

## Supplement to: Model evaluation of NO<sub>3</sub> secondary organic aerosol (SOA) source and heterogeneous organic aerosol (OA) sink in the western United States

Juliane L. Fry and Kathryn Sackinger

Chemistry Department, Reed College, Portland, OR, USA

This supplemental information shows the experimental data used to determine average NO<sub>3</sub> uptake onto saturated (Fig. S1) and unsaturated (Fig. S2) organic in aerosol, the surface area of each aerosol size bin (Fig. S3), and a representative WRF/Chem namelist file (Appendix). All western US and Pacific Northwest monthly and daily OA source and sink spatial maps are available upon request (fry@reed.edu).

### Supplementary materials

Substrate	Experimental conditions	$\gamma$	Source
Octadecanethiol monolayer	None specified	$(8.8 \pm 2.5) \times 10^{-4}$	[1]
Diethyl sebacate	298 K, liquid	$(4.1 \pm 0.3) \times 10^{-3}$	[2]
	288 K, liquid	$(3.6 \pm 0.1) \times 10^{-3}$	[2]
	278 K, liquid	$(4.1 \pm 0.5) \times 10^{-3}$	[2]
	272 K, solid	$(3.6 \pm 0.5) \times 10^{-4}$	[2]
	263 K, solid	$(2.5 \pm 0.2) \times 10^{-4}$	[2]
Glycerol	293 K, liquid	$(1.4 \pm 0.3) \times 10^{-3}$	[2]
	286 K, liquid	$(9.2 \pm 0.4) \times 10^{-4}$	[2]
	268 K, liquid	$(8.3 \pm 0.5) \times 10^{-4}$	[2]
n-hexadecane	293 K, liquid	$(2.6 \pm 0.8) \times 10^{-3}$	[3]
	283-289 K, solid	$(3.8 \pm 1.0) \times 10^{-4}$	[3]
n-octanoic acid	290 K, liquid	$(2.1 \pm 0.3) \times 10^{-3}$	[3]
	287 K, solid	$(4.4 \pm 0.8) \times 10^{-4}$	[3]
Heptamethyl nonane	253 K, liquid	$(2.1 \pm 0.8) \times 10^{-3}$	[3]
	234 K, solid	$(2.6 \pm 0.9) \times 10^{-3}$	[3]
1-octanol	258 K, liquid	$(7.1 \pm 1.6) \times 10^{-3}$	[3]
	248 K, solid	$(5.2 \pm 2.0) \times 10^{-3}$	[3]

Table S2: Uptake coefficients for nitrate radical onto unsaturated organic substrates			
Substrate	Temperature and phase	$\gamma$	Source
Terminal C <sub>11</sub> alkene monolayer	None specified	0.034	[4]
Oleic acid	302 K, liquid	0.21 (+0.79/-0.11)	[2]
	295 K, liquid	0.18 (+0.82/-0.06)	[2]
	288 K, liquid	0.16 (+0.84/-0.06)	[2]
	285 K, liquid	0.17 (+0.83/-0.09)	[2]
	283 K, solid	0.053 ± 0.011	[2]
	277 K, solid	0.051 ± 0.014	[2]
	268 K, solid	0.076 ± 0.026	[2]
Conjugated linoleic acid	298 K, liquid	0.37 (+0.63/-0.24)	[2]
	288 K, liquid	0.33 (+0.67/-0.21)	[2]
	278 K, liquid	0.62 (+0.38/-0.47)	[2]
	263 K, solid	0.08 ± 0.03	[2]
	273 K, liquid	(7.9 ± 0.2) × 10 <sup>-3</sup>	[3]
	253-263 K, solid	(7.8 ± 1.4) × 10 <sup>-3</sup>	[3]
Nonconjugated linoleic acid	298 K, liquid	0.29 (+0.71/-0.15)	[2]
	288 K, liquid	0.33 (+0.67/-0.20)	[2]
	278, liquid	0.41 (+0.59/-0.27)	[2]
	263, liquid	0.13 (+0.67/-0.05)	[3]
	288, liquid	(1.5 ± 0.2) × 10 <sup>-2</sup>	[3]
	248-263 K, solid	(1.1 ± 0.3) × 10 <sup>-2</sup>	[3]
1-octadecene	293 K, liquid	(1.6 ± 0.3) × 10 <sup>-3</sup>	[3]
	283 K, solid	(1.4 ± 0.1) × 10 <sup>-3</sup>	[3]
1-hexadecene	277 K, liquid	(2.3 ± 0.9) × 10 <sup>-3</sup>	[3]
	254-274 K, solid	(1.8 ± 0.3) × 10 <sup>-3</sup>	[3]
7-tetradecene	246 K, liquid	(5.8 ± 2.0) × 10 <sup>-3</sup>	[3]
	238 K, solid	(5.2 ± 2.0) × 10 <sup>-3</sup>	[3]

1. Knopf, D.A., J. Mak, S. Gross, and A.K Bertram (2006): Does atmospheric

- processing of saturated hydrocarbon surfaces by NO<sub>3</sub> lead to volatilization?, *Geophys. Res. Lett.*, *33*, L17816, doi:10.1029/2006GL026884.
2. Gross, S., R. Iannone, S. Xiao, and A.K. Bertram (2009): Reactive uptake studies of NO<sub>3</sub> and N<sub>2</sub>O<sub>5</sub> on alkenoic acid, alkanoate, and polyalcohol substrates to probe nighttime aerosol chemistry, *Phys. Chem. Chem. Phys.*, *11*, 7792-7803, doi: 10.1039/b904741g.
  3. Moise, T., R.K. Talukdar, G.J. Frost, R.W. Fox, and Y. Rudich (2002): Reactive uptake of NO<sub>3</sub> by liquid and frozen organics, *J. Geophys. Res.*, *107(D2)*, 4014, doi:10.1029/2001JD000334.
  4. Gross, S., and A.K. Bertram (2009): Products and kinetics of the reactions of an alkane monolayer and a terminal alkene monolayer with NO<sub>3</sub> radicals, *J. Geophys. Res.*, *114*, D02307, doi:10.1029/2008JD010987.

Table S3. Drop diameters and surface-area-to-volume ratios of aerosol bins				
Bin number	Minimum diameter (μm)	Maximum diameter (μm)	Average radius (μm)	Average SA/vol (m)
a01	0.0390625	0.15625	0.048828	6.1440 x 10 <sup>7</sup>
a02	0.15625	0.625	0.195	1.54 x 10 <sup>7</sup>
a03	0.625	2.5	0.78	3.8 x 10 <sup>6</sup>
a04	2.5	10	3.1	9.7 x 10 <sup>5</sup>

## Appendix: Representative WRF/Chem namelist file.

```
&time_control
run_days      = 31,
run_hours     = 0,
run_minutes   = 0,
run_seconds   = 0,
start_year    = 2008,
start_month   = 05,
start_day     = 1,
start_hour    = 00,
start_minute  = 00,
start_second  = 00,
end_year      = 2008,
end_month     = 05,
end_day       = 31,
end_hour      = 00,
end_minute    = 00,
end_second    = 00,
interval_seconds = 21600,
input_from_file = .true.,
history_interval = 60,
frames_per_outfile = 24,
restart       = .false.,
restart_interval = 1440,
auxinput5_interval_m = 60,
io_form_history = 2,
io_form_restart = 2,
io_form_input = 2,
io_form_boundary = 2,
io_form_auxinput4 = 2,
io_form_auxinput5 = 2,
debug_level   = 0,
/

&domains
time_step      = 30,
time_step_fract_num = 0,
time_step_fract_den = 1,
max_dom        = 1,
s_we           = 1,
e_we           = 90,
s_sn           = 1,
e_sn           = 180,
s_vert         = 1,
e_vert         = 28,
p_top_requested = 10000,
num_metgrid_levels = 30,
num_metgrid_soil_levels = 4,
dx             = 12000,
dy             = 12000,
grid_id        = 1,
parent_id      = 0,
i_parent_start = 1,
j_parent_start = 1,
```

```
parent_grid_ratio = 1,  
parent_time_step_ratio = 1,  
feedback = 1,  
smooth_option = 0,  
/  

```

```
&physics  
mp_physics = 3,  
ra_lw_physics = 1,  
ra_sw_physics = 1,  
radt = 30,  
sf_sfclay_physics = 1,  
sf_surface_physics = 3,  
bl_pbl_physics = 1,  
bldt = 0,  
cu_physics = 1,  
cudt = 5,  
isfflx = 1,  
ifsnow = 0,  
icloud = 1,  
surface_input_source = 1,  
num_soil_layers = 4,  
sf_urban_physics = 0,  
maxiens = 1,  
maxens = 3,  
maxens2 = 3,  
maxens3 = 16,  
ensdim = 144,  
/  

```

```
&fdda  
/  

```

```
&dynamics  
w_damping = 0,  
diff_opt = 1,  
km_opt = 4,  
diff_6th_opt = 0, 0, 0,  
diff_6th_factor = 0.12, 0.12, 0.12,  
base_temp = 290.  
damp_opt = 0,  
zdamp = 5000., 5000., 5000.,  
dampcoef = 0.2, 0.2, 0.2  
khdif = 0, 0, 0,  
kvdif = 0, 0, 0,  
non_hydrostatic = .true., .true., .true.,  
moist_adv_opt = 2, 1, 1,  
scalar_adv_opt = 2, 1, 1,  
/  

```

```
&bdy_control  
spec_bdy_width = 5,  
spec_zone = 1,  
relax_zone = 4,  

```

```
specified      = .true.,
periodic_x     = .false.,
symmetric_xs   = .false.,
symmetric_xe   = .false.,
open_xs        = .false.,
open_xe        = .false.,
periodic_y     = .false.,
symmetric_ys   = .false.,
symmetric_ye   = .false.,
open_ys        = .false.,
open_ye        = .false.,
nested         = .false.,
/
```

```
&grib2
/
```

```
&chem
kemit          = 19,
chem_opt       = 7,
bioemdt        = 30,
photdt         = 30,
chemdt         = 2.,
frames_per_emissfile = 12,
io_style_emissions = 1,
emiss_inpt_opt = 102,
emiss_opt      = 3,
chem_in_opt    = 0,
phot_opt      = 3,
drydep_opt    = 1,
bio_emiss_opt = 1,
dust_opt      = 0,
dmsemis_opt   = 0,
seas_opt      = 0,
gas_bc_opt    = 1,
gas_ic_opt    = 1,
aer_bc_opt    = 1,
aer_ic_opt    = 1,
aer_op_opt    = 2,
gaschem_onoff = 1,
aerchem_onoff = 1,
wetscav_onoff = 0,
cldchem_onoff = 0,
vertmix_onoff = 1,
chem_conv_tr  = 1,
biomass_burn_opt = 0,
plumerisefire_frq = 30,
aer_ra_feedback = 1,
have_bcs_chem = .false.,
/
```

```
&namelist_quilt
nio_tasks_per_group = 0,
```

nio\_groups = 1,  
/