

1 Microphysical simulations of new particle formation in the 2 upper troposphere and lower stratosphere

3
4 J. M. English¹, O. B. Toon¹, M. J. Mills², and F. Yu³

5 [1] {Laboratory for Atmospheric and Space Physics, Department of Atmospheric and Oceanic
6 Sciences, University of Colorado, Boulder, CO}

7 [2] {NCAR Earth System Laboratory, National Center for Atmospheric Research, Boulder, CO}

8 [3] {Atmospheric Sciences Research Center, State University of New York, Albany, NY}

9 Correspondence to: J. English (englishj@colorado.edu)

10 11 Abstract

12 Using a three-dimensional general circulation model with sulfur chemistry and sectional aerosol
13 microphysics (WACCM/CARMA), we studied aerosol formation and microphysics in the upper
14 troposphere and lower stratosphere (UTLS) as well as the middle and upper stratosphere based
15 on three nucleation schemes (two binary homogeneous schemes and an ion-mediated scheme
16 related to one of the binary schemes). Simulations suggest that ion-mediated nucleation rates in
17 the UTLS are 25% higher than its related binary scheme, but that the rates predicted by the two
18 binary schemes vary by two orders of magnitude. None of the nucleation schemes is superior at
19 matching the limited observations available at the smallest sizes. However, it is found that
20 coagulation, not nucleation, controls number concentration at sizes greater than approximately
21 10 nm. Therefore, based on this study, processes relevant to atmospheric chemistry and radiative
22 forcing in the UTLS are not sensitive to the choice of nucleation schemes. The dominance of
23 coagulation over other microphysical processes in the UTLS is consistent with other recent work
24 using microphysical models. Simulations using all three nucleation schemes compare reasonably
25 well to observations of size distributions, number concentration across latitude, and vertical
26 profiles of particle mixing ratio in the UTLS. Interestingly, we find that we need to include Van
27 der Waals forces in our coagulation scheme to match the UTLS aerosol concentrations. We
28 conclude that this model can reasonably represent sulfate microphysical processes in the UTLS,
29 and that the properties of particles at atmospherically relevant sizes appear to be inssensitive to
30 the details of the nucleation scheme. We also suggest that micrometeorites, which are not

Jay 7/12/11 5:14 PM

Deleted: tropical

Jay 7/12/11 5:14 PM

Deleted: rates

MacD63 LASP 7/14/11 10:24 AM

Deleted: atmospherically relevant

MacD63 LASP 7/14/11 10:24 AM

Deleted: in the UTLS (

MacD63 LASP 7/14/11 10:24 AM

Deleted:)

yfq 7/20/11 11:06 PM

Deleted: accurately

yfq 7/20/11 11:07 PM

Deleted: are not

31 included in this model, dominate the aerosol properties in the upper stratosphere above about 30
32 km.

34 1 Introduction

35 The tropical upper troposphere is known to be a net source region of new particles (e.g. Brock et
36 al., 1995, Clarke and Kapustin, 2002). These particles may cross the tropopause and accelerate
37 stratospheric ozone destruction via heterogeneous chemistry (Hofmann and Solomon, 1989),
38 impact climate by modifying cirrus cloud properties in the upper troposphere (Jensen et al.,
39 1996), and possibly descend to the marine boundary layer and act as cloud condensation nuclei
40 (CCN) there (Clark, 1993). The UTLS region does not have a standard spatial definition, but we
41 generally refer to the region between 50 and 500 hPa. Climate geoengineering schemes are
42 receiving increased attention in recent years especially those related to UTLS aerosols such as
43 stratospheric sulfur injection (e.g. Crutzen, 2006, Robock et al., 2008, Heckendorn et al., 2009)
44 and cirrus cloud modification (e.g. Rasch et al., 2008). However, mechanisms of UTLS new
45 particle formation (NPF) continue to be poorly understood.

46 Classical nucleation theory suggests binary homogeneous nucleation (BHN) of sulfuric acid and
47 water is favored in the UTLS due to cold temperatures and availability of supersaturated sulfuric
48 acid and water (Brock et al., 1995). Other nucleation processes are also possible but are generally
49 associated with nucleation at warmer temperatures or closer to surface sources, including ternary
50 nucleation mediated by ammonia (Coffman and Hegg, 1995) or by organic molecules (Zhang et
51 al., 2004). Ion-mediated nucleation (IMN) of sulfuric acid and water has received increased
52 attention in recent years as a possible link between solar activity and climate (Yu and Turco,
53 2001, Lovejoy et al., 2004). Ions produced by cosmic rays entering earth's atmosphere may
54 stabilize molecular clusters, increasing the formation rate and number of new particles. As these
55 ions are produced in much of the earth's atmosphere (Usoskin et. al., 2009), they can potentially
56 influence nucleation rates in any region. While numerous modeling and observational studies
57 have investigated IMN in the lower troposphere (e.g. Yu and Turco 2001, 2011, Lovejoy et. al.,
58 2004, Eisele et. al., 2006), study of the UTLS region is more limited. Kanawade and Tripatha
59 (2006) calculated IMN with a sectional aerosol model and found agreement with UTLS
60 observations, but did not compare with BHN simulations. Pierce and Adams (2009) and Snow-

Jay 7/12/11 5:14 PM

Deleted: descend into the marine boundary layer and act as cloud condensation nuclei (CCN) there (Clark, 1993),

Jay 7/12/11 5:14 PM

Deleted: and

Jay 7/12/11 5:14 PM

Deleted:

MacD63 LASP 7/14/11 10:30 AM

Deleted: in rare cases

61 | [Kropla et al. \(2011\)](#) calculated [changes in](#) IMN from solar cycle changes using a [sectional](#) model
62 | and found the IMN contribution to cloud condensation nuclei to be two orders of magnitude too
63 | small to account for observed changes in cloud properties. Yu and Luo (2009) calculated IMN
64 | with a sectional microphysical aerosol model and found reasonable agreement in the troposphere,
65 | but did not compare to BHN, and did not compare to observations in the UTLS. [Yu et al. \(2010\)](#)
66 | [compared nucleation rates and number concentration from IMN and two different BHN schemes](#)
67 | [in the troposphere to aircraft observations, but did not study](#)[look into the aerosol evolution \(size,](#)
68 | [mass, effective radius\) and did not study stratospheric properties.](#) Kazil et al. (2010) found that
69 | simulations agree best with observations in the lower and mid-troposphere when IMN and BHN
70 | are included across the entire model domain and organic cluster formation is [included but](#) limited
71 | to the [continental](#) boundary layer, but did not compare to observations in the stratosphere. We
72 | present the first simulations using a sectional aerosol microphysical model that includes two
73 | different binary homogeneous nucleation schemes and an ion-induced nucleation scheme. We
74 | compare our simulations with UTLS observations of size distribution (Lee et al., 2003, Deshler
75 | et al. 2003), number concentration (Borrmann et al. 2010, Brock et al., 1995, Heintzenberg et al.,
76 | 2003), and Stratospheric Aerosol and Gas Experiment (SAGE) II aerosol extinctions and
77 | effective radii (Chu et al., 1989).

78

79 | **2 Model description**

80 | We have constructed a three-dimensional general circulation model with sulfur chemistry and
81 | sectional aerosol microphysics. We use the Whole Atmosphere Community Climate Model
82 | (WACCM) (Garcia et al., 2007) coupled with the Community Aerosol and Radiation Model for
83 | Atmospheres (CARMA) (Toon et al., 1988). Fig. 1 illustrates the processes treated in the
84 | coupled model. Model coupling is done by implementing a single column version of CARMA
85 | as a WACCM physics package. Results of this coupling for meteoric dust (Bardeen et. al., 2008),
86 | noctilucent clouds (Bardeen et. al., 2010) and black carbon (Mills et al., 2008; Ross et al., 2010)
87 | have been published previously. Other versions of the code have been used to simulate sea salt
88 | and dust (Fan and Toon, 2010; Su and Toon, 2011). The WACCM state is passed to CARMA
89 | one column at a time. CARMA calculates changes to the constituents, and the tendencies are sent
90 | back to WACCM where they are used to adjust the model's state. Each CARMA aerosol size bin

Jay 7/12/11 5:14 PM

Deleted: modal

yfq 7/20/11 11:12 PM

Deleted: study

91 is added as a unique WACCM constituent (Bardeen et. al., 2008). Although CARMA is capable
92 of interacting radiatively and chemically with WACCM, for these studies the interactions were
93 mainly disabled. This version of WACCM utilizes SAGE II sulfate surface area densities for
94 radiative transfer and ozone heterogeneous chemistry calculations.

95

96 **2.1 WACCM with sulfur chemistry**

97 We use WACCM3 version 3.1.9 tag 9 with 30-minute time steps at 4° latitude by 5° longitude
98 horizontal resolution with 66 vertical levels based on hybrid-sigma coordinates, providing 15
99 vertical levels in the UTLS between 50 and 500 hPa. We use the WACCM mass-conserving
100 finite volume dynamical core based on a flux-form semi-Lagrangian transport scheme (Lin and
101 Rood 1996, 1997). The vertical diffusion algorithm in WACCM handles eddy and molecular
102 diffusion for gases. A 63-species chemistry module is implemented. We utilize WACCM's
103 standard 56-species chemical package which includes Ox, NOx, HOx, ClOx, and BrOx chemical
104 families along with CH₄ and its products and 7 ions (Kinnison et al., 2006), and add 7 sulfur-
105 bearing gases: S, SO, SO₂, SO₃, HOSO₂, H₂SO₄, and OCS. Their reaction rates and photo-
106 dissociation rates are given in Table 1.

107 The model includes emissions of carbonyl sulfide (OCS) and sulfur dioxide (SO₂), two primary
108 sulfur emissions of importance to the UTLS region. OCS is specified with a constant surface
109 concentration of 510 pptv. SO₂ is specified from a two-dimensional monthly mean surface
110 emissions dataset (Lamarque et al., 2010, Smith et al., 2010). The model does not include
111 emissions of dimethylsulfide or volcanic SO₂, which are large natural sources of sulfur to the
112 UTLS, but minor contributions to total UTLS aerosol. We compare with data from a period with
113 relatively little volcanic activity. Wet deposition for all constituents (including the aerosol bins
114 from CARMA) is calculated using WACCM's existing techniques (Barth et al., 2000). All of
115 the aerosol bins are assumed to have a constant 0.3 solubility parameter. WACCM treats dry
116 deposition of gases (Barth et al., 2000), while dry deposition of aerosols is not treated in this
117 model.

118

119 **2.2 CARMA**

Jay 7/12/11 5:14 PM

Deleted: 2000.

120 We use CARMA 2.3 configured for one-dimensional columns using the same vertical grid as
121 WACCM. Split-time stepping is enabled for nucleation and growth routines when sulfuric acid
122 is supersaturated. [Nucleation and growth are treated simultaneously in the model](#). If sulfuric acid
123 gas concentrations become unstable (negative), the CARMA time step is retried with double the
124 number of substeps. On rare occurrences the model reached over 1000 substeps, but typically
125 the model ran with only 1 or 2 substeps. [We have found that increasing the number of timesteps
126 past the point at which negative gas amounts were found did not significantly change results.](#)
127 [Additionally, we limited nucleation to 40% of the sulfuric acid available.](#) We specify 38 sulfuric
128 acid mass bins ranging from 0.2 nm to 1 micron radius, with mass doubling between bins. The
129 particles are assumed to have spherical shape. Sulfate surface tension is calculated using the
130 constants from Sabinina and Terpigov (1935). Since the bins only carry sulfate, the equivalent
131 sulfate aerosol size (including sulfuric acid and water) is determined by the technique of
132 Tabazadeh et al., (1997), which calculates weight percent sulfuric acid as a function of
133 temperature and water activity. Weight percent sulfuric acid is assumed to be independent of
134 particle size. We did not include any other types of aerosols. Although other aerosols, such as
135 organics, are known to compose a significant fraction of the [sulfate](#) aerosol mass in the UTLS
136 ([Froyd et al., 2009](#), [Murphy et al., 2007](#)), sulfates are believed to be the primary source of new
137 particles in this region, and [the primary aerosol in the lower and middle stratosphere \(Murphy et
138 al., 2007\)](#).

Jay 7/12/11 5:14 PM

Deleted: ,

Jay 7/12/11 5:14 PM

Deleted: we wanted to clearly isolate the impact of sulfates.

139 Fall velocities are calculated by assuming a Stokes-Cunningham equation with Knudsen number
140 corrections from Fuchs (1964), using the equivalent aerosol size (sulfuric acid plus water). Since
141 WACCM handles advection by winds as well as eddy diffusion, no additional eddy diffusion of
142 aerosol particles is added by CARMA. However, CARMA treats Brownian diffusion of aerosols,
143 [which becomes important above 100 km as the heterosphere is approached](#), and which is not well
144 treated by algorithms in WACCM. CARMA calculates the effect of coagulation of particles of
145 equivalent aerosol size using the numerical approach described in Toon et al. (1988).
146 Coagulation coefficients are calculated to include Brownian, convective and gravitational effects.
147 A sticking coefficient of 1 is used, which assumes that all particles stick together upon colliding.
148 A correction for the impact of inter-particle Van der Waals forces on coagulation is included
149 (Chan and Mozurkewich, 2001). Sulfate aerosol growth and evaporation is calculated via
150 sulfuric acid equilibrium vapor pressure over binary solution using the method of Ayers et al.

Jay 7/12/11 5:14 PM

Deleted: important above 100 km, and

151 (1980) with a temperature correction by Kulmala (1990) and thermodynamic constants from
152 Giauque (1959). Numerical calculations for fall velocity and growth/evaporation are solved by
153 CARMA using the piecewise parabolic method of Colella and Woodward (1984).

154

155 **2.3 Description of nucleation schemes**

156 Three nucleation schemes are implemented in CARMA: two binary homogeneous nucleation
157 (BHN) schemes and one ion-mediated nucleation (IMN) scheme.

158

159 **2.3.1 Zhao BHN scheme**

160 The “Zhao BHN” scheme predicts the binary homogeneous nucleation rate of sulfuric acid and
161 water using classical nucleation theory (e.g. Flood 1934, Reiss 1950, Hamill et al., 1977) with
162 modifications for calculating the saddle point in Gibbs free energy by Zhao and Turco (1995).
163 Instead of searching for the Gibbs free energy saddle point in two-dimensional space, the
164 coordinate system is transformed to a function of cluster volume and sulfuric acid weight
165 fraction of a solution droplet. This provides for a unique solution in a 1-dimensional parameter
166 space. Water equilibrium vapor pressure over a binary solution is calculated using the technique
167 of Lin and Tabazadeh (2001). Sulfuric acid equilibrium vapor pressure over binary solution is
168 calculated in the same manner as aerosol growth/evaporation: using the method of Ayers et al.,
169 (1980) with a temperature correction by Kulmala (1990) and thermodynamic constants from
170 Giauque (1959). Many similar schemes for BHN have been published (Kulmala 1990) with
171 varying choices for thermodynamic parameters. Since these schemes are analytic they generally
172 are well behaved over the entire range of parameter space covered in WACCM, although their
173 predicted nucleation rates are known to vary by an order of magnitude or more.

174

175 **2.3.2 Yu BHN scheme**

176 The “Yu BHN” scheme predicts the binary homogeneous nucleation rate of sulfuric acid and
177 water by assuming the “quasi-unary” nucleation of sulfuric acid in equilibrium with water vapor
178 (Yu, 2008), also known as kinetic nucleation theory. Since the growth of clusters is largely
179 determined by the availability of sulfuric acid, the binary nucleation can be reduced to unary
180 nucleation of sulfuric acid except that the clusters containing different numbers of sulfuric acid

181 molecules also contain a semi-fixed number of water molecules at a given temperature and
182 relative humidity. The kinetically self-consistent “Yu BHN” model is constrained by the
183 measured bonding energetics of H₂SO₄ monomers with hydrated sulfuric acid dimers and trimers
184 (Hanson and Lovejoy, 2006; Kazil et al., 2007) and gives BHN nucleation rates in good
185 agreement with available experimental data. While the laboratory data used to constrain the “Yu
186 BHN” model substantially reduces the model uncertainty, they were measured under
187 tropospheric conditions and can’t be extrapolated to dry stratospheric conditions. As a result, the
188 application of the present “Yu BHN” scheme should be limited to the troposphere. The “Yu
189 BHN” scheme is available as a set of two lookup tables, a low temperature table for temperatures
190 between 180 and 250 K, and a high temperature table for temperatures between 250 and 300 K.
191 The low temperature table is five-dimensional, with inputs for sulfuric acid concentration,
192 relative humidity, temperature, pre-existing aerosol surface area, and ion-pairs. (Ion pairs are set
193 to zero for BHN simulations). The high temperature table is three-dimensional, with inputs for
194 sulfuric acid concentration, relative humidity, and temperature. If input values are outside of the
195 limits of the table, the values are adjusted to the minimum values. As mentioned earlier, this
196 scheme was developed for the troposphere and the tables do not cover the full range of relative
197 humidity, surface area and temperature found [above the middle stratosphere](#). It was discovered
198 that the tables predict unrealistic nucleation rates in the middle and upper stratosphere, due to
199 relative humidity being adjusted from calculated values as low as 10⁻⁸ to the table minimum
200 values (0.1% for the low temperature table and 1% for the high temperature table). The problem
201 of extending the range of the tables was largely resolved by setting the nucleation rate to zero if
202 RH was less than the table minimums. However, there may remain some unrealistic nucleation
203 rates in the [middle](#) stratosphere and above due to the boundary conditions of the lookup tables.
204 The tables should be used with caution [in these regions](#).

205

206 2.3.3 Yu IMN scheme

207 The “Yu IMN” scheme predicts ion-mediated nucleation rates of sulfuric acid and water. Ions of
208 positive and negative charge stabilize the molecular cluster due to molecular attractions of
209 opposite polarity. This scheme uses the same low temperature lookup table as the Yu BMN
210 scheme, while a different high temperature table is used, as described in Yu (2010). For the Yu
211 IMN scheme, a globally constant input value of 10 ion-pairs per cm³ is prescribed. Although the

yfq 7/20/11 11:24 PM
Deleted: in the atmosphere
LASP 7/22/11 2:06 PM
Deleted: .
LASP 7/22/11 3:15 PM
Deleted: tropopause

Jay 7/12/11 5:14 PM
Deleted: above the tropopause

212 ionization rate varies spatially and temporally, it is relatively constant in the UTLS and is
213 estimated to be between 5 and 20 ion-pairs per cm^3 (Usoskin et. al., 2009). This table was
214 developed for tropospheric conditions as well and thus has the same limitations as the Yu BHN
215 scheme. Thus, nucleation was set to zero if relative humidity was below the table minimum
216 values (0.1% for the low temperature table or 0.3% for the high temperature table).

217

218 | 3 Model Evaluation

219 We have found that for most particle properties all three nucleation schemes produce nearly
220 identical results. Therefore, below we first compare simulations using one nucleation scheme
221 (Zhao BHN) with observations. Later we highlight where the schemes differ. Initial values for
222 atmospheric state, gas properties and aerosol properties are read in from a baseline run with a 5-
223 year spinup time. A three-year simulation was conducted, with the third year analyzed. Analysis
224 of sulfate mass and number concentration indicate that the model achieved steady state in less
225 than one year when using the common spin-up file.

226

227 | 3.1 Simulations of sulfur gas precursors

228 Calculated OCS is uniformly mixed in the troposphere (Fig. 2a), due to its long photochemical
229 lifetime there. In the stratosphere, its mixing ratio decreases with altitude due to photolytic
230 conversion of OCS to SO_2 . Fig. 2b shows OCS correlated with N_2O , a long-lived tracer with
231 well-understood chemistry, from our calculations as well as from balloon-borne observations
232 (Geoff Toon, private communication). The close agreement in the slope of this correlation
233 indicates that the model correctly treats photochemical losses of OCS.

234 Calculated surface SO_2 concentrations vary by five orders of magnitude across the earth's
235 surface (Fig. 3a), with highest concentrations in the industrial mid-latitudes, particularly in
236 eastern Asia, eastern United States, and Europe. SO_2 mixing ratio decreases with altitude in the
237 troposphere (Fig. 3b), with the highest concentrations near 30°N , correlating with the peak
238 latitude of surface emissions. SO_2 mixing ratios decrease rapidly just above the tropopause due
239 to slow vertical transport relative to chemical loss mainly by reaction with OH. A peak occurs in

Jay 7/12/11 5:14 PM

Deleted: Validation

MacD63 LASP 7/14/11 10:42 AM

Deleted: W

240 the tropics above 25 km where OCS is converted into SO₂ and SO₂ increases again in the upper
241 stratospheric due to photolytic conversion of H₂SO₄ back to SO₂ (Mills et al., 2005).

Jay 7/12/11 5:14 PM

Deleted: .

242 Fig. 4a shows latitudinal variation in SO₂ mixing ratios in the Pacific upper troposphere between
243 8 and 12 km from model calculations and PEM-TA, PEM-TB, and ACE-2 aircraft observations
244 (Thornton et al., 1999). Our calculations are slightly lower than the observations, but generally
245 within or close to the observed variability. Fig. 4b shows vertical profiles of SO₂ mixing ratios
246 compared to 6 ACE-2 aircraft observations (Curtius et al., 2001). Our calculations are generally
247 within the measurement variability expressed by the error bars although the model overpredicts
248 the SO₂ observations in the upper troposphere above 300 hPa. The observations are limited to a
249 narrow latitude range of 28-32°N. As shown by the calculations from various latitudes, SO₂
250 concentrations are a strong function of latitude (and longitude) in this range. Compared to PEM-
251 TB aircraft observations in the tropics (Wang et al., 2001) (Fig. 4c), the model overpredicts near
252 the surface and underpredicts above 700 hPa. It is possible that the lack of dimethylsulfide
253 emissions in the model contributes to lower SO₂ concentrations in the mid-troposphere at non-
254 industrial latitudes. Overall, considering the limited number of observations and the high
255 variability of available observations, we conclude that the model SO₂ emissions and chemistry
256 are generally well-behaved. It would be extremely valuable for UTLS studies to obtain SO₂
257 observations above 200 hPa.

258 Sources of H₂SO₄ vapor in the model are SO₂ and OCS oxidation and aerosol evaporation, while
259 sinks include nucleation, condensation, and photolysis at high altitudes. As shown in Fig. 5a,
260 calculated H₂SO₄ vapor mixing ratios increase from 25 to 35 km due to sulfate aerosol
261 evaporation. It is so warm and dry above 35 km that the sulfuric acid vapor pressure exceeds the
262 total (gas + particle) mixing ratio of sulfuric acid; hence, the particles completely evaporate.
263 H₂SO₄ also has a local maximum in the Northern Hemisphere sub-tropical upper troposphere due
264 to availability of SO₂ and OH for chemical conversion. As Fig. 5b shows, calculated H₂SO₄
265 mixing ratios are generally within the standard deviation of PEM-TA aircraft observations
266 (Lucas and Prinn, 2003). H₂SO₄ averages about 0.1 pptv throughout most of the tropical
267 troposphere and lower stratosphere. As Fig. 5c shows, calculated H₂SO₄ concentrations in the
268 stratosphere at 43°N closely match balloon-borne observations (Arnold et al., 1981, Reiner and
269 Arnold, 1997, Schlager and Arnold, 1987, Viggiano and Arnold, 1981), with a peak near 35 km.

Michael Mills 7/21/11 4:02 PM

Deleted: so

MacD63 LASP 7/14/11 10:45 AM

Deleted: (H₂SO₄ is very sub-saturated in this relatively warm and dry region).

Jay 7/12/11 5:14 PM

Deleted: mixing ratios

271 **3.2 Sulfate aerosol properties**

272 Calculated sulfate mass mixing ratios versus N_2O are compared to a compilation of NASA
273 aircraft observations (Wilson et al., 2008) in Fig. 6. Calculated sulfate mass mixing ratio
274 generally is within the variability in the mid-latitude UTLS (220 to 300 ppbv N_2O). Calculated
275 sulfate mass mixing ratio is about 50% too high at low N_2O values (polar mid-stratosphere). It is
276 possible that simulated sedimentation rates are too slow in the midlatitude stratosphere. Recent
277 calculations with our WACCM/CARMA model including both sulfates and meteoric dust
278 improves this correlation. However, as we will discuss below our model underpredicts aerosol
279 volume (also measured by Wilson et al., 2008) versus CO. It is possible that these conflicting
280 differences in aerosol volume and sulfate mass versus tracer abundance are related to transport or
281 tracer chemistry issues within WACCM. It is also possible there are errors in the observations,
282 or errors with comparing aircraft flights on particular days with averaged model data.

283 Model calculations are compared to SAGE II extinction measurements at two wavelengths in
284 Fig. 7. Here, the calculation is within about 50% of the observations at both wavelengths for all
285 three latitude regions from the tropopause through the mid-stratosphere. Below the tropopause,
286 SAGE II has higher extinction than the calculations, with high variability. It is likely clouds are
287 interfering with measurements below the tropopause, as has been noted in prior analyses of
288 SAGE II extinction measurements (Wang et al., 1995, 1996). In the upper stratosphere,
289 WACCM extinctions decline sharply with higher altitude, while SAGE II extinctions level off at
290 about 10^{-6} km^{-1} at 1024 nm near 35 km. Hervig et al. (2009) have observed from AIM solar
291 occultation measurements that micrometeorites, sedimenting down from the mesopause, have an
292 extinction near 10^{-6} at a 1037 nm near 35 km. Recent calculations with our WACCM/CARMA
293 model including both sulfates and meteoric dust improves this correlation between the model and
294 SAGE II data above 35 km, reinforcing this suggestion. Hunten et al. (1980) originally suggested
295 the presence of these particles, and they have long been sought in rocket measurements with little
296 quantitative success. It is [interesting](#) that SAGE II has seen them throughout its observational
297 record, but their presence was not recognized.

298 While WACCM extinction is within 50% of SAGE II in the mid-latitudes, WACCM is higher
299 than SAGE II at 1024 nm in the tropics, suggesting that the WACCM particles are slightly too

Jay 7/12/11 5:14 PM

Deleted: ironic

300 large in the tropics. Indeed, calculated effective particle radius (Fig. 8) is about 25% higher than
301 SAGE II in the UTLS. Model calculations of size distributions in the UTLS at 41°N are
302 compared to balloon-borne observations (Deshler et al., 2003) in Fig. 9. Vertical profiles of
303 calculated particle concentration are within 50% of observations at the smallest size (>0.01
304 microns), while at larger sizes the model underpredicts number concentration below the
305 tropopause and overpredicts number concentration in the mid-stratosphere.

306 Vertical profiles of calculated sulfate number concentration in the nm size range are compared to
307 aircraft observations (Borrmann et al., 2010, Brock et al., 1995) in Fig. 10. Here, two Zhao BHN
308 calculations are compared: a run with Brownian coagulation and no inter-particle forces
309 (noVW), and our base case in which the Brownian coagulation kernels are adjusted to include
310 the effect of a Van der Waals forces between the particles. Van der Waals forces have been
311 observed to be important for sulfuric acid aerosols in several laboratory studies (Schmid-Ott and
312 Burtscher, 1982, Alam, 1987, Huang et al., 1990) and we include a size-dependent expression for
313 the Hamaker constant based on laboratory measurements (Chan and Mozurkewich, 2001). In
314 both the tropics (Fig. 10a) and extratropics (Fig. 10b), calculated and observed particle mixing
315 ratios increase in the troposphere, peak near the tropopause where the highest nucleation rates
316 are observed, and decrease in the stratosphere, as expected. Including the effect of Van der
317 Waals forces on coagulation results in calculations that are within the error bars of the
318 observations. Prior to including the effect of Van der Waals forces on coagulation, the model had
319 always overpredicted particle mixing ratio in the mid-stratosphere despite modifications to
320 nucleation schemes, sulfur emissions, fall velocity schemes, and growth equations. The impact
321 of including Van der Waals forces highlights how important coagulation rates are to
322 stratospheric aerosol properties.

323 Calculated aerosol number, area, and volume versus carbon monoxide (CO) are compared to an
324 average of 13 flights in the tropical UTLS between 2004 and 2006 in Fig. 11. Again, the Zhao
325 BHN calculations with and without the Van der Waals coagulation correction are compared.
326 Calculated number concentrations (Fig. 11b) for both simulations are within the error bars at
327 lower CO, but increase to up to an order of magnitude too high above 55 ppbv CO. In the
328 model, this CO region is present near 200 hPa and 20°N, where the model predicts peak
329 nucleation. It is possible that the model and data do not have corresponding geographical areas
330 with the same CO values. Model output were too limited in the range of longitudes covered by

331 aircraft observations, so all longitudes were included. The Zhao no VW scheme predicts higher
332 aerosol number where peak nucleation is observed, but lower number outside this region.
333 Calculated aerosol area (Fig. 11c) and volume (Fig. 11d) are about half the observations. It is
334 odd that the model underestimates aerosol area and volume in the Northern Hemisphere between
335 3°S and 23°N, yet overestimates aerosol mass in the Northern Hemisphere between 60 and 90°N
336 (Fig. 6c). The comparisons are made with coordinates of CO and N₂O, respectively, rather than
337 geographic location, so this discrepancy may be due to differences between modeled and
338 observed geographic locations or errors in the observations. [Additionally, the Zhao no VW
339 predicts higher nucleation at higher CO but lower nucleation at lower CO, as shown in Fig. 11b.
340 The Zhao no VW area plot \(Fig. 11b\) is higher than with the VW correction due to slower
341 growth rates.](#)

342

343 **4 Simulations with different nucleation schemes**

344 We analyze model output for three simulations, one for each nucleation scheme: Zhao BHN, Yu
345 BHN, and Yu IMN. Simulations are each 3 years in length (with initialized values from a
346 shared 5-year spin-up simulation). All three schemes were computationally similar in efficiency,
347 requiring approximately 11 hours to complete 3-years of simulation time on 96 dual-core
348 processors on the NASA Pleiades supercomputer. The microphysical model reduced computing
349 speed by about a factor of 2. Unless otherwise indicated, comparisons are done using the
350 average of the 3rd year of the simulations.

351

352 **4.1 Differences in Nucleation Rates**

353 Contour plots of calculated nucleation rates are provided in Fig. 12, while peak and average
354 nucleation rates and critical radii sizes in the UTLS are provided in Table 2. All three schemes
355 predict similar patterns of nucleation – the highest rates are predicted in the tropical upper
356 troposphere, with lower nucleation rates predicted in other parts of the troposphere and polar
357 stratosphere, but their magnitudes differ significantly. Near the surface, the Yu IMN scheme
358 predicts several orders of magnitude higher nucleation rates than the Yu BHN, suggesting that
359 ion nucleation from cosmic rays can have a large influence on new particle formation in this
360 region, in agreement with many other studies (Yu and Turco, 2001, Lovejoy et. al., 2004, Eisele
361 et. al., 2006, Kazil et al., 2010, [Snow-Kropla et al., 2011](#)). In the UTLS, where conditions are

362 favorable for BHN, the Yu IMN scheme predicts only 25% higher nucleation rates than the Yu
363 BHN scheme, consistent with some studies (e.g. Kazil et al., 2010) but different from others (e.g.,
364 Yu et al., 2010). Since BHN is favored under highly supersaturated H₂SO₄ and H₂O
365 environments, it is likely that the differences between studies are due to differences in UTLS
366 temperatures and availability of H₂SO₄ and H₂O. Regardless, the present results do suggest that
367 ions produced from cosmic rays may impact nucleation rates in the UTLS to a small degree.
368 Note also that the Zhao BHN scheme (classical nucleation theory) predicts two orders of
369 magnitude higher rates than the Yu BHN scheme (kinetic nucleation theory), suggesting that the
370 uncertainty associated with BHN computations is much larger than the effects of ions on
371 nucleation in the UTLS. The large differences between BHN schemes have been documented
372 | previously ([Korhonen et al., 2003](#), Yu et al., 2010), and can be partially explained by differences
373 | in predicted critical radii - the Yu schemes predict critical radii that are 60% larger than the Zhao
374 | BHN scheme. The larger sizes of new particles predicted by the Yu schemes partially offsets
375 | lower nucleation rates, resulting in a less substantial difference in sulfate mass and number
376 | concentration at larger sizes (e.g. Zhao BHN predicts a 100 times higher particle number creation
377 | rate from nucleation but only 17 times higher particle mass creation rate). Finally, note that the
378 | Yu BHN and Yu IMN lookup tables, designed for tropospheric conditions, were originally found
379 | to predict unrealistic nucleation rates in the middle and upper stratosphere due to exceptionally
380 | low relative humidities (<0.1%) being outside the table bounds. While setting nucleation to zero
381 | if relative humidity was less than the table minimum resolved much of this issue, it is possible
382 | that this approach may predict too little nucleation in certain regions. [However, an analysis of](#)
383 | [the input parameters has found that the tables behave well in the UTLS region and below.](#)

384

385 **4.2 Effects of nucleation rates on aerosol size distributions**

386 Calculated size distributions are compared with data from 56 aircraft flights from a range of
387 NASA field programs summarized by Lee et al. (2003). Size distributions are calculated for
388 three regions: Tropical troposphere (7-17 km), mid-high latitude UTLS (7-13 km), and high-
389 latitude stratosphere (17-21 km), and the data are separated into series with or without recent
390 new particle formation (NPF). Recent NPF was defined as meeting two conditions: i) number
391 concentrations with diameter 4-6 nm exceeds that of number concentration with diameter 6-9
392 nm, and ii) number concentrations with diameter 4-9 nm exceed 1 cm⁻³. Simulation size bin

393 ranges are selected based on the closest bins available to the size specified in Lee et al. (2003).
394 Calculated 1-day averages of the third year are checked for NPF conditions and segregated into
395 two sets of data (with and without recent NPF). Simulation “data” points include values for 360
396 days in the third simulation year. The model outputs daily averages, so these criteria will not
397 provide instantaneous indicators of recent NPF. Additionally, since the model output is across
398 the entire year, while the aircraft data are obtained on specific days, differences may be due to
399 temporal variability. A summary of the number of simulation data points considered NPF days
400 and no NPF days is provided in Table 3. All three schemes reported all days were NPF days in
401 the tropical troposphere, most days were NPF days in the high-latitude UTLS, and very few NPF
402 days occurred in the high-latitude stratosphere. Lee et al. reported 16% of size distributions to be
403 considered NPF events.

404
405 Size distributions for each of the three regions are provided in Fig. 13. In the tropical
406 troposphere (Fig. 13a), high numbers of particles are observed and predicted, as expected due to
407 this region being conducive to NPF. Relative differences in number concentrations between the
408 three schemes at the smallest sizes are explained by differences in nucleation rates. All three
409 schemes predict approximately two times too many of the smallest particles, and are missing the
410 observed size mode at 30 nm. It is possible that this discrepancy is due to the lack of other
411 aerosol types in the model. Advection of aged aerosol from other regions could contribute to
412 growth rather than nucleation, reducing the number concentration at the smallest sizes and
413 possibly creating a mode near 30 nm. A similar trend is observed in the mid-high latitude UTLS
414 region where NPF was observed (Fig. 13b); all three of the simulations predict two times too
415 many particles at the smallest sizes. In the mid-high latitude UTLS where NPF was not observed
416 (Fig. 13d), only the Yu BHN predicted days with no NPF. This simulation replicates both the 10
417 nm and 100 nm modes that are observed, albeit with a larger 10 nm mode and a smaller 100 nm
418 mode. In the high-latitude stratosphere (Fig. 13c), all three schemes reproduce the observed 100
419 nm mode, with particle number within a factor of two of that observed. All three schemes predict
420 a broader mode than observed, with the Yu BHN scheme better reproducing the mode at the
421 small end. However, only 14, 2, and four days, respectively for the Zhao BHN, Yu BHN, and
422 Yu IMN schemes met the criteria for stratospheric NPF. When plotting all simulation output
423 regardless of whether the grid cells met the criteria for recent NPF (Fig 14), all three simulations

Jay 7/12/11 5:14 PM
Deleted: .
MacD63 LASP 7/14/11 10:51 AM
Deleted: , in agreement with commonly known understanding of NPF in these regions.

424 predict a similar size mode in the stratosphere. In most cases, the three nucleation schemes
425 produce simulations that differ at the smallest sizes due to differing nucleation rates, but become
426 nearly indistinguishable from one another at sizes larger than 10 nm. On the other hand, the
427 Zhao no VW curve has higher number concentrations than any of the nucleation schemes. This
428 reinforces the conclusion that coagulation, not nucleation is the dominant process determining
429 aerosol number at atmospherically relevant sizes.

LASP 7/14/11 1:38 PM

Deleted: , suggesting that coagulation, not nucleation is the dominant process determining aerosol number at atmospherically relevant sizes

LASP 7/14/11 1:38 PM

Deleted:

430

431 4.3 Effects of nucleation rate on aerosol number concentrations

432 We illustrate vertical profiles of calculated sulfate number concentration (>8 nm) compared to
433 aircraft observations (Borrmann et al., 2010, Brock et al., 1995) in Fig. 10. All three simulations
434 with varying nucleation rate correlate very well with observations in both the tropics (Fig. 10a)
435 and extratropics (Fig. 10b). Although nucleation rates differ by up to two orders of magnitude,
436 there is very little difference in number concentration of particles >8 nm. Likewise, comparisons
437 of aerosol number (>4 nm) versus CO suggest differences due to nucleation schemes (Fig. 11b),
438 but aerosol area (Fig. 11c) and volume (Fig. 11d) are virtually unaffected by nucleation scheme.
439 Again, this suggests that the choice of nucleation scheme is nearly irrelevant compared to the
440 impacts of coagulation at atmospherically relevant sizes.

441 Similar trends are seen when comparing calculated zonal-averaged number concentrations in the
442 upper troposphere to aircraft observations from the CARIBIC campaign (Heintzenberg et al.,
443 2003). In the 4-12 nm size range (Fig. 15b), predicted number concentrations vary by a factor of
444 5 between nucleation simulations, with the highest nucleation rates (Zhao BHN) being associated
445 with the highest number concentration. But at sizes above 12 nm (Figs. 15c and 15d), the
446 differences in number concentration between the nucleation schemes become muted as
447 coagulation become dominant. When comparing the model to the observations, however, in this
448 comparison there are numerous discrepancies. Observed number concentrations peak near the
449 equator while the simulations peak near industrial latitudes. In the 4-12 nm size range (Fig.
450 15b), both observations and simulations peak at about 20,000 cm⁻³ Standard Temperature and
451 Pressure (STP) (273 K, 1013.25 hPa), while at the larger sizes the simulations underpredict
452 number concentration by up to an order of magnitude. The discrepancy at larger sizes may be
453 due to the model treating only sulfates and not other types of aerosols, such as organics or
454 biomass burning products. The underprediction could also be due to uncertainties in the

Jay 7/12/11 5:14 PM

Deleted: ,

Jay 7/12/11 5:14 PM

Deleted: or

455 emission of [gaseous precursors, particularly dimethylsulfide or volcanic emissions, which we do](#)
456 [not have in our model](#). It could also be due to the model not including other types of nucleation
457 near the surface, which is known to be important in this region (Kazil et. al. 2010). Any of these
458 discrepancies may be due to a mismatch between the spatial scale of the observations and
459 simulations. The observations are taken across a flight path at a specific altitude, latitude, and
460 longitude, while the model simply averages a region over the entire altitude and latitude range.
461 A contour plot of calculated number concentration for the Zhao BHN case as a function of
462 latitude and longitude between 216-316 hPa (Fig. 15a) shows that number concentration can vary
463 up to two orders of magnitude at the same altitude across the globe.

Jay 7/12/11 5:14 PM

Deleted: .

465 4.4 Effect of nucleation rate on Effective Radius and Extinction

466 Vertical profiles of effective radii and extinctions for each of the three calculated nucleation
467 schemes are compared to SAGE II satellite observations in Figs. 8 and 16. All three nucleation
468 schemes yield essentially identical results. Effective radius is important for radiative forcing,
469 while extinction is proportional to surface area, which is important to heterogeneous chemistry.
470 Hence the choice of nucleation rate should not be important to radiative forcing or atmospheric
471 chemistry.

473 5 Conclusions

474 We have implemented a three-dimensional general circulation model with sulfur chemistry and
475 sectional aerosol microphysics (WACCM/CARMA). Three nucleation schemes are available in
476 this model: two BHN schemes - one based on classical nucleation theory (Zhao BHN) and one
477 based on kinetic nucleation theory lookup tables (Yu BHN) – as well as an IMN scheme look-up
478 table (Yu IMN). The two Yu schemes often are found to be out of the table limits [in the middle](#)
479 [stratosphere and above](#) due to limits on boundary conditions of table inputs; this problem was
480 mostly resolved by setting nucleation to zero if relative humidity is less than the table
481 minimum value, but the tables should be used with caution [in these regions](#). Further
482 thermodynamic data are needed to extend the Yu schemes into dry stratospheric conditions
483 (RH<0.1%). Calculations suggest that ion-mediated nucleation rates in the UTLS are 25%
484 higher than binary only, [consistent with some studies \(e.g. Kazil et al., 2010\) but different from](#)

Jay 7/12/11 5:14 PM

Deleted: above

Jay 7/12/11 5:14 PM

Deleted: tropopause

Jay 7/12/11 5:14 PM

Deleted: above the tropopause.

Jay 7/12/11 5:14 PM

Deleted: ; however

485 | [others \(e.g., Yu et al., 2010\).](#) However, the two binary schemes vary by two orders of
486 | magnitude, [consistent with other studies \(Korhonen et al., 2003, Yu et al., 2010\).](#) More
487 | importantly, it is found that coagulation, not nucleation, controls number concentration at sizes
488 | greater than approximately 10 nm, [in the UTLS and the middle and upper stratosphere.](#) The
489 | dominance of coagulation over other microphysical processes is consistent with other recent
490 | work using microphysical models; Pierce and Adams (2009) found coagulation to be more
491 | important than nucleation in tropospheric studies, and Timmreck et al. (2010) found coagulation
492 | to drive stratospheric particle sizes from the eruption of Mount Toba to much larger values than
493 | previously assumed. [Lee et al. \(2003\) suggested that ion nucleation was important in the UTLS,](#)
494 | [on the basis of their ability to match observed size distributions with a model based on ion](#)
495 | [clusters. In contrast, we find that identical size distributions are produced for each type of](#)
496 | [nucleation, due to the dominance of coagulation. Hence fitting the size distribution is not](#)
497 | [diagnostic of the type of nucleation occurring \(binary or ion\). Unfortunately, the rate of](#)
498 | [nucleation is not easily determined from data either, because all of the particle properties for](#)
499 | [particles larger than 10nm are not altered even for two-order-of-magnitude changes in the](#)
500 | [nucleation rate.](#)

501 | We compared our calculations to observations from the tropopause to the mid-stratosphere.
502 | Above about 30 km, the model underpredicts SAGE extinctions, which we suggest is due to the
503 | importance of micrometeorites, as observed by Hervig et al (2009). We also found that
504 | including Van der Waals forces improved the model calculations for the numbers of particles in
505 | the UTLS. We conclude that this model contains the sulfate microphysical processes needed for
506 | simulations in the UTLS, and that the properties of particles with sizes relevant to climate, cloud
507 | physics and heterogeneous chemistry are not sensitive to the details of the nucleation scheme or
508 | to the presence or absence of ion nucleation.

509

510 Acknowledgements

511 | We thank Charles Bardeen for help coupling CARMA and CAM. We thank V. Lynn Harvey for
512 | help with the SAGE II extinction data. We thank Ryan Neely for help converting Deshler
513 | balloon data. This work was supported by NSF Awards ATM-0856007 and AGS-0942106,

Jay 7/12/11 5:14 PM

Deleted: .

Jay 7/12/11 5:14 PM

Deleted: .

Jay 7/12/11 5:14 PM

Deleted: Lee et al. (2003) suggested that ion nucleation was important in the UTLS, on the basis of their ability to match observed size distributions with a model based on ion clusters. In contrast we find that identical size distributions are produced for each mode of nucleation.

Jay 7/12/11 5:14 PM

Deleted: 2007

Michael Mills 7/21/11 3:43 PM

Deleted:

Michael Mills 7/21/11 3:43 PM

Deleted:

Michael Mills 7/21/11 3:43 PM

Deleted:

MacD63 LASP 7/14/11 10:13 AM

Deleted: Lee et al. (2003) suggested that ion nucleation was important in the UTLS, on the basis of their ability to match observed size distributions with a model based on ion clusters. In contrast, we find that identical size distributions are produced for each mode of nucleation, due to the dominance of coagulation

514 NASA Award NNX09AK71G, and NASA GSRP Fellowship NNX-09AM38H. NCAR is
515 sponsored by the National Science Foundation.

516

517 **References**

518 Alam, M. K.: The effect of Van der Waals and viscous forces on aerosol coagulation, *Aerosol*
519 *Sci. Tech.*, 6(1), 41-52, doi:10.1080/02786828708959118, 1987.

520 Arnold, F., Fabian, R., and Joos, W.: Measurements of the height variation of sulfuric-acid vapor
521 concentrations in the stratosphere, *Geophys. Res. Lett.*, 8, 293–296, 1981.

522 Ayers, G. P., Gillett, R. W., and Gras, J. L.: On the vapor-pressure of sulfuric acid, *Geophys.*
523 *Res. Lett.*, 7, no. 6, 433-436, 1980.

524 Bardeen, C. G., Toon, O. B., Jensen, E. J., Marsh, D. R., and Harvey, V. L.: Numerical
525 simulations of the three-dimensional distribution of meteoric dust in the mesosphere and upper
526 stratosphere, *J. Geophys. Res.*, 113, D17202, doi 10.1029/2007jd009515, 2008.

527 Bardeen, C. G., Toon, O. B., Jensen, E. J., Hervig, M. E., Randall, C. E., Benze, S., Marsh, D.
528 R., and Merkel, A.: Numerical simulations of the three-dimensional distribution of polar
529 mesospheric clouds and comparisons with Cloud Imaging and Particle Size (CIPS) experiment
530 and the Solar Occultation For Ice Experiment (SOFIE) observations, *J. Geophys. Res.*, 115,
531 D10204, doi:10.1029/2009JD012451, 2010.

532 Barth, M. C., Rasch, P. J., Kiehl, J. T., Benkovitz, C. M., and Schwartz, S. E.: Sulfur chemistry
533 in the National Center for Atmospheric Research Community Climate Model: Description,
534 evaluation, features, and sensitivity to aqueous chemistry, *J. Geophys. Res.-Atmospheres*, 105,
535 no. D1, 1387-1415, 2000.

536 Bates, T. S., Huebert, B., Gras, J., Griffiths, F., and Durkee, P.: The International Global
537 Atmospheric Chemistry (IGAC) Project's First Aerosol Characterization Experiment (ACE-1):
538 Overview, *J. Geophys. Res.*, 103, 16,297–16,318, 1998.

539 Blitz, M. A., McKee, K. W., and Pilling, M. J.: Temperature dependence of the reaction of OH
540 with SO, *Proceedings of the Combustion Institute*, 28, 2491 – 2497, 2000.

541 Blitz, M. A., Hughes, K. J., and Pilling, M. J.: Determination of the high-pressure limiting rate
542 coefficient and the enthalpy of reaction for OH+SO₂, *J. Phys. Chem. A*, 101, 1971-1978, 2003.

543 Borrmann, S., Kunkel, D., Weigel, R., Minikin, A., Deshler, T., Wilson, J. C., Curtius, J., Volk,
544 C. M., Homan, C. D., Ulanovsky, A., Ravegnani, F., Viciani, S., Shur, G. N., Belyaev, G. V.,
545 Law, K. S., and Cairo, F.: Aerosols in the tropical and subtropical UT/LS: in-situ measurements
546 of submicron particle abundance and volatility, *Atmos. Chem. Phys.*, 10, no. 12, 5573-5592,
547 2010.

548 Brock, C. A., Hamill, P., Wilson, J. C., Jonsson, H. H., and Chan, K. R.: Particle formation in the
549 upper tropical troposphere – A source of nuclei for the stratospheric aerosol, *Science*, 270, 1650–
550 1653, 1995.

551 Brunning, J. and Stief, L. J.: Kinetic studies of the reaction of the SO radical with NO₂ and ClO
552 from 210 to 363 K, *J. Chem. Phys.*, 84, 4371-4377, 1986.

553 Brunning, J. and Stief, L. J.: Rate constant for the reaction SO + BrO → SO₂ + Br, *J. Chem.*
554 *Phys.*, 85(5), 2591, doi:10.1063/1.451066, 1986b.

555 Burkholder, J. B. and McKeen, S.: UV absorption cross-section for SO₃, *Geophys. Res. Lett.*,
556 24(24), 3201–3204, 1997.

557 Chan, T. W. and Mozurkewich, M.: Measurement of the coagulation rate constant for sulfuric
558 acid particles as a function of particle size using tandem differential mobility analysis, *J. Aerosol*
559 *Sci.*, 32, no. 3, 321-339, 2001.

560 Cheng, B. M. and Lee, Y. P.: Rate constant of OH + OCS reaction over the temperature range
561 255-483K, *Int. J. Chem. Kinet.*, 18, 1303-1314, 1986.

562 Chu, W. P., McCormick, M.P., Lenoble, J., Brogniez, C., and Pruvost, P.: SAGE II inversion
563 algorithm, *J. Geophys. Res.*, 94, 8339 – 8351, 1989.

564 Clarke, A. D.: Atmospheric nuclei in the Pacific midtroposphere – their nature, concentration,
565 and evolution, *J. Geophys. Res. – Atmospheres*, 98, no. D11, 20633-20647, 1993.

566 Clarke, A. D. and Kapustin, V. N.: A Pacific aerosol survey. part I: A decade of data on particle
567 production, transport, evolution, and mixing in the troposphere. *J. Atmos. Sci.*, 52, 363–382,
568 2002.

569 Clyne, M. A. A. and MacRobert, A. J.: Kinetic-studies of free-radical reactions by mass-
570 spectrometry, *Int. J. Chem. Kinet.*, 13, 187-197, 1981.

571 Clyne, M. A. A. and Townsend, L. W.: Rate constant measurements for rapid reactions of
572 ground-state sulfur $3P_4(P-3(J))$ atoms, *Int. J. Chem. Kinet., Symp. 1*, 73-84, 1975.

573 Coffman, D. J. and Hegg, D. A.: A preliminary study of the effect of ammonia on particle
574 nucleation in the marine boundary layer, *J. Geophys. Res.*, 100(D4), 7147–7160, 1995.

575 Colella, P. and Woodward, P. R.: The piecewise parabolic method (PPM) for gas-dynamical
576 simulations, *J. Comput. Phys.*, 54, no. 1, 174-201, 1984.

577 Crutzen, P. J.: Albedo enhancement by stratospheric sulfur injections: A contribution to resolve a
578 policy dilemma?, *Climatic Change*, 77, no. 3-4, 211-219, 2006.

579 Curtius, J., Sierau, B., Arnold, F., de Reus, M., Strom, J., Scheeren, H. A., and Lelieveld, J.:
580 Measurement of aerosol sulfuric acid 2. Pronounced layering in the free troposphere during the
581 second Aerosol Characterization Experiment (ACE 2), *J. Geophys. Res.-Atmospheres*, 106, no.
582 D23, 31975-31990, 2001.

583 Davis, D. D., Klemm, R. B. and Pilling, M.: A flash photolysis-resonance fluorescence kinetics
584 study of ground-state sulfur atoms: I. Absolute rate parameters for reaction of $S(^3P)$ with $O_2(^3\Sigma)$,
585 *Int. J. Chem. Kinet.*, 4, 367-382. doi: 10.1002/kin.550040402, 1972.

586 Deshler, T., Hervig, M. E., Hofmann, D. J., Rosen, J. M., and Liley, J. B.: Thirty years of in-situ
587 stratospheric aerosol size distribution measurements from Laramie, Wyoming (41N), using
588 balloon-borne instruments, *J. Geophys. Res.*, 108(D5), 4167, doi:10.1029/2002JD002514, 2003.

589 Eisele, F. L., Lovejoy, E. R., Kosciuch, E., Moore, K. F., Mauldin III, R. L., Smith, J. N.,
590 McMurry, P. H., and Iida, K.: Negative atmospheric ions and their potential role in ion-induced
591 nucleation, *J. Geophys. Res.*, 111, D04305, doi:10.1029/2005JD006568, 2006.

592 Fan, T. and Toon, O. B.: Modeling sea-salt aerosol in a coupled climate and sectional
593 microphysical model: mass, optical depth and number concentration, *Atmos. Chem. Phys.*
594 *Discuss.*, 10, 1–63, doi:10.5194/acpd-10- 24499 -2010, 2010.

595 Flood, H.: Droplet formation in oversaturated ethanol-water vapour mixtures, *Zeit. Chem.*
596 *Phys.*, 170(3/4), 286-294, 1934.

597 Fuchs, N. A.: *The Mechanics of Aerosols*, Pergamon, New York, 1964.

598 Garcia, R. R., Marsh, D. R., Kinnison, D. E., Boville, B. A., and Sassi, F.: Simulation of secular
599 trends in the middle atmosphere, 1950–2003, *J. Geophys. Res.-Atmos.*, 112(D9), D09301,
600 doi:10.1029/2006JD007485, 2007.

601 Giaque, W. F., Hornung, E. W., Kunzler, J. E., and Rubin, T. R.: The thermodynamic properties
602 of aqueous sulfuric acid solutions and hydrates from 15-degrees-K to 300-degrees-K, *J. Amer.*
603 *Chem. Soc.*, 82, no. 1, 62-70, 1960.

604 Hamill, P., Toon, O. B., and Kiang, C. S.: Microphysical processes affecting stratospheric
605 aerosol particles, *J. Atmos. Sci.*, 34, 1104-1119, 1977.

606 Hanson, D. R., and Lovejoy, E. R.: Measurement of the thermodynamics of the hydrated dimer
607 and trimer of sulfuric acid, *J. Phys. Chem. A-Letters*, 110, 9525-9528, 2006.

608 Heckendorn, P., Weisenstein, D., Fueglistaler, S., Luo, B. P., Rozanov, E. Schraner, M.,
609 Thomason, L. W., and Peter, T.: The impact of geoengineering aerosols on stratospheric
610 temperature and ozone, *Environ. Res. Lett.*, 4, 045108, doi:10.1088/1748-9326/4/4/045108,
611 2009.

612 Heintzenberg, J., Hermann, M., and Theiss, D.: Out of Africa: High aerosol concentrations in the
613 upper troposphere over Africa, *Atmos. Chem. Phys.*, 3, 1191–1198, doi:10.5194/acp-3-1191-
614 2003, 2003.

615 Hervig, M. E., Gordley, L. L., Deaver, L. E., Siskind, D. E., Stevens, M. H., Russell III, J. M.,
616 Bailey, S. M., Megner, L., and Bardeen, C. G.: First satellite observations of meteoric smoke in
617 the middle atmosphere, *Geophys. Res. Lett.*, 36, L18805, doi:10.1029/2009GL039737, 2009.

618 Hoell, J. M., Davis, D. D., Jacob, D. J., Rodgers, M. O., Newell, R. E., Fuelberg, H. E., McNeal,
619 R. J., Raper, J. L., and Bendura, R. J.: Pacific Exploratory Mission in the tropical Pacific: PEM-
620 Tropics A, August-September 1996, *J. Geophys. Res.*, 104, no. D5, 5567-5583, 1999.

621 Hofman, D. J., and Solomon, S.: Ozone destruction through heterogeneous chemistry following
622 the eruption of El Chichon, *J. Geophys. Res.-Atmospheres*, 94, no. D4, 5029-5041, 1989.

623 Huang, D. D., Seinfeld, J. H., and Marlow, W. H.: BGK equation solution of coagulation for
624 large Knudsen number aerosols with a singular attractive contact potential, *J. Colloid Interf. Sci.*,
625 140(1), 258-276, doi:10.1016/0021-9797, 1990.

626 Hunten, M. D., Turco, R. P., and Toon, O. B.: Smoke and dust particles of meteoric origin in the
627 mesosphere and stratosphere, *J. Atmos. Sci.*, 37, 1342-1357, 1980.

628 Jensen, E. J., Toon, O. B., Selkirk, H. B., Spinhirne, J. D., and Schoeberl, M. R.: On the
629 formation and persistence of subvisible cirrus clouds near the tropical tropopause, *J. Geophys.*
630 *Res.*, 101, No. D16, 21361-21375, 1996.

631 Jourdain, J. L., Le Bras, G., and Combourieu, J.: Kinetic study of reactions of 1,1,1 trifluoro-2
632 chloroethane with chlorine atoms and oxygen atoms, *J. Chim. Phys.*, 75, 318-323, 1978.

633 Kanawade, V. and Tripathi, S. N.: Evidence for the role of ion-induced particle formation during
634 an atmospheric nucleation event observed in Tropospheric Ozone Production about the Spring
635 Equinox (TOPSE), *J. Geophys. Res.-Atmos.*, 111(D2), D02209, doi:10.1029/2005JD006366,
636 2006.

637 Kazil, J., Lovejoy, E. R., Jensen, E. J., and Hanson, D. R.: Is aerosol formation in cirrus clouds
638 possible?, *Atmos. Chem. Phys.*, 7, 1407-1413, 2007.

639 Kazil, J., Stier, P., Zhang, K., Quaas, J., Kinne, S., O'Donnell, D., Rast, S., Esch, M., Ferrachat,
640 S., Lohmann, U., and Feichter, J.: Aerosol nucleation and its role for clouds and Earth's radiative
641 forcing in the aerosol-climate model ECHAM5-HAM, *Atm. Chem. Phys.*, 10, no. 22, 10733-
642 10752, 2010.

643 Kinnison, D. E., Brasseur, G. P., Walters, S., Garcia, R. R., Marsh, D. R., Sassi, F., Harvey, V.
644 L., Randall, C. E., Emmons, L., Lamarque, J. F., Hess, P., Orlando, J. J., Tie, X. X., Randel, W.,
645 Pan, L. L., Gettelman, A., Granier, C., Diehl, T., Niemeier, U., and Simmons, A. J.: Sensitivity of
646 chemical tracers to meteorological parameters in the MOZART-3 chemical transport model, *J.*
647 *Geophys. Res.*, 112, D20302, doi:10.1029/2006JD007879, 2007.

648 [Korhonen, H., K., Lehtinen, E. J., Pirjola, L., Napari, J., Vehkamäki, H., Noppel, M., and](#)
649 [Kulmala, M.: Simulation of atmospheric nucleation mode: A comparison of nucleation models](#)
650 [and size distribution representations, *J. Geophys. Res.*, 108, D15, doi:10.1029/2002JD003305,](#)
651 [2003.](#)

652 Kulmala, M. and Laaksonen, A.: Binary nucleation of water sulfuric-acid system – comparison
653 of classical-theories with different H₂SO₄ saturation vapor-pressures, *J. Chem. Phys.*, 93, no. 1,
654 696-701, 1990.

655 Lamarque J.-F., Bond, T. C., Eyring, V., Granier, C., Heil, A., Klimont, Z., Lee, D., Liousse, C.,
656 Mieville, A., Owen, B., Schultz, M. G., Shindell, D., Smith, S. J., Stehfest, E., Van Aardenne, J.,
657 Cooper, O. R., Kainuma, M., Mahowald, N., McConnell, J. R., Naik, V., Riahi, K., van Vuuren,
658 D. P.: Historical (1850-2000) gridded anthropogenic and biomass burning emissions of reactive
659 gases and aerosols: methodology and application, *Atmos. Chem. Phys.*, 10, no. 15, 7017-7039,
660 2010.

661 Lee, S.-H., Reeves, J. M., Wilson, J. C., Hunton, D. E., Viggiano, A. A., Miller, T. M.,
662 Ballenthin, J. O., and Lait, L. R.: Particle formation by ion induced nucleation in the upper
663 troposphere and lower stratosphere, *Science*, 301, 1886–1889, 2003.

664 Lin, S. J., and Rood, R. B: Multidimensional flux-form semi-Lagrangian transport schemes,
665 *Monthly Weather Rev.*, 124, no. 9, 2046-2070, 1996.

666 Lin, S. J., and Rood, R. B.: An explicit flux-form semi-Lagrangian shallow-water model on the
667 sphere, *Q. J. R. Meteorol. Soc.*, 123, 2477-2498, 1997.

668 Lin, J. S., and Tabazadeh, A.: A parameterization of an aerosol physical chemistry model for the
669 NH₃/H₂SO₄/HNO₃/H₂O system at cold temperatures, *J. Geophys. Res.-Atmospheres* 106, no.
670 D5, 4815-4829, 2001.

671 Lovejoy, E. R., Hanson, D. R., and Huey, L. G.: Kinetics and products of the gas-phase reaction
672 of SO₃ with water, *J. Phys. Chem.*, 100, 51, 19911-19916, 1996.

673 Lovejoy, E. R., Curtius, J., and Froyd, K. D.: Atmospheric ion-induced nucleation of sulfuric
674 acid and water, *J. Geophys. Res.*, 109, D08204, doi:10.1029/2003JD004460, 2004.

675 Lucas, D. D. and Prinn, R. G.: Tropospheric distributions of sulfuric acid-water vapor aerosol
676 nucleation rates from dimethylsulfide oxidation, *Geophys. Res. Lett.*, 30(22), 2136,
677 doi:10.1029/2003GL018370, 2003.

678 Mills, M. J., Toon, O. B., Turco, R. P., Kinnison, D. E., and Garcia, R. R.: Massive global ozone
679 loss predicted following regional nuclear conflict, *PNAS*, 105(14), 5307-5312,
680 doi:10.1073/pnas.0710058105, 2008.

681 [Mills, M. J., Toon, O. B., Vaida, V., Hintze, P. E., Kjaergaard, H. G., Schofield, D. P., and](#)
682 [Robinson, T. W.: Photolysis of sulfuric acid vapor by visible light as a source of the polar](#)
683 [stratospheric CN layer, J. Geophys. Res., 110, D08201, doi:10.1029/2004JD005519, 2005.](#)

684 Molina, L. T. and Molina, M. J.: UV absorption cross-sections of HO₂NO₂ vapor, J.
685 Photochem., 15, 97-108, 1981.

686 [Murphy, D. M., Cziczo, D. J., Hudson, P. K., and Thomson, D. S.: Carbonaceous material in](#)
687 [aerosol particles in the lower stratosphere and tropopause region, J. Geophys. Res., 112, D04203,](#)
688 [doi:10.1029/2006JD007297, 2007.](#)

689 Okabe, H.: In Photochemistry of Small Molecules, John Wiley and Sons Inc.: New York, pp.
690 248-249, 1978.

691 Palmer, K. F., and Williams, D.: Optical constants of sulfuric acid; Application to the clouds of
692 Venus?, Appl. Optics, 14, 208-219, 1975.

693 Pierce, J. R. and Adams, P. J.: Can cosmic rays affect cloud condensation nuclei by altering new
694 particle formation rates?, Geophys. Res. Lett., 36, L09820, doi:10.1029/2009GL037946, 2009.

695 Rasch, P. J., Tilmes, S., Turco, R. P., Robock, A., Oman, L., Chen, C., Stenchikov, G. L., and
696 Garcia, R. R.: An overview of geoengineering of climate using stratospheric sulphate aerosols,
697 Philos. Trans. of the Royal Soc. A-Math. Phys. and Eng. Sci., 366, no. 1882, 4007-4037, 2008.

698 Reiner, T., and Arnold, F.: Stratospheric SO₃: Upper limits inferred from ion composition
699 measurement: Implications for H₂SO₄ and aerosol formation, Geophys. Res. Lett., 24, 1751–
700 1754, 1997.

701 Reiss, H.: The kinetics of phase transitions in binary systems, J. Chem. Phys., 18, no. 6, 840-848,
702 1950.

703 Robock, A., Oman, L, and Stenchikov, G. L.: Regional climate responses to geoengineering with
704 tropical and Arctic SO₂ injections, J. Geophys. Res.-Atmos., 113, D16101,
705 doi:10.1029/2008JD010050, 2008.

706 Ross, M., Mills, M., and Toohy, D.: Potential climate impact of black carbon emitted by
707 rockets, Geophys. Res. Lett., 37, L24810, doi:10.1029/2010GL044548, 2010.

708 Sabinina, A. L. and Terpuogow, L.: Die oberflächenspannung de systems schwefelsaure-wasser,
709 Z. Phys. Chem. A, 173, 237–241, 1935.

710 Sander, S. P., Finlayson-Pitts, B. J., Friedl, R. R., Golden, D. M., Huie, R. E., Keller-Rudek, H.,
711 Kolb, C. E., Kurylo, M. J., Molina, M. J., Moortgat, G. K., Orkin, V. L., Ravishankara, A. R., and
712 Wine, P. H.: Chemical Kinetics and Photochemical Data for Use in Atmospheric Studies,
713 Evaluation Number 15, JPL Publication 06-2, Jet Propulsion Laboratory, Pasadena, 2006.

714 Schlager, H. and Arnold, F.: Balloon-borne composition measurements of stratospheric negative-
715 ions and inferred sulfuric-acid vapor abundances during the Map-Globus 1983 campaign, Planet.
716 Space Sci., 35, 693– 701, 1987.

717 Schmidt-Ott, A. and Burtscher, H.: The effect of Van der Waals forces on aerosol coagulation,
718 J. Colloid Interf. Sci., 89(2), 353-357, doi:10.1016/0021-9797(82)90187-4, 1982.

719 Smith, S. J., van Aardenne, J., Klimont, Z., Andres, R., Volke, A., and Delgado Arias, S.:
720 Anthropogenic sulfur dioxide emissions: 1850–2005, Atmos. Chem. Phys. Discuss., 10, 16111-
721 16151, doi:10.5194/acpd-10-16111-2010, 2010.

722 [Snow-Kropla, E. J., Pierce, J. R., Westervelt, D. M., and Trivitayanurak, W.: Cosmic rays,](#)
723 [aerosol formation and cloud-condensation nuclei: sensitivities to model uncertainties, Atmos.](#)
724 [Chem. Phys., 11, 4001-4013, doi:10.5194/acp-11-4001-2011, 2011.](#)

725 Su, L., and Toon, O. B.: Saharan and Asian Dust: Similarities and Differences Determined by
726 CALIPSO, AERONET, and a Coupled Climate-Aerosol Microphysical Model, Atmos. Chem.
727 Phys. Discuss., 10, 29513-29567, 2010.

728 Tabazadeh, A., Toon, O. B., Clegg, S. L., and Hamill, P.: A new parameterization of
729 H₂SO₄/H₂O aerosol composition: Atmospheric implications. Geophys. Res. Lett., 24, no. 15,
730 1931-1934, 1997.

731 Thornton, D. C., Bandy, A. R., Blomquist, B. W., Driedger, A. R., and Wade, T. P.: Sulfur
732 dioxide distribution over the Pacific Ocean 1991–1996, J. Geophys. Res., 104, 5845–5854, 1999.

733 Timmreck, C., Graf, H.-F., Lorenz, S. J., Niemeier, U., Zanchettin, D., Matei, D., Jungclaus, J.
734 H., and Crowley, T. J.: Aerosol size confines climate response to volcanic super-eruptions,
735 Geophys. Res. Lett., 37, L24705, doi:10.1029/2010GL045464, 2010.

736 Toon, O. B., Turco, R. P., Westphal, D., Malone, R., and Liu, M. S.: A multidimensional model
737 for aerosols - description of computational analogs, *J. Atmos. Sci.*, 45, 2123-2143, 1988.

738 Usoskin, I. G., Desorgher, L., Velinov, P., Storini, M., Fluckiger, E. O., Butikofer, R.,
739 Kovaltsov, G. A.: Ionization of the earth's atmosphere by solar and galactic cosmic rays." *Acta*
740 *Geophysica*, 57, no. 1, 88-101, 2009.

741 Vaida, V., Kjaergaard, H. G., Hintze, P. E., and Donaldson, D. J.: Photolysis of sulfuric acid
742 vapor by visible solar radiation, *Science*, 299, 1566-1568, 2003.

743 Viggiano, A. A. and Arnold, F.: Extended sulfuric acid vapor concentration measurements in the
744 stratosphere, *Geophys. Res. Lett.*, 8, 583-586, 1981.

745 Wang, P. H., Minnis, P., and Yue, G. K.: Extinction coefficient (1 μm) properties of high-altitude
746 clouds from solar occultation measurements (1985-1990): Evidence of volcanic aerosol effect, *J.*
747 *Geophys. Res.*, 100, No. D2, 3181-3199, 1995.

748 Wang, P. H., Kent, G. S., and McCormick, M. P.: Retrieval analysis of aerosol-size distribution
749 with simulated extinction measurements at SAGE III wavelengths, *Applied Optics*, 35, 3, 433-
750 440, 1996.

751 Wang, Y. H., Liu, S. C., Wine, P. H., Davis, D. D., Sandholm, S. T., Atlas, E. L., Avery, M. A.,
752 Blake, D. R., Blake, N. J., Brune, W. H., Heikes, B. G., Sachse, G. W., Shetter, R. E., Singh, H.
753 B., Talbot, R. W., Tan, D.: Factors controlling tropospheric O₃, OH, NO_x and SO₂ over the
754 tropical Pacific during PEM-Tropics B, *J. Geophys. Res.-Atmospheres*, 106, no. D23, 32733-
755 32747, 2001.

756 Wilson, J. C., Lee, S.-H., Reeves, J. M., Brock, C. A., Jonsson, H. H., Lafleur, B. G.,
757 Loewenstein, M., Podolske, J., Atlas, E., Boering, K., Toon, G., Fahey, D., Bui, T. P., Diskin, G.,
758 and Moore, F.: Steady-state aerosol distributions in the extra-tropical, lower stratosphere and the
759 processes that maintain them, *Atmos. Chem. Phys.*, 8, 6617-6626, doi:10.5194/acp-8-6617-
760 2008, 2008.

761 Yu, F.: Updated H₂SO₄-H₂O binary homogeneous nucleation look-up tables, *J. Geophys. Res.*,
762 113, D24201, doi:10.1029/2008JD010527, 2008.

763 Yu, F.: Ion-mediated nucleation in the atmosphere: Key controlling parameters, implications,
764 and look-up table, *J. Geophys. Res.*, 115, D03206, doi:10.1029/2009JD012630, 2010.

765 Yu, F. and Luo, G.: Simulation of particle size distribution with a global aerosol model:
766 contribution of nucleation to aerosol and CCN number concentrations, *Atmos. Chem. Phys.*, 9,
767 10597-10645, 2009.

768 Yu, F., Luo, G., Bates, T. S., Anderson, B., Clarke, A., Kapustin, V., Yantosca, R. M., Wang,
769 Y., and Wu, S.: Spatial distributions of particle number concentrations in the global troposphere:
770 Simulations, observations, and implications for nucleation mechanisms, *J. Geophys. Res.*, 115,
771 D17205, doi:10.1029/2009JD013473, 2010.

772 Yu, F. and Turco, R. P.: From molecular clusters to nanoparticles: The role of ambient ionization
773 in tropospheric aerosol formation, *J. Geophys. Res.*, 106, 4797-4814, 2001.

774 [Yu, F., and R. P. Turco, The size-dependent charge fraction of sub-3-nm particles as a key](#)
775 [diagnostic of competitive nucleation mechanisms under atmospheric conditions, *Atmos. Chem.*](#)
776 [Phys. Discuss., 11, 11281-11309, 2011.](#)

777 Yung, Y.L. and Demore, W. B.: Photochemistry of the stratosphere of Venus: Implications for
778 atmospheric evolution, *Icarus*, 51, 199-247, 1982.

779 Zhang, R., Suh, I., Zhao, J., Zhang, D., Fortner, E. C., Tie, X., Molina, L. T., and Molina, M. J.:
780 Atmospheric new particle formation enhanced by organic acids, *Science*, 304(5676), 1487-1490,
781 2004.

782 Zhao, J. and Turco, R. P.: Nucleation simulations in the wake of a jet aircraft in stratospheric
783 flight, *J. Aerosol Sci.* 26, no. 5, 779-795, 1995.

784

784
 785 Table 1. Sulfur reactions included in WACCM/CARMA. Reactions R1 through R12 obtained
 786 from NASA Jet Propulsion Laboratory Chemical Kinetics 15th evaluation (Sander et al., 2006)
 787 with original citations also noted. Sander et al. combined numerous sources to determine rates
 788 R1, R7, R8, and R13.
 789

Binary Reactions			
	<u>Reaction</u>	<u>Rate (cm³/s)</u>	<u>Source</u>
R1	OCS + O → SO + CO	2.1e ⁻¹¹ exp(-2200/T)	Sander et al. (2006)
R2	OCS + OH → SO ₂ + {C} + H	1.1e ⁻¹³ exp(-1200/T)	Cheng and Lee (1986)
R3	S + OH → SO + H	6.6e ⁻¹¹	Jourdain et al. (1978)
R4	S + O ₂ → SO + O	2.3e ⁻¹²	Davis et al. (1972)
R5	S + O ₃ → SO + O ₂	1.2e ⁻¹¹	Clyne and Townsend (1975)
R6	SO + OH → SO ₂ + H	2.70e ⁻¹¹ exp(335/T)	Blitz et al. (2000)
R7	SO + O ₂ → SO ₂ + O	1.25e ⁻¹³ exp(-2190/T)	Sander et al. (2006)
R8	SO + O ₃ → SO ₂ + O ₂	3.4e ⁻¹² exp(-1100/T)	Sander et al. (2006)
R9	SO + NO ₂ → SO ₂ + NO	1.4e ⁻¹¹	Brunning and Stief (1986-1)
R10	SO + CLO → SO ₂ + CL	2.8e ⁻¹¹	Brunning and Stief (1986-1)
R11	SO + BRO → SO ₂ + BR	5.7e ⁻¹¹	Brunning and Stief (1986-2)
R12	SO + OCLO → SO ₂ + CLO	1.9e ⁻¹²	Clyne and MacRobert (1981)
R13	HSO ₃ + O ₂ → SO ₃ + HO ₂	1.3e ⁻¹² exp(-330/T)	Sander et al. (2006)
R14	SO ₃ + H ₂ O → H ₂ SO ₄	2.26e ⁻²³ * T * exp(6544/T)	Lovejoy (1996)
Ternary Reactions			
	<u>Reaction</u>	<u>Rate</u>	<u>Source</u>
T1	SO ₂ + OH + M → HSO ₃ + M	k _o = 3.0e ⁻³¹ (T/300) ^{4.3} k _{infinity} : 1.6e ⁻¹²	Blitz et al. (2003)
Photolytic Reactions			

Reaction		Source
J1	$\text{H}_2\text{SO}_4 + \text{h}\nu \rightarrow \text{SO}_3 + \text{H}_2\text{O}$	Vaida et al. (2003)
J2	$\text{SO}_2 + \text{h}\nu \rightarrow \text{SO} + \text{O}$	Okabe (1978), Yung and DeMore (1982)
J3	$\text{SO}_3 + \text{h}\nu \rightarrow \text{SO}_2 + \text{O}$	Burkholder and McKeen (1997)
J4	$\text{OCS} + \text{h}\nu \rightarrow \text{S} + \text{CO}$	Molina et al (1981)
J5	$\text{SO} + \text{h}\nu \rightarrow \text{S} + \text{O}$	Yung and Demore (1982)

790

791 Table 2. Modeled critical radii and nucleation rates in the UTLS (between 50 and 500 hPa and
792 50° S and 50° N). All calculations are for the annual average of the third simulation year, except
793 for 30-min peak rate, which is the maximum rate found across all 30-minute timesteps for the
794 third simulation year. Annual average is the average across all grid boxes in the specified UTLS
795 range. Annual peak rate is for the grid box with the highest annual rate in the specified UTLS
796 range.

797

Simulation	Critical radius (nm)	Annual average rate (cm⁻³ s)	Annual peak rate (cm⁻³ s)	30-min peak rate (cm⁻³ s)
Zhao BHN	0.45	53.08	5099	432279
Yu BHN	0.72	0.50	71	2587
Yu IMN	0.72	0.75	70	2761

798

799 Table 3. Count and percent of model simulation days that meet conditions for NPF as defined by
800 Lee et al. (2003). Lee et al. reported 16% of total size distributions to be considered NPF events.

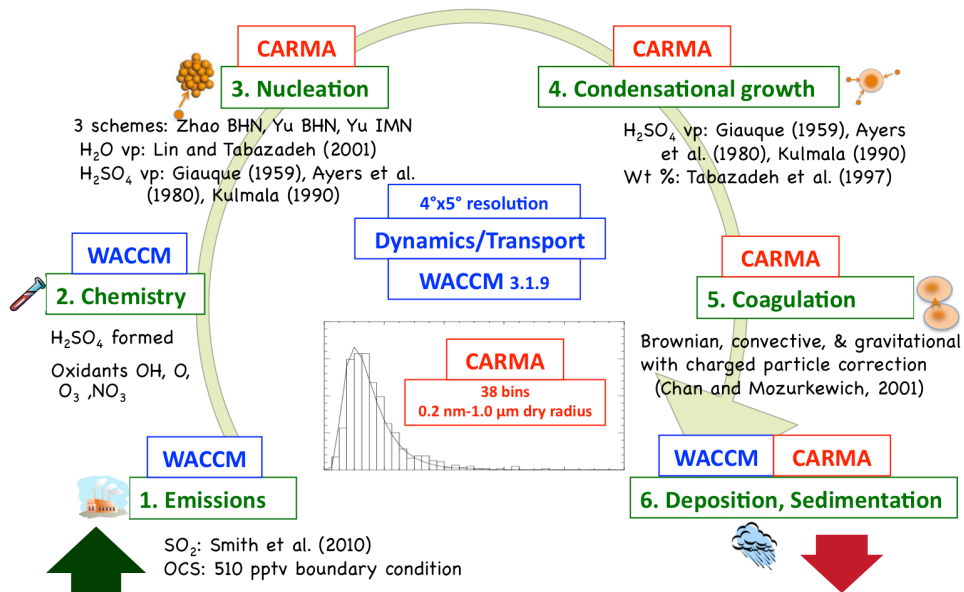
	Zhao BHN		Yu BHN		Yu IMN	
	Count	%	Count	%	Count	%
Tropical Troposphere	360/360	100%	360/360	100%	360/360	100%
Mid-High Latitude UTLS	360/360	100%	348/360	97%	360/360	100%
High Latitude Stratosphere	14/360	3.9%	2/360	0.6%	4/360	1.1%

801

802

802

803



804

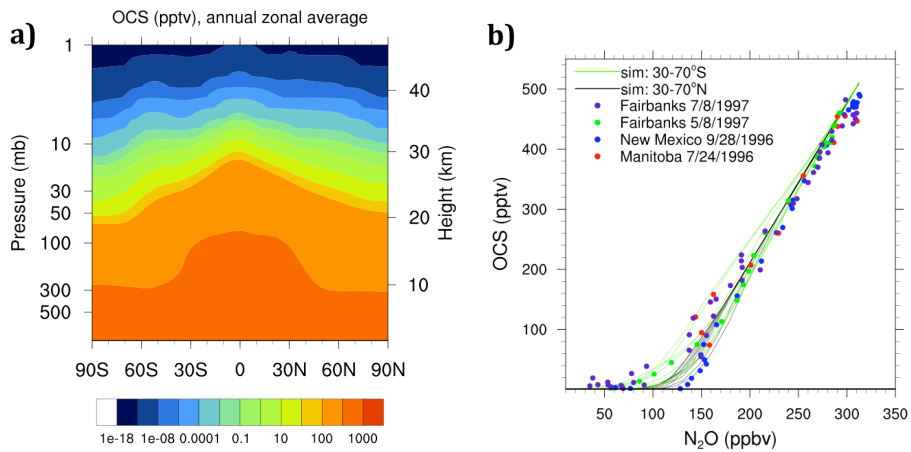
805 Figure 1. Diagram of the WACCM/CARMA Model. WACCM simulates emissions, chemistry,

806 dynamics and wet deposition. CARMA simulates nucleation, condensational growth,

807 coagulation, and sedimentation.

808

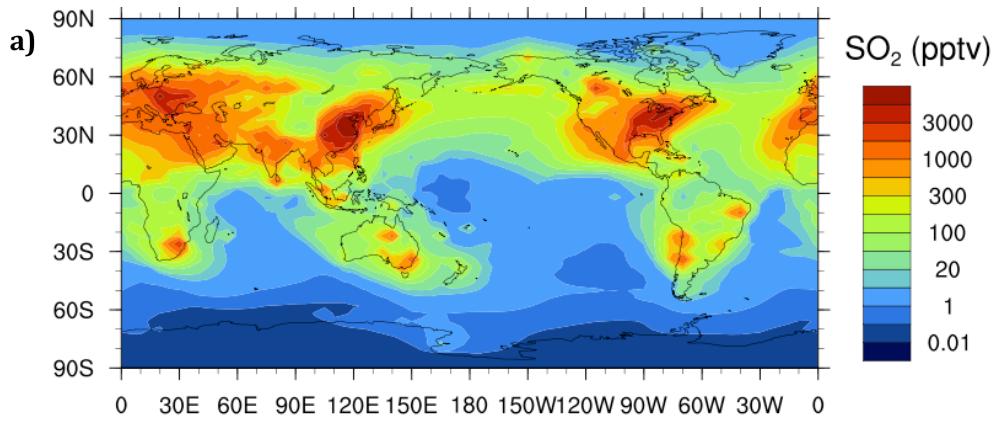
809



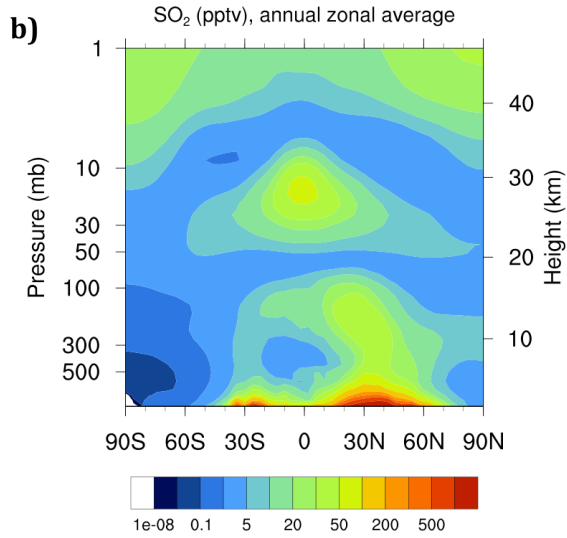
810

811 Figure 2. (a) Calculated OCS mixing ratio; annual and zonal average as a function of
 812 atmospheric pressure and latitude. (b) Calculated OCS versus N₂O, compared to balloon
 813 observations by Geoff Toon (private communication). Observations are from ascent and descent
 814 profiles at dates and locations noted. Simulation lines are an average of JJA at the latitude ranges
 815 noted. [Each simulation line represents model output at a specific latitude at 4° increments](#)
 816 [between 30° and 70° at all longitudes.](#)

817

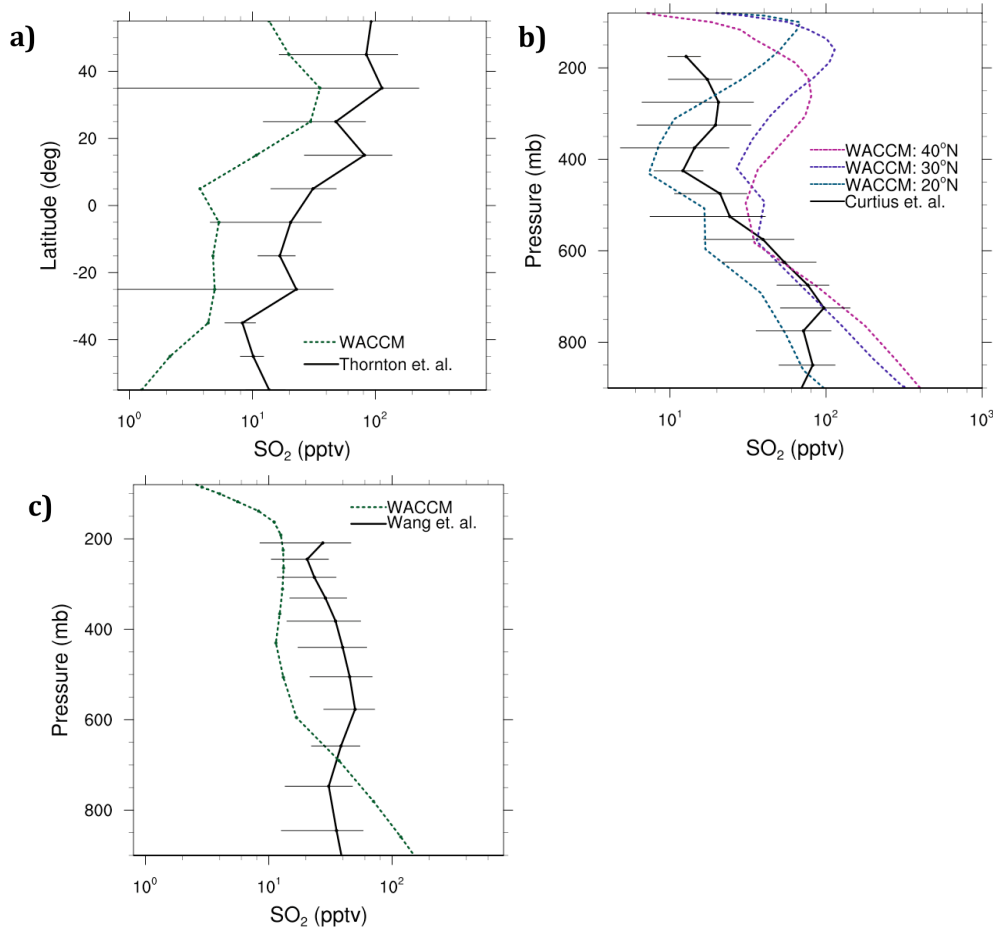


818



819

820 Figure 3. (a) Calculated surface SO₂ mixing ratio; annual average as a function of latitude and
 821 longitude. (b) Calculated SO₂ mixing ratio; annual and zonal average as a function of
 822 atmospheric pressure and latitude.



823

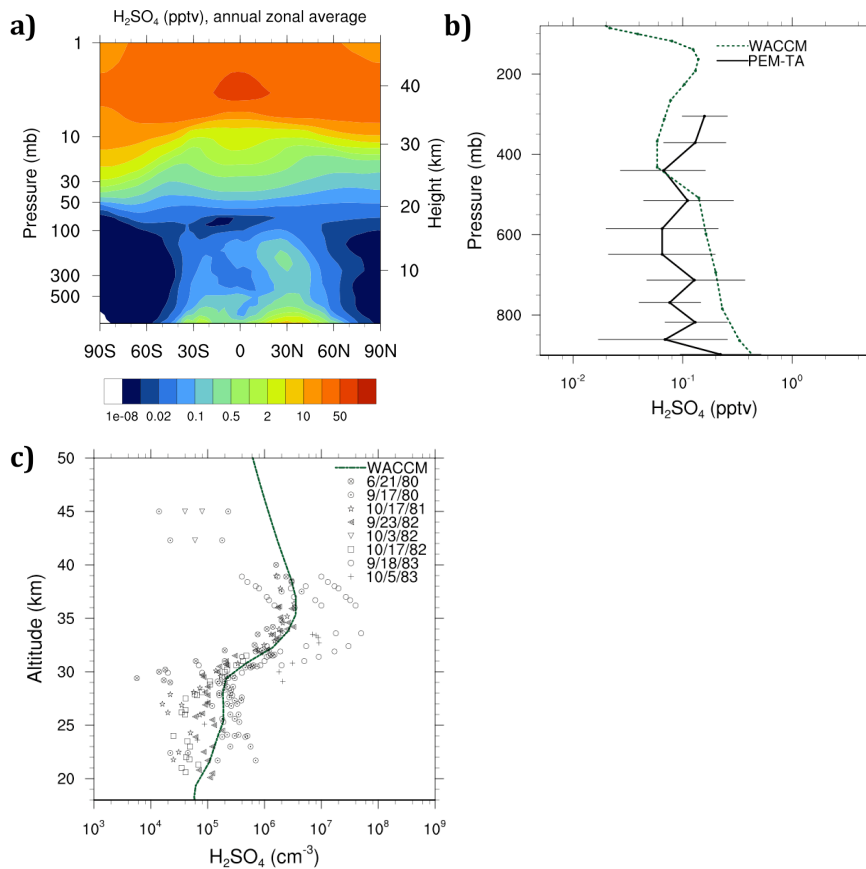
824

825

826 Figure 4. (a) Calculated SO₂ concentration between 8-12 km at varying latitudes compared to an
 827 average of Pacific Exploratory Mission (PEM)-West A, PEM-West B, and Atmospheric
 828 Chemistry Experiment (ACE)-2 aircraft observations over the Pacific Ocean between 110°E and
 829 80°W, binned into 10-degree segments with error bars representing plus/minus one standard
 830 deviation (Thornton et al., 1999). Model calculations are an annual average in the same
 831 longitude and altitude region. (b) Vertical profiles of calculated SO₂ mixing ratio at different
 832 latitudes compared to ACE-2 aircraft observations (Curtius et al., 2001). Observations are an
 833 average and standard deviation of 6 flights in July 1997 from 28°N to 32°N. Simulation lines are

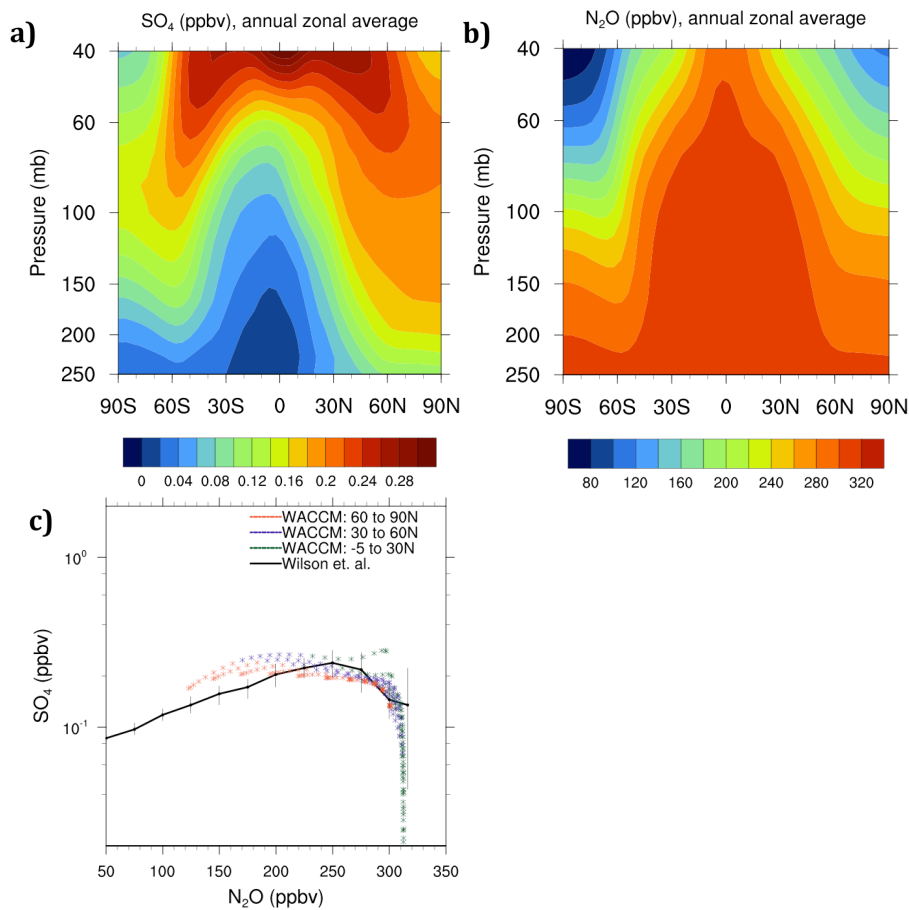
Jay 7/12/11 5:14 PM
 Deleted: (b
 Jay 7/12/11 5:14 PM
 Deleted:) Vertical profile of calculated SO₂ mixing ratio between 30°N and 30°S compared to PEM-TB aircraft observations (Wang et al., 2001). Observations are an average and standard deviation of DC-8 flights between 38°N and 36°S in March and April 1999. Simulation is an average of March and April from 38°N to 36°S. (c

834 an average of the month of July at each of the latitudes noted. (c) Vertical profile of calculated
 835 SO_2 mixing ratio between 30°N and 30°S compared to PEM-TB aircraft observations (Wang et
 836 al., 2001). Observations are an average and standard deviation of DC-8 flights between 38°N
 837 and 36°S in March and April 1999. Simulation is an average of March and April from 38°N to
 838 36°S .
 839



841
 842 Figure 5. (a) Calculated H_2SO_4 mixing ratio; annual and zonal average as a function of
 843 atmospheric pressure and latitude. (b) Vertical profile of calculated H_2SO_4 mixing ratio
 844 compared to PEM-TA (Bates et al., 1998, Hoell et al., 1999) aircraft observations binned into
 845 averages plus/minus standard deviation at different altitudes (Lucas and Prinn, 2003) from flights
 846 in August and September 1996 between 24°N and 24°S . Simulation lines are an average of July

847 and August between 24 N and 24 S. (c) Vertical profile of calculated H₂SO₄ number
 848 concentration at 43 N compared to various balloon-borne observations at the same latitude
 849 (Arnold et al., 1981, Reiner and Arnold, 1997, Schlager and Arnold, 1987, Viggiano and Arnold,
 850 1981). Simulation is an average of the months of September and October while the observations
 851 were taken at the dates listed.

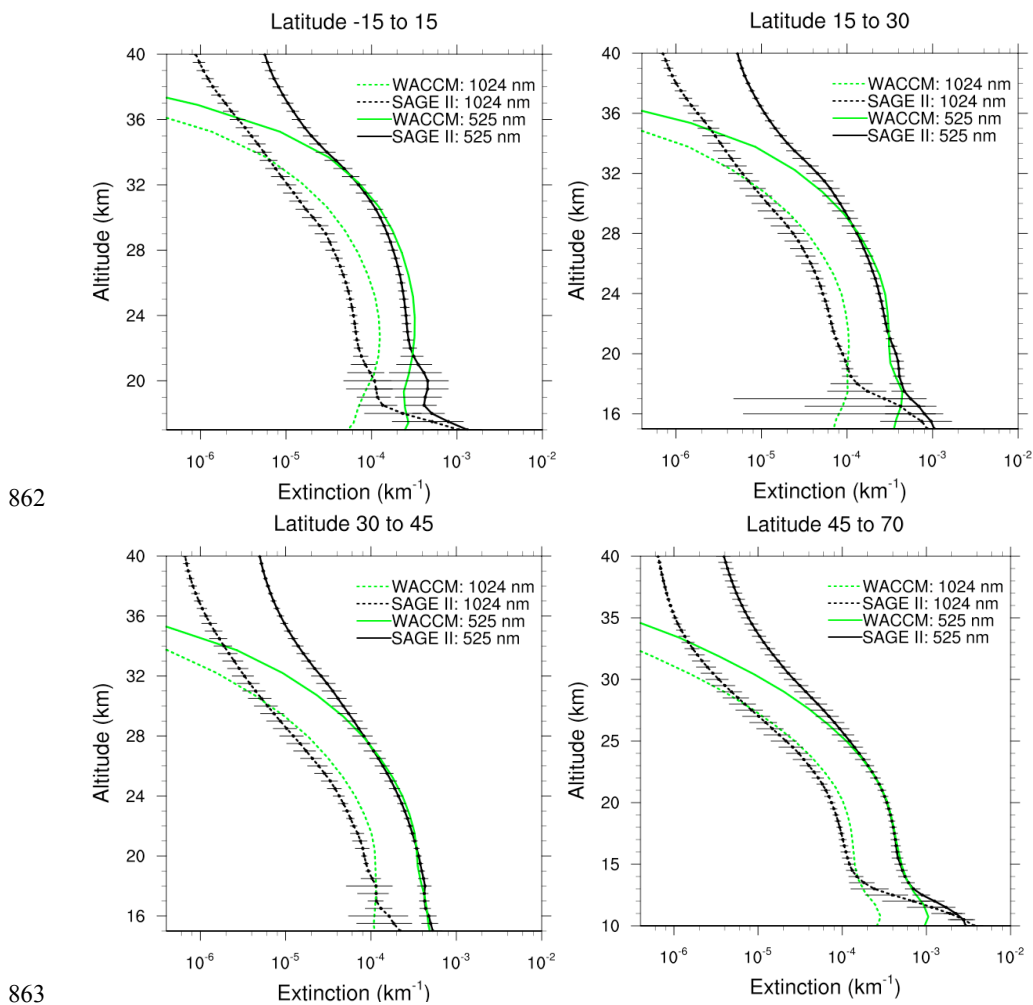


852

853

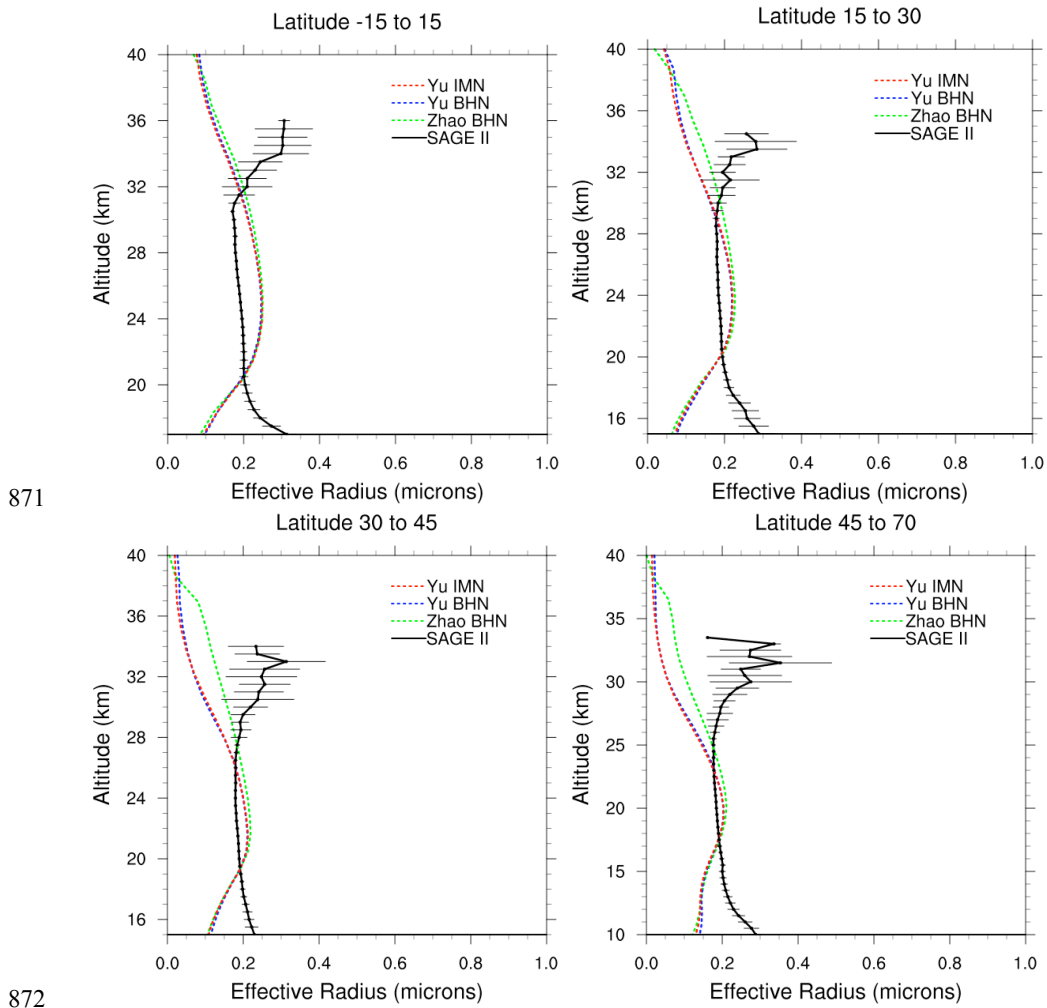
854 Figure 6. (a) Calculated SO₄ mixing ratio; annual and zonal average as a function of atmospheric
 855 pressure and latitude. (b) Calculated N₂O mixing ratio; annual and zonal average as a function
 856 of atmospheric pressure and latitude. (c) Calculated SO₄ mixing ratio versus N₂O compared to a
 857 compilation of aircraft observations (Wilson et al., 2008). Observations are an average of
 858 volcanically quiescent aircraft observations between 1999 and 2004 taken from NASA ER-2,

859 WB-57 and DC-8. The aircraft measurements spanned 5 S to 90 N latitude and approximately
860 40 to 250 hPa. Simulation points are annual zonal average of each grid cell between 40 and 250
861 hPa at the latitudes specified.



863
864
865 Figure 7. Calculated sulfate extinction compared to Stratospheric Aerosol and Gas Experiment
866 (SAGE) II satellite retrievals (Chu et. al., 1989) at two wavelengths (525 nm and 1024 nm) and
867 the latitude regions specified. SAGE data are averaged from years 2000-2005. Simulations are

868 1-year average. Simulated extinction coefficients are calculated as a function of weight percent
869 and wavelength using the refractive indices of Palmer and Williams (1975).
870



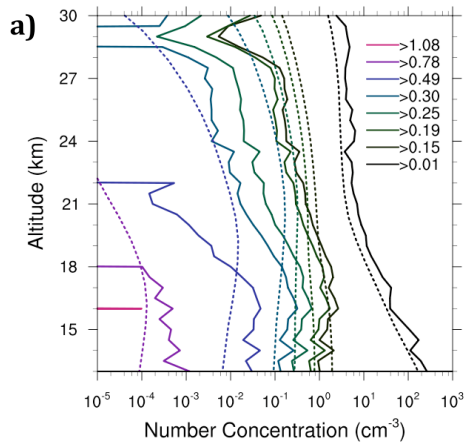
871

872

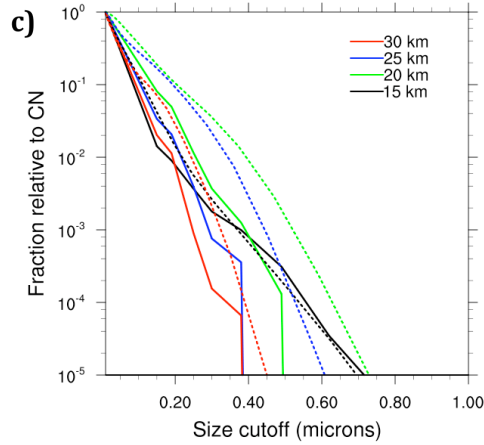
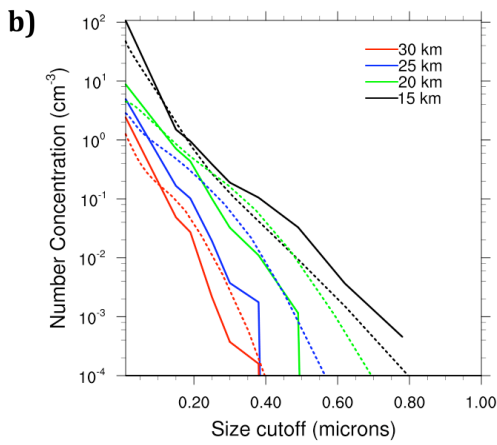
873

874 Figure 8. Calculated sulfate effective radius compared to SAGE II satellite retrievals at the
875 latitude regions specified. SAGE data are averaged from years 2000-2005. Simulations are 1-
876 year average, using wet particle radii for calculations. Solid lines represent standard calculation
877 for effective radius using wet particle radii.

878



879

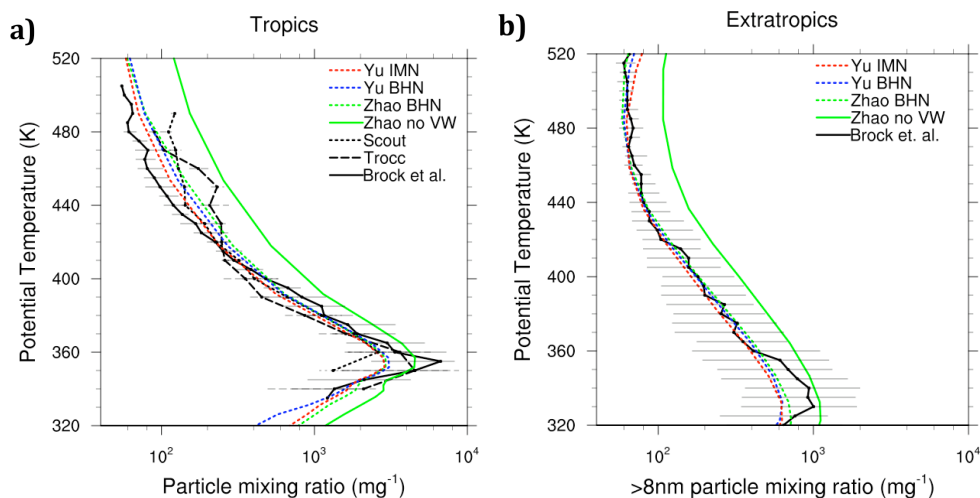


880

881

882 Figure 9. Calculated size distributions for the Zhao BHN simulation (dotted lines) compared to
 883 balloon-borne observations above Wyoming (Deshler et al. 2003) (solid lines) at specified wet
 884 particle radii in microns. Observations are an average of years 2000-2010. Simulations are an
 885 annual average at 41°N and 104°W. (a) Vertical profile of number concentration larger than
 886 specified minimum radii. Calculated number concentrations were calculated using the closest
 887 bin to the specified wet radius. (b) Number concentration larger than a specified wet radius
 888 cutoff at the altitudes specified. (c) Ratio of number concentrations larger than the specified wet
 889 radius to number concentrations of condensation nuclei (>0.01 microns wet radius).

Jay 7/12/11 5:14 PM
 Deleted: at

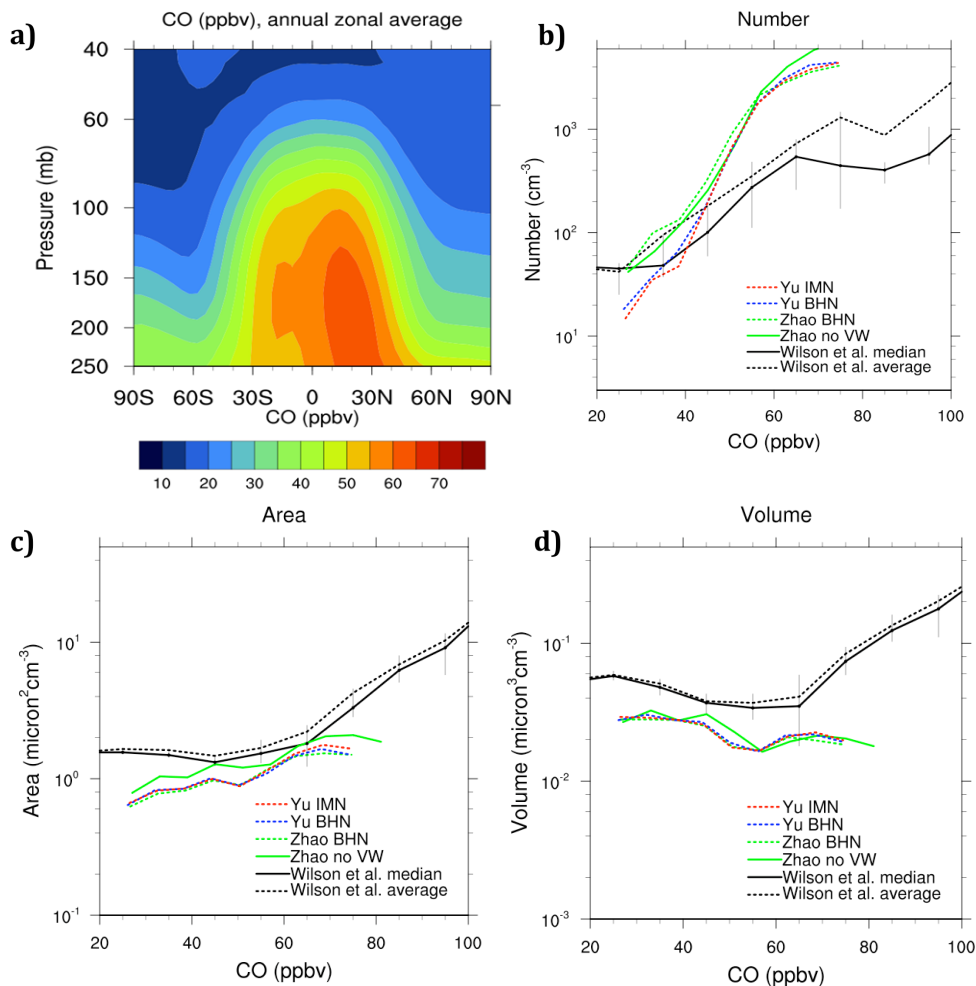


891

892

893 Figure 10. (a) Vertical profile of calculated sulfate particle mixing ratios in the tropics (30°S to
 894 30°N) compared to aircraft observations (Borrmann et al. 2010). (b) Vertical profile of calculated
 895 sulfate particle mixing ratios in the extratropics (average of 30°S to 90°S and 30°N to 90°N)
 896 compared to aircraft observations (Brock et al. 1995). Observations are medians plus/minus
 897 $25^{\text{th}}/75^{\text{th}}$ percentiles. Brock et al. mixing ratios include particles greater than 8 nm diameter and
 898 are based on 5 worldwide aircraft campaigns between 1987 and 1994, with data points attributed
 899 to the eruption of Mt. Pinatubo removed. Scout (SCOUT-O3; Stratospheric-Climatic Links with
 900 Emphasis on the Upper Troposphere and Lower Stratosphere) and Trocc (TROCCINOX;
 901 Tropical Convection, Cirrus, and Nitrogen Oxides Experiment) mixing ratios include particles
 902 greater than 6 nm diameter and are from 2005 aircraft campaigns. It is assumed that no water is
 903 present on the particles. Calculated mixing ratios in both regions include 8.0 nm dry diameter
 904 and larger. The “Zhao no VW” simulation uses Brownian coagulation kernels based on collision
 905 theory, while the other three simulations include the effect of Van der Waals forces on the
 906 collision cross section using the calculations of Chan and Mozurkewich (2001).

907

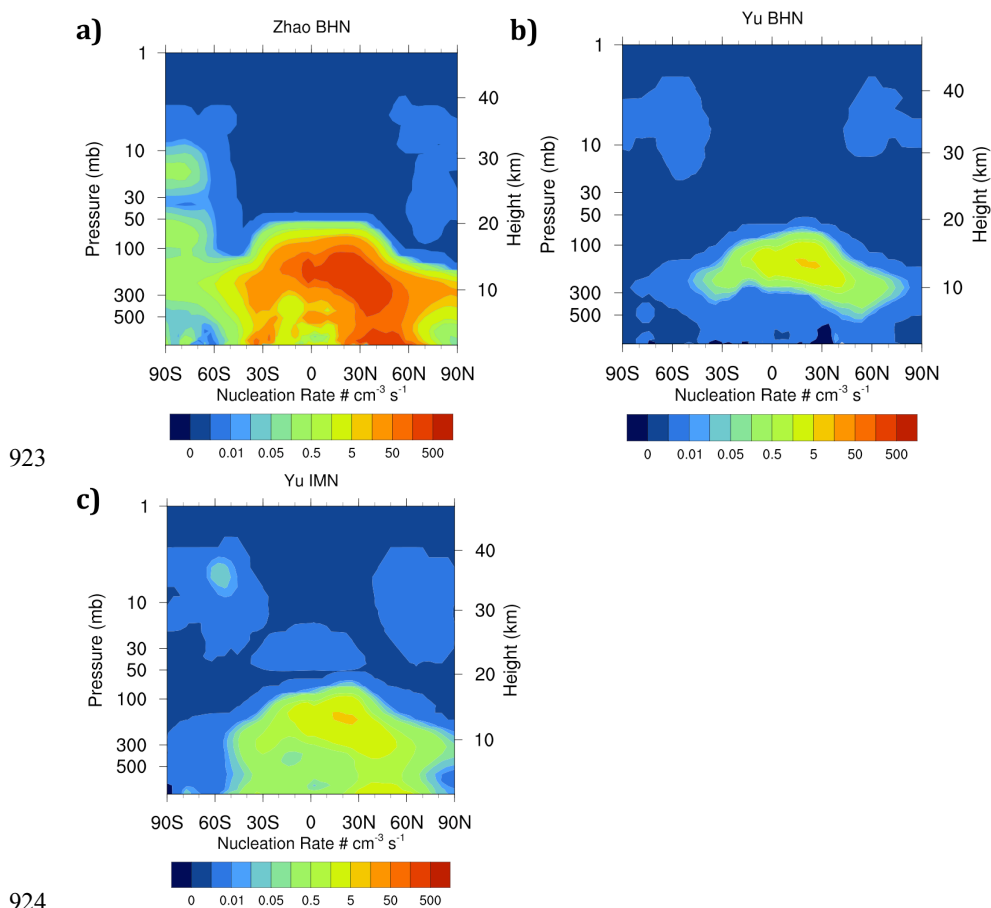


908

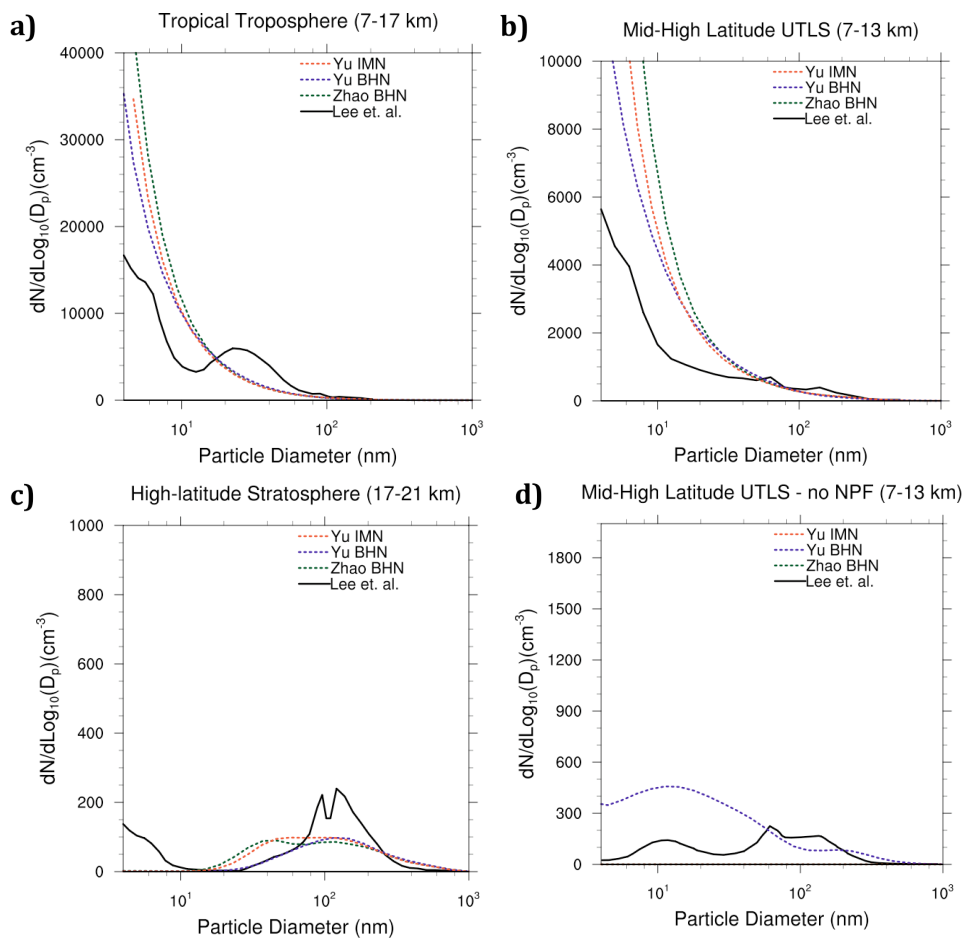
909

910 Figure 11. (a) Calculated CO mixing ratio; annual and zonal average as a function of
 911 atmospheric pressure and latitude. (b) Calculated number, (c) Calculated area, and (d) Calculated
 912 volume versus CO compared to a compilation of aircraft observations provided by J. C. Wilson
 913 and J. M. Reeves. Average (solid) and median (dotted) observations are drawn from Earth
 914 Science Project Office Archives database of 2577 data points from 13 flights between January
 915 2004 and August 2007 spanning 3°S to 21°N, 78°W to 95°W, and 8 to 19 km altitude. The
 916 particle collector has a lower cutoff of 4 nm diameter. It is assumed that no water is present on
 917 the particles when area and volume are computed. Simulations are annual averages spanning

918 3°S to 21°N, 60 to 250 hPa, and all longitudes, binned into increments of 6 ppbv CO.
919 Simulations include particles with 4.0 nm dry diameter and higher. The Zhao no VW simulation
920 does not have the Van der Waals correction, while the three nucleation scheme comparisons
921 (Zhao BHN, Yu BHN, and Yu IMN) include the correction.
922



926 Figure 12. Calculated nucleation rates; annual and zonal average as a function of atmospheric
927 pressure and latitude. (a) Zhao BHN. (b) Yu BHN. (c) Yu IMN.

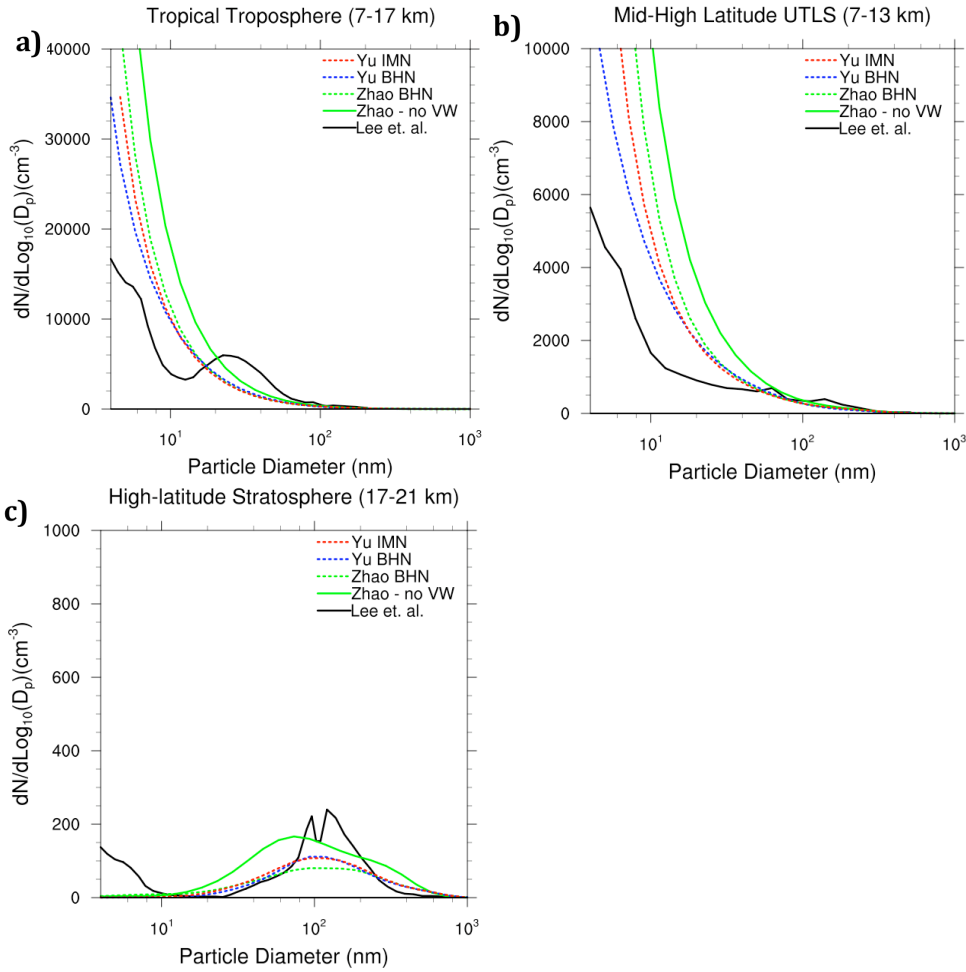


928

929

930 Figure 13. Size distributions, corrected for STP, for the 3 calculated nucleation schemes
 931 compared to observed size distributions from 56 NASA flights between 1998-2000 as reported
 932 by Lee et al. (2003). Size distributions when recent NPF was observed at three different regions:
 933 (a) tropical troposphere, (b) mid-high latitude UTLS, and (c) high-latitude stratosphere.
 934 Observations define recent NPF when two conditions are met: i) number concentrations with
 935 diameter 4-6 nm exceeding that of number concentration with diameter 6-9 nm, and ii) number
 936 concentrations with diameter 4-9 nm exceeding 1 cm⁻³. Simulations were subjected to the same
 937 criteria by analyzing daily averages across the third simulation year for all grid boxes within the
 938 region. Simulation size bin ranges are selected based on the closest bins available to the size

939 specified in Lee et al. (d) Size distributions in the mid-high latitude UTLS when recent NPF
 940 was not observed. Here, only the Yu BHN simulation is plotted because the Zhao BHN and Yu
 941 IMN schemes did not predict any days with no NPF.

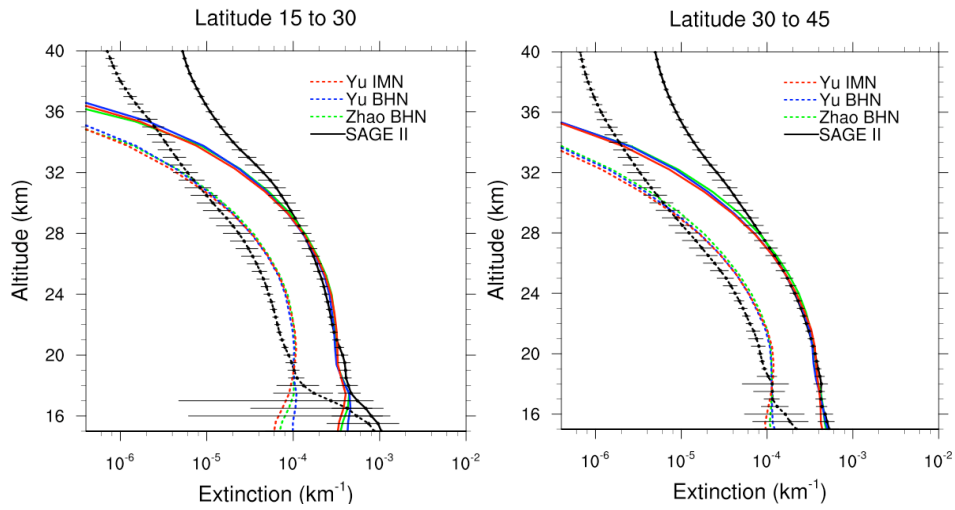


942

943

944 Figure 14. Same as Fig. 13 except all simulation grid boxes are averaged, regardless of whether
 945 they met the criteria for recent NPF.

946



947
 948 Figure 16. Calculated sulfate extinctions for the three simulations compared to SAGE II at two
 949 wavelengths (525 nm; solid lines, and 1024 nm; dotted lines) in the latitude regions specified.

950
 951
 952
 953
 954
 955
 956
 957
 958
 959
 960