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Chemical characteristics and environmental drivers of nitrogen-containing organic aerosol formation in coastal and inland urban atmospheres in Myanmar

Ning Zhang et al.

Correspondence to: Junfeng Wang (wangjunfeng@nuist.edu.cn)

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S1. Setting for UHPLC separation

A Waters Acquity HSS T3 column (1.8 μm, 100 × 2.1 mm) was used to fulfill the separation with mixed mobile phase of (A) 0.1% formic acid solution and (B) 0.1% formic acid and acetonitrile. The gradient elution program was as follows: it started with 99% of A and 1% of B for 2 minutes, followed by a linear gradient to 100% of B over the next 11 minutes, continued with 100% of B for 2 minutes, then returned to 1% of B in 0.1 minutes, and maintained 1% of B for 6.9 minutes. The flow rate of the A and B mixtures was 300 μL min-1 (Wang et al., 2017).

S2. UHPLC-Orbitrap MS data analysis

Non-target analysis was performed using the software MZmine 2.39 to analyze the UHPLC-Orbitrap MS data. This software provides essential functions for MS data processing, including raw data import, peak detection, shoulder peak filtering, chromatogram deconvolution, and peak identification (Hu et al., 2016; Katajamaa et al., 2006; Pluskal et al., 2010). The detailed processing steps and parameter settings can be found in the previously published articles (Fuller et al., 2012; Kind and Fiehn, 2007; Lin et al., 2012a; Lin et al., 2012b; Wang et al., 2016). The determined molecular formulas are expressed as CxHyOzNmSn, where x, y, z, m, and n correspond to the number of C, H, O, N, and S atoms in the molecular formula, respectively. Molecular formula determination is primarily based on the detected m/z values, with a tolerance of 2 ppm in both ESI+/- modes. The detected compounds in the ambient samples were compared with those in the blank samples, and only the compounds with a sample/blank peak area ratio of $\geqslant$10 were retained.

To obtain more reasonable molecular formulas, the following specific rules were applied (Koch and Dittmar, 2006): (1) Atom counts: 1-40 for 12C, 1-100 for 1H, 0-40 for 16O, 0-5 for 14N, 0-2 for 32S; (2) Element ratios: in ESI- mode, 0.3-3.0 H/C, 0-3 O/C, 0-0.5 N/C, 0-2.0 S/C; in ESI+ mode, 0.3-3.0 H/C, 0-1.2 O/C, 0-1.0 N/C, 0-0.8 S/C; (3) Double-Bond Equivalent (DBE value, calculated as 1+(2C-H+N)/2): ranges from 0 to 25.

S3. Semi-quantification of the detected organic compounds

In this study, the peak areas of detected organic compounds were semi-quantified

using 3-nitrophenol (3-NP) as an external standard, with estimated concentrations reported in ng m^{-3}. It is important to note that different organic compounds exhibit varying ionization efficiencies in mass spectrometry, introducing inherent uncertainties when comparing peak areas across different molecular species. To enable relative comparisons between samples, we assumed a uniform mass spectrometric response factor for all detected compounds.

For calibration, a standard solution of 3-NP (m/z 138) at 1000 ppb in acetonitrile was analyzed in ESI- mode (**Fig. S2**). A linear calibration curve (y = 893,319x, R^2 = 0.999) was established using a dilution series of 3-NP (5, 10, 30, 50, 100, 500, and 1000 ppb; **Fig. S3**).

Fig. S1 The map of Myanmar sampling location

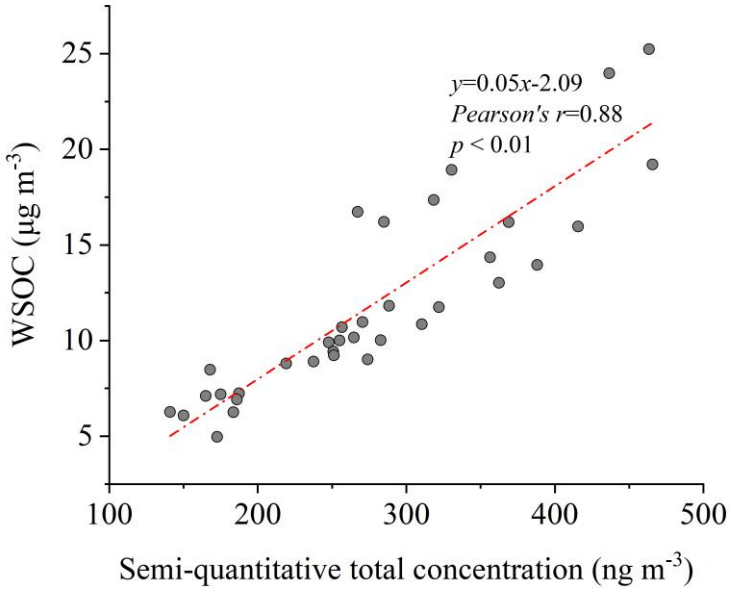

Fig. S2 The correlation between WSOC and semi-quantitative total concentration

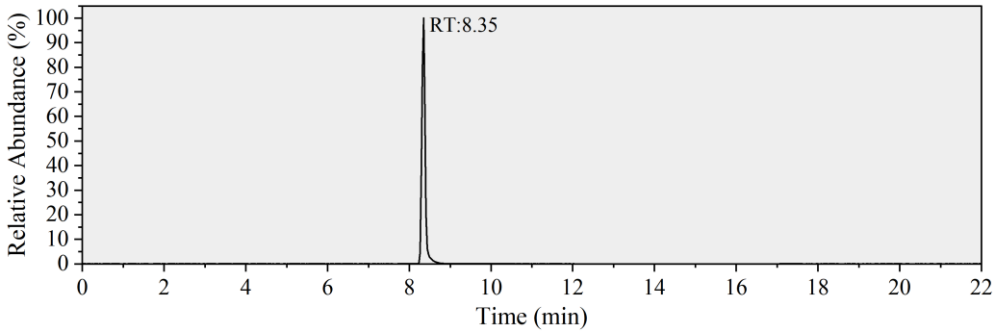

Fig. S3 Chromatographic peaks of 3-NP standard substance in ESI- mode in Myanmar

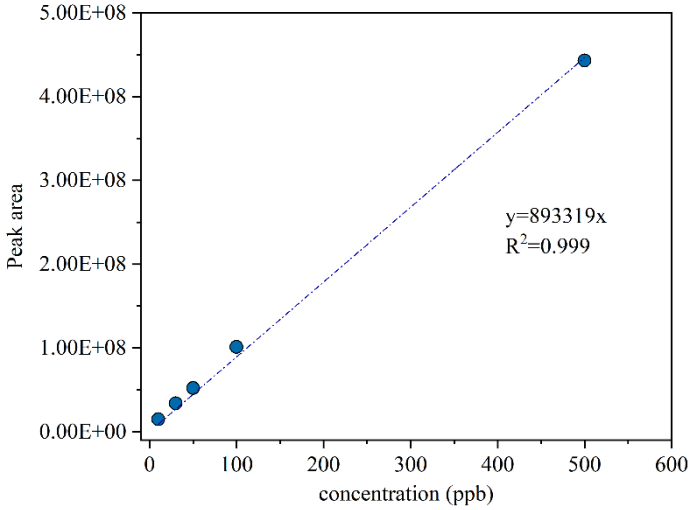

Fig. S4 Calibration curve of 3-NP in acetonitrile

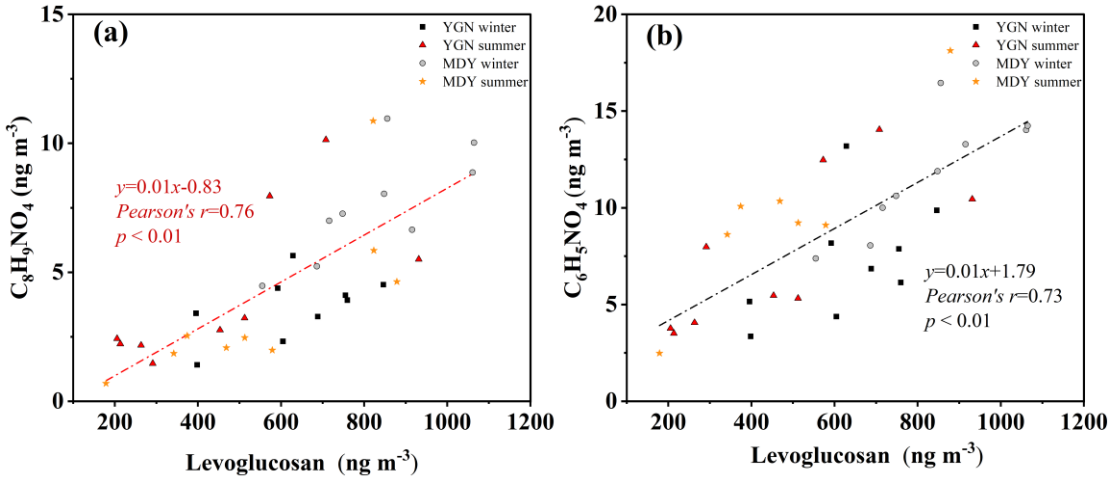

Fig. S5 The correlation between $C_8H_9NO_4$ (a), $C_6H_5NO_4$ (b) and levoglucosan. Color bar is the range of RH.

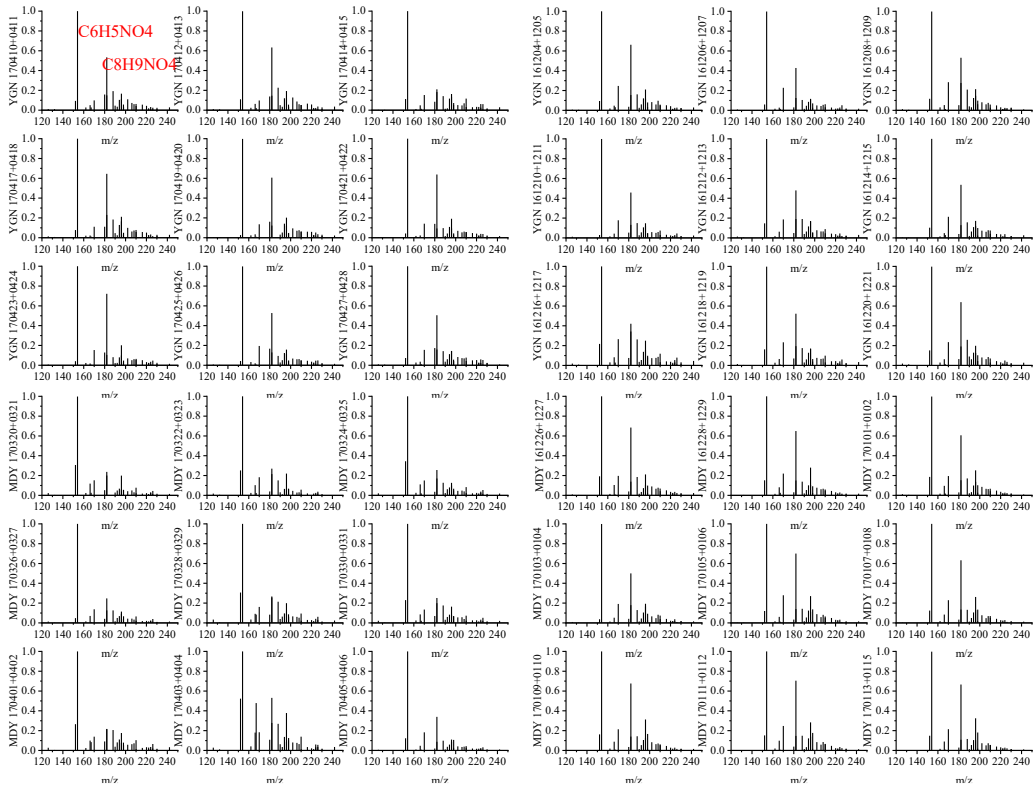

Fig. S6 Mass spectra reconstructions of CHON same peaks from extracted ion chromatograms in ESI- mode. The vertical axis represents the semi-quantitative normalized concentration for each compound.

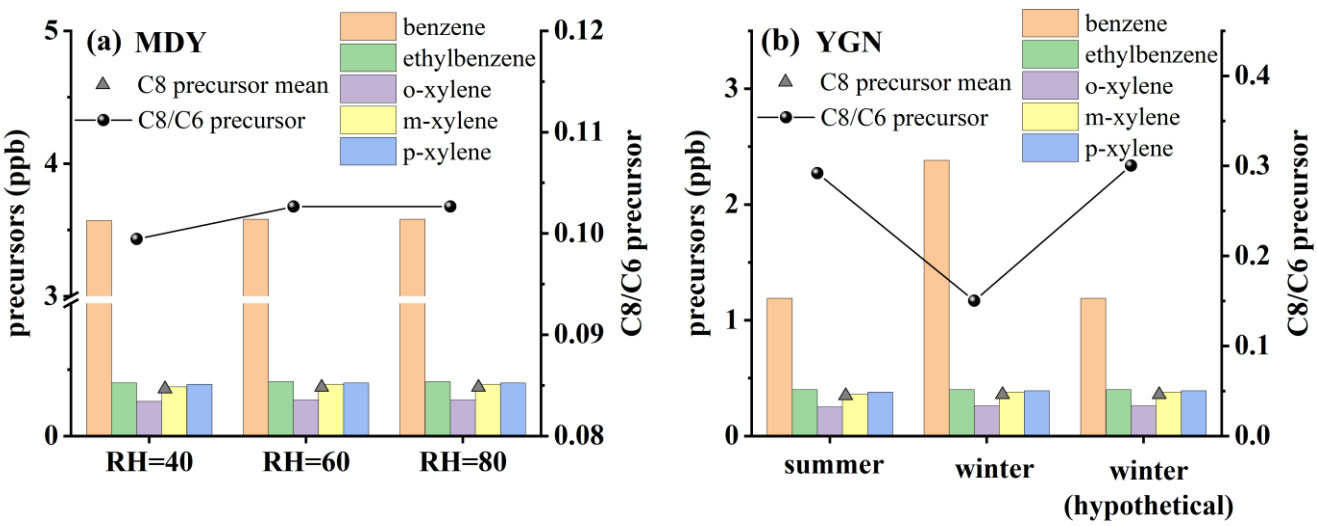

Fig. S7 Distributions of simulated precursor mass concentrations for $C_8H_9NO_4$ and $C_6H_5NO_4$ under different RH conditions. In the model, the precursors of $C_8H_9NO_4$ include ethylbenzene, o-xylene, m-xylene, and p-xylene, whereas benzene is the sole precursor of $C_6H_5NO_4$. Triangles denote the average precursor concentration of $C_8H_9NO_4$.

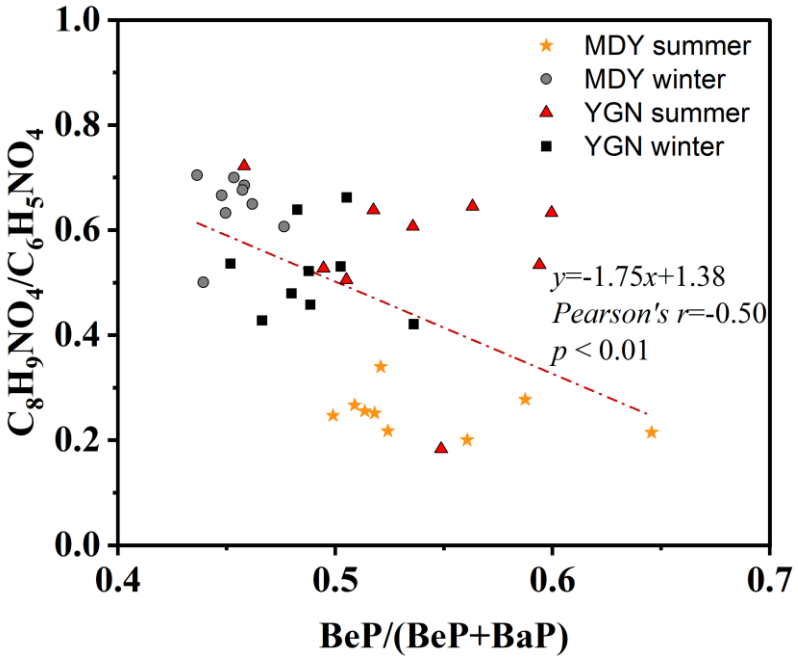

Fig. S8 The correlation between BeP/(BeP+BaP) and $C_8H_9NO_4/C_6H_5NO_4$.

Table S1 The average number, peak area and corresponding compound species of different types of molecular formulas in Myanmar

Ion mode	subgroup	number	Percentage (%)	Mass concentration (ng m^{-3})	Mass concentration Percentage (%)
ESI-	CHO-	520	54.4	164.3	58.8
	CHON-	166	16.9	62.3	22.3
	CHOS-	195	19.6	43.8	15.7
	CHONS-	73	7.0	6.8	2.4
	CHNS-	19	1.9	2.3	0.8
	CHN-	2	0.2	0.2	0.1

Table S2 Summary of the number, concentration, and ratio parameters of compounds in different subgroups of $PM_{2.5}$ in Yangon and Mandalay during the winter and summer seasons.

Sample ID	Subgroup	Number of compounds	Mass concentration (ng m^{-3})
YGN-Summer	Total	772	196.8
	CHO	473	127.4
	CHON	139	44.2
	CHOS	117	20.9
	CHONS	31	2.2
	CHN	1	0.1
	CHNS	11	2.0
YGN-Winter	Total	938	241.4
	CHO	507	148.5
	CHON	154	43.7
	CHOS	191	41.6
	CHONS	67	5.7
	CHN	1	0.0
	CHNS	18	1.8
MDY-Summer	Total	1064	342.7
	CHO	558	197.2
	CHON	198	84.3
	CHOS	201	49.1
	CHONS	83	9.4
	CHN	2	0.1
	CHNS	22	2.5
MDY-Winter	Total	1125	337.8
	CHO	544	184.2
	CHON	172	76.9
	CHOS	269	63.6
	CHONS	109	9.9

Sample ID	Subgroup	Number of compounds	Mass concentration (ng m^{-3})
	CHN	3	0.4
	CHNS	27	2.7

Reference:

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