

**Observation of  
aerosol size  
distribution and new  
particle formation**

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**Observation of aerosol size distribution  
and new particle formation at a mountain  
site in subtropical Hong Kong**

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## Abstract

In order to investigate the atmospheric particle formation and growth processes, and to quantify the particle number (PN) concentration and size distributions in Hong Kong, a three-month intensive field measurement was conducted from September to November in 2010 near the mountain summit of Tai Mo Shan, a suburban site approximately the geographical centre of the New Territories in Hong Kong. The mean total number concentration in the size range of 5.5–350 nm was  $7.86 \pm 0.66 \times 10^3 \text{ cm}^{-3}$  (mean  $\pm$  95% confidence interval), with a maximum value in November. New particle formation (NPF) events were observed on 12 out of 35 days in October/November 2010 with the formation rate from 0.29 to  $4.53 \text{ cm}^{-3} \text{ s}^{-1}$ , and the average growth rates from 1.53 to  $9.44 \text{ nm h}^{-1}$ . The events usually began at 10:00 ~ 11:00 local time characterized by the occurrence of a nucleation mode with a peak diameter of 6 ~ 10 nm. The observed linear or non-linear correlations between nucleation mode PN concentration (5.5–10 nm) and ozone, volatile organic compounds (VOCs) and/or ( $\text{UV} \times \text{SO}_2$ ) suggested critical roles of sulfuric acid and biogenic VOCs (e.g. isoprene,  $\alpha$ -pinene and  $\beta$ -pinene) in the NPF events.

## 1 Introduction

Over the past decades, atmospheric submicrometer aerosols ( $< 1 \mu\text{m}$  in diameter) have been found to impact global climate (e.g., Twomey et al., 1984), air quality (USEPA, 1996), visibility (e.g., Mohan and Payra, 2009) and human health (e.g., Oberdorster et al., 2005). Typically, the sizes of submicrometer particles are categorized into three modes: a nucleation mode, representing quite newly formed particles usually have diameters smaller than 10 nm; Aitken and accumulation modes, representing aged particles generally ranged in size from 10 to 100 nm and from 100 to 1000 nm, respectively (e.g., McMurry et al., 2000; Seinfeld and Pandis, 2006). Especially, as the newly formed particles, parts of them can grow into cloud condensation nuclei (CCN) sizes,

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significantly increasing the indirect radioactive effect of cloud (e.g., McMurry et al., 2011). Moreover, due to their nanometer-scale size, the newly formed particles can enter deep into the ciliated and alveolar sections of the lung and even bloodstream (Oberdorster et al., 2005), contributing to negative health effects. Consequently, it is critical to understand formation and growth processes of these new particles in the atmosphere.

To date, numerous measurements have been conducted in different locations to study the size distribution of submicrometer particles and new particle formation (NPF) processes, including free troposphere (e.g., Weber et al., 2001), boreal forests (e.g., Vehkamäki et al., 2004; Dal Maso et al., 2007; Asmi et al., 2011), coastal rural areas (e.g., Lee et al., 2008), high mountain-top (e.g., Gannet et al., 2011) and Antarctic/Arctic areas (e.g., Park et al., 2004). In addition, an increased number of studies were conducted in continental and coastal urban areas, i.e. Atlanta (Woo et al., 2001), St. Louis (Shi et al., 2002), East St. Louis (Qian et al., 2007), Pittsburgh (Stanier et al., 2004a,b), Birmingham, UK (Alam et al., 2003), Mexico City (Dunn et al., 2004), and Brisbane (Guo et al., 2008; Cheung et al., 2011). In China, the field measurements for particle size distribution were also carried out in several cities and regions in recent years. Short-term intensive studies on NPF were performed in Beijing (Yue et al., 2009), Pearl River Delta (PRD) (Liu et al., 2008), Yangtze River Delta (YRD) (Gao et al., 2009) and Hong Kong (Yao et al., 2010). Moreover, a couple of long-term continuous measurements were conducted in Beijing (e.g., Wu et al., 2008; Shen et al., 2011) and a remote mountain site, Mt. Waliguan (Kivekäs et al., 2009). Generally, nucleation events are much more frequently observed in rural/remote areas than in urban locations.

The production of atmospheric particulate matters normally requires two processes: the nucleation of stable atmospheric clusters (1–3 nm) and the growth of these clusters into observable sizes (> 3 nm) (Kulmala et al., 2000). In the first step of NPF, three classical nucleation theories are assumed to induce most NPF: (i) binary nucleation of sulfuric acid and water ( $\text{H}_2\text{SO}_4\text{--H}_2\text{O}$ ); (ii) ternary nucleation involving a third molecule,

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i.e. ammonia ( $\text{NH}_3$ ) ( $\text{H}_2\text{SO}_4\text{--H}_2\text{O--NH}_3$ ) and (iii) ion-induced nucleation. While the exact mechanisms and compounds participating in atmospheric NPF have remained unsolved. Recent studies (e.g., Kerminen et al., 2010) found that (i) nucleation is driven by sulfuric acid and possibly other low-volatile vapors (e.g. organics), and there was no indication from field measurements that water vapor participates into the nucleation; (ii) dimethylamine enhances the nucleation more effectively than ammonia in the atmosphere; and (iii) some organic acids produced from monoterpene oxidation can nucleate with sulfuric acid to form very stable clusters. The second step of particle growth is controlled by the competition between particle growth via complex mechanisms, and scavenging on the larger pre-existing particles (Kerminen et al., 2004). Actually, particle growth probably occurs due to the condensable vapors such as low VOCs. Both theoretical and experimental evidences suggest that a large number of charged and neutral clusters are almost always present in the atmosphere (Kulmala et al., 2007). Thus, the second step may be the limiting step in the atmospheric NPF.

Hong Kong is a coastal city located in the PRD region with a total land area of  $1104\text{ km}^2$  and a population of 7 million. Due to Hong Kong's special geographical location and complex atmospheric conditions, both local emissions and regional transport influence the Hong Kong's air quality. Previous studies indicate that air pollution caused by fine particulate matters and ozone ( $\text{O}_3$ ) in the atmosphere of Hong Kong is severe (e.g., Wang et al., 2005; Guo et al., 2009a). It is well-known that both biogenic and anthropogenic VOCs together with nitrogen oxides ( $\text{NO}_x$ ) are important precursors of surface-level  $\text{O}_3$ , and the oxidation products of photochemical reaction may play a key role in determining the spatial and temporal features of the nucleation events. Hence, NPF events likely occur along with photochemical  $\text{O}_3$  production. To our best knowledge, only one study was conducted on atmospheric new particle growth and shrinkage at a coastal site in Hong Kong (Yao et al., 2010). No study on the characteristics of NPF events initiated or influenced by atmospheric photochemistry has been undertaken in Hong Kong so far. Therefore, it is urgent to enrich the database of particle number

(PN) concentration and size distribution, and to investigate the NPF processes in Hong Kong.

In this study, an intensive field measurement was conducted from 6 September to 30 November in 2010 near the summit of Mt. Tai Mo Shan (TMS; 640 m a.s.l.). The PN concentration and size distribution were characterized and the NPF events were classified based on their strength and growth of geometric mean diameter (GMD). Moreover, the favorable meteorological conditions and possible mechanisms for the nucleation and growth of new particles were discussed by analyzing concurrently-observed meteorological parameters, gaseous pollutants, major hydrocarbons (e.g. isoprene,  $\alpha$ - and  $\beta$ -pinenes, myrcene, limonene, toluene and *m/p/o*-xylenes), intermediate products i.e. methacrolein (MAC) and methyl vinyl ketone (MVK) generated from isoprene oxidation, carbonyl compounds i.e. formaldehyde, acetaldehyde and acetone, and 24–48 h PM<sub>2.5</sub> data.

## 2 Methodology

### 2.1 The topography and climate of Hong Kong

Hong Kong (22°15'N, 114°10'E) is surrounded by the South China Sea on the east, south, and west, and borders the Guangdong city of Shenzhen to the north over the Shenzhen River. The territory's 1104 km<sup>2</sup> area consists of Hong Kong Island, the Kowloon Peninsula, the New Territories, and over 200 offshore islands, of which the largest is Lantau Island where a power plant is located. The highest elevation in the territory is at TMS (957 m a.s.l.).

Hong Kong climate can be affected by severe weather phenomena including tropical cyclones, strong winter monsoon, and thunderstorms with associated squalls that are most frequent from April to September. Generally the dry monsoon season begins in mid to late September and brings three months of pleasantly warm days and clean air

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masses. Hence, we chose these three months (September, October and November) as a sampling period for field measurement at TMS.

## 2.2 The sampling site

Mt. TMS is claimed to be Hong Kong's most misty area due to high daily clouds coverage. The highest point on Mt. TMS is occupied by a Hong Kong Observatory weather radar station. Our monitoring site was located at the waist of Mt. TMS near the mountain summit (22.405° N, 114.118° E, about 640 m a.s.l.) (see Fig. 1).

Surrounding the foot of the mountain are urban centers with a population of 2.23 million. The straight distances between the mountain summit and the urban centers at the foot are about 5–10 km. Farther to the south are the urban centers of the partial New Territory, Kowloon peninsula and Hong Kong Island. To the southwest is the newly-developed residential area of Tung Chung, the international airport and the South China Sea. To the west are the Tuen Mun residential areas and to the south is the South China Sea. To the north and northeast are the city clusters of the inland PRD region. Due to prevailing north/northeast synoptic winds in September–November, polluted air from inland PRD often reaches the sampling site. In addition, because of its unique topography, mountain-valley breezes and sea-land breezes are frequently observed at TMS. These mesoscale circulations enhance the interaction of polluted urban air and the mountain air. The sampling site was set on the rooftop of a building at the waist of TMS (22.405° N, 114.118° E; 640 m a.s.l.).

## 2.3 Measurement techniques

The PN concentration and size distribution, VOCs, carbonyl compounds PM<sub>2.5</sub>, gaseous pollutants and meteorological parameters were measured in this study. The detailed information is presented as follows:

Particle size distributions in the range of 5.5 to 350 nm were continuously measured by a Scanning Mobility Particle Sizer (SMPS, model 5.400, GRIMM, Germany) and

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5 a Condensation Particle Counter (CPC, model 5.400, GRIMM, Germany) system with 44 size bins at 4-min scan intervals. The SMPS was equipped with a nano-differential mobility analyzer (DMA), which separated poly-disperse particles into selected mono-disperse particles according to the electrical mobility of the particles, a function of  
10 their size. Then the CPC counted the number concentration of mono-disperse particles by laser light scattering. A sample air flow rate of  $0.3 \text{ l min}^{-1}$  was fixed for the set of SMPS + CPC with a sheath air flow rate of  $3.0 \text{ l min}^{-1}$ . To minimize the particle loss, ambient aerosols were drawn into the SMPS+CPC system through a 1 m long flexible and conductive tube with an inner diameter of 0.8 cm. Due to significant diffusion  
15 loss of particles with size below 10 nm and diffusion broadening effect in the transfer function, it is important to understand the efficiency (or penetration rate) of the DMA for the accurate measurement of airborne particles. An experimental system determining the aerosol losses in the DMA and the calibration of the SMPS was established and discussed in Wang et al. (2012). In addition, since the Grimm CPC is not widely used,  
20 the discrepancy between Grimm CPC and TSI CPC reported by Schlatter (2006) was used to compare our results with other studies.

To understand the relationships among newly formed particles and trace gases, ambient  $\text{O}_3$ , carbon monoxide (CO), sulfur dioxide ( $\text{SO}_2$ ) and nitric oxide (NO) concentrations were monitored.  $\text{O}_3$  was measured using a commercial UV photometric instrument (Model 400E, Teledyne instrument, CA).  $\text{SO}_2$  was measured by pulsed UV fluorescence (Model 100E, Teledyne instrument, CA). CO was measured with a gas filter correlation, non-dispersive infrared analyzer (Model 300EU, Teledyne instrument, CA) with a heated catalytic scrubber to convert CO to carbon dioxide ( $\text{CO}_2$ ) for baseline determination. NO was detected with a modified commercial MoO/chemiluminescence  
25 analyzer (Model 200E, Teledyne instrument, CA). These analyzers were calibrated daily by injecting scrubbed ambient air (TEI, Model 111) and a span gas mixture. A data logger (Environmental Systems Corporation, Model 8832) was used to control the zero/span calibration and to collect 1-s data, which were averaged to 1-min values.

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To investigate the potential associations of individual VOCs and carbonyl compounds with NPF and particle growth, non-methane hydrocarbons (NMHCs) and carbonyl samples were collected on high O<sub>3</sub> episode days (O<sub>3</sub> ≥ 100 ppbv) i.e. on 27 October–2 November 2010, based on the prediction of meteorological conditions and the review of the PN concentrations on previous days, which were usually related to strong solar radiation, cool and dry air, and low number concentrations of pre-existing large size particles.

Ambient VOC samples were collected using cleaned and evacuated 2-l electro-polished stainless steel canisters. A flow-controlling device was used to collect 1-h integrated samples. Hourly samples were consecutively collected from 09:00 to 16:00 with additional samples collected at 00:00, 03:00, 07:00, 18:00 and 21:00 per day. Finally, 91 VOC samples were collected at TMS on the seven days. The detailed analytical systems and the quality control and assurance for VOC analyses are described in Guo et al. (2009a).

Carbonyl samples were collected on the same sampling days using the silica cartridges impregnated with acidified 2,4-dinitrophenylhydrazine (DNPH). Air samples were drawn through the cartridge at a flow rate of 0.8–0.9 l min<sup>-1</sup> for 120 min (2 h); the flow rate through the cartridges was monitored with a rotameter which was calibrated before and after each sampling. An O<sub>3</sub> scrubber was connected to the inlet of the DNPH–silica cartridge to prevent interference from O<sub>3</sub>. During these sampling days, carbonyl samples were consecutively collected from 07:00 to 21:00 every two hours with additional samples collected at 00:00 and 03:00. Finally, 70 carbonyl samples were collected during this sampling period. All cartridges were stored in a refrigerator at 4 °C after sampling. The sampled carbonyl cartridges were eluted slowly with 2 ml of acetonitrile into a 2-ml volumetric flask. A 20-μl aliquot was injected into the high performance liquid chromatography (HPLC) system through an auto-sampler. The operating conditions of the HPLC and the quality control and detection limit for carbonyl compounds are also provided in Guo et al. (2009a).



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Pre-fired and pre-weighed quartz fabric filters were used to collect  $PM_{2.5}$  samples for 24–48 h by a high-volume sampler ( $1.15\text{--}1.18\text{ m}^3\text{ min}^{-1}$ ). The collected filters were weighted to determine the  $PM_{2.5}$  mass using the standard gravimetric method, and then were stored in  $4^\circ\text{C}$  refrigerator before chemical analysis.

To analyze inorganic ionic species in  $PM_{2.5}$ , i.e. ammonium ( $NH_4^+$ ), sulfate ( $SO_4^{2-}$ ), and nitrate ( $NO_3^-$ ), sample filters were ultrasonically extracted with 10 ml of Milli-Q water for 30 min. The water extracts were filtered through a syringe filter ( $0.2\ \mu\text{m}$  PTFE membrane) to remove filter debris and suspended insoluble particles and then stored in a bottle. The  $100\ \mu\text{l}$  extracts were injected into an ion chromatography (Dionex Corporation, ICS-1000) system for analysis. The detailed operations of the IC instrument are described in Wu and Wang (2007). In addition, organic carbon (OC) and elemental carbon (EC) were measured on a  $0.526\text{ cm}^2$  punch from each filter by thermal optical reflectance (TOR) following the IMPROVE protocol on a DRI Model 2001 thermal/optical carbon analyzer (Atmoslytic Inc., Calabasas, CA, USA) (Chow et al., 2005).

The meteorological conditions were continuously monitored using a mini-weather station (Vantage Pro TM & Vantage Pro 2 – TM Weather Stations, Davis Instruments), which consists of two components: a sleekly designed integrated sensor suite (ISS) to measure outdoor weather conditions, and a data-receiving console to receive the weather data wirelessly. The weather station was placed on the rooftop of a building at the TMS site. Temperature, relative humidity, solar radiation, UV light, wind speed and direction were recorded during the sampling period.

## 2.4 Data processing and analysis

### 2.4.1 Data processing

The first step was to divide the size distribution into 5.5–10 (nucleation mode,  $N_{\text{nuc}}$ ), 10–100 (Aitken mode,  $N_{\text{ait}}$ ), and 100–350 nm (accumulation mode,  $N_{\text{acc}}$ ), and to calculate

the total concentration of particles in each size class through the formula:

$$N_{p-q} = \alpha \sum_p^q N(dp) \quad (1)$$

where  $N_{p-q}$  is the number concentration in each size class,  $N(dp)$  is the number concentration in the size interval  $dp + \Delta dp$  and  $\alpha$  is a normalization factor obtained from the equation:

$$\alpha = \log(dp + \Delta dp) - \log(dp) \quad (2)$$

The SMPS data covered 44 channels spaced in the logarithmic scale. Calculation of the log differences between size channels in these three size modes gave an average  $\alpha$ -value of 0.03812, 0.040202, 0.049418 for  $N_{\text{nucl}}$ ,  $N_{\text{Ait}}$  and  $N_{\text{acc}}$ , respectively. The total PN concentration represented PN concentration from 5.5 to 350 nm in mobility diameter.

In this study, some data of particle size distributions from late September to early October were unavailable due to the malfunction of the DMA instrument caused by very high humidity during this period. The mean PN concentrations in different size classes were calculated with a 95 % confidence level ( $p < 0.05$ ). The linear and power correlations between the tested parameters were indicated by a squared correlation coefficient  $R^2$ .

### 2.4.2 Definition of a particle formation event

In this study, nucleation events were classified into three groups (Classes Ia, Ib and II) according to the classification scheme developed by Dal Maso et al. (2005), following the criteria that the particle mode must start in the distinct nucleation mode size and the particle mode must show signs of growth with a time span of hours. Class I presents the days when the formation and growth rate are determined with a good confidence level. Class I is divided into Class Ia and Ib. The Class Ia event has clear and strong particle formation with little or no pre-existing particles, while a Class Ib

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event is any other Class I event where the particle formation and growth rate still can be determined. A Class II event represents the event where the accuracy of formation rate calculation is questionable due to data fluctuation even though the banana shapes are still observable.

### 2.4.3 Calculation of formation rate, growth rate and condensational sink

In this study, particle formation rates (FR) were represented and quantified using the number concentrations of particles (5.5–10 nm) at the start and end of a linear increase period ( $dN_{10}/dt$ ), without considering the effects of both coagulation and transport due to the small influence on particle production of homogenous nucleation in relatively clean air masses (Jeong et al., 2010). In this study  $dN_{10}/dt$  is a measure of the intensity of nucleation mode particle formation in the event, rather than the nucleation rate, which is typically defined as the number of nuclei clusters growing larger than 1 nm.

The observed growth rate (GR) of the newly formed particles was quantified by the evolution of geometric mean diameters (GMD) of particles from the start to the end of the particle formation events: GMD for each size distribution was used to examine particle growth processes (Dal Maso et al., 2005):

$$GR = \frac{\Delta GMD}{\Delta t} \quad (3)$$

The condensation sink (CS) of newly formed particles, presented as the loss rate of molecules onto existing particles, was estimated by integrating over the aerosol size distribution as follows (Kulmala et al., 2005):

$$CS = 2\pi D \int D_p \beta_M(D_p) n(D_p) dD_p = 2\pi D \sum_i \beta_{Mi} D_{pi} N_i \quad (4)$$

where  $D$  is the diffusion coefficient of the condensing vapor,  $\beta_M$  is the transitional regime correction factor,  $D_{pi}$  is the particle diameter in size class  $i$ , and  $N_i$  is the PN concentration in the respective size class.

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### 3 Results and discussion

#### 3.1 Overview of meteorological conditions and major pollutants

##### 3.1.1 Day-to-day variations

Figures S1–S3 show the temporal variations of PN concentration (5.5–350 nm), trace gases (SO<sub>2</sub>, NO, CO and O<sub>3</sub>) and meteorological conditions (UV, T, RH and wind) at the TMS site in September, October and November 2010, respectively. In general, the PN concentration was the highest in November and the lowest in September, consistent with O<sub>3</sub> mixing ratio. The increase in PN concentration was often associated with elevated CO, NO, SO<sub>2</sub> and O<sub>3</sub> (see Figs. S1–S3), such as on 8–9 and 16–21 September, 23–24 and 27–31 October, 1–3, 8–12, 17–19 and 26–27 November, 2010. These fine particle increase events were probably attributed to primary emissions from local sources and regional transport, and secondary formation via chemical reactions.

During the sampling period, September is the wettest month in 2010 (RH:  $91.2 \pm 1.3\%$ , mean  $\pm 95\%$  C.I.) and the hottest month among these three months ( $23.8 \pm 0.5^\circ\text{C}$ ). More than half of the rainfall was attributed to the two tropical cyclones, namely Lionrock and Fanapi. At the beginning of the sampling campaign, the weather in Hong Kong was fine and very hot on 6 and 7 September owing to the subsiding air associated with severe tropical storm Lionrock. On 15 September, Fanapi was formed over the Western North Pacific to the east of the Luzon Strait. Due to the subsiding air ahead of Fanapi, local weather was generally fine and hot from 16 to 19 September. At the last week of September, the northeast monsoon brought generally fine condition to Hong Kong. Affected by the tropical cyclones and northeast monsoon, the PN concentration and SO<sub>2</sub>, CO, O<sub>3</sub> mixing ratio increased on 8–9 September and 16–21 September, likely associated with primary emissions and regional transport (see Fig. S1).

The mean temperature and relative humidity in October 2010 were lower than in previous years, due to an intense northeast monsoon dominating over Southern China

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during the last six days of the month. The monsoon mainly brought fine and dry weather on 23 and 24 October, and reached the South China coast on the morning of 26 October. Affected by the intense monsoon, local temperature dropped progressively to a minimum of 15.7 °C on 28 October – the lowest of the month. A clear and typical NPF and growth event occurred on 28 October (see Sect. 3.2.1). Cool and very dry conditions with occasionally strong winds on high ground persisted for the last 5 days of the month, accompanied with the events of NPF and growth observed on 27–31 October and the O<sub>3</sub> episode days occurred on 29–31 October (maximum O<sub>3</sub>: 110.4 ppbv on 31 October) (see Fig. S2).

The mean temperature and relative humidity in November were the lowest ( $T$ : 17.1 ± 0.5 °C; RH: 77.7 ± 1.6 %) among the three months. A dry northeast monsoon over Southern China brought fine and dry weather to Hong Kong on the first three days of the month. An episode of NPF events was consistently observed from 1–3 November. Affected by a broad cloud band over the coastal areas of Guangdong, the weather became rainy and cooler from 4 to 6 November. It remained fine and dry from 8 to 12 November during which another episode of NPF events occurred. Dominated by the northeast monsoon, local weather was mainly fine apart from some haze between 17 and 24 November, during which an ozone episode was recorded mainly due to regional transport on 17–19 November. On 26 November, another NPF event was observed under a cool and dry weather condition, mainly caused by the reinforcement of the northeast monsoon reaching the South China coastal areas (see Fig. S3 and Sect. 3.2).

### 3.1.2 Diurnal variations

#### Meteorological conditions and gaseous pollutants

Figure 2 shows mean diurnal variations of SO<sub>2</sub>, NO, CO, O<sub>3</sub> and meteorological conditions (UV, RH,  $T$  and wind) at TMS, respectively.

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The diurnal variation of solar radiation indicated the time of sunrise (06:00 local time LT) and sunset (18:00 LT) was similar in September, October and November. Diurnal patterns of  $O_3$  exhibited a high peak in the afternoon and relatively low concentrations at night and in the morning. The mean daily  $O_3$  mixing ratio was the highest in November ( $67.4 \pm 3.0$  ppbv), and the lowest in September ( $37.7 \pm 3.6$  ppbv). The daily maximum  $O_3$  mixing ratio appeared at 14:00 ~ 15:00 LT, likely due to photochemical reactions and/or regional transport. The average daily maximum  $O_3$  mixing ratio was  $70.2 \pm 6.3$  ppbv.

In contrast, broad peaks of  $SO_2$  and NO were observed at 10:00 ~ 15:00 LT and at 10:00 LT, respectively, a slight delay compared to the time (08:00 ~ 09:00) of peaks usually appeared at the urban site, perhaps due to the influence of valley breeze which brought the urban primary pollutants from mountain foot to the summit. The peak NO value ( $5.3 \pm 0.5$  ppbv) at TMS was much lower than those at urban sites (Guo et al., 2009a).

## Particle number concentration and size distributions

### i) Mean particle number concentration

In this study, the mean total number concentration in the size range of 5.5–350 nm at the mountain site was  $7.86 \pm 0.66 \times 10^3 \text{ cm}^{-3}$ , falling into the range of typical values ( $5\text{--}25 \times 10^3 \text{ cm}^{-3}$ ) reported for the suburban/urban sites of North America and Europe. However, the mean PN levels at TMS were lower than that at a coastal site in Hong Kong (Yao et al., 2010), and was about 50 % less than those at some rural sites in China such as Yufa ( $17.0 \times 10^3 \text{ cm}^{-3}$ ) (Yue et al., 2009) and Xinken ( $16.3 \times 10^3 \text{ cm}^{-3}$ ) (Liu et al., 2008), probably due to the different altitudes of sampling sites and the emission sources near the sites. In contrast, the PN level at TMS was much higher than that ( $2.03 \times 10^3 \text{ cm}^{-3}$ ) at a remote background station i.e. Mt. Waliguan in North-western China (Kivekäs et al., 2009), mainly attributed to the prevailing northeastern

synoptic winds over the TMS site, which brought polluted air masses from mainland China to Hong Kong. Furthermore, the mean values for nucleation mode ( $N_{\text{nuc}}$ , 5.5–10 nm), Aitken mode ( $N_{\text{Ait}}$ , 10–100 nm), and accumulation mode ( $N_{\text{acc}}$ , 100–350 nm) were  $1.27 \pm 0.18 \times 10^3$ ,  $1.63 \pm 0.35 \times 10^3$ , and  $0.75 \pm 0.08 \times 10^3 \text{ cm}^{-3}$  in September, respectively;  $0.63 \pm 0.05 \times 10^3$ ,  $5.65 \pm 0.76 \times 10^3$ , and  $0.74 \pm 0.06 \times 10^3 \text{ cm}^{-3}$  in October, respectively; and  $1.74 \pm 0.13 \times 10^3$ ,  $6.06 \pm 0.85 \times 10^3$ , and  $1.11 \pm 0.12 \times 10^3 \text{ cm}^{-3}$  in November, respectively. The mean highest  $N_{\text{nuc}}$  ( $1.74 \pm 0.13 \times 10^3 \text{ cm}^{-3}$ ) was observed in November, suggesting frequent nucleation-mode particle formation at the TMS site during this period.

## ii) Diurnal variations of particle number concentration and size distribution

Figure 3 shows the diurnal variations of PN concentration (5.5–350 nm) in September, October and November at TMS. As illustrated, the diurnal pattern of PN concentration was different for each of the three months. In September, the PN concentration began to rise at 08:00, reached its peak at noon, and then gradually decreased. In October, the PN concentration did not show significant difference between daytime and nighttime, likely due to many rainy and cloudy days in early and middle October. The relatively higher PN concentration in October than in September was possibly caused by the  $\text{O}_3$  episode event occurred in late October (i.e. 27–31 October). In November, the PN concentration increased at 08:00, reached the peak value at 16:00, and then decreased. The typical diurnal pattern of PN concentration in September and November i.e. maxima at noon and afternoon and minima at night and early morning, similar to that of  $\text{O}_3$ , implied that there might be photochemical formation of new particles.

To further investigate the influence of photochemical reaction on NPF at TMS, the diurnal patterns of PN concentration and size distribution on  $\text{O}_3$  episode and non- $\text{O}_3$  episode days during the sampling period are drawn in Fig. 4a, b. It is noteworthy that a day with the maximum  $\text{O}_3$  mixing ratio higher than 100 ppb was defined as an  $\text{O}_3$  episode day, while a day with the maximum  $\text{O}_3$  mixing ratio lower than 50 ppb was defined as a non- $\text{O}_3$  episode day. Significant difference in diurnal pattern was apparently

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observed between  $O_3$  and non- $O_3$  episode days (see Fig. 4a). During daytime hours (i.e. 08:00 ~ 18:00), the PN concentration on  $O_3$  episode days was much higher than that on non- $O_3$  episode days. The PN concentration started to increase at 08:00 and reached a peak value at 16:00 on  $O_3$  episode days, whereas only a low and broad peak was observed on non- $O_3$  episode days. The high and sharp peak at daytime on  $O_3$  episode days was likely attributed to local and/or regional photochemical particle formation and, to a lesser extent, the effect of valley breeze which brought PN in urban area (foot of the mountain) to the sampling site (i.e. summit of the mountain). On the other hand, the low and broad peak on non- $O_3$  episode days was probably due to the influence of valley breeze and regional transport. Indeed, further inspection on the size distributions of PN on  $O_3$  episode days found that there was a small peak at nucleation mode, which indicated nucleation mode particle burst occurred on  $O_3$  episode days. Moreover, the peak diameter for the Aitken mode was 32 nm, smaller than 65 nm on non- $O_3$  episode days, indicating that Aitken mode particles on  $O_3$  episode days were probably from condensational growth and coagulation of nucleation mode particles, and from primary emissions such as traffic.

### 3.2 Classification of new particle formation events

According to the classification scheme developed by Dal Maso et al. (2005), the typical Class Ia events occurred on 28 October, 1, 9 and 12 November, 2010 at TMS. The examples of Class Ib observed in this study included 29 October, 2, 8 and 26 November, 2010. On 27 and 30 October, 11 and 28 November, Class II events were mainly observed.

The mean growth rate for the nucleation events was calculated by the slope of GMD against time during the period of particle growth until 25 nm. By computation, the growth rates of Class I events observed at TMS were from 2.95 to 5.00  $\text{nm h}^{-1}$  (average 3.90  $\text{nm h}^{-1}$ ), in line with the range of new particle growth rates observed in rural/suburban locations i.e. 1–10  $\text{nm h}^{-1}$  (Kulmala et al., 2004a).

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These NPF events were often associated with elevated SO<sub>2</sub> and O<sub>3</sub>. Hence, two Class Ia cases were studied to identify the sources and mechanisms that were responsible for NPF and growth in this study. Detailed description was provided as follows:

### Case I – new particle formation under a clean background

5 New particle formation event (nucleation mode) on 28 October was clearly observed after raining on 27 October, in line with the fact that nucleation events often occur after a sharp decrease in concentrations of pre-existing particles with large surface area. The minimum temperature of 12.5 °C recorded on 28 October was the lowest in October since 1988. The NFP started when wind speed increased to achieve maximum (5.26 ms<sup>-1</sup>). As shown in Fig. 5, the nucleation mode appeared in the measurement range at about 10:00 and increased substantially with an observable growth rate (2.95 nm h<sup>-1</sup>). The nucleation mode particle concentration ( $N_{\text{nuc}}$ ) increased rapidly from ~ 1.28 × 10<sup>3</sup> cm<sup>-3</sup> to 23.5 × 10<sup>3</sup> cm<sup>-3</sup> at noon with a formation rate of 4.11 cm<sup>-3</sup> s<sup>-1</sup>. The nucleation modes shifted towards Aitken modes during the afternoon until 16:00, perhaps due to the condensation growth and coagulation of the nucleation mode particles with larger particles. Since the concentration of around 0.2–1 × 10<sup>6</sup> cm<sup>-3</sup> of sub-5-nm particles is needed to induce a growth rate of 1 nm h<sup>-1</sup> by self-coagulation (Kulmala et al., 2004b), in this event the initial nucleation mode particle concentration of below 3 × 10<sup>4</sup> cm<sup>-3</sup> indicated that self-coagulation only made a minor contribution to the particle growth. In addition, after 14:00, the GR decreased with the condensation sink (CS) increased from 0.0186 s<sup>-1</sup> to 0.0377 s<sup>-1</sup>, indicating increased large surface area particles scavenged potential condensable vapors responsible for particle growth. After 16:00, the total PN concentration decreased rapidly, perhaps due to the dilution effect and coagulation losses. Obviously synchronous increases of O<sub>3</sub> with increased PN concentration suggested the photochemical reaction may contribute to the particle formation. Meanwhile, simultaneously slow increase of SO<sub>2</sub>, decrease of CO, and rise of the calculated (HYSPLIT 4) boundary layer height were observed at the start of this event (Figs. 5 and S4), indicating that these favorite conditions for NPF were likely

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a result of breakup of the morning inversion that mixed SO<sub>2</sub>-enriched and CO-depleted air from aloft and diluted the accumulated urban emissions (Zhang et al., 2004).

The correlations of organic aerosol precursors i.e. VOCs and carbonyls with nucleation mode PN concentrations were analyzed to preliminarily investigate the formation mechanism of the event on 28 October. The concentrations of biogenic VOC (BVOC) and anthropogenic VOC (AVOC) species, which were demonstrated to be the potential precursors of secondary organic aerosols (SOA) reached the maxima at 07:00 ~ 09:00 (i.e. BVOCs: isoprene 96 pptv,  $\alpha$ -pinene 24 pptv,  $\beta$ -pinene 6 pptv, myrcene 5 pptv and limonene 329 pptv, and AVOCs: propene 510 pptv, toluene 6079 pptv, ethyltoluenes 180 pptv and triethylbenzenes 246 pptv), 1–3 h earlier than the time when the 5.5 nm nucleation mode particles were observed. It is known that the oxidation products of BVOCs and AVOCs have a potential contribution to the new aerosol formation and growth (Griffin et al., 1999; Kulmala et al., 2004a; Hatch et al., 2011); and there is a time delay before VOC turns to condensable species which promote nucleation. Hence, the correlation of hourly PN with VOC species measured an hour earlier was explored in this study. The BVOCs i.e. isoprene,  $\alpha$ -pinene and  $\beta$ -pinene and the AVOCs i.e. propene, ethyltoluenes and triethylbenzenes showed stronger correlations with  $N_{\text{nuc}}$  than other VOCs species (see Table 1). Good correlations of  $N_{\text{nuc}}$  with isoprene,  $\alpha$ - and  $\beta$ -pinenes, propene and aromatics suggested that BVOCs and AVOCs could contribute to the NPF event. Indeed, chamber study suggested that the possible mechanism was the oxidation of VOCs which generated condensable oxygenated VOCs (e.g., Ng et al., 2008). In the atmosphere, methacrolein (MAC) and methyl vinyl ketone (MVK) are the major intermediate products generated from isoprene oxidation. In this event MAC and MVK showed moderate linear correlations with  $N_{\text{nuc}}$  ( $R^2 = 0.414$  and  $0.677$ , respectively), suggesting that oxidation of BVOCs was a pathway to the nucleation mode particles.

Additionally, the correlation of  $N_{\text{nuc}}$  with O<sub>3</sub> mixing ratio ( $R^2 = 0.485$ ) was much higher than that with SO<sub>2</sub> ( $R^2 = 0.030$ ), CO ( $R^2 = 0.004$ ) and NO ( $R^2 = 0.052$ ) (see

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Table 2). All the evidences above suggested that the NPF observed on 28 October was likely due to the photochemical formation via VOC oxidation processes.

## Case II – new particle formation under a polluted background

Another NPF event (nucleation mode) was clearly observed on 1 November (Fig. 6). Compared to the nucleation event occurred on 28 October, the background PN concentration on 1 November before particle nucleation was higher, which might inhibit nucleation process due to high concentrations of the pre-existing large surface area particles. However, on the morning of 1 November, the winds turned to north direction with speeds from calm to light wind, and the CO levels sharply decreased at the start of this event, probably due to the rise of the boundary layer height which caused the breakup of the morning inversion and diluted the accumulated pollutants, providing a relatively clean atmosphere (Figs. 6 and S5). Consequently, with rapid increase of solar radiation intensity and temperature, the nucleation mode appeared in the observable range at about 10:00 and the nucleation mode PN concentration increased rapidly from  $\sim 3.93 \times 10^3 \text{ cm}^{-3}$  to  $5.47 \times 10^3 \text{ cm}^{-3}$  in two hours (growth rate:  $5.0 \text{ nm h}^{-1}$ ). Same as that on 28 October, the nucleation modes shifted towards Aitken modes in the afternoon until 14:00 on 1 November, indicating the growth of nucleation mode particles into Aitken mode particles. At 14:00 ~ 18:00, the Aitken mode PN concentration remained a stable and high level. No obvious increase in  $N_{\text{nuc}}$  in these four hours might be explained by the fact that  $N_{\text{nuc}}$  were anti-correlated with aerosol surface area concentrations, i.e. lower  $N_{\text{nuc}}$  were recorded during the periods of higher aerosol surface area concentrations. This is consistent with the observations reported in other studies (Guo et al., 2008; Liu et al., 2008). It was also supported by the condensation sink in this event that increased from  $0.0228 \text{ s}^{-1}$  to  $0.0455 \text{ s}^{-1}$  during the condensation process, which inhibited the nanometer-scale particle nucleation causing a decreased  $N_{\text{nuc}}$  concentration from  $5.47 \times 10^3 \text{ cm}^{-3}$  to  $1.36 \times 10^3 \text{ cm}^{-3}$ . After 18:00, the wind direction switched to the east which might bring aged air masses to the mountain summit. Different from the NPF and growth freshly formed at the site, the aged air masses

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transported from local urban areas and/or PRD region had high number concentrations of Aitken mode and accumulation mode particles.

In this event, only isoprene and limonene showed strong correlations with  $N_{\text{nuc}}$  (see Table 1). No correlations were found between anthropogenic VOCs and  $N_{\text{nuc}}$ . Different from the event on 28 October, the concentrations of most BVOC species started to increase at 07:00 and reached the maximum concentrations at 13:00, i.e. isoprene 118 pptv,  $\alpha$ -pinene 10 pptv, and  $\beta$ -pinene 6 pptv, though myrcene and limonene reached the maximum concentrations at 07:00, i.e. myrcene 6 pptv and limonene 33 pptv. The various temporal variations and correlations of different BVOCs with the  $N_{\text{nuc}}$  indicated different oxidation processes of VOCs on these two events (i.e. 28 October and 1 November). Additionally, the correlation between  $\text{O}_3$  mixing ratio and  $N_{\text{nuc}}$  ( $R^2 = 0.491$ ) was slightly higher than that of  $\text{SO}_2$  ( $R^2 = 0.314$ ), while CO ( $R^2 = 0.013$ ) and NO ( $R^2 = 0.121$ ) both had no obvious correlations with  $N_{\text{nuc}}$ , confirming that the nucleation mode particles observed on 1 November were likely formed during the processes of photochemical reactions (Table 2). Given that the transformation from gas-phase pollutants to nucleation mode particles requires sufficient time, there may be a delay between the measured gaseous pollutant i.e.  $\text{SO}_2$  and nucleation mode particles. Indeed, a much better correlation between  $\text{SO}_2$  mixing ratio and the  $N_{\text{nuc}}$  with 2-h lag was found ( $R^2 = 0.637$ ), suggesting the possible involvement of  $\text{H}_2\text{SO}_4$  vapor or/and  $\text{SO}_2$  oxidation products in this nucleation event. Since the  $\text{SO}_2$  background concentration in the PRD was much higher than that required in nucleation events observed in clean environments, the nucleation of sulfuric acid should be dominant in this event apart from the contributions of oxidation of BVOCs. More detailed discussion is given in Sect. 3.3.2.1.

Previous studies on new particle growth indicate that particle growth rate depends on temperature and concentrations of available condensable vapor (Kulmala et al., 2004a). It is also reported that the semi- or no-volatile oxidation products of VOCs account for more than 70 % of the materials for the particle growth, compared to  $\text{H}_2\text{SO}_4$  condensation which only accounts for 8.8 % of the observed growth (Boy et al., 2005).

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the solar radiation intensity was as low as  $46 \text{ Wm}^{-2}$ , indicating that the occurrence of NPF depended not only on the presence of intense sunlight but also on the properties of the present air masses (Wehner and Wiedensohler, 2003).

In this study, temperatures on the NPF days were lower ( $12.5 \sim 19.7^\circ\text{C}$ ) than the average temperature of the whole sampling period. For example, the minimum temperature on 28 October was  $12.5^\circ\text{C}$  when NPF occurred. This study also found that NPF took place at relative humidity of 55–80%, similar to those reported by O'Dowd et al. (2002).

In the NPF events, the northeast winds were prevailing, mainly due to intense monsoon over Southern China, which brought cool and very dry weather conditions to Hong Kong. Nevertheless, wind speeds did not show any difference between NPF days and the other days.

By inspecting the start time of the NPF events occurred at TMS, it was obvious that all the NPF events were initiated at about 10:00 ~ 11:00, consistent with other studies (Stanier et al., 2004b; Guo et al., 2008; Modini et al., 2009). Stanier et al. (2004b) claimed that nucleation was typically observed starting around 09:00 EST in Pittsburgh, USA. In Eastern Australia, the NPF events usually began at 10:00 ~ 11:00 local time (Guo et al., 2008; Modini et al., 2009), except one case in which a particularly strong nucleation event began early morning (08:00), suggesting some different, or stronger precursor sources than the sources for all other nucleation events (Modini et al., 2009). In this study, the similar start time for all the NPF events indicated that the events were mainly induced and controlled by some certain sources and pathways, which were likely to be the homogeneous and/or heterogeneous nucleation of low volatile condensable gases such as sulfuric acid and some dominated organic compounds. For instance, the diurnal variations of BVOCs i.e. isoprene,  $\alpha$ -pinene and  $\beta$ -pinene showed that their mixing ratios steadily increased at 07:00, reached a maximum level in early afternoon, and then gradually decreased to low levels at night, which were consistent with the diurnal pattern of  $N_{\text{nuc}}$  or with 1–2 h time lag, indicating that the BVOC and its oxidation products had a close association with  $N_{\text{nuc}}$  formation.

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In addition, low SO<sub>2</sub>, NO and CO mixing ratios on the NPF days revealed that clean air masses were favorable to the NPF. Moreover, high O<sub>3</sub> mixing ratios in these events suggested that photochemical reaction occurred during the NPF processes.

### 3.3.2 Potential precursors of new particle formation

#### Trace gases

Table 2 shows the growth rate, formation rate and condensation sink of each NPF and growth event observed at TMS, and the correlations of SO<sub>2</sub>, CO, NO and O<sub>3</sub> with the  $N_{\text{nuc}}$  for each NPF event. The  $N_{\text{nuc}}$  showed a better correlation with O<sub>3</sub> than with SO<sub>2</sub>, CO and NO for most events, except for the event on 26 November, 2010, suggesting that photochemical reactions of VOCs may play a key role in the NPF. In addition, the events observed on 17 October, 1 and 8 November, 2010, showed that  $N_{\text{nuc}}$  had relatively high correlations with both O<sub>3</sub> and SO<sub>2</sub> mixing ratios (two hours ahead  $N_{\text{nuc}}$ ), especially on 1 November, the SO<sub>2</sub> even had a better correlation with  $N_{\text{nuc}}$  than with O<sub>3</sub>. This may indicate that SO<sub>2</sub> made an important contribution to the NPF in the above event. Since hydroxyl radicals (OH) and H<sub>2</sub>SO<sub>4</sub> vapor were not measured, we used the product of ultraviolet radiation and SO<sub>2</sub> (i.e. UV × SO<sub>2</sub>) as a surrogate parameter for H<sub>2</sub>SO<sub>4</sub> production (Petäjä et al., 2009). Better correlation between  $N_{\text{nuc}}$  and the value of (UV × SO<sub>2</sub>) (a proxy for the H<sub>2</sub>SO<sub>4</sub> production rate) during the nucleation events was observed, suggesting a critical role played by H<sub>2</sub>SO<sub>4</sub> in the nucleation and early growth (see Table 2).

Figure 7 shows the correlation of (UV × SO<sub>2</sub>) with condensation sink during the NPF events at TMS. It was used to predict the required H<sub>2</sub>SO<sub>4</sub> production rates for nucleation and growth of particles as a function of CS, RH, and temperature (Stanier et al., 2004b). The panel of Fig. 7 was divided into 2 qualitative regions by a 45° line. The lower right region indicated more favorable conditions to nucleation than upper left region. The threshold was calculated at representative ground-level RH and temperature values. As the UV × SO<sub>2</sub> product increased and the condensational sink values

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decreased, the nucleation events were likely observed. It can be seen that most of data points fell below the dividing line. Compared to the result reported by Stanier et al. (2004b), most values of  $UV \times SO_2$  in this study were higher, confirming that this mountain site was a sulfur-rich environment, providing abundant precursor to the NPF.

Indeed, the data points with high  $N_{nuc}$  concentrations fell into the right side of the figure, suggesting that nucleation events occurred at high  $UV \times SO_2$ . In other word, the  $SO_2$  oxidation products may be the trigger of NPF at TMS. Moreover, the low and relatively constant condensational sink values with low temperature and relative humidity (cold and dry) were the favorable conditions for nucleation at the site.

## VOCs and carbonyl compounds

Generally, oxidation of compounds with different positions of unsaturated bonds and number of carbon atoms in the molecules can result in different aerosol growth rates and SOA mass yields (Griffin et al., 1999). In this study, it was found that the BVOCs showed much stronger correlations with  $N_{nuc}$  than the AVOCs in most cases, mainly due to the higher reactivity of BVOCs against the important atmospheric oxidants OH,  $NO_3$  and  $O_3$  (Table 1) (e.g., Atkinson, 2000). Particularly, isoprene,  $\alpha$ -pinene and  $\beta$ -pinene had a good to moderate correlation with  $N_{nuc}$  ( $R_{isoprene}^2 = 0.303\text{--}0.653$ ,  $R_{\alpha\text{-pinene}}^2 = 0.307\text{--}0.498$  and  $R_{\beta\text{-pinene}}^2 = 0.418\text{--}0.705$ ), indicating that these species might be the major potential contributors to NPF on these events at TMS. The higher correlations of isoprene with  $N_{nuc}$  than those of  $\alpha$ -pinene and  $\beta$ -pinene in most NPF events found in this study are consistent with a previous study, in which the contributions of isoprene to SOA were found to be higher than those of  $\alpha$ - and  $\beta$ -pinenes (Fu and Kawamura, 2011). Occasionally, other BVOCs i.e. myrcene and limonene (i.e.  $R_{myrcene}^2 = 0.672$  on 2 November and  $R_{limonene}^2 = 0.705$  on 1 November), and some AVOC species i.e. xylenes and toluene also had high correlations with  $N_{nuc}$  (i.e.  $R_{toluene}^2 = 0.732$  and  $R_{xylene}^2 = 0.567$  on 30 October), indicating that the NPF was not

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dominated by one or two BVOC species, but by a number of factors such as the mixing ratios of BVOCs, AVOCs, O<sub>3</sub>, OH and NO<sub>x</sub> levels, and the degree of oxidation due to the different reaction sensitivity of individual VOCs (Hallquist et al., 2009).

Additionally, it was found that the correlation coefficient  $R^2$  for the power curves of BVOCs vs.  $N_{\text{nuc}}$  was much higher than that of their linear correlation, and the power exponent of isoprene,  $\alpha$ -pinene and  $\beta$ -pinene concentrations was from 1.05 to 2.54, 0.99 to 2.23 and 1.12 to 2.31, respectively. The recent studies on SOA formation by BVOCs found that as the BVOC species were oxidized by O<sub>3</sub> and OH, various products with differing volatilities were formed. When these products exceeded their saturation vapor pressure, homomolecular and heteromolecular homogeneous nucleation between H<sub>2</sub>SO<sub>4</sub> and organic vapors could occur with an exponent coefficient in the range of 1 to 2 (e.g., Paasonen et al., 2010). In this study, the exponents of isoprene,  $\alpha$ -pinene and  $\beta$ -pinene with  $N_{\text{nuc}}$  mainly fell into the range of 1–2, suggesting that both sulfuric acid and organic vapor may play important roles on NPF at TMS.

Laaksonen et al. (2008) concluded that oxidation products of  $\alpha$ -pinene and other monoterpenes were not only present in the freshly nucleated particles, but also determined their growth rates. In addition, increased studies focused on the potential SO<sub>2</sub>/acidity contribution to secondary aerosol formation and growth through heterogeneous reactions in the presence of an acid catalyst (e.g., Hatch et al., 2011). Moreover, Villani et al. (2009) reported that the particle growth can be drastically influenced by some volatile compounds in a humid environment, although these organic compounds present at low concentrations. In this study, two NPF events occurred on 27 and 30 October, with the highest GR (9.44 nm h<sup>-1</sup>) and lowest GR (1.53 nm h<sup>-1</sup>), respectively, among the six NPF events were compared. The ambient air was cleaner and humidity was higher on 27 October than that on 30 October, and the correlation coefficient ratio of BVOCs to AVOCs ( $R_{\text{BVOCs}}^2/R_{\text{AVOCs}}^2$ ) and  $R_{\text{UV}\times\text{SO}_2}^2$  were higher on 27 October than that on 30 October, indicating that the GR on 27 October was mainly influenced by the oxidation products of BVOCs, sulfuric acid, and to a lesser extent, the ambient

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humidity, which was related to GR by a positive linear correlation ( $R^2 = 0.68$ ) for the six NPF events.

Different from the NPF event on 27 October, the GR on 30 October may be dominated by the interactions between biogenic and anthropogenic aerosol precursors. Recent evidence suggested that chemical interactions between anthropogenic and biogenic VOCs were important in the formation of secondary aerosols (e.g., Brock et al., 2003). In this study, based on strong correlations between  $N_{\text{nuc}}$  and AVOC species and moderate correlations between  $N_{\text{nuc}}$  and BVOC species, the SOA formation may be dominated by AVOCs oxidation and/or be mediated by anthropogenic pollutants on the transformation of BVOCs to SOA on 30 October, 2010. The low formation rate and growth rate may be related to the increased condensation and the coagulation sinks caused by a large number of preexisting aerosol particles, inhibiting the formation and growth of freshly nucleated particles on 30 October 2010.

## Particulate matter

Further investigation on the major inorganic chemical species i.e. sulfate ( $\text{SO}_4^{2-}$ ), nitrate ( $\text{NO}_3^-$ ) and ammonium ( $\text{NH}_4^+$ ) and OC/EC in  $\text{PM}_{2.5}$  collected from 7 September to 26 November 2010 showed that the correlations between ion mass concentrations in  $\text{PM}_{2.5}$  and PN concentration were weak, i.e.  $R^2 = 0.214$  for  $\text{NH}_4^+$ ,  $R^2 = 0.025$  for  $\text{NO}_3^-$  and  $R^2 = 0.252$  for  $\text{SO}_4^{2-}$ . Due to strong and frequent photochemical formation from VOC species may contribute a significant fraction to the PN concentration in the atmosphere at TMS. Consequently, the correlations between inorganic ion mass concentrations and the PN concentration were weakened. This can be further evidenced by the fact that the OC/EC ratios obtained during the NPF events were 2.16 ~ 2.53. Typically, EC is emitted from combustion sources with limited chemical transformations, while OC can be emitted from primary sources and/or generated from chemical

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reactions among primary gaseous VOCs in the atmosphere. Therefore, The OC/EC ratios exceeding 2.0 have been used to indicate the presence of SOA (e.g., Turpin et al., 1990). The average OC/EC ratio at TMS (i.e. 2.32), indicated a certain contribution of SOAs to the increase in PN concentration.

## 4 Conclusions

Particle number concentrations and size distributions were measured at a suburban site near the mountain summit of TMS from September to November 2010. The overall average concentration of nucleation mode ( $N_{\text{nuc}}$ , 5.5–10 nm), Aitken mode ( $N_{\text{Ait}}$ , 10–100 nm), and accumulation mode ( $N_{\text{acc}}$ , 100–350 nm) were  $0.65 \pm 0.05 \times 10^3 \text{ cm}^{-3}$ ,  $4.75 \pm 0.63 \times 10^3 \text{ cm}^{-3}$  and  $0.86 \pm 0.07 \times 10^3 \text{ cm}^{-3}$ , respectively. The high PN concentration was likely attributed to local photochemical particle formation and/or regional transport of aged particles. The dry monsoon season begun in late October and early November created favorable conditions for NPF. The NPF events were observed on 12 out of 35 days at TMS based on the evolution of particle size distributions and PN concentrations. Back trajectory analyses suggested that particle nucleation and growth events were associated with less polluted cool and dry air from the PRD region caused by the winter monsoon.

Good to moderate correlations between  $\text{O}_3$  and  $N_{\text{nuc}}$  were found for almost all NPF events, suggesting that NPF events were closely related to photochemical reactions. Initial inspection on the potential precursors of the NPF events at TMS revealed that  $N_{\text{nuc}}$  had good correlations with  $(\text{UV} \times \text{SO}_2)$  on some days i.e. 8 November, with BVOCs on other days i.e. 29 October and 2 November, with both  $(\text{UV} \times \text{SO}_2)$  and BVOCs on 1 November, with both BVOCs and AVOCs on 28 October, and with VOCs and  $(\text{UV} \times \text{SO}_2)$  on 27 October. The results suggested that  $\text{SO}_2$  products, BVOCs and AVOCs might be the main contributors to  $N_{\text{nuc}}$ . The  $\text{SO}_2$  products might dominate the nucleation process while the VOCs could mainly involve in the condensation growth. Further investigation on correlations between  $N_{\text{nuc}}$  and VOC species showed that

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isoprene,  $\alpha$ -pinene and  $\beta$ -pinene might be the major potential BVOCs responsible for the NPF events at the mountain site.

**Supplementary material related to this article is available online at:**

<http://www.atmos-chem-phys-discuss.net/12/12119/2012/>

[acpd-12-12119-2012-supplement.zip](#).

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**Table 1.** Power correlations of BVOCs, AVOCs and carbonyls with the  $N_{\text{nuc}}$  for each NPF event observed during the sampling campaign.

Date	Biogenic VOCs						Anthropogenic VOCs					Carbonyl Compounds		
	$R^2$ (Isoprene)	$R^2$ ( $\alpha$ -Pinene)	$R^2$ ( $\beta$ -Pinene)	$R^2$ (Myrcene)	$R^2$ (Limonene)	$R^2$ (Propene)	$R^2$ (Benzene)	$R^2$ (Toluene)	$R^2$ (Xylene*)	$R^2$ (Ethyltoluene*)	$R^2$ (Trimethylbenzene*)	$R^2$ (Formaldehyde)	$R^2$ (Acetaldehyde)	$R^2$ (Acetone)
Class Ia														
28 Oct 2010	0.645	0.465	0.705	0.101	0.013	0.825	0.057	0.318	0.086	0.643	0.760	0.531	0.108	0.400
1 Nov 2010	0.653	0.204	0.231	0.015	0.705	0.369	0.272	0.310	0.275	0.200	0.080	0.597	0.470	0.435
Class Ib														
29 Oct 2010	0.520	0.362	0.116	0.060	0.352	0.064	0.014	0.034	0.029	0.054	0.169	0.227	0.336	0.425
2 Nov 2010	0.754	0.498	0.423	0.672	0.195	0.052	0.044	0.016	0.019	0.056	0.105	0.673	0.609	0.640
Class II														
27 Oct 2010	0.303	0.307	0.418	0.470	0.08	0.572	0.155	0.155	0.087	0.031	0.080	0.104	0.101	0.030
30 Oct 2010	0.620	0.440	0.488	0.269	0.083	0.135	0.383	0.732	0.567	0.433	0.196	0.955	0.776	0.698

\* Xylene includes *m*-xylene, *p*-xylene and *o*-xylene; Ethyltoluene includes 2-ethyltoluene, 3-ethyltoluene and 4-ethyltoluene; Trimethylbenzene includes 1,3,5-trimethylbenzene, 1,2,4-trimethylbenzene and 1,2,3-trimethylbenzene.

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**Table 2.** Linear correlations of  $O_3$ ,  $SO_2$ ,  $SO_2^*$  (two hours lag),  $UV \times SO_2$ , NO and CO with the  $N_{nuc}$  and GR, FR, CS for each NPF event observed during the sampling campaign.

Date	$R^2$ ( $O_3$ )	$R^2$ ( $SO_2$ )	$R^2$ ( $SO_2^*$ )	$R^2$ ( $UV \times SO_2$ )	$R^2$ (NO)	$R^2$ (CO)	Growth Rate ( $nm h^{-1}$ )	Formation Rate ( $cm^{-3} s^{-1}$ )	Condensation Sink ( $s^{-1}$ )
Class Ia									
28 Oct 2010	0.485	0.030	0.063	0.263	0.052	0.004	2.95	4.11	0.0282
1 Nov 2010	0.491	0.314	0.637	0.694	0.121	0.013	5.00	0.29	0.0342
9 Nov 2010	0.486	0.074	0.187	0.387	0.064	0.092	3.02	0.34	0.0354
12 Nov 2010	0.562	0.121	0.264	0.273	0.084	0.189	4.63	4.53	0.0234
Class Ib									
29 Oct 2010	0.386	0.052	0.153	0.267	0.042	0.064	4.00	4.32	0.0165
2 Nov 2010	0.576	0.145	0.248	0.314	0.121	0.147	2.69	0.86	0.0364
8 Nov 2010	0.764	0.354	0.589	0.613	0.007	0.165	4.52	0.42	0.0296
26 Nov 2010	0.173	0.097	0.262	0.312	0.074	0.121	2.41	3.89	0.0144
Class II									
27 Oct 2010	0.586	0.245	0.475	0.486	0.116	0.231	9.44	3.23	0.0197
30 Oct 2010	0.684	0.102	0.253	0.261	0.053	0.115	1.53	0.51	0.0398
11 Nov 2010	0.783	0.062	0.185	0.348	0.021	0.051	8.43	3.01	0.0165
28 Nov 2010	0.316	0.013	0.174	0.264	0.008	0.031	6.45	4.01	0.0069

$(SO_2)^*$  means that  $SO_2$  mixing ratio is two hours ahead of  $N_{nuc}$ .

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**Table 3.** Summary of chemical and meteorological conditions during the NPF events observed at TMS.

Date	Start Time (LT)	Mean hourly solar radiation (08:00–16:00) ( $\text{Wm}^{-2}$ )	Mean temperature ( $^{\circ}\text{C}$ )	Mean relative humidity (%)	Mean wind speed ( $\text{ms}^{-1}$ )	Mean wind direction ( $^{\circ}$ )	Mean $\text{SO}_2$ (ppb)	Mean $\text{O}_3$ (ppb)	Mean NO (ppb)	Mean CO (ppb)	Mean PN concentration ( $\text{cm}^{-3}$ )
Class Ia											
28 Oct 2010	1000	560.8	12.5	68.5	3.0	184.0	4.3	58.0	4.3	420.5	$7.0 \times 10^3$
1 Nov 2010	1015	565.5	16.0	76.8	1.0	198.6	4.8	81.6	4.1	519.7	$14.1 \times 10^3$
9 Nov 2010	1035	514.5	18.0	55.0	2.0	167.8	6.2	83.5	4.2	514.6	$8.6 \times 10^3$
12 Nov 2010	1030	525.4	18.5	56.3	1.1	164.9	5.1	71.8	4.2	404.2	$9.4 \times 10^3$
Class Ib											
29 Oct 2010	1030	546.2	13.4	64.8	2.6	128.2	6.0	69.5	4.2	473.7	$9.8 \times 10^3$
2 Nov 2010	1015	542.0	16.4	76.7	1.3	186.1	4.2	88.9	4.1	514.3	$8.7 \times 10^3$
8 Nov 2010	1030	527.8	19.4	60.5	2.2	115.5	4.6	77.9	4.2	509.2	$11.1 \times 10^3$
26 Nov 2010	1045	424.1	15.2	76.1	1.2	146.5	5.1	72.9	4.2	498.6	$7.4 \times 10^3$
Class II											
27 Oct 2010	1100	522.5	13.5	80.1	3.4	205.2	3.9	46.1	4.3	433.9	$8.9 \times 10^3$
30 Oct 2010	1045	546.0	14.1	56.6	2.8	137.8	8.0	78.4	4.1	578.5	$11.7 \times 10^3$
11 Nov 2010	1100	513.0	18.8	41.4	1.1	179.4	3.7	83.0	4.2	431.4	$8.7 \times 10^3$
28 Nov 2010	1100	389.3	17.2	75.6	1.9	113.7	2.7	61.8	4.1	305.6	$3.5 \times 10^3$

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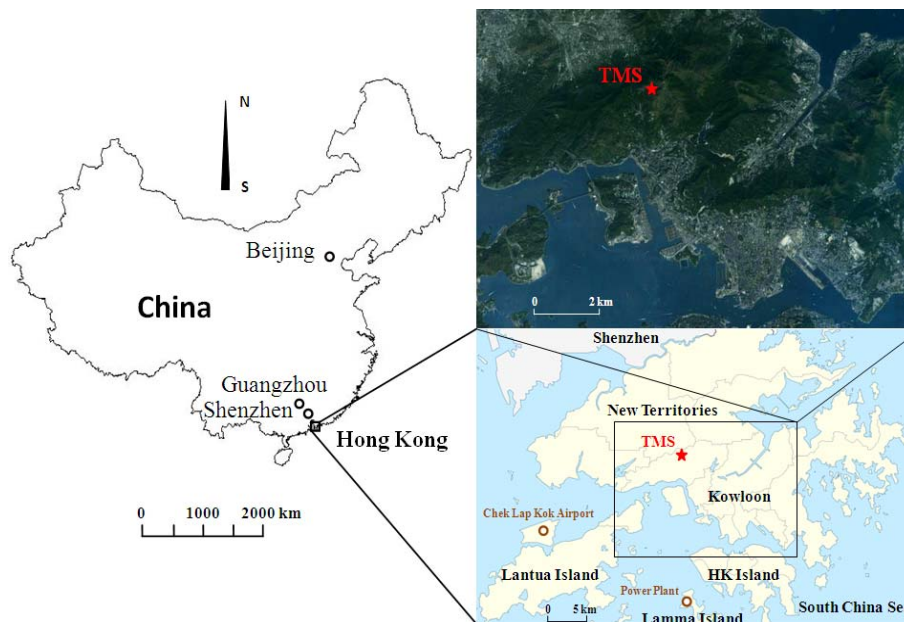
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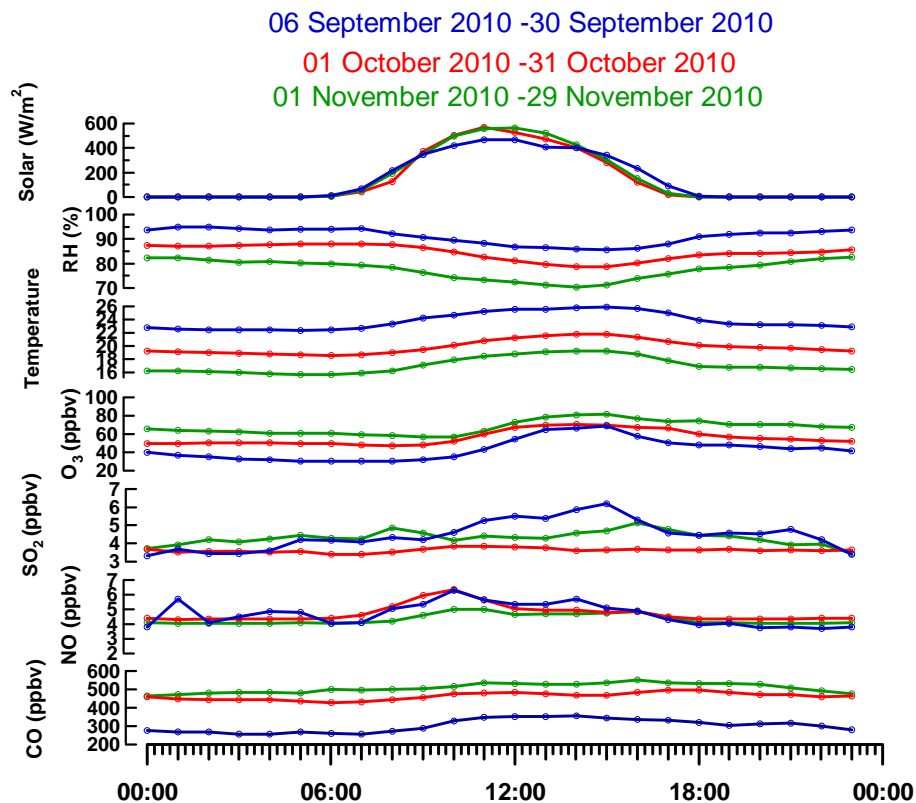
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**Fig. 1.** Locations of Hong Kong and the TMS sampling site.[Title Page](#)[Abstract](#)[Introduction](#)[Conclusions](#)[References](#)[Tables](#)[Figures](#)[⏪](#)[⏩](#)[◀](#)[▶](#)[Back](#)[Close](#)[Full Screen / Esc](#)[Printer-friendly Version](#)[Interactive Discussion](#)

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**Fig. 2.** Mean diurnal variations of trace gases ( $\text{SO}_2$ , NO, CO and  $\text{O}_3$ ) and meteorological conditions (UV, RH, T and wind) at the TMS site in September, October and November, 2010.

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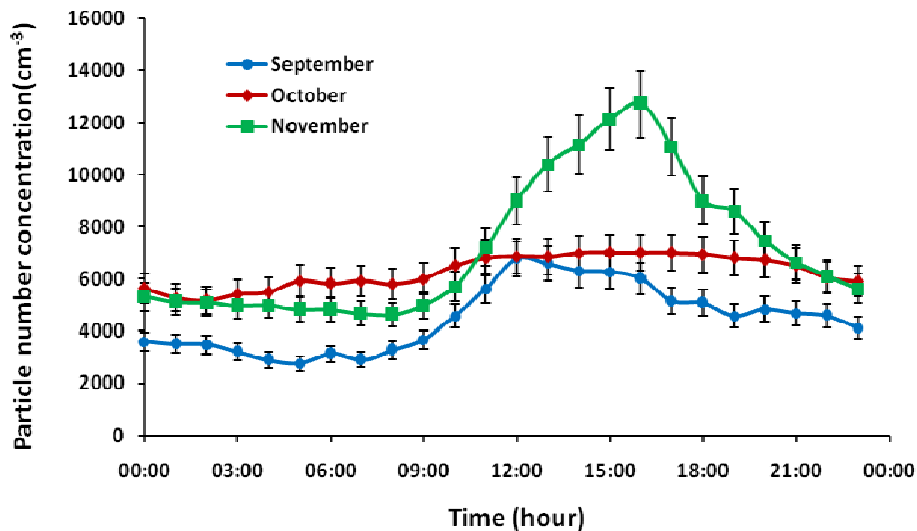
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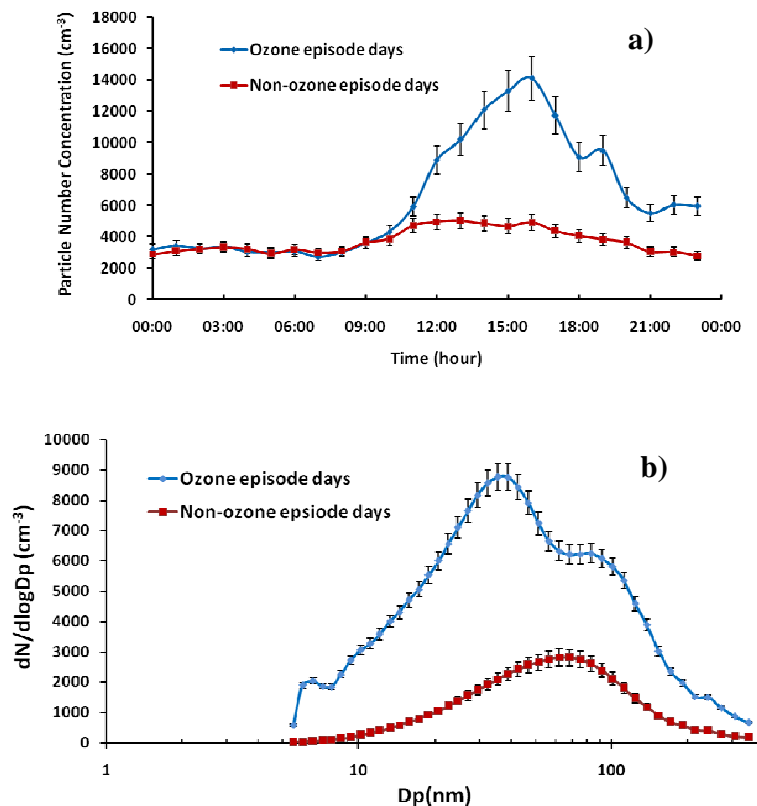
**Fig. 3.** Diurnal variations of PN concentration (5.5–350 nm) in September, October and November 2010 at TMS. The error bars show the standard error of the mean.

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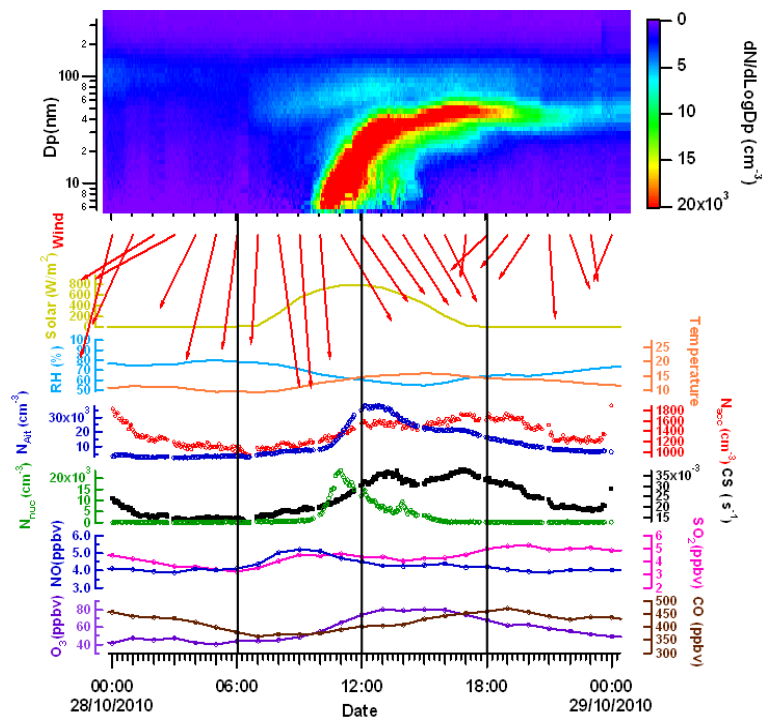
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**Fig. 4.** Particle number concentration and size distribution on O<sub>3</sub> episode days (O<sub>3</sub> > 100 ppbv) and non-O<sub>3</sub> episode days (O<sub>3</sub> < 50 ppbv) in October and November at TMS; **(a)** diurnal variations of particle number concentration; **(b)** mean particle size distributions. The error bars show the standard error of the mean.

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**Fig. 5.** Contour plots of the average diurnal variations of particle size distribution, the meteorological conditions and the concentrations of airborne pollutants on 28 October 2010 at TMS. From top to bottom, the parameters are: (i) particle size distribution; (ii) wind direction and speed (iii) solar radiation; (iv) temperature and relative humidity; (v) number concentrations of Aitken mode and accumulation mode particles; (vi) number concentration of nucleation mode particles and condensation sink, (vii) no and SO<sub>2</sub>; (viii) O<sub>3</sub> and CO.

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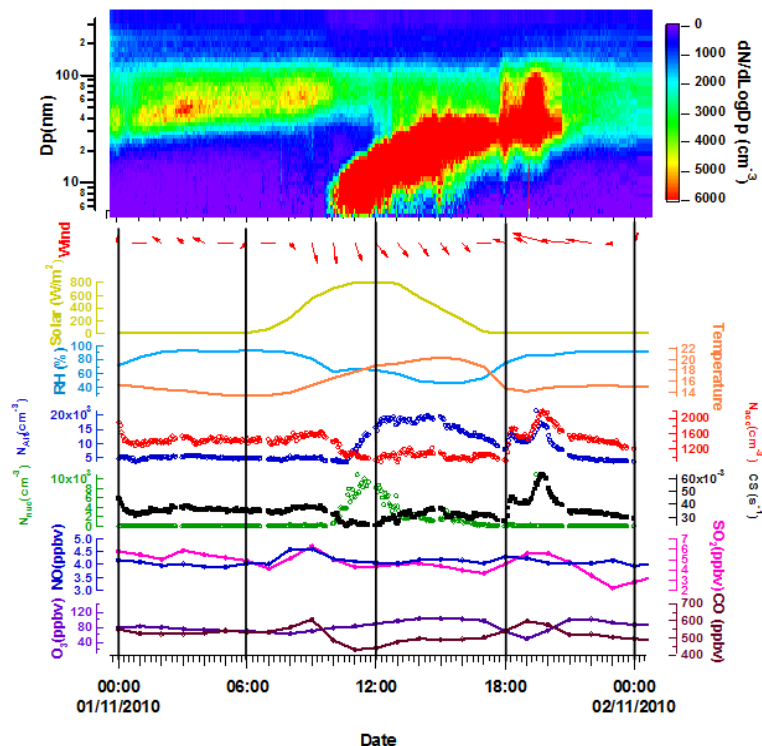
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**Fig. 6.** Contour plots of the average diurnal variations of particle size distribution, the meteorological conditions and the concentrations of airborne pollutants on 1 November 2010 at TMS. From top to bottom, the parameters are: (i) particle size distribution; (ii) wind direction and speed (iii) solar radiation; (iv) temperature and relative humidity; (v) number concentration of Aitken mode and accumulation mode particles; (vi) number concentration of nucleation mode particles and condensation sink, (vii) NO and SO<sub>2</sub>; (viii) O<sub>3</sub> and CO.

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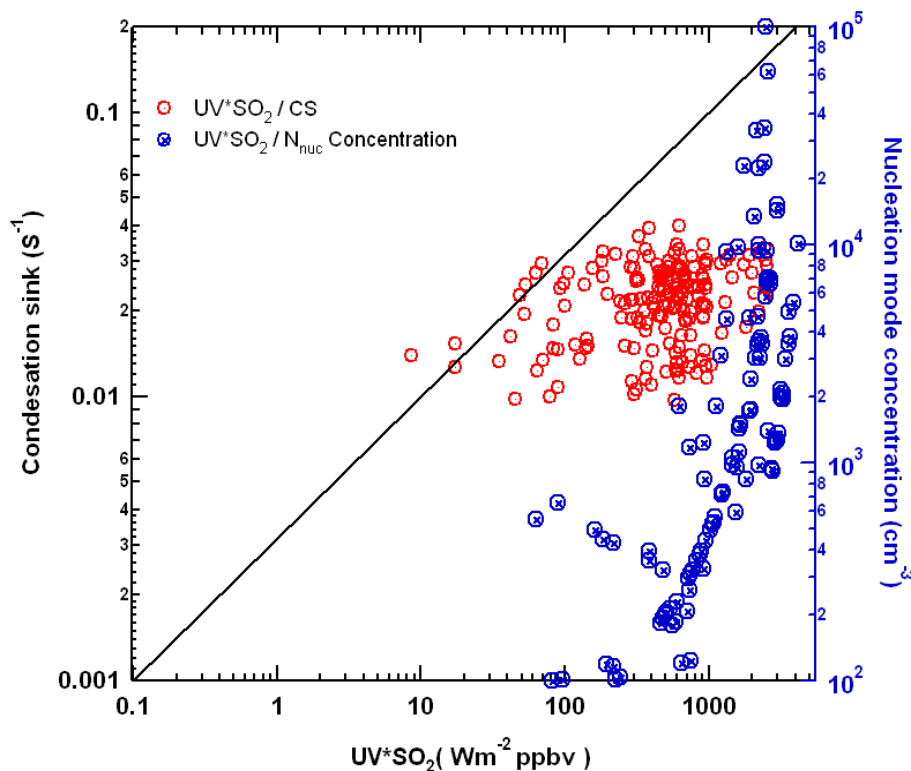
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**Fig. 7.** Scatter plot of  $UV \times SO_2$  versus condensation sink (red dots) and  $UV \times SO_2$  versus  $N_{nuc}$  concentration (blue dots) during the NPF events. Condensation sink (y-axis – left) and nucleation mode particle concentration (y-axis – right) is plotted against the product of ultraviolet light intensity and  $SO_2$  concentration (x-axis), respectively. Four-minute averaged values are plotted for the onset of all NPF events during this study.

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