

1 Supporting Information

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3 **Condensational Uptake of Semivolatile Organic Compounds in Gasoline**
4 **Engine Exhaust onto Pre-existing Inorganic Particles**

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22 **AURAMS Description**

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24 AURAMS (version 1.4.0) is an off-line chemical transport model (CTM) that is
25 driven by the Canadian operational weather forecast model, GEM (Global Environmental
26 Multiscale model). GEM (version 3.2.2) was used to produce meteorological fields with a
27 15-km horizontal grid spacing. GEM was run for 12-hr periods from reanalysis files with
28 a 6-hr spin-up and 6-hr of simulation stored for the CTM. AURAMS was run with a 15-
29 km horizontal grid spacing for a domain covering the northeastern U.S. and eastern
30 Canada and using climatological chemical boundary conditions.

31 Gridded hourly anthropogenic point, area and on-road mobile emissions files were
32 prepared for the CTM with the 2005 Canadian and 2005 U.S. national criteria-air-
33 contaminant emissions inventories and version 2.2 of the SMOKE emissions processing
34 system. Total gasoline exhaust organic vapour was treated as an additional gas-phase
35 species in the on-road mobile emissions stream of the emissions processing system. This
36 species was emitted, transported, lost by gas-phase chemistry and allowed to reach an
37 equilibrium partitioning with sulphate aerosol based on the effective uptake coefficient fit

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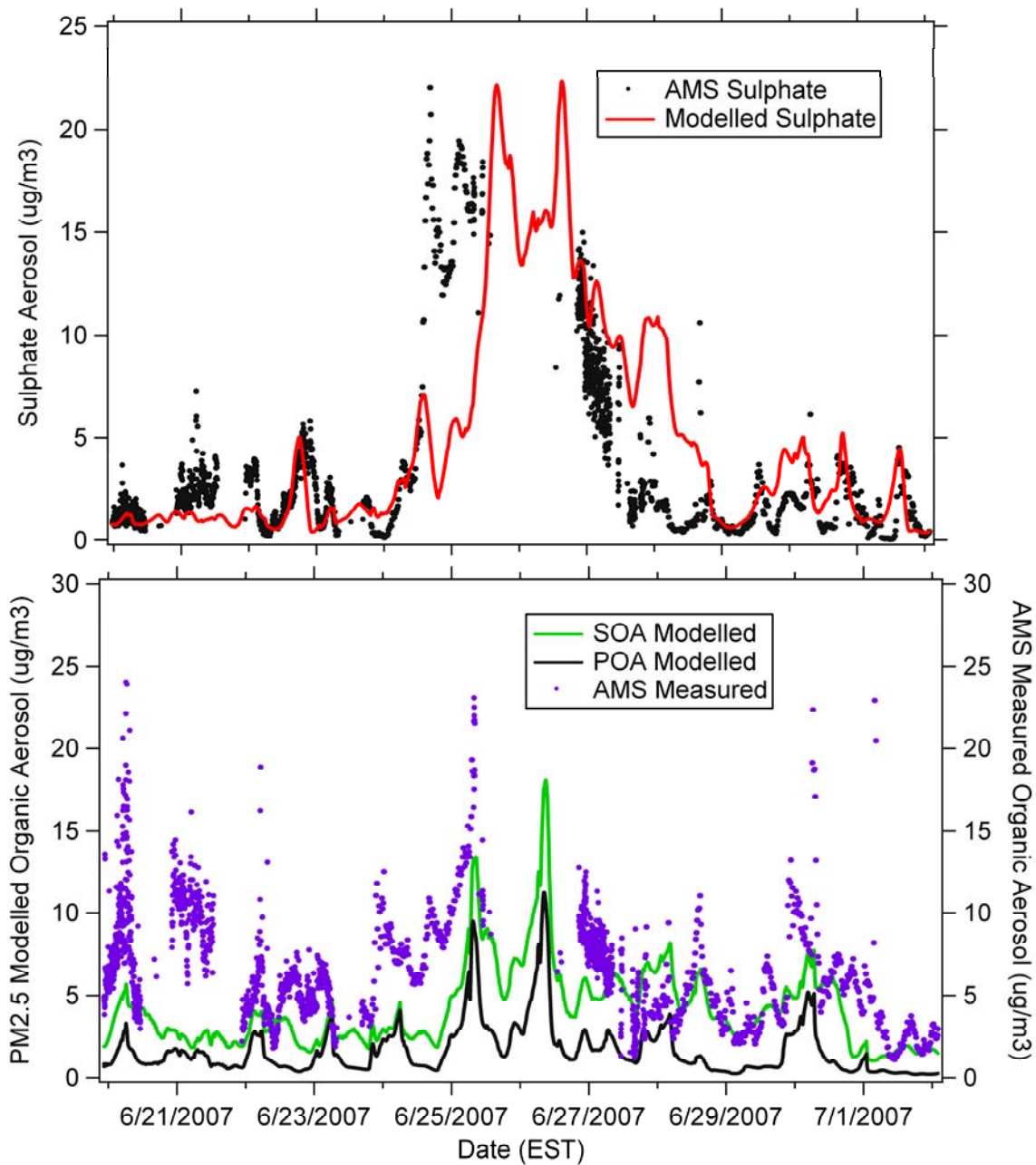
38 of Equation (6) ($S=0.012+0.000137*THC^{2.53}$) where uptake has units of kg organic
39 particle per kg sulfate and GTHC has units of $\mu\text{g m}^{-3}$). A Newton iteration method was
40 used to calculate the equilibrium solution with a 1% convergence criteria for the GTHC
41 vapour. Gas-phase loss by oxidation with OH, NO₃ and O₃ was calculated with rate
42 coefficients of 1.2E-11, 1.2E-14 and 6.7E-18 cm³ molec⁻¹ sec⁻¹, respectively. These rate
43 coefficients are reactivity-weighted averages from the individual species rate coefficients
44 in the VOC emissions profile for gasoline exhaust. Gasoline exhaust primary organic
45 aerosol emissions were also modelled in AURAMS as a separate tracer aerosol species.

46 Biogenic emissions were calculated on-line by AURAMS using BEIS version
47 3.09, the Biogenic Emissions Landcover Database (BELD3) vegetation data set (30 tree
48 species, 20 crop species), and meteorological fields (temperature and irradiance) from
49 GEM. Biogenic VOC emissions are speciated into four groups: isoprene; monoterpenes;
50 sesquiterpenes; and “other VOCs”. Sesquiterpene emissions were calculated by scaling
51 monoterpene emissions, as described in Helmig et al., (2007) (e.g., sesquiterpene
52 emissions were a factor of 0.16 lower than monoterpenes at 30C).

53 The gas-phase mechanism in AURAMS is an updated version of the ADOM-II
54 mechanism (Lurmann et al., 1986; Stockwell et al., 1989; Kuhn et al., 1998) that is solved
55 using a vectorized version of the rodas3 solver (Sandu and Sander, 2006). A detailed
56 description of the ADOM-II VOC lumping scheme can be found in Stroud et al. (2008).
57 In this study, a lumped monoterpene species was separated from the original ADOM-II
58 anthropogenic long-chain alkene species and assigned the OH/O₃/NO₃ kinetics of α -
59 pinene. A lumped sesquiterpene species was added to the mechanism and modelled with
60 β -caryophyllene OH/ O₃/NO₃ kinetics. Benzene was separated from the original ADOM-
61 II lumped species, propane (sum of propane, acetylene and benzene), and reacted in the
62 modified mechanism with OH kinetics. The overall organic aerosol yield approach was
63 applied to the following VOC precursor species: isoprene (ISOP), monoterpenes (PINE),
64 sesquiterpenes (SESQ), benzene (BENZ), mono-substituted aromatics (TOLU), multi-
65 substituted aromatics (AROM), long chain anthropogenic alkenes (ALKE), long chain
66 anthropogenic alkanes (ALKA). Aerosol yields were calculated for low and high NO_x
67 limits as a function of existing organic aerosol loadings (sum of primary and secondary)

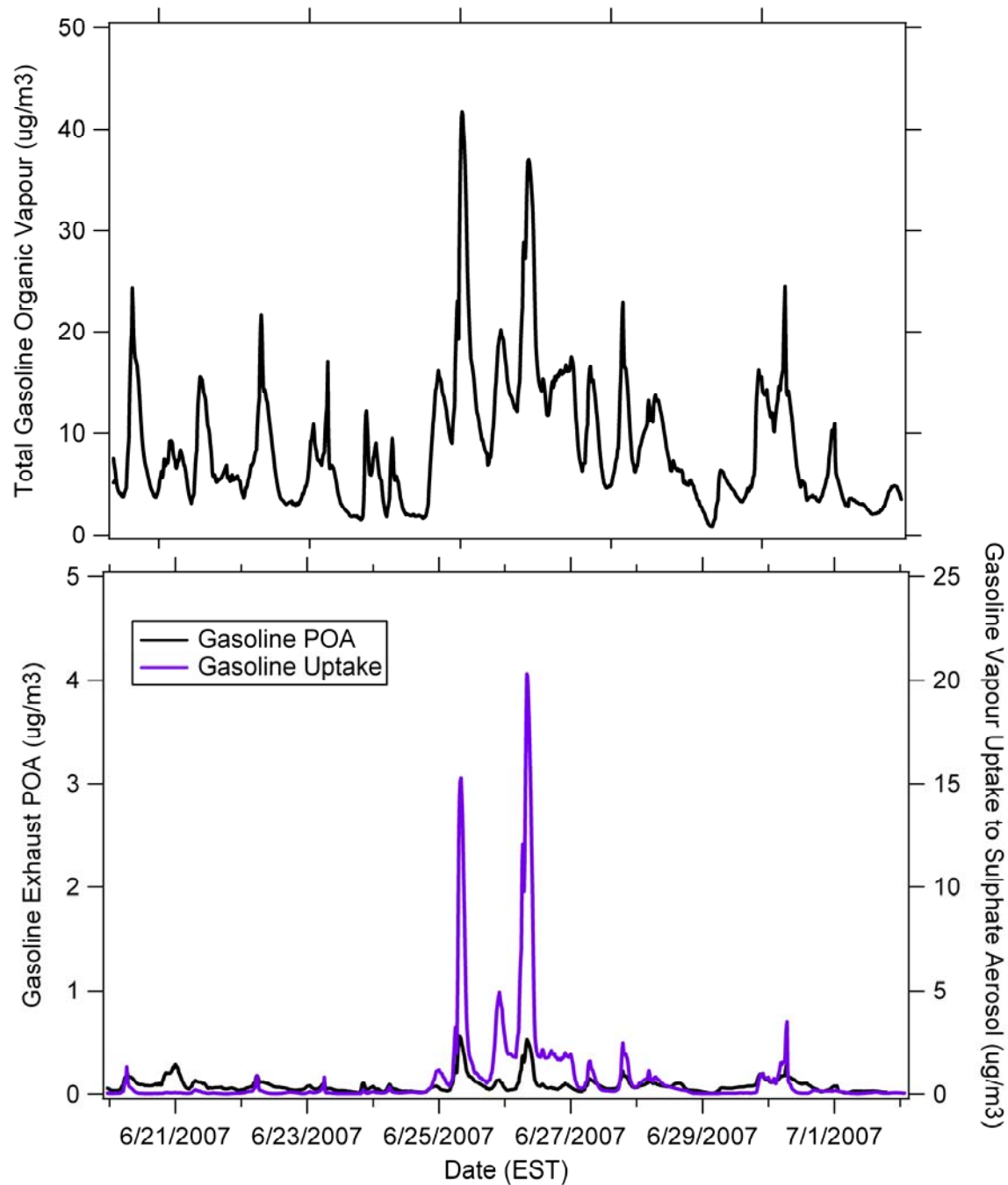
68 and temperature. Updated α_i and K_i values were based on recent literature studies (ISOP,
69 Kroll et al. (2006) and Lane et al. (2008); PINE, Pathak et al. (2007), Griffin et al., (1999)
70 and Zhang et al. (2006); SESQ, Lane et al. (2008); BENZ, Ng et al. (2006); TOLU,
71 Hildebrandt et al. (2009); AROM, Ng et al. (2007); ALKE, Lane et al. (2008); and
72 ALKA, Lane et al. (2008)). An incremental increase in SOA mass was calculated from
73 decreases in precursor VOC concentrations for a given time step under both low and high
74 NO_x conditions. A linear interpolation between the low NO_x and high NO_x incremental
75 SOA mass was performed based on the fraction of the RO2 radicals that react with HO_x
76 vs NO_x (Presto and Donahue, 2006; Henze et al., 2008). An organic particle density of
77 1.5 g cm^{-3} was assumed for conversion of normalized aerosol yield data. The particle
78 size distribution is represented in the CTM by 12 size bins ranging from 0.01 to 40.96 μm
79 in Stokes diameter, with the 8 lower bins corresponding to sizes below 2.5 μm . Particle
80 composition is represented by nine chemical species (sulfate, nitrate, ammonium, black
81 carbon, POA, SOA, crustal material, sea salt, and particulate water), which are assumed
82 to be internally mixed within each size bin (14). Condensation of the SOA to the particle
83 size distribution is described by a modified Fuchs-Sutugin equation as described by
84 equation A14 in Gong et al. (2003).

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Figure S1. Windsor time series for measured and modelled sulphate aerosol (top panel) and organic aerosol (bottom panel).



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Figure S2. Windsor time series for total gasoline organic vapour (top panel), gasoline exhaust primary organic aerosol (bottom panel) and gasoline vapour uptake to sulphate aerosol (bottom panel).

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