from -0.9% to -1.4%), but improve the predictions of OLR, 762 CF, COT, and CWP slightly (with 0.6% - 10.4% decreases in their NMBs) and CDNC 763 significantly (reducing NMBs from -57.5% up to -13.4%). Although the CCN predictions are 764 somewhat degraded in MAM CON and MAM CON/IMN, they are improved significantly in 765 MAM CON/ISO and MAM NEWA (reducing NMBs from -61.6% to 1.8-6.3%). As shown in 766 Table A1, changes in most radiative and cloud variables between MAM SIM and MAM NEWA 767 are statistically significant, indicating the significant impacts of new and modified treatments on 768 predictions of radiation and cloud. Among all new and modified model treatments, the new gas-769 phase chemistry simulates more gaseous species and improves the predictions of NH₃ over 770 Europe, PM_{2.5} over CONUS and PM₁₀ over East Asia. The modified condensation and aqueous-771 phase chemistry simulate more aerosol species (NO₃⁻ and Cl⁻) and improve the prediction of 772 HNO₃. MAM CON also improves J in the PBL due to more available H₂SO₄ involving in the 773 homogeneous nucleation using an accommodation coefficient of 0.02 for H₂SO₄ condensation, 774 and improves the predictions of CDNC and AOD significantly. MAM CON/IMN increases 775 PM_{num} above PBL and PM_{2.5} and PM₁₀ over Europe and improves the prediction of PM_{2.5} over CONUS and Europe. MAM CON/ISO improves the predictions of HNO₃, NH₄⁺, PM_{2.5}, NO₃⁻, 776 777 and Cl⁻ over CONUS, NO₃⁻ and Cl⁻ over Europe, and CCN over globe, and improves the 778 predictions of SWCF most (with an NMB of 1.6%). 779 Large biases in some variables remain in MAM NEWA due to uncertainties in model

780 inputs (e.g., meteorology and emissions) and model treatments (e.g., multi-phase chemistry, dust

781 emission scheme, cloud microphysics, aerosol activation, SOA formation, and dry and wet deposition). The large NMBs of CO and SO₂ over East Asia, SO₂, NH₃, and NO₂ over Europe, 782 SO₂, and BC over CONUS are likely due to the uncertainties of emissions and the interpolation 783 784 of emissions from a fine-grid scale in the original emission inventories (e.g., county-based 785 emissions over CONUS) to a large-grid scale used in this work, which can result in large NMBs in secondary aerosols (e.g., $SO_4^{2^-}$, NH_4^+ , NO_3^- , thus $PM_{2.5}$ and PM_{10}). Heterogeneous reactions 786 787 are not included in this work, which may help explain to some extent less oxidation and 788 underpredictions for PM species predictions (e.g., sulfate and nitrate) and overpredictions for 789 gaseous species. The large NMB of O_3 predictions over Europe in MAM NEWA (with an NMB) 790 of 62.7%) is mainly due to a lack of NO_x titration (as indicated by large underpredictions in NO_2) 791 and more production of O₃ from the photolytic reaction of NO₂ resulted from overpredictions of 792 SWD particularly in autumn and winter. Table 5 shows the seasonal statistics for O₃, NO₂, and 793 HNO₃ over Europe in MAM NEWA. During autumn and winter, O₃ is overpredicted by about 794 $100\% \sim 140\%$, whereas NO₂ is underpredicted by about -85% \sim -20%, indicating insufficient NO_x for titration of O₃ titration. SWD is overpredicted by 45.0 W m⁻² (or by 58.4%), favoring the 795 796 photolytic reactions of NO₂ to produce O_3 . Due to the uncertainties in the NO_x emissions, NO₂ is 797 underpredicted, causing less NO₂ to be oxidized to produce HNO₃, which results in an underprediction of HNO₃ in winter. In autumn, SWD is overpredicted by 42.8 W m⁻² (or by 798 799 37.9%). However, in autumn, although NO₂ is underpredicted due to the uncertainties in the NO_x 800 emissions, HNO₃ mixing ratios are overpredicted. SWD is stronger in autumn than in winter, and 801 mixing ratios of OH are higher due to photolytic reactions of overpredicted O_3 and additional 802 photolytic reactions of VOCs. Therefore, OH can oxidize NO₂ to produce HNO₃, resulting in the 803 overprediction of HNO₃. Simple aqueous-phase chemistry is included in this work, which could

804	result in high uncertainty in predicting aerosols in clouds. Decreased aerosol number
805	concentrations can result in a decrease of CCN and AOD directly. The underpredictions of
806	CDNC are likely due to uncertainties in the model treatments for aerosol activation and cloud
807	microphysics, which then result in large NMBs in COT and CWP. The large biases in OC and TC
808	indicate the uncertainties in the emissions of BC and primary OC, and the treatments for SOA
809	formation. The large NMB in particle formation rate J is likely due to uncertainties in model
810	inputs (e.g., SO_2 emissions) and model treatments (e.g., the accommodation coefficient of H_2SO_4
811	and missing participants in the current nucleation schemes).
812	
813	5.6 Impacts of Adjusted Emissions
814	The evaluation and analyses of MAM_NEWA indicate that some large biases are caused
815	by inaccuracies in the emissions of CO, SO ₂ , BC, OC, and NH ₃ . The sensitivity simulation with
816	adjusted emissions of CO, SO ₂ , BC, OC, and NH ₃ (MAM_NEW/EMIS) is performed to further
817	look into such impacts. For example, with 30% increase in CO emissions and 20% increase in
818	NH ₃ emissions over Europe, the NMBs of surface concentrations of CO and NH ₃ change from-
819	3.4% to 12.1%, -84.3% to -77.5%, respectively. On a global scale, the increased CO emissions
820	result in 3.0% absolute reduction in the NMB of column CO. The 30% reduction in SO_2
821	emissions and 20% increase in OC and BC emissions over CONUS result in 139.6%, 8.6%, and
822	24.9% absolute reduction in their NMBs. The 30% increase in CO emissions and 20% increase
823	in SO ₂ over East Asia result in 3.3% and 7.8% absolute reduction in their NMBs.
824	As shown in Table 4, compared with MAM_NEWA, MAM_NEW/EMIS shows an
825	improved performance in the concentrations of SO ₂ , HNO ₃ , SO ₄ ^{2^-} , NH ₃ , and NH ₄ ⁺ over Europe,
826	SO ₂ , HNO ₃ , BC, OC, TC, NO ₃ ⁻ , and Cl ⁻ over CONUS, CO and SO ₂ over Asia, and column CO

over globe. However, it degrades to some extent the performance of SO_4^{2-} and NH_4^+ over 827 CONUS, PM_{2.5} and PM₁₀ over Europe, PM₁₀ over Asia, and J over globe. Decreased SO₂ 828 emissions over CONUS result in a decrease of H_2SO_4 and therefore a decrease of SO_4^{2-} . Based 829 on aerosol thermodynamic treatments, decreased SO₄²⁻ will result in decreased NH₄⁺. Therefore, 830 PM_{2.5} and PM₁₀ decrease as well. Adjusted emissions can affect secondary aerosol formations 831 832 and therefore radiative variables can be affected due to the direct and indirect effects of aerosols. 833 As shown in Table 3, compared with MAM NEWA, MAM NEW/EMIS reduces MB of LWD 834 by 9.3%, SWD by 37.5%, and CF by 18.9%, leading to 0.1% - 1.6% absolute reduction in their 835 NMBs. This illustrates the sensitivity of radiation to the perturbations in emissions through 836 chemistry feedbacks to the climate system. As shown in Table A1, only column CCN5 and AOD 837 are significantly different between MAM NEWA and MAM NEW/EMIS, indicating the 838 impacts of emissions are more significant on predictions of gas and aerosol than radiative 839 variables.

840

841 6. Evaluation of the Five-Year Simulations

842 6.1. Performance Evaluation

Tables 6 and 7 show the statistical performance for radiative/cloud variables and chemical predictions, respectively, from the 5-yr simulations using three different configurations. Compared with MAM_SIM_5Y, MAM_NEW_5YA improves the predictions of aerosol and cloud variables such as AOD, COT, CWP, CCN5, and CDNC (with 4.8% to 23.4% absolute reduction in their NMBs), and radiative variables such as SWD, LWD, OLR, and SWCF (with 0.4-4.2% absolute reduction in their NMBs). MAM_NEW_5YA also shows slight improvement for the predictions of SO₄²⁻ and BC over CONUS and SO₂ over East Asia (with 0.3-2.3%

850	absolute reduction in their NMBs), but moderate-to-large improvements for the predictions of
851	OC, TC, and $PM_{2.5}$ over CONUS, PM_{10} over East Asia, and SO ₂ , $PM_{2.5}$, and PM_{10} over Europe
852	(with 5.2-20.1% absolute reduction in their NMBs). Compared to TOR calculated based on O_3
853	climatology used in MAM_SIM_5Y, TOR predicted from MAM_NEW_5YA is slightly
854	improved with 1.2%, 1.3%, and 0.3 absolute reduction in its NMB, NME, and RMSE,
855	respectively. Evaluation of major radiative/cloud variables and chemical predictions are also
856	conducted for June, July, and August (JJA) of 2001-2005, which is shown in Tables A2 and A3 in
857	the supplementary material. Compared with full 5-year (2001-2005) average, the simulation for
858	JJA gives similar predictions for chemical species but better model predictions for radiation (e.g.,
859	LWD, SWD, and OLR) and cloud (e.g., COT, CWP, column CCN5, and CDNC) variables.
860	Tables 6 and 7 also show the performance of MAM_NEW_5YB in which CAM5 is fully
861	coupled with land, ocean, and ice models. The performance is overall similar for all radiative
862	variables and most chemical species between MAM_NEW_5YA and MAM_NEW_5YB (most
863	within 5% differences in the absolute values of their NMBs). The performance of HNO ₃ over
864	CONUS and Europe, NH_4^+ , NO_3^- , and Cl^- over Europe, PM_{10} over Europe and East Asia is
865	improved appreciably (with 4.2-17.9% reduction in the absolute values of their NMBs), and that
866	of SO ₂ over CONUS and Europe and NH_4^+ , NO_3^- , and Cl^- over CONUS degrades appreciably
867	(with 4.3-8.5% increase in the absolute values of their NMBs). Those changes are mainly due to
868	the interactions among Earth's components, particularly at the interface of earth components
869	(e.g., air-sea, air-land, and sea-ice interfaces) and feedbacks to the climate system, which in turn
870	affects gaseous and aerosol concentrations in the coupled system.
871	Large biases remain for some variables in MAM_NEW_5YA and MAM_NEW_5YB due

to uncertainties in model inputs (e.g., meteorology and emissions) and model treatments (e.g.,

multi-phase chemistry, dust emission scheme, cloud microphysics, aerosol activation, SOA
formation, and dry and wet deposition), which have been illustrated in Section 5.5. Large biases
in Cl⁻ predictions over Europe are likely due to the combined effects of a low concentration of
observed Cl⁻, uncertainties in HCl emissions, and inaccurate predictions of coarse Cl⁻ in the
model since ISORROPIA II is only implemented for fine particles. Uncertainties in the mass
accommodation coefficients of volatile gas species can also result in uncertainties in predictions
of condensable gases.

880

6.2 Impact of New and Modified Treatments on 5-year (2001-2005) Simulations

Figure 9 shows the absolute differences of surface SO₂, NH₃, SO₄²⁻, NH₄⁺, TC, PM_{2.5}, 882 883 PM_{10} , J, and aerosol number (PM_{num}) and Figure 10 shows the absolute differences of radiative 884 variables between MAM SIM 5Y and MAM NEW 5YA. The new and modified model 885 treatments in MAM NEW 5YA cause changes in the concentrations of PM and precursor gases, 886 which affect radiative variables through aerosol direct and indirect effects. The changes in 887 radiative variables in turn affect gas-phase chemistry and aerosol processes. As shown in Figure 888 9, the difference of SO_2 between the two simulations varies from -1.7 to 3.8 ppb, with a global 889 mean difference of 4.2 ppt. The decrease of SO₂ over most oceanic area is mainly due to the decrease of DMS resulted from less oxidation by OH radicals. The increase of SO_4^{2-} over East 890 Asia and eastern U.S. drives more NH_3 from gas-phase to particulate phase to form NH_4^+ 891 through thermodynamic equilibrium, increasing the concentrations of NH_4^+ over these regions. 892 However, the concentrations of SO_4^{2-} decrease over Europe due in part to less oxidation of SO_2 . 893 Despite such a decrease, the concentrations of NH_4^+ are higher over Europe due to the 894 895 neutralization of NH₃ by Cl⁻ and NO₃⁻ that are treated in MAM NEW 5YA but not treated in in

896 MAM_SIM_5Y. Compared with MAM_SIM_5Y, J from MAM_NEW_5YA increases over 897 globe with a global mean difference of 0.066 cm⁻³ s⁻¹, due to the use of a lower mass 898 accommodation coefficient of H_2SO_4 in MAM_NEW_5YA, resulting in more available H_2SO_4 899 vapor participating in nucleation. The increases in J result in an increase in aerosol mass and 900 number concentrations and thus higher concentrations of $PM_{2.5}$ and PM_{10} (which improve 901 appreciably their performance, see Table 6).

902 As shown in Figure 10, compared with MAM SIM 5Y, AOD increases by 0.007, column CCN5 increases by 3.8×10^7 cm⁻², and CDNC increases by 16.1 cm⁻³ in MAM NEW 5YA. 903 904 Higher PM_{num} in MAM NEW 5YA allows more aerosol to grow into the CCN size, leading to 905 higher CCN in MAM NEW 5YA. Higher aerosol concentrations in MAM NEW 5YA result in 906 higher AOD. The increased aerosol number and mass concentration result in an increase in the 907 predictions of cloud variables through the aerosol-cloud interactions. For example, with all the modified and new treatments, COT increases by 0.8, CWP increases by 4.1 g m⁻², and PWV 908 909 increases by 0.026 cm on global average. Due to the aerosol direct and indirect effects, the difference in simulated SWD varies from -19.3.0 to 10.4 W m⁻² and decreases by 3.4 W m⁻² (\sim 910 2.0%) on a global average. The difference in LWD varies from -4.2 to 8.5 W m⁻² and increases by 911 1.0 W m^{-2} (~ 0.4%) on a global average (Figure not shown). The difference in SWCF varies from 912 -8.4 to 17.9 W m⁻², with a net increase of 2.7 W m⁻² ($\sim 6.4\%$) on a global average. The absolute 913 914 differences of surface chemical species and major cloud/radiative variables for JJA average of 915 2001-2005 are shown in Figures A1 and A2 in the supplementary material. Compared with 5-916 year average, the absolute changes of most radiative variables are smaller in JJA. The absolute 917 changes of PM₁₀ are smaller in JJA, which is mainly due to the dust events during other months 918 (e.g., March-May over East Asia).

920 6.3 Global Burden Analysis

921	Table 8 shows the simulated global burdens of major gas and aerosol species for 2001-
922	2005. The global burdens of most gaseous precursors of aerosol are higher in MAM_NEW_5YA
923	than MAM_SIM_5Y (except for NH_3), due mainly to the incorporation of ISORROPIA II in
924	MAM_NEW_5YA. The global burden of tropospheric O_3 is higher in MAM_NEW_5YA than
925	MAM_SIM_5Y, which is due to higher mixing ratios of O ₃ precursors (e.g., NO ₂ and VOCs) that
926	are simulated in MAM_NEW_5YA. The global burdens of most gas species are comparable with
927	previous studies (Horowitz et al.2006; Larmarque et al., 2006; Williams et al., 2009; Liu et al.,
928	2012) with absolute differences of less than 20%. One exception is H_2SO_4 , which is higher by a
929	factor 5 in MAM_NEW_5YA than in MAM_SIM_5Y. The higher burden of H_2SO_4 in
930	MAM_NEW_5YA is likely due to the less condensation of H_2SO_4 resulted from the use of a
931	lower mass accommodation coefficient. SO_4^{2-} burden is higher by 8.3% in MAM_NEW_5YA
932	than MAM_SIM_5Y, which is likely due to greater SO ₂ oxidation in MAM_NEW_5YA. Higher
933	SO_4^{2-} burden results from higher SO_2 burden. Higher SO_2 burden leads to more SO_2 to be
934	oxidized to produce SO_4^{2-} , which overweighs the impacts from less H_2SO_4 condensation due to
935	lower mass accommodation coefficient. More SO_4^{2-} results in more NH_4^+ . The burdens of BC
936	and POM are lower by 16.5% and 23.8%, respectively, in MAM_NEW_5YA than in
937	MAM_SIM_5Y, which is likely due in part to greater dry deposition fluxes and in part to a
938	slower primary carbon aging rate resulted from reduced condensation of gas species in
939	MAM_NEW_5YA. Condensation onto the primary carbon mode produces aging of the particles
940	in this mode. A lower accommodation coefficient is used in MAM_NEW_5YA, which results in
941	less condensation. Therefore, the fraction of aged particles has decreased. The global burdens of

942	most aerosol species are in the range of previous studies. For example, global burdens of SO_4^{2-}
943	and NH_4^+ from MAM_SIM_5Y and MAM_NEW_5YA are 23.4% and 17.0%, respectively, and
944	16.7% and 12.5%, respectively, lower than Liu et al. (2012), which is likely because
945	MAM_SIM_5Y contains no SO_4^{2-} emissions but Liu et al. (2012) included additional SO_4^{2-}
946	emissions of 1.66 Tg S yr ⁻¹ . Higher SO_4^{2-} emission leads to more SO_4^{2-} concentrations thus more
947	$\rm NH_4^+$ in Liu et al. (2012). Compared with Horowitz et al. (2006), global burdens of BC and OC
948	from MAM_NEW_5YA are lower by 72.9% and 52.3%, respectively. Compared with Liu et al.
949	(2012), MAM_NEW_5YA gives comparable BC and POM burdens but much lower SOA (by a
950	factor of 3.0). Compared with Textor et al. (2006), POM burden is a factor of 3.5 lower in
951	MAM_NEW_5YA. The lower BC, OC, POM, and SOA burdens are likely due to the
952	uncertainties in the BC and OC emissions used as well as differences in the model treatments for
953	SOA formation and POM aging.

954

955 **7. Conclusions and Future work**

956 In this work, a new gas-phase chemistry mechanism and several advanced inorganic 957 aerosol treatments have been incorporated into CESM/CAM5.1-MAM7. These include (1) the 958 CB05 GE gas-phase chemical mechanism coupled with MAM7; (2) the condensation and 959 aqueous-phase chemistry involving HNO₃/ NO₃⁻ and HCl/ Cl⁻; (3) an ion-mediated nucleation 960 (IMN) parameterization for the new particle formation from ions, (4) an inorganic thermodynamic module, ISORROPIA II, that explicitly simulates thermodynamics of SO_4^{2-} , 961 NH_4^+ , NO_3^- , CI^- , and Na^+ as well as the impact of crustal species, such as Ca^{2+} , K^+ , and Mg^{2+} , on 962 963 aerosol thermodynamics. CB05 GE with new and modified inorganic aerosol treatments in MAM7 simulates 139 species with 273 chemical reactions, which is more accurate than simple 964

965 gas chemistry coupled with default MAM7. Seven 1-yr simulations for 2001 and three 5-yr 966 simulations for 2001-2005 with different model configurations are performed to evaluate the 967 capabilities of the original and improved CESM/CAM5 and the mechanisms underlying 968 differences among model predictions. 969 Comparing to the simple gas-phase chemistry, the 2001 simulation with CB05 GE can 970 predict many more gaseous species, and give improved performance for predictions of organic carbon and PM_{2.5} over CONUS, NH₃ and SO₄²⁻ over Europe, SO₂ and PM₁₀ over East Asia, and 971 972 cloud properties such as CF, CDNC, and SWCF. MAM CON simulates NO₃⁻ and Cl⁻, which are 973 important inorganic aerosols. With species-dependent accommodation coefficients for gas 974 condensation, more H_2SO_4 can participate in homogeneous nucleation, resulting in the 975 improvement of predictions of PM_{2.5}, PM₁₀, J, CDNC, and SWCF. IMN can increase the 976 predictions of J and PM_{num} in the upper atmosphere and thus improve the predictions of AOD, 977 CCN, and cloud properties, and SWCF over globe, PM_{2.5} over CONUS and Europe, PM₁₀ over 978 Europe and East Asia, and PM composition over Europe. The 2001 simulation with ISORROPIA 979 II can improve the predictions of major gas and aerosol species significantly, including HNO_3 , NH₄⁺, NO₃⁻, Cl⁻, BC, OC, TC, and PM_{2.5} over CONUS, SO₂, NH₃, NO₂, SO₄²⁻, NH₄⁺, NO₃⁻, and 980 981 Cl⁻ over Europe, and CO and SO₂ over East Asia. Such improvements lead to improved 982 predictions of SWD, SWCF, and CCN5 over globe. The 2001 simulation with the new and 983 modified inorganic aerosol treatments appreciably improve the predictions of OLR, CF, COT, CWP, PWV, CCN, CDNC, SWCF, J over globe, and HNO₃, NH₄⁺ (CONUS), PM₂₅, and PM₁₀. 984 985 The 2001 sensitivity simulation with adjusted emissions further improves model predictions of 986 CO and SO₂ over East Asia, SO₂, HNO₃, NO₃⁻, Cl⁻, BC, OC, and TC over CONUS, SO₂, NH₃, NH4⁺, HNO3, NO3⁻, and Cl⁻ over Europe, and column CO and SWD over globe. The change of 987

emissions can affect primary gaseous precursors directly, and secondary gaseous species
indirectly through gas-phase chemistry. Meanwhile, secondary aerosols can be affected by
gaseous precursors, and therefore have impacts on cloud properties as well as direct and indirect
effects on radiation. Reducing the uncertainty of emissions can thus help reduce the model biases

significantly.

993 The comparison of the 5-yr simulations with prescribed SST shows that

994 MAM_NEW_5YA with CB05_GE can appreciably improve the predictions of AOD, COT, CWP,

995 CCN5, CDNC, SWD, LWD, OLR, and SWCF on global scale and OC, TC, and PM_{2.5} over

996 CONUS, PM₁₀ over East Asia, and SO₂, PM_{2.5}, and PM₁₀ over Europe. The performance is

997 overall similar for all radiative variables and most chemical species between MAM_NEW_5YA

998 with prescribed SST and MAM_NEW_5YB in a fully-coupled mode.

999 In addition to uncertainties in emissions, additional uncertainties exist in the model 1000 treatments. For example, the large biases in the predictions of O₃ over Europe and East Asia are 1001 mainly due to insufficient NO_x titration resulting from the uncertainties in the NO_x emissions. 1002 The large biases in PM_{10} over East Asia and Europe may be mainly due to the inaccurate 1003 predictions of dust. The large bias in Cl⁻ over Europe may be due to the inaccurate predictions of 1004 HCl and coarse Cl⁻ as well as the uncertainty in the mass accommodation coefficient of HCl 1005 used. In the default and modified nucleation treatments, it only considers H₂SO₄, NH₃, H₂O, and 1006 ions involving in the new particle formation. Missing species (e.g., organics, iodine compounds, 1007 and DMS) may also contribute to the new particle formation. Uncertainties in treating organic 1008 gas-aerosol partitioning may contribute to the inaccurate predictions of SOA, OC, TC, and PM. 1009 The large biases in CDNC, COT, and LWP indicate the uncertainties in cloud microphysics 1010 schemes and aerosol-cloud interaction parameterizations, which also limit the ability of climate

1011 and Earth system models to quantify aerosol indirect effects (Stephens, 2005; Lohmann et al., 1012 2007; Gettelman et al., 2008). In addition to uncertainties in the model treatments, uncertainties 1013 in the model simulation settings such as the use of a coarse grid resolution and a large model 1014 time step of 1800 seconds for solving the chemical system in this work may contribute to the 1015 model biases. The representations of some of the aforementioned uncertain processes in 1016 CESM/CAM5.1 are being further improved by the authors' group. Decadal simulations using 1017 improved CESM/CAM5.1 will be conducted in the future to study the interactions among 1018 atmospheric chemistry, aerosol, and climate change and reduce associated uncertainties

1019

1020 Acknowledgments

1021 This work is sponsored by the U.S. NSF EaSM program AGS-1049200. The authors would like

1022 to thank Fangqun Yu for providing the IMN scheme, Athanasios Nenes for providing

1023 ISORROPIA II, Xiaohong Liu for providing a version of MAM7 that works in CAM5.0 and

1024 CAM5.1, Ralf Bennartz and John Rausch for providing CDNC data, Steve J. Ghan and Richard

1025 C. Easter for insightful discussions, and Shuai Zhu, a former postdoc researcher of the air quality

1026 forecasting laboratory at NCSU for early work on the incorporation of CB05_GE and its

1027 coupling with MAM3. The authors would also like to thank the four reviewers for their valuable

1028 suggestions that help improve the technical quality of this work. MODIS data and CERES data

1029 are provided by NASA via http://ladsweb.nascom.nasa.gov and

1030 http://ceres.larc.nasa.gov/order_data.php, respectively. Other surface network data were

1031 downloaded from their respective web sites. We would like to acknowledge high-performance

1032 computing support from Yellowstone (ark:/85065/d7wd3xhc) provided by NCAR's

- 1033 Computational and Information Systems Laboratory, sponsored by the U.S. National Science
- 1034 Foundation.
- 1035
- 1036 References
- Abdul-Razzak, H. and Ghan, S. J.: A Parameterization of Aerosol Activation, Part 2: Multiple
 Aerosol Types, J. Geophys. Res., 105, 6837-6844, 2000.
- Adams, P. J. and Seinfeld, J. H.: Predicting global aerosol size distributions in general circulation
 models, J. Geophys. Res., 107(D19), 4370, doi:10.1029/2001JD001010, 2002.
- 1041 Appel, K.W., Pouliot, G.A., Simon, H., Sarwar, G., Pye, H. O. T., Napelenok, S. L., Akhtar, F.,
- and Roselle, S. J.: Evaluation of dust and trace metal estimates from the Community
- 1043 Multiscale Air Quality (CMAQ) model version 5.0, Geosci. Model Dev., 6, 883-899, 2013.
- Arstila, H., Korhonen, P., and Kulmala, M.: Ternary nucleation: Kinetics and application to
 water-ammonia-hydrochloric acid system, J. Aerosol Sci., 30, 131-138, doi:10.1016/S00218502(98)00033-0, 1999.
- Barth, M. C., Rasch, P. J., Kiehl, J. T., Benkovitz, C. M., and Schwartz, S. E.: Sulfur chemistry in
 the National Center for Atmospheric Research Community Climate Model: Description,
 evaluation, features and sensitivity to aqueous chemistry, J. Geophys. Res., 105, 1387-1415,
 2000.
- Berndt, T., Sipilä, M., Stratmann, F., Petäjä, T., Vanhanen, J., Mikkilä, J., Patokoski, J., Taipale,
 R., Lee Mauldin III, R., and Kulmala, M.: Enhancement of atmospheric H₂SO₄/H₂O
 nucleation: organic oxidation products versus amines, Atmos. Chem. Phys. Discuss., 13,
 16301-16335, doi:10.5194/acpd-13-16301-2013, 2013.
- Bennartz, R.: Global assessment of marine boundary layer cloud droplet number concentration
 from satellite, J. Geophys. Res., 112, D02201, doi: 10.1029/2006JD007547, 2007.
- Bey, I., Jacob, D. J., Yantosca, R. M., Logan, J. A., Field, B. D., Fiore, A. M., Li, Q., Lui, H. Y.,
 Mickley, L. J., and Schultz, M. G.: Global modeling of tropospheric chemistry with
 assimilated meteorology: Model description and evaluation, J. Geophys. Res., 106(D19),
 23073-23095, 2001.
- Bretherton, C. S. and Park, S.: A new moist turbulence parameterization in the community
 atmosphere model, J. Climate, 22, 3422-3448, 2009.
- Burkholder, J. B., Curtius, J., Ravishankara, A. R., and Lovejoy, E. R.: Laboratory studies of the
 homogeneous nucleation of iodine oxides, Atmos. Chem. Phys., 4, 19-34, 2004.
- Byun, D. W. and Schere, K. L.: Review of the governing equations, computational algorithms,
 and other components of the Models-3 Community Multiscale Air Quality (CMAQ)
 Modeling System, Appl. Mech. Rev., 59, 51-77, 2006.

- Cappa, C. D., Lovejoy, E. R., and Ravishankara, A. R.: Evidence for liquid-like and nonideal
 behavior of a mixture of organic aerosol components, P. Natl. Acad. Sci. USA, 105, 1868718691, doi:10.1073/pnas.0802144105, 2008.
- Collins, W. D., Bitz, C. M., Blackmon, M. L., Bonan, G. B., Bretherton, C. S., Carton, J. A.,
 Chang, P., Doney, S. C., Hack, J. J., Henderson, T. B., Kiehl, J. T., Large, W. G., McKenna, D.
 S., Santer, B. D., and Smith, R. D.: The Community Climate System Model version3
 (CCSM3), J. Clim., 19, 2122-2143, doi:10.1175/JCLI3761.1, 2006.
- Dunne, J. P., John, J. G., Adcroft, A. J., Griffies, S. M., Hallberg, R. W., Shevliakova, E.,
 Stouffer, R. J., Cooke, W., Dunne, K. A., Harrison, M. J., Krasting, J. P., Malyshev, S. L.,
 Milly, P. C. D., Phillipps, P. J., Sentman, L. T., Samuels, B. L., Spelman, M. J., Winton, M.,
 Wittenberg, A. T., and Zadeh, N.: GFDL's ESM2 global coupled climate-carbon earth system
 models. Part I: Physical formulation and baseline simulation characteristics, J. Climate, 25,
 6646-6665, 2012.
- Dunne, J. P., John, J. G., Shevliakova, E., Stouffer, R. J., Krasting, J. P., Malyshev, S. L., Milly,
 P. C. D., Sentman, L. T., Adcroft, A. J., Cooke, W., Dunne, K. A., Griffies, S. M., Hallberg, R.
 W., Harrison, M. J., Levy, H., Wittenberg, A. T., Phillips, P. J., and Zadeh, N.: GFDL's ESM2
 global coupled climate-carbon earth system models. Part II: Carbon system formulation and
 baseline simulation characteristics, J. Climate, 26, 2247-2267, 2013.
- Dutkiewicz, S., Sokolov, A. P., Scott, J., and Stone, P. H.: A Three-Dimensional Ocean-Seaice Carbon Cycle Model and its Coupling to a Two-Dimensional Atmospheric Model: Uses in
 Climate Change Studies. MIT JPSPGC Report 122, May, 47 p, 2005.

Emmons, L. K., Walters, S., Hess, P. G., Lamarque, J.-F., Pfister, G. G., Fillmore, D., Granier, C.,
Guenther, A., Kinnison, D., Laepple, T., Orlando, J., Tie, X., Tyndall, G., Wiedinmyer, C.,
Baughcum, S. L., and Kloster, S.: Description and evaluation of the Model for Ozone and
Related chemical Tracers, version 4 (MOZART-4), Geosci. Model Dev., 3, 43-67,
doi:10.5194/gmd-3-43-2010, 2010.

- ENVIRON: Comprehensive Air Quality Model with extensions User's Guide, Novato,
 California, USA, 5.3 edn., 2010.
- Faraji, M., Kimura, Y., McDonald-Buller, E., and Allen, D.: Comparison of the carbon bond and
 SAPRC photochemical mechanisms under conditions relevant to southeast Texas, Atmos.
 Environ., 42, 5821–5836, doi:10.1016/j.atmosenv.2007.07.048, 2008.
- Fast, J. D., Gustafson Jr., W. I., Easter, R. C., Zaveri, R. A., Barnard, J. C., Chapman, E. G.,
 Grell, G. A., and Peckham, S. E.: Evolution of ozone, particulates, and aerosol direct radiative
 forcing in the vicinity of Houston using a fully coupled meteorology-chemistry-aerosol
 model, J. Geophys. Res., 111, D21305, doi:10.1029/2005JD006721, 2006.
- 1103Fountoukis, C and Nenes, A.: ISORROPIA II: a computationally efficient thermodynamic1104equilibrium model for K^+ Ca^{2+} - Mg^{2+} - NH_4^+ - Na^+ - SO_4^{2-} - NO_3^- - Cl^- + H_2O aerosols, Atmos.1105Chem. Phys., 7, 4639–4659, doi:10.5194/acp-7-4639-2007, 2007.

- Fountoukis, C., Nenes1, A., Sullivan, A., Weber, R., Farmer, D., and Cohen, R. C., 2009,
 Thermodynamic characterization of Mexico City aerosol during MILAGRO 2006, Atmos.
 Chem. Phys., 9, 2141–2156, 2009.
- Gent, P. R., Yeager, S. G., Neale, R. B., Levis, S., and Bailey, D. A.: Improvements in a half degree atmosphere/land version of the CCSM, Climate Dyn., 34, 819-833,
- 1111 doi:10.1007/s00382-009-0614-8, 2010.
- Gettelman, A., Morrison, H., and Ghan, S. J.: A new two-moment bulk stratiform cloud
 microphysics scheme in the community atmosphere model, version 3 (CAM3). Part II:
 Single-column and global results, J. Climate, 21(15), 3660-3679, 2008.
- Ghan, S. J., and Zaveri, R. A.: Parameterization of optical properties for hydrated internally
 mixed aerosol, J. Geophys. Res., 112, D10201, doi:10.1029/2006JD007927,2007.
- Ghan, S. J., Liu, X., Easter, R. C., Zaveri, R., Rasch, P. J., and Yoon, J.-H.: Toward a minimal
 representation of aerosols in climate models: comparative decomposition of aerosol direct,
 semidirect, and indirect radiative forcing, J. Climate, 25, 6461-6476, 2012.
- Guenther A., Karl, T., Harley, P., Wiedinmyer, C., Palmer, P. I., and Geron, C.: Estimates of
 global terrestrial isoprene emissions using MEGAN (Model of Emissions of Gases and
 Aerosols from Nature), Atmos. Chem. Phys., 6, 3181-3210, 2006.
- Heald, C. L., Henze, D. K., Horowitz, L. W., Feddema, J., Lamarque, J.-F., Guenther, A., Hess, P.
 G., Vitt, F., Seinfeld, J. H., Goldstein, A. H., and Fung, I.: Predicted change in global
 secondary organic aerosol concentrations in response to future climate, emissions, and land
 use change, J. Geophys. Res., 113, D05211, doi:10.1029/2007JD009092, 2008.
- 1127 Heintzenberg, J.: Fine particles in the global troposphere, A review. Tellus, 41B, 149-160, 1989.
- Hoffmann, T., O'Dowd, C. D., and Seinfeld, J. H.: Iodine oxide homogeneous nucleation: An
 explanation for coastal new particle production, Geophys. Res. Lett., 28(10), 1949-1952,
 2001.
- Horowitz, L. W.: Past, present, and future concentrations of tropospheric ozone and aerosols:
 methodology, ozone evaluation, and sensitivity to aerosol wet removal, J. Geophys. Res., 111,
 D22211, doi:10.1029/2005JD006937, 2006.
- Iacono, M. J., Delamere, J. S., Mlawer, E. J., and Clough, S. A.: Evaluation of upper tropospheric
 water vapor in the NCAR Community Climate Model (CCM3) using modeled and observed
 HIRS radiances, J. Geophys. Res., 108(D2), 4037, doi:10.1029/2002jd002539, 2003.
- Iacono, M. J., Delamere, J. S., Mlawer, E. J., Shephard, M. W., Clough S. A., and Collins, W. D.:
 Radiative forcing by long-lived greenhouse gases: Calculations with the AER radiative
 transfer models, J. Geophys. Res., 113(D13), D13103, doi:10.1029/2008jd009944, 2008.
- Jacobson, M. Z.: Studying the effect of calcium and magnesium on size-distributed nitrate and
 ammonium with EQUISOLV II, Atmos. Environ., 33, 3635-3649, 1999.

- Jacobson, M. Z.: Short-term effects of Controlling Fossil-Fuel Soot, Biofuel Soot and Gases, and
 Methane on Climate, Arctic Ice, and Air Pollution Health, J. Geophys. Res., 115, D14209,
 doi:10.1029/2009JD013795, 2010.
- Karamchandani, P., Zhang, Y., Chen, S.-Y., and Balmori-Bronson, R.: Development of an
 extended chemical mechanism for global-through-urban applications, Atmospheric Pollution
- 1147 Research, 3, 1-24, 2012.
- Kim, Y., Sartelet, K., and Seigneur, C.: Formation of secondary aerosols: impact of the gas-phase
 chemical mechanism, Atmos. Chem. Phys., 11, 583-598, doi:10.5194/acp-11-583-2011,
 2011a.
- Koloutsou-Vakakis S., Rood, M. J., Nenes, A., and Pilinis, C.: Modeling of aerosol properties
 related to direct climate forcing, J. Geophys. Res., 103(D14), 17009-17032, doi:
 10.1029/98JD00068, 1998.
- Kuang, C., McMurry, P. H., McCormick, A. V., and Eisele, F. L.: Dependence of nucleation rates
 on sulfuric acid vapor concentration in diverse atmospheric locations, J. Geophys. Res., 113,
 D10209, doi:10.1029/2007JD009253, 2008.
- Kuang, C., McMurry, P. H., and McCormick, A. V.: Determination of cloud condensation nuclei
 production from measured new particle formation events, Geophys. Res. Lett., 36, L09822,
 doi:10.1029/2009GL037584, 2009.
- Kulmala, M., Lehtinen, K., and Laaksonen, A.:, Cluster activation theory as an explanation of the
 linear dependence between formation rate of 3 nm particles and sulphuric acid concentration,
 Atmos. Chem. Phys., 6, 787-793, doi:10.5194/acp-6-787-2006, 2006.
- Kulmala, M., Vehkamaki, H., Petaja, T., Dal Maso, M., Lauri, A., Kerminen, V.-M., Birmili, W.,
 and McMurry, P.: Formation and growth rates of ultrafine atmospheric particles: A review of
 observations, J. Aerosol Sci., 35, 143-176, 2004.
- Kurtén, T., Loukonen, V., Vehkamaki, H., and Kulmala, M.: Amines are likely to enhance neutral
 and ion-induced sulfuric acid-water nucleation in the atmosphere more effectively than
 ammonia, Atmos. Chem. Phys., 8, 4095-4103, doi:10.5194/acp-8-4095-2008, 2008.
- Lamarque, J.-F., Kiehl, J. T., Hess, P. G., Collins, W. D., Emmons, L. K., Ginoux, P., Luo, C., and
 Tie, X. X.: Response of a coupled chemistry-climate model to changes in aerosol emissions:
 global impact on the hydrological cycle and the tropospheric burdens of OH, ozone, and NOx,
 J. Geophys. Res., 32, L16809, doi:10.1029/2005GL023419, 2005.
- Lamarque, J. F., Shinedell, D. T., Josse, B., Young, P. J., Cionni, I., Eyring, V., Bergmann, D.,
 Cameron-Smith, P., Collins, W. J., Doherty, R., Dalsoren, S., Faluvegi, G., Folberth, G., Ghan,
 S. J., Hiriwutz, L. W., Lee, Y. H., MacKenzie, I. A., Nagashima, T., Naik, V., Plummer, D.,
 Righi, M., Rumbold, S. T., Schulz, M., Skeie, R. B., Stevenson, D. S., Strode, S., Sudo, K.,
 Szopa, S., Voulgarakis, A., and Zeng, G.: The atmospheric chemistry and climate model
- 1178 intercomparison project: overview and description of models, simulations and climate
- 1179 diagnostics, Geosci. Model Dev., 6, 179-206, doi:10.5194/gmd-6-179-2013, 2013.

- Lamarque, J. F., Emmons, L. K., Hess, P. G., Kinnison, D. E., Tilmes, S., Vitt, F., Heald, C. L.,
 Holland, E. A., Lauritzen, P. H., Neu, J., Orlando, J. J., Rasch, P. J., and Tyndall, G. K.:CAMchem: description and evaluation of interactive atmospheric chemistry in CESM, Geosci.
 Model Dev., 5, 369-411, doi:10.5194/gmd-5-369-2012, 2012.
- 1184 Lawrence, D. M., Oleson, K. W., Flanner, M. G., Thornton, P. E., Swenson, S. C., Lawrence, P.
- 1185 J., Zeng, X., Yang, Z.-L., Levis, S., Sakaguchi, K., Bonan, G. B., and Slater, A. G.:
- Parameterization improvements and functional and structural advances in version 4 of the Community Land Model, J. Adv. Model. Earth Syst., 3, Art. 2011MS000045, 27 pp, doi:
- 1188 10.1029/2011MS000045, 2011.
- Liao, H., Adams, P. J., Chung, S. H., Seinfeld, J. H., Mickley, L. J., and Jacob, D. J.: Interactions
 between tropospheric chemistry and aerosols in a unified general circulation model, J.
 Geophys. Res., 108(D1), 4001, doi:10.1029/2001JD001260, 2003.
- Liu, X., Easter, R. C., Ghan, S. J., Zaveri, R., Rasch, P., Shi, X., Lamarque, J.-F., Gettleman, A.,
 Morrison, H., Vitt, F., Conley, A., Park, S., Neale, R., Hannay, C., Ekman, A. M. L., Hess, P.,
 Mahowald, N., Collins, W., Iacono, M.J., Bretherton, C. S., Flanner, M. G., and Mitchell,
 D.L.: Toward a minimal representation of aerosols in climate models: description and
 evaluation in the Community Atmosphere Model CAM5, Geosci. Model Dev., 5, 709-739,
- 1196 evaluation in the Community Atmosphere Model CAM5, Geosci. Model Dev., 5, 709-759, 1197 2012.
- Lohmann, U., Stier, P., Hoose, C., Ferrachat, S., Kloster, S., Roeckner, E., and Zhang, J.: Cloud
 microphysics and aerosol indirect effects in the global climate model ECHAM5-HAM,
 Atmos. Chem. Phys., 7, 3425-3446, 2007.
- Luecken, D. J., Phillips, S., Sarwar, G., and Jang, C.: Effects of using the CB05 vs. SAPRC99 vs.
 CB4 chemical mechanism on model predictions: Ozone and gas-phase photochemical
 precursor concentrations, Atmos. Environ., 42, 5805–5820,
 doi:10.1016/j.atmosenv.2007.08.056, 2008.
- Marsh, A. R. W., and McElory, W. J.: The dissociation constant and Henry's law constant of HCl
 in aqueous solution, Atmos. Environ., 19, 1075-1080, 1985.
- Martensson, E. M., Nilsson, E. D., deLeeuw, G., Cohen, L. H., and Hansson, H. C.: Laboratory
 simulations and parameterization of the primary marine aerosol production, J. Geophys. Res.,
 108(D9), 4297, doi:10.1029/2002JD002263, 2003.
- Meng, Z. and Seinfeld, J. H.: Time scales to achieve atmospheric gas-aerosol equilibrium for
 volatile species, Atmos. Environ., 30, 2889-2900, 1996.
- Meng, Z, Dabdub, D., and Seinfeld, J.H.: Size- and chemically-resolved model of atmospheric
 aerosol dynamics. J. Geophys. Res., 103, 3419-3435, 1998.

 Merikanto, J., Napari, I., Vehkamaki, H., Anttila, T., and Kulmala, M.: New parameterization of sulfuric acid-ammonia-water ternary nucleation rates at tropospheric conditions, J. Geophys.
 Res., 112, D15207, doi: 10.1029/2006JD007977, 2007.

- Metzger, S. M., Dentener, F. J., Lelieveld, J., and Pandis, S. N.: Gas/aerosol partitioning I: A
 computationally efficient model, J. Geophys. Res., 107(D16), 4312, doi:
 10.1029/2001JD001102, 2002.
- Metzger, S. and Lelieveld, J.: Reformulating atmospheric aerosol thermodynamics and
 hygroscopic growth into fog, haze and clouds, Atmos. Chem. Phys., 7, 3163-3193, doi:
 10.5194/acp-7-3163-2007, 2007.
- Metzger, S., Steil, B., Xu, L., Penner, J. E., and Lelieveld, J.: New representation of water
 activity based on a single solute specific constant to parameterize the hygroscopic growth of
 aerosols in atmospheric models, Atmos. Chem. Phys., 12, 5429-5446, doi:10.5194/acp-125429-2012, 2012.
- Mlawer, E. J., Taubman, S. J., Brown, P. D., Iacono, M. J., and Clough, S. A.: Radiative transfer
 for inhomogeneous atmospheres: RRTM, a validated correlated-k model for the longwave, J.
 Geophys. Res., 102(D14), 16663-16682, 1997.
- Monier, E., Scott, J. R., Sokolov, A. P., Forest, C. E., and Schlosser, C. A.: An integrated
 assessment modeling framework for uncertainty studies in global and regional climate
 change: the MIT IGSM-CAM (version 1.0), Geosci. Model Dev. Discuss., 6, 2213-2248,
 2013.
- Morrison, H. and Gettelman, A.: A new two-moment bulk stratiform cloud microphysics scheme
 in the community atmosphere model, version 3 (CAM3). Part I: Description and numerical
 tests, J. Climate, 21(15), 3642-3659, 2008.
- Nenes, A., Pandis, S.N., and Pilinis, C.: ISORROPIA: A new thermodynamic equilibrium model
 for multiphase multicomponent inorganic aerosols, Aquatic Geochemistry, 4, 123-152, 1998.
- O'Dowd, C. D., Jimenez, J. L., Bahreini, R., Flagan, R. C., Seinfeld, J. H., Hämeri, K., Pirjola,
 L., Kulmala, M., Jennings, S. G., and Hoffmann, T.: Marine aerosol formation from biogenic
 iodine emissions, Nature, 417, 632-636, doi:10.1038/nature00775, 2002.
- Olerud, D. and Sims, A. (2004). MM5 2002 Modeling in Support of VISTAS (Visibility
 Improvement State and Tribal Association of the Southeast), Report, Baron Advanced
 Meteorological Systems, LLC, Raleigh, NC, August.
- Park, S. and Bretherton, C. S.: The university of Washington shallow convection and moist
 turbulence schemes and their impact on climate simulations with the community atmosphere
 model, J. Climate, 22, 3449-3469, 2009.
- Pechtl, S., Lovejoy, E. R., Burkholder, J. B., and von Glasow, R.: Modeling the possible role
 iodine oxides in atmospheric new particle formation, Atmos. Chem. Phys., 6, 505-523, 2006.
- Pierce, J. R. and Adams, P. J.: Uncertainty in global CCN concentrations from uncertain aerosol
 nucleation and primary emission rates, Atmos. Chem. Phys., 9, 1339-1356, 2009.
- Raes, F., Augustin, J., and Vandingenen, R.: The role of ion-induced aerosol formation in the
 lower atmosphere, J. Aerosol Sci., 17, 466-470, doi: 10.1016/0021-8502(86)90135-7, 1986.

- 1254 Reiter, R.: Phenomena in atmospheric and environmental electricity, Elsevier, New York, 1992.
- Roeckner, E., Bauml, G., Bonaventura, L., Brokopf, R., Esch, M., Giorgetta, M., Hagemann, S.,
 Kirchner, I., Kornblueh, L., Manzini, E., Rhodin, A., Schlese, U., Schulzweida, U., and
 Tompkins, A.: The atmospheric general circulation model ECHAM 5. PART I: model
 description, MPI Technical Report 349, Max Planck Institute for Meteorology, Hamburg,
 Germany, 2003.
- Roeckner, E., Brokopf, R., Esch, M., Giorgetta, M. A., Hagemann, S., Kornblueh, L., Manzini,
 E., Schlese, U., and Schulzweida, U. : Sensitivity of Simulated Climate to Horizontal and
 Vertical Resolution in the ECHAM5 Atmosphere Model, Journal of Climate, 19, 3771-3791,
 2006a.
- Sander S. P., Friedl, R. R., Golden, D. M., Kurylo, M. J., Huie, R. E., Orkin, V. L., Moortgat, G.
 K., Ravishankara, A. R., Kolb, C. E., Molina, M. J., and Finlayson-Pitts, B. J.: Chemical
 Kinetics and Photochemical Data for Use in Atmospheric Studies, National Aeronautics and
 Space Administration, Jet Propulsion Laboratory California Institute of Technology Pasadena,
 California, 2003.
- Sarwar, G., Luecken, D., Yarwood, G., Whitten, G., and Carter, W. P. L.;, Impact of an updated
 carbon bond mechanism on predictions from the Community Multiscale Air Quality Model, J.
 Appl. Meteorol. Climatol., 47, 3-14, doi:10.1175/2007JAMC1393.1, 2008.
- Schwartz, S. E.: Gas- and aqueous-phase chemistry of HO₂ in liquid water clouds, J. Geophys.
 Res., 89, 11589-11598, 1984.
- Seinfeld, J. H. and Pandis, S. N.: Atmospheric chemistry and physics: From air pollution to climate change, 2 ed., John Wiley & Sons, Inc, 2006.
- Shindell, D. T., Lamarque, J.-F., Schulz, M., Flanner, M., Jiao, C., Chin, M., Young, P. J., Lee, Y.
 H., Rotstayn, L., Mahowald, N., Milly, G., Faluvegi, G., Balkanski, Y., Collins, W. J., Conley,
 A. J., Dalsoren, S., Easter, R., Ghan, S., Horowitz, L., Liu, X., Myhre, G., Nagashima, T.,
 Naik, V., Rumbold, S. T., Skeie, R., Sudo, K., Szopa, S., Takemura, T., Voulgarakis, A., Yoon,
 J.-H., and Lo, F.:, Radiative forcing in the ACCMIP historical and future climate simulations,
 Atmos. Chem. Phys., 13, 2939-2974, doi:10.5194/acp-13-2939-2013, 2013.
- Sihto, S. L., Kulmala, M., Kerminen, V. M., Maso, M. D., Petaja, T., Riipinen, I., Korhonen, H.,
 Arnold, F., Janson, R., Boy, M., Laaksonen, A., and Lehtinen, K. E. J.: Atmospheric sulphuric
 acid and aerosol formation: implications from atmospheric measurements for nucleation and
 early growth mechanisms, Atmospheric Chemistry and Physics, 6, 4079-4091, 2006.
- Sokolov, A. P., Schlosser, C. A., Dutkiewicz, S., Paltsev, S., Kicklighter, D., Jacoby, H. D., Prinn,
 R. G., Forest, C. E., Reilly, J. M., Wang, C., Felzer, B., Sarofim, M. C., Scott, J., Stone, P. H.,
 Melillo, J. M., and Cohen, J.: The MIT Integrated Global System Model (IGSM) Version 2:
 Model Description and Baseline Evaluation. MIT JPSPGC Report 124, July, 40 p, 2005.
- Spracklen, D. V., Carslaw, K. S., Kulmala, M., Kerminen, V.-M., Mann, G. W., and Sihto, S.-L.:
 The contribution of boundary layer nucleation events to total particle concentrations on
 regional and global scales, Atmos. Chem. Phys., 6, 5631-5648, 2006.

- Stephens, G. L.: Cloud feedbacks in the climate system: A critical review, J. Climate, 18(2), 237273, 2005.
- Stier, P., Feichter, J., Kinne, S., Kloster, S., Vignati, E., Wilson, J., Ganzeveld, L., Tegen, I.,
 Werner, M., Balkanski, Y., Schulz, M., Boucher, O., Minikin, A., and Petzold, A.: The aerosolclimate model ECHAM5-HAM, Atmos. Chem. Phys., 5, 1125-1156, doi:10.5194/acp-5-11252005, 2005.
- Textor, C., Schulz, M., Guibert, S., Kinne, S., Balkanski, Y., Bauer, S., Berntsen, T., Berglen, T.,
 Boucher, O., Chin, M., Dentener, F., Diehl, T., Easter, R., Feichter, H., Fillmore, D., Ghan, S.,
 Ginoux, P., Gong, S., Grini, A., Hendricks, J., Horowitz, L., Huang, P., Isaksen, I., Iversen, I.,
 Kloster, S., Koch, D., Kirkevag, A., Kristjansson, J. E., Krol, M., Lauer, A., Lamarque, J. F.,
 Liu, X., Montanaro, V., Myhre, G., Penner, J., Pitari, G., Reddy, S., Seland, Ø., Stier, P.,
 Takemura, T., and Tie, X.: Analysis and quantification of the diversities of aerosol life cycles
 within AeroCom, Atmos. Chem. Phys., 6, 1777–1813, doi:10.5194/acp-6-1777-2006, 2006.
- Tsigaridis. K., Krol, M., Dentener, F. J., Balkanski, Y., Lathiere, J., Metzger, S., Hauglustaine, D.
 A., and Kanakidou, M.: Change in global aerosol composition since preindustrial times,
 Atmos. Chem. Phys., 6, 5143-5162, 2006.
- Usoskin, I. G. and Kovaltsov, G. A.: Cosmic ray induced ionization in the atmosphere: full
 modeling and practical applications, J. Geophys. Res., 111, D21206,
 doi:10.1029/2006JD007150, 2006.
- van Dingenen, R. and Raes, F.: Ternary nucleation of methane sulphonic acid, sulphuric acid and
 water vapour, J. Aerosol Sci., 24, 1-17, doi:10.1016/0021-8502(93)90081-J, 1993.
- Van Pelt, R. S. and Zobeck, T. M.: Chemical constituents of fugitive dust, Environ. Monit.
 Assess., 130, 3-16, doi:10.1007/s10661-006-9446-8, 2007.
- Vehkamaki, H., Kulmala, M., Napari, I., Lehtinen, K. E. J., Timmreck., C., Noppel, M., and
 Laaksonen, A.: an improved parameterization for sulfuric acid-water nucleation rates for
 tropospheric and stratospheric conditions, Journal of Geophysical Research-Atmospheres,
 107(D22), 4622, doi:10.1029/2002JD002184, 2002.
- Wang, M. and Penner, J. E.: Aerosol indirect forcing in a global model with particle nucleation,
 Atmos. Chem. Phys., 9, 239-260, 2009.
- Wang, K., Zhang, Y., Nenes, A., and Fountoukis, C.: Implementation of dust emission and
 chemistry into the Community Multiscale Air Quality modeling system and initial application
 to an Asian dust storm episode, Atmos. Chem. Phys., 12, 10209-10237, doi:10.5194/acp-1210209-2012, 2012.
- Wexler, A. S. and Seinfeld, J. H.: Second-generation inorganic aerosol model, Atmos. Environ.,
 25A, 2731-2748, 1991.
- Williams, J. E., Scheele, M. P., van Velthoven, P. F. J., Cammas, J.-P., Thouret, V. Galy-Lacaux,
 C., and Volz-Thomas, A.: The influence of biogenic emissions from Africa on tropical

tropospheric ozone during 2006: a global modeling study, Atmos. Chem. Phys., 9, 5729-5749,2009.

Young, P. J., Archibald, A. T., Bowman, K. W., Lamarque, J.-F., Naik, V., Stevenson, D. S.,
Tilmes, S., Voulgarakis, A., Wild, O., Bergmann, D., Cameron-Smith, P., Cionni, I., Collins,

- 1334 W. J., Dalsøren, S. B., Doherty, R. M., Eyring, V., Faluvegi, G., Horowitz, L. W., Josse, B.,
- 1335 Lee, Y. H., MacKenzie, I. A., Nagashima, T., Plummer, D. A., Righi, M., Rumbold, S. T.,
- 1336 Skeie, R. B., Shindell, D. T., Strode, S. A., Sudo, K., Szopa, S., and Zeng, G.: Pre-industrial to
- end 21st century projections of tropospheric ozone from the Atmospheric Chemistry and
- Climate Model Intercomparison Project (ACCMIP), Atmos. Chem. Phys., 13, 2063-2090,
 doi:10.5194/acp-13-2063-2013, 2013.
- Yu, F.: From molecular clusters to nanoparticles: Second generation ion-mediated nucleation
 model, Atmos. Chem. Phys., 6, 5193-5211, 2006.
- Yu, F.: Ion-mediated nucleation in the atmosphere: Key controlling parameters, implications, and
 look-up table, J. Geophys. Res., 115, D03206, doi: 10.1029/2009JD012630, 2010.

Yu, F. and Luo, G.: Simulation of particle size distribution with a global aerosol model:
Contribution of nucleation to aerosol and CCN number concentrations, Atmos. Chem. Phys.,
9, 7691-7710, 2009.

- Yu, F., Luo, G., Liu, X., Easter, R. C., Ma, X., and Ghan, S. J.: Indirect radiative forcing by ionmediated nucleation of aerosol, Atmos. Chem. Phys., 12, 11451-11463, 2012.
- Yu, F., Luo, G., Bates, T. S., Anderson, B., Clarke, A., Kapustin, V., Yantosca, R. M., Wang, Y.,
 and Wu, S.: Spatial distributions of particle number concentrations in the global troposphere:
 simulations, observations, and implications for nucleation mechanisms, J. Geophys., Res.,
 115, D17205, doi:10.1029/2009JD013473, 2010.
- Yu, F. and Turco, R. P.: Ultrafine aerosol formation via ion-mediated nucleation, Geophys. Res.
 Lett., 27, 883-886, doi: 10.1029/1999GL011151, 2000.
- Yu, F. and Turco, R. P.: From molecular clusters to nanoparticles: The role of ambient ionization
 in tropospheric aerosol formation, J. Geophys., Res., 106, 4797-4814, doi:
 10.1029/2000JD900539, 2001.
- Yu, F., Wang, Z., Luo, G., and Turco, R. P.: Ion-mediated nucleation as an important global source of tropospheric aerosols, Atmos. Chem. Phys., 8, 2537-2554, 2008.
- Zaveri, R. A., Easter, R. C., and Peters, L. K.: A computationally efficient multicomponent
 equilibrium solver for aerosols (MESA), J. Geophys. Res., 110, D24203, doi:
 10.1029/2004JD005618, 2005.
- Zender, C. S., Bian, H., and Newman, D.: The mineral Dust Entrainment And Deposition
 (DEAD) model: Description and 1990s dust climatology, J. Geophys. Res., 108(D14), 4416,
 doi: 10.1029/2002JD002775, 2003.

- Zhang, G. J., and McFarlane, N. A.: Sensitivity of climate simulations to the parameterization of
 cumulus convection in the Canadian Climate Centre general circulation model, Atmosphere Ocean, 33, 407-446, 1995.
- Zhang, Y., Bischof, C. H., Easter, R. C., and Wu, P.-T.: Sensitivity analysis of a mixed-phase
 chemical mechanism using automatic differentiation, J. Geophys. Res., 103 (D15), 18,95318,979, 1998.
- Zhang, Y., Seigneur, C., Seinfeld, J. H., Jacobson, M., Clegg, S. L., Binkowski, F. S.: A
 comparative review of inorganic aerosol thermodynamic equilibrium modules: similarities,
 differences, and their likely causes, Atmos. Environ., 34, 117-137, 2000.
- Zhang, Y., McMurry, P. H., Yu, F., and Jacobson, M. Z.: A comparative study of nucleation
 parameterizations: 1. Examination and evaluation of the formulations, J. Geophys. Res., 115,
 D20212, doi: 10.1029/2010JD014150, 2010.
- Zhang, Y., Chen, Y., Sarwar, G., and Schere, K.: Impacts of gas-phase mechanisms on weather
 research forecasting model with chemistry (WRF/Chem) predictions: Mechanism
 implementation and comparative evaluation, J. Geophys. Res., 117, D01301, doi:
 10.1029/2011JD015775, 2012a.
- Zhang, Y., Karamchandani, P., Glotfelty, T., Street, D. G., Grell, G., Nenes, A., Yu, F., and
 Bennartz, R.: Development and initial application of the global-through-urban weather
 research and forecasting model with chemistry (GU-WRF/Chem), J. Geophys. Res., 117,
 D20206, doi:10.1029/2012JD017966, 2012b.
- Zuend, A., Marcolli, C., Peter, T., and Seinfeld, J. H.: Computation of liquid-liquid equilibria and
 phase stabilities: implications for RH-dependent gas/particle partitioning of organic-inorganic
 aerosols, Atmos. Chem. Phys., 10, 7795-7820, doi:10.5194/acp-10-7795-2010, 2010.

	Table 1. Simulation design an	d purposes
Run Index	Model Configuration	Purpose
MAM_SIM	Simple gas-phase chemistry coupled with default MAM7	A baseline run for the 1 st set of simulations (see text)
MAM_CB05_GE	CB05_GE coupled with default MAM7	Differences of MAM_SIM and MAM_CB05_GE indicate the impacts of gas-phase chemical mechanisms
MAM_CON	Same as MAM_CB05_GE, but with explicit treatments for NO ₃ ⁻ , Cl ⁻ , and Na ⁺ ; HNO ₃ and HCl condensation and aqueous-phase chemistry; species-dependent accommodation coefficients	A baseline run for the 2 nd set of simulations; differences of MAM_SIM and MAM_CB05_GE indicate the impact of modified condensation and aqueous-phase chemistry treatments
MAM_CON/IMN	Same as MAM7_CON, but combine IMN with modified default nucleation parameterizations with a prefactor of 1.0×10^{-8}	Differences of MAM_CON and MAM_CON/IMN indicate the impacts of IMN and the lower prefactor for WP09
MAM_CON/ISO	Same as MAM7_CON, but with ISORROPIA II for aerosol thermodynamics under metastable conditions	Differences between MAM_CON and MAM_IMN/ISO indicate the impacts of explicit aerosol thermodynamics
MAM_NEWA	Same as MAM7_CON, but with all modified and new treatments and using a prefactor of 1.0×10^{-9} for default nucleation parameterization	Differences between MAM_CB05_GE and MAM_NEWA indicate the impacts of all new and modified treatments for inorganic aerosols
MAM_NEWB	Same as MAM_NEWA, but with ISORROPIA II under stable condition	Differences between MAM_NEWA and MAM_NEWB indicate the impacts of thermodynamic conditions on gas-aerosol partitioning
MAM_NEW/EMIS	Same as MAM7_NEW, but with adjusted emissions of SO ₂ , NH ₃ , BC, POM, and CO over CONUS, Europe, and East Asia	Differences between MAM_NEWA and MAM_NEW/EMIS indicate the impact of emissions
MAM_SIM_5Y	Same as MAM_SIM, but with prescribed SST for 2001-2005	A baseline run for 4 th set of simulations
MAM_NEW_5YA	Same as MAM_NEW/EMIS, but with prescribed SST for 2001- 2005	Differences between MAM_SIM_5Y and MAM_NEW_5YA indicate the indicate the impacts of all new and modified treatments for inorganic aerosols
MAM_NEW_5YB	Same as MAM_NEW/EMIS, but with fully-coupled model for 2001-2005	Difference between MAM_NEW_5YB and MAM_NEW_5YA indicate the impacts of processes from component models in the fully- coupled Earth system
	56	

Table 2. Datasets for	model evaluation
Species/Variables	Dataset
Downwelling longwave radiation (LWD)	BSRN
Downwelling shortwave radiation (SWD)	BSRN
Outgoing longwave radiation (OLR)	NOAA/CDC
Cloud fraction (CF)	MODIS
Cloud optical thickness (COT)	MODIS
Cloud water path (CWP)	MODIS
Precipitating water vapor (PWV)	MODIS
Aerosol optical depth (AOD)	MODIS
Column cloud condensation nuclei (ocean) at $S = 0.5\%$ (CCN5)	MODIS
Cloud droplet number concentration (CDNC)	BE07
Shortwave cloud radiative forcing (SWCF)	CERES
	Europe: EMEP
Carbon monoxide (CO)	East Asia: NIES of Japan, TAQMN
	CONUS: CASTNET
Ozone (O_3)	Europe: Airbase, BDQA, EMEP
	East Asia: TAQMN
	CONUS: CASTNET
Sulfur dioxide (SO ₂)	Europe: Airbase, BDQA, EMEP
	East Asia: MEP of China, NIES of Japan, TAQMN
	CONUS: CASTNET
Nitric acid (HNO ₃)	Europe: EMEP
Ammonia (NH ₃)	Europe: Airbase, EMEP
	Europe: Airbase, BDQA, EMEP
Nitrogen dioxide (NO ₂)	East Asia: NIES of Japan, TAQMN
$(2,1)(4,1)(2,0)^{2}$	CONUS: CASTNET, IMPROVE, STN
Sulfate (SO_4^{2-})	Europe: Airbase, EMEP
· · · · · · · · · · · · · · · · · · ·	CONUS: CASTNET, IMPROVE, STN
Ammonium (NH_4^+)	Europe: Airbase, EMEP
	CONUS: CASTNET, IMPROVE, STN
Nitrate (NO_3^-)	Europe: Airbase, EMEP
Chlorida (Cl ⁻)	CONUS: IMPROVE
Chloride (Cl ⁻)	Europe: Airbase, EMEP
Organic carbon (OC), Black carbon (BC), Total carbon (TC)	CONUS: IMPROVE, STN
Particulate matter with diameter less than 2.5 µm	CONUS: IMPROVE, STN
(PM _{2.5})	Europe: BDQA, EMEP
(2.5)	Europe: Airbase, BDQA, EMEP
Particulate matter with diameter less than 10 μ m (PM ₁₀)	East Asia: MEP of China, NIES of Japan, TAQMN
Column CO	Globe: MOPITT
Column NO ₂	Globe: GOME
Tropospheric ozone residual (TOR)	Globe: TOMS/SBUV
New particle formation rate (J)	
New particle formation rate (J)	Globe: Kulmala et al. (2004); Yu et al. (2008)

Table 2. Datasets for model evaluation

BSRN: Baseline Surface Radiation Network; NOAA/CDC: National Oceanic and Atmospheric Administration Climate Diagnostics Center; MODIS: Moderate Resolution Imaging Spectroradiometer; BE07: Bennartz, 2007; CERES: Clouds and Earth's Radiant Energy System; TOMS/SBUV: the Total Ozone Mapping Spectrometer/the Solar Backscatter UltraViolet; MOPITT: the Measurements Of Pollution In The Troposphere; GOME: Global Ozone Monitoring Experiment; CASTNET: Clean Air Status and Trends Network; IMPROVE: Interagency Monitoring of Protected Visual Environments; STN: Speciation Trends Network; EMEP: European Monitoring and Evaluation Program; BDQA: Base de Données sur la Qualité de l'Air; AirBase: European air quality database; MEP of China: Ministry of Environmental Protection of China; TAQMN: Taiwan Air Quality Monitoring Network; NIES of Japan: National Institute for Environmental Studies of Japan.

, , , ,									
Species/Variables	Dataset	Obs.	MAM SIM_	MAM_ CB05_GE	MAM_ CON_	MAM_ CON/IMN	MAM_ CON/ISO	MAM_ NEWA_	MAM_ NEW/EMIS
I W/D /W/ ⁻² /a	DON	3175	309.2/	309.6/	308.4/	308.0/	308.3/	308.7/	309.1/
	NNCO	0.710	-3.4/-1.1 ^c	-2.9/-0.9	-4.2/-1.3	-4.5/-1.4	-4.2/-1.3	-3.8/-1.2	-3.5/-1.1
SWD (W m-2)b	BCDN	C 101	179.2/	177.0/	169.4/	170.2/	177.3/	174.5/	177.0/
		101.2	-2.0/-1.1	-4.2/-2.3	-11.8/-6.5	-11.0/-6.1	-3.9/-2.2	-6.8/-3.7	-4.2/-2.3
OLP (W ²)	-NOAA-	ν ν ι C	223.2/	222.4/	219.3/	219.3/	220.7/	221.2/	221.2/
OLIN (W III)	CDC	<i>L</i> 14.4	8.8/4.1	8.1/3.8	4.9/2.3	4.9/2.3	6.2/2.9	6.9/3.2	6.9/3.2
CUICE (III2)	CEDES	0.11	-37.8/	-38.4/	-43.2/	-43.3/	-40.4/	-40.7/	-40.5/
ower (w III)	CENES	-41.0	3.2/7.9	2.7/6.5	-2.2/-5.3	2.3/-5.6	-0.7/1.6	-0.4/0.9	-0.6/1.4
CE (07)	SICOM	0 77	65.6/	65.9/	67.5/	67.6/	66.4/	66.5/	66.6/
	CITOM	6.00	-1.4/-2.0	-1.0/-1.5	0.5/0.8	0.7/1.0	-0.5/-0.8	-0.4/-0.6	-0.3/-0.5
TOD	SICON		6.9/	7.1/	8.7/	8.8/	7.T/	1.7/	7.7/
701	CIUUM	1./1	-10.2/-59.5	-10.1/-58.8	-8.4/-49.2	-8.3/-48.4	-9.4/-55.1	-9.4/-54.9	-9.4/-55.2
			33.0/	33.5/	42.3/	42.7/	36.4/	36.5/	36.2/
$CWP (g m^{-2})$	MODIS	148.1	-115.1/	-114.7/	-105.8/	-105.4/	-111.7/	-111.7/	-111.9/
			-77.7	-77.4	-71.4	-71.2	-75.4	-75.4	-75.5
			1.9/	1.9/	1.9/	1.9/	1.9/	1.9/	1.9/
PWV (cm)	MODIS	1.9	-2.5×10^{-2}	-1.8×10^{-2}	-3.3×10 ⁻² /	-3.9×10 ⁻² /	-1.8×10^{-2}	-1.4×10^{-2}	$-1.2 \times 10^{-2}/$
			-1.3	6.0-	-1.7	-2.0	6.0-	-0.7	9.0-
			9.8×10^{-2}	1.0×10^{-1}	1.2×10^{-1}	1.3×10^{-1}	1.0×10^{-1}	1.0×10^{-1}	1.0×10^{-1}
AOD	MODIS	1.5×10^{-1}	-5.5×10^{-2}	-5.2×10^{-2}	-3.0×10 ⁻² /	-2.6×10^{-2}	-5.3×10^{-2}	-5.0×10^{-2}	-5.2×10^{-2}
			-36.1	-33.9	-19.8	-17.1	-34.4	-32.9	-34.0
			5.8×10^7	5.2×10^7	$1.8{ imes}10^8/$	2.0×10^{8}	9.1×10^{7}	8.5×10^7	8.2×10^7
Column CCINS (ocean) (cm ⁻²)	MODIS	$2.4{ imes}10^8$	-1.9×10^{8}	$-1.9{ imes}10^{8}$	-6.7×10^7	-4.6×10^7	$-1.5 \times 10^8/$	$-1.6 \times 10^8/$	$-1.6 \times 10^8/$
			-76.4	-78.6	-27.5	-18.8	-62.7	-65.3	-66.6
$CDNC (cm^{-3})$	BE07	112 1	45.5/	46.7/	89.7/	93.1/	65.0/	66.7/	67.0/
	DEVI	1.611	-67.7/-59.9	-66.5/-58.8	-23.4/-20.7	-20.0/-17.7	-48.1/-42.5	-46.4/-41.0	-46.1/-40.8

Table 3. Mean Bias (MB) and Normalized Mean Bias (NMB, in %) of Radiative/Cloud Predictions for the 2001 Simulations

(<u>http://www.pangaea.de</u>). ^cThe values of modeled results (Sim), MBs, and NMBs are expressed as Sim/MB/NMB.

		lable 4. Mu	ean Bias (MB)	1able 4. Mean Blas (MIB) and Normalized Mean Blas (NMIB, III %) of Chemical Predictions for the 2001	Mean Blas (NMI	<u>3, 11 %) 01 Unemical</u>	nical Fredictions	TOT THE ZUUT SIN	Simulations	
Species/						Inuic				
variables ^a	Domain	Obs.	MAM_SIM	MAM_CB05_GE	MAM_CON	MAM_CON/IMN	MAM_CON/ISO	MAM_NEWA	MAM_NEWB	MAM_NEW/ EMIS
00	Europe	123.0	-	112.4/-10.6/-8.6	115.0/-8.0/-6.5	107.9/-15.1/-12.3	114.0-9.0/-7.3	118.8/-4.2/-3.4	113.6/-9.4/-7.6	137.9/14.9/12.1
00	East Asia	0.6	-	0.1/-0.5/-82.1	0.1/-0.5/-82.0	0.1/-0.5/-81.8	0.1/-0.5/-81.8	0.1/-0.5/-82.0	0.1/-0.5/-81.7	0.1/-0.5/-78.7
	CONUS	3.9	14.2/10.3/264.8 ^b	14.4/10.5/270.1	15.6/11.7/301.2	15.1/11.2/286.1	15.4/11.5/295.8	15.3/11.4/291.8	15.3/11.4/293.0	9.8/5.9/152.2
SO_2	Europe	6.8	13.4/6.6/97.5	13.8/7.0/103.2	15.2/8.4/123.0	13.6/6.8/100.3	14.6/7.8/114.7	15.7/8.9/130.7	14.5/7.7/114.0	6.8/0.0/0.3
	East Asia	12.5	4.6/-7.9/-63.0	4.8/-7.7/-61.3	4.8/-7.7/-61.4	4.8/-7.7/-61.8	4.9-7.6/-61.0	4.8/-7.7/-61.2	4.8/-7.7/-61.2	5.8/-6.7/-53.4
NH ₃	Europe	9.4	1.7/-7.7/-82.0	1.8/-7.6/-80.8	1.2/-8.2/-86.8	1.1/-8.3/-87.8	1.4/-8.0/-84.7	1.5/-7.9/-84.3	1.1/-8.3/-84.0	2.1/-7.3/-77.5
- ON	Europe	20.2		4.6/-15.6/-77.0	5.2/-15.0/-74.1	4.7/-15.5/-76.5	5.0/-15.2/-75.2	5.2/-15.0/-74.1	4.9/-15.3/-75.7	4.9/-15.3/-75.9
2001	East Asia	14.0		1.6/-12.4/-88.4/	1.7/-12.3/-88.0	1.7/-12.3/-88.2	1.6/-12.4/-88.4	1.7/-12.3/-88.3	1.6/-12.4/-88.5	1.7/-12.3/-88.2
	CONUS	34.6	-	44.6/10.0/28.9	42.6/8.0/23.0	42.5/7.9/22.7	44.4/9.8/28.4	44.1/9.5/27.4	43.7/9.1/26.4	44.4/9.8/28.1
03	Europe	53.5		90.2/36.7/68.6	84.4/30.9/57.7	84.5/31.0/58.0	87.6/34.1/63.7	87.0/33.5/62.7	87.7/34.2/63.9	88.4/34.9/65.2
	East Asia	26.4		42.8/16.4/62.2	42.7/16.3/61.7	40.7/14.3/54.3	42.6/16.2/65.9	42.1/15.7/59.6	43.0/16.6/63.0	42.5/16.1/61.2
ONH	CONUS	1.5		2.5/1.0/68.1	0.6/-0.9/-60.2	0.6/-0.9/-59.7	1.7/0.2/15.8	1.8/0.3/17.7	1.8/0.3/19.0	1.6/0.1/4.1
1 103	Europe	0.5	1	1.8/1.3/268.5	0.3/-0.2/-34.1	0.3/-0.2/-35.8	0.9/0.4/86.1	0.9/0.4/83.6	1.0/0.5/103.8	0.9/0.4/73.8
CO 2-	CONUS	2.6	2.5/-0.1/-5.1	2.4/-0.2/-7.2	$2.6/4.4 \times 10^{-2}/1.7$	2.6/4.2×10 ⁻² /1.6	2.4/-0.2/-7.9	2.4/-0.2/-6.3	2.5/-0.1/-5.5	1.9/-0.7/-28.4
504	Europe	2.2	3.0/0.8/36.5	2.9/0.7/33.1	3.1/0.9/40.3	3.0/0.8/35.8	2.9/0.7/32.6	3.1/0.9/39.4	3.0/0.8/36.8	2.0/-0.2/-7.2
+ 110	CONUS	1.4	1.0/-0.4/-32.1	0.8/-0.6/-39.6	1.7/0.3/20.0	1.7/0.3/19.7	1.3/-0.1/-6.4	1.3/-0.1/-6.5	1.3/-0.1/-4.3	1.2/-0.2/-13.1
	Europe	1.2	1.1/-0.1/-9.1	1.0/-0.2/18.3	2.2/1.0/85.0	2.0/0.8/65.7	1.8/0.6/49.4	1.9/0.7/54.8	1.7/0.5/37.7	1.6/0.4/32.5
- 01	CONUS	1.0		-	3.0/2.0/198.2	2.9/1.9/192.7	$1.0/-4.8 \times 10^{-2}/-4.8$	0.9/-0.1/-9.6	$1.0/-2.2 \times 10^{-2}/-2.1$	$1.0/4.0 \times 10^{-3}/0.4$
103	Europe	2.0		-	3.4/1.4/67.8	3.0/1.0/49.4	1.9/-0.1/-4.3	$2.0/-4.0 \times 10^{-2}/-2.0$	1.8/-0.2/-12.5	2.1/0.1/5.2
-5	CONUS	0.1		-	0.5/0.4/359.9	0.5/0.4/373.1	$0.1/-1.5 \times 10^{-2}/-14.5$	$0.1/-1.8 \times 10^{-2}/-17.5$	$0.1/-1.5 \times 10^{-2}/-12.1$	$0.1/\text{-}2.8{\times}10^{\text{-}3}/\text{-}2.8$
5	Europe	0.7		-	1.4/0.7/102.8	1.3/0.6/89.9	$0.7/2.1 \times 10^{-3}/0.3$	$0.7/1.4 \times 10^{-2}/2.0$	0.6/-0.1/-14.5	$-4.7 \times 10^{-2}/-6.7$
BC	CONUS	0.6	0.3/-0.3/-54.6	0.3/-0.3/-55.8	0.3/-0.3/-54.7	0.3/-0.3/-54.6	0.3/-0.3/-53.8	0.3/-0.3/-54.3	0.3/-0.3/-54.9	0.4/-0.2/-29.4
00	CONUS	1.1	0.8/-0.3/-22.7	1.0/-0.1/-12.1	1.0/-0.1/-11.4	1.0/-0.1/-11.9	1.0/-0.1/-8.6	1.0/-0.1/-9.1	1.0/-0.1/-11.3	$1.0/5.6 \times 10^{-3}/0.5$
TC	CONUS	2.5	1.3/-1.2/-47.9	1.4/-1.1/-43.1	1.4/-1.1/-42.2	1.4/-1.1/-42.5	1.4/-1.0/-40.9	1.5/-1.0/-41.1	1.4/-1.1/-42.5	1.6/-0.9/-35.0
$PM_{2.5}$	CONUS	7.9	4.9/-3.0/-37.6	5.0/-2.9/-36.8	9.5/1.6/20.1	6.6/1.3/16.7	7.8/-0.1/-1.7	6.9/-1.0/-13.2	7.2/-0.7/-8.8	6.8/-1.1/-13.5
	Europe	14.5	8.4/-6.1/-41.8	7.9/-6.6/-45.3	13.7/-0.8/-5.5	14.4/-0.1/-0.9	11.0/-3.5/-24.4	11.9/-2.6/-17.7	10.9/-3.6/-24.9	10.6/-3.9/-27.2
PM_{10}	Europe	25.7	17.5/-8.2/-31.8	16.5/-9.2/-35.8	22.5/-3.2/-12.3	23.0/-2.7/-10.5	20.1/-4.8/-18.5	21.4/-4.3/-16.6	20.7/-5.0/-19.4	20.9/-4.8/-18.8
	East Asia	118.5	38.5/-80.0/-67.5	44.9/-73.6/-62.1	55.9/-62.6/-52.8	58.8/-57.7/-48.7	48.5/-70.0/-59.1	65.5/-53.0/-44.7	55.6/-62.9/-53.1	48.2/-70.3/-59.3
Col.CO	Globe	1.3×10^{18}	I	1.2×10^{18}	$1.2 \times 10^{18}/$	$1.2 \times 10^{18}/$	$1.2 \times 10^{18}/$	$1.2 \times 10^{18}/$	$1.2 \times 10^{18}/$	$1.3 \times 10^{18}/$
			I	-7.4×10 ¹⁶ /-5.7	-5.7×10 ¹⁶ /-4.4	-6.3×10 ¹⁶ /-4.8	-6.4×10 ¹⁶ /-4.9	-6.3×10 ¹⁶ /-4.8	-5.6×10 ¹⁶ /-4.3	2.3×10 ¹⁶ /1.8
Col.NO ₂	Globe	4.7×10^{14}	ı	6.7×10^{14}	6.2×10^{14}	6.2×10^{14}	6.5×10^{14}	6.5×10^{14}	6.5×10^{14}	$6.5 \times 10^{14}/$
				$1.9 \times 10^{14}/40.5$	$1.4 \times 10^{14}/30.4$	$1.4 \times 10^{14}/30.0$	$1.8 \times 10^{14}/37.5$	$1.8 \times 10^{14}/37.2$	$1.8 \times 10^{14}/37.9$	1.8×10 ¹⁴ /37.3
TOR	Globe	30.3	29.8/-0.5/1.6	29.2/-1.1/-3.7	27.6/-2.7/-9.0	27.4/-2.9/-9.6	28.8/-1.5/-4.9	28.7/-1.6/-5.2	28.6/-1.5/-5.0	28.6-1.5/-4.9
ſ	Globe	0.6	0.003/-0.6/-99.6	0.1/-0.5/-99.5	0.5/-0.1/-12.8	0.3/-0.3/-49.6	0.8/0.2/36.1	0.3/-0.3/-53.1	0.3/-0.3/-51.7	0.3/-0.3/-62.0
^a The units are cm ⁻² ; TOR, DU	CO, ppb (over J; J, cm ⁻³ s ⁻¹ . Al	Europe) and pr l other concent	pm (over East Asia); rations are in $\mu g m^{-3}$.	^a rthe units are CO, ppb (over Europe) and ppm (over East Asia); SO ₂ , ppb (over East Asia) and µg m ⁻³ (over CONUS and Europe); O ₃ , ppb (over CONUS) and µg m ⁻³ (over Europe); column CO and NO ₂ , molecules cm ⁻² ; TOR, DU; J, cm ⁻³ s ⁻¹ . All other concentrations are in µg m ⁻³ . ^b The values of modeled results (Sim), MBs, and NMBs are expressed as Sim/MB/NMB.	sia) and µg m ⁻³ (over (ed results (Sim), MBs,	CONUS and Europe); , and NMBs are expre	O ₃ , ppb (over CONU) ssed as Sim/MB/NME	S) and µg m ⁻² (over Eu 3.	ırope); column CO aı	nd NO ₂ , molecules
					59					

Table 4 Mean Bias (MB) and Normalized Mean Bias (NMB in %) of Chemical Predictions for the 2001 Simulations

predictions of O ₃		$\frac{1}{2}$, and mix	O3 mixing face	bs over Europe III I	
	Network		Obs (µg m ⁻³)	Sim (µg m ⁻³)	MB/NMB
				75.2	37.5/99.6 ^a
	Airbase	NO ₂	26.0	7.6	-18.4/-70.9
	DDOA	O ₃	31.0	74.2	43.2/139.2
Winter	BDQA	NO ₂	30.6	5.6	-25.0/-81.9
vv IIItel		O ₃	50.7	75.7	25.0/49.3
	EMEP	NO ₂	9.0	8.3	-0.7/-7.8
		HNO ₃	0.5	0.5	-4.9×10 ⁻³ /1.0
	Airbase	O ₃	63.1	100.8	37.7/59.7
	Allbase	NO ₂	20.0	4.6	-15.4/-77.1
	DDOA	O ₃	59.6	98.9	39.3/65.9
Spring	BDQA	NO ₂	23.6	3.1	-20.5/-87.0
		O ₃	75.0	101.9	26.9/35.9
	EMEP	NO ₂	5.9	4.9	-1.0/-17.2
		HNO ₃	0.4	0.9	0.5/144.5
	Ainhaga	O ₃	64.9	93.5	28.6/44.0
	Airbase	NO ₂	16.2	4.4	-11.8/-72.8
	BDQA	O ₃	64.5	94.5	30.0/46.5
Summer	BDQA	NO ₂	18.7	3.6	-15.1/-80.9
		O ₃	72.2	91.2	19.0/26.3
	EMEP	NO ₂	4.7	4.4	-0.3/-6.2
		HNO ₃	0.5	1.3	0.8/169.6
	Ainhaga	O ₃	40.5	79.5	39.0/96.4
	Airbase	NO ₂	21.7	5.3	-16.4/-75.6
		O ₃	35.7	80.9	45.2/126.5
Autumn	BDQA	NO ₂	24.8	3.7	-21.1/-85.2
		O ₃	51.7	78.2	26.5/51.2
	EMEP	NO ₂	6.6	5.2	-1.4/-21.1
		HNO ₃	0.6	0.9	0.3/45.0

Table 5. The observed values and the mean bias (MB) and normalized mean bias (NMB, in %) of predictions of O₃ NO₂, and HNO₃ mixing ratios over Europe in MAM_NEWA

^aThe values of MBs and NMBs are expressed as MB/NMB.

Snecies/Variables	Dataset	Ohs.		Simulations	
		2	MAM_SIM_5Y	MAM_NEW_SYA	MAM_NEW_5YB
LWD (W $m^{-2})^a$	CERES	307.6	302.9/-4.7/-1.5/2.9/11.6°	303.9/-3.6/-1.1/2.8/11.3	304.4/-3.1/-1.0/2.9/11.3
$SWD (W m^{-2})^b$	CERES	163.9	169.9/5.9/3.6/7.0/14.1	166.5/2.5/1.5/6.5/13.8	167.0/3.1/1.9/6.7/13.7
OLR (W m ⁻²)	NOAA-CDC	215.9	222.5/6.6/3.1/3.5/8.9	220.7/4.8/2.2/3.4/9.1	221.4/5.5/2.6/3.5/9.0
SWCF (W m ⁻²)	CERES	-41.0	-38.8/2.2/-5.4/-21.5/12.0	-41.5/-0.5/1.2/-21.4/12.5	-40.8/0.2/-0.5/-22.2/12.4
CF (%)	MODIS	67.1	66.6/-0.6/-0.8/15.2/13.3	67.3/0.2/0.3/14.7/13.0	66.6/-0.6/-0.9/15.5/13.7
COT	MODIS	17.3	7.1/-10.3/-59.3/70.2/15.1	7.9/-9.4/-54.5/65.7/14.6	7.8/-9.6/-55.2/65.6/14.5
$CWP (g m^{-2})$	MODIS	86.0	38.2/-47.8/-55.5/55.7/52.9	43.2/-42.8/-49.8/50.0/49.2	43.4/-42.6/-49.5/49.7/49.2
PWV (cm)	SIDOM	1.93	1.96/0.03/1.5/11.6/0.3	1.99/0.06/2.9/10.9/0.3	1.97/0.04/1.8/13.8/0.3
AOD	MODIS	0.2	0.1/-0.07/-44.1/54.5/0.1	0.1/-0.06/-39.2/51.3/0.1	0.1/-0.06/-36.3/49.5/0.1
Column CCN5 (ocean) (cm ⁻²)	MODIS	2.5×10^{8}	5.3×10 ⁷ /-1.9×10 ⁸ / -78.6/78.6/5.7×10 ⁸	$8.6 \times 10^7/1.6 \times 10^8/$ -65.2/65.2/5.5×10 ⁸	8.6×10 ⁷ /1.6×10 ⁸ / -65.3/65.3/5.5×10 ⁸
$CDNC (cm^{-3})$	BE07	112.6	44.2/-68.3/-60.7/61.6/84.3	69.2/-43.4/-38.6/44.2/66.8	68.8/-43.8/-38.9/45.5/67.9
^a The pair of observation and simulation is removed in	n and simulation is re	emoved in the	statistical calculation if the observ	the statistical calculation if the observed LWD value is lower than 50 W m ⁻² or higher than 700 W m ⁻²	m ⁻² or higher than 700 W m ⁻²

^oThe values are expressed as Sim/MB/NMB/NME/RMSE. Sim: simulated values; MB: mean bias; NMB: normalized mean bias (%); NME: normalized mean error (%); RMSE: root mean squared error.

Species/			Sint	llations) Simulations	
variables ^a	Domain	Obs.	MAM_SIM_5Y	MAM_NEW_5YA	MAM_NEW_5YB
СО	East Asia	562.0		 139.7/-422.3/-75.1/75.1/451.8 ^b	137.0/-425.0/-75.6/75.6/454.0
	CONUS	3.4	9.6/6.2/183.9/184.6/9.9	10.1/6.7/198.8/199.1/10.6	10.3/6.9/203.1/203.5/10.9
SO ₂	Europe	6.6	6.0/-0.6/-9.3/73.3/7.9	6.6/-0.06/-0.9/77.2/8.3	6.2/-0.4/-5.5/74.6/8.0
	East Asia	3.4	3.4/0.04/1.1/76.0/5.0	3.4/0.01/0.4/76.2/5.0	3.4/-0.05/-1.6/73.1/4.8
NH ₃	Europe	6.3	3.0/-3.3/-52.0/81.0/25.3	2.4/-3.9/-61.3/79.7/25.3	2.4/-3.9/-62.0/79.3/25.3
	Europe	23.5	-	5.8/-17.7/-75.4/76.5/21.5	5.5/-18.0/-76.7/77.7/21.7
NO ₂	East Asia	13.5	-	2.3/-11.2/-83.3/83.3/12.2	2.3/-11.2/-83.6/83.6/12.2
	CONUS	35.1	-	43.9/8.8/25.1/27.3/11.3	44.1/9.0/25.7/27.7/11.6
O ₃	Europe	52.7	-	86.6/33.9/64.5/64.6/36.4	89.2/36.5/69.3/69.4/38.8
	East Asia	27.4	-	45.6/18.2/66.4/66.4/19.2	45.5/18.1/66.0/66.0/19.1
	CONUS	1.4	-	1.6/0.2/16.3/39.5/0.7	1.6/0.2/12.1/38.2/0.7
HNO ₃	Europe	0.7	-	1.0/0.3/45.8/83.5/0.8	1.0/0.3/37.9/79.8/0.8
so 2-	CONUS	2.6	2.3/-0.3/-13.4/26.9/1.0	2.3/-0.3/-13.1/23.0/0.8	2.3/-0.3/-12.8/24.2/0.9
SO4 ²⁻	Europe	2.3	2.3/-0.04/-1.9/37.3/1.4	2.0/-0.3/-11.1/34.1/1.3	2.0/-0.3/-13.0/35.5/1.4
NUT +	CONUS	1.2	0.9/-0.3/-20.8/33.4/55.0	1.5/0.3/22.2/43.2/0.8	1.5/0.3/26.4/44.3/0.8
NH4 ⁺	Europe	1.0	0.8/-0.2/-16.8/36.9/0.5	1.6/0.6/62.8/68.7/0.9	1.5/0.5/53.8/60.3/0.8
NOT	CONUS	1.1	-	1.6/0.5/41.3/85.4/1.4	1.6/0.5/49.8/90.2/1.5
NO ₃ -	Europe	1.8	-	2.3/0.5/30.3/51.1/1.2	2.2/0.4/24.7/47.0/1.1
CI-	CONUS	0.1	-	0.1/3.1×10 ⁻³ /2.7/105.8/0.4	0.1/8.7×10 ⁻³ /7.8/110.1/0.4
Cl	Europe	0.3	-	2.4/2.1/681.2/681.2/2.9	2.3/2.0/663.3/663.6/2.8
BC	CONUS	0.4	0.3/-0.1/-17.9/44.4/0.3	0.3/-0.1/-15.6/44.0/28.2	0.3/-0.1/-17.7/44.3/0.2
OC	CONUS	1.2	0.9/-0.3/-23.2/59.3/1.0	1.1/-0.1/-7.7/56.7/1.0	1.1/-0.1/-11.0/54.3/0.9
ТС	CONUS	3.1	1.4/-1.7/-54.4/62.8/2.8	1.7/-1.4/-45.7/57.1/2.6	1.6/-1.5/-47.1/57.1/2.7
PM _{2.5}	CONUS	8.8	7.2/-1.6/-17.9/37.0/4.3	9.2/0.4/4.1/33.5/3.9	8.7/-0.1/-1.1/29.4/3.6
	Europe	14.6	6.7/-7.9/-53.9/54.6/10.6	9.7/-4.9/-33.8/37.6/8.6	10.0/-4.6/-31.7/36.1/8.4
PM ₁₀	Europe	26.3	15.1/-11.2/-42.6/46.8/15.9	18.7/-7.6/-28.8/36.1/13.9	19.9/-6.4/-24.4/33.5/13.1
-	East Asia	107.9	45.4/-62.5/-58.0/59.3/70.7	52.5/-57.4/-53.2/54.2/66.0	57.8/-50.1/-46.5/50.0/61.6
Col.CO	Globe	1.4×10 ¹⁸		1.3×10 ¹⁸ /-1.4×10 ¹⁷ /	1.2×10^{18} /- 1.5×10^{17} /
	01000	1.4^10	-	-10.2/16.5/3.1×10 ¹⁷	-11.0/17.2/3.2×10 ¹⁷
Col.NO ₂	Globe	5.3×10 ¹⁴	-	8.4×10 ¹⁴ /3.1×10 ¹⁴ /	8.3×10 ¹⁴ /3.0×10 ¹⁴ /
				59.2/70.0/5.4×10 ¹⁴	57.6/69.2/5.4×10 ¹⁴
TOR	Globe	30.4	29.9/-0.5/1.6/16.3/6.1	30.5/0.1/0.4/15.0/5.8	29.9/-0.5/-1.7/16.4/6.1

Table 7. Statistical Performance of Chemical Predictions (Average of from the 5-yr (2001-2005) Simulations)

^aThe units are CO, ppm (over East Asia); SO₂, ppb (over East Asia) and µg m⁻³ (over CONUS); O₃, ppb (over CONUS) and µg m⁻³ (over Europe); column CO and NO₂, molecules cm⁻²; TOR, DU. All other concentrations are in µg m⁻³. ^bThe values are expressed as Sim/MB/NME/RMSE. MB: mean bias; NMB: normalized mean bias (%); NME: normalized

mean error (%); RMSE: root mean squared error.

	Sim	ulations	
	MAM_SIM_5Y	MAM_NEW_5YA	Previous studies
Tropospheric CO (Tg) ^a	N.A. ^c	322.06	337-354 ^d
Tropospheric O ₃ (DU) ^a	29.7 °	30.5	34.04 ^e
Tropospheric O ₃ (Tg) ^a	324.14 °	332.87	372 ^e
DMS (Tg S)	0.051	0.058	0.067 ^f
SO ₂ (Tg S)	0.276	0.281	0.34 ^f
H_2SO_4 (Tg S)	3.8×10 ⁻⁴	1.9×10 ⁻³	4.2×10 ^{-4 f}
Tropospheric NO _x ^{a, b}	N.A. ^c	0.116 Tg N	7.6×10^{14}
Γ ropospheric NO_x		$(8.24 \times 10^{14} \text{ molecules cm}^{-2})$	molecules cm ^{-2 g}
NOy (Tg N) ^b	N.A. ^c	3.26	N.A. ^c
NH3 (Tg N)	0.074	0.059	0.064 ^f
VOCs (Tg C) ^b	N.A. ^c	7.63	N.A. ^c
Tropospheric HCHO (Tg C) ^a	N.A. ^c	0.391	0.335-0.349 ^d
SO_4^{2-} (Tg S)	0.36	0.39	0.84 °,0.47 ^f , 0.66 ^h
$NO_3^-(Tg N)$	N.A. ^c	0.11	0.01-0.14 i
NH_4^+ (Tg N)	0.20	0.21	0.24 ^f , (0.27-0.44) ⁱ
$Na^{+}(Tg)$	2.93	3.04	2.98 ^e , (0.38-5.19) ⁱ
Cl ⁻ (Tg)	4.52	4.47	4.60 ^e , (0.59-8.02) ⁱ
BC (Tg)	0.091	0.076	0.28^{e} , 0.093^{f}
OC (Tg)	0.45	0.61	1.28 ^e
POM (Tg)	0.63	0.48	$0.68^{\rm f}, 1.70^{\rm h}$
SOA (Tg)	N.A. ^c	0.38	1.15 ^f , 0.59 ^j
Dust (Tg)	25.78	26.43	24.7 ^f , (7.9-35.9) ⁱ

Table 8. Global Burdens of Major Gaseous and Aerosol Species from the 5-yr (2001-2005) Simulations

^a CESM/CAM5 simulations use 30 model layers, with atmospheric pressures from ~1000 mb (layer 30) to ~3 mb (layer 1). Troposphere refers to model layers below tropopause height.

^b $NO_x = NO + NO_2$; $NO_y = NO_x$ + nitrogen trioxide (NO_3) + dinitrogen pentoxide (N_2O_5) + nitrous acid (HONO) + nitric acid (HNO_3) + pernitric acid (HNO_4) + peroxyacyl nitrate (PAN) + $\geq C_3$ peroxyacyl nitrate (PANX) + other organic nitrate (NTR); VOCs-volatile organic compounds including acetaldehyde (ALD2), carboxylic acid(AACD), long-chain alkanes (ALKH), Cresol and higher phenols (CRES), ethene (ETH), ethane (ETHA), ethanol (ETOH), formaldehyde (FORM), internal olefinic carbon bond (IOLE), methanol (MEOH), olefinic carbon bond (OLE), paraffin carbon bond (PAR), polycyclic aromatic hydrocarbons (PAH), toluene (TOL), xylene (XYL), isoprene (ISOP), and terpene (TERP).

 $^{\circ}$ N.A – not available, it refers to the species that are not treated in MAM_SIM_5Y or species having no burden data from previous studies. Tropospheric O₃ burden in MAM_SIM_5Y is from climatology. N.A. in SOA is due to no SOAG emission for MAM_SIM_5Y.

^d Williams et al. (2009)

^e Horowitz et al. (2006)

^f Liu et al. (2012)

^g Larmarque et al. (2006)

^h Textor et al. (2006)

ⁱ Tsigaridis et al. (2006)

^j Heald et al. (2008)

Figure captions

Figure 1a. Absolute differences of H_2O_2 , SO_2 , SO_4^{2-} , and SOA between MAM_CB05_GE and MAM_SIM for 2001.

Figure 1b. Absolute differences between the mixing ratios of surface OH, HO₂, NO₃, and O₃ predicted from MAM_CB05_GE and climatology values used in MAM_SIM for 2001.

Figure 2. Spatial distributions of CO, O₃, NO₂, HNO₃, HCl, and isoprene (ISOP) at surface simulated by MAM CB05 GE for 2001.

Figure 3. Spatial distributions of total ammonium, total sulfate, total nitrate, total chloride, PM_{2.5}, NH₃, SO₂, H₂SO₄, HNO₃, and HCl at surface between MAM_CON and MAM_CB05_GE for summer (June, July, and August (JJA)), 2001.

Figure 4. Vertical distributions of new particle formation rate (J) (row 1) and aerosol number (PM_{num}) (row 3) simulated by MAM_CON/IMN for 2001. The overlay plots in row 2 show the distributions of simulated and observed J in bottom 1000-m in the atmosphere. Circles on overlay plots represent observations for J. Different colors of circles represent different values of J, using the same color scale as simulated J.

Figure 5. Absolute differences of PM_{2.5}, AOD, column CCN5, CF, COT, and SWCF between MAM_CON/IMN and MAM_CON for 2001.

Figure 6a. Absolute differences of major PM species and their gas precursors between MAM_CON/ISO and MAM_CON for summer, 2001.

Figure 6b. Absolute differences of major PM species and their gas precursors between MAM_CON/ISO and MAM_CON for winter, 2001.

Figure 7. The spatial distributions of TCAT/TSO4 at surface in MAM_CON and MAM_CON/ISO and their absolute differences (MAM_CON/ISO - MAM_CON) for summer and winter, 2001.

Figure 8. Absolute differences of major aerosol species and their gas precursors between metastable (MAM_NEWA) and stable (MAM_NEWB) conditions.

Figure 9. Absolute differences of major aerosol species and their gas precursors, new particle formation rate (J), and aerosol number between MAM_NEW_5YA and MAM_SIM_5Y for 2001-2005.

Figure 10. Absolute differences of major cloud and radiative variables between MAM_NEW_5YA and MAM_SIM_5Y for 2001-2005.

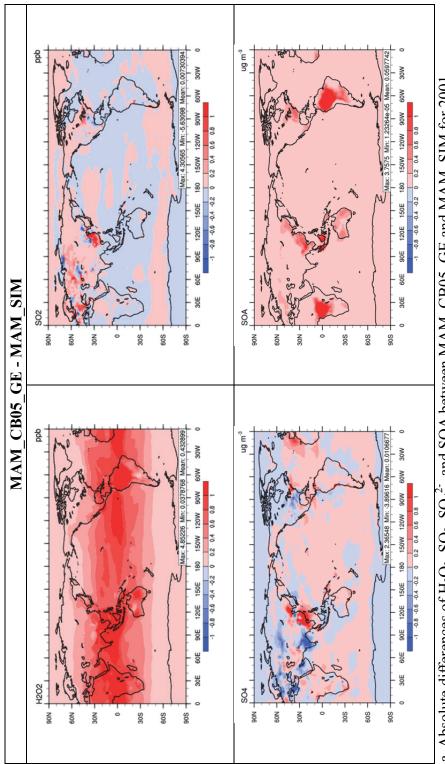
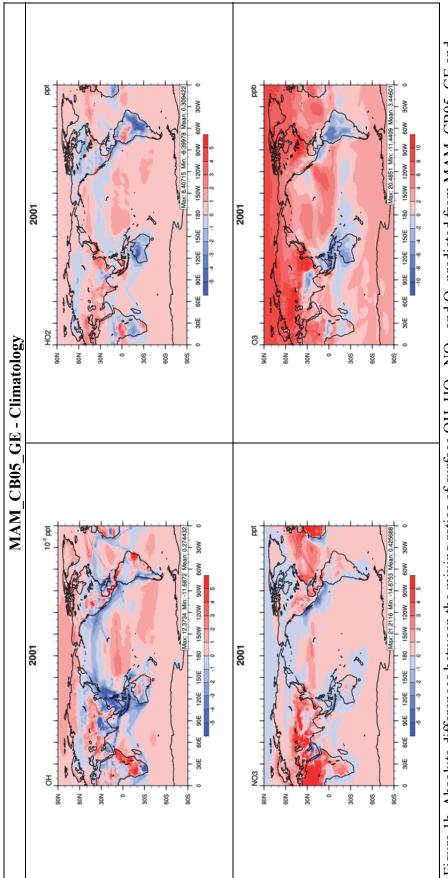


Figure 1a. Absolute differences of H₂O₂, SO₂, SO₄²⁻, and SOA between MAM_CB05_GE and MAM_SIM for 2001.





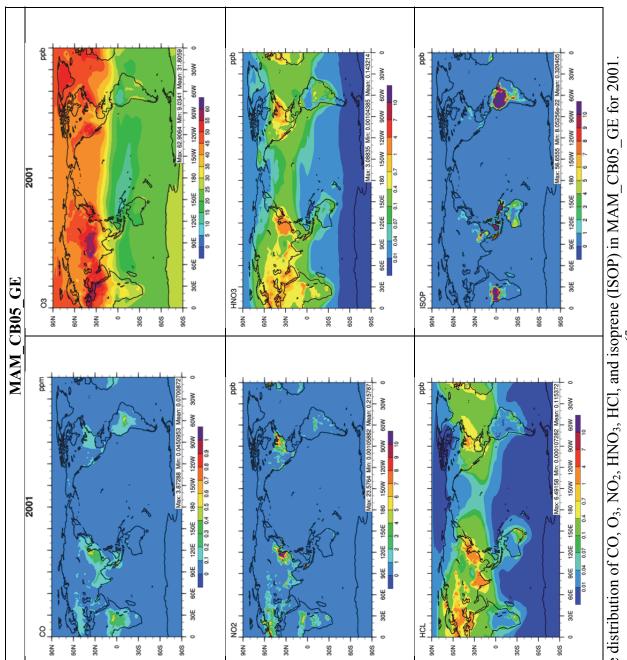


Figure 2. Surface distribution of CO, O₃, NO₂, HNO₃, HCl, and isoprene (ISOP) in MAM_CB05_GE for 2001.

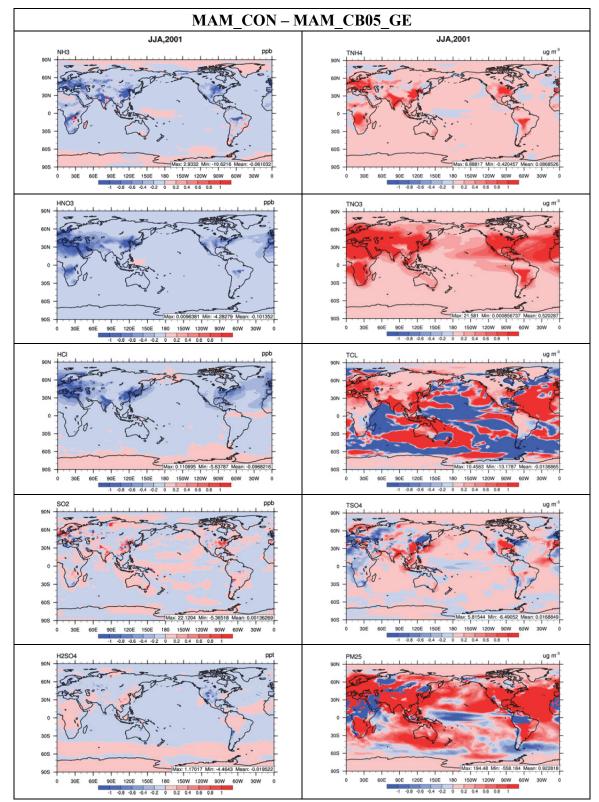


Figure 3. Surface distribution of total ammonium, total sulfate, total nitrate, total chloride, PM_{2.5}, NH₃, SO₂, H₂SO₄, HNO₃, and HCl between MAM_CON and MAM_CB05_GE for summer (June, July, and August (JJA)), 2001.

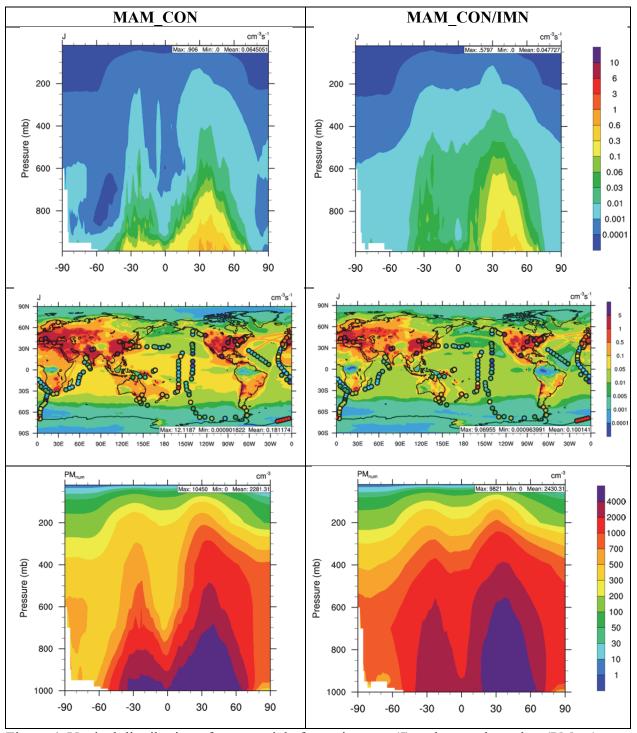


Figure 4. Vertical distribution of new particle formation rate (J) and aerosol number (PM_{num}) simulated by MAM_CON/IMN for 2001. The overlay plots show the distribution of J in bottom 1000-m. Circles on overlay plots represent observations for J. Different colors of circles represent different values of J, using the same color scale as simulated J.

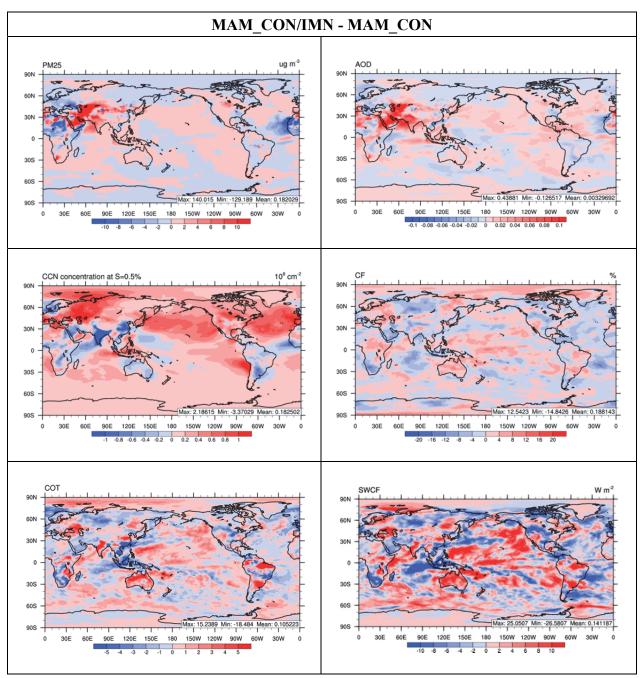


Figure 5. Absolute differences of PM_{2.5}, AOD, column CCN5, CF, COT, and SWCF between MAM_CON/IMN and MAM_CON for 2001.

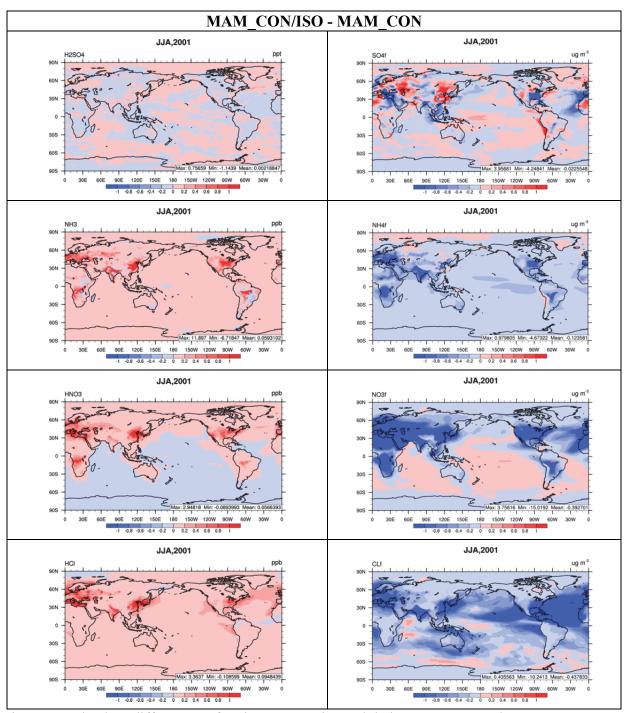


Figure 6a. Absolute differences of major PM species and their gas precursors between MAM_CON/ISO and MAM_CON for summer, 2001.

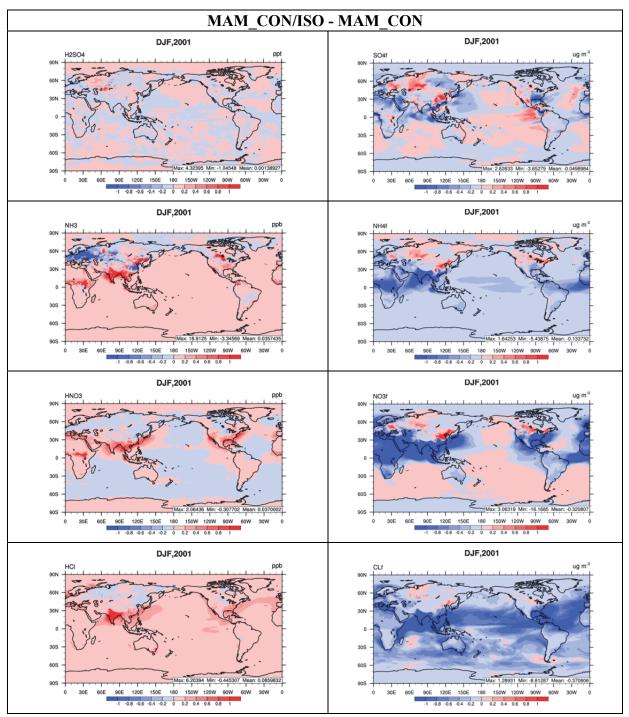
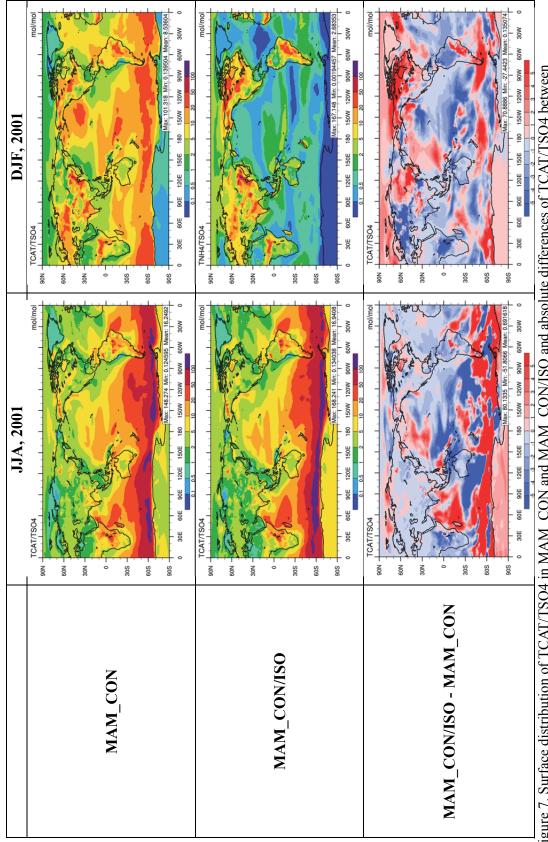


Figure 6b. Absolute differences of major PM species and their gas precursors between MAM_CON/ISO and MAM_CON for winter, 2001.





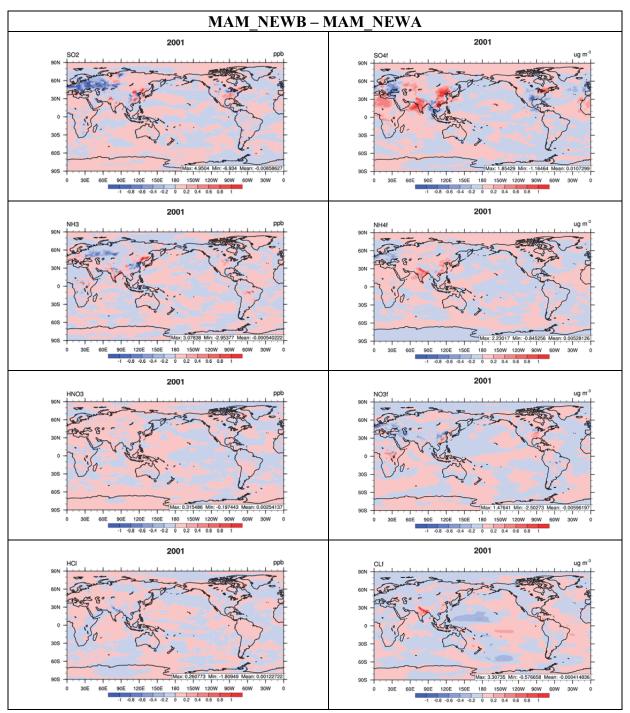


Figure 8. Absolute differences of major aerosol species and their gas precursors between metastable and stable conditions.

