



Supplement of

Measurement report: Variations and environmental impacts of atmospheric N_2O_5 concentrations in urban Beijing during the 2022 Winter Olympics

Tiantian Zhang et al.

Correspondence to: Weili Lin (linwl@muc.edu.cn) and Chunxiang Ye (c.ye@pku.edu.cn)

The copyright of individual parts of the supplement might differ from the article licence.

Sect. S1 Technical details of CRDS

Sect. S1.1 Determination of the Limit of Detection (LOD)

The LOD of the CRDS analyzer for N₂O₅ was determined via a 4-hour zero-air injection experiment: zero air (free of target analytes) was continuously introduced into the instrument cavity (1 s time resolution), and Allan variance analysis was applied to evaluate system stability (Figure S1). The LOD was calculated using Eq. (S1) below, yielding a value of 2.9 pptv.

$$[A] = \frac{R_L}{c\sigma} \left(\frac{1}{\tau} - \frac{1}{\tau_0} \right) \quad (\text{S1})$$

where [A] denotes the analyte concentration, " R_L " is the ratio of the cavity length to the length over which the absorber is present, c is the speed of light, σ is the absorption cross-section, τ is the cavity ring-down time with sample gas, and τ_0 is the background ring-down time with zero air. During the zero-air experiment, the instrument background level (τ_0) was measured as 50 μ s.

Sect. S1.2 Calibration

To ensure measurement accuracy, we performed regular calibrations using stable, calibrated NO₃ and N₂O₅ standard sources (generated via a dynamic standard gas system, NO₃ generated by the reaction of NO₂ and O₃, N₂O₅ generated by the equilibrium of NO₃ and NO₂, as referenced in our prior work: Zhang et al., 2026). These calibrations quantified two critical correction factors addressing sampling losses

- Tubing loss: Measured at $11.4 \pm 1.3\%$.
- Filter membrane loss: Quantified as $4.5 \pm 0.5\%$

The final concentrations of [NO₃ + N₂O₅] and derived NO₃ were obtained by dividing raw instrument measurements by the sum of these loss ratios (i.e., correcting for signal reduction during sample transport).

Sect. S1.3 Uncertainty

- Parameter uncertainties in concentration calculation: The absorption cross-section (σ) and effective absorption cavity length—key parameters in Eq. (1)—contribute 13% and 4% uncertainty, respectively.

- Data correction uncertainties: Corrections for membrane loss ($4.5 \pm 0.5\%$) and pipeline loss ($11.4 \pm 1.3\%$) during sample transport introduce an additional 1.4% uncertainty.

The total combined uncertainty was calculated following the method described in our prior work (Zhang et al., 2024), confirming the 13.7% value.

References:

Zhang, T., Zuo, P., Ma, J., Ye, C., Lin, W., and Zhu, T.: Characterization and Application of an Online Measurement System for NO_3 and N_2O_5 Based on Cavity Ring-Down Spectroscopy, *Acta Sci. Nat. Univ. Pekin.*, 60, 563–574, <https://doi.org/10.13209/j.0479-8023.2024.030>, 2024

Zhang, T., Ma, J., Liu, T., Lin, W., Zuo, P., and Ye, C.: A dynamic generation system for NO_3 and N_2O_5 standard gases, *Environ. Chem.*, 45, 1–7, <https://doi.org/10.7524/j.issn.0254-6108.2024091302>, 2026.

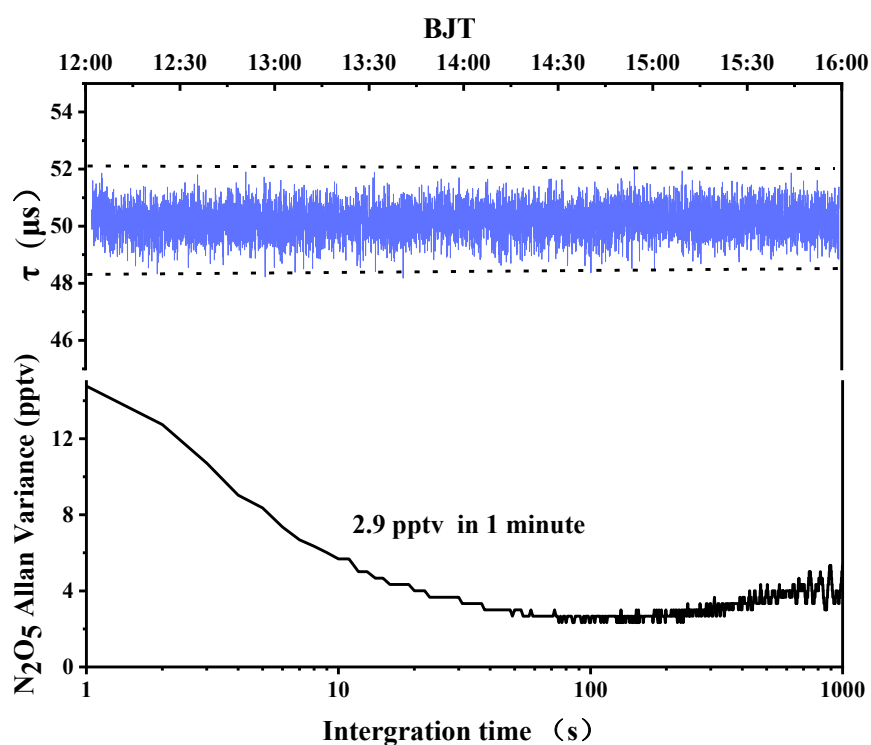


Figure S1. Limit of detection (LOD) and background signal of the instrument (blue line).

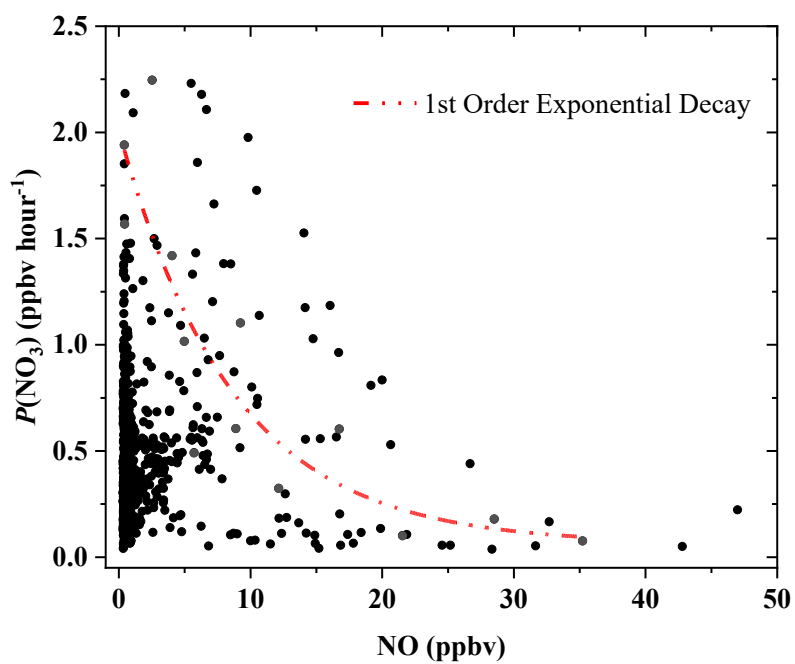


Figure S2. Schematic diagram of the correlation between NO and $P(\text{NO}_3)$.

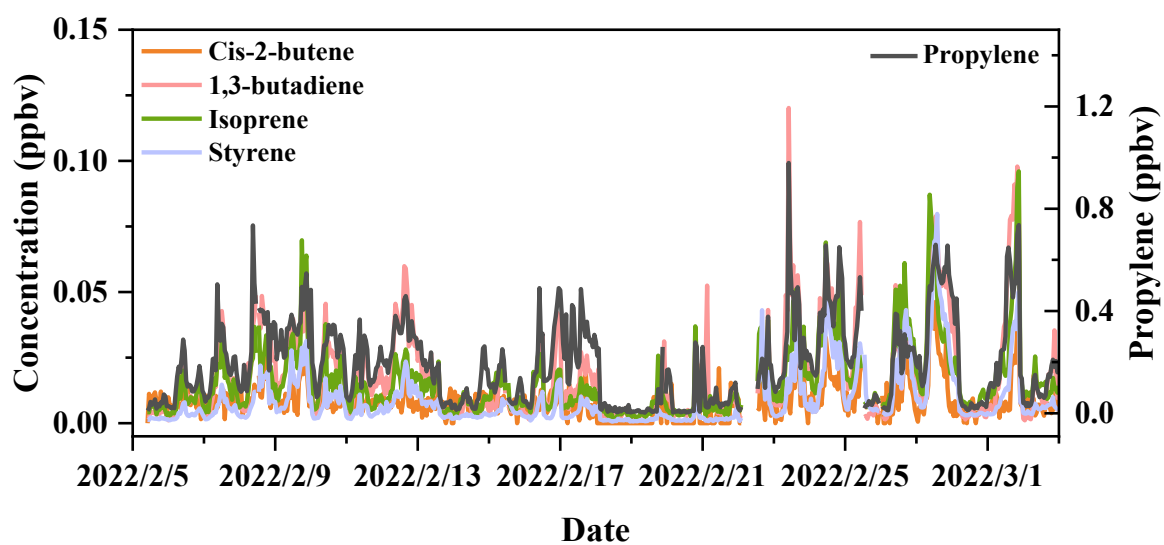


Figure S3. Temporal trends of highly reactive VOC concentrations during the observation period.

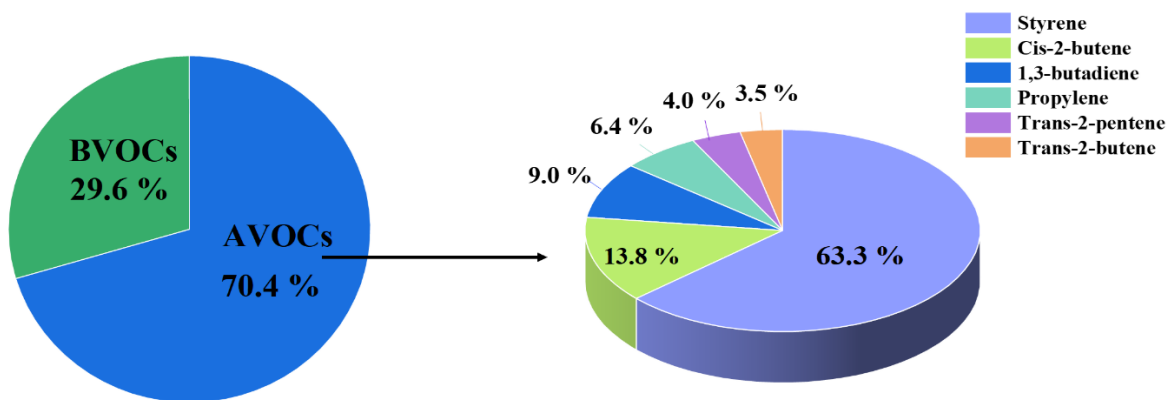


Figure S4. The proportion of reaction activity of NO₃ with different VOCs.

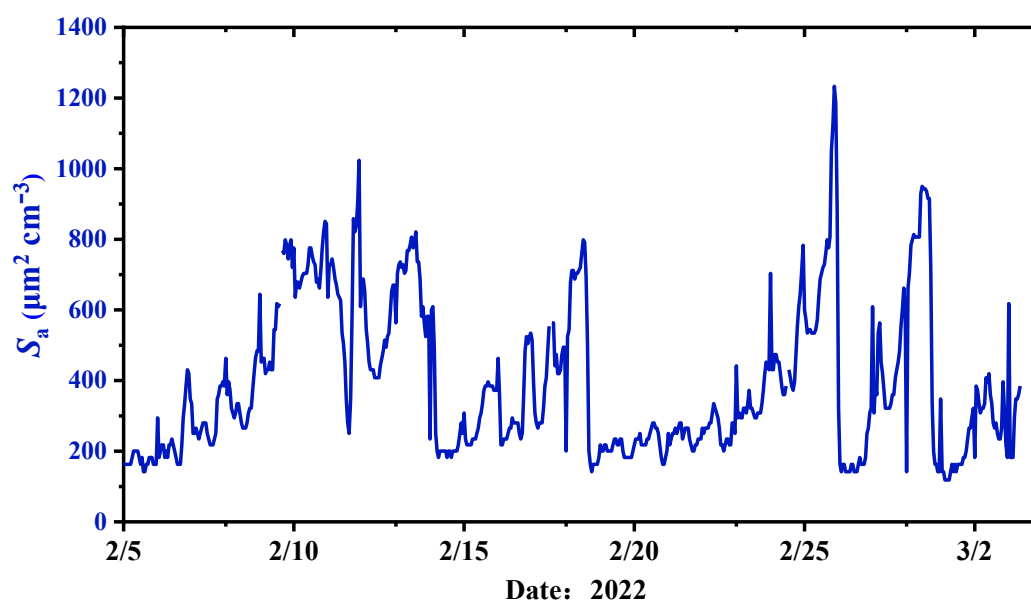


Figure S5. Time-series variations in 1h-mean S_a .

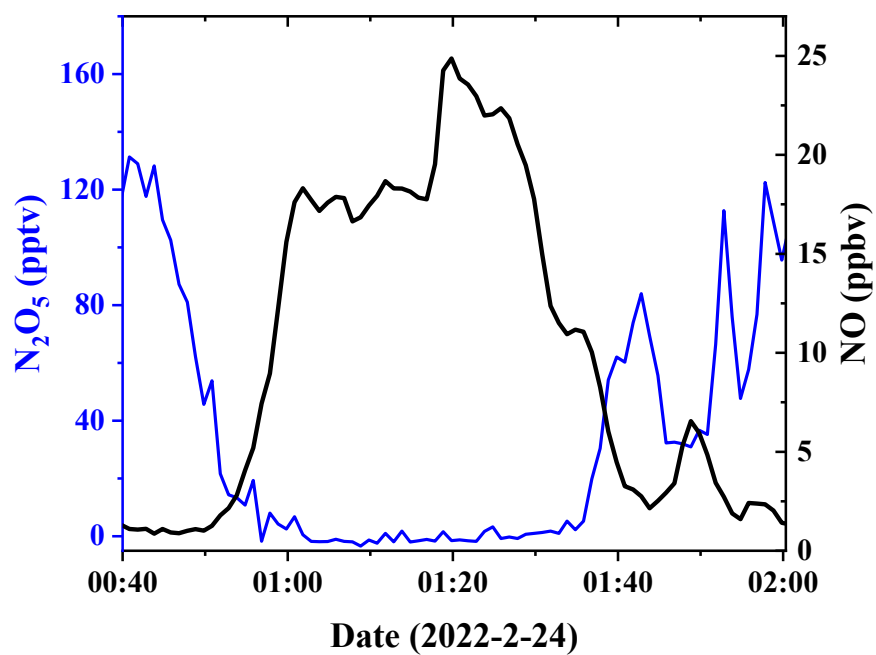


Figure S6. Variations in NO and N₂O₅ mixing ratios on Feb. 24, 2022.

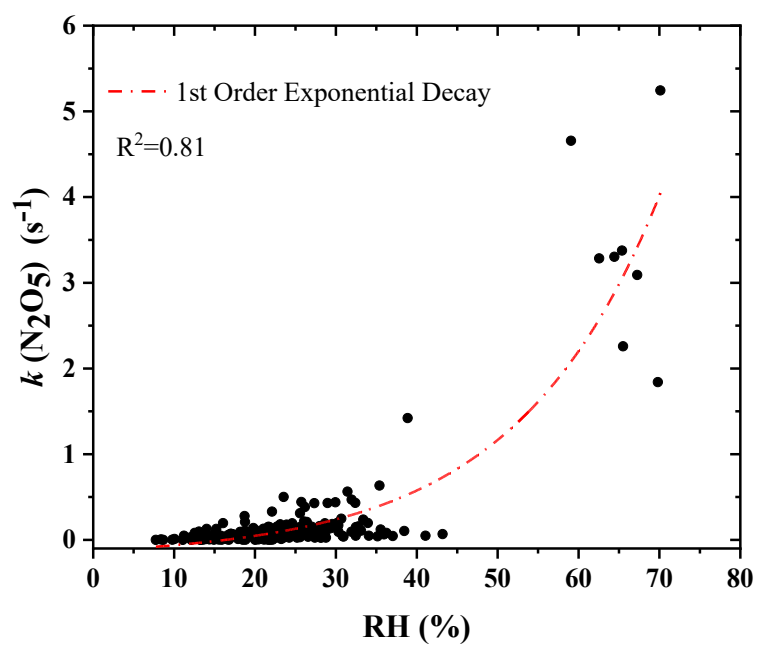


Figure S7. Schematic diagram of the correlation between RH and $k_{\text{N}_2\text{O}_5}$.

55 **Table S1. Reaction rate coefficients of VOCs with respect to NO₃.**

VOCs	OGP	POP	<i>k</i> (298K)
	Concentration (ppbv)	Concentration (ppbv)	(10 ⁻¹⁵ cm ³ molecule ⁻¹ s ⁻¹)
Ethane	3.61±1.55	4.03±2.12	0.01
Propane	1.91±0.97	2.46±1.67	<0.07
Acetylene	1.41±1.04	1.61±1.09	0.21
Acetone	1.27±0.43	1.55±0.71	<0.03
Ethylene	1.27±1.08	1.44±1.22	0.0049
Styrene	0.006±0.01	0.01±0.01	1500
Cis-2-butene	0.007±0.01	0.01±0.01	352
1,3-butadiene	0.02±0.01	0.02±0.02	100
Propylene	0.18±0.15	0.22±0.19	9.49
Isoprene (BVOC)	0.01±0.01	0.02±0.02	696

57 **Table S2. N₂O₅ uptake coefficients and NO₃ loss rate coefficients from a nightly steady state**
 58 **analysis.**

Start Time-End Time	$\gamma(\text{N}_2\text{O}_5)$	$k(\text{NO}_3)(\text{s}^{-1})$	Start Time-End Time	$\gamma(\text{N}_2\text{O}_5)$	$k(\text{NO}_3)(\text{s}^{-1})$
2/06 0:00 - 2/06 05:40	0.0325	0.0713	2/19 20:40 - 2/20 06:00	0.0222	0.1214
2/06 19:00 - 2/07 05:40	0.0216	0.0999	2/20 19:20 - 2/21 06:20	0.011	0.1395
2/07 19:00 - 2/08 06:40	0.0198	0.0868	2/21 20:20 - 2/22 06:00	0.0285	0.1368
2/08 19:20 - 2/09 05:20	0.0296	0.2163	2/22 19:40 - 2/23 06:40	0.0108	0.1847
2/09 18:40 - 2/10 04:00	0.0042	0.3108	2/23 19:00 - 2/24 00:20	0.0053	0.2358
2/10 19:40 - 2/11 04:40	0.0108	0.0384	2/25 19:00 - 2/26 06:40	0.0101	0.062
2/11 18:40 - 2/12 06:40	0.0155	0.0097	2/26 18:20 - 2/27 02:40	0.024	0.1655
2/13 19:00 - 2/14 07:00	0.2244	0.7001	2/27 18:40 - 2/28 20:20	0.0345	0.2158
2/14 18:40 - 2/15 06:20	0.0162	0.3441	2/28 22:00 - 3/01 06:40	0.1232	0.0111
2/15 19:20 - 2/16 05:40	0.0125	0.1554	3/01 18:20 - 3/02 03:00	0.0395	0.0887
2/16 18:20 - 2/17 06:20	0.0043	0.1996	3/02 18:40 - 3/03 04:40	0.0014	0.1186
2/18 19:20 - 2/19 06:40	0.0402	0.1112			