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Interactive comment on "The HNO $_3$ forming branch of the HO_2

+NO*reaction* : *pre* – *industrial* – *to* – *presenttrendsinatmosphericspeciesandradiative* forcings"b

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As pointed out correctly in the text, the results of Butkovskaya et al. (2005) indicated that the abundance of water vapour increases the importance of the HNO₃-forming channel in the reaction of HO₂ with NO. However, the assertion that "there is no parameterisation of this effect which would be suitable for incorporation into an atmospheric model" is incorrect. Butkovskaya et al. (2009) presented a new experimental study where the effect of H₂O has been determined at 298 K. A rate constant of $6 \cdot 10^{-13}$ cm³ molecule⁻¹ s⁻¹ was derived for the reaction involving the HO₂·H₂O complex. The complexed fraction of the hydroperoxyl radical HO₂ can be estimated from the esti-

C3392

mated equilibrium constant of $HO_2 + H_2O \leftrightarrow HO_2 \cdot H_2O$: see e.g. Kanno et al. (2005) (and references therein) : $1.57 \cdot 10^{-24} \exp(3775/T)$ molec. cm³. In Tropical regions, near the surface, the complexed fraction reaches about up to about 25%. In these conditions, the total rate of the HNO₃-forming channel is increased from ca. $0.45 \cdot 10^{-13}$ without H₂O-assistance to $1.6 \cdot 1.9 \cdot 10^{-13}$ molec. $^{-1}$ cm³ when H₂O-assistance is taken into account.

We performed 3 simulations with the global chemistry-transport model IMAGESv2 (see e.g. Stavrakou et al., 2009 for a description of the CTM):

- A : without the reaction $\text{HO}_2\text{+}\text{NO}\rightarrow\text{HNO}_3$
- + B : with the reaction HO_2+NO \rightarrow HNO_3, without H_2O-assistance
- C : id. with H_2O -assistance

The impact of the reaction on NOx concentrations is displayed in Fig. 1 below. The calculated effect of the reaction without water-assistance is consistent with the modelling results shown in this study with the Oslo CTM. The effect of water-assistance is large in tropical regions, where water vapour is abundant. The lifetime of methane is increased from 7.7 years in run A to 8.8 in run B and 9.8 years in run C. Water-assistance has also a large impact of surface ozone mixing ratio (up to ca. 30% e.g. over Amazonia).

It is worth noting that the vertically integrated columns of NO₂ (with averaging kernels of satellite instruments such as GOME) are substantially decreased as a result of the reaction (see Fig. 2). Therefore, the top-down estimation of NOx emissions based on satellite (e.g. GOME, OMI etc.) should be greatly affected by the implementation of this reaction. In particular, tropical emissions (especially soil NO emissions) will need to be increased by 10-30%.

References

Butkovskaya, N., M.-T. Rayez, J.-C. Rayez, A. Kukui, and G. Le Bras, Water vapor effect on the HNO_3 yield in the HO_2 + NO reaction: Experimental and theoretical evidence, J. Phys. Chem. A, 113, 11327-11342, 2009.

Kanno, N., K. Tonokura, A. Tezaki, and M. Koshi, Water dependence of the HO_2 self reaction: Kinetics of the HO_2 - H_2O complex, J. Phys. Chem. A, 109, 3153-3158, 2005.

Stavrakou, T., J.-F. Müller, I. De Smedt, M. van Roozendael, G. van der Werf., L. Giglio, and A. Guenther, Evaluating the performance of pyrogenic and biogenic emission inventories against one decade of space-based formaldehyde columns, Atmos. Chem. Phys., 9, 1037-1060, 2009.





Fig. 1. Effect of HNO3-forming channel of HO2+NO reaction on zonally average NOx (in %) without H2O-assistance (left) and with H2O-assistance (right), for the month of July.

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Fig. 2. Effect of HNO3-forming channel of HO2+NO reaction on vertically integrated NO2 column for the month of July. The impact of the averaging kernel of the instrument (GOME) is taken into account.

C3396