

# REVISED

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**Title: Trace gas fluxes of CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O in a permanent grassland soil exposed to elevated CO<sub>2</sub> in the Giessen FACE study**

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Short title: Gaseous emissions under elevated CO<sub>2</sub>

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43 **Abstract**

44 Long-term field observations showed that N<sub>2</sub>O fluxes observed shortly after N application were not  
45 significantly affected by elevated CO<sub>2</sub> in the Giessen Free Air Carbon dioxide Enrichment (FACE) study.  
46 To further investigate this unexpected result a <sup>15</sup>N tracer study was carried out under controlled  
47 conditions where in parallel treatments either the NH<sub>4</sub><sup>+</sup> pool (<sup>15</sup>NH<sub>4</sub>NO<sub>3</sub>) or the NO<sub>3</sub><sup>-</sup> pool (NH<sub>4</sub><sup>15</sup>NO<sub>3</sub>)  
48 was enriched with <sup>15</sup>N. Fluxes of CO<sub>2</sub>, CH<sub>4</sub>, and N<sub>2</sub>O as well as the <sup>15</sup>N enrichment of the N<sub>2</sub>O were  
49 measured. Denitrifying Enzyme Activity (DEA), total denitrification (N<sub>2</sub>+N<sub>2</sub>O) and N<sub>2</sub>-to-N<sub>2</sub>O ratios  
50 were quantified in separate experiments. Over the 57 day incubation, N<sub>2</sub>O fluxes averaged 0.090 ng  
51 N<sub>2</sub>O-N g<sup>-1</sup> h<sup>-1</sup> under ambient and 0.083 ng N<sub>2</sub>O-N g<sup>-1</sup> h<sup>-1</sup> under elevated CO<sub>2</sub> (not significantly  
52 different). The N<sub>2</sub>O production processes were identified by a two-source model. Results showed that  
53 N<sub>2</sub>O must have also been produced by a third source – possibly related to organic N transformation –  
54 which was stimulated by elevated CO<sub>2</sub>. Soil CO<sub>2</sub> fluxes were approximately 20% higher under elevated  
55 CO<sub>2</sub> than soil from ambient but the differences were not significant. CH<sub>4</sub> oxidation rates were on  
56 average -1.75 ng CH<sub>4</sub>-C g<sup>-1</sup> h<sup>-1</sup> in the elevated and -1.17 ng CH<sub>4</sub>-C g<sup>-1</sup> h<sup>-1</sup> in the ambient indicating  
57 that elevated CO<sub>2</sub> increased the CH<sub>4</sub> oxidation by 49% compared to ambient CO<sub>2</sub> under controlled  
58 conditions. N fertilization increased CH<sub>4</sub> oxidation by 3-fold in both CO<sub>2</sub> treatments. CO<sub>2</sub> did not have  
59 any significant effect on DEA while total denitrification and N<sub>2</sub>-to-N<sub>2</sub>O ratios increased by 36 and 33%,  
60 respectively. The results indicate that shortly after N application elevated CO<sub>2</sub> must have stimulated  
61 both the N<sub>2</sub>O production and reduction to N<sub>2</sub> to explain the increased N<sub>2</sub>-to-N<sub>2</sub>O ratio and at the  
62 same time explain the non-responsiveness of the N<sub>2</sub>O emissions. Thus, the observed variation of the  
63 CO<sub>2</sub> effect on N<sub>2</sub>O emissions throughout the year is possibly governed by the dynamics of the N<sub>2</sub>O  
64 reductase activity.

65

## 66 **1 Introduction**

67 The level of earth's atmospheric carbon dioxide (CO<sub>2</sub>) concentration has risen from ~280μL L<sup>-1</sup> at the  
68 start of the industrial revolution to greater than 385μL L<sup>-1</sup> today, and is expected to exceed 700μL L<sup>-1</sup>  
69 by the end of this century (IPCC, 2007). Elevated atmospheric CO<sub>2</sub> increases the plant productivity  
70 and aboveground biomass resulting in a substantial allocation of carbon (C) to belowground that may  
71 lead to a general increase in C inputs in soil. This additional C is likely to fuel belowground microbial  
72 processes and may alter both C and N cycling in soil. Any change in C and N flow and transformation  
73 will affect the soil-atmosphere exchange of biogenic trace gases. The accumulation of greenhouse  
74 gases (GHG) in the atmosphere does alter the earth's radiative balance and is likely responsible for  
75 climate change (Watson et al., 1992; IPCC, 2007; Smith et al., 2010). Although CO<sub>2</sub> is by far the most  
76 abundant greenhouse gas, N<sub>2</sub>O and CH<sub>4</sub> are important atmospheric trace gases because of their  
77 unique radiative properties and their long residence time in the atmosphere resulting in global  
78 warming potential of 296 and 21 times that of CO<sub>2</sub>, respectively (IPCC, 2007). In addition, N<sub>2</sub>O and  
79 CH<sub>4</sub> participate in other atmospheric reactions (e.g. stratospheric ozone depletion) of global  
80 environmental significance. Their concentration in the atmosphere is continuously rising and since  
81 the pre-industrial era it has increased by 15 and 145%, respectively (Watson et al., 1992; Houghton et  
82 al., 1996; IPCC, 2007).

83 Soil plays a major role in the global accounting of C not only due to large amount of C stored in  
84 soil, but also since soil contribution to the annual flux of CO<sub>2</sub> to the atmosphere is 10 times that  
85 contributed by fossil fuel burning (Post et al., 1990). Respiration fluxes of CO<sub>2</sub> in grassland  
86 ecosystems under elevated CO<sub>2</sub> varied from a 10% decline to a 162% increase with a mean response  
87 of 51% increase (Zak et al., 2000). Reich et al. (2001) found a 13% greater CO<sub>2</sub> fluxes per unit mass  
88 under elevated atmospheric CO<sub>2</sub>. Similarly, Smith et al. (2010) reported that soil CO<sub>2</sub> flux in an arable

89 soil was significantly greater under elevated CO<sub>2</sub> being in the range of 15% to 50% compared to  
90 ambient CO<sub>2</sub>.

91 In addition to soil CO<sub>2</sub> flux, elevated atmospheric CO<sub>2</sub> can affect other greenhouse and  
92 reactive trace gases i.e. CH<sub>4</sub> and N<sub>2</sub>O and studies so far provide contradictory results. [Ineson et al.  
93 \(1998\)](#) measured fluxes of N<sub>2</sub>O, CH<sub>4</sub> and CO<sub>2</sub> from soils under ambient and elevated CO<sub>2</sub> at the Swiss  
94 FACE experiment in plots of *Lolium perenne* and reported increased N<sub>2</sub>O emissions by 27% under  
95 elevated CO<sub>2</sub> while ambient plots oxidized consistently more CH<sub>4</sub> than the elevated plots indicating  
96 that elevated CO<sub>2</sub> may result in the inhibition of CH<sub>4</sub> oxidation. [Cheng et al. \(2006\)](#) reported a 58%  
97 increase in CH<sub>4</sub> flux from rice paddies under elevated CO<sub>2</sub>. This increase was attributed to greater  
98 root exudates and numbers of tillers, resulting in more surface area for the release of CH<sub>4</sub> to the  
99 atmosphere ([Ziska et al., 1998](#); [Inubushi et al., 2003](#)). In another study, [Arnon and Bohlen \(1998\)](#) and  
100 [Baggs et al. \(2003a\)](#) reported that both N<sub>2</sub>O and CO<sub>2</sub> fluxes under elevated CO<sub>2</sub> were 2–3 times higher  
101 than those observed under ambient CO<sub>2</sub>. This increase was attributed to increased belowground C  
102 allocation in elevated CO<sub>2</sub> providing energy for denitrifiers or that there is increased O<sub>2</sub> consumption  
103 under elevated CO<sub>2</sub>. However, [Mosier et al. \(2002\)](#) conducted an open-top-chamber CO<sub>2</sub> enrichment  
104 study in the Colorado shortgrass steppe and reported that even though both C<sub>3</sub> and C<sub>4</sub> plant biomass  
105 increased and soil moisture content was typically higher under elevated CO<sub>2</sub>, none of the trace gas  
106 fluxes were significantly altered by CO<sub>2</sub> enrichment over the 43 months period of observation.  
107 Similarly, N<sub>2</sub>O fluxes were not affected by elevated CO<sub>2</sub> in a paddy, arable and grassland fields ([Cheng  
108 et al., 2006](#); [Smith et al., 2010](#); [Dijkstra et al., 2010](#)). However, [Kettunen et al. \(2006\)](#) showed that  
109 elevated CO<sub>2</sub> increased both N<sub>2</sub>O flux from soil and soil water content.

110 A significant increase of N<sub>2</sub>O emissions under elevated atmospheric CO<sub>2</sub> has been observed in  
111 the Giessen FACE study ([Kammann et al., 2008](#)). The more than 9-year data set allowed for the first

112 time the investigation of different time periods throughout the year. Unexpectedly, N<sub>2</sub>O stimulation  
113 by elevated CO<sub>2</sub> in this N limited grassland ecosystem occurred throughout the vegetation period  
114 when mineral N supply was limited, while in the period following N application no significant  
115 difference in N<sub>2</sub>O emissions was detected. Differences in N cycling and/or stimulation of different  
116 microbial groups under elevated CO<sub>2</sub> were made responsible for the observed results. A <sup>15</sup>N tracing  
117 study with soil taken from the Giessen FACE study showed that under elevated CO<sub>2</sub> the turnover of N  
118 changed towards a higher N cycling speed (Müller et al., 2009). To explain the N<sub>2</sub>O response to CO<sub>2</sub> it  
119 is particularly important to study in detail the periods following N fertilizer application because these  
120 are times when high N<sub>2</sub>O emissions occur and when most of the annual N<sub>2</sub>O is produced. Thus the  
121 objective of this study was to determine the extent to which elevated CO<sub>2</sub> concentration may change  
122 soil-atmosphere exchange of GHG (CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O) from grassland soil that had been under the  
123 influence of elevated CO<sub>2</sub> for more than 6 years.

## 124 **2 Material and Methods**

### 125 **2.1 Site description**

126 The grassland site (Environmental Monitoring Climate Impact Research Station) is located 50°32′N  
127 and 8°41.3′E at an elevation of 172 m above sea level near Giessen, Germany. The semi-natural non-  
128 grazed grassland has been managed extensively as a meadow for at least 50 years, fertilized with 40  
129 kg N ha<sup>-1</sup> annum<sup>-1</sup> as calcium ammonium nitrate and mown twice per year. The annual mean  
130 precipitation and temperature (last 35 years) are 644 mm and 9.9°C. The vegetation, an  
131 *Arrhenatheretum elatioris* Br.Bl. *Filipendula ulmaria* sub-community, is dominated by 12 grass  
132 species, 2 legumes and 15 non-leguminous herbs. The soil is classified as a Fluvic Gleysol and has a  
133 sandy clay loam texture over a clay layer, with a mean C and N content of 4.5% and 0.45%,  
134 respectively and a pH of 6.2 (Müller et al., 2009, note the organic C content was not significantly

135 different between the two CO<sub>2</sub> treatments, [see Lenhart \(2008\)](#). In May 1998, the long-term Giessen  
136 FACE system was established ([Jäger et al., 2003](#)).

## 137 **2.2 Soil sampling and experimental set-up**

138 Soil for the experiments reported here was sampled from the top 12 cm of the old grassland soil. The  
139 soil was taken from the ambient and elevated FACE rings where also soil had been sampled for the  
140 <sup>15</sup>N tracing study described by [Müller et al. \(2009\)](#) (see this publication for more details). Fresh soil  
141 was sieved (5 mm) and sub-samples were taken for determining initial gravimetric moisture content  
142 at 105°C for 24 h. The soil was stored for a week at 4°C before the start of the incubation experiment.  
143 A set of twelve jars (Weck®) was arranged according to the treatments: i) two soils i.e. elevated CO<sub>2</sub>  
144 soil and ambient soil; ii) two N sources i.e. <sup>15</sup>NH<sub>4</sub>NO<sub>3</sub> or NH<sub>4</sub><sup>15</sup>NO<sub>3</sub> (60 atom%) with three repetitions  
145 per treatment. Soil portions of 200 g (fresh wt. equivalent) were weighed out and filled into each jar.  
146 The soil was adjusted to a water content of 0.40 g H<sub>2</sub>O g<sup>-1</sup> dry soil with distilled water and incubated  
147 for a week at 20 °C prior to fertilizer application. Both the soils (either from plots under elevated or  
148 ambient CO<sub>2</sub>) were labelled with <sup>15</sup>N at a rate of 100 µg N g<sup>-1</sup> fresh soil in 10.5 mL per jar using a  
149 seven-needle applicator to assure an even distribution of the applied N in soil. The resulting water  
150 content was on average 0.45 g H<sub>2</sub>O g<sup>-1</sup> dry soil. The jars were covered with parafilm that was  
151 perforated with a needle to ensure natural gas exchange and incubated at 20 °C. Samples were  
152 weighed at regular intervals during the incubation; water loss under present experimental set-up was  
153 almost negligible (~0.2 mL).

## 154 **2.3 Gas samplings and measurements**

155 In total, 13 gas samplings were carried out at day 0 (shortly after N application) and 1, 2, 4, 9, 14, 18,  
156 24, 29, 35, 39, 48 and 57 days after N application. Four samplings were carried out (3, 4, 5, 7 days)  
157 before fertilizer application (control). At each sampling time the jars were closed for 0.5 to 2 hours

158 with a glass lid. Gas samples were taken through a septum in the lid with 60 ml disposable syringes at  
159 time zero and at the end of the incubation period. A 12 ml sub sample at the end of the incubation  
160 samples were transferred to evacuated exetainer (Labco, England) for  $^{15}\text{N}$  analysis. Gas samples were  
161 analyzed on a gas chromatograph equipped with ECD ( $\text{N}_2\text{O}$ ,  $\text{CO}_2$ ) and FID ( $\text{O}_2$ ,  $\text{CH}_4$ ) detector by  
162 standard gas chromatographic method (Mosier and Mack, 1980). The gas chromatograph (Shimadzu  
163 14a) was equipped with a  $^{63}\text{Ni}$ -electron capture detector ECD for  $\text{N}_2\text{O}$  and  $\text{CO}_2$  (oven, valve and  
164 detector temperatures were operated at 65, 100 and  $280^\circ\text{C}$ ) and flame ionization detector (FID) for  
165  $\text{O}_2$  and  $\text{CH}_4$  estimation. The  $^{15}\text{N}$  excess in  $\text{N}_2\text{O}$  was determined in separate samples by isotope-ratio  
166 mass-spectrometry (Stevens et al., 1993). The determination of the relative contribution of  
167 denitrification to the overall  $\text{N}_2\text{O}$  flux was calculated based on the Method by Stevens et al (1997).  
168 Briefly, the procedure assumes that  $\text{N}_2\text{O}$  is produced either by nitrification ( $\text{NH}_4^+$  oxidation) and/or  
169 denitrification ( $\text{NO}_3^-$  reduction) using the following equation:  $d = (a_m - a_n) / (a_d - a_n)$  with  $d$ , the fraction  
170 derived from the denitrification pool;  $(1-d)$  = fraction derived from nitrification,  $a_d$ ,  $^{15}\text{N}$  fraction of the  
171  $\text{NO}_3^-$  pool;  $a_n$ ,  $^{15}\text{N}$  fraction of the  $\text{NH}_4^+$  pool,  $a_m$ ,  $^{15}\text{N}$  fraction of the  $\text{N}_2\text{O}$  (mixture).

172 If the calculation results in a negative value then the  $^{15}\text{N}$  abundance of the  $\text{N}_2\text{O}$  must have  
173 been lower than the  $^{15}\text{N}$  abundance from either the  $\text{NO}_3^-$  or the  $\text{NH}_4^+$  pool. Thus providing an  
174 indication that  $\text{N}_2\text{O}$  was produced by a third process that is not associated with the turnover of  $\text{NH}_4^+$   
175 and/or  $\text{NO}_3^-$  (see also, Rütting et al., 2010).

#### 176 **2.4 Denitrification Enzyme activity (DEA)**

177 A set of twelve flasks (Brand) per sampling date (total of 8 sets) was arranged according to the  
178 treatments: i) two soils i.e. elevated  $\text{CO}_2$  soil and ambient soil; ii) two  $\text{C}_2\text{H}_2$  levels ( $-\text{C}_2\text{H}_2$ ;  $+\text{C}_2\text{H}_2$ ) with  
179 three repetitions per treatment. Acetylene was used to inhibit the reduction of  $\text{N}_2\text{O}$  to  $\text{N}_2$  during  
180 denitrification and thus allowing estimation of total denitrification by measurement of the

181 accumulated  $\text{N}_2\text{O}$  (Abbasi and Adams, 2000a). Prior to DEA analysis, twenty grams of soil at a  
182 moisture content of  $0.41 \text{ g H}_2\text{O g}^{-1}$  dry soil was pre-incubated at  $20^\circ\text{C}$  for 7 days after adding  $100 \mu\text{g N}$   
183  $\text{g}^{-1}$  fresh soil (as  $\text{NH}_4\text{NO}_3$ ) following experiment 1. DEA analysis was carried out in 250 mL flasks  
184 (Brand) with a septum fitted in the lid for gas sampling, using an anaerobic slurry technique as  
185 described by Müller et al. (2002). At the start of the assay 50 mL of a nitrate-glucose solution were  
186 applied to each flask resulting in concentrations of  $50 \mu\text{g NO}_3^- \text{N g}^{-1}$  (as  $\text{KNO}_3$ ) and  $300 \mu\text{g C g}^{-1}$  soil (as  
187 glucose). The bottles were immediately closed, evacuated and the headspace flushed (to atmosphere  
188 pressure) with pure  $\text{N}_2$  with a double needle. Each evacuation and/or flushing lasted for 2 min and  
189 the internal atmosphere did not contain detectable oxygen, as confirmed by gas chromatography. In  
190  $\text{C}_2\text{H}_2$  treated flasks, 10% of headspace gas was removed and replaced by adding 10 mL of  $\text{C}_2\text{H}_2$  with a  
191 syringe and internal pressure was equilibrated to atmospheric pressure. The samples were placed at  
192  $20^\circ\text{C}$  on a rotary shaker at 120 rpm for a total of 40 min. The headspace atmosphere was removed  
193 (first sample) with 60 mL gas-tight syringes at 20 min. The extracted gas after the first sample was  
194 replaced by the same amount of  $\text{N}_2$ . Following continuous shaking, a second sample was taken after  
195 40 min. Gas samples were analyzed for  $\text{O}_2$ ,  $\text{CH}_4$ ,  $\text{CO}_2$ , and  $\text{N}_2\text{O}$  on a gas chromatograph (GC) equipped  
196 with an FID and ECD detector (Mosier and Mack, 1980). DEA was calculated as the difference in  $\text{N}_2\text{O}$   
197 concentration increase during a 20 min incubation (40-20 min), accounting for bottles, soil, media  
198 and water volume. The concentrations of the sampling were adjusted for dissolved gas in soil solution  
199 using the Bunsen coefficient (Moraghan and Buresh, 1977).

## 200 **2.5 Statistical analysis**

201 Statistical analysis was carried out with Sigmaplot in combination with Sigmastat (version 3.1, SPSS,  
202 Inc.). During the analysis test for normality and equal variance are carried out before running the  
203 ANOVA and tests to determine significant differences via the Holm-Sidak test.

## 204 **3. Results**

### 205 **3.1 Effect of elevated atmospheric CO<sub>2</sub> on CO<sub>2</sub> emissions**

206 Soil carbon dioxide fluxes before N application were 1.00–1.47  $\mu\text{g CO}_2\text{-C g}^{-1}$  under ambient and 1.13–  
 207 1.42 under elevated CO<sub>2</sub> (Fig. 1). During 7 days samplings (average), the fluxes were 1.22 and 1.27  $\mu\text{g}$   
 208 CO<sub>2</sub>-C g<sup>-1</sup> in ambient and elevated CO<sub>2</sub> soils, respectively showing a non-significant response of  
 209 elevated CO<sub>2</sub>. Application of N fertilizer did not alter the CO<sub>2</sub> fluxes in both the soils: The maximum  
 210 fluxes occurred during the first 14 days and thereafter CO<sub>2</sub> fluxes continuously decreased with  
 211 incubation time. Over 57 days' sampling, CO<sub>2</sub> fluxes were on average 0.77  $\mu\text{g CO}_2\text{-C g}^{-1}$  and 0.93  $\mu\text{g}$   
 212 CO<sub>2</sub>-C g<sup>-1</sup> in ambient and elevated CO<sub>2</sub> soil, respectively indicating approximately 20% higher soil CO<sub>2</sub>  
 213 emissions under elevated CO<sub>2</sub> than soil from ambient CO<sub>2</sub> but the differences were not significant  
 214 ( $p>0.05$ ).

### 215 **3.2 Effect of elevated atmospheric CO<sub>2</sub> on CH<sub>4</sub> fluxes**

216 Net CH<sub>4</sub> oxidation was observed in all samplings before and after N application (Fig. 1). The CH<sub>4</sub>  
 217 oxidation rates before N application were -0.29 to -0.34 ng CH<sub>4</sub>-C g<sup>-1</sup> h<sup>-1</sup> in ambient and -0.46 to -  
 218 0.76 ng CH<sub>4</sub>-C g<sup>-1</sup> h<sup>-1</sup> in elevated CO<sub>2</sub> soil indicating about a 22% higher oxidation rate in soil that had  
 219 been under elevated CO<sub>2</sub>. After N application, the rate of CH<sub>4</sub> oxidation increased from -0.21 to -3.1  
 220 ng CH<sub>4</sub>-C g<sup>-1</sup> h<sup>-1</sup> in ambient and -0.45 to -4.26 ng CH<sub>4</sub>-C g<sup>-1</sup> h<sup>-1</sup> in elevated CO<sub>2</sub>. Maximum oxidation  
 221 rates were observed 1 day after fertilizer application and increased steadily till 18–24 days of  
 222 incubation. During this period the oxidation rates in the ambient control were -1.19 to -3.07 ng CH<sub>4</sub>-  
 223 C g<sup>-1</sup> h<sup>-1</sup> while under elevated CO<sub>2</sub> the rates were -1.79 to -4.18 ng CH<sub>4</sub>-C g<sup>-1</sup> h<sup>-1</sup>. After day 24, the  
 224 oxidation potential of soil decreased consistently to background level till the end of the incubation.  
 225 On average over the incubation time, CH<sub>4</sub> oxidation rates before N application were -0.40 ng CH<sub>4</sub>-C g<sup>-1</sup>  
 226 h<sup>-1</sup> and became -1.46 ng CH<sub>4</sub>-C g<sup>-1</sup> h<sup>-1</sup> after N application indicating a substantial increase in CH<sub>4</sub>

227 oxidation with N fertilization. Average rates over sampling dates revealed that CH<sub>4</sub> oxidation in  
228 elevated CO<sub>2</sub> soil was  $-1.75 \text{ ng CH}_4\text{-C g}^{-1} \text{ h}^{-1}$  while the CH<sub>4</sub> oxidation in the ambient soil was  $-1.17 \text{ ng}$   
229 CH<sub>4</sub>-C  $\text{g}^{-1} \text{ h}^{-1}$  indicating a 49% higher CH<sub>4</sub> oxidation under elevated compared to ambient CO<sub>2</sub>.

### 230 **3.3 Effect of elevated atmospheric CO<sub>2</sub> on N<sub>2</sub>O emissions**

231 In the week before fertilizer N application N<sub>2</sub>O emissions were  $0.019 \text{ ng N}_2\text{O-N g}^{-1} \text{ h}^{-1}$  in the ambient  
232 and  $0.023 \text{ ng N}_2\text{O-N g}^{-1} \text{ h}^{-1}$  in the elevated CO<sub>2</sub> soils (Fig. 2). N<sub>2</sub>O fluxes did not show any consistent  
233 pattern with time. Likewise, N<sub>2</sub>O fluxes did not differ between elevated CO<sub>2</sub> and ambient treatments  
234 and both showed similar fluxes. After N application the flux rates increased substantially and reached  
235  $0.280$  and  $0.240 \text{ ng N}_2\text{O-N g}^{-1} \text{ h}^{-1}$  at day 0. Over the 57 days, N<sub>2</sub>O fluxes averaged  $0.090 \text{ ng N}_2\text{O-N g}^{-1}$   
236  $\text{h}^{-1}$  in ambient and  $0.083 \text{ ng N}_2\text{O-N g}^{-1} \text{ h}^{-1}$  in elevated CO<sub>2</sub> (not significantly different) resulted in a 3-  
237 to 4-fold increase after N application. The highest fluxes of  $0.281$  and  $0.240 \text{ ng N}_2\text{O-N g}^{-1} \text{ h}^{-1}$  were  
238 measured from ambient and elevated CO<sub>2</sub> treatments, respectively just after N application (day 0).  
239 The increase in emissions was short-lived (3-4 days) with fluxes returning to 'background' levels 30  
240 days after N application.

### 241 **3.4 <sup>15</sup>N enrichment of the N<sub>2</sub>O**

242 The <sup>15</sup>N enrichment of the N<sub>2</sub>O in the soil increased one day after N fertilizer application together  
243 with the increase in N<sub>2</sub>O concentrations (Fig. 3). Ten days after fertilizer N application, the  
244 enrichment of the N<sub>2</sub>O was close to the enrichment in the applied N, indicating that the observed N<sub>2</sub>O  
245 originated from the applied fertilizer. Comparing the <sup>15</sup>N enrichments in the N<sub>2</sub>O from the ambient  
246 and elevated CO<sub>2</sub> soils, no significant difference was observed between the two soils labeled either  
247 with NH<sub>4</sub><sup>15</sup>NO<sub>3</sub> or <sup>15</sup>NH<sub>4</sub>NO<sub>3</sub>. The <sup>15</sup>N enrichment of the N<sub>2</sub>O in the treatments where NO<sub>3</sub><sup>-</sup> was  
248 labelled, were relatively higher than the treatment where NH<sub>4</sub><sup>+</sup> was labeled. The contribution of  
249 denitrification for N<sub>2</sub>O production estimated by the 2-pool model of Stevens et al. (1997) indicated on

250 day 1 after  $^{15}\text{N}$  application a contribution of 16 and 32% under ambient and elevated  $\text{CO}_2$   
251 respectively. Negative values after 15 days showed that apart from  $\text{N}_2\text{O}$  contribution related to  $\text{NH}_4^+$   
252 and  $\text{NO}_3^-$  turnover a third process must have been in operation which was responsible for a dilution of  
253 the  $^{15}\text{N}$   $\text{N}_2\text{O}$  abundance below the  $^{15}\text{N}$  abundance of  $\text{NH}_4^+$  and  $\text{NO}_3^-$ .

### 254 **3.5 Denitrification enzyme Activity, total denitrification and ratio of $\text{N}_2$ -to- $\text{N}_2\text{O}$**

255 The measurement of denitrification enzyme activity (DEA) by measuring  $\text{N}_2\text{O}$  emissions during short  
256 incubation periods (anaerobic), total denitrification ( $\text{N}_2\text{O}+\text{N}_2$ ) and  $\text{N}_2/\text{N}_2\text{O}$  ratios was carried-out from  
257 both  $\text{CO}_2$  treatments (Fig. 4). Before N application, one measurement was taken and DEA rates were  
258 0.137 in ambient and 0.172  $\mu\text{g N}_2\text{O-N g}^{-1} \text{ h}^{-1}$  in elevated  $\text{CO}_2$  soil while total denitrification ( $\text{N}_2\text{O}+\text{N}_2$ )  
259 was 0.456 in ambient and 0.514  $\mu\text{g N}_2\text{O-N g}^{-1} \text{ h}^{-1}$  in elevated  $\text{CO}_2$  soil. The  $\text{N}_2/\text{N}_2\text{O}$  ratios were 3.33 for  
260 ambient and 2.99 for elevated  $\text{CO}_2$  treatment. After N application, DEA rates (both  $\text{N}_2\text{O}$  and  $\text{N}_2\text{O}+\text{N}_2$ )  
261 increased in the first two samplings (day 0 and 1) but thereafter the rates continuously declined over  
262 time. DEA rates ( $\text{N}_2\text{O}$  fluxes) in the elevated  $\text{CO}_2$  treatment were on average (20 days incubation) 16  
263 % higher (0.149 vs 0.128  $\mu\text{g N}_2\text{O-N g}^{-1} \text{ h}^{-1}$ ) than  $\text{N}_2\text{O}$  fluxes in the ambient  $\text{CO}_2$  treatment. But the  
264 values of both treatments across different sampling days were not-significantly different. Total  
265 denitrification rates ( $\text{N}_2\text{O}+\text{N}_2$ ) indicated significantly higher fluxes (36%) in elevated  $\text{CO}_2$  treatment  
266 than in ambient  $\text{CO}_2$  ( $P \leq 0.05$ ). Similarly, the  $\text{N}_2$  production was consistently higher under elevated  
267  $\text{CO}_2$  treatment and on average 54% higher than the  $\text{N}_2$  production in the ambient  $\text{CO}_2$  treatment (Fig.  
268 5). The  $\text{N}_2/\text{N}_2\text{O}$  ratio was 1.02 in the ambient and 1.36 in the elevated  $\text{CO}_2$  treatment showing a 33%  
269 higher ratio under elevated  $\text{CO}_2$ . Contribution of  $d$  ( $\text{NO}_3^-$  reduction) to total  $\text{N}_2\text{O}$  production at  
270 ambient and elevated  $\text{CO}_2$  is shown in Fig. 5. Results indicated that shortly after N application  $\text{N}_2\text{O}$   
271 production and reduction to  $\text{N}_2$  substantially increased both in ambient and elevated  $\text{CO}_2$  and the

272 emissions decreased sharply with time. Elevated CO<sub>2</sub> stimulated both the N<sub>2</sub>O production and  
273 reduction to N<sub>2</sub> compared to ambient CO<sub>2</sub>.

## 274 **4 Discussion**

### 275 **4.1. CO<sub>2</sub> production and methane oxidation**

276 Over the 57-day period of observation, CO<sub>2</sub> flux between the ambient and elevated CO<sub>2</sub> treatments  
277 was not significant suggesting that CO<sub>2</sub> flux was not affected by elevated atmospheric CO<sub>2</sub>. This was  
278 unexpected because in the field 25% higher CO<sub>2</sub> fluxes were observed under CO<sub>2</sub> enrichment possibly  
279 caused by the enhanced biomass and root biomass production and general higher activity under  
280 elevated CO<sub>2</sub> (Kammann et al., 2008). In this laboratory study soil from both ambient and elevated  
281 CO<sub>2</sub> treatments was incubated under similar conditions thus, any discrepancy between laboratory  
282 and field studies can be associated with plant effects which was also confirmed by <sup>13</sup>C studies in the  
283 FACE rings (Lenhart, 2008). Higher CO<sub>2</sub> fluxes under elevated CO<sub>2</sub> were also be observed in a number  
284 of other studies (Hungate et al. 1997; Arnone and Bohlen, 1998; Ambus and Robertson, 1999; Reich  
285 et al., 2001; Smith et al., 2010). However, there are also reports showing that ecosystem respiration  
286 (CO<sub>2</sub> flux) was not affected by elevated CO<sub>2</sub> (Ineson et al., 1998; Mosier et al., 2003). Hu et al. (2001)  
287 suggested that in the long term, soil microbial decomposition is slowed under elevated CO<sub>2</sub> because  
288 of N limitation and CO<sub>2</sub> production is either not affected or limited. In our study it was not the N  
289 limiting factor affecting CO<sub>2</sub> production in elevated CO<sub>2</sub> soils but some other unknown control factors  
290 which affected soil respiration.

291 Throughout the course of the experiment, the CH<sub>4</sub> oxidation potential was significantly  
292 greater in the elevated CO<sub>2</sub> (49%) than the ambient CO<sub>2</sub>. These results were in contrast to the earlier  
293 studies where CH<sub>4</sub> consumption i.e. oxidation was lowered by an average of 17µg CH<sub>4</sub>-C m<sup>-2</sup> h<sup>-1</sup>  
294 under elevated CO<sub>2</sub> (Ineson et al., 1998), more oxidation in ambient than elevated CO<sub>2</sub> soil (Mosier et

295 [al., 2003](#)), or no effect of elevated CO<sub>2</sub> on CH<sub>4</sub> oxidation was observed ([Mosier et al., 2002](#); [Smith et](#)  
296 [al., 2010](#)). Most of these studies were conducted under field conditions where two possibilities may  
297 tend to increase CH<sub>4</sub> production and decrease CH<sub>4</sub> oxidation i) increased soil moisture under elevated  
298 CO<sub>2</sub> constrain and slow down the diffusive CH<sub>4</sub> (and O<sub>2</sub>) transport from the atmosphere to the water-  
299 film covered microbial population and therefore inhibit CH<sub>4</sub> oxidation ([Dorr et al., 1993](#)), ii) [increased](#)  
300 [inhibition of CH<sub>4</sub> oxidation under elevated CO<sub>2</sub> or increased CH<sub>4</sub> production due to greater C](#)  
301 [availability in the soil under elevated CO<sub>2</sub>](#) ([van Kessel et al., 2000](#)). In our study similar atmospheric  
302 conditions to both ambient and elevated CO<sub>2</sub> soils in the laboratory were maintained so that  
303 conditions for gas diffusion in the two soils were similar. Thus, the results from the study showed that  
304 the mechanism responsible for inhibiting CH<sub>4</sub> oxidation under elevated CO<sub>2</sub> as observed under field  
305 conditions, were not operative under laboratory conditions.

306 Average net CH<sub>4</sub> oxidation rates after N application were 3-fold compared to fluxes before N  
307 application. These results are in contrast to earlier findings that the application of NH<sub>4</sub><sup>+</sup> reduced CH<sub>4</sub>  
308 oxidation rates almost immediately (forest soils, [Stuedler et al., 1989](#); short-grass steppe, [Mosier et](#)  
309 [al. 1991](#); laboratory incubations, [Hütsch, 1998](#); [Tlustos et al., 1998](#); [Ullah et al., 2008](#)). The reduced  
310 CH<sub>4</sub> oxidation in the field was attributed to suppression in the population growth and lower  
311 abundance of methane oxidizers and to an inhibition of de-novo enzyme synthesis ([Kolb et al., 2005](#)).  
312 The higher oxidation rates by N addition in the present study are difficult to explain but they show  
313 that either the activity of methanotrophic bacteria was enhanced under elevated CO<sub>2</sub> possibly  
314 increased methanogenesis under conditions when mineral N was available ([Kammann et al., 2001](#)).  
315 Therefore, the kinetics of CH<sub>4</sub> oxidation/production is complex and their dependence on soil N status  
316 or moisture are crucial for an accurate prediction of net CH<sub>4</sub> oxidation.

317

## 318 **4.2 N<sub>2</sub>O emissions**

319 The present investigation indicated a substantial increase in N<sub>2</sub>O emissions after N application  
320 in both CO<sub>2</sub> treatments. Furthermore, results indicated that N<sub>2</sub>O emissions in both the treatments  
321 (ambient and elevated) appeared to be limited by available N as fluxes in N fertilized soils increased  
322 3-to 4-fold. Application of fertilizer N had a direct effect on N<sub>2</sub>O production by supplying N for both  
323 nitrification and denitrification (e.g. [Mosier, 1994](#); [Clayton et al., 1997](#); [Abbasi and Adams, 2000b](#))  
324 which may occur simultaneously ([Abbasi and Adams, 2000a,b](#)). Calculations of the contribution of  
325 denitrification on the total N<sub>2</sub>O was carried out according to the method of [Stevens et al. \(1997\)](#).  
326 Results indicated that in the first 8 days of the experiment (when calculated contributions were still  
327 positive) the average contribution of denitrification to N<sub>2</sub>O production was 17% higher under  
328 elevated compared to ambient CO<sub>2</sub> (i.e. ambient CO<sub>2</sub>: 8% and under elevated CO<sub>2</sub>: 25%) ([Fig. 5](#)). This  
329 method is based on the assumption that only nitrification and denitrification contribute to the  
330 observed N<sub>2</sub>O production. However, as shown by [Rütting et al. \(2010\)](#) via a <sup>15</sup>N tracing study in the  
331 New Zealand grassland FACE not only nitrification and denitrification but also heterotrophic  
332 processes, metabolizing organic N, may contribute to N<sub>2</sub>O production. This is also the reason for  
333 negative values found in this study from day 15–20 onwards ([Fig. 5](#)), which occur in situation when  
334 the N<sub>2</sub>O <sup>15</sup>N abundance is below the <sup>15</sup>N abundance of NH<sub>4</sub><sup>+</sup> and NO<sub>3</sub><sup>-</sup>, thus indicating that a third  
335 source at natural or low abundance contributed to the N<sub>2</sub>O emissions. [Rütting et al. \(2010\)](#) showed  
336 that denitrification increased from 4.7% to 8% under elevated CO<sub>2</sub>, a similar trend was observed in  
337 our study.

338 The N<sub>2</sub>O emissions observed before and after N application showed that elevated CO<sub>2</sub> did not  
339 show any significant effect on N<sub>2</sub>O fluxes, rates of fluxes (average) were almost similar. The pre-  
340 existing organic fractions and resulting differences in microbial activity and dynamics under elevated

341 CO<sub>2</sub> treatment (Kammann et al., 2008) could have had an effect on N<sub>2</sub>O production but did not  
342 contribute to higher N<sub>2</sub>O fluxes under elevated CO<sub>2</sub>. This finding is in line with Rütting et al. (2010)  
343 who found no statistical evidence that elevated CO<sub>2</sub> stimulated N<sub>2</sub>O production in grassland soil  
344 exposed to elevated CO<sub>2</sub> for 10 years. Similarly, Barnard et al. (2005) concluded from a review of 20  
345 experiments that field N<sub>2</sub>O fluxes were not altered by elevated CO<sub>2</sub>. However, in several studies it has  
346 been shown that N<sub>2</sub>O flux rates were increased by elevated CO<sub>2</sub>. Ineson et al. (1998) found 27%  
347 higher N<sub>2</sub>O emissions in grassland exposed to elevated CO<sub>2</sub>. Similarly, in perennial grassland N<sub>2</sub>O  
348 fluxes under elevated CO<sub>2</sub> were double than those observed under ambient CO<sub>2</sub> (Arnone and Bohlen,  
349 1998). In contrast, there are also reports that elevated CO<sub>2</sub> either did not alter N<sub>2</sub>O fluxes, or even  
350 reduced N<sub>2</sub>O emissions (Hungate et al., 1997; Mosier et al., 2002, 2003; Welzmler et al., 2008).  
351 Baggs and Blum (2004) reported that the response of N<sub>2</sub>O emissions to elevated CO<sub>2</sub> in grass swards  
352 depend on the rate of N application. The response was non-significant at low rates while N<sub>2</sub>O  
353 emissions significantly increased under elevated CO<sub>2</sub> when high rates of N fertilizer were applied.  
354 Observations in the Giessen FACE study were unexpected because enhanced N<sub>2</sub>O emissions in the  
355 elevated CO<sub>2</sub> treatment (vs ambient) were only observed during times of low N availability (Kammann  
356 et al., 2008). After N fertilizer application N<sub>2</sub>O emissions were not different between ambient and  
357 elevated CO<sub>2</sub>. As highlighted by Kammann et al. (2008) it is important to take the temporal dynamics  
358 of N<sub>2</sub>O emissions into account, which may identify time periods when N<sub>2</sub>O emissions are significantly  
359 higher and those which are not. Thus our results from the laboratory study and from other studies  
360 are only representative for certain time periods and are not representative of general response  
361 patterns.

362 N<sub>2</sub>O production and its concentration in atmosphere are linked to the soil N turnover  
363 (mineralization, nitrification, denitrification) (Müller et al., 2009; Rütting et al., 2010). The net and

364 gross nitrification rates decreased while DEA (which is carried out under non-N limiting conditions)  
365 did not show any significant increase under elevated CO<sub>2</sub> (Kammann et al., 2008). Thus the potential  
366 for net N<sub>2</sub>O production is not affected by elevated CO<sub>2</sub>. Together with the increase in the contribution  
367 of denitrification to the overall N<sub>2</sub>O flux this result indicates that the N<sub>2</sub>O reductase activity must have  
368 been higher under elevated CO<sub>2</sub>. Thus to confirm this, the N<sub>2</sub>-to-N<sub>2</sub>O ratio was determined under  
369 optimum conditions for denitrification.

### 370 **4.3 Total denitrification and N<sub>2</sub>-to-N<sub>2</sub>O ratio**

371 Stimulation of denitrification and N<sub>2</sub>-to-N<sub>2</sub>O ratios was also observed from the soil incubation  
372 studies. However, apart from CO<sub>2</sub> the magnitude of emissions varies depending on type and timing of  
373 inorganic fertilizer application, soil temperature, moisture content, soil type which will vary  
374 throughout the year (Baggs et al., 2003a; Kammann et al., 2008).

375 Denitrification enzyme activity (DEA) was on average 16% higher in the elevated CO<sub>2</sub> than in  
376 the ambient treatment but the difference between the two CO<sub>2</sub> treatments was not significantly  
377 different, suggesting that elevated CO<sub>2</sub> had only a limited effect on the quantity of active denitrifying  
378 enzymes present in the soil. These results were in line with findings of Barnard et al. (2004) who also  
379 reported very little response of DEA to CO<sub>2</sub> treatment in German grassland soils. However, the total  
380 denitrification (N<sub>2</sub>O+N<sub>2</sub>) and the ratio of N<sub>2</sub>-to-N<sub>2</sub>O were significantly higher (36% and 33%) under  
381 elevated CO<sub>2</sub> compared to the ambient treatment. In general the ratio under elevated CO<sub>2</sub> (average  
382 1.358) was similar to the ranges reported by Rolston et al. (1976) (0.1–40) but were lower than the  
383 ratios (345 and 410) reported by Baggs et al. (2003b). Thus, under elevated CO<sub>2</sub> and under non-N  
384 limiting conditions most likely a stimulation of the N<sub>2</sub>O reductase occurred while under N limiting  
385 conditions the higher N<sub>2</sub>O emissions under elevated CO<sub>2</sub> may be related to a higher nirK/nosZ ratio  
386 (nitrite/nitrous oxide reductase) as observed by Regan et al. (2011) for this soil. Thus our studies

387 provide indirect evidence that the kinetics of the reductase systems during denitrification in this  
388 grassland soils are linked to the enhanced C input in connection to N-oxide availability ([Dendooven et](#)  
389 [al., 1994](#)).

390 The maximum ratios in both the treatments were found shortly after N application at day 0 in  
391 contrast to [Baggs et al. \(2003b\)](#) who found very low  $N_2$ -to- $N_2O$  ratios till 8 days after fertilizer  
392 application and proposed different times lag for  $N_2$  and  $N_2O$  production. Our results are in line with  
393 observations by [Welzmilller et al. \(2008\)](#) who also found constantly higher  $N_2$  -to-  $N_2O$  ratios under  
394 elevated  $CO_2$ . The higher  $N_2$ -to- $N_2O$  ratios under elevated  $CO_2$  emphasize the need for the  
395 consideration of  $N_2$  measurements in future denitrification studies and showed that despite a non  
396 significant response to  $N_2O$  total denitrification may be altered. This might be due to a shift of the  
397 denitrifier community under elevated  $CO_2$  which exhibit different of the different reductase dynamics  
398 during denitrification ([Regan et al., 2011](#)).

## 399 **5. Conclusions**

400 Most of the studies conducted so far have suggested higher  $N_2O$  emissions under elevated  $CO_2$  while  
401 very few reported no response. We observed no statistically significant  $CO_2$  enrichment effect on  
402 fluxes of  $CO_2$  and  $N_2O$  in the laboratory study which was carried out under non-N limiting conditions.  
403 However, the relative rate of  $N_2O$  from denitrification and the  $N_2$ -to- $N_2O$  ratio changed under  
404 elevated  $CO_2$ . Thus, elevated  $CO_2$  appears to have an impact on the denitrification kinetics in this  
405 grassland soil which was also confirmed by molecular studies of this soil ([Regan et al., 2011](#)) The  
406 enhanced  $CH_4$  oxidation under elevated  $CO_2$  is surprising and shows that the potential for  $CH_4$   
407 oxidation may increase in this soil. However, this effect was not observed in the field suggesting that  
408 the combination of N application in combination with the environmental regulators (e.g. moisture,  
409 temperature) which were held constant in the current study have an impact under field conditions.

410 The understanding of the stimulation of the microbial populations and activity of methanogenics and  
411 methanotrophic bacteria in response to changing substrate and abiotic factors are essential to predict  
412 the net CH<sub>4</sub> oxidation in terrestrial ecosystem under elevated CO<sub>2</sub>. A mechanistic understanding of  
413 changes in the N cycle and associated GHG production under elevated atmospheric CO<sub>2</sub>  
414 concentrations is essential to predict GHG dynamics under climate change. Therefore, while this  
415 study does not directly contribute to a better understanding of atmospheric processes, it still can  
416 elucidate some of main drivers governing the exchange of GHGs between soil and the atmosphere  
417 which will aid the development of models that are aiming to simulate GHG dynamics.

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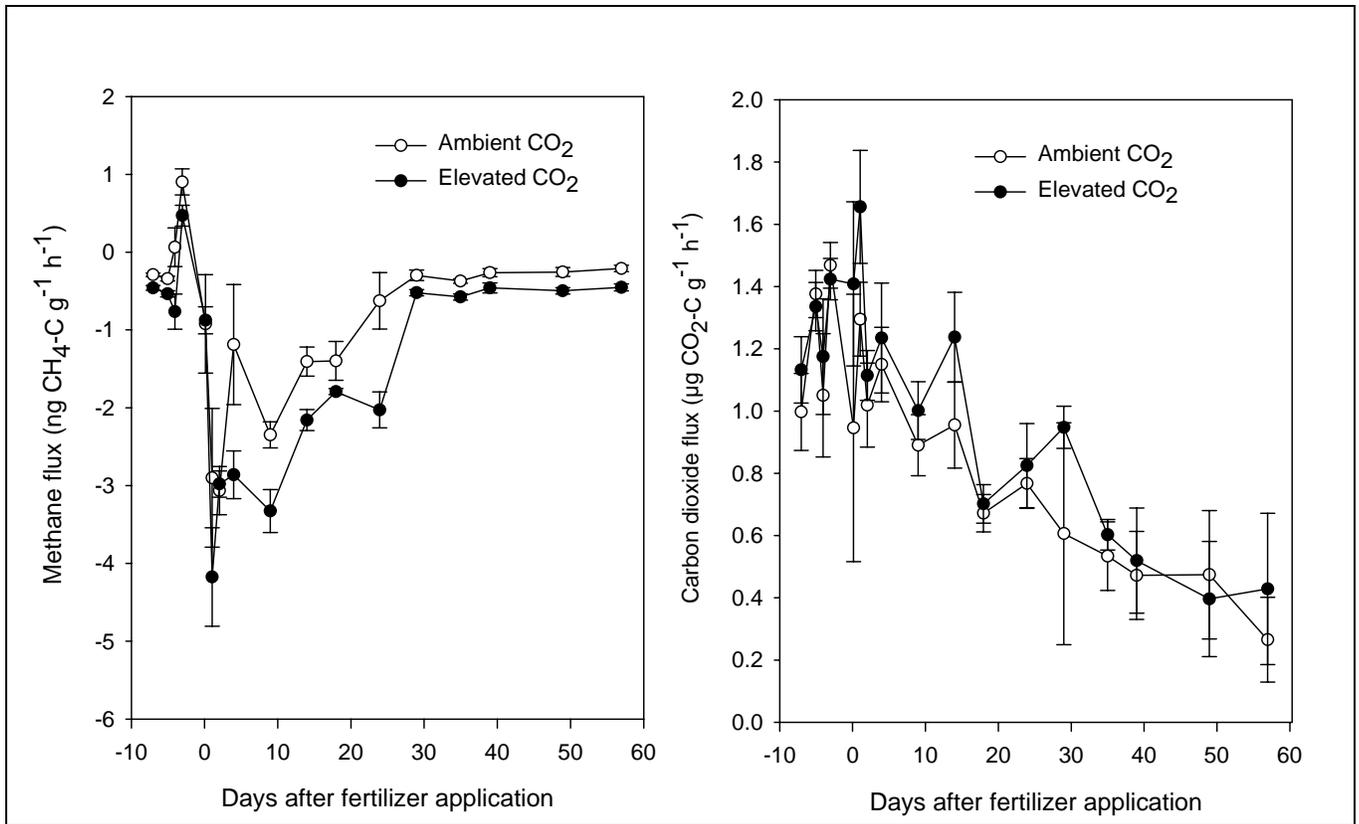
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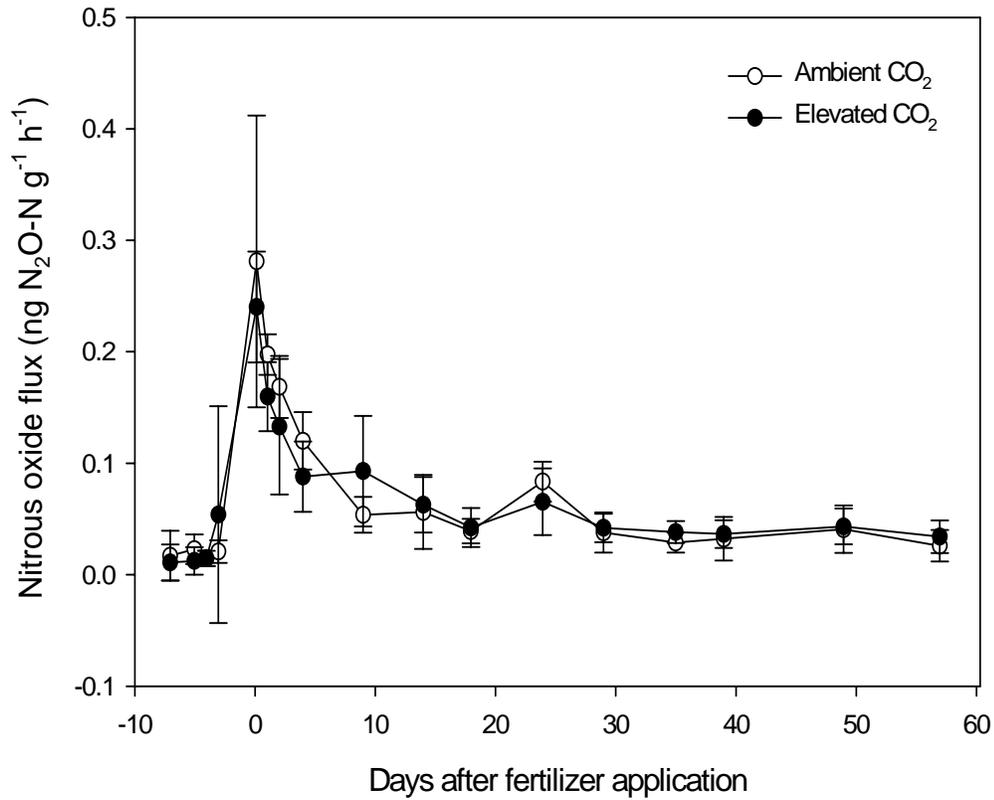
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559 **Fig. 1.** Daily fluxes of CH<sub>4</sub> (ng CH<sub>4</sub>-C g<sup>-1</sup> h<sup>-1</sup>) and CO<sub>2</sub> (µg CO<sub>2</sub>-C g<sup>-1</sup> h<sup>-1</sup>) (Avg. ±SD) from temperate  
 560 grassland soil exposed to elevated CO<sub>2</sub> and soil without elevated CO<sub>2</sub> treatment i.e. ambient  
 561 incubated under controlled laboratory conditions following the application of NH<sub>4</sub><sup>15</sup>NO<sub>3</sub> and  
 562 <sup>15</sup>NH<sub>4</sub>NO<sub>3</sub>

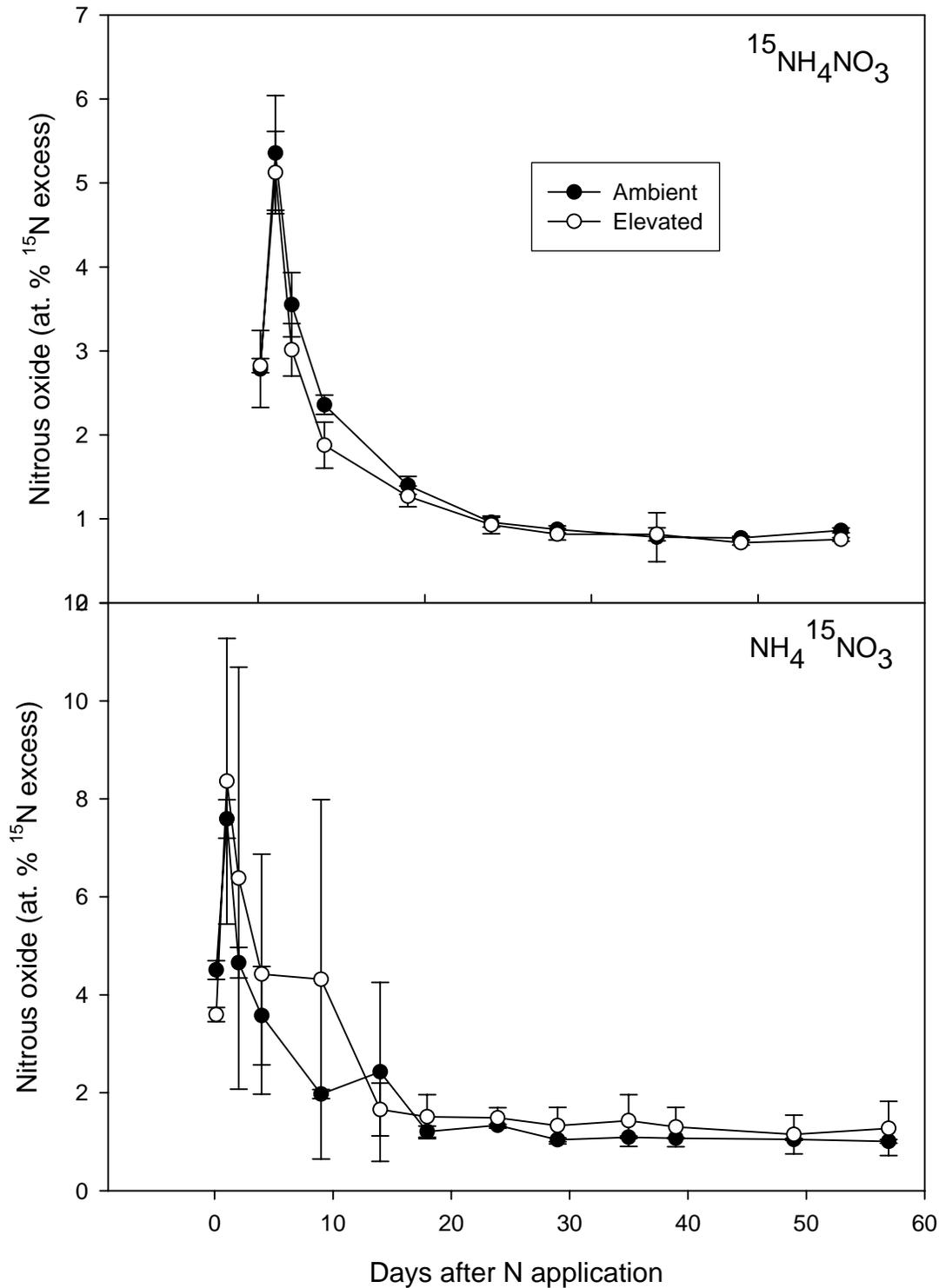
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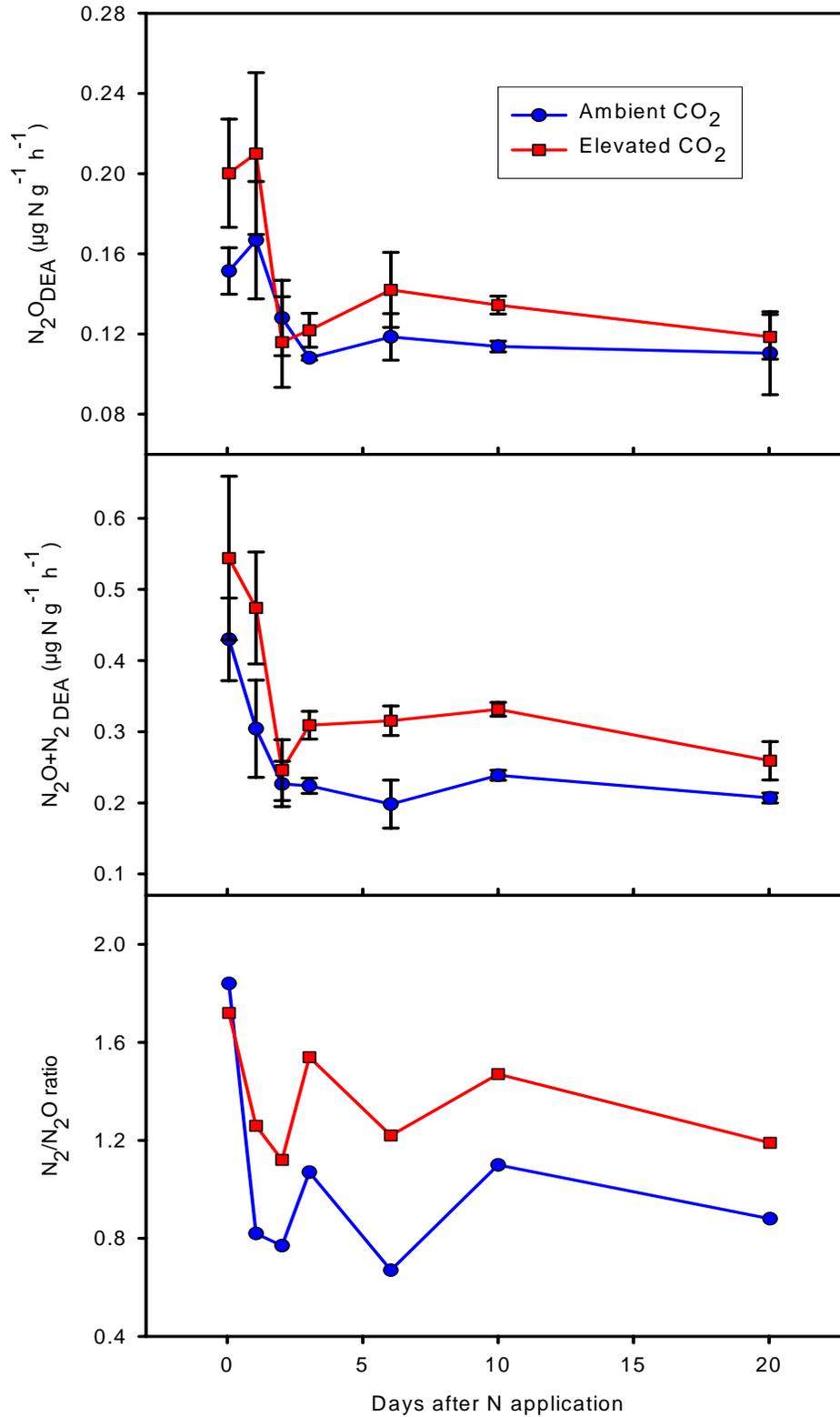
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**Fig. 2.** Daily fluxes of N<sub>2</sub>O (ng N<sub>2</sub>O-N g<sup>-1</sup> h<sup>-1</sup>) (Avg. ±SD) from temperate grassland soil exposed to elevated CO<sub>2</sub> and soil without elevated CO<sub>2</sub> treatment i.e. ambient incubated under controlled laboratory conditions following the application of NH<sub>4</sub><sup>15</sup>NO<sub>3</sub> and <sup>15</sup>NH<sub>4</sub>NO<sub>3</sub>



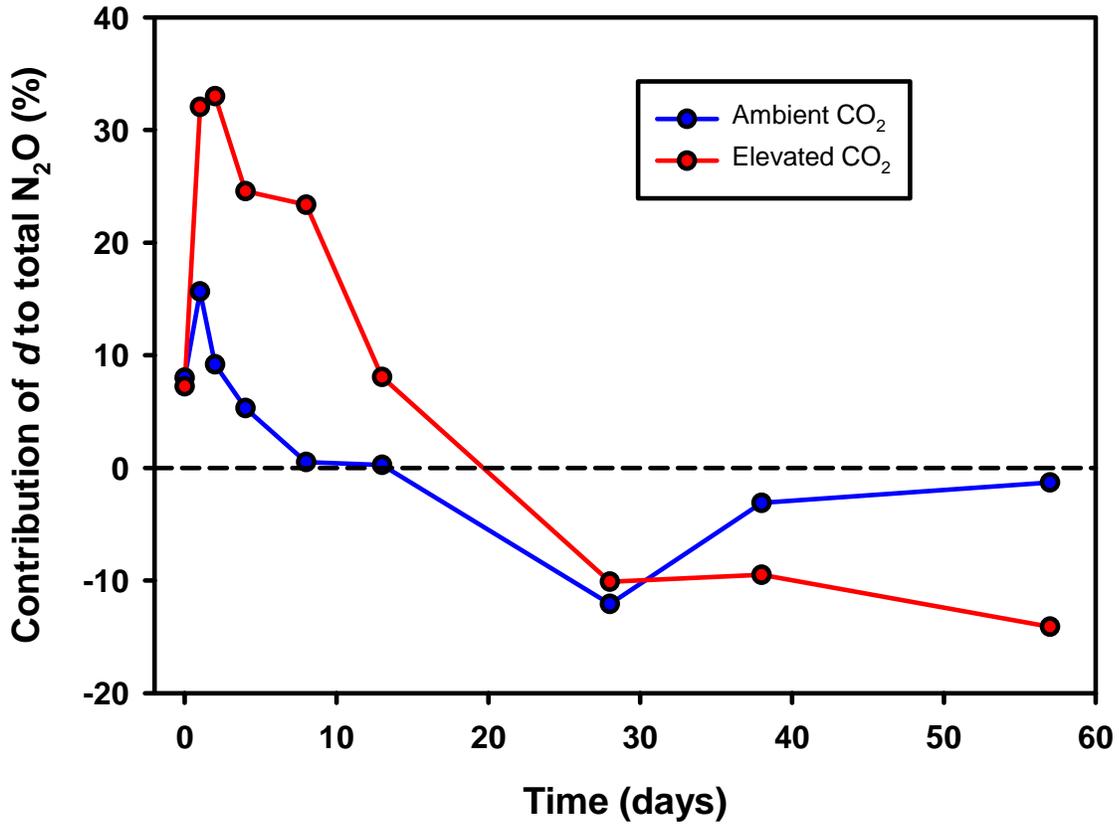
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**Fig. 3.** Nitrous oxide ( $\text{N}_2\text{O}$ ) enrichments (Avg.  $\pm$ SD) in a temperate grassland soil exposed to elevated  $\text{CO}_2$  and soil without elevated  $\text{CO}_2$  treatment i.e. ambient following N fertilizer application where the nitrate pool ( $\text{NH}_4^{15}\text{NO}_3$ ) and the ammonium pool ( $^{15}\text{NH}_4\text{NO}_3$ ) were labelled with  $^{15}\text{N}$  at 60 atom% excess



575  
 576 **Fig. 4.** Emission of N<sub>2</sub>O, total denitrification (N<sub>2</sub>O+N<sub>2</sub>) (μg N g<sup>-1</sup>h<sup>-1</sup>) and N<sub>2</sub>/N<sub>2</sub>O ratio (AVG ±SD) from  
 577 temperate grassland soil exposed to elevated CO<sub>2</sub> and soil without elevated CO<sub>2</sub> treatment i.e.  
 578 ambient incubated under controlled laboratory conditions following the application of NH<sub>4</sub><sup>15</sup>NO<sub>3</sub> and  
 579 <sup>15</sup>NH<sub>4</sub>NO<sub>3</sub>

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**Fig 5.** Contribution of *d* (NO<sub>3</sub><sup>-</sup> reduction) to total N<sub>2</sub>O production in grassland soil at ambient and elevated CO<sub>2</sub>.