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Characterisation of sub-micron particle number concentrations and formation events in the western Bushveld Igeneous Complex, South Africa

A. Hirsikko^{1,2}, V. Vakkari², P. Tiitta^{3,4}, H. E. Manninen², S. Gagné^{2,*}, H. Laakso², M. Kulmala², A. Mirme⁵, S. Mirme⁵, D. Mabaso⁶, J. P. Beukes⁴, and L. Laakso^{1,2,4}

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Correspondence to: A. Hirsikko (anne.hirsikko@fmi.fi)

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¹Finnish Meteorological Institute, Research and Development, P.O. Box 503, 00101, Finland

²Department of Physics, Univ. of Helsinki, P.O. Box 64, 00014 Univ. of Helsinki, Finland

³University of Eastern Finland, PL 1627, 70211 Kuopio, Finland

⁴School of Physical and Chemical Sciences, North-West Univ., Potchefstroom, South Africa

⁵Institute of Physics, University of Tartu, 18 Ülikooli Str., 50090 Tartu, Estonia

⁶Rustenburg Local Municipality, Rustenburg, South Africa

now at: Dalhousie University and Environment, Canada

Africa.

South Africa holds significant mineral resources, with a substantial fraction of these reserves occurring in a large geological structure termed the Bushveld Igeneous Complex (BIC). The majority of the world's platinum group metals (PGMs) and chromium originate from the BIC. Considering the importance of PGMs in the manufacturing of automotive catalytic converters, as well as the relatively poor current state of air quality and the general lack of atmospheric research in the BIC, atmospheric related research in this geographical area is of local (South African) and of international interest. The western limb of the BIC is the most exploited, with at least eleven pyrometallurgical smelters occurring within a 55 km radius. Due to the lure of employment in the industrialised BIC, the area is populated by informal, semi-formal and formal residential developments. In order to investigate the characteristics and processes affecting sub-micron particle number concentrations and formation events, air ion and aerosol particle size distribution and concentration measurements were conducted for over two years at Marikana in the heart of the western BIC. Our results indicated that high amounts of Aitken and accumulation mode particles originated from domestic burning for heating and cooking in the morning and evening, while during daytime SO₂-based nucleation (from industrial emissions) was the most probable source for large number concentrations of nucleation and Aitken mode particles. Nucleation event day frequency was extremely high, i.e. 86% of the analysed days, which to the knowledge of the authors is the highest frequency ever reported. Secondary particle formation was influenced both by local pollution sources and regional ambient conditions. Therefore, our observation of the annual cycle and magnitude of the particle formation and growth rates during nucleation events were similar to the results from a semi-clean savannah site in South

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Introduction

Aerosol particles may affect climate directly by scattering and reflecting radiation, or indirectly via cloud-aerosol interactions (Lohmann and Feichter, 2005; Myhre, 2009; Quaas et al., 2009; Hand and Malm, 2007; IPCC, 2007). Aerosol particles are also known to have an influence on human health and, in polluted environments, particles pose a serious health threat for the population (e.g. Pope and Dockery, 2006; Russell and Brunekreef, 2009; Breitner et al., 2011).

Particle sources and formation mechanisms have been intensively studied (e.g. Kulmala et al., 2004, 2011), but the number of observations in the developing world is still very limited. Previously, such observations have been carried out in Mexico City (Mexico), Beijing (China), New Delhi (India), Gaborone (Botswana) (e.g. Jayaratne and Verma, 2001; Dunn et al., 2004; Mönkkönen et al., 2004, 2005; Yue et al., 2009; Wu et al., 2008, 2011; Kalafut-Pettibone et al., 2011) and few other sites. To our knowledge, there are no previous observations of ultrafine particle number concentrations or size distributions from residential area from southern Africa. However, in a relatively clean background savannah area in South Africa new particle formation has been studied by Laakso et al. (2008) and Vakkari et al. (2011). These studies revealed new particle formation taking place every sunny day throughout the year. As new particle formation was found to be a regional phenomenon (Laakso et al., 2008 and Vakkari et al., 2011), it is expected that new particle formation will also occurs in polluted areas.

South Africa holds major mineral assets. A significant proportion of these reserves are located in a geological structure known as the Bushveld Igeneous Complex (BIC), out of which ca. 80 and 45% of the world's platinum group metals (PGMs) and chromium, respectively, are produced (Cramer et al., 2004; Xiao and Laplante, 2004; Beukes et al., 2010). The most important use of PGMs is for the production of catalytic converters in automotive exhaust systems, which improve air quality. The chromium produced in this region is mostly in the form of ferrochrome (a crude alloy between iron and chromium), which in turn is mostly used for the production of stainless steel,

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a vital modern alloy. The BIC is divided into five different limbs, of which the western limb is the most intensely mined and industrialised. In a recent study it has been proposed that the western BIC should be declared as an air pollution hotspot by the South African government (Scott, 2010). However, notwithstanding the international importance of the western BIC in terms of the supply of commodities critical for the modern day society, nor the afore-mentioned indication that air quality in this region might be problematic, very little has been published in the peer reviewed scientific literature on atmospheric related matters for the western BIC.

Due to the lure of employment in the mining and metallurgical sectors in the western BIC, the area is populated by informal and semi-formal settlements. The observations discussed in this paper were carried out in a semi-informal and informal residential area at Marikana village in the heart of the western BIC, as part of an atmospheric research and air quality monitoring project initiated in 2006 (Laakso et al., 2008). Recently, Venter et al. (2012) analysed sources and temporal variation of gaseous compounds and particulate matter (PM₁₀) at Marikana. The aims of this work were to investigate the characteristic behaviour of sub-micron particles by investigating ion and particle concentrations, as well as new particle formation based on nucleation event classification, formation and growth rate analyses. As far as the authors could asses this is the first paper investigating sub-micron and ultrafine particle characteristics and new particle formation in this internationally important area. Additionally, the measurement site also afforded the opportunity to investigate the interaction of pollution from semi-formal and informal settlements, with that of mining and pyrometallurgical industries.

Site description

The measurements at Marikana (25°41′54.51" S and 27°28′50.05" E, 1170 m a.s.l.) were carried out from 8 February 2008 to 17 May 2010. Venter et al. (2012) gave a detailed description of the measurement site and its location, which is therefore not repeated here. In general, the Rustenburg-Brits mining and residential area is located

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in a shallow valley surrounded by mountains rising 500 m above the central area. The valley is approximately 60 km long in west-east direction and 20–30 km wide in north-south direction. Figure 1 shows the location of Marikana within a regional context. The extent of the BIC in central South Africa and the location of major point sources (i.e. pyrometallurgical smelters) within the western BIC are also indicated in Fig. 1. Additionally, the Google image shows the immediate surroundings of the measurement site. The distributions of semi-formal and informal settlements in the immediate vicinity of the measurement site are indicated on this image with white blocks.

On the Highveld of South Africa two distinct seasonal periods can be identified, i.e. dry and wet periods, from May to September and October to April, respectively. The dry period (winter and adjacent months on autumn and spring) is characterised by cold nights and low relative humidity with no rain for several months, while during the wet period (summer and adjacent months in autumn and spring) the air is warm and heavy rains, especially in the form of thunderstorms, occur.

3 Measurements

All the measurements were carried out from a specially fitted trailer. The trailer and measurement devices have been introduced by Laakso et al. (2008), Vakkari et al. (2011) and Venter at al. (2012). Here we only give a short summary of the devices relevant for this specific study. To ensure data quality, the different instruments were compared against reference instruments at Hyytiälä, Finland before transportation to South Africa.

3.1 Ion and particle size distribution measurements

An Air Ion Spectrometer (AIS, Mirme et al., 2007) measures naturally charged air ions in the mobility range 3.2– $0.0013\,\mathrm{cm}^2\,\mathrm{V}^{-1}\,\mathrm{s}^{-1}$, which corresponds to diameter range of ca. 0.8– $42\,\mathrm{nm}$. The AIS consists of two identical cylindrically symmetric differential mobility analysers (DMA), each with 21 electrometers. One DMA measures the positive

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ion spectrum and the other DMA measures the negative ion spectrum simultaneously. The air that flows through each DMA consists of a sample (30 lpm) and sheath (60 lpm) flow. These air flows are controlled by ball valves, which are placed after the DMAs, and measured by venturi tubes, which include a dense grid to produce laminar flow. The AIS data is averaged and recorded in five minute cycles. The accuracy of the AIS data of this specific measurement campaign will be discussed in Sect. 4.1.

The aerosol particle size distributions were measured with a Differential Mobility Particle Sizer (DMPS) in the diameter range 10-840 nm. The DMPS had a 28 cm long Vienna-type DMA (Winklmayr, 1991), a condensation particle counter (TSI 3010, Mertes, 1995) and it operated in a closed loop configuration (Hoppel, 1978; Jokinen and Mäkelä, 1997). The aerosol sample was dried with a Permapure MD-110-48 (Permapure LLC, USA) to less than 30% relative humidity, which was also continuously monitored. The total concentration obtained with this CPC was compared to a reference CPC (TSI 3772). The difference between the two CPCs was 7 %, which was considered in the data inversion.

3.2 Supporting data

Ancillary data included gas concentrations and basic meteorological parameters. For ozone (O₃) concentration monitoring, an Environment analyser (S.A. O341M) was deployed, correspondingly sulphur dioxide (SO₂) concentrations were monitored with Thermo analyser (Environmental Instruments Inc. 43S), nitrogen oxides (NO_x) concentrations with a Teledyne analyser (Instruments 200AU), black carbon (BC) concentrations with a multi-angle absorption photometer (MAAP, model 5012), and carbon monoxide (CO) with a Horiba analyser (APMA 360). Temperature and relative humidity were measured with Rotronic (MP 101A), while wind speed and direction were measured with Vector equipment (A101ML and A200P/L, respectively). Although not included in this work, additionally particulate matter concentrations (i.e. PM₁, PM_{2.5} and PM₁₀) were also monitored and trace metal content of the particulate matter is also available.

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Data measured both with the AIS and the DMPS included some gaps or bad data, which were due to malfunctioning of the instruments, as well as service and/or calibrations being conducted. Before data analysis commenced, raw data was cleaned according to methods indicated by Laakso et al. (2008) and Venter et al. (2012). Aerosol data was cleaned by visually selecting and deleting periods and size intervals when data was noisy or otherwise suspicious. Gas data was corrected based on in-situ calibration procedures. All the data was averaged over 15 min. However, as higher temporal resolution particle and ion data are needed for the formation and growth rate analysis, the original time resolution of DMPS (9 min) and AIS (5 min) data were kept.

Effect of flow rate and contamination on size distributions obtained with the AIS

During the measurements at Marikana we noticed that the air pathways of the AIS got partially blocked due to the accumulation of soot and coarse particles, mainly in the areas with small apertures and on the venturi tube nets (for a technical drawing of AIS, please see Mirme et al., 2007). Fortunately no particulate build-up occurred in the blower. As a result of accumulation of particulate matter, the flow rates of the two DMAs may become different over time, since both of the DMAs have their own postventuri air flow paths. The sheath air flow rates were also affected by blockages in the nets before the electrical filters, which produce ion free air.

Decreasing flow rate broadens and shifts mobility distributions. The best way to correct the data would have been to calculate new transfer functions, which correspond to new flow rates. However, we were uncertain of the relative changes of the different flows (i.e. two sheath flows, two sample flows and a combined exhaust flow), since only the combined exhaust flow rate was recorded by default. All the individual flow rates were measured only after the measurement campaign at Marikana was completed. These post campaign measurements indicated that the decrease of all flows were

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proportional. Due to the limited amount of knowledge on the effect of reduced flow rates on the mobility distribution, we conducted theoretical and experimental studies to evaluate the effect.

The theoretical investigation included calculation of ion currents (i.e. raw data) as a function of the electrometer number using the AIS inversion algorithm by reducing the flow rate by 10, 20 and 30 % of the expected value. For simplicity, we assumed that the sheath and sample flow rates in both analysers dropped proportionally, which may not always be the case and requires further experimental investigation. The results showed that the small ion (less than ca. 1.9 nm in diameter) distribution was immediately shifted towards smaller sizes when flow rate began to decrease. However, the distributions of larger ions (ca. >3 nm) began to shift only when flow rates had decreased by ca. 30 % or more. Signals of the electrometers in all sizes decreased immediately with any decrease in flow rate (Fig. 2). The ratios in Fig. 2 were obtained by including signal readings of three adjacent electrometers, which are taken into account in the data inversion.

Experimental investigation carried out in Finland in May–June 2010 included both laboratory calibration and field study. The laboratory calibrations are explained and analysed in detail by Gagné et al. (2011). The results showed that the AIS (serial nr. 2 utilised here) classified the mobility accurately. However, the AIS overestimated concentrations compared to a reference instrument. The concentration difference was almost constant and independent of size. We therefore calculated a correction factor for concentration based on the calibration.

The second experiment was to reduce flow rates artificially by burning a paraffin torch, which generated soot and coarse particles, close to the inlet. In this way, the out flow rate was decreased by 10 and 20 %. The observations (Fig. 3) confirmed the results from the theoretical evaluation (Fig. 2). Upon reduced flow, small ion mode immediately began shifting towards smaller sizes. Larger particle distribution shift could not be identified with the applied flow rates and method. However, concentrations of ions in all size ranges dropped with reduced flow. The functioning of the AIS electrometers

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eventually became unstable as an increased amount of soot gathered on their surfaces, therefore, only a qualitative comparison was possible.

Based on the above-mentioned results, it was decided to use the AIS data as follows: (1) sub-3 nm ion data was used only if the flow rate was at the desired level (60 lpm ± 5%), (2) data of ions larger than 3 nm in diameter was used if the flow rate had decreased by at most 20 %, and (3) all the data affected by high electrometer noise (due to the accumulation of dirt on the electrometers) were excluded. Additionally, we multiplied the entire positive and negative ion spectrum having diameters larger than 3 nm with concentration correction factors of 0.76 and 0.79, respectively, which were obtained based on the calibration with the reference instrument (Gagné et al., 2011). An alteration to the inversion procedure was implemented on 7 May 2008 and the calibration was valid for the applied inversion procedure. Therefore only AIS data from 7 May 2008 to 17 May 2010 (end of the sampling campaign at Marikana) is analysed and considered further in this paper. Due to above-presented analysis we utilised ion data only when classifying the particle formation events (Sect. 4.3.1) and analysing the growth rates (Sect. 4.3.2).

Altitude and temperature effects on the AIS data

Our measurements were performed at a relatively high altitude (1170 m a.s.l.), where annual average temperature and pressure were 19°C and 886 hPa, respectively. The mobility of an ion depends on both ambient pressure and temperature. Thus at higher altitudes the ions may become more mobile compared to sea level. In this work we decided to use Millikan diameter conversion (Millikan, 1923) calculated at ca. 23°C and 1000 hPa for our data to be comparable to the majority of ion size distribution studies already published (e.g. Manninen et al., 2010; Hirsikko et al., 2011). This led to a ca. 6% underestimation of the diameter in the mobility-diameter conversion compared to DMPS diameters, which were calculated by using temperature and pressure dependent Millikan-diameter.

4.3.1 Particle formation

Particle formation events were classified visually according to guidelines by Yli-Juuti et al. (2009) and Vakkari et al. (2011) for the AIS data and by Dal Maso et al. (2005) for the DMPS data. We divided days into particle formation (i.e. nucleation), undefined and non-event days. Our aim was to analyse formation and growth rates for all nucleation events if possible, therefore no further division of particle formation events were made. Based on our earlier experience we know that this type of visual classification is subject to small discrepancies due to the judgement of the analyst(s).

Due to accuracy issues of concentration measurements and lack of sub-3 nm ion data we were unable to calculate formation rates for 2-nm ions (J_2) as is typically done when investigating particle formation processes (e.g. Kulmala et al., 2007; Manninen et al., 2010). The DMPS system was set to measure particles from 10 nm up. Unfortunately, the lowest size channel of the DMPS had to be rejected as the DMPS high voltage power source could not keep the voltage stable below 8 V. Therefore we began our analysis of formation and growth rates from 12 nm (i.e. J_{12} and GR_{12-30}).

Formation rates for 12 nm particles (J_{12}) were calculated following Dal Maso et al. (2005, and references therein):

$$J_{12} = \frac{dN_{12-30}}{dt} + \text{CoagS} \cdot N_{12-30} + F_{\text{growth}}. \tag{1}$$

Where N_{12-30} is the concentration of nucleation mode particles obtained by DMPS measurement, CoagS the particle removal rate due to coagulation and $F_{\rm growth}$ is the flux of particle growth out of the nucleation mode. The latter was assumed to be equal to zero due to the assumption that particles do not grow larger than 30 nm during the time the nucleation event is investigated (Dal Maso et al., 2005).

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We estimated the growth rates of particle population based on AIS and DMPS data. Ion growth rates from the AIS were analysed by using a maximum concentration method described by Hirsikko et al. (2005). The method follows the timing of maximum concentration of every size bin, and the growth rates are obtained as a slope of a linear fit to time-diameter pairs. However, this method works best when the growing mode is narrow enough (i.e. for sub-20 nm particle mode). Therefore, we calculated growth rates based on the DMPS data by using a mode fitting method introduced by Hussein et al. (2005) and Dal Maso et al. (2005). The mode fitting method cannot typically be used to estimate initial growth, thus the fitting often begins close to a diameter of 5 nm, which makes this method unsuitable for growth rate analysis with the AIS data.

Major error sources for growth rate analysis are the stretching of size distributions due to decreasing flow rates in the AIS (as was discussed in the Sect. 4.1) and an inhomogeneous/changing air mass during a growth event. The first source of error was taken into account by using only data measured when the flow rates were high enough. The later effect was dealt with by visually checking how well the obtained slope followed the growing mode.

The analysis by Yli-Juuti et al. (2011) showed that the maximum concentration method typically results in somewhat higher growth rates than the mode fitting method. This was assumed to be due to high and changing coagulation sinks during growth, especially if the growth is slow (Leppä et al., 2011). However, growth rates at Marikana were high (Sect. 5.3), which reduced somewhat the possible overestimation. The obtained growth rates also depend on the instrument and the charge polarity of the particles (Yli-Juuti et al., 2011). The first arises from inaccurate determinations of flow rates, and the latter is due to the combination of the fact that activation of one charge polarity occurs before the other and continuous charging in the atmosphere. We will present growth and formation rates statistically, therefore qualitative error estimation is assumed to be sufficient for our purposes.

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There are strong indications that sulphuric acid is involved in the atmospheric particle formation and growth (e.g. Weber et al., 1996; Fiedler et al., 2005; Sihto et al., 2006; Riipinen et al., 2007; Kuang et al., 2008; Sipilä, 2010; Paasonen et al., 2010; Brus et al., 2011). Vakkari et al. (2011) also observed that the nucleation event day frequency correlated with the temporal cycle of the sulphuric acid proxy concentrations in a semiclean South African savannah. Several proxies for sulphuric acid have been derived (Petäjä et al., 2009; Mikkonen et al., 2011). However, none of them have been tested against sulphuric acid data from environments comparable to Marikana. We decided to deploy parameterisation by Petäjä et al. (2009), as was previously done by Vakkari et al. (2011):

$$H_2SO_4 - proxy = 1.4 \times 10^{-7} \cdot Glob^{-0.70} \cdot \frac{[SO_2] \cdot Glob}{CS}$$
 (2)

Where Glob indicates global radiation in $W\,m^{-2}$, $[SO_2]$ is the concentration in molecules cm⁻³ and CS is a condensation sink (in s⁻¹) due to background aerosol particles. The condensation sink was calculated based on particle size distribution data according to Dal Maso et al. (2005, and references therein). While utilising the proxy developed for the clean Finnish background station, it should be kept in mind that there is no confirmation it works for the conditions at Marikana. Thus the absolute values are uncertain and most probably only the difference between the seasons can be considered reliable.

5 Results

5.1 Variation of particle number concentration

The concentrations of Aitken (20–100 nm in diameter) and accumulation (100–840 nm in diameter) mode particles had a clear annual cycle, i.e. concentrations were higher during the dry period compared to the wet period (Fig. 4). Median diurnal cycles (Fig. 5)

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indicated that accumulation mode had peaks in the morning (06:30-08:00) and evening (19:00-21:00), while Aitken mode had three peak periods, i.e. the first in the morning (06:00-07:30), the second in the afternoon (11:00-14:30) and the third in the evening (18:30-20:00). However, the timing of the concentration peaks was related to the du-5 ration of daylight and followed the length of the day.

Nucleation mode (12–20 nm in diameter) particle concentrations also had three concentration peaks, dominated by high concentrations at or before midday (Fig. 5), but the results showed that nucleation mode particle concentrations had the opposite annual cycle compared to larger particles (Fig. 4). The nucleation mode midday concentrations were higher during the wet period compared to observations during the dry period.

We calculated small ion concentrations during the periods (ca. 10% of the measurement time) when the AIS flow rate was at the desired level. The concentrations were observed to follow the evolution of the mixing layer. The median concentrations were 320 and 450 cm⁻³ for positive and negative small ions respectively, which are typical for polluted environment (Hirsikko et al., 2011).

Condensation sink, which is mainly caused by large particles, may be used as a tracer for air quality. The CS values were high and typical of polluted urban or industrial environments (e.g. Mönkkönen et al., 2005; Kulmala et al., 2005; Yue et al., 2009). The CS had a clear annual cycle with maximum values during the dry period (Fig. 6), which was in accordance to observations by Vakkari et al. (2011) in the South African savannah.

Nucleation, Aitken and accumulation mode particle concentrations were associated with BC, CO and NO_x concentrations during both day and night throughout the year, and with SO₂ during the daytime. The concentrations of nucleation mode particles were independent of the condensation sink, while Aitken and accumulation mode particles concentrations correlated with CS, as expected. Venter et al. (2012) indicated that NO_v, CO and BC mainly originated from local residential combustion for heating and cooking at Marikana, while SO₂ mainly originated from high stack industry emissions.

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The above-mentioned correlations of the various particle modes with the NO_x, CO, BC and SO₂ respectively, as well as the temporal variation of the particle concentrations indicate that the morning and evening concentration peaks (Fig. 5) were due to the local residential combustion rather than industrial emissions. During the dry periods, which correspond to winter and the colder autumn and spring months, domestic burning is a major particle source. Residential combustion mainly takes place during the early morning and evening. Evening concentration peaks (Fig. 5) are usually more pronounced due to the need for space heating. The morning residential combustion period is usually curtailed due to warming temperatures and residents leaving for work. During the wet periods, which correspond to summer and the wetter autumn and spring months, domestic heating probably plays a smaller role as a particle source hence the observed Aitken and accumulation mode particles concentrations were higher in the dry periods (Fig. 4).

In contrast, the midday peak in particle concentrations is expected to be due to particle formation as a result of industrial emissions, as will be explained in Sect. 5.2. The South African Highveld is well known for the formation of several well developed inversion layers, at different heights, that typically form during the night and persist until later morning (Garstang et al., 1996; Tyson et al., 1996; Wenig et al., 2003). High stack industry emissions can therefore accumulate between two inversion layers during nighttime, which is then released after the break-up of the inversion layers in the morning, hence resulting in a release of pollution at ground level. It is known that the PGM industry in this area has relatively high SO₂ emissions, since this industry utilises sulphite ore (Xiao et al., 2004) that generate substantial SO₂ emissions. Thus, based on our observations we have identified two particle sources: SO₂-based nucleation before or at midday and domestic burning during the morning and evening.

5.2 Frequency of new particle formation days

In this section, unlike everywhere else in this paper, particle nucleation event classification was done on the entire, un-cleaned ion (AIS) and particle (DMPS) datasets. This

was done, since the procedure is visual and distinction between particle nucleation, undefined and non-events can be done even if the distributions may be disturbed or broadened (Sect. 4.1). This decision allowed us to use as much data as possible and to obtain the most representative result.

The nucleation events were observed on 86% (DMPS), 84% (negative ions) and 78% (positive ions) of the analysed days at Marikana (Table 1, Fig. 7). With the ion spectrometer we were able to identify local charged particle events, and may therefore conclude that both local and regional nucleation events were detected. Despite the strong local pollution sources, regional new particle formation was still significant at Marikana. This is understandable, as the typical transport time above the BIC area is small (1–2 h maximum), and oxidation of SO₂ to H₂SO₄ also takes some time.

At Marikana, the frequency of nucleation event days was higher than is reported from any other location (e.g. Kulmala et al., 2004; Jaatinen et al., 2009; Hirsikko et al., 2011). As an example, in the South African savannah, the monthly nucleation event day frequency was 55-90% (Vakkari et al., 2011), and in the Australian Eucalyptus forest the corresponding frequency was ca. 52 % (Suni et al., 2008). In contrast, in the clean rural boreal environments annual nucleation event day frequency is typically less than 30% (e.g. Dal Maso et al., 2005, 2007; Hirsikko et al., 2007; Asmi et al., 2011), and only one event per day is observed. Over the polar areas the nucleation event day frequency is slightly lower (e.g. Virkkula et al., 2007; Asmi et al., 2010).

Typically, particle formation events at Marikana began earlier in the morning during the wet period than during the dry period, due to the earlier sunrise (Fig. 5). Although the difference of nucleation event day frequency between the various months was small, we may conclude that the DMPS nucleation event day frequency was slightly higher in summer and early autumn (wetter period) than in winter and spring (dryer period). In summer, the air was cleaner, but the estimated sulphuric acid concentration showed lower monthly median values (Fig. 6). In contrast, based on the ion spectrometer the nucleation event day frequency was slightly higher during late autumn and winter having similar annual cycle as H₂SO₄-proxy (Fig. 6). These observations

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indicate that the regional nucleation events were also affected by prevailing conditions outside the western BIC.

The observations indicated that all the particle formation events could not be detected with the ion spectrometers, and the concentrations of sub-3 nm ions were typi-5 cally low during particle formation events. These results indicate that contributions of electrically neutral particles are dominating on such days. This is in accordance with the theory and earlier observations that ion-mediated particle formation requires environment having low nucleation rates, low temperature and aerosol content but high small ion concentration, and preferably being rich with sulphuric acid (e.g. Laakso et al., 2002; Curtius et al., 2006; Iida, 2006; Yu, 2010). None of these requirements were met at Marikana.

Despite the slightly lower frequency of positive ion events we cannot conclude with any certainty a clear polarity difference based on the visual classification (Table 1, Fig. 7). In contrast, in rural boreal environment negative ion events have been observed to be more frequent and stronger compared to positive ion events (Hirsikko et al., 2007).

A very notable observation was that multiple consecutive nucleation events (i.e. more than one event per day), starting from the small ion sizes (cs. 1.5 nm) followed by subsequent growth, were observed on ca. 100 days. All of these multiple events were observed during daylight. In addition to daytime nucleation, nocturnal nucleation events (Junninen et al., 2008; Lehtipalo et al., 2011), which were suppressed before diameter of ca. 3 nm. were also observed at Marikana.

Svenningsson et al. (2008) have also observed several nucleation events, characterised by very fast growth rates, to occur during a day over wetland in a clean Sub-Arctic Abisko, where the nucleation event day frequency was 40 % between spring and autumn. Svennigsson et al. (2008) observed such consecutive nucleation events related to homogeneous air masses to occur also during night-time in summer. However, they were not able to explain their observations of the night-time or multiple events. Further investigations of the mechanisms behind these observations are required in future.

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Particle formation and growth rates

Particle formation and growth rates (J_{12} and GR_{12-30}) were high from spring to autumn (Fig. 8, compare also with Fig. 5). Thus, ambient conditions were less favourable for 12-30 nm particle formation and subsequent growth in winter (Table 2). High formation rates were observed to occur simultaneously with fast growth rates (Fig. 9). Both formation and growth rates were observed to correlate with concentrations of Aitken mode particles during nucleation event, but not with the accumulation mode particle concentrations. When comparing our results of J_{12} and GR_{12-30} (Fig. 8) to observations by Vakkari et al. (2011) in the semi-clean South African savannah, we notice that both the magnitude and temporal variation of these parameters were similar.

Eventhough sub-3 nm ion data was acceptable during 10 % of the time, we were able to estimate growth rates of sub-3 nm ions only on 13 days. This may be an indication that ion-mediated particle formation pathways are not an important contributor to new particle formation at Marikana. Growth rate analysis for 7-20 nm ions was possible only during half of the months due to data quality issues. Charged particle growth rates in all size ranges were slightly higher during months from autumn to late spring (Figs. 10 and 11, Table 2). Growth rates of charged particles were also compared to Aitken and accumulation mode particle concentrations during the nucleation events. The analysis showed that the faster growth rates were observed when Aitken mode particle concentrations were higher.

The magnitude of presented formation and growth rates is in accordance with earlier observations in urban and polluted environments (Kulmala et al., 2004 and references therein). The observed formation (up to 100 cm⁻³ s⁻¹) and growth rates are typically faster in polluted environments compared to rural or polar sites (e.g. Kulmala et al., 2004; Dal Maso et al., 2005, 2007; lida et al., 2008; Jaatinen et al., 2009; Kalafut-Pettibone et al., 2011), due to more pronounced concentrations of H₂SO₄ vapour and stronger competition between growth and coagulation loss. Earlier observations suggest that high growth rate and CS exist at the same time since slow growth cannot be

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observed due to quick scavenging by pre-existing aerosol (e.g. Kulmala et al., 2005). The CS in Marikana was 3-5 times higher than in rural Botsalano, which may balance the potentially increased particle growth rates.

Concluding remarks

The aim of this work was to characterise particle sources and formation events at a residential area surrounded by mining and pyrometallurgical smelting industries in South Africa. The aerosol particle number concentrations were guite high (on average 10⁴ cm⁻³) and typical for urban or industrial environments, although, higher particle number concentrations have also been reported in literature (e.g. Ruuskanen et al., 2001; Woo et al., 2001; Wehner and Wiedensohler, 2003; Hussein et al., 2004; Stanier et al., 2004; Mönkkönen et al., 2005; Wu et al., 2008; Cheung et al., 2011). Our results, together with the analysis by Venter et al. (2012) indicate that the major source for nucleation mode particles was new particle formation (most likely SO₂-based) around midday. Aitken and accumulation particles originated from domestic burning for heating and cooking, but in the afternoon also from nucleation from industrial pollution.

Nucleation event day frequency was very high, i.e. 86 % of the analysed days, which is the highest thus far recorded in any environment (e.g. Kulmala et al., 2004; Hirsikko et al., 2011). Both the local sources and regional conditions affected the particle formation and growth. The formation and growth rates were high and typical for urban environments (Kulmala et al., 2004; Dal Maso et al., 2005, 2007; lida et al., 2008; Jaatinen et al., 2009). However, the temporal cycle of the new particle formation characteristics was similar as have been observed by Vakkari et al. (2011) over a semi-clean savannah in South Africa.

From the results presented it is clear that the industrial and residential activities in the western BIC are sources of particulates. Our results combined with the results of Venter et al. (2012) provide valuable information for the assessment of local air quality and the possible health threat to the community in the western BIC, as well as supply

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information for decision making processes in the future for limiting air pollution. The results presented here were the most thorough analysis of particle formation and growth in an informal and semi-formal residential area, influenced by mining and pyrometallurgical industries.

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Table 1. Statistics of nucleation event day classification based on the DMPS and the AIS data.

	Analysed	Events	Non-events	Undefined
DMPS				
Number of days Fraction of analysed days (%)	649 100	559 86.1	2 0.3	88 13.6
AIS(-)				
Number of days Fraction of analysed days (%)	606 100	509 84.0	9 1.5	88 14.5
AIS(+)				
Number of days Fraction of analysed days (%)	606 100	476 78.5	18 3.0	112 18.5

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Table 2. Mean, median, standard deviation, and 10th/90th percentiles of growth rate (GR) and formation rate (J_{12}) separately for dry and wet periods based on the DMPS and the AIS. The growth rates based on the AIS data were calculated separately for the positive and the negative polarity.

	Mean \pm std	Median	10th/90th percentile
Dry period			
DMPS			
$J_{12} (\text{cm}^{-3} \text{s}^{-1})$	4.5 ± 6.1	2.5	0.6/10.7
$GR_{12-30\text{nm}}(\text{nm}\text{h}^{-1})$	8.0 ± 4.1	7.1	4.2/13.1
AIS(+)			
GR _{3-7 nm} (nm h ⁻¹)	12.5 ± 8.2	10.5	4.7/24.5
GR _{7-12 nm} (nm h ⁻¹)	19.0 ± 11.3	16.0	7.0/38.3
AIS(-)			
GR _{3-7 nm} (nm h ⁻¹)	11.9 ± 7.3	10.1	4.7/22.6
$GR_{7-12\text{nm}}(\text{nm}\text{h}^{-1})$	19.7 ± 10.5	17.5	8.2/34.5
Wet period			
DMPS			
$J_{12} (\text{cm}^{-3} \text{s}^{-1})$	7.6 ± 7.3	5.7	1.5/16.4
$GR_{12-30\text{nm}}(\text{nm}\text{h}^{-1})$	11.1 ± 5.2	10.2	5.8/16.2
AIS(+)			
GR _{3-7 nm} (nm h ⁻¹)	13.1 ± 7.7	11.1	5.4/24.2
GR _{7-12 nm} (nm h ⁻¹)	19.1 ± 11.0	15.4	7.9/35.4
AIS(-)			
GR _{3-7 nm} (nm h ⁻¹)	13.3 ± 7.4	11.3	5.5/23.7
$GR_{7-12\text{nm}}(\text{nm}\text{h}^{-1})$	17.3 ± 8.8	15.7	7.6/29.9

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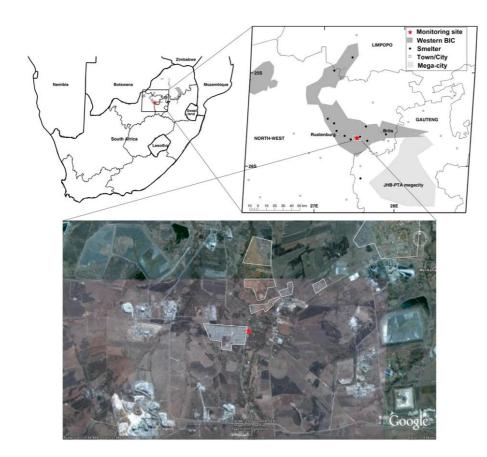


Fig. 1. A geographical map of southern Africa indicating the extent of the Bushveld Igeneous Complex (BIC, in grey). The enlarged section of the geographical map indicates the western BIC, the proximity of the Johannesburg/Pretoria mega-city and the location of major point sources (pyrometallurgical smelters) in the western BIC. The Google map indicates the immediate surroundings of the measurement site (25°41′54.51" S and 27°28′50.05" E). The white outlined blocks indicate the occurrence of informal and semi-formal residential settlements in-between the mining and metallurgical industries.

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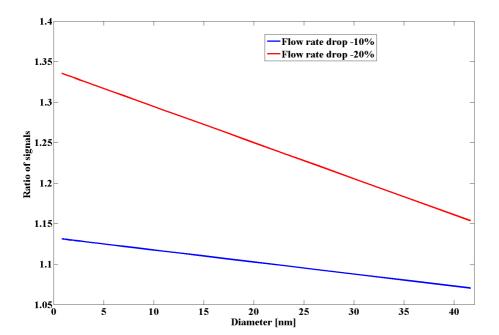


Fig. 2. Ratio of raw signals during normal flow rate to reduced flow rate. The peak signal and both adjacent signals from the electrometers were included in the analysis.

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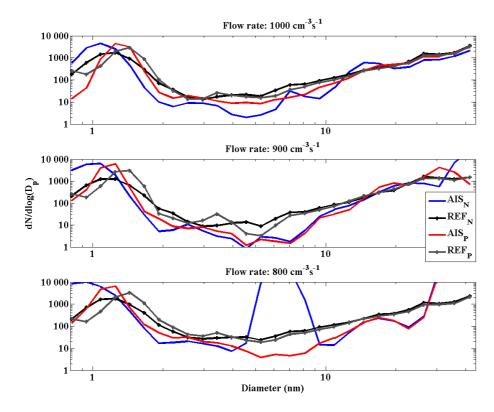


Fig. 3. Average size distributions measured with the AIS and a reference ion spectrometer at different flow rates during flow rate experiments. The negative polarity is symbolised with the letter N and the positive polarity with P.



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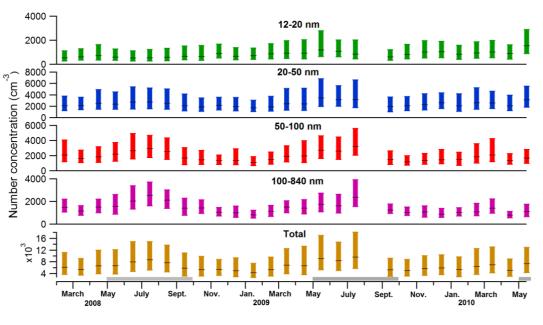


Fig. 4. Monthly median, 25 and 75 % percentile concentrations of 12-20, 20-50, 50-100, 100-840, and 12-840 nm (total) particles over two years. The dry periods are indicated with grey bars at the bottom of the figure.

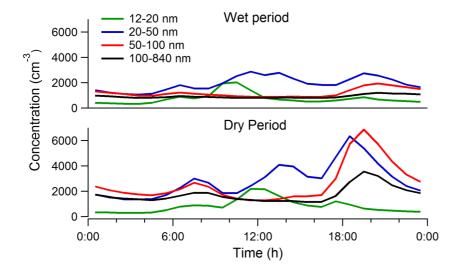


Fig. 5. Diurnal cycle of 12–20, 20–50, 50–100 and 100–840 nm particle concentrations, during wet (October–April, upper panel) and dry (May–September, lower panel) periods.

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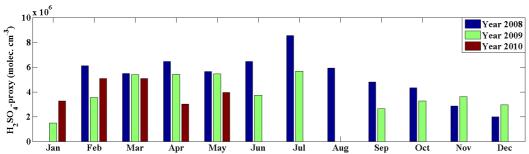




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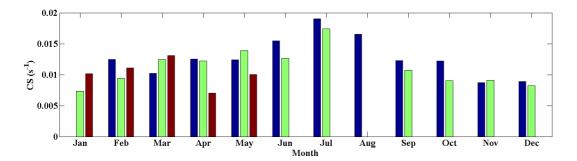


Fig. 6. Annual cycle of monthly medians of H₂SO₄-proxy and condensation sink. The figure includes observations made during February-December 2008, January-July 2009, September-December 2009, and January-May 2010.

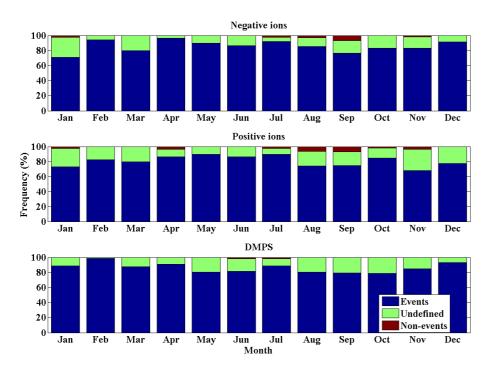


Fig. 7. Monthly distribution of particle formation, undefined and non-event days based on the DMPS (lower panel), positive (middle panel) and negative (upper panel) ion data (AIS).

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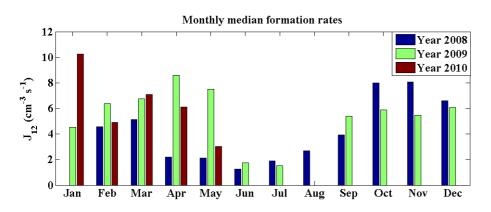
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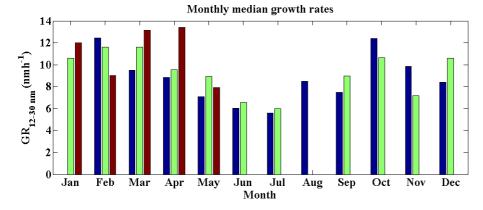


Fig. 8. Monthly median formation rates of 12–30 nm particles (J_{12}) and growth rates (GR_{12-30 nm}) based on the DMPS data. The figure includes observations made during February–December 2008, January–July 2009, September–December 2009, and January–May 2010.

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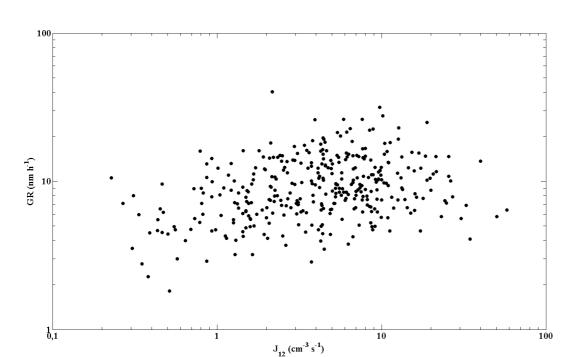


Fig. 9. Formation rate of 12 nm particles as a function of growth rate (12–30 nm) based on the DMPS data.

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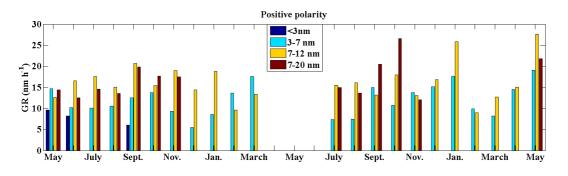
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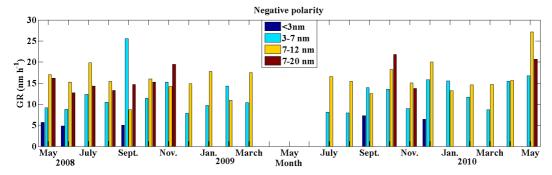


Fig. 10. Monthly median growth rates of positive and negative ions (based on the AIS data) for different size fractions indicated with different colours.

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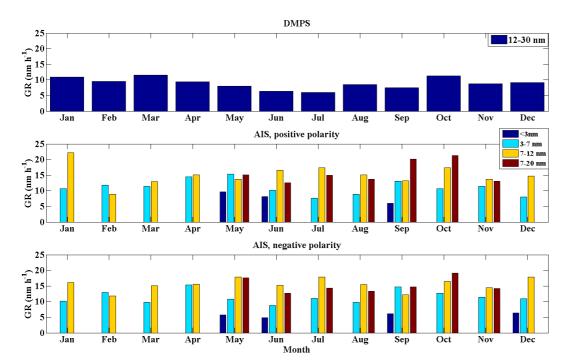


Fig. 11. Annual monthly median growth rates (GR) based on DMPS (upper panel), positive (middle panel) and negative (lower panel) ion data (AIS).

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