1 2 **Supplementary Material to** 3 4 Mass-spectrometric identification of primary biological particle markers: indication for 5 low abundance of primary biological material in the pristine submicron aerosol of 6 **Amazonia** 7 J. Schneider¹, F. Freutel¹, S. R. Zorn^{1,2}, O. Chen², D. K. Farmer³, J. L. Jimenez³, S. T. 8 Martin², P. Artaxo⁴, A. Wiedensohler⁵, and S. Borrmann^{1,6} 9 10 [1] Particle Chemistry Department, Max Planck Institute for Chemistry, Mainz, Germany School of Engineering and Applied Sciences and Department of Earth and Planetary 11 [2] 12 Sciences, Harvard University, Cambridge, MA, USA 13 [3] Dept. of Chem. & Biochem. & CIRES, University of Colorado, Boulder, CO, USA Applied Physics Department, Institute of Physics, University of São Paulo, Brazil 14 [4] 15 [5] Leibniz Institute for Tropospheric Research, Leipzig, Germany [6] Institute for Atmospheric Physics, Johannes Gutenberg University Mainz, Germany 16 17 Correspondence to: J. Schneider (johannes.schneider@mpic.de) 18 19

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Details on the evaluation of AMS data from the AMAZE field campaign

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- 22 1 Modifications made to fragmentation table (SQUIRREL v1.49)
- 23 Gas phase correction derived from blank measurements:
- 24 frag air[29] = 0.845*0.00736* frag air[28]
- 25 frag_CO2[44] = 0.83 * 0.00037 * 1.36 * 1.28 * 1.14 * frag_air[28]
- 26 frag_RH[18] = 0.8 * 0.01 * frag_air[28]

- 27 frag_O16[16] = 1.10 * 0.353 * frag_air[14]
- 28
- 29 Changes to account for the high contribution of organic nitrates:
- 30 frag_nitrate[46] = 46,
- frag_nitrate[30] = 2* frag_nitrate[46],
- frag_organic[30] = 30,-frag_nitrate[30],-frag_air[30]
- 33
- 34 This is based on the following assumptions: m/z 46 is only due to NO_2^+ from inorganic nitrate,
- 35 the ratio of m/z 30 to m/z 46 is 2:1 for inorganic ammonium nitrate (Allan et al., 2003;
- Hogrefe et al., 2004), and therefore the rest of m/z 30 is due to organic nitrate or other organic
- ions (as CH₄N⁺ and CH₂O⁺). The high-resolution data (12 h averages) show that between 20
- and 60% (on average 35%) of m/z 30 is due to NO⁺ (see Figure S1, lower panel).
- 39 For the calculation of the mass concentration standard relative ionization efficiencies were
- 40 used (nitrate: 1.1; sulfate: 1.2; organics: 1.4, ammonium: 4; chloride: 1.3). The applied
- 41 collection efficiency (CE) was 1.0, which is consistent with the intercomparisons with other
- 42 instruments and the liquid character of the submicron particles (see Chen et al., (2009)
- 43
- 2 Contributions of the marker m/z to the UMR mass peaks:
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- 46 Figure S1 shows the ratios of the marker peak intensities to the respective UMR peak (m/z 30
- and 42 for amino acids, upper panel (a); m/z 60, 61, and 73 for carbohydrates, middle panel,
- 48 (b)). The lower panel (c) shows the ratio of NO^+ to the UMR peak at m/z 30.
- Figure S2 shows the high resolution peak fitting for m/z 30 and 42 for the examples for March
- 50 05, 12 h (local time), when the amino acid markers showed maximum values.
- 51

Supplementary Figures

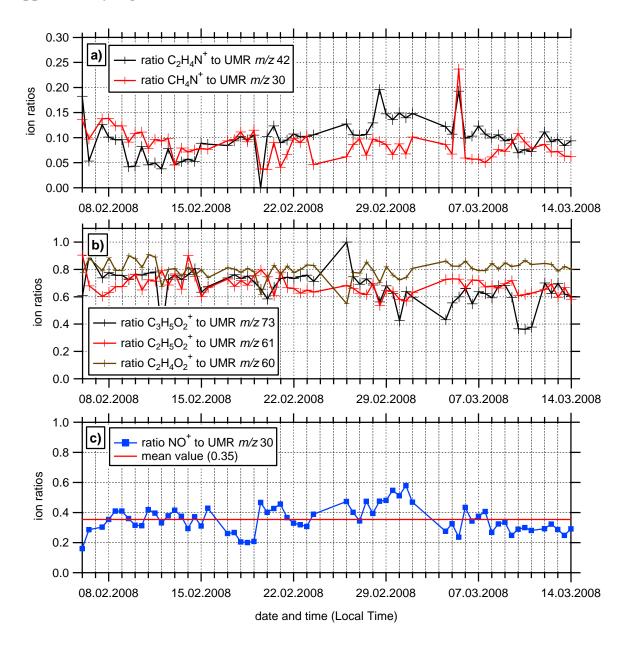


Figure S1. Fraction of marker peaks to the total UMR peak at the nominal m/z ratio measured during AMAZE-08. a) amino acid markers, b) carbohydrate markers, c) fraction of NO⁺ to UMR m/z 30.

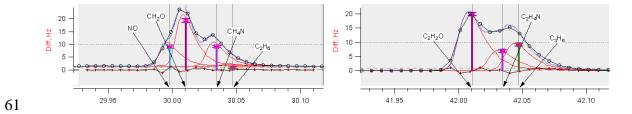


Figure S2. High resolution peak at m/z 30 and m/z 42 from March 05, 2008, 12 h (local time).

During this time period CH_4N^+ and $C_2H_4N^+$ contribute significantly more to the respective

UMR peak than during other times.