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Long-lived halocarbon trends and budgets from atmospheric chemistry modelling constrained with measurements in polar firn

P. Martinerie¹, E. Nourtier-Mazauric¹, J.-M. Barnola¹, W. T. Sturges², D. R. Worton^{2,*}, E. Atlas³, L. K. Gohar^{4,**}, K. P. Shine⁴, and G. P. Brasseur⁵

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Abstract. The budgets of seven halogenated gases (CFC-11, CFC-12, CFC-113, CFC-114, CFC-115, CCl₄ and SF₆) are studied by comparing measurements in polar firn air from two Arctic and three Antarctic sites, and simulation results of two numerical models: a 2-D atmospheric chemistry model and a 1-D firn diffusion model. The first one is used to calculate atmospheric concentrations from emission trends based on industrial inventories; the calculated concentration trends are used by the second one to produce depth concentration profiles in the firn. The 2-D atmospheric model is validated in the boundary layer by comparison with atmospheric station measurements, and vertically for CFC-12 by comparison with balloon and FTIR measurements. Firn air measurements provide constraints on historical atmospheric concentrations over the last century. Age distributions in the firn are discussed using a Green function approach. Finally, our results are used as input to a radiative model in order to evaluate the radiative forcing of our target gases. Multi-species and multi-site firn air studies allow to better constrain atmospheric trends. The low concentrations of all studied gases at the bottom of the firn, and their consistency with our model results confirm that their natural sources are small. Our results indicate that the emissions, sinks and trends of CFC-11, CFC-12, CFC-113, CFC-115 and SF₆ are well constrained, whereas it is not the case for CFC-114 and CCl₄. Signif-



Correspondence to: P. Martinerie (patricia@lgge.obs.ujf-grenoble.fr)

icant emission-dependent changes in the lifetimes of halocarbons destroyed in the stratosphere were obtained. Those result from the time needed for their transport from the surface where they are emitted to the stratosphere where they are destroyed. Efforts should be made to update and reduce the large uncertainties on CFC lifetimes.

1 Introduction

This study presents a trend and budget analysis of seven halogenated gases: CFC-11 (CCl₃F), CFC-12 (CCl₂F₂), CFC-113 (CCl₂FCClF₂), CFC-114 (CClF₂CClF₂), CFC-115 (CClF₂CF₃), CCl₄ and SF₆. These species are greenhouse gases (GHGs) and a new evaluation of their radiative impact is provided. Chlorofluorocarbons (CFCs) and CCl₄ are also ozone-depleting substances. Our target gases have very long lifetimes because they are destroyed above the troposphere. They are presumed to be predominantly, if not totally anthropogenic (WMO, 2007).

Originally developed in 1930 as refrigerants, CFCs have also been used as aerosol propellants, foam blowing agents, and cleaning solvents (especially in the electronics industry). Carbon tetrachloride (CCl₄) was first used from the early 1900s to the mid 1960s as a fire extinguishing and a degreasing agent. More recently, it has been used in many other industrial processes. Its atmospheric emissions are thought to arise mainly from its use in the production of CFC-11 and

¹Laboratoire de Glaciologie et Géophysique de l'Environnement (UMR CNRS/INSU 5183), CNRS, Université Joseph Fourier-Grenoble, BP 96, 38 402 Saint Martin d'Hères, France

²School of Environmental Sciences, University of East Anglia, Norwich, NR4 7TJ, UK

³Rosenstiel School of Marine and Atmospheric Science, University of Miami, Miami, FL 33149, USA

⁴Department of Meteorology, University of Reading, Reading RG6 6BB, UK

⁵National Center for Atmospheric Research, Boulder, CO 80307-3000, USA

^{*}now at: Department of Environmental Science, Policy and Management, University of California, Berkeley, CA 94720-3110, USA

^{**}now at: Met Office Hadley Centre, FitzRoy Road, Exeter. EX1 3PB, UK

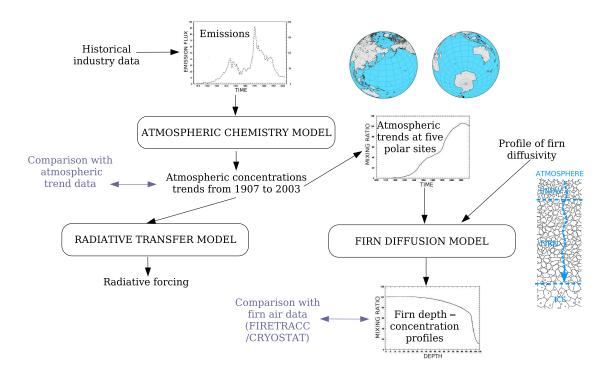


Fig. 1. Combined use of 3 numerical models.

CFC-12. First introduced in the 1960's, sulphur hexafluoride (SF₆) became widespread in the 1980's. It is used mostly as an insulator in electrical transformers and in magnesium production. Targeted by the Montreal protocol, CFC emissions started to decrease in 1998 and are now close to zero (AFEAS, 2007). Also controlled by the Montreal Protocol, CCl₄ emissions decrease started earlier: 1972 (Simmonds et al., 1998), whereas SF₆ emissions (controlled under the Kyoto Protocol) likely decreased temporarily around 1997–1999 before increasing again (Maiss and Brenninkmeijer, 2000; Smythe, 2004).

While CFCs and CCl₄ are nearly inert in the troposphere, they are destroyed in the stratosphere by UV photolysis and oxidation with O(¹D) radicals. However, CCl₄ may have two important surface sinks: oceanic (Yvon-Lewis and Butler, 2002) and soil (Happell and Roche, 2003) losses could play a major role in the destruction and lifetime of CCl₄ (WMO, 2007). Sulphur hexafluoride has a very long lifetime (3200 years) as it is destroyed only in the mesosphere, through photolysis and ion reactions (Ravishankara et al., 1993; WMO, 2007).

Whereas all the studied gases are emitted mainly in the Northern Hemisphere, their concentrations are similar in both hemispheres because their very long lifetimes allow for their mixing in the whole atmosphere. Such long lifetimes also explain the important time lag between the emission decrease of CFCs and CCl₄, and the decrease of their concentrations. Air measurements performed at ground stations

in both hemispheres provided atmospheric trends of most of the studied gases for the last thirty years at best (e.g. Elkins et al., 1998 and Prinn et al., 2000 for the major ESRL (Earth System Research Laboratory, previously called CMDL) and AGAGE (Advanced Global Atmospheric Gases Experiment, previously called GAGE and ALE) networks. These records can be extended back to the earlier part of the 20th century by using measurements in polar firn.

Trace gases diffuse slowly from the atmosphere through the open channels of polar firn down to the ice forming zone, where they are trapped in air bubbles. The time scale of this process is of the order of decades to a century. It depends primarily on the tortuosity of the open porosity at the considered site, and on the time trend of the gas concentration at the surface, which drives the concentration gradient (and thus diffusion speed) between the top and the bottom of the firn. Pumping firn air at different depths allows the extraction of relatively large amounts of air which permits the measurements of low concentration gases (Schwander et al., 1993). Several studies have used firn diffusion models to infer our target gases concentration history from firn air measurements (Schwander et al., 1993; Butler et al., 1999; Sturges et al., 2000; Sturrock et al., 2002; Trudinger et al., 2002). The originality of our approach is to use an atmospheric chemistry model in order to constrain the concentration trends used as input to the firn diffusion model.

Figure 1 illustrates our combined use of three numerical models in order to interpret firn air data in terms of species trends, budgets and radiative forcings. Historical emission data are used to constrain a 2-D model of tropospheric and stratospheric chemistry run on the century time scale (Brasseur et al., 1990; Martinerie et al., 1995). The model calculates the transport and chemical destruction of trace gases, and the balance of these processes controls the species atmospheric concentration fields. Model results are validated by comparison with atmospheric data sets: surface trends, total column, and balloon-based measurements.

The predicted atmospheric concentration trends are used as input to a direct model of gas diffusion in interstitial air of polar firn (Rommelaere et al., 1997), in order to be compared with trace gas concentrations in firn measured within the FIRETRACC and CRYOSTAT EC projects. Measurements were made at five Arctic and Antarctic sites: Devon Island and North Greenland Ice Coring Project (North GRIP) in the Northern Hemisphere, and Dronning Maud Land, Dome Concordia and Berkner Island in the Southern Hemisphere. The firn diffusion model has to be constrained with a site-specific effective depth-diffusivity profile. It is calculated using both the known long term atmospheric trend and concentrations in firn air of a reference gas (such as CO₂ or CH₄) to constrain an inverse version of our firn model (Fabre et al., 2000; Rommelaere et al., 1997).

The combined use of an atmospheric chemistry model and a firn diffusion model allows the discussion of the model/data comparison in terms of budget understanding for our target species. The uncertainties vary largely from one species to another, and we discuss the impact of different factors such as emissions, lifetime, measurement calibration, and diffusivity in polar firn.

The atmospheric chemistry model based halocarbon concentrations are also used for calculating a time history of radiative forcing for all studied gases.

This manuscript is organized as follows. Section 2 describes the firn diffusion and atmospheric chemistry models. Sections 3 and 5 present their validation. The trace gas measurements in firn air (Sect. 4) are introduced just before the firn model validation (Sect. 5). In Sect. 6, we discuss our understanding of trace gas ages in the firn (Sect. 6.1), and our target gas trends (Sect. 6.2), lifetimes (Sect. 6.3) and budgets (Sect. 6.4). Section 6.5 deals with their radiative impact.

The supplementary material (http://www.atmos-chem-phys.net/9/3911/2009/acp-9-3911-2009-supplement.pdf) presents firn diffusivities and additional firn age data discussed in terms more firn-oriented than atmosphere-oriented. It also contains additional chemistry model results which are presented below only for selected species.

2 Numerical models

2.1 The atmospheric chemistry model

Simulations of atmospheric chemistry are performed with a global zonally averaged model with 5° latitude bands from 85° S to 85° N and 1km vertical layers from the Earth surface to 85 km (Brasseur et al., 1990; Martinerie et al., 1995). The time step is 15 days, allowing for 100 years simulations to be performed in a few hours on a personal computer (PC). Radiation, transport and chemistry of the middle atmosphere are fully coupled. Mean monthly climatological winds and temperatures (Randel, 1987) are imposed in the troposphere. The model includes an improved version of the NCAR CCM1 radiative code (Wang et al., 1991). Chemistry comprises HO_x, NO_v, CH₄, CO, PAN, Cl_v, and Br_v, represented by 56 long and 28 short lived species. SF₆ is destroyed at very high altitudes and longer time scales than our 100 years simulations. Thus its lifetime is not calculated but imposed as 3200 years (WMO, 2007).

A new photochemistry code has been coupled to the model: TUV3.9 (Madronich, 1993, Madronich, personal communication, 1998). Its spectral grid includes 183 intervals between 116.3 and 735 nm. The chemical and photochemical reaction rates were updated to take into account recent evaluations (Sander et al., 2006; IUPAC, 2006).

The model ability to simulate large scale meridional gradients and long term trends has been tested using ⁸⁵Kr. The model was constrained by emission data since the beginning of ⁸⁵Kr production (1945). Calculated concentrations were compared to observations (the data used as well as the approach are described in Jacob et al., 1987; Zimmermann et al., 1989; Levin and Hesshaimer, 1996). Although the low spatial and time resolution of the model does not allow to achieve similar precision as the above 3-D models, a satisfactory representation of ⁸⁵Kr trend and meridional gradient is obtained.

The estimates of halocarbon historical anthropogenic emissions are based on production/sales data provided by chemical manufacturers, and emission functions. These functions represent the timing and rate of halocarbon emissions to the atmosphere, and primarily depend on halocarbon usage categories. For example, emissions from open cell foams will be much faster than those from closed-circuit refrigeration. All major manufacturers from developed countries report their CFC productions. Yearly updated emission data were derived from these data (AFEAS, 2007). However, companies responding to AFEAS represent a diminishing fraction (less than 16% in 2004) of global CFC sales, thus only global production is reported for 2004 and there will be no data in the future. Production and emission data from non reporting countries (mainly from Eastern Europe and Asia) are less precise and updated on a less regular basis.

For CFCs (CFC-11, CFC-12, CFC-113, CFC-114, CFC-115), the model was constrained with emissions reported to

AFEAS (2007). Recent estimates of non-reported emissions were used when available (CFC-11: McCulloch et al., 2001; CFC-12 and CFC-113: McCulloch et al., 2003). To our knowledge, no estimate of non-reported emissions is available for CFC-114 and CFC-115, thus we neglect them in the first instance. For CCl₄, emission data from Simmonds et al. (1998) were used until 1995, and extrapolated using the same methodology of scaling to CFC production. This gives more reliable results than a simple extrapolation of CCl₄ emission data. SF₆ emissions were initially taken from Maiss and Brenninkmeijer (1998, 2000). However this resulted in an overestimation of the slope of the SF₆ trend around 1998-2002. Thus the more recent estimate from the EDGAR database (V3.2 and V3.2 FT2000, EDGAR, 2007) has been used. This dataset covers the 1949–1995 period and year 2000, it was completed using Smythe (2004). For all compounds, the meridional distribution of the emissions was taken from the EDGAR (2007) database: EDGAR V2.0 for all CFCs (data for the year 1986), and EDGAR v3.2 FT2000 (year 1990) for SF₆. In the absence of data, the meridional distribution of CCl₄ emissions was assumed to be the same as for CFC-12. The surface ocean is not represented in the model, and the meridional distribution of CCl₄ soil sink still has to be quantified. Besides, WMO (2007) states that the CCl₄ ocean and soil sinks are difficult to reconcile with its concentration records, thus these sinks were not taken into account in our model. Their potential effect is discussed in Sect. 6.4.

As in Martinerie et al. (1995), the model boundary conditions are imposed concentrations at the ground for CO_2 , CH_4 and N_2O and emission fluxes for other major source gases, which control the oxidation capacity of the troposphere but have a minor influence on the stratosphere. The effect of the trends in CO_2 , CH_4 and N_2O on halocarbon concentrations was evaluated by comparison with a test simulation constrained with the 3 major greenhouse gases having constant (1990) concentrations over the whole 100 years simulation. The induced change in our target species concentration trends is negligible (less than 0.2%).

2.2 The firn diffusion model

A firn diffusion model is a dating tool, calculating trace gas depth-concentration profiles as a function of their atmospheric trends. Three depth zones are distinguished in our model (Rommelaere et al., 1997). In the upper part of the firn (a few meters), a "well mixed zone" can be defined. Due to mechanical (wind) and thermal mixing, it has the same composition as the above atmospheric air. Below, in the "transport zone", concentration gradients between the top and the bottom of the firn induce trace gas diffusion in firn air. Gravitational fractionation also occurs. For our target gases, which are heavier than air, gravitational fractionation leads to mixing ratios increased by 2.5% to 9% at the bottom of the firn, depending on the firn close-off depth and

trace gas molecular mass. Trace gas diffusion and gravitational fractionation induce a trace gas flux in the firn which lead to vertical concentration gradients. At the bottom of the firn, namely the "close-off zone" or "lock in zone", densification leads to the formation of air bubbles. The ratio of open over closed porosity decreases rapidly with depth. The duration of the close-off process mainly depends on the snow accumulation rate, which moves the firn downward. Steeper depth-concentration gradients are observed in the lowest part of the firn where diffusion is very slow and open porosity decreases. The ice zone starts where porosity is totally closed. The air composition is then fixed, and the air bubbles are simply advected down with the ice. The most important processes occurring in the firn are trace gas diffusion and bubble close-off, which eliminates old air from the deepest part of the firn. Our firn model and the mathematical representation of the above processes are described in Rommelaere et al. (1997).

Several assumptions are made in our representation of trace gases transport in firn. The snow accumulation rate, temperature and firn structure at a given depth (characterized by porosity, density, tortuosity, etc...) are assumed to be independent of time. This is justified here because the studied period is short and climatically stable. Firn temperature is also considered as uniform and close to mean annual temperature. This is consistent with direct observations in polar firm below ten meters depth. Thermal diffusion is not taken into account; it would mainly affect the upper ten meters of the firn. Considering trace gas transport, all gases are assumed to verify ideal gas law. Molecules are assumed to collide with each other much more than with firn surface, because their mean free path is much smaller than the dimension of pores in firn. For the studied inert gases having low solubilities in water, gas exchanges through the surface of firn pores are

An essential parameter used to constrain firn diffusion models is the effective diffusivity-depth profile. It depends on the tortuosity of the open firn porosity. Direct measurements of trace gas diffusion coefficients in firn samples have been performed (Schwander et al., 1988; Fabre et al., 2000). These measurements are time consuming, thus relationships between diffusivity and porosity (or its inverse: density) have been used to extrapolate them to other sites (Schwander et al., 1993; Trudinger et al., 1997). In this study, effective diffusivity profiles are retrieved by inverse modelling of a trace gas having a well known atmospheric trend and a firn depth profile at the studied site (Fabre et al., 2000; Rommelaere et al., 1997). CH₄ was used for diffusivity calculation at all sites except Dome C, where CO₂ leads to better results around 95m depth (see supplement http://www.atmos-chem-phys.net/9/3911/2009/ acp-9-3911-2009-supplement.pdf). We should note that the effective diffusivity is constrained to be an increasing function of open porosity, and therefore a decreasing function of depth. This does not allow to explicitly take into account the

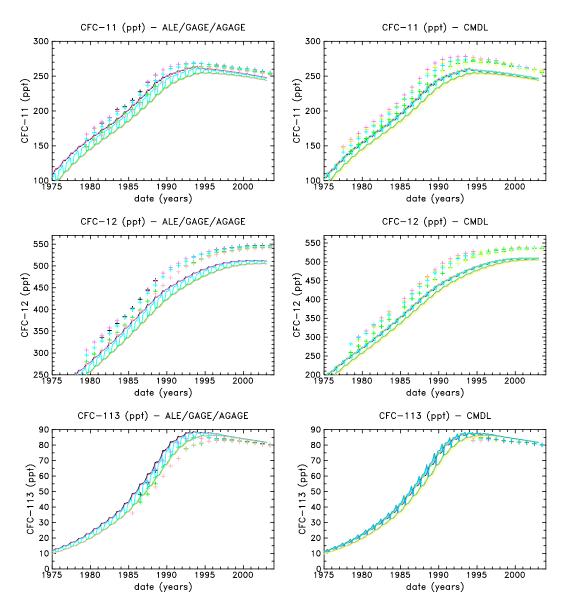


Fig. 2. Comparison of the chemistry model results (lines) for major CFCs with annual mean atmospheric data (plus signs). The different colours depict trends at ground level for different latitudes. Left panel: comparison with AGAGE data: violet: Adrigole (52° N) and Mace Head (53° N), Ireland, black: Oregon (45° N) and Trinidad Head (41° N), USA, turquoise: Barbados (13° N), green: Samoa (14.3° S), pink: Cape Grim (40.7° S), Australia. Right panel: comparison with ESRL data: black: Alert (82.5° N), Canada, violet: Pt. Barrow (71.3° N), Alaska, USA, orange: Niwot Ridge (40.1° N), USA, turquoise: Mauna Loa (19.5° N), Hawaii, blue: Cape Kumukahi, (19.5° N), Hawaii, green: Samoa (14.3° S), pink: Cape Grim (40.7° S), Australia, yellow: South Pole (90° S), Antarctica. The sites represented in turquoise (Barbados and Hawaii) show a high seasonal variability because the inter-tropical convergence zone shifts seasonally North and South of these measurement stations. A similar variability is observed in the monthly mean (instead of annual mean) observations.

effect of refrozen melt layers at Devon Island or other heterogeneities in firn (Martinerie et al., 1992). Fabre et al. (2000) have shown that the inverse modelling approach provides a better representation of diffusivity variations from one site to another than direct measurements on firn samples. Indeed, firn diffusivity appears to be sensitive to heterogeneities occurring at larger scale than the samples used for direct measurements.

Once the diffusivity profile at a given site is established for a reference gas, it can be used for another gas by simply scaling it by the ratio of molecular diffusion coefficients of the two gases. Therefore, a bias shared by all gases at one site may result from a bias in the diffusivity profile at this site. As in most previous studies, molecular diffusion coefficients for halocarbons are estimated from molecular volume calculations (Gilliland, 1934, described in Perry and Chilton, 1973).

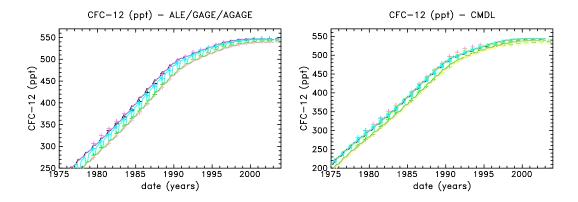


Fig. 3. Comparison of the chemistry model results (lines) for CFC-12 using adjusted emissions with annual mean atmospheric data (plus signs). The colours are defined in Fig. 2.

They are provided and evaluated against measurements in the supplementary material (http://www.atmos-chem-phys.net/9/3911/2009/acp-9-3911-2009-supplement.pdf).

3 Validation of the atmospheric chemistry model

The quality of the atmospheric chemistry model results is evaluated by comparison with surface, total column, and balloon-based measurements.

3.1 Validation of surface concentrations

Our surface concentration trends are compared with atmospheric measurements from multiple long-term observation sites from the ALE/GAGE/AGAGE (noted AGAGE in the following) and NOAA-CMDL/ESRL (further noted ESRL) monitoring networks (Montzka et al., 1999; Prinn et al., 2000). The major sources of uncertainties in the atmospheric cycle of long lived halocarbons are related to their emissions, their lifetimes (or loss rates) and their measurement calibration. These uncertainties are relatively small for CFC-11, CFC-12 and CFC-113 (WMO, 2007), thus these gases are used to discuss the validation of our chemistry model. A similar comparison for our other target gases is presented in the supplementary material (http://www.atmos-chem-phys.net/ 9/3911/2009/acp-9-3911-2009-supplement.pdf). According to WMO (2007), inter-calibration differences between ESRL and AGAGE are lower than 2% for CFC-11, CFC-12 and CFC-113. Our results (Fig. 2) under-predict the observations by 3-6% for CFC-11 and 6.5-9.5% for CFC-12, whereas they over-predict the measurements for CFC-113 by 2–7%. These differences vary with time but stay within the uncertainty limits of emission estimates: between 5 and 9% (Mc-Culloch et al., 2003) or 7.5–18% (WMO, 2007).

Meridional distributions are well reproduced by our model results (Fig. 2). For these long lifetime gases, meridional gradients reflect their transport time from their production (mostly in the Northern hemisphere) to their measurement sites. The drastic reduction of their production rate following the Montreal and Kyoto protocols explains the strong decrease in their North to South concentration gradient over recent years. For CFC-11, the difference between model and data is higher with the ESRL record than with the AGAGE record (Fig. 2). As the model results are basically the same on both plots, this difference is due to inter-calibration differences between the two datasets.

Additional chemistry model simulations were performed with halocarbon emissions adjusted to fit the atmospheric measurements. The results for CFC-12 are shown in Fig. 3. Here emissions were increased by 10% before 1991 and decreased by 10% between 1991 and 1995, no adjustment was performed afterwards. Such chemistry model results with halocarbon emissions adjusted to fit atmospheric measurements will be used to constrain the firn diffusion model in order to check the consistency between firn air and atmospheric measurements.

3.2 Comparison with balloon and column measurements for CFC-12

The validation of our chemistry model has been completed for CFC-12 by comparison with stratospheric balloon measurements and vertical column measurements. This aims at further testing the model transport and CFC-12 destruction in the stratosphere. A significant bias in model vertical transport would likely affect the destruction rate and global sink of our target gases, which are destroyed only in the middle atmosphere.

Engel et al. (1998) collected balloon data since 1978 at Aire sur l'Adour, Gap (44° N, France) and Kiruna (68° N, Sweden) in order to derive historical trends of CFC-12 in the stratosphere. The CFC-12 mixing ratio trends were compiled for three N₂O levels, considered as a surrogate for altitude (lowermost stratosphere, around 19 km and around 21.5 km). In the model, the N₂O surface concentration

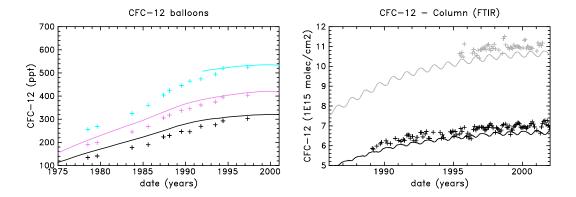


Fig. 4. Comparison of modelled CFC-12 concentrations (lines) with stratospheric balloon (left panel) and vertical column (right panel) measurements (plus signs). Left panel: CFC-12 mixing ratios at N_2O levels of 310 ppb (turquoise), 250 ppb (violet) and 200 ppb (black). Right panel: Vertical column abundances of CFC-12 above the Jungfraujoch (46° N, 3580 m, in black) and Wollongong (34° S, 30 m, in grey) measurement stations. The model results shown here use adjusted emissions (as in Fig. 3).

scenario was built using AGAGE atmospheric measurements at Cape Grim, Australia from 1978 to 2002 (Prinn et al., 1990, 2000) and Antarctic firn and ice core data before 1978. Two smoothed scenarios were built using the records of Machida et al. (1995) and MacFarling Meure et al. (2006). They are consistent within 0.5 ppb. A graphic comparison with Fig. 3 in Battle et al. (1996) indicates that the atmospheric trend inferred from the South Pole firn record is about 2-3 ppb lower than the two above scenarios, although South Pole data are within the range of the variability in the Machida et al. (1995) record. The N₂O meridional gradient is positive from South to North and is stronger close to the Equator (IPCC, 2001). Modelled N₂O mixing ratios are consistent with AGAGE measurements at all latitudes. Model results for CFC-12 at the three N₂O levels are virtually identical for the two launch site latitudes (44° N and 68° N). They are compared with observation on the left panel of Fig. 4.

Model results in Fig. 4 were computed with CFC-12 emissions adjusted to obtain surface mixing ratios consistent with surface station measurements (see Sect. 3.1), they slightly overestimate the measurements, but the results obtained with raw CFC-12 emissions fit them almost perfectly. Considering the 5% precision of balloon measurements (Engel et al., 1998), our model results are globally consistent with observations. The results obtained with the two N₂O ice core scenarios show differences in CFC-12 concentrations lower than 1ppt. Model results at the 310 ppb N₂O level are plotted only after 1992 because modelled N₂O concentrations did not reach 310 ppb before that time. This is consistent with yearly average AGAGE surface data, which reach this level in 1991. Thus we suggest that early measurements of CFC-12 at 310 ppb N₂O were made on tropospheric polluted air intruded into the lower stratosphere.

Our model results were further compared with total column densities of CFC-12 from ground-based FTIR spectrometers at Jungfraujoch in Switzerland and Wollongong in Australia (Zander et al., 2000; NCEP, 2006). Since the elevation at the Jungfraujoch station is much higher than Wollongong, column densities are higher at Wollongong than at Jungfraujoch (right panel in Fig. 4). This hides slight differences due to meridional variations of CFC-12 mixing ratio, described in Sect. 3.1. Model results underestimate the experimental data by approximately 6%. However, considering the uncertainties reported for the Jungfraujoch (±4%) and Wollongong (15%) measurements, as well as the uncertainties on emissions used to constrain the model, the modelled column densities are still consistent with measurements.

Our model results somewhat overestimate the balloon measurements, but underestimate total column data, thus no clear bias of the vertical structure of our modelled CFC-12 concentrations can be detected.

4 Trace gas measurements in firn air

Numerous trace gas measurements were performed at five Arctic and Antarctic drilling sites within the FIRETRACC and CRYOSTAT EC projects (FIRETRACC, 2007; CRYO-STAT, 2007). The major characteristics of these sites are presented in Table 1. The experimental setup used to extract firn air from boreholes in ice sheets is similar to the original technique described in Schwander et al. (1993). The modifications (Sturges et al., 2001) include on-site continuous measurements of CO₂ (LI-COR analysis) allowing the absence of contamination with contemporary air to be monitored. Only a few surface air samples were taken during the short (about a month) firn drilling periods. Our surface values are thus less reliable than monthly mean data from atmospheric measurement networks which involve numerous samples. We should note that samples from the two last depth levels at Berkner Island are thought to be contaminated

Table 1. Characteristics of the firn drilling sites. Temp. stands for mean annual temperature, and Accu. for snow accumulation rate in water equivalent units. Drill site abbreviations: DI: Devon Island, NGR: North GRIP, BKN: Berkner Island, DML: Dronning Maud Land, DC: Dome C.

Site	Drill date	Lat./Lon. (degrees)	Alt. (m).	Temp.	Accu. (cm/yr)
DI	04/1998	75 N/82 W	1800	-23	30
NGR	07/2001	75 N/42 W	2975	-32	17
BKN	01/2003	79 S/45 W	900	-26	13
DML	01/1998	$77\mathrm{S}/10\mathrm{W}$	2300	-38	7
DC	01/1999	75 S/124 E	3240	-53	3.5

by a leak in the air pumping system (Worton et al., 2007) and should be considered with caution. Halocarbon and SF₆ measurements presented in this paper were performed at both the University of East Anglia (UEA) and National Center for Atmospheric Research (NCAR) using gas chromatography combined with mass spectroscopy (GC-MS). These analytical techniques are described in Sturges et al. (2001) and references therein. Detection limits are about 0.002 ppt for UEA measurements and 0.02-0.03 ppt for NCAR measurements. Among the measured concentrations reported here, only CFC-115 results at the deepest levels of Northern Hemisphere sites fall below NCAR detection limits. Additional measurements of SF₆ were done at the Max Planck Institute of Mainz on large volume samples (Sturges et al., 2000). Most of the data presented in this paper are publicly available in the FIRETRACC and CRYOSTAT databases (FIRE-TRACC, 2007; CRYOSTAT, 2007).

The firn air data dispersion (or spreading) was evaluated using multiple measurement at the same depth level, by calculating the standard deviation resulting from these concentration differences. Although data from all drilling sites were taken into account, the overall values in Table 2 are based on only about 20 replicates. However, we should note that higher values of concentration differences frequently occur in the upper part of the firn (top 15–20 m), where firn air concentration are potentially influenced by short term atmospheric variations. The inter-calibration differences between UEA and NCAR measurements were carefully examined on samples measured at the same site and depth. The resulting values (column 4 in Table 2) were used to convert NCAR data to values consistent in scale with UEA measurements.

5 Validation of the firn diffusion model

The major uncertainties on simulated vertical profiles in the firn arise from the atmospheric trends used as input, which rely on somewhat uncertain emission trends (Sect. 3.1), uncertainties on firn air measurements (Sect. 4) and on the firn model, which is mainly constrained by the firn diffu-

Table 2. Quality parameters for trace gas measurements in firn. STD stands for standard deviation (ppt), Calibr. stands for intercalibration differences, and WMO refers to WMO (2007). When two numbers are given in the STD UEA column, the first one refers to measurements made with the PLOT chromatographic column, whereas the second one refers to the DB5 column (Sturges et al., 2000, 2001).

Gas	STD UEA	STD NCAR	Calibr. UEA-NCAR	Calibr. WMO
CFC-11	3	1	+1%	<2 %
CFC-12	7/13	3	+1.5%	<2 %
CFC-113	1/1	3	+1%	<2 %
CFC-114	0.3	0.2	+20%	
CFC-115	0.2	0.3	+20%	
CCl ₄	4	6	+6%	3–4 %
SF_6	0.1			

sivity profiles (Sect. 2.2). CFC-11, CFC-12 and CFC-113 have long atmospheric trend records and low uncertainties on their budget, thus they are again used here for firn model evaluation. The adjusted atmospheric trends calculated with the chemistry model (Sect. 3.1) were used as input to the firn diffusion model and are compared with measured concentrations in firn (Fig. 5). The overall agreement between model results and observations is very good. Previous estimates of firn diffusivities (FIRETRACC, 2007; CRYOSTAT, 2007) produced systematic differences between model results and observations for all gases at the bottom of the Devon Island firn and at intermediate depths of the Dronning Maud Land firn. Such differences were attributed to a bias induced by the firn diffusivity profiles (see Sect. 2.2). New calculations of the firn diffusivities were performed for all firn drilling sites with less detailed adjustments to the reference gas. The resulting smoother diffusivity profiles (see supplement http://www.atmos-chem-phys.net/9/3911/2009/ acp-9-3911-2009-supplement.pdf) produce a better agreement with halocarbon measurements. At Devon Island, a model/data discrepancy remains between 25 and 30 m depth: the data show a drop in the concentrations of major CFCs (Fig. 5), whereas the model produces a smooth decrease. 197 refrozen melt layers with thicknesses ranging between 0.5 and 6.5 cm were logged in the Devon Island firn core, and the thickest melt layer was found at 28 m depth. Although the 25–30 m depth range is not the only interval where thick melt layers are observed, the 13-30 m region shows the highest accumulation of melt layers (in terms of total number as well as cumulated thickness). The effect of impermeable layers depends on their horizontal extent and continuity. In our model, the firn is considered as a homogeneous medium and singular layers cannot be reproduced. The drop in observed concentrations around 25-30 m depth is thus likely due to the presence of melt layers at Devon Island. Model/data discrepancies showing parallel signals (similar slopes) may be

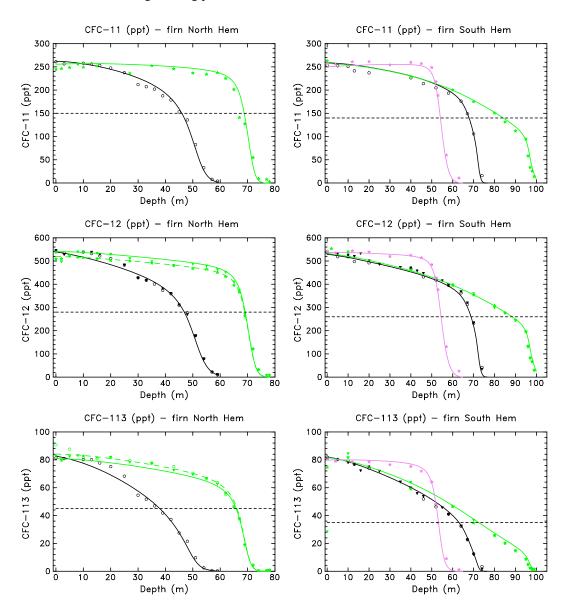


Fig. 5. Comparison of the firn model results (continuous lines) for major CFCs with FIRETRACC and CRYOSTAT data (symbols). Left panel: comparison with North Hemisphere drill sites: Devon Island in black, North GRIP in green. Right panel: comparison with South hemisphere drill sites: Berkner Island in violet, Dronning Maud land in black, Dome C in green. Stars and triangles show UEA measurements performed respectively with a PLOT and DB5 chromatographic columns, open circles depict NCAR measurements. Horizontal black dashed lines illustrate the lowest atmospheric concentrations measured in the relevant hemisphere by the AGAGE or ESRL networks. The green dashed lines show the impact of reducing atmospheric CFC-12 concentrations by 4% and increasing atmospheric CFC-113 concentrations by 4%.

due to calibration issues on the firn measurements. A calibration bias would likely affect all measurements by the same percent change. The dashed green lines for CFC-12 and CFC-113 at North GRIP illustrate this effect: the atmospheric concentration scenario was reduced by 4% for CFC-12 and increased by 4% for CFC-113. These scenario adjustments produce model results in agreement with the firn data. Such concentration adjustments have a negligible impact on the lower part of the firn record (Fig. 5).

Overall, the modelled and observed slopes are in excellent agreement in the upper part of the firn. The gentle slopes at small depths indicate that the upper firn is a fast diffusive medium. Steeper slopes are obtained at the three sites drilled in 1998–1999 (Table 1) than at North GRIP and Berkner, drilled in 2001–2003. This might be due to the recent reduced increase or decrease in major CFC trends. However, a sensitivity test performed by running the diffusion model only until January 1998 at North GRIP and Berkner showed

Table 3. Comparison of ages in the firn with atmospheric scenario ages at the deepest measurement level(s). Scen age is the age in the atmospheric scenario corresponding to the concentration measured in the firn. Firn age presents the trace gas mean ages (sum over time of ages multiplied by their probability) in the firn, and between brackets: ages at 15%, 50% (median age) and 85% accumulated probability. Drilling sites abbreviations are DI: Devon Island, NGR: North GRIP, BKN: Berkner, DML: Dronning Maud Land, DC: Dome C.

Site	DI	NGR	NGR	BKN	BKN	DML	DC	DC
Depth	59 m	74 m	78 m	57 m	63 m	74 m	98 m	99 m
CH ₄								
Scen age	45	39	56	28	52	32	21	
Firn age	41 (31/39/50)	38 (29/36/47)	60 (50/58/69)	32 (13/25/53)	62 (35/56/90)	37 (22/34/52)	21 (10/18/32)	
CO_2								
Scen age	34	37	43	28	45	34	23	
Firn age	46 (35/44/57)	43 (33/41/54)	65 (54/63/76)	38 (17/31/61)	72 (43/66/101)	44 (28/41/61)	29 (15/25/44)	
SF ₆								
Scen age	36	38	38	29	37	33		29
Firn age	55 (41/52/69)	52 (39/50/65)	74 (60/71/87)	48 (24/41/72)	85 (55/79/115)	57 (37/53/76)		51 (28/46/75)
CFC-12								
Scen age	45	41	50	32	43	35	33	38
Firn age	55 (41/52/69)	52 (39/50/65)	74 (60/72/87)	48 (24/41/72)	85 (56/80/115)	57 (37/53/76)	45 (23/40/67)	51 (28/46/75)
CFC-11								
Scen age	43	41	46	32	41	34	33	36
Firn age	56 (42/53/71)	54 (40/51/67)	75 (62/73/89)	49 (25/43/74)	87 (58/82/117)	59 (39/55/79)	48 (25/42/72)	55 (30/49/80)
CFC-113								
Scen age	46	46	43	28	41	32	36	36
Firn age	60 (44/56/76)	57 (42/54/72)	79 (64/76/94)	53 (28/46/78)	92 (62/86/122)	64 (43/60/85)	55 (29/49/82)	62 (35/56/90)

that the slopes in the upper part of the firn remained lower at these two sites. Thus this behaviour may also be a characteristic of medium accumulation rate sites where no melting occurs. Continuous atmospheric measurements of CFC-11 and CFC-12 started in 1979, and in 1985 for CFC-113. The comparison of concentrations measured at that time with firn air records suggests that only the deepest part of the firn records contain significant information about the period pre-dating ambient air measurements for these gases. In this earlier part of the records, the model/data agreement remains correct although the high slopes and age mixing in the firn lead to increased uncertainties.

The uncertainties in the lower part of the firn records can be better characterized by calculating trace gas age distributions in the firn. Following the classical Green function approach, the age distributions in the firn can be established by studying its response to a pulse in trace gas concentration. Such an approach has already been used in previous studies (e.g. Schwander et al., 1993; Trudinger et al., 1997). It allows the calculation of the probability to have trace gas of a given age at a given firn depth. The ages in the firn depend on the particular trace gas through its diffusion coefficient. Our target gases are slow diffusing species with relative diffusion coefficients with respect to CO₂ ranging between 0.495 (for CFC-113) and 0.62 (for SF₆), whereas methane is a small and thus fast diffusing molecule. Table 3 presents the firn model derived mean ages, as well as the ages at 15%, 50% and 85% accumulated probability. Age distributions in the firn are asymmetric (see Sect. 6.1), this is why the mean age is located between the 50% and 85%

accumulated probability ages. By contrast, the maximum probability age (mode age) is located between the 15% and 50% (median) accumulated probability ages. In Table 3, the trace gases are presented in decreasing order of their diffusion coefficients. Thus the model-derived "firn ages" at a given depth are increasing when going down the table. The modelled firn ages are compared with the atmospheric "scenario ages" corresponding to the mean concentrations measured at the lowest sampled levels in the firn. We should note that mean age and scenario age (or effective age) are not theoretically identical for trace gases undergoing a nonlinear time-trend in the atmosphere (Trudinger et al., 2002). As trace gas concentrations at the bottom of polar firns are low, they are subject to maximum experimental errors. Such errors can be visualized in Table 3 by the fact that scenario ages show inconsistencies between the different trace gases by not following the theoretical anti-correlation between diffusion coefficient and trace gas age. In this frame, firn ages and scenario ages are reasonably consistent except at the bottom of North GRIP (78 m) and Berkner (63 m) firns. At these sites, the agreement is satisfying at 74 m and 57 m depth respectively. At Berkner Island, the measurements below 60 m depth are suspected of contamination. Thus both the scenario ages and the firn ages (based on CH₄-derived diffusivity) are unreliable. At North GRIP, the scenario age difference between 74 m and 78 m varies strongly from one trace gas to another. The SF₆ scenario age is invariant and the CFC-113 age decreases between 74 m and 78 m. It reflects unchanged (SF₆) or slightly increasing (CFC-113, see Fig. 5) concentrations. We thus suggest that measurements at the lowest

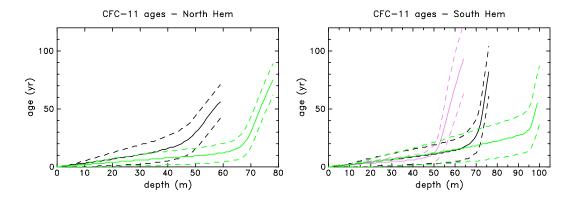


Fig. 6. CFC-11 age versus depth. Continuous lines: mean age, lower dashed lines: age at 15% accumulated probability, higher dashed lines: age at 85% accumulated probability. Colours: same as Fig. 5.

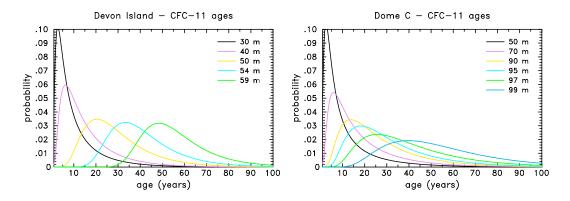


Fig. 7. CFC-11 age distributions at selected depths in the Devon Island and Dome C firns.

level of the North GRIP firn might be slightly contaminated. At this depth, the air flow rate through the sampling apparatus was reduced by two thirds, indicating a nearly impermeable firn. This may have caused ambient air leaks into the sampling system. Similar unchanged or increased concentration anomalies at deepest firn levels were obtained by Butler et al. (1999), who sampled three Antarctic and Greenland sites with 1 m to 1.5 m depth resolution (detailed results can be found in the ESRL database Elkins et al., 1998). The deep Dome C firn shows surprisingly young ages. Scenario ages and firn ages look generally consistent at this site (Table 3), however we should note that the firn model systematically overestimates measured concentrations in the lower part of the Dome C firn (Fig. 5). This may induce an underestimation of the firn ages.

6 Results and discussion

6.1 Age distributions in the firn

As our target species have similar diffusion coefficients, their age distributions show relatively small variations from one gas to another (Table 3). Figure 6 illustrates the depth profile of CFC-11 ages which has an intermediate relative diffusion coefficient of $D_{\rm CFC-11}/D_{\rm CO_2}$ =0.575. All drill sites show a distinct behaviour in the upper firn (slow increase in age with depth) and the lower firn (fast increase in age). The slope change in age-depth profile is less marked at Devon Island than other sites. We attribute this feature to the presence of refrozen melt layers, which decrease the diffusivity in the upper part of the firn. Devon Island is thus our best site if we consider the regularity of the depth-age profile (i.e. the least contrasted gentle/steep slope areas). However the age distribution at Devon Island is wider than at North GRIP (Table 3). This might be related to the low diffusivity of the Devon Island firn, which could lead to increased age mixing at equivalent depths.

Mean ages plotted in Fig. 6 reach their maximum at the last depth before full close-off. If we consider the last measurement depth and exclude the lowest levels at North GRIP and Berkner Island, where firn ages are inconsistent with atmospheric scenario ages (Table 3), maximum CFC-11 mean ages vary between 49 and 59 years before time of drilling.

In the upper firn, the age distribution width is smaller at North GRIP and Berkner Island than at other sites. Upper firn age distributions are highly asymmetric (Fig. 7). This asymmetry decreases with depth. The lower firn age distributions show a general tendency of widening for sites with lower snow accumulation rates, as illustrated in Fig. 7. However, the detailed site to site variations are somewhat complex and discussed in the supplement (http://www.atmos-chem-phys.net/9/3911/2009/acp-9-3911-2009-supplement.pdf).

Overall, the two Arctic sites show no significant influence of air less than 30 years old at the deepest reliable measurement level, whereas Antarctic sites show an influence of 15 to 25 years old air at all reliable measurement depths. On the other hand, at the deepest levels of Antarctic sites, 100 years old air still has an influence on trace gas concentrations in firm. Therefore, site to site differences in age distributions imply that a multi site approach provides more constraints on past trace gas trends.

6.2 Historical concentration trends

6.2.1 Major CFCs (CFC-11, CFC-12)

Sturrock et al. (2002) used firn model inversion techniques to derive atmospheric trends from measurements in Law Dome (Antarctica) firn for long lived halocarbons. For the recent past, the Sturrock et al. (2002) trends are partially constrained with atmospheric and Cape Grim air archive data (Trudinger et al., 2002). Figure 8 compares these concentration based scenarios to our chemistry model and emissions based trends. An almost perfect agreement is obtained for CFC-11 and CFC-12. These results are also qualitatively consistent with the trend histories calculated by Butler et al. (1999) using industrial emission models and atmospheric measurements. We should note that during the atmospheric record period, our approach is somewhat equivalent to Sturrock et al. (2002) because some adjustments were performed on emission scenarios in order to fit the atmospheric records (see Sect. 3.1). Uncorrected and corrected emissions are plotted in Fig. 9, the exact correction factors are given in the supplementary material (http://www.atmos-chem-phys.net/ 9/3911/2009/acp-9-3911-2009-supplement.pdf). justment to atmospheric data in Sturrock et al. (2002) and our study represents a strong constraint for CFC-11 and CFC-12 which have a long atmospheric record. The continuing agreement before 1978 indicates a good consistency between the firn records and emission histories for these gases. The firn drilling site of Sturrock et al. (2002) has an exceptionally high temperature and low close-off depth, which induce a very narrow age distribution (standard deviation of about five years (Sturrock et al., 2002)). On the other hand, only six depth levels were sampled for halocarbons in this shallow firm. Three of these levels show CFC-11 and CFC-12 concentrations lower than the 1978 atmospheric levels, two of them have near zero concentrations. Atmospheric emission histories used to constrain our chemistry model have a much higher time resolution (one year). Thus the consistency of our results with Sturrock et al. (2002) arises from both the consistency of early emissions with their firn record and the relative smoothness of the emission trends.

Measured concentrations at the bottom of the firn range between 4 and 10 ppt for CFC-11, and 9 and 14 ppt for CFC-12 (Fig. 5) at the narrow age distribution Arctic sites. Higher values of 11–18 and 27–44 ppt for CFC-12 are observed at the Antarctic sites. This is likely to be due to the higher impact of recent air on bottom level concentrations at our Antarctic sites (Sect. 6.1). Our lowest concentration levels are consistent with those obtained by Butler et al. (1999). Lower values (0.4–0.7 ppt for CFC-11 and 1.6–2.1 ppt for CFC-12) were observed at the very narrow age distribution Law Dome site sampled by Sturrock et al. (2002). The consistency between our emission based model results and firn measurements (Fig. 5) confirms the absence of significant CFC-11 and CFC-12 concentrations in the atmosphere before the start of their industrial emissions.

6.2.2 Other CFCs (CFC-113, CFC-114 and CFC-115)

Unlike major CFCs, our modelled CFC-113, CFC-114 and CFC-115 atmospheric trends show some differences with the Sturrock et al. (2002) scenarios (Fig. 8). The Sturrock et al. (2002) CFC-113 data suggest higher concentrations than ours in the 1970–1980 period (its AGAGE atmospheric record starts in 1985). To our knowledge, there are no CFC-114 and CFC-115 atmospheric trend records before 2003. Thus our emission based scenario was only linearly modified to fit the Cape Grim air archive data. This correction is important (15%) for CFC-114 (Fig. 9). Sturrock et al. (2002) applied a correction to put UEA (Oram, 1999) Cape Grim air archive data on the AGAGE calibration scale. This rescaling (AGAGE/UEA=0.94 for CFC-114 and 0.96 for CFC-115) probably explains the 1 ppt difference in recent CFC-114 concentrations (Fig. 8) and affects less CFC-115 (0.35 ppt). The Sturrock et al. (2002) trend scenario for CFC-114 starts in 1960 because the two lowest depths sampled at their Law Dome sites showed undetectable concentrations. On the other hand, CFC-114 emissions show a peculiar behaviour with an abrupt increase between 1944 and 1946, followed by a stabilization in 1950–1960 (Fig. 9). This results in a steeper and more constant slope at the beginning of our trend scenario than for other CFCs. Overall, our model results are consistent with Sturrock et al. (2002) for the recent past, but we obtain higher levels of CFC-114 before 1978 and lower levels of CFC-115 in 1975-1985 (Fig. 8). We should note that for CFC-113 and CFC-115, the Law Dome firn air record suggests higher concentrations than the early Cape Grim air archive (around 1980) (Sturrock et al., 2002).

In order to test if the above differences in time trends significantly affect our firn depth profiles, we ran our firn diffusion model with the Sturrock et al. (2002) atmospheric scenarios (Fig. 10). The relatively small difference in CFC-113 trend scenarios mostly impact the DML and Dome C

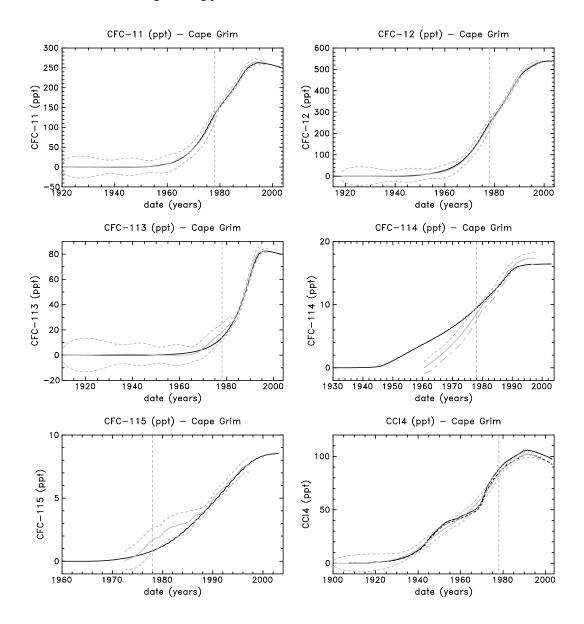


Fig. 8. Comparison of our concentration trends (in black) with the Sturrock et al. (2002) trends (in grey) at the latitude of Cape Grim. Continuous black lines represent 2-D chemistry model trends calculated with emissions adjusted to AGAGE data except for CCl₄, for which the AGAGE derived trend is represented by the black dashed line, whereas the black continuous line represents the trend calculated with emissions adjusted to ESRL data (this scale is more consistent with UEA data for CCl₄). The grey lines represent the Sturrock et al. (2002) mean (continuous line), minimum and maximum (dashed lines) scenarios. The vertical grey line illustrates the start date of the Cape Grim air archive record used by Sturrock et al. (2002) to constrain their trends.

firns, where relatively shallow depth levels (before the steep part of the firn records) are affected. The data-model agreement is better with our emission based trend than with the Sturrock et al. (2002) trend for this gas. However, the effect of the difference between scenarios on CFC-113 depth profiles in firn is of similar magnitude as the uncertainty on CFC-113 diffusion coefficient used in firn models (see supplement http://www.atmos-chem-phys.net/9/3911/2009/acp-9-3911-2009-supplement.pdf). The larger difference in

scenarios for CFC-115, affecting the same age range (20–25 years old air for the sites drilled before year 2000), also affect Devon Island results. At this site, the Sturrock et al. (2002) trend slightly improves the model-data agreement. The Sturrock et al. (2002) scenario somewhat overestimates the data at DML and Dome C whereas our scenario underestimates them, especially at Dome C. CFC-115 atmospheric history thus may have an intermediate behaviour between the two tested scenarios. Although the data dispersion is smaller

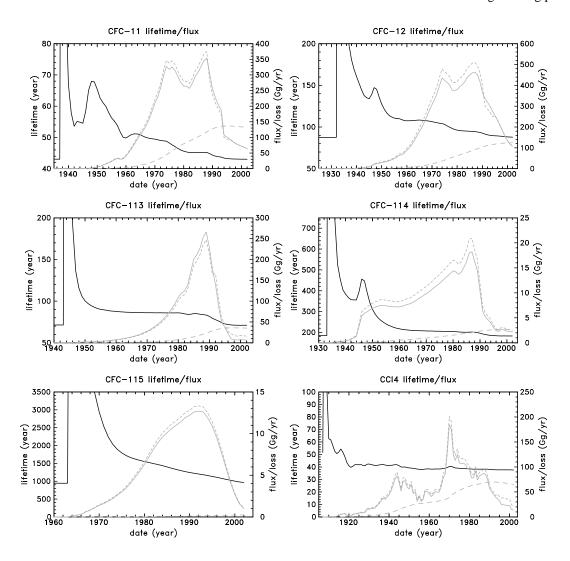


Fig. 9. Halocarbon emissions, atmospheric losses and lifetimes. Continuous grey lines: industrial inventory based emissions, short dashed lines: emissions corrected to fit recent atmospheric concentration trends, long dashed lines: atmospheric losses, black lines: atmospheric lifetimes (see Sect. 6.3).

for CFC-114 than for CFC-115, the model-data differences are larger for CFC-114 than for other CFCs, especially at Dome C. This trace gas is the only species for which our emission based scenario clearly differs from the envelope defined in Sturrock et al. (2002). The Sturrock et al. (2002) scenario improves the model data agreement at Arctic sites. The bottom part of Berkner record, showing a very pronounced slope change around 50 m depth, shows a small sensitivity to the large difference in scenarios. Both the Sturrock et al. (2002) and our scenario overestimate the DML and Dome C records in the depth range sensitive to 20-25 years old air. Although different, the two scenarios both induce important differences with Dome C data around 70-95 m depth. Thus no scenario can be considered as fully satisfactory. Overall, considering the large uncertainties on the CFC-114 and CFC-115 budgets (see Sect. 6.4), the Sturrock et al. (2002)

scenarios derived from Law Dome firn data are mostly consistent with our emission based scenarios.

CFC-113 and CFC-114 concentrations at the deepest firn levels show a behaviour similar to CFC-11 and CFC-12: very low concentrations are measured at the bottom of all firn sites, with lower levels at Arctic sites (0.4–1 ppt CFC-113 and 0.2–0.8 ppt CFC-114) than Antarctic sites (1.5–3.5 ppt CFC-113 and 0.7–1 ppt CFC-114). Our lowest CFC-115 values show very low concentrations at all sites: below the detection limit of NCAR measurements in the Arctic and <0.25 ppt in Antarctica, suggesting a later start of its atmospheric emissions than for CFC-114, which is confirmed by the emission data (Fig. 9). The Sturrock et al. (2002) CFC-114 and CFC-115 concentrations at Law Dome fall below their 0.1 ppt detection limit. Unlike other CFCs, CFC-113 shows similar concentrations near the firn-ice transition at Law Dome

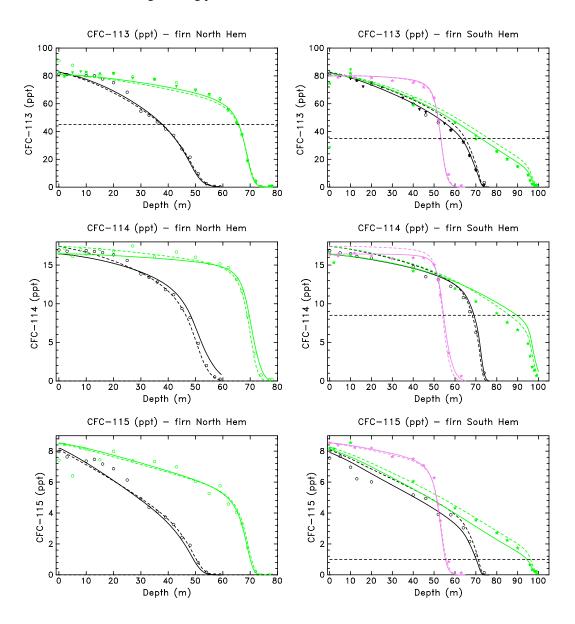


Fig. 10. Comparison of the firn model results with our emission-based scenarios (continuous lines) and the Sturrock et al. (2002) concentration-based mean scenarios (dashed lines) for CFCs 113, 114 and 115 with FIRETRACC and CRYOSTAT data (symbols). See Fig. 5 for drilling sites and measurement lab. symbols. The Sturrock et al. (2002) trend scenarios were extrapolated from 1997 to 2003 in order to simulate the North GRIP and Berkner firn data.

(0.38–0.74 ppt) and at our Arctic sites. Thus we suggest that the Sturrock et al. (2002) value might be somewhat overestimated. This would be qualitatively consistent with the overestimation of our DML and Dome C records by their scenario.

6.2.3 Carbon tetrachloride (CCl₄)

CCl₄ has the longest emission history, starting in 1908 (Fig. 9). On the other hand, experimental uncertainties are high for this species: our measured concentrations show a

significant spreading, and the AGAGE-adjusted and ESRL-adjusted chemistry model scenarios differ by 3.7 to 5% due to inter-calibration differences between the two records. Both these scenarios are represented in Fig. 11 because the ESRL-adjusted trend is more consistent with our observed concentrations, whereas the AGAGE-adjusted trend is consistent with the Sturrock et al. (2002) scenario. Although these last two scenarios are very similar (Fig. 8), the Dome C firn simulations show higher values around 90 m depth with the Sturrock et al. (2002) trend than with our AGAGE-adjusted scenario. This is likely due to higher val-

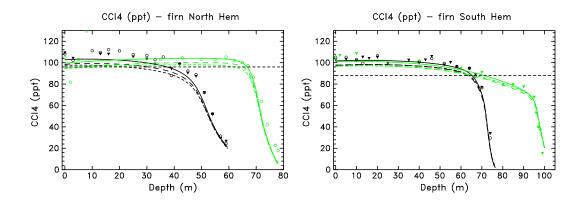


Fig. 11. Comparison of the firn model results with our emission-based scenarios (continuous lines: emissions adjusted to ESRL data, long dashed lines: emissions adjusted to AGAGE data) and the Sturrock et al. (2002) concentration-based mean scenarios (short dashed lines) for CCl₄ with FIRETRACC and CRYOSTAT data (symbols). See Fig. 5 for drilling sites and measurement lab. symbols. There is no reliable CCl₄ record at Berkner Island.

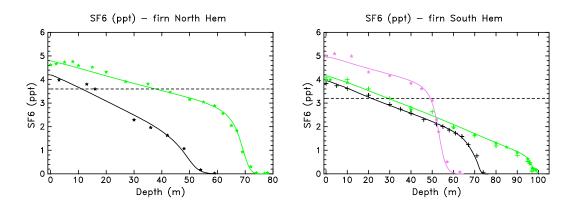


Fig. 12. Comparison of the firn model results (continuous lines) for SF_6 with FIRETRACC and CRYOSTAT data (symbols). Measurements made at UEA are shown as stars, those made at MPI (Sturges et al., 2000; FIRETRACC, 2007) as plus signs. Drilling sites representation: see Fig. 5.

ues in the smooth Sturrock et al. (2002) scenario around 1965–1970 (Fig. 8). Taking into account the rather high firn data dispersion, our ESRL-adjusted scenario is globally consistent with the data except at Devon Island. We should remember that the Devon Island firn contains numerous refrozen melt layers which may affect its diffusivity (especially around 27 m depth, see Sect. 5). The CCl4 atmospheric record shows a pronounced maximum around 1990 (see supplement http://www.atmos-chem-phys.net/9/3911/2009/acp-9-3911-2009-supplement.pdf). The modeldata discrepancies in the Devon Island firn may result from an incorrect modelling of this feature due to higher uncertainties in the diffusivity at this site.

This early-emitted gas shows higher concentrations at the firn bottom than CFCs (18–35 ppt). Non zero CCl₄ levels were also obtained by Butler et al. (1999) and Sturrock et al. (2002). Our atmospheric scenarios start with zero concentrations in 1907 and also lead to non zero modelled concentra-

tions at the last measurement depths in the firn for all sites. Taking into account the uncertainties on the deep firn data and modelling discussed in Sect. 5, the overall consistency between the model and measured concentrations indicates that an atmospheric scenario with null CCl₄ concentrations before 1908 can account for the measurements in firn. However, the high uncertainties on the CCl₄ lifetime and budget (Sect. 6.3 and Sect. 6.4) preclude a quantification of CCl₄ natural versus anthropogenic sources using our firn air data and models. Performing CCl₄ measurements in ice cores is likely the most direct way of better constraining the pre-industrial CCl₄ concentration.

6.2.4 Sulphur hexafluoride (SF₆)

As for CFC-11 and CFC-12 (Fig. 5) our model and SF₆ concentration measurements show an excellent agreement at our five firn drilling sites (Fig. 12). Such a good agreement is

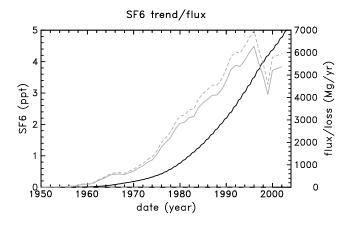


Fig. 13. SF₆ concentration trends at the latitude of Cape Grim (in black) and emission scenarios (in grey) constructed from industrial inventories (continuous line, see Sect. 2.1), and corrected to fit the ESRL atmospheric record (dashed line).

obtained only if a stabilization of SF₆ emissions in 1997– 2002 (Fig. 13) is taken into account. Maiss and Brenninkmeijer (2000) observed a slope reduction in the SF₆ atmospheric trend at Izana after 1996–1997, which suggest a 27% reduction in SF₆ emissions. Smythe (2004) inventoried sales data from 7 major companies and conclude that a 33% decrease in SF₆ sales occurred in 1996-1999, followed by a 27% increase in 1999-2003. Smythe (2004) data were used to extrapolate the EDGAR (2007) emission trend from 1996 to 2003 (Fig. 13). This emission scenario had to be increased by 11% to reach ESRL SF₆ concentration levels in 1995-2002, and model/data agreement on the recent slope of the trend could be reached (see supplement http://www.atmos-chem-phys.net/9/3911/2009/ acp-9-3911-2009-supplement.pdf). In the firn, an emission scenario which does not take into account a decrease in SF₆ emissions around 1997-1999 would simulate correctly the firn data at the sites drilled in 2002-2003 but underestimate the upper firm SF_6 data at the sites drilled in 1998–1999.

Our modelled SF₆ trend (Fig. 13) is consistent with the trend derived by Maiss et al. (1996) from the Cape Grim air archive and atmospheric measurements although slightly overestimating it (by less than $0.2\,\mathrm{ppt}$, see supplement http://www.atmos-chem-phys.net/9/3911/2009/acp-9-3911-2009-supplement.pdf). It is also qualitatively consistent with those calculated by Butler et al. (1999); Sturges et al. (2000) with similar methodologies.

 SF_6 concentrations measured at the deepest firn sampling level range between 0.03 and 0.05 ppt at Arctic sites and 0.09–0.18 ppt at Antarctic sites. Similarly low concentrations were measured by Butler et al. (1999). A small natural SF_6 source from crustal degassing has been identified (e.g. Deeds et al., 2008, and references therein). Deeds et al. (2008) estimate the total lithospheric flux of SF_6 to be 13–

 $80\,\text{kg/yr}$, which would sustain an atmospheric equilibrium concentration of 0.0004–0.0098 ppt with an SF_6 lifetime of 800–3200 years. To our knowledge, the lowest SF_6 concentration measured in firn air is 0.0063 ppt at Law Dome (Vollmer and Weiss, 2002). Our North GRIP record indicates a stabilization of SF_6 concentrations in the 74–78 m depth range, however the suspicion of some contamination at 78 m depth (Sect. 4) and our detection limits (Sect. 5) preclude a firm estimate of the pre-industrial SF_6 concentration from this site. Our North-GRIP data suggest that future high-precision SF_6 measurements in central Greenland firn air could help confirming the Law Dome result.

6.3 Halocarbon lifetimes

Atmospheric lifetimes are often considered as constants. Locally (or in box models), lifetime is defined as the species concentration (e.g. in mol/cm³) divided by its chemical loss (e.g. in mol/cm³/s). As the chemical loss rate is always proportional to the species concentration, the above ratio can be simplified and the local lifetime is independent from the species concentration. However, atmospheric chemical loss rates undergo temporal (e.g. in relation to day/night cycles) as well as spatial (e.g. in relation with the UV increase with altitude) variations. In chemistry models, which explicitly calculate atmospheric losses of chemical species, lifetimes are calculated as the global average concentration divided by the global average loss. Our target gases are destroyed only in the middle atmosphere, thus their lifetimes in the troposphere are virtually infinite. Since these gases are emitted at the ground, it takes them about a decade to reach the stratosphere where they are destroyed. This effect is visible in Fig. 9: the lifetimes show out of scale values at the beginning of the species emissions. Once halocarbons have reached their destruction region, their lifetimes decrease but remain sensitive to temporal changes in their emissions (with delay) (Fig. 9). Species destroyed in the troposphere (in the vicinity of their emission regions), such as methane or hydrogenated CFC substitutes do not exhibit such temporal variations of their lifetimes.

CFCs are commonly modelled without an explicit calculation of their lifetimes, but by imposing a homogeneous and constant loss rate based on literature values of their lifetimes. For example, this is the case in tropospheric models which use CFCs as transport tracers. In order to evaluate the effect of halocarbon lifetime variations on their concentration trends, three test simulations were performed. The first simulation aimed at evaluating the equilibrium lifetimes of our target species. The model was initialized with homogeneous halocarbon concentrations and run for 100 years with zero halocarbon emissions. In their destruction region, the local halocarbon lifetimes are much shorter than their global average lifetimes, thus the results stabilize rapidly. This is illustrated by the constant lifetimes obtained before the start of emissions in Fig. 9. In the second simulation, our halocarbon

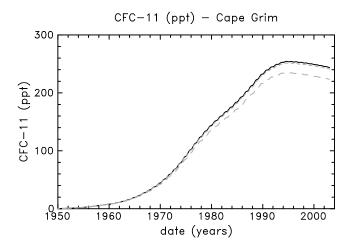


Fig. 14. Impact of spatial and temporal lifetime changes on CFC-11 concentration trend. Continuous line: reference concentration trend. Long dashed line: concentration calculated with a constant and homogeneous lifetime (equilibrium lifetime). Short dashed line: concentration calculated by imposing the equilibrium lifetime as a scaling factor to the calculated CFC-11 sink.

loss rates were calculated as the inverse of the above equilibrium lifetimes times their concentration, thus imposing constant and homogeneous lifetimes. Finally, in order to separate the effects of spatial and temporal lifetime variations, a third simulation was performed: the chemical losses were calculated, and then rescaled so that the global average loss is consistent with the equilibrium lifetime. This suppresses the temporal changes in lifetimes while preserving their spatial structure. The results for CFC-11 are shown in Fig. 14. The concentration trend calculated with our third test run remains close to the reference simulation. Thus the temporal changes in lifetimes, which mainly occur in the early stage of historical emissions, have a small impact on the concentration trend. On the other hand, imposing a spatially homogeneous lifetime (second test run) leads to a significant decrease in CFC-11 concentrations. Thus tropospheric models or transport models, which calculate CFC concentrations with imposed lifetimes can produce a significant bias. Similar results for other halocarbons are provided in the supplementary material (http://www.atmos-chem-phys.net/9/3911/ 2009/acp-9-3911-2009-supplement.pdf). They show that this bias is most important for the relatively short lived gases, which have the most important chemical loss at the time scale of our simulation, whereas it is negligible for the very long lived CFC-115.

Our modelled equilibrium lifetimes for CFC-11, CFC-12 and CFC-113 are roughly consistent with the recommendations from WMO (2007) (Table 4). The atmospheric loss of these major CFCs is mostly due to photodissociation. CFC-114 and CFC-115 show larger differences with recommended values, which were established by Ravishankara

Table 4. Comparison of our chemistry model lifetimes with recommended lifetimes from WMO (2007) ("Life. WMO"). "Life. eq." is the equilibrium chemical lifetime, "Life. photo." and "Life. $O(^1D)$ " display the equilibrium lifetimes due to photodissociation only and $O(^1D)$ reaction only respectively. "Life. 1975" and "Life. 2002" are our transient modelled chemical lifetimes for these years. The WMO (2007) lifetime for CCl_4 includes the effect of its oceanic loss.

Gas	Life.	Life.	Life.	Life.	Life.	Life.
	WMO	eq.	Photo.	O(¹ D)	1975	2002
CFC-11	45	43	44	273	48	43
CFC-12	100	88	91	351	103	88
CFC-113	85	72	74	287	86	71
CFC-114	300	186	224	411	205	183
CFC-115	1700	948	1540	2130	1793	965
CCl ₄	26	38	38	200	39	38

et al. (1993). More recent cross section data from Sander et al. (2006) were used for photodissociations, whereas the O(¹D) reaction rates have not been re-evaluated and the recommended values are still from Ravishankara et al. (1993). Although the 85 km upper limit of our model is lower than the 116 km in Ravishankara et al. (1993) and we thus miss some non negligible processes such as the Lyman-alpha photodissociation of CFC-115, we obtain lower lifetimes (i.e. higher atmospheric losses) than Ravishankara et al. (1993). However, at the timescale of our simulations, chemical losses play a minor role on the budget of these very long lifetime gases (Fig. 9). Our chemical equilibrium lifetime for CCl₄ is consistent with the 35 years WMO (2007) estimate. However these lifetimes do not take into account an important oceanic sink which reduces CCl₄ lifetime to approximately 26 years, nor a more recently discovered soil sink which could be of similar magnitude (WMO, 2007). WMO (2007) suggests that such short lifetimes would be difficult to reconcile with emissions deduced from high frequency atmospheric measurements or lifetimes deduced from CCl₄ meridional gradients. On the other hand, the long-term inventory-based emissions of CCl₄ are not well constrained, and the overall sources/sinks balance is poorly understood for CCl₄.

In 1975, long lived halocarbon emissions were still increasing or fairly close to their peak. Our modelled lifetimes for this period are higher than equilibrium lifetimes and generally closer to the recommended values. In 2002, halocarbon emissions had been strongly reduced and our model lifetimes are very close to their equilibrium values.

6.4 Species budgets and budget uncertainties

Because our long-lived target gases have slow destruction rates, their budgets have long been dominated by their emissions. Following the strong decrease of these emissions

Table 5. Halocarbon and SF₆ budget analysis. The first three columns provide estimates of uncertainties on emissions accumulated over time: WMO (2007) values (from Table 1–7) are compared with the difference between our inventory-based and adjusted emissions to fit the AGAGE and ESRL records. The uncertainties on losses are the WMO (2007) $(1/\tau_{min}-1/\tau_{max})/(1/\tau_{best})$ calculated from Table 1–7 (column WMO), and the maximum differences in our modelled cumulated losses when using kinetic and photochemical constants from the last three JPL evaluations (DeMore et al., 1997; Sander et al., 2003, 2006) (column JPL). Calibration uncertainties estimated by WMO (2007) are compared with the UEA-NCAR differences in our firn air samples. Concentration uncertainties show the impact on year 2002 concentrations of our emission corrections to adjust AGAGE and ESRL measurements. Finally the last two columns display our chemistry model best estimate of emissions and losses accumulated over time until 2002.

Gas	Δ Emission (%)		Δ Loss (%)		Δ Calibration (%)		Δ Concentration (%)		Cumul. budg. (10 ⁹ kg)		
	WMO	AGAGE	ESRL	WMO	JPL	WMO	UEA-NCAR	AGAGE	ESRL	Emission	Loss
CFC-11	18	+3.6	+7.3	50	<2	<2	+1	+3.0	+6.3	9.0	3.5
CFC-12	7.5	+7.1	+6.3	38	<1	<2	+1.5	+6.5	+5.7	14.4	2.9
CFC-113	8.4	-2.4	-2.3	43	<1	<2	+1	-2.0	-1.8	3.3	0.6
CFC-114		+15	+15	_	< 0.5		+20	_	+14	0.59	0.07
CFC-115		_	+5.0	_	< 0.2		+20	_	+4.9	0.25	0.003
CCl ₄	34	_	+11	77	<1.5	3–4	+6	+6.4	+11	4.9	2.8
SF ₆		_	+11	_	-			_	+10.5	0.14	0.0005

under the Montreal protocol, our results indicate that photochemical destruction exceeds emissions for most of these species (Fig. 9). This change occurred approximately in 1993 for CFC-11, 2001 for CFC-12, 1995 for CFC-113, and 1996 for CFC-114. For the shorter lived CCl₄, atmospheric destruction dominates known emissions since 1990, but photochemical destruction has been influential since the early 1950's (Fig. 9). This is consistent with the recent stabilization or decrease of atmospheric concentrations of this gas (WMO, 2007). On the other hand the budgets of the longest lived gases, CFC-115 and SF₆, are still largely dominated by their emissions. For these two compounds, the comparison of our modelled concentration trends with measurements is thus mostly indicative of the reliability of emissions. It is not the case for the recent part of other species records, which progressively become mostly influenced by atmospheric losses and thus become controlled by the species lifetimes.

CFC-113 has the most reduced emissions, however an important increase in these emissions (+140% after 1996) was needed to reconcile our model results with the recent observed trend in CFC-113. It reflects the fact that our initial model results showed a too strongly decreasing slope in recent years (Fig. 2). As chemical losses largely dominate emissions at that time (Fig. 9), this discrepancy could be due to a smaller bias in CFC-113 lifetime. A test simulation showed that a 40% reduction of its chemical losses during that period also fits the concentration data. CFC-113 lifetime is then 84 years (instead of 71 years) in 2002.

By contrast, no emission correction was needed for CFC-11 and 12 after 1995 and 1994 respectively. This suggests that the lifetimes of these compounds are consistent with their recent trends. However, their emissions in 2002 are still equivalent to 48% of the CFC-11 loss and 86% of CFC-12 loss (Fig. 9). Thus we cannot firmly conclude on the consistency of major CFC lifetimes with their concentration trends.

We should note that the increasing difficulty in estimating CFC emissions due to the dominant role of non reporting countries (see Sect. 2.1) may limit our ability to validate CFC lifetimes with their trends in the future. CCl₄ emissions were increased by 5% until 1993 and 25% afterwards in order to fit its concentration trend. A much larger increase would be required if its oceanic and soil sinks were taken into account. However, the overall consistency of our firn model results with CCl₄ data suggests that the historical variations in CCl₄ emissions are generally correct.

Table 5 summarizes budget uncertainties for our target gases. Calibration uncertainties play a minor role except for CFC-114 and CFC-115. Our estimates of uncertainties on emissions and concentrations are not independent: they both reflect the emission adjustments necessary to fit atmospheric measurements. These values vary from one gas to another, they are small (<2.5%) for CFC-113, intermediate (5–10%) for CFC-11, CFC-12 and CFC-115 and exceed 10% for CFC-114, CCl₄ and SF₆. Our uncertainties on calibration and emissions are globally consistent with WMO (2007). The highest uncertainties reported in WMO (2007) concern the halocarbon lifetimes. In Table 5, they are expressed in terms of atmospheric losses. The main sink of our target halocarbons is photodissociation. Their cross sections have been re-evaluated several times over the last decade. We compared modelled CFC losses calculated with three different chemical schemes derived from DeMore et al. (1997); Sander et al. (2003, 2006). The differences obtained are always lower than 2%. On the other hand, WMO (2007) large uncertainties on atmospheric losses for major CFCs (38–50%) are based on somewhat older estimates derived from modelbased values (WMO, 1999) and stratospheric measurements performed in 1994 (Volk et al., 1997). Part of these values are probably inconsistent with the recent concentration trends, which are more dependent on loss processes, thus

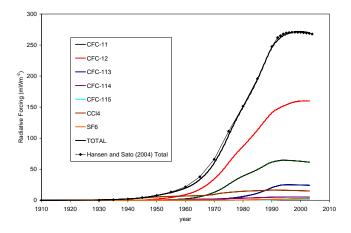


Fig. 15. Time series of radiative forcing for the seven gases in this study, together with the total forcing, compared with the forcing derived by Gohar and Shine (2007) using the Hansen and Sato (2004) concentration time series.

an updated evaluation of CFC lifetimes and lifetime uncertainties would improve the understanding of these species budgets. The very high WMO (2007) uncertainty for CCl₄ (77%) integrates the soil and oceanic sinks not taken into account in this study.

The last two columns in Table 5 provide our best estimates of the total emissions and losses accumulated over time between 1908 and 2002. Although atmospheric losses now equal or exceed annual emissions for most of our species, the previous domination of emissions still has a strong imprint on the accumulated budgets. This effect increases with the species lifetime: 57% of the emitted CCl₄ has been destroyed whereas only 39% of CFC-11, 20% of CFC-12 and 18% of CFC-113 had been eliminated from the atmosphere by 2002. These percentages fall to 11%, 1%, and 0.4% for the longer lived CFC-114, CFC-115 and SF₆ respectively.

6.5 Radiative forcing

The concentrations generated by the 2-D chemistry model can be used to calculate radiative forcings. The method used here follows Gohar and Shine (2007), where radiative efficiencies (W m⁻² ppbv⁻¹), together with the global-mean surface values derived from the model, are combined to give the forcings. Figure 15 shows the time-series of the radiative forcing for each of the 7 gases and the total. As expected, the total forcing in 2003 is dominated by CFC-12, which has the third largest radiative forcing among well mixed greenhouse gases (IPCC, 2007), followed by CFC-11, CFC-113 and CCl₄ which also contribute significantly. The current contribution from the other gases: CFC-114, CFC-115 and SF₆ is negligible, but their long lifetimes render their small contribution more permanent.

The 2-D model results constrained by the firn data gives improved confidence in the time variation of the forcing from these gases during the 20th century. As expected, given the linear relationship between the radiative forcings and concentrations of these gases, the forcing is seen to rise rapidly from the 1950s before starting to level off in the mid-1980s. For comparison, the forcings derived in Gohar and Shine (2007) using the time-series of gas concentrations used in the Goddard Institute of Space Studies climate model (Hansen and Sato, 2004, 2005) are also shown in Fig. 15. The CFC-11 and CFC-12 concentration histories built by these authors are based on ESRL measurements for the instrumental period and industrial production data prior to 1978, together with lifetimes of 50 and 100 years for CFC-11 and CFC-12 respectively. Overall agreement in the total radiative forcing is very good for the instrumental period. Our slightly lower values in 1990-1995 are probably due to differences between the AGAGE and ESRL CFC-12 records for this period (Fig. 3). Higher differences, reaching around $6 \,\mathrm{mWm}^{-2}$. are obtained in the 1970s. However, differences in forcing for the individual gases can approach 15%.

Although our target gases are responsible of only about 10% of the long-lived greenhouse gases radiative forcing (2.63±0.26 Wm⁻², IPCC (2007)), the control of the non-CO₂ greenhouse gas emissions plays an important role in efforts to avoid dangerous anthropogenic interference in the Earth's climate. As has been shown by Hansen and Sato (2004) and Velders et al. (2007), the control of CFCs emissions under the Montreal Protocol has played a significant role in reducing anthropogenic radiative forcing – indeed, as pointed out by Velders et al. (2007), the Montreal Protocol has likely had a greater influence than the Kyoto Protocol. As the Montreal Protocol has been successful in strongly reducing CFC emissions (Fig. 9), their future radiative impact should be increasingly dependent on their atmospheric loss rate (Sect. 6.4).

7 Conclusions

Polar firn is a large reservoir of old air, which allows for the measurement of numerous low concentration species. Trace gas measurements in firn air provide a unique tool to assess their concentration trends over the pre-instrumental period of the 20th century. The use of firn models allows the translation of firn-air records into atmospheric trends. Multi-species studies permit to better constrain the main model parameter: firn diffusivity, through the detection of systematic bias for a given depth range. On the other hand, multi-site studies better constrain the atmospheric trend scenarios, because their different characteristics (accumulation rate, temperature, etc.) lead to different trace gas age structures in firn. Globally, high accumulation and temperature sites provide results with best time resolution whereas low accumulation

and temperature sites are sensitive to earlier variations in atmospheric trends.

Two kinds of approaches have been used in order to infer atmospheric concentration trends: firn model inversion techniques or "top-down" methods (e.g. Rommelaere et al., 1997; Sturrock et al., 2002) and emission-based scenarios used in direct firn models and compared with data or "bottomup" methods (e.g. Butler et al., 1999). Our study used the "bottom-up" approach with a full atmospheric chemistry model which further allows to discuss the lifetimes and budgets of our target species. Our best-fit scenarios are directly compared with the Sturrock et al. (2002) "top-down" scenarios. The two approaches show a remarkable agreement for CFC-11 and CFC-12. It is due to both the consistency of early major CFC emissions with the firn records, and the relative smoothness of these emissions which lead to smooth atmospheric scenarios that can be derived from a small number of firn air measurements. By contrast, the two approaches lead to somewhat different scenarios for CFC-113, CFC-114, CFC-115 and CCl₄. Our emission based scenario leads to a better fit of our CFC-113 firn data, whereas an intermediate trend between the "bottom-up" and "topdown" derived scenarios would best fit the CFC-115 data. A less satisfactory model/data agreement is achieved with either scenarios for CFC-114 and CCl₄, which have higher data dispersions and inter-calibration corrections, and insufficiently constrained budgets. However this scenario comparison shows that fairly small and old differences in the atmospheric trends (e.g. CFC-113) can leave an imprint in the firn, especially at low-accumulation sites. As for CFC-11 and CFC-12, an excellent model/data agreement is obtained for SF₆, suggesting that the emissions of this very long lived gas are well constrained. However only an emission scenario which fits the slope reduction in the SF₆ concentration trend around 1998–2002 can be consistent with both firn sites drilled in 1998–1999 and those sampled in 2002–2003. Improved estimates of halocarbon historical concentration trends might be derived in the future from a "top-down" approach (inverse modelling of firn) constrained with data from ideally all available firn air pumping sites. The comparison with "bottom-up" studies would then allow to better assess the uncertainties on the species budgets (i.e. emissions and/or lifetimes).

The low concentrations observed at the last depth levels of all firn sampling sites confirm that pre-industrial concentrations of our target compounds were small (WMO, 2007). The consistency between these data and simulated concentrations in firn further suggests that non-zero emissions in the early 20th century are not required to account for our firn air records. However, a full determination of pre-industrial concentrations of our target gases requires high precision measurements in ice cores.

Our chemistry model results show important temporal variations in CFC and CCl₄ lifetimes. Such pollutants emitted at the Earth's surface and destroyed only in the strato-

sphere have no chemical loss in the troposphere, therefore their lifetime is virtually infinite at the beginning of their emission, since their transport to the stratosphere requires several years before chemical destruction can begin. The trends and budgets of halocarbons destroyed in the stratosphere are often calculated using tropospheric-only models. In such cases, the chemical sinks are not calculated but a constant and homogeneous sink (the inverse of the equilibrium lifetime) is imposed in the whole atmosphere. Specific tests simulations showed that this can induce a significant bias on the CFC and CCl₄ trends. Following the Montreal Protocol, the emissions of these ozone depleting substances have been strongly reduced, thus their trends and budgets are becoming increasingly dominated by their sinks. We believe that in order to better predict the ozone layer recovery, efforts should be made to update and reduce the large uncertainties on recommended CFC lifetimes. This would also improve the estimate of future radiative forcing by these gases.

Our results confirm that the budget histories of CFC-11, CFC-12, CFC-113, and SF₆ are well understood. However, our emission-constrained CFC-113 scenario shows a too quickly decreasing trend over recent years. It could result either from a strong underestimation of its sources or a smaller overestimation of its losses. The long-lived CFC-115 emissions are reasonably consistent with the Cape Grim air archive data and firn air records. On the other hand, the CFC-114 and CCl₄ budgets and trends are still insufficiently understood. When accumulated over time, the total emissions still largely dominate the total losses for our target species, due to their long lifetimes and the overall strong rise of their emissions during the twentieth century.

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