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## ***Interactive comment on “Anthropogenic imprints on nitrogen and oxygen isotopic composition of precipitation nitrate in a nitrogen-polluted city in southern China” by Y. T. Fang et al.***

### **Anonymous Referee #2**

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Review of “Anthropogenic imprints on nitrogen and oxygen isotopic composition of precipitation nitrate in a nitrogen-polluted city in southern China” by Fang et al. Atmospheric Chemistry and Physics Discussions

### **General comments:**

Fang et al present an impressive dataset of two years of wet deposition (e.g. rain)  $\text{NO}_3^-$  N and O isotopic composition in a region of China characterized by heavy air pollution. The authors use concentration along with  $\delta^{15}\text{N}$  and  $\delta^{18}\text{O}$  in order to provide some constraint on seasonality of N sources to this landscape as well as

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atmospheric cycling of  $\text{NO}_x$ . Overall the manuscript is well written, well organized and clearly presented. The data seem to be of sound quality and the interpretations are, for the most part, well justified, within the bounds of our ability to characterize N sources and understand chemical cycling.

My primary criticism of this work, presented below, involves the invocation of the influence of a poorly characterized isotopic influence by a peroxy radical pathway for explaining 'anomalous'  $\delta^{18}\text{O}$  composition as well as only minimal attention given to the potential influence of other mechanisms that might be controlling variability of  $\delta^{15}\text{N}$  in  $\text{NO}_3^-$  found in rain.

### Specific Comments:

1. The role of a peroxy radical associated pathway of  $\text{NO}_x$  oxidation on  $\text{NO}_3^-$  isotopic composition is intriguing. However, I feel that, particularly because this may be a relatively new idea and/or novel to this type of highly polluted environment, there needs to be more support, references and discussion. Is this complete speculation? What is known about reactions involving peroxy radicals and their transfer of O atoms (and isotopic composition) to a  $\text{NO}_x$  pool? Is there any other evidence that supports this as a potentially important player in the  $\text{NO}_x$  cycle for this environment or any others? Again I feel that this may be an important finding/discussion point for the paper and the manuscript would benefit (and readers too!) from a more thorough discussion of the potential influence of this pathway.

2. What about other factors that might control  $\delta^{15}\text{N}$  of  $\text{NO}_3^-/\text{HNO}_3$  deposition and also then the potential for seasonality of these controls (e.g., temperature, humidity, etc.) for exercising some power over the seasonality observed in  $\delta^{15}\text{N}$  of  $\text{NO}_3^-$ . For example,

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- a. Higher ozone in polluted areas promotes higher  $\text{NO}_2$  concentrations and lower  $\delta^{15}\text{N}$  of  $\text{NO}_3^-$  (Freyer et al., 1993). Thus, could seasonal variation in the  $\delta^{15}\text{N}$  of  $\text{NO}_3^-$  be related to changes in the relative sizes of the  $\text{NO}$  and  $\text{NO}_2$  pools as it relates to  $\text{O}_3$ ?
- b. Changes in temperature and/or humidity can influence interactions between  $\text{HNO}_3$  vapor and aerosol  $\text{NO}_3^-$ . Could seasonal changes in temperature be influencing the partitioning of N between these two forms and hence influence the isotopic composition of the component in rain?
- c. How high is the particulate load? Is the  $\text{NO}_3^-$  particle associated? Particles can be important reactive surfaces – any insight into variability of the particulate composition and/or seasonality or concentrations?
- d. The precipitation samples contained substantial  $\text{NH}_4^+$ . Therefore, I wonder about the dynamic equilibrium between  $\text{NH}_4\text{NO}_3$  and  $\text{HNO}_3$  and the implications for isotopic exchange? Under conditions of high temperatures and high humidity the model of Morino et al. 2006 would predict that most of the  $\text{NO}_3^-$  would be in the form of particulate  $\text{NO}_3^-$  (vs  $\text{HNO}_3$  vapor). The isotopic discrepancy between these two pools could have important implication for measured  $\text{NO}_3^-$  in rain as several studies have shown (Baker et al., 2007; Elliott et al., 2009; Freyer 1991).

3. There seems to be a lot of attention and comparison with other studies of atmospheric  $\text{NO}_3^-$  found in precipitation. However, I feel that there was more attention given to high latitude studies than needed. In fact, reference to the studies of arctic  $\text{NO}_x$  cycling may not be pertinent to your study at all. There are data available from similar latitudes (Morin et al., 2009 - wide range of latitudes; Hastings et al., 2003, Bermuda 32N; Wankel et al., 2010, Israel 29N) that might provide a better basis for comparison in your discussion.

### Technical Corrections:

P21440, L4: Don't need to mention the use of the denitrifier method here.

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P21440, L17: Should this be “altitude” or “latitude?”

P21441, L6: You define  $\text{NO}_x$  as a ‘sum of  $\text{NO}$  and  $\text{NO}_2$ ’ as is traditionally done. However, a sum is a singular object and should be used grammatically so throughout the paper. “ $\text{NO}_x$  contributes”, “ $\text{NO}_x$  is”, “ $\text{NO}_x$  dominates” – and so on.

P21441, L7: “. . .and are thus important factors. . .”

P21441, L17: Awkward sentence. Rephrase “The dominant source of  $\text{NO}_x$  emissions is expected to have shifted. . .”

P21441, L19: “. . .due to increased vehicle usage and power plant emissions.”

P21441, L23: Not sure the Figure 1 is needed.

P21441, L24: Not clear how you can expect to control anthropogenic production of  $\text{NO}_x$  from natural processes such as lightning?

P21442, L2: “ $\text{NO}_x$  is. . .”

P21442, L5: “ $\text{NO}_x$  is. . .”

P21442, L22: “Nitrogen stable isotopic composition of atmospheric deposition. . .”

P21443, L2: “vehicular  $\text{NO}_x$  emissions measured from tailpipe. . .”

P21443, L29: “This explains the higher  $\delta^{18}\text{O}$  . . .”

P21444, L1: “in winter than in summer in previous studies”

P21444, L7: “and proximity of stationary source. . .”

P21444, L9: Elliott et al did not use ‘dual’ isotopes – only  $\delta^{15}\text{N}$ .

P21447, L20: “Annual N load from precipitation is the sum. . .”

P21448, L16: falling

P21448, L18: falling

P21450, L7: Is this mean a flux weighted value? P21450, L26: “. . .was also found to significantly influence. . .”

P21452, L22: “Partitioning of  $\text{NO}_x$  sources using. . .”

P21453, L14: “using a simple two end-member mixing model.”

P21543, L14: Were flux-weighted average used for this mixing analysis?

P21453, L28: “may be higher than our assumed value. . .”

P21455, L9: “donates fewer O atoms”

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P21455, L22: “When the air temperature...” P21456, L12: provide reference for this statement.

P21456, L15: provide reference for statement.

P21456, L17: provide reference for this statement. Please explain why the DMS/HC pathway would necessarily lead to higher  $\delta^{18}\text{O}$  than the  $\text{N}_2\text{O}_5$  pathway.

P21457, L4-11: Invoking a peroxy radical pathway is a potentially important explanation/finding/conclusion – but there needs to be more discussion about the supporting work done on the influence of this pathway on the isotopic composition of  $\text{NO}_3^-$ .

Table 1, 2, and 3: No need to report arithmetic mean. Should be using flux-weighted means for all calculations involving  $\delta^{15}\text{N}$  – in particular the mixing model calculations. There also appears to be some slight disagreement among Tables 1, 2, and 3 regarding weighted  $\delta^{15}\text{N}$  values.

Table 4: Not sure if this is needed. Table excludes Michalski et al., 2003, 2004; Hastings et al., 2004; Jarvis et al., 2008; Wankel et al., 2010. Savarino is misspelled.

Fig 1: Not sure if this is needed.

Fig 2: Recommend using shading for warm vs cool seasons.

Fig 3: Maybe also provide a map of the region?

Fig 6: When looking at this figure, I wondered whether this spread of data could be thought of as mixing? While recognizing a general lack of knowledge of isotope effects from the various oxidation reactions (which greatly limits our understanding of  $\Delta^{17}\text{O}$  of atmospheric  $\text{NO}_3^-$  in general), if this were to be plotted against inverse concentration (e.g. a Keeling plot) could this be used to help bracket the endmember composition for a mixing model? I also recognize that the concentration of  $\text{NO}_3^-$  in precipitation results from more complex processes than simple mixing of two endmembers, but perhaps this could be a point off of which to build some discussion (just a thought).

## REFERENCES

Baker, A., et al. (2007), Dry and wet deposition of nutrients from the tropical At-

lantic atmosphere: Links to primary productivity and nitrogen fixation, Deep Sea Research I, 54, 1704-1720.

Elliott, E. M., et al. (2009), Dual nitrate isotopes in dry deposition: Utility for partitioning  $\text{NO}_x$  source contributions to landscape nitrogen deposition, Journal of Geophysical Research, 114(G04020).

Freyer, H. (1991), Seasonal variations in  $^{15}\text{N}/^{14}\text{N}$  ratios in atmospheric nitrate species, Tellus, 43, 34-44.

Freyer, H., et al. (1993), On the interaction of isotopic exchange processes with photochemical reactions in the atmospheric oxides of nitrogen, Journal of Geophysical Research, 98, 14791-14796.

Hastings, M. G., et al. (2003), Isotopic evidence for source changes of nitrate in rain at Bermuda, Journal of Geophysical Research, 108(D24), 4790.

Jarvis, J. C., et al. (2008), Influence of local photochemistry on isotopes of nitrate in Greenland snow, Geophysical Research Letters, 35(L21804).

Michalski, G., et al. (2003), First measurements of  $\Delta^{17}\text{O}$  in atmospheric nitrate, Geophysical Research Letters, 30(16), 1870.

Michalski, G., et al. (2004), Tracing atmospheric nitrate deposition in a complex semi-arid ecosystem using  $\Delta^{17}\text{O}$ , Environmental Science Technology, 38, 2175-2181.

Morin, S., et al. (2009), Comprehensive isotopic composition of atmospheric nitrate in the Atlantic Ocean boundary layer from 65S to 79N, Journal of Geophysical Research, 114.

Morino, Y., et al. (2006), Partitioning of  $\text{HNO}_3$  and particulate nitrate over Tokyo: Effect of vertical mixing, Journal of Geophysical Research, 111(D15215).

Wankel, S. D., et al. (2010), Sources of aerosol nitrate to the Gulf of Aqaba: Evidence from  $\delta^{15}\text{N}$  and  $\delta^{18}\text{O}$  of nitrate and trace metal chemistry, Marine Chemistry, 120, 90-99.

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