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Source apportionment of carbonaceous aerosol in southern Sweden

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Source apportionment of carbonaceous aerosol

J. Genberg et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

⏪

⏩

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion

Abstract

A one-year study was performed at the Vavihill background station in southern Sweden to estimate the anthropogenic contribution to the carbonaceous aerosol. Weekly samples of the particulate matter PM₁₀ were collected on quartz filters, and the amounts of organic carbon, elemental carbon, radiocarbon (¹⁴C) and levoglucosan were measured. This approach enabled source apportionment of the total carbon in the PM₁₀ fraction using the concentration ratios of the sources. The sources considered in this study were emissions from the combustion of fossil fuels and biomass, as well as biogenic sources. During the summer, the carbonaceous aerosol mass was dominated by compounds of biogenic origin (82 %), which are associated with biogenic primary and secondary organic aerosols. During the winter months, biomass combustion (38 %) and fossil fuel combustion (33 %) were the main contributors to the carbonaceous aerosol. Elemental carbon concentrations in winter were about twice as large as during summer, and can be attributed to biomass combustion, probably from domestic wood burning. The contribution of fossil fuels to elemental carbon was stable throughout the year, although the fossil contribution to organic carbon increased during the winter. Thus, the organic aerosol originated mainly from natural sources during the summer and from anthropogenic sources during the winter. The result of this source apportionment was compared with results from the EMEP model. The model and measurements were generally consistent for total atmospheric organic carbon, however, the contribution of the sources varied substantially. E.g. the biomass burning contributions of OC were underestimated by the model by a factor of 8.2 compared to the measurements.

1 Introduction

Aerosol particles are known to have adverse effects on human health (Dockery et al., 1993) by affecting not only the respiratory tract, but also the cardiovascular organs (Pope III et al., 2002). Aerosol particles also influence the Earth's climate due to their optical properties and their ability to act as condensation nuclei for cloud droplets

Source apportionment of carbonaceous aerosol

J. Genberg et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

⏪

⏩

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



(Andreae et al., 2004). Aerosol composition must, therefore, be taken into consideration when modelling changes in the Earth's climate. However, it is not known in detail how human activities affect aerosol composition, or the degree to which they can alter the climate.

5 According to aerosol mass spectrometry measurements performed in Europe, the organic aerosol (OA) constitutes a considerable fraction of the submicron particle mass, corresponding to between 30 and 60 % of the aerosol mass, depending on location and season (Jimenez et al., 2009). The large fraction of organic compounds in aerosol particles motivates studies of these compounds because of their adverse effects on
10 our environment. Also, the sources of these particles are highly uncertain and variable (Gelencsér et al., 2007; May et al., 2009). Organic aerosols originate mainly from three sources: combustion of fossil fuels, biomass burning and biogenic sources. All these sources can contribute to both primary and secondary organic aerosols (POAs and SOAs). Biogenic POAs consist of pollen and plant debris, fungal spores and microor-
15 ganisms. Biogenic SOAs originate from volatile organic compounds (VOCs) emitted by plants, e.g. monoterpenes. The emission of biogenic VOCs is related to the growing season and temperature, and is significantly greater during the warmer months of the year (Zemankova and Brechler, 2010). Primary fossil-fuel-based emissions from traffic and industry contribute to organic aerosols, as do secondary particles arising from the
20 emission of fossil-based VOCs. Biomass burning, e.g. forest fires, is common during dry periods and is a large contributor to aerosol particles globally (Crutzen and Andreae, 1990) and occasionally in Nordic areas (Saarikoski et al., 2008). The burning of biomass for domestic heating is also a major source of particles, especially in South Asia (Gustafsson et al., 2009), but also in some European cities e.g. Zurich (Switzer-
25 land), where approximately 65 % of the winter carbonaceous aerosol has been found to originate from biomass burning (Szidat et al., 2006). During wintertime biomass burning is also a large contributor to aerosol mass in the Scandinavian countries; Sweden (Hedberg Larsson et al., 2006; Szidat et al., 2009), Denmark (Glasius et al., 2006) and Norway (Yttri et al., 2009, 2011).

Source apportionment of carbonaceous aerosol

J. Genberg et al.

[Title Page](#)[Abstract](#)[Introduction](#)[Conclusions](#)[References](#)[Tables](#)[Figures](#)[Back](#)[Close](#)[Full Screen / Esc](#)[Printer-friendly Version](#)[Interactive Discussion](#)

**Source
apportionment of
carbonaceous
aerosol**J. Genberg et al.

[Title Page](#)[Abstract](#)[Introduction](#)[Conclusions](#)[References](#)[Tables](#)[Figures](#)[Back](#)[Close](#)[Full Screen / Esc](#)[Printer-friendly Version](#)[Interactive Discussion](#)

The carbonaceous aerosol consists of numerous compounds with different chemical and physical properties. The total carbon content (TC) is divided into organic carbon (OC) and elemental carbon (EC). The organic carbon fraction includes organic molecules and polymers, e.g., cellulose, alkanes and organic acids, which do not absorb visible light. EC consists of compounds with graphite-like structures formed during combustion, e.g. char and soot, which can absorb light (Pöschl, 2005). This means that EC has important climate impact, as light-absorbing particles in the atmosphere lead to a warmer climate (Ramanathan and Carmichael, 2008; Menon et al., 2002). The split of combustion products into OC and EC is, however, not clear as they form a continuum from OC to EC (Masiello, 2004). With higher combustion temperatures, carbon compounds generally become more light absorbing and less reactive, all properties of EC. Numerous methods of separating OC from EC by thermal analysis have been proposed (Cachier et al., 1989; Fung, 1990). These methods utilise the difference in thermal stability between OC and EC. However, during the analysis, some organic compounds char due to heating and form “false” EC. This can be corrected for by monitoring the transmission of a laser beam through the filter sample during heating. The NIOSH protocol (Birch and Cary, 1996) uses this transmission method which is also the basis for the protocol developed within the EU-funded project EUSAAR (European Supersites for Atmospheric Aerosol Research) (Cavalli et al., 2010).

Radiocarbon (^{14}C) is a powerful tracer for modern carbon. With a half-life of 5730 yr, ^{14}C is depleted in fossil fuels while present at traceable amounts in carbon from contemporary sources. The amount of ^{14}C in a sample can therefore be used to determine the fractions of carbon with modern and fossil origin. However, the method is complicated by the release of ^{14}C from detonations of thermonuclear weapons in the 1950s and 1960s (the bomb effect, Levin and Hesshaimer, 2000), as well as dilution of atmospheric modern carbon by fossil carbon emitted from combustion of fossil fuels (the Suess effect, Suess, 1955). ^{14}C has been used in several aerosol source apportionment studies to determine the influence of fossil fuels on the organic aerosol (Szidat et al., 2006, 2007; Gelencsér et al., 2007). Since both the biogenic OA and the biomass

burning carbonaceous aerosol contain ^{14}C , it is difficult to separate these two sources by relying solely on carbon isotope measurements. The tracer molecule levoglucosan can however be used to determine whether the particles originate from the burning of biomass, as it is formed when cellulose is heated to above 300°C . For the purpose of this study, it was considered to be atmospherically stable, although this is still being debated (Locker, 1988; Fraser and Lakshmanan, 2000; Hoffmann et al., 2010).

In this study, 51 aerosol samples collected in southern Sweden over one year were analysed to determine the concentrations of ^{14}C , levoglucosan, EC and OC. These measurements were combined to provide the first full-year source apportionment of the carbonaceous aerosol in Sweden. The measurement based source apportionment results for OC were compared with corresponding results from the EMEP MSC-W chemical transport model. This is an important step to evaluate how well the OA models and their input data represent the measurements.

2 Methods

2.1 Sampling

The sampling site Vavihill is a EUSAAR and EMEP (European Monitoring and Evaluation Programme) background station situated in southern Sweden ($56^\circ 01' \text{N}$, $13^\circ 09' \text{E}$, 172 m a.s.l.). The station is not close to any large local pollution sources, although the distances to the densely populated areas of Malmö, Copenhagen and Helsingborg, west to southwest of the station, are only 45, 50 and 25 km, respectively. Continental Europe in the south also influences the air at Vavihill, whereas air masses from the north and northeast are generally clean (Kristensson et al., 2008).

Samples were collected weekly between April 2008 and April 2009. The PM_{10} fraction was collected on 47 mm quartz filters (Pall TissuquartzTM, binder-free) using an aerosol flow of 38 L min^{-1} . The filters were baked at 900°C for 4 h prior to sampling. After sampling, the filters were stored in a Petri dish, wrapped in aluminium foil in a

Source apportionment of carbonaceous aerosol

J. Genberg et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

⏪

⏩

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



refrigerator (+8°C) or a freezer (−30°C). Samples were generally analysed for OC and EC within 12 weeks of sampling. ¹⁴C and levoglucosan analyses were conducted within two years of sampling.

Most of the samples were collected using the EUSAAR sampling train, consisting of two filters in series following a VOC denuder in order to remove VOCs. Some samples were collected without the use of the denuder and therefore the OC values had to be corrected for positive artefacts due to adsorbed VOCs. Samples collected with the denuder were not corrected for positive or negative artefacts since the back-filter loadings were close to that of the field blanks, indicating that the artefacts were low. These back-filter results also demonstrate that storage of the filters did not lead to any significant positive artefacts. Two tests were performed at Vavihill to determine the denuder efficiency in order to estimate positive artefacts; one in February 2008 and one in August 2008. Samples were collected simultaneously in three sampling lines: one with both a denuder and a Teflon filter in front of two quartz filters, one with only a Teflon filter in front of two quartz filters, and one with two quartz filters without a denuder or a Teflon filter. The results from these samples were in this study used to evaluate the magnitude of the positive artefacts on samples collected without a denuder.

2.2 OC/EC analysis

OC/EC analyses were conducted on a 0.5 cm² area of the filter using a DRI Model 2001 OC/EC Carbon Analyzer (Atmoslytic, Calabasas, CA, USA). The organic and elemental carbon fractions were separated according to a novel method developed within the framework of the EUSAAR project called the EUSAAR_2 protocol. This was developed to reduce measurement uncertainty in aerosol samples from background sites around Europe (Cavalli et al., 2010). According to the EUSAAR protocol, the filter sample is heated in helium in four steps to 650°C. The filter is then left to cool to 500°C, oxygen (2%) is added and the temperature is raised in four steps to 850°C. The split point between OC and EC is determined by monitoring the transmission of a laser beam through the filter (He/Ne, 633 nm). When the transmission reaches its initial

Source apportionment of carbonaceous aerosol

J. Genberg et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

⏪

⏩

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



value, i.e. the value of the start of the analysis, the carbon left on the filter is regarded as EC.

2.3 ^{14}C determination

The ^{14}C content in the aerosol samples was analysed using accelerator mass spectrometry. Prior to analysis, the carbon was extracted using a newly implemented graphitisation system optimised for μg -sized samples (Genberg et al., 2010). Briefly, the sample is combusted in vacuum using CuO as oxidation agent. The CO_2 formed is cryogenically purified and mixed with H_2 in a small-volume reduction reactor, and thereafter heated to 600°C . In the reaction, the CO_2 is transformed into pure graphite on a heated iron catalyst. The $^{14}\text{C}/^{12}\text{C}$ ratio in the graphite is analysed using the 250 kV single-stage accelerator mass spectrometer at Lund University (Skog, 2007; Skog et al., 2010) and expressed as fraction modern ($F^{14}\text{C}$) (Reimer et al., 2004). A $F^{14}\text{C}$ value of 1.0 is equivalent to the hypothetical concentration of ^{14}C in atmospheric carbon from 1950 when human influences are not taken into account. The true atmospheric ^{14}C concentration has however been altered due to emissions of fossil CO_2 (Suess effect) and formation of ^{14}C from thermonuclear weapons (bomb effect).

Due to the Suess and bomb effect $F^{14}\text{C}$ values had to be deduced for the different modern sources of carbon (biogenic and biomass). For biogenic emissions a value of $F^{14}\text{C} = 1.04$ was used, taken from measurements performed by Levin in 2007 (Levin, 2008). It is complicated to determine the $F^{14}\text{C}$ of biomass burning ($F^{14}\text{C}_{\text{bb}}$) since the trees used as firewood in Sweden today were growing during the 1960s, when the amount of ^{14}C in the atmosphere was up to twice that of today (Levin and Hesshaimer, 2000). We used the method described by Lewis et al. (2004) to model the amount of ^{14}C accumulated in trees. The method uses the Chapman–Richards growth model (Richards, 1959) to simulate the growth of the tree, and atmospheric ^{14}C concentrations (Stuiver and Quay, 1981; Levin et al., 2008; Levin, 2008) to estimate the $F^{14}\text{C}$ of the whole tree. According to this method, a 60- to 80-yr-old tree harvested in 2008 has an estimated $F^{14}\text{C}$ between 1.21 and 1.23 (Fig. 1).

Source apportionment of carbonaceous aerosol

J. Genberg et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

⏪

⏩

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



2.4 Levoglucosan analysis

Levoglucosan was analysed using the method of Zdrahal et al. (2002) with some modifications. A filter sample, 1 cm², was further divided into small pieces and placed in a 50 mL conical Pyrex flask. Fifteen mL of a solvent mixture of dichloromethane and methanol (3:1) was added to the flask. Extraction was carried out for 45 min in a Branson 3200 sonicator (Branson, Danbury, CT, USA). The extract was collected, and two separate aliquots of 10 mL of the dichloromethane:methanol mixtures were added for re-extraction for 30 and 15 min. The extract was pooled and concentrated to dryness by evaporation at 60 °C under a gentle stream of N₂. Dichloromethane (CH₂Cl₂) was added to the solid material remaining, to a volume of 1 mL.

Fifty µL of the extract solution and 5 µL DTE (1,4-dithioerythritol, 0.5 µg mL⁻¹, 99 % pure, Sigma-Aldrich) were placed in a gas chromatography vial and evaporated to dryness under a stream of nitrogen at 60 °C. Thirty µL pyridine (99.5 % purity, Acros Organics) and 10 µL N-methyl-N-(trimethylsilyl)trifluoroacetamide (MSTFA, ≥98 % excluding trimethylsilyl chloride, Acros Organics) containing 1 % trimethylsilyl chloride was added to the vial. The vial was sealed with a screw cap with Teflon septa. Derivatisation was carried out at 80 °C for one hour, after which the sample was allowed to cool to room temperature. The solution containing derivatised levoglucosan was evaporated to dryness at 60 °C under a continuous flow of nitrogen, and thereafter dissolved in 50 µL dichloromethane containing 5 µg mL⁻¹ 1-phenyldodecane (97 % Acros Organics, internal standard).

Samples were analysed using an Agilent 6890 series gas chromatograph with an Agilent 5973 network mass detector. The column used was Varian VF-1ms (30 m × 0.32 mm ID 0.25). The injection volume was 2 µL, and splitless injection was used. The inlet temperature was 280 °C and constant flow mode was used. The gas (He) flow through the column was 1.3 mL min⁻¹. The temperature programme was as follows: initial temperature 60 °C for three minutes, the temperature was then raised to 190 °C at a rate of 10 °C min⁻¹, and finally the temperature was raised to 300 °C at a rate

Source apportionment of carbonaceous aerosol

J. Genberg et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

⏪

⏩

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion

of 30 °C min⁻¹ (Hsu et al., 2007). The MSD transfer line temperature was 280 °C, the MS source temperature was 250 °C, and the MS quadropole temperature was 180 °C. Levoglucosan was identified by comparison with the mass spectrum and retention time of standards, and quantified using selected-ion monitoring. The amount of levoglucosan was quantified using $m/z = 217$, and 1-phenyldodecane (internal standard) using $m/z = 246$. Standards were prepared using 99 % pure 1,6-anhydro- β -glucopyranose (Alfa Aesar).

2.5 Source apportionment method

To model the origin of the carbon fractions, the carbon was assumed to originate from one of three sources: biogenic emissions, biomass burning and combustion of fossil fuels. These are the same sources as those used by Szidat et al. (2006). The total carbon (TC) content is divided into OC and EC:

$$TC = EC + OC \quad (1)$$

EC is determined as the sum of elemental carbon arising from biomass burning (EC_{bb}) and fossil fuel combustion (EC_{ff}):

$$EC = EC_{ff} + EC_{bb} \quad (2)$$

EC_{bb} is calculated using the measured levoglucosan concentration (lev) and the estimated ratio of EC_{bb}/lev :

$$EC_{bb} = lev \cdot \frac{EC_{bb}}{lev} = lev \cdot \left(\frac{EC}{OC} \right)_{bb} / \left(\frac{lev}{OC_{bb}} \right) \quad (3)$$

The ratio EC_{bb}/lev is determined using two ratios from earlier studies (Sect. 2.6): $(EC/OC)_{bb}$ and (lev/OC_{bb}) . EC_{bb} is limited so as not to exceed the total measured EC, to prevent negative values of EC_{ff} . EC_{ff} is then determined by subtraction (Eq. 2).

Source apportionment of carbonaceous aerosol

J. Genberg et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

⏪

⏩

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion

OC is separated into three sources: biomass burning (OC_{bb}), non-fossil OC (OC_{bio}) and OC originating from fossil fuels (OC_{ff}):

$$OC = OC_{bb} + OC_{bio} + OC_{ff} \quad (4)$$

OC_{bb} includes primary particles emitted from forest fires and agricultural clearing, as well as domestic wood burning and is based on the formation of levoglucosan when cellulose is heated during the combustion. OC_{bio} in this study includes both primary and secondary biogenic particles, as well as secondary particles formed from other non-fossil sources, i.e. SOA originating from biomass burning.

OC_{bb} is calculated using the levoglucosan levels and the lev/ OC_{bb} ratio, which yields the organic primary particle emission from biomass burning:

$$OC_{bb} = lev / \frac{lev}{OC_{bb}} \quad (5)$$

Since levoglucosan is not an ideal quantitative tracer, and the lev/ OC_{bb} ratio is uncertain, as will be discussed in Sect. 2.6, the value of OC_{bb} was limited in two ways in the calculations. Firstly, it was not allowed to exceed the total OC. Secondly, it was limited by the radiocarbon measurement in order not to introduce so much OC_{bb} that the ^{14}C level could not be balanced by OC_{ff} . The maximum amount of OC originating from biomass burning ($OC_{bb_{max}}$) corresponds to the amount of OC needed to allocate all ^{14}C in the OC fraction to biomass burning:

$$OC_{bb_{max}} = OC \cdot \frac{F^{14}C_{OC}}{F^{14}C_{bb}} \quad (6)$$

$F^{14}C_{OC}$ is calculated from the measured ^{14}C concentration in the sample ($F^{14}C_{TC}$), OC/EC measurements, and the levoglucosan value:

$$F^{14}C_{OC} = \frac{F^{14}C_{TC} \cdot TC - F^{14}C_{EC} \cdot EC}{OC} \quad (7)$$

Source apportionment of carbonaceous aerosol

J. Genberg et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



Source apportionment of carbonaceous aerosol

J. Genberg et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

⏪

⏩

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



In Eq. (7), $F^{14}C_{EC}$ is determined by the relative contribution of biomass burning to EC:

$$F^{14}C_{EC} = F^{14}C_{bb} \cdot \frac{EC_{bb}}{EC} \quad (8)$$

$F^{14}C_{bb}$ is the ^{14}C concentration in wood smoke, modelled using the method described by Lewis et al. (2004) as described above (Sect. 2.3). The value of $F^{14}C_{OC}$ is, however, close to the value of $F^{14}C_{TC}$ since OC is the dominant contributor to the total carbon.

OC_{bio} is calculated by balancing the ^{14}C content which was not attributed to OC_{bb} (Eq. 5):

$$OC_{bio} = \frac{OC \cdot F^{14}C_{OC} - OC_{bb} \cdot F^{14}C_{bb}}{F^{14}C_{bio}} \quad (9)$$

$F^{14}C_{bio}$ is the ^{14}C concentration of the atmosphere and is set to 1.04 (Levin, 2008).

The remaining OC is considered to originate from fossil fuels (OC_{ff}) and is obtained by subtraction (Eq. 4).

2.6 Sensitivity analysis

Some of the variables in Eqs. (1–9) are measured and some are taken from the literature, all with uncertainties. To evaluate the effect of the uncertainties of the different variables on the outcome of source apportionment, a random sampling model (McKay et al., 1979) inspired by the Latin hypercube sampling method (Gelencsér et al., 2007) was used. All variables were allowed to vary within a set distribution, and calculations were performed with 3000 random sets of variables. The median value was considered the best estimate since the mean value may be affected by the use of limitations in the calculations (Eqs. 3 and 5).

The measurements of ^{14}C concentration ($F^{14}C$) and levoglucosan used in the equations were allowed to vary by ± 2 standard deviations in a broad beta distribution ($\alpha = 1.5$, $\beta = 1.5$). OC and EC measurements were varied by ± 10 and 20 %, respectively, in a beta distribution ($\alpha = 1.5$, $\beta = 1.5$). The value of the ^{14}C concentration

**Source
apportionment of
carbonaceous
aerosol**

J. Genberg et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



resulting from biomass combustion was modelled according to Lewis et al. (2004), as described in Sect. 2.3. In Sweden, forests are normally clear cut at an age of 60 to 80 yr, rendering an $F^{14}C_{bb}$ value of 1.21 to 1.23 for the majority of the fire wood. However, fire wood used for domestic burning (which normally has less effective combustion) may be derived from thinning of the forest, which provides younger firewood (20–30 yr), and therefore lower $F^{14}C_{bb}$ values (1.06 to 1.09). To cover this variation in age a broad beta distribution of $F^{14}C_{bb}$ was used ($\alpha = 1.8$, $\beta = 1.2$). This distribution slightly favours the higher values to emphasize the higher values as most of the firewood used belongs to the older category.

The ratios of levoglucosan to OC_{bb} and EC_{bb} to OC_{bb} are dependent on the type of wood being burnt and the burning conditions (Oros and Simoneit, 2001a, b), which vary in the south of Sweden (and throughout Europe). Both ratios have been investigated for numerous types of wood fuels used in the US (Fine et al., 2001, 2002, 2004a, b; Engling et al., 2006). The results from these studies, together with the results presented by Zdráhal et al. (2002) are presented in Table 1. The values reported by Zdráhal et al. are from measurements in Brazil during the dry season, where most of the particulate carbon is presumed to originate from biomass burning.

Similar investigations have been performed in order to determine the best values for these ratios. Szidat (2006) concluded that suitable values of the levoglucosan/ OC_{bb} and $(EC/OC)_{bb}$ ratios were 0.15 ± 0.09 and 0.16 ± 0.05 , respectively, while Gelencsér et al. (2007) used 0.08–0.167 for levoglucosan/ OC_{bb} and 0.07–1.0 for $(EC/OC)_{bb}$. Puxbaum et al. (2007) used 0.136 for levoglucosan/ OC_{bb} ratio after having undertaken an extensive literature survey. The ratios and distributions used in the present study for the sensitivity analysis are presented in Table 2. A uniform distribution between 0.08 and 0.2 was used as the value for the levoglucosan/ OC_{bb} ratio, which covers most of the variation in the measurements, as well as the ratios used in previous studies. Estimates of the $(EC/OC)_{bb}$ ratio vary by more than one order of magnitude, and although the mean values are close to 0.15, the additional uncertainty of the OC/EC ratio analyses must be taken into account. A large portion of the range of estimates given by Fine

(Table 1, 0.06–0.45) were used in a broad beta distribution ($\alpha = 1.12$, $\beta = 1.52$), which slightly favours the lower values with a maximum probability of 0.15.

2.7 Source apportionment model of OC

The calculated sources based on measurements were compared with modelled sources apportionment of OC from the same time periods. A recently developed particulate carbonaceous matter (PCM) version of the EMEP MSC-W chemical transport model (Simpson et al., 2007, 2011) was used. The EMEP PCM model includes an organic aerosol scheme based on the Volatility Basis Set (VBS) approach (Donahue et al., 2006).

The emitted fossil fuel POA is distributed over different volatilities (9-bin VBS) and partitions between the gas and particulate phases. The emissions are assumed to be accompanied by emissions of low-vapour pressure (that is, partitioning) gases, which are currently not captured in either the POA or the VOC inventories. Following Shrivastava et al. (2008) the total emissions of condensable material (including POA) are assumed to amount to 2.5 times the POA inventory.

Both POA and anthropogenic SOA undergo gas phase reactions with OH (aging) and each reaction leads to a shift of the reacting OA to a lower volatility bin ($k_{\text{OH-POA}} = 4 \times 10^{-11} \text{ cm}^3 \text{ molecule}^{-1} \text{ s}^{-1}$, $k_{\text{OH-ASOA}} = 10^{-11} \text{ cm}^3 \text{ molecule}^{-1} \text{ s}^{-1}$, based on Murphy and Pandis, 2009).

Primary organic aerosol emissions were taken from the EUCAARI anthropogenic carbonaceous aerosol emission inventory (Visschedijk et al., 2009). Other anthropogenic emissions are taken from the standard EMEP emission inventory (Mareckova et al., 2009). Biogenic emissions of isoprene and monoterpenes are based on Guenther et al. (1993) and Simpson et al. (1999). A background concentration of $0.5 \mu\text{g m}^{-3}$ of biogenic OC is included in the model. This represents sources of OC not directly included in the emission inventories (primary biogenic, oceanic, and wildfire OC). Further details about the EMEP PCM model are given in Bergström et al. (2011).

Source apportionment of carbonaceous aerosol

J. Genberg et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion

3 Results and discussion

3.1 Positive artefact correction

Most of the samples analysed in this study were collected with a denuder in front of two quartz filters. This set-up was considered artefact-free in this study, despite the fact that some positive artefacts have been found during the EUSAAR project (Putaud et al., 2010). The denuder tests conducted at Vavihill indicate a denuder efficiency of 90 to 95 %. The results from parts of the Vavihill denuder tests are presented in Table 3. The back quartz filter loading on the line without a Teflon filter corresponds well to the loading of the back filter behind Teflon collected simultaneously. This demonstrates that the positive artefact is not affected by the Teflon filter. Samples where denuders were not used were corrected by using the carbon found on the back filter in accordance with the results of tests carried out in the winter and summer of 2008 (Table 3). Cheng et al. (2009), found that correcting by a factor of one times the amount of carbon on the back filter was inadequate compared with a denuded quartz filter sample. In this study a correction factor of 2.16 was used. However, the measurements range from 1.85 to 2.61 (and one outlier at 3.84 according to Grubb's test ($P < 0.05$)). No significant difference was observed between the two seasons, thus all the samples were corrected using the same factor. The correction was only conducted on the OC values and not to the corresponding $F^{14}C$ values. The $F^{14}C$ of the positive artefacts were assumed to be equal to the $F^{14}C$ of the OC fractions.

3.2 OC/EC results

The filter analyses of OC and EC show that both fractions increased in concentration during winter compared with the summer (Fig. 2). The seasons were determined by fitting a line to daily mean temperature. Summer and winter were determined as the period where the line was above $10^{\circ}C$ and below $0^{\circ}C$, respectively (Table 4). Spring and autumn were defined as the period between winter and summer. The amount of

Source apportionment of carbonaceous aerosol

J. Genberg et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

⏪

⏩

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



EC in the winter was twice that in the summer ($0.15 \mu\text{g m}^{-3}$ vs. $0.3 \mu\text{g m}^{-3}$), which is significant ($P < 0.001$). The amount of OC was higher during the winter ($P < 0.1$), and the top value was recorded at the beginning of February 2009 ($4.4 \mu\text{g OC m}^{-3}$). As can be seen in Fig. 2, the EC concentration increased already in October, and remained stable throughout the autumn and most of the winter. The OC concentration, on the other hand, was fairly stable before it peaked in a few samples collected in January and February 2009.

3.3 ^{14}C and levoglucosan results

The concentration of ^{14}C in the organic aerosol during the summer was generally high, as can be seen in Fig. 3a. A ^{14}C concentration between 0.9 and 1.0, expressed as $F^{14}\text{C}$, means that most of the carbon is of modern origin, i.e. biogenic sources or biomass burning. The simultaneously low levoglucosan concentrations ($2\text{--}12 \text{ ng m}^{-3}$) (Fig. 3b) imply that the effect of biomass burning was very low, from which it can be concluded that biogenic sources dominated during the summer. The amounts of levoglucosan in the summer were lower compared to summer measurements in Norway (Yttri et al., 2007, 2011).

During the winter months, the amount of levoglucosan in the air peaked at more than ten times the concentration found during the summer. At the same time, the ^{14}C content in TC decreased to between 0.7 and 0.8 expressed as $F^{14}\text{C}$. These two observations suggest that the influence of both fossil fuel and biomass burning is increased during winter. The wintertime concentrations of levoglucosan at Vavihill are similar to what has been reported for Norwegian urban samples collected during the winter (Yttri et al., 2011) but low compared to urban background samples (Yttri et al., 2009).

3.4 Best estimate source apportionment

The calculations were performed 3000 times for each filter sample, as explained in Sect. 2.6. The best estimates presented here are the median values calculated from

Source apportionment of carbonaceous aerosol

J. Genberg et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

⏪

⏩

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



those simulations. Then discussing seasonal concentrations the mean of the given season's medians are used. The results outlined above (Sects. 3.2 and 3.3) are confirmed by the best estimates, as can be seen in Fig. 4 where the seasonal sources of OC and EC are presented. The results of the sensitivity analysis are presented in Figs. 5 and 6, and will be discussed further in Sect. 3.5.

3.4.1 Biomass burning contribution to TC

Biomass burning only contributes slightly to the organic aerosol during the summer (<5%), while it is the major contributor during the winter, accounting for an average of 38% of TC (Fig. 4 and Table 5). The absolute contribution of biomass burning to TC is on average about $1 \mu\text{g m}^{-3}$ during winter (Table 5). This is low compared to results from Norwegian suburban and urban background stations (Yttri et al., 2009), the Hungarian continental site K-Puszt (May et al., 2009) and the Portuguese site Aveiro (Gelencsér et al., 2007).

3.4.2 Fossil contribution to TC

The contribution to the carbonaceous aerosol from fossil fuels is lower during the summer than in the winter, while EC resulting from fossil fuel combustion is rather stable throughout the year (Table 5). This leads to a varying OC/EC ratio of the fossil carbon from approximately 1 in the summer to >5 in the winter. This could be explained by the lower temperature in the winter altering the gas-particle equilibrium and thus suggests that a larger portion of the fossil OC during winter is secondary aerosol. It could also indicate a seasonal change in fossil fuel sources of the organic aerosol over the year. In this study, no method to distinguish between different fossil sources was applied.

The determination of fossil influence on TC is relying on the ^{14}C concentration of the sample. The lower $F^{14}\text{C}$ found in the winter cannot be interpreted as anything else but increased influence by fossil fuels.

Source apportionment of carbonaceous aerosol

J. Genberg et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

⏪

⏩

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



3.4.3 Biogenic contribution to TC

OC_{bio} is, as expected, higher during summer than in winter. In the summer, 82 % of the total carbon in the organic aerosol is of non-fossil origin, and is due to SOA from biogenic VOCs and primary particles, e.g. pollen. However, even in the coldest months, the model attributes on average 29 % of TC to biogenic sources. There may be several reasons why biogenic OC is found during the winter when most plants in Sweden are dormant. One reason could be long-range transport of biogenic SOAs or VOCs from the European continent, where VOCs are emitted during the winter months. However, the winter emission of monoterpene in central Europe is estimated to be only about 20 % of the summer peak (Zemankova and Brechler, 2010), and therefore emissions alone cannot fully explain why the biogenic organic aerosol in southern Sweden during the winter is one third of the summer concentration. Increased condensation may be a factor, with colder temperature driving the semivolatile organic compounds (SVOCs) into the particle phase. Such a possibility was demonstrated by (Simpson et al., 2007) but as noted there, there are too few measurements to verify or falsify such effects. Another explanation could be that the source apportionment method used here does not separate secondary OC from by precursors emitted by biomass burning from non-anthropogenic sources of OC_{bio} . Some of the carbon, which according to the model originates from biogenic sources, may thus be SOA formed from VOCs emitted by the burning of wood or biofuels e.g. ethanol combustion. This would further increase the effect of biomass burning on the organic aerosol mass loading during the winter.

3.4.4 Results of the sensitivity analysis

Figure 5 shows the results of the sensitivity analyses for four samples representing the four seasons. Each source of carbon in the sample is illustrated as a probability distribution. The lines represent the contribution of each of the sources to TC in that individual sample. The spread in the results increases drastically with increasing influence of biomass burning, i.e. during the winter, when the levoglucosan levels are higher. This is most prominent for the source apportionment of EC, since both of the

Source apportionment of carbonaceous aerosol

J. Genberg et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



**Source
apportionment of
carbonaceous
aerosol**

J. Genberg et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion

most uncertain parameters, $\text{lev}/\text{OC}_{\text{bb}}$ and $(\text{EC}/\text{OC})_{\text{bb}}$, are being used (Table 2). Both these parameters are dependent on the burning conditions and the type of biomass in question, as discussed in Sect. 2.6. The highly uncertain source apportionment of EC can be seen in the winter sample in Fig. 5a where fossil sources are estimated to contribute to between 0 to almost 100% of EC. The use of levoglucosan as a tracer for biomass combustion is also affected by its atmospheric stability. In a recent publication, the modelled chemical half-life for levoglucosan was found to be between 72 and 84 h in winter conditions (Hoffmann et al., 2010). This estimate is lower than values reported earlier (Locker, 1988; Fraser and Lakshmanan, 2000), which indicates a further uncertainty in source apportionment based on levoglucosan. More reliable data would be obtained if ^{14}C analyses could be performed on the pure EC fractions of the samples. In this study, the OC and EC were not separated prior to ^{14}C analysis, although this has been done in earlier studies (Szidat et al., 2004, 2006). Separating OC and EC has the advantage that the EC resulting from biomass burning is not dependent on the levoglucosan values. It also allows the OC from biomass burning to be calculated from both levoglucosan values and the $(\text{OC}/\text{EC})_{\text{bb}}$ ratio. The considerable uncertainty in source apportionment during periods with high levoglucosan values demonstrates the importance of separating OC and EC prior to ^{14}C analysis. However, no reliable technique for this separation was available at the time of this study. During periods of low levoglucosan concentrations, e.g. the summer sample in Fig. 5c, the source apportionment uncertainty is much smaller, which shows the low uncertainty in the non-biomass burning related factors used in the calculations.

Figure 6 shows the result of the total sensitivity analysis where the best estimates (median values) are presented as filled diamonds and the error bars represent 5th and 95th percentiles of the simulation results. As shown in Fig. 5, the uncertainties of all components are higher when levoglucosan is found in the samples. However, the data show conclusively that the increase in EC during the winter is mainly due to biomass burning, while the influence of fossil fuel combustion on EC is more stable over the year. It is also evident that OC_{ff} increases during the winter, based on ^{14}C data.

3.5 Model comparison

The calculated sources of OC were compared with results from the EMEP PCM model (Fig. 7). The modelled total OC and the measurements of OC coincide well although the model tends to underestimate the OC, especially for winter and autumn (Fig. 7a).

The model underestimates the biomass burning OC by a factor of about eight compared to the measurements, as an average for the whole year (Fig. 7b). This large underestimation is especially important for the winter samples where levoglucosan levels, and thus biomass burning OC, are high. The underestimation can be due to large-scale problems in the emission inventory or to some local wood-burning activities near the measurement site, which are not captured in the regional scale EMEP model. The biofuel emission factors, used in the emission inventory, are much lower for Sweden than for the neighbouring countries (Nielsen et al., 2010). This discrepancy can partly explain the model's underestimation of total OC during winter and autumn.

Similar problems, with too low modelled biomass burning OC, were found by Simpson et al. (2007) for other sites in Europe. Like Simpson et al. (2007), we have investigated the possible importance of improving the wood-burning emission estimate by making a simple rescaling of the modelled biomass burning OC, using the yearly average ratio between observed and modelled concentrations (i.e. increasing the modelled biomass burning OC by a factor of 8.2), and looked at the impact on total OC (Fig. 8). Without rescaling, the model underestimates total OC by about 27% and the correlation coefficient between observed and modelled OC is low ($r = 0.36$). Rescaling the modelled biomass burning OC leads to a small bias for total OC (+0.2%) and also better correlation ($r = 0.59$). This suggests that improvements in the national wood burning emission inventory may be an important step towards more accurate modelling of carbonaceous aerosol. Measurements at other sites, and higher resolution modelling, are needed to determine if the emission estimates are too low in general or if it is a local scale problem.

The fossil contributions to OC in the EMEP model consist of anthropogenic SOA and primary OC emissions from fossil fuel combustion. For the summer period the model

Source apportionment of carbonaceous aerosol

J. Genberg et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion

systematically overestimated fossil OC compared to the measurements (Fig. 7c). However for most of the winter and some autumn samples the fossil OC is underestimated. It should be noted that although ^{14}C measurements from a filter are precise, the possibility of contamination with “hot-carbon” (carbon containing much higher concentration than found in the atmosphere) exists close to nuclear installations, facilities for incineration of low-level radioactive waste and anthropogenic activities using ^{14}C as a tracer, and this would lead to an overestimation of the biogenic sources (Bench et al., 2007; Stenström et al., 2010). Contamination of the samples to imply a higher fossil influence is only possible if large amounts of fossil carbon were added to the samples, which would appear as elevated TC values. A more likely explanation for the incorrect seasonal variation of the model fossil OC is that the emissions of primary OC may be overestimated in summer and under-estimated in winter. Given the paucity of experimental verification of such emissions, and the myriad and seasonally differing sources of OC, such uncertainties are not surprising.

In summer the modelled biogenic contributions to OC, which are generally the largest fraction of TC, are reasonably well correlated with observations, but the model underpredicts biogenic OC by up to a factor of about two (Fig. 7d). This level of agreement is well within the uncertainties of the emissions of BVOC (Simpson et al., 1999; Rinne et al., 2009). In this comparison the model's biogenic SOA and background OC is considered to be biogenic OC. Most of the model background OC is assumed to originate from primary biogenic particles, but it also includes particles emitted from the ocean and wild fires. For winter and autumn, the correlation between the model and the measurements of biogenic OC is very low ($r = 0.08$). This is probably partly due to the very simplified model treatment of the background OC, which dominates the models biogenic OC during these periods. The discrepancy may also be related to errors in the measured biogenic OC. During winter, when wood burning is an important source of OC, the measured biogenic OC fraction is dependent on the highly uncertain conversion factors used to calculate OC_{bb} from levoglucosan concentrations.

Source apportionment of carbonaceous aerosol

J. Genberg et al.

[Title Page](#)[Abstract](#)[Introduction](#)[Conclusions](#)[References](#)[Tables](#)[Figures](#)[Back](#)[Close](#)[Full Screen / Esc](#)[Printer-friendly Version](#)[Interactive Discussion](#)

4 Conclusions

The sources of carbonaceous aerosol particles in background air in southern Sweden change significantly with the seasons. During summer, the total carbon is dominated by biogenic sources of OC, while fossil fuel combustion is the predominant source of EC. In winter, OC and EC originate mainly from biomass burning, as well as fossil fuel combustion. These findings are similar to those found in earlier studies. Gelencsér et al. (2007) reported similar results for the low altitude sites K-Pusztá (Hungary) and Aveiro (Portugal), although those summer aerosols were not as dominated by biogenic sources as at Vavihill. Just as for Vavihill, the winter samples collected at K-Pusztá and Aveiro were dominated by biomass burning. The fractions are similar at the continental and Swedish sites, although the TC mass is substantially higher at the continental sites. The present study shows that the sources of carbonaceous aerosols in southern Sweden are similar to those at other low altitude sites in Europe. This study adds a full year of source apportionment of carbonaceous aerosols at a part of Europe where such data are previously unavailable.

Comparison between the measurements and the EMEP model showed good agreement for total OC, except for the winter period. However, the model and measurement do not agree to which extent the individual sources of OC contribute. Biomass burning is a large source during winter according to the measurements, with contributions about eight times larger than that suggested by the model. This indicates a strong underestimation of some Swedish and possibly international biomass burning emissions. When the biomass burning contribution in the model was scaled to fit the measured OC_{bb} concentration, the total OC estimation was significantly improved as well. Fossil sources are also underestimated by the model during winter while they are overestimated during summer. More work is needed to resolve the remaining discrepancy between measurements and models of organic aerosols, both regarding the models and the measurement based source apportionment. An important area for future work is evaluation and improvement of emission inventories of carbonaceous aerosols, especially for wood-burning and biogenic emissions of particles and VOCs.

Source apportionment of carbonaceous aerosol

J. Genberg et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



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Source apportionment of carbonaceous aerosol

J. Genberg et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



Source apportionment of carbonaceous aerosol

J. Genberg et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

⏪

⏩

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



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Source apportionment of carbonaceous aerosol

J. Genberg et al.

[Title Page](#)
[Abstract](#)
[Introduction](#)
[Conclusions](#)
[References](#)
[Tables](#)
[Figures](#)




[Back](#)
[Close](#)
[Full Screen / Esc](#)
[Printer-friendly Version](#)
[Interactive Discussion](#)

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Source apportionment of carbonaceous aerosol

J. Genberg et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

⏪

⏩

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion

2009.

Kristensson, A., Dal Maso, M., Swietlicki, E., Hussein, T., Zhou, J., Kerminen, V. M., and Kulmala, M.: Characterization of new particle formation events at a background site in Southern Sweden: relation to air mass history, *Tellus B*, 60, 330–344, doi:10.1111/j.1600-0889.2008.00345.x, 2008.

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Source apportionment of carbonaceous aerosol

J. Genberg et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



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J. Genberg et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

⏪

⏩

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



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Source apportionment of carbonaceous aerosol

J. Genberg et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



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**Source
apportionment of
carbonaceous
aerosol**

J. Genberg et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures



Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion

Source apportionment of carbonaceous aerosol

J. Genberg et al.

Table 1. Ratios of levoglucosan/OC_{bb} and (EC/OC)_{bb} from other studies.

Levoglucosan/OC _{bb}			(EC/OC) _{bb}			Reference
Minimum	Maximum	Mean	Minimum	Maximum	Mean	
0.05	0.17	0.1±0.04	0.04	0.43	0.15±0.15	Fine et al. (2001)
0.036	0.159	0.1±0.05	0.08	0.18	0.13±0.05	Fine et al. (2002)
0.1	0.33	0.16±0.1	0.016	0.45	0.09±0.14	Fine et al. (2004a)
0.125	0.409	0.24±0.1	0.06	0.38	0.21±0.13	Fine et al. (2004b)
0.23	0.52	0.34±0.16	0.025	0.059	0.046±0.02	Schauer et al. (2001)
0.036	0.54	0.28±0.21	–	–	–	Engling et al. (2006)
0.047	0.159	0.108	–	–	0.04	Zdráhal et al. (2002)

[Title Page](#)
[Abstract](#)
[Introduction](#)
[Conclusions](#)
[References](#)
[Tables](#)
[Figures](#)
[Back](#)
[Close](#)
[Full Screen / Esc](#)
[Printer-friendly Version](#)
[Interactive Discussion](#)

Source apportionment of carbonaceous aerosol

J. Genberg et al.

Table 2. Parameters used in the randomised sampling simulation.

Parameter	Low	Most probable value	High	Distribution
$(\text{EC}/\text{OC})_{\text{bb}}$	0.06	0.15	0.45	Beta ($\alpha = 1.12$, $\beta = 1.52$)
$\text{lev}/\text{OC}_{\text{bb}}$	0.08	–	0.2	uniform
$\text{F}^{14}\text{C}_{\text{bb}}$	1.06	1.2	1.23	Beta ($\alpha = 1.8$, $\beta = 1.2$)
OC factor	0.9	1	1.1	Beta ($\alpha = 1.5$, $\beta = 1.5$)
EC factor	0.8	1	1.2	Beta ($\alpha = 1.5$, $\beta = 1.5$)
$\text{F}^{14}\text{C}_{\text{TC}}$	$\bar{x} - 2\sigma$	\bar{x}	$\bar{x} + 2\sigma$	Beta ($\alpha = 1.5$, $\beta = 1.5$)
Levogluconan	$\bar{x} - 2\sigma$	\bar{x}	$\bar{x} + 2\sigma$	Beta ($\alpha = 1.5$, $\beta = 1.5$)

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



Source apportionment of carbonaceous aerosol

J. Genberg et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



Table 3. Results from denuder tests at Vavihill in 2008. Values of OC are presented as $\mu\text{g m}^{-3}$. The filters were positioned behind a Teflon filter which removes particulate matter but not VOCs. The material collected on the filters is therefore the positive artefact from VOCs. The correction factor for undenuded filter samples was determined to be 2.16.

Season	Front OC	Back OC	Correction factor
Winter	0.73	0.30	2.43
Winter	0.48	0.26	1.85
Winter	0.73	0.19	3.84*
Summer	0.60	0.23	2.61
Summer	0.36	0.18	2.00
Summer	0.90	0.46	1.96
Summer	0.96	0.44	2.18
Summer	0.70	0.34	2.06
Mean	0.68	0.32	2.16

* This sample was considered an outlier using Grubbs' test ($P < 0.05$), and was therefore not used in the calculation of the mean.

**Source
apportionment of
carbonaceous
aerosol**

J. Genberg et al.

[Title Page](#)[Abstract](#)[Introduction](#)[Conclusions](#)[References](#)[Tables](#)[Figures](#)[Back](#)[Close](#)[Full Screen / Esc](#)[Printer-friendly Version](#)[Interactive Discussion](#)**Table 4.** Definition of seasons determined from a line fitted to temperature data.

Season	Start	Stop
Spring 2008	24 April 2008	3 May
Summer	3 May	25 September
Autumn	25 September	19 December
Winter	19 December	4 March 2009
Spring 2009	4 March	22 April

Source apportionment of carbonaceous aerosol

J. Genberg et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

⏪

⏩

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



Table 5. Seasonal means of medians, 5th and 95th percentiles of the source apportionment simulations (unit: $\mu\text{g m}^{-3}$). “% of TC” represents the means of medians’ contribution to TC.

Season		OC	EC	OC _{bio}	OC _{bb}	OC _{ff}	EC _{bb}	EC _{ff}
Winter	Median	2.25	0.31	0.73	0.79	0.71	0.18	0.13
	5th percentile	2.07	0.26	0.36	0.57	0.55	0.06	0.04
	95th percentile	2.43	0.36	1.08	1.13	0.89	0.29	0.27
	% of TC	88.7	12.3	28.8	31.1	27.9	7.2	4.9
Spring	Median	1.13	0.28	0.78	0.19	0.15	0.04	0.23
	5th percentile	1.04	0.23	0.59	0.13	0.09	0.01	0.16
	95th percentile	1.22	0.32	0.92	0.32	0.23	0.09	0.29
	% of TC	81.0	19.8	56.2	13.7	10.6	3.0	16.6
Summer	Median	1.53	0.14	1.37	0.07	0.09	0.01	0.13
	5th percentile	1.41	0.12	1.24	0.05	0.07	0.01	0.10
	95th percentile	1.65	0.17	1.49	0.11	0.12	0.03	0.15
	% of TC	91.8	8.6	81.9	4.0	5.6	0.9	7.6
Autumn	Median	1.50	0.27	0.88	0.39	0.22	0.08	0.19
	5th percentile	1.38	0.23	0.61	0.26	0.14	0.03	0.09
	95th percentile	1.62	0.31	1.11	0.59	0.33	0.17	0.26
	% of TC	85.1	15.4	50.2	21.9	12.6	4.7	10.6

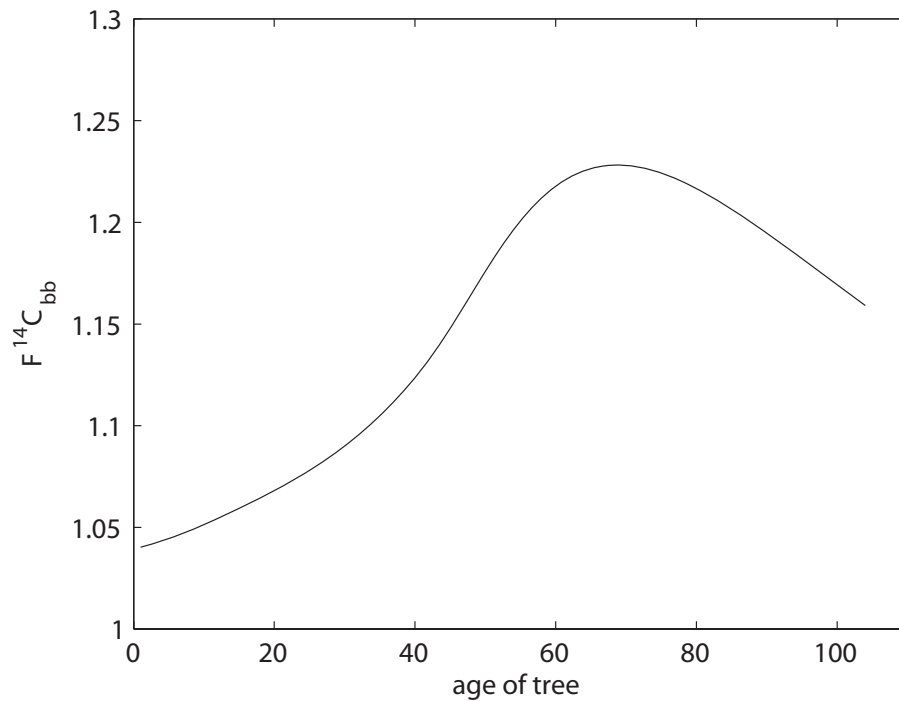


Fig. 1. Simulated $F^{14}C$ values for trees harvested in 2008 as a function of tree age.

**Source
apportionment of
carbonaceous
aerosol**

J. Genberg et al.

[Title Page](#)

[Abstract](#) | [Introduction](#)

[Conclusions](#) | [References](#)

[Tables](#) | [Figures](#)

[⏪](#) | [⏩](#)

[◀](#) | [▶](#)

[Back](#) | [Close](#)

[Full Screen / Esc](#)

[Printer-friendly Version](#)

[Interactive Discussion](#)



Source apportionment of carbonaceous aerosol

J. Genberg et al.

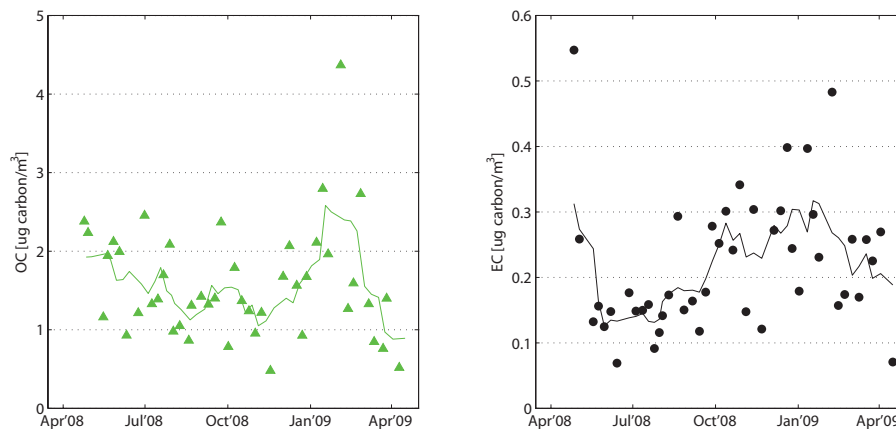


Fig. 2. Organic carbon (OC) on the left hand side and elemental (EC) carbon concentrations on the right hand side measured from filters collected at Vavihill in southern Sweden. The lines illustrate the average of the closest five points. Note the difference in scales in the two graphs. EC generally constitutes 10 to 15% of the total carbon, the rest being OC.

[Title Page](#)[Abstract](#)[Introduction](#)[Conclusions](#)[References](#)[Tables](#)[Figures](#)[⏪](#)[⏩](#)[◀](#)[▶](#)[Back](#)[Close](#)[Full Screen / Esc](#)[Printer-friendly Version](#)[Interactive Discussion](#)

Source apportionment of carbonaceous aerosol

J. Genberg et al.

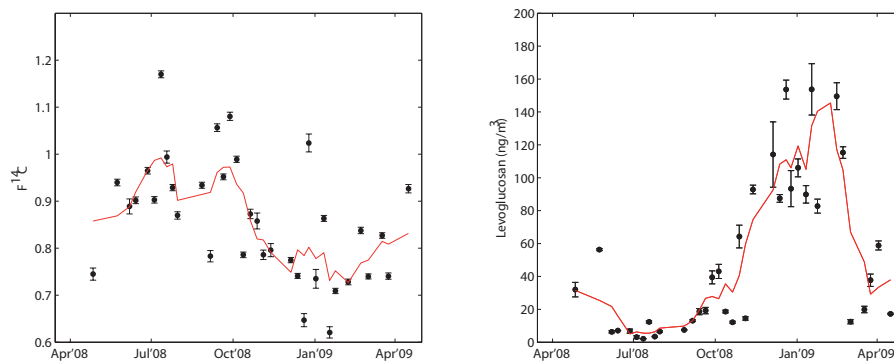


Fig. 3. ^{14}C concentration (expressed as $F^{14}C$) on the left side and levoglucosan concentration on the right. Measurements from filters collected at Vavihill from April 2008 to April 2009. The error bars illustrate one standard deviation and the red lines illustrate the average of the closest five points.

[Title Page](#)[Abstract](#)[Introduction](#)[Conclusions](#)[References](#)[Tables](#)[Figures](#)[◀](#)[▶](#)[◀](#)[▶](#)[Back](#)[Close](#)[Full Screen / Esc](#)[Printer-friendly Version](#)[Interactive Discussion](#)

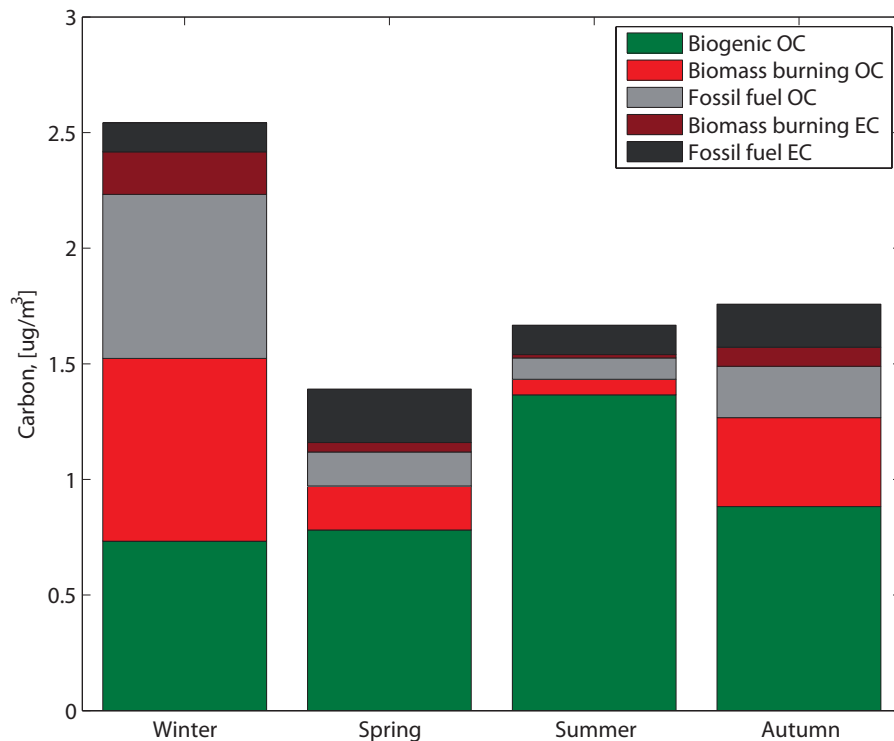


Fig. 4. Absolute contributions to the total carbon from the five sources investigated: biogenic OC, biomass burning OC and EC, and fossil fuel OC and EC. The mean of the median values from the sensitivity analyses (Table 5) were used in the diagram.

Source apportionment of carbonaceous aerosol

J. Genberg et al.

Title Page

Abstract Introduction

Conclusions References

Tables Figures

⏪ ⏩

◀ ▶

Back Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



Source apportionment of carbonaceous aerosol

J. Genberg et al.

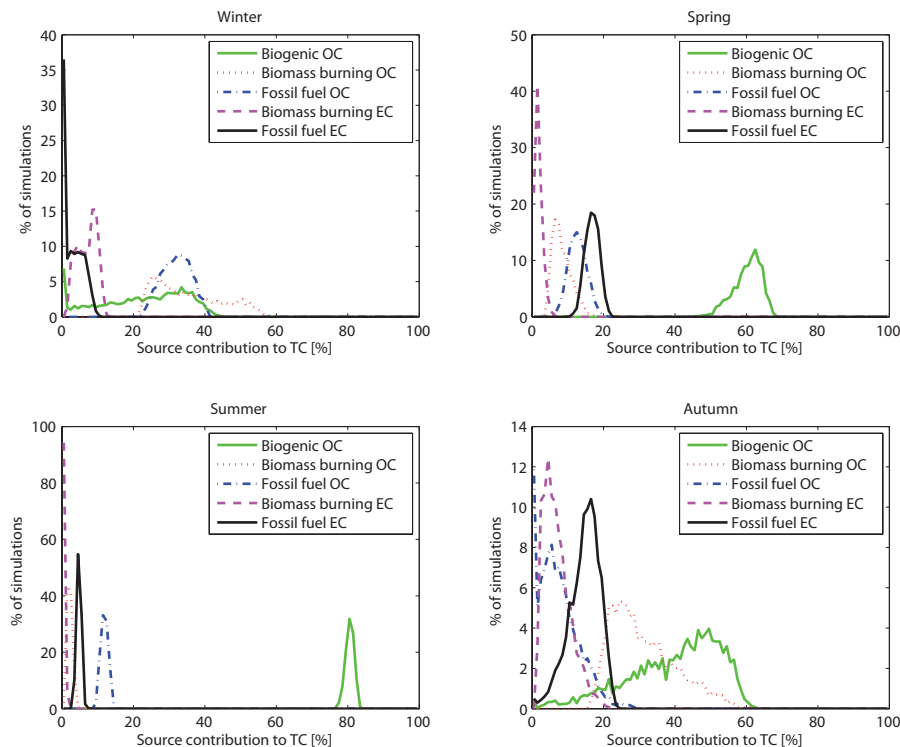


Fig. 5. Results from the sensitivity analyses of the source contributions of OC and EC as probability distributions. Each figure represents one representative filter sample from each season, **(a)** winter (3–11 February 2009), **(b)** spring (24–28 April 2008), **(c)** summer (28 July–1 August 2008) and **(d)** autumn (24–31 October 2008). The lines correspond to all the outcome of the simulation for the contribution of each source to TC.

[Title Page](#)
[Abstract](#)
[Introduction](#)
[Conclusions](#)
[References](#)
[Tables](#)
[Figures](#)
[⏪](#)
[⏩](#)
[◀](#)
[▶](#)
[Back](#)
[Close](#)
[Full Screen / Esc](#)
[Printer-friendly Version](#)
[Interactive Discussion](#)

**Source
apportionment of
carbonaceous
aerosol**

J. Genberg et al.

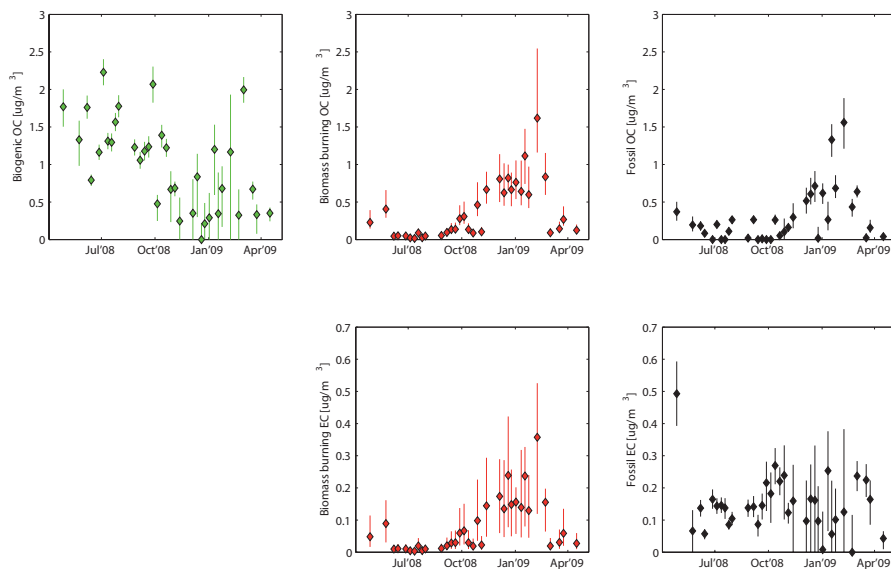


Fig. 6. Full results from the sensitivity analysis, **(a)** biogenic OC, **(b)** biomass burning OC, **(c)** fossil OC, **(d)** biomass burning EC, **(e)** fossil EC. The diamonds represent the median of the simulation results and the error bars represent 5th and 95th percentiles.

[Title Page](#)[Abstract](#)[Introduction](#)[Conclusions](#)[References](#)[Tables](#)[Figures](#)[◀](#)[▶](#)[◀](#)[▶](#)[Back](#)[Close](#)[Full Screen / Esc](#)[Printer-friendly Version](#)[Interactive Discussion](#)

**Source
apportionment of
carbonaceous
aerosol**

J. Genberg et al.

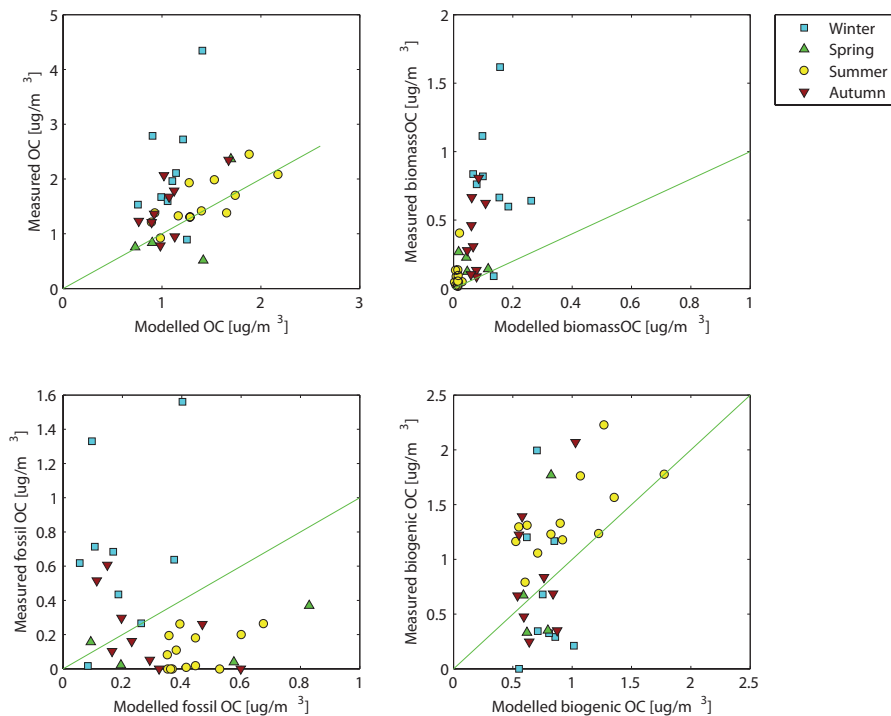


Fig. 7. Comparison between modelled (x-axis) and “measured” (y-axis) contribution to OC. Each figure represents one or more sources of OC: **(a)** total OC, **(b)** biomass burning OC, **(c)** fossil OC and **(d)** biogenic OC. The green lines represent the 1:1 lines.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion

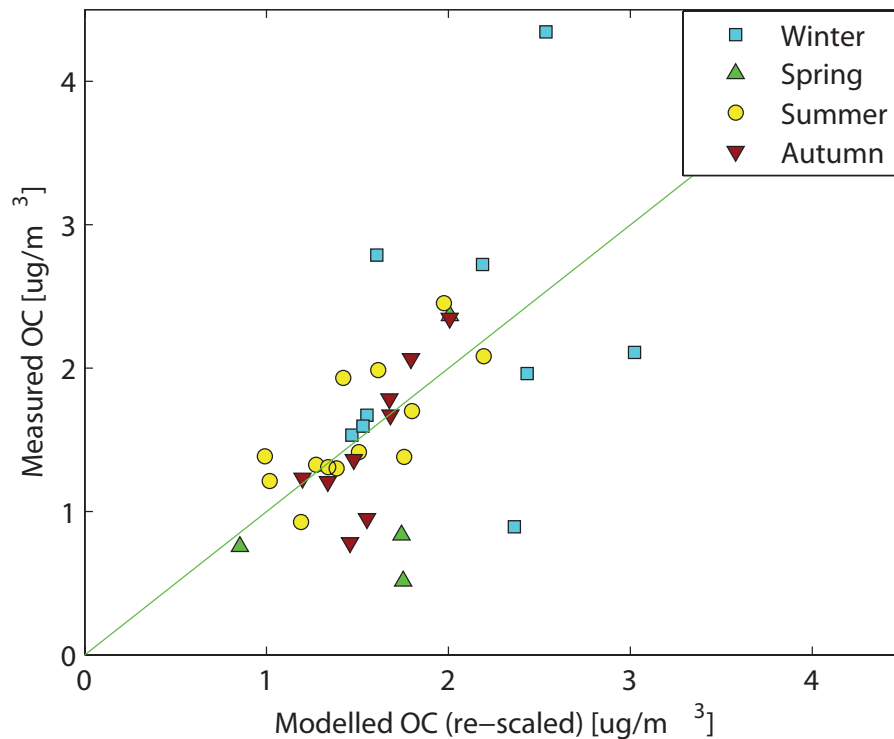


Fig. 8. Comparison between re-scaled modelled (x-axis) and “measured” (y-axis) contribution to total OC. The modelled biomass burning contribution to OC has for the purpose of this figure been multiplied by 8.2 to fit the measured OC_{bb} concentrations.

Source apportionment of carbonaceous aerosol

J. Genberg et al.

Title Page

Abstract Introduction

Conclusions References

Tables Figures

⏪ ⏩

◀ ▶

Back Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion

