

1 **Supplement for:**

2 **Terpenylic acid and related compounds: precursors for dimers in**
3 **secondary organic aerosol from the ozonolysis of α - and β -pinene**

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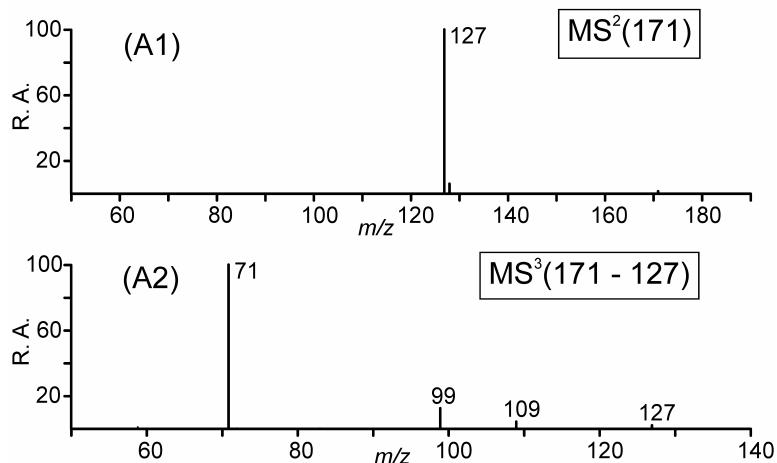
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20 **1. The K-puszta sampling site and 2006 BIOSOL summer field campaign**

21 K-puszta is a rural measurement site located on the Great Hungarian Plain (46°58'N,
22 19°35'E, 125 m a.s.l.), 15 km northwest from the nearest town Kecskemét, and 80 km
23 southeast from Budapest. The surroundings of the measurement site are dominated by
24 mixed forest (62% coniferous, 28% deciduous) and grassland (10%). The site is
25 characterized by intensive solar radiation during summer. The BIOSOL (Formation
26 mechanisms, marker compounds, and source apportionment for biogenic atmospheric
27 aerosols) campaign took place from 24 May to 29 June 2006. Maenhaut et al. (2008)
28 studied the composition of atmospheric particulate matter during the BIOSOL campaign
29 and observed that the campaign time could be divided into two periods: from the start of
30 the campaign until 11 June 2006 when it was unusually cold with daily maximum
31 temperatures between 12 and 23 °C, and from 12 June 2006 onward when the
32 temperatures were considerably higher with daily maxima ranging from 24 to 36 °C.
33 During the cold period the air masses came mainly from the northwest over the North Sea
34 or the Atlantic Ocean whereas after 12 June 2006, i.e., during the warm period, the air
35 masses were continental.

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37 **2. Mass spectral characterization of *cis*-norpinic acid**

38 (–)ESI-MS data for the MW 172 compound eluting at RT 19.8 min (Figs. 1 and 2), which
39 is attributed to *cis*-norpinic acid based on detailed interpretation of the (–)ESI-MS data
40 and comparison with LC/MS data reported in the literature (Glasius et al., 2000; Warnke
41 et al., 2006), are presented in Figure S1 and Scheme S1.

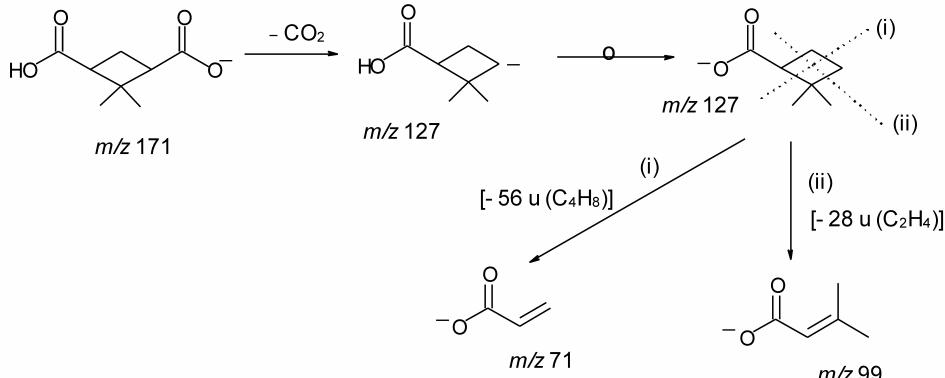


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43 **Figure S1:** (-)ESI-MSⁿ ($n = 2, 3$) data for the minor MW 172 compound
 44 attributed to *cis*-norpinic acid.

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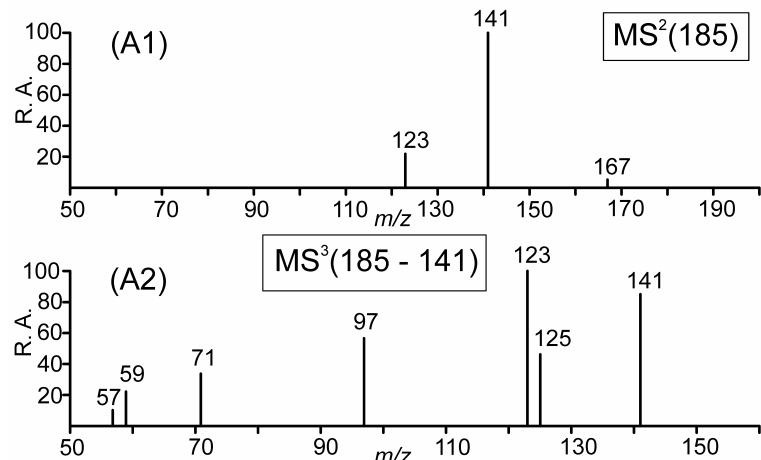
48 **Scheme S1:** Proposed fragmentation pathways for deprotonated *cis*-norpinic acid.

49 **3. Mass spectral characterization of homoterpenylic acid, a specific β -pinene SOA
 50 tracer, and possible formation pathway from β -pinene**

51 (-)ESI-MS data obtained for the MW 186 compound present in β -pinene SOA and
 52 tentatively assigned to homoterpenylic acid based on interpretation of (-)ESI-MS data are
 53 presented in Figure S2. Proposed fragmentation pathways are given in Scheme S2. The
 54 formation of terpenylic acid and diaterpenylic acid from α -pinene have been explained in
 55 previous work through both OH radical- and ozone-initiated reactions (Claeys et al.,
 56 2009); however, in the case of β -pinene we were only able to explain the formation of
 57 homoterpenylic acid through OH radical-initiated reactions (Scheme S3). We therefore
 58 suggest that the OH radical pathway is more likely to operate under ozone-initiated
 59 reactions, where no OH scavenger is used, as well as under ambient conditions. An
 60 additional question that arises is the formation of terpenylic acid and diaterpenylic acid
 61 acetate from β -pinene. A simple but possible explanation is that there is a fast OH

radical-initiated isomerization of β -pinene to α -pinene, prior to OH radical attack on the double bond.

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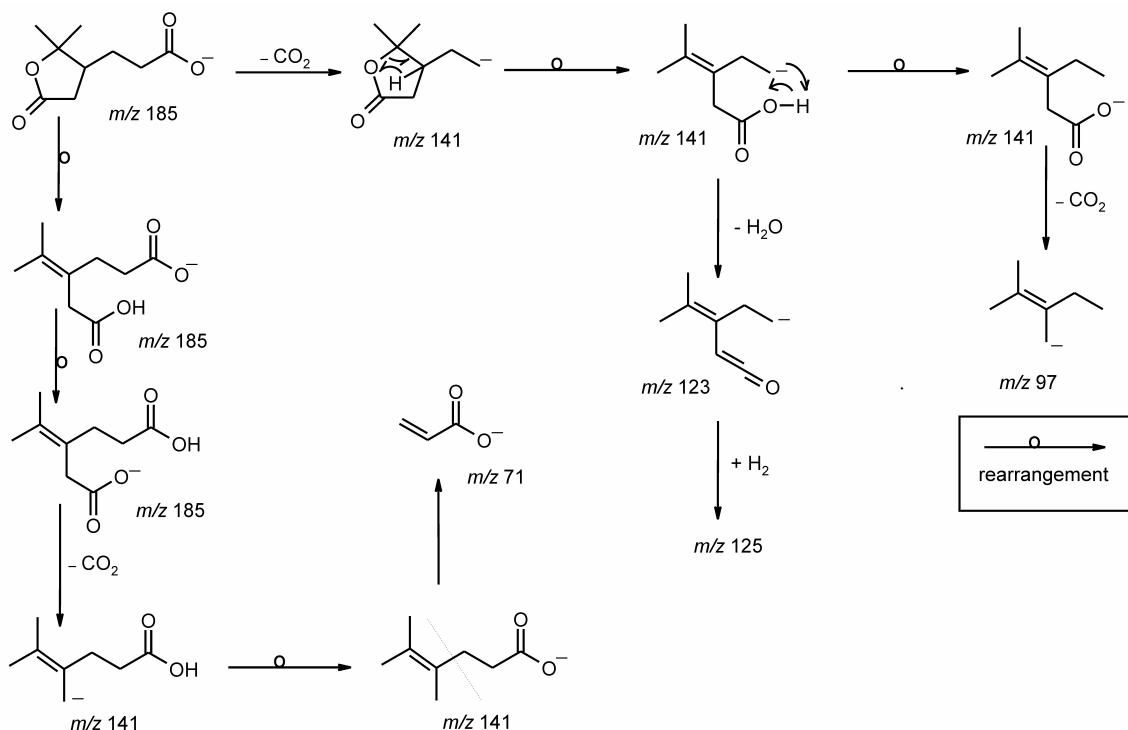
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Figure S2: (-)ESI-MSⁿ (n = 2, 3) data obtained for the MW 186 compound present in β -pinene SOA and tentatively assigned to homoterpenylic acid.



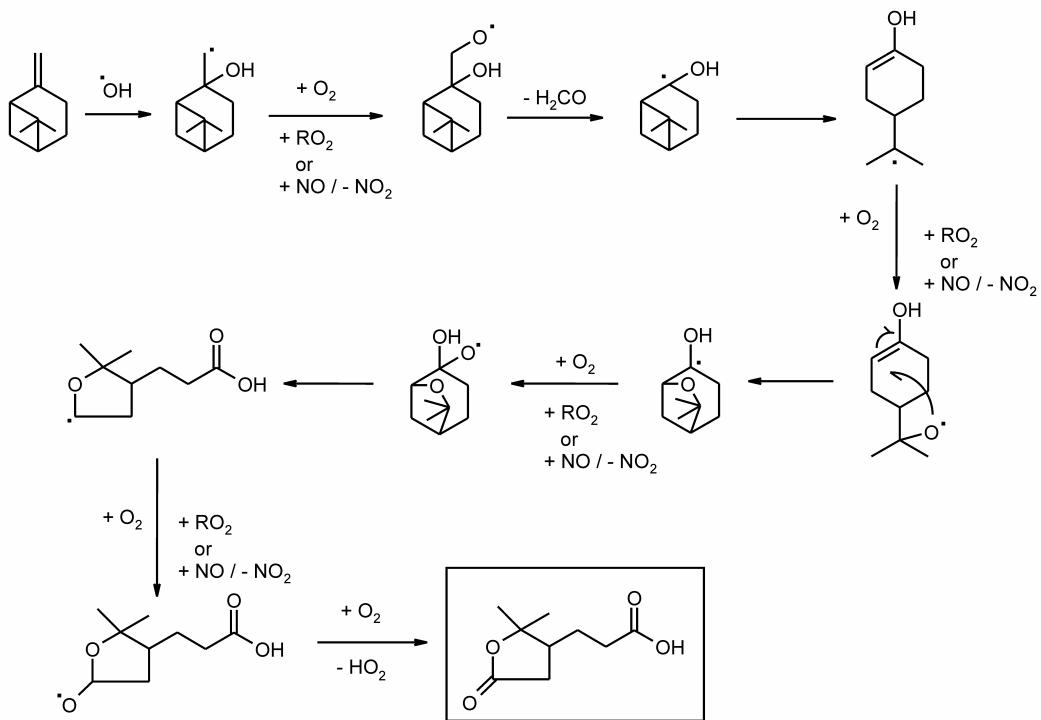
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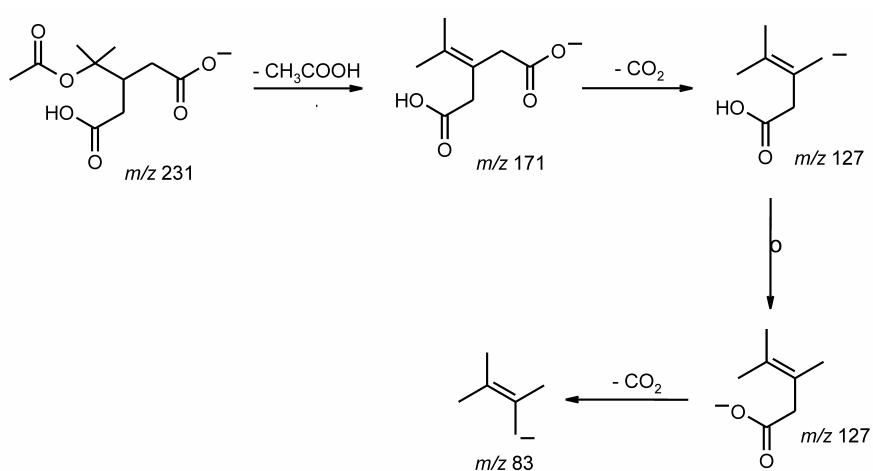
Scheme S2: Proposed fragmentation pathways for deprotonated homoterpenylic acid.



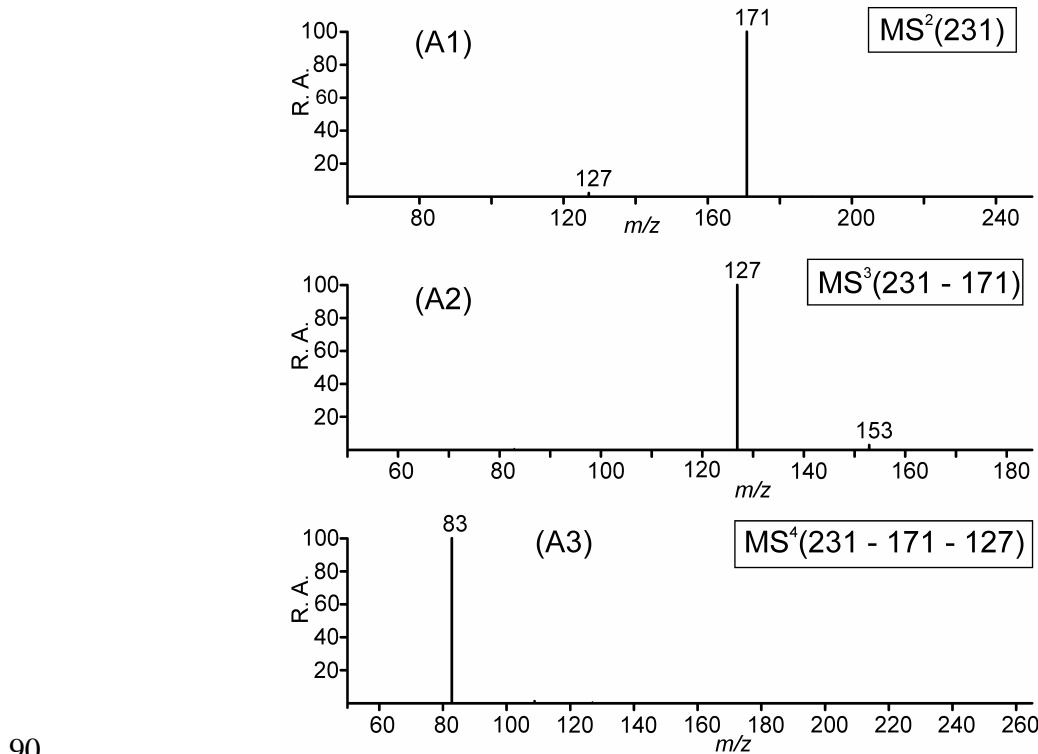
Scheme S3: Pathway leading to the formation of homoterpenylic acid from β -pinene through OH radical-initiated reactions.

4. Mass spectral characterization of diaterpenylic acid acetate and an isomeric MW 232 product

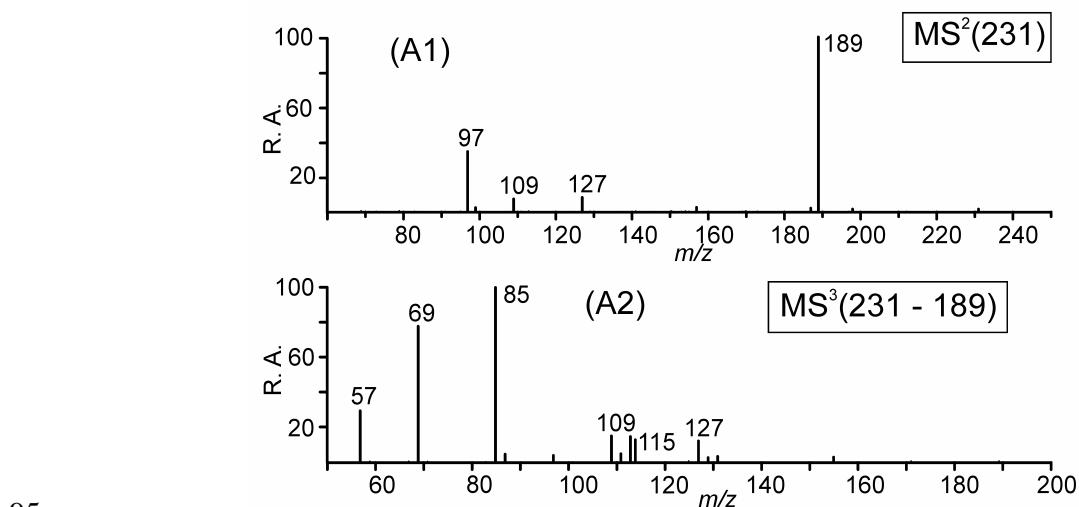
(-)ESI-MS data obtained for synthesized diaterpenylic acid acetate and for the isomeric MW 232 product with RT 22.1 min present in α -pinene SOA (Fig. 1) are presented in Figures S3 and S4, respectively. Proposed fragmentation pathways for deprotonated diaterpenylic acid acetate are given in Scheme S4.



Scheme S4: Proposed fragmentation pathways for deprotonated diaterpenylic acid acetate.

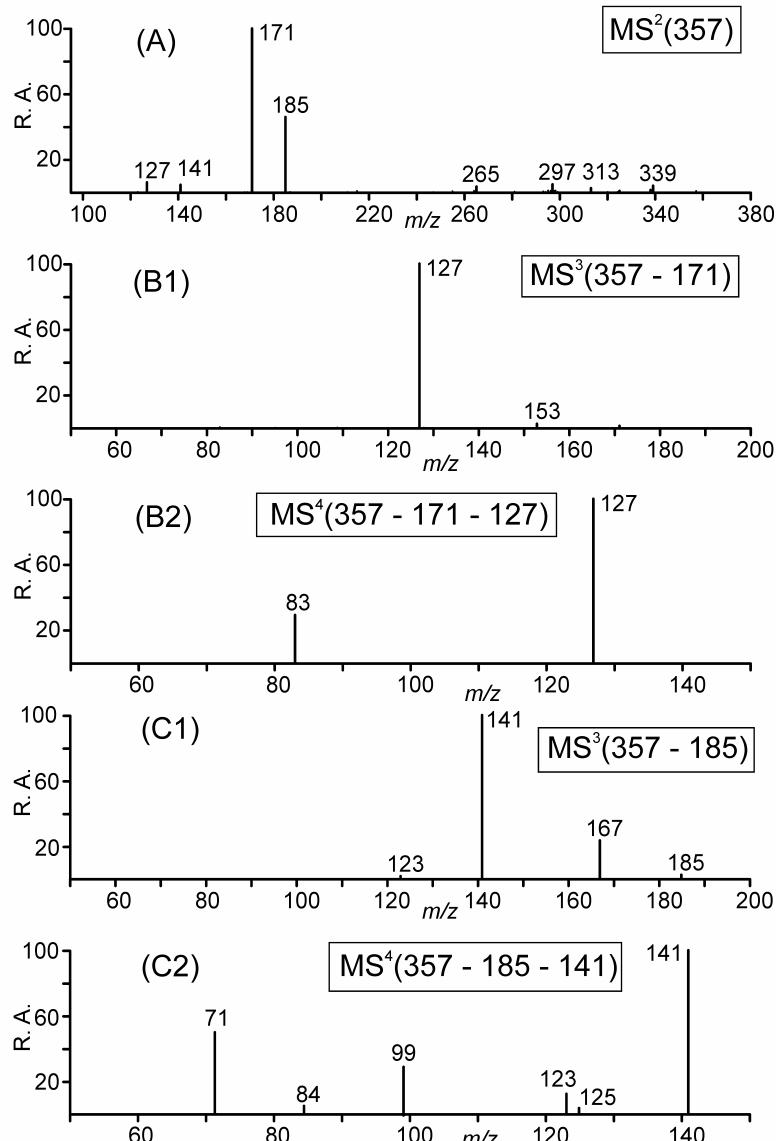


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92 **Figure S3:** (-)ESI-MSⁿ ($n = 2 - 4$) data obtained for deprotonated diaterpenylic acid
93 acetate.
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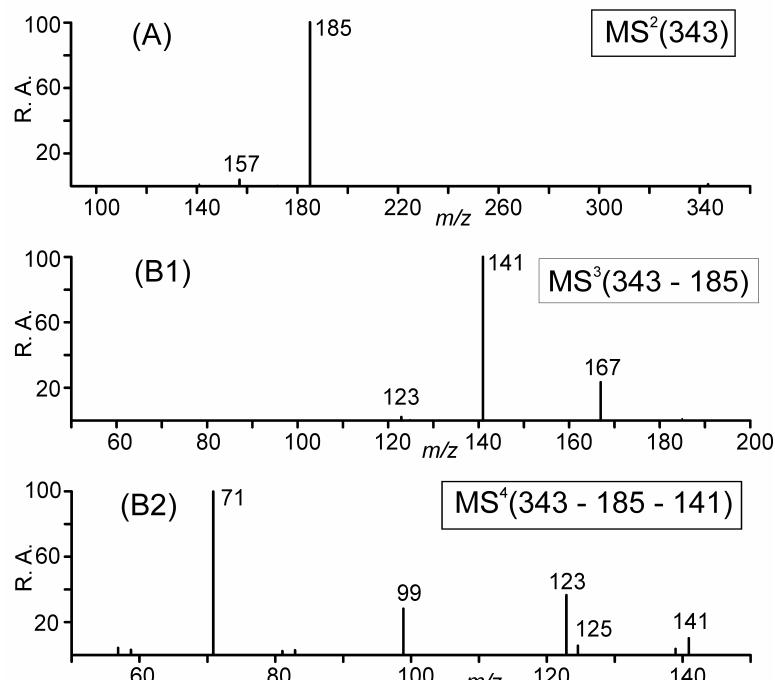
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97 **Figure S4:** (-)ESI-MSⁿ ($n = 2, 3$) data obtained for the deprotonated MW 232
98 compound with RT 22.1 min present in α -pinene ozonolysis SOA (Fig. 1). The
99 fragmentation behavior is distinctly different from that of deprotonated diaterpenylic acid
100 acetate, and starts with the loss of 44 u (CO_2) instead of 60 u (acetic acid) (Scheme S4).
101
102

103 5. Mass spectral characterization of the major MW 358 product from ambient
104 aerosol
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107 **Figure S5:** (-)ESI-MSⁿ data obtained for the MW 358 compound present in ambient fine
108 aerosol (RT 24.6 min, Fig. 2): (A) m/z 357 MS^2 spectrum; (B1) and (B2) higher-order m/z
109 $357 \rightarrow m/z$ 171 MS^n ($n = 3, 4$) spectra; and (C1) and (C2) higher-order m/z 357 $\rightarrow m/z$
110 185 MS^n ($n = 3, 4$) spectra.
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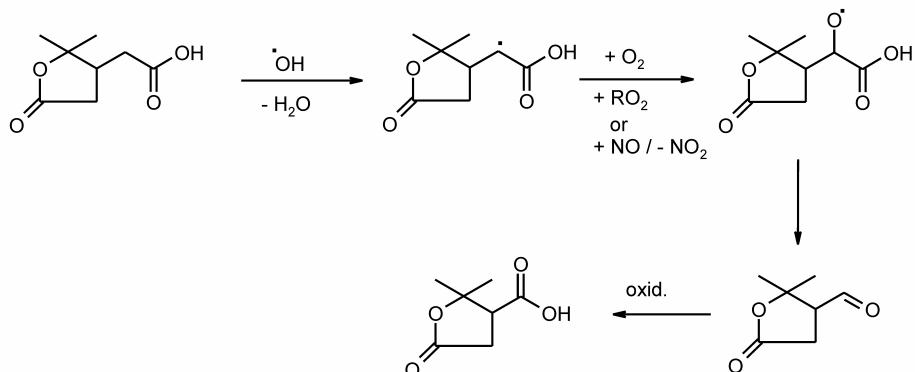
112 **6. Mass spectral characterization of the minor MW 344 β -pinene SOA product co-**
 113 **eluting with the major MW 358 product**



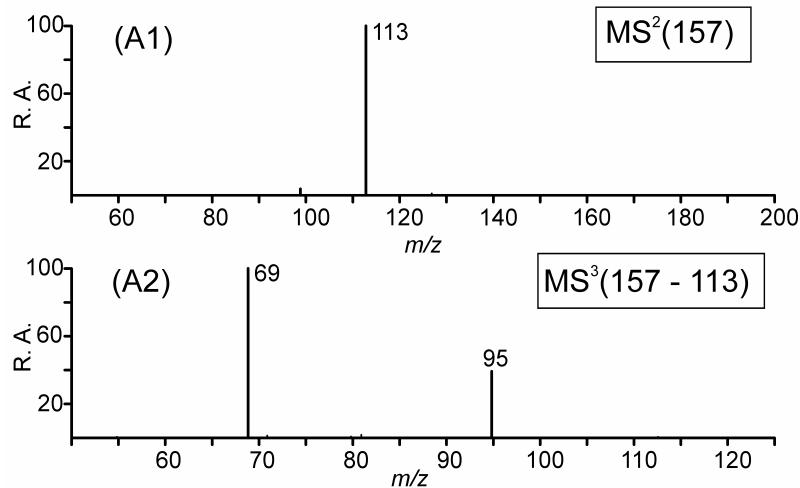
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 115 **Figure S6:** (-)ESI-MSⁿ data obtained for the MW 344 compound present in β -pinene
 116 ozonolysis SOA that co-elutes with the major MW 358 compound (RT 24.7 min, Fig. 1):
 117 (A) m/z 343 MS^2 spectrum; and (B1) and (B2) higher-order m/z 343 \rightarrow m/z 185 MS^n ($n =$
 118 3, 4) spectra. The higher-order m/z 343 \rightarrow m/z 157 MS^n ($n = 3, 4$) spectra (not shown)
 119 were consistent with terebic acid (Fig. S7). The MW 344 dimer is assigned to the diester
 120 formed between diaterebic acid and pinic acid.
 121

122 **7. Formation pathway of terebic acid**

123 (-)ESI-MSⁿ data obtained for terebic acid (reference compound) are presented in Figure
 124 S7, while pathways leading to terebic acid through further processing of terpenylic acid
 125 are given in Scheme S5.

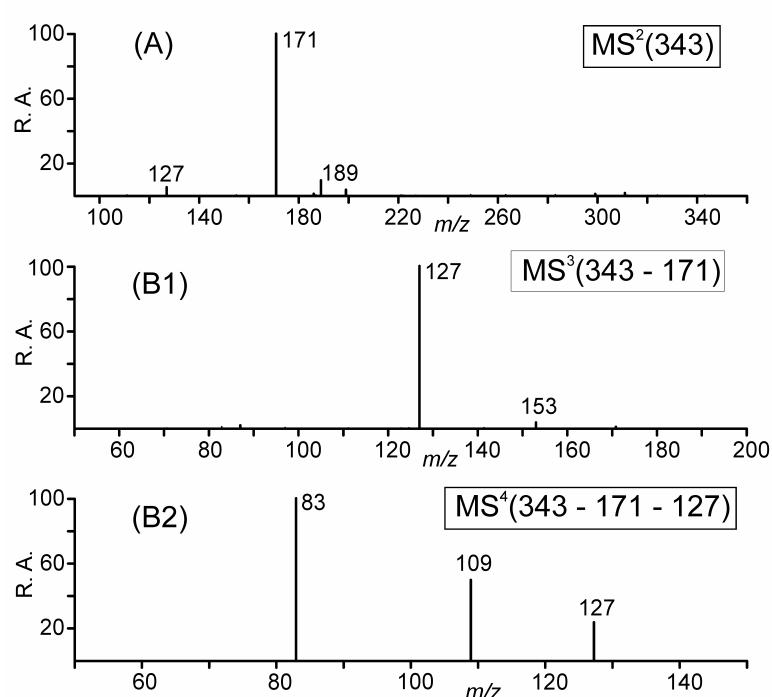


126
 127 **Scheme S5:** Proposed formation of terebic acid from terpenylic acid through
 128 an OH radical-initiated reaction.



129
130 **Figure S7:** (-)ESI- MS^n ($n = 2, 3$) data obtained for terebic acid (reference).
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132 **8. Mass spectral characterization of a minor MW 344 α -pinene SOA product
133 eluting before the major MW 358 product**
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136
137 **Figure S8:** (-)ESI- MS^n data obtained for the MW 344 compound present in α -pinene
138 ozonolysis SOA that elutes before the major MW 358 compound (RT 23.6 min, Fig. 1):
139 (A) m/z 343 MS^2 spectrum; and (B1) and (B2) higher-order m/z 357 \rightarrow m/z 171 MS^n ($n =$
140 3, 4) spectra. The dimer is tentatively assigned to the diester formed between
141 diaterpenylic acid and terpenylic acid.
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143 **References**

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