

## Separation of emitted and photochemical formaldehyde in Mexico City using a statistical analysis and a new pair of gas-phase tracers

A. R. Garcia<sup>1,\*</sup>, R. Volkamer<sup>1</sup>, L. T. Molina<sup>1</sup>, M. J. Molina<sup>1</sup>, J. Samuelson<sup>2</sup>, J. Mellqvist<sup>2</sup>, B. Galle<sup>2</sup>, S. C. Herndon<sup>3</sup>, and C. E. Kolb<sup>3</sup>

<sup>1</sup>Dept. of Earth, Atmospheric and Planetary Sciences, Massachusetts Inst. of Technology, Cambridge, MA 01239-4307, USA <sup>2</sup>Chalmers Tekniska Hoegskola, Goeteborg, Sweden

<sup>3</sup>Center for Atmospheric and Environmental Chemistry, Aerodyne Research, Inc. Billerica, MA 01821-3976, USA

Received: 2 August 2005 – Published in Atmos. Chem. Phys. Discuss.: 11 November 2005 Revised: 21 August 2006 – Accepted: 3 October 2006 – Published: 9 October 2006

Abstract. Photochemical pollution control strategies require an understanding of photochemical oxidation precursors, making it important to distinguish between primary and secondary sources of HCHO. Estimates for the relative strengths of primary and secondary sources of formaldehyde (HCHO) were obtained using a statistical regression analysis with time series data of carbon monoxide (CO) and glyoxal (CHOCHO) measured in the Mexico City Metropolitan Area (MCMA) during the spring of 2003. Differences between Easter week and more typical weeks are evaluated. The use of CO-CHOCHO as HCHO tracers is more suitable for differentiating primary and secondary sources than  $CO-O_3$ . The application of the  $CO-O_3$  tracer pair to mobile laboratory data suggests a potential in-city source of background HCHO. A significant amount of HCHO observed in the MCMA is associated with primary emissions.

#### 1 Introduction

Improving the air quality in many large cities require a better understanding of the sources and transformation of pollutants in the atmosphere (see, e.g., Molina and Molina, 2002). Formaldehyde (HCHO) is the simplest and most abundant carbonyl in urban air. HCHO is carcinogenic (Vaughn et al., 1986), and highly water soluble (Dean, 1985), making respiration of air containing elevated levels of formaldehyde an efficient pathway for human exposure. Industrial emissions (Carlier et al., 1986) and mobile sources of directly emitted (primary HCHO) have been identified (Hoekman, 1992; Dodge, 1990; Carter, 1995; Anderson, 1996; Janning et al., 1996; Williams et al., 1990; Kean et al., 2001; Reyes et al., 2006) as sources for HCHO in the urban atmo-

Correspondence to: A. R. Garcia

(agustin@atmosfera.unam.mx)

sphere. Realtime on-road measurements can quantify average fleet HCHO emissions (Kolb et al., 2004; Zavala et al., 2006) and quantify variable emissions among individual vehicles (Shorter et al., 2005). However, integrated estimates of primary HCHO from both the vehicle fleet and industrial sources are largely unavailable.

Formaldehyde can also be produced from the atmospheric oxidation of numerous VOCs (secondary HCHO) (Finlayson-Pitts and Pitts, 1986; Seinfeld, 1986; Altshuller, 1993; Dodge, 1990; Carter, 1995). Primary emission and secondary formation of formaldehyde both contribute to the ambient concentration of this compound. In urban air, ambient concentrations of HCHO typically range from  $1-20 \text{ ppb}_v$ (Carlier et al., 1986); lower concentrations are found in semi-urban and rural air masses inside the boundary layer (0.5 ppbv, Finlayson-Pitts and Pitts, 1986) and the background troposphere (0.20 ppbv, Zhou et al., 1996). These variations are due to formaldehyde's relatively short daytime atmospheric life time (generally few hours), primarily determined by reaction with OH-radicals and photolysis.

HCHO photolysis leads to the formation of HO<sub>2</sub>-radicals, which react with NO during the morning hours, rapidly converting to OH-radicals. In the MCMA, HCHO photolysis is the dominant radical source sustaining photochemical smog formation in the late morning and throughout the day (Volkamer et al., 2005a<sup>1</sup>). For control and regulatory purposes, it is important to differentiate between primary (directly emitted) and secondary (photochemically produced) sources of HCHO, in order to identify those control policies that efficiently enable to reduce ambient HCHO concentrations and ultimately improve urban air pollution.

<sup>&</sup>lt;sup>1</sup>Volkamer, R., Molina, L. T., Molina, M. J., et al.: Experimental quantification of the primary and secondary HOx formation in Mexico City, Atmos. Chem. Phys. Discuss., in preparation, 2005a.

During the MCMA-2003 field campaign, held in April and early May 2003 in the Mexico City Metropolitan Area, measured HCHO concentrations ranged from 1 to 36 ppb<sub>v</sub> with a monthly average of  $(8.2\pm4.6)$  ppb<sub>v</sub> at the Mexican National Center for Environmental Research and Training (CENICA) supersite (Volkamer et al., 2005b), in good agreement with the monthly average concentration of  $(6.0\pm4.7)$  ppb<sub>v</sub> measured in downtown La Merced (Grutter et al., 2005).

It is not possible to distinguish between primary and secondary sources directly from ambient HCHO concentration measurements. Previous attempts to estimate their relative strengths have used a statistical analysis approach (Friedfeld et al., 2002) that aimed to reproduce the concentration-time series (ct-series) of ambient HCHO from a linear combination of ct-series of CO and O<sub>3</sub> (or mathematical homologues) as tracers for primary and secondary HCHO, respectively. In this paper, a similar approach is employed, using data from real-time open-path measurements conducted at the CENICA supersite during MCMA-2003. Besides the previous CO-O<sub>3</sub> tracer pair, a new tracer pair (CO-CHOCHO) is used as input for the statistical analysis, and the performance of both tracer pairs is compared. In both cases CO is an excellent indicator for primary HCHO (mobile sources contribute 99% of ambient CO; stationary point and area sources contribute only 1% in the MCMA - Secretaria del Medio Ambiente, 2002); the statistical weight of the HCHO to CO ratio is used to estimate the primary contribution to HCHO ambient concentrations. Secondary HCHO is correlated with a photochemical products, either O<sub>3</sub> or glyoxal (CHOCHO). Unlike O<sub>3</sub>, CHOCHO does not react with nitric oxide (NO), and is essentially unaffected by traffic emissions. Its time-resolved direct detection in the atmosphere was demonstrated for the first time during MCMA-2003, presenting a novel indicator molecule for VOC oxidation processes (Volkamer et al., 2005b). A further advantage of CHOCHO as indicator for secondary HCHO lies in the close analogy of their atmospheric sink reactions. Like HCHO, CHOCHO is primarily removed by OH-radical and photolysis loss processes (Atkinson, 2000) and daytime atmospheric residence times under typical MCMA conditions of both species are conditions comparable, i.e. about 2.2 h for CHOCHO (Volkamer et al., 2005b), and about twice as long for HCHO

During MCMA-2003 two different emissions regimes were sampled: (1) before and after Easter week and (2) during Easter week, when emissions across the MCMA were considerably lower. The objective of this paper is to assess the relative contributions of primary and secondary HCHO to ambient HCHO concentrations at the CENICA supersite, and to assess the spatial variability of HCHO background concentrations by utilizing mobile laboratory measurements (Kolb et al., 2004) made throughout the MCMA.

## 2 Method

## 2.1 CENICA central site

The CENICA site is located approximately 10 km southeast of downtown Mexico City in the Iztapalapa campus of the Metropolitan Autonomous University (UAM). The area features commercial, residential and industrial buildings as well as the conservation area, Cerro de la Estrella.

During the MCMA-2003 campaign, the temperature at CENICA ranged from 10°C to 32°C and the relative humidity from 5% to 98%. The measurement period was characterized by clear to partially clouded skies during the morning and afternoon hours, with some evenings featuring mild convective rainfalls.

## 2.2 CENICA measurements

Differential Optical Absorption Spectroscopy (DOAS) and Fourier Transform Infrared (FTIR) open path systems were deployed at CENICA. These instruments measured continuously gas phase pollutants as aromatics, ozone, carbon monoxide, carbon dioxide, formaldehyde, acetaldehyde and glyoxal, among others. The time series data include days from 4 April to 5 May 2003.

The FTIR system consisted of a medium resolution  $(1 \text{ cm}^{-1})$  spectrometer (Bomem MB104) coupled to a custom fabricated transmitting and receiving telescope. At the other side of the light path, a cubecorner array was mounted at a tower, making up a total folded path of 860 m. The system provided data with 5-min integration time. Spectra were analyzed using the latest HITRAN database cross sections (Rothman et al., 2000) and a nonlinear fitting algorithm.

The DOAS technique has been described elsewhere (Platt, 1994). The long-path DOAS (LP-DOAS) measured HCHO, O<sub>3</sub>, CHOCHO and other species by detection of their unique specific narrow-band (<5 nm) absorption structures in the ultraviolet and visible spectral ranges. The LP-DOAS was installed on the rooftop of the CENICA building, where light of a broadband UV/vis lightsource (Xe-short arc lamp) was projected into the open atmosphere towards an array of retro reflectors located in south-westerly direction on top of the local hill Cerro de la Estrella, folding the lightbeam back into the instrument, where spectra were recorded using a Czerny-Turner type spectrometer coupled to a 1024-element PDA detector. The average height of the light path was 70 m above ground and a total path length was 4.42 km. The experimental setup, evaluation algorithms and data are described in detail elsewhere (Volkamer et al., 2005b).

## 2.3 Mobile laboratory measurements

The mobile laboratory was deployed to three additional fixed sites in the MCMA. The highly urban Merced site is close to the center of the city. The residential Pedregal site, located to the southeast of the city center, typically receives middle and late afternoon aged emissions from the city. The rural Santa Ana site is located just outside the MCMA to the south and is more representative of the background conditions. The instrumentation aboard the mobile laboratory and methods used for its analysis have been described elsewhere (Canagaratna et al., 2004; Kolb et al., 2004; and Herndon et al., 2005b). Formaldehyde was measured using tunable infrared differential absorption spectroscopy (TILDAS). During the MCMA 2003 campaign, a pair of absorptions lines at 1774.67 and  $1774.83 \text{ cm}^{-1}$  were used to determine HCHO concentrations (Herndon et al., 2005). Two relatively weak water lines bracket these features, and a very small water line is present in the gap between. There are no other known strongly absorbing species found in an urban setting in this region. As operated during these measurements, the 1 s rms precisions for HCHO was normally less than 1.1 ppb. CO and O<sub>3</sub> were measured using conventional commercial infrared and ultraviolet absorption instruments, respectively. Due to a problem discovered in the O<sub>3</sub> monitor deployed aboard the mobile laboratory, a correction has been applied to the mobile data (Dunlea et al., 2005). When other fixed

site data was available for  $O_3$ , it was used. As part of the quality assurance of ambient HCHO measurements, adjustments to the absorption cross-section of HCHO in the UV and IR- spectral ranges were performed to assure matching calibration of HCHO measurements by DOAS and TILDAS as discussed in (Volkamer et al., 2005c<sup>2</sup>).

#### 2.4 Statistical analyses

All statistical analyses were performed using Intercooled STATA  $8^{\textcircled{O}}$ . Linear regressions models were applied to the time series data set for HCHO, CO, O<sub>3</sub> and CHOCHO. In order to identify the most suitable model for the time series data, two evaluations were performed. The first used HCHO, CO and O<sub>3</sub> series applying ten different transformations: linear (no transformation), natural log, square root, second power, 3rd power, inverse, inverse of natural log, inverse of square root, inverse of second power and inverse of 3rd power. The second approach used the same transformations with HCHO, CO and CHOCHO.

The special case of a linear model was with the second approach, is represented by the equation:

$$[\text{HCHO}] = \beta_0 + \beta_1 [\text{CHOCHO}] + \beta_2 [\text{CO}] \tag{1}$$

For every unit increase in CHOCHO concentration there is a  $\beta_1$  unit increase in HCHO; for every unit increase in CO concentration there is a  $\beta_2$  unit increase in HCHO;  $\beta_0$  represents

residual HCHO, which is associated with either primary nor secondary HCHO, and is referred henceforth as background HCHO (in units of ppb<sub>v</sub>). Background HCHO may be due to alternative HCHO sources not represented by the indicator surrogates for primary and secondary HCHO, or due to a different lifetime of HCHO and these surrogates, which may introduce a bias in the  $\beta$ -values of Eq. (1).

To estimate the contribution of CO and CHOCHO to primary and secondary HCHO to the ambient concentration to HCHO, it is necessary to scale the values of CO and CHO-CHO by their respective  $\beta$  value.

Contributions due to the photochemistry, emissions and background HCHO can be computed with the following equations:

% pho=
$$\frac{\sum \left(\beta_1 [\text{CHOCHO}]_i\right)}{\sum \left(\beta_1 [\text{CHOCHO}]_i + \beta_2 [\text{CO}]_i + \beta_0\right)} \times 100$$
(2)

% emiss=
$$\frac{\sum \beta_2 [\text{CO}]_i}{\sum (\beta_1 [\text{CHOCHO}]_i + \beta_2 [\text{CO}]_i + \beta_0)} \times 100 \quad (3)$$

% bkgr=
$$\frac{\beta_0}{\sum \left(\beta_1 [\text{CHOCHO}]_i + \beta_2 [\text{CO}]_i + \beta_0\right)} \times 100 \quad (4)$$

Where % pho represents the contribution of secondary HCHO (photochemical VOC oxidation) and % emiss is the contribution from primary sources (vehicle emissions); % bkgr indicates HCHO, which can neither be accounted as primary or secondary HCHO. [CHOCHO]<sub>*i*</sub> and [CO]<sub>*i*</sub> represent the concentrations of CHOCHO and CO respectively at time i and  $\beta_0$ ,  $\beta_1$  and  $\beta_2$  are the coefficients obtained by the linear regression (see Eq. 1).

As part of the sensitivity tests, the effect of remaining differences in atmospheric lifetime of CHOCHO and HCHO on the results of the statistical analysis was investigated using alternative concentration time profiles for CHOCHO and HCHO concentrations, which were corrected for losses due to photolysis loss and OH reaction and were derived as follows:

$$[X]_{corr} = [X] \left(1 + J_x \Delta t + K_{OH-X} \left[OH\right] \Delta t\right)$$
(5)

Where X represents either HCHO or CHOCHO,  $J_x$  is the photolysis frequency for X as measured by spectroradiometry (Volkamer et al., 2005b),  $K_{OH-X}$  is the reaction rate between OH and X (Sander et al., 2002), [OH] the measured concentration of OH-radicals (Shirley et al., 2005), and  $\Delta t$ , the chosen time interval for correction, was (varied between 1–120 min).

Further sensitivity tests aimed to correct the CO concentration time profile for influence due to the vertical mixing of air masses, which are likely depleted in their HCHO/CO ratio. An extreme case was estimated assuming the HCHO/CO ratio to be zero outside the planetary boundary layer, and corrections were calculated as follows:

$$[CO]_{corr} = [CO]_{measured} - [CO]_{back} \frac{(D-1)}{D}$$
(6)

<sup>&</sup>lt;sup>2</sup>Volkamer, R., Zavala, M., Molina, L. T., Molina, M. J., Samuelsson, J., Mellqvist, J., Galle, B., Herndon, S., Kolb, C., Knighton, B., Flores, E., and Grutter, M.: Open-path emission factors derived from DOAS and FTIR Measurements in the Mexico City Metropolitan Area, Atmos. Chem. Phys. Discuss., in preparation, 2005c.



**Fig. 1.** Traffic count over Insurgentes Ave. 8 km towards west from CENICA during the 2003 April campaign.

Where  $[CO]_{measured}$  is the CO concentration from the ground measurements,  $[CO]_{back}$  is the CO concentration aloft the mixing height (taken as 60 ppbv, the value measured in the background sites outside the MCMA). *D* is the dilution factor, which was calculated using CO as a chemically inert tracer for the variability of mixing height as follows:

$$D = \frac{[CO]_{traffic}}{[CO]_{measured}}$$
(7)

Where *D* represents a relative factor by which the mixing height increased relative to the morning hours. (D-1) thus represents the volume of air that contained [CO]<sub>back</sub>. [CO]<sub>traffic</sub> is the estimated CO concentration under the mixing height due to vehicular emissions related to the traffic activity.

#### 3 Results and discussion

Emissions activities varied over the field campaign; from Thursday to Sunday during the Easter week magnetic traffic counters at Insurgentes avenue measured 38% fewer cars (see acknowledgment) than during same days in prior and subsequent weeks. Figure 1 shows the traffic variation during 10 days spanning the Easter holidays; the activity decrease during four-day period is noticeable. Reductions in the average concentrations of CO, O<sub>3</sub>, CHOCHO and HCHO were observed; between Easter week, and the averages from the prior and subsequent weeks: accounting to 36%, 21%, 44% and 36%, respectively (on a 24h average basis). Averages from 18 to 20 April were compared with averages from 15 to 16 and 21, 22 April. It is interesting to note that reductions in traffic have different effects in the concentrations of the different pollutants. While CO, CHOCHO and HCHO more or less reflect the reduction in traffic on Insurgentes, the reduction in O<sub>3</sub> is much less pronounced, though significant. The ct-series of O<sub>3</sub>, CO, HCHO and CHOCHO mixing ratios, which are the basis for the statistical analysis, are shown in Fig. 2a. Easter week was from 18 to 20 April. Figure 2b presents an expansion of O<sub>3</sub>, HCHO and CO concentrations (approach 1) at CENICA, shown for a two day period shortly after Easter week. Similarly, in Figure 2c concentrations of CHOCHO, HCHO and CO (approach 2) are shown for the same time interval. Open path data smooth out possible influence from any local sources near the sampling site. Figure 2 presents plots of O<sub>3</sub>, HCHO, CHOCHO and CO concentrations at CENICA during mid and late April. The CO generally has lower concentrations during the Easter holidays (18–20 April).

Amplification of the time series in Figs. 2b and c show how CO and HCHO concentrations start to build up before sunrise, i.e. during the early morning rush hour. When the sun rises photochemical VOC transformations generate both CHOCHO, HCHO and also lead to the production of  $O_3$ . However, initial  $O_3$  production is suppressed due to the efficient titration reaction of  $O_3 + NO$ , producing NO<sub>2</sub>. This process delays peak ozone to near noon. Furthermore, efficient  $O_3$  production in the MCMA compensates its dilution in the rising boundary layer. The result is accumulating amounts of  $O_3$  in the later morning, which is typical for  $O_3$ , but is not observed for the other species. In comparing panel b and c, the better match of the CHOCHO and HCHO time profiles is visible, reflecting similar removal pathways from OH and photolysis, as well as dilution for these gases.

#### 3.1 Statistical modeling results at CENICA

Statistical tests were performed to determine the mathematical transformation that yielded best correlations for both approaches. As the best model for the CO-O<sub>3</sub> tracer pair, a square root transformation was identified ( $r^2=0.53$ , see Table 1), while the CO-CHOCHO tracer pair was best fit using a linear relation ( $r^2=0.76$ , see Table 1), covering both the Easter and non-Easter weeks. Table 2 shows results of these regression tests for the tested mathematical transformations. A comparison between the measured and fitted concentration time profiles of ambient HCHO concentrations is shown in Fig. 3. It is possible to observe that HCHO modeled by CO-CHOCHO pair follows more accurate the diurnal profile than the HCHO obtained by CO-O<sub>3</sub> pair (see Fig. 4). Also in the scatter plots the confidence interval area are smaller in the CO-CHOCHO pair than CO-O<sub>3</sub> pair, finally the  $r^2$  coefficients are superior for CO-CHOCHO pair than CO-O3.

A summary of results and the statistical significance of the multiple regression analysis for the CO-CHOCHO tracer pair (Eq. 1) are presented in Table 2. This table also shows the coefficients from the entire campaign, before, during and after Easter week. In order to compare results the analysis was run using subsets of data, which comprise about comparable number of data points for each time period considered (about 2 days). By dividing the data in 3 periods of interest, it is possible to improve the  $r^2$ -values, and lower the residual  $\beta_0$  values, suggesting that the  $\beta$ -values are not fixed, but vary from day to day.



Fig. 2. Time series plots of  $O_3$ , HCHO, CHOCHO and CO measured in CENICA supersite during April 2003. Times is in UTC, each day is marked with 0 and 12 h.

In an attempt to characterize the day-by-day variability in the  $\beta$ -values, daily subsets of data were analyzed using the CO-CHOCHO tracer pair. Of the 24 days analyzed, only 13 days met the quality assurance criteria to be considered in this analysis, i.e. had a minimum of 17 h of data coverage per day, and  $r^2 > 0.7$ . The average ratio of primary HCHO to CO determined from individual days  $(0.005\pm0.001 \text{ ppb}_v/\text{ppb}_v)$  and the ratio of secondary HCHO to CHOCHO  $(0.040\pm0.026 \text{ ppb}_v/\text{ppb}_v)$  both agree within error-limits with the value derived from the global data-set (see Table 1). The  $\beta_1$  and  $\beta_2$  parameters were found to be independent of each other, and no correlations with traffic counts, temperature and wind speed were observed. Also, the HCHO background was found independent from traffic counts and wind speed. All significant correlations are shown in Fig. 5. Significant correlations were observed for the HCHO background with  $\beta_1$ ,  $\beta_2$  (both negative correlations),  $O_3$  and temperature (both positive correlations). The negative correlations of the HCHO background with  $\beta_1$  and  $\beta_2$  reflect lower  $\beta_0$ -parameters during days, when stronger coupling of the ambient HCHO with the primary and secondary indicator molecules is observed, in general agreement with the interpretation of  $\beta_0$  as a residual HCHO concentration. The positive correlations of HCHO background with  $O_3$  and temperature are indicative of some effect during conditions favorable for photochemical activity, which may either point to a possible role of the HCHO background in promoting photochemical smog, or be an indication for higher HCHO concentrations during days of more active photochemistry, where imperfections in the approach are likely to become more significant, or both.

	CO-HCHO-O <sub>3</sub>				СО-НСНО-СНОСНО			
	$\beta_1$	$\beta_2$	$\beta_0$	$R^2$	$\beta_1$	$\beta_2$	$\beta_0$	$r^2$
Linear	0.064	0.002	1.69	0.41	0.014	0.0015	2.4	0.76
ln	0.290	0.569	-3.06	0.45	0.221	0.390	-1.85	0.53
Sqrt	0.167	0.042	0.17	0.53	0.060	0.027	1.01	0.69
$x^2$	5.89E-03	3.22E-06	47.785	0.14	3.80E-04	2.78E-06	30.329	0.73
x <sup>3</sup>	5.11E-04	3.84E-09	844.194	0.03	9.27E-06	3.77E-09	334.178	0.71
1/x	2.64E-03	38.441	0.123	0.12	6.15E-03	38.968	0.116	0.13
$1/\ln(x)$	7.87E-04	5.505	-0.215	0.07	3.78E-02	5.464	-0.230	0.07
1/sqrt(x)	0.102	4.525	0.229	0.21	0.158	4.213	0.236	0.24
$1/x^2$	-2.77E-05	2019.471	0.033	0.02	-1.34E-04	1982.638	0.030	0.02
$1/x^{3}$	-7.96E-07	50016.25	0.010	0.00	-4.61E-06	50232.12	0.009	0.00

Table 1. Result from different linear regressions using different data transformations.

Table 2. Model Results using the CO-CHOCHO tracer pair.

	$\beta_1$	$\beta_2$	$\beta_0$	$r^2$	N. Obs
Pre Easter	$0.0143 {\pm} 0.0005$	$0.0017 {\pm} 0.00011$	$1.44{\pm}0.31$	0.77	257
Easter	$0.0197 {\pm} 0.0008$	$0.0027 {\pm} 0.00021$	$1.33 {\pm} 0.17$	0.90	256
Post Easter	$0.0149 {\pm} 0.0006$	$0.0024{\pm}0.00014$	$1.75 {\pm} 0.23$	0.84	264

In an attempt to assess whether differences in the atmospheric lifetime of HCHO and CHOCHO affect the results, corrections to CHOCHO and HCHO due to OH and photolysis loss were made (Eq. 5). The time intervals for integration,  $\Delta t$ , was varied from 1 to 120 min, and corresponding corrections in the concentrations of HCHO and CHOCHO varied from 10% and 13% (1 min) to 74% and 192% (120 min). Use of the loss-corrected time series of HCHO and CHO-CHO tended to improve the  $r^2$ -values (from 0.76 to 0.8 in global analysis), but did not significantly affect the percent fractions of primary and secondary HCHO. Those improvements reflect a lower HCHO background concentration of about 5%.

A series of sensitivity tests were carried out, where a variable offset-concentrations (between 10 to 150 pptv) was added to the ct-profile of glyoxal to reflect measurement uncertainties in glyoxal concentrations. Additionally, in some runs a modified ct-profile of CO was used to reflect a possible influence that arises from the vertical mixing of air masses. Air from aloft the planetary boundary layer (PBL) mixes into the PBL during the morning, and likely has a different HCHO/CO ratio that would create a bias in the model. These corrections (Eq. 6) decrement the measured CO concentration, and were generally small, i.e. 0–2% between 5:00 to 9:00 am, and up to 12% around noon time. The results from these sensitivity tests are summarized in Fig. 6, where the percent contribution of primary, secondary

and background HCHO from individual runs are plotted as a function of the  $\beta_0$ -parameter. While the percent fraction of primary HCHO was found to be almost constant, the percent fractions of background and secondary HCHO contributions are anti-correlated, and scale linearly with the  $\beta_0$ parameter. The fact that secondary HCHO decreases as the background HCHO increases, suggests a relation of the  $\beta_0$ parameter with photochemistry rather than emission sources. Furthermore, weak negative correlations of the percent fraction of secondary HCHO with O3 and temperature were observed (despite no such relation was observed for the  $\beta_1$  and  $\beta_2$  parameters). For both parameters, the percent fraction of secondary HCHO showed essentially no variation at low temperatures (below 20 C) and moderately low O<sub>3</sub> (below 110 ppbv), but tended to decrease for higher values of temperature and O<sub>3</sub>. This is taken as an indication that at least during days of intense photochemistry, the  $\beta_0$ -parameter is at least partly affected by remaining imperfections in the CO-CHOCHO tracer approach. This also introduces a slight bias in the interpretation of  $\beta_0$ -parameter as HCHO background, indicating that at least some of the HCHO background is in fact erroneously interpreted secondary HCHO. Such HCHO could be form from the photochemistry of anthropogenic VOC which may not form CHOCHO, or be the result of advected air masses that are depleted in CHOCHO during transport. The  $r^2$  values from the statistical analysis indicate the proportion of HCHO mixing ratio measure-



Fig. 3. Comparison between HCHO measured and fitted values from statistical regression of HCHO-CO-CHOCHO data for each time period. Upper row shows results from pre-Easter week, center row for Easter week and bottom for post-Easter week. In the scatter plots shadow present the 95% confidence interval for each regression. Pre Easter week has  $r^2 = 0.77$ , Easter week 0.90 and post Easter  $r^2 = 0.85$ .

ments that can be predicted by the regression model (Green, 1998). The results in Table 2 indicate that this proportion is in all cases well above 75%, and up to 90%, reflecting that the CO-CHOCHO tracer pair – while not perfect – generally replicates well the observations.

Using the Eqs. (2), (3) and (4) it is possible to compute the ct-profiles of the percentage of primary, secondary and background HCHO. Figure 7 shows the diurnal profiles of the contribution of primary, secondary and background HCHO before, during and after Easter week. For each time period, primary HCHO is the dominant source of HCHO in



**Fig. 4.** Comparison between HCHO measured and fitted values from statistical regression models using HCHO-CO-O<sub>3</sub> for each time period. Upper row shows results from pre-Easter week, center row for Easter week and bottom for post-Easter week. In the scatter plots shadow present the 95% confidence interval for each regression. Pre Easter week has  $r^2 = 0.44$ , Easter week 0.74 and post Easter  $r^2 = 0.47$ .

the early morning hours (before 8:00 am). The photochemistry of VOC becomes active about one hour after sunrise (Volkamer et al., 2005b), and becomes the dominant source of HCHO around 8:30 am, slightly earlier under the traffic conditions of Easter week. Secondary HCHO is the dominant source for most of the day, until in the late afternoon primary sources gain relative importance and become comparable around 3–5 pm. Primary HCHO is the dominant source in the late afternoon and at night. The percent fractions of different sources contributing to ambient HCHO concentra-



Fig. 5. Correlation of HCHO background vs. temperature (a) and O<sub>3</sub> concentration (b). Positive correlation is observed. In panel (c) coefficients  $\beta_1$ ,  $\beta_2$  from Eq. (1) vs. HCHO background concentration.



Fig. 6. Percent-contributions from photochemistry, emissions and the background as a function of background concentration.

**Table 3.** Contributions of photochemistry and emissions to ambientHCHO.

	Seco	Primary	
	СНОСНО	Const	СО
Pre Easter	$41.0 \pm 6.5$	$17.5 \pm 6.5$	$41.5 {\pm} 0.2$
Easter	$38.9 \pm 7.1$	$23.7 {\pm} 6.8$	$37.5 {\pm} 0.3$
Post Easter	$33.5 \pm 6.9$	$20.9 \pm 4.8$	$45.6{\pm}2.1$
Global average	$37.8 {\pm} 6.8$	$20.7 \pm 6.0$	$41.5\pm0.8$
4:00-14:00 Average	$42.3 \pm 2.9$	$18.1 \pm 1.45$	39.6±1.6

tions are listed for the investigated time-periods in Table 3. On average, the contributions of primary, secondary and unaccounted sources (background) to ambient HCHO concentrations (on a 24 h basis) were 42%, 38% and 21%. Respectively using an average from 4:00 to 14:00 – this covers rushhour and photochemistry, but excludes venting and recirculation within the MCMA – as we can see the contributions are comparable.



**Fig. 7.** Average diurnal contributions of primary, secondary and residual HCHO to ambient HCHO for three time periods: before (left), during (middle), after (right) Easter week. The top row of panels gives contributions in units of ppbv, and lower row of panels in percent contributions to ambient HCHO.

Primary HCHO, emitted during early morning hours will help "jump start" photochemistry since formaldehyde photolysis is a major source of prompt radical production in the MCMA (Volkamer et al.,  $2005a^1$ ). The resulting photochemistry will be enhanced, causing more effective VOC oxidation and, therefore, more photochemical HCHO production. Thus, the emitted HCHO fraction and its photochemical production fraction are not decoupled, and it will require a comprehensive photochemical modeling study to determine degree of coupling.

3.2 Application to mobile laboratory measurements at other the MCMA sites

The coefficients determined for the CO-O<sub>3</sub> tracer pair were applied to the simultaneous measurements of HCHO, CO and O<sub>3</sub> sampled on board the ARI mobile laboratory at CENICA, and three additional fixed sites, in Merced, Pedregal and Santa Ana to assess the spatial distribution of the background HCHO concentration within the MCMA. The approach in this analysis was to use the global  $\beta_1$  and  $\beta_2$ coefficient (approach 1, linear model) determined from the open path measurements at CENICA to subtract primary and secondary HCHO from the ambient HCHO data collected by ARI mobile laboratory. The background HCHO is defined as the difference between the measured HCHO and the inferred contributions from direct (determined by measured CO) and photochemical (determined by measured  $O_3$ ) sources was calculated as:

$$[\text{HCHO}]^{\text{background}} = [\text{HCHO}]^{\text{measured}} - \left(0.063 \, [\text{O}_3]^{\text{measured}} + 0.0028 \, [\text{CO}]^{\text{measured}}\right) \tag{8}$$

The background HCHO values, determined via this analysis are depicted in Fig. 8 for several sites throughout the city. The red triangles represent fixed sites where a full diurnal pattern of HCHO, O<sub>3</sub> and CO was available for analysis. The other black circles, in Fig. 8, represent selected mobile measurements where the ambient concentrations were not grossly influenced by primary emissions. Each of the points includes approximately 30 min of data, the resulting background HCHO concentration is portrayed by the size of the data point. This analysis approach must be caveated with a couple of remarks. First, the site-to-site variability of the  $\beta$  parameters is assumed to be nil. This would be problematic if the HCHO/CO emission ratio for traffic typical of the sampling footprint at CENICA was very different from that at the other fixed sites. Second, the sampled airmasses in the mobile laboratory are often influenced by very "fresh" emissions. These emissions are not in photochemical equilibrium to the same extent that the open path CENICA airmass is which determined the  $\beta$  parameters used. Finally,

the focus of this paper is the treatment of the superior glyoxal – carbon monoxide tracer pair, but because a glyoxal measurement was not available on the mobile laboratory, the  $O_3$ -CO pair was used in order to examine the potential spatial variability in the parameter interpreted as "background" formaldehyde.

The first observation is that the CENICA openpath HCHO background concentration (between 1.5 and 2.4 ppbv) for the whole campaign, agrees within error limits with the application of the CO-O<sub>3</sub> tracer data (3.7 ppbv) to the mobile laboratory data at CENICA. The data from the other sites indicate there may be a decrease in the background HCHO with increasing distance from the city center. For example at Santa Ana, located well to the south of the city, the background HCHO drops to about 1 ppbv.

The pattern in the mobile data is not as clear. The difficulty in applying the CO-O<sub>3</sub> tracer pair to mobile data lies with the direct emissions of NO from mobile sources. The influence of primary NO on NO<sub>x</sub> partitioning and O<sub>3</sub> pushes the system away from the presumed photochemical equilibrium. Typically, the presence of any automobile or truck traffic within 50 m upwind influences the CO and NOx concentrations measured by the mobile laboratory. As a result, most of the on-road data has not been included in this analysis.

The spatial difference in background HCHO in MCMA suggests the presence of direct HCHO emissions within the city that do not have a concomitant CO emission. The additional HCHO could be coming directly from certain adhesives used in wood products, insect and fungus control products, corrosion inhibitors in metal processing facilities as well as numerous other industrial applications, and may increase the value of background HCHO inside the city relative to the Santa Ana site.

# 3.3 Comparison of primary vs. secondary HCHO from the literature

Previous studies employed a range of techniques to separate primary and secondary formation of HCHO (Li et al., 1997) employed principal component analysis, and found the portion of primary HCHO about 20% during summer near Vancouver, Canada. Possanzini et al. (2002) compared the ratio of HCHO/toluene in fresh emissions and ambient air in Rome, Italy, and found that secondary HCHO accounted for 80–90% in the summer, while this portion fell below 35% in the winter. Friedfeld et al. (2002) used the O<sub>3</sub>-CO tracer pair, and found secondary HCHO to account for roughly 63% of ambient HCHO during June 2000 in Houston, TX. Jiménez et al. (2005) used the O<sub>3</sub>-NO<sub>x</sub> tracer pair to separate secondary and primary HCHO and find on average 80% secondary HCHO, consistent with a photochemical model during summer in Grenoble, France. None of these studies address the issue of residual HCHO in their analysis, which is expected highest during times of low photochemical activity. In this study, despite active photochemistry and the improve-



**Fig. 8.** Apparent background HCHO using the CO and  $O_3$  correlations. The circular data points represent "mobile data" periods where sampling criteria indicated a "background" ambient HCHO estimate could be attempted. The larger diamonds represent stationary site data. See text for additional details.

ments in the tracer pair used, residual HCHO accounts for a non-negligible 20% of ambient HCHO. Even if all this portion was counted as secondary HCHO (in sum 58%), the portion of primary HCHO (42%) in Mexico City is higher than in any previous study conducted during a time of year when photochemistry was comparable and active.

## 4 Conclusions

A statistical regression model has been employed to separate ambient HCHO concentrations into contributions from mobile emission sources (primary HCHO) and the airborne source from volatile organic compound (VOC) photochemistry (secondary HCHO). Two sets of tracer molecules were employed, and their performance is compared. In both cases CO was used as an indicator for primary HCHO. As indicator for secondary HCHO we used O<sub>3</sub> in once case, and in the other case employ a novel indicator molecule, glyoxal (CHOCHO, the simplest  $\alpha$ -dicarbonyl), for which direct time-resolved measurements have recently been demonstrated (Volkamer et al., 2005b). Analyses were performed for three separate time-regimes: (1) before, (2) during, and (3) after Easter week.

The CO-CHOCHO tracer pair is found to better represent ambient HCHO concentrations than the CO-O<sub>3</sub> tracer pair. This is attributed to the efficient titration reaction of O<sub>3</sub> with NO from traffic emissions, as well as the rather indirect link of O<sub>3</sub> and secondary HCHO forming VOC chemistry, which is highly non-linear. In general, the CO-CHOCHO tracer pair is found to explains a larger portion of ambient HCHO (up to 90%). Also, this tracer pair explains ambient HCHO best by using a rather intuitive "linear" model, which represents ambient HCHO as a linear combination of CO and CHO-CHO concentrations. The amount of HCHO not accounted for by the CO-CHOCHO tracer pair (unaccounted HCHO) tends to be lower than for the CO-O<sub>3</sub> tracer pair, and only for the former pair of tracers is "background HCHO" in general agreement with measured concentrations of HCHO outside the MCMA.

Reduced traffic intensity during Easter week results in a reduction in the percent contributions of primary HCHO; it affects the percent contribution of secondary HCHO to a lesser extent. In all cases studied, diurnal variations of the percent-contributions of primary HCHO peak before sunrise, when primary HCHO accounts for up to 80% of the source of ambient HCHO. Shortly after sunrise, secondary HCHO sources rapidly gain importance, and become comparable to primary sources within few hours. Secondary HCHO sources are the largest daytime HCHO source for ambient HCHO, accounting for up to 80% of ambient HCHO sources before and around solar noon (1:45 pm). In the later afternoon (3:00 to 5:00 pm) primary sources at night.

Direct emissions from not combustion related sources (e.g. from industrial processing) may not be correlated with the indicator molecules used in this study, and may be in part responsible for some of the up to 21% of unaccounted HCHO. These unaccounted HCHO sources are responsible for about a 2 ppbv HCHO "background" concentration throughout the MCMA. However, analyses indicate this unaccounted HCHO may be at least in part, due to unaccounted secondary HCHO (e.g. due to VOC photochemistry that forms HCHO but not CHOCHO). This derived contribution of secondary HCHO is a lower limit, and the percentage of unaccounted HCHO sets an upper limit to the relative importance of industrial emission sources of HCHO in the MCMA. On a 24 h average basis, for the contributions to ambient HCHO from primary, secondary and unaccounted sources were 42%, >38% and <21%, respectively.

A very important portion of the ambient HCHO measured during the MCMA-2003 campaign is related to mobile source emissions. Positive feed-backs from reductions in primary HCHO sources to yield reductions also in secondary HCHO sources are expected, and deserve further investigation.

Acknowledgements. This research was supported by funding from the Mexican Metropolitan Environmental Commission, the US National Science Foundation (ATM- 308748) and the Alliance for Global Sustainability. R. V. acknowledges a postdoctoral fellowship by the Henry and Camille Dreyfus Foundation. We are grateful to V. H. Paramo and R. Ramos for their help in obtaining traffic count data and to Enrique Gonzalez-Villaseñor from Dirección de Operación y Tránsito e Ingeniería (Secretaria de Seguridad Publica) and INMER SA de CV magnetic traffic counters operator. We thank C. Pastor for help with the GIS work and W. Brune from OH data. Edited by: U. Pöschl

## References

- Altshuller, A. P.: Production of aldehydes as primary emissions and from secondary atmospheric reactions of alkenes during the night and early morning hours, Atmos. Environ. 27A, 21–32, 1993.
- Anderson, L. G., Janning, J. A., Barrell, R., Miyagishima, J., Jones, R. H., and Wolfe, P.: Sources and sinks of formaldehyde and acetaldehyde: an analysis of Denvers ambient concentration data, Atmos. Environ., 30(12), 2114–2123, 1996.
- Atkinson, R.: Atmospheric chemistry of VOCs and NOx, Atmos. Environ., 34, 2063–2101, 2000.
- Box, G., Jenkings, G., and Reinsel, G.: Time Series Analysis: Forecasting and Control, Pretince-Hall, Englewood Cliffs, NJ, 1994.
- Canagaratna, M. R., Jayne, J. T., Ghertner, A., Herndon, S. C., Shi, Q., Jimenez, J. L., Silva, P., Williams, P., Lanni, T., Drewnick, F., Demerjian, K. L., Kolb, C. E., and Worsnop, D. R.: Chase Studies of Particulate Emissions from in-use New York City Vehicles, Aerosol Sci. Technol, 38, 555–573, 2004.
- Carlier, P., Hannachi, H., and Mouvier, G.: The chemistry of carbonyl compounds in the atmosphere, Atmos. Environ., 20, 2079– 2099, 1986.
- Carter, W. P. L.: Computer Modeling of Environmental Chamber Measurements of Maximum Incremental Reactivities of Volatile Organic Compounds, Atmos. Environ., 29, 2513–2527, 1995.
- Dean J.: Lange's handbook of chemistry, 13th ed. New York, NY, McGraw-Hill, 1985.
- Dodge, M. C.: Formaldehyde production in photochemical smog as predicted by three state-of-the-science chemical oxidant mechanisms, J. Geophys. Res., 95, 3635–3648, 1990.
- Dunlea, E. J., Herndon, S. C., Nelson, D. D., Volkamer, R. M., Lamb, B. K., Allwine, E. J., Grutter, M., Ramos Villegas, C. R., Marquez, C., Blanco, S., Cardenas, B., Kolb, C. E., Molina, L. T., and Molina, M. J.: Technical note: Evaluation of standard ultraviolet absorption ozone monitors in a polluted urban environment, Atmos. Chem. Phys., 6, 3163–3180, 2006, http://www.atmos-chem-phys.net/6/3163/2006/.
- Finlayson-Pitts, B. J. and Pitts, J. N.: Atmospheric Chemistry, Wiley, New York, 1986.
- Friedfeld, S., Fraser, M., Ensor, K., et al.: Statistical analysis of primary and secondary atmopsheric formaldehyde, Atmos. Environ., 36, 477–4775, 2002.
- Granby, K., Christensen, C. S., and Lohse, C.: Urban and semi-rural observations of carboxylic acids and carbonyls, Atmos. Environ., 35, 2096–2106, 1997.
- Green, W. H.: Econometric Analysis, 3rd Ed. Pretince Hall Inc., 1998.
- Grutter, M., Flores, E., Andraca-Ayala, G., and Baez, A.: Formaldehyde levels in downtown Mexico City during 2003, Atmos. Environ., 39, 1027–1034, 2005.
- Herndon, S. C., Shorter, J. H., Zahniser, M. S., Nelson, D. D. J., Jayne, J. T., Brown, R. C., Miake-Lye, R. C., Waitz, I. A., Silva, P., Lanni, T., Demerjian, K. L., and Kolb, C. E.: NO and NO<sub>2</sub> Emissions Ratios Measured from in use Commercial Aircraft during Taxi and Take-Off, Environ. Sci. Technol., 38, 6078– 6084, 2004.
- Herndon, S. C., Yongquan, L., Nelson, D. D., and Zahniser, M. S.: Determination of line strengths for selected transitions in the

 $v_2$  band relative to the U<sub>1</sub> and N<sub>2</sub> bands of H<sub>2</sub>CO, J. Quant. Spectros. Rad. Transfer, 90, 207–216, 2005.

- Hoekman, S. K.: Speciated measurements and calculated reactivities of vehicle exhaust emissions from conventional and reformulated gasolines, Envir. Sci. Technol., 26, 1206–1216, 1992.
- Kean, A. J., Grosjean, E., Grosjean, D., and Harley, R. A.: On-Road Measurement of Carbonyls in California Light-Duty Vehicle Emissions, Environ. Sci. Technol., 35, 4198–4204, 2001.
- Kolb, C. E., Herndon, S. C., McManus, J. B., Shorter, J. H., Zahniser, M. S., Nelson, D. D., Jayne, J. T., Canagaratna, M. R., and Worsnop, D. R.: Mobile Laboratory with Rapid Response Instruments for Real-time Measurements of Urban and Regional Trace Gas and Particulate Distributions and Emission Source Characteristics, Environ. Sci. Technol., 38, 5694–5703, 2004.
- Li, S. M., Anlauf, K. G., Wiebe, H. A., Bottenheim, J. W., Shepson, P. B., and Biesenthal, T.: Emission ratios and photochemical production efficiencies of nitrogen oxides, ketones, and aldehydes in the Lower Fraser Valley during the summer Pacific 1993 oxidant study, Atmos. Environ., 31, 2037–2048, 1997.
- Molina, L. T. and Molina, M. J.: Air Quality in the Mexico Megacity, Kluwer Academia Publishers, Netherlands, 2002.
- Müller, K.: Determination of aldehydes and ketones in the atmosphere. A comparative long-time study at an urban and a rural site in Eastern Germany, Chemosphere, 35, 2096–2106, 1997.
- Platt, U.: in: Monitoring by Spectroscopic Techniques, Differential Optical Absorption Spectroscopy (DOAS) edited by: Sigrist, M. W., Wiley & Sons, New York, chap. 2, 1994.
- Possanzini, M., Di Palo, V., and Cecinato, A.: Sources and photodecomposition of formaldehyde and acetaldehyde in Rome ambient air, Atmos. Environ., 36, 3195–3201, 2002.
- Reyes, F., Grutter, M., Jazcilevich, A., González-Oropeza, R.: Analysis of non-regulated vehicular emissions by extractive FTIR spectrometry: tests on a hybrid car in Mexico City, Atmos. Chem. Phys. Discuss., 6, 5773–5796, 2006,
- http://www.atmos-chem-phys-discuss.net/6/5773/2006/.
- Rothman, L. S., Barbe, A., Benner, D. C., Brown, L. R., et al: The HITRAN molecular spectroscopic database: Edition of 2000 including updates through 2001, Journal of Quantitative Spectroscopy and Radiative Transfer, 82(1–4), 5–44, 2003.
- Sander, S. P., Friedl, R. R., Ravishankara, A. R., Golden, D. M., Kolb, C. E., Kurylo, M. J., Huke, R. E., Orkin, V. L., Molina, M. J., Moortgat, G. K., and Finlayson-Pitts, B.: Chemical Kinetics and Photochemical Data for Use in Atmospheric Studies, JPL Publication 02-25, Evaluation Number 14, http://jpldataeval.jpl. nasa.gov/, 2002.

- Satsumabayashi, H., Kurita, H., Chang, Y.-S., Carmichael, G. R., and Ueda, H..: Photochemical formations of lower aldehydes and lower 15 fatty acids under long-range transport in central Japan, Atmos. Environ., 29, 255–266, 1995.
- Secretaria del Medio Ambiente, S. M. A.: Inventario de Emisiones. Mexico DF, http://www.sma.df.gob.mx/sma/modules. php?name=Biblioteca&dop=getit&lid=374, 2002 (2002).
- Seinfeld, J. H.: Atmospheric Chemistry and Physics of Air Pollution, John Wiley & Sons Inc., pp. 768, 1986.
- Shirley, T. R., Brune W. H., Ren, X., Mao, J., Lesher, R., Cardenas, B., Volkamer, R., Molina, L. T., Molina, M. J., Lamb, B., Velasco, E., Jobson, T., and Alexander, M.: Atmospheric oxidation in the Mexico City Metropolitan Area (MCMA) during April 2003, Atmos. Chem. Phys., 6, 2753–2765, 2006, http://www.atmos-chem-phys.net/6/2753/2006/.
- Shorter, J. H., Herndon, S., Zahniser, M. S., Nelson, D. D., Wormhoudt, J., Demerjian, K. L., and Kolb, C. E.: Real-time measurements of nitrogen oxide emissions from in-use New York City transit buses using a chase vehicle Environ. Sci. Technol., 39(20), 7991–8000, 2005.
- Vaughn, T. L., Strader, C., Davis, S., and Daling, J. R.: Formaldehyde and cancers of the pharynx, sinus and nasal cavity: I. Occupational exposures, Int. J. Cancer, 38, 677–683, 1986.
- Viskari, E. L., Vartiainen, M., and Pasanen, P.: Seasonal and diurnal variation in formaldehyde and acetaldehyde concentrations along a highway in Eastern Finland, Atmos. Environ., 34(6), 917–923, 2000.
- Volkamer, R., Molina, L. T., Molina, M. J., Shirley, T., and Brune, W. H.: DOAS measurement of glyoxal as an indicator for fast VOC chemistry in urban air, Geophys. Res. Lett., 32, L08806, doi:10.1029/2005GL022616, 2005b.
- Williams, R. L., Lipari, F., and Potter, R. A.: Formaldehyde, methanol and hydrocarbon emissions from methanol-fueled cars, J. Air Waste Manage. Assoc., 40, 747–756, 1990.
- Zavala, M., Herndon, S. C., Slott, R. S., Dunlea, E. J., Marr, L. C., Shorter, J. H., Zahniser, M., Knighton, W. B., Rogers, T. M., Kolb, C. E., Molina, L. T., and Molina, M. J.: Characterization of on-road vehicle emissions in the Mexico City Metropolitan Area using a mobile laboratory in chase and fleet average measurement modes during the MCMA-2003 field campaign, Atmos. Chem. Phys. Discuss., 6, 4689–4725, 2006,

http://www.atmos-chem-phys-discuss.net/6/4689/2006/.

Zhou, X., Mopper, K., Yin-Nan, Lee, Newman, L., and Xiaohur, Chen: Tropospheric formaldehyde concentration at the Mauna Loa Observatory during the Mauna Loa Observatory Photochemistry Experiment 2, J. Geophys. Res., 101(D9), 14711–14719, 1996.