



## Abstract

To identify the relative contribution of local versus regional sources of particulate matter (PM) in the Greater Athens Area (GAA), simultaneous mass and chemical composition measurements of size segregated particulate matter (PM: PM<sub>1</sub>, PM<sub>2.5</sub> and PM<sub>10</sub>) were carried out from September 2005 to August 2006 at three locations: one urban (Goudi, Central Athens) and one suburban (Lykovrissi, Athens) in GAA and the third in a regional background site (Finokalia, Crete).

The two stations in GAA exceeded the EU-legislated PM<sub>10</sub> limit values, both in terms of annual average (59.0 and 53.6  $\mu\text{g m}^{-3}$  for Lykovrissi and Goudi, respectively) and of 24-h value, while the concentration levels at the remote site of Finokalia indicated an elevated background. High levels of PM<sub>2.5</sub> and PM<sub>1</sub> were also found at all locations (23.5 and 18.6 for Lykovrissi, while 29.4 and 20.2  $\mu\text{g m}^{-3}$  for Goudi, respectively).

Significant correlations were observed between same PM fractions at both GAA sites indicating important spatial homogeneity within GAA. During the warm season, the PM<sub>1</sub> ratio between the GAA and the background site ranged from 1.1 to 1.3. On the other hand this ratio was significantly higher (1.6–1.7) during the cold season highlighting the role of long-range transport and local sources during the warm and cold seasons respectively. Similar seasonal and geographical patterns were observed for nss-SO<sub>4</sub><sup>2-</sup>, a secondary compound characteristic of regional sources, confirming the above hypothesis.

Regarding the coarse fraction no such seasonal trend was observed for both GAA sites with their ratio (GAA site/Finokalia) being higher than 2 indicating significant contribution from local sources such as road dust and/or constructions as confirmed by Ca<sup>2+</sup> measurements.

Chemical speciation data showed that on a yearly basis, ionic and crustal mass represent up to 78% of the gravimetrically determined mass for PM<sub>10</sub> samples in GAA. The unidentified mass might be attributed to organic carbon (OC) and elemental carbon (EC), in agreement with the results reported by earlier studies in central Athens.

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The contribution of local sources at both GAA sites was also estimated by considering mass and chemical composition measurements at Finokalia as representative of the regional background. Carbonaceous material, Particulate Organic Matter (POM) and EC, seemed to be the main contributor of the local PM mass within GAA (up to 62% in  $PM_{10}$ ). Dust from local sources (mainly resuspension) contributed also significantly to the local  $PM_{10}$  mass (up to 33%).

## 1 Introduction

The interest on aerosols widely increased the last years due to their impact on air quality, human health and climate change. Legislation regarding air pollution based on atmospheric particulate matter (PM) is becoming gradually more stringent, as a result of the high levels of aerosols during intense episodes of either natural or anthropogenic origin. Such episodes could lead to formation and accumulation of aerosol pollutants on regional or even continental scales since they can be associated with synoptic and mesoscale meteorological conditions (Querol et al., 2009).

Long term monitoring of PM concentrations in the Greater Athens Area (GAA; Chaloulakou et al., 2003, 2005; Grivas et al., 2004a, 2008a) registered the occurrence of a significant number of PM exceedances above the limits set by EU legislation and point to the need for abatement strategies. However to propose such strategy precise knowledge of PM sources is a prerequisite. For this the relative contribution of natural and anthropogenic sources and the role of local versus long range sources need to be determined. The EU directive (2008/50/EC) specifically requires information for particles deriving from natural sources for the assessment of PM-related air quality since EU recognizes the weakness of individual countries in reducing PM levels that are maintained by long range transport.

To address these critical issues for the GAA, mass and chemical composition of size segregated aerosols simultaneously collected at 3 locations were analyzed; Two sites in Athens, representing the urban and suburban environment and a natural background

site (Finokalia, Crete) for which previous studies (Mihalopoulos et al., 1997; Gerasopoulos et al., 2007; Koulouri et al., 2008b) documented its ability to represent the Eastern Mediterranean regional background.

## 2 Experimental

### 2.1 Sampling site

Simultaneous  $PM_1$ ,  $PM_{2.5}$  and  $PM_{10}$  sampling was conducted in the GAA, at Lykovrissi and Goudi during the period September 2005–August 2006. Lykovrissi (LYK), is a residential suburban site at the Northern part of GAA, affected by primary emission sources and particle transport from the southern part of the GAA basin. Goudi (GOU) is an urban roadside located in the center of GAA and it is affected by heavy traffic-related activities. Details on the sites characteristics can be found in Grivas et al. (2004b) and Grivas and Chaloulakou (2006).

Size-segregated aerosol samples were also collected at Finokalia (FKL), a regional background site located on Crete Island in the Eastern Mediterranean during the period July 2004–July 2006. FKL is situated 70 km northeast of Heraklion, the site characteristics and the prevailing meteorology can be found in Mihalopoulos et al. (1997), Gerasopoulos et al. (2007) and Theodosi et al. (2010a).

### 2.2 Sampling and analytical techniques

In GAA, simultaneous 24-h  $PM_1$ ,  $PM_{2.5}$ , and  $PM_{10}$  measurements were conducted using Partisol low volume samplers with cyclonic separators for  $PM_1$  at both sites and for  $PM_{2.5}$  at LYK, while Harvard impactors were used for  $PM_{10}$  at both sites and for  $PM_{2.5}$  at GOU. Particle mass was collected on pre-weighted Teflon-coated glass-fiber filters. Measurements were conducted on a regular basis of one sample every-3 days, collecting on average 10 samples per month for both sites and all three size fractions.

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At FKL aerosol samples were collected using a virtual impactor (VI; Loo and Cork, 1988) modified to divide particles into two size fractions: fine (aerodynamic particle diameter  $D_a < 1.3 \mu\text{m}$ , hereafter as  $\text{PM}_{10}$ ) and coarse particles ( $D_a > 1.3 \mu\text{m}$ ). The inlet situated before the VI has a cut-off size of  $10 \mu\text{m}$ . Polytetrafluoroethylene (PTFE) filters (Millipore Fluoropore; pore size  $3.0 \mu\text{m}$ ; diameter  $47 \text{mm}$ ) were used in the VI sampler (henceforth PTFE-VI). More details on sample collection can be found in Koulouri et al. (2008b). In addition, size resolved aerosol samples were also collected using a Small-Deposit-area low-volume-Impactor (SDI) (Maenhaut et al., 1996). The inlet preceding the SDI has a cut-off size of  $10 \mu\text{m}$ . The SDI has 12 collecting stages over the particle size range  $0.041\text{--}10 \mu\text{m}$  with cut-offs at  $0.041, 0.085, 0.138, 0.225, 0.346, 0.585, 0.762, 1.06, 1.66, 2.68, 4.08$  and  $8.39 \mu\text{m}$ . The average sampling time for both VI and SDI was 2 days (ranging from 1 up to 3 days). Both blank and field filter samples were conditioned at constant temperature ( $22 \pm 3^\circ\text{C}$ ) and relative humidity ( $40 \pm 5\%$ ), for at least 24 h prior to weighting, before and after sampling.

In total, 127  $\text{PM}_{10}$ , 127  $\text{PM}_{2.5}$  and 126  $\text{PM}_{10}$  samples were collected at LYK, 114  $\text{PM}_{10}$ , 109  $\text{PM}_{2.5}$  and 107  $\text{PM}_{10}$  samples at GOU and 90 aerosol samples at FKL for both VI and SDI. The filters were analyzed for water-soluble ions and elements. One quarter of each filter was extracted using 10 ml of nanopure water. The solutions obtained were analyzed by ion chromatography (IC) for anions ( $\text{Cl}^-$ ,  $\text{Br}^-$ ,  $\text{NO}_3^-$ ,  $\text{SO}_4^{2-}$ ,  $\text{C}_2\text{O}_4^{2-}$ ,  $\text{MS}^-$ ) and cations ( $\text{Na}^+$ ,  $\text{NH}_4^+$ ,  $\text{K}^+$ ,  $\text{Mg}^{2+}$  and  $\text{Ca}^{2+}$ ). More details on the IC method are given by Bardouki et al. (2003). An acid microwave digestion procedure followed by Inductively Coupled Plasma Mass Spectrometry (ICP-MS, Thermo Electron X Series) was applied to measure PM elemental concentrations (V, Cr, Mn, Fe, Ni, Cu, Cd and Pb; Theodosi et al., 2010b).

## 3 Results and discussion

### 3.1 PM mass concentrations

#### 3.1.1 Levels

The monthly mean observations of  $PM_1$ ,  $PM_{2.5}$  and  $PM_{10}$  at the three stations are shown in Figs. 1a–c. The arithmetic mean and standard deviation of  $PM_1$ ,  $PM_{2.5}$  and  $PM_{10}$  ( $\mu\text{g m}^{-3}$ ) for the samples collected at the three locations are given in Table 1. There was a noteworthy upward trend for  $PM_1$  and  $PM_{2.5}$  levels when moving from natural background to suburban and then to urban sites. On the contrary, for the  $PM_{10}$  fraction no pronounced difference was found between the GAA stations.

Annual mean  $PM_{10}$  values at both GAA sites ( $53.6 \mu\text{g m}^{-3}$  and  $59.0 \mu\text{g m}^{-3}$  at GOU and LYK), exceeded the limit value of  $50 \mu\text{g m}^{-3}$  in 44% and 51% of the samples from GOU and LYK respectively. Even at the background station of FKL, with average  $PM_{10}$  of  $37 \mu\text{g m}^{-3}$ , the  $50 \mu\text{g m}^{-3}$  value was exceeded 6% during the year suggesting elevated background aerosol levels in the area under specific meteorological conditions.

Statistically significant correlations were observed between the GAA stations for all PM fractions ( $\rho = 1.6 \times 10^{-10}$ ,  $6.7 \times 10^{-10}$  and  $2.7 \times 10^{-10}$  for  $PM_1$ ,  $PM_{2.5}$  and  $PM_{10}$  respectively) indicating the existence of significant spatial homogeneity of PM within the GAA.

#### 3.1.2 PM ratios

The variability of the PM fractions was examined at all locations. Strong correlations were observed between  $PM_{2.5}$  and  $PM_{10}$  ( $r = 0.63$ ) for both sites in GAA. The significant percentage of crustal elements content in  $PM_{2.5}$  (see Sect. 3.4) is one for the factors responsible. The  $PM_{2.5}/PM_{10}$  ratios for the suburban, urban and natural background sites were 0.37, 0.56 and 0.58 respectively. The smaller ratio at LYK is relevant to the land use characteristics of the surrounding area (mainly unpaved roads and non

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built land), which favor the suspension of soil particles, as demonstrated by the high nss-Ca<sup>2+</sup> levels (see Sect. 3.2.3). The PM<sub>2.5</sub>/PM<sub>10</sub> ratio observed for central Athens site (GOU) is in agreement with the ratios reported by Chaloulakou et al. (2003, 2005), Sillanpää et al. (2005) and Karageorgos and Rapsomanikis (2007) for measurements conducted at same-type locations.

On the other hand the regression between PM<sub>1</sub> and PM<sub>10</sub> presented an important degree of scattering (Fig. 2). For instance at FKL the relation between PM<sub>1</sub> and PM<sub>10</sub> (Koulouri et al., 2008b) revealed the existence of two data subsets, corresponding to non-dust and dust events. The first had a slope of 0.90 manifesting the dominance of pollution particles, while the poorer correlation ( $r = 0.49$ ) depicted the variability of submicron particle sources in the area. The second demonstrated higher PM<sub>10</sub> levels for the same order of PM<sub>1</sub> values (slope 9.6) as the non-dust case with a significant correlation ( $r = 0.89$ ) related to transported dust.

As in FKL, PM<sub>1</sub>/PM<sub>10</sub> ratios between the two GAA stations revealed also the existence of two data pairs corresponding to dust and non-dust events with PM<sub>1</sub>/PM<sub>10</sub> ratios similar to those observed at FKL (1:1.1 to 1:2.1 for non-dust, and 1:11 to 1:5.8 for dust events, for Gou and LYK respectively). These results provide further support to the conclusion drawn by Gerasopoulos et al. (2007) and Querol et al. (2009) that in the Mediterranean due to high crustal content of PM, PM<sub>1</sub> rather than PM<sub>2.5</sub> monitoring can provide an estimation of the anthropogenic fraction. On that account in the following discussion fine and coarse fractions will correspond to PM<sub>1</sub> and PM<sub>10-1</sub> respectively.

### 3.1.3 Seasonal variation of PM fractions

The monthly variations of PM<sub>1</sub>, PM<sub>2.5</sub> and PM<sub>10</sub> at all sites ( $\mu\text{g m}^{-3}$ ) are displayed in Fig. 1. To better understand the PM variability a brief presentation of the prevailing meteorology in the studied area is presented. On a yearly basis, more than half (66%) of the air masses arriving in the area originate from the North sector, which covers Central and Eastern Europe as well as part of western Turkey. This northern contribution reaches almost 85% during summer. On the other hand, Sahara dust events are very

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frequent in the area during spring and autumn contributing up to 25% of the prevailing air masses.

Very high levels of  $PM_{10}$  and  $PM_{2.5}$  were recorded in February and April at all locations, due to dust transport from North Africa. These events are so intense that they can influence the monthly mean PM levels in such extent that they can almost mask differences between urban and background sites (case of April 2006 for both  $PM_{10}$  and  $PM_{2.5}$ ). Note that the dust events were not visible in  $PM_1$ .

At all locations, the seasonal variation of PM fractions was examined after separation in warm (April to September) and cold season (October to March). During the warm season the  $PM_1$  ratio between the GAA and background site ranged from 1.1 to 1.3, whilst during the cold the ratio was significantly higher from 1.6 to 1.7. More intense anthropogenic activities in GAA during the winter (e.g. heating), associated with a more shallow boundary layer can be accounted for the difference during the cold season. On the other hand during the warm season, traffic related emissions were comparably less intense (this is a stable temporal pattern for the GAA) and the increased atmospheric turbulence favored ventilation of the GAA but also long range conditions (see below Sect. 3.2). An example of regional influence on  $PM_1$  levels recorded at all locations is depicted in Fig. 3.

Regarding the coarse fraction ( $PM_{10-1}$ ) no seasonal trend was observed. Indeed the GOU/FKL ratio varied from 2.3 to 2.4 during the warm and cold season respectively. For LYK the corresponding values are 2.8 and 3.1, respectively. As discussed in Sect. 3.1.2 the higher LYK/FKL ratio is due to the higher coarse fraction at LYK location.

## 3.2 Ionic composition

### 3.2.1 Levels and size segregated distribution of water soluble ions

As in the case of FKL (Koulouri et al., 2008b),  $SO_4^{2-}$  in all PM fractions along with  $NO_3^-$  and  $Ca^{2+}$  mainly in coarse mode, have the higher contribution to the total ionic

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mass. In  $PM_{10}$ ,  $SO_4^{2-}$  and  $NO_3^-$  together account for about two thirds of the total ionic mass at both stations, while for  $PM_{10}$  mass they account for 53 and 56% at LYK and GOU, respectively. The third most dominant ion particularly in  $PM_{10}$ , was  $Ca^{2+}$  with contribution of 30% at LYK and 23% at GOU.

In  $PM_{10}$  and  $PM_{2.5}$  levels of water soluble ions are in good agreement with the values reported in previous works conducted in Athens over shorter period (Sillanpää et al., 2006; Karageorgos and Rapsomanikis, 2007). For  $PM_{10}$  to our knowledge this is the first time that water soluble ions are reported.

The  $PM_{10}/PM_{10}$  ratios for all studied water soluble ions are shown in Fig. 4d. Water-soluble ions ( $Cl^-$ ,  $Mg^{2+}$  and  $Ca^{2+}$ ) were extensively found in the  $PM_{10-1}$  fraction (66–95% for LYK and 61–92% for GOU), as expected on the basis of their source and formation mechanism. Nitrate ( $NO_3^-$ ) presented different behavior at FKL and in GAA. At FKL on average, about 94% of particulate nitrate ( $NO_3^-$ ) was associated with coarse particles, strongly indicating that it was chemically bounded mainly with alkaline ions (Mamane and Gottlieb, 1992; Pakkanen et al., 1999). The most likely formation pathway for particulate nitrate ( $NO_3^-$ ) is the reaction of gaseous nitric acid or some other nitrogen compounds with sea salt and mineral dust particles (Metzger et al., 2006). In GAA, a significant portion of  $NO_3^-$  (about 30%) was found in fine mode, indicating ammonium nitrate formation (see further Sect. 3.2.3 on  $NO_3^-$ ).

The mass concentrations of oxalate ( $C_2O_4^{2-}$ ) and non-sea salt sulfate ( $nss-SO_4^{2-}$ , estimated using  $Na^+$  as a sea salt tracer) were mainly found in the fine mode (70–75% and 72–88%, respectively in the GAA). For  $NH_4^+$  it is interesting to note the lower values observed in  $PM_{10}$  compared to  $PM_{2.5}$  and  $PM_{10}$ . This behavior has been also reported by other authors (Querol et al., 2009) and was due to  $NH_4^+$  volatilisation from  $PM_{10}$  owing to their alkaline character and ammonium nitrate formation. Such behavior is not seen at FKL, where temperatures are higher and observations do not indicate the presence of ammonium nitrate (Mihalopoulos et al., 1997). On the other hand as no intercomparison between the different samplers used was performed, the use of VI

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impactors at FKL could partly explain the difference observed between this site and the GAA ones for the ammonium  $PM_1/PM_{10}$  ratio.

### 3.2.2 Ionic balance

The ionic balance can be employed to determine potentially missing ionic species. For this purpose, ionic balance was calculated both for the fine and  $PM_{10}$  fractions of the aerosol particles for both GAA stations. Plots of total anions equivalents (eq) against total cations equivalents for each size class are presented in Figs. 5a and b. The slope of the regression line for  $PM_{10}$  particles indicated a value smaller than unity (slope = 0.3 to 0.5,  $r = 0.63$  to 0.77), which may be due to the existence of  $CO_3^{2-}$  in this size fraction, which has not been measured using ionic chromatography. In contrast, the slope of the regression line for the fine fraction is higher than unity (slope = 1.20 to 1.26;  $r = 0.90$  to 0.92), which may be attributed to the presence of  $H^+$  (not measured) in the aerosol samples. If this is the case,  $CO_3^{2-}$  is expected to associate with  $Ca^{2+}$  and  $H^+$  with  $SO_4^{2-}$  in the  $PM_{10}$  and fine fractions, respectively (Figs. 5c and d). A statistically significant correlation (slope =  $-0.77$  to  $-0.82$ ;  $r = 0.89$  to 0.96) was found when  $Ca^{2+}$  concentrations were plotted against the anions deficiency (sum cations - sum anions, in equivalents), indicating that  $CO_3^{2-}$  was most probably the missing anion in the coarse aerosol fraction. Similar findings for the ionic balance in coarse particles were reported by Karageorgos and Rapsomanikis (2007) for a site in central Athens. Silanpää et al. (2005) also identified carbonate associated with  $Ca^{2+}$ , as an important constituent of coarse PM in the atmosphere of GAA.

A significant correlation was also found between  $SO_4^{2-}$  and the cation deficiency in the fine fraction, indicating possible association of  $H^+$  with  $SO_4^{2-}$ . Indeed nss- $SO_4^{2-}$  neutralization by  $NH_4^+$  was incomplete throughout the year with  $NH_4^+$  versus nss- $SO_4^{2-}$  slope being below unity (varying from 0.69 to 0.73 during warm and cold periods respectively), indicating that 27–31% of  $SO_4^{2-}$  could be associated with  $H^+$ . Similar

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temporal pattern for  $H^+$  in fine particles was also reported by Siskos et al. (2001) based on one year of measurements in central Athens.

### 3.2.3 Temporal variability of the main ionic species

As mentioned, the three main ions observed in all locations are  $SO_4^{2-}$ ,  $NO_3^-$  and  $Ca^{2+}$ . These ions represent three different major sources categories: secondary particle formation processes ( $SO_4^{2-}$ ), primary anthropogenic sources related to fuel combustion and vehicular circulation ( $NO_x$ ) and finally natural sources producing geological particles (Ca), respectively. In addition these ions followed a very similar spatial distribution pattern in the GAA (Fig. 6). Indeed for the  $PM_1$  fraction, the correlations ( $r$ ) between LYK and GOU were 0.93 for  $SO_4^{2-}$  with slope of 1.1 and 0.81 for  $NO_3^-$  with slope of 1.0. For  $PM_{10}$  the correlations ( $r$ ) were 0.94 for  $SO_4^{2-}$  with slope of 0.9 and 0.81 with slope of 0.9 for  $NO_3^-$ . Whilst for  $Ca^{2+}$  the slope was 0.6 and the correlation was weaker ( $r = 0.54$ ). All the above significant correlations with slopes close to 1, indicate high spatial homogeneity for sulfate, nitrate and in a lesser extend for  $nss-Ca^{2+}$  within GAA. Local sources of crustal particles at LYK could account for slope smaller than 1 observed for  $nss-Ca^{2+}$ .

Below the temporal variability for each of the main ions is presented.

*Non-sea-salt sulfate* ( $nss-SO_4^{2-}$ ):  $nss-SO_4^{2-}$  presented a prominent peak in winter (only in GAA) and summer (in all sites; Fig. 6a). The summer peak could be related to enhanced photochemistry, lack of precipitation, low air mass renovation at regional scale or the increment of the summer mixing layer depth favouring the regional mixing of polluted air masses (Mihalopoulos et al., 2007). The secondary maxima of  $SO_4^{2-}$  concentration commonly recorded during winter could concur with the anticyclonic pollution episodes as also indicated by the high nitrate levels (Querol et al., 2009).

In the case of  $nss-SO_4^{2-}$  there was a clear decreasing gradient from urban to sub-urban and to natural sites ( $5.3$ ,  $5.0$  and  $4.2 \mu g m^{-3}$  in  $PM_1$  for GOU, LYK and FKL, respectively). In addition a clear seasonal variation was observed in the GAA/FKL ratio

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the GAA/FKL ratio was reaching values up to 2. Similar trend also existed for  $\text{NO}_3^-$  in  $\text{PM}_1$  fraction. However the GAA/FKL ratio during winter was much pronounced and reached values as high as 8 that is almost double the factor of 4–5 during the warm season. This observation is in agreement with ammonium nitrate formation in GAA during winter.

*Non-sea-salt calcium* ( $\text{nss-Ca}^{2+}$ ):  $\text{nss-Ca}^{2+}$  is considered as an effective tracer of crustal sources in the area (Sciare et al., 2008; Vrekoussis et al., 2005). Despite the vicinity of FKL to N. Africa,  $\text{nss-Ca}^{2+}$  levels in  $\text{PM}_{10}$  were significantly higher at the GAA sites compared to background one (up to 3.5 times higher; Fig. 6g). Additional sources of  $\text{nss-Ca}^{2+}$  in GAA such as dust resuspension from traffic and/or local activities (case of  $\text{PM}_{10}$  at LYK) can explain this trend. By plotting  $\text{nss-Ca}^{2+}$  as a function of coarse mass for all sites although a significant correlation was obtained from all sites ( $r^2$  of 0.67–0.81), the slopes differ significantly ranging from 0.03 at FKL to 0.12 at LYK and GOU, respectively. Considering that all  $\text{nss-Ca}^{2+}$  at FKL was due to regional dust, the remaining part for the GAA could be explained by “local dust”, most probably resuspension from traffic. Thus local sources in GAA (mainly dust re-suspension from traffic) can account for almost 75% of the observed  $\text{nss-Ca}^{2+}$ . The temporal variation of  $\text{Ca}^{2+}$  concentrations in GAA reveal higher levels during the warm season, when prevailing weather conditions (reduced relative humidity, leading to the drying up of surfaces) favor road dust resuspension (Nicholson and Branson, 1990).

### 3.3 Metals speciation

Trace metals are emitted in the atmosphere from a variety of sources and subsequently can be used as tracers of specific sources such as Earth’s crust, combustion etc. Trace metals were analyzed in  $\text{PM}_{10}$  samples collected at LYK and GOU and well as in the  $\text{PM}_1$  fraction sampled at LYK. The annual average concentrations of all studied metals are presented in Table 2 and Fig. 7a. In general, levels of the studied trace metals in  $\text{PM}_{10}$  are in good agreement with values reported for Athens (Manalis et al., 2005; Karegeorgos and Rapsomanikis, 2007; Karanasiou et al., 2007). Lead concentrations

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have further reduced through the years (2001 to 2006), since non-catalyst equipped vehicles are gradually removed from circulation. For  $PM_{10}$  to our knowledge this is the first time that trace metals are reported. It is therefore interesting to analyze their size distribution.

The ratios of trace metal concentrations in  $PM_{10}$  to those in  $PM_{10}$  for all studied trace metals are shown in Fig. 7c for LYK where they are compared with those reported for FKL (Koulouri et al., 2008b). The majority of the measured trace levels were found in the coarse mode (ratios lower than 0.5) as in the case of the crust originated trace metals (e.g. Fe and Mn). Only Cr, V and Pb were confined in the fine mode indicative of anthropogenic sources. Generally metals distribution in GAA presented similarities that were observed at FKL (Koulouri et al., 2008b).

Figure 7b depicts the ratios of the levels of trace metals in  $PM_{10}$  measured in GAA during this work to those reported at FKL. Based on the GAA/FKL ratio metals can be divided in two categories: Those having a ratio below 2 and those above. Ratio lower than 2 or even closer to 1 is calculated for Fe, Mn and Pb and Cr. Long range transport from arid areas of Sahara represented the main source of Fe, Mn and Cr. Trajectory analysis confirmed our assumption since the above trace metals presented the highest values in air masses originating from that area. Pb is also associated with transport from the Southern sector (N. African countries) as they also used it as additive in gasoline.

On the other hand V, Ni, Cd and especially Cu appear to have a local origin since the GAA/FKL ratio is higher to 2. Stationary combustion of fossil fuels for V, Ni and Cd and cars vehicular circulation for Cu (Weckwerth, 2001) are considered as the main sources of the above metals. This was also confirmed by the lack of correlation between these elements within GAA indicating significant contribution from local rather regional sources.

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### 3.4 Chemical mass closure

Chemical composition was defined by three main classes: Dust, Ionic Mass (IM) and “unidentified”, the later accounting mainly for elemental (EC) and organic carbon (OC). IM is the sum of the anions and cations measured. Dust levels have been calculated using Fe or Mn as indicator of crustal material assuming an upper crust relative ratio as described by Guieu et. al., 2002 and Wedepohl, 1995. Both estimations gave comparable results for dust within 20%. Figure 8 presents monthly source apportionment analysis for PM<sub>10</sub> fraction for both GAA sites (IM and Dust) in comparison to the PM<sub>10</sub> mass fraction.

In total crustal elements can account for 36–46% of total mass (depending on the element used and location) in both GAA sites with maximum during the transition period (spring and autumn). This high percentage is related to the frequent occurrence of Saharan dust transport to the eastern Mediterranean during transition season (Kalivitis et al., 2007 and references therein) and to dust resuspension due to vehicle circulation. IM, mainly secondary aerosol contribution (nss-SO<sub>4</sub><sup>2-</sup>, NO<sub>3</sub><sup>-</sup> and NH<sub>4</sub><sup>+</sup>) showed lower contributions that minimized in winter and sharply increased from winter to summer. On a yearly basis, ionic composition, account for about 32% of the total mass at both GAA sites. On average ionic and crustal mass represent 77% and 78% of the gravimetrically determined mass for PM<sub>10</sub> samples in LYK and GOU respectively. The unidentified mass might be attributed to aerosol water and OC, EC which was not measured and accounted for the remaining of the PM<sub>10</sub> mass. This percentage is in agreement with results reported by earlier studies in central Athens. Prosmittis et al. (2004) have accounted the contribution of elemental and organic carbon to PM<sub>10</sub> at 43% at Goudi. Sillanpää et al. (2005, 2006) estimated POM contribution to PM at about 35 and 25% for PM<sub>2.5</sub> and PM<sub>10-2.5</sub> respectively. Grivas et al. (2008b) calculated the contribution of the sum of POM and EC to PM<sub>2.5</sub> at 31% for central Athens.

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### 3.4.1 Local versus regional contribution to PM mass

By considering mass and chemical composition measurements at FKL as representative of the regional background, the contribution of local sources at both GAA sites can be estimated for both  $PM_1$  and  $PM_{10}$  fractions. The results are presented in Fig. 9. In this figure two columns are presented for each GAA site for both  $PM_1$  and  $PM_{10}$  fractions. The first column corresponds to PM regional background (FKL) and the second to the local sources ( $PM_{GAA} - PM_{FKL}$ ). As expected dust from local sources (mainly re-suspension) contributed significantly to the local  $PM_{10}$  mass (up to 33% of PM mass). The contribution of IM ranged from around 20% in local  $PM_{10}$  to less than 40% in local  $PM_1$ ). The reduced contribution of IM was somehow expected given the role of long range transport on  $SO_4^{2-}$  and in lesser extent on  $NO_3^-$  levels in the area. As mentioned before these anions contribute significantly to the measured IM mass.

Carbonaceous material (POM and EC) seemed to be the main contributor of the missing local PM mass (up to 62% in  $PM_1$ ). Simultaneous OC and EC measurements performed during winter and summer of 2008 in various locations around GAA and at FKL as background site confirmed the above conclusion (Mihalopoulos, unpublished data, 2011).

## 4 Conclusions

The chemical composition of size-segregated aerosols ( $PM_1$ ,  $PM_{2.5}$  and  $PM_{10}$ ) was determined at three sites: natural background FKL, suburban LYK and urban GOU from September 2005 to August 2006 in order to identify the major factors controlling levels and chemical composition of aerosols in the three regions and to evaluate the role of local versus regional sources.

The EU annual limit value of  $40 \mu g m^{-3}$  for  $PM_{10}$  was exceeded on yearly basis at both urban and suburban sites and frequent exceedances of the 24-h limit value of  $50 \mu g m^{-3}$  were recorded at all locations.

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Simultaneous measurements of  $PM_1$  and  $PM_{10}$  can highlight natural contributions, since  $PM_1$  is closely related to anthropogenic aerosol and thus better represent the anthropogenic particle fraction. For air quality monitoring policy our results imply that  $PM_1$  would be a better indicator for fine/anthropogenic aerosols and should be also continuous monitored.

During the warm season the no significant difference in  $PM_1$  and sulfur between urban and natural locations, highlighting the role of long-range transport. On the other hand local anthropogenic sources dominated during the cold season.

Regarding the coarse fraction a significant contribution from soil was found in urban locations all over the year. The difference in the slope of  $Ca^{2+}$  versus PM between rural and urban locations indicated that about 1/3 is of natural origin suggesting traffic-related aerosol sources at both sites in Athens in addition to the regional background of FKL.

Chemical speciation data showed that PM in GAA was characterized by relatively constant contribution of ionic mass, 32% of the  $PM_{10}$  mass, with  $SO_4^{2-}$  and  $NO_3^-$  as the dominant ionic species. Crustal material was accounting for almost half of the mass contribution (46%), with maximum during the transition period (spring and autumn). On a yearly basis, ionic and crustal mass represent 77% and 78% of the gravimetrically determined mass for  $PM_{10}$  samples in LYK and GOU, respectively. The unidentified mass might be attributed to OC and EC, in agreement with the results reported by earlier studies in central Athens.

The contribution of local sources at both GAA sites was also estimated by considering mass and chemical composition measurements at FKL as representative of the regional background. Carbonaceous material (POM and EC) seemed to be the main contributor of the local PM mass (up to 62% in  $PM_1$ ). Dust from local sources (mainly resuspension) contributed significantly to the local  $PM_{10}$  mass (up to 33%). The contribution of local IM ranged from around 20% in  $PM_{10}$  to less than 40% in  $PM_1$  given the role of regional sources on the measured  $SO_4^{2-}$  and  $NO_3^-$  levels.

*Acknowledgements.* This work presented results from the research project PYTHAGORAS I & II. The project was co-funded by the European Social Fund (75%) and National Resources (25%). CT, PZ and NM acknowledge also support by the EU project CITYZEN.

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**Table 1.** Annual mean concentrations of PM fractions ( $\mu\text{g m}^{-3}$ ) measured in Lykovrissi, Goudi and Finokalia from September 2005 to August 2006 and basic descriptive statistics.

Particulate Matter ( $\mu\text{g m}^{-3}$ )		Average	Stdev	Median	Min	Max
This Study, Lykovrissi	PM <sub>1</sub>	18.6	9.1	16.9	3.1	58.9
	PM <sub>2.5</sub>	23.5	10.8	21.8	4.6	71.2
	PM <sub>10</sub>	59.0	28.4	51.8	11.3	197.3
This Study, Goudi	PM <sub>1</sub>	20.2	7.4	18.7	8.2	43.8
	PM <sub>2.5</sub>	29.4	10.3	28.4	11.5	67.7
	PM <sub>10</sub>	53.6	29.6	47.7	18.7	299.9
Koulouri et al. (2008a, b), Finokalia	PM <sub>1</sub>	10.1	5.0	9.8	2.7	27.8
	PM <sub>2.5</sub>	18.2	16.2	14.9	4.1	124.5
	PM <sub>10</sub>	37.0	54.2	24.3	13.7	307.5

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**Table 2.** Annual mean concentrations of metals ( $\mu\text{g m}^{-3}$ ) measured at Lykovrissi ( $\text{PM}_1$  and  $\text{PM}_{10}$ ) and Goudi ( $\text{PM}_{10}$ ) from September 2005 to August 2006.

PM ( $\mu\text{g m}^{-3}$ )	Lykovrissi		Goudi
	$\text{PM}_1$	$\text{PM}_{10}$	$\text{PM}_{10}$
V	0.025	0.025	0.024
Cr	0.010	0.010	0.011
Fe	0.164	1.304	1.024
Mn	0.006	0.020	0.019
Ni	0.001	0.011	0.011
Cu	0.010	0.032	0.041
Cd	0.001	0.001	0.001
Pb	0.015	0.023	0.016

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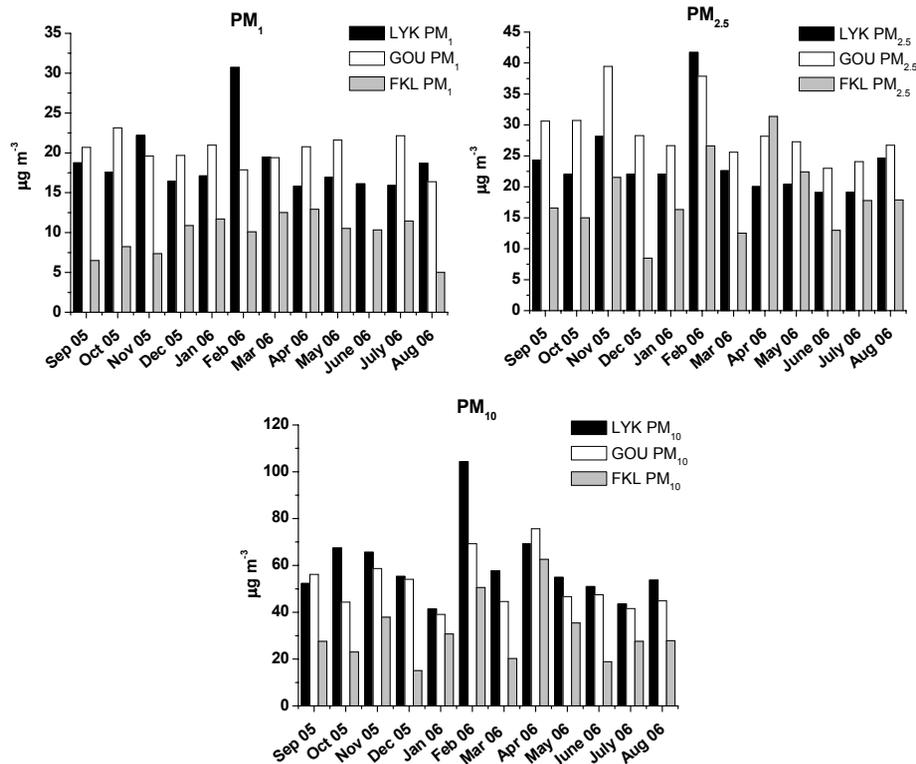
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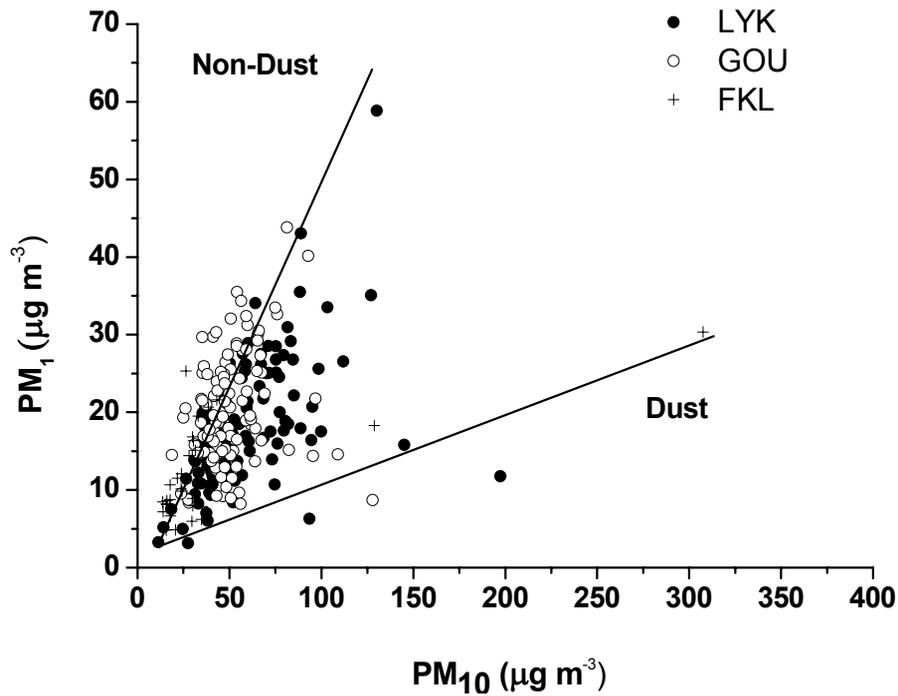


**Fig. 1.** Temporal variability of size segregated concentrations ( $\mu\text{g m}^{-3}$ ) (a)  $\text{PM}_{10}$ , (b)  $\text{PM}_{2.5}$  and (c)  $\text{PM}_{10}$  for all three stations: LYK, GOU and FKL.

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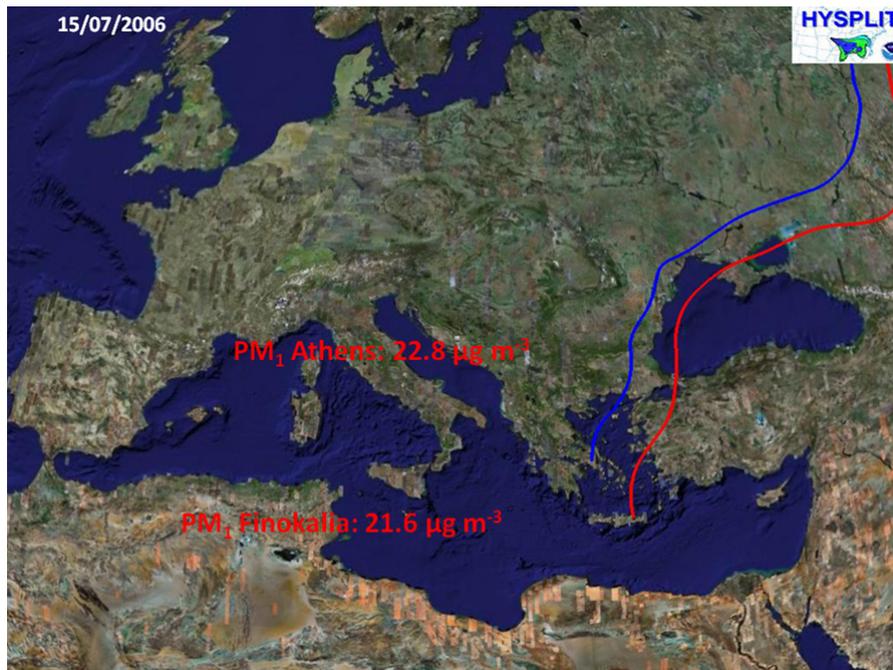
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**Fig. 2.**  $PM_1/PM_{10}$  ratios for all three stations during dust and non-dust events.

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**Fig. 3.** A characteristic long range transport influence on  $PM_{10}$  levels in the studied sites.

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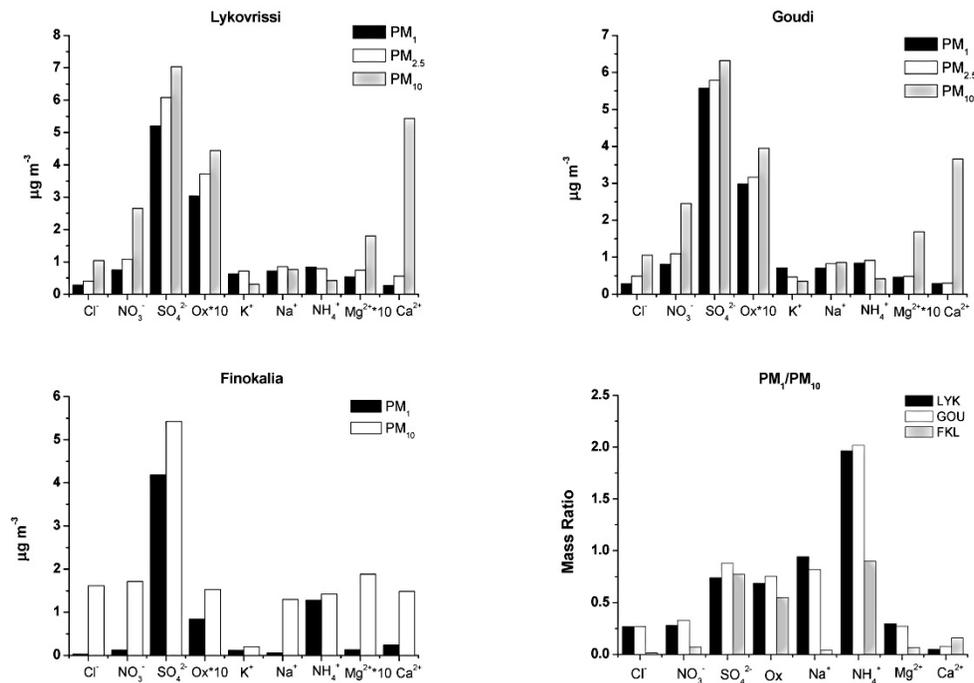
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**Fig. 4.** Annual average concentrations ( $\mu\text{g m}^{-3}$ ) of various ions for all three stations **(a)** LYK, **(b)** GOU, **(c)** FKL and **(d)** their  $\text{PM}_1/\text{PM}_{10}$  mass ratios.

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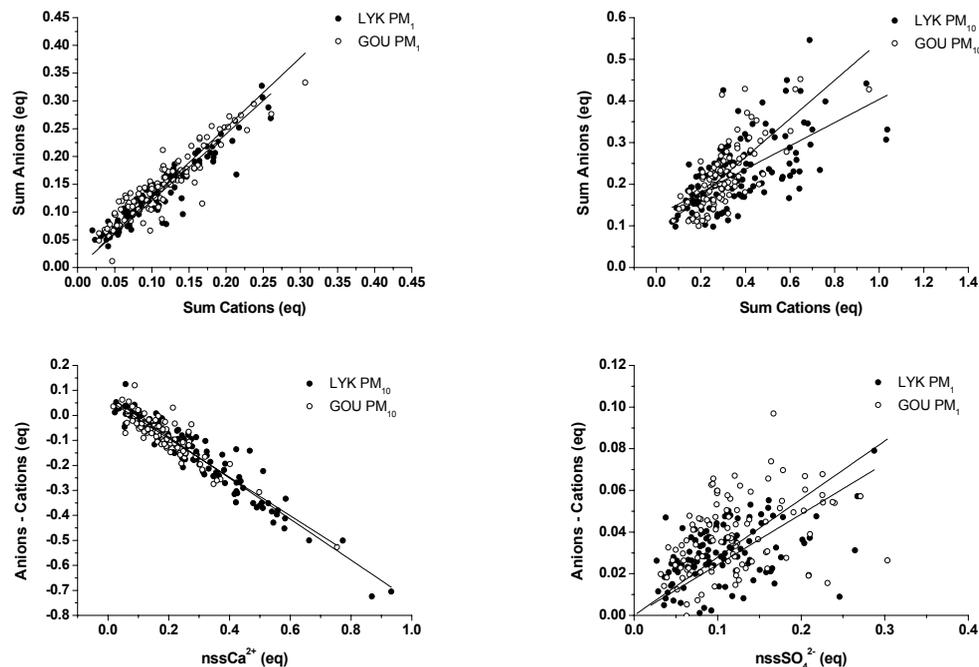
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**Fig. 5.** Ionic balance (equivalents) for both GAA stations for **(a)** fine and **(b)** PM<sub>10</sub> fractions. Anions deficiency versus nssCa<sup>2+</sup> in PM<sub>10</sub> samples **(c)** and nssSO<sub>4</sub><sup>2-</sup> in PM<sub>1</sub> samples **(d)**.

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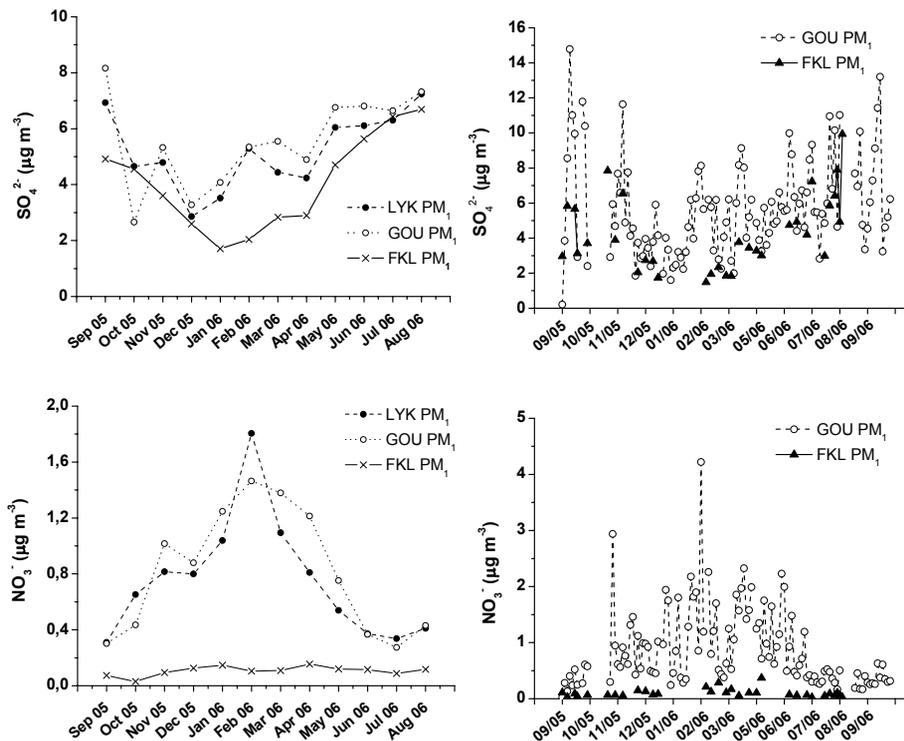
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**Fig. 6.** Seasonal (a, c, e and g) and daily (b, d, f and h) variations ( $\mu\text{g m}^{-3}$ ) for  $\text{SO}_4^{2-}$ ,  $\text{NO}_3^-$  and  $\text{Ca}^{2+}$  in  $\text{PM}_{10}$  and for  $\text{NO}_3^-$  in  $\text{PM}_{10}$  samples.

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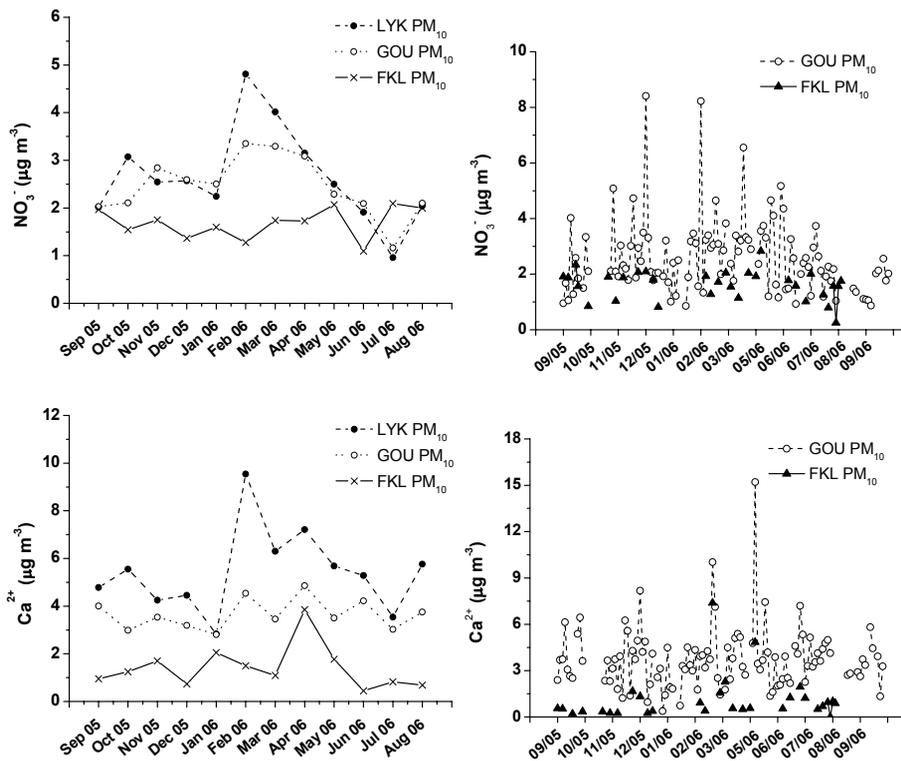


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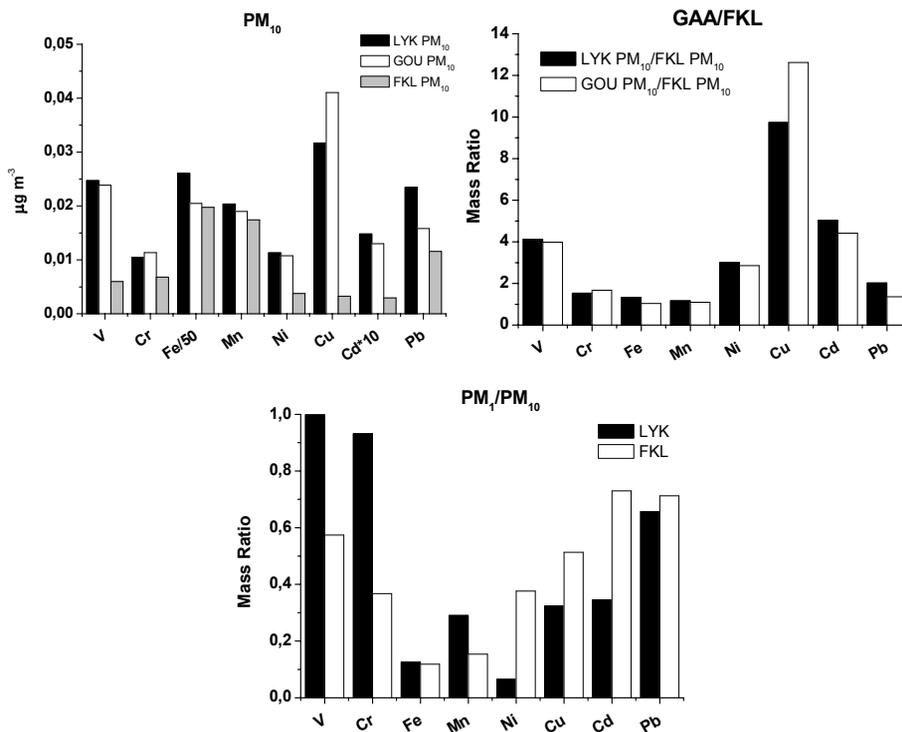
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**Fig. 7.** Annual average concentrations ( $\mu\text{g m}^{-3}$ ) of the studied metals for the three stations (a) in  $\text{PM}_{10}$ , (b) GAA/FKL mass ratio and (c)  $\text{PM}_1/\text{PM}_{10}$  mass ratio.

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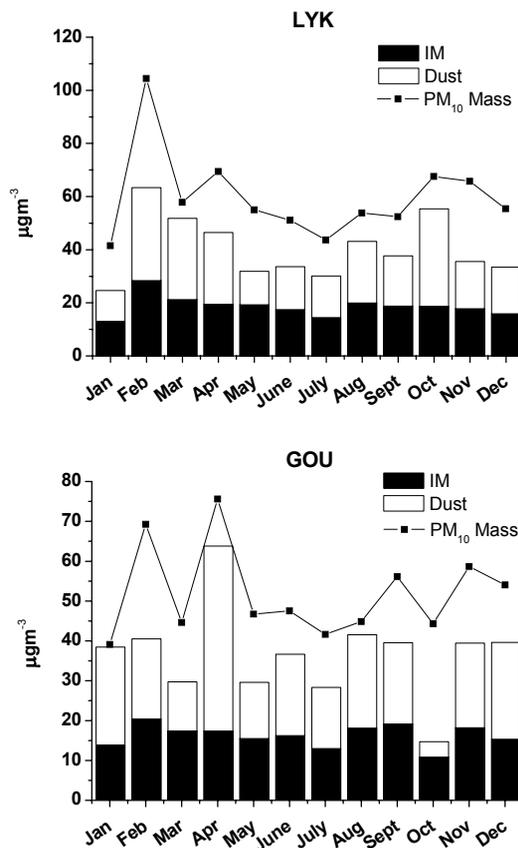
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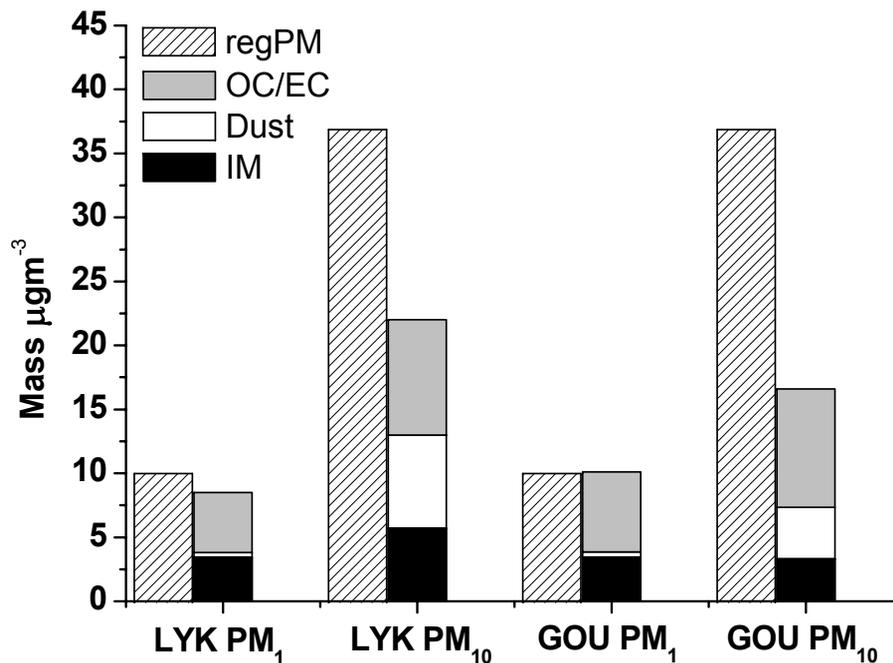


**Fig. 8.** Mass Closure (Ionic Mass and Dust) versus measured PM<sub>10</sub>: (a) LYK and (b) GOU.

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**Fig. 9.** Relative contribution of regional and local sources to PM levels in GAA. Regional PM (regPM) corresponds to PM measurements at FKL.

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