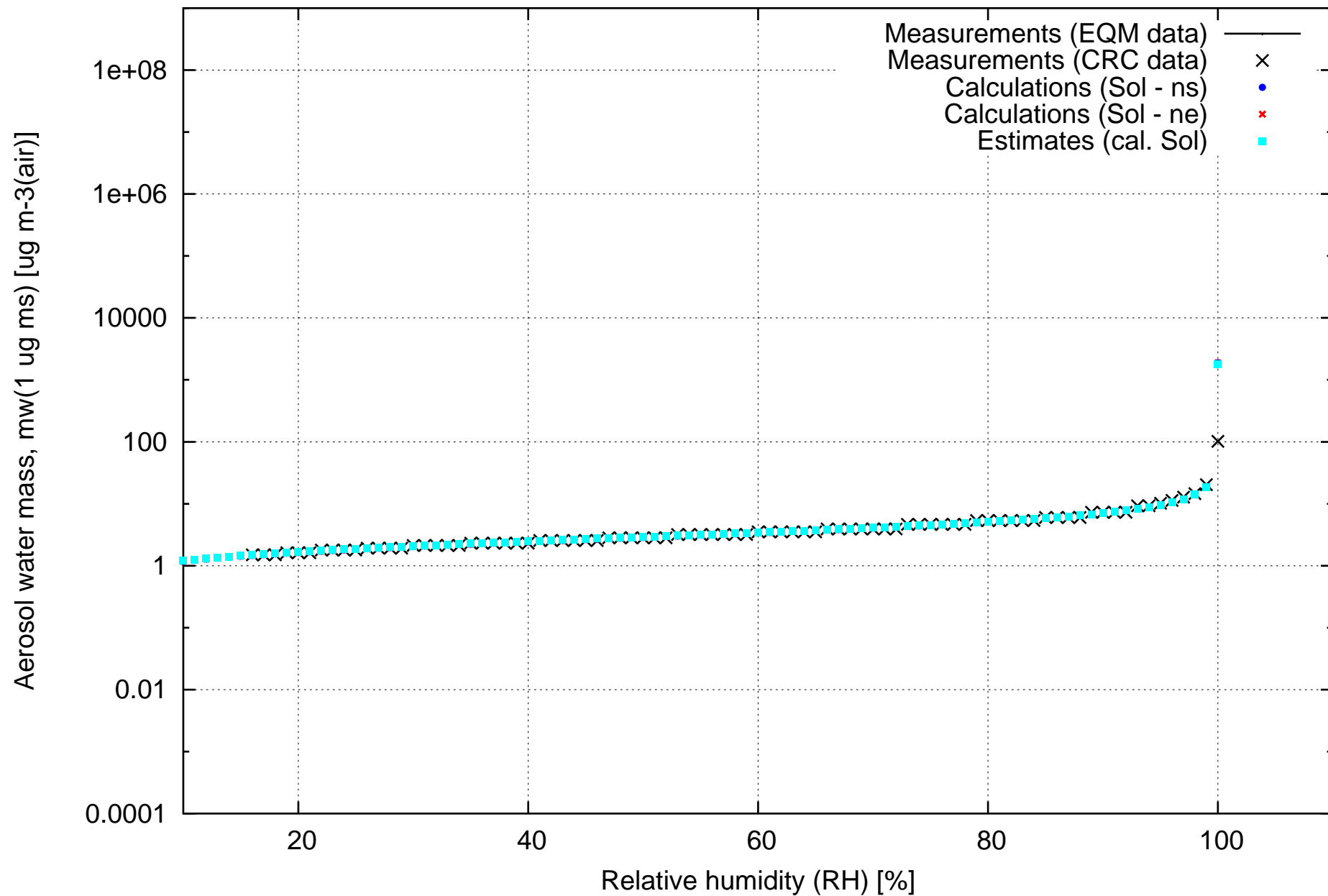
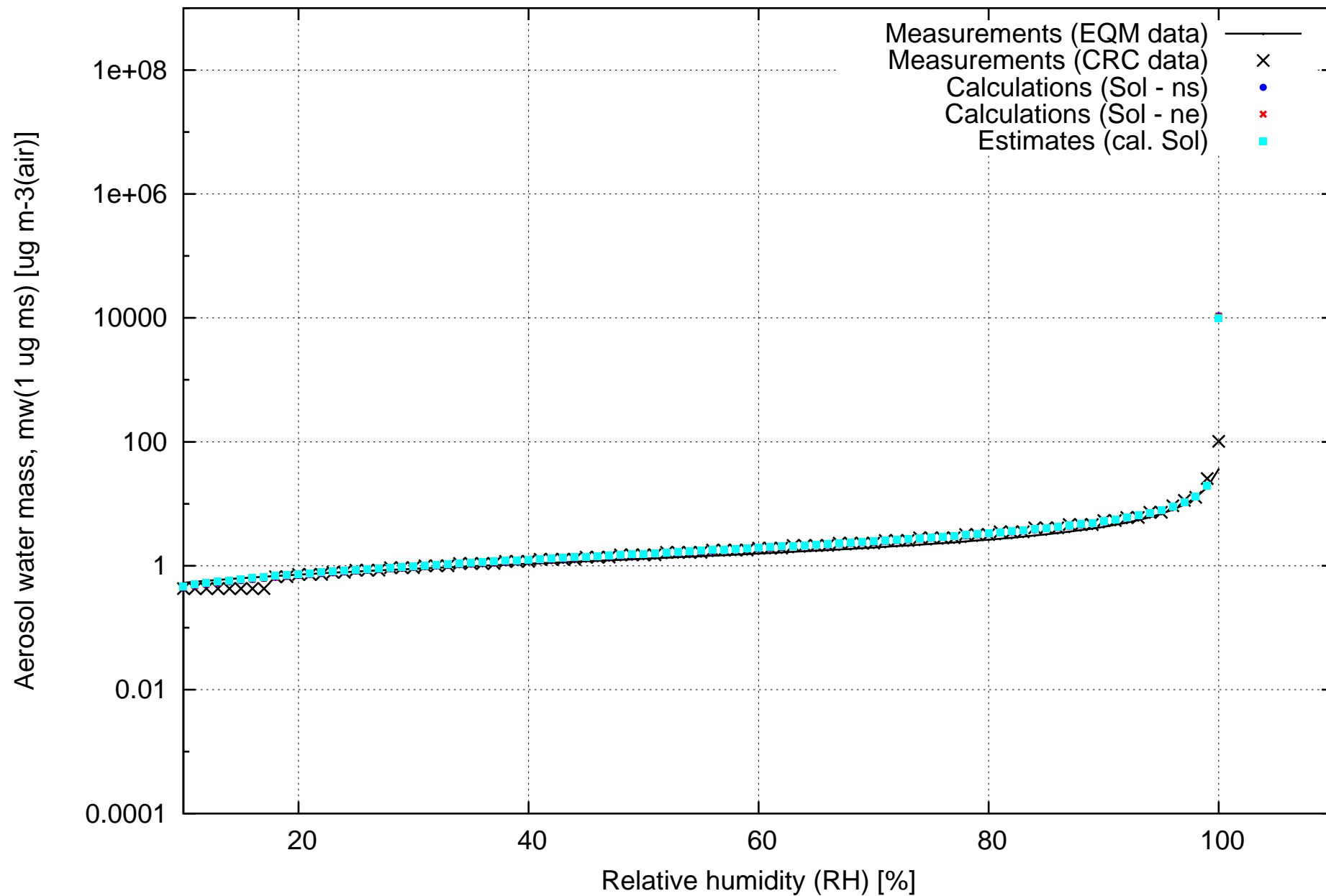


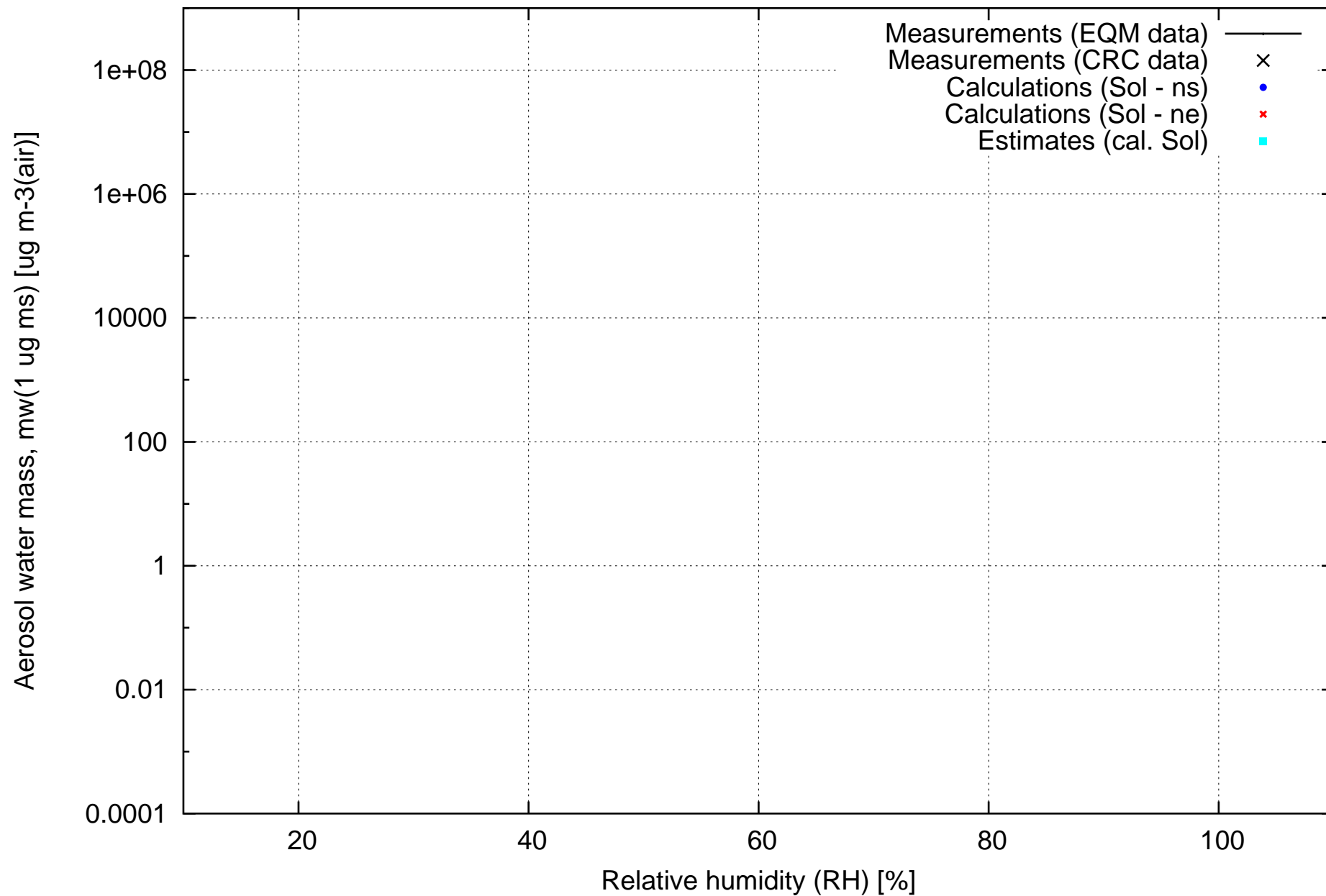
Phosphoric acid - H₃PO₄



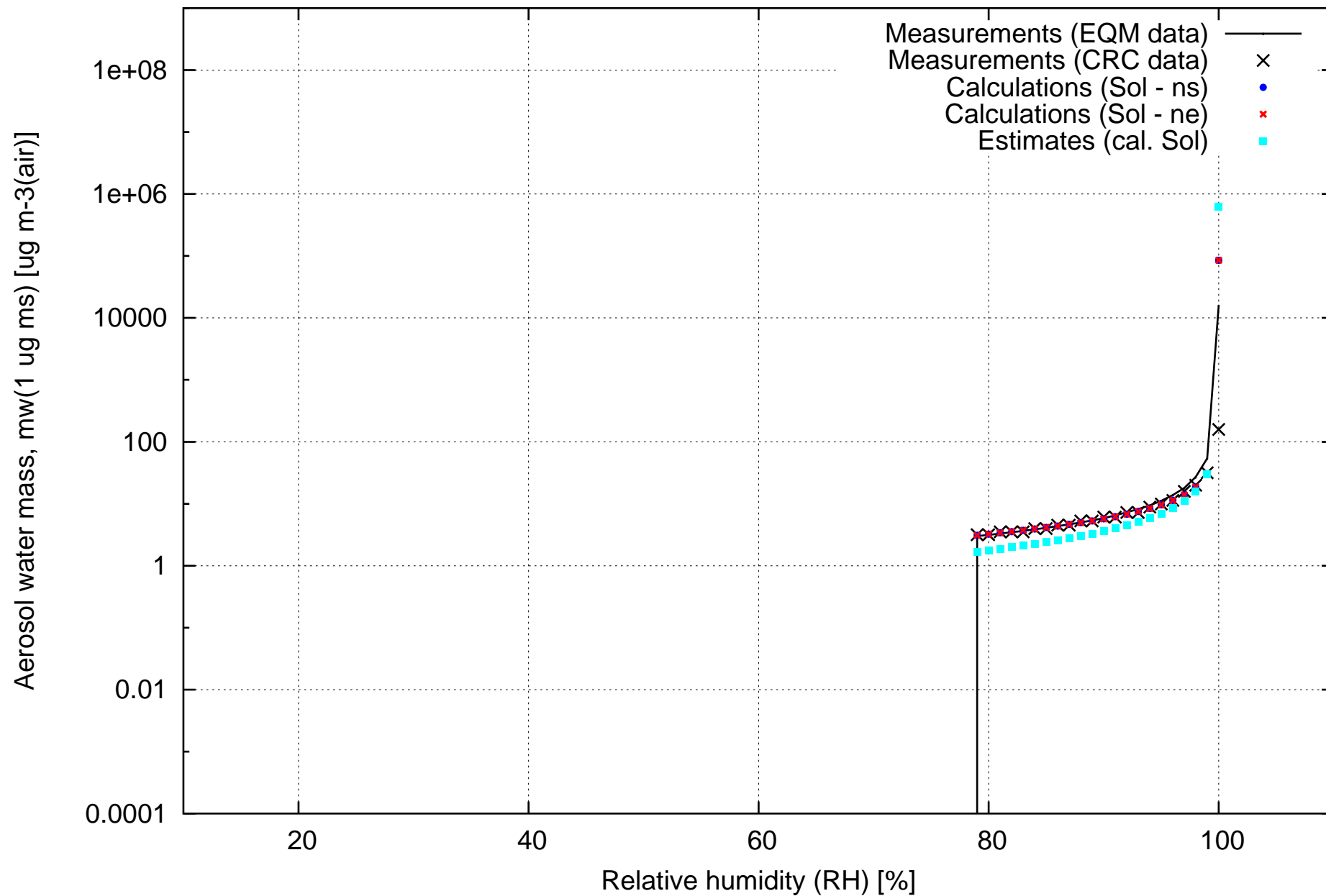
Sulfuric acid - H₂SO₄



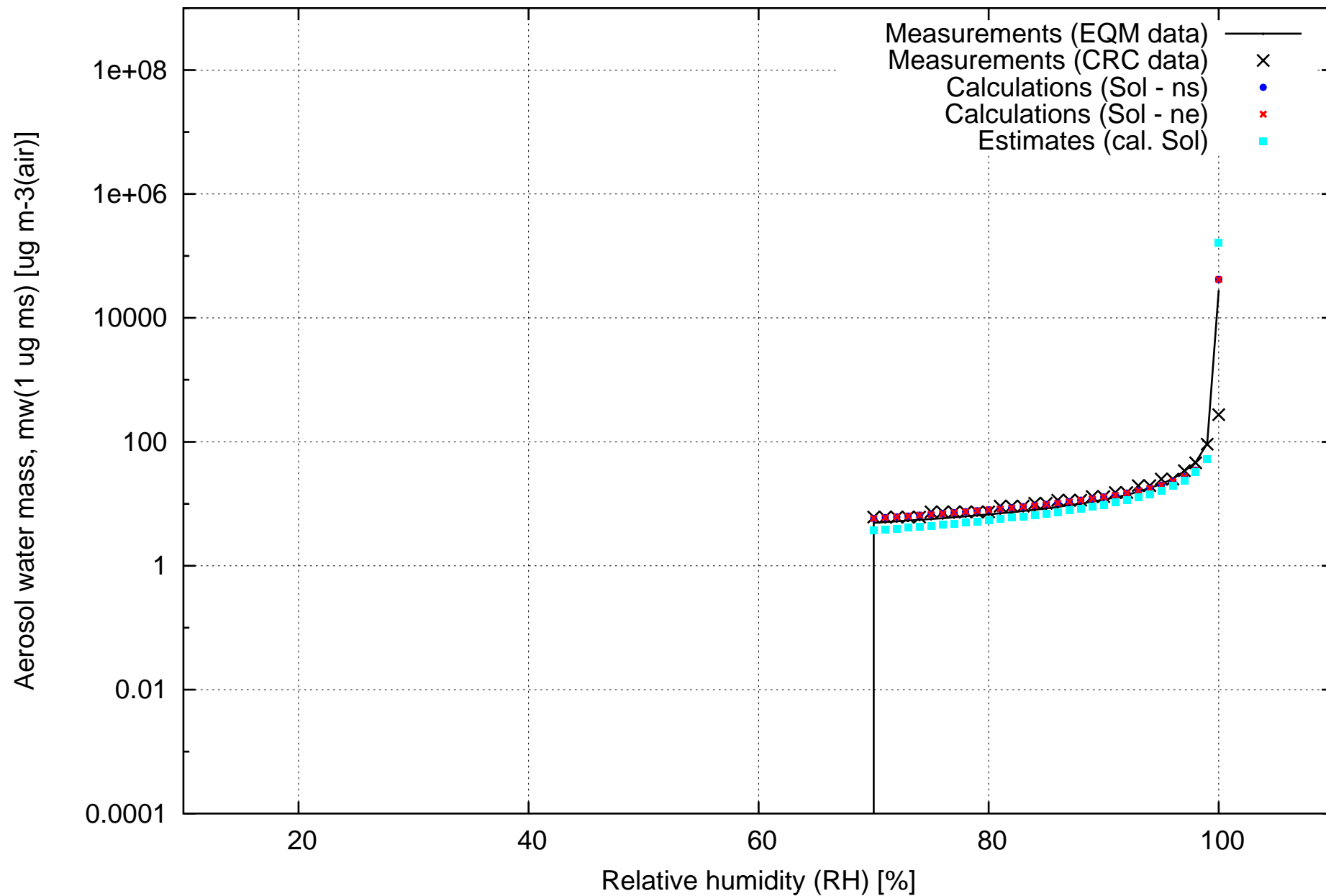
Hydrogen - dummy 03



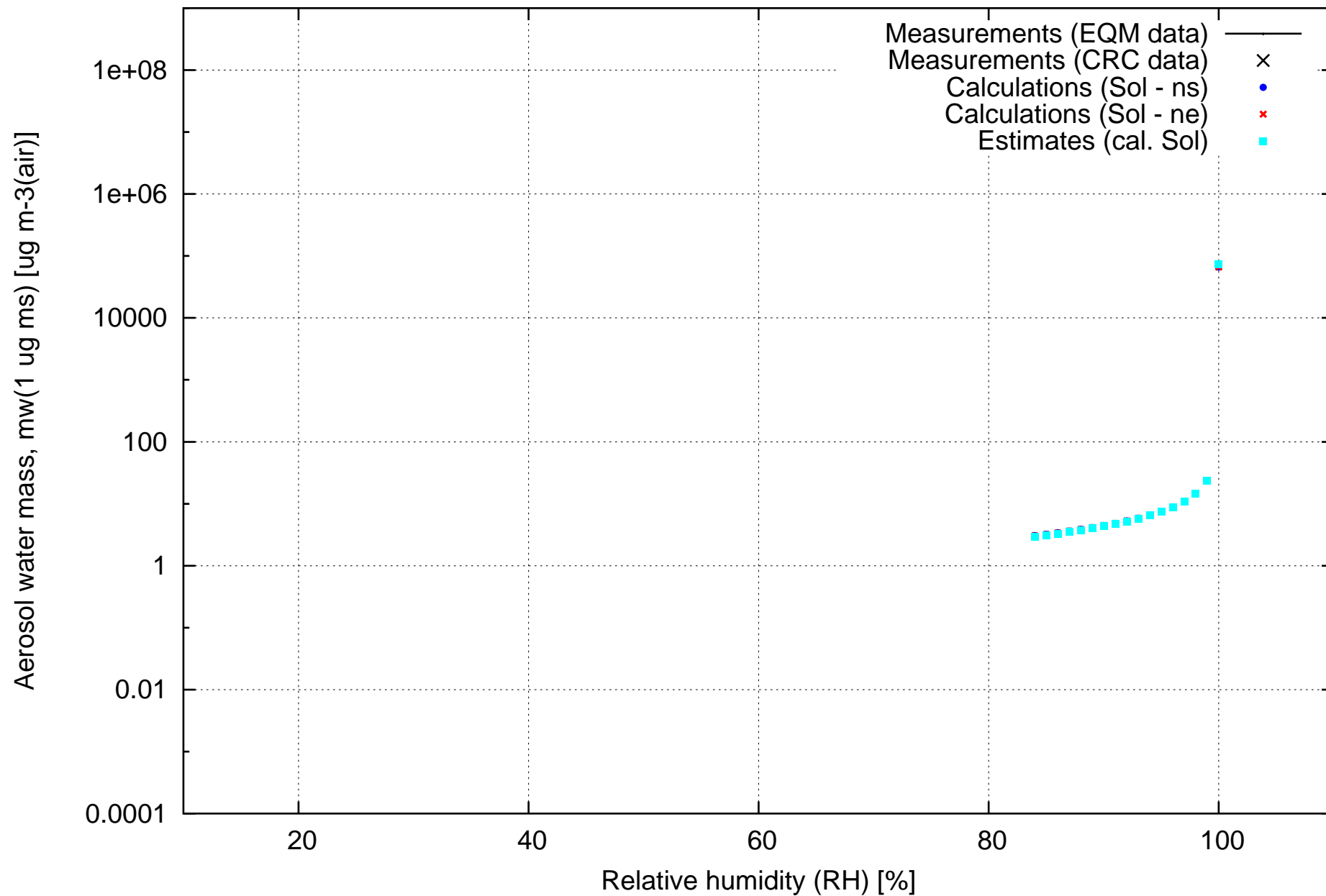
Nitric acid - HNO₃



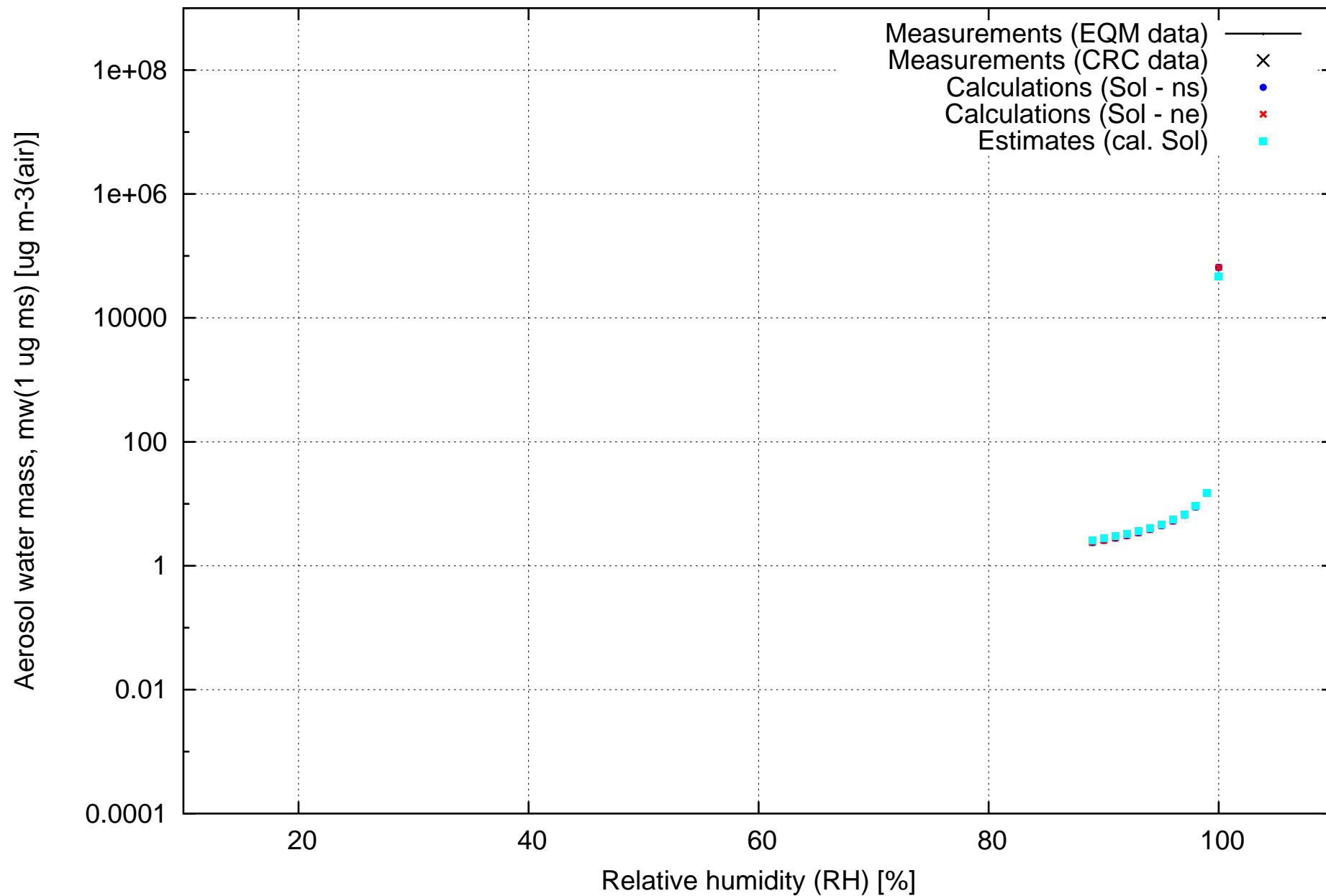
Hydrogen chloride - HCl



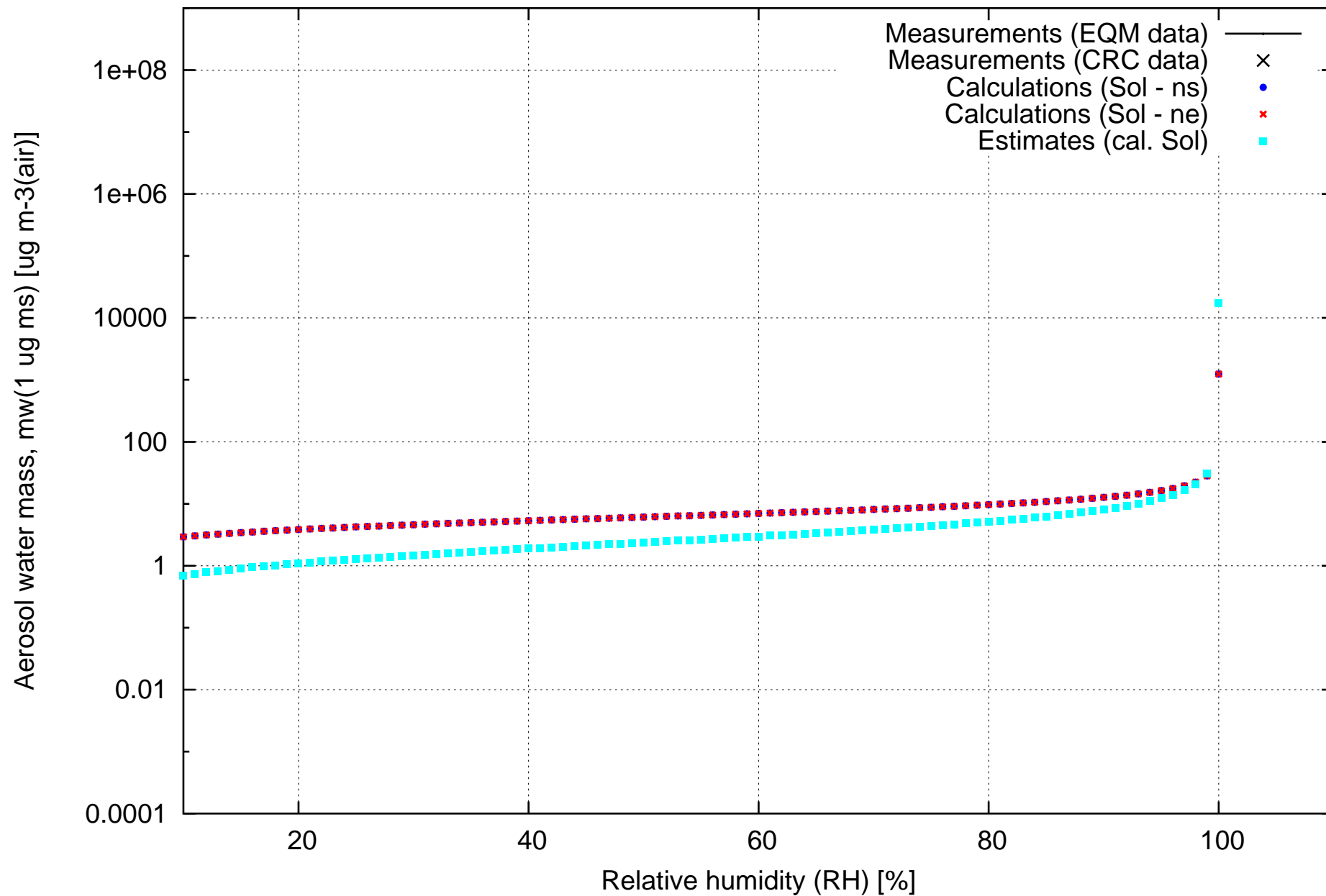
Hydrogen bromide - HBr



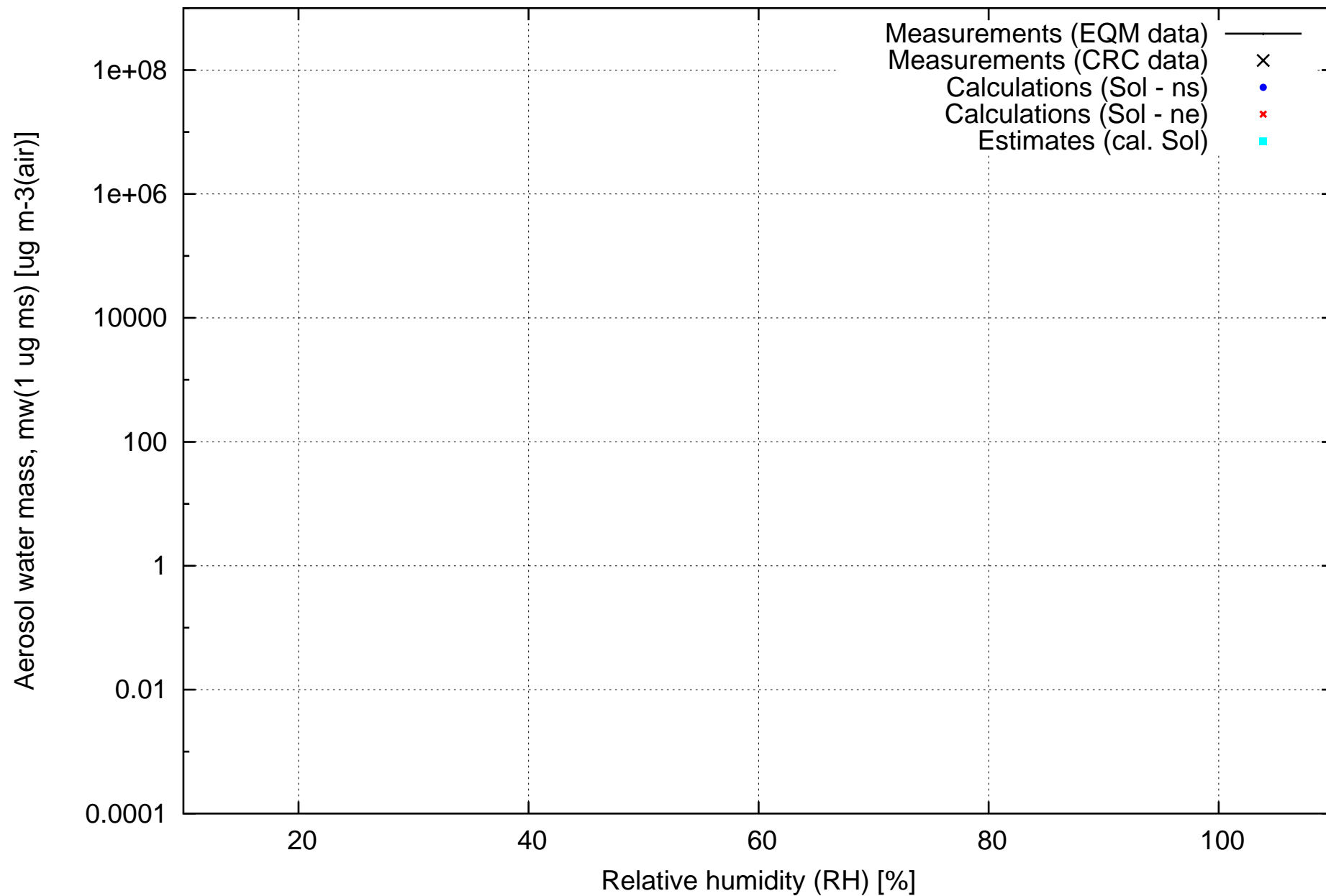
Hydrogen iodide - HI



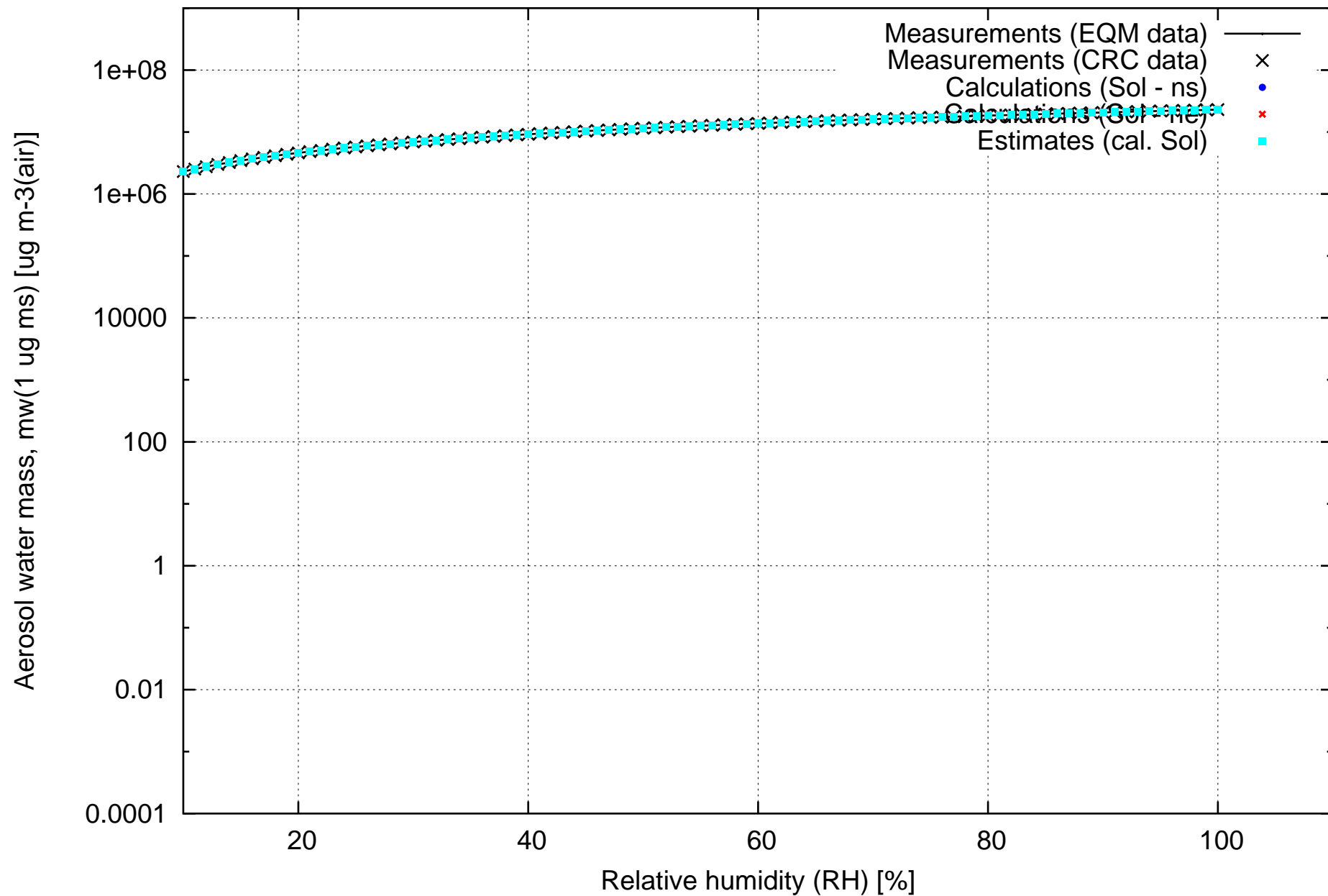
Carbonic acid - H2CO3



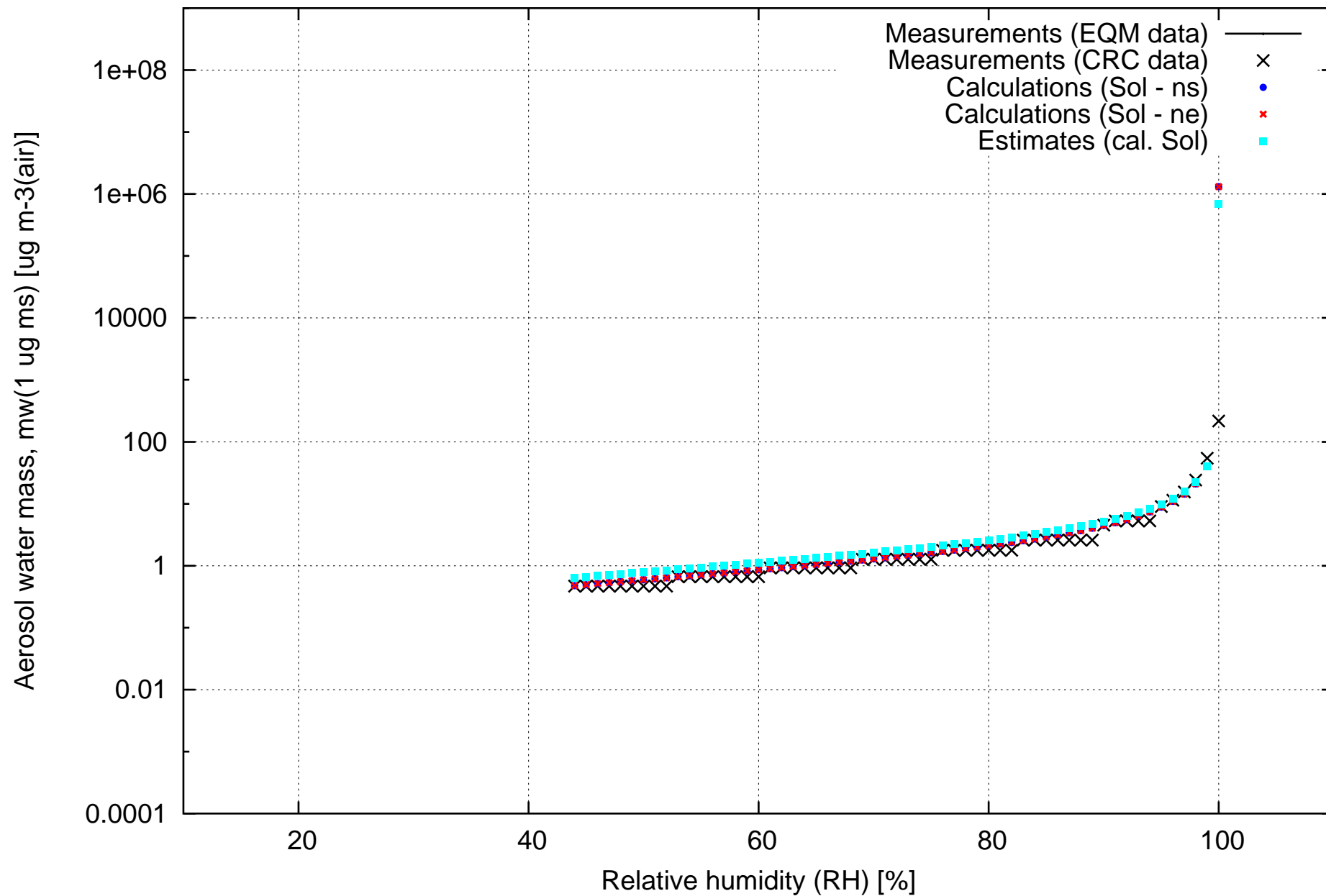
Hydrogen - dummy 09



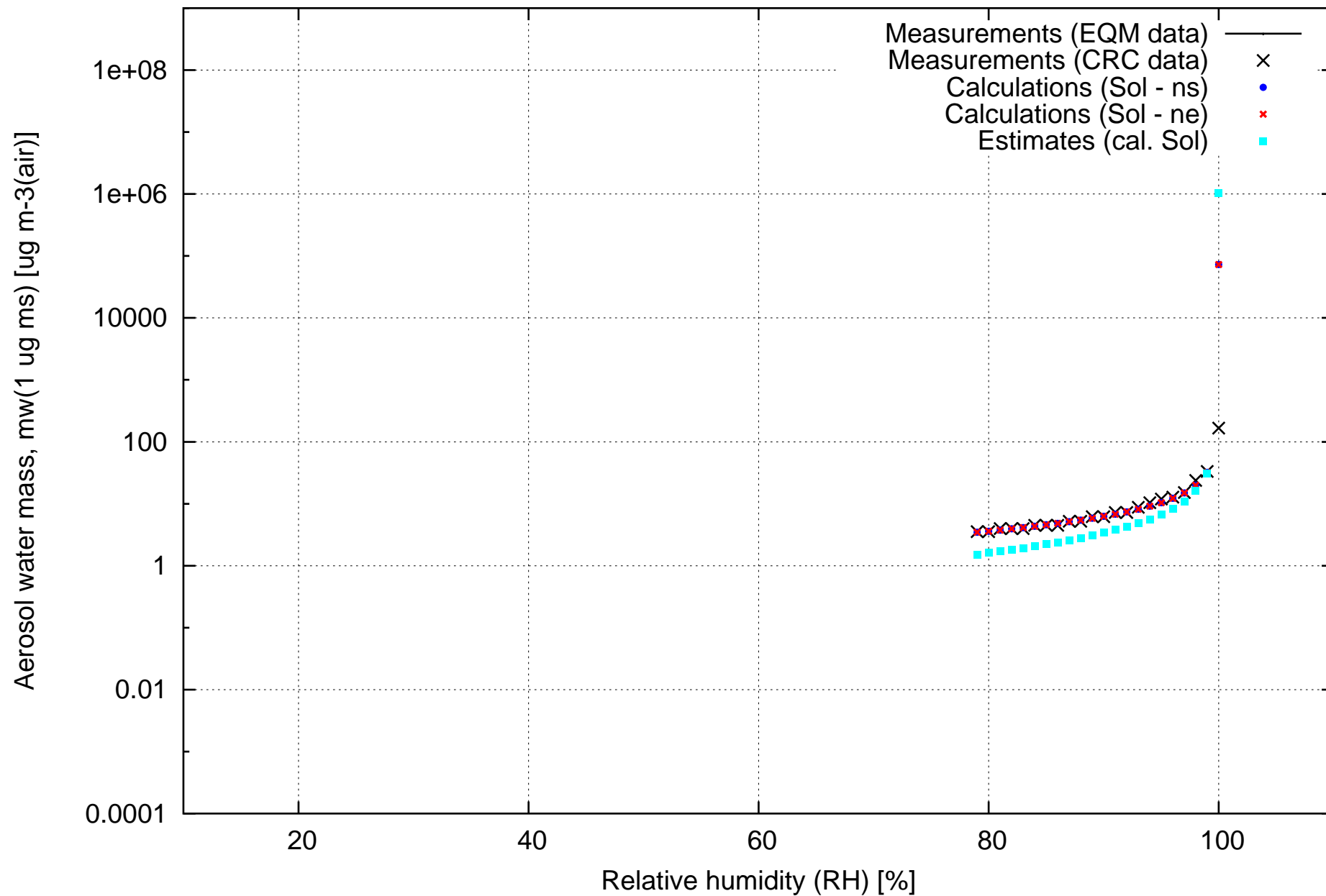
Water - H2O



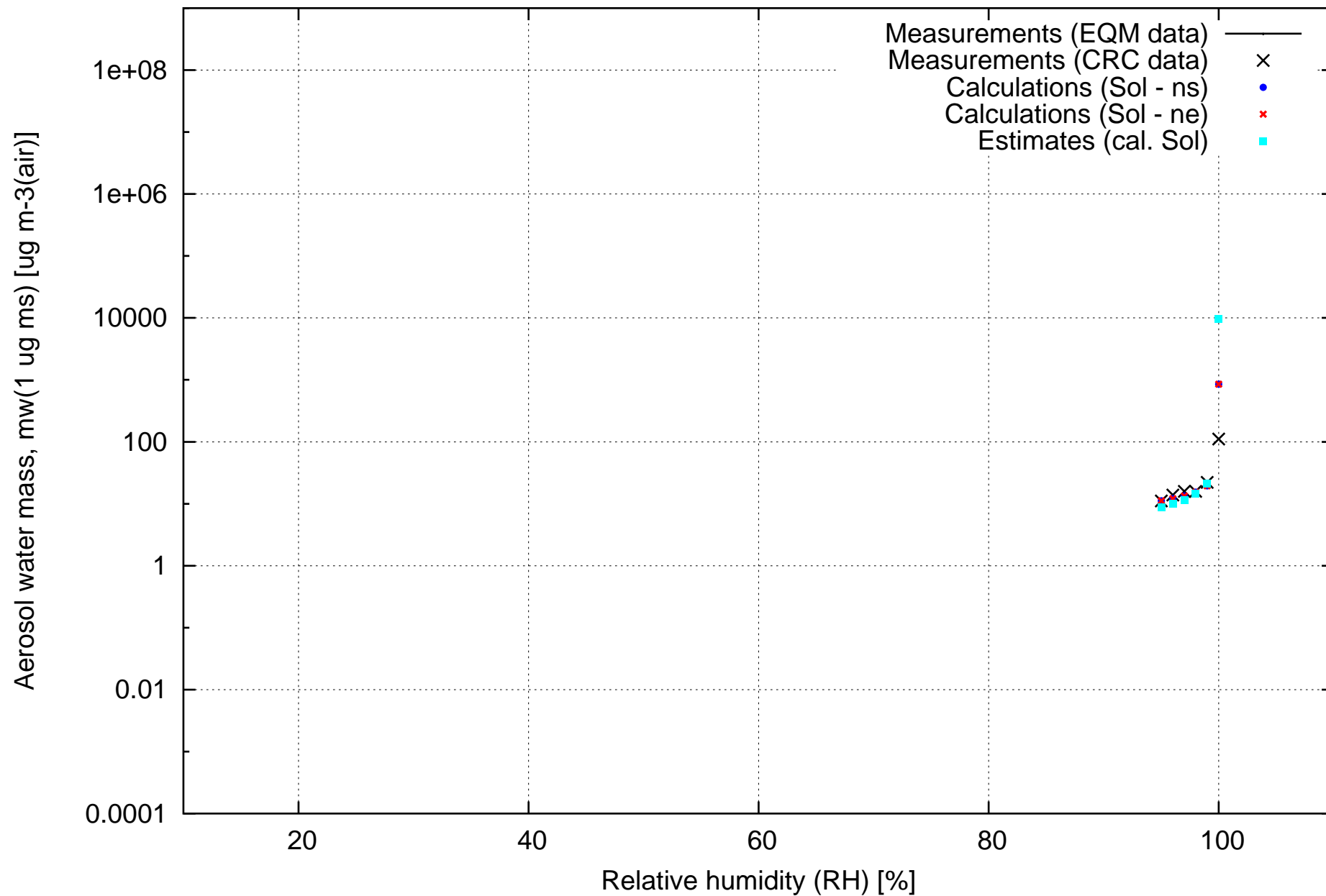
Formic acid - CH₂O₂ = H(COOH)



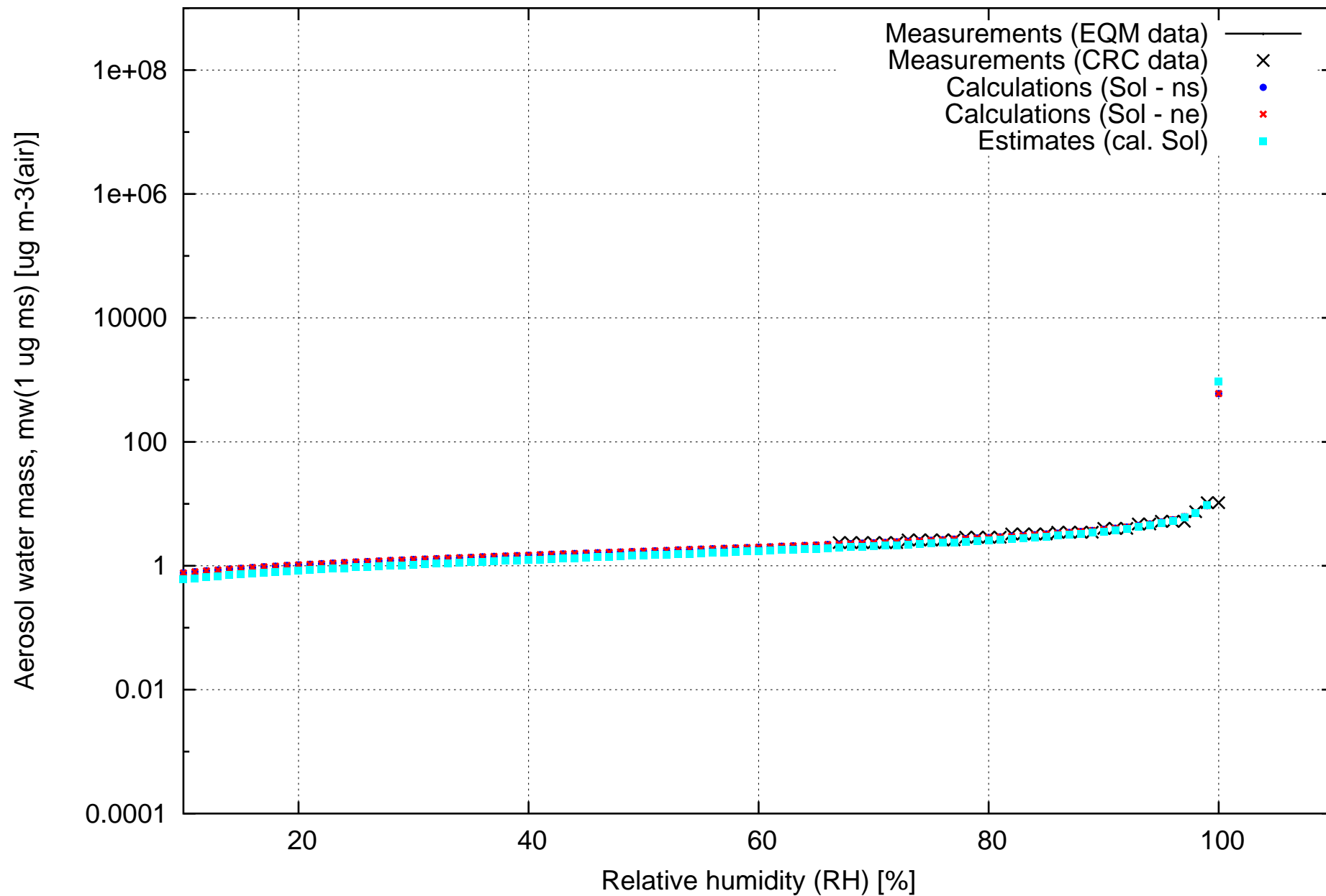
Acetic acid - C₂H₄O₂ = CH₃(COOH)



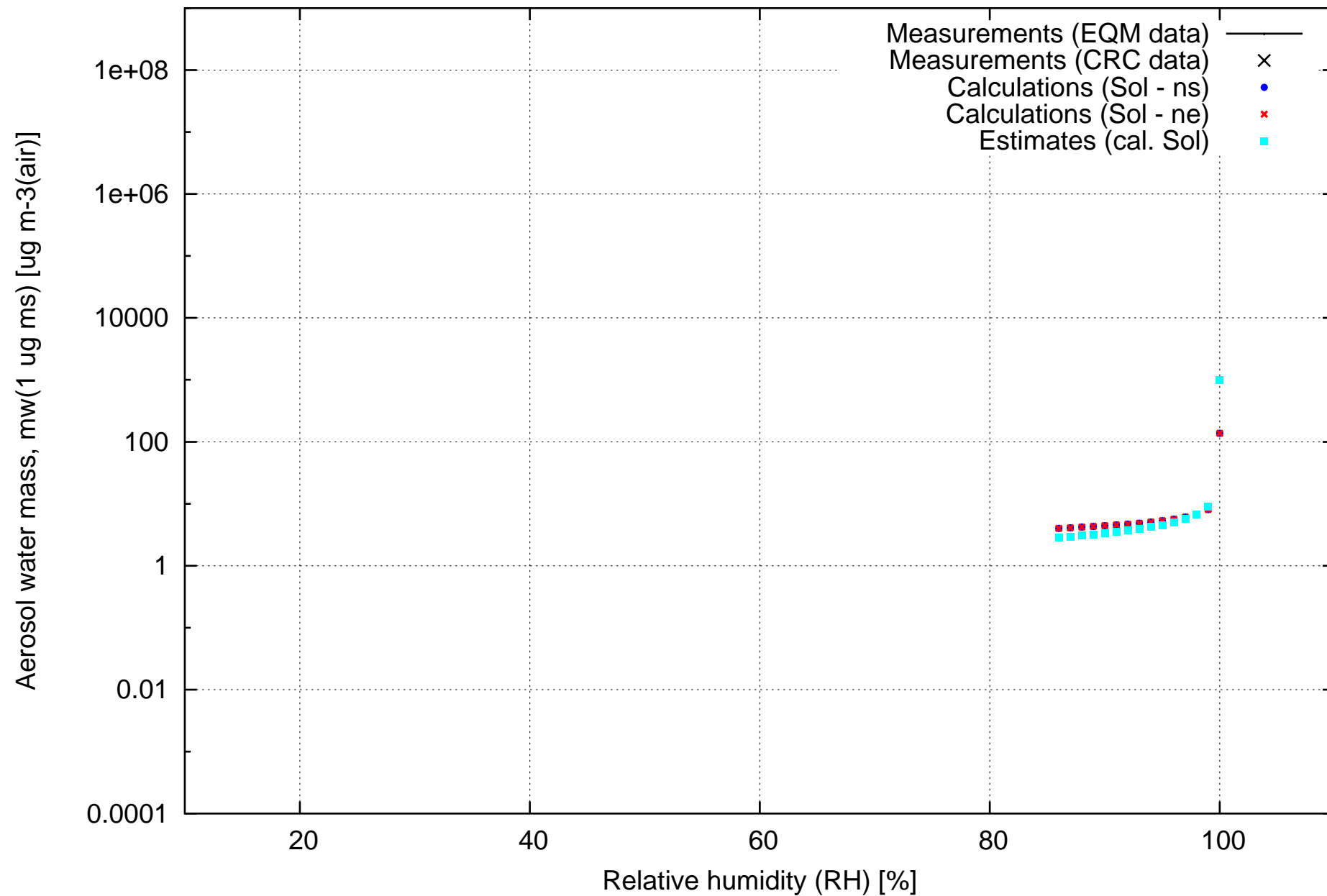
Oxalic acid - $\text{C}_2\text{H}_2\text{O}_4 = (\text{COOH})_2$



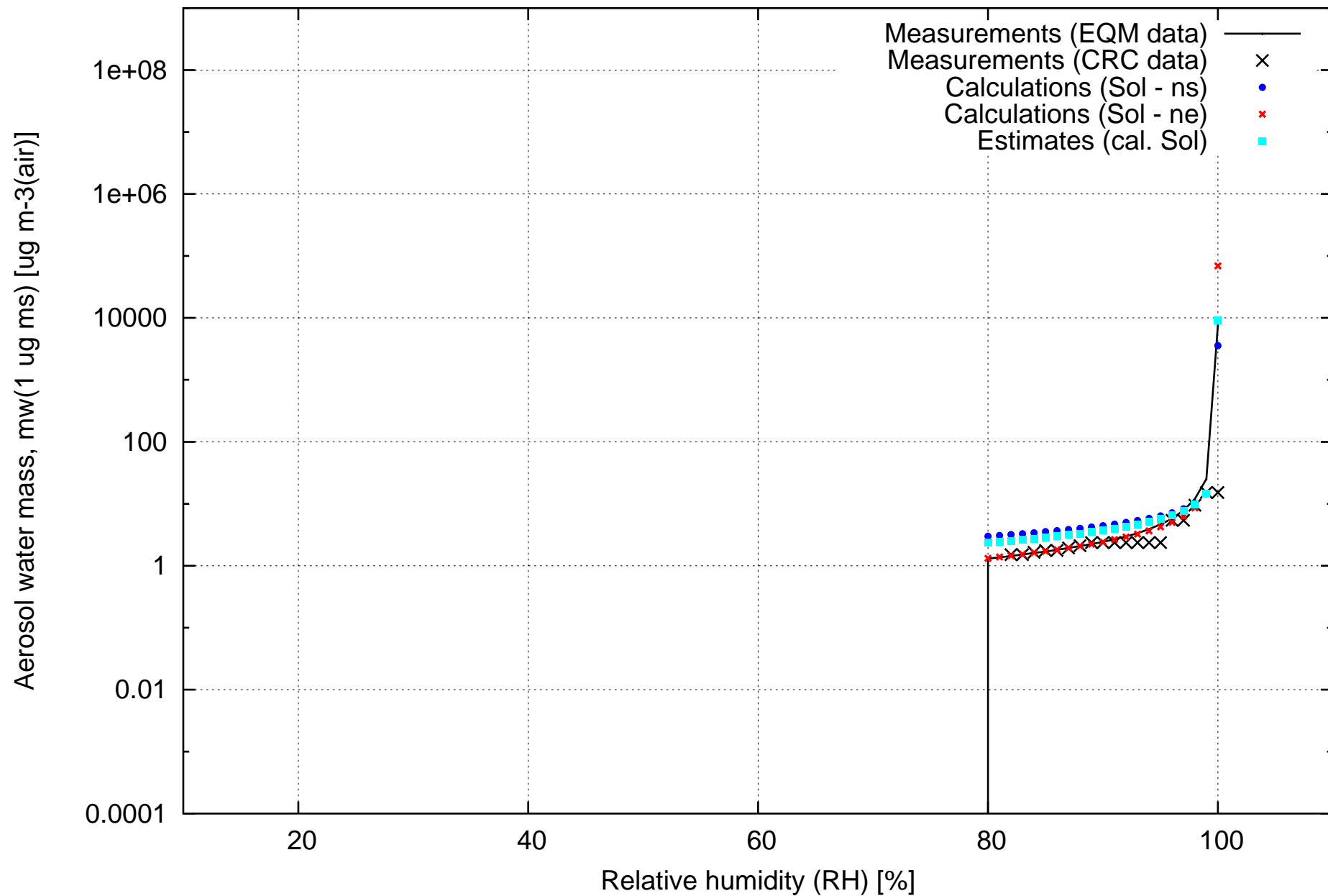
Citric acid - C₆H₈O₇ = (HO)C(COOH)₃



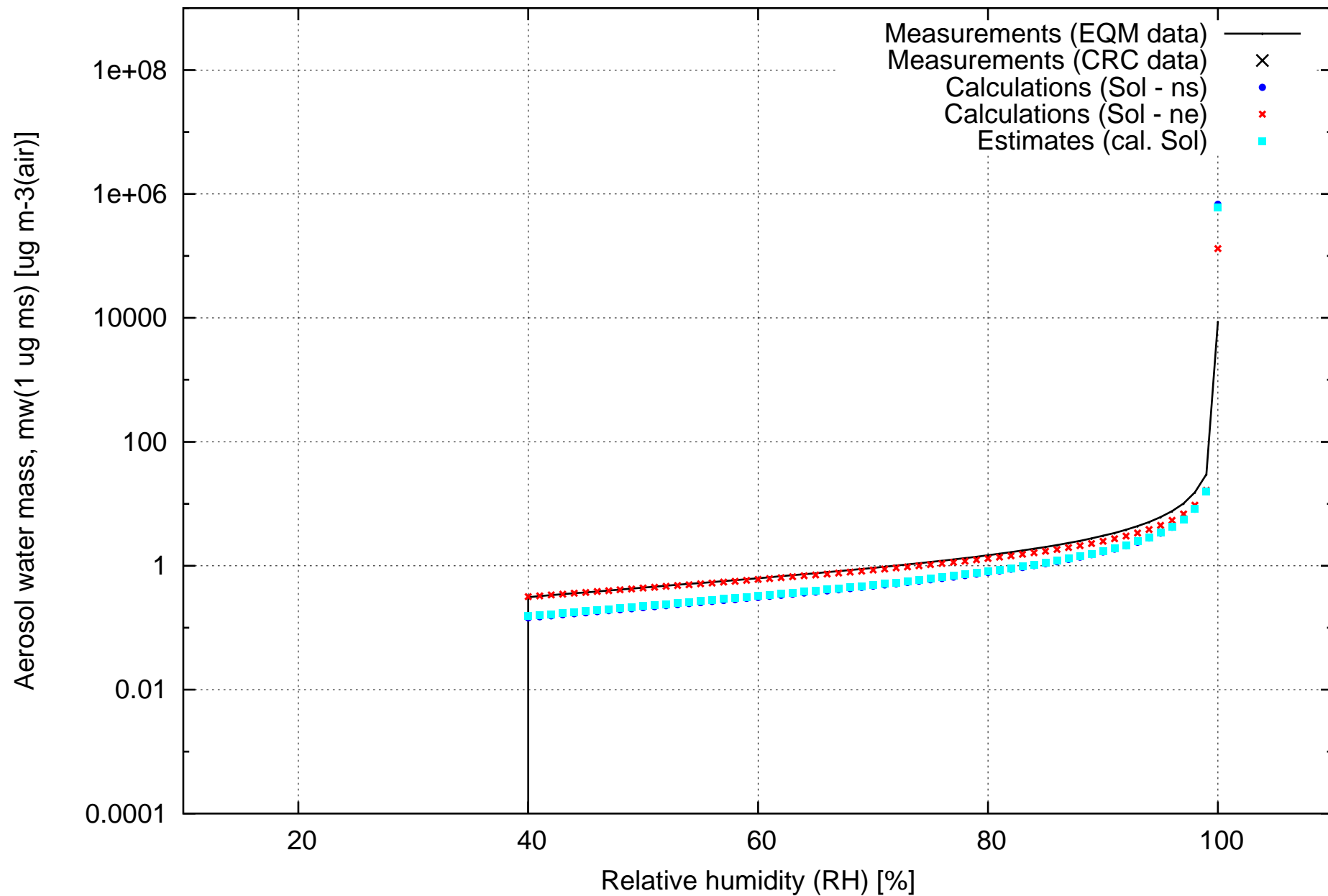
Ammonium phosphate - (NH₄)₃PO₄



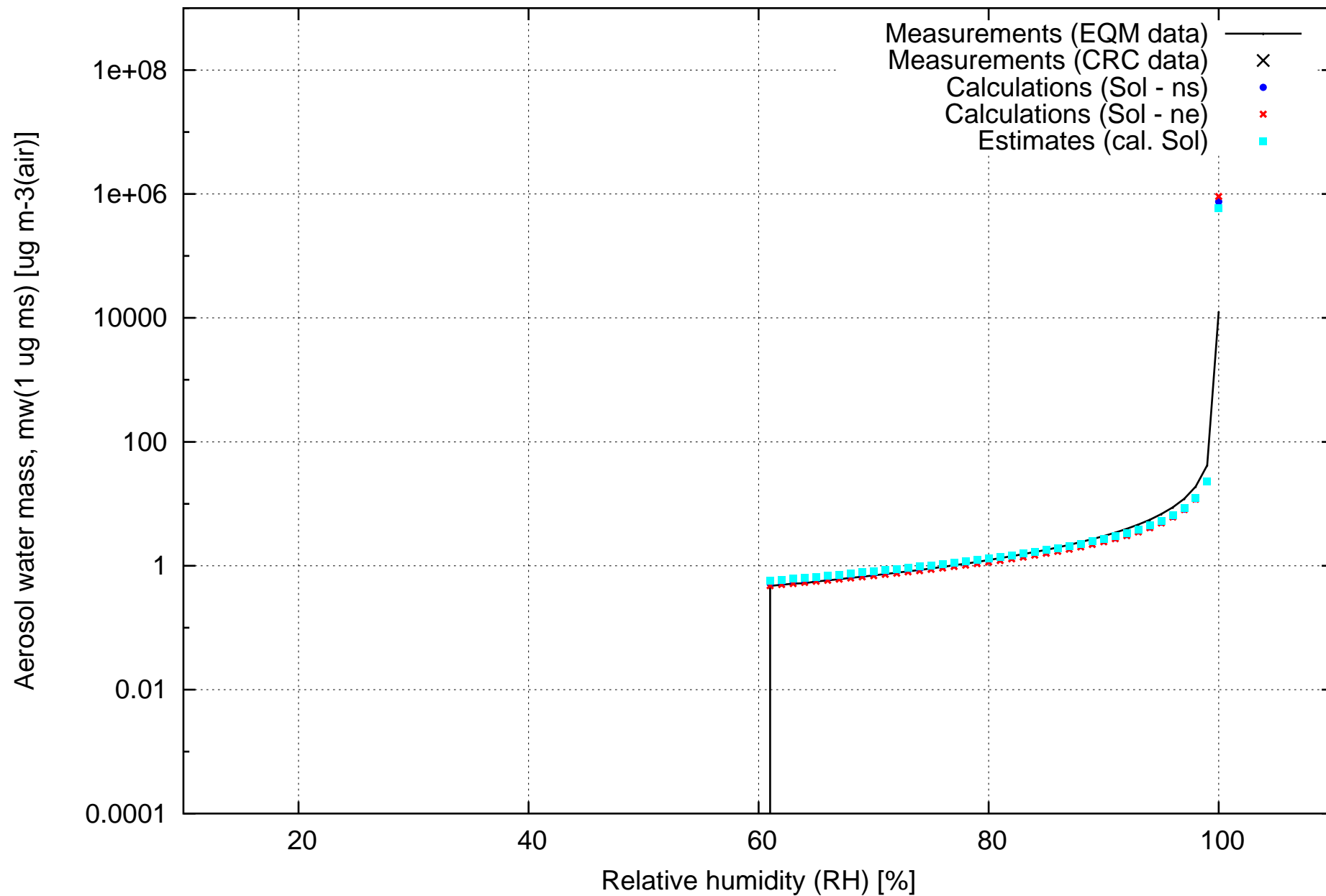
Ammonium sulfate - (NH₄)₂SO₄



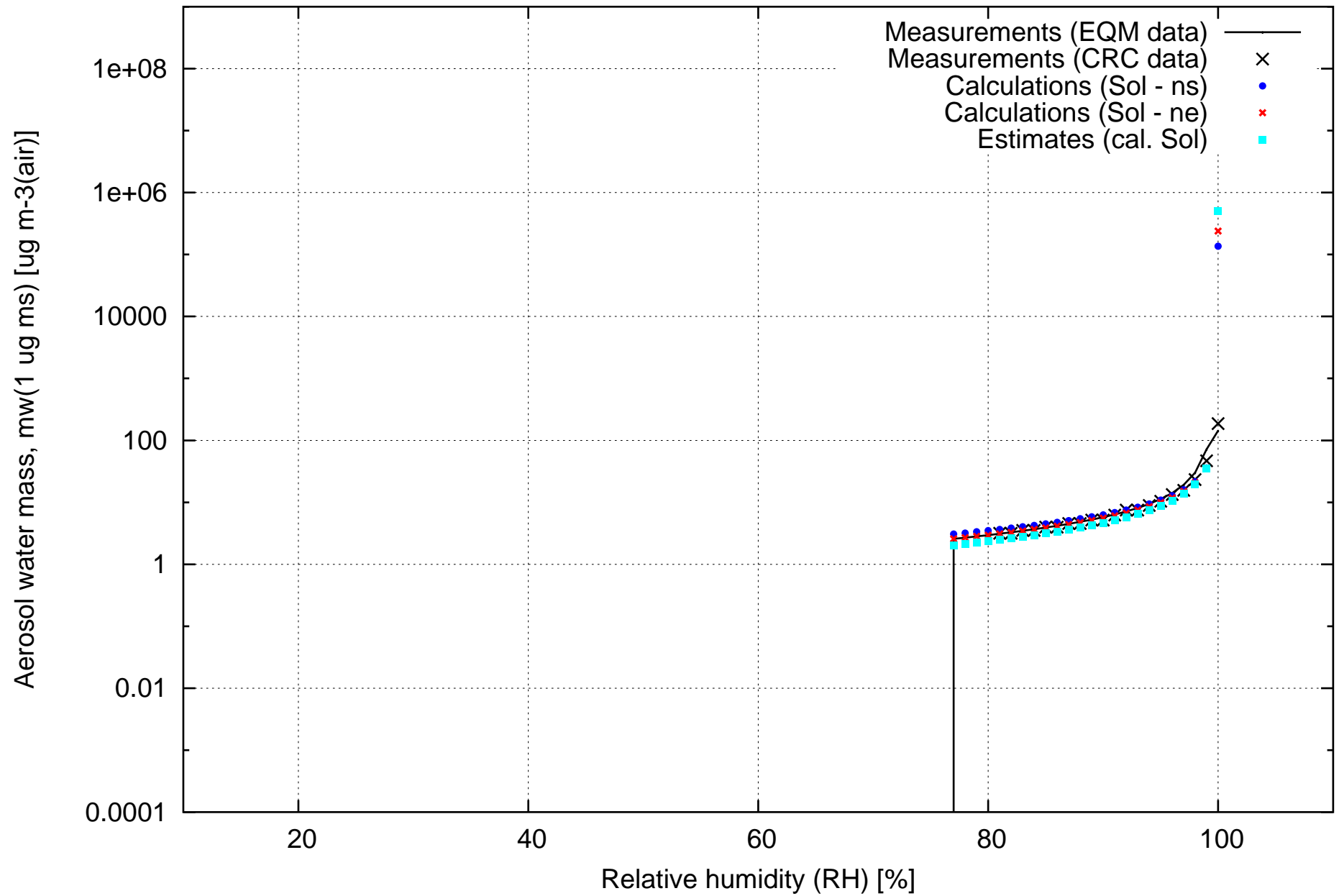
Ammonium hydrogen sulfate - NH_4HSO_4



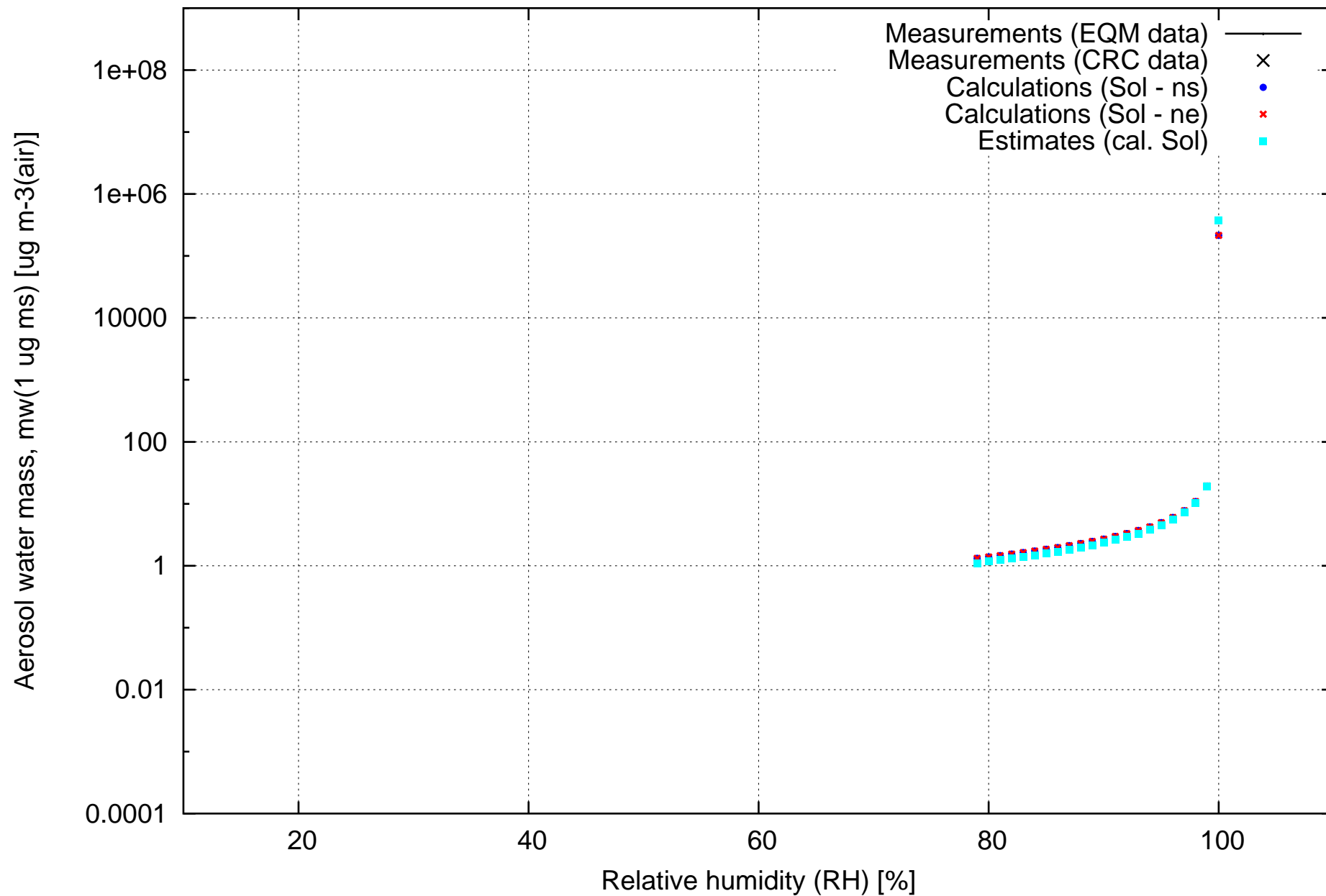
Ammonium nitrate - NH_4NO_3



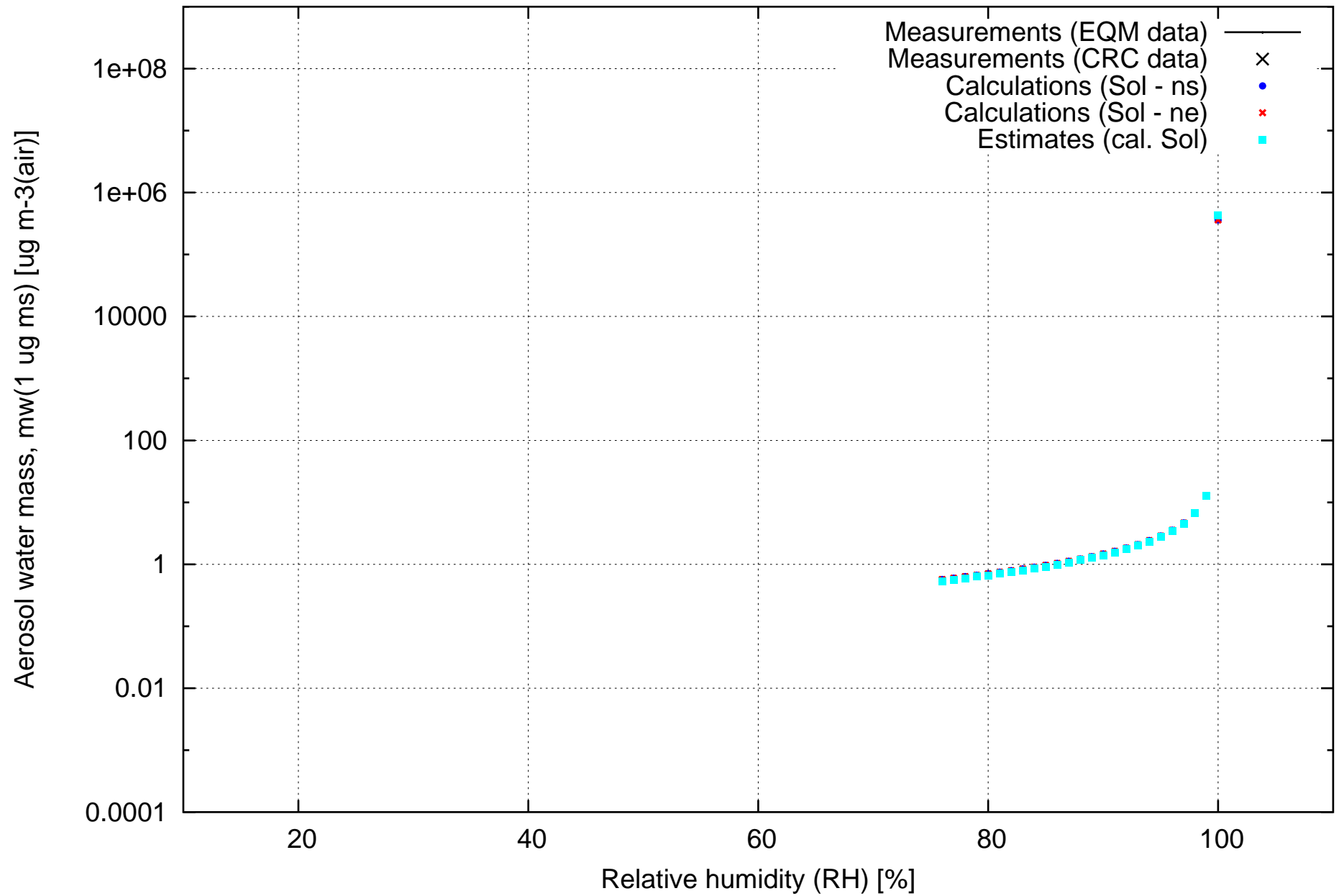
Ammonium chloride - NH₄Cl



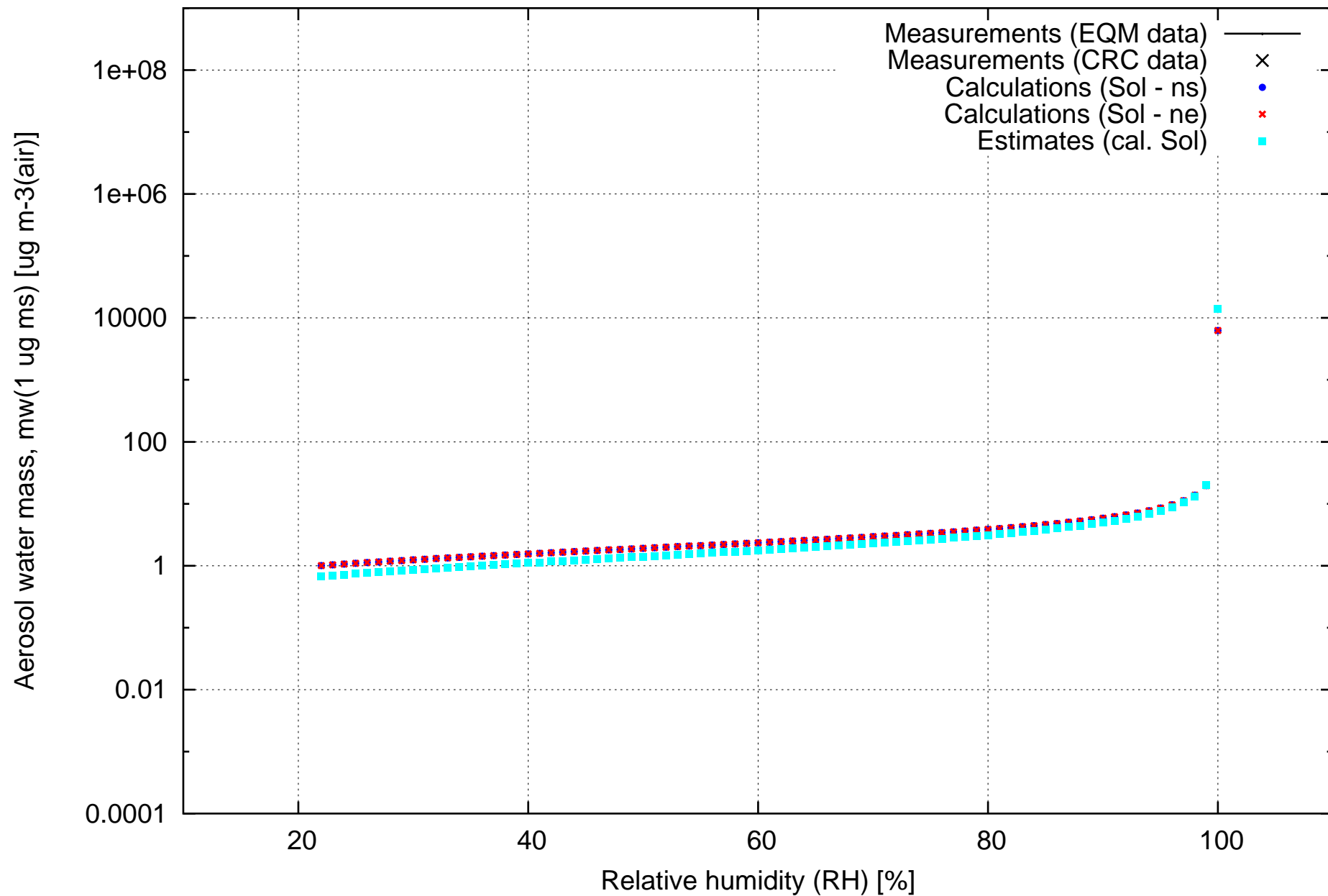
Ammonium bromide - NH₄Br



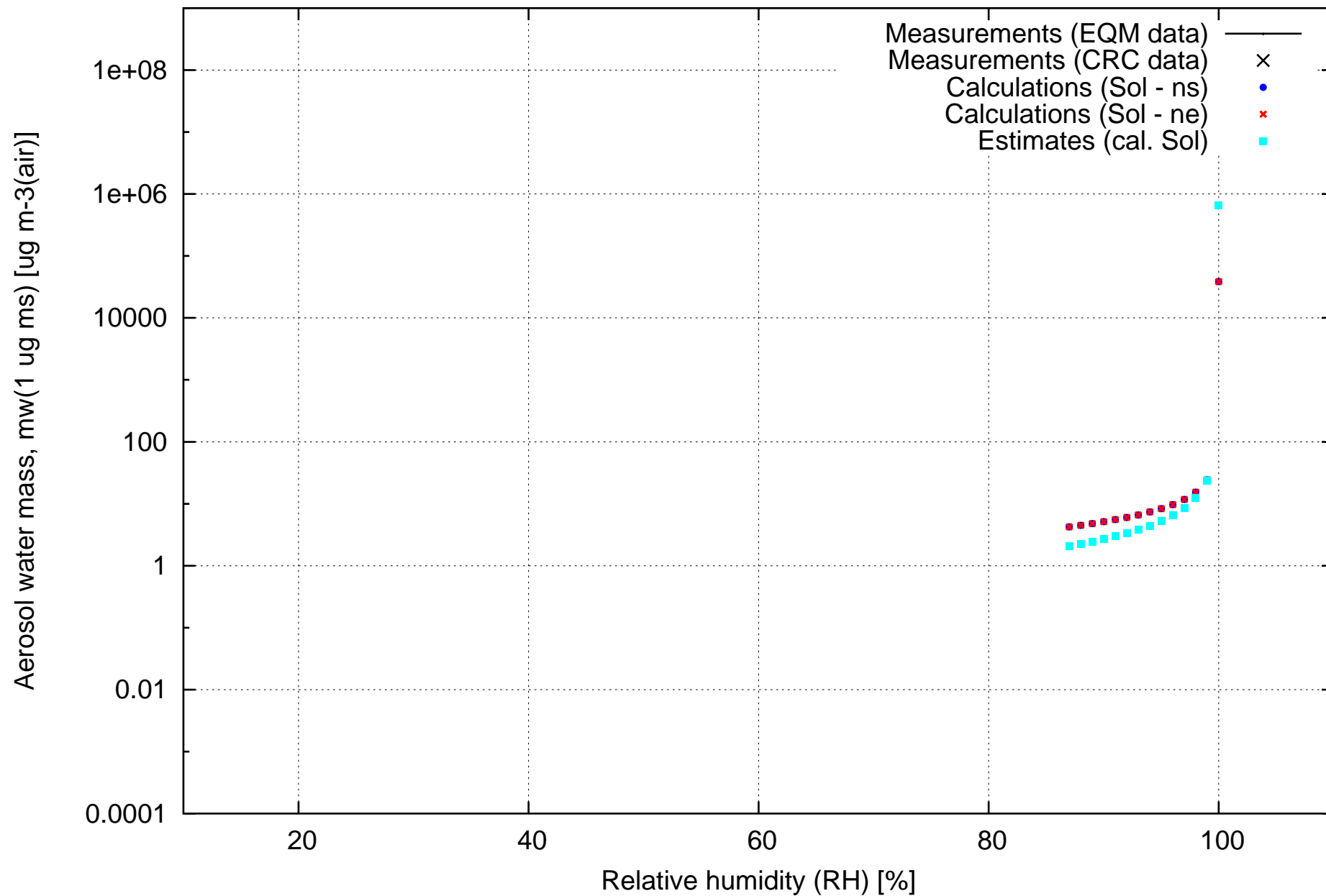
Ammonium iodide - NH₄I



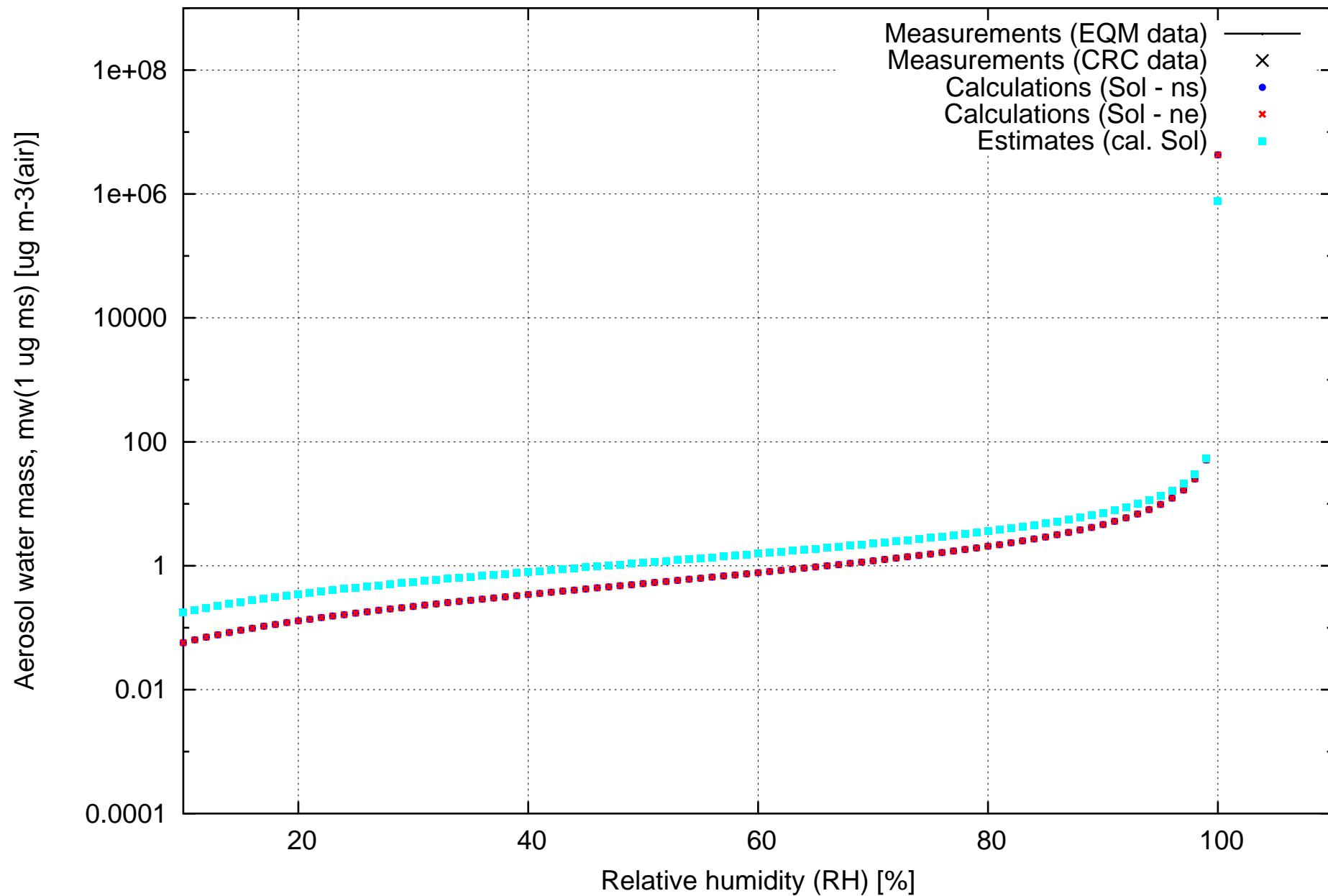
Ammonium carbonate - (NH₄)₂CO₃



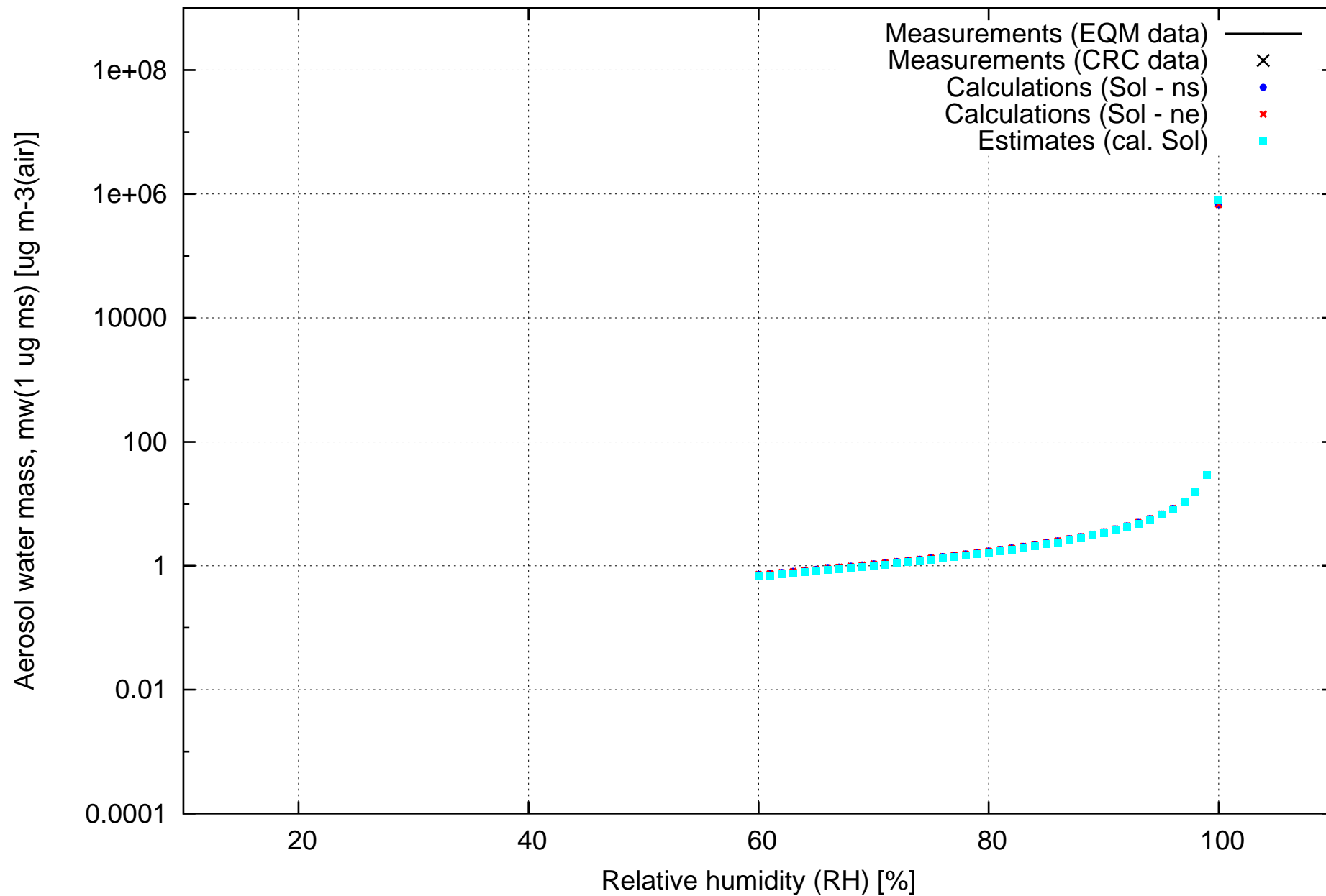
Ammonium hydrogen carbonate- NH_4HCO_3



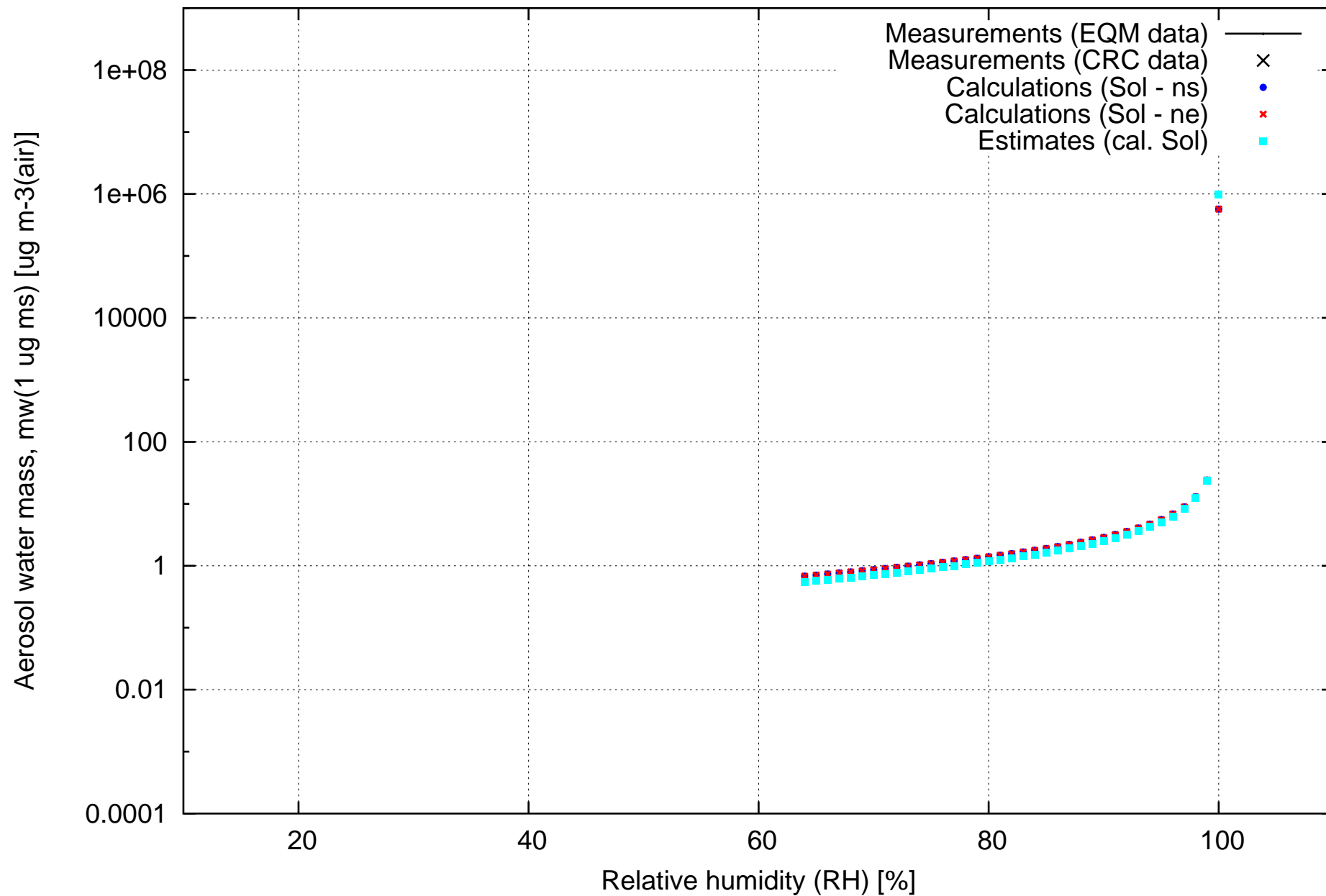
Ammonium hydroxide - NH₄OH



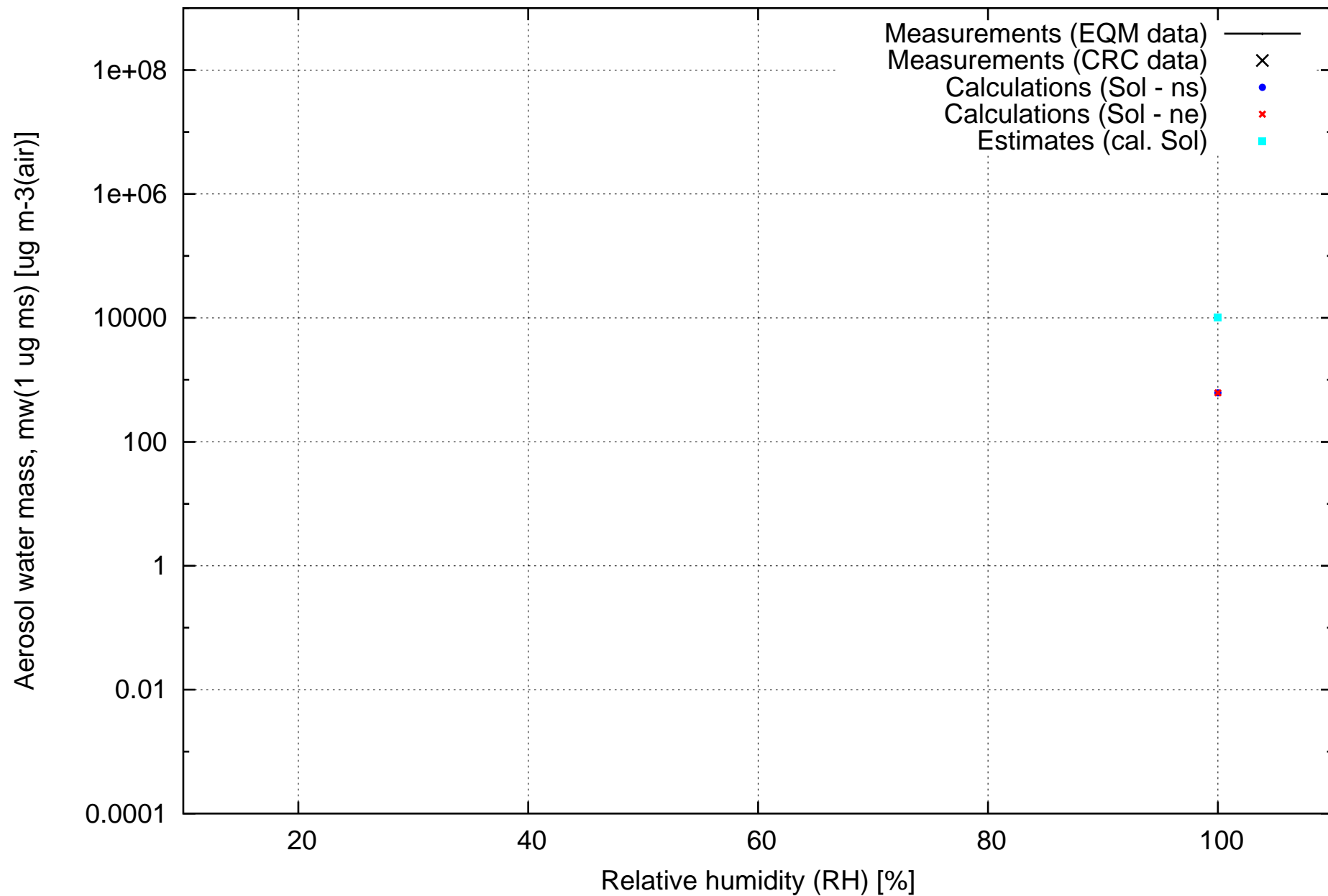
Ammonium formate - $\text{NH}_4\text{CHO}_2 = \text{H}(\text{COONH}_4)$



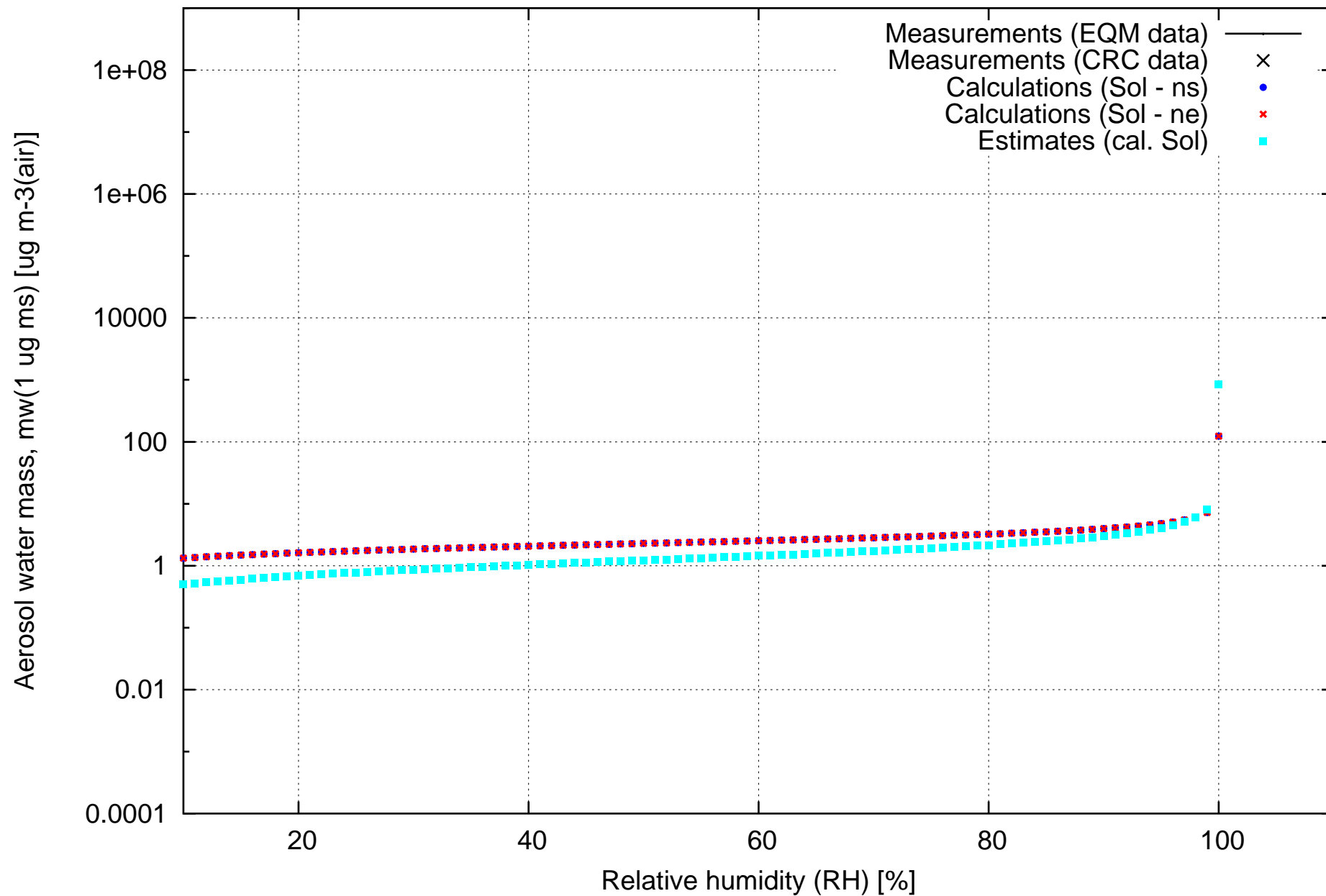
Ammonium acetate - $\text{NH}_4\text{C}_2\text{H}_3\text{O}_2 = \text{CH}_3(\text{COONH}_4)$



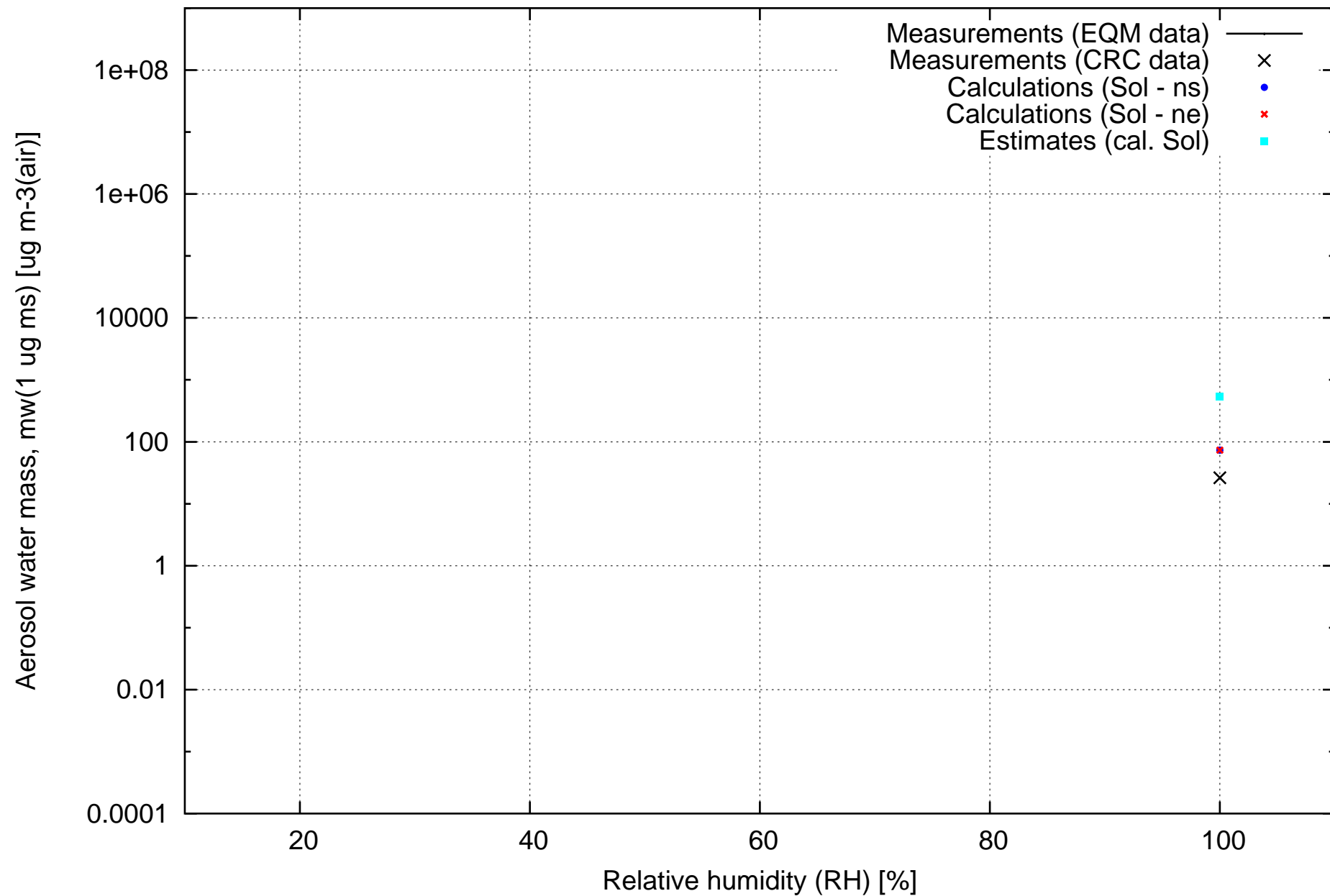
Ammonium oxalate - $(\text{NH}_4)_2\text{C}_2\text{O}_4 = (\text{COONH}_4)_2$



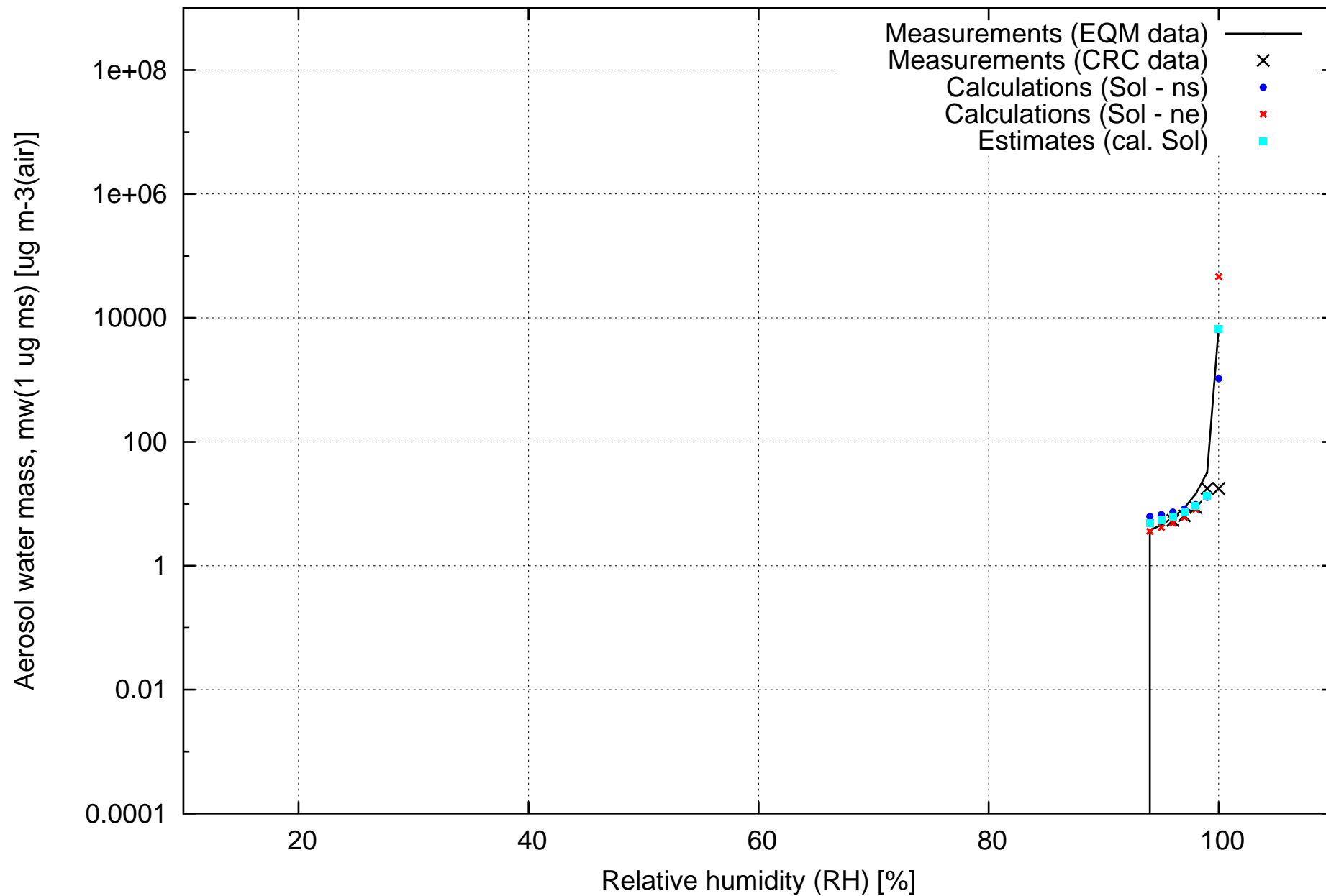
Ammonium citrate - $(\text{NH}_4)_2\text{HC}_6\text{H}_5\text{O}_7 = (\text{HO})\text{C}(\text{COONH}_4)_3$



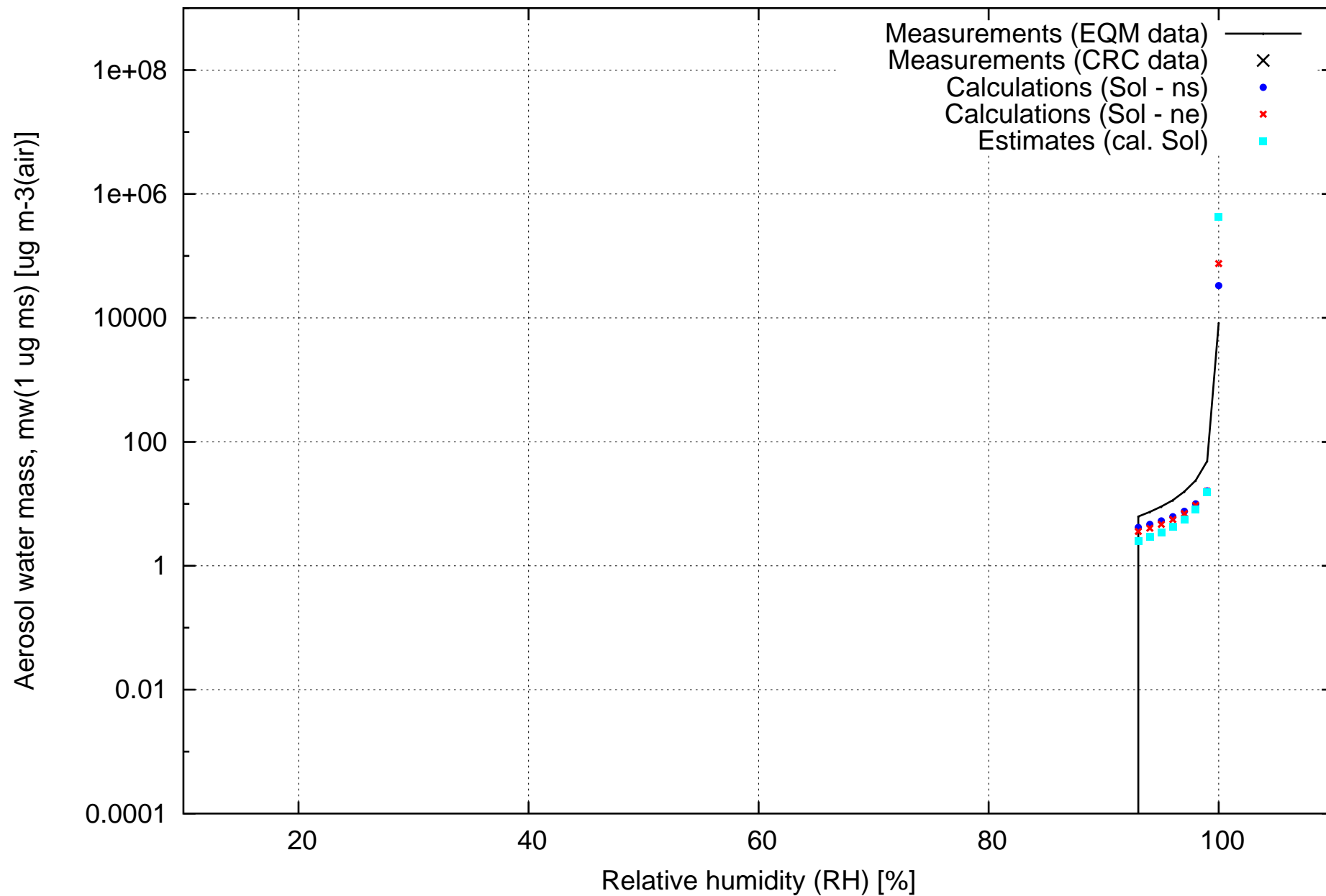
Sodium phosphate - Na₃PO₄



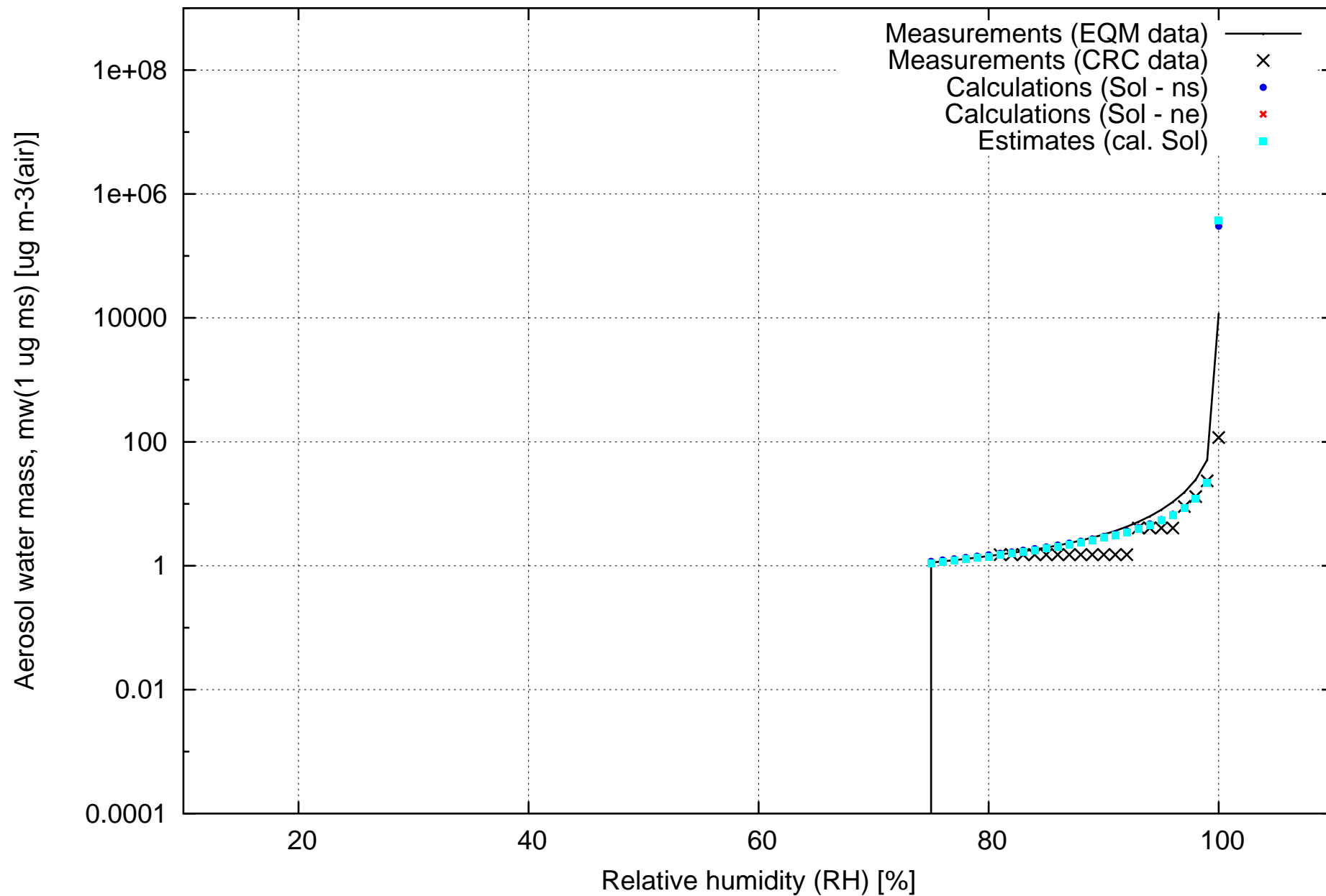
Sodium sulfate - Na₂SO₄



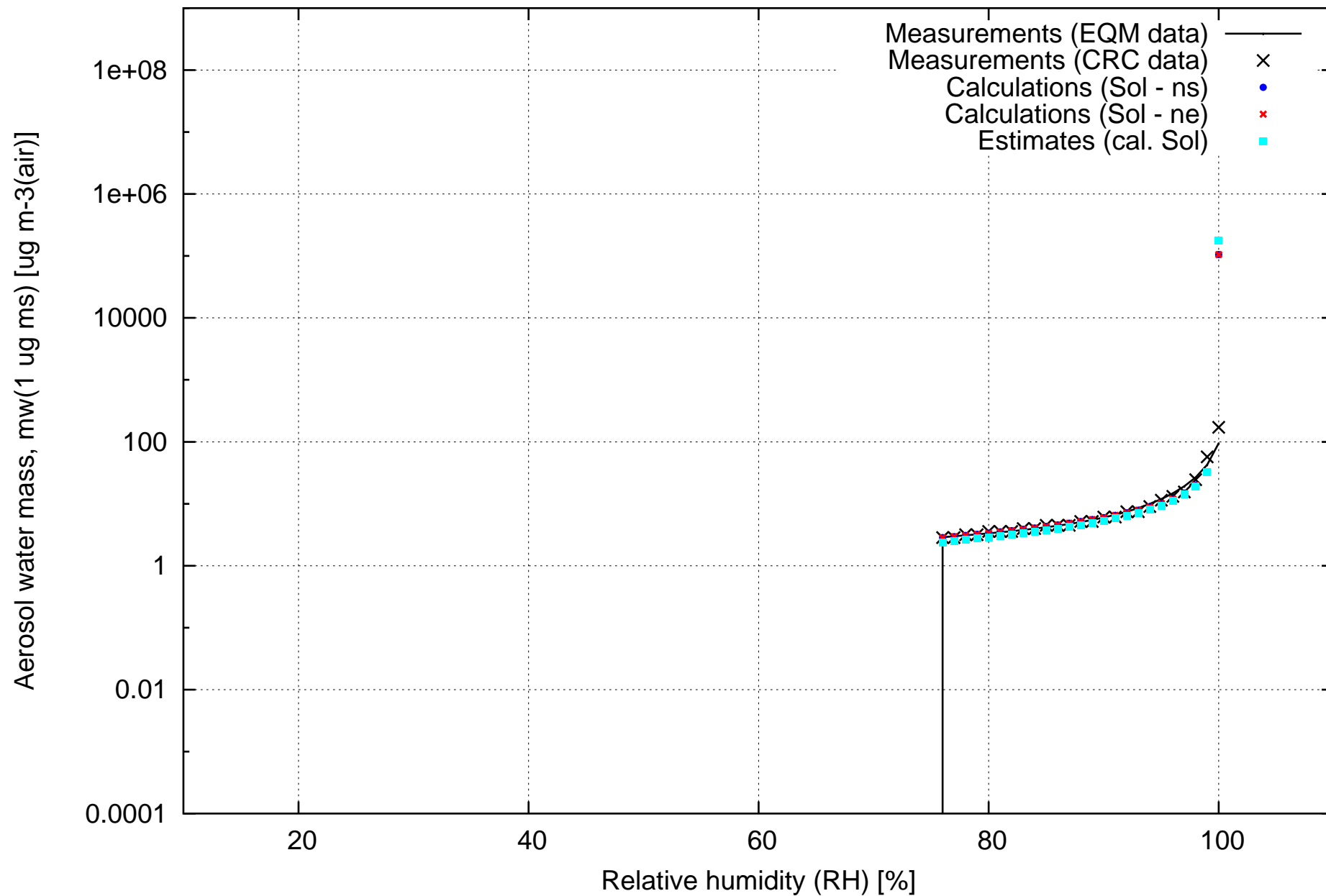
Sodium hydrogen sulfate - NaHSO₄



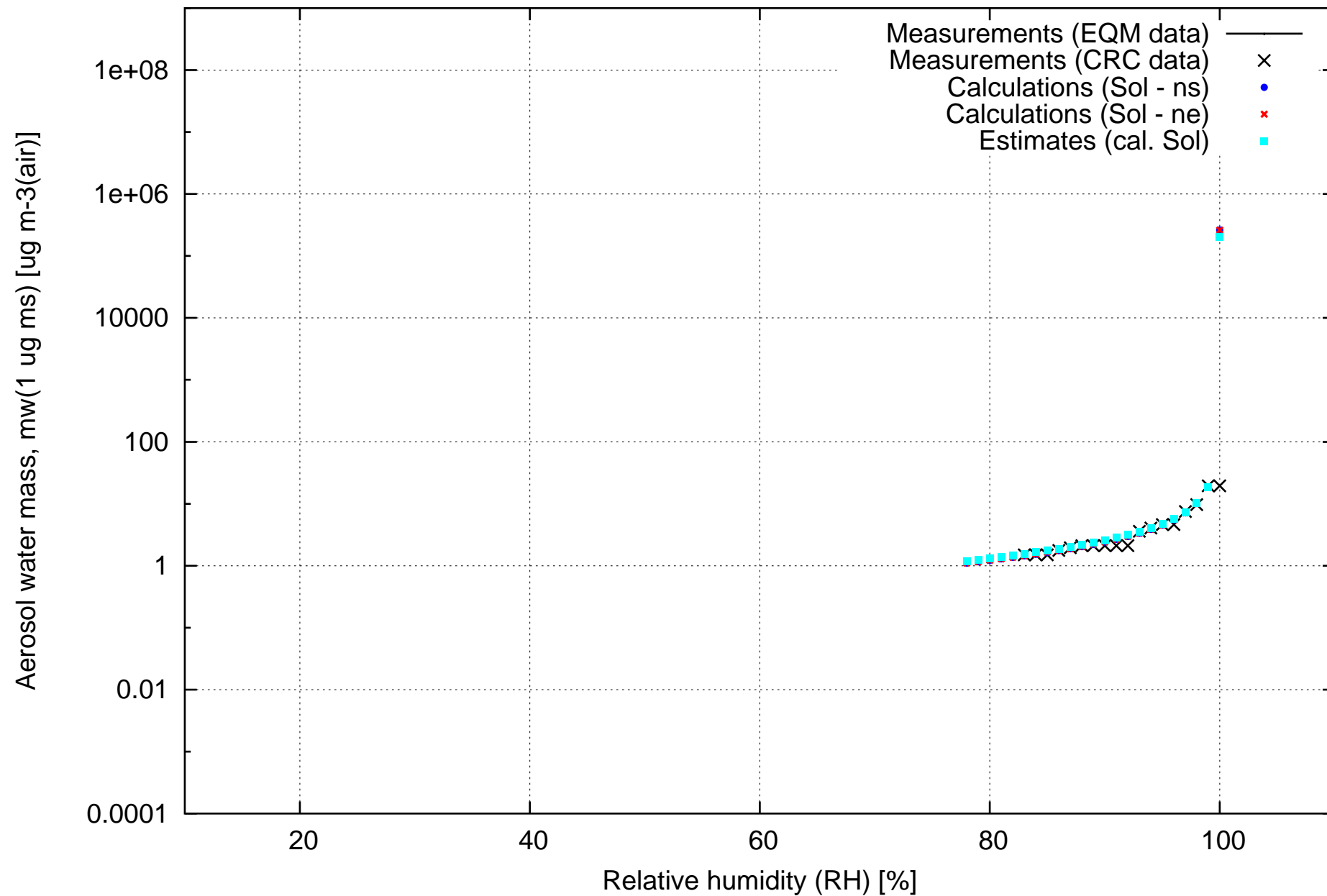
Sodium nitrate - NaNO₃



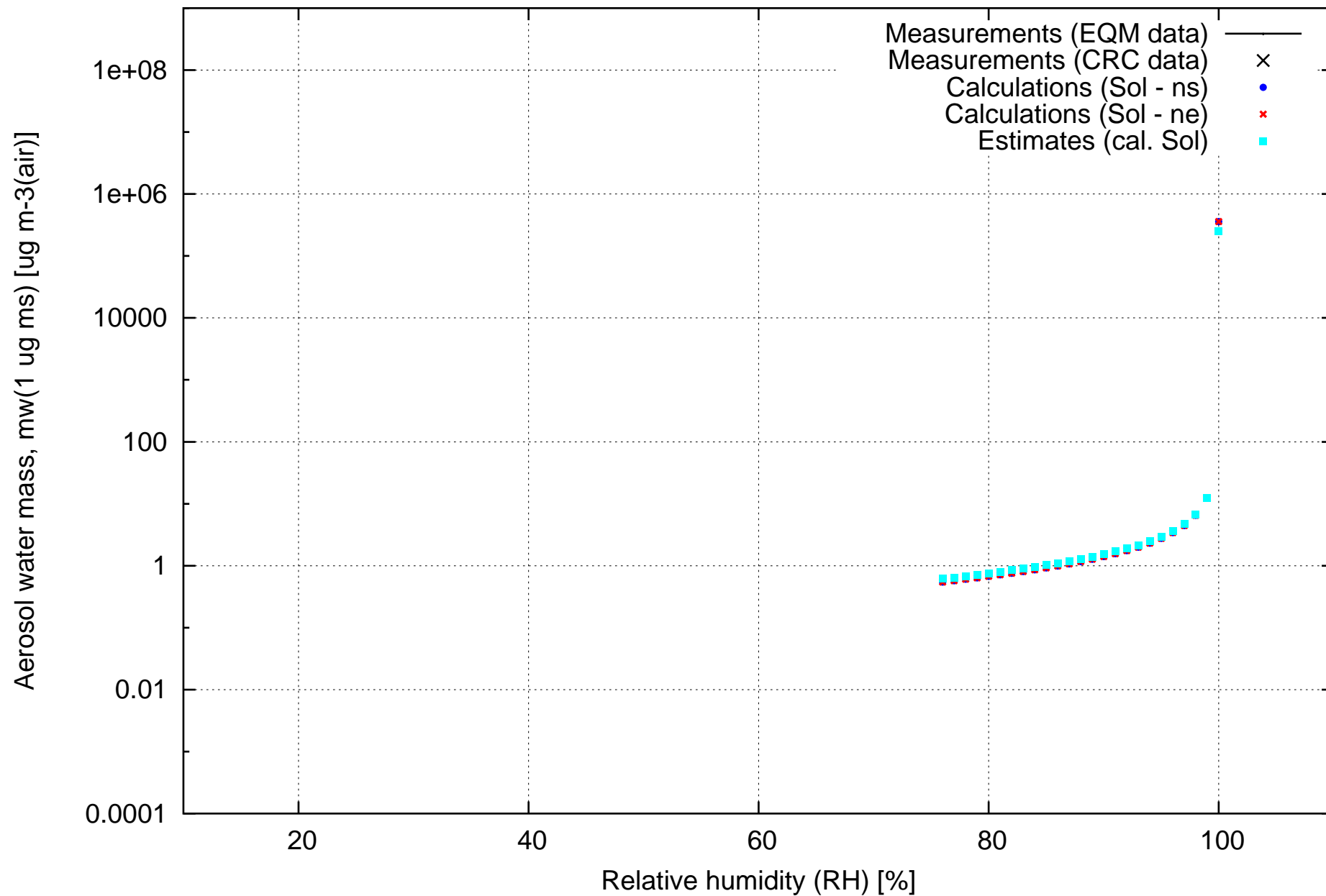
Sodium chloride - NaCl



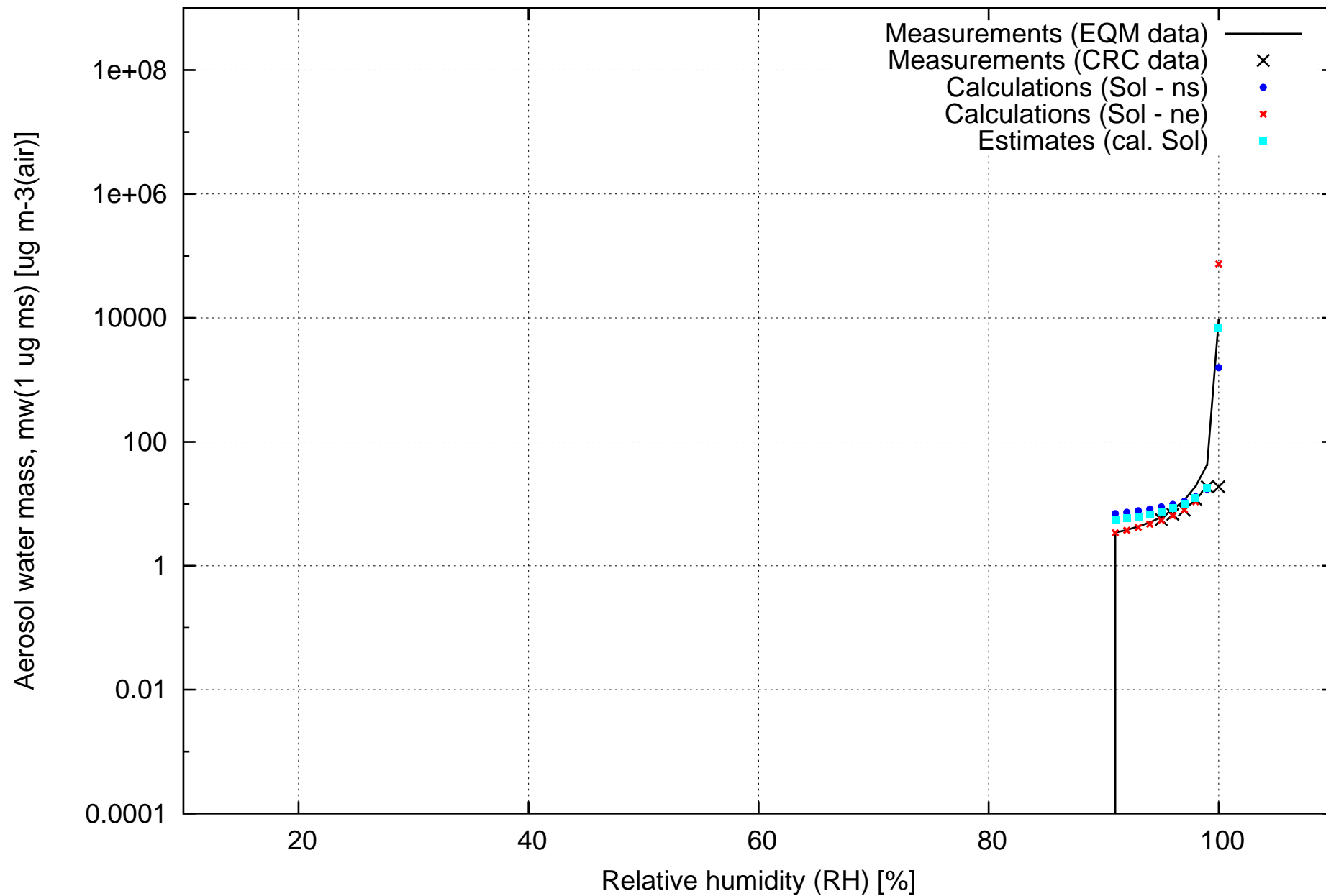
Sodium bromide - NaBr



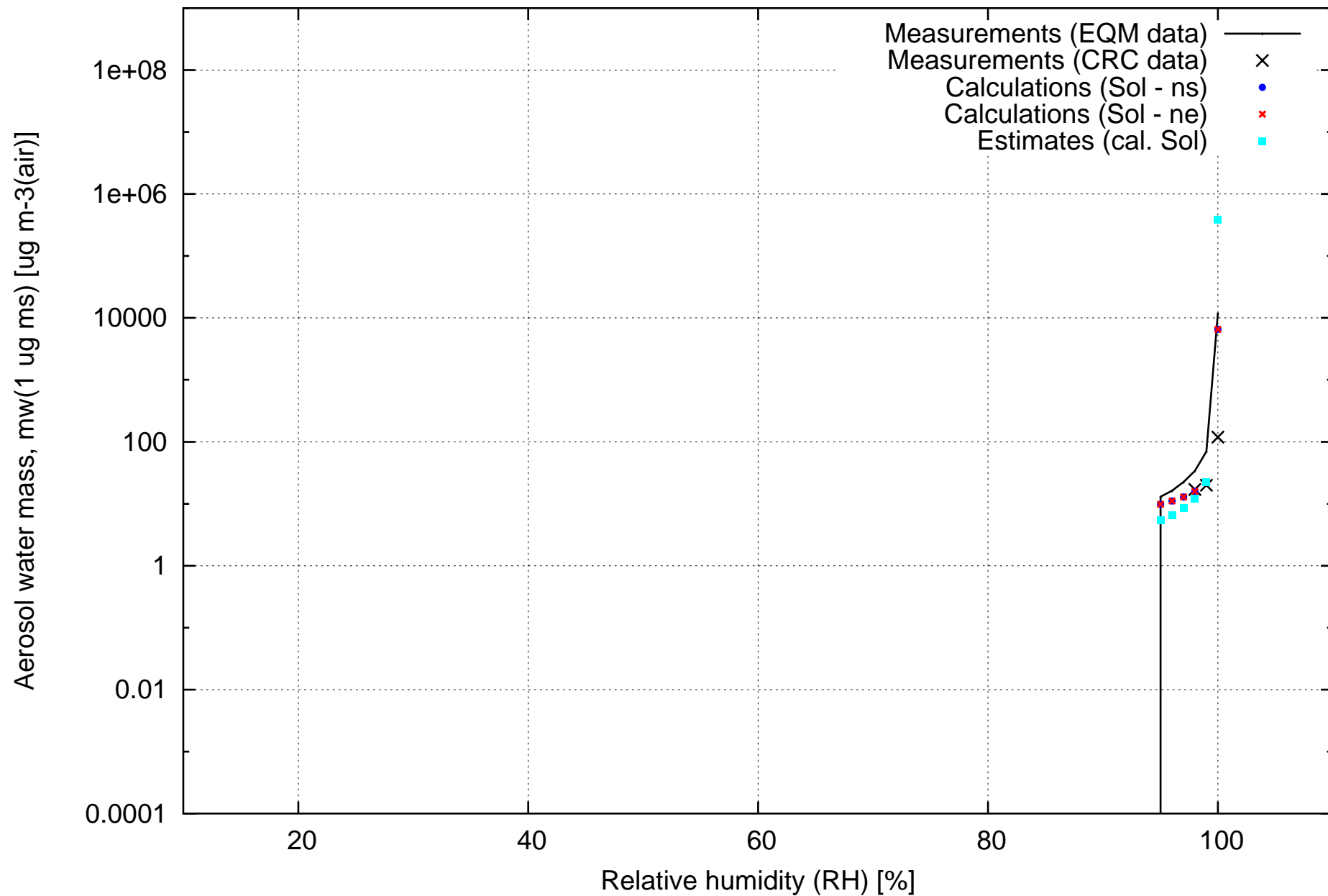
Sodium iodide - NaI



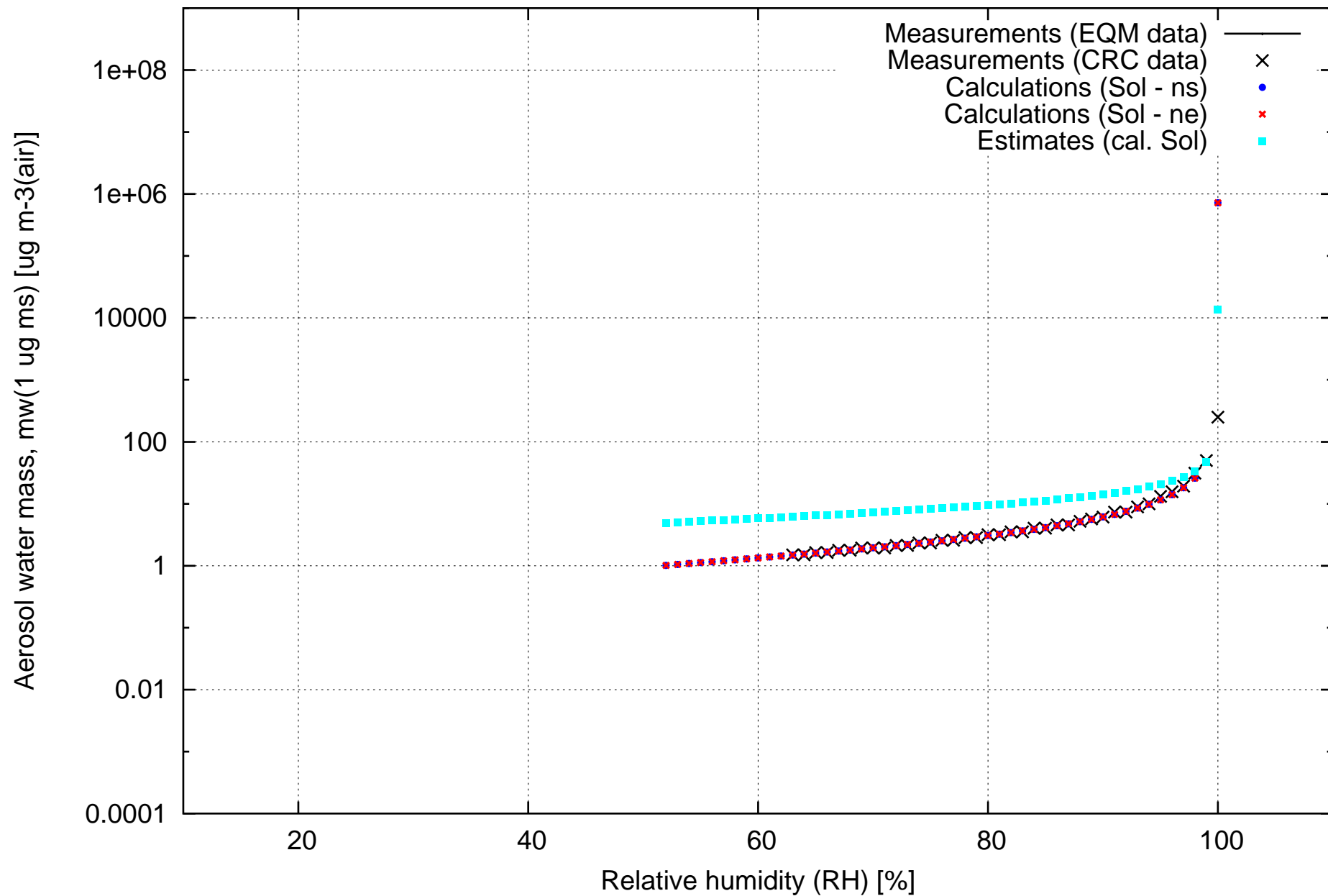
Sodium carbonate - Na₂CO₃



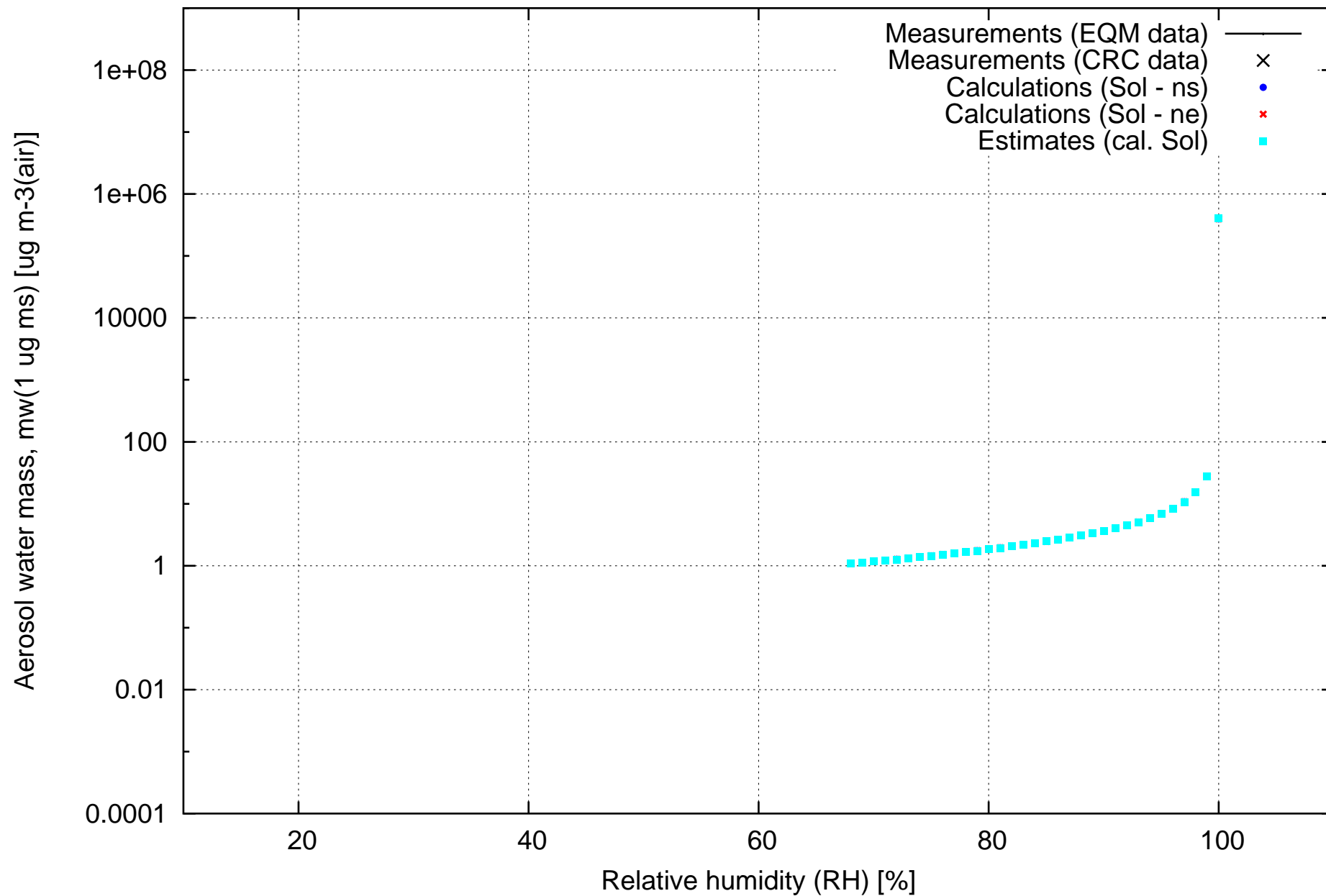
Sodium hydrogen carbonate - NaHCO₃



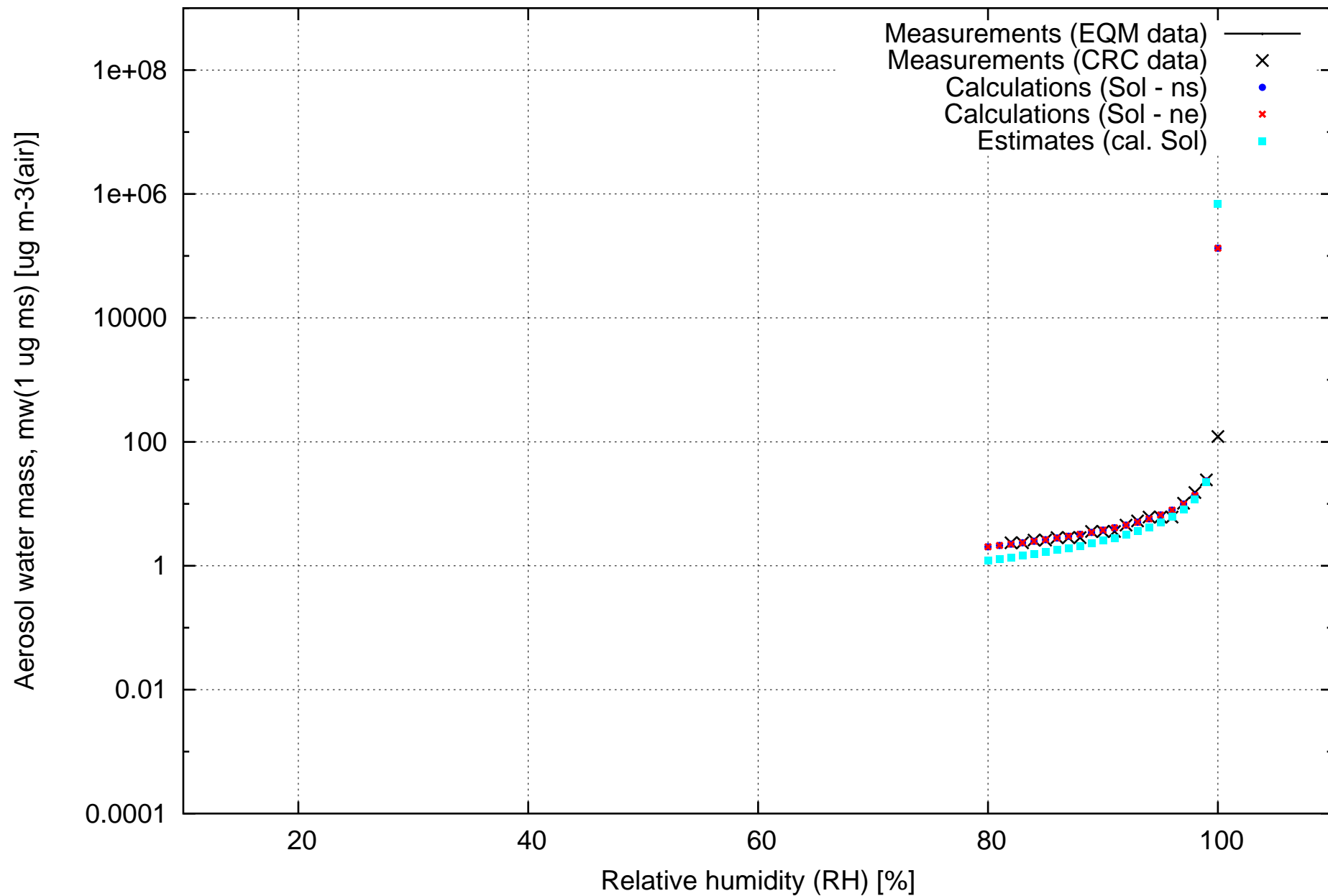
Sodium hydroxide - NaOH



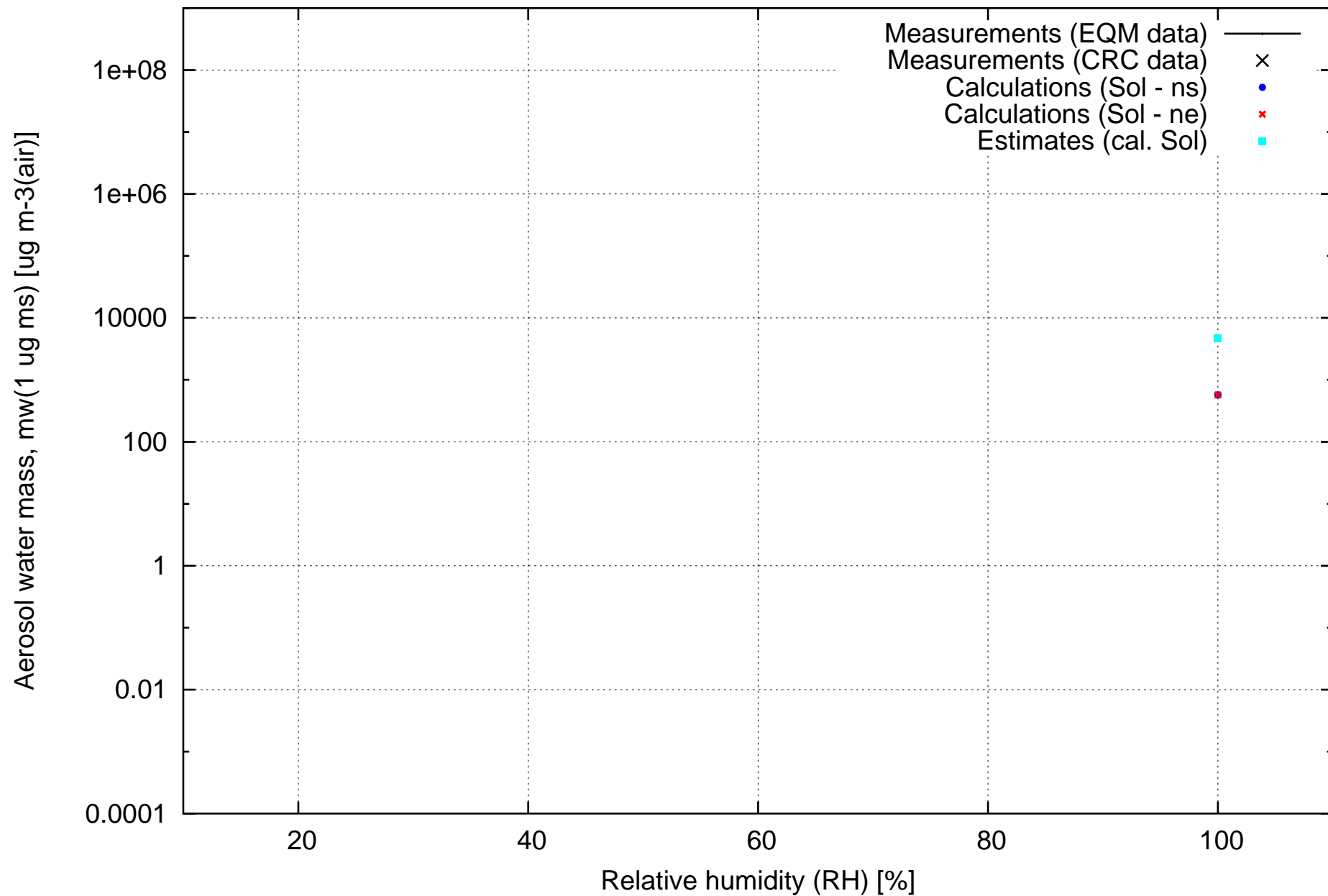
Sodium formate - $\text{NaCHO}_2 = \text{H}(\text{COONa})$



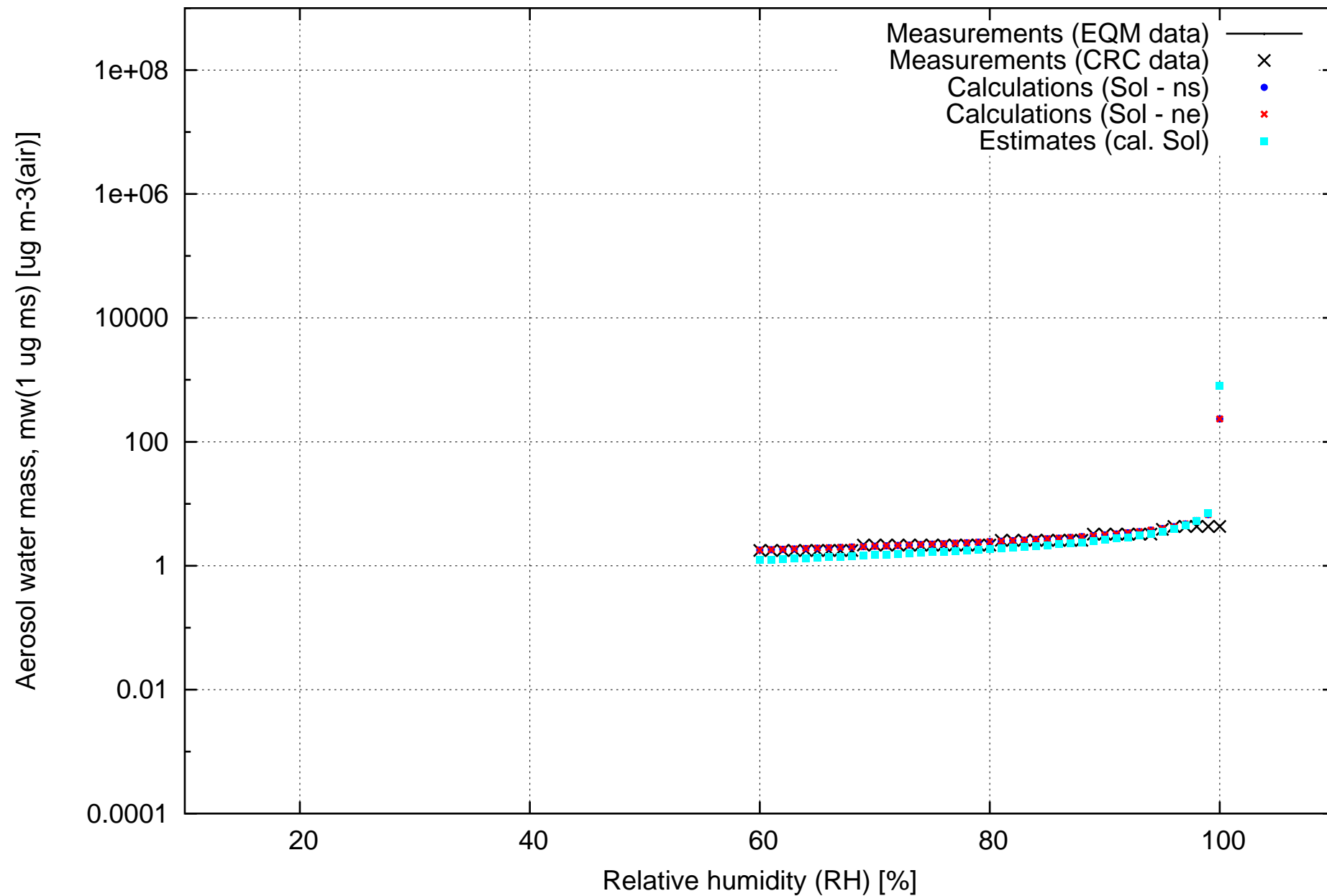
Sodium acetate - $\text{NaC}_2\text{H}_3\text{O}_2 = \text{CH}_3(\text{COONa})$



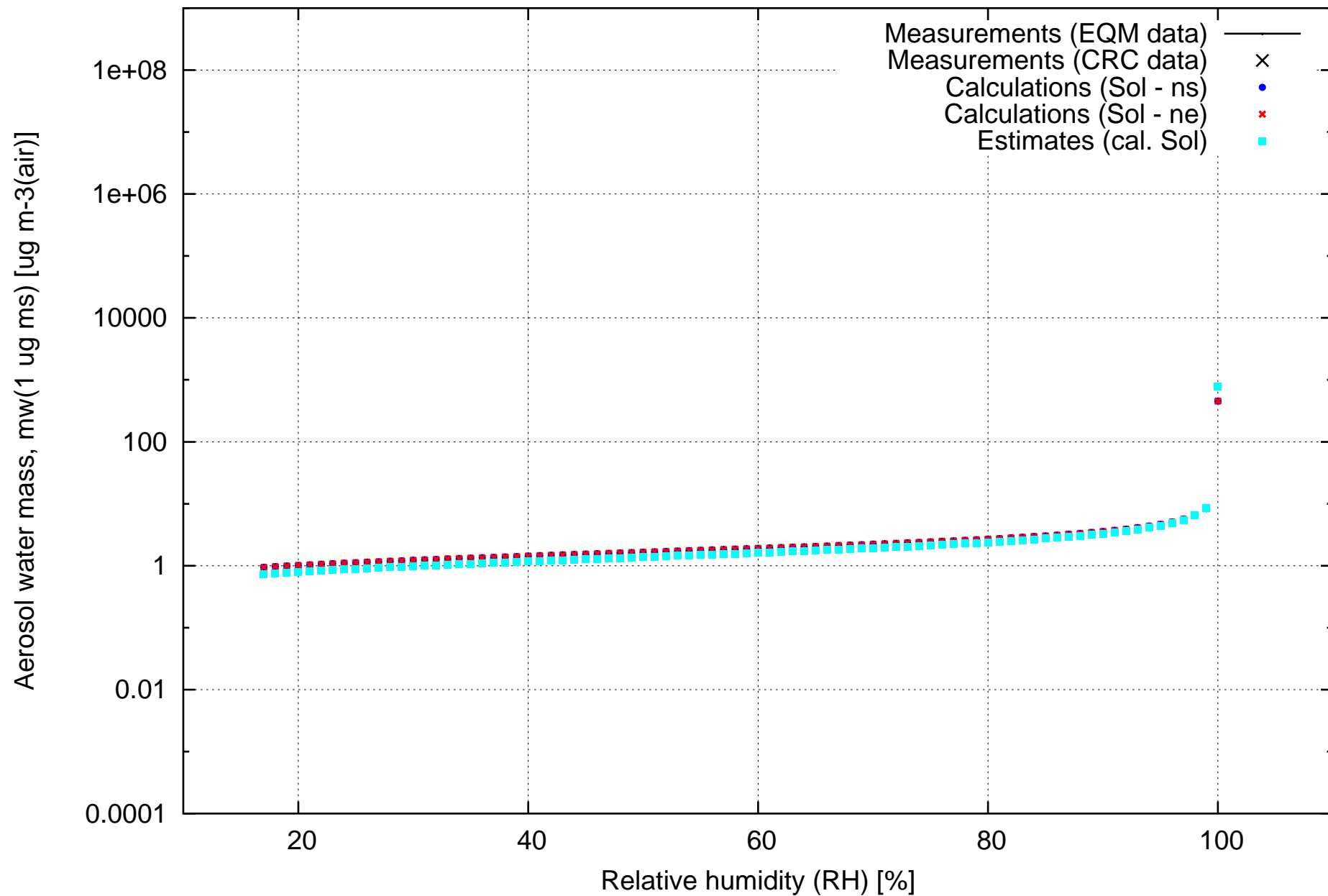
Sodium oxalate - $\text{Na}_2\text{C}_2\text{O}_4 = (\text{COONa})_2$



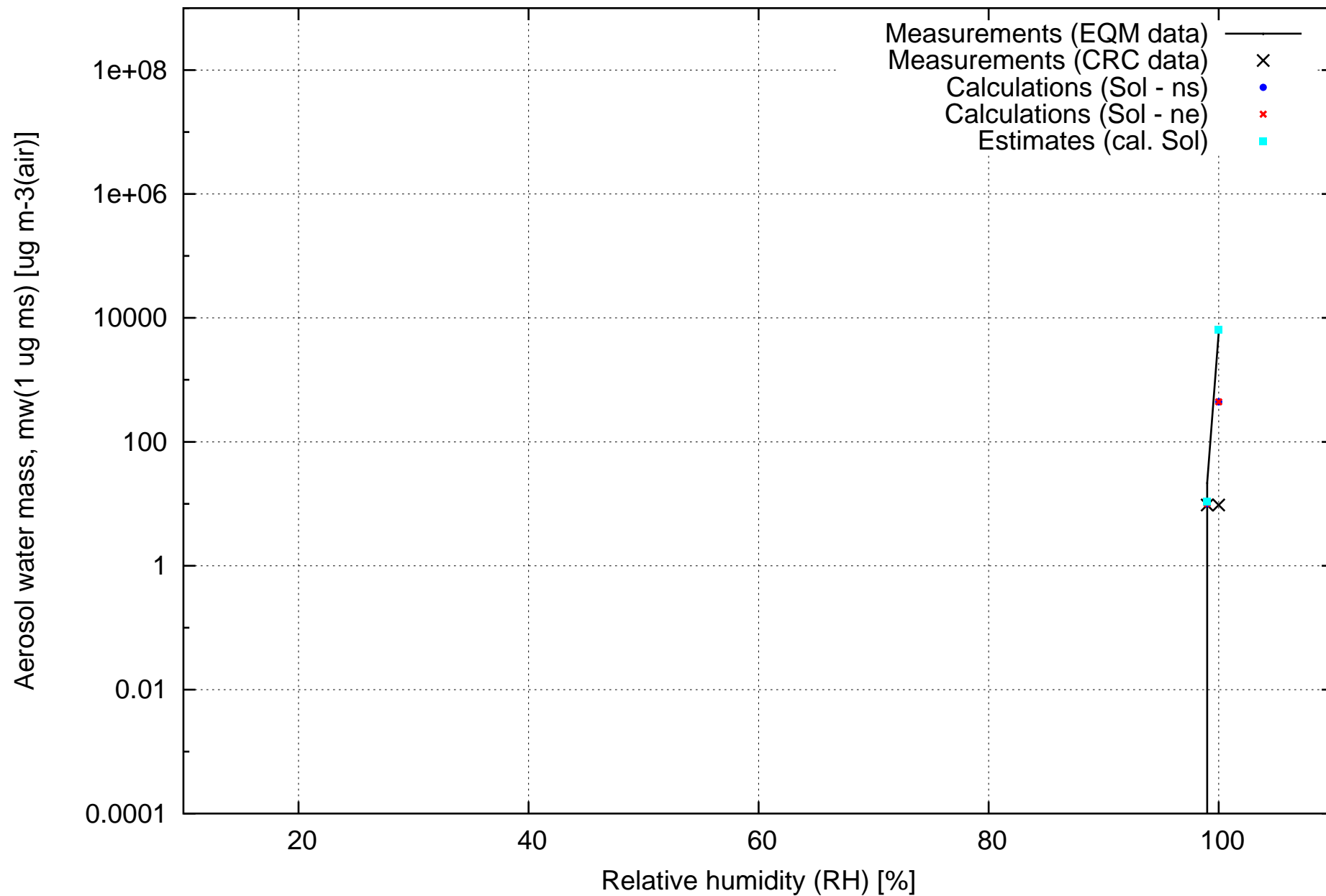
Sodium citrate - $\text{Na}_3\text{C}_6\text{H}_5\text{O}_7 = (\text{HO})\text{C}(\text{COONa})_3$



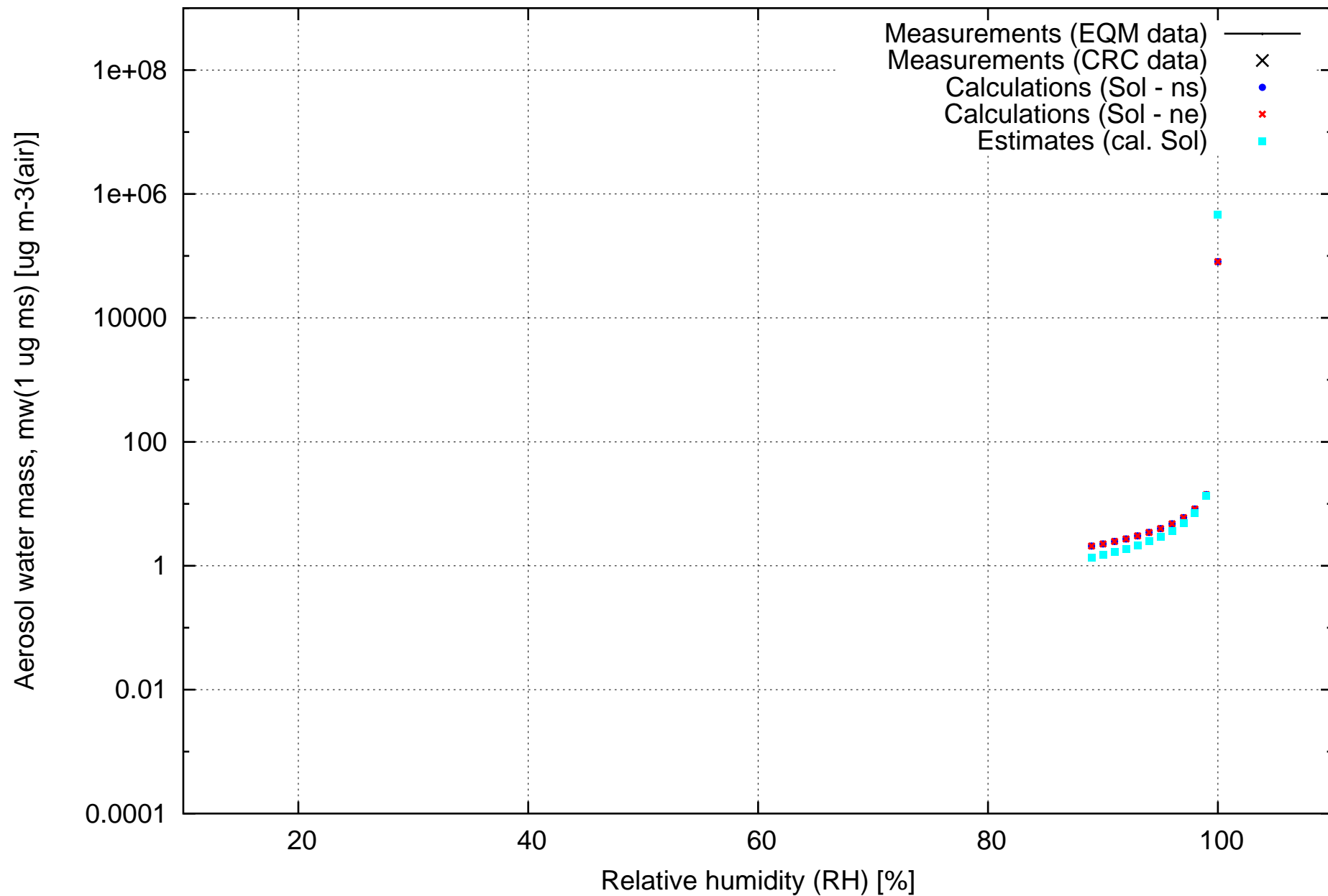
Potassium phosphate - K₃PO₄



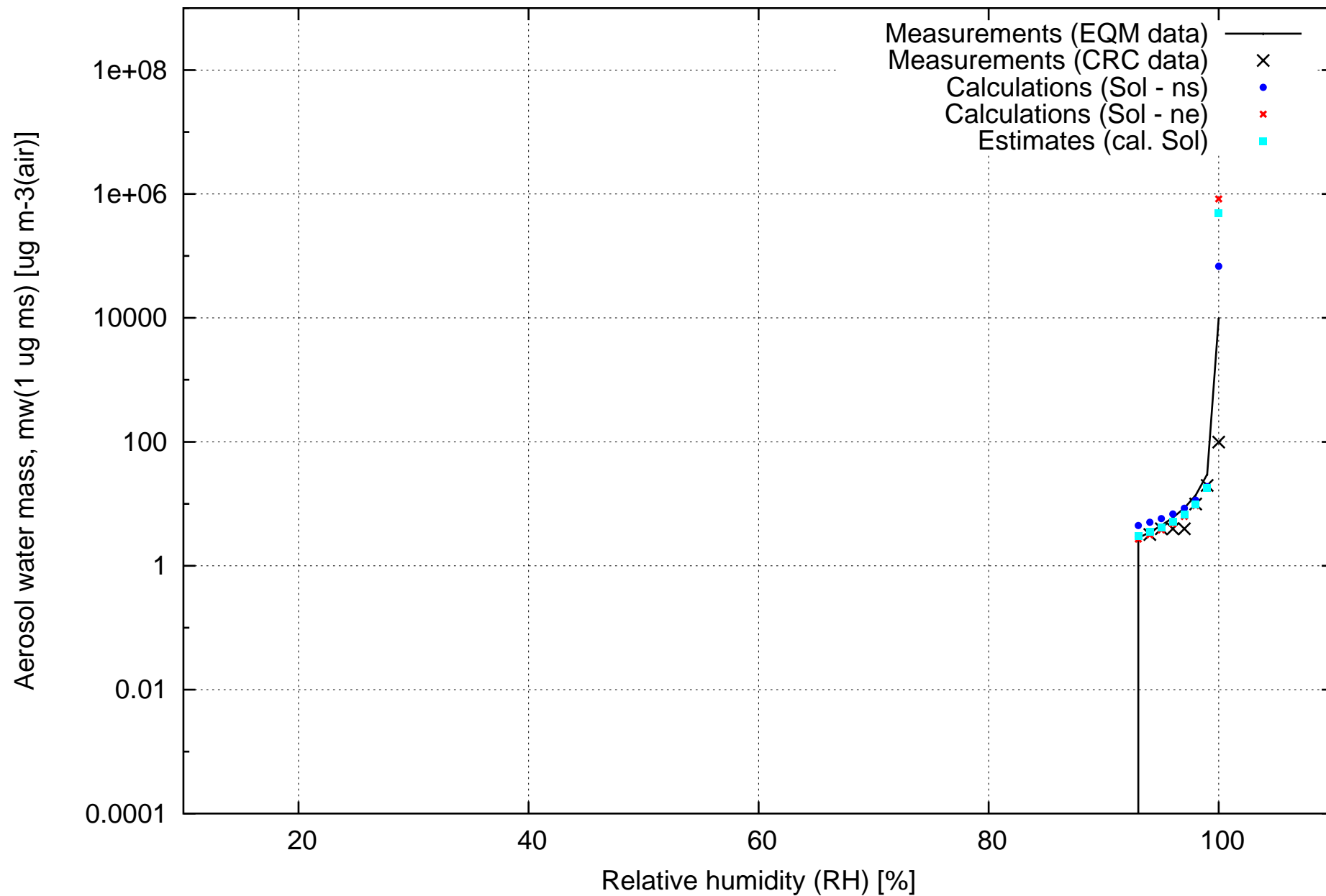
Potassium sulfate - K₂SO₄



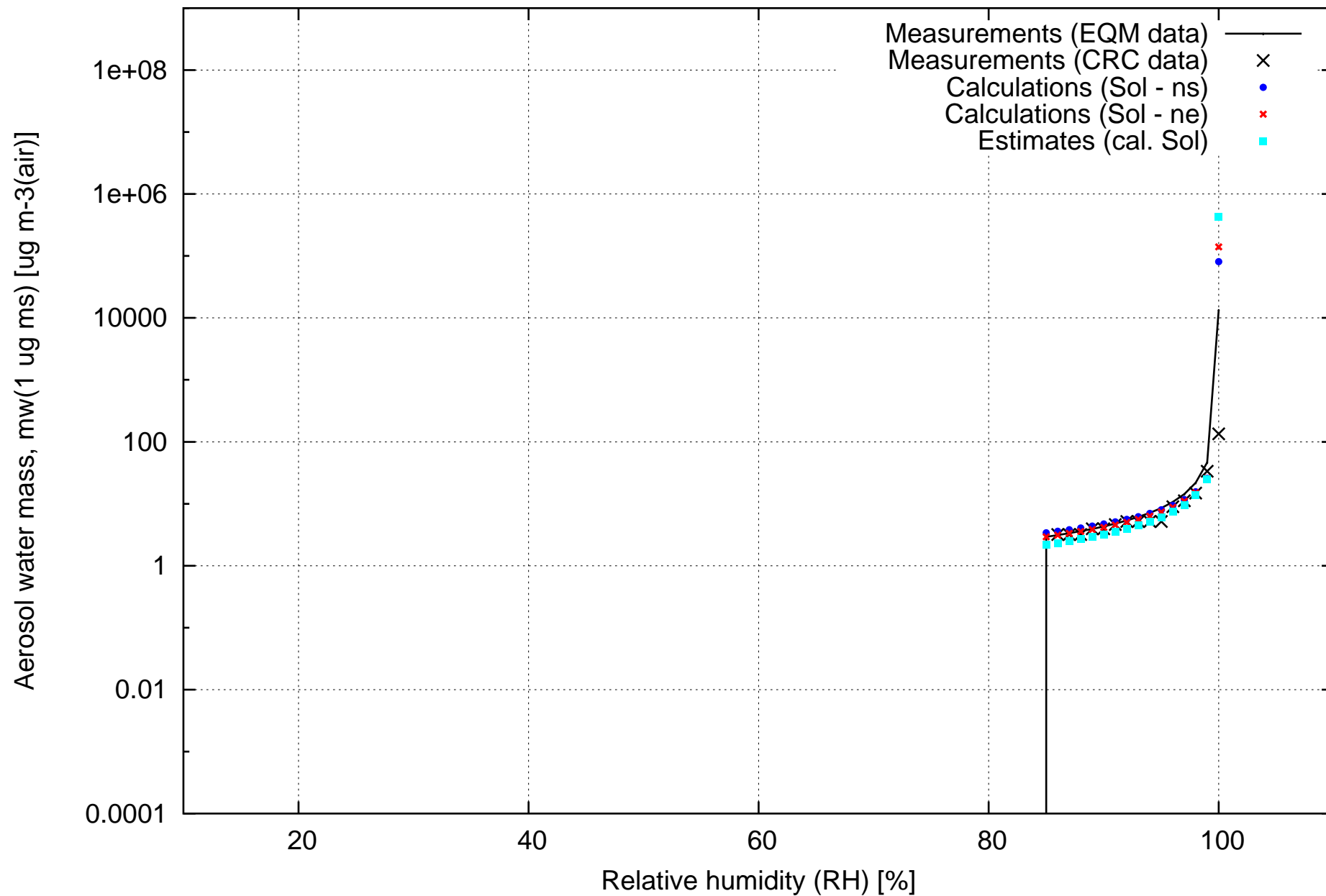
Potassium hydrogen sulfate - KHSO4



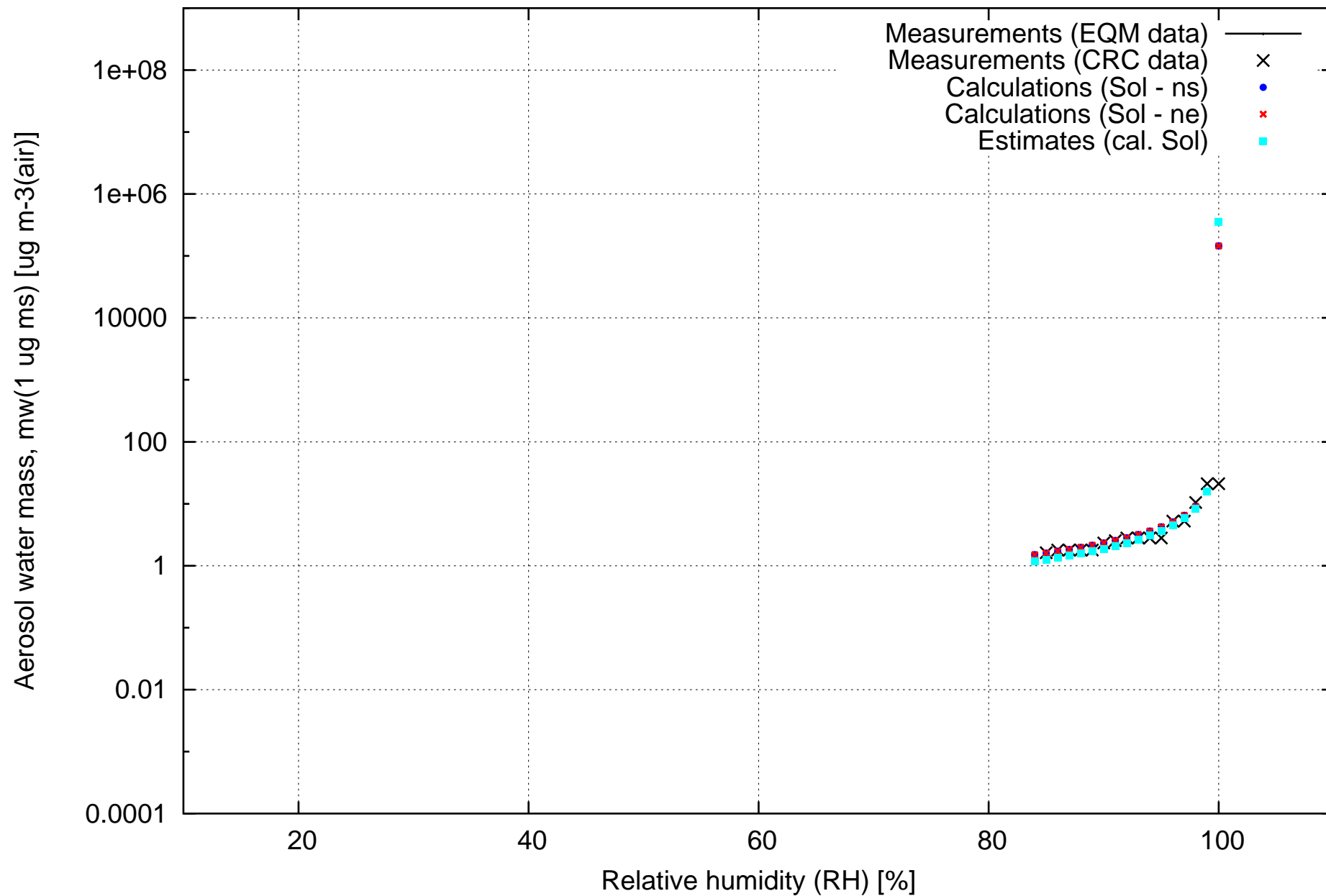
Potassium nitrate - KNO₃



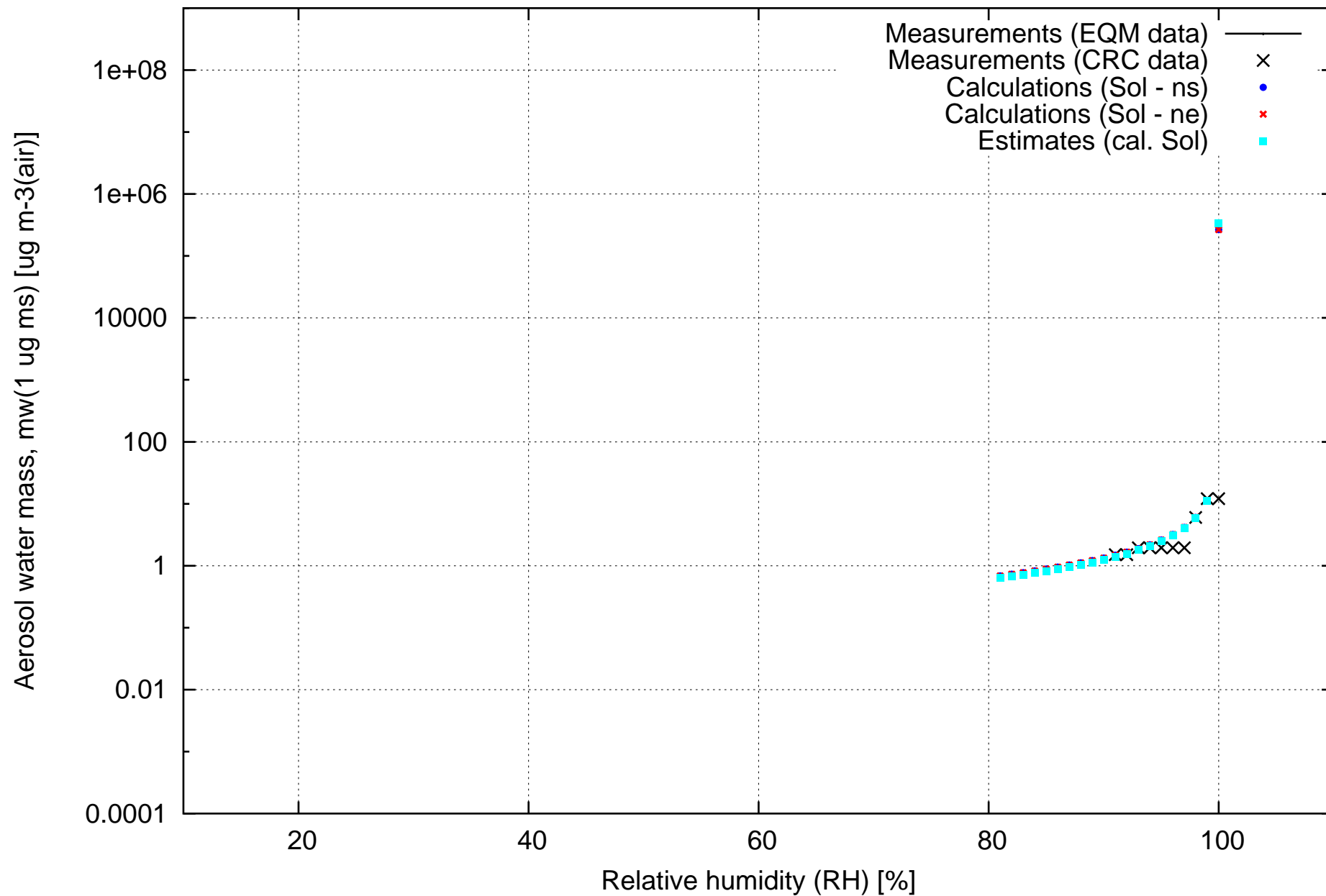
Potassium chloride - KCl



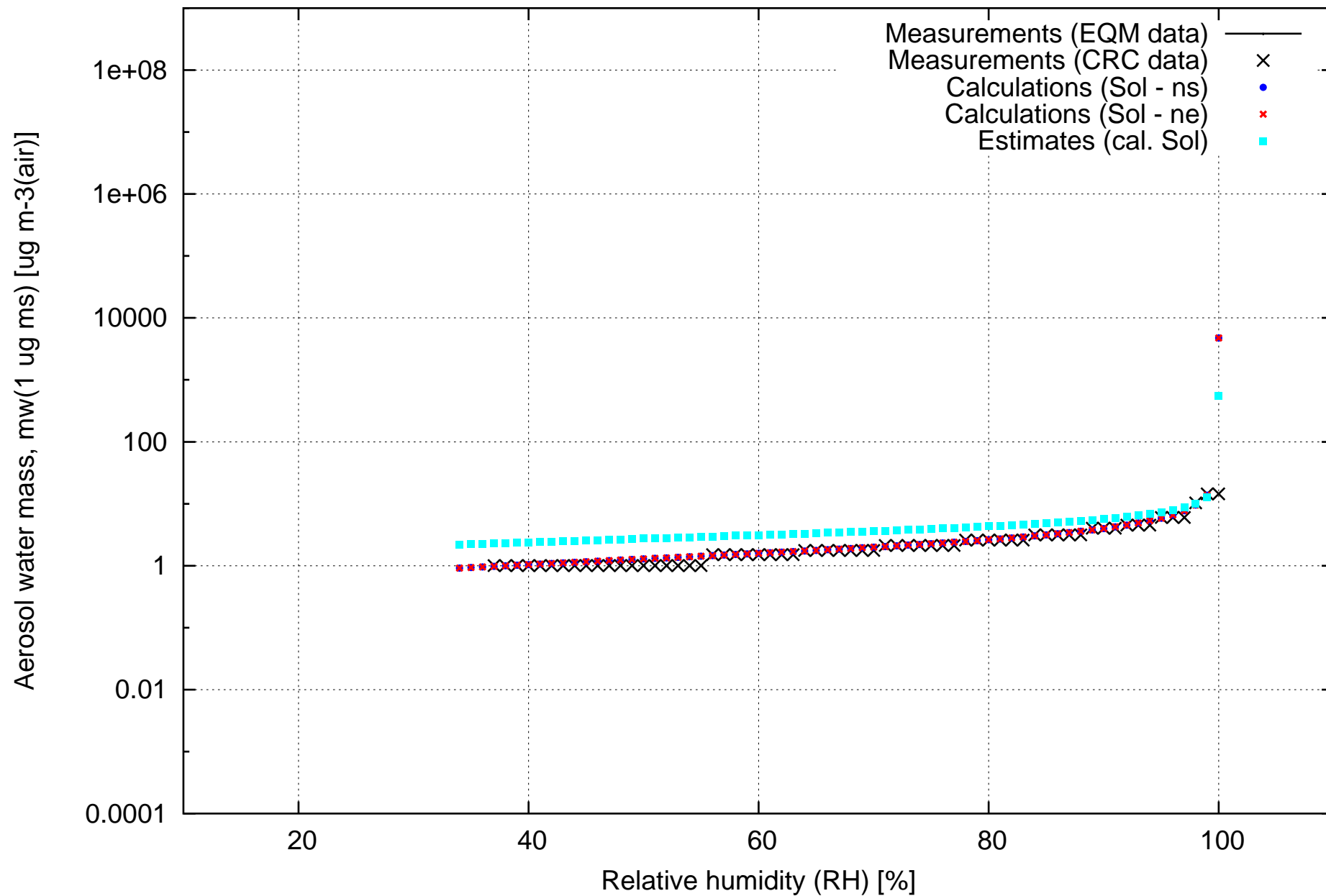
Potassium bromide - KBr



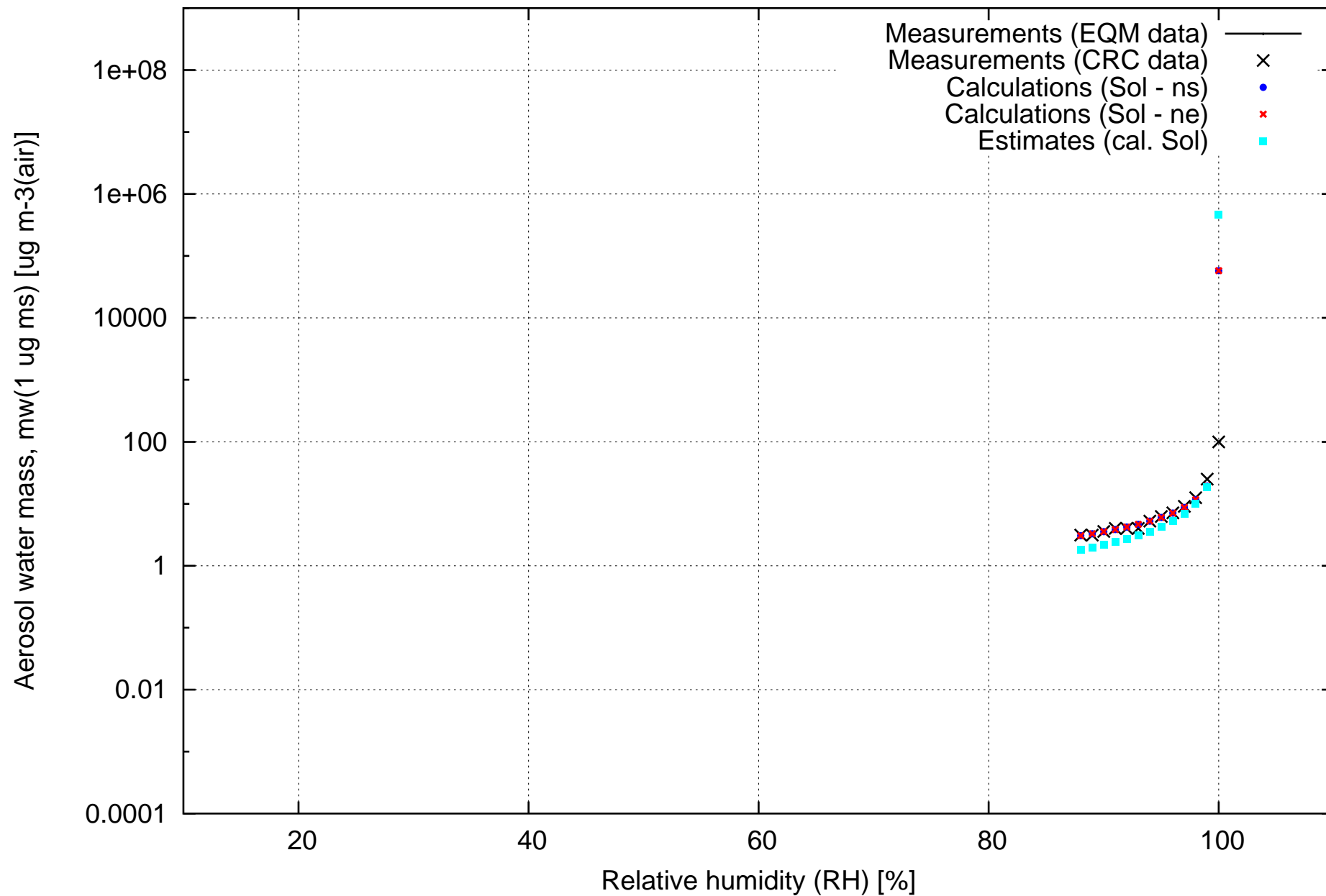
Potassium iodide - KI



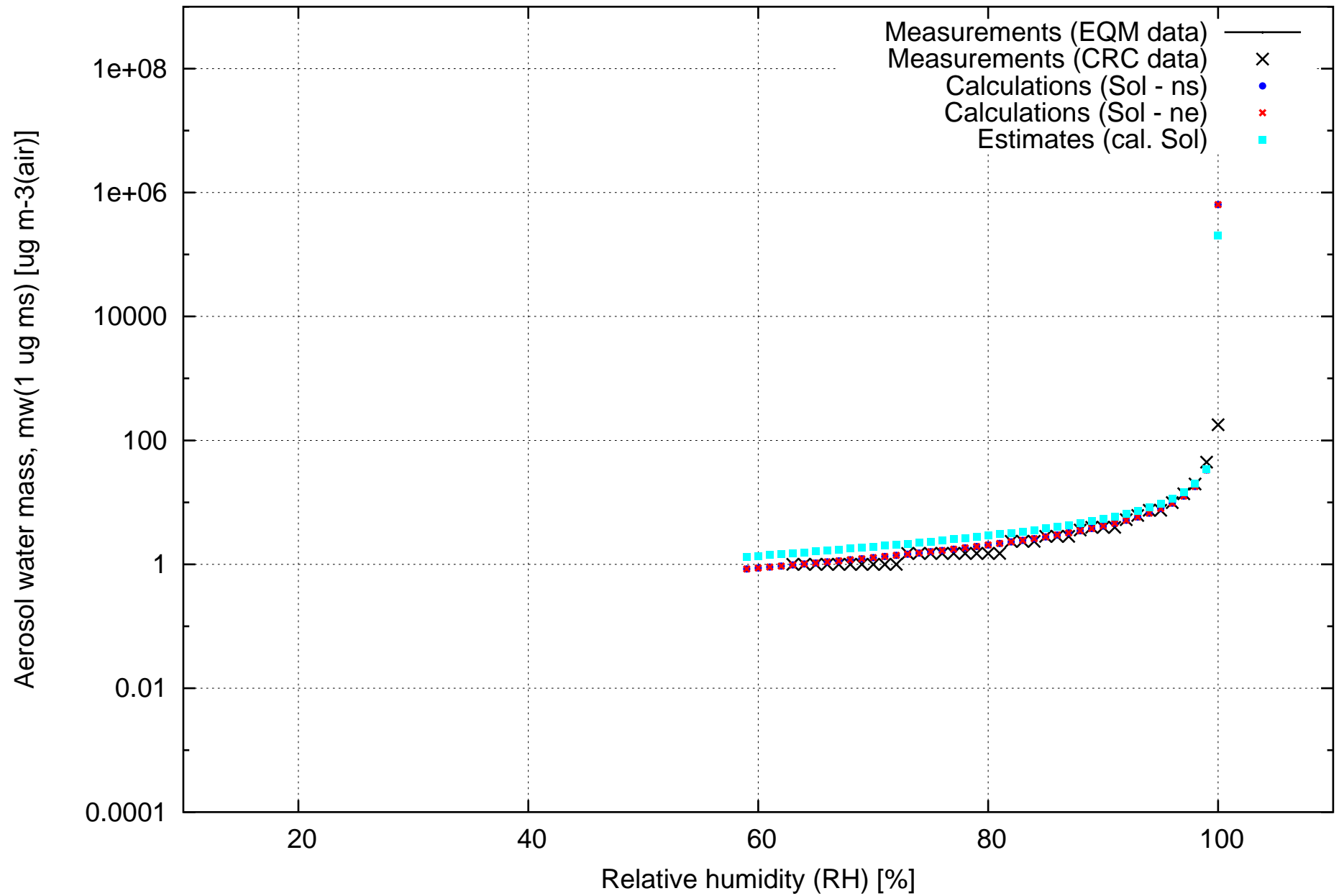
Potassium carbonate - K₂CO₃



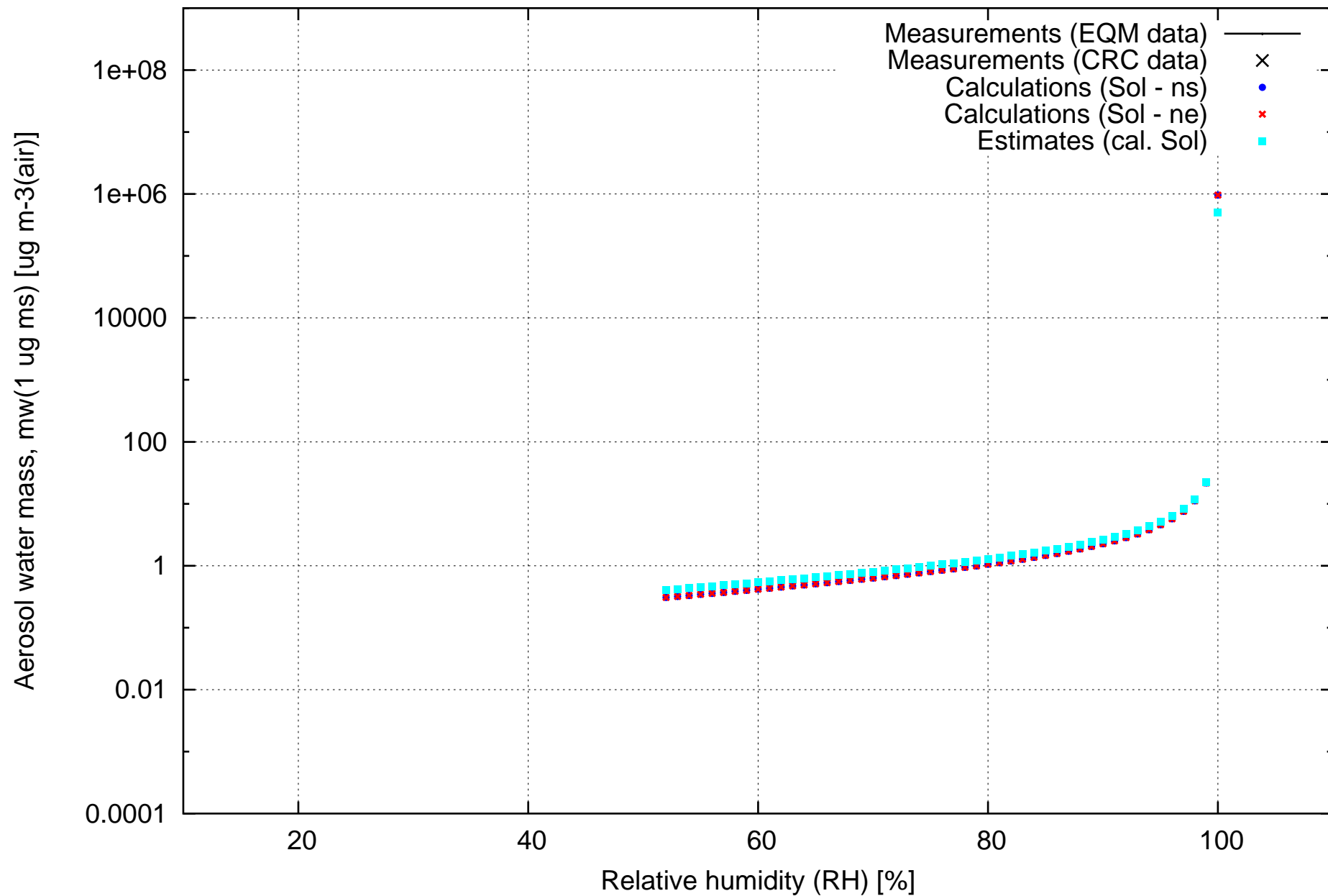
Potassium hydrogen carbonate - KHCO₃



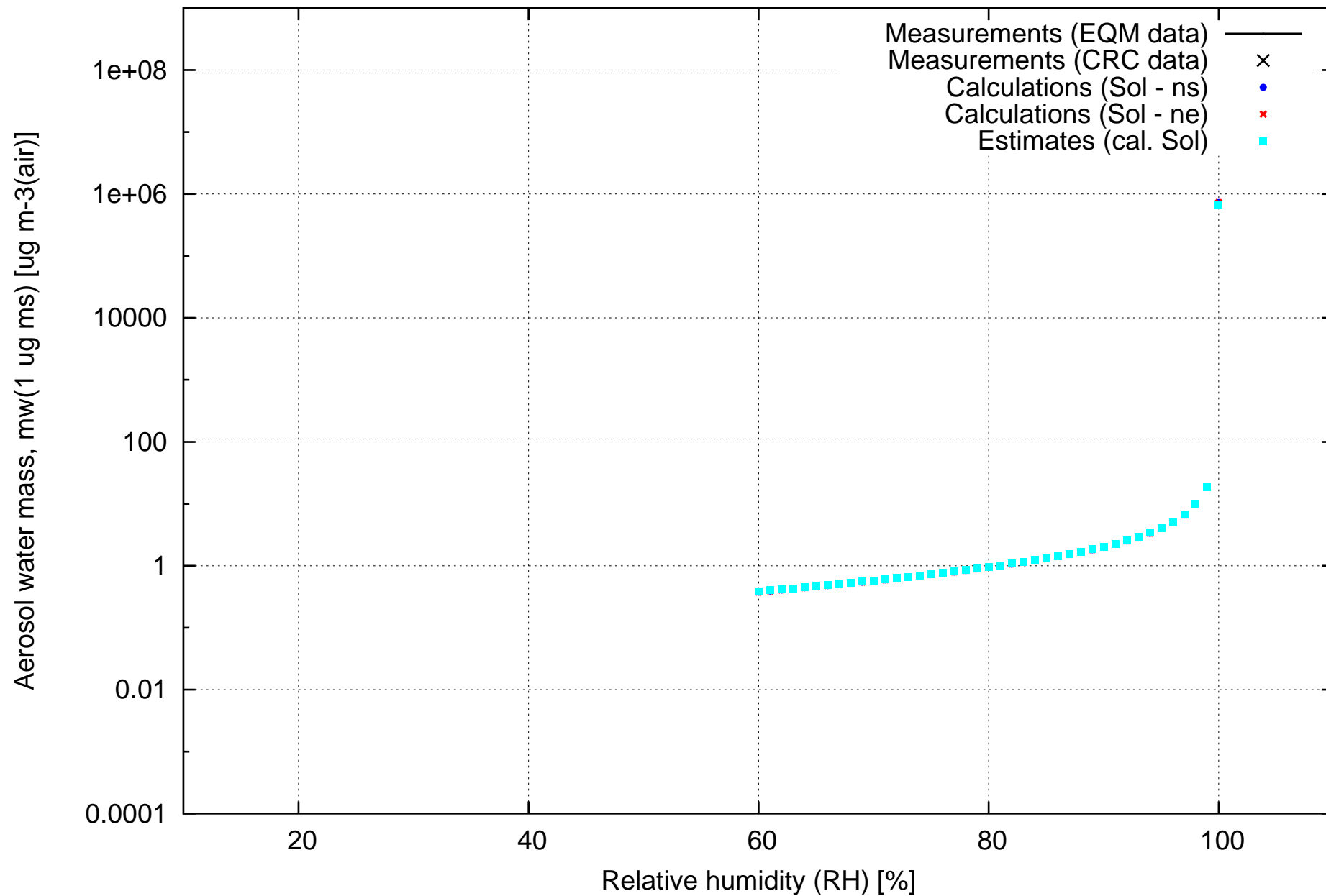
Potassium hydroxide - KOH



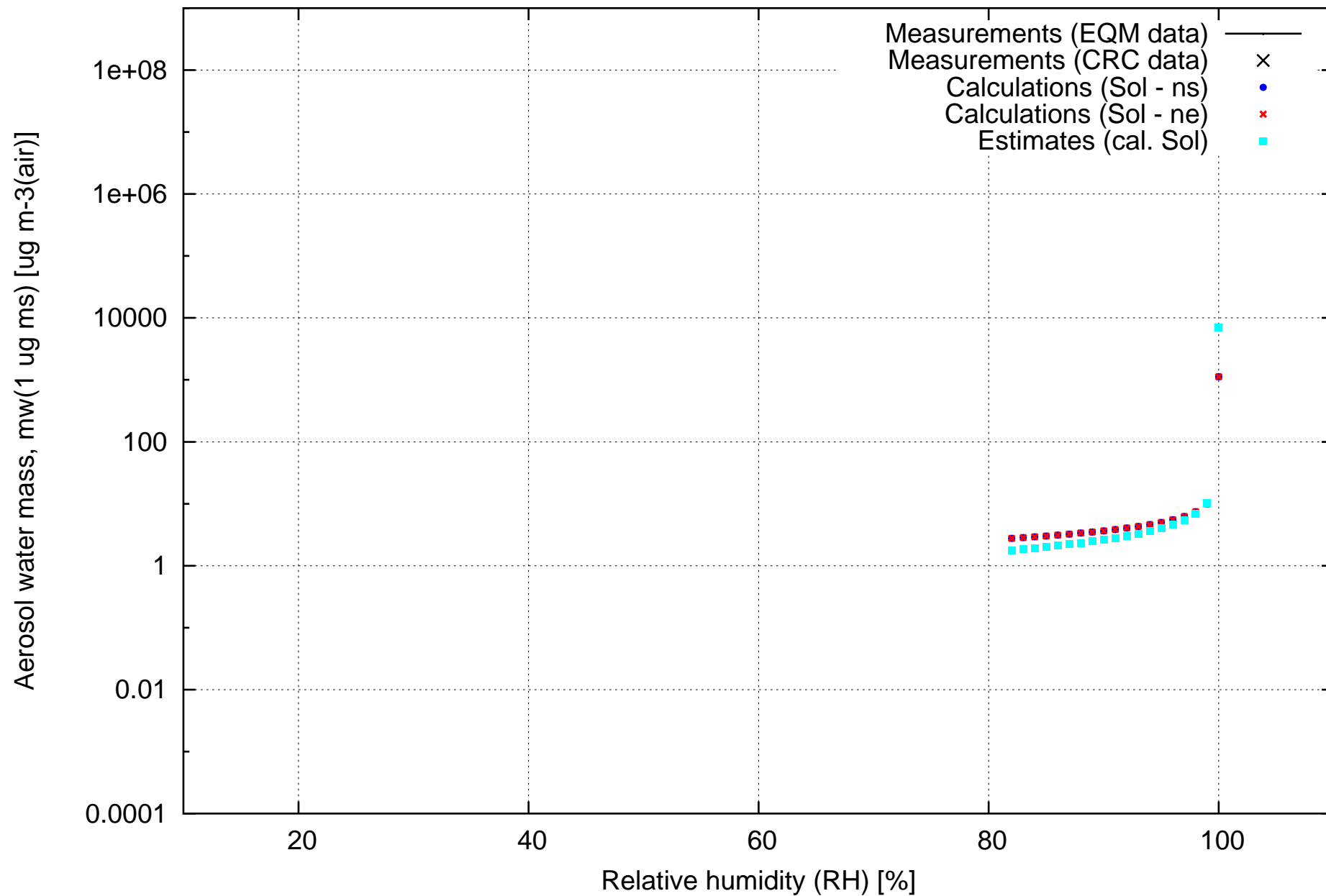
Potassium formate - $\text{KCHO}_2 = \text{H}(\text{COOK})$



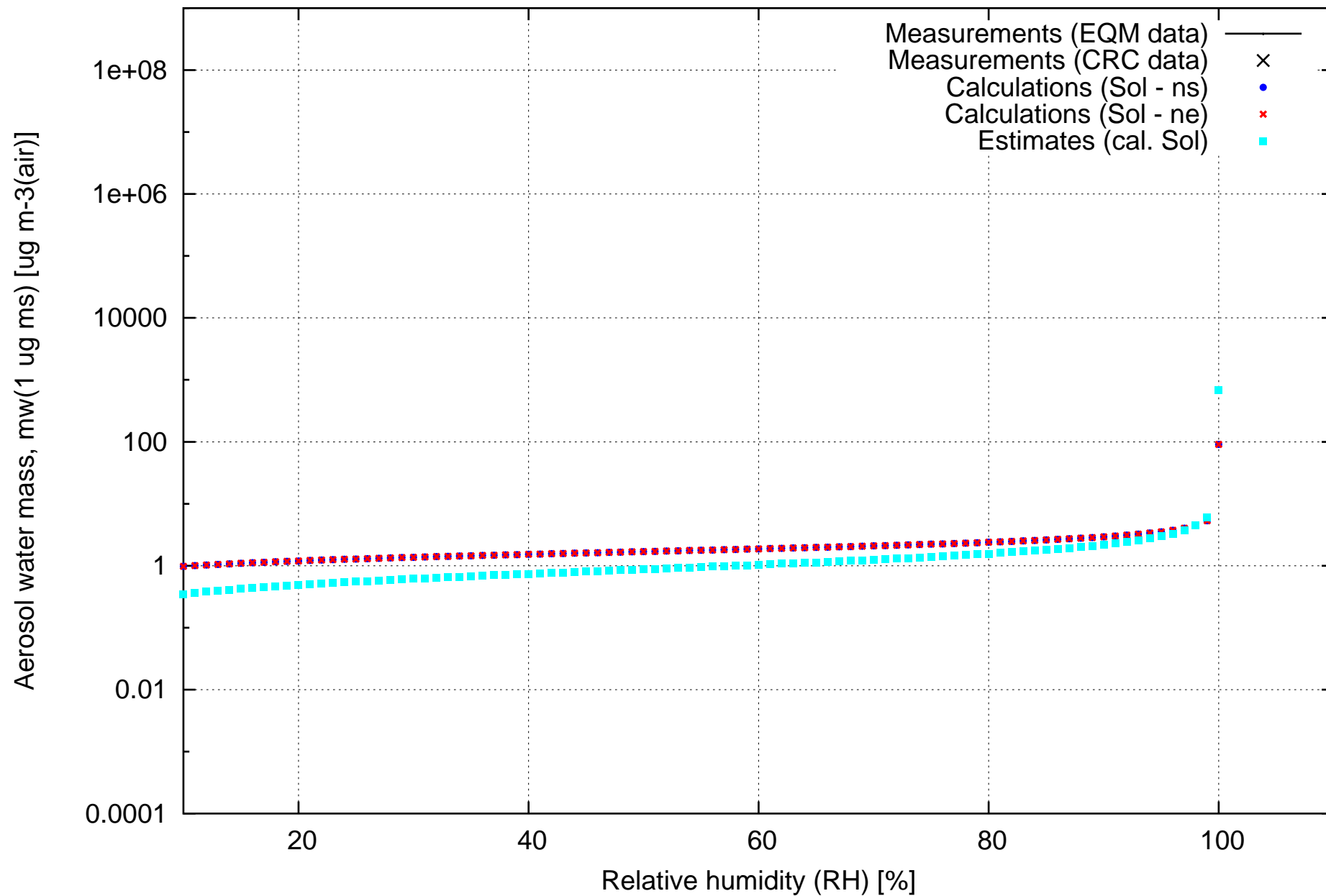
Potassium acetate - $\text{KC}_2\text{H}_3\text{O}_2 = \text{CH}_3(\text{COOK})$



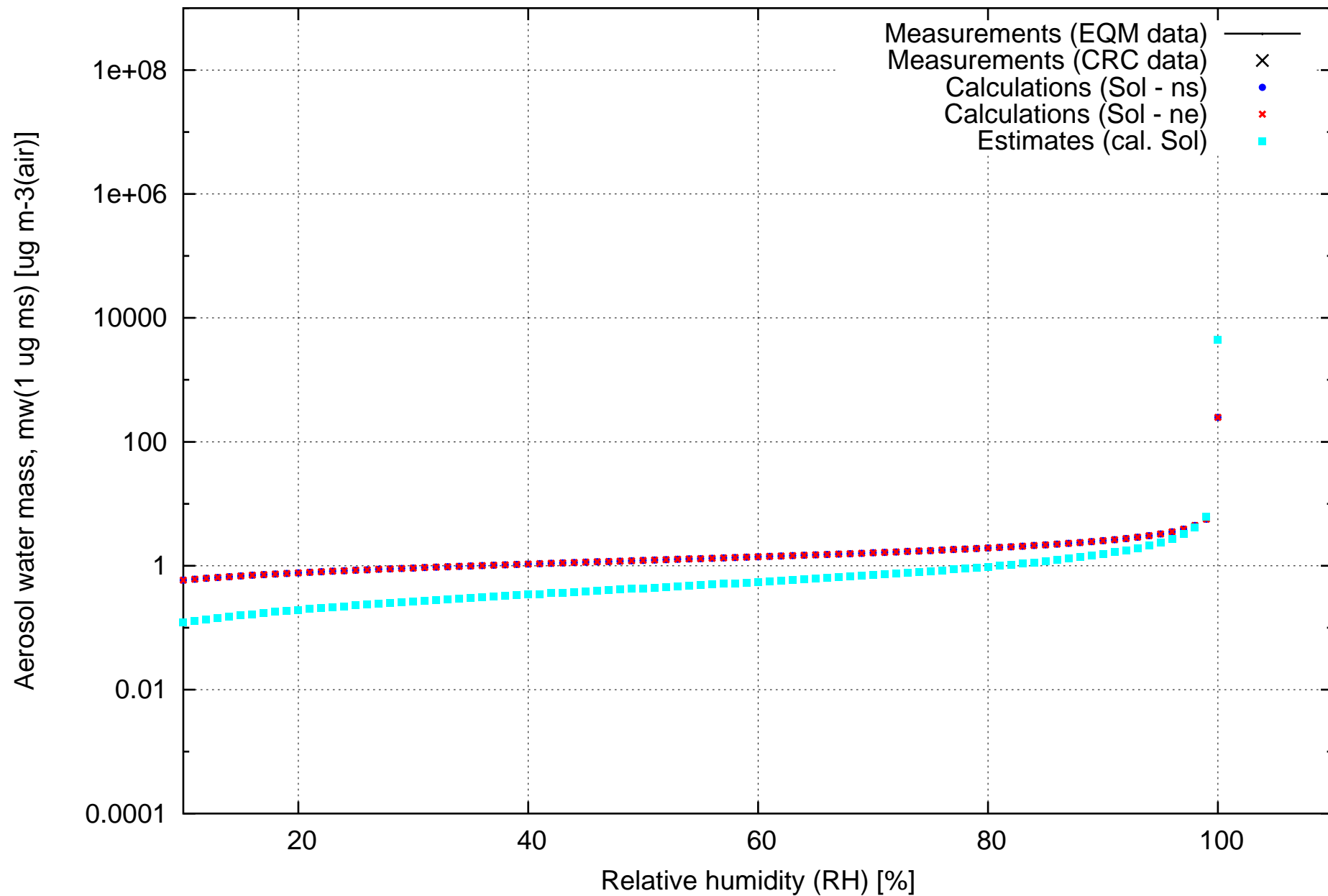
Potassium oxalate - $\text{K}_2\text{C}_2\text{O}_4 = (\text{COOK})_2$



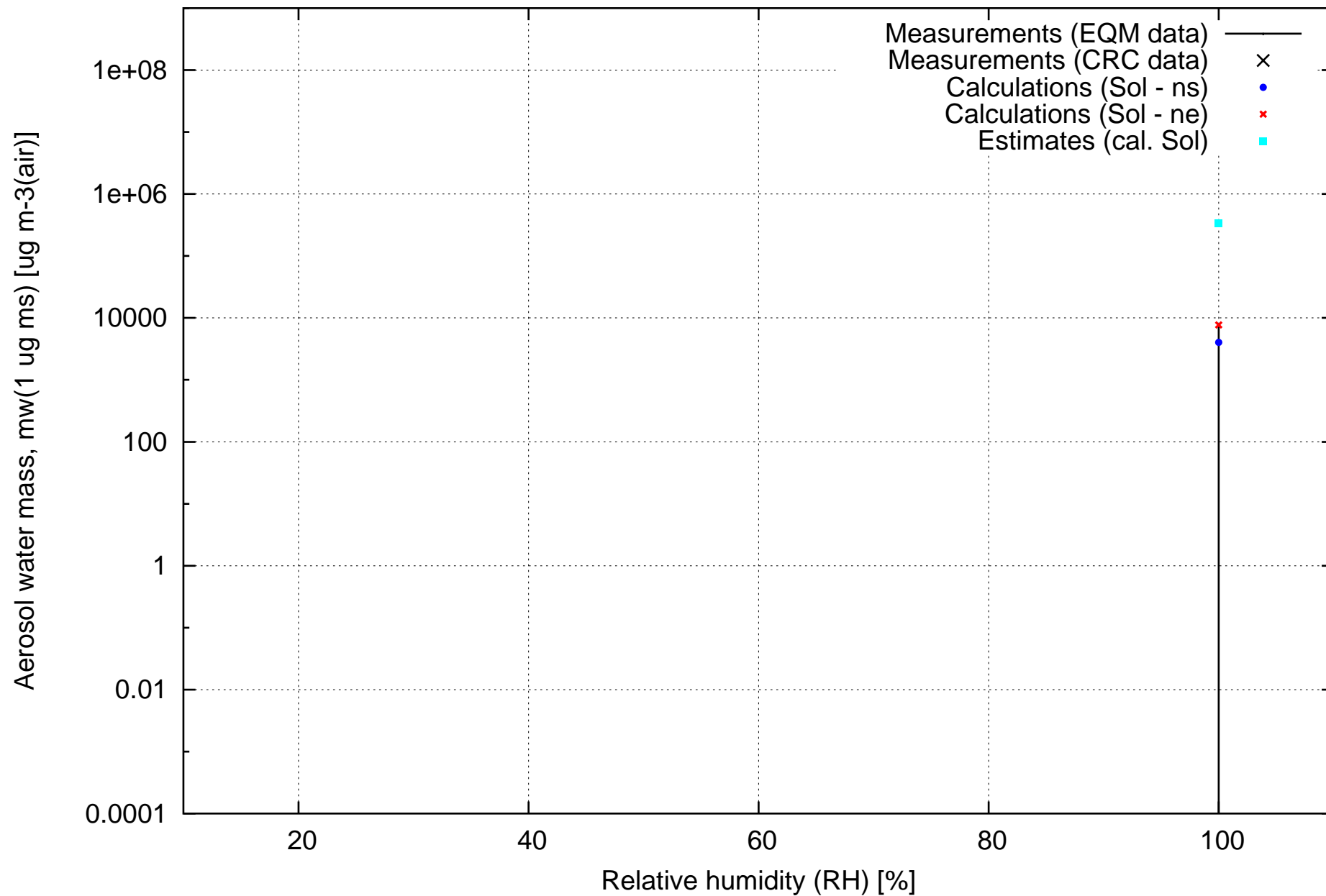
Potassium citrate - $\text{K}_3\text{C}_6\text{H}_5\text{O}_7 = (\text{HO})\text{C}(\text{COOK})_3$



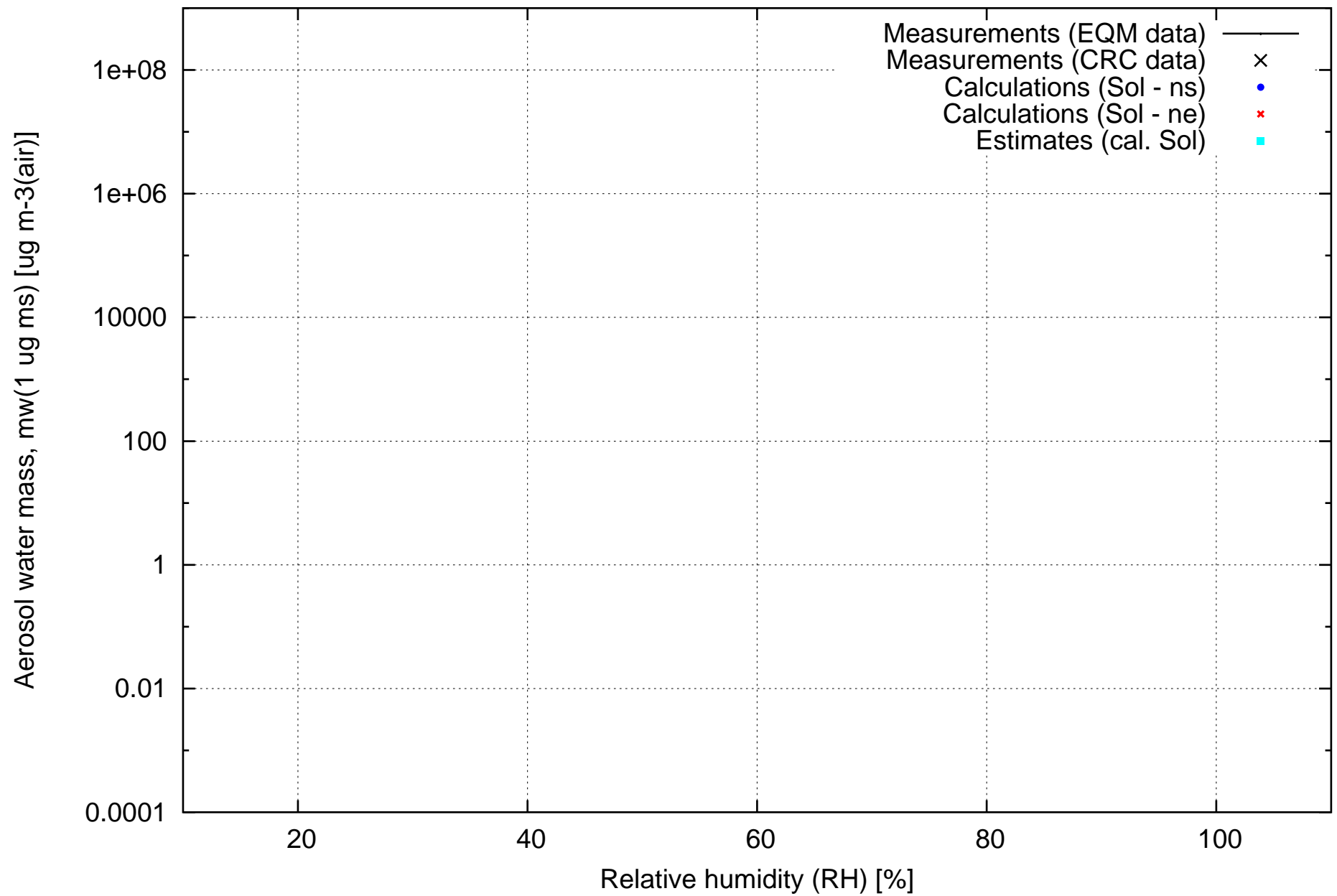
Calcium phosphate - $\text{Ca}_3(\text{PO}_4)_2$



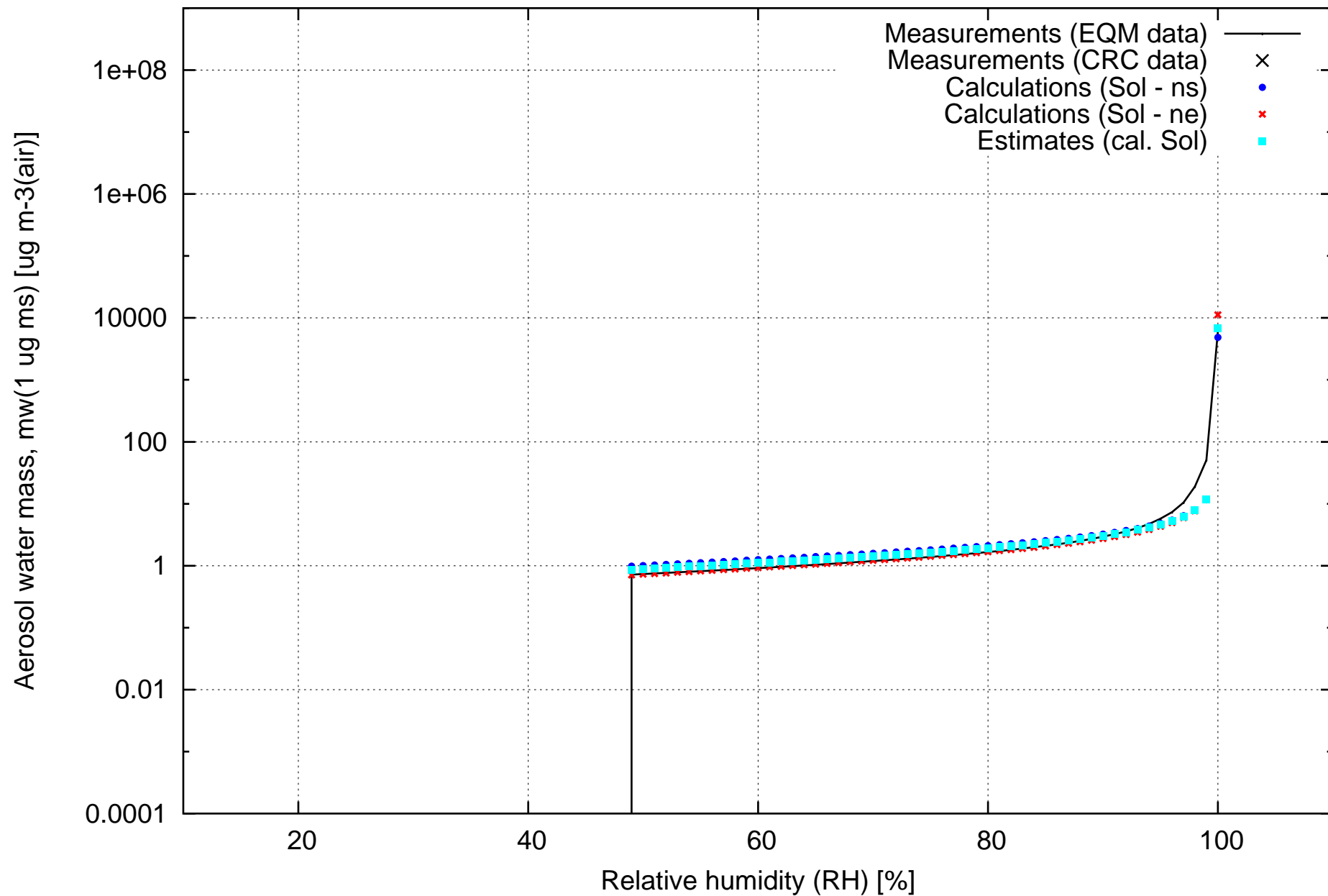
Calcium sulfate - CaSO₄



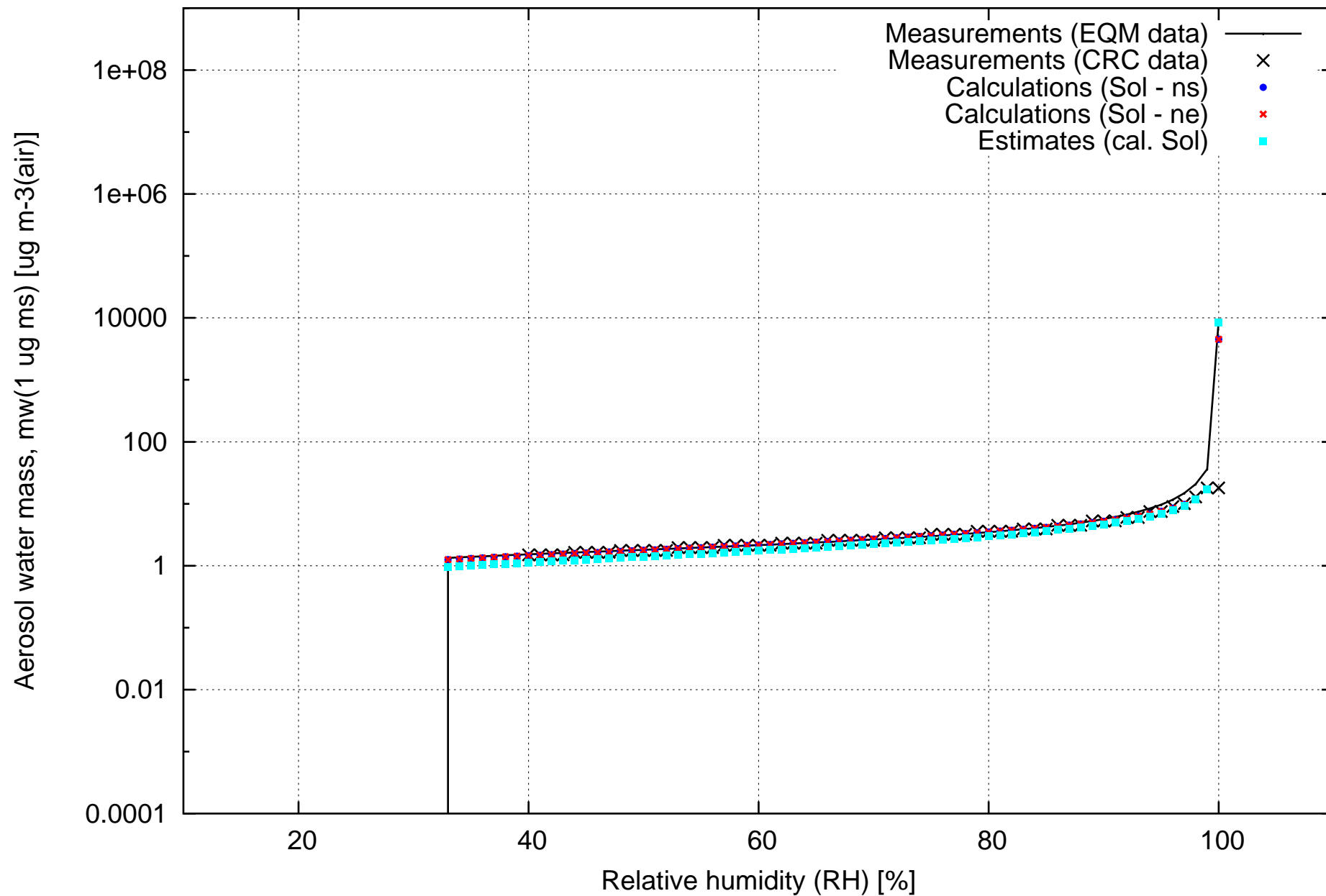
Calcium - dummy 03



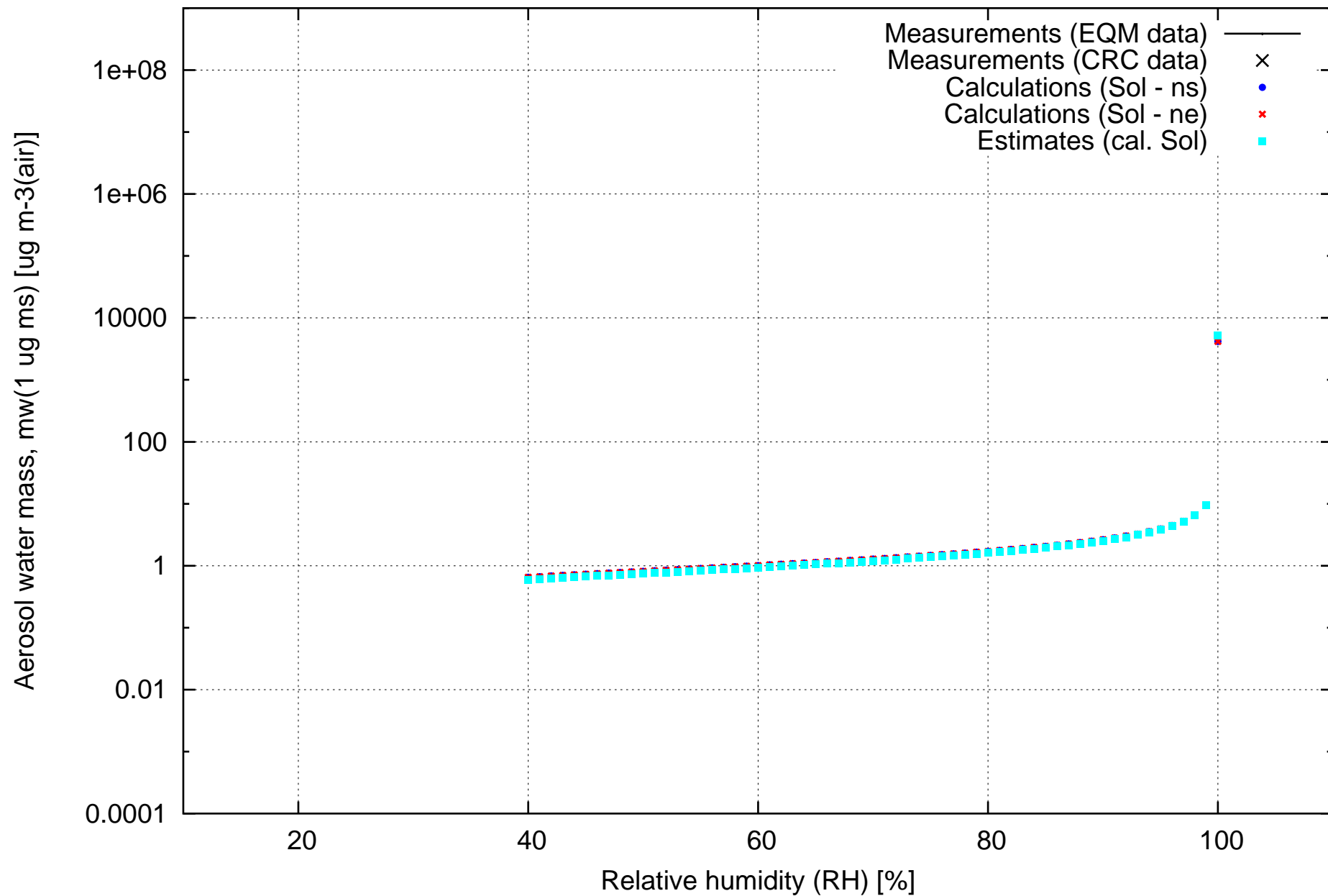
Calcium nitrate - $\text{Ca}(\text{NO}_3)_2$



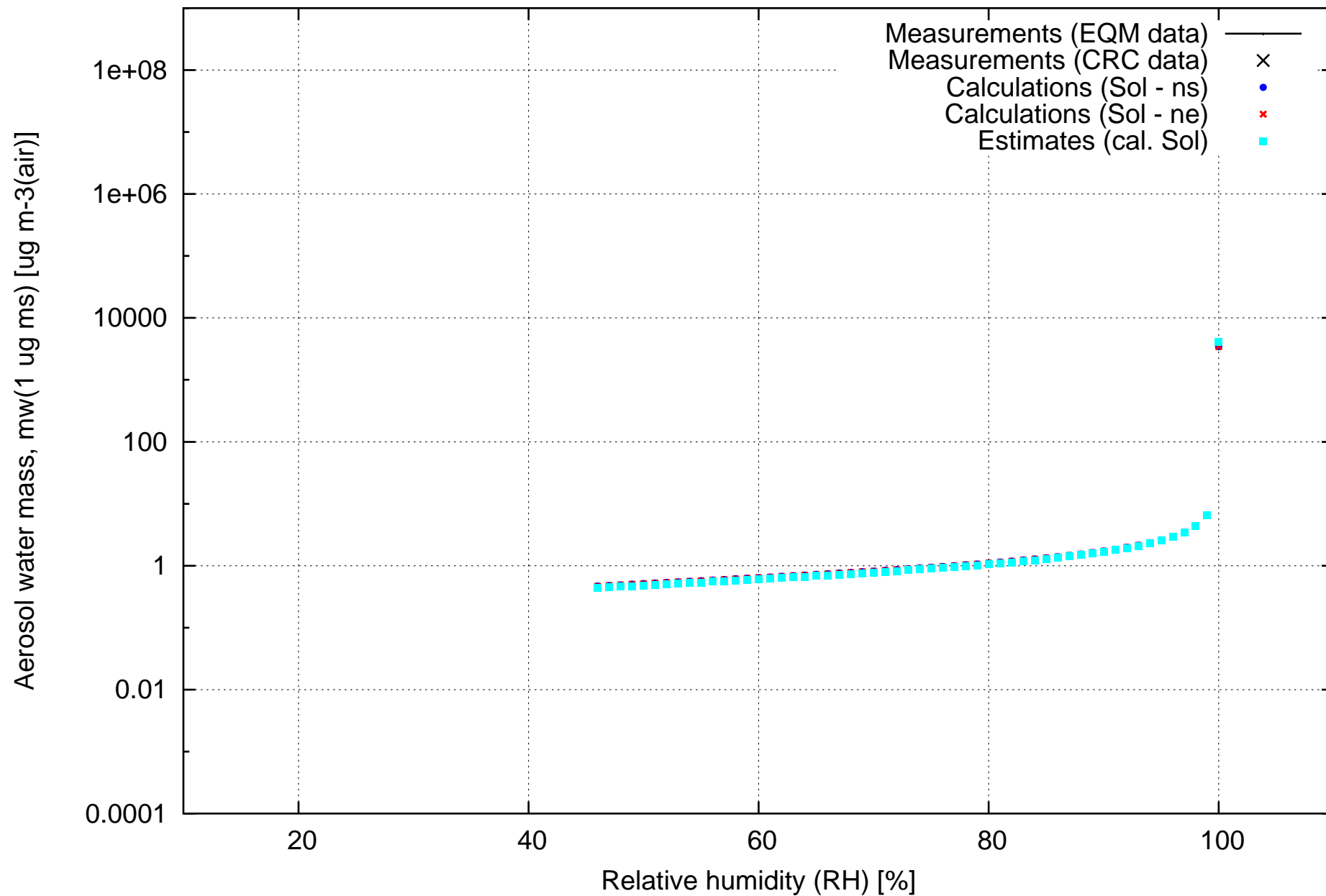
Calcium chloride - CaCl₂



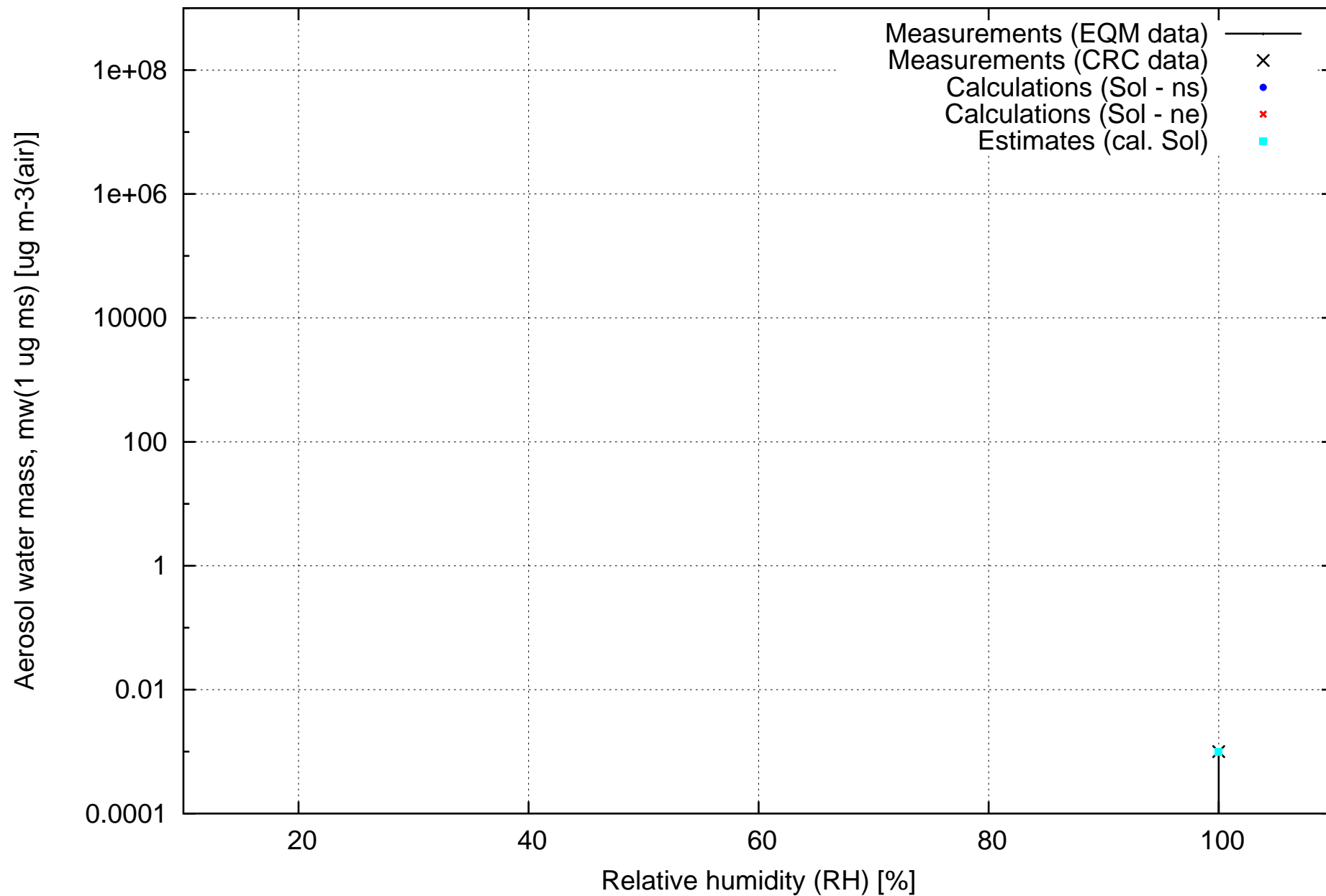
Calcium bromide - CaBr₂



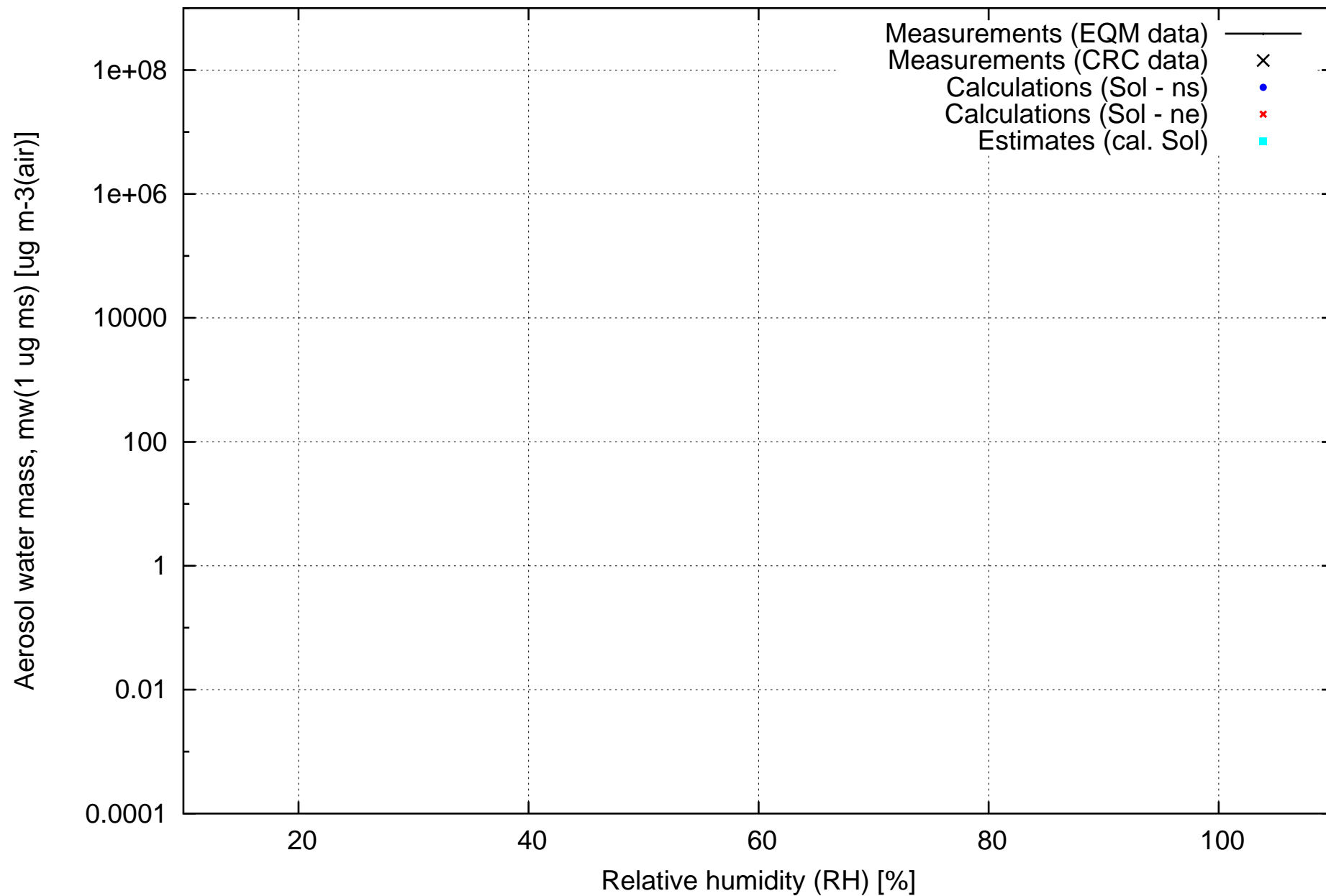
Calcium iodide - Cal2



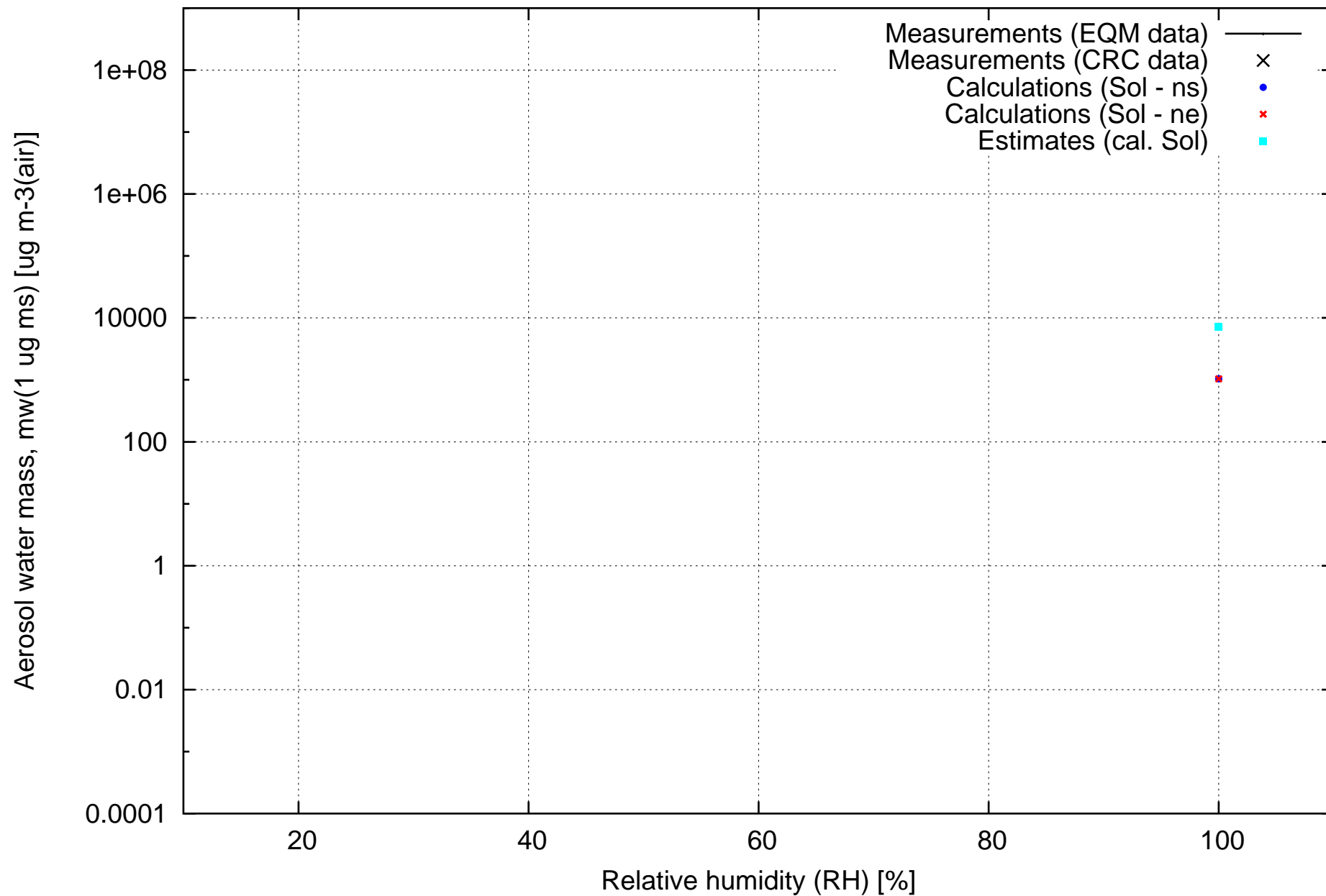
Calcium carbonate - CaCO₃



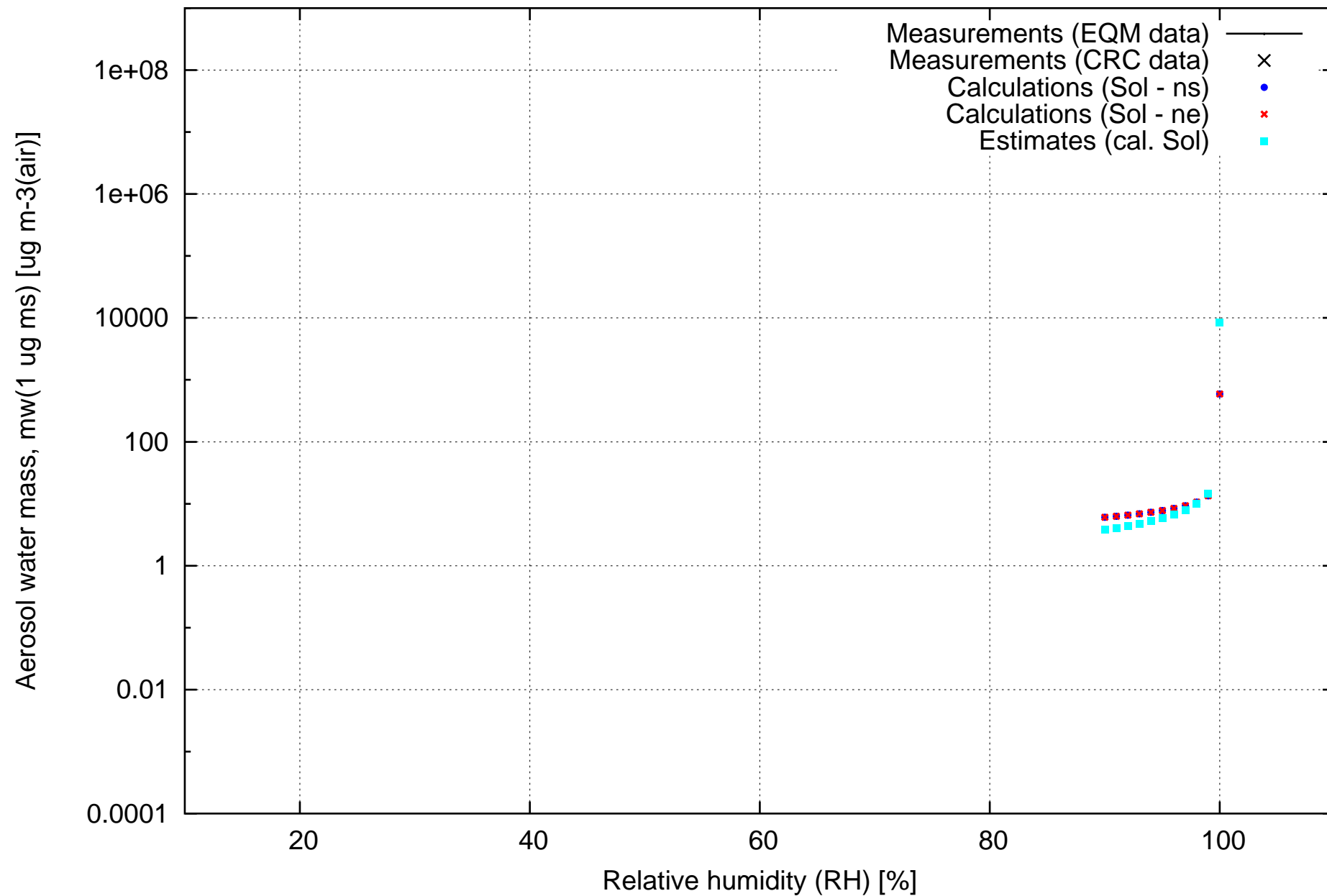
Calcium - dummy 09



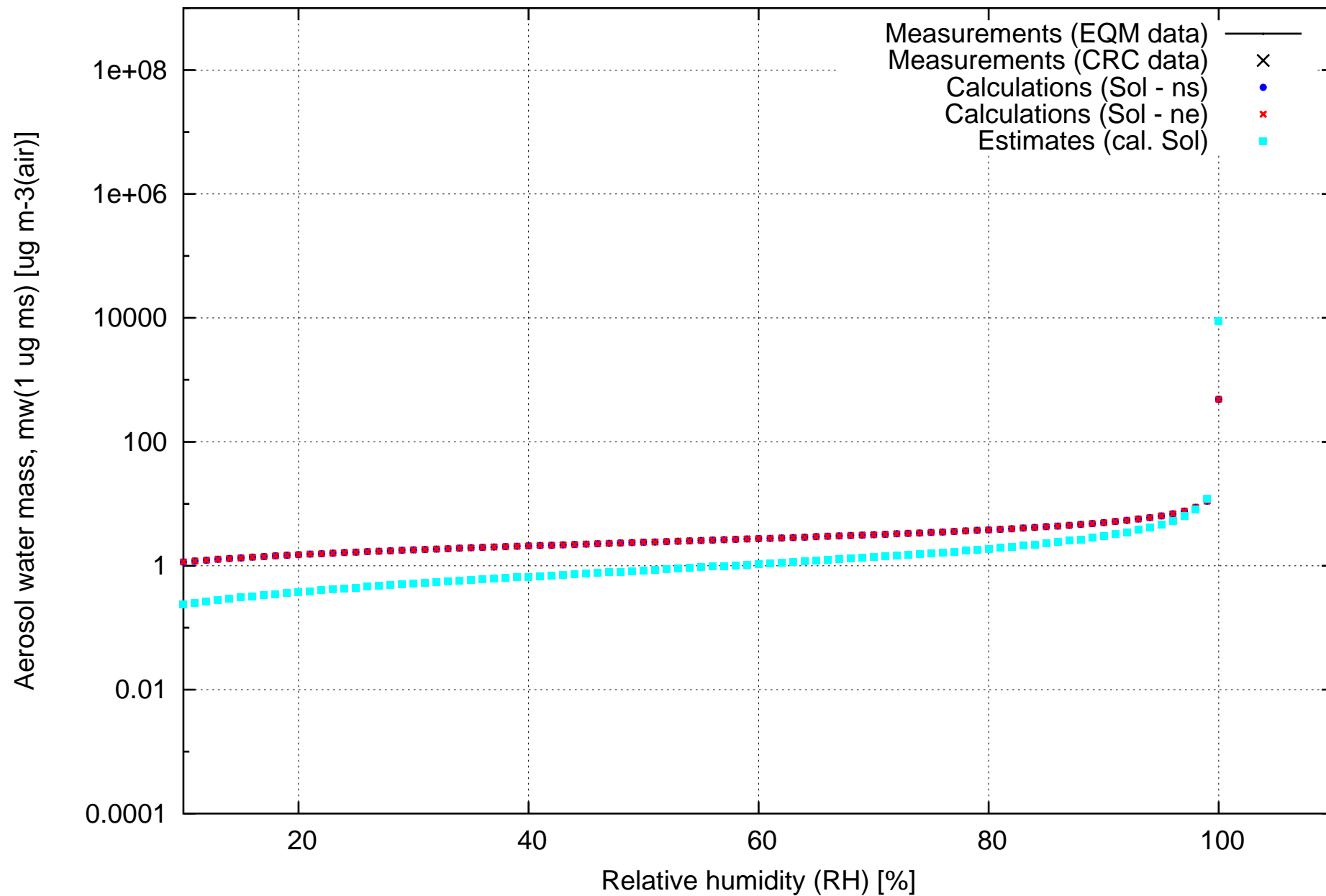
Calcium hydroxide - Ca(OH)₂



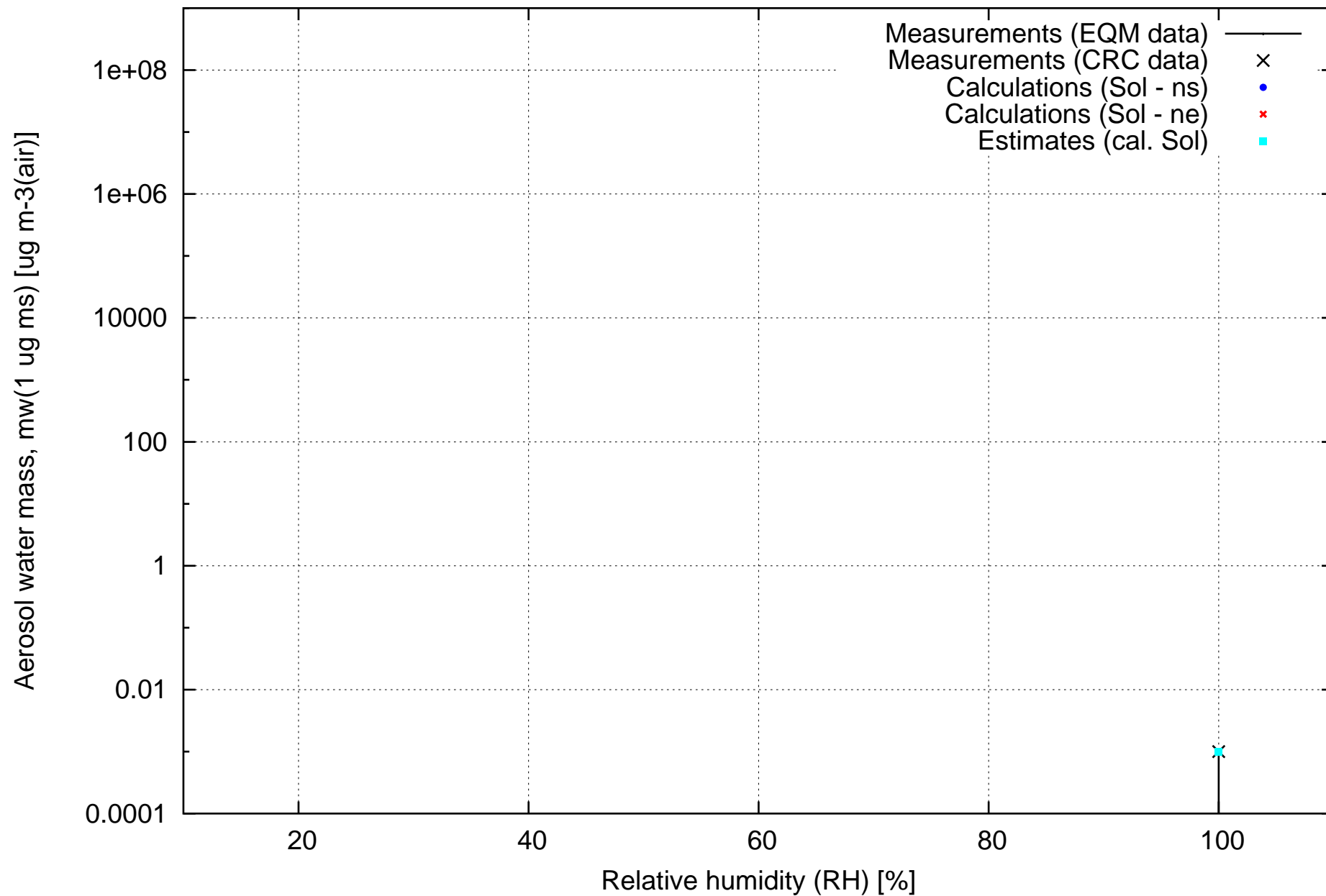
Calcium formate - $\text{Ca}(\text{CHO}_2)_2$



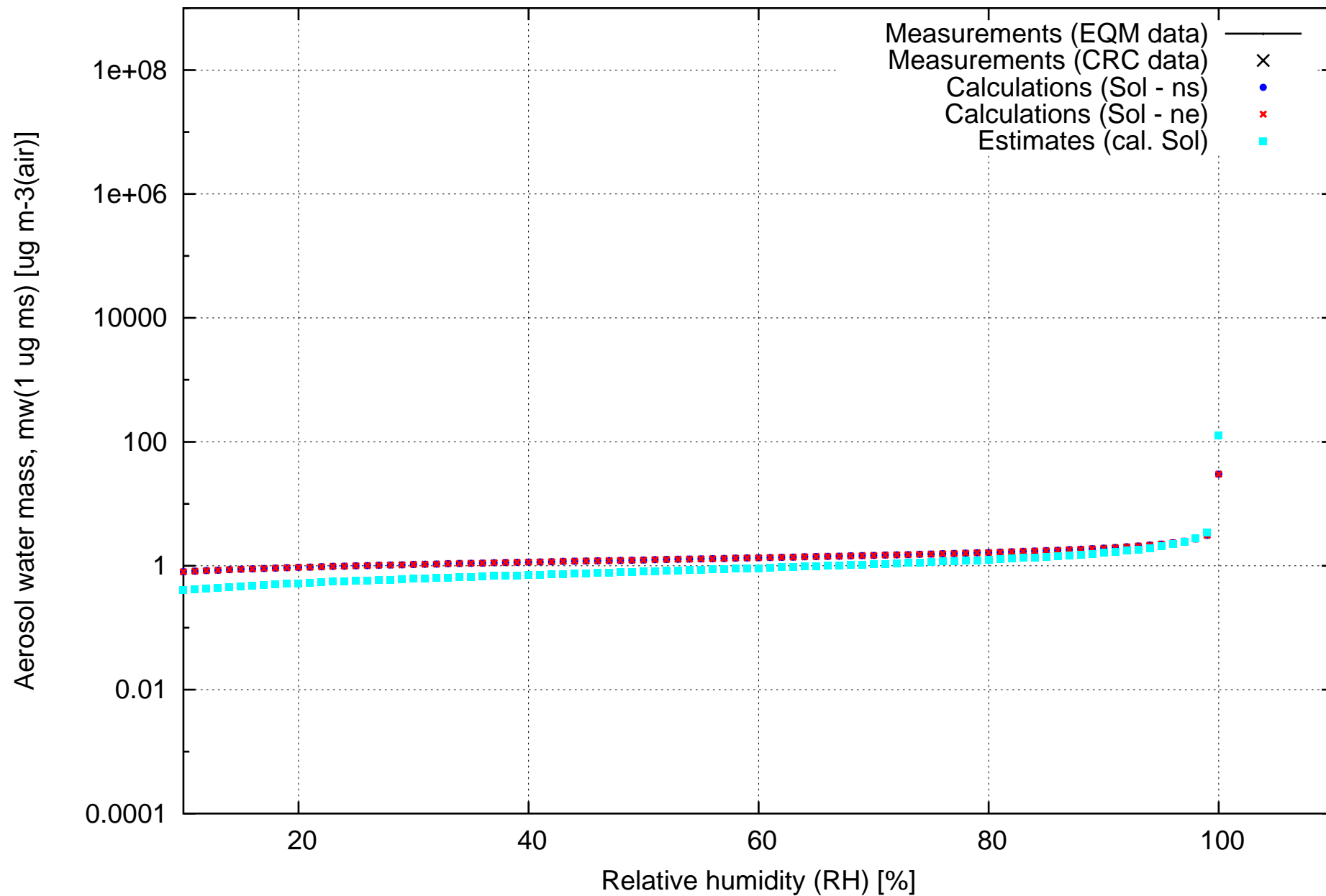
Calcium acetate - $\text{Ca}(\text{C}_2\text{H}_3\text{O}_2)_2$



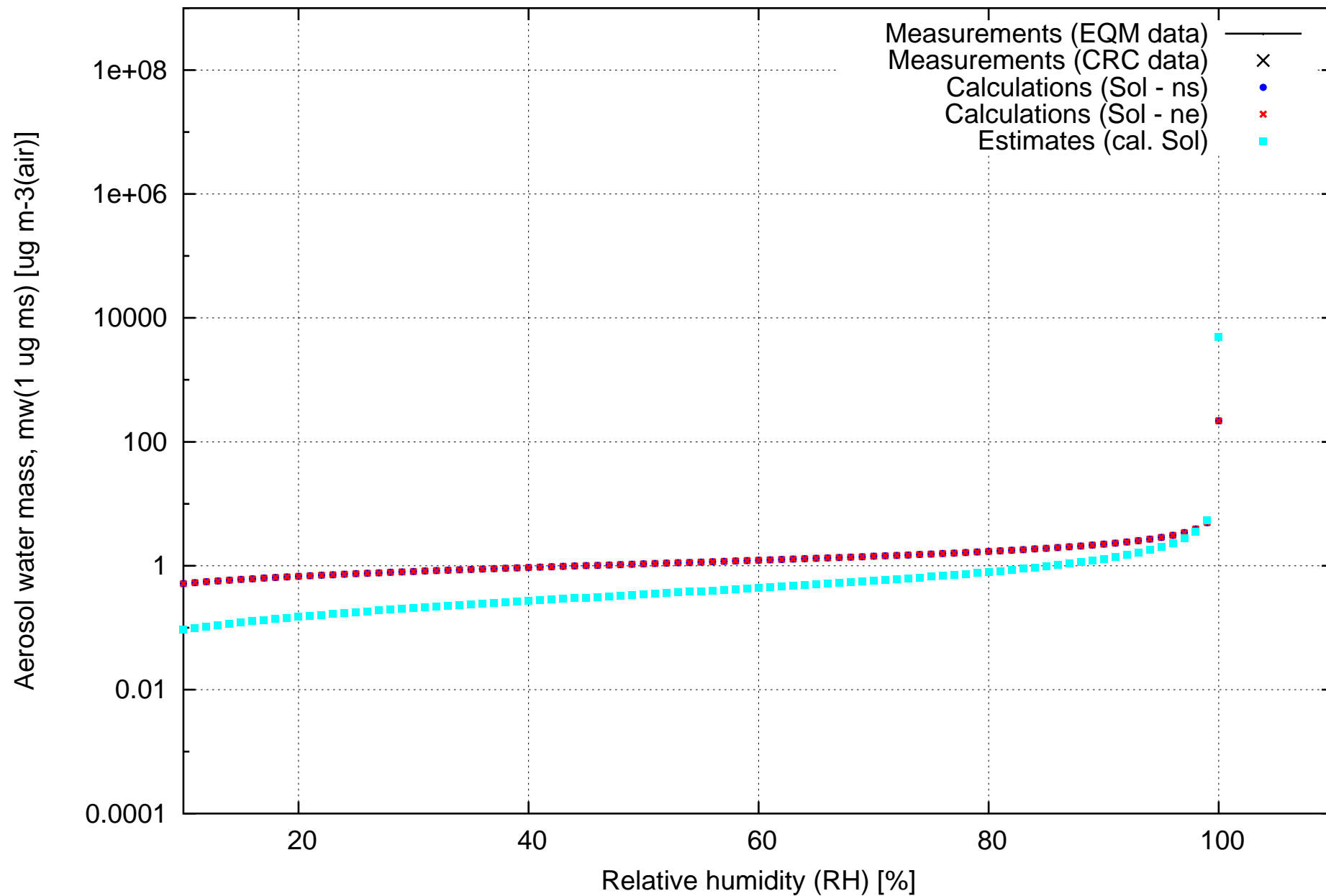
Calcium oxalate - CaC₂O₄



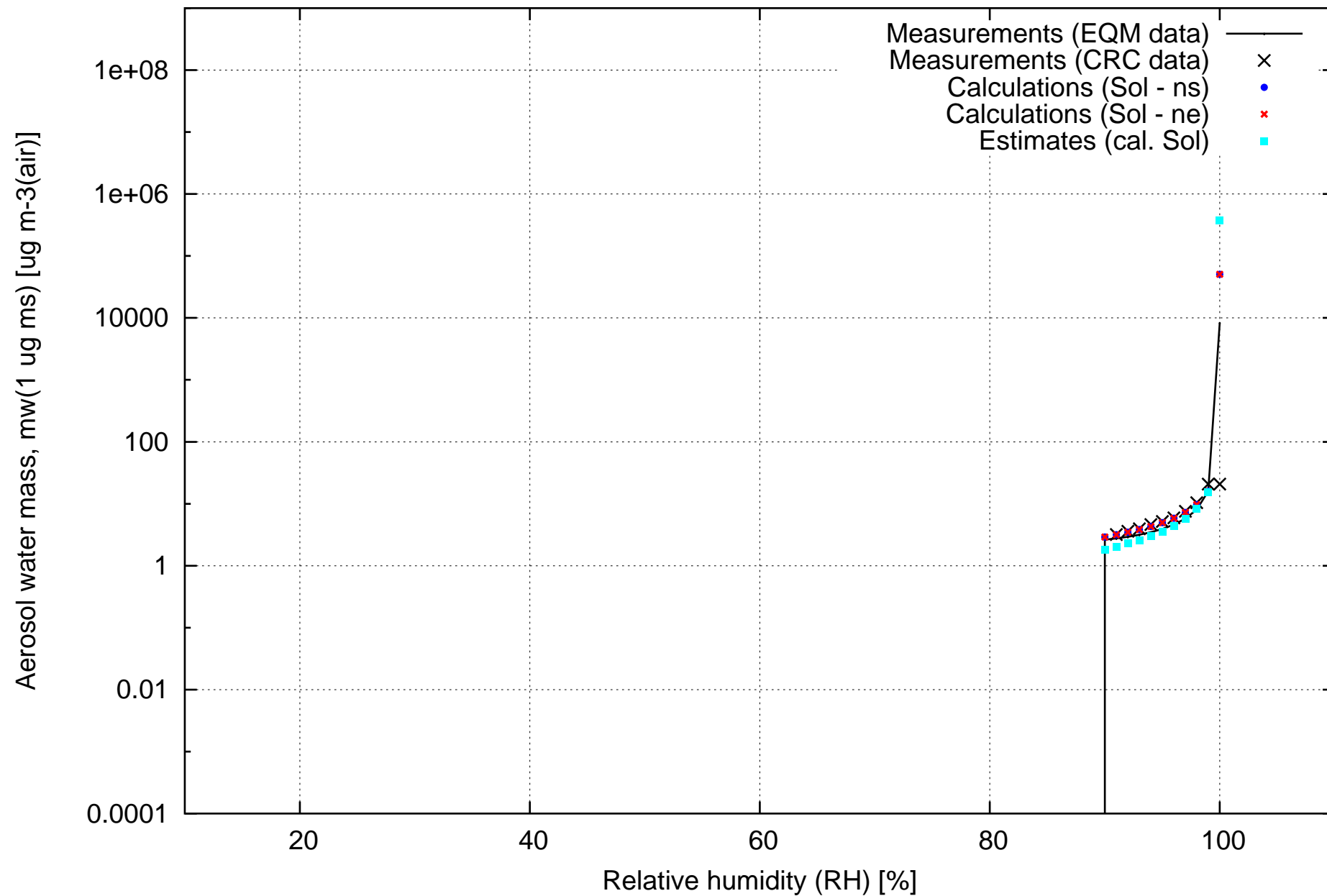
Calcium citrate - $\text{Ca}_3(\text{C}_6\text{H}_5\text{O}_7)_2$



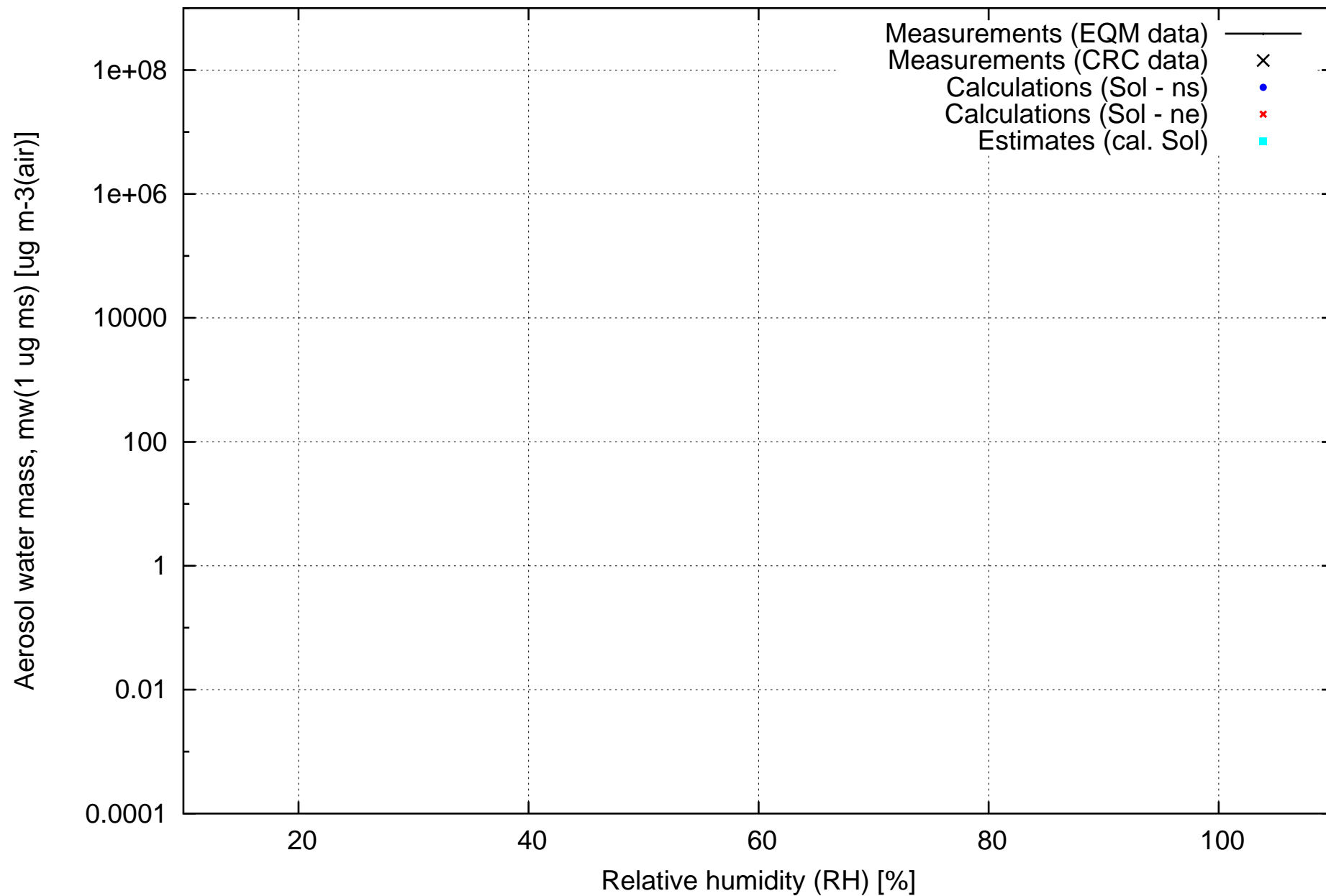
Magnesium phosphate - $\text{Mg}_3(\text{PO}_4)_2$



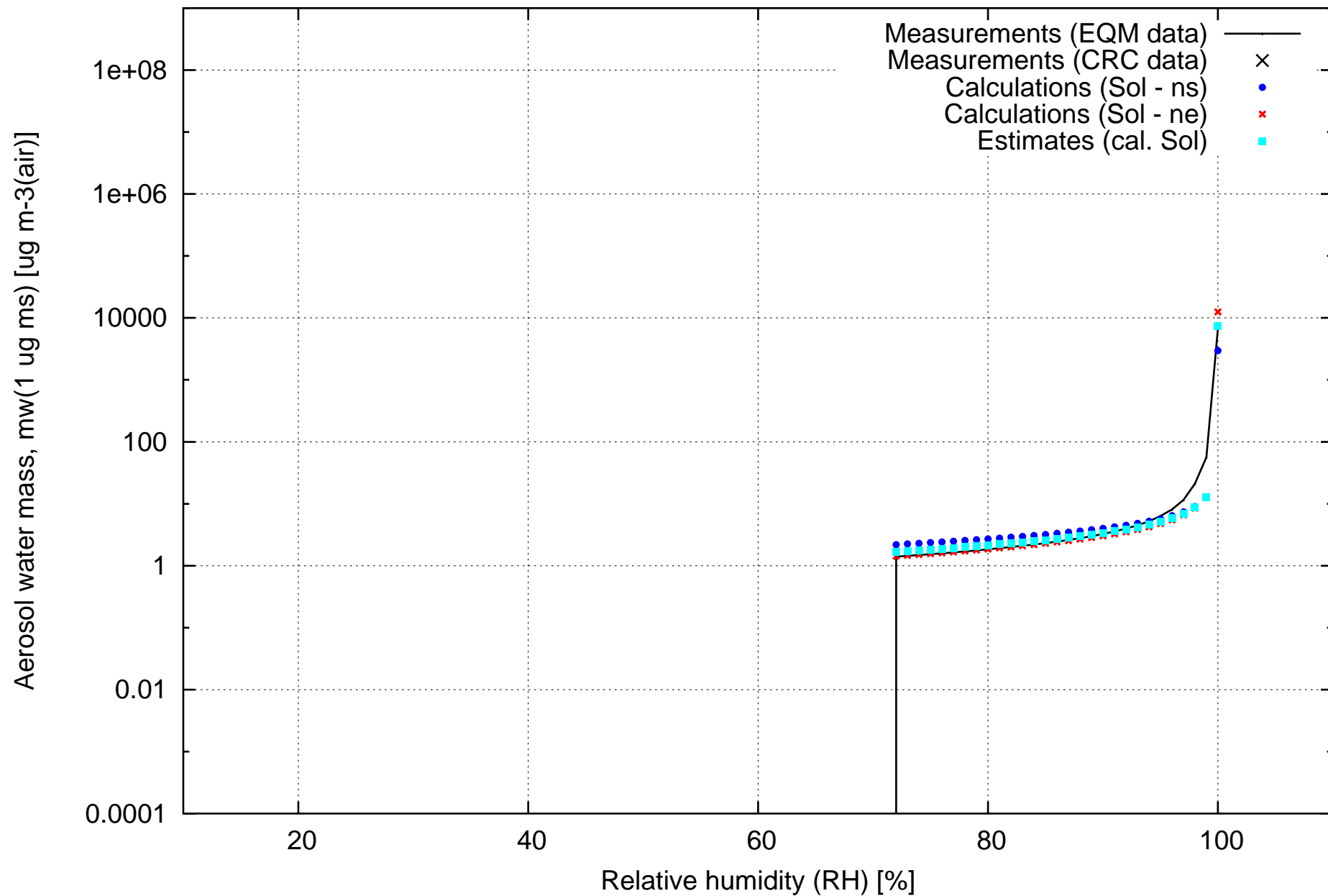
Magnesium sulfate - MgSO₄



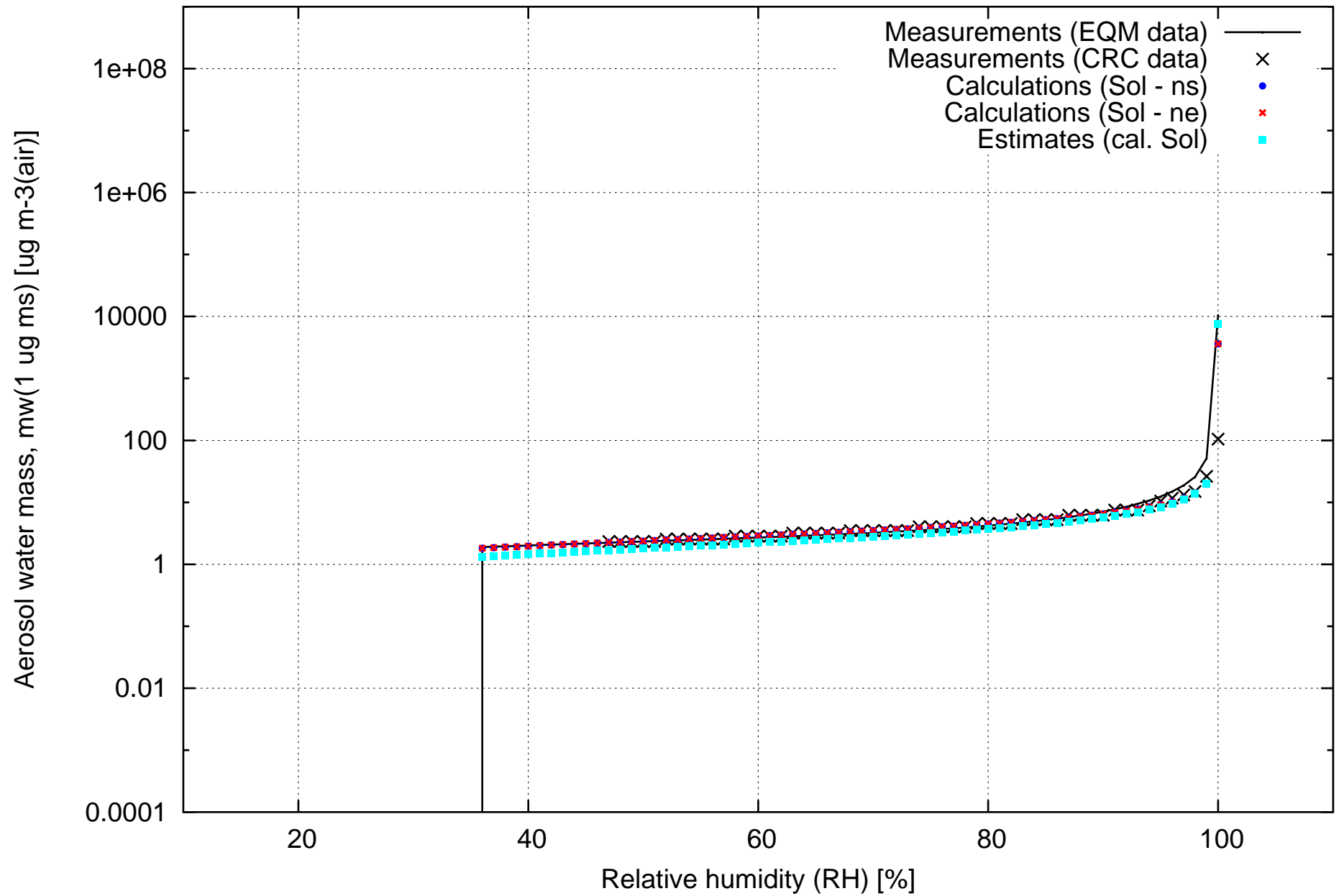
Magnesium - dummy 03



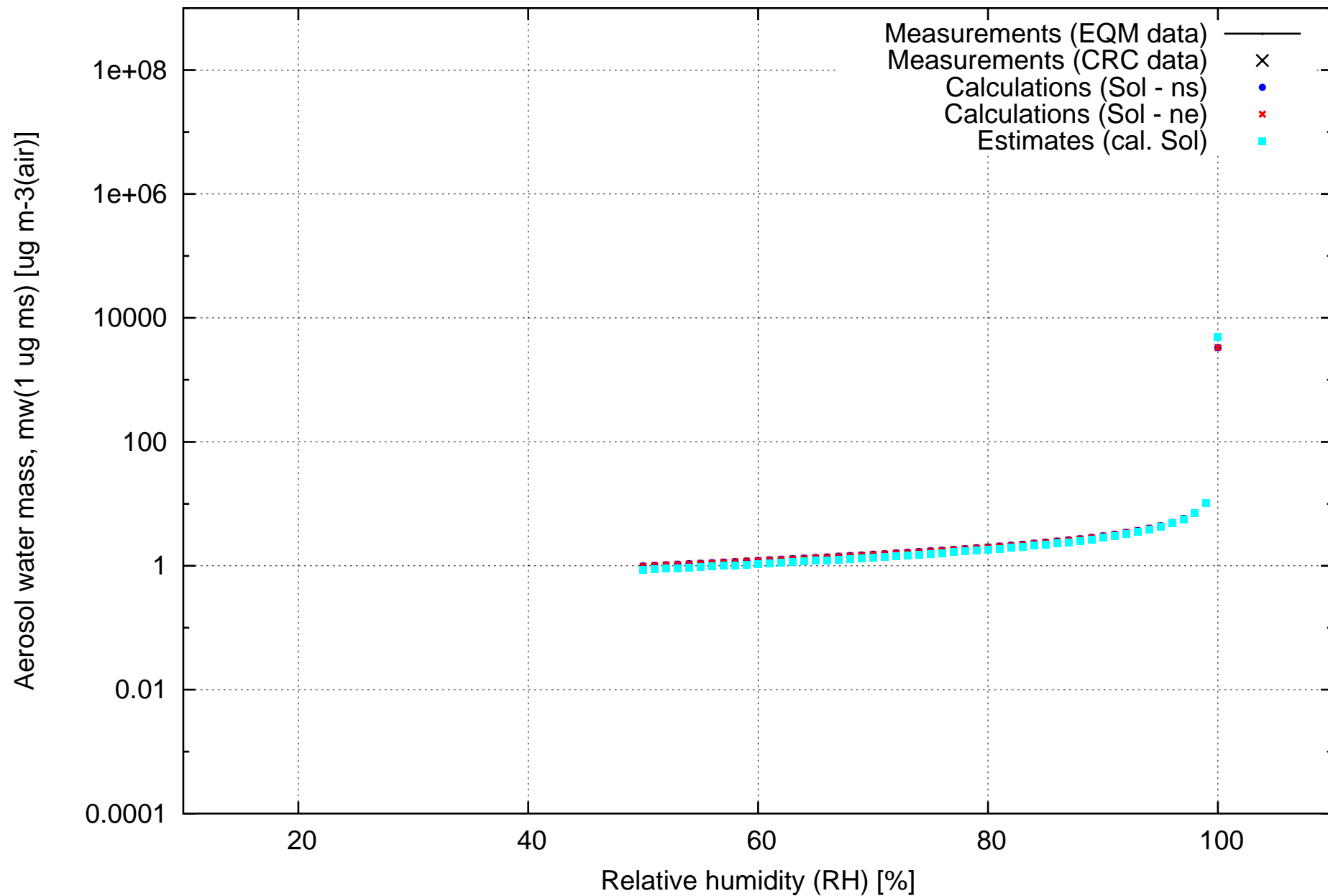
Magnesium nitrate - $\text{Mg}(\text{NO}_3)_2$



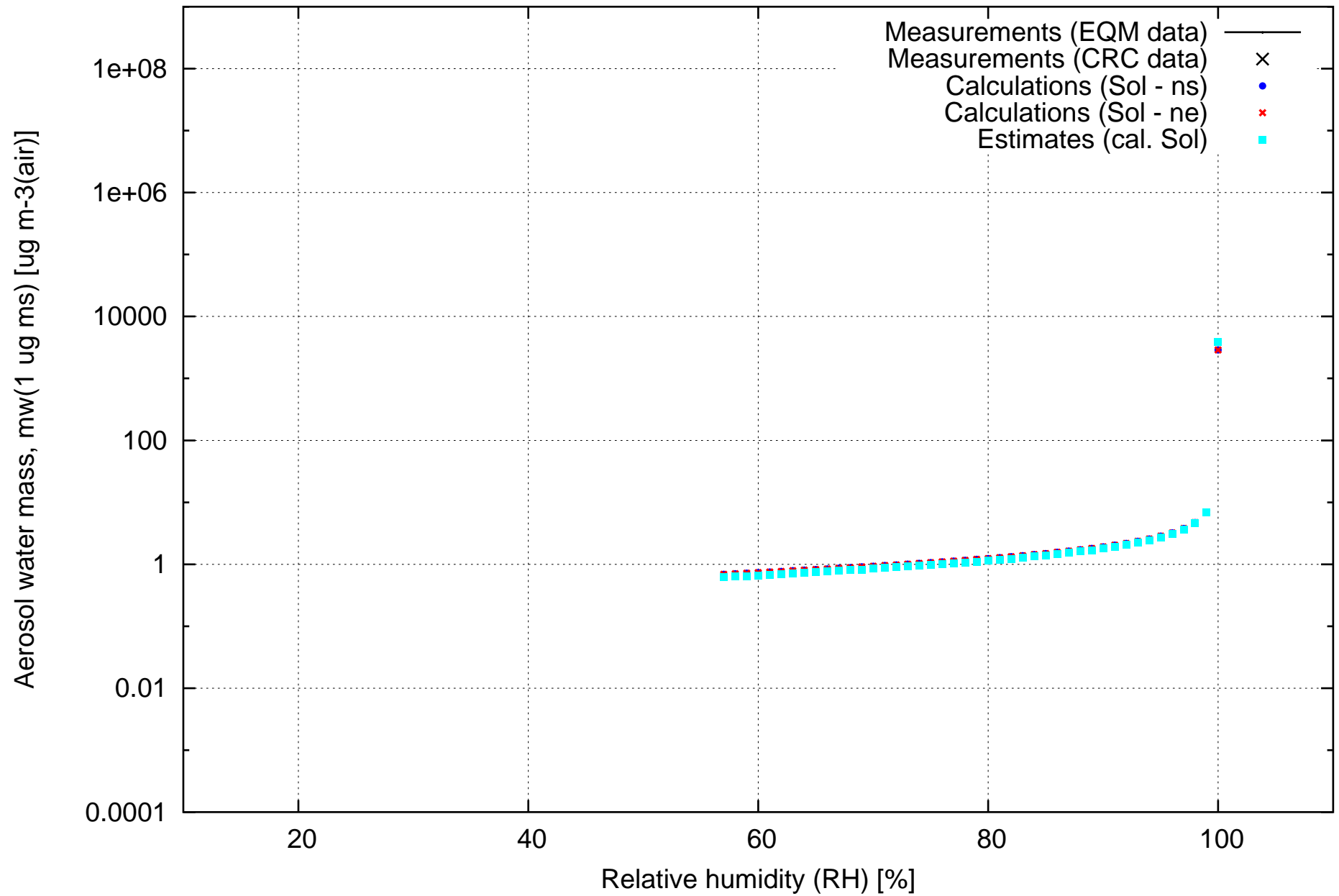
Magnesium chloride - MgCl_2



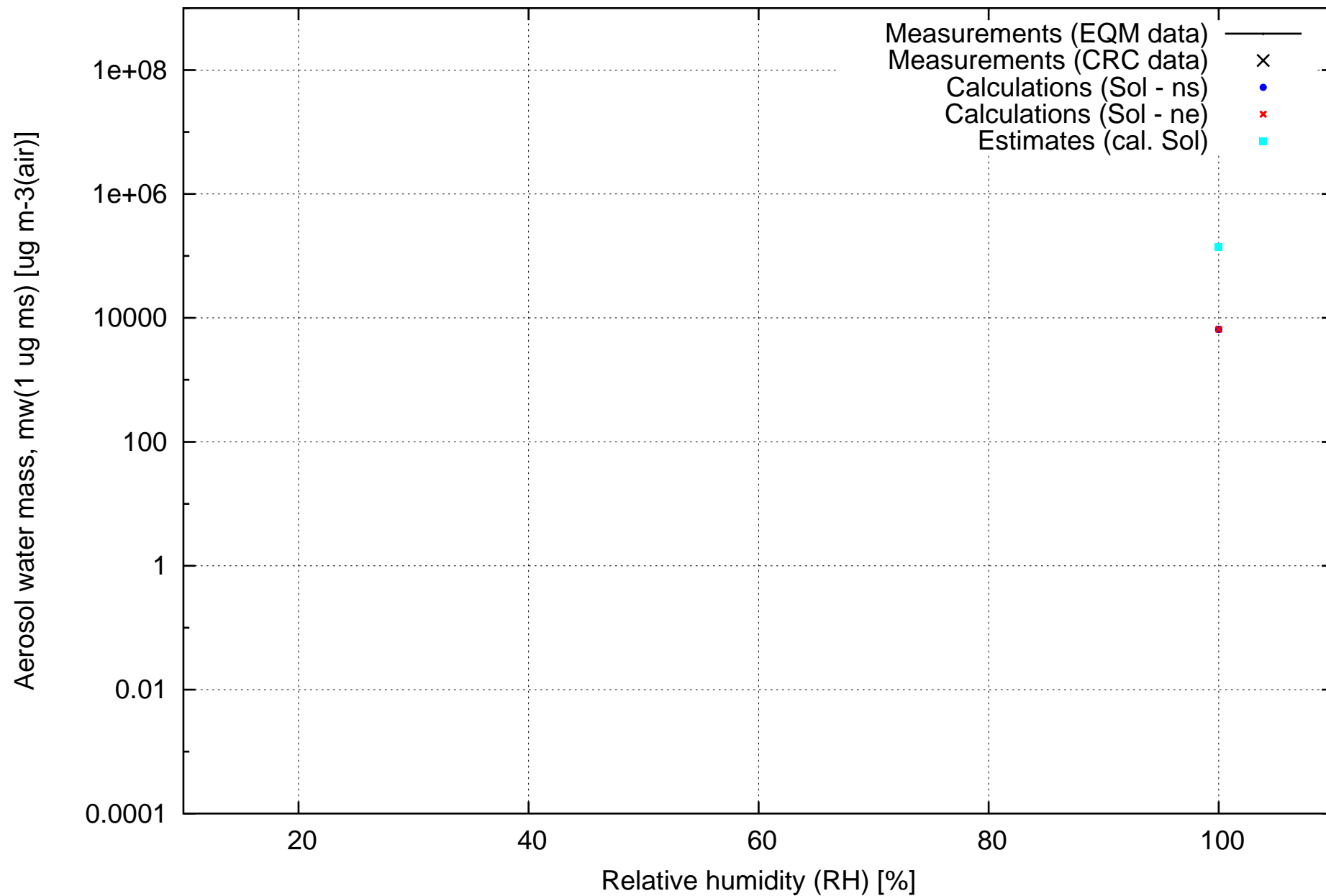
Magnesium bromide - MgBr₂



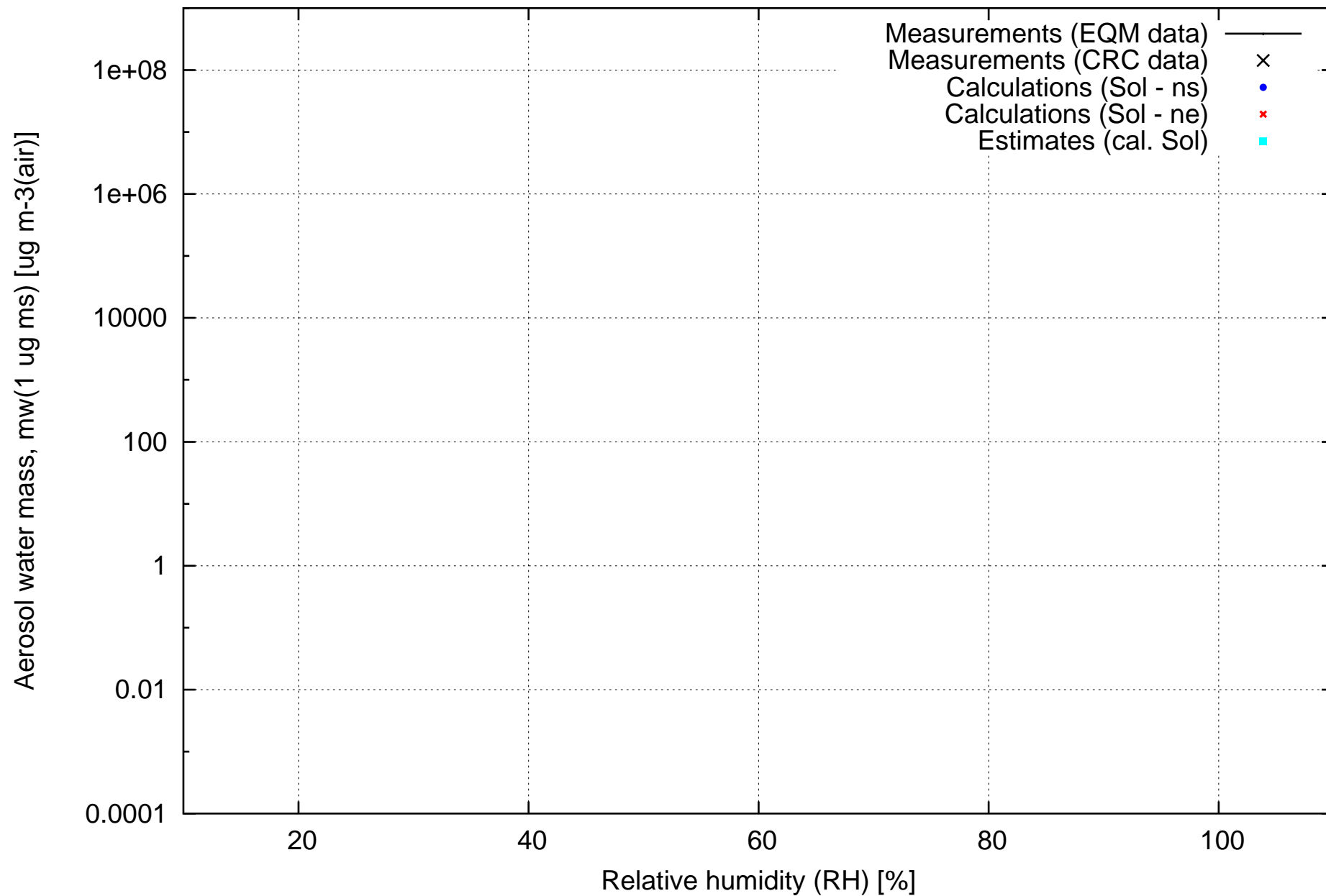
Magnesium iodide - MgI2



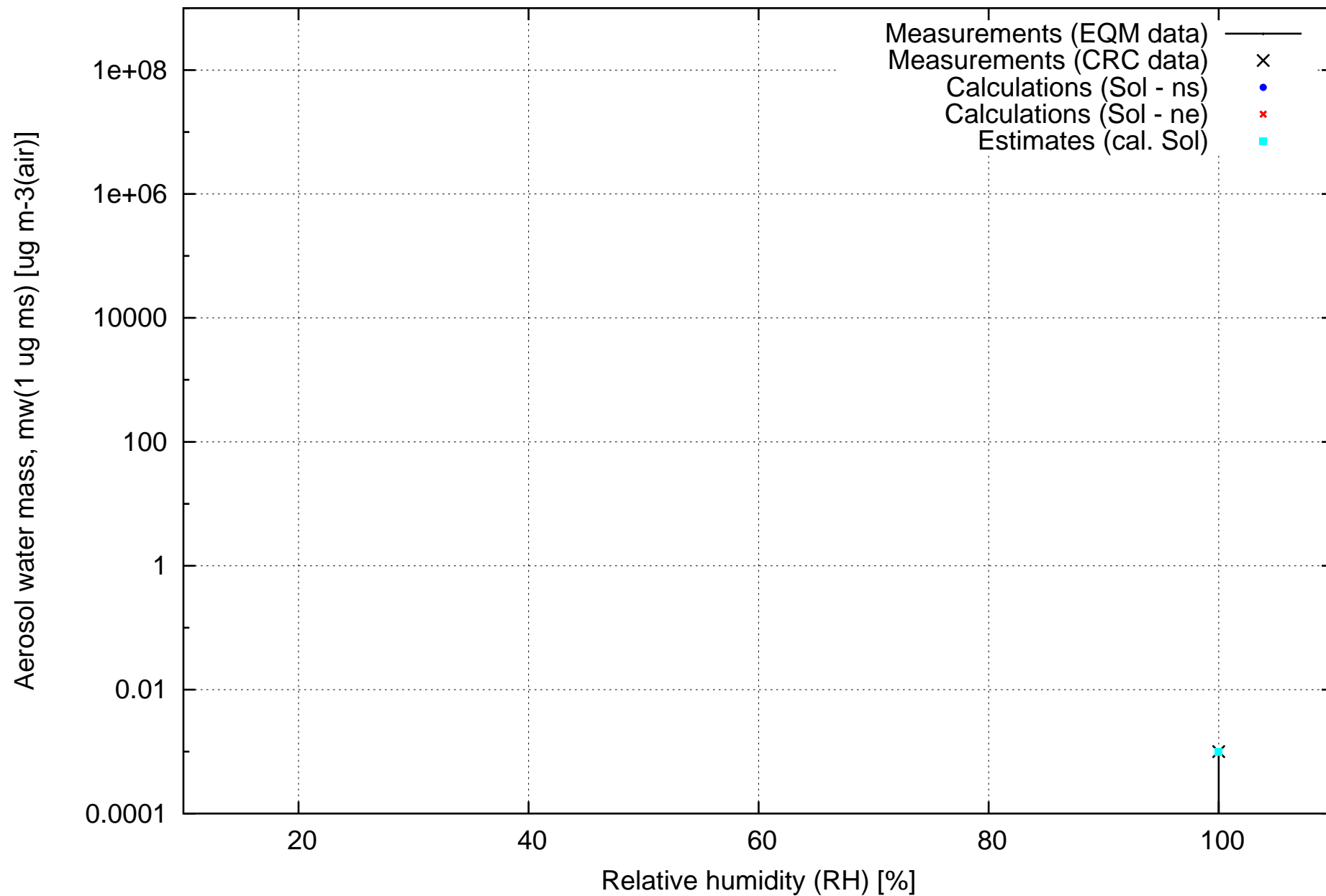
Magnesium carbonate - MgCO_3



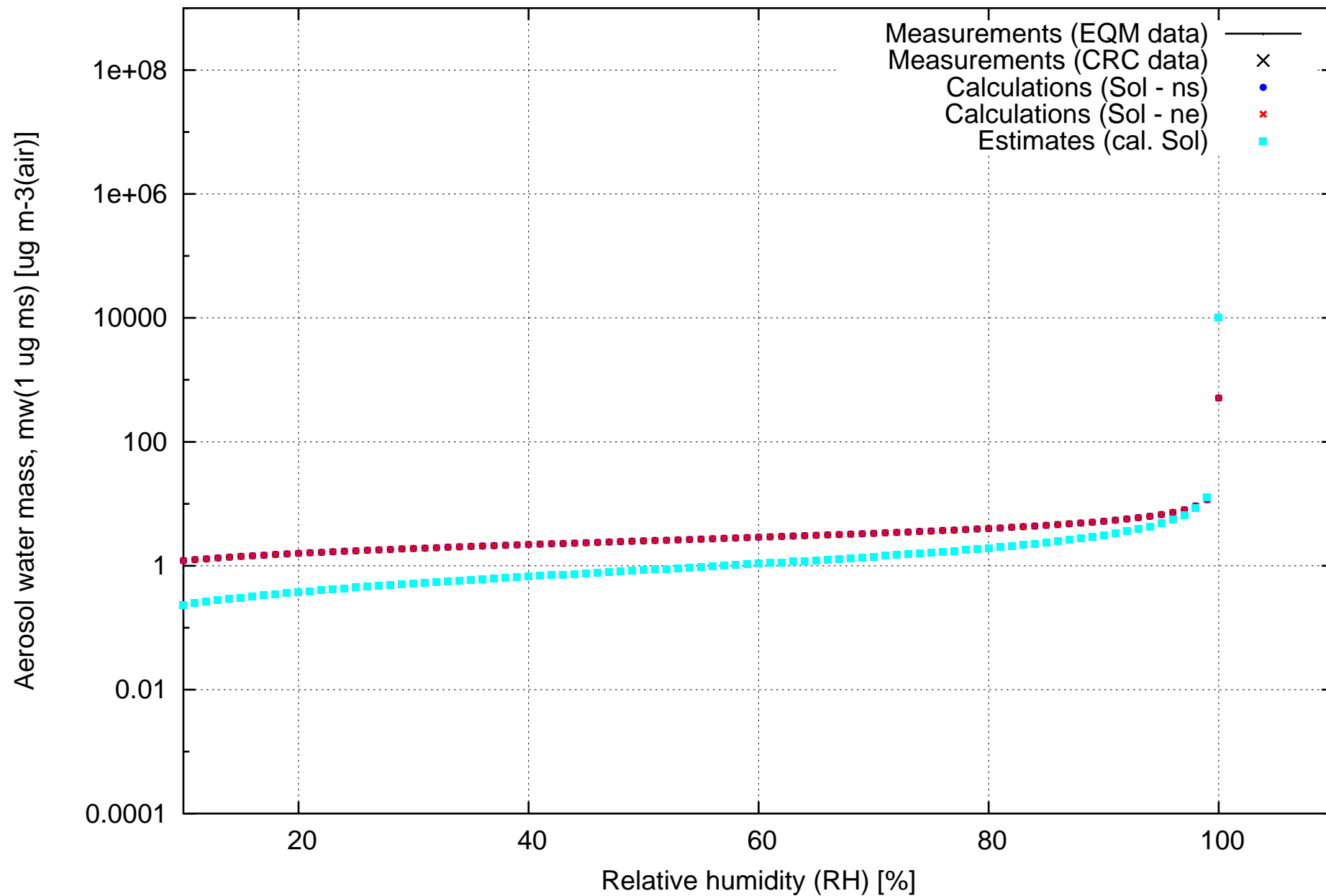
Magnesium - dummy 09



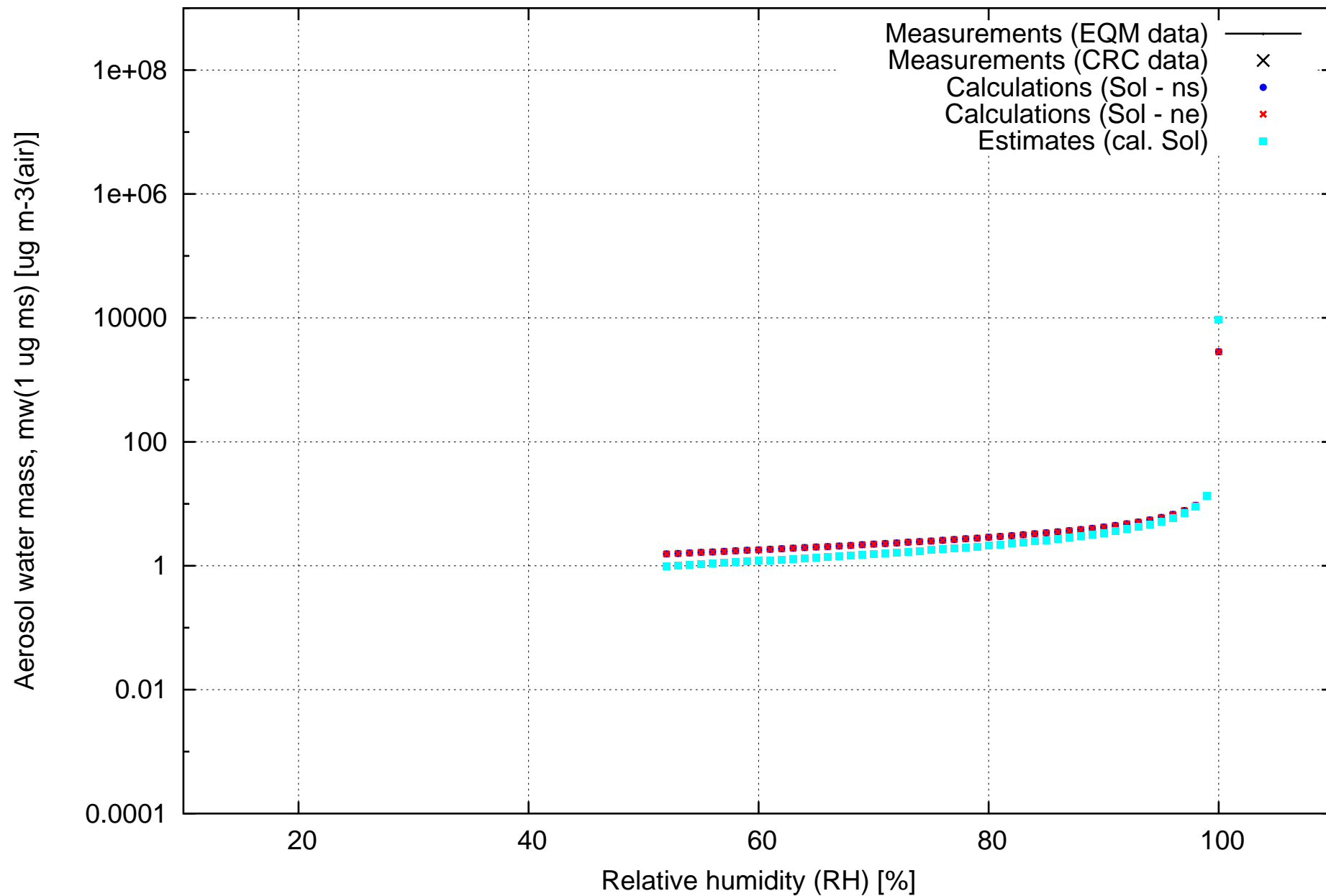
Magnesium hydroxide - Mg(OH)₂



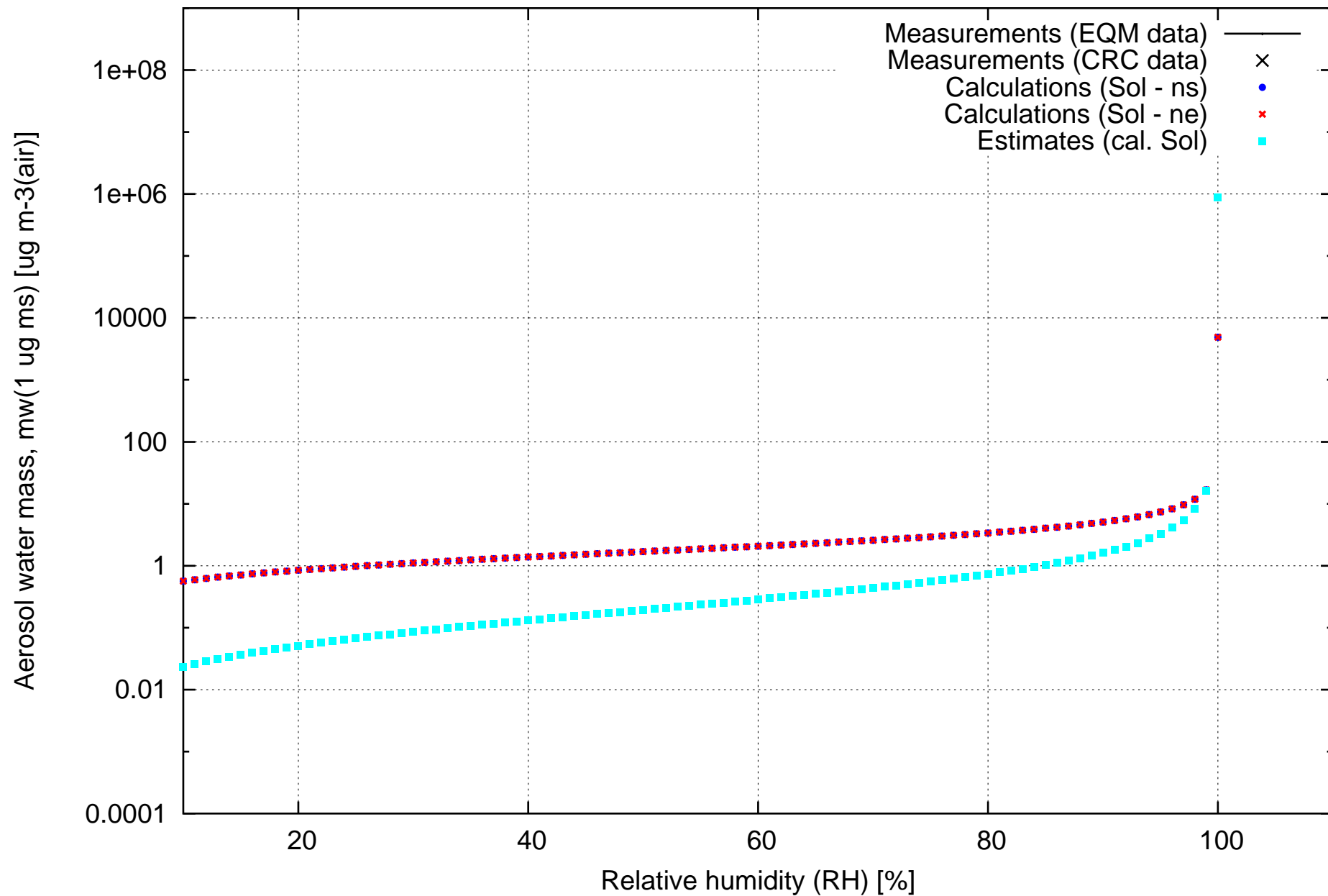
Magnesium formate - $\text{Mg}(\text{CHO}_2)_2$



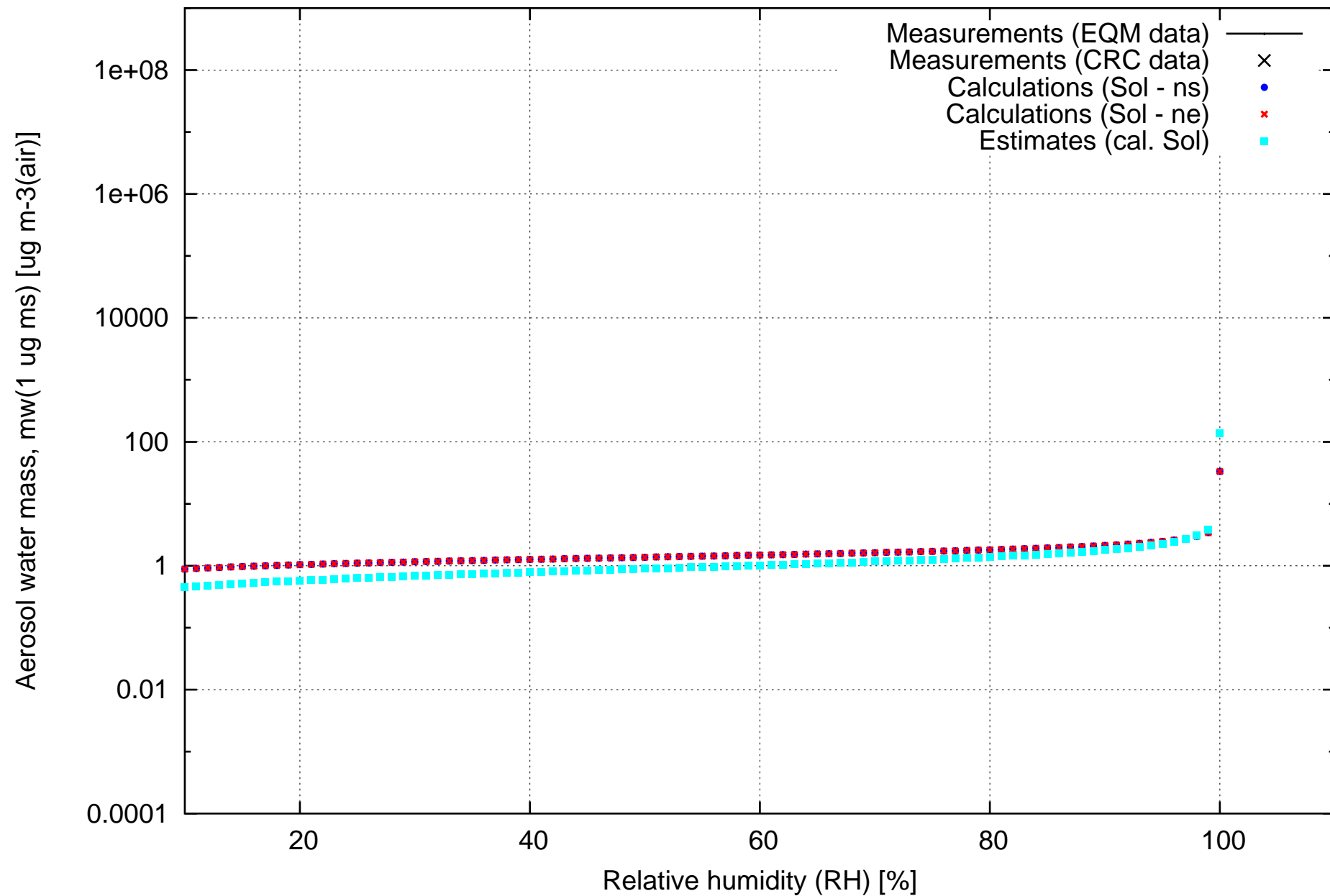
Magnesium acetate - $\text{Mg}(\text{C}_2\text{H}_3\text{O}_2)_2$



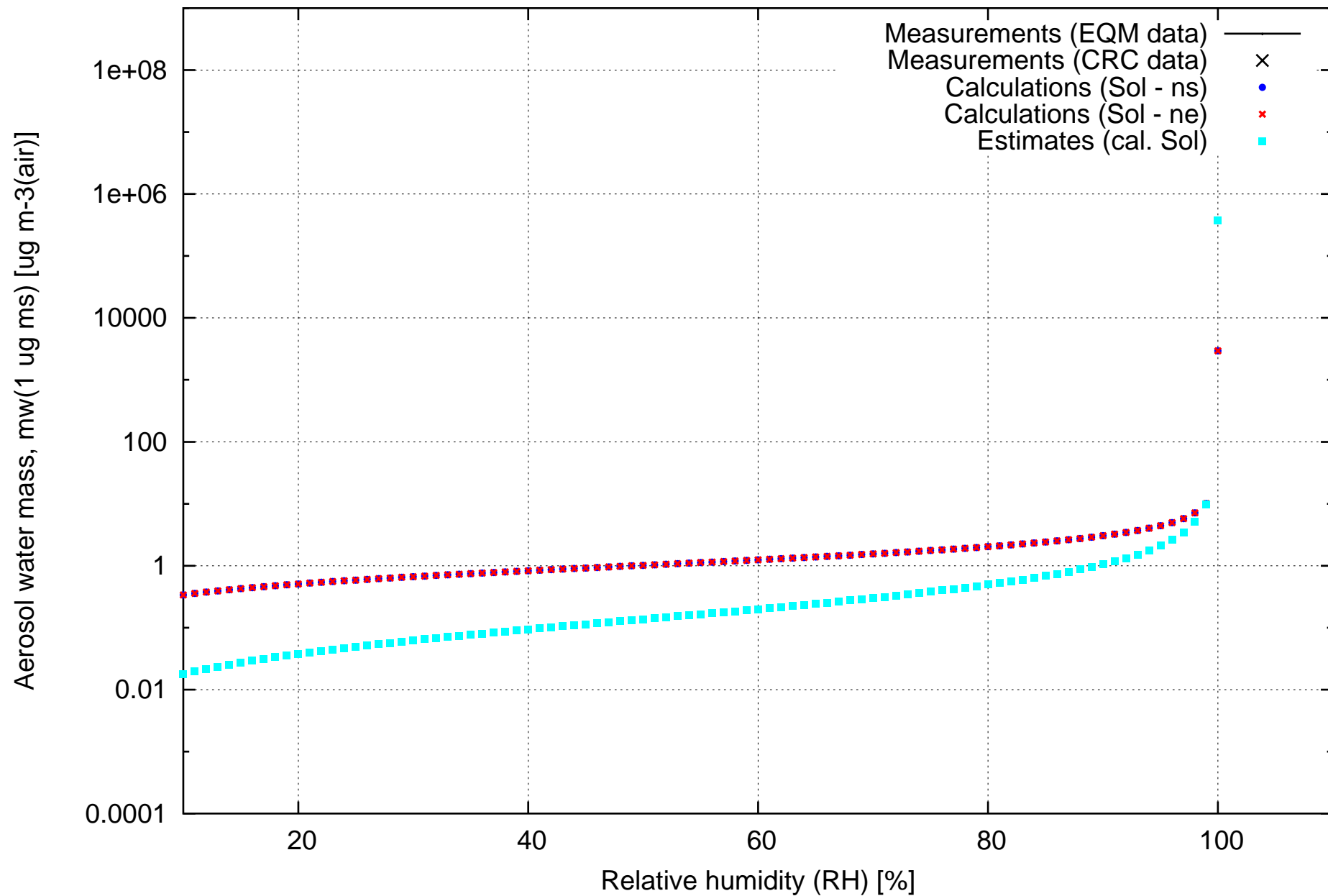
Magnesium oxalate - MgC_2O_4



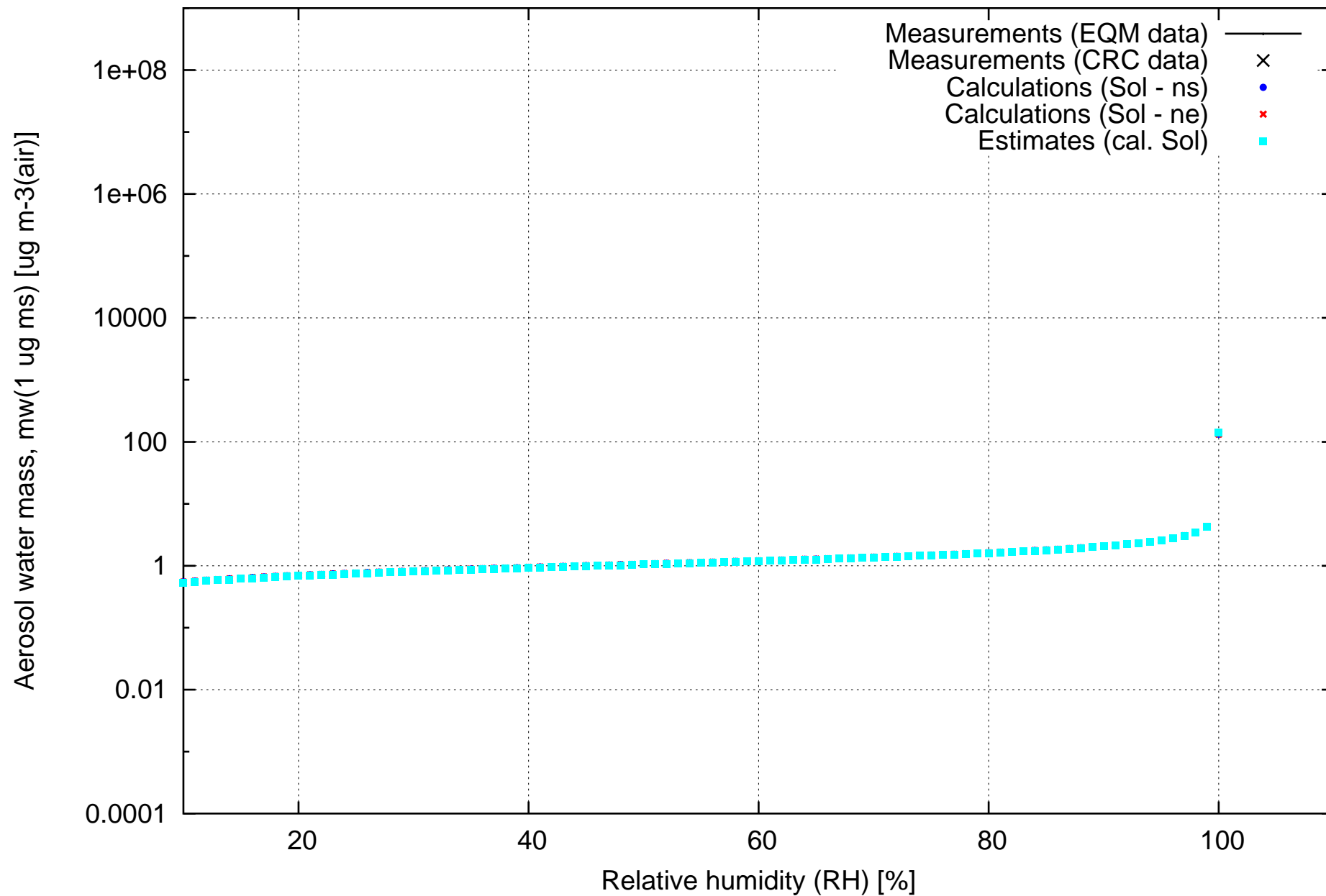
Magnesium citrate - $\text{Mg}_3(\text{C}_6\text{H}_5\text{O}_7)_2$



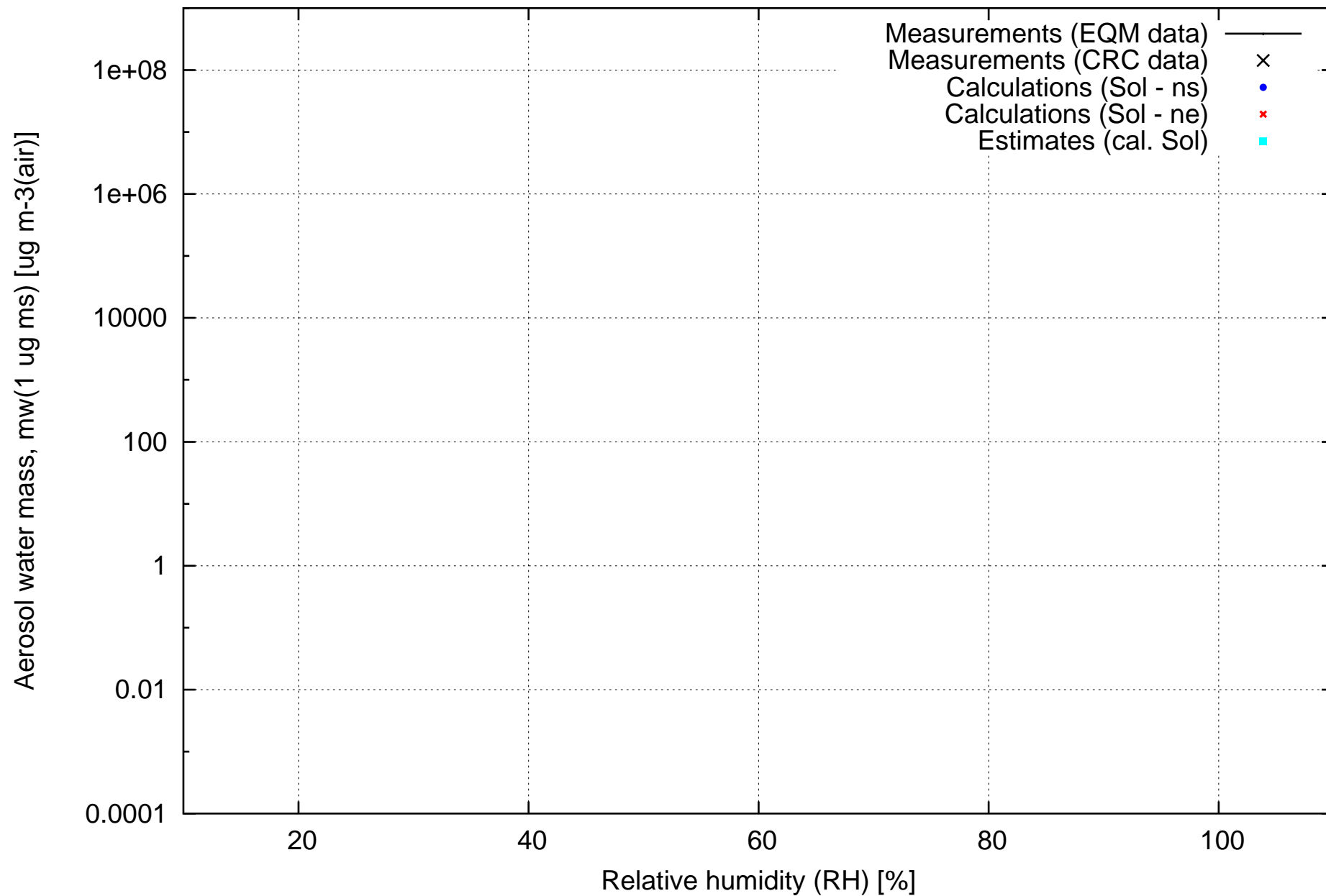
Iron(III) phosphate - FePO4



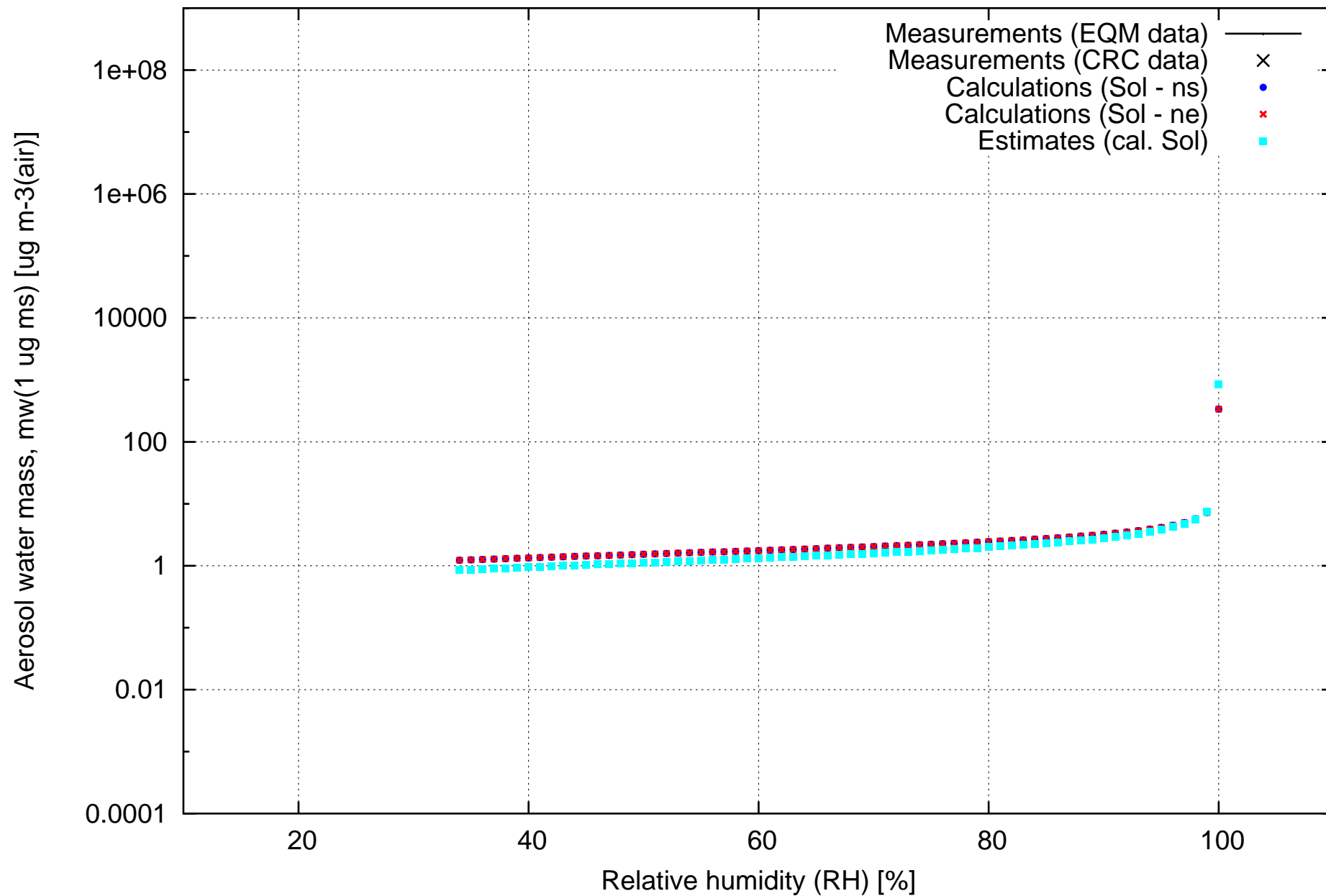
Iron(III) sulfate - $\text{Fe}_2(\text{SO}_4)_3$



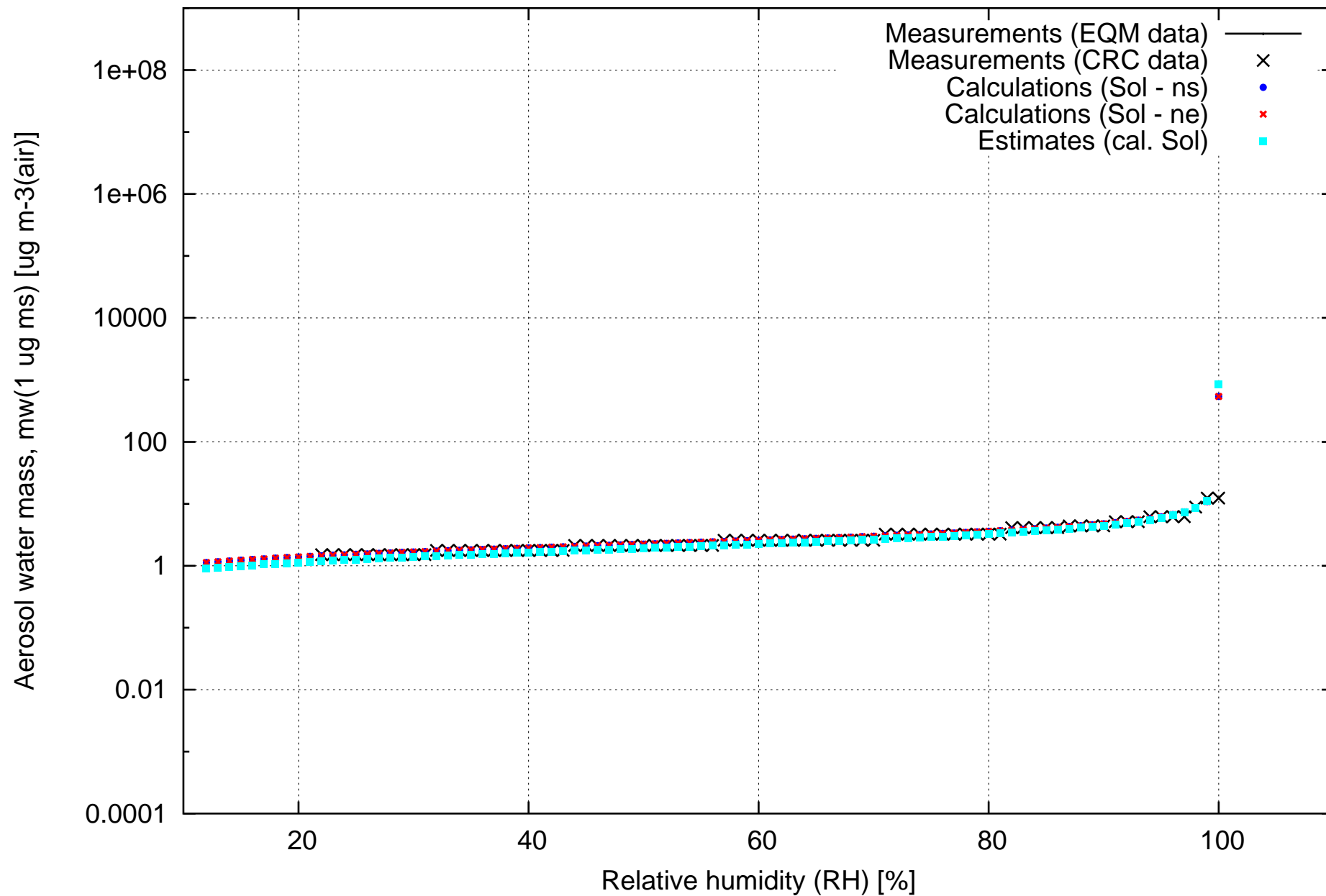
Iron(III) - dummy 03



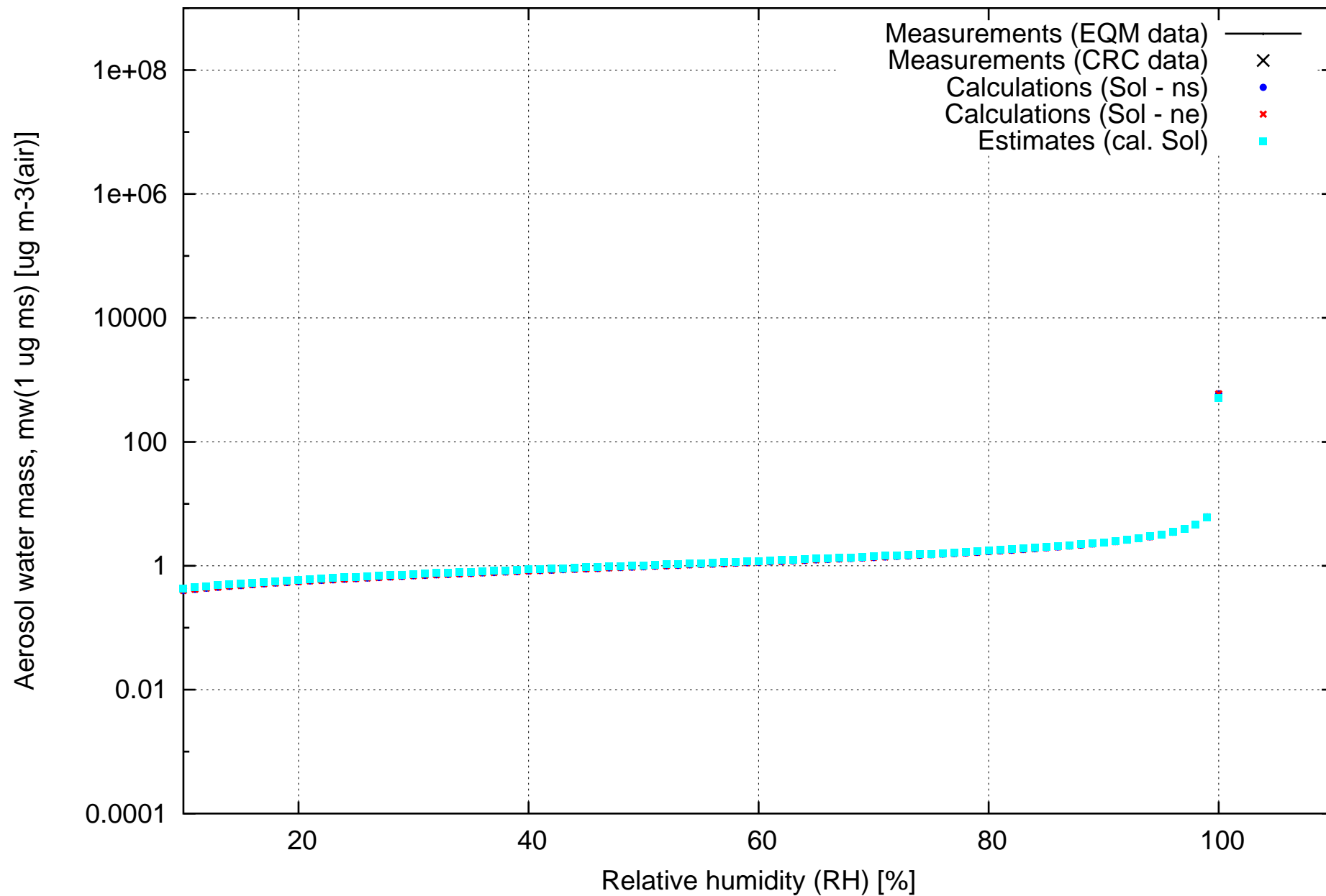
Iron(III) nitrate - Fe(NO₃)₃



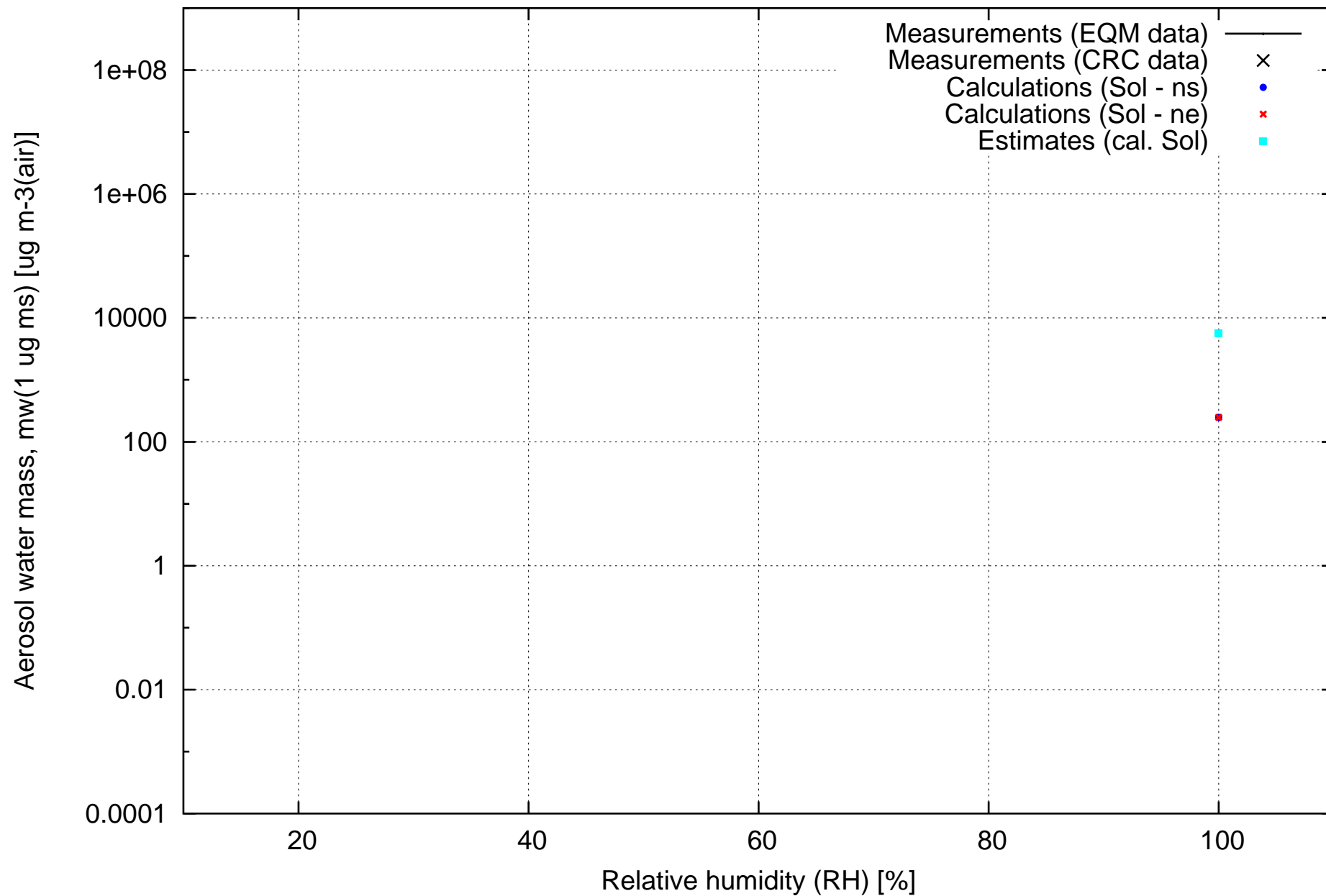
Iron(III) chloride - FeCl3



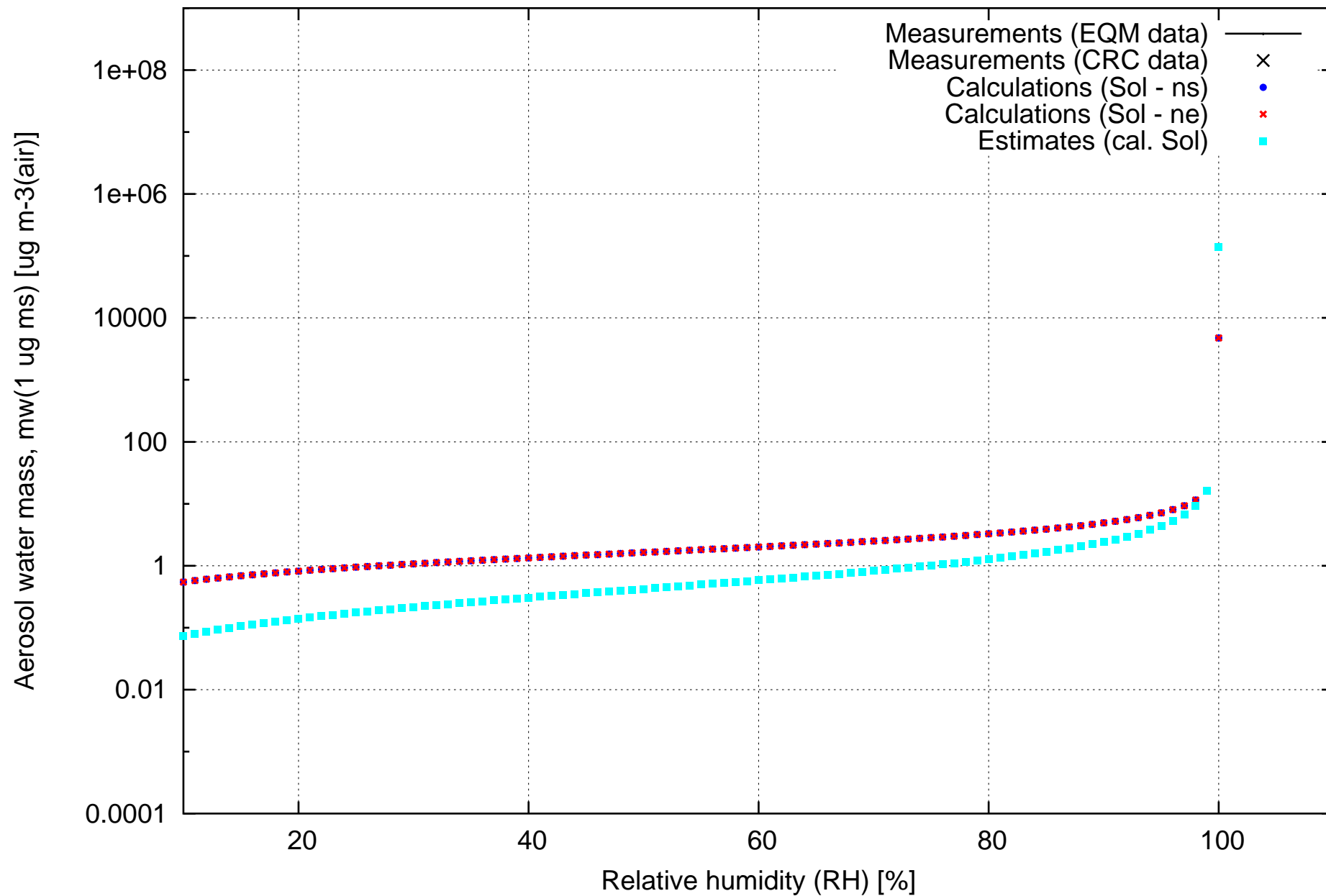
Iron(III) bromide - FeBr3



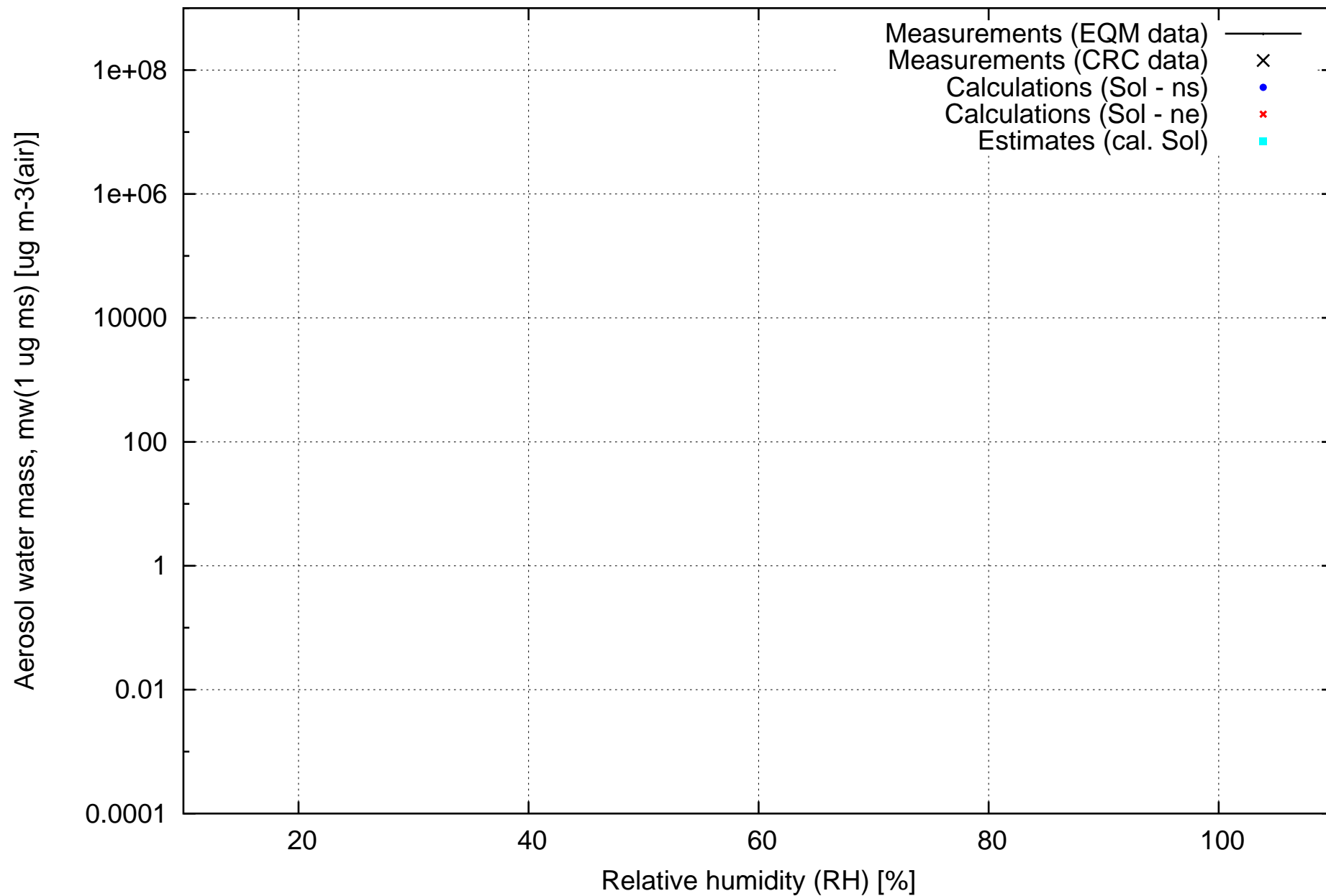
Iron(III) iodide - FeI3



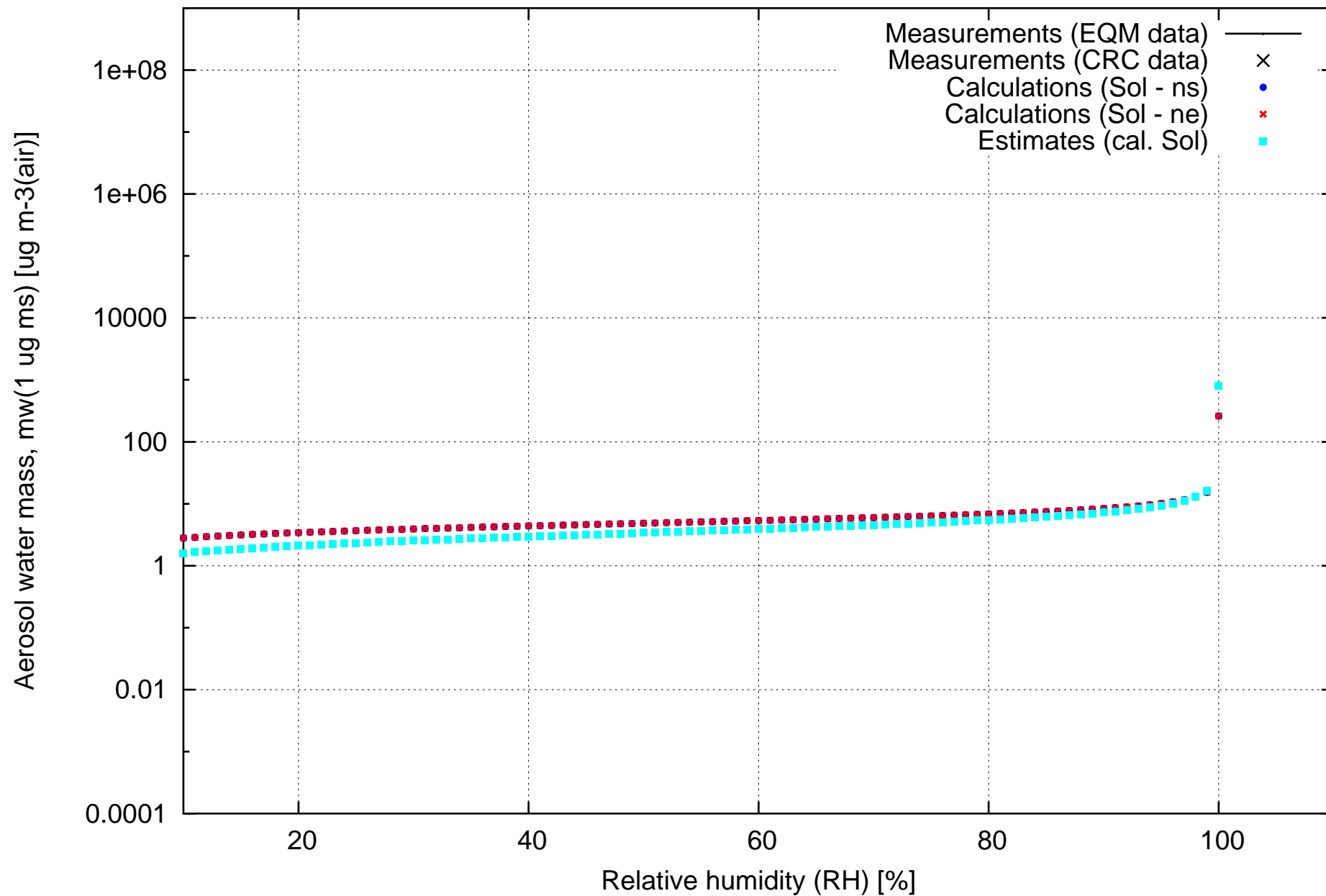
Iron(II) carbonate - FeCO₃



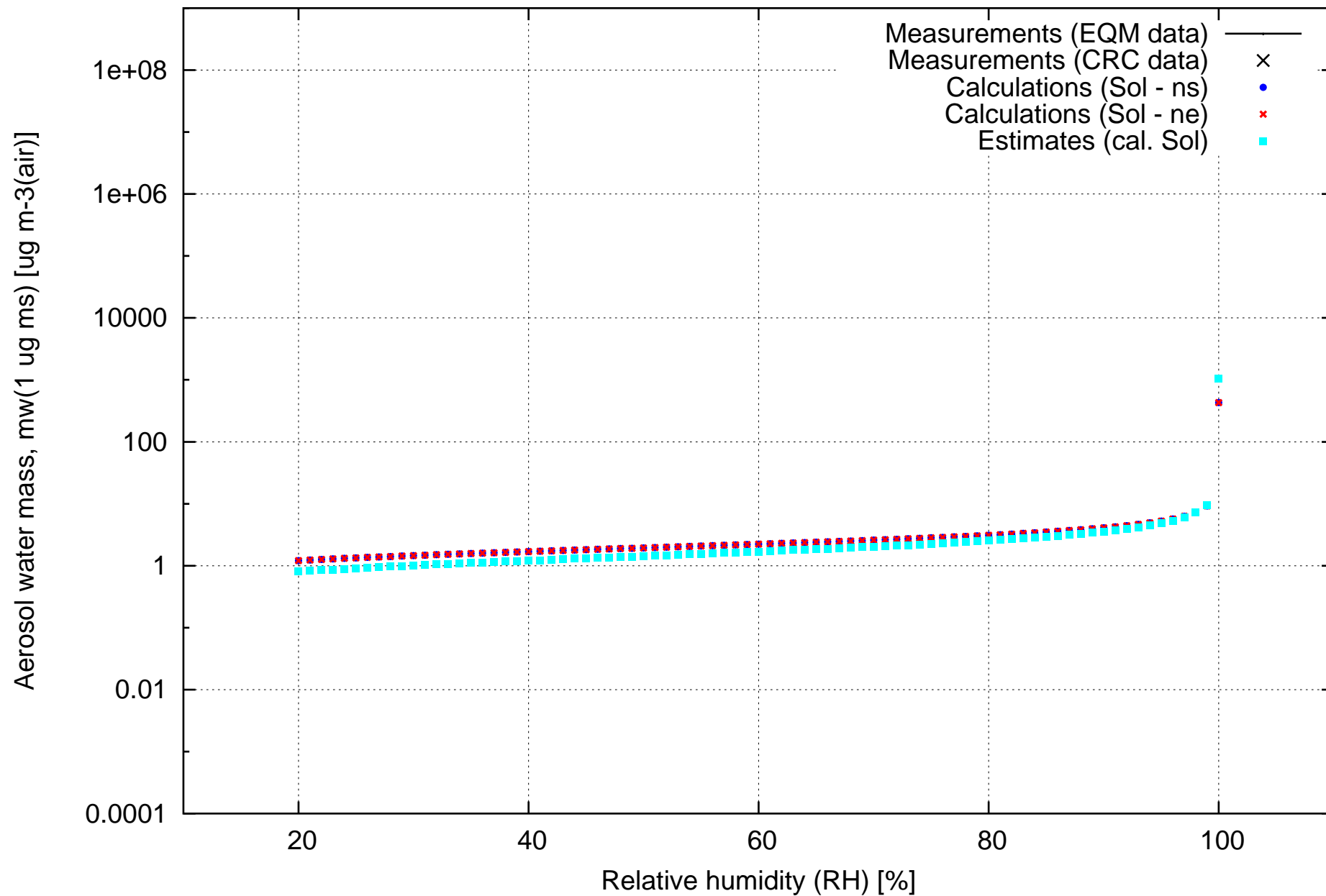
Iron(III) - dummy



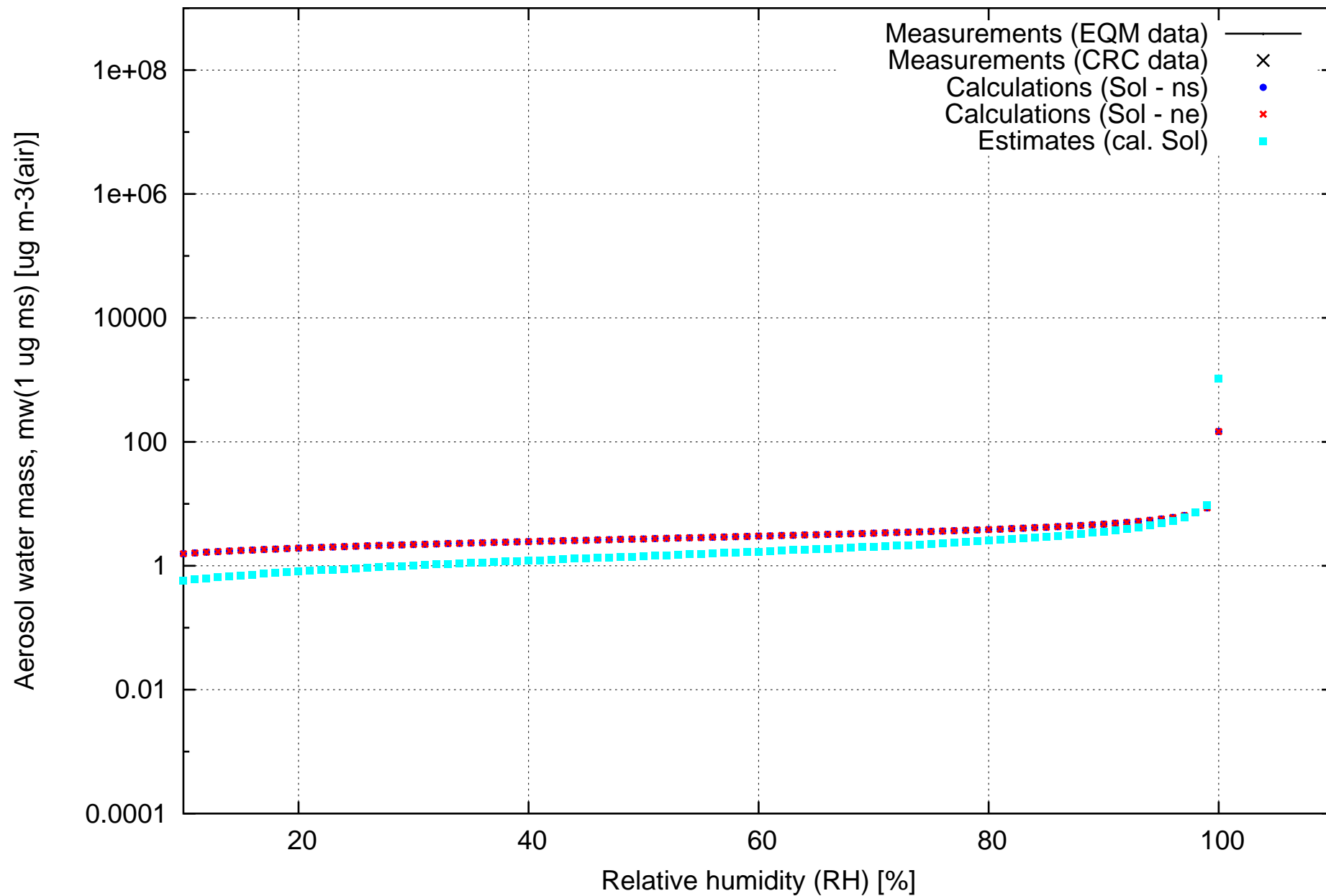
Iron(III) hydroxide - Fe(OH)₃



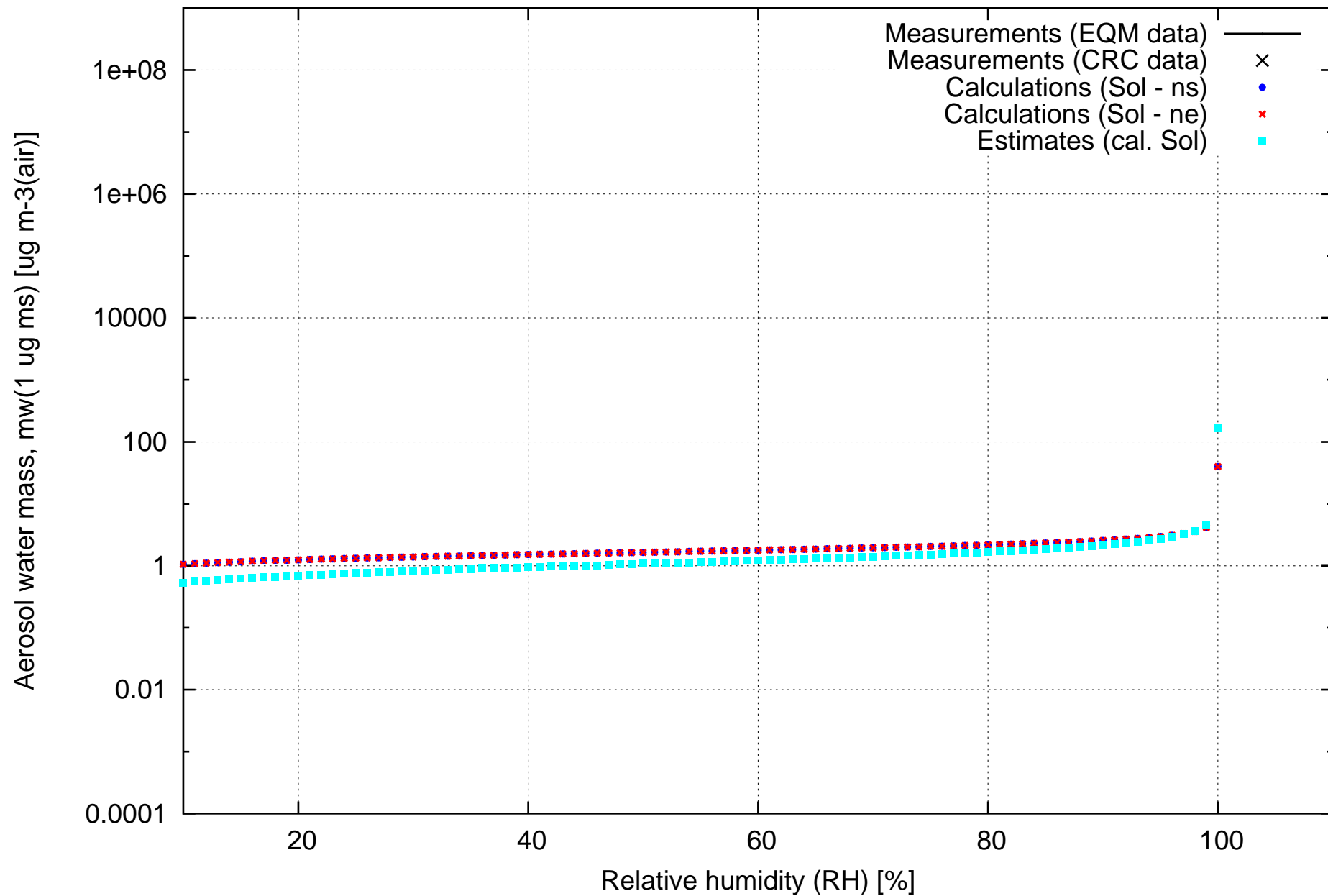
Iron(III) formate - Fe(CHO₂)₃



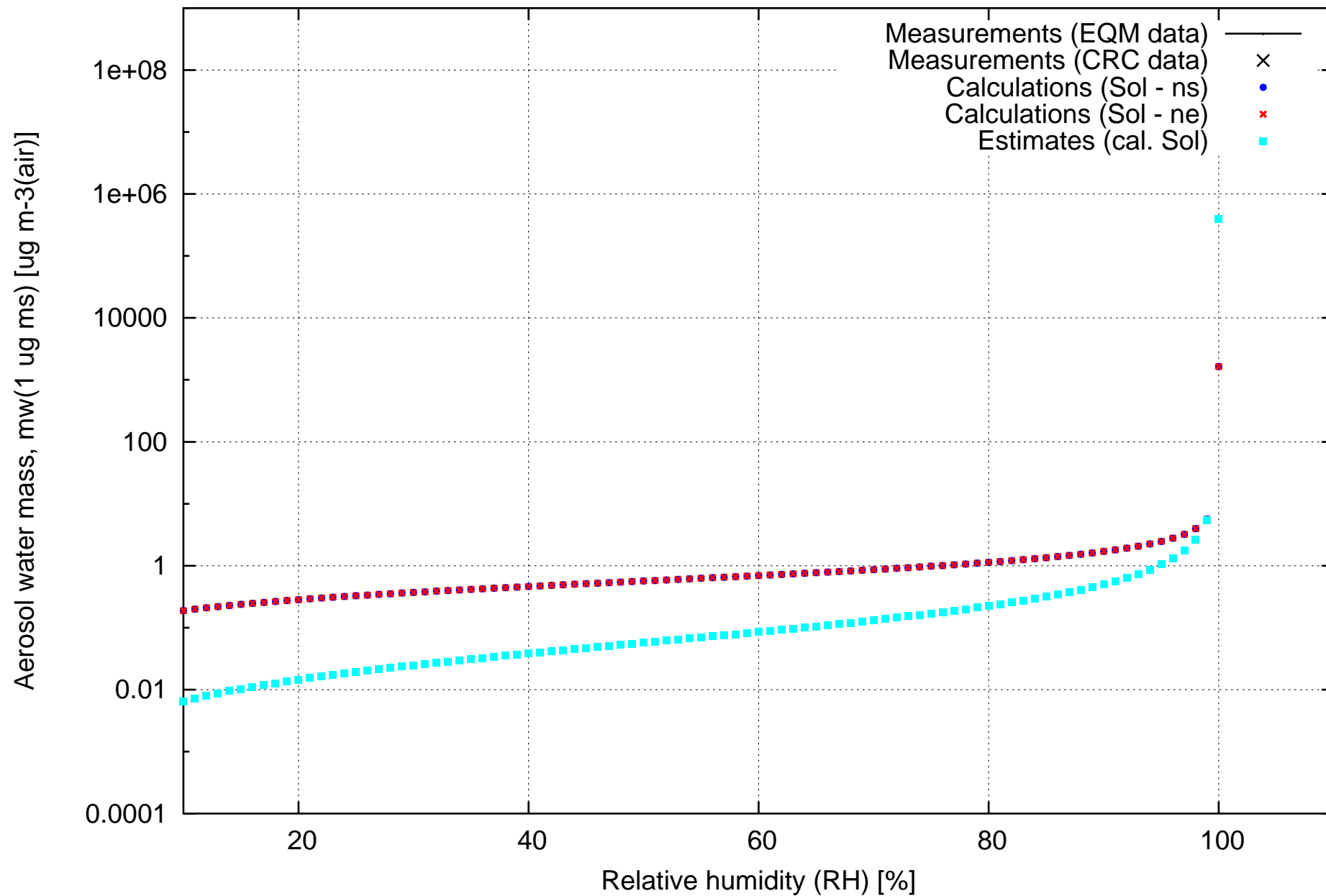
Iron(III) acetate - $\text{FeOH}(\text{C}_2\text{H}_3\text{O}_2)_2$



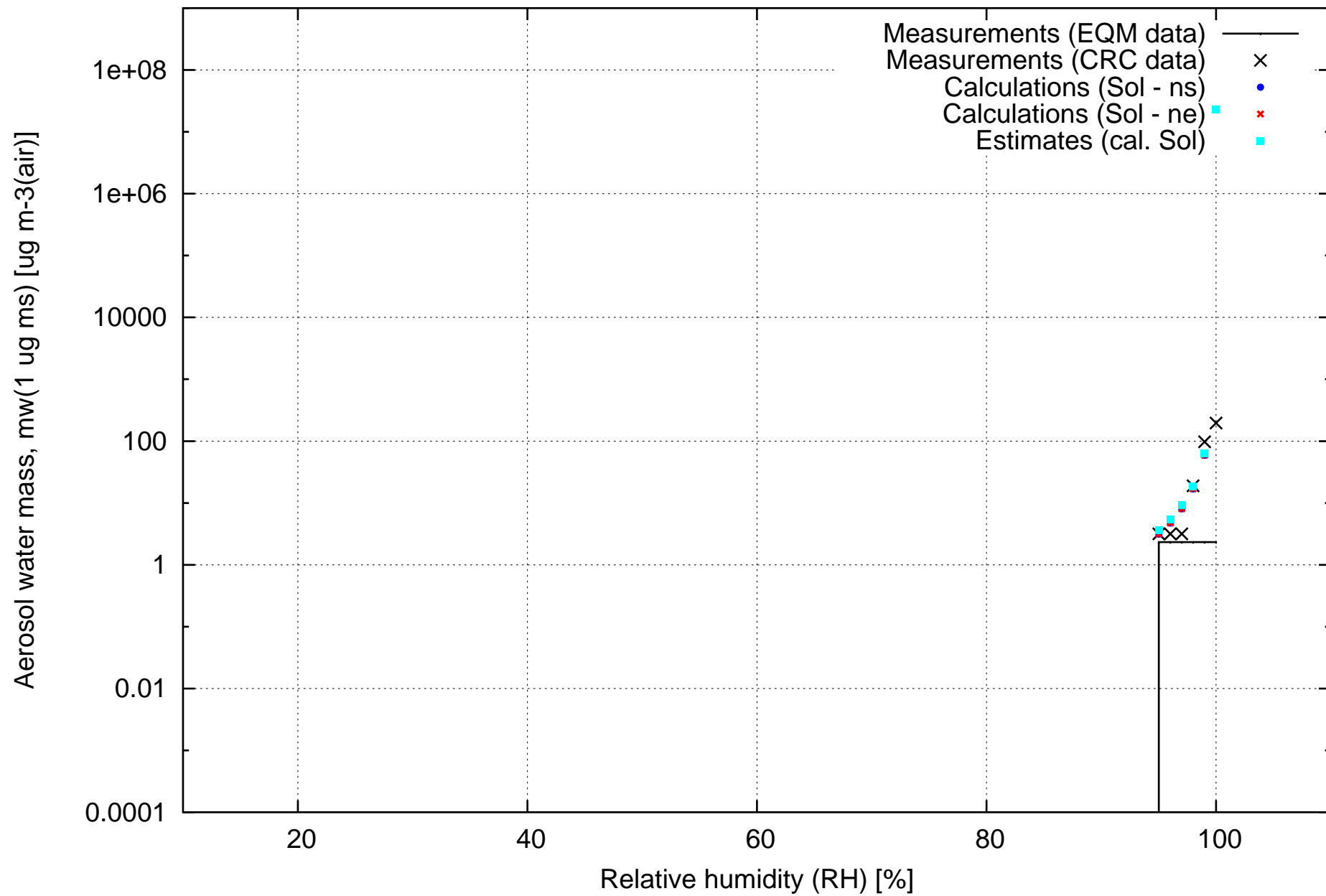
Iron(III) oxalate - $\text{Fe}_2(\text{C}_2\text{O}_4)_3$



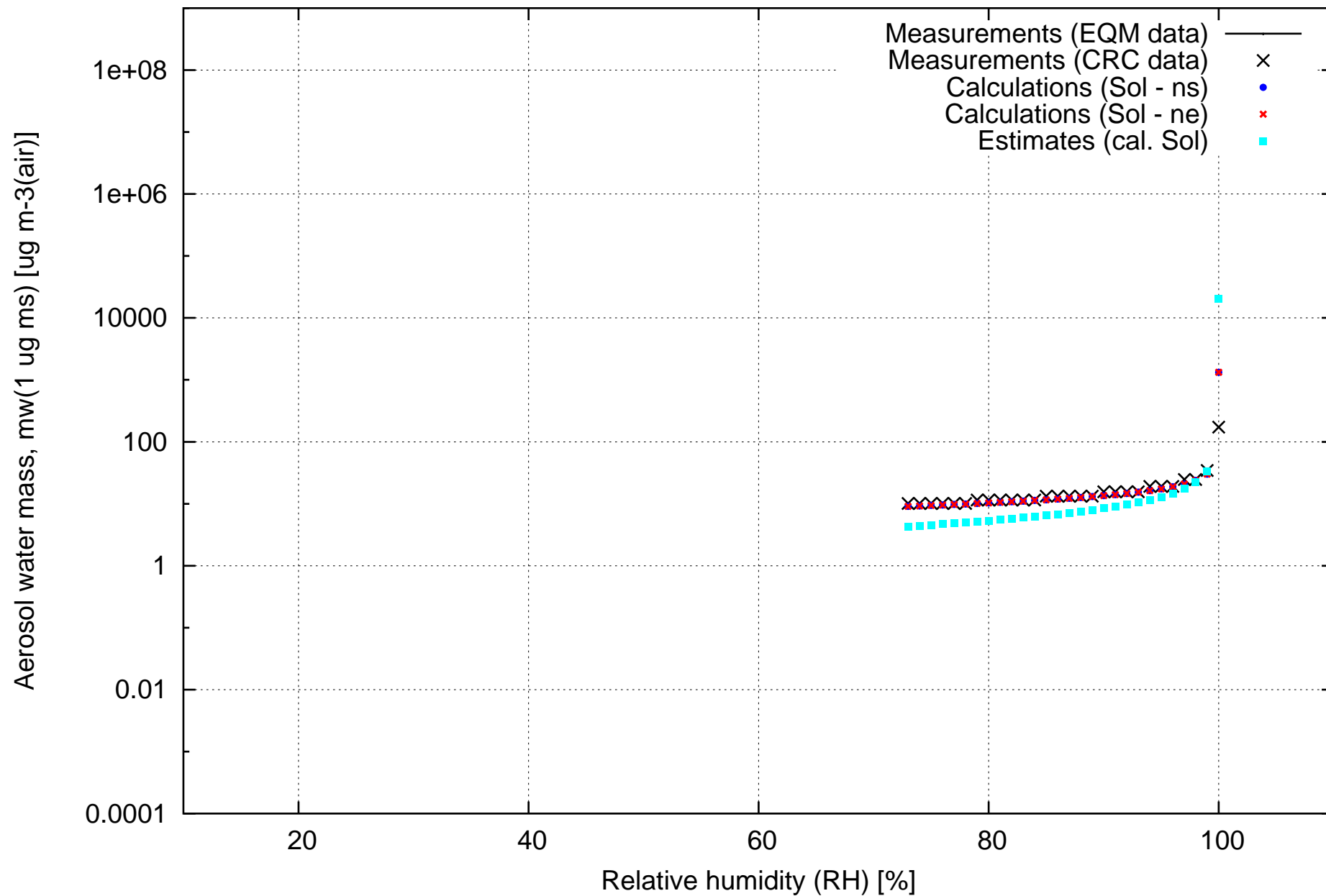
Iron(III) citrate - FeC6H5O7



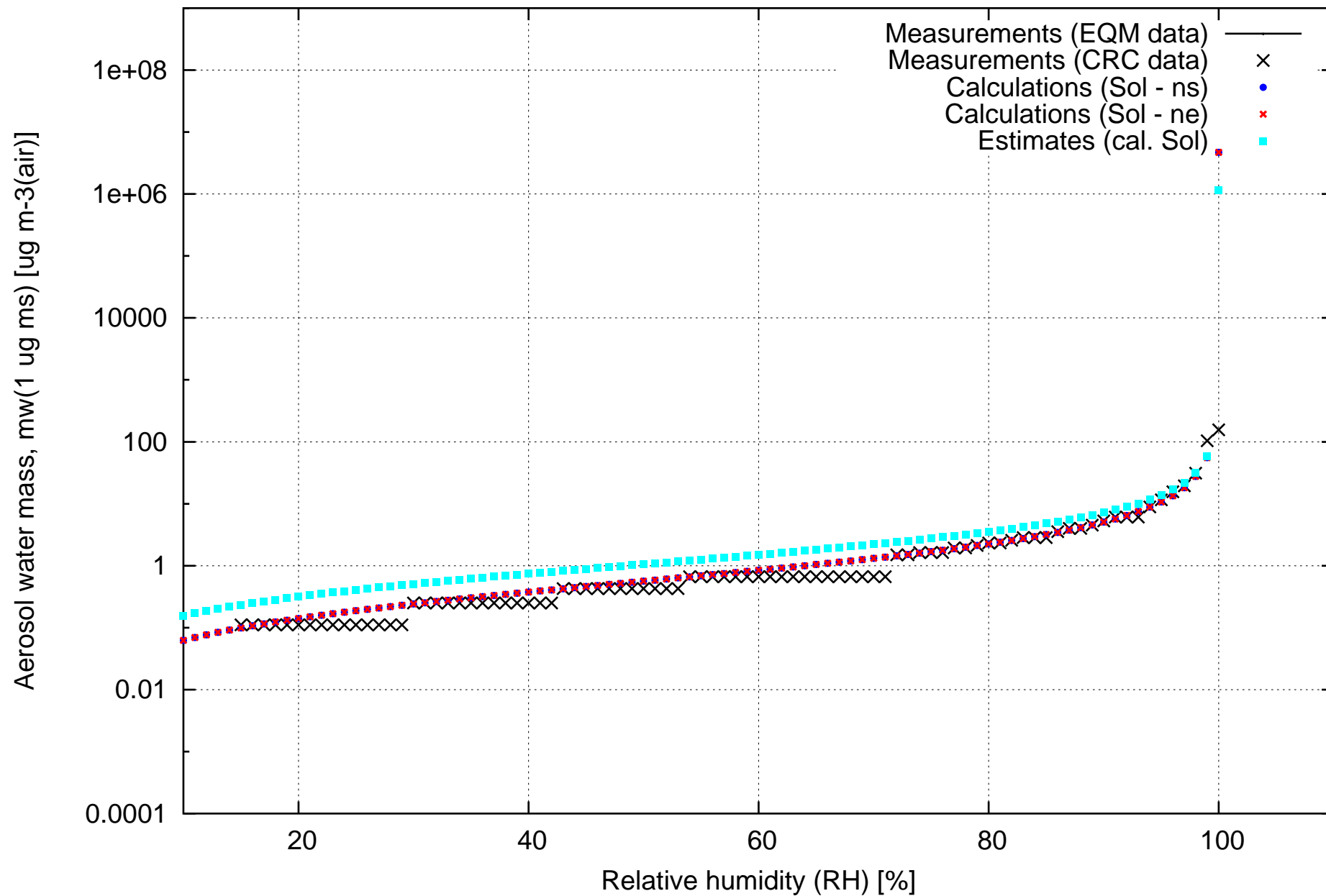
Ammonia - NH_3



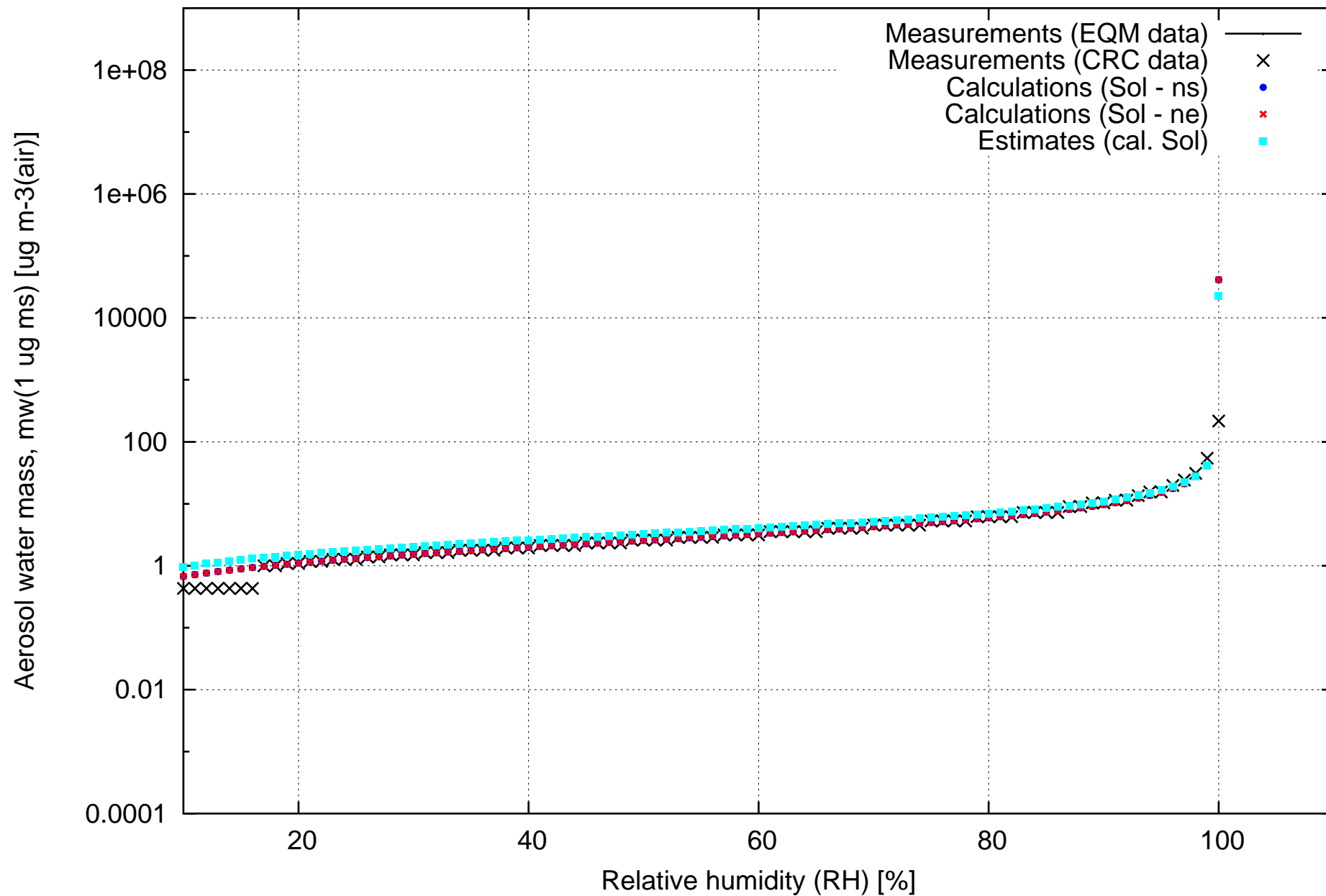
Acetone - (CH₃)₂CO



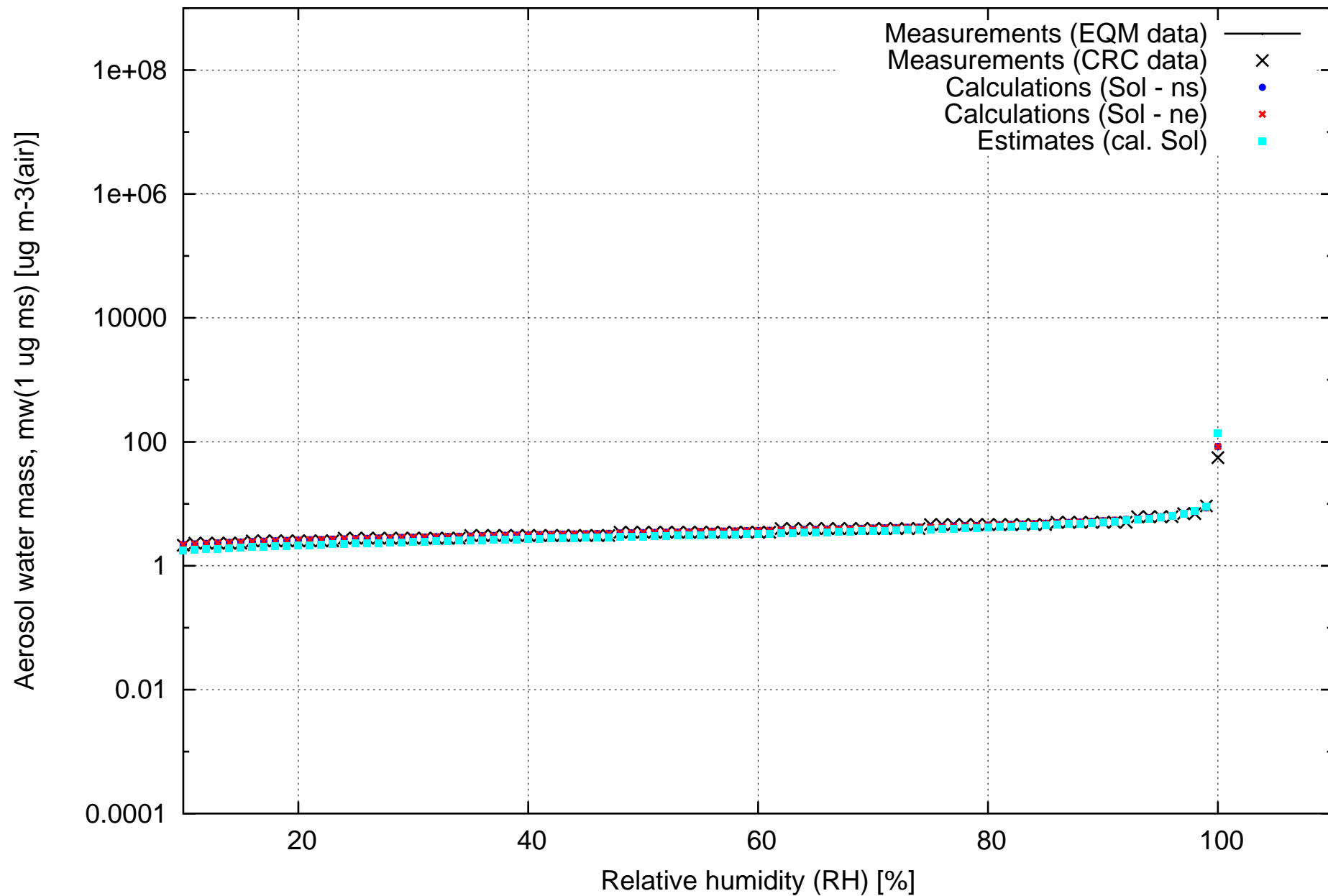
Methanol - CH₃OH



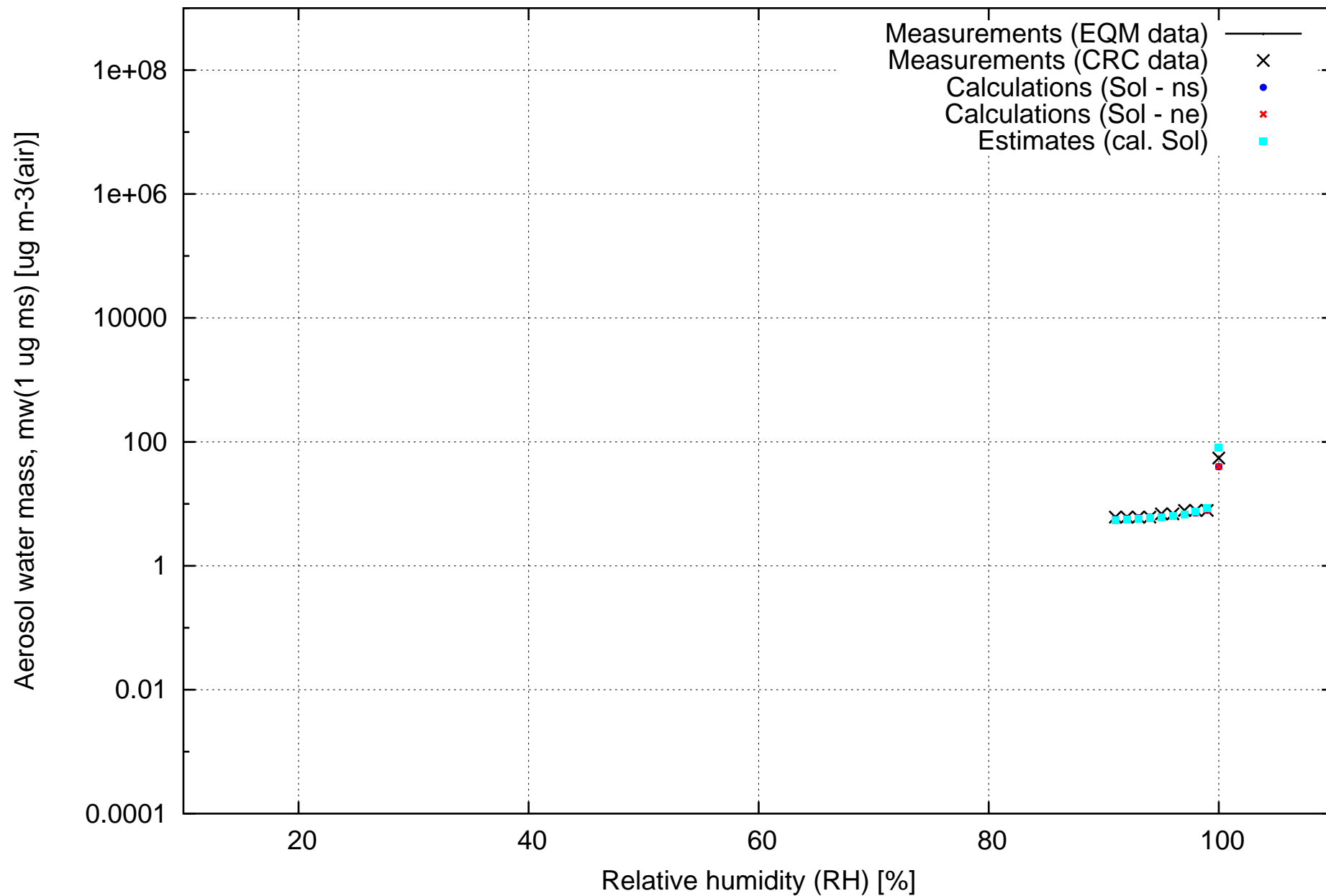
Ethanol - CH₃CH₂OH



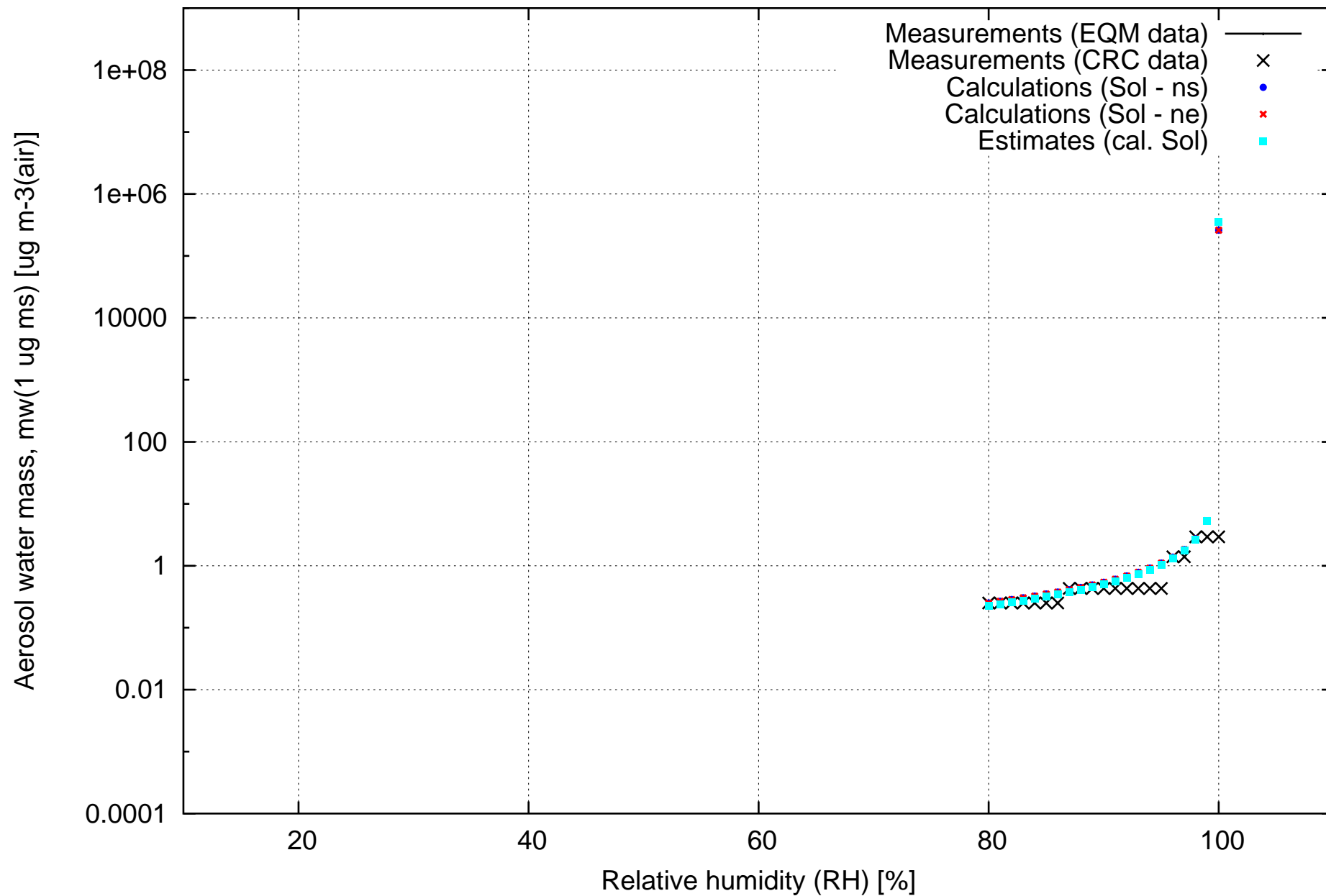
D-Fructose - C₆H₁₂O₆



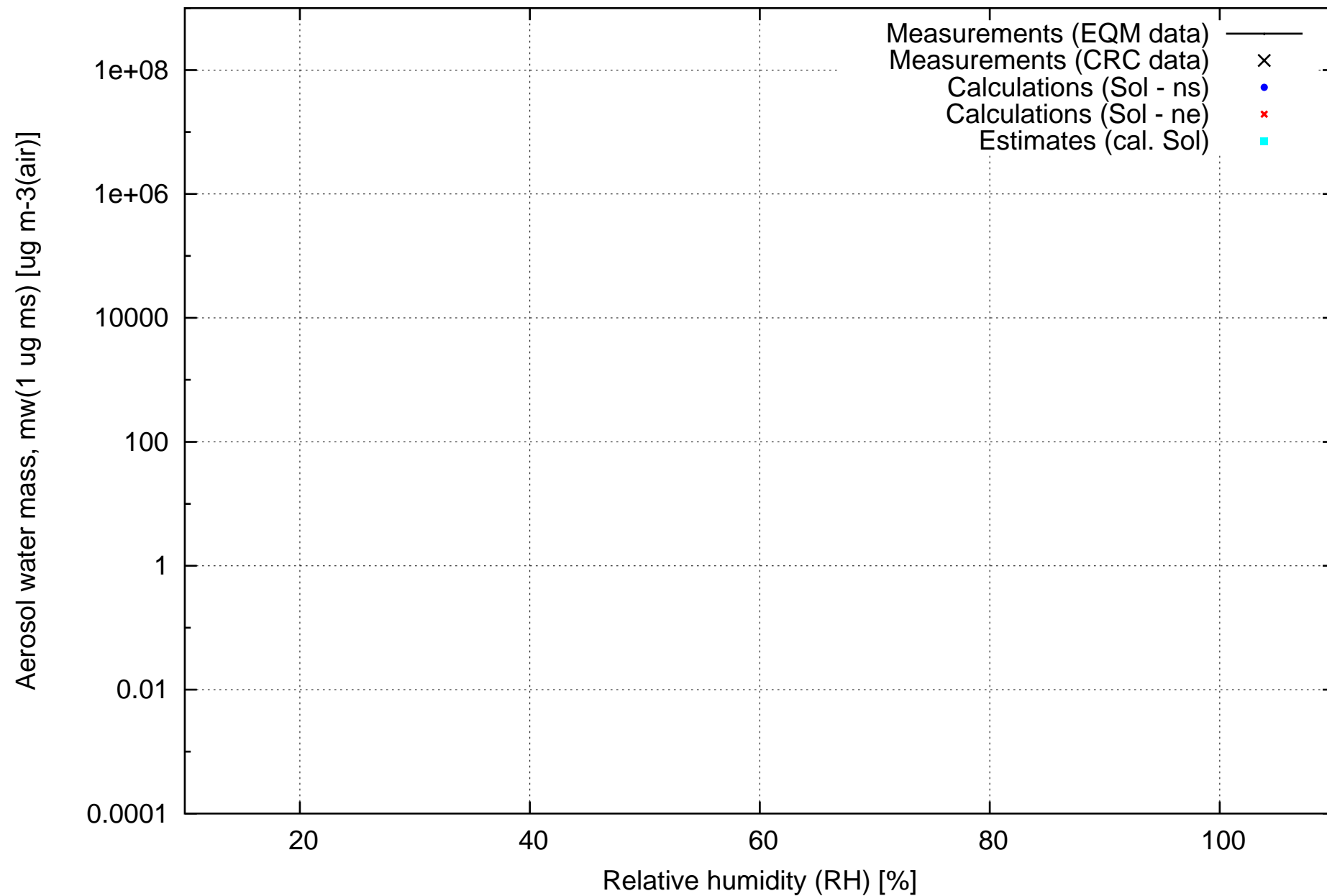
D-Mannitol - C₆H₁₄O₆



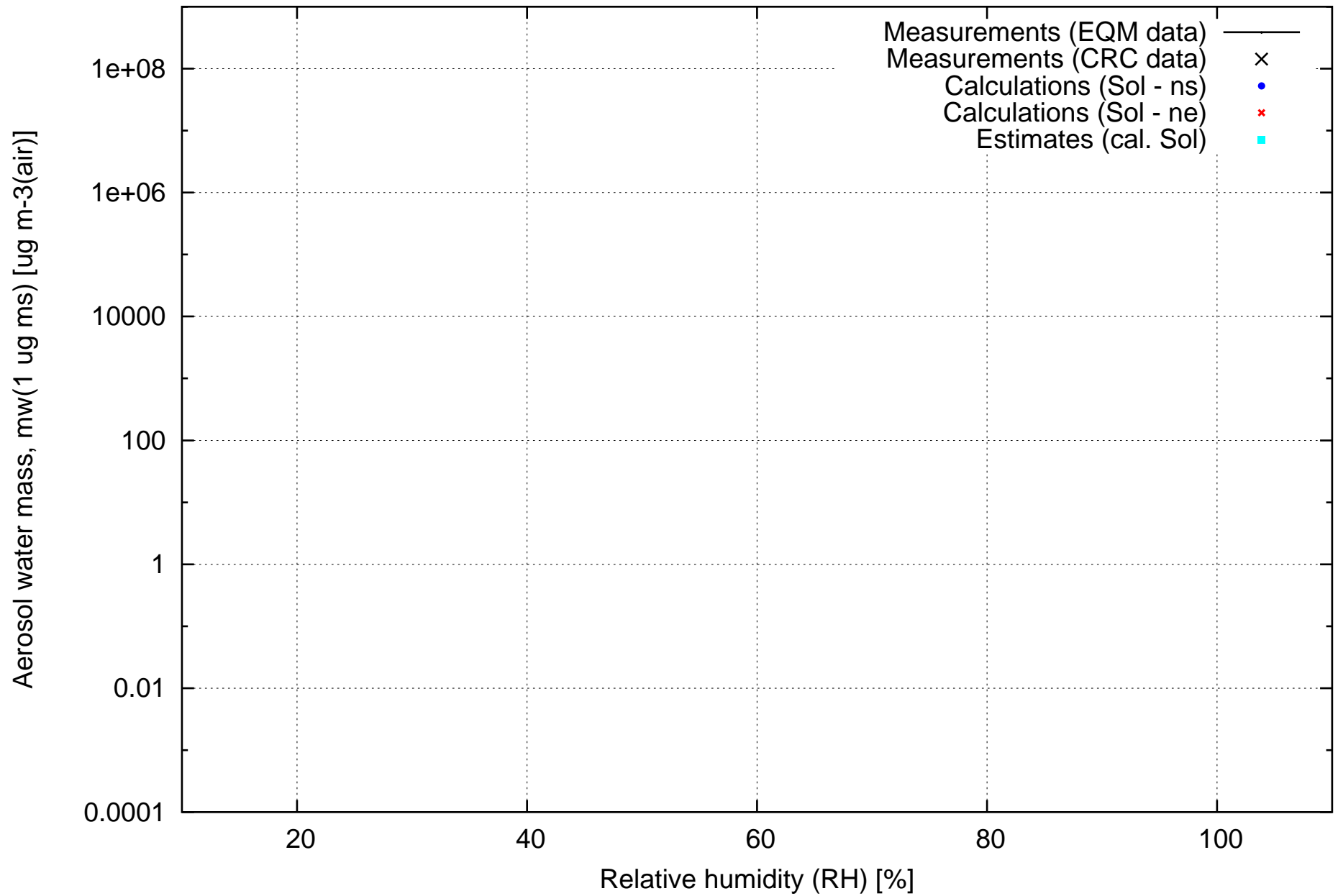
Sucrose - C12H22O11



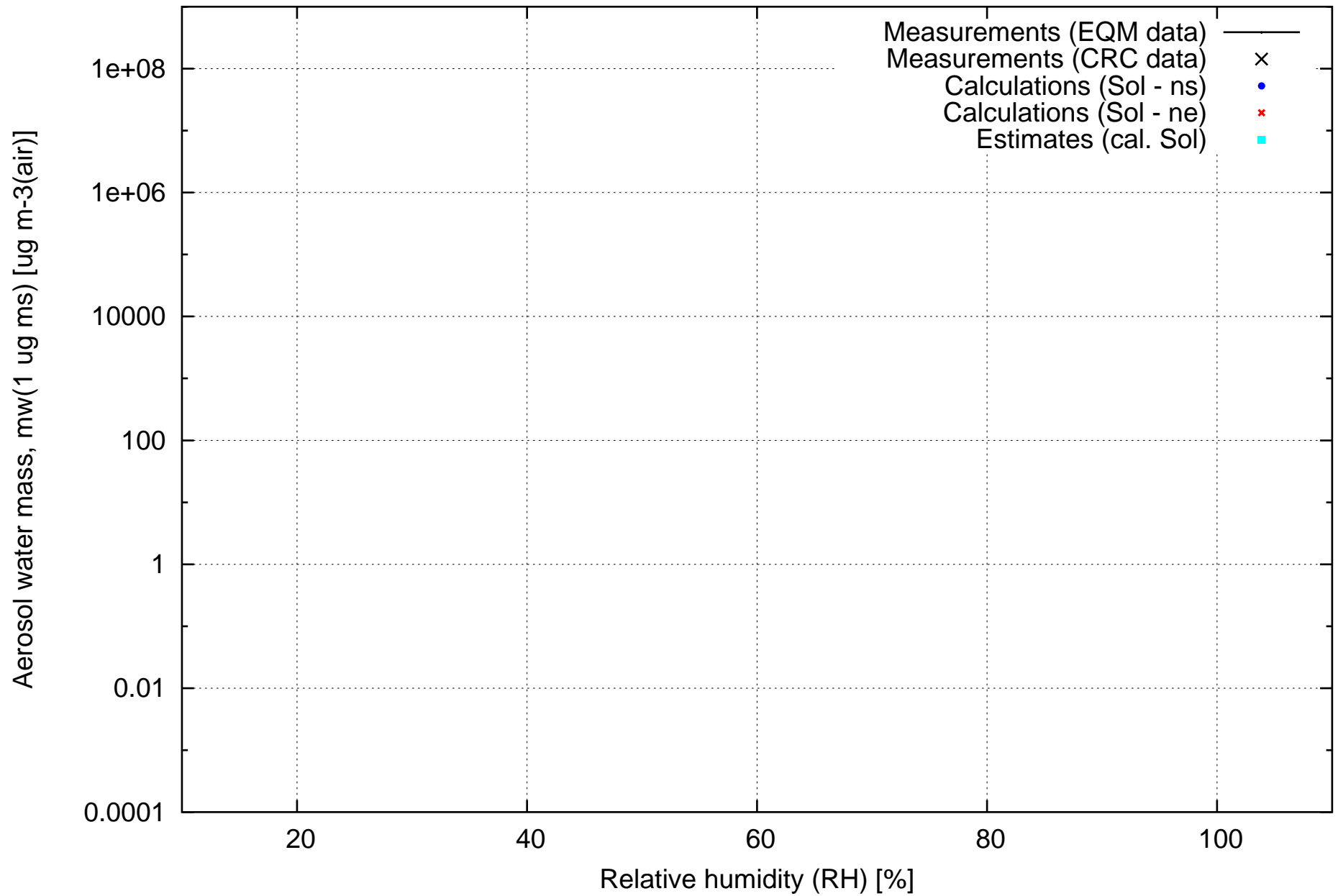
Solute - dummy



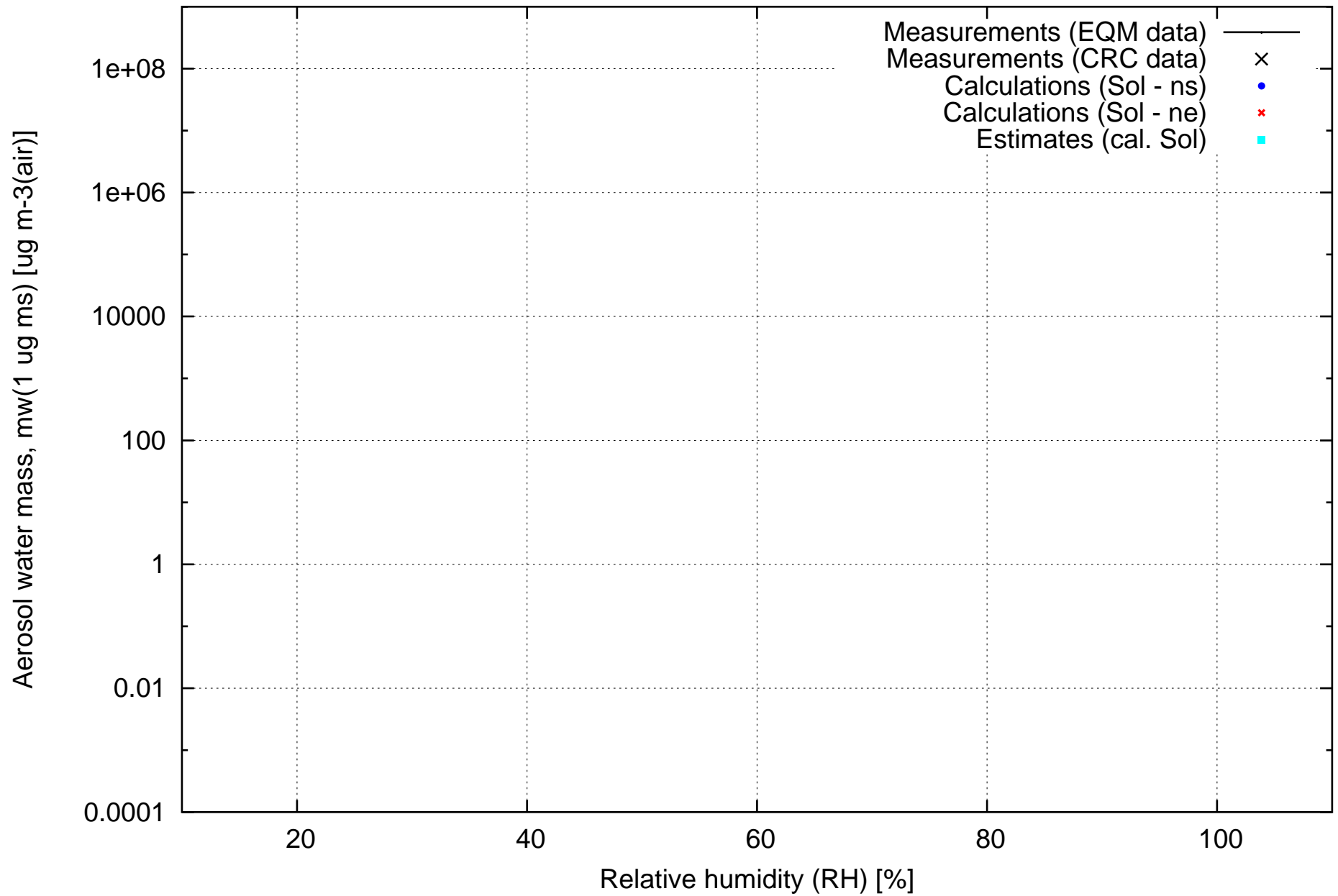
Solute - dummy



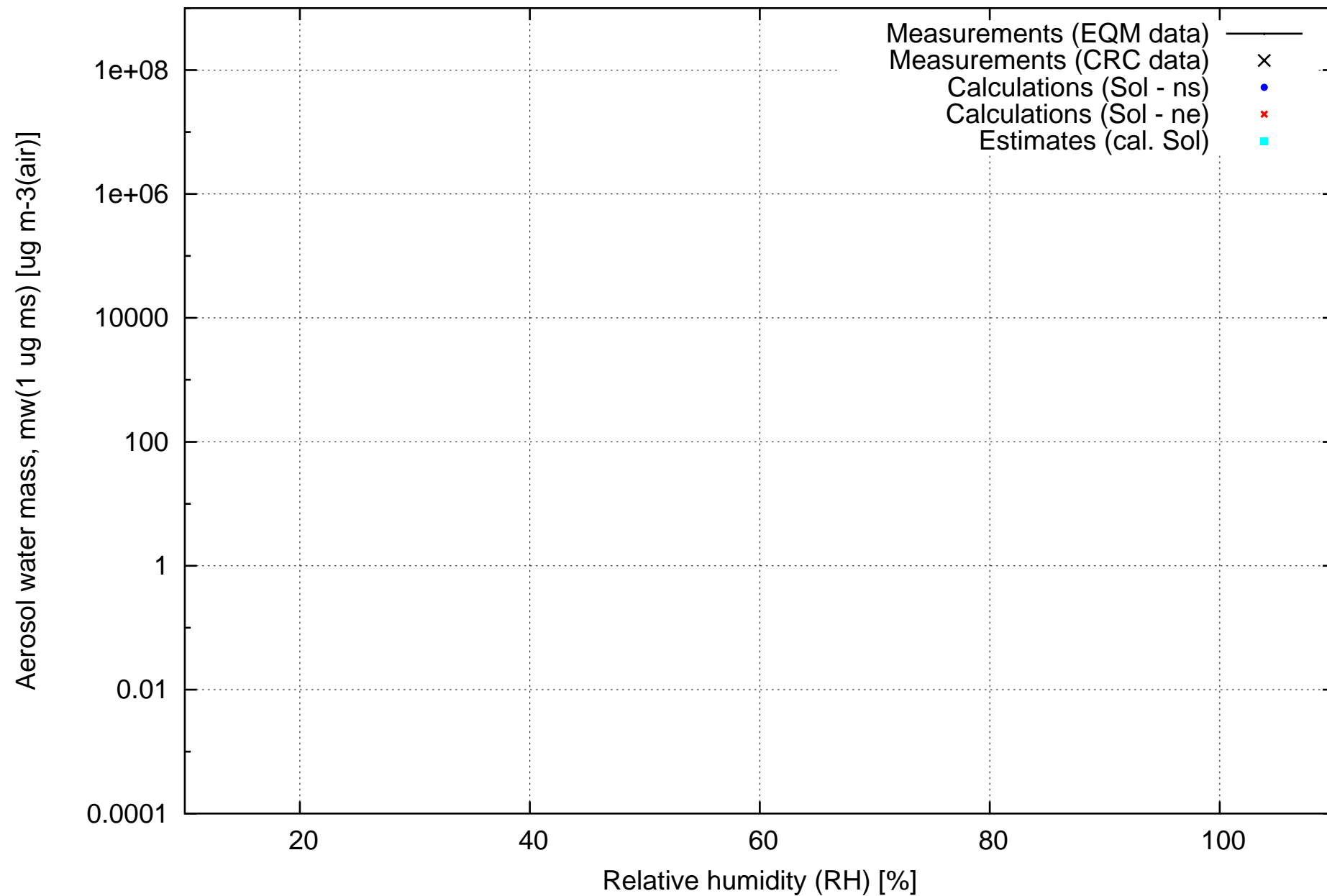
Solute - dummy



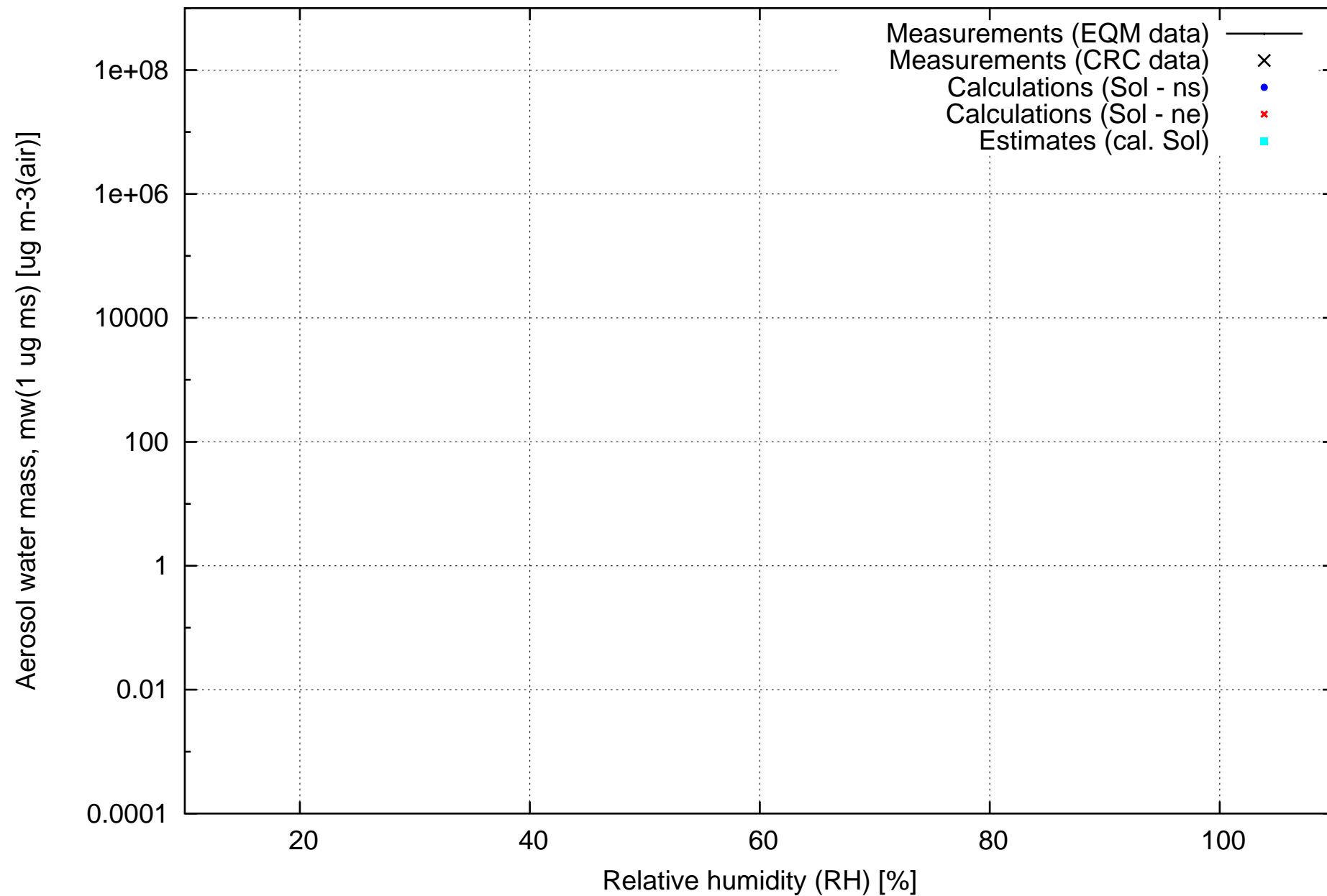
Solute - dummy



Solute - dummy



Solute - dummy



Solute - dummy

