

1 **Supplementary material to “Influence of Surface Morphology on the Immersion Mode Ice**  
2 **Nucleation Efficiency of Hematite”**

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28 Möhler. **Influence of Surface Morphology on the Immersion Mode Ice Nucleation Efficiency**  
29 **of Hematite**, for *Atmospheric Chemistry and Physics*

30 This supplementary information provides additional details in the measurement of  
31 absolute number of charges by polyelectrolyte titrations with PVS and PDADMAC (Table S1) as  
32 well as representative AIDA adiabatic expansion experiments (Figure S1 and S2).

33 Below, we briefly describe an experimental procedure to estimate maximal charge  
34 densities of hematite particles. First, we generated the maximal interface potentials in hematite  
35 suspensions by adding 0.01 mol L<sup>-1</sup> NaOH or HCl solution. Compensations of the developed  
36 charges in the suspensions were directly followed and carried out by addition the oppositely  
37 charged polyelectrolyte solution (PVS or PDADMAC) to zero potential to identify the absolute  
38 number of surface charges. With measured and known parameters summarized in Table S1, we  
39 calculated the maximal charge surface densities,  $a$  (nm<sup>-2</sup>), according to

$$41 \quad a = \frac{c_{eq} \cdot V \cdot N_A}{m \cdot A_{BET}} \quad (S1)$$

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43 where  $c_{eq}$  is the polyelectrolyte charge equivalent concentration (mol L<sup>-1</sup>),  $V$  is the titrated  
44 volume to isoelectric point (L),  $N_A$  is the Avogadro's constant (mol<sup>-1</sup>),  $m$  is the hematite mass (g)  
45 and  $A_{BET}$  is the BET specific surface area (m<sup>2</sup> g<sup>-1</sup>). We note that 1 mol L<sup>-1</sup> of polyelectrolyte  
46 concentration compensates 400 charges.

47 Figure S1 displays the time-series of ice crystal concentration ( $N_{ice}$ ) as well as liquid  
48 droplet concentration ( $N_{droplet}$ ), temperature ( $T$ ), relative humidity with respect to ice and water  
49 measured by the TDL, and particle phase inferred by particles' backscattered intensities to the  
50 incident polarisation state of the laser light during the immersion mode freezing experiment for  
51 cubic hematite and milled hematite particles shown in Manuscript Figure 2 (INUIT04\_13 and  
52 INUIT04\_15, respectively). It is noteworthy that the observed early increases in depolarisation  
53 ratio before the full droplet formation at water saturation are the indicator of deposition mode  
54 freezing (Figure S1 A. iv. and B. iv.). As prescribed in Manuscript Section 3.2, the contributions  
55 of depositional ice formation to the total ice crystals formed through an expansion (up to 27%)  
56 was quantitatively minor and did not inhibit new ice formation in the immersion mode after  
57 reaching to the water supersaturation condition (i.e. no indication of water depletion until  
58 homogeneous freezing emerges). Therefore, the ice crystals formed through deposition mode  
59 freezing were simply subtracted from the total number of ice crystals measured within

60 heterogeneous freezing regime to compute  $n_s$  solely accounting for the immersion mode ice  
61 nucleation.

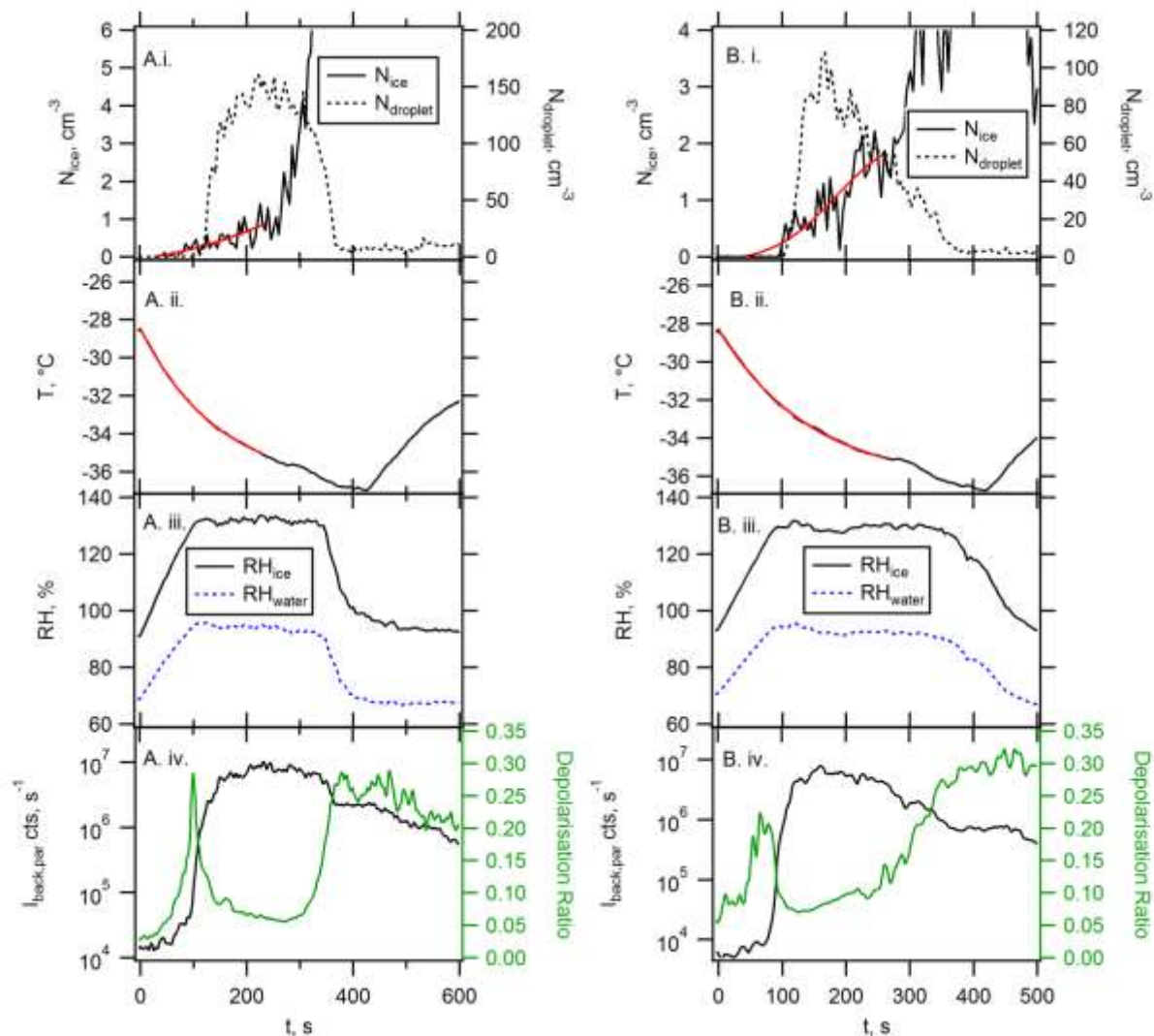
62 Figure S2 illustrates the size distributions of particles, droplets, and ice crystals measured  
63 by the welas. Observed size growth at 100 s was triggered by droplet formation. Particles above  
64 20  $\mu\text{m}$  diameter were counted as ice crystals. The contributions from homogeneous ice  
65 nucleation appear below  $-35\text{ }^\circ\text{C}$ . Soon after that point, abrupt increase in depolarisation ratio and  
66 quick decrease in water saturation were observed, which implies the presence of pure ice cloud  
67 (Figure S1).

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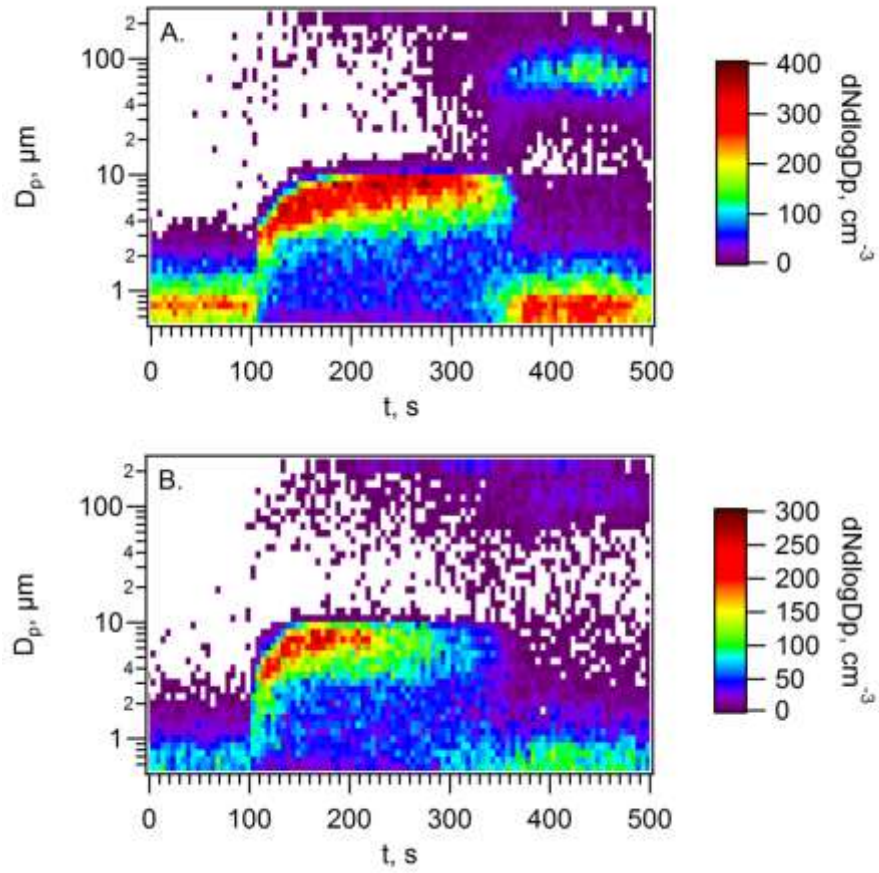
96 Table S1. Summary of parameters used to calculate the charge densities,  $a$  ( $\text{nm}^{-2}$ ), of cubic and  
 97 milled hematite particles. PVS and PDADMAC solutions were used to obtain maximal positive  
 98 and maximal negative charge densities, respectively.  
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Hematite	$c_{eq}$ , $10^{-5}$ mol L $^{-1}$	$V$ , $10^{-3}$ L	$m$ , $10^{-3}$ g	$A_{BET}$ , $\text{m}^2 \text{g}^{-1}$	$a$ , $\text{nm}^{-2}$
Cubic (max. positive)	1 $\pm$ 0.009	1.91 $\pm$ 0.01	10.1 $\pm$ 0.1	2.2 $\pm$ 0.1	0.36 $\pm$ 0.03
Cubic (max. negative)	10 $\pm$ 0.090	1.15 $\pm$ 0.01	10.1 $\pm$ 0.1	2.2 $\pm$ 0.1	1.39 $\pm$ 0.03
Milled (max. positive)	1 $\pm$ 0.009	1.83 $\pm$ 0.01	8.2 $\pm$ 0.1	3.7 $\pm$ 0.1	0.52 $\pm$ 0.05
Milled (max. negative)	1 $\pm$ 0.009	7.02 $\pm$ 0.01	8.2 $\pm$ 0.1	3.7 $\pm$ 0.1	3.13 $\pm$ 0.05

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 112 Figure S1. Typical experimental profiles, including i. ice crystal concentration ( $N_{ice}$ ) and liquid  
 113 droplet concentration ( $N_{droplet}$ ), ii. temperature ( $T$ ), iii. TDL, and iv. SIMONE measurements, of  
 114 the AIDA immersion mode ice nucleation experiment for A. cubic hematite particles  
 115 (INUIT04\_13) and B. milled hematite particles (INUIT04\_15). Note that the red lines represent  
 116 interpolated data. The  $I_{back,par}$  in Panel A.iv and B.iv denotes backscattered light scattering  
 117 intensity parallel to the incident polarisation state (log-scaled).  
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 120 Figure S2. Time-series of the wet size distribution of the AIDA immersion mode ice  
 121 nucleation experiment for A. cubic hematite particles (INUIT04\_13) and B. milled hematite  
 122 particles (INUIT04\_15).  
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