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Field measurements of trace gases emitted by prescribed fires in southeastern US pine forests using an open-path FTIR system

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We report trace-gas emission factors from three pine-understory prescribed fires in South Carolina, US measured during the fall of 2011. The fires were more intense than many prescribed burns because the fuels included mature pine stands not subjected to prescribed fire in decades that were lit following an extended drought. The emission factors were measured with a fixed open-path Fourier transform infrared (OP-FTIR) system that was deployed on the fire control lines. We compare these emission factors to those measured with a roving, point sampling, land-based FTIR and an airborne FTIR that were deployed on the same fires. We also compare to emission factors measured by a similar OP-FTIR system deployed on savanna fires in Africa. The data suggest that the method used to sample smoke can strongly influence the relative abundance of the emissions that are observed. The majority of the fire emissions were lofted in the convection column and they were sampled by the airborne FTIR along with the downwind chemistry. The roving, ground-based, point sampling FTIR measured the contribution of actively located individual residual smoldering combustion fuel elements scattered throughout the burn site. The OP-FTIR provided a ~ 30 m path-integrated sample of emissions transported to the fixed path via complex ground-level circulation. The OP-FTIR typically probed two distinct combustion regimes, "flaming-like" (immediately after adjacent ignition and before the adjacent plume achieved significant vertical development) and "smoldering-like." These two regimes are denoted "early" and "late", respectively. The emission factors from all three systems were plotted versus modified combustion efficiency and for some species (e.g. CH₄ and CH₃OH) they fit a single trend suggesting that the different emission factors for these species were mainly due to the specific mix of flaming and smoldering that each system sampled. For other species, the different fuels sampled also likely contributed to platform differences in emission factors. The path-integrated sample of the ground-level smoke layer adjacent to the fire provided by the OP-FTIR also provided our best estimate of fire-line exposure to smoke for wildland fire personnel. We provide a table of estimated fire-line

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Introduction

Biomass burning is a significant, global source of trace gases and particles that impact the chemical composition and radiative balance of the atmosphere (Crutzen and Andreae, 1990). Biomass burning includes open fires in forests, savannas, crop residues, and peatlands as well as biofuel and garbage burning (Akagi et al., 2011). In the US, wild and prescribed fires in forests account for a significant fraction of the total fire activity (Hardy et al., 2001; Melvin, 2012). In the southeastern US, prescribed fires are ignited in some wildlands to restore or maintain the natural, beneficial role that fire plays in fire-adapted ecosystems (Biswell, 1989; Carter and Foster, 2004; Keeley et al., 2009). These fires also help reduce the risk of wildfire and smoke impacts by consuming accumulated fuels under weather conditions that allow smoke production and dispersion to be at least partially controlled (Hardy et al., 2001; Wiedinmyer and Hurteau, 2010; Cochrane et al., 2012). The ideal "smoke management" scenario occurs when the majority of the smoke is produced by flaming combustion, lofted via convection, and directed away from major population centers. This requires that fuel conditions, boundary layer depth, wind speed, and wind direction are within specific limits. Land managers try to minimize prolonged smoldering outside the envelope of convection from the flame front, which is often termed "residual smoldering combustion", or RSC (Bertschi et al., 2003). This type of combustion typically produces un-lofted smoke that accounts for many of the local-scale air quality impacts of prescribed burning (Bertschi et al., 2003; Achtemeier, 2006). There are very few peer-reviewed field measurements of the emissions from RSC (Bertschi et al., 2003; Burling et al., 2011; Akagi et al., 2013) and these measurements are becoming more desirable with increased recognition that RSC is a major fuel consumption process in some ecosystems (Christian

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et al., 2007; Greene et al., 2007; Hyde et al., 2011; Turetsky et al., 2011; Benscoter et al., 2011).

This work is part of a series of studies of the smoke emissions from prescribed fires on US Department of Defense (DoD) bases. Previous studies from this series include ₅ Burling et al. (2010) who sampled the emissions from fuels collected on DoD bases and burned in a large laboratory combustion facility; Burling et al. (2011) and Akagi et al. (2012, 2013) who described airborne and ground-based smoke measurements on bases in the western and southeastern US; and Yokelson et al. (2013) who synthesized the laboratory and field results. In the previous studies, Burling et al. (2011) and Akagi et al. (2013) actively located and measured RSC using a rolling, cart-based FTIR system for point source samples of various fuel types (e.g. smoldering logs, stumps, litter, etc.) that were scattered throughout the site. In this study we focus on "passive" ground level emissions measurements using a static, open-path Fourier transform infrared (OP-FTIR) gas analyzer system that measured any smoke (including both flaming and smoldering emissions) that drifted through the fixed measurement path of ~ 30 m. Griffith et al. (1991) was first to employ an OP-FTIR system to study biomass burning emissions. More recently, OP-FTIR has been used to study polluted air in challenging environmental or industrial conditions, such as measuring volcanic emissions or aircraft exhaust (Gosz et al., 1998; Oppenheimer and Kyle, 2007; Schafer et al., 2004). Recently, Wooster et al. (2011) revived the use of OP-FTIR for field measurements of biomass burning, reporting emission ratios (ER) and emission factors (EF) for CO₂, CO, CH₄, HCHO, and NH₃ from savanna fires in Kruger National Park, South Africa.

In this study we describe the OP-FTIR system employed for these fires and the data reduction approach. We present a time series of OP-FTIR results with the simultaneous observations/activities of the other FTIR instruments noted for perspective. We calculate OP-FTIR EF for the trace gases detected. These EF are then compared to EF from the other FTIRs on the same fires and to EF measured by an OP-FTIR system deployed on savanna fires. Finally, we combine the OP-FTIR mixing ratio measure**ACPD**

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2 Experimental details

2.1 Open-path FTIR measurements

Measurements of ground-level smoke on the perimeter of three prescribed fires at Fort Jackson near Columbia, South Carolina (SC), US were made using a Bruker OPAG-22 OP-FTIR (Fig. 1a). The OPAG-22 is a tripod-mounted, field-portable FTIR system that can be used to monitor trace gas species in the atmosphere across distances of tens to hundreds of meters. An active configuration was used with an unmodulated SiC glowbar source and sender telescope at one end of the light path and the FTIR with receiver telescope at the other. The source was powered (~ 20 W) using a 12.6 V DC automobile battery. The 1200°C SiC source was mounted at the focal point of an f/4 Newtonian telescope with a 150 mm clear aperture. The sender telescope directed a collimated, broadband IR beam to a 150 mm receiver telescope coupled to the OPAG-22 FTIR spectrometer. Pathlengths of 29.3–32.2 m were used to optimize infrared intensity and sensitivity (Fig. 1b). On the receiving end, the OPAG-22 was powered by two automobile batteries in series to provide ~ 25 V DC. The nominal field of view of the spectrometer is 30 milliradians (mrad), which was reduced to 10 mrad by the receiver telescope. The interferometer uses dual retro-reflecting cube corner mirrors in an inverted pendulum mechanism that does not need alignment in the field. The FTIR used a Stirling-cycle cooled mercury cadmium telluride (MCT) detector with a proprietary software correction for nonlinearity (Keens, 1990). Spectra were recorded at a resolution of 1.5 cm⁻¹ and 50 scans were co-added to give increased signal-to-noise ratio at a time resolution of 134s per spectrum. After aligning the telescopes, an ambient emission spectrum was recorded with the source turned off. This spectrum accounts for emission from the ambient-temperature environment which is modulated by the inACPD

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terferometer and detected in the FTIR spectra. The ambient emission spectrum was subtracted from all measured globar spectra before further analysis.

The emission-corrected sample spectra were then analyzed either directly as singlebeam spectra, or as transmission spectra ratioed to a background air spectrum taken before the fire. Ratioing to background was used only in spectral regions where the continuum spectrum of the source-telescope-interferometer system was complex and could not be fitted well by the analysis procedure. The background spectrum was also used to characterize the composition of the pre-fire atmosphere. Analysis was by iterative non-linear least-squares fitting of the measured spectra by calculated spectra as described in previous work (Griffith, 1996; Yokelson et al., 2007a; Burling et al., 2011; Griffith et al., 2012). The calculated spectra are based on HITRAN (Rothman et al., 2009) and Pacific Northwest National Laboratory (Sharpe et al., 2004; Johnson et al., 2006, 2010) spectral databases, and include the effects of environmental pressure and temperature as well as the instrument line shape and resolution. Spectra were analyzed in domains of typically 10-200 cm⁻¹ width, with each region targeting one or more trace gases (see Table A1 in Appendix A for all species reported and the spectral analysis regions from which they were retrieved). Typical precision (repeatability) of measurements is < 1 % for dominant species such as CO₂, CO and CH₄, but accuracy may be a few percent, varying from species to species; Smith et al. (2011) provide a detailed analysis of the accuracy of OP-FTIR measurements. Detection limits for trace species are typically 1–10 ppb. Excess mixing ratios for any species Xdetected when smoke filled the optical path (denoted ΔX , the mixing ratio of species X in a smoke plume/layer minus its mixing ratio in background air) were obtained directly from the transmission spectra or by difference between the appropriate single beam retrievals for H₂O, CO₂, CO, and CH₄ in the plume and pre-fire. All the retrieved excess mixing ratios are listed in the Supplement by individual species for each fire (Table S1).

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In addition to measurements made by the OP-FTIR, two closed-cell FTIR systems were employed: (1) an Airborne FTIR (AFTIR) to sample lofted fresh and photochemically aged smoke (Fig. 1c), and (2) a mobile, LAnd-based FTIR (LAFTIR) system to sample point sources of smoldering smoke (Akagi et al., 2013). This work will focus primarily on gas-phase species measured by the OPAG-22 (hereafter referred to as the OP-FTIR) system, but it is instructive to compare with the other FTIRs at times. Whole air sampling (WAS) canisters were also used on the ground and in the air to measure an extensive suite of gases (mostly non-methane organic compounds, NMOCs) and are reported in Akagi et al. (2013).

2.3 Calculation of emission ratios (ERs) and emission factors (EFs)

Excess mixing ratios (EMRs) for FTIR species were calculated following the procedure in Sect. 2.1. The molar emission ratio (ER) is calculated by dividing ΔX by the EMR of a reference species ΔY , usually ΔCO or ΔCO_2 , measured in the same fresh smoke sample as "X". Since all species are retrieved from the same spectrum at the same time, emission ratios can be determined for any pair of species at each spectrum time-step (for the OP-FTIR ~ 134s). In this study, we first combined all the OP-FTIR measurements from each fire to compute a single fire-averaged initial emission ratio (and $1-\sigma$ standard deviation) for each fire. We computed the fire-averaged ERs from the slope of the linear least-squares regression line with the intercept forced to zero when plotting ΔX against ΔY (Yokelson et al., 1999). The intercept is forced to zero because the background concentration is typically well known and variability in the plume can affect the slope and intercept if the intercept is not forced. This method heavily weights the large excess mixing ratios that may reflect higher rates of fuel consumption and data that have higher signal-to-noise. For NH₃ and CH₃COOH, for unknown reasons, there was a large positive intercept in the plots versus CO so the intercept was not forced. For comparison we also summed the excess amounts of X and Y over time

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and took the ratio $\Sigma\Delta X/\Sigma\Delta Y$ as an alternate estimate of the ER. The ERs calculated by this summation method were within 20% of those calculated using the regression method. For example, the ER(Δ CH $_3$ OH/ Δ CO) on the first fire was 0.0209 or 0.0193 using the summation or regression method, respectively. The summation method is intrinsically more sensitive to the duration of the measurements as opposed to peak emissions because each datum is weighted equally. Both methods give similar ERs and we choose the regression method to emphasize measurements collected with high SNR during the most intense periods of combustion.

The source ERs can be used to calculate EFs expressed as grams of compound emitted per kilogram of biomass burned (on a dry weight basis). A set of ERs obtained at any point during the fire could be used to calculate a set of EFs relevant to the time of the sample. For this study we use fire-averaged ERs (obtained as described above) to calculate a set of fire-averaged EFs for each fire using the carbon mass-balance method (Yokelson et al., 1996, 1999) illustrated by Eq. (1):

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$$\mathsf{EF}(\mathsf{g}\,\mathsf{kg}^{-1}) = F_{\mathsf{C}} \times 1000 \times \frac{\mathsf{MM}_{X}}{\mathsf{MM}_{\mathsf{C}}} \times \frac{C_{X}}{C_{\mathsf{T}}} \tag{1}$$

where $F_{\rm C}$ is the mass fraction of carbon in the fuel, ${\rm MM_X}$ is the molecular mass of compound X, ${\rm MM_C}$ is the molecular mass of carbon (12.011 g mol⁻¹), and $C_{\rm X}/C_{\rm T}$ is the number of emitted moles of compound X divided by the total number of moles of carbon emitted. $C_{\rm X}/C_{\rm T}$ can be calculated directly from the fire-averaged ERs and consideration of the number of carbon atoms in a species. This method is most accurate when the mass fraction of carbon in the fuel is precisely known and all the burnt carbon is volatilized and detected. Based on literature values for similar fuels (Susott et al., 1996; Burling et al., 2010) we assumed a carbon fraction of 0.50 by mass on a dry weight basis for fuels burned in this campaign. The actual fuel carbon fraction was likely within 5–10 % of this value. Note that EFs scale linearly with the assumed fuel carbon fraction. Total emitted carbon in this study was determined from the sum of the carbon from species quantified from the OP-FTIR spectra. This sum underesti-

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Because the emissions from flaming and smoldering processes differ, we use the modified combustion efficiency, or MCE, to describe the relative contribution of each of these combustion processes, where higher MCEs indicate more flaming combustion (Ward and Radke, 1993; Yokelson et al., 1996) (Eq. 2):

$$MCE = \frac{\Delta CO_2}{\Delta CO_2 + \Delta CO}$$
 (2)

2.4 Field campaign site description

Fort Jackson is located at 34.05° latitude and -80.83° longitude just northeast of Columbia, SC in the southeastern US. The fires took place on 30 October, 1 November, and 2 November 2011 and are hereafter referred to as the Blocks 6, 9b, and 22b fires, respectively. Information regarding fuels, weather, size, location, etc. for the three prescribed fires sampled in this study can be found in Akagi et al. (2013).

Fort Jackson Army Base lies at the inland edge of the South Carolina coastal plain in the Sandhills ecosystem, which supports a distinctive type of vegetation. The overstory is dominated by two native pine species, longleaf pine (*Pinus palustris*) and loblolly pine (*Pinus taeda*), and also features turkey oak (*Quercus cerris*). In low density pine areas the understory has a diverse herbaceous layer with little bluestem (*Schizachyrium scoparium*) and scrub oak (turkey oak) regeneration. In high density pine areas associated with fire exclusion there is high degree of canopy closure, which results in less understory vegetation and relatively more duff/litter composed primarily of pine needles. In Block 9b (burned on 1 November) there was significant growth of farkleberry (*Vaccinium arboretum*) intermixed with the mature stands of pine. This fire-adapted community typically is burned every 5–10 yr to maintain forest health and also provide suitable army training grounds (www.dnr.sc.gov/cwcs/pdf/habitat/SandhillsHabitat.pdf), but the plots burned in this study were specifically selected to include stands with no

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recent prescribed fire. Blocks 6, 9b, and 22b had not been burned since 1957, 1956, and 2003, respectively. In addition, the plots were ignited under drought conditions in an effort to create a scenario closer to that of a wildfire. Thus, the Fort Jackson fires provide a contrast to the Camp Lejeune fires sampled earlier in this series of studies by Burling et al. (2011), which occurred on regularly burned sites during a wet spring.

2.5 Three-pronged sampling approach

Three FTIRs – the OP-FTIR, LAFTIR, and AFTIR – were used collectively at different temporal and spatial scales to provide an enhanced understanding of smoke emissions from different combustion processes over the duration of the fire. The OP-FTIR successfully sampled smoke generated by ignition activities near the measurement path; post-ignition, wind-blown smoke from the wake of the receding, local flame front; occasional smoke from more distant combustion; and any upwind residual smoldering combustion emissions directed through the open path. The OP-FTIR initially captured mostly flaming emissions and then a mix of flaming and smoldering emissions. All these emissions originated in a difficult to define upwind source area that probed a constantly changing portion of the burn unit. This is partly because fires can create local wind gusts and downdrafts in a complex manner and partly due to inherently light winds on the days of these measurements. The AFTIR sampled flaming emissions mixed with entrained smoldering emissions in the single convection column that was generated by each burn. The convection column is not developed enough for airborne sampling until sufficient fire has been applied to site. AFTIR measurements indicate that overall emissions from the burn blocks peaked ~ 170-200 min after initial fire ignition while the OP-FTIR concentrations peaked approximately 10-30 min following local ignition. Finally, the ground-based LAFTIR system captured point-sources of RSC smoke after the flame front had passed through the sample area. Our initial expectation was that both ground-based FTIRs would observe much lower MCEs than the AFTIR and that the OP-FTIR data would help us weight the relative contribution of the point sources sampled by the LAFTIR to the overall ground-level smoke layer. However, because of

the rarity of RSC point sources on these prescribed fires, the LAFTIR system often roved to locations whose emissions were not probed by the OP-FTIR, which in turn often sampled drift smoke whose source was not sampled by the LAFTIR. In summary, these three FTIR approaches provide a comprehensive probe of the combustion emissions, each with its own advantage. Each instrument followed different protocols as detailed below.

2.5.1 **OP-FTIR**

Unlike the LAFTIR and AFTIR, the OP-FTIR was set up before the burns on a preselected portion of the fire perimeter. For each fire the OP-FTIR was positioned to capture the downwind smoke emitted shortly after the fire ignition commenced. Figure 2 shows the burn blocks at Fort Jackson and the relative placement of the OP-FTIR for each fire. After ignition, the OP-FTIR sampled a variety of emissions as discussed above. Figure 3 shows the OP-FTIR time series of MCE and excess CO (ppm) that can be used as indicators of combustion type/intensity for each fire. OP-FTIR, AFTIR and LAFTIR sampling time periods and fire ignition times are also shown.

For the Block 6 fire, light and variable winds were from the northeast and the OP-FTIR was positioned along the southwest perimeter of the fire area with an optical path of 32.2 m (Fig. 2a). A backing fire was started at 12:24 p.m. local time (LT, EDT) on the southwestern perimeter of the burn block along the same firebreak as the OP-FTIR setup. The heading fire was initiated at the opposite end of the block at 13:35 LT, with more backfires lit to increase the fire intensity at \sim 15:20 LT. The most intense column of smoke of the day measured by the airborne platform was sampled \sim 25 min later around 15:46 LT (Fig. 3a).

For the Block 9b fire, light winds (typically $3-4\,\mathrm{m\,s^{-1}}$) were from the north and the OP-FTIR was placed on the south side along an east—west road with an optical path of 29.3 m (Fig. 2b). A backing fire was lit near the OP-FTIR at \sim 11:15 LT and produced very heavy smoke with the high intensity reflected in the high starting MCE and high levels of excess CO (Fig. 3b). The headfire was ignited at 13:46 LT.

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2.5.2 **AFTIR**

The AFTIR airborne sampling strategy is detailed in Akagi et al. (2013). To measure the initial emissions, lofted smoke less than several minutes old was sampled by penetrating the smoke column 150 to several thousand meters from the flame front (Fig. 1c). The smoke sampled by the AFTIR was produced mainly from flaming emissions and any smoldering emissions that became entrained in the single main updraft core. AFTIR sampling periods and peak smoke samples are seen in Fig. 3.

2.5.3 LAFTIR

The LAFTIR ground-based sampling protocol was similar to that described in Burling et al. (2011) and Akagi et al. (2013). Backgrounds were acquired before the fire. Ground-based sampling access was sometimes precluded during ignition, but sampling access then continued through late afternoon until the fire was effectively out. During post-ignition access, numerous point sources of RSC were sought out and sampled with the LAFTIR system minutes to hours after passage of a flame front. The spot sources of white smoke, mainly produced from pure smoldering combustion, included smoldering stumps, fallen logs, litter layers, etc., and they contributed to a dense smoke layer usually confined below the canopy. The LAFTIR sometimes sampled in the vicinity of the OP-FTIR, but frequently roved to other areas. The LAFTIR sampling period for each fire is shown in Fig. 3.

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3.1 Initial emissions

Fire-average MCEs and emission factors measured by OP-FTIR for each fire are shown in Table 1. The MCEs across all fires showed minimal variability with a fire-average of 0.912 ± 0.004 . The fire-averaged MCEs from the LAFTIR (0.842 ± 0.046) and AFTIR (0.929 ± 0.008) indicate larger contributions from smoldering and flaming combustion, respectively. The intermediate OP-FTIR MCE is consistent with both smoldering and flaming emissions being transported to the OP-FTIR path.

Fire-average EFs are important when assessing overall fire characteristics, or when comparing to other fire-average EF in the literature. That being said, the drop in OP-FTIR MCE seen partway through each Fort Jackson fire (Fig. 3) suggests that EF computed separately for "early" and "late" time blocks would be mainly indicative of flamingand smoldering-dominated combustion, respectively, and in fact, the calculation of OP-FTIR EF for early and late phases did improve the comparison to EF measured from other platforms. It should be noted that not all fire measurements show a fast transition between high and low MCE (Yokelson et al., 1996) and/or the division between "early" and "late" can be indistinct. However, this informal separation is one useful way to probe the dynamic mix of flaming and smoldering combustion and compare to other platforms. For the Block 6 fire, "early" was defined from the first OP-FTIR sample (at the start of the fire, 12:38:25 LT) until 13:47:00 LT when a noticeable drop in MCE is observed (Fig. 3a, black, upper trace). This signifies a change in the composition of the sampled smoke from mostly flaming combustion to more smoldering combustion, as discussed earlier. This change is also noted in the ER plots for several species, including CO and methanol (Fig. 4). Both species are primarily produced from smoldering combustion and thus, a higher ratio of ΔCO/ΔCO₂ and ΔCH₃OH/ΔCO was observed when sampling "late" smoke that had a greater contribution from unlofted RSC (shown in blue).

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It is of interest to compare the emission factors from all three FTIRs employed during the Fort Jackson burns as each FTIR had a different spatial and temporal perspective on the overall combustion emissions. Figure 5 shows a side-by-side comparison of OP-FTIR, LAFTIR, and AFTIR fire-averaged emission factors from the Fort Jackson fires. The study-average MCEs (average MCE of all three Fort Jackson fires) were 0.842 ± 0.046, 0.912 ± 0.004, and 0.929 ± 0.008 for the LAFTIR, OP-FTIR, and AFTIR platforms, respectively. We observe a trend for some smoldering species whose emissions depend more strongly on MCE than fuel type (e.g. CH₄, CH₂OH, furan) – namely: EF(LAFTIR) > EF(OP-FTIR) > EF(AFTIR), which is consistent with the trend in FTIR fire-averaged MCEs. For other compounds whose emissions are typically more fuel dependent, an EF(OP-FTIR) > EF(LAFTIR) > EF(AFTIR) trend was observed (e.g. CH₃COOH, HCHO, NH₃), except for EF(C₂H₂) (a flaming compound) where EF(AFTIR) was greater than EF(LAFTIR). Some of the EF that were higher for OP-FTIR were higher despite an intermediate MCE and they are also known as "sticky" compounds that are difficult to sample in closed-cell systems (NH3 and HCOOH). Losses on the cell walls were measured and corrected for in both closed cell FTIR systems according to a protocol developed by Yokelson et al. (2003) who compared AFTIR and OP-FTIR systems under controlled conditions in well-mixed laboratory smoke samples. If the passivation corrections (40–100%) were accurate, then the higher study-average EF by OP-FTIR for some species in this work may largely be due to sampling emissions from a different mix of fuels. This idea is supported by the fact that EFs for HCHO and C₂H₄, which are smoldering compounds that do not suffer from wall losses, are also higher in OP-FTIR than the closed cell systems. In addition, the NH₃ EFs agree well for the LAFTIR and OP-FTIR "late" period on one fire (Block 9b). However, we must also consider the clear advantage of open-path measurements for such species and if the closed cell correction factors are too small, then fires may emit more NH₃ than our previous closed-cell measurements indicate (Williams et al.,

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1992). Resolving this question would require more controlled tests. Fuel differences also could explain some internal inconsistency in the OP-FTIR results. For instance, on the Block 9b fire, the EF for NH₃ and CH₃COOH are twice as large for the early "flaming dominated" OP-FTIR samples as they are for the later "smoldering dominated" samples; despite the fact that these compounds are well-known to be associated with smoldering emissions. The OP-FTIR may be relatively more influenced by recirculated emissions from burning live fuels early in the fire. Part of any difference in fuels probed by the two ground-based systems can likely be traced to working in different areas and the sampling procedure of the LAFTIR, which is based on seeking and sampling sources of visible white smoke. Glowing combustion also occurs during RSC and produces high levels of some compounds (e.g. NH₂), but less visible smoke (Fig. 3 in Yokelson et al., 1997). Regardless of the reason for the study-average differences (for NH₃ and HCOOH in particular) between the OP-FTIR and the other FTIRs, the EF from the OP-FTIR are significant, because the OP-FTIR gives a relevant view of the groundlevel smoke. The fraction of unlofted emissions produced by a fire is hard to measure, likely variable, and could be large for some fires. Further, the experiment suggests that the ground level smoke layer is not essentially a linear combination of the output from visible RSC sources.

Emission factors for "early" and "late" smoke measured by OP-FTIR from the Fort Jackson fires are shown in Table 2. CO had a large EF range, with EF(CO) "late" being almost twice as large as EF(CO) "early". As mentioned above, we see higher EF for some smoldering compounds like methane and methanol late in the fire associated with lower MCE, but with mixed, somewhat anomalous results likely rooted in fuel differences for other species such as: ammonia, ethylene, acetic acid, formaldehyde, and formic acid. We also represent data in Table 2 as a bar chart (Fig. 6). Methane and methanol EF appear to follow a decreasing "step-wise" trend from smolderingdominant to flaming-dominant platforms, correlating with low to high MCE. Trends are not so straightforward for EF(C₂H₄), EF(CH₃COOH), or EF(NH₃) whose emissions typically depend more on fuels.

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The extent to which EFs correlate with MCE for a given species across all three platforms is best shown by EF vs. MCE plots (Fig. 7), where a strong negative correlation of EF with increasing MCE is seen for species like CH₄ and CH₃OH. The EF that appear to deviate from the trend-line are OP-FTIR EF sampled late in the fire's progression (OP-FTIR "late"), especially for EF(CH₄). Nonetheless, we observe reasonable correlation across different instruments and time periods. Burling et al. (2011) report strong correlations (high R^2 values) for these same compounds when comparing airborne measurements in temperate conifer forests from US and Mexico, suggesting EF(CH₄) and EF(CH₃OH) versus MCE relationships are fairly robust across studies in nominally similar fuel types. Additionally, the high correlations shown in Fig. 7 suggest a consistent relationship between the EF obtained and the flaming to smoldering ratio each instrument can sample.

3.3 OP-FTIR comparisons with the literature

We can compare the OP-FTIR EF with those from a study that employed a similar open-path FTIR to measure biomass burning emissions from South African savanna fires (Wooster et al., 2011). The fire-averaged MCEs or ΔCO/ΔCO₂ from Wooster et al. $(0.913 \pm 0.026, 0.095)$ are similar to those in this work $(0.912 \pm 0.004, 0.095)$. This similarity in fire-average MCE or $\Delta CO/\Delta CO_2$ is surprising considering pineunderstory and savanna fuels are intrinsically guite different and have been measured from airborne platforms at different MCE and $\Delta CO/\Delta CO_2$ (0.931 ± 0.016 (0.074) and 0.944 ± 0.012 (0.059) respectively, Akagi et al., 2011, 2013). Savannas are usually dominated by fine fuels that burn at high combustion efficiency (Akagi et al., 2011) and do not often include large diameter fuels that are highly susceptible to prolonged smoldering. Temperate pine understory ecosystems often have more dead/down debris and below-ground fuels like organic soils that tend to burn by smoldering and/or RSC although that is minimized in prescribed fires. The Wooster et al. (2011) fires were not sampled by an airborne platform, thus, we cannot compare both OP-FTIR and AFTIR MCEs between the studies.

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We can also compare emissions for several other species from this work and Wooster et al. (2011) (Fig. 8). Emission factors from this work are all within the natural variability of those reported by Wooster et al. (2011), except for NH₃. Multiple factors can affect ammonia emissions, the most important being the nitrogen content 5 of the fuel. Measurements in Wooster et al. (2011) were acquired at Kruger National Park where elephant dung is a major fuel component. Dung is known to have a higher nitrogen content compared with other biomass types (Christian et al., 2007; Keene et al., 2006). While the N content of fuels sampled in this work and in Wooster et al. is unknown, higher fuel N could explain why EF(NH₃) was significantly higher in Wooster et al. (2011).

3.4 Estimating fire-line exposure to air toxics

Smoke can affect human health via numerous, complex, and poorly understood mechanisms. Firefighters, burn managers, and other wildland fire personnel are subjected to a complex mixture of combustion-generated gases and respirable particles that includes toxic and carcinogenic compounds. Woodsmoke contains at least five chemical groups that are classified as known human carcinogens by the International Agency for Research on Cancer (IARC), other species classified by the IARC as probable or possible human carcinogens, and at least 26 chemicals listed by the US EPA as hazardous air pollutants (Naeher et al., 2007). Adverse health effects caused by smoke emitted during a fire can include upper respiratory symptoms (Swiston, 2008), neurological symptoms, and potentially cancer (though previous studies have not found a strong link between the two, Demers et al., 1994). Only a few studies in the literature have evaluated occupational exposure to smoke among firefighters (Materna et al., 1992; Reinhardt and Ottmar, 1997; Reinhardt and Ottmar, 2004; Adetona et al., 2011).

In many countries, laws and guidelines have been established to reduce exposures to potentially harmful levels of toxins. Average concentrations over time and peak exposures are both of concern: In the US, the Occupational Safety and Health Administration (OSHA) sets legal exposure limits known as permissible exposure limits (PELs)

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and short-term exposure limits (STELs) for these two cases, respectively. A PEL is a time-weighted average (TWA) concentration not to be exceeded for routine 8 h exposure while a STEL should not be exceeded for any 15-30 min period. The National Institute of Occupational Safety and Health (NIOSH) provide Recommended Exposure Limits, or RELs, as TWA concentrations for an 8 or 10 h workday. NIOSH also reports STELs as a 15 min maximum exposure. NIOSH limits, being guidelines, are often more conservative than those enforced by OSHA (Sharkey, 1997). In addition to NIOSH, the American Conference of Industrial Hygienists (ACGIH) sets exposure guidelines known as Threshold Limit Values (TLVs). The ACGIH TLV is an 8 h TWA and the TLV STEL is a 15 min maximum exposure. In our analysis we report a range when more than one exposure limit/quideline is available.

Measuring fire-line exposures to various toxins present in smoke for comparison to OSHA, NIOSH, or ACGIH exposure limits, is not simple. Fire intensity, fuel composition, and weather conditions are constantly changing and directly affect the fire dynamics and smoke dilution occurring in the work environment (Sharkey, 1997). Different fire types also pose different conditions; several studies have shown that exposures to pollutants were higher among firefighters at prescribed fires than at wildfires (Reinhardt and Ottmar, 2004; Sharkey, 1997). In addition to fire type, smoke exposure can vary by work activity (e.g. direct attack, lighting, mop-up) (Reinhardt and Ottmar, 2004). For the typical morning prescribed burn, increasing afternoon winds may increase smoke distribution and risk of smoke overexposure for some workers. Various measurement techniques, including electronic dosimeters, liquid chromatography, gas chromatography/flame ionization detection (FID) have been employed to measure different species. This work is the first to assess fire-line exposure using the open-path FTIR technique.

Table 3 shows measured TWA burn-average and peak exposures for CO and HCHO from this work, other works (Reinhardt and Ottmar, 2004), and the recommended TWA (8 h) and STEL exposure ranges. We first compare OP-FTIR burn-average TWA concentrations to those from Reinhardt and Ottmar (2004), who report a frequency distribution of fire-line exposures as a cumulative percent of sampled firefighters measured

from prescribed burns in the Northwest. The CO burn-average mixing ratio exposure for firefighters in the 50th percentile from Reinhardt and Ottmar (2004) was slightly higher (by 8.6%) than the burn-average concentration measured in this work, while their HCHO 50th percentile concentration was approximately a factor of two lower than in our work. Location, fuel, weather, and fuel moisture are just some of the variables that could have created very different burn conditions between our study and that of Reinhardt and Ottmar (2004). Burn-averaged exposures from the OP-FTIR can also be compared with recommended TWA exposures. Our burn-average ΔCO was below all the recommended exposure levels while our burn-average ΔHCHO was near the lower end of exposure guidelines (0.016–0.75 ppm range). Thus, Fort Jackson ΔCO and ΔHCHO did not exceed OSHA guidelines suggesting that prolonged exposures were a limited problem for these compounds during the Fort Jackson fires.

The average peak mixing ratios for CO and HCHO measured by the OP-FTIR and LAFTIR for the three fires and the recommended STEL (15 min) exposure ranges are also shown in Table 3. OP-FTIR peak CO levels are a factor of 20 lower than the peak point exposures measured by Reinhardt and Ottmar (2004); which are 3.6 times lower than LAFTIR peak Δ CO point values. OP-FTIR Δ CO and Δ HCHO peak mixing ratios fall below the range of recommended STEL mixing ratios, but the LAFTIR peaks (at upper end of range) exceed CO and HCHO STELs by factors of 3.2 and 3.8, respectively. While these exceedances are important, we note that LAFTIR values represent a mostly avoidable upper limit, as these mixing ratios were measured by placing the sample line within several feet of smoldering point sources.

Thus far we have limited our discussion of air toxins to CO and HCHO, though many others exist. Exposure to the other air toxins not measured by the OP-FTIR can be estimated using published normalized excess mixing ratios ($\Delta X/\Delta CO$, or NEMR, where "X" is an air toxin whose ratio to CO in smoke was measured in another study) multiplied by the OP-FTIR burn-average CO. Exposure estimates have previously been derived this way by Austin (2008) who used published emission factors and ceiling limits to calculate "hazard ratios". We use a slightly different approach: we estimate

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TWA and peak exposures of high risk compounds using a recent comprehensive set of pine-understory prescribed fire emission ratios from Yokelson et al. (2013) and multiply those ER by the OP-FTIR burn-average and peak ΔCO. For air toxins measured both by OP-FTIR and Yokelson et al. (2013) we can "test" this approach by comparing "calculated" vs. "measured" exposures (for HCHO, CH₃OH, NH₃, see Table A2 in Appendix A). In most cases the calculated mixing ratios are lower than the measured mixing ratios (by up to 65%), except for HCHO and NH₃ measured by the LAFTIR; e.g., the greatest deviation from 1 was the calculated/measured value of 6.60 for the NH2 LAFTIR peak exposure. Given such a high ratio (based on comparison to AFTIR measurements from 2010) it is clear that this estimation technique is less applicable for Ncontaining compounds since their emissions depend strongly on fuel N (Burling et al., 2011). It is also important to note that the emissions data from Yokelson et al. (2013) are for the 2010 pine understory prescribed fires at Camp Lejeune that were lit after a wet spring versus old growth stands lit after a prolonged drought that the OP-FTIR sampled at Fort Jackson. Excluding the high calculated NH₃ value mentioned above, the average calculated/measured ratio and 1- σ standard deviation is 0.69 ± 0.38. Thus, smoke is variable, but this method is still useful to estimate exposures for unmeasured compounds of interest.

Based on this methodology we present estimated exposure to many air toxins not measured in this work, but measured in the lab or air for prescribed fires and reported in Yokelson et al. (2013) (Table 4). All of the species listed in Table 4 are designated as hazardous air pollutants, or harmful and potentially harmful constituents in tobacco smoke as noted by Yokelson et al. (2013). Our estimated fire-line TWA exposures based on OP-FTIR burn average CO are significantly lower than recommended TWA exposure limits (a factor of 10 lower at the least), suggesting that reasonably cautious personnel on the Fort Jackson fires likely did not exceed individual recommended exposure limits for the hazardous compounds listed in Table 4. Even estimated peak exposures based on LAFTIR peak CO were lower than recommended STELs except for acrolein and HCN, which exceeded STELs by factors of 3.7 and 1.2, respectively.

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We also show estimated exposures divided by the recommended TWA exposure limits, or E_{ν} , where X is a given compound of interest. E_{ν} can be used to calculate a unitless irritant exposure mixture term $E_{\rm m}$, where $E_{\rm m} = E_{x1} + E_{x2} + E_{x3+} \dots$ (Reinhardt and Ottmar, 2004). For example, E_x for compounds such as acrolein and formaldehyde 5 can be summed and if E_m exceeds 1, then the combination of the irritants exceeds the combined exposure limit (Sharkey, 1997). Only considering acrolein (Table 4) and formaldehyde (Table 3), we report a TWA combined irritant exposure E_m of 0.31 which is not in exceedance of OSHA limits but only lower by a factor of ~3, showing that combined TWA exposures are a greater concern than TWA exposures assessed individually. However, we note that the exposure mixture equation is a simplification of complex phenomena and it is unlikely that the effects of toxins add linearly (Yokelson et al., 2013; Menser and Heggestad, 1966; Mauderly and Samet, 2009). E_m is used as an estimate of combined exposure effects as the actual synergistic effects of a given pollutant combination are unknown. Additionally, we ignore the effects of particles which likely affect exposure limits for individual and combined species (Pope and Dockery, 2006; Adetona et al., 2011).

In this work, measured OP-FTIR TWA fire-line mixing ratios and calculated fire-line mixing ratios based on OP-FTIR CO did not exceed recommended exposure limits for any individual species. Combined exposure limits were also not exceeded, but they were more likely to approach recommended limits. Peak mixing ratios measured by the LAFTIR violated STELs for CO and HCHO and calculated peak exposures based on LAFTIR peak CO levels were higher than the STELs for acrolein and HCN. These compounds are some of the more serious irritants and carcinogens that the fire-line may be exposed to. While peak exposures were often observed during initial attack in a study of wildland firefighters (Sharkey, 1997), we show that smoldering combustion measured after the flame front has passed through an area could cause problematic exposures if not carefully avoided. This work agrees with previous works that "shiftaverage" TWA exposures may be less of a problem than peak exposures (Sharkey,









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1997; Reinhardt and Ottmar, 2004; Austin, 2008), however, combined TWA exposures must be considered for a more realistic assessment of fire-line risk.

4 Conclusions

We measured trace gas emission factors for three prescribed fires at Fort Jackson, SC using an open-path FTIR. The fires occurred outside the common range of conditions for southeastern US prescribed fires because they mainly were in stands that had not been burned by prescribed fire in decades and the stands had recently been subject to drought. Thus, the emissions may be somewhat relevant to a scenario where frequency of prescribed fire is reduced, or to that of a wildfire.

The OP-FTIR provided a ~ 30 m, path-integrated sample of upwind smoldering emissions and some flaming emissions produced primarily after adjacent ignition and before the local flame front achieved significant vertical column development. We compared OP-FTIR fire-average emission factors (EF) with fire-average EF for residual smoldering combustion measured on the same fires by a roving, land-based FTIR (LAFTIR) and fire-average EF measured with an airborne FTIR (AFTIR) that sampled in the single convection columns produced by the fires (Akaqi et al., 2013). We observed a consistent trend in EF for the smoldering species that were not highly fuel dependent (i.e. CH₄, CH₃OH), where EF(LAFTIR) > EF(OP-FTIR) > EF(AFTIR). We also observed a decrease in MCE between the "early" and "late" phases of the OP-FTIR measurements indicative of a general shift from flaming-dominated combustion (immediately after adjacent ignition) to smoldering dominated combustion. Emission factors were calculated separately for these "early" and "late" time blocks to further probe the temporal change in the emissions transported to the fixed OP-FTIR path and facilitate a more detailed comparison with other EF measurements made on the same fires, but from other platforms. For CH₄ and CH₃OH, the "early" OP-FTIR EF were most similar to EF(AFTIR) while "late" OP-FTIR EF were most similar to EF(LAFTIR), which is not surprising given the MCE dependence of these species. For other gases there was

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large scatter in the fire-to-fire and species-to-species comparisons, suggesting that the various platforms preferentially sampled the emissions from different fuels. This is not surprising given the high natural variability of the fire environment coupled with the spatial separation between the systems. The largest differences were seen for NH₃, which was higher by ground-based OP-FTIR than from an aircraft. The fraction of unlofted emissions is not highly constrained suggesting that some prescribed fires, or other fires, may produce higher overall NH₃ emissions than would be implied by airborne measurements (Griffith, 1991; Wooster 2011). Overall, data from the three FTIR sampling methods employed showed that the method in which the smoke was sampled strongly influenced the relative abundance of the emissions that were observed.

We also compared our pine-understory prescribed fire EF to EF measured on prescribed African savanna fires by a system similar to the OP-FTIR. The EF were very similar between the two studies despite the fact that the fires burned in very different ecosystems, fuel types, weather conditions, etc. This provides further evidence that MCE and trace gas EFs can be highly dependent on measurement platform.

Our initial expectation was that both ground-based FTIRs would observe much lower MCEs than the AFTIR and that the OP-FTIR data would help us weight the relative contribution of the point sources sampled by the LAFTIR to the ground-level smoke layer. Our results show that the open path system measured an average MCE closer to that of the airborne system than the point sources that could be sampled on the ground. This suggests that local ignition before plume development and to a lesser extent, downdrafts after plume development, contribute significantly to the ground level smoke layer and not just a filling of the sub canopy layer by local, visible point sources. While the airborne measurements provide the best fire integrated sample in the absence of abundant RSC, the characterization of the ground-level smoke layer is more interesting than we anticipated. While the LAFTIR enables modeling of specific RSC fuels, the OP-FTIR is likely a less biased sample of the ground-level smoke layer and there was not a simple linear combination of the LAFTIR-measured fuel-specific EF that matched the OP-FTIR EF. This disconnect resulted at least partially from the rarity of RSC on

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these prescribed fires and therefore the need for the LAFTIR system to rove to locations whose emissions were not probed by the OP-FTIR. Despite the uncertainty due to high spatial variability in the ground level environment, it seems likely that active sampling of visible smoke sources may bias a measurement of the overall ground level smoke layer (this bias could occur by undersampling both glowing combustion and recirculated smoke from more distant flaming sources). More coordinated ground-based sampling of emissions and fuel consumption would be of value in future experiments.

Average and peak OP-FTIR mixing ratios and peak LAFTIR mixing ratios were compared to recommended TWA and peak guidelines put forth by OSHA, NIOSH, and ACGIH. We also estimated TWA and peak exposures for many air toxins not measured in this work by ratioing NEMR from another more comprehensive study to our real fireline CO data. This is an important approach to estimating exposures since it would be difficult to deploy large amounts of advanced instrumentation on a fire-line. TWA individual and combined estimated exposures did not exceed recommended guidelines although measured and calculated LAFTIR peak mixing ratios, which represent avoidable exposures, did exceed STELs for four compounds: CO, HCHO, acrolein, and HCN. Finally, our data support previous findings that peak exposures are more likely to challenge permissible exposure limits than average exposures suggesting it is important for wildland fire personnel to avoid concentrated smoldering smoke to minimize their risk of overexposure.

Supplementary material related to this article is available online at: http://www.atmos-chem-phys-discuss.net/13/18489/2013/ acpd-13-18489-2013-supplement.zip.

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Table 1. MCE and emission factors (g kg⁻¹) for three pine understory burns measured by OP-FTIR.

MCE		Block 6 0.917	Block 9b 0.911	Block 22b 0.909	Average 0.912	Std dev 0.004
Species	Formula					
Carbon Dioxide	CO ₂	1652.8	1642.5	1645.3	1646.9	5.3
Carbon Monoxide	CO	94.8	102.1	104.9	100.6	5.2
Methane	CH ₄	2.62	2.70	2.72	2.68	0.05
Ethylene	C_2H_4	1.67	1.58	1.69	1.65	0.06
Ammonia	NH_3	0.54	0.38	0.97	0.63	0.30
Hydrogen Cyanide	HCN	2.11	0.78	_	1.44	0.94
Formaldehyde	HCHO	2.31	2.48	2.69	2.49	0.19
Acetic Acid	CH ₃ COOH	2.96	3.88	2.76	3.20	0.60
Formic Acid	HCOOH	0.40	0.38	0.31	0.36	0.04
Methanol	CH ₃ OH	2.09	2.00	1.88	1.99	0.11
Acetylene	C_2H_2	0.81	0.15	_	0.48	0.46
Carbonyls as glyoxal*	$C_2H_2O_2$	1.60	2.01	_	1.80	0.29
Furan	C_4H_4O	_	0.48	_	0.48	_
Sum NMOC		11.83	12.96	9.34	12.46	0.21

^{*} The residual spectrum from 2820 to 2850 cm⁻¹ (after fitting HCHO, CH₄, and H₂O) contained features similar to glyoxal, but shifted by several wavenumbers. The feature may have been due to a mixture of oxygenated compounds (most likely carbonyls), but was analyzed using the glyoxal IR cross-section (Profeta et al., 2011).

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Table 2. Emission factors and MCE for select compounds measured during "early" and "late" blocks by OP-FTIR. Fire-averaged emission factors (g kg⁻¹) from the LAFTIR and AFTIR (Akagi et al., 2013) are also shown.

Fire			OP-FTIR ("early")	OP-FTIR ("late")	LAFTIR	AFTII
Block 6	MCE		0.927	0.869	0.876	0.932
	Carbon Dioxide	CO ₂	1673.2	1574.8	1554	1674
	Carbon Monoxide	CO	83.7	150.8	140	78
	Methane	CH₄	2.16	2.39	5.20	1.74
	Ethylene	C ₂ H ₄	1.75	1.00	0.89	1.21
	Ammonia	NH ₃	0.50	0.61	0.09	0.11
	Hydrogen Cyanide	HCN	1.86	_	0.95	0.74
	Formaldehyde	HCHO	2.25	1.55	1.79	1.87
	Acetic Acid	CH ₃ COOH	2.71	2.26	1.03	1.24
	Formic Acid	HCŎOH	0.41	0.24	_	0.08
	Methanol	CH ₃ OH	1.66	1.99	2.35	1.18
	Acetylene	C_2H_2	0.74	0.50	0.25	0.35
	Carbonyls as glyoxal*	$C_2H_2O_2$	1.40	1.34	_	-
Block 9b	MCE		0.923	0.849	0.858	0.919
	Carbon Dioxide	CO ₂	1665.6	1545.3	1496	1643
	Carbon Monoxide	CO	89.0	174.4	158	92
	Methane	CH₄	2.41	2.11	11.50	2.08
	Ethylene	C ₂ H ₄	1.59	0.98	1.53	1.23
	Ammonia	NH ₃	0.62	0.29	0.23	0.13
	Hydrogen Cyanide	HCN	0.61	_	0.85	0.82
	Formaldehyde	HCHO	2.16	_	2.42	2.11
	Acetic Acid	CH ₃ COOH	3.94	2.03	3.84	0.75
	Formic Acid	HCŎOH	0.32	0.31	_	0.09
	Methanol	CH ₃ OH	1.69	1.69	6.42	1.45
	Acetylene	C ₂ Ḧ ₂	0.13	_	0.22	0.24
	Carbonyls as glyoxal*	C ₂ H ₂ O ₂	1.75	_	_	_
	Furan	C ₄ H ₄ O	0.42	_	-	0.20
Diagle OOL	MOE		0.005	0.007	0.700	0.005
Block 22b	MCE	00	0.935	0.897	0.789	0.935
	Carbon Dioxide	CO ₂	1701.4	1630.5	1305	1679
	Carbon Monoxide	CO	75.5	118.9	222	74
	Methane	CH ₄	1.53	1.94	10.34	2.01
	Ethylene	C ₂ H ₄	1.49	1.12	1.25	0.94
	Ammonia	NH ₃	0.87	0.66	0.33	0.14
	Formaldehyde	HCHO	_	1.86	2.51	1.70
	Acetic Acid	CH₃COOH	2.17	1.89	2.42	1.25
	Formic Acid	HCOOH	0.20	0.22	-	0.11
	Methanol	CH ₃ OH	1.12	1.33	3.60	1.16

^{*} The residual spectrum from 2820 to 2850 cm⁻¹ (after fitting HCHO, CH_A, and H₂O) contained features similar to glyoxal, but shifted by several wavenumbers. The feature may have been due to a mixture of oxygenated compounds (most likely carbonyls), but was analyzed using the glyoxal IR cross-section (Profeta et al., 2011).

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Table 3. Average TWA and peak exposures measured in this work and other studies and recommended TWA and peak exposures.

	CO (ppm)	HCHO (ppm)
Average TWA exposures		
OP-FTIR (burn-average) ^{a, b}	6.351	0.147 ^c
Reinhardt and Ottmar (2004) (burn-average, 50th percentile) ^d	6.9	0.075
Reinhardt and Ottmar (2004) (burn-average, 90th percentile) ^d	23	0.18
Recommended TWA (8 h average) exposure range	25–50 ^e	0.016–0.75 ^f
Peak exposures		
OP-FTIR (max) ^{a, g}	32.16	0.825
LAFTIR (max) ^{a, g}	641.6	7.665
Reinhardt and Ottmar (2004) (max)	> 179	1.460
Recommended STEL (15 min) peak exposure range	200 ^h	0.1–2.0 ⁱ

^a Reported as excess mixing ratios. Absolute values will be slightly higher to account for background concentrations.

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 $^{^{\}rm b}$ The time at the prescribed burns averaged 4:13 h (range \sim 4–5 h)

^c Since we do not report HCHO measured from the start to end of the Fort Jackson fires, this value was estimated as ER(HCHO/CO) × OP-FTIR (burn-average) ΔCO.

^d The time at the prescribed burns averaged 7 h (range 2–13 h)

^e Low and high CO values represent ACGIH TWA TLV and OSHA TWA PEL, respectively.

f Low and high HCHO values represent NIOSH TWA REL and OSHA TWA PEL, respectively.

^g Peak exposures represent the average maximum peak exposure from the three different fires measured.

h NIOSH ceiling and OSHA STEL (5 min)

Low and high values represent NIOSH STEL and OSHA STEL, respectively.

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Table 4. Estimated OP-FTIR TWA burn-averaged and peak concentrations, LAFTIR peak concentrations, and recommended TWA and peak exposures.

	Estimated OP-FTIR TWA exposure (ppm) ^a	Recommended TWA exposure (ppm) ^b	Ex (estimated exposure/ Recommended exposure) ^c	Estimated OP-FTIR peak exposure (ppm) ^a	Estimated LAFTIR peak exposure (ppm) ^a	Recommended STEL peak exposure (ppm) ^d
Acrolein (C ₃ H ₄ O)	0.0109	0.1	1.09E-01	0.055	1.102 ^e	0.3
Ammonia (NH ₃) ^f	0.206	25-50	4.12E-03	0.493	1.106	35
Benzene (C ₆ H ₆)	0.0058	0.1-1.0	5.81E-03	0.029	0.587	1.0-5.0
Hydrogen Cyanide (HCN)	0.0540	10	5.40E-03	0.273	5.456 ^e	4.5
Hydrochloric Acid (HCI)	0.0043	2.0-5.0	8.68E-04	0.022	0.438	3.0-7.0
Acetonitrile (CH ₃ CN)	0.0079	20-40	1.98E-04	0.040	0.801	60
Acetaldehyde (CH ₃ CHO)	0.0385	100	3.85E-04	0.195	3.885	150
Formaldehyde (HCHO) ^f	0.147	0.016-0.75	1.96E-01	0.825	7.665	0.1-2.0
Methanol (CH ₃ OH) ^f	0.1200	200	6.00E-04	0.560	15.65	250
Acrylonitrile (C ₃ H ₃ N)	0.0010	1.0-2.0	5.07E-04	0.005	0.102	10
1,3-Butadiene (C ₄ H ₆)	0.0001	1.0-2.0	7.48E-05	0.0004	0.008	5
Propanal (C ₃ H ₆ O)	0.0043	20	2.14E-04	0.022	0.433	-
Acetone (C ₃ H ₆ O)	0.0150	250-1000	1.50E-05	0.076	1.514	1000
1,1-Dimethylhydrazine (C ₂ H ₈ N ₂)	0.0014	0.5	2.70E-03	0.007	0.136	_
Crotonaldehyde (C ₄ H ₆ O)	0.0074	2.0	3.68E-03	0.037	0.743	_
Acrylic Acid (C ₃ H ₄ O ₂)	0.0013	2.0-10.0	1.33E-04	0.007	0.134	-
Methyl Ethyl Ketone (MEK, C ₄ H ₈ O)	0.0041	200	2.07E-05	0.021	0.418	300
n -Hexane (C_6H_{14})	0.0006	50-500	1.21E-06	0.003	0.061	510
Toluene (C ₆ H ₅ CH ₃)	0.0038	50-200	1.89E-05	0.019	0.381	500
Phenol (C ₆ H ₅ OH)	0.0088	5	1.76E-03	0.044	0.887	15.6
Methyl Methacrylate (C ₅ H ₈ O ₂)	0.0009	50-100	9.21E-06	0.005	0.093	100
Styrene (C ₈ H ₈)	0.0012	20–100	1.16E-05	0.006	0.117	40–200
Xylenes (C ₈ H ₁₀)	0.0031	100	3.07E-05	0.016	0.310	150-200
Ethylbenzene (C ₈ H ₁₀)	0.0009	100	8.95E-06	0.005	0.090	125
Naphthalene (C ₁₀ H ₈)	0.0038	10	3.83E-04	0.019	0.387	15
Isocyanic Acid (HNCO) ^g	0.0052	-	_	0.026	0.524	_

^a Estimated values reported as excess mixing ratios. Absolute values will be slightly higher to account for background concentrations.

Beported as OSHA TWA PEL, NIOSH TWA REL, and/or ACGIH TWA TLV

^c Estimated exposures (ppm) were divided by the recommended OSHA TWA exposures (ppm) to aid in the estimation of combined exposure limits. When OSHA TWA were not available, ACGIH TWA TLV were used.

^d Reported as OSHA STEL, NIOSH STEL, and/or ACGIH TLV STEL

e Exceeds recommended STEL peak exposure limit.

f Measured values from Table 3 are shown instead of estimated values.

⁹ Roberts et al. (2011) suggest mixing ratios above 0.001 ppm may have physiological effects, but no recommendations have been established.

Table A1. Spectral regions used to retrieve excess mixing ratios reported in this work.

Target species	Spectral region (cm ⁻¹)	Other species fitted	Single Beam (SB) or Transmission (TR)
CO, CO ₂	2050–2330	H ₂ O	SB
CH₄ ¯	2990-3105	H ₂ O	SB
C_2H_4 , NH_3	922-975	H ₂ O	TR
CH ₃ OH	1020-1055	NH_3 , H_2O	TR
CH ₃ COOH, HCOOH	1100-1230	H_2O , CH_4 , NH_3	TR
HCŇ	709–717	H ₂ O	TR
C ₂ H ₂ , Furan	725–755	H ₂ O, CO ₂ , 2-Methylfuran	TR
HCHO, Glyoxal	2740-2850	CH ₄ , H ₂ O	TR

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Table A2. Estimated and measured exposures for species measured both by the OP-FTIR and in Yokelson et al. (2013) reported as excess mixing ratios (see Sect. 3.3 for discussion).

		OP-FTIR TWA exposure (ppm)	OP-FTIR peak exposure (ppm)	LAFTIR peak exposure (ppm)
Formaldehyde (HCHO)	Calculated ^a	0.12	0.63	12.52
	Measured ^b	0.147 ^c	0.825	7.665
	Calculated/Measured	0.82	0.76	1.63
Methanol (CH ₃ OH)	Calculated ^a	0.081	0.409	8.165
	Measured	0.120	0.56	15.65
	Calculated/Measured	0.67	0.73	0.52
Ammonia (NH ₃)	Calculated ^a Measured Calculated/Measured	0.072 0.206 0.35	0.366 0.493 0.74	7.304 1.106 6.60

 $[^]a$ Calculated from pine-understory fire ER($\Delta X/\Delta CO$) from Yokelson et al. (2013) multiplied by the burn-average ΔCO measured by the OP-FTIR (Table 3).

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b Shown in Table 3.

^c Since we do not report HCHO measured from the start to end of the Fort Jackson fires, this value was estimated as ER(ΔHCHO/ΔCO) × OP-FTIR (burn-average) ΔCO.





Fig. 1. (a) Photograph of the OPAG-22 spectrometer system with receiver telescope in the field during the 2 Nov fire. **(b)** Photograph of the sender and receiver telescopes separated by an optical path of $\sim 30 \, \text{m}$ taken in clean air before ignition on 30 Oct. **(c)** Photograph of the Nov 2 fire from the airborne platform used by the airborne FTIR system. Pictures of fuels sampled by the LAFTIR can be found in Akagi et al. (2013).

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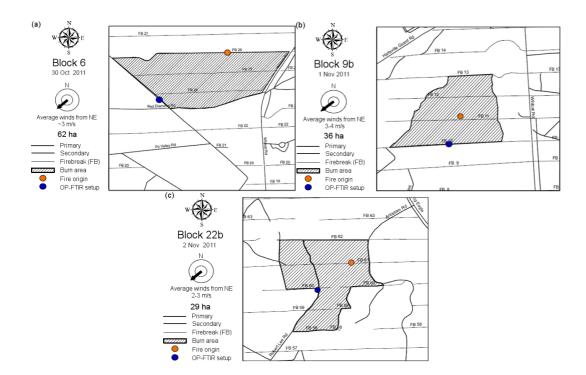


Fig. 2. Detailed burn maps of **(a)** Block 6, **(b)** Block 9b, and **(c)** Block 22b prescribed fires at Fort Jackson, SC. The location of the OP-FTIR is shown as a blue circle. The location where the fire was first lit is shown by the orange circle. Fires were typically lit along firebreaks in a continuous line with the "fire origin" representing where the fire-line was initiated.

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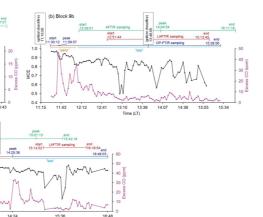


Fig. 3. MCE (black) and excess CO (ppm, pink) time series from OP-FTIR on the three Fort Jackson fires. Above the time series, AFTIR (green), LAFTIR (red), and OP-FTIR (blue) sampling time frames are shown to denote the start and end of measurement collection and when the "peak" intensity signal was observed from a given measurement platform. "Early" and "late" periods of OP-FTIR sampling are denoted in orange and light blue, respectively. Ignition times are shown in black to mark the lighting of headfires and backfires.

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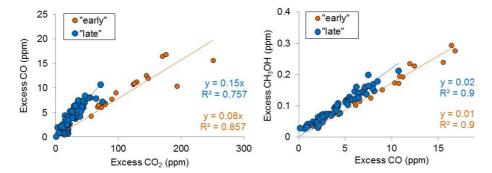


Fig. 4. ER plots of (a) $\Delta \text{CO}/\Delta \text{CO}_2$ and (b) $\Delta \text{CH}_3 \text{OH}/\Delta \text{CO}$ from the Block 6 (30 Oct) fire with two trend-lines shown: samples collected "early" in the fire are shown as orange circles and those collected "late" in the fire are shown as blue circles. Different trends observed "early" and "late" in the fire's progression imply changes in the sampled smoke over time and a decrease in MCE.

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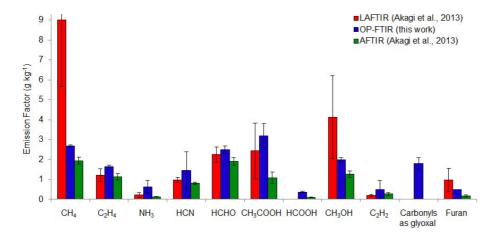


Fig. 5. Side-by-side comparison of study-average emission factors between the LAFTIR (red), OP-FTIR (blue), and AFTIR (green) FTIRs employed during the Fort Jackson campaign. The EFs represent the averages over all three of the Fort Jackson fires.

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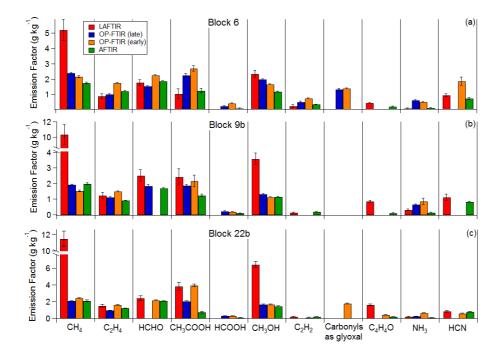


Fig. 6. Emission factors (g kg⁻¹) measured by the OP-FTIR, LAFTIR (red), and AFTIR (green) from the three Fort Jackson fires: **(a)** Block 6, **(b)** Block 9b, and **(c)** Block 22b. The OP-FTIR EF have been broken down into "late" (blue) and "early" (orange) as shown in Fig. 5.

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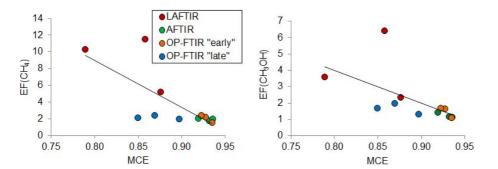


Fig. 7. Fire-averaged emission factors (g kg⁻¹) for CH₄ and CH₃OH as a function of MCE for the three Fort Jackson burns as measured by FTIR, with LAFTIR (red), OP-FTIR "early" (orange), OP-FTIR "late" (light blue), and AFTIR (green).

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Interactive Discussion



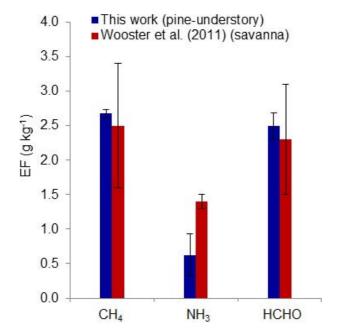


Fig. 8. Comparison of emission factors from this work (blue) and Wooster et al. (2011) (red). EF from this work have been slightly recalculated using a similar mass balance of carbon as dictated by measured species from Wooster et al. (2011), and are thus slightly different than EF shown in Table 1.

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